

1 **Supplement for**

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3 **The Secondary Formation of Organosulfates under the Interactions**  
4 **between Biogenic Emissions and Anthropogenic Pollutants in**  
5 **Summer of Beijing**

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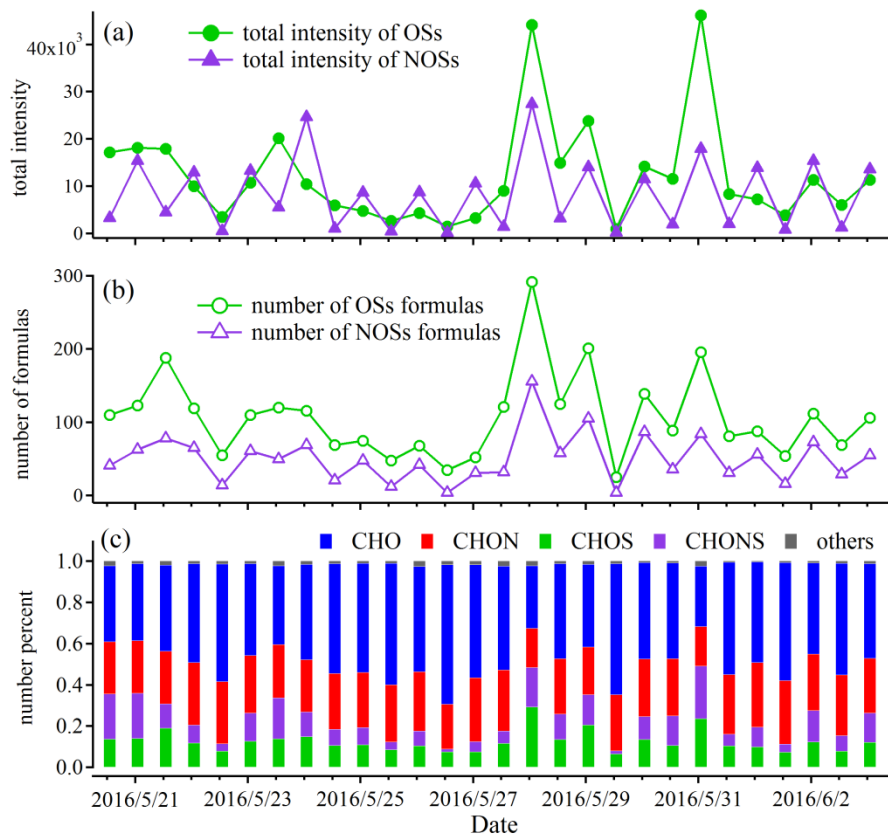
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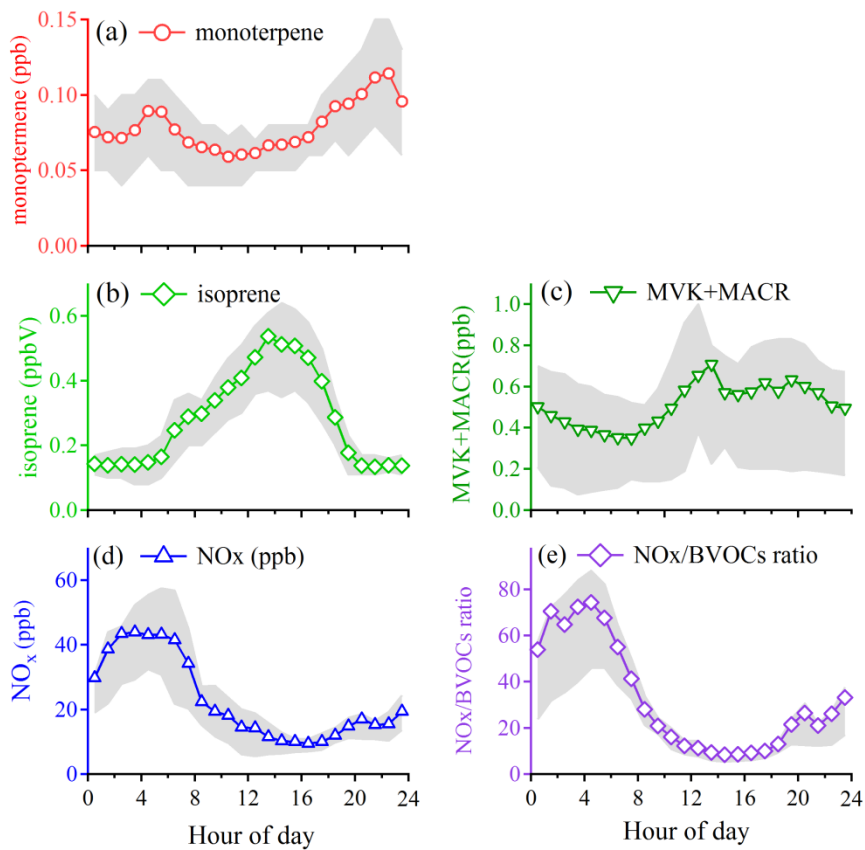
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19 Fig. S1 Temporal variation of the (a) total intensity and (b) total number of OSs and NOSs, and (c)  
 20 temporal variation of the number percent of different compound categories.

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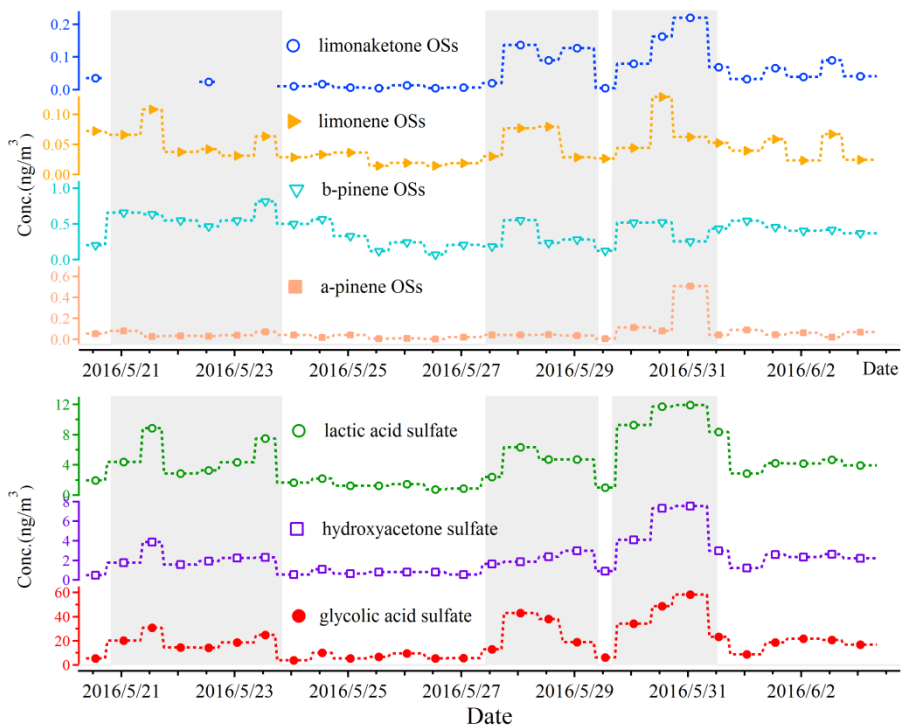
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Fig. S2 The diurnal variations of monoterpane, isoprene, NO<sub>x</sub> and NO<sub>x</sub>/BVOCs ratios



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30 Fig. S3 Time series of  $\alpha$ -pinene OSs,  $\beta$ -pinene OSs, limonene OSs, limonaketone OSs, lactic acid  
 31 sulfate, glycolic acid sulfate and hydroxyacetone sulfate. The pollution episodes were marked by gray  
 32 shadow.

Table S1 List of some most abundant OSs and NOSs peaks observed in the mass spectra

No.	m/z [M-H] <sup>-</sup>	Formula [M-H] <sup>-</sup>	relative intensity (%)			precursor/ formation	references
			05/24N	05/30D	05/30N		
<b>organosulfates</b>							
1	267.05440	C <sub>9</sub> H <sub>15</sub> SO <sub>7</sub> <sup>-</sup>	2.1	19.4	100.0	alkane/ isoprene	(Riva et al., 2016b;Riva et al., 2016a)
2	281.07005	C <sub>10</sub> H <sub>17</sub> SO <sub>7</sub> <sup>-</sup>	2.6	13.3	40.0	diesel fuel SOA	(Blair et al., 2017)
3	253.07513	C <sub>9</sub> H <sub>17</sub> SO <sub>6</sub> <sup>-</sup>	2.0	11.1	38.0	diesel fuel SOA	(Blair et al., 2017)
4	239.05948	C <sub>8</sub> H <sub>15</sub> SO <sub>6</sub> <sup>-</sup>		11.1	29.4	diesel fuel SOA	(Blair et al., 2017)
5	309.10135	C <sub>12</sub> H <sub>21</sub> SO <sub>7</sub> <sup>-</sup>		11.0	25.7	diesel fuel SOA	(Blair et al., 2017)
6	295.08570	C <sub>11</sub> H <sub>19</sub> SO <sub>7</sub> <sup>-</sup>	1.2	9.0	25.6	diesel fuel SOA	(Blair et al., 2017)
7	251.05948	C <sub>9</sub> H <sub>15</sub> SO <sub>6</sub> <sup>-</sup>	3.7	9.5	24.7	monoterpene	(Wang et al., 2017)
8	223.06457	C <sub>8</sub> H <sub>15</sub> SO <sub>5</sub> <sup>-</sup>	3.1	9.7	16.8	diesel fuel SOA	(Blair et al., 2017)
9	279.05440	C <sub>10</sub> H <sub>15</sub> SO <sub>7</sub> <sup>-</sup>	3.9	4.8	15.9	alkane	(Riva et al., 2016b)
10	269.07005	C <sub>9</sub> H <sub>17</sub> SO <sub>7</sub> <sup>-</sup>	0.3	4.0	15.8	alkane	(Riva et al., 2016b)
11	225.04383	C <sub>7</sub> H <sub>13</sub> SO <sub>6</sub> <sup>-</sup>	0.4	2.3	14.0	diesel fuel SOA	(Blair et al., 2017)
12	265.07513	C <sub>10</sub> H <sub>17</sub> SO <sub>6</sub> <sup>-</sup>	3.0	7.3	13.9	alkane	(Riva et al., 2016b)
13	297.06496	C <sub>10</sub> H <sub>17</sub> SO <sub>8</sub> <sup>-</sup>		4.2	13.5	alkane	(Riva et al., 2016b)
14	267.09078	C <sub>10</sub> H <sub>19</sub> SO <sub>6</sub> <sup>-</sup>	1.6	6.0	13.4	diesel fuel SOA	(Blair et al., 2017)
15	279.09078	C <sub>11</sub> H <sub>19</sub> SO <sub>6</sub> <sup>-</sup>	1.9	5.5	13.3	diesel fuel SOA	(Blair et al., 2017)
16	283.04931	C <sub>9</sub> H <sub>15</sub> SO <sub>8</sub> <sup>-</sup>	0.4	3.2	12.3	diesel fuel SOA	(Blair et al., 2017)
17	151.00705	C <sub>4</sub> H <sub>7</sub> SO <sub>4</sub> <sup>-</sup>	1.0	4.8	10.3	diesel fuel SOA	(Blair et al., 2017)
18	325.09626	C <sub>12</sub> H <sub>21</sub> SO <sub>8</sub> <sup>-</sup>			10.2	diesel fuel SOA	(Blair et al., 2017)
19	293.07005	C <sub>11</sub> H <sub>17</sub> SO <sub>7</sub> <sup>-</sup>	0.9		10.1	diesel fuel SOA	(Blair et al., 2017)
20	283.08570	C <sub>10</sub> H <sub>19</sub> SO <sub>7</sub> <sup>-</sup>	0.3	3.7	10.1	diesel fuel SOA	(Blair et al., 2017)
21	237.04383	C <sub>8</sub> H <sub>13</sub> SO <sub>6</sub> <sup>-</sup>	1.4	3.3	10.0	diesel fuel SOA	(Blair et al., 2017)
22	307.08570	C <sub>12</sub> H <sub>19</sub> SO <sub>7</sub> <sup>-</sup>	0.6		9.7		
23	293.10643	C <sub>12</sub> H <sub>21</sub> SO <sub>6</sub> <sup>-</sup>	0.9		9.5		
24	237.08022	C <sub>9</sub> H <sub>17</sub> SO <sub>5</sub> <sup>-</sup>	2.0	5.4	9.4	alkane	(Riva et al., 2016b)
25	351.11191	C <sub>14</sub> H <sub>23</sub> SO <sub>8</sub> <sup>-</sup>			9.4	diesel fuel SOA	(Blair et al., 2017)

26	235.06457	$C_9H_{15}SO_5^-$	12.9	8.7	9.1	diesel fuel SOA	(Blair et al., 2017)
27	373.09626	$C_{16}H_{21}SO_8^-$	5.0	9.5	8.7	diesel fuel SOA	(Blair et al., 2017)
28	281.10643	$C_{11}H_{21}SO_6^-$	0.6	3.6	8.2	diesel fuel SOA	(Blair et al., 2017)
29	339.07553	$C_{12}H_{19}SO_9^-$			8.0	diesel fuel SOA	(Blair et al., 2017)
30	321.17412	$C_{15}H_{29}SO_5^-$	1.6	6.0	7.7	diesel fuel SOA	(Blair et al., 2017)
31	307.15847	$C_{14}H_{27}SO_5^-$	2.1		7.6	diesel fuel SOA	(Blair et al., 2017)
32	365.12756	$C_{15}H_{25}SO_8^-$	0.3		7.6	diesel fuel SOA	(Blair et al., 2017)
33	321.10135	$C_{13}H_{21}SO_7^-$	0.5		7.5	diesel fuel SOA	(Blair et al., 2017)
34	251.09587	$C_{10}H_{19}SO_5^-$	1.9	5.3	7.5	alkane	(Riva et al., 2016b)
35	209.04892	$C_7H_{13}SO_5^-$	1.0	2.2	7.4	alkane	(Riva et al., 2016b)
36	307.12208	$C_{13}H_{23}SO_6^-$			7.2		
37	295.04931	$C_{10}H_{15}SO_8^-$	1.0	2.5	7.1	alkane	(Riva et al., 2016b)
38	279.12717	$C_{12}H_{23}SO_5^-$	2.2	3.3	6.5	alkane	(Riva et al., 2016b)
39	295.12208	$C_{12}H_{23}SO_6^-$	0.7	3.2	6.2	diesel fuel SOA	(Blair et al., 2017)
40	249.08022	$C_{10}H_{17}SO_5^-$	2.7	2.6	4.2	monoterpene	(Surratt et al., 2008; Wang et al., 2017)

#### nitrooxy-organosulfates

1	294.06530	$C_{10}H_{16}NO_7S^-$	67.3	5.0	82.9	monoterpene	(Surratt et al., 2008)
2	326.05513	$C_{10}H_{16}NO_9S^-$	23.9	4.0	29.2	monoterpene	(Surratt et al., 2008)
3	342.05004	$C_{10}H_{16}NO_{10}S^-$	8.3	3.3	23.3	monoterpene	(Surratt et al., 2008)
4	300.03948	$C_8H_{14}NO_9S^-$	3.9	2.5	19.2		
5	314.05513	$C_9H_{16}NO_9S^-$	1.9		14.6		
6	312.07586	$C_{10}H_{18}NO_8S^-$	10.1		11.5		
7	284.04456	$C_8H_{14}NO_8S^-$	6.9	3.0	10.4		
8	328.07078	$C_{10}H_{18}NO_9S^-$	3.3	2.5	9.4	monoterpene	(Surratt et al., 2008)
9	276.01835	$C_9H_{10}NO_7S^-$	3.2		9.3		
10	296.04456	$C_9H_{14}NO_8S^-$	23.2	1.4	9.0	monoterpene	(Surratt et al., 2008)

Table S2 The pearson correlations between individual OSs and NOSs species quantified by HPLC-MS (n=28)

	C <sub>3</sub> H <sub>5</sub> O <sub>5</sub> S <sup>-</sup>	C <sub>2</sub> H <sub>3</sub> O <sub>6</sub> S <sup>-</sup>	C <sub>3</sub> H <sub>5</sub> O <sub>6</sub> S <sup>-</sup>	C <sub>4</sub> H <sub>7</sub> O <sub>7</sub> S <sup>-</sup>	C <sub>5</sub> H <sub>7</sub> O <sub>7</sub> S <sup>-</sup>	C <sub>5</sub> H <sub>11</sub> O <sub>7</sub> S <sup>-</sup>	C <sub>5</sub> H <sub>10</sub> NO <sub>9</sub> S <sup>-</sup>	α-pinene OSs C <sub>10</sub> H <sub>17</sub> O <sub>5</sub> S <sup>-</sup>	β-pinene OSs C <sub>10</sub> H <sub>17</sub> O <sub>5</sub> S <sup>-</sup>	Limonene OSs C <sub>10</sub> H <sub>17</sub> O <sub>5</sub> S <sup>-</sup>	C <sub>9</sub> H <sub>15</sub> O <sub>6</sub> S <sup>-</sup>	C <sub>10</sub> H <sub>16</sub> NO <sub>7</sub> S <sup>-</sup>	C <sub>9</sub> H <sub>14</sub> NO <sub>8</sub> S <sup>-</sup>
C <sub>3</sub> H <sub>5</sub> O <sub>5</sub> S <sup>-</sup>	1.00												
C <sub>2</sub> H <sub>3</sub> O <sub>6</sub> S <sup>-</sup>	<b>0.88</b>	1.00											
C <sub>3</sub> H <sub>5</sub> O <sub>6</sub> S <sup>-</sup>	<b>0.92</b>	<b>0.91</b>	1.00										
C <sub>4</sub> H <sub>7</sub> O <sub>7</sub> S <sup>-</sup>	<b>0.97</b>	<b>0.86</b>	<b>0.86</b>	1.00									
C <sub>5</sub> H <sub>7</sub> O <sub>7</sub> S <sup>-</sup>	<b>0.96</b>	<b>0.95</b>	<b>0.93</b>	<b>0.95</b>	1.00								
C <sub>5</sub> H <sub>11</sub> O <sub>7</sub> S <sup>-</sup>	<b>0.91</b>	<b>0.73</b>	<b>0.72</b>	<b>0.93</b>	<b>0.85</b>	1.00							
C <sub>5</sub> H <sub>10</sub> NO <sub>9</sub> S <sup>-</sup>	<b>0.82</b>	<b>0.83</b>	<b>0.79</b>	<b>0.83</b>	<b>0.88</b>	<b>0.76</b>	1.00						
α-pinene OS (C <sub>10</sub> H <sub>17</sub> O <sub>5</sub> S <sup>-</sup> )	0.68	0.63	0.59	0.67	0.67	<b>0.71</b>	<b>0.84</b>	1.00					
β-pinene OS (C <sub>10</sub> H <sub>17</sub> O <sub>5</sub> S <sup>-</sup> )	0.22	0.29	0.44	0.20	0.31	(0.03)	0.26	0.01	1.00				
Limonene OS (C <sub>10</sub> H <sub>17</sub> O <sub>5</sub> S <sup>-</sup> )	0.62	0.69	0.71	0.56	0.67	0.37	0.39	0.20	0.44	1.00			
C <sub>9</sub> H <sub>15</sub> O <sub>6</sub> S <sup>-</sup>	<b>0.84</b>	<b>0.76</b>	<b>0.73</b>	<b>0.82</b>	<b>0.79</b>	<b>0.86</b>	<b>0.74</b>	<b>0.76</b>	(0.04)	0.48	1.00		
C <sub>10</sub> H <sub>16</sub> NO <sub>7</sub> S <sup>-</sup>	0.11	0.26	0.18	0.13	0.25	0.10	0.48	0.40	0.29	(0.14)	0.10	1.00	
C <sub>9</sub> H <sub>14</sub> NO <sub>8</sub> S <sup>-</sup>	0.02	0.07	0.02	0.02	0.10	0.06	0.26	0.32	0.19	(0.18)	0.07	<b>0.79</b>	1.00

## References

- Blair, S. L., MacMillan, A. C., Drozd, G. T., Goldstein, A. H., Chu, R. K., Pasa-Tolic, L., Shaw, J. B., Tolic, N., Lin, P., Laskin, J., Laskin, A., and Nizkorodov, S. A.: Molecular Characterization of Organosulfur Compounds in Biodiesel and Diesel Fuel Secondary Organic Aerosol, *Environ. Sci. Technol.*, 51, 119-127, 10.1021/acs.est.6b03304, 2017.
- Riva, M., Budisulistiorini, S. H., Zhang, Z., Gold, A., and Surratt, J. D.: Chemical characterization of secondary organic aerosol constituents from isoprene ozonolysis in the presence of acidic aerosol, *Atmos. Environ.*, 130, 5-13, 10.1016/j.atmosenv.2015.06.027, 2016a.
- Riva, M., Da Silva Barbosa, T., Lin, Y.-H., Stone, E. A., Gold, A., and Surratt, J. D.: Chemical characterization of organosulfates in secondary organic aerosol derived from the photooxidation of alkanes, *Atmos. Chem. Phys.*, 16, 11001-11018, 10.5194/acp-16-11001-2016, 2016b.
- Surratt, J. D., Gomez-Gonzalez, Y., Chan, A. W., Vermeylen, R., Shahgholi, M., Kleindienst, T. E., Edney, E. O., Offenberg, J. H., Lewandowski, M., Jaoui, M., Maenhaut, W., Claeys, M., Flagan, R. C., and Seinfeld, J. H.: Organosulfate formation in biogenic secondary organic aerosol, *J. Phys. Chem. A*, 112, 8345-8378, 10.1021/jp802310p, 2008.
- Wang, Y., Ren, J., Huang, X. H. H., Tong, R., and Yu, J. Z.: Synthesis of Four Monoterpene-Derived Organosulfates and Their Quantification in Atmospheric Aerosol Samples, *Environ. Sci. Technol.*, 51, 6791-6801, 10.1021/acs.est.7b01179, 2017.