



Is water vapor a key player of the wintertime haze in North China Plain? 1 2 Jiarui Wu^{1,4,7}, Naifang Bei², Bo Hu³, Suixin Liu^{1,4}, Meng Zhou⁵, Qiyuan Wang^{1,4}, Xia Li^{1,4,7}, Lang Liu^{1,4,7}, Tian Feng¹, Zirui Liu³, Yichen Wang¹, Junji Cao^{1,4}, Xuexi Tie^{1,4}, Jun Wang⁵, Luisa T. Molina⁶, and Guohui Li^{1,4*} 3 4 5 ¹Key Lab of Aerosol Chemistry and Physics, SKLLQG, Institute of Earth Environment, Chinese Academy of 6 7 Sciences, Xi'an, Shaanxi, China 8 ²School of Human Settlements and Civil Engineering, Xi'an Jiaotong University, Xi'an, Shaanxi, China 9 ³State Key Laboratory of Atmospheric Boundary Layer Physics and Atmospheric Chemistry, Institute of 10 Atmospheric Physics, Chinese Academy of Sciences, Beijing, 100029, China ⁴CAS Center for Excellence in Quaternary Science and Global Change, Xi'an, China 11 ⁵Department of Chemical and Biochemical Engineering & Interdisciplinary Graduate Program in 12 Geo-Informatics, University of Iowa, Iowa City, Iowa, USA 13 14 ⁶Molina Center for Energy and the Environment, La Jolla, California, USA 15 ⁷University of Chinese Academy of Science, Beijing, China *Correspondence to: Guohui Li (ligh@ieecas.cn) 16 17 18 Abstract. Water vapor has been proposed to amplify the severe haze pollution in China by enhancing the aerosol-radiation feedback (ARF). Observations have revealed that the near-surface $PM_{2.5}$ concentrations ([PM_{2.5}]) generally exhibits an increasing trend with relative humidity (RH) in North China Plain (NCP) during 2015 wintertime, indicating that the aerosol liquid water (ALW) caused by hygroscopic growth could play an important role in the PM_{2.5} formation and accumulation. Simulations during a persistent and heavy haze

19 20 21 22 23 24 pollution episode from 05 December 2015 to 04 January 2016 in NCP were conducted using 25 the WRF-CHEM model to comprehensively quantify contributions of the ALW effect to 26 near-surface [PM2.5]. The WRF-CHEM model generally performs reasonably well in 27 28 simulating the temporal variations of RH against measurements in NCP. The factor separation approach (FSA) was used to evaluate the contribution of the ALW effect on the ARF, 29 photochemistry, and heterogeneous reactions to $[PM_{2,5}]$. The ALW not only augments particle 30 sizes to enhance aerosol backward scattering, but also increases the effective radius to favor 31 aerosol forward scattering. The contribution of the ALW effect on the ARF and 32 photochemistry to near-surface $[PM_{2,5}]$ is not significant, generally within 1.0 µg m⁻³ on 33 average in NCP during the episode. Serving as an excellent substrate for heterogeneous 34 reactions, the ALW substantially enhances the secondary aerosol (SA) formation, with an 35 average contribution of 71%, 10%, 26%, and 48% to near-surface sulfate, nitrate, ammonium, 36 37 and secondary organic aerosol concentrations. Nevertheless, the SA enhancement due to the ALW decreases the aerosol optical depth and increases the effective radius to weaken the 38 ARF, reducing near-surface primary aerosols. The contribution of the ALW total effect to 39 near-surface [PM2.5] is 17.5% on average, which is overwhelmingly dominated by enhanced 40 41 SA. Model sensitivities also show that when the RH is less than 80%, the ALW progressively





- 42 increases near-surface $[PM_{2.5}]$, but commences to decrease when the RH exceeding 80% due
- 43 to the high occurrence frequencies of precipitation.
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49 1 Introduction

Atmospheric aerosols or fine particle matters (PM2.5) influence the climate directly by 50 scattering and absorbing the solar radiation, and indirectly by serving as cloud condensation 51 52 nuclei and ice nuclei (Ackerman, 1977; Ackerman and Baker, 1977; Jacobson, 1998, 2002; Penner et al., 2001). Moreover, high levels of $PM_{2.5}$ in the atmosphere also cause severe haze 53 54 pollution, impairing visibility and exerting deleterious effect on ecological system and human health (Chan and Yao, 2008; Zhang et al., 2013; Kurokawa et al., 2013; Weinhold, 2008; 55 Parrish and Zhu, 2009). In addition to anthropogenic emissions, the poor air quality is 56 57 generally influenced by stagnant meteorological situations with weak winds and high relative humidity (RH) (Flocas et al., 2009; Quan et al., 2013; Zhang et al., 2014; Bei et al., 2016; Wu 58 et al., 2017; Ding et al., 2017). RH, as an important meteorological factor in the atmosphere, 59 60 considerably affects the formation, chemical composition, and physical properties of atmospheric aerosols (Seinfeld et al., 2001; Hallquist et al., 2009; Poulain, 2010; Nguyen et 61 al., 2011). 62

63 As the main constituent in the atmosphere, water vapor directly participates in the atmospheric physical and chemical processes. Since many components of atmospheric 64 aerosols are hygroscopic, they take up water as RH increases (Covert et al., 1972; Pilinis et 65 al., 1989), thereby influencing the aerosol size distribution, chemical composition, mass 66 concentration, and corresponding optical properties as well as radiative effects (Im et al., 67 2001; Carrico et al., 2003; Randles et al. 2004; Cheng et al., 2008). Wang et al. (2016) have 68 indicated that the ratio of SO_4^{2-} to SO_2 exhibits an exponential increase with RH. Tie et al. 69 (2017) have shown that the sulfate, nitrate, and ammonium concentrations increase from 16 70 to 25 μ g m⁻³, 15 to 23 μ g m⁻³, and 11 to 17 μ g m⁻³, respectively, when RH increases from 60% 71 to 80%. Field measurements in Beijing have demonstrated that the inorganic aerosol fraction 72 73 increases with increasing RH (Wu et al., 2018). In addition, water vapor also serves as an





important medium in the formation of secondary aerosols (SA) through liquid-phase 74 reactions and heterogeneous reactions (Seinfeld and Pandis, 1986; Pilinis et al., 1989). For 75 example, Li et al. (2017) have indicated that the aerosol liquid water (ALW) induced by the 76 77 wet growth could play a significant role in the sulfate formation and emphasized the importance of bulk aqueous-phase oxidation of SO₂ in ALW and heterogeneous reaction of 78 79 SO₂ on aerosol surfaces involving ALW. ALW also plays an important role in secondary organic aerosol (SOA) formation (Hastings et al., 2005; Healy et al., 2009; Kamens et al., 80 2011; Koehler et al., 2004). Numerous studies have investigated the effect of RH on SOA 81 82 formed from different aromatics during their photochemical oxidation processes (Blando and Turpin, 2000; Cocker et al., 2001; Seinfeld et al., 2001; Zhou et al., 2011; Jia and Xu., 2014). 83 Furthermore, Zhang et al. (2015) have revealed that, as the RH increases from 40% to 85% in 84 85 the Yangtze River Delta of China, the aerosol scattering and backscattering coefficients increase by 58% and 25%, respectively, and the calculated aerosol direct radiative forcing 86 caused by hygroscopic growth is increased by 47%. 87

88 In recent years, China has experienced persistent haze pollution with unprecedentedly high PM_{2.5} concentrations during wintertime, particularly in North China Plain (NCP) (Chan 89 and Yao, 2008; He et al., 2001; Kan et al., 2012; Guo et al., 2014; Wang et al., 2014; Fu et al., 90 91 2014). A conceptual model based on the aerosol radiation feedback (ARF) has been 92 established to interpret the wintertime heavy haze formation, in which water vapor is considered to play a key role in the progressive accumulation and formation of $PM_{2.5}$. In 93 winter, when the atmospheric condition is stagnant, air pollutants commence to accumulate in 94 the planetary boundary layer (PBL), favorable for the $PM_{2.5}$ formation. Increasing $PM_{2.5}$ 95 scatters or absorbs the incoming solar radiation to lower the surface temperature and cause 96 anomalous temperature inversion, subsequently suppressing the vertical turbulent diffusion 97 98 and decreasing the planetary boundary layer height (PBLH) to further trap more air pollutants





and water vapor to increase the RH in the PBL. Increasing RH enhances aerosol hygroscopic
growth and multiphase reactions and augments the particle size and mass, causing further
dimming and decrease of the surface temperature and PBL height (Quan et al., 2013; Tie et
al., 2017; Ding et al., 2017). However, few studies have been performed in China to
comprehensively quantify the effect of water vapor in the atmospheric physical and chemical
process on the PM_{2.5} pollution to further verify the haze formation.

The purpose of the present study is to quantitatively evaluate the contribution of aerosol
water induced by the aerosol wet growth to PM_{2.5} concentrations in NCP using the Weather
Research and Forecast model with Chemistry (WRF-CHEM). The model configuration and
methodology are described in Section 2. Results and discussions are presented in Section 3,
and conclusions and summaries are given in Section 4.

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111 2 Model and methodology

112 2.1 WRF-CHEM Model and Configuration

A persistent air pollution episode with high levels of PM_{2.5} from 05 December 2015 to 04 January 2016 in NCP was simulated using the WRF-CHEM model with modifications by Li et al. (2010, 2011a, b, 2012) from the Molina Center for Energy and the Environment. Figure 1 shows the WRF-CHEM model simulation domain and Table 1 provides the model configurations. Detailed model description can be found in Wu et al. (2018a).

118 2.2 Data and Methodology

The model performance of RH was validated using the hourly measurements in
Luancheng, Yucheng, and Jiaozhouwan observed from the Chinese Ecosystem Research
Network (CERN). Furthermore, the NCEP reanalysis data was used to compare to the
simulated RH distribution. The detailed information of other data used for validation can be
found in Wu et al. (2018a).





The mean bias (MB), root mean square error (RMSE) and the index of agreement (IOA) 124 were utilized to evaluate the performance of the WRF-CHEM model simulations against 125 126 measurements. To assess the contributions of ALW to the near-surface concentrations of air 127 pollutants in NCP, the factor separation approach (FSA) was used in this study (Stein and Alpert, 1993; Gabusi et al., 2008; Li et al., 2014). Generally, the formation of the secondary 128 129 atmospheric pollutants, such as O₃, secondary organic aerosol, and nitrate, is a complicated nonlinear process in which its precursors from various emissions sources and transport react 130 chemically or reach equilibrium thermodynamically. Nevertheless, it is not straightforward to 131 132 evaluate the contributions from different factors in a nonlinear process (Wu et al., 2017). The factor separation approach (FSA) proposed by Stein and Alpert (1993) can be used to isolate 133 the effect of one single factor from a nonlinear process and has been widely used to evaluate 134 135 source effects. The total effect of one factor in the presence of others can be decomposed into 136 contributions from the factor and that from the interactions of all those factors. Considering that there are two factors X and Y that influence the formation of secondary pollutants in the 137 atmosphere and also interact with each other. Denoting f_{XY} , f_X , f_Y , and f_0 as the 138 simulations including both of two factors, factor X only, factor Y only, and none of the two 139 factors, respectively. The contributions of factor X and Y can be defined as $f_{XY} - f_Y$ and 140 $f_{XY} - f_X$, respectively. Detailed description of the methodology can be found in Wu et al. 141 (2017). 142

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144 3 Results and discussions

145 3.1 Relationship between RH and near-surface PM_{2.5} concentrations

High RH has been suggested to be an important factor facilitating the SA formation in
the atmosphere and aggravating the haze pollution (Sun et al., 2013; Cheng et al., 2015).
Figure 2 presents the scatter plot of the near-surface PM_{2.5} concentrations ([PM_{2.5}]) and RH in





the winter of 2015 at six typical polluted cities in NCP, including Beijing, Tianjin, 149 Shijiazhuang, Tangshan, Baoding, and Chengde. The observed near-surface $[PM_{25}]$ at those 150 six cities display a growing trend with increasing RH, suggesting that the ALW induced by 151 152 the hygroscopic growth under high RH conditions has potentials to accelerate the PM2.5 formation and accumulation. Increasing RH facilitates the aerosol hygroscopic growth and 153 154 further enhances the ALW, which serves as an efficient medium for promoting the liquid-phase and heterogeneous reactions and accelerating the transformation of reactive 155 gaseous pollutants into the particle phase. Increased ALW also augments the particle size, 156 157 enhancing the ARF to increase the near-surface [PM_{2.5}]. However, the attenuation of incoming solar radiation caused by the ALW also decreases the photolysis rates, unfavorable 158 for photochemical activities and lowering the atmospheric oxidation capability (AOC). Field 159 160 measurements show that large fraction of SA in PM_{2.5} has been observed in NCP during wintertime (Sun et al., 2013; Guo et al., 2014; Xu et al., 2015). Therefore, decreased AOC 161 generally does not facilitate the SA formation, particularly with regards to SOA and nitrate, to 162 163 partially counteract the PM_{2.5} enhancement caused by the ALW. It is also worth noting that since high RH frequently corresponds to atmospheric stagnation, near-surface [PM_{2.5}] also 164 165 build up under high RH conditions. For example, the humid air mass is subject to being 166 transported from south to NCP under the stagnant weather with weak south winds and 167 meanwhile the $PM_{2.5}$ also accumulates due to the unfavorable dispersion condition. Additionally, when the RH is very high, there also exist the low near-surface [PM_{2.5}] shown 168 169 in Figure 2, demonstrating that other factors, such as emissions, horizontal transport, vertical exchange, and precipitation, also substantially influence near-surface [PM_{2.5}]. Generally, high 170 occurrence frequency of precipitation coincides with high RH, thus the precipitation washout 171 might constitute one of the most possible reasons for the low near-surface [PM2.5] under high 172 173 RH situations. Therefore, it is still imperative to verify quantitatively the contribution of the





174 ALW to near-surface $[PM_{2.5}]$.

175 3.2 Model validation

The WRF-CHEM model simulation of the haze pollution episode in NCP has been comprehensively validated using available measurements in Wu et al. (2018a). In general, the model simulates well the spatial distribution and temporal variation of PM_{2.5}, O₃, NO₂, SO₂ and CO mass concentrations compared to observations in NCP. The predicted aerosol species are also in good agreement with the measurement in Beijing. Moreover, the model performs reasonably well in simulating the aerosol optical depth and single scattering albedo, PBL height and downward shortwave flux against measurements.

In order to verify the effect of the ALW on near-surface [PM2.5] during the haze 183 pollution episode, the simulated temporal variation of RH was first compared to 184 185 measurements at Luancheng, Yucheng, and Jiaozhouwan in NCP from 05 December 2015 to 04 January 2016 (Figure 3). The WRF-CHEM model generally performs well in simulating 186 the hourly variation of RH in these three cities, with IOAs of 0.73, 0.83, and 0.69, 187 respectively. RH is a key meteorological component, sensitive to the atmospheric 188 thermodynamic (e.g., temperature) and dynamic (e.g., winds) conditions. Even when the 189 simulated water vapor content is the same as the observation, the overestimation or 190 191 underestimation of temperature still causes underestimation or overestimation of RH. Biases 192 of wind speeds and directions considerably influence the origination of air mass at the observation site. In general, the northerly wind carries dry air, and is opposite for the 193 194 southerly wind during wintertime in NCP. Therefore, the uncertainties from meteorological field simulations might constitute one of the most possible reasons for the RH bias (Bei et al., 195 2017). Figure 4 presents the pattern comparison of the average simulated RH and the NCEP 196 reanalysis during the episode. The simulated RH distribution is generally consistent with that 197 198 from the reanalysis, e.g., dry air in West China and fairly humid air in South China. However,





the air over NCP in the simulation is more humid than the analysis, and the average simulated
RH is 70.6%, about 20% higher than the reanalyzed RH, which might be caused by the
temperature decrease due to the ARF considered in the simulation.

202 3.3 Sensitivity studies

The ALW not only enlarges the particle size to increase the aerosol optical depth (AOD), likely enhancing the ARF to facilitate the PM_{2.5} accumulation or to alter photolysis rates to affect the AOC, but also influences the SA formation serving as a medium for multiphase reactions. Therefore, sensitivity studies are used to quantitatively evaluate the effect of the ALW on the PM_{2.5} concentration during the haze pollution episode.

The FSA method was used to evaluate the contribution of the ALW to near-surface [PM_{2.5}] by differentiating two model simulations with and without the ALW effect. Besides the base case with all the ALW effect (hereafter referred as to f_{base}), additional four sensitivity simulations were performed, in which the ALW effect on the ARF, photolysis, multiphase reactions, and the total were excluded, respectively (hereafter referred as to $f_{alw-rad0}$, f_{alw-j0} , $f_{alw-het0}$, and $f_{alw-tot0}$, respectively)

214 3.3.1 ALW effect on the ARF

215 The ALW, caused by the aerosol hygroscopic growth, augments the particle size to increase AOD, potentially enhancing the ARF and aggravating the haze pollution. Figure 5 216 217 shows the distribution of the average AOD contribution due to the ALW during the haze episode, evaluated by differentiating f_{base} and $f_{alw-rad0}$. Apparently, the ALW 218 219 substantially increases the AOD in NCP, with the contribution ranging from 30% to more than 50%, indicating that ALW is an important contributor of the AOD. Substantial increase 220 of the AOD due to the ALW is anticipated to attenuate the incoming solar radiation, 221 decreasing the surface temperature and suppressing the PBL development, therefore, 222 223 deteriorating the haze pollution, as proposed by recent studies (Tie et al., 2017; Liu et al.,





224 2018).

Figure 6 presents the distribution of the average near-surface PM_{2.5} contribution of the 225 ALW effect on the ARF (hereafter referred as to ALW-ARF) during the haze episode. 226 227 Interestingly, the ALW-ARF does not increase the near-surface [PM2.5] consistently in NCP, as expected. The ALW-ARF enhances the near-surface $[PM_{2,5}]$ most strikingly in the south of 228 Hebei, with a contribution of more than 15 µg m⁻³ (less than 7%). However, in some areas, 229 230 such as the east of Shandong, the near-surface PM_{2.5} contribution of the ALW-ARF is negative, or the ALW-ARF decreases [PM2.5]. On average, the ALW-ARF increases 231 near-surface $[PM_{2,5}]$ in NCP during the episode by about 1.1 µg m⁻³, so cannot constitute an 232 important factor for the heavy haze formation. 233

It is worth noting that enlarged particles due to the ALW not only increase the AOD to 234 235 enhance aerosol backward scattering, but also cause the aerosol spectrum to successively 236 shift toward larger sizes. Based on the Mie scattering theory, when the particle size is similar to the wavelength of incoming solar radiation, the radiation is favored to be scattered in the 237 238 forward directions. In order to further verify the ALW effect on solar radiation and near-surface [PM_{2.5}], an ensemble method is employed similar to that reported in Wu et al. 239 (2018a). The daytime near-surface [PM_{2.5}] in NCP during the episode in f_{base} are first 240 subdivided into 30 bins with the interval of 20 µg m⁻³. The AOD at 550 nm, aerosol effective 241 radius (Reff), downward shortwave radiation at the surface (SWDOWN), surface temperature 242 (TSFC), PBLH, and near-surface [PM_{2.5}] in f_{base} and $f_{alw-rad0}$ in the same grid cell are 243 assembled as the bin [PM2.5], respectively, and an average of these variables in each bin are 244 calculated. Figure 7 shows the variation of AOD and Reff in f_{base} and $f_{alw-rad0}$ as a 245 function of bin [PM2.5]. The ALW not only significantly enhances the AOD, with an average 246 contribution of 46% in NCP during the episode, but also increases the Reff considerably 247 (Figures 7a and 7b). The Reff enhancement due to the ALW is the most striking with 248





near-surface [PM2.5] between 40 and 160 µg m⁻³, exceeding 60%. On average, the ALW 249 increases the Reff from 0.31 µm to 0.48 µm, close to the peak band of solar radiation. 250 Therefore, the ALW increases the AOD to scatter more incoming solar radiation, decreasing 251 252 the SWDOWN, but augments particle sizes to favor the forward scattering, increasing the SWDOWN. Generally, the decrease of the SWDOWN caused by the ALW is not significant, 253 less than 1% with near-surface [PM_{2.5}] less than 200 μ g m⁻³ and ranging from 2% to 3% with 254 [PM_{2.5}] exceeding 240 µg m⁻³ (Figure 8a). Correspondingly, the ALW-ARF effect on the 255 daytime TSFC is also marginal and the TSFC is decreased by less than 0.2°C (Figure 8b). 256 Furthermore, the ALW-ARF generally increases the daytime PBLH with near-surface [PM2.5] 257 less than 140 µg m⁻³, but are opposite with [PM_{2.5}] exceeding 140 µg m⁻³ (Figure 8c). Hence, 258 the contribution of the ALW-ARF to near-surface $[PM_{2,5}]$ is highly uncertain (Figure 8d), 259 260 depending on the relative importance of the ALW induced enhancement of aerosol backward 261 and forward scattering.

262 3.3.2 ALW effect on the photochemistry

263 In addition to the ARF, the ALW also exerts an impact on the photochemistry by altering the aerosol backward and forward scattering to affect the photolysis, further influencing the 264 265 ozone (O_3) and SA formation. Previous studies have shown that the ALW modifies the vertical profile of photolysis rate of NO₂ (J_{NO_2}), inhibiting it at the ground level and 266 accelerating it in the upper PBL (Tao et al., 2014; Dickerson et al., 1997). The combination 267 reaction between the ground state oxygen atom (O³P), produced from NO₂ photolysis, and 268 oxygen molecules (O_2) forms O_3 , representing the only important source of O_3 in the 269 270 troposphere:

$$NO_2 + h\nu \rightarrow NO + O(^3P) \tag{1}$$

272
$$O(^{3}P) + O_{2} + M \to O_{3} + M$$
 (2)

273 Thus, the variation of J_{NO_2} considerably affects the O₃ formation in the troposphere,





changing the AOC and further the SA formation.

Figures 9a and 9b present the distribution of the percentage variation of average daytime 275 J_{NO_2} due to the ALW during the haze episode at the 1st and 5th model layer, respectively. At 276 the 1st model layer, except in the north of Jiangsu, the ALW generally decreases J_{NO_2} 277 slightly in NCP. However, at 5th model layer, the region with enhanced J_{NQ_2} caused by the 278 ALW is obviously increased compared to that at 1st model layer. Apparently, the ALW 279 induced enlargement of particle sizes increases the AOD and enhances the aerosol backward 280 scattering to reduce the solar radiation reaching the ground level, decreasing the photolysis 281 282 rate, but the enhanced forward scattering still potentially accelerates the photolysis, such as in 283 the north of Jiangsu. In addition, the enhanced aerosol backward scattering also increases the photolysis rate in the upper and above PBL, which is consistent with previous studies (Tao et 284 285 al., 2014; Dickerson et al., 1997). The variation of average daytime O₃ concentrations is not consistent with that of J_{NO_2} at the 1st model layer (Figure 9c). For example, although the 286 J_{NO_2} is decreased in Hebei and Shandong, the O₃ concentration is still enhanced by the ALW 287 in some areas of the two provinces. One of the possible reasons is the vertical transport of O_3 288 from the upper layers where the O_3 is plausibly enhanced due to the increased photolysis rate 289 (Figure 9b). At the 5th model layer, the area with enhanced O₃ concentrations is much larger 290 than that at 1^{st} model layer, which is in agreement with the variation of the J_{NO_2} (Figure 9d). 291 On average, the ALW decreases near-surface (1st model layer) daytime O₃ concentrations by 292 about 0.2 μ g m⁻³ (or 0.45%), playing a minor role in O₃ formation. 293

Figure 10 shows the distribution of the average near-surface $PM_{2.5}$ contribution of the ALW effect on the photolysis frequencies (hereafter referred as to ALW-J) during the haze episode. Except in some areas in Shandong and Anhui, the ALW-J slightly decreases near-surface $[PM_{2.5}]$ in NCP, with an average reduction of about 0.87 µg m⁻³ (or 0.64%). Therefore, the ALW-J does not play an important role in mitigating the haze pollution.





299 3.3.3 ALW effect on heterogeneous reactions

300 The ALW provides an excellent substrate for heterogeneous reactions in the atmosphere,

301 which have been proposed to play a key role in the SA formation during haze days (Li et al.,

302 2017; Xing et al., 2018).

Figure 11 presents the distribution of contributions of the ALW on heterogeneous 303 304 reactions (hereafter referred to ALW-HET) to near-surface sulfate, nitrate, and ammonium concentrations averaged during the haze episode. A parameterization of sulfate heterogeneous 305 formation involving ALW has been developed and implemented into the WRF-CHEM model, 306 307 which has successfully reproduced the observed rapid sulfate formation during haze days (Li et al., 2017). The sulfate heterogeneous formation from SO₂ is parameterized as a first order 308 irreversible uptake by ALW surfaces, with a reactive uptake coefficient of 0.5×10^{-4} assuming 309 that there is enough alkalinity to maintain the high iron-catalyzed reaction rate. The 310 contribution of the ALW-HET to the sulfate formation is substantial in NCP, exceeding 5 μ g 311 m^{-3} (or 50%) in NCP (Figures 11a and 11b). The ALW-HET contributes about 71.3% of the 312 313 sulfate in NCP during the episode, indicating that the heterogeneous formation involving the ALW is the dominant sulfate source during haze days. 314

The heterogeneous hydrolysis of N_2O_5 on the surface of deliquescent aerosols leads to 315 the HNO₃ formation, which is the key contributor to the nitrate aerosol loading (Riemer et al., 316 2003; Pathak et al., 2011; Chang et al., 2016). In this study, the parameterization of the 317 heterogeneous hydrolysis of N_2O_5 , proposed by Riemer et al. (2003), has been included in the 318 319 WRF-CHEM model for considering the effect of ALW-HET on the nitrate formation. The N_2O_5 uptake coefficient on wet aerosol surfaces ranges from 0.002 to 0.02, depending on the 320 sulfate and nitrate aerosol mass. The ALW-HET generally increases the near-surface nitrate 321 mass concentration by 2 to 8 μ g m⁻³ in NCP, with an average contribution of 2.8 μ g m⁻³ (or 322 323 10%) (Figures 11c and 11d). It is worth noting that the ALW-HET does not consistently





increase the nitrate formation and the nitrate concentration is considerably decreased in some 324 areas in east China. One of the possible reasons is that the ALW-HET substantially enhances 325 the sulfate formation and the increased sulfate competes with nitrate for ammonia (NH₃), 326 327 suppressing the nitrate formation. Figure 12 shows the distribution of the NH₃ emission rate in December. High NH₃ emissions are concentrated in NCP, Central China, Sichuan basin, 328 329 and Northeast China, where the nitrate concentration is generally increased by the ALW-HET. Ammonium serves as the main alkali in the atmosphere to neutralize acidic aerosols 330 such as sulfate and nitrate, so its formation is not only dependent on its precursor (NH_3) , but 331 332 is also influenced by acid aerosols. The ALW-HET enhances the near-surface ammonium mass concentration most strikingly in NCP with high NH₃ emissions and increased sulfate 333 and nitrate aerosols. The contribution of the ALW-HET to the ammonium concentration 334 varies from 2 to more than 10 μ g m⁻³, with an average of 4.2 μ g m⁻³ (or 25.6%), showing that 335 the heterogeneous formation constitutes an important ammonium source (Figures 11e and 336 11f). 337

338 Heterogeneous reactions are also an important pathway for the SOA formation (Fu et al., 2008; Li et al., 2013). Laboratory and field studies have indicated that glyoxal and 339 methylglyoxal play an important role in SOA formation via aerosol uptake or cloud 340 processing (Liggio et al., 2005; Volkamer et al., 2007; Li et al., 2011). In this study, the 341 heterogeneous reaction of SOA formation from glyoxal and methylglyoxal is parameterized 342 as a first-order irreversible uptake by aerosol particles, with a reaction uptake coefficient of 343 3.7×10^{-3} (Liggio et al., 2005; Zhao et al., 2006; Volkamer et al., 2007; Li et al., 2011). 344 During the haze episode, the average near-surface SOA contribution of the ALW-HET is 7.4 345 μg m⁻³ (or 48%) in NCP, ranging from 3 to over 15 μg m⁻³ (or 30 to over 50%) in NCP, 346 showing that heterogeneous reactions of glyoxal and methylglyoxal on wet aerosol surfaces 347 348 play a critical role in SOA formation (Figure 13). Xing et al. (2018) have demonstrated that,





in Beijing-Tianjin-Hebei, the near-surface SOA contribution from glyoxal and methylglyoxal
is around 30% during a haze episode in January 2014, much less than that (48%) in the study.
The possible reason is that the RH in the episode is higher than that in Xing et al. (2018),
causing more SOA formation from glyoxal and methylglyoxal.

High SA contribution to the haze pollution has been observed in China (e.g., Huang et 353 354 al., 2014), so the ALW-HET substantially enhances the SA formation, constituting an important factor for heavy haze formation. Figure 14 shows the contribution of the 355 ALW-HET to average near-surface [PM_{2.5}] during the study episode. The ALW-HET 356 enhances near-surface [PM2.5] by more than 40 µg m-3 in the central part of NCP, and on 357 average, the PM_{2.5} contribution of the ALW-HET is 21.7 μ g m⁻³ (or 15.9%), less than the total 358 enhancement of 25.1 µg m⁻³ (or 18.4%) from sulfate, nitrate, ammonium, and SOA. The 359 inconsistency indicates that the ALW-HET induced SA enhancement causes a decrease in 360 primary aerosols. Figure 15a and 15b show the percentage decrease of black carbon (BC) and 361 primary organic aerosols (POA) as a function of bin $[PM_{25}]$, respectively. Interestingly, the 362 363 average BC and POA concentrations are decreased by 5.1% and 5.2% due to the ALW-HET, respectively. The ALW-HET induced SA enhancement augment the particle size and should 364 increase the AOD. However, on the contrary, the ALW-HET decreases the AOD, although 365 considerably increases the Reff (Figures 16a and 16b). The reason is that the enhanced SA 366 enlarges particle sizes, facilitating the coagulation to decrease the aerosol number 367 concentration and surface area, as shown in Figures 16c and 16d. Hence, the decreased AOD 368 369 and increased Reff due to the ALW-HET enhance the SWDOWN and TSFC, further increasing the PBLH and decreasing near-surface primary aerosols, as shown in Figure 17. 370

371 3.3.4 ALW total effect

Above discussions have shown that the ALW influences near-surface [PM_{2.5}] through
 complicated physical and chemical processes, which interact with each other. Figure 18





shows the near-surface PM2.5 contribution of the total ALW effect (hereafter referred as to 374 ALW-TOT) during the haze episode, evaluated by differentiating f_{base} and $f_{alw-tot0}$. The 375 376 ALW-TOT plays an important role in the PM_{2.5} formation during the wintertime haze pollution in NCP, with an average contribution of 23.8 µg m⁻³ (or 17.5%), ranging from 5 to 377 over 40 µg m⁻³. About 78% of the enhanced near-surface [PM_{2.5}] due to the ALW-TOT is 378 379 contributed by secondary inorganic aerosols, in which the contributions from sulfate, nitrate, 380 and ammonium are 45%, 14% and 19%, respectively, and around 32% is contributed by SOA. 381 Therefore, about 10% of near-surface $[PM_{2,5}]$ enhancement from SA is counterbalanced by 382 the decrease in the primary aerosols. Therefore, the ALW induced enhancement of near-surface [PM_{2.5}] is overwhelmingly determined by the ALW-HET, and the ALW-ARF 383 384 and ALW-J are subject to decreasing $[PM_{2,5}]$.

Figure 19 shows the variation of near-surface [PM2.5] caused by the ALW-TOT, 385 386 ALW-HET, ALW-ARF, and ALW-J, respectively, as a function of bin near-surface RH in NCP during the haze episode to further assess the ALW effect. The hourly near-surface RH in 387 f_{base} is first divided into 28 bins, ranging from 30% to 100%, with interval of 2.5%. The 388 $PM_{2.5}$ contribution from the four sensitivity simulations is assembled in the same grid cell as 389 the bin RH, and an average of PM_{2.5} contribution in each bin is calculated. The ALW does not 390 continuously increase near-surface [PM2.5] with the RH. When the RH is less than 80%, the 391 near-surface PM_{2.5} contribution of the ALW-TOT generally increases rapidly with the RH. 392 393 However, when the RH exceeds 80%, the contribution commences to decrease and fluctuates between 20 and 30 µg m⁻³, showing the effect of high occurrence frequencies of precipitation. 394 In addition, the ALW-HET dominates the PM_{2.5} contribution, particularly with RH less than 395 396 50%. The ALW-RAD generally decreases [PM_{2.5}] slightly with the RH less than 52.5% and vice versa with the RH more than 52.5%. The $PM_{2.5}$ contribution of the ALW-J is negative 397 and less than 1.5 μ g m⁻³, except when the RH is between 92.5% and 97.5%. 398





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400 4 Conclusions and summaries

The good relationships between near-surface [PM2.5] concentration and RH during the 401 402 wintertime of 2015 in NCP indicate the possibility that high RH plays an important role in the $PM_{2.5}$ formation during the haze pollution. A severe haze pollution episode from 05 403 404 December 2015 to 04 January 2016 is simulated using the WRF-CHEM model to investigate the impact of ALW caused by the accumulated moisture on near surface [PM2.5] in NCP. The 405 air over NCP during the haze episode is humid, with an average simulated RH of about 71%. 406 407 In general, the WRF-CHEM model reproduces reasonably well the temporal variations of RH compared to observations at three sites in NCP, although the model biases still exist due to 408 the uncertainties in simulated meteorological fields. 409

410 The FSA method is used to evaluate the contribution of the ALW effect on ARF, photochemistry and heterogeneous reactions to the wintertime PM2.5 concentration in NCP. 411 Model results show that the ALW substantially increases the daytime AOD with an average 412 413 contribution of 46% in NCP during the episode, enhancing the aerosol backward scattering, and also augments the Reff from 0.31 µm to 0.48 µm, approaching the peak band of solar 414 radiation, favoring the aerosol forward scattering. The ALW does not significantly attenuate 415 416 the incoming solar radiation at the ground surface to enhance the ARF, and the average near-surface PM2.5 contribution of the ALW-ARF is about 1.1 µg m⁻³ in NCP during the 417 episode. Therefore, the ALW-ARF is not an important factor for heavy haze formation and its 418 419 contribution relies on the relative importance of the ALW induced enhancement of aerosol backward and forward scattering. The ALW generally decreases the photolysis rate at the 420 surface level due to enhanced backward aerosol scattering, but the favored forward scattering 421 still possibly accelerates the photolysis. Additionally, the ALW increases the photolysis rate 422 423 in the upper and above PBL. On average, the ALW decreases near-surface daytime O_3





- 424 concentrations by 0.20 μg m⁻³ and [PM_{2.5}] by 0.87 μg m⁻³, playing a minor role in mitigating
 425 the haze pollution.
- 426 The ALW substantially enhances the SA formation by serving as an important medium 427 for liquid-phase and heterogeneous reactions. The ALW contribution to near-surface sulfate, nitrate, ammonium, and SOA concentrations is 71%, 10%, 26%, and 48% in NCP during the 428 429 episode, respectively. However, the enhanced SA due to the ALW-HET enlarges particle sizes to facilitate the coagulation, decreasing the aerosol number concentration and surface area 430 and further AOD. Therefore, the decreased AOD and increased Reff enhance the incoming 431 432 solar radiation reaching the ground surface, further increasing the surface temperature and PBLH and decreasing near-surface primary aerosols. The ALW-HET contributes 15.9% of 433 near-surface $[PM_{2.5}]$, less than the total SA enhancement of 18.4%, and the rest is 434 435 counterbalanced by the decrease in primary aerosols.
- The near-surface PM_{2.5} contribution of the ALW total effect is 17.5% in NCP, indicating
 that ALW plays an important role in the PM_{2.5} formation during the wintertime haze pollution.
 Moreover, the ALW-HET overwhelmingly dominates the PM_{2.5} enhancement due to the ALW.
 The ALW does not consistently enhance near-surface [PM_{2.5}] with increasing RH. When the
 RH exceeds 80%, the contribution of the ALW commences to decrease caused by the high
 occurrence frequencies of precipitation.

Although the model performs reasonably well in simulating air pollutants, aerosol species and optical properties, and RH during the episode in NCP, the uncertainties from meteorological fields and emission inventory still exist, leading to model biases. In addition, simulation period of one month might not be sufficient to provide a comprehensive view of the ALW effect on the PM_{2.5} formation. More wintertime case studies will need to be performed in the future to further investigate the ALW effect.





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476	Ackerman, T., and Baker, M. B.: Shortwave radiative effects of unactivated aerosol-particles
477	in clouds, J. Appl. Meteorol., 16, 63-69,
478	10.1175/1520-0450(1977)016<0063:sreoua>2.0.co;2, 1977.
479	Ackerman, T. P.: Model of effect of aerosols on urban climates with particular applications to
480	Los-angeles basin, J. Atmos. Sci., 34, 531-547,
481	10.1175/1520-0469(1977)034<0531:amoteo>2.0.co;2, 1977.
482	Bei, N. F., Li, G. H., Huang, R. J., Cao, J. J., Meng, N., Feng, T., Liu, S. X., Zhang, T., Zhang,
483	Q., and Molina, L. T.: Typical synoptic situations and their impacts on the wintertime air
484	pollution in the Guanzhong basin, China, Atmos. Chem. Phys., 16, 7373-7387,
485	10.5194/acp-16-7373-2016, 2016a.
486	Bei, N. F., Xiao, B., Meng, N., and Feng, T.: Critical role of meteorological conditions in a
487	persistent haze episode in the Guanzhong basin, China, Sci. Total Environ., 550,
488	273-284, 10.1016/j.scitotenv.2015.12.159, 2016b.
489 490 491 492 493	Bei, N. F., Wu, J. R., Elser, M., Feng, T., Cao, J. J., El-Haddad, I., Li, X., Huang, R. J., Li, Z. Q., Long, X., Xing, L., Zhao, S. Y., Tie, X. X., Prevot, A. S. H., and Li, G. H.: Impacts of meteorological uncertainties on the haze formation in Beijing-Tianjin-Hebei (BTH) during wintertime: a case study, Atmos. Chem. Phys., 17, 14579-14591, 10.5194/acp-17-14579-2017, 2017.
494	Blando, J. D., and Turpin, B. J.: Secondary organic aerosol formation in cloud and fog
495	droplets: a literature evaluation of plausibility, Atmos. Environ., 34, 1623-1632,
496	10.1016/s1352-2310(99)00392-1, 2000.
497 498 499	Carrico, C. M., Kus, P., Rood, M. J., Quinn, P. K., and Bates, T. S.: Mixtures of pollution, dust, sea salt, and volcanic aerosol during ACE-Asia: Radiative properties as a function of relative humidity, J. Geophys. ResAtmos., 108, 18, 10.1029/2003jd003405, 2003.
500 501	Chan, C. K. and Yao, X.: Air pollution in mega cities in China, Atmos. Environ., 42, 1–42, 2008.
502	Chang, W. L., Brown, S. S., Stutz, J., Middlebrook, A. M., Bahreini, R., Wagner, N. L., Dube,
503	W. P., Pollack, I. B., Ryerson, T. B., and Riemer, N.: Evaluating N ₂ O ₅ heterogeneous
504	hydrolysis parameterizations for CalNex 2010, J. Geophys. ResAtmos., 121,
505	5051-5070, 10.1002/2015jd024737, 2016.
506	Chen, F. and Dudhia, J.: Coupling an advanced land surface-hydrology model with the Penn
507	State-NCAR MM5 modeling system. Part I: Model implementation and sensitivity, Mon.
508	Weather Rev., 129(4), 569-585, 2001.
509 510 511	Cheng, Y., He, K. B., Du, Z. Y., Zheng, M., Duan, F. K., and Ma, Y. L.: Humidity plays an important role in the PM _{2.5} pollution in Beijing, Environ. Pollut., 197, 68-75, 10.1016/j.envpol.2014.11.028, 2015.
512	Cheng, Y. F., Wiedensohler, A., Eichler, H., Heintzenberg, J., Tesche, M., Ansmann, A.,
513	Wendisch, M., Su, H., Althausen, D., Herrmann, H., Gnauk, T., Bruggemann, E., Hu, M.,
514	and Zhang, Y. H.: Relative humidity dependence of aerosol optical properties and direct
515	radiative forcing in the surface boundary layer at Xinken in Pearl River Delta of China:
516	An observation based numerical study, Atmos. Environ., 42, 6373-6397,
517	10.1016/j.atmosenv.2008.04.009, 2008.





518 519	Chou, MD. and Suarez, M. J.: A solar radiation parameterization for atmospheric studies, NASA Tech. Rep. NASA/TM-1999- 10460, 15, 38 pp., 1999.
520 521	Chou, MD. and Suarez, M. J.: A thermal infrared radiation parameterization for atmospheric studies, NASA/TM-2001-104606, 19, 55 pp., 2001.
522 523 524	Cocker, D. R., Clegg, S. L., Flagan, R. C., and Seinfeld, J. H.: The effect of water on gas-particle partitioning of secondary organic aerosol. Part I: alpha-pinene/ozone system, Atmos. Environ., 35, 6049-6072, 10.1016/s1352-2310(01)00404-6, 2001a.
525	Cocker, D. R., Flagan, R. C., and Seinfeld, J. H.: State-of-the-art chamber facility for
526	studying atmospheric aerosol chemistry, Environ. Sci. Technol., 35, 2594-2601,
527	10.1021/es0019169, 2001b.
528 529 530 531	Cocker, D. R., Mader, B. T., Kalberer, M., Flagan, R. C., and Seinfeld, J. H.: The effect of water on gas-particle partitioning of secondary organic aerosol: II. m-xylene and 1,3,5-trimethylbenzene photooxidation systems, Atmos. Environ., 35, 6073-6085, 10.1016/s1352-2310(01)00405-8, 2001c.
532	Covert, D. S., Charlson, R. J., and Ahlquist, N. C.: A Study of the Relationship of Chemical
533	Composition and Humidity to Light Scattering by Aerosols, J. Appl. Meteorol., 11,
534	968-976, 1972.
535	Dickerson, R. R., Kondragunta, S., Stenchikov, G., Civerolo, K. L., Doddridge, B. G., and
536	Holben, B. N.: The impact of aerosols on solar ultraviolet radiation and photochemical
537	smog, Science, 278, 827-830, 10.1126/science.278.5339.827, 1997.
538	Ding, Y. H., Wu, P., Liu, Y. J., and Song, Y. F.: Environmental and Dynamic Conditions for
539	the Occurrence of Persistent Haze Events in North China, Engineering, 3, 266-271,
540	10.1016/j.eng.2017.01.009, 2017.
541	Flocas, H., Kelessis, A., Helmis, C., Petrakakis, M., Zoumakis, M., and Pappas, K.: Synoptic
542	and local scale atmospheric circulation associated with air pollution episodes in an urban
543	Mediterranean area, Theo. Appl. Climatol., 95, 265-277, 10.1007/s00704-008-0005-9,
544	2009.
545	Fu, G. Q., Xu, W. Y., Yang, R. F., Li, J. B., and Zhao, C. S.: The distribution and trends of fog
546	and haze in the North China Plain over the past 30 years, Atmos. Chem. Phys., 14,
547	11949-11958, 10.5194/acp-14-11949-2014, 2014.
548	Fu, T. M., Jacob, D. J., Wittrock, F., Burrows, J. P., Vrekoussis, M., and Henze, D. K.: Global
549	budgets of atmospheric glyoxal and methylglyoxal, and implications for formation of
550	secondary organic aerosols, J. Geophys. ResAtmos., 113, 17, 10.1029/2007jd009505,
551	2008.
552	Gabusi, V., Pisoni, E., and Volta, M.: Factor separation in air quality simulations, Ecol.
553	Model., 218, 383-392, 10.1016/j.ecolmodel.2008.07.030, 2008.
554 555 556 557	Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates of global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and Aerosols from Nature), Atmos. Chem. Phys., 6, 3181-3210, https://doi.org/10.5194/acp-6-3181-2006, 2006.
558	Guo, S., Hu, M., Zamora, M. L., Peng, J. F., Shang, D. J., Zheng, J., Du, Z. F., Wu, Z., Shao,
559	M., Zeng, L. M., Molina, M. J., and Zhang, R. Y.: Elucidating severe urban haze
560	formation in China, P. Natl. Acad. Sci. USA., 111, 17373-17378,





561	10.1073/pnas.1419604111, 2014.
562 563 564 565 566 567 568	 Hallquist, M., Wenger, J. C., Baltensperger, U., Rudich, Y., Simpson, D., Claeys, M., Dommen, J., Donahue, N. M., George, C., Goldstein, A. H., Hamilton, J. F., Herrmann, H., Hoffmann, T., Iinuma, Y., Jang, M., Jenkin, M. E., Jimenez, J. L., Kiendler-Scharr, A., Maenhaut, W., McFiggans, G., Mentel, T. F., Monod, A., Prevot, A. S. H., Seinfeld, J. H., Surratt, J. D., Szmigielski, R., and Wildt, J.: The formation, properties and impact of secondary organic aerosol: current and emerging issues, Atmos. Chem. Phys., 9, 5155-5236, 10.5194/acp-9-5155-2009, 2009.
569 570 571 572	Hastings, W. P., Koehler, C. A., Bailey, E. L., and De Haan, D. O.: Secondary organic aerosol formation by glyoxal hydration and oligomer formation: Humidity effects and equilibrium shifts during analysis, Environ. Sci. Technol., 39, 8728-8735, 10.1021/es0504461, 2005.
573 574 575	He, K. B., Yang, F. M., Ma, Y. L., Zhang, Q., Yao, X. H., Chan, C. K., Cadle, S., Chan, T., and Mulawa, P.: The characteristics of PM _{2.5} in Beijing, China, Atmos. Environ., 35, 4959-4970, 10.1016/s1352-2310(01)00301-6, 2001.
576 577 578 579 580 581 582 583 584 583 584	 Healy, R. M., Temime, B., Kuprovskyte, K., and Wenger, J. C.: Effect of Relative Humidity on Gas/Particle Partitioning and Aerosol Mass Yield in the Photooxidation of p-Xylene, Environ. Sci. Technol., 43, 1884-1889, 10.1021/es802404z, 2009. Hong, SY. and Lim, JO. J.: The WRF Single-Moment 6-Class Microphysics Scheme (WSM6), Asia-Pacific J. Atmos. Sci., 42, 129-151, 2006. Horowitz, L. W., Walters, S., Mauzerall, D. L., Emmons, L. K., Rasch, P. J., Granier, C., Tie, X. X., Lamarque, J. F., Schultz, M. G., Tyndall, G. S., Orlando, J. J., and Brasseur, G. P.: A global simulation of tropospheric ozone and related tracers: Description and evaluation of MOZART, version 2, J. Geophys. ResAtmos., 108, 4784, https://doi.org/10.1029/2002jd002853, 2003.
586 587 588 589 590 591	Huang, RJ., Zhang, Y., Bozzetti, C., Ho, KF., Cao, JJ., Han, Y., Daellenbach, K. R., Slowik, J. G., Platt, S. M., Canonaco, F., Zotter, P., Wolf, R., Pieber, S. M., Bruns, E. A., Crippa, M., Ciarelli, G., Piazzalunga, A., Schwikowski, M., Abbaszade, G., Schnelle-Kreis, J., Zimmermann, R., An, Z., Szidat, S., Baltensperger, U., El Haddad, I., and Prévôt, A. S. H.: High secondary aerosol contribution to particulate pollution during haze events in China, Nature, 514, 218-222, doi: 10.1038/nature13774, 2014.
592 593 594	Im, J. S., Saxena, V. K., and Wenny, B. N.: An assessment of hygroscopic growth factors for aerosols in the surface boundary layer for computing direct radiative forcing, J. Geophys. ResAtmos., 106, 20213-20224, 10.1029/2000jd000152, 2001.
595 596 597	Jacobson, M. Z.: Studying the effects of aerosols on vertical photolysis rate coefficient and temperature profiles over an urban airshed, J. Geophys. ResAtmos., 103, 10593-10604, 10.1029/98jd00287, 1998.
598 599 600 601 602 603	 Jacobson, M. Z.: Analysis of aerosol interactions with numerical techniques for solving coagulation, nucleation, condensation, dissolution, and reversible chemistry among multiple size distributions, J. Geophys. ResAtmos., 107, 23, 10.1029/2001jd002044, 2002. Janjic['], Z. I.: Nonsingular Implementation of the Mellor -Yamada Level 2.5 Scheme in the NCEP Meso Model, Ncep Office Note, 436, 2002.

604 Jia, L., and Xu, Y. F.: Effects of Relative Humidity on Ozone and Secondary Organic Aerosol





605 606	Formation from the Photooxidation of Benzene and Ethylbenzene, Aerosol Sci. Technol., 48, 1-12, 10.1080/02786826.2013.847269, 2014.
607	Kamens, R. M., Zhang, H. F., Chen, E. H., Zhou, Y., Parikh, H. M., Wilson, R. L., Galloway,
608	K. E., and Rosen, E. P.: Secondary organic aerosol formation from toluene in an
609	atmospheric hydrocarbon mixture: Water and particle seed effects, Atmos. Environ., 45,
610	2324-2334, 10.1016/j.atmosenv.2010.11.007, 2011.
611	Kan, H. D., Chen, R. J., and Tong, S. L.: Ambient air pollution, climate change, and
612	population health in China, Environ. Int., 42, 10-19, 10.1016/j.envint.2011.03.003, 2012.
613	Koehler, C. A., Fillo, J. D., Ries, K. A., Sanchez, J. T., and De Haan, D. O.: Formation of
614	secondary organic aerosol by reactive condensation of furandiones, aldehydes, and water
615	vapor onto inorganic aerosol seed particles, Environ. Sci. Technol., 38, 5064-5072,
616	10.1021/es034672b, 2004.
617	Kurokawa, J., Ohara, T., Morikawa, T., Hanayama, S., Janssens- Maenhout, G., Fukui, T.,
618	Kawashima, K., and Akimoto, H.: Emissions of air pollutants and greenhouse gases over
619	Asian regions during 2000–2008: Regional Emission inventory in Asia (REAS) version
620	2, Atmos. Chem. Phys., 13, 11019–11058, https://doi.org/10.5194/acp-13-11019-2013,
621	2013.
622	Li, G., Lei, W., Zavala, M., Volkamer, R., Dusanter, S., Stevens, P., and Molina, L. T.:
623	Impacts of HONO sources on the photochemistry in Mexico City during the
624	MCMA-2006/MILAGO Campaign, Atmos. Chem. Phys., 10, 6551-6567,
625	10.5194/acp-10-6551-2010, 2010.
626	Li, G., Bei, N., Tie, X., and Molina, L. T.: Aerosol effects on the photochemistry in Mexico
627	City during MCMA-2006/MILAGRO campaign, Atmos. Chem. Phys., 11, 5169-5182,
628	10.5194/acp-11-5169-2011, 2011a.
629	Li, G., Zavala, M., Lei, W., Tsimpidi, A. P., Karydis, V. A., Pandis, S. N., agaratna, M. R., and
630	Molina, L. T.: Simulations of organic aerosol concentrations in Mexico City using the
631	WRF-CHEM model during the MCMA-2006/MILAGRO campaign, Atmos. Chem.
632	Phys., 11, 3789-3809, 10.5194/acp-11-3789-2011, 2011b.
633	Li, G., Lei, W., Bei, N., and Molina, L. T.: Contribution of garbage burning to chloride and
634	PM _{2.5} in Mexico City, Atmos. Chem. Phys., 12, 8751-8761, 10.5194/acp-12-8751-2012,
635	2012.
636	Li, G. H., Bei, N. F., Zavala, M., and Molina, L. T.: Ozone formation along the California
637	Mexican border region during Cal-Mex 2010 field campaign, Atmos. Environ., 88,
638	370-389, 10.1016/j.atmosenv.2013.11.067, 2014.
639	Li, G., Bei, N., Cao, J., Huang, R., Wu, J., Feng, T., Wang, Y., Liu, S., Zhang, Q., Tie, X., and
640	Molina, L. T.: A possible pathway for rapid growth of sulfate during haze days in China,
641	Atmos. Chem. Phys., 17, 3301–3316, https://doi.org/10.5194/acp-17-3301-2017, 2017a.
642	Li, N., Fu, T. M., Cao, J. J., Lee, S. C., Huang, X. F., He, L. Y., Ho, K. F., Fu, J. S., and Lam,
643	Y. F.: Sources of secondary organic aerosols in the Pearl River Delta region in fall:
644	Contributions from the aqueous reactive uptake of dicarbonyls, Atmos. Environ., 76,
645	200-207, 10.1016/j.atmosenv.2012.12.005, 2013.
646 647	Liggio, J., Li, S. M., and McLaren, R.: Reactive uptake of glyoxal by particulate matter, J. Geophys. ResAtmos., 110, doi: 10.1029/2004jd005113, 2005.





648 649 650 651	Liu, Q., Jia, X. C., Quan, J. N., Li, J. Y., Li, X., Wu, Y. X., Chen, D., Wang, Z. F., and Liu, Y. G.: New positive feedback mechanism between boundary layer meteorology and secondary aerosol formation during severe haze events, Sci. Rep., 8, 8, 10.1038/s41598-018-24366-3, 2018.
652	Nguyen, T. B., Roach, P. J., Laskin, J., Laskin, A., and Nizkorodov, S. A.: Effect of humidity
653	on the composition of isoprene photooxidation secondary organic aerosol, Atmos. Chem.
654	Phys., 11, 6931-6944, 10.5194/acp-11-6931-2011, 2011.
655	Parrish, D. D. and Zhu, T.: Clean Air for Megacities, Science, 326, 674–675,
656	https://doi.org/10.1126/science.1176064, 2009.
657	Pathak, R. K., Wang, T., and Wu, W. S.: Nighttime enhancement of PM _{2.5} nitrate in
658	ammonia-poor atmospheric conditions in Beijing and Shanghai: Plausible contributions
659	of heterogeneous hydrolysis of N ₂ O ₅ and HNO ₃ partitioning, Atmos. Environ., 45,
660	1183-1191, 10.1016/j.atmosenv.2010.09.003, 2011.
661	Penner, J. E., Hegg, D., and Leaitch, R.: Unraveling the role of aerosols in climate change,
662	Environ. Sci. Technol., 35, 332A-340A, 10.1021/es0124414, 2001.
663	Poulain, L., Wu, Z., Petters, M. D., Wex, H., Hallbauer, E., Wehner, B., Massling, A.,
664	Kreiden- weis, S. M., and Stratmann, F.: Towards closing the gap between hygroscopic
665	growth and CCN activation for secondary organic aerosols - Part 3: Influence of the
666	chemical compo- sition on the hygroscopic properties and volatile fractions of aerosols,
667	Atmos. Chem. Phys., 10, 3775–3785, doi:10.5194/acp-10-3775-2010, 2010.
668	Pilinis, C., Seinfeld, J. H., and Grosjean, D.: Water content of atmospheric aerosols, Atmos.
669	Environ., 23, 1601–1606, 1989.
670	Quan, J. N., Gao, Y., Zhang, Q., Tie, X. X., Cao, J. J., Han, S. Q., Meng, J. W., Chen, P. F.,
671	and Zhao, D. L.: Evolution of planetary boundary layer under different weather
672	conditions, and its impact on aerosol concentrations, Particuology, 11, 34-40,
673	10.1016/j.partic.2012.04.005, 2013.
674	Randles, C. A., Russell, L. M., and Ramaswamy, V.: Hygroscopic and optical properties of
675	organic sea salt aerosol and consequences for climate forcing, Geophys. Res. Lett., 31, 4,
676	10.1029/2004gl020628, 2004.
677	Riemer, N., Vogel, H., Vogel, B., Schell, B., Ackermann, I., Kessler, C., and Hass, H.: Impact
678	of the heterogeneous hydrolysis of N ₂ O ₅ on chemistry and nitrate aerosol formation in
679	the lower troposphere under photosmog conditions, J. Geophys. ResAtmos., 108, 21,
680	10.1029/2002jd002436, 2003.
681	Seinfeld, J. H., Erdakos, G. B., Asher, W. E., and Pankow, J. F.: Modeling the formation of
682	secondary organic aerosol (SOA). 2. The predicted effects of relative humidity on
683	aerosol formation in the alpha-pinene-, beta-pinene-, sabinene-, Delta(3)-Carene-, and
684	cyclohexene-ozone systems, Environ. Sci. Technol., 35, 1806-1817, 10.1021/es001765+,
685	2001.
686	Seinfeld, J. H. and Pandis, S. N.: Atmospheric Chemistry and Physics: From Air Pollution to
687	Climate Change, John Wiley & Sons, USA, 1986.
688	Sun, Y., Wang, Z., Fu, P., Jiang, Q., Yang, T., Li, J., and Ge, X.: The impact of relative
689	humidity on aerosol composition and evolution processes during wintertime in Beijing,
690	China, Atmos. Environ., 77, 927-934, 10.1016/j.atmosenv.2013.06.019, 2013.





691	Stein, U., and Alpert, P.: Factor separation in numerical simulations, Journal of the
692	Atmospheric Science, 50, 2107-2115, 10.1175/1520-0469(1993)
693	050<2107:fsins>2.0.co;2, 1993.
694	Tao, J. C., Zhao, C. S., Ma, N., and Liu, P. F.: The impact of aerosol hygroscopic growth on
695	the single-scattering albedo and its application on the NO ₂ photolysis rate coefficient,
696	Atmos. Chem. Phys., 14, 12055-12067, 10.5194/acp-14-12055-2014, 2014.
697	Tie, X. X., Huang, R. J., Cao, J. J., Zhang, Q., Cheng, Y. F., Su, H., Chang, D., Poschl, U.,
698	Hoffmann, T., Dusek, U., Li, G. H., Worsnop, D. R., and O'Dowd, C. D.: Severe
699	Pollution in China Amplified by Atmospheric Moisture, Sci. Rep., 7, 8,
700	10.1038/s41598-017-15909-1, 2017.
701 702 703	Volkamer, R., Martini, F. S., Molina, L. T., Salcedo, D., Jimenez, J. L., and Molina, M. J.: A missing sink for gas-phase glyoxal in Mexico City: Formation of secondary organic aerosol, Geophys. Res. Lett., 34, 5, 10.1029/2007gl030752, 2007.
704	Wang, G., Zhang, R., Gomez, M. E., Yang, L., Zamora, M. L., Hu, M., Lin, Y., Peng, J., Guo,
705	S., and Meng, J.: Persistent sulfate formation from London Fog to Chinese haze, P. Natl.
706	Acad. Sci. USA., 113, 13630–13635, 2016.
707	Wang, Y. S., Yao, L., Wang, L. L., Liu, Z. R., Ji, D. S., Tang, G. Q., Zhang, J. K., Sun, Y., Hu,
708	B., and Xin, J. Y.: Mechanism for the formation of the January 2013 heavy haze
709	pollution episode over central and eastern China, Sci. China-Earth Sci., 57, 14-25,
710	10.1007/s11430-013-4773-4, 2014.
711 712	Weinhold, B.: Ozone nation – EPA standard panned by the people, Environ. Health. Persp., 116, A302–A305, 2008.
713	Wu, J. R., Li, G. H., Cao, J. J., Bei, N. F., Wang, Y. C., Feng, T., Huang, R. J., Liu, S. X.,
714	Zhang, Q., and Tie, X. X.: Contributions of trans-boundary transport to summertime air
715	quality in Beijing, China, Atmos. Chem. Phys., 17, 2035-2051,
716	10.5194/acp-17-2035-2017, 2017.
717	Wu, P., Ding, Y. H., and Liu, Y. J.: Atmospheric circulation and dynamic mechanism for
718	persistent haze events in the Beijing-Tianjin-Hebei region, Adv. Atmos. Sci., 34,
719	429-440, 10.1007/s00376-016-6158-z, 2017.
720	Wu, Z. J., Wang, Y., Tan, T. Y., Zhu, Y. S., Li, M. R., Shang, D. J., Wang, H. C., Lu, K. D.,
721	Guo, S., Zeng, L. M., and Zhang, Y. H.: Aerosol Liquid Water Driven by Anthropogenic
722	Inorganic Salts: Implying Its Key Role in Haze Formation over the North China Plain,
723	Environ. Sci. Technol. Lett., 5, 160-166, 10.1021/acs.estlett.8b00021, 2018.
724	Xing, L., Wu, J., Elser, M., Tong, S., Liu, S., Li, X., Liu, L., Cao, J., Zhou, J., El-Haddad, I.,
725	Huang, R., Ge, M., Tie, X., Prévôt, A. S. H., and Li, G.: Wintertime secondary organic
726	aerosol formation in Beijing-Tianjin-Hebei (BTH): Contributions of HONO sources and
727	heterogeneous reactions, Atmos. Chem. Phys. Discuss., 2018, 1-25,
728	10.5194/acp-2018-770, 2018.
729	Xu, J. W., Martin, R. V., van Donkelaar, A., Kim, J., Choi, M., Zhang, Q., Geng, G., Liu, Y.,
730	Ma, Z., Huang, L., Wang, Y., Chen, H., Che, H., Lin, P., and Lin, N.: Estimating
731	ground-level PM2.5 in eastern China using aerosol optical depth determined from the
732	GOCI satellite instrument, Atmos. Chem. Phys., 15, 13133-13144,
733	10.5194/acp-15-13133-2015, 2015.





734 735 736 737	Zhang, L., Sun, J. Y., Shen, X. J., Zhang, Y. M., Che, H., Ma, Q. L., Zhang, Y. W., Zhang, X. Y., and Ogren, J. A.: Observations of relative humidity effects on aerosol light scattering in the Yangtze River Delta of China, Atmos. Chem. Phys., 15, 8439-8454, 10.5194/acp-15-8439-2015, 2015.
738	Zhang, Q., Streets, D. G., Carmichael, G. R., He, K. B., Huo, H., Kannari, A., Klimont, Z.,
739	Park, I. S., Reddy, S., Fu, J. S., Chen, D., Duan, L., Lei, Y., Wang, L. T., and Yao, Z. L.:
740	Asian emissions in 2006 for the NASA INTEX-B mission, Atmos. Chem. Phys., 9,
741	5131-5153, https://doi.org/10.5194/acp-9-5131-2009, 2009.
742 743 744	Zhang, R. H., Li, Q., and Zhang, R. N.: Meteorological conditions for the persistent severe fog and haze event over eastern China in January 2013, Sci. China-Earth Sci., 57, 26-35, 10.1007/s11430-013-4774-3, 2014.
745	Zhang, R., Jing, J., Tao, J., Hsu, SC., Wang, G., Cao, J., Lee, C. S. L., Zhu, L., Chen, Z.,
746	Zhao, Y., and Shen, Z.: Chemical characterization and source apportionment of PM _{2.5} in
747	Beijing: seasonal perspective, Atmos. Chem. Phys., 13, 7053–7074,
748	https://doi.org/10.5194/acp-13-7053-2013, 2013.
749	Zhao, J., Levitt, N. P., Zhang, R. Y., and Chen, J. M.: Heterogeneous reactions of
750	methylglyoxal in acidic media: implications for secondary organic aerosol formation,
751	Environ. Sci. Technol., 40, 7682–7687, 2006.
752 753 754 755 756 757	Zhou, Y., Zhang, H. F., Parikh, H. M., Chen, E. H., Rattanavaraha, W., Rosen, E. P., Wang, W. X., and Kamens, R. M.: Secondary organic aerosol formation from xylenes and mixtures of toluene and xylenes in an atmospheric urban hydrocarbon mixture: Water and particle seed effects (II), Atmos. Environ., 45, 3882-3890, 10.1016/j.atmosenv.2010.12.048, 2011.





Table 1 WRF-CHEM model configurations

Regions	East Asia
Simulation period	December 05, 2015 - January 04, 2016
Domain size	400×400
Domain center	35°N, 114°E
Horizontal resolution	12km × 12 km
Vertical resolution	35 vertical levels with a stretched vertical grid with spacing ranging from 30 m near the surface, to 500 m at 2.5 km and 1 km above 14 km
Microphysics scheme	WSM 6-class graupel scheme (Hong and Lim, 2006)
Boundary layer scheme	MYJ TKE scheme (Janjić, 2002)
Surface layer scheme	MYJ surface scheme (Janjić, 2002)
Land-surface scheme	Unified Noah land-surface model (Chen and Dudhia, 2001)
Longwave radiation scheme	Goddard longwave scheme (Chou and Suarez, 2001)
Shortwave radiation scheme	Goddard shortwave scheme (Chou and Suarez, 1999)
Meteorological boundary and initial conditions	NCEP 1°×1° reanalysis data
Chemical initial and boundary conditions	MOZART 6-hour output (Horowitz et al., 2003)
Anthropogenic emission inventory	Developed by Zhang et al. (2009) and Li et al. (2017), 2012 base year, and SAPRC-99 chemical mechanism
Biogenic emission inventory	MEGAN model developed by Guenther et al. (2006)
Model spin-up time	28 hours





765	Figure Captions
766 767 768 769 770 771 772	 Figure 1 (a) WRF-CHEM simulation domain with topography and (b) North China Plain. In (a), the blue circles represent centers of cities with ambient monitoring sites and the size of circles denotes the number of ambient monitoring sites of cities. In (b), the red capitals denote six typical polluted cities in NCP. A: Beijing; B: Tianjin; C: Shijiazhuang; D: Baoding; E: Tangshan; F: Chengde. The blue numbers denote the CERN sites with the hourly RH measurement. 1: Jiaozhouwan; 2: Yucheng; 3: Luancheng.
773 774 775 776	Figure 2 Scatter plots of near-surface [PM _{2.5}] and RH at six typical polluted cities in NCP during the 2015 wintertime. The red diamond shows the bin average of near-surface [PM _{2.5}], and the red line denotes the variation of the bin average of near-surface [PM _{2.5}] with RH.
777 778 779	Figure 3 Comparison of measured (black dots) and predicted (red line) diurnal profiles of the RH in (a) Luancheng, (b) Yucheng, and (c) Jiaozhouwan from 05 December 2015 to 04 January 2016.
780 781	Figure 4 Spatial distribution of (a) NCEP reanalyzed and (b) simulated RH averaged from 05 December 2015 to 04 January 2016.
782 783	Figure 5 (a) absolute and (b) relative AOD contribution caused by the ALW, averaged from 05 December 2015 to 04 January 2016.
784 785	Figure 6 Near-surface [PM _{2.5}] contribution caused by the ALW-ARF, averaged from 05 December 2015 to 04 January 2016.
786 787 788	Figure 7 Average variations of AOD and Reff in f_{base} (red line) and $f_{alw-rad0}$ (blue line) as a function of bin [PM _{2.5}] in NCP during daytime from 05 December 2015 to 04 January 2016.
789 790 791 792	Figure 8 Average (a) percentage decrease of SWDOWN at the ground surface, (b) decrease of TSFC, (c) percentage decrease of PBLH, and (d) percentage contribution of near-surface [PM _{2.5}] caused by the ALW-ARF, as a function of the near-surface [PM _{2.5}] in NCP during daytime from 05 December 2015 to 04 January 2016.
793 794 795	Figure 9 Average variations of daytime NO ₂ photolysis and O ₃ concentration at 1 st and 5 th model layer (around 18 m and 420 m above the ground surface, respectively) caused by ALW from 05 December 2015 to 04 January 2016 in NCP.
796 797	Figure 10 Near-surface PM _{2.5} contribution caused by the ALW-J, averaged from 05 December 2015 to 04 January 2016 in NCP.
798 799	Figure 11 Near-surface sulfate, nitrate, and ammonium contribution caused by the ALW-HET, averaged from 05 December 2015 to 04 January 2016.
800	Figure 12 Spatial distribution of NH ₃ emission rate in December.
801 802	Figure 13 Near-surface SOA contribution caused by the ALW-HET, averaged from 05 December 2015 to 04 January 2016.
803 804	Figure 14 Near-surface PM _{2.5} contribution caused by the ALW-HET, averaged from 05 December 2015 to 04 January 2016.
805	Figure 15 Average percentage decrease of (a) BC and (b) POA concentrations caused by the





806 807		ALW-HET, as a function of the near-surface [PM _{2.5}] in NCP from 05 December 2015 to 04 January 2016.
808 809 810 811	Figure	16 Average variations of (a) AOD and (b) Reff in f_{base} (red line) and $f_{alw-het0}$ (blue line), respectively, and average percentage decrease of near-surface (c) aerosol number concentration and (d) surface area caused by the ALW-HET, as a function of bin [PM _{2.5}] in NCP from 05 December 2015 to 04 January 2016.
812 813 814 815	Figure	17 Average (a) percentage decrease of SWDOWN at the ground surface, (b) decrease of TSFC, and (c) percentage decrease of PBLH caused by the ALW-HET, as a function of the near-surface $[PM_{2.5}]$ in NCP during daytime from 05 December 2015 to 04 January 2016.
816 817	Figure	18 Near-surface PM _{2.5} contribution caused by the ALW-TOT, averaged from 05 December 2015 to 04 January 2016 in NCP.
818 819 820	Figure	19 Average contributions to near-surface $[PM_{2.5}]$ caused by the ALW-TOT (black line), ALW-HET (red line), ALW-RAD (green line), and ALW-J (blue line), respectively, as a function of the RH in NCP from 05 December 2015 to 04 January 2016.
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(a), the blue circles represent centers of cities with ambient monitoring sites and the size of
blue circles denotes the number of ambient monitoring sites of cities. In (b), the red capitals
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841Figure 2 Scatter plots of near-surface $[PM_{2.5}]$ and RH at six typical polluted cities in NCP842during the 2015 wintertime. The red diamond shows the bin average of near-surface $[PM_{2.5}]$,843and the red line denotes the variation of the bin average of near-surface $[PM_{2.5}]$, with RH.







Figure 3 Comparison of measured (black dots) and predicted (red line) diurnal profiles of the
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Figure 4 Spatial distribution of (a) NCEP reanalyzed and (b) simulated RH averaged from 05December 2015 to 04 January 2016.







Figure 5 (a) absolute and (b) relative AOD contribution caused by the ALW, averaged from05 December 2015 to 04 January 2016.







Figure 6 Near-surface PM_{2.5} contribution caused by the ALW-ARF, averaged from 05
December 2015 to 04 January 2016 in NCP.







888 Figure 7 Average variations of AOD and Reff in f_{base} (red line) and $f_{alw-rad0}$ (blue line) **889** as a function of bin [PM_{2.5}] in NCP during daytime from 05 December 2015 to 04 January **890** 2016.

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Figure 8 Average (a) percentage decrease of SWDOWN at the ground surface, (b) decrease of
TSFC, (c) percentage decrease of PBLH, and (d) percentage contribution of near-surface
[PM_{2.5}] caused by the ALW-ARF, as a function of the near-surface [PM_{2.5}] in NCP during
daytime from 05 December 2015 to 04 January 2016.







Figure 9 Average variations of daytime NO₂ photolysis and O₃ concentration at 1st and 5th
model layer (around 18 m and 420 m above the ground surface, respectively) caused by ALW
from 05 December 2015 to 04 January 2016.







Figure 10 Near-surface PM_{2.5} contribution caused by the ALW-J, averaged from 05
December 2015 to 04 January 2016.







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Figure 11 Near-surface sulfate, nitrate, and ammonium contribution caused by the ALW-HET,averaged from 05 December 2015 to 04 January 2016.







933 Figure 12 Spatial distribution of NH₃ emission rate in December.







Figure 13 Near-surface SOA contribution caused by the ALW-HET, averaged from 05December 2015 to 04 January 2016.







Figure 14 Near-surface [PM_{2.5}] contribution caused by the ALW-HET, averaged from 05
December 2015 to 04 January 2016.







Figure 15 Average percentage decrease of (a) BC and (b) POA concentrations caused by the ALW-HET, as a function of the near-surface $[PM_{2.5}]$ in NCP from 05 December 2015 to 04 January 2016.

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969Figure 16 Average variations of (a) AOD and (b) Reff in f_{base} (red line) and $f_{alw-het0}$ 970(blue line), respectively, and average percentage decrease of near-surface (c) aerosol number971concentration and (d) surface area caused by the ALW-HET, as a function of bin $[PM_{2.5}]$ in972NCP from 05 December 2015 to 04 January 2016.







Figure 17 Average (a) percentage decrease of SWDOWN at the ground surface, (b) decrease
of TSFC, and (c) percentage decrease of PBLH caused by the ALW-HET, as a function of the
near-surface [PM_{2.5}] in NCP during daytime from 05 December 2015 to 04 January 2016.







Figure 18 Near-surface PM_{2.5} contribution caused by the ALW-TOT, averaged from 05
December 2015 to 04 January 2016.







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Figure 19 Average contributions to the near-surface [PM_{2.5}] caused by the ALW-TOT (black
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