

1 **Characteristic of atmospheric mercury in a suburban area of East China: sources,**
2 **formation mechanisms, and regional transport**

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13

14 **Abstract**

15 Speciated atmospheric mercury including gaseous elemental mercury (GEM), gaseous oxidized
16 mercury (GOM), and particulate-bound mercury (PBM) were measured continuously for one year
17 period at a suburban site, representing a regional transport intersection zone, in East China. Annual
18 mean concentrations of GEM, PBM, and GOM reached 2.77 ng/m³, 60.8 pg/m³, and 82.1 pg/m³,
19 respectively. GEM concentrations were elevated in all the seasons except autumn. High mercury
20 concentrations were related to winds from the south, southwest, and north of the measurement site.
21 Combining analysis results from using various source apportionment methods, it was found that
22 GEM concentration was higher when quasi-local sources dominated over long-range transport. Six
23 source factors belonging to the anthropogenic sources of GEM were identified, including the
24 common sectors previously identified (industrial and biomass burning, coal combustion, iron and
25 steel production, cement production, and incineration), as well as an additional factor of shipping
26 emission (accounting for 19.5% of the total), which was found to be important in East China where
27 marine vessel shipping activities are intense. Emissions of GEM from natural surfaces were also
28 found to be as important as those from anthropogenic sources for GEM observed at this site.
29 Concurrences of high GOM concentrations with elevated O₃ and temperature, along with the lagged

30 variations of GEM and GOM during daytime demonstrated that the very high GOM concentrations
31 were partially ascribed to intense in situ oxidation of GEM. Strong gas-particle partitioning was also
32 identified when $PM_{2.5}$ was above a threshold value, in which case GOM decreased with increasing
33 $PM_{2.5}$.

34

35 **1. Introduction**

36 Mercury (Hg) is a global pollutant of great concerns for environment and human health (Wright
37 et al., 2018). Atmospheric mercury is operationally divided into three forms, i.e. gaseous elemental
38 mercury (GEM), particulate-bound mercury (PBM), and gaseous oxidized mercury (GOM). GEM
39 is the predominant form in the atmosphere (>90%), while GOM and PBM generally have similar
40 concentrations (Xu et al., 2017; Mao et al., 2016). GEM in the atmosphere is relatively stable, having
41 a long lifetime of 0.5-2 year, and can thus transport globally before being oxidized and/or removed
42 from the atmosphere via wet and dry depositions (Schroeder and Munthe, 1998). In contrast, GOM
43 and PBM can be rapidly wiped out from the atmosphere due to their high reactivity and solubility.

44 Both natural processes and anthropogenic activities release mercury into the atmosphere
45 (Zhang et al., 2017; Pirrone et al., 2010). Natural sources of mercury include the ocean volatilization,
46 volcanic eruption, evasion from soils and vegetation, geothermal activities, and weathering minerals.
47 Re-emissions of mercury that previously deposited onto the environmental surfaces are also
48 considered as natural sources. As for the anthropogenic emission sources of mercury, coal
49 combustion, non-ferrous smelters, cement production, waste incineration, and mining are
50 considered to be the main sources. After being emitted into the atmosphere, mercury undergoes
51 speciation which plays an important role in its biogeochemical cycle. Previous studies suggest that
52 the oxidation of GEM in the terrestrial environments was generally initiated by O_3 and OH radicals
53 (Mao et al., 2016). Atomic bromine (Br) and bromine monoxide (BrO) are two additional oxidation
54 agents in the marine atmosphere (Tang et al., 2018). Observational studies of GOM in the polar
55 regions (Wang et al., 2009; Ye et al., 2016) and in the subtropical marine boundary layer (Kim et al.,
56 2009) as well as atmospheric modeling studies about mercury cycling (Wang et al., 2016; Zhang et
57 al., 2015) have considered Br to be an important oxidant of GEM. (Wang et al., 2014) even reported
58 that Br is the primary oxidant of GEM in tropical marine boundary layer (MBL). However, it still
59 remains unknown and controversial about the speciation and quantification of the $GEM+O_3$

60 products, and the reaction of GEM+OH is still under huge debate between theoretical and
61 experimental studies due to the lacking of mechanisms consistent with thermochemistry (Zhu et al.,
62 2015). The levels of GEM in the atmosphere are mainly controlled by various emission sources,
63 redox reactions, and foliar uptake (Streets et al., 2005; Wright et al., 2016; Zhu et al., 2016). A portion
64 of GOM in air will be adsorbed onto particulate matter due to its high water solubility and relatively
65 strong surface adhesion properties (Liu et al., 2010). The levels of GOM and PBM in the atmosphere,
66 especially in areas far away from anthropogenic mercury sources, are mainly controlled by GEM
67 oxidation processes and atmospheric particulate matter levels, with the former factors affecting the
68 production of GOM and the latter affecting the gas-particle partitioning (Zhang et al., 2013).

69 Anthropogenic mercury emissions in Asia accounted for more than 50% of the global total
70 mercury emissions (Pacyna et al., 2016), among these approximately 27% were from mainland
71 China (Hui et al., 2017). The Yangtze River Delta (YRD) is one of the most industrialized and
72 urbanized regions in China. Early field measurements in urban Shanghai suggested that TGM was
73 most likely derived from coal fired power plants, smelters and industrial activities (Friedli et al.,
74 2011). Field measurements in urban Nanjing indicated that natural sources were important while
75 most sharp peaks of TGM were caused by anthropogenic sources (Zhu et al., 2012). Modeling
76 studies of atmospheric mercury in eastern China showed that natural emissions, accounting for 36.6%
77 of the total emissions, were the most important source for GEM in eastern China (Zhu et al., 2015).
78 Measurements made at Chongming (an island belonging to Shanghai) observed a downward trend
79 of GEM concentrations from 2014 to 2016, likely caused by the reduction of domestic emissions
80 (Tang et al., 2018). Source apportionment studies for atmospheric mercury are very limited in this
81 region (Zhu et al., 2015), although some studies were made in the other regions of China (Wan et
82 al., 2009; Fu et al., 2019)

83 In this study, one-year continuous measurements of GEM, GOM, and PBM were conducted at
84 Dianshan Lake Station (DSL), a suburban site in Shanghai. DSL is located in the junction area of
85 Shanghai, Zhejiang, and Jiangsu provinces and is close to the East China Sea (ECS). There are few
86 local mercury sources in this area, making it a unique station for studying the main pollution sources
87 and transport pathways of Hg. The mercury data were analyzed for (1) investigating the impact of
88 meteorology on mercury distribution, (2) exploring the GEM oxidation process, (3) revealing the
89 GOM adsorption process to ambient particles, and (4) identifying potential mercury sources.

90 Knowledge gained from this study provides scientific basis for establishing future mercury emission
91 control policies in this region.

92

93 **2. Materials and methods**

94 **2.1. Site description**

95 Field sampling was conducted on the top of a four-story building (~ 14 m above ground) at a
96 super site (DSL) located in the west area of Shanghai, and nearby the Dianshan Lake in Qingpu
97 district, noting that the lake is the largest freshwater body in Shanghai with a total area of 62 square
98 kilometers (Fig. 1). This supersite is carefully maintained by Shanghai Environmental Monitoring
99 Center (SEMC) and the building is purely used for atmospheric monitoring. There are no large point
100 sources within ~20 kilometers of the site, with the total GEM emissions within a 20 km of the site
101 being estimated as ~100kg/yr. The distance between the sampling site and the coastal lines is ~50
102 km so the site is capable of capturing land-sea circulation. Its special geographical location, i.e. at
103 the junction of Shanghai, Zhejiang and Jiangsu provinces, makes it possible to receive the air masses
104 from all the populous regions. In addition, the site is located at the typical outflow path from East
105 China to the Pacific Ocean. The red dots in Fig. 1 represent the amount of atmospheric Hg emitted
106 by anthropogenic activities in each provinces in 2014 (Wu et al., 2016). The emission intensities of
107 anthropogenic Hg in China were higher in the north and lower in the south. The atmospheric Hg
108 emissions by province in 2014 are listed in Table S1.

109

110 **2.2 Measurements of atmospheric mercury**

111 GEM, GOM, and PBM were measured from June 2015 to May 2016 using the Tekran
112 2537B/1130/1135 system (Tekran Inc., Canada), an instrument that has been widely used worldwide
113 (Landis and Keeler, 2002). In general, GEM, GOM, and PBM in the atmosphere were collected by
114 dual gold cartridges, KCl-coated annular denuder, and regenerable quartz fiber filter, respectively.
115 In this study, GEM was collected at an interval of 5 minutes with a flow rate of 1 L/min, while GOM
116 and PBM were collected at an interval of 2 hours with a flow rate of 10L/min. After the collection,
117 all mercury species were thermally decomposed to Hg⁰ immediately and measured by cold vapor
118 atomic fluorescence spectroscopy (CVAFS). GEM concentrations were expressed in ng/m³, while
119 GOM and PBM were in pg/m³ at standard temperature of 273.14K and pressure of 1013 hPa.

120 Quality control procedures are strictly followed. The KCl-coated denuder, Teflon coated glass
121 inlet, and impactor plate were replaced weekly and quartz filters were replaced monthly. Before
122 sampling, denuders and quartz filters were prepared and cleaned according to the methods in Tekran
123 technical notes. The Tekran 2537B analyzer was routinely calibrated using its internal permeation
124 source every 47 hours, and was also cross-calibrated every 3 months against an external temperature
125 controlled Hg vapor standard. Two-point calibrations (including zero calibration and span
126 calibration) were performed separately for each pure gold cartridge. Manual injections were
127 performed to evaluate these automated calibrations using the standard saturated mercury vapor.
128 Extremely high concentrations (especially for GOM and PBM) were occasionally observed. If the
129 values were several times higher than the previous hour, those data were regarded as outliers and
130 were excluded in the data analysis.

131

132 **2.3 Measurements of other air pollutants and meteorological parameters**

133 Water-soluble inorganic anions (SO_4^{2-} , NO_3^- , Cl^-) and cations (K^+ , Mg^{2+} , Ca^{2+} , NH_4^+) in $\text{PM}_{2.5}$
134 were simultaneously monitored by the Monitor for Aerosols and Gases in ambient Air (MARGA)
135 at the resolution of one hour. Ambient air was drawn into a sampling box at a flow rate of 16.7L/min.
136 After removing the water-soluble gases by an absorbing liquid, a supersaturation of water vapor
137 induced the particles in the airflow to grow into droplets, and then the droplets were collected and
138 transported into the analytical box which contains two ion chromatograph systems for the
139 determination of the water-soluble ions in $\text{PM}_{2.5}$.

140 Heavy metals (Pb, Fe, K, Ba, Cr, Se, Cd, Ag, Ca, Mn, Cu, As, Hg, Ni, Zn, V) in $\text{PM}_{2.5}$ were
141 measured hourly using the Xact-625 Ambient Metals Monitor (Cooper Environmental services,
142 Beaverton, OR, USA), which is a sampling and analyzing X-ray fluorescence spectrometer designed
143 for online measurements of particulate elements. In this study, ambient air was sampled at a flow
144 rate of 16.7L/min and the particles were collected onto a Teflon filter tape. Then the filter tape was
145 moved into the spectrometer, where it was illuminated with an X-ray tube under three excitation
146 conditions and the excited X-ray fluorescence was measured by a silicon drift detector. Daily
147 advanced quality assurance checks were performed during 30 minutes after midnight to monitor
148 shifts in the calibration.

149 The hourly meteorological data including air temperature, relative humidity (RH), wind speed,

150 and wind direction were simultaneously monitored at the observation site by the automatic weather
151 station (AWS). The data of planetary boundary layer (PBL) height were retrieved from the U.S.
152 National Oceanic and Atmospheric Administration (<https://ready.arl.noaa.gov/READYamet.php>).
153 Atmospheric ozone (O₃) concentration was continuously measured using Thermo Fisher 49i, which
154 operates on the principle that ozone molecules absorb UV light at a wavelength of 254 nm. The
155 ambient carbon monoxide (CO) and PM_{2.5} concentrations were measured by Thermo Fisher 481tle
156 and Thermo Fisher 1405F, respectively. All the data were averaged into hour values.

157

158 **2.4 Potential source contribution function (PSCF)**

159 PSCF is a useful tool to diagnose the possible source areas with regard to the levels of air
160 pollutants when setting a contamination concentration threshold at the receptor site. Back trajectory
161 models are used to simulate the airflows. The principle of PSCF is to calculate the ratio of the total
162 number of back trajectory segment endpoints in a grid cell (*i, j*) exceeding the threshold
163 concentration (*m_{ij}*) to the total number of back trajectory segment endpoints in this grid cell (*i, j*)
164 during the whole sampling period (*n_{ij}*) (Equation (1)) (Schroeder and Munthe, 1998; Cheng et al.,
165 2015):

$$166 \text{PSCF}_{ij} = \frac{m_{ij}}{n_{ij}} \quad (1)$$

167 When a particular cell is associated with a small number of endpoint, weighting function (*w_{ij}*)
168 is applied to reduce this uncertainty and the value of *w_{ij}* is shown in Equation (2) (Fu et al., 2011),
169 in which *N_{ave}* is the mean *n_{ij}* of all grid cells with *n_{ij}* greater than zero. *PSCF_{ij}* is multiplied by *w_{ij}* to
170 derive the weighted PSCF values.

171

$$172 w_{ij} = \begin{cases} 1.0, & N_{ij} > 3N_{ave} \\ 0.7, & 3N_{ave} > N_{ij} > 1.5N_{ave} \\ 0.4, & 1.5N_{ave} > N_{ij} > N_{ave} \\ 0.2, & N_{ave} > N_{ij} \end{cases} \quad (2)$$

173 In this study, we set the threshold concentration as the mean value of the whole sampling period.
174 The mean GEM, PBM, and GOM concentrations were 2.77 ng/m³, 60.8 pg/m³, and 82.1 pg/m³,
175 respectively. The HYSPLIT (HYbrid Single-Particle Lagrangian Integrated Trajectory) model is
176 applied for calculating air mass backward trajectories (Draxler and Rolph, 2012). The model was
177 run online at the NOAA ARL READY Website using the meteorological data archives of Air

178 Resource Laboratory (ARL). The meteorological input data used in the model was obtained from
179 NCEP (National Centers for Environmental Prediction)'s global data assimilation system (GDAS)
180 with a horizontal resolution of $0.5^\circ \times 0.5^\circ$. In this study, 72-hours back trajectories were calculated
181 at 500m AGL (above ground level) and the cell size was set as $0.5^\circ \times 0.5^\circ$.

182

183 2.5 Positive matrix factorization (PMF)

184 The PMF model is widely used to quantitatively determine the source contributions of specific
185 air pollutants (Gibson et al., 2015). The essential principle of PMF is that the concentration of each
186 sample is determined by source profiles and different contributions. The equation of the PMF model
187 is shown as Equation (3):

$$188 \quad X_{ij} = \sum_{k=1}^P g_{ik} f_{kj} + e_{ij} \quad (3)$$

189 Where X_{ij} is the concentration of the j th contamination at the receptor site in the i th sample. g_{ik}
190 represent the contribution of the k th factor on the i th sample, f_{kj} is used to express the mass fraction
191 of the j th contamination in the k th factor, P is the number of factors, which represent pollution
192 sources, and e_{ij} is the residual for each measurement or model error.

193 Before the model determines the optimal non-negative factor contributions and factor profiles,
194 an objective function, which is the sum of the square difference between the measured and modeled
195 concentrations weighted by the concentration uncertainties, has to be minimized (Cheng et al., 2015).

196 The equation that determines the objective function is given by Equation (4):

$$197 \quad Q = \sum_{i=1}^n \sum_{j=1}^m \left(\frac{X_{ij} - \sum_{k=1}^P A_{ik} F_{kj}}{S_{ij}} \right)^2 \quad (4)$$

198 where X_{ij} is the ambient concentration of the j th pollutant in the i th sample (m and n represent the
199 total number pollutants and samples, respectively). A_{ik} is the contribution of the k th factor on the i th
200 sample and F_{kj} is the mass fraction of the j th pollutant in the k th factor. S_{ij} is the uncertainty of the
201 j th pollutant on the i th measurement, and P is the number of factors, which imply the pollutant
202 sources. In this study, the number of factors from 3 to 8 was examined with the optimal solutions
203 determined by the slope of the Q value versus the number of factors. For each run, the stability and
204 reliability of the outputs were assessed by referring to the Q value, residual analysis, and correlation
205 coefficients between observed and predicted concentrations. Finally, a 6-factor solution, which
206 showed the most stable results and gave the most reasonable interpretation, was chosen. Before
207 running the model, a dataset including unique uncertainty values of each data point was created and

208 digested into the model. The error fraction was assumed to be 15% of concentrations for GEM and
209 10% of concentrations for the other compounds (Xu et al., 2017), the missing data were excluded
210 and the total number of samples was 3526.

211

212 **3. Results and discussion**

213 **3.1. Characteristics of atmospheric mercury species**

214 Fig. 2 displays the time series of atmospheric GEM, PBM, and GOM concentrations during 1
215 June, 2015 to 31 May, 2016 at DSL. The annual average concentrations of GEM, PBM, and GOM
216 at DSL were 2.77 ± 1.36 ng/m³, 60.8 ± 67.4 pg/m³, and 82.1 ± 115.4 pg/m³, respectively. As shown in
217 Table 1, the levels of GEM and PBM in this study were lower than at some sites in China by a factor
218 of 2-7, such as rural Miyun, suburban Xiamen, and urban Guiyang (Xu et al., 2015;Fu et al., 2011).
219 However, compared to the studies conducted in urban and rural areas abroad such as New York
220 (Wang et al., 2009), Chicago (Gratz et al., 2013), and Nova Scotia (Cheng et al., 2014b), the
221 concentrations of GEM and PBM in the suburbs of Shanghai were much higher by a factor of 1-3
222 and 3-8, respectively. Different from GEM and PBM, the GOM concentrations at DSL were higher
223 than at all the other Chinese sites and sites around the world listed in Table 1. The mean GOM
224 concentration in this study (82.1 pg/m³) was even higher than that in Guiyang (35.7 pg/m³), where
225 the emissions of GEM and GOM were quite intense due to the massive primary emission sources
226 such as coal-fired power plants and cement plants (Fu et al., 2011). The abnormally high GOM
227 concentrations observed in this study were likely dominated by strong primary emissions.

228 The monthly patterns of GEM, PBM, and GOM during the whole sampling period are shown
229 in Fig. 3. The seasonal mean GEM concentrations were slightly higher in winter (2.88 ng/m³) and
230 summer (2.87 ng/m³) than in spring (2.73 ng/m³) and autumn (2.63 ng/m³), with the highest monthly
231 mean value of 3.19 ng/m³ in June and the lowest of 2.39 ng/m³ in March. Statistical test showed that
232 no significant differences of the seasonal variations of GEM concentrations among spring, summer,
233 and winter were observed (Table S2). This was different from many urban and remote sites in China,
234 such as Guiyang, Xiamen, and Mt. Changbai, where GEM showed significantly high concentrations
235 in cold than warm seasons (Wang et al., 2016;Xu et al., 2015;Fu et al., 2012). The relatively high
236 GEM concentrations during the cold season in China should be attributed to the increases of energy
237 consumption (Fu et al., 2015). In this study, GEM concentration in summer was comparable to that

238 in winter, which was likely attributed to the strong natural mercury emissions (e.g. soils, vegetations,
239 and water) due to the elevated temperature in summer (Liu et al., 2016). The seasonal mean PBM
240 concentrations were the highest in winter (93.5 pg/m^3) while the lowest in summer (35.7 pg/m^3),
241 and moderate in autumn (56.8 pg/m^3) and spring (51.6 pg/m^3), with the highest monthly mean value
242 of 109.4 pg/m^3 in January and the lowest of 28.9 pg/m^3 in September. This seasonal pattern was
243 consistent with those at the other sites in China such as Beijing and Nanjing (Kim et al., 2009; Lynam
244 and Keeler, 2005). PBM concentrations at low altitude sites in the Northern Hemisphere were
245 commonly enhanced in winter, which was ascribed to intense emissions from residential heating,
246 the reduction of wet scavenging processes, enhanced gas-particle partitioning of atmospheric
247 mercury under low temperature, etc. (Rutter and Schauer, 2007). As for GOM, its seasonal mean
248 concentrations were the highest in winter (124.0 pg/m^3), followed by summer (77.3 pg/m^3), spring
249 (68.1 pg/m^3), and autumn (61.0 pg/m^3). GOM concentrations in summer were much lower than that
250 in winter, but were higher than in spring and autumn. Secondary transformation from GEM in
251 warm/hot seasons and slow dispersion in winter can explain such season variations, as further
252 discussed below.

253 Fig. 4 shows the diurnal variations of GEM, PBM, and GOM during the whole sampling period.
254 To ensure the time resolutions were consistent among all the three mercury species, the temporal
255 resolution of measured GEM was converted from 5 minutes to the two-hour average. As shown in
256 Fig. 4, GEM concentrations were higher during daytime with the maximum in the morning at around
257 10:00 and minimum in the midnight at around 02:00. The diurnal trends of GOM were similar to
258 that of GEM, except that the minimum GOM occurred at around 20:00 in the evening. The diurnal
259 trends of PBM were different from those of GEM and GOM, exhibiting relatively higher
260 concentrations during nighttime. The PBM maximum occurred in the early morning at around 6:00
261 and the minimum in the afternoon at 18:00. The diurnal trends of GEM, PBM, and GOM were
262 similar to those observed in Nanjing (Zhu et al, 2012), but different from those in Guiyang, Xiamen,
263 and Guangzhou (Wang et al., 2016; Chen et al., 2013). Since DSL and Nanjing both belong to the
264 Yangtze River Delta region, the similar meteorology and emission characteristics within the Yangtze
265 River Delta region may explain the similar diurnal patterns of Hg species between these two sites.
266 The elevated GEM concentrations at DSL during daytime were likely related to the stronger
267 emissions from both human activities and natural releases. GOM and GEM showed similar diurnal

268 variations and both peaked at 10:00, probably suggesting that GOM and GEM were affected by
269 common sources (e.g. coal-fired power plants and industrial boilers). The high PBM concentrations
270 at night were likely derived from the adsorption of Hg species onto the preexisting particles and the
271 subsequent accumulation in the shallow nocturnal boundary layer. Wind speed was relatively low
272 while relative humidity was high at night, which were conducive to the adsorption of GOM onto
273 the particles (Fig. 4).

274

275 **3.2. Relationship between Hg species and meteorological factors**

276 Fig. 5 shows the relationship between wind direction/speed and atmospheric mercury species.
277 One-third of the prevailing winds came from the east and 16% from the north at DSL during the
278 study period (Fig. 5a). Wind speed from all directions during the study period was mainly in the
279 range of 0-6 m/s, of which wind speed higher than 4 m/s mainly derived from the east. The highest
280 GEM concentrations, averaged at 3.92 ng/m³, were linked to the winds from the south and southwest,
281 in comparison of the average concentration of 2.71 ng/m³ from the other wind sectors (Fig. 5b). A
282 similar wind-concentration pattern was also seen for GOM, but not for PBM for which high
283 concentrations were from the north/northwest and south/southwest (Fig. 5c & 5d). Anthropogenic
284 emissions were generally higher in northern than southern China (Fig 1), which explained largely
285 the observed high concentrations from the north wind sector. However, even higher atmospheric Hg
286 concentrations were observed from the south and southeast wind sector, implying other factors
287 dominated the Hg concentrations in these wind sectors. (Wang et al., 2016) reported that Hg
288 concentrations in the surface soils were generally higher in southern than northern China. A
289 modeling study estimated that the mean annual Hg air-soil flux in the southwest region of our
290 sampling site (e.g. Zhejiang province) ranged from 8.75 to 15 ng m⁻² h⁻¹, while that in the
291 north/northwest region (e.g. Jiangsu province) ranged from 2.5 to 8.75 ng m⁻² h⁻¹ (Wang et al., 2016).
292 Hence, emissions from natural sources, such as soils, vegetations and water, should play important
293 roles in the observed high atmospheric Hg concentrations from the south to southeast wind sectors.

294 In order to confirm this hypothesis, the relationship between temperature and Hg
295 concentrations at DSL was investigated. Seasonal temperature in ascending order was divided into
296 different groups and the corresponding mean Hg concentrations are plotted in Fig. 6. In spring,
297 GEM concentration increased with increasing temperature. In the other seasons, similar trends were

298 observed when temperature increased to a certain value. Such a phenomenon was likely caused by
299 temperature-dependent surface emissions, noting that the increasing PBL height with increasing
300 temperature should have offset some of the increased mercury concentration. GOM concentration
301 showed a clearly positive correlation with temperature in summer, largely due to the in situ oxidation
302 of GEM under high temperature heavy ozone pollution (Qin et al., 2018). Such a good correlation
303 did not exist in the other seasons. As for PBM, it appeared to have weakly negative correlation with
304 PBL height, suggesting the atmospheric diffusion conditions were influential on the concentrations
305 of PBM.

306

307 **3.3. Source-apportionment analyses**

308 **3.3.1. Potential source regions for mercury**

309 PSCF analysis results suggested that the potential sources for GEM were mainly located in
310 Anhui, Jiangxi, and Zhejiang provinces, and also possibly in Shandong province (Fig. 7a). Seasonal
311 dominant sources affecting GEM at DSL site were those in Jiangsu and Zhejiang provinces in spring,
312 in Anhui and Jiangxi provinces in summer, In Jiangsu province in autumn, and in Anhui and
313 Zhejiang provinces in winter (Fig. S1). In addition, sources in Henan and Shandong provinces
314 seemed to also play a role on GEM in winter, suggesting the importance of long-range transport in
315 this season. There existed substantial high PSCF signals in the southern areas, stronger than those
316 in the northern areas, despite lower anthropogenic emissions in the south. For example, southern
317 provinces such as Zhejiang and Jiangxi were estimated to release 25 tons/yr atmospheric Hg from
318 anthropogenic activities (Fig. 1), far less than those from the northern provinces such as Jiangsu and
319 Shandong (77 tons/yr) (Wu et al., 2016). Based on the mean annual Hg air-soil flux of 8.75 to 15 ng
320 $\text{m}^{-2} \text{h}^{-1}$ in the southwest region of our sampling site (Wang et al., 2016), it was estimated that the
321 total Hg emissions from soils were in the range of 6.9 - 13.9 tons/yr from Zhejiang province, which
322 was comparable to the anthropogenic Hg emission of approximately 15tons/yr in 2014 in this
323 province (Wu et al., 2016). Thus, the soil emission of GEM from natural surfaces was an importance
324 source in southern areas, corroborating the discussion in Section 3.2. In addition, the East China Sea
325 (including the offshore areas and open ocean) showed sporadic high PSCF signals of GEM in all
326 the four seasons (Fig. S1), indicating possible influences from shipping activities (more discussion
327 below).

328 The PSCF pattern of PBM was quite different from that of GEM (Fig. 7b). The annual potential
329 source regions for PBM were mainly from the northern areas of Jiangsu and Anhui provinces, and
330 from the northeastern China including Shandong and Hebei provinces. These provinces were
331 regarded as the main Hg sources areas in China and accounted for approximately 25.2% of the
332 Chinese anthropogenic atmospheric Hg emissions (Wu et al., 2016). Seasonal sources for PBM were
333 similar in spring, autumn, and winter, but not in summer for which high PSCF values were mainly
334 located in the southern areas of Shanghai, likely due to the prevailing winds in summer from the
335 south, southeast, and southwest where Zhejiang and Jiangxi provinces were important mercury
336 source regions.

337 The annual potential sources for GOM were mainly located in Anhui and Zhejiang provinces
338 and the coastal areas along Jiangsu province (Fig. 7c). The PSCF pattern of GOM was similar to
339 that of GEM but not PBM. The potential source regions of GOM were more from southern than
340 northern China, likely due to the higher atmospheric oxidant levels in the southern regions. Seasonal
341 PSCF patterns for GOM showed dominant sources in Zhejiang and Jiangxi provinces in summer,
342 major sources in inland areas and moderate sources over the East China Sea and Yellow Sea in
343 spring, major sources in Zhejiang province and moderate sources over the Yellow Sea in autumn,
344 and major sources from the coastal areas of Jiangsu to a vast ocean of the Yellow Sea in winter. One
345 previous study suggested that the marine boundary layer could provide considerable amounts of
346 oxidants such as chlorine and bromine, which were beneficial for the production of GOM by
347 oxidizing GEM (van Donkelaar et al., 2010) and this mechanism may explain the substantial PSCF
348 signals over the ocean.

349 The PSCF analysis results discussed above demonstrated the relative contributions from the
350 short-range transport from the adjacent areas of Shanghai, and regional and long-range transport
351 from northern and southern China as well as from the ocean on contributing to speciated
352 atmospheric mercury at DSL (more discussion below).

353

354 **3.3.2. Quasi-local sources versus regional/long-range transport on atmospheric mercury**

355 According to the relationships between wind direction and Hg concentration as well as the
356 PSCF analysis results discussed above, the elevated GEM, GOM, and PBM concentrations at DSL
357 were generally associated with the wind sectors from the southwest and north directions. To reveal

358 the relative importance of local sources and regional transport, the ratio of GOM/PBM was applied
359 as an indicator based on the fact that the residence time of GOM is generally considered to be shorter
360 than that of PBM (Lee et al., 2016). Lower ratios of GOM/PBM indicate the dominance of
361 regional/long-range transport over local sources, and vice versa. In the following discussion, the
362 ratios of GOM/PBM during the whole study period were grouped into four categories, i.e. 0-1, 1-2,
363 2-3, and higher than 3 with their corresponding wind sector distribution shown in Fig. 8. Higher
364 GOM/PBM ratios were mostly associated with winds from the east and southeast, e.g., with the
365 frequency of these two wind sectors increased from 27% for the GOM/PBM ratios of less than 1 to
366 52% for the ratios higher than 3. Winds from these wind sectors were typically characterized of
367 relatively clean air masses, suggesting the dominance of local sources over the regional transport at
368 the observational site. In contrast, lower GOM/PBM ratios were mostly associated with winds
369 clockwise from the west to the north, with the frequency of these wind sectors decreased
370 significantly from 44% for the GOM/PBM ratios less than 1 to 21% for the ratios higher than 3,
371 indicating the importance of the long-range/regional transport from northern China associated with
372 the relatively low GOM/PBM ratios. According to the PSCF results above, the potential source areas
373 of Hg species (GEM, GOM and PBM) derived mostly from the south and southwest of the sampling
374 site. As shown in Fig. 8, the frequency of south, southwest, west, and northeast winds showed no
375 clear trend as the GOM/PBM ratios increased, suggesting complicated interactions that cannot be
376 explained solely by the local sources or regional transport. In general, the GOM/PBM ratio can be
377 used as a qualitative tracer for identifying the relative importance of long-range transport vs. local
378 sources if one factor dominates over the other.

379 The relationships among GEM, CO, secondary inorganic aerosols (SNA) and GOM/PBM ratio
380 were further investigated. Fig. 9 displays the concentrations of GEM as a function of GOM/PBM
381 ratio colored by CO. The size of the circles represents the corresponding concentration of SNA in
382 PM_{2.5}. CO has been commonly used as a tracer of fuel combustion, and SNA were derived from
383 secondary particle formation via the gas-to-particle conversion. CO and SNA were collectively used
384 as proxies of the extent of regional/long-range transport of anthropogenic air pollutants. As shown
385 in Fig. 9, GEM generally increased with increasing GOM/PBM ratio. In addition, lower GOM/PBM
386 ratios were associated with higher CO and SNA concentrations and vice versa. This corroborated
387 the discussion above that the GOM/PBM ratio was a reliable tracer for assessing the relative

388 importance of regional/long-range transport vs. local atmospheric processing.

389 GEM concentration fluctuated with a mean value of less than 2.6 ng/m³ when the GOM/PBM
390 ratio is less than 2.5, increased from 2.61 ng/m³ in the GOM/PBM ratio bin of 2.5-3.0 to 2.8 ng/m³
391 in the bin of 3.0-3.5, and then remained relatively stable when the GOM/PBM ratio is higher than
392 3.0. Generally, GEM showed an increasing trend as the GOM/PBM ratio increased while both SNA
393 and CO decreased. The elevated GEM concentrations tended to be associated with quasi-local
394 sources. In contrast, under the high SNA and CO conditions and when GOM/PBM ratios were lower,
395 GEM showed relatively low concentrations, suggesting that regional/long-range transport didn't
396 favor the elevation of GEM concentration. It has been recognized that the common regional/long-
397 range transport pathways contributing to the particulate pollution events in Shanghai were from the
398 north and northwest, which were originating mostly from the North China Plain. The relatively
399 lower GEM concentrations under the regional/long-range transport conditions corroborated the
400 PSCF analysis that only moderate probabilities of GEM source regions were from northern China
401 (Fig. 7a).

402

403 3.3.3. Source apportionment by the combined PMF and PCA analyses

404 PMF modeling has been widely used to apportion the sources of atmospheric pollutants. In this
405 study, GEM together with heavy metals and soluble ions, measured online synchronously, were
406 introduced into the EPA PMF5.0 model to apportion the major anthropogenic sources of GEM. A
407 six-factor solution was selected based on the results of multiple model runs, which can well explain
408 the measured concentrations of the introduced species. The profiles of the six identified PMF factors
409 and contributions of major anthropogenic sources to GEM are shown in Fig. 10. It has to be noted
410 that since no tracers for the natural emissions (e.g. soil, vegetation, and ocean) were available in this
411 study, the identification of natural mercury sources was not possible based on the PMF modeling.

412 Factor 1 had high loadings of Se, As, Pb, NO₃⁻, SO₄²⁻, and NH₄⁺. Se, As, and Pb were typical
413 tracers of coal combustion. SO₄²⁻ and NO₃⁻ were also formed from the gaseous pollutants emitted
414 from coal burning. Hence, this factor was defined as coal combustion sources and accounted for
415 12.3% of the anthropogenic GEM. Fig. S4 plotted wind roses of specific aerosol species. SO₄²⁻ and
416 As shared similar patterns with high concentrations mainly from the southwest. Since combustion
417 was an important source of GEM, this confirmed that the major sources of GEM were located in the

418 southwest region.

419 Factor 2 displayed particularly high loadings of Ni and V. The major sources of Ni in the
420 atmosphere can be derived from coal and oil combustions (Tian et al., 2012), and oil combustion
421 accounted for 85% of anthropogenic V emissions in the atmosphere (Duan and Tan, 2013). In
422 general, Ni and V have been considered as good tracers of heavy oil combustion, which has been
423 commonly used in marine vessels (Viana et al., 2009). Thus, this factor was identified as shipping
424 emissions. The sampling site is adjacent to the East China Sea and is located in Shanghai which has
425 the largest port in the world. Fig. S4 showed similar patterns of Ni and V with high concentrations
426 mainly from the northeast, east, and southeast. To further validate this factor, the time-series of GEM
427 concentrations from the shipping factor based on the PMF modeling were extracted and digested
428 into the PSCF modeling (Fig. S5), which showed that the potential sources regions were mainly
429 located over the East China Sea and coastal regions. This indicated factor 2 from PMF should be
430 representative of the shipping sector as well as oil combustion in motor vehicles and inland shipping
431 activities. Overall, this factor accounted for 19.6% of anthropogenic GEM and ranked as the second
432 largest emission sector, highlighting the urgent need of controlling the marine vessel emissions.

433 Factor 3 showed high loadings of Ca and moderate loadings of Ba and Fe. Ca and Fe are rich
434 elements in crust that can be used for cement production. As mercury could be released during
435 industrial processes of cement production, Factor 3 was assigned as cement production and
436 accounted for a minor fraction of 6.3 % of the anthropogenic GEM.

437 Factor 4 was characterized by high loadings of Cr and moderate loadings of Mn, Fe, Ni, and
438 Cu. These species together served as markers of metals smelting. Metals smelting were known to
439 be large sources of Hg emitted to the atmosphere (Pirrone et al., 2010), especially in the YRD, one
440 of the most developed and industrialized areas in China. This factor accounted for 7.6% of the
441 anthropogenic GEM.

442 Factor 5 had high loadings of Cl⁻. Waste incineration is an important source of enriched
443 chloride over land. Factor 5 was identified to be waste incineration, which contributed 6.4% of the
444 anthropogenic GEM.

445 Factor 6 was characterized by high loadings of Cd, Ag, K⁺, and Na⁺. The major sources of Cd
446 in China were iron and steel smelting industries (Duan and Tan, 2013). Ag was mainly used in
447 industrial applications, including electronic appliances and photographic materials. K⁺ was a typical

448 tracer of biomass burning, which often stemmed from agriculture waste burning over the Yangtze
449 River Delta and the North China Plain. Factor 6 was considered as a combined source of the
450 industrial and biomass burning emission sectors. It was estimated to contribute 47.8% of the
451 anthropogenic GEM.

452 As discussed above, surface emissions were likely to be important sources of the observed
453 atmospheric mercury. As PMF modeling didn't resolve the natural sources of mercury, the PCA
454 (Principal Component Analysis) was applied for further source apportionment by introducing the
455 temperature parameter. Four factors are resolved, which totally explained 75.32% of the variance as
456 shown in Table 2. Factor 1 accounted for 34.15% of the total variance with high loadings for SO₂,
457 SO₄²⁻, NH₄⁺, K⁺, Pb, Se, and As, which was explained as coal combustion mixed with biomass
458 burning. Factor 2 accounted for 14.85% of the total variance with high loadings for temperature, O₃,
459 and NH₃, which was explained as surface emissions. Factor 3 accounted for 13.43% of the total
460 variance and showed high loadings for Ni and V, which indicated the contribution of ship emissions.
461 Factor 4 accounted for 12.89% of the total variance and showed high loading for Fe and Ca,
462 indicating the contribution of cement production.

463

464

465 **3.4. Factors affecting the formation and transformation of mercury species**

466 **3.4.1. Factors affecting the formation of GOM**

467 A typical case from July 24 to July 27, 2015 was chosen to investigate the possible formation
468 process of GOM. As shown in Fig. 11, the shaded episodes represented nighttime from 18:00 to
469 6:00 the next day. Both GEM and GOM exhibited rising trends during nighttime (Fig. 11a), which
470 was ascribed to nighttime accumulation effect due to the very shallow boundary layer (Fig. 11c).
471 Starting from 6:00 in the morning, GEM concentration began to gradually decline as the boundary
472 layer developed. In contrast, the concentration of GOM continued to rise from 6:00 until it reached
473 the peak value at around 10:00. During this period, ozone and temperature also kept rising until
474 surpassed 200 µg/m³ and 34°C, respectively. Accordingly, as an anthropogenic emitting tracer, the
475 concentration of carbon monoxide was basically stable and even showing a downward trend, which
476 suggested that some other factors caused the increase of GOM in addition to the anthropogenic
477 emissions. This phenomenon clearly revealed the acceleration of the conversion process of GEM to

478 GOM under favorable atmospheric conditions of higher O₃ concentration and ambient temperature.
479 In the case of atmosphere dilution by the rise of PBL, the fact that GOM was not falling but rising
480 suggested the great influence of this process on ambient GOM concentration. Similar observation
481 has been found at the high-altitude Pic du Midi observatory in southern France (Fu et al., 2016),
482 which was almost impervious to anthropogenic emission sources. The important role of GEM
483 oxidation in our sampling site, which is located in one of the most developed industrial areas in
484 China, was most likely due to the presence of sufficient oxidants in this area. Severe ozone pollution
485 frequently occurred in the YRD due to strong anthropogenic emission intensities (Duan et al., 2017).
486 Previous studies suggested that the primary oxidants in the terrestrial environment were O₃ and OH
487 radicals (Zhang et al., 2015), while Br was an important oxidant in the subtropical marine boundary
488 layer (Obrist et al., 2011). It was possible that, in addition to O₃ and OH radicals, Br might also be
489 an important species to oxidize GEM as the DSL site is adjacent to the East China Sea.

490 Statistical relationships among GOM, O₃, and temperature are shown in Fig. 12. Temperature
491 was plotted against a range of GOM/PBM bins colored by O₃ and the size of the circles represents
492 the concentration of GOM. In general, as temperature and O₃ increased, the concentration of GOM
493 was subject to substantial enhancement. For instance, when temperature (O₃) was below 12°C (65.7
494 µg/m³), GOM was averaged at 37.8 pg/m³. While temperature (O₃) increased to above 20°C (91.5
495 µg/m³), GOM rose to 168.8 pg/m³, yielding a factor of 1- 5 increases. This further confirmed the
496 hypothesis from the case study above that the levels of oxidants under favorable environmental
497 conditions were crucial for the formation of GOM. It can also be seen from Fig. 12 that the lower
498 ratios of GOM/PBM were associated with lower temperatures and O₃ concentrations, indicating
499 relatively weak photochemistry during the cold seasons. On the contrary, the higher ratios of
500 GOM/PBM were associated with higher temperature and O₃ concentrations, indicating relatively
501 strong photochemistry during the warm seasons. Thus, the formation of GOM was more favored by
502 local atmospheric processing rather than the transport. This study demonstrated the abnormally high
503 GOM concentrations observed at DSL were largely ascribed to local oxidation reactions. However,
504 the explicit formation mechanism of GOM need to be investigated by measuring more detailed
505 components of GOM and atmospheric oxidants.

506

507 **3.4.2. Factors affecting the transformation of PBM**

508 Previous studies have shown that PBM can be emitted directly from various anthropogenic
509 sources such as coal-fired power plants and industries (Cheng et al., 2014a;Wu et al., 2016). In
510 addition, gas-particle partitioning was considered to be an important pathway for the formation of
511 PBM (Zhang et al., 2015;Amos et al., 2012). Since most of the areas in the YRD belong to non-
512 attainment areas in regard of particulate pollution and the concentrations of GOM were particularly
513 high at DSL as discussed above, the role of gas-particle partitioning in the formation of PBM should
514 be investigated. Previous studies reported that the concentrations of PM_{2.5} in eastern and northern
515 China are highest in the world (van Donkelaar et al., 2010), for which the elevated atmospheric
516 particulate matter probably facilitates the formation of PBM (Fu et al., 2015). Fig. 13 shows the
517 statistical pattern of the variation of PBM and GOM in the ascending bins of PM_{2.5}. The
518 concentration of PBM increased with increasing PM_{2.5}, likely due to the gas-particle portioning of
519 oxidized Hg (Amos et al., 2012;Cheng et al., 2014a). The trend of GOM was somehow different
520 from that of PBM. When PM_{2.5} concentrations were at relatively low levels under 75 µg/m³, GOM
521 concentrations increased with PM_{2.5}. However, when PM_{2.5} concentrations increased to 75-105
522 µg/m³, GOM exhibited a clear decreasing trend as PM_{2.5} increased. This suggests that low levels
523 PM_{2.5} may not have an apparent effect on GOM, but higher levels of PM_{2.5} may directly affect GOM
524 through adsorption of GOM onto the particles. When PM_{2.5} concentrations exceeded 105 µg/m³,
525 GOM exhibited a slightly increasing trend as PM_{2.5} increased. High PM_{2.5} concentrations in China
526 always related to severe anthropogenic emissions (van Donkelaar et al., 2010), so the moderate
527 increasing trend of GOM in these bins should be attributed to the impact of strong primary emissions.

528 A short episode from December 30, 2015 to January 1, 2016 was chosen to further investigate
529 this phenomenon. As shown in Fig. 14, in Stage 1, the concentrations of PM_{2.5} were below 100
530 µg/m³, PBM and GOM shared the similar temporal variation as PM_{2.5}. In Stage 2, as PM_{2.5} kept
531 climbing, GOM began to show somewhat negative correlation with PM_{2.5}, but not significant. The
532 reason might be that the relatively high temperature and low humidity during this period were not
533 conducive to the transfer of GOM to particle matters. In Stage 3, GOM decreased as PM_{2.5} continued
534 to increase, showing a clear anti-correlation. During this period, PBM showed a consistent trend
535 with PM_{2.5} and CO. Temperature was relatively low but with relatively high humidity. This
536 phenomenon clearly demonstrated the process of gas-particle partitioning of PBM formation. In
537 stage 4, GOM and PBM showed similar decreasing trend with PM_{2.5} and CO. The low GOM

538 concentrations, low humidity, and high temperature resulted in no significant signs of GOM
539 adsorption to PM_{2.5} in this stage. In general, under high PM_{2.5}, GOM concentration and humidity,
540 and low temperature conditions at DSL, clear processes of gas-particle partitioning were observed.

541

542 **4. Conclusions**

543 In this study, a year-long observation of speciated atmospheric Hg concentrations was
544 conducted at the Dianshan Lake (DSL) Observatory, located on the typical transport routes from
545 mainland China to the East China Sea. Different from many sites in China, GEM at DSL exhibited
546 high concentrations in winter, summer, and spring, which was due to the strong re-emission fluxes
547 from natural surfaces in summer and enhanced coal combustion for residential heating over northern
548 China in winter. The relatively high GOM concentrations in summer indicated that the formation of
549 GOM from GEM oxidation was likely crucial. PBM exhibited high concentrations in winter,
550 indicating the impact of long-range transport. The diurnal patterns of GEM and GOM were similar
551 with relatively high levels during daytime. For GEM, this was likely attributed to both human
552 activities and re-emission from natural surfaces during daytime. For GOM, in addition to direct
553 emissions, high concentrations during daytime were partially ascribed to photochemical oxidation
554 of GEM. The PBM concentrations were higher during nighttime, which was ascribed to the
555 accumulation effect within the shallow nocturnal boundary layer.

556 The relationship between meteorological factors and atmospheric Hg species showed that the
557 high Hg concentrations were generally related to the winds from the south, southwest, and north
558 and positively correlated with temperature. Both anthropogenic sources and natural sources
559 contributed to the atmospheric mercury pollution at DSL. Higher GOM/PBM ratios corresponded
560 to lower CO and SNA concentrations and vice versa. The ratio of GOM/PBM can be used as a tracer
561 for distinguishing local sources and regional/long-range transport based on the fact that the
562 residence time of GOM was shorter than that of PBM. GEM as a function of the GOM/PBM ratio
563 indicated that when the quasi-local sources dominated, GEM concentrations were relatively higher
564 than those events under the regional/long-range transport conditions. According to the PMF source
565 apportionment results, six sources of anthropogenic GEM and their contributions were identified,
566 i.e. industrial and biomass burning (47.8%), shipping emission (19.6%), coal combustion (12.3%),
567 iron and steel production (7.6%), incineration (6.4%), and cement production (6.3%). The

568 significant contribution of shipping emission suggested that in coastal areas mercury emitted from
569 marine vessels can be significant. In addition, a considerable natural source of GEM was identified
570 by digesting temperature into the principle component analysis.

571 The formation processes of GOM and PBM based on episodic studies were also investigated.
572 The high GOM concentrations were partially attributed to strong local photochemical reactions
573 under the conditions of high O₃ and temperature. Under high PM_{2.5} concentration and humidity and
574 low temperature conditions, the gas-particle partitioning processes were observed.

575

576 **Author contributions.** XQ, KH, and CD conceived the study and wrote the paper. XW, YL, and
577 DW performed the measurements and collected data. All have contributed to the data analysis and
578 review of the manuscript.

579

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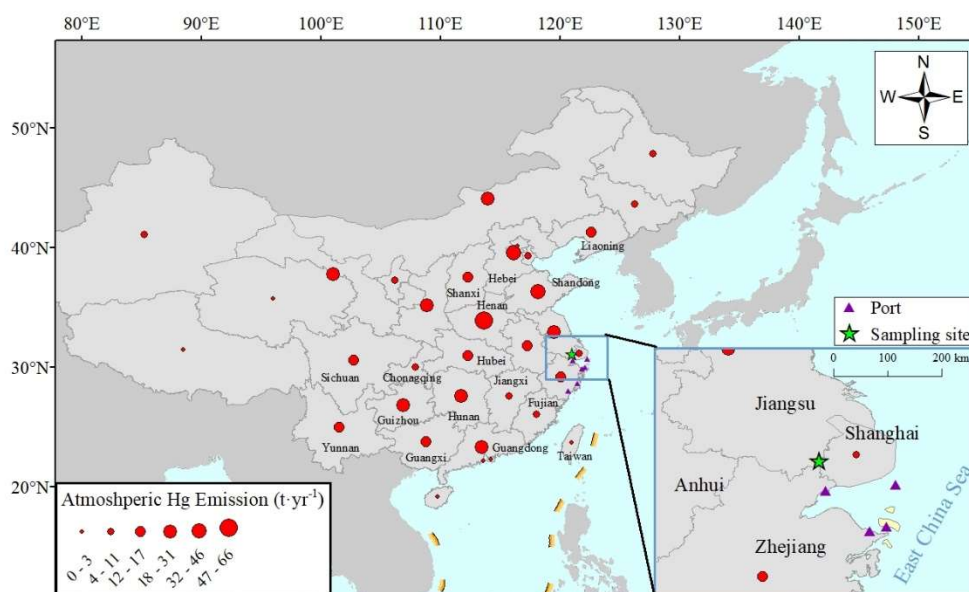
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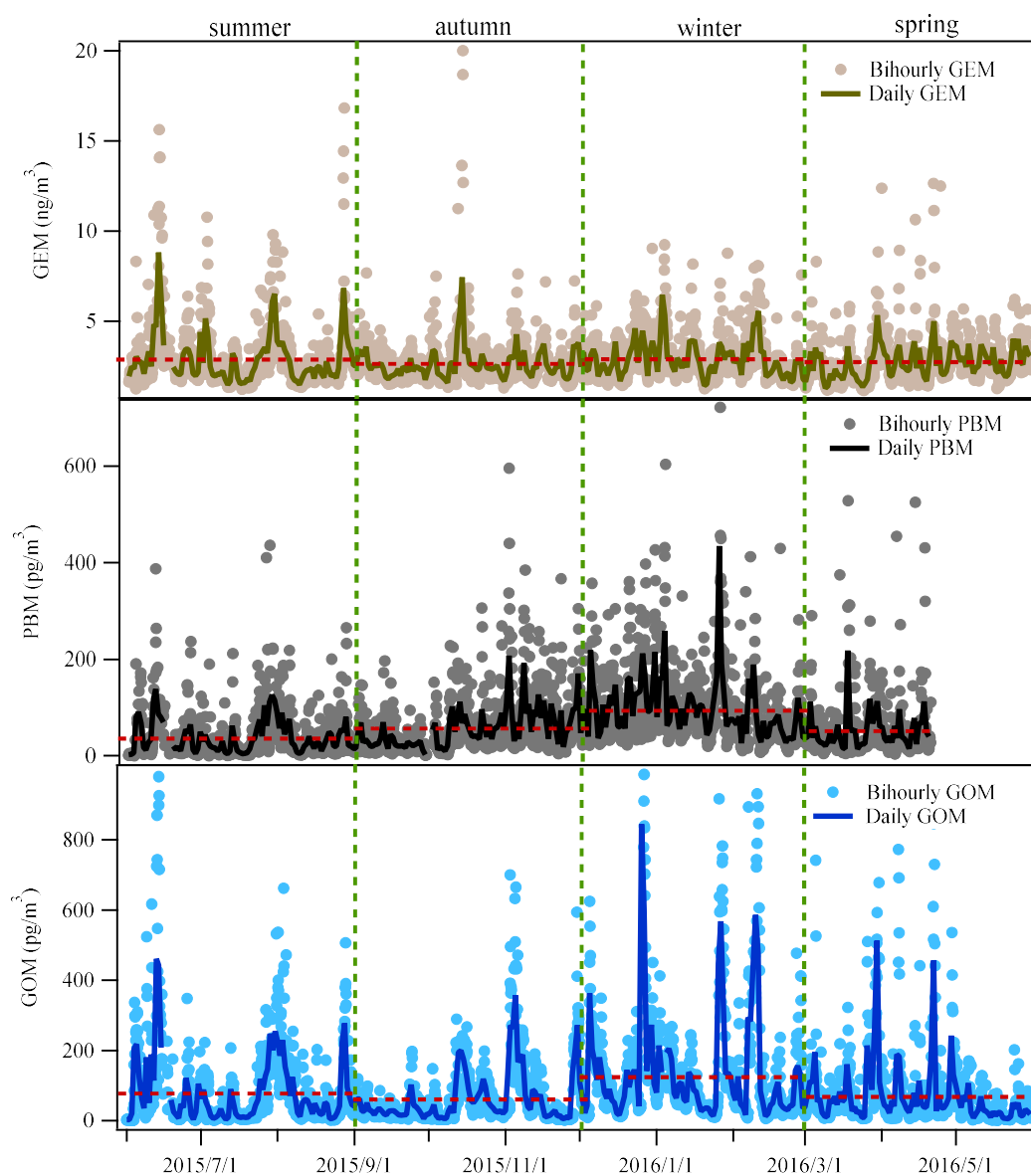


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783 Figure 1. The location of the Dianshan Lake (DSL) site in Shanghai, China. The red dots in the
784 map represent the anthropogenic atmospheric Hg emissions by each province in 2014 (Wu et al.,
785 2016).

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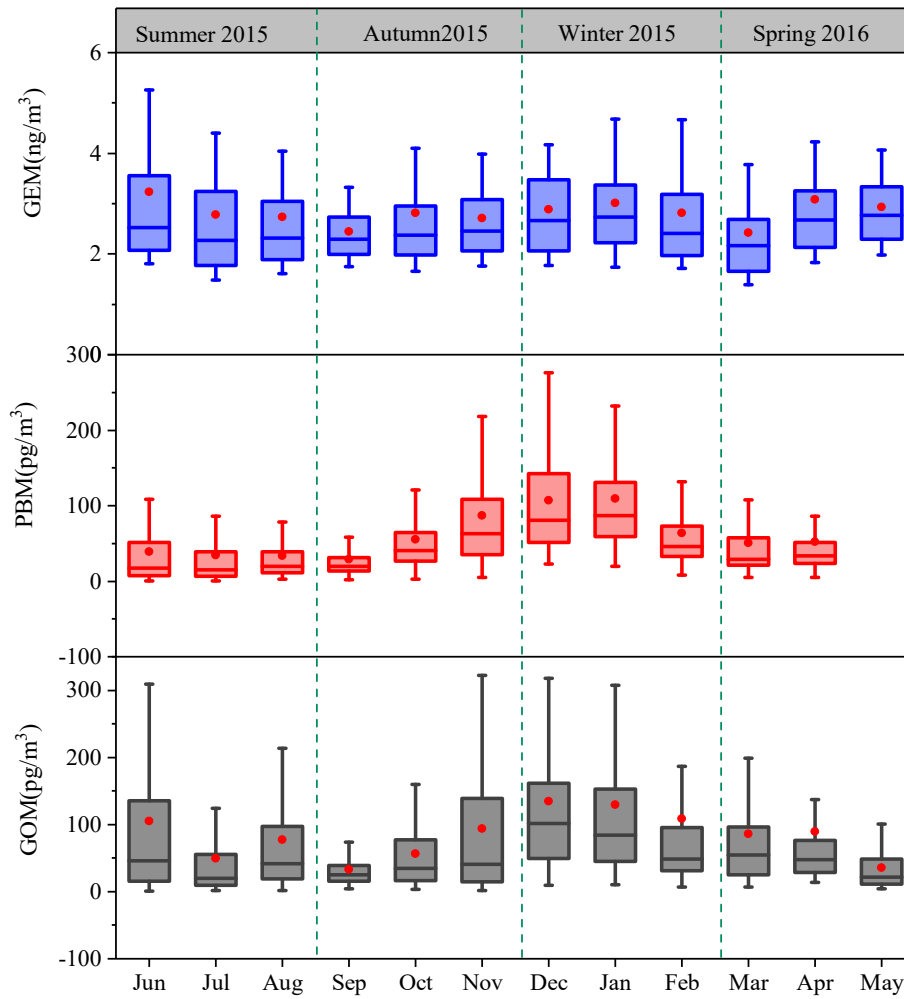
789 Figure 2. Time series of atmospheric Hg (GEM, PBM and GOM) concentrations during the whole
 790 study period at DSL. The red-dashed lines represent the mean concentrations of Hg species in
 791 each season.

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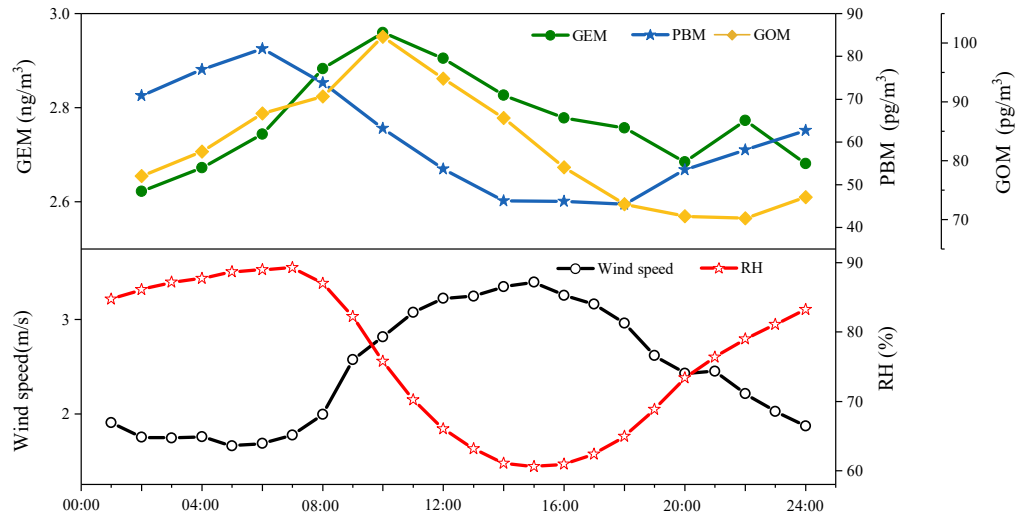
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797 Figure 3. Monthly variation of GEM, PBM, and GOM concentrations. The 10th, 25th, median, 75th
 798 and 90th percentile values are indicated in the box plots. The red dots represent the mean values.

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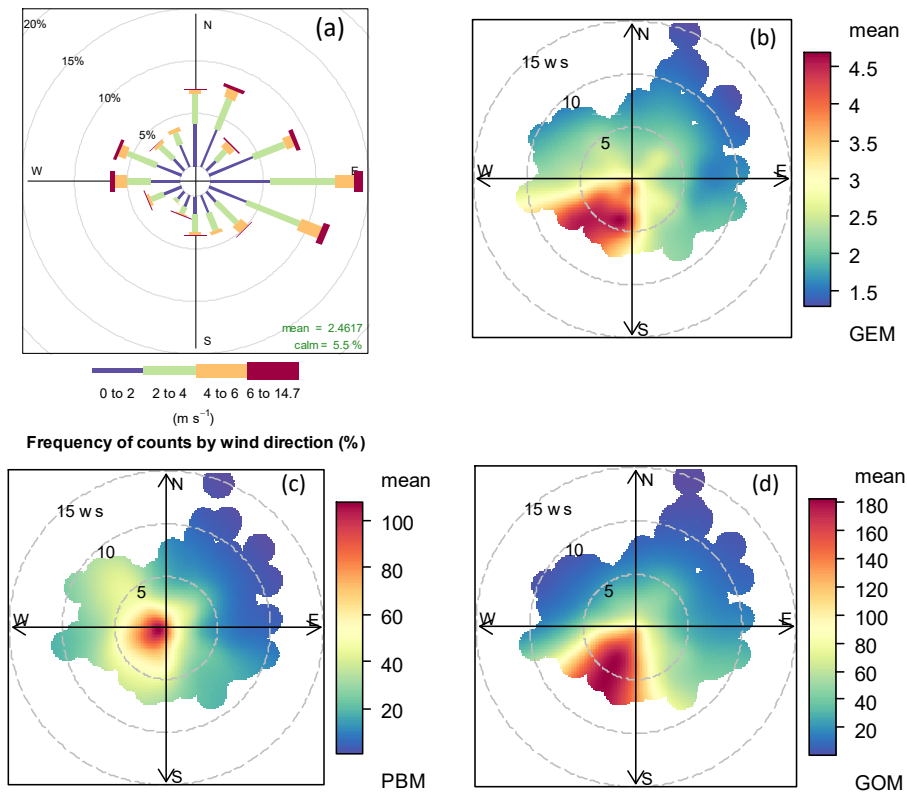
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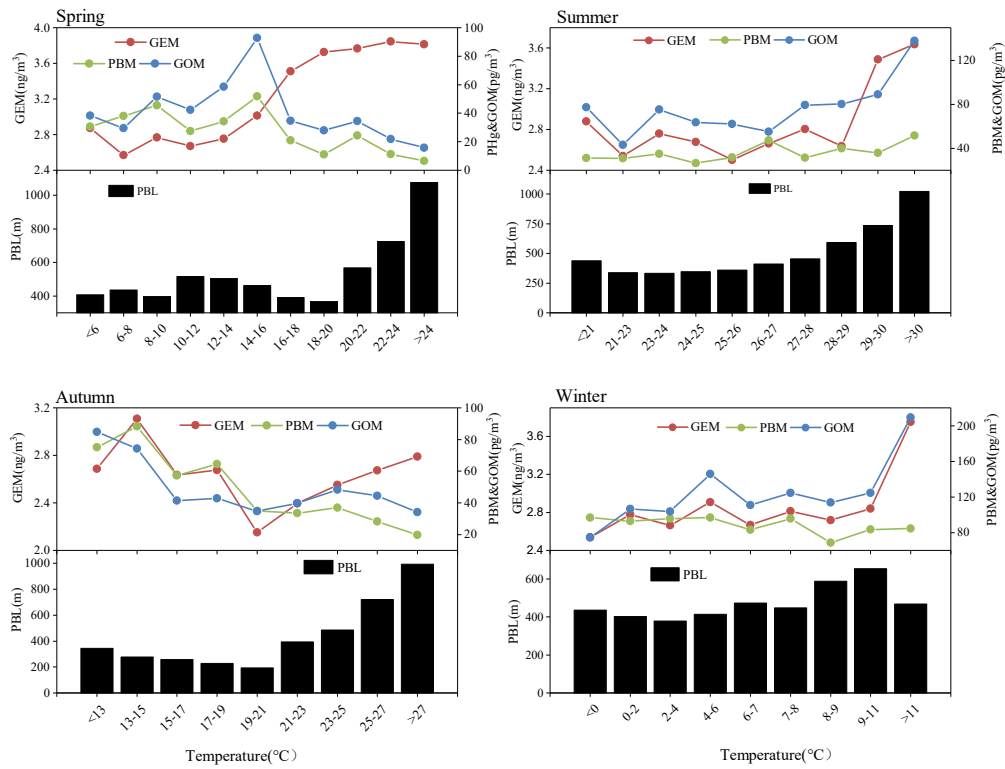
Figure 4. Annual mean diurnal variation of GEM, PBM, and GOM concentrations.

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806 Figure 5. (a) Wind rose plot during the study period. Mean concentrations of (b) GEM, (c) PBM,
807 and (d) GOM as a function of wind speed and wind directions. The radii of the circle in Figure 5 (a)
808 represent the frequency of wind directions and the radii of the circle in Figure 5 (b), (c), and (d)
809 represent the value of wind speed.

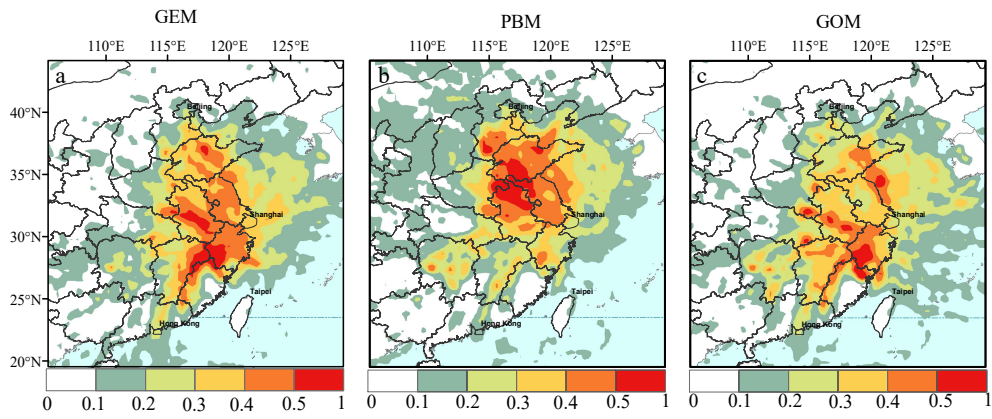


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811 Figure 6. The variation of atmospheric Hg (GEM, PBM, and GOM) and PBL as a function of

812 temperature in all four seasons.

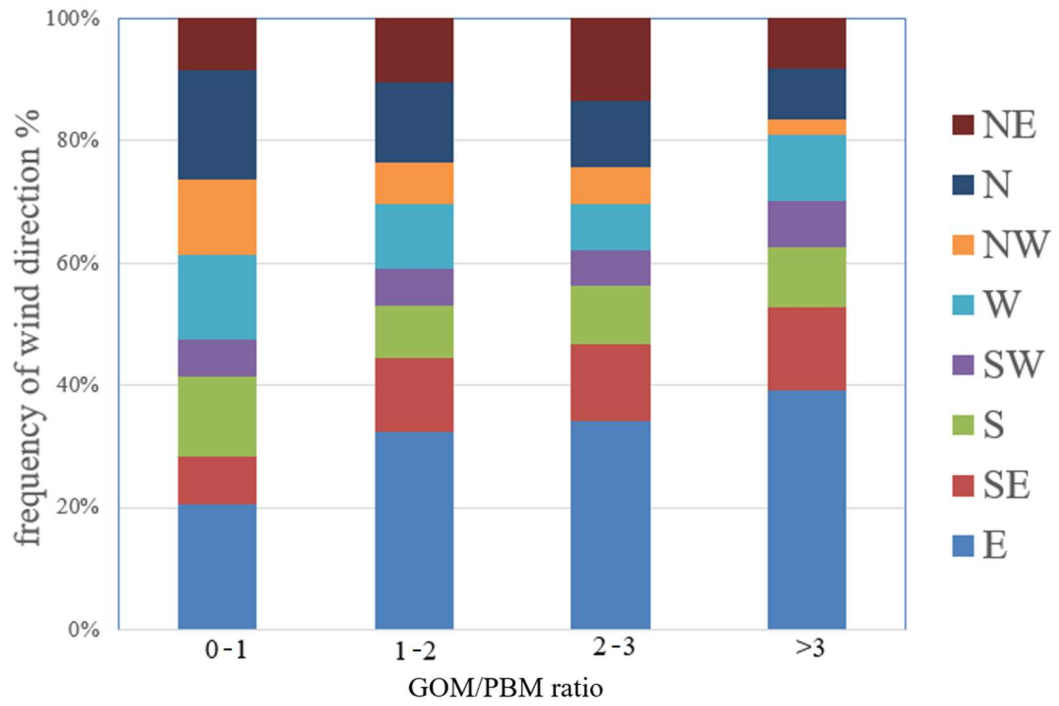
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815 Figure 7. Potential source regions of atmospheric Hg (GEM, PBM, and GOM) at the observational
 816 site according to PSCF analysis.

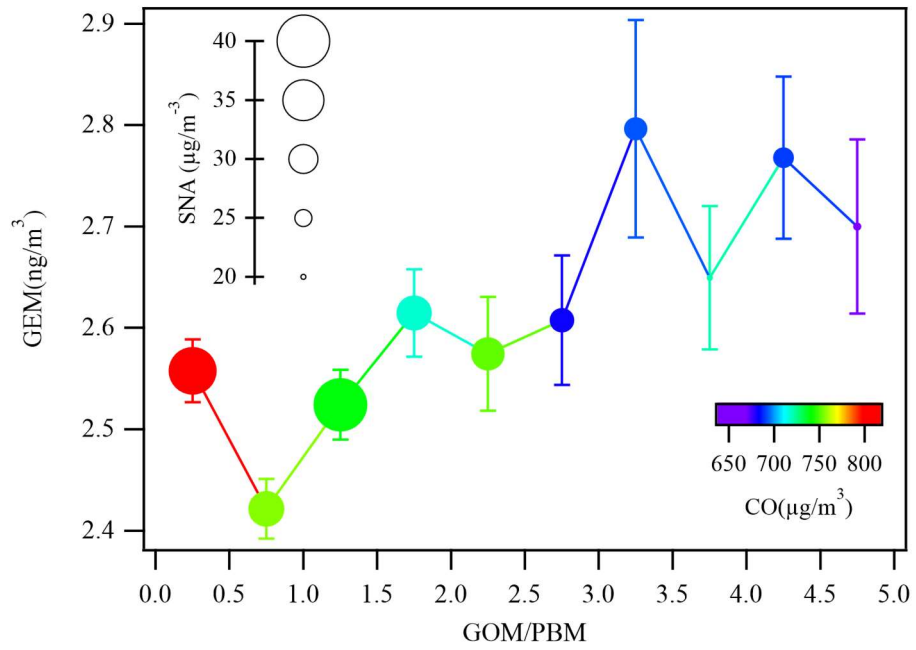
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819 Figure 8. Frequency of wind directions under different ranges of GOM/PBM ratios.

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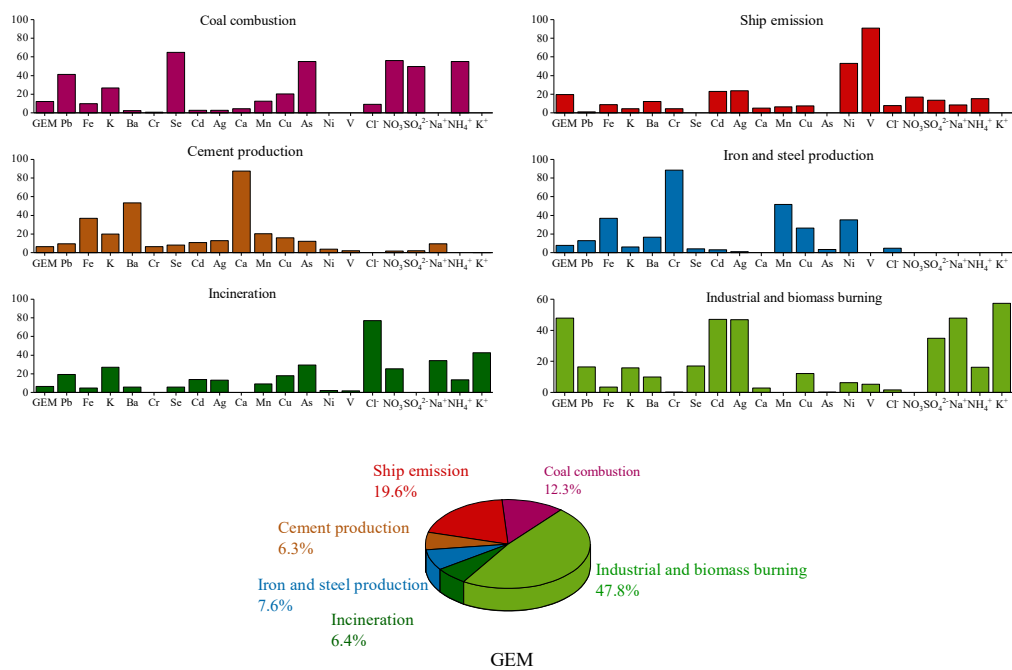
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822 Figure 9. The GEM concentrations as a function of the GOM/PBM ratios in each bin of 0.5. The

823 dots are colored by the concentrations of CO and the sizes of the dots represent the concentrations

824 of SNA in PM_{2.5}. The bars represent one standard error of GEM concentration in each bin.

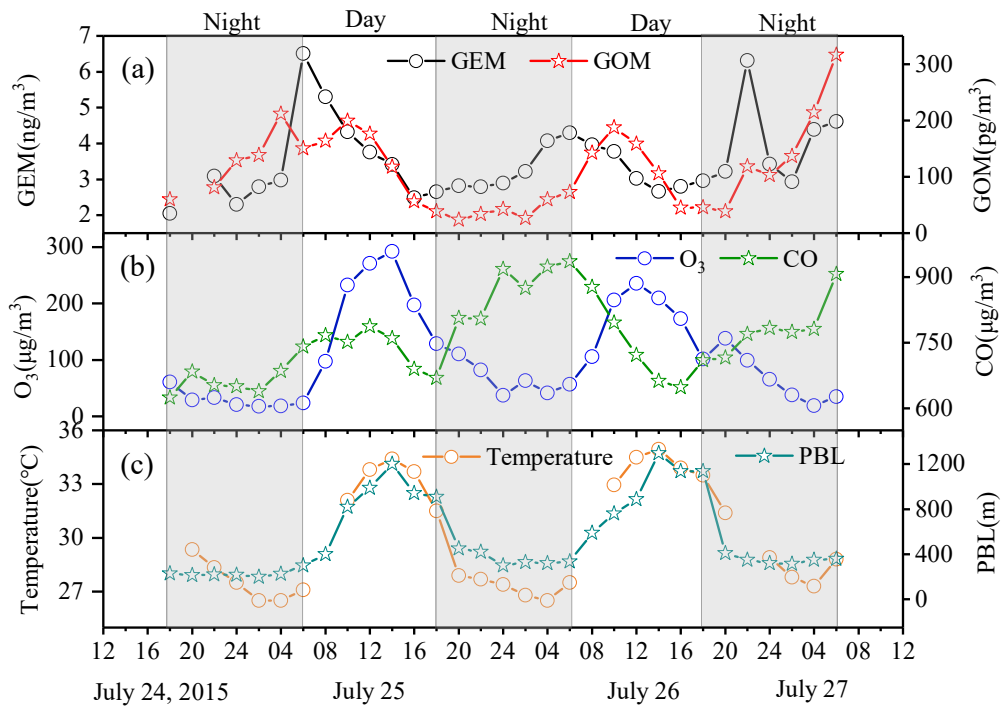
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827 Figure 10. A six factor source apportionment for anthropogenic GEM based on PMF analysis.

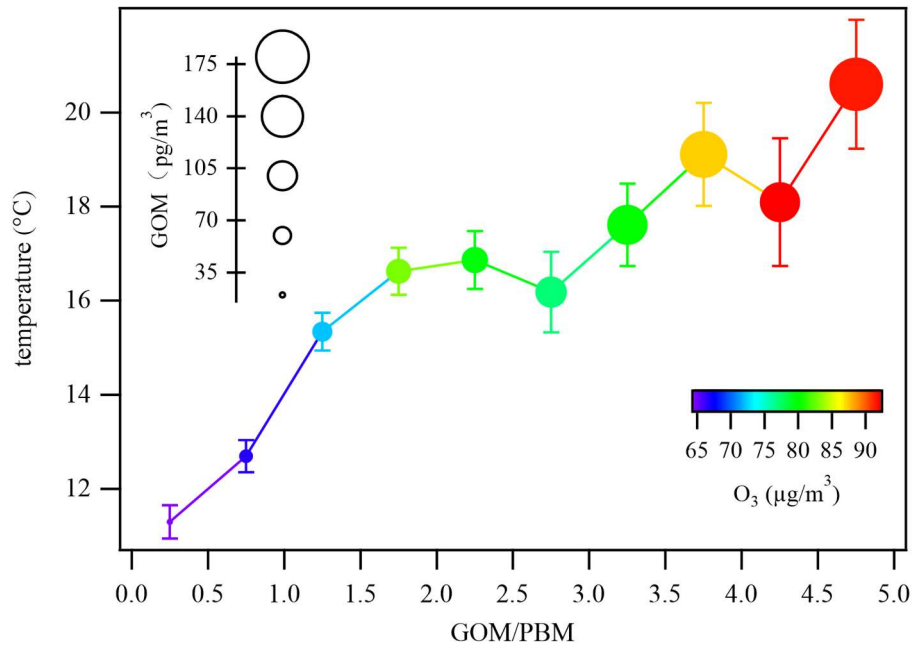
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830 Figure 11. A case study of GEM oxidation from July 24 to 27, 2015. The time-series of GEM, GOM,
 831 O₃, CO, temperature, and PBL are plotted. The shaded parts represent nighttime.

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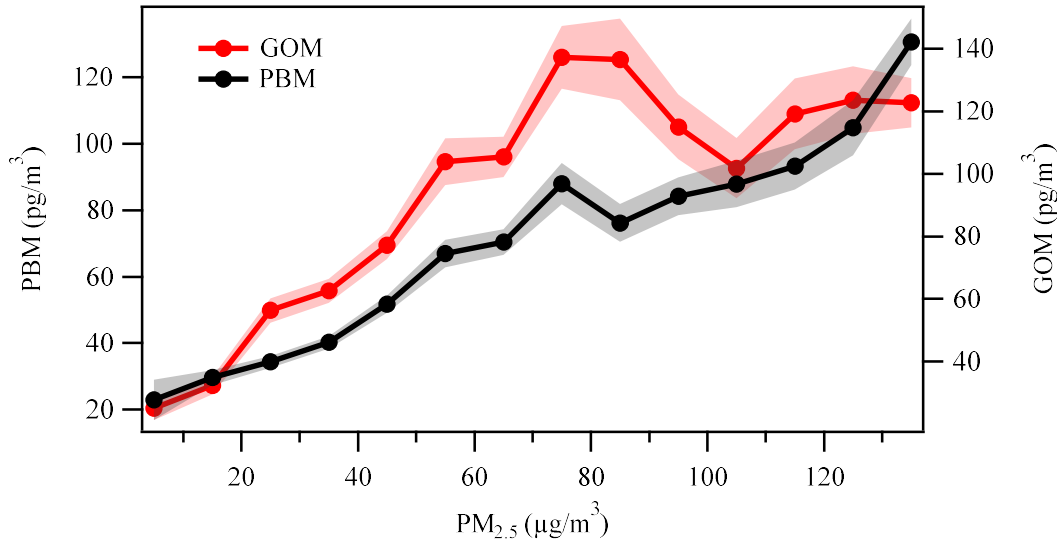


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834 Figure 12. Temperature variations in each bin of the GOM/PBM ratios. The dots are colored by the
 835 concentrations of O₃ and the sizes of the dots represent the concentrations of GOM. The bars
 836 represent one standard error of temperature in each bin.

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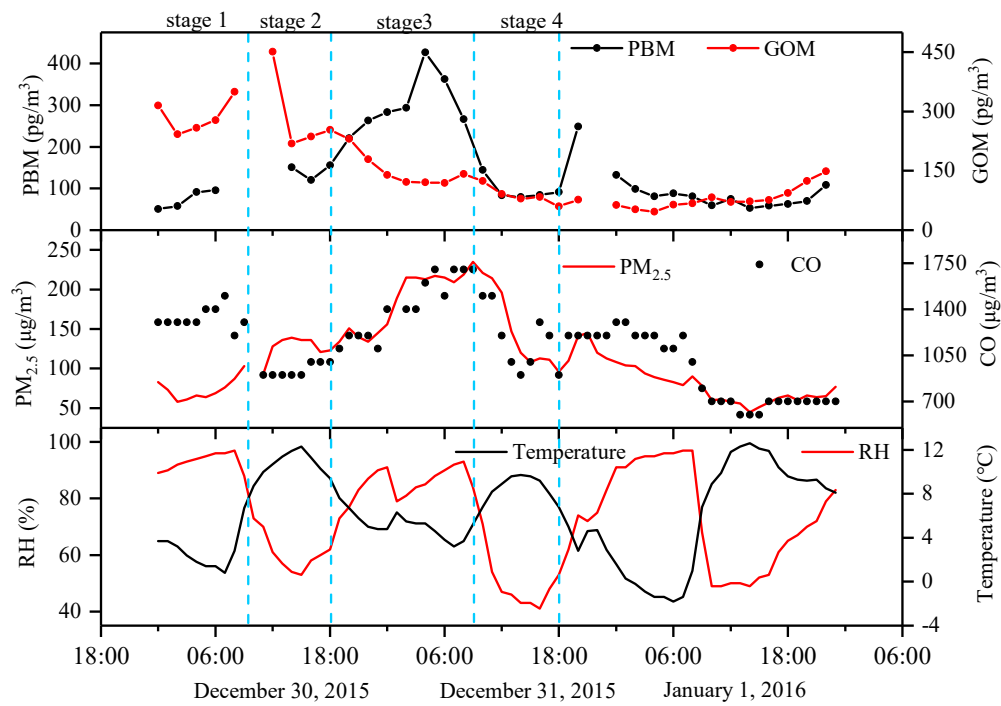
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840 Figure 13. The variations of PBM and GOM as a function of PM_{2.5} in each bin of 10 µg/m³. The
 841 shaded areas represent one standard error of GOM and PBM concentrations.

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844 Figure 14. A case study of gas-particle partitioning between GOM and PBM from December 30, 2015
 845 to January 1, 2016, which was divided into different stages. The time-series of PBM, GOM, $\text{PM}_{2.5}$,
 846 CO, temperature, and RH are plotted.

Table 1. The concentrations of speciated atmospheric mercury in this study and other sites around the world.

Location	Remarks	Year	GEM (ng/m ³)	PHg (pg/m ³)	RGM (pg/m ³)	Reference
Dianshan Lake Shanghai, China	Suburban	2015-2016	2.77±1.36	60.8±67.4	82.13±115.5	This study
Chongming Shanghai, China	Suburban	2009-2012	2.65±1.73	21.5±25.4	8.0±8.8	Zhang et al. (2017)
Xiamen, China	Suburban	2012-2013	3.5	174.41	61.05	Xu et al. (2015)
Guiyang, China	Urban	2009	9.72±10.2	368±676	35.7±43.9	Fu et al. (2011)
Miyun, China	Rural	2008-2009	3.23	98.2	10.1	Zhang et al. (2013)
Mt. Waliguan, China	Remote	2007-2008	1.98±0.98	19.4±18.1	7.4±4.8	Wan et al. (2009)
Seoul, Korea	Urban	2005-2006	3.22±2.10	23.9±19.6	27.2±19.3	Kim et al. (2009)
Nova Scotia, Canada	Rural	2010-2011	1.38±0.2	0.4±1.0	3.5±4.5	Cheng et al. (2014)
Elora, Ontario, Canada	Rural	2006-2007	1.17	16.40	15.1	Baya and Van Heyst (2010)
Chicago, USA	Urban	2007	2.5±1.5	9±20	17±87	Gratz et al. (2013)
Reno, USA	Suburban	2007-2009	2.0±0.7	7±7	18±22	Lyman and Gustin (2009)
Rochester, NY, USA	Urban	2008-2009	1.49	6.57	4.08	Huang et al. (2010)

Table 2. PCA (Principal Component Analysis) analysis for GEM at DSL.

	Factor 1	Factor 2	Factor 3	Factor 4
GEM	0.50	0.25	0.11	0.07
SO ₂	0.69	-0.20	-0.18	0.35
NO ₂	0.38	-0.49	0.35	0.39
SO ₄ ²⁻	0.84	0.13	0.15	0.00
NH ₄ ⁺	0.88	-0.12	0.18	0.07
K ⁺	0.77	-0.25	0.04	0.39
Pb	0.80	-0.17	0.04	0.32
Se	0.87	-0.05	0.01	0.29
As	0.82	-0.23	0.06	0.33
O ₃	0.06	0.79	-0.30	0.03
NH ₃	0.03	0.73	0.36	-0.04
Temperature	-0.23	0.82	0.17	-0.03
Ni	0.24	-0.02	0.85	0.22
V	-0.03	0.11	0.90	-0.05
Fe	0.50	-0.12	0.24	0.74
Ca	0.26	0.08	0.00	0.90
Explained variance %	34.15	14.85	13.43	12.89

