

1 **Inter-comparison of O₃ formation and radical chemistry in the past decade at a suburban site in**
2 **Hong Kong**

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10 **Abstract**

11 Hong Kong, as one of the densely populated metropolises in East Asia, has been suffering
12 from severe photochemical smog in the past decades, though the observed nitrogen oxides
13 (NO_x) and total volatile organic compounds (TVOCs) were significantly reduced. This study,
14 based on the observation data in the autumns of 2007, 2013 and 2016, investigated the
15 photochemical ozone (O₃) formation and radical chemistry during the three sampling periods
16 in Hong Kong with the aid of a Photochemical Box Model incorporating the Master
17 Chemical Mechanism (PBM-MCM). While the simulated locally produced O₃ remained
18 unchanged ($p=0.73$) from 2007 to 2013, the observed O₃ increased ($p<0.05$) at a rate of 1.78
19 ppbv/yr driven by the rise in regionally transported O₃ (1.77 ± 0.04 ppbv/yr). Both the
20 observed and locally produced O₃ decreased ($p<0.05$) from the VOC sampling days in 2013
21 to those in 2016 at a rate of -5.31 ± 0.07 and -5.52 ± 0.05 ppbv yr⁻¹, respectively. However, a
22 levelling-off ($p=0.32$) was simulated for the regionally transported O₃ during 2013 – 2016.
23 The mitigation of autumn O₃ pollution in this region was further confirmed by the continuous
24 monitoring data, which has never been reported in previous studies. Benefited from the air
25 pollution control measures taken in Hong Kong, the local O₃ production rate decreased

26 remarkably ($p < 0.05$) from 2007 to 2016, along with the lowering of recycling rate of
27 hydroxyl radical (OH). Specifically, VOCs emitted from the source of liquefied petroleum
28 gas (LPG) usage and gasoline evaporation decreased in this decade at a rate of -2.61 ± 0.03
29 ppbv yr^{-1} , leading to a reduction of the O_3 production rate from $0.51 \pm 0.11 \text{ ppbv h}^{-1}$ in 2007 to
30 $0.10 \pm 0.02 \text{ ppbv h}^{-1}$ in 2016. In addition, solvent usage made decreasing contributions to both
31 VOCs (rate = $-2.29 \pm 0.03 \text{ ppbv yr}^{-1}$) and local O_3 production rate (1.22 ± 0.17 and 0.14 ± 0.05
32 ppbv h^{-1} in 2007 and 2016, respectively) in the same period. All the rates reported here were
33 for the VOC sampling days in the three sampling campaigns. It is noteworthy that
34 meteorological changes also play important roles in the inter-annual variations of the
35 observed O_3 and the simulated O_3 production rates. Evaluations with more data in longer
36 periods are therefore recommended. The analyses on the decadal changes of the local and
37 regional photochemistry in Hong Kong in this study may be a reference for combating
38 China's national-wide O_3 pollution in near future.

39 **Keywords:** Ozone formation; Volatile organic compounds; Radical chemistry; Source
40 apportionment; Control measures

41 **1 Introduction**

42 Ground-level ozone (O_3) is one of the most representative air pollutants in photochemical
43 smog, produced through photochemical reactions between volatile organic compounds
44 (VOCs) and nitrogen oxides (NO_x) in presence of sunlight (NRC, 1992; Jacob et al., 1999;
45 Guo et al., 2017). It is well documented that O_3 is harmful to human health (Bell et al., 2004),
46 crops (Wang et al., 2005) and natural ecosystems (Ashmore, 2005). Through the last 30 years,
47 extensive efforts have been made by the local and federal governments to alleviate the
48 tropospheric O_3 pollution around the world (NRC, 1992; NARSTO, 2000; Wang et al., 2017a;
49 Wang et al., 2018a). Effectiveness has gradually shown in some countries/regions, such as

50 Switzerland, Germany, Ireland and eastern North America (Lefohn et al., 2010; Cui et al.,
51 2011; Derwent et al., 2013; Parrish et al., 2014; Lin et al., 2017). In contrast, the O₃ levels in
52 many places are still increasing or not decreasing at the expected rates, particularly in East
53 Asia (Ding et al., 2008; Xu et al., 2008; Parrish et al., 2014; Xue et al., 2014a; Wang et al.,
54 2017a).

55 Hong Kong, as one of the densely populated metropolises in East Asia, has been suffering
56 from severe photochemical smog in the past decades, though the locally-emitted NO_x and
57 total VOCs (TVOCs) were significantly reduced (Xue et al., 2014a; Ou et al., 2015; Lyu et al.,
58 2016a; Wang et al., 2017a). On one hand, this indicates the non-linear relationship between
59 O₃ and its precursors. On the other hand, in addition to local O₃ formation, the observed O₃ in
60 Hong Kong is also influenced by the regional transport due to the proximity of the highly
61 industrialized Pearl River Delta (PRD) region. Earlier studies revealed that the local O₃
62 production is typically limited by VOCs in urban and some suburban areas in Hong Kong
63 (Zhang et al., 2007; Ling et al., 2014; Wang et al., 2017b). Namely, cutting VOCs emissions
64 will reduce O₃ production, while the reduction of NO_x may cause an O₃ increment (Cheng et
65 al., 2010, 2013; Guo et al., 2011; Wang et al., 2017a). Previous studies also documented that
66 photochemical O₃ formation is dependent upon the ratios between TVOCs and NO_x (Sillman,
67 1999; Guo et al., 2013a; Ling et al., 2013), reactivity of VOC species (Zhang et al., 2007; Liu
68 et al., 2008; Cheng et al., 2010) and the composition of NO_x (*i.e.* relative abundances of NO₂
69 and NO) (Richter et al., 2005; Xu et al., 2008; Wang et al., 2018a). Moreover, located in the
70 subtropical region, Hong Kong has relatively high temperature and strong solar radiation,
71 which are favourable for local O₃ formation. For regional transport, studies (Wang et al.,
72 2001; Ding et al., 2004; Wang et al., 2017b) indicated that O₃ was generally built up in Hong
73 Kong under the northerly winds, whereas it was often driven down by the sea breeze from
74 South China Sea (SCS) and by the southwest monsoon in warm seasons. The contribution of

75 regional transport to O₃ in Hong Kong even reached 70% under the dominance of tropical
76 cyclone (Huang et al., 2005), a typical synoptic condition conducive to severe O₃ pollution in
77 the Northern Hemisphere (So and Wang, 2003; Huang et al., 2005; Lam et al., 2005). To
78 improve the air quality in Hong Kong, a series of control measures aiming at restriction of
79 VOC emissions have been implemented by Hong Kong government since 2007, which
80 effectively reduced the concentrations of some VOCs, such as propane and *i/n*-butanes
81 emitted from taxis and public light buses fuelled by liquefied petroleum gas (LPG) (Lyu et al.,
82 2016b), the aromatics mainly attributable to solvent usage, and the alkenes in association
83 with diesel exhaust (Lyu et al., 2017a). As a result, Xue et al. (2014a) and Wang et al.
84 (2017a) found that the locally produced O₃ decreased. However, the regional and super-
85 regional transport of O₃ and its precursors from PRD and eastern China to Hong Kong had
86 offset the decrease of the local O₃ production, resulting in an overall increase of the observed
87 O₃ in Hong Kong from 2005 to 2013. Overall, the previous studies have greatly enhanced our
88 understanding on O₃ pollution in Hong Kong, and details about the studies can be referred to
89 in Table S1.

90 Despite many previous studies (Xue et al., 2014a, 2016; Ou et al., 2015; Lyu et al., 2016a;
91 Wang et al., 2017a; Wang et al., 2018a), the inter-annual variations of the O₃ formation
92 regimes and radical chemistry have yet been fully understood in Hong Kong. Additionally,
93 the online measurement data used in previous long-term O₃ study might hamper the exact
94 understanding of the local O₃ formation mechanisms, due to the unavailability of many
95 reactive VOCs, such as formaldehyde. Besides, the trends of the local production and
96 regional transport of O₃ were only updated to 2013 in previous studies (Xue et al., 2014a;
97 Wang et al., 2017a). In fact, many measures were taken to reduce air pollutants' emissions in
98 the latest years in Hong Kong and PRD. For examples, nearly 75% of the old catalytic
99 converters on LPG-fuelled vehicles were renewed during September 2013 - May 2014. A

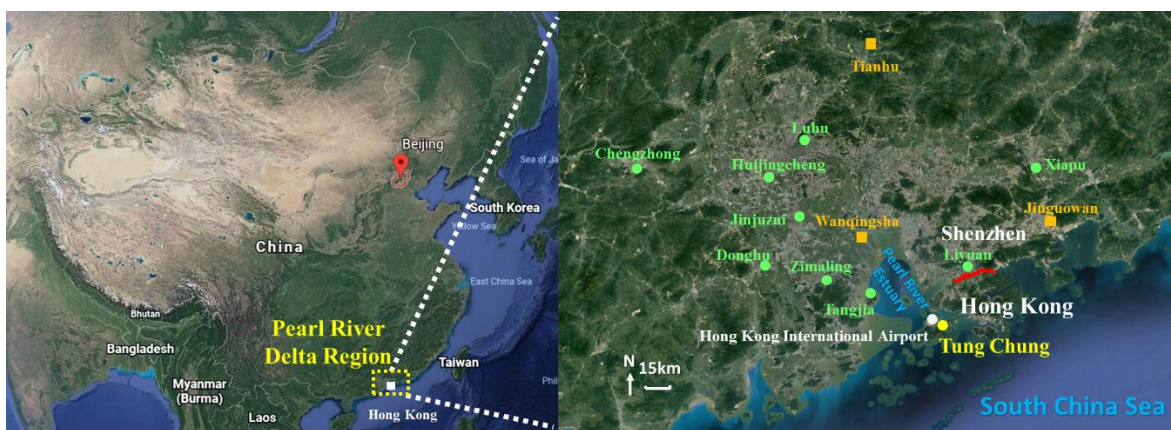
100 program to eliminate the pre-Euro IV diesel vehicles or to upgrade their emission standards to
101 Euro IV was initiated in March 2014 and is still ongoing till 2019 at its third phase. In PRD,
102 the second stage of the clean air controlling program was implemented in 2013 - 2015
103 (DGEPC, 2013). In 2014, the Guangdong provincial government has launched an Action
104 Plan for Air Pollution Prevention and Control (MEE PRC, 2014), putting the emphases on the
105 emission control of traffics, coal-fired power plants and industrial sources. Investigations on
106 the post-2013 variations of the local O₃ production in Hong Kong and the regional impacts
107 provide a good opportunity for us to examine the effectiveness of these local and regional
108 measures.

109 The objectives of this study were to re-examine the O₃ trend in the pre-2013 and trace the O₃
110 evolution in the post-2013 in Hong Kong, and to explore the underlying mechanisms for the
111 variations of O₃ formation and radical chemistry. With the aid of a photochemical box model,
112 the locally produced and regionally transported O₃, as well as their variation trends, were
113 determined (see section 2.5). Under the assumption that the local O₃ production in these years
114 was changed due to a series of control measures in Hong Kong, we also aimed to evaluate the
115 actual effectiveness of these control measures. China is suffering from severe O₃ pollution,
116 almost second to none over the world. While O₃ began to decrease in most areas of North
117 America and Europe, China's O₃ pollution was even aggravated in recent years. A series of
118 air pollution control strategies have been implemented in China, though most of them were
119 not specifically designed for O₃ abatement. Investigations on O₃ trends and the potential
120 causes in Hong Kong would provide a good example of assessing the evolution of O₃
121 pollution and the effects of artificial interventions in China. In addition, the changes in the
122 regional contribution to O₃ in Hong Kong determined in this study would throw light upon
123 the variations of O₃ in China, particularly in South China. It is expected that this study would
124 have some inspiration to the O₃ pollution control in other cities and regions in China.

125 **2 Methodolgy**

126 **2.1 Sampling site**

127 Hong Kong is located on the southern coast of China with Guangdong province to the north
128 and Pearl River Estuary (PRE) to the northwest. The sampling site (22.29N, 113.94E), Tung
129 Chung (TC), was in a newly-developed suburban area in western Hong Kong, with a
130 population of ~77,400 in 2016 (CSD, 2011, 2018). The urban centre of Hong Kong is ~20 km
131 northeast of TC. Hong Kong is dominated by the subtropical oceanic monsoon climate.
132 During warm seasons, the prevailing winds mainly come from SCS at a relatively low speed
133 (southwest winds). In cold seasons, the east and northeast winds are predominant. Generally,
134 the sampling site receives relatively polluted air masses from mainland China, *i.e.* PRD
135 region, Yangtze River Delta region and even North China between October and March, when
136 high O₃ levels are often observed (Wang et al., 2009). Therefore, the samplings were mainly
137 conducted in October and November in this study, except for 4 out of 45 sampling days in
138 September. The sampling site was close to a highway linking to the Hong Kong International
139 Airport (HKIA), and the HKIA was around 3 km to the north of the site. In addition, the local
140 emissions from residential activities may modulate the air quality at this site. It was expected
141 that O₃ at TC would be significantly influenced by NO emitted from the aforementioned
142 sources. As shown in Figure S1, the O₃ titrated by NO ($\Delta\text{O}_3\text{-1}$) was equivalent to $29.9\pm 8.8\%$,
143 $26.7\pm 12.7\%$ and $32.5\pm 16.6\%$ of the photochemically formed O₃ ($\Delta\text{O}_3\text{-2}$) in the 2007, 2013
144 and 2016 sampling campaigns, respectively, confirming the importance of NO titration in
145 modulating O₃ at the site. Figure 1 shows the locations of the sampling site (TC) and the 12
146 air quality monitoring stations in PRD, which witnessed the evolution of air quality in PRD
147 over the last decade and is used to demonstrate the variations of regional O₃ in this study.
148 More detailed description of the site can be found in our previous studies (Jiang et al., 2010;
149 Cheng et al., 2010; Ling et al., 2013; Ou et al., 2015).



150

151 Figure 1. Location of the sampling site (yellow circle) and the surrounding environment. The
 152 red line in the right panel shows the border between Hong Kong and Shenzhen, Guangdong.
 153 The three regional and nine urban air quality monitoring stations in PRD are symbolized by
 154 orange blocks and green circles, respectively.

155 **2.2 Continuous measurements of trace gases and collection of VOC/OVOC samples**

156 Trace gases (SO_2 , CO, NO, NO_2 and O_3) and meteorological conditions were continuously
 157 measured at TC site for three autumn periods in 2007, 2013 and 2016 (see Table S2 for the
 158 specific sampling periods), including 25 O_3 episode days with the maximum hourly average
 159 O_3 exceeding 100 ppbv (Level II of China National Ambient Air Quality Standard) and 185
 160 non-episode days. VOC and OVOC samples were selectively collected on 8, 19 and 18 days
 161 in 2007, 2013 and 2016, respectively (see Table S2 for the specific sampling dates). The three
 162 sampling periods were used as representatives of the autumns in the three years in this study,
 163 and the rationality will be discussed in section 3.1.

164 Trace gases were continuously measured at the TC air quality monitoring station operated by
 165 the Hong Kong Environmental Protection Department (HKEPD), ~0.8 km to our sampling
 166 site. The instruments were the same as those used in the US air quality monitoring program
 167 (HKEPD, 2017a). Table S3 summarizes the instruments, analysis techniques, detection limits
 168 and the time resolutions for measurements of the trace gases. The high resolution data were

169 collected and averaged into the hourly averages. All the analysers except O₃ analyser were
170 zeroed daily by analysing scrubbed ambient air and calibrated every two weeks by a span gas
171 mixture with a NIST (National Institute of Standards and Technology) traceable standard,
172 while the O₃ analyser was calibrated using a transfer standard (Thermo Environmental
173 Instruments (TEI) 49PS) every two weeks. Details about the quality assurance and control
174 procedures can be found in Ling et al. (2016a). The meteorological parameters, including
175 temperature, relative humidity, pressure, wind speed, wind direction, precipitation and solar
176 radiation, were also continuously monitored by a mini weather station (Vantage Pro TM &
177 Vantage Pro 2 Plus TM Weather Stations, Davis Instruments) during the sampling periods.
178 Data were integrated into 30-minute averages by a built-in program in the weather station.
179 The collection and analysis of VOCs and OVOCs were detailed in our previous studies (Guo
180 et al., 2009; Wang et al., 2018b). Briefly, pre-cleaned and evacuated 2 L electropolished
181 stainless-steel canisters were used to collect VOC samples. On O₃ episode days, one-hour
182 sample was collected in each hour during the daytime (07:00-19:00 LT), generating 13
183 samples per day, while 5-7 one-hour samples were collected every other hour on non-O₃
184 episode days from 07:00 to 19:00 LT in the 2013 and 2016 sampling campaigns. However,
185 12 one-hour samples were collected on each VOC sampling day between 07:00 and 18:00 in
186 2007, regardless of O₃ episodes or non-episodes. The O₃ episode days were predicted prior to
187 sampling based on weather forecast and numerical simulation of O₃. Overall, the O₃ episodes
188 were usually associated with high temperature, strong solar radiation, low humidity, and
189 weak or northerly winds. A total of 414 canister samples, including 96 samples in 2007, 146
190 samples in 2013 and 172 samples in 2016, were collected and analysed during the three
191 sampling periods (Table S2).

192 In addition to VOC samples, OVOC samples were also collected on the same days as those
193 for the collection of VOCs. Dinitrophenylhydrazine (DNPH)-silica cartridges (Waters Sep-

194 Pak DNPH-Silica, Milford, MA) were used to collect the OVOC samples. An ozone scrubber
195 (Sep-Pak; Waters Corporation, Milford, MA) was connected in front of the DNPH cartridge
196 to prevent interference of ozone. The ozone scrubber was replaced every two OVOC samples.
197 For each OVOC sample, air was drawn to pass the O₃ scrubber and the cartridge for 2 hours
198 (2.5 hours in 2007 sampling campaign) at a flow rate of 0.5 L min⁻¹, which was controlled by
199 a rotameter. During the sampling periods in 2013 and 2016, 5-7 OVOC samples were
200 collected every two hours from 06:00-20:00 LT on both O₃ episode and non-episode days. In
201 2007, only 2 samples were collected on non-O₃ episode days at 10:30-13:00 and 13:00-15:30,
202 and 4 samples between 08:00 and 18:00 on O₃ episode days. In total, 275 OVOC samples (28
203 in 2007, 124 in 2013 and 124 in 2016) were collected and analysed in the three sampling
204 campaigns (Table S2).

205 **2.3 Chemical analysis**

206 2.3.1 Analysis of VOCs

207 The concentrations of 48 speciated non-methane hydrocarbons (NMHCs) in the canisters
208 were determined with an Entech Model 7100 Preconcentrator (Entech Instruments Inc.,
209 California, USA) coupling with a gas chromatography-mass selective detector (Model 5973N,
210 Agilent Technologies, USA), a flame ionization detector, and an electron capture detector
211 (GC-MSD/FID/ECD). The NMHCs were analysed in Guangzhou Institute of Geochemistry
212 (GIG), Chinese Academy of Sciences for the samples collected in both 2007 and 2013, and in
213 The Hong Kong Polytechnic University (HKPolyU) for the samples collected in 2016. It
214 should be noted that the GC-MSD/FID/ECD system in the latter two institutes was the same
215 as that at UCI, and inter-comparisons were performed regularly among the three institutes,
216 which showed reasonably good agreements (Ling et al., 2014; Wang et al., 2018b; Zeng et al.,
217 2018). Detailed information about the analysis procedures and quality assurance and control

218 can be found in Colman et al. (2001) and Simpson et al. (2010). Table S4 summarizes the
219 limits of detection (LoDs), precisions and accuracies of the VOC analyses in the three
220 institutes.

221 The OVOC samples were stored in a refrigerator at 4°C after sampling. For analyses of
222 OVOCs, the cartridges were eluted slowly with 2 ml of acetonitrile into a 2-ml volumetric
223 flask. A high-performance liquid chromatography (HPLC) system (Perkin Elmer Series 2000,
224 MA, USA) coupled with an ultraviolet (UV) detector operating at 360 nm was used for
225 analysis. The instrument was calibrated using standards of 5 gradient concentrations covering
226 the concentrations of interest for different OVOCs in ambient air. Good linear relationships
227 ($R^2 > 0.999$) between the standard concentrations and responses of the instrument were
228 obtained for the 16 analysed OVOC species. The built-in computerized programs of quality
229 control systems such as auto-linearization and auto-calibration were used to guarantee the
230 data quality. Detailed information about the analysis and quality control of OVOC samples
231 was provided in Cheng et al. (2014), Cui et al. (2016) and Ling et al. (2016b). Due to the low
232 detection rate of many OVOCs, this study only focused on formaldehyde, acetaldehyde,
233 acetone and propionaldehyde, which had relatively high concentrations.

234 **2.4 Model description**

235 **2.4.1 Positive matrix factorization (PMF)**

236 PMF is a receptor model that has been extensively used for source apportionment of airborne
237 particulate matters and VOCs (Lee et al., 1999; Brown et al., 2007). In this study, US EPA
238 PMF 5.0 model (US EPA, 2017) was applied to identify the sources of O₃ precursors,
239 according to Equation (1) (Paatero, 1997; Ling et al., 2014).

$$240 \quad x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad \text{Equation (1)}$$

241 where x_{ij} is the measured concentration of j th species in i th sample, g_{ik} represents the
242 contribution of k th source to i th sample, f_{kj} denotes the fraction of j th species in k th source,
243 and e_{ij} is the residual for j th species in i th sample. p stands for the total number of
244 independent sources (Paatero, 2000a, b).

245 The uncertainties of the concentrations applied to PMF were set in the same way as Polissar
246 et al. (1998) and Reff et al. (2007). Values below or equal to the LoD were replaced by half
247 of the LoDs and the uncertainties for these values were set as 5/6 of the corresponding LoDs.
248 For the values greater than LoDs, the uncertainties were calculated as $[(\text{Error Fraction} \times$
249 $\text{concentration})^2 + (\text{LoD})^2]^{1/2}$ where 10% was assigned as the error fraction. Missing values
250 (mainly due to maintenance or malfunction of the instruments) were replaced by the
251 geometric mean of the measured values and their accompanying uncertainties were set as
252 four times the geometric mean value. More details about the settings of the uncertainty were
253 provided in Norris et al. (2008) and Zhang et al. (2012).

254 The model was run for 20 times with a random seed, and tests with different number of
255 factors were conducted. The optimum solution was finally determined based on both a good
256 fit to the observed data and the most reasonable and interpretable results according to the
257 knowledge on the sources of O₃ precursors in Hong Kong (Ling et al., 2011, 2014; Ou et al.,
258 2015).

259 **2.4.2 Observation-based model (OBM)**

260 A photochemical box model coupled with the Master Chemical Mechanism (PBM-MCM)
261 was used to simulate the photochemical O₃ formation on the VOC sampling days. In this
262 study, MCM v3.2, a near explicit chemical mechanism consisting of 5,900 species and
263 16,500 reactions which fully describes the homogeneous gas phase reactions in the
264 atmosphere (Jenkin et al., 1997, 2003; Saunders et al., 2003), was used. The observation data

265 of temperature, relative humidity, O₃, SO₂, CO, NO, NO₂ and 52 C₂-C₁₀ VOCs/OVOCs were
266 input into the model. Specifically, the 52 VOCs/OVOCs included 19 alkanes, 16 alkenes, 13
267 aromatics and 4 OVOCs, as shown in Table S5, where the statistics of the mixing ratios of
268 VOCs/OVOCs are also presented. Though previous studies (Guo et al., 2013b; Ling et al.,
269 2016b) indicated that secondary formation dominated the sources of OVOCs in Hong Kong,
270 the primary emissions could not be neglected. Therefore, formaldehyde, acetaldehyde,
271 acetone and propionaldehyde with relatively high abundances were constrained to the
272 observed concentrations in the model, while the other OVOCs with low concentrations and
273 low detection rates were simulated by the model. Nitrous acid (HONO) was not monitored in
274 this study. The average diurnal cycle of HONO mixing ratios measured at the same site in
275 autumn in 2011 (Xu et al., 2015) was input into the model to roughly represent its role in O₃
276 formation and atmospheric radical chemistry. Due to the data limitation, the trends of HONO
277 at TC in the three sampling campaigns were not traceable. However, the measurements at a
278 background site in Hong Kong indicated comparable levels of HONO ($p > 0.1$) between the
279 autumn in 2012 and in 2018 (unpublished data). Therefore, adopting the HONO measured in
280 2011 as the inputs of the simulations in the three sampling campaigns was likely a plausible
281 assumption, despite some uncertainties. The model was also tailored to the real situations in
282 Hong Kong. Specifically, the height of the planetary boundary layer was allowed to vary
283 from 300 m at night to 1400 m at noon. The photolysis rates were calculated according to the
284 measured solar radiations by the Tropospheric Ultraviolet and Visible Radiation model
285 (Madronich and Flocke, 1999; Wang et al., 2017a), with the detailed method described in
286 Lyu et al. (2017b). In addition to the chemical processes, the exchange between the lower
287 troposphere and free troposphere, and dry deposition were also considered in the model. The
288 concentrations of air pollutants in the free troposphere were set according to the observations
289 at a mountainous site in Hong Kong (Lam et al., 2013). The dry deposition rates were

290 adopted from the previous studies (Saunders et al., 2003; Lam et al., 2013). The other
291 physical processes were not included in the model, which might lead to insufficient
292 description of the transport. However, since the model was constrained to the observations
293 which included the transported air pollutants, the regional transport was partially considered.
294 Besides, the observations at 07:00 on each day were used to initiate each day's modelling,
295 through which the effect of regional transport before the daytime modelling was also
296 considered. We admit that the PBM-MCM cannot perfectly reproduce the real atmospheric
297 processes. However, it performed well in describing the in-situ photochemistry in previous
298 studies (Lam et al., 2013; Ling et al., 2014; Lyu et al., 2017b; Wang et al., 2017a). Actually,
299 the deficiency of PBM-MCM in consideration of the atmospheric dynamics enabled us to
300 assess the contributions of regional transport to O₃ in Hong Kong, based on the differences
301 between the observed and simulated O₃ (Wang et al., 2017a).

302 **2.5 Simulation scenarios**

303 The PBM-MCM simulates the in-situ O₃ photochemistry based on the observed O₃ precursors.
304 Figure S2 shows the average mixing ratios of some O₃ precursors in different wind sectors.
305 The higher levels of CO, ethyne, ethane, propane and toluene under northwest winds
306 indicated the transport of these species from PRD to Hong Kong. Meanwhile, O₃ might also
307 be transported to Hong Kong. Text S1 discusses the determination of the locally produced
308 and regionally transported O₃, as well as the uncertainties. Furthermore, to evaluate the
309 contributions of VOC sources to the local O₃ production, two scenarios of model simulation
310 were performed, *i.e.*, Scenario A and Scenario B. The scenario A simulated the O₃
311 photochemistry in the whole air, which was constrained by the observed concentrations of all
312 the O₃ precursors. The model simulations in scenario B (including six assumed sub-scenarios)
313 were constrained by the concentrations of O₃ precursors with those contributed by individual
314 sources being subtracted from the observed concentrations. Text S2 elaborates the set-up of

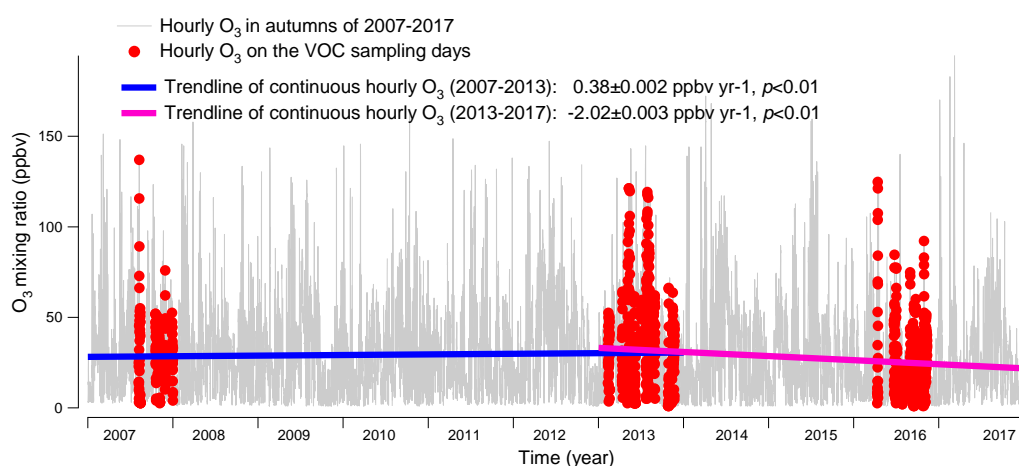
315 these scenarios. The simulated O₃ in scenario A was regarded as the locally produced O₃, as
316 the observed O₃ concentrations were not input to constrain the model. Bearing in mind that
317 the regional effects cannot be completely eliminated in this approach, due to the impacts of
318 regional air on the observed concentrations of O₃ precursors. The differences between the
319 scenario A and scenarios B reflected the contributions of the individual sources to the
320 simulated O₃ production rate. It should be noted that due to the nonlinear relationships
321 between O₃ and its precursors, the subtraction approach only qualitatively rather than
322 quantitatively evaluated the contributions of VOC sources to O₃ production.

323 **3 Results and discussion**

324 **3.1 Observation overview**

325 Figure 2 shows the hourly mixing ratios of O₃ observed at TC in the autumns of 2007-2017
326 with the data on VOC sampling days being highlighted in red. It was found that the autumn
327 O₃ increased significantly from 2007 to 2013 ($p < 0.01$), with a rate of 0.34 ± 0.002 ppbv yr⁻¹.
328 This was consistent with Wang et al. (2017a) who reported an overall increase rate of autumn
329 O₃ of 0.67 ± 0.07 ppbv yr⁻¹ at the same site for the period of 2005-2013. On one hand, the
330 discrepancy in O₃ increasing rates might be due to the different statistics used to draw the
331 rates, *i.e.* hourly values in this study and monthly averages in Wang et al. (2017a). On the
332 other hand, the autumn O₃ increased substantially from 23.9 ± 0.97 ppbv in 2005 to 30.2 ± 0.97
333 ppbv in 2007, much quicker than the increase between 2007 and 2013. Without the inclusion
334 of the period of 2005-2007 might be another reason of the less O₃ enhancement calculated
335 here. In contrast to the increased autumn O₃ during 2007-2013, the autumn O₃ decreased
336 obviously from 2013 to 2017 ($p < 0.01$), at a rate of -2.27 ± 0.003 ppbv yr⁻¹, indicating a
337 fundamental alleviation of O₃ pollution in Hong Kong in the latest 5 years. Overall, a
338 statistically significant decreasing trend (rate = -0.44 ± 0.001 ppbv yr⁻¹) was observed for the
339 autumn O₃ at TC through 2007 to 2017 ($p < 0.05$). The average O₃ on VOC sampling days in

340 the three sampling campaigns also followed the same pattern, which increased from
 341 32.8 ± 2.6 ppbv in 2007 to 36.9 ± 2.3 ppbv in 2013, while decreased to 24.4 ± 1.9 ppbv in 2016.
 342 Further, we investigated the number of O₃ episode days in the autumns of the three VOC
 343 sampling years (see Figure S3) and identified 15 (16.5% of the autumn days, same below)
 344 and 16 (17.6%) O₃ episode days in 2007 and 2013, respectively. However, there was only 5
 345 (5.5%) O₃ episode days in the autumn of 2016. Similarly, the O₃ episode days accounted for
 346 12.5%, 26.3% and 5.6% of the 2007, 2013 and 2016 sampling campaigns, respectively.
 347 Therefore, the increase of O₃ from 2007 to 2013 and the decrease in the following years
 348 could be represented by O₃ observed in the three sampling periods.



349

350 Figure 2. Long-term trends of the observed O₃ at TC from 2007 to 2017. Hourly O₃ values on
 351 the VOC sampling days in the autumns of 2007, 2013 and 2016 are marked in red. The
 352 hourly variation rates of O₃ are converted to yearly rates in periods of 2007 – 2013 and 2013
 353 – 2017.

354 Table 1 and Table S6 present the observed O₃, CO, NO, NO₂, SO₂ and TVOCs, as well as the
 355 meteorological conditions averaged on the VOC sampling days in 2007, 2013 and 2016,
 356 respectively. From 2007 to 2013, the TVOCs decreased by nearly a half, which was expected
 357 to result in the reduction of O₃ in view of the VOC-limited regime of O₃ formation at TC
 358 (Cheng et al., 2010; Wang et al., 2017a). However, the increases of CO and the notable

359 decrease of NO in 2013 could enhance the O₃ production. The higher O₃ in 2013 indicated
360 that this effect overrode the reduction of TVOCs in influencing the O₃ production. In
361 particular, the decrease of NO meant the reduced NO titration to O₃, which has been
362 recognized as a primary reason of O₃ increase in VOC-limited regime (Chou et al., 2006;
363 Wang et al., 2018b). From 2013 to 2016, the decrease of O₃ was accompanied by the
364 reductions of TVOCs and NO₂, though CO remained increasing at the same time. NO₂, as a
365 direct source of O₃ through photolysis, plays important role in modulating the O₃ variation.
366 Though the causes of NO₂ reduction are unknown to us, it might be one of the critical factors
367 contributing to the decline of O₃ in Hong Kong in recent years. On the contrary, the increase
368 of CO was also confirmed by the continuous monitoring data at TC, with a rate of
369 33.9 ± 0.7 ppbv yr⁻¹ between 2013 and 2016. In fact, the consistent increasing trend ($p < 0.05$)
370 was also observed at the roadside sites in Hong Kong (not shown here). While the causes of
371 CO increase in Hong Kong may be complicated, the increased vehicle emission is a plausible
372 explanation. Studies (Johnson, 2008; Yao et al., 2008) revealed that while the new engine
373 technologies performed well in reducing NO_x emission, they might lead to the increased
374 emission of CO, with the application of lower air-to-fuel ratio and engine temperature.

375 In addition, studies have confirmed that continental anticyclones and tropical cyclones are
376 conducive to severe O₃ pollution in Hong Kong, because these synoptic systems are often
377 accompanied with northerly winds, high temperature, strong solar radiation, and relatively
378 high pressure in Hong Kong (Ding et al., 2004; Huang et al., 2005; Jiang et al., 2015).
379 Table S7 summarizes number of O₃ episode days with tropical cyclone, continental
380 anticyclone and low pressure trough in the autumns of 2007, 2013 and 2016. In autumn 2007,
381 8, 8 and 1 O₃ episode day(s) were found to be related to the tropical cyclone, continental
382 anticyclone and low-pressure trough, respectively, with 2 O₃ episode days under the
383 combined influence of tropical cyclone and continental anticyclone. There were also 11 and 5

384 O₃ episode days in association with tropical cyclone and continental anticyclone in autumn
 385 2013, respectively (Wang et al., 2018b). However, 4 out of the 5 episode days found in
 386 autumn 2016 were associated with tropical cyclone, with the other one related to low-
 387 pressure trough. Therefore, the lower O₃ and less O₃ episode days in 2016 were also benefited
 388 from the meteorological conditions.

389 Table 1. Mixing ratios of the measured trace gases and TVOCs averaged on the selective 45
 390 VOC sampling days in 2007, 2013 and 2016.

Unit: ppbv	2007		2013		2016	
	Mean ± 95% C.I.	Max.	Mean ± 95% C.I.	Max.	Mean ± 95% C.I.	Max.
O ₃	32.8±2.6	137.0	36.9±2.2	121.2	24.4±1.9	124.9
CO	456.3±19.8	847.0	585.0±11.9	1047.9	691.8±9.5	1074.7
NO	17.2±3.2	124.7	10.9±1.3	98.6	11.3±1.4	94.6
NO ₂	27.7±2.1	69.6	31.5±1.4	80.8	22.0±1.1	103.2
SO ₂	6.9±0.4	21.8	7.0±0.2	18.0	3.0±0.1	10.7
TVOCs	49.7±4.4	111.1	25.1±1.4	68.0	21.1±1.4	71.9

391

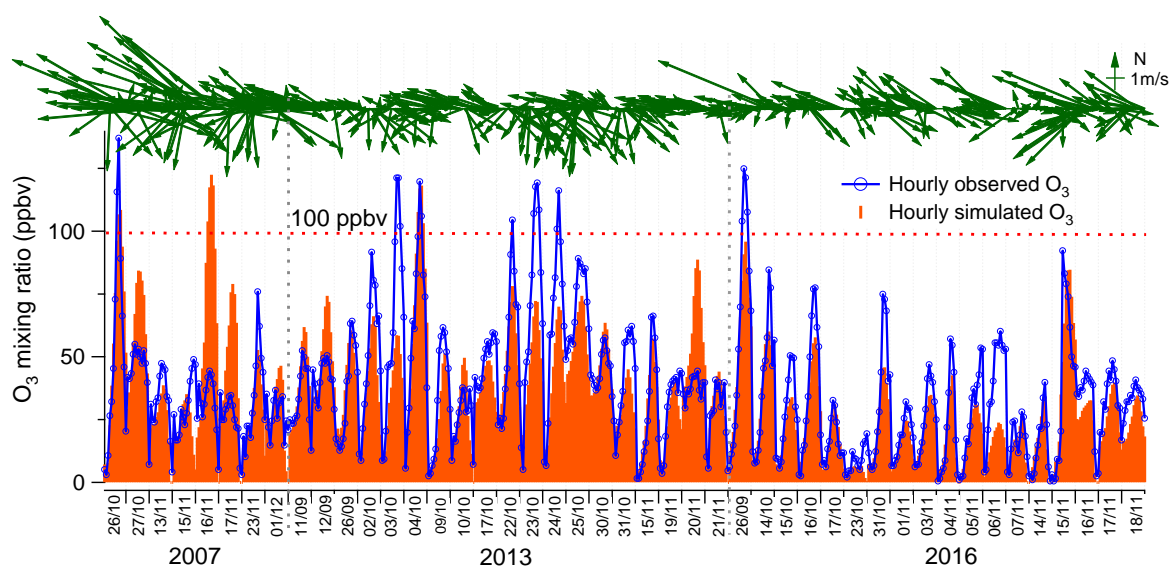
392 3.2 Model simulation of O₃

393 3.2.1 Model validation

394 Figure 3 compares the simulated O₃ in scenario A and the observed O₃ on the VOC sampling
 395 days. Overall, both the magnitudes and the temporal patterns of the observed O₃ were
 396 reasonably reproduced, though the mean of the simulated O₃ (33.8±1.9 ppbv) was slightly
 397 lower than the observed average (37.1±2.0 ppbv). To quantitatively evaluate the model
 398 performance, the index of agreement (IOA) was used to examine the goodness of fit between
 399 simulated and observed O₃. Within the range of 0-1, higher IOA represents better agreement

400 between the simulated and observed values (Willmott, 1982). In this study, the overall IOA
401 for the three sampling periods was 0.74, within the range of IOA (0.67-0.89) accepted by the
402 previous studies (Wang et al., 2015; Lyu et al., 2015, 2016a, c; Wang et al., 2017a, 2018a).
403 Good correlations ($R^2=0.61$) were also shown between the simulated and observed hourly O_3 .
404 Bearing in mind the deficiencies of the box model in describing the atmospheric dynamics,
405 we believed that the modelling results were acceptable, but special attention and explanation
406 to the discrepancies between the simulated and observed O_3 was needed.

407 It was found that the discrepancies were most likely caused by the transport processes, *i.e.*,
408 vertical and horizontal transport, which were not fully represented in the PBM-MCM model
409 (George et al., 2013; Lakey et al., 2015; Wang et al., 2017a). For example, the simulated O_3
410 (maximum: 122.6 ppbv) was much higher than the observed O_3 (maximum: 44.3 ppbv) on
411 November 16, 2007, when the strong southeast winds (wind direction: 90° - 180°) with the
412 highest wind speed of 5.3 m s^{-1} prevailed in Hong Kong. The south sector winds from SCS
413 might dilute the locally produced O_3 and the O_3 precursors/intermediates (such as the radicals)
414 which were not constrained by the observations. The same circumstances were also observed
415 on October 27, November 17, 2007 and September 11-12, November 20, 2013, with
416 southeast winds dominated (74.4%) during the daytime (Figure 3). For those days with the
417 simulated O_3 lower than the observed O_3 , *i.e.* October 3, 22-25, 2013 and November 6, 2016,
418 69.3% of the winds during the daytime came from the north (wind directions: 0° - 90° and
419 270° - 360°), which might transport the air masses laden with O_3 and/or O_3
420 precursors/intermediates not constrained to the observations from inland PRD to the sampling
421 site. The observed O_3 mixing ratios are plotted against the wind fields in Figure S4. It is
422 obvious that O_3 were higher under the north winds, while lower in the south wind sectors,
423 confirming the effects of dilution and regional transport of the south and north winds on O_3
424 pollution in Hong Kong, respectively.



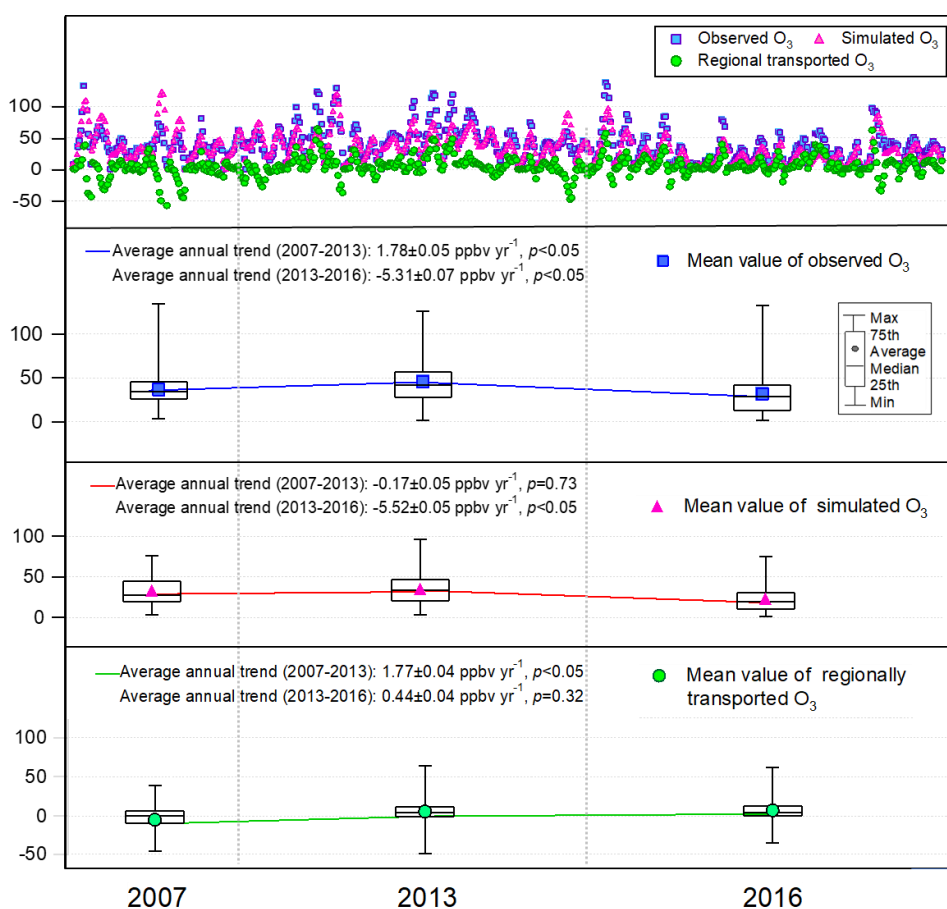
425

426 Figure 3. Hourly mixing ratio of the simulated and observed O₃ at TC during the VOC
 427 sampling periods in 2007, 2013 and 2016. The arrows represent the hourly wind sectors
 428 monitored at the sampling site.

429 **3.2.2 Inter-annual variations of the locally produced and regional transported O₃**

430 As discussed in section 2.5, the simulated O₃ in scenario A could be regarded as the locally
 431 produced O₃. Therefore, the differences between the observed O₃ and O₃ simulated in
 432 scenario A were treated as the regionally transported O₃ (Wang et al., 2017a). It is
 433 noteworthy that some negative values were generated with this method, corresponding to the
 434 dilution of the south winds to the locally produced O₃ as elaborated in section 3.2.1. Figure 4
 435 shows the hourly mixing ratios of the observed, local and regional O₃ at TC in daytime hours
 436 (07:00-19:00 LT) of the three sampling campaigns. Overall, the observed O₃ was mainly
 437 (88.7±2.5%) contributed by the local photochemical production, with regional transport only
 438 accounting for 11.3±2.5% of the observed daily maximum O₃. However, regional transport
 439 was responsible for as high as 58.0±5.4% of the observed daily maximum O₃ in Hong Kong
 440 on the O₃ episode days when northerly winds prevailed, indicating the heavy O₃ burden
 441 superimposed by regional air masses from PRD. From 2007 to 2013, the simulated locally-
 442 produced O₃ remained statistically unchanged ($p>0.1$), in contrast to the increase of observed

443 O₃ and regional O₃ at rates of 1.78 ± 0.05 ppbv yr⁻¹ ($p < 0.05$) and 1.77 ± 0.04 ppbv yr⁻¹ ($p < 0.05$),
 444 respectively, the same trends as those reported by Wang et al. (2017a) for the autumn O₃
 445 during 2005-2013. However, the decrease of the locally produced O₃ in the same period as
 446 that simulated by Wang et al. (2017a) was not seen here according to the simulated O₃ in the
 447 2007 and 2013 sampling campaigns. This discrepancy was likely caused by the limited
 448 samples in this study, no OVOCs considered in Wang et al. (2017a) and/or the inexactly
 449 same study periods between the two studies. Instead, we found that the locally produced O₃
 450 showed a significant decline at a rate of -5.52 ± 0.05 ppbv yr⁻¹ during 2013-2016 ($p < 0.05$),
 451 when the regionally transported O₃ did not change ($p = 0.32$), resulting in a downward trend ($-$
 452 5.31 ± 0.07) of the observed O₃. As such, the increase of the observed O₃ from 2007 to 2013
 453 was reversed by the decrease between 2013 and 2016, leading to an overall decreasing trend
 454 of the observed O₃ during 2007-2016 (rate = -0.57 ± 0.03 ppbv yr⁻¹, $p < 0.05$).

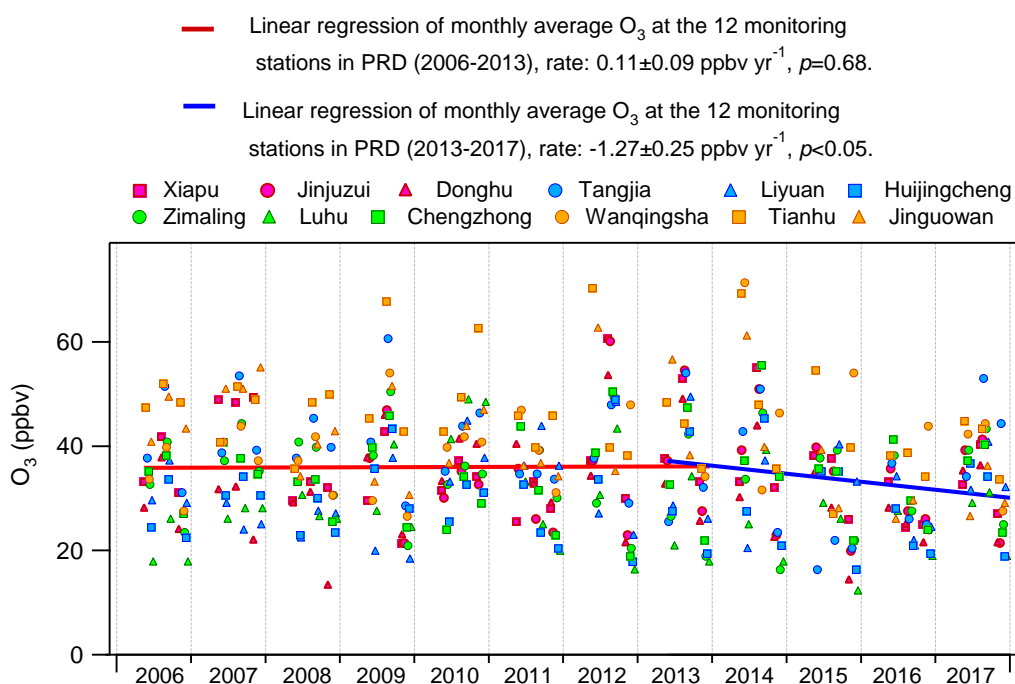


455

456 Figure 4. Hourly values (first panel) and the throughout-campaign statistical results (second
457 to fourth panels) of the observed, simulated (locally-produced) and regional O₃ mixing ratios
458 in daytime hours (07:00 – 19:00 LT) in the three sampling campaigns.

459 The significant alleviation of O₃ pollution in Hong Kong from 2013 to 2016 might be related
460 to the measures taken to control the emissions of O₃ precursors in Hong Kong and in
461 mainland China. The effectiveness of the actions launched by Hong Kong government in O₃
462 abatement was fully demonstrated in previous studies (Xue et al., 2014a; Lyu et al., 2017a;
463 Wang et al., 2017a), and would be further evaluated in this study (section 3.4). Besides, the
464 emission controls in mainland China might contribute to the decrease of O₃ or at least lessen
465 the regional O₃ burden in this period. For example, the China's NO_x emissions for the first
466 time showed a decreasing trend from 2013, benefited from the implementation of the China's
467 Clean Air Action Plan (Zheng et al., 2018). Furthermore, we looked into the monthly average
468 O₃ observed at the 12 air quality monitoring stations across the inland PRD, including three
469 regional monitoring stations, *i.e.* Tianhu, Wanqingsha and Jinguowan, and nine urban
470 monitoring stations, *i.e.* Xiapu, Jinjuzui, Donghu, Tangjia, Liyuan, Huijingcheng, Zimaling,
471 Luhu and Chengzhong (https://www.epd.gov.hk/epd/sc_chi/resources_publications/m_report.html). As shown in Figure 5, O₃ at these stations remained relatively
472 stable ($p=0.68$) during 2006-2013, which however showed a contrastively decreasing trend at
473 a rate of -1.27 ± 0.25 ppbv yr⁻¹ from 2013 to 2016. This corroborated our modelling results
474 that the regional contribution to O₃ in Hong Kong ceased increasing or even began to
475 decrease since 2013. While the substantial decrease of NO_x was a plausible reason for the
476 alleviated regional O₃ pollution, meteorological variations might also play roles in
477 modulating O₃ variations in these years (Li et al., 2019). However, analyses on the causes are
478 out of the scope of this study. In addition to the reduced local formation and regional

480 transport of O₃, the more favourable meteorological conditions in 2016 might be another
481 reason of the O₃ decrease, as discussed in section 3.1.



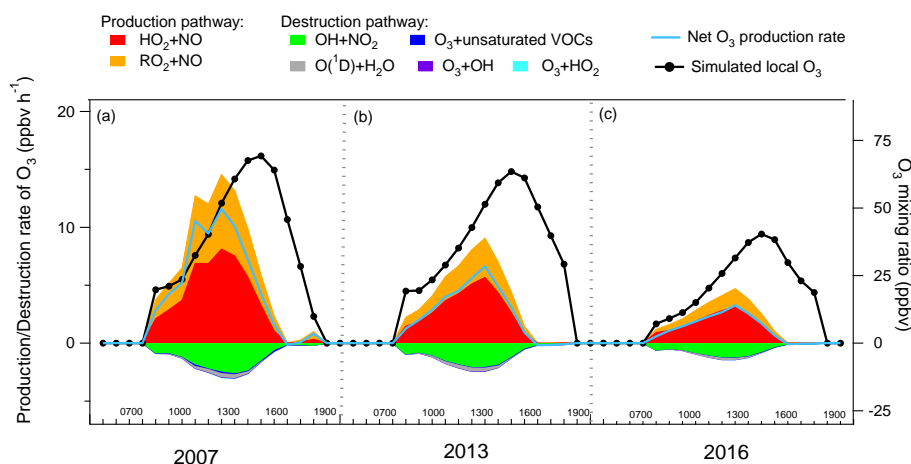
482
483 Figure 5. Trends of the observed monthly average O₃ at the 12 air quality monitoring stations
484 in inland PRD.

485 3.3 Local production and destruction pathways of O₃ and OH radical

486 3.3.1 In-situ net O₃ production

487 Figure 6 shows the average diurnal profiles of the simulated O₃ production and destruction
488 pathways during the three sampling campaigns. Also shown are the average diurnal cycles of
489 the simulated O₃. The shift of the peaks between the net O₃ production rate and the simulated
490 O₃ was due to the accumulation of the newly generated O₃ over time in the model, which was
491 also true in the real situations. The reactions between NO₂ and O₃, leading to the formation of
492 NO₃ and N₂O₅, in addition to dry deposition and aloft exchange, were the main depletions of
493 the simulated O₃ in the late afternoon. Consistent with previous studies (Kanaya et al., 2009;

494 Liu et al., 2012; Xue et al., 2014b), these pathways were not included in the calculation of the
495 net O₃ production rate, because we mainly focused on the photochemical processes in the
496 hours when O₃ was accumulated. It was found that the reaction between HO₂ with NO
497 dominated the O₃ production rates in all the cases, with an average rate of 3.7±0.7 ppbv h⁻¹
498 (56.5±1.1%, percentage of the total O₃ production rate, same below), 2.5±0.3 ppbv h⁻¹
499 (64.3±0.8%) and 1.4±0.2 ppbv h⁻¹ (67.7±0.7%) in the 2007, 2013 and 2016 sampling
500 campaigns, respectively. In addition, the sum of the reaction rates between RO₂ radicals and
501 NO contributed 3.0±0.6 ppbv h⁻¹ (43.5±1.1%), 1.5±0.2 ppbv h⁻¹ (35.7±0.8%) and
502 0.7±0.1 ppbv h⁻¹ (32.3±0.7%) to the O₃ production rate in 2007, 2013 and 2016, respectively.
503 The formation of HNO₃ though the reaction between OH and NO₂ served as the main
504 scavenger pathway of O₃, as NO₂ would be photolyzed and produce O₃ otherwise. On
505 average, O₃ was consumed in this way at a rate of -1.3±0.2 ppbv h⁻¹ (80.7±3.3%, percentage
506 of the total O₃ destruction rate, same below), -1.0±0.1 ppbv h⁻¹ (79.3±1.8%)
507 and -0.6±0.07 ppbv h⁻¹ (81.6±2.0%) in 2007, 2013 and 2016, respectively. The photolysis of
508 O₃ was the second contributor to O₃ destruction, with an average contribution of -0.11±0.01
509 ppbv h⁻¹ (8.5±0.5%) for the three sampling periods. Besides, the ozonolysis of unsaturated
510 VOCs and the reactions between O₃ and radicals (OH and HO₂) were responsible for 3.5±0.3%
511 and 1.7±0.2% of the total destruction rate of the locally produced O₃, respectively.
512 Overall, the net local O₃ production rate decreased from 5.2±1.1 ppbv h⁻¹ in 2007, to
513 2.7±0.4 ppbv h⁻¹ in 2013, till 1.4±0.3 ppbv h⁻¹ in 2016, corresponding to the decline of the
514 locally produced O₃ through 2007 to 2016 (Section 3.2.2).



515

516 Figure 6. Average diurnal profiles of the local O₃ production and destruction rates in the
 517 sampling campaigns of (a) 2007, (b) 2013 and (c) 2016.

518 3.3.2 Recycling of OH radical

519 As one of the most important radicals in the atmosphere, OH initiates the oxidation of VOCs,
 520 leading to O₃ formation. Figure 7 presents the average diurnal profiles of the simulated OH
 521 and the formation and loss pathways dominating the recycling of OH during the three
 522 sampling periods, which roughly followed the typical pattern of the intensities of
 523 photochemical reactions, *i.e.* higher at noon and lower at the beginning and end of the day.

524 On average, the simulated OH concentration was comparable ($p=0.4$) between the 2007
 525 sampling campaign ($1.6\pm 0.3\times 10^6$ molecules cm^{-3}) and the 2013 sampling campaign
 526 ($1.5\pm 0.2\times 10^6$ molecules cm^{-3}), but it decreased ($p<0.05$) to $1.0\pm 0.2\times 10^6$ molecules cm^{-3} in the
 527 2016 sampling campaign.

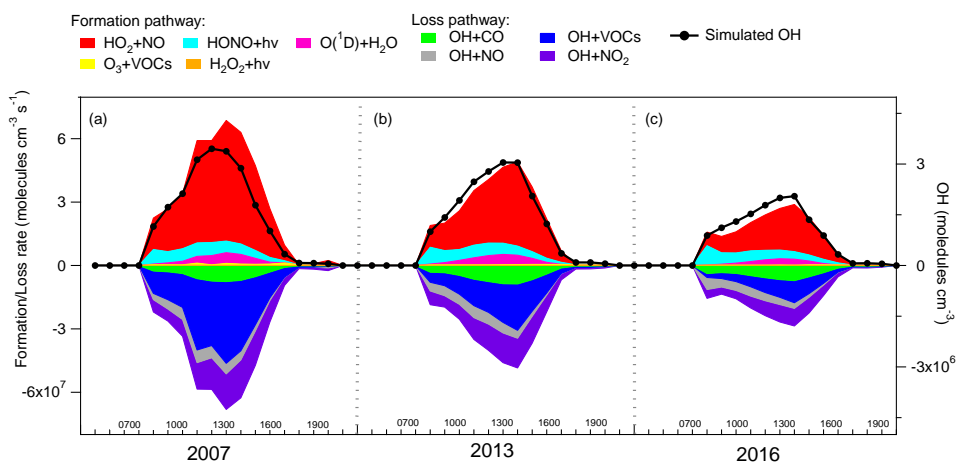
528 As expected, the formation and loss rates of OH were basically balanced in all the cases. OH
 529 was mainly formed from the reaction of HO₂+NO, which accounted for $69.8\pm 1.1\%$ of the
 530 total OH production rate over the three sampling campaigns. The photolysis of HONO ranked
 531 the second in supplying OH with the contribution of $22.0\pm 1.4\%$. As stated in section 2.4.2,
 532 the average diurnal cycle of HONO measured at TC in 2011 was adopted in the simulations.
 533 To assess the uncertainties, we also calculated the HONO concentrations according to the

534 measured HONO/NO_x ratios and the NO_x concentrations at TC in the three sampling
535 campaigns (Figure S5). The uncertainties in HONO concentrations and in the contributions of
536 HONO to OH formation and loss rates are discussed in Text S3. The formation of OH from
537 HONO photolysis was most efficient in the early morning, which was explained by the
538 morning peak of HONO concentration, due to the nocturnal heterogeneous formation and the
539 vehicle emissions in morning rush hours. Apart from the two dominant pathways, O₃
540 photolysis (6.3±0.2%), ozonolysis of unsaturated VOCs (1.5±0.2%) and H₂O₂ photolysis
541 (0.2±0.01%) also made some contributions to the formation of OH, with the highest rates at
542 noon or in the early afternoon when the productions of O₃ and H₂O₂ were the most intensive.
543 To sum up, the total formation rates of OH from the primary sources (photolysis of HONO,
544 O₃ and H₂O₂, and ozonolysis of VOCs) were lower than the recycling rates of OH (HO₂+NO)
545 throughout the day at TC, consistent with the results in Xue et al. (2016) simulated at the
546 same site. The dominant role of HO₂+NO in OH formation at TC (average contribution of
547 69.8±1.1%) might be related to the abundant NO at this site. The same pathway was
548 simulated and accounted for only 42.7±0.2% of the total OH formation rate at an island more
549 than 40 km away from Hong Kong with very low NO concentrations, *i.e.* maximum of 0.56
550 ppbv (Wang et al., 2018a).

551 OH was mainly depleted by the reactions with VOCs (32.3±1.2%), NO₂ (31.9±0.9%), CO
552 (19.3±0.6%) and NO (16.5±1.1%). The reaction rates of OH+NO (formation rates of HONO)
553 had the highest values in the morning, approximately in line with the diurnal pattern of the
554 HONO photolysis rates, which however were not completely balanced due to the constraint
555 of HONO to observations in the model. The average net photolysis rates of HONO
556 (differences between the HONO photolysis and formation rates) were 0.68±0.21×10⁶,
557 0.70±0.12×10⁶ and 0.87±0.12×10⁶ molecules cm⁻³ s⁻¹ in the 2007, 2013 and 2016 sampling
558 campaigns, respectively. The losses of OH through the other pathways all exhibited the

559 highest efficiencies at noon or in the early afternoon. It should be noted that the reaction
 560 between OH and NO₂ was not only the sink of OH but also a termination reaction in the
 561 photochemical system. In comparison, the termination reaction rates were lower than the OH
 562 formation rates from the primary sources (photolysis of HONO, O₃ and H₂O₂, and ozonolysis
 563 of VOCs) in the morning (7:00 – 10:00 LT), which were reversed in the following hours of
 564 the day due to the increases in OH concentrations.

565 Consistent with the variations of the local O₃ production, both the local formation and loss
 566 rates of OH decreased through 2007 to 2016 ($p < 0.05$), with much more obvious reductions in
 567 the later phase (2013-2016). On one hand, the continuous reduction of VOCs resulted in
 568 lower HO₂ and RO₂ concentrations (Figure S6), hence the lower production rate of OH
 569 through the reaction of HO₂+NO. At the same time, the destruction rates of OH also
 570 decreased due to the reductions of OH and the O₃ precursors, except for CO (Figure 7 and
 571 Table 1). The decreases of the OH production and destruction rates indicated that the
 572 propagation of the reaction cycles, namely the recycling of OH, became slower from 2007 to
 573 2016. This also explained why the locally produced O₃ decreased in these ten years, since O₃
 574 is formed with the consumption and recycling of OH radical.



575
 576 Figure 7. Average diurnal cycles of the OH formation and loss rates during the sampling

577 periods in (a) 2007, (b) 2013 and (c) 2016.

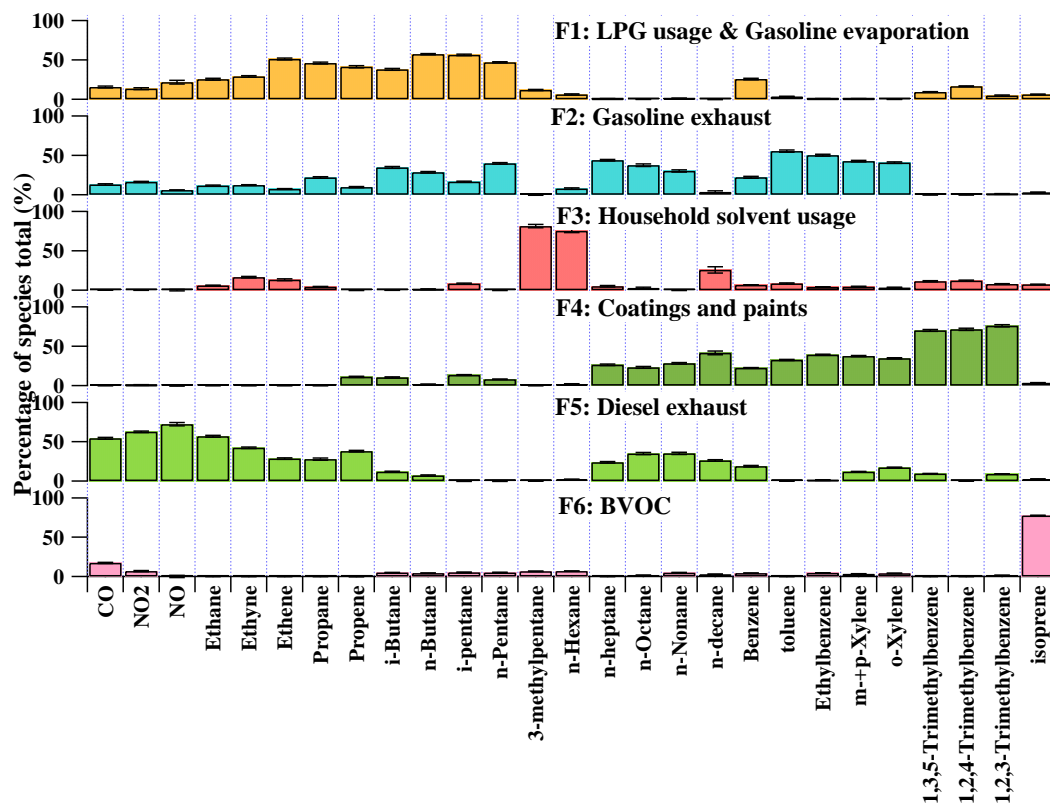
578 **3.4 Source contributions to the production of O₃ and radicals**

579 **3.4.1 Source apportionment**

580 To resolve the sources of O₃ precursors, 27 species, including CO, NO, NO₂, 12 alkanes, 4
581 alkenes and 8 aromatics, were applied to PMF for source apportionment. These species were
582 either of high abundances or typical tracers of VOC sources in Hong Kong. Source
583 apportionment was conducted for a total of 414 samples covering the three sampling periods,
584 so that the uncertainty of the source apportionment results could be reduced, compared to
585 separate source apportionments for each of the three sampling periods. Figure 8 shows the
586 average profiles of the six sources resolved by PMF. The modelling errors were estimated
587 with the bootstrap method integrated in PMF (Brown et al., 2015).

588 Factor 1 was assigned as the combination of LPG usage and gasoline evaporation, in view of
589 the high loadings of C₂-C₅ hydrocarbons. Specifically, propane and *i/n*-butanes are the main
590 components of LPG in Hong Kong, and gasoline evaporation generally contains large
591 quantities of *i/n*-pentanes, in particularly *i*-pentane (Guo et al., 2013a; Lyu et al., 2017a).
592 Factor 2 was characterized by moderate to high percentages of *i/n*-pentanes and TEX
593 (toluene, ethylbenzene and xylenes). These species are commonly seen in gasoline exhausts.
594 Therefore, we defined this factor as gasoline exhausts. Both the third and fourth factors
595 indicated solvent-related emissions. While Factor 3 likely represented household solvent
596 usage, due to the dominance of hexane and hexane isomer (3-methylpentane) (Ling and Guo,
597 2014; Ou et al., 2015), Factor 4 was more related to emissions from coatings and paints, in
598 view of the dominance of the aromatics (Ling and Guo, 2014). Factor 5 was distinguished by
599 the high concentrations of ethane, ethene, ethyne and benzene, together with the relatively
600 heavy (C₇-C₁₀) alkanes, which are typical species in diesel exhausts (Schauer et al., 1999;
601 Kashdan et al., 2008; Sahoo et al., 2011). Therefore, this factor was designated as diesel

602 exhausts. The last factor denoted for biogenic emissions (BVOCs), due to the exclusive
 603 dominance of isoprene (Guenther, 2006).



604
 605 Figure 8. Average profiles of the O₃ precursors sources at TC in the three sampling
 606 campaigns. The uncertainties were estimated with the bootstrap method in PMF.

607 Figure S7 presents the total mixing ratio of VOCs emitted from each individual source
 608 extracted from PMF during the three sampling periods in Hong Kong. The VOC emissions
 609 from LPG usage and gasoline evaporation decreased significantly ($p < 0.05$) at a rate of -
 610 2.61 ± 0.03 ppbv yr⁻¹ from 2007 to 2016. However, the VOCs in association with gasoline
 611 exhausts experienced an increase (rate = 1.32 ± 0.02 ppbv yr⁻¹, $p < 0.05$) in these years,
 612 indicating that the reduction of VOC emissions from LPG usage and gasoline evaporation
 613 was not attributable to the change in emissions of gasoline-fuelled vehicles. Insight into the
 614 mixing ratios of propane and *i/n*-butanes (LPG tracers) in this source revealed a significant
 615 decline from 3.51 ± 0.52 ppbv in the 2007 sampling campaign to 1.27 ± 0.11 ppbv in the 2016

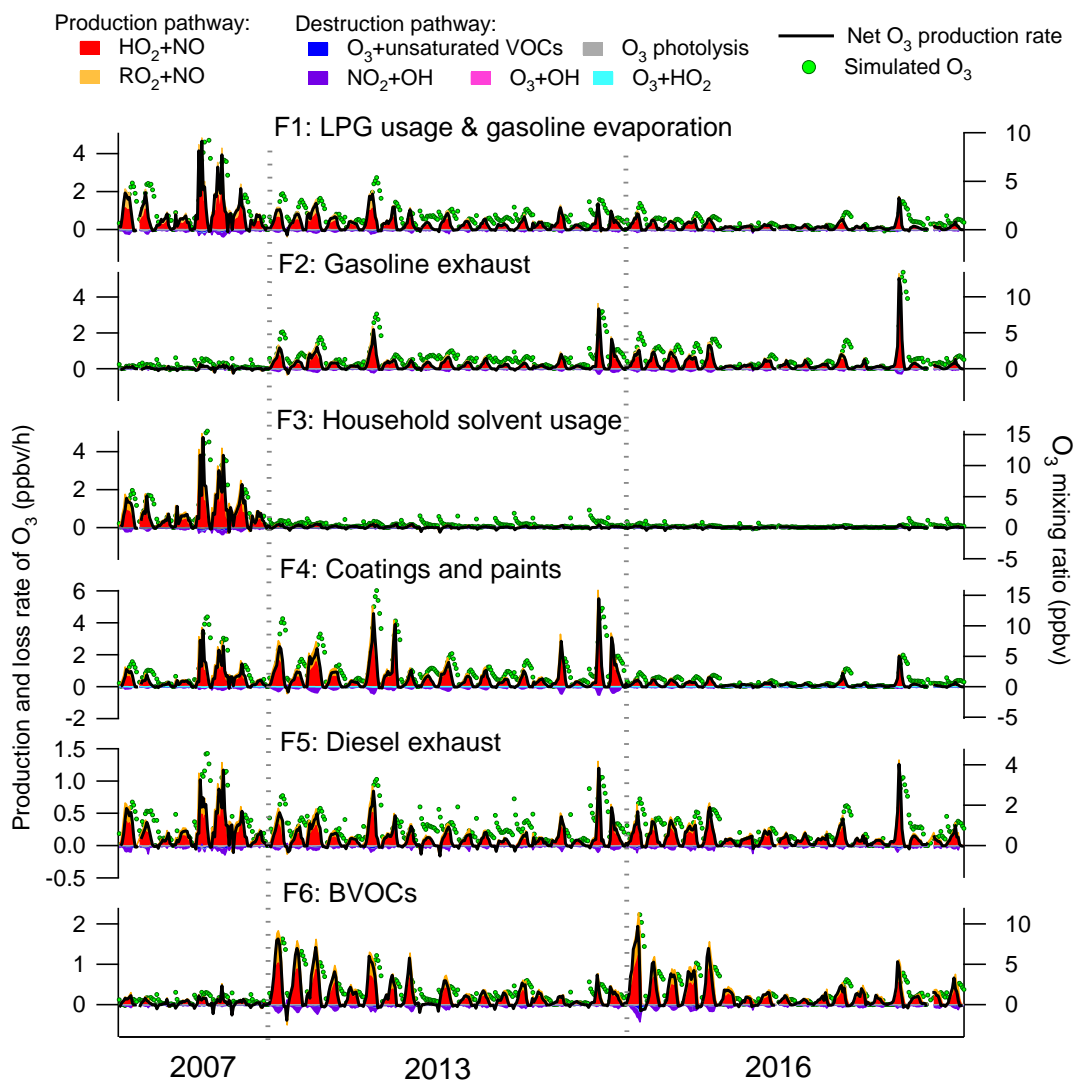
616 sampling campaign. Therefore, the reduction of VOC emissions from LPG usage was most
617 likely the reason of the decrease of VOCs allocated to the source of LPG usage and gasoline
618 evaporation. In fact, it was confirmed by our previous studies (Lyu et al., 2016b; Yao et al.,
619 2019) that the replacement of catalytic converters on LPG-fuelled vehicles during September
620 2013-May 2014 effectively reduced the VOC emissions from LPG-fuelled vehicles in Hong
621 Kong. In addition, the variations in LPG usage in inland PRD, where LPG was extensively
622 used as vehicular and domestic fuels (Liu et al., 2008), might also contribute to the emission
623 reduction of VOCs, in view of the decrease of LPG tracers in this source from 2007
624 (3.51 ± 0.52 ppbv) to 2013 (2.04 ± 0.27 ppbv), when no control was performed against LPG
625 fuelled vehicle emissions in Hong Kong. The VOCs emitted from solvent usage (including
626 the household solvent, coatings and paints) also decreased significantly ($p<0.05$) from 2007
627 to 2016, likely benefiting from the actions taken to restrict the VOC contents in solvent
628 products starting from 2007 (phase I) and 2010 (phase II) in Hong Kong (Lyu et al., 2017a).
629 VOCs attributable to diesel exhausts decreased ($p<0.05$) from the 2007 (2.6 ± 0.3 ppbv) to
630 2013 sampling campaign (2.0 ± 0.2 ppbv), which however were unchanged between 2013 and
631 2016 (2.2 ± 0.2 ppbv). In fact, a subsidy program has been implemented in Hong Kong since
632 2007 to progressively eliminate the pre-Euro IV diesel vehicles or to upgrade their emission
633 standards to Euro IV (HKEPD, 2017b), and the effectiveness of this program in VOC
634 reductions till 2013 was confirmed by Lyu et al. (2017a) with the online measurement data at
635 the same site. However, while the phase III of this program (2014-2019) is still ongoing, the
636 VOCs emitted from diesel vehicles remained stable between the 2013 and 2016 sampling
637 campaigns. This undesirable result might be due to the fact that the actions were mainly
638 targeted at the pre-Euro, Euro I and Euro II diesel vehicles before 2013, whereas the phase III
639 of the program initiated in 2014 focused on the Euro III vehicles (HKEPD, 2017b, 2018).
640 Since the former were vehicles with higher emissions, it is not unreasonable that reduction of

641 VOCs was more discernible between 2007 and 2013. Further, the effectiveness of the phase
642 III program might be somewhat offset by the wearing-out of the pre-existing vehicles and the
643 increase of diesel vehicle populations (Competition Commission, 2017). Further evaluation
644 with more data in a longer period is recommended. At last, the increase of BVOCs from 2007
645 to 2013 but comparable levels between 2013 and 2016 seemed to be related to the lower
646 ($p < 0.05$) temperature in the 2007 sampling campaign (Figure S8 and Table S6). Besides, the
647 more frequent (62.8%) southeast winds from SCS with higher wind speeds ($2.3 \pm 0.2 \text{ m s}^{-1}$)
648 might dilute BVOCs emitted from the terrestrial plants in the 2007 sampling campaign.

649 **3.4.2 Source contributions to O₃ production**

650 Figure 9 presents the contributions of VOCs emitted from individual sources to the
651 production and destruction rates of O₃, as well as the simulated contributions to the O₃
652 mixing ratios. NO_x was not included in these analyses, because of its relatively high
653 uncertainties in source apportionment results due to the short lifetimes. Consistent with the
654 O₃ production and destruction in the whole air, the pathway of HO₂+NO dominated over the
655 reactions between RO₂ and NO in O₃ production for all the individual sources. The
656 destruction of O₃ was mainly driven by NO₂ reacting with OH. For the net O₃ production rate,
657 VOCs attributable to the coatings and paints made the largest contribution ($0.38 \pm 0.05 \text{ ppbv h}^{-1}$)
658 ¹), followed by gasoline exhausts ($0.22 \pm 0.03 \text{ ppbv h}^{-1}$), LPG and gasoline evaporation
659 ($0.21 \pm 0.03 \text{ ppbv h}^{-1}$), BVOCs ($0.19 \pm 0.03 \text{ ppbv h}^{-1}$), household solvent usage ($0.15 \pm 0.04 \text{ ppbv}$
660 h^{-1}) and diesel exhausts ($0.13 \pm 0.01 \text{ ppbv h}^{-1}$). Despite some peak shifts for the reasons
661 illustrated in section 3.3.1, the O₃ mixing ratios elevated by the individual sources followed
662 the same pattern as the net O₃ production rates, with the highest O₃ enhancement (1.92 ± 0.21
663 ppbv) by the source of coatings and paints and the lowest increase by household solvent
664 usage ($0.86 \pm 0.06 \text{ ppbv}$) and diesel exhausts ($0.83 \pm 0.06 \text{ ppbv}$). The contributions of source-
665 specific VOCs to O₃ production, particularly the importance of solvent usage in O₃ formation

666 in Hong Kong, were generally in line with previous studies (Ling and Guo, 2014; Ou et al.,
 667 2015). This was actually expected according to the reactivity of major VOCs in each source.
 668 For example, the TEX in the source of coatings and paints (Figure 8) have been identified to
 669 be of high O₃ formation potentials (Lau et al., 2010; Ling et al., 2011, 2013). However, the
 670 PBM-MCM model simulations enabled us to quantitatively evaluate the contributions of
 671 VOC sources to O₃ production rates.



672
 673 Figure 9. Contributions of VOCs in individual sources to the production and destruction rates
 674 of O₃ and to the O₃ mixing ratios in the three sampling campaigns.

675 From a historical perspective, we found that the contribution of LPG usage and gasoline
676 evaporation to O₃ production significantly decreased ($p<0.05$) from 2007 to 2016 sampling
677 campaign (2007: 0.51 ± 0.11 ppbv h⁻¹; 2013: 0.20 ± 0.03 ppbv h⁻¹; 2016: 0.10 ± 0.02 ppbv h⁻¹),
678 which coincided with the variations of VOCs emitted from LPG-fuelled vehicles as discussed
679 above. Gasoline exhaust contributed much less ($p<0.05$) to the net O₃ production rate in 2007
680 (0.02 ± 0.01 ppbv h⁻¹), than those in 2013 (0.26 ± 0.05 ppbv h⁻¹) and 2016 (0.27 ± 0.07 ppbv h⁻¹),
681 in line with the variations of VOCs emitted from this source. The reductions of VOC
682 emissions from solvents also resulted in the consistent decrease of the net O₃ production rate
683 from 1.22 ± 0.17 ppbv h⁻¹ in the 2007 to 0.14 ± 0.05 ppbv h⁻¹ in the 2016 sampling campaign.
684 The O₃ production rates contributed by VOCs in diesel exhausts were reduced from 2007
685 (0.21 ± 0.05 ppbv h⁻¹) to 2013 (0.11 ± 0.02 ppbv h⁻¹) and remained unchanged thereafter (2016:
686 0.11 ± 0.02 ppbv/h). The O₃ production rate traceable to BVOCs showed a significant increase
687 from 2007 (0.04 ± 0.02 ppbv h⁻¹) to 2016 (0.22 ± 0.04 ppbv h⁻¹), since the mixing ratios of
688 BVOCs significantly increased ($p<0.05$) in these years. It is noteworthy that the changes in
689 meteorological conditions in these three sampling campaigns might also partially account for
690 the variations in the source contributions to O₃ production. For example, the 2013 sampling
691 campaign was characterized by the relatively higher temperature and lowest relative humidity
692 among the three sampling periods, which favoured O₃ formation in 2013 (Table S6). Besides,
693 due to limited samples in this study, we recommend further assessments with more data in
694 longer periods to be carried out in future study.

695 **4 Conclusions**

696 Photochemical pollution with high and increasing concentrations of O₃ has been an important
697 environmental issue in South China. With the observation data of O₃ and its precursors at a
698 suburban site in Hong Kong, downwind of South China, this study analysed the inter-annual
699 variations of O₃ and its photochemistry, as well as the contributions of VOC sources to the

700 local O₃ production rates in 2007, 2013 and 2016. To our knowledge, this is the first time that
701 a substantial alleviation of O₃ pollution in this region was identified between 2013 and 2016,
702 in contrast to the repeatedly confirmed O₃ increase before 2013. In addition to the changes in
703 meteorological conditions among the three sampling campaigns, the termination of the rise in
704 regionally transported O₃ and the decrease of the local O₃ production rate contributed to the
705 decline of O₃ in the later period. The emission reductions (particularly for NO_x) in mainland
706 China starting from 2013, the year when the China's Clean Air Action Plan was launched,
707 might more or less play a role in ceasing the increase of regional O₃. In Hong Kong, the
708 replacement of catalytic converters and the constraints of VOC contents in solvent products
709 led to the reductions of VOC emissions from LPG-fuelled vehicles and solvent usage,
710 respectively. As a result, the local O₃ production rate and the recycling rate of OH radical
711 decreased substantially from 2013 to 2016. Though the variations in meteorological
712 conditions and the limited sample size might somewhat introduce uncertainties to the
713 conclusions drawn from the present study, it is plausible that the local and regional
714 interventions were effective on the control of O₃ pollution in Hong Kong. Nevertheless,
715 studies with more data in longer periods should be conducted, not only in Hong Kong but
716 also in mainland China where O₃ is still increasing in most of the territories.

717 **Author contribution**

718 Hai Guo and Fei Jiang initiated and designed the experiments, and Xufei Liu and Xiaopu Lyu
719 carried them out. Xiaopu Lyu and Yu Wang developed the model code and performed the
720 simulations. Xufei Liu and Xiaopu Lyu prepared the manuscript and Hai Guo finalized the
721 manuscript with contributions from all co-authors.

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