

1 Characterization of urban amine-containing particles in Southwestern China: seasonal
2 variation, source, and processing

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22

Abstract

23 Amine-containing particles were characterized in an urban area of Chongqing during
24 both summer and winter using a single particle aerosol mass spectrometer (SPAMS).
25 Among the collected particles, 12.7% were amine-containing in winter and 8.3% in
26 summer. Amines were internally mixed with elemental carbon (EC), organic carbon
27 (OC), sulfate, and nitrate. Diethylamine (DEA) was the most abundant among amine-
28 containing particles. Wintertime amine-containing particles were mainly from the
29 northwest direction where a forest park was located; in summer, they were from the
30 northwest and southwest (traffic hub) directions. These origins suggest that vegetation
31 and traffic were the primary sources of particulate amines. The average relative peak
32 area of DEA depended strongly on humidity, indicating that the enhancement of DEA
33 was possibly due to increasing aerosol water content and aerosol acidity. Using an
34 adaptive resonance theory neural network (ART-2a) algorithm, four major types of
35 amine-containing particles were clustered including amine-organic-carbon (A-OC), A-
36 OCEC, DEA-OC, and A-OCEC-aged. The identified particle types implied that amines
37 were taken up by particles produced from traffic and biomass burning. Knowledge
38 gained in this study is useful to understand the atmospheric processing, origin, and
39 sources of amine-containing particles in the urban area of Chongqing.

40

41 **1. Introduction**

42 Amines are ubiquitous in the atmosphere and have both natural (ocean, biomass burning,
43 and vegetation) and anthropogenic (animal husbandry, industry, combustion, and traffic)
44 emission sources (Ge et al., 2011a). Trimethylamine (TMA) is one of the most abundant
45 amines with an estimated global emission flux of 170Gg year⁻¹ (Ge et al., 2011a).
46 Gaseous amines compete with ammonia in acid-base reactions, participate in gas-
47 particle partitioning, and contribute to wet and dry deposition (Angelino et al., 2001;
48 Monks, 2005; Gómez Alvarez et al., 2007; De Haan et al., 2011; Huang et al., 2012;
49 You et al., 2014). Gaseous amines also play an essential role in new particle formation
50 via enhancing the ternary nucleation of H₂SO₄-H₂O clusters in remote areas (Bzdek et
51 al., 2012; Kirkby et al., 2011). In polluted areas, H₂SO₄-diethylamine (DMA)-water
52 clusters were important during the new particle formation events (Yao et al., 2018).
53 Amines are also essential in the growth of ambient particles. For example, particulate
54 aminium salts, which were produced via amine-acid neutralization, tended to prevent
55 the coagulation between pre-existing particles leading to increased particle number
56 concentrations (Wang et al., 2010; Smith et al., 2010). Moreover, the enrichment of
57 TMA had been observed in cloud and fog processing (Zhang et al., 2012; Rehbein et
58 al., 2011). Characterization of amine-containing particles is important to evaluate their
59 processing and impact.

60 Single particle mass spectrometers (SPMS), such as aerosol time-of-flight mass
61 spectrometer (ATOFMS) and Single Particle Aerosol Mass Spectrometer (SPAMS),
62 have been widely used in real-time measurements of amine-containing particles for
63 chemical composition and mixing state (Li et al., 2017). SPAMS is different from the
64 Aerodyne soot-particle aerosol mass spectrometer (SP-AMS), which is a type of

65 aerosol mass spectrometer (AMS) for detecting black carbon, sulfate, nitrate,
66 ammonium, chloride, and organics (Onasch et al., 2012; Wang et al., 2016). The
67 chemical composition and mixing state of TMA-containing particles have been
68 reported worldwide, such as in California, USA (Denkenberger et al., 2007; Qin et al.,
69 2012)); Ontario, Canada (Tan et al., 2002; Rehbein et al., 2011); Mexico City (Moffet
70 et al., 2008)); European cities (Barcelona, Cork, Zurich, Paris, Dunkirk and Corsica
71 (Healy et al., 2015; Dall'Osto et al., 2016)), and Chinese cities such as Guangzhou,
72 Shanghai and Xi'an (Zhang et al., 2012; Chen et al., 2016; Huang et al., 2012). In the
73 five European cities such as Cork, Paris, Dunkirk, Corsica, and Zurich, amines were
74 found internally mixed with sulfate and nitrate; but in Corsica, amines were internally
75 mixed with methanesulfonate (Healy et al., 2015). In Barcelona, five unique types of
76 amine-containing particles were observed, including *amine-POA58* (composed of
77 amines, sulfate, and nitrate), *amine-EST84*(environment tobacco smoke), *amine-*
78 *SOA59* (composed of TMA and organics), *amine-SOA114*, and organic nitrogen
79 amines (Dall'Osto et al., 2016; Dall'Osto et al., 2013). In a rural area site in the Pearl
80 River Delta (China), the marker ion, $(C_2H_5)_2NH_2^+$, was the most abundant (90% and
81 86% of amine-containing particles in summer and winter) (Cheng et al., 2018). In
82 Guangzhou, TMA-containing particles were important, taking up to 7% in number
83 fraction during clear days and 35% during fog events (Zhang et al., 2012). In previous
84 studies, reported high RH conditions and fog processing were favorable for the
85 enhancement of trimethylamine in the particle phase. Zhang et al. found that, during
86 fog events, the number fraction of TMA-containing particles took up to 35%; in the
87 size range of 0.5-2.0 μm , the fraction accounted up to 60% (Zhang et al., 2012). Thus
88 the location-specific studies in the varied environments are still necessary.

89 The knowledge of amine-containing particles is limited in southwestern China. In this
90 region, Chongqing is a megacity with a population of 8.23 million. The city is
91 subtropical, industrial, and polluted (Chen et al., 2017; Tao et al., 2017). Fog events
92 frequently occurred in this area, and hence, it is known as the “fog city” in China. The
93 effect of high relative humidity (RH) on the processing of amine-containing particles
94 needs investigation. This study aims to characterize the amine-containing particles,
95 including chemical composition, mixing state, atmospheric processing, and source in
96 Chongqing during winter and summer.

97 **2. Methods**

98 **2.1 Sampling site**

99 Ambient single particles were collected at an urban air quality supersite from
100 07/05/2016 to 08/14/2016 (referred to as a summer season) and from 01/21/2016 to
101 02/25/2016 (referred to as a winter season). The supersite has been described in our
102 previous studies (Chen et al., 2017). Briefly, the supersite is located on the rooftop of a
103 commercial office building (106.51°E, 29.62°N) at a height of 30 m above the ground
104 (Figure S1). The building is surrounded by business and residential communities and is
105 15 km from the city center. A 3 km² forest park is located northwest of the sampling
106 site and a traffic hub in the southwest.

107 **2.2 Instrumentation**

108 A SPAMS (Hexin Inc. Guangzhou, China, model 0515) was deployed for single
109 particle sampling, and the technical description of SPAMS is available in the literature
110 (Li et al., 2011; Chen et al., 2017). Briefly, after passing through a diffusive dryer,
111 particles in a size range of 0.1–2.0 μm are sampled via an aerodynamic lens and form

112 a particle beam. Particles in the beam come across two pre-positioned laser beams (Nd:
113 YAG, 532 nm) one-by-one, and the vacuum aerodynamic diameter (D_{va}) of each
114 particle is determined via its time-of-flight. Particles are ionized using an Nd: YAG
115 laser operating at a wavelength of 266 nm. The ions are analyzed using a bipolar time-
116 of-flight mass spectrometer. Due to the limitation of SPAMS, quantification of
117 particulate amines was not attempted.

118 **2.3 Data analysis**

119 The SPAMS data were imported into the YAADA toolkit (Software Toolkit to Analyze
120 Single-Particle Mass Spectral Data, v 2.11) to form a single particle dataset. The query
121 was conducted using the marker ions of amines.: m/z 59 $[(CH_3)_3N]^+$ (TMA), 74
122 $[(C_2H_5)_2NH_2]^+$ (diethylamine, DEA), 86 $[(C_2H_5)_2NCH_2]^+$ or $[C_3H_7NHC_2H_4]^+$ (DEA or
123 DPA), 101 $[(C_2H_5)_3N]^+$ (TEA), 102 $[(C_3H_7)_2NH_2]^+$ (DPA), 114 $[(C_3H_7)_2NCH_2]^+$ (DPA),
124 and 143 $[(C_3H_7)_3N]^+$ (TPA) (Healy et al., 2015). Firstly, m/z 59 was used for querying
125 the TMA-containing particles; m/z 74 for the DEA-containing particles and m/z 86 for
126 TEA-containing particles, and so on. The query strategy resulted in duplicate particles
127 in the result when various amines co-existed in one single amine-containing particle.
128 After the duplicate particles were removed from the multiple query results described
129 above, all amine-containing particles were combined into an amine-containing particle
130 cluster. Various amines could be both internally and externally mixed in these particle
131 clusters.

132 An adaptive resonance theory based neural network algorithm (ART-2a) was applied
133 to cluster the amine-containing particle types using a vigilance factor of 0.70, a learning
134 rate of 0.05, and 20 iterations (Song et al., 1999). This procedure produced 67 clusters
135 in summer and 75 clusters in winter; many of these clusters exhibited identical mass

136 spectra with slight differences in specific ion intensities. A well-established combining
137 strategy, on the basis of similar mass spectra, temporal trends, and size distribution, was
138 adopted to merge these particle clusters into the finalized particle types (Dallosto and
139 Harrison, 2006). In addition, the relative peak area (RPA) is defined as the peak area of
140 each m/z divided by the total dual-ion mass spectral peak areas of each particle (Healy
141 et al., 2013). To calculate the overall RPA of amines, the relative peak area of amines
142 in each particle were extracted and summed up.

143 **3. Results and discussion**

144 **3.1 Single particle chemical composition and seasonal variation**

145 The percentage of amine-containing particles was 12.7% in the winter SPAMS dataset
146 and 8.3% in the summer dataset. The DEA-containing particles were dominant,
147 accounting for 70% and 78% of all amine-containing particles in winter and summer,
148 respectively; while TMA-containing particles were minor, accounting for up to 7% in
149 winter and 3% in summer among all the amine-containing particles. The average mass
150 spectra of DEA-, DPA, and TMA-containing particles are provided in Figure S2, and
151 these spectra showed strong homogeneity. The determination coefficient (R^2) between
152 DEA- and DPA- containing particles was 0.98, and R^2 between DEA- and TMA-
153 containing particles was 0.83.

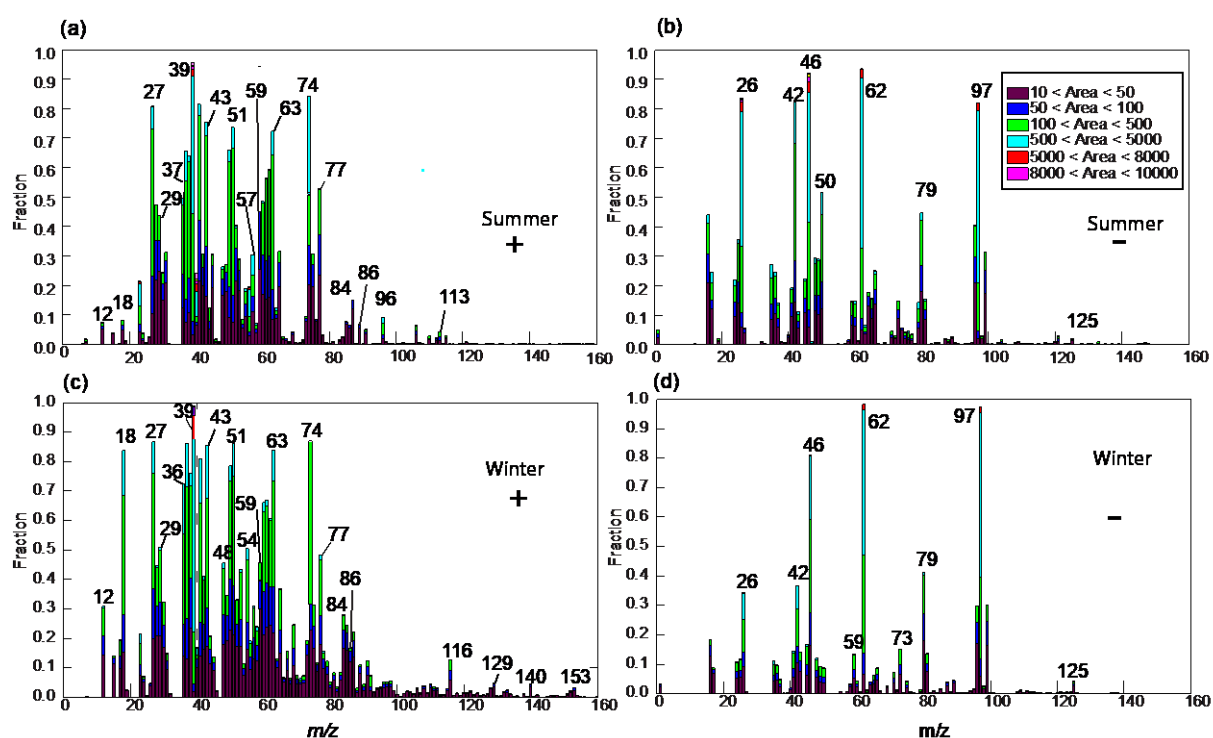
154 Figure 1 shows the digital mass spectra of amine-containing particles in two seasons.
155 In each spectrum, the ion height indicates its fraction in the amine-containing particle
156 dataset, and the stacked color map shows the corresponding ion intensity ranges. The
157 assignment of ions is shown in Table S1. In both seasons, the dominant ions were K^+
158 (m/z 39 and 41), amines (m/z 59, 74, and 86), and organics (m/z 43, 51, 63, and 77). The

159 mixing ratios of ammonium (NH_4^+ , m/z 18) and polycyclic aromatic hydrocarbons (e.g.,
160 m/z 116 ($[\text{C}_9\text{H}_8]^+$), 129 ($[\text{C}_{10}\text{H}_9]^+$), 140 ($[\text{C}_{11}\text{H}_8]^+$), and 153 ($[\text{C}_{12}\text{H}_9]^+$)) were higher in
161 winter than in summer. The strong signal of NH_4^+ was possibly due to the lower
162 temperature (8°C) in winter than in summer (31°C). The mixing ratios of m/z 59 were
163 45% and 44% during summer and winter, respectively.

164 In the negative mass spectra of two seasons (Figures 1(b) and 1(d)), the dominant ions
165 were CN^- (m/z -26), CNO^- (m/z -42), nitrate (m/z -46 and -62), phosphate (-79), and
166 sulfate (m/z -80 and -97). Primary species, such as CN^- and CNO^- were commonly
167 from biomass burning (BB) and organonitrogen (Pratt et al., 2011). Levoglucosan
168 markers from BB, such as -45, -59, and -71 were also detected. Dust markers, such as
169 $[\text{SiO}_2]^-$ (m/z -60), $[\text{}^{28}\text{SiO}_3]^-$ or $[\text{AlO}_2(\text{OH})]^-$ (-76), and $[\text{PO}_3]^-$, were also detected during
170 summertime, suggesting the influence of dust particles.

171 Prior to comparison, the ion peak was normalized using the method developed by Qin
172 et al. (2012). Briefly, the peak area of each m/z was divided by the total mass spectral
173 peak area matrix. The normalized ion intensity of the wintertime particles was
174 subtracted from that of the summertime particles. A positive value indicates the
175 normalized ion intensity was greater in the summer, whereas a negative value indicates
176 that the normalized ion intensity was greater in the winter. As shown in Figure S3, Ca^+
177 (m/z 40) and Fe^+ (m/z 56) were more prevalent during summer. Organic species, such
178 as C_2H_3^+ (m/z 27), C_4H_3^+ (m/z 51), C_5H_3^+ (m/z 63), and C_6H_5^+ (m/z 77) typically from
179 aromatic hydrocarbons, were also more abundant in summer. During wintertime,
180 signals of sulfate (m/z -97), NO_3^- (m/z -62), NH_4^+ (m/z 18), and K^+ (m/z 39) were more
181 prominent than in summer, suggesting that the wintertime particles contained more
182 secondary species than those in summer.

183 The unscaled size distribution of amine-containing particles also showed strong
 184 seasonal variations (Figure S4). Generally, amine-containing particles had monomodal
 185 size distributions in the droplet mode; and the distributions peaked at a larger D_{va} in
 186 summer than winter. For example, DEA-containing particles peaked at 0.6 μm in winter
 187 and 0.8 μm in summer, and DPA-containing particles at 0.7 μm in winter and 0.9 μm
 188 in summer. The size distributions of the major amine-containing particles suggested
 189 that these particles had undergone substantial aging processes.



190

191 Figure 1. (a) and (c): the positive digital mass spectrum of amine-containing particles
 192 during summer and wintertime, respectively; (b) and (d): the negative digital mass
 193 spectrum during summer and wintertime, respectively. The ion height indicates its
 194 fraction in the amine-containing particle dataset, and the stacked color map indicates
 195 the ion peak area range.

196 3.2 Temporal trend, diurnal pattern, and origin of amine-containing particles

197 Figure 2 shows the temporal trends of RH, temperature, number count, and the peak area
198 of amine-containing particles. The winter temperature was lower ($8.0\pm 4.0^\circ\text{C}$) than
199 summer ($31\pm 4^\circ\text{C}$), and RH in the winter was slightly higher ($70\pm 14\%$ versus $64\pm 16\%$)
200 (Table 1). Stagnant air conditions occurred in both seasons due to the low wind speeds
201 (Huang et al., 2017), and the winter wind speed was lower than in summer. The hourly
202 count of amine-containing particles was typically ten times higher in winter than
203 summer.

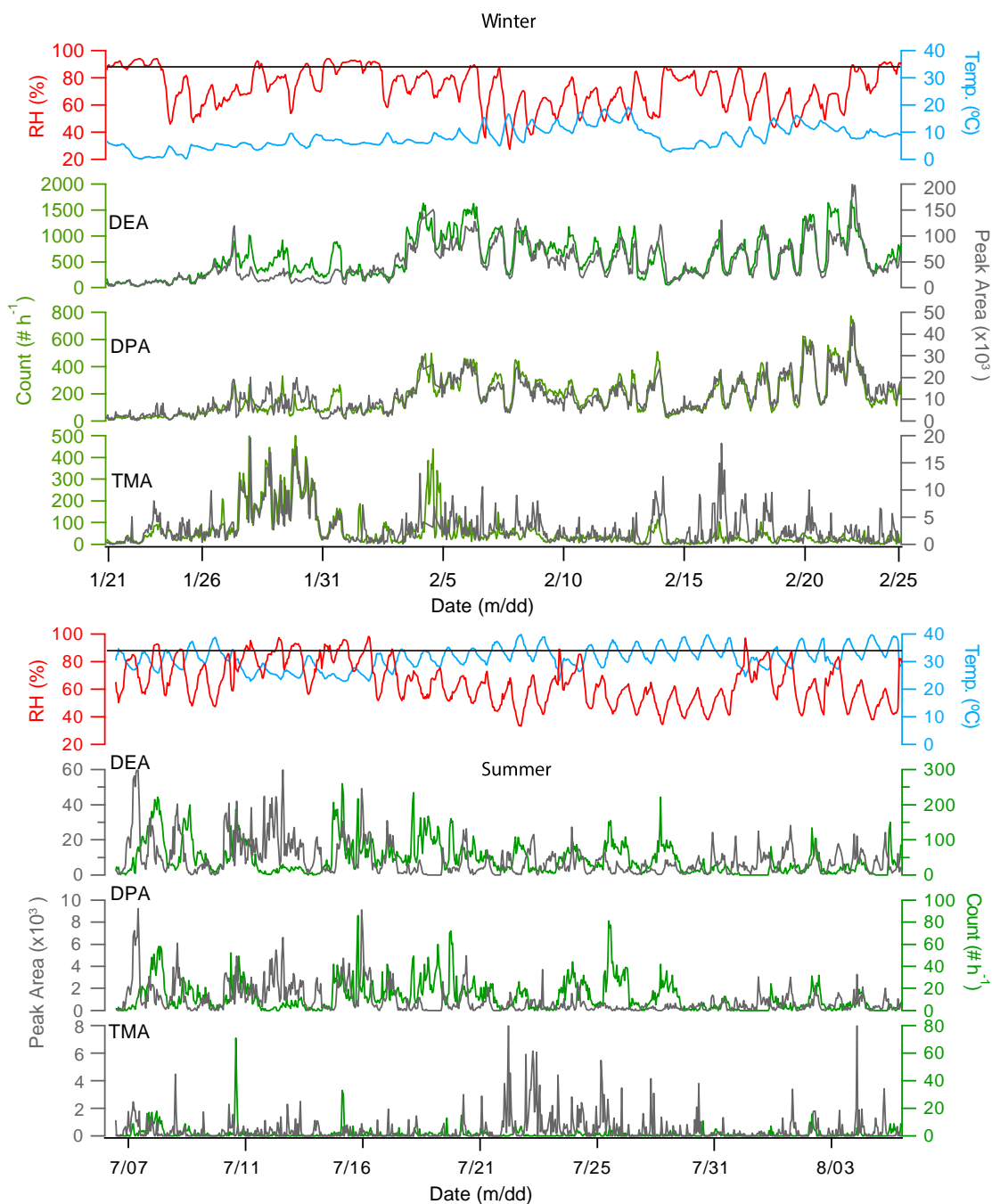
204 In winter, a good correlation existed between the temporal trends of hourly number
205 count and peak area of DEA-containing particles ($R^2 = 0.86$). The corresponding R^2 in
206 wintertime DPA-containing particles was 0.88. No such correlation for TMA-
207 containing particles was observed in winter ($R^2 = 0.22$) or summer (Figure 2). The
208 hourly counts of DEA- and DPA-containing particles were well correlated in both
209 summer ($R^2 = 0.63$) and winter ($R^2 = 0.87$), but a weak correlation ($R^2 = 0.25$) existed
210 between DEA- and TMA-containing particles. These results suggest DEA- and DPA-
211 containing particles were possibly from the same sources.

212 Table 1. Meteorological factors and particle counts in summer and winter.

	Winter	Summer
213		
214	Temperature ($^\circ\text{C}$)	8 ± 4 31 ± 4
	Relative humidity (%)	70 ± 14 64 ± 16
215	Wind Speed	1.2 ± 0.7 1.5 ± 1.0
216	Amine-particle Count ($\# \text{ h}^{-1}$)	587 ± 384 47 ± 26

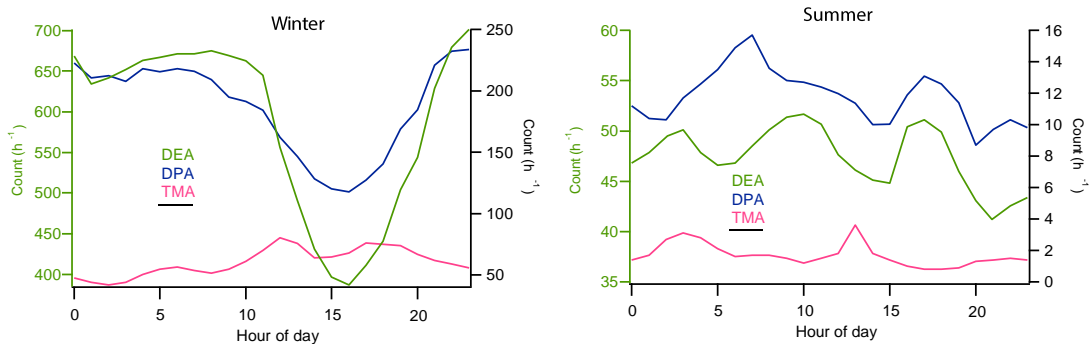
217 DEA- and DPA-containing particles remained at low levels from 1/20/2016 to
218 01/26/2016 and averaged at 109 and 26 count h⁻¹, respectively. During this period, wind
219 speed was relatively high, commonly above 1.5 ms⁻¹. TMA-, DEA-, and DPA-
220 containing particles started accumulating after 01/26/2016 when wind speed was low
221 (0.8 ms⁻¹) and wind direction from the northwest. After 02/03/2016, DEA- and DPA-
222 containing particles showed regular diurnal patterns with high levels of hourly count
223 during daytime on most days and minimum levels at 15:00. A similar diurnal pattern
224 was also observed for DPA-containing particles during wintertime (Figures 3). TMA-
225 containing particles presented a complex diurnal profile with peaks in the early morning
226 (4:00), at noon (12:00) and in the afternoon (18:00). The chemical composition and
227 diurnal pattern of TMA-containing particles were strongly connected to traffic
228 emissions.

229 Wind direction and number count of amine-containing particles were analyzed together
230 using bivariate polar plots (Figure 4). During wintertime, the dominant direction for
231 amine-containing particles was from the northwest where a forest park was located.
232 After being emitted from vegetation (plants, grass, and trees) (Ge et al., 2011a), DEA
233 could partition to the pre-existing particles before arriving at the sampling site. The
234 transport of these particles to the sampling site caused the elevation in the morning.
235 Based on the excellent correlation between DEA- and DPA-containing particles, DPA-
236 containing particles could also be from vegetation. It can be concluded that the major
237 source of amines in DEA- and DPA-containing particles was vegetation from the
238 northwest.



239

240 Figure 2. Temporal trends of relative humidity (RH), temperature (Temp.), hourly peak area (dark gray), and particle count (green) of DEA (m/z 74), DPA (m/z 86), and TMA
 241 (m/z 59) -containing particles in winter (top panel) and summer (bottom panel). The
 242 black lines in the two panels indicate RH of 90%.
 243



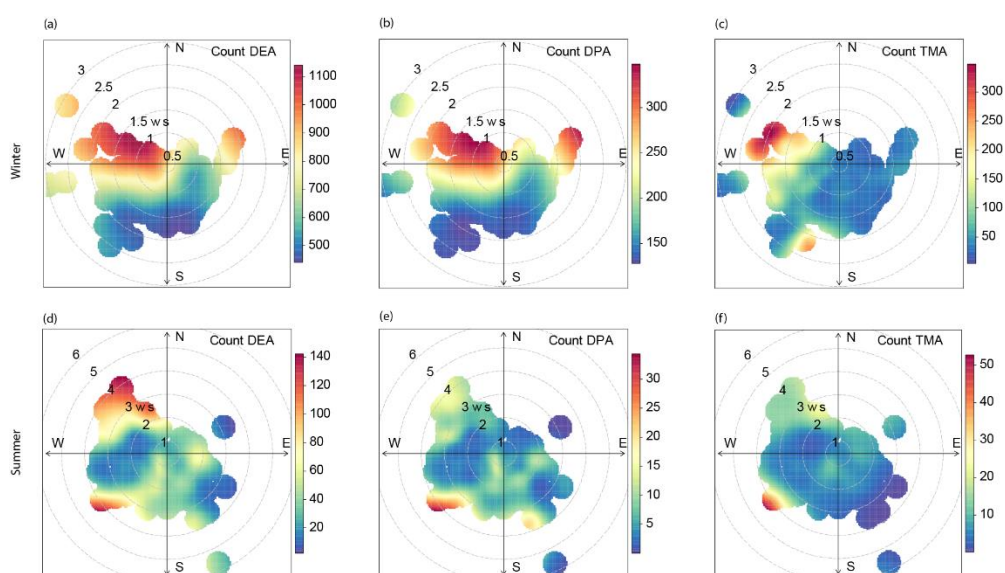
244

245 Figure 3. Diurnal profiles of amine-containing particles during both winter (left panel)
 246 and summer (right panel). The green left axis in each panel indicates the average
 247 number count of DEA-containing particles, while the right-axis represents the number
 248 count of both DPA- and TMA-containing particles.

249 During summer, the amine particles appeared in several episodes; each episode lasted
 250 for 1~3 days. In these episodes, DPA-containing particles had two rush-hour peaks
 251 (7:00 and 17:00), likely from traffic (Dall'Osto et al., 2016). Besides traffic, vegetation
 252 is also a source of DPA-containing particles (from the southwest, Figure 4e). The DPA-
 253 containing particles peaked 0.84 μm , suggesting that they were not freshly-emitted and
 254 had undergone substantial aging processes. Moreover, as shown in Figure S2, the mass
 255 spectra of the amines were present with aromatic hydrocarbon fragments, such as C_4H_3^+
 256 (m/z 51), C_5H_3^+ (m/z 63), C_6H_5^+ (m/z 77), and C_9H_8^+ (m/z 116), as well as with alkanes
 257 fragments such as C_4H_7^+ (m/z 55), C_4H_9^+ (m/z 57), and C_5H_9^+ (m/z 69). The chemical
 258 composition of DPA-containing particles contained markers associated with traffic
 259 emissions. Similar amine-containing particle type has been reported in the literature
 260 (Dall'Osto et al., 2016).

261 In summer, DEA-containing particles had a diurnal pattern of three peaks appearing at
 262 3:00, 9:00 and 17:00. TMA-containing particles had an early morning (4:00) and a noon
 263 peak (12:00). The morning peaks of DEA- and TMA-containing particles could be due

264 to the local traffic emissions; specifically, heavy-duty vehicles were only allowed to
265 enter the urban area between 00:00 and 6:00 (Chen et al., 2017). The polar plots showed
266 that DEA-containing particles were from the northwest and southwest, passing through
267 the forest park and traffic hub, respectively. This scenario seemed to be inconsistent
268 with the wintertime results because of the limited traffic contributions to particle levels
269 in winter. In addition, due to the competition between vegetation and traffic in summer,
270 the number count and peak area of all three amine-containing particles were poorly
271 correlated with each other.



272

273 Figure 4. Polar plots of amine-containing particles during winter- and summertime. The
274 axes in each figure indicate hourly count of each particle type and the colors within the
275 circles represent wind speed (ws)

276 3.3 Effect of RH on the enrichment of DEA-containing particles

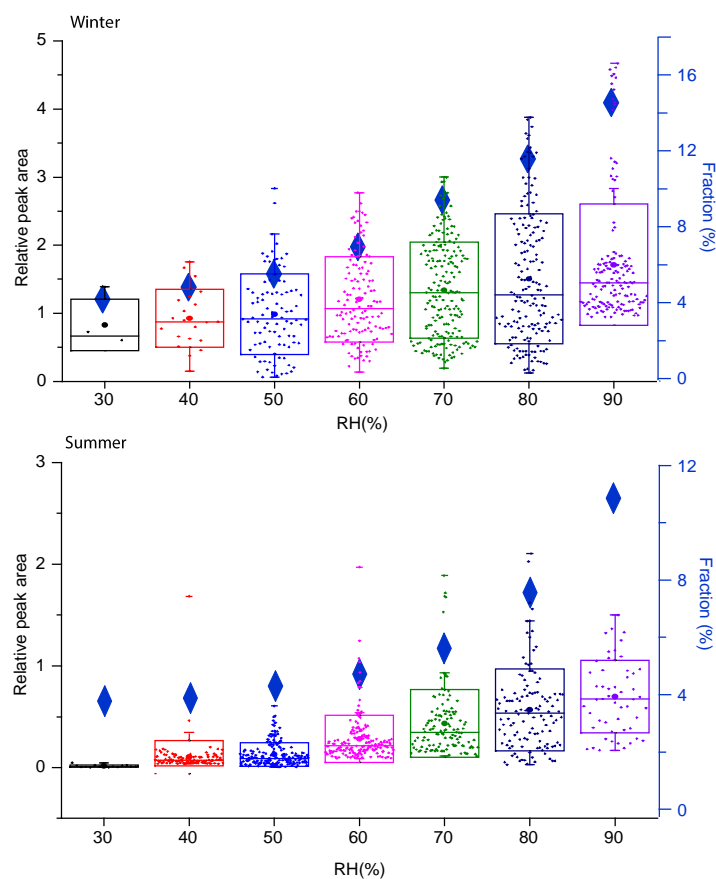
277 DEA-containing particles were predominant in both winter and summer, providing a
278 unique opportunity for investigating DEA processing. Indeed, the effect of RH on
279 aerosol chemical processing should be treated cautiously and the influences of wind

280 speed, wind direction, temperature, and planetary boundary layer reduction should be
281 removed. As described above, the average wind speed in both winter and summer was
282 1.2 ms^{-1} and 1.5 ms^{-1} , respectively. In these stagnant air conditions, the sampled
283 particles were generally local. Temperature could influence the gas-particle phase
284 partitioning. Assuming the Henry's Law constants (K_H) and the enthalpy change
285 $\Delta_r H_o(K_H)$ of DEA are constant, a variation of 10°C in both summer and winter has
286 negligible influence on the partitioning of amines from the gas phase to the particle
287 phase, according to the Clapeyron equation (Ge et al., 2011b). In addition, the shift in
288 planetary boundary layer (PBL) height could affect the number count and concentration
289 of PM. Using the temporal trends of RPA, the influence of PBL height can be removed
290 because it only shows the relative changes between different species which are all
291 simultaneously influenced by the shift in the PBL height.

292 Box plots of DEA relative peak area under different RH are shown in Figure 5. In winter,
293 the median RPA of amine-containing particles increased by two times when RH
294 increased from 35% to 95%. Meanwhile, the fraction of DEA-containing particles
295 increased from 4.0% to 16.6%. In summer, the average RPA of DEA increased by three
296 times (from 0.25 to 0.75) and the fraction of DEA-containing particles ramped from
297 3.8% to 12.1% when RH increased from 60% to 90%. These results suggest that RH is
298 important to the enrichment of DEA in the particle phase. When DEA reacts with HCl,
299 H_2SO_4 , and HNO_3 , it tends to form aminium salts, which are soluble in aerosol water.
300 Along with the influence of aerosol water content, Ge et al. (2011a) also proposed that
301 strong aerosol acidity could also enhance the partitioning of DEA in the aqueous phase.
302 As particles are dried in the SPAMS, the amount of aerosol water content and pH were
303 unavailable. The values of the anion/ cation ratio ((sulfate +nitrate)/ammonium, Yao et
304 al. (2011) were in a range of 20-150 suggesting that the particles might have been acidic

305 which favors the dissolution of DEA. Overall, these results implied that high RH
306 conditions in Chongqing was favorable for particle uptake of DEA, and the resulting
307 formation of aminium salts stabilized pre-existing particles; thus, increased their
308 number concentrations.

309 Rehbein et al. (2011) and Zhang et al. (2012) observed direct links between fog
310 processing and enhancement of TMA-containing particles. High RH conditions were
311 favorable for TMA entering the particle phase via gas -particle partitioning (Rehbein et
312 al., 2011; Zhang et al., 2012). Ge et al. (2011b) argued that TMA in the aerosol phase
313 was in the form of free base, e.g., amine, not aminium salt; TMA could be dissolved in
314 the aerosol water; the formation of TMA-HSO₄ salt was possible, but the formation of
315 TMA-NO₃ and TMA-Cl was impossible due to the competition with ammonia. Thus,
316 TMA could enter the aerosol phase by gas-aqueous partitioning, or in the form of TMA-
317 HSO₄ salt. The mechanism of DEA entering the aerosol phase might be different from
318 TMA. DEA salts were favorable for forming in aerosol phase (Ge et al., 2011b).
319 Besides, Pankow (2015) proposed that the absorptive uptake of atmospheric amines
320 could also be possible on organic aerosols. In the context of single particle mixing state,
321 the amine-containing particles were internally mixed with hygroscopic species, e.g.,
322 sulfate, nitrate, POA species (C_xH_y⁺, see section 3.4), and SOA species (oxalate,
323 C₂H₃O⁺). Therefore, the mixing state of amine-containing particles was also favorable
324 for the uptake of amines via different pathways: the aqueous dissolution of aminium
325 salts and the absorptive uptake on OA.



326

327 Figure 5. Box plots of the hourly relative peak area of DEA under different RH
 328 conditions in winter (top panel) and summer (bottom panel). The boxes indicate the
 329 25th and 75th percentiles; the dots indicate mean value with each data point representing
 330 a datum of RPA in an hour size bin. Right axis in each panel and the blue diamonds
 331 show the average number fraction of amine-containing particles among the whole
 332 SPAMS dataset.

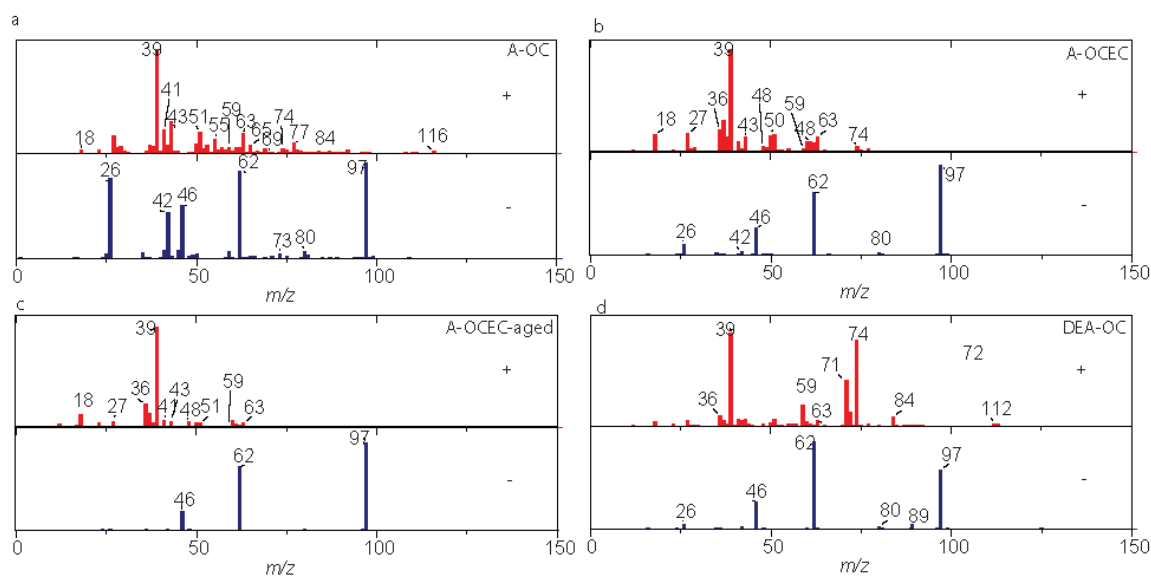
333 3.4 Particle types of amine-containing particles

334 As shown in Figure 6, four amine-containing particle types were resolved, including
 335 amine-OC (A-OC, 41%), A-ECOC (39%), DEA-OC (11%), and A-ECOC-aged (9%).
 336 All of these particle types had strong signals of amines, and the amines were internally
 337 mixed with sulfate, nitrate, elemental carbon, and organics.

338 In the A-OC particles, amines were mixed with aromatic hydrocarbon fragments, such
339 as $C_4H_3^+$ (m/z 51), $C_5H_3^+$ (m/z 63), $C_6H_5^+$ (m/z 77), and $C_9H_8^+$ (m/z 116), as well as with
340 alkanes fragments such as $C_4H_7^+$ (m/z 55), $C_4H_9^+$ (m/z 57), and $C_5H_9^+$ (m/z 69). In the
341 negative mass spectrum of A-OC, strong signals from CN^- (m/z -26) and CNO^- (m/z
342 -42) were typically primary species, along with levoglucosan (Silva et al., 1999). The
343 amine fragments, such as TMA (m/z 59), DEA (m/z 74), and DPA (m/z 86), were very
344 abundant in this particle type (76%, 95%, and 88%, respectively). The parent particles
345 of A-OC were a kind of OC particles from biomass burning; then they mixed with
346 amines via uptake. Amines could enter the A-OC particle type via dissolution in the
347 aerosol water content or uptake due to absorptive uptake on the organic aerosol
348 (Pankow, 2015).

349 In A-ECOC mass spectra, strong signals of amines (m/z 59 and 74), along with the
350 major aromatic hydrocarbon fragments and EC components (i.e., m/z 36, 48, 60) were
351 detected. In the negative mass spectra, nitrate and sulfate were also dominant. The A-
352 ECOC-aged particle type had a similar chemical composition to A-ECOC ($R^2 = 0.53$)
353 but with weaker relative intensities of $C_xH_y^+$ and amine ions, suggesting it could be
354 more secondary.

355 In the positive mass spectra of DEA-OC, DEA fragment (m/z 74) was dominant and
356 present with organic fragments described above. The secondary organic marker ions,
357 such as m/z 43 ($[C_2H_3O]^+$) and -89 (oxalic acid), were found in the mass spectra.
358 Besides, DEA-OC was not sensitive to wind speed ($R^2 = 0.18$), implying they were local.



359

360 Figure 6. Average mass spectra of major particle types clustered from amine-containing
 361 particles.

362 The summertime amine-containing particles were similar to the particle types during
 363 winter (all $R^2 > 0.7$), except a Ca-rich particle type was also resolved (Figure S5). A-
 364 Ca-OC particle type was mainly composed of calcium (Ca^+ and CaO^+), sodium (m/z
 365 23), potassium (m/z 39), TMA (m/z 59), sulfate, nitrate, and phosphate. An ion signal
 366 of zinc (m/z 64) was observed in the positive mass spectrum. Zn is a marker for tire
 367 wear on roads (Grigoratos and Martini, 2015; Thorpe and Harrison, 2008). The A-Ca-
 368 OC particle type was possibly from traffic activities (Chen et al., 2017).

369 The amine-containing particle types reported in this study were different from those in
 370 literature. Cheng et al. (2018) reported that m/z 74 amine-containing particles were most
 371 abundant in the Pearl River Delta, China, but the chemical composition and mixing
 372 state of amine particles were different from this study. For example, the mixing ratio of
 373 DPA was much stronger (~ 0.2) in Guangdong than in Chongqing (< 0.1). In most related
 374 studies, TMA-containing particles were dominant, while the present study showed

375 DEA-containing particles were dominant (Rehbein et al., 2011; Zhang et al., 2012;
376 Healy et al., 2015; Dall'Osto et al., 2016).

377 **4. Conclusions**

378 Amine-containing particles were collected and analyzed during winter and summer in
379 the urban area of Chongqing. Generally, amine-containing particles were more
380 abundant in winter than in summer. DEA-containing particles (m/z 74) were the most
381 important particle type during both summer and winter. Amines were internally mixed
382 with EC components, organics, sulfate, and nitrate, suggesting particle aging was
383 significant in both seasons. Amine-containing particles had monomodal size
384 distributions in the droplet mode, and the distributions peaked at a larger D_{va} in summer
385 than winter. DEA- and DPA-containing particles showed strong homogeneity, and
386 good correlations between the hourly number count and peak area were observed during
387 winter. The amine-containing particles were mostly from vegetation located southwest
388 of the sampling area, and traffic sources in the northwest. An enrichment of DEA-
389 containing particles under high RH conditions was revealed. Reduction of
390 anthropogenic amines, such as DEA and TMA, would improve the air quality in this
391 region, which can be achieved by decreasing the emissions of on-road fuel-powered
392 automobiles.

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398 carried them out; HR, CY, ZL, CJ, and GD analyzed the experiment data; CY prepared
399 the manuscript with contributions from all co-authors.

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