1	Water adsorption and hygroscopic growth of six anemophilous pollen species: the
2	effect of temperature
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### 23 Abstract

Hygroscopicity largely affects environmental and climatic impacts of pollen grains, one 24 important type of primary biological aerosol particles in the troposphere. However, our knowledge 25 in pollen hygroscopicity is rather limited, and especially the effect of temperature has rarely been 26 explored before. In this work three different techniques, including a vapor sorption analyzer, 27 diffusion reflectance infrared Fourier transform spectroscopy (DRIFTS) and transmission Fourier 28 transform infrared spectroscopy (transmission FTIR) were employed to characterize six 29 anemophilous pollen species and to investigate their hygroscopic properties as a function of 30 31 relative humidity (RH, up to 95%) and temperature (5 or 15, 25 and 37 °C). Substantial mass increase due to water uptake was observed for all the six pollen species, and at 25 °C the relative 32 mass increase at 90% RH, when compared to that at <1% RH, ranged from  $\sim30$  to  $\sim50\%$ , varying 33 with pollen species. It was found that the modified  $\kappa$ -Köhler equation can well approximate mass 34 hygroscopic growth of all the six pollen species, and the single hygroscopicity parameter ( $\kappa$ ) was 35 determined to be in the range of 0.034±0.001 to 0.061±0.007 at 25 °C. In-situ DRIFTS 36 measurements suggested that water adsorption by pollen species was mainly contributed by OH 37 groups of organic compounds they contained, and good correlations were indeed found between 38 hygroscopicity of pollen species and the amount of OH groups, as determined using transmission 39 FTIR. Increase in temperature would in general lead to decrease in hygroscopicity, except for 40 pecan pollen. For example,  $\kappa$  values decreased from 0.073±0.006 at 5 °C to 0.061±0.007 at 25 °C 41 and to  $0.057\pm0.004$  at 37 °C for populus tremuloides pollen, and decreased from  $0.060\pm0.001$  at 42 15 °C to 0.054±0.001 at 25 °C to 0.050±0.002 at 37 °C for paper mulberry pollen. 43

### 45 **1 Introduction**

Primary biological aerosol particles (PBAPs), an important type of aerosol particles in the 46 47 troposphere, are directly emitted from the biosphere and include pollen, fungal spores, bacteria, viruses, algae, and so on (Després et al., 2012; Fröhlich-Nowoisky et al., 2016). Emission and 48 abundance of PBAPs are quite uncertain, and annual emission fluxes are estimated to be in the 49 range of <10 to ~1000 Tg for total PBAPs and 47-84 Tg for pollen (Després et al., 2012). Pollen, 50 and PBAPs in general, are of great concerns due to their various impacts on the Earth system (Sun 51 and Ariya, 2006; Ariya et al., 2009; Georgakopoulos et al., 2009; Morris et al., 2011; Morris et al., 52 2014; Fröhlich-Nowoisky et al., 2016). For example, they can be allergenic, infectious or even 53 toxic, affecting the health of human and other species in the ecological systems over different 54 scales (Douwes et al., 2003; Reinmuth-Selzle et al., 2017; Shiraiwa et al., 2017). The geographical 55 dispersion of anemophilous plants largely relies on pollen dispersal, which in turn depends on the 56 emission, transport and deposition of pollen grains; therefore, pollen plays a key role in the 57 evolution of many ecosystems (Womack et al., 2010; Fröhlich-Nowoisky et al., 2016). In addition, 58 PBAPs can serve as giant cloud condensation nuclei (CCN) and ice nucleating particles (INPs), 59 significantly impacting the formation and properties of clouds and thus radiative balance and 60 precipitation (Möhler et al., 2007; Ariya et al., 2009; Pratt et al., 2009; Pope, 2010; Pummer et al., 61 2012; Gute and Abbatt, 2018). It has also been proposed that PBAPs may have significant impacts 62 on chemical composition of aerosol particles via heterogeneous and multiphase chemistry 63 (Deguillaume et al., 2008; Estillore et al., 2016; Reinmuth-Selzle et al., 2017; Shiraiwa et al., 2017). 64 Hygroscopicity is one of the most important physicochemical properties of pollen (as well 65 as aerosol particles in general). Hygroscopicity largely impacts the transport and deposition of 66 pollen grains (Sofiev et al., 2006), therefore affecting their lifetimes, abundance and 67

spatiotemporal distribution. In addition, hygroscopicity is closely linked to the ability of aerosol 68 particles to serve as CCN and INPs (Petters and Kreidenweis, 2007; Kreidenweis and Asa-Awuku, 69 2014; Laaksonen et al., 2016; Tang et al., 2016). Several previous studies have measured the 70 hygroscopicity and CCN activities of pollen (Diehl et al., 2001; Pope, 2010; Griffiths et al., 2012; 71 Lin et al., 2015; Steiner et al., 2015; Prisle et al., 2018) and other PBAPs such as bacteria (Pasanen 72 73 et al., 1991; Reponen et al., 1996; Franc and DeMott, 1998; Ko et al., 2000; Lee et al., 2002; Bauer et al., 2003). For example, water uptake of eleven pollen species was studied using an analytical 74 balance (Diehl et al., 2001), and the mass of pollen was found to be increased by 3-16% at 73% 75 76 RH and by ~100-300% at 95% RH, compared to that at 0% RH. An electrodynamic balance was employed to investigate hygroscopic growth of eight types of pollen (Pope, 2010; Griffiths et al., 77 2012), and it was found that their hygroscopic growth can be approximated by the modified  $\kappa$ -78 Köhler equation, with single hygroscopicity parameters being around 0.1 (depending on the 79 assumed pollen density, as discussed in Section 3.2). 80

81 Previous measurements were mostly carried out at or close to room temperature, and the effects of temperature on hygroscopic properties of pollen and other types of PBAPs are yet to be 82 elucidated. To our knowledge, only one previous study (Bunderson and Levetin, 2015) explored 83 84 the effect of temperature (4, 15 and 20 °C) on water uptake by Juniperus ashei, Juniperus monosperma and Juniperus pinchotii pollen. It is important to account for the temperature effects, 85 because ambient temperatures range from below -70 to >30 °C in the troposphere. In particular, 86 87 the altitude of 0.5-2.0 km to which pollen can be easily transported (Noh et al., 2013) may have temperatures close to or lower than the chilling temperatures for vegetative species (up to 16.5 °C) 88 89 (Melke, 2015). Moreover, the temperature in the respiratory tract can reach up to of 37 °C (the 90 physiological temperature). In the work presented here, a vapor sorption analyzer (VSA) was

employed to investigate hygroscopic growth of pollen grains at different temperature (5 or 15, 25, and 37 °C), a range covering the chilling temperature to the physiological temperature. Water uptake by pollen was also examined using diffusion reflectance infrared Fourier transform spectroscopy at room temperature to complement the VSA results. Furthermore, transmission Fourier transformation infrared spectroscopy was used to characterize functional groups of dry pollen grains, in an attempt to seek potential links between chemical composition of pollen grains and their hygroscopic properties.

### 98 **2 Experimental sections**

99 Six pollen species, all from anemophilous plants, were investigated in this work, including 100 populus tremuloides and populus deltoides (provided by Sigma Aldrich) as well as ragweed, corn, 101 pecan and paper mulberry (provided by Polysciences, Inc.). The six pollen species were chosen in 102 our work primarily because they were commercially available. Furthermore, these plants are also 103 widely distributed in the globe. For example, corn is the most produced grain in the world 104 (International-Grains-Council, 2019), and up to 50% of pollen-related allergic rhinitis cases in 105 North America are caused by ragweed pollen (Taramarcaz et al., 2005).

### 106 **2.1 Fourier transformation infrared spectroscopy**

The adsorption of water by pollen species were studied using in-situ diffusion reflectance infrared Fourier transform spectroscopy (DRIFTS) at room temperature (~25 °C). This technique was described in details in our previous work (Ma et al., 2010), and similar setups have also been used by other groups to investigate adsorption of water by mineral dust (Joshi et al., 2017; Ibrahim et al., 2018). Infrared spectra were recorded using a Nicolet 6700 Fourier transformation infrared spectrometer (FTIR, Thermo Nicolet Instrument Corporation), equipped with an in-situ diffuse reflection chamber and a high-sensitivity mercury cadmium telluride (MCT) detector cooled by

liquid nitrogen. A pollen sample (about 10 mg for each sample) under investigation was placed 114 into a ceramic crucible which was located in the in-situ chamber. A dry air flow and a humidified 115 air flow were first mixed and then delivered into the chamber, and the total flow rate was set to 116 200 mL/min (standard condition). Relative humidity (RH) in the chamber could be adjusted by 117 varying the flow rate ratio of the dry flow to the humidified flow, and was monitored online using 118 119 a moisture meter (CENTER 314) with an absolute uncertainty of  $\pm 2\%$ . Prior to each experiment, the sample was flushed with dry air for 3 h at 25 °C, and the reference spectrum was recorded after 120 the pretreatment. Infrared spectra were collected and analyzed using OMNIC 6.0 software (Nicolet 121 Corp.). All the spectra reported here were recorded with a wavenumber resolution of 4 cm<sup>-1</sup>, and 122 100 scans were averaged to produce a spectrum. Water adsorption was equilibrated for at least 30 123 min at each RH to ensure that the equilibrium between water vapor and adsorbed water was 124 reached. 125

Pollen samples used in this work were also characterized using transmission FTIR equipped with a deuterated triglycine sulfate detector (DTGS) detector. Pollen grains and KBr were mixed with a mass ratio of approximately 1:100 and ground in an agate mortar, and the mixture was then pressed into a clear disc. Transmission FTIR was employed to examine these discs, and a pure KBr disc was used as the reference. All the spectra, each of which was the average of 100 scans, were also recorded with a wavenumber resolution of 4 cm<sup>-1</sup>.

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#### 2.2 Vapor sorption analyzer

Hygroscopic growth of pollen grains was further investigated using a vapor sorption analyzer (Q5000 SA, TA Instruments, New Castle, DE, USA) described in our previous work (Gu et al., 2017; Guo et al., 2018; Jia et al., 2018). In brief, this instrument measured the sample mass as a function of RH under isothermal conditions. The instrument can be operated in the temperature

137	range of 5-85 °C with a temperature accuracy of $\pm 0.1$ °C and in the RH range of 0-98 % with an
138	absolute accuracy of $\pm 1\%$ . RH in the humidity chamber was regulated by using two mass flow
139	controllers to control the dry and humidified nitrogen flows very precisely. The accuracy in RH
140	control was routinely checked by measuring the DRH values for a series of standard compounds,
141	e.g., NaCl, (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , KCl, and etc., and the difference between the measured and theoretical
142	DRH was always $<1\%$ . The mass measurement had a range of 0-100 mg and a sensitivity of $\pm 0.01$
143	$\mu$ g. The initial mass of each sample used in this work was in the range of 0.5-1 mg. For each of
144	the first three types of pollen species (populus tremuloides, populus deltoides and ragweed pollen),
145	three samples in total were investigated, and each sample was studied under isothermal conditions
146	at 5, 25 and 37 °C. For each of the other three types of pollen species (corn, pecan and paper
147	mulberry pollen), experiments were carried out at 15 °C instead of 5 °C, because the instrument
148	could only be cooled down to 15 °C due to a technical problem after we finished experiments for
149	the first three pollen species.

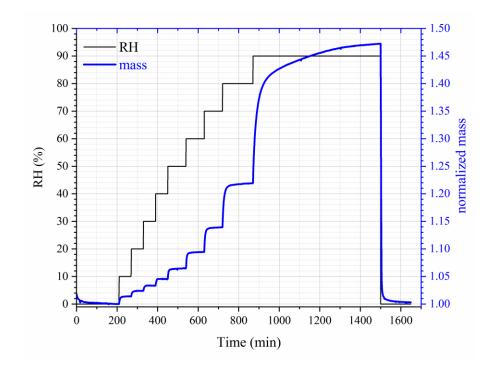


Figure 1. Change of RH (black curve, left *y*-axis) and normalized sample mass (blue curve, right *y*-axis) with time for a typical experiment in which hygroscopic growth of pollen grains was measured. In this figure a dataset for paper mulberry pollen at 25 °C is plotted as an example.

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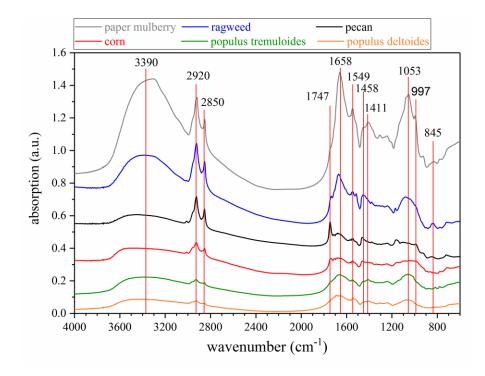
For the first sample, at each temperature the sample was first dried at 0% RH (the actual 155 RH was measured to be <1%); after that, RH was increased stepwise to 95% with an increment of 156 5% per step and then switched back to <1% to dry the sample again. At each RH, the sample was 157 equilibrated with the environment (i.e. until the sample mass became stable) before RH was 158 159 changed to the next value, and the sample mass was considered to be stabilized when the mass change was <0.05% within 30 min. Such a measurement at one temperature could take several 160 days. In order to reduce experimental time, the second and third samples were investigated in a 161 similar way as the first sample, except that RH was increased stepwise to 90% with an increment 162 of 10% per step. A typical experimental dataset is displayed in Figure 1 as an example to illustrate 163 the change of RH and normalized sample mass with experimental time. 164

## 165 **3 Results and discussion**

### 166 **3.1 FTIR characterization of pollen samples**

### 167 **3.1.1 Infrared spectra of dry pollen samples**

Figure 2 shows the transmission FTIR spectra of the six pollen species investigated in our work, and peak assignments can be found in Table 1. A broad band in the range of 3600-3000 cm<sup>-1</sup>, attributed to O-H stretching vibration (Stuart, 2004; Pummer et al., 2013), and two sharp peaks at 2920 and 2850 cm<sup>-1</sup>, attributed to C-H stretching (Eliason et al., 2003; Stuart, 2004; Pummer et al., 2013), were observed for all the pollen species. The two peaks at 1747 and 1658 cm<sup>-1</sup> were assigned to alkyl ester carbonyls (Pappas et al., 2003; Najera et al., 2009; Pummer et al., 2013), and the two peaks at 1549 and 1458 cm<sup>-1</sup> (1411 cm<sup>-1</sup> for paper mulberry pollen) were assigned to C=C stretching and H-C-H deformation (Stuart, 2004; Pummer et al., 2013). In addition, the three peaks at 1053, 997 and 845 cm<sup>-1</sup> were assigned to C-O stretching, C-C stretching, and C-H out-ofplane bending, respectively (Stuart, 2004; Najera et al., 2009; Pummer et al., 2013).



178

179 **Figure 2.** Transmission FTIR spectra of six pollen species investigated in this work.

180

OH groups and C-H groups in organic compounds are generally considered to be 181 hydrophilic and hydrophobic, and one may expect that the amount of OH groups (relative to that 182 of C-H groups) that organic samples contain may affect their hygroscopicity. For example, it was 183 found in many previous studies (Eliason et al., 2003; Asad et al., 2004; Hung et al., 2005; Najera 184 et al., 2009) that heterogeneous reactions of organic materials with O<sub>3</sub> and OH radicals would 185 increase the IR absorption intensity for the O-H stretching mode and decrease the IR absorption 186 intensity for the C-H stretching mode, meanwhile leading to the enhancement in their 187 hygroscopicity. Therefore, in this work we use the intensity ratio of the O-H stretching vibration 188

band (3000-3600 cm<sup>-1</sup>) to the C-H stretching mode (2920 cm<sup>-1</sup>) to qualitatively represent the 189 amount of OH groups pollen samples contain. As shown in Figure 2, the six pollen species 190 examined in our work can be roughly classified into two catalogues: 1) for populus deltoides, 191 populus tremuloides and paper mulberry pollen, the O-H stretching vibration band is more 192 intensive than the C-H stretching mode, indicating that they contain relatively high levels of OH 193 194 groups; 2) for ragweed, pecan and corn pollen, the O-H stretching vibration band is less intensive than the C-H stretching mode, indicating that they contain relatively low levels of OH groups. The 195 relation between the amount of OH groups that pollen species contain and their hygroscopicity 196 197 will be further discussed in Section 3.3.

198

**Table 1.** Vibrational mode assignment for six pollen species investigated in this work.

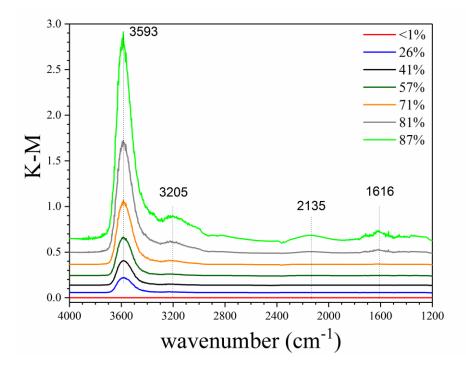
wavenumber (cm <sup>-1</sup> )	vibrational mode
3600-3000	O-H stretching
2920 and 2820	C-H stretching
1747 and 1658	alkyl ester carbonyls
1549	C=C stretching
1458 and 1411	H-C-H deformation
1053	C-O stretching
997	C-C stretching
845	C-H out-of-plane bending
1549 1458 and 1411 1053 997	alkyl ester carbonyls C=C stretching H-C-H deformation C-O stretching C-C stretching

200

### **3.1.2 Infrared spectra of pollen samples at different RH**

In-situ DRIFTS was employed to explore adsorption of water by pollen grains. Typical spectra of populous deltoides pollen as a function of RH up to 87%, relative to that at <1% RH, are displayed in Figure 3. DRIFTS spectra of other pollen samples at different RH can be found in Figures S1-S5 in the supplement, and are very similar to those for populous deltoides pollen. As

evident from Figure 3, several IR peaks (e.g., 3593, 3205, 2135, and 1616 cm<sup>-1</sup>) appeared in the 206 spectra at elevated RH, when compared with that at <1% RH, and their intensities increased with 207 increasing RH. The peaks at 3205, 2135 and 1616 cm<sup>-1</sup> can be assigned to the stretching, 208 association and bending modes of adsorbed water (Goodman et al., 2001; Schuttlefield et al., 209 2007a; Ma et al., 2010; Hatch et al., 2011; Song and Boily, 2013; Yeşilbaş and Boily, 2016; Joshi 210 et al., 2017; Ibrahim et al., 2018). 211



212

Figure 3. In-situ DRIFTS spectra of populous deltoides pollen as a function of RH (<1, 26, 41, 57, 213 71, 81 and 87%) at 25 °C. 214

215

216

The peak at  $\sim$ 3600 cm<sup>-1</sup> was the most intensive one observed in the spectra, as shown in Figure 3. For comparison, the IR peaks assigned to the stretching mode of adsorbed water on 217 mineral dust and NaCl appeared at lower wavenumbers, typically at around or lower than 3400 218 cm<sup>-1</sup> (Schuttlefield et al., 2007a; Ma et al., 2010; Tang et al., 2016; Ibrahim et al., 2018). As a 219 result, the peak at  $\sim$ 3600 cm<sup>-1</sup> may be assigned to the asymmetric stretching mode of water which 220

interacted with OH groups in pollen samples (Iwamoto et al., 2003). These results imply that water adsorption by pollen samples could be mainly contributed by OH groups of organic compounds they contained; in addition, other factors, such as porosity and internal structure, may also be important for hygroscopic properties of pollen grains. The intensities of IR peaks at ~3600 cm<sup>-1</sup> were used to represent the amount of water adsorbed by pollen samples. Table 2 summarizes integrated areas of IR peaks at 3600 cm<sup>-1</sup> as a function of RH for the six pollen species examined in our work, suggesting that the amount of adsorbed water by pollen samples increased with RH.

Table 2. Integrated areas of IR peaks (at  $\sim 3600 \text{ cm}^{-1}$ ) of adsorbed water as a function of RH for the six pollen species investigated in this work. Wavenumber ranges used for integration are 3750-3300 cm<sup>-1</sup> for populus deltoides pollen, 3750-3350 cm<sup>-1</sup> for populus tremuloides pollen, 3750-3400 cm<sup>-1</sup> for ragweed pollen, 3750-3500 cm<sup>-1</sup> for corn pollen, 3750-3450 cm<sup>-1</sup> for pecan pollen, and 3750-3300 cm<sup>-1</sup> for paper mulberry pollen.

RH (%)	peak area	RH (%)	peak area	RH (%)	peak area
populus deltoides		populus tremuloides		ragweed	
0	0	0	0	0	0
26	22.7	24	5.5	26	10.1
41	36.9	41	16.4	42	18.9
57	57.4	56	35.4	50	24.5
71	93.6	70	66.5	56	30.2
79	137.6	78	91.2	69	49.7
81	164.7	87	156.9	88	104.6
87	293.1				
corn		pe	can	paper n	nulberry
0	0	0	0	0	0
26	10.0	26	8.6	26	10.2
42	21.5	43	16.9	43	17.7
58	41.9	58	29.5	51	23.1

73	87.5	73	60.0	59	29.8
89	222.2	89	338.9	71	46.7
				86	105.1

234

#### **3.2 Mass hygroscopic growth**

### 236 3.2.1 Hygroscopicity parameterizations

The single hygroscopicity parameter,  $\kappa$ , is widely used to describe the hygroscopicity of aerosol particles under both subsaturation and supersaturation (Petters and Kreidenweis, 2007). When the Kelvin effect is negligible (this is valid for pollen grains which are typically >1 µm), the dependence of diameter-based growth factor (GF) on RH can be linked to  $\kappa$  via Eq. (1) (Petters and Kreidenweis, 2007; Tang et al., 2016):

242 
$$RH = \frac{GF^3 - 1}{GF^3 - 1 + \kappa} \quad (1)$$

If we further assume that the particle is spherical, Eq. (1) can be transformed to Eq. (2):

244 
$$\frac{1}{RH} = 1 + \frac{\kappa}{GF^3 - 1} = 1 + \frac{\kappa}{\frac{V}{V_0} - 1} = 1 + \kappa \frac{V_0}{V - V_0} = 1 + \kappa \frac{V_0}{V_W}$$
(2)

where V,  $V_0$  and  $V_w$  are the volumes of the particle at the given RH, the dry particle and water associated with the particle at the given RH. In order for Eq. (2) to be valid, it is also assumed that at a given RH, V is equal to the sum of  $V_0$  and  $V_w$ . Eq. (2) can be further transformed to Eqs. (3-4):

248 
$$\frac{1}{RH} = 1 + \kappa \frac{\rho_w}{\rho_p} \frac{m_0}{m_w} \quad (3)$$

249 
$$\frac{m_w}{m_0} = \kappa \cdot \frac{\rho_w}{\rho_p} / (\frac{1}{RH} - 1) \quad (4)$$

where  $\rho_{\rm w}$  and  $\rho_{\rm p}$  are the density of water and the dry particle, and  $m_0$  and  $m_{\rm w}$  are the mass of the dry particle and water associated with the particle at the given RH. Since the particle mass, *m*, is equal to the sum of  $m_0$  and  $m_{\rm w}$ , Eq. (5) can be derived from Eq. (4):

253 
$$\frac{m}{m_0} = 1 + \kappa \frac{\rho_W}{\rho_p} / (\frac{1}{RH} - 1) \quad (5)$$

254 Using an electrodynamic balance, Pope and co-workers (Pope, 2010; Griffiths et al., 2012) measured hygroscopic growth of eight types of pollen grains, and found that their mass change 255 with RH can be approximated by Eq. (5). It should be noted that the original equation derived by 256 257 Pope and co-workers (Pope, 2010; Griffiths et al., 2012) has a different format from but is essentially equivalent to Eq. (5). Eq. (5) relates mass growth experimentally measured in our work 258 to the single hygroscopicity parameter ( $\kappa$ ), which has been widely used in atmospheric science to 259 describe hygroscopic properties of aerosol particles under subsaturation as well as their CCN 260 activities under supersaturation; nevertheless, a few assumptions are needed to derive Eq. (5), as 261 discussed. 262

The Freundlich adsorption isotherm is another widely used equation to describe the change of sample mass with RH due to water uptake (Atkins, 1998; Skopp, 2009; Hatch et al., 2011; Tang et al., 2016):

$$\frac{m}{m_0} = 1 + A_f \cdot \sqrt[B_f]{RH} \quad (6)$$

where  $A_{\rm f}$  and  $B_{\rm f}$  are empirical Freundlich constants related to the adsorption capacity and strength. 267 One advantage of the Freundlich adsorption isotherm is that it provides a direct relationship 268 between RH and mass growth which was experimentally measured in our work, without any 269 additional assumptions. In addition, the BET (Brunauer-Emmett-Teller) adsorption isotherm is 270 also widely used to describe water adsorption by insoluble solid particles (Brunauer et al., 1938; 271 Goodman et al., 2001; Henson, 2007; Ma et al., 2010; Tang et al., 2016; Joshi et al., 2017). While 272 the BET adsorption isotherm typically works well for water adsorption of a few monolayers, the 273 mass of adsorbed water, as shown in Section 3.2.2, can reach up to 50% of the dry pollen mass at 274 high RH; therefore, in this work we did not attempt to use the BET adsorption isotherm to describe 275

water adsorption by pollen grains. Another reason that we did not attempt to use the BET adsorption isotherm is that the BET adsorption isotherm is mathematically more complex and requires the BET surface area to be known.

### **3.2.2 Mass hygroscopic growth at room temperature**

Figure 4 displays the sample mass (normalized to that at 0% RH) as a function of RH for pecan pollen at 25 °C. Significant increase in sample mass was observed at elevated RH due to uptake of water. Compared to that at 0% RH, the sample mass increased by  $(2.3\pm0.3)$ % at 30% RH,  $(6.4\pm0.2)$ % at 60% RH,  $(30.3\pm0.4)$ % at 90% RH, and up to ~72% at 95% RH. As shown by the data compiled in Tables S1-S3 in the supplement, substantial increases in sample mass were also observed for the other five types of pollen species at 25 °C (as well as 5 and 37 °C).

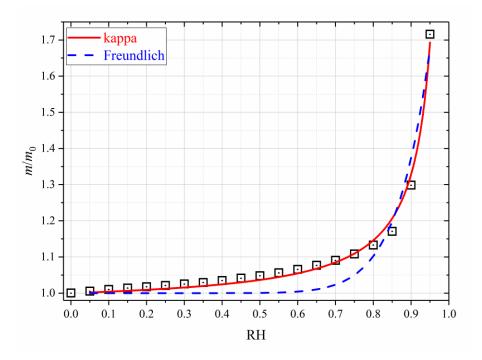


Figure 4. Measured change of sample mass (normalized to that at dry conditions, i.e.  $m/m_0$ ) of pecan pollen as a function of RH (0-0.95) at 25 °C. The experimental data are fitted with the modified  $\kappa$ -Köhler equation (solid red curve) and the Freundlich adsorption isotherm (dashed blue curve).

Hygroscopic properties exhibited considerable variations among different pollen species. 292 Figure 5a compares the measured ratios of sample mass at 90% RH to that at 0% RH,  $m(90\%)/m_0$ , 293 for the six pollen species investigated in this work. We specifically discuss mass changes of pollen 294 grains at 90% RH (relative to that at 0% RH) because aerosol hygroscopic growth at 90% RH was 295 widely reported by laboratory and field studies (Kreidenweis and Asa-Awuku, 2014). As shown 296 in Figure 5a,  $m(90\%)/m_0$  determined at 25 °C ranged from 1.293±0.028 (ragweed pollen) to 297 1.476±0.094 (populus deltoides pollen), i.e. the amount of water adsorbed/absorbed by the six 298 different pollen species at 90% RH varied between ~30% to ~50% of the dry mass. 299

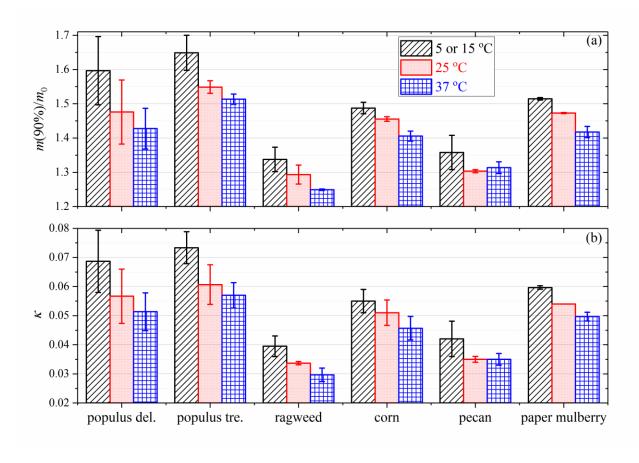


Figure 5. Measured ratios of sample mass at 90% RH to that at 0% RH (a) and derived  $\kappa$  values (b) for six pollen species at different temperatures. The lowest temperatures were 5 °C for populus

deltoides (populus del.), populus tremuloides (populus tre.) and ragweed pollen, and 15 °C for corn,
 pecan and paper mulberry pollen.

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As shown in Figure 4, the increase of pecan pollen mass with RH at 25 °C could be 306 satisfactorily described by the modified  $\kappa$ -Köhler equation for the entire RH range (up to 95%). 307 308 On the contrary, the Freundlich adsorption isotherm significantly underestimated the sample mass at low RH, although it represented the experimental data at high RH reasonably well. In addition, 309 we found that the modified  $\kappa$ -Köhler equation could also approximate the dependence of sample 310 311 mass on RH for all the six types of pollen species investigated in this work at different temperatures. If we use Eq. (5) to fit  $m/m_0$  against RH,  $\kappa \cdot \rho_w/\rho_p$  can be derived. The bulk densities of dry pollen 312 grains were found to vary with species but typically fall into the range of  $0.5-2 \text{ g cm}^{-3}$  (Harrington 313 and Metzger, 1963; Hirose and Osada, 2016), and for simplicity  $\rho_p$  was assumed to be 1 g cm<sup>-3</sup> in 314 this work (i.e.  $\rho_w/\rho_p$  is equal to 1). With the assumptions on dry particle density and also particle 315 sphericity,  $\kappa$  could then be derived from the measured RH-dependent sample mass at a given 316 temperature. 317

Table 3 summarizes the average  $\kappa$  values at different temperatures for the six pollen species 318 investigated in this work. At 25 °C, the  $\kappa$  values were found to increase from 0.034±0.001 for 319 ragweed pollen to 0.061±0.007 for populus tremuloides pollen, varied by almost a factor of 2. The 320  $\kappa$  values measured by Pope and co-workers (Pope, 2010; Griffiths et al., 2012) were approximately 321 322 in the range of 0.05-0.11 (assuming that  $\rho_w/\rho_p$  is equal to 1), in reasonably good agreement with these reported in our work. It should be noted that in order to convert the measured mass growth 323 324 to diameter growth and  $\kappa$  values, one key assumption is particle sphericity; nevertheless, pollen 325 grains are known to be non-spherical and porous, and therefore our derived  $\kappa$  values might be smaller than the actual values. For example, although the mass increase was substantial (around 30-50 % at 90% RH) for the six pollen species examined, their  $\kappa$  values at 25 °C were derived to be in the range of 0.034-0.061, significantly smaller than those (0.1-0.2) for typical secondary organic aerosols produced in smog chamber studies (Petters and Kreidenweis, 2007; Kreidenweis and Asa-Awuku, 2014).

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Table 3. Single hygroscopicity parameters ( $\kappa$ ) derived in this work for the six pollen species at different temperatures. All the errors given in this work are standard deviations.

pollen type	$T(^{\circ}\mathrm{C})$	sample 1	sample 2	sample 3	average
populus	5	0.071±0.001	0.078±0.001	0.057±0.002	0.069±0.011
	-				
deltoides	25	$0.054 \pm 0.001$	$0.067 \pm 0.002$	$0.049 \pm 0.002$	$0.057 \pm 0.009$
	37	$0.058 \pm 0.002$	$0.051 \pm 0.001$	$0.045 \pm 0.002$	0.051±0.007
populus	5	$0.068 \pm 0.001$	0.073±0.001	$0.079 \pm 0.001$	0.073±0.006
tremuloides	25	$0.053 \pm 0.002$	$0.063 \pm 0.002$	$0.066 \pm 0.002$	0.061±0.007
	37	$0.052 \pm 0.002$	$0.059 \pm 0.002$	$0.060 \pm 0.002$	0.057±0.004
ragweed	5	$0.042 \pm 0.001$	$0.037 \pm 0.002$		0.040±0.004
	25	$0.033 \pm 0.002$	$0.034 \pm 0.003$	$0.034 \pm 0.002$	0.034±0.001
	37	$0.027 \pm 0.001$	$0.031 \pm 0.002$	$0.031 \pm 0.002$	0.030±0.002
corn	15	$0.051 \pm 0.001$	$0.059 \pm 0.002$	$0.055 \pm 0.002$	0.055±0.004
	25	$0.046 \pm 0.002$	$0.053 \pm 0.002$	$0.054 \pm 0.002$	0.051±0.004
	37	$0.041 \pm 0.002$	$0.048 \pm 0.002$	$0.048 \pm 0.002$	0.046±0.004
pecan	15	$0.049 \pm 0.001$	$0.038 \pm 0.001$	0.039±0.001	0.042±0.006
	25	$0.036 \pm 0.001$	$0.034 \pm 0.001$	$0.035 \pm 0.001$	0.035±0.001
	37	0.033±0.001	$0.035 \pm 0.002$	$0.037 \pm 0.001$	0.035±0.002
paper	15	$0.059 \pm 0.002$	$0.060 \pm 0.002$	$0.060 \pm 0.002$	0.060±0.001
mulberry	25	$0.054 \pm 0.001$	$0.054 \pm 0.001$	$0.054 \pm 0.001$	0.054±0.001
	37	$0.048 \pm 0.002$	$0.050 \pm 0.002$	$0.051 \pm 0.002$	0.050±0.002

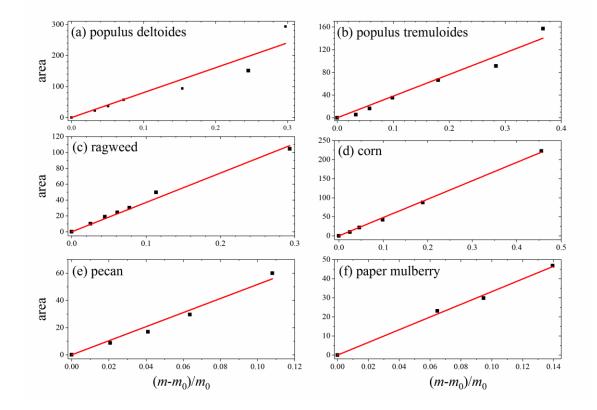
#### 335 **3.3 Discussion**

#### 336 **3.3.1 Reconciliation between IR and VSA results**

Our in-situ DRIFTS measurements, as discussed in Section 3.1.2, suggested that water 337 uptake by pollen samples was mainly contributed by OH groups of organic compounds they 338 339 contained; therefore, one may expect that pollen species which contain higher levels of OH groups 340 would exhibit higher hygroscopicity. Transmission FTIR characterization of pollen species (Section 3.1.1) showed that populus deltoides, populus tremuloides and paper mulberry pollen 341 contained relatively high levels of OH groups, and indeed their hygroscopicity ( $\kappa$ : 0.053-0.054 at 342 343 25 °C) was higher than the other three pollen species, as shown in Figure 5 and Table 3. For comparison, ragweed and pecan pollen contained relatively low levels of OH groups and 344 correspondingly exhibited lower hygroscopicity ( $\kappa$ : 0.033-0.036 at 25 °C). Corn pollen appeared 345 to be an exception: it contained relatively low levels of OH group but displayed medium 346 347 hygroscopicity ( $\kappa$ : ~0.046 at 25 °C). As a result, our results may imply that in addition to chemical composition, other physicochemical properties, such as porosity and internal structure of pollen 348 grains, could also play an important role in determining the hygroscopicity of pollen species. One 349 350 clue came from environmental scanning electron microscopy observations (Pope, 2010), revealing that pollen grains started to swell internally before significant water uptake on the surface took 351 place. 352

In our work two complementary techniques were employed to explore hygroscopic properties of pollen species. VSA measured the amount of water absorbed/adsorbed by pollen grains as a function of RH in a quantitative manner, whereas the intensities of IR peaks of adsorbed water at different RH, as characterized by DRIFTS, can be used semi-quantitatively to represent the amount of water associated with particles (Goodman et al., 2001; Schuttlefield et al., 2007b;

Ma et al., 2010; Yeşilbaş and Boily, 2016; Joshi et al., 2017; Ibrahim et al., 2018). We compare our VSA results (i.e. the relative mass change due to water uptake) to the DRIFTS results (i.e. integrated area of IR peaks at ~3600 cm<sup>-1</sup>). As shown in Figure 6, good correlations between VSA and DRIFTS results are found for all the six pollen species, suggesting that DRIFTS can be used to represent the amount of adsorbed water, at least in a semi-quantitative manner.



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Figure 6. Integrated areas of IR peaks at ~3600 cm<sup>-1</sup> versus relative mass increase due to water uptake,  $(m-m_0)/m_0$ , for six pollen species: (a) populus deltoides; (b) populus tremuloides; (c) ragweed; (d) corn; (e) pecan; (f) paper mulberry.

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### 368 **3.3.2 Effect of temperature**

Figure 5a shows the comparison of the measured ratios of sample mass at 90% RH to that at 0% RH,  $m(90\%)/m_0$ , at different temperatures for the six pollen species. It can be concluded from Figure 5a that except for pecan pollen for which a small increase in  $m(90\%)/m_0$  occurred when temperature increased from 25 to 37 °C, increase in temperature would lead to small but nevertheless significant decrease in  $m(90\%)/m_0$ . For example,  $m(90\%)/m_0$  decreased from  $1.597\pm0.100$  at 5 °C to  $1.476\pm0.094$  at 25 °C and to  $1.427\pm0.060$  at 37 °C for populus deltoides pollen, and from  $1.338\pm0.036$  at 5 °C to  $1.293\pm0.028$  at 25 °C and to  $1.249\pm0.002$  at 37 °C for ragweed pollen.

We further derived  $\kappa$  values at different temperatures for the six pollen species, and the results are plotted in Figure 5b and summarized in Table 3. Increase in temperature would lead to decrease in  $\kappa$  values, except for pecan pollen. For example,  $\kappa$  decreased from 0.073±0.006 at 5 °C to 0.057±0.004 at 37 °C for populus tremuloides pollen, and decreased from 0.060±0.001 at 15 °C to 0.050±0.002 at 37 °C for paper mulberry pollen.

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# **4** Conclusion and implications

Pollen grains are one of the most abundant types of primary biological aerosol particles in the troposphere and play important roles in many aspects of the Earth system. Hygroscopicity is among the most important physicochemical properties of pollen grains and largely affect their environmental, health and climatic impacts. However, our knowledge in their hygroscopicity is still quite limited, and especially the temperature effect has been rarely explored.

In this work we investigated hygroscopic properties of six types of pollen species as a function of RH (up to 95%) at 5 (or 15), 25 and 37 °C. Substantial increase in pollen mass was observed at elevated RH due to water uptake for all the six pollen species. Therefore, change in the mass of pollen grains and their aerodynamic properties at different RH should be taken into account to better understand their transport and deposition in the troposphere. It was found that the mass hygroscopic growth of pollen grains can be well approximated by the modified  $\kappa$ -Köhler

394 equation. The derived  $\kappa$  values at 25 °C ranged from 0.034±0.001 to 0.061±0.007, varying with pollen species. DRIFTS measurements indicated that water adsorption by pollen species were 395 mainly contributed by OH groups of organic compounds contained by pollen grains, and indeed 396 pollen species that contained lower levels of OH groups (relative to C-H groups, as determined 397 using transmission FTIR) showed lower hygroscopicity. One exception was corn pollen which 398 399 contained low levels of OH group but exhibited medium hygroscopicity, suggesting that in addition to chemical composition, other physicochemical properties, such as porosity and internal 400 structure, might also play an important role in determining the hygroscopicity of pollen grains. 401 402 Due to their moderate hygroscopicity as well as large sizes, pollen grains can thus act as efficient giant CCN which may have significant impacts on cloud and precipitation (Johnson, 1982; 403 Feingold et al., 1999; Yin et al., 2000; Posselt and Lohmann, 2008). It is worth noting that only 404 six different pollen species were examined in our work, and hygroscopic properties of other pollen 405 species commonly found in the troposphere should be further investigated. 406

The effect of temperature on the hygroscopicity of pollen grains was systematically 407 investigated in this work. Increase in temperature (from 5 or 15 °C to 25 and 37 °C), a range 408 covering chilling temperature to physiological temperature, led to small but detectable decrease in 409 410 pollen hygroscopicity. For example,  $\kappa$  values were found to decrease from 0.073±0.006 at 5 °C to 0.061±0.007 at 25 °C and to 0.057±0.004 at 37 °C for populus tremuloides pollen, and decrease 411 from 0.060±0.001 at 15 °C to 0.054±0.001 at 25 °C to 0.050±0.002 at 37 °C for paper mulberry 412 413 pollen. Our measurements at 37 °C (physiological temperature) provide very valuable parameters which can be used in numerical models to better understand the transport and deposition of pollen 414 415 particles in the respiratory system and thus their impacts on human health (Yeh et al., 1996; Broday 416 and Georgopoulos, 2001; Park and Wexler, 2008; Lambert et al., 2011; Longest and Holbrook,

2012; Tong et al., 2014). Nevertheless, it should be noted that due to the short residence time in
the respiratory system, pollen grains and other inhaled particles in general, may not reach
equilibrium with water vapor in the respiratory tract.

Due to technical challenges, the lowest temperature we could reach in this work was 5 °C, 420 in the range of normal chilling temperatures for vegetative species and also in the expected 421 422 temperature range at the altitudes of 0.5-2.0 km to which pollen grains can be easily transported. Temperatures in the upper troposphere can be as low as below -70 °C, and it is yet to be explored 423 whether further decrease in temperature to far below 0 °C will lead to large increase in pollen 424 425 hygroscopicity. As a result, experimental measurements of pollen hygroscopicity at lower temperatures are warranted and would significantly help better understand the transport of pollen 426 grains in the troposphere. Since water vapor has to be adsorbed or condensed on ice nucleating 427 particles before heterogeneous ice nucleation can take place (Laaksonen et al., 2016), knowledge 428 in hygroscopicity and water uptake at temperatures below 0 °C would provide fundamental insights 429 into atmospheric ice nucleation, in which pollen grains may play an important role (Pratt et al., 430 2009; Prenni et al., 2009; Hoose et al., 2010; Pöschl et al., 2010; Murray et al., 2012; Creamean et 431 al., 2013; Tang et al., 2018). 432

433 Author contribution

434 MT, QM and YJL designed the research; WG, CZ, SL and XY did the measurements; MT, 435 QM, YJL and RJH analyzed the results; MT, QM, YJL and RJH wrote the manuscript with 436 contribution from all the co-authors.

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