Comments by referees are in blue.

Our replies are in black.

Changes to the manuscript are highlighted in red both in here and in the revised manuscript.

Reply to Ref #1

Tang et al. investigated the hygroscopic growth of six pollen species and its temperature dependence. This study measured water uptake and growth factor by pollen grains using a vapor sorption analyzer and characterize pollen grains using FTIR. A hygroscopic parameter (k) was calculated from the measurements. The subject of this manuscript is within the scope of this journal. There are some minor issues that the authors may want to address before it can be accepted for publication.

Reply: We would like to thank Ref #1 for his/her insightful and detailed comments, which have largely helped us improve our manuscript. We have addressed all the comments adequately in the revised manuscript, as detailed below.

1. P5, L98-100, are these pollen species atmospheric relevant? Justification of using these pollen species needs further discussion.

Reply: These pollen species have been chosen in our work primarily because of their commercial availability; nevertheless, these plants are widely distributed in the globe and some of the pollen species, such as ragweed pollen, are well-known due to their impacts on human health. In the revised manuscript (page 5, line 101-105) we have added a few sentences to further justify why these pollen species were chosen in our work: "The six pollen species were chosen in our work primarily because they were commercially available. Furthermore, these plants are also widely distributed in the globe. For example, corn is the most produced grain in the world (International-Grains-Council, 2019), and up to 50% of pollen-related allergic rhinitis cases in North America are caused by ragweed pollen (Taramarcaz et al., 2005)."

2. P6, L114, what is the uncertainty of this moisture meter?

Reply: The sensor has an absolute uncertainty of ±2%. In the revised manuscript (page 6, line 118-119) the sentence has been changed to "…and was monitored online using a moisture meter (CENTER 314) with an absolute uncertainty of ±2%."

3. P6, L130-132, which kind of temperature and humidity sensors that can achieve such high accuracy (+-0.1 K and 1% RH) at this temperature and RH range?

Reply: The temperature was monitored using a thermocouple, which could achieve a temperature accuracy of ±0.1 K easily. The high accuracy of RH control was achieved by using mass flow controllers to precisely control the flow rates of the dry and humidified nitrogen flows used to regulate RH in the humidity chamber; the accuracy of RH control was routinely checked by measuring the DRH of standard compounds, and the difference in measured and theoretical DRH was always <1%. In the revised manuscript (page 7, line 138-142) we have added a few sentences to explain how high accuracy in RH control was achieved: "RH in the humidity chamber was regulated by using two mass flow controllers to control the dry and humidified nitrogen flows very precisely. The accuracy in RH control was routinely checked by measuring the DRH values for a series of standard compounds, e.g., NaCl, $(NH_4)_2SO_4$, KCl, and etc., and the difference between the measured and theoretical DRH was always <1%."

4. P7, L137-138, Although it may be fine, I do not think this is an excuse that left the other temperature out. One can simply conduct a few more experiments for the missing points.

Reply: We agree with the referee, and would like to carry out measurements at 5 ^oC for the other three pollen species. However, unfortunately due to a technical problem, the lowest temperature our instrument could reach was 15 ^oC since the technical problem occurred. In the revised manuscript (page 7, line 146-149) the following change has been implemented for further clarification: "For each of the other three types of pollen species (corn, pecan and paper mulberry pollen), experiments were carried out at 15 ^oC instead of 5 ^oC, because the instrument could only be cooled down to 15 ^oC due to a technical problem after we finished experiments for the first three pollen species."

5. P8, L157-165, it is suggested to list these peak assignments in a table.

Reply: As suggested, we have included a table in the revised manuscript (Table 1, page 10) to summarize these peak assignments, and also made corresponding changes to text in Section 3.1.1 (page 8, line 168-169).

6. P9, L169-173, please justify the use of such ratio as a qualitative representation of OH groups. Have any other studies been using such proxy?

Reply: As stated in our original manuscript, one may expect that the amount of OH groups (relative to that of C-H groups) that pollen samples contain may affect their hygroscopicity. In addition, a number of previous studies found that heterogeneous aging of organic materials would lead to increase in hygroscopicity; in addition, they also found that the IR absorption intensity for

the O-H stretching mode would increase and the IR absorption intensity for the C-H stretching mode would decrease upon heterogeneous oxidation. Therefore, we used the intensity ratio of the O-H stretching vibration band to the C-H stretching mode to represent the amount of OH groups in a qualitative manner (not quantitatively, however) and explored if there was any correlation between this intensity ratio and measured hygroscopicity. In the revised manuscript (page 9-10, line 181-190) we have expanded our discussion to provide further justification: "OH groups and C-H groups in organic compounds are generally considered to be hydrophilic and hydrophobic, and one may expect that the amount of OH groups (relative to that of C-H groups) that organic samples contain may affect their hygroscopicity. For example, it was found in many previous studies (Eliason et al., 2003; Asad et al., 2004; Hung et al., 2005; Najera et al., 2009) that heterogeneous reactions of organic materials with O_3 and OH radicals would increase the IR absorption intensity for the O-H stretching mode and decrease the IR absorption intensity for the C-H stretching mode, meanwhile leading to the enhancement in their hygroscopicity. Therefore, in this work we use the intensity ratio of the O-H stretching vibration band (3000-3600 cm^{-1}) to the C-H stretching mode (2920 cm^{-1}) to qualitatively represent the amount of OH groups pollen samples contain."

7. P11, L201-203, It only indicates that there is a correlation between water adsorption and OH groups in pollen samples. As discussed in L302-316, there may be other factors contribute to the water uptake. It is suggested to revise these related statements.

Reply: As suggested, in the revised manuscript (page 12, line 221-224) we have revised our discussion: "These results imply that water adsorption by pollen samples could be mainly contributed by OH groups of organic compounds they contained; in addition, other factors, such as porosity and internal structure, may also be important for hygroscopic properties of pollen grains."

8. P12, L215, k parameter is just a fitting from the data. As for now there is no physical meaning for such equation. It is not really a theory.

Reply: We agree with the referee. In the revised manuscript (page 13, line 236) the title of Section 3.2.1 has been from "Theories" to "Hygroscopicity parameterizations", and throughout the revised manuscript (e.g., page 1, line 34) "κ-Köhler equation" has been changed to "κ-Köhler equation".

9, P16, L279-L284, As mentioned above, the k value is obtained from the fitting of these data points, of course, this should fit it well, otherwise k value is wrong. As for Freundlich approach, what are the A and B values? To compare these two different approaches, further discuss is needed.

Reply: We respect but do not quite agree with this comment, and would like to clarify it here. In our work we attempted to use both the κ-Köhler equation and the Freundlich adsorption isotherm to fit our experimental data. For the experimental data shown in Figure 4, if fitted with the κ-Köhler equation, the best fitting gave a $\kappa \cdot \rho_w/\rho_p$ value of 0.036±0.001, and as shown in Figure 4, the κ-Köhler equation fitted the experimental data very well; if fitted with the Freundlich adsorption isotherm, the best fitting gave an A value of 1.19 and a B value of 0.091, and as shown in Figure 4, the Freundlich adsorption isotherm failed to fit the experimental data. This is why we have stated in our original manuscript that the κ-Köhler equation described our experimental data very well but the Freundlich adsorption isotherm did not.

10, P17, L286-289, If use density of 1 g/cm3, that means k values may be 2 times higher or lower when considering range of 0.5 -2 g/cm3. This is a huge uncertainty. That mean you cannot really compare k values for different species unless they have very similar density.

Reply: As pointed out by the referee, our derived κ values have large uncertainties, mainly due to large uncertainties in pollen density. However, as shown in Eq. (5), at a given RH mass hygroscopic growth factors, m/m_0, depend on $\kappa \cdot \rho_w/\rho_p$ rather than κ; to compare the hygroscopicity of different pollen species, we should compare $\kappa \cdot \rho_w/\rho_p$ values rather than κ values; in our work to make the comparison mathematically simpler, we assume a pollen density of 1 g cm^{-3} and thus $\kappa \cdot \rho_w/\rho_p$ is equal to κ. As a result, when we compare hygroscopicity of the six pollen species examined in our work, we do not need to assume that they have very similar density.

11, P18, L298, It is not clear what does "All the errors are statistical only." mean.

Reply: We actually mean that all the errors given here are standard deviations. In the revised manuscript (page 18, line 333) we have changed this sentence to "All the errors given in this work are standard deviations."

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Our replies are in black.

Changes to the manuscript are highlighted in red both in here and in the revised manuscript.

Reply to Ref #2

The hygroscopicity of pollen species is not well-recognized. The authors investigated six different type of pollen particles using two methods. This work provides valuable dataset for hygroscopicity study community. I have two major comments, which should be addressed and implemented in the revised manuscript. Afterwards, I would like to review another round.

Reply: We would like to thank Ref #2 for his/her insightful and detailed comments, which have largely helped us improve our manuscript. We have addressed all the comments adequately in the revised manuscript, as detailed below.

(1) In 3.2.1 Theories, the authors assumed the pollen grains are spherical, then, build the link between kappa and mass hygroscopic growth. While, the pollen gains may not the case and are porous in real world. Assuming a spherical particle could lead to a big bias, for example, higher increase in mass, but, smaller hygroscopic growth in diameter. Actually, the mass growth is significant, but the kappa is very small value compared the atmospheric secondary organic aerosols. The authors only mentioned in line 362-364 that porosity and internal structure, might play an important role in determining the hygroscopicity of pollen grains. But no any discussion in theory part. A detail discussion on the non-spherical situation and its effects on the relationship between kappa and mass growth should be given.

Reply: We agree with the referee, and as suggested, in the revised manuscript (page 17-18, line 323-330) when we discuss κ values of pollen species we have added a few sentences to further discuss the particle sphericity assumption and its implications for the derived κ value: "It should be noted that in order to convert the measured mass growth to diameter growth and κ values, one key assumption is particle sphericity; nevertheless, pollen grains are known to be non-spherical and porous, and therefore our derived κ values might be smaller than the actual values. For example, although the mass increase was substantial (around 30-50 % at 90% RH) for the six pollen species examined, their κ values at 25 ^{o}C were derived to be in the range of 0.034-0.061, significantly smaller than those (0.1-0.2) for typical secondary organic aerosols produced in smog chamber studies (Petters and Kreidenweis, 2007; Kreidenweis and Asa-Awuku, 2014)."

(2) For the kappa theory proposed by Petters, 2007, the particles being studied should be assume as solution. Differently, Freundlich adsorption isotherm is water adsorption by materials. The principles between two theories are quite different. The authors may clarify the purpose by using two different theories to fit the observed curve. Which method is more suitable to explain the water uptake of pollen?

Reply: First of all, as discussed in Section 3.2.2 in the original manuscript, it was concluded in our work that the modified κ-Köhler equation is more suitable to explain water uptake by pollen because when compared to the Freundlich adsorption isotherm, it fits the experimental data much better.

Furthermore, in Section 3.1.1 of the revised manuscript, we have explained further why we attempted to use these two different equations/theories to fit the experimental data, as detailed below.

We tried to use the modified κ-Köhler equation because it relates our measured mass growth to the single hygroscopicity parameter. In the revised manuscript (page 14, line 258-262) we have added a few sentences to provide further explanation: "Eq. (5) relates mass growth experimentally measured in our work to the single hygroscopicity parameter (κ), which has been widely used in atmospheric science to describe hygroscopic properties of aerosol particles under subsaturation as well as their CCN activities under supersaturation; nevertheless, a few assumptions are needed to derive Eq. (5), as discussed."

We also tried to use the Freundlich adsorption isotherm to fit our data because it provides a direct relationship between RH and our measured mass growth, without any additional assumptions. In the revised manuscript (page 14, line 258-262) we have added one sentence to provide further explanation: "One advantage of the Freundlich adsorption isotherm is that it provides a direct relationship between RH and mass growth which was experimentally measured in our work, without any additional assumptions."

Water adsorption and hygroscopic growth of six anemophilous pollen species: the effect of temperature

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Abstract

Hygroscopicity largely affects environmental and climatic impacts of pollen grains, one important type of primary biological aerosol particles in the troposphere. However, our knowledge in pollen hygroscopicity is rather limited, and especially the effect of temperature has rarely been explored before. In this work three different techniques, including a vapor sorption analyzer, diffusion reflectance infrared Fourier transform spectroscopy (DRIFTS) and transmission Fourier transform infrared spectroscopy (transmission FTIR) were employed to characterize six anemophilous pollen species and to investigate their hygroscopic properties as a function of relative humidity (RH, up to 95%) and temperature (5 or 15, 25 and 37 ^{o}C). Substantial mass increase due to water uptake was observed for all the six pollen species, and at 25 ^{o}C the relative mass increase at 90% RH, when compared to that at <1% RH, ranged from ~30 to ~50%, varying with pollen species. It was found that the modified κ-Köhler equation can well approximate mass hygroscopic growth of all the six pollen species, and the single hygroscopicity parameter (κ) was determined to be in the range of 0.034±0.001 to 0.061±0.007 at 25 ^{o}C. In-situ DRIFTS measurements suggested that water adsorption by pollen species was mainly contributed by OH groups of organic compounds they contained, and good correlations were indeed found between hygroscopicity of pollen species and the amount of OH groups, as determined using transmission FTIR. Increase in temperature would in general lead to decrease in hygroscopicity, except for pecan pollen. For example, κ values decreased from 0.073±0.006 at 5 ^{o}C to 0.061±0.007 at 25 ^{o}C and to 0.057±0.004 at 37 ^{o}C for populus tremuloides pollen, and decreased from 0.060±0.001 at 15 ^{o}C to 0.054±0.001 at 25 ^{o}C to 0.050±0.002 at 37 ^{o}C for paper mulberry pollen.

1 Introduction

Primary biological aerosol particles (PBAPs), an important type of aerosol particles in the troposphere, are directly emitted from the biosphere and include pollen, fungal spores, bacteria, viruses, algae, and so on (Després et al., 2012; Fröhlich-Nowoisky et al., 2016). Emission and abundance of PBAPs are quite uncertain, and annual emission fluxes are estimated to be in the range of <10 to ~1000 Tg for total PBAPs and 47-84 Tg for pollen (Després et al., 2012). Pollen, and PBAPs in general, are of great concerns due to their various impacts on the Earth system (Sun and Ariya, 2006; Ariya et al., 2009; Georgakopoulos et al., 2009; Morris et al., 2011; Morris et al., 2014; Fröhlich-Nowoisky et al., 2016). For example, they can be allergenic, infectious or even toxic, affecting the health of human and other species in the ecological systems over different scales (Douwes et al., 2003; Reinmuth-Selzle et al., 2017; Shiraiwa et al., 2017). The geographical dispersion of anemophilous plants largely relies on pollen dispersal, which in turn depends on the emission, transport and deposition of pollen grains; therefore, pollen plays a key role in the evolution of many ecosystems (Womack et al., 2010; Fröhlich-Nowoisky et al., 2016). In addition, PBAPs can serve as giant cloud condensation nuclei (CCN) and ice nucleating particles (INPs), significantly impacting the formation and properties of clouds and thus radiative balance and precipitation (Möhler et al., 2007; Ariya et al., 2009; Pratt et al., 2009; Pope, 2010; Pummer et al., 2012; Gute and Abbatt, 2018). It has also been proposed that PBAPs may have significant impacts on chemical composition of aerosol particles via heterogeneous and multiphase chemistry (Deguillaume et al., 2008; Estillore et al., 2016; Reinmuth-Selzle et al., 2017; Shiraiwa et al., 2017).

Hygroscopicity is one of the most important physicochemical properties of pollen (as well as aerosol particles in general). Hygroscopicity largely impacts the transport and deposition of pollen grains (Sofiev et al., 2006), therefore affecting their lifetimes, abundance and

spatiotemporal distribution. In addition, hygroscopicity is closely linked to the ability of aerosol particles to serve as CCN and INPs (Petters and Kreidenweis, 2007; Kreidenweis and Asa-Awuku, 2014; Laaksonen et al., 2016; Tang et al., 2016). Several previous studies have measured the hygroscopicity and CCN activities of pollen (Diehl et al., 2001; Pope, 2010; Griffiths et al., 2012; Lin et al., 2015; Steiner et al., 2015; Prisle et al., 2018) and other PBAPs such as bacteria (Pasanen et al., 1991; Reponen et al., 1996; Franc and DeMott, 1998; Ko et al., 2000; Lee et al., 2002; Bauer et al., 2003). For example, water uptake of eleven pollen species was studied using an analytical balance (Diehl et al., 2001), and the mass of pollen was found to be increased by 3-16% at 73% RH and by ~100-300% at 95% RH, compared to that at 0% RH. An electrodynamic balance was employed to investigate hygroscopic growth of eight types of pollen (Pope, 2010; Griffiths et al., 2012), and it was found that their hygroscopic growth can be approximated by the modified κ-Köhler equation, with single hygroscopicity parameters being around 0.1 (depending on the assumed pollen density, as discussed in Section 3.2).

Previous measurements were mostly carried out at or close to room temperature, and the effects of temperature on hygroscopic properties of pollen and other types of PBAPs are yet to be elucidated. To our knowledge, only one previous study (Bunderson and Levetin, 2015) explored the effect of temperature (4, 15 and 20 ^{o}C) on water uptake by Juniperus ashei, Juniperus monosperma and Juniperus pinchotii pollen. It is important to account for the temperature effects, because ambient temperatures range from below -70 to >30 ^{o}C in the troposphere. In particular, the altitude of 0.5-2.0 km to which pollen can be easily transported (Noh et al., 2013) may have temperatures close to or lower than the chilling temperatures for vegetative species (up to 16.5 ^{o}C) (Melke, 2015). Moreover, the temperature in the respiratory tract can reach up to of 37 ^{o}C (the physiological temperature). In the work presented here, a vapor sorption analyzer (VSA) was

employed to investigate hygroscopic growth of pollen grains at different temperature (5 or 15, 25, and 37 ^{o}C), a range covering the chilling temperature to the physiological temperature. Water uptake by pollen was also examined using diffusion reflectance infrared Fourier transform spectroscopy at room temperature to complement the VSA results. Furthermore, transmission Fourier transformation infrared spectroscopy was used to characterize functional groups of dry pollen grains, in an attempt to seek potential links between chemical composition of pollen grains and their hygroscopic properties.

2 Experimental sections

Six pollen species, all from anemophilous plants, were investigated in this work, including populus tremuloides and populus deltoides (provided by Sigma Aldrich) as well as ragweed, corn, pecan and paper mulberry (provided by Polysciences, Inc.). The six pollen species were chosen in our work primarily because they were commercially available. Furthermore, these plants are also widely distributed in the globe. For example, corn is the most produced grain in the world (International-Grains-Council, 2019), and up to 50% of pollen-related allergic rhinitis cases in North America are caused by ragweed pollen (Taramarcaz et al., 2005).

2.1 Fourier transformation infrared spectroscopy

The adsorption of water by pollen species were studied using in-situ diffusion reflectance infrared Fourier transform spectroscopy (DRIFTS) at room temperature (~25 ^{o}C). This technique was described in details in our previous work (Ma et al., 2010), and similar setups have also been used by other groups to investigate adsorption of water by mineral dust (Joshi et al., 2017; Ibrahim et al., 2018). Infrared spectra were recorded using a Nicolet 6700 Fourier transformation infrared spectrometer (FTIR, Thermo Nicolet Instrument Corporation), equipped with an in-situ diffuse reflection chamber and a high-sensitivity mercury cadmium telluride (MCT) detector cooled by

liquid nitrogen. A pollen sample (about 10 mg for each sample) under investigation was placed into a ceramic crucible which was located in the in-situ chamber. A dry air flow and a humidified air flow were first mixed and then delivered into the chamber, and the total flow rate was set to 200 mL/min (standard condition). Relative humidity (RH) in the chamber could be adjusted by varying the flow rate ratio of the dry flow to the humidified flow, and was monitored online using a moisture meter (CENTER 314) with an absolute uncertainty of ±2%. Prior to each experiment, the sample was flushed with dry air for 3 h at 25 ^{o}C, and the reference spectrum was recorded after the pretreatment. Infrared spectra were collected and analyzed using OMNIC 6.0 software (Nicolet Corp.). All the spectra reported here were recorded with a wavenumber resolution of 4 cm^{-1}, and 100 scans were averaged to produce a spectrum. Water adsorption was equilibrated for at least 30 min at each RH to ensure that the equilibrium between water vapor and adsorbed water was reached.

Pollen samples used in this work were also characterized using transmission FTIR equipped with a deuterated triglycine sulfate detector (DTGS) detector. Pollen grains and KBr were mixed with a mass ratio of approximately 1:100 and ground in an agate mortar, and the mixture was then pressed into a clear disc. Transmission FTIR was employed to examine these discs, and a pure KBr disc was used as the reference. All the spectra, each of which was the average of 100 scans, were also recorded with a wavenumber resolution of 4 cm^{-1}.

2.2 Vapor sorption analyzer

Hygroscopic growth of pollen grains was further investigated using a vapor sorption analyzer (Q5000 SA, TA Instruments, New Castle, DE, USA) described in our previous work (Gu et al., 2017; Guo et al., 2018; Jia et al., 2018). In brief, this instrument measured the sample mass as a function of RH under isothermal conditions. The instrument can be operated in the temperature

range of 5-85 °C with a temperature accuracy of ±0.1 °C and in the RH range of 0-98 % with an absolute accuracy of ±1%. RH in the humidity chamber was regulated by using two mass flow controllers to control the dry and humidified nitrogen flows very precisely. The accuracy in RH control was routinely checked by measuring the DRH values for a series of standard compounds, e.g., NaCl, $(NH_4)_2SO_4$, KCl, and etc., and the difference between the measured and theoretical DRH was always <1%. The mass measurement had a range of 0-100 mg and a sensitivity of ±0.01 μg. The initial mass of each sample used in this work was in the range of 0.5-1 mg. For each of the first three types of pollen species (populus tremuloides, populus deltoides and ragweed pollen), three samples in total were investigated, and each sample was studied under isothermal conditions at 5, 25 and 37 °C. For each of the other three types of pollen species (corn, pecan and paper mulberry pollen), experiments were carried out at 15 °C instead of 5 °C, because the instrument could only be cooled down to 15 °C due to a technical problem after we finished experiments for the first three pollen species.

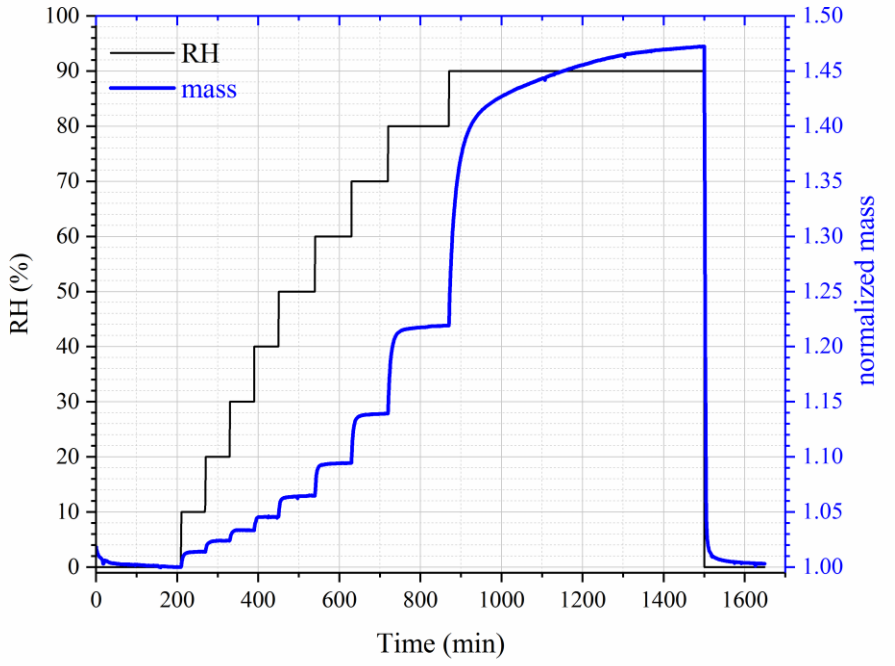

Figure 1. Change of RH (black curve, left *y*-axis) and normalized sample mass (blue curve, right *y*-axis) with time for a typical experiment in which hygroscopic growth of pollen grains was measured. In this figure a dataset for paper mulberry pollen at 25 °C is plotted as an example.

For the first sample, at each temperature the sample was first dried at 0% RH (the actual RH was measured to be <1%); after that, RH was increased stepwise to 95% with an increment of 5% per step and then switched back to <1% to dry the sample again. At each RH, the sample was equilibrated with the environment (i.e. until the sample mass became stable) before RH was changed to the next value, and the sample mass was considered to be stabilized when the mass change was <0.05% within 30 min. Such a measurement at one temperature could take several days. In order to reduce experimental time, the second and third samples were investigated in a similar way as the first sample, except that RH was increased stepwise to 90% with an increment of 10% per step. A typical experimental dataset is displayed in Figure 1 as an example to illustrate the change of RH and normalized sample mass with experimental time.

3 Results and discussion

3.1 FTIR characterization of pollen samples

3.1.1 Infrared spectra of dry pollen samples

Figure 2 shows the transmission FTIR spectra of the six pollen species investigated in our work, and peak assignments can be found in Table 1. A broad band in the range of 3600-3000 cm^{-1}, attributed to O-H stretching vibration (Stuart, 2004; Pummer et al., 2013), and two sharp peaks at 2920 and 2850 cm^{-1}, attributed to C-H stretching (Eliason et al., 2003; Stuart, 2004; Pummer et al., 2013), were observed for all the pollen species. The two peaks at 1747 and 1658 cm^{-1} were assigned to alkyl ester carbonyls (Pappas et al., 2003; Najera et al., 2009; Pummer et al., 2013),

and the two peaks at 1549 and 1458 cm^{-1} (1411 cm^{-1} for paper mulberry pollen) were assigned to C=C stretching and H-C-H deformation (Stuart, 2004; Pummer et al., 2013). In addition, the three peaks at 1053, 997 and 845 cm^{-1} were assigned to C-O stretching, C-C stretching, and C-H out-of-plane bending, respectively (Stuart, 2004; Najera et al., 2009; Pummer et al., 2013).

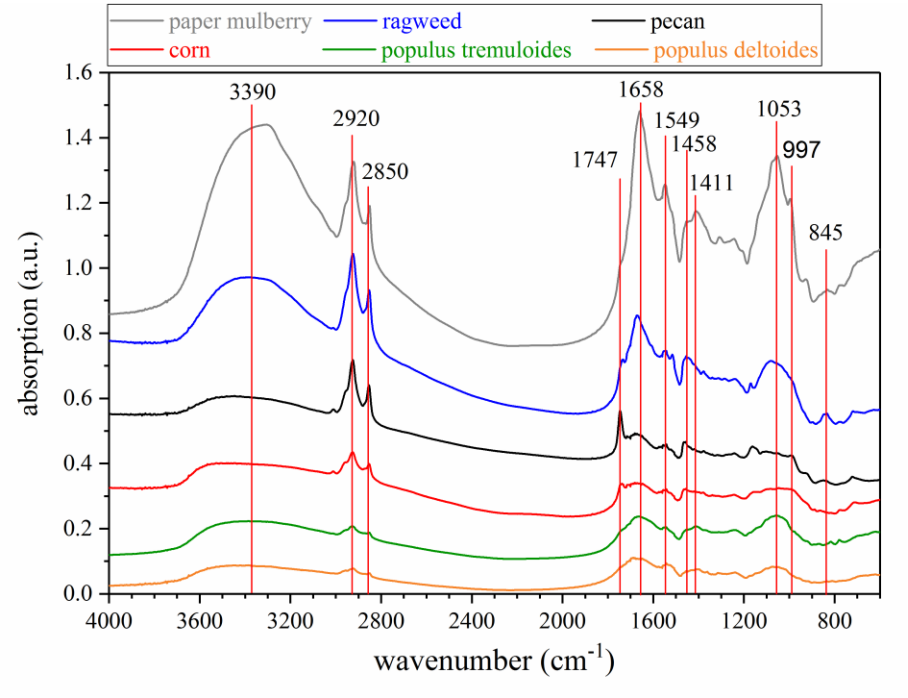

Figure 2. Transmission FTIR spectra of six pollen species investigated in this work.

OH groups and C-H groups in organic compounds are generally considered to be hydrophilic and hydrophobic, and one may expect that the amount of OH groups (relative to that of C-H groups) that organic samples contain may affect their hygroscopicity. For example, it was found in many previous studies (Eliason et al., 2003; Asad et al., 2004; Hung et al., 2005; Najera et al., 2009) that heterogeneous reactions of organic materials with O_3 and OH radicals would increase the IR absorption intensity for the O-H stretching mode and decrease the IR absorption intensity for the C-H stretching mode, meanwhile leading to the enhancement in their hygroscopicity. Therefore, in this work we use the intensity ratio of the O-H stretching vibration

band (3000-3600 cm^{-1}) to the C-H stretching mode (2920 cm^{-1}) to qualitatively represent the amount of OH groups pollen samples contain. As shown in Figure 2, the six pollen species examined in our work can be roughly classified into two catalogues: 1) for populus deltoides, populus tremuloides and paper mulberry pollen, the O-H stretching vibration band is more intensive than the C-H stretching mode, indicating that they contain relatively high levels of OH groups; 2) for ragweed, pecan and corn pollen, the O-H stretching vibration band is less intensive than the C-H stretching mode, indicating that they contain relatively low levels of OH groups. The relation between the amount of OH groups that pollen species contain and their hygroscopicity will be further discussed in Section 3.3.

Table 1. Vibrational mode assignment for six pollen species investigated in this work.

wavenumber (cm^{-1})	vibrational mode
3600-3000	O-H stretching
2920 and 2820	C-H stretching
1747 and 1658	alkyl ester carbonyls
1549	C=C stretching
1458 and 1411	H-C-H deformation
1053	C-O stretching
997	C-C stretching
845	C-H out-of-plane bending

3.1.2 Infrared spectra of pollen samples at different RH

In-situ DRIFTS was employed to explore adsorption of water by pollen grains. Typical spectra of populous deltoides pollen as a function of RH up to 87%, relative to that at <1% RH, are displayed in Figure 3. DRIFTS spectra of other pollen samples at different RH can be found in Figures S1-S5 in the supplement, and are very similar to those for populous deltoides pollen. As

evident from Figure 3, several IR peaks (e.g., 3593, 3205, 2135, and 1616 cm^{-1}) appeared in the spectra at elevated RH, when compared with that at <1% RH, and their intensities increased with increasing RH. The peaks at 3205, 2135 and 1616 cm^{-1} can be assigned to the stretching, association and bending modes of adsorbed water (Goodman et al., 2001; Schuttlefield et al., 2007a; Ma et al., 2010; Hatch et al., 2011; Song and Boily, 2013; Yeşilbaş and Boily, 2016; Joshi et al., 2017; Ibrahim et al., 2018).

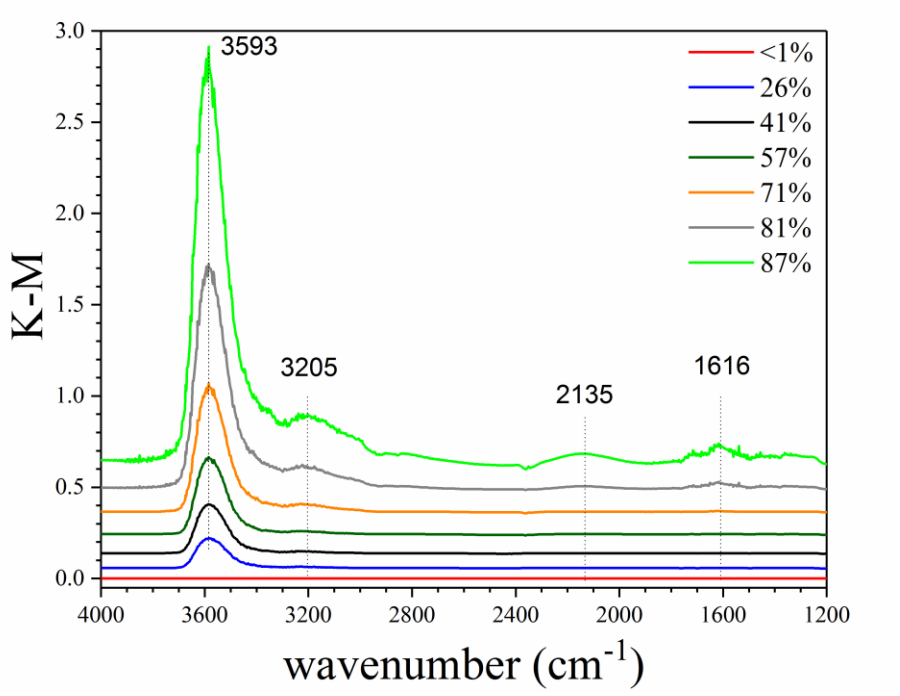

Figure 3. In-situ DRIFTS spectra of populous deltoides pollen as a function of RH (<1, 26, 41, 57, 71, 81 and 87%) at 25 ºC.

The peak at ~3600 cm^{-1} was the most intensive one observed in the spectra, as shown in Figure 3. For comparison, the IR peaks assigned to the stretching mode of adsorbed water on mineral dust and NaCl appeared at lower wavenumbers, typically at around or lower than 3400 cm^{-1} (Schuttlefield et al., 2007a; Ma et al., 2010; Tang et al., 2016; Ibrahim et al., 2018). As a result, the peak at ~3600 cm^{-1} may be assigned to the asymmetric stretching mode of water which

interacted with OH groups in pollen samples (Iwamoto et al., 2003). These results imply that water adsorption by pollen samples could be mainly contributed by OH groups of organic compounds they contained; in addition, other factors, such as porosity and internal structure, may also be important for hygroscopic properties of pollen grains. The intensities of IR peaks at ~3600 cm^{-1} were used to represent the amount of water adsorbed by pollen samples. Table 2 summarizes integrated areas of IR peaks at 3600 cm^{-1} as a function of RH for the six pollen species examined in our work, suggesting that the amount of adsorbed water by pollen samples increased with RH.

Table 2. Integrated areas of IR peaks (at ~3600 cm^{-1}) of adsorbed water as a function of RH for the six pollen species investigated in this work. Wavenumber ranges used for integration are 3750-3300 cm^{-1} for populus deltoides pollen, 3750-3350 cm^{-1} for populus tremuloides pollen, 3750-3400 cm^{-1} for ragweed pollen, 3750-3500 cm^{-1} for corn pollen, 3750-3450 cm^{-1} for pecan pollen, and 3750-3300 cm^{-1} for paper mulberry pollen.

RH (%)	peak area	RH (%)	peak area	RH (%)	peak area
populus deltoides		populus tremuloides		ragweed	
0	0	0	0	0	0
26	22.7	24	5.5	26	10.1
41	36.9	41	16.4	42	18.9
57	57.4	56	35.4	50	24.5
71	93.6	70	66.5	56	30.2
79	137.6	78	91.2	69	49.7
81	164.7	87	156.9	88	104.6
87	293.1				
corn		pecan		paper mulberry	
0	0	0	0	0	0
26	10.0	26	8.6	26	10.2
42	21.5	43	16.9	43	17.7
58	41.9	58	29.5	51	23.1

73	87.5	73	60.0	59	29.8
89	222.2	89	338.9	71	46.7
				86	105.1

3.2 Mass hygroscopic growth

3.2.1 Hygroscopicity parameterizations

The single hygroscopicity parameter, κ, is widely used to describe the hygroscopicity of aerosol particles under both subsaturation and supersaturation (Petters and Kreidenweis, 2007). When the Kelvin effect is negligible (this is valid for pollen grains which are typically >1 μm), the dependence of diameter-based growth factor (GF) on RH can be linked to κ via Eq. (1) (Petters and Kreidenweis, 2007; Tang et al., 2016):

$$RH = \frac{GF^3-1}{GF^3-1+\kappa} \quad (1)$$

If we further assume that the particle is spherical, Eq. (1) can be transformed to Eq. (2):

$$\frac{1}{RH} = 1 + \frac{\kappa}{GF^3-1} = 1 + \frac{\kappa}{\frac{V}{V_0}-1} = 1 + \kappa\frac{V_0}{V-V_0} = 1 + \kappa\frac{V_0}{V_w} \quad (2)$$

where V, V_0 and V_w are the volumes of the particle at the given RH, the dry particle and water associated with the particle at the given RH. In order for Eq. (2) to be valid, it is also assumed that at a given RH, V is equal to the sum of V_0 and V_w. Eq. (2) can be further transformed to Eqs. (3-4):

$$\frac{1}{RH} = 1 + \kappa\frac{\rho_w}{\rho_p}\frac{m_0}{m_w} \quad (3)$$

$$\frac{m_w}{m_0} = \kappa \cdot \frac{\rho_w}{\rho_p} / (\frac{1}{RH} - 1) \quad (4)$$

where ρ_w and ρ_p are the density of water and the dry particle, and m_0 and m_w are the mass of the dry particle and water associated with the particle at the given RH. Since the particle mass, m, is equal to the sum of m_0 and m_w, Eq. (5) can be derived from Eq. (4):

$$\frac{m}{m_0} = 1 + \kappa \frac{\rho_w}{\rho_p} / (\frac{1}{RH} - 1) \quad (5)$$

Using an electrodynamic balance, Pope and co-workers (Pope, 2010; Griffiths et al., 2012) measured hygroscopic growth of eight types of pollen grains, and found that their mass change with RH can be approximated by Eq. (5). It should be noted that the original equation derived by Pope and co-workers (Pope, 2010; Griffiths et al., 2012) has a different format from but is essentially equivalent to Eq. (5). Eq. (5) relates mass growth experimentally measured in our work to the single hygroscopicity parameter (κ), which has been widely used in atmospheric science to describe hygroscopic properties of aerosol particles under subsaturation as well as their CCN activities under supersaturation; nevertheless, a few assumptions are needed to derive Eq. (5), as discussed.

The Freundlich adsorption isotherm is another widely used equation to describe the change of sample mass with RH due to water uptake (Atkins, 1998; Skopp, 2009; Hatch et al., 2011; Tang et al., 2016):

$$\frac{m}{m_0} = 1 + A_f \cdot \sqrt[B_f]{RH} \quad (6)$$

where A_f and B_f are empirical Freundlich constants related to the adsorption capacity and strength. One advantage of the Freundlich adsorption isotherm is that it provides a direct relationship between RH and mass growth which was experimentally measured in our work, without any additional assumptions. In addition, the BET (Brunauer-Emmett-Teller) adsorption isotherm is also widely used to describe water adsorption by insoluble solid particles (Brunauer et al., 1938; Goodman et al., 2001; Henson, 2007; Ma et al., 2010; Tang et al., 2016; Joshi et al., 2017). While the BET adsorption isotherm typically works well for water adsorption of a few monolayers, the mass of adsorbed water, as shown in Section 3.2.2, can reach up to 50% of the dry pollen mass at high RH; therefore, in this work we did not attempt to use the BET adsorption isotherm to describe

water adsorption by pollen grains. Another reason that we did not attempt to use the BET adsorption isotherm is that the BET adsorption isotherm is mathematically more complex and requires the BET surface area to be known.

3.2.2 Mass hygroscopic growth at room temperature

Figure 4 displays the sample mass (normalized to that at 0% RH) as a function of RH for pecan pollen at 25 ^{o}C. Significant increase in sample mass was observed at elevated RH due to uptake of water. Compared to that at 0% RH, the sample mass increased by (2.3±0.3)% at 30% RH, (6.4±0.2)% at 60% RH, (30.3±0.4)% at 90% RH, and up to ~72% at 95% RH. As shown by the data compiled in Tables S1-S3 in the supplement, substantial increases in sample mass were also observed for the other five types of pollen species at 25 ^{o}C (as well as 5 and 37 ^{o}C).

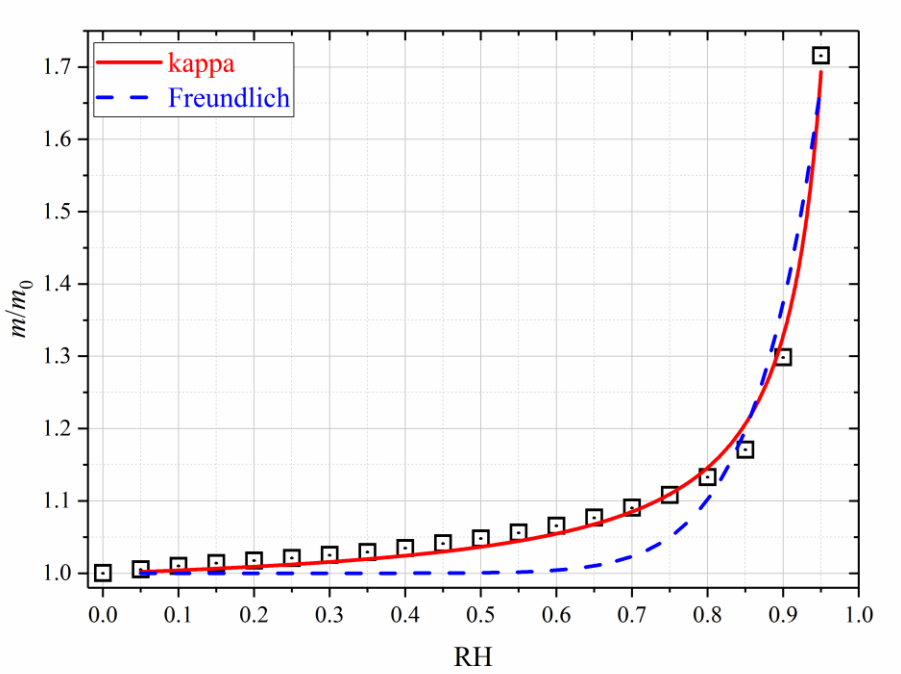

Figure 4. Measured change of sample mass (normalized to that at dry conditions, i.e. m/m_0) of pecan pollen as a function of RH (0-0.95) at 25 ^{o}C. The experimental data are fitted with the modified κ-Köhler equation (solid red curve) and the Freundlich adsorption isotherm (dashed blue curve).

Hygroscopic properties exhibited considerable variations among different pollen species. Figure 5a compares the measured ratios of sample mass at 90% RH to that at 0% RH, $m(90\%)/m_0$, for the six pollen species investigated in this work. We specifically discuss mass changes of pollen grains at 90% RH (relative to that at 0% RH) because aerosol hygroscopic growth at 90% RH was widely reported by laboratory and field studies (Kreidenweis and Asa-Awuku, 2014). As shown in Figure 5a, $m(90\%)/m_0$ determined at 25 °C ranged from 1.293±0.028 (ragweed pollen) to 1.476±0.094 (populus deltoides pollen), i.e. the amount of water adsorbed/absorbed by the six different pollen species at 90% RH varied between ~30% to ~50% of the dry mass.

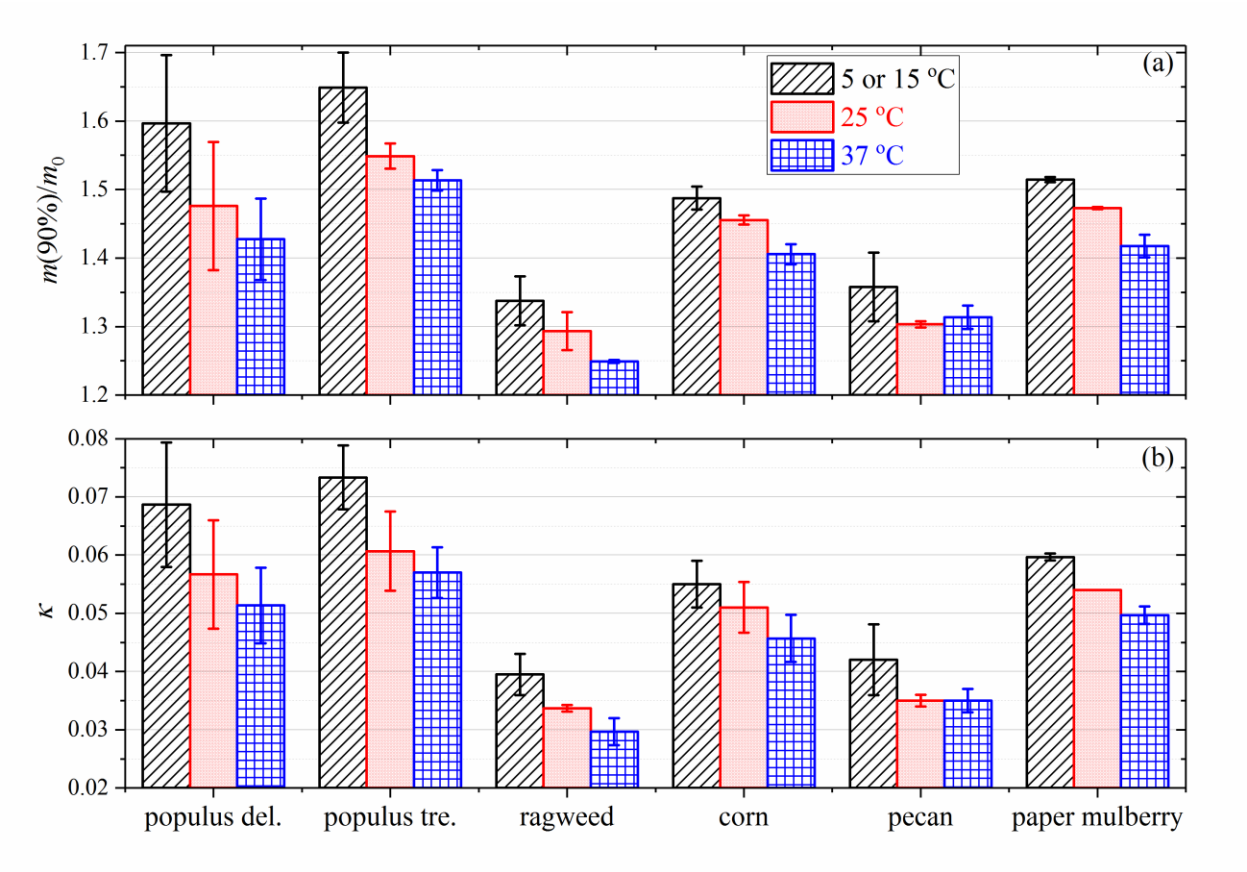

Figure 5. Measured ratios of sample mass at 90% RH to that at 0% RH (a) and derived κ values (b) for six pollen species at different temperatures. The lowest temperatures were 5 °C for populus

deltoides (populus del.), populus tremuloides (populus tre.) and ragweed pollen, and 15 ^{o}C for corn, pecan and paper mulberry pollen.

As shown in Figure 4, the increase of pecan pollen mass with RH at 25 ^{o}C could be satisfactorily described by the modified κ-Köhler equation for the entire RH range (up to 95%). On the contrary, the Freundlich adsorption isotherm significantly underestimated the sample mass at low RH, although it represented the experimental data at high RH reasonably well. In addition, we found that the modified κ-Köhler equation could also approximate the dependence of sample mass on RH for all the six types of pollen species investigated in this work at different temperatures. If we use Eq. (5) to fit m/m_0 against RH, $\kappa \cdot \rho_w/\rho_p$ can be derived. The bulk densities of dry pollen grains were found to vary with species but typically fall into the range of 0.5-2 g cm^{-3} (Harrington and Metzger, 1963; Hirose and Osada, 2016), and for simplicity ρ_p was assumed to be 1 g cm^{-3} in this work (i.e. ρ_w/ρ_p is equal to 1). With the assumptions on dry particle density and also particle sphericity, κ could then be derived from the measured RH-dependent sample mass at a given temperature.

Table 3 summarizes the average κ values at different temperatures for the six pollen species investigated in this work. At 25 ^{o}C, the κ values were found to increase from 0.034±0.001 for ragweed pollen to 0.061±0.007 for populus tremuloides pollen, varied by almost a factor of 2. The κ values measured by Pope and co-workers (Pope, 2010; Griffiths et al., 2012) were approximately in the range of 0.05-0.11 (assuming that ρ_w/ρ_p is equal to 1), in reasonably good agreement with these reported in our work. It should be noted that in order to convert the measured mass growth to diameter growth and κ values, one key assumption is particle sphericity; nevertheless, pollen grains are known to be non-spherical and porous, and therefore our derived κ values might be

smaller than the actual values. For example, although the mass increase was substantial (around 30-50 % at 90% RH) for the six pollen species examined, their κ values at 25 ^{o}C were derived to be in the range of 0.034-0.061, significantly smaller than those (0.1-0.2) for typical secondary organic aerosols produced in smog chamber studies (Petters and Kreidenweis, 2007; Kreidenweis and Asa-Awuku, 2014).

Table 3. Single hygroscopicity parameters (κ) derived in this work for the six pollen species at different temperatures. All the errors given in this work are standard deviations.

pollen type	T (^{o}C)	sample 1	sample 2	sample 3	average
populus	5	0.071±0.001	0.078±0.001	0.057±0.002	0.069±0.011
deltoides	25	0.054±0.001	0.067±0.002	0.049±0.002	0.057±0.009
	37	0.058±0.002	0.051±0.001	0.045±0.002	0.051±0.007
populus	5	0.068±0.001	0.073±0.001	0.079±0.001	0.073±0.006
tremuloides	25	0.053±0.002	0.063±0.002	0.066±0.002	0.061±0.007
	37	0.052±0.002	0.059±0.002	0.060±0.002	0.057±0.004
ragweed	5	0.042±0.001	0.037±0.002	--	0.040±0.004
	25	0.033±0.002	0.034±0.003	0.034±0.002	0.034±0.001
	37	0.027±0.001	0.031±0.002	0.031±0.002	0.030±0.002
corn	15	0.051±0.001	0.059±0.002	0.055±0.002	0.055±0.004
	25	0.046±0.002	0.053±0.002	0.054±0.002	0.051±0.004
	37	0.041±0.002	0.048±0.002	0.048±0.002	0.046±0.004
pecan	15	0.049±0.001	0.038±0.001	0.039±0.001	0.042±0.006
	25	0.036±0.001	0.034±0.001	0.035±0.001	0.035±0.001
	37	0.033±0.001	0.035±0.002	0.037±0.001	0.035±0.002
paper	15	0.059±0.002	0.060±0.002	0.060±0.002	0.060±0.001
mulberry	25	0.054±0.001	0.054±0.001	0.054±0.001	0.054±0.001
	37	0.048±0.002	0.050±0.002	0.051±0.002	0.050±0.002

3.3 Discussion

3.3.1 Reconciliation between IR and VSA results

Our in-situ DRIFTS measurements, as discussed in Section 3.1.2, suggested that water uptake by pollen samples was mainly contributed by OH groups of organic compounds they contained; therefore, one may expect that pollen species which contain higher levels of OH groups would exhibit higher hygroscopicity. Transmission FTIR characterization of pollen species (Section 3.1.1) showed that populus deltoides, populus tremuloides and paper mulberry pollen contained relatively high levels of OH groups, and indeed their hygroscopicity (κ: 0.053-0.054 at 25 ^{o}C) was higher than the other three pollen species, as shown in Figure 5 and Table 3. For comparison, ragweed and pecan pollen contained relatively low levels of OH groups and correspondingly exhibited lower hygroscopicity (κ: 0.033-0.036 at 25 ^{o}C). Corn pollen appeared to be an exception: it contained relatively low levels of OH group but displayed medium hygroscopicity (κ: ~0.046 at 25 ^{o}C). As a result, our results may imply that in addition to chemical composition, other physicochemical properties, such as porosity and internal structure of pollen grains, could also play an important role in determining the hygroscopicity of pollen species. One clue came from environmental scanning electron microscopy observations (Pope, 2010), revealing that pollen grains started to swell internally before significant water uptake on the surface took place.

In our work two complementary techniques were employed to explore hygroscopic properties of pollen species. VSA measured the amount of water absorbed/adsorbed by pollen grains as a function of RH in a quantitative manner, whereas the intensities of IR peaks of adsorbed water at different RH, as characterized by DRIFTS, can be used semi-quantitatively to represent the amount of water associated with particles (Goodman et al., 2001; Schuttlefield et al., 2007b;

Ma et al., 2010; Yeşilbaş and Boily, 2016; Joshi et al., 2017; Ibrahim et al., 2018). We compare our VSA results (i.e. the relative mass change due to water uptake) to the DRIFTS results (i.e. integrated area of IR peaks at ~3600 cm^{-1}). As shown in Figure 6, good correlations between VSA and DRIFTS results are found for all the six pollen species, suggesting that DRIFTS can be used to represent the amount of adsorbed water, at least in a semi-quantitative manner.

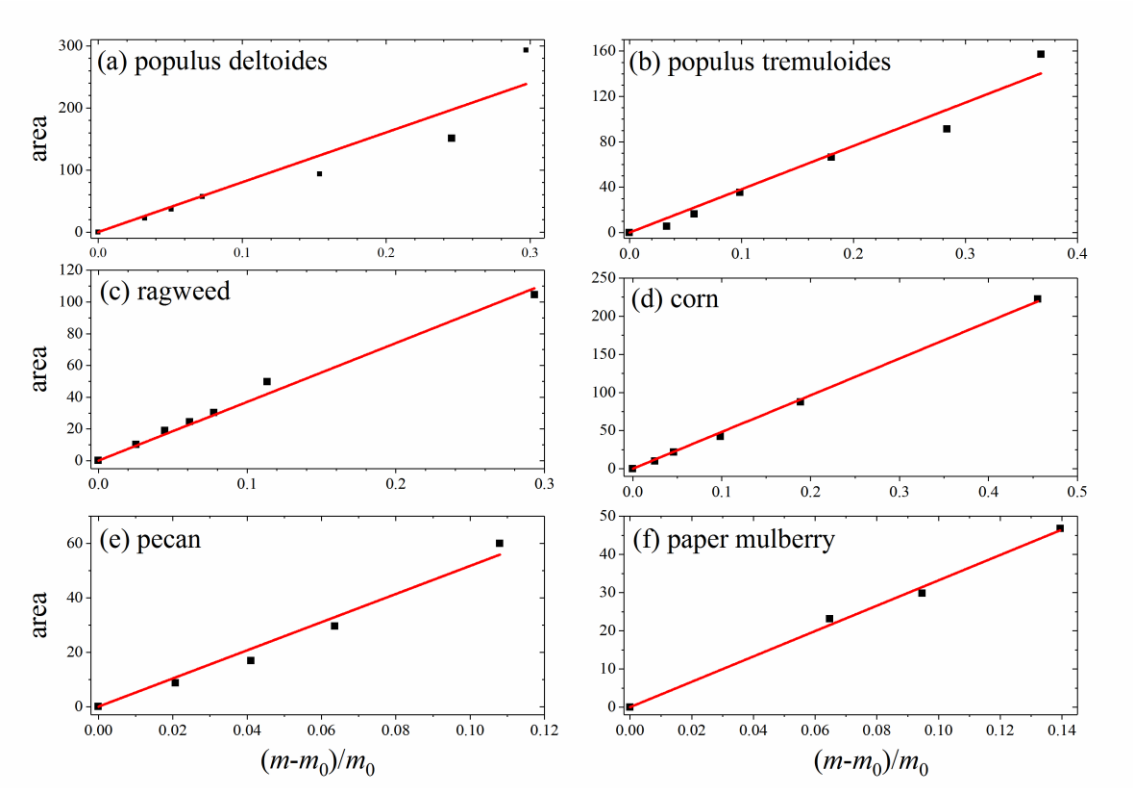

Figure 6. Integrated areas of IR peaks at ~3600 cm^{-1} versus relative mass increase due to water uptake, $(m\text{-}m_0)/m_0$, for six pollen species: (a) populus deltoides; (b) populus tremuloides; (c) ragweed; (d) corn; (e) pecan; (f) paper mulberry.

3.3.2 Effect of temperature

Figure 5a shows the comparison of the measured ratios of sample mass at 90% RH to that at 0% RH, $m(90\%)/m_0$, at different temperatures for the six pollen species. It can be concluded

from Figure 5a that except for pecan pollen for which a small increase in $m(90\%)/m_0$ occurred when temperature increased from 25 to 37 ^{o}C, increase in temperature would lead to small but nevertheless significant decrease in $m(90\%)/m_0$. For example, $m(90\%)/m_0$ decreased from 1.597±0.100 at 5 ^{o}C to 1.476±0.094 at 25 ^{o}C and to 1.427±0.060 at 37 ^{o}C for populus deltoides pollen, and from 1.338±0.036 at 5 ^{o}C to 1.293±0.028 at 25 ^{o}C and to 1.249±0.002 at 37 ^{o}C for ragweed pollen.

We further derived κ values at different temperatures for the six pollen species, and the results are plotted in Figure 5b and summarized in Table 3. Increase in temperature would lead to decrease in κ values, except for pecan pollen. For example, κ decreased from 0.073±0.006 at 5 ^{o}C to 0.057±0.004 at 37 ^{o}C for populus tremuloides pollen, and decreased from 0.060±0.001 at 15 ^{o}C to 0.050±0.002 at 37 ^{o}C for paper mulberry pollen.

4 Conclusion and implications

Pollen grains are one of the most abundant types of primary biological aerosol particles in the troposphere and play important roles in many aspects of the Earth system. Hygroscopicity is among the most important physicochemical properties of pollen grains and largely affect their environmental, health and climatic impacts. However, our knowledge in their hygroscopicity is still quite limited, and especially the temperature effect has been rarely explored.

In this work we investigated hygroscopic properties of six types of pollen species as a function of RH (up to 95%) at 5 (or 15), 25 and 37 ^{o}C. Substantial increase in pollen mass was observed at elevated RH due to water uptake for all the six pollen species. Therefore, change in the mass of pollen grains and their aerodynamic properties at different RH should be taken into account to better understand their transport and deposition in the troposphere. It was found that the mass hygroscopic growth of pollen grains can be well approximated by the modified κ-Köhler

equation. The derived κ values at 25 ^{o}C ranged from 0.034±0.001 to 0.061±0.007, varying with pollen species. DRIFTS measurements indicated that water adsorption by pollen species were mainly contributed by OH groups of organic compounds contained by pollen grains, and indeed pollen species that contained lower levels of OH groups (relative to C-H groups, as determined using transmission FTIR) showed lower hygroscopicity. One exception was corn pollen which contained low levels of OH group but exhibited medium hygroscopicity, suggesting that in addition to chemical composition, other physicochemical properties, such as porosity and internal structure, might also play an important role in determining the hygroscopicity of pollen grains. Due to their moderate hygroscopicity as well as large sizes, pollen grains can thus act as efficient giant CCN which may have significant impacts on cloud and precipitation (Johnson, 1982; Feingold et al., 1999; Yin et al., 2000; Posselt and Lohmann, 2008). It is worth noting that only six different pollen species were examined in our work, and hygroscopic properties of other pollen species commonly found in the troposphere should be further investigated.

The effect of temperature on the hygroscopicity of pollen grains was systematically investigated in this work. Increase in temperature (from 5 or 15 ^{o}C to 25 and 37 ^{o}C), a range covering chilling temperature to physiological temperature, led to small but detectable decrease in pollen hygroscopicity. For example, κ values were found to decrease from 0.073±0.006 at 5 ^{o}C to 0.061±0.007 at 25 ^{o}C and to 0.057±0.004 at 37 ^{o}C for populus tremuloides pollen, and decrease from 0.060±0.001 at 15 ^{o}C to 0.054±0.001 at 25 ^{o}C to 0.050±0.002 at 37 ^{o}C for paper mulberry pollen. Our measurements at 37 ^{o}C (physiological temperature) provide very valuable parameters which can be used in numerical models to better understand the transport and deposition of pollen particles in the respiratory system and thus their impacts on human health (Yeh et al., 1996; Broday and Georgopoulos, 2001; Park and Wexler, 2008; Lambert et al., 2011; Longest and Holbrook,

2012; Tong et al., 2014). Nevertheless, it should be noted that due to the short residence time in the respiratory system, pollen grains and other inhaled particles in general, may not reach equilibrium with water vapor in the respiratory tract.

Due to technical challenges, the lowest temperature we could reach in this work was 5 ^{o}C, in the range of normal chilling temperatures for vegetative species and also in the expected temperature range at the altitudes of 0.5-2.0 km to which pollen grains can be easily transported. Temperatures in the upper troposphere can be as low as below -70 ^{o}C, and it is yet to be explored whether further decrease in temperature to far below 0 ^{o}C will lead to large increase in pollen hygroscopicity. As a result, experimental measurements of pollen hygroscopicity at lower temperatures are warranted and would significantly help better understand the transport of pollen grains in the troposphere. Since water vapor has to be adsorbed or condensed on ice nucleating particles before heterogeneous ice nucleation can take place (Laaksonen et al., 2016), knowledge in hygroscopicity and water uptake at temperatures below 0 ^{o}C would provide fundamental insights into atmospheric ice nucleation, in which pollen grains may play an important role (Pratt et al., 2009; Prenni et al., 2009; Hoose et al., 2010; Pöschl et al., 2010; Murray et al., 2012; Creamean et al., 2013; Tang et al., 2018).

Author contribution

MT, QM and YJL designed the research; WG, CZ, SL and XY did the measurements; MT, QM, YJL and RJH analyzed the results; MT, QM, YJL and RJH wrote the manuscript with contribution from all the co-authors.

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Supplements of

Water adsorption and hygroscopic growth of six anemophilous pollen species: the effect of temperature

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Table S1. Sample mass (normalized to that at 0% RH) as a function of RH at three different temperatures (37, 25 and 5 ºC) for populus deltoides pollen and populus tremuloides pollen. All the errors (±1 σ) are statistical only.

	populus deltoides			populus tremuloides		
RH (%)	37 ºC	25 ºC	5 ºC	37 ºC	25 ºC	5 ºC
0 [a]	1.000	1.000	1.000	1.000	1.000	1.000
5 [b]	1.008	1.008	1.007	1.004	1.006	1.001
10	1.014±0.001	1.013±0.002	1.012±0.004	1.010±0.011	1.008±0.004	1.002±0.001
15 [b]	1.020	1.021	1.020	1.016	1.019	1.006
20	1.024±0.002	1.024±0.004	1.024±0.006	1.022±0.002	1.021±0.004	1.010±0.004
25 [b]	1.030	1.032	1.033	1.035	1.033	1.016
30	1.035±0.001	1.035±0.003	1.036±0.008	1.038±0.005	1.037±0.003	1.019±0.004
35 [b]	1.042	1.044	1.048	1.054	1.049	1.024
40	1.050±0.003	1.051±0.001	1.057±0.004	1.058±0.007	1.058±0.002	1.044±0.009
45 [b]	1.057	1.059	1.074	1.079	1.070	1.041
50	1.070±0.008	1.072±0.006	1.085±0.006	1.084±0.009	1.085±0.002	1.085±0.008
55 [b]	1.077	1.079	1.108	1.112	1.099	1.110
60	1.098±0.014	1.104±0.015	1.123±0.013	1.121±0.011	1.122±0.004	1.126±0.009
65 [b]	1.109	1.110	1.156	1.159	1.144	1.159
70	1.144±0.022	1.154±0.025	1.181±0.021	1.176±0.012	1.181±0.005	1.190±0.010
75 [b]	1.164	1.170	1.233	1.231	1.218	1.238
80	1.226±0.033	1.246±0.041	1.288±0.038	1.273±0.012	1.284±0.007	1.308±0.012
85 [b]	1.276	1.297	1.401	1.373	1.368	1.405
90	1.427±0.060	1.476±0.094	1.597±0.100	1.514±0.015	1.549±0.018	1.649±0.051
95 [b]	1.986±	2.045	2.340	1.872	1.923	2.267

a: Sample mass was normalized to that at 0% RH, and therefore the uncertainty for the data at 0% RH was not reported.

b: Only one measurement was carried out at each of these RH, and therefore the uncertainties were not reported.

Table S2. Sample mass (normalized to that at 0% RH) as a function of RH at different temperatures for ragweed pollen (37, 25 and 5 ºC) and corn pollen (37, 25, and 15 ºC). All the errors (±1 σ) are statistical only.

	ragweed			corn		
RH (%)	37 ºC	25 ºC	5 ºC	37 ºC	25 ºC	15 ºC
0 [a]	1.000	1.000	1.000	1.000	1.000	1.000
5 [b]	1.006	1.005	1.003	1.001	1.004	1.000
10	1.011±0.002	1.011±0.002	1.012±0.007	1.005±0.002	1.009±0.001	1.002±0.001
15 [b]	1.016	1.015	1.011	1.006	1.013	1.002
20	1.020±0.003	1.021±0.005	1.021±0.008	1.015±0.003	1.018±0.001	1.011±0.006
25 [b]	1.026	1.025	1.020	1.018	1.025	1.009
30	1.029±0.003	1.031±0.005	1.030±0.006	1.028±0.003	1.030±0.002	1.025±0.007
35 [b]	1.036	1.039	1.036	1.035	1.041	1.029
40	1.040±0.002	1.045±0.006	1.043±0.001	1.047±0.002	1.046±0.003	1.048±0.010
45 [b]	1.048	1.056	1.055	1.061	1.058	1.047
50	1.053±0.002	1.061±0.007	1.060±0.009	1.072±0.004	1.067±0.002	1.076±0.006
55 [b]	1.064	1.078	1.080	1.093	1.083	1.081
60	1.074±0.001	1.082±0.009	1.083±0.017	1.105±0.006	1.099±0.002	1.112±0.005
65 [b]	1.086	1.108	1.113	1.134	1.121	1.132
70	1.100±0.002	1.114±0.012	1.115±0.015	1.152±0.008	1.147±0.001	1.161±0.012
75 [b]	1.119	1.151	1.156	1.195	1.189	1.199
80	1.148±0.002	1.167±0.014	1.177±0.024	1.231±0.010	1.243±0.002	1.253±0.013
85 [b]	1.188	1.234	1.253	1.311	1.318	1.332
90	1.249±0.002	1.293±0.028	1.338±0.036	1.406±0.015	1.456±0.007	1.488±0.017
95 [b]	1.473	1.566	1.776	1.691	1.813	1.920

a: Sample mass was normalized to that at 0% RH, and therefore the uncertainty for the data at 0% RH was not reported.

b: Only one measurement was carried out at each of these RH, and therefore the uncertainties were not reported.

Table S3. Sample mass (normalized to that at 0% RH) as a function of RH at different temperatures (37, 25, and 5 ºC) for pecan pollen and paper mulberry pollen. All the errors (±1 σ) are statistical only.

	pecan			paper mulberry		
RH (%)	37 ºC	25 ºC	15 ºC	37 ºC	25 ºC	15 ºC
0 [a]	1.000	1.000	1.000	1.000	1.000	1.000
5 [b]	1.003	1.005	1.002			
10	1.008±0.002	1.008±0.004	1.004±0.003	1.011±0.001	1.014±0.001	1.013±0.001
15 [b]	1.011	1.014	1.008			
20	1.016±0.001	1.015±0.004	1.015±0.001	1.021±0.001	1.024±0.001	1.024±0.001
25 [b]	1.020	1.021	1.021			
30	1.026±0.002	1.023±0.003	1.027±0.001	1.033±0.001	1.034±0.001	1.037±0.001
35 [b]	1.031	1.029	1.035			
40	1.038±0.002	1.032±0.003	1.041±0.003	1.050±0.001	1.046±0.001	1.054±0.001
45 [b]	1.047	1.041	1.052			
50	1.054±0.002	1.046±0.003	1.058±0.005	1.073±0.001	1.065±0.001	1.077±0.001
55 [b]	1.067	1.056	1.075			
60	1.076±0.004	1.064±0.002	1.081±0.009	1.103±0.001	1.095±0.001	1.111±0.001
65 [b]	1.096	1.077	1.108			
70	1.108±0.007	1.089±0.002	1.116±0.013	1.148±0.004	1.139±0.001	1.163±0.001
75 [b]	1.141	1.108	1.161			
80	1.161±0.013	1.132±0.002	1.179±0.022	1.226±0.004	1.221±0.003	1.255±0.001
85 [b]	1.230	1.171	1.275			
90	1.314±0.017	1.303±0.004	1.358±0.050	1.418±0.017	1.473±0.001	1.515±0.001
95 [b]	1.574	1.716	1.927			

a: Sample mass was normalized to that at 0% RH, and therefore the uncertainty for the data at 0% RH was not reported.

b: Only one measurement was carried out at each of these RH, and therefore the uncertainties were not reported.

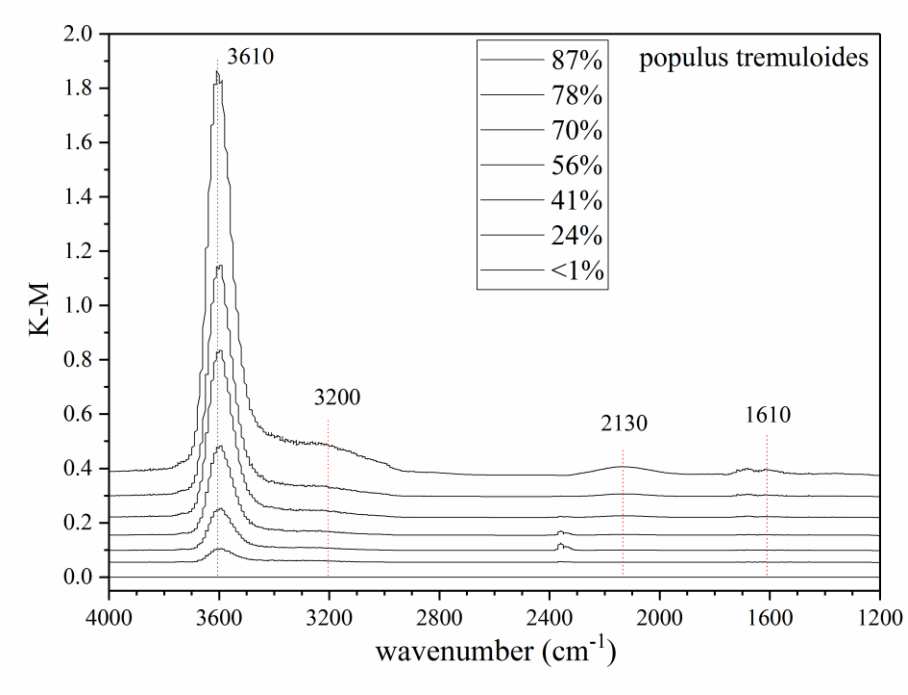


Figure S1. In-situ DRIFTS spectra of populous deltoides pollen as a function of RH at 25 ºC, relative to that at <1% RH.

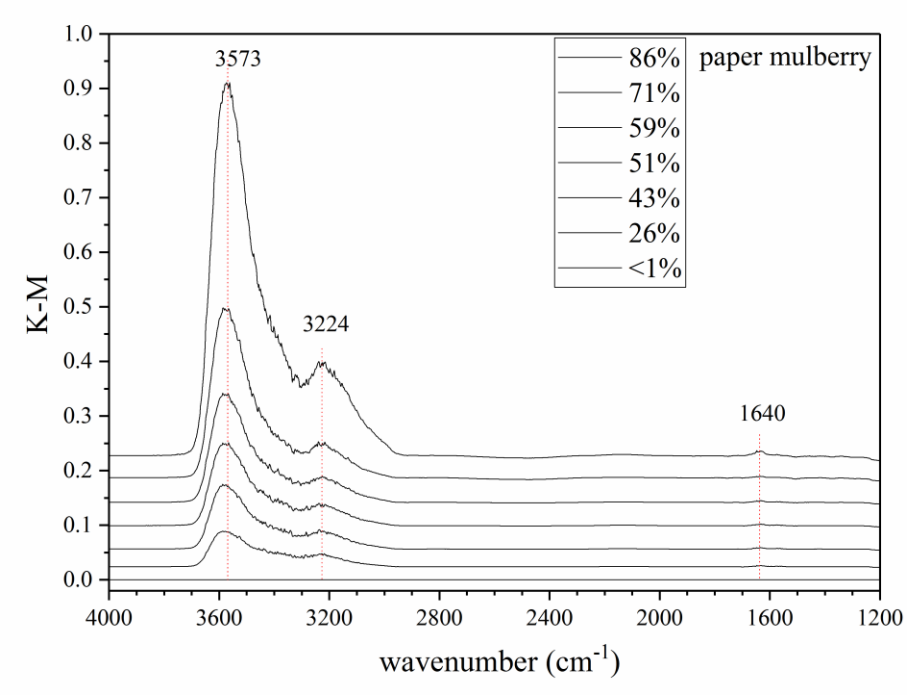
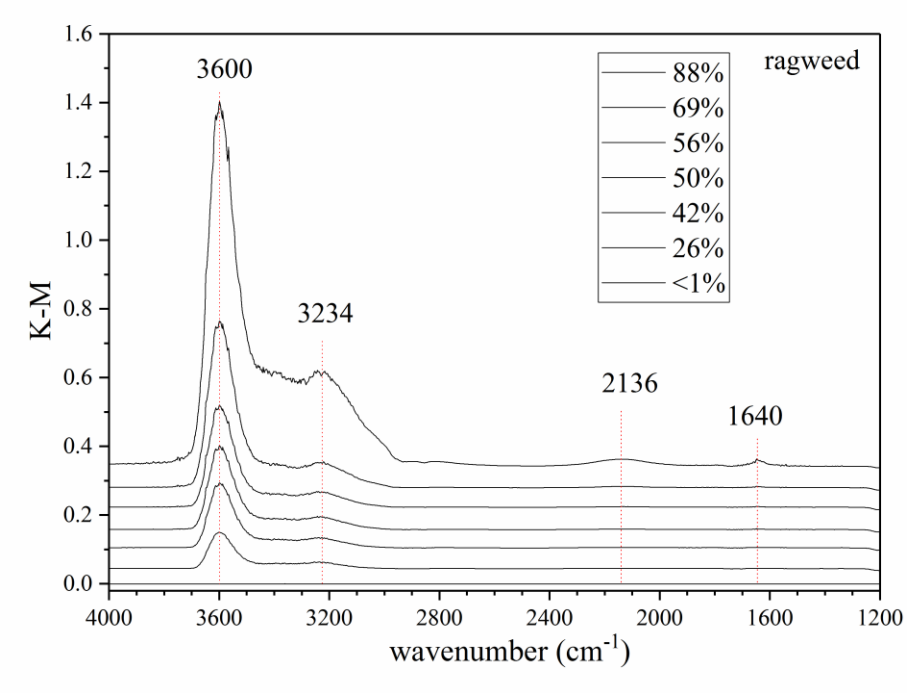


Figure S2. In-situ DRIFTS spectra of paper mulberry pollen as a function of RH at 25 ºC, relative to that at <1% RH.



40

41 **Figure S3.** In-situ DRIFTS spectra of ragweed pollen as a function of RH at 25 ºC, relative to that

42 at <1% RH.

43

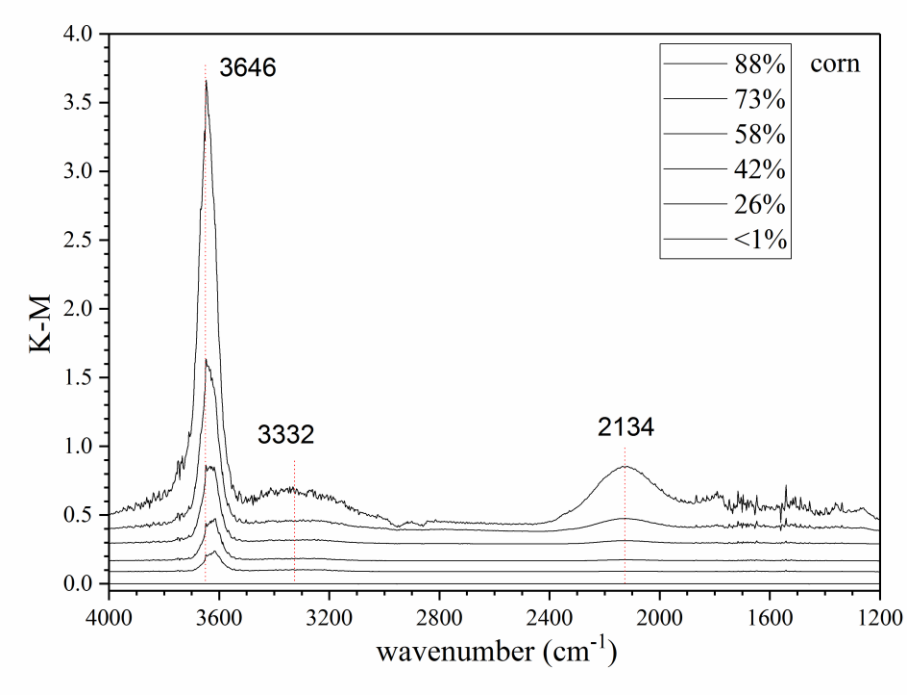


Figure S4. In-situ DRIFTS spectra of corn pollen as a function of RH at 25 ºC, relative to that at <1% RH.

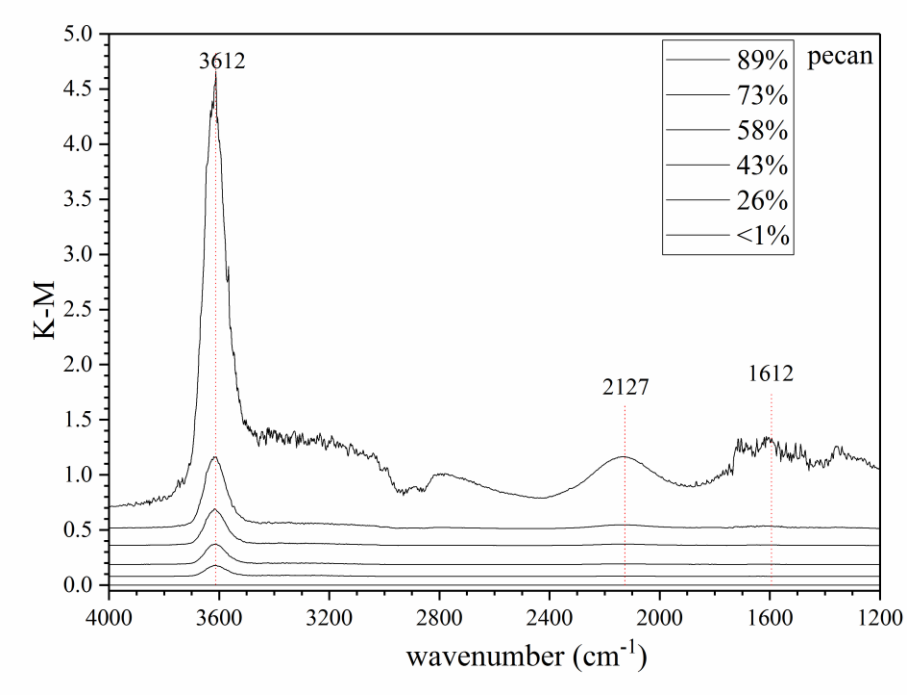


Figure S5. In-situ DRIFTS spectra of pecan pollen as a function of RH at 25 ºC, relative to that at <1% RH.