

Response to referees:

We thank both referees for their insightful and thorough reviews. The manuscript is greatly improved as a result of their efforts.

Referee #1:

General comments:

We are well aware of the merits of the “per meg” scale for reporting O₂ measurements and have used it in all of our past publications. We had chosen to present this research using mole fraction because the Local Biospheric Exchange Ratio (“ α_B ” in the original draft, see our reply to comment #1 of Referee #2 below) characterizes a mole-for-mole flux balance. However, we had overlooked the importance of converting to per meg and then back to ppm equivalent. We have now done this throughout the paper and added an explanatory paragraph on units.

The second general comment suggests the use of CO and ¹⁴C observations to assess the influence of fossil fuel combustion on our dataset. We agree completely that these would be ideal tools for quantifying the role of combustion at the forest. Unfortunately, this simply isn’t an option. For most of the time we were collecting data, there were no contemporaneous measurements of either species. While CO is now routinely measured at the forest, it comes too late for our dataset. Routine ¹⁴C measurements are prohibitively expensive. We did explore the use of SF₆ as an indicator of anthropogenic signals, but were convinced by Jocelyn Turnbull’s work (Turnbull et al. GRL VOL. 33, L01817, 2006) that this was not a suitable tool. This is why we turned to modelling to gain insight into non-local influences.

Specific comments:

All suggestions have been adopted with the following exceptions and/or notes

P3, line 21-23: The tanks and regulators are now described in the caption of Table 1, as well as rough estimates of the uncertainties of the composition. Based on the intercomparison of the various tanks, we see no clear evidence of drift in the O₂ composition of the tanks over the years of use. However, any quantitative estimate of stability is beyond the scope of this paper.

P6, line 7-10: Perhaps we are simply missing something important, but it seems to us that the changes in O₂ and CO₂ over a six-hour period will be faithfully recorded if the standard tanks are essentially constant over that same six-hour period. If the O₂ drift in the tanks is (for example) 50 per meg over a year, the drift will be about 0.03 per meg over the 6-hour period from which each slope is determined. While 50 per meg drift would doom an O₂-based land-sink calculation, the 6-hour drift would have no impact on the slopes. Since this paper focuses exclusively on slopes, we have chosen to omit any discussion of long term stability of the standards.

P6, line 27: Again, this kind of analysis is beyond the scope of this paper. However, your question prompted us to make a quick linear fit to the low-intake O₂ time series shown in Figure 3. It does reveal a drop of about 9 ppm. Given that we have made no effort to develop a long-term calibration curve for our tanks, this is reassuring.

P7, line 29: The Lagrangian back-trajectory model provides us with information that helps us use our 1-dim forward box model, but the two models are not linked in any specific way. We run HYSPLIT to tell us (in general terms) how far a parcel of air might move in 6 hours, and from whence the parcels have come when they arrive at our study site. Informed by the HYSPLIT results (in a general sense), we designed our forward model to carry a parcel of air a realistic distance over a fictitious landscape and record the influence of the idealized fluxes to/from the landscape. We have tried to clarify this with minor changes in the text.

P8, lines 3-4: The box heights used in our 1D model are indeed derived from PBL values extracted from the NAM12 dataset. Again, we have tried to clarify this.

P8, lines 11-12 and 16-17: Regarding the particulars of the sensitivity tests: Our 1D model is so simplified that we feel its output values for slopes and O₂ and CO₂ ranges are not particularly meaningful. This is why we haven't included a table of values from the model. What we do consider meaningful and significant is that the model is consistent with our intuition: A mix of forest and urban fluxes usually yields a slope intermediate between 1.4 and 1.05, taller boxes yield smaller signals, and the seasonal cycle in fluxes leads to a seasonal cycle in the ranges of O₂ and CO₂ values predicted for any particular 6-hour period. These intuitive results hold despite (largely independent) diurnal cycles in the various fluxes and the depth of the PBL. We have changed the text to emphasize this point.

Regarding the citation of Conley (2018), the referee raises an interesting point. While Conley (2018) is not peer reviewed, it is freely available electronically. For this reason, we were careful not to use it as a reference for results. Instead, we cite it only for more complete descriptions of the model, or the experiments carried out with the model. Essentially, we have used it as an appendix or supplement to this manuscript. If this kind of citation is inappropriate, we are certainly willing to excerpt the relevant information and turn it into a formal supplement.

Page 8, Section 4.2: The determination of the "footprint" or "region of influence" has been a challenging part of this analysis from the beginning. Flux footprints (which have been thoroughly studied at Harvard Forest) are quite different from the concentration footprints that are relevant to our study. The work of Gerbig et al. (2006) is the closest match to our present needs (Figs. 2 & 4 of that paper), but even that is geared toward quantifying the cumulative influence of many small contributions integrated over a large area. Nonetheless, Gerbig et al. does make it clear that a 6-hour period corresponds to a primarily forested region within ~100km or less of our study site.

Page 8-9: The calculation is indeed simple and straightforward, but it probably isn't meaningful in this case. On account of its many simplifications, we don't trust the model enough to quantify the contributions of fossil fuels and biosphere. As discussed above, we only consider the model a tool for testing the end-member mixing framework. Instead, as discussed now in section 4.3, we use observationally based estimates of carbon fluxes, both biogenic and anthropogenic to argue that our summer daytime slopes are the closest thing we have to a pure biogenic signal.

Page 9, line 4-5: We sought, and received from the authors, some clarification of ambiguous language in the Sargent et al. paper. Based on this, and in response to questions from both reviewers, this section has been significantly revised.

Page 10, Lines 2-3: We have substantially reworked the discussion of observations and the information gleaned from data subsets (high/low, day/night, summer/winter). We believe this new treatment will address your question.

Page 10, Section 5.3: We have replaced the Lomb-Scargle periodograms with a year-long climatology of slopes (averaged weekly), as well as a time series of slopes for the entire study period (averaged monthly). In addition, the discussion of observations now contains a substantial examination of the possibility of variation in the local biospheric exchange ratio (formerly α_B).

Page 17, Figure 1: All standard tanks were stored horizontally in an insulated blue box (for the reasons you describe). We have changed the text and figure to indicate this.

Page 23, Figure 7: As discussed above, we used HYSPLIT only to establish the area of influence. This is not expected to exhibit secular variation between the beginning of our study period and the present. Furthermore, the HYSPLIT back-trajectories were not linked to our 1-D forward model runs. Thus, any one trajectory is as good as any other and we simply chose a representative example.

Page 24, Figure 8: Thank you for pointing this out. The figure was left over from an earlier draft. We have now removed it entirely.

Referee #2 (Brit Stephens):

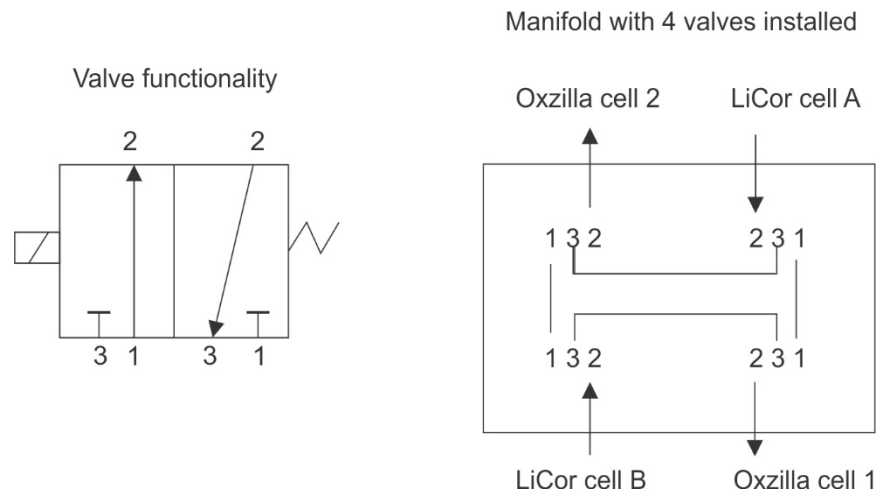
- 1) Extrapolation to α_B : Essentially all suggestions incorporated. Your comments prompted us to realize that we had been unwittingly treating α_B as a prognostic concept, rather than a diagnostic one. As you correctly point out, α_B is a catch-all parameter, much like the “land sink” in the GCP budget is really just a collection of left-over fluxes that haven’t been captured in the accounting of ocean and atmospheric measurements. Thus, α_B is not something that we measure directly. Instead, we measure something (now named the Local Biospheric Exchange Ratio or LBER in our paper) that contributes to α_B and with luck, is an important determinant of it. Thanks for pointing out our misrepresentation.
- 2) As requested, we have replaced the periodograms with time series of monthly mean values, and a separate climatological year of weekly mean values. The text in Section 5.2 has been altered to address this with more clarity.
- 3) We have substantially reworked the “Discussion of Observations” section to address these concerns. We have added figures giving graphical comparisons for average slopes calculated in various ways. To complement this, we organized the discussion by agency (flux sizes, atmospheric dynamics, changing physiology) rather than data category (high/low, day/night, winter/summer). We think this leads to a much more coherent examination of possible mechanisms.

Minor comments: We have adopted your suggestions in most cases. Exceptions, or further explanation as needed, are below:

P1 Line 10: Changed, to avoid the use of αB and instead, refer to “local biotic influences”.

P3 line 17: Our understanding is that, in principle at least, tiny ice crystals could be entrained/suspended in the airstream and not melt/sublimate before reaching the filters (a bit like frazil ice in a river). Perhaps this is impossible at the low flow rates in our system, but this was the thinking behind the original statement. However, whether such ice crystals evaporate/sublimate while airborne or in the filter, they’ll never raise the dewpoint much above -90C. In any event, we have changed the text to focus on keeping particulates out of the downstream equipment.

P3 line 30: Because we used simple dual-action valves and wanted completely symmetric flow paths, we used four valves connected to a custom manifold. Here is a diagram:



A changeover occurs when all four valves switch simultaneously from energized to de-energized (or vice versa). We can include this diagram in the paper if such a level of detail is deemed appropriate.

P5 line 12: We have changed the text to refer to both lines (which have stale air in them and need flushing) and regulators (which are the components that need purging, as you point out). The details of the tanks and regulators are now given in the caption of the table listing tank compositions.

P5 line 15: We have added the transition time, as suggested and tried to clarify our (admittedly arcane) averaging protocol. As we now explicitly state in the text, we don’t actually use all of the values that postdate the 70% transition threshold. When developing our analysis protocol, we considered a range of post-transition “deadtimes”, and finally settled on using the last ~70% of the post-transition data. This value minimizes the standard error on each calibration run. That is to say, working back from the end of the cal run, as we retained more and more values the standard error fell and then began to climb. The climb was a result of gradually including remnants of the previous analyte. Based on multiple cal runs, we find minimal standard error when using 5, 4 or 3 of the 24-second changeover-blocks of data for transition times in the first 51 secs, 103 secs or 180 secs, respectively. If the transition happens after 180s, we discard that cal run. This algorithm works well for all cal runs except the Low Span (LS) tank. In

those runs, where the LS tanks has a CO₂ mixing ratio roughly 100ppm higher than the working tank (WT), 6 minutes is simply insufficient for the system to come to equilibrium. Consequently, all LS-WT comparisons are biased low. For our 6-hour slope calculations, this bias is not a problem. Looking ahead to work on low-frequency variations and intercomparisons with other labs, we are developing ways to correct for this bias.

P5 line 20 and P6 line 26: The CO₂ scale is now correctly referenced, as suggested. The O₂ scale is referred to as the “S2” scale by Keeling et al. 2007. When originally preparing this manuscript we did indeed check with Ralph Keeling and this was his preferred name for the scale (and preferred citation).

P5 lines 30-31: You have (perhaps inadvertently) drawn attention to a mistake in our reasoning. The statement about the loose packing nut was true, but we realize now it was likely irrelevant. The crossover valve is only involved in the intake of sample air. However, the instrumental performance shown in Figure 2 is based on standard tanks. Thus, while a loose packing nut might well degrade analyses of samples, it can't explain noisy standard data. At this point, our best explanation for poor standard tank precision is that a leak developed around or within the 6-port selector (Valco) valves. We won't be able to test this hypothesis until the instrument is fully operational again. Whether the tightening of the packing nut changes sample precision also remains to be seen, as the instrument has not been operating since we made that repair. In the meantime, we have adjusted the text accordingly, and omitted mention of the loose packing nut.

P6 Line 2: Done. Please see response to reviewer #1.

P6 Line 6: In this case, we believe the uncertainties on the measurements are the right quantities to use in the Deming regressions. It is certainly true that atmospheric variability from one measurement to the next is much larger than the uncertainties, but unlike analytic errors, the atmospheric variability in O₂ and CO₂ is correlated. Whatever the reason ambient O₂ might be high or low from one moment to the next, CO₂ will be low or high. This is the essence of the linear model we are applying (with the slope as a free parameter) and fundamentally, there's no difference between these variations and phenomena we are trying to measure. In contrast, the errors are uncorrelated, non-negligible and unequal for the two species.

P6 Line 16: We have substantially rewritten this subsection and changed our approach to calculating the errors on our measurements. Before, we had started with the scatter on a single calibration tank (and then compared this with inter-tank scatter). Now, we start with inter-tank scatter and work backward to errors on single measurements, which we then use to estimate errors on the atmospheric values. We believe this is a more conservative approach, since inter-tank scatter captures additional potential sources of systematic error.

P8 Line 3: We have changed the text to explain how $\frac{1}{2}$ the PBL height is used in the STILT model. Since we use STILT to determine the provenance of our signal, we use $\frac{1}{2}$ the PBL height in our own 1-D model for consistency. Furthermore, Eq. 4 shows that changing the box height would change the magnitude of all of our signals in the same way. Thus, while a better value for box height might give different slopes, and would certainly give different ranges of variability for each species, it would make no difference to the conclusion we draw from the 1-D model: A simple mix of fluxes gives a simple mix of slopes.

Section 4.2: Our guess is that we are using “region of influence” differently than you are. We aren’t trying to identify all parts of the country that contribute to the observed concentrations at Harvard Forest. This would indeed extend over an enormous area. Instead, we are asking where is the parcel of air at $t = 0$ that will be at our study site at $t = 6\text{hr}$ (the end of a sampling interval). Likewise, where is the parcel of air at $t = -6$ that will be at our site at $t = 0$. For this reason, we’ve replaced “region of influence” with “6-hour provenance”. Naturally, the composition of the air at $t = 0$ was set by the processes upwind dating back to $t = -6\text{hr}$. We try to capture this with model spin-up. We also assume that during the 6-hour interval ($t = -6$ through $t = 0$), the composition of air arriving in the box farthest upwind is essentially constant. This assumption is supported by the relatively constant wind direction and speed we see in the HYSPLIT back-trajectories (Fig. 7). Furthermore, if we had big shifts in wind direction bringing big changes in the “input” concentration, we wouldn’t have the nice linear relationships we see evolving in most of our O_2 vs. CO_2 plots. Please see also our response to Referee #1, above.

P8, line 32: We sought, and received from the authors, some clarification of ambiguous language in the Sargent et al. paper. Based on this, and in response to questions from both reviewers, this section has been significantly revised.

P9, lines 4-7: As we mention above in reply to your question on P8, line 32, we have revised this section of the text with more specific information. You are correct that a concentrated plume of combusted natural gas would skew our results. Referring to ACES and the EPA’s Greenhouse Gas Reporting Program (<https://ghgdata.epa.gov/ghgp/main.do>) there are few power plants in the general area of Harvard Forest, and fewer still that are natural gas fired. Nonetheless, without contemporaneous CO or C_2H_2 data, we can’t rule out this possibility so we have made a brief mention of it.

P9 Lines 11-12: The choice to employ a 3σ iterative cut was the outcome of much study and deliberation. To include documentation of all the supporting work (even in an appendix) would be excessive.

Fundamentally, we are just trying to get rid of “bad slopes”: slopes that are poorly constrained by the data or that are unduly influenced by a small subset of anomalous points. Bad slopes are pretty easy to identify visually, but with ~ 3700 6-hour high- and low-intake subsets, this is simply out of the question. When we actually look at some of the slopes with unphysical values (>0 or <-2), they have extreme values for many reasons: one point with very low O_2 but reasonable CO_2 (presumably an Oxzilla glitch), low variability in both O_2 and CO_2 (yielding a cluster of points with no clear correlation), etc. We tried to flag these slopes using statistical indicators of fit quality (χ^2 and statistical uncertainty in the slope) as well as scatter plots of slope vs. CO_2 range and slope vs. O_2 range but found nothing particularly useful. You are correct that periods with low CO_2 variability tend to have a wider range of slopes, but plenty of these data subsets actually have tight O_2 - CO_2 correlations. Furthermore, a cut on CO_2 range would not be entirely agnostic, as there does seem to be a weak correlation between CO_2 range and slope, and this correlation changes with day and night, summer and winter.

For these reasons, we adopted the iterative cut. As Figure 6 shows, there are significant non-gaussian tails even after it is employed so it’s clear the cut is not too draconian. While a discussion like the one above is inappropriate for the paper, we have added prose about the purpose of the cut and we have added values to Table 2 that clarify the impact of the cuts.

Page 9, lines 29-32: We agree that the flux of nitrogen from leaves to soil will have an influence on LBER. The question is one of timing and duration of disequilibrium. Even in the idealized scenario outlined in section 4.1.2, Severinghaus explicitly acknowledges that the system must eventually yield the same average exchange ratios for photosynthesis and respiration due to mass balance. Instead of his Fig. 4.1, imagine a pair of trees. One of them (shown in Fig. 4.1) is standing, and PQ for the tree is lower than OR for the soil. The other tree has fallen over and its trunk is decaying, with OR lower than PQ. In other words, to invoke different exchange ratios for day and night requires an ecosystem that (in this simplified example) only has standing trees. Given that Harvard Forest is measured to be a net carbon sink (i.e. a system with photosynthesis and respiration unequivocally out of balance) it is possible that Severinghaus's original Fig. 4.1 is a better analogue for Harvard Forest than the two-tree picture we suggested above, but it is also possible that this patch of forest is diverse enough in age and organismal composition, that it contains both standing and fallen trees (so to speak). We have changed the text to acknowledge these two possibilities.

In a more realistic analogy, there isn't one standing tree; there are several, each with a different age and a different elemental ratio. These represent the many pools into which carbon may be allocated (leaves, roots, stem wood, etc.) and their various carbon residence times. Again, to invoke different day/night exchange ratios requires the "standing trees" to be different from the "fallen trees".

Page 10, lines 4-8: This is an interesting question and led us to correspond with Rick Wehr. He pointed out that more recent data (Wehr et al 2016) show the cycle of respired ^{13}C is somewhat damped, relative to the photosynthetic cycle. This suggests that the Wehr and Trumbore work might be consistent if most respired material was synthesized only days before (thus, no visible lag in the seasonal cycle), but a substantial amount of respired carbon came from a much older pool with no seasonality in its ^{13}C signature (causing damping of the respired ^{13}C cycle). Wehr points out that Trumbore's values have no uncertainties, making it hard quantify (in)consistency.

This analysis is reinforced by an unusual ^{14}C -based study described by Trumbore et al. (Eos V.83 N.24, 2002) in which they say "Previous studies with deliberate ^{14}C labeling [Hanson et al., 2000] have shown that >90% of the added ^{14}C may be respired within a few days, with <10% allocated to longer-lived carbon pools such as leaves, roots, and wood".

We have added a brief discussion of these issues to the text.

Typographical comments and suggestions: All done except...

Page 1, Line 9: The two datasets in question are very different sizes. We have significantly fewer data in the "summer, low, day" subset, so precision is reduced. For this reason, we prefer to leave the presentation unchanged.

Page 8, line 2: We have stuck with 22.4 but acknowledged the potential problems with this value.

Atmospheric measurements of the terrestrial O₂:CO₂ exchange ratio of a mid-latitude forest

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Abstract.

Measurements of atmospheric O₂ have been used to quantify large-scale fluxes of carbon between the oceans, atmosphere and land since 1992 (Keeling and Shertz, 1992). With time, datasets have grown and estimates of fluxes have become more precise, but a key uncertainty in these calculations is the exchange ratio of O₂ and CO₂ associated with the net land carbon sink (α_B). We present measurements of atmospheric O₂ and CO₂ collected over a six-year period from a mixed deciduous forest in central Massachusetts, USA (42.537°N, 72.171°W). Using a differential fuel-cell based instrument for O₂ and a non-dispersive infrared analyzer for CO₂, we analyzed airstreams collected within and ~ 5 m above the forest canopy. Averaged over the entire period of record, we find these two species covary with a slope of -1.081 ± 0.007 moles of O₂ per mole of CO₂ (the mean and standard error of 6-hour periods). If we limit the data to values collected on summer days within the canopy, the slope is -1.03 ± 0.01 . These are the conditions in which biotic influences are most likely to dominate. This result is significantly different from value of -1.1 widely used in O₂-based calculations of the global carbon budget, suggesting the need for a deeper understanding of the exchange ratios of the various fluxes and pools comprising the net sink.

atmospheric O₂/N₂, oxidative ratio, terrestrial carbon sink

Copyright statement. TEXT

15 1 Introduction

Since the pioneering work of Keeling and Shertz (1992), measurements of the abundance of atmospheric O₂ and CO₂ have been used extensively for constraining fluxes of carbon to and from the land biosphere and the oceans. In a conceptual form, the budgets for atmospheric O₂ and CO₂ can be written

$$\frac{dO_2}{dt} = \alpha_{ff} f_{ff} + \alpha_B f_{land} + Z_{ocean} \quad (1)$$

$$20 \quad \frac{dCO_2}{dt} = f_{ff} + f_{land} + f_{ocean} \quad (2)$$

where f_{xx} is the flux of CO_2 from reservoir xx to the atmosphere, α_{ff} is the global average oxidative ratio of fossil fuels and α_B is the average oxidative ratio of the global net land carbon sink. Z_{ocean} describes the outgassing of O_2 from the ocean due to warming. Knowing f_{ff} and α_{ff} from industrial inventories, calculating Z_{ocean} from changes in ocean heat storage, and measuring the changes in O_2 and CO_2 , we can solve these two equations for f_{land} and f_{ocean} in terms of α_B (for a rigorous treatment of these equations, see for example Keeling and Manning (2014)). This O_2 -based method has become increasingly sophisticated with refinements in methodology, and increasingly precise with better instruments and ever-longer datasets (Keeling and Manning, 2014). Nonetheless, all estimates of global carbon fluxes based on atmospheric O_2 require a value for α_B .

The molar ratio of O_2 released to the atmosphere and CO_2 removed from it during photosynthesis (one contribution to α_B) will depend on the type of organic material being synthesized. This ratio is set by the relative amounts of carbon and hydrogen in the material, as well as the amount and source of any nitrogen included. Similarly, the ratio of CO_2 release and O_2 uptake during respiration depends on the composition of the material being respired and whether nitrogen is oxidized to nitrate. For example, synthesis and respiration of the simplest sugar (glucose: $\text{C}_6\text{H}_{12}\text{O}_6$) will have a flux ratio of 1.0. However, if plants are making shoots and other nitrogen-rich tissue, the oxidative ratio might range from 1.0 to 1.26, depending on the source of the nitrogen (nitrate vs. ammonium) (Bloom et al., 1989). Clearly, any value of α_B that is used in a large-scale calculation of carbon fluxes must represent the net imbalance of many processes in many ecosystems over long times.

One approach to this problem is to survey the elemental ratios of various biogenic materials. This was done by Severinghaus (1995); Randerson et al. (2006); Masiello et al. (2008); Hockaday et al. (2009); Worrall et al. (2013) and Gallagher et al. (2017). Based on a variety of techniques and a range of materials, estimates of α_B at present seem to be around 1.05 ± 0.1 (Keeling and Manning, 2014). Propagating this uncertainty to the global land-ocean partition of carbon uptake corresponds yields a value of $\pm 0.1 \text{ PgCyr}^{-1}$.

One method that can illuminate the processes responsible for α_B , and the one we have chosen, was first published by Seibt et al. (2004); Sturm et al. (2005) and Stephens et al. (2007). It uses direct measurements of O_2 and CO_2 changes in ambient air within, and close to, an ecosystem. Given sufficient precision and temporal resolution, the covariation of these two atmospheric components should reflect the exchange ratio for the particular mixture of fluxes to/from the ecosystem present during the study period. We refer to this quantity as the local biotic exchange ratio (LBER), recognizing that it is closely related to, yet distinct from, α_B . α_B may turn out to be close to measured values of LBER, but it is a distinct quantity by definition: α_B is the ratio of global net CO_2 and O_2 exchanges from all terrestrial biological processes.

The most groundbreaking feature of the work of Stephens et al. (2007) was the use of a newly-developed, continuous O_2 analyzer deployed at the study site, allowing precise, high-frequency measurements. These techniques have the potential to better characterize contributions to α_B as well as improve our understanding of the underlying physiology of the ecosystem generating the fluxes.

Following Stephens et al. (2007), we have used a fuel-cell-based differential O_2 analyzer and a non-dispersive infrared (NDIR) CO_2 analyzer to make quasi-continuous measurements of the air in and above the canopy at the Harvard Forest Environmental Measurement Site (Urbanski et al., 2007). In following sections of this manuscript, we begin with a description of the instrument and our methods of data collection and analysis. We then present data acquired between 2006 and 2013 and

make a first effort at interpreting these data in the context of both the ecosystem's process-specific ratios and the response of the ecosystem to environmental controls.

A note on units: Because O_2 is a major atmospheric constituent, its abundance is customarily reported in "per meg" defined as

$$\delta(O_2/N_2) \equiv \left(\frac{O_2/N_2}{O_2/N_2_{st}} - 1 \right) \times 10^6 \quad (3)$$

where O_2/N_2_{st} is O_2/N_2 for the standard gas on which the scale is based. This avoids complications arising from dilution effects (Keeling et al., 1998). Because our measurements are made using an instrument that reports differences in apparent mole fraction, we convert to per meg using Eq. 3 of Stephens et al. (2007). Finally, when working with exchange ratios, we convert per meg to "ppm equivalent" by multiplying per meg by 0.209500 (the O_2 mole fraction of atmospheric air) (Keeling et al., 1998). When using ppm equivalent for O_2 and $\mu\text{mol mol}^{-1}$ (or ppm) for CO_2 , the synthesis of glucose (described above) would increase O_2 and decrease CO_2 by an equal number of units, directly translating our measurements to the underlying stoichiometry, as desired.

2 Methods

In August 2005 we installed a measurement system at the Harvard Forest Environmental Measurement Site (EMS). Located at 42.5377°N , 72.1714°W and 340 m above sea level, we collected air from two intake tubes mounted on a tower 7.50 m and 29.0 m above the ground. The lower intake is within the forest canopy, the upper about 5 m above it. The tower and site are described in more detail by Urbanski et al. (2007). Our system is very similar to the one installed at WLEF in Wisconsin and documented very thoroughly by Stephens et al. (2007), so we focus primarily on differences from the WLEF instrument in our description here.

2.1 System design and operation

A schematic of the system is shown in Fig. 1. Unless otherwise mentioned, all wetted surfaces are stainless steel. At both points of collection, air is drawn in through downward-facing aspirated inlets (R.M. Young, Part #43408) to reduce thermal fractionation (Blaine et al., 2006). A nominal flow rate of $50 \text{ cm}^3 \text{ min}^{-1}$ is maintained at all times in both lines using pumps with viton diaphragms and PTFE valves (KNF Neuberger N05-ATI) in concert with mass flow controllers (MKS M100B). Tubing on the tower ($\frac{3}{8}'' O.D.$ Dekabon[®]) is routed into the instrument enclosure without local low spots to minimize buildup of condensed water. Within the building, water is removed from the airstream using a series of three coaxial stainless steel traps. The first is kept at 5°C with a Peltier cooler (M&C model ECP2000-SS), while the second and third traps are filled with borosilicate glass beads and immersed in a heat-transfer fluid (Syltherm-XLT[®]) held at -90°C using a mechanical refrigeration system (FTS VT490). Filters with a nominal pore size of $7 \mu\text{m}$ (Swagelok SS-6F-7) protect downstream equipment from particulates. Since our analyzers can only measure a single airstream, the high and low intake lines are alternately sent to

analysis or waste. The intake selector valve is a pneumatically actuated, crossover ball valve with PTFE packing (Swagelok SS-43YFS2), switched every 720 seconds.

Both the O₂ and CO₂ analyzers make differential measurements of the chosen stream of ambient air relative to a cylinder of dried air (the working tank, WT). Data are recorded at 1 Hz. Periodically, the ambient air stream is diverted to waste and the WT is analyzed against one of four calibration tanks: HS, LS, MF and LF. Details of the various tanks are given in Table 1. The frequency of the calibration runs was determined in part from pre-deployment tests in which we observed instrumental drift when running pairs of (sacrificial) tanks against one another. Other considerations include consumption of standard gases and lost time for atmospheric measurements.

The ambient air stream (high or low) or the calibration tank is chosen for analysis with a pair of electrically actuated 6-port selector valves (VICI EMT2SF6MWE and EMT2SDMWE). From these valves it passes through another mass flow controller, the NDIR CO₂ analyzer (Licor LI-7000), a 0.5 μm filter (Swagelok SS-6F-05), a needle-tipped metering valve (Swagelok SS-4BMG), a changeover valve assembly (four Numatics TM101V24C2 solenoid valves on a custom-built aluminum manifold), and the fuel-cell based O₂ analyzer (Sable Systems Oxzilla II with customized stainless steel internal plumbing), then exits through a mass-flow meter (Honeywell AWM3100V). Absolute pressures are recorded upstream of the CO₂ analyzer in three locations (Omega PX303-030A10V) and within the Oxzilla (see below).

The response of the Licor CO₂ analyzer depends on the pressure difference between the two cells. Thus, we use a differential manometer and electronically controlled metering valve with active feedback (MKS 223BD, 247D and 248) to ensure balanced pressures in the Licor.

The response of the Oxzilla O₂ analyzer depends on both pressure and flow differences between the cells. To provide sufficient control of these parameters, we have two metering valves upstream of the Oxzilla and one downstream. We use the metering valves (as well as the active pressure-difference control upstream of the Licor) to tune the system so that pressures and flows are as close to equal as possible regardless of the state of the changeover valve (see below). For this tuning, and subsequent monitoring of the system, we use flow data from the Honeywell meters and pressure data recorded within the Oxzilla immediately downstream of the cells. Prior to deployment, we modified the Oxzilla's internal plumbing and electronics so that pressure transducers (Honeywell SCX15N) were teed off of each of the outlets of the fuel cells. The two transducers are read out with the single high-precision digitizer built into the Oxzilla motherboard using a multiplexer to alternately select the transducer signal sent to the digitizer.

To further improve measurement precision of O₂, we follow Stephens et al. (2007) and use a changeover valve to alternate the fuel cells into which the two gas streams flow. This allows for a difference-of-differences calculation that eliminates cell-to-cell biases in the instrument. However, the elimination of bias occurs only if that bias varies little during the two measurement periods being considered. Achieving optimal precision would seem to require high frequency changeovers to reduce variation in the bias. However, each changeover is followed by a “dead time” when flows and instrument response need to settle. For a given dead time, more rapid changeovers mean fewer measurements are available for averaging, leading to a greater impact of random noise and thus, poorer precision.

We adopt a method for quantifying this trade-off and settling on an optimal changeover frequency that was developed by Keeling et al. (2004). We determined dead time for our instrument by analyzing tank air (relative to a working tank) with the changeover running very slowly. We then measured the time after each changeover required for O₂ values to settle to within 1 σ of the post-changeover equilibrium value. This “dead time” was 14 seconds for our instrument. We then determined the optimal changeover time by analyzing two tanks of air against each other with the changeover valve disabled. We processed these measurements as if the changeover valve were operating, calculating difference-of-difference values for various (hypothetical) changeover rates, imposing a 14 s dead time. This exercise indicated an optimal changeover rate of 24 s between switches.

2.2 Data reduction

2.2.1 Atmospheric data

10 In order to allow the instrument to achieve equilibrium after the selector valves have changed state (at the end of a calibration run), we discard the first 432 s of atmospheric data after such an actuation.

The remaining atmospheric data are naturally divided into 12-minute subsets by the switching of the intake selector between high and low intake lines. We discard the first 432 s of data after an intake switch to allow time for the instrument to re-equilibrate after the disruption in flows and pressures. The surviving data are processed slightly differently for O₂ and CO₂.

15 For O₂, we break the surviving 288 s of data into 24 s blocks defined by the cycling of the changeover valve. We average the 1 Hz data collected between 14 s and 24 s after the valve changes state. We then calculate difference-of-difference values for consecutive pairs of averages. This typically yields 6 values. These are in turn averaged, yielding a single O₂ value for the chosen intake height at a time corresponding to the middle of the interval of data used in the average.

For CO₂, we consider every 1 Hz CO₂ value collected at the same time as the 1 Hz O₂ values that contribute to the final O₂ mixing ratio for each of the 12-minute intake intervals. After the cuts for deadtime following valve actuations (gas selector, intake selector and changeover), there are 120 s of live data, so our CO₂ mixing ratio is an average of 120 measurements.

2.2.2 Calibration data

Roughly four times daily, atmospheric sampling is interrupted for 6-minute runs of calibration tanks. To more effectively use the information from these runs, we adopt a protocol that differs from our treatment of atmospheric data.

25 To flush stale gas from the lines and purge the regulators, air from the calibration tank is vented to waste for 6 minutes prior to measurement. Once the selector valves direct the calibration air to the analyzers, we consider all 6 minutes of CO₂ measurements. We look for a transition in the CO₂ value as tank air reaches the analyzer, displacing the atmospheric air that was in the optical cell. We find the time at which 70% of the eventual change in CO₂ has been achieved (the transition time).

30 To determine O₂ in these calibration runs, we calculate difference-of-differences values for each pair of changeover blocks as with our atmospheric data. Then, based on the timing of the transition in the CO₂ record, we average the last 3, 4, or 5 difference-of-difference values of the calibration run. The transition time typically occurs \sim 2.6 minutes after moving the

selector valve. In this case, we average 3 difference-of-difference pairs, corresponding to the last 2.4 min of data. This translates to averaging the last $\sim 40\%$ of each cal run (or $\sim 70\%$ of the post-transition data).

To determine CO_2 , we average all CO_2 values collected at the same times as the values that were used in the O_2 calculation.

Our standards are calibrated on the Scripps S2 O_2 scale (Keeling et al., 1998, 2007) and the NOAA/WMO CO_2 scale (Zhao and Tans, 2006), thanks to the generous assistance of these laboratories. Thus it is on these scales that we report our measurements.

2.3 System performance

While we have made great efforts to calibrate our measurements to internationally accepted CO_2 and O_2 scales using our suite of reference tanks (Table 1), we defer a detailed treatment of long-term measurement stability and accuracy for a later publication. Our focus in this manuscript is on the local biotic exchange ratio, as expressed in a series of short-term averages of CO_2 and O_2 covariation. Thus, our primary concern is instrumental precision and stability over periods of roughly 6 hours.

To determine this instrumental precision, we use runs of calibration tanks (see Sec. 2.2.2).

There are two possible approaches: We could use the observed scatter in the 3, 4 or 5 difference-of-difference values for O_2 that are calculated during a 6-min run of a calibration tank, and likewise for the 60, 80 or 100 CO_2 measurements retained during the same run. These are the values plotted in Fig. 2. We refer to this metric as “scatter on one tank” (SOOT).

Not surprisingly, the performance of the instrument varies over time. At times, we could link a loss of precision to problems with an individual analyzer, while on other occasions, we found water (liquid or solid) in the system to be the cause. The source of the major degradation in the precision of both analyzers that develops in late 2012 and persists until the end of our dataset remains under investigation.

An alternative approach, and the one that we adopt here, uses the observed scatter in the difference of paired standard tanks that were run immediately following one another. The calculation of the difference between the HS and LS tanks run in succession emulates the treatment of the atmospheric data by comparing one gas stream (HS in place of atmosphere) to another (LS in place of a “calibration curve”) without the involvement of the working tank. The scatter in the HS-LS difference is a conservative estimate of uncertainty in atmospheric data because it includes any variability that might be introduced by operation of the gas-selection valve, a valve that remains fixed in position during atmospheric sampling.

We find the the scatter in the HS-LS differences for 12-day periods is around ± 2 ppm equivalent for O_2 and $\pm 0.2 \mu\text{mol mol}^{-1}$ for CO_2 until the middle of 2008. From 2012 onward, the scatter was ± 9 ppm equivalent and $\pm 0.6 \mu\text{mol mol}^{-1}$.

Each HS and LS O_2 value was typically the average from 3 difference-of-difference pairs, while the CO_2 values were usually an average of 60 measurements. In contrast, a 12-minute atmospheric measurement on either the high or low intake is an average of 6 O_2 difference-of-difference pairs and 120 CO_2 measurements (Sec. 2.2.1). Thus, the atmospheric measurements should have uncertainties given by

$$\sigma_{\text{O}_2 \text{ atm}} = \frac{\sigma_{\text{O}_2 \text{ HS-LS}}}{\sqrt{2}} \frac{\sqrt{3}}{\sqrt{6}} \quad (4)$$

$$\sigma_{CO_2_{atm}} = \frac{\sigma_{CO_2_{HS-LS}}}{\sqrt{2}} \frac{\sqrt{60}}{\sqrt{120}} \quad (5)$$

Based on these equations, we believe that the atmospheric measurements carry uncertainties of $\sigma_{O_2_{atm}} = 1$ ppm equivalent and $\sigma_{CO_2_{atm}} = 0.1 \mu\text{mol mol}^{-1}$ prior to mid-2010. For measurements beginning in 2012, the uncertainties were $\sigma_{O_2_{atm}} = 4.5$ ppm equivalent and $\sigma_{CO_2_{atm}} = 0.3 \mu\text{mol mol}^{-1}$.

- 5 Our choice to work backwards from the observed scatter in HS-LS pairs may not be the optimal technique for extracting uncertainties, but we are confident it yields O_2 and CO_2 errors that bear the correct *relative* size. Thus, we use the value listed above when calculating slopes in our Deming regressions (Sec. 3).

3 Observations

The measurements we present here commenced on September 27, 2006 and continued until January 9, 2013. The record is approximately continuous until the middle of 2010, when a series of hardware failures, due in part to a lightning strike, caused an extended hiatus in data collection. Measurements resumed in May of 2012 and continued without interruption through the end of that year, when failure of the oxygen analyzer led to a suspension of data collection.

Measurements of O_2 and CO_2 for the entire period of record are shown in Fig. 3, while Fig. 4 gives an example of a single day's cycle. The values shown here were derived from the raw data using the calibration runs and the algorithm described in Sec. 2.2. They represent our best estimates of the O_2 and CO_2 content of the atmosphere on the Scripps S2 and NOAA/WMO scales, respectively. In these figures we can see a very gradual rise in CO_2 and drop in O_2 (mainly due to fossil fuel combustion), a strong diurnal cycle with inverse variation in CO_2 and O_2 , and a great deal of variability on a range of time scales. Results from the upper and lower intakes are quite similar, but not identical.

While these records contain a wealth of information, our particular focus is on the covariation of O_2 and CO_2 . We expect to see different results from the high and low intakes, particularly at night. The lower intake is more likely to show stronger influences of soil and canopy fluxes, particularly at night when the atmosphere is more stable and the surface layer is more likely to be decoupled from the planetary boundary layer. For this reason, we separate the data from the high and low intakes, create day and night subsets for each intake, plot O_2 vs. CO_2 for each subset, perform a linear regression, and take LBER from the slope of the fit.

The 6-hour subsets making up each plot are 9am - 3pm (day) and 10pm - 4am (night). These intervals are chosen to be times of maximal and minimal coupling (day and night, respectively) between the air within and above the canopy. We define the intervals based on the climatological diurnal cycle in the friction velocity u^* measured at the EMS tower (Munger and Wofsy, 2017). Subsets with fewer than three measurements are ignored. Because uncertainties in both species are comparable, we use a Deming regression (Deming, 2011). Furthermore, to reduce any influence of non-Gaussian tails in the dataset, we perform an iterative fit, in which we calculate the standard deviation of the residuals to the fit (σ_{res}) and discard any points lying more than $4\sigma_{res}$ from the fit. Of the 3760 fits, only 50 had any outliers and only 11 of these required more than one iteration to converge. Representative plots and fits are shown in Fig. 5.

In total, we have slope values for 3760 6-hr subsets. While the great majority of slope values are close to -1.0 and are normally distributed (see Fig. 6), a small number are implausibly large (up to +2035.9) or small (down to -102.6). Since we seek a representative slope value for each high/low day/night subset, we choose to limit the impact of these outliers by performing an iterative calculation of the means. We calculate a mean and standard deviation, discard all slopes more than 3σ from the mean and iterate to convergence. Table 2 shows the results of these calculations, as well as sensitivity studies varying the length of the day and night intervals and the tightness of the cut in our iterative averaging.

4 The relationship between local $O_2:CO_2$ ratios and the local biotic exchange ratio (LBER)

In its simplest form, the variability in O_2 and CO_2 in the continental interior is an end-member mixing problem: Fossil fuel fluxes will give slopes ranging from -1.17 (coal) to -2.0 (methane) (Keeling, 1988; Steinbach et al., 2011), while biogenic fluxes will give slopes closer to -1.0. Thus, we expect air masses exposed to both types of fluxes will evolve with intermediate slopes.

Within this conceptual framework, a successful effort to extract LBER from the data depends on the relative sizes of the biogenic and anthropogenic fluxes. Only at times when photosynthesis and respiration dominate the land-air fluxes will observed slopes be close to LBER.

To test these assertions, we first use a Lagrangian transport model to estimate the region over which surface fluxes influence signals at our sampling site. We then use a simple 1-dimensional box model to establish the validity of the end-member mixing framework. Finally, we estimate the relative size of the biogenic and anthropogenic contributions.

4.1 The 6-hour provenance of atmospheric variability

We use a Lagrangian transport model to estimate the surface regions to which air parcels are exposed during the six hours prior to their arrival at the EMS tower. Specifically, we created back-trajectories for a range of dates and times, day and night, across seasons and years using the HYSPLIT Lagrangian transport model (NOAA, 2018) forced with NAM12 meteorology. We generated many sets of six 6-hour back-trajectories originating hourly during the daytime or nighttime intervals to determine the provenance of the air we analyze at Harvard Forest. An example of one such set of trajectories is shown in Fig. 7.

In general, the 6-hour provenance ranges from 50 km to 200 km, with little spatial variation over the 6-hour interval. Given the location of our study site, a ~ 100 km provenance, primarily from the west, means the air we analyze has travelled over a mostly forested landscape, occasionally passing small urban centers. This analysis is consistent with the work of Gerbig et al. (2006) who found a 10-fold (or greater) fall-off in the influence of fluxes more than ~ 100 km from the study site.

4.2 “Mixing” of slopes

To test the applicability of an end-member approach, we considered the simplest model with two contributors: generic fossil fuel, with an exchange ratio of -1.4, and biotic activity, with an exchange ratio of -1.05. These characterized the stoichiometry of surface fluxes used in a 1-dimensional box model, in which parcels of air travel forward in time in 12-minute steps, passing

over a fictitious landscape with forested, suburban and urban components. In 30 of these time steps the parcels exchange CO₂ and O₂ with the underlying sources and sinks, changing the composition of the parcel according to

$$C_{end} = C_{start} + \frac{F \cdot t}{h} \frac{22.4}{1000} \quad (6)$$

where C is the mixing ratio of O₂ or CO₂ in $\mu\text{mol mol}^{-1}$, F is the flux from the landscape underneath the box in $\mu\text{mol m}^{-2} \text{s}^{-1}$, t is the time step in seconds, h is the height of the box in m , and the factors of 22.4 (the molar volume) and 1000 convert moles to liters and liters to m^3 , respectively. For h we use one half of the planetary boundary layer (PBL) height. We spin up the model for 24 hours and then record O₂ and CO₂ concentrations of consecutive parcels as they “arrive” in their final time steps over the same 6-hour day and night intervals used for the observations.

The CO₂ surface fluxes vary with time of day and day of year and are based on observations at Harvard Forest (Munger and Wofsy, 2017) and in urban (Bergeron and Strachan, 2011; Velasco and Roth, 2010; Ward et al., 2015) and suburban (Bergeron and Strachan, 2011) settings. The O₂ surface fluxes were derived from the CO₂ fluxes by assuming exchange ratios of 1.4 for fossil fuel combustion and 1.05 for biospheric activity. The PBL values are climatological hourly average values, calculated separately for summer and winter from the NAM12 dataset (NOAA, 2018). Within the STILT model (Gerbig et al., 2006), only air within the lower half of the PBL is influenced directly by surface fluxes. Air above this height is closely matched to the composition of the free troposphere. Because we use the STILT model to infer the provenance of air we analyze, we define our box height as half the height of the PBL so that the two models (STILT and our 1-dim box model) are conceptually consistent. We performed sensitivity studies, varying the landscape composition (ranging from 100% forest to 100% urban), surface flux magnitude (values characteristic of winter and summer seasons) and PBL height (25-75% of the NAM12 values). In all cases, the CO₂ and O₂ values of the parcels entering the model were held constant, ensuring that any variability predicted by the model was a result of the fluxes from the landscape over which the parcels travelled in the previous 6 hours and/or changes in PBL height.

We emphasize that the purpose of the 1-D model is to explore the validity of the end-member mixing approach. In particular, we ask whether a mix of sources will relate simply to a mix of slopes, despite the fluxes and the box height having their own diel cycles.

In all but a few special circumstances, the model predicts slopes that are completely consistent with the specified admixture of the source fluxes over which the air parcels have traveled. Exceptions arise when the covariation of the boundary layer height and diel cycles in the fluxes conspire to bias slopes by a few percent. Results of various sensitivity tests and a discussion of these exceptions can be found in Table 5.1 of Conley (2018), along with a more complete description of the model and its forcing fluxes.

The 1-D model results give us confidence that slopes close to -1.0 are a closer representation of the LBER, while more negative slopes have ever larger contributions from fossil fuel combustion. Though reassuring, we acknowledge the model results are not definitive due to the many simplifications involved. More sophisticated studies are needed to give a truly robust confirmation of the end-member mixing framework.

Aside from the slopes, the full range of O_2 and CO_2 variability predicted by the model is in the general range of our observations. Because of the crude nature of our model, this result is somewhat surprising. Our values for the box height and the molar volume are almost certainly wrong, so the data-model agreement may be coincidental.

5 Discussion of observation

5 Adopting the approach outlined above, we interpret slopes of our plots as averages of LBER (a number near -1.0) and external fossil-fuel influences (numbers ranges from -1.17 to -2.0), weighted by the relative contributions of these fluxes in the Harvard Forest “footprint.”

Taking all data together, using 6-hour intervals, we find a slope of -1.081 ± 0.007 (see Table 2). This value is a little higher than the oxidative ratio of organic material calculated by Worrall et al. (2013) and at the lower end of the range of possible values for the global exchange ratio quoted by Severinghaus (1995).

10 Ideally, our dataset would not include “poor” slopes – those that are weakly constrained by the data or unduly influenced by a small number of points. We found the most effective tool for excluding such slopes was a 3σ iterative cut. In practice, this cut only eliminates 6-hr intervals with slopes that are positive, or very negative (< -2.0). Inspection of a subset of the fits failing the cut shows they are indeed poor fits, as defined above. While the cut does not catch all poor slopes, we believe the slopes that are cut are poor ones.

15 We emphasize that this all-inclusive average slope is not a direct measurement of the LBER. Instead, it is a measure of the LBER mixed with the exchange ratio of fossil fuel combustion.

We also considered 2-week aggregation periods (instead of 6 hours). These slopes average -1.19 ± 0.02 . This is consistent with nearly equal contributions from fossil and biotic fluxes, as expected for an aggregation period with a provenance encompassing the entire continental United States (Gerbig et al., 2006).

5.1 Information from subsets of data

Considering particular subsets of the data (Table 2) may provide insight into the processes determining observed slope values. Thus, we categorize our 6-hour slopes by day and night, high and low, winter (Dec, Jan, Feb) and summer (Jun, Jul, Aug).

25 We begin with data aggregated in broad categories. Each group’s average slope, calculated using the 3σ cut described above, is shown in Fig. 8. These groupings convey some unequivocal messages: Summer slopes are less negative than winter slopes. Day slopes are less negative than night slopes. Low-intake slopes are less negative than high-intake. Daytime data from the low intake are least negative of all.

We also consider narrower categories, each with a unique set of slopes. The average slopes for each category are shown in Fig. 9. For these averages, we impose a 2.5σ iterative cut because a 3σ cut fails to reject several unphysical fits (with slopes > 2.0) in the winter subsets, badly biasing the mean.

The most striking feature of this plot is the large scatter in the winter data. Even with the 2.5σ iterative cut, the slopes within a category range widely and are difficult to interpret. In contrast, the summer data comprise much tighter groups and have well-defined mean values, showing the same day/night and high/low relationships as the full dataset.

Explaining these patterns is challenging. Possible factors determining the observed slope include the magnitudes of the fluxes, the dynamics of the atmosphere, and variation of the biospheric exchange ratio over time.

5.1.1 The relative size of fluxes

The relative weighting for combustion and biogenic influence in the slope will depend in part on the magnitudes of the respective fluxes across the region. Our best estimate of combustion fluxes comes from Sargent et al. (2018), who use an inverse analysis of CO_2 concentration data to infer a value of $0.8 \text{ kg C m}^{-2} \text{ y}^{-1}$ (i.e. $2.4 \mu\text{mol C m}^{-2} \text{ s}^{-1}$) for Boston and its suburbs. We can relate this to Harvard Forest using the ACES high-resolution emission inventory developed specifically for the Northeastern US (Gately and Hutyra, 2017). ACES shows a strong decrease in anthropogenic emissions moving from Boston to Harvard Forest and beyond to the west and north, with the area around the tower showing annual mean emissions closer to $0.3 \mu\text{mol C m}^{-2} \text{ s}^{-1}$.

For the biogenic fluxes, we use eddy-flux measurements of net ecosystem exchange (NEE) at Harvard Forest (Munger and Wofsy, 2017). In the summer, the daytime mean NEE at Harvard Forest is of order 10 or $20 \mu\text{mol m}^{-2} \text{ s}^{-1}$ for coniferous and deciduous forest, respectively, and around $5 \mu\text{mol m}^{-2} \text{ s}^{-1}$ at night. Winter NEE averages range from 2 to $3 \mu\text{mol m}^{-2} \text{ s}^{-1}$.

Collectively, these data show the regionally averaged biogenic influence on CO_2 (and by extension O_2) is substantially larger than the fossil fuel source during the growing season, and dwarfs it during daytime. In winter months, anthropogenic CO_2 emissions have, at most, a magnitude comparable to the biogenic flux. Further evidence that combustion sources are a minor player at Harvard Forest during the growing season comes from the observation by Potosnak et al. (1999) that CO_2 and CO are correlated in winter but not summer. Finally, we note that the modeling work of Gerbig et al. (2006) shows short-term variability in CO_2 is dominated by the predicted influence of vegetation.

While US EPA's inventory (Agency, 2017) shows that the primary upwind sector (west-northwest) (Moody et al., 1998) is essentially free of power plants for 100km or more, because we lack contemporaneous measurements of CO (a tracer of combustion), it remains a possibility that plumes of power plant exhaust might occasionally influence our measurements.

In summary, based on the estimated magnitudes of the biotic and combustion fluxes, we expect the summer daytime observations of the $\text{O}_2:\text{CO}_2$ slopes to be most strongly linked to the LBER. Furthermore, the link between winter variability in CO_2 and CO observed by Potosnak et al. (1999) offers one possible explanation for the much greater scatter we see in our winter data.

Comparing day and night data, fossil fuel influence may be more apparent at night when biotic fluxes are reduced (Munger and Wofsy, 2017) but fossil fuel combustion falls only slightly (Bergeron and Strachan, 2011; Velasco and Roth, 2010; Ward et al., 2015).

5.1.2 Atmospheric Dynamics

With the diel cycle of insolation, the planetary boundary layer cycles between nocturnal stability and diurnal convective overturning. This is clearly evident in the climatological diel cycle of u^* at Harvard Forest (Munger and Wofsy, 2017). Based on these data, we defined our 6-hour daytime period to capture full photosynthetic activity and maximal vertical mixing, while our night interval captures pure respiration and maximal stratification. These dynamics suggest that air at both of our intakes would be closer in composition to air aloft during the middle of the day. Thus, daytime slopes might show more influence from distant fluxes including more fossil fuel combustion (and thus, be more negative).

In addition, we expect nocturnal stability to trap fluxes from soil respiration close to the ground, amplifying their impact, particularly on the lower intake (Munger, 1996; Urbanski et al., 2007). This will lead to biologically dominated nocturnal slopes. Assuming the respiration-driven slope is closer to -1.0 than fossil fuel, we would expect to find the least negative slopes at night.

The fact that we see less negative slopes during the day than at night (-1.02 ± 0.01 vs. -1.12 ± 0.01) suggests that either other factors are more important (Sec. 5.1.1 and Sec. 5.1.3) or this picture of atmospheric motion is oversimplified.

Another implication of diurnal convection and nocturnal stratification would be similar behaviour of the high and low intakes during daylight hours and greater differences at night. We do see a suggestion of this in our summer data, when the high-low difference appears to be smaller during the day (-0.015 ± 0.02 vs. -0.03 ± 0.01), but uncertainties are substantial. We also expect any differences driven by stratification to be reduced during the winter. Unfortunately, our winter data are too variable to meaningfully test this expectation (-0.1 ± 0.1 and -0.16 ± 0.08).

Yet another possibility is a daytime temperature inversion within the forest canopy. This arises because air is warmed from above by incoming sunlight striking leaves (Jacobs et al., 1992). However, profiles of CO_2 and O_2 on the tower show that at Harvard Forest these inversions are often eroded by gusts from above penetrating the canopy and usually only persist in the first meter or two above ground level.

Regardless of time of day or year, velocities within the canopy will be frictionally reduced. This might suggest an enhancement of local signals (presumably with less fossil influence and thus, less negative slopes) on the lower intake. This is very much consistent with our observations: With the 3σ cut, low and high slopes are -1.04 ± 0.01 and -1.11 ± 0.01 , respectively. Results are very similar with the 2.5σ cut.

Finally, we note that u^* varies seasonally as well as daily. Not only is u^* roughly twice as high in the winter as the summer, but it is far more variable. This may explain why the slopes in the winter months exhibit so much variation.

5.1.3 Possible variation in the biotic exchange ratio

Another possible driver of variability in slopes is variation in the biotic exchange ratio over time. For example, nocturnal plant respiration might consume molecules richer in nitrogen with higher oxidative ratios than the molecules being produced during photosynthesis (Severinghaus, 1995). Even in the complete absence of dynamical influences or fossil fuel contamination, this would lead to more negative slopes at night.

This scenario is built on a violation of mass balance that cannot persist indefinitely. Eventually, the nitrogen-poor material that was synthesized above ground must also be respired, changing atmospheric O_2 and CO_2 with slopes closer to -1.0.

5 The viability of this explanation for our more negative nocturnal slopes becomes a question of equilibrium. In illustrative (and highly simplified) terms: Is the forest as a whole out of balance, engaged in a long-term accumulation of nitrogen-poor, refractory trunk wood and a steady diversion of nitrogen into labile leaves and rootlets? Or is the forest in balance, with as many tree trunks being decomposed as are being synthesized?

10 Evidence pointing to a balanced forest are the isotopic measurements by Wehr and Saleska (2015) and Wehr et al. (2016), which suggest much of Harvard Forest's respired carbon was assimilated no more than a week or two earlier. On the other hand, Trumbore (2000) used radiocarbon measurements to show that the mean age of respired soil carbon is 3 years. These two studies are not entirely incompatible if Trumbore's mean age is comprised of, for example, 60% very young carbon with the balance coming from much older pools.


At this point, we are not in a position to say whether the day-night difference in slopes is due to a forest that is out of equilibrium, or whether we are measuring respiration of the full range of substrates and something else is responsible for the day-night difference in slopes.

15 5.2 Temporal variability in slopes

Our data are sufficiently abundant that we can examine the values of slopes as a function of time on both interannual and seasonal time scales. Figure 10 shows that one year looks much like the next, but stacked years reveal a seasonal variation in the slopes.

20 The seasonal cycle in slopes may be due to seasonal changes in LBER itself. Trees synthesize many different compounds throughout the year (Seibt et al., 2004), each with their own oxidative ratios (Bloom et al., 1989). However, direct measurements of leaves and tree-rings indicate oxidative ratios in these tissues appear invariant through the year (Gallagher et al., 2017).

25 On the other hand, the total amount of biological activity unquestionably varies seasonally, while the fossil fuel contribution remains roughly constant. As mentioned above, earlier work at Harvard Forest using carbon monoxide and acetylene (Potosnak et al., 1999) shows clearly that fossil fuel combustion is a driver of CO_2 variability in the winter, but is overwhelmed by biotic signals during the summer. Thus, the simple seasonal shift in the biotic/fossil-fuel balance is a likely explanation for some, if not all, of the seasonal cycle in slopes.

Spectral analysis (using Lomb-Scargle periodograms to accommodate our irregularly spaced data (Press et al., 2007)) of various subsets of the data shows some evidence of the seasonal cycle mentioned above. It appears to be mostly in the night/high-intake subset. There is also a suggestion of multi-year variation (0.6 cycles/year) in the day-low dataset and 3-week periodicity in the day-high and night-low data. Identical spectral analyses of environmental variables measured at Harvard Forest (temperature, net ecosystem exchange, ecosystem respiration, photosynthetically active radiation and u^*) all show highly significant seasonal and diel cycles, as expected. 

The three-week cycle in the day-high and night-low records is perplexing. No such cycle appears in the environmental variables listed above. Thus, our working hypothesis is that the 3-week cycles are an experimental artifact resulting from the timing of runs of calibration tanks and swapping of working tanks.

5.3 Selecting periods of heightened biogenic signal

5 Given the competition between biogenic and anthropogenic influences in our observations, we searched for periods when the biogenic contribution might be most dominant. We selected times when the fluxes in the immediate neighborhood were likely to dominate. Our assumption was that background variation (i.e. compositional variability due to influences beyond the ~ 100 km provenance of our measurements) is likely to carry signals of continental-scale gradients with more substantial fossil fuel influences embedded therein.

10 One approach we considered was based on u^* . We hypothesized that slopes calculated when u^* was low were more likely to be biogenic, since the air at our intakes was exchanging minimally with the overlying boundary layer. However, a scatter plot of slopes as a function of u^* showed no correlation. Similarly, slope values were insensitive to discrete cuts on average u^* ($< 40 \text{ cm s}^{-1}$), maximum u^* ($< 50 \text{ cm s}^{-1}$) and the scatter in u^* ($\sigma_{u^*} < 10 \text{ cm s}^{-1}$).

We also explored the possibility of a link between slopes and the ranges of CO_2 and O_2 during the 6-hour periods. Again,
15 no correlation was apparent, whether with discrete cuts during summer and winter periods separately, or with scatter plots of slope vs. range.

We tested the hypothesis that summer periods of extended high barometric pressure with languid anticyclonic circulation might be dominated by local fluxes. Again, we observed no correlation between slope and barometric pressure, whether with discrete cuts or with scatter plots.

20 Finally, we looked for correlations between slopes and wind speed, direction and variability. We found none, even when winds were from the relatively polluted sector to the southwest.

In short, relatively simple indicators of reduced advective transport show no correlation with shallower slopes. This insensitivity of our result adds support for our model of a geographically modest footprint comprised of a landscape in which biogenic fluxes are much stronger than anthropogenic ones.

25 6 Conclusions

Having measured changes in atmospheric CO_2 and O_2 in a mixed deciduous, mid-latitude forest with a precision of a few $\mu\text{mol mol}^{-1}$, we find that these species covary with an all-data average molar ratio over 6-hr periods of -1.081 ± 0.007 ($\text{O}_2 : \text{CO}_2$). In the absence of fossil-fuel influences, our measured ratios reflect the local signature of photosynthesis and respiration (LBER). If our data represent a mix of forest and fossil fuel fluxes, -1.081 is a lower (most negative) limit on
30 LBER. Adopting a two-end-member mixing model, the best value would be taken from the subset of the data with the smallest fossil fuel contribution and presumably, the shallowest measured slope. Restricting our dataset to daytime values from the low

intake in summer (when we expect local influences to be maximal due to vigorous biotic activity and modest coupling with the air above the canopy), we find a slope of -1.03 ± 0.01 .

Our values are consistent with measurements in other temperate forests. The consistency of elemental ratios in the work of Worrall et al. (2013) across a range of study sites suggests the Harvard Forest result may have broad applicability. Furthermore, basic constraints in plant physiology argue against wildly divergent values across species (Björkman and Demmig, 1987). Nonetheless, similar studies at other sites are required before concluding a value very close to 1.0 is globally representative.

Also, it is still possible that gross photosynthetic and respiratory exchanges could exhibit $O_2 : CO_2$ ratios individually that differ from the (much smaller) net uptake of CO_2 and release of O_2 . For this reason it is important to track O_2 in process-based ecosystem models, enabling predictions of both LBER and α_B while matching observations.

Even with this qualification, our measurements provide valuable insight into the nature of O_2 and CO_2 exchange in a temperate forest. They also suggest that the best value for α_B in calculations partitioning global carbon fluxes may be lower than 1.1, the value widely used to date (Keeling and Manning, 2014). At the least, studies that rely on α_B should explore sensitivity to values as low as 1.0.

If our results do indeed match the ratio of the global net carbon sink, the use of 1.03 for α_B leads to a 7% increase in terrestrial carbon uptake and an equivalent reduction in oceanic carbon storage (see Eq. 8 of Keeling and Manning (2014)). While this adjustment is within the uncertainties on these terms in the global carbon budget, it would nonetheless be an important correction.

Data availability. Battle M, Munger W. 2018. Atmospheric Oxygen and Carbon Dioxide at Harvard Forest EMS Tower since 2006. Harvard Forest Data Archive: HF306.

Author contributions. MB conceived of the experiment. MB built and operated the hardware and wrote software with assistance from JWM, ES, RP, RH, ZD, JS, JW, KG, SS and JC. MB and MC performed data analysis. MB wrote the manuscript with assistance from JWM and MC.

Competing interests. The authors have no competing interests to declare.

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References

- Agency, U. S. E. P.: Greenhouse Gas Reporting Program, <https://www.epa.gov/ghgreporting>, 2017.
- Bergeron, O. and Strachan, I. B.: CO₂ sources and sinks in urban and suburban areas of a northern mid-latitude city, *Atmospheric Environment*, 45, 1564–1573, 2011.
- 5 Björkman, O. and Demmig, B.: Photon yield of O₂ evolution and chlorophyll fluorescence characteristics at 77K among vascular plants of diverse origins, *Planta*, 170, 489–504, <https://doi.org/10.1007/BF00402983>, 1987.
- Blaine, T., Keeling, R., and Paplawsky, W.: An improved inlet for precisely measuring the atmospheric Ar/N₂ ratio, *Atmospheric Chemistry and Physics*, 6, 1181–1184, 2006.
- Bloom, A., Caldwell, R., Finazzo, J., Warner, R., and Weissbart, J.: Oxygen and carbon dioxide fluxes from barley shoots depend on nitrate
10 assimilation, *Plant Physiology*, 91, 352–356, 1989.
- Conley, M. M.: Using atmospheric O₂ and CO₂ to determine the stoichiometry of photosynthesis and respiration in a temperate forest, Bachelor of Arts thesis, Bowdoin College, <https://digitalcommons.bowdoin.edu/honorsprojects/87/>, 2018.
- Deming, W. E.: Statistical adjustment of data, Dover Publications, New York, 2011.
- Gallagher, M. E., Liljestrand, F. L., Hockaday, W. C., and Masiello, C. A.: Plant species, not climate, controls aboveground
15 biomass O₂:CO₂ exchange ratios in deciduous and coniferous ecosystems, *Journal of Geophysical Research: Biogeosciences*, 122, 2017JG003 847, 2017.
- Gately, C. K. and Hutyrá, L. R.: Large Uncertainties in Urban-Scale Carbon Emissions, *Journal of Geophysical Research, Atmospheres*, 122, 11 242–11 260, 2017.
- Gerbig, C., Lin, J. C., Munger, J. W., and Wofsy, S. C.: What can tracer observations in the continental boundary layer tell us about surface-
20 atmosphere fluxes?, *Atmospheric Chemistry and Physics*, 6, 539–554, 2006.
- Hall, B.: Calculation of expanded uncertainties for CO₂, https://www.esrl.noaa.gov/gmd/ccl/ccl_uncertainties_co2.html, 2017.
- Hockaday, W., Masiello, C., Randerson, J., Smernik, R., Baldock, J., Chadwick, O., and Harden, J.: Measurement of soil carbon oxidation state and oxidative ratio by ¹³C nuclear magnetic resonance, *Journal of Geophysical Research: Biogeosciences*, 114, <https://doi.org/10.1029/2008JG000803>, 2009.
- 25 Jacobs, A. F. G., van Boxel, J. H., and Shaw, R. H.: The dependence of canopy layer turbulence on within-canopy thermal stratification, *Agricultural and Forest Meteorology*, 58, 247–256, 1992.
- Keeling, R. and Manning, A.: Studies of Recent Changes in Atmospheric O₂ Content, in: *Treatise on Geochemistry*, pp. 385–404, Elsevier Inc., 2 edn., <https://doi.org/10.1016/B978-0-08-095975-7.00420-4>, 2014.
- Keeling, R., Manning, A., McEvoy, E., and Shertz, S.: Methods for measuring changes in atmospheric O₂ concentration and their application
30 in southern hemisphere air, *Journal of Geophysical Research*, 103, 3381–3397, 1998.
- Keeling, R., Paplawsky, W., and Cox, A.: On the long-term stability of reference gases for atmospheric O₂/N₂ and CO₂ measurements, *Tellus Series B – Chemical and Physical Meteorology*, 59, 3–14, 2007.
- Keeling, R. F.: Development of an Interferometric Oxygen Analyzer for Precise Measurement of the Atmospheric O₂ Mole Fraction, Ph.D. thesis, Harvard University, http://bluemoon.ucsd.edu/publications/ralph/34_PhDthesis.pdf, 1988.
- 35 Keeling, R. F. and Shertz, S. R.: Seasonal and interannual variations in atmospheric oxygen and implications for the global carbon cycle, *Nature*, 358, 723–727, 1992.

- Keeling, R. F., Blaine, T., Paplawsky, B., Katz, L., Atwood, C., and Brockwell, T.: Measurements of changes in atmospheric Ar/N₂ ratio using a rapid-switching, single-capillary mass spectrometer system, *Tellus*, 56B, 322–338, 2004.
- Masiello, C., Gallagher, M., Randerson, J., Deco, R., and Chadwick, O.: Evaluating two experimental approaches for measuring ecosystem carbon oxidation state and oxidative ratio, *Journal of Geophysical Research: Biogeosciences*, 113, <https://doi.org/10.1029/2007JG000534>, 2008.
- Moody, J., Munger, J., Goldstein, A., Jacobs, D., and Wofsy, S.: Harvard Forest regional-scale air mass composition by Patterns in Atmospheric Transport History (PATH), *Journal of Geophysical Research*, 103, 13 181–13 194, 1998.
- Munger, J. W.: Atmospheric deposition of reactive nitrogen oxides and ozone in a temperate deciduous forest and a subarctic woodland I. Measurements and mechanisms, *Journal of Geophysical Research Atmospheres*, 101, 12 639–12 657, 1996.
- 10 Munger, W. and Wofsy, S.: Canopy-Atmosphere Exchange of Carbon, Water and Energy at Harvard Forest EMS Tower since 1991, Harvard Forest Data Archive: HF004, 2017.
- NOAA: HYSPLIT - Hybrid Single Particle Lagrangian Integrated Trajectory model, <https://ready.arl.noaa.gov/HYSPLIT.php>, 2018.
- Potosnak, M. J., Wofsy, S. C., Denning, A. S., Conway, T. J., Munger, J. W., and Barnes, D. H.: Influence of biotic exchange and combustion sources on atmospheric CO₂ concentrations in New England from observations at a forest flux tower, *Journal of Geophysical Research*, 15 104, 9561–9569, 1999.
- Press, W. H., Teukolsky, S. A., Vetterling, W. T., and Flannery, B. P.: 13.8 Spectral Analysis of Unevenly Sampled Data, in: *Numerical Recipes: The Art of Scientific Computing*, pp. 685–692, Cambridge University Press, Cambridge, third edn., 2007.
- Randerson, J., Masiello, C., Still, C., Rahn, T., Poorter, H., and Field, C.: Is carbon within the global terrestrial biosphere becoming more oxidized? Implications for trends in atmospheric O₂, *Global Change Biology*, 12, 260–271, 20 104, <https://doi.org/10.1111/j.1365-2486.2006.01099.x>, 2006.
- Sargent, M., Barrera, Y., Nehrkorn, T., Hutyra, L., Gatley, C., Jones, T., McKain, K., Sweeney, C., Hegarty, J., Hardiman, B., and Wofsy, S.: Anthropogenic and Biogenic CO₂ fluxes in the Boston urban region, *Proceedings of the National Academy of Sciences*, 115, 7491–7496, 2018.
- Seibt, U., Brand, W., Heimann, M., Lloyd, J., Severinghaus, J., and Wingate, L.: Observations of O₂: CO₂ exchange ratios during ecosystem gas exchange, *Global Biogeochemical Cycles*, 18, 1–18, <https://doi.org/10.1029/2004GB002242>, 2004.
- 25 Severinghaus, J. P.: Studies of the Terrestrial O₂ and Carbon Cycles in Sand Dune Gases and in Biosphere 2, Ph.D. thesis, Columbia University, 1995.
- Steinbach, J., Gerbig, C., Rödenbeck, C., Karstens, U., Minejima, C., and Mukai, H.: The CO₂ release and Oxygen uptake from Fossil Fuel Emission Estimate (COFFEE) dataset: effects from varying oxidative ratios, *Atmospheric Chemistry and Physics*, 11, 6855–6870, 30 <https://doi.org/10.5194/acp-11-6855-2011>, 2011.
- Stephens, B., Bakwin, P., Tans, P., Teclaw, R., and Baumann, D.: Application of a differential fuel-cell analyzer for measuring atmospheric oxygen variations, *Journal of Atmospheric and Oceanic Technology*, 24, 82–94, <https://doi.org/10.1175/JTECH1959.1>, 2007.
- Sturm, P., Leuenberger, M., and Schmidt, M.: Atmospheric O₂, CO₂ and δ¹³C observations from the remote sites Jungfraujoch, Switzerland, and Puy de Dôme, France, *Geophysical Research Letters*, 32, 1–4, <https://doi.org/10.1029/2005GL023304>, 2005.
- 35 Trumbore, S.: Age of Soil Organic Matter and Soil Respiration: Radiocarbon Constraints on Belowground C Dynamics, *Ecological Applications*, 10, 399–411, 2000.

- Urbanski, S. P., Barford, C. C., Wofsy, S. C., Kucharik, C. J., Pyle, E. H., Budney, J., McKain, K., Fitzjarrald, D. R., Czikowsky, M. J., and Munger, J. W.: Factors controlling CO₂ exchange on time scales from hourly to decadal at Harvard Forest, *Journal of Geophysical Research-Biogeosciences*, 112, 2007.
- 5 Velasco, E. and Roth, M.: Cities as net sources of CO₂ : Review of atmospheric CO₂ exchange in urban environments measured by eddy covariance technique, *Geography Compass*, 4, 1238–1259, 2010.
- Ward, H. C., Kotthaus, S., Grimmond, C. S. B., Bjarkegren, A., Wilkinson, M., Morrison, W. T. J., Evans, J. G., Morrison, J. I. L., and Iamarino, M.: Effects of urban density on carbon dioxide exchanges: Observations of dense urban, suburban and woodland areas of southern England, *Environmental Pollution*, 198, 186–200, 2015.
- 10 Wehr, R. and Saleska, S.: An improved isotopic method for partitioning net ecosystem-atmosphere CO₂ exchange, *Agricultural and Forest Meteorology*, 214-215, 515–531, 2015.
- Wehr, R., Munger, J., McManus, J., Nelson, D., Zahniser, M., Davidson, E., Wofsy, S., and Saleska, S.: Seasonality of temperate forest photosynthesis and daytime respiration, *Nature*, 534, 680–683, <https://doi.org/10.1038/nature17966>, 2016.
- Worrall, F., Clay, G. D., Masiello, C. A., and Mynheer, G.: Estimating the oxidative ratio of the global terrestrial biosphere carbon, *Biogeochemistry*, 115, 23–32, 2013.
- 15 Zhao, C. and Tans, P.: Estimating uncertainty of the WMO mole fraction scale for carbon dioxide in air, *Journal of Geophysical Research Atmospheres*, 111, <https://doi.org/10.1029/2005JD006003>, 2006.

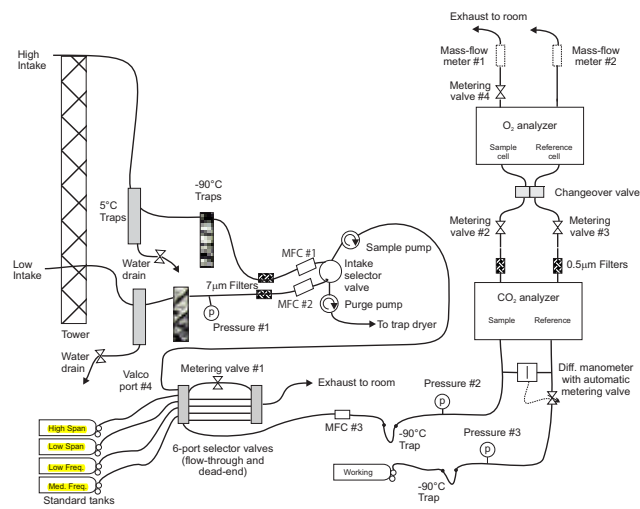


Figure 1. Schematic diagram of the instrumentation installed at Harvard Forest.

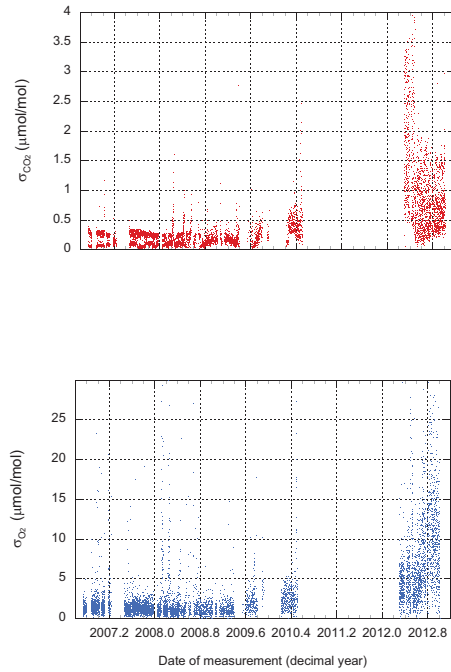


Figure 2. The scatter (standard deviation from the mean) of routine CO₂ and O₂ measurements of calibration tanks. Values plotted here apply to individual CO₂ measurements (1 Hz) and single difference-of-difference O₂ values (1/48s).

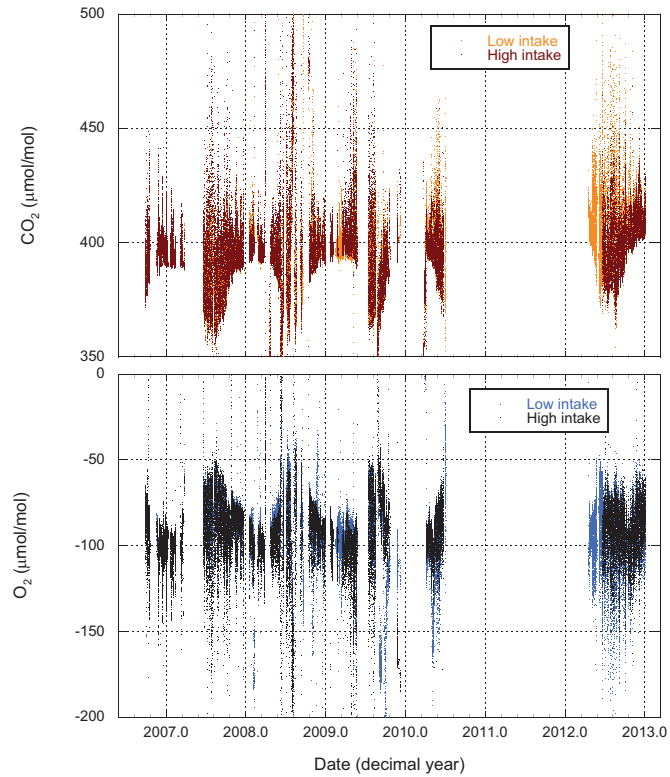


Figure 3. CO₂ and O₂ measured from both high and low intakes at Harvard Forest. The data have been processed (Sec. 2.2) and are presented on the WMO and S2 scales. Each point (spaced 11 min apart) represents the average atmospheric composition for a 4.8 min period.

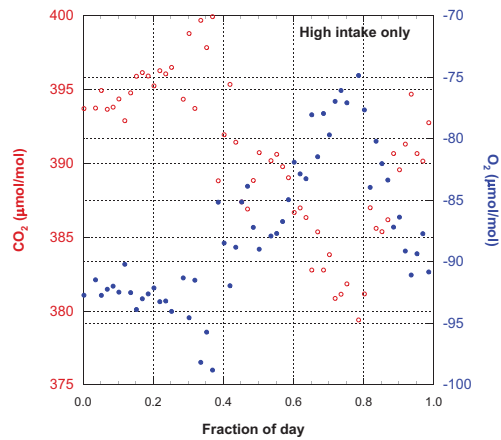


Figure 4. A detail from Fig. 3: CO₂ and O₂ measured from the high intake on Sept. 29, 2006. This is simply one example of the diel cycle of these species. Note the strict inverse variation in O₂ and CO₂, even when there are large deviations from a smooth temporal evolution.

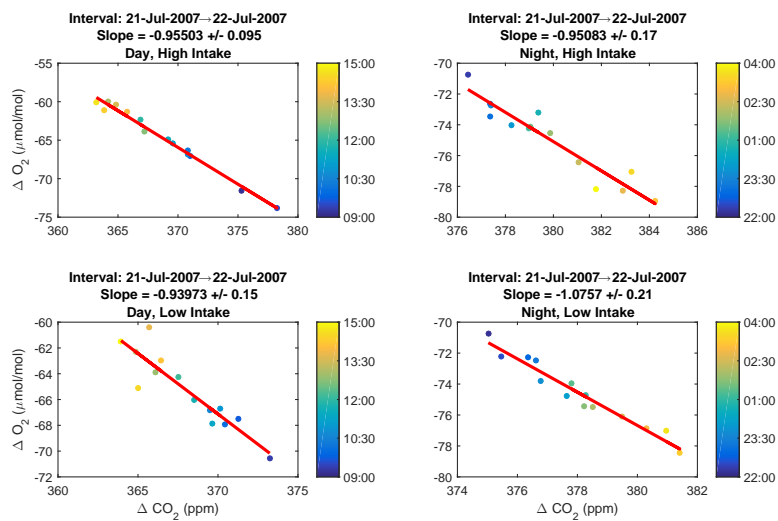


Figure 5. Representative examples of slope-plots: data collected during day and night periods from the high and low intakes on July 21-22, 2007. The colors of the points indicate the time of collection within the 6-hr intervals, as keyed in the color bar to the right of each plot. Errors on the slopes are purely statistical.

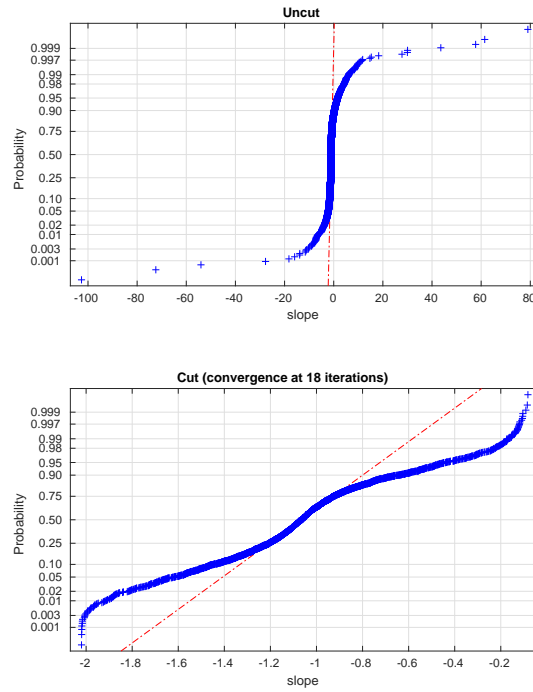


Figure 6. Probability plots of slopes for the entire dataset, before and after a 3σ iterative cut. The dash-dot lines indicate the shape expected for a purely Gaussian distribution. In the interests of clarity, we omit five points with extremely large or small slopes from the upper panel. See Table 2 for statistical details of these distributions.

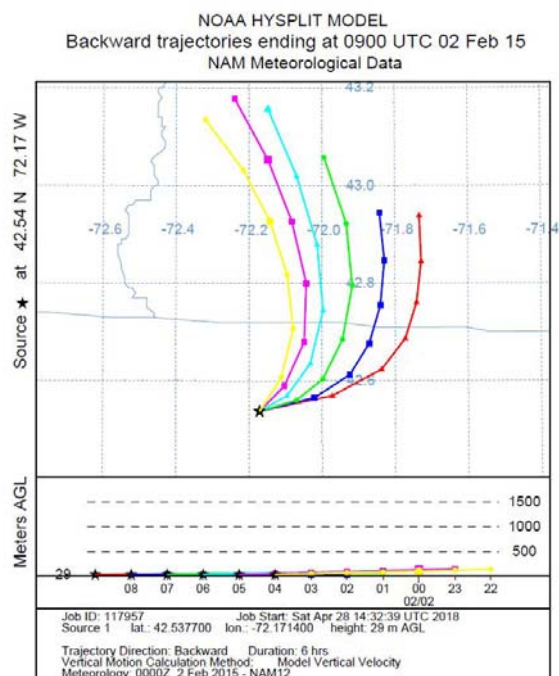


Figure 7. An example of 6-hour back-trajectories for Harvard Forest on Feb. 2, 2015. Trajectories start hourly between 10pm and 4am EST, corresponding to the nighttime interval for slope plots. State lines for New Hampshire, Vermont and Massachusetts are in gray. On this particular night, the source region was fairly constant, so changes in O_2 and CO_2 are relatively likely to be primarily due to local influences.

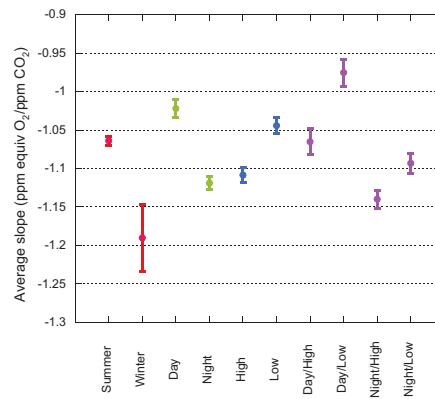


Figure 8. Average slopes for various subsets of the data. Error bars show the standard error on the mean. Subsets within each group (i.e. sharing a color) are non-overlapping.

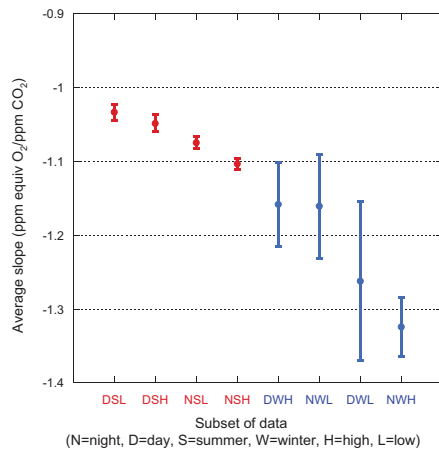


Figure 9. Average slopes for various non-overlapping subsets of the data. Error bars show the standard error on the mean. Winter data are shown in blue; summer in red. Categories are ordered by increasingly negative mean values for convenience.

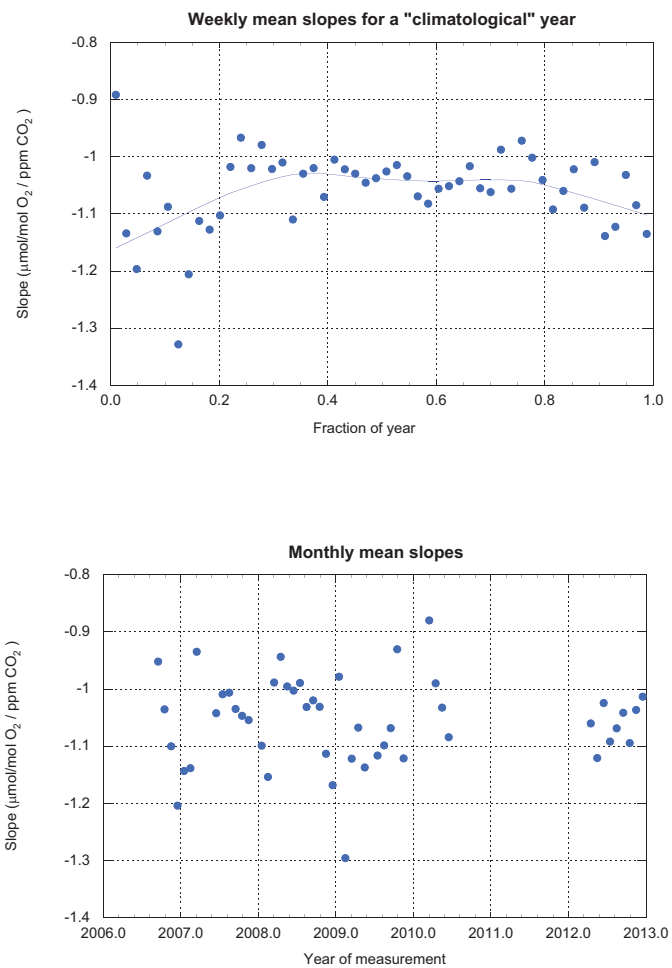


Figure 10. Upper panel: Slopes (averaged weekly) for a climatological year. Lower panel: Slopes (averaged monthly) for the entire period of study. The gray line in the upper panel is a locally weighted least-squares fit with 46% smoothing. It is simply intended as an objective guide to the eye.

Table 1. Nominal composition of calibration tanks. Actual values vary slightly as cylinders are used and refilled. The O₂ values quoted here were measured in the lab of R. Keeling, while the CO₂ values were measured by the CCGG lab of NOAA-ESRL-GMD. Based on replicate measurements, the O₂ values are typically determined to within roughly ±1.5 per meg on the S2 scale, while the CO₂ values on the WMO X2007 scale are known to about ±0.2 μmol mol⁻¹ (Hall, 2017). The cylinders are Luxfer N265 uncoated aluminum, with 51-14B regulators from Scott Specialty Gases (now Air Liquide). Tanks and regulators are stored horizontally in an insulated box to minimize thermal and gravitational fractionation.

Tank name	O ₂ (per meg on S2 scale)	CO ₂ (μmol mol ⁻¹)	Analysis frequency
High Span (HS)	+125	350	4/day
Low Span (LS)	-850	525	4/day
Medium Freq. (MF)	-360	375	1/3 days
Low Freq. (LF)	-470	390	1/week

Table 2. Table of average values for all slopes and four data subsets. σ is 1 standard deviation, SEM is the standard error of the mean (σ/\sqrt{n}), and n is the number of slope values surviving the iterative cut.

Interval length and cut	Data set	mean	σ	SEM	n
6-hour (3σ)	All data	-1.081	0.38	0.007	2950
	Day, high intake	-1.07	0.46	0.02	774
	Day, low intake	-0.98	0.46	0.02	684
	Night, high intake	-1.14	0.33	0.01	801
	Night, low intake	-1.09	0.36	0.01	725
4-hour (3σ)	All data	-1.071	0.66	0.008	3070
	Day, high intake	-1.04	0.58	0.02	751
	Day, low intake	-0.94	0.66	0.03	679
	Night, high intake	-1.15	0.34	0.01	808
	Night, low intake	-1.10	0.37	0.01	725
6-hour (4σ)	All data	-1.005	0.77	0.01	3347
	Day, high intake	-1.03	0.95	0.03	877
	Day, low intake	-0.85	0.92	0.03	782
	Night, high intake	-1.11	0.63	0.02	898
	Night, low intake	-1.01	0.71	0.02	833
6-hour (no cut)	All data	5.93	205	3	3760
	6 outliers cut	³¹ -0.05	37	0.6	3754