

Interactive comment on “Source apportionment of atmospheric aerosol in a marine dusty environment by Ionic/composition Mass Balance (IMB)” by João Cardoso et al.

Anonymous Referee #1

Received and published: 2 April 2018

This study applied a method called IMB for source apportionment in Cape Verde, and compared the results with those from PMF. Essentially, the IMB method is nothing more than linear combinations of weighted concentrations of PM components. It might be acceptable to identify some factors with known and relatively fixed ratios between species, e.g. Na and Cl in sea spray and metals and oxygen in dust, however, it is of limited use when handling sources with similar components but varied weights.

Answer: We do not agree with the referee. Of course that the IMB is not a full solution to Source Apportion the atmospheric aerosol. However, partial mass balance has been used in the past in a multitude of situations to partially apportion the atmospheric aerosol with success. Here we extend this methodology further in order to apportion the totality of the aerosol mass, obtaining aerosol fractions that fit completely in the total PM mass and that give interesting and important information concerning sources and formation processes. This information is complementary to information obtained from PMF and is quite useful to test the accuracy of the PMF obtained solution. The “European Guide on Air Pollution Source Apportionment with Receptor Models” (<http://publications.jrc.ec.europa.eu/repository/bitstream/JRC83309/lb-na-26-080-en-n.pdf>) clearly alleged in page 19 that “Receptor Models can be used in combination with independent methodologies (e.g. emission inventories, chemical transport models (CTMs)) to achieve more robust estimations by mutual validation of the outputs”. In this paper we wanted to show that IMB may help in the obtaining of a more real and accurate solution than that the one obtained only with PMF. This will be especially important when the various source emission impacts in the receptor covary in time, situation in which methodologies such as PMF are unable of completely separating the different sources of Secondary Inorganic Aerosol (SIA), as it was the present case. We would like to emphasize that in the review of ambient particulate matter source apportionment results obtained using receptor models achieved by Belis et al. (Atmospheric Environment 69 (2013), 94-108) it was stressed that “SIA” was the strongest source for PM mass concentrations over Europe. These authors also declared that “SIA” contributions increased significantly when PM<sub>10</sub> and PM<sub>2.5</sub> concentrations, respectively were above the normative limit values. However, in most of the studies that were reviewed “SIA” was mainly composed of ammonium- sulphate and nitrate and only one third of them reported sulphate and nitrate contributions separately. It should also be noted that SIA levels thus determined could be underestimated since other secondary inorganic contributions were included in the classical “Sea/Road Salt” and the “Crustal/Mineral Dust” categories (Belis et al., Atmospheric Environment 69 (2013), 94-108; Atmospheric Environment 85 (2014), 275-276). From these results it can be concluded that in general terms RMs currently used were not able to provide a rather-full discrimination of different secondary inorganic contributions to the PM mass registered at a given receptor site.

The factor of “noncarbonate Carbonaceous elemental and organic matter” showed exactly the limitation of this IMB method, as OC and EC may come from multiple anthropogenic sources but this method failed to separate them. OC is additionally contributed by secondary process which was also not able to identify by IMB. If performing well, PMF should be able to identify the multiple sources for OC and EC, as evidenced in many other places in the world.

Answer: We agree with the referee in relation to the limitations of the IMB method to completely separate and identify all the sources responsible for the carbonaceous matter, because these sources contain other substances besides carbonaceous material. But we continue to disagree with the referee about the capability of the IMB method to provide useful information concerning carbonaceous sources and formation processes. In the present case we did not invest much in separating the non-carbonated, carbonaceous component into fractions, because concentrations are quite low and, for EC, very near the limit of detection of the system analysis capabilities (very low concentrations of EC in a matrix with quite high concentrations of interfering colored dust). In the present case, the total contribution of non-carbonate carbonaceous matter to PM<sub>10</sub> is only of the order of 2%. However in another situation where we are applying IMB (concerning urban pollution), where contribution of carbonaceous matter is of the order of 50%, we could separate the carbonaceous matter into biomass burning, primary fossil fuel emission by cars and secondary formation processes. For that, we used levoglucosan measurements, edge lines for EC, versus OC versus fine K, versus levoglucosan and known ratios taken from specialized literature. In this urban data the biomass burning mass contribution compared very well between IMB and PMF (results to be submitted for publication, soon). This is a demonstration of the capabilities of IMB in “more complex” situations.

The fact that the IBM results happen to be consistent with PMF ones is that the PM contributing sources in Cape Verde is quite simple, which cannot sufficiently justify that this method is also applicable in other places with complicated contributing sources. In contrast, PMF has seen successful applications around the world with distinct environmental conditions. Therefore, I consider the approach applied in this study had significant drawback, and the limited scientific significance of this study does not meet the standard of Atmospheric Chemistry and Physics.

Answer: Part of the answer to this commentary has been given in the two previous answers. Here we will only complement the previous exposition. First, we want to stress that we do not have the intention of showing that the IMB should substitute the PMF, neither that it is better. We defend that IMB is a good methodology that extends and deepens frequently used methodologies in the past, which by being able of explaining the total PM mass is therefore more constrained and secured than previous incomplete attempts and that is complementary and useful to evaluate PM sources and formation processes in conjunction with other employed methods, such as PMF. The present evaluated aerosol is simple in what concerns the number of source contributions, taking into account classic air pollution considerations. But it is complex in which concerns collinearity of concentration variations because anthropogenic industrial and transport pollution are originated from the same directions than the sources of dust. Also, dust is emitted with different compositions in different parts of Africa and is important to know the dust contributions from each region. For telling the truth, neither the IMB, nor the PMF, were capable of fully

discriminating the regional origins and contributions of the African dust. As demonstrated in the text now added to the manuscript, in the present situation the PMF could not completely separate even sources such as Sea-salt and Dust, because of the common composition of dust and sea salt, concerning major ions. In many occasions PMF calculated zero sea salt contribution (impossible for a place in the middle of the ocean); in other occasions PMF considered important contributions of Al and Si in the sea salt component. So, this is not a “quite simple” situation.

Anonymous Referee #2

Received and published: 19 March 2018

Some improvements have been made in this revised version and I recommend publication in ACP. The paper reported the development of mass balance models for source apportionment of aerosol particles and the models were compared with PMF model. Results showed good agreements between the two models. I have following questions about the ionic mass balance (IMB) model as highlighted below.

(1) Is that possible to compare the mass balanced model developed in the present study with previous works, such as Malm et al., 1994 (JGR, 99, 1347), and other mass balance model thereby to further demonstrate the usefulness of new model.

**Answer:** The reference in consideration was added to the paper and several sentences were added to the text comparing IMB with previously published Mass Balance attempts.

(2) I would suggest to add some discussions about the robustness of the new IMB model C1as compared with PMF.

**Answer:** Text is added to the manuscript where the subject is discussed.

(3) Authors mentioned uncertainties in multivariate methods, such as PMF, are there any uncertainties in IMB model developed and applied in this study?

**Answer:** We have added a subsection in the manuscript where uncertainties for the IMB method are presented and discussed

Anonymous Referee #3

Received and published: 28 March 2018

This paper analyzed elemental and some ion data in PM<sub>10</sub> acquired from Cape Verde, an island west of the African Continent, with the attempt to determine the contributions from Saharan dust, sea salt, and their derivatives due to atmospheric transformation. Their approach is mainly based on ion balance, assuming known crustal and sea water chemical composition and the sequence of cation/anion neutralization. The mass and ion closure result from the approach generally make sense, especially when accounting for the residual water content. However, uncertainties were not provided for each contribution estimate, as PMF and other receptor models usually did. In many cases, the authors picked a middle value from a range of possible ratios, such as the water/soluble dust ratio, Fe/Na<sup>+</sup> ratio, Mg<sup>2+</sup>ss/Na<sup>+</sup>ss, ratio, etc. to carry out their calculation. Is it possible to propagate uncertainties in these assumptions throughout the calculation and give an overall uncertainty estimate in Table 3? The uncertainties should be compared with those from PMF based on the bootstrapping or DISP methods.

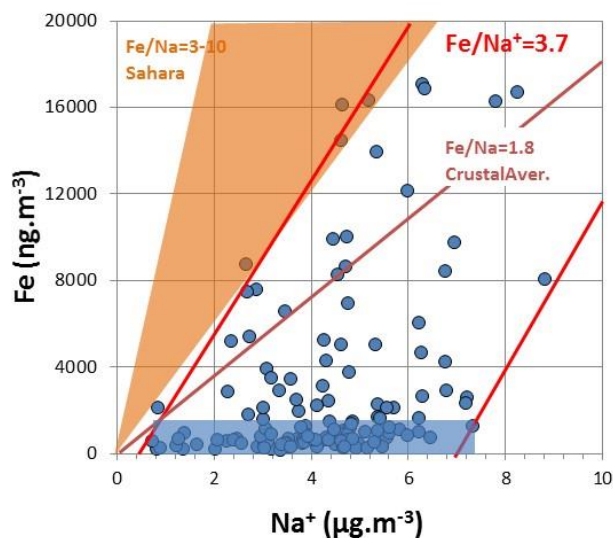
[Answer: We have added a subsection in the manuscript where uncertainties for the IMB method are presented and discussed.](#)

In fact, the dust Fe/Na<sup>+</sup> ratio (3.7) determined from the edge line (Figure 2) is based on only 3 points and the ratio falls at the low end of possible range of Fe/Na ratio for Sahara dust. It is nearly impossible to find a period with zero sea salt contribution at the island, and therefore using the edge line to determine the dust Na<sup>+</sup> fraction is risky. At the very least, uncertainty in this method should be given and propagated into all subsequent calculations.

[Answer:](#)

[Answer- In this point we do not agree with the referee. This methodology is much better than the known alternative, which is to attribute all Na<sup>+</sup> to sea salt production.](#)

[In reality the edge line is not based in 3 points only \(see Figure\).](#)



If we use an estimation taking into account predictable measuring analytical errors, the Fe/Na<sup>+</sup> edge ratio is based in 10 points. As important, the edge line intercepts the x axis at Na<sup>+</sup> levels within the range of values (represented by the blue rectangle) referring to periods without significant dust intrusions, in accordance with the fact that there is always sea salt spray in this marine atmosphere (The subject is further discussed in the new added sub-section 4.5).

When the authors compare IMB and PMF source contribution, they only compare the average contributions for the three periods. It is also meaningful to compare: 1) correlation of respective source contributions determined by the two methods across individual samples and 2) chemical composition of corresponding sources/classes, particularly the sea salt and dust sources, to examine whether the authors' assumptions in IMB, such as the aforementioned Fe/Na<sup>+</sup> ratio, are consistent with PMF. These comparisons should be presented and discussed explicitly.

Answer: New figures with individual comparisons between IMB and PMF (Figure 6a, b, c and d): Comparisons were discussed in added text, showing the capabilities of IMB to source apportion the aerosol.

Additional comments:

Page 10, Line 16. What is the range of this F factor? Can this be propagated into the uncertainty estimate?

Answer: The subject is discussed in new sub-section 4.5.

Page 10, Line 33. Does K<sup>+</sup> (and perhaps Cl<sup>-</sup>) generated from biomass burning need to be considered in the IMB for the region?

Answer: The amounts of possible bio-mass burning material are so small by comparison with other sources that any tentative to detail this source is mostly speculative.

Page 13, Line 12-19. How does the assumption of average inorganic and organic water growth factor influence the mass closure, and the contribution of each source to PM10? Can the range be estimated?

Answer: [The subject is discussed in new added 4.5 sub-section.](#)

# Source apportionment of atmospheric aerosol in a marine dusty environment by Ionic/composition Mass Balance (IMB)

João Cardoso<sup>1,2</sup>, Susana M. Almeida<sup>3</sup>, Teresa Nunes<sup>1</sup>, Marina Almeida-Silva<sup>3</sup>, Mário Cerqueira<sup>1</sup>, Célia Alves<sup>1</sup>, Fernando Rocha<sup>4</sup>, Paula Chaves<sup>3</sup>, Miguel Reis<sup>3</sup>, Pedro Salvador<sup>5</sup>, Begoña Artiñano<sup>5</sup>, Casimiro Pio<sup>1</sup>

<sup>1</sup>CESAM & Dep. Environ, Aveiro University, Aveiro, 3810-193, Portugal

<sup>2</sup>Cape Verde University, Praia, 279, Cape Verde

<sup>3</sup>C<sup>2</sup>TN, Instituto Superior Técnico, Lisbon University, Bobadela, 2695-066, Portugal.

<sup>4</sup>Geobiotec & Dep. Geosciences, Aveiro University, Aveiro, 3810-193, Portugal

10 <sup>5</sup>Environ Dep, CIEMAT, Madrid, 28040, Spain

*Correspondence to:* Casimiro Pio (casimiro@ua.pt)

**Abstract.** PM<sub>10</sub> aerosol was sampled in Santiago, the largest island of Cape Verde, for one year, and analysed for elements, ions and carbonaceous material. Very high levels of dust were measured during the winter months, as a result of the direct transport of dust plumes from the African continent. Ionic and Mass Balances (IMB) were applied to the analysed compounds, permitting the determination of 6-7 different processes and source contributions to the aerosol loading: insoluble and soluble dust, sea-salt, carbonaceous material and secondary inorganic compounds resulting from the reaction of acidic precursors with ammonia, sea-salt and dust. The mass balance could be closed by the consideration and estimation of sorbed water that constituted 20-30% of the aerosol mass. The balance methodology was compared with Positive Matrix Factorization (PMF), giving comparable results throughout the sampling campaign. This gives confidence in the capability of both methods, which are complementary, for the source apportionment of aerosol particles.

## 1 Introduction

The atmospheric aerosol has an important role in atmospheric physics and chemistry (Lohmann and Feichter, 2005; Poschl, 2005) and significant impacts on climate (Buseck and Pósfai, 1999; Ramanatham et al., 2001) and human health (Pope, 2000; Brunekreef and Fosberg 2005; Tobias et al., 2011).

25 Atmospheric aerosol is both the result of gas-to-particle transformation from natural, or anthropogenically, induced precursors and of direct emissions from the Earth's surface by the action of wind on soil and water. Gas-to-particle transformation gives fundamentally fine (sub-micrometre) particles, while wind-induced mechanic processes on the planet's surface originate mostly coarse particles, as sea-salt over the oceans, or soil dust over deserts and other bare soil areas (Seinfeld and Pandis, 1998). On a global scale, sea-spray and dust are highly dominant, in terms of suspended mass and regions affected, by comparison with other aerosol sources (Raes et al., 2000; Tanaka and Chiba, 2006). Although mostly natural, coarse aerosol

particles associated with dust emissions are frequently affected by anthropogenic activities such as industry, transportation and agricultural practices, at urban and regional/hemispheric scales (Almeida et al., 2006a; Ginoux et al., 2012).

It is important to know quantitatively the sources of atmospheric aerosols in order to correctly implement strategies and measures to control and reduce atmospheric particulate pollution and its effects on nature and humankind. There is an array of methodologies to evaluate the impact of sources and precursors on atmospheric particulate loading that range from emission, dispersion and transport modelling, to source apportionment techniques (Blanchard, 1999). Source apportionment techniques use information on aerosol atmospheric composition and concentrations, at one or several locations, to infer quantitatively the sources and processes responsible for the particulate loading at the receptor (Almeida et al., 2006b; Belis et al., 2013).

In source apportionment, when the number and composition of the sources are unknown, multivariate analysis, based on particulate composition variability at the receptor(s) as a function of time, is a very common and useful methodology to quantitatively evaluate the main sources of atmospheric particulate matter (Ashbaugh et al., 1984; Henry et al., 2004; Hopke, 1985). Presently the most used multivariate analysis methodology is Positive Matrix Factorization, (PMF), because it allows the discrimination of only positive values in both source profiles and contributions (Paatero and Tapper, 1994; Paatero, 1999; Reff et al., 2007; Amato et al., 2016).

Source apportionment multivariate methodologies permit frequently to identify the impact of the majority of direct sources and gas-to-particle conversion processes and their variability in time, during the aerosol measured period. If associated with statistical backward trajectory analysis the method also permits the determination of source regions during regional and, principally, long-range transport (Salvador et al., 2016).

Multivariate methods, although very useful, are not perfect and have uncertainties resulting from collinearity of sources, the evolution of composition during transport, etc., that need to be detected and estimated (Belis et al., 2013). When a large number of compounds and elements is determined by chemical analysis of aerosol samples, an alternative methodology can be used to infer the total aerosol composition, which takes into account that the total aerosol mass is the sum of the mass of the individual components. Also from the chemical analysis, it is possible to inter-compare analysed anions and cations, which have to obey the principle of electro-neutrality. From the mass and ionic composition, it is frequently viable to infer quantitatively the origin of the aerosol, because many of the analysed constituents are tracers of specific sources.

Ionic and Mass Balances (IMB) rely mostly on the direct measurement of aerosol constituents and therefore are less affected by indemonstrable assumptions, as it happens in the assignment of the number of factors and their identities, in multivariate methods, such as PMF (Belis et al., 2014). Mass balance has been frequently applied in the past but mostly as a screening tool (Watson et al., 2002). However, if properly applied, ionic and mass balances have the potential to correctly perform the source apportionment of atmospheric aerosol. We would like to emphasize that the “European Guide on Air Pollution Source Apportionment with Receptor Models” (Belis et al., 2014) exhorts to use Receptor Models in combination with independent methodologies to achieve more robust estimations by mutual validation of the outputs. Our objective in this paper is thus, to develop and apply a detailed ionic and mass balance to aerosol particles in a dusty marine environment, demonstrating the



capability of this methodology to determine the aerosol sources with an accuracy as good as that of the most developed multivariate methods, such as PMF.

## 2 Mass balance methodologies

Composition and mass balance is feasible when the main components of the aerosol sources, such as soil elements, sea-salt constituents, inorganic water soluble ions and carbonaceous mass, are measured and quantified (Sciare et al., 2005; Guinot et al, 2007; Grigoratos et al., 2014; Genga et al., 2017). Even when there is a thorough quantification of aerosol constituents, it is not often possible to apportion more than 70 to 80% of measured aerosol total mass, because important elemental constituents of particulate matter, such as oxygen, in water, soil and organic carbon, are not usually analysed ([Malm et al., 1994](#), Andrews et al., 2000; Rees et al., 2004; Perrino et al., 2013; Grigoratos et al., 2014). Oxygen is the most abundant element in the Earth's crust constituting on average 47% of the continental crust mass (Wedepohl, 1995).

For mass balance purposes, the determination of soil contribution is, usually, inferred from the analysis of the major soil elements measured in the aerosol samples (Si, Al, Fe, Mn, Ti, etc.), taking into account the presence of their oxides:

$$\text{Soil Dust Mass} = \text{SiO}_2 + \text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3 + \text{MnO} + \text{TiO}_2 + \text{etc.} \quad \text{Equation 1}$$

Depending on the completeness of the soil elemental analysis and the composition knowledge of the source soils, it is possible to adapt the above equation ~~in order~~ to better apportion the soil mass by mass balance (Formenti et al., 2001; Eldred, 2003; Andrews et al., 2000, and references there in):

$$\text{Soil Dust Mass} = F(2.14\text{Si} + 1.89\text{Al} + 1.43\text{Fe} + 1.39\text{Mn} + 1.67\text{Ti} + \text{etc.}), \quad \text{Equation 2}$$

where F is a multiplying factor that takes into account the unmeasured material (such as elements and the presence of hydrated water) in the soil dust.

The sea-salt contribution is evaluated by considering that emitted sea-salt spray has the composition of salt in sea-water. A possible exception is chloride that frequently appears in lower ratios due to sea-salt spray interaction with atmospheric acids, such as  $\text{HNO}_3$ ,  $\text{H}_2\text{SO}_4$ , or  $\text{SO}_2$ , which results in the evaporation of  $\text{Cl}^-$ , as  $\text{HCl}$ . The  $\text{Cl}^-$  in the particulate phase can be, partially, or totally, substituted by  $\text{NO}_3^-$ , or  $\text{SO}_4^{2-}$ , in the form of secondary sodium and magnesium nitrates and sulphates (Pio and Lopes, 1998). A similar reaction can take place between these atmospheric strong acids and dust, resulting, for example, in the partial or total vaporisation of carbonates, as  $\text{CO}_2$ , and the formation of secondary calcium nitrates and sulphates ([Pio et al., 1994](#); Goodman et al., 2000).

Soil carbonates are part of the carbonaceous aerosol. As they are, infrequently, analysed, in source apportionment they have to be approximately inferred from calcium and magnesium measurements. In dusty environments, the measurement of carbonates is important to permit a more correct composition/mass balance source apportionment.

Another component of the carbonaceous aerosol is the organic mass. This component is usually calculated from the measurement of organic carbon, by applying a multiplication factor to take into account other unmeasured elements such as nitrogen, sulphur and, principally, oxygen. Factors ranging from 1.2 to 2.3 have been employed for this purpose (Countess et

al., 1980; Japar et al., 1984; Rogge et al., 1993a; Rogge et al., 1993b; Sempere and Kawamura, 1994; Russel, 2003; Chen and Yu, 2007; El-Zanan et al., 2009). The highest values are commonly used in sites affected by biomass burning emissions, rich in sugars and organic acids, or away from emission sources, because under these conditions the precursor organic material had plenty of time to be strongly oxidised (Turpin and Lim, 2001; Sciare et al. 2005; Ervens et al., 2011). Genga et al., (2017) used  
5 variable values between 1.8 and 2.1, depending on the direction of the wind, to best fit the mass closure process, in a Mediterranean Port city.

Water is a common and important component of atmospheric aerosol that may constitute up to 20% of the total PM mass (Canepari et al., 2013; Perrino et al., 2013). Model calculations estimate that particle-bound water constitutes 20–35% of the annual mean European atmospheric PM concentrations (Tsyro, 2005). In spite of that, only in a few [studies](#) aerosol particulate  
10 water has been indirectly or directly estimated (Dick et al., 2000; Rees et al, 2004; Stanier et al., 2004; Speer et al., 1997; Speer et al., 2003; Kitamori et al., 2009).

Several attempts have been done and published, to account for water in sampled aerosols. Using a thermodynamic equilibrium ion modelling, temperature, humidity and inorganic ions concentrations, Chen et al. (2014) estimated that water constituted up to 38% of the PM<sub>2.5</sub> mass in the heavily polluted atmosphere of Beijing, for aerosols weighted at 40% [relative humidity \(RH\)](#).  
15 To estimate strongly bound water, Harrison et al. (2003), in samples weighted at 45-50% RH, applied a hydration multiplication factor of 1.29 to the measured sulphates and nitrates (as ammonium or/and sodium compounds). Sciare et al. (2005) and Genga et al. (2017) successfully used this methodology to close the mass balance in Mediterranean aerosols.

[During](#) laboratory studies with water and sea-salt particles, Tang et al. (1997) found the presence of a hysteresis super saturation when decreasing relative humidity (~~RH~~), with suddenly efflorescence at 47% [RH](#). Depending on whether the particles were in  
20 a dry, or wet state, the ratio water/dry sea-salt masses observed at 50% RH was 0.4, or 1.4, respectively.

Tang and Munkelwitz (1994) and Shu et al. (1998) determined [the water content](#) in ammonium sulphate. [A water/salt ratio of 0.4 was obtained at 50% RH](#), in liquid meta-state equilibrium. A ratio of 0.45 was employed to calculate the water [contribution to](#) ammonium sulphate aerosols by Speer et al. (2003).

Speer et al. (2003) also estimated the water content in organic aerosol particles. A relationship between the excess liquid water and the measured organic carbon mass was found. Through modelling it was determined that, on average, about 80% of the liquid water in the PM<sub>2.5</sub> could be accounted for by inorganic ions, with the remaining 20% associated with organic compounds.  
25 The liquid water to organic carbon mass ratio, at 50%, was estimated as 0.2 (an OM/OC=2 was considered).

### 3 Experimental

The present work uses data from the field campaign of the CVDUST Project, which took place in Santiago Island, Cape Verde,  
30 between January 2011 and January 2012. Atmospheric aerosol and ancillary measurements were performed on the roof platform of the Cape Verde Meteorological Institute, on the outskirts of Praia city (14.92N; 23.48W), 98 meters above sea level and approximately 1.7 kilometres from the sea border. During the sampling period, daily averaged temperatures and RH

ranged from 21 to 29 °C and from 50 to 86%, respectively. Total rainfall was only 152mm, concentrated in the months of August to October.

PM<sub>10</sub> aerosol particles were collected, in parallel, onto three filter types (quartz fibre, Teflon and Nuclepore membranes) with high volume and low volume samplers, equipped with PM<sub>10</sub> inlets. A total of 140 events were sampled, with filtering periods ranging from 6 to 96 hours (the low sampling periods during Saharan dust episodes) [allowing the collection of](#) enough aerosol material for all necessary analysis without the [risk](#) of clogging the filters during dust storms. Taking into account the variable extension of sampling periods, in this publication all the calculated concentration averages are weighted by the sampling time. Details of sampling and filter analysis are given elsewhere (Almeida-Silva, 2014; Salvador et al., 2016); here we only provide a summary of published information. Filters were weighted with semi-micro, or micro, balances to determine PM<sub>10</sub> total mass, at 50% RH and 20°C. Mass concentrations measured in the three parallel sampling lines compared quite well, ( $R^2=0.98-0.99$ ; best-fit lines with  $y/x=0.98-1.02$  – for details consult Figure A1 in Annex), a confirmation that the filters were sampling the same aerosol population.

Nuclepore filters were employed to determine particulate elemental content using Particle Induced X-Ray emission (PIXE) and/or k<sub>0</sub>-Instrumental Neutron Activation Analysis (k<sub>0</sub>-INAA) (Almeida et al., 2013; Almeida-Silva, 2014). A total of 26 elements was measured by the two techniques, although some light elements, such as Na and Cl, could only be quantified with large uncertainties, characteristic of the analytical conditions and techniques.

The quartz filters were used to determine carbonates by acid evolution and non-dispersive infrared analysis of evolved CO<sub>2</sub> (Pio et al., 1994), and elemental carbon (EC) plus organic carbon (OC), with a homemade thermo-optical carbon analyser, [after pre-removal of carbonates with HCl fumes](#), (Pio et al., 2011).

Water-soluble anions and cations, sampled in Teflon filters, were measured by ion chromatography. The method permitted the quantification of NH<sub>4</sub><sup>+</sup>, Na<sup>+</sup>, Mg<sup>2+</sup> and K<sup>+</sup> cations, and SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, and Cl<sup>-</sup> anions. Comparison between total cation and anion equivalents indicate a clear excess of cations (42%, on average). Inclusion of independently measured carbonates in the ionic balance, brings the ratio cations/anions to 0.93, demonstrating the importance of carbonate measurements for a more complete aerosol characterisation in dusty environments (see Figure A2 in Annex, for details).

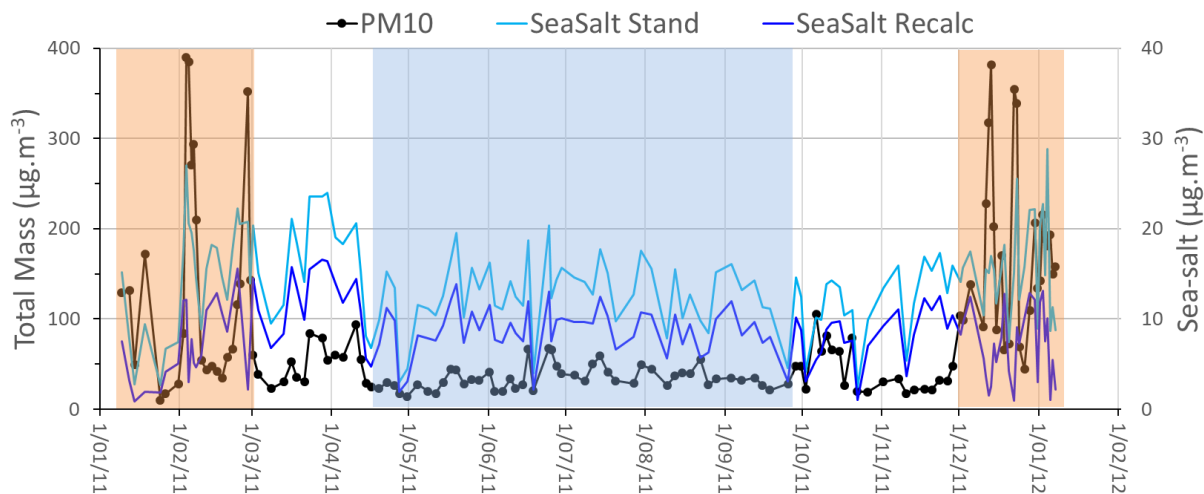
## 4 Results and discussion

### 4.1 PM<sub>10</sub> mass and components

As reported elsewhere (Almeida-Silva, 2013; Pio et al., 2014; Salvador et al., 2016) Cape Verde has two distinct atmospheric pollution seasons. During winter months (December-February) the atmospheric boundary layer is impacted by important dust intrusions from the Saharan region, with daily averaged PM<sub>10</sub> concentrations going up to hundreds of µg.m<sup>-3</sup> (see Figure 1 and Table 1); This period is locally designated by “Bruma-Seca”, meaning “Dry-Haze”.

During May-September there is no direct intrusion of dust plumes from Africa, at lower atmospheric levels, in the boundary layer, (represented by a blue shade mask in Figure 1), and we call it “Dust-Free” season. During the “Dust-Free” period the

atmosphere contains still important quantities of dust, originated, either from the island arid surface, or from continuous dust transport from Africa into the region across the free troposphere, which partially sediments to lower atmospheric levels (Gama et al., 2015). The months of March, April, October and November have intermittent direct intrusions of Sahara dust, with  $PM_{10}$  concentrations sometimes reaching one hundred  $\mu\text{g}\cdot\text{m}^{-3}$ . Throughout this document we present average results for the total sampling period and for the two “Dry-Haze” and “Dust-Free” seasons.



**Figure 1: Levels of  $PM_{10}$  and sea-salt along the annual sampling period. Shades of brown and blue represent, respectively, periods clearly with (Dry-Haze) and without (Dust-Free) dust plume direct intrusions. “SeaSalt Stand” are concentrations of sea-salt spray calculated by considering  $Na^+$  as an exclusive tracer of marine emissions. “SeaSalt Recalc” represent sea-salt levels estimated after removal of soil dust  $Na^+$  and  $Mg^{2+}$ .**

#### 4.2 Sea-salt and soil sources

$PM_{10}$  in Cape Verde is mainly influenced by emissions from sea and soil surfaces (Gama et al., 2015). The determination of the soil contribution, by composition/mass balance, can be improved from the knowledge of source soil composition. When information on dust originated soil is [unavailable](#), average global upper continental crust composition is frequently employed (Manson and Moore, 1982; Wedepohl, 1995). In our case, dust is almost exclusively [originating](#) from North African Sahara and Sahel, and information on regional soil composition [for these regions](#) is available.

There is a handful of publications on soil composition in North Africa, which [provide evidence](#) of a wide [composition](#) variability across the Saharan and sub-Saharan regions (Guieu et al., 2002; Journet et al., 2013; Scheuven et al., 2013, and references there in). One of the most complete Saharan soil data sets was given by the IDEA-CSIC Research Group (Moreno

et al., 2006). The publication provides the concentrations of 47 elements in the bulk soil of 9 locations, across four regions, in North Africa (Hogar Massif, Chad Basin, Niger and Western Sahara). Castillo et al. (2008) provides soil size distributed composition information for the same sites. We used this data set to infer the composition and mass balance of soil dust in our samples (Table A1, in Annex, adapted from Moreno et al., (2006), shows compound average contributions).

- 5 Saharan soil composition in Moreno et al. (2006) reveals some differences by comparison with the average crust/upper crust (Manson and Moore, 1986; Wedepohl, 1995), with a relative enrichment of Si and Ti, probably as a result of intense weathering of Sahara desert soils. Si and Ti form rather hard crystals (silica and rutile), resistant to physical weathering. The size chemical speciation of Saharan soils by Castillo et al. (2008) revealed Al, Mg, and Fe moderate enrichments, in suspended finer particles, in opposition to K that had increased concentrations at coarser sizes.

10

**Table 1: Weighted average concentrations of PM<sub>10</sub>, major elements, ions and carbon fractions, for the total campaign and for Dry-Haze and Dust-Free seasons. Na and Cl (in red) were measured with important inaccuracies, as evidenced by comparison with the respective water-soluble ion concentrations.**

	<b>Year Average</b> (ng.m <sup>-3</sup> )	<b>Dry-Haze</b> (ng.m <sup>-3</sup> )	<b>Dust- Free</b> (ng.m <sup>-3</sup> )
<b>PM10</b>	59,602	117,000	33,900
<b>Si</b>	6,595	17,100	2,150
<b>Al</b>	3,814	9,930	1,220
<b>Na</b>	3,230	3,440	3,040
<b>Fe</b>	1,835	4,560	721
<b>Ca</b>	1,450	2,920	783
<b>Mg</b>	969	2,130	454
<b>K</b>	772	1,580	380
<b>Ti</b>	197	454	93
<b>Ba</b>	31	66	21
<b>Mn</b>	31	78	12
<b>Cl</b>	4,660	5,810	3,930
<b>S</b>	852	941	792
<b>Na<sup>+</sup></b>	4,047	4,360	3,760
<b>Ca<sup>2+</sup></b>	818	1,540	475
<b>Mg<sup>2+</sup></b>	386	443	351
<b>K<sup>+</sup></b>	240	251	166
<b>NH<sub>4</sub><sup>+</sup></b>	213	101	257
<b>Cl<sup>-</sup></b>	5,344	5,770	4,840
<b>SO<sub>4</sub><sup>2-</sup></b>	1,898	1,880	1,880
<b>NO<sub>3</sub><sup>-</sup></b>	1,191	1,250	115
<b>CO<sub>3</sub><sup>2-</sup></b>	816	2,230	169
<b>EC</b>	188	110	270
<b>OC</b>	980	1,340	870

Aerosols generated by suspension from Sudan desert soil, have [also](#) shown an Al enrichment, while there has been an impoverishment in Si and Ti elements, in smaller particles (Eltayeb et al., 2001). The ratio Al/Si in suspended dust decreased with increasing particle size. A similar behaviour happened to the ratio Al/Ti, for particles <45 µm. The authors attributed this  
5 behaviour to the presence of large quartz crystals in soil and their substitution in dust by smaller particles of alkali-plagioclase and clay minerals.

Journet et al. (2014) concluded that, in desert soils, silica minerals [accumulate](#) preferably in the silt fraction ( $2\mu\text{m} < D_p < 65\mu\text{m}$ ), while kaolinite and other clay minerals [are mostly concentrated](#) in the clay size fraction ( $D_p < 2\mu\text{m}$ ); kaolinite that has a ratio Al/Si=0.95 is the main mineral of desert areas. The authors [assumed](#) that the mineral composition of airborne dust is broadly  
10 similar to that of the clay size fraction in the desert soil.

Taking into account the previous information we speculate that, as a result of soil weathering, particles containing Si are heavier/larger than other soil particles and, therefore, are more difficultly suspended by the wind and exported to other regions, enriching local soils.

Consequently, it is expectable that Saharan suspended dust will be impoverished in Si and Ti, by comparison with less hard  
15 minerals containing Al, Fe, Mg, Na, etc. This is observed in our aerosol samples where there is a quite constant Al/Si mass ratio of 0.61, independent from the period of the year or the air mass trajectories ( $\text{Al}=0.61\text{Si}-254\text{ ng}\cdot\text{m}^{-3}$ ;  $R^2=0.99$ ). Comparison between prevalently soil originated elements, showed that, for Al, Fe and Mn, the concentration ratios in the aerosol are similar to those in average crustal material and within the limits of Sahara data from Moreno et al. (2006). In contrast, the ratio between particulate Al and Si (or Ti) levels is 2 to 4 times higher than in Moreno et al. (2006), indicating a Si (Ti) deficit, by comparison  
20 with other major Saharan soil elements (for clarification see Figure A3, in Annex). After aerosol measurements performed in Southern Morocco, Kandler et al. (2006) concluded that the major dust constituents were quartz, potassium feldspar, plagioclase, calcite, hematite and clay minerals. Particles in the range of 0.5-50 µm consisted mainly of silicates and, above 50 µm, quartz was dominant.

Published information concerning the Al/Si mass ratios in atmospheric dust from the Saharan region is still scarce. Al/Si  
25 average ratios of 0.43-0.49, with values, depending on the air mass origin, were measured by Chiapello et al. (1997), using bulk filtration, in Sal Island, Cape Verde. Formenti et al. (2003) determined Al/Si ratios of the order of 0.5, in particles larger than 1 µm, during aircraft measurements performed in the Cape Verde region. Remoundaki et al. (2013) found Al/Si ratios of  $0.44\pm 0.12$  in PM<sub>2.5</sub> aerosols collected in Greece under the influence of air mass transport from the Saharan region. In South America, Formenti et al. (2001) observed Al/Si values of  $0.48\pm 0.08$  in fine particles and of  $0.77\pm 0.18$  in the coarse aerosol  
30 fraction. Aerosol measurements over western Atlantic and the eastern coast of USA present an Al/Si ratio value coincident with our measurements (Eldred, 2003). From this information we hypothesise that during long-range transport of Saharan dust there is a prevalent loss, by sedimentation (or non-emission), of Si (and Ti), in comparison with other particulate dust components, which becomes more evident for larger particles.

Because of the Al/Si behaviour in our samples, we felt more confident in using Fe as a representative tracer of soil contribution in composition/mass balance calculations. In addition, Fe is the soil element that showed a better correlation ( $R^2=0.99$ ) with  $PM_{10}$  total mass during dust events (see Figure A4, in Annex, for visualisation).

In coastal, or marine, non-dusty, environments it is common, and correct, to infer the mass contribution of particulate sea spray to the atmospheric aerosol by using  $Mg^{2+}$  or, predominantly,  $Na^+$ , as an exclusive sea-salt tracer. However, these ions are also present in the soil and, during dust episodes, the soil contribution cannot be neglected. From Figure 1 it is possible to observe that sea-salt concentrations calculated in this way increase steeply during dust pollution episodes, which is not reasonable.

To eliminate the soil interference in sea-salt estimation, we employed  $Fe/Mg^{2+}$  and  $Fe/Na^+$  mass ratios superior edge lines (see Figure 2). These edge lines represent the minimum fractional contribution of sea-salt to  $Mg^{2+}$  and  $Na^+$  in the aerosol, compatible with experimental data. Therefore, it is expectable that they represent, reasonably well, the ratios between the ions and Fe, in soil dust, as long as it is acceptable that, in such a location, these are the unique major sources of Fe and  $Na^+$ . Based on these edge line ratios, the amounts of soil  $Mg^{2+}$  and  $Na^+$  ions can thus be determined and subtracted to the total ion concentrations, permitting a first estimation of sea-salt  $Mg^{2+}$ s, and  $Na^+$ s. As these edge lines only approximatively represent the soil ratios, the calculation of sea-salt contributions may consequently suffer from these inaccuracies/variabilities.

15

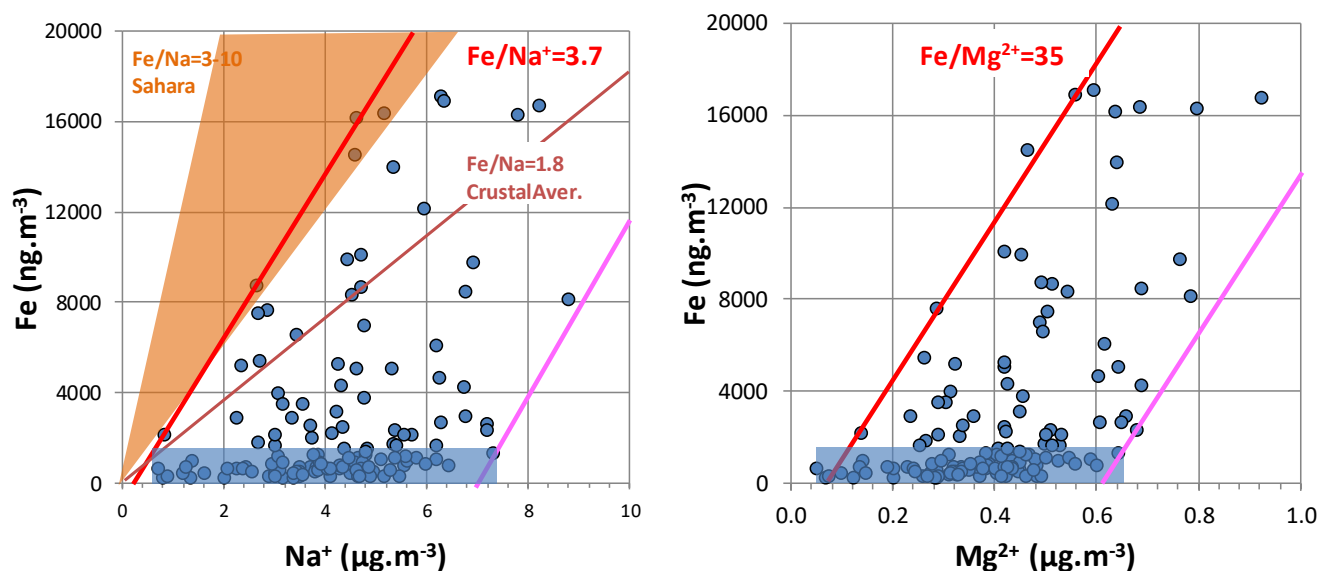


Figure 2: Edge lines (in red) for Fe versus  $Na^+$  and Fe versus  $Mg^{2+}$  inter-comparisons. The blue rectangle represents periods without significant dust intrusions. The pink lines are parallels to the red edge lines in the maximum ion concentration regions. Also shown Fe/Na ratio ranges in Sahara and global soils, for total sodium, taken from Moreno et al., (2006), and Manson and Moore, (1982).

20

A further refinement of sea-salt calculation can be implemented by looking at the ratios ( $\text{Na}^+_{\text{s}}/\text{Mg}^{2+}_{\text{s}}$ ), calculated from  $\text{Na}^+_{\text{s}}$  and  $\text{Mg}^{2+}_{\text{s}}$  in each sample, and comparing them with those in seawater ( $\text{Mg}^{2+}_{\text{ss}}/\text{Na}^+_{\text{ss}}=0.12 \mu\text{g}\cdot\mu\text{g}^{-1}$ , Turekian, 1968). Since it is not possible to have less  $\text{Mg}^{2+}_{\text{ss}}$  (or more  $\text{Na}^+_{\text{ss}}$ ) ion mass sea-salt than the one given by the 0.12 ratio, -if  $\text{Mg}^{2+}_{\text{s}}/\text{Na}^+_{\text{s}}\geq 0.12$ ,  $\text{Na}^+_{\text{s}}$  is chosen as the true sea salt  $\text{Na}^+_{\text{ss}}$  concentration. Otherwise,  $\text{Mg}^{2+}_{\text{s}}$  is chosen as true  $\text{Mg}^{2+}_{\text{ss}}$  tracer. The contributions of other sea-salt ions are, consequently, estimated from the chosen  $\text{Na}^+_{\text{ss}}$  or  $\text{Mg}^{2+}_{\text{ss}}$ , using the salt ratios present in seawater (Turekian, 1968). In this way, it is assured that we are not imposing excessive soil  $\text{Mg}^{2+}$ , or  $\text{Na}^+$ , removal. However, this methodology may not take into account all the water-soluble  $\text{Mg}^{2+}$  and  $\text{Na}^+$  in the soil dust component.

Figure 1 presents the estimation of sea-salt contribution to the aerosol (“SeaSalt Recalc”), considering the methodology described above. The Figure shows that, with the modified methodology, there is no increase of sea-salt aerosol loading during dust intrusions, in contrast with the standard methodology. During the dust periods there is, even, a small decrease in the contribution of sea-salt to the aerosol that may result either from excessive calculation of soil  $\text{Na}^+$  and  $\text{Mg}^{2+}$ , or, more probably, from an increased deposition rate of sea-salt during dusty periods, by co-sedimentation of dust and sea-salt particles. Because of the application of our adapted alternative methodology, the amount of calculated sea-salt contribution decreases by 46% in the Dry-Haze season and 32% in the rest of the year.

The correct determination of sea-salt ion concentrations permits the estimation of the remaining common elements, supposedly from soil origin. In this way, it is possible to calculate  $\text{Mg}_{\text{soil}}$ ,  $\text{K}_{\text{soil}}$  and  $\text{Ca}_{\text{soil}}$  concentrations. Determination of total  $\text{Na}_{\text{soil}}$  and  $\text{Cl}_{\text{soil}}$  is not feasible in this work because of analytical limitations.

Taking into account that we did not, or could not, measure with accuracy P and Na, the calculation of soil contribution was done by adapting Equation 2 to:

$$\text{Soil Dust}=1.15(2.14\text{Si}+1.89\text{Al}+1.43\text{Fe}+1.39\text{Mn}+1.67\text{Ti}+1.66\text{Mg}_{\text{soil}}+1.40\text{Ca}_{\text{soil}}+1.20\text{K}_{\text{soil}}), \quad \text{Equation 3}$$

where the Factor 1.15 is an average for the nine sites studied by Moreno et al., (2006), (see Table A1 in Annex, for detail and clarification). This factor is higher than the corresponding values for the average continental / upper crust (1.05-1.06), taken from Manson and Moore (1982) and Wedepohl (1995).

### 4.3 Secondary formation processes

The attribution of analysed water-soluble anions and cations to different sources and formation processes can be done using the sequential ionic balance proposed by Alastuey et al. (2005), adapted and developed by Mirante et al. (2014) for the Madrid urban aerosol. The present situation, with very large inputs of dust and marine aerosols, imposes a further adaptation of the ionic balance method, because gas-to-particle reactions involving precursor pollutants and sea-salt spray, or dust, cannot be neglected, and, from the evaluation of dust and sea-salt composition, the amounts of soluble ions of sea-salt and dust origin have to be initially calculated. Therefore, the ionic balance applied to the present samples is the following:

- 1-Start by calculating soil  $\text{Na}^+_{\text{soil}}$  and soil  $\text{Mg}^{2+}_{\text{soil}}$  from  $\text{Fe}/\text{Na}^+$  and  $\text{Fe}/\text{Mg}^{2+}$  edge lines in Figure 2.
- 2-Calculate sea-salt  $\text{Na}^+_{\text{ss}}$  and  $\text{Mg}^{2+}_{\text{ss}}$  from differences between total and soil  $\text{Na}^+_{\text{soil}}$  and  $\text{Mg}^{2+}_{\text{soil}}$ .
- 3-Recalculate sea-salt  $\text{Na}^+_{\text{ss}}$  and  $\text{Mg}^{2+}_{\text{ss}}$ , using minimum values and the  $\text{Na}^+/\text{Mg}^{2+}$  ratio in seawater.



4-Calculate sea-salt mass concentration and composition from  $\text{Na}^+_{\text{ss}}$  and  $\text{Mg}^{2+}_{\text{ss}}$ , and seawater ion ratios, taking into account available  $\text{Cl}^-$ .

5- Calculate non-sea-salt  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$  and  $\text{Cl}^-$ .

6-Associate, sequentially, free non-sea-salt  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  with  $\text{NH}_4^+$ , until all  $\text{NH}_4^+$  is neutralized.

7-From the differences between total and sea-salt cations, calculate soil  $\text{Na}^+_{\text{soil}}$ ,  $\text{Mg}^{2+}_{\text{soil}}$ ,  $\text{K}^+_{\text{soil}}$  and  $\text{Ca}^{2+}_{\text{soil}}$ .

8-Associate free  $\text{NO}_3^-$ , sequentially, with free sea-salt and soil cations.

9-Associate, totally,  $\text{CO}_3^{2-}$ , sequentially, with free soil  $\text{Ca}^{2+}_{\text{soil}}$ ,  $\text{Mg}^{2+}_{\text{soil}}$ ,  $\text{Na}^+_{\text{soil}}$  and  $\text{K}^+_{\text{soil}}$ .

10-Associate free  $\text{SO}_4^{2-}$ , sequentially, with free  $\text{Ca}^{2+}_{\text{soil}}$ ,  $\text{Mg}^{2+}_{\text{soil}}$ ,  $\text{Na}^+_{\text{soil}}$  and  $\text{K}^+_{\text{soil}}$ .

11-Associate free  $\text{Cl}^-$  with free  $\text{Na}^+_{\text{soil}}$  and  $\text{Mg}^{2+}_{\text{soil}}$ . Any excess  $\text{Cl}^-$  is associated with  $\text{K}^+_{\text{soil}}$ .

12-Calculate the total masses of water-soluble soil sulphate, nitrate and chloride.

13-Edge line ratios between Fe and sulphate, nitrate, or chloride, permit a rough calculation of the fraction of these ionic compounds that are present in the soil, or that result from secondary reaction with atmospheric produced acids (for visualization [see](#) Figure A4, in Annex).

Using an Excel spreadsheet, the 13 steps were applied, sequentially, in order to attribute all measured cations and anions to sea-salt, soil and secondary inorganic compounds. The first 4 steps were described in detail in the beginning of this section. Due to space limitations, only results for the remaining 8 steps are presented.

**Table 2: Soil and secondary inorganic compounds resulting from the ionic balance**

Inorganic Compounds	SIC ss+du							
	SIC am		SoilDust sol		SIC du		SIC ss	
	Dust-Free ( $\mu\text{g}\cdot\text{m}^{-3}$ )	Dry-Haze ( $\mu\text{g}\cdot\text{m}^{-3}$ )	Dust-Free ( $\mu\text{g}\cdot\text{m}^{-3}$ )	Dry-Haze ( $\mu\text{g}\cdot\text{m}^{-3}$ )	Dust-Free ( $\mu\text{g}\cdot\text{m}^{-3}$ )	Dry-Haze ( $\mu\text{g}\cdot\text{m}^{-3}$ )	Dust-Free ( $\mu\text{g}\cdot\text{m}^{-3}$ )	Dry-Haze ( $\mu\text{g}\cdot\text{m}^{-3}$ )
$(\text{NH}_4)_2\text{SO}_4$	0.94	0.37						
$\text{NH}_4\text{NO}_3$	0.00	0.01						
$\text{CaCO}_3$			0.27	3.15				
$\text{MgCO}_3$			0.01	0.24				
$\text{Na}_2\text{CO}_3$			0.00	0.29				
$\text{K}_2\text{CO}_3$			0.00	0.01				
$\text{NaNO}_3$			0.83	1.24	0.72	0.59	0.45	0.18
$\text{Mg}(\text{NO}_3)_2$			0.09	0.11	0.08	0.05	0.09	0.04
$\text{KNO}_3$			0.06	0.10	0.05	0.05	0.01	0.00
$\text{Ca}(\text{NO}_3)_2$			0.01	0.00	0.01	0.00	0.02	0.01
$\text{CaSO}_4$			0.59	0.43	0.42	0.10		
$\text{MgSO}_4$			0.01	0.13	0.01	0.03		
$\text{Na}_2\text{SO}_4$			0.12	0.80	0.08	0.20		
$\text{K}_2\text{SO}_4$			0.05	0.06	0.04	0.01		
$\text{NaCl}$			0.25	1.52	0.15	0.53		
$\text{MgCl}_2$			0.00	0.06	0.00	0.02		
$\text{KCl}$			0.00	0.06	0.00	0.02		
<b>TOTAL</b>	<b>0.94</b>	<b>0.37</b>	<b>2.28</b>	<b>8.20</b>	<b>1.55</b>	<b>1.61</b>	<b>0.57</b>	<b>0.22</b>

Using a ratio of OM/OC=2.0 and measured EC and OC, it is also possible to estimate the total non-carbonated, carbonaceous aerosol.

With this IMB methodology, it is possible to account for the [presence](#) of seven source classes: Sea-Salt spray (SeaSalt), Insoluble Soil Dust (SoilDust ins), Soluble Soil Dust (SoilDust sol), Secondary Inorganic Compounds from the reaction of atmospheric acids with ammonia (SIC am), Sea-Salt (SIC ss) and Dust (SIC du), and non-carbonate Carbonaceous elemental and organic matter (Carbon).

The results of water-soluble compounds are presented in Table 2, for the Dry-Haze and Dust-Free seasons. Concentrations of secondary ammonium salts (SIC am) more than double during the Dust-Free season, probably as a result of higher temperatures and transport of air masses from non-desert polluted areas, or the removal by co-sedimentation with dust, during dust episodes. Soluble soil dust (SoilDust sol) more than triple during the Dry-Haze season, being formed mainly by calcium carbonates and sulphates, sodium nitrates and sulphates, and by sodium chloride.

Reaction of acid precursors with soil dust (SIC du) produces equivalent amounts of secondary compounds during the two seasons, probably because of limited availability of acidic precursors. Secondary processes resulting from acidic reactions with sea-salt (SIC ss) produce more sea-salt secondary material during the dust-free season. As, in the present conditions, it is difficult to clearly differentiate between the two processes, because it is not possible to give a priority in the competitive acidic reaction with sea-salt, or dust, in the rest of the publication [the](#) two source processes are treated together, as SIC ss+du.

#### 4.4 Particulate water

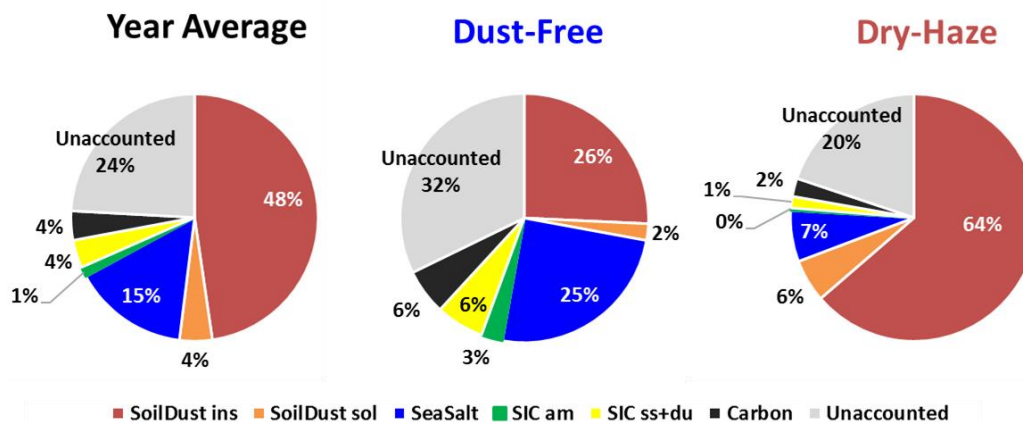
The fractional contribution of the 6, adapted, source classes is given in Figure 3, for the two seasons and for the total campaign. The figure reveals that the sum of the quantified sources only accounts for 76% of the measured PM<sub>10</sub> total mass concentration, on average, during the total measurement campaign. The value decreases to 68% during the Dust-Free season. These fractions are of the same order of values published in literature (Perrino et al., 2013; Rees et al., 2004; Andrews et al., 2000; Grigoratos et al., 2014).

As in our case carbonates were directly measured, it is predictable that most of the unaccounted mass results from the aerosol water content, in the form of adsorbed/absorbed water (hydrates in soil constituents were already considered with the application of Factor F), now that PM<sub>10</sub> total mass measurement was performed at 20 °C and 50% RH, in conditions of equilibrium between the laboratory atmospheric water vapour and the particulate material in the filter.

To estimate the amount of sorbed water in the aerosol we consider that, by approximation, there is a thermodynamic equilibrium between the controlled room atmosphere, at 20°C and 50% RH, and the aerosol deposited on the filter, during the mass weighting in the Laboratory, and also that the behaviour of different compounds is independent of internal or external mixture conditions.

For sea-salt, thermodynamic information from Tang et al. (1997) was used, which, at 50% RH and 20 °C, indicates a water/salt mass equilibrium ratio of 0.4 for a dry phase state, or 1.4 for a metastable deliquescent liquid phase state. The ISORROPIA

thermodynamic equilibrium model (Nenes et al., 1998a, 1998b) was applied, at 20°C and 50%RH, both to the calculated sea-salt, alone, (SeaSalt) and considering, together, the sea-salt and the secondary inorganic compounds formation by the attack of atmospheric acids (SeaSalt + SIC ss). ISORROPIA output gave water fractions very similar, in both cases, and also coincident with the values taken from Tang et al. (1997) for wet or dry containing phases. As information about phase status during weighting is unavailable and the weighting was performed at RH very near the forced efflorescence point (47% RH), we used for the water/salt ratio the arithmetic average of the two equilibrium values (0.9).



**Figure 3: Contribution of different components to the PM<sub>10</sub> measured total mass, estimated by IMB, for the whole sampling campaign and for the “Dust-Free” and “Dry-Haze” seasons.**

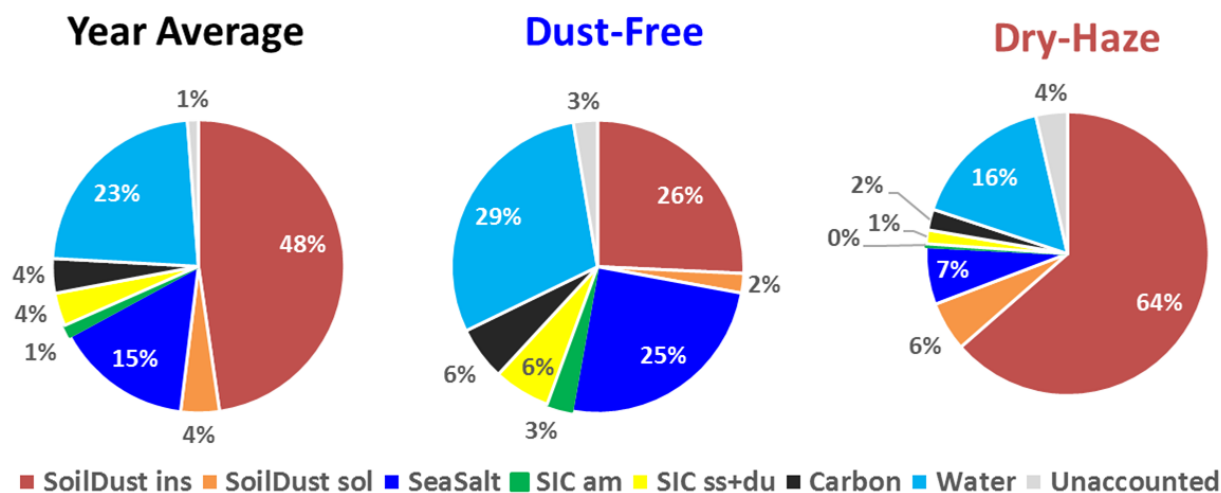
For secondary inorganic water-soluble ammonium salts (mainly ammonium sulphate), thermodynamic information from Xu et al. (1998) and Tang and Munkelwitz (1994) was applied, which shows an equilibrium water/ammonium sulphate mass ratio of 0.4 at 50% RH.

The information concerning the water content of organic polar matter was taken from Speer et al. (2003) who used water/OC ratio of 0.2.

Suspended soil also sorbs water, principally the water-soluble ionic component. We used ISORROPIA, version 2.1, which includes ions from crustal origin, to estimate the water content in soluble dust. The ISORROPIA II version was run for a liquid metastable assumption and applied to the soluble soil dust (SoilDust sol) and to the sum of soluble soil dust and secondary inorganic compounds formed from the attack of atmospheric acids on dust particles (SoilDust sol + SIC du). In the first case, the average water/salt mass ratio calculated was 0.22 in the Dust-Free season and 0.47 in the Dry-Haze season. In the mixture, the calculated ratios were comparable (0.14 and 0.50, respectively). Due to the lack of more specific information, water/soluble dust ratio values of 0.2, during the Dust-Free season, and 0.5, during the rest of the year, were employed.

Various important components of the insoluble fraction of dust are hygroscopic, such as clay minerals. Schuttlefield et al. (2007) measured water adsorption by clay minerals having found a large variability in the water uptake by different clay

mineral species, with water/clay mass ratios varying from 0.02-0.06 for kaolinite, going up to 0.17 for illite and 0.08-0.7 for montmorillonite, at 23 °C and 50% RH. Based on these data, a round value of 0.1 for the ratio water/SoilDust ins was adopted for our samples.



5

**Figure 4: Inclusion of estimated water in the IMB, for the total sampling campaign and for the Dust-Free and Dry-Haze seasons.**

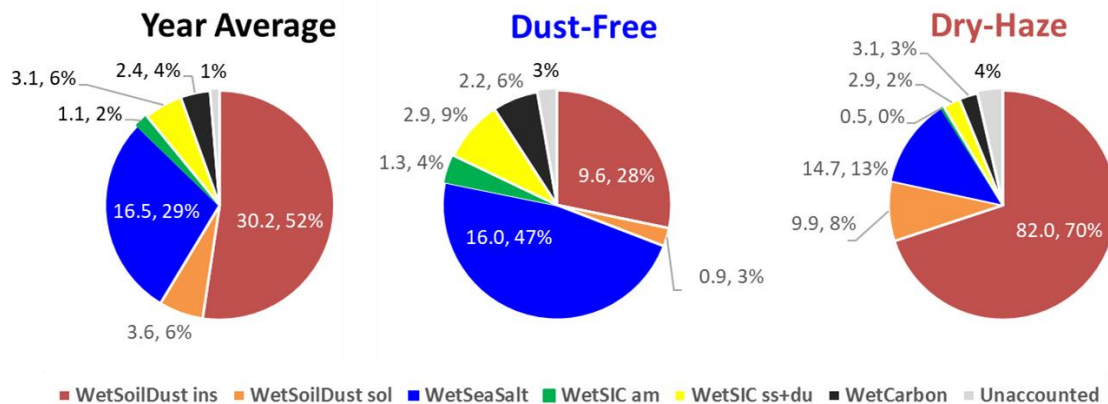
The estimation of total water content in the collected aerosols using the above referred ratio assumptions is presented in Figure 4. The figure shows that, by using this methodology, there is almost a perfect accounting of the PM<sub>10</sub> total mass. Water represents an average contribution of 23% to the aerosol mass. During the Dry-Haze period, the calculated water, on average, accounts for 16% of the PM<sub>10</sub> mass, with a maximum contribution of 29% during the Dust-Free season. By including particulate water, only around 1-4% of the PM total mass is unaccounted, with the applied IMB methodology.

#### 4.5 Comparison with PMF

15 The final ionic and mass balance calculations are presented in Figure 5, for the total campaign and the two seasons, considering the components associated with the respective water uptake. The addition of sorbed water reinforces the impact of hygroscopic /soluble components, such as sea-salt, in the atmospheric aerosol loading, which, for example, during the Dust-Free season increases its contribution from 25 to 47%.

20 The results of IMB can be evaluated and compared with PMF results applied to the same data set and already published by Salvador et al. (2016). Here, the published PMF results were reorganised, in order to explicit the PMF contributions during the two Dry-Haze and Dust-Free seasons and to show unaccounted/excess calculated PM mass, as represented in Figure 6.

The PMF analysis could differentiate 7 aerosol sources: three concerning soil contamination, two considering Secondary Inorganic Components, plus sea-salt and combustion processes. The dust sources comprised “Mineral1”, associated mainly with Al, Si and Fe, “Mineral2”, associated with  $\text{CO}_3^{2-}$  and  $\text{Ca}^{2+}$ , and “Dust+Ind” containing both soil elements (Al, Si, Fe) and tracers of industrial emissions (V, Ni, Cu and Cr). The secondary inorganic components included “SIC1”, associated with  $\text{NH}_4^+$ ,  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{Na}^+$ ,  $\text{K}^+$  and  $\text{Mg}^{2+}$ , and “SIC2” containing  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ . The “SeaSalt” source represented most of the variability of  $\text{Na}^+$ ,  $\text{Cl}^-$ ,  $\text{Mg}^{2+}$ ,  $\text{Br}^-$  and  $\text{K}^+$ . The “Combustion” source included mainly EC and OC variability [\(Complementary information concerning PMF sources compositions and contributions can be found in Figure A8, in the annex section\)](#).



10 **Figure 5: IMB obtained by attributing the calculated water content to the respective source classes, for the entire campaign and the Dust-Free and Dry-Haze seasons. Values are in  $\mu\text{g.m}^{-3}$  and percentage of measured total mass.**

Table 3 compares both methodologies for the two seasons and the total campaign (for individual sampling events consult Figures A6 and A7, in Annex). There is a ~~quite~~ good agreement between the two source apportionment techniques. Both methodologies reproduce almost totally the measured  $\text{PM}_{10}$  total mass.

15 [Table 3 shows](#) a good comparability between total soil dust estimated by both methods in any season. In the IMB, SoilDust sol fraction is approximately equivalent to the Mineral2 component of PMF. The mass balance method could not discriminate the Dust+Ind from the total insoluble dust fraction, as [was possible](#) with PMF.

Contribution of sea-salt was also equally estimated by the two techniques, [on average](#), during the Dry-Haze season, but during the Dust-Free free period the IMB estimated somehow higher sea-salt levels than the PMF.

20 SIC values are similar in both source apportionment methodologies, but during the Dust-Free period the PMF estimated higher SIC contributions. This was mainly due to higher estimations of ammonium secondary salts by the PMF, which only can happen if other compounds are included in the source component, as evidenced by the PMF source profile.

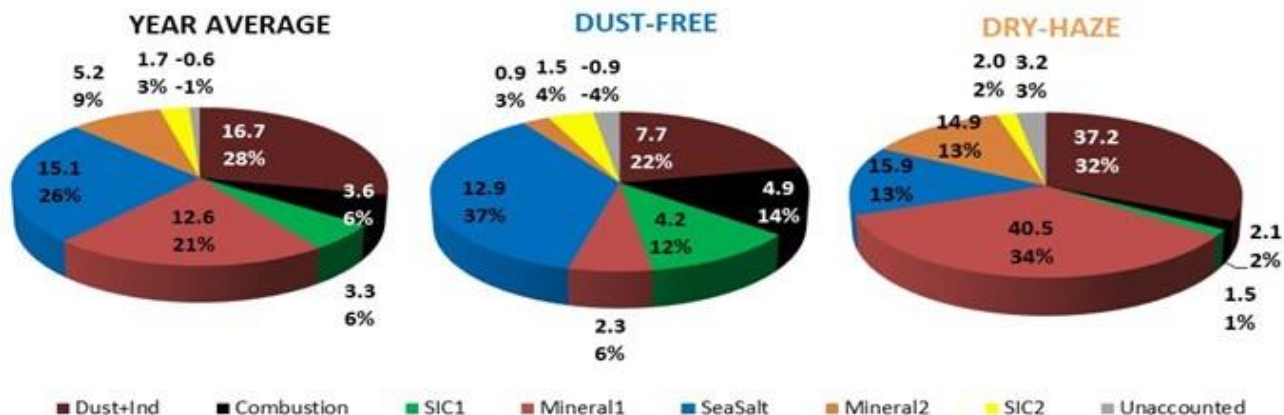


Figure 6: PMF source apportionment results, for the total campaign and during the two pollution seasons. Values are in  $\mu\text{g.m}^{-3}$  and percentage of measured total mass.

5 There is also a higher contribution from carbonaceous/organic/combustion material in PMF, by comparison with IMB, principally during the dry season, although a high factor of 2 was applied to the OM/OC ratio in the mass balance approach. The inclusion by the PMF of other constituents from combustion processes in West Africa is, probably, the reason for the discrepancy.

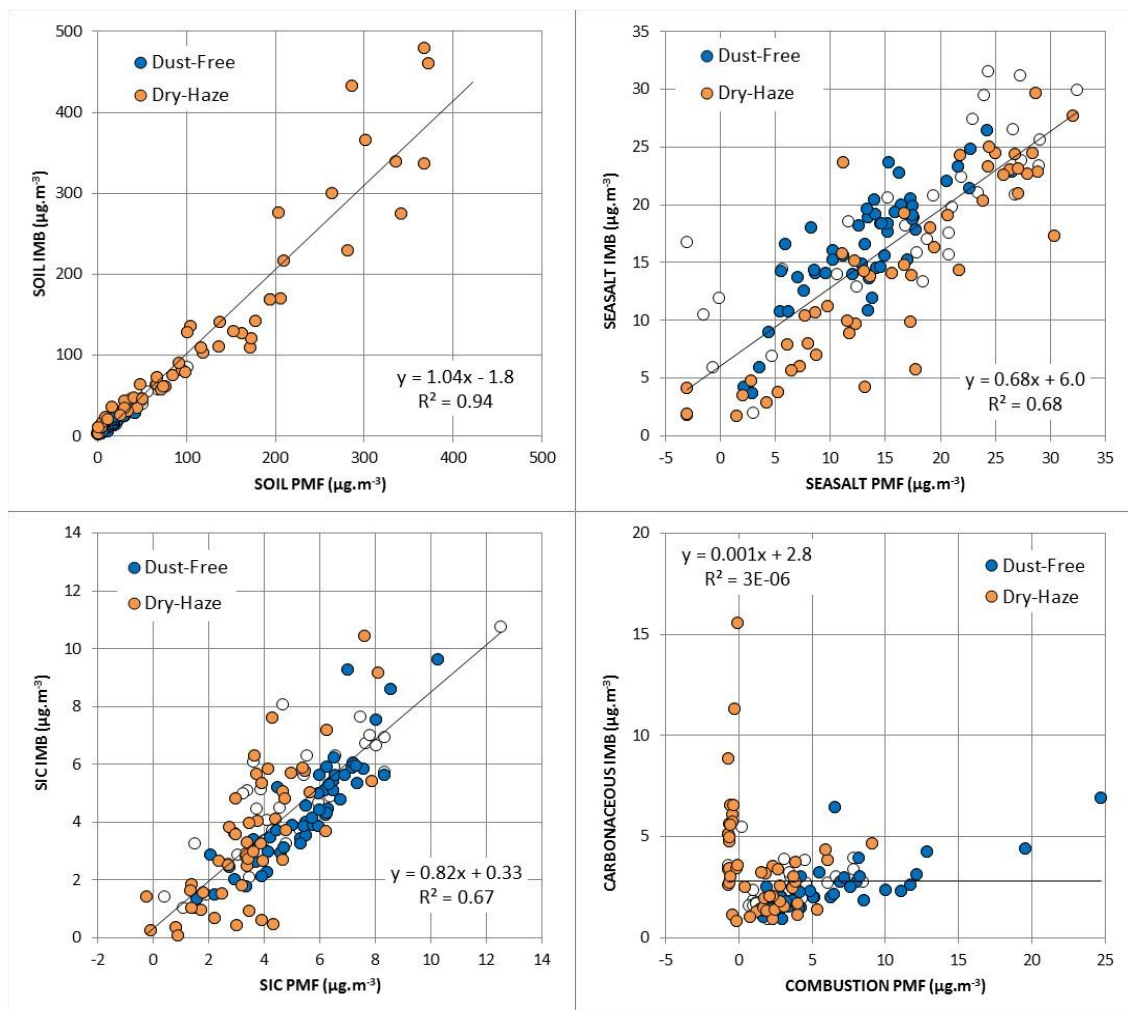
10 **Table 3: Comparison between IMB and PMF results, grouped into four main classes. PM10 represents gravimetric measurements of total mass-** (in IMB, Soil=WetSoilDust ins+WetSoilDust sol; in PMF, Soil=Mineral1+Mineral2+Dust+Ind; in IMB, SIC=WetSIC am+WetSIC ss+du; in PMF, SIC=SIC1+SIC2)

	Year Average		Dry-Haze		Dust-Free	
	PMF ( $\mu\text{g.m}^{-3}$ )	IMB ( $\mu\text{g.m}^{-3}$ )	PMF ( $\mu\text{g.m}^{-3}$ )	IMB ( $\mu\text{g.m}^{-3}$ )	PMF ( $\mu\text{g.m}^{-3}$ )	IMB ( $\mu\text{g.m}^{-3}$ )
Soil	34.5	33.9	92.6	91.9	10.9	10.4
Sea-salt	15.1	16.5	15.9	14.7	12.9	16.0
SIC	4.9	4.2	3.5	3.4	5.7	4.3
Carbon/Combust	3.6	2.4	2.1	3.1	4.9	2.2
<b>SUM</b>	<b>58.2</b>	<b>57.0</b>	<b>114.1</b>	<b>113.1</b>	<b>34.5</b>	<b>32.9</b>
<b>PM<sub>10</sub></b>	<b>57.7</b>		<b>117.3</b>		<b>33.9</b>	

15 Further insight into the capabilities and limitations of IMB versus PMF can be attained by comparing source classes for each individual sample, as presented in Figure 6. From this figure it is possible to confirm the good comparability between the Soil source estimations by both methods, with a linear ratio estimation of 1.04 and a correlation  $R=0.97$ . For Sea-salt estimation, the comparison is not so good with IMB/PMF ratio estimation of only 0.68 and an  $R=0.82$ . This happens principally because, for several samples, PMF gives zero or negative Sea-salt contributions, while the IMB estimates

important Sea-salt values. For a location in the middle of the ocean, it is not expectable to have absence of sea-salt and therefore, in our opinion, PMF fails, attributing, probably, the available sea-salt to a Soil source. At high concentration levels there is a tendency of PMF to give higher sea-salt values than IMB. An inspection of PMF compounds contribution to Sea-salt source shows that, in average, there is a mass inclusion of about 20% of elements such as Si, Al, Fe, etc., in this source (see Figure A8, graph 8, for clarification). Therefore, it is clear that PMF could not completely separate Soil from Sea-salt sources, probably because of the overwhelming presence of soil during dust episodes. During the “Dust-Free” season IMB tends to give somehow higher Sea-salt contributions than PMF (Sea-salt IMB= 0.73 Sea-salt PMF+6.8; R=0.84). One of the possible reasons may be a too high estimation of sea-salt sorbed water, in IMB.

5



10

**Figure 6: Comparison between the apportionment of individual samples using IMB and PMF, for four source classes. Circles filled in white represent periods with intermittent dust intrusions. Linear best fits are presented for the total sampling campaign.**

- 5 Both methodologies also compare reasonably well in what concerns Secondary Inorganic Compounds (SIC) contributions to the aerosol loading, with a ratio IMB/PMF=0.82 and a correlation coefficient R=0.82. Where the comparison fails completely is in the Carbonaceous (IMB) versus Combustion (PMF) sources that present no clear relation. This results from several facts as exposed in the following text. IMB source profile represents only non-carbonate carbonaceous matter, irrespective of the source, while PMF factor intends to represent all emissions from combustion sources, besides carbonaceous matter. Therefore,
- 10 from Figure 6 it is possible to observe in several samples important contributions of Carbonaceous matter estimated by IMB, while PMF gives zero to negative Combustion emission estimations. Most probably this results from soil contribution to organic matter that in PMF is attributed to dust or anthropogenic sources (Ant+Dust, see Graphs C and H in Figure A8, for clarification). Also in PMF the Combustion source has, in average, an important contribution (around 50%; see Figure A8, for clarification) of elements, such as Si, Al, Fe, etc., from soil origin and, therefore, in our opinion, this PMF Combustion source
- 15 is highly contaminated with soil, with PMF not fully capable of separating Combustion from Ssoil dust, due to the overwhelming presence of soil particles during dust episodes.

From Table 3 and Figure 6 we may then conclude that the IMB solution compares very favourably with the outcome of PMF apportionment, being able of correctly discriminating source and formation processes that PMF has difficulty in separating correctly. The good agreement between these two independent source apportionment methodologies adds

20 confidence to the apportionment of an atmospheric aerosol with quite specific and uncommon characteristics. We can rely in the complementarity of both methods for the evaluation of sources contributing to atmospheric contamination, in circumstances of very high natural inputs of sea-salt and desert dust particles, subject to atmospheric transformation during long-range transport.

#### 25 **4.5 Accuracy and Errors**

IMB and PMF are subject to a number of errors that affect the precision and accuracy of the sources estimation. Fully acquainting for all errors is difficult, because some used information (bibliographic or experimental) has no available accuracy or is subject to unknown and unexpected errors. PMF applies several statistical tests to evaluate the influence of random errors and the rotational ambiguity of the obtained solution. Although these tests indicated that the PMF solution was robust they

30 could not identify collinearity problems that resulted in the significant contamination of Combustion and Sea-salt sources with soil dust.

Both methodologies are equally affected by errors in the aerosol chemical analysis. Probably, the higher relative analytical errors are related with elements and EC evaluation, in conditions of high dust concentrations, although care was taken to



sample for shorter periods during dust episodes. EC is frequently near the detection limits and it is quite difficult to fully evaluate the interference of coloured dust during the optical-thermo analytical process. This affects, in an unknown value, principally, the evaluation of the PMF Combustion source that uses EC as its principal tracer (see Figure A8 for clarification). In IMB, there are probably four estimations where the errors influencing the source apportionment are higher: a) calculation of water sorbed in sea-salt; b) the estimation of total soil content based in factor F; c) Calculation of sea-salt Na<sup>+</sup>; d) the estimation of organic matter from OC. It is necessary to take into account however that total errors are controlled by -measured PM<sub>10</sub> total mass constraint (Malm et al., (1994) uses this comparison as a validation and self-consistency check for their reconstructed aerosol mass balance). Falling the sum of estimated masses within 96-99% of measured PM<sub>10</sub> total mass, the individual errors are probably limited (or of opposite directions, with mutual compensation).

Estimation of water content in sea-salt depends mainly on metastable equilibrium considerations that give a water/dry sea-salt mass ratio varying from 0.4 to 1.4. An average value of 0.9 was applied in our calculations. Application of the two extreme values would vary the fractional contribution of wet sea-salt by, approximately, ±6%, ±7% or ±3.5%, for the total sampling campaign, the Dust-Free, or the Dry-Haze periods. It is necessary to consider, however, that by choosing the two ratio extremes the closure of the total PM mass would be affected, resulting in a maximum unaccounted mass of 16%, for the choice of a 0.4 value and a maximum over-prediction of 9%, for the choice of 1.4 ratio and therefore the correctness of these extreme values is questionable.

The calculation of the total soil dust was based in equation 3 that uses an average factor F value of 1.15. F values taken from Moreno et al. (2006) (see Table A1 in annex, for clarification) vary in the range 1.09-1.27 (standard deviation=0.21) and the factor F for global composition is 1.05-1.06. A range of approximately 16% will result from the calculation of soil dust contribution by application of the two extreme F factors. Use of extreme F values would give a maximum unaccounted PM mass of 8% or an excess total PM mass calculation of 6%.

The Fe/Na<sup>+</sup> edge line methodology used to estimate Na<sup>+</sup> and Mg<sup>2+</sup> was conservative, minimizing the subtraction from sea-salt. This was partially compensated by using minimum values and the Na<sup>+</sup><sub>ss</sub>/Mg<sup>2+</sup><sub>ss</sub> ratio in seawater. The edge line intercepts the x axe at Na<sup>+</sup> levels within the range of values observed for periods without significant dust intrusions, in accordance with the fact that there is always sea salt spray in this marine atmosphere (see Figure 2a, where air masses without dust intrusions are represented by a blue rectangle). The points to the right of the edge line have excess Na<sup>+</sup>, by comparison with Fe in the edge line. This excess can have two origins: either it results from variability in the relative content of Na<sup>+</sup> in soil dust, or it is resultant from the variability in the contributions of sea salt. It is clear that the points at low Fe levels, within the blue rectangle, are only consequence from variability in sea salt spray loading. On the right border of the blue rectangle, a pink straight line is drawn, parallel to the Fe/Ion edge lines. As all the measured points are within both lines, this is a strong indicator that Na<sup>+</sup> and Mg<sup>2+</sup> increase, in relation to the edge lines, result mostly from variability in sea-salt input. The real value of the errors in this methodology is difficult to establish but a change in the Fe/Na<sup>+</sup> edge ratio of 10% would have no visible effect in the estimation of sea-salt or soil dust, by IMB.

5 The estimated value for non-carbonate carbonaceous matter depends strongly from the OM/OC ratio. In literature values in the range 1.2-2.3 have been proposed. We used a ratio of 2.0 in the high end of the range because the sampling was performed at a background location, away from primary combustion sources, with plenty of time for oxidation and secondary formation. However, during dust episodes, important fractions of the organic material have a soil origin and for that less reliable information exists. As the Carbonaceous fraction in PM mass is only 2-6%, errors in OM/OC ratio have only a small effect in total mass attribution, but important errors, of the order of 40%, can result in the estimation of the Carbonaceous mass if a OM/OC ratio of 1.6 is the correct assumption, instead of 2.0.

## 5 Conclusions

10 Atmospheric aerosol was collected during one year, as PM<sub>10</sub>, in air masses transported from North Africa to Cape Verde islands and submitted to total mass, elemental, ionic and carbon content analysis. Two clear different aerosol seasons were observed, one in December-February, with frequent intrusions of dust from Africa (denominated Dry-Haze), and other in May-September without direct African dust contamination (Dust-Free).

15 The application of IMB to the collected aerosol permitted the determination of 6 to 7 source classes: insoluble dust, soluble dust, sea-salt, secondary inorganic compounds from the reaction of atmospheric acidic precursors with sea-salt, dust and ammonia, and carbonaceous matter.

20 The sum of calculated components only partially closed the mass balance, being 20-30% of the measured PM<sub>10</sub> total mass unaccounted. Consideration and estimation of particulate water content, based in bibliographic and thermodynamic assumptions, permitted an almost total closure of the mass balance. This outcome is diverse from most previous mass balance studies, such those referred in the section Mass Balance Methodologies, where mass closure was only partial. Therefore the present IMB methodology permits a more well based mass account and apportionment of formation processes and sources.

During the Dry-Haze season dust contributed with around 80% to the aerosol mass loading, while in the Dust-Free period the main aerosol component was sea-salt that constituted approximately 50% of the aerosol mass.

25 The IMB methodology was compared with PMF results applied to the same data set. The outcomes were quite similar in most cases, for both seasons, with IMB being able to discriminate sources for which PMF has difficulties in separating correctly. This which permits to gain confidence in the joint capability of the two independent source apportionment approaches in evaluating the sources and processes responsible for particulate pollution in conditions of high contributions from natural desert dust and sea spray.

30 Otherwise, source composition and contribution knowledge obtained with IMB can be used to constrain a model run with the new versions of the PMF model in order to get more refined solutions than the original ones. Constraints can be created using specific ratios between two different species or mass balance equations derived from IMB techniques such as those performed in this study.

## Acknowledgements

The authors gratefully acknowledge the Portuguese Science Foundation through the project CV-DUST - Atmospheric aerosol in Cape Verde region: seasonal evaluation of composition, sources and transport (PTDD/AAC-CLI/100331/2008) and the PhD scholarship of João Cardoso. C2TN/IST/ULisboa authors gratefully acknowledge the FCT support through the  
5 UID/Multi/04349/2013 project.

## References

- Alastuey, A., Querol, X., Castillo, S., Escudero, M., Avila, A., Cuevas, E., Torres, C., Romero, P.-M., Exposito, F., García, O., Pedro Diaz, J., Dingenen, R.V. and Putaud, J.P.: Characterisation of TSP and PM<sub>2.5</sub> at Izaña and Sta. Cruz de Tenerife (Canary Islands, Spain) during a Saharan Dust Episode (July 2002), *Atmos. Environ.* 39, 4715–4728, doi:  
10 10.1016/j.atmosenv.2005.04.018, 2005.
- Almeida, S.M., Pio, C.A., Freitas, M.C., Reis, M.A. and Trancoso, M.A.: Source apportionment of atmospheric urban aerosol based on weekdays / weekend variability: evaluation of road re-suspended dust contribution, *Atmos. Environ.*, 40, 2058-67, doi: 10.1016/j.atmosenv.2005.11.046, 2006a.
- Almeida, S.M., Pio, C.A., Freitas, M.C., Reis, M.A. and Trancoso, M.A.: Approaching PM<sub>2.5</sub> and PM<sub>2.5-10</sub> source apportionment  
15 by mass balance analysis, principal component analysis and particle size distribution, *Sci. Total Environ.*, 368, 663-74, doi: 10.1016/j.scitotenv.2006.03.031, 2006b.
- Almeida-Silva, M., Almeida, S.M., Cardoso, J., Nunes, T., Reis, M.A., Chaves, P.C. and Pio, C.A.: Characterization of the aeolian aerosol from Cape Verde by k(0)-INAA and PIXE, *J. Radioanal. Nucl. Chem.*, 300, 2, 629-635, doi: 10.1007/s10967-014-2957-9, 2014.
- 20 Almeida-Silva, M., Almeida, S.M., Freitas, M.C., Pio, C.A., Nunes, T. and Cardoso, J.: Impact of Sahara dust transport on Cape Verde atmospheric element particles. *J. Toxicol. Env. Heal. A*, 76, 240-251, doi: 10.1080/15287394.2013.757200, 2013.
- Almeida, S.M., Freitas, M.C., Reis, M., Pinheiro, T., Felix, P.M. and Pio, C.A.: Fifteen years of nuclear techniques application to suspended particulate matter studies, *J. Radioanal. Nucl. Chem.*, 297, 347-356, doi: 10.1007/s10967-012-2354-1, 2013
- Amato, F., Alastuey, A., Karanasiou, A., Lucarelli, F., Nava, S., Calzolari, G., Severi, M., Becagli, S., Gianelle, V.L., Colombi,  
25 C., Alves, C., Custódio, D., Nunes, T., Cerqueira, M., Pio, C., Eleftheriadis, K., Diapouli, E., Reche, C., Minguillón, M.C., Manousakas, M., Maggos, T., Vratolis, S., Harrison, R.H. and Querol, X.: AIRUSE-LIFE+: a harmonized PM speciation and

- source apportionment in 5 Southern European cities, *Atmos. Chem. Phys.*, 16, 3289-3309, doi: 10.5194/acp-16-3289-2016, 2016.
- Andrews, E., Saxena, P., Musarra, S., Hildemann, L.M., Koutrakis, P., McMurry, P.H., Olmez, I. and White, W.H.: Concentration and Composition of Atmospheric Aerosols from the 1995 SEAVS Experiment and a Review of the Closure  
5 between Chemical and Gravimetric Measurements, *J. Air & Waste Manage. Assoc.*, 50, 648-664, doi: 10.1080/10473289.2000.10464116, 2000.
- Ashbaugh, L.L., Myrup, L.O. and Flocchini, R.G.: A principal component analysis of sulfur concentrations in the western United States, *Atmos. Environ.*, 18, 783-791, doi: 10.1016/0004-6981(84)90262-2, 1984.
- Blanchard, C.: Methods for attributing ambient air pollutants to emission sources, *Annu. Rev. Energy Environ.*, 24, 329-365,  
10 doi: 10.1146/annurev.energy.24.1.329, 1999.
- Belis, C.A., Karagulian, F., Larsen, B.R. and Hopke, P.K.: Critical review and meta-analysis of ambient particulate matter source apportionment using receptor models in Europe, *Atmos. Environ.*, 69, 94-108, doi: 10.1016/j.atmosenv.2012.11.009, 2013.
- Belis, C.A., Larsen, B.R., Amato, F., El Haddad, I., Favez, O., Harrison, R.M., Hopke, P.K., Nava, S., Paatero, P., Prevot, A.,  
15 Quass, U., Vecchi, R. and Viana, M.: European Guide on Air Pollution Source Apportionment with Receptor Models, JRC Reference Report EUR 26080. Publication Office of the European Union, ISBN 978-92-79-32514-4, doi: 10.2788/9332, 2014.
- Brunekreef, B. and Fosberg, B.: Epidemiological evidence of effects of coarse airborne particles on health, *Eur. Respir. J.*, 26, 309-318, doi: 10.1183/09031936.05.00001805, 2005.
- Buseck, P.R. and Pósfai, M.: Airborne minerals and related aerosol particles: Effects on climate and the environment, *PNAS*,  
20 96, 3372-3379, doi: 10.1073/pnas.96.7.3372, 1999.
- Castillo, S., Moreno, T., Querol, X., Alastuey, A., Cuevas, E., Herrmann, L., Mounkaila, M. and Gibbons, W.: Trace element variation in size-fractionated African desert dusts, *J. Arid Environ.*, 72, 1034-1045, doi: 10.1016/j.jaridenv.2007.12.007, 2008.
- Chen, X. and Yu J.Z.: Measurement of organic mass to organic carbon ratio in ambient aerosol samples using a gravimetric technique in combination with chemical analysis. *Atmos. Environ.*, 41, 8857-8864, doi: 10.1016/j.atmosenv.2007.08.023,  
25 2007.
- Chiapello, I., Bergametti, G., Chatenet, B., Bousquet, P. and Santos Soares, E.: Origins of African dust transported over northeastern tropical Atlantic. *J. Geophys. Res.*, 102, 13,701-13,709, doi: 10.1029/97JD00259, 1997.

- Countess, R.J., Wolff, G.T. and Cadle, S.H.: The Denver Winter Aerosol: A Comprehensive Chemical Characterization, *J. Air Pollut. Control Assoc.*, 30, 1194–1200, doi: 10.1080/00022470.1980.10465167, 1980.
- Canepari, S., Farao, C., Marconi, E., Giovannelli, C. and Perrino C.: Qualitative and quantitative determination of water in airborne particulate matter, *Atmos. Chem. Phys.*, 13, 1193-1202, doi: 10.5194/acp-13-1193-2013, 2013.
- 5 Dick, W. D., Saxena, P., and McMurry, P. H.: Estimation of water uptake by organic compounds in submicron aerosols measured during the Southeastern aerosol and visibility study. *J. Geophys. Res.*, 105, 1471–1479, doi: 10.1029/1999JD901001, 2000.
- Eldred B (2003) Internal memo to IMPROVE Staff.  
[http://vista.cira.colostate.edu/improve/Publications/GrayLit/023\\_SoilEquation/Soil\\_Eq\\_Evaluation.pdf](http://vista.cira.colostate.edu/improve/Publications/GrayLit/023_SoilEquation/Soil_Eq_Evaluation.pdf).
- 10 Eltayeb, M.A.H., Injuk, J., Maenhaut, W. and Van Grieken, R.E.: Elemental Composition of Mineral Aerosol Generated from Sudan Sahara Sand, *J. Atmos. Chem.*, 40, 247–273, doi: 10.1023/A:1012272208129, 2001.
- El-Zanan, H.S., Zielinska, B., Mazzoleni, L.R., Hansen, D.A.: Analytical determination of the aerosol organic mass-to-organic carbon ratio, *J. Air Waste Manag. Assoc.*, 59(1), 58-69, doi: 10.3155/1047-3289.59.1.58, 2009.
- Ervens, B., Turpin, B.J. and Weber, R.J.: Secondary organic aerosol formation in cloud droplets and aqueous particles (aqSOA): a review of laboratory, field and model studies, *Atmos. Chem. Phys.*, 11, 11069–11102, doi: 10.5194/acp-11-11069-2011, 2011.
- Formenti, P., Andreae, M.O., Lange, L., Roberts, G., Cafineyer, J., Rajta, I., Maenhaut, W., Holben, B.N., Artaxo, P. and Lelieveld, J.: Saharan dust in Brazil and Suriname during the Large-Scale Biosphere-Atmosphere Experiment in Amazonia (LBA) - Cooperative LBA Regional Experiment (CLAIRE) in March 1998. *J. Geophys. Res.*, 106, 14,919-14,934, doi:  
20 10.1029/2000JD900827, 2001.
- Formenti, P., Elbert, W., Maenhaut, W., Haywood, J. and Andrea, M.O.: Chemical composition of mineral dust aerosol during Saharan Dust Experiment (SHADE) airborne campaign in the Cape Verde region, September 2000, *J. Geophys Res.*, 108, D18, 8576, doi: 10.1029/2002JD002648, 2003.
- Gama, C., Tchepel, O., Baldasano, J., Basart, S., Ferreira, J., Pio, C., Cardoso, J. and Borrego, C.: Seasonal patterns of Saharan  
25 dust over Cape Verde - a combined approach using observations and modelling, *Tellus B*, 67:24410, doi: 10.3402/tellusb.v67.24410, 2015.

- Genga, A., Ielpo, P., Siciliano, T. and Siciliano, M.: Carbonaceous particles and aerosol mass closure in PM<sub>2.5</sub> collected in a port city, *Atmos. Res.*, 183,245-254, doi: 10.1016/j.atmosres.2016.08.022, 2017.
- Ginoux, P., Prospero, J.M., Gill, T.E. Hsu, N.C. and Zhao, M.: Global-scale attribution of anthropogenic and natural dust sources and their emission rates based on MODIS deep blue aerosol products, *Rev. Geophys.*, 50, RG3005, doi: 10.1029/2012RG000388, 2012.
- 5
- Goodman, A.L., Underwood, G.M. and Grassian, V.H.: A laboratory study of the heterogeneous reaction of nitric acid on calcium carbonate particles, *J. Geophys. Res.*, 105, 29,053-29,064, doi: 10.1029/2000JD900396, 2000.
- Grigoratos, T., Samara, C., Voutsas, D., Manoli, E. and Kouras, A.: Chemical composition and mass closure of ambient coarse particles at traffic and urban-background sites in Thessaloniki, Greece, *Environ. Sci. Pollut. Res.*, 21, 7708-7722, doi: 10.1007/s11356-014-2732-z, 2014.
- 10
- Guieu, C., Loye-Pilot, M.-D., Ridame, C. and Thomas, C.: Chemical characterization of the Saharan dust end-member: Some biogeochemical implications for the western Mediterranean Sea, *J. Geophys. Res.*, 107, D15, doi: 10.1029/2001JD000582, 2002.
- Guinot, B., Cachier, H. and Oikonomou, K.: Geochemical perspectives from a new aerosol chemical mass closure, *Atmos. Chem. Phys.*, 7, 1657–1670, doi: 10.5194/acp-7-1657, 2007.
- 15
- Harrison, R.M., Jones, A.M. and Lawrence, R.G.: A pragmatic mass closure model for airborne particulate matter at urban background and roadside sites, *Atmos. Environ.*, 37, 4927–4933, doi: 10.1016/j.atmosenv.2003.08.025, 2003.
- Henry, R.C., Lewis, C.W., Hopke, P.K. and Williamson, H.J.: Review of receptor model fundamentals, *Atmos. Environ.*, 18, 1507-1515, doi: 10.1016/0004-6981(84)90375-5, 1984.
- 20
- Hopke, P.K.: *Receptor Modelling in Environmental Chemistry*. Wiley & Sons, NY, 1985.
- Japar, S.M., Szkarlat, A.C., Gorse, Jr. R.A., Heyerdahl, E.K., Johnson, R.L., Rau, J.A., and Huntzicker, J.J.: Comparison of Solvent Extraction and Thermal Optical Carbon Analysis Methods: Application to Diesel Vehicle Exhaust Aerosol, *Environ. Sci. Technol.*, 18, 231–234, doi: 10.1021/es00122a004, 1984.
- Journet, E., Balkanski, Y. and Harrison, S.P.: A new data set of soil mineralogy for dust-cycle modeling. *Atmos. Chem. Phys.*, 14, 3801–3816, doi: 10.5194/acp-14-3801-2014, 2014.
- 25

- Kandler, K., Schütz, L., Deutscher, C., Eber, M., Hofmann, H., Jäckel, S., Jaenicke, R., Knippertz, P., Lieke K., Massling, A., Petzold, A., Schladitz, A., Weinzierl, B., Wiedensohler, A., Zorn, S. and Weinbruch, S.: Size distribution, mass concentration, chemical and mineralogical composition and derived optical parameters of the boundary layer aerosol at Tinfou, Morocco, during SAMUM 2006, *Tellus B*, 61, 32–50, doi: 10.1111/j.1600-0889.2008.00385.x, 2009.
- 5 Kitamori, Y., Mochida, M., and Kawamura, K.: Assessment of the aerosol water content in urban atmospheric particles by the hygroscopic growth measurements in Sapporo, Japan, *Atmos. Environ.*, 43, 3416–3423, doi: 10.1016/j.atmosenv.2009.03.037, 2009.
- Lohmann, U. and Feichter, J.: Global indirect aerosol effects: a review, *Atmos. Chem. Phys.*, 5, 715–737, doi: 10.5194/acp-5-715-2005, 2005.
- 10 Mirante, F., Salvador, P., Pio, C., Alves, C., Artinano, B., Caseiro, A., and Revuelta, M.A.: Size fractionated aerosol composition at roadside and background environments in the Madrid urban atmosphere. *Atmos. Res.*, 138, 278–292. 1873-2895, doi: 10.1016/j.atmosres.2013.11.024, 2014.
- Malm, W.C., Sisler, J.F., Huffman, D., Eldred, R.A. and Cahill, T.A.: Spatial and seasonal trends in particle concentration and optical extinction in the United States. *J. Geophys. Res.*, 9, 1347-1370, doi:10.1029/93JD02916.
- 15 Mason B., Moore C.B.: *Principles of Geochemistry* (4<sup>th</sup> ed) Wiley & Sons, 1982.
- Moreno, T., Querol, X., Castillo, S., Alastuey, A. and Cuevas, E.: Geochemical variations in Aeolian mineral particles from the Sahara-Sahel dust corridor. *Chemosphere*, 65, 261–270, doi: 10.1016/j.chemosphere.2006.02.052, 2006.
- Nenes, A., Pandis, S.N. and Pilinis, C.: ISORROPIA: A new thermodynamic equilibrium model for multiphase multicomponent inorganic aerosols, *Aquat. Geoch.*, 4, 123–152, doi: 10.1023/A:1009604003981, 1998a.
- 20 Nenes, A., Pilinis, C. and Pandis, S.N.: Continued Development and Testing of a New Thermodynamic Aerosol Module for Urban and Regional Air Quality Models, *Atmos. Env.*, 33, 1553–1560, doi: 10.1016/S1352-2310(98)00352-5, 1998b.
- Paatero, P. and Tapper, U.: Positive matrix factorization: A non-negative factor model with optimal utilization of error estimates of data values. *Environmetrics* 5, 111–126, doi: 10.1002/env.3170050203, 1994.
- Paatero, P.: The multilinear engine—a table-driven least squares program for solving multilinear problems, including the n-
- 25 way parallel factor analysis model. *J. Comput. Graph Stat.*, 8, 854–888, doi: 10.1080/10618600.1999.10474853, 1999.

- Perrino, C., Canepari, S., Catrambone, M.: Comparing the Performance of Teflon and Quartz Membrane Filters Collecting Atmospheric PM: Influence of Atmospheric Water, *Aerosol Air Qual. Res.*, 13, 137–147, doi: 10.4209/aaqr.2012.07.0167, 2013.
- Pio, C., Cerqueira, M., Harrison, R., Nunes, T., Mirante, F., Alves, C., Oliveira, C., Sanchez de la Campa, A., Artfiano, B.,  
5 and Matos, M.: OC/EC Ratio Observations In Europe: Re-Thinking The Approach For Apportionment Between Primary And Secondary Organic Carbon, *Atmos. Environ.*, 45, 6121–6132, doi: 10.1016/j.atmosenv.2011.08.045, 2011.
- Pio, C.A. and Lopes, D.: Chlorine loss from marine aerosol in a coastal atmosphere. *J. Geophys. Res.*, 103, 25263–25269, doi: 10.1029/98JD02088, 1998.
- Pio, C.A., Ramos, M.O. and Duarte, A.C.: Measurement of Carbonates in Atmospheric Aerosols by Acidification and NDIR  
10 Analysis of Evolved CO<sub>2</sub>. In *Physico-Chemical Behaviour of Atmospheric Pollutants*. Edit Angeletti G. and Restelli G., Report EUR 15609/1 EN. European Commission. Brussels, Vol 1, pp 712-717, 1994.
- Pope, C.A. III : Review: Epidemiological basis for particulate air pollution health standards. *Aerosol Sci. Tech.*, 32, 4–14, doi: 10.1080/027868200303885, 2000.
- Poschl, U.: Atmospheric aerosols: Composition, transformation climate and health effects, *Angew Chem. Int. Ed.*, 44, 7520–  
15 7540, doi: 10.1002/anie.200501122, 2005.
- Raes, F., Van Dingenen, R., Vignati, E., Wilson, J., Putaud, J.P., Seinfeld, J.H. and Adams, P. : Formation and cycling of aerosols in the global troposphere, *Atmos. Environ.*, 34, 4215–4240, doi: 10.1016/S1352-2310(00)00239-9, 2000.
- Ramanathan, V., Crutzen, P.J., Kiehl, J.T. and Rosenfeld, D.: Aerosols, Climate, and the Hydrological Cycle, *Science*, 294, 2119–2124, doi: 10.1126/science.1064034, 2001.
- 20 Rees, S.L., Robinson, A.L., Khlystov, A., Stanier, C.O. and Pandis, S.N.: Mass Balance Closure and the Federal Reference Method for PM 2.5 in Pittsburgh, Pennsylvania, *Atmos. Environ.*, 38, 3305–3318, doi : 10.1016/j.atmosenv.2004.03.016, 2004.
- Reff, A., Eberly, S.I. and Bhave, P.V.: Receptor Modeling of Ambient Particulate Matter Data Using Positive Matrix Factorization: Review of Existing Methods, *J. Air Waste Man. Assoc.*, 57, 146–154, doi: 10.1080/10473289.2007.10465319,  
25 2007.



- Remoundaki, E., Papayannis, A., Kassomenos, P., Mantas, E., Kokkalis, P. and Tsezos, M.: Influence of Saharan Dust Transport Events on PM 2.5 Concentrations and Composition over Athens, *Water Air Soil Pollut.* 224: 1373, doi: 10.1007/s11270-012-1373-4, 2013.
- Rogge, W.F., Hildemann, L.M., Mazurek, M.A., Cass, G.R. and Simoneit B.R.T.: Sources of Fine Organic Aerosol. 4. Particulate Abrasion Products from Leaf Surfaces of Urban Plants, *Environ. Sci. Technol.*, 27, 2700–2711, doi: 10.1021/es00049a008, 1993a.
- Rogge, W.F., Mazurek, M.A., Hildemann, L.M., and Cass, G.R.: Quantification of Urban Organic Aerosols at a Molecular Level: Identification, Abundance and Seasonal Variation, *Atmos. Environ.*, 27, 1309–1330, doi: 10.1016/0960-1686(93)90257-Y, 1993b.
- 10 Russel L.M.: Aerosol Organic-Mass-to-Organic-Carbon Ratio Measurements. *Environ. Sci. Technol.*, 37 (13), 2982–2987, doi: 10.1021/es026123w, 2003.
- Salvador, P., Almeida, S.M., Cardoso, J., Almeida-Silva, M., Nunes, T., Cerqueira, M., Alves, C., Reis, M.A., Chaves, P.C., Artinano, B. and Pio, C.: Composition and origin of PM10 in Cape Verde: Characterization of long-range transport episodes, *Atmos. Environ.*, 127, 326–339, doi: 10.1016/j.atmosenv.2015.12.057, 2016.
- 15 Sciare, J., Oikonomou, K., Cachier, H., Mihalopoulos, N., Andreae, M. O., Maenhaut, W. and Sarda-Estève, R.: Aerosol mass closure and reconstruction of the light scattering coefficient over the Eastern Mediterranean Sea during the MINOS campaign, *Atmos. Chem. Phys.*, 5, 2253–2265, doi: 10.5194/acp-5-2253-2005, 2005.
- Scheuvens, D., Kandler, K., Ebert, M. and Weinbruch, S.: Bulk composition of northern African dust and its source sediments — A compilation, *Earth-Sci. Rev.*, 116, 170–194, doi: 10.1016/j.earscirev.2012.08.005, 2013.
- 20 Seinfeld, J.H. and Pandis, S.N.: *Atmospheric Chemistry and Physics: From Air Pollution to Climate Change*, 2<sup>nd</sup> Edition available in Wiley, Hoboken, NJ, ISBN: 978-0471178163, 1998.
- Sempere, R., and Kawamura, K.: Comparative Distributions of Dicarboxylic Acids and Related Polar Compounds in Snow, Rain and Aerosols from Urban Atmosphere, *Atmos. Environ.*, 28:449–459, doi: 10.1016/1352-2310(94)90123-6, 1994.
- Schuttlefield, J. D., Cox, D. and Grassian, V. H.: An Investigation of water uptake on Clays Minerals Using ATR-FTIR Spectroscopy Coupled with Quartz Crystal microbalance measurements, *J. Geophys. Res.*, 112, D21303, doi: 10.1029/2007JD008973, 2007.
- 25

- Speer, R. E., Barnes, H. M., and Brown, R.: An instrument for measuring the liquid content of aerosols, *Aerosol Sci. Technol.*, 27, 50–61, doi: 10.1080/02786829708965457, 1997.
- Speer R.E., Edney E.O. and Kleindienst T.E.: Impact of organic compounds on the concentrations of liquid water in ambient PM<sub>2.5</sub>, *Aerosol Science*, 34, 63–77, doi: 10.1016/S0021-8502(02)00152-0, 2003.
- 5 Stanier, C., Khlystov, A., Chan, W. R., Mandiro, M., and Pandis, S. N.: A method for the in-situ measurement of aerosol water content of ambient aerosols: The Dry-Ambient Aerosol Size Spectrometer (DAASS), *Aerosol Sci. Technol.*, 38, 215-228, doi: 10.1080/02786820390229525, 2004.
- Tanaka, T.Y. and Chiba, M.: A numerical study of the contributions of dust source regions to the global dust budget. *Global Planet Change*, 52, 88-104, doi: 10.1016/j.gloplacha.2006.02.002, 2006.
- 10 Tang, I.N., Munkelwitz, H.R.: Water activities, densities, and refractive indices of aqueous sulfates and sodium nitrate droplets of atmospheric importance. *J. Geophys. Res.*, 99, 18801–18808, doi: 10.1029/94JD01345, 1994.
- Tang, I.N., Tridico, A.C. and Fung, K.H.: Thermodynamic and optical properties of sea salt aerosols. *J. Geophysical Res.*, 102, 23,269-23,275, doi: 10.1029/97JD01806, 1997.
- Tobias, A., Perez, L., Diaz J., Linares, C., Pey, J., Alastuey, A., Querol, X.: Short term effects of particulate matter on total mortality during Saharan dust outbreaks: a case-crossover analysis in Madrid (Spain), *Sci. Total Environ.* 412–413, 386-389, doi: 10.1016/j.scitotenv.2011.10.027, 2011.
- 15 Turekian, K.K.: *Oceans*. Prentice-Hall, 1968.
- Turpin, B.J. and Lim, H.J.: Species Contributions to PM 2.5 Mass Concentrations: Revisiting Common Assumptions for Estimating Organic Mass, *Aerosol Sci. Technol.*, 35, 602–610, doi: 10.1029/97JD01806, 2001.
- 20 Tsyro, S.G.: To what extent can aerosol water explain the discrepancy between model calculated and gravimetric PM 10 and PM 2.5, *Atmos. Chem. Phys.* 5, 515–532, doi: 10.5194/acp-5-515-2005, 2005.
- Watson, J.G., Zhu, T., Chow, J.C., Engelbrecht, J., Fujita, E.M. and Wilson, W.E.: Receptor modeling application framework for particle source apportionment. *Chemosphere*, 49, 1093-1136, doi:10.1016/S0045-6535(02)00243-6, 2002.
- Wedepohl, K.H.: The composition of the continental crust. *Geochim. Cosmochim. Ac.*, 59, 1217-1232, doi: 10.1016/0016-25 7037(95)00038-2, 1995.

Xu, J., Imre, D., McGraw, R. and Tang, I.: Ammonium sulphate: Equilibrium and metastability phase diagrams from 40 to -50 °C. *J. Phys. Res. B*, 102, 7462–7469, doi: 10.1021/jp981929x, 1998.

## ANNEX

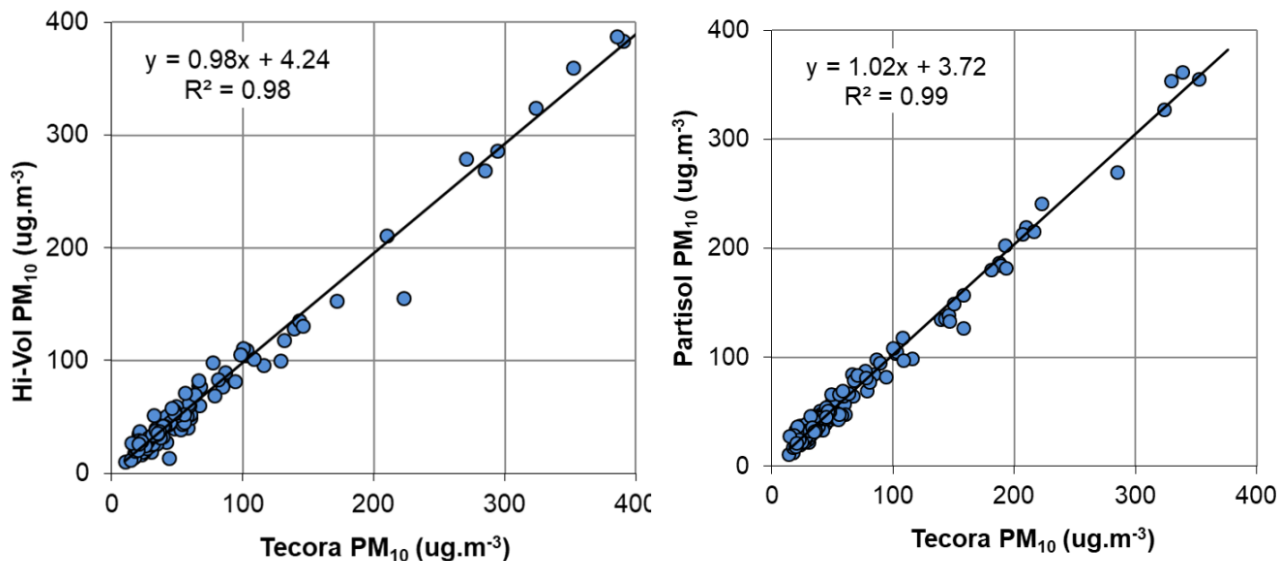


Figure A1: Comparison between PM<sub>10</sub> concentration values, in atmospheric aerosol parallel sampling with Nuclepore (Tecora),  
5 Teflon (Partisol) and Quartz (Hi-Vol) filters.

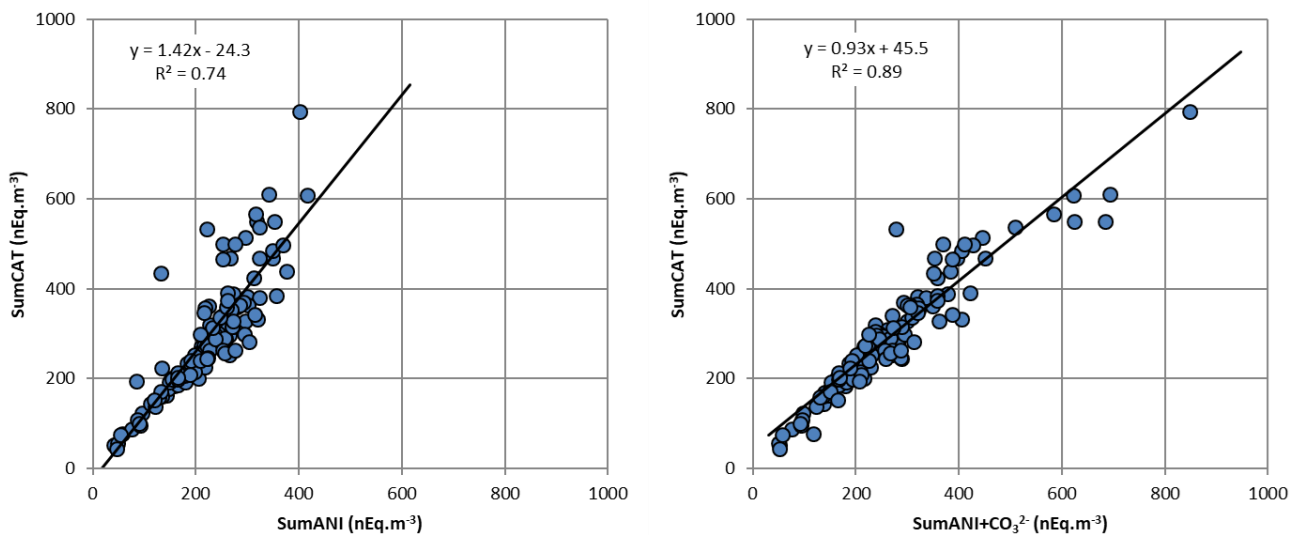


Figure A2: Ion balance between total analysed cations and anions, without and with the inclusion of measured carbonates (ANI- anions; CAT- cations).

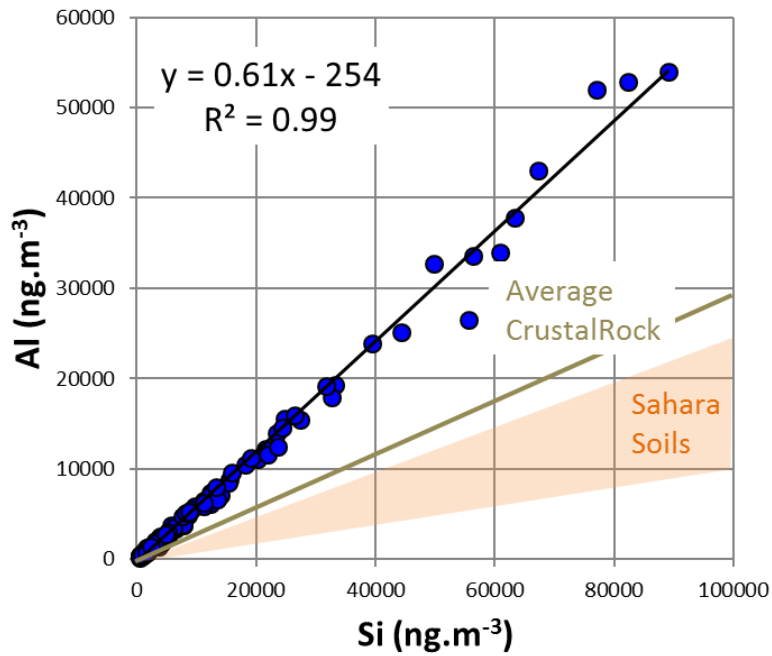
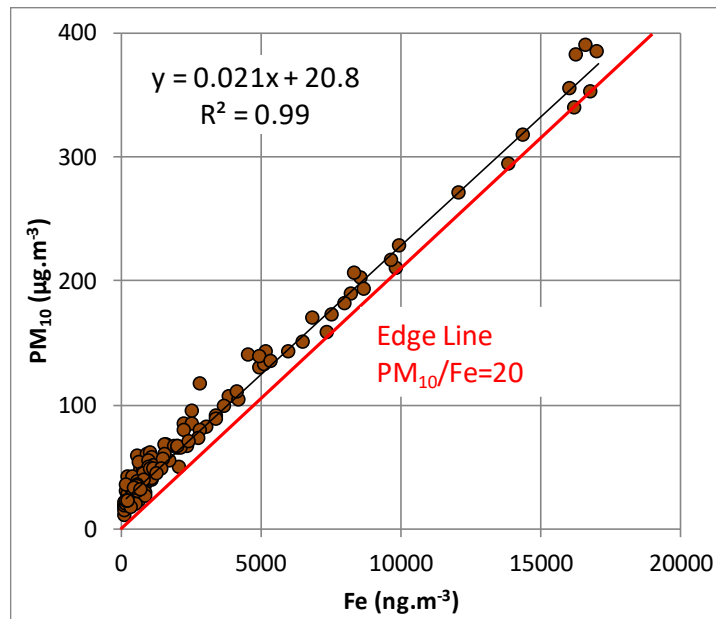


Figure A3: Relation between Si and Al concentrations in atmospheric aerosols. Also shown Al/Si ratio ranges, in Sahara, and average value for global soils, taken from Moreno et al., (2006), and Manson and Moore, (1982), respectively.



5

Figure A4: Relation between Fe and  $\text{PM}_{10}$  concentrations. The red line is the low edge representing, approximately, the amount of dust in  $\text{PM}_{10}$

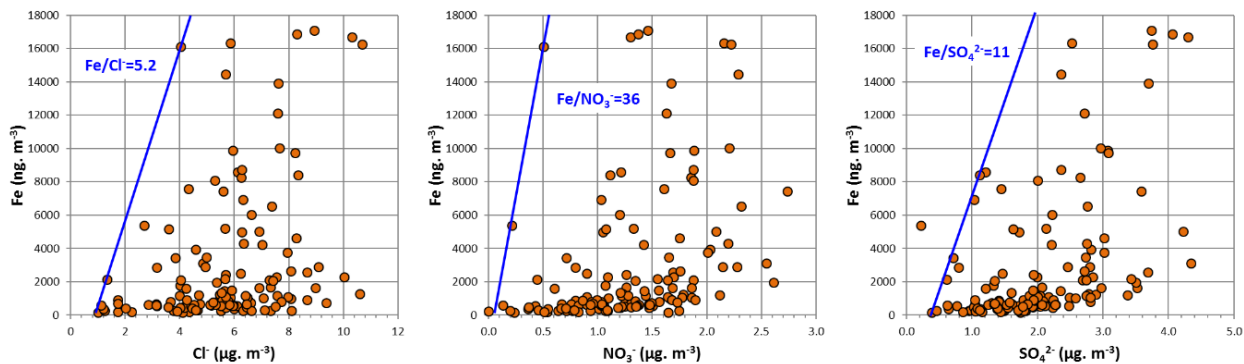
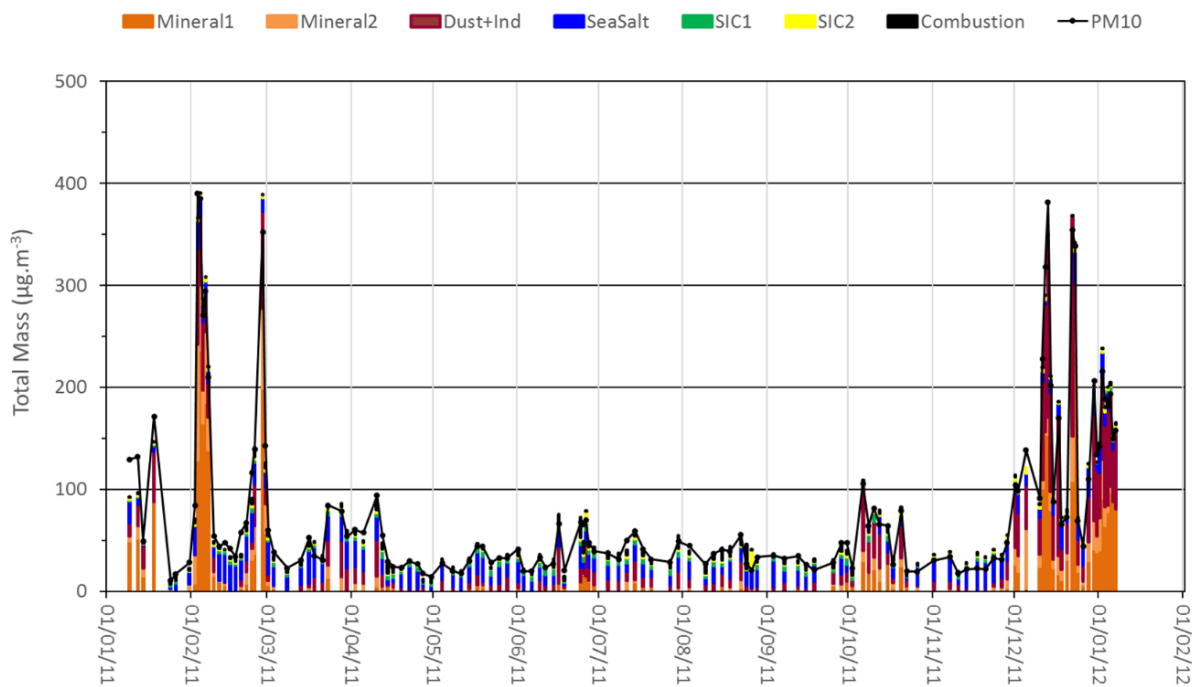


Figure A5: Edge lines for chloride, nitrate and sulphate, representing, coarsely, the amount of these anions in soil dust.



5 Figure A6: Source contributions for individual sampling events, calculated with PMF

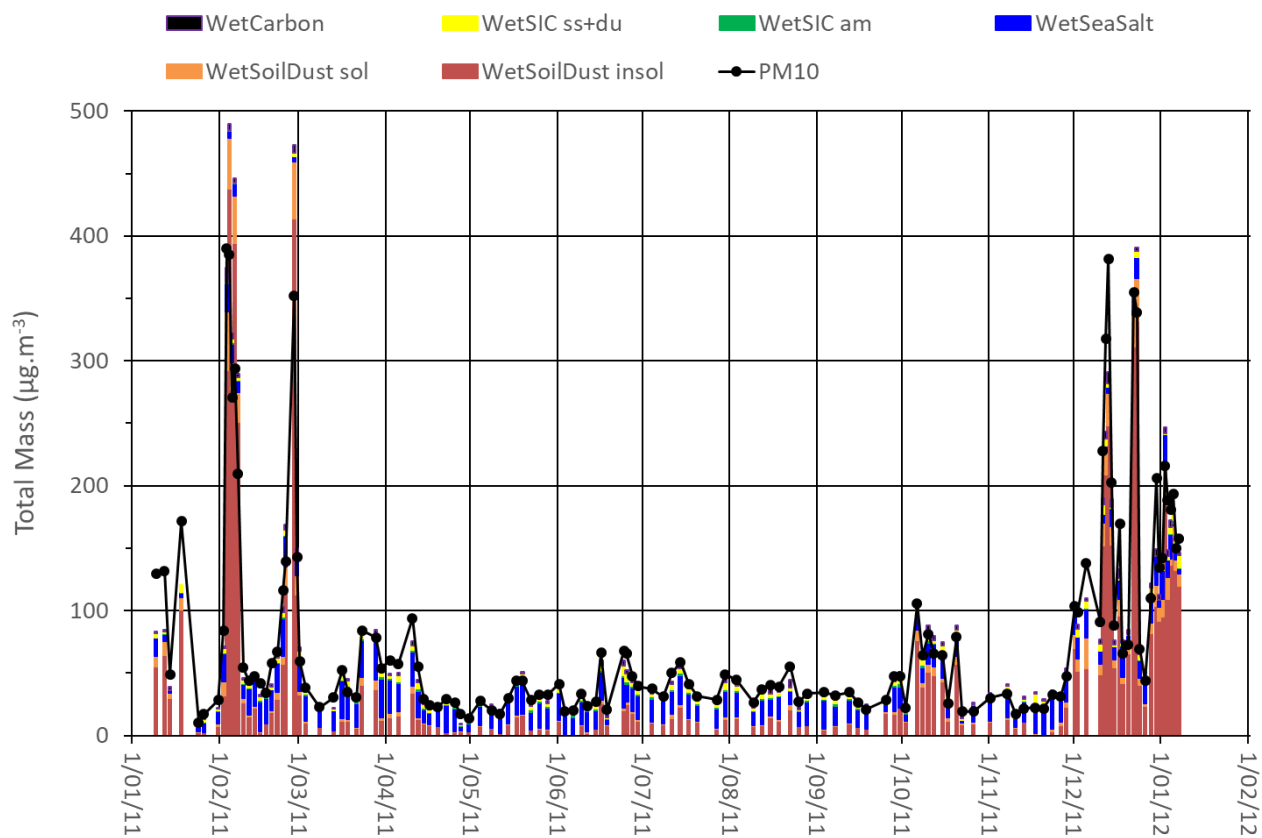


Figure A7: Source contributions for individual sampling events, calculated with the IMB methodology.

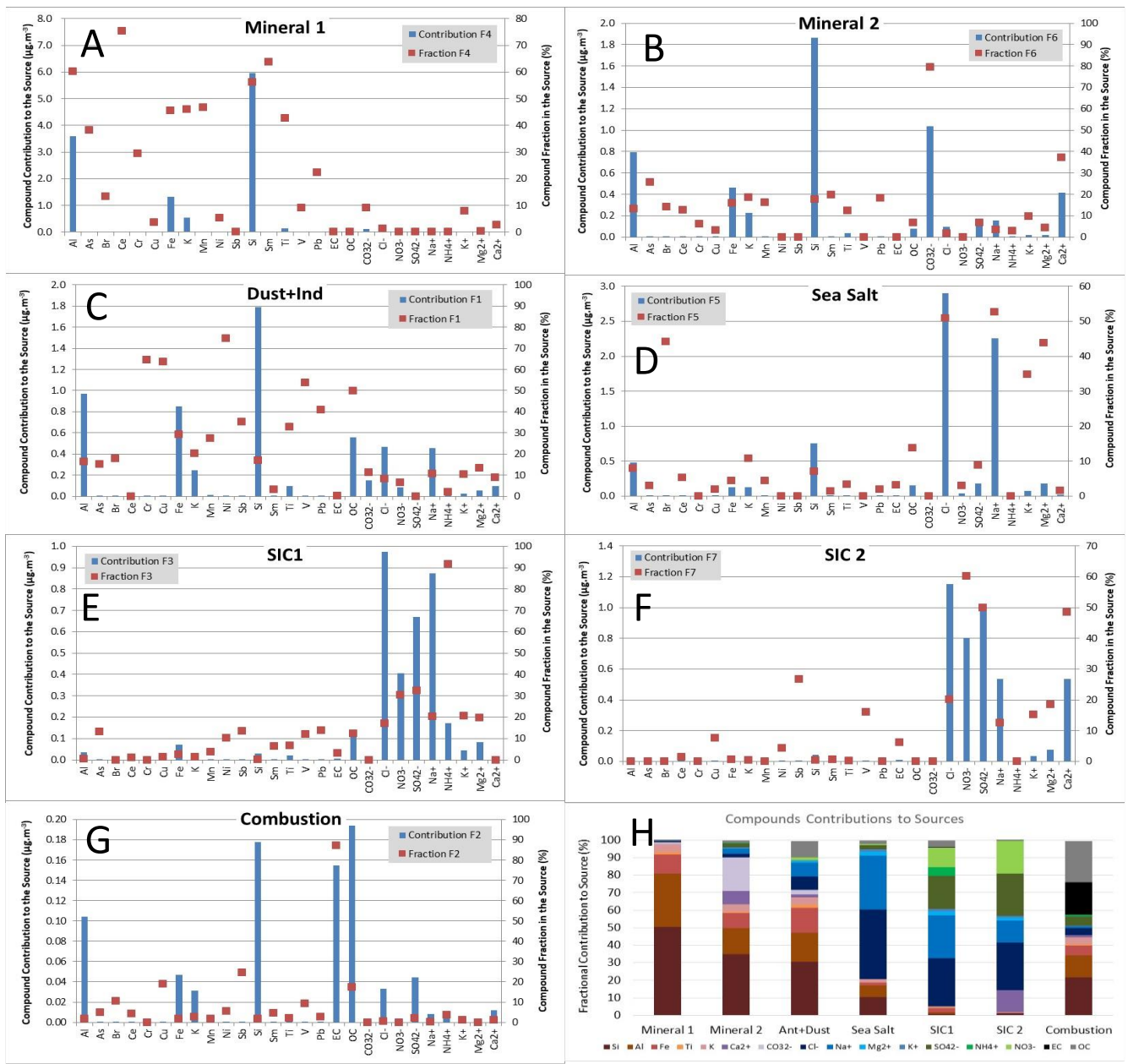


Figure A8: PMF Source Contributions and Profiles for the seven identified factors ( Graphs A-G). Graph H presents the fraction contribution of each main compound to each source.



Table A1: Concentrations of major elemental species (as oxides) at nine sites in Sahara region, according to Moreno et al., (2006), and average crustal levels taken from Manson and Moore, (1982). The table presents also the estimation of factor F used in Equation 3.

	CrustalAver	Hoggar Massif		Chad Basin		Niger		Western Sahara			Sahara
	HM1	HM2	CB1	CB2	MON	HAR	WS1	WS2	WS3	Average	
	(%)	(%)	(%)	(%)	(%)	(%)	(%)	(%)	(%)	(%)	(%)
SiO <sub>2</sub>	59.3	62.1	62.5	63.2	70.2	69.5	66.7	49.4	46.2	56.8	60.7
TiO <sub>2</sub>	0.7	1.4	1.2	1.0	1.2	1.4	1.2	0.6	0.4	0.9	1.0
Al <sub>2</sub> O <sub>3</sub>	15.4	14.7	13.6	14.0	11.7	11.6	12.2	8.8	7.0	5.1	11.0
Fe <sub>2</sub> O <sub>3</sub>	7.2	6.1	5.2	6.7	4.7	4.4	5.7	4.4	2.8	4.2	4.9
MnO	0.1	0.1	0.1	0.0	0.1	0.0	0.1	0.1	0.1	0.1	0.1
MgO	3.5	1.8	1.6	1.2	0.8	0.5	0.9	2.9	2.6	1.9	1.6
CaO	5.1	2.4	2.0	1.4	1.4	0.4	1.6	12.9	17.7	12.2	5.8
K <sub>2</sub> O	3.1	2.5	2.5	1.5	1.5	1.2	1.8	1.9	1.7	1.4	1.8
<b>SUM</b>	<b>94.4</b>	<b>91.0</b>	<b>88.6</b>	<b>89.0</b>	<b>91.5</b>	<b>88.9</b>	<b>90.1</b>	<b>80.9</b>	<b>78.5</b>	<b>82.6</b>	<b>86.8</b>
Na <sub>2</sub> O	3.8	1.6	1.9	1.1	0.5	0.4	0.6	0.9	0.7	0.8	0.9
P <sub>2</sub> O <sub>5</sub>	3.1	2.5	2.5	1.5	1.5	1.2	1.8	1.9	1.7	1.4	1.8
SO <sub>3</sub>	0.1	0.2	0.2	0.3	0.2	0.0	0.1	0.1	0.1	0.1	0.1
<b>TOTAL</b>	<b>98.5</b>	<b>93.1</b>	<b>91.0</b>	<b>90.6</b>	<b>92.3</b>	<b>89.4</b>	<b>91.0</b>	<b>82.2</b>	<b>79.7</b>	<b>83.9</b>	<b>88.1</b>

$$F=100/86.8= 1.15$$