Answers to the report by anonymous Referee #1

We thank Referee 1 for reviewing our manuscript and also for useful hints and suggestions. Below, comments from the referee are given in blue while our answers are given in black, with passages including new text given in italic. Additionally, the new text is marked yellow in the revised version of the manuscript.

General Comment:
This manuscript reports the ice nucleating abilities of urban aerosol particles from Beijing (China) using two different ice nuclei counters (i.e., LINA and INDA) during "clean" and heavy haze episodes. The authors did not find a major influence of the air pollution in Beijing on the ice nucleating particle (NP) concentrations as the INP levels did not correlate neither with PM2.5 nor with black carbon (BC) concentrations. Additionally, the predicted INP concentrations from the concentration of aerosol particles larger than 500 nm using DeMott et al. (2010) and DeMott et al. (2015) parametrizations did not correlate with the measured INP concentrations. The authors suggest that the INP concentrations in Beijing may have biogenic sources or non-urban dust and that is why high levels of pollution did not increase the INP concentrations. This is one of the very few studies that measures the INP concentrations in an urban location, and it could be useful to help the ice nucleation community to understand the role that urban aerosol could play in ice cloud formation. However, the paper requires major corrections before it can be accepted for its publication in ACP.

Major Comments:
1. Given that the "insignificant" influence of air pollution in the INP concentrations is not clearly supported I suggest to soften the tone of the Title. How about: "Ice nucleating particle concentrations under urban air pollution in Beijing, China"

Thank you for your suggestion, but we prefer not to change our title since we do think that our title is well supported by our results. Combined with the newly added information on meteorology, our laboratory results and different parameterizations, we show that the number concentration of INP has no correlation to some vital components of urban aerosol, where clean and heavily polluted days were examined. These components include, for example, BC concentration, particle number concentration and $PM_{2.5}$ mass concentration. We hope that you can agree to this for the new revised version of the manuscript.

2. The reviewer is surprised the authors completely ignored meteorology in this study. A detailed analysis of the meteorological variables and air masses is required to explain ambient observations, even in urban areas.

We added two plots showing trajectories and also wind direction and wind speed, to show the meteorological condition during the sampling period, together with the following text (line 234-245):

"*Additionally, Fig.2 shows 2-day back-trajectories obtained by the NOAA HYSPLIT model, with one trajectory related to each sampled filter, starting at the median sampling time of each filter. Fig. 3 shows minutely recorded data for wind-direction and wind–speed collected by (Met One 591) and (Met One 590) located on the same roof top as the aerosol sampling equipment. Both pictures are colored-coded with respect to $PM_{2.5}$ mass concentrations. The air masses that came from north or north-western directions were generally coincident with higher wind-speeds. They brought clean air with lower $PM_{2.5}$ mass concentrations. They did cross desert regions, however, Beijing was reported to be affected by desert dust in mainly only spring (Wu et al., 2009). Typically, the air masses coming from south and south-west of*

Beijing moved slowly and spent much more time over industrialized regions, resulting in high particulate matter mass concentrations. This here observed pattern is typical for Beijing, and these connections between wind-direction and pollution levels in Beijing have been analyzed in detail previously in Wehner et al. (2008)."

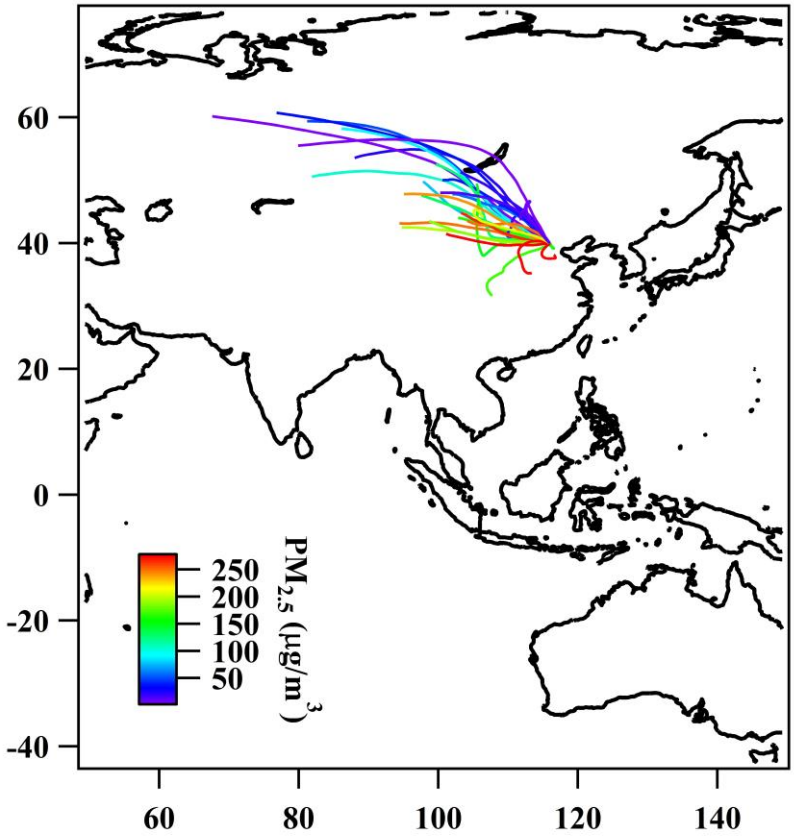

Figure 2. The 2-day back-trajectories obtained by the NOAA HYSPLIT model colored-coded with respect to $PM_{2.5}$ mass concentration determined by PTEF filter.

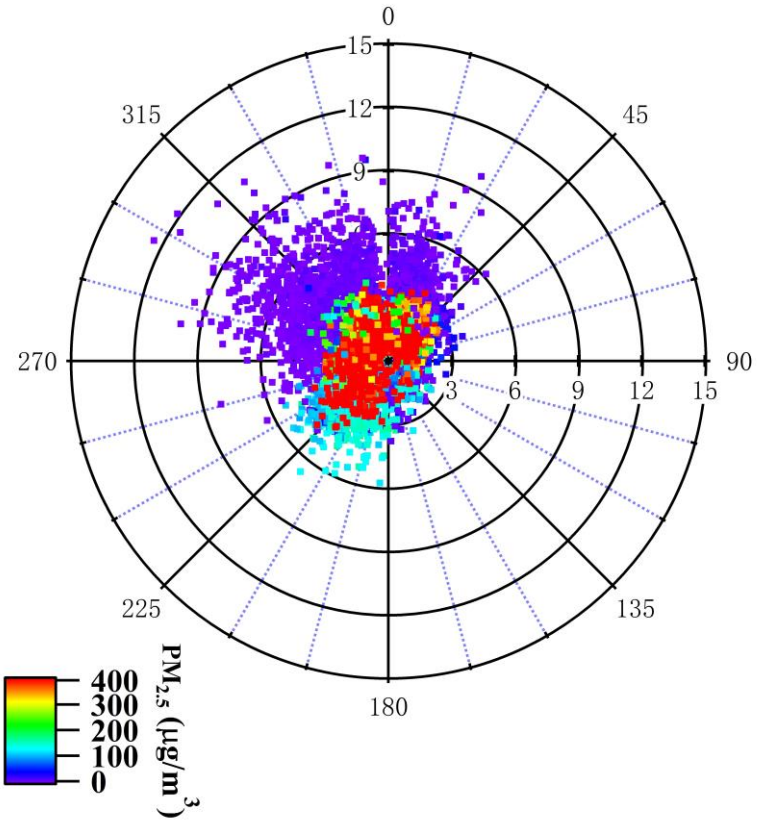

Figure 3. Minutely recorded data for wind-direction and wind-speed colored-coded with respect to $PM_{2.5}$ mass concentration.

3. I am not sure if the comparison of the BC and INP concentration is completely fair given that the INP concentrations were obtained from particles collected on 8-h filters, while the BC data was obtained in-situ. Is it possible that the BC particles collected of the filters may have change their ice nucleating abilities during the 8-h period (i.e., aging, coagulation, oxidation, and coating)?

It is right that particles on the filters might age. But both BC measurements, in-situ and from filters, can be expected to yield the same results, as those properties of BC that would be detected from a filter measurement will not change with aging on the filters. So in general, the concern here would not be the comparison of 12-h-filter samples to averages of much more highly resolved aerosol data, but aging of BC such that a possible ice activity of the BC might be destroyed in general. Aging could happen due to loss of semi-volatile materials, such as ammonium nitrate, or through oxidation. A loss of semi-volatile material from BC is not to be expected. Concerning oxidation, during wintertime, atmospheric oxidants such as OH radicals and ozone typically have low concentrations. Ozone measurements done in parallel to our sampling showed that during nighttime, concentrations of ozone, one of the most important oxidants, was close to zero. During daytime, the ozone concentration was below 20 ppbv, indicating the oxidation capacity was very weak during our sampling in general. For coating, we assume you aim at the formation of SOA from the gas phase. These would be substances that would dissolve during INDA measurements when the sampled filters get in contact with the water, or when washing off the Teflon filters for LINA measurements. In general, during INDA measurements, water can be expected to surround all available surfaces, so particles being located close to each should not noticeably reduce the results, either.

Summarizing, we expect that the aging of BC particles collected on filters or other sampling artifacts concerning BC can be ignored for our study for INDA measurements. Concerning LINA (i.e., washing off the filters), see our answer to your minor comment 8.

4. It is mentions that secondary particle formation did not contribute to the INP concentrations. However, it is complete unclear how secondary particles were measured or identified in this study.

According to the chemical composition analysis in Figure 1, there was a notable increase in sulfate, nitrate and ammonium during times with high pollution. These substances are typically present in increased concentrations during secondary formation of particulate matter as shown in many studies, e.g., Zheng et al. (2016) and Guo et al. (2010). This formation of sulfate, nitrate and ammonium (SNA in short) is what we refer to as secondary formation of particulate matter (see the previous version of the manuscript, line 209). This and other related mentions might have been misleadingly formulated in the first version of our manuscript. We did the following changes:

formerly line 212, now line 221: We added "as it has previously been described in Guo et al. (2010) and Zheng et al. (2016). In this study, when we refer to secondarily formed particulate matter, this will always stand mainly for SNA and secondary organic substances."

formerly line 211, now line 220: We replaced "secondary transformation could be" by "*generation of secondary particulate mass is*"

formerly line 220, now line 231: We replaced "secondary and primary organic aerosols" by "*secondarily and primarily formed organic particulate matter*"

formerly line 292, now line 340: We replaced "secondary formation" by "*formation of secondary particulate matter*"

formerly line 316, now line 370: We replaced "secondary processes" by "*formation of secondary particulate matter*"

At the end of section 3.3, we explicitly state: "*Additionally, also no correlation was found between any of the water-soluble constituents that were analyzed with ion chromatography and INP concentrations.*" This means that also none of the components that formed secondary particulate mass contributed to INP.

5. The ice nucleation activation scans from the INDA and LINA are directly compared. However, given that the operational principle from both instruments is different, I am wondering if this is a fair comparison. Did the PTFE and quartz filters collect the same particles mass? Would it be necessary to normalize the INPs concentrations?

The samplers sampling filters for INDA and LINA were installed behind the same air inlet on a roof top, and the sampling length (12 hours) and times for filter changes on both samplers were the same. Therefore, it can be safely assumed that they sampled the same air mass. Please check what was already written in the manuscript (line 114 to 116 in the previous version, line 121 now): "Particles with an aerodynamic diameter less than or equal to 2.5 micro-meters (PM2.5) were collected on quartz fiber (Whatman, 1851-865) and PTFE filters (Whatman,7592-104) using a 4-channel sampler with 2.5µm impactors …"

If we would compare the measured frozen fractions obtained from INDA and LINA directly, indeed, the results could not be compared directly. However, by applying equation (2) (line 195 in the previously

submitted version, now line 204), a normalization is done with respect to the amount of air sampled per examined droplet. Therefore, the parameter we compare is the number of INP per volume of collected air for each of the methods. This makes data from the different methods directly comparable. Nothing changed.

6. There are not uncertainties reported in this study at all. The correlations performed in this study do not present any statistical analysis.

Thank you for this hint. We prefer to not add error bars to the figures in the main text as this will make them unnecessarily messy and it would be difficult to see anything. Instead, we added a figure and text to the appendix and we also added error bars to the time series of N_{INP} that is now shown in Fig. 7 (see our comment to your minor comment 2). Additionally we added data on the statistical analysis you were asking for.

In detail:

Uncertainties were added for measured frozen fractions in a new plot that appears in the appendix, together with descriptive text and the figure caption:

"*The highest and lowest freezing curved detected with INDA are shown exemplarily in Fig. A3 together with the measurement uncertainty. The derivation of the uncertainty was based on the fact that at each temperature, all INP that are ice active at that or any higher temperature are Poission distributed to the examined droplets. It followed a method described in Harrison et al. (2016). For LINA, no uncertainties are given, as we know that washing off from the filters was incomplete, and the fraction of particles that was retained on the filters cannot be determined. The largest deviation that we allowed between LINA and INDA, i.e., a factor of 4.4 (see Sec. 3.2), is the base for the maximum uncertainty for f_{ice} detected with LINA. For both, INDA and LINA, the temperature uncertainty is 0.5K.*"

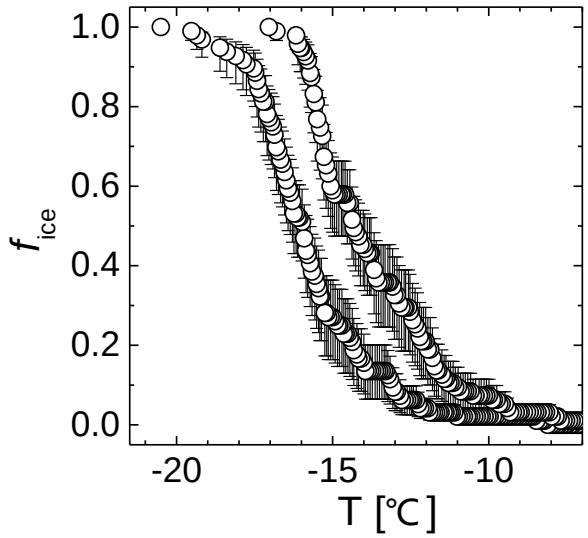

Fig. A3. The highest and lowest freezing curved detected with INDA together with the measurement uncertainty.

As far as a statistical analysis is concerned, we added the following table, giving R^2 and p values for the 6 scatter-plots presented in what was Fig. 4 (now Fig. 6) and we added the respective lines to the figure (see

below), together with some text. It can clearly be seen by the values given in the newly added table, that there is no correlation.

The following text was added:
"*Linear fits are included in all panels of Fig. 6, and values for R^2 and p for these fits are shown in Table 1.*" (line 305)

"*Also the R^2 and p values given in Table 1 clearly show that there was no correlation between N_INP and any of the examined parameters.*" (line 309)

Table 1: Coefficient of determination (R2) and a measure for the statistical significance of the assumption of a linear correlation (p) for the comparison of N_{INP} at -16°C with the different parameters shown in Fig.6.

parameter	**R^2**	**p**
(a) BC concentration	0.003	0.79
(b) $PM_{2.5}$ concentration	0.006	0.71
(c) N_{total}	0.005	0.73
(d) $N_{>500nm}$ at -16°C	0.008	0.67
(e) N_{INP} at -16°C, based on DeMott et al. (2010)	0.005	0.73
(f) N_{INP} at -16°C, based on DeMott et al. (2015)	0.007	0.67

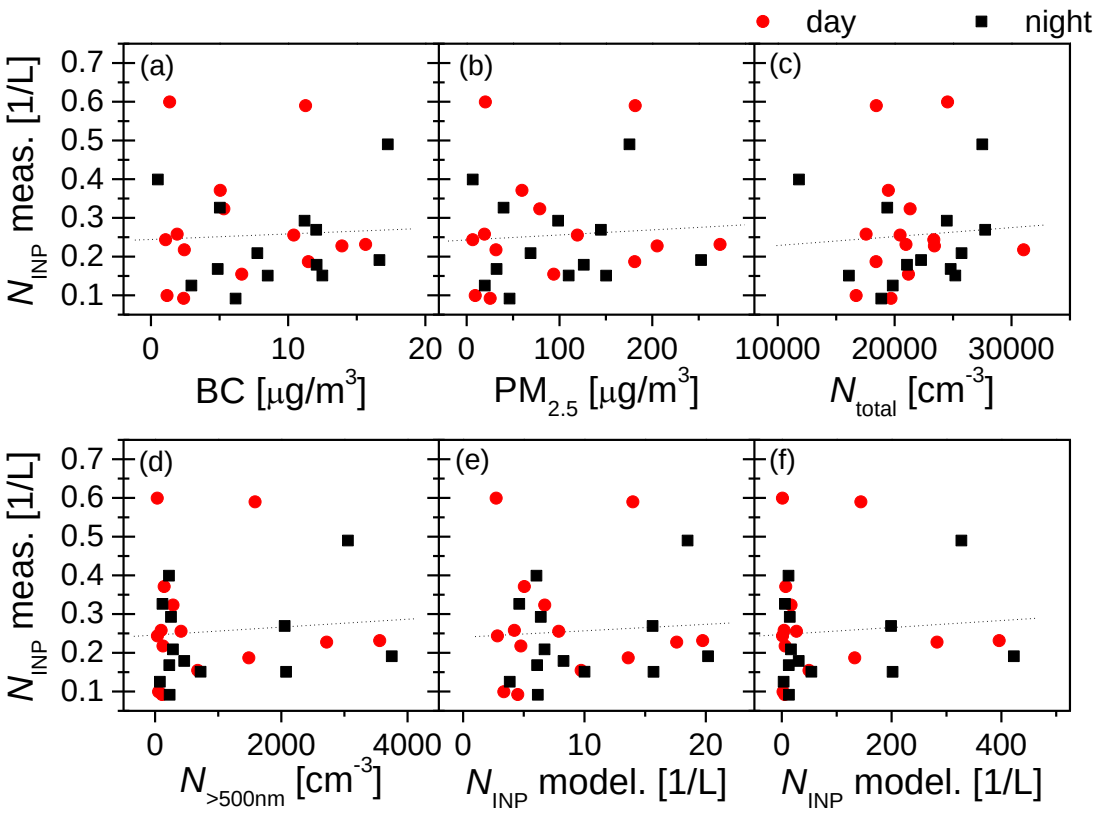

Figure 6. N_{INP} **at -16°C as function of mass concentrations of BC (a) and** $PM_{2.5}$ **(b), and of 12h-average values of** N_{total} **(c). Furthermore, we show** $N_{>500nm}$ **(d), and** N_{INP} **at -16°C derived based on (DeMott et al., 2010) (e) and DeMott et al. (2015) (f) for daytime (red circles) and nighttime (black squares) samples.**

7. I am not sure why the results from this study were compared with the Petters and Wright (2015) precipitation data. I would rather compare the present data with the results from Knopf et al. (2010) and Corbin et al. (2012) that were obtained in urban ambient air instead of precipitation samples.

In the study of Petters and Wright (2015), INP concentrations obtained per volume of precipitation are converted to INP concentrations per volume of air, i.e., to the same parameter we derive from our samples. Petters and Wright (2015) explain the uncertainties in their assumptions due to this conversion at great length, and we feel it is justified to use these data for the kind of comparison we are doing here. Also, Petters and Wright (2015) offer one of the largest compilations on atmospheric INP concentrations that we are aware off.

Concerning the publications from Knopf et al. (2010) and Corbin et al. (2012), as we already said in the introduction of the previously submitted version, their results were obtained for water sub-saturated conditions (see also a more detailed comment on Knopf et al., 2010 below, at your minor comment 5), i.e., no immersion freezing was examined, which impedes a direct comparison. Also, measurements in Corbin et al. (2012) were only done at -34°C, i.e., at a lower temperature than the range we examined in our study, while Knopf et al. (2010) only reports temperatures for ice nucleation onsets and no concentrations of INP, i.e., a comparison is not possible. This was added to the manuscript in the summarizing sentence given below (line 390 ff), while, for the here given multitude of reasons, nothing else was changed in our manuscript concerning this remark.

"*A comparison with Corbin et al. (2012) and Knopf et al. (2010), who both also examined INP in urban air in Toronto and Mexico City, respectively, is not possible due to different examined ice nucleation modes, and also because they only measured at -34°C (Corbin et al., 2012), i.e., outside of the temperature range examined in this study, or only reported ice onset temperatures (Knopf et al., 2010).*"

8. The authors claim that the measured INPs are non-urban and they suggest that the sources of the INPs could be dust or bioparticles which are non-urban. Do the authors think that is it not possible to have urban dust and urban bioparticles?

Indeed, there may be urban dust and also urban bioparticles. But these are no major contributor to the increase in PM2.5 mass concentrations during winter times – rather, it is well known that this increase is related to anthropogenic pollution. On the other hand, considering biogenic and dust particles, these particles emitted from urban areas will only contribute little to the overall atmospheric dust and biogenic particle load, as the non-urban sources are much more dominant for these types of particles. Therefore explicitly mentioning that dust and biogenic particles might also be emitted from urban sources does not really make sense. If there is, however, a specific passage in the text that you feel is miss-formulated, please tell us where this is exactly and why precisely you think this is wrong. For the time being, nothing was changed.

9. The conclusions are not well supported by the shown data. They are mainly qualitative, in comes cases speculative, given the lack of meteorological analysis, and the non-detection of secondary organic particles, dust, and bioparticles.

We do not agree with this rather simplified statement of the reviewer. We show that INP concentrations did neither correlate to concentrations of $PM_{2.5}$, BC, N_{total} or $N_{>500nm}$ (and related to the latter also not to INP concentrations derived from parameterizations based on $N_{>500nm}$ taken from literature). It is correct that we do not know the nature of the INP we detect, but given their small total number (and hence mass), a chemical analysis to detect what they are is currently, and will be for quite some time, rather impossible, not only for us but for the community in general. Hence an indirect method as we show in the current work already adds a lot of understanding to connections between INP and different aerosol sources. Pollution in Beijing, at least for our data-set, did not add INP to the atmospheric aerosol.

We hope that by adding the trajectories and statistics on the correlations to the manuscript will help to convince the reviewer to step back from the statement he made here. As all respective changes were already discussed above, no additional changes were made.

Minor Comments:

1. It is unclear how BC was calculated/determined for the $PM_{2.5}$ reported in Figure 1.

The BC was measured by a multi-angle absorption photometer (5012, MAAP, Thermo Fisher Scientific, Waltham, MA, USA) which got the sampled air through an inlet with a 2.5µm cut-off. This had been included in the previous version of the manuscript. The instrument measures the absorption of particles collected on a filter with a time resolution of 5 min and automatically derives BC mass concentration from the measurement while accounting for multiple scattering occurring on the filter. The MAAP is a well known and often used instrument for the measurement of absorption coefficients and BC mass concentrations.

The respective retrieval of BC values is now added in line 143 ff:"*The instrument measures the absorption of particles collected on a filter with a time resolution of 5 min and automatically derives BC mass concentration from the measurement while accounting for multiple scattering occurring on the filter.*"

2. Why the measured INP concentration time series is not included in Figure 2?

We showed what was previously Figure 2 in a similar manner as we show Figure 1, i.e., to describe the general situation concerning the atmospheric aerosol. In this part of the text, derived INP number concentrations have not been shown, yet, and showing them here would twist the line of thought followed in the text. To be able to include the measured INP concentrations in the lower panel of the figure you refer to, here, we made that lower panel an extra figure (see below) which now appears as Figure 7. Related necessary changes were made in the text.

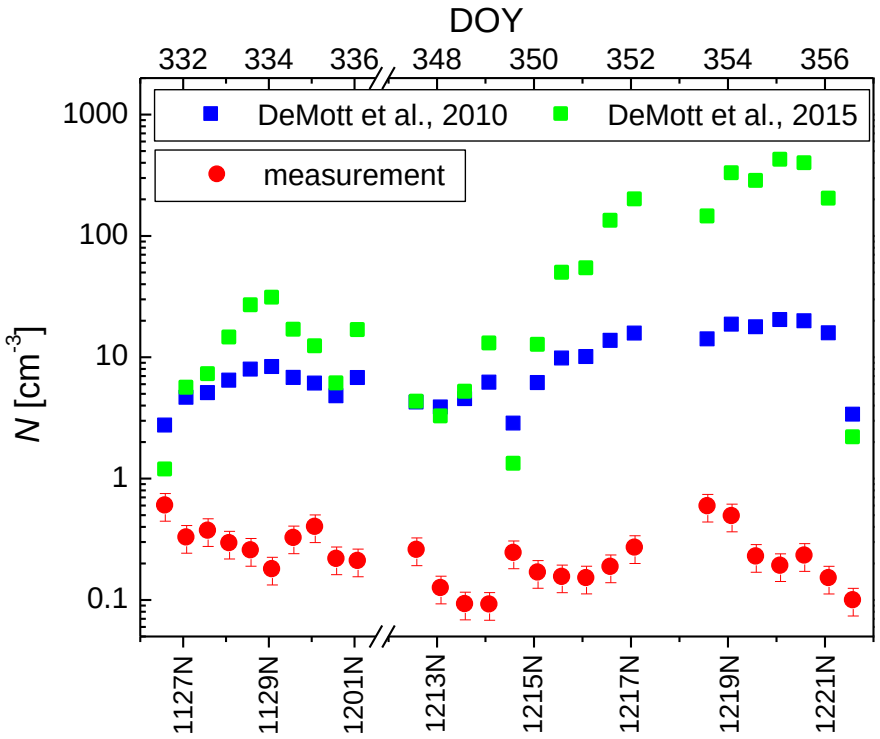

Figure 7. The time series of measured N_{INP} and N_{INP} parameterized according to DeMott et al. (2010, 2015) at -16°C.

3. There are several sentences and paragraphs that require a citation (e.g., Lines 34, 44, 250, 252, 264, 275, 280, and 352).

We were somewhat astounded by the list given to us here, as one of these lines concerns a description we make in the text: "we decided to use a subset of the therewith obtained data. For our analysis, ten LINA measurements from different days were selected," (line 251-253 in the previously submitted manuscript). Another one already was a citation: "…, whereas some described BC particles as possible INPs (Cozic et al., 2008; Cozic et al., 2007)." (line 264 in the previously submitted manuscript). Where possible, we added citations where they were asked for:

34: We added the review by DeMott et al. (2010) here to support this statement: “This results in a significant impact on the cloud extent, lifetime, formation of precipitation, and radiative properties of clouds *(DeMott et al., 2010)*.”

44: We added the review by Kanji et al. (2017) here to support this statement: “However, it has become obvious that many fundamental questions in this field are still unsolved *(Kanji et al., 2017).*”

250: See our answer to your minor comment 8.

275: We added: “*While mineral dust and biological particles are generally assumed to be the most abundant INP in the atmosphere (Murray et al., 2012, Kanji et al., 2017), the role of particles from combustion, i.e., of soot and ash particles, as INP is still controversial (Kanji et al., 2017).*”

280: We edited it and it now is: “*In the atmosphere of Beijing, the aging timescale is much shorter than in cleaner urban environments, which was shown in Peng et al. (2016).* For example, to achieve a complete morphology modification for BC particles in Beijing, the aging timescale was estimated to be 2.3 h *compared to 9 h in Houston* (Peng et al., 2016).”

352: The sentence referred to You et al. (2002) which was already given in Table 1 and is explicitly mentioned in the text now as well.

4. The introduction is quite disorganized. It jumps between bioparticles, dust, bioparticles, ash, soot, urban, soot and ash. I suggest to re-organize it and to focus on urban particles only. When introducing literature studies, make sure the ice nucleation modes are clearly stated.

Focussing only on urban particles makes no sense, as we show in this work that urban pollution particles do not contribute to INP. Indeed, as you remarked elsewhere, the urban environment can add mineral dust and bioparticles. But the fact that our derived INP concentrations are well within those reported in Petters & Wright (2015), where the latter were derived from non-urban environments, suggests that we are measuring a typical mid-latitude continental background, where non-urban sources might be the largest contributors. Hence it makes sense to at least mention all these possible types of INP particles.

We have reorganized the introduction, now shown at line 31 ff. Please see the revised version of the manuscript. New text is marked in yellow, but parts of the text that were simply shifted to improve the flow of the text are not marked particularly.

5. Lines 85-87: Knopf et al. (2010) also performed immersion freezing experiments relevant to mixed-phase clouds.

We are a bit confused by your statement, as Fig. 1 in Knopf et al. (2010), which presents all the data obtained for ice nucleation in that study, shows that data was not obtained at water saturation, besides for mainly the background measurements (called “H2O uptake substrate”) and a tiny fraction of one bar representing some data that barely touches the water saturation line. Indeed, there is data in there that is called “immersion mode”, however, as these data were generally obtained at relative humidities well below saturation (wrt. liquid water), this is not really immersion freezing. As one of us showed in a previous publication (Wex et al., 2014), for the type of freezing called “immersion freezing” in Knopf et al. (2010) an additional freezing point depression would have to be considered. Also, results on INP in Knopf et al. (2010) are only presented in terms of temperatures for the onset for ice nucleation, and the

text only mention cirrus clouds. The relevance of this paper for mixed-phase clouds eludes us. Nothing changed.

6. Was the PM2.5 time series obtained from the particles collected on the PTFE or quartz filters?

We point you towards the previously submitted manuscript where we already said: "Two PTEF filters were always sampled in parallel, and while one was used for INP analysis, the other was selected for the total mass and water-soluble ion analysis." (Previously line 139-140, now line 149-150.)

7. Were the freezing experiments performed with the LINA recorded with pictures taken every 6 seconds? My experience is that droplets freeze very quick and if pictures are taken every 6 seconds very important information can be missed.

Yes, pictures were taken every 6 seconds. And yes, droplets freeze very quickly. But we do not have to detect the exact second at which a droplet freezes, and once it is frozen, it stays frozen during the experiment. For a cooling rate of 1 K/min, data-points can "only" be given for a temperature resolution of 0.1 K, which, however, is still a high temperature resolution. Each time a picture is taken, the cumulated number concentration of INP that are ice active at the respective temperature (that is currently effectively valid) or above can be derived (by counting all frozen droplets).

This is a procedure that has been used in the past and currently has seen a revival, as described e.g. in the literature by Budke & Koop (2015) and Conen et al. (2012), on which our set-ups and methods are based, as we describe in the manuscript.

8. It was said that particle removal by washing the filters was frequently incomplete. Can the authors indicate by how much? What percentage of the particles was not possible to be removed from the filters? This calculation is very important for the direct comparison of the LINA and INDA data.

As we said in the previously submitted version of our manuscript, of the 30 examined filter sets, for ten of the filters for which washing off was done the deviation factor between INP concentrations from INDA and LINA were between 1.3 and 4.4 (INDA was always higher), and only results from the analysis with LINA for these ten filters are shown in the manuscript. This had been said in lines 252 to 254 in the previous version.

We edited the respective paragraph and added some additional information, so this whole passage now is as follows (with new text in italic) (see line 270 ff):
"Washing particles off from the PTFE filters was more complete for some filters than for others. This showed in differently large deviations in N_{INP} from INDA and LINA measurements in the overlapping temperature range, where results determined from INDA were always similar to or higher than those from LINA, as particle removal by washing the filters was frequently incomplete. *It is mentioned in Conen et al. (2012), that a quantitative extraction of particles from quartz fiber filters was not possible without also extracting large amounts of quartz fibers. We tried to overcome this issue by using PTFE filters, as degradation of the PTFE filter during washing does not occur due to the hydrophobic properties of the filter material. But we observed that not all particles were released into the water during the washing procedure (likely those collected deep within the filter), as filters frequently still looked greyish after washing, independent from the washing procedure (we experimented with different washing times up to 4 hours and with the use of an ultrasonic bath). For our INDA measurements, punches of quartz filters were*

measured after they were immersed in water, representing the ice nucleating properties of all collected particles (Conen et al., 2012). However, as already mentioned above, N_{INP} derived from LINA measurements were lower than those from INDA, due to particles that did not come off during washing. Based on our observations, we cannot recommend the use of sampling on PTFE filters followed by particle extraction in water. But we still decided to select those data from LINA measurements that showed the lowest deviation to the respective INDA results in the overlapping temperature range for use in this study. After calculating the deviation between INDA and LINA results, represented as the factor (N_{INP} of INDA / N_{INP} of LINA), ten LINA measurements from different days were selected to be used. For these measurements, the factor representing the deviation was in a range from 1.3 to 4.4."

9. Be consistent with the references format (lines 495, 505, 517, 553, 574, 604, 626, 629, 635, 642, 666, and 669).

Thank you for your careful reading, we corrected these citations.

10. Table 1. (Now Table 2!) How is it possible to perform deposition ice nucleation at water saturation? Given that S_i is higher than S_w, how is it possible to obtain conditions with S_i= S_w?

It is difficult to convey all the information from the literature cited here in a simple table. This caused the information on S_i and S_w to be somewhat messy. We decided to delete this information, as it does not add any additional value on top of the freezing mode that was examined in the different cited papers and that is given in the table, anyway.

11. Figure 2. (Now Figure 4) Are the INP concentrations in std L-1? Add here the measured INPs with their corresponding uncertainty.

Yes, the volume we used was already given as standard volume. This is explicitly mentioned in the text now.

As mentioned above at our answer to your minor comment 2, the data was added, but for the flow of the text, the lower panel to which the data was added is now an extra figure which now appears as Figure 7. (See our answer to your minor comment 2, above.)

As mentioned above at our answer to your major comment 6, a new figure and related text was added in the appendix.

12. Figure 3. (Now Figure 5) Add all four panels to one single figure. I mean, one figure with 4 panels in one page.

This is a topic for setting the final version of the manuscript. Nothing changed.

13. Figure 4. (Now Figure 6) Axis and symbols are too small. Add r2 and p-values.

The figure was edited, script has been enlarged, and a table with R2 and p values has been added. See our answer to your major comment 6.

14. Figure 5. (Now Figure 8) I don't see the purpose of this figure given that Petters and Wrifgt (2015) study focused on precipitation samples.

As we already argued in our comment to your major comment 7, in the study of Petters and Wright (2015), INP concentrations obtained per volume of precipitation are converted to INP concentrations per volume of air, i.e., to the same parameter we derive from our samples. Petters and Wright (2015) explain the uncertainties in their assumptions due to this conversion at great length, and we feel it is justified to use these data for the kind of comparison we are doing here. Also, Petters and Wright (2015) offer one of the largest compilations on atmospheric INP concentrations that we are aware off. We feel that it is an important information that INP number concentrations in such a strongly polluted location as Beijing city in November and December did not exceed the respective concentrations measured in more rural environments.

Literature:

Conen, F., Henne, S., Morris, C. E., and Alewell, C.: Atmospheric ice nucleators active>=12°C can be quantified on PM10 filters, Atmos. Meas. Tech., 5, 321-327, doi:10.5194/amt-5-321-2012, 2012.

Corbin, J. C., Rehbein, P. J. G., Evans, G. J., and Abbatt, J. P. D.: Combustion particles as ice nuclei in an urban environment: Evidence from single-particle mass spectrometry, Atmos. Environ., 51, 286-292, doi:10.1016/j.atmosenv.2012.01.007, 2012.

Cozic, J., Verheggen, B., Mertes, S., Connolly, P., Bower, K., Petzold, A., Baltensperger, U., and Weingartner, E.: Scavenging of black carbon in mixed phase clouds at the high alpine site Jungfraujoch, Atmos. Chem. Phys., 7, 1797-1807, doi:10.5194/acp-7-1797-2007, 2007.

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Answers to the report by anonymous Referee #2

We thank Referee 2 for reviewing our manuscript and also for useful hints and suggestions. Below, comments from the referee are given in blue while our answers are given in black, with passages including new text given in italic. Additionally, the new text is marked yellow in the revised version of the manuscript.

This paper presents measurements of ice nucleating particles in Beijing, China. The ice nucleation activity of particles sampled on filters was quantified using ice-nucleating droplet arrays, LINA and INDA. This information was supplemented by ion chromatography measurements of the filters and in-situ measurements of black carbon and particle size distributions. The authors find no correlation between filter-based INP concentrations and PM2.5 or black carbon measurements. As the authors correctly state, there are few measurements of ice nucleating particles in urban areas, particularly in China. This paper is therefore of interest to the community. I recommend the publication in ACP after the following concerns are addressed:

1. It would be useful to see a time series of INP concentrations alongside Figure 1.

We included such a time series in the lower panel of what was Figure 2 (now Fig. 7), and, for the flow of the text, made that lower panel an extra figure which now appears as Figure 7. Related necessary changes were made in the text.

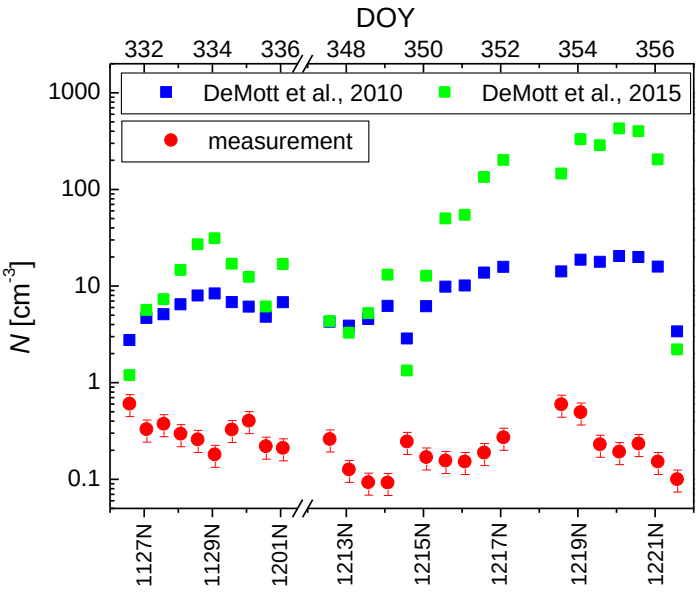


Figure 7. The time series of measured N_{INP} and N_{INP} parameterized according to DeMott et al. (2010, 2015) at -16ºC.

2. On p. 10, lines 250-252, the authors mention difficulty in washing particles off of PTFE filters and state that this procedure cannot be recommended in general. Why

was the non-recommended technique used here? Can the authors provide an estimate of the uncertainty that this contributes to LINA measurements? Further, in line 252, the authors mention that they heave used a "subset" of PTFE filter LINA measurements. Which measurements were left off and why?

While doing the sampling, we had planned to focus on the quartz fiber filters, knowing that this will yield results for a restricted temperature range, only. But as we had a four-channel sampler, we thought we could try to sample on additional filter material. Polycarbonate filters did not work out, as they cause such a large pressure drop and the pump that was available was not sufficiently strong to collect a reasonable air volume. Hence the PTFE filters were used. We were not sure if these filters would work, but tried, nevertheless. It turned out that indeed, due to the fibers, particles stick in them even after washing (filters sometimes were still colored after the washing), but we decided to use at least results from those filters for which, in the temperature range where data was obtained from both filters, results were not more than a factor of 4.4 lower for LINA than for INDA data, which left data from ten filters analyzed with LINA.

We edited the text concerning this and added some additional information, so this whole passage now is as follows (with new text in italic) (see line 270):

"Washing particles off from the PTFE filters was more complete for some filters than for others. This showed in differently large deviations in N_{INP} from INDA and LINA measurements in the overlapping temperature range, where results determined from INDA were always similar to or higher than those from LINA, as particle removal by washing the filters was frequently incomplete. *It is mentioned in Conen et al. (2012), that a quantitative extraction of particles from quartz fiber filters was not possible without also extracting large amounts of quartz fibers. We tried to overcome this issue by using PTFE filters, as degradation of the PTFE filter during washing does not occur due to the hydrophobic properties of the filter material. But we observed that not all particles were released into the water during the washing procedure (likely those collected deep within the filter), as filters frequently still looked greyish after washing, independent from the washing procedure (we experimented with different washing times up to 4 hours and with the use of an ultrasonic bath). For our INDA measurements, punches of quartz filters were measured after they were immersed in water, representing the ice nucleating properties of all collected particles (Conen et al., 2012). However, as already mentioned above,* N_{INP} *derived from LINA measurements were lower than those from INDA, due to particles that did not come off during washing. Based on our observations, we cannot recommend the use of sampling on PTFE filters followed by particle extraction in water. But we still decided to select those data from LINA measurements that showed the lowest deviation to the respective INDA results in the overlapping temperature range for use in this study. After calculating the deviation between INDA and LINA results, represented as the factor (*N_{INP} *of INDA /* N_{INP} *of LINA), ten LINA measurements from different days were selected to be used. For these measurements, the factor representing the deviation was in a range from 1.3 to 4.4."*

3. In Figure 4c, the x-axis label is "Ntotal at -16 degC". Should this be just the Ntotal average values from Figure 1? Why is the "at -16 degC" there?

You are right, thanks for the hint. Corrected!

4. Several (a and c, for example) of the plots in Figure 4 arguably show some weak

correlations. Some statistical tests of significance would help to strengthen the authors' case.

We added the following table giving R^2 and p values for the 6 scatter-plots presented in in what was Fig. 4 (now Fig. 6) and added the respective lines to the figure (see below), together with some text. It can clearly be seen by the values given in the newly added table, that there is no correlation.

The following text was added:
"*Linear fits are included in all panels of Fig. 6, and values for R^2 and p for these fits are shown in Table 1.*" (line 305)

"*Also the R^2 and p values given in Table 1 clearly show that there was no correlation between N_INP and any of the examined parameters.*" (line 309)

Table 1 Coefficient of determination (R^2) and a measure for the statistical significance of the assumption of a linear correlation (p) for the comparison of N_{INP} at -16°C with the different parameters shown in Fig. 6.

parameter	R^2	p
(a) BC concentration	0.003	0.79
(b) $PM_{2.5}$ concentration	0.006	0.71
(c) N_{total}	0.005	0.73
(d) $N_{>500nm}$ at -16°C	0.008	0.67
(e) N_{INP} at -16°C, based on DeMott et al. (2010)	0.005	0.73
(f) N_{INP} at -16°C, based on DeMott et al. (2015)	0.007	0.67

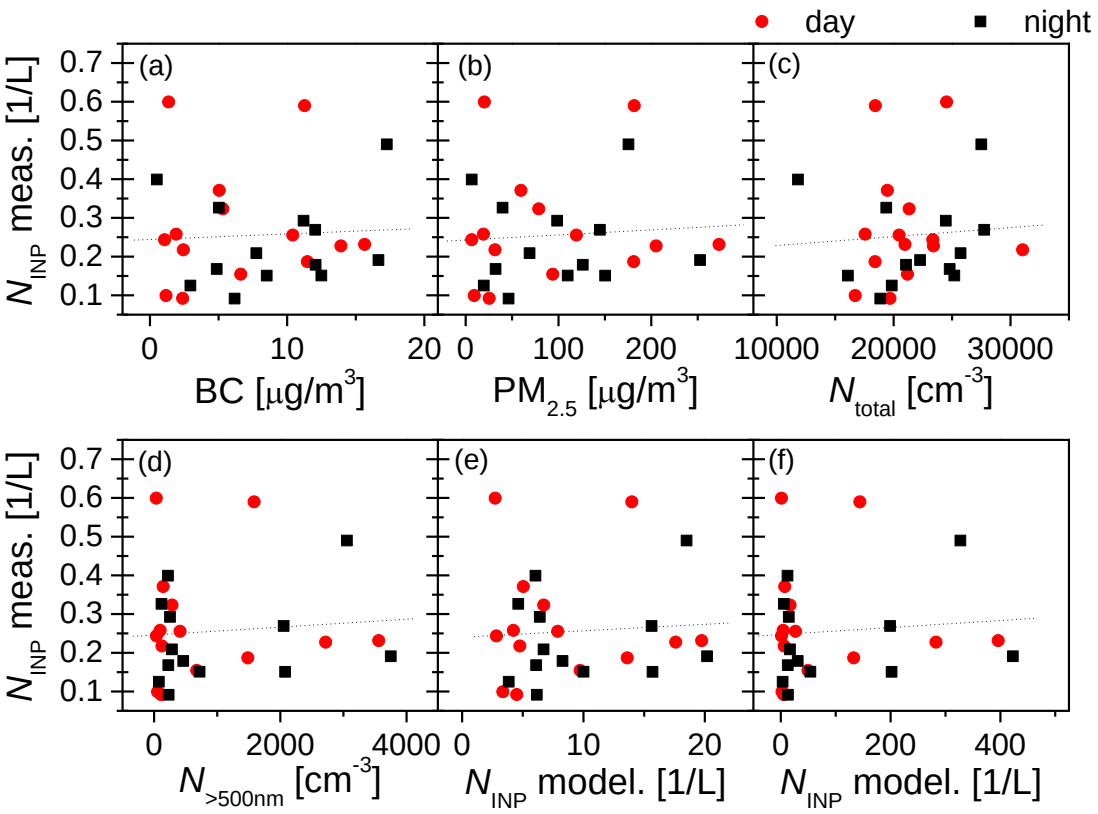


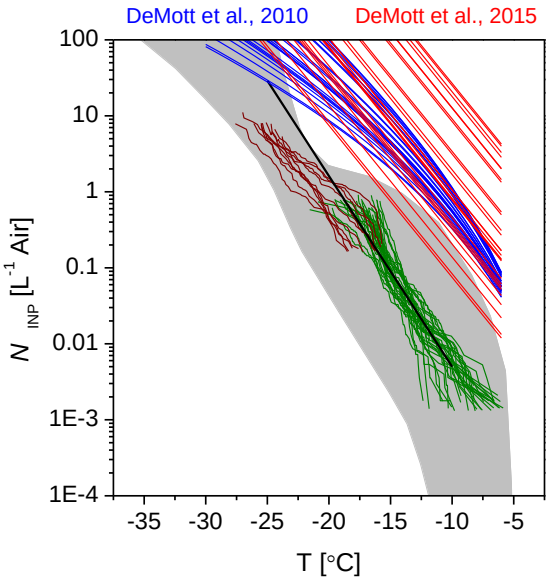
Figure 6. N_{INP} at -16°C as function of mass concentrations of BC (a) and $PM_{2.5}$ (b), and of 12h-average values of N_{total} (c). Furthermore, we show $N_{>500nm}$ (d), and N_{INP} at -16°C derived based on (DeMott et al., 2010) (e) and DeMott et al. (2015) (f) for daytime (red round symbols) and nighttime (green square symbols) samples.

5. Figure 4 (Now Figure 6) only shows INP concentrations at -16 degC, were there any differences for other temperatures?

No, there were no correlations at other temperatures, either (which can already be seen by the freezing curves not crossing much). This had been said before in the manuscript (previously line 268, now line 306), therefore nothing was changed: "Our results discussed in the following, based on N_{INP} at -16°C, are similarly valid for all other temperatures down to -25°C."

6. It would be useful to plot DeMott, et al. (2010 and 2015) parametrizations alongside the Fletcher (1962) parametrization in Figure 5.

The DeMott parameterizations were added to the respective figure as shown below. However, we do prefer to not add this to the main manuscript as we do not want to stress it too much that these parameterizations do not fit, as this is caused by the pollution aerosol, which also included pollution particles larger than 500 nm, and not by the parameterizations. It does not say anything about the applicability of the parameterizations. Should you be of a different opinion, please let us know.



7. The authors conclude that the INPs detected here are "background" INPs, likely originating from dust, based on some previous measurements in China, which show enhancements in ice nucleation during dust events. Since the calcium ion is used here as a tracer for dust, do the INP concentrations correlate with it? Do they correlate with any of the chemical constituents measured with ion chromatography?

We did not find a correlation with any single component we looked at, please see the comparison with Ca^{2+} and K^{+} as examples below (this is not included in the new version of the manuscript, we only show it to you here). It has been assumed that the feldspar fraction (and there maybe only the K-feldspar fraction) of dust may be responsible for the ice nucleation activity of atmospheric mineral dust, therefore it is not too astounding that no correlation with Ca^{2+} was found (see both figures below). As for K^{+}, this is also emitted during biomass burning and hence is influenced by anthropogenic pollution (as can be seen in the second figure below, it follows the trends of the pollution), and hence, it also will not serve as a tracer for mineral dust. It should also be mentioned, that we only examined the soluble fractions of these components, while in dust particles, they might occur in non-soluble compounds.

Furthermore, INP make up a very small amount of both, total mass and total number concentration (a general value that is often given is that one in a million particles is ice active at -20°C), so it might be impossible to correlate atmospheric INP with a chemical compound based on the chemical analysis we did here.

And last but not least, the broad temperature range over which INP concentrations in the atmosphere are observed to increase generally suggest that a number of different types of INP participate (it is often assumed that more rarely occurring biological particles cause ice nucleation at higher temperatures (>-15°C) while mineral dust particles are responsible for the observed ice nucleation at lower temperatures), so we might have to look at a combination of different trace components.

But all this was not the aim of this study. We wanted to see if anthropogenic pollution might add INP, which we showed is not the case. Still, we added a paragraph summarizing what we answered to you above, which can be found at the end of section 3.3, from line 371 onward:

"Additionally, also no correlation was found between any of the water-soluble constituents that were analyzed with ion chromatography and INP concentrations. This is not too astounding, as INP make up only a small fraction of all particles, as can be seen when comparing number concentrations from Fig. 4 and Fig. 7, and hence they make up only a small fraction of the mass, likely too small to be detected. Furthermore, a number of different components might contribute to INP, e.g., biological INP that are generally ice active at higher temperatures (> -15°C, Murray et al., 2012) and mineral dusts which are ice active at lower temperatures, therefore one common tracer for INP might not be applicable. As far as K is concerned, which might be connected to K-feldspar containing mineral dust particles with high ice activity (Atkinson et al., 2013), we only analyzed the water soluble fraction, i.e., K related to feldspar would not have been analyzed. Moreover, K is also emitted by biomass burning and hence influenced by anthropogenic pollution. It remains to be seen if a simple correlation between chemical constituents of the atmospheric aerosol and INP concentrations can be established at all."

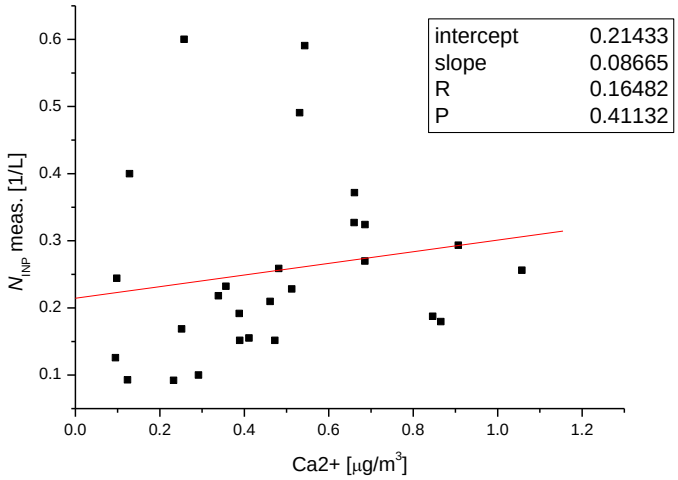


Figure above: Measured INP number concentrations at -16°C plotted versus Ca^{2+} mass concentrations derived from ion chromatography analyzing water soluble ions.

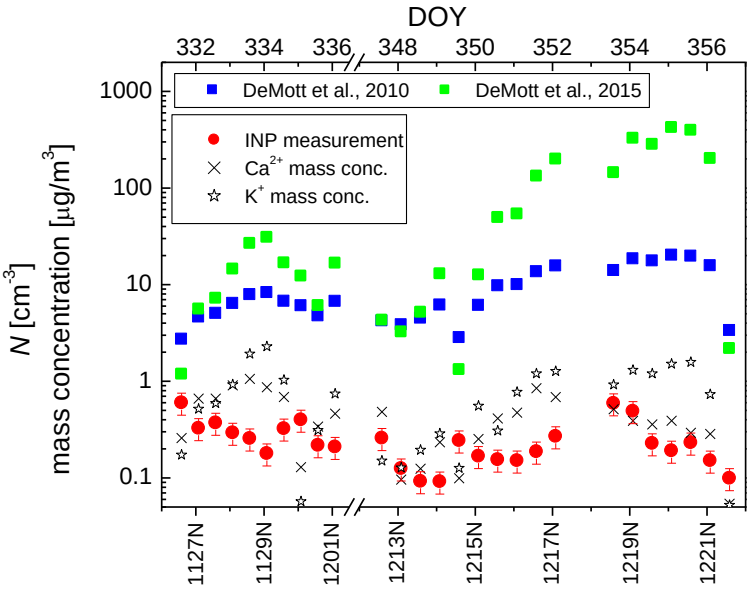


Figure above: The time series of measured N_{INP} and N_{INP} parameterized according to DeMott et al. (2010, 2015) at -16ºC, together with Ca^{2+} and K^{+} mass concentrations derived from ion chromatography analyzing water soluble ions.

Literature:

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Ice nucleating particle concentrations unaffected by urban air pollution in Beijing, China

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Key Points:

- Ice nucleation
- Urban aerosol
- Immersion mode

Abstract

Exceedingly high levels of $PM_{2.5}$ with complex chemical composition occur frequently in China. It has been speculated if anthropogenic $PM_{2.5}$ may significantly contribute ice nucleating particles (INP). However, few studies have focused on the ice-nucleating properties of urban particles. In this work, two ice-nucleating droplet arrays have been used to determine the atmospheric number concentration of INP (N_{INP}) in the range from -6 ºC to -25 ºC in Beijing. No correlations between N_{INP} and neither $PM_{2.5}$ nor black carbon mass concentrations were found, although both varied by more than a factor of 30 during the sampling period. Similarly, there were no correlations between N_{INP} and either total particle number concentration or number concentrations for particles with diameters > 500 nm. Furthermore, there was no clear difference between day and night samples. All these indicate that Beijing air pollution did not increase or decrease INP concentrations in the examined temperature range above values observed in non-urban areas, hence, the background INP concentrations might not be anthropogenically influenced as far as urban air pollution is concerned, at least in the examined temperature range.

1 Introduction

Formation of the ice phase in clouds can be modulated by aerosols emitted from anthropogenic and natural sources (Morris et al., 2014; Murray et al., 2012; Rosenfeld et al., 2008) via heterogeneous ice nucleation (Pruppacher et al., 1998). This results in a significant impact on the cloud extent, lifetime, formation of precipitation, and radiative properties of clouds (DeMott et al., 2010). Currently, four mechanisms are proposed for the heterogeneous ice nucleation in mixed-phase clouds: deposition ice nucleation, condensation freezing, immersion freezing, and contact freezing (Vali et al., 2015; Hoose and Möhler, 2012). It is under discussion if condensation freezing is different from immersion freezing on a fundamental level (Wex et al., 2014) and if at least some of the observed deposition ice nucleation can be attributed to pore condensation and freezing (Marcolli, 2014). For mixed-phase clouds, immersion freezing has been widely reported to be the most important ice nucleation mechanism (Ansmann et al., 2008; Murray et al., 2012; Westbrook and Illingworth, 2013). During the past decades, great efforts have been dedicated to understanding heterogeneous ice nucleation. However, it has become obvious that many fundamental questions in this field are still unsolved (Kanji et al., 2017).

Numerous studies have attempted to quantify the ice nucleation ability of selected aerosol particles of a specific composition in immersion mode, such as dust (DeMott et al., 2015; Kaufmann et al., 2016; DeMott et al., 2003), marine (Wilson et al., 2015; DeMott et al., 2016; Alpert et al., 2011) and biological particles (Pummer et al., 2012; Hartmann et al., 2013; Fröhlich-Nowoisky et al., 2015). Szyrmer and Zawadzki (1997), Hoose and Möhler (2012), Murray et al. (2012) and Kanji et al. (2017) are all reviews which give a more extensive overview over materials that can induce ice nucleation. In general, biogenic particles have been assumed to provide atmospheric ice nucleating particles (INP) which are ice active in the immersion mode at comparably high temperatures (above -15°C, Murray et al., 2012; Petters & Wright, 2015). Ice activity at lower temperatures is attributed to mineral dust particles (Murray et al., 2012) while the role of soot particles in atmospheric ice nucleation is still debated (Kanji et al., 2017).

Biogenic particles in general have long been known to be able to induce ice nucleation at comparably high temperatures above -10°C (e.g. Schnell and Vali, 1972). It has been widely accepted that biological particles can act as efficient INP, with some bacteria and fungi reported to possess the ability to arouse freezing at temperatures as high as -2°C to -5°C (Lundheim, 2002). Fungal spores (O'Sullivan et al., 2016; Pummer et al., 2015) and lichen (Moffett et al., 2015) are known to nucleate ice in the temperature range above -10°C, while pollen (Augustin et al., 2013; Pummer et al., 2012) may compete with mineral dust particles in terms of their ability to nucleate ice, albeit not in terms of their atmospheric abundance.

Recognized as the dominant INPs in mixed-phase clouds (Kamphus et al., 2010), particles from various mineral dusts were found to catalyse ice formation effectively in chamber experiments (Murray et al., 2012; Kanji et al., 2017). Among mineral dust particles, those containing K-feldspar might be particularly ice active (Atkinson et al., 2013).

In general, burning of liquid fuels produces soot particles (i.e., particles that are mostly organic), while burning of solid material as e.g., biomass or coal will also produce ash particles which contain the inorganic components that made up the fuel. Umo et al. (2015) and Grawe et al. (2016) examined the ice activity of ash particles from wood and coal burning in the immersion mode and both find that these particles are ice active. In Grawe et al. (2016), ash particles with atmospherically relevant sizes of 300 nm were examined and the most active particles came from a sample of fly ash from a coal burning power plant, inducing immersion freezing below -22°C. Both, Umo et al. (2015) and Grawe et al. (2016) suggest that ash particles might play a role in the atmosphere, however, point to a lack of knowledge of their

atmospheric abundance. Also, different ash samples showed different ice activities, and also large differences in the results between the methods used for the examination were described, i.e., it is still inconclusive if ash particle might play an important role as atmospheric INP.

Although there has been a considerable number of studies aimed at understanding the ability of black carbon (BC)-containing particles acting as INP, the results are still controversial. Some studies show that BC-containing particles did not act as good INP (Schill et al., 2016; Chou et al., 2013). Chou et al. (2013) observed that soot particles from diesel engines and wood burning form ice at -40°C, and unrealistically high relative humidity (RH) was needed for freezing initiation above this temperature. Schill et al. (2016) showed that neither fresh nor aged emissions from diesel engines contributed appreciably to atmospheric INP concentrations. However, some studies considered BC-containing particles as possible INPs (Cozic et al., 2008; Levin et al., 2016; Cozic et al., 2007). Observation of abundant BC in ice particle residuals in field experiments suggested that some BC-containing particles may preferentially act as INP (Cozic et al., 2008). In the experiments conducted by Levin et al. (2016), emissions of different types of biomass fuel produced measurable concentrations of INPs (0.1-10 cm^{-3}) associated with higher BC concentration accounting for about 0-70%. Determination of ice nucleating properties of physically and chemically aged soot particles suggests that the heterogeneous ice nucleation activity of freshly emitted diesel soot particles is sensitive to some of the aging processes (Kulkarni et al., 2016).

In the atmosphere of urban areas with dense population, various sources and complex aging transformations (such as coagulation, condensation of vapor, chemical reaction) of particles can be found. Particularly, urban aerosol may be rich in BC-containing particles resulting from anthropogenic activities, such as fossil fuel combustion and biomass burning (Bond et al., 2013), which were speculated to play a role for the formation of ice in clouds (Kanji et al., 2017). However, the ice nucleating properties of particles produced in urban regions have rarely been the focus of previous studies. Exceptions are Knopf et al. (2010) and Corbin et al. (2012), examining the ice nucleation activity of particles in the anthropogenically influenced atmospheric aerosol in Mexico City and Toronto, respectively. In both studies the relative humidity at which measurements were made were below water vapor saturation (with respect to liquid water). Using filter samples, Knopf et al. (2010) state that organic particles included in their samples might potentially induce ice nucleation at conditions relevant to cirrus formation. Corbin et al. (2012) used a CFDC (Continuous-Flow Diffusion Chamber) operating at -33°C together with a

particle mass spectrometer. Statistical limitations impeded a statistical sound analysis, but their data suggests that dust particles, particles from biomass burning and particles containing elemental carbon might be sources of INP at their experimental conditions. They explicitly encourage further studies of these particles types concerning their role as possible INP.

In the present study, we measured the ice nucleating activity of urban aerosols in parallel with BC and $PM_{2.5}$ mass concentration and particle number concentrations in the atmosphere of the mega-city Beijing, which is frequently experiencing heavy pollution. During heavy haze episodes, $PM_{2.5}$ mass concentration can be several hundred micrograms per cubic meter and composed of a complex mixture of different chemical components (organic matter, inorganic ions and black carbon) (Zheng et al., 2016). The goal of this project is to find out if anthropogenic sources which are dominant in the urban atmosphere significantly contribute to the local INP concentrations, focusing particularly on the ice nucleating ability of BC in urban aerosols.

2 Materials and Methods

2.1 Sample collection and particle number measurement

The sampling site for particle collection was on the roof of a six-floor building (about 30 m above ground level) on the campus of Peking University (39°59′20″N, 116°18′26″E), located in the north-western urban area of Beijing.

Particles with an aerodynamic diameter less than or equal to 2.5 micro-meters ($PM_{2.5}$) were collected on quartz fiber (Whatman, 1851-865) and PTFE filters (Whatman,7592-104) using a 4-channel sampler with 2.5µm impactors from 27th November 2016 to 1st December 2016 and 13th December 2016 to 22th December 2016. Daytime filters were collected from 8:00 am to 8:00 pm and nighttime filters were collected from 8:00 pm to 8:00 am with an air flow rate of 16.7 L min^{-1}, resulting in a total volume of air sampled on each filter of ~12000 L. Note that all sample volumes used herein were converted to standard volumes. The quartz filters were treated before the sampling by heating them to 550 ºC for 6 h. After sampling, all filters were kept at ≤ -18 ºC during storage, and the INP analysis was done within 20 days, starting on 5th February in 2017.

A scanning mobility particle sizer (SMPS, TSI Inc., USA) system was used to obtain particle number distribution in the 3-700 nm (electrical mobility diameter) size range during the sampling period

while an aerodynamic particle sizer (APS, TSI model 3321, TSI Inc., USA) measured particle number size distributions between 800 nm and 2.5µm (aerodynamic diameter). The APS results were transformed from aerodynamic diameter to Stokes diameter with a particle density of 1.5 g cm^{-3} which were measured by a CPMA (centrifugal particle mass analyzer) and combined with the measured and inverted size distributions obtained from the SMPS. From these combined size distributions, we calculated the total particle number concentration of particles in the diameter range from 3nm to 2.5µm (N_{total}) and number concentrations of particles larger than 500nm ($N_{>500nm}$). When comparing with filter results, we use 12h-average values of N_{total} and $N_{>500nm}$., where the averages were always made from 8:00 am to 8:00 pm for daytime data and from 8:00 pm to 8:00 am for nighttime data. $N_{>500nm}$ was derived, as in general larger particles are expected to be more efficient INP, and also as this size range was selected in DeMott et al. (2010, 2015) to serve as a base for parameterizations of INP number concentrations.

Concentrations of BC were continuously measured by a multi-angle absorption photometer (5012 MAAP, Thermo Fisher Scientific, Waltham, MA, USA) utilizing a 637 nm LED as a light source (Müller et al., 2011). The instrument measures the absorption of particles collected on a filter with a time resolution of 5 min and automatically derives BC mass concentration from the measurement while accounting for multiple scattering occurring on the filter.

2.2 Chemical analysis

Two PTEF filters were always sampled in parallel, and while one was used for INP analysis, the other was selected for the total mass and water-soluble ion analysis. $PM_{2.5}$ mass concentration was obtained with an analytical balance by the gravimetric method (Mettler Toledo AG285) (Yang et al., 2011). As for water-soluble inorganic compounds analysis, Guo et al. (2012) described the method for seven major ions (K^+, Mg^{2+}, Ca^{2+}, NH_4^+, NO_3^-, SO_4^{2-} and Cl^-) measured by ion-chromatograph (DIONEX, ICS-2500/2000) based on the usage of PTEF filters. Post-sampling, all filters were stored in the refrigerator at -18 ºC before analysis.

2.3 INDA and LINA analysis

Two devices called INDA (Ice Nucleation Droplet Array) and LINA (Leipzig Ice Nucleation Array) have been set up at the Leibniz Institute for Tropospheric Research (TROPOS) in Germany following

159 the design described in Conen et al. (2012) and in Budke & Koop (2015), respectively. INDA was used
160 to investigate the immersion freezing properties of the quartz fibre filter samples while LINA was used
161 to test the particles on PTFE filters.

162 INDA consists of a thermostat (JULABO FP40) with a 16 L cooling bath. 96 circles (1mm in
163 diameter each) of each quartz filter were cut out by a punch and immersed separately in the tubes of a
164 PCR (Polymerase chain reaction) tray which each contained 50 µl distilled water. While Conen et al.
165 (2012) originally used separate Eppendorf Tubes®, the use of PCR trays for immersion freezing studies
166 has been suggested before in Hill et al. (2016) and was adapted in the LINA set-up. The PCR trays were
167 placed on a sample-holder and exposed to decreasing temperatures with a cooling rate of approximately
168 1 K min^{-1} in the cooling bath down to -30 ºC. Real time images of the PCR trays were recorded every 6
169 seconds by a CCD (Charge Coupled Device) camera. A flat light that was fixed at the bottom of the
170 cooling bath helped to yield proper contrast between frozen and unfrozen droplets on the recorded
171 pictures, so that frozen droplets could be identified according to the brightness change during the freezing
172 process. A program recorded the current temperature of the cooling bath and related it to the real-time
173 images from the CCD camera. The temperature in the PCR trays had been calibrated previously as
174 described in section 1.1 of the appendix.

175 For the measurement of ice nucleating particles at lower temperature, LINA was built according
176 to an optical freezing array named BINARY, which was described in detail by Budke & Koop (2015).
177 PTFE filters collected during the same period as quartz fibre filters were used for LINA. Half of the
178 PTFE filter of each day was immersed in 10 ml distilled water and shaken for 1 h to wash particles off.
179 For each measurement, 90 droplets with the volume of 1 µl were pipetted from the resulting suspension
180 onto a thin hydrophobic glass slide (diameter 40 mm, thickness 0.13-0.16 mm, obtained from Marienfeld-
181 superior), with each droplet being contained in a separate compartment. These compartments were round
182 holes with diameters of 3 mm, drilled into an aluminium plate with a diameter of 40 mm and a thickness
183 of 14 mm. Both, hydrophobic glass slide and the aluminium plate with the compartments were
184 surrounded by an aluminium ring with an inner diameter of 40 mm, which acted to keep glass slide and
185 aluminium plate in place. Slide, plate and ring were all arranged before the droplets were pipetted. A
186 second thin glass slide was put on top of the plate so that the compartments were all separated from each
187 other and that evaporation of the droplets was prevented. The droplets were cooled on a Peltier element
188 with a cooling rate of 1 K min^{-1}. There was a thin oil (squalene) film between the hydrophobic glass slide

and the Peltier element for optimal heat conductivity. The temperature on the glass slide had been determined previous to the experiments as described in section 1.2 of the appendix, and the temperature shift between that set on the Peltier element and that observed on the glass slide was accounted for in the data presented herein. The freezing process again was recorded by taking pictures with a CCD camera every 6 seconds and detecting the freezing based on a change in the reflectance of the droplets upon freezing.

As mentioned above, the temperature calibration for these two instruments is described in detail in the section 1.1 and 1.2 of the appendix. The background freezing signal of pure distilled water and circles cut from clean filters were tested as well. These results are shown in the section 2 of the appendix.

The measurements resulted in frozen fractions (f_{ice}) as defined in Eq. (1):

$$f_{ice} = \frac{N_{frozen}}{N_t} \qquad (1)$$

where N_{frozen} is the number of frozen tubes or droplets at a certain temperature and N_t is the total number of tubes in PCR trays (i.e., 96) or droplets on the slides (i.e., 90).

The temperature dependent cumulative number concentration of INP (N_{INP}) per volume of sampled air was calculated according to Eq. (2), similarly to Vali (1971) and Conen et al. (2012):

$$N_{INP}(T) = -\frac{\ln(1-f_{ice}(T))}{V_{sampled}} \qquad (2)$$

where $N_{unfrozen}$ is the number of tubes or droplets still unfrozen (liquid) at a certain temperature, and $V_{sampled}$ is the volume of air converted to standard conditions (0°C and 1013hPa) from which the particles were collected that were suspended in each of the droplets in LINA or that were collected on each filter punch used for INDA measurements, respectively.

The chemical ion analysis in section 3.1 and the determination of the $PM_{2.5}$ mass concentration was done at Peking University. The filters used for INP measurements were brought to TROPOS where then INP measurements were done. Filters were continuously cooled below 0°C in a portable ice box during transport.

3 Results and Discussion

3.1 Severe $PM_{2.5}$ pollution in Beijing

Fig. 1 shows the time series of $PM_{2.5}$ mass concentrations and chemical composition during the sampling period. The $PM_{2.5}$ mass concentration with a mean value of 97.30±77.9 μg m^{-3} ranged from

6.54 μg m^{-3} up to 273.06 μg m^{-3}. Here, the cases with $PM_{2.5}$ above 50 μg m^{-3} were defined as polluted days, whereas the rest was defined as clean days. On average, the sulfate, nitrate, and ammonia (SNA) accounted for around 35% of $PM_{2.5}$ during the whole period with an obvious enhancement in polluted days (53%), indicating that generation of secondary particulate mass is one major contributor to the formation of particulate pollution, as it has previously been described in Guo et al. (2010) and Zheng et al. (2016). In this study, when we refer to secondarily formed particulate matter, this will always stand mainly for SNA and secondary organic substances. Dust particles are in the coarse mode, and only contribute little to the total $PM_{2.5}$ load (Lu et al., 2015; Li and Shao, 2009). In these studies, Ca^{2+} as a tracer for dust particles showed a low proportion in $PM_{2.5}$, suggesting that the dust particles also only contributed little to $PM_{2.5}$ during our observations as well.

During the sampling period, BC mass concentrations varied from 0.50 μg m^{-3} on clean days up to 17.26 μg m^{-3} on polluted days. On average, the mean mass concentration of BC, 7.77±5.23 μg m^{-3}, accounted for about 13% of $PM_{2.5}$. During night time, BC concentrations were higher than those during daytime due to stronger diesel engine emissions and a lower boundary layer (Guo et al., 2012). Our previous studies showed that secondarily and primarily formed organic particulate matter contributed to around 36% of non-refractory PM_1 detected by an aerosols mass spectrometer during wintertime in the atmosphere of Beijing (Hu et al., 2017).

Additionally, Fig. 2 shows 2-day back-trajectories obtained by the NOAA HYSPLIT model, with one trajectory related to each sampled filter, starting at the median sampling time of each filter. Fig. 3 shows minutely recorded data for wind-direction and wind-speed collected by an Auto weather station (Met One Instruments Inc.) located on the same roof top as the aerosol sampling equipment. Both pictures are colored-coded with respect to $PM_{2.5}$ mass concentrations. The air masses that came from north or north-western directions were generally coincident with higher wind-speeds. They brought clean air with lower $PM_{2.5}$ mass concentrations. They did cross desert regions, however, Beijing was reported to be affected by desert dust mainly only in spring (Wu et al., 2009). Typically, the air masses coming from south and south-west of Beijing moved slowly and spent much more time over industrialized regions, resulting in high particulate matter mass concentrations. This here observed pattern is typical for Beijing, and these connections between wind-direction and pollution levels in Beijing have been analyzed in detail previously in Wehner et al. (2008).

3.2 Particle number concentrations

Fig. 4 shows the time series of the total number concentration of particles from 3 nm up to 2.5µm (N_{total}) and the number concentration of particles larger than 500 nm ($N_{>500nm}$), where N_{total} varied from $3*10^3$-$7*10^4$ cm^{-3} and $N_{>500nm}$ varied from 10 to $4*10^3$ cm^{-3}. Obviously, in the atmosphere of Beijing during the sampling period, small particles less than 500 nm account for a large faction of the total particle number concentration, but during strong pollution events, also a large increase in $N_{>500nm}$ is seen.

Fig. 5(a) and Fig. 5(b) show INP number concentrations (N_{INP}) as a function of temperature for INDA measurements. The lines are colour coded depending on the $PM_{2.5}$ mass concentration (Fig. 5(a)) and 12h-average $N_{>500nm}$ (Fig. 5(b)) during the respective filter sampling, where each line (30 in total) represents an individual result of a filter. Exemplary measurement uncertainties are given in section 3 of the appendix. All filter samples had INP that were active at -12.5°C and the highest freezing temperature was observed to be -6°C. Overall, N_{INP} varied from 10^{-3} to 1 L^{-1}. Already at a first glance, there is no clear trend in N_{INP} with $PM_{2.5}$ mass concentration and 12h-average $N_{>500nm}$, indicating that the dominant pollutants of urban atmosphere may not significantly contribute to INPs active down to roughly -16°C in an urban region.

To verify the results observed in INDA at lower temperatures, $PM_{2.5}$ collected by PTEF filters in the same period were used for LINA which can test the ice nucleating properties of droplets down to below -20°C. Washing particles off from the PTFE filters was more complete for some filters than for others. This showed in differently large deviations in N_{INP} from INDA and LINA measurements in the overlapping temperature range, where results determined from INDA were always similar to or higher than those from LINA as particle removal by washing the filters was frequently incomplete. It is mentioned in Conen et al. (2012), that a quantitative extraction of particles from quartz fiber filters was not possible without also extracting large amounts of quartz fibers. We tried to overcome this issue by using PTFE filters, as degradation of the PTFE filter during washing does not occur due to the hydrophobic properties of the filter material. But we observed that not all particles were released into the water during the washing procedure (likely those collected deep within the filter), as filters frequently still looked greyish after washing, independent from the washing procedure (we experimented with different washing times up to 4 hours and with the use of an ultrasonic bath).

For our INDA measurements, punches of quartz filters were measured after they were immersed in water, representing the ice nucleating properties of all collected particles (Conen et al., 2012). However, as already mentioned above, N_{INP} derived from LINA measurements were lower than those from INDA, due to particles that did not come off during washing. Based on our observations, we cannot recommend the use of sampling on PTFE filters followed by particle extraction in water. But we still decided to select those data from LINA measurements that showed the lowest deviation to the respective INDA results in the overlapping temperature range for use in this study. After calculating the deviation between INDA and LINA results, represented as the factor (N_{INP} of INDA / N_{INP} of LINA), ten LINA measurements from different days were selected to be used. For these measurements, the factor representing the deviation was in a range from 1.3 to 4.4. These data are shown in Fig. 5(c) and Fig. 5(d). The LINA data is represented by the dotted lines and the respective INDA data from the same sampling periods is represented by solid lines. In the temperature from -20°C to -25°C, results of LINA also show no clear trend in N_{INP} with $PM_{2.5}$ mass concentration and 12h-average $N_{>500nm}$, even though a lower temperature has been involved, extending our statement that urban pollution might not contribute to INP down to -25°C.

3.3 Correlation of N_{INP} with $PM_{2.5}$, and BC mass concentration and particle number concentrations

There have been many studies carried out in field and laboratory focusing on the ice nucleating properties of BC particles, however with inconclusive results. Some held the view that BC is not an efficient ice nucleation active species (Kamphus et al., 2010; Schill et al., 2016), whereas some described BC particles as possible INPs (Cozic et al., 2008; Cozic et al., 2007).

Here we selected N_{INP} derived from INDA measurements at -16°C and plotted them against BC (Fig. 6 (a)), $PM_{2.5}$ mass concentrations (Fig. 6 (b)) and 12h-average values of N_{total} (Fig. 6 (c)), $N_{>500nm}$ (Fig. 6 (d)), and N_{INP} at -16°C derived from DeMott et al. (2010) (Fig.6 (e)) and DeMott et al. (2015) (Fig.6 (f)). To determine the latter two, the 12h-averages of $N_{>500\ nm}$ shown in Fig. 3 were used, following parameterizations suggested by DeMott et al. (2010, 2015). Linear fits are included in all panels of Fig. 6, and values for R^2 and p for these fits are shown in Table 1 Our results discussed in the following, based on N_{INP} at -16°C, are similarly valid for all other temperatures down to -25°C.

Fig. 6(a) to (f) show that there was no clear trend between N_{INP} and any of the displayed parameters, be it BC or $PM_{2.5}$ mass concentration or any of the 12h-average particle number concentrations. Also the R^2 and p values given in Table 1 clearly show that there was no correlation between N_{INP} and any of the examined parameters In the urban region of Beijing during winter, the INP could be assumed to be soot or ash particles from traffic emissions, biomass burning and coal combustion, or to be dust particles advected from the desert regions during prevailing northern and north-western wind, or to originate from the biosphere. While mineral dust and biological particles are generally assumed to be the most abundant INP in the atmosphere (Murray et al., 2012, Kanji et al., 2017), the role of particles from combustion, i.e., of soot and ash particles, as INP is still controversial (Kanji et al., 2017). Our results indicate that BC particles did not correlate with INP concentrations in the urban atmosphere. It is possible that the BC particles emitted from coal burning, biomass burning, and traffic emissions are not ice active in the first place, or that they underwent atmospheric aging processes (such as coagulation, condensation upon vapor, and chemical reaction) resulting in more internally mixed particles after emission (Pöschl, 2005), which might inactivate their potential to act as INP. In the atmosphere of Beijing, the aging timescale is much shorter than in cleaner urban environments, which was shown in Peng et al. (2016). For example, to achieve a complete morphology modification for BC particles in Beijing, the aging timescale was estimated to be 2.3 h, compared to 9 h in Houston (Peng et al., 2016). $PM_{2.5}$ chemical composition indicated that the BC particles may be aged and coated by secondarily formed chemical components (SNA and other secondary organic materials) during the heavy haze episodes (Peng et al., 2016), thereby, resulting in weakened heterogeneous ice nucleation activity of freshly emitted diesel soot particles (Kulkarni et al., 2016).

However, if a possible coating was soluble, it would dissolve both during immersion freezing and during our experiments and would not impede the ice activity of BC particles, unless it reacted chemically with an ice active site. It has been observed that a coating did not impede the ice activity of mineral dust particles coated with nitric acid in Sullivan et al. (2010) and coated with succinic acid or levoglucosan in Wex et al. (2014).

Another study conducted in Ulaanbaatar in Mongolia, a city suffering from severe air pollution, showed a low ice activity towards heterogeneous ice nucleation when the sulphur content of particles was highest (Hasenkopf et al., 2016). It is interesting to note that we observe the opposite in our study, i.e., the increase of $PM_{2.5}$ mass concentration and percentage of SNA in $PM_{2.5}$ during haze periods also

seem to have no negative impact on INP concentrations. Not only did increased BC mass concentrations not increase the observed INP concentrations, but also were INP concentrations not particularly low during pollution episodes. Furthermore, we conclude that the strong formation of secondary particulate matter during haze days would not contribute to INP. In addition, there is no clear difference of ice nucleation between day and night time samples.

The size distribution measurements show that the largest fraction of all particles occurred in the size range below 500 nm. However, during the strongest pollution event towards the end of our measurement period (Dec. 17 during daytime (1217D) till the night from Dec. 21 to Dec. 22 (1221N)), also $N_{>500nm}$ increased noticeably to much larger values than before. In general, also particles in this size range were affected by the pollution, e.g., by an increase in size of pre-existing particles via atmospheric aging processes (such as coagulation, condensation, chemical reaction) where particles advected from southern industrial areas of Beijing might also contribute. This is at the base of the explanation why the parameterizations for N_{INP} by DeMott et al. (2010, 2015) were not able to describe the measured values, as seen in Fig. 6 (e) and (f). Additionally, the time series of N_{INP} at -16 ºC, based on DeMott et al. (2010, 2015) and are shown as blue and green squares in Fig. 7, respectively. Also shown are values for N_{INP} at -16°C as measured by LINA (red circles), i.e., the same values used in Fig. 7. Mostly, the parameterization by DeMott et al. (2015) yields larger values and a larger spread, compared to the parameterization by DeMott et al. (2010), but naturally both follow the trends in $N_{>500nm}$. A correction factor of 3, as suggested in DeMott et al. (2015), was not applied, as this would simply increase all respective values by this factor, i.e., it will not change the results. Indeed, during the pollution phase, the parameterizations overestimate the observed values by more than two orders of magnitude. But also during clean phases, neither $N_{>500nm}$ nor the parameterizations by DeMott et al. (2010, 2015) correlate with N_{INP}. Summarizing, this shows that pollution events not only did not add INP, but also that for the aerosol observed during our study, a parameterization of N_{INP} based on particles in the size range > 500 nm is not feasible. Interestingly, as will be shortly discussed in the next section, a much older parameterization by Fletcher (1962) captures N_{INP} as measured in this study rather surprisingly well, at least within one order of magnitude (Fig. 8). In summary, during polluted days, the increase of BC concentration, secondary components (SNA) and other compounds contributing to $PM_{2.5}$, as well as particle concentrations have no impact on INP concentrations down to -25°C in the urban region we examined in our study. This means that anthropogenic pollution did not contribute to the INP

concentration. But it also indicates that that anthropogenic pollution in Beijing did not deactivate the present INP, as polluted periods did not show particularly low INP concentrations, although aging and formation of secondary particulate matter typically are intense during times of strong pollution.

Additionally, also no correlation was found between any of the water-soluble constituents that were analyzed with ion chromatography and INP concentrations. This is not too astounding, as INP make up only a small fraction of all particles, as can be seen when comparing number concentrations from Fig. 4 and Fig. 7, and hence they make up only a small fraction of the mass, likely too small to be detected. Furthermore, a number of different components might contribute to INP, e.g., biological INP that are generally ice active at higher temperatures (> -15°C) and mineral dusts which are ice active at lower temperatures, therefore one common tracer for INP might not be applicable. As far as K is concerned, which might be connected to K-feldspar containing mineral dust particles with high ice activity (Atkinson et al., 2013), we only analyzed the water soluble fraction, i.e., K related to feldspar would not have been analyzed. Moreover, K is also emitted by biomass burning and hence influenced by anthropogenic pollution. It remains to be seen if a simple correlation between chemical constituents of the atmospheric aerosol and INP concentrations can be established at all.

3.4 Comparison with literature

First, we compare our results with results of N_{INP} derived from precipitation samples as collected in Petters and Wright (2015) as shown in Fig. 8. These literature data were collected in various locations in North America and Europe, and none of these locations was one with strong anthropogenic pollution, different from the sample location in the present study. The N_{INP} in our study varied from 10^{-3}-10 L^{-1}·air at the temperature range of -10°C to -25°C. The data of this study (dark green and brownish lines) are within the range of values given in Petters and Wright (2015), in the whole temperature range for which INP concentrations were derived here. A comparison with Corbin et al. (2012) and Knopf et al. (2010), who bothexamined INP also in urban air in Toronto and Mexico City, respectively, is not possible due to different examined ice nucleation modes, and also because they only measured at -34°C (Corbin et al., 2012), i.e., outside of the temperature range examined in this study, or only reported ice onset temperatures (Knopf et al., 2010). But we want to point towards the fact that an older parameterization based on Fletcher (1962), which has been used for large scale modelling, agrees well with our data (see

Fig. 8) down to -20°C. It should, however, also be pointed out that the occurring variability in the data certainly cannot be captured by such a single line. But the increase in N_{INP} towards lower temperatures as parameterized in Fletcher (1962) is similar to that of our data, where it should also be said that this parameterization is known to overestimate atmospheric observations at lower temperatures (roughly below -25°C, see e.g., Meyers et al., 1992). A similar observation was recently described in Welti et al. (2017), where down to -20°C the temperature trend of N_{INP} derived from filter samples taken on the Cape Verde islands also agreed well with the parameterisation by Fletcher (1962), while at lower temperatures, the parameterization exceeded the measurements. In general, for the case of the Beijing air masses examined in this study, both the range of N_{INP} given in Petters and Wright (2015) as well as the parameterization by Fletcher (1962) agree better with our measurements than the parameterizations by DeMott et al. (2010, 2015).

All of this is again indicative for the fact that Beijing severe air pollution did not increase or decrease INP concentrations above or below values typically observed in other, non-urban areas on the Earth, and hence, that the background INP concentrations, at least down to -25°C might in general not be directly anthropogenically influenced.

Measurements of N_{INP} in China have been done as early as 1963 by You and Shi (1964), and a few further studies listed in Table 2 have been carried out in recent years. Table 2 includes some campaigns finished in different regions of China including mountains, plateaus and suburban districts with low $PM_{2.5}$ concentration and BC-containing particles. In contrast to these observations, our study shows N_{INP} detected in an urban region during highly polluted days with complex particle sources. In our study, immersion freezing was examined, while not all studies listed in Table 2 examined this ice nucleation mode. But due to the scarcity of data, we include the results from all these studies in our discussion here. Apparently, compared with results in Table 2, N_{INP} determined for the urban site of this study (1 L^{-1} Air at -20°C) was on the lower end of reported values, which were up to roughly 20 L^{-1} Air at -20°C for non-dust events. Highest concentrations were observed for dust events with values up to 604 L^{-1} ·Air at -20°C detected at a suburban site in Beijing, showing that INP from mineral dust contribute to the overall N_{INP} already at this temperature (You et al., 2002). Despite the difference among methods and ice nucleating modes, this again suggests that urban aerosol particles might not be efficient immersion freezing INP and that the ice nucleating ability of particles in urban aerosols might originate from the non-urban

419 background aerosol particles that are included in the urban aerosol, i.e., that INP observed in urban
420 environments might have the same sources among bioaerosols and dust particles as non-urban INP.

421 **4 Conclusions**

422 INP concentrations down to -25°C determined from $PM_{2.5}$ samples collected at an urban site of the
423 megacity Beijing, China, in winter were found to not be influenced by the highly variable amount of air
424 pollution, both in mass and particle number concentrations, that was present during the sampling period.
425 Therefore, we conclude that neither BC nor other pollutants contributed to INP, including secondarily
426 formed particulate mass. On the other hand, we also conclude that the present INP were not noticeably
427 deactivated during strong pollution events. Particle number concentrations for particles with diameters >
428 500nm were affected by pollution events, and INP concentrations did not correlate with these
429 concentrations. Therefore, as can be expected, parameterizations based on these concentrations (DeMott
430 et al., 2010, 2015) do not reproduce the INP concentrations under these extreme conditions and yield
431 values which are up to more than two orders of magnitude higher than the measured values. On the other
432 hand, INP concentrations were in the middle of the range reported for atmospheric, non-urban,
433 concentrations in Petters and Wright (2015), and on the lower end of reported values collected from
434 previous atmospheric observations in China, while they were much lower than observations during dust
435 events in China. From this, we conclude that INP concentrations might not be influenced directly by
436 anthropogenic activities, at least not down to roughly -25°C and maybe even below, and that particularly
437 natural mineral dust sources might effect INP concentrations observed in China. It should be noted that
438 ice nucleation observed at high freezing temperatures (particularly above -10°C, but maybe as low as -
439 20°C) is typically attributed to biogenic ice activity. But while identifying the nature of the INP detected
440 here is beyond the reach of our study, we assume that they originated from natural sources and not from
441 anthropogenic combustion sources. However, it should be kept in mind that an indirect anthropogenic
442 influence on INP concentrations is still possible due to land use changes and related changes in
443 atmospheric dust loadings as well as due to vegetation changes and related changes in the biosphere.
444
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Appendix

1. Temperature calibration and background of INDA and LINA

1.1 Temperature calibration of INDA

The bath of the thermostat was well mixed during the cooling cycle, and the cooling rate was at 1 K min^{-1}. PCR trays were immersed into the cooling liquid such that the water level in the tubes was below the level of the liquid in the thermostat. The temperature inside the tubes was determined before the experiments by putting a temperature sensor into a tube during cooling. This was repeated for tubes in several locations. This worked down to -7°C, below which the sensor induced freezing. In this temperature range, generally a small constant shift of 0.2 K was observed which was assumed to be overall valid and was incorporated in the data at all temperatures. A comparison of data obtained for suspensions of Snomax with previous work done at TROPOS with LACIS (Leipzig Aerosol Cloud Interaction Simulator) and within INUIT (Ice Nuclei Research Unit, (Wex et al., 2015)) showed good agreement down to the lowest temperature at which the experiments for the comparison were run, which was -16°C.

1.2 Temperature calibration of LINA

The temperature on the glass slide in LINA was obtained by feeding an air flow with a known dew point temperature through the instrument, while the instrument cooled down with 1 K min^{-1}, i.e., with the same freezing rate used in the experiments. The humidified air flow was obtained by mixing a dry air flow with an air flow that was humidified in a Nafion humidifier (Perma Pure MH-110-12S-4, Perma Pure, Toms River, New Jersey, USA) which was connected to a thermostat (HAAKE C25P, HAAKE GmbH, Karlsruhe, Germany) that kept the temperature in the humidifier at 10°C. By mixing the two air streams, dew point temperatures below 0°C were obtained. The dew point temperature was measured with a dew point mirror (Dew Prime I-S2, Edge Tech, Milford, Massachusetts, USA). The overall setup is based on the principle of a dew point mirror, i.e., the glass slide on the Peltier element in LINA started to fog when its temperature reached the dew point temperature adjusted in the air flow. Optical detection by the CCD camera was deployed similar to how it is used during measurements, i.e., taking a picture every 6 s. Subsequently detected greyscale images were compared to an image that was taken well before fogging began. Brightness differences between this original picture and the following pictures were taken and resulted in a S-shaped curve, reaching the maximum plateau once the glass slide was fogged over

completely. A fit was applied to the curve in order to find the temperature where 50% are fogged, which was taken to represent the actual temperature. Using this principle, the temperature on the glass plate in LINA was calibrated repeatedly at 5 different temperatures in the range from -2.3°C to -22.3°C. A comparison of data obtained for suspensions of pollen washing water with previous work done at TROPOS with LACIS (Augustin et al., 2013) showed good agreement down to the lowest temperature at which the experiments for the comparison were run, which was -25°C.

2. Background measurement of INDA and LINA

In the background experiments of INDA, clean filters in distilled water froze from -17°C to -26°C, while filters with atmospheric particles froze from -6°C to -22°C. The f_{ice} of the clean filters was 5 to 14 times lower than that of atmospheric samples at the same temperature, showing a low impact. In LINA measurements, the background of clean filters washed with distilled water was even lower, as droplets started to freeze at -22°C. Figure A1 and A2 show the measured frozen fractions of the samples and the background from pure water and the water with clean filters for both INDA and LINA, to corroborate that the measurements were well separated from the background.

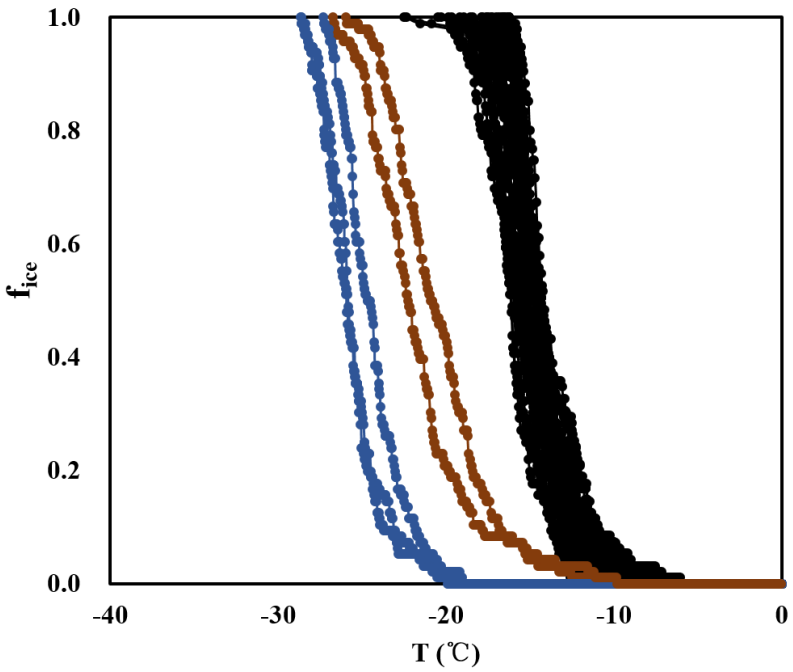

Fig. A1 Frozen fractions determined from INDA (black lines), together with background signals determined for pure water (blue lines) and for pure water containing punches of a clean filter (brown lines).

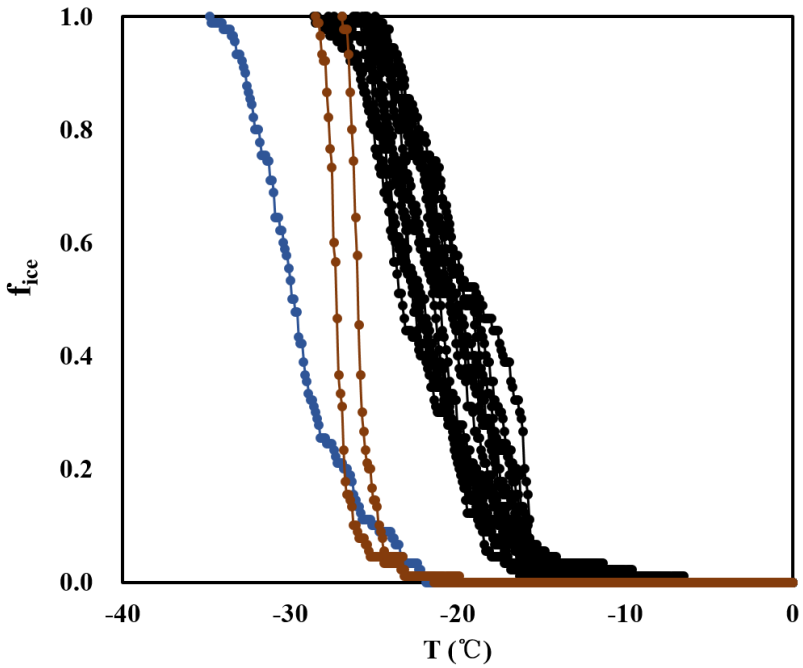

Fig. A2 Frozen fractions determined from LINA, together with background signals determined for pure water and for pure water in which a clean filter was put and washed, similar to the procedure for the samples.

3. Measurement uncertainty for INP measurements

The highest and lowest freezing curved detected with INDA are shown exemplarily in Fig. A3 together with the measurement uncertainty. The derivation of the uncertainty was based on the fact that at each temperature, all INP that are ice active at that or any higher temperature are Poission distributed to the examined droplets. It followed a method described in Harrison et al. (2016). For LINA, no uncertainties are given, as we know that washing off from the filters was incomplete, and the fraction of particles that was retained on the filters cannot be determined. The largest deviation that we allowed between LINA and INDA, i.e., a factor of 4.4 (see Sec. 3.2), is the base for the maximum uncertainty for fice detected with LINA. For both, INDA and LINA, the temperature uncertainty is 0.5K.

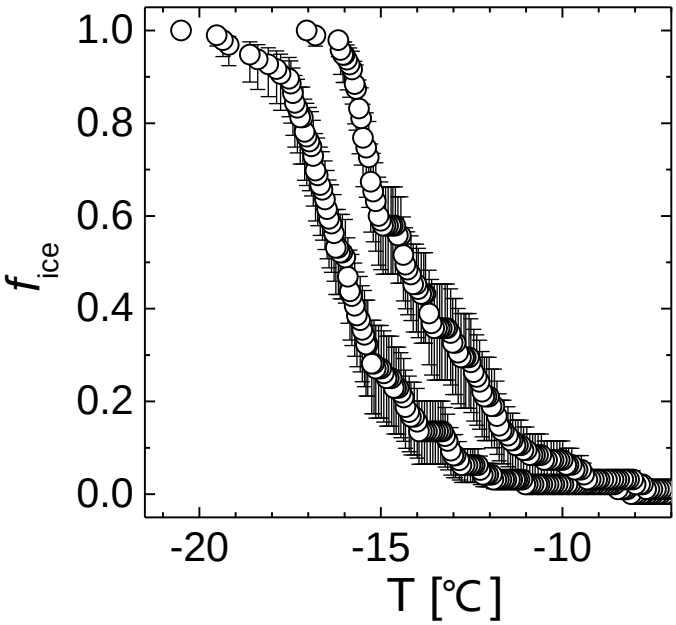

Fig. A3. The highest and lowest freezing curved detected with INDA together with the measurement uncertainty.

Acknowledgments

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Table and Figures:

Table 1 Coefficient of determination (R^2) and a measure for the statistical significance of the assumption of a linear correlation (p) for the comparison of N_{INP} at -16°C with the different parameters shown in Fig. 5.

parameter	R^2	p
(a) BC concentration	0.003	0.79
(b) $PM_{2.5}$ concentration	0.006	0.71
(c) N_{total}	0.005	0.73
(d) $N_{>500nm}$ at -16°C	0.008	0.67
(e) N_{INP} at -16°C, based on DeMott et al. (2010)	0.005	0.73
(f) N_{INP} at -16°C, based on DeMott et al. (2015)	0.007	0.67

826 **Table 2. Comparison of INP measurements in different regions of China, including N_{INP} (i.e., INP number**
827 **concentrations) and corresponding temperature**

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Sampling site	Citation	Sampling Date	Instruments	Temperature (°C)	Average INP (L^{-1}·Air)	Mode
Huangshan (mountain site)	(Jiang et al., 2015)	September-October,2012	Vacuum water vapor diffusion chamber	-15~-23	0.27~7.02	Deposition
Huangshan (mountain site)	(Jiang et al., 2014)	May-September,2011	Mixing cloud chamber The static diffusion cloud chamber	-20	16.6	Deposition/ Condensation
Huangshan (mountain site)	(Hang et al., 2014)	May-September,2011; September-October,2012	Static vacuum water vapor diffusion cloud chamber	-20	18.74	All modes
Tianshan (mountain site)	(Jiang et al., 2016)	14-24 May, 2014	Vacuum water vapor diffusion chamber;	-20	11(non-dust) Hundreds(dust)	Deposition
Nanjing (suburban site**)**	(Yang et al., 2012)	May-August,2011	Mixing cloud chamber; The statistic diffusion chamber;	-20	20.11	All modes
Qing Hai (plateau site)	(Shi et al., 2006)	5-26 October, 2003	The Bigger mixing cloud chamber	-15, -20, -25	23.3~85.4	Deposition
Beijing (suburban site)	(You and Shi, 1964)	18 March-30 April,1963	Mixing cloud chamber	-20	3.9~4.8	All modes
Beijing (suburban site)	(You et al., 2002)	18 March-30 April,1995	The Bigger mixing cloud chamber	-15, -20	21,78.9(non-dust) 604(dust)	All modes
Beijing (urban site)	This study	27 November-22 December, 2016	Ice Nucleation droplets Array	-10 ~ -28	0.001~10	Immersion

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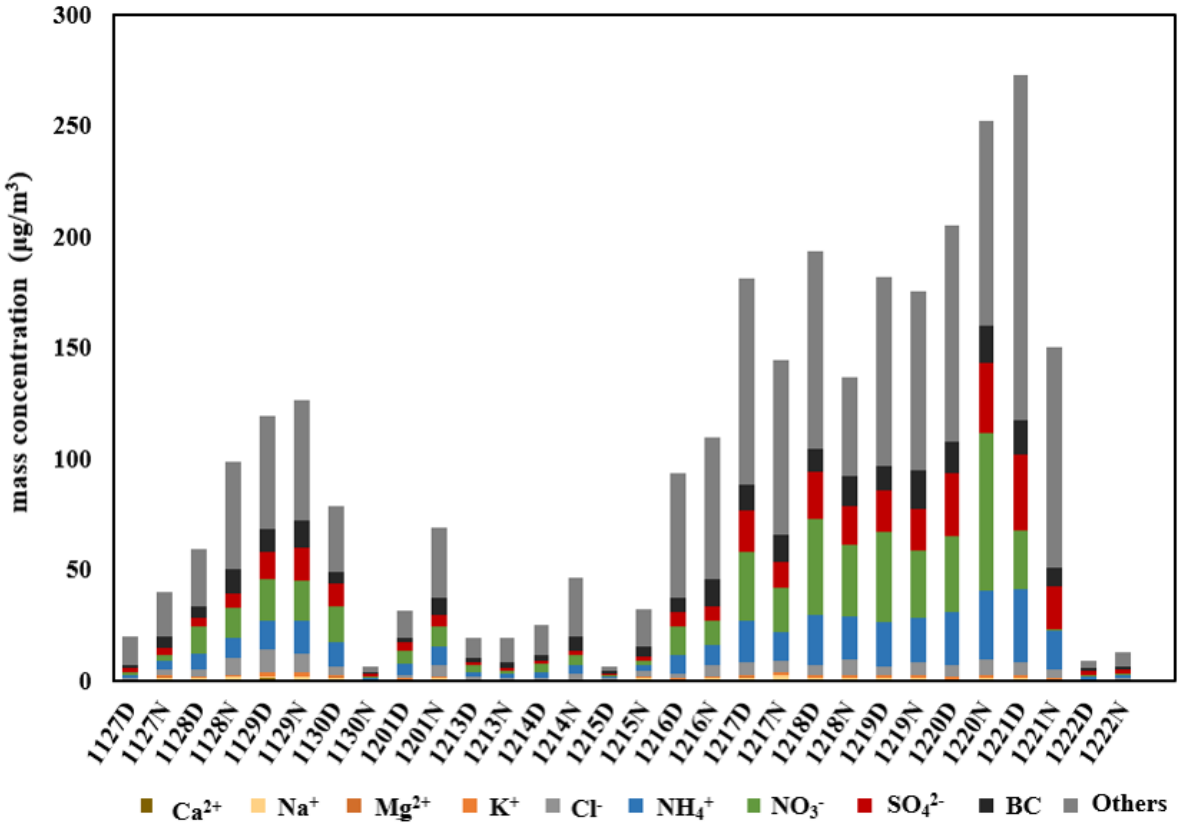


Figure 1. The time series of $PM_{2.5}$ concentrations and chemical composition. Data are shown for 15 different days where the dates are indicated in the x-axis-labeling and "D" and "N" stand for daytime and nighttime, respectively.

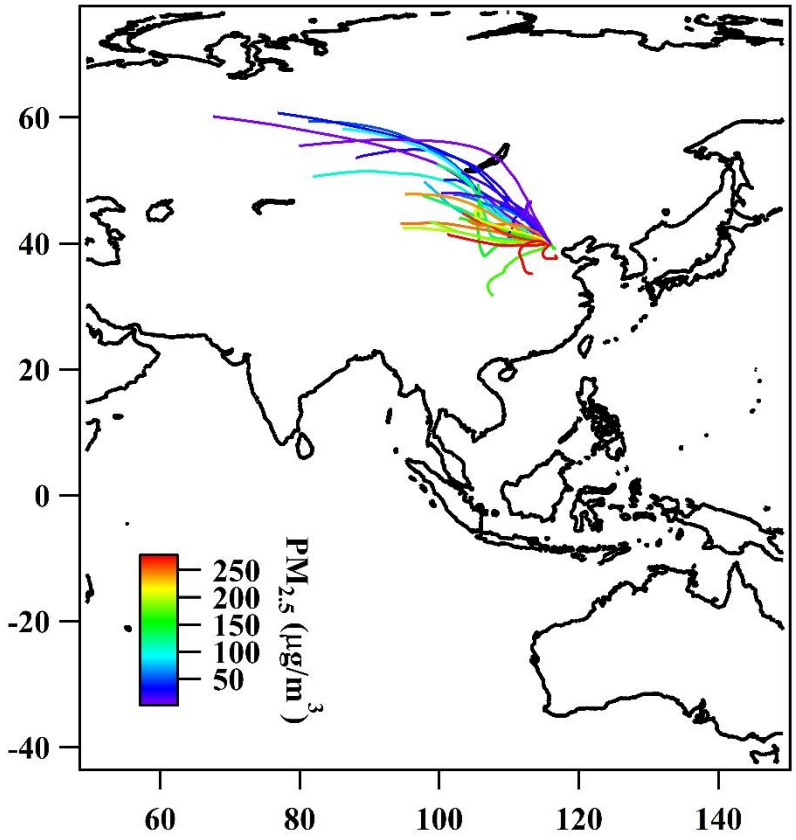


Figure 2. The 2-day back-trajectories obtained by the NOAA HYSPLIT model colored-coded with respect to $PM_{2.5}$ mass concentration determined by PTEF filter.

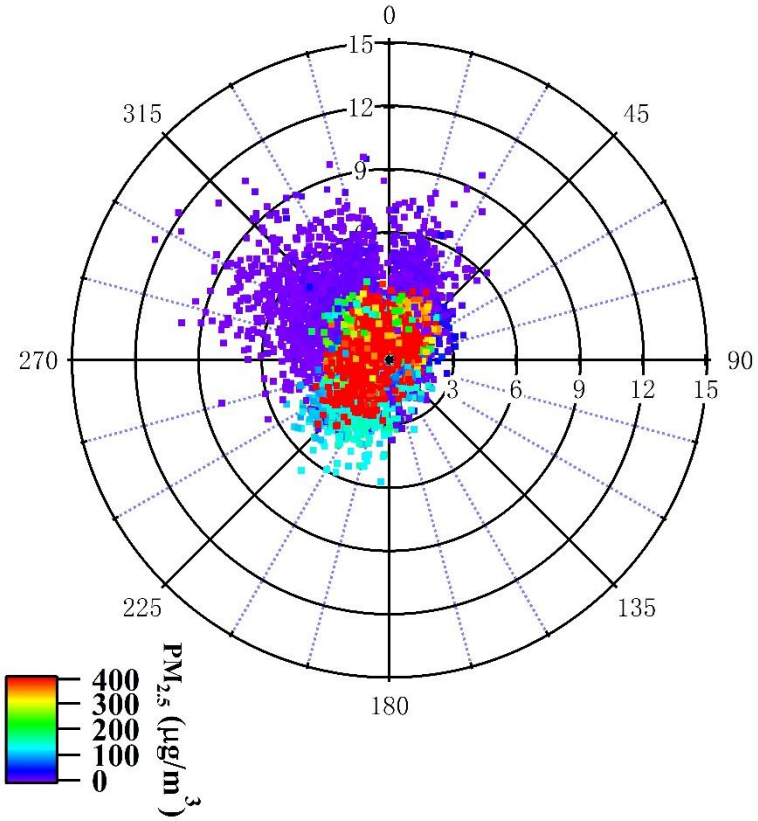
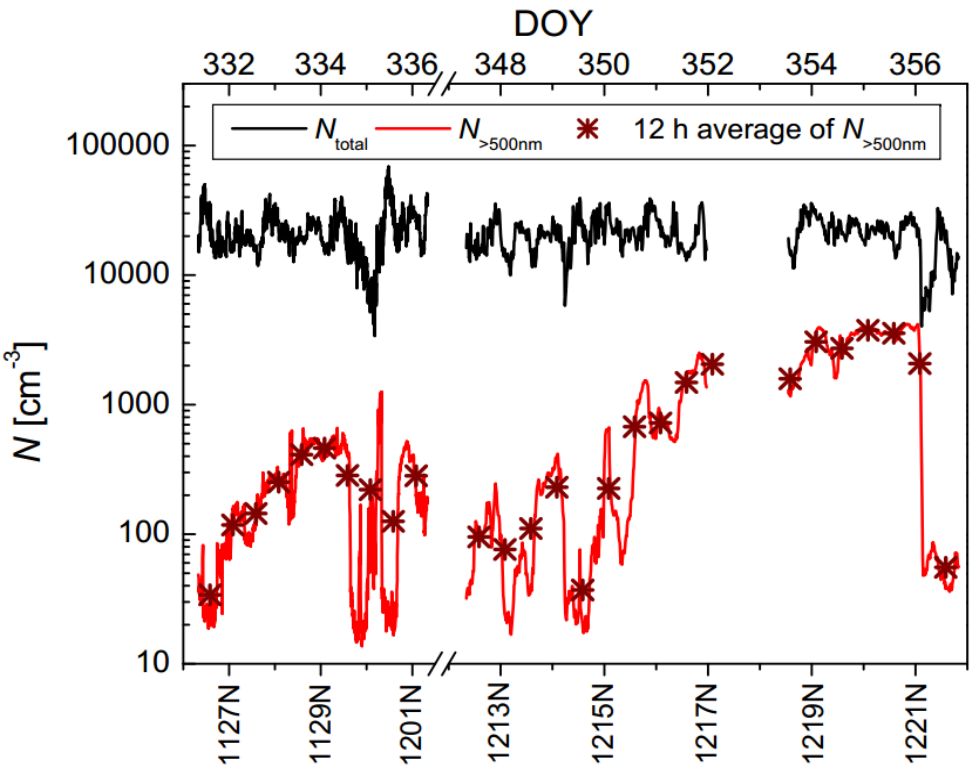


Figure 3. Minutely recorded data for wind-direction and wind-speed colored-coded with respect to $PM_{2.5}$ mass concentration.



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860 **Figure 4. The time series of N_{total}, $N_{>500nm}$ and 12-h average $N_{>500nm}$ at -16°C.**

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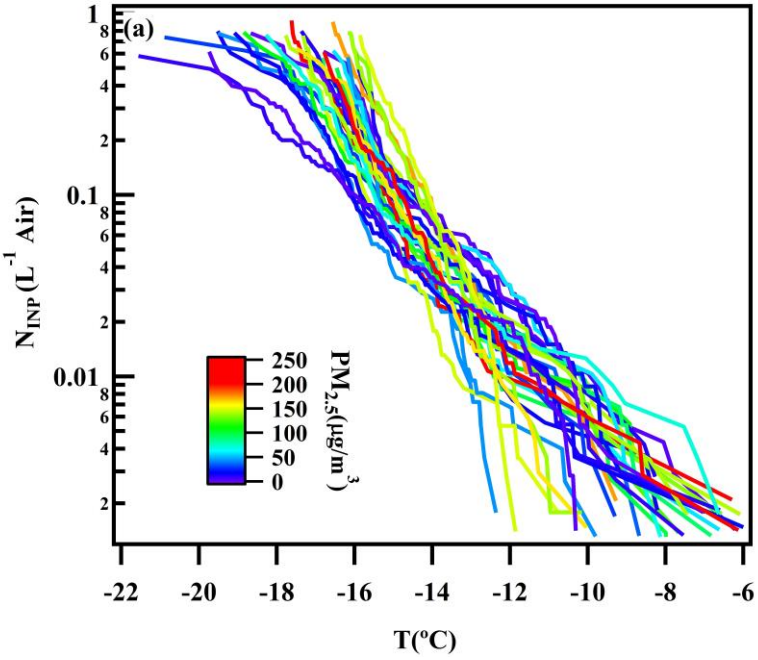
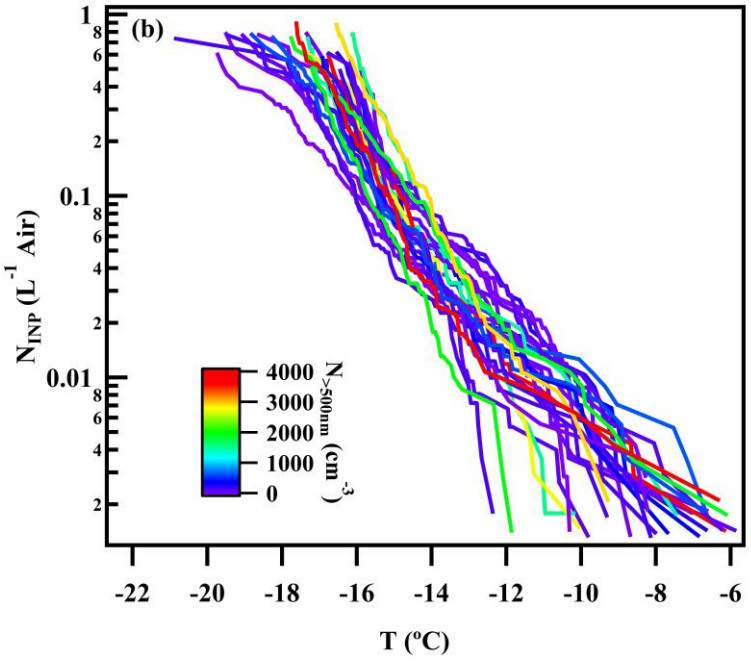
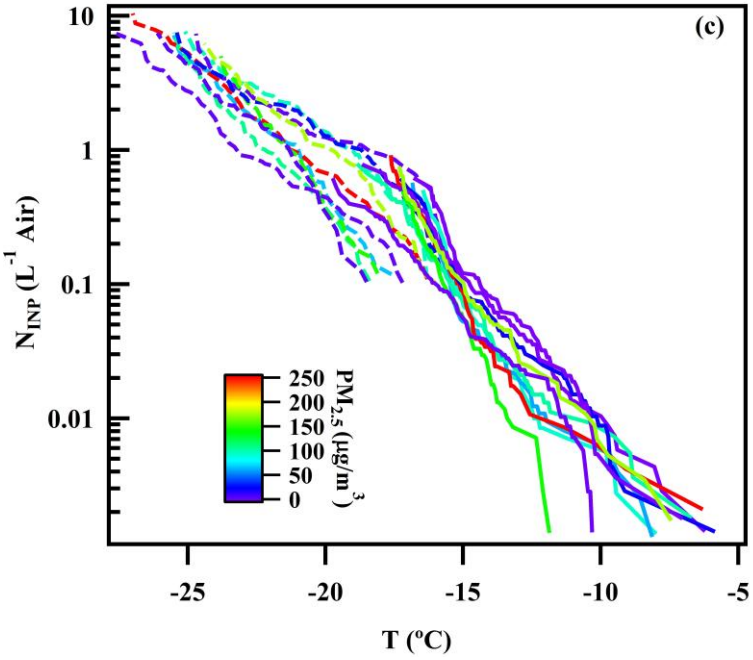

(a)
N_{INP} (L^{-1} Air)
1
0.1
0.01
250
200
150
100
50
0
$PM_{2.5}$($\mu g/m^3$)
-22
-20
-18
-16
-14
-12
-10
-8
-6
T(ºC)

863


(b)
N_{INP} (L^{-1} Air)
1
0.1
0.01
4000
3000
2000
1000
0
$N_{>500nm}$ (cm^{-3})
-22
-20
-18
-16
-14
-12
-10
-8
-6
T (ºC)

864



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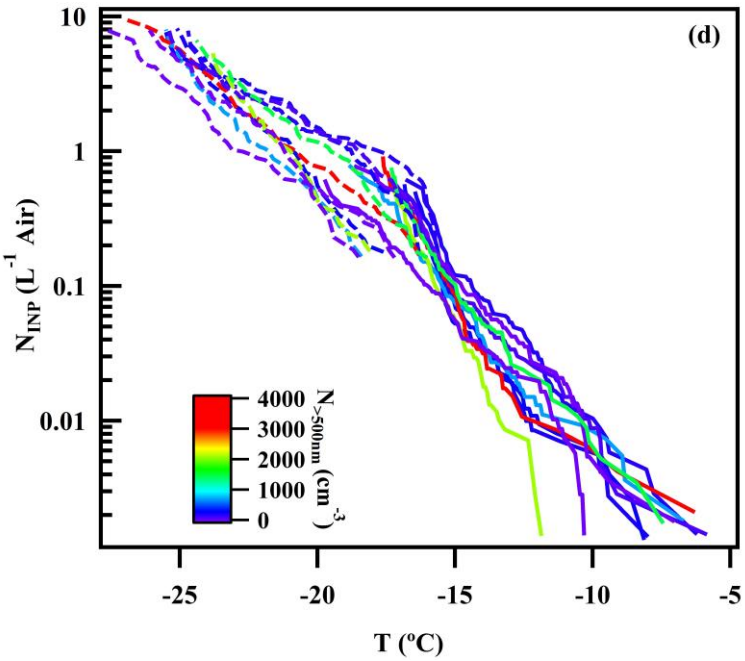

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868 **Figure 5.** N_{INP} **as function of temperature, panel (a) and (b) show INDA results coloured by** $PM_{2.5}$ **mass**
869 **concentration and 12h-average** $N_{>500nm}$**, (c) and (d) for 10 comparable results of INDA and LINA coloured by**
870 $PM_{2.5}$ **mass concentration and 12h-average** $N_{>500nm}$**, dotted lines represents LINA results while solid lines**
871 **represents INDA results.**

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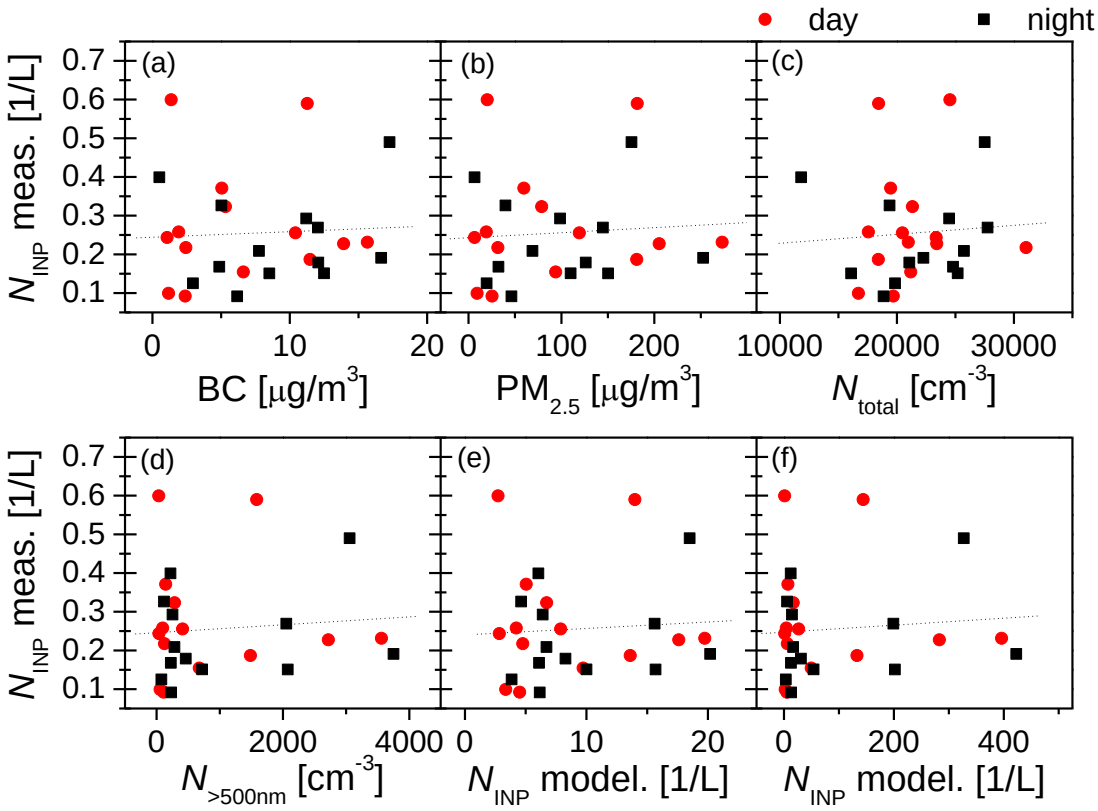

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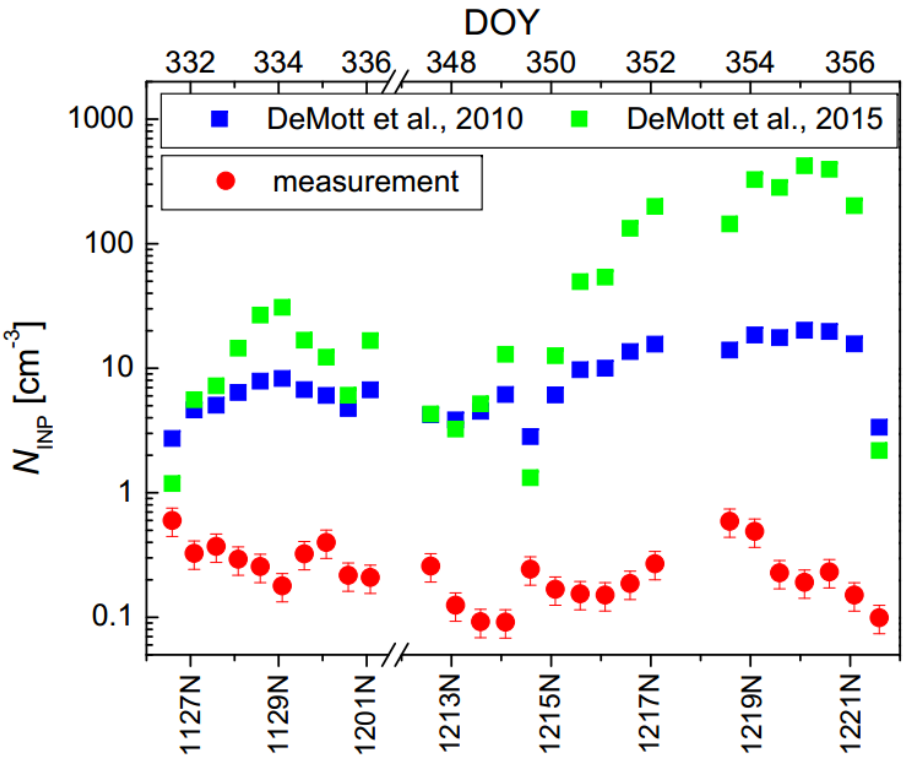
875 **Figure 6.** N_{INP} **at -16°C as function of mass concentrations of BC (a) and** $PM_{2.5}$ **(b), and of 12h-average values**
876 **of** N_{total} **(c). Furthermore, we show** $N_{>500nm}$ **(d), and** N_{INP} **at -16°C derived based on (DeMott et al., 2010) (e)**
877 **and DeMott et al. (2015) (f) for daytime (red round symbols) and nighttime (green square symbols) samples.**



878

879 **Figure 7. The time series of measured N_{INP} and N_{INP} parameterized according to DeMott et al. (2010, 2015) at**
880 **-16°C.**

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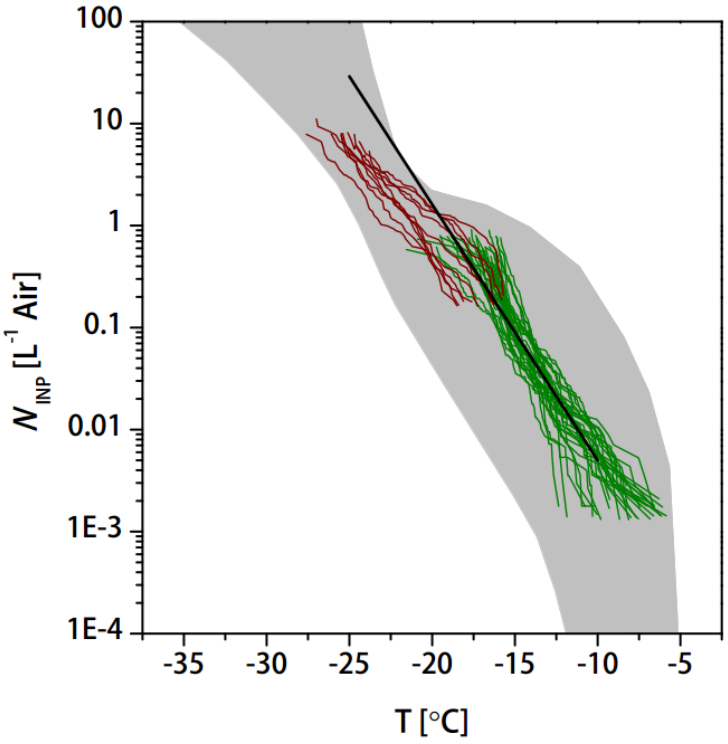

885

886 **Figure 8.** N_{INP} **as derived from precipitation samples collected in Petters and Wright (2015) (grey area) and a**
887 **parameterization based on Fletcher (1962) (black line), together with our results (dark green and brownish**
888 **lines from INDA and LINA measurements, respectively).**

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