



Primary and secondary organic aerosols in 2016 summer of Beijing

Rongzhi Tang¹, Zepeng Wu¹, Xiao Li¹, Yujue Wang¹, Dongjie Shang¹, Yao Xiao¹, Mengren Li¹, Limin Zeng¹, Zhijun Wu¹, Mattias Hallquist², Min Hu¹, Song Guo^{1,*}

¹State Key Joint Laboratory of Environmental Simulation and Pollution Control, College of Environmental Sciences and Engineering, Peking University, Beijing, 100871, PR China

² Atmospheric Science, Department of Chemistry and Molecular Biology, University of Gothenburg, Sweden

* Correspondence to: Song Guo, <u>songguo@pku.edu.cn</u>





1 Abstract

2 To improve the air quality, Beijing government has employed several air pollution 3 control measures since 2008 Olympics. In order to investigate the organic aerosol 4 sources after the implementation of these measures, ambient fine particulate matters 5 were collected at a regional site Changping (CP) and an urban site Peking University 6 Atmosphere Environment MonitoRing Station (PKUERS) during the "Photochemical 7 Smog in China" field Campaign in summer of 2016. A chemical mass balance (CMB) 8 modeling and the tracer yield method were used to apportion the primary and 9 secondary organic sources. Our results showed that the particle concentration 10 decreased significantly during the last a few years. The apportioned primary and 11 secondary sources explained $62.8 \pm 18.3\%$ and $80.9 \pm 27.2\%$ of the measured OC at 12 CP and PKUERS, respectively. Vehicular emissions served as the dominant sources. 13 Except gasoline engine emission, the contributions of all the other primary sources 14 decreased. Besides, the anthropogenic SOC, i.e. toluene SOC, also decreased, 15 implying that deducting primary emission can reduce anthropogenic SOA. Different 16 from the SOA from other regions in the world, where biogenic SOA was dominant, 17 anthropogenic SOA was the major contributor to SOA, implying that deducting 18 anthropogenic VOCs emissions is an efficient way to reduce SOA in Beijing. Back 19 trajectory cluster analysis results showed that high mass concentrations of OC were 20 observed when the air mass was from south. However, the contributions of different 21 primary organic sources were similar, suggesting the regional particle pollution. The 22 ozone concentration and temperature correlated well with the SOA concentration. 23 Different correlations between day and night samples suggested the different SOA 24 formation pathways. Significant enhancement of SOA with increasing particle water 25 content and acidity were observed in our study, suggesting the aqueous phase 26 acid-catalyzed reactions may be the important SOA formation mechanism in summer 27 of Beijing.





28 1. Introduction

29 Beijing is the capital and a major metropolis of China. With the rapid economic 30 growth and urbanization, Beijing is experiencing serious air pollution problems, and 31 became one of the hotspots of PM_{2.5} (particular matters with size smaller than 2.5µm) 32 pollution in the world (Guo et al., 2014; Xiang et al., 2017; Tian et al., 2016). Due to 33 the frequent haze events in Beijing, Beijing government has taken a series of control 34 measures in recent years, especially after 2008 Olympics, which may greatly 35 influence the primary and secondary particle sources. Therefore, elucidating the 36 current contributions of primary particle sources as well as secondary particle sources 37 is of vital importance. It is also important to compare with the previous results to 38 evaluate the effectiveness of the control measures and shed light on the influence of 39 the primary source emission control on the secondary aerosol formation.

40 Several studies regarding to the source apportionment of fine particles in Beijing have 41 been conducted using multifarious methods during the last few years (Yu et al., 2013; 42 Gao et al., 2014; Zheng et al., 2016b; Tan et al., 2014; Wang et al., 2009; Guo et al., 43 2013). Receptor model is a commonly used method to apportion the particle sources 44 (Zhang et al., 2017; Zhou et al., 2017; Zhang et al., 2013; Song et al., 2006; Zheng et 45 al., 2005). Elemental tracers were previously used to apportion particulate matter sources (Yu et al., 2013; Gao et al., 2014; Zheng et al., 2016b). However, elemental 46 47 tracer-based method was unable to distinguish sources that mostly emit organic compounds instead of specific elements such as diesel/gasoline engines. Among all 48 49 the apportionment methods, chemical mass balance (CMB) model was one of the 50 most commonly used methods to apportion the primary organic sources of fine 51 particulate matter (Zhang et al., 2017; Hu et al., 2015; Schauer et al., 1996). Organic 52 tracers have been successfully used in several studies which aimed to quantify the 53 main sources of Beijing (Liu et al., 2016; Guo et al., 2013; Wang et al., 2009). Wang 54 et al. assessed the source contributions of carbonaceous aerosol during 2005 to 2007





55 (Wang et al., 2009). Guo et al. (Guo et al., 2013) and Liu et al. (Liu et al., 2016) apportioned the organic aerosol sources using CMB model in summer of 2008 and a 56 57 severe haze event in winter of 2013. Both studies found that vehicle emission and coal combustion were the dominant primary sources of fine organic particles. Tracer-yield 58 59 method has been considered as a useful tool to semi-quantify SOA derived from 60 specific VOCs precursors (Guo et al., 2012; Zhu et al., 2017; Zhu et al., 2016; Tao et 61 al., 2017; Hu et al., 2008). However, only a few studies have estimated secondary 62 organic aerosol in Beijing. Yang et al. (Yang et al., 2016) estimated the biogenic SOC to OC during CAREBEIJING-2007 field campaign, and found that the SOC 63 accounted for 3.1% of the measured OC. Guo et al. (Guo et al., 2012) illustrated the 64 65 SOA contributions in 2008, and found that secondary organic carbon could contribute 66 a great portion ($32.5 \pm 15.9\%$) to measured organic carbon at the urban site. Ding et al. (Ding et al., 2014) used the tracer-yield method to investigate the SOA loading on a 67 68 national scale and found that SOA, especially anthropogenic SOA played great role in 69 major city clusters of China.

70 In this study, we quantified 144 kinds of particulate organic species, including 71 primary and secondary organic tracers, at a regional site and an urban site of Beijing. 72 A CMB modeling and the tracer yield method were used to apportion the primary and 73 secondary sources of the organic aerosols in the 2016 summer of Beijing. The results 74 were compared with the previous studies to evaluate the effectiveness of control 75 measures on primary as well as secondary organic aerosols. Moreover, source 76 apportionment results from different air mass origins according to the back trajectory 77 clustering analysis were shown to investigate the influences of air mass from different 78 directions on the fine organic particle sources. Influencing factors of SOA formation, 79 i.e. temperature, oxidant concentration, aerosol water content, as well as particle 80 acidity were also discussed in this study to improve our understanding of SOA 81 formation under polluted environment.





82 **2. Experimental**

83 2.1 Sampling and Chemical Analysis

The measurements were conducted simultaneously at an urban site Peking University Atmosphere Environment MonitoRing Station (PKUERS, 39°59'21" N, 116°18'25" E) and a regional site Changping (CP, 40°8'24"N, 116°6'36" E) 40km north of PKUERS site during "Photochemical Smog in China" campaign, from May 16th to June 5th, 2016 (see Fig. S1). The PKUERS site is set on the roof at an academic building on the campus of Peking University in the northwest of Beijing. CP site is located on the fourth floor of a building on the Peking University Changping campus of Changping.

91 Four-channel samplers (TH-16A, Tianhong, China) consisting of three quartz filter 92 channel and one Teflon filter channel, were employed to collect 12-h aerosol samples at PKUERS and CP, respectively. The sampling flow rate was 16.7 L min⁻¹. Teflon 93 94 filters were weighed by a microbalance (Toledo AX105DR, USA) after a 24 h balance in an environmental controlled room (temperature 20 \pm 1°C, relative humidity 40 \pm 95 3%) for gravimetric analysis. Teflon-based samples were extracted by deionized water 96 to measure water-soluble inorganic compounds (WSICs), namely Na⁺, NH₄⁺, K⁺, 97 Mg^{2+} , Ca^{2+} , NO_3^- , SO_4^{2-} and Cl^- by DIONEX ICS-2500 and ICS-2000 98 99 ion-chromatograph. One punch (1.45 cm^2) of quartz-based sample was then cut off to 100 analyze the EC and OC via thermal-optical method using Sunset Laboratory-based 101 instrument (NIOSH protocol, TOT). The other two quartz filters were then extracted 102 and analyzed for chemical composition and particulate organic matters. Some daytime 103 and nighttime samples were combined to ensure the detection of most organic 104 compounds. To better understand the chemical speciation, daytime samples were 105 separated from nighttime samples.

Authentic standards were used to identify and quantify the organic compounds. The
analytical methods used in this study referred to the previous work (Song et al., 2014).
Briefly, the samples were first spiked with a mixture of internal standard, including





109 ketopinic acid (KPA), 20 kinds of deuterated compounds, and one carbon isotope 110 ¹³C-substituted compound. The filters were then ultrasonically extracted with methanol: dichloromethane (v:v=1:3) solvent in water bath (temperature < 30 °C) for 111 112 3 times. Each time was 20 min. The extracts were filtered, and then concentrated 113 using a rotary vacuum evaporator. An ultra-pure nitrogen flow was used to further 114 concentrate the extracts into 0.5-1 ml. Each extracted solution was divided into two 115 portions, one of which added BSTFA (BSTFA/TMCS = 99:1, Supelco) to convert 116 polar organic compounds into trimethylsilanized derivatives. Afterwards, the 117 derivatized and the untreated samples were analyzed by an Agilent 6890 GC-MS System (MSD GC-5973N) equipped with an Agilent DB-5MS GC column (30 m \times 118 119 $0.25 \text{ mm} \times 0.5 \text{ }\mu\text{m}$).

120 2.2 Source Apportionment

121 A chemical mass balance modelling developed by the U.S. Environmental Protection 122 Agency (EPA CMB version 8.2) was applied to determine the apportion of the 123 primary contribution of OC (Schauer et al., 1996). This receptor model solved a set of 124 linear equations using ambient concentrations and chemical source profiles. CMB 125 approach depends strongly on the representativeness of the source profile. In this 126 study, five primary source profiles including vegetative detritus (Rogge et al., 1993), 127 coal combustion (Zheng et al., 2005), gasoline engines (Lough et al., 2007), diesel 128 engines (Lough et al., 2007) as well as biomass burning (Sheesley et al., 2007) were input into the model. Fitting species including EC, n-alkanes, levoglucosan, 129 130 $17\beta(H)-21\alpha(H)$ -norhopane, $17\alpha(H)-21\beta(H)$ -hopane, benzo(b)fluoranthene, 131 benzo(k)fluoranthene, benzo(e)pyrene, benzo(ghi)perylene, indeno(1,2,3-cd)pyrene. 132 The criteria for acceptable fitting results included the square regression coefficient of the regression equation $R^2 > 0.85$ as well as the sum of square residual Chi-square 133 value $\chi^2 < 4$. 134





135 The tracer yield method was used to estimate the contributions of biogenic and 136 anthropogenic secondary organic aerosols using fixed tracers to SOC ratio (f_{SOC}) 137 based on laboratory experiments, which was 0.155 ± 0.039 for isoprene, 0.231 ± 0.111 for α -pinene, 0.0230 \pm 0.0046 for β -caryophyllene and 0.0079 \pm 0.0026 for toluene 138 139 (Kleindienst et al., 2007). The uncertainties of the estimation mainly derived from the 140 imparity in choosing the organic tracer compounds as well as the idealization of the 141 single-valued tracer mass fractions, which have been comprehensively discussed by 142 previous studies (Yttri et al., 2011;El Haddad et al., 2011) (Song et al., 2014). Despite 143 of its large uncertainties, the tracer-yield method was considered as a useful approach 144 up to now to roughly estimate the SOA contributions derived from individual 145 hydrocarbon precursors.

146 **3. Gaseous pollutants and particle chemical composition**

147 **3.1** Gaseous pollutants and meteorological conditions of the observation period

148 Mixing ratios of gaseous pollutants and meteorological conditions during the 149 observation period were shown in Fig. S2 and Table S1. Compared with the results in 150 summer of 2010 (Zheng et al., 2016a), the gaseous mixing ratios SO_2 and CO were 151 lower than before owing to the desulfurization efforts made by the government. 152 Higher concentrations of NO and NO₂ were caused by the increasing number of 153 vehicles. The increment of ozone indicated the importance of secondary pollution. 154 Clearly, ozone concentration at CP was higher than that of PKUERS while other 155 pollutants were lower.

During the campaign, the average wind speed was low, showing average values of 2.3
± 1.4 m/s and 2.4 ± 1.5 m/s at CP and PKUERS, respectively. The diurnal variations
of wind directions and speeds are exhibited in Fig. S2. The prevailing wind was from
south, with higher wind speed during the daytime.

160 To explore the influence of the air masses from different directions on fine particle161 loading and sources, back trajectory analysis was performed using National Oceanic





162 and Atmospheric Administration (NOAA) Hybrid Single Particle Lagrangian Integrated Trajectory (HYSPLIT) model. We calculated 36 h air mass back 163 164 trajectories arriving at two sampling site during the observation period using the HYSPIT-4 model with a $1^{\circ}\times1^{\circ}$ latitude-longitude grid and the final meteorological 165 database. The model was run with the starting time of 0:00, 4:00, 8:00, 12:00, 16:00, 166 167 and 20:00 UTC). The arrival level was set at 200 m above ground level. The method 168 used in trajectory clustering was based on GIS-based software TrajStat 169 (http://www.meteothinker.com/TrajStatProduct.aspx). 36-h back trajectories staring at 170 200 m above ground level in CP and PKUERS were calculated every 4 hours during 171 the entire campaign and then clustered according to their similarity in spatial 172 distribution using the HYSPLIT4 software. Three-cluster solution was adopted as 173 shown in Fig. S3. The three clusters were defined as Far North West (Cluster 1, Far 174 NW), Near West North (Cluster2, Near WN), and South (Cluster 3). South cluster was 175 found to be the most frequent one, accounting for 52% at CP and 64% at PKUERS. 176 Clusters Far NW and Near NW accounted for 17% and 31%, 17% and 19% at CP and 177 PKUERS, respectively.

178 **3.2 Overview of PM_{2.5} chemical composition**

179 In this study, daily $PM_{2.5}$ concentrations fluctuated dramatically from 6.7 µg m⁻³ to 80.3 μ g m⁻³ at CP, and from 9.6 to 82.5 μ g m⁻³ at PKUERS, respectively. A paired 180 181 t-test was used to compare the mass concentrations at two sites. The results indicate 182 that the mass concentrations showed statistically non-significant difference, 183 suggesting the regional particle pollution in Beijing. PM_{25} mass concentrations during 184 the summer of 2008 to 2016 in Beijing are summarized in Table 1. Guo et al. (Guo et 185 al., 2013) reported the average PM2.5 concentrations during the summers of 2000 to 186 2008, which showed distinct decreasing tendency during 2000-2006 and then slightly 187 increased in 2007 due to unfavorable meteorological conditions. To better understand 188 the variation tendency of the PM_{2.5} in the summer of Beijing, we compared the fine 189 particle matter data since 2008. Compared with 2008, the PM2.5 concentrations





decreased from 92.3 \pm 44.7 µg m⁻³ to 88.2 µg m⁻³ in 2009 and 62.7 µg m⁻³ in 2010. 190 The mass concentration continued falling to 45.5 μ g m⁻³ in 2016. This decreasing is 191 192 attributed to the drastic emission control measures implemented by the Beijing 193 government since 2012. In spite of the prominent decrease of the PM2.5 mass 194 concentrations, the aerosol loading in Beijing was still much higher than that in 195 developed countries (Tai et al., 2010; Barmpadimos et al., 2012; Park and Cho, 2011). 196 Fig. S4 showed the chemical composition of PM_{2.5}. In general, organic particulate 197 matters (OM, OC*1.6) and sulfate were the two dominant components, accounting for 198 more than 50% of the PM2.5 mass concentration during the field campaign. The

average concentration of total WSICs for CP was $17.4 \pm 11.5 \ \mu g \ m^{-3}$, higher than that of PKUERS ($12.2 \pm 8.5 \ \mu g \ m^{-3}$). Among the WSICs, secondary inorganic ions (sulfate, nitrate, and ammonium) were the most abundant compounds, indicating secondary particles played great roles in the summer of Beijing. The higher sulfate proportion could be explained by the increased photochemical conversion of sulfur dioxide to sulfate aerosol (Xiang et al., 2017).

205 Carbonaceous aerosols, i.e. organic carbon (OC) and elemental carbon (EC) were also 206 great contributors to PM2.5 concentrations. Higher proportion of OC and EC at 207 PKUERS demonstrated severe carbonaceous pollution in urban Beijing, which might 208 have close correlation with the higher traffic flow, coal/wood combustion by residents 209 and industrial emissions (Wang et al., 2006; Dan et al., 2004; Cao et al., 2004). 210 Comparison of the OC, EC concentrations from 2008 to 2016 were also listed in Table 1. Unlike PM_{2.5}, OC concentration at PKUERS showed a higher OC concentration 211 $(11.0 \pm 3.7 \ \mu g \ m^{-3})$ compared with that in 2008 (9.2 $\pm 3.3 \ \mu g \ m^{-3})$, suggesting organic 212 213 aerosol pollution becomes more and more important. EC concentration decreased dramatically to 0.7 \pm 0.5 µg m⁻³ at CP and 1.8 \pm 1.0 µg m⁻³ at PKUERS, which 214 showed the lowest value since 2000. This could be attributed to the implementation of 215 216 air pollution prevention and control action plan enacted by the state council since 217 2013.





To evaluate the influences of the air masses from different directions on the $PM_{2.5}$ loadings during the campaign, three categories were divided according to the back trajectory clustering analysis (See Fig. S5). In general, cluster south represented the most polluted air mass origin followed by clusters Near WN and Far NW, which was in accordance with previous studies demonstrating severe aerosol pollution in southerly air flow in summer of Beijing (Huang et al., 2010; Sun et al., 2010).

224 **3.3** Concentration of particulate organic species from different air mass origins

225 The organic species (except secondary organic tracers) were divided into 12 226 categories. Their concentrations in different directions according to the back trajectory 227 clustering were shown in Fig. S6. Detailed information for each class at the two sites 228 could be found in the supplementary material (Fig. S7). Cluster south showed higher 229 particulate organic matter concentration, followed by cluster near WN and far NW, 230 indicating more severe aerosol pollution from the south. Our result consists with the 231 previous studies that more pollution emissions are from the south area of Beijing than 232 those from the north (Wu et al., 2011; Zhang et al., 2009).

233 Dicarboxylic acid was the most abundant species among all the components, 234 demonstrating the great contribution of the secondary formation to the organic 235 aerosols in the summer of Beijing (Guo et al., 2010). A series of n-alkanes ranging 236 from C12 to C36 were analyzed. Their distribution during the observation period was 237 shown in Fig. S7 (a). The maximum-alkane concentration species (C_{max}) were C27 and C29. The odd carbon preference was an indicative of biogenic sources (vegetative 238 239 matters and biomass burning) (Huang et al., 2006; Rogge et al., 1993). In this study, 240 total PAHs were much lower than previous studies in summer of Beijing, suggesting 241 the effectiveness of the control strategies since 2013 (Wang et al., 2009). According to 242 Fig. S7 (c), five ring PAHs were dominant species among all the species, followed by 243 four-ring and six-ring PAHs. In total, four to six ring PAHs had higher abundancy, 244 accounting for more than 60% of the total PAHs. The result was much similar with





245 previous studies that the distribution of PAHs was impacted by the volatility of PAHs 246 and the temperature (Wang et al., 2009; Guo et al., 2013). Saccharide was considered 247 to originate from biomass burning (Simoneit et al., 1999). In this study, we quantified three sugar compounds including levoglucosan, manosan and galactosan, in which 248 249 levoglucosan was considered as a good tracer for biomass burning. The average daily mass concentration of levoglucosan at CP and PKUERS were 53.03 \pm 39.26 ng m⁻³ 250 and 59.87 \pm 38.93 ng m⁻³, respectively. It's worth mentioning that the levoglucosan 251 252 concentration was the lowest in recent years (Cheng et al., 2013; Guo et al., 2013). 253 Hopanes have been considered as markers for oil combustion (Lambe et al., 2009), 254 vehicles (i.e. gasoline-powered and diesel-powered engine) (Cass, 1998; Lough et al., 255 2007) and coal combustion(Oros and Simoneit, 2000). Nevertheless, contributions of 256 coal combustion to hopanes were much less than that of vehicle exhaustion. Concentrations of quantified hopanes including 17a(H)-22,29,30-trishopane, 257 258 $17\beta(H)-21\alpha(H)$ -norhopane, and $17\alpha(H)-21\beta(H)$ -hopane of CP and PKUERS are 259 shown in Fig. S7(d). The total average concentrations of hopanes were 3.05 ± 1.53 ng m^{-3} for CP and 3.90 ± 2.06 ng m^{-3} for PKUERS. The hopanes concentrations at urban 260 261 site PKUERS were much higher than that of CP, which could probably explained by 262 the heavier vehicle emissions in the urban area. The concentrations of primary organic 263 tracers used in CMB model were listed in Table S2.

264 **3.4 Biogenic and anthropogenic SOA tracers**

Table S3 compared the SOA tracers measured in this work with those in other regions in the world as well as that observed in Beijing 2008. The sites for comparison include an urban background site at Indian Institute of Technology Bombay, Mumbai, India (IITB) (Fu et al., 2016), an outflow region of Asian aerosols and precursors Cape Hedo, Okinawa, Japan (CH) (Zhu et al., 2016), a residential site at Yuen Long, Hong Kong (YL) (Hu et al., 2008), three industrial sites at Cleveland Ohio (CL, data was averaged among the three sites), a suburban site in the Research Triangle Park





- 272 North California (RTP). The detailed information about these sites were summarized
- in the supplementary material.
- Three isoprene-SOA tracers i.e. two 2-methyltetrols (2-methyltheitol and 2-methylerythritol) and 2-methylglyceric acid were detected. The summed concentration of the isoprene-SOA tracers ranged from 3.7 to 62.3 ng m⁻³ at CP and 8.6 to 46.5 ng m⁻³ at PKUERS. The concentration was higher than that of IITB and CH. Compared with the isoprene-SOA tracers in 2008, the concentrations in 2016 were lower.

Nine α -pinene tracers were identified. The sum of the tracers ranged from 20.9 to 281 282.3 ng m⁻³ at CP and 50.0 to 161.4 ng m⁻³ at PKUERS, which had similar 282 distribution pattern with that measured in 2008 Beijing and YL. The total α -pinene 283 tracer concentrations were lower than those in 2008, while still much higher than the 284 concentrations in other regions of the world.

 β -caryophyllinic acid is one of the oxidation products of β-caryophyllene which is considered as a tracer for β-caryophyllene SOA. In this study, β-caryophyllinic acid concentrations ranged from 1.4 to 16.7 ng m⁻³ at CP, and 0.9 to 12.0 ng m⁻³ at PKUERS, with average daily average concentrations of 6.1 ± 3.5 ng m⁻³ and 6.0 ± 2.8 ng m⁻³ for CP and PKUERS, respectively. The values were lower than those at YL and RPT, higher than that measured at Yufa and PKUERS in 2008.

291 2,3-Dihydroxy-4-oxopentanoic acid is deemed as a tracer for toluene SOA. Our 292 results showed that the 2,3-Dihydroxy-4-oxopentanoic acid concentration was $9.7 \pm$ 293 7.3 ng m⁻³ at CP and 11.0 \pm 3.7 ng m⁻³ at PKUERS. Compared with other regions of 294 the world, the concentrations of 2,3-Dihydroxy-4-oxopentanoic acid was much higher, 295 implying higher contributions of anthropogenic sources at Beijing. However, the 296 concentrations in CP were lower than that of PKUERS.





297 4. Primary sources and secondary formation of organic aerosols

298 4.1 Contributions of primary and secondary organic aerosols

299 A CMB model and the tracer-yield method were used to quantify the contributions of 300 primary and secondary sources to the ambient organic carbon (See Fig. 1). On 301 average, the primary sources accounted for $42.6 \pm 15.4\%$ and $50.4 \pm 19.1\%$ of the 302 measured OC at CP and PKUERS, respectively. The vehicle emissions were the 303 dominant primary sources, with the contributions of 28.8 \pm 14.8% and 37.6 \pm 19.3% 304 at PKUERS and CP, respectively, implying the urgency to control vehicular 305 exhaustion in urban areas. Despite of the lower contribution of the gasoline exhaust at 306 PKUERS, the mass concentration of the gasoline exhaust was higher compared with 307 that of CP given the higher OC loading at PKUERS. The contributions of biomass 308 burning were $3.9 \pm 2.6\%$ and $5.0 \pm 2.2\%$ at CP and PKUERS, respectively, showing 309 the higher concentrations at night. The drastic change of the biomass burning 310 contribution in CP at night was due to occasional burning activities at night. Coal 311 combustion contributed $5.8 \pm 5.5\%$ and $4.6 \pm 2.6\%$ of the measured OC at CP and 312 PKUERS. The higher contribution at CP was due to more burning activities in the 313 suburban areas.

314 The secondary organic sources accounted for $20.2 \pm 6.7\%$ of the organic carbon at CP, 315 with 1.6 \pm 0.4% from isoprene, 4.4 \pm 1.5% from α -pinene, 2.7 \pm 1.0% from 316 β -caryophyllene and 12.5 ± 3.4% from toluene. As for PKUERS, the secondary 317 organic sources took up $30.5 \pm 12.0\%$ of the measured OC, in which isoprene was 318 responsible for 2.3 \pm 0.9%, α -pinene for 5.6 \pm 1.9%, β -caryophyllene for 3.6 \pm 2.6% 319 and toluene for $19.0 \pm 8.2\%$. Haque et al. (Haque et al., 2016) used tracer-based 320 method to apportion the organic carbon and results showed that the biogenic SOC was 321 responsible for 21.3% of the total OC with isoprene SOC contributing 17.4%, 322 α/β -pinene SOC contributing 2.5% and β -caryophyllene SOC contributing 1.4% in the 323 summer of Alaska, implying the significant contributions of the biogenic SOA to the





324 loading of the organic aerosol. Our results exhibited that the biogenic SOA 325 concentration was comparable or even high than that at some forest sites in other 326 places of the world (Miyazaki et al., 2012; Stone et al., 2012; Claeys et al., 2004; 327 Kourtchev et al., 2008). The SOA formation mechanism is complicated. A possible 328 reason is the high oxidation capacity in China. More work is still needed to 329 investigate the SOA formation mechanism under Air Pollution Complex in China.

330 Stone et al. (Stone et al., 2009) discovered that primary and secondary sources 331 accounted for $83 \pm 8\%$ of the measured organic carbon, with primary sources 332 accounted for $37 \pm 2\%$ and SOC contributed for $46 \pm 6\%$ with $16 \pm 2\%$ from biogenic 333 gas-phase precursors and $30 \pm 4\%$ from toluene using CMB model and tracer-based 334 method at Cleveland with heavy industries, implying that anthropogenic sources 335 played great roles in the formation of SOA. Our results showed a similar with the 336 results published by Stone et al., where anthropogenic sources i.e. toluene derived 337 SOC dominated the apportioned SOC. Our research revealed an important point that 338 controlling SOA seems feasible in the developing countries like China. It is difficult 339 to control SOA in developed countries, since biogenic SOA are dominant. However, 340 deducting anthropogenic precursors may be an efficient way to reduce the SOA 341 pollution where anthropogenic SOA is dominant. On average, $62.8 \pm 18.3\%$ and 80.9 \pm 27.2% of the measured OC were apportioned at CP and PKUERS, respectively. 342 343 About $36.3 \pm 18.1\%$ and $29.3 \pm 15.6\%$ of the OC sources remained unknown, which 344 were probably composed of uncharacterized primary or secondary sources. Further 345 research is needed to explain the unapportioned sources of OC.

Due to the drastic emission control measures taken by the Beijing government, the primary and secondary sources in Beijing may change greatly. Fig. 2 displayed the comparison of the sources between 2008 and 2016 at the same site PKUERS. We compared the average contributions by percentage rather than the mass concentration. In general, primary sources contributed $50.4 \pm 19.1\%$ of the measured OC in 2016, closely correlated to the increasing contribution of vehicular emissions. Gasoline





engines accounted for 18% of the measured OC, showing an enhancement of 80% 352 353 with respect to 2008. This might be related to the rising number of the vehicles in 354 Beijing. In comparison, diesel exhaust decreased by 12.5% due to the strict control 355 measures made by the government. A 28.5% and 20% reduction of coal combustion 356 and biomass burning could also be found due to the drastic measures made by the government. Compared with 2008, contributions of secondary organic aerosol 357 358 decreased by 29.4%. However, the contribution of toluene SOC was the highest 359 among the apportioned SOC, which was different from the results of the most developed countries in the world. In summary, the contributions of most POA 360 decreased in recent years, except for gasoline exhaust, indicating more efforts should 361 362 be made to control the gasoline emission. The apportioned SOC was also decreased 363 with toluene SOC served as the dominant source. Our results revealed that deducting 364 anthropogenic precursors may be an efficient way to control SOA pollution in China.

365 **4.2 Organic aerosol sources from different air mass origins**

366 The regional sources and transport of air pollutants exert profound impacts on air 367 quality of Beijing. To better understand the regional impacts on the primary and secondary aerosol sources of Beijing, source apportionment results when air mass 368 369 from different origins were shown in Fig. 3. Vehicular emissions i.e. gasoline and 370 diesel exhaust showed identical contributions from different air mass origins (31.0% 371 from south vs 31.3% from Near WN vs 31.7% from Far NW) at PKUERS, demonstrating the vehicular pollution could mostly be attributed to the vehicular 372 373 emission at the local site. However, the contribution of vehicular emission at CP 374 showed significant difference from different air mass origins, with lowest contribution 375 when air mass was from far northwest. This might be explained by regional transport 376 from different directions. Comparable contributions of coal combustion and biomass 377 burning were found at CP and PKUERS from different air mass origins, implying the 378 regional pollution in Beijing. Similarly, biogenic SOC showed similar contributions 379 from different air mass origins both at the regional site and the urban site. From all the





- 380 directions, the toluene SOC (anthropogenic source) was the largest contributor to
- 381 apportioned SOC, with higher concentrations at the urban site PKUERS. On the
- 382 whole, most of the sources showed comparable contribution from different air mass
- 383 origins, implying the pollution in Beijing was regional.

384 **4.3 Influencing factors for secondary organic aerosol formation in the summer of**

385 Beijing

Laboratory experiments have revealed that several factors can influence the SOA formation, e.g. oxidants (OH radical, ozone etc.), temperature, humidity, particle water content and acidity. In this work, the relationships between estimated SOA and these factors were investigated to better understand the SOA formation in Beijing.

SOA formation from ozonolysis

391 Ozone is considered as an important oxidant for SOA formation. Fig. 4 (a)(b) showed the correlation with ozone mixing ratio and SOC. It is clear that SOC increased 392 393 significantly with the increasing of ozone mixing ratio, which is consistent with 394 previous studies in Beijing (Guo et al. 2012). Different correlations were found 395 between day and night samples, with better correlation for the daytime samples at 396 both sites, implying SOA may have other formation mechanism at night other than 397 ozonolysis. At CP, the growth rate of SOC with O₃ was similar for day and night samples, which was 0.02 µg m⁻³ per ppbv ozone. For PKUERS, the increment rate of 398 the SOC towards ozone was 0.04 µg m⁻³ and 0.02 µg m⁻³ per ppbv ozone at day and 399 400 night, respectively.

401 Influence of temperature and relative humidity on SOA formation

Temperature was considered as a great influencing factor on SOA formation. On the one hand, higher temperature promoted the evaporation of the semi volatile SOA. On the other hand, high-temperature conditions would favor the oxidation, which would accelerate the SOA formation (Saathoff et al., 2009). Fig. 4 (c) (d) showed the





406 variation of SOC concentrations with the temperature. In this study, SOC
407 concentration showed positive correlation with temperature at CP and PKUERS,
408 demonstrating that temperature favors the SOA formation in the summer of Beijing.
409 Moreover, different correlation of the day and the night samples imply the different
410 pathways of SOA formation. However, poor relations could be found between SOC
411 and RH.

412 Effects of aqueous-phase acid catalyzed reactions on SOA formation

413 Aerosol water and acidity have been considered to have great impact on the 414 aqueous-phase SOA formation (Cheng et al., 2016). To figure out the influences of 415 water content and aerosol acidity on the aqueous-phase reactions, ISORROPIA-II 416 thermodynamic equilibrium model was used (Surratt et al., 2007). The model was set 417 at forward mode, based on the concentrations of particle phase Na⁺, NH₄⁺, K⁺, Mg²⁺, 418 Ca²⁺, NO₃⁻, SO₄²⁻, Cl⁻ and gaseous NH₃ as well as ambient temperature and RH.

Results showed that the average aerosol water content at CP was $3.87 \pm 3.73 \ \mu g \ m^{-3}$, 419 higher than that at PKUERS (1.83 \pm 1.81 µg m⁻³). The water content was lower in 420 421 2016 than that in 2008. The estimated SOC concentration showed good correlations 422 with water content at both sites. Compared with CP, the correlation factor in PKUERS 423 was better, implying that aqueous phase reaction may be more important in the urban 424 area. Different correlation could be found at different liquid water contents, especially 425 for CP, where liquid water contented spanned a wide range, implying that different mechanisms may exist at different liquid water content. 426

In this study, modeled H^+ concentration and SOC showed significant correlation (p<0.05) at the two places, which indicated that acid-catalyzed reaction might provide a crucial pathway for the SOA formation in the summer of Beijing. Laboratory studies showed that acid-catalyzed reactive uptake might play great role in the enhancement of SOA (Zhang et al., 2014; Surratt et al., 2010; Jang et al., 2002). However, contrary conclusions were made by other group, demonstrating the inconsistence of the aerosol





433 acidity and the SOA formation (Wong et al., 2015; Kristensen et al., 2014). The 434 contradiction might give the facts that the impacts of the acidity on the SOA loading 435 varied from place to place, determined by the specific environmental conditions. Linear regression showed that the enhancement of SOC with modeled H^+ 436 concentration were at a value of 0.02 μ g m⁻³ per nmol H⁺, which was lower than the 437 previous results (0.046 for PKUERS, and 0.041 for Yufa, 2008). Offenberg et al. 438 439 (Offenberg et al., 2009) discovered good correlation between SOC and $[H^+]_{air}$, with R^2 value of 0.815. Besides, a one nmol m⁻³ [H⁺]_{air} would give rise to 0.015 µg m⁻³ 440 441 SOC increase from the oxidation of a-pinene in the chamber experiment. In the 442 present work, different correlations could be found at different modeled H⁺ 443 concentrations where apportioned SOC increased significantly as the H⁺ 444 concentration increased and then increased slowly at a certain level, showing gradient 445 growth at different hydrogen-ion concentrations. Therefore, aqueous phase 446 acid-catalyzed reactions may influence the SOA formation through different 447 mechanisms at varied level of liquid water concentration and aerosol acidity.

448 **5.** Conclusion

449 High concentrations of fine particles were observed during the "Campaign on 450 Photochemical Smog in China", with the average mass concentrations of 45.48 \pm 19.78 μ g m⁻³ and 42.99 \pm 17.50 μ g m⁻³, at CP site and PKUERS site, respectively. 451 452 Compared with previous studies, the concentrations of PM_{2.5}, EC and estimated SOC 453 decreased significantly, due to the drastic measures implemented by the government 454 in the recent years. However, OC showed a higher concentration, suggesting 455 particulate organic matters become more and more important in Beijing. CMB 456 modeling and tracer-yield method were used to apportion the primary and secondary 457 organic aerosol sources. The apportioned primary and secondary OC accounted for $62.8 \pm 18.3\%$ and $80.9 \pm 27.2\%$ of the measured OC at CP and PKUERS, respectively. 458 459 Vehicle emissions i.e. diesel and gasoline engine emissions were the major primary 460 organic aerosol sources, which contributed to $28.8 \pm 14.8\%$ and $37.6 \pm 19.3\%$ of the





461 OC at CP and PKUERS, respectively. Compared with the results of the previous work, 462 the gasoline engine emission contributed almost twice of that in 2008 (18.0% vs 463 10.0%), while the contribution of diesel engine emission decreased by 12.5% compared with the result in 2008. Besides, the contributions of biomass burning and 464 465 coal combustion both decreased. The apportioned biogenic and anthropogenic SOC can explain 20.2 \pm 6.7% and 30.5 \pm 12.0% of the measured OC at CP and PKUERS, 466 467 respectively. The contribution of toluene SOC is the highest among the apportioned 468 SOC, which is different from the results of the most developed countries in the world. 469 Our results revealed an important point, which is that controlling SOA seems feasible 470 in the developing countries like China. It is difficult to control SOA in developed 471 countries, since biogenic SOA are dominant. However, deducting anthropogenic 472 precursors may be an efficient way to reduce the SOA pollution where anthropogenic SOA is dominant. Back trajectory clustering analysis showed that the particle source 473 474 contributions were similar when air masses were from different directions, suggesting 475 the regional organic particle pollution in Beijing. However, the higher organic particle loading from south cluster indicates that there were more emissions from southern 476 477 region of Beijing. The present work also implied that the aqueous phase 478 acid-catalyzed reactions may be an important SOA formation mechanism in summer 479 of Beijing.

19





480 Acknowledgement

- 481 This research is supported by the National Key R&D Program of China: Task
- 482 3 (2016YFC0202000), the National Natural Science Foundation of China (21677002),
- 483 framework research program on 'Photochemical smog in China" financed by Swedish
- 484 Research Council (639-2013-6917).





485	References
486	Barmpadimos, I., Keller, J., Oderbolz, D., Hueglin, C., and Pr év ôt, A.: One decade of
487	parallel fine ($PM_{2.5}$) and coarse (PM_{10} - $PM_{2.5}$) particulate matter measurements in
488	Europe: trends and variability, Atmos Chem Phys, 12, 3189-3203, 2012.
489	Cao, J. J., Lee, S. C., Ho, K. F., Zou, S. C., Fung, K., Li, Y., Watson, J. G., and Chow,
490	J. C.: Spatial and seasonal variations of atmospheric organic carbon and
491	elemental carbon in Pearl River Delta Region, China, Atmospheric Environment,
492	38, 4447-4456, http://doi.org/10.1016/j.atmosenv.2004.05.016, 2004.
493	Cass, G. R.: Organic molecular tracers for particulate air pollution sources,
494	Trac-Trends in Analytical Chemistry, 17, 356-366,
495	10.1016/s0165-9936(98)00040-5, 1998.
496	Cheng, Y., Engling, G., He, K. B., and Duan, F. K.: Biomass burning contribution to
497	Beijing aerosol, Atmospheric Chemistry & Physics, 13, 7765-7781, 2013.
498	Cheng, Y., Zheng, G., Wei, C., Mu, Q., Zheng, B., Wang, Z., Gao, M., Zhang, Q., He,
499	K., and Carmichael, G.: Reactive nitrogen chemistry in aerosol water as a source
500	of sulfate during haze events in China, Science Advances, 2, e1601530, 2016.
501	Claeys, M., Graham, B., Vas, G., Wang, W., Vermeylen, R., Pashynska, V., Cafmeyer,
502	J., Guyon, P., Andreae, M. O., and Artaxo, P.: Formation of Secondary Organic
503	Aerosols Through Photooxidation of Isoprene, Science, 303, 1173, 2004.
504	Dan, M., Zhuang, G., Li, X., Tao, H., and Zhuang, Y.: The characteristics of
505	carbonaceous species and their sources in PM2.5 in Beijing, Atmospheric
506	Environment, 38, 3443-3452, http://doi.org/10.1016/j.atmosenv.2004.02.052,
507	2004.
508	Ding, X., He, Q. F., Shen, R. Q., Yu, Q. Q., and Wang, X. M.: Spatial distributions of
509	secondary organic aerosols from isoprene, monoterpenes, beta-caryophyllene,
510	and aromatics over China during summer, Journal of Geophysical
511	Research-Atmospheres, 119, 11877-11891, 10.1002/2014jd021748, 2014.

21





- 512 El Haddad, I., Marchand, N., Temime-Roussel, B., Wortham, H., Piot, C., Besombes,
- 513 J. L., Baduel, C., Voisin, D., Armengaud, A., and Jaffrezo, J. L.: Insights into the
- 514 secondary fraction of the organic aerosol in a Mediterranean urban area:
- 515 Marseille, Atmos Chem Phys, 11, 2059-2079, 2011.
- 516 Fu, P., Aggarwal, S. G., Chen, J., Li, J., Sun, Y., Wang, Z., Chen, H., Liao, H., Ding,
- 517 A., Umarji, G. S., Patil, R. S., Chen, Q., and Kawamura, K.: Molecular Markers
- of Secondary Organic Aerosol in Mumbai, India, Environ. Sci. Technol., 50,
 4659-4667, 10.1021/acs.est.6b00372, 2016.
- 520 Gao, J. J., Tian, H. Z., Cheng, K., Lu, L., Wang, Y. X., Wu, Y., Zhu, C. Y., Liu, K. Y.,
- Zhou, J. R., Liu, X. G., Chen, J., and Hao, J. M.: Seasonal and spatial variation
 of trace elements in multi-size airborne particulate matters of Beijing, China:
 Mass concentration, enrichment characteristics, source apportionment, chemical
 speciation and bioavailability, Atmospheric Environment, 99, 257-265,
 10.1016/j.atmosenv.2014.08.081, 2014.
- Guo, S., Hu, M., Wang, Z. B., Slanina, J., and Zhao, Y. L.: Size-resolved aerosol
 water-soluble ionic compositions in the summer of Beijing: implication of
 regional secondary formation, Atmos Chem Phys, 10, 947-959, 2010.
- 529 Guo, S., Hu, M., Guo, O., Zhang, X., Zheng, M., Zheng, J., Chang, C. C., Schauer, J.
- J., and Zhang, R.: Primary Sources and Secondary Formation of Organic
 Aerosols in Beijing, China, Environmental Science & Technology, 46,
 9846-9853, 10.1021/es20425641, 2012.
- Guo, S., Hu, M., Guo, Q., Zhang, X., Schauer, J., and Zhang, R.: Quantitative
 evaluation of emission controls on primary and secondary organic aerosol
 sources during Beijing 2008 Olympics, Atmos Chem Phys, 13, 8303-8314, 2013.
- 536 Guo, S., Hu, M., Zamora, M. L., Peng, J., Shang, D., Zheng, J., Du, Z., Wu, Z., Shao,
- 537 M., and Zeng, L.: Elucidating severe urban haze formation in China, Proceedings
- 538 of the National Academy of Sciences, 111, 17373-17378, 2014.





- 539 Haque, M. M., Kawamura, K., and Kim, Y.: Seasonal variations of biogenic
- 540 secondary organic aerosol tracers in ambient aerosols from Alaska, Atmospheric
- 541 Environment, 130, 95-104, 10.1016/j.atmosenv.2015.09.075, 2016.
- 542 Hu, D., Bian, Q., Li, T. W., Lau, A. K., and Yu, J. Z.: Contributions of isoprene,
- 543 monoterpenes, β caryophyllene, and toluene to secondary organic aerosols in
 544 Hong Kong during the summer of 2006, Journal of Geophysical Research:
 545 Atmospheres, 113, 2008.
- Hu, M., Guo, S., Peng, J.-f., and Wu, Z.-j.: Insight into characteristics and sources of
 PM2. 5 in the Beijing–Tianjin–Hebei region, China, National Science Review, 2,
 257-258, 2015.
- Huang, X.-F., He, L.-Y., Hu, M., and Zhang, Y.-H.: Annual variation of particulate
 organic compounds in PM 2.5 in the urban atmosphere of Beijing, Atmospheric
 Environment, 40, 2449-2458, 2006.
- Huang, X. F., He, L. Y., Hu, M., Canagaratna, M. R., Sun, Y., Zhang, Q., Zhu, T., Xue,
 L., Zeng, L. W., and Liu, X. G.: Highly time-resolved chemical characterization
 of atmospheric submicron particles during 2008 Beijing Olympic Games using
 an Aerodyne High-Resolution Aerosol Mass Spectrometer, Atmospheric
 Chemistry & Physics Discussions, 10, 8933-8945, 2010.
- Jang, M. S., Czoschke, N. M., Lee, S., and Kamens, R. M.: Heterogeneous
 atmospheric aerosol production by acid-catalyzed particle-phase reactions,
 Science, 298, 814-817, 10.1126/science.1075798, 2002.
- 560 Kleindienst, T. E., Jaoui, M., Lewandowski, M., Offenberg, J. H., Lewis, C. W.,
- 561 Bhave, P. V., and Edney, E. O.: Estimates of the contributions of biogenic and
- 562 anthropogenic hydrocarbons to secondary organic aerosol at a southeastern US
- location, Atmospheric Environment, 41, 8288-8300, 2007.
- Kourtchev, I., Warnke, J., Maenhaut, W., Hoffmann, T., and Claeys, M.: Polar organic
 marker compounds in PM2.5 aerosol from a mixed forest site in western
- 566 Germany, Chemosphere, 73, 1308-1314, 2008.





567	Kristensen, K., Cui, T., Zhang, H., Gold, A., Glasius, M., and Surratt, J.: Dimers in
568	α -pinene secondary organic aerosol: effect of hydroxyl radical, ozone, relative
569	humidity and aerosol acidity, Atmos Chem Phys, 14, 4201-4218, 2014.
570	Lambe, A. T., Miracolo, M. A., Hennigan, C. J., Robinson, A. L., and Donahue, N. M.:
571	Effective Rate Constants and Uptake Coefficients for the Reactions of Organic
572	Molecular Markers (n-Alkanes, Hopanes, and Steranes) in Motor Oil and Diesel
573	Primary Organic Aerosols with Hydroxyl Radicals, Environ. Sci. Technol., 43,
574	8794-8800, 10.1021/es901745h, 2009.
575	Liu, Q. Y., Baumgartner, J., Zhang, Y., and Schauer, J. J.: Source apportionment of
576	Beijing air pollution during a severe winter haze event and associated
577	pro-inflammatory responses in lung epithelial cells, Atmospheric Environment,
578	126, 28-35, 10.1016/j.atmosenv.2015.11.031, 2016.
579	Lough, G. C., Christensen, C. G., Schauer, J. J., Tortorelli, J., Mani, E., Lawson, D. R.,
580	Clark, N. N., and Gabele, P. A.: Development of molecular marker source
581	profiles for emissions from on-road gasoline and diesel vehicle fleets, J. Air
582	Waste Manage. Assoc., 57, 1190-1199, 10.3155/1047-3289.57.10.1190, 2007.
583	Miyazaki, Y., Jung, J., Fu, P., Mizoguchi, Y., Yamanoi, K., and Kawamura, K.:
584	Evidence of formation of submicrometer water - soluble organic aerosols at a
585	deciduous forest site in northern Japan in summer, Journal of Geophysical
586	Research: Atmospheres, 117, 2012.
587	Offenberg, J. H., Lewandowski, M., Edney, E. O., Kleindienst, T. E., and Jaoui, M.:
588	Influence of Aerosol Acidity on the Formation of Secondary Organic Aerosol
589	from Biogenic Precursor Hydrocarbons, Environ. Sci. Technol., 43, 7742-7747,
590	10.1021/es901538e, 2009.
591	Oros, D. R., and Simoneit, B. R. T.: Identification and emission rates of molecular
592	tracers in coal smoke particulate matter, Fuel, 79, 515-536, Doi
593	10.1016/S0016-2361(99)00153-2, 2000.

24





- 594 Park, S. S., and Cho, S. Y.: Tracking sources and behaviors of water-soluble organic
- 595 carbon in fine particulate matter measured at an urban site in Korea, Atmospheric
- 596 environment, 45, 60-72, 2011.
- 597 Rogge, W. F., Hildemann, L. M., Mazurek, M. A., Cass, G. R., and Simoneit, B. R. T.:
- 598 Sources of fine organic aerosol. 4. Particulate abrasion products from leaf 599 surfaces of urban plants, Environ. Sci. Technol., 27, 2700-2711, 1993.
- 600 Saathoff, H., Naumann, K.-H., Möhler, O., Jonsson, Å. M., Hallquist, M.,
- Kiendler-Scharr, A., Mentel, T. F., Tillmann, R., and Schurath, U.: Temperature
 dependence of yields of secondary organic aerosols from the ozonolysis of
 α-pinene and limonene, Atmos Chem Phys, 9, 1551-1577, 2009.
- 604 Schauer, J. J., Rogge, W. F., Hildemann, L. M., Mazurek, M. A., Cass, G. R., and
- Simoneit, B. R. T.: Source apportionment of airborne particulate matter using
 organic compounds as tracers, Atmospheric Environment, 30, 3837-3855,
 10.1016/1352-2310(96)00085-4, 1996.
- Sheesley, R. J., Schauer, J. J., Zheng, M., and Wang, B.: Sensitivity of molecular
 marker-based CMB models to biomass burning source profiles, Atmospheric
 Environment, 41, 9050-9063, 2007.
- 611 Simoneit, B. R., Schauer, J. J., Nolte, C., Oros, D. R., Elias, V. O., Fraser, M., Rogge,
- W., and Cass, G. R.: Levoglucosan, a tracer for cellulose in biomass burning and
 atmospheric particles, Atmospheric Environment, 33, 173-182, 1999.
- 614 Song, G., Min, H., Qingfeng, G., and Dongjie, S.: Comparison of secondary organic
- aerosol estimation methods, ACTA CHIMICA SINICA, 72, 658-666, 2014.
- 616 Song, Y., Zhang, Y., Xie, S., Zeng, L., Zheng, M., Salmon, L. G., Shao, M., and
- 617 Slanina, S.: Source apportionment of PM2. 5 in Beijing by positive matrix
- 618 factorization, Atmospheric Environment, 40, 1526-1537, 2006.
- 619 Stone, E. A., Nguyen, T. T., Pradhan, B. B., and Dangol, P. M.: Assessment of
- 620 biogenic secondary organic aerosol in the Himalayas, Environmental Chemistry,
- 621 9, 263-272, 2012.





- 622 Sun, J., Zhang, Q., Canagaratna, M. R., Zhang, Y., Ng, N. L., Sun, Y., Jayne, J. T.,
- 623 Zhang, X., Zhang, X., and Worsnop, D. R.: Highly time- and size-resolved
- 624 characterization of submicron aerosol particles in Beijing using an Aerodyne
- 625 Aerosol Mass Spectrometer, Atmospheric Environment, 44, 131-140, 2010.
- 626 Surratt, J. D., Kroll, J. H., Kleindienst, T. E., Edney, E. O., Claeys, M., Sorooshian, A.,
- 627 Ng, N. L., Offenberg, J. H., Lewandowski, M., Jaoui, M., Flagan, R. C., and
- 628 Seinfeld, J. H.: Evidence for organosulfates in secondary organic aerosol,
- 629 Environ. Sci. Technol., 41, 517-527, 10.1021/es062081q, 2007.
- 630 Surratt, J. D., Chan, A. W., Eddingsaas, N. C., Chan, M., Loza, C. L., Kwan, A. J.,
- 631 Hersey, S. P., Flagan, R. C., Wennberg, P. O., and Seinfeld, J. H.: Reactive
- 632 intermediates revealed in secondary organic aerosol formation from isoprene,
- 633 Proceedings of the National Academy of Sciences, 107, 6640-6645, 2010.
- Tai, A. P., Mickley, L. J., and Jacob, D. J.: Correlations between fine particulate matter
 (PM 2.5) and meteorological variables in the United States: Implications for the
 sensitivity of PM 2.5 to climate change, Atmospheric Environment, 44,
 3976-3984, 2010.
- Tan, J. H., Duan, J. C., Chai, F. H., He, K. B., and Hao, J. M.: Source apportionment
 of size segregated fine/ultrafine particle by PMF in Beijing, Atmospheric
 Research, 139, 90-100, 10.1016/j.atmosres.2014.01.007, 2014.
- Tao, J., Zhang, L., Cao, J., Zhong, L., Chen, D., Yang, Y., Chen, D., Chen, L., Zhang,
 Z., Wu, Y., Xia, Y., Ye, S., and Zhang, R.: Source apportionment of PM2.5 at
 urban and suburban areas of the Pearl River Delta region, south China With
 emphasis on ship emissions, Science of the Total Environment, 574, 1559-1570,
 10.1016/j.scitotenv.2016.08.175, 2017.
- Tian, S. L., Pan, Y. P., and Wang, Y. S.: Size-resolved source apportionment of
 particulate matter in urban Beijing during haze and non-haze episodes, Atmos
 Chem Phys, 16, 1-19, 10.5194/acp-16-1-2016, 2016.





- 649 Wang, Q., Shao, M., Zhang, Y., Wei, Y., Hu, M., and Guo, S.: Source apportionment
- of fine organic aerosols in Beijing, Atmospheric Chemistry and Physics, 9,
- 651 8573-8585, 2009.
- 652 Wang, X., Bi, X., Sheng, G., and Fu, J.: Chemical composition and sources of PM10
- and PM2. 5 aerosols in Guangzhou, China, Environmental Monitoring andAssessment, 119, 425-439, 2006.
- 655 Wong, J. P. S., Lee, A. K. Y., and Abbatt, J. P. D.: Impacts of Sulfate Seed Acidity and
- Water Content on Isoprene Secondary Organic Aerosol Formation, Environ. Sci.
 Technol., 49, 13215-13221, 10.1021/acs.est.5b02686, 2015.
- 658 Wu, Y., Guo, J., Zhang, X., and Li, X.: Correlation between PM concentrations and
- Aerosol Optical Depth in eastern China based on BP neural networks,
- 660 Geoscience and Remote Sensing Symposium, 2011, 5876-5886.
- 661 Xiang, P., Zhou, X. M., Duan, J. C., Tan, J. H., He, K. B., Yuan, C., Ma, Y. L., and
- Zhang, Y. X.: Chemical characteristics of water-soluble organic compounds
 (WSOC) in PM2.5 in Beijing, China: 2011-2012, Atmos. Res., 183, 104-112,
- 664 10.1016/j.atmosres.2016.08.020, 2017.
- 665 Yang, F., Kawamura, K., Chen, J., Ho, K. F., Lee, S. C., Gao, Y., Cui, L., Wang, T. G.,
- and Fu, P. Q.: Anthropogenic and biogenic organic compounds in summertime
 fine aerosols (PM2.5) in Beijing, China, Atmospheric Environment, 124,
 166-175, 10.1016/j.atmosenv.2015.08.095, 2016.
- 669 Yttri, K. E., Simpson, D., Nojgaard, J. K., Kristensen, K., Genberg, J., Stenstrom, K.,
- 670 Swietlicki, E., Hillamo, R., Aurela, M., Bauer, H., Offenberg, J. H., Jaoui, M.,
- 671 Dye, C., Eckhardt, S., Burkhart, J. F., Stohl, A., and Glasius, M.: Source
- apportionment of the summer time carbonaceous aerosol at Nordic rural
- background sites, Atmos Chem Phys, 11, 13339-13357, 2011.
- 674 Yu, L. D., Wang, G. F., Zhang, R. J., Zhang, L. M., Song, Y., Wu, B. B., Li, X. F., An,
- 675 K., and Chu, J. H.: Characterization and Source Apportionment of PM2.5 in an





- 676 Urban Environment in Beijing, Aerosol Air Qual Res, 13, 574-583,
- 677 10.4209/aaqr.2012.07.0192, 2013.
- 678 Zhang, H., Zhang, Z., Cui, T., Lin, Y.-H., Bhathela, N. A., Ortega, J., Worton, D. R.,
- 679 Goldstein, A. H., Guenther, A., Jimenez, J. L., Gold, A., and Surratt, J. D.:
- 680 Secondary Organic Aerosol Formation via 2-Methyl-3-buten-2-ol Photooxidation:
- 681 Evidence of Acid-Catalyzed Reactive Uptake of Epoxides, Environmental
- 682 Science & Technology Letters, 1, 242-247, 10.1021/ez500055f, 2014.
- 683 Zhang, Q., Streets, D. G., Carmichael, G. R., He, K. B., Huo, H., Kannari, A.,
- 684 Klimont, Z., Park, I. S., Reddy, S., Fu, J. S., Chen, D., Duan, L., Lei, Y., Wang, L.
- T., and Yao, Z. L.: Asian emissions in 2006 for the NASA INTEX-B mission,
 Atmos Chem Phys, 9, 5131-5153, 2009.
- 687 Zhang, R., Jing, J., Tao, J., Hsu, S.-C., Wang, G., Cao, J., Lee, C. S. L., Zhu, L., Chen,
- Z., and Zhao, Y.: Chemical characterization and source apportionment of PM 2.5
 in Beijing: seasonal perspective, Atmos Chem Phys, 13, 7053-7074, 2013.
- Zhang, Y., Cai, J., Wang, S., He, K., and Zheng, M.: Review of receptor-based source
 apportionment research of fine particulate matter and its challenges in China,
 The Science of the total environment, 586, 917-929,
 10.1016/j.scitotenv.2017.02.071, 2017.
- Zheng, J., Hu, M., Peng, J., Wu, Z., Kumar, P., Li, M., Wang, Y., and Guo, S.: Spatial
 distributions and chemical properties of PM 2.5 based on 21 field campaigns at
 17 sites in China, Chemosphere, 159, 480-487, 2016a.
- 697 Zheng, M., Salmon, L. G., Schauer, J. J., Zeng, L., Kiang, C., Zhang, Y., and Cass, G.
- R.: Seasonal trends in PM2. 5 source contributions in Beijing, China,
 Atmospheric Environment, 39, 3967-3976, 2005.
- 700 Zheng, X. X., Guo, X. Y., Zhao, W. J., Shu, T. T., Xin, Y. A., Yan, X., Xiong, Q. L.,
- Chen, F. T., and Lv, M.: Spatial variation and provenance of atmospheric trace
 elemental deposition in Beijing, Atmos. Pollut. Res., 7, 260-267,
 10.1016/j.apr.2015.10.006, 2016b.





- 704 Zhou, J. B., Xiong, Y., Xing, Z. Y., Deng, J. J., and Du, K.: Characterizing and
- 705 sourcing ambient PM2.5 over key emission regions in China II: Organic
- 706 molecular markers and CMB modeling, Atmospheric Environment, 163, 57-64,
- 707 10.1016/j.atmosenv.2017.05.033, 2017.
- 708 Zhu, C., Kawamura, K., and Fu, P.: Seasonal variations of biogenic secondary organic
- aerosol tracers in Cape Hedo, Okinawa, Atmospheric Environment, 130, 113-119,
- 710 10.1016/j.atmosenv.2015.08.069, 2016.
- 711 Zhu, Y., Yang, L., Kawamura, K., Chen, J., Ono, K., Wang, X., Xue, L., and Wang, W.:
- 712 Contributions and source identification of biogenic and anthropogenic
- 713 hydrocarbons to secondary organic aerosols at Mt. Tai in 2014, Environmental
- 714 Pollution, 220, 863-872, 10.1016/j.envpol.2016.10.070, 2017.





Table

Table 1. Summer $PM_{2.5}$ mass concentrations in Beijing from 2008-2016, average \pm

Year/Mont	2008/07	2009/07	2010/05	2016/05-06	2016/05-06
h					
Site	PKUERS	PKUERS	PKUERS	СР	PKUERS
	(µg m ⁻³)	(µg m ⁻³)	(µg m-3)	(µg m ⁻³)	(µg m ⁻³)
PM _{2.5}	92.3±44.7	88.2±52.3	62.7±36.5	43.0±17.5	45.5±19.8
OC	10.4±2.9	8.5±2.5	8.9±4.5	8.9±3.2	11.0±3.7
EC	3.3±1.5	2.5±0.9	2.1±1.1	0.7±0.5	1.8±1.0
Ref.	(Guo et al.,	(Zheng et	(Zheng et	This study	This study
	2012)	al., 2016a)	al., 2016a)		

standard deviation ($\mu g m^{-3}$).





Figure captions

Fig. 1 Concentrations of organic carbon from primary and secondary organic sources at (a) CP and (b) PKUERS as well as their contributions to the measured organic carbon at (c) CP and (d) PKUERS (%).Fig. 2 Comparison of the sources at PKUERS between 2016 and 2008

Fig. 3 Particle sources from different air mass origins

Fig. 4 Correlations between SOC and different influencing factors (a)-(b) ozone,

(c)-(d) temperature, (e)-(f) water and (g)-(h) H^+ concentratio







Fig. 1 Concentrations of organic carbon from primary and secondary organic sources at (a) CP and (b) PKUERS as well as their contributions to the measured organic carbon at (c) CP and (d) PKUERS (%).







Fig.2 Comparison of the sources at PKUERS between 2016 and 2008 (Guo et al. 2012)







Fig. 3 Particle sources from different air mass origins







Fig. 4 Correlations between SOC and different influencing factors (a)-(b) ozone, (c)-(d) temperature, (e)-(f) water and (g)-(h) H^+ concentration