

1 We thank both reviewers for their positive and constructive comments. We address their
2 specific points below:

3

4 **Reviewer 1: Bill Asher**

5 *I think the authors are glossing over a potential problem in that in the system they are studying,*
6 *CO₂ is an invasive flux (air-to-ocean) and DMS is an evasive flux (ocean-to-air). My hunch is*
7 *that Equation 5 is only strictly true when both gases are far from equilibrium *and* the flux is*
8 *in the same direction. Problems arise in applying Equation 5 for a mixed system, where one*
9 *gas is invading and one is evading, because the bubble gas flux is not the same, even when*
10 *normalized to a common diffusivity/solubility. In the case of invasion, the bubble overpressure*
11 *drives more gas than expected (based on the bulk air-ocean concentration difference) into the*
12 *water.*

13 This is true, and we are glad that you brought it up. However, it is a small effect. Woolf (1997)
14 provides the means to calculate the magnitude of this effect. The average fractional extra
15 pressure on the gas in contact with the sea (Δ) can be estimated from:

16
$$\Delta = (U/U_i)^2\%$$

17 where U_i is the wind speed at which the supersaturation of a particular gas equals 1% (49 m s⁻¹
18 for CO₂). A high wind speed (20 m s⁻¹) gives $\Delta = 0.167\%$. An atmospheric pCO₂ of 400 ppm
19 implies that the bubble overpressure would be 0.67 ppm. This is a ~2% enhancement of the
20 CO₂ flux when the air/sea concentration gradient is small (minimum for this study = 30 ppm).
21 Larger air/sea concentration gradients would diminish the magnitude of the bubble
22 overpressure further.

23

24 We have added the following text to our revised manuscript after equation 5:

25 “Strictly speaking, Equation 5 should also account for the influence of bubble overpressure,
26 which alters the gas flux due to bubbles when the concentration gradient is toward the ocean.
27 The extra pressure on the gas in the bubbles is calculated following Woolf (1997): $\Delta = (U_{10}/U_i)^2$
28 % where U_i is the wind speed at which the supersaturation of a particular gas equals 1% (49 m
29 s⁻¹ in the case of CO₂). A high wind speed (20 m s⁻¹) gives $\Delta = 0.167\%$, which would lead to a

30 ~2% enhancement of the CO₂ flux when the air/sea concentration gradient is 30 ppm (minimum
31 for this study) and into the ocean. The magnitude of this effect would be larger for gases less
32 soluble than CO₂ but we are able to ignore it for the purposes of this study.”

33

34 *one thing is clear from looking at the material in the supplements, is that using the Asher et al.*
35 *(2002) relationship for both CO₂ (invasion) and DMS (evasion) is not correct. The Asher et*
36 *al. (2002) relation is only for invasion. For evasion, there is a separate equation in Asher and*
37 *Wanninkhof (1998).*

38 This is also true and we are glad that it has been pointed out. As the bubble term for DMS is
39 small, there will be negligible impact upon our data. To be absolutely correct, we have adjusted
40 our data and the relevant equations in the revised manuscript.

41

42 *Line 54: "These processes include ..." Comment: Buoyancy effects are not a process. It might*
43 *be better in this sentence to say something like "These processes include diffusion, surface*
44 *renewal, and bubble-mediated transport. In turn, turbulence can be generated by wind stress,*
45 *wave-induced mixing, buoyancy currents, and wave breaking."*

46 Change made.

47

48 *Line 56: "A variety of theoretical, laboratory, and field ..." Comment: I don't think this sentence*
49 *is strictly true. My opinion is we have a fairly good understanding of the factors that affect gas*
50 *exchange from a phenomenological standpoint (the authors list them just a couple of sentences*
51 *earlier). What we lack is how to determine which of those processes are important under a*
52 *given set of circumstances.*

53 Changed sentence to:

54 “A variety of theoretical, laboratory, and field approaches have been used to study the
55 processes that control air/sea transfer, but we do not yet have a firm understanding of their
56 relative importance under a range of atmospheric and oceanic conditions.”

57

58 *Line 60: "Gas transfer via bubbles (k_{bub}) ..." Comment: It would be good to define k_{bub}*
59 *here. The point is that there are a couple of different ways to do this, you can go the Memery*
60 *and Merlivat (Memery, L. and L. Merlivat (1985). "Modeling of the gas flux through bubbles*
61 *at the air-water interface." Tellus, Ser. B 37: 272-285) approach and use the bulk air-water*
62 *concentration difference and accept that k_{bub} for invasion and evasion are different (e.g., I*
63 *used this approach in Asher et al. (1996, JGR-Oceans)) or you can redefine the air-water*
64 *concentration difference in terms of how bubbles would affect the equilibrium and have a*
65 *common k_{bub} (but then it might get complicated relating k_{bub} for invasion and evasion) as*
66 *done by Woolf (1997).*

67 We have clarified the definition of k_{bub} here and cover the issue of invasion vs evasion in
68 detail on Line 161 (to address the comment below).

69

70 *Line 78: "These measurements typically show DMS gas transfer velocities that are lower and*
71 *exhibit more linear wind speed dependence than those estimated for CO₂ based on dual tracer*
72 *studies (e.g. Bell et al., 2015; Yang et al., 2011; Goddijn-Murphy et al., 2012)." Comment: I*
73 *think the authors should be clear here that there are no CO₂ measurements from dual-tracer*
74 *studies. There are DT measurements for SF₆/He, which get related to CO₂ through diffusivity.*
75 *Then there are EC measurements for CO₂. Comparison of the DT-derived CO₂ transfer*
76 *velocities with CO₂ transfer velocities produced by EC measurements of CO₂ fluxes shows*
77 *relatively good agreement. It is the transfer velocities produced by EC measurements of DMS*
78 *fluxes that show different behavior.*

79 Change made.

80

81 *Line 87: Comment: maybe want to note that they agree when normalized to a common*
82 *diffusivity.*

83 Change made.

84

85 *Line 126: "The air side gas transfer contributes about 5% on average to the total resistance*
86 *for DMS." Comment: The air-side resistance fraction is a function of wind speed. Does this*

87 5% increase as U increases? COAREG must reproduce this, it was measured by McGillis et
88 al. a while back.

89 McGillis et al. (2000) used a non-linear relationship between waterside transfer velocity and
90 wind speed. Based on more recent measurements of DMS gas transfer velocity it is more
91 appropriate to assume a linear relationship. As a result, the relative contribution of airside
92 resistance to total resistance does not change substantially as a function of wind speed (see
93 Supplemental material in Bell et al 2015).

94

95 *Line 161: the relation in the text showing $k_w = k_{int} + k_{bub}$. Comment: I wonder if maybe*
96 *it is time to stop writing this as a general expression (I know, I am guilty of this as well). What*
97 *is generally true is that the total gas flux is equal to the sum of interfacial flux and the bubble*
98 *flux. Saying the overall transfer velocity is equal to the sum of the two transfer velocities really*
99 *only works if the concentration difference is far from equilibrium. Work through David's*
100 *relations from the 1997 paper and you'll find they are a bit convoluted in terms of how exactly*
101 *the pieced (his Delta term) fit together to make a coherent physical picture. If you start by*
102 *assuming it is the fluxes, not the transfer velocities, which sum linearly, the assumptions*
103 *required to get to the various relations proposed are more easily understood.*

104 We have changed the text to be more accurate.

105

106 *Technical Comments:*

107 *1. Multiple citations are not in any recognizable order. Sometimes they are chronological,*
108 *sometimes alphabetical. I don't remember what the ACP style guide says, but I am sure it is*
109 *not "random."*

110 Agreed! We will correct this.

111

112 *2. Line 74: "... studies indicate a non-linear dependence ..."*

113 Changed to "studies observed a non-linear wind speed dependence"

114

115 *Line 91: Shouldn't cite papers that are not published or submitted.*

116 OK, removed.

117

118 *Line 175: The two f values are opposite from what is given in the supplement. Not sure*
119 *which is correct, but it should be consistent (and correct).*

120 Good spot. Changed in the supplemental material.

121

122 *Line 323: I know this is petty, but I don't think Woolf (1997) is based on laboratory data.*

123 Woolf (1997) did use laboratory data from other studies. We will clarify this.

124

125 *Equation 7: Figure caption says "cubic" and equation 7 is quadratic. Resolve this difference.*

126 Changed.

127

128 *Line 384: The citation to Asher and Wanninkhof (1998) should be to Asher et al. (1996).*

129 *If you really must cite Asher and Wanninkhof (1998) in this context, which you shouldn't,*

130 *at least make it the other Asher and Wanninkhof (1998) paper that is directly relevant*

131 *(see citation above).*

132 Changed.

133

134 **Reviewer 2: Ian Brooks**

135

136 *Line 86-87: "In that study, no statistically significant difference was observed in gas transfer-*
137 *wind speed relationships of CO₂ and DMS for winds below 10 m s⁻¹" – need to clarify this*
138 *statement, was a significant difference found for winds above 10 m s⁻¹? Was 10 m s⁻¹ the*
139 *maximum wind speed in the study?*

140 Changed to:

141 “In that study, no data were collected for winds greater than 10 m s⁻¹ and no statistically
142 significant difference was observed in the CO₂ and DMS gas transfer-wind speed
143 relationships.”

144

145 *Line 104-105: as noted in the comment from Blomquist, the air-side resistance is not a function*
146 *of solubility (though its contribution to the derivation of waterside transfer velocity is*
147 *dependent on solubility).*

148 This has been corrected.

149

150 *Line 122-123: The use of the COAREG 3.1 model to calculate air-side transfer velocities in*
151 *order to derive the waterside transfer velocity introduces an assumption that COAREG is*
152 *providing valid values of k_a . Any uncertainty in this will impact the later results and should be*
153 *acknowledged and if possible quantified.*

154 We have added the following sentence:

155 “Note that the use of the COAREG 3.1 model introduces a small uncertainty in our estimates
156 of waterside DMS gas transfer velocity (approximately $\pm 2\%$ when wind speed = 20 m s⁻¹).”

157

158 *Line 126-127: “The air side gas transfer contributes about 5% on average to the total*
159 *resistance for DMS” – do you mean ‘air-side resistance’ rather than air-side transfer?*

160 Yes. Changed.

161

162 *Line 140: “...(Equation 4)...” should be “...(Equation 3)...”*

163 *Line 170: “...into Equation 6 yields...” should be “...into Equation 5 yields...”*

164 Changed.

165

166 *Line 192 - figures: Figures 1 and 2 are introduced here, but figure 2 is not actually discussed*
167 *until after discussion of figure 4 breaking the flow of discussion and figures, and leaving me,*
168 *initially, confused as to how I’d missed the discussion of figure 2. In fact, only the general*

169 *environmental conditions shown in figure 1 are discussed here, not the gas flux results, which*
170 *are discussed much later. Figure 1 would be better split into 2, separating the gas fluxes into*
171 *a figure matching the format of current figure 2. The figures showing the gas flux results could*
172 *then be placed in a logical order within the discussion. Since wave state is a relevant parameter*
173 *in the later discussion, it would be useful to add a time series of at least significant wave height*
174 *to figure 1.*

175 We agree. Figure 1 has been split and a timeseries of Hs has been added. Figures are reordered
176 so the gas flux/transfer velocity figures follow the whitecapping figures.

177

178 *Line 207: The authors note that their estimates of whitecap fraction as a function of wind speed*
179 *are substantially lower than other recently published values – at times an order of magnitude*
180 *lower. A likely reason for this is the exposure settings on the camera. During the HiWinGS*
181 *project cruise in 2013 two independent sets of cameras were used for whitecap imaging. They*
182 *were initially found to give whitecap fractions that differed by a factor of several. Tests were*
183 *conducted during the final transit of the cruise, in which a pair of identical cameras were run*
184 *side by side; one with fixed exposure settings, the other having the exposure settings changed*
185 *every few hours. The exposure settings were found to make a substantial difference to the*
186 *whitecap fraction calculated using the same Callaghan and White (2009) algorithm used here*
187 *– up to a factor of 4 for the range of settings tested. It was found that almost all of this difference*
188 *(both between the 2 cameras in the exposure trial, and between the two sets of cameras used*
189 *throughout the cruise) was removed if the images were first ‘normalised’ to remove any*
190 *brightness gradient across the image. Brief details of these tests will be given in*

191 *Brumer, S. E., C. J. Zappa, I. M. Brooks, H. Tamura, S. M. Brown, B. Blomquist, C. W. Fairall,*
192 *A. Cifuentes-Lorenzen, 2017: Whitecap coverage dependence on wind and wave statistics as*
193 *observed during SO GasEx and HiWinGS, J. Phys. Oceanogr. (under revision)*

194 Many of the potential issues highlighted above (exposure settings, intercomparability of image
195 processing) have actually been addressed. We have added additional information about the
196 whitecap image processing to the Supplemental material.

197

198 The lower bounds of the Knorr W data do fall below recent parameterisations, most notably in
199 the wind speed range of $\sim 7.5 \text{ m s}^{-1}$ to 12.5 m s^{-1} . However, when the Knorr W data are binned

200 by wind speed (squares in Figure 3a, main manuscript), the binned W data compare favourably
201 with the Schwendeman and Thomson (2015) estimate of whitecap fraction at all wind speeds,
202 with the exception of the lowest bin of 3 m s^{-1} . The binned Knorr W data also agree favourably
203 with Callaghan et al., (2008), except for the 9 m s^{-1} and 11 m s^{-1} binned datapoints, for which
204 the Knorr binned W data are lower. When binned by wind speed, the Knorr W data and the two
205 parameterisations generally fall within a factor of 2 across the wind speeds examined.

206

207 There is clearly quite a lot of scatter in the Knorr W dataset, and many data points lie below the
208 Schwendeman and Thomson (2015) and Callaghan et al., (2008) parameterisations. However,
209 we do not believe that the primary driver of the differences observed is due to the image
210 processing methodology employed. Rather, it could be that other sources of variability (at a
211 given wind speed) have caused the observed differences. Examples include: (i) water chemistry
212 (surfactants); (ii) total wave field energy dissipation; and (iii) energy dissipation by microscale
213 breaking waves as opposed to air-entraining whitecaps. It is beyond the scope of this paper to
214 address these potential causes.

215

216 *Line 213-215: “Stage A whitecap fraction data is highly variable at $\sim 11 \text{ m s}^{-1}$ 213 wind speeds*
217 *(Figure 3b), which is driven by the difference in the wind-wave conditions during Knorr_11*
218 *(ST184 vs ST191, Figure 4a)” – two points:*

219 *(1) the difference is ascribed to different wind-wave conditions at the two stations, but no wave*
220 *data are shown. As noted above, relevant wave parameters need to be added to figure 1.*

221 Done.

222

223 *(2) A similar broad range of stage A whitecaps is evident at around $U = 6 \text{ m s}^{-1}$, also resulting*
224 *from grouping of high/low values by different stations...are the wind-wave conditions similarly*
225 *different in this case?*

226 We agree that there is a grouping at $U=6 \text{ m s}^{-1}$. It is unclear what caused this (although ST181
227 was in the Gulf Stream and thus had much higher water temperatures). However, this
228 discussion is beyond the scope of this paper.

229

230 *Lines 218-223: The authors first note that where stage A whitecap fractions is $< 10^{-4}$ the*
231 *relationship with R_H is more scattered than at higher fractions; they then note a number of*
232 *factors that affect wave breaking and so whitecap fraction, but don't make a coherent link back*
233 *to their initial point about the scatter in the stage-A whitecap / R_H relationship. This reads as*
234 *an almost unconnected series of statements...all true, but leaving the reader wondering what*
235 *the point being made is.*

236 We agree that this text needed adjusting. We also decided that the text discussing Stage A
237 whitecap variability and R_H was not essential to the manuscript and have moved the modified
238 text and the related Figure to Supplemental information.

239

240 *Line 282: "... (Figure 5)" -> "... (Figure 5b,d)" – again text refers to high wave conditions for*
241 *ST191 but no wave data provided for reader to assess.*

242 In this instance, we feel that it is sufficient for us to reference the in depth discussion in Bell et
243 al. (2013).

244

245 *Figure 5 and the discussion of it have some general issues:*

246 *- It's hard to see the pink/green lines against the mass of pink/green dots on panels a and b –*
247 *it*

248 *would help here to plot the dots in a paler shade of pink/green to allow the lines to stand out.*

249 The figure has been adjusted to improve readability.

250

251 *- The curves shown, for the COAREG3.1 model are a useful reference, but fits to the actual*
252 *data*

253 *are also needed; these would allow a much clearer assessment of how closely the COARE*
254 *model*

255 *agrees with the observations.*

256 We tried this but it made the plots too busy. We decided it best not to include both.

257

258 - lines 282-284: “Under the high wind, high wave conditions encountered during ST191, the
259 wind speed-dependence of $k_{DMS,Sc}$ was lower than expected, with a slope roughly half that of
260 the rest of the cruise data. This effect was not observed at ST184.” – since the ST191 data are
261 not highlighted in any way it is not possible for the reader to judge the behaviour here. Of note
262 perhaps is not simply the high wind and wave conditions during ST191 but the different time
263 history of the winds – a sustained period of high winds during ST191 vs a very short period in
264 ST184 where the wind rises rapidly, spikes, and decreases rapidly. These two periods are likely
265 to produce very different wave fields at the same wind speeds – again, reason to plot the wave
266 parameters in figure 1 – which might explain the very different whitecap fractions seen in
267 Figure 1b for these periods.

268 In addition to plotting the wave data, we have added the word ‘sustained’ to this sentence and
269 referenced Bell et al. 2013)

270

271 Line 294 & 305: the phrasing “until 11 m s⁻¹ wind speed” is rather clumsy; ‘until’ implies a
272 variation over time, which is not what is meant – “...up to wind speeds of 11 m s⁻¹” would read
273 better.

274 Change made.

275

276 Line 326: “... Δk_w is near zero at very low wind speeds ($U_{10} \leq 4.5$ m s⁻¹)...” – this is hard to
277 judge. Eyeballing the data points I would agree; however, there are only 3 points at $U < 4.5$
278 m s⁻¹, and all of those at $U > 3$ m s⁻¹. Their mean Δk is ~5 cm/hr and the fitted curve approaches
279 a Δk_w of ~3 cm/hr at $U = 0$, not zero. One might argue for an alternative functional form in
280 which the exponents were not prescribed might better represent the data; the quadratic used
281 here implicitly assumes the functional dependence.

282 We have tested a variety of functional forms and number of exponents for our best fit: linear,
283 polynomial (n=2,3), power (n=1,2) and exponential (n=1,2). The goodness of fit was extremely
284 comparable between all fits (from $R^2 = 0.66$ to $R^2 = 0.67$) with the exception of the linear fit
285 ($R^2 = 0.62$). As the goodness of fit did not help with our decision, the choice of fit is subjective.
286 Having considered the different fit lines, we feel that a simple power law fit ($\Delta k_w =$

287 $0.177U_{10}^{1.928}$) represents the low Δk_w data better than our original choice of fit. We have
288 changed the manuscript accordingly.

289 The fit line to the data tells us that Δk_w is indeed positive at low wind speeds. This could be
290 driven by effects such as chemical enhancement and sea surface skin temperature (see
291 Discussion), which we have not controlled for here. We used the term ‘near zero’ intentionally
292 but now clarify this in the revised manuscript with “(< 4.5 cm/hr)”

293

294 *What is the reasoning behind using 4-hour averages of transfer velocities here (and*
295 *elsewhere)? 4*

296 *hours is quite a long time relative to the time period over which significant changes in forcing*
297 *can take place. Granted it greatly reduces scatter, but I would wary of averaging over periods*
298 *much long than ~1 hour.*

299 Some averaging of the data was necessary to reduce the scatter, but we accept that minimising
300 the averaging time period is important. We have changed our averaging period to two hours.

301

302 *Line 358: “...the relationship between Δk_w and whitecap areal extent appears to be linear.” –*
303 *I’m not convinced this is entirely true – approximately so over the range $0.005 < W < 0.05$,*
304 *but ~half the data points lie at $W < 0.005$, and seem to drop off rather more rapidly than the*
305 *fit to the high values of W would indicate. Plotting W on a log scale might give a rather different*
306 *impression. Not that IF the relationship is linear as suggested then eyeballing a fit (if you claim*
307 *a linear fit, it would help to show it!) suggests $\Delta k = 10-15$ cm/hr at $W = 0$, which raises*
308 *questions as to why that should be when there are no bubbles to account for a difference in k ,*
309 *and why this minimum difference is several times higher than that derived as a function of wind*
310 *speed. If on the other hand a roughly linear fit of Δk to $\log(W)$ existed (which I think the rapid*
311 *drop off in Δk at very low W might support) then Δk would approach zero at low W .*

312 We have revised the statement to:

313 “The functional form of the relationship between Δk_w and whitecap areal extent appears to be
314 linear for $W_T > 0.005$. However, the Knorr_11 dataset is small and quite scattered, particularly
315 when $W_T < 0.005$.”

316

317 *Line 366-367: “In this case, Δkw should be more strongly correlated with W_A than W_B or*
318 *W_T .” – in a general sense, this is true, but is only if the various factors affecting foam decay*
319 *vary. If foam decay rate is constant then W_T should be proportional to W_A .*

320 We agree. We feel this case is broadly covered later in the paragraph. We have adjusted the
321 text slightly to take into account a constant decay rate:

322 “ W_T and W_A may be equally good (or poor) proxies for bubbles because: (i) surfactant activity
323 was either insignificant or sufficiently invariant in the study region (despite high biological
324 productivity) that W_B does not confound the relationship between W_T and W_A ”

325

326 *Line 387: “Both models significantly underestimate k_{bub,CO_2} at wind speeds below about 11*
327 *m s⁻¹.” Actually both models rather underestimate the observationally derived K_{bub,CO_2} at*
328 *all windspeeds; however, note that both models are driven by the observed wind-whitecap*
329 *relationship, which has already been stated to be low compared to other recent estimates. Is*
330 *the agreement better using a whitecap function that agrees more closely with the recent*
331 *consensus?*

332 The agreement is slightly improved (at intermediate wind speeds) using a different whitecap
333 function. Although we see no reason to doubt our whitecap measurements, it is useful to
334 observe the importance of the wind speed-whitecap fraction relationship to the output from
335 these models, so we have added these k_{bub,CO_2} estimates to the manuscript.

336

337 *While the Asher et al. model is lower than the observations, it is not wildly so (essentially*
338 *matching the lower boundary of the observed values), and the agreement in both values and*
339 *functional behaviour is rather convincing.*

340 We are possibly less optimistic on this point and hopefully can agree to disagree!

341

342 *Line 455: “...eddy covariance setup...” -> “...eddy covariance system...”*

343 Change made.

344

345 *Line 463: reference to paper in preparation not generally allowed...if it's not 'in press' by the*
346 *time this manuscript is copy edited, the copy editor will want the reference cut.*

347 Change made.

348

349 *Figure 3: it's hard to pick out the curves and black square points against the mass of black*
350 *dots.*

351 *Suggest changing black dots to mid-grey and ensuring everything else is plotted over them. It*
352 *would also be good to see a functional fit to this data as well as the functions from previous*
353 *studies...especially as this function is later used to drive the kbub,co2 models plotted in figure*
354 *8.*

355 Revised accordingly.

356

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369 **Estimation of bubbled-mediated air/sea gas exchange from**
370 **concurrent DMS and CO₂ transfer velocities at intermediate-**
371 **high wind speeds**

372

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386 **Abstract**

387 Simultaneous air/sea fluxes and concentration differences of dimethylsulfide (DMS) and
388 carbon dioxide (CO₂) were measured during a summertime North Atlantic cruise in 2011. This
389 dataset reveals significant differences between the gas transfer velocities of these two gases
390 (Δk_w) over a range of wind speeds up to 21 m s⁻¹. These differences occur at and above the
391 approximate wind speed threshold when waves begin breaking. Whitecap fraction (a proxy for
392 bubbles) was also measured and has a positive relationship with Δk_w , consistent with enhanced
393 bubble-mediated transfer of the less soluble CO₂ relative to that of the more soluble DMS.
394 However, the correlation of Δk_w with whitecap fraction is no stronger than with wind speed.
395 Models used to estimate bubble-mediated transfer from *in situ* whitecap fraction under-predict
396 the observations, particularly at intermediate wind speeds. Examining the differences between

397 gas transfer velocities of gases with different solubilities is a useful way to detect the impact
398 of bubble-mediated exchange. More simultaneous gas transfer measurements of different
399 solubility gases across a wide range of oceanic conditions are needed to understand the factors
400 controlling the magnitude and scaling of bubble-mediated gas exchange.

401 **1 Introduction**

402 Air/sea exchange is a significant process for many compounds that have biogeochemical and
403 climatic importance. Approximately 25% of the carbon dioxide (CO₂) released into the
404 atmosphere by anthropogenic activities has been taken up by the world oceans, which has
405 tempered its climate forcing while leading to ocean acidification (Le Quéré et al., 2015). The
406 biogenic gas dimethylsulfide (DMS) is a major contributor to the mass of marine atmospheric
407 aerosol (Virkkula et al., 2006). Volatile organic compounds (VOCs) such as isoprene, acetone
408 and acetaldehyde alter the oxidising capacity of the troposphere (Carpenter et al., 2012). The
409 solubility differences between these VOCs mean that their exchange is controlled to differing
410 degrees by processes on the water and air side of the air/sea interface (Yang et al., 2014). Many
411 of the factors influencing air/sea gas exchange will be altered by future changes in climate,
412 ocean circulation and biology. Earth system models and air quality models require more
413 accurate understanding of the processes that influence air/sea gas transfer.

414 Air/sea gas exchange is typically parameterised as a function of the ocean/atmosphere bulk
415 concentration difference (ΔC) and the physical mixing induced by wind stress at the interface
416 (Liss and Slater, 1974). The air/sea flux is typically described using the expression:

$$417 \quad \text{Flux} = K(C_w - \alpha C_a) \quad \text{Equation 1}$$

418 where C_w and C_a are the trace gas bulk concentration on either side of the interface, α is the
419 dimensionless water/air solubility of the gas in seawater and K is the gas transfer velocity. The
420 physics of gas transfer are implicitly represented by the gas transfer velocity, which is
421 commonly expressed in water-side units of velocity (cm hr⁻¹) and parameterized as a function
422 of wind speed (U_{10}) and Schmidt number (Sc). The simplicity of Equation 1 belies the
423 complexity of the processes involved in air/sea gas transfer. These processes include diffusion,
424 surface renewal, ~~buoyancy effects, wave-induced mixing, wave breaking~~ and bubble-mediated
425 transport. In turn, turbulence can be generated by wind stress, wave-induced mixing,
426 buoyancy, currents and wave breaking. A variety of theoretical, laboratory, and field

427 approaches have been used to study these processes that control air/sea transfer, but we do not
428 yet have a firm understanding of their relative importance factors that control air/sea transfer
429 under a range of atmospheric and oceanic conditions.

430 The gas transfer-wind speed relationships for gases of different solubility may be affected by
431 breaking waves and bubbles (Keeling, 1993; Woolf, 1993, 1997). Gas ~~transfer~~ invasion and
432 evasion via bubbles (k_{bub}) is sensitive to the void fraction (ratio of air volume to total volume)
433 of the bubble plume as well as the bubble size distribution. Bubble injection depth and
434 cleanliness of the surface (influenced by surfactants) affect bubble rise velocity and residence
435 time. Bubble residence time determines the time available for equilibration to occur while
436 bubble volume, pressure and gas diffusivity (Sc) govern the time needed for a bubble to
437 equilibrate. The magnitude of k_{bub} is expected to be greater for sparingly soluble gases (e.g.
438 CO₂, dimensionless solubility ~1) than for more soluble gases such as DMS (dimensionless
439 solubility ~15), particularly when bubbles are fully equilibrated. Bubble-mediated gas transfer
440 has been studied in the laboratory (Asher et al., 1996; Rhee et al., 2007) and using models (e.g.
441 Woolf, 2005; Woolf et al., 2007; Fairall et al., 2011; Goddijn-Murphy et al., 2016).

442 Deliberate, dual-tracer techniques have estimated gas transfer by measuring the evasion of a
443 pair of sparingly soluble gases with different diffusivity (³He and SF₆, dimensionless solubility
444 ≤0.01). These studies ~~indicate observed a~~ non-linear wind speed dependence of the gas transfer
445 velocity, in qualitative agreement with earlier studies in wind-wave tanks (e.g. Wanninkhof et
446 al., 1985; Liss and Merlivat, 1986; Watson et al., 1991). Direct, shipboard measurements of
447 waterside gas transfer have also been made by eddy covariance (e.g. McGillis et al., 2001;
448 Huebert et al., 2004; Marandino et al., 2007; Miller et al., 2010; Bell et al., 2013). These
449 measurements typically show DMS gas transfer velocities that are lower and exhibit more
450 linear wind speed dependence than ~~those the CO₂ transfer velocity-wind speed relationship~~
451 inferred from estimated for CO₂-based on dual tracer studies (e.g. Yang et al., 2011; Goddijn-
452 Murphy et al., 2012; Bell et al., 2015). It has been suggested that the difference between the
453 open ocean gas transfer velocities of CO₂ and DMS is due to the reduced importance of bubble-
454 mediated exchange for DMS (Blomquist et al., 2006; Fairall et al., 2011; Goddijn-Murphy et
455 al., 2016).

456 Only one set of concurrent CO₂ and DMS gas transfer velocity measurements have been
457 published to date (Miller et al., 2009). In that study, no data were collected for winds greater

458 ~~than 10 m s⁻¹ and no statistically significant difference was observed in the CO₂ and DMS gas~~
459 ~~transfer-wind speed relationships after normalising both gases to a common diffusivity.~~~~In that~~
460 ~~study, no statistically significant difference was observed in gas transfer wind speed~~
461 ~~relationships of CO₂ and DMS for winds below 10 m s⁻¹.~~ This study presents a more extensive
462 set of CO₂ and DMS gas transfer velocities that were measured simultaneously aboard the R/V
463 Knorr in the 2011 summertime North Atlantic in both oligotrophic and highly productive
464 waters. The DMS ~~and CO₂~~-gas transfer velocities are discussed separately in detail by Bell et
465 al. (2013) ~~and Miller et al., In Prep.~~ Here we focus specifically on what can be learned about
466 gas transfer from the differences in behaviour of two different solubility gases at intermediate
467 and high wind speeds.

468 **2 Methods**

469 **2.1 Seawater, atmospheric and flux measurement systems**

470 The measurement setups for DMS and CO₂ concentrations in air and water and the eddy
471 covariance flux systems have been discussed in detail elsewhere (Miller et al., 2008; Saltzman
472 et al., 2009; Miller et al., 2010; Bell et al., 2013; Landwehr et al., 2014; Bell et al., 2015;
473 Landwehr et al., 2015). We provide a summary ~~plus~~and some additional details in the
474 Appendix ~~(Section 6)~~.

475 **2.2 Gas transfer velocity calculations**

476 In this section we describe the calculation of DMS and CO₂ gas transfer velocities from the
477 Knorr_11 cruise data. Measured gas transfer velocities are transformed into water side only
478 gas transfer velocities in order to remove the influence of air-side resistance. The relative
479 contribution of air-side resistance to the total resistance is a function of solubility and thus
480 different for the two gases. Finally, we discuss the most appropriate approach for comparing
481 the water-side gas transfer velocities, given that the two gases have different molecular
482 diffusivity and solubility.

483 Total gas transfer velocities (K) are calculated for CO₂ and DMS for each 10-minute flux
484 interval of the Knorr_11 cruise using Equation 1. The temperature-dependent dimensionless
485 solubilities of CO₂ and DMS in seawater ~~is~~are calculated following Weiss (1974) and Dacey
486 et al. (1984) ~~respectively~~. These gas transfer velocities reflect the result of resistance on both

487 sides of the interface (Liss and Slater, 1974). The water side contribution to the total resistance
488 is determined as follows:

$$489 \quad k_w = \left[\frac{1}{K} - \frac{\alpha}{k_a} \right]^{-1} \quad \text{Equation 2}$$

490 where k_w and k_a are the air side and water side gas transfer velocities and α is dimensionless
491 water/air solubility. Note that we use the α reported by Dacey et al. (1984) in these calculations
492 rather than H as there appears to be an error in conversion between α and H in that study (see
493 Supplemental information for discussion). CO_2 solubility is sufficiently low that air side
494 resistance is negligible and the water side gas transfer is assumed equal to the total transfer
495 velocity ($k_{\text{CO}_2} = K_{\text{CO}_2}$). The air side resistance for DMS needs to be accounted for because it is
496 a moderately soluble gas (McGillis et al., 2000). Air side gas transfer velocities (k_a) for DMS
497 were calculated for each 10 minute flux interval with the NOAA COAREG 3.1 model, using
498 sea surface temperature (SST) and horizontal wind speed measured during the cruise. The
499 NOAA COAREG 3.1 model (Fairall et al., 2011) is an extension of the COARE bulk
500 parameterization for air/sea energy and momentum fluxes to simulate gas transfer (Fairall et
501 al., 1998; Fairall et al., 2000). The air side gas-transfer-resistance contributes about 5% on
502 average to the total resistance for DMS. NOAA COAREG 3.1 model calculations were carried
503 out using a turbulent/molecular coefficient, $A = 1.6$, and bubble-mediated coefficient, $B = 1.8$
504 (Fairall et al., 2011). Knorr_11 measurements of SST, air temperature, relative humidity, air
505 pressure, downward radiation and wind speed were used as input parameters to the model. Note
506 that the use of the COAREG 3.1 model introduces a small uncertainty in our estimates of
507 waterside DMS gas transfer velocity (approximately $\pm 2\%$ when wind speed = 20 m s^{-1}).

508 To facilitate comparison of transfer coefficients for the two gases across a range of sea surface
509 temperatures, gas transfer velocities are corrected for changes in molecular diffusivity and
510 viscosity. The correction typically involves the normalisation of water side gas transfer
511 velocities to a common Schmidt number ($Sc=660$), equivalent to CO_2 in seawater at 20°C :

$$512 \quad k_{X,660} = k_X \cdot \left(\frac{660}{Sc_X} \right)^{-0.5} \quad \text{Equation 3}$$

513 where subscript x refers to CO₂ or DMS (i.e. $k_{DMS,660}$ and $k_{CO_2,660}$). Temperature-dependent
514 Sc_{CO_2} and Sc_{DMS} were obtained using the *in situ* seawater temperature from the ship's bow
515 sensor and parameterisations from Wanninkhof (1992) and Saltzman et al. (1993).

516 The Sc number normalization (Equation 34) is commonly used across the whole range of wind
517 speeds. In fact, it is only appropriate only for low or moderate winds in which when interfacial
518 gas transfer dominates over bubble-mediated gas exchange. If bubbles are an important
519 component of gas transfer then solubility also plays a role and normalization based on Sc alone
520 may not be sufficient.

521 To develop a more rigorous comparison of k_{DMS} and k_{CO_2} , we normalized the water side
522 transfer velocities of DMS to the Schmidt number of CO₂ at the *in situ* sea surface temperature
523 of each 10-minute flux interval, as follows:

$$524 \quad k_{DMS,Sc} = k_{DMS} \left(\frac{Sc_{CO_2}}{Sc_{DMS}} \right)^{-0.5} \quad \text{Equation 4}$$

525 where Sc_{CO_2} and Sc_{DMS} are the Schmidt numbers of CO₂ and DMS at the *in situ* sea surface
526 temperature. Compared to normalizing both DMS and CO₂ to $Sc=660$, this approach has the
527 advantage of correcting only k_{DMS} , with no correction to k_{CO_2} . The Sc correction for DMS
528 should be reasonably accurate, assuming that the bubble-mediated transfer for the more soluble
529 DMS is relatively small.

530 On the Knorr_11 cruise, the variability in sea surface temperature was small ($1\sigma = \pm 1^\circ\text{C}$). As
531 a result, there is little difference in the variability or wind speed dependence of Sc -corrected
532 k_{CO_2} compared to k_{CO_2} at the *in situ* temperature (Figure 5 vs. Figure S4-S5 in Supplemental
533 information). In Section 3.4, the relationship between CO₂ and DMS gas transfer velocities
534 and wind speed is examined using $k_{DMS,Sc}$ and k_{CO_2} .

535 2.3 Calculation of k_{bub,CO_2}

536 The flux of a water-side controlled gas transfer velocity (k_w) is comprised is equal to the sum
537 of the interfacial flux and the bubble-mediated flux. For gases with significant air/sea
538 disequilibrium these processes are often considered as parallel transfer velocities, i.e. transfer
539 mechanisms, which operate in parallel, total transfer velocity i.e. $k_w = k_{int} + k_{bub}$. See Woolf

540 (1997) for a more complete discussion of bubble-mediated transfer for gases close to
 541 ocean/atmosphere equilibrium. We assume that turbulence and diffusive mixing at the sea
 542 surface operate similarly upon the interfacial air/sea transfer of CO₂ and DMS (i.e. $k_{int,CO_2} =$
 543 $k_{int,DMS}$), given appropriate normalization for the differences in molecular diffusivity. Observed
 544 differences between $k_{DMS,Sc}$ and k_{CO_2} should therefore be a measure of the difference between
 545 the bubble-mediated contributions to DMS and CO₂ gas transfer:

$$546 \quad \Delta k_w = k_{bub,CO_2} - k_{bub,DMS} \quad \text{Equation 5}$$

547 Strictly speaking, Equation 5 should also account for the influence of bubble overpressure,
 548 which alters the gas flux due to bubbles when the concentration gradient is into the ocean. The
 549 extra pressure on the gas in the bubbles is calculated following Woolf (1997): $\Delta = (U_{10}/U_i)^2 \%$
 550 where U_i is the wind speed at which the supersaturation of a particular gas equals 1% (49 m s⁻¹
 551 in the case of CO₂). A high wind speed (20 m s⁻¹) gives $\Delta = 0.167\%$, which would lead to
 552 only a ~2% enhancement of the CO₂ flux when the air/sea concentration gradient is 30 ppm
 553 (minimum for this study) and into the ocean. The magnitude of this effect would be larger for
 554 gases less soluble than CO₂ but we are able to ignore it for the purposes of this study.

555 k_{bub,CO_2} and $k_{bub,DMS}$ are related by the influence of solubility and diffusivity upon bubble-
 556 mediated transfer. We parameterize this relationship simply as $k_{bub,DMS} = f \cdot k_{bub,CO_2}$.
 557 Substitution into Equation 5 yields:

$$558 \quad k_{bub,CO_2} = \frac{\Delta k_w}{1 - f} \quad \text{Equation 6}$$

559 The value of f depends on seawater temperature and the complex dynamics of bubble formation
 560 and cycling (size distributions, surfactants, etc.). At the mean SST encountered in this study
 561 (9.8°C), the bubble gas transfer models of Woolf (Woolf, 1997) and Asher (Asher and
 562 Wanninkhof, 1998; Asher et al., 2002) yield values for f of 0.14 and 0.287, respectively (see
 563 Supplemental information for model equations).

564 **2.4 Sea surface imaging**

565 Whitecap areal fraction was measured using images of the sea surface recorded with a digital
 566 camera (5 mega pixel Arecont Vision, 16 mm focal length lens) mounted 14.6 m above the

567 ocean surface at an angle of $\sim 75^\circ$ from the nadir. Image footprints represent $\sim 7600 \text{ m}^2$ of sea
568 surface. Images were collected at a sample period of about 1 second and post-processed for
569 whitecap fraction according to the Automated Whitecap Extraction algorithm method
570 (Callaghan and White, 2009). [More detail on the methodology, camera exposure settings and](#)
571 [data comparability are provided in the Supplemental information.](#) Images were further
572 processed to distinguish whitecap pixels as either stage A or stage B whitecaps by applying a
573 spatial separation technique (Scanlon and Ward, 2013). The whitecap fraction measurements
574 were averaged in the same way as the gas transfer velocities (i.e. time-averaged mean values
575 as well as 2 m s^{-1} wind speed bins).

576 **3 Results**

577 **3.1 Cruise location and environmental conditions**

578 This study took place in the summertime North Atlantic (June 24 – July 18, 2011; DOY 175-
579 199), departing and returning to Woods Hole, MA. Most of the data were collected north of
580 50°N , including the occupation of four 24-36 hr stations – ST181, ST184, ST187 and ST191
581 (Figures 1–& 2). The cruise track was designed to sample regions with high biological
582 productivity and phytoplankton blooms, with large air/sea concentration differences for CO_2
583 and DMS. The cruise meteorology and physical oceanography is discussed in detail by (Bell
584 et al., 2013). A series of weather systems travelling from West to East passed over the region
585 during the cruise. Wind speeds ranged from ~ 1 to 22 m s^{-1} , with strongest winds during the
586 frontal passages at stations ST184 and ST191 (Figure 1b). Atmospheric boundary layer
587 stability was close to neutral for most of the cruise ($|z/L| < 0.07$; 75% of the time), with
588 infrequent stable conditions ($z/L > 0.05$; $< 8\%$ of the time, [Figure 1a](#)). There was no evidence
589 that the stable periods affected the flux measurements (Bell et al., 2013). Whitecap areal
590 fraction increased up to a maximum of ~ 0.06 in response to high wind speeds (Figure 1b).

591 **3.2 Whitecaps**

592 Whitecaps were observed during Knorr_11 when wind speeds exceeded 4.5 m s^{-1} , a typical
593 wind speed threshold for whitecap formation in the open ocean (Callaghan et al., 2008;
594 Schwendeman and Thomson, 2015). Whitecap areal fraction is a strong, non-linear function
595 of wind speed (Figure 23a). The whitecap vs. wind speed relationship for Knorr_11 is similar
596 in shape, ~~but considerably lower than to recently-previously published,~~ wind speed-based

597 whitecap parameterisations (Callaghan et al., 2008; Schwendeman and Thomson, 2015). At
598 intermediate wind speeds the Knorr_11 whitecap data are ~~as much as an order of magnitude~~
599 lower than the parameterisations (Figure ~~23~~a).

600 Total whitecap coverage is a function of (i) active ‘stage A whitecaps’ (W_A) produced from
601 recent wave breaking and (ii) maturing ‘stage B whitecaps’ (W_B) that are decaying foam from
602 previous breakers. The Stage A whitecap fraction data is highly variable at $\sim 11 \text{ m s}^{-1}$ wind
603 speeds (Figure ~~23~~b), which is driven by the difference in the wind-wave conditions during
604 Knorr_11 (~~ST184 vs ST191, Figure 4~~~~see discussion in Supplemental information~~). ~~Stage A~~
605 ~~whitecap fraction data does not show the same differences between ST184 and ST191 when~~
606 ~~plotted against the dimensionless Reynolds number, R_H , which describes breaking waves using~~
607 ~~Knorr_11 measurements of significant wave height (Zhao and Toba, 2001). The relationship~~
608 ~~between Stage A whitecap fraction and R_H is more scattered when Stage A whitecaps are below~~
609 ~~$\sim 10^4$ (Figure 4b). Wave development and steepness (slope) influence the likelihood of~~
610 ~~breaking waves. Breaking waves are more closely associated with steep, young waves. At a~~
611 ~~given wind speed and wave height, older, swell dominated waves do not produce as many stage~~
612 ~~A whitecaps compared to young wave systems (Sugihara et al., 2007; Callaghan et al., 2008).~~

613 3.3 Concentrations, fluxes and gas transfer velocities

614 Seawater pCO_2 was consistently lower than the overlying atmosphere throughout the study
615 region due to biological uptake (Figure ~~3a~~e). As a result, the air/sea concentration difference
616 (ΔpCO_2) was large and always into the ocean, with $\Delta\text{pCO}_2 < -45$ ppm for more than 80% of the
617 measurements. Periods with particularly enhanced ΔpCO_2 into the ocean were during the
618 transit between ST181 and ST184 (ΔpCO_2 as large as -120 ppm) and during ST191 (ΔpCO_2
619 consistently -75 ppm).

620 Seawater DMS levels were much higher than atmospheric levels, reflecting the biogenic
621 sources in seawater and the relatively short atmospheric lifetime (~ 1 day; Kloster et al., 2006).
622 The largest air/sea DMS concentration differences (ΔDMS) of 6-12 ppb were observed during
623 DOY 185-190 (Figure ~~4~~2a). The ΔDMS and ΔpCO_2 did not co-vary (Spearman $\rho = 0.11$,
624 $n=918$, $p < 0.001$). This is not surprising because, although seawater DMS and CO_2 signals are
625 both influenced by biological activity, they are controlled by different processes. Seawater
626 CO_2 levels reflect the net result of community photosynthesis and respiration, while DMS

627 production is related to metabolic processes that are highly species-dependent (Stefels et al.,
628 2007).

629 CO₂ fluxes (F_{CO_2}) were generally into the ocean, as expected given the direction of the air/sea
630 concentration difference (Figure 3b+d). The variability in F_{CO_2} observed on this cruise reflects
631 dependence on both wind speed and ΔpCO_2 . For example, during DOY182 air-to-sea CO₂
632 fluxes increase due to a gradual increase in ΔpCO_2 with fairly constant wind speed. More
633 commonly, ΔpCO_2 was fairly constant and variability in F_{CO_2} reflected changes in wind speed.
634 For example, from DOY 185-187 wind speeds gradually declined from ~10 to 5 m s⁻¹ with a
635 concurrent decline in F_{CO_2} . DMS eddy covariance fluxes were always out of the ocean. Ten
636 minute averaged DMS fluxes (F_{DMS}) clearly show the influence of ~~both~~ ΔDMS (e.g. DOY 188)
637 and wind speed (e.g. DOY 184).

638 Gas transfer velocities of CO₂ and DMS from this cruise exhibit two systematic differences: i)
639 k_{DMS} values are generally lower than k_{CO_2} , particularly during episodes of high wind speed;
640 and ii) k_{CO_2} is characterized by much larger scatter than k_{DMS} . We attribute the large scatter in
641 k_{CO_2} to the greater random uncertainty associated with the eddy covariance measurement of
642 air/sea CO₂ fluxes compared to those of DMS. As shown by Miller et al. (2010), the analytical
643 approach used in this study (dried air, closed path LI7500) has sufficient precision to
644 adequately resolve the turbulent fluctuations in atmospheric pCO₂ associated with the surface
645 flux over most of the cruise ($\Delta pCO_2 < -30$ ppm). The scatter in the CO₂ flux measurements is
646 more likely due to environmental variability resulting from fluctuations in boundary layer CO₂
647 mixing ratio arising from horizontal and/or vertical transport unrelated to air/sea flux (Edson
648 et al., 2008; Blomquist et al., 2014). These effects likely have a much smaller effect on air/sea
649 DMS fluxes, because the air/sea DMS concentration difference is always much larger than the
650 mean atmospheric DMS concentration (due to the short atmospheric lifetime of DMS). For
651 example, a ΔpCO_2 of 100 ppm at a wind speed of 10 m s⁻¹ will produce turbulent fluctuations
652 that are ~0.02% of the background CO₂ on average. In contrast, a typical seawater DMS
653 concentration (2.6 nM) at a wind speed of 6 m s⁻¹ generates fluctuations ~~of that are~~ 20% of the
654 background (Table 1; Blomquist et al., 2012). Thus, F_{CO_2} measurements are highly sensitive
655 to small fluctuations in background CO₂ and the relative uncertainty is expected to be much
656 larger than that for F_{DMS} .

657 3.4 Comparison of k_{CO_2} and $k_{DMS,Sc}$

658 The differences between CO_2 and DMS gas transfer velocities observed in the time series are
659 also evident when the data are examined as a function of wind speed. From the 10-minute
660 averaged data, it is clear that k_{CO_2} is greater than k_{DMS} and has a stronger wind speed-
661 dependence over most of the wind speed range (Figure 5a,b). These broad trends are also easily
662 seen in longer time-averaged data. Flux and ΔC measurements were averaged into ~~42~~ hour
663 periods (minimum of 3 flux intervals per ~~24~~ hour period), which reduced the scatter in F_{CO_2}
664 while preserving the temporal variability ([see Figure S3S7 in Supplemental information](#)). Gas
665 transfer velocities were then recalculated from the ~~42~~ hour averaged data. 10-minute k_{CO_2} and
666 $k_{DMS,Sc}$ data were also averaged into 2 m s^{-1} wind speed bins, with a minimum of ~~five5 10ten-~~
667 minute periods per bin. The ~~24~~ hour averaged data and the wind speed binned data show k_{CO_2}
668 and $k_{DMS,Sc}$ diverging at intermediate wind speeds, differing by a factor of roughly two at 10 m
669 s^{-1} (Figure 5c,d).

670 DMS gas transfer velocities on this cruise exhibit complex behaviour at intermediate to high
671 wind speeds, as discussed in Bell et al. (2013). $k_{DMS,Sc}$ increases linearly with wind speed up
672 to $\sim 11 \text{ m s}^{-1}$ (Figure 5). Under the [sustained](#) high wind, high wave conditions encountered
673 during ST191, the wind speed-dependence of $k_{DMS,Sc}$ was lower than expected, with a slope
674 roughly half that of the rest of the cruise data. This effect was not observed at ST184 – [for](#)
675 [detailed discussion, see](#) Bell et al. (2013). Such coherent spatial-temporal variation means that
676 wind speed bin averaging of the higher wind speed $k_{DMS,Sc}$ may mask real variability in the
677 relationship with wind speed. Relationships developed from wind speed bin-averaged gas
678 transfer data should be interpreted with caution, especially when it comes to developing
679 generalizable air/sea gas transfer models.

680 The Knorr_11 k_{CO_2} data also demonstrate a clear wind speed dependence (Figure 5). The
681 NOAA COARE model for CO_2 has been tuned to previous eddy covariance flux measurements
682 (McGillis et al., 2001), with bubble-mediated transfer determining the non-linear relationship
683 with wind speed (Fairall et al., 2011). There is reasonable agreement between the COARE
684 model gas transfer velocity predictions and the Knorr_11 k_{CO_2} data ~~until up to~~ $\sim 11 \text{ m s}^{-1}$ wind
685 speed. Above 11 m s^{-1} , the COARE model over predicts k_{CO_2} . This could be interpreted as
686 indicating high wind speed suppression of gas transfer for CO_2 as observed for DMS (as

687 discussed by Bell et al., 2013). However, it is important to note that the number of high wind
688 speed ($>15 \text{ m s}^{-1}$) gas transfer measurements in this study is limited to 9 hours and 16 hours of
689 data for DMS and CO_2 respectively. Much more data are needed in order to firmly establish
690 the high wind speed behaviour.

691 The COAREG 3.1 model parameterizes interfacial gas transfer by scaling to Sc and friction
692 velocity and estimates bubble-mediated gas transfer following Woolf (1997). The lower
693 solubility of CO_2 leads to enhanced gas transfer relative to that of DMS at high wind speeds
694 where bubble transport is significant (Fairall et al., 2011). There is good agreement between
695 the COAREG model gas transfer velocity predictions and the Knorr_11 k_{CO_2} and k_{DMS} data
696 until $\sim 11 \text{ m s}^{-1}$ wind speed.

697 Earlier in this paper we introduced the quantity Δk_w as an observational measure of the
698 difference in gas transfer velocities of CO_2 and DMS (Section 2.3, ~~e~~Equation 6). The
699 relationship between Δk_w and wind speed is positive and shows no systematic differences
700 related to temporal variability (Figure 6). Sea surface temperature (SST) is indicated by symbol
701 size. Some of the scatter in Figure 6 could be driven by changes in Sc due to SST variability.
702 Nearly all of the data in Figure 6 ~~is-are~~ from periods when SST was relatively constant
703 ($9.78 \pm 1.10^\circ\text{C}$). Many of the k_{CO_2} data with warm seawater (i.e. ST181, $\text{SST} > 12^\circ\text{C}$) were
704 rejected by our quality control criteria (see Appendix A.3). These data were collected when
705 wind speeds were low, which resulted in small CO_2 fluxes with large variability at low
706 frequencies. Of the periods with $\text{SST} > 12^\circ\text{C}$ that passed the quality control criteria, the
707 majority contributed fewer data within a 24 hour averaging period than the minimum threshold
708 (three 10-minute averaged data points). ~~Only one 4 hour period passed the thresholds for flux~~
709 ~~quality control and number of points, and this was associated with the most negative Δk_w value.~~

710 4 Discussion

711 The bubble-mediated component of gas transfer is a strong function of wind speed and breaking
712 waves. Previous estimates of bubble-mediated air/sea gas exchange have ~~been based on used~~
713 ~~data from~~ laboratory experiments (Keeling, 1993; Asher et al., 1996; Woolf, 1997). The
714 differences between gas transfer velocities for DMS and CO_2 provide a unique way to constrain
715 the importance of bubble-mediated transfer under natural conditions. This study shows that
716 Δk_w is near zero ($< 4.5 \text{ cm hr}^{-1}$) at ~~very~~ low wind speeds ($U_{10} \leq 4.5 \text{ m s}^{-1}$), which is consistent

717 with the wind speed at which whitecap fraction becomes significant ($W_T > 10^{-5}$, Figure 23a).
 718 Above 4.5 m s^{-1} , Δk_w increases non-linearly, consistent with an increase in bubble-mediated
 719 CO_2 transfer associated with wave breaking. The relationship between Δk_w and wind speed is
 720 non-linear, and ~~the quadratic power law~~ wind speed-dependence yields a good fit ($R^2 =$
 721 0.6677 ; Figure 6):

$$722 \quad \Delta k_w = 0.1570.177U_{10}^2 - 0.535U_{10}^{1.928} + 4.289 \quad \text{Equation 7}$$

723 The functional form of this relationship is qualitatively consistent with those found between
 724 U_{10} and breaking waves/wave energy dissipation (Melville and Matusov, 2002) and U_{10} vs.
 725 whitecap areal fraction (e.g. Callaghan et al., 2008; Schwendeman and Thomson, 2015).
 726 Bubble-mediated gas transfer is the only viable explanation for the magnitude and wind-speed
 727 dependence of Δk_w . The only alternative explanation would require a large systematic bias in
 728 the measurement of relative gas transfer velocities of DMS and CO_2 . There are no obvious
 729 candidates for such biases.

730 During strong wind/large wave conditions, the Knorr_11 data suggest that bubble-mediated
 731 exchange is a dominant contributor to the total transfer of CO_2 . For example, when wind speeds
 732 were $11\text{-}12 \text{ m s}^{-1}$, Δk_w was about 50% of the total CO_2 gas transfer (k_{CO_2}). A significant
 733 contribution by bubbles to the total gas transfer velocity means that bubble-mediated exchange
 734 must be included and adequately parameterised by gas transfer models. The Schmidt number
 735 (Sc) normalisation (Equation 4) assumes that the gas transfer velocity is purely interfacial. An
 736 alternative normalisation (involving Sc and solubility) is required when bubble-mediated
 737 transfer is significant. Our data suggest that the current Sc normalisation should be applied with
 738 caution to gas transfer data for different solubility gases at wind speeds greater than 10 m s^{-1} .

739 If Δk_w reflects the difference between the bubble-mediated contribution to the transfer of CO_2
 740 and DMS, one would expect Δk_w to correlate with wave-breaking, and hence with the areal
 741 coverage of whitecaps. Breaking waves generate plumes of bubbles (Stage A whitecaps, W_A),
 742 which then rise to the surface and persist for a short period as foam (Stage B whitecaps, W_B).
 743 Almost all whitecap measurements represent the fraction of the sea surface that is covered by
 744 bubble plumes and/or foam i.e. $W_T = W_A + W_B$. Δk_w is positively correlated with both W_T
 745 (Spearman $\rho = 0.6581$, $n=4332$, $p < 0.001$) and W_A (Spearman $\rho = 0.7482$, $n=3226$, $p < 0.001$)
 746 (Figure 7a,b). These correlations are approximately the same strength as the correlation
 747 between Δk_w and wind speed (Spearman $\rho = 0.7383$, $n=8855$, $p < 0.001$). The functional form

748 of the relationship between Δk_w and whitecap areal extent appears to be linear for $W_T > 0.005$.
749 However, the Knorr_11 dataset is small and quite scattered, particularly when $W_T < 0.005$.
750 More data are required to fully test the validity of whitecap areal fraction as a proxy for bubbles
751 and bubble-mediated exchange.

752 Observations of the decaying white cap signal (W_B) suggest that the persistence of surface foam
753 is related to both bubble plume depth (deeper bubble plumes take longer to degas) and sea
754 surface chemistry (Callaghan et al., 2013). As measured here, W_B is approximately an order of
755 magnitude larger than W_A and thus dominates the W_T signal. It is often assumed that gas
756 exchange takes place in bubble plumes formed by active wave breaking (i.e. W_A), while W_B
757 may vary widely due to surfactant concentration with little or no impact upon bubble-mediated
758 gas exchange (e.g. Pereira et al., 2016). In this case, Δk_w should be more strongly correlated
759 with W_A than W_B or W_T . The Knorr_11 data do not suggest that W_A is an improvement upon
760 either W_T or even wind speed as a measure of bubble mediated exchange. This may be because
761 whitecaps do not fully represent the bubbles facilitating gas exchange as these may dissolve
762 before they reach the sea surface. Alternatively, W_T and W_A may be equally good (or poor)
763 proxies for bubbles because: (i) surfactant activity was either insignificant minimal or
764 sufficiently invariant in the study region (despite high biological productivity) ~~such~~ that W_B
765 does not confound the relationship between W_T and W_A ; (ii) W_A is no better than W_T at
766 representing the volume of air entrained by breaking waves; and/or (iii) bubbles residing at the
767 surface (i.e. W_B) continue to contribute to gas transfer (Goddijn-Murphy et al., 2016).

768 As shown earlier, the bubble-mediated contribution to gas transfer (k_{bub,CO_2}) can be obtained
769 from Δk_w using information from mechanistic bubble gas transfer models (f , see Section 2.3).
770 The k_{bub,CO_2} datasets derived from the Knorr_11 data using the Asher (Asher and Wanninkhof,
771 1998; Asher et al., 2002) and Woolf (Woolf, 1997) models differ by about 15% (Figure 8). The
772 field-based estimates of k_{bub,CO_2} can also be compared to model-only estimates for the Knorr_11
773 conditions using the Asher and Woolf models. Both models are based on total whitecap areal
774 fraction, W_T . A non-linear fit of the Knorr_11 W_T and wind speed measurements ($W_T = 1.9 \times 10^{-6} U_{10m}^{3.36}$)
775 was used to drive both models (Figure 8). Asher et al. (2002) is based on laboratory
776 tipping bucket gas evasion experiments (Asher et al., 1996) and the model was then adjusted
777 to represent the flux of CO_2 into the ocean (invasion). Woolf (1997) scaled a single bubble
778 model to the open ocean based on laboratory experiments.

779 Both models significantly underestimate k_{bub,CO_2} at wind speeds below about 11 m s⁻¹. At higher
780 wind speeds, the Asher et al. (2002) model increases rapidly with wind speed to agree **slightly**
781 better with the Knorr_11 data. In contrast, Woolf (1997) consistently underestimates k_{bub,CO_2}
782 at all wind speeds. Both k_{bub,CO_2} models depend on the choice of wind speed-whitecap
783 parameterisation. Using the Schwendeman and Thomson (2015) whitecap parameterisation
784 instead of the Knorr_11 best fit makes some difference to the model output, but not enough to
785 adequately fit to the data (Figure 8). A ‘dense plume model’ was also developed by Woolf et
786 al. (2007) to take account of the interaction of a bubble plume with the interstitial water
787 between bubbles. This model yields estimates of k_{bub,CO_2} that are even lower than the original
788 Woolf (1997) ‘single bubble model’ (data not shown).

789 It is likely that the Knorr_11 cruise data will be compared with estimates of k_{bub,CO_2} derived
790 from future field campaigns, which will be conducted under different environmental
791 conditions. Our k_{bub,CO_2} data is at *in situ* seawater temperature (~10°C) and thus *in situ* CO₂
792 solubility ($\alpha=1.03$) and diffusivity ($Sc=1150$). We use the Asher et al. (2002) and Woolf (1997)
793 bubble models to make estimates of k_{bub,CO_2} normalised to a standard seawater temperature of
794 20°C ($k_{bub,CO_2,20^\circ C}$, where $\alpha=0.78$ and $Sc=666$). The 24 hour averaged Knorr_11 cruise data,
795 including estimates of Δk_w , k_{bub,CO_2} and $k_{bub,CO_2,20^\circ C}$, are provided in Supplemental Table S1.

796 The approach used in this study to estimate Δk_w and k_{bub,CO_2} from the Knorr_11 field data
797 neglects the effect of sea surface skin temperature and CO₂ chemical enhancement. Skin
798 temperature is typically only a few tenths of a degree less than bulk seawater under the
799 conditions encountered in this study (Fairall et al., 1996). The impact upon k_{CO_2} due to skin
800 temperature effects on CO₂ solubility and carbonate speciation is likely on the order of 3%
801 (Woolf et al., 2016). There is a chemical enhancement of the CO₂ flux due to ionization at the
802 sea surface (Hoover and Berkshire, 1969). The effect on k_{CO_2} has been estimated to be up to
803 about 8% at a wind speed of 4-6 m s⁻¹ (Wanninkhof and Knox, 1996), which amounts to a
804 maximum impact of a few cm hr⁻¹. By neglecting these effects we have slightly overestimated
805 Δk_w and k_{bub,CO_2} , but the magnitude of these corrections would be small relative to the
806 environmental scatter or measurement uncertainty.

807 **5 Conclusions**

808 The Knorr_11 concurrent measurements of DMS and CO₂ gas transfer velocities show
809 significant differences in gas transfer between the two gases at intermediate-high wind speeds.
810 These data indicate that: i) bubble-mediated gas transfer becomes significant for CO₂ at or
811 above the threshold for wave-breaking; and ii) the wind speed-dependence is non-linear, with
812 a similar functional form to proposed relationships predicting whitecap areal extent from wind
813 speed. However, existing models of bubble-mediated gas transfer using the Knorr_11 *in situ*
814 observations of whitecap fraction significantly underestimate the importance of this process.

815 There are a number of assumptions behind model estimates of bubble-mediated gas exchange
816 (Goddijn-Murphy et al., 2016). Model bias can be crudely split into: i) uncertainties in the
817 scaling of whitecap fraction to the bubble population (e.g. using Cipriano and Blanchard,
818 1981); and ii) the relationship between gas exchange and bubble properties, which are predicted
819 as a function of air entrainment into the surface ocean by a breaking wave, bubble injection
820 depth, size distribution and mobility through the water (a function of surface cleanliness and
821 surfactants). The underestimation of bubble-mediated CO₂ gas transfer by both models is
822 particularly apparent at low-intermediate wind speeds and low whitecap fraction. This could
823 indicate that either bubble production during microscale breaking is an important process for
824 gas transfer or the relationship between whitecap fraction and the bubble population is poorly
825 constrained.

826 In summary, the approach of using simultaneous measurements of multiple gases with different
827 solubility appears to be a viable way to constrain the magnitude of bubble-mediated gas
828 transfer. Analysis of additional sparingly soluble gases, such as methane or oxygenated
829 hydrocarbons would further strengthen this approach. A much larger data set, under a wider
830 range of oceanographic conditions is certainly needed. In particular, it would be useful to
831 examine DMS and CO₂ gas transfer velocities in ocean regions with different temperatures,
832 where the solubility of each gas is significantly different from this study.

833

834 **Appendix A**

835 **A.1 Seawater CO₂ and DMS measurements**

836 Seawater CO₂ and DMS were monitored in the supply of seawater pumped continuously
837 through the ship from an intake on the bow located 6 m below the sea surface. CO₂ was
838 equilibrated with air in a recirculating showerhead-type system. Alternate air and water side
839 pCO₂ were each measured for 5 min by the same Infrared Gas Analyser (IRGA). Seawater
840 DMS was equilibrated with DMS-free air in a tubular porous membrane equilibrator, operated
841 in a single-pass, counterflow mode. DMS was measured at 1 Hz using chemical ionization
842 mass spectrometry and bin-averaged at 1 minute intervals (UCI miniCIMS; Saltzman et al.,
843 2009). DMS was calibrated by continuously pumping an internal standard of tri-deuterated,
844 DMS (d₃-DMS) into the seawater flow just before the equilibrator. Details of the methods and
845 instrumentation used for equilibration and detection of seawater DMS are described in
846 Saltzman et al. (2009).

847 **A.2 Mast-mounted instrumentation and data acquisition**

848 The eddy covariance ~~setup system~~ was mounted 13.6 m above the sea surface on the bow mast.
849 Platform angular rates and accelerations were measured by two Systron Donner Motion Pak II
850 (MPII) units. Three dimensional winds and sonic temperature were measured by two Campbell
851 CSAT3 sonic anemometers. Air sampling inlets for DMS and CO₂ were located at the same
852 height as the anemometers and within 20 cm of the measurement region. GPS and digital
853 compass output were digitally logged at 1 Hz. Winds were corrected for ship motion and
854 orientation as described in Miller et al. (2008) and Landwehr et al. (2015). The eddy covariance
855 data streams were logged in both analog and digital format as described in Bell et al. (2013)
856 ~~and Miller et al., In Prep.~~

857 **A.3 High frequency atmospheric DMS and CO₂ measurements**

858 Atmospheric DMS measurements were made at 10 Hz using an atmospheric pressure chemical
859 ionisation mass spectrometer located in a lab van (UCI mesoCIMS; Bell et al. (2013)). Air
860 was drawn to the instrument through a 28 m long ½ in OD Teflon tube. A subsample of the
861 air stream was passed through a Nafion drier prior to entering the mass spectrometer. The

862 measurement was calibrated using an internal gas standard of tri-deuterated DMS added to the
863 inlet (see Bell et al., 2013).

864 Atmospheric CO₂ measurements were made on air drawn at 8 L min⁻¹ through a filtered inlet
865 (90 mm diameter with 1 micron pore size, Savillex) near the sonic anemometers on the bow
866 mast, through 5 m of 5.9 mm ID polyethylene-lined Dekabon tubing to two fast-response
867 CO₂/H₂O IRGAs in an enclosure on the bow mast. The IRGAs were open-path style sensors
868 (LI7500, Licor Inc.) converted to a closed-path configuration (see Miller et al., 2010) and were
869 plumbed in series. A Nafion multi-tube membrane drier (PD-200T, PermaPure) with 6 L min⁻¹
870 dry air counter flow was installed between the two IRGAs such that the upstream IRGA
871 sampled undried air and the downstream IRGA sampled the same air after drying. This
872 technique removes 97% of the Webb Correction from the measured CO₂ flux (first shown by
873 Miller et al. (2010) and confirmed by Landwehr et al. (2014)).

874 The air flow through both the CO₂ and DMS inlets was fully turbulent (Re > 10,000). The inlets
875 used in this study introduced a small delay ($\Delta t = 2.2$ s for DMS, $\Delta t = 1.2$ s for CO₂) between
876 measured wind and atmospheric measurements, as well as minor loss of covariance at high
877 frequencies (<5%). The methods used to estimate the delay and loss of flux are given in Bell
878 et al. (2013).

879 Eddy covariance fluxes were computed for DMS and CO₂ as F_{DMS} or $F_{CO_2} = \sigma_{air} \langle w'c' \rangle$

880 where σ_{air} is the dry air density, w' is the fluctuation in vertical winds and c' is the delay-
881 adjusted fluctuation in gas concentration. Average covariance fluxes were processed in 10
882 minute and 9.5 minute intervals for DMS and CO₂, respectively (hereafter referred to as 10
883 minute intervals). Momentum and sensible heat fluxes were also computed for 10 minute
884 intervals (see Bell et al., 2013).

885 Sampling intervals with a mean wind direction relative to the bow of >90° were excluded from
886 the final data set. CO₂ fluxes were also excluded from intervals when either: i) relative wind
887 direction changed excessively (SD > 10°); ii) relative wind speed was low (< 1 m s⁻¹); or iii)
888 ΔCO_2 was low (< |30| ppm). DMS and CO₂ fluxes were quality controlled for excessive low
889 frequency flux as described in the Supplemental information of Bell et al. (2013). These quality
890 control criteria excluded 62% of the intervals for CO₂ and 55% for DMS and significantly
891 reduced the scatter in the data.

892

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904

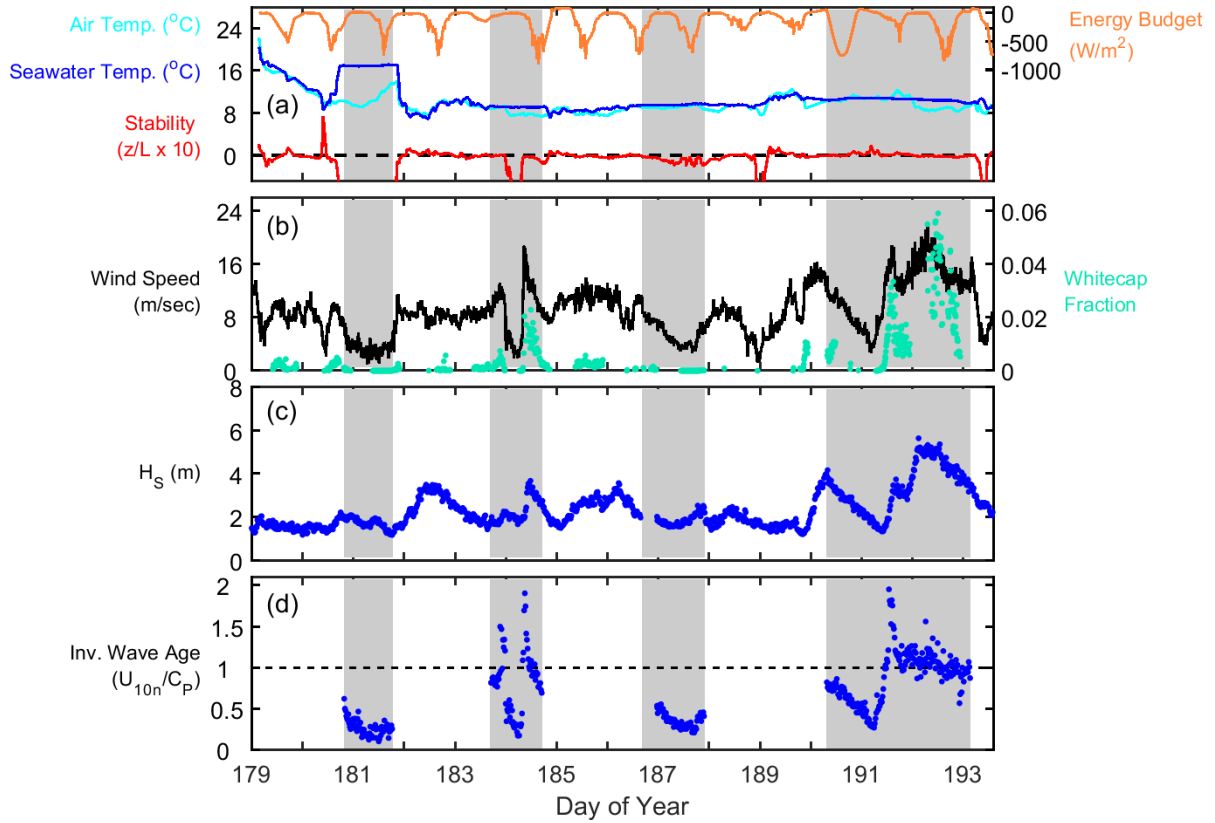
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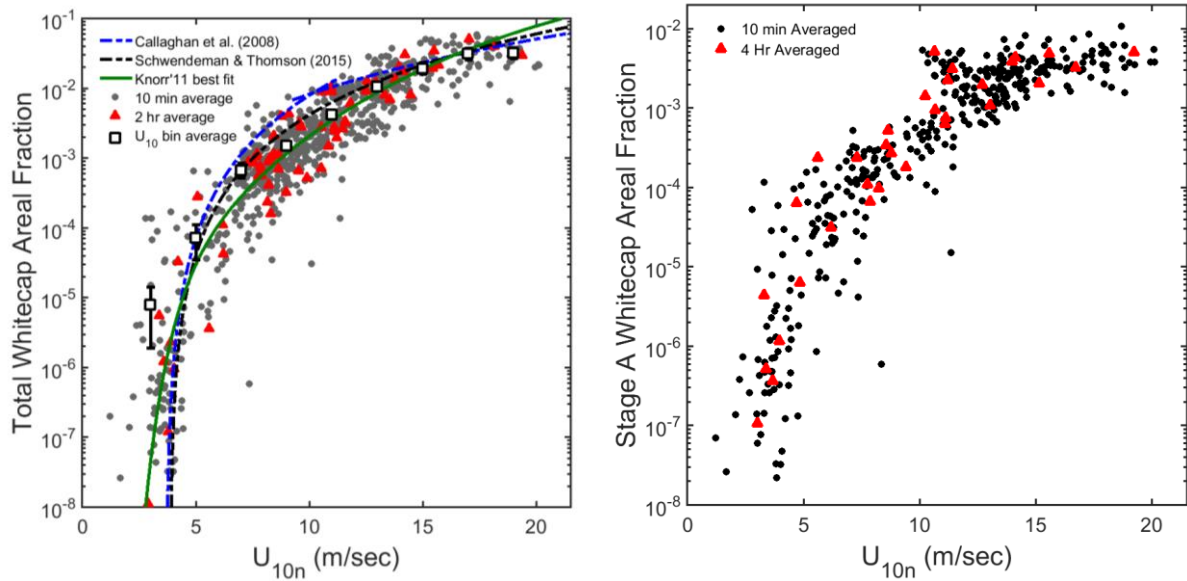
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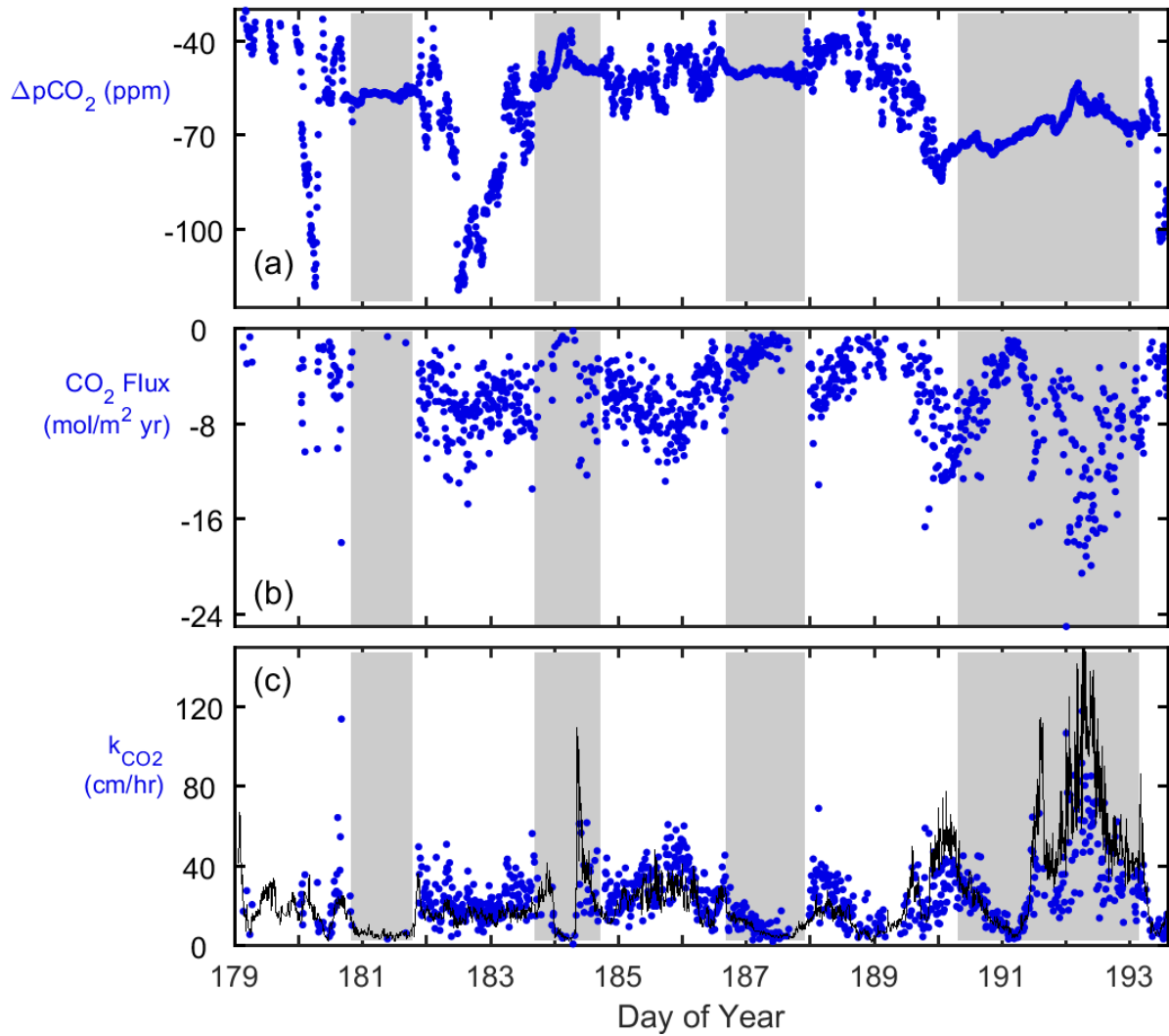
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Figure 1: Time series of ten minute averaged data collected during the Knorr_11 cruise. Dashed black line in panel (a) indicates neutral atmospheric stability ($z/L = 0$). Grey shaded regions represent intervals when the ship occupied stations ST181, ST184, ST187, and ST191. Measured wave properties (see Bell et al., 2013) are presented in panel (c) and (d): Panels (c), (d) and (e) are the significant wave height (H_s , c) and inverse wave age (d). $U_{10m}/C_p \geq 1$ represent younger seas and $U_{10m}/C_p < 1$ represent older seas. $\Delta p\text{CO}_2$, flux (F_{CO_2}) and gas transfer velocity (k_{CO_2}) (water-side only, no Sc correction), respectively. Panel (e) also shows k_{CO_2} calculated using the NOAA-COARE model (black line). Note that negative k_{CO_2} data points in (e) were omitted for clarity (see Supplemental Figure S2 for full data set).



1060 **Figure 2:** Semi-log plots of whitecap areal fraction as a function of mean horizontal wind speed at 10
 1061 m above the sea surface (U_{10n}) during the Knorr 11 cruise. 10 min average (grey dots) and 2 hour
 1062 average (red triangles) data are shown on both panels. Left panel shows total whitecap area (W_T) versus
 1063 U_{10n} bin averaged data (open squares, 2 m s^{-1} bins). The best fit line to Knorr'11 2 hr average data
 1064 (green; $\log_{10}(W_T) = -42.19e^{(-0.95U)} - 6.5e^{(-0.0886U)}$) and wind speed parameterisations from the recent
 1065 literature are shown for reference. Right panel is the whitecap area considered to be solely from wave
 1066 breaking (Stage A whitecaps (W_A), see text for definition).
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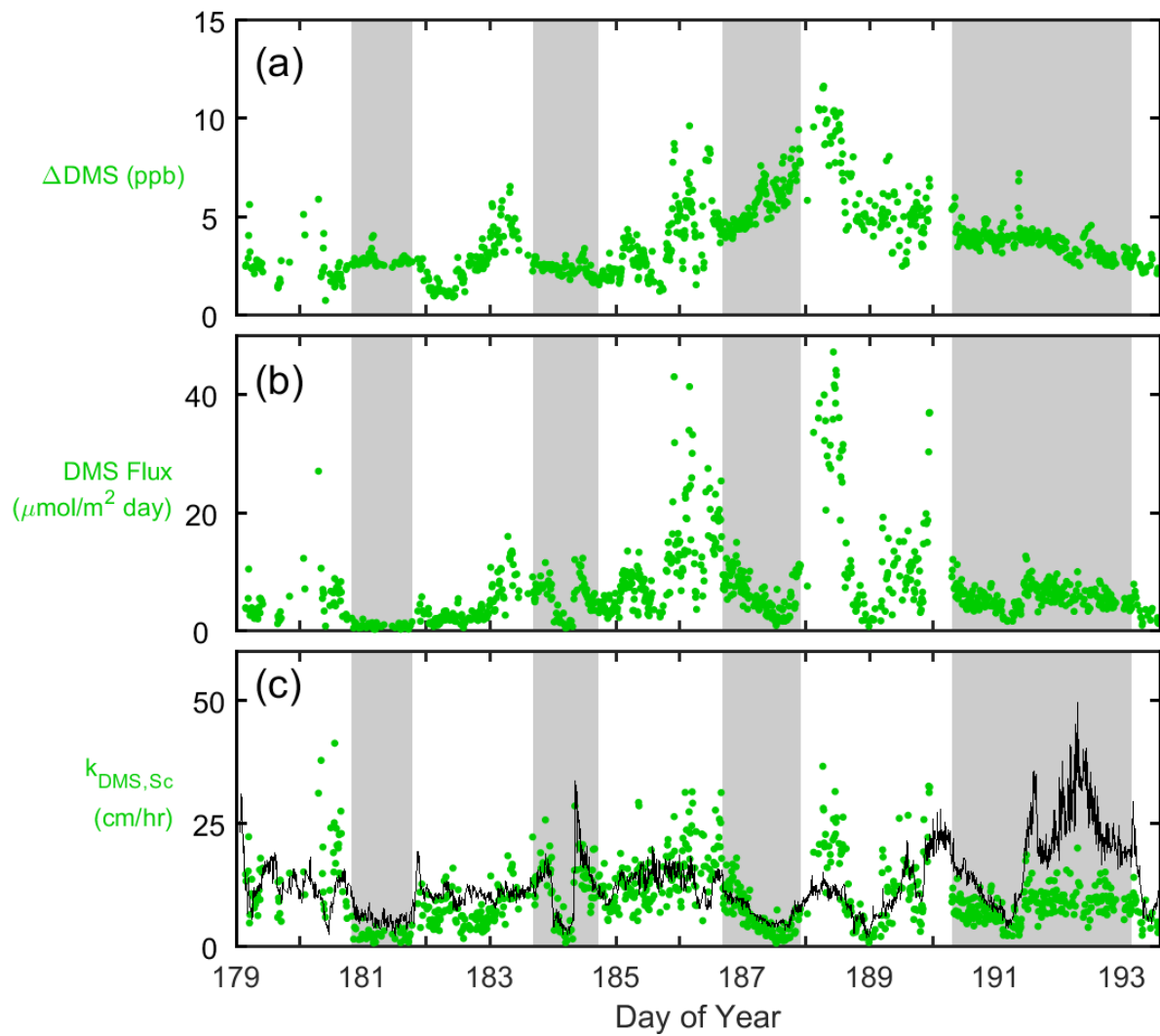
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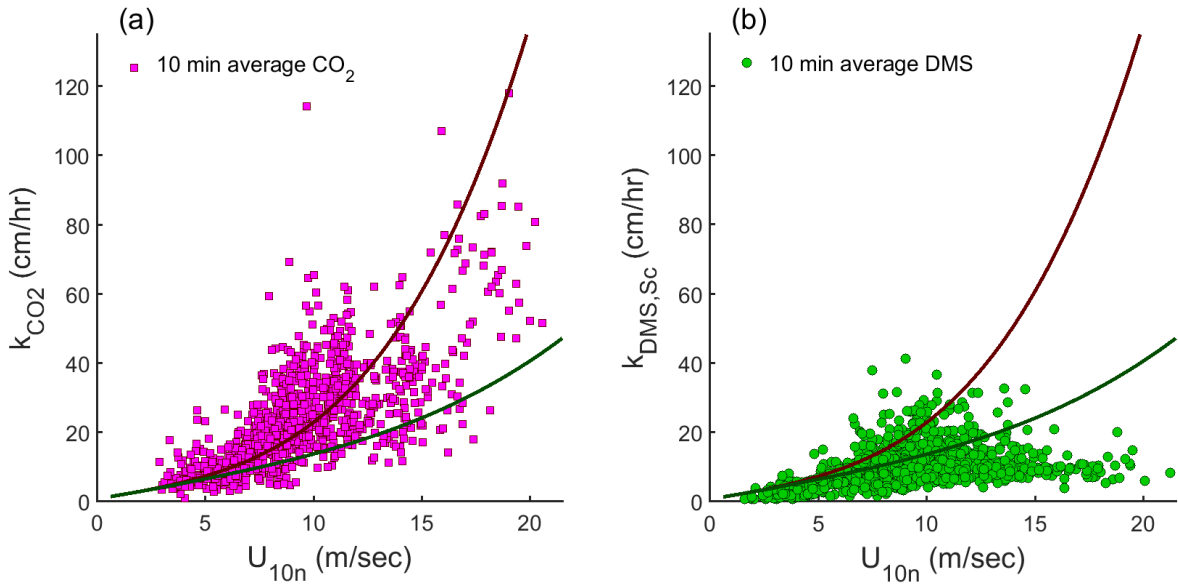
Figure 3: Knorr_11 cruise time series of ten minute averaged CO_2 : (a) air/sea concentration difference ($\Delta p\text{CO}_2$); (b) flux (F_{CO_2}); and (c) gas transfer velocity (k_{CO_2}) (water-side only, no Sc correction). Panel (c) also shows k_{CO_2} calculated using the NOAA COARE model (black line). Note that negative k_{CO_2} data points in (c) were omitted for clarity (see Supplemental Figure S6 for full data set). Grey shaded regions represent periods on station.



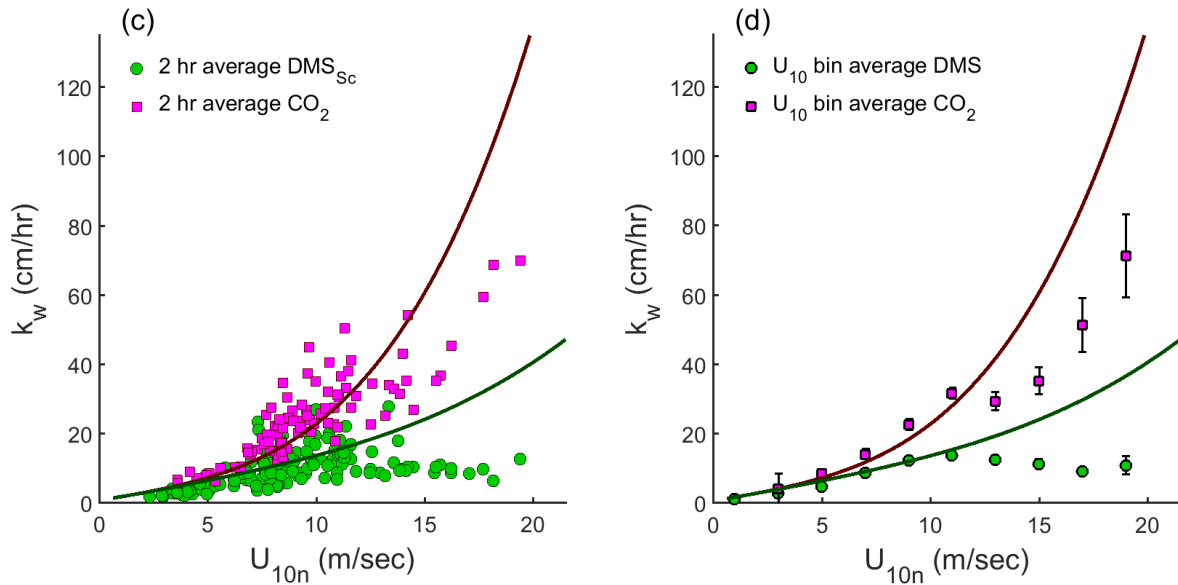
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1076 **Figure 24:** Knorr_11 cruise time series of ten minute averaged DMS: (a) air/sea concentration
 1077 difference (Δ DMS); (b) flux (F_{DMS}); and (c) gas transfer velocity normalised to the *in situ* CO_2 Sc
 1078 number ($k_{DMS,Sc}$). Panel (c) shows $k_{DMS,Sc}$ calculated using NOAA COARE model output (black line).
 1079 Grey shaded regions represent periods on station.
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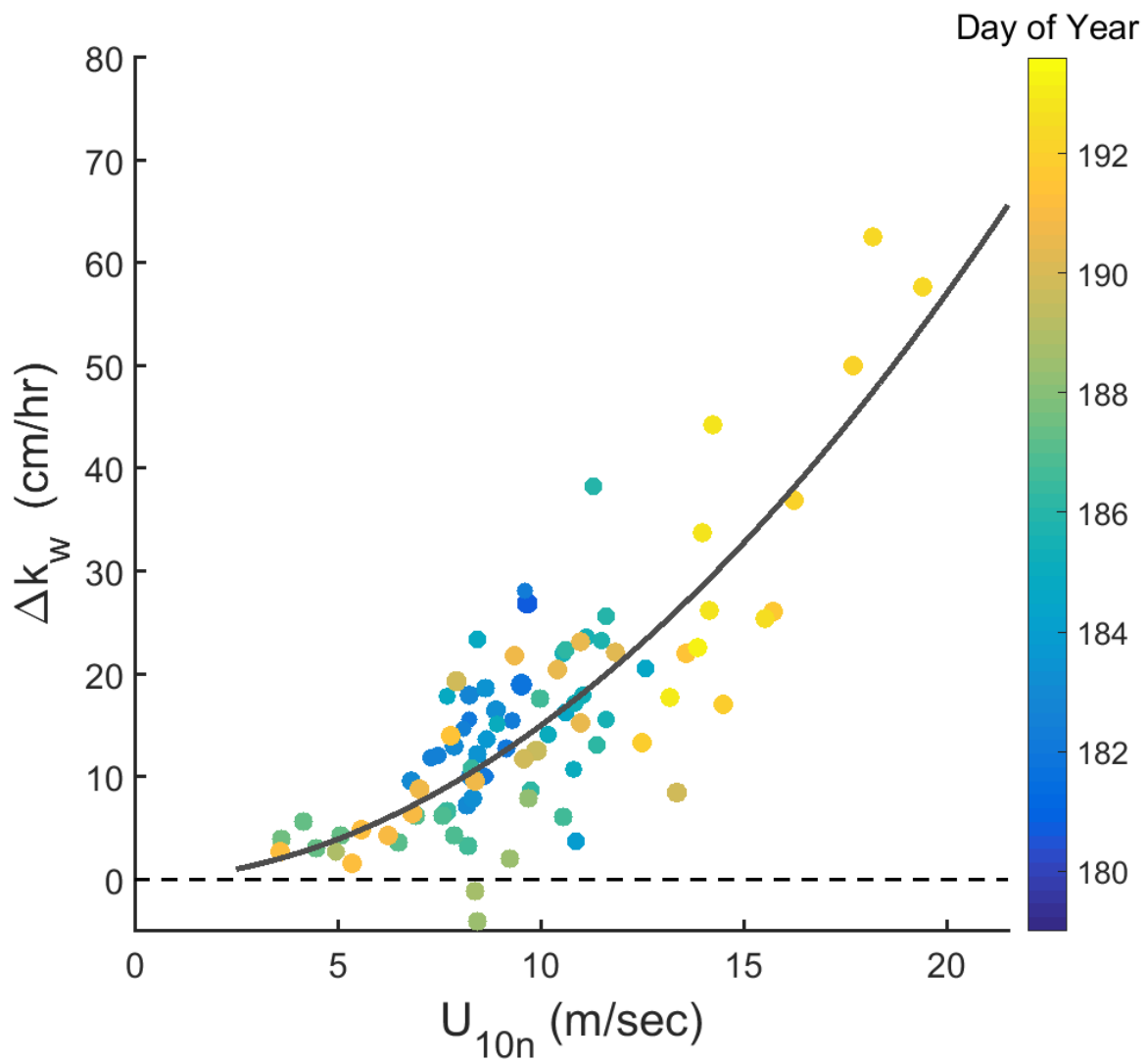
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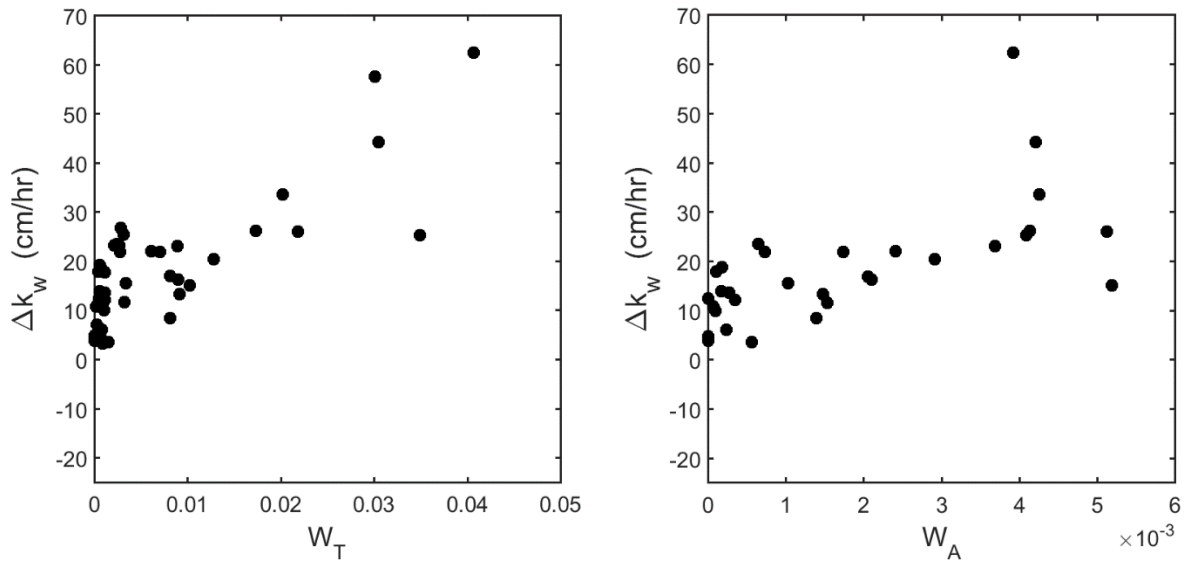
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Figure 5: Gas transfer velocities plotted against mean horizontal wind speed (U_{10}) from the Knorr_11 cruise. Ten minute average data for CO_2 (a) and DMS (b). DMS gas transfer velocities are normalised to the *in situ* CO_2 Sc number. Data are averaged into 24 hour periods (c) and 2 m s^{-1} wind speed bins (d). Note that negative k_{CO_2} data in (a) and (e) have not been plotted for clarity (see Supplemental Figure S4-S8 for full data set). For reference, the NOAA COAREG3.1 model output for CO_2 (magenta line) and DMS (green line) is plotted on all four panels. The COARE model was run with the turbulent/molecular coefficient, $A = 1.6$, and the bubble-mediated coefficient, $B = 1.8$, and used mean Knorr_11 data for the input parameters.



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Figure 6: Difference (Δk_w) between 24 hour average k_{CO_2} and $k_{DMS,sc}$ plotted against U_{10} . Data are coloured by the date of measurement (Day of Year). The solid grey line describes the power law fit to the data (see [text for coefficients Equation 7](#)).



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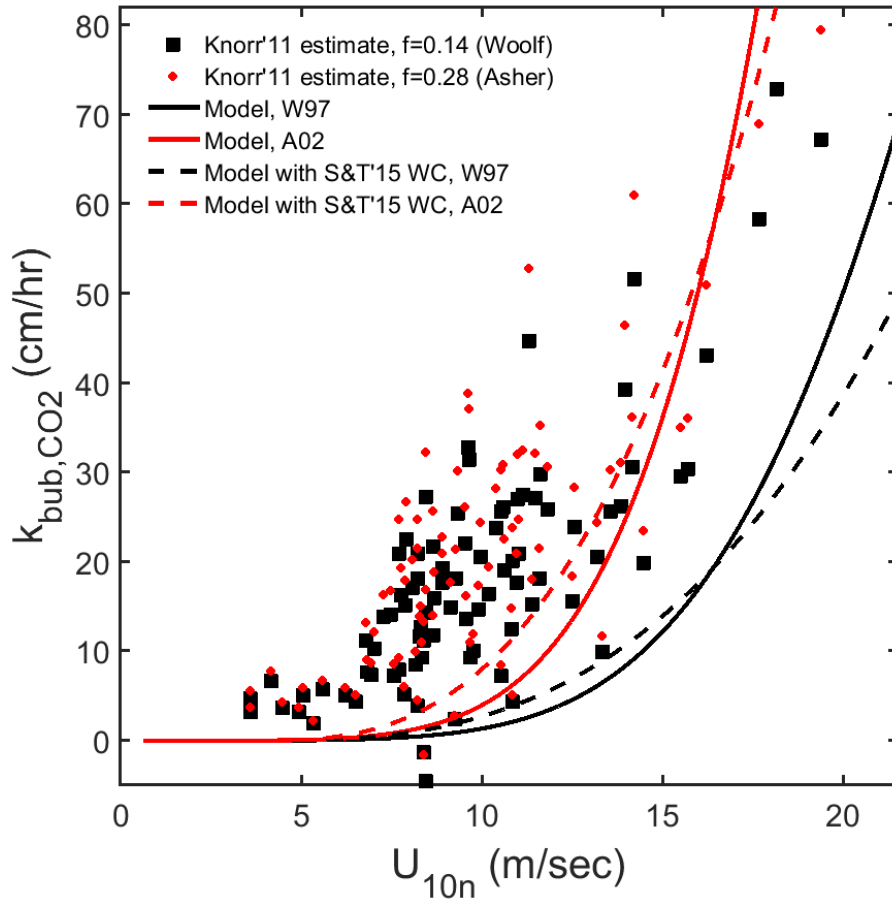
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Figure 7: Knorr_11 Δk_w data plotted against total whitecap areal fraction (left panel) and against Stage A whitecap areal fraction (right panel). Each point is a 24 hour average of coincident measurements of whitecap fraction and DMS and CO₂ gas transfer.



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Figure 8: Bubble-mediated transfer velocity of CO_2 (k_{bub,CO_2}) as a function of wind speed. Individual points are Knorr_11 observations using solubility and diffusivity scaling from Woolf (1997) (black squares) and Asher et al. (2002) (red circles). Continuous lines are model calculations of k_{bub,CO_2} using the Knorr_11 wind speed-whitecap areal fraction relationship (see Figure 2) and mean SST (Woolf (1997), black; Asher et al. (2002), red). Model calculations were also performed using the Schwendeman and Thomson (2015) wind speed-whitecap areal fraction relationship (dashed lines).