

Point by point reply to Referee #1

This paper presents a very interesting and important data set on Nr aerosol concentrations as well as the $\delta^{15}\text{N}$ of total N and NO_3^- in aerosols. The aerosols were collected on ship transects from China to the Northwest Pacific Ocean, and as such are presented as marine aerosols. This is a dataset that should be published and it is important work.

Reply: Thanks for reviewer's appreciation of the merit of our work.

However, given the incredibly strong anthropogenic source strength of Nr in this region, I do not think these are representative of marine aerosols. Rather, this is a presentation of the impact of continental/anthropogenic aerosols on the coastal/near shore marine environment. It is a subtle, but important distinction.

Reply: Thanks for the suggestion. We added statements clearly present that the marine aerosol sampled in ECSs was co-influenced by both sea salt and continental/anthropogenic aerosols (Section 3.1 Paragraph 3). But in the main text, we still use "marine aerosol".

"The much higher water-soluble nitrogen species in the ECSs marine aerosol (compared to that in the NWPO aerosol) indicates that continental/anthropogenic Nr strongly affected the marine aerosol. However, the amounts of sea-salt ions (such as Na^+) in the ECSs aerosols sampled in both 2014 ($123 \pm 98 \text{ nmol m}^{-3}$; Luo et al., 2016) and 2015 ($151 \pm 164 \text{ nmol m}^{-3}$; Luo et al., unpublished data) were higher than those in land aerosol sampled during spring ($23 \pm 7.8 \text{ nmol m}^{-3}$ in Beijing; Zhang et al., 2013), which implies that those aerosols sampled in the ECSs were also significantly influenced by sea salt. Thus, we define the aerosol collected by ship over the ECSs as marine aerosol."

Section 2.2.5: The aerosol extracts do not contain salts, which is typically why SPE is used prior to FT-ICR MS analysis. It is not clear why this procedure was followed and some justification should be provided? It will lead to loss of organic carbon and organic nitrogen, indeed the % recoveries are $< 50\%$ and it does not seem appropriate. FT-ICR MS analysis using negative ion mode means it is not comparable to Altieri et al., or Wozniak et al. In those studies they analyzed samples in the positive ion mode. The negative ion mode would detect organonitrate compounds, whereas the amine and amino-acid compounds are detected in the positive ion mode. The listed elemental ratios based on positive ion mode analysis are not at all applicable in this work. Given the use of SPE and the negative ion mode analysis, the FT-ICRMS analysis needs to be removed from the manuscript, or interpreted in a completely different manner.

Reply: This comment is well taken. We removed all the text about the FT-ICRMS from the manuscript, and the conclusion is not altered.

I agree with page 1 line 24-26 – the anthropogenic signal is so strong that there is no way these aerosols represent a background signal at all. The classification needs to be explained in more detail, and changed to something more appropriate. It seems impossible that aerosols classified as "background" could have

such high concentrations of Nr. It is more likely that all of these aerosol samples are heavily influenced by anthropogenic pollution, with the signal declining as the polluted air is mixed with clean air off shore. Any local marine signal would be swamped by the large continental sources.

Reply: Agree with the reviewer. We defined the term “background” in the Section 2.1 Paragraph 2 prior to Discussion. “Hereafter, we define background aerosol as aerosol not impacted by either dust or sea fog, rather than representing pristine conditions, the background is an environmental baseline collected within the study area during the investigation period.”

The similarity in 2014 and 2015 WSON concentrations is interesting given different chlorophyll fields. It is critical that some basic sampling information such as aerosol size is presented.

Reply: The sampling information has been added into the Section 2.1 Paragraph 1. “Total suspended particulate samples were collected using a high-volume sampler (TE-5170D; Tisch Environmental, Inc.) with Whatman®41 cellulose filters ...”

In addition, the authors need to provide information on field blanks and procedural blanks, especially for WSON and ammonium concentrations.

Reply: Descriptions of the blank were added into the Section 2.2.1 to describe the field blanks and procedural blanks.

“Eight filters of the same type as those used to collect samples were taken as blanks. All blank filters and aerosol samples were stored at $-20\text{ }^{\circ}\text{C}$ during the sampling periods and underwent the same extraction procedure. The NO_3^- , NH_4^+ and WSON content of the blank filters comprised less than 1%, 4% and 9%, respectively, of the average concentration of the corresponding N species in the aerosol samples.”

In Section 2.2.3, please explain the recoveries of WSTN and TDN? How does $n=6$ if there were 44 and 39 aerosol samples analyzed?

Reply: The recoveries of WSTN and TDN represent the oxidation efficiency of prepared solution of N-containing organic and inorganic compounds standards (glycine, urea, ethylene diamine tetraacetic acid and ammonium sulfate) by using the alkaline potassium persulfate.

The following sentences had been added into the Section 2.2.3.

“To verify the WSTN and TDN oxidation efficiency, N-containing organic and inorganic compound standards (specifically, glycine, urea, ethylene diamine tetraacetic acid, and ammonium sulphate) were prepared in solution at a concentration of $800\text{ }\mu\text{M-N}$ for oxidation analysis. The recoveries of the N-containing compound standards under oxidation by alkaline potassium persulfate were within 95 ~ 105%

(n = 6).”

What dry deposition velocities are used? There are large uncertainties associated with dry deposition velocities, which are size specific. If the aerosols are not size segregated, how can you apply a size-specific dry deposition velocity? These are going to be highly uncertain estimates and should be treated as such.

Reply: We agree. The dry deposition velocity varies by more than 3 orders of magnitude with particle size ranging from 0.1 to 100 μm (Hoppel et al., 2002). Thus, it is really hard to accurately estimate the Nr dry deposition by using TSP sample.

In general, ammonium appears in submicron mode from 0.1 to 1 μm , with a small fraction residing in the coarser mode, by contrast, nitrate is distributed mainly in supermicron size ranging from 1 to 10 μm while WSON appears in a wide size spectrum. Thus, for any water-soluble nitrogen species, using a fixed deposition velocity to calculate the dry deposition might cause under- or over- estimation. In our observation, wind speed ranged from 0.8 to 18 m s^{-1} under wide RH ranging from 40 to 100%. Thus, it is not possible to provide variable dry deposition velocities for estimation Nr dry deposition under a wide range of environmental conditions; assumptions were made based on existing knowledge. Here, deposition velocity of 2 cm s^{-1} was applied for nitrate, 0.1 cm s^{-1} for ammonium and 1.0 cm s^{-1} for WSON. We have detailed the issue of size and deposition velocity of nitrate, ammonium and WSON in our previous ACP paper (Luo et al., 2016). Following the reviewer's suggestion, the descriptions of deposition velocities of water-soluble nitrogen species were briefed in the Section 2.3.

“The deposition velocities of water-soluble nitrogen species used herein were 2 cm s^{-1} for nitrate, 0.1 cm s^{-1} for ammonium, and 1.0 cm s^{-1} for WSON, which were consistent with our previous studies (Luo et al., 2016).”

Page 7 line 199-201: It's not clear what size the aerosols are, and so this comparison is difficult to understand. Regardless, these concentrations are incredibly high, and indicative of strong pollution sources.

Reply: Except the case for the Xi'an city, which was PM_{10} , all other aerosols were collected comparably by TSP. We explicitly described the difference in the revised version in the Section 3.1 Paragraph 2.

If the authors want to claim that any of these aerosols are “background” aerosols, they need to find other background sites that have such high Nr concentrations.

Reply: Replied above.

Also, the data are very difficult to see on the log scale, it's a very large range of concentrations to present on one figure. This should be separated somehow.

Reply: Thanks for this suggestion. In this version, we added one more figure (i.e., y-axis is presented in liner scale) into the supplementary material (Fig. S4).

Page 7 line222, you can't tell the difference between 2014 and 2015 in the figures due to the log scale. A different way of presenting the data would help.

Reply: The y-axis now is in liner scale (attached). We also added descriptions of statistical significance ($p < 0.05$ for all cases) into the figure 3 caption.

I agree with the authors conclusion on page 8 paragraph 2 – the “background” aerosols should be re-labeled as it is very misleading.

Reply: We added the definition of “background” in the Section 2.1 Paragraph 2.

Page 8 paragraph 3 – it is important to note that the surface seawater DON is the most likely source of primary WSON aerosols, but secondary WSON aerosols have many other sources, including e.g., surface ocean VOC emissions that go on to oxidize in the atmosphere and form secondary N-containing SOA.

Reply: Thanks for the suggestion. We added description of secondary WSON aerosols into the Section 3.1 Paragraph 6.

“However, the sources of marine aerosol WSON are complex mixture which composed of primary marine organic N and secondary N-containing organic aerosol. Biogenic organic material in SSW can be injected into the atmosphere to form an ice cloud via bubble bursting at the atmosphere-ocean interface (Wilson et al., 2015), this is probably the primary WSON aerosol source. Volatile organic compounds emitted from the surface ocean can react with NO_x and NH_x in the atmosphere to form secondary N-containing organic aerosol (Fischer et al., 2014; Liu et al., 2015).”

Section 3.2 should be removed or reinterpreted given the focus on negative mode CHON compounds identified here.

Reply: The part of FT-ICRMS has been removed.

Page 9 section 3.3 second paragraph. There is a large difference in WSTN _15N from 2014 to 2015 in these aerosols. This should be discussed. Figure 6 and discussion thereof: This is not a valid approach to

understanding what is driving the $\delta^{15}\text{N}$ of WSTN. A cross-plot of $\delta^{15}\text{N-NO}_3^-$ vs $\delta^{15}\text{N-WSTN}$ would provide more information on the influence of nitrate on the total N isotopic composition. Looking at figure 5, it looks like nitrate $\delta^{15}\text{N}$ is a main control on the $\delta^{15}\text{N-WSTN}$. The lack of correlation between $\delta^{15}\text{N WSTN}$ and the relative concentration of NO_3^- is not useful. The relationship between the $\delta^{15}\text{N RN}$ and the relative concentrations of NH_4^+ would be useful, but is not presented.

Reply: Thanks for the suggestions. The scatter plots of $\delta^{15}\text{N-WSTN}$ vs. $\delta^{15}\text{N-NO}_3^-$, $\delta^{15}\text{N-WSTN}$ vs. NO_3^- , $\delta^{15}\text{N-RN}$ vs. NH_4^+ and $\delta^{15}\text{N-RN}$ vs. NH_4^+/RN have been added as Figure S7 and Figure S8 in the new version, and we redraw the Figure 5 and we rewrite the Section 3.2 Paragraph 1-3.

Page 10 paragraph line 300, it is also a possibility that the aerosol WSON is secondary organic aerosol, which may have had its $\delta^{15}\text{N}$ altered by transport or chemical reactions. This is a very over-simplified approach to the interpretation of the $\delta^{15}\text{N-WSON}$ data.

Reply: We agree with the reviewer, however, there are limited studies of $\delta^{15}\text{N}$ about the marine aerosol WSON. The multiple sources of marine aerosol WSON and $\delta^{15}\text{N}$ -WSON fractionation during the processes of secondary N-containing organic aerosol formation are not clear to date. We discussed the possible causes to modulate $\delta^{15}\text{N-WSON}$ in the last paragraph in Section 3.2 and highlighted more studies in future are needed for the secondary marine N-containing organic aerosol, particularly from the $\delta^{15}\text{N}$ scope.

“these high $\delta^{15}\text{N-RN}$ values may be attributable to $\delta^{15}\text{N}$ fractionation and $\delta^{15}\text{N}$ enrichment in the WSON during processes such as secondary N-containing organic aerosol formation by the reaction of NH_x or NO_x with organic aerosol (Fischer et al., 2014; Liu et al., 2015), complex atmospheric chemical reactions (i.e. the photolysis of organic nitrogen into ammonium; Paulot et al., 2015), aerosol WSON aging process, and in-cloud scavenging (Altieri et al., 2016). More studies are needed to explore nitrogen transformation processes, especially those focusing on secondary N-containing organic aerosol in the atmosphere from an isotopic perspective.”

Section 3.4 is too speculative given the limited information presented. Are the ammonium, nitrate, and WSON concentrations statistically different from 2014 to 2015 and between the three classifications? Is there a statistically significant relationship between the $\delta^{15}\text{N}$ of DON in seawater or $\delta^{15}\text{N NO}_3^-$ in seawater and the $\delta^{15}\text{N}$ of TN, NO_3^- , or RN in the aerosols?

Reply: Thanks for the suggestion. The scatter plot of $\delta^{15}\text{N-DON}$ in SSW vs. $\delta^{15}\text{N}$ of aerosol NO_3^- , WSTN and RN in 2015 cruise are attached below. There were no significant relationship in scatter plots of the $\delta^{15}\text{N-DON}$ in SSW against the aerosol $\delta^{15}\text{N-NO}_3^-$, RN and WSTN sampled correspondingly in time and space. According to the reviewer's suggestion, we rewrite the Section 3.3 Paragraph 2.

Table 1. How are aerosol concentrations volume weighted? Is this a mass weighted average?

Reply: The volume weighted mean (C) calculated by the following equation:

$$C = \frac{\sum_{i=1}^n C_i V_i}{\sum_{i=1}^n V_i}$$

where C_i is the concentration of water-soluble nitrogen species in aerosol, V_i is the sampling volume for an aerosol, n is the number of sample.

Figure 1. It is not clear what “regional wind streamlines” are or where they came from. The blue on the background of the figure makes it difficult to see the symbols.

Reply: We enlarged streamlines and modified background color accordingly in the Figure 1.

Figure 3. The caption says a is nitrate and b is ammonium, but they are labeled in the opposite manner.

Reply: Corrected.

Figure 4 should be removed.

Reply: Removed.

Figure 6. The caption says a is concentration and b is $_{-15}N$ but the plots are the opposite.

Reply: Corrected.

Abstract: Line 14 insert “of” between transport and anthropogenic, line 15 “continents may exert a profound impact”, line 16 should read “surface ocean” instead of “marine biogenic”,

Reply: Modified as suggested.

line 18 do the authors mean in the open ocean or do they mean in the atmosphere?,

Reply: Modified as suggested.

Line 26 are the concentrations statistically higher in 2014? If so this should be presented in the text.

Reply: We presented the statistical significance ($p < 0.05$ for all cases) between 2014 and 2015 in this version.

Introduction: Define SSW on first use.

Reply: Defined in the Introduction.

Page 3 paragraph 1 should clearly state that they are referring to primary WSON aerosols.

Reply: Thanks. Primary WSON aerosol has been clearly stated in the new version.

Page 8 second paragraph. It’s not clear what is meant by “atmospheric diffusion”

Reply: “Atmospheric diffusion” has been changed into “atmospheric long-range transport”

Altieri, K. E., Fawcett, S. E., Peters, A. J., Sigman, D. M., and Hastings, M. G.: Marine biogenic source of atmospheric organic nitrogen in the subtropical North Atlantic, Proceedings of the National

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- Luo, L., Yao, X. H., Gao, H. W., Hsu, S. C., Li, J. W., and Kao, S. J.: Nitrogen speciation in various types of aerosols in spring over the northwestern Pacific Ocean. *Atmospheric Chemistry and Physics*, 16, 325-341, doi:10.5194/acp-16-325-2016, 2016.
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- Zhang, R., Jing, J., Tao, J., Hsu, S.-C., Wang, G., Cao, J., Lee, C. S. L., Zhu, L., Chen, Z., Zhao, Y., and Shen, Z.: Chemical characterization and source apportionment of PM_{2.5} in Beijing: seasonal perspective, *Atmos. Chem. Phys.*, 13, 7053–7074, doi:10.5194/acp-13-7053-2013, 2013.

Point by point reply to Referee#2

This manuscript describes ship-board measurements of marine aerosols collected during two cruises around the East China Sea and the northwestern Pacific Ocean in 2014 and 2015. In this manuscript, authors reported concentrations of water-soluble total nitrogen (WSTN), water-soluble organic nitrogen (WSO_N), nitrate (NO₃⁻) and ammonium (NH₄⁺), and values of $\delta^{15}\text{N}$ -WSTN and $\delta^{15}\text{N}$ -NO₃⁻ in aerosols as well as dissolved organic nitrogen (DON) and NO₃⁻ concentrations and $\delta^{15}\text{N}$ -DON and $\delta^{15}\text{N}$ -NO₃⁻ values in sea surface water, which provide good indications where future studies can understand possible sources of atmospheric WSO_N and air-sea exchange of N species. I believe that the contents, including data, of the manuscript should be eventually published because of scarcity of atmospheric WSO_N observation and its significance in biogeochemical N cycle.

Reply: Thanks for reviewer's appreciation of our data and the scientific significance.

I recommend publication of the manuscript after a major revision and the improvement of English.

Reply: We paid for editing service.

1. The title of this manuscript is "Sources of reactive nitrogen in marine aerosol over the Northwest Pacific Ocean in spring"; however, the authors mainly described the spatial distributions and concentrations of atmospheric reactive N species and potential sources of WSO_N. I recommend the authors to describe in their manuscript the sources of atmospheric inorganic N species, although they are relatively well-known compared to that of WSO_N.

Reply: Thanks for suggestion. We added more discussions of the sources of atmospheric inorganic N into the Section 3.1 Paragraph 5.

2. (Page 4, line 98-116) It is not clear that how many, what kind of aerosol samplers and filters were used during the cruises, and how avoided contamination from ship's exhaust. Was a impactor used to separate PM_{2.5} and PM₁₀? More detail information on sampling method should be described in the manuscript, although the authors referred to Luo et al. (2016).

Reply: Aerosol sampling information including the instrument model, company, aerosol type and sampling filter have been added into the Section 2.1.

No separation for PM_{2.5} and PM₁₀.

Follow the reviewer's suggestion, the following explanation added into the Section 2.1. "To avoid self-contamination from the research vessel, the TSP sampler was installed on the top of the tower at the ship head, and aerosols were sampled only during travel. More information about self-contamination from ship exhaust can be found in Luo et al. (2016)."

In general, a pre-combusted glass fiber or a quartz filter is used for determination of WSON. If the authors used the same method for aerosol sampling described in Luo et al. (2016), the authors should explain about the treatment of aerosol filter samples and field blank concentrations and blank correction, because Luo et al. (2016) used a Whatman 41 cellulose filter.

Reply: Descriptions of the blank were added into the Section 2.2.1 to describe the field blanks and procedural blanks.

“ Eight filters of the same type as those used to collect samples were taken as blanks. All blank filters and aerosol samples were stored at $-20\text{ }^{\circ}\text{C}$ during the sampling periods and underwent the same extraction procedure. The NO_3^- , NH_4^+ and WSON content of the blank filters comprised less than 1%, 4% and 9%, respectively, of the average concentration of the corresponding N species in the aerosol samples.”

For determination of DON in sea surface water, the authors mentioned that a $0.2\text{ }\mu\text{m}$ filter was used to remove particulate matters in sea surface water. Usually, a pre-combusted GF/F filter is used to remove particulate matter and minimize the influence of organic matter from the filters on DON concentration in seawater. Please update that what kind of filter was used for filtration of seawater sample. I am also wondering if any consensus reference material (CRM, e.g., deep Florida Strait water from Hansell lab, University of Miami) was used during DON measurement to check the accuracy of analysis.

Reply: The filter information has been added into the Section 2.1.

The measured accuracy verified by our laboratory standard rather than other reference material, and the oxidation efficiency also has been added into the Section 2.2.3.

(Page 6, line 188-Page 7, line 191) Dry deposition velocity. It is unclear if marine aerosols are segregated into PM_{2.5} and PM₁₀ during the aerosol sampling as mentioned in question 2. Although size distributions of atmospheric N species can vary on meteorological conditions, it is known that, in the marine atmosphere, both atmospheric NH_4^+ and WSON primarily exist on fine mode aerosols, whereas atmospheric NO_3^- is predominantly associated with coarse mode aerosols (e.g., Nakamura et al., 2006). I recommend the authors to describe more detail that what dry deposition velocity was used for each N species.

Reply: Thanks for the suggestion. The detail deposition velocity has been added into the Section 2.3.

“ The deposition velocities of water-soluble nitrogen species used herein were 2 cm s^{-1} for nitrate, 0.1 cm s^{-1} for ammonium, and 1.0 cm s^{-1} for WSON, which were consistent with our previous studies (Luo et al., 2016).”

4. (Page 8, line 228-232) The authors compared their NH_4^+ and NO_3^- concentrations with those by Miyazaki et al. (2011) to explain why higher concentrations of inorganic N species were observed during the period of this study (spring). The authors mentioned that the study by Miyazaki et al. (2011) was carried out over the same regions. I doubt about it. The cruise by Miyazaki et al. (2011) was conducted from 44°N to 10°N along 155°E , which covers the subarctic to subtropical northwestern Pacific region.

Although the study by Miyazaki et al. (2011) was carried out in summer, different sampling season is not the only reason why the authors observed high inorganic N species during their study period.

Reply: Agree. Since there were no more data of NH_4^+ and NO_3^- in aerosol sampled on the same season and adjacent area, we can only compare with aerosol collected cover the western North Pacific Ocean. Base on the statistical significance ($p < 0.05$ for all cases), and we rewrite the Section 3.1 Paragraph 5 to discuss the variations of inorganic N in marine aerosol over the NWPO.

5. (Page 8, line 233-242) The authors described that “Likely the source of WSON in background aerosol did not share the same source with NH_4^+ and NO_3^- ” (line 234-235), as if DON in sea surface water is the only source of atmospheric WSON in the open ocean. What is the grounds for this? Because high atmospheric WSON and inorganic N species concentrations were observed in the East China Sea and inorganic N was also detected in the open ocean, the long range transport of anthropogenic WSON to the open ocean should be considered.

Reply: Thanks for suggestion. We agree with that DON in sea surface water is not the only source of atmospheric WSON in the open ocean and aerosol WSON collected in the NWPO also influenced by the anthropogenic emission. Thus, we rewrite this sentence to make it read clearly, and the anthropogenic WSON to the open ocean also added in the Section 3.1 Paragraph 6.

6. (Page 8, line 252-Page 9, line 263) The results of characteristics of CHON molecular compounds shows that 13%, 3% and 19% of marine aerosols collected in the East China Sea, northwestern Pacific Ocean during dust period and northwestern Pacific Ocean during non dust period, were derived from biological sources, respectively. Does this mean that 87%, 97%, and 81% of marine aerosols collected in the same regions were affected by anthropogenic sources? It seems like the contribution of biogenic sources to atmospheric WSON is still low in the open ocean. What is the contribution of biologically-derived atmospheric WSON in the other oceanic regions?

Reply: Part of FT-ICRMS has been removed, the conclusion is not altered.

7. (Page 11, line 331-Page 12, line 355) The authors described that atmospheric reactive N dry deposition flux can account for 14%-58% of the low $\delta^{15}\text{N}-\text{NO}_3^-$ in the northwestern Pacific Ocean during the spring. It is surprising to me that atmospheric reactive N deposition has a significant influence on $\delta^{15}\text{N}-\text{NO}_3^-$ values. My question is that dry deposition of atmospheric reactive N is strong enough to affect or change $\delta^{15}\text{N}-\text{NO}_3^-$ values below the thermocline in the northwestern Pacific Ocean? What is the depth of thermocline in the northwestern Pacific Ocean in the spring season? I recommend the authors to estimate the contribution of atmospheric reactive N dry deposition to primary production in their study area. I think most primary production in the East China Sea and northwestern Pacific Ocean is controlled by nutrients in seawater, which implies that main factor for controlling $\delta^{15}\text{N}-\text{NO}_3^-$ values in the ocean is marine N cycle.

Reply: Thanks for the suggestion. We rewrite the Section 3.3 Paragraph 2.

1. (Page 5, line 136-137) How did the authors obtain the recovery efficiency (i.e.,95-105% (n = 6)) of WSTN and TDN?

Reply: The recoveries of WSTN and TDN are the oxidation efficiency of prepared solution of N-containing organic and inorganic compounds standards (glycine, urea, ethylene diamine tetraacetic acid and ammonium sulfate) by the alkaline potassium persulfate.

The following sentences had been added into the Section 2.2.3.

“To verify the WSTN and TDN oxidation efficiency, N-containing organic and inorganic compound standards (specifically, glycine, urea, ethylene diamine tetraacetic acid, and ammonium sulphate) were prepared in solution at a concentration of 800 $\mu\text{M-N}$ for oxidation analysis. The recoveries of the N-containing compound standards under oxidation by alkaline potassium persulfate were within 95 ~ 105% (n = 6)”

2. (Page 5, 155-156) The authors mentioned that the extraction efficiency on a carbon basis was on average $46 \pm 24\%$ (n = 44). Does it mean that 64% of organic compounds in the extract was not identified?

Reply: This part has been removed.

3. (Page 6, line 159-163) The uncertainty of WSON estimated from propagating errors of WSTN, NO_3^- and NH_4^+ should be added.

Reply: The errors propagation has been added into the Section 2.3.

“The standard errors propagated through the WSON calculation for the 2014 data can be found in Luo et al. (2016). For 2015, the standard errors propagated through WSON calculation varied from sample to sample from 7 to 210%; the average standard error of all samples was 33%.”

(Page 6, line 175) The authors mentioned that $[\text{NH}_4^+]$ in sea surface water typically less than $0.05 \mu\text{mol L}^{-1}$. Is this a common condition in the East China Sea and the Northwestern Pacific Ocean during the sampling period (i.e., spring season)? Sea surface $[\text{NH}_4^+]$ can vary depending on sampling season and locations.

Reply: Reviewer is right. $[\text{NH}_4^+]$ in sea surface water varies depending on sampling season and locations. However, NH_4^+ is much less than DON in this cruise agreeing with common sense for open oceans due to high bio-affinity of NH_4^+ .

In this version, we eliminate “typically” in old statement to avoid confusion. The revised statement is “Since the average $[\text{NH}_4^+]$ in SSW at the selected sites during the 2015 cruise (12 sites and 23 samples) was $0.05 \mu\text{M}$, which is much less than DON in μM level...”

List of all the relevant changes made in the manuscript

- 1, Aerosol sampling information including the instrument model, company, aerosol type and sampling filter have been added into the Section 2.1 Paragraph 1.
- 2, The following sentences were added into the Section 2.1 Paragraph 1. “ To avoid self-contamination from the research vessel, the TSP sampler was installed on the top of the tower at the ship head, and aerosols were sampled only during travel. More information about self-contamination from ship exhaust can be found in Luo et al. (2016).”
- 3, We define the term “background” in the Section 2.1 Paragraph 2.
- 4, Descriptions of the blank were added into the Section 2.2.1 to describe the field blanks and procedural blanks.
- 5, The “oxidation efficiency” also has been added into the Section 2.2.3.
- 6, The errors propagation of WSON calculation has been added into the Section 2.3 Paragraph 1.
- 7, We redescribe the $[\text{NH}_4^+]$ in sea surface water in the Section 2.3 Paragraph 3.
- 8, The detail deposition velocity has been added into the Section 2.3 Paragraph 4.
- 9, We explicitly describe the difference of aerosol size in the Section 3.1 Paragraph 2.
- 10, We add statements clearly present that the marine aerosol sampled in ECSs was co-influenced by both sea salt and continental/anthropogenic aerosols in the Section 3.1 Paragraph 3.
- 11, We rewrite the Section 3.1 Paragraph 5.
- 12, We rewrite the Section 3.1 Paragraph 6.
- 13, Part of FT-ICRMS has been removed from the revised manuscript.
- 14, We rewrite the Section 3.2 Paragraph 1-3.
- 15, We rewrite the Section 3.2 Paragraph 6.
- 16, We rewrite the Section 3.3 Paragraph 2.
- 17, We modified the Figure 1.
- 18, We modified the Figure 3.
- 19, We redraw the Figure 5.
- 20, Figure S4, Figure S7, Figure S8 were added into the supplement.
- 21, Other minor changes have been done.
- 22, we paid for editing service.

Sources of reactive nitrogen in marine aerosol over the Northwest Pacific Ocean in spring

Li Luo^{1,3}, Shuh-Ji Kao^{2*}, Hongyan Bao², Huayun Xiao^{1,3}, Hongwei Xiao^{1,3}, Xiaohong Yao⁴, Huiwang Gao⁴, Jiawei Li⁵, Yangyang Lu²

5 ¹Jiangxi Province Key Laboratory of the causes and control of Atmospheric pollution, East China University of Technology, Nanchang 330013, China

²State Key Laboratory of Marine Environmental Science, Xiamen University, Xiamen 361102, China

³School of Water Resources and Environmental Engineering, East China University of Technology, Nanchang 330013, China

10 ⁴Key laboratory of Marine Environmental Science and Ecology, Ministry of Education, Ocean University of China, Qingdao 266100, China

⁵Key Laboratory of Regional Climate-Environment for Temperate East Asia, Institute of Atmospheric Physics, Chinese Academy of Sciences, Beijing 100029, China

Correspondence to: Shuh-Ji Kao (sjkao@xmu.edu.cn)

Abstract Atmospheric deposition of long-range transport of anthropogenic reactive nitrogen (Nr, mainly comprised of NH_x, NO_y and water-soluble organic nitrogen (WSON)) from continents may have profound impact on marine biogeochemistry. On the other hand, surface ocean dissolve organic nitrogen (DON) may also contribute to aerosol WSON in the overlying atmosphere. Despite of the importance of off-continent dispersion and Nr interactions at the atmosphere-ocean boundary, our knowledge of the sources of various nitrogen species in the atmosphere over the open ocean remains limited due to insufficient observations. In the spring of 2014 and 2015, we conducted two cruises from the coast of China through the East China Seas (ECSs, i.e. the Yellow Sea and East China Sea) to the open ocean (i.e. the Northwest Pacific Ocean, NWPO). Concentrations of water-soluble total nitrogen (WSTN), NO₃⁻ and NH₄⁺, as well as the δ¹⁵N of WSTN and NO₃⁻ in marine aerosol were measured during both cruises. In the spring of 2015, we also analysed the concentrations and δ¹⁵N of NO₃⁻ and the DON of surface sea water (SSW; at a depth of 5 m) along the cruise track. Aerosol NO₃⁻, NH₄⁺ and WSON decreased logarithmically (1-2 orders of magnitude) with distance from the shore, reflecting strong anthropogenic emission sources of NO₃⁻, NH₄⁺ and WSON in China. Average aerosol NO₃⁻ and NH₄⁺ concentrations were significantly higher in 2014 (even in the remote NWOP) than in 2015 due to the stronger wind field in 2014, underscoring the role of the Asian winter monsoon in seaward transport of anthropogenic NO₃⁻ and NH₄⁺. However, the background aerosol WSON over the NWPO in 2015 (12.7 ± 8.7 nmol m⁻³) was similar to that in 2014 (10.7 ± 7.0 nmol m⁻³), suggesting an additional non-anthropogenic WSON source in the open ocean. Obviously, marine DON emissions should be considered in model and field assessments of net atmospheric WSON deposition in the open ocean. This study contributes parallel isotopic marine DON composition and aerosol Nr datasets; however, more research is required to explore complex Nr sources and deposition processes in order to advance our understanding of anthropogenic influences on the marine nitrogen cycle and nitrogen exchange at land-ocean and atmosphere-ocean interfaces.

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1 Introduction

Atmospheric transport and deposition of anthropogenic reactive nitrogen (Nr) to global oceans have increased considerably since the industrial revolution (Duce et al., 2008). Due to accumulated atmospheric Nr deposition, the stoichiometric relationship between nitrogen and phosphorous in the upper North Pacific Ocean (where nitrogen is the limiting nutrient in surface ocean) has been significantly altered (Kim et al., 2011). Such alterations may in turn impact pristine oceanic ecosystems and biogeochemical cycles. The Nr species deposited in the ocean include inorganic reduced nitrogen species (NH_3 and NH_4^+), oxidized nitrogen species (HNO_3 and NO_3^-), and organic nitrogen compounds (Erisman et al., 2002). The depositional fluxes (both dry and wet) of atmospheric Nr to global oceans have been studied previously through models (Duce et al., 2008; Doney et al., 2010). Recent model (Kanakidou et al., 2012) and observational (Altieri et al., 2014, 2016) studies have also reported that the ocean may be a source of atmospheric WSON and NH_3 . Nevertheless, field observations in the open ocean are still scarce; thus, more observations and new approaches, such as stable nitrogen isotopic composition studies, are urgently needed to trace the sources of Nr and investigate Nr exchange at the atmosphere-ocean interface.

Using organic nitrogen compounds, Cape et al. (2011) revealed several possible sources of WSON in the atmosphere, including livestock and animal husbandry, fertilizers, vehicle exhaust, biomass burning, secondary pollutants, and marine biological sources. Cape et al. (2011) also noted explicitly that complex atmospheric chemical process may obscure source identification for individual organic N compounds in atmospheric WSON. Stochastic analysis coupled with molecular characterization using FT-ICR-MS revealed that biological organic nitrogen in surface sea water can be a source of atmospheric WSON over the open ocean (Wozniak et al., 2014; Altieri et al., 2016); similar conclusions have been drawn from the positive correlation between marine aerosol WSON concentration and wind speed during a cruise in the Northwest Pacific Ocean (Luo et al., 2016).

The stable nitrogen isotopic composition ($\delta^{15}\text{N}$, $\delta^{15}\text{N}(\text{‰})_{\text{sample}} = ((^{15}\text{N}/^{14}\text{N})_{\text{sample}} / (^{15}\text{N}/^{14}\text{N})_{\text{standard}} - 1) \times 1000$) may be used to discriminate the sources of atmospheric NO_x and NH_x . This approach (i.e. the use of $\delta^{15}\text{N}\text{-NO}_x$) has successfully distinguished fossil fuel burning NO_x from soil biogenic activity NO_x (Felix and Elliott, 2014), as well as coal combustion emissions (Felix et al., 2012) from vehicle exhaust (Walters et al., 2015). Similarly, atmospheric NH_x can be measured and traced using $\delta^{15}\text{N}\text{-NH}_x$ (Freyer, 1978; Heaton, 1987; Jickells, 2003; Altieri et al., 2014). However, direct measurements of atmospheric $\delta^{15}\text{N}\text{-WSON}$ are currently highly impractical due to difficulties in completely separating organic and inorganic nitrogen. Via isotope mass conservation, a few previous studies have reported $\delta^{15}\text{N}\text{-WSON}$ values in precipitation collected from urban, rural and remote regions ranging from -7.3 to $+7.3\text{‰}$ (Cornell et al., 1995), which is consistent with values from precipitation sampled in a metropolis surrounded by agricultural areas in southern Korea (-7.9 to $+3.8\text{‰}$, with annual means of $+0.3\text{‰}$ and $+0.2\text{‰}$ in 2007 and 2008, respectively; Lee et al., 2012), but lower than the $\delta^{15}\text{N}\text{-WSON}$ values (-0.5 to $+14.7\text{‰}$, with a median of $+5\text{‰}$ and non-significant seasonal variation) reported in precipitation over the US East Coast area (Russell

et al., 1998). Compared with those for precipitation, aerosol $\delta^{15}\text{N}$ -WSON values reported in various rural regions in the UK cover a wider range (mainly caused by low values; the range covers -14.6 to $+12.5\%$, with medians of -2% and -5% for the fine and coarse mode, respectively; Kelly et al., 2005) (Fig. S1). Using $\delta^{15}\text{N}$, it is more difficult to identify the sources of atmospheric WSON than it is to identify NO_x and NH_x sources. However, the relatively uniform $\delta^{15}\text{N}$ values ($+2.2$ to $+5.4\%$) of dissolved organic nitrogen (DON) in surface sea water worldwide (Knapp et al., 2005; Knapp et al., 2011) enable the use of the isotope end member mixing approach for **primary WSON aerosol**; unfortunately, no cruises to date have undertaken parallel marine aerosol sampling and surface sea water $\delta^{15}\text{N}$ identification.

In terms of the hemispheric wind field, the East Asian monsoon transition from winter (October to April) to summer (May to September) influences the whole East Asian region. During the East Asian winter monsoon period, strong cold air masses mobilize rapidly through north-eastern China to the NWPO; in contrast, summer monsoon air masses arise primarily from the tropical Pacific Ocean (Wang et al., 2003). Air masses originating from China in winter have been reported to contain higher concentrations of NO_x and NH_x than air masses arising from remote Pacific Ocean regions in summer (Kunwar et al., 2014). Monitoring over the NWPO at Hedo Island and Ogawasara Island also shows that the dry deposition of aerosol NO_3^- and NH_4^+ varies inter-annually by a factor of 2-5 due to variable monsoon intensity (<http://www.eanet.asia/>). Meanwhile, dust storms occur frequently during monsoonal transition periods, and long-range transport in the upper- and mid-troposphere through northern China to the remote Pacific Ocean (Yang et al., 2013); these dust plumes contain abundant crustal elements in addition to NO_x and NH_x (Duce et al., 1980; Kang et al., 2009). In order to evaluate the seaward gradient of atmospheric Nr concentrations and explore the sources and fates of atmospheric NO_3^- , NH_4^+ and WSON from China (which features the largest emissions of such species worldwide), we conducted cruises from China to the Northwest Pacific Ocean during spring, during which the East Asian monsoon transition period occurs; cruises were complete during two different years to allow comparison.

In this study, we measured WSTN, NO_3^- and NH_4^+ concentrations, as well as $\delta^{15}\text{N}$ -WSTN and $\delta^{15}\text{N}$ - NO_3^- in marine aerosols collected over the ECSs and the NWPO during spring 2014 and 2015. The concentrations and $\delta^{15}\text{N}$ of DON and NO_3^- in **SSW (surface sea water, collected at 5 m depth)** were analysed in parallel along the cruise track in 2015. The purposes of this study are (1) to investigate the spatial distributions of concentrations of various Nr species in marine aerosol from the ECSs to the NWPO, (2) to explore possible sources of atmospheric WSON in marine environments, and 3) to advance our understanding of atmospheric Nr transport at the land-ocean boundary and potential Nr exchanges between the atmosphere and the ocean.

2 Material and Methods

2.1 Sampling and background weather during cruises

Total suspended particulate samples were collected using a high-volume sampler (TE-5170D; Tisch Environmental, Inc.) with Whatman[®]41 cellulose filters during two research cruises (Fig. 1) aboard the R/V *Dongfanghong II*. The first cruise (Fig. 1a) spanned 17 March to 22 April 2014 (44 samples were collected in total; detailed sampling information can be found in Luo et al., 2016), and the second cruise (Fig. 1b) lasted from 30 March to 3 May 2015 (38 samples were collected in total; detailed sampling information, including the date, time period and location for each sample are listed in Table S1). To avoid self-contamination from the research vessel, the TSP sampler was installed on the top of the tower at the ship head, and aerosols were sampled only during travel. More information about self-contamination from ship exhaust can be found in Luo et al. (2016). Both cruises were undertaken during the East Asian monsoon transition period. The 2-month average (March and April) wind streamlines at 1000 hPa over the NWPO show that the wind speed ranged from 2 to 6 m s⁻¹ in 2014 (Fig. 1a) and from 1 to 3 m s⁻¹ in 2015 (Fig. 1b). In general, the wind was stronger in 2014 than in 2015 over the open ocean during the sampling periods.

Meteorological data, including wind speed, direction, relative humidity (RH) and ambient temperature, is shown in Figure S2 for the 2015 cruise (data for the 2014 cruise is reported in Luo et al., 2016). In the ECSs, both cruises encountered sea fog, which inevitably influenced aerosol sampling and, of course, aerosol chemistry. Because of the analogous weather conditions experienced during the two cruises, we used the techniques of Luo et al. (2016) to classify the 2015 marine aerosol samples into three types (namely, sea-fog-modified aerosol (orange triangles) collected in the ECSs, dust aerosol (pink circles), and background aerosol (black squares) sampled in the NWPO; Fig. 1) based on the meteorological conditions (Fig. S2), concentrations of aluminium (data not shown), and the Lidar browse images from NASA (Fig. S3). Hereafter, we define background aerosol as aerosol not impacted by either dust or sea fog; rather than representing pristine conditions, the background is an environmental baseline collected within the study area during the investigation period.

To examine the relationships between the isotopic compositions of WSON in marine aerosol and DON in SSW, we collected SSW (at a depth of 5 m; sampling locations are shown in Fig. 1b as open blue circles) using Niskin bottles during the 2015 cruise. The SSW samples were filtered using a 0.22 μm MILLEX•GP filter and kept frozen at -20 °C in 50 mL 450 °C pre-combusted brown glass tubes until analysis.

2.2 Chemical analyses

2.2.1 NO₃⁻ and NH₄⁺ in marine aerosol

The marine aerosol samples were extracted in Milli-Q water (specific resistivity of 18.2 MΩ cm⁻¹) following Luo et al. (2016). The aerosol extracts were analysed using an ion chromatograph (model ICS-1100 for anions and model ICS-900 for

cations) equipped with a conductivity detector (ASRS-ULTRA) and suppressor (ASRS-300 for the ICS-1100 and CSRS-300 for the ICS-900). The precision was better than 5% for all ionic species. Details of the analytical processes can be found in Hsu et al. (2014). Only five of the aerosol samples contained detectable NO_2^- , and these accounted for < 1% of the WSTN. Eight filters of the same type as those used to collect samples were taken as blanks. All blank filters and aerosol samples were stored at $-20\text{ }^\circ\text{C}$ during the sampling periods and underwent the same extraction procedure. The NO_3^- , NH_4^+ and WSON content of the blank filters comprised less than 1%, 4% and 9%, respectively, of the average concentration of the corresponding N species in the aerosol samples.

2.2.2 NO_3^- in SSW

The SSW NO_3^- concentration was measured using a chemiluminescence method (Braman and Hendrix, 1989). Briefly, the solution containing NO_3^- was injected into a heated solution of acidic Vanadium (III), in which the NO_3^- was reduced to nitric oxide (NO) to be measured by a NO_x analyser (MODEL T200U, Teledyne Technologies Incorporated, USA). Working standards were injected after every 10 samples. The relative standard deviation for the standard replicate was < 5%. The concentration of NO_2^- in the SSW was below the $0.1\text{ }\mu\text{mol L}^{-1}$ detection limit throughout the cruise, as reported previously (Adornato et al., 2005).

2.2.3 WSTN in marine aerosol and total dissolve nitrogen in SSW

Aerosol WSTN and SSW total dissolved nitrogen (TDN, i.e. $\text{NO}_3^- + \text{NH}_4^+ + \text{DON}$) were measured using the alkaline potassium persulfate oxidation method to convert WSTN and TDN to NO_3^- (Luo et al., 2016; Knapp et al., 2005). The NO_3^- content of the digested solution was then measured via chemiluminescent detection (Braman and Hendrix, 1989). To verify the WSTN and TDN oxidation efficiency, N-containing organic and inorganic compound standards (specifically, glycine, urea, ethylene diamine tetraacetic acid, and ammonium sulphate) were prepared in solution at a concentration of $800\text{ }\mu\text{M-N}$ for oxidation analysis. The recoveries of the N-containing compound standards under oxidation by alkaline potassium persulfate were within 95 ~ 105% ($n = 6$).

2.2.4 Stable nitrogen isotope

The $\delta^{15}\text{N-NO}_3^-$ was analysed using the denitrifier method described by Sigman et al. (2001) and Casciotti et al. (2002), which has been widely used to analyse the $\delta^{15}\text{N}$ in NO_3^- in aerosol, rainwater and sea water (Altieri et al., 2013; Buffam and McGlathery, 2003; Gobel et al., 2013; Hastings et al., 2003; Sigman et al., 2005), as well as that in NO_3^- in solutions digested with alkaline potassium persulfate (Knapp et al., 2005; Knapp et al., 2010; Knapp et al., 2011; Knapp et al., 2012). Detailed stable nitrogen isotope analysis procedures can be found in Archana et al. (2016) and Yang et al. (2014). Briefly, NO_3^- was reduced to N_2O by the denitrifying bacteria *Pseudomonas aureofaciens* (ATCC 13985); then, the stable nitrogen isotope of N_2O was analysed using a GasBench II connected to a continuous flow isotope ratio mass spectrometer (IRMS, Thermo Delta

V Advantage). Two international standards, USGS34 and IAEA-N3 (Böhlke et al., 2003), and two nitrate laboratory working standards were used to verify instrument stability. After the WSTN and TDN were oxidized to NO_3^- , the $\delta^{15}\text{N}$ -WSTN and $\delta^{15}\text{N}$ -TDN were analysed using the same procedures employed for NO_3^- . The pooled standard deviations for replicates were $\pm 0.2\%$, $\pm 0.5\%$ and $\pm 0.5\%$ for $\delta^{15}\text{N}$ - NO_3^- , $\delta^{15}\text{N}$ -WSTN and $\delta^{15}\text{N}$ -TDN, respectively.

5 2.3 Data analysis

The concentrations of WSON in marine aerosol, which cannot be measured directly (as mentioned previously), were calculated using the following equation:

$$[\text{WSON}] = [\text{WSTN}] - [\text{NO}_3^-] - [\text{NH}_4^+], \quad (1)$$

where $[\text{WSTN}]$, $[\text{NO}_3^-]$ and $[\text{NH}_4^+]$ are the molar concentrations (nmol N m^{-3}) of the given water-soluble nitrogen species in marine aerosol. The standard errors propagated through the WSON calculation for the 2014 data can be found in Luo et al. (2016). For 2015, the standard errors propagated through WSON calculation varied from sample to sample from 7 to 210%; the average standard error of all samples was 33%.

Reduced nitrogen (RN, i.e. $\text{NH}_4^+ + \text{WSON}$) and the $\delta^{15}\text{N}$ -RN in the aerosol were calculated by mass balance via:

$$[\text{RN}] = [\text{WSTN}] - [\text{NO}_3^-], \quad (2)$$

$$\delta^{15}\text{N-RN} = (\delta^{15}\text{N-WSTN} * [\text{WSTN}] - \delta^{15}\text{N-NO}_3^- * [\text{NO}_3^-]) / [\text{RN}], \quad (3)$$

where $[\text{WSTN}]$ and $[\text{NO}_3^-]$ are the molar concentrations (nmol N m^{-3}) of the given water-soluble nitrogen species in the marine aerosol. The average propagated standard error for RN was 9% for both 2014 and 2015. The propagated error for the calculation of $\delta^{15}\text{N-RN}$ was $\pm 0.6\%$.

Similar to aerosol WSON, the SSW DON concentration and $\delta^{15}\text{N}$ -DON were calculated using the following equations:

$$[\text{DON}] = [\text{TDN}] - [\text{NO}_3^-], \quad (4)$$

$$\delta^{15}\text{N-DON} = (\delta^{15}\text{N-TDN} * [\text{TDN}] - \delta^{15}\text{N-NO}_3^- * [\text{NO}_3^-]) / [\text{DON}], \quad (5)$$

where $[\text{TDN}]$ and $[\text{NO}_3^-]$ are the molar concentrations ($\mu\text{mol N L}^{-1}$) of the given species in SSW. The standard error propagated through the DON calculation was 5.3%. Since the average $[\text{NH}_4^+]$ in SSW at the selected sites during the 2015 cruise (12 sites and 23 samples) was $0.05 \mu\text{M}$, which is much less than DON in μM level, thus, $[\text{NH}_4^+]$ is neglected in Equations 4 and 5.

Unfortunately, most of the NO_3^- concentrations in the 5 m SSW samples were $< 0.5 \mu\text{mol L}^{-1}$, which is too low for the measurement of $\delta^{15}\text{N-NO}_3^-$. We attempted to evaluate the interference from nitrate in the $\delta^{15}\text{N}$ -DON calculations. For all the SSW samples on average, NO_3^- comprised 5.7% of the total NO_3^- plus DON; the $\delta^{15}\text{N-NO}_3^-$ in SSW ranged from +8.2‰ to +16.4‰ in our measurements, and the bias of the calculated $\delta^{15}\text{N-DON}$ varied from +0.5‰ to +0.9‰.

The dry deposition N fluxes were calculated via:

$$F = C_i \times V_i, \quad (7)$$

where C_i is the concentration of a given water-soluble nitrogen species in the aerosol, and V_i is the given dry deposition velocity

of the given nitrogen species. The deposition velocities of water-soluble nitrogen species used herein were 2 cm s^{-1} for nitrate, 0.1 cm s^{-1} for ammonium, and 1.0 cm s^{-1} for WSON, which were consistent with our previous studies (Luo et al., 2016).

3 Results and Discussion

3.1 Spatial and temporal variations of water-soluble nitrogen species in the aerosol

Overall, significant logarithmic decreases can be seen from shore seaward for all water-soluble nitrogen species and WSTN in both 2014 and 2015 (Fig. 2). The seaward gradient was caused primarily by continental emissions influenced by sea fog (Luo et al., 2016); thus, concentrations were high in the ECSs (orange triangles in Fig. 2) and low offshore in the NWPO background aerosol (black squares in Fig. 2). Dust aerosol (pink circles in Fig. 2) appeared sporadically in the NWPO and generally featured higher NH_4^+ and NO_3^- (but not WSON) values (Table 1 and Fig. 2).

The measured WSTN concentrations in TSP varied from 21 to 2411 nmol m^{-3} (Table 1 and Fig. 2a and 2b), lower than those in PM_{10} sampled during spring in Xi'an, China (which ranged from 786 to 3000 nmol m^{-3} ; Wang et al., 2013), but higher than those in TSP sampled in Sapporo, Japan (which ranged from 20.9 to 108.6 nmol m^{-3} ; Pavuluri et al., 2015), Okinawa Island (which ranged from 5 to 216 nmol m^{-3} ; Kunwar and Kawamura, 2014), and the North Pacific (which ranged from 1.4 to 64.3 nmol m^{-3} in May-July; Hoque et al., 2015). This wide range of aerosol WSTN content illustrates the influence of the distance between sampling locations and emission sources (Matsumoto et al., 2014), seasonality (Kunwar and Kawamura, 2014), and meteorological conditions, such as sea fog (Luo et al., 2016).

The concentrations of marine aerosol WSTN in the ECSs ranged from 444 to 2411 nmol m^{-3} in 2014 (with a volume-weighted mean of 1136 nmol m^{-3}) and from 92.9 nmol m^{-3} to 1195 nmol m^{-3} in 2015 (with a volume-weighted mean of 287 nmol m^{-3}), which were clearly higher than those in dust aerosol (with volume-weighted means of 242 nmol m^{-3} in 2014 and 154 nmol m^{-3} in 2015) and background aerosol (with volume-weighted means of 85.6 nmol m^{-3} in 2014 and 42.3 nmol m^{-3} in 2015) collected in the NWPO (Table 1). The air mass backward trajectories (see Fig. S5 for 2015 and Luo et al., 2016 for 2014) reveal that the high aerosol water-soluble nitrogen species in the ECSs arose from anthropogenic Nr emissions from eastern China (Gu et al., 2012). In addition, frequent formation of sea fog in the ECSs in spring (Zhang et al., 2009) may also enrich the amount of water-soluble nitrogen in sea-fog-modified aerosol via chemical processing (Luo et al., 2016). The much higher water-soluble nitrogen species in the ECSs marine aerosol (compared to that in the NWPO aerosol) indicates that continental/anthropogenic Nr strongly affected the marine aerosol. However, the amounts of sea-salt ions (such as Na^+) in the ECSs aerosols sampled in both 2014 ($123 \pm 98 \text{ nmol m}^{-3}$; Luo et al., 2016) and 2015 ($151 \pm 164 \text{ nmol m}^{-3}$; Luo et al., unpublished data) were higher than those in land aerosol sampled during spring ($23 \pm 7.8 \text{ nmol m}^{-3}$ in Beijing; Zhang et al., 2013), which implies that those aerosols sampled in the ECSs were also significantly influenced by sea salt. Thus, we define the aerosol collected by ship over the ECSs as marine aerosol.

In the NWPO, higher WSTN values were observed in dust aerosol than in background aerosol in both 2014 and 2015; the dust aerosol WSTN consisted predominantly of NH_4^+ and NO_3^- rather than WSON (Table 1, pink circles in Fig. 2), which implies that dust can carry more NH_4^+ and NO_3^- during long-range transport from East Asia to the NWPO during the Asian winter monsoon in spring. The air mass back trajectories of those dust aerosols arose mainly from high-Nr regions, as evidenced by dust plumes captured by Lidar browse images from NASA (Fig. S3 for 2015 and Luo et al., 2016 for 2014).

In our observations, the concentrations of NH_4^+ and NO_3^- were higher in all aerosols in 2014 than they were in 2015 (Fig. 3a and 3b; Table 1). The difference between the two years was caused by a stronger Asian winter monsoon in 2014. Additionally, the cruise in 2014 (17 March to 22 April) occurred during a period of intensive coal and/or fossil fuel combustion for heat supply in northern China; in contrast, the 2015 cruise started on 30 March and finished on 3 May, during a hiatus in heating demand. The influence of heating on aerosol emissions can be seen in the atmospheric aerosol optical depth over heat-generating areas in China; for example, Xiao et al. (2015) reported that the aerosol optical depth was five times higher during heat generation periods than during non-generation period. The consistent variations between heat supply in northern China and higher NO_3^- in marine aerosol sampled in 2014 than 2015 underscore the influence of anthropogenic NO_x emissions on marine aerosol NO_3^- . Our spring observations in 2014 and 2015 showed average concentrations of NH_4^+ and NO_3^- in background aerosol (black boxes in Fig. 3a and 3b; Table 1) higher than the average concentrations ($3.5 \pm 3.3 \text{ nmol m}^{-3}$ for NH_4^+ and $2.1 \pm 1.5 \text{ nmol m}^{-3}$ for NO_3^-) reported in the western Pacific Ocean in summer (blue boxes in Fig. 3a and 3b, data from Miyazaki et al., 2011 and Jung et al., 2013), which suggests far-reaching influence of anthropogenic emissions during the monsoon transition. Moreover, concentrations of NH_4^+ and NO_3^- were higher in 2014 due to the stronger Asian winter monsoon, which further supports the idea that the monsoon exerts an important control on annual and seasonal variations in marine aerosol Nr via atmospheric long-range transport.

Unlike NH_4^+ and NO_3^- , WSON concentrations in the background aerosol sampled in the NWPO in 2014 (average = $10.7 \pm 7.0 \text{ nmol m}^{-3}$) were similar to those in 2015 (average = $12.7 \pm 8.7 \text{ nmol m}^{-3}$; black boxes in Fig. 3c). In the open ocean, apart from terrestrial and anthropogenic WSON long-range transport (Mace et al., 2003; Lesworth et al., 2010), the ocean itself is the most likely source of marine aerosol WSON. For instance, in situ observations in the subtropical North Atlantic found that aerosol WSON had strong positive relationships with surface ocean primary productivity and wind speed (Altieri et al., 2016). Another study in the South Atlantic Ocean showed that the WSON in marine aerosol associated with high SSW chlorophyll-*a* was 9 times higher than that associated with low SSW chlorophyll-*a* (Violaki et al., 2015). In our observations, the WSON in the background aerosol (black bar in Fig. 3c) was significantly higher in spring than in summer (blue bar in Fig. 3c), which is consistent with the higher SSW chlorophyll-*a* concentration over the NWPO in spring relative to that in summer (Fig. S6). However, the sources of marine aerosol WSON are complex mixture which composed of primary marine organic N and secondary N-containing organic aerosol. Biogenic organic material in SSW can be injected into the atmosphere to form an ice cloud via bubble bursting at the atmosphere-ocean interface (Wilson et al., 2015), this is probably the primary WSON

aerosol source. Volatile organic compounds emitted from the surface ocean can react with NO_x and NH_x in the atmosphere to form secondary N-containing organic aerosol (Fischer et al., 2014; Liu et al., 2015).

3.2 Isotopic composition of nitrogen species

Aerosol $\delta^{15}\text{N-NO}_3^-$ values over the ECSs and NWPO in 2014 and 2015 ranged from -9.2 to $+10.2\text{‰}$ (Table 2 and Fig. 4a and 4b). All the observed $\delta^{15}\text{N-NO}_3^-$ values fell within the ranges previously reported for atmospheric $\delta^{15}\text{N-NO}_3^-$ over land (Elliott et al., 2009; Fang et al., 2011; Felix and Elliott, 2014) and in the marine boundary layer (Altieri et al., 2013; Gobel et al., 2013; Hastings et al., 2003; Morin et al., 2009; Savarino et al., 2013). The mass-weighted mean aerosol $\delta^{15}\text{N-NO}_3^-$ values in 2014 ($+1.6\text{‰}$ in the ECSs, -1.1‰ for dust aerosol and -2.6‰ for background aerosol sampled in the NWPO) were similar to those in 2015 ($+1.9\text{‰}$ in the ECSs, -3.8‰ for dust aerosol and -1.1‰ for background aerosol sampled in the NWPO; Table 2) for all the aerosols, suggesting that the aerosol NO_3^- arose from similar origins and atmospheric chemical pathways in both 2014 and 2015.

The $\delta^{15}\text{N-WSTN}$ values for all the aerosols ranged from -10.7 to $+5.6\text{‰}$ (Table 2, Fig 4c and 4d), which is consistent with the $\delta^{15}\text{N-WSTN}$ ranges reported in precipitation, which include -4.2 to $+12.3\text{‰}$ in the Baltic Sea (Rolff et al., 2008); -8 to $+8\text{‰}$ in Bermuda (Knapp et al., 2010); -4.9 to $+3.2\text{‰}$ in a forest in southern China (Koba et al., 2012); and -12.1 to $+2.9\text{‰}$ in Cheju, Korea (Lee et al., 2012). In contrast, our results were lower than the $\delta^{15}\text{N-WSTN}$ in TSP sampled in Sapporo, Japan ($+12.2$ to $+39.1\text{‰}$; Pavuluri et al., 2015) and the Sapporo Forest ($+9.0$ to $+26.0\text{‰}$; Miyazaki et al., 2014). These authors attributed higher isotopic values to biogenic sources, nitrogenous aerosol aging, and fossil fuel combustion. However, WSTN is, in fact, composed of various nitrogen species, and the relative proportions of NH_4^+ , NO_3^- and WSON to WSTN, coupled with their isotopic compositions (i.e. $\delta^{15}\text{N-NH}_4^+$, $\delta^{15}\text{N-NO}_3^-$ and $\delta^{15}\text{N-WSON}$), jointly mediate variations in aerosol $\delta^{15}\text{N-WSTN}$ (where $\delta^{15}\text{N-WSTN} \cdot [\text{WSTN}] = \delta^{15}\text{N-NO}_3^- \cdot [\text{NO}_3^-] + \delta^{15}\text{N-NH}_4^+ \cdot [\text{NH}_4^+] + \delta^{15}\text{N-WSON} \cdot [\text{WSON}]$). Taking our study as an example, the inconsistent trends in both the positive relationships between $\delta^{15}\text{N-WSTN}$ and $\delta^{15}\text{N-NO}_3^-$ in the ECSs aerosols and NWPO background aerosols (Fig. S7 a and c) and the negative relationship between $\delta^{15}\text{N-WSTN}$ and the NO_3^- concentration (Fig. S7 d and f) imply that the $\delta^{15}\text{N}$ of other species (NH_4^+ and WSON) in WSTN affected the $\delta^{15}\text{N-WSTN}$.

The $\delta^{15}\text{N-WSTN}$ values in 2014 (-10.7 to $+1.0\text{‰}$) were lower than those in 2015 (-5.6 to $+5.6\text{‰}$; Table 2), whereas the $\text{NH}_4^+/\text{WSTN}$ ratios were higher in 2014 than in 2015 (Table 1) for all aerosol types. The negative linear relationships between $\text{NH}_4^+/\text{WSTN}$ and $\delta^{15}\text{N-WSTN}$ for all aerosol types (Fig. 5a, 5b, and 5c) may be attributed to higher proportions of NH_4^+ in WSTN and negative $\delta^{15}\text{N-NH}_4^+$ values, which sourced from the anthropogenic and marine emissions (Altieri et al., 2014; Jickells, 2003; Koba et al., 2012; Liu et al., 2014; Xiao et al., 2012; Yeatman et al., 2001). In fact, low $\delta^{15}\text{N-NH}_4^+$ values have been reported in precipitation collected from many places, such as Beijing (-33.0 to $+14.0\text{‰}$ with an arithmetic mean of -10.8‰ ; Liu et al., 2014); Guiyang City in southwestern China (-38.0 to $+5.0\text{‰}$ with an average of -15.9‰ ; Xiao et al., 2012); Gwangju, Korea (-15.9 to $+2.9\text{‰}$ with volume-weighted means of -6.0‰ in 2007 and -6.8‰ in 2008; Lee et al., 2012); and

a forest in southern China (-18.0 to $+0.0\text{‰}$ with a concentration-weighted mean of -7.7‰ ; Koba et al., 2012). Low atmospheric $\delta^{15}\text{N-NH}_4^+$ has also been associated with marine air masses (e.g. -8 to -5‰ , Jickells et al. 2003; $-9 \pm 8\text{‰}$, Yeatman et al., 2001; and $-4.1 \pm 2.6\text{‰}$, Altieri et al., 2014). Together, this low atmospheric average $\delta^{15}\text{N-NH}_4^+$ (-15.9 to -4.1‰) supports our findings of higher $\text{NH}_4^+/\text{WSTN}$ and lower $\delta^{15}\text{N-WSTN}$ in aerosol.

5 There were positive linear relationships between the WSON/WSTN ratio and $\delta^{15}\text{N-WSTN}$ for all the aerosols types (Fig. 5d, 5e, and 5f). This implies that the aerosol $\delta^{15}\text{N-WSON}$ may be positive. The WSON in the marine aerosol either originated from terrestrial long-range transport or the DON from SSW and N-containing secondary organic marine aerosol as discussed in Section 3.1. Terrestrial aerosol $\delta^{15}\text{N-WSON}$ was reported in a wide range (-15.0 to $+14.7\text{‰}$) with mean values from -3.7 to $+5.0\text{‰}$, and $\delta^{15}\text{N-WSON}$ in marine aerosol with more positive $\delta^{15}\text{N}$ (Fig. S1). In addition, our own observations for $\delta^{15}\text{N-}$
10 DON in SSW (sampled at 5 m depth) showed positive $\delta^{15}\text{N}$ in the ECSs (varying from $+5.1$ to $+12.9\text{‰}$, with an average $+7.9 \pm 2.3\text{‰}$) and NWPO (ranging from $+1.9$ to $+11.6\text{‰}$, with an average $+5.7 \pm 2.0\text{‰}$; Fig. 6a). At the same time, concentrations of DON in our observations ranged from 4.4 to $11.8 \mu\text{mol L}^{-1}$ (Fig. 6b), which is within the DON concentration range reported in global SSW (Knapp et al., 2011; Letscher et al., 2013; Li et al., 2009; L nborg et al., 2015; Van Engeland et al., 2010). This high DON concentration in SSW may be ejected into the atmosphere during bubble bursting (Wilson et al., 2015).

15 To better clarify the sources of aerosol WSON, we removed the aerosol NO_3^- and its $\delta^{15}\text{N}$ effect from the WSTN and its $\delta^{15}\text{N-WSTN}$, respectively, by mass balance. The remaining NH_4^+ and WSON was defined as reduced nitrogen ($\text{RN} = \text{NH}_4^+ + \text{WSON}$). The $\delta^{15}\text{N-RN}$ ranged from -11.8 to $+7.4\text{‰}$ for all aerosol types in both 2014 and 2015 (Table 2), which is consistent with the reported $\delta^{15}\text{N-RN}$ in precipitation (-12.6 to $+7.8\text{‰}$) collected in Bermuda, in the Atlantic Ocean (Knapp et al., 2010).
20 The negative relationships between $\delta^{15}\text{N-RN}$ and NH_4^+ concentration (Fig. S8 a, b, and c), and between $\delta^{15}\text{N-RN}$ and the ratios of NH_4^+/RN (Fig. S8 d, e, and f) for all aerosol types further support the low values of $\delta^{15}\text{N-NH}_4^+$ in marine aerosol. Three important end members, compiled atmospheric $\delta^{15}\text{N-NH}_4^+$ (-15.9 to -4.1‰ , green bars in Fig. 7) and continental $\delta^{15}\text{N-WSON}$ (-3.7 to $+5.0\text{‰}$, grey bars) versus the $\delta^{15}\text{N-DON}$ observed in SSW in our cruise ($+7.9 \pm 2.3\text{‰}$ in the ECSs and $+5.7 \pm 2.0\text{‰}$ in the NWPO, red dots with error bars), were added to Figure 7 to facilitate discussion. Note that nearly all the aerosol $\delta^{15}\text{N-RN}$ values in 2014 were lower than those in 2015, which may be attributed to higher NH_4^+ concentrations in 2014 than
25 in 2015 (Table 1). Moreover, higher values of $\delta^{15}\text{N-RN}$ can be seen with higher WSON/RN ratios for all aerosol types in Fig. 7. The high values of $\delta^{15}\text{N}$ in continental WSON and marine DON in SSW may cause these positive relationships. Thus, although higher WSON/RN values accompany with higher $\delta^{15}\text{N-RN}$, we may still not conclude a significant DON in surface sea water contribution to aerosol WSON. In the open ocean, to some extent, background aerosol WSON was more likely influenced by DON in surface sea water judging by nitrogen isotopic information.

30 Note that some data points collected in 2015 for the open ocean case in Fig. 7b and 7c fell outside the mixing field, deviating toward higher $\delta^{15}\text{N-RN}$ values; these high $\delta^{15}\text{N-RN}$ values may be attributable to $\delta^{15}\text{N}$ fractionation and ^{15}N enrichment in the WSON during processes such as secondary N-containing organic aerosol formation by the reaction of NH_x

or NO_x with organic aerosol (Fischer et al., 2014; Liu et al., 2015), complex atmospheric chemical reactions (i.e. the photolysis of organic nitrogen into ammonium; Paulot et al., 2015), aerosol WSON aging process, and in-cloud scavenging (Altieri et al., 2016). More studies are needed to explore nitrogen transformation processes, especially those focusing on secondary N-containing organic aerosol in the atmosphere from an isotopic perspective.

5 3.3 Dry Nr deposition and its biogeochemical role

The dry deposition of aerosol NH_4^+ , NO_3^- and WSON is summarized in Table 3. The calculated depositional fluxes of water-soluble nitrogen species in the ECSs were significantly higher than those in the NWPO (Fig. 8). The averaged dry depositional fluxes of NH_4^+ and NO_3^- in 2014 were 2 to 5 times higher than those in 2015 for all aerosol types (Table 3). The dry depositional fluxes of NH_4^+ and NO_3^- in dust aerosol were clearly higher than those in background aerosol in the NWPO (Table 3, Fig. 8a, 8b, 8c and 8d). Comparisons of these dry fluxes with other similar studies and estimations of the contribution of atmospheric Nr deposition to primary production in the NWPO are discussed in Luo et al. (2016), specifically for 2014; here, we focus on the influence of atmospheric Nr deposition on the nitrogen cycle in the ocean.

The influences of atmospheric Nr deposition on the marine nitrogen cycle are obvious over long time scales. For example, by analysing concentrations of NO_3^- and phosphorus in sea water over the NWPO from 1980 to 2010, Kim et al. (2010) reported that the higher N/P ratio in the upper ocean (in contrast to the deep ocean) in the NWPO was caused primarily by the accumulation of atmospheric anthropogenic Nr deposition. Another recent study found higher atmospheric anthropogenic Nr deposition to be associated with lower $\delta^{15}\text{N}$ in surface sediment over the NWPO (Kim et al., 2017), the authors also posited that atmospheric anthropogenic Nr deposition can reach as far down as the deep ocean through biological action and lower the $\delta^{15}\text{N}$ in surface sediment. The atmospheric $\delta^{15}\text{N}$ values for water-soluble nitrogen species in our observations (Table 2) are clearly lower than the $\delta^{15}\text{N}\text{-NO}_3^-$ in deep ocean water (+5.6‰; unpublished data from Kao); thus, it is possible that atmospheric $\delta^{15}\text{N}\text{-Nr}$ to lower $\delta^{15}\text{N}\text{-NO}_3^-$ in the thermocline as mentioned in previous studies (Knapp et al., 2010; Yang et al., 2014). However, it is hard to quantify the contribution of atmospheric $\delta^{15}\text{N}\text{-Nr}$ to $\delta^{15}\text{N}\text{-NO}_3^-$ in the thermocline from the perspective of ^{15}N for the follow reasons. First, there are large spatial and temporal uncertainties in the dry and wet deposition fluxes of atmospheric Nr. For example, the dry depositional fluxes of NH_4^+ and NO_3^- (23.1 to 43.3 $\mu\text{mol N m}^{-2} \text{d}^{-1}$; Table 3) in our observations are significantly higher than those in summer (4.9 $\mu\text{mol N m}^{-2} \text{d}^{-1}$; Jung et al. 2013). Moreover, according to a previous study, wet Nr deposition is 2 to 3 times higher than dry deposition (Jung et al., 2013), but Nr wet deposition in spring over the NWPO is unknown. Second, the $\delta^{15}\text{N}$ of N-fixation (-2 to 0‰; summarized by Knapp et al., 2010) is similar to the atmospheric $\delta^{15}\text{N}\text{-Nr}$ (Table 2), but the fluxes of N-fixation in the global ocean also vary considerably both spatially and temporally (Mulholland and Bernhardt, 2005; Needoba et al., 2007; Karl et al., 1997). Third, ^{15}N fractionation always occurs in the complicated marine nitrogen cycle (Knapp et al., 2005, 2010, 2011, 2012), which hampers the use of ^{15}N in estimating the influence of atmospheric deposition Nr on marine $\delta^{15}\text{N}$ under our limited current understanding.

4 Conclusions

Concentrations of water-soluble total nitrogen (WSTN), nitrate (NO_3^-) and ammonium (NH_4^+), as well as the stable nitrogen isotopes of $\delta^{15}\text{N}$ -WSTN and $\delta^{15}\text{N}$ - NO_3^- , were measured in marine aerosols sampled between the ECSs and the NWPO in spring 2014 and 2015. Dissolve organic nitrogen (DON) and $\delta^{15}\text{N}$ -DON were also analysed in SSW collected at a depth of 5 m along the cruise route in the spring of 2015. The highest concentrations of water-soluble nitrogen species were found in aerosol sampled in the ECSs, which suggests significant influence from anthropogenic emissions on aerosol Nr. The higher NO_3^- and NH_4^+ in all aerosol types in 2014 (relative to 2015) may be attributed to the stronger Asian winter monsoon in 2014, as well as the intensity of residential heating in spring in northern China.

Negative linear relationships were found between the NH_4^+ /WSTN ratios and $\delta^{15}\text{N}$ -WSTN for all aerosol types. In contrast, positive linear relationships were observed between the WSON/WSTN ratios and $\delta^{15}\text{N}$ -WSTN. The distinctive nitrogen species compositions and isotopic compositions suggest that aerosol $\delta^{15}\text{N}$ -WSTN values were mediated synergistically by NO_3^- , NH_4^+ , and WSON in our observations. Meanwhile, our isotope mixing model indicates that DON in SSW is likely to be a source of primary WSON in aerosol, especially over the open ocean. Many uncertainties remain concerning Nr in the marine boundary layer and surface sea water, let alone Nr exchange at the atmosphere-ocean interface; further study of Nr exchange between the low atmosphere and the upper ocean is needed in the future.

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Table 1. Concentration ranges and means for WSTN, NH₄⁺, NO₃⁻, WSON and RN in aerosols.

		WSTN (nmol m ⁻³)			NH ₄ ⁺ (nmol m ⁻³)			NH ₄ ⁺ /	NO ₃ ⁻ (nmol m ⁻³)			NO ₃ ⁻ /	WSON (nmol m ⁻³)			WSON/	RN (nmol m ⁻³)		
		Range	Mean ^a	Mean ^b	Range	Mean ^a	Mean ^b	WSTN	Range	Mean ^a	Mean ^b	WSTN	Range	Mean ^a	Mean ^b	WSTN	Range	Mean ^a	Mean ^b
ECSs (Sea fog)	2014	444 ~ 2411	1126	1136	228 ~ 777	442	437	0.42±0.09	160 ~ 1118	536	550	0.48±0.07	23 ~ 517	149	148	0.10±0.06	267 ~ 1294	591	585
	2015	92.9 ~ 1195	321	287	25.7 ~ 564	126	113	0.35±0.12	30.1 ~ 239	93.2	87.6	0.35±0.15	6.9 ~ 392	102	85.7	0.29±0.18	32.6 ~ 956	228	199
NWPO (Dust)	2014	205 ~ 297	245	242	94.4 ~ 163	138	137	0.56±0.07	78.6 ~ 145	100	98.3	0.41±0.05	7.7 ~ 16.9	11.2	11	0.05±0.02	111 ~ 173	145	144
	2015	81.4 ~ 340	147	154	20.7 ~ 143	58.3	61.4	0.39±0.13	34.7 ~ 126	57.2	58.6	0.41±0.1	5.8 ~ 80.7	31.7	34.4	0.21±0.13	45.0 ~ 214	89.9	95.8
NWPO (Bgd.)	2014	31.4 ~ 411	88.8	85.6	16.1 ~ 244	54	51.8	0.60±0.11	6.4 ~ 166	25.8	25.1	0.27±0.09	0.9 ~ 27.1	10.7	9.7	0.14±0.08	19.2 ~ 245	63.1	60.5
	2015	21.0 ~ 68.7	41.7	42.3	6.9 ~ 29.5	16.3	15.5	0.41±0.15	2.8 ~ 35.1	12.7	13.4	0.29±0.13	1.0 ~ 40.3	12.7	13.4	0.30±0.16	15.2 ~ 48.2	29	28.9

a, arithmetic mean

b, volume-weighted mean

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Table 2. Ranges and means for stable nitrogen isotopes of WSTN, NO₃⁻ and RN in aerosols.

		$\delta^{15}\text{N-WSTN}$			$\delta^{15}\text{N-NO}_3^-$			$\delta^{15}\text{N-RN}$		
		Range	Mean ^a	Mean ^b	Range	Mean ^a	Mean ^b	Range	Mean ^a	Mean ^b
ECSs (Sea fog)	2014	-5.3 ~ -1.7	-3.4	-3.4	-1.7 ~ +4.3	+1.0	+1.6	-11.8 ~ -2.5	-7.3	-7.9
	2015	-4.3 ~ +0.9	-1.1	-2.1	-1.3 ~ +10.2	+2.8	+1.9	-5.2 ~ -0.6	-3.3	-3.7
NWPO (Dust)	2014	-6.9 ~ -3.1	-4.7	-4.4	-3.0 ~ +1.3	-1.0	-1.1	-9.4 ~ -5.8	-7.1	-7.3
	2015	-3.0 ~ +0.8	-1.5	-1.3	-7.3 ~ -2.1	-3.9	-3.8	-2.7 ~ +5.5	+0.6	+0.3
NWPO (Bgd.)	2014	-10.7 ~ +1.0	-5.6	-5.5	-7.6 ~ +4.3	-2.3	-2.6	-11.7 ~ +1.5	-6.9	-6.7
	2015	-5.6 ~ +5.6	+0.8	+0.9	-9.2 ~ +1.2	-1.6	-1.1	-5.0 ~ +7.4	+1.9	+1.7

a, arithmetic mean

b, mass-weighted mean

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Table 3. Dry deposition fluxes of water-soluble nitrogen species.

		NH ₄ ⁺ (μmol N m ⁻² d ⁻¹)			NO ₃ ⁻ (μmol N m ⁻² d ⁻¹)			WSO _N (μmol N m ⁻² d ⁻¹)		
		Range	Mean ^a	Mean ^b	Range	Mean ^a	Mean ^b	Range	Mean ^a	Mean ^b
ECSs (Sea fog)	2014	19.7 ~ 67.2	38.1	37.8	277 ~ 1931	926	951	19.7 ~ 446	129	128
	2015	2.2 ~ 48.7	10.9	9.8	52.1 ~ 412	161	151	5.9 ~ 339	88.5	74.1
NWPO (Dust)	2014	8.2 ~ 14.1	11.9	11.9	136 ~ 250	172	170	~ 14.6	6.5	5.5
	2015	1.8 ~ 12.4	5.0	5.3	60.0 ~ 218	98.8	101	5.0 ~ 69.7	27.3	29.7
NWPO (Bgd.)	2014	1.4 ~ 21.1	4.7	4.5	11.0 ~ 287	44.6	43.3	~ 23.4	7.6	7.5
	2015	0.6 ~ 2.6	1.4	1.3	4.9 ~ 60.6	21.9	23.1	0.9 ~ 34.8	10.9	11.6

a, arithmetic mean

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b, volume-weighted mean

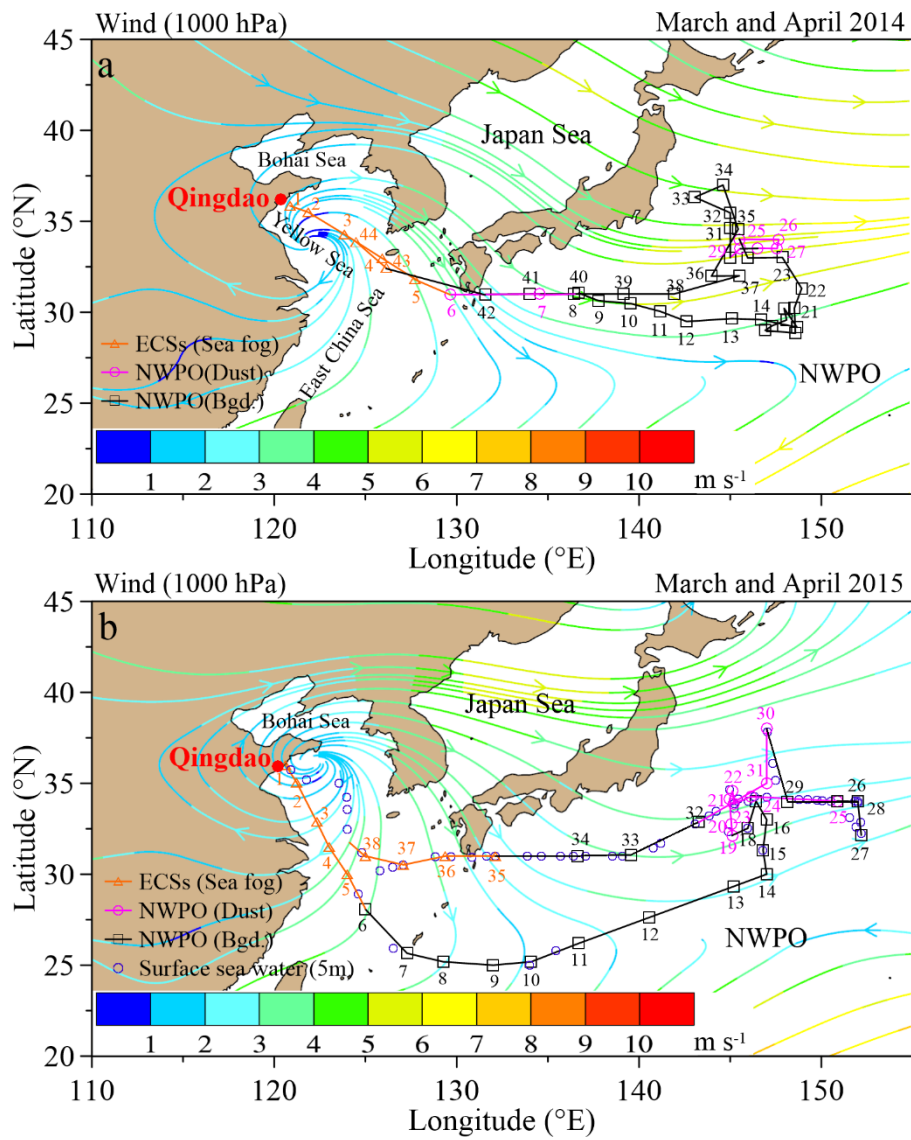


Figure 1. Regional wind streamlines (in m s^{-1}) at 1000 hPa during the Asian winter monsoon period (a, March and April in 2014 and b, March and April in 2015) based on an NCEP dataset. Cruise tracks are also shown (orange, pink, and black indicate sea fog, dust, and background aerosol, respectively). The aerosol number and collection range are shown in orange, pink, and black for sea-fog modified, dust, and background aerosol, respectively. The blue open circles in (b) indicate the locations of surface sea water sample (at 5 m depth) during the 2015 cruise.

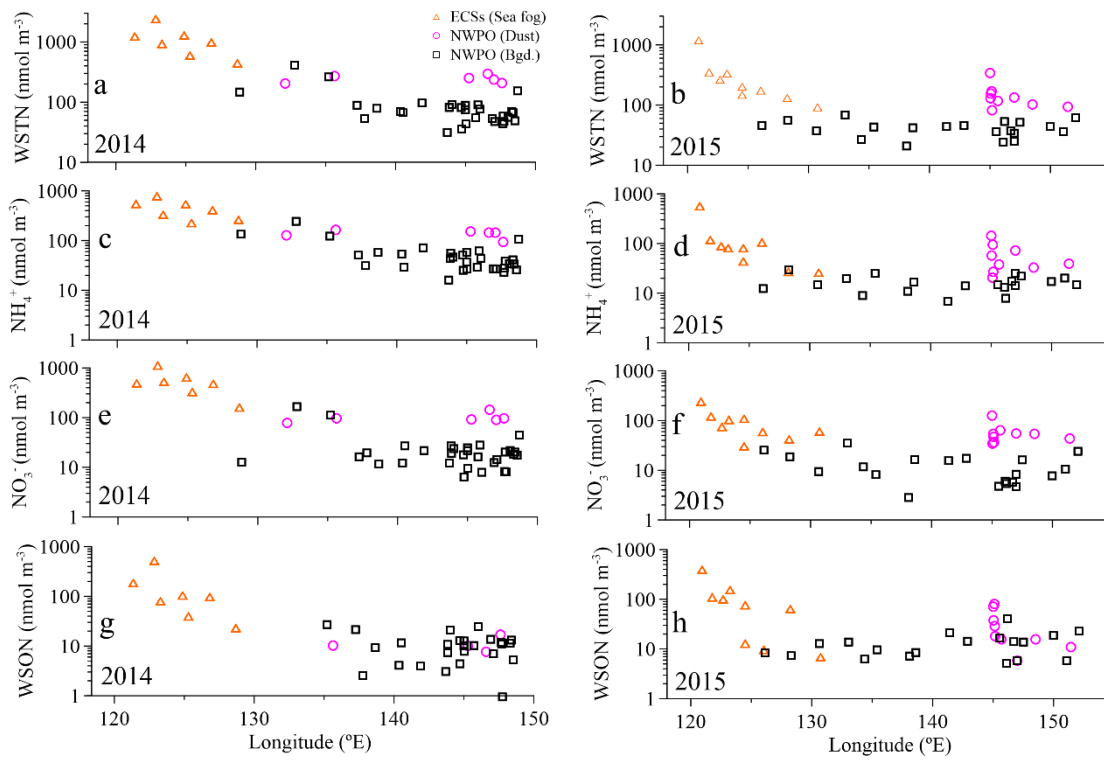


Figure 2. Concentrations of aerosol WSTN (a and b), NH_4^+ (c and d), NO_3^- (e and f), and WSON (g and h) with longitude for the 2014 and 2015 cruises. The orange open triangles denote sea-fog-modified aerosol in the ECSs, the pink circles denote dust aerosol, and the black open squares denote background aerosol in the NWPO.

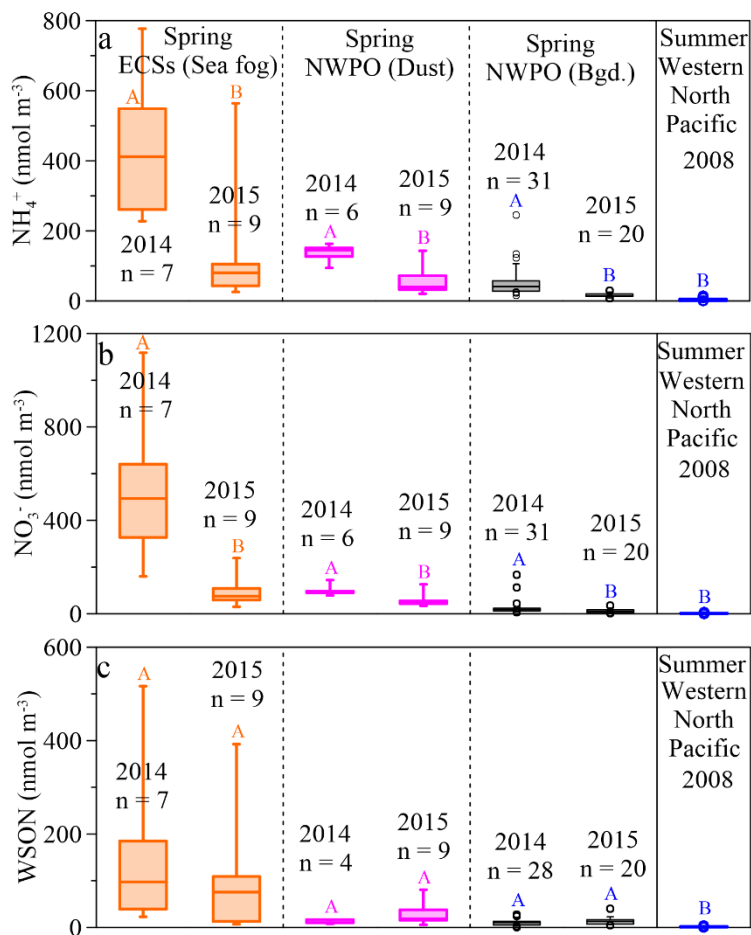


Figure 3. Box plots for spring concentrations of aerosol NH_4^+ (a), NO_3^- (b) and WSON (c) in the ECSs (the orange boxes denote sea-fog-modified aerosol) and NWPO (the pink boxes denote dust aerosol and the black boxes denote background aerosol), and summer aerosol (the blue box denotes NH_4^+ and NO_3^- from Miyazaki et al. (2011) and Jung et al. (2013) and the WSON from Miyazaki et al. (2011)). The large boxes represented the interquartile range from the 25th to 75th percentile, the line inside the box indicates the median value, and the whiskers extend upward to the 90th and downward to the 10th percentiles. Significant differences at the $p < 0.05$ level between different years are marked with coloured uppercase letters.

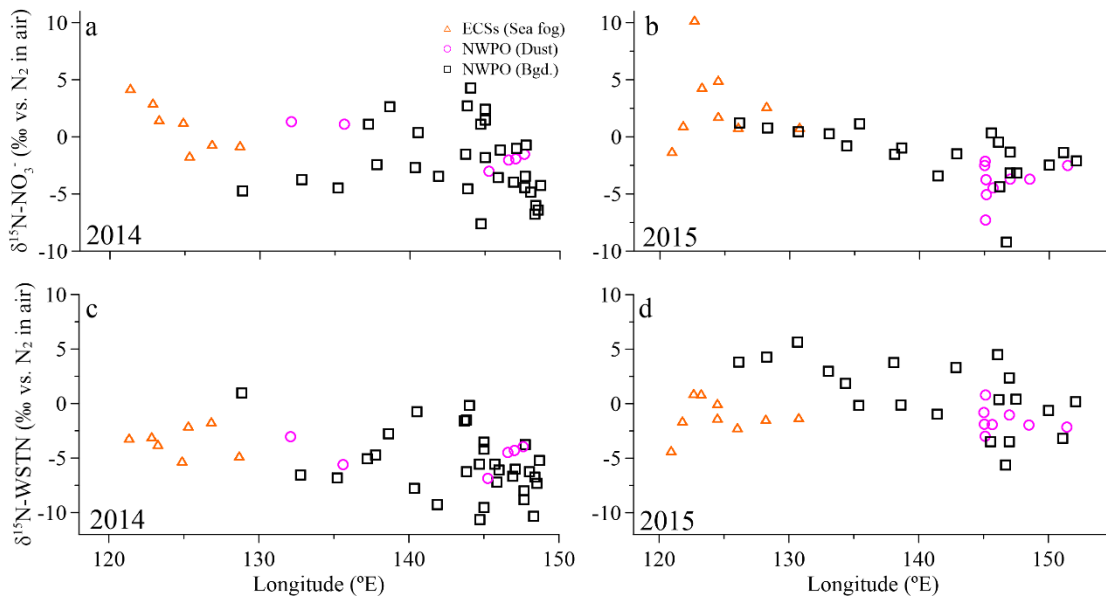


Figure 4. Aerosol $\delta^{15}\text{N-NO}_3^-$ (a and b) and $\delta^{15}\text{N-WSTN}$ (c and d) in the ECSs (the orange open triangles denote for sea-fog-modified aerosol), and in the NWPO (the pink open circles denote dust aerosol and the black open squares denote background aerosol) with longitude in the 2014 and 2015 cruise, respectively.

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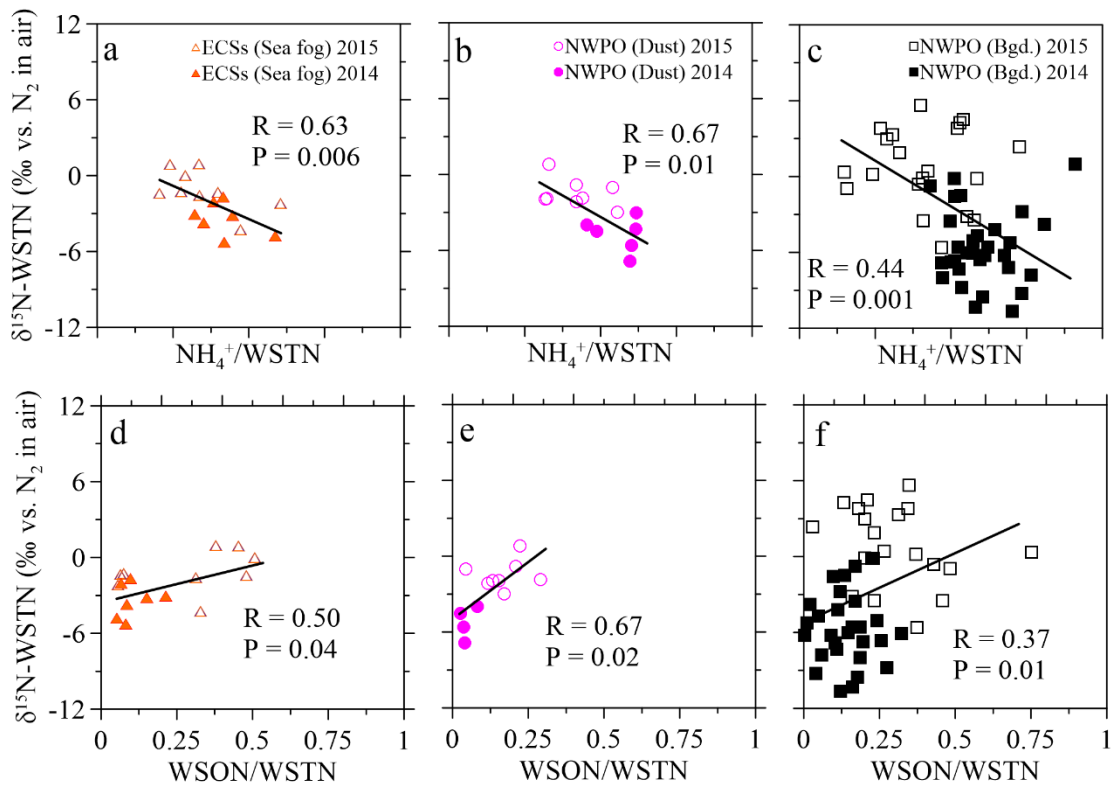


Figure 5. Scatter plots of $\delta^{15}\text{N-WSTN}$ against the $\text{NH}_4^+/\text{WSTN}$ ratio in (a) aerosol sampled in the ECSs, (b) dust aerosol and (c) background aerosol collected in the NWPO. Scatter plots of aerosol $\delta^{15}\text{N-WSTN}$ against the WSON/WSTN ratio in the (d) ECSs, (e) dust aerosol and (f) background aerosol in the NWPO. The solid and open symbols indicate aerosol sampled in 2014 and 2015, respectively.

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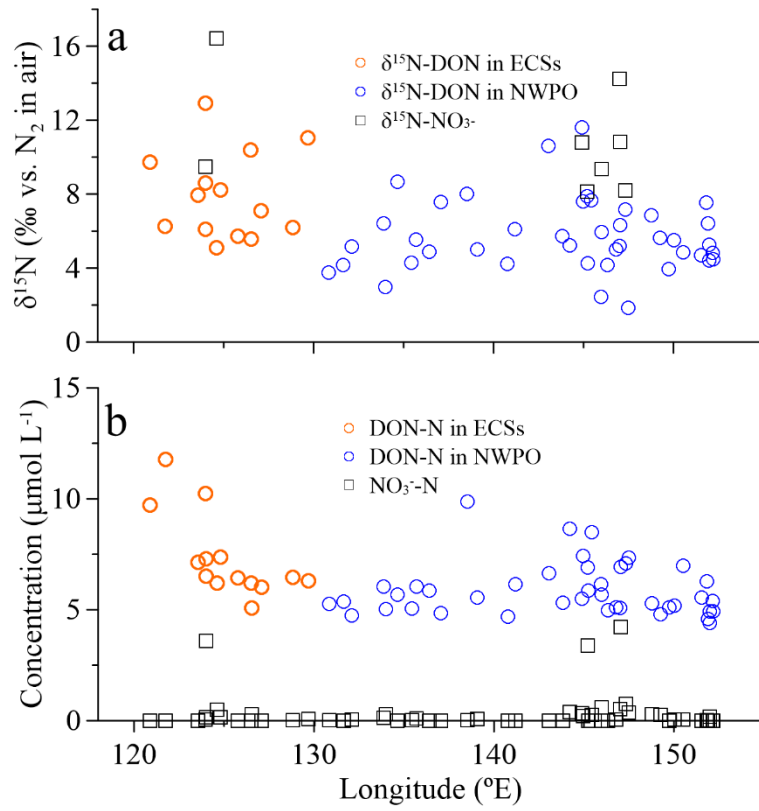


Figure 6. (a) $\delta^{15}\text{N}$ -DON (open circles) and $\delta^{15}\text{N}$ - NO_3^- (black squares), and (b) concentrations of DON (open circles) and NO_3^- (black squares) in SSW with longitude during the 2015 cruise.

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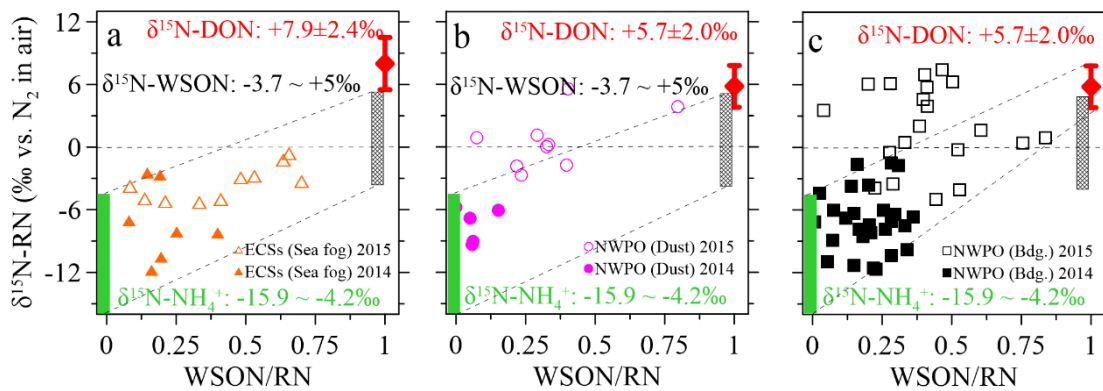


Figure 7. Scatter plots of aerosol $\delta^{15}\text{N-RN}$ against the WSON/RN ratio in the (a) ECSs, (b) dust aerosol in the NWPO, and (c) background aerosol in the NWPO. The green bar indicates the sources of anthropogenic, terrestrial, and oceanic $\delta^{15}\text{N-NH}_4^+$, the gray bar indicates the sources of terrestrial and anthropogenic $\delta^{15}\text{N-WSON}$, and the red bar indicates the $\delta^{15}\text{N-DON}$ in SSW.

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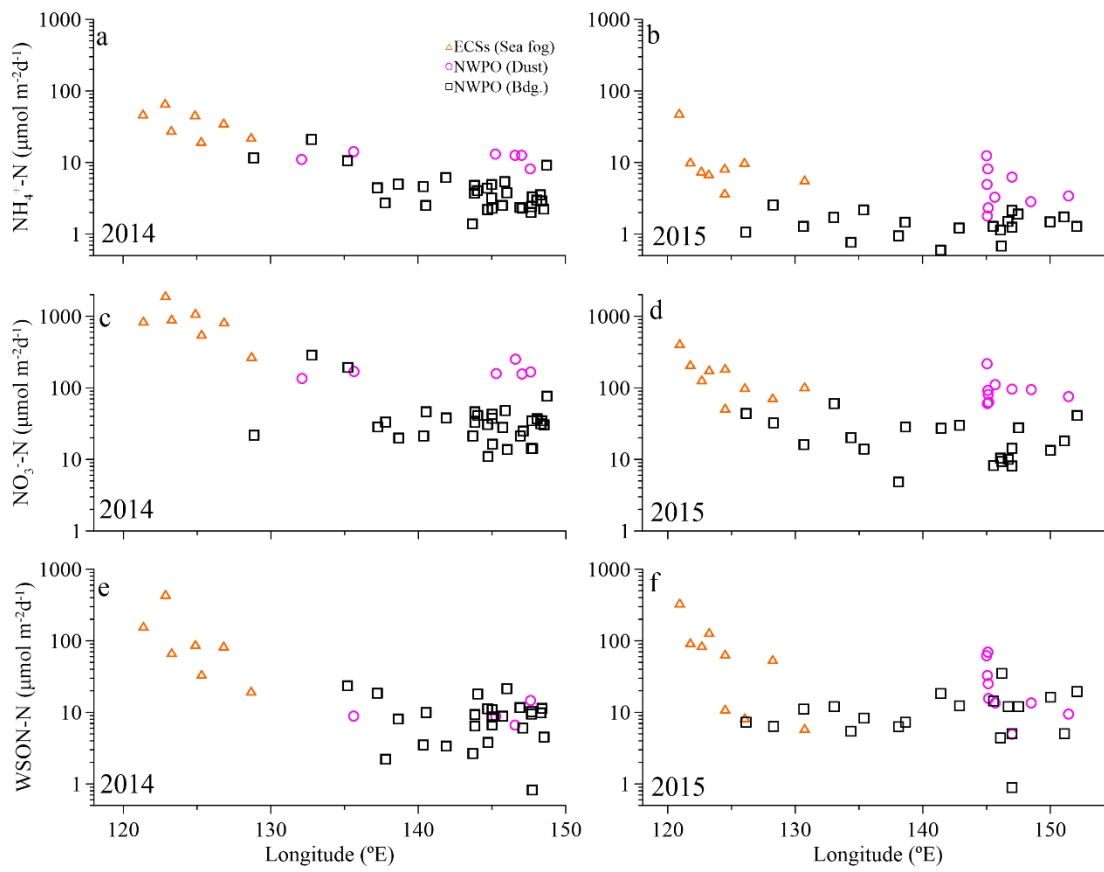


Figure 8. Dry deposition of aerosol $\text{NO}_3^-\text{-N}$ (a and b), $\text{NH}_4^+\text{-N}$ (c and d) and WSON-N (e and f) in the ECSs (orange open triangles) and in the NWPO (pink open circles for dust aerosol and black open squares for background aerosol) along longitude during the 2014 and 2015 cruise, respectively.