

RESPONSES TO REVIEWERS' COMMENTS

General Response to the Reviewers:

We appreciate your comments on the manuscript entitled “Using different assumptions of aerosol mixing state and chemical composition to predict CCN concentrations based on field measurements in urban Beijing”. All comments were valuable and helped us improve our manuscript. Below are point-by-point responses to the reviewers’ comments.

This is an interesting and important study, where a closure between directly measured and calculated CCN concentrations is presented. Overall, the authors did a good job addressing the issues from the previous submission. However, several minor issues have remained. I suggest a minor revision before the paper can be accepted.

In the abstract, some of the abbreviations are left undefined. For instance, it takes some effort to realize that 'EI-SR' stands for 'external-internal size-resolved', as none of those words appear in the abstract.

21 Re: thanks a lot for the comments. 'EI-SR' is defined as “an assumption of a size
22 dependent chemical composition for which sulfate, nitrate, secondary organic aerosols
23 and aged black carbon are internally mixed with each other but externally mixed with
24 primary organic aerosol and fresh black carbon (EI-SR scheme)”. We have revised
25 this in the abstract.

26 In CCN equations, a spherical particle geometry is used and hence no
27 assumptions about the type of particle diameter needs to be made.
28 However, the instrument, in general, produce different diameters:
29 mobility diameter from SMPS, volume (or mass) equivalent diameter
30 from SP2, and aerodynamic diameter from AMS. For spherical
31 particles, this makes no difference, but not all particles are spheres.
32 For instance, significant fractions of fresh and aged BC are present in
33 Beijing air. Fresh BC particles are not spheres and neither are some of
34 the partially aged BC. This issue must be addressed.

35 Re: As the reviewer commented, the instruments produce different diameters. In this
36 paper, we have unified both the aerodynamic diameter from AMS and volume
37 equivalent diameter from SP2 to be mobility diameter. In addition, actual fresh BC
38 particles are not spheres and neither are some of the partially aged BC, but because
39 both the diameter measured from SP2 and the BC size distribution from the literatures
40 are with assumption of the particles being spheres, the fresh and aged BC in this study
41 are thereby assumed to be spherical particles. This issue has been addressed in the
42 revised manuscript (lines 165-175) .

43 The discussion in lines 159-162 is cryptic and so is the related
44 discussion in lines 435-444. I understand how SP2 underestimates
45 the aged BC particle diameter, but how would it overestimate the
46 diameter? Also, what diameter are we talking about here?

47 Re: thanks a lot for the comments. Because the SP2 measures BC core diameter
48 instead of the diameter of the BC-containing particle, it would overestimate the BC
49 mass concentration of smaller particles. Such overestimation would likely lead to an
50 underestimation of N_{CCN} due to the increased mass fraction of BC of total particles.
51 Note that, here, it is the “BC mass concentration” rather than “diameter” being
52 overestimated.

53 Affiliations must be corrected (letters point to wrong institutions)

54 Re: Revised.

55 Other minor corrections

56 L327: remove ‘While’

57 Re: Revised.

58 L334: replace ‘activation’ with ‘activated’

59 Re: Revised.

60 L339: add ‘respectively’ before ‘S*’

61 Re: Revised.

62 L386: replace ‘most’ with ‘at least some’

63 Re: Revised.

64 L406: replace `slight` with `slightly`

65 [Re: Revised.](#)

66 L408: replace `is` with `are`

67 [Re: Revised.](#)

68 L465: replace `internal-` with `internally-`

69 [Re: Revised.](#)

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81 **Using different assumptions of aerosol mixing state and chemical**
82 **composition to predict CCN concentrations based on field**
83 **measurements in urban Beijing**

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106 **Abstract**

107 Understanding the impacts of aerosol chemical composition and mixing state on
108 cloud condensation nuclei (CCN) activity in polluted areas is crucial for accurately
109 predicting the CCN number concentrations (N_{CCN}). In this study, we predict N_{CCN}
110 under five assumed schemes of aerosol chemical composition and mixing state based
111 on field measurements in Beijing during the winter of 2016. Our results show that the
112 best closure is achieved with an assumption of a size dependent chemical composition
113 for which sulfate, nitrate, secondary organic aerosols and aged black carbon-~~(BC)~~
114 are internally mixed with each other but externally mixed with primary organic
115 aerosol (POA) and fresh ~~black carbon~~BC ([external-internal size-resolved, abbreviated](#)
116 [as](#) EI-SR scheme). The resulting ratios of predicted-to-measured N_{CCN} ($R_{CCN_p/m}$) were
117 0.90–0.98 under both clean and polluted conditions. Assumption of an internal
118 mixture and bulk chemical composition (INT-BK scheme) shows good closure with
119 $R_{CCN_p/m}$ of 1.01–1.16 under clean conditions, implying that it is adequate for CCN
120 prediction in continental clean regions. On polluted days, assuming the aerosol is
121 internally mixed and has a chemical composition that is size dependent (INT-SR
122 scheme) achieves better closure than the INT-BK scheme due to the heterogeneity and
123 variations in particle composition at different sizes. The improved closure achieved
124 using the EI-SR and INT-SR assumptions highlight the importance of measuring
125 size-resolved chemical composition for CCN predictions in polluted regions. N_{CCN} is

126 significantly underestimated (with $R_{CCN_p/m}$ of 0.66–0.75) when using the schemes of
127 external mixtures with bulk (EXT-BK [scheme](#)) or size-resolved composition (EXT-SR,
128 [scheme](#)), implying that primary particles experience rapid aging and physical mixing
129 processes in urban Beijing. However, our results show that the aerosol mixing state
130 plays a minor role in CCN prediction when the κ_{org} exceeds 0.1.

131 **1 Introduction**

132 Atmospheric aerosol particles can serve as cloud condensation nuclei (CCN) and,
133 in turn, affect the optical and microphysical properties of clouds ([Twomey, 1977](#);
134 [Albrecht, 1989](#); [Charlson et al., 1992](#)). Additionally, an increase in the aerosol number
135 concentration may suppress precipitation in shallow clouds and promote it in deep
136 convective clouds ([Rosenfeld et al., 2008](#); [Li et al., 2011](#)). A key challenge to
137 understanding indirect aerosol effects is quantifying CCN spectra and their spatial and
138 temporal variations.

139 The ability of particles to act as CCN mainly depends on their size, chemical
140 composition, and mixing state ([McFiggans et al., 2006](#); [Dusek et al., 2006](#); [Ma et al.,](#)
141 [2013](#)). The impacts of the size distribution and chemical composition on CCN activity
142 has been discussed in previous studies ([Dusek et al., 2006](#), [Ervens et al., 2007](#);
143 [Broekhuizen et al., 2006](#); [Yum et al., 2005, 2007](#); [Wiedensohler et al., 2009](#); [Deng et](#)
144 [al., 2013](#); [Zhang et al., 2014, 2016](#); [Kawana et al., 2016](#)). The effect of chemical
145 composition can be represented by a hygroscopicity parameter (κ) ([Petters and](#)
146 [Kreidenweis, 2007](#)) that is often used to predict N_{CCN} ([Moore et al., 2012](#); [Zhang et al.,](#)

147 2014). However, particle composition may vary from single species to a mixture of
148 multiple species for a given size. A description of size-resolved chemical composition
149 thus leads to a better prediction of N_{CCN} because it allows variation of κ with size
150 (Medina et al., 2007; Wang et al., 2010; Meng et al., 2014). Variations in mixing state
151 also impact N_{CCN} prediction, with the effect dependent on the hygroscopicity of the
152 organic component (Wang et al., 2010). The assumption of internal mixtures has been
153 demonstrated to predict N_{CCN} well (Ervens et al., 2007; Chang et al., 2007; Andreae
154 and Rosenfeld, 2008; Gunthe et al., 2009; Rose et al., 2008; Meng et al., 2014; Zhang
155 et al., 2014; Li et al., 2017). However, some studies have shown that detailed
156 information about the chemical composition and the mixing state is required because
157 of the complexity of the hygroscopicity of organics (Broekhuizen et al., 2006; Bhattu
158 and Tripathi, 2015) and the differences in the CCN activity between fresh and aged
159 aerosols (Gunthe et al., 2011). Therefore, the impact of different assumptions
160 concerning the mixing state and chemical composition on accurately quantifying CCN
161 concentrations needs further investigation, especially in heavily polluted regions.

162 Beijing, a typical polluted city, frequently experiences severe haze pollution
163 episodes (Sun et al., 2013; Guo et al., 2014; Zheng et al., 2015), particularly in winter.
164 Several recent studies have focused on studying particle hygroscopicity (Wu et al.,
165 2016; Wang et al., 2017) and chemical composition (Gunthe et al., 2011), and using
166 bulk κ to predict CCN in Beijing (e.g., Liu et al., 2014; Zhang et al., 2017). However,
167 to our knowledge, a comprehensive CCN closure test considering chemical
168 composition and mixing state is lacking for this polluted urban area. In particular, the

169 transformation of the particle mixing state may be very quick during severe pollution
170 conditions (Wu et al., 2016). During pollution events, the hygroscopicity of organics
171 and the CCN activity are often enhanced rapidly with the aging process (Gunthe et al.,
172 2011; Kawana et al., 2016). Therefore, the characterization and parameterization of
173 CCN activation may be more challenging in polluted regions due to the impacts of
174 organics (Wang et al., 2010; Meng et al., 2014; Che et al., 2016; Zhang et al., 2016).

175 In this study, we use size-resolved measurements of CCN activity and
176 size-resolved chemical composition information to predict N_{CCN} using field
177 measurement data collected in Beijing during the winter of 2016. The CCN closure
178 study is carried out using five schemes with different assumptions of particle mixing
179 state and chemical composition. By classifying the data into three different periods
180 (nighttime, noontime, and the evening rush hour), we also investigate the variations in
181 aerosol mixing state from fresh to relatively aged aerosols. The sensitivity of
182 predicted N_{CCN} to the particle mixing state and organic volume fraction with the aging
183 of organic particles is also presented in the last section of the study.

184 **2 Measurements and data**

185 Data used here were measured from 15 November to 14 December 2016
186 during the Air Pollution and Human Health (APHH) field campaign at the Institute of
187 Atmospheric Physics (IAP), Chinese Academy of Sciences (39.97°N, 116.37°E),
188 which is a typical urban site with influences from traffic and cooking emissions (Sun
189 et al., 2015). The sampling instruments were placed in a container at ground level.

190 The particle number size distribution (PNSD) was measured by a Scanning
191 Mobility Particle Sizer (SMPS; Wang et al., 2003). The SMPS consists of a
192 differential mobility analyzer (DMA; model 3081, TSI Inc.) and a condensation
193 particle counter (CPC; model 3772, TSI Inc.). Measurements of size-resolved CCN
194 efficiency spectra were made by an integrated system combining the SMPS (Wang et
195 al., 2003) and a Droplet Measurement Technologies CCN counter (DMT-CCNc;
196 Lance et al., 2006). The procedure to couple the SMPS and the DMT-CCNc
197 developed by Moore et al. (2010) was followed. Atmospheric particles were sampled
198 from an inlet located 1.5 m above the roof of the container and were then passed
199 through a silica gel desiccant drying tube and into the SMPS. The relative humidity
200 of the sample flow was below 30%. The sample flow exiting the DMA was divided
201 into 0.5 lpm for the CCNc and 0.5 lpm for the CPC. Before and after the field
202 campaign ammonium sulfate was used to calibrate the supersaturation (SS) levels of
203 the CCNc with longitudinal temperature differences of 2, 3, 5, 8, 10, 13, and 15 K as
204 shown in Fig. S1. Based on this calibration, the five effective SS levels were 0.12,
205 0.14, 0.23, 0.40, and 0.76%.

206 The PNSD spanned the size range of 10–550 nm with a measurement scan time
207 of 5 min. Total particle or condensation nuclei (CN) size distributions were calculated
208 with the multiple charge correction and transfer function used in the TSI-AIM
209 software. The CN number concentration (N_{CN}) is the total aerosol number
210 concentration and is obtained by integrating the PNSD over the size range of 10–550
211 nm. The full measurement cycle of the CCNc for the five SS levels took one hour (20

212 min for 0.12% and 10 min for each higher SS). Size-resolved CCN efficiency data
213 were inverted with a multiple charge correction (Moore et al., 2010). The CCN
214 number size distribution was calculated by multiplying the CCN efficiency spectrum
215 by the particle number size distribution. The total CCN concentration was then
216 calculated by integrating the size-resolved N_{CCN} . The bulk activation ratio (AR) was
217 calculated as N_{CCN}/N_{CN} . The results were stratified between polluted and background
218 conditions with an assumed threshold PM1 mass concentration of $50 \mu\text{g m}^{-3}$.

219 An Aerodyne High-Resolution Time-of-Flight Aerosol Mass Spectrometer
220 (HR-ToF-AMS; DeCarlo et al., 2006) was housed in a sampling room on the rooftop
221 of a two-story building to measure size-resolved non-refractory submicron aerosols,
222 including organics, sulfate, nitrate, ammonium, and chloride with a time resolution of
223 ~5 min. More details about the HR-ToF-AMS and the measurement site have been
224 described in previous studies (Sun et al., 2010; Sun et al., 2016). The organics are
225 classified by using Positive Matrix Factorization (PMF) (Paatero and Tapper, 1994),
226 considering as being composed of two components: primary organic aerosol (POA)
227 representing non-hygroscopic particles ($\kappa=0$) and secondary organic aerosol (SOA)
228 representing hygroscopic particles. The first ~~component factor~~ consists mainly of is
229 hydrocarbon-like organic aerosol (HOA) ~~which is considered~~, a surrogate of primary
230 organic aerosol (POA) from local urban combustion sources. ~~And t~~The size
231 distribution of the primary OA HOA was ~~calculated~~ measured from by the estimated
232 size-distribution of the C_4H_9^+ fragment ~~which is generally dominated by HOA~~ (Aiken
233 et al., 2009; Zhang et al., 2005). The size distribution of the SOA was calculated is-

234 ~~estimated~~ as the difference between ~~the those of~~ total OA and ~~PH~~OA.

235 The black carbon (BC) mass concentration was measured using a
236 seven-wavelength aethalometer (AE33, Magee Scientific Corp.). [Zhao et al. \(2017\)](#)
237 provides details about this instrument and the measurements it makes. Due to an
238 absence of size-resolved BC measurements, the BC size distribution was calculated
239 from the combination of an approximately lognormal distribution measured by a
240 single particle soot photometer (SP2, DMT) ([Wu et al., 2017](#)) and the total BC mass
241 concentration. Note that because the SP2 measures BC core diameter instead of the
242 diameter of the BC-containing particle, it would overestimate the BC mass
243 concentration of smaller particles but underestimate that of the larger ones. [Such](#)
244 [overestimation would likely lead to an underestimation of \$N_{CCN}\$ due to the increased](#)
245 [mass fraction of BC of total particles.](#) –The uncertainty of this effect is evaluated in
246 Section 4.3. ~~The fresh and aged BC size distributions are determined from the total~~
247 ~~BC size distribution measured by the SP2 (Wu et al., 2017) and from the dependence~~
248 ~~of the fraction of internally mixed soot (F_{in}) on particle diameter (D_p) observed in~~
249 ~~urban Beijing by Cheng et al. (2012). The instruments produce different diameters. In~~
250 ~~this paper, we have unified both the aerodynamic diameter from AMS and volume~~
251 ~~equivalent diameter from SP2 to be mobility diameter. In addition, actual fresh BC~~
252 ~~particles are not spheres and neither are some of the partially aged BC, but because~~
253 ~~both the diameter measured from SP2 and the BC size distribution from the literatures~~
254 ~~are with assumption of the particles being spheres, the fresh and aged BC in this study~~
255 ~~are thereby assumed to be spherical particles. And the fresh and aged BC are all~~

256 assumed to be spherical particles. Due to the different diameters producing by the
257 instrument, the aerodynamic diameter from AMS and volume equivalent diameter
258 from SP2 are adjusted to be mobility diameter.

259 **3 Theory**

260 **3.1 Calculation of CCN concentration using κ -Köhler theory**

261 In this study, we used the critical or cutoff particle diameter (D_{cut}) and particle
262 number size distribution to calculate N_{CCN} . The method to derive D_{cut} is based on
263 κ -Köhler theory (Petters and Kreidenweis, 2007), with the water vapor saturation ratio
264 over the aqueous solution droplet S given by:

$$265 \quad S = \frac{D^3 - D_p^3}{D^3 - D_p^3(1 - \kappa)} \exp\left(\frac{4\sigma_w M_w}{RT\rho_w D}\right), \quad (1)$$

266 where D is the droplet diameter, D_p is the dry diameter of the particle, M_w is the
267 molecular weight of water, σ_w is the surface tension of pure water, ρ_w is the density of
268 water, R is the gas constant, and T is the absolute temperature. When $\kappa > 0.1$ it can be
269 approximately expressed as:

$$270 \quad \kappa = \frac{4A^3}{27D_p^3 \ln^2 S_c}, \quad (2)$$

$$271 \quad A = \frac{4\sigma_w M_w}{RT\rho_w}, \quad (3)$$

272 where S_c is the particle critical supersaturation. The other variables in the equations

273 are: $T = 298.15 \text{ K}$, $R = 8.315 \text{ J K}^{-1} \text{ mol}^{-1}$, $\rho_w = 997.1 \text{ kg m}^{-3}$, $M_w = 0.018015 \text{ kg mol}^{-1}$,
274 and $\sigma_w = 0.072 \text{ J m}^{-2}$ (Rose et al., 2008).

275 For internally-mixed particles, κ is calculated as follows (Petters and
276 Kreidenweis, 2007; Gunthe et al., 2009):

$$277 \quad \kappa_{chem} = \sum_i \varepsilon_i \kappa_i, \quad (4)$$

$$278 \quad \kappa_{org} = f_{POA} \cdot \kappa_{POA} + f_{SOA} \cdot \kappa_{SOA}, \quad (5)$$

279 where κ_i and ε_i are the hygroscopicity parameter and volume fraction for the
280 individual components in the mixture, and f_{POA} and f_{SOA} are the primary organic
281 aerosol (POA) and secondary organic aerosol (SOA) mass fractions in the mixture.
282 The Aerosol Mass Spectrometer (AMS) mainly measured the particle mass size
283 distributions of SO_4^{2-} , NO_3^- , NH_4^+ and organic compounds, while the
284 Zdanovskii-Stokes-Robinson relation requires the volume fractions of the particle
285 chemical composition (Stokes and Robinson, 1966; Zdanovskii, 1948). A simplified
286 ion pairing scheme is used to calculate the mass concentrations of the inorganic salts,
287 which includes only NH_4NO_3 and $(\text{NH}_4)_2\text{SO}_4$ as possible salts (Gysel et al., 2007). In
288 this study, we considered five components: NH_4NO_3 , $(\text{NH}_4)_2\text{SO}_4$, SOA, POA, and BC.
289 The $\kappa_{(\text{NH}_4\text{NO}_3)}$ is equal to 0.67 and $\kappa_{(\text{NH}_4)_2\text{SO}_4}$ is equal to 0.61 (Petters and Kreidenweis,
290 2007; Gunthe et al., 2009). The κ_{org} is estimated using the linear function derived by
291 Mei et al. (2013a), namely, $\kappa_{org} = 2.10f_{44} - 0.11$, where f_{44} is dependent upon organics
292 oxidation level. The mean κ_{org} is 0.10 in our case. The organics are classified ~~into by~~

293 ~~using Positive Matrix Factorization (PMF; Paatero and Tapper, 1994)~~, and considered
294 ~~to be composed of~~ two factor components: POA representing non-hygroscopic
295 particles ($\kappa = 0$) and SOA representing hygroscopic species. In our study, the average
296 contributions of POA and SOA to total organics were 0.53 and 0.47, respectively. On
297 the basis of equation (5), $\kappa_{(\text{SOA})}$ is assumed to be 0.2. Also, $\kappa_{(\text{BC})}$ is assumed to be 0.

298 **3.2 Assumptions about mixing state and chemical composition**

299 To examine the influence of the mixing state and chemical composition on
300 CCN activation, five assumptions (Fig. 1) are used to predict N_{CCN} . Although the
301 assumption of completely internal or external mixing for ambient aerosols represents
302 two extremely simplified schemes and may be atmospherically unrealistic, it allows
303 us to understand the importance of the particle mixing state for predicting N_{CCN} . In
304 addition, size independent and dependent compositions are derived from the mass
305 concentrations of different species measured by the AMS so that the impact of
306 chemical composition on CCN activity can be examined. A detailed introduction of
307 the five assumption schemes follows.

308 **Assumption 1: internal mixture with bulk chemical composition (INT-BK)**

309 In this scheme, submicron particles are assumed to be internally mixed with bulk
310 chemical composition, where the mass fraction of each component (e.g. NH_4NO_3 ,
311 $(\text{NH}_4)_2\text{SO}_4$, SOA, POA, and BC) is uniform throughout the full size range as shown in
312 Fig. 1a. The overall κ is calculated from the bulk chemical composition measured by

313 the AMS based on the simple mixing rule (Equation 4) to obtain the critical diameter
 314 at a given SS. For calculating N_{CCN} all (and only) particles with diameters greater than
 315 D_{cut} are considered CCN-active. The total N_{CCN} is then calculated from the step-wise
 316 integration of the PNSD for $D_p > D_{cut}$. The equations used in the calculations are as
 317 follows,

$$318 \quad CCN_{pre} = \int_{D_{cut}}^{D_{end}} n(\log D_p) d \log D_p \quad (6)$$

$$319 \quad D_{cut} = \sqrt[3]{\frac{4A^3}{27 \sum_i \varepsilon_i \kappa_i \ln^2 S_c}} \quad (7)$$

320 where D_{cut} is the critical diameter, D_{end} is the upper size limit of the PNSD, $n(\log D_p)$
 321 is the function of the aerosol number size distribution, i is the chemical component
 322 element, and the other parameters are the same as those presented in Equations (2), (3)
 323 and (4).

324 **Assumption 2: internal mixture with size-resolved chemical composition**

325 **(INT-SR)**

326 For this scheme submicron particles are assumed to be internally mixed and the
 327 chemical composition is size-dependent as shown in Fig. 1d. The fractional
 328 contributions of the components at each size bin are derived from mass size
 329 distributions of the five species considered, i.e., NH_4NO_3 , $(NH_4)_2SO_4$, SOA, POA,
 330 and BC.

331 For this assumption, the critical diameter is derived from the total hygroscopic
 332 parameter, κ , at each size bin, j . For each size bin for which $D_{p,j}$ is $>$ than the
 333 calculated $D_{cut,j}$ the activated fraction was assumed to be 1.0 and for all others it was
 334 0.0. The N_{CCN} is calculated as follows:

$$335 \quad CCN_{pre} = \int_{D_{begin}}^{D_{end}} n(\log D_p) d \log D_p \quad (8)$$

$$336 \quad D_{cut,j} = \sqrt[3]{\frac{4A^3}{27 \sum_i \varepsilon_{ij} \kappa_{ij} \ln^2 S_c}} \quad (9)$$

337 where D_{begin} and D_{end} are the first and last diameters of the PNSD, $n(\log D_p)$ is
 338 the function of the aerosol number size distribution, i is the chemical component
 339 element, j is the PNSD size bin, and the other parameters are the same as those
 340 presented in Equations (2), (3) and (4).

341 **Assumption 3: external mixture with bulk chemical composition (EXT-BK)**

342 For this scheme the submicron aerosol is treated as an external mixture. This
 343 means that there are five types of particles, i.e., NH_4NO_3 , $(NH_4)_2SO_4$, SOA, POA, and
 344 BC, and each particle consists of a single component. The volume fraction of each
 345 component, which is derived from bulk mass concentrations, does not vary with size
 346 (as shown in Fig. 1b).

347 At a given S , the critical diameter of each particle type is retrieved from the κ of
 348 each component. The N_{CCN} of each aerosol type is calculated as the CCN-active

349 particle number concentration multiplied by the bulk volume fraction of the
 350 components as expressed in Equation (10). The N_{CCN} of the five particle types are
 351 finally summed to obtain the total N_{CCN} . The specific equations are as follows,

$$352 \quad CCN_{pre} = \sum_i \left(\int_{D_{icut}}^{D_{end}} n(\log D_p) d \log D_p * V_i \right) \quad (10)$$

$$353 \quad D_{cut,i} = \sqrt[3]{\frac{4A^3}{27\kappa_i \ln^2 S_c}} \quad (11)$$

354 where $D_{cut,i}$ is calculated for each component, i , at a given SS, V_i is the volume
 355 fraction of each aerosol type, $n(\log D_p)$ is the function of the aerosol number size
 356 distribution, i is the chemical component element, and the other parameters are the
 357 same as those presented in Equations (2), (3) and (4).

358 **Assumption 4: external mixture with size-resolved chemical composition**

359 **(EXT-SR)**

360 As with the EXT-BK scheme the same five particle types are considered and
 361 their relative concentrations selected to match the measured composition. But unlike
 362 with the EXT-BK scheme the relative concentrations of the five particle types vary
 363 with particle size to capture the size-dependence of the measured composition, as is
 364 depicted in Fig. 1e. The volume fraction of each particle type at each size is first
 365 multiplied by the total particle number size distribution (PNSD) to get the $PNSD_i$ of
 366 each aerosol type. The N_{CCN} of each particle type is then obtained from the step-wise

367 integration of the PNSD_{*i*} for D_p > D_{cut,*i*}, and then summed to get the total N_{CCN}, as
 368 described by Equation (12). Similar to EXT-BK, the critical diameter of each particle
 369 type is also derived from the κ of each pure component at a given S.

$$370 \quad CCN_{pre} = \sum_i \left(\int_{D_{begin}}^{D_{end}} (n(\log D_p) * V_{ij}) d \log D_p \right) \quad (12)$$

$$371 \quad D_{cut,i} = \sqrt[3]{\frac{4A^3}{27\kappa_i \ln^2 S_c}} \quad (13)$$

372 where V_{*i*} is the volume fraction of each particle type in a size bin, n (log D_p) is the
 373 function of the aerosol number size distribution, *i* is the chemical component element,
 374 *j* is the particle size bin, and the other parameters are the same as those presented in
 375 Equations (2), (3) and (4).

376 **Assumption 5: sulfate, nitrate, SOA and aged BC internally mixed, and POA and**
 377 **fresh BC externally mixed, and all components with size-resolved chemical**
 378 **composition (EI-SR)**

379 At each particle size sulfate, nitrate, and SOA with BC-aged are treated as
 380 internally mixed, but POA and BC-fresh are present in separate particles and are
 381 non-hygroscopic. As with INT-SR and EXT-SR the chemical composition is
 382 size-dependent, as shown in Fig. 1c. The EI-SR scheme likely represents a case that is
 383 most similar to that of actual atmospheric aerosols in locations such as Beijing. The
 384 fresh and aged BC size distributions are determined from the total BC size

385 distribution measured by the SP2 (Wu et al., 2017) and from the dependence of the
 386 fraction of internally mixed soot (F_{in}) on particle diameter (D_p) observed in urban
 387 Beijing by Cheng et al. (2012).

388 In this assumption the fresh BC and POA particles can serve as CCN only if their
 389 diameter is larger than 200 nm; otherwise they are CCN-inactive. Thus, the total N_{CCN}
 390 of those externally mixed components (N_{CCN_EXT}) is calculated from the step-wise
 391 integration of the product of the PNSD and the volume fraction of the fresh BC and
 392 POA in each size bin larger than 200 nm.

393 The N_{CCN} of the remaining components (sulfate, nitrate, and SOA with BC-aged)
 394 that are treated as an internal mixture, denoted as N_{CCN_INT} , is predicted in the same
 395 way as for the INT-SR scheme, with the only difference being that the PNSD is first
 396 multiplied by the volume fraction of the mixed component particles for each size bin.
 397 The total N_{CCN} is thus calculated as the sum of N_{CCN_EXT} and N_{CCN_INT} . The
 398 specific equations are as follows,

$$399 \quad CCN_{pre} = \int_{D_{begin}}^{D_{200}} (n(\log D_p) * r_j) d \log D_p + \int_{D_{200}}^{D_{end}} n(\log D_p) d \log D_p \quad (14)$$

$$400 \quad D_{cut,j} = \sqrt[3]{\frac{4A^3}{27 \sum_i \varepsilon_{ij} \kappa_{ij} \ln^2 S_c}} \quad (15)$$

401 where D_{begin} and D_{end} are the first and last diameters of the PNSD, $n(\log D_p)$ is the
 402 function of the aerosol number size distribution, r is the volume fraction of the
 403 internal (hygroscopic) mixture at each size, i is the chemical component element, j is

404 the particle size bin, and the other parameters are the same as those presented in
405 Equations (2), (3) and (4).

406 **4 Results and discussion**

407 **4.1 Diurnal variations in aerosol properties**

408 Diurnal variations in mean PNSD and bulk chemical composition under
409 polluted and background conditions are shown in Fig. 2. Significant diurnal variations
410 in PNSD are observed during the campaign. For both polluted and background cases
411 the abrupt increases in concentration of small particles ($D_p < 100$ nm) from
412 1700–2000 local time (LT) are likely related to fresh primary emissions from cooking
413 and traffic sources (Wang et al., 2017; Zhao et al., 2017), which is also evident in the
414 significant increase in mass concentration of non-hygroscopic POA (Fig. 2d and 2e).
415 The peak amplitude in the PNSD that occurs from about 0800 to 1200 LT is probably
416 associated with secondary formation processes, which is indicated by an apparent
417 increase of nitrate, SOA and f_{44} (oxidation level of organics) in the morning (0800 LT)
418 when photochemistry becomes significant. The effect is more apparent on clean days.
419 In addition, the PNSD amplitude and BC and POA concentrations are high at
420 nighttime, suggesting an influence from the diurnal variation of the planetary
421 boundary layer (PBL) height. In particular, on polluted days the PBL plays a key role
422 in regulating the diurnal variation of primary components like POA and BC (e.g.,
423 Dzepina et al., 2009; Cross et al., 2009). ~~While o~~On clean days secondary formation

424 and primary sources play dominant roles in regulating diurnal variations. The PNSD
 425 in clean cases has peaks at smaller D_p (~30–40 nm, Fig. 1c) compared to polluted
 426 cases (~100 nm), which is associated with particle growth accompanying atmospheric
 427 chemistry processes during haze evolution (Guo et al., 2014; Wang et al., 2016).

428 4.2 Cumulative Gaussian distribution function fit and parameters derived from 429 the CCN efficiency

430 The ~~activation~~-activated fractions measured at the five supersaturation levels were
 431 fitted using the following two functions (Rose et al., 2008; Mei et al., 2013b):

$$432 \quad R_a(S) = \frac{E}{2} \cdot \left(1 + \operatorname{erf}\left(\frac{\ln S - \ln S^*}{\sqrt{2}\sigma_s}\right)\right), \quad (16)$$

$$433 \quad f_{N_{CCN}/N_{CN}} = a \left(1 + \operatorname{erf}\left(\frac{D - D_a}{\sigma_a \sqrt{2}}\right)\right), \quad (17)$$

434 where $R_a(S)$ and $f_{(N_{CCN}/N_{CN})}$ are the CCN activation fractions, the maximum activation
 435 fraction (MAF) is equal to E or $2a$, S^* and D_a are the midpoint activation
 436 supersaturation and diameter, respectively, and σ_s and σ_a are the cumulative
 437 distribution function (CDF) standard deviations. During this field campaign, 2580
 438 size-resolved CCN efficiency spectra at five SS levels were measured. To illustrate
 439 the characteristics of the activation spectra, the CDF fits are shown in Fig. 3 and in
 440 Tables S1-2. A gradual increase in size-resolved AR with SS suggests that particles
 441 had different hygroscopicities even at the same diameter. The heterogeneity of particle

442 chemical composition can be represented by the ratio of σ_a and D_a (i.e., σ_a/D_a), where
443 σ_a is the standard deviation derived from the cumulative Gaussian distribution
444 function (Eqn. 12) and D_a is the activation diameter (Rose et al., 2010). The ratio of
445 σ_a/D_a during the three periods is shown in Fig. 3b.

446 **4.2.1 CCN activation curves and heterogeneity of chemical components**

447 For larger particles with $D_p > 100$ nm, no significant differences were observed
448 in the CCN efficiency spectra (Fig. 3a), suggesting a similar hygroscopicity during the
449 three periods. For particles with $D_p < 100$ nm, the CCN efficiency spectrum observed
450 during the evening rush hour period showed a much more gradual increase (with
451 smaller slopes) in size-resolved AR than that derived for the other two periods. This is
452 attributed to the strong influence of POA emissions, which consist of less hygroscopic
453 and externally-mixed smaller particles mainly from cooking and traffic during the
454 evening rush hour period (also indicated by the increased σ_a/D_a). Particles with $D_p <$
455 100 nm emitted during the evening rush hour period require a higher SS to reach the
456 same AR. However, for $D_p > 100$ nm the slope of AR with respect to SS was steep
457 and near the instrumental limit obtained for a pure ammonium sulfate aerosol. Che et
458 al. (2016) have reported that particles larger than about 150 nm have relatively
459 uniform composition. This suggests that particles become more internally mixed with
460 growth from the Aitken mode to the accumulation mode. This feature is also
461 suggested by the decreasing σ_a/D_a with increasing particle diameter.

462 **4.2.2 Mean critical activation diameter**

463 The critical activation diameter at different SS levels under background and
464 polluted conditions is shown in Fig. 4. The difference in critical diameter between
465 polluted and background cases are calculated as $D_p_POL - D_p_BG$. At lower SS levels,
466 the critical diameters for polluted cases were slightly smaller than those observed on
467 clean days, suggesting larger particles are more CCN-active on polluted days. This is
468 expected based on HTDMA measurements that showed that particles in the
469 accumulation mode on polluted days are more hygroscopic than those on clean days
470 in urban Beijing (Wang et al., 2017). At higher SS the critical diameter on polluted
471 days was a little higher than that obtained under clean conditions, suggesting that
472 particles with D_p of ~40 nm are less CCN active. This is likely because a high
473 concentration of small and hygroscopic particles in the Aitken mode arise from the
474 photochemistry-driven nucleation process on clean days. However, in polluted cases,
475 small particles are mostly composed of hydrophobic POA from cooking and traffic
476 sources. This was also observed by Wang et al., (2017) who showed that 40 nm
477 particles are less hygroscopic on polluted days. However, the differences in critical
478 diameter between polluted and background cases are small, reflecting a relatively
479 minor influence of hygroscopicity on CCN activity.

480 **4.2.3 MAF**

481 As shown in Fig. 5, the maximum activated fractions on clean and polluted days

482 during the campaign are less than 1, which suggests that ~~at least some~~ most sampled
483 aerosols were externally mixed (Gunthe et al., 2011). For example, the MAF for
484 particles with D_p of ~180 nm was around 0.78 at SS = 0.12% under background
485 conditions, indicating that ~22% of the particles are non-hygroscopic. The higher
486 MAFs under polluted conditions suggest a more internally mixed aerosol (Wu et al.,
487 2016; Wang et al., 2017). The MAF during the 1200–1400 LT period was highest,
488 which is likely due to strong photochemical aging processes that lead to more internal
489 mixing of the aerosol.

490 **4.3 CCN closure study and the sensitivity of predicted N_{CCN} to assumed aerosol** 491 **mixing state and chemical composition**

492 Fig. 6 shows the comparisons between predicted and measured N_{CCN} at different
493 SS levels under background and polluted conditions. The ratios of
494 predicted-to-measured N_{CCN} ($R_{CCN_p/m}$) ranged from 0.66 to 1.16, suggesting
495 significant influences of the different assumptions on CCN prediction. The EI-SR
496 assumption scheme predicts N_{CCN} very well, with $R_{CCN_p/m}$ of 0.90–0.98
497 (corresponding to a slight underestimation of 2-10%). For the EI-SR scheme,
498 hydrophobic POA and a portion of the BC are assumed to be externally mixed while
499 the other species (sulfate, nitrate, SOA and aged BC) are assumed to be internal
500 mixtures. The assumption is physically sound, and the result just implies that the
501 EI-SR represents well the actual mixing state and compositions of the particles. The
502 slight underestimation may due to an overestimation of fresh BC caused by the

503 method (see Section 3.1) that we used to retrieve it. Also, a slight^{ly} larger
504 underestimation of N_{CCN} for BG case in EI-SR scheme showed in Figure 6 may
505 suggest that aerosols during clean periods ^{are} mostly aged and internal-mixed.

506 The INT-SR and INT-BK schemes that assume the aerosol is internally mixed also
507 predict N_{CCN} reasonably well at lower SS. The prediction is better on background days,
508 reflecting the more homogenous aerosol composition in clean conditions. With
509 increasing SS this overestimation became more pronounced, which is likely due to
510 limitations of the AMS measurements. The AMS distributions show that the mass
511 concentration was most impacted by particles with diameters near ~100–400 nm.
512 Because particles in that size range tended to be more hygroscopic than those with
513 diameters < 100 nm, this leads to an overestimation of κ (underestimation of the
514 critical diameter) and a resulting overestimation of N_{CCN} at high SS. With decreasing
515 SS the critical diameter increased and the deviation using the INT-BK and INT-SR
516 schemes decreased. Detailed explanations about this effect have been given by Wang
517 [et al. \(2010\)](#) and Zhang [et al. \(2017\)](#). Overall, the INT-BK and INT-SR schemes
518 achieve CCN closure within what is deemed here an acceptable overprediction of
519 0-16%. The EXT-BK and EXT-SR schemes underestimated N_{CCN} , with $R_{CCN_p/m}$ of
520 0.66-0.75.

521 Overall, the internal-mixing schemes achieve much better closure than do those
522 assuming external mixtures. Our results suggest that freshly-emitted particles in
523 urban Beijing may experience a quick conversion and mixing with pre-existing

524 secondary particles, e.g. converting from externally mixed to internally mixed (or
525 from hydrophobic to hydrophilic, along with a decrease in the volume of POA and BC)
526 as reported previously (Riemer et al., 2004; Aggarwal and Kawamura, 2009; Jimenez
527 et al., 2009; Wu et al., 2016; Peng et al., 2016). In summary, under background
528 conditions, the INT-BK scheme achieved the best CCN closure, implying that the
529 INT-BK assumption is likely sufficient to predict CCN in clean continental regions.
530 However, in polluted regions, the EI-SR and INT-SR schemes may achieve better
531 closure.

532 As mentioned in Section 2.2, because the SP2 measures BC core diameter and
533 not the diameter of the BC-containing particle, the method would overestimate the BC
534 mass concentration of smaller particles but underestimate that of the larger ones. This
535 effect adds uncertainty to the CCN prediction when using the EXT-SR scheme and is
536 evaluated here (Fig. 7). For the evaluation, we predict N_{CCN} with the retrieved fresh
537 BC size distribution only in the EXT-SR scheme, which represents an upper limit of
538 the overestimation of the fresh BC size distribution due to the SP2 measurement.
539 Therefore, the result represents the largest underestimation of N_{CCN} caused by the
540 BC-containing particle effect. Our result shows that the underestimation of N_{CCN} is
541 reduced from 28% to 25% by changing the total BC size distribution to that of just the
542 fresh BC. That means that the overestimation of fresh BC due to the BC-containing
543 particle effect in the SP2 measurements would lead to a maximum underestimation of
544 3% of N_{CCN} . The minimal uncertainty contributed by uncertainty in the BC size
545 distribution could be explained by the small fractional contribution of BC to the total

546 particle concentration. In conclusion, such an effect is quite small or negligible
547 compared to the overall large underestimation of N_{CCN} with the EXT-SR assumption.

548 **4.4 Performance of the five schemes at different times of the day**

549 To investigate the performance of the five schemes at different times of the day,
550 the diurnal variations in the $R_{CCN_p/m}$ (SS = 0.23%) derived by the schemes are shown
551 in Fig. 8. In general, the INT-BK, INT-SR, and EI-SR schemes can predict N_{CCN}
552 well during all periods of the day under polluted or background conditions. $R_{CCN_p/m}$
553 values are within the acceptable $\pm 20\%$ uncertainty range (Wang et al., 2010; Zhang
554 et al., 2017). Compared with other periods, the predicted N_{CCN} during the morning
555 and evening rush hour periods showed the most sensitivity to the different assumption
556 schemes, especially on clean days (Fig. 8b). For example, the $R_{CCN_p/m}$ derived using
557 the INT-SR schemes reaches values up to >1.2 , and the $R_{CCN_p/m}$ obtained using the
558 EXT-BK scheme decreased to a minimum value of ~ 0.5 . The INT-SR, INT-BK and
559 EI-SR assumptions overestimate N_{CCN} for the evening rush hour period by up to
560 $\sim 20\%$. This may be because most freshly emitted POA and BC particles during
561 evening traffic hours are hydrophobic and do not contribute to the N_{CCN} . In addition,
562 for EIS assumption, a portion of BC is assumed aged and internally-mixed with
563 sulfate, nitrate and SOA, as may reduce the actual fraction of fresh BC during rush
564 hour period and thereby lead to an overestimation of N_{CCN} .

565 Use of the EXT-BK or EXT-SR assumption for the polluted case resulted in a
566 predicted N_{CCN} that was underestimated by $\sim 30\text{-}40\%$ at night (0000–0600 LT).

567 Expectedly, the prediction using the two schemes improved during the daytime and
568 evening rush hours, e.g., the $R_{CCN_p/m}$ changed from about 0.6 to 0.8. This is likely
569 associated with heavy urban traffic emissions/residential cooking sources during the
570 daytime that lead to more externally-mixed particles under polluted conditions; while,
571 at night, the particles are less influenced by those local primary sources (Zhao. et al.,
572 2017). Wang et al. (2017) showed that the probability density function of κ during
573 rush hour has a bimodal distribution and a hydrophobic mode from locally-emitted
574 particles. This also leads to reasonably accurate estimates of N_{CCN} during nighttime
575 with larger error during the daytime when using the internal mixing assumptions
576 (INT-BK, INT-SR and EI-SR) for polluted cases (Fig. 8).

577 **4.5 Impact of mixing state and organic volume fraction on predicted N_{CCN} and** 578 **their variation with aerosol aging**

579 To further examine the sensitivity of predicted N_{CCN} to the particle mixing state
580 and organic volume fraction with the aging of organic particles, the relative deviation
581 between N_{CCN} predicted assuming internal and external mixtures as a function of κ_{org}
582 was calculated, with the results shown in Fig. 9. The schemes that assume internal and
583 external mixtures use bulk composition of organics, sulfate, and nitrate, which
584 simplifies the analysis and interpretation of the results. For the data collected
585 throughout the field campaign, the organic volume fraction is categorized as <50%,
586 50-60%, and >60%. The deviation between the concentrations predicted assuming
587 internal and external mixtures is calculated as $[(N_{CCN,INT-BK} - N_{CCN,EXT-BK})$

588 ($N_{CCN,EXT-BK}^{-1}$]. The result shows that the relative deviation increased as the
589 organic volume fraction increased. For organic volume fractions less than 50% the
590 maximum difference can only reach up to 20% (SS=0.76%). This is consistent with
591 previous studies that reported differences less than 20% when $x_{org} < 30\%$
592 (Sotiropoulou et al., 2006; Wang et al., 2010). The maximum deviation approaches to
593 100% for x_{org} of >60% at SS = 0.76%. Overall, the deviation is largest when the
594 organics are less or non-hygroscopic, i.e., when $\kappa_{org} < 0.05$. The deviation decreased
595 rapidly as κ_{org} increased to 0.05 in all cases. For κ_{org} of 0.1 the differences were less
596 than 20%, even with high organic fractions. Moreover, differences were 10% or less
597 for κ_{org} of 0.15, suggesting that the mixing state plays a minor role when κ_{org} exceeds
598 0.1.

599 **5 Conclusions**

600 In this study, we have investigated the importance of aerosol chemical
601 composition and mixing state on CCN activity based on measurements made during a
602 field campaign carried out in Beijing in the winter of 2016. The N_{CCN} was predicted
603 by applying κ -Köhler theory and using five schemes that assume different mixing
604 state and chemical composition combinations.

605 We show that there is a significant impact of the different assumptions on CCN
606 prediction, with $R_{CCN,p/m}$ ranging from 0.66 to 1.16. The best estimates of N_{CCN} under
607 both background and polluted conditions were obtained when using the EI-SR scheme,
608 with a resulting $R_{CCN,p/m}$ of 0.90–0.98. Under background conditions, the INT-BK

609 scheme also provided reasonable estimates, with $R_{CCN_p/m}$ ranging from 1.00–1.16.
610 This suggests that the INT-BK assumption is likely sufficient to predict CCN in clean
611 continental regions. On polluted days, the EI-SR and INT-SR schemes are believed to
612 achieve better closure than the INT-BK scheme due to the heterogeneity in particle
613 composition across different sizes. The improved closure obtained using the EI-SR
614 and INT-SR assumptions highlights the importance of knowing the size-resolved
615 chemical composition for CCN prediction in polluted regions. The EXT-SR and
616 EXT-BK schemes markedly underestimate N_{CCN} on both polluted and clean days,
617 with an $R_{CCN_p/m}$ of 0.66–0.75. The diurnal variations in $R_{CCN_p/m}$ show that the
618 predicted N_{CCN} during the evening rush hour period is most sensitive to the mixing
619 state assumptions. The $R_{CCN_p/m}$ ranged from ~ 0.5 to ~ 1.2 , reflecting the impact from
620 evening traffic and cooking sources (both with large amounts of hydrophobic POA).
621 But we also find that the particle mixing state plays a minor role when κ_{org} exceeds
622 0.1, even with a high organic fraction.

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629

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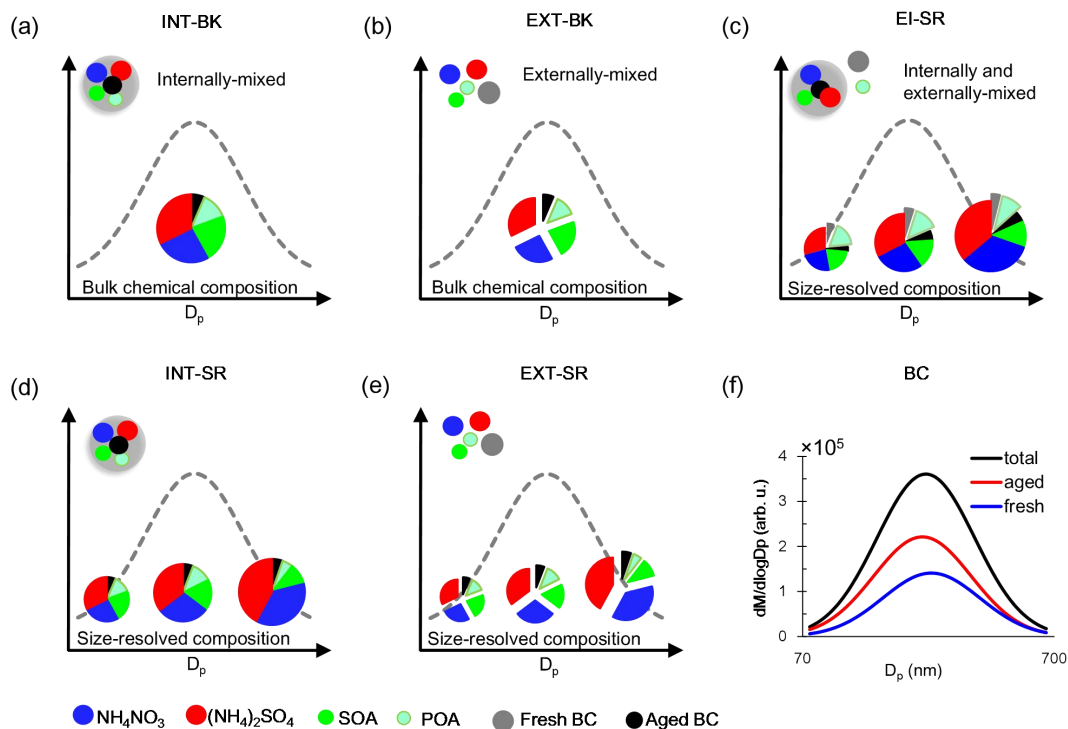
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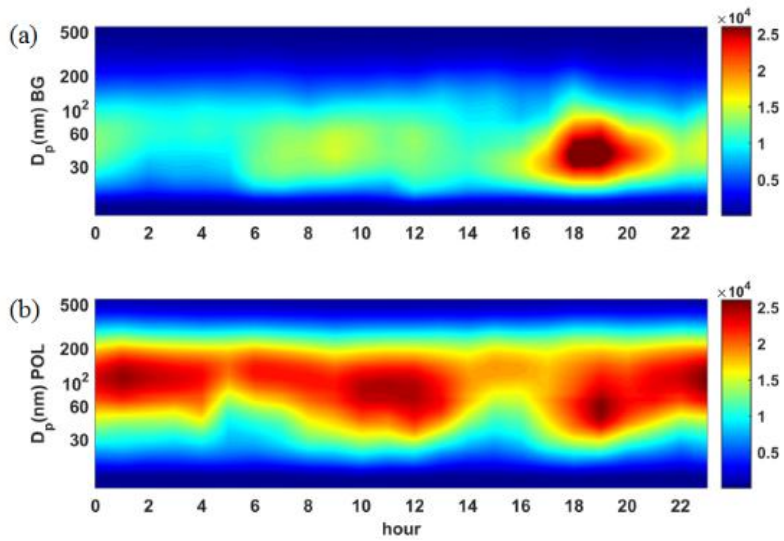
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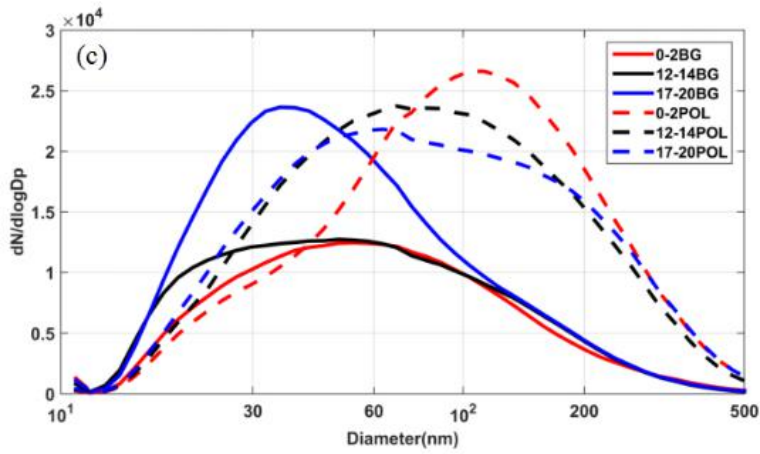
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881 **Figures**

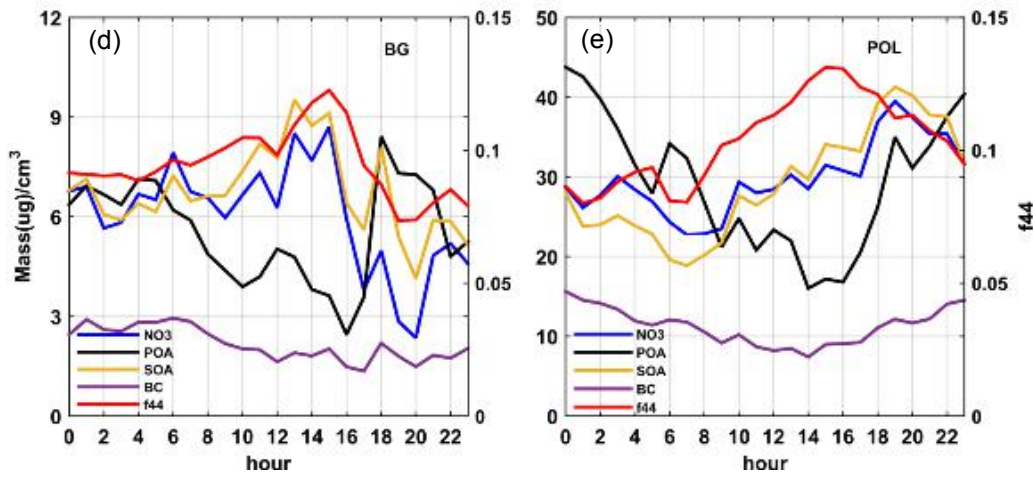




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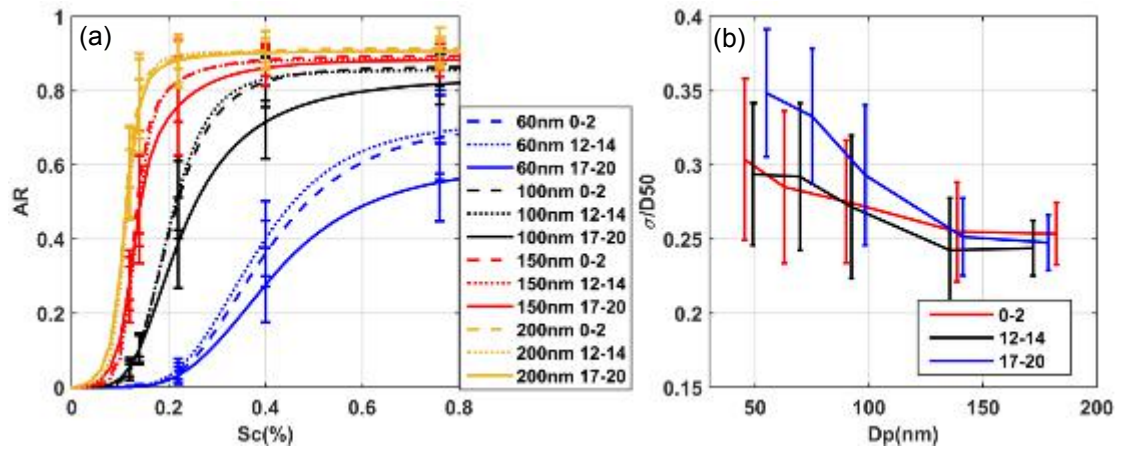


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897 **Figure 2.** Diurnal variations in aerosol properties at the IAP site during the APHH
 898 field experiment, including the particle number size distribution measured by the

899 SMPS under (a) background (BG) and (b) polluted (POL) conditions; (c) mean
900 particle number size distribution measured by the SMPS during three periods
901 (0000–0200 LT, 1200–1400 LT, and 1700–2000 LT) under BG and POL conditions;
902 bulk chemical component mass concentrations (NO_3 , POA, SOA, and BC) and f_{44}
903 made under (d) BG and (e) POL conditions.

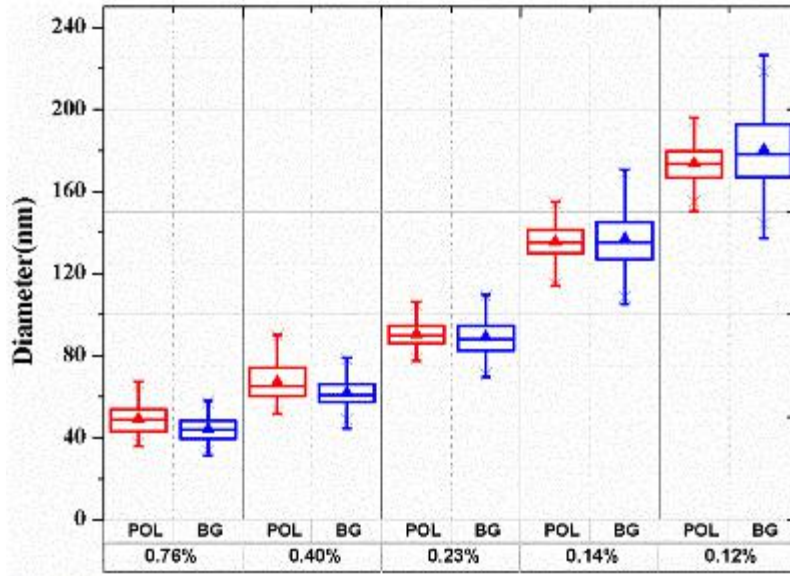
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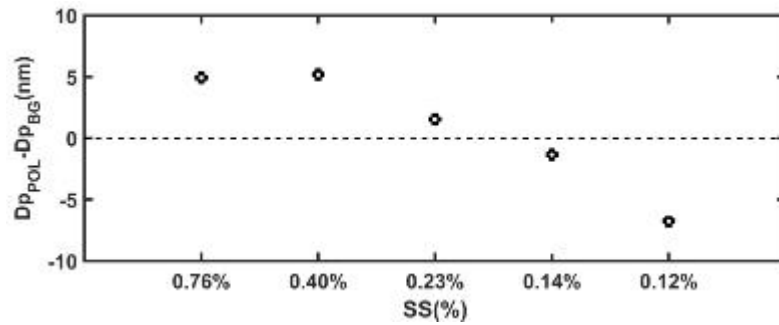
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906 **Figure 3.** (a) Averaged fitted CCN efficiency spectra during the nighttime period
 907 (0000–0200 LT, dashed lines), the noontime period (1200–1400 LT, dotted lines) and
 908 the evening rush hour period (1700–2000 LT, solid lines) for different diameters (60,
 909 100, 150, and 200 nm); (b) the heterogeneity of aerosol particles (σ_a/D_a) derived from
 910 Equation (7) during the three selected periods.

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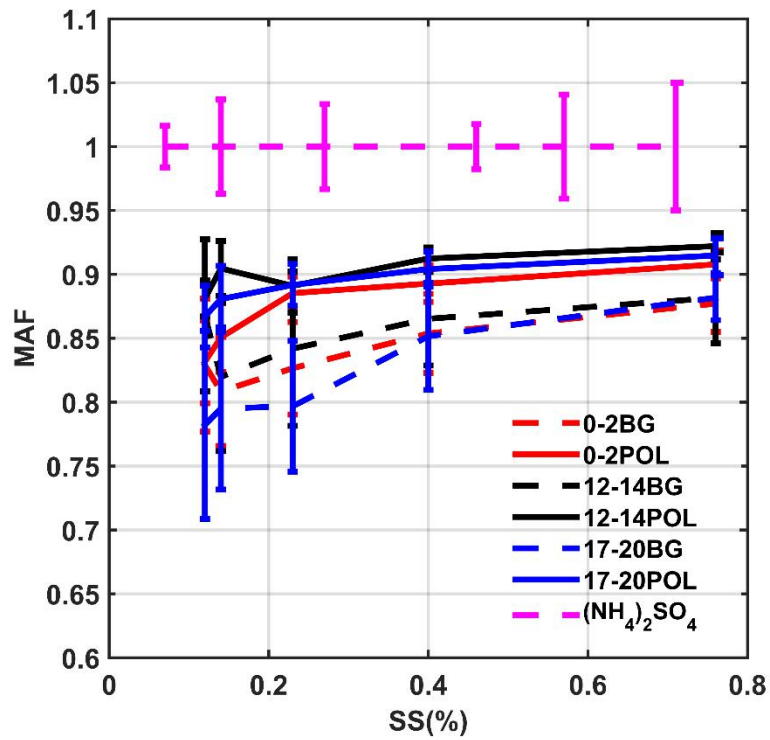


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914 **Figure 4.** Top: Retrieved mean critical activation diameters at SS = 0.12, 0.14, 0.23,
 915 0.40, and 0.76% under background (BG) and polluted (POL) conditions. The box
 916 plots show mean critical activation diameters at the 25th, 50th, and 75th percentiles.
 917 Bottom: Difference in the mean critical activation diameter between BG and POL
 918 cases.

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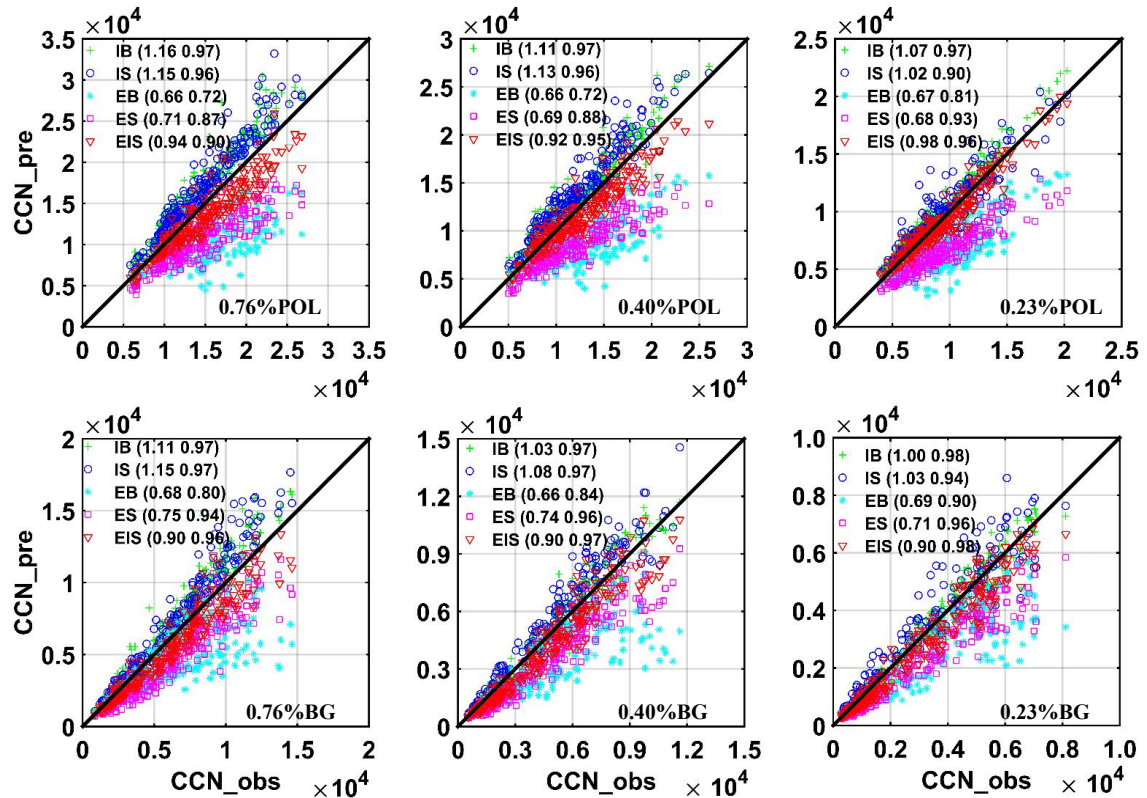
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922 **Figure 5.** Mean maximum active fractions (MAFs) of CCN activation spectra under
 923 polluted (POL) and background (BG) conditions during the three periods, i.e.,
 924 0000–0200 LT, 1200–1400 LT, and 1700–2000 LT. The MAF of pure (NH₄)₂SO₄
 925 particles at the different SS levels (magenta line) is also plotted.

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929 + INT-BK Internal mixture, bulk composition

930 o INT-SR Internal mixture, size-resolved composition

931 * EXT-BK External mixture, bulk composition

932 □ EXT-SR External mixture, size-resolved composition

933 ▽_EI-SR External mixture, POA and BC externally mixed, size-resolved composition

934 **Figure 6.** Predicted N_{CCN} as a function of measured N_{CCN} using the five assumptions

935 (colored symbols) at three supersaturation levels (0.23, 0.40, and 0.76%) under

936 polluted (POL) and background (BG) conditions. The numbers in parentheses are the

937 slope (first number) and the correlation coefficient (second number).

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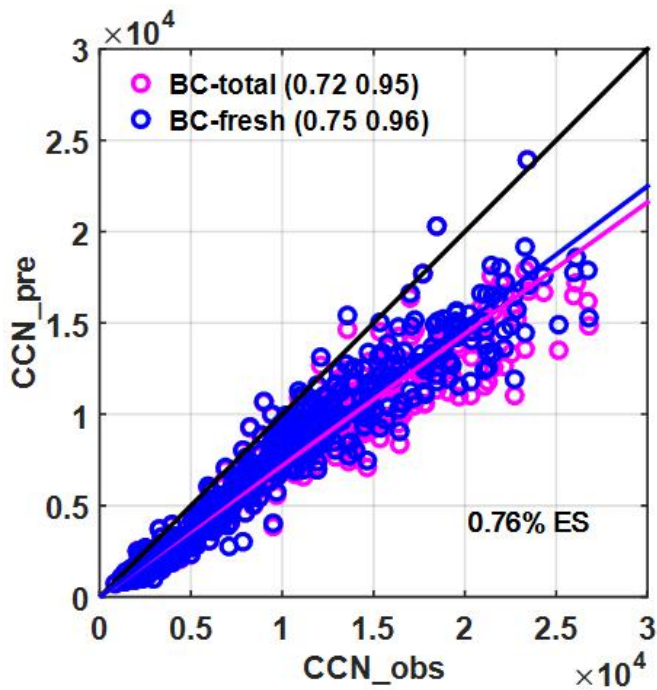
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945 **Figure 7.** Predicted N_{CCN} as a function of measured N_{CCN} using the EXT-SR
946 assumption (colored symbols) at $S=0.76\%$. The pink and blue circles denote the
947 results predicted by using total and fresh BC size distributions, respectively. The
948 numbers in parentheses are the slope (first number) and the correlation coefficient
949 (second number).

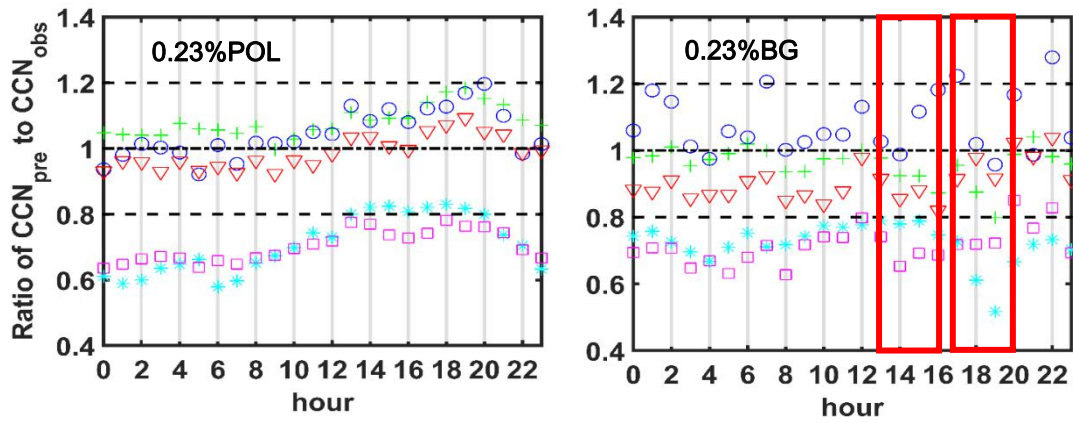
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956 + INT-BK Internal mixture, bulk composition

957 o INT-SR Internal mixture, size-resolved composition

958 * EXT-BK External mixture, bulk composition

959 □ EXT-SR External mixture, size-resolved composition

960 ▽ EI-SR External mixture, POA and BC externally mixed, size-resolved composition

961 **Figure 8.** Diurnal variations in the ratio of predicted-to-measured N_{CCN} at a

962 supersaturation level of 0.23% under background (BG) and polluted (POL)

963 conditions.

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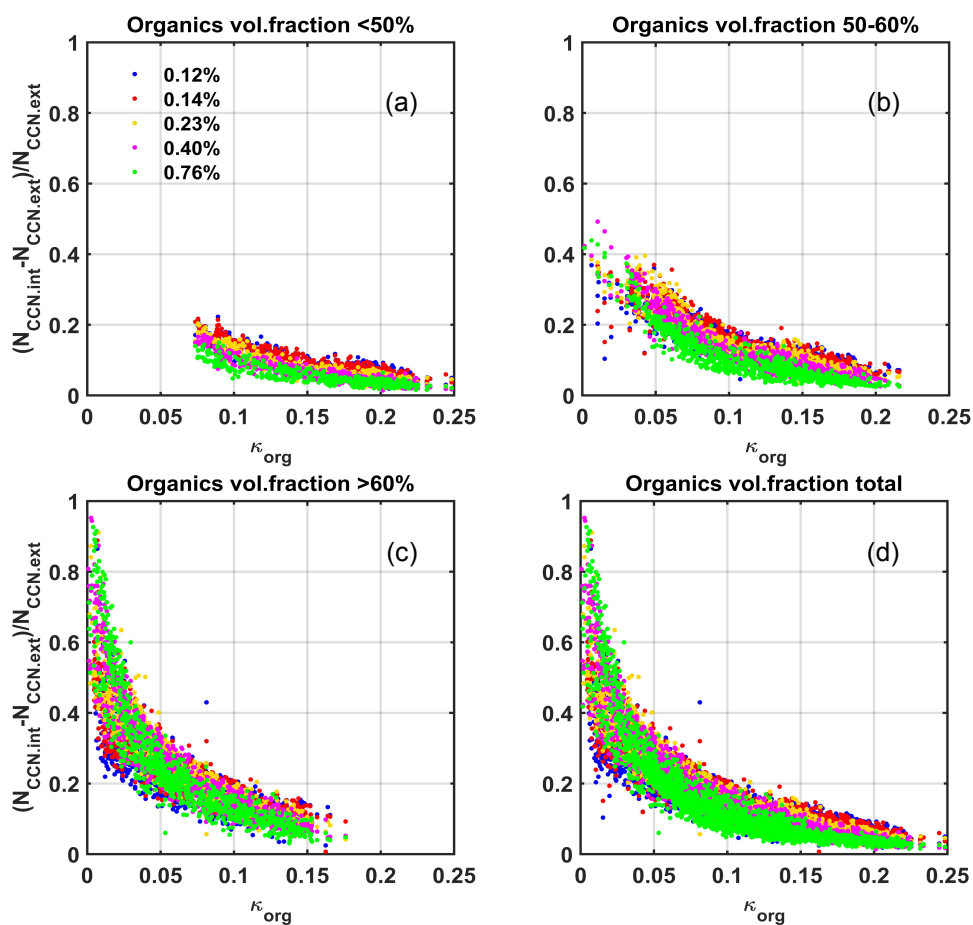
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Figure 9. Relative deviations between N_{CCN} predicted under the assumptions of

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internal (INT-BK) and external (EXT-BK) mixtures $[(N_{CCN,INT-BK} - N_{CCN,EXT-BK})$

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$(N_{CCN,EXT-BK})^{-1}]$ as a function of κ_{org} when organic volume fractions of <50 (a),

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50-60 (b), >60% (c) and all observed data points (d). The solid with different colors

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represent different supersaturation levels. The different colors denote the different

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organic fractions.

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