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- 1 First simultaneous measurements of peroxyacetyl nitrate
- 2 (PAN) and ozone at Nam Co in the central Tibetan Plateau:
- 3 impacts from the PBL evolution and transport processes

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- 5 Xiaobin Xu¹, Hualong Zhang^{1,*}, Weili Lin^{1,2}, Ying Wang¹, and Shihui Jia^{1,**}
- 6 [1]{State Key Laboratory of Severe Weather & Key Laboratory for Atmospheric Chemistry
- 7 of China Meteorological Administration, Chinese Academy of Meteorological Sciences,
- 8 Beijing, China}
- 9 [2]{Meteorological Observation Center, China Meteorological Administration, Beijing, China}
- 10 [*]{now at : Guangdong Meteorological Observatory, Guangzhou, Guangdong, China}
- 11 [**] {now at: School of Environment and Energy, South China University of Technology,
- 12 Guangzhou, Guangdong, China}

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14 Correspondence to: Xiaobin Xu (xuxb@camscma.cn)

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16 Abstract

- 17 Both peroxyacetyl nitrate (PAN) and ozone (O₃) are key photochemical products in the
- 18 atmosphere. Most of the previous in-situ observations of both gases have been made in
- 19 polluted regions and at low altitude sites. Here we present first simultaneous measurements of
- 20 PAN and O₃ at Nam Co (NMC, 90 °57′E, 30 °46′N, 4745 m a.s.l.), a remote site in the central
- 21 Tibetan Plateau (TP). The observations were made during summer periods in 2011 and 2012.
- 22 The PAN concentrations averaged 0.36 ppb (range: 0.11-0.76 ppb) and 0.44 ppb (range: 0.21-
- $23\,$ 0.99 ppb) during 16-25 August 2011 and 15 May to 13 July 2012, respectively. The O_3
- 24 concentration varied from 27.9 ppb to 96.4 ppb, with an average of 60.0 ppb. Profound

diurnal cycles of PAN and O₃ were observed, with minimum values around 5:00 LT, steep

- rises in the early morning, and broader platforms of high values during 9:00-20:00 LT. We
- 27 find that the evolution of planetary boundary layer (PBL) played a key role in shaping the
- 28 diurnal patterns of both gases, particularly the rapid increases of PAN and O₃ in the early

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1 morning. Air entrainment from the free troposphere into the PBL seemed to cause the early 2 morning increase and be a key factor of sustaining the daytime high concentrations of both 3 gases. The days with higher daytime PBL (about 3 km) showed stronger diurnal variations of 4 both gases and were mainly distributed in the drier pre-monsoon period, while those with 5 shallower daytime PBL (about 2 km) showed minor diurnal variations of both gases and were 6 mainly distributed in the humid monsoon period. Episodes of higher PAN levels were 7 observed occasionally at NMC. These PAN episodes were caused either by rapid downward 8 transport of air masses from the middle/upper troposphere or by long-range transport of PAN 9 plumes from North India. The PAN level in the downward transport cases ranged from 0.5 10 ppb to 0.7 ppb and may indicate the PAN abundance in the middle/upper troposphere. In the 11 long-range transport case, the PAN level varied in the range of 0.6-1.0 ppb. This long-range 12 transport process influenced most of the western and central TP region for about a week in 13 early June 2012. Our results suggest that polluted air masses from South Asia can 14 significantly enhance the PAN level over the TP. As PAN act as a reservoir of NO_x, the 15 impacts of pollution transport from South Asia on tropospheric photochemistry over the TP 16 region deserve further studies.

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1 Introduction

Peroxyacetyl nitrate (PAN) and ozone (O₃) are important oxidants in the troposphere. They are toxic for human and vegetation. Tropospheric O₃ contributes significantly to global warming with a radiative forcing of +40 (±0.20) W/m2 (Myhre et al., 2013). Tropospheric O₃ originates mainly from the photochemical reactions within the troposphere and to a lesser extent from the stratosphere (Lelieveld and Dentener, 2000), while PAN in the troposphere is nearly exclusively formed in oxidation of volatile organic compounds (VOCs) in the presence of NO_x (Fischer et al., 2014). PAN is produced in the association reaction between CH₃C(O)O₂ (PA) and NO₂. As one of the key radicals, PA is produced by oxidation of a number of VOCs (Roberts, 2007; LaFranchi et al., 2009; Fischer et al., 2014). Since both VOCs and NO_x are largely emitted by anthropogenic sources, PAN is primarily produced in and downwind of industrial and populated areas. With different lifetimes at different temperatures (Cox and Roffey, 1977), PAN is instable under warm conditions, but stays longer in colder environment, thus is revealed ubiquitous in the mid to upper troposphere (Singh, 1987; Talbot et al., 1999; Russo et al., 2003), ultimately resulting in a global scale

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1 transport. PAN can decompose and release NO₂ when it reaches warm environment,

2 becoming one of the key sources of NO_x in remote areas. This makes PAN an important

3 reservoir of NO₂. Inter-comparisons among models and between model and observation show

4 very large PAN differences in many regions of the atmosphere (Thakur et al., 1999; Sudo et

5 al., 2002; von Kuhlmann et al., 2003; Singh et al., 2007), but confirm the important role for

6 PAN in sustaining O₃ production over remote regions (Hudman et al., 2004; Zhang et al.,

7 2008). Since tropospheric O₃ and OH are principally controlled by the abundance of NO_x,

8 decomposition of PAN would have great implications for the budget of these key atmospheric

9 oxidants. It has been indicated that regional increase of O₃ can be attributed to an

intercontinental and even global transport of PAN (Hudman et al., 2004; Fischer et al., 2011)

11 and most of the conveying paths are in the free troposphere, driving PAN plumes travelling to

remote areas (Roiger et al., 2011; Pandey Deolal et al., 2013). Thus a considerable amount

of PAN has been detected in remote areas with sparse anthropogenic emission (Zanis, 2007).

14 Up to now the main methods to directly obtain the PAN concentration are ground-based and

15 aircraft observations. Although PAN has been measured in a great deal of campaigns during

16 past decades, the observational data of PAN are very inhomogeneously distributed over the

17 world, with most of them being from North America, West Europe, and Pacific region

18 (Fischer et al., 2014). PAN measurements are extremely lacking in many areas over Eurasian

19 continent, northeastern African, Oceanic regions, the Indian Ocean, and the Tibetan Plateau

20 (TP) region.

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21 The TP region covers an area of about 2,500,000 km², with an average elevation of about

22 4000 m above sea level. The world's highest plateau acts as a heat core in summer, heating

the air above and prompting its ascending motion (Yeh et al., 1957). In addition to the thermal

effect, the South Asian monsoon also exerts a convergence effect driving the ascending

25 motion (Chen et al., 2012). Accompanied by the ascending motion, water vapor and air

26 pollutants emitted or formed in the boundary layer can be rapidly transported to the upper

27 troposphere and lower stratosphere (UTLS) (Dessler and Sherwood, 2004; Gettelman and

28 Kinnison, 2004, Fu et al., 2006; Lelieveld et al., 2007; Law et al., 2010). Convective transport

29 over the TP and surrounding areas can be clearly tracked by satellite observations of some

30 longer-lived species, such as CO (Park et al., 2007, 2009), PAN (Ungermann et al., 2016),

31 CH₄ (Xiong et al., 2009) and HCN (Randel et al., 2010). Elevated concentrations of some

32 relatively short-lived anthropogenic pollutants in the UTLS region are also reported (Park et

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- 1 al., 2008, Tian et al., 2008; Gu et al., 2016). Such rapid, upward transport of pollutants and
- 2 water vapor may have great implications on atmospheric composition and climate of regional
- 3 and global scales. Efforts have been made to understand the impacts of upward transport of
- 4 air masses over the TP, among which is the potential relationship with the ozone valley over
- 5 the TP reported by Zhou et al. (1995).
- 6 The TP region is very sparsely populated with nearly no industrial emissions of pollutants.
- 7 Although the TP has been nearly unpolluted, the high altitude and the correspondingly
- 8 intensified UV radiation make it an interesting region for studies of photochemical products,
- 9 such as O₃ and PAN. However, there have been only sparse reports of measurements of O₃
- 10 and related species from the TP mainly due to the poor accessibility and logistic difficulties of
- 11 this vast region. So far, most of the published measurements of O₃ and its precursors over the
- 12 TP have been from sites at the edges of the TP (Ma et al., 2002a, 2002b; Ding and Wang,
- 13 2006; Wang et al., 2006; Zhu et al., 2006; Cristofanelli et al., 2010; Xue et al., 2011; Zheng et
- 14 al., 2011; Ma et al., 2014; Wang et al., 2015b; Xu et al., 2016, 2017). Only three publications
- 15 present measurements of O₃ and related species from sites in the central TP, with one
- 16 reporting data from urban observations (Ran et al., 2014) and two showing results from
- 17 remote sites (Lin et al., 2015; Yin et al., 2017).
- 18 Observational data of PAN from the TP are extremely lacking. The only field observation of
- 19 ambient PAN in the TP was made by Xue et al. (2011), who measured PAN and other
- 20 reactive species at Mt. Waliguan, a global atmosphere watch (GAW) station located at the
- 21 northeast edge of the TP. The average level of PAN was 0.44 (± 0.14) ppb for a two-week
- 22 period in summer 2006. This observation offers a preliminary detection of ambient PAN over
- 23 the northeast TP. So far, there has been no published in-situ measurement of PAN from the
- 24 central TP. In addition to the traditional observation methods, remote sensing techniques can
- also be applied to acquire the global PAN distribution from satellites (Remedios et al., 2007;
- Moore and Remedios, 2010; Wiegele et al., 2012; Tereszchuk et al., 2013; Fadnavis et al.,
- 27 2014). However, the PAN data retrieved from satellite observations need further validations.
- Here we present the first simultaneous measurements of PAN and O₃ at a site in the central
- 29 TP. We study the diurnal variations of observed concentrations and the links to the evolution
- 30 of planetary boundary layer (PBL). We also investigate the vertical and horizontal transport
- 31 and discuss the implications of our measurements.

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1 2 Observations

2 **2.1 Site**

- 3 The observations of PAN and other species were made from 11 July to 31 August 2011 and
- 4 from 15 May to 13 July 2012 at the Nam Co Comprehensive Observation and Research
- 5 Station, Chinese Academy of Sciences (CAS) (NMC, 90 °57'E, 30 °46'N, 4730 m a.s.l.). West
- 6 and north of the NMC site is the Nam Co Lake, with the nearest distance to the lake being
- 7 about 1.5 km. The Nyainqentanglha mountains (about 5000-6800 m a.s.l.) stand south and
- 8 east of the site, with the mountain ridge being more than 20 km distant from the site. The TP
- 9 region has a population density of less than 2 person/km² (http://sedac.ciesin.columbia.
- 10 edu/gpw/). The largest city of Tibet, Lhasa, is about 120 km south from the NMC site, far
- beyond the continuous ridges of the Nyainqentanglha Mountains. The nearest population
- 12 center, Dangxiong township is located about 35 km southeast of the NMC site. The direct
- transport of air pollutants from Lhasa and Dangxiong is hardly possible due to the blocking of
- the high mountain ridges. There is a road about 1.3 km southeast of the NMC site, connecting
- 15 the tourism site of the Nam Co Lake to Dangxiong and the No. 109 National Road. More
- details about NMC and its surrounding can be found in literature (Ma et al., 2011; Lin et al.,
- 17 2015; Yin et al. 2017).

18 2.2 Instruments and data correction

- 19 Ambient PAN was observed using a PAN analyzer (Meteorologie Consult GmbH, Germany),
- 20 which consists of an automated gas chromatograph (GC) equipped with an electron capture
- 21 detector (ECD) and a calibration unit. The equipment is the same one used in previous
- 22 observation in Beijing (Zhang et al., 2014), with identical setup details depicted in the paper.
- 23 In respect of the calibration system, an NO reference gas (4.5 ppm) in nitrogen was
- 24 introduced in and reacts with acetone vapor under the UV irradiation. Under the same
- 25 condition, the PAN standard is produced with the efficiency of $93\% \pm 7\%$ (Volz-Thomas et
- 26 al., 2002). Surface O₃ was simultaneously observed using an O₃ analyzer (TE 49C, Thermo
- 27 Environmental Instruments, Inc., USA), which was regularly calibrated using an O₃ calibrator
- 28 (TE 49C PS) traceable to the Standard Reference Photometer (SRP) maintained by WMO
- 29 World Calibration Centre in EMPA, Switzerland. All instruments were housed in a simply
- 30 constructed one-storey building, located 0.15 km southeast of the station's main building.

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- 1 Meteorological data were collected using automatic meteorological station systems installed
- 2 at different levels on a tower near the observation building.
- 3 Although measurements of PAN have been made previously at some high altitude sites in
- 4 other areas using methods similar to ours (Ford et al., 2002; Fischer et al., 2010; Xue et al.,
- 5 2011; Pandey Deolal et al., 2013), this is the first report of using the GC-ECD instrument for
- 6 PAN measurement under the conditions of a high altitude site over 4700 m a.s.l. During the
- 7 observation period in 2011, the instrument performance was somewhat instable, probably
- 8 affected by the extreme ambient conditions at the site. The variation of environment
- 9 temperature is suspected to have made it hard to keep the ECD inner temperature constant.
- 10 This resulted in abrupt fluctuations in the 10-min chromatographic PAN signals sometimes
- during the measurement period in 2011. The instable performance of ECD caused varying
- detection sensitivity. Normally, we convert PAN signals of air samples to concentration data
- 13 based on ratios of signals to theoretical PAN concentration of the standard gas produced
- during the calibration. However, the jumping sensitivity makes it improper to obtain PAN
- 15 concentrations using the normal method. Thus we applied another method, which we call the
- 16 indirect calibration. Our GC-ECD instrument is optimized for the separation and detection of
- both PAN and CCl₄. Therefore, it is possible to indirectly calculate the PAN concentration,
- 18 i.e., by using the ratio of the PAN to CCl4 signal. Details about the indirect calibration are
- 19 given in the supplement.
- 20 Although the indirect calibration is a viable way to obtain PAN concentrations, the
- 21 uncertainty of final data could be larger primarily due to the two assumptions mentioned in
- 22 the supplement and some technical problems with the observation system. We are more
- 23 confident of the data from 16 to 25 August 2011. During this period, the instruments
- 24 performed well and the two calibrations before and after this period gave almost identical
- 25 sensitivity. In view of this, we report and analyze in this paper mainly data from 16 to 25
- 26 August 2011, together with those obtained from 15 May to 10 July 2012, where our
- 27 instruments worked stably.

28 2.3 Meteorological data and analysis

- 29 Local meteorological variables, including temperature, relative humidity, 3-dimensional
- 30 winds, etc., were observed by corresponding sensors installed at 2 m, 10 m, and 20 m of the
- 31 meteorological tower at the NMC station. The National Centers for Environmental Prediction

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- 1 (NCEP) reanalysis data, together with the local meteorological data, are used in this paper to
- 2 facilitate the interpretation of our PAN and O₃ measurements. Global Data Assimilation
- 3 System (GDAS, 3 hourly, 1°×1° in longitude and latitude, and 26 pressure levels,
- 4 http://ready.arl.noaa.gov/gdas1.php) was obtained from National Oceanic and Atmospheric
- 5 Administration (NOAA) Air Resources Laboratory (ARL). The GDAS data were used as
- 6 input to the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model (V4.9)
- 7 for simulating backward air trajectories ending at 500 m and 1500 m above the NMC site.
- 8 The HYSPLIT model is developed by NOAA/ARL (Draxler and Hess, 1997). In addition,
- 9 NCEP FNL(final) Operational Global Analysis data (6 hourly, 1°×1° in longitude and
- 10 latitude, and 26 pressure levels, http://rda.ucar.edu/datasets/ds083.2/#!description) were
- 11 acquired from National Center for Atmospheric Research (NCAR). These data were used to
- 12 obtain meteorological fields for analyzing weather patterns and air circulations over the TP.

13 3 Results and discussion

14 3.1 Surface concentrations of PAN and O₃

- 15 The PAN concentration averaged 0.36 ppb in the period of 16-25 August 2011, ranging from
- 16 0.11 ppb to 0.76 ppb. A clear increasing trend is found in the time series of PAN data in this
- 17 period. The origin of increasing PAN in such period will be discussed in section 3.2. In 2012,
- 18 the effective observation covered nearly two months (from 15 May to 13 July), long enough
- 19 to obtain the PAN levels under different atmospheric conditions during the South Asian
- 20 Monsoon period. The observed PAN concentration in this period varied from 0.16 ppb to 0.99
- 21 ppb, with an average of 0.44 ppb. This result is close to the PAN levels observed in summer
- 22 2006 at Waliguan (WLG), a remote site at the northeastern edge of the TP (Xue et al., 2011).
- 23 The O₃ concentration varied from 27.9 ppb to 96.4 ppb, with an average of 60.0 ppb, nearly
- 24 identical to the average O₃ level at WLG. There were little day-to-day and diurnal variations
- 25 when the PAN and O₃ measurements from WLG were not impacted by relatively polluted
- 26 airmasses from the eastern sector (Xue et al., 2011). In contrast, our PAN and O₃
- 27 measurements from NMC show profound variations. The reasons of the variations,
- 28 particularly the diurnal variations, should be investigated.

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3.2 Diurnal cycles of PAN and O₃ and potential impacts from the PBL evolution

3 The 10-min PAN and O₃ concentrations observed in 2012 were used to obtain the averaged diurnal patterns (Fig. 2). As can be seen in Fig. 2, during night time both PAN and O₃ show a 4 5 decreasing trend and reach the valley around 5:00 Local Time (LT, here LT=Beijing Time -6 2h), demonstrating their steady loss during night. From 5:00 LT to 10:00 LT, both gases can 7 be characterized by rapid increase, with the average levels of PAN and O₃ being lifted over 8 0.10 ppb and 15.0 ppb, respectively. Subsequently, O₃ increases at a much lower rate before 9 reaching its peak around 16:00 LT and then starts to decline. Unlike O₃, PAN behaves more 10 fluctuating after its peak time (around 12:00 LT), with a larger deviation from the trace of O₃. 11 It is noteworthy to see the sharp early-morning increase of PAN and O₃ shown in Fig. 2. If the 12 observed increase of both gases had been caused by photochemical productions, considerable 13 amount of their precursors would be required to fuel the photochemical reactions. However, 14 according to the EDGAR 3.2FT2000 database, anthropogenic emission in TP is extremely low, with emissions of NOx and CO being respectively no more than $0.1\times10^{-12}\ kg/m^2/s$ and 1 15 16 ×10⁻¹² kg/m²/s in the surrounding areas (http://themasites.pbl.nl/tridion/en/themasites/edgar 17 /emission data/edgar 32ft2000/index-2.html). Surface NO_x at NMC was below the lower 18 detect limit of the commercial NO_x analyzers like TE42CTL and Eco Physics CLD88p that 19 we deployed there. In addition, the key condition for the photochemistry, i.e., the UV 20 radiation, is not expected to be strong enough to drive the photochemistry in the very early morning (say around 5:00 LT), as the sunrise in that TP area occurs around 6:00 LT in 21 22 summer. Therefore, it is hypothesized that the main factor driving PAN and O₃ ascending in 23 the early morning is not photochemistry but the mixing process during the PBL evolution. To 24 prove this hypothesis, we display scatter plots in Fig. 3, showing the correlations between the 25 increment of O_3 concentration (ΔO_3) and that of PAN concentration (ΔPAN) for two time 26 periods of the day, and the correlation between the increments of O_3 and temperature (ΔT). 27 Figure 3(a) represents data from the 5:00-9:00 LT period, when the solar radiation becomes 28 gradually intensive. Figures 3(b) and 3 (c) show data from the 2:00-4:00 LT period, when no 29 solar radiation is available for the local photochemical reactions. 30 Significant linear correlation between ΔO_3 and ΔPAN is found for both the early morning

period (Fig. 3(a)) and the dark period (Fig. 3(b)), with correlation coefficients of 0.745 and

0.711, respectively. Although photochemical reactions, in which both O₃ and PAN are

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- 1 produced, can lead to the ΔO_3 - ΔPAN correlation, they cannot occur during the dark period.
- 2 Therefore, the significant correlation in Fig. 3(b) should be attributed to some meteorological
- 3 processes instead of photochemical process. Moreover, the ΔO_3 - ΔT correlation shown in Fig.
- 4 3(c) further indicates that surface O₃ and PAN at the site may be changed purely due to some
- 5 meteorological processes that change air temperature as well. The net change of O₃ could be
- 6 positive before dawn, and occurred on those days with simultaneously rising PAN and
- 7 temperature. The rising temperature could be related to the dry adiabatic heating process
- 8 during air masses sinking. Such a process happens when the PBL is extended, not necessarily
- 9 driven by solar radiation. Downward transport of PAN and O₃ may accompany such process.
- 10 Therefore, the PBL evolution might have significantly impacted the diurnal variations of PAN
- 11 and O₃ at NMC.

12

3.3 Insight into the PBL evolution

- 13 The evolution of PBL plays one of the key roles in the diurnal variations of surface
- 14 meteorological parameters and air pollutants, and is influenced by the dominating synoptic
- 15 situation. It has different diurnal patterns under different synoptic situations. Here we take the
- 16 O₃ enhancement in the early morning as an indicator quantity to find out major differences in
- 17 the evolution of the PBL and some related parameters under different synoptic situations. We
- 18 selected 30 days from the observation period in 2012 and separated them into two groups,
- 19 with group one including 15 days with the greatest ΔO_3 values (High ΔO_3) and group two
- 20 including 15 days with the smallest ΔO_3 values (Low ΔO_3). For the two groups, average
- 21 diurnal variations were calculated of PAN, O₃ and some meteorological parameters, i.e., wind
- 22 speed at 2 m above ground (Ws), U wind speed at 2 m above ground (Us), V wind speed at 2
- 23 m (Vs), the ratio between the 2-m and 10-m wind speeds (WSR), the temperature difference
- 24 between 20 m and 10 m (TD), and water vapor pressure (WVP). The obtained diurnal
- 25 variations are plotted in Fig. 4.
- 26 A stable nocturnal boundary layer (NBL) forms gradually in the night (Stull, 1988). A
- 27 temperature inversion occurs in the NBL, with the air temperature increasing with height. A
- 28 nocturnal jet may form over the NBL so that a larger gradient of wind speed exists in the NBL.
- 29 Such stratification prevents the air from being vertical mixed in the night and is broken in the
- 30 early morning. As a result, the concentrations of O₃ and PAN at the ground-level decrease
- 31 largely in the nighttime because of chemical and physical losses and increase rapidly in the
- 32

early morning because of the downward mixing of upper-level air containing more O₃ and

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- 1 PAN. This evolution of PBL, however, can be strongly impacted by some systematic
- 2 processes so that the day-night differences of PBL are weakened or even disappear. We
- 3 believe that the two groups of data presented in Fig. 4 represent approximately two
- 4 circumstances of the PBL evolution, with the High- ΔO_3 group being less or not impacted and
- 5 the Low- ΔO_3 group being strongly impacted by the systematic processes.
- 6 As can be seen in Fig. 4, the Low- ΔO_3 group showed much smaller diurnal variations of PAN,
- 7 O₃, Ws, WVP, and WSR, suggesting a weak day-night cycle of the PBL. Compared with the
- 8 values in the Low-ΔO₃ group, the nighttime values of PAN, O₃, Ws, and WSR in the High-
- 9 ΔO_3 group were much lower, and that of TD much higher. Lower WSR and higher TD in the
- 10 night indicate a more stable NBL, which explains the lower PAN and O₃ levels as discussed
- 11 above. After dawn the values of PAN, O_3 , Ws, WSR, and TD in the High- ΔO_3 group changed
- 12 rapidly back to their daytime levels, indicating the break of the stable NBL. It is noteworthy
- 13 that there were virtually no or only minor differences in the daytime values of PAN, Ws,
- WSR, and TD between the two groups. The daytime O_3 in the High- ΔO_3 group reached
- 15 significantly higher levels than that in the Low- ΔO_3 group. Moreover, the WVP value in the
- 16 High- ΔO_3 group was lower than that in the Low- ΔO_3 group during the entire day. These
- 17 phenomena imply that the High-ΔO₃ group is related to drier days and PBL conditions
- 18 favoring the increase of surface O₃ during daytime (e.g., through downward mixing) and
- 19 destruction during nighttime, while the Low- ΔO_3 group is related to more humid days and
- 20 PBL conditions that inhibit the variation of surface O₃.
- 21 The PBL evolution was investigated in previous field experiments in the TP. Li et al. (2011)
- 22 found that there were some differences in the diurnal evolution of the PBL structure between
- dry and rainy seasons. In the dry season, namely the pre-monsoon period, a shallow but strong
- 24 inversion layer could be clearly observed at night. The occurrence of the inversion layer is
- 25 high in the pre-monsoon period, simply because the PBL structure is primarily driven by
- sensible heat (Ma et al., 2005). The outflow of sensible heat at night is massive according to
- 27 thermal analysis. In the rainy season, a shallower but more persistent wet convection evolves,
- 28 forcing efficient exchange of quantities and also comparably smaller gradients of
- 29 meteorological elements. The daytime PBL height can reach 4-5 km above the ground in the
- 30 dry pre-monsoon period, while it is usually about 1-2 km above the ground in the wet
- 31 monsoon period (Li et al., 2011; Chen et al., 2013). In our case, prevailing monsoonal
- 32 features are perceivable in meteorological measurements associated with the Low-ΔO₃ group,

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1 such as the stronger southerly wind (Fig. 4(h)) and higher WVP (Fig. 4(d)). Unlike the dry

2 season, the convection intensity in the wet season has a much smaller diurnal variation, as

3 suggested by the smaller day-night differences of WSR and TD. Thus in the wet season,

4 downward transport of PAN and O₃ during nighttime may be much more effective than that in

5 the dry season. This can explain the observed nighttime differences in the PAN and O₃

6 concentrations between the Low- ΔO_3 and High- ΔO_3 groups (Figs. 4(a) and 4(b)).

7 To know more details about the two groups of days discussed above, the distribution of the

8 two groups of days, together with parameters including the PBL height, precipitable water of

9 entire atmosphere (PWAT), WVP, and the PAN and O₃ concentrations are shown in Fig. 5.

10 The PBL height and PWAT values are obtained from the NCEP FNL reanalysis data. It can

11 be seen that the surface measured WVP is in good accordance with the PWAT in trend. The

whole observation period in 2012 can be divided into dry period and wet period. The

13 transition between the wet and dry periods can be easily identified based on the changes of the

14 PBL height, and the PWAT and WVP values. It can also be seen in the variation of the daily

15 rainfall at NMC (Fig. S1). We can see a sudden seasonal change in the middle of June, when

the depth of PBL was suppressed after 16 June 2012 (marked with green bar in Fig. 5) and the

water amount became more abundant, suggesting the onset of the South Asian monsoon. The

18 distributions of the two groups of days are labeled on Fig. 5(a). Although there are some

irregular cases, the High- ΔO_3 days (group 1) are mostly distributed in the dry period and the

20 Low- ΔO_3 days (group 2) in the wet period. This supports our analysis in previous paragraph.

21 The time series of the PBL height indicates that the daily maximum PBL heights in the dry

22 period were much higher than those in the wet period, with only a few exceptions. Such

23 phenomenon agrees with the observational results from Naqu, about 230 km northeast of

24 NMC (Li et al., 2011). The nocturnal PBL height in the dry period could be extremely low

25 (frequently lower than 200 m). This explains the lower nighttime PAN and O₃ values in the

26 High- ΔO_3 group (Fig. 4).

17

27 In the pre-monsoon there may be episodes with monsoon features. An example of this is the

28 period of a few days around early June 2012, where the PBL height was considerably

29 suppressed, and the PWAT, WVP as well as the concentrations of PAN and O₃ were

30 significantly enhanced (Fig. 5). In this relatively humid episode, the nighttime concentrations

31 of PAN and O₃ were largely elevated, which may be attributable to the PBL structure and

32 airmasses transported from the polluted region (see section 3.5).

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- 1 In conclusion, the South Asian monsoon brings not only more water vapor over the central
- 2 Tibet area but also effectively drives the PBL evolution, which plays an important role in
- 3 shaping the diurnal patterns of PAN and O₃ at the NMC site.

4 3.4 PAN abundance in upper levels

- 5 It is noticeable in Fig. 4 that the levels of daytime O₃ were considerably different between the
- 6 two groups, while those of daytime PAN were close to each other. The highest hourly O₃
- 7 concentrations for the two groups were 69.7±2.4 ppb and 59.0±2.5 ppb, respectively, and the
- 8 highest hourly PAN concentrations were 0.48 ±0.02 ppb and 0.49 ±0.05 ppb, respectively. Xue
- 9 et al. (2011) pointed out that air masses from higher altitudes (upper troposphere/lower
- 10 stratosphere) could have significant impact on surface O₃ in the plateau. As shown in Fig. 5,
- 11 the daytime PBL heights in group 1 could extend to much higher altitudes than those in group
- 12 2, indicating a higher probability of downward mixing of O₃-richer air from the middle and
- 13 upper troposphere on the days in group 1. The air masses from higher altitudes seemed not to
- 14 cause additional increase in the daytime level of surface PAN, as suggested by the negligible
- distinction of daytime PAN between groups 1 and 2.
- 16 To gain more insight in air masses from upper origins, we attempt to label the upper air
 - masses in the whole observation periods. Following the categorizing way in section 3.3,
- 18 scatter plots of PAN- O₃, WVP- O₃, and WVP-PAN are shown in Fig. 6 for the two groups.
- 19 Since we have confirmed the meteorological features associated with both groups, some
- 20 relationship characteristics could be well comprehended. Suppose photochemistry did not
- 21 vary considerably, the amount of O_3 could in some degree represent the impact of upper
- 22 airmass. Using the Tropospheric Emission Spectrometer (TES) observations, Worden et al.
- 23 (2009) elaborately depicted the 3-dimensional distribution of tropospheric O₃ over the TP and
- 24 its surrounding areas, which shows a gradual increase of the O₃ concentration with height
- 25 under tropopause and a steep increase from upper troposphere to lower stratosphere. In some
- 26 cases, air masses with O₃ higher than 100 ppb can be downward transported to near ground,
- 27 causing high surface O_3 events. Thus surface O_3 concentration in the TP region can be used as
- an indicator of air masses from the higher altitudes and also reflects the depth of developed
- 29 PBL. As the WVP profile shows a clear decrease with height (Chen et al., 2013), air masses
- 30 from high altitudes can also be indicated by lower WVP. The data points within the red
- 31 rectangle in Fig. 6(c) are measurements associated with higher O₃ levels and lower WVP. We
- 32 consider these as measurements with significant features of middle/upper tropospheric air.

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1 Fig. 6(b) displays a good positive PAN-O₃ correlation for group 2, which is consistent with

2 simultaneous photochemical production of both secondary oxidants. However, the dataset

3 from group 1 shows a much weaker PAN-O₃ correlation (Fig. 6(a)), indicating a more obscure

4 relationship between PAN and altitude in group 1. Nearly no correlation between PAN and

5 WVP is found (Fig. 6(e)), supporting the above viewpoint. At present, the actual causes of the

6 poor PAN-O₃ and PAN-WVP correlations are unknown. However, it is reasonable to believe

7 that on the days in group 1, the observed O₃ level was more influenced by air masses from

8 upper troposphere/lower stratosphere, where the O₃ level is high (Worden et al., 2009) but the

9 PAN level is low (Moore and Remedios, 2010). In addition, it is suspected that the horizontal

variability of PAN is larger than that of O₃.

11 The ambiguity in the relationship between PAN and altitude makes it impossible to estimate

12 the PAN abundance in upper levels simply from the data shown in Fig. 6. Nevertheless, we

13 can make use of some cases with deep convection and apparent downward transport activities

14 in the dry period. Here we try to deduce the origins of air masses in two cases and roughly

15 estimate the PAN concentrations in upper levels. The two cases chosen for analysis are 25

16 May 2012 and 24 August 2011. Figure 7 displays the omega field and horizontal wind vectors

17 at different time and pressure height, with the two cases labeled by red rectangles (termed as

18 Case 1 and Case 2). Both cases are segments of the dry periods, when the PBL could reach

19 higher heights and favor the entrainment of upper air-masses.

20 Figure 7(a) shows that positive omega dominated the PBL from early 25 May 2012 to early

21 26 May 2012 (Case 1), with the range of higher omega (>0.1 m/s) extending from surface to

22 350 hPa, and a distinct valley of specific humidity line of 2g/kg, indicating a strong

23 downward motion core. In response to this downward transport, PAN and O₃ were both

24 elevated to higher levels and WVP dropped (Fig. 5). A similar downward motion core

25 occurred on 22-23 August 2011 (Case 2), as shown in Fig. 7(b). The downdraft motion core

26 on 22 August extended up to 300 hPa and lasted all day long, with very high intensity

27 (omega > 0.4 m/s) after the noontime. For better understanding of Case 2, we display the time

28 series of O₃, PAN, and related meteorological parameters in Fig. 8. There was a rapid rising

29 trend of the O₃ and PAN levels on 22 August 2011, as indicated by the arrow in Fig. 8(b). In

30 parallel with this trend, relative humidity and wind vector changed sharply, with the former

31 dropping dramatically from 80% to about 30% and the later turning from southerly to

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1 northerly. Similar rapid variations were also observed partly during 23-24 August 2011,

2 corresponding to the subsiding of dry air masses (Fig. 7(b)).

3 It is noticeable that the daytime levels of O₃ and PAN did not show much distinction among

4 the days from 22 to 25 August 2011. This suggests that the air masses measured during the

5 period might originate from the similar height and area. To prove this, we calculated

6 backward trajectories with endpoints at 500 m and 1500 m above the ground of the NMC site.

7 Some of the trajectories for the two selected cases, 25 May 2012 (Case 1) and 22 August 2011

8 (Case 2), are plotted in Figs. 9 and 10, respectively, overlaying on the 250 hPa potential

9 vorticity (PV) fields at three timepoints during 23-24 May 2012 (for Case 1) and during 20-22

10 August 2011 (for Case 2), respectively. Similar plots with the same trajectories and 350 hPa

11 PV fields are shown in Figs. S2 and S3 for Case 1 and Case 2, respectively. In both cases

12 stratospheric intrusions occurred as indicated by the higher PV values. In Case 1 (Figs. 9 and

13 S2) higher PV covered the zone from 30 N to beyond 50 N. In Case 2 (Figs. 10 and S3)

14 higher PV extended from about 40 N to beyond 50 N. In both case air masses arrived at the

15 NMC site originated from or travelled through the zones obviously impacted by stratospheric

16 intrusions. Therefore, the PAN and O₃ measurements in both cases were influenced by upper

17 level air masses that cantained stratospheric air. In addition to this transport feature, the

18 rapidly increasing O₃ and decreasing water amount also indicate that surface air masses

19 originated from high levels. For Case 1 and Case 2, the PAN concentration was elevated

20 respectively up to 0.5 ppb and 0.6-0.7 ppb, which can be regarded as the estimated level of

21 PAN at high altitudes (around 350 hPa).

28

22 Table 1 summarizes the PAN levels in the upper troposphere over the TP. A prominent

23 progress in detection of global upper-tropospheric PAN is based on the application of infra-

24 red spectrometers. There exists some detection error as retrieval band of PAN would

25 inevitably be contaminated by irrelevant compounds, such as water vapor and CCl₄ and the

26 variability of some atmospheric parameters would also bring interferences (Remedios et al.,

27 2007). In summer, there exists more uncertainty using remote sensing method due to higher

amount of water vapor over the TP region; hence a direct comparison with field measurement

29 is required. According to the satellite results (Moore and Remedios, 2010), the retrieved PAN

30 levels at 333 hPa and 234 hPa in March 2003 were in the ranges of 0.3-0.5 ppb and 0.15-0.2

31 ppb, respectively, implying a decreasing trend of the PAN concentration along height in the

32 upper troposphere. In August 2003 the PAN levels at 278 hPa and 185 hPa did not show

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1 evident increment compared with March 2003. Model results from Fischer et al. (2014) show

2 similar ranges of PAN levels. The trajectories shown in Figs. 9, 10, S2 and S3 indicate that air

3 masses could originate from the upper troposphere (~350 hPa), thus could well match the

4 other observation heights listed in Table 1. Our estimates of PAN level are close to or greater

5 than the upper ranges of the PAN levels reported in the literature (Glatthor et al., 2007; Moore

6 and Remedios, 2010; Wiegele et al., 2012; Fadnavis et al., (2014; Fischer et al., 2014). Taking

7 into account the thermal decomposition and photolysis of PAN during the transport, the actual

8 PAN levels in the upper troposphere could be higher than our estimates. Therefore, the

9 retrieval from MIPAS-E measurements might have underestimated the PAN level in the upper

10 troposphere.

Moreover, an enhancement of PAN was observed during 21-26 August 2011, with a range of

12 0.6-0.7 ppb, indicating that PAN was accumulated in an elevated level in later summer. Such

13 PAN level was not acquired by Moore and Remedios (2010) who reported a maximum of

14 PAN of 0.5 ppb in August, implying again the possibility of underestimation of PAN in

15 satellite retrieval. It is noted that the satellite observations were made years before our

16 measurements. Any potential long-term changes or interannual variations in PAN may also

17 cause some differences. Fadnavis et al. (2014) reported a maximum increase rate of 4.5 ± 3.1

18 ppt yr⁻¹ for PAN in the UTLS over the Asian summer monsoon region in the summer

19 monsoon season during 2002-2011. However, such small trend in PAN cannot explain the

above discrepancy.

21 3.5 A PAN episode driven by South Asian monsoon

22 In warm environment, PAN is short-lived. Below 7 km, thermal decomposition is the main

23 loss process of PAN (Talukdar, 1995). Thus although polluted air masses from south of the

24 Himalayas can be transported to TP along the monsoon stream, PAN in the air masses may

25 experience great loss during the travelling. The level of PAN observed at our site was the

26 remaining PAN in the air masses, which should be significantly lower than that in the

27 formation area. Nevertheless, PAN episodes can be observed under certain meteorological

28 conditions. One of such episodes occurred in early June 2012. As can be seen in Fig. 5, the

29 PAN level humped during 1-6 June 2012, with some diurnal variations. The maximum PAN

30 level reached 1.0 ppb, and the diurnal minima on these days were even higher than the diurnal

31 maxima on many of other observation days. The origin of the high PAN concentrations

32 deserves investigation.

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1 Data in Fig. 5 indicate that the monsoon flow prevailed persistently after the middle of June 2 2012, and there were also some features of monsoon impact during 1-6 June 2012 when the 3 PAN level was increased to near 1 ppb. After this abrupt rising, PAN dropped down to much 4 lower level, suggesting a substantial change in air mass. To understand this phenomenon, we 5 calculated average fields of wind, relative humidity, and omega at sigma=0.995 for the 6 periods 30-31 May using the FNL reanalysis data. During 30-31 May 2012, the major part of 7 Indian subcontinent was controlled by an anticyclone system and a large convergence zone 8 formed over the central TP (see Fig. S4). The NMC site was within this convergence zone and 9 obviously influenced by airflow from North India. Figure 10 shows the average wind fields 10 for 12:00 (UTC) of 4, 5, 7 and 8 June 2012. These wind fields can help us to understand the 11 high level of PAN observed during 1-6 June 2012. As can be seen in Figs. 11 and S4, the 12 week after 30 May the NMC site was under the control of a large convergence zone and 13 influenced by westerly and southwesterly winds, which could transport air masses from North 14 India to the NMC site. After this period, the site was influenced by significantly different air 15 masses. For example, the average wind field shown in Figs. 11c and 11d indicate that after 7 16 June 2012, strong southerly and southeasterly winds developed over East India and Southeast 17 Nepal, and the area surrounding NMC was controlled by southerly wind, which promotes the 18 transport of air masses from the Bay of Bengal. Although most of the central and western TP 19 was still under the control of the convergence zone, NMC and its surrounding were outside of 20 its direct impact. Such change in air masses arriving NMC inevitably caused substantial 21 differences in photochemistry. As indicated by observations (Ghude et al., 2008; Fishman et 22 al., 2003), Northern India suffers photochemical pollution. Emission inventories (Ohara et al., 23 2007; Zhang et al., 2009) suggest that North India is one of the Asian emission centers for 24 pollutants including NO_x and VOCs. Tropospheric NO₂ and HCHO can be derived from 25 satellite observations based on the DOAS technique (Boersma et al., 2011; De Smedt et al., 2008, 2012). Figure S5 shows tropospheric NO2 and HCHO columns during June, 2012, 26 27 made available by the Tropospheric Emission Monitoring Internet Service (TEMIS) at The 28 Netherlands Meteorological (KNMI), The Royal Institute Netherlands 29 (http://www.temis.nl/index.php). As can be seen in this figure, both NO₂ and HCHO in the 30 troposphere over North India were highly abundant in June, 2012. The high NO2 and HCHO 31 concentrations suggest that the concentrations of NO_x and VOCs over the North India region 32 were high. NO_x and VOCs are species for photochemical reactions leading to formation of O₃,

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- 1 PAN, etc. Therefore, it is likely that the PAN episode observed at our site during 1-6 June
- 2 2012 was mainly caused by plumes with high PAN and its precursors from North India.
- 3 To further support the above view, we made calculations of backward air trajectories. The
- 4 results are presented in Fig. 12. The 5-day trajectories were calculated for endpoints at 500 m
- 5 and 1000 m above ground for 1-6 June and 7-10 June 2012, respectively. Obviously, air
- 6 trajectories arriving NMC during 1-6 June were quite different from those during 7-10 June,
- 7 particularly those with endpoints at 500 m (Figs. 12a and 12c). About a half of the trajectories
- 8 during 1-6 June originated from or moved through the PBL over North India (Fig. 12a), while
- 9 nearly none of the trajectories during 7-10 June had an opportunity to pass through the PBL
- 10 over North India (Fig. 12c). Most of the trajectories during 7-10 June originated either from
- 11 the free troposphere over western Asia and Indian subcontinent or from the PBL south of
- 12 NMC. These can explain the sudden decrease of the PAN level after 6 June 2012 on one hand,
- and on the other hand support the idea that the PAN episode observed during 1-6 June 2012
- was mainly caused by plumes from North India.
- 15 Although the TP is one the cleanest regions of the world, transport of anthropogenic
- 16 pollutants deserves attention. Some recent studies have showed that air pollutants can be
- 17 transported to the Himalayas or to the TP region through passes like river valleys from the
- surroundings (Cong et al., 2007; Cong et al., 2009; Bonasoni et al., 2010; Kopacz et al., 2011;
- 19 Lüthi et al., 2015; Shen et al., 2015; Zhang et al., 2015). The main source regions are South
- 20 and East Asia. During the South Asian monsoon, the TP is predominately influenced by air
- 21 masses from the Indian subcontinent. Impacts of transported pollutants on atmospheric
- 22 environment over the Himalayas and TP, particularly the climate and hydrological effects of
- 23 deposition of black carbon and other substances on Himalayan glaciers, have caused concerns
- 24 (Ramanathan et al., 2007; Ming et al., 2012; Zhao et al., 2013; Qu et al., 2014; Wang et al.,
- 25 2015; Zhang et al., 2015).
- 26 So far, studies of transport of pollutants to the TP and its effect are mainly about aerosols
- 27 (compositions and optical depth) and less attention has been paid to gaseous pollutants. There
- 28 has been no previous report about impacts of long-range transport of pollutants on
- 29 tropospheric photochemistry over the TP region. Our results show, for the first time, that
- 30 long-range transport of polluted airmasses from North India can significantly enhance
- 31 ambient level of PAN at NMC. Although we have no observational data of PAN from other
- 32 sites in the TP, it is likely that the entire convergence zone in the central and western TP (Figs.

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- 1 10 and S4) was more or less impacted by the pollutants from North India. This implies that
- 2 photochemistry over a large area in the TP was probably disturbed for at least ten days in the
- 3 cases shown in Figs. 11 and S4. PAN transported to the TP region may be thermally and/or
- 4 photolytically decomposed to release NO_x, acting as a chemical source of atmospheric NO_x
- 5 over the TP, a region with very little anthropogenic emission of NO_x. The impacts of the
- 6 transport of PAN and other related species on tropospheric photochemistry over the TP need
- 7 to be studied in the future.

8 4 Conclusions

- 9 For the first time, we made simultaneous ground-based measurements of two photochemical
- 10 products, PAN and O₃ at Nam Co, a remote site in the central Tibetan Plateau (TP) region.
- 11 Our effective PAN data cover two summer periods, i.e., 16-25 August 2011 and 15 May to 13
- 12 July 2012. The average concentrations of PAN were 0.36 ppb (range: 0.11-0.76 ppb) and 0.44
- ppb (range: 0.21-0.99 ppb) in the 2011 and 2012 periods, respectively. During the observation
- in 2012, the O₃ concentration varied from 27.9 ppb to 96.4 ppb, with an average of 60.0 ppb,
- 15 very close to the summertime O₃ level found at Waliguan, a global baseline station at the
- 16 northeastern edge of the TP.
- 17 PAN and O₃ showed profound and similar diurnal cycles, with valleys around 5:00 LT, steep
- 18 rises in the early morning, and broader platforms of high values during 9:00-20:00 LT. Such
- 19 patterns of diurnal variations of both gases, particularly the sharp increases even before
- 20 sunrise, cannot be attributed solely to local photochemistry. Our analysis suggests that the
- 21 PBL evolution played a key role in shaping the diurnal patterns of both oxidants. PAN and O_3
- 22 in the shallow nocturnal PBL were significantly removed by their sinks, such as chemical
- 23 reactions and dry deposition. In the early morning, the elevation of the PBL height caused
- 24 downward mixing of upper air containing higher PAN and O₃, leading to steep rises of the
- 25 concentrations of these gases. The downward mixing and photochemistry sustained the higher
- 26 levels of PAN and O₃ in the daytime. However, there were day-to-day differences in the PBL
- 27 evolution, which could cause large differences in the diurnal variations of PAN and O₃. We
- 28 identified two groups of days with different meteorological conditions and different diurnal
- 29 patterns of trace gases and meteorological parameters. Days in group 1 were mainly
- 30 distributed in the pre-monsoon period, with higher daytime height of PBL (about 3 km), lower
- 31 humidity, and larger day-night variations of PAN and O₃. Days in group 2 were mainly

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- 1 distributed in the monsoon period, with shallower daytime PBL (about 2 km), higher
- 2 humidity, and much smaller day-night variations of PAN and O₃.
- 3 There were some cases with obvious rapid transport of air masses during our observations.
- 4 We identified two cases of rapid downward transport of air masses from the middle and upper
- 5 troposphere. The observed PAN levels during these two cases ranged from 0.5 ppb to 0.7 ppb.
- 6 They may represent those in the middle and upper troposphere. These PAN levels are higher
- 7 than those retrieved from satellite measurements. In addition to vertical transport of PAN, we
- 8 also identified a case of strong long-range transport of PAN plumes. During this case,
- 9 relatively polluted air masses from the PBL over North India were able to be transported over
- 10 the western and central TP to NMC, causing a profound episode of PAN with maximum close
- 11 to 1 ppb during 1-6 June 2012. In contrast, significantly lower PAN levels were observed
- when air masses originated from other areas. Although transport of aerosols from South and
- 13 Southeast Asia and its impacts on atmospheric environment over Himalayas and the TP have
- 14 been intensively studied in recent years, gaseous pollutants have received less attention. Our
- 15 results show, for the first time, that polluted air masses from South Asia can significantly
- 16 enhance the ambient level of PAN at NMC. The space scale and frequency of this
- 17 phenomenon and its impacts on tropospheric photochemistry over the TP region remain to be
- studied in the future.
- 19 Data availability. The observational data analyzed in this paper can be made available for
- 20 scientific purposes by contacting the corresponding author (xuxb@camscma.cn).
- 21 **Competing interests.** The authors declare that they have no conflict of interest.
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1 References

- 2 Bonasoni, P., Laj, P., Marinoni, A., Sprenger, M., Angelini, F., Arduini, J., Bonaf & U.,
- 3 Calzolari, F., Colombo, T., Decesari, S., Di Biagio, C., di Sarra, A. G., Evangelisti, F., Duchi,
- 4 R., Facchini, M. C., Fuzzi, S., Gobbi, G. P., Maione, M., Panday, A., Roccato, F., Sellegri, K.,
- 5 Venzac, H., Verza, G. P., Villani, P., Vuillermoz, E., and Cristofanelli, P.: Atmospheric
- 6 Brown Clouds in the Himalayas: first two years of continuous observations at the Nepal
- 7 Climate Observatory-Pyramid (5079 m), Atmos. Chem. Phys., 10, 7515-7531, 2010.
- 8 Boersma, K.F., Eskes, H.J., Dirksen, R. J., van der A, R. J., Veefkind, J. P., Stammes, P.,
- 9 Huijnen, V., Kleipool, Q. L., Sneep, M., Claas, J., Leitao, J., Richter, A., Zhou, Y., and
- 10 Brunner, D.: An improved retrieval of tropospheric NO₂ columns from the Ozone Monitoring
- 11 Instrument, Atmos. Meas. Tech., 4, 1905-1928, 2011
- 12 Chen, B., Xu, X. D., Yang, S., and Zhao, T. L.: Climatological perspectives of air transport
- 13 from atmospheric boundary layer to tropopause layer over Asian monsoon regions during
- boreal summer inferred from Lagrangian approach, Atmos. Chem. Phys., 12, 5827-5839,
- 15 10.5194/acp-12-5827-2012, 2012.
- 16 Chen, X., An, J. A., Su, Z., Torre, L., Kelder, H., Peet, J., Ma, Y., and Fu, R.: The Deep
- 17 Atmospheric Boundary Layer and Its Significance to the Stratosphere and Troposphere
- Exchange over the Tibetan Plateau, PloS One, 8, e56909, 10.1371/journal.pone.005690, 2013.
- 19 Cong, Z. Y., Kang, S. C., Liu, X. D., and Wang, G. F.: Elemental composition of aerosol in
- 20 the Nam Co region, Tibetan Plateau, during summer monsoon season, Atmos. Environ., 41,
- 21 1180–1187, 2007.
- 22 Cong, Z. Y., Kang, S., Smirnov, A., and Holben, B.: Aerosol optical properties at Nam Co, a
- remote site in central Tibetan Plateau, Atmos. Res., 92, 42–48, 2009.
- 24 Cox, R.A. and Roffey, M.J.: Thermal decomposition of peroxyacetylnitrate in the presence of
- 25 nitric oxide. Environ. Sci. & Technol., 11(9), 900–906,1977.
- 26 Cristofanelli, P., Bracci, A., Sprenger, M., Marinoni, A., Bonafè, U., Calzolari, F., Duchi, R.,
- 27 Laj, P., Pichon, J. M., Roccato, F., Venzac, H., Vuillermoz, E., and Bonasoni, P.:
- 28 Tropospheric ozone variations at the Nepal Climate Observatory-Pyramid (Himalayas, 5079m
- 29 a.s.l.) and influence of deep stratospheric intrusion events, Atmos. Chem. Phys., 10, 6537-
- 30 6549, 2010.

Manuscript under review for journal Atmos. Chem. Phys.





- 1 De Smedt, I., Van Roozendael, M., Stavrakou, T., Müller, J.-F., Lerot, C., Theys, N., Valks,
- 2 P., Hao, N., and van der, N.A.: Improved retrieval of global tropospheric formaldehyde
- 3 columns from GOME-2/MetOp-A addressing noise reduction and instrumental degradation
- 4 issues. Atmos. Meas. Tech., 5, 2933-2949, 2012.
- 5 De Smedt, I., Müller, J.-F., Stavrakou, T., van der, A., R. J., Eskes, H. J., and Van Roozendael,
- 6 M.: Twelve years of global observations of formaldehyde in the troposphere using GOME
- 7 and SCIAMACHY sensors, Atmos. Chem. Phys., 8(16), 4947-4963, 2008.
- 8 Dessler, A. E., and Sherwood, S. C.: Effect of convection on the summertime extratropical
- 9 lower stratosphere, J. Geophys. Res., 109, D23301-D23310, 10.1029/2004jd005209, 2004.
- 10 Ding, A. and Wang, T.: Influence of stratosphere-to-troposphere exchange on the seasonal
- 11 cycle of surface ozone at Mount Waliguan in western China, Geophys. Res. Lett., 33, L03803,
- 12 doi:10.1029/2005GL024760, 2006.
- 13 Draxler, R.R. and Hess, G.D.: Description of the HYSPLIT 4 modeling system. NOAA
- 14 Technical Memorandum. ERLARL-224, NOAA Air Resources Laboratory, Silver Spring,
- 15 MD. 24, 1997.
- 16 Fadnavis, S., Schultz, M.G., Semeniuk, K., Mahajan, A.S., Pozzoli, L., Sonbawne, S., Ghude,
- 17 S.D., Kiefer, M., and Eckert, E.: Trends in peroxyacetyl nitrate (PAN) in the upper
- 18 troposphere and lower stratosphere over southern Asia during the summer monsoon season:
- 19 regional impacts, Atmos. Chem. Phys., 14, 12725–12743, 2014.
- 20 Fischer, E. V., Jaffe, D. A., Reidmiller, D. R., and Jaegle, L.: Meteorological controls on
- 21 observed peroxyacetyl nitrate at Mount Bachelor during the spring of 2008, J. Geophys. Res.,
- 22 115, D03302, doi:10.1029/2009jd012776, 2010.
- 23 Fischer, E. V., Jaffe, D. A., and Weatherhead, E. C.: Free tropospheric peroxyacetyl nitrate
- 24 (PAN) and ozone at Mount Bachelor: potential causes of variability and timescale for trend
- 25 detection, Atmos. Chem. Phys., 11, 5641-5654, 10.5194/acp-11-5641-2011, 2011.
- 26 Fischer, E. V., Jacob, D. J., Yantosca, R. M., Sulprizio, M. P., Millet, D. B., Mao, J., Paulot,
- 27 F., Singh, H. B., Roiger, A. E., Ries, L., Talbot, R. W., Dzepina, K., and Pandey Deolal, S.:
- 28 Atmospheric peroxyacetyl nitrate (PAN): a global budget and source attribution, Atmos.
- 29 Chem. Phys., 14, 2679-2698, 2014.

Manuscript under review for journal Atmos. Chem. Phys.





- 1 Fishman, J., Wozniak, A.E., and Creilson, J.K.: Global distribution of tropospheric ozone from
- 2 satellite measurements using the empirically corrected tropospheric ozone residual technique:
- 3 Identification of the regional aspects of air pollution, Atmos. Chem. Phys., 3, 893–907, 2003.
- 4 Ford, K. M., Campbell, B. M., Shepson, P. B., Bertman, S. B., Honrath, R. E., Peterson, M.,
- 5 and Dibb, J. E.: Studies of Peroxyacetyl nitrate (PAN) and its interaction with the snowpack
- 6 at Summit, Greenland, J. Geophys. Res., 107, 4102-4111, 2002.
- Fu, R., Hu, Y., Wright, J. S., Jiang, J. H., Dickinson, R. E., Chen, M., Filipiak, M., Read, W.
- 8 G., Waters, J. W., Wu, D. L., and Affiliations, A.: Short circuit of water vapor and polluted
- 9 air to the global stratosphere by convective transport over the Tibetan Plateau, PNAS, 103,
- 10 5664-5669, 10.1073/pnas.0601584103, 2006.
- 11 Gettelman, A., and Kinnison, D. E.: Impact of monsoon circulations on the upper troposphere
- 12 and lower stratosphere, J. Geophys. Res., 109, D22101-D22114, 10.1029/2004jd004878,
- 13 2004.
- 14 Ghude, S.D., Jain, S.L., Arya, B.C., Beig, G., Ahammed, Y.N., Kumar, A., Tyagi, B.: Ozone
- 15 in ambient air at a tropical megacity, Delhi: characteristics, trends and cumulative ozone
- 16 exposure indices, J. Atmos. Chem., 60, 237–252, 2008.
- 17 Glatthor, N., von Clarmann, T., Fischer, H., Funke, B., Grabowski, U., Hopfner, M.,
- 18 Kellmann, S., Kiefer, M., Linden, A., Milz, M., Steck, T., and Stiller, G. P.: Global
- 19 peroxyacetyl nitrate (PAN) retrieval in the upper troposphere from limb emission spectra of
- 20 the Michelson Interferometer for Passive Atmospheric Sounding (MIPAS), Atmos. Chem.
- 21 Phys., 7, 2775–2787, doi:10.5194/acp-7-2775-2007, 2007.
- 22 Gu, Y., Liao, H., and Bian, J.: Summertime nitrate aerosol in the upper troposphere and lower
- 23 stratosphere over the Tibetan Plateau and the South Asian summer monsoon region, Atmos.
- 24 Chem. Phys., 16, 6641-6663, doi:10.5194/acp-16-6641-2016, 2016.
- Hudman, R.C., Jacob, D.J., Cooper, O.R., Evans, M.J., Heald, C.L., Park, R.J., Fehsenfeld, F.,
- Flocke, F., Holloway, J., Hübler, G., Kita, K., Koike, M., Kondo, Y., Neuman, A., Nowak, J.,
- 27 Oltmans, S., Parrish, D., Roberts, J.M., and Ryerson, T.: Ozone production in transpacific
- 28 Asian pollution plumes and implications for ozone air quality in California, J. Geophys. Res.,
- 29 109, D23S10-D23S23, 10.1029/2004jd004974, 2004.

Manuscript under review for journal Atmos. Chem. Phys.





- 1 Kopacz, M., Mauzerall, D. L., Wang, J., Leibensperger, E. M., Henze, D. K., and Singh, K.:
- 2 Origin and radiative forcing of black carbon transported to the Himalayas and Tibetan Plateau,
- 3 Atmos. Chem. Phys., 11, 2837-2852, 10.5194/acp-11-2837-2011, 2011.
- 4 LaFranchi, B. W., Wolfe, G. M., Thornton, J. A., Harrold, S. A., Browne, E. C., Min, K. E.,
- 5 Wooldridge, P. J., Gilman, J. B., Kuster, W. C., Goldan, P. D., de Gouw, J. A., McKay, M.,
- 6 Goldstein, A. H., Ren, X., Mao, J., and Cohen, R. C., Closing the peroxy acetyl nitrate budget:
- 7 observations of acyl peroxy nitrates (PAN, PPN, and MPAN) during BEARPEX 2007, Atmos.
- 8 Chem. Phys., 9(19), 7623–7641, 2009.
- 9 Law, K. S., Fierli, F., Cairo, F., Schlager, H., Borrmann, S., Streibel, M., Real, E., Kunkel, D.,
- 10 Schiller, C., Ravegnani, F., Ulanovsky, A., D'Amato, F., Viciani, S., Volk, C.M.: Air mass
- 11 origins influencing TTL chemical composition over West Africa during 2006 summer
- 12 monsoon, Atmos. Chem. Phys., 10, 10753–10770, 2010.
- Lelieveld, J., Brühl, C., Jöckel, P., Steil, B., Crutzen, P. J., Fischer, H., Giorgetta, M. A., Hoor,
- 14 P., Lawrence, M. G., Sausen, R., and Tost, H.: Stratospheric dryness: model simulations and
- satellite observations, Atmos. Chem. Phys., 7, 1313-1332, 2007.
- 16 Lelieveld, J. and Dentener, F. J.: What controls tropospheric ozone? J. Geophys. Res.,
- 17 105(D3), 3531–3551, 2000.
- 18 Li, M., Y. Ma, W. Ma, H. Ishiakawa, F. Sun, S. Ogino, Structural difference of atmospheric
- 19 boundary layer between dry and rainy seasons over the central Tibetan Plateau (in
- 20 Chinese), Journal of Glaciology and Geocryology, 33, 72-79, 2011.
- 21 Lin, W., Xu, X., Zheng, X., Jaxi, D., Ciren, B., Two-year measurements of surface ozone at
- 22 Dangxiong, a remote highland site in the Tibetan Plateau, J. Environ. Sci., 31, 133-145, 2015.
- 23 Lüthi, Z.L., Škerlak, B., Kim, S.-W., Lauer, A., Mues, A., Rupakheti, M., and Kang, S.:
- 24 Atmospheric Brown Clouds reach the Tibetan Plateau by crossing the Himalayas, Atmos.
- 25 Chem. Phys., 15, 6007-6021, 2015.
- 26 Ma, J., Liu, H., and Hauglustaine, D.: Summertime tropospheric ozone over China simulated
- 27 with a regional chemical transport model 1. Model description and evaluation, J. Geophys.
- 28 Res., 107, ACH 27-21–ACH 27-13, doi:10.1029/2001JD001354, 2002a.
- 29 Ma, J., Tang, J., Zhou, X., and Zhang, X.: Estimates of the Chemical Budget for Ozone at
- 30 Waliguan Observatory, J. Atmos. Chem., 41, 21–48, doi:10.1023/A:1013892308983, 2002b.

Manuscript under review for journal Atmos. Chem. Phys.





- 1 Ma, J., Lin, W. L., Zheng, X.D., Xu, X.B., Li, Z., and Yang, L.L.: Influence of air mass
- 2 downward transport on the variability of surface ozone at Xianggelila Regional Atmosphere
- 3 Background Station, southwest China, Atmos. Chem. Phys., 14, 5311–5325, 2014.
- 4 Ma, Y., Fan, S., Ishikawa, H., Tsukamoto, O., Yao, T., Koike, T., Zuo, H., Hu, Z., and Su, Z.:
- 5 Diurnal and inter-monthly variation of land surface heat fluxes over the central Tibetan
- 6 Plateau area, Theoret. Appl. Climat., 80, 259-273, 2005.
- 7 Ma, W., Ma, Y., and Bob, S.: Feasibility of Retrieving Land Surface Heat Fluxes from
- 8 ASTER Data Using SEBS: a Case Study from the NamCo Area of the Tibetan Plateau, Arctic,
- 9 Antarctic, and Alpine Research, 43(2), 239-245, 2011.
- 10 Ming, J., Du, Z., Xiao, C., Xu, X., and Zhang, D.: Darkening of the mid-Himalaya glaciers
- 11 since 2000 and the potential causes, Environ. Res. Lett. 7, 014021, doi:10.1088/1748-
- 12 9326/7/1/014021, 2012.
- 13 Moore, D.P. and Remedios, J.J.: Seasonality of Peroxyacetyl nitrate (PAN) in the upper
- troposphere and lower stratosphere using the MIPAS-E instrument, Atmos. Chem. Phys., 10,
- 15 6117-6128, 10.5194/acp-10-6117-2010, 2010.
- 16 Myhre, G., Shindell, D., Bréon, F.-M., Collins, W., Fuglestvedt, J., Huang, J., Koch, D.,
- 17 Lamarque, J.-F., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G., Takemura, T.,
- and Zhang, H.: Anthropogenic and Natural Radiative Forcing. In: Climate Change 2013: The
- 19 Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of
- the Intergovernmental Panel on Climate Change [Stocker, T.F., D. Qin, G.-K. Plattner, M.
- 21 Tignor, S.K. Allen, J. Boschung, A. Nauels, Y. Xia, V. Bex and P.M. Midgley (eds.)].
- 22 Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 2013.
- 23 Ohara, T., Akimoto, H., Kurokawa, J., Horii, N., Yamaji, k., Yan, X., and Hayasaka, T.: An
- 24 Asian emission inventory of anthropogenic emission sources for the period 1980–2020,
- 25 Atmos. Chem. Phys., 7, 4419–4444, 2007.
- 26 Pandey Deolal, S., Staehelin, J., Brunner, D., Cui, J., Steinbacher, M., Zellweger, C., Henne,
- 27 S., and Vollmer, M. K.: Transport of PAN and NOy from different source regions to the
- 28 Swiss high alpine site Jungfraujoch, Atmos. Envron., 64, 103-115,
- 29 doi:10.1016/j.atmosenv.2012.08.021, 2013.

Manuscript under review for journal Atmos. Chem. Phys.





- 1 Park, M., Randel, W. J., Gettelman, A., Massie, S. T., and Jiang, J. H.: Transport above the
- 2 Asian summer monsoon anticyclone inferred from Aura Microwave Limb Sounder tracers, J.
- 3 Geophys. Res., 112, 10.1029/2006jd008294, 2007.
- 4 Park, M., Randel, W. J., Emmons, L. K., Bernath, P. F., Walker, K. A., and Boone, C. D.:
- 5 Chemical isolation in the Asian monsoon anticyclone observed in Atmospheric Chemistry
- 6 Experiment (ACE-FTS) data, Atmos. Chem. Phys., 8, 757-764, 10.5194/acp-8-757-2008,
- 7 2008.
- 8 Park, M., Randel, W. J., Emmons, L. K., and Livesey, N. J.: Transport pathways of carbon
- 9 monoxide in the Asian summer monsoon diagnosed from Model of Ozone and Related
- Tracers (MOZART), J. Geophys. Res., 114, D08303-D08313, 10.1029/2008jd010621, 2009.
- 11 Qu, B., Ming, J., Kang, S.-C., Zhang, G.-S., Li, Y.-W., Li, C.-D., Zhao, S.-Y., Ji, Z.-M., and
- 12 Cao, J.-J.: The decreasing albedo of the Zhadang glacier on western Nyainqentanglha and the
- 13 role of light-absorbing impurities, Atmos. Chem. Phys., 14, 11117-11128, doi:10.5194/acp-
- 14 14-11117-2014, 2014.
- 15 Ramanathan, V., Ramana, M. V., Roberts, G., Kim, D., Corrigan, C., Chung, C., Winker, D.:
- Warming trends in Asia amplified by brown clouds solar absorption, Nature, 448, 575–578,
- 17 2007.
- 18 Ran, L., Lin, W.L., Deji, Y.Z., La, B., Tsering, P.M., Xu, X.B., and Wang, W.: Surface gas
- 19 pollutants in Lhasa, a highland city of Tibet: current levels and pollution implications, Atmos.
- 20 Chem. Phys. 14, 10721–10730, 2014.
- 21 Randel, W. J., Park, M., Emmons, L., Kinnison, D., Bernath, P., Walker, K. A., Boone, C.,
- 22 and Pumphrey, H.: Asian Monsoon Transport of Pollution to the Stratosphere, Science
- 23 Magazine, 328, 611-613, 10.1126/science.1182274, 2010.
- 24 Remedios, J. J., Allen, G., Waterfall, A. M., Oelhaf, H., Kleinert, A., and Moorel, D. P.:
- 25 Detection of organic compound signatures in infra-red, limb emission spectra observed by the
- 26 MIPAS-B2 balloon instrument, Atmos. Chem. Phys., 7, 1599-1613, 10.5194/acp-7-1599-
- 27 2007, 2007.
- 28 Roiger, A., Aufmhoff, H., Stock, P., Arnold, F., and Schlager, H.: An aircraft-borne chemical
- 29 ionization ion trap mass spectrometer (CI-ITMS) for fast PAN and PPN measurements,
- 30 Atmos. Meas. Tech., 4, 173-188, 10.5194/amt-4-173-2011, 2011.

Manuscript under review for journal Atmos. Chem. Phys.





- 1 Roberts, J. M.: PAN and Related Compounds, in: Volatile Organic Compounds in the
- 2 Atmosphere, edited by: Koppmann, R., Blackwell Publishing, 500, Oxford, UK, 2007.
- 3 Russo, R. S., Talbot, R. W., Dibb, J. E., Scheuer, E., Seid, G., Jordan, C. E., Fuelberg, H. E.,
- 4 Sachse, G. W., Avery, M. A., Vay, S. A., Blake, D. R., Blake, N. J., Atlas, E., Fried, A.,
- 5 Sandholm, S. T., Tan, D., Singh, H. B., Snow, J., and Heikes, B. G.: Chemical composition of
- 6 Asian continental outflow over the western Pacific: Results from Transport and Chemical
- 7 Evolution over the Pacific (TRACE-P), J. Geophys. Res., 108, 10.1029/2002jd003184, 2003.
- 8 Shen, R.-Q., Ding, X., He, Q.-F., Cong, Z.-Y., Yu, Q.-Q., and Wang, X.M.: Seasonal
- 9 variation of secondary organic aerosol tracers in Central Tibetan Plateau, Atmos. Chem. Phys.,
- 10 15, 8781-8793, 2015.
- Singh, H.B.: Reactive nitrogen in the troposphere. Environ. Sci. & Technol., 21(4), 320–327,
- 12 1987.
- 13 Singh, H.B., Salas, L., Herlth, D., Kolyer, R., Czech, E., Avery, M., Crawford, J.H., Pierce,
- 14 R.B., Sachse, G.W., Blake, D.R., Cohen, R. C., Bertram, T.H., Perring, A., Wooldridge, P.J.,
- 15 Dibb, J., Huey, G., Hudman, R.C., Turquety, S., Emmons, L.K., Flocke, F., Tang, Y.,
- 16 Carmichael, G.R., and Horowitz, L.W.: Reactive nitrogen distribution and partitioning in the
- 17 North American troposphere and lowermost stratosphere, J. Geophys. Res., 112, D12S04,
- 18 doi:10.1029/2006JD007664, 2007.
- 19 Sudo, K., Takahashi, M., and Akimoto, H.: CHASER: A global chemical model of the
- 20 troposphere 2. Model results and evaluation, J. Geophys. Res, 107, 4586,
- 21 doi:10.1029/2001jd001114, 2002.
- 22 Stull, R.B.: An Introduction to Boundary Layer Meteorology, Kluwer Academic, Dordrecht,
- The Netherlands, 1988.
- 24 Talbot, R. W., Dibb, J. E., Scheuer, E. M., Kondo, Y., Koike, M., Singh, H. B., Salas, L. B.,
- 25 Fukui, Y., Ballenthin, J. O., Meads, R. F., Miller, T. M., Hunton, D. E., Viggiano, A. A.,
- Blake, D. R., Blake, N. J., Atlas, E., Flocke, F., Jacob, D. J., and Jaegle, L.: Reactive nitrogen
- 27 budget during the NASA SONEX Mission, Geophys. Res. Lett., 26, 3057-3060,
- 28 10.1029/1999GL900589, 1999.
- 29 Talukdar, R. K., Burkholder, J. B., Schmoltner, A.-M., Roberts, J. M., Wilson, R. R., and
- 30 Ravishankara, A. R.: Investigation of the loss processes for peroxyacetyl nitrate in the

Manuscript under review for journal Atmos. Chem. Phys.





- 1 atmosphere: UV photolysis and reaction with OH, J. Geophys. Res., 100, 14163–14173,
- 2 10.1029/95JD00545, 1995.
- 3 Thakur, A.N., Singh, H.B., Mariani, P., Chen, Y., Wang, Y., Jacob, D.J., Brasseur, G., Müller,
- 4 J.F., and Lawrence, M.: Distribution of reactive nitrogen species in the remote free
- 5 troposphere: data and model comparisons, Atmos. Environ., 33, 1403–1422,
- 6 doi:10.1016/s1352-2310(98)00281-7, 1999.
- 7 Tereszchuk, K.A., Moore, D.P., Harrison, J.J., Boone, C.D., Park, M., Remedios, J.J., Randel,
- 8 W.J., and Bernath, P.F.: Observations of peroxyacetyl nitrate (PAN) in the upper troposphere
- 9 by the Atmospheric Chemistry Experiment-Fourier Transform Spectrometer (ACE-FTS),
- 10 Atmos. Chem. Phys., 13, 5601-5613, 10.5194/acp-13-5601-2013, 2013.
- 11 Tian, W., Chipperfield, M., and Huang, Q.: Effects of the Tibetan Plateau on total column
- 12 ozone distribution, Tellus B, 60, 622-635, 10.1111/j.1600-0889.2008.00338.x, 2008.
- 13 Ungermann, J., Ern, M., Kaufmann, M., Müller, R., Spang, R., Ploeger, F., Vogel, B., and
- 14 Riese, M.: Observations of PAN and its confinement in the Asian summer monsoon
- anticyclone in high spatial resolution, Atmos. Chem. Phys., 16, 8389–8403, 2016.
- 16 Volz-Thomas, A., Xueref, I., and Schmitt, R.: An automatic gas chromatograph and
- 17 calibration system for ambient measureme, Environ. Sci. Pollut. Res., 9, 72-76, 2002.
- 18 von Kuhlmann, R., Lawrence, M.G., Crutzen, P.J., and Rasch, P.J.: A model for studies of
- 19 tropospheric ozone and nonmethane hydrocarbons: Model evaluation of ozone-related species,
- 20 J. Geophys. Res, 108, 4729, doi:10.1029/2002jd003348, 2003.
- 21 Wang, M., Xu, B., Cao, J., Tie, X., Wang, H., Zhang, R., Qian, Y., Rasch, P. J., Zhao, S., Wu,
- 22 G., Zhao, H., Joswiak, D. R., Li, J., and Xie, Y.: Carbonaceous aerosols recorded in a
- 23 southeastern Tibetan glacier: analysis of temporal variations and model estimates of sources
- 24 and radiative forcing, Atmos. Chem. Phys., 15, 1191-1204, doi:10.5194/acp-15-1191-2015,
- 25 2015.
- Wang, T., Wong, H.L.A., Tang, J., Ding, A., Wu, W.S., and Zhang, X.C.: On the origin of
- 27 surface ozone and reactive nitrogen observed at a remote mountain site in the northeastern
- 28 Qinghai-Tibetan Plateau, western China, J. Geophys. Res., 111, D08303,
- 29 doi:10.1029/2005JD006527, 2006.

Manuscript under review for journal Atmos. Chem. Phys.





- 1 Wang, Q.Y., Gao, R.S., Cao, J.J., Schwarz, J.P., Fahey, D.W., Shen, Z.X., Hu, T.F., Wang, P.,
- 2 Xu, X.B., and Huang, R.J.: Observations of high level of ozone at Qinghai Lake basin in the
- 3 northeastern Qinghai-Tibetan Plateau, western China, J. Atmos. Chem., 72, 19–26, 2015.
- 4 Wiegele, A., Glatthor, N., Höpfner, M., Grabowski, U., Kellmann, S., Linden, A., Stiller, G.,
- 5 and von Clarmann, T.: Global distributions of C₂H₆, C₂H₂, HCN, and PAN retrieved from
- 6 MIPAS reduced spectral resolution measurements, Atmos. Meas. Tech., 5, 723-734,
- 7 10.5194/amt-5-723-2012, 2012.
- 8 Worden, J., Jones, D.B.A., Liu, J., Parrington, M., Bowman, K., Stajner, I., Beer, R., Jiang, J.,
- 9 Thouret, V., Kulawik, S., Li, J.-L. F., Verma, S., and Worden, H.: Observed vertical
- 10 distribution of tropospheric ozone during the Asian summertime monsoon, J. Geophys. Res.,
- 11 114, D13304-D13320, 10.1029/2008jd010560, 2009.
- 12 Xiong, X., Houweling, S., Wei, J., Maddy, E., Sun, F., and Barnet, C.: Methane plume over
- 13 south Asia during the monsoon season: satellite observation and model simulation, Atmos.
- 14 Chem. Phys., 9, 783-794, 2009.
- 15 Xu, W., Lin, W., Xu, X., Tang, J., Huang, J., Wu, H., and Zhang, X., Long-term trends of
- surface ozone and its influencing factors at the Mt. Waliguan GAW station, China Part 1:
- 17 Overall trends and characteristics, Atmos. Chem. Phys., 16, 6191–6205, 2016.
- 18 Xu, W., Xu, X., Lin, M., Lin, W., Tang, J., Tarasick, D., Ma, J., and Zheng, X.: Long-term
- 19 trends of surface ozone and its influencing factors at the Mt. Waliguan GAW station, China,
- 20 Part 2: Variation mechanism and links to some climate indices, Atmos. Chem. Phys. Discuss.,
- 21 https://doi.org/10.5194/acp-2017-483, 2017.
- 22 Xue, L.K., Wang, T., Zhang, J.M., Zhang, X.C., Deliger, Poon, C.N., Ding, A.J., Zhou, X.H.,
- 23 Wu, W.S., Tang, J., Zhang, Q.Z., and Wang, W.X.: Source of surface ozone and reactive
- 24 nitrogen speciation at Mount Waliguan in western China: New insights from the 2006
- 25 summer study, J. Geophys. Res., 116, 10.1029/2010jd014735, 2011.
- 26 Yeh, T.-C., Lo, S.-W., and Chu, P.-C.: The wind structure and heat balance in the lower
- 27 troposphere over Tibetan Plateau and its surroundings, Acta Meteor. Sinica (in Chinese), 28,
- 28 108-121, 1957.
- 29 Yin, X., Kang, S., de Foy, B., Cong, Z., Luo, J., Zhang, L., Ma, Y., Zhang, G., Rupakheti, D.,
- 30 and Zhang, Q.: Surface ozone at Nam Co (4730 m a.s.l.) in the inland Tibetan Plateau:

Manuscript under review for journal Atmos. Chem. Phys.





- 1 variation, synthesis comparison and regional representativeness, Atmos. Chem. Phys.
- 2 Discuss., https://doi.org/10.5194/acp-2017-175, 2017.
- 3 Zanis, P., Ganser, A., Zellweger, C., Henne, S., Steinbacher, M., and Staehelin, J.: Seasonal
- 4 variability of measured ozone production efficiencies in the lower free troposphere of Central
- 5 Europe, Atmos. Chem. Phys., 7, 223-236, 10.5194/acp-7-223-2007, 2007.
- 6 Zhang, H., Xu, X., Lin, W., and Wang, Y.: Wintertime peroxyacetyl nitrate (PAN) in the
- 7 megacity Beijing: Role of photochemical and meteorological processes, J. Environ. Sci., 26,
- 8 83–96, 10.1016/S1001-0742(13)60384-8, 2014.
- 9 Zhang, L., Jacob, D.J., Boersma, K.F., Jaffe, D.A., Olson, J.R., Bowman, K.W., Worden, J.R.,
- 10 Thompson, A.M., Avery, M.A., Cohen, R.C., Dibb, J.E., Flock, F.M., Fuelberg, H.E., Huey,
- 11 L.G., McMillan, W.W., Singh, H.B., and Weinheimer, A.J.: Transpacific transport of ozone
- 12 pollution and the effect of recent Asian emission increases on air quality in North America: an
- 13 integrated analysis using satellite, aircraft, ozonesonde, and surface observations, Atmos.
- 14 Chem. Phys., 8, 6117–6136, doi:10.5194/acp-8-6117-2008, 2008.
- 15 Zhang, Q., Streets, D.G., Carmichael, G.R., He, K.B., Huo, H., Kannari, A., Klimont, Z., Park,
- 16 I.S., Reddy, S., Fu, J.S., Chen, D., Duan, L., Lei, Y., Wang, L.T., Yao, Z.L., Asian emissions
- 17 in 2006 for the NASA INTEX-B mission, Atmos. Chem. Phys. 9, 5131-5153, 2009.
- 18 Zhang, R., Wang, H., Qian, Y., Rasch, P. J., Easter, R. C., Ma, P.-L., Singh, B., Huang, J.,
- 19 and Fu, Q.: Quantifying sources, transport, deposition and radiative forcing of black carbon
- over the Himalayas and Tibetan Plateau, Atmos. Chem. Phys., 15, 6205–6223, 2015.
- 21 Zhao, S., Ming, J., Sun, J., and Xiao, C.: Observation of carbonaceous aerosols during 2006–
- 22 2009 in Nyaing êntanglha Mountains and the implications for glaciers, Environ. Sci. Pollut.
- 23 Res., DOI 10.1007/s11356-013-1548-6, 2013.
- 24 Zheng, X. D., Shen, C. D., Wan, G. J., Liu, K. X., Tang, J., and Xu, X. B.: ¹⁰Be/⁷Be implies
- 25 the contribution of stratosphere-troposphere transport to the winter-spring surface O₃ variation
- observed on the Tibetan Plateau, Chinese Sci. Bull., 56, 84–88, 2011.
- 27 Zhou, X., Lou, C., Li, W.L. and Shi, J.E. Ozone changes over China and low center over
- 28 Tibetan Plateau, Chin. Sci. Bull., 40, 1396–1398, 1995.

Discussion started: 25 September 2017 © Author(s) 2017. CC BY 4.0 License.





- 1 Zhu, T., Lin, W.L., Song, Y., Cai, X.H., Zou, H., Kang, L., Zhou, L.B., and Akimoto, H.:
- 2 Downward transport of ozone-rich air near Mt. Everest, Geophys. Res. Lett. 33 (23), L23809,
- 3 doi: 10.1029/2006GL027726, 2006.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 25 September 2017 © Author(s) 2017. CC BY 4.0 License.





- 1 Table 1 Concentrations of PAN at different heights above the TP. Data from literature are
- 2 acquired either through remote sensing method or global modeling, while data from this work
- 3 are estimates based on analysis of ground measurements.

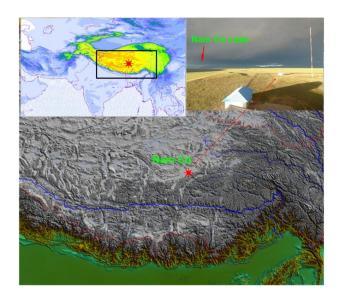
Average PAN concentration(ppb)	Period	Height	Reference
around 0.5	25 May 2012	~350 hPa	this work
0.6-0.7	22 August 2011	~350 hPa	
0.35-0.45	March 2003	333 hPa	Moore and Remedios (2010)
0.15-0.23		234 hPa	
0.35-0.5	August 2003	278 hPa	
0.15-0.23		185 hPa	
0.1-0.15	October 2007	12km	Wiegele et al.(2012)
0.1-0.2	21 October 2003	12km	Glatthor et al.(2007)
0.30-0.50	June-August 2008	2-6km	Fischer et al.(2014)
0.2-0.4		6-10km	
0.15-0.2	June-September 1995-2004	6-10km	Fadnavis et al. (2014)

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2 Fig.1 Map showing location of the observation site and local environment.

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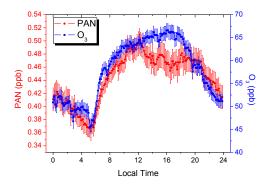


Fig.2 Diurnal patterns of PAN and O₃. All data are processed as 10 minutes resolution. The vertical bars represent one standard error of the mean.

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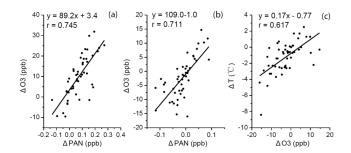


Fig.3 Scatter plots of ΔPAN (variation of the PAN concentration), ΔO_3 (variation of the O_3 concentration) and ΔT (variation of temperature) in specific time spans: (a) from 5:00 LT to 9:00 LT; (b, c) from 2:00 LT to 4:00 LT. All correlations shown in the figures are statistically significant at α =0.01.

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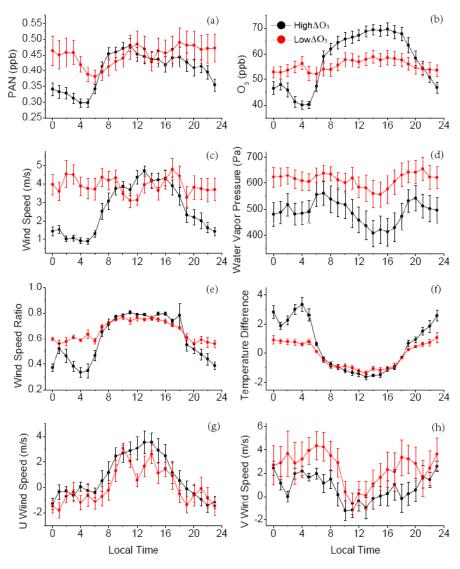


Fig. 4 Diurnal patterns of PAN (a), O_3 (b), Wind Speed (c), Water Vapor Pressure (d), Wind Speed Ratio (e, ratio of 10 meters height wind speed and 2 meters height wind speed), Temperature Difference (f, subtraction of 20 meters height temperature and 10 meters height temperature), U wind speed (g) and V wind speed (h). Black curves represent diurnal curves of 15 days with greatest ΔO_3 from 7:00 BJT to 11:00 BJT, and red curves represent diurnal curves of 15 days with smallest ΔO_3 correspondently. The vertical bars represent one standard error of the mean.

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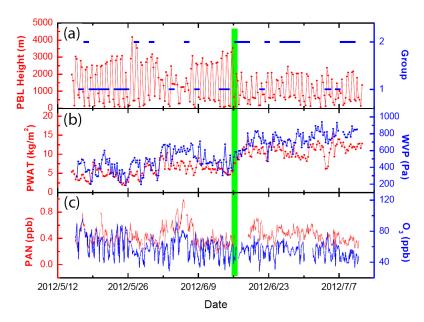


Fig. 5 Distribution of two groups of days (classified following Fig. 4) and time series the PBL height, PWAT (Precipitable Water of Entire Atmosphere), WVP (Water Vapor Pressure), PAN and O₃. The PBL Height and PWAT are acquired from the FNL data with temporal resolution of 6 hours. WVP are calculated and processed as 6-hours resolution data from field observation. PAN and O₃ concentrations are processed as hourly data.

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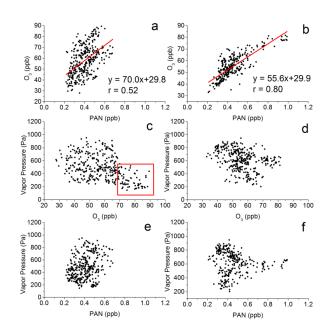


Fig. 6 Scatter plots of hourly O_3 -PAN, Vapor Pressure- O_3 , Vapor Pressure-PAN of group 1 (a,c,e) and group 2 (b,d,f), following Fig. 4. The correlation shown in Figs. 6(a) and 6(b) are significant at α =0.01. The data points within the red rectangle in Fig. 6(c) represent O_3 levels higher than 70 ppb and WVP lower than 500 hPa.

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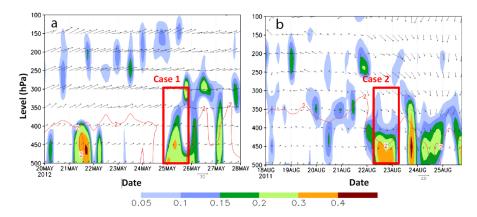


Fig. 7 Omega (shaded), specific humidity (red line) and horizontal wind field in dependence of time and height in two time frames. (a) From 20 to 28 May 2012; (b) From 15 to 25 August 2011. Case 1 and Case 2 correspond to two significant downdraft events.

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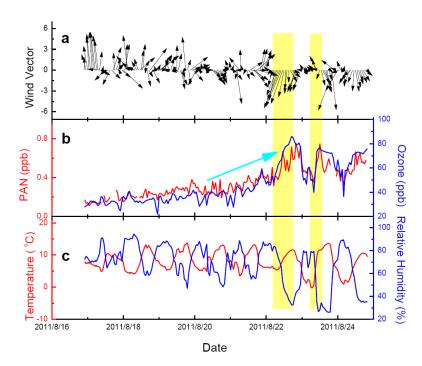


Fig. 8 Time series of (a) surface wind vectors, (b) PAN and O_3 , and (c) temperature and relative humidity during 16-25 August 2011. Yellow shadows represent the short periods controlled by downward motion. The blue arrow indicates the increasing trend of PAN and O_3 .

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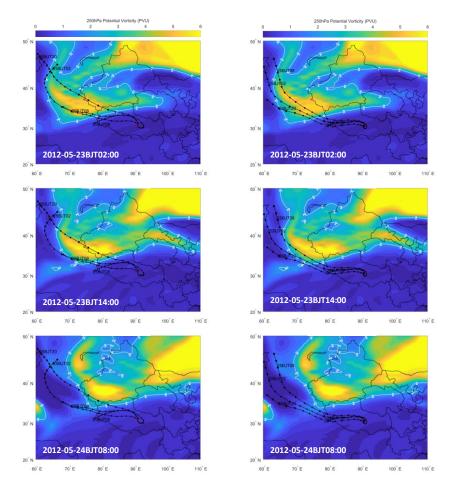


Fig. 9 Plots showing 250 hPa potential vorticity fields at three timepoints during 23-24 May 2012 and back trajectories of air masses arriving at 500 m (left) and 1500 m (right) above ground of the NMC site (red star) during 25-26 May 2012.

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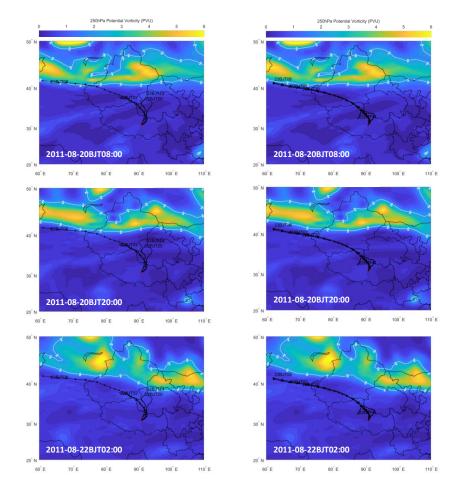


Fig. 10 Plots showing 250 hPa potential vorticity fields at three timepoints during 20-22 August 2011 and back trajectories of air masses arriving at 500 m (left) and 1500 m (right)

4 above ground of the NMC site (red star) during 22-23 August 2011.

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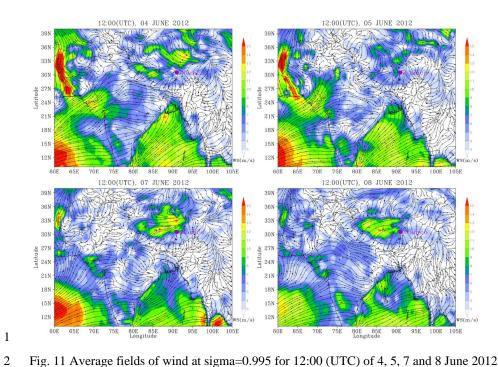


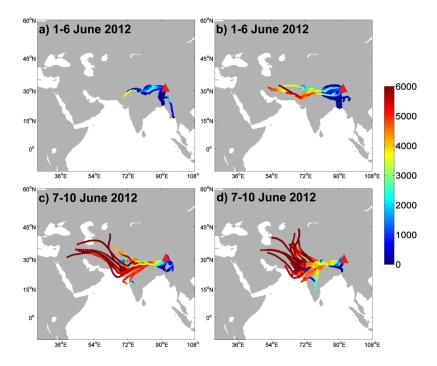
Fig. 11 Average fields of wind at sigma=0.995 for 12:00 (UTC) of 4, 5, 7 and 8 June 2012.

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- 2 Fig. 12 Backward air trajectories arriving at NMC with endpoint heights of 500 meters (a,c)
- 3 1500 meters (b,d) for the periods 1-6 June 2012 (a,b) and 7-10 June 2012 (c,d).