Morphological transformation of soot: investigation of microphysical processes during the condensation of sulfuric acid and limonene ozonolysis product vapors

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Abstract.

The morphological transformation of soot particles via condensation of low-volatility materials constitutes a dominant atmospheric process with serious implications for the optical and hygroscopic properties, and atmospheric lifetime of the soot. We consider the morphological transformation of soot aggregates under the influence of condensation of vapors of sulfuric acid, and/or limonene ozonolysis products. This influence was systematically investigated using a Differential Mobility Analyzer-Aerosol Particle Mass Analyzer (DMA-APM) and the Tandem DMA techniques integrated with a laminar flow-tube system. We hypothesize that the morphology transformation of soot results (in general) from a two-step process, i.e., (i) filling of void space within the aggregate; (ii) growth of the particle diameter. Initially, the transformation was dominated by the filling process followed by growth, which led to the accumulation of sufficient material that exerted surface forces, which eventually facilitating further filling. The filling of void space was constrained by the initial morphology of the fresh soot as well as the nature and the amount of condensed material. This process continued in several sequential steps until all void space within the soot aggregate was filled. And then "growth" of a spherical particle continued as long as vapors condensed on it. We developed a framework for quantifying the microphysical transformation of soot upon the condensation of various materials. This framework used experimental data and the hypothesis of "ideal sphere growth" and void filling to quantify the distribution of condensed materials in the complementary filling and growth processes. Using this framework, we quantified the percentage of material consumed by these processes at each step of the transformation. For the largest coating experiments, 6%, 10%, 24% and 58% of condensed material went to filling process, while 94%, 90%, 76% and 42% of condensed material went to growth process for 75 nm, 100 nm, 150 nm and 200 nm soot particles, respectively. We also used the framework to estimate the fraction of internal voids and open voids. This information was then used to estimate the volume equivalent diameter of the soot aggregate containing internal voids and to calculate the dynamic shape factor, accounting for internal voids. The dynamic shape factor estimated based on the traditional assumption

(of no internal voids) differed significantly from the value obtained in this study. Internal voids are accounted for in the experimentally derived dynamic shape factor determined in the present study. In fact, the dynamic shape factor adjusted for internal voids was close to 1 for the fresh soot particles considered in this study, indicating the particles were largely spherical. The effective density was strongly correlated with the morphological transformation responses to the condensed material on the soot particle and the resultant effective density was determined by the (i) nature of the condensed material; (ii) morphology and size of the fresh soot. In this work we quantitatively tracked *in-situ* microphysical changes in soot morphology, providing details of both fresh and coated soot particles at each step of the transformation. This framework can be applied to model development with significant implications for quantifying the morphological transformation (from the viewpoint of hygroscopic and optical properties) of soot in the atmosphere.

1 Introduction

Soot containing black carbon (BC) produced from incomplete combustion of fossil-fuel or biomass is ubiquitous in the atmosphere, and represents a major constituent of atmospheric aerosols (Bond et al., 2013). Through direct radiative forcing, soot influences the climate significantly as it efficiently absorbs light in the visible and infrared spectral ranges.

Once emitted into the atmosphere, soot particles undergo morphological transformations due to several ageing processes. Soot ageing includes adsorption and condensation of semi-volatile vapors, coagulation with pre-existing aerosol particles, heterogeneous reactions with atmospheric gaseous species, and in-cloud processing (Khalizov et al., 2009). The physical and chemical properties of soot particles change considerably during these ageing processes. For example, exposure to sulfuric acid changes the morphology, enhances the absorption and scattering, and increases the hygroscopicity of fresh soot particles (Zhang and Zhang, 2005; Zhang et al., 2008). The cloud forming properties of soot can be altered, and climate forcing can be influenced (albeit indirectly), by coating the soot with soluble material. For example, atmospheric ageing of coated soot enhances the hygroscopicity of the coatings and, hence, aged soot particles can act as efficient cloud condensation nuclei (CCN) or ice nuclei (IN) (Ammann et al., 1998; Henning et al., 2012). Transformation from fresh hydrophobic soot to processed hygroscopic soot can also have adverse health effects as, compared with hydrophobic fresh soot, aged hygroscopic soot has an altered deposition in the lungs (Löndahl et al., 2008).

Soot particles are emitted as complex aggregates. Their morphology and life-span in the atmosphere are directly correlated with the atmospheric processing (e.g., the type and the amount of material condensed on the aggregate). The mass mobility exponent (*Dfm*) is an indirect measure of the morphology of irregularly shaped aggregated particles (DeCarlo et al., 2004). The DMA-APM technique characterizes the properties of fresh or coated soot aggregates by performing simultaneous measurements with a differential mobility analyzer (DMA) and an aerosol particle mass analyzer (APM) (McMurry et al., 2002). With this technique, the mobility diameter and mass of the same aggregate particles are accurately measured, and the

mass mobility exponent is determined through fitting of the mass-mobility relationship. The mass mobility exponent, *Dfm*, is defined as:

$$Dfm = \frac{\ln(m_p) - \ln(k)}{\ln(D_p)},$$ (1)

where D_p, m_p, and k are the particle mobility diameter, particle mass, and a proportionality constant, respectively.

The effective density of a particle, ρ_{eff}, can also be determined from simultaneous measurements of the mass and mobility diameter, and is given as follows:

$$\rho_{eff} = \frac{6m_p}{\pi D_p^3},$$ (2)

where m_p is the particle mass and D_p is the mobility diameter.

A *Dfm* of 3.0 is obtained for spherical particles, whereas values of 2.2–2.4 are typically obtained for fresh diesel soot (Park et al. 2003). This range of values is indicative of an open-structure aggregate. Schneider et al. (2006) found that particles resulting from biomass burning have variable *Dfm* values: for dry beech fuel *Dfm* = 2.1, whereas for humid oak fuel *Dfm* was close to 3.0. Fresh soot from various sources, such as a diesel engine, diffusion-type burner, and premixed-type burner,

15 have effective densities of 0.2–1.2 g cm^{-3}, which typically decrease with increasing mobility diameter (Ghazi et al., 2013; Rissler et al., 2013).

Previous studies (Zhang et al., 2008; Weingartner et al., 1995) have found that the mobility diameter of fresh hydrophobic soot below water saturation (relative humidity < 100%) undergoes a relatively small amount of hygroscopic growth. This

20 indicates that, at sub-saturation, water vapor has little influence on the soot morphology. However, soot particles will become hydrophilic if water-soluble compounds, such as sulfuric acid or organics condense on the surface, and then collapse into a spherical shape when the relative humidity (RH) is increased (Zhang et al., 2008; Pagels et al., 2009). The effective density of coated and annealed soot increases, consistent with restructuring of the soot cores after they acquire sufficient fractions of coating material (Cross et al., 2010; Qiu et al., 2012; Saathoff et al., 2003; Pagels et al., 2009). The extent of

25 restructuring depends on the production conditions of the soot, as well as on the mass fraction and type of the coating material. Carbon soot aggregates produced from spark discharge collapse to relatively compact shapes at sub-saturation, whereas diesel soot undergoes limited restructuring (Weingartner et al., 1997). Different dicarboxylic acids have different effects on soot restructuring. Exposure to glutaric acid (C$_5$, low RH$_{deliquescence}$) leads to significant collapse of the soot cores, whereas condensation of succinic acid (C$_4$, high RH$_{deliquescence}$) has no influence on the restructuring (Xue et al., 2009).

30 Secondary organic aerosol (SOA) produced from OH-initiated oxidation of both toluene and isoprene results in an increase in the effective density and a significant decrease in the dynamic shape factor of the cores (Qiu et al., 2012; Khalizov et al., 2013). However, the effect of limonene SOA on soot ageing has yet to be studied, despite the fact that limonene comprises up to 10% of the total global monoterpene emissions (Geron et al., 2000; Guenther et al., 2012) and has very high potential

for SOA formation. Limonene is efficiently oxidized to SOA with little or no carbon loss, generating water-soluble products (Pathak et al., 2012) due to two different unsaturated bonds in the molecule: an endocyclic tri-substituted double bond and an exocyclic terminal double bond. In this work, we determine the effect of limonene ozonolysis products on soot morphology.

5   Sulfuric acid – formed during the combustion processes, condenses on soot and provides active acidic surfaces for the condensation of semi-volatile SOA. The acid-catalyzed SOA reactions can enhance SOA formation significantly, as shown in laboratory experiments (Jang et al. 2002). However, soot – SOA interactions mediated by surface acidity have barely been investigated. For e.g., the effect of an SOA-added acidified soot surface on the morphology, density, and, hence, the lifetime of soot in the atmosphere as well as the radiative properties remains unexplored.

Condensation of such low-volatility materials on soot particles leads to morphological transformation of these particles with serious implications for the corresponding optical and hygroscopic properties, and atmospheric lifetime. Nevertheless, many aspects of this transformation (including the mechanism of void-space filling within fractal soot and the restructuring and *in-situ* growth upon condensation of different materials) remain unclear. The correlation (i.e., mutual dependence and/or
15   sequential dependence) or lack thereof between these processes must also be clarified. To date, these microphysical aerosol processes are neither described in current models nor fundamentally studied and quantified. Therefore, major knowledge gaps remain, thereby hindering the development of robust modeling tools for improved climate predictions.

To address these issues, a study of chemical and physical processes encompassing the interaction of SOA with soot on the
20   acidic surfaces is performed using a flow-tube reactor at the University of Gothenburg, Sweden. In this study, soot processing is investigated using controlled condensation of sulfuric acid and SOA onto a laboratory-generated propane flame soot. We perform direct measurements of the change in mobility diameter and mass upon processing, using the TDMA and DMA-APM techniques. The mass-mobility relationship is then used to calculate the mass mobility exponent, effective density, and dynamic shape factor of soot subjected to processing of atmospheric relevance. Subsequently, the results are
25   compared with those of previous studies. A framework is developed for quantifying the morphological state and subsequent transformation of soot, i.e., the utilization of material for filling and growth upon the condensation of material. Using this framework, a method is derived for (i) quantifying the fraction of internal voids and open voids in the soot aggregate, and (ii) determining the volume equivalent diameter inclusive of unfilled voids.

2 Experimental methods

30   Soot particles emitted from a premixed-diffusion propane flame soot generator with and without coatings were studied. The coating processes were systematically performed in a temperature and RH-controlled laminar flow tube as shown in Fig. 1.

The key instruments used included an APM, two DMAs, and two condensation particle counters (CPCs). A schematic of the experimental setup is shown in Fig. 1.

2.1 Soot generation

Laboratory-scale soot-SOA interaction process experiments were performed using a versatile propane premixed-diffusion flame soot generator built, in-house, to produce sub-micron soot particles with desired particle size and concentration. Different flame conditions were realized by regulating the fuel to air equivalence ratios. Details of the flame selection and flame conditions are provided elsewhere (Pei et al, 2018).

The soot from the propane flame generator was dried with a silica gel diffusion drier, and a Thermo-Denuder (TD) was then used to remove each of the co-produced primary organic components, as shown in Fig. 1. The TD was maintained at 400 °C and the particle residence time within the TD was ~1.6 min. Further downstream, NOx was removed using a NOx denuder filled with charcoal granite (Activated Charcoal Powder USP, Spectrum Chemical Mfg. Corp., USA). The poly-dispersed pure soot cores were charged by a bipolar charger (^{63}Ni) and then selected using the first DMA (DMA1, model 3081, TSI Inc., Shoreview, MN, USA, aerosol flow rate 0.4 L min^{-1}, and sheath flow rate 6.0 L min^{-1}) for size-resolved modulation and subsequent characterization experiments.

2.2 Soot modification

Soot particles were modulated using the experimental setup of the Tandem Differential Mobility Analyzer (TDMA) with the integrated flow-tube system and then characterized using the APM, DMAs, and CPC, as shown in Fig. 1. The size selected mono-disperse soot core surface was modified by sulfuric acid coatings using a cylindrical glass mixing chamber (length: 47 cm, diameter: 10 cm) equipped with a temperature-regulated bath system. The modified soot particles were then further modulated by SOA generated via limonene ozonolysis in a low NOx, dark laminar flow tube (flow rate: 1.52 L min^{-1} and effective particle residence: 4.8 min at 25 °C and regulated RH).

Four nominal sizes of soot core, i.e., 75, 100, 150, and 200 nm, were selected using DMA1. The experimental matrix included four major procedural steps: 1) characterization of the pure soot core; 2) modification of the soot core surface with sulfuric acid and characterization of the modulated soot; 3) modification of soot core surface with SOA and characterization of the modulated soot; 4) modulation of the sulfuric acid modified soot with limonene ozonolysis products and subsequent characterization.

The acid-induced surface modifications were performed at three different temperatures namely, 1 °C (small=S), 5 °C (medium=M), and 25 °C (large=L), using a temperature-regulated bath system. In the SOA modulation steps, the limonene

concentrations in the flow tube were regulated using flowing zero air over limonene in a diffusion vial submersed in a temperature-regulated bath system (Jonsson et al., 2008). The limonene mixing ratios in the flow-tube (56 ppb, 73 ppb, 138 ppb) were precisely regulated by setting the temperatures of the limonene vial bath system to 1 °C (small=S), 5 °C (medium=M), and 15 °C (large=L) at a constant zero air flow rate (0.002 L min^{-1}). Ozone was generated by passing zero air (Ultra Zero Air Generator GT 3000, LNI SCHMIDLIN AG) through a UV-lamp unit (SOG-3, UVP). The ozone concentration used in the experiments was set to a constant value of 1432 µg m^{-3} (~730 ppb) and monitored using a UV photometric ozone analyzer (model 49C, Thermo Environmental Instruments Inc.). A low constant RH (5±1%) was maintained in all cases. The nucleation of limonene ozonolysis products may happen during the modification process; however, since the homogeneous nucleated SOA particles did not have a charged soot core, they cannot be selected by DMA2 and will not influence the soot characterization results.

For (i) pure soot core characterization, the modulation steps involving sulfuric acid and limonene SOA were all bypassed (see Fig. 1), (ii) determining the effect of sulfuric acid on soot mobility and morphological properties, only the modulation steps involving limonene SOA were bypassed (by turning off the ozone production through the UV-lamp power supply; see Table S1 of the supplementary information for the experimental matrix). (iii) determining the effect of pure limonene SOA on soot mobility and morphological properties, the modulation steps involving sulfuric acid were bypassed

2.3 Aerosol characterization

2.3.1 DMA+CPC for the particle-size number distributions

The post modulation peak diameters of the mono-disperse soot cores were scanned using a combination of a DMA (DMA2, Vienna type, length 280 mm, inner/outer radius 25.0/33.3 mm, aerosol flow rate 2.25 L min^{-1}, and sheath flow rate 10.0 L min^{-1}) and a condensation particle counter (CPC – model 3775, TSI Inc., Shoreview, MN, USA, flow rate 0.3 L min^{-1}), which served as a scanning mobility particle sizer (SMPS).

2.3.2 DMA-APM for the size-resolved particle mass

After DMA2, the mass of the mono-disperse particle population was measured using a series combination of an APM (model APM-3600, Kanomax) and a CPC (model 3775, TSI Inc., Shoreview, MN, USA, flow rate: 0.3 L min^{-1}, see Fig. 1). The APM consists of two concentric cylinders rotating at the same angular speed. A voltage was applied over the cylinders and the particles introduced in the gap between these cylinders experienced a centrifugal force. The APM transmitted only those particles with a specific mass at which the electric force is equal to the centrifugal force. These particles were then counted with a CPC (Rissler et al., 2013). The DMA-APM system was calibrated using two sizes of polystyrene latex spheres (Duke Scientific Corp., USA), in accordance with previously described methodology (McMurry et al., 2002).

Moreover, the mass sensitivity of multiply charged particles was determined and for all soot particle sizes, the mass was overestimated by <5% (see supplementary information).

2.4 Data analysis

The particle diameter growth factor, *Gfd*, is calculated from:

5   $Gfd = \frac{D_p}{D_{p,0}}$, $\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad$ (3)

where D_p and $D_{p,0}$ are the mobility diameter of particles at a given reaction time and the mobility diameter of the fresh soot particles, respectively.

The particle effective density ρ_{eff} can be calculated from Eq. (2). The particle organic mass fraction fm_{org} is determined from:

10   $(m_p - m_{SA} - m_0)/m_p$, and the particle sulfuric acid mass fraction fm_{SA} is determined as m_{SA}/m_p, where m_p, m_{SA} and m_0 are mass of coated particle, mass of sulfuric acid coating and mass of fresh soot, respectively. Similarly, the material density ρ_m is calculated from the material density of the soot (ρ_{soot}: 1.77 g cm^{-3}), organic coating (ρ_{org}), and sulfuric acid (ρ_{SA}) if volumetric additivity is assumed. The material density of limonene ozonolysis products (ρ_{org}: 1.20 g cm^{-3}) measured by Chen et al. (2010) is used in this study. In this study, aerosol mass spectrometer (AMS) was also used in parallel with DMA-

15   APM system. The O:C and H:C determined via AMS measurements were used to estimate the organic aerosol density (1.26±0.04 g cm^{-3}) with the method given by Kuwata et al. (2012). The material density of limonene ozonolysis products (1.20 g cm^{-3}) used in this study is similar to the AMS results and the value (1.3±0.2 g cm^{-3}) reported by Saathoff et al., (2009). The material density of sulfuric acid (sulfuric acid-water mixture) at 5±1% RH (ρ_{SA}: 1.84 g cm^{-3}) is estimated using the same method employed by Pagels et al. (2009). The material density ρ_m is calculated from:

20   $m_{SA} = m_{p,soot+SA} - m_0$, $\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad$ (4)

$fm_{SA} = \frac{m_{SA}}{m_p}$, $\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad$ (5)

$fm_{org} = \frac{m_p - m_{SA} - m_0}{m_p}$, $\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad$ (6)

$\frac{1}{\rho_m} = \frac{fm_{org}}{\rho_{org}} + \frac{fm_{SA}}{\rho_{SA}} + \frac{1 - fm_{org} - fm_{SA}}{\rho_{soot}}$, $\qquad\qquad\qquad\qquad$ (7)

The mass equivalent diameter D_{me} corresponds to a spherical particle of the same mass, and can be calculated from the

25   particle mass m_p and the material density ρ_m:

$D_{me} = \sqrt[3]{\frac{6m_p}{\pi \rho_m}}$, $\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad\qquad$ (8)

For fresh soot ρ_m is the inherent material density of the soot, whereas for coated particle ρ_m is the average inherent material density over all the components of the particle, which can be calculated from Eq. (4)–(7). It should be pointed out that this definition of the mass equivalent diameter is identical to the definition of volume equivalent diameter used by McMurry et al. (2002) and Park et al. (2003).

The change in particle D_{me} is expressed as the mass equivalent coating thickness Δr_{me}:

$$\Delta r_{me} = \frac{D_{me} - D_{me,0}}{2} \, , \tag{9}$$

where $D_{me,0}$ and D_{me} are the mass equivalent diameters of the fresh and the coated soot particles, respectively.

10   The dynamic shape factor χ can be calculated from the measured mobility diameter D_p and the mass equivalent diameter D_{me} (Baron and Willeke, 2001):

$$\chi = \frac{D_p C_{me}}{D_{me} C_p} \, , \tag{10}$$

where C_{me} and C_p are the Cunningham slip correction factors for particles with diameters D_{me} and D_p, respectively. The dynamic shape factor is derived directly from Stokes' law. It is the ratio of the drag force exerted on the irregular particle

15   divided by the drag force exerted on its mass equivalent sphere, when travelling at the same speed.

The void space fraction (F_{vs}), i.e., volume of voids/total volume of particles derived from the mobility diameter, is estimated from the D_{me} and D_p of fresh and coated soot (Baron and Willeke, 2001; Zhang et al., 2016):

$$F_{vs} = 1 - \frac{D_{me}{}^3}{D_p{}^3} \, , \tag{11}$$

In this study, the model developed by Sorensen (2011) is used to describe the relation among the particle mobility diameter D_p, primary particle (soot spherules) diameter d_{pp}, and number of primary particles (soot spherules) in a soot aggregate (N). We assume that the primary particles of soot aggregates are in point contact and the material density of these particles is the same as that (1.77 g cm^{-3}) of the soot.

25   $$m_p = N \cdot \frac{\pi}{6} \rho_m d_{pp}^3 \, , \tag{12}$$

$$D_p = d_{pp} N^x, N < 1000 \, , \tag{13}$$

where $x = 1/Dfm = 0.46$ assuming a Dfm of 2.17 for the soot aggregate. In the original Sorensen model, a fixed exponent x was used for $N < 100$. However, Rissler et al. (2013) suggested that the formula should also be used for N values of up to 1000.

2.5 Framework for quantifying the morphological transformation

A framework was developed for quantifying the state of the morphological transformation of soot, i.e., the utilization of material for filling and growth, as shown in Fig. 2. Fig. 2(a) shows the particle diameter growth factor (Gfd) as a function of the coating thickness Δr_{me} associated with each measurement point. The ideal growth line, denoted by a black solid line, describes condensation of material on a perfect incompressible solid sphere with the same initial mobility diameter and mass as a fresh soot particle. We hypothesize that when material condenses on the soot aggregate, the growth process will be described by Gfd parallel to the ideal growth line, while void filling will be described by a line parallel to the x-axis. The coating thickness for void filling ($\Delta r_{me,f}$) and the coating thickness for particle growth ($\Delta r_{me,g}$) are indicated by the red arrow (parallel to the x-axis) and the green arrow (parallel to the ideal growth curve), respectively. From this framework, the fraction of void space filled ($F_{vs,f}$), volumes of material utilized for void filling (V_f) and diameter growth (V_g), and the percentages of material utilized for filling (P_f) and growth (P_g) can be derived as follows (based on the assumption of a concentric core shell structure; see Fig. 2(b) for illustration):

$$F_{vs,f} = \frac{(D_{me,0}+2\Delta r_{me,f})^3 - D_{me,0}{}^3}{D_p{}^3 F_{vs}}, \tag{14}$$

$$V_f = \frac{\pi}{6}\left[(D_{me,0} + 2\Delta r_{me,f})^3 - D_{me,0}{}^3\right], \tag{15}$$

$$V_g = \frac{\pi}{6}\left[(D_{me,0} + 2\Delta r_{me,f} + 2\Delta r_{me,g})^3 - (D_{me,0} + 2\Delta r_{me,f})^3\right], \tag{16}$$

$$P_f = \frac{V_f}{V_f + V_g}, \tag{17}$$

$$P_g = \frac{V_g}{V_f + V_g}, \tag{18}$$

During the experiment, the soot undergoes step-wise morphological transformation (see section 3.3 for details). The fraction of internal voids (F_i) and the fraction of open voids (F_o) in the soot aggregate are determined based on the following hypotheses: (i) in the case of SOA, the open voids are filled prior to the onset of growth, and (ii) at the onset of growth, F_o is equal to the fraction of void space filled ($F_{vs,f}$), whereas $F_i = 1 - F_o$. The volume equivalent diameter including internal voids ($D_{ve,i}$) is then given as:

$$D_{ve,i} = \sqrt[3]{D_{me}{}^3 + D_p{}^3 F_{vs} F_i}, \tag{19}$$

Then $D_{ve,i}$ is used to calculate the dynamic shape factor with internal voids (χ_i):

$$\chi_i = \frac{D_p C_{ve,i}}{D_{ve,i} C_p}, \tag{20}$$

where $C_{ve,i}$ is corresponding Cunningham slip correction factors for $D_{ve,i}$.

3 Results

3.1 Fresh soot properties

As previously mentioned, fresh pure soot is obtained by heating the soot particles in the TD at 400 °C to remove co-produced organics from the soot generator. For the soot transformation study, four fresh pure soot particle sizes with nominal mobility diameters (D_p) of 75, 100, 150, and 200 nm are selected using DMA1. The corresponding actual mobility diameters (actual D_p) are scanned using DMA2 and CPC, as shown in Fig. 1. The D_p, mass equivalent diameter (D_{me}), mass (m_p), effective density (ρ_{eff}), and the dynamic shape factor (χ) of the particles are listed in Table 1. As the table shows, the final measured D_p of the modified soot differs only ~5% from the instrumental accuracy of the DMA1 and the SMPS (DMA2+CPC) system. The calculated D_{me} is smaller than the D_p corresponding to each size, i.e., D_{me} of 57.1, 70.8, 95.3, and 116.2 nm are obtained for D_p of 75, 100, 150, and 200 nm, respectively. This suggests that the soot particle contains significant fractions of void space (F_{vs}). In fact, the estimated fraction of void space within a fresh soot particle (listed in Table 1) is fairly large, e.g., F_{vs} values of 58%, 63%, 72%, and 78% are obtained for D_p of 75, 100, 150, and 200 nm, respectively. These void fractions account for both internal and external voids (see section 3.4). However, the corresponding dynamic shape factor (χ), a parameter describing the sphericity of the particle (see Table 1), is estimated based on the assumption of no internal voids. The value of χ increases from 1.66 to 2.29 with increasing (from 75 to 200 nm) D_p. This indicates, as reported in previous studies (Xue et al., 2009; Khalizov et al., 2013), that the shape irregularity of the particle increases with increasing mobility diameter.

The morphological structure of the soot aggregate may also be described via the mass-mobility exponent (Dfm), which characterizes the primary particles (spherules) and describes their arrangement within the aggregate. Averaged over all four sizes, i.e., 75, 100, 150, and 200 nm, a Dfm of 2.28 is derived by fitting the actual D_p and the soot particle mass. This value is consistent with aggregates reported for fresh soot particles from a diffusion propane burner (Pagels et al., 2009). We used the Dfm of pure BC particles in Eq. (11) and Eq. (12) to determine the size of the primary spherules (d_{pp}), and a value of ~28 nm is obtained for all four sizes of the soot aggregates. The d_{pp} reported in previous studies vary significantly and are typically lower than the value obtained in this study. For e.g., Pagels et al. (2009) and Zhang et al. (2008) obtained a d_{pp} of ~15 nm via TEM image analysis. Similarly, Rissler et al. (2013) obtained, via TEM analysis, values of 24, 27, 28, 11, and 13 nm, for particles (geometric mean diameter: 50 nm) associated with a heavy-duty transient diesel engine, heavy-duty idling diesel engine, light-duty idling diesel engine, candle, and propane flame soot, respectively. Our d_{pp} (i.e., 28 nm) is similar to that obtained from field measurements performed by Rissler et al. (2014), indicating that the soot aggregates considered in this study are similar to real-world soot particles. According to the Sorensen model, the d_{pp} determines the void fraction in the

aggregate. Therefore, quantifying the d_{pp} is essential for a complete description of the morphological state of the soot aggregate.

Generally, comparison with previous studies is difficult, owing to the use of different fresh soot material as the substrate for subsequent soot transformation studies. The properties reported in selected studies are summarized in Table S4. As the table shows, fresh soot particles with similar mobility diameter (D_p ~100 nm) have similar Dfm (2.14–2.28), but widely varying (15 nm–45 nm) d_{pp}. This indicates that the soot aggregates are formed through a similar coagulation process after the nascent primary spherules are formed in the flame. The results reported in the literature can be classified into two groups: (1) Pagels et al. (2009), Xue et al. (2009), Qiu et al. (2012), Khalizov et al. (2013), and Peng et al. (2016) generated similar soot with χ of ~2.2 and d_{pp} of 15–21 nm; (2) Guo et al. (2016) and Ghazi and Olfert (2013) generated similar soot with relatively compact morphology characterized by $\chi = 1.5$ and $d_{pp} = 45$ nm. The compactness of the soot considered in this study ($\chi = 1.81$ and $d_{pp} = 28$ nm) represents an intermediate of these two groups. This results from the fact that, for any given soot core mobility diameter, the porosity of the soot aggregate increases with decreasing d_{pp}. In previous studies, χ was calculated based on $D_{me} = D_{ve}$. In this work, D_{ve} increases whereas χ_i decreases when we attempt to adjust for internal voids (see sections 3.3 and 3.4 for the experimentally determined open-void fractions and relevant discussions).

3.2 Formation of limonene ozonolysis SOA

Limonene SOA was formed on the soot surface regulated by VOC, ozone and sulfuric acid concentrations in the experiments. In general, compared with those formed from the reactions of VOC and ozone, significantly higher amounts of SOA formed when the reactions occurred on the acid-coated soot surfaces. The levels of acidity on the surface play an important role in this formation. Fig. 3 shows the masses of different levels of sulfuric acid and SOA coatings associated with a 200 nm soot core. The amounts of sulfuric acid and limonene SOA coatings on a soot particle are classified as small (S), medium (M), and large (L). The mixed coating experiments cover the full matrix, i.e., BC coated with each of the small, medium, and large amounts of sulfuric acid is also co-coated with small, medium, and large amounts of limonene SOA. Soot cores of other sizes (75, 100, and 150 nm) exhibit similar trends to those observed for the 200 nm soot. Acidity-induced enhancement of the SOA mass is attributed to acid-catalyzed heterogeneous reaction mechanisms (such as hemiacetal and acetal formation) of carbonyls in SOA, polymerization, and aldol condensation (Jang et al., 2002). The mass of SOA formed on the BC particle surface under large SOA and large sulfuric conditions was significantly higher than the total mass of large SOA formed under large sulfuric acid only and on the acid-free surface. Therefore, we achieved various particle coating thicknesses, which allowed a thorough investigation of the morphological transformation of the BC particle.

3.3 Growth of the particle mobility diameter

The mobility diameter growth factor (Gfd) implies that the morphological transformation of soot aggregates results from condensation of the material. Fig. 4(a)–(d) shows Gfd, associated with initial fresh soot mobility diameters of 75, 100, 150, and 200 nm, as a function of the mass equivalent coating thickness (Δr_{me}). The Gfd values are measured for the full experimental matrix described above. In Fig. 4, the ideal growth curve is denoted by the black dashed line and the pie chart denoting each data point shows the chemical composition of fresh and processed soot particles. The black, red, and green colors in each pie chart represent the mass fraction of black carbon, sulfuric acid, and organics, respectively. We hypothesize that the morphological transformation of soot particles result from the: (i) filling of void space within a fractal soot aggregate; (ii) growth of mobility diameter identical to that of the ideal sphere (parallel to black dashed line); (iii) rearrangement of primary spherules, i.e., collapse, within a fractal soot aggregate due to surface forces exerted by the condensed material. Condensational growth of the material will increase the mobility size and mass of the particles. Adding material to the void space of the soot structure will increase the mass, but the mobility size will remain the same or decrease due to collapse. We have observed all three effects (see Fig. 4). The morphological transformation of the soot fraction results from a combination of two or more of these processes and, when expressed as the Gfd, exhibits a strong particle size dependence. The morphological properties associated with the four sizes of fresh soot are shown in Table 1. Furthermore, the Gfd associated with different coatings are illustrated by purple lines and the trends are summarized as follows, the: (i) line parallel to the ideal sphere growth line (dashed) indicates that all material added was utilized for growth of the particle diameter; (ii) line parallel to the x-axis indicates that all condensed material was utilized for filling the void space within a fractal soot aggregate; (iii) negative slope indicates a combination of void space filling and collapse of the soot particle.

The soot structure of fresh soot particles with diameter and mass of 75 nm and 0.17 fg (Exp. 1), respectively, is relatively resistant to collapse. The major morphological transformation occurs sequentially in a step-wise manner, with filling of void space (shown by the solid purple line parallel to the x-axis) and growth of the diameter (solid purple line parallel to the black dashed line) as a function of Δr_{me} (mass equivalent coating thickness; see Fig. 4(a)). When the heaviest coating is applied to the 75 nm soot aggregate with large sulfuric acid (0.26 fg) + large SOA (2.37 fg) (Exp. 15), 94% of the total volume of the condensed material is utilized for diameter growth (see Fig. 4(a)). Only 6% is utilized for filling the void space, but this 6% has filled 97% of the void space within the soot particle (Table 2). The morphological evolution of the soot appears to be independent of the type of condensed material. In other words, in the present study, pure sulfuric acid, pure SOA, and acidity-mediated SOA participate indiscriminately in the process of void filling and growth of the 75 nm soot particle. In the (i) small and medium pure sulfuric acid experiments (Exp. 4, 0.02 fg and Exp. 8, 0.03 fg, respectively), 93% and 100%, respectively, of the condensed sulfuric acid are utilized for filling. (ii) large pure sulfuric acid experiment (Exp. 12, 0.26 fg), 23% of the condensed material is utilized for void filling, resulting in the filling of 25% of the voids. (iii) small and medium pure SOA (Exp. 2, 0.08 fg and Exp. 3, 0.11 fg, respectively), only 23% and 35% of the condensed material are utilized for

void filling, resulting in 12% and 26%, respectively, of filled voids. These results show that the growth of material on the soot plays a dominant role in soot transformation. We assume that the microphysical structure of the soot aggregate considered in this study contains mainly two types of voids (as outlined by DeCarlo et al. (2004)) namely, (i) internal voids, i.e., space shielded by soot primary spherules, and (ii) open voids, i.e., space open to the atmosphere (shown in Fig. 2(c)). We assume that both types of voids lie within a hypothetical sphere of mobility diameter. Furthermore, we interpret our data in light of the knowledge that the open voids will be filled over the internal voids, due to the shielding of primary spherules. This filling characterizes the microphysical transformation of the soot considered in this work. The space within the internal voids is probably bottlenecked due to the narrow opening, thereby preventing the entry of material except for large masses (5–15 times of the soot particle mass) of the condensed material (as observed in Exp.11 and Exp. 13–15). These occurrences seem to control the step-wise filling and subsequent growth process, which proceed until the voids are all filled. The indiscriminate behavior of pure sulfuric acid and pure SOA in the morphological transformation of 75 nm soot particles is attributed to the same growth and filling patterns (see Fig. 4(a)). This suggests that most of the void space is occupied by internal voids, as evidenced by the filling of only 10% of voids prior to the onset of sharp growth. The percentage of voids filled and the growth of the particle diameter are quantified (see Table S2) at each step shown in Fig. 4(a). Thus, we have quantified the state of morphological transformation (from beginning to end) as a function of the amount and type of material condensed on the soot aggregate.

We also consider 100 nm soot consisting of particles with mass of 0.33 fg. Upon condensation of the material, the major morphological transformation occurs sequentially in a step-wise fashion, i.e., filling of void space and growth of diameter, as previously described for the 75 nm soot (see Fig. 4(b)). When condensed large sulfuric acid (0.34 fg) + large SOA (1.50 fg) is applied (Exp. 31), 90% of the total volume of the condensed material is utilized for diameter growth. The remaining 10% is utilized for filling, resulting in filling of 44% of the void space. In contrast to that observed for the 75 nm soot, the response on the morphological transformation of the 100 nm fresh soot exhibits a strong dependence on the type of material condensed. Most of the volume (54%) of the large pure sulfuric acid (Exp. 28) is utilized for initial filling (up to 10 nm Δr_{me} filled 30% of the void space). For large pure SOA (Exp. 19), only 17% of void filling occurs, but the diameter growth is greater (80% of material utilized for growth and only 20% for filling) than in the case of large pure sulfuric acid. This eventually leads to two different pathways of morphological transformation of the soot particle, leading to a hysteresis of transformation (see Fig. 4(b) and Table S2 of the supplementary information). This hysteresis constitutes a response to the nature of condensing material and the type of microphysical void space (internal voids or open voids) within the soot particle. For e.g., the condensation of medium (Exp. 24) followed by large (Exp. 28) sulfuric acid on pure soot lead to a void filling of 30%, due to the higher surface tension relative to that associated with large SOA. This higher tension yields efficient filling of some (21%) of the internal voids prior to the onset of major particle growth. Similar forces and similar filling (26%) can be achieved with additional large SOA and small sulfuric acid (Exp. 23) on the soot by increasing the amount of material. The type of growth and void filling associated with the hysteresis suggest that the 100 nm soot aggregate contains mainly

internal voids, which require a considerable amount of condensable material to fill. For example, only 44% (maximum) of void space is filled when almost five times the soot core mass consisting of acidity-mediated SOA (1.84 fg) is condensed (Exp. 31). This suggests that, in the case of pure SOA, the internal void structure inhibits void-space filling (growth begins at 10% of void-space filling). However, 30% of void filling is achieved for large sulfuric acid (0.34 fg, Exp. 28).

5   Morphological transformation from the large sulfuric acid experiment (Exp. 28) to large sulfuric acid + large SOA experiment (Exp. 31), is dominated by growth of the diameter (the extent of this growth ranges from 7% to 59% relative to void filling, i.e., 30% (large sulfuric acid) to 44% (large sulfuric acid + large SOA)). Owing to the surface tension of sulfuric acid, its vapors can efficiently enter the narrow spaces of internal voids within the soot aggregate, thereby resulting in filling prior to the onset of growth. However, in the case of SOA, growth begins with complete filling of the open voids, indicating

10   that the initial diameter growth (leading to 10% of void filling) determined the open-void fraction (i.e., 10%).

In the case of 150 nm fresh soot with a mass of 0.75 fg, 76% of the total volume of the condensed material is utilized for diameter growth. In contrast, 24% is utilized for filling when a mixture of large sulfuric acid + large SOA (Exp. 46) is condensed, leading to 39% of void-space filling and 30% diameter growth. The major morphological transformation occurs

15   sequentially in a step-wise manner (see Fig. 4(c)), i.e., with sequential filling and growth of the diameter. Furthermore, the response to the nature of the material and the hysteresis of the transformation differ significantly from those of the 100 nm particle. For example, 2% growth is initially achieved in three experiments namely, medium sulfuric acid (Exp. 39), large sulfuric acid (Exp. 43), and small SOA (Exp. 33). Void filling of 3%, 25%, and 9%, respectively, are achieved for these experiments. Increasing the amount of SOA, e.g., in Exp. 39 (medium sulfuric acid) and Exp. 43 (large sulfuric acid), yields

20   a significant change in the response to morphological transformation pathways, although similar growth is achieved. In other words, additional large SOA in Exp. 39 (Exp. 42) leads to 13% growth and 29% void filling, while additional small SOA to large sulfuric acid in Exp. 43 (Exp. 44) leads to 11% growth and 42% void filling. This leads to hysteresis of the transformation, as shown in Fig. 4(c). The process of void filling and particle growth is quantified at each step shown in Fig. 4(c) (see Table S2 of the supplement). The same level of growth (i.e., 2%) occurs in these three experiments irrespective of

25   the nature of the condensed material and the percentage of the volume of the condensed material that is utilized for void filling (39%, 81%, and 70% for Exp. 39, Exp. 43, and Exp. 33, respectively). This suggests that most of the initial filling occurs for open voids. When open voids are filled completely, the filling of internal voids occurs prior to growth by high surface tension species, such as pure sulfuric acid (e.g., 56 N m^{-1}). SOA with low surface tension (e.g., 25–30 N m^{-1}) will, however, lead to growth. Compared with that required for growth of the diameter (high surface energy barriers), a smaller

30   surface area is required for the filling of voids (lower energy barriers). Sulfuric acid tends to fill even the internal voids prior to growth of the diameter, but SOA fills these voids only after growth has begun. In other words, the ability of sulfuric acid to fill some internal voids prior to growth can be attributed to its high surface tension and low vapor pressure. Owing to these characteristics, the relatively low surface energy barriers of the internal voids are easily overcome. The filling of open voids will largely be completed when SOA growth begins. This hypothesis is verified by quantifying the fraction of internal to

open voids within the boundary of the hypothetical sphere of mobility diameter. As in the case of the 75 nm and 100 nm soot aggregates, the 150 nm soot contains a smaller fraction (i.e., 10%) of open voids than internal voids. All (100%) of these voids are filled, whereas only 33% of the internal voids (constituting 90% of all voids in the soot) is filled during the heaviest coating experiment (Exp. 46). The morphological transformation process of the 150 nm soot differs (e.g., from the viewpoint of hysteresis) from that of the 75 nm and 100 nm aggregates. For e.g., during the heaviest coating experiments, 6%, 10%, and 24% of the condensed material are utilized for filling in the 75 nm, 100 nm, and 150 nm soot, respectively. However, in terms of absolute fractions, condensational growth still dominated over void filling for both sulfuric acid and SOA.

200 nm soot particles with a mass of 1.46 fg is also considered. When coated with large sulfuric acid (0.68 fg) + large SOA (2.00 fg) (Exp. 62), 42% of the total volume of condensed material is utilized for diameter growth and 58% is used for filling. This filling resulted in 41% of the void space filled. The percentage of condensed material utilized for filling increases significantly when the soot size is increased from 75 nm to 200 nm, confirming that more void space occurs in the larger size soot. This is consistent with the estimated F_{vs} result. As anticipated, the major morphological transformation results from sequential step-wise filling and growth of the 200 nm soot particle. Furthermore, the response to the nature of the condensed material and the hysteresis of the transformation are similar to those of the 150 nm soot particle. Internal re-arrangement of primary spherules within the hypothetical sphere of mobility diameter that shrinks or collapses ($Gfd < 1$) is also possible. Upon the addition of small/medium/large sulfuric acid (Exp. 51, Exp. 55, Exp. 59) and small SOA (Exp. 48) to the 200 nm soot aggregates, indiscriminate shrinkage occurs, owing to the rearrangement of primary spherules. The fraction of open voids may increase with this rearrangement and shrinkage of 4% and 2% occur for large sulfuric acid (Exp. 59) and SOA (Exp. 48). The initial fraction of open voids (10%) is filled by small SOA (Exp. 48), and additional (7%) open voids are created with subsequent condensation of the material (Exp. 48). These additional voids are filled during Exp. 56 and Exp. 57 (as indicated by negligible growth of the diameter). However, compared with SOA, sulfuric acid is more efficient at filling the internal voids when more material is condensed (Exp. 59–62) and 31% of these voids (constituting a void fraction of 83%) is filled. This filling accounts for 26% of the total void space. Detailed estimation associated with each step of filling and shrinkage of the 200 nm soot is provided in Table S2 of the supplement.

Previous studies considering the morphological transformation of soot using similar techniques are also evaluated with the framework described above (Guo et al. 2016; Qiu et al. 2012). Fig. 5 shows the Gfd as a function of Δr_{me} for the 100 nm soot transformation measured in a smog chamber by Guo et al. (2016) and Qiu et al. (2012). The major difference between those experiments and the experiments employed in this study is that a flow tube (rather than a smog chamber) is used in the present work. The smog chamber allows dynamic particle growth, i.e., particle properties including mobility diameter and mass are measured in 40-min intervals while particles are continuously growing. This may introduce some shift in the mapping of the mass and mobility size, leading to either underestimation or overestimation of Δr_{me}. Nevertheless, the

framework provides a fairly good depiction of the features associated with the morphological transformation (i.e., filling and growth) of their soot. These features are consistent with the ideal sphere growth theory. Our framework reveals that their soot is composed mainly of open voids (84–95%) as indicated by their growth curve, where the growth line is parallel to the x-axis (no growth). In their studies, growth and filling occur sequentially, but over very short time intervals, indicating that

5   internal voids are relatively easy to fill. This highlights the major differences (see Table 3) between the present study and previous studies. These results show that the framework developed in this study is quite capable of evaluating the mass and mobility data required for interpreting the morphological transformation of soot of various sizes.

3.4 Dynamic shape factor

10   The dynamic shape factor, χ, is an important parameter used to represent the shape of both fresh and coated soot particles. $\chi = 1$ and $\chi > 1$ correspond to perfectly spherical particles and irregularly shaped particles, respectively. Fig. 6(a)–(d) shows χ as a function of the coating thickness Δr_{me} for four different fresh soot sizes and the same soot cores coated with either pure sulfuric acid, pure SOA or acidity-mediated SOA. These are calculated from Eq. (10) assuming that the soot aggregate is free of internal voids. However, as previously stated, our experimental results when interpreted with the new framework

15   suggest that internal voids dominate the total void space in all four cases. Void fractions of 88%, 91%, 89%, and 83% are obtained for the 75 nm, 100 nm, 150 nm, and 200 nm soot, respectively. Therefore, the χ values obtained based on the *no internal voids* assumption (χ_n) differ significantly (see Fig. 6) from the values where we attempted to adjust for internal voids (χ_i).

20   This assumption stipulates (based on the notion of a void-less sphere) that the equivalent volume is equal to the sum of all the primary spherules. These findings have implications for atmospheric surface processes that are considered critical for modeling-based studies. As shown in Fig. 6(a)–(d), a significant amount of material is condensed on the soot particles, but a perfect sphere remains elusive. For e.g., the thickness of the coating on the 75 nm particle is at least two times larger than the initial mobility diameter, but χ_n still deviates from unity (Exp. 15). This results from the fact that χ_n is calculated (Eq. 10)

25   based on the assumption that the voids measured in our experiment are all open voids.

Our framework highlights that the high values of the dynamic shape factor found at high coating thicknesses can not only be caused by truly non-spherical particles (for example a few chains of the soot core sticking out from a spherical droplet) but in addition internal porosity in the soot core that blocks full penetration of the condensed species. Thus, internal porosity

30   causes uncertainty when judging the particle shape with conventional approaches. The main uncertainty in the new framework comes from the determination of F_{vs} and F_i. For example, when assigning a volume fraction of voids based on mobility measurements and when dividing F_{vs} into internal (F_i) and open (F_o) void space based on the growth curves. In addition,

when interpreting the data using our framework one finds that the low values of *Dfm* as well as size-dependent effective density do not always indicate soot aggregates have an open-structure, but instead compact soot cores with increasing internal porosity can also have these features and match the mass-mobility data.

5   **3.5 Effective density**

Fig. 7 (a)–(d) shows the effective density (ρ_{eff}) as a function of the coating thickness (Δr_{me}) for four different fresh soot sizes. Ideal growth curves of effective density correspond to growth of material on a perfect sphere with the same mass and mobility diameter as fresh soot grown by either pure sulfuric acid or pure SOA. The coating thickness of the ideal sphere is based on the corresponding mass-equivalent fresh soot. The method of calculating the ideal growth curve is described in 10   detail in the supplement.

As described in section 3.3, the morphological transformation (mostly the growth) responds indiscriminately to the addition of pure sulfuric acid and pure SOA to the 75 nm soot. The same response occurs for the effective density of the coated particles. For example, the pure sulfuric acid points are consistent with the ideal growth curve shown in Fig. 7(a) and most of 15   the SOA and acidity-mediated SOA points lie on the ideal growth curve of SOA. These observations support our description of the morphological transformation of the 75 nm soot (see section 3.3). Unlike those associated with the 75 nm soot, the sulfuric acid as well as the SOA and acidity mediated SOA points of the 100 nm soot lie slightly above the respective ideal growth curves of pure sulfuric acid and SOA (see Fig. 7(b)). This results from the fact that more material is utilized for void filling (than for diameter growth), leading to an increase in the particle mass without diameter growth (see section 3.3 for 20   details of the 100 nm particle). The increase in mass results in a sharp increase in ρ_{eff}. However, when particle growth occurs, ρ_{eff} increases only modestly and follows (for the most part) the ideal curve. In the case of acidity-mediated SOA, most of the points lie between the ideal growth curves of two pure substances, indicative of combined sulfuric acid and SOA growth as well as filling. The 100 nm particle transformation is still dominated by growth and, hence, points from the acidity-mediated SOA heavy coating experiment are close to ~1.1 g m^{-3} (see section 3.3), which approaches the pure SOA ideal curve. For the 25   150 nm soot, large sulfuric acid points lie above the sulfuric acid ideal growth curve, which describes the response to filling. Large sulfuric acid + small/medium/large SOA points result from filling and growth, as discussed in section 3.3. However, the corresponding ρ_{eff} values evolve in a complex manner as the materials (pure soot, pure sulfuric acid, and SOA) in the system have different material densities and mass fractions. The ρ_{eff} increases moderately in these experiments when coatings from small to large SOA are applied. Other points, representing mainly the growth-dominant transformation of the 150 nm 30   soot via SOA condensation, lie close to the SOA ideal growth curve, as shown in Fig. 7(c). In the 200 nm soot, most of the initially condensed material is utilized for filling and growth is negligible. Therefore, points from the large sulfuric acid experiment lie above the sulfuric acid ideal growth curve. Similarly, several points from the SOA coating experiment lie

above the SOA ideal growth curve. The occurrence of points above the curves is consistent with morphological transformation linked to the filling of voids. The morphological transformation occurring during the heavy acidity-mediated SOA coating experiments (Exp. 60–62) is also dominated by the filling of voids. This filling yields increased ρ_{eff} for points well above the SOA ideal growth curve and around the sulfuric acid ideal growth curve. In each case, ρ_{eff} results from the

5   microphysical transformation of the soot aggregate and changes in this value are consistent with the findings described in section 3.3.

4 Conclusion

Soot particles generated from a propane flame were aged via condensation of sulfuric acid, limonene ozonolysis products or a mixture of both, in a laminar flow tube system. A framework is developed for quantifying the microphysical

10   transformation of soot due to the condensation of material. This framework utilizes experimental data and the hypothesis of ideal sphere growth and filling of voids to quantify the distribution of condensed materials during a two-step process consisting of diameter growth and void filling. Using this framework, we quantify the percentage of material that is utilized for particle growth and void filling at each step. In the initial stages, filling is the dominant process followed by some growth, which leads to the accumulation of sufficient material. This material exerts a large surface force that facilitates

15   further filling. The process continues in several sequential steps depending on the initial morphology of the fresh soot and the nature as well as the amount of condensed material. Using the same framework, we estimate the fraction of internal voids and open voids and use this information to derive the volume equivalent diameter of a soot aggregate containing internal voids. The dynamic shape factor calculations where we attempted to account for internal voids (χ_i) is also calculated from the parameters derived by the framework. The dynamic shape factor estimated from traditional assumptions and methods (χ_n)

20   differs significantly from the value determined in this study. In fact, the dynamic shape factor adjusted for internal voids (χ_i) was close to 1 for the fresh soot particles considered in this study, indicating the particles were largely spherical. The effective density is strongly correlated with the morphological transformational responses to the condensed material on the soot particle and the resultant effective density is determined by the (i) nature of the condensed material and (ii) morphology and size of the fresh soot.

This work quantifies (*in-situ*) microphysical changes in soot morphology, providing details of both the fresh and coated soot particles. The employed framework may be useful for developing a model that determines the morphological transformation and microphysical properties (including hygroscopic and optical properties) of soot in the atmosphere.

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Table 1. Mobility diameter, mass equivalent diameter, mass, effective density, dynamic shape factor, void space fraction, primary particle size, and number of primary particles of different sizes of fresh soot particles (errors: standard deviation of all the values).

Nominal D_p (nm)	Actual D_p (nm)	D_{me} (nm)	mass (fg)	ρ_{eff} (g cm^{-3})	$\chi^{\#}$ without internal voids	χ with internal voids	F_{vs} (%)	d_{pp} (nm)	N
75	76.0±1.4	57.1±0.2	0.17±0.01	0.77±0.08	1.66±0.03	1.04	58	27.9	8
100	99.0±1.1	70.8±1.2	0.33±0.02	0.65±0.07	1.81±0.04	1.03	63	28.5	15
150	146.2±1.4	95.3±1.1	0.77±0.03	0.48±0.04	2.10±0.03	1.04	72	27.7	39
200	193.3±3.4	116.2±4.2	1.46±0.16	0.39±0.09	2.29±0.09	1.08	78	28.0	72

[#] these dynamic shape factors were calculated assuming that internal voids were absent from the soot aggregate

Table 2. Quantification of volume of condensed material utilized for filling and growth of the mobility diameter when coated with large sulfuric acid and large SOA. Results of other cases are shown in Table S2 of the supplement.

Soot core D_p (nm)	Soot Core F_{vs} (%)	Δr_{me} of coated particle (nm)	Δr_{me} for filling (nm)	Δr_{me} for growth (nm)	Volume of total coating material ($\times 10^5$ nm^3)	Volume of material used for filling ($\times 10^5$ nm^3)	Volume of material used for growth ($\times 10^5$ nm^3)	Volume of soot core void space ($\times 10^5$ nm^3)	Fraction of void space filled (%)	% of material went to filling	% of material went to growth	Fraction of internal voids (%)	Fraction of open voids (%)
75	58	52.1	9.3	42.8	21.00	1.30	19.70	1.33	97	6	94	90	10
100	63	36.9	7.3	29.6	13.90	1.40	12.50	3.20	44	10	90	91	9
150	72	35.2	12.4	22.8	19.29	4.54	14.75	11.78	38	24	76	91	9
200	78	30.5	20.5	10.0	20.92	12.13	8.79	29.50	41	58	42	83	17

10 **Table 3. Literature values reported for fraction of void space, internal voids, open voids, and void space filled at $\Delta r_{me} = 20$ nm compared with the values obtained in this study.**

	Fraction of void space, F_{vs} (%)	Fraction of internal voids, F_i (%)	Fraction of open voids, F_o (%)	Fraction of void space filled, $F_{vs,f}$ (%) at $\Delta r_{me} = 20$ nm
Qiu et al. (2012)	76	16	84	126
Guo et al. (2016)	77	5	95	169
This study	63	91	9	18
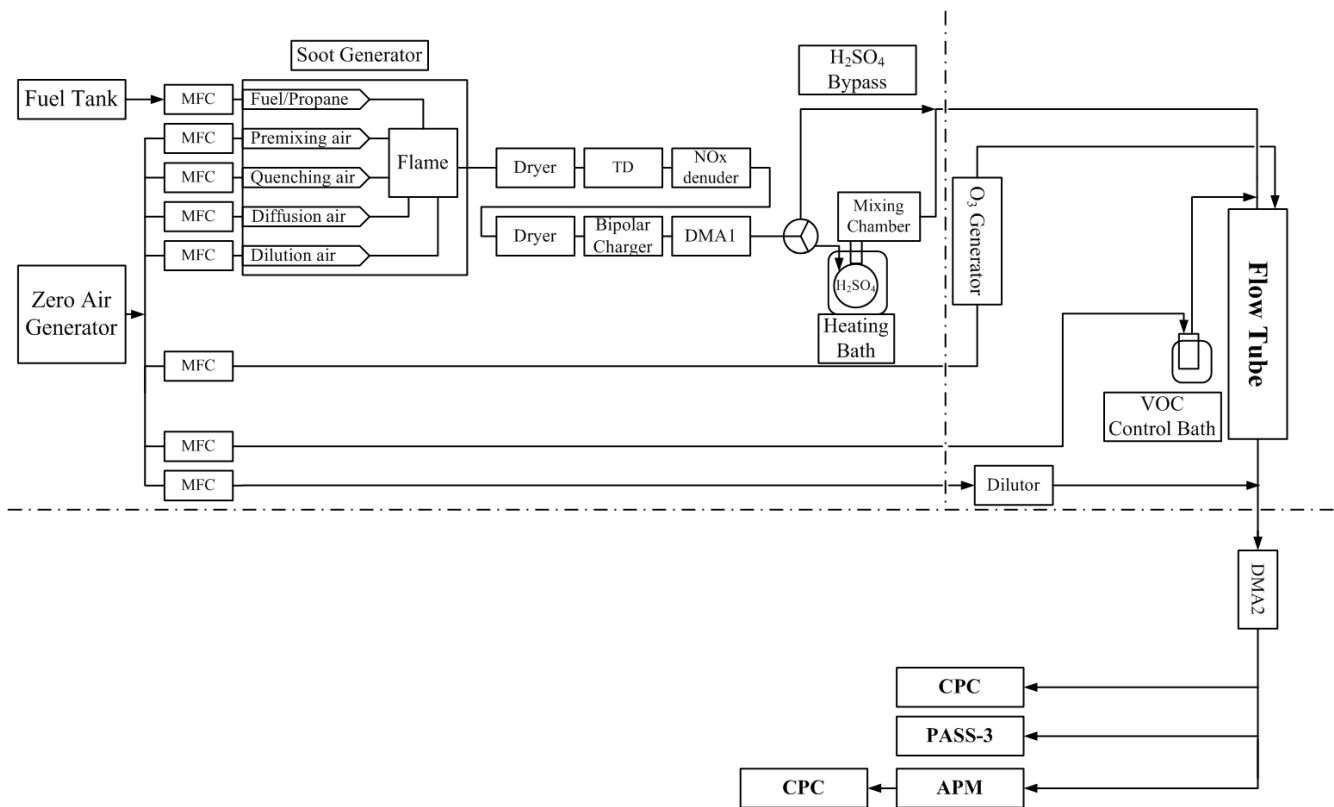
Figure 1. Schematic of the experimental setup.
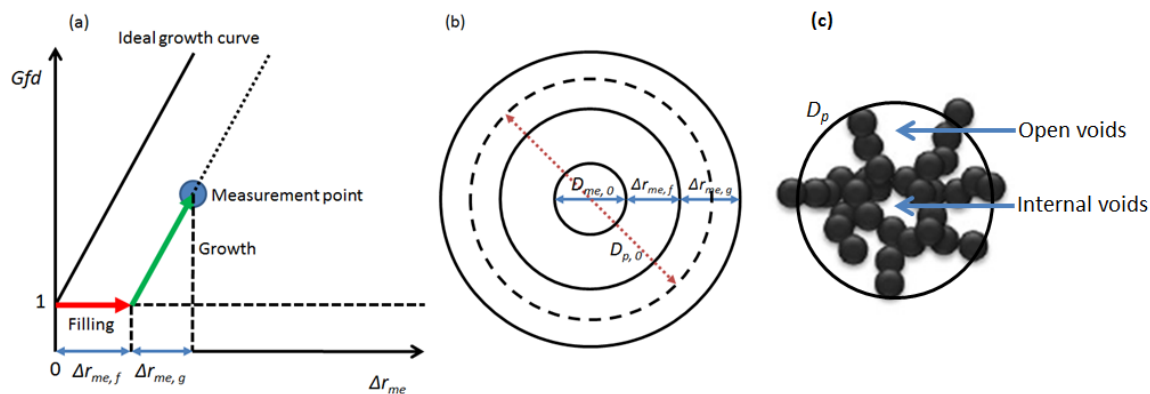
Figure 2. Illustration of the framework for quantifying the morphological transformation and structure of soot aggregate.
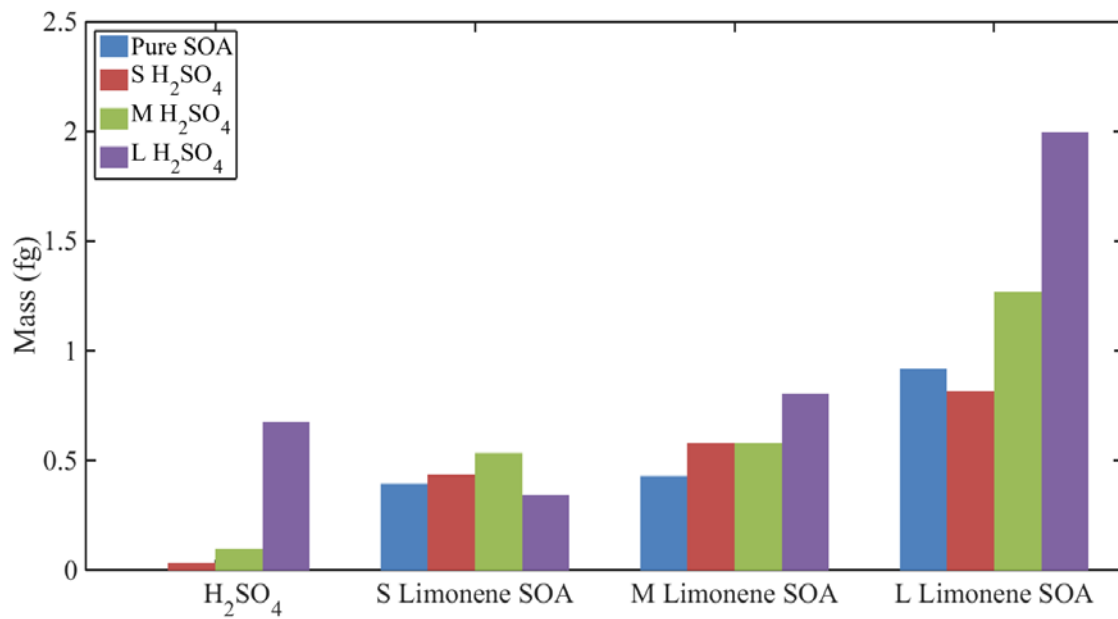
Figure 3. Mass associated with different levels of sulfuric acid and SOA coatings for the 200 nm soot core.
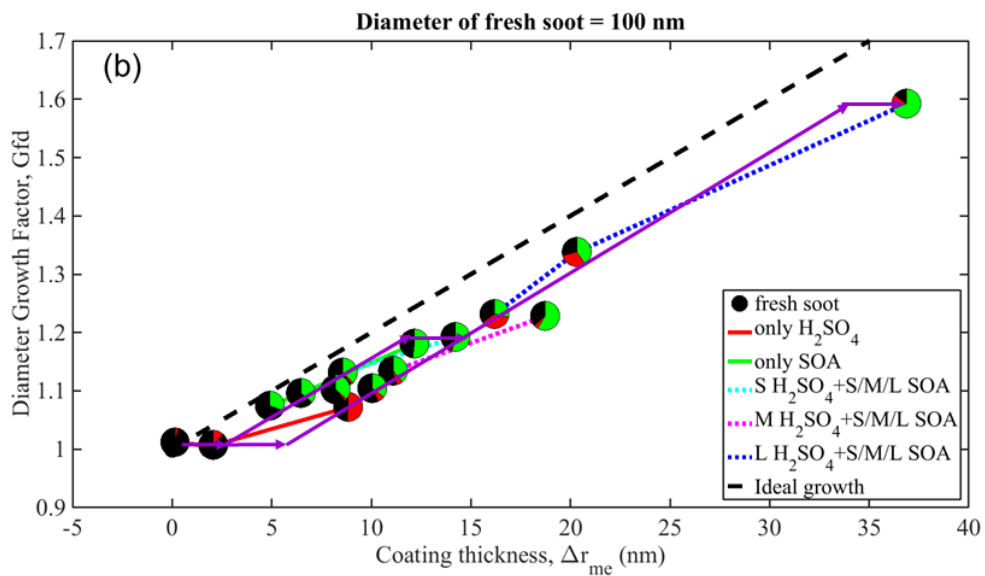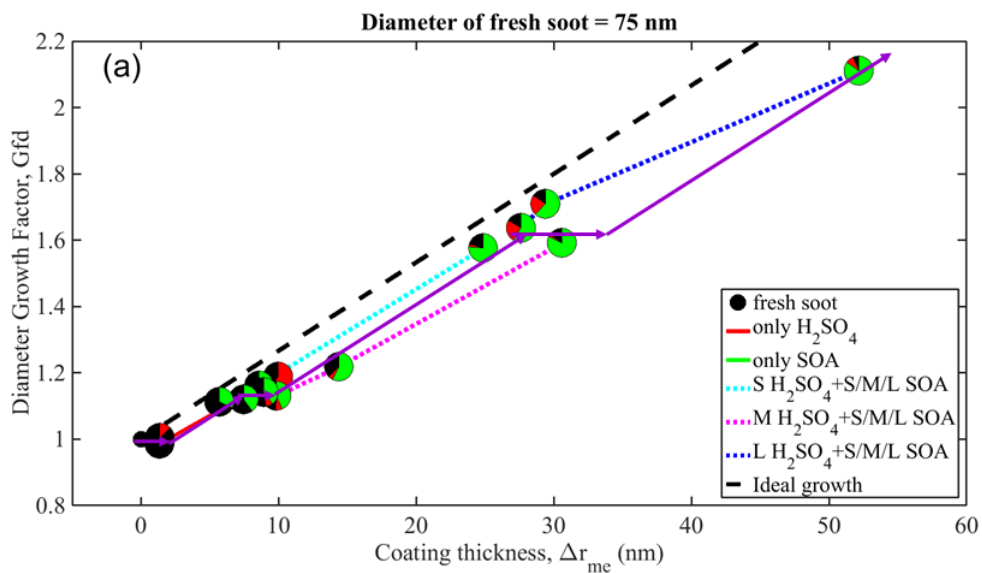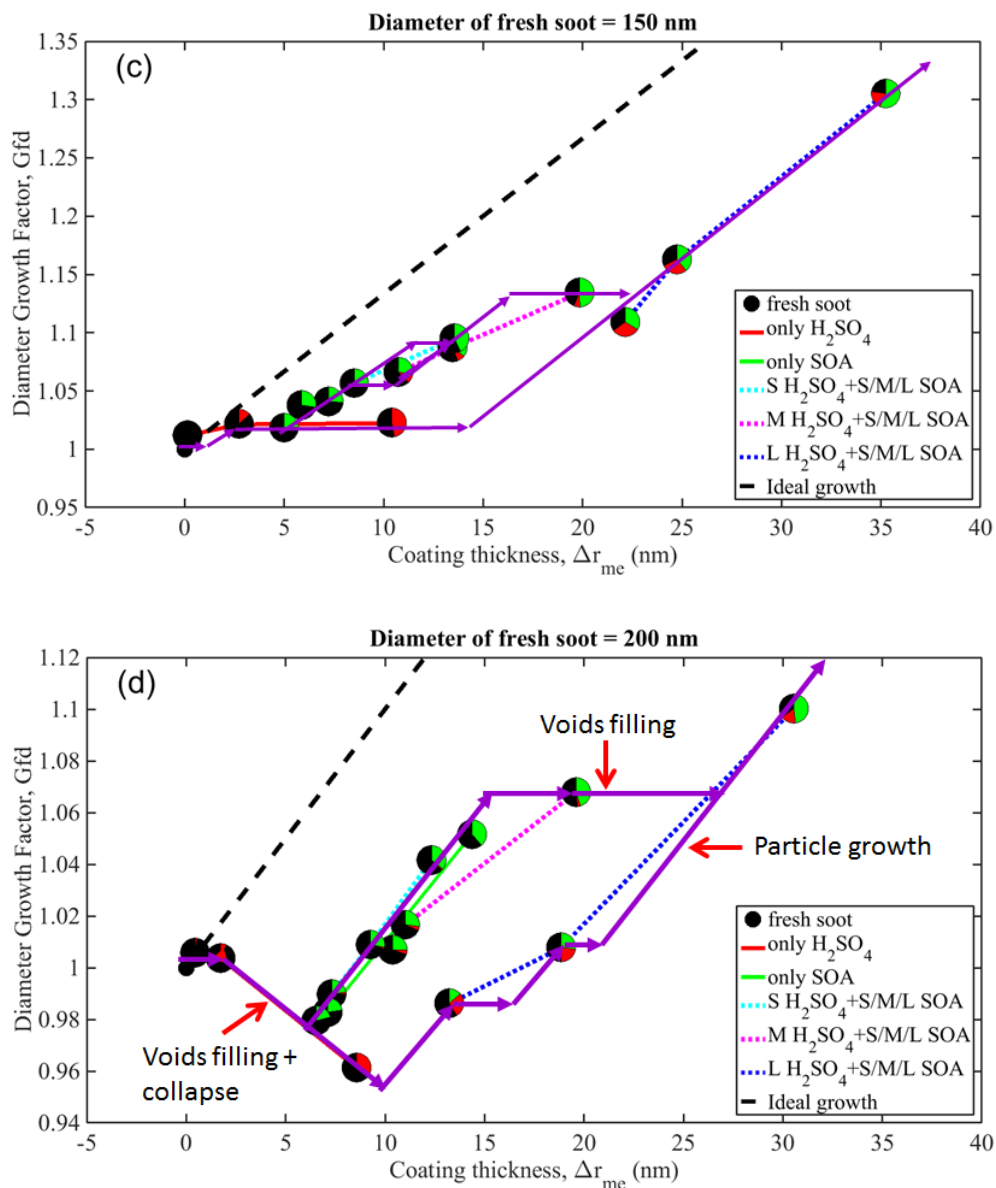
Figure 4. Mobility diameter growth factor (*Gfd*) with initial fresh soot mobility diameter of 75, 100, 150, and 200 nm as a function of the mass equivalent coating thickness (*Δr_{me}*). The black, red, and green colors in each pie chart represent the mass fraction of black carbon, sulfuric acid, and organics calculated from Eq. (4)–(6), respectively. Purple lines parallel to the ideal sphere growth line (dashed black) represent growth of the particle diameter; purple lines parallel to the x-axis represent filling of voids; purple lines with negative slope indicates a combination of voids filling and collapse of the soot particle.
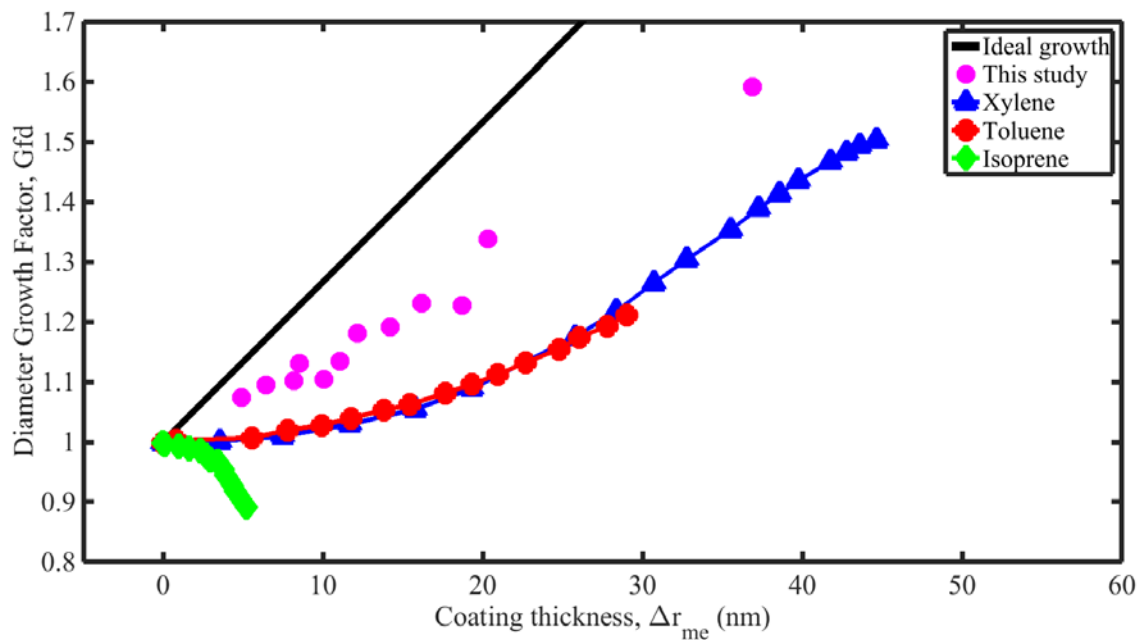
Figure 5. Mobility diameter growth factor (*Gfd*) with initial fresh soot mobility diameter of 100 nm as a function of the mass equivalent coating thickness (*Δr~me~*) reported in the literature and obtained in this study.
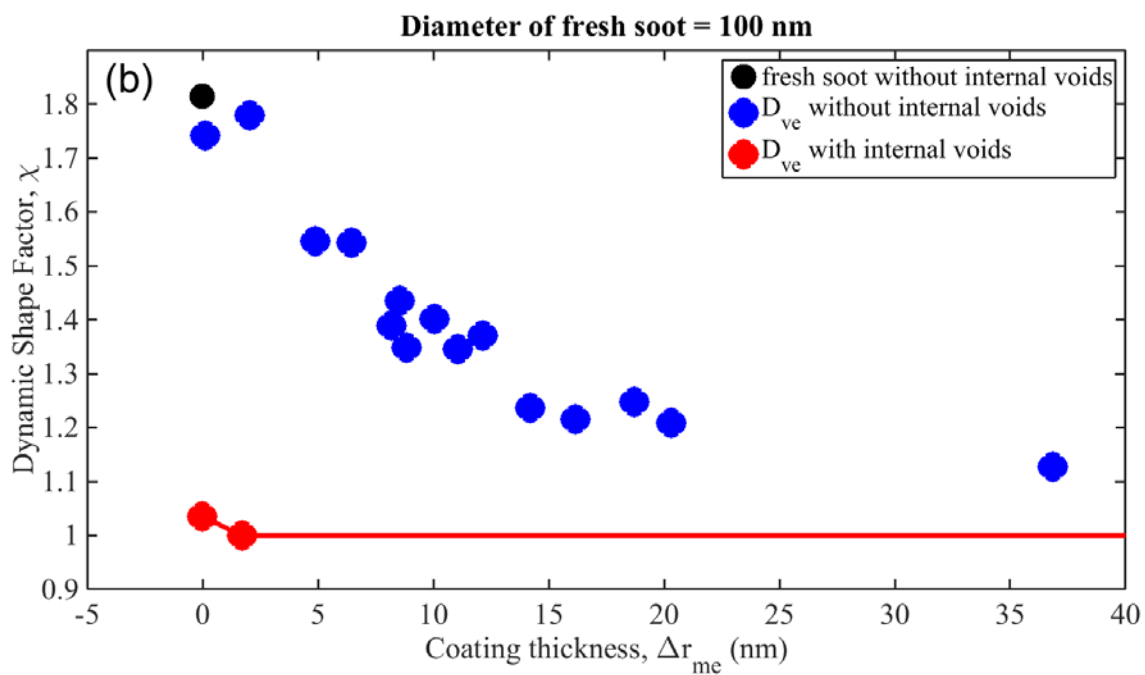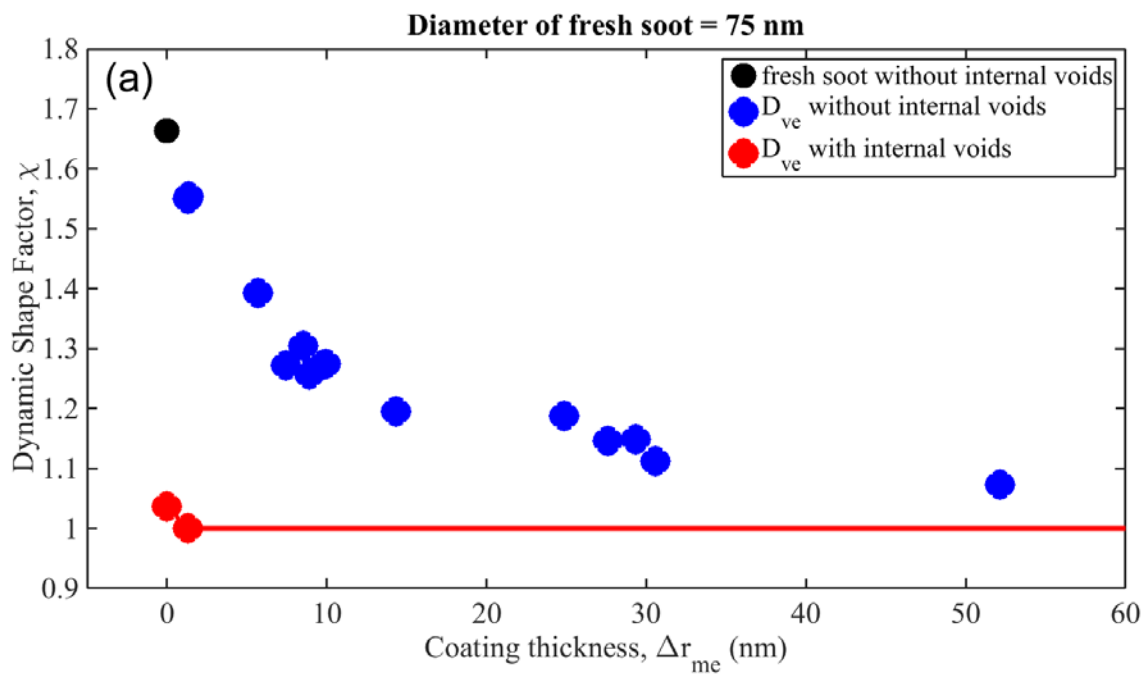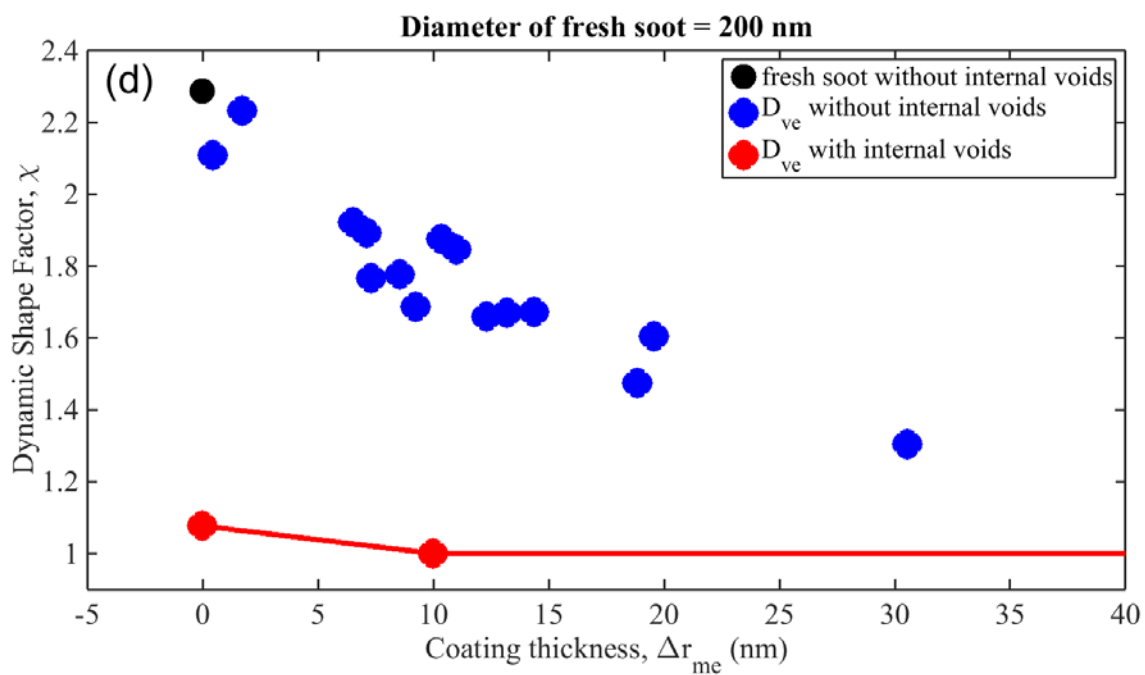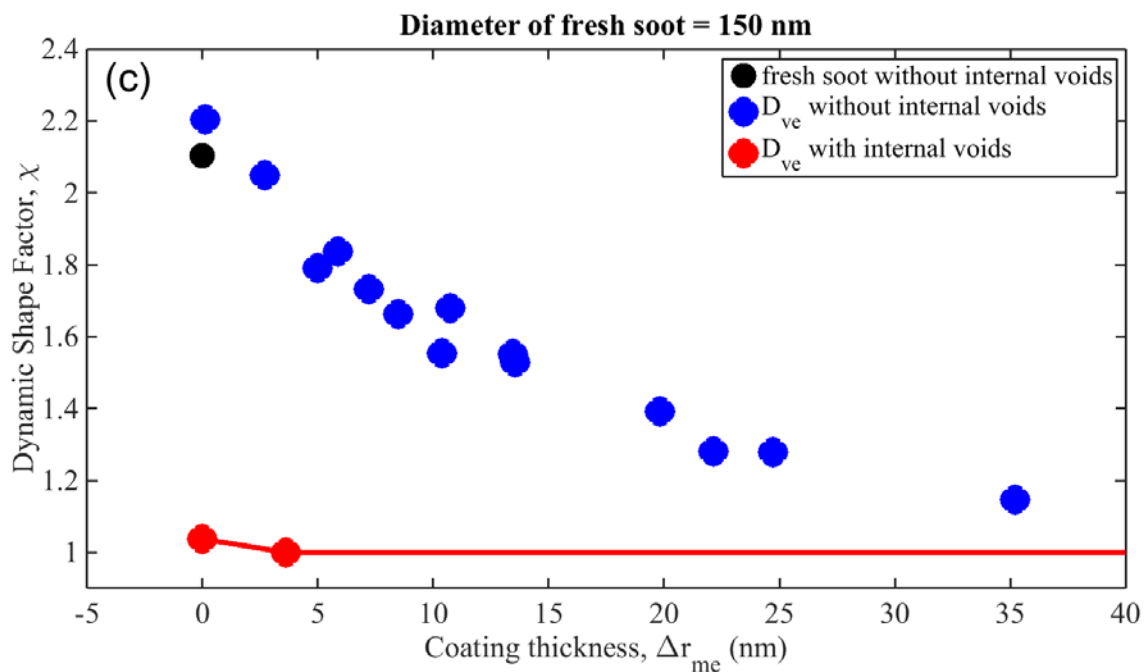
Figure 6. Dynamic shape factor (χ) of fresh and processed soot particles as a function of the mass equivalent coating thickness (Δr_{me}).
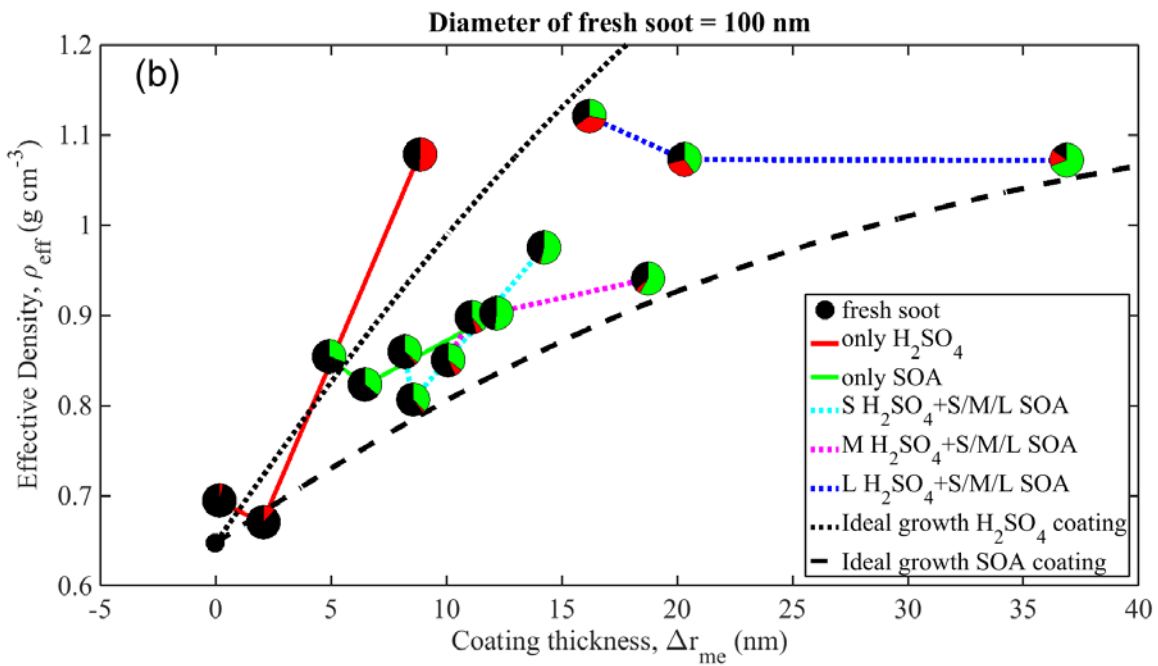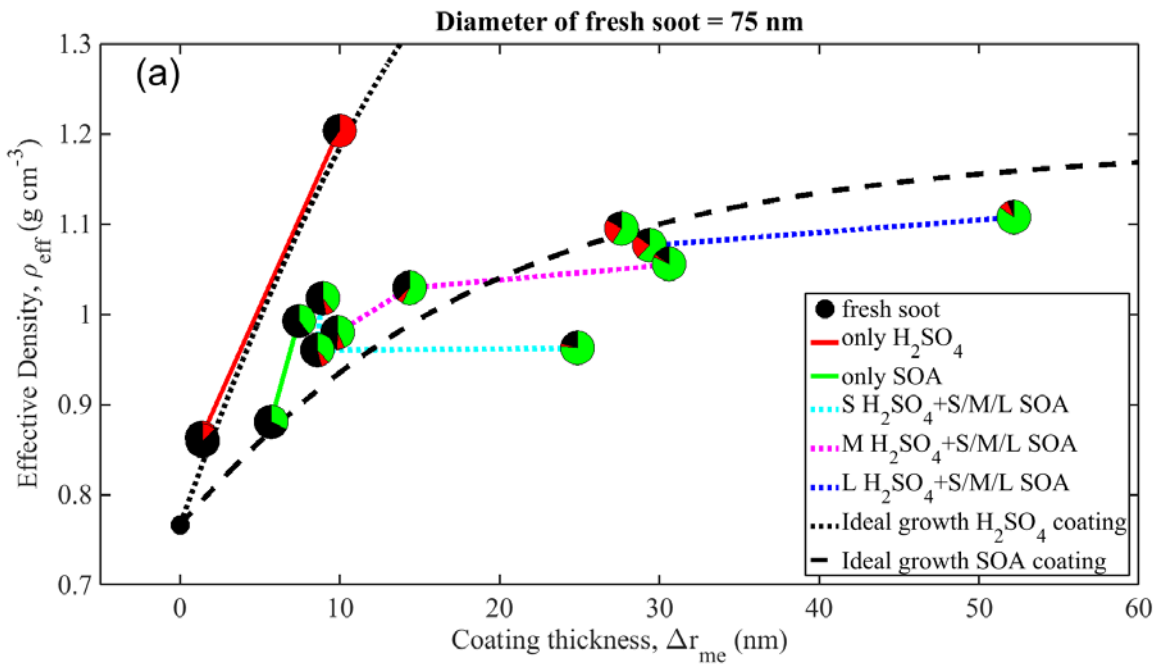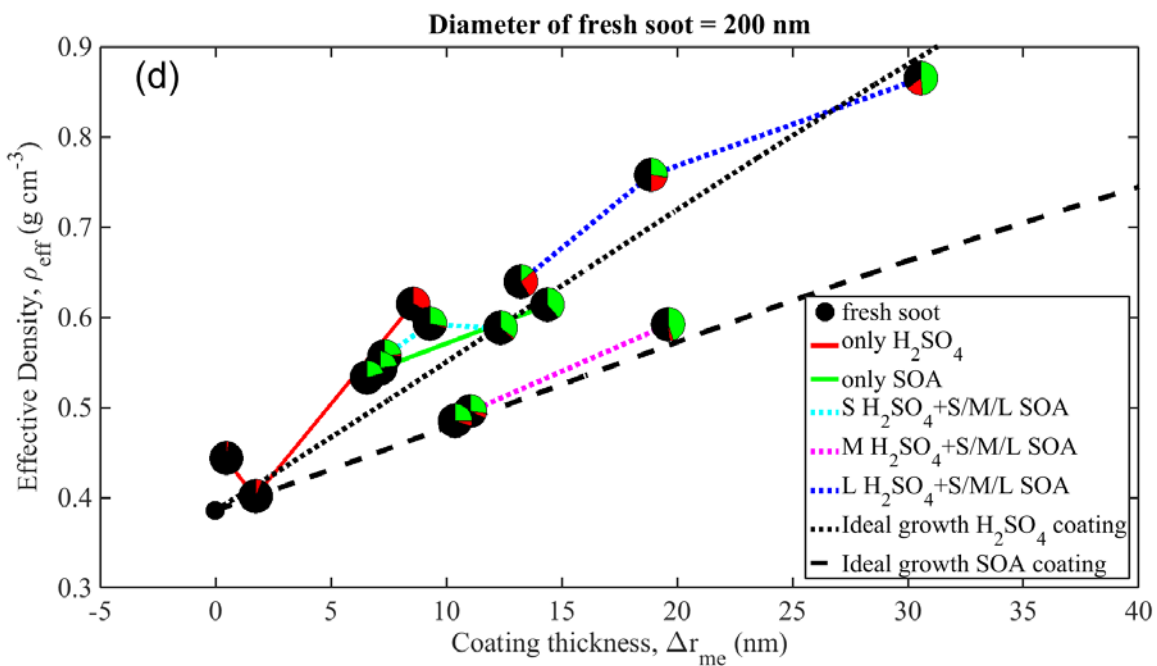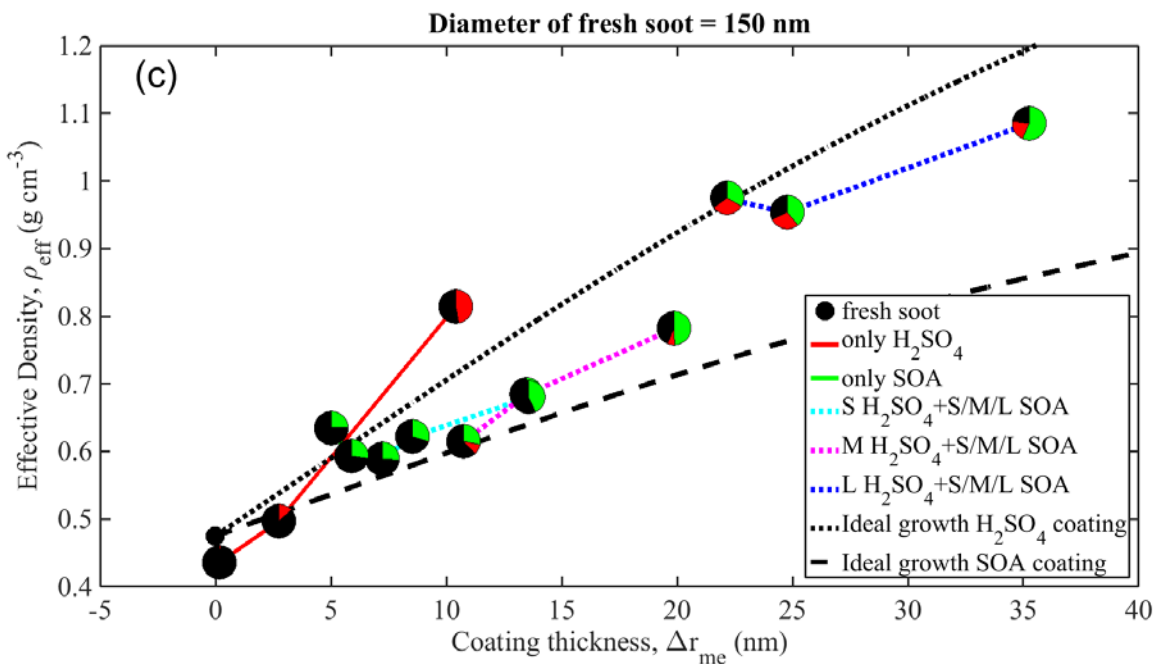
Figure 7. Effective density (ρ_{eff}) of fresh and processed soot particles as a function of the mass equivalent coating thickness (Δr_{me}).