1	Global-scale combustion sources of organic aerosols: Sensitivity to
2	formation and removal mechanisms
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13 Abstract

Organic compounds from combustion sources such as biomass burning and fossil 14 fuel use are major contributors to the global atmospheric load of aerosols. We 15 analyzed the sensitivity of model-predicted global-scale organic aerosols (OA) to 16 parameters that control primary emissions, photochemical aging and the scavenging 17 efficiency of organic vapors. We used a computationally efficient module for the 18 description of OA composition and evolution in the atmosphere (ORACLE) of the 19 global chemistry-climate model EMAC. A global dataset of aerosol mass 20 21 spectrometer measurements was used to evaluate simulated primary (POA) and secondary OA (SOA) concentrations. Model results are sensitive to the emission rates 22 of intermediate volatility organic compounds (IVOCs) and POA. Assuming enhanced 23 reactivity of semi-volatile organic compounds (SVOCs) and IVOCs with OH 24 25 substantially improved the model performance for SOA. Use of a hybrid approach for the parameterization of the aging of IVOCs had a small effect on predicted SOA 26 levels. The model performance improved by assuming that freshly emitted organic 27 28 compounds are relatively hydrophobic and become increasingly hygroscopic due to 29 oxidation.

30

31 **1 Introduction**

32 Organic aerosol (OA) is an important constituent of the atmosphere, contributing about 50% of the total submicron dry aerosol mass (Zhang et al., 2011) with major 33 34 impacts on human health and climate (IPCC, 2013; Lelieveld et al., 2015). OA 35 comprises a large number of compounds with a wide range in volatility and oxidation 36 states. The material that is in the particulate phase upon emission is called primary organic aerosol (POA). The co-emitted organic vapors can undergo one or more 37 chemical transformations, which can alter their volatility due to functionalization 38 (reducing their volatility) or fragmentation (increasing their volatility). The oxidation 39 products with lower volatility can be transferred to the particulate phase forming 40 secondary organic aerosol (SOA). 41

42 Several regional-scale modeling studies have accounted for the semi-volatile 43 nature and chemical aging of organic compounds by using the volatility based set 44 (VBS) approach (Donahue et al., 2006), demonstrating improvements in the accuracy 45 of the predicted concentrations of organic aerosols (OA) and their chemical properties 46 (Robinson et al., 2007; Shrivastava et al., 2008; Murphy and Pandis, 2009; Hodzic et al., 2010; Tsimpidi et al., 2010; Fountoukis et al., 2011; Li et al., 2011; Tsimpidi et 47 al., 2011; Bergstrom et al., 2012; Athanasopoulou et al., 2013; Zhang et al., 2013; 48 Fountoukis et al., 2014). However, only few global modeling studies have adopted the 49 VBS approach (Pye and Seinfeld, 2010; Jathar et al., 2011; Jo et al., 2013; Tsimpidi et 50 al., 2014). According to these studies, the modeled global tropospheric burden of 51 POA is 0.03-0.23 Tg and of SOA 1.61-2.77 Tg, with SVOCs and IVOCs contributing 52 0.71-1.57 Tg to the total. 53

54 The VBS approach is a flexible framework for simulating OA formation and removal; however, there are several uncertainties in the parameters used. The first 55 source of uncertainty is related to the emissions of organic particles and vapors 56 (Kanakidou et al., 2005). The volatility distribution of the fresh POA is important in 57 the VBS as it determines the initial evaporation of POA. Part of the IVOC emissions 58 is not included in conventional inventories, even if it is important for the predicted 59 SOA (Shrivastava et al., 2008; Grieshop et al., 2009; Tsimpidi et al., 2010). Several 60 studies have assumed a 50% addition to the traditional emission inventory (e.g., 61 Shrivastava et al., 2008; Jathar et al., 2011; Tsimpidi et al., 2014) for IVOC emissions 62 63 but enhancements up to a factor 6.5 have been used in the literature (e.g., Shrivastava et al., 2011). Furthermore, most previous modeling studies typically assumed the 64 65 same volatility distributions of all emissions independent of their source (e.g. Robinson et al., 2007). However, recent investigations reported significant differences 66 67 in the volatility distribution of particles emitted from biomass burning, diesel and gasoline vehicle exhausts (May et al., 2013a; May et al., 2013c, b). 68

69 The second source of uncertainty is related to the oxidation of the emitted SVOCs and IVOCs. The parameters used by the VBS to simulate this process are the 70 71 oxidation rate constant, the volatility distribution of the products, and the oxygen mass added per generation of oxidation. The VBS volatility resolution used to 72 represent the SVOC/IVOC volatility range $(3.2 \times 10^{-1} \ \mu g \ m^{-3} < C^* < 3.2 \times 10^6 \ \mu g \ m^{-3})$ 73 affects these parameters as well. A coarse volatility resolution requires a lower 74 effective oxidation rate constant and a more rapid addition of oxygen and reduction in 75 volatility than a finer volatility resolution. A common representation for the oxidation 76 77 of SVOCs and IVOCs, mainly used by regional models (e.g. Murphy and Pandis, 2009; Tsimpidi et al., 2010; Fountoukis et al., 2011; Tsimpidi et al., 2011; Bergstrom 78

et al., 2012; Athanasopoulou et al., 2013; Fountoukis et al., 2014), is based on the 79 work of Robinson et al. (2007) and Shrivastava et al. (2008) and includes 9 volatility 80 bins with saturation concentrations ranging from 10^{-2} to 10^{6} µg m⁻³, an oxidation rate 81 constant of 4×10^{-11} cm³ molec⁻¹ s⁻¹ based on Atkinson and Arey (2003), a reduction in 82 volatility by one order of magnitude after each reaction, and a 7.5% net increase in 83 mass to account for the added oxygen. This formulation is rather conservative 84 compared to other studies which have assumed higher reduction in volatility and/or 85 increase in mass. Shrivastava et al. (2011) assumed a 15% increase in mass due to the 86 87 added oxygen, while Grieshop et al. (2009) and Hodzic et al. (2010) assumed a 40% increase in mass and two orders of magnitude reduction in volatility in each reaction 88 step. Pye and Seinfeld (2010) simulated the POA emissions using two SVOCs (with 89 C^* equal to 20 and 1646 µg m⁻³) and one IVOC (10⁵ µg m⁻³) and used an oxidation 90 rate constant of 2×10^{-11} cm³ molec⁻¹ s⁻¹, two orders of magnitude reduction in 91 volatility in each reaction, and 50% increase in mass per reaction. Shrivastava et al. 92 (2011) used only two surrogate species (C^* equal to 10^{-2} and $10^5 \,\mu g \,m^{-3}$), an oxidation 93 rate constant of 0.57×10^{-11} cm³ molec⁻¹ s⁻¹, seven orders of magnitude reduction in 94 volatility, and 50% increase in mass per reaction. Tsimpidi et al. (2014) used a lower 95 resolution VBS scheme with 4 surrogate species (with $C^* 10^{-1}$, 10^1 , 10^3 , and $10^5 \mu g$ 96 m⁻³), an oxidation rate constant of 2×10^{-11} cm³ molec⁻¹ s⁻¹, two orders of magnitude 97 reduction in volatility, and 15% increase in mass per reaction. All of the above 98 schemes should be viewed as parameterizations of the complex reactions that actually 99 100 take place; the oxidation products can be up to four orders of magnitude lower in volatility than the precursor (Kroll and Seinfeld, 2008). To address this limitation, 101 102 Jathar et al. (2012) developed a hybrid method to represent the formation of SOA from non-speciated SVOC and IVOC vapors. According to this framework, the first 103 104 generation of oxidation of SVOC and IVOC is parameterized by fitting to SOA data from smog chamber experiments. Subsequently, the generic multi-generational 105 oxidation scheme of Robinson et al. (2007) was used for the subsequent generation 106 steps. 107

108 The third source of uncertainty is related to the scavenging efficiency of gas-phase 109 oxidized SVOCs and IVOCs. The water solubility of these organic vapors is largely 110 unknown and in most OA modeling studies a fixed effective Henry's law constant 111 (e.g., $H = 10^5$ M atm⁻¹) is used for all organic compounds. However, organic vapors become increasingly more hydrophilic during their atmospheric lifetime. Pye and Seinfeld (2010) treated the freshly emitted gas-phase SVOCs as relatively hydrophobic (H= 9.5 M atm⁻¹) and their oxidation products as moderately hydrophilic (H= 10⁵ M atm⁻¹). Hodzic et al. (2014) argued that Henry's law constants have a strong negative correlation with the saturation vapor pressures and depend on the precursor species, the extent of photochemical processing, and the NO_x levels during the formation.

In this work we use ORACLE, a computationally efficient module for the 119 120 description of OA composition and evolution in the atmosphere (Tsimpidi et al., 2014), to quantify the impact of the main VBS parameters on the model OA 121 predictions. Our main focus is the formation of OA from anthropogenic combustion 122 and open biomass burning sources. We conducted different tests to study the 123 sensitivity of the model predictions to emissions, photochemical aging and scavenging 124 efficiency of LVOCs, SVOCs and IVOCs. The results are compared to the reference 125 simulation and aerosol mass spectrometer (AMS) measurements at multiple locations 126 worldwide following Tsimpidi et al. (2016a). Results from these sensitivity tests help 127 identify the major uncertainties of the VBS formulations and give rise to suggestions 128 129 about potential model improvements.

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131 **2** Reference model description and application

132 **2.1 EMAC Model**

133 The ECHAM/MESSy Atmospheric Chemistry (EMAC) model is a numerical chemistry and climate simulation system that includes sub-models describing lower 134 135 and middle atmosphere processes and their interaction with oceans, land and human influences (Jöckel et al., 2006). EMAC includes submodels that describe gas-phase 136 chemistry (MECCA; Sander et al., 2011), inorganic aerosol microphysics (GMXe; 137 Pringle et al., 2010), cloud microphysics (CLOUD; Jöckel et al., 2006), aerosol 138 optical properties (AEROPT; Lauer et al., 2007), dry deposition and sedimentation 139 (DRYDEP and SEDI; Kerkweg et al., 2006a), cloud scavenging (SCAV; Tost et al., 140 2006), emissions (ONLEM and OFFLEM; Kerkweg et al., 2006b), and organic 141 aerosol formation and growth (ORACLE; Tsimpidi et al., 2014). EMAC model has 142 been extensively described and evaluated against in situ observations and satellite 143 retrievals (Pozzer et al., 2012; Karydis et al., 2016a; Karydis et al., 2016b; Tsimpidi et 144

al., 2016b). The spectral resolution used in this study is T63L31, corresponding to a
horizontal grid spacing of 1.875°x1.875° and 31 vertical layers extending to 25 km
altitude. The thickness of the first vertical layer is 68 m. The 11-year period between
2000 and 2010 is simulated, with the first year used as spin-up.

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150 **2.2 ORACLE Module**

ORACLE is a computationally efficient submodel for the description of organic 151 aerosol composition and evolution in the atmosphere (Tsimpidi et al., 2014). ORACLE 152 153 simulates a wide variety of semi-volatile organic products separating them into bins of logarithmically spaced effective saturation concentrations. In this study, primary 154 organic emissions from biomass burning and fuel combustion sources are taken into 155 account using separate surrogate species for each source category. These surrogates are 156 subdivided into three groups of organic compounds: LVOCs ($C^*=10^{-2} \mu g m^{-3}$), SVOCs 157 $(C^*=10^0 \text{ and } 10^2 \ \mu\text{g m}^{-3})$ and IVOCs $(C^*=10^4 \text{ and } 10^6 \ \mu\text{g m}^{-3})$. These organic 158 159 compounds are allowed to partition between the gas and aerosol phases resulting in the formation of POA. Anthropogenic and biogenic VOCs are simulated separately, and 160 their oxidation results in products distributed in four volatility bins with effective 161 saturation concentrations 10⁰, 10¹, 10², and 10³ µg m⁻³. Gas-phase photochemical 162 reactions that modify the volatility of the organics are taken into account and the 163 164 oxidation products (SOA-sv, SOA-iv, and SOA-v) of each group of precursors (SVOCs, IVOCs, and VOCs) are simulated separately in the module to keep track of 165 166 their origin. We have assumed that functionalization and fragmentation processes result in a net average decrease of volatility for SOA produced by SVOC/IVOC and 167 anthropogenic VOC, without a net average change of volatility for SOA produced by 168 biogenic VOC (Murphy et al., 2012). The volatilities of SVOCs and IVOCs are 169 reduced by a factor of 10^2 as a result of the OH reaction with a rate constant of 2×10^{-11} 170 cm³ molecule⁻¹ s⁻¹ and a 15% increase in mass to account for two added oxygen atoms 171 (Tsimpidi et al., 2014). LVOCs are not allowed to participate in photochemical 172 reactions since they are already in the lowest volatility bin. In total 52 organic 173 compounds are simulated explicitly (26 in each of the gas and aerosol phases). The 174 model set-up and the different aerosol types and chemical processes that simulated by 175 ORACLE in this study are illustrated in Figure 1a. More details about ORACLE can 176 be found in Tsimpidi et al. (2014). 177

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179 **2.3 Emission inventory**

The CMIP5 RCP4.5 emission inventory (Clarke et al., 2007) is used for the 180 anthropogenic primary organic aerosol emissions from fuel combustion and biomass 181 burning. The open biomass burning emissions from savanna and forest fires are based 182 on the Global Fire Emissions Database (GFED v3.1; van der Werf et al., 2010). In 183 order to convert the emitted organic carbon (OC) to organic mass (OM), OM/OC 184 factors of 1.3 and 1.6 have been used for the anthropogenic and biomass burning 185 186 emissions, respectively (Aiken et al., 2008; Canagaratna et al., 2015). Furthermore, emission fractions are used to distribute the OM to the volatility bins used by 187 ORACLE. The sum of the emission fractions used for the volatility bins with $C^* \leq$ 188 10^4 is unity since current emission inventories are based on samples collected at 189 aerosol concentrations up to $10^4 \ \mu g \ m^{-3}$ (Shrivastava et al., 2008; Robinson et al., 190 2010). Additional emission fractions can be assigned to the volatility bins with $C^* > C^*$ 191 10^4 based on dilution experiments (Robinson et al., 2007). 192

In this study we assume that anthropogenic fuel (fossil and biofuel) combustion 193 emissions cover a range of volatilities from 10^{-2} to 10^{6} µg m⁻³ and the additional 194 195 IVOC emissions are 1.5 times the traditional POA emissions (Robinson et al., 2007); therefore, the sum of the emission fractions for the fuel combustion emissions is 2.5 196 (Figure 2a). Biomass burning emissions are assumed to cover a range of volatilities 197 from 10^{-2} to 10^{4} (May et al., 2013a) and no additional IVOC emissions are assumed 198 from biomass burning sources. Therefore, the sum of their emission factors is unity 199 (Figure 2a). Overall, the decadal average global emission flux of primary organic 200 emissions is 44 Tg yr⁻¹ from anthropogenic combustion sources and 28 Tg yr⁻¹ from 201 202 open biomass burning sources.

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204 3 Sensitivity Simulations

All sensitivity calculations are conducted for the same 11-year period as the reference simulation, the results of which have been analyzed by (Tsimpidi et al., 207 2016a). Table 1 summarizes the general characteristics of the sensitivity simulations. A detailed description is provided below.

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210 **3.1 Sensitivities to emissions**

The emissions of LVOCs, SVOCs and IVOCs are a key input for the accurate description of atmospheric OA. To quantify the sensitivity of the reference case results to the LVOC, SVOC and IVOC emissions, three simulation tests have been designed. Figure 2 summarizes the emission factors used for the volatility distribution of the emissions and the emission rate of each volatility bin for the reference and the sensitivity tests. More specifically:

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Low volatility: In this sensitivity simulation, we assume zero emissions of IVOCs to quantify their contribution to the formation of global SOA. Therefore, the fuel combustion and biomass burning emissions are distributed only in the LVOCs (10^{-2} µg m⁻³) and SVOCs (10^{0} and 10^{2} µg m⁻³) volatility bins and the sum of their emission fractions is equal to unity (Figure 2b). The decadal average global emission flux of primary organic emissions in this test is 18 Tg yr⁻¹ from anthropogenic combustion sources and 28 Tg yr⁻¹ from open biomass burning sources (Table 1).

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High IVOCs: To estimate an upper limit of the IVOC contribution to the formation 226 of SOA, a sensitivity simulation is conducted in which the emissions of IVOCs are 227 228 increased by 1.5 times the original POA emissions. These emissions are distributed in the volatility bins with C^* of 10^4 and $10^6 \mu g m^{-3}$ (Figure 2c) by applying an additional 229 emission factor of 0.5 and 1 respectively. The LVOC and SVOC emissions are the 230 same as in the reference simulation. Overall, the total anthropogenic and biomass 231 232 burning emissions are 4 and 2.5 times higher respectively than the original POA emission inventory. The decadal average global emission flux of primary organic 233 emissions in this sensitivity test is 71 Tg yr⁻¹ for both anthropogenic and open 234 biomass burning sources (Table 1). 235

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Alternative POA emissions: To investigate the sensitivity of the model results to the magnitude of the POA emissions, we have utilized the AEROCOM database for the POA emissions from anthropogenic combustion sources (Dentener et al., 2006) and the CMIP5 RCP4.5 emission inventory for the POA emissions from open biomass burning sources. These emission inventories include 36% lower POA emissions from anthropogenic combustion sources and 33% higher POA emissions from open biomass burning sources on average over the decade 2000-2010 compared to the

reference simulation. The assumed volatility distributions are the same as in the reference simulation. The decadal average global emission flux of primary organic emissions in this case is 29 Tg yr⁻¹ from anthropogenic combustion sources and 38 Tg yr⁻¹ from open biomass burning sources (Table 1).

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249 **3.2 Sensitivity to chemistry**

The photooxidation of SVOCs and IVOCs emitted from fuel combustion and biomass-burning sources can lead to the formation of substantial SOA mass on a global scale (Jathar et al., 2011; Tsimpidi et al., 2014). To evaluate the sensitivity of the model to the parameters used to describe the aging process we have conducted three sensitivity simulations described below.

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High reaction rate constant: In this simulation we investigate the sensitivity of the results to the rate constant used for the gas-phase photooxidation of SVOCs and IVOCs with OH. We assume that the corresponding oxidation rate constant is twice that of the reference simulation and equal to 4×10^{-11} cm³ molecules⁻¹ s⁻¹. All other parameters remained the same as in the reference simulation (Table 1).

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Alternative aging scheme: To quantify the sensitivity of the results to the aging 262 263 scheme, we designed a sensitivity case in which the aging scheme of Robinson et al. (2007) is used (Figure 1b). Based on this implementation, we are using nine volatility 264 265 bins (compared to 5 in the reference simulation) to distribute the primary emissions into LVOCs (10^{-2} and $10^{-1} \mu g m^{-3}$), SVOCs (10^{0} , 10^{1} , and $10^{2} \mu g m^{-3}$), and IVOCs 266 $(10^3, 10^4, 10^5, \text{ and } 10^6 \text{ }\mu\text{g m}^{-3})$. This model set up is based on the formulation 267 proposed by (Shrivastava et al., 2008). The volatility distribution of anthropogenic 268 269 combustion and open biomass burning emissions is shown in Figure 2d. The sum of these emission factors is the same as in the reference simulation (2.5 for fuel 270 combustion and 1 for biomass burning). However, the relative importance of SVOC 271 and IVOC to total OA emissions is changed compared to the reference simulation. In 272 the sensitivity simulation the fraction of SVOCs to the total emissions is 20% for fOA 273 and 60% for bbOA (Figure 2d), compared to 32% and 70%, respectively, in the 274 reference simulation (Figure 2a). Furthermore, the saturation concentration of the 275 organic vapors reacting with OH is reduced by a factor of 10 (instead of 100 in the 276

reference simulation) with a rate constant of $4x10^{-11}$ cm³ molec⁻¹ (double the value 277 used in the reference simulation) and a 7.5% increase in mass to account for one 278 added oxygen (half the value used in the reference simulation). The formation of SOA 279 from LVOCs is possible in this configuration (contrary to the reference simulation) 280 due to the presence of two species in the LVOC volatility range ($C^* < 3.2 \times 10^{-1} \mu g$ 281 m⁻³). Overall, in this simulation, 46 surrogate organic aerosol species are used to track 282 the source- and volatility-resolved OA components compared to 26 aerosol species in 283 284 the reference simulation.

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Hybrid aging scheme: The reference and alternative aging scheme simulations 286 assume that the volatility of the organic vapor precursors is reduced by two and one 287 orders of magnitude, respectively, after each oxidation step. However, photooxidation 288 reactions of IVOCs can create products with a volatility 1 to 4 orders of magnitude 289 lower (Kroll and Seinfeld, 2008). Furthermore, recent experiments indicate that the 290 reduction in volatility due to oxidation reactions changes as the organic molecules 291 become more oxygenated and fragmentation becomes important (Chacon-Madrid et 292 293 al., 2013). To investigate the effect of these assumptions on the predicted global SOA 294 burden, we have modified the OA chemistry mechanism to include a hybrid method to calculate the SOA formation from the oxidation of IVOCs based on the approach of 295 296 (Jathar et al., 2012). The SVOC oxidation scheme remains the same as in the reference. The hybrid scheme distributes the IVOC first generation oxidation products 297 298 over a range of volatilities, with larger reductions in volatility compared to the reference simulation. The oxidation of each IVOC is assumed to result in the 299 300 formation of two condensable organic gases with four and six orders of magnitude lower volatility and aerosol yields equal to 0.71 and 0.115, respectively (Jathar et al., 301 302 2014) (Figure 1c). Then, the reference oxidation scheme is used for subsequent oxidation of these products assuming a factor of 100 reduction in volatility with 15% 303 increase in mass. The photo-oxidation of SVOCs and IVOCs in the hybrid aging 304 scheme is described by the following reactions: 305

306

 $307 \qquad SVOC_i + OH \rightarrow 1.15 SOG-sv_{i-1} \qquad (R1)$

 $308 \qquad SOG-sv_i + OH \rightarrow 1.15 SOG-sv_{i-1} \qquad (R2)$

 $309 \qquad \qquad SOG-sv_i \leftrightarrow SOA-sv_i \qquad \qquad (R3)$

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311	$IVOC_i + OH \rightarrow 0.71 \text{ SOG-i}v_{i-2} + 0.115 \text{ SOG-i}v_{i-3}$ (R4)	
312	$SOG-iv_i + OH \rightarrow 1.15 SOG-iv_{i-1}$ (R5)	
313	$SOG-iv_i \leftrightarrow SOA-iv_i$ (R6)	
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315	This representation is more consistent with SOA formation from VOCs and prov	vides
316	in principle at least a more realistic representation of SOA formation from IVOCs.	
317		
318	3.3 Sensitivities to scavenging	
319	The wet and dry removal the organic vapours from the atmosphere depend	s on
320	their ability to partition into water which is commonly expressed by their Henry's	law
321	constant (H). Two sensitivity simulations where performed to investigate the effective effect	ct of
322	this uncertain parameter.	
323		
324	Low solubility: To test the sensitivity of the results to the solubility of the SVOC	and
325	IVOC vapors, we have conducted a simulation using a Henry's law constant	
326	orders of magnitude lower than the reference and equal to 10^3 M atm ⁻¹ for	both
327	primary and secondary SVOCs/IVOCs.	
328		
329	Variable solubility: The photochemical aging of organic vapors results on average	ge in
330	less volatile and more hydrophilic products (Jimenez et al., 2009). To quantify	the
331	effect of this change on the model results we have conducted a sensitivity simula	
332	in which the fresh SVOCs and IVOCs are hydrophobic with $H = 10$ M atm ⁻¹	and

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atm⁻¹.

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336 4 Reference simulation results and evaluation

The predicted decadal average surface concentrations of total OA, POA, SOA-sv, and SOA-iv for the reference simulation are shown in Figure 3. High POA concentrations are predicted over regions affected by biomass burning (i.e., the tropical and boreal forests) as well as over the industrialized regions of the Northern Hemisphere where strong fossil and biofuel combustion sources are located (i.e., Eastern and Southern Asia, Central and Eastern Europe, Western and Eastern US).

become more hydrophilic after their photochemical oxidation with an $H = 10^5$ M

343 Further downwind of the sources, the POA concentration decreases substantially due to dilution and evaporation (Figure 3b). On the other hand, the predicted SOA-sv and 344 SOA-iv concentrations are high over a wide area downwind of the polluted urban 345 areas and the major rainforests (Figure 3c and 3dc) due to the transport of IVOCs and 346 SVOCs and their continued chemical transformations. Since IVOC emissions from 347 anthropogenic sources are assumed to be two times higher than SVOC emissions 348 (Figure 1a), predicted SOA-iv is higher than SOA-sv over populated areas (Figure 3c 349 and 3d). On the other hand, over the tropical rainforests, SOA-sv and SOA-iv 350 351 concentrations are similar due to the low fraction of IVOCs assumed for the open biomass burning OA emissions. Overall, the reference simulation yields a 352 tropospheric OA burden of 1.98 Tg consisting of 12% POA, 18% SOA-sv, 32% SOA-353 iv, and 38% SOA-v. More details about the reference case results can be found in 354 Tsimpidi et al. (2016). 355

A comprehensive AMS dataset from field campaigns performed in the Northern 356 357 Hemisphere during 2001-2010 (Tsimpidi et al., 2016) has been used to evaluate the model performance for each simulation. The mean bias (MB), mean absolute gross 358 error (MAGE), normalized mean bias (NMB), normalized mean error (NME), and the 359 360 root mean square error (RMSE) are used to assess the model performance for POA (versus AMS hydrocarbon-like aerosol (HOA); Table 2) and SOA (versus AMS 361 362 oxygenated organic aerosol (OOA); Table 3). Tsimpidi et al. (2016) have shown that, as expected the model underestimates the concentrations of POA and SOA over urban 363 364 locations due to its coarse resolution and missing sources in the emission database (e.g., cold vehicle start and wood burning emissions in winter). Therefore, urban 365 locations are excluded from our analysis in order to avoid misinterpretation of the 366 sensitivity results and their effects on OA model performance. A comprehensive 367 analysis of the model evaluation based on the reference scenario results can be found 368 in Tsimpidi et al. (2016) and will be used here as a reference for analysing the effect 369 of each sensitivity scenario on the performance of the model. EMAC reproduces POA 370 levels with very little bias (NMB= -3%; Table 2). On the other hand, OOA 371 concentrations are underpredicted (-31%; Table 3) indicating that the model may be 372 missing an important source or formation pathway of SOA especially in winter 373 (Tsimpidi et al., 2016) or may be removing the corresponding pollutants faster. 374 Another possible reason for the underprediction of OOA is the uncertainty in SOA 375

yields due to wall losses in laboratory chambers. Zhang et al. (2014) demonstrated
that while the particle losses are routinely accounted for, losses of semivolatile vapors
are not well evaluated and can lead to substantial underestimations of the SOA
formation.

380

381 **5** Sensitivity to emission factors

382 **5.1 Low volatility**

In the first sensitivity test, the IVOC emissions are set to zero and only semi-383 384 volatile organic compounds are emitted. This is accompanied by an increase of SVOC emissions from anthropogenic and open biomass burning sources by 100% and 40%, 385 respectively. This initial partitioning of the emissions favors the particulate phase, 386 resulting in an increase of POA compared to the reference scenario (Figure 4a). The 387 largest fPOA and bbPOA increases are predicted over Eastern China (4.3 µg m⁻³) and 388 the Congo Basin (3.9 µg m⁻³), respectively. The higher SVOC emissions in the 389 sensitivity simulation result in an increase of the simulated SOA-sv concentrations as 390 well (Figure 5a). However, since a large fraction of the emitted SVOCs remains in the 391 392 particle phase, the SOA-sv concentration increase is smaller than the corresponding 393 changes in POA. Relatively strong fSOA-sv and bbSOA-sv increases are found over the Indo-Gangetic Plane (IGP) (0.4 μ g m⁻³) and the Congo Basin (1.3 μ g m⁻³), 394 respectively. The "low volatility" simulation does not predict any SOA-iv as it 395 assumes zero IVOC emissions. Therefore, SOA-iv concentrations are zero around the 396 397 globe, resulting in substantial decreases in areas where the reference simulation predicts high SOA-iv levels (Figures 3dc and 6a). 398

399 The significant decrease of organic emissions from anthropogenic sources (Table 1) due to the lack of IVOC emissions results in an overall decrease of total OA 400 concentrations by up to 5 μ g m⁻³ over anthropogenically polluted regions (Figure 7a). 401 On the other hand, organic emissions from open biomass burning sources remain at 402 the same level as the reference simulation (Table 1), however, they are assumed to 403 have lower volatility. This results in an increase of total OA concentrations in the 404 sensitivity simulation by up to 2 μ g m⁻³ over the tropical and boreal forests. Overall, 405 the calculated tropospheric burden of POA in the sensitivity simulation increases by 406 around 50% due to the increase of the SVOC emissions (Table 2). For the same 407 reason, the tropospheric fSOA-sv and bbSOA-sv burdens increase by 14% and 39%, 408

409 respectively. Nevertheless, the absence of IVOC emissions, and thus the significant 410 decrease of anthropogenic organic compound emissions, results in a decrease of the 411 total OA tropospheric burden by 23%. This result emphasizes the importance of the 412 volatility distributions used in the simulation and the contribution of IVOC emissions 413 to SOA formation on a global scale.

The simulated POA in the reference model configuration is very close to the 414 average HOA concentrations derived from the AMS measurements (Table 3). 415 Therefore, assuming lower volatility of the organic emissions results in overprediction 416 417 (NMB=43%). However, the performance of the model is significantly improved during winter (Figure 8) since POA concentrations during that season were 418 underpredicted (NMB=-37%; Tsimpidi et al., 2016). On the other hand, during spring 419 the overestimate of POA increases in the sensitivity simulation (NMB=86%) 420 compared to the reference (NMB=26%). For summer and autumn, the performance of 421 the model changes from a slight underestimation of POA in the reference (NMB=-422 15%) to a slight overprediction in the sensitivity test (NMB=30%). The performance 423 of the model in reproducing the OOA concentrations worsens in this sensitivity 424 simulation (Table 4). OOA was underpredicted by the model reference simulation 425 426 (NMB=-31%), therefore, by neglecting SOA formation from IVOC emissions in the sensitivity run results in an even larger OOA underestimation (NMB=-52%). The 427 428 performance of the model does not change significantly during winter (Figure 8) since the simulated SOA formation during this season is low (Tsimpidi et al., 2016). The 429 430 highest change in model performance occurs during spring when SOA is predicted to reach the annual maximum (Tsimpidi et al., 2016); the predicted underestimation of 431 432 OOA increases from 20% in the reference to 50% in the sensitivity simulation. These results indicate that the omission of IVOCs as a source of SOA in atmospheric models 433 can result in a significant underestimation of OA concentrations, especially during 434 periods where formation of SOA is strong. 435

436

437 **5.2 High IVOCs**

In the second sensitivity simulation, the increased IVOC emissions result in an increase of total organics by 60%, and 150% from anthropogenic and open biomass burning sources, respectively (Table 1). These additional organic emissions are distributed only in the intermediate volatility bins, therefore, their impact on the 442 simulated POA and SOA-sv levels is marginal (Figures 4b and 5b, respectively). POA increases, up to 0.6 µg m⁻³ over Eastern China, while SOA-sv decreases, up to 0.3 µg 443 m⁻³ over the Congo Basin. This effect can be explained by the assumption that SOA-444 sv and SOA-iv form a pseudo-ideal solution. As a result, the increased SOA-iv 445 concentrations calculated in the sensitivity simulation favor the partitioning of the 446 fresh SVOCs into the aerosol phase, forming additional POA. At the same time, 447 SVOCs decrease in the gas-phase and therefore the formation of SOA-sv is reduced in 448 the sensitivity simulation. As expected, the largest effect is found for SOA-iv (Figure 449 6b). The significant increase of IVOC emissions results in large changes of SOA-iv 450 over areas close to anthropogenic sources (up to 5.7 µg m⁻³ over the IGP) and biomass 451 burning regions (up to 5.3 µg m⁻³ over the Congo Basin). The increase of SOA-iv 452 dominates the effect on total OA concentrations that increase up to 6 μ g m⁻³ (Figure 453 7b). Overall, the predicted changes of the tropospheric burden of POA and SOA-sv 454 are small (Table 2). However, the tropospheric burdens of fSOA-iv and bbSOA-iv 455 increase by 88% and 115%, respectively, resulting in an increase of the total OA 456 burden by 38%. 457

The additional IVOC emissions assumed in this sensitivity test do not affect the 458 459 performance of the model for POA. On the other hand, these additional emissions bring the predicted SOA concentrations closer to the measured OOA levels (Table 4; 460 Figure 8). The NMB improves from -31% in the reference simulation to -10%. With 461 the exception of winter, where the model still underpredicts OOA levels (MB= $2.2 \mu g$ 462 463 m⁻³, Figure 8), the performance of the model for SOA improves with seasonal NMB ranging from -16% (during summer) to 11% (during spring); compared to -33% and 464 465 -20% for the reference model, respectively. The improved performance of the model due to the increase of IVOC emissions supports the hypothesis that the IVOC 466 emissions may have been underestimated in previous modeling studies that assumed 467 IVOC/POA =1.5 (Ots et al., 2016). 468

469

470 **5.3 Alternative POA emissions**

The final emission sensitivity test is used to estimate the uncertainty introduced by the choice of emission database. The inventories used in the sensitivity simulation assume 36% lower fuel combustion OA emissions and 33% higher biomass burning OA emissions compared to the reference simulation, while the total OA emissions are 475 only reduced by 9%. Since the volatility distribution of the emissions is identical to the reference simulation, the fractional changes of the calculated POA, SOA-sv, SOA-476 iv are also similar (Table 4). The tropospheric burden of fOA (the sum of fPOA, 477 fSOA-sv, and fSOA-iv) decreases by 34%. On the other hand, bbOA (the sum of 478 bbPOA, bbSOA-sv, and bbSOA-iv) increases by 11%. Overall, the total tropospheric 479 OA burden increases by only 4%. The changes in fOA and bbOA concentrations, 480 however, are not spatially uniform. Over Europe, fOA decreases everywhere, up to 481 3.3 µg m⁻³, except in Paris where fOA increases by 0.24 µg m⁻³. Over the US fOA 482 slightly increases (mostly over the northeast by up to 0.6 µg m⁻³), while it decreases 483 over Mexico by as much as 1.7 µg m⁻³. The largest fOA change is predicted over Asia 484 where fOA decreases significantly, up to 8.3 μ g m⁻³, mostly over East Asia and the 485 IGP. bbOA decreases over the boreal forests (up to 3.6 μ g m⁻³), while it increases 486 significantly over the Southeast Asian tropical forests by up to 14 μ g m⁻³. Over the 487 Amazon and Congo forests, bbOA concentrations change significantly (the bbOA 488 changes vary from -2.4 to 3.3 μ g m⁻³ in the Amazon, and from -5.3 to 7.8 μ g m⁻³ in 489 Congo) but the average bbOA concentration over both regions remains the same. 490 491 Overall, the fOA and bbOA emission changes lead to total OA increases over the 492 tropical and boreal forests and decreases over anthropogenic areas (Figure 7c).

The lower OA emissions used in the sensitivity simulation (especially over China and Europe) result in a reduction of both total POA and SOA concentrations (Tables 2 and 3). Consequently, the model now underestimates POA with NMB=-25% and SOA with NMB=-40%. These results suggest that the use of the CMIP5 RCP4.5 emission inventory in EMAC results in OA concentrations that agree more closely to the measurements compared to the AEROCOM database. It also underscores the large uncertainty associated with primary OA emissions.

500

501 6 Sensitivity to aging reactions

502 **6.1 Higher aging reaction rate**

503 In this sensitivity simulation, the photochemical reaction rate constant for SVOCs 504 and IVOCs has been doubled compared to the reference. This results in an increase of 505 SOA-sv and SOA-iv concentrations worldwide (Figures 5d and 6d). SOA-sv 506 increases, up to 0.65 μ g m⁻³, mostly over the tropics and the polluted regions of 507 Eastern China and the IGP (Figure 5d). The effect on SOA-iv concentrations is even

508 more significant since IVOCs undergo more oxidation steps before forming SOA than SVOCs. SOA-iv increased by up to 2.4 μ g m⁻³ mostly over the IGP and Eastern China 509 (Figure 6d). The SOA-iv increase over the tropics is smaller (up to $0.8 \ \mu g \ m^{-3}$) due to 510 the assumed low fraction of IVOCs in biomass burning emissions. Overall, the 511 tropospheric burdens of SOA-sv and SOA-iv both increase by 0.04 Tg (or 11% and 512 7%, respectively). POA is not expected to be affected directly by the change of the 513 reaction rate constant. However, the substantial reduction of gas-phase SVOCs (due to 514 their increased reactivity) results in the re-evaporation of POA to achieve equilibrium, 515 reducing its concentration (Figure 4d) mainly over the tropics (up to 0.21 µg m⁻³). 516 This results in an overall decrease of the tropospheric POA burden by 8%. Following 517 the significant increase of both SOA-sv and SOA-iv, total OA increases worldwide by 518 up to 3 μ g m⁻³ (Figure 7d). Overall, the tropospheric burden of total OA increases by 519 4%. 520

The model performance for POA is not affected by the change of the reaction rate 521 constant (Table 2) since POA remains largely unchanged over the Northern 522 Hemisphere (Figure 4d). On the other hand, the performance of the model regarding 523 SOA is significantly improved (Table 3). The underestimation of SOA by the model 524 525 is reduced (NMB=-22%) compared to the reference (NMB=-31%). The best performance is found during spring (NMB=-7%) when the calculated SOA is almost 526 527 unbiased. However, during winter, the model still severely underestimates SOA (NMB=-77%), which indicates that the gas-phase oxidation of SVOCs and IVOCs 528 529 does not suffice to explain the underprediction of SOA in winter.

530

531 6.2 Alternative aging scheme

In this sensitivity simulation we used the chemical aging scheme of Robinson et al. 532 (2007) which is currently the most commonly used in VBS models. This aging 533 scheme is accompanied by changes in the number of volatility bins used and the 534 assigned emission factors, the oxidation rate constant, the volatility reductions after 535 each oxidation step, and the increase in mass due to added oxygen (as discussed in 536 Sect. 3.2). The changes in the number of volatility bins and the emission factors used 537 for the SVOCs (Figure 2d) result in reduced condensation of SVOCs into the 538 particulate phase during the initial partitioning and therefore to a significant decrease 539 of POA (Figure 4e). The decrease of POA is global and most prominent over Eastern 540

541 China (up to 9.3 μ g m⁻³). This reflects a significant change in the tropospheric burdens 542 of both fPOA and bbPOA by 65% and 38%, respectively.

Furthermore, the reduced fraction of SVOCs to total OA emissions (see Section 543 3.2) results in a worldwide decrease of SOA-sv (Figure 5e) and an increase of SOA-iv 544 (Figure 6e). SOA-sv decreases up to 1.8 µg m⁻³ over the Congo Basin and the IGP. 545 Similar to POA, the tropospheric burden of fSOA-sv and bbSOA-sv decreases by 546 68% and 47%, respectively. On the other hand, the increase in SOA-iv, due to the 547 increase in the IVOC fraction of the emissions, is not as strong as the decrease of 548 549 SOA-sv (Table 4). This is due to the slower aging in the sensitivity simulation (Figure 1b), compared to the reference (Figure 1a), which limits the formation of SOA from 550 IVOCs. SOA-iv increases up to $0.9 \ \mu g \ m^{-3}$ over the Congo Basin and the IGP, while it 551 locally decreases by 0.1 µg m⁻³ over Beijing, for example. The tropospheric burden of 552 fSOA-iv and bbSOA-iv increases by 14% and 30%, respectively. Overall, the sum of 553 SOA-sv and SOA-iv decreases by 7% due to the slower aging in this sensitivity 554 simulation. Following the simultaneous decrease of both POA and SOA, total OA 555 decreases worldwide by up to 11 μ g m⁻³ (Figure 7e) and its tropospheric burden is 556 reduced by 0.2 Tg (or 10%). 557

The reduction of both modelled POA and SOA results in reduced agreement of the model with AMS measurements. Especially for POA, the modeled concentrations decrease by 67% in the sensitivity simulation, resulting in a significant underprediction of AMS-HOA (NMB=-67%). Modelled SOA also decreases (by 10%) in the sensitivity simulation, which degrades the model agreement with AMS-OOA measurements (NMB=-38%). This sensitivity test underscores the significance of the volatility distribution of the organic emissions and the associated aging scheme.

565

566 6.3 Hybrid aging scheme

567 The final chemistry sensitivity simulation focuses on the photochemical aging of 568 IVOCs and assumptions regarding the first oxidation step. The approach used here is 569 similar to the oxidation of the traditional VOCs, in contrast with the reference where 570 the oxidation of IVOCs produces only one product with two orders of magnitude 571 reduced volatility. However, the stoichiometric coefficient used in the reference 572 (equal to 1.15) is higher than the aerosol yields used in the sensitivity simulation 573 (Section 3.2). This results in a reduction of SOA-iv concentrations by up to 2.2 µg m⁻³

574 (Figure 6f). Since the chemical scheme for SVOCs is identical in both the reference and the sensitivity simulations, no significant change is found in either SOA-sv or 575 POA (Figure 5f and 4f, respectively). The decrease of SOA-iv concentrations has a 576 marginal effect on the initial partitioning of SVOC emissions resulting in slightly less 577 POA and more SOA-sv (by up to 0.1 µg m⁻³ in either case). Therefore, total OA 578 concentrations are reduced worldwide following the decrease of SOA-iv. Overall, the 579 tropospheric burden of SOA-iv decreases by 37% in the sensitivity simulation 580 resulting in a decrease of total OA by 13% (Table 4). 581

- The simulated POA concentrations remain almost unchanged in the sensitivity simulation; therefore, similar to the reference, the calculated POA is unbiased compared to measurements (Table 2). On the other hand, the lower SOA-iv concentrations calculated by the model in this sensitivity test aggravate the underestimation of OOA by the model (NMB=-39%). The decrease of modelled SOA-iv concentrations is larger during spring (13%) and the calculated NMB for SOA deteriorates from -20% in the reference to -30% in the sensitivity simulation.
- 589

590 7 Sensitivity to wet/dry removal of organic vapors

591 7.1 Reduced Henry's law constant

In this sensitivity test we used a Henry's law constant that is two orders of 592 593 magnitude lower than in the reference simulation (see Section 3.3) for the gas-phase SVOCs and IVOCs. This change decreases their removal rate, thus increasing their 594 595 lifetime and the concentrations of both POA (due to the condensation of the fresh SVOCs) and SOA (due to the condensation of the chemically aged SVOCs and 596 IVOCs). POA increases up to 0.7 µg m⁻³ over Eastern China (Figure 4g) where POA 597 concentrations are relatively high (Figure 3b), however, the increase of POA in the 598 rest of the world is less than 0.2 μ g m⁻³ (Figure 4g). SOA-sv increases up to 0.2 μ g 599 m⁻³ mostly over the Congo Basin and the IGP (Figure 5g). The most significant 600 change is calculated for SOA-iv. SOA-iv is formed from gases (i.e., IVOCs) that need 601 to go through more than two oxidation steps to be able to condense to the aerosol 602 phase (in comparison to only one oxidation step for SVOCs). Therefore, by lowering 603 the Henry's law constant of IVOCs we prolong the lifetime of SOA-iv precursors, and 604 their ability to undergo multiple oxidation steps and produce aerosols. This results in a 605 significant increase of SOA-iv by up to 1.2 μ g m⁻³ (Figure 6g). Total OA increases by 606

up to 2 μ g m⁻³ due to the simultaneous increase of both POA and SOA (Figure 7g). 607 Overall, the tropospheric burden of SOA-iv increases by 17% and of total OA by 8%. 608 It is also worth noticing that the tropospheric burden of fOA (sum of fPOA, fSOA-sv, 609 and fSOA-iv) increases by 18% compared to an increase of 5% of the bbOA (sum of 610 bbPOA, bbSOA-sv, and bbSOA-iv). The above results emphasize the significance of 611 the removal of organic vapors for the calculated OA concentrations, and corroborate 612 the importance of constraining the Henry' law constants of SVOCs and more 613 importantly of IVOCs. 614

The change of Henry's law constant of SVOCs does not affect the model 615 performance for POA significantly. POA slightly increases (by 4%), eliminating the 616 already low model bias (Table 2). The SOA increase (by 12%) in the sensitivity 617 simulation (mainly due to the increased SOA-iv) results in reduced SOA 618 underestimation (Table 2). In both POA and SOA cases the effect is more important 619 during winter, when wet removal is most efficient, and lower during summer. POA 620 increases during winter by 10% while during summer it remains unchanged. SOA 621 increases during winter by 26% and during summer by only 3%, with spring and 622 autumn in between (~12%). Despite the wintertime POA and SOA increase in this 623 624 sensitivity simulation, the model still underestimates POA (NMB=-31%) and SOA (NMB=-78%) during this season (Figure 8). 625

626

627 7.2 Different Henry's law constant for POA and SOA

628 In the last sensitivity test we assume that the freshly emitted SVOCs and IVOCs are hydrophobic (with the Henry's law constant H being 4 orders of magnitude lower 629 630 than the reference) while after photochemical aging H increases to match the value used in the reference (see Section 3.3). POA increases up to 0.7 µg m⁻³, mostly over 631 Eastern China and to a lesser degree over Eastern Europe and Russia (Figure 4h). 632 SOA-sv increases up to 0.2 µg m⁻³, mostly over the tropical forests of Central Africa 633 and Southeastern Asia, as well as over Eastern China and the IGP (Figure 5h). SOA-634 iv also increases by up to 1 μ g m⁻³ (Figure 6h) because fresh IVOCs are more 635 hydrophobic in the sensitivity simulation, therefore, the time available to react with 636 OH is extended, forming additional SOA-iv. Total OA concentrations increase by up 637 to 2 µg m⁻³ over Eastern China (Figure 7h). The tropospheric burden of total OA 638

639 increases by 8% in this sensitivity test with the strongest increase coming from fSOA-640 iv (21%).

Both the predicted POA and SOA increase in the sensitivity simulation by 6% and 641 12% respectively. This results in a small overprediction of POA (NMB=4%), 642 compared to a small underprediction in the reference (NMB=-3%). For SOA, NMB 643 improves in the sensitivity simulation (NMB=-23%) compared to the reference (-644 31%). Similar to the previous sensitivity test (Section 7.1) the effect is more relevant 645 during winter (POA and SOA increase by 9% and 36%, respectively), followed by 646 647 spring (POA and SOA increase by 8% and 16%, respectively) and autumn (POA and SOA increase by 7% and 10%, respectively), and is small during summer (POA and 648 SOA increase by 2% and 5%, respectively) (Figures 8). This results in an improved 649 model performance for both POA and SOA during all seasons. The highest 650 improvement is found for SOA during spring when the NMB is reduced to -6% from 651 -20% in the reference. Despite the significant increase of SOA concentrations during 652 winter (by 36%), the model still strongly underestimates SOA (NMB=-76%), 653 indicating that the model underperprediction of OOA cannot be attributed solely to 654 errors in the simulation of removal processes. Therefore, we expect that the 655 656 discrepancy in this season is related to sources that are missing or underestimated in emission inventories (e.g., residential wood combustion in winter (Denier van der 657 658 Gon et al., 2015)), to additional oxidation pathways (e.g., aqueous-phase and heterogeneous oxidation reactions), and to uncertainties in SOA yields due to wall 659 660 losses in laboratory chambers.

661

662 8 Summary and conclusions

We investigated the effect of parameters and assumptions that control the emissions, photochemical aging, and scavenging efficiency of LVOCs, SVOCs and IVOCs on the simulated OA concentrations. We used the organic aerosol module ORACLE, based on the VBS framework, in the EMAC global chemistry-climate model. A global dataset of AMS measurements has been used to evaluate the predicted POA and SOA concentrations, based on a number of sensitivity tests.

669 The results show that total OA concentrations are sensitive to the emissions of 670 IVOCs. By neglecting these emissions, the model produces unrealistically low SOA 671 concentrations resulting in the poorest model performance (NMB=-52%) compared to 672 the other eight simulations conducted (Table 3). Conversely, increasing the IVOC emissions substantially improved the SOA model results, leading to the best model 673 performance (NMB=-10%). These results emphasize the need to accurately estimate 674 the IVOC emissions independently. The use of a more accurate POA emission 675 inventory is found to be of prime importance for the model performance, especially to 676 improve simulated POA concentrations in winter. In our tests, using an alternative 677 POA emission inventory led to a NMB of -25% compared to a low bias in the 678 performance of the reference model. 679

680 Sensitivity tests of the photochemical aging of SVOCs and IVOCs indicate the importance of the OH-reaction rate. Assuming an increased reactivity of SVOC and 681 IVOC with OH improves the model results for SOA (NMB=-22%). This is even more 682 important for the IVOCs, which participate in a larger number of photochemical 683 reactions during atmospheric transport compared to the SVOCs. Another assumption 684 tested is that oxidation reactions of IVOCs are similar to many other VOCs, and 685 produce partly oxidized compounds with several orders of magnitude lower 686 volatilities. Despite the strong volatility reduction of the IVOC oxidation products, the 687 performance of the model was similar to the reference simulation since the IVOC 688 689 aerosol yields were lower compared to the stoichiometric coefficient used in the reference. The use of an alternative aging scheme (based on Robinson et al., 2007) 690 691 resulted in lower SOA concentrations since the photochemical aging of SVOCs and IVOCs was less effective. This led to a slight reduction in model performance for 692 693 SOA (Table 3). In this sensitivity test the fraction of SVOCs to total OA emissions was lower compared to the reference, resulting in a significant reduction of POA and 694 695 a reduced model performance (NMB=-67%). This underscores the significance of the assumed volatility distribution of OA emissions. 696

697 The calculated OA concentrations are highly sensitive to the scavenging efficiency of the gas-phase SVOCs and IVOCs, expressed by the Henry's law 698 constant (H). Reducing H resulted in an increase of both POA and SOA 699 concentrations, especially from the oxidation of IVOCs. This increase yielded 700 improved model performance, particularly for SOA (Table 3). Assuming different 701 hygroscopicity for the freshly emitted and the photochemically processed SVOCs and 702 703 IVOCs resulted in similar improvement of the model results (Tables 2 and 3). In this sensitivity test, the simulated POA improved substantially during winter (NMB=-704

29%) during which the model has difficulties reproducing AMS observations
(Tsimpidi et al., 2016). Nevertheless, SOA was still underpredicted during winter
(NMB=-76%) indicating that other processes (e.g., seasonally dependent residential
wood combustion emissions, aqueous-phase oxidation paths, uncertainties in SOA
yields due to wall losses in chambers) are a main cause of the inadequate
performance.

Our results indicate that IVOCs can be major contributors to OA formation on a global scale. However, their abundance and physicochemical properties are poorly known, and more research is needed to determine the parameters that control their emissions, chemistry, and atmospheric removal. According to the model results, a combination of increased IVOC emissions, enhanced photochemical aging of IVOCs, and decreased hygroscopicity of the freshly emitted IVOCs can help reduce discrepancies between simulated SOA and observed OOA concentrations.

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725 **10. References**

- Aiken, A. C., Decarlo, P. F., Kroll, J. H., Worsnop, D. R., Huffman, J. A., Docherty,
 K. S., Ulbrich, I. M., Mohr, C., Kimmel, J. R., Sueper, D., Sun, Y., Zhang, Q.,
 Trimborn, A., Northway, M., Ziemann, P. J., Canagaratna, M. R., Onasch, T. B.,
 Alfarra, M. R., Prevot, A. S. H., Dommen, J., Duplissy, J., Metzger, A.,
 Baltensperger, U., and Jimenez, J. L.: O/C and OM/OC ratios of primary,
 secondary, and ambient organic aerosols with high-resolution time-of-flight
 aerosol mass spectrometry, Environmen. Sci. & Technol., 42, 4478-4485, 2008.
- Athanasopoulou, E., Vogel, H., Vogel, B., Tsimpidi, A. P., Pandis, S. N., Knote, C.,
 and Fountoukis, C.: Modeling the meteorological and chemical effects of
 secondary organic aerosols during an EUCAARI campaign, Atmos. Chem. Phys.,
 13, 625-645, 2013.
- Atkinson, R. and Arey, J.: Atmospheric degradation of volatile organic compounds,
 Chemical Reviews, 103, 4605-4638, 2003.
- Bergstrom, R., van der Gon, H. A. C. D., Prevot, A. S. H., Yttri, K. E., and Simpson,
 D.: Modelling of organic aerosols over Europe (2002-2007) using a volatility
 basis set (VBS) framework: application of different assumptions regarding the
 formation of secondary organic aerosol, Atmos. Chem. Phys., 12, 8499-8527,
 2012.

- Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P.,
 Ruiz, L. H., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue,
 N. M., Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of organic
 compounds using aerosol mass spectrometry: characterization, improved
 calibration, and implications, Atmospheric Chemistry and Physics, 15, 253-272,
 2015.
- Chacon-Madrid, H. J., Henry, K. M., and Donahue, N. M.: Photo-oxidation of
 pinonaldehyde at low NOx: from chemistry to organic aerosol formation,
 Atmospheric Chemistry and Physics, 13, 3227-3236, 2013.
- Clarke, L., Edmonds, J., Jacoby, H., Pitcher, H., Reilly, J., and Richels, R.: Scenarios
 of greenhouse gas emissions and atmospheric concentrations (Part A) and review
 of integrated scenario development and application (Part B). A report by the U.S.
 climate change science program and the subcommittee on global change research,
 Department of Energy, Office of Biological & Environmental Research,
 Washington, D.C., USA, 2007.
- Donahue, N. M., Robinson, A. L., Stanier, C. O., and Pandis, S. N.: Coupled
 partitioning, dilution, and chemical aging of semivolatile organics, Environ. Sci.
 Technol., 40, 2635-2643, 2006.
- Fountoukis, C., Megaritis, A. G., Skyllakou, K., Charalampidis, P. E., Pilinis, C.,
 Denier van der Gon, H. A. C., Crippa, M., Canonaco, F., Mohr, C., Prévôt, A. S.
 H., Allan, J. D., Poulain, L., Petäjä, T., Tiitta, P., Carbone, S., Kiendler-Scharr,
 A., Nemitz, E., O'Dowd, C., Swietlicki, E., and Pandis, S. N.: Organic aerosol
 concentration and composition over Europe: insights from comparison of regional
 model predictions with aerosol mass spectrometer factor analysis, Atmos. Chem.
 Phys. Discuss., 14, 7597-7635, 2014.
- Fountoukis, C., Racherla, P. N., van der Gon, H. A. C. D., Polymeneas, P.,
 Charalampidis, P. E., Pilinis, C., Wiedensohler, A., Dall'Osto, M., O'Dowd, C.,
 and Pandis, S. N.: Evaluation of a three-dimensional chemical transport model
 (PMCAMx) in the European domain during the EUCAARI May 2008 campaign,
 Atmos. Chem. and Phys., 11, 10331-10347, 2011.
- Grieshop, A. P., Logue, J. M., Donahue, N. M., and Robinson, A. L.: Laboratory
 investigation of photochemical oxidation of organic aerosol from wood fires 1:
 measurement and simulation of organic aerosol evolution, Atmos. Chem. Phys.,
 9, 1263-1277, 2009.
- Hodzic, A., Aumont, B., Knote, C., Lee-Taylor, J., Madronich, S., and Tyndall, G.:
 Volatility dependence of Henry's law constants of condensable organics:
 Application to estimate depositional loss of secondary organic aerosols,
 Geophysical Research Letters, 41, 4795-4804, 2014.
- Hodzic, A., Jimenez, J. L., Madronich, S., Canagaratna, M. R., DeCarlo, P. F.,
 Kleinman, L., and Fast, J.: Modeling organic aerosols in a megacity: potential
 contribution of semi-volatile and intermediate volatility primary organic
 compounds to secondary organic aerosol formation, Atmos. Chem. Phys., 10,
 5491-5514, 2010.
- Jathar, S. H., Farina, S. C., Robinson, A. L., and Adams, P. J.: The influence of semi-volatile and reactive primary emissions on the abundance and properties of global organic aerosol, Atmos. Chem. Phys., 11, 7727-7746, 2011.
- Jathar, S. H., Gordon, T. D., Hennigan, C. J., Pye, H. O. T., Pouliot, G., Adams, P. J.,
 Donahue, N. M., and Robinson, A. L.: Unspeciated organic emissions from
 combustion sources and their influence on the secondary organic aerosol budget

- in the United States, Proceedings of the National Academy of Sciences of theUnited States of America, 111, 10473-10478, 2014.
- Jathar, S. H., Miracolo, M. A., Presto, A. A., Donahue, N. M., Adams, P. J., and
 Robinson, A. L.: Modeling the formation and properties of traditional and nontraditional secondary organic aerosol: problem formulation and application to
 aircraft exhaust, Atmospheric Chemistry and Physics, 12, 9025-9040, 2012.
- 799 Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prevot, A. S. H., Zhang, Q., Kroll, J. H., DeCarlo, P. F., Allan, J. D., Coe, H., Ng, N. L., Aiken, A. C., 800 Docherty, K. S., Ulbrich, I. M., Grieshop, A. P., Robinson, A. L., Duplissy, J., 801 802 Smith, J. D., Wilson, K. R., Lanz, V. A., Hueglin, C., Sun, Y. L., Tian, J., Laaksonen, A., Raatikainen, T., Rautiainen, J., Vaattovaara, P., Ehn, M., 803 Kulmala, M., Tomlinson, J. M., Collins, D. R., Cubison, M. J., Dunlea, E. J., 804 805 Huffman, J. A., Onasch, T. B., Alfarra, M. R., Williams, P. I., Bower, K., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K., 806 Salcedo, D., Cottrell, L., Griffin, R., Takami, A., Miyoshi, T., Hatakeyama, S., 807 Shimono, A., Sun, J. Y., Zhang, Y. M., Dzepina, K., Kimmel, J. R., Sueper, D., 808 809 Jayne, J. T., Herndon, S. C., Trimborn, A. M., Williams, L. R., Wood, E. C., Middlebrook, A. M., Kolb, C. E., Baltensperger, U., and Worsnop, D. R.: 810 Evolution of organic aerosols in the atmosphere, Science, 326, 1525-1529, 2009. 811
- Jo, D. S., Park, R. J., Kim, M. J., and Spracklen, D. V.: Effects of chemical aging on
 global secondary organic aerosol using the volatility basis set approach, Atmos.
 Environ., 81, 230-244, 2013.
- Jöckel, P., Tost, H., Pozzer, A., Bruehl, C., Buchholz, J., Ganzeveld, L., Hoor, P.,
 Kerkweg, A., Lawrence, M. G., Sander, R., Steil, B., Stiller, G., Tanarhte, M.,
 Taraborrelli, D., Van Aardenne, J., and Lelieveld, J.: The atmospheric chemistry
 general circulation model ECHAM5/MESSy1: consistent simulation of ozone
 from the surface to the mesosphere, Atmos. Chem. Phys., 6, 5067-5104, 2006.
- Kanakidou, M., Seinfeld, J. H., Pandis, S. N., Barnes, I., Dentener, F. J., Facchini, M.
 C., Van Dingenen, R., Ervens, B., Nenes, A., Nielsen, C. J., Swietlicki, E.,
 Putaud, J. P., Balkanski, Y., Fuzzi, S., Horth, J., Moortgat, G. K., Winterhalter,
 R., Myhre, C. E. L., Tsigaridis, K., Vignati, E., Stephanou, E. G., and Wilson, J.:
 Organic aerosol and global climate modelling: a review, Atmos. Chem. Phys., 5,
 1053-1123, 2005.
- Karydis, V. A., Tsimpidi, A. P., Bacer, S., Pozzer, A., Nenes, A., and Lelieveld, J.:
 Global impact of mineral dust on cloud droplet number concentration, Atmos.
 Chem. Phys. Discuss., 2016, 1-40, 2016a.
- Karydis, V. A., Tsimpidi, A. P., Pozzer, A., Astitha, M., and Lelieveld, J.: Effects of
 mineral dust on global atmospheric nitrate concentrations, Atmos. Chem. Phys.,
 16, 1491-1509, 2016b.
- Kerkweg, A., Buchholz, J., Ganzeveld, L., Pozzer, A., Tost, H., and Jöckel, P.:
 Technical Note: An implementation of the dry removal processes DRY
 DEPosition and SEDImentation in the Modular Earth Submodel System
 (MESSy), Atmos. Chem. Phys., 6, 4617-4632, 2006a.
- Kerkweg, A., Sander, R., Tost, H., and Jöckel, P.: Technical note: Implementation of
 prescribed (OFFLEM), calculated (ONLEM), and pseudo-emissions (TNUDGE)
 of chemical species in the Modular Earth Submodel System (MESSy), Atmos.
 Chem. Phys., 6, 3603-3609, 2006b.

- Kroll, J. H. and Seinfeld, J. H.: Chemistry of secondary organic aerosol: Formation
 and evolution of low-volatility organics in the atmosphere, Atmos. Environ., 42,
 3593-3624, 2008.
- Lauer, A., Eyring, V., Hendricks, J., Joeckel, P., and Lohmann, U.: Global model
 simulations of the impact of ocean-going ships on aerosols, clouds, and the
 radiation budget, Atmos. Chem. Phys., 7, 5061-5079, 2007.
- Li, G., Zavala, M., Lei, W., Tsimpidi, A. P., Karydis, V. A., Pandis, S. N.,
 Canagaratna, M. R., and Molina, L. T.: Simulations of organic aerosol
 concentrations in Mexico City using the WRF-CHEM model during the MCMA2006/MILAGRO campaign, Atmospheric Chemistry and Physics, 11, 3789-3809,
 2011.
- May, A. A., Levin, E. J. T., Hennigan, C. J., Riipinen, I., Lee, T., Collett, J. L.,
 Jimenez, J. L., Kreidenweis, S. M., and Robinson, A. L.: Gas-particle partitioning
 of primary organic aerosol emissions: 3. Biomass burning, Journal of
 Geophysical Research-Atmospheres, 118, 11327-11338, 2013a.
- May, A. A., Presto, A. A., Hennigan, C. J., Nguyen, N. T., Gordon, T. D., and
 Robinson, A. L.: Gas-particle partitioning of primary organic aerosol emissions:
 (1) Gasoline vehicle exhaust, Atmospheric Environment, 77, 128-139, 2013b.
- May, A. A., Presto, A. A., Hennigan, C. J., Nguyen, N. T., Gordon, T. D., and
 Robinson, A. L.: Gas-Particle Partitioning of Primary Organic Aerosol
 Emissions: (2) Diesel Vehicles, Environmental Science & Technology, 47, 82888296, 2013c.
- Murphy, B. N., Donahue, N. M., Fountoukis, C., Dall'Osto, M., O'Dowd, C.,
 Kiendler-Scharr, A., and Pandis, S. N.: Functionalization and fragmentation
 during ambient organic aerosol aging: application of the 2-D volatility basis set to
 field studies, Atmospheric Chemistry and Physics, 12, 10797-10816, 2012.
- Murphy, B. N. and Pandis, S. N.: Simulating the formation of semivolatile primary
 and secondary organic aerosol in a regional chemical transport model, Environ.
 Sci. Technol., 43, 4722-4728, 2009.
- Ots, R., Young, D. E., Vieno, M., Xu, L., Dunmore, R. E., Allan, J. D., Coe, H.,
 Williams, L. R., Herndon, S. C., Ng, N. L., Hamilton, J. F., Bergström, R., Di
 Marco, C., Nemitz, E., Mackenzie, I. A., Kuenen, J. J. P., Green, D. C., Reis, S.,
 and Heal, M. R.: Simulating secondary organic aerosol from missing dieselrelated intermediate-volatility organic compound emissions during the Clean Air
 for London (ClearfLo) campaign, Atmos. Chem. Phys. Discuss., 2016, 1-36,
 2016.
- Pozzer, A., de Meij, A., Pringle, K. J., Tost, H., Doering, U. M., van Aardenne, J., and
 Lelieveld, J.: Distributions and regional budgets of aerosols and their precursors
 simulated with the EMAC chemistry-climate model, Atmos. Chem. Phys., 12,
 961-987, 2012.
- Pringle, K. J., Tost, H., Message, S., Steil, B., Giannadaki, D., Nenes, A., Fountoukis,
 C., Stier, P., Vignati, E., and Leieved, J.: Description and evaluation of GMXe: a
 new aerosol submodel for global simulations (v1), Geoscientific Model
 Development, 3, 391-412, 2010.
- Pye, H. O. T. and Seinfeld, J. H.: A global perspective on aerosol from low-volatility
 organic compounds, Atmos. Chem. Phys., 10, 4377-4401, 2010.
- Robinson, A. L., Donahue, N. M., Shrivastava, M. K., Weitkamp, E. A., Sage, A. M.,
 Grieshop, A. P., Lane, T. E., Pierce, J. R., and Pandis, S. N.: Rethinking organic

- aerosols: Semivolatile emissions and photochemical aging, Science, 315, 12591262, 2007.
- Robinson, A. L., Grieshop, A. P., Donahue, N. M., and Hunt, S. W.: Updating the
 conceptual model for fine particle mass emissions from combustion systems, J.
 Air Waste Manage., 60, 1204-1222, 2010.
- Sander, R., Baumgaertner, A., Gromov, S., Harder, H., Joeckel, P., Kerkweg, A.,
 Kubistin, D., Regelin, E., Riede, H., Sandu, A., Taraborrelli, D., Tost, H., and
 Xie, Z. Q.: The atmospheric chemistry box model CAABA/MECCA-3.0,
 Geoscientific Model Development, 4, 373-380, 2011.
- Shrivastava, M., Fast, J., Easter, R., Gustafson, W. I., Jr., Zaveri, R. A., Jimenez, J. L.,
 Saide, P., and Hodzic, A.: Modeling organic aerosols in a megacity: comparison
 of simple and complex representations of the volatility basis set approach, Atmos.
 Chem. Phys., 11, 6639-6662, 2011.
- Shrivastava, M. K., Lane, T. E., Donahue, N. M., Pandis, S. N., and Robinson, A. L.:
 Effects of gas particle partitioning and aging of primary emissions on urban and regional organic aerosol concentrations, J. Geophys. Res.-Atmos., 113, 2008.
- Tost, H., Jöckel, P., Kerkweg, A., Sander, R., and Lelieveld, J.: Technical note: A
 new comprehensive SCAVenging submodel for global atmospheric chemistry
 modelling, Atmos. Chem. Phys., 6, 565-574, 2006.
- Tsimpidi, A. P., Karydis, V. A., Pandis, S. N., and Lelieveld, J.: Global combustion
 sources of organic aerosols: Model comparison with 84 AMS factor analysis data
 sets, Atmos. Chem. Phys. Discuss., 2016, 1-51, 2016a.
- Tsimpidi, A. P., Karydis, V. A., Pandis, S. N., and Lelieveld, J.: Global combustion
 sources of organic aerosols: model comparison with 84 AMS factor-analysis data
 sets, Atmos. Chem. Phys., 16, 8939-8962, 2016b.
- Tsimpidi, A. P., Karydis, V. A., Pozzer, A., Pandis, S. N., and Lelieveld, J.: ORACLE
 (v1.0): module to simulate the organic aerosol composition and evolution in the
 atmosphere, Geoscientific Model Development, 7, 3153-3172, 2014.
- Tsimpidi, A. P., Karydis, V. A., Zavala, M., Lei, W., Bei, N., Molina, L., and Pandis,
 S. N.: Sources and production of organic aerosol in Mexico City: insights from
 the combination of a chemical transport model (PMCAMx-2008) and
 measurements during MILAGRO, Atmos. Chem. Phys., 11, 5153-5168, 2011.
- Tsimpidi, A. P., Karydis, V. A., Zavala, M., Lei, W., Molina, L., Ulbrich, I. M.,
 Jimenez, J. L., and Pandis, S. N.: Evaluation of the volatility basis-set approach
 for the simulation of organic aerosol formation in the Mexico City metropolitan
 area, Atmos. Chem. Phys., 10, 525-546, 2010.
- van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla,
 P. S., Morton, D. C., DeFries, R. S., Jin, Y., and van Leeuwen, T. T.: Global fire
 emissions and the contribution of deforestation, savanna, forest, agricultural, and
 peat fires (1997-2009), Atmos. Chem. Phys., 10, 11707-11735, 2010.
- Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Ulbrich, I. M., Ng, N. L., Worsnop, D.
 R., and Sun, Y. L.: Understanding atmospheric organic aerosols via factor
 analysis of aerosol mass spectrometry: a review, Anal. Bioanal. Chem., 401, 3045-3067, 2011.
- Zhang, Q. J., Beekmann, M., Drewnick, F., Freutel, F., Schneider, J., Crippa, M.,
 Prevot, A. S. H., Baltensperger, U., Poulain, L., Wiedensohler, A., Sciare, J.,
 Gros, V., Borbon, A., Colomb, A., Michoud, V., Doussin, J. F., van der Gon, H.
 A. C. D., Haeffelin, M., Dupont, J. C., Siour, G., Petetin, H., Bessagnet, B.,
 Pandis, S. N., Hodzic, A., Sanchez, O., Honore, C., and Perrussel, O.: Formation

- 937 of organic aerosol in the Paris region during the MEGAPOLI summer campaign:
 938 evaluation of the volatility-basis-set approach within the CHIMERE model,
 939 Atmos. Chem. Phys., 13, 5767-5790, 2013.
- Zhang, X., Cappa, C. D., Jathar, S. H., McVay, R. C., Ensberg, J. J., Kleeman, M. J.,
 and Seinfeld, J. H.: Influence of vapor wall loss in laboratory chambers on yields
 of secondary organic aerosol, Proceedings of the National Academy of Sciences
 of the United States of America, 111, 5802-5807, 2014.

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Table 1. Parameters used in the sensitivity simulations

Simulation Emission factor		Emission rate (Tg yr ⁻¹)		Volatility bins	Reduction in volatility (µg m ⁻³)	Stoichiometric coefficient of aging reactions	Oxidation rate constant (cm ³ molec ⁻¹ s ⁻¹)	Henry's law constant (mol L ⁻¹ atm ⁻¹)		
	fPOA	bbPOA	fPOA	bbPOA					Freshly emitted	Aged
Reference	2.5	1	44.2	28.4	5	10 ²	1.15	2x10 ⁻¹¹	105	10 ⁵
Low volatility	1	1	17.7	28.4	5	10 ²	1.15	2x10 ⁻¹¹	10 ⁵	10 ⁵
High IVOCs	4	2.5	70.7	71	5	10 ²	1.15	2x10 ⁻¹¹	10 ⁵	10 ⁵
Alternative POA emissions	2.5	1	28.5	37.8	5	10 ²	1.15	2x10 ⁻¹¹	10 ⁵	10 ⁵
High reaction rate constant	2.5	1	44.2	28.4	5	10 ²	1.15	4x10 ⁻¹¹	10 ⁵	10 ⁵
Alternative aging scheme	2.5	1	44.2	28.4	9	10	1.075	4x10 ⁻¹¹	10 ⁵	10 ⁵
Hybrid aging scheme	2.5	1	44.2	28.4	5	SVOCs:10 ² IVOCs:10 ⁴ -10 ⁶	SVOCs:1.15 IVOCs:1.115-0.71	SVOCs:2x10 ⁻¹¹ IVOCs:1.2x10 ⁻¹¹	10 ⁵	10 ⁵
Low solubility	2.5	1	44.2	28.4	5	10 ²	1.15	2x10 ⁻¹¹	10 ³	10 ³
Variable solubility	2.5	1	44.2	28.4	5	10 ²	1.15	2x10 ⁻¹¹	10	10 ⁵

949 Table 2. Statistical evaluation of EMAC POA (sum of fPOA and bbPOA) against
950 AMS POA (sum of HOA and BBOA) using 61 data sets in urban downwind and rural
951 areas during 2001-2010.

	Mean	Mean	MAGE	MB	NME	NMB	RMSE
Simulation Name	Observed	Predicted	(µg m ⁻³)	(µg m ⁻³)	(%)	(%)	(µg m ⁻³)
	(µg m ⁻³)	(µg m ⁻³)					
Reference		0.51	0.38	-0.02	71	-3	0.50
Low volatility		0.75	0.46	0.22	88	43	0.64
High IVOCs		0.52	0.38	-0.01	73	0	0.51
Alternative POA emissions		0.39	0.33	-0.14	63	-25	0.44
High reaction rate constant	0.53	0.50	0.37	-0.03	70	-5	0.49
Conservative aging scheme		0.17	0.42	-0.36	79	-67	0.60
Hybrid aging scheme		0.50	0.38	-0.03	72	-4	0.50
Low solubility		0.53	0.38	0	72	1	0.50
Variable solubility		0.54	0.38	0.01	73	4	0.51

Table 3. Statistical evaluation of EMAC SOA against AMS OOA using 61 data sets

956 in downwind urban and rural areas during 2001-2010.

	Mean	Mean	MAGE	MB	NME	NMB	RMSE
Simulation Name	Observed	Predicted	(µg m ⁻³)	(µg m ⁻³)	(%)	(%)	(µg m ⁻³)
	(µg m ⁻³)	(µg m ⁻³)					
Reference		1.91	1.39	-0.87	50	-31	2.02
Low volatility		1.32	1.69	-1.46	61	-52	2.30
High IVOCs	2.78	2.50	1.47	-0.28	53	-10	2.05
Alternative POA emissions		1.66	1.55	-1.12	56	-40	2.15
High reaction rate constant		2.16	1.32	-0.62	48	-22	1.97
Conservative aging scheme		1.73	1.49	-1.05	53	-38	2.09
Hybrid aging scheme		1.71	1.46	-1.08	53	-39	2.08
Low solubility		2.10	1.33	-0.68	48	-25	1.98
Variable solubility		2.14	1.32	-0.64	48	-23	1.97

Table 4. Percentage change of the tropospheric burden of organic aerosol components for
each sensitivity simulation relative to the reference simulation during the decade 2001-2010.
Positive change corresponds to an increase. The predicted tropospheric burden in Tg of the
reference simulation is also shown.

	fPOA	bbPOA	fSOA-sv	bbSOA-sv	fSOA-iv	bbSOA-iv	Total OA
Tropospheric burden of reference (Tg)	0.06	0.18	0.13	0.21	0.44	0.2	1.98
			Percentage	e Change (%)	from refere	nce	
Simulation Name							
Low volatility	53	48	14	39	-100	-100	-23
High IVOCs	7	5	-3	-4	88	165	38
Alternative POA emissions	-39	10	-33	11	-34	11	-8
High reaction rate constant	-10	-7	11	11	8	6	4
Alternative aging scheme	-65	-38	-68	-47	14	30	-10
Hybrid aging scheme	-2	-1	2	2	-37	-36	-13
Low solubility	6	1	11	4	21	8	8
Variable solubility	9	2	14	5	22	7	8
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965							
966							
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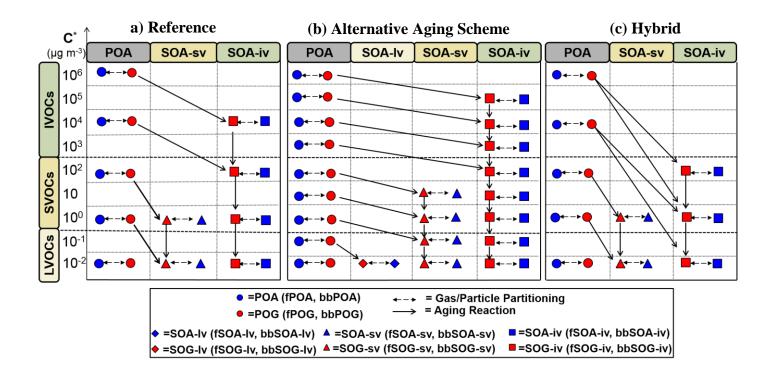


Figure 1: Schematic of the VBS resolution and the formation of SOA from SVOCs and IVOCs in the: (a) reference simulation, (b) alternative aging scheme and (c) hybrid case. SOA from LVOCs (SOA-lv) is only formed in the alternative aging scheme (b). Red indicates that the organic compound is in the vapor phase and blue in the particulate phase. The circles correspond to primary organics emitted as gases or particles. Diamonds symbolize the formation of SOA from LVOC emissions by fuel combustion and biomass burning. Triangles indicate SOA formation from SVOC emissions by fuel combustion and biomass burning, while the squares show SOA from IVOC by the same sources. Gas-aerosol partitioning, aging reactions, and names of species are also shown.

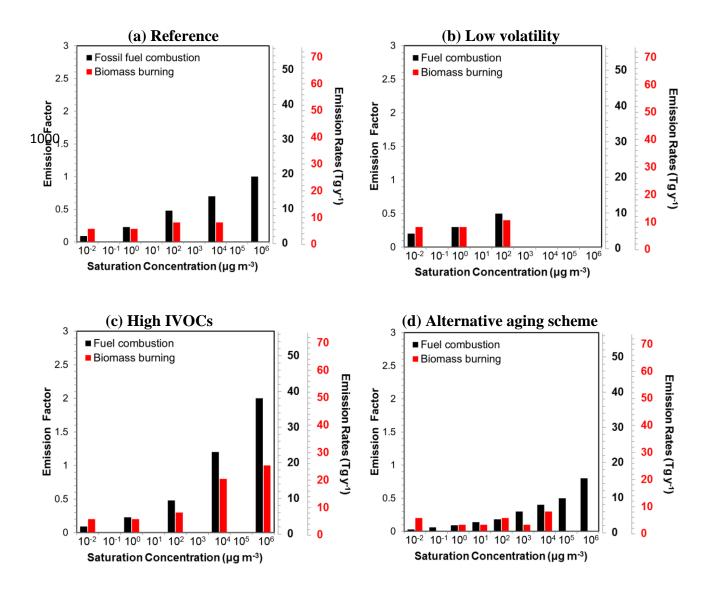


Figure 2. Volatility distribution for fuel combustion (black) and biomass burning OA (red) for the (a) reference, (b) low volatility, (c) high IVOCs and (d) conservative aging scheme simulations. The reference emission factors are from Robinson et al. (2007) for fPOA and May et al. (2013) for bbPOA emissions. The emission rates of fPOA and bbPOA are also shown on the right axis.

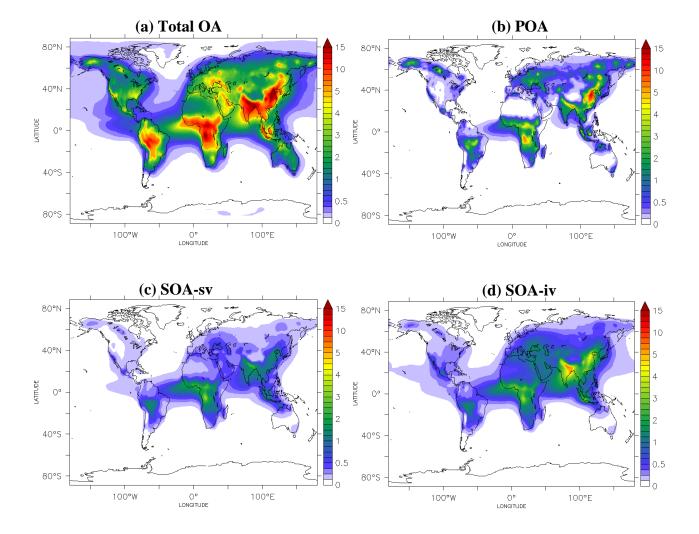


Figure 3: Predicted average surface concentrations (in μ g m⁻³) of: (a) Total OA (sum of POA, SOA-sv, SOA-iv and SOA-v), (b) POA and (c) SOA from the oxidation of SVOCs (SOA-sv) and (d) SOA from the oxidation of IVOCs (SOA-iv) for the reference simulation during the 2001-2010 period.

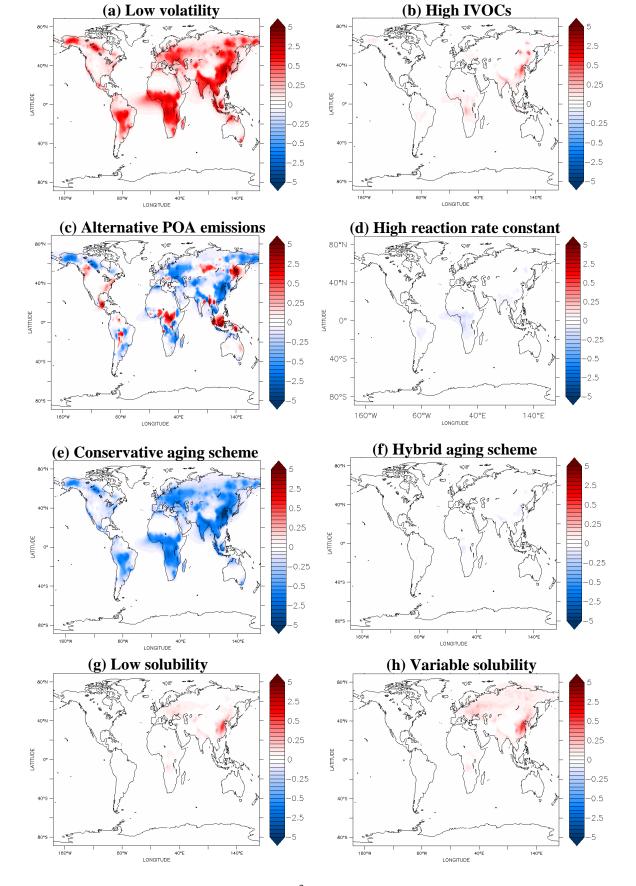


Figure 4: Absolute changes (in μ g m⁻³) of the average surface POA concentrations between the reference and the (a) low volatility, (b) high IVOCs, (c) alternative POA emissions, (d) high reaction rate constant, (e) conservative aging scheme, (f) hybrid aging scheme, (g) low solubility, and, (h) hybrid solubility simulations during the period 2001-2010. A positive change indicates an increase in the sensitivity test.

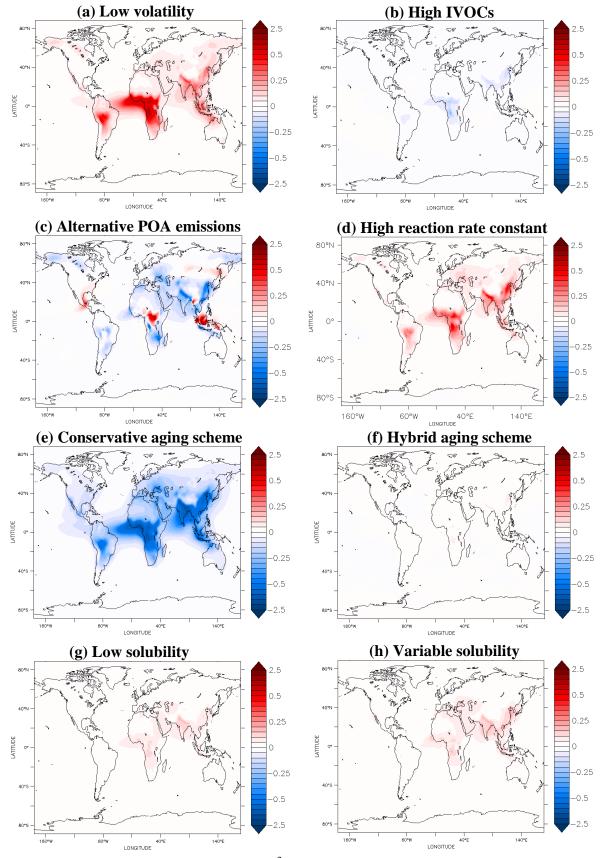


Figure 5: Absolute changes (in μ g m⁻³) of the average surface SOA concentrations from SVOCs (SOA-sv) between the reference and the (**a**) low volatility, (**b**) high IVOCs, (**c**) alternative POA emissions, (**d**) high reaction rate constant, (**e**) conservative aging scheme, (**f**) hybrid aging scheme, (**g**) low solubility, and, (**h**) hybrid solubility simulations during the period 2001-2010. A positive change indicates an increase in the sensitivity test.

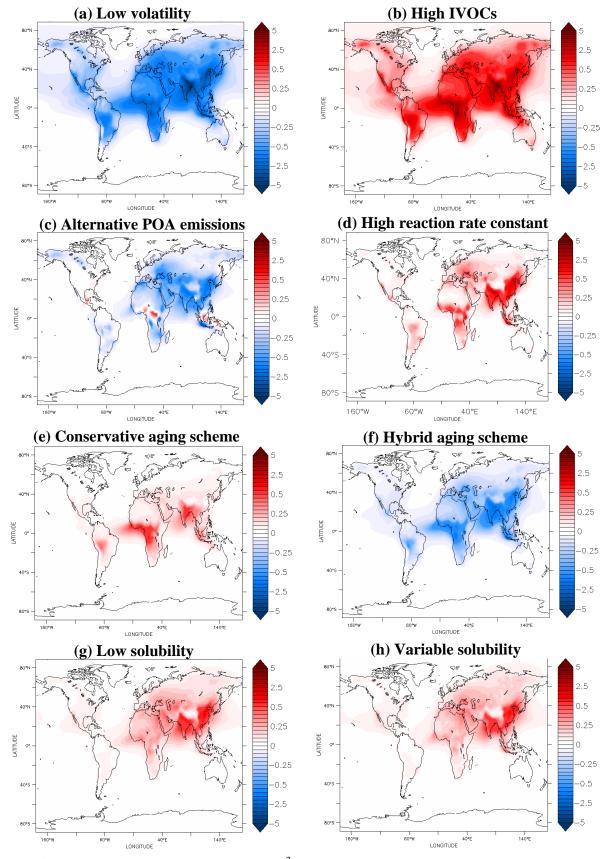


Figure 6: Absolute changes (in μ g m⁻³) of the average surface SOA concentrations from IVOCs (SOA-iv) between the reference and the (**a**) low volatility, (**b**) high IVOCs, (**c**) alternative POA emissions, (**d**) high reaction rate constant, (**e**) conservative aging scheme, (**f**) hybrid aging scheme, (**g**) low solubility, and, (**h**) hybrid solubility simulations during the period 2001-2010. A positive change indicates an increase in the sensitivity test.

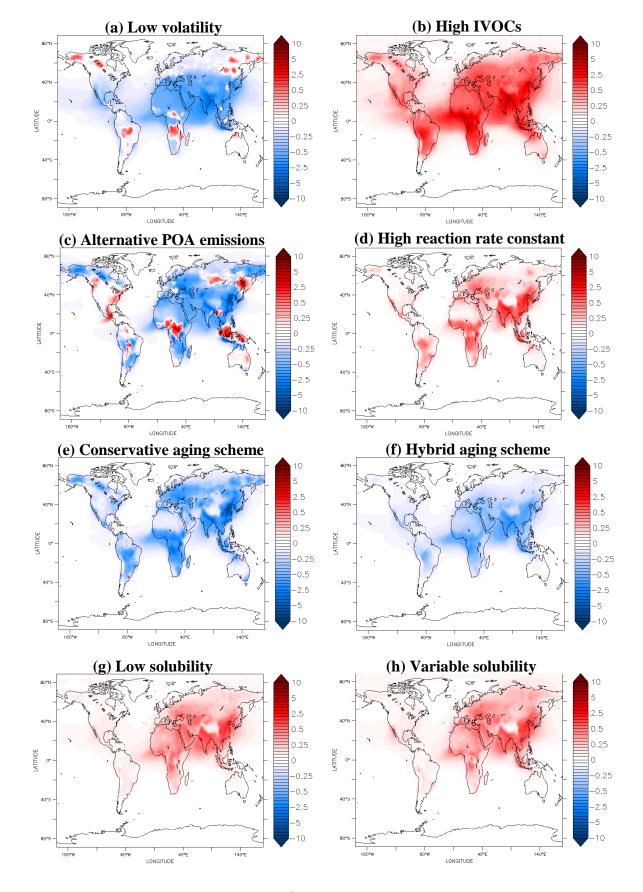


Figure 7: Absolute changes (in μ g m⁻³) of the average surface total OA concentrations between the reference and the (a) low volatility, (b) high IVOCs, (c) alternative POA emissions, (d) high reaction rate constant, (e) conservative aging scheme, (f) hybrid aging scheme, (g) low solubility, and, (h) hybrid solubility simulations during the period 2001-2010. A positive change indicates an increase in the sensitivity test.

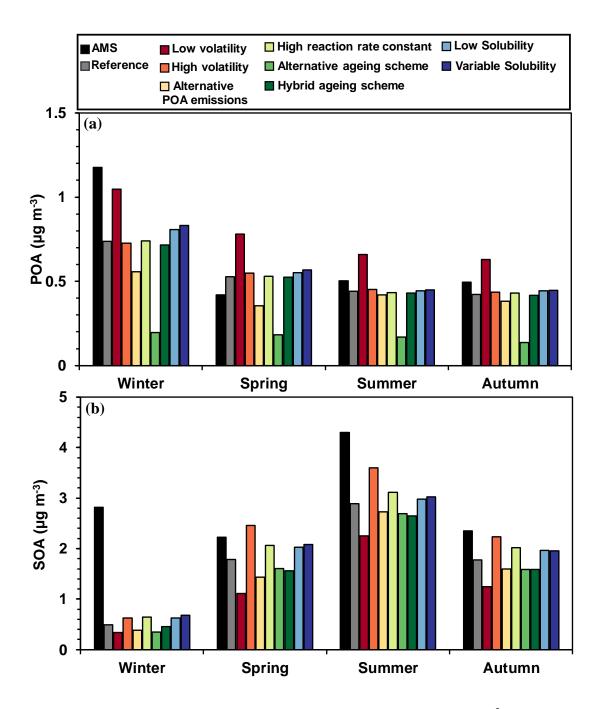


Figure 8: Average (a) POA and (b) SOA concentrations (in μ g m⁻³) measured and predicted in the reference and sensitivity simulations during winter, spring, summer, and autumn in urban-downwind and rural areas of the continental Northern Hemisphere.