1	Chemical characteristics of submicron particles at the central
2	Tibet Plateau: insights from aerosol mass spectrometry
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17	Abstract

18 Recent studies have revealed a significant influx of anthropogenic aerosol from south Asia to Himalayas 19 and Tibet Plateau (TP) during pre-monsoon period. In order to characterize the chemical composition, 20 sources, and transport processes of aerosol in this area, we carried out a field study during June 2015 by 21 deploying a suite of online instruments including an Aerodyne high-resolution time-of-flight aerosol mass 22 spectrometer (HR-AMS) and a multi-angle absorption photometer (MAAP) at Nam Co Station (90°57'E, 23 30°46'N, 4730m a.s.l) at the central of the TP. The measurements were made at a period when the 24 transition from pre-monsoon to monsoon occurred. The average ambient mass concentration of submicron particulate matter (PM₁) over the whole campaign was ~2.0 μ g m⁻³, with organics accounting for 68%, 25 26 followed by sulfate (15%), black carbon (8%), ammonium (7%), and nitrate (2%). Relative higher aerosol 27 mass concentration episodes were observed during the pre-monsoon period, whereas persistently low 28 aerosol concentrations were observed during the monsoon period. However, the chemical composition of 29 aerosol during the higher aerosol concentration episodes in the pre-monsoon season was on a case-by-case 30 basis, depending on the prevailing meteorological conditions and air mass transport routes. Most of the 31 chemical species exhibited significant diurnal variations with higher values occurring during afternoon and 32 lower values during early morning, whereas nitrate peaked during early morning in association with higher 33 relative humidity and lower air temperature. Organic aerosol (OA) was more oxidized with an oxygen-to-34 carbon ratio (O/C) of 0.94 during the pre-monsoon period than during monsoon (average O/C of 0.72), and 35 an average O/C was 0.88 over the entire campaign period suggesting overall highly oxygenated aerosol in

36 the central TP. Positive matrix factorization of the high resolution mass spectra of OA identified two

37 oxygenated organic aerosol (OOA) factors: a less oxidized OOA (LO-OOA) and a more oxidized OOA

- 38 (MO-OOA). The MO-OOA dominated during the pre-monsoon period, whereas LO-OOA dominated
- during monsoon. The sensitivity of air mass transport during pre-monsoon with synoptic process was also
- 40 evaluated with a 3-D chemical transport model.
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42 **1. Introduction**

The Tibet Plateau (TP) and Himalayas is a vast and elevated highland in Central Asia that extends over the area of 27-45°N, 70-105°E with a mean elevation of more than 4000 m above sea level (a.s.l.). It is a sparsely populated area with minimal local pollution. The TP is an ideal area for observations of free tropospheric air masses and aerosol transported from polluted areas surrounding the TP after long distances. Determination of the chemical characteristics of aerosol particles in the TP is important for assessments of their influences on atmospheric chemistry and climate (Li et al., 2016), which are so far poorly understood due to harsh conditions and logistic limitations.

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51 Over recent decades, an increasing number of field studies have been conducted in these regions to 52 characterize aerosol physical and chemical features from mountain observatories, e.g., the Nepal Climate 53 Observatory-Pyramid (5079 m), which is set up for long-term monitoring and synchronous observation 54 (Bonasoni et al., 2010; Liu et al., 2017). There are significant seasonal variations in aerosol mass loading in 55 the southern TP and Himalayas. Higher aerosol concentration was often found during pre-monsoon due to 56 less precipitation and favorable atmospheric circulation (Bonasoni et al., 2010; Marinoni et al., 2010; 57 Marinoni et al., 2013; Zhao et al., 2013). For example, the concentration of carbonaceous species at the 58 Qomolangma (Mt. Everest) Station (4276 m a.s.l.), northern Himalayas during pre-monsoon was found to 59 be 3 - 5 times higher than that during the monsoon periods (Cong et al., 2015). Since this seasonal 60 variation of aerosol loading is consistent at both the southern and northern Himalayas (Xu et al., 2014a), 61 aerosol plume was thought to be able to across the Himalayas, a finding which is also supported by model 62 results (Lu et al., 2012; Lüthi et al., 2015; Zhang et al., 2015). During monsoon period, ambient aerosol 63 from the upwind sources is significantly scavenged during long-range transport and air mass is mainly 64 originated from marine area, which can lead to aerosol chemical differences between pre-monsoon and 65 monsoon. The distinct seasonal variation of aerosol loading is mainly attributed to the change of the 66 dominated climatic systems and then weather conditions. During pre-monsoon, the cold and dry southern 67 Westerliers dominated the southern TP and Himalayas, while the South Asia Monsoon covers most of 68 South Asia, Himalayas, and the southern TP during summer period.

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Most of the studies conducted in these regions focused on some specific species, such as black carbon,
which has strong light absorption. Burning of biomass fuels and wildfires in the south Himalaya and South
Asia are thought to be important sources for the black carbon (Stone et al., 2010; Engling et al., 2011;

73 Kumar et al., 2011). However, a recent study shows that sources of black carbon in the region of South 74 Asia are highly complex, including emissions from low efficient transport tools and cooking using cow 75 dung and biogas as well (Stockwell et al., 2016). These burning activities also emit other species such as 76 organic and inorganic particulate species and volatile organic compounds, which generate a well-mixed 77 aerosol plume eventually via processes such as coagulation, evaporation, oxidation, and condensation. Fang 78 et al. (2015) recently suspected that biogenic aerosol could also be an important contribution for aerosol in 79 the TP during summer. However, these mixed plumes have been rarely, if ever, characterized by 80 comprehensive field measurements.

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82 Filter-based sampling method with a low time resolution (days) were widely adopt in these remote regions 83 due to logistical difficulties with deployment of real-time instruments. The low time resolution made the 84 understanding of chemical processes of aerosol during transport challenging. Secondary species such as 85 sulfate and water soluble organic carbon (WSOC) are normally the dominated species in aerosol. For 86 example, WSOC accounted for about 60% of OC and the ratio between OC and EC could be up to 10 87 (Zhao et al., 2013; Cong et al., 2015), suggesting a dominant contribution of secondary organic aerosol 88 (SOA) to carbonaceous aerosol loading in the TP. The ambient conditions at high elevation regions are 89 characterized by higher solar radiation and concentrations of oxidants such as O₃, which makes 90 photochemical processing in this high elevation remote region intense and likely dominant. High-time 91 resolution measurement is thus necessary in this region for detecting short-term events and the evolution of 92 pollutants. In addition, the high time resolution data are useful for constraining atmospheric chemical 93 transport models.

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The aerodyne aerosol mass spectrometer (AMS) has been widely used to study the chemical composition of 95 96 non-refractory submicron particle (e.g., Xu et al., 2014b; Xu et al., 2016). There are two merits of the AMS 97 including its high time resolution and bulk measurement. The high time resolution is usually in minutes 98 which has advanced our understanding of fundamental chemical processes of fine aerosols at different 99 regions of the world (Canagaratna et al., 2007). The bulk measurements enable the observation to obtain 100 rich information of various aerosol chemical species simutaneously. The organic aerosol, which is often the 101 most important component of aerosol, can be further analyzed to determine the average elemental ratios 102 and by positive matrix factorization (PMF) analysis to determine the sources and atmospheric processes 103 (Ulbrich et al., 2009; Zhang et al., 2011). The atomic elemental ratios of oxygen and hydrogen to carbon 104 (O/C and H/C) calculated from the OA mass spectra can provide information about the sources and evolution processes of OA in the atmosphere (Aiken et al., 2008; Heald et al., 2010; Kroll et al., 2011b; Ng 105 106 et al., 2011). They also often closely correlate with key OA properties such as hygroscopicity, density, and 107 phase separation (Jimenez et al., 2009; Bertram et al., 2011). In addition, due to high sensitivity and low 108 detection limits, AMS has been successfully deployed at many remote sites with low aerosol mass loading 109 such as Antarctica (Schmale et al., 2013; Giordano et al., 2017).

111 The study here presents results from measurements using an AMS at the central TP during the transitional 112 period from pre-monsoon to monsoon. The study was designed to characterize aerosol chemical 113 composition, temporal variations, transport processes, and emission sources. During the campaign, besides 114 the AMS, multiple other real-time instruments were also deployed.

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116 **2. Methodology**

117 **2.1 Site description**

118 The field study was conducted between May 31 and July 1, 2015 at a high altitude observatory, i.e., Nam 119 Co Station for Multisphere Observation and Research, Chinese Academy of Sciences (Nam Co station, 120 90°57′E, 30°46′N 4730m a.s.l), at the central part of the TP (Figure 1). The Nam Co station is located 121 nearby Nam Co lake (Figure 1b), the second largest inland lake in the TP (area: 1920 km²) which is located 122 at the northern border of Nyainqentanglha Mountains. The melted glacier from Nyainqentanglha Mountains 123 supply water to the lake each year during warm seasons. The Nam Co station and its surrounding is a 124 pristine region except for a small village for local people that is about 10 km west to the station with a 125 population of 300 to 500. In the past several years, tourism for this beautiful lake has grown. A highway for 126 tourists was built about 3 km south of the station with a low traffic intensity (less than 300 cars per day) 127 during June. The capital city (Lhasa) of the Tibet Autonomous Region is about 100 km southeast of the 128 station with an average elevation of 3600 m a.s.l. between which is Nyainqentanglha Mountains (higher 129 than 6000 m a.s.l) (Figure 1c). The closest town, Dangxiong, is alongside the famous highway, Qinghai-130 Tibet Highway, and about 70 km east of the station with an average elevation of 4200 m a.s.l. Overall, the 131 station is surrounded by the mountains in the south and east, and the lake at the west. The ecology of 132 surrounding area is semi-arid land dominated by alpine meadow and barren areas. The precipitation is 133 mainly occurred during summer monsoon period. The cooking and heating at the station is supplied by the 134 power and natural gas.

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136 **2.2 Instrument setup**

137 The study was conducted at the observatory field of the station using a customer-made trailer with inlet 138 stepped out of the top with the height of ~5m above ground. All the instruments were arranged inside the 139 trailer where the air temperature was controlled at $\sim 20^{\circ}$ C by two air conditioners. The air flow of inlet was 140 maintained by a vacuum pump with a flow rate of 10 Lmin^{-1} , and several instruments were taking samples 141 from this inlet including a high-resolution time-of-flight aerosol mass spectrometer (HR-AMS, Aerodyne 142 Research Inc., Billerica, MA, USA), a custom-made scanning mobility particle sizer (SMPS), a soot 143 particle Aerodyne mass spectrometer (SP-AMS, Aerodyne Research Inc., Billerica, MA, USA), a single-144 particle intracavity laser-induced incandescence photometer (SP2, DMT, Inc., Boulder, CO, USA), and a 145 seven wavelength Athelometer (model AE31, Magee Scientific, Berkeley, CA, USA). The total flow rate of

146 the inlet was maintained at ~16 L min⁻¹ and a $PM_{2.5}$ cyclone was used in the front of the inlet (model URG-

2000-30EH, URG Corp., Chapel Hill, NC, USA) to remove coarse particles. Several additional instruments
were also co-located but with their own inlets, such as a Nephelometer (model 3563, TSI, Shoreview, MN,
USA), a PM_{2.5} sampler (model PQ200, BGI, USA), and a multi-angle absorption photometer (MAAP,
model 5012, Thermo Electron Cooperation, USA). The meteorological data was recorded at the tower of
the Nam Co station at a height of 20 m above ground.

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153 HR-AMS was used to measure the non-refractory PM_1 (NR-PM₁), which was thermally vaporized at 154 ~600 °C, ionized with a 70eV electron impact and determined by using a time-of-flight mass spectrometry. 155 The details of the instrument have been described elsewhere (DeCarlo et al., 2006). For our study, HR-156 AMS was only operated with V-mode with 5 min resolution due to the low aerosol mass loading. Due to 157 malfunction of the chopper, size distribution of NR-PM₁ was not determined. The HR-AMS was calibrated 158 for ionization efficiency (IE) with the ammonium nitrate following standard procedures (Jayne et al., 2000) 159 at the beginning, in the middle and end of the study. Particulate-free air was sampled twice during the study 160 period to adjust the influences of air on the fragmentation table and determine the detection limits (DLs) of 161 aerosol species. The DLs were calculated as 3 times the standard deviations (3σ) of the measured values 162 during the particle-free sampling periods. The 2.5-min DLs for organic, sulfate, nitrate, ammonium, and chloride were determined at 0.108, 0.014, 0.007, 0.002, 0.010 μ g m⁻³, which are comparable to the values 163 164 reported in previous studies.

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166 **2.3 Data processing**

167 The HR-ToF-AMS data were processed using the standard software of SQUIRREL (v1.56) and PIKA 168 (v1.15c) written in IGOR (Wavemetrics, Inc., Lake Oswego, OR, USA) (http://cires.colorado.edu/jimenez-169 group/ToFAMSResources/ToFSoftware/index.html) to determine the mass concentrations of the NR-PM₁ 170 species and the ion-speciated mass spectra of organics. An empirical particle collection efficiency (CE) of 171 0.5 was used, which has been widely used in field studies employing AMS with a dryer installed in front of 172 the equipment's particle inlet. This CE value was further validated by the consistency and reasonable slope 173 between HR-AMS measured mass concentrations and SMPS-determined particle volumes (section 3.1.2, 174 $R^2 = 0.9$, slope = 1.48). The elemental ratios of OA (O/C, H/C, and organic matter to carbon (OM/OC)) in 175 this study was determined using the "improved-ambient" method (Canagaratna et al., 2015). Default 176 relative ionization efficiency (RIE) values were used for organics (1.4), nitrate (1.1), and chloride (1.3), 177 while an RIE value of 3.8 was determined for ammonium and 1.1 for sulfate based on the calibration for 178 pure NH₄NO₃ and (NH₄)₂SO₄, respectively.

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180 The source apportionment of organics was conducted by PMF with the robust engine. First, organic matrix 181 was analysed using the PMF2.exe algorithm in robust mode (Paatero and Tapper, 1994) and explored using 182 the PMF Evaluation Toolkit (PET) (Ulbrich et al., 2009). The PMF solution was evaluated following the 183 procedures outlined in Table 1 of Zhang et al. (2011) including modification of the error matrix and downweight of low signal-to-noise ions. Moreover, based on the AMS fragmentation table, some organic ions were not directly measured but scaled to the organic signal at m/z 44, which were downweighted by increasing their errors by a factor of 3. A two-factor solution with fPeak = 0 was chosen in this study, as it is able to reconstruct the total OA mass and temporal profiles very well. The results of three-factor solution

- 188 with fPeak = 0 are shown in supplementary material (Figure S1), which show splitting of the solutions.
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190 **3 Results and discussions**

3.1 The meteorological conditions during the study

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193 During the field study, the meteorological conditions were generally cold and windy (Figure 2a and b). The 194 average air temperature was 8.4 ± 3.6 °C with a diurnal hourly average ranging from 4.8 to 12.3 °C; the diurnal average wind speed (WS) ranged from 4.5 to 6.5 m s⁻¹. The dominant wind directions were south 195 196 and southwest, although they varied during different periods (Figure 2b and S2). The wind direction (WD) 197 had distinct diurnal variation with air mass from south during night to morning and west during afternoon. 198 Based on the report of the Climate Diagnostics Bulletin of India 199 (http://imdpune.gov.in/Clim_RCC_LRF/Climate_Diagnostic_Bulletins/cdbi_apr_2015.pdf), the 200 precipitation from Indian monsoon had covered most part of India on June 12 to June 14, and these days 201 are treated as the onset of South Asian monsoon. Indeed, precipitation was recorded at the Nam Co station 202 on June 13 and lasted for several days (Figure 2a). Based on this weather condition, the period before June 203 14 was classified as the pre-monsoon period and afterwards as the monsoon period. The RH during the pre-204 monsoon and monsoon periods were $48.4 \pm 19.2\%$ and $58.8 \pm 16.5\%$, respectively. The origins of the air 205 masses were also different during these two periods. Figure 3 shows the air mass from west during pre-206 monsoon period accounted about 30-40%, while it was only 7% during monsoon period, of which ~80% 207 was from south.

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3.2 The temporal variations of chemical species

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The average mass concentration of PM₁ (NR-PM₁ + BC) was 2.0 μ g m⁻³ during the whole study period 211 212 with 68% OA, 15% sulfate, 2% nitrate, 7% ammonium, and 8% BC. The NR-PM₁ mass loading (1.84 µg m⁻³) at Nam Co Station was lower than the values observed at some high elevation sites such as Menyuan 213 214 $(10.8 \ \mu g \ m^{-3})$ (Du et al., 2015), Montsec (4.9 \ \mu g \ m^{-3}) (Ripoll et al., 2015), Mt. Cimone (4.5 \ \mu g \ m^{-3}) 215 (Rinaldi et al., 2015), Puy de Done $(5 - 27 \ \mu g \ m^{-3})$, and Mt. Bachelor (15.1 $\ \mu g \ m^{-3})$ (Zhou et al., 2017), but higher than those in some other sites, such as sub-Antarctic (0.46 μ g m⁻³) (Schmale et al., 2013) and 216 Jungfraujoch (0.55 μ g m⁻³) (Fröhlich et al., 2015), similar with that at Mt. Whistler (1.91 μ g m⁻³) (Sun et 217 218 al., 2009) (Table 1 and Figure S3). Although the sampling seasons and altitudes of these sampling sites are 219 different (Table 1), the results of these studies could basically represent the level of aerosol at these sits due 220 to their relatively long sampling spans (from one month to ten months). The contribution of OA at these 221 high-elevation sites ranged from 50% - 90%; the highest value was at Mt. Bachelor which was frequently 222 influenced by transported biomass burning plume, while the relative low OA contribution (38%) at sub-223 Antarctic was due to the dominant source from marine emission containing high sulfate. The mass 224 concentration of PM₁ varied dynamically during our study with distinct difference between pre-monsoon 225 and monsoon periods. The average mass concentrations of PM_1 for these two periods were 2.6 and 1.2 μg 226 m^{-3} , respectively. For comparison, the concentrations of various species for these two periods were side-by-227 side displayed in Figure 4a. The ratios between pre-monsoon and monsoon for all the species were higher 228 than 1 with the maxima for ammonium (3.1) and sulfate (2.8); the contribution of OA was thus slightly 229 higher during the monsoon period than during the pre-monsoon (71% vs. 64% with a mass concentration 230 ratio of 1.9). Apart from the potential scavenging effect, these results could also be influenced by the 231 sources, transport route, and chemical processes during different periods.

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233 Based on the mass concentration and temporal variations of PM_1 species and weather conditions, the pre-234 monsoon period could be further divided into two periods, i.e., period 1 (P1: May 30 to June 7) and period 235 2 (P2: June 8 to June 13). P1 was characterized by relatively high sulfate concentration (0.46 vs. 0.41 μ g m⁻ ³) and sunny days, while P2 was characterized by high PM₁ concentration (2.9 vs. 2.1 μ g m⁻³; P< 0.005, 236 237 chi-square test), high nitrate contribution, and wet and cold days (Figure S4). The air masses for P2 had 238 higher contribution from west and north than those during P1 (Figure 3). Figure 4b shows the comparisons 239 of mass concentration of different species between these two periods. For chemical species, from P1 to P2, 240 nitrate and OA increased dramatically by a factor of 1.6 and 2.2, respectively, and BC and ammonium also 241 increased, suggesting the influence of elevated air plume. However, sulfate and chloride decreased slightly 242 (the ratios between P2 and P1 were ~0.95). The variations of sulfate and nitrate during these two periods 243 could be related to the photochemical conditions and origination of air mass (see section 3.3).

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245 The particles were generally neutralized as illustrated by the scatter plot between predicted and measured ammonium (slope = 0.91) (Figure S5a). The neutralized PM₁ were likely due to the high availability of 246 247 ammonia from agriculture emission in North India (Clarisse et al., 2009; Van Damme et al., 2015). The 248 slope was ~0.75 during P2 (green circles, Figure S5a), which suggests that there was more ammonium than 249 needed to neutralize sulfate and nitrate. We checked whether the high ammonium was from fragmentation 250 of organic nitrogen such as amines. The ratios of NH^+ to NH_2^+ were consistently during the study (Figure 251 S5b), suggesting there was no significant influences from amino compounds. This excess ammonium 252 determined by the neutralization could be related to the presence of significant amounts of organic anions 253 in aerosol (such as carboxylic acids) or variation of RIE for ammonium which could have higher value in 254 the mixed acidic particles.

3.3 Diurnal variation and chemical process of different species

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258 The diurnal cycles of OA, sulfate, nitrate, ammonium, and BC during different periods are shown in Figure 259 5. All these species unexpectedly present dramatic diurnal variations, especially during P2. OA, sulfate, 260 ammonium, and BC showed a similar pattern with low values during nighttime to early morning and high 261 values during afternoon which suggest their common sources or similar transport pathways (Zhang et al., 262 2001). We checked the diurnal variation of the origination of air masses and found that there were 263 increased air masses from south during nighttime and from west during afternoon (Figure S6). This 264 phenomenon could be related to the plateau monsoon during summer, which the strong heating at the 265 surface resulted in a shallow cyclonic circulation and converge center (Tang and Reiter, 1984). The 266 enhanced air mass transportation from west during afternoon could favor the aerosol plume transport. The 267 enhanced WS during afternoon was also observed with the increase of air temperature (Figure 5). The 268 diurnal variations of chemical species during monsoon period were relatively flat comparing with those 269 during P1 and P2 which may relate with the relatively consistent air mass origination during monsoon. 270 Nitrate presented a significant different diurnal variation with high values during nighttime to early 271 morning and low values during afternoon. These features were highly correlated with that of the RH and air 272 temperature (Figure 5) suggesting the importance of thermodynamically-driven gas/particle partitioning of 273 ammonium nitrate and heterogeneous production of nitrate due to hydrolysis of N_2O_5 . In addition, during 274 the early morning time (6:00 - 8:00), there was a peak for most species, which was accompanied with the 275 lowest air temperature and the highest RH, and the lower plenary boundary layer (PBL), which could 276 concentrate all the aerosol species (Yanai and Li, 1994). Overall, the diurnal variations of aerosol species 277 at Nam Co may be dominated by the variabilities of both long-range transport air mass and local 278 meteorological conditions.

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280 In order to further elucidate the chemical processes and potential sources of the aerosol species, the 281 relationships of chemical species with wind conditions were analyzed based on bivariate polar plot analysis 282 (Figure 6). During P1, sulfate and ammonium had hotspots from all directions across a wide range of wind 283 speed $(0 - 10 \text{ m s}^{-1})$; while OA was prevailing from southeast and northwest winds; nitrate had hotspots when the wind speed was relatively slow $(0 - 8 \text{ m s}^{-1})$ and from southeast/east mainly; BC had hotspots 284 from south, west and northwest at high wind speeds $(4 - 12 \text{ m s}^{-1})$. During P2, all species except nitrate had 285 hotspots from south/southwest at high wind speeds $(6 - 14 \text{ m s}^{-1})$; Nitrate also had hotspots at a relative 286 287 low wind speed from southwest. These results suggested that the sources and formations of aerosols during 288 P1 and P2 could be different, albeit the diurnal variations of species between these two periods were similar. 289 During monsoon period, all species had similar distributions of the hotspots, which could be from all 290 directions but relatively weak from southeast.

292 The significant difference of diurnal variation between sulfate and nitrate suggested the different chemical 293 evolution of these two species. Nitrate and sulfate during three periods all had weak correlations. During P1, 294 the ratios of sulfate versus nitrate had peaks (> 20) during afternoon, while during P2, the ratios all kept at 295 low values (< 20) (not shown). In order to further investigate the chemical evolutions of these two species, 296 we selected the high concentration periods based on nitrate concentration (> average + 2σ). Figure 7 297 showed all periods with high nitrate during the study and the corresponding meteorological conditions. It is 298 easy to find that most of the high mass loading periods occurred during nighttime. The wind speed and 299 wind direction varied dynamically and most of them were from southwest. Higher wind speed from this 300 wind direction could transport more aerosol plumes to Nam Co as illustrated from the event 1 (E1) during 301 which all the species (OA, sulfate, ammonium, and BC) except nitrate increased; the weather during this 302 type of event was accompanied with warm and dry air conditions. When the wind direction was from 303 southwest with lower wind speed (E2), the RH increased to higher than 90% accompanying with deceased 304 air temperature, and the mass concentrations of nitrate, sulfate and OA increased significantly and BC 305 decreased significantly. These results further suggest the different chemical evolution and sources for 306 different aerosol episodes at Nam Co Station.

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308 3.4 The average chemical feature of organic aerosol

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310 The average mass spectrum of OA was shown in Figure 8a. The organic mass was on average composed of 311 51% oxygen, 44% carbon, and 5% hydrogen with an average nominal formula being $C_1H_{1,33}O_{0.88}N_{0.001}$. On average, $C_x H_v O_1^+$ (38.1%) and $C_x H_v^+$ (37.3%) ions dominated the total OA following by $C_x H_v O_2^+$ (19.8%), 312 313 and $H_vO_1^+$ (4.6%). m/z 44 was the base peak in the OA spectrum and mainly composed of CO_2^+ (99%). m/z 314 43 had contributions from both $C_2H_3O^+$ (82%) and $C_3H_7^+$ (18%). m/z 55, likely an important primary 315 fragment was consisted of 51% $C_4H_7^+$ and 49% $C_3H_3O^+$. The OA was highly oxidized with an O/C of 0.88 316 on average. The OA was more oxidized during pre-monsoon than monsoon with higher O/C ratio (0.94 vs. 317 0.72) and lower H/C ratio (1.28 vs. 1.44) (Figure 8b). The oxidation states of OA during two pre-monsoon 318 periods were also different. The O/C during P2 (0.98) was higher than P1 (0.90) and the H/C was higher 319 during P1 than P2 (1.30 vs. 1.27, significant, P< 0.005, chi-square test). Correspondingly, the OA during P2 320 contained higher contribution of $C_x H_y O_1^+$ (40.3 vs. 39.1%, significant, P< 0.005, chi-square test) and 321 $C_xH_vO_2^+$ (25.2 vs. 23.6%, significant, P< 0.005, chi-square test) ions than those during P1 (Figure 8c).

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The diurnal variations of both O/C and OM/OC ratios presented higher values during late morning to afternoon and lower values during early morning (Figure 9a). H/C presented an opposite trend. These patterns suggested that OA was more oxidized during daytime which could be due to photo oxidation and/or transport of highly oxidized OA during daytime. We examined the variation of elemental ratios with OA mass concentrations, and found that the O/C increase was accompanied with the increase of OA mass concentration (Figure 9b). This relationship could be related with the different influences of wet scavenging 329 on more oxidized and less oxidized OA. In addition, this result likely suggested the importance of 330 transportation on the oxidized OA during afternoon due to the higher mass concentration occurring 331 frequently on afternoon time. However it was not possible that aerosol plume arrived at Nam Co within 332 several hours due to the long distance from source regions (more than 1000 km from India). Previous 333 studies have reported the presence of an aerosol layer between 6 - 18 km a.s.l altitude over the Tibet 334 Plateau during summer (Tobo et al., 2007; Vernier et al., 2011). He et al. (2014) examined the vertical 335 profiles of aerosol extinction coefficients measured with a Micro Pulse Lidar at Naqu, about 100 km east of 336 Nam Co Station, and observed a maximum in aerosol extinction coefficient between 18 - 19 km a.s.l 337 during summer 2011. Recently, Gu et al. (2016) examined the aerosol compositions using the global three-338 dimensional Goddard Earth Observing System chemical transport model (GEOS-Chem) and found elevated 339 concentrations of sulfate, nitrate, ammonium, BC, and organic carbon over the TP. Further, observational 340 and modeling studies have also shown that deep convection over the TP during daytime is one of the 341 important routes for tropospheric and stratospheric exchange of aerosols (Cristofanelli et al., 2009; 342 Cristofanelli et al., 2010; Lin et al., 2016). Thus the enhanced aerosol concentrations during afternoon 343 could be possibly attributed to the mixed downward of aerosol layer from the upper troposphere during the 344 growth of TP boundary layer. Recently, Xu et al. (2017) elucidate this mechanism at Nam Co Station to 345 explain the source of O_3 and peroxyacetyl (PAN), and found strong downward motion core events during 346 pre-monsoon and monsoon periods which accompanied with increased O_3 and PAN at the ground level. 347 We also performed a test using the Weather Research and Forecasting model (WRF) model to check this 348 kind of transport during P2 which indeed showed a downward motion core at the Nam Co Station (Figure 349 S7). Nevertheless, this hypothesis needs further validation in the future in this region.

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351 **3.5 The PMF apportionment on the OA**

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353 The two OA factors determined by PMF have distinctly different mass spectra and diurnal patterns (Figure 354 10), including a less oxidized OOA (LO-OOA; O/C = 0.49) and a more oxidized OOA (MO-OOA; O/C =355 0.96). Both factors appeared to be secondary in nature. Our inability to separate a primary traffic-related OA factor is consistent with the fact that $C_4H_9^+$ was a minor peak in the OA spectra (0.6% of the total 356 signal) and its organic-equivalent concentration averaged only 0.008 µg m⁻³ during this study suggesting a 357 358 minimal contribution from traffic-related primary OA. Collier et al. (2015) reported that the average $fC_4H_9^+$ 359 (fraction of total organic signal accounted for by $C_4H_9^+$) was 8.1% in the mass spectrum of primary OA 360 from vehicle emissions. Based on this relationship, even assuming $C_4H_9^+$ was completely contributed by vehicle-derived OA, the average vehicle-OA concentration would be only ~ 0.1 μ g m⁻³, or 7% of total OA 361 362 mass, during our study. Note that $C_4H_9^+$ fitted well in the PMF analysis with very small residual, indicating 363 that the signal of this ion was properly apportioned between the two OOA factors.

365 The mass spectra of MO-OOA and LO-OOA were characterized by high peaks at m/z 44 (mostly CO_2^+) 366 and LO-OOA had a relative large peak at m/z 43 (mostly $C_2H_3O^+$) as well (Figure 10a). The contributions 367 of the $C_x H_y O_2^+$ ion category in these two mass spectra were 15.1% and 28.6%, respectively, while the contributions of $C_x H_y O_1^+$ were 37.8% and 41.5%, respectively. The time series of LO-OOA correlated well 368 with $C_x H_v O_1^+$ and $C_x H_v^+$ ions, while MO-OOA correlated well with $C_x H_v O_2^+$ ions (Figure S8). In addition, 369 the time series of MO-OOA correlated well with sulfate ($R^2 = 0.55$), BC ($R^2 = 0.54$) and less well with 370 nitrate ($R^2 = 0.33$), while LO-OOA correlated poorly with sulfate, BC and nitrate (Figure 10b and S9). 371 372 These results highlight the oxidation degree of MO-OOA and LO-OOA. LO-OOA and MO-OOA 373 accounted for 41% and 59% on average, respectively, of total OA mass during this study (Figure 11a), and 374 their relative contributions varied across the study (Figure 2e). For example, LO-OOA accounted for 41% 375 of the total OA mass during P1, 24% during P2, and 67% during the monsoon period (Figure 11b). The 376 diurnal pattern of MO-OOA was characterized by higher concentrations during afternoon similar to those 377 of sulfate and BC. While the diurnal pattern of LO-OOA peaked at early evening time (Figure 10c). The 378 polar plot showed concentrated hotspots to the northwest of the sampling site for MO-OOA, and southeast 379 for LO-OOA (Figure 6), indicating that the sources of these two components were different. Based on 380 trajectory analysis (Figure 3), MO-OOA was likely closely related to long-range transport of air mass from 381 southwest, while LO-OOA could from relative shorter distance transport such as marine air mass from 382 south and regional background aerosol during the nighttime. Shen et al. (2015) reported that there were 383 significant aerosol source from biogenic emission during summer near the Nam Co Station. As shown in 384 Figure 12a and b, MO-OOA, which was highly oxidized (O/C = 0.96), appeared on the up-left corner of the 385 triangle plot while LO-OOA was in the middle part with an O/C ratio of 0.49. The high oxidation degree of 386 MO-OOA was likely related to extensive aging processes occurred during long-range transport. The slope 387 of linear fitting of all the data points in V-K diaphragm is -0.76 suggesting the evolution of OA as 388 carboxylic acid functionalization (Figure 12c).

389

390 Biomass burning emission is an important source in South Asia and could be transported to Himalayas and 391 TP during pre-monsoon (Engling et al., 2011; Kumar et al., 2011; Sang et al., 2013; Cong et al., 2015), 392 however there are not significant signals of biomass burning aerosol in our AMS results such as signals at 393 m/z 60 and 73 in mass spectrum of OA which were found to be associated with levoglucosan formed from 394 the pyrolysis of cellulose (Alfarra et al., 2007). The contribution of f60 (fraction of total organic signal 395 accounted for by m/z 60) for LO-OOA and MO-OOA were 0.2% and 0.3%, respectively, which were 396 similar with the global background level (less than 0.3%) suggested by Cubison et al. (2011). These results 397 suggest that OA, if ever partly originated from biomass burning emission, could have been highly oxidized 398 during transport. This behavior had been observed in a few studies that levoglucosan could be quickly 399 (within a few hours) oxidized after being emitted (Ortega et al., 2013). In addition, Zhou et al. (2017) 400 recently reported the observation of a highly aged BBOA factor with f60 < 0.3% in its mass spectrum, in 401 aged wildfire plumes that had gone through extensive photochemical oxidation.

3.6 Sensitivity of the aerosol transport to synoptic process

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405 Synoptic process is an important factor determining if the regional emissions can be transported to the TP. 406 It is interesting to know what kind of synoptic process is favorable for transporting aerosol plume to the 407 Himalayas and the TP. A 3-D Regional chemical transport Model (REAM) coupled with WRF was used to 408 examine the chemical evolution and regional transport of pollutants such as aromatics in this study. REAM 409 has been used in previous studies of the Tibet Plateau, and details about the model can be found in Zhang et 410 al. (2017) and supplementary material. REAM could capture some synoptic processes which cannot be 411 simulated by the normally used reanalysis data due to their low-resolution and the complexity of terrain in 412 the Tibet Plateau (Zhang et al., 2017). Figure 13 shows the distribution of simulated daily surface wind, 413 300 hPa geopotential height fields and concentrations of reactive aromatics over the Tibetan Plateau during 414 30 May – 13 June, 2015. During 30 May to 7 June corresponding to P1, there was a trough over the north 415 propagated from west to east and this low pressure induced increasingly stronger surface wind from India 416 to the TP, which could lead to transport of aerosol plume as illustrated by the results of model and AMS. 417 The simulated concentration of reactive aromatics showed a peak during this period (Figure S10). During 418 8-11 June, there was a weak ridge system over the north. Weak wind from west and south was observed 419 and the simulated concentrations of reactive aromatics were sharply decreased (Figure S10). After that, a 420 weak low-pressure trough system was observed again. The increased concentrations of reactive aromatics 421 were also observed accompanying with intensified southern wind. Although these trends are basically 422 consistent with our AMS results, there were also significant differences (Figure S10). The possible reason 423 was that the weak ridge during P2 enhanced the wind from north and weakened the wind from west and 424 south where a lot of biomass burning emission sources located (Figure 3b). Zhang et al., (2017) suggested 425 that a cut-off low system from the upper troposphere could be an important driver for aerosol plume 426 transport into the TP. In our study, the trough/ridge system seems to be also an important factor affecting 427 the transport of aerosol plume from south and west, although this effect tends to be weaker in summer than 428 in the other seasons because the tropopause is higher and stratospheric wave activity is weaker in summer.

429

430 **3.7 Atmospheric implications**

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Our results have several potential implications to the atmospheric studies in the TP and Himalayas. Firstly, it is useful for the accurate estimation of the radiative forcing of aerosols in this region and validation of current model simulation results based on our observed chemical composition and mass loadings of fine aerosols. Ji et al. (2015) estimate the radiative forcing from aerosols over the TP and Himalayas at the surface level using a regional climate model (RegCM4.3); for carbonaceous aerosols, there are several literatures that tested the model results, but all other species were referred to data available in the published inventory with a coarse spatial resolution. Secondly, our findings have implications for aerosol deposition 439 on the glacier of this region and evaluation of subsequent impacts on the melting of snow/ice (Yasunari et 440 al., 2010). Thirdly, highly-time resolved aerosol data is very scarce in this remote plateau, thus our data are 441 valuable to validate modeling results regarding the transport of aerosol plume as demonstrated in section 442 3.6. At last, the transport mechanism of aerosol to the inland of TP is less understood so far. Hindman and 443 Upadhyay (2002) suggested that the vertical lifting due to convection and subsequent horizontal mountain-444 valley wind could lead to the transport of aerosol from Nepal to Tibet. Dumka et al. (2010) also highlighted 445 the important role of mountain-valley wind in the aerosol transport in the central Himalayas. The dynamic 446 variations of aerosol chemical species measured here, are likely helpful to elucidate the transport 447 mechanism of high concentration air mass. Nevertheless, this scientific issue required further detailed 448 investigations in the future.

449

450 **4. Conclusion**

451

The average PM₁ loading measured at Nam Co during June 2015 was 2.0 µg m⁻³ with organics accounting 452 453 for 68%, followed by sulfate (15%), black carbon (8%), ammonium (7%), and nitrate (2%). This mass 454 concentration was comparable to some AMS observations from mountain-top sites. The mass concentration of PM₁ varied during different weather conditions with higher concentration during pre-monsoon and lower 455 456 concentration during monsoon. The pre-monsoon period could also be divided into two periods (P1 and P2) 457 based on meteorological conditions and aerosol chemical composition. During P1, PM₁ was characterized 458 with high contribution from OA and sulfate, while increased contribution of nitrate was observed during P2 459 with wet and cold weather conditions. All PM₁ species had clear diurnal variations with OA, sulfate, BC, 460 and ammonium peaking during afternoon due to photochemical production of these species coupled with 461 transport of aerosol plumes. Nitrate, however, peaked during the nighttime and early morning, which was 462 related to the high RH condition and low air temperature. The formation of nitrate was highly correlated 463 with transport of air masses from southwest under very low wind speeds, while the mass concentrations of 464 sulfate, OA, and BC were highly correlated with air masses from northwest and southeast under higher 465 wind speed conditions. OA was overall highly oxidized during the entire study with higher O/C ratios 466 during the pre-monsoon period. Based on PMF analysis, the OA was found to be composed of a LO-OOA 467 and a MO-OOA. LO-OOA was mainly associated with air masses originated from south, while MO-OOA 468 was mainly from northwest. MO-OOA dominated OA during the pre-monsoon period, while LO-OOA 469 dominated during the monsoon period. The transport mechanism of aerosol plume was further investigated 470 by using the REAM chemical model coupled with the WRF model. The aerosol plume was found to be 471 more easily transported to the TP and Himalayas during low pressure trough weather.

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Sites	Instruments	Lat.	Long.	Elev.	Date	Mass Conc.	Reference
Nam Co	HR-AMS	30.77°N	90.9°E	4730	Jun-15	1.84	
Menyuan	ACSM	37.61°N	101.26°E	3925	5 September - 15 October, 2013	10.8	Du et al. (2015)
Montsec	ACSM	42.05°N	0.73°E	1570	2011 - 23 April 2012	4.9	Rinaldi et al. (2015)
Jungfraujoch	ToF- ACSM	46.55°N	07.98°E	3580	27 July 2012 - 2 October 2013	0.55	Fröhlich et al. (2015)
Mt. Cimone	HR-AMS	44.18°N	10.7°E	2165	June - July 2012	4.5	Rinaldi et al. (2015)
Mt. Whistler	HR-AMS	50.01°N	122.95°E	2182	Apr - May 2006	1.91	Sun et al. (2009)
Puy de Done	cToF-AMS	45.77°N	2.95°E	1465	Sep 2008 - June 2010	2008 Autumn: 7.82 2008 Winter: 5.58 2009 Summer: 27.594	Freney et al. (2011)
Mt. Bachelor	HR-AMS	43.98°S	121.69W	2800	25 July - 25 August, 2013	15.1	Zhou et al. (2017)
Sub- Antarctic Bird Island	HR-AMS	54.00°S	38.00°W	ND	November and December, 2010	0.46	Schmale et al. (2013)

Table 1. Summary of AMS measurement at high elevation sites around the world. The massconcentration was for NR-PM₁.



Figure 1. Location map for (a) the Tibetan Plateau and (b) Nam Co Station colored by the altitude.

- (c) The profile of altitude from costal area to Nam Co Station (vertical dash line). The red arrow
- in the map represent the possible wind direction.





Figure 2. The combo plot of the data of the Nam Co study including (a) the meteorological conditions (T: air temperature; RH: relative humidity; Precip.: precipitation), (b) the variation of WS (wind speed) colored by WD (wind direction), (c) the temporal variation of mass concentration of PM_1 species and the average contribution each species (pie chart), (d) the mass contribution of each PM_1 species and the total mass concentration of PM_1 , and (e) the mass contribution of PMF results (section 3.5). Three periods based on the meteorological conditions were marked.



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764 Figure 3. 72-hour backward air mass trajectory and class statistics over NOCS for (a) P1, (b) P2, and (c) 765 monsoon period. The back trajectories at 1000 m above ground level were calculated at 1h intervals using 766 the NOAA Hybrid Single Particle Lagrangian Integrated Trajectory (HYSPLIT) model (Draxler and Hess, 767 1998) and one degree resolution Global Data Assimilation System (GDAS) dataset 768 (https://ready.arl.noaa.gov/gdas1.php). The classes of trajectories are colored by pressure. Fire spot 769 observed by MODIS (https://firms.modaps.eosdis.nasa.gov) and average wind rose plot during each period 770 are also shown.





Figure 4. The comparisons of PM₁ species of the mass concentration (left axis) and the ratio between them

- (right axis) between (a) pre-monsoon and monsoon and (b) P1 and P2.
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- 775







Figure 5. Diurnal variations of each species and weather conditions (RH: relative humidity; Tem: air
temperature; and WS: wind speed) during three periods of the study. Note that the signals of sulfate,
ammonium, and BC are increased by a factor for comparison.



783 Figure 6. Bivariate polar plots that illustrate the variations of the concentrations (colored) of each species as

784 a function of wind speed (m s^{-1}) and wind direction during different periods of the study.



meteorological data.



Figure 8. (a) The average mass spectrum of organic aerosol, (b) the average ratios of H/C and O/C during

- different periods, and (c) the average contribution of six ion categories during different periods.





Figure 9. (a) Diurnal variations of elemental ratios and (b) the variations of elemental ratios as the function

798 of mass concentration of organic aerosol.



Figure 10. PMF results of (a) high resolution mass spectra colored by six ion categories for LO- and MOOOA, respectively, (b) temporal variations of the OA factors and corresponding comparison with tracer
species (sulfate and nitrate), and (c) diurnal variations of the mass concentration of the OA factors and
tracer species.





810 Figure 11. (a) The average mass contributions of two factors (a) during the study, (b) during the different

- 811 periods, and (c) the diurnal mass contribution of two factor (left axis) accompanying with the total organics
- 812 (right axis).
- 813



Figure 12. Scatter plots of (a) f44 vs. f43, (b) fCO_2^+ vs. $fC_2H_3O^+$, and (c) H/C vs. O/C for the OA. The dash lines in (a) and (b) refer to a triangular region that encompasses ambient OOA factors determined from PMF analyses of 43 AMS datasets ((Ng et al., 2010). The light solid lines with slopes of 0, -1, and -2 in (c) indicate the changes of H/C against O/C due to adding specific functional groups to an aliphatic carbon (Heald et al., 2010); The light dash lines indicate the oxidation state (OS) of -2, -1, 0, 1, and 2, respectively (Kroll et al., 2011a); The red and blue dash lines are derived from the right and left lines in the triangle plot; The heavy solid line indicate the slope of our measured data. The corresponding values of the OOAs

822 identified in this study are also shown in each plot.



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824 Figure 13. Daily distribution of WRF-simulated surface wind and REAM-simulated

825 concentrations of reactive aromatics over the Tibet Plateau during 30 May to 13 June, 2015.

Chemical characteristics of submicron particles at the central 1 **Tibet Plateau: insights from aerosol mass spectrometry** 2

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18 Recent studies have revealed a significant influx of air pollutionanthropogenic aerosol from south Asia to 19 Himalayas and Tibet Plateau (TP) during pre-monsoon period. In order to characterize the chemical 20 composition, sources, and transport processes of polluted air massaerosol in this area, we carried out a field 21 study during June 2015 by deploying a suite of online instruments including an Aerodyne high-resolution 22 time-of-flight aerosol mass spectrometer (HR-AMS) and a multi-angle absorption photometer (MAAP) at 23 Nam Co Station (90°57'E, 30°46'N, 4730m a.s.l) at the central of the TP. The measurements were made at 24 a period when the transition from pre-monsoon to monsoon occurred. The average ambient mass 25 concentration of submicron particulate matter (PM₁) over the whole campaign was ~2.0 μ g m⁻³, with 26 organics accounting for 68%, followed by sulfate (15%), black carbon (8%), ammonium (7%), and nitrate (2%). Air pollutionRelative higher aerosol mass concentration episodes were observed during the pre-27 28 monsoon period, whereas persistently low aerosol concentrations were observed during the monsoon period. 29 However, the chemical composition of aerosol during the air pollutionhigher aerosol concentration 30 episodes in the pre-monsoon season was on a case-by-case basis, depending on the prevailing 31 meteorological conditions and air mass transport routes. Most of the chemical species exhibited significant 32 diurnal variations with higher values occurring during afternoon and lower values during early morning, 33 whereas nitrate peaked during early morning in association with higher relative humidity and lower air 34 temperature. Organic aerosol (OA) was more oxidized with an oxygen-to-carbon ratio (O/C) of 0.94 during 35 the pre-monsoon period than during monsoon (average O/C of 0.72). The), and an average O/C of OA-was

0.88 over the entire campaign period-<u>suggesting overall highly oxygenated aerosol in the central TP.</u>
Positive matrix factorization of the high resolution mass spectra of OA identified two oxygenated organic
aerosol (OOA) factors: a less oxidized OOA (LO-OOA) and a more oxidized OOA (MO-OOA). The MOOOA dominated during the pre-monsoon period, whereas LO-OOA dominated during monsoon. The
sensitivity of air pollutionmass transport during pre-monsoon with synoptic process was also evaluated
with a 3-D chemical transport model.

42

43 1. Introduction

The Tibet Plateau (TP) and Himalayas is a vast and elevated highland in Central Asia that extends over the area of 27-45°N, 70-105°E with a mean elevation of more than 4000 m above sea level (a.s.l.). It is a sparsely populated area with minimal local pollution. The TP is an ideal area for observations of free tropospheric air masses and pollutantsaerosol transported from polluted areas surrounding the TP after long distances. Determination of the chemical characteristics of aerosol particles in the TP is important for assessments of their influences on atmospheric chemistry and climate (Li et al., 2016), which are so far poorly understood due to harsh conditions and logistic limitations.

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52 Over recent decades, an increasing number of field studies have been conducted in these regions to 53 characterize aerosol physical and chemical features from mountain observatories, e.g., the Nepal Climate 54 Observatory-Pyramid (5079 m), which is set up for long-term monitoring and synchronous observation 55 (Bonasoni et al., 2010; Liu et al., 2017). There are significant seasonal variations in aerosol mass loading in 56 the southern TP and Himalayas. Higher aerosol concentration was often found during pre-monsoon due to 57 less precipitation and favorable atmospheric circulation (Bonasoni et al., 2010; Marinoni et al., 2010; 58 Marinoni et al., 2013; Zhao et al., 2013). For example, the concentration of carbonaceous species at the 59 Qomolangma (Mt. Everest) Station (4276 m a.s.l.), northern Himalayas during pre-monsoon was found to 60 be 3 - 5 times higher than that during the monsoon periods (Cong et al., 2015). Since this seasonal 61 variation of aerosol loading is consistent at both the southern and northern Himalayas (Xu et al., 2014a), 62 polluted air massaerosol plume was thought to be able to across the Himalayas, a finding which is also 63 supported by model results (Lu et al., 2012; Lüthi et al., 2015; Zhang et al., 2015). During monsoon period, 64 ambient aerosol from the upwind sources is significantly scavenged during long-range transport and air 65 mass is mainly originated from marine area, which can lead to aerosol chemical differences between pre-66 monsoon and monsoon. The distinct seasonal variation of aerosol loading is mainly attributed to the change 67 of the dominated climatic systems and then weather conditions. During pre-monsoon, the cold and dry 68 southern WestliersWesterliers dominated the southern TP and Himalayas, while the South Asia Monsoon 69 covers most of South Asia, Himalayas, and the southern TP during summer period.

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71 Most of the studies conducted in these regions focused on some specific species, such as black carbon,
72 which has strong light absorption. Burning of biomass fuels and wildfires in the south Himalaya and South

73 Asia are thought to be important sources for the black carbon (Stone et al., 2010; Engling et al., 2011; 74 Kumar et al., 2011). However, a recent study shows that sources of black carbon in the region of South 75 Asia are highly complex, including emissions from low efficient transport tools and cooking using cow 76 dung and biogas as well (Stockwell et al., 2016). These burning activities also emit other species such as 77 organic and inorganic particulate species and volatile organic compounds, which generate a well-mixed 78 aerosol plume eventually via processes such as coagulation, evaporation, oxidation, and condensation. Fang 79 et al. (2015) recently suspected that biogenic aerosol could also be an important contribution for aerosol in 80 the TP during summer. However, these mixed plumes have been rarely, if ever, characterized by 81 comprehensive field measurements.

83 Filter-based sampling method with a low time resolution (days) were widely adopt in these remote regions 84 due to logistical difficulties with deployment of real-time instruments. The low time resolution made the 85 understanding of chemical processes of aerosol during transport challenging. Secondary species such as 86 sulfate and water soluble organic carbon (WSOC) are normally the dominated species in aerosol. For 87 example, WSOC accounted for about 60% of OC and the ratio between OC and EC could be up to 10 88 (Zhao et al., 2013; Cong et al., 2015), suggesting a dominant contribution of secondary organic aerosol 89 (SOA) to carbonaceous aerosol loading in the TP. The ambient conditions at high elevation regions are 90 characterized by higher solar radiation and concentrations of oxidants such as OH radicals and O3, which 91 makes photochemical processing in this high elevation remote region intense and likely dominant. High-92 time resolution measurement is thus necessary in this region for detecting short-term events and the 93 evolution of pollutants. In addition, the high time resolution data are useful for constraining atmospheric 94 chemical transport models.

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96 The aerodyne aerosol mass spectrometer (AMS) has been widely used to study the chemical composition of 97 non-refractory submicron particle (e.g., Xu et al., 2014b; Xu et al., 2016). There are two merits of the AMS 98 including its high time resolution and bulk measurement. The high time resolution is usually in minutes 99 which has advanced our understanding of fundamental chemical processes of fine aerosols at different 100 regions of the world (Canagaratna et al., 2007). The bulk measurements enable the observation to obtain 101 rich information of various aerosol chemical species simutaneously. The organic aerosol, which is often the 102 most important component of aerosol, can be further analyzed to determine the average elemental ratios 103 and by positive matrix factorization (PMF) analysis to determine the sources and atmospheric processes 104 (Ulbrich et al., 2009; Zhang et al., 2011). The atomic elemental ratios of oxygen and hydrogen to carbon 105 (O/C and H/C) calculated from the OA mass spectra can provide information about the sources and 106 evolution processes of OA in the atmosphere (Aiken et al., 2008; Heald et al., 2010; Kroll et al., 2011b; Ng 107 et al., 2011). They also often closely correlate with key OA properties such as hygroscopicity, density, and 108 phase separation (Jimenez et al., 2009; Bertram et al., 2011). In addition, due to high sensitivity and low

detection limits, AMS has been successfully deployed at many remote sites with low aerosol mass loadingsuch as Antarctica (Schmale et al., 2013; Giordano et al., 2017).

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112 The study here presents results from measurements using an AMS at the central TP during the transitional 113 period from pre-monsoon to monsoon. The study was designed to characterize aerosol chemical 114 composition, temporal variations, transport processes, and emission sources. During the campaign, besides 115 the AMS, multiple other real-time instruments were also deployed.

116

117 2. Methodology

118 **2.1 Site description**

119 The field study was conducted between May 31 and July 1, 2015 at a high altitude observatory, i.e., Nam 120 Co Station for Multisphere Observation and Research, Chinese Academy of Sciences (Nam Co station, 121 90°57′E, 30°46′N 4730m a.s.l), at the central part of the TP (Figure 1). The Nam Co station is located 122 nearby Nam Co lake (Figure 1b), the second largest inland lake in the TP (area: 1920 km²) which is located 123 at the northern border of Nyaingentanglha Mountains. The melted glacier from Nyaingentanglha Mountains 124 supply water to the lake each year during warm seasons. The Nam Co station and its surrounding is a 125 pristine region except for a small countyvillage for local people that is about 10 km west to the station-with 126 a population of 300 to 500. In the past several years, tourism for this beautiful lake has grown. A 127 freewayhighway for tourists was built about 53 km south of the station-with a low traffic intensity (less 128 than 300 cars per day) during June. The capital city (Lhasa) of the Tibet Autonomous Region is about 100 129 km southeast of the station with an average elevation of 3600 m a.s.l. between which is Nyainqentanglha 130 Mountains (higher than 6000 m a.s.l) (Figure 1c). The closest town, Dangxiong, is alongside the famous 131 highway, Qinghai-Tibet Highway, and about 70 km east of the station with an average elevation of 4200 m 132 a.s.l. Overall, the station is surrounded by the mountains in the south and east, and the lake at the west. The 133 ecology of surrounding area is semi-arid land dominated by alpine meadow and barren areas. The 134 precipitation is mainly occurred during summer monsoon period. The cooking and heating at the station is 135 supplied by the power and natural gas.

137 2.2 Instrument setup

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138 The study was conducted at the observatory field of the station using a customer-made trailer with inlet 139 stepped out of the top with the height of ~5m above ground. All the instruments were arranged inside the 140 trailer where the air temperature was controlled at ~20°C by two air conditioners. The air flow of inlet was maintained by a vacuum pump with a flowrateflow rate of 10 L/_min, following by-1, and several 141 142 instruments were taking samples from this inlet including a high-resolution time-of-flight aerosol mass 143 spectrometer (HR-AMS, Aerodyne Research Inc., Billerica, MA, USA), a custom-made scanning mobility 144 particle sizer (SMPS), a soot particle Aerodyne mass spectrometer (SP-AMS, Aerodyne Research Inc., 145 Billerica, MA, USA), a single-particle intracavity laser-induced incandescence photometer (SP2, DMT,

146 Inc., Boulder, CO, USA), and a seven wavelength Athelometer (model AE31, Magee Scientific, Berkeley, 147 CA, USA). The total flowrate flow rate of the inlet was maintained at ~16 L^{\perp} min⁻¹ and a PM_{2.5} cyclone was 148 used in the front of the inlet (model URG-2000-30EH, URG Corp., Chapel Hill, NC, USA) to remove 149 coarse particles. Several additional instruments were also co-located but with their own inlets, such as a 150 Nephelometer (model 3563, TSI, Shoreview, MN, USA), a PM_{2.5} sampler (model PQ200, BGI, USA), and 151 a multi-angle absorption photometer (MAAP, model 5012, Thermo Electron Cooperation, USA). The 152 operations of the instruments used in this study were described below. The meteorological data was recorded 153 at the tower of the Nam Co station at a height of 20 m above ground.

155 2.2.1 HR AMS

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156 HR-AMS was used to measure the non-refractory PM₁ (NR-PM₁), which was thermally vaporized at 157 ~600 °C, ionized with a 70eV electron impact and determined by using a time-of-flight mass spectrometry. 158 The details of the instrument have been described elsewhere (DeCarlo et al., 2006). For our study, HR-159 AMS was only operated with V-mode with 5 min resolution due to the low aerosol mass loading. Due to 160 malfunction of the chopper, size distribution of NR-PM₁ was not determined. The HR-AMS was calibrated 161 for ionization efficiency (IE) with the ammonium nitrate following standard procedures (Jayne et al., 2000) 162 at the beginning, in the middle and end of the study. Particulate-free air was sampled twice during the study 163 period to adjust the influences of air on the fragmentation table and determine the detection limits (DLs) of 164 aerosol species. The DLs were calculated as 3 times the standard deviations (3σ) of the measured values 165 during the particle-free sampling periods. The 2.5-min DLs for organic, sulfate, nitrate, ammonium, and chloride were determined at 0.108, 0.014, 0.007, 0.002, 0.010 µg m⁻³, which are comparable to the values 166 167 reported in previous studies.

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169 2.2.2 Other instruments

The Aethalometer was used to measure the equivalent black carbon mass concentration using seven
wavelength lights. The data for BC is commonly calculated from measurement at 880 nm using
recommended MAC from the manufacture. The instrument was operated at the time resolution of 5 min
with a flow rate of 5 L/min, which was calibrated once a week.

PM_{2.5}-sampler was set up at the top of the trailer. Filter samples were collected between 2 June and 1 July
2015. Each aerosol sample was collected on a 47 mm quartz filter (Whatman, Maidstone, UK) using the
aerosol sampler with a flowrate of 16.7 L/min (BGI, USA, model PQ 200). The meteorological data was
recorded at the tower of the Nam Co station at a height of 20 m above ground.

180 **2.3 Data processing**

181 The HR-ToF-AMS data were processed using the standard software of SQUIRREL (v1.56) and PIKA
182 (v1.15c) written in IGOR (Wavemetrics, Inc., Lake Oswego, OR, USA) (<u>http://cires.colorado.edu/jimenez-</u>

183 group/ToFAMSResources/ToFSoftware/index.html) to determine the mass concentrations of the NR-PM1 184 species and the ion-speciated mass spectra of organics. An empirical particle collection efficiency (CE) of 185 0.5 was used, which has been widely used in field studies employing AMS with a dryer installed in front of 186 the equipment's particle inlet. This CE value was further validated by the consistency and reasonable slope 187 between HR-AMS measured mass concentrations and SMPS-determined particle volumes (section 3.1.2, 188 $R^2 = 0.9$, slope = 1.48). The elemental ratios of OA (O/C, H/C, and organic matter to carbon (OM/OC)) in this study was determined using the "improved-ambient" method (Canagaratna et al., 2015). Default 189 190 relative ionization efficiency (RIE) values were used for orgnaics (1.4), nitrate (1.1), and chloride (1.3), 191 while an RIE value of 3.8 was determined for ammonium and 1.1 for sulfate based on the calibration for 192 pure NH₄NO₃ and (NH₄)₂SO₄, respectively.

193

194 The source apportionment of organics was conducted by PMF with the robust engine. First, organic matrix 195 was analysed using the PMF2.exe algorithm in robust mode (Paatero and Tapper, 1994) and explored using 196 the PMF Evaluation Toolkit (PET) (Ulbrich et al., 2009). The PMF solution was evaluated following the 197 procedures outlined in Table 1 of Zhang et al. (2011) including modification of the error matrix and 198 downweight of low signal-to-noise ions. Moreover, based on the AMS fragmentation table, some organic 199 ions were not directly measured but scaled to the organic signal at m/z 44, which were downweighted by 200 increasing their errors by a factor of 3. A two-factor solution with fPeak = 0 was chosen in this study, as it 201 is able to reconstruct the total OA mass and temporal profiles very well. The results of three-factor solution

- with fPeak = 0 are shown in supplementary material (Figure S1), which show splitting of the solutions.
- 203

3 Results and discussions

3.1 The meteorological conditions during the study

206

207 During the field study, the meteorological conditions were generally cold and windy (Figure 2a and b). The 208 average air temperature was 8.4 ± 3.6 °C with a diurnal hourly average ranging from 4.8 to 12.3 °C; the 209 diurnal average wind speed (WS) ranged from 4.5 to 6.5 m s⁻¹. The dominant wind directions were south 210 and southwest, although they varied during different periods (Figure 2b and S2). The wind direction (WD) 211 had distinct diurnal variation with air mass from south during night to morning and west during afternoon. 212 Based the report of the Climate Diagnostics Bulletin on of India 213 (http://imdpune.gov.in/Clim_RCC_LRF/Climate_Diagnostic_Bulletins/cdbi_apr_2015.pdf), the 214 precipitation from Indian monsoon had covered most part of India on June 12 to June 14, and these days 215 are treated as the onset of IndiaSouth Asian monsoon. Indeed, precipitation was recorded at the Nam Co 216 station on June 13 and lasted for several days (Figure 2a). Based on this weather condition, the period 217 before June 14 was classified as the pre-monsoon period and afterwards as the monsoon period. The RH 218 during the pre-monsoon and monsoon periods were $48.4 \pm 19.2\%$ and $58.8 \pm 16.5\%$, respectively. The 219 origins of the air masses were also different during these two periods. Figure 3 shows the air mass from

220 west during pre-monsoon period accounted about 30-40%, while it was only 7% during monsoon period, of

- which ~80% was from south.
- 222

3.2 The temporal variations of chemical species

224

The average mass concentration of PM_1 (NR-PM₁ + BC) was 2.0 $\mu g m^{-3}$ during the whole study period 225 226 with 68% OA, 15% sulfate, 2% nitrate, 7% ammonium, and 8% BC. The NR-PM1 mass loading (1.84 µg m^{-3}) at Nam Co Station was lower than the values observed at some high elevation sites such as Menyuan 227 228 (10.8 μg m⁻³) (Du et al., 2015), Montsec (4.9 μg m⁻³) (Ripoll et al., 2015), Mt. Cimone (4.5 μg m⁻³) 229 (Rinaldi et al., 2015), Puy de Done $(5 - 27 \ \mu g \ m^{-3})$, and Mt. Bachelor (15.1 $\ \mu g \ m^{-3})$ (Zhou et al., 2017), but 230 higher than those in some other sites, such as sub-Antarctic (0.46 μ g m⁻³) (Schmale et al., 2013) and Jungfraujoch (0.55 μ g m⁻³) (Fröhlich et al., 2015), similar with that at Mt. Whistler (1.91 μ g m⁻³) (Sun et 231 232 al., 2009) (Table 1 and Figure S3). Although the sampling seasons and altitudes of these sampling sites are 233 different (Table 1), the results of these studies could basically represent the level of pollutionaerosol at 234 these sits due to their relatively long sampling spans (from one month to ten months). The contribution of 235 OA at these high-elevation sites ranged from 50% - 90%; the highest value was at Mt. Bachelor which was frequently influenced by transported biomass burning plume, while the relative low OA contribution (38%) 236 237 at sub-Antarctic was due to the dominant source from marine emission containing high sulfate. The mass 238 concentration of PM₁ varied dynamically during our study with distinct difference between pre-monsoon 239 and monsoon periods. The average mass concentrations of PM_1 for these two periods were 2.6 and 1.2 µg 240 m^{-3} , respectively. For comparison, the concentrations of various species for these two periods were side-by-241 side displayed in Figure 4a. The ratios between pre-monsoon and monsoon for all the species were higher 242 than 1 with the maxima for ammonium (3.1) and sulfate (2.8); the contribution of OA was thus slightly 243 higher during the monsoon period than during the pre-monsoon (71% vs. 64% with a mass concentration 244 ratio of 1.9). Apart from the potential scavenging effect, these results could also be influenced by the 245 sources, transport route, and chemical processes during different periods.

246

247 Based on the mass concentration and temporal variations of PM₁ species and weather conditions, the pre-248 monsoon period could be further divided into two periods, i.e., period 1 (P1: May 30 to June 7) and period 249 2 (P2: June 8 to June 13). P1 was characterized by relatively high sulfate concentration (0.46 vs. 0.41 μ g m⁻ 250 ³) and sunny days, while P2 was characterized by high PM₁ concentration (2.9 vs. 2.1 μ g m⁻³; P< 0.005, 251 chi-square test), high nitrate contribution, and wet and cold days (Figure S4). The air masses for P2 had 252 higher contribution from west and north than those during P1 (Figure 3). Figure 4b shows the comparisons 253 of mass concentration of different species between these two periods. For chemical species, from P1 to P2, 254 nitrate and OA increased dramatically by a factor of 1.6 and 2.2, respectively, and BC and ammonium also 255 increased, suggesting the influence of transported pollutedelevated air mass.plume. However, sulfate and 256 chloride decreased slightly (the ratios between P2 and P1 were ~0.95). The variations of sulfate and nitrate

257 during these two periods could be related to the photochemical conditions and origination of air mass (see 258 section 3.3).

259

260 The particles were generally neutralized as illustrated by the scatter plot between predicted and measured 261 ammonium (slope = 0.91) (Figure S5a). The neutralized PM₁ were likely due to the high availability of 262 ammonia from agriculture emission in North India (Clarisse et al., 2009; Van Damme et al., 2015). The 263 slope was ~0.75 during P2 (green circles, Figure S5a), which suggested suggests that there was excessione 264 ammonium than needed to neutralize sulfate and nitrate. We checked whether the high ammonium was 265 from fragmentation of organic nitrogen such as amines. The ratios of NH^+ to NH_2^+ were consistently during 266 the study (Figure S5b), suggesting there was no significant influences from amino compounds. This excess 267 ammonium determined by the neutralization could be related to the presence of significant amounts of 268 organic anions in aerosol (such as carboxylic acids) or variation of RIE for ammonium which could have 269 higher value in the mixed acidic particles.

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271

3.3 Diurnal variation and chemical process of different species

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273 The diurnal cycles of OA, sulfate, nitrate, ammonium, and BC during different periods are shown in Figure 274 5. All these species unexpectedly present dramatic diurnal variations, especially during P2. OA, sulfate, 275 ammonium, and BC showed a similar pattern with low values during nighttime to early morning and high 276 values during afternoon which suggest their common sources or similar transport pathways (Zhang et al., 277 2001).(Zhang et al., 2001). We checked the diurnal variation of the origination of air masses and found that 278 there were increased air masses from south during nighttime and from west during afternoon (Figure S6)-279 which-). This phenomenon could be related to the plateau monsoon during summer, which the strong 280 heating at the surface resulted in a shallow cyclonic circulation and converge center (Tang and Reiter, 281 1984). The enhanced air mass transportation from west during afternoon could favor the transportation of 282 polluted air mass.aerosol plume transport. The enhanced WS during afternoon was also observed with the 283 increase of air temperature (Figure 5). The diurnal variations of chemical species during monsoon period 284 were relatively flat comparing with those during P1 and P2 which may relate with the relatively consistent 285 air mass origination during monsoon. Nitrate presented a significant different diurnal variation with high values during nighttime to early morning and low values during afternoon. These features were highly 286 287 correlated with that of the RH and air temperature (Figure 5) suggesting the importance of 288 thermodynamically-driven gas/particle partitioning of ammonium nitrate and heterogeneous production of 289 nitrate due to hydrolysis of N_2O_5 . In addition, during the early morning time (6:00 – 8:00), there was a peak 290 for most species, which was accompanied with the lowest air temperature and the highest RH, and the 291 lower plenary boundary layer (PBL), which could concentrate all the air pollutants aerosol species (Yanai 292 and Li, 1994). Overall, the diurnal variations of aerosol species at Nam Co may be dominated by the 293 variabilities of both long-range transport air mass and local meteorological conditions.

295 In order to further elucidate the chemical processes and potential sources of the aerosol species, the 296 relationships of chemical species with wind conditions were analyzed based on bivariate polar plot analysis 297 (Figure 6). During P1, sulfate and ammonium had hotspots from all directions across a wide range of wind 298 speed $(0 - 10 \text{ m s}^{-1})$; while OA was prevailing from southeast and northwest winds; nitrate had hotspots 299 when the wind speed was relatively slow $(0 - 8 \text{ m s}^{-1})$ and from southeast/east mainly; BC had hotspots 300 from south, west and northwest at high wind speeds $(4 - 12 \text{ m s}^{-1})$. During P2, all species except nitrate had 301 hotspots from south/southwest at high wind speeds (6 – 14 m s⁻¹); Nitrate also had hotspots at a relative 302 low wind speed from southwest. These results suggested that the sources and formations of aerosols during 303 P1 and P2 could be different, albeit the diurnal variations of species between these two periods were similar. 304 During monsoon period, all species had similar distributions of the hotspots, which could be from all 305 directions but relatively weak from southeast.

306

307 The significant difference of diurnal variation between sulfate and nitrate suggested the different chemical 308 evolution of these two species. Nitrate and sulfate during three periods all had weak correlations. During P1, 309 the ratios of sulfate versus nitrate had peaks (> 20) during afternoon, while during P2, the ratios all kept at 310 low values (< 20) (not shown). In order to further investigate the chemical evolutions of these two species, 311 we selected the high concentration periods based on nitrate concentration (> average + 2σ). Figure 7 312 showed all periods with high nitrate during the study and the corresponding meteorological conditions. It is 313 easy to find that most of the high mass loading periods occurred during nighttime. The wind speed and 314 wind direction varied dynamically and most of them were from southwest. Higher wind speed from this wind direction could transport more polluted air massaerosol plumes to Nam Co as illustrated from the 315 316 event 1 (E1) during which all the species (OA, sulfate, ammonium, and BC) except nitrate increased; the 317 weather during this type of event was accompanied with warm and dry air conditions. When the wind 318 direction was from southwest with lower wind speed (E2), the RH increased to higher than 90% 319 accompanying with deceased air temperature, and the mass concentrations of nitrate, sulfate and OA 320 increased significantly and BC decreased significantly in contrast to the increase of nitrate. The decreased mass concentration of BC indicated no primary aerosol transportation, thus the increased species were 321 322 mainly secondarily formed. These results further suggest the different chemical evolution and sources for 323 different aerosol episodes at Nam Co Station.

324

325 **3.4 The average chemical feature of organic aerosol**

326

327The average mass spectrum of OA was shown in Figure 8a. The organic mass was on average composed of32851% oxygen, 44% carbon, and 5% hydrogen with an average nominal formula being $C_1H_{1,33}O_{0.88}N_{0.001}$. On329average, $C_xH_yO_1^+$ (38.1%) and $C_xH_y^+$ (37.3%) ions dominated the total OA following by $C_xH_yO_2^+$ (19.8%),330and $H_yO_1^+$ (4.6%). m/z 44 was the base peak in the OA spectrum and mainly composed of CO_2^+ (99%). m/z

331 43 had contributions from both $C_2H_3O^+$ (82%) and $C_3H_7^+$ (18%). m/z 55, likely an important primary 332 fragment was consisted of 51% C₄H₇⁺ and 49% C₃H₃O⁺. The OA was highly oxidized with an O/C of 0.88 333 on average. The OA was more oxidized during pre-monsoon than monsoon with higher O/C ratio (0.94 vs. 334 0.72) and lower H/C ratio (1.28 vs. 1.44) (Figure 8b). The oxidation states of OA during two pre-monsoon 335 periods were also different. The O/C during P2 (0.98) was higher than P1 (0.90) and the H/C was higher 336 during P1 than P2 (1.30 vs. 1.27)-, significant, P< 0.005, chi-square test). Correspondingly, the OA during 337 P2 contained higher contribution of $C_xH_vO_1^+$ (40.3 vs. 39.1%) significant, P< 0.005, chi-square test) 338 and $C_xH_yO_2^+$ (25.2 vs. 23.6%)%, significant, P< 0.005, chi-square test) ions than those during P1 (Figure 339 8c).

340

341 The diurnal variations of both O/C and OM/OC ratios presented higher values during late morning to 342 afternoon and lower values during early morning (Figure 9a). H/C presented an opposite trend. These 343 patterns suggested that OA was more oxidized during daytime which could be due to photo oxidation 344 and/or transport of highly oxidized OA during daytime. We examined the variation of elemental ratios with 345 OA mass concentrations, and found that the O/C increase was accompanied with the increase of OA mass 346 concentration (Figure 9b). This relationship could related with the different influences of wet 347 scavenging on more oxidized and less oxidized OA. In addition, this result likely suggested the importance 348 of transportation on the oxidized OA during afternoon due to the higher mass concentration occurring 349 frequently on afternoon time. However it was not possible that the polluted air massaerosol plume arrived 350 at Nam Co within several hours due to the long distance from source regions (more than 1000 km from 351 India). Previous studies have reported the presence of an aerosol layer between 6 - 18 km a.s.l altitude over 352 the Tibet Plateau during summer (Tobo et al., 2007; Vernier et al., 2011). He et al. (2014) examined the 353 vertical profiles of aerosol extinction coefficients measured with a Micro Pulse Lidar at Naqu, about 100 354 km east of Nam Co Station, and observed a maximum in aerosol extinction coefficient between 18-19 km 355 a.s.l during summer 2011. Recently, Gu et al. (2016) examined the aerosol compositions using the global 356 three-dimensional Goddard Earth Observing System chemical transport model (GEOS-Chem) and found 357 elevated concentrations of sulfate, nitrate, ammonium, BC, and organic carbon over the TP. Further, 358 observational and modeling studies have also shown that deep convection over the TP during daytime is 359 one of the important routes for tropospheric and stratospheric exchange of aerosols (Cristofanelli et al., 360 2009; Cristofanelli et al., 2010; Lin et al., 2016). Thus the enhanced aerosol concentrations during 361 afternoon could be possibly attributed to the mixed downward of aerosol layer at 16 18 km alti-362 during the growth of TP boundary layer. This type of transportation could not be captured by re-analysis of 363 data used in the back trajectory analysis likely due to the low time and spatial resolution.from the upper 364 troposphere during the growth of TP boundary layer. Recently, Xu et al. (2017) elucidate this mechanism at 365 Nam Co Station to explain the source of O3 and peroxyacetyl (PAN), and found strong downward motion 366 core events during pre-monsoon and monsoon periods which accompanied with increased O3 and PAN at 367 the ground level. We also performed a test using the Weather Research and Forecasting model (WRF)

368 model to check this kind of transport during P2 which indeed showed a downward motion core at the Nam

369 Co Station (Figure S7). Nevertheless, this hypothesis needs further validation in the future in this region.

371 3.5 The PMF apportionment on the OA

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373 The two OA factors determined by PMF have distinctly different mass spectra and diurnal patterns (Figure 374 10), including a less oxidized OOA (LO-OOA; O/C = 0.49) and a more oxidized OOA (MO-OOA; O/C = 0.49) 375 0.96). Both factors appeared to be secondary in nature. Our inability to separate a primary traffic-related 376 OA factor is consistent with the fact that $C_4H_9^+$ was a minor peak in the OA spectra (0.6% of the total 377 signal) and the averageits organic-equivalent concentration of $C_4H_{2}^+$ was averaged only 0.008 μ g m⁻³ during 378 this study (Figure 5a)-suggesting a minimal contribution from traffic-related primary OA. Collier et al. 379 (2015) reported that the average $fC_4H_9^+$ (fraction of total organic signal accounted for by $C_4H_9^+$) was 8.1% 380 in the mass spectrum of primary OA from vehicle emissions. Based on this relationship, even assuming 381 $C_4H_9^+$ was completely contributed by vehicle-derived OA, the average vehicle-OA concentration would be 382 only ~ 0.1 μ g m⁻³, or 7% of total OA mass, during our study. Note that C₄H₉⁺ fitted well in the PMF 383 analysis with very small residual, indicating that the signal of this ion was properly apportioned between 384 the two OOA factors.

385

386 The mass spectra of MO-OOA and LO-OOA were characterized by high peaks at m/z 44 (mostly CO_2^+) 387 and LO-OOA had a relative large peak at m/z 43 (mostly $C_2H_3O^+$) as well (Figure 10a). The contributions 388 of the $C_xH_yO_2^+$ ion category in these two mass spectra were 15.1% and 28.6%, respectively, while the 389 contributions of $C_x H_v O_1^+$ were 37.8% and 41.5%, respectively. The time series of LO-OOA correlated well 390 with $C_x H_v O_1^+$ and $C_x H_v^+$ ions, while MO-OOA correlated well with $C_x H_v O_2^+$ ions (Figure <u>\$758</u>). In 391 addition, the time series of MO-OOA correlated well with sulfate ($R^2 = 0.55$), BC ($R^2 = 0.54$) and less well 392 with nitrate ($R^2 = 0.33$), while LO-OOA correlated poorly with sulfate, BC and nitrate (Figure 10b and 393 \$859). These results highlight the oxidation degree of MO-OOA and LO-OOA. LO-OOA and MO-OOA 394 accounted for 41% and 59% on average, respectively, of total OA mass during this study (Figure 11a), and 395 their relative contributions varied across the study (Figure 2e). For example, LO-OOA accounted for 41% 396 of the total OA mass during P1, 24% during P2, and 67% during the monsoon period (Figure 11b). The 397 diurnal pattern of MO-OOA was characterized by higher concentrations during afternoon similar to those 398 of sulfate and BC. While the diurnal pattern of LO-OOA peaked at early evening time (Figure 10c). The 399 polar plot showed concentrated hotspots to the northwest of the sampling site for MO-OOA, and southeast 400 for LO-OOA (Figure 6), indicating that the sources of these two components were totally different. Based 401 on trajectory analysis (Figure 3), MO-OOA was likely closely related to long-range transport of air mass 402 from South Asiasouthwest, while LO-OOA could from relative shorter distance transport such as marine air 403 mass from south and regional background aerosol during the nighttime. Shen et al. (2015) reported that 404 there were significant aerosol source from biogenic emission during summer near the Nam Co Station. As

405 shown in Figure 12a and b, MO-OOA, which was highly oxidized (O/C = 0.96), appeared on the up-left 406 corner of the triangle plot while LO-OOA was in the middle part with an O/C ratio of 0.49. The high 407 oxidation degree of MO-OOA was likely related to extensive aging processes occurred during long-range 408 transport. The slope of linear fitting of all the data points in V-K diaphragm is -0.76 suggesting the 409 evolution of OA as carboxylic acid functionalization (Figure 12c).

410

411 Biomass burning emission is an important source in South Asia and could be transported to Himalayas and 412 TP during pre-monsoon (Engling et al., 2011; Kumar et al., 2011; Sang et al., 2013; Cong et al., 2015), 413 however there are not significant signals of biomass burning aerosol in our AMS results such as signals at 414 m/z 60 and 73 in mass spectrum of OA which were found to be associated with levoglucosan formed from 415 the pyrolysis of cellulose (Alfarra et al., 2007). The contribution of f60 (fraction of total organic signal 416 accounted for by m/z 60) for LO-OOA and MO-OOA were 0.2% and 0.3%, respectively, which were 417 similar with the global background level (less than 0.3%) suggested by Cubison et al. (2011). These results 418 suggest that OA, if ever partly originated from biomass burning emission, could have been highly oxidized 419 during transport. This behavior had been observed in a few studies that levoglucosan could be quickly 420 (within a few hours) oxidized after being emitted (Ortega et al., 2013). In addition, Zhou et al. (2017) 421 recently reported the observation of a highly aged BBOA factor with f60 < 0.3% in its mass spectrum, in 422 aged wildfire plumes that had gone through extensive photochemical oxidation.

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424 425

3.6 Sensitivity of the <u>aerosol</u> transport-of pollutants to synoptic process

426 Synoptic process is an important factor determining if the regional emissions can be transported to the TP. 427 It is interesting to know what kind of synoptic process is favorable for transporting the polluted air 428 massaerosol plume to the Himalayas and the TP. A 3-D Regional chemical transport Model (REAM) 429 coupled with the Weather Research and Forecasting model (WRF) was used to examine the chemical 430 evolution and regional transport of pollutants such as aromatics in this study. REAM has been used in 431 previous studies of the Tibet Plateau, and details about the model can be found in Zhang et al. (2017) and 432 supplementary material. REAM could capture some synoptic processes which cannot be simulated by the 433 normally used reanalysis data due to their low-resolution and the complexity of terrain in the Tibet Plateau 434 (Zhang et al., 2017). Figure 13 shows the distribution of simulated daily surface wind, 300 hPa 435 geopotential height fields and concentrations of reactive aromatics over the Tibetan Plateau during 30 May 436 - 13 June, 2015. During 30 May to 7 June corresponding to P1, there was a trough over the north 437 propagated from west to east and this low pressure induced increasingly stronger surface wind from India 438 to the TP, which could lead to transport of polluted air massaerosol plume as illustrated by the results of 439 model and AMS. The simulated concentration of reactive aromatics showed a peak during this period 440 (Figure <u>\$9\$10</u>). During 8-11 June, there was a weak ridge system over the north. Intensified Weak wind from northwest and south was observed as illustrated by HYSPLIT results (Figure 5b) and the simulated 441

442 concentrations of reactive aromatics were sharply decreased (Figure S9S10). After that, a weak low-443 pressure trough system was observed again. The increased concentrations of reactive aromatics were also 444 observed accompanying with intensified southern wind. Although these trends are basically consistent with 445 our AMS results, there were also significant differences (Figure S9S10). The possible reason was that the 446 weak troughridge during P2 intensified enhanced the wind from north and weakened the wind from west 447 other thanand south where a lot of biomass burning emission sources locatelocated (Figure 5b3b). Zhang et 448 al., (2017) suggested that a <u>cut-off</u> low-eut system from stratospherethe upper troposphere could be an 449 important driver for pollutionaerosol plume transport into the TP. In our study, the trough/ridge system 450 seems to be also an important factor affecting the transport of air pollutionaerosol plume from south and 451 west, although this effect tends to be weaker in summer than in the other seasons because the tropopause is 452 higher and stratospheric wave activity is weaker in summer.

453

454 **3.7 Atmospheric implications**

455

456 Our results have several potential implications to the atmospheric studies in the TP and Himalayas. Firstly, 457 it is useful for the accurate estimation of the radiative forcing of aerosols in this region and validation of 458 current model simulation results based on our observed chemical composition and mass loadings of fine 459 aerosols. Ji et al. (2015) estimate the radiative forcing from aerosols over the TP and Himalayas at the 460 surface level using a regional climate model (RegCM4.3); for carbonaceous aerosols, there are several 461 literatures that tested the model results, but all other species were referred to data available in the published 462 inventory with a coarse spatial resolution. Secondly, our findings are implicate to the estimation of have 463 implications for aerosol deposition on the glacier of this region and evaluation of subsequent impacts on the 464 melting of snow/ice (Yasunari et al., 2010). Thirdly, highly-time resolved aerosol data is very scarce in this 465 remote plateau, thus our data are valuable to validate modeling results regarding the transport of polluted 466 air massaerosol plume as demonstrated in section 3.6. At last, the transport mechanism of aerosol to the 467 inland of TP is less understood so far. Hindman and Upadhyay (2002) suggested that the vertical lifting due 468 to convection and subsequent horizontal mountain-valley wind could lead to the transport of aerosol from 469 Nepal to Tibet. Dumka et al. (2010) Dumka et al. (2010) also highlighted the important role of mountain-470 valley wind in the aerosol transport in the central Himalayas. The dynamic variations of aerosol chemical 471 species measured here, are likely helpful to elucidate the transport mechanism of pollutedhigh 472 concentration air mass. Nevertheless, this scientific issue required further detailed investigations in the 473 future.

474

475 **4. Conclusion**

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The average PM_1 loading measured at Nam Co during June 2015 was 2.0 µg m⁻³ with organics accounting for 68%, followed by sulfate (15%), black carbon (8%), ammonium (7%), and nitrate (2%). This mass 479 concentration was comparable to some AMS observations from mountain-top sites. The mass concentration 480 of PM₁ varied during different weather conditions with higher concentration during pre-monsoon and lower 481 concentration during monsoon. The pre-monsoon period could also be divided into two periods (P1 and P2) 482 based on meteorological conditions and aerosol chemical composition. During P1, PM₁ was characterized 483 with high contribution from OA and sulfate, while increased contribution of nitrate was observed during P2 484 with wet and cold weather conditions. All PM1 species had clear diurnal variations with OA, sulfate, BC, 485 and ammonium peaking during afternoon due to photochemical production of these species coupled with 486 transport of polluted air mass.aerosol plumes. Nitrate, however, peaked during the nighttime and early 487 morning, which was related to the high RH condition and low air temperature. The formation of nitrate was 488 highly correlated with transport of air masses from southwest under very low wind speeds, while the mass 489 concentrations of sulfate, OA, and BC were highly correlated with air masses from northwest and southeast 490 under higher wind speed conditions. OA was overall highly oxidized during the entire study with higher 491 O/C ratios during the pre-monsoon period. Based on PMF analysis, the OA was found to be composed of a 492 LO-OOA and a MO-OOA. LO-OOA was mainly associated with air masses originated from south, while 493 MO-OOA was mainly from northwest. MO-OOA dominated OA during the pre-monsoon period, while 494 LO-OOA dominated during the monsoon period. The transport mechanism of polluted airaerosol plume 495 was further investigated by using the REAM chemical model coupled with the WRF model. The polluted 496 airaerosol plume was found to be more easily transported to the TP and Himalayas during low pressure 497 trough weather.

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769

771 Table 1. Summary of AMS measurement at mountainhigh elevation sites around the world. The

772	mass concentration was for NR-PM ₁ .

Sites	Instruments	Lat.	Long.	Elev.	Date	Mass Conc.	Reference
Nam Co	HR-AMS	30.77°N	90.9°E	4730	Jun-15	1.84	
Menyuan	ACSM	37.61°N	101.26°E	3925	5 September – 15 October, 2013	10.8	Du et al. (2015)
Montsec	ACSM	42.05°N	0.73°E	1570	14 July 2011 - 23 April 2012	4.9	Rinaldi et al. (2015)
Jungfraujoch	ToF- ACSM	46.55°N	07.98°E	3580	27 July 2012 - 2 October 2013	0.55	Fröhlich et al. (2015)
Mt. Cimone	HR-AMS	44.18°N	10.7°E	2165	June - July 2012	4.5	Rinaldi et al. (2015)
Mt. Whistler	HR-AMS	50.01°N	122.95°E	2182	Apr - May 2006	1.91	Sun et al. (2009)
Puy de Done	cToF-AMS	45.77°N	2.95°E	1465	Sep 2008 - June 2010	2008 Autumn: 7.82 2008 Winter: 5.58 2009 Summer: 27.594	Freney et al. (2011)
Mt. Bachelor	HR-AMS	43.98°S	121.69W	2800	25 July - 25 August, 2013	15.1	Zhou et al. (2017)
Sub- Antarctic Bird Island	HR-AMS	54.00°S	38.00°W	ND	November and December, 2010	0.46	Schmale et al. (2013)



Figure 1. Location map for (a) the Tibetan Plateau and (b) Nam Co Station colored by the altitude.

- (c) The profile of altitude from costal area to Nam Co Station (vertical dash line). The red arrow
- in the map represent the possible wind direction.



781

Figure 2. The combo plot of the data of the Nam Co study including (a) the meteorological conditions (T: air temperature; RH: relative humidity; Precip.: precipitation), (b) the variation of WS (wind speed) colored by WD (wind direction), (c) the temporal variation of mass concentration of PM_1 species and the average contribution each species (pie chart), (d) the mass contribution of each PM_1 species and the total mass concentration of PM_1 , and (e) the mass contribution of PMF results (section 3.5). Three periods based on the meteorological conditions were marked.



796 colored by pressure. Fire spot observed by MODIS (https://firms.modaps.eosdis.nasa.gov) and average

797 wind rose plot during each period are also shown.



800 Figure 4. The comparisons of PM_1 species of the mass concentration (left axis) and the ratio between them

- 801 (right axis) between (a) pre-monsoon and monsoon and (b) P1 and P2.
- 802
- 803





806 Figure 5. Diurnal variations of each species and weather conditions (RH: relative humidity; Tem: air

temperature; and WS: wind speed) during three periods of the study. Note that the signals of sulfate,ammonium, and BC are increased by a factor for comparison.



811 Figure 6. Bivariate polar plots that illustrate the variations of the concentrations (colored) of each species as

812 a function of wind speed (m s^{-1}) and wind direction during different periods of the study.



Figure 7. The high aerosol loading periods based on nitrate (> average + 2σ) accompanying with meteorological data.



820 Figure 8. (a) The average mass spectrum of organic aerosol, (b) the average ratios of H/C and O/C during

821 different periods, and (c) the average contribution of six ion categories during different periods.





828 of mass concentration of organic aerosol.







ב 15

O/C = 0.49, H/C = 1.34, N/C = 0.000, OM/OC = 1.77

O/C = 0.96, H/C = 1.04, N/C = 0.001, OM/OC = 2.36

LO-OOA

MO-OOA

1.0

0.5

0.0 3

2

1

0

0.1

0.0 0.4

0.3 0.2

LO-OOA

MO-OOA

1.0

0.5 0.0 3

2

0

0.1

0.0 0.4

0.3

0.2

24

Nitrate

Sulfate

Nitrate

Sulfa

LO-OOA

MO-OOA

LO-OOA

MO-OOA

Nitrate

Sulfate

Nitrate

Sulfate

LO-OOA

MO-OOA

LO-OOA

MÓ-OOA

hoooc

20 24

16

11

200

833	Figure 10. PMF results of (a) the high resolution mass spectra colored by six ion categories for LO- and	Formatted: Justified
834	MO-OOA, respectively, (b) the-temporal variation and (c) diurnal-variations of twothe OA factors. The	
835	temporal and diurnal variations of corresponding comparison with tracer species (sulfate and nitrate-are	
836	also shown for comparisons), and (c) diurnal variations of the mass concentration of the OA factors and	
837	tracer species.	
838		







842 Figure 11. (a) The average mass contributions of two factors (a) during the study, (b) during the different

843 periods, and (c) the diurnal mass contribution of two factor (left axis) accompanying with the total organics

844 (right axis).



848 Figure 12. Scatter plots of (a) f44 vs. f43, (b) fCO₂⁺ vs. fC₂H₃O⁺, and (c) H/C vs. O/C for the OA. The dash 849 lines in (a) and (b) refer to a triangular region that encompasses ambient OOA factors determined from 850 PMF analyses of 43 AMS datasets ((Ng et al., 2010). The light solid lines with slopes of 0, -1, and -2 in (c) indicate the changes of H/C against O/C due to adding specific functional groups to an aliphatic carbon 851 (Heald et al., 2010); The light dash lines indicate the oxidation state (OS) of -2, -1, 0, 1, and 2, respectively 852 853 (Kroll et al., 2011a); The red and blue dash lines are derived from the right and left lines in the triangle 854 plot-; The heavy solid line indicate the slope of our measured data. The corresponding values of the OOAs 855 identified in this study are also shown in each plot.



Figure 13. Daily distribution of WRF-simulated surface wind and REAM-simulatedconcentrations of reactive aromatics over the Tibet Plateau during 30 May to 13 June, 2015.