



MIPAS observations of volcanic sulphate aerosol and sulphur dioxide in the stratosphere

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Abstract

10 Volcanic eruptions can increase the stratospheric sulphur content by orders of magnitude above the background level and are the most important source of variability of stratospheric sulphur loading. We present a set of vertical profiles of sulphate aerosol volume densities and derived liquid-phase H₂SO₄ mole-fractions for 2005–2012, retrieved from infrared limb emission measurements by the Michelson Interferometer for Passive Atmospheric Sounding (MIPAS) on board of the Environmental Satellite. The MIPAS aerosol dataset has been corrected for a possible altitude-dependent bias by comparison
15 with balloon-borne in situ aerosol measurements at Laramie, Wyoming. The MIPAS data of stratospheric sulphate aerosol is linked to MIPAS observations of sulphur dioxide (SO₂) with the help of Chemical Transport Model simulations. We investigate the production of sulphate aerosol and its fate from volcanically emitted SO₂ for two volcanic case studies: the eruptions of Kasatochi in 2008 and Sarychev in 2009, which both occurred in the Northern Hemisphere mid-latitudes during boreal summer. We show that the MIPAS sulphate aerosol and SO₂ data are qualitatively and quantitatively consistent to
20 each other. Further, we demonstrate that the lifetime of SO₂ is well described by its oxidation by hydroxyl radicals. While sedimentation of the sulphate aerosol plays a role, we find that the dominant mechanism controlling the stratospheric lifetime of sulphur after these volcanic eruptions at mid-latitudes is transport in the Brewer-Dobson circulation. Sulphur emitted by the two mid-latitude volcanoes resides mostly north of 30° N at altitudes of ~10–16 km, while at higher altitudes (~18–22 km) part of the volcanic sulphur is transported towards the equator where it is lifted into the stratospheric 'overworld' and
25 can further be transported into both hemispheres.

1 Introduction

Aerosol particles are omnipresent in the atmosphere and can affect climate, air quality, and atmospheric chemistry. In the stratosphere, aerosol particles are mainly composed of sulphuric acid and water (Kremser et al., 2016; Thomason and Peter, 2006), though organic material is demonstrated to also play a significant role in the upper troposphere and lower stratosphere
30 (Yu et al., 2016; Murphy et al., 2014). Stratospheric sulphate aerosol has the potential to directly cool surface temperatures



by backscattering parts of the incoming solar radiation. Estimates of the amount of stratospheric aerosol, and their evolution with time, are therefore important for climate change modelling studies. The negative radiative effect of stratospheric aerosol (Andersson et al., 2015; Brühl et al., 2015; Ridley et al., 2014; Santer et al., 2014; Solomon et al., 2011; Vernier et al., 2011) is discussed to be one of the causes for the global warming 'hiatus' experienced during the first decade of this century
5 (Haywood et al., 2014; Santer et al., 2014; Fyfe et al., 2013). Hofmann et al. (2009) observed an increase of stratospheric aerosol and speculated that this is due to anthropogenic emissions. Newer studies, however, show this increase to be connected more likely to a series of smaller and medium sized tropical volcanic eruptions (e.g. Neely et al., 2013). Following Vernier et al. (2011), the increase of stratospheric aerosol levels since 2002 is connected to a series of moderate eruptions of volcanoes especially in the tropics. These volcanoes directly injected sulphur up to 20 km into the stratosphere. Ridley et al.
10 (2014), and Andersson et al. (2015) noticed a strong contribution of aerosols in the lowermost stratosphere of the mid- and high latitudes to the volcanic aerosol forcing during the last decade. Understanding of stratospheric sulphur, its sources and sinks, and the processes involved in its conversion and transport is important in the framework of proposed climate engineering schemes (e.g. Niemeier and Timmreck, 2015; Rasch et al., 2008).

The main source gases of stratospheric sulphate aerosol during background conditions are sulphur dioxide (SO_2) and
15 carbonyl sulphide (OCS). Due to the longer lifetime of OCS compared to SO_2 , carbonyl sulphide has a relatively high flux across the tropical tropopause layer (TTL), its main entry pathway into the stratosphere. Crutzen (1976) first stated the essential role of OCS for stratospheric aerosol. E.g., Chin and Davis (1995), Thomason and Peter (2006), Brühl et al. (2012), and Sheng et al. (2015), agree on a major contribution of OCS. However, its exact contribution to stratospheric aerosol during background conditions is still in discussion. During volcanically perturbed times volcanically emitted SO_2 is the
20 dominant source for stratospheric sulphate aerosol and causes most of the variability in the stratospheric sulphur content. Some eruptions directly injected SO_2 into the stratosphere (e.g. Vernier et al., 2011). In volcanic emissions, SO_2 is the third most abundant emitted gas, after water vapour and carbon dioxide (von Glasow et al., 2009).

To study stratospheric sulphur a combination of observations and model simulations is useful. When analysing the vertical extent of the SO_2 and aerosol plumes after volcanic eruptions and their transport at different altitudes, vertically
25 resolved observations are needed. These are available from satellite limb measurements, such as the Michelson Interferometer for Passive Atmospheric Sounding (MIPAS), an instrument that was operational on Envisat (Environmental Satellite). From 2002 to 2012 the instrument delivered limb emission measurements in the infrared spectral region. From MIPAS several datasets that are relevant to the stratospheric sulphur content are already available. These are volume mixing ratios (VMRs) of OCS (Glatthor et al., 2015 and 2017), and SO_2 (Höpfner et al., 2013 and 2015). Here we present an
30 additional dataset of sulphate aerosol from MIPAS, and combine the MIPAS SO_2 and liquid-phase H_2SO_4 measurements with Chemical Transport Model (CTM) simulations to analyse the consistency of the two datasets, and the fate of volcanically emitted sulphur.

In Sect. 2 we first provide basic information on MIPAS, the MIPAS SO_2 dataset, and balloon-borne in situ data of aerosol volume densities used in this study. This is followed by a short description of the CTM and our model implementations. This



paper has several purposes. We introduce a new aerosol dataset, retrieved from MIPAS measurements in Sect. 3, and compare the data to independent measurements of aerosols. We further study the distribution of MIPAS sulphate aerosol in the period 2005 to 2012 and compare it to MIPAS SO₂. In Sect. 4 we perform a case study for two of the largest volcanic eruptions of the last decade in Northern Hemisphere mid-latitudes, which were measured by MIPAS. The volcanoes are Kasatochi (52.2° N/175.5° W) that erupted in August 2008, and Sarychev (48.1° N/153.2° E), which erupted in June 2009. We analyse MIPAS observations of SO₂ and stratospheric sulphate aerosol in comparison to CTM simulations, and study the sulphur mass contained in SO₂ and sulphate aerosol, together with the transport of their volcanic plumes. Finally, in Sect. 5 we draw final conclusions on the consistency between the MIPAS SO₂ and the new MIPAS sulphate aerosol dataset, in combination with our model results, and give a short summary of our findings.

10 2 Datasets and methods

2.1 MIPAS

2.1.1 Instrument

MIPAS (Fischer et al., 2008) is an infrared (IR) limb emission sounder that operated on ESAs satellite Envisat. The Fourier transform spectrometer measured high-resolution spectra emitted by the constituents of the atmosphere in the thermal IR, in the region 685 to 2410 cm⁻¹ (ESA, 2000). The instrument operated from July 2002 to April 2012, separated in two measurement periods. Here we only study data from the second measurement period. During this period, from January 2005 to April 2012, radiance profiles from 7 to 72 km altitude were measured, with an unapodized spectral resolution of 0.0625 cm⁻¹, a latitudinal distance of 420 km between two subsequent limb scans, and a vertical sampling step of 1.5 km in the upper troposphere / lower stratosphere region. Installed on a sun-synchronous polar orbiting satellite, at an altitude of about 800 km, MIPAS delivered data at around 10am and 10pm, local time. For the retrieval of sulphate aerosol volume densities, described in this paper, MIPAS level 1b calibrated radiances version 5 were used, as provided by ESA.

2.1.2 The SO₂ dataset

In this study, we use the MIPAS SO₂ dataset as described by Höpfner et al. (2015). Error estimations and a validation of the SO₂ dataset by comparison with satellite data from the Atmospheric Chemistry Experiment Fourier Transform Spectrometer (ACE-FTS) and other available SO₂ observations are provided by Höpfner et al. (2015). Single SO₂ profiles of this dataset have a total estimated error of around 70–100 pptv and a vertical resolution of 3–5 km. The MIPAS data are shown to be consistent with independent measurements from several aircraft campaigns within ± 50 pptv. With respect to satellite-borne data from ACE-FTS, MIPAS SO₂ shows a bias of -20 to 50 pptv during volcanically perturbed periods that differs with altitude, which is a reasonably small bias, given the natural variability of volcanic SO₂. For background conditions this bias lies within -10 to 20 pptv in the altitude region 10–20 km. For the analysis of volcanically enhanced



periods it is necessary to stress complications of the MIPAS SO₂ data up to a few weeks directly after the eruption when due to aerosol-related sampling artefacts the total mass of SO₂ was found to be strongly underestimated (Höpfner et al., 2015). Their study comprises a dataset of volcanically emitted SO₂ for 30 volcanic eruptions, as seen in the MIPAS measurements.

2.2 Aerosol in situ measurements from Laramie, Wyoming

5 To validate the new MIPAS aerosol dataset, we use aerosol volume density profiles that were derived from in situ measurements of stratospheric aerosol (Deshler et al., 2003) above Laramie, Wyoming (41° N, 105° W). The in situ measurements were completed with balloon-borne University of Wyoming optical aerosol counters and consist of size resolved aerosol concentrations from the surface to approximately 30 km. Measurements usually occurred between 6 and 9am, local time, with the measurement frequency varying from monthly to bi-monthly. Data are available from 1971 to
10 present. Over this time period three different primary instrument types were used. The latest style of the three was used initially in 2006, became the standard Laramie instrument in 2008, and was flown on quasi-Lagrangian balloons in Antarctica in 2010 (Ward et al., 2014). For the MIPAS validation the Wyoming measurements were confined to those made with this final instrument (Laser based counter, LPC), which measures particles with radii > 0.08–4.2 μm in eight size classes. To derive geophysical quantities from the size resolved aerosol concentration measurements it requires fitting a size
15 distribution to the data. In the past this has been done by choosing a subset of the measurements to fit either a unimodal or bimodal lognormal size distribution. The final size distribution selected is from that subset of the measurements which minimizes the root mean square error when the fitted distribution is compared to all the measurements. This approach has recently been changed to use laboratory measurements of the counting efficiency at each channel and then search the lognormal parameter space for the lognormal coefficients, which minimizes the error of the fitted distribution compared to
20 the measurements. In our study we use the volume density profiles that are derived from the fitted lognormal size distributions (unimodal or bimodal) to the measurements. The precision of these volume estimates is ± 40 % (Deshler et al., 2003).

2.3 Chemical Transport Model

The isentropic Chemical Transport Model used in our study (e.g. Sinnhuber et al., 2003; Kiesewetter et al., 2010) is
25 forced by temperature, wind fields, and diabatic heating rates from the ERA-Interim reanalysis (Dee et al., 2011). Horizontal transport on levels of constant potential temperature is derived from the wind fields, while vertical transport is calculated using the diabatic heating rates. The CTM employs the second order moments advection scheme by Prather (1986). The model domain covers 29 isentropic levels between 330 and 2700 K (~10–55 km), with a horizontal resolution of about 2.5° latitude x 3.75° longitude (Gaussian latitude grid).

30 A simple sulphur chemistry scheme is implemented, including OCS, SO₂ and H₂SO₄ as advected tracers. The H₂SO₄ tracer consists of gas-phase and liquid-phase H₂SO₄. In the presented simulations, we consider volcanic SO₂ from one



volcanic eruption as the only sulphur source per simulation, in order to study the SO₂ and sulphate aerosol after the eruptions of Kasatochi in 2008 and Sarychev in 2009 individually. The bottom boundary concentrations and initial fields are zero for SO₂ and H₂SO₄. The volcanic SO₂ is injected instantaneously into the model column of the volcano. Different SO₂ masses are injected into three altitude ranges (see Sect. 4.1), and SO₂ is uniformly distributed to the grid boxes per altitude range. The sulphur released from volcanic SO₂ by its reaction with OH builds H₂SO₄. As the interim product sulphur trioxide (SO₃), formed during the oxidation of SO₂, combines rapidly with water vapour to form sulphuric acid, it is not explicitly considered in the model. The sinks for the species are advective transport out of the model domain into the troposphere, removal of SO₂ by its reaction with OH, and gravitational settling for sulphate aerosol. An OH climatology of daily distributions was derived in a full chemistry run of the CTM (2003–2006), and for reaction rates the recommendations from JPL are used (Burkholder et al., 2015). For the sedimentation, equilibrium partitioning between gas- and liquid-phase is assumed for H₂SO₄ (Ayers et al., 1980) at each simulation time step. To determine the terminal velocity for the part of the sulphate aerosol that settles, velocity calculations follow Jacobson (1999). In the simulations presented here the sedimentation radius is fixed to one effective settling radius. The solution density of the aerosol is calculated online from the fraction of liquid-phase H₂SO₄ in the binary solution of the H₂SO₄–H₂O aerosol. The sulphur scheme hydrates the sulphate aerosol based on ambient water vapour content, which is fixed at 4.5 ppmv for the stratosphere. Sedimentation transports sulphate aerosol into the grid box below, or finally out of the model domain. All model results shown for H₂SO₄ only consist of the sulphate aerosol droplets, as the MIPAS measurements do not consider gas-phase H₂SO₄.

3 The MIPAS aerosol dataset

3.1 Aerosol retrieval from MIPAS limb-spectra

In previous analyses of mid-infrared observations by MIPAS/Balloon and MIPAS/Envisat it has been demonstrated that the limb radiance due to particles has two major contributing terms: (1) the thermal emission of the particles, and (2) the scattered radiation from the atmosphere and Earth's surface from below the tangent point (Höpfner et al., 2002 and 2006). The relative weights of these contributions differ with particle size and wavenumber. For particles sufficiently small compared to the wavelength ($d < \sim 1 \mu\text{m}$ in the mid-IR; Höpfner, 2004), the scattered contribution can be neglected such that only the thermal emission remains as major source of IR radiation. In this regime the radiance only depends on the total aerosol volume density. Typical sizes of the stratospheric aerosol layer particles are less than 1 μm in case of background and enhanced conditions due to medium sized volcanic eruptions (e.g. Deshler et al., 2003). Thus, our retrieval target is the altitude profile of volume densities, derived from each set of calibrated MIPAS limb-scan spectra.

For this study we have concentrated on the second MIPAS measurement period between January 2005 and April 2012. The retrieval model used is the KOPRA/KOPRAFIT suite, allowing to directly retrieve aerosol parameters from observed radiances by coupling a Mie-model with the line-by-line radiative transfer scheme (Stiller et al., 2002; Höpfner et al., 2002 and 2006). For aerosol composition we assume a 75 wt% H₂SO₄–H₂O solution, as the stratospheric sulphuric aerosol



composition typically varies between around 70 and 80 %, as obtained by equilibrium calculations (Carslaw et al., 1995) and observations (e.g. Doeringer et al., 2012). E.g. Kleinschmitt et al. (2017), calculating aerosol optical properties, Kremser et al. (2016), calculating sulphur fluxes, and Gao et al. (2007), calculating atmospheric volcanic aerosol loadings, also use a 75 wt% H₂SO₄–H₂O composition in their studies. The imaginary parts of various refractive index datasets in the mid-IR are displayed in Fig. 1. Here the used optical constants by Niedziela et al. (1999) for 75 wt% and 230 K (bold red line) are compared to data at other concentrations and temperatures by Niedziela et al. (1999) (upper panel), and Myhre et al. (2003) (lower panel). This particular data set has been chosen because in an evaluation of optical constants for sulphuric acid, Wagner et al. (2003) have found those datasets to be best consistent with observations in the aerosol chamber AIDA (Aerosol Interactions and Dynamics in the Atmosphere). The spectral range selected for the retrieval (1,216.5–1,219.5 cm⁻¹) is situated at the long wavelength end of MIPAS band B as indicated by the two vertical lines in Fig. 1. It lies within one of the atmospheric windows as can be seen by comparison to the limb-transmission curve (light grey) in Fig. 1. We have not chosen the windows at around 830 cm⁻¹ and 950 cm⁻¹ since at 1,220 cm⁻¹ the absorption by H₂SO₄ droplets is higher and the relative difference between the various sets of refractive indices is lower.

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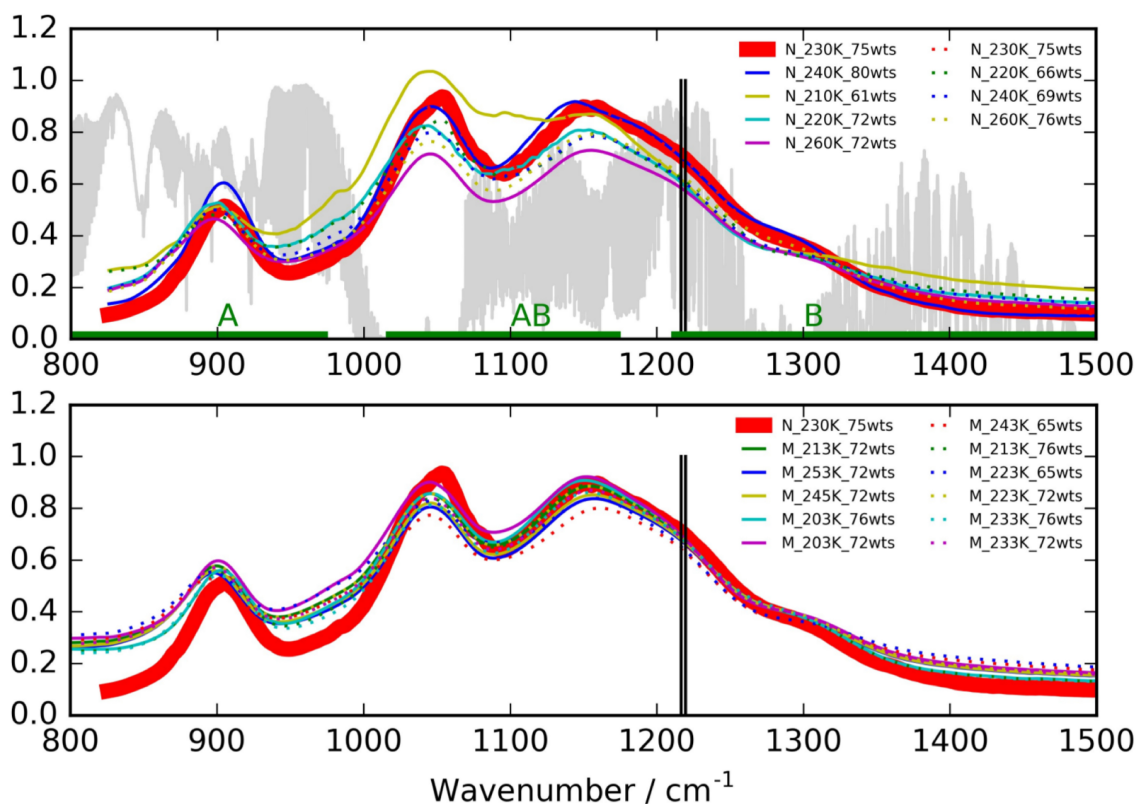


Figure 1. Imaginary parts of refractive indices for aqueous H_2SO_4 solutions of different concentrations (wts: weight % H_2SO_4) and different temperatures in the mid-infrared region. **(Top)** thin solid and dotted curves: Niedziela et al (1999), **(bottom)** Myhre et al. (2003). The bold red line indicates the dataset, and the two vertical lines the spectral window used in this study. A simulated limb transmission spectrum for 10 km tangent altitude for standard mid-latitude conditions is additionally plotted in the top row as well as the wavenumbers covered by MIPAS bands A, AB and B.

The retrieval has been set up as a multiparameter nonlinear least-squares fit of the calculated to the measured limb radiances of entire limb-scans (e.g. von Clarmann et al., 2003). Besides the target parameter, namely sulphate aerosol volume densities, further atmospheric fit-parameters of the retrieval are vertical profiles of spectrally interfering trace gases CH_4 , H_2O , O_3 , and HNO_3 . While zero initial guess profiles have been used for the volume densities, results from the IMK routine processing are taken for the trace gases (von Clarmann et al., 2009). As the atmospheric parameters are represented at denser altitude levels (1 km) than the vertical field-of-view (~ 3 km) and the vertical tangent point spacing (1.5 km) of MIPAS, constraints on the smoothness of the profile shape are introduced by regularization (Tikhonov, 1963; Steck, 2002). The retrieval of aerosol volume density is restricted to altitudes up to 33 km and the regularization strength has been adjusted such that its resulting vertical resolution is around 3 to 4 km. To cover instrumental uncertainties a spectral shift parameter and a radiance offset, constant over all wavenumbers and tangent altitudes, has been retrieved simultaneously to the



atmospheric quantities. For the analysis in this paper only data at altitudes with averaging kernel diagonal elements larger than 0.05 which are at least 1 km above the lowest tangent height are used.

An overview of the leading error components is presented in Fig. 2, with the assumed parameter uncertainties listed in the caption. The error contributions are estimated from a subset of a few hundred single cases by sensitivity studies using modified parameters or, in case of spectral noise, directly from the retrieval diagnostics. The total error changes with altitude from around 20 % ($0.09 \mu\text{m}^3\text{cm}^{-3}$) at 10 km up to over 40 % ($0.005 \mu\text{m}^3\text{cm}^{-3}$) at 30 km. It is dominated by the uncertainty of the optical constants resulting in 10–20 % error, followed by tangent pointing knowledge with 5–15 %. The error component resulting from spectral noise is rather constant with altitude in absolute terms of volume density and amounts to about $0.01 \mu\text{m}^3\text{cm}^{-3}$. Other instrumental errors that have been investigated but are not listed in Fig. 2 are uncertainties due to the knowledge of the instrumental line shape, and radiometric gain and offset calibration error. In the estimation of the radiative error no offset variations with tangent altitudes were considered, and, thus, compensated for by the retrieval approach. However, a tangent altitude dependent radiometric offset error caused e.g. by straylight in the instrument cannot be excluded (López-Puertas et al., 2009). We have not handled this uncertainty in the framework of error estimation but we have tried instead to compensate for it through a de-biasing of the dataset based on validation with in situ observations as described in Sect. 3.2.

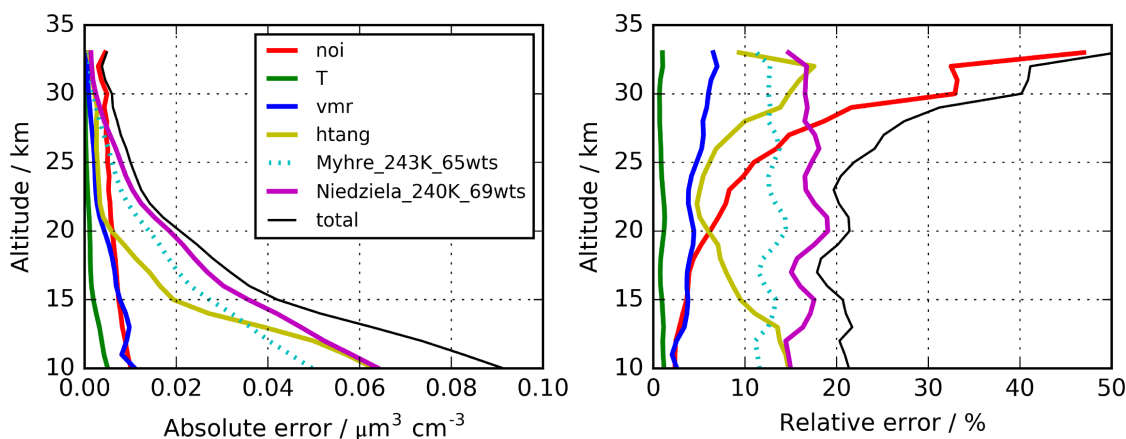


Figure 2. Altitude dependent estimated errors for the retrieval of $\text{H}_2\text{SO}_4\text{-H}_2\text{O}$ aerosol volume densities. Solid lines indicate the uncertainties used to calculate the ‘total’ error. Indicated errors are: ‘noi’: single scan spectral noise, ‘T’: temperature uncertainty 2 K, ‘vmr’: 10 % uncertainty of volume mixing ratios of interfering gases, ‘htang’: 300 m tangent altitude uncertainty, ‘Niedziela_240K_60wts’: use of optical constants by Niedziela et al. (1999) for 240 K and 69 wt% instead of Niedziela et al. (1999) for 230 K and 75 wt% H_2SO_4 . The dotted curves (‘Myhre_243K_65wts’) show the results when the optical constants of Myhre et al. (2003) for a temperature of 243 K and a concentration of 65 wt% are used instead of those by Niedziela et al. (1999) ($T=230$ K and 75 wt%).



Prior to the retrieval, a deselection of spectra affected by clouds has been performed via application of an established cloud filter method for MIPAS by Spang et al. (2004). To sort out optically thick clouds, but not all aerosol-affected spectra, this cloud filter has been applied with a cloud index limit of 1.7. Due to this loose setting of the cloud-filter artefacts caused e.g. by thin cirrus clouds, polar stratospheric clouds (PSCs) or volcanic ash remain in the dataset, which are all attributed to the retrieved 75 wt% H₂SO₄-H₂O aerosol volume density. Thus, further filtering of affected profiles has been necessary after completing the retrieval.

Two distinct features of strong enhancements with an annual cycle show up in the unfiltered dataset. The first feature is due to strong enhancements in presence of PSCs at the winter pole. To deselect PSC-affected profiles a filter is applied when temperatures in the altitude range 17–23 km drop below a threshold of 195 K polewards of 40°, for the Northern Hemisphere from 15 Nov–15 Apr, and for the Southern Hemisphere from 1 Apr–30 Nov. This temperature represents the nitric acid trihydrate (NAT) existence temperature at around 20 km, under typical stratospheric conditions for nitric acid (HNO₃) and water vapour (H₂O).

The second feature is assumed to be induced by thin cirrus clouds. It is present mainly in the tropics, at around 25° S–25° N, at altitudes between about 13–21 km. It reaches highest altitudes and is most intense above and in the vicinity of continents, and above the western Pacific. The vertical extent is smallest in boreal summer, and its vertical gradient towards lower aerosol volume densities is relatively strong, with no upward transport being observed. Both, the location in the tropics, in regions of strong vertical motions and convective clouds, and the relatively sharp decrease at higher altitudes towards increasing temperatures suggest that it is connected to the influence of ice particles. The ice-filter for MIPAS data by Griessbach et al. (2016) is applied on all retrieved MIPAS aerosol profiles to reduce the effect of spectra influenced by ice in the present dataset. This method consists of two steps to detect whether MIPAS spectra are influenced by aerosols, ice, clouds, ashes, or a clear sky (Griessbach et al., 2014 and 2016). First aerosols and clouds are identified, using a spectral window region that is sensitive to aerosols and clouds. Then ice clouds and aerosols are discriminated, using spectral windows with contrasting behaviour for ice and aerosols. This is then combined to brightness temperature difference correlations. In our dataset, we consider only retrieved values starting 4 km above the altitude of the uppermost spectrum that was flagged to have been influenced by ice. Further, the ash filter for MIPAS spectra by Griessbach et al. (2014), based on an ash detection threshold function, is applied in the same way as the ice filter, to filter out volcanic ash and mineral dust.

3.2 Validation and bias correction

To validate the new dataset, we compare the profiles of MIPAS aerosol volume density to in situ balloon measurements (Deshler et al., 2003). In situ measurements were carried out with laser based aerosol spectrometers (LPCs) from Laramie, Wyoming (41° N/105° W), between 6 and 9am, local time. In Fig. 3, profiles of the balloon measurements are shown. In comparison, MIPAS mean aerosol volume density profiles are presented, selected from a restricted area around Laramie, together with their normalised standard deviations (Fig. 3), which show the variability of the underlying profiles. MIPAS profiles are chosen for the day of the balloon flight, within a 5° latitude x 10° longitude distance around Laramie. Further, in



Fig. 4a we show the mean over all profiles that were retrieved from LPC measurements, as shown in Fig. 3 and the corresponding MIPAS profiles, together with the absolute (b) and relative difference (c) of the mean MIPAS profile to the mean profile from balloon measurements. In Fig. 4a the standard errors of the mean show the uncertainty of the bias.

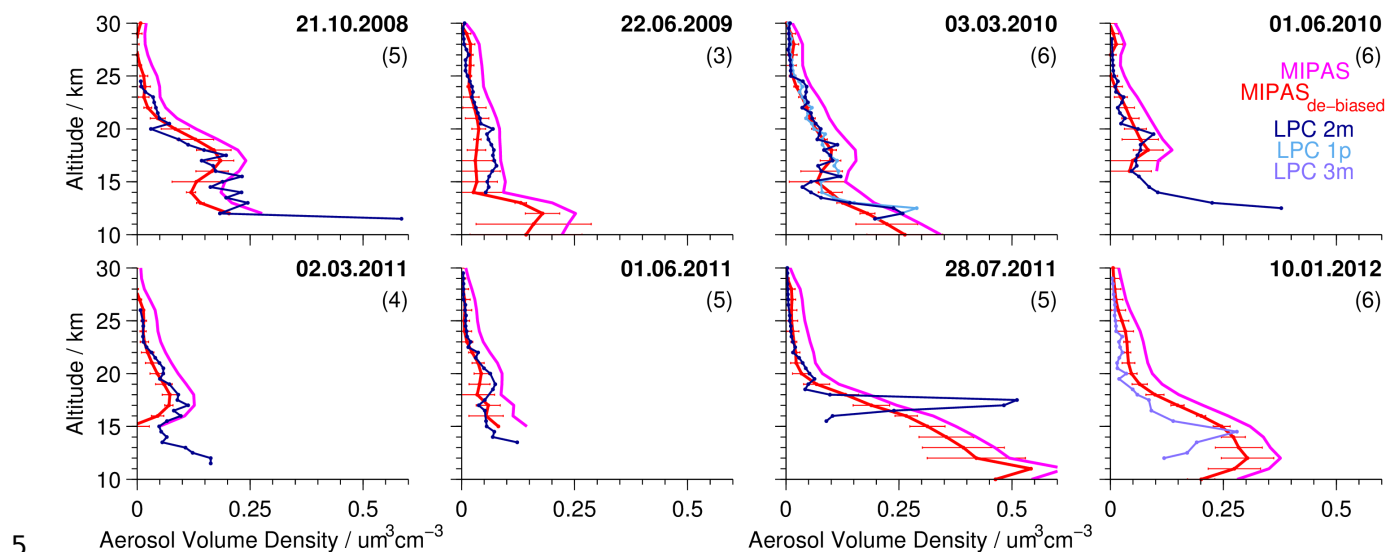


Figure 3. Profiles of aerosol volume densities from in situ and MIPAS satellite measurements. In situ data are balloon-borne data above the lapse rate tropopause for Laramie, Wyoming ($41^\circ\text{N}/105^\circ\text{W}$), measured by Laser Particle Counters (LPCs). MIPAS satellite data in pink (original data) and red (de-biased data). For all MIPAS data the ice filter by Griessbach et al. (2016) and ash filter by Griessbach et al. (2014) are applied. MIPAS profiles are averages of all available profiles for the indicated day, averaged in a $\pm 5^\circ$ lat \times 10° lon area. In brackets are the number of MIPAS profiles included in the average. For the de-biased MIPAS values the standard deviations of the mean profiles are shown.

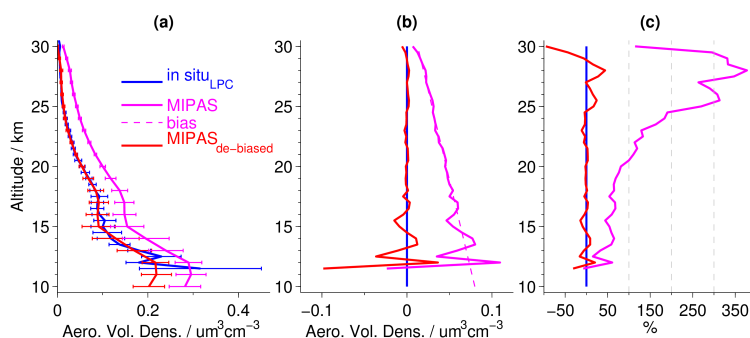


Figure 4. Aerosol volume densities by MIPAS and in situ data from Laramie, Wyoming. (a) mean profiles for data in Fig. 3, and their standard error. In red are de-biased MIPAS values, in pink the original data. (b) absolute and (c) relative differences of MIPAS to LPC in situ data. Differences are calculated for the mean profiles shown in (a). Dashed pink line: linear fit to the original MIPAS data between 18 and 30 km (pink solid line in panel b), used to reduce the bias towards the LPC in situ measurements.



Generally, the aerosol volume densities (Fig. 3 and 4) are highest in the lower stratosphere and then decrease towards zero at higher altitudes. As the balloon data has a higher vertical resolution, and the retrieval process for MIPAS profiles includes smoothing, the in situ data show finer structures. Compared to the balloon data, the original MIPAS aerosol volume densities show a positive bias in most profiles (Fig. 3) as well as in the mean profile (Fig. 4). This is most easily detectable at higher altitudes where profiles are relatively smooth. The offset amplifies towards lower altitudes (Fig. 4b). Aiming on a reduction of this positive offset, a height dependent de-biasing is performed on all single MIPAS profiles. The de-biasing is based on the in situ measurements carried out with laser based particle counters. MIPAS profiles show a consistent variation with height, compared to the LPC measurements. An additive linear de-biasing is applied, rather than a multiplicative correction, as the offset is expected to be caused by a possible altitude-dependent additive stray light error in the radiances (see Sect. 3.1). The offset-correction is represented in Fig. 4b (dashed line). It is the difference between a linear fit to MIPAS values at 18 to 30 km altitude, and the in situ data (Fig. 4b, pink and blue solid line, respectively). At lower altitudes, where profiles show more variability, this linear fit also suits well. The mean de-biased MIPAS profile (Fig. 4a) matches the in situ data and lies mostly in the range of the standard error of the mean of the in situ data. Further the absolute and relative differences to the balloon data are reduced significantly (Fig. 4b and c). Percentage differences are mostly below $\pm 25\%$. For the non-de-biased profile, at altitudes above around 20 km percentage differences increase strongly, due to very low aerosol volume densities, while at lower altitudes percentage differences are below about 100 %.

3.3 Time series of MIPAS sulphate aerosol and SO₂ for 2005 to 2012

To study the distribution of sulphate aerosol, as measured by MIPAS from 2005 to 2012, Fig. 5 (left) shows latitudinally resolved time series of liquid-phase H₂SO₄ mole-fractions, for various altitudes from 10 to 22 km. From the retrieved aerosol volume densities, the mole-fractions are calculated by assuming all the aerosol to be sulphate aerosol with a composition of 75 wt% H₂SO₄ and 25 wt% H₂O, and an aerosol density of 1,700 kg m⁻³. Strongest variability in the MIPAS sulphate aerosol data is caused by volcanic eruptions. In the Northern Hemisphere, strongest signatures of volcanic eruptions are due to the eruptions of Kasatochi (52.2° N/175.5° W) in August 2008, Sarychev (48.1° N/153.2° E) in June 2009, and Nabro (13.4° N/41.7° E) in June 2011, at altitudes from 10 to about 20 km. At low latitudes, tropical volcanoes such as Manam (4.1° S/145.0° E) in January 2005, Soufrière Hills (16.7° N/62.2° W) in May 2006, and Rabaul (4.3° S/152.2° E) in October 2006, increase the sulphate aerosol mole-fractions at higher altitudes, above 16 km. In the Southern Hemisphere the eruption of Puyehue-Cordón Caulle (40.6° S/72.1° W) in June 2011 has the strongest impact on the measurements, but is restricted to lower altitudes, below about 14/15 km. During the preceding years the mole-fractions are relatively low in the mid-latitudes of the Southern Hemisphere, at 10 to 12 km.

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Sulphur is released from OCS mainly in the tropics at altitudes between about 25 and 35 km (Brühl et al., 2012) and the sulphate aerosol that is built is transported towards mid-latitudes and lower altitudes.

3. In the mid-latitudes of the Northern Hemisphere, the sulphate aerosol is increased during boreal summer at around 10–12 km.

5 4. In the tropics at around 14–16 km, aerosol values are elevated, while they are very low below and above these altitudes, unless influenced by volcanic eruptions.

As SO₂ is the main precursor for stratospheric sulphate aerosol during volcanically perturbed times, we analyse similarities and discrepancies between the distribution of latitudinally resolved time series of MIPAS SO₂ at various altitudes for 2005–2012 (Fig. 5, right), and the new aerosol data (Fig. 5, left). Prominent features seen in the distribution of SO₂ mole-
10 fractions from single limb scans are described by Höpfner et al. (2015). It should be noted, that these data are useful mainly for the analysis of enhanced SO₂, rather than for background conditions, for which the MIPAS SO₂ dataset by Höpfner et al. (2013) is more suited. The distributions of SO₂ and sulphate aerosol show clear similarities, especially concerning volcanic plumes. Differences in the patterns result mostly from the longer residence time of sulphate aerosol in the stratosphere, compared to SO₂. Sulphate aerosol can reside in the stratosphere for several months up to several years, if it is neither being
15 transported back to the troposphere nor evaporated at higher altitudes. On the contrary, SO₂ has a stratospheric lifetime of a few weeks. Different point sources, such as volcanic eruptions, can therefore be distinguished more easily in the SO₂ measurements than in the aerosol data. Further discrepancies arise from the fact that sulphur is released from SO₂ over weeks, during the exponential decay of the latter, and can then be converted into sulphate aerosol. Elevated SO₂ amounts are therefore not instantly leading to elevated sulphate aerosol amounts, and the curve of enhanced sulphate aerosol is broader
20 and flatter than for SO₂.

In the Northern Hemisphere at low altitudes (< 12/13 km) during boreal summer a similar feature of increased VMRs is present in the SO₂ as in the sulphate aerosol data (point 3). A closer look at the monthly distribution of SO₂ and sulphate aerosol reveals no distinct patterns (for SO₂ see Höpfner et al., 2015); enhancements are spread over the entire Northern Hemisphere. In Höpfner et al. (2015) this feature could not be confirmed due to a lack of SO₂ in situ data. The presence of
25 similar enhancements in the aerosol data supports the hypothesis of the increased sulphur content at low altitudes in the Northern Hemisphere not being a retrieval artefact. Further, elevated values in the tropics at around 14–16 km, as seen in the MIPAS aerosol (point 4) are also present in the SO₂ data. These are localized mostly in continental regions, and the western Pacific, both for MIPAS sulphate aerosol and SO₂ (for SO₂ see Höpfner et al., 2015).

4 Volcanic eruptions of Kasatochi in 2008 and Sarychev in 2009

30 In this section a case study of MIPAS SO₂ and sulphate aerosol measurements and CTM model simulations for the two volcanic eruptions of Kasatochi and Sarychev (Kasatochi: 7 Aug 2008, 52.2° N/175.5° W; Sarychev: 12 Jun 2009, 48.1° N/153.2° E) is presented. Both volcanoes erupted at Northern Hemisphere mid-latitudes during boreal summer. MIPAS



satellite measurements are compared to CTM simulations, to study the fate of the emitted sulphur in terms of conversion from SO₂ to sulphate aerosol, and its transport and removal at altitudes between 10 and 22 km. As our intention is to study explicitly the sulphur per volcanic eruption, background values per model simulation are set to zero for both SO₂ and H₂SO₄, and no other sources than the volcanically emitted SO₂ of one volcanic eruption is included.

5 4.1 Sulphur mass in the Northern Hemisphere mid- and high-latitudes

In this section we aim at testing the agreement between measured SO₂ and liquid-phase H₂SO₄ masses, together with modelled data, in terms of the increase and decline of sulphur emitted by the volcanic eruptions of Kasatochi in August 2008 and Sarychev in June 2009, and the influence of sedimentation radius on the residence time of sulphate aerosol. Good accordance between the modelled and measured SO₂ masses is essential to test, by comparison with modelled sulphate aerosol, if the measured aerosol is quantitatively and qualitatively consistent with its measured precursor.

In Table 1, SO₂ masses for three altitude regions, as used for the CTM simulations in the present study, are given (labelled 'present study'), together with comparisons to volcanic SO₂ masses from the literature. The simulations result in good agreement between measured and modelled SO₂. The upper injection limit for the volcanic emissions in the CTM is set to 19 km, an altitude limit derived from comparisons with MIPAS SO₂. The main part of SO₂ per eruption is emitted into the altitude region from 10 to 18 km, and only few percent of the SO₂ masses are injected into altitudes above 18 km. In the case of Kasatochi, our model is run with the SO₂ masses by Höpfner et al. (2015), reduced by their given uncertainties. Höpfner et al. (2015) derived volcanic SO₂ masses for three altitude regions from 10–14 km, 14–18 km, and 18–22 km by exponential extrapolation of the MIPAS SO₂ masses back to the eruption day. This method was applied as in the first month after the eruption MIPAS underestimates the SO₂ (Höpfner et al., 2015). Error bars of the extrapolated values were found to be relatively large (Höpfner et al., 2015; presented also in Table 1). In the case of Sarychev, the SO₂ mass used in the present study lies below the given error bars.

When comparing the SO₂ masses from different studies, it has to be pointed out, that the SO₂ masses are generally not derived for the same altitude regions. Höpfner et al. (2015), Brühl et al. (2015), and the present study are not totally independent from each other, as they are entirely or partly based on the same MIPAS SO₂ data by Höpfner et al. (2015). The SO₂ masses in our study lie below all studies but Brühl et al. (2015) for Kasatochi, and in the range of the other publications for Sarychev. The wide range of SO₂ masses in Table 1 shows the difficulties and uncertainties related to the determination of volcanically emitted SO₂.



Table 1. Volcanically emitted SO₂ masses from various publications for Kasatochi in 2008 and Sarychev in 2009. For Höpfner et al. (2015) the given total uncertainty is the sum of the uncertainties per altitude range (Table 3 therein). In the case of Pumphrey et al. (2015) pressure levels are given. These represent the highest pressure used for vertical integration.

	Kasatochi in 2008		Sarychev in 2009		
	SO ₂ mass Gg	Height range	SO ₂ mass Gg	Height range	
Present study	677	10–19 km	Present study	768	10–19 km
	518	10–14 km		401	10–14 km
	124	14–18 km		362	14–18 km
	35	18–19 km		5	18–19 km
Höpfner et al. (2015)	898 (± 221)	10–22 km	Höpfner et al. (2015)	1,474 (± 357)	10–22 km
	645 (± 127)	10–14 km		888 (± 293)	10–14 km
	210 (± 86)	14–18 km		542 (± 60)	14–18 km
	43 (± 8)	18–22 km		44 (± 4)	18–22 km
Brühl et al. (2015)	376		Brühl et al. (2015)	562	
Pumphrey et al. (2015)	1,350 (± 38)	215 hPa	Pumphrey et al. (2015)	1,160	147 hPa
Clarisse et al. (2012)	1,600			571 (± 42)	215 hPa
Karagulian et al. (2010)	1,700		Clarisse et al. (2012)	900	
Prata et al. (2010)	1,200		Haywood et al. (2010)	1,200 (± 200)	
Kristiansen et al. (2010)	1,000	> 10 km	Carn et al. (2016)	1,200	
Krotkov et al. (2010)	2,200				
Corradini et al. (2010)	900–2,700				
Thomas et al. (2011)	1,700				
Carn et al. (2016)	2,000				

5 Time series of sulphur mass contained in SO₂ and sulphate aerosol are shown in Fig. 6, for observations by MIPAS and simulations by the CTM. The sulphur content is shown for three altitude regions, from 10.5–14.5 km, 14.5–18.5 km and 18.5–22.5 km, for 30° N–90° N, including the latitude of the eruptions (~50° N). For liquid-phase H₂SO₄, four model results are presented for each volcanic eruption. The simulations differ concerning the implemented effective sedimentation radius (0, 0.1, 0.5, and 1 µm). In Fig. 6a–c, the sulphur mass is shown for SO₂ and sulphate aerosol separately, while Fig. 6d–f is
 10 the total sulphur contained in SO₂ and sulphate aerosol. The total simulated sulphur mass is not influenced by chemical sulphur removal, but only by removal due to transport by advection and sedimentation.

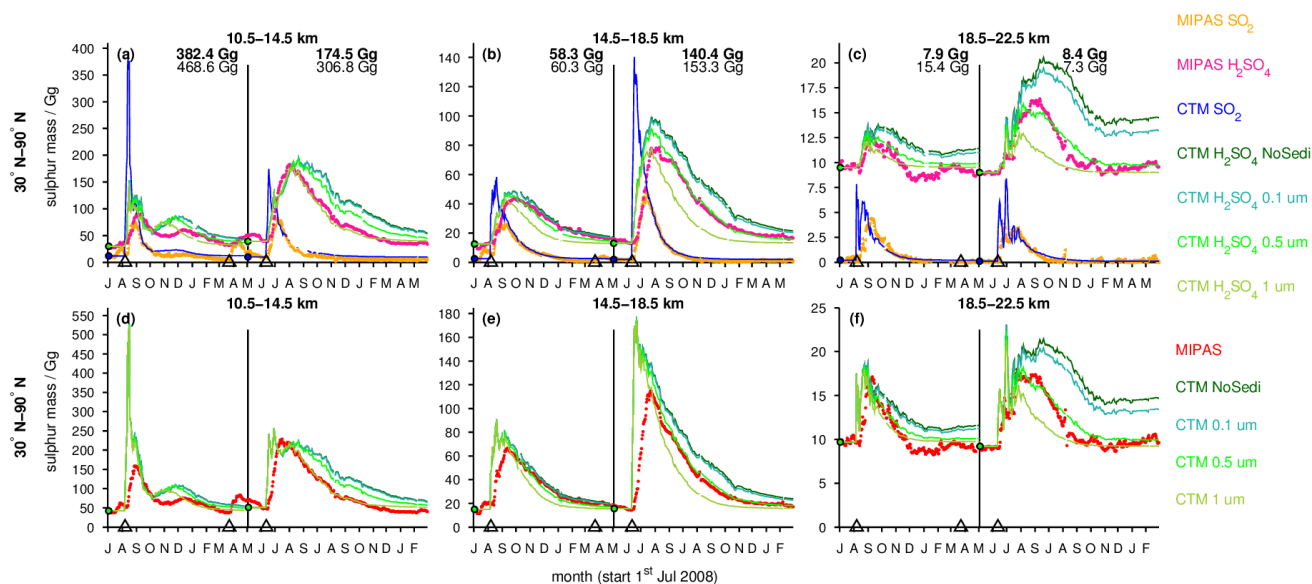


Figure 6. Sulphur mass contained in SO₂ and sulphate aerosol after the eruptions of Kasatochi (7 Aug 2008) and Sarychev (12 Jun 2009), from MIPAS measurements and CTM simulations. Presented are 5-days running means for 30° N to 90° N, in three altitude regions (10.5–14.5 km, 14.5–18.5 km, 18.5–22.5 km). For the CTM four simulations for H₂SO₄(liq) are shown with different effective sedimentation radii (0, 0.1, 0.5, and 1 μm). All simulations are carried out with the volcanic SO₂ masses from Table 1 ('present study'). A constant background is derived from the measured data and added to the simulated sulphur masses per altitude region (circles). Model simulations for Kasatochi and Sarychev are separated by a vertical black line in May 2009. **(a) to (c):** sulphur mass per species. **(d) to (f):** sum of sulphur mass in SO₂ and sulphate aerosol. The numbers in (a) to (c) show the peak values of sulphur mass contained in simulated SO₂ on locations covered also by MIPAS (bold), and for all available model data. Indicated by black triangles are the eruption days of Kasatochi (7 Aug 2008, 52.2° N/175.5° W), Redoubt (23 Mar 2009, 60.5° N/152.7° W), and Sarychev (12 Jun 2009, 48.1° N/153.2° E).

To analyse the measured and simulated data, datasets of sulphur mass densities (SMD = mass per unit volume) are re-sampled on a common grid with 1 km vertical spacing and a horizontal resolution that equals the model grid. On this new grid, the same data basis is used for the measured and simulated data, dismissing all 'grid cells' for which either only MIPAS or only CTM data are available. For MIPAS aerosols, SMDs are calculated from the primarily retrieved volume densities, using an assumed aerosol density of 1,700 kg m⁻³, and a binary solution of 75 wt% H₂SO₄–H₂O, while for MIPAS SO₂ and the CTM data SMDs are calculated from the measured and simulated mole-fractions. Sulphur masses are then derived from 5-days running zonal means of SMDs, by multiplication with the corresponding air volume of the new grid.

Generally, when calculating an integrated mass, high data coverage is crucial to prevent underestimation, therefore we use a 5-days running zonal means. Zonal mean values, used to calculate sulphur masses, are derived using a method of increasing area averaging (see Appendix), to reduce the bias of mean values due to a non-uniform data coverage. Even though high data coverage is very important, we use a method that dismisses available data and information, as the same



basis of available values is used for MIPAS and the CTM. This is appropriate when analysing the agreement between the data. Data is especially dismissed for the CTM. Thus we also provide some information on modelled sulphur masses derived from the non-co-located data (Fig. 6a–c). The impact of missing data is strongest in the lowermost altitude region presented here. For MIPAS this is mainly due to the presence of clouds and ash, which were filtered out using the cloud filter by Spang et al. (2004) in the case of SO₂ and partly filtered out in the case of aerosol, and additionally the ice and ash filters by Griessbach et al. (2016 and 2014, respectively) for the aerosol retrieval. The CTM has low data coverage at lower altitudes due to its isentropic vertical grid. Interpolation to geometric heights starting at 10 km, produces missing values at altitudes up to 13 km, especially in mid- to high-latitudes, and in the presence of strong vortices.

To ease visual comparisons of measured and modelled sulphur mass in Fig. 6, a constant background is added to the model results, as only volcanic sulphur is considered in these simulations. The background mass is chosen considering the mass derived by MIPAS before the volcanic eruption in the region of interest, per altitude and latitude bin. This does not necessarily represent normal background conditions, but unmasks the anomalies caused by the volcanoes.

Concerning the measured and modelled SO₂ masses after the eruptions of Kasatochi and Sarychev (Fig. 6), comparisons show that until about one month after the eruptions, the SO₂ mass is by far underestimated by MIPAS. This underestimation of SO₂ was stressed by Höpfner et al. (2015), when comparing MIPAS SO₂ to measurements by the Microwave Limb Sounder (MLS), on board Aura (Pumphrey et al., 2015). It is mainly due to the presence of particles, that hinders MIPAS SO₂ measurements in largely eruption-affected air-parcels and causes a sampling bias towards less volcano-affected air parcels. Through our model simulations we confirm this bias, and the related time scale found by Höpfner et al. (2015).

The decay of SO₂ is well simulated by the CTM, in comparison to the MIPAS measurements. Only oxidation by OH is considered in the model, and we see that the decay of SO₂ can adequately be described by this mechanism. Other processes, as decay by photolysis or reaction with O are not considered, and following the good agreement between measurements and model results, can be neglected at the temporal scale and region of interest. Inside volcanic plumes chemistry interactions might lead to changes in SO₂-lifetimes (Bekki, 1995). When a high amount of SO₂ gets depleted by hydroxyl radicals, the concentration of the radicals might decrease, which could reduce the speed of further depletion. The good accordance between MIPAS measurements and CTM simulations, which do not account for any feedback on the OH concentrations, suggests, that even if such interactions occurred, they did not produce a strong impact on the timescale of months and larger spatial scales.

To investigate the effect of particle sedimentation on the residence time of sulphur after the volcanic eruptions, model simulations with different effective sedimentation radii are performed, including one simulation without any sedimentation. The radii lie in the range of aerosol size distributions, as observed by Deshler et al. (2003) and Deshler (2008) for volcanically perturbed periods, and one constant radius is applied for all H₂SO₄ droplets per simulation. Fig. 6 shows the influence of varying the gravitational settling between no settling, and radii of 0.1, 0.5, and 1 μm. The amount of sulphate aerosol removed by sedimentation increases with growing particle size, while the time needed for the removal increases for smaller sedimentation radii. The sulphur mass contained in liquid-phase H₂SO₄ from a simulation with a particle radius of



0.1 μm differs little from a simulation without sedimentation, while radii of 0.5, and 1 μm show an increasing impact. In the middle and uppermost altitude region, the best agreement between simulated and measured aerosols is found for a radius of 0.5 μm , for both eruptions. At 10.5–14.5 km, especially in the case of Sarychev, the simulations show temporal disagreement to the decrease of measured aerosol. Here, sulphate aerosol simulated with a radius of 1 μm compares better. A larger
5 sedimentation radius seems more appropriate at lower altitudes, as heavier particles can settle faster, and can be removed more rapidly than smaller and lighter particles. These can float in the atmosphere or undergo ascent. The particle size distributions of aerosols can further show natural variation for different volcanic eruptions; therefore some differences in the agreement between modelled and measured data when studying different volcanic eruptions can be expected. In general, we conclude that a particle radius of 0.5 μm gives a satisfactory fit between the measurements and simulations for the purpose
10 of studying sulphur mass and sulphur transport in the Northern Hemisphere. Hence, we base all following model results on the CTM runs with a sedimentation radius of 0.5 μm .

We conclude from the comparisons between measured and simulated SO_2 and sulphate aerosol, that the amplitude of the peak of liquid-phase H_2SO_4 and its removal from the studied altitude regions, as measured by MIPAS, is consistent with the measured SO_2 , both qualitatively and quantitatively. In the model, sulphur is released from SO_2 , due to its reaction with OH,
15 and sulphate aerosol is consequently built. Modelled SO_2 , that fits well to MIPAS SO_2 measurements in terms of amplitude and decay, releases sulphur and builds H_2SO_4 that in turn matches well to MIPAS sulphate aerosol in terms of amplitude and decrease.

Further, we find that the dominating process on the fate of volcanic sulphur is transport by the Brewer–Dobson circulation out of the region of interest. This becomes obvious when comparing the total modelled sulphur with and without
20 sedimentation to the observed sulphur mass (Fig. 6d–f). In the case of the CTM this excludes all influence by chemical reactions on the removal of volcanic sulphur. Even though consideration of sedimentation of sulphate aerosol with a sedimentation radius between about 0.5 and 1 μm further improves the agreement between model results and observations in 10–22 km altitude, the modelled sulphur mass without sedimentation already compares rather well with the measured sulphur mass. Hereby we see that the removal is dominated by advection rather than sedimentation.

25 In the altitude region of interest, from around 10 to 22 km height, supplementary processes, as the photolysis of gas-phase H_2SO_4 , that is important at altitudes above 30 km (Vaida et al., 2003; Brühl et al., 2015), or a meteoritic dust sink (Brühl et al., 2015), are not considered. Other processes, such as the evolution of sulphate aerosol through microphysical processes, as nucleation, coagulation, condensation, or evaporation, and sedimentation of particles with different sizes can play a role in our region of interest. However, comparisons of simulations and measurements show that these processes are
30 not essential to study the fate of sulphur emitted by Kasatochi in 2008 and Sarychev in 2009.

4.2 Sulphur transport

The Kasatochi eruption directly injected a large amount of SO_2 especially into the altitude region between 10 and 14 km (Table 1). Fig. 7 displays vertically resolved time series of SO_2 and liquid-phase H_2SO_4 mole-fractions, for 30° N to 90° N,



as measured by MIPAS (7a–c) and modelled by the CTM (7d–f), together with their sum (c and f). Both, SO₂ and sulphate aerosol show a separation of the plume into a lower and upper part, in the measured and simulated data. While the lower part is removed relatively fast from the altitude range of observations, starting at 10 km, the upper part moves downward with time, following the Brewer-Dobson circulation, and reaches 10 km after a few months. As the descent is seen in sulphate aerosol and SO₂, it is not primarily driven by sedimentation. In Fig. 6, the second increase of sulphur mass in the altitude region 10.5–14.5 km is also due to the descending part of the volcanic plume. A similar pattern of subsidence was found by Andersson et al. (2015, Fig. 3 therein) when studying aerosol scattering measured by CALIPSO (Cloud-Aerosol Lidar and Infrared Pathfinder Satellite Observations). Additionally shown in Fig. 7 is the eruption of Sarychev in 2009. No separation is noticeable, vertically the sulphur is distributed rather homogeneously. In terms of downward transport in the extra-tropics from higher altitudes and the removal from the stratosphere, SO₂ and sulphate aerosol from the Sarychev eruption evolve quite similar to the Kasatochi eruption.

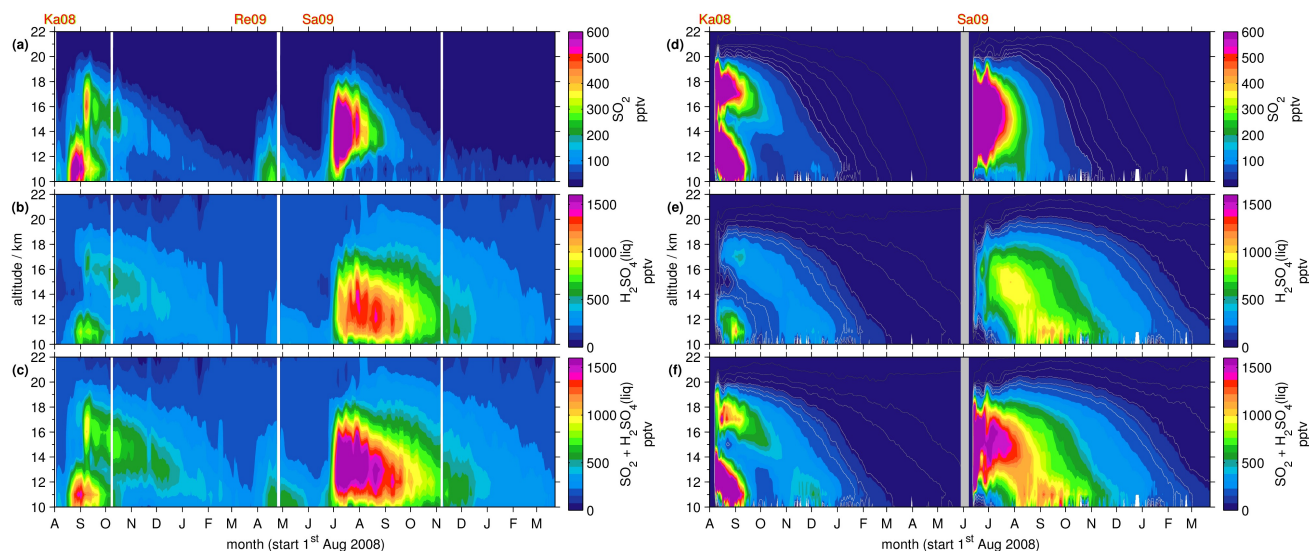


Figure 7. SO₂, liquid-phase H₂SO₄, and total sulphur (SO₂ + H₂SO₄) volume mixing ratios by MIPAS (a–c) and the CTM (d–f). Vertically resolved time series of SO₂ and sulphate aerosol for 30° N–90° N, starting on the 1 Aug 2008 (area weighted 5d running means). Indicated are the days of the eruptions of the volcanoes Kasatochi in Aug 2008, Redoubt in Mar 2009, and Sarychev in Jun 2009. The colour scale is restricted to 0–800 pptv (SO₂), and 0–1,600 pptv (sulphate aerosol, and sum of SO₂ and sulphate aerosol). Lower and higher values are assigned with the respective limiting value. For the CTM contour lines in dark to light grey represent 1, 5, 10, 20, 30, and 40 pptv in the case of SO₂, and 5, 25, 50, 100, 150, and 200 pptv for sulphate aerosol, and the sum of SO₂ and sulphate aerosol. For the CTM two simulations are shown, from August 2008 to May 2009, and from June 2009 to May 2010, separated by a bold grey line.

Figure 8 shows vertically resolved measurements and simulations of SO₂ and liquid-phase H₂SO₄ for 0° N–30° N. Some of the sulphur emitted at around 50° N reaches low latitudes. While the SO₂ is removed rather fast, compared to sulphate



aerosol, the sulphate aerosol resides in the tropics for many months and moves upwards with time. The modelled sulphate aerosol with a sedimentation radius of 0.5 μm behaves in a similar way as the measurements, moving slightly upwards with time. In comparison, a simulation with 1 μm effective sedimentation radius shows a rather horizontal transport and faster removal, while the simulated lifting is stronger than in the measurements, when not considering sedimentation. Due to

5 uncertainties in modelled ascent speeds (e.g. Pommrich et al., 2010; Liu et al., 2013), the particle radius that is most suitable to reproduce MIPAS measurements by CTM simulations is not necessarily the best estimate when performing similar analyses with different models or meteorological driving data. In the MIPAS data of the tropics, where the tropopause height is relatively constant at around 16–17 km, a clear transition from elevated sulphur mole-fractions in the troposphere to lower sulphur content in the stratosphere is observed during times of weak volcanic influence (May–Jun 2009, Fig. 8c).

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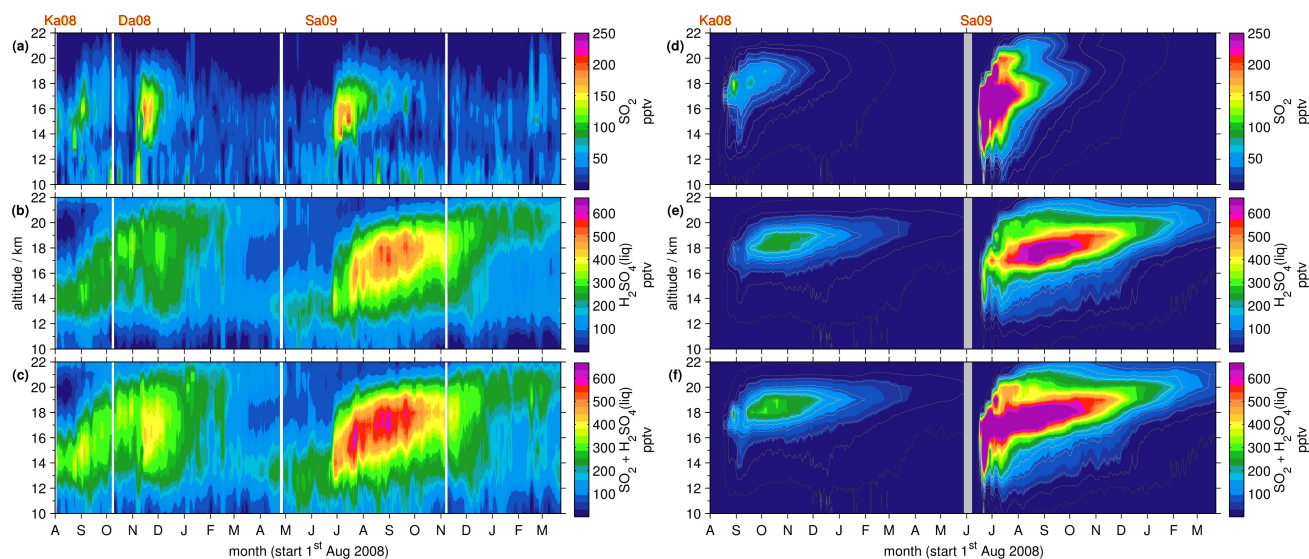


Figure 8. As Fig. 7, but for 0° N–30° N. The colour scales are restricted to 0–200 pptv (SO_2), and 0–800 pptv (sulphate aerosol, and sum of SO_2 , and $\text{H}_2\text{SO}_4(\text{liq})$). Additionally the volcano Dalafilla is indicated in Nov 2008.

In Fig. 9 and 10 time series of latitudinally resolved mole-fractions show the transport of SO_2 and sulphate aerosol at different altitudes. We present time series of 5d running zonal mean mole-fractions for the Northern Hemisphere, at altitudes from 10 to 22 km, both for MIPAS measurements and CTM simulations. As the model has very poor coverage at 10 km,

15 results are not shown for the model at this altitude. For the eruption of Kasatochi, the separation of the plume and downward transport of the upper part is notable at mid- to high-latitudes, most easily visible for sulphate aerosol (Fig. 10).

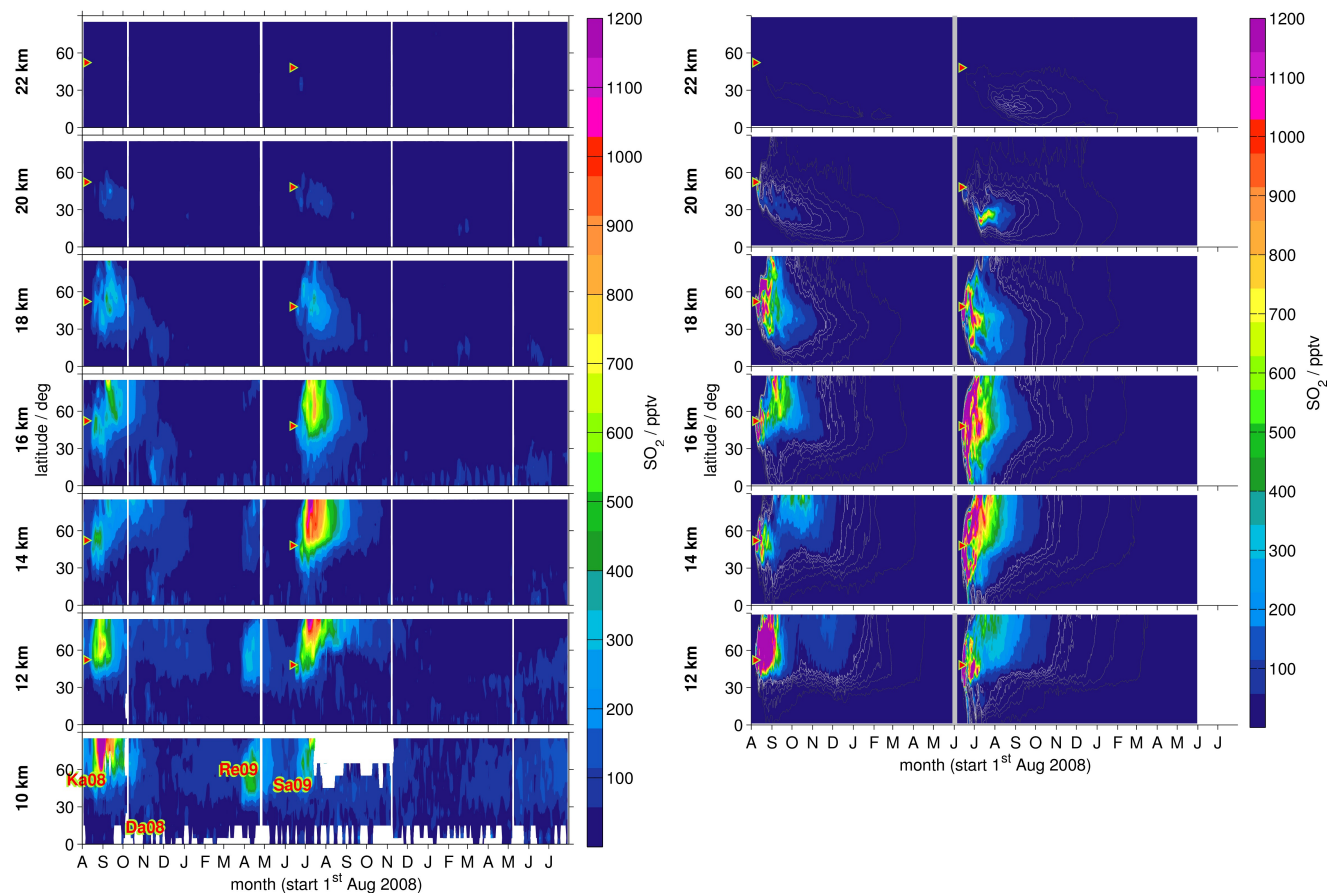


Figure 9. Time series of SO_2 from MIPAS measurements (**left panels**) and CTM simulations (**right panels**). Shown are volume mixing ratios after the eruption of Kasatochi (7 Aug 2008) and Sarychev (12 Jun 2009), as zonal mean for the Northern Hemisphere, at various altitudes (10 to 22 km). For MIPAS 5d running means for 10° latitude bins are calculated, while for the CTM daily values are shown for $\sim 2.5^\circ$ latitude bins. The colour code is restricted to 0–1,200 pptv. Values exceeding these limits are assigned with the limiting value, respectively. Time and location of the eruptions are indicated by red triangles. (left) Indicated are the eruptions of Kasatochi and Dalaffilla in 2008, and Redoubt and Sarychev in 2009. (right) The two simulations are separated by a grey line. Black to white contour-lines denote 1, 5, 10, 20, 30, and 40 pptv.

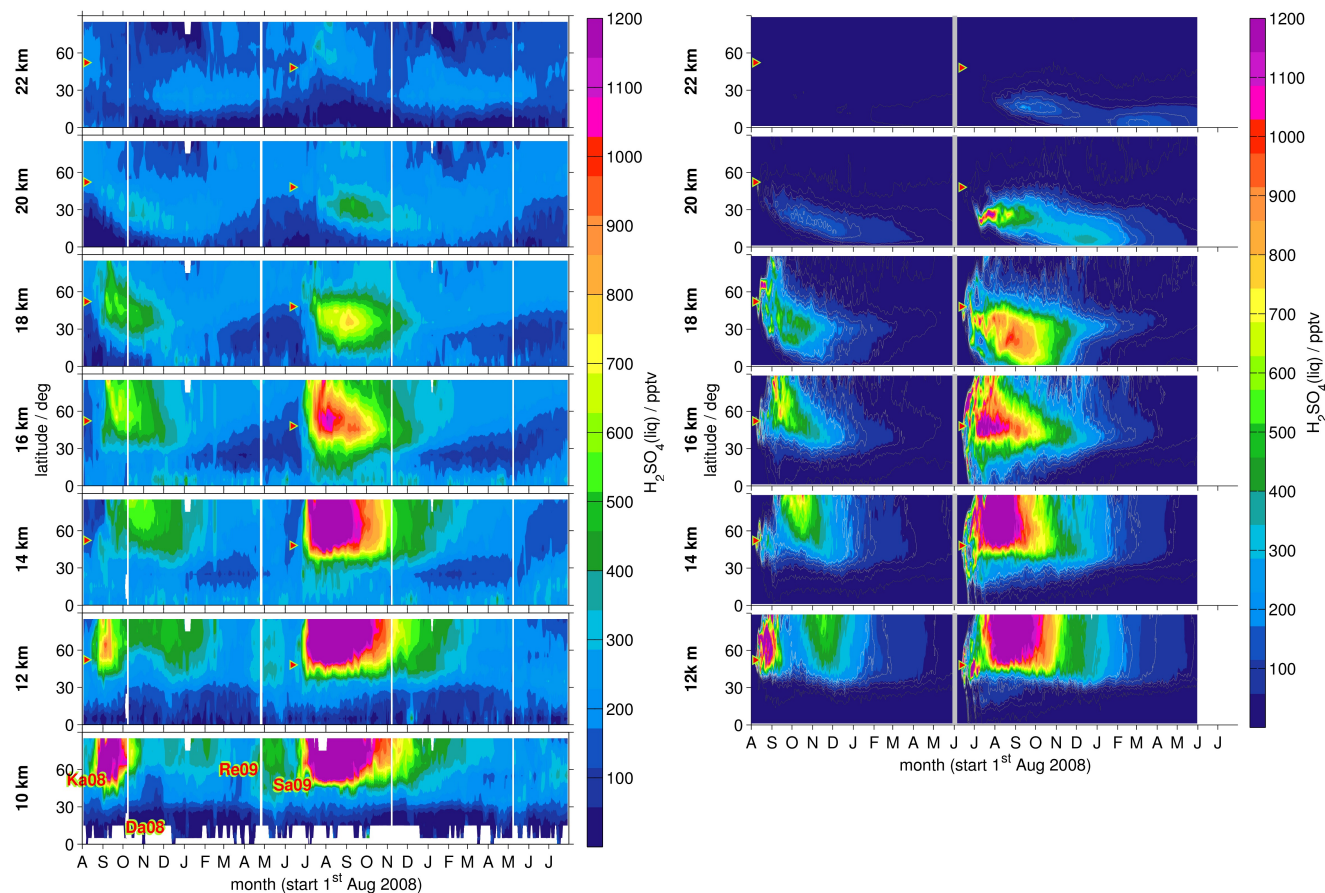


Figure 10. As Fig. 9, but for sulphate aerosol. (right) Black to white contour-lines denote 5, 25, 50, 100, 150, and 200 pptv.

Both, in the measurements and simulations, most of the sulphur contained in SO_2 and sulphate aerosol stays north of 20 30° N at lower altitudes up to around 16 km (Fig. 9 and 10). Especially at low altitudes we find a mixing barrier at ~30° N, with a strong gradient between low values in the tropics and high values in the extra-tropics, which weakens towards higher altitudes. This gradient is due to the subtropical jet stream, and is most easily detectable in the contour lines shown for modelled liquid-phase H_2SO_4 (Fig. 10, right panel), but similar patterns are observed by MIPAS. At around 16–18 km, especially in the longer lived sulphate aerosol, this forms a “tongue” of relatively high mole-fractions, which persists over a 25 longer period than in the surrounding latitudes. An additional transport process starts at an altitude of about 18 km in the case of Kasatochi and ~16 km in the case of Sarychev (Fig. 9 and 10). At these altitudes, both the MIPAS measurements and CTM simulations show that sulphur from mid-latitude volcanic eruptions can reach the tropics, predominantly in the form of sulphate aerosol. In the tropics sulphur is then lifted in the ‘tropical pipe’ (denotation following e.g. Plumb 1996), and can reach the stratospheric ‘overworld’ (denotation following e.g. Hoskins, 1991), also seen in Fig. 8. Measurements of the 30 stratospheric optical depth by OSIRIS/Odin also show that in the months after the eruptions of Kasatochi in 2008 and



Sarychev in 2009 their impact extended to lower latitudes (Bourassa et al., 2012). Wu et al. (2017) studied the equatorward dispersion of the Sarychev volcanic plume together with the influence of the Asian summer monsoon on the transport pattern. They find, that at 360–400 K potential temperature, the southward transport was primarily caused by anticyclonic Rossby wave breaking, intensified by the Asian summer monsoon during Northern Hemisphere summer. Above 400 K less aerosol is transported into the tropics. They further show an 'aerosol hole' in the anticyclone, surrounded by aerosol-rich air. Following Wu et al. (2017), a strong subtropical jet in combination with weak Rossby wave breaking events would hinder the southward transport of the volcanic plume during winter conditions. The weaker southward transport in the case of the Kasatochi eruption that starts at higher altitudes, compared to the Sarychev eruption, could be due to the eruption having been later during the monsoon season, leading to enhanced southward transport by the Asian summer monsoon for a shorter period of time. In model studies of the Sarychev eruption, Haywood et al. (2010) find that sulphate aerosol is transported around the entire globe in around 14 days. We see that the bulk of the aerosol that moves southwards reaches the equator about 2 to 3 months after the volcanic eruption. To some small extent this sulphur crosses the equator and can hereby influence the sulphur content of the Southern Hemisphere (see also Wu et al., 2017). Generally, similarities in the geographic pattern between Kasatochi in 2008 and Sarychev in 2009 have also been noticed by Haywood et al. (2010), caused by the agreement between the season of the eruptions, the injection altitude and latitude of the eruptions. The model results and MIPAS measurements of SO₂ and sulphate aerosol presented in our study confirm similarities in the transport patterns of sulphur after the volcanic eruptions of Kasatochi in August 2008 and Sarychev in June 2009, and a southward transport of the volcanic plumes, towards the equator, where sulphur can then ascent in the Brewer-Dobson circulation.

5 Discussion and conclusions

In this study a new dataset of MIPAS/Envisat global aerosol volume densities and liquid-phase H₂SO₄ VMR distributions is presented for 2005 to 2012, covering the altitude range of 10 to 30 km, with up to 1,300 profiles per day, derived from single limb-scans. The MIPAS aerosol volume densities have been corrected for possible instrumental radiance baseline offsets by comparison to coincident balloon-borne in situ observations from Laramie, Wyoming. With absolute differences below $\pm 0.003 \mu\text{m}^3 \text{cm}^{-3}$ at 20–25 km the bias corrected MIPAS profiles compare well with the in situ data. The strongest variability in the MIPAS sulphate aerosol is caused by various volcanic eruptions. Liquid-phase H₂SO₄ patterns from MIPAS are in general agreement with MIPAS SO₂ profiles from single limb-scans during volcanically perturbed and quiescent periods.

In a case study, we investigate the fate of volcanic sulphur after two major mid-latitude volcanic eruptions of the last decade (Kasatochi in 2008 at 51.2° N, and Sarychev in 2009 at 48.1° N) by combining this new dataset with simultaneously observed profiles of SO₂ from the same instrument with the help of CTM simulations. Liquid-phase H₂SO₄ from the MIPAS aerosol retrieval is not only qualitatively, but also quantitatively consistent with the MIPAS SO₂ observed after the two volcanic eruptions. One of the advantages of deriving aqueous H₂SO₄ and SO₂ from one instrument is that no sampling



biases occur due to different geolocations and measurement times. Remaining sampling biases are caused by different filter methods, which depend partly on the retrieved species. The datasets provide a valuable tool for further analyses of the stratospheric sulphur content. The new H₂SO₄ aerosol observations enable us to further constrain the total sulphur emitted into the stratosphere by the Kasatochi and Sarychev eruptions and to revise our previous estimates that were based on SO₂ observations only. The decay of SO₂ after the volcanic eruptions and the formation of sulphate aerosol are consistent with known SO₂ chemical lifetimes due to reaction with OH, under OH background conditions. While sedimentation of sulphate aerosol does play a role, the residence time of sulphur in the mid-latitude lower stratosphere following the volcanic eruptions of Kasatochi and Sarychev is dominated by advective transport and transport by the Brewer-Dobson circulation. Sensitivity simulations with the CTM with different effective aerosol radii indicate that the observed decay of sulphate aerosol is best described by sedimentation of aerosol particles with an effective radius of about 0.5 μm. Most of the sulphur emitted by the two volcanic eruptions resides in the extra tropical lowermost stratosphere where it is transported downward across the tropopause. However, at higher altitudes (at about 16 to 22 km) part of the volcanically emitted sulphur from these Northern Hemisphere mid-latitude volcanoes is transported equatorwards where it is lifted in the 'tropical pipe' into the stratospheric 'overworld' and even enters the Southern Hemisphere.

Our findings of the residence time and transport pathways of enhanced sulphate aerosol in the mid-latitude lower stratosphere have implications for the forcing of surface climate by moderate sized mid-latitude volcanoes and proposed geoengineering schemes. Sulphur injections into the lowermost stratosphere at mid-latitudes can affect not only the mid- to high-latitudes of the respective hemisphere, but are potentially transported towards the tropics, where they can undergo uplift and further transport by the Brewer-Dobson circulation, and can hereby reach the other hemisphere.

6 Data availability

The MIPAS datasets for aerosol volume densities and liquid-phase H₂SO₄ mole-fractions are available upon request from the authors or at <http://www.imk-asf.kit.edu/english/308.php>. Model results are available upon request from the authors.

7 Appendix

Method of increasing areas for zonal averages:

To reduce biasing of zonal averages due to non-uniformly distributed data, we use a method of increasing areas. It is based on the horizontal grid of our Chemical Transport Model, which has a resolution of ~2.5° latitude x 3.75° longitude. For MIPAS daily arithmetic means are calculated for these ~2.5° x 3.75° boxes. These are then averaged to 32 boxes of ~10° x 11.25°. The area is further increased longitudinally by a factor of 2 in each step, while the number of boxes decreases by the same factor (32 x 11.25° → 16 x 22.5° → 8 x 45° → 4 x 90° → 2 x 180° → 1 x 360°). The result is not changed, compared to normal averaging (sum of values divided by number of values), when an equal number of values is available per ~



2.5°x3.75° grid-cell, which is the case for the CTM, in altitudes above 13 km. By interpolating the model results to a vertical grid with 1 km resolution, starting at an altitude of 10 km, we find missing values up to 13 km. Therefore the method of increasing areas for zonal averages is applied to the simulated data in these altitudes as well. As the surface decreases with increasing latitude, the latitude bin is chosen to be 11.25° and not broader, as increasing latitude bins can give a too high weight to values corresponding to relatively smaller areas.

8 Author contributions

A. Günther developed and performed the model simulations, wrote most of the paper and conducted most of the analyses. M. Höpfner developed the MIPAS aerosol retrieval and provided the retrieval sensitivity studies and error estimations, and their description (Sect. 3.1). B.-M. Sinnhuber provided advice with the chemical transport modelling. T. von Clarmann, G. Stiller and M. Höpfner provided advice for the analyses of MIPAS data. S. Griessbach provided the flags for the MIPAS ice and ash filter. T. Deshler provided the balloon-borne in situ data and their description (Sect. 2.2). All authors contributed to the discussion of the results.

9 Competing interests

The authors declare that they have no conflict of interest.

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