1 A growing threat to the ozone layer from short-lived anthropogenic 2 chlorocarbons

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19 Abstract

20 Large and effective reductions in emissions of long-lived ozone-depleting substance (ODS) are being achieved through the Montreal Protocol, the effectiveness of which can be seen in 21 22 the declining atmospheric abundances of many ODS. An important remaining uncertainty 23 concerns the role of very short lived substances (VSLS) which, owing to their relatively short 24 atmospheric lifetimes (less than 6 months), are not regulated under the Montreal Protocol. 25 Recent studies have found an unexplained increase in the global tropospheric abundance of one VSLS, dichloromethane (CH_2CI_2), which has increased by around 60% over the past 26 decade. Here we report dramatic enhancements of several chlorine-containing VSLS (Cl-27 28 VSLS), including CH₂Cl₂ and CH₂ClCH₂Cl (1,2-dichloroethane), observed in surface and upper tropospheric air in East and South East Asia. Surface observations were, on occasion, 29 30 an order of magnitude higher than previously reported in the marine boundary layer, whilst upper tropospheric data were up to 3 times higher than expected. In addition we provide 31 32 further evidence of an atmospheric transport mechanism whereby substantial amounts of industrial pollution from East Asia, including these chlorinated VSLS, can rapidly, and 33 regularly, be transported to tropical regions of the western Pacific and subsequently uplifted 34 to the tropical upper troposphere. This latter region is a major provider for air entering the 35 stratosphere and so this mechanism, in conjunction with increasing emissions of CI-VSLS 36 from East Asia, could potentially slow the expected recovery of stratospheric ozone. 37

38 1. Introduction

Large-scale ozone depletion in the stratosphere is a persisting global environmental problem. It is predominantly caused by the release of reactive chlorine and bromine species from halogenated organic compounds. Although the basic science is well established, there remains significant uncertainty surrounding the long-term recovery of the ozone layer (Hegglin et al., 2015). One important issue is the recent, unexplained increase in the global tropospheric abundance of dichloromethane (CH₂Cl₂), which has increased by ~60% over the past decade (Leedham-Elvidge et al., 2015; Hossaini et al., 2015a; Carpenter and Reimann et al., 2015). 46 CH₂Cl₂ is one of a large group of halogenated compounds known as VSLS (very short-lived 47 substances). Owing to their relatively short atmospheric lifetimes (typically less than 6 months) and their correspondingly low Ozone Depletion Potentials (ODPs), VSLS are not currently 48 regulated by the Montreal Protocol. It is however estimated that a significant fraction of VSLS 49 and their atmospheric degradation products reach the stratosphere (>80% in the case of 50 51 chlorinated VSLS; Carpenter and Reimann et al., 2015) and, furthermore, halogenated VSLS have been shown to have a disproportionately large impact on radiative forcing and climate 52 53 due to their atmospheric breakdown, and the subsequent depletion of ozone, occurring at 54 lower, climate sensitive altitudes (Hossaini et al., 2015b). According to the most recent 55 Scientific Assessment of Stratospheric Ozone Depletion (Carpenter and Reimann et al., 2015) over the period 2008-2012 the total chlorine from VSLS increased at a rate of approximately 56 57 1.3 ± 0.2 ppt Cl yr⁻¹, the majority of this increase being due to CH₂Cl₂, and this has already 58 begun to offset the decline in total tropospheric chlorine loading over the same period (13.4 \pm 59 0.9 ppt Cl yr⁻¹) caused by the reduced emissions of substances controlled by the Montreal Protocol. 60

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62 In recent years much attention has been focussed on the potential of bromine-containing VSLS to contribute to stratospheric ozone depletion (Law and Sturges, 2007; Montzka and 63 Reimann, 2011). This is primarily due to the large observed discrepancy between the 64 65 measured inorganic bromine in the stratosphere and the amount of bromine available from known, longer lived source gases, namely the halons and methyl bromide (Dorf et al., 2006). 66 In contrast, the role of very short-lived chlorine compounds (CI-VSLS) in ozone depletion has 67 68 been considered relatively minor because they are believed to contribute only a few percent 69 to the total chlorine input to the stratosphere, the majority of which is supplied by long-lived 70 compounds such as the chlorofluorocarbons (CFCs), methyl chloroform (CH₃CCl₃) and carbon 71 tetrachloride (CCl₄). Since 1987 the consumption of these long-lived anthropogenic 72 compounds has been controlled by the Montreal Protocol and the sum of total organic chlorine 73 in the troposphere has been falling since its peak of around 3660 parts per trillion (ppt) in 74 1993/94 to ~3300 ppt in 2012 (Carpenter and Reimann et al., 2015). Because of its relatively 75 short atmospheric lifetime (~5 years) and its high chlorine content (3 chlorine atoms per 76 molecule), the main contributor to this decline has been CH₃CCl₃. However, most CH₃CCl₃ 77 has now been removed from the atmosphere with a present day abundance of less than 5 ppt. 78 Consequently the rate of decline in total organic chlorine has fallen to 13.4 ppt/year (2008-79 2012), which is around 50% smaller than the maximum seen in the late 1990s (Carpenter and 80 Reimann et al., 2015).

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82 Owing to their short atmospheric lifetimes and their hitherto low background concentrations, 83 chlorinated VSLS have not been considered of major importance for ozone depletion. Indeed the contribution of VSLS to the total chlorine entering the stratosphere is estimated to be only 84 85 55 (38–95) ppt (Carpenter and Reimann et al., 2015), which is between 1% and 3% of the present day (2012) total (3300 ppt). However, because of their short lifetimes, the potential 86 87 impact of VSLS on stratospheric ozone is highly dependent on the location of their sources, with emissions close to the major stratospheric input regions being of far greater significance 88 89 for ozone depletion.

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91 The transport of trace gases and aerosols from the troposphere into the stratosphere occurs
92 primarily in the tropics, where convective activity and vertical uplift are most intense. In order

to get to the stratosphere an air parcel has to pass through the tropical tropopause layer (TTL),

the region of the atmosphere between the level of maximum convective outflow (~12 km
altitude, 345K potential temperature) and the cold-point tropopause (~17 km, 380K) (see Box **1-3**, Figure 1 in Carpenter and Reimann et al., 2015). The vertical flux into the TTL is thought
to be dominated by two main regional pathways, (1) ascent above the western Pacific during
Northern Hemispheric (NH) winter and (2) the circulation of the Asian (Indian) Monsoon during
NH summer (Fueglistaler et al., 2009). The latter has been suggested as the most important
region for transport of anthropogenic pollution (Randel et al., 2010).

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102 Because of their short lifetimes, to be able to accurately determine the VSLS contribution to 103 total organic halogen loading in the stratosphere it is highly desirable to collect data in the TTL. Surface measurements alone, particularly in regions outside the tropics where most long-104 term surface stations are sited, are not sufficient. Furthermore, because of the distribution and 105 106 seasonality of stratospheric entry points it is also essential to measure in specific locations 107 and at specific times of year, i.e. in the Indian summer monsoon and over the winter western Pacific. Unfortunately there are very few available measurements of VSLS in the TTL generally 108 109 as it is above the maximum altitude of most research aircraft, and, furthermore, there is a 110 paucity of both ground and aircraft data available in these two key regions of interest. Where recent TTL data is available it is primarily from different regions and focussed on brominated 111 VSLS (e.g. Sala et al., 2014; Navarro et al., 2015). 112

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The focus of the present study is the western Pacific and, in particular, the region of the South 114 China Sea. During NH winter the region is heavily influenced by the large anticyclone that 115 116 forms over Siberia each year which gives rise to strong north-easterly winds that impact deep into the tropics as far south as Malaysia, Singapore and Indonesia. These north-easterly winds 117 typically prevail for 4-5 months (November-March) and form part of the East Asian winter 118 119 monsoon circulation. Superimposed on this seasonal synoptic flow are transient disturbances known as cold surges, which are triggered by a southward shift of the anticyclone and lead to 120 sudden drops in surface air temperatures and increased wind speeds (Zhang et al., 1997; 121 Garreaud, 2001). It has been proposed that during these events significant amounts of 122 pollution from continental East Asia (>35°N) can be transported rapidly to the tropics (Ashfold 123 et al., 2015). Furthermore, these events, which can last for many days, occur regularly each 124 125 winter and are associated with some of the strongest convective activity in the western Pacific region. Indeed, trajectory calculations show that it can take less than 10 days for air masses 126 127 to travel from the East Asian boundary layer (>35°N) to the upper tropical troposphere 128 (altitudes > 200 hPa), thereby providing a fast route by which VSLS (and many other 129 pollutants) may enter the lower stratosphere, despite their relatively short atmospheric 130 lifetimes (Ashfold et al., 2015).

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Here we provide strong evidence to support this proposed transport mechanism based on new 132 atmospheric observations in the East and SE Asia region. We will present new CI-VSLS 133 measurements from recent ground-based and aircraft campaigns in the region during which 134 135 we have observed dramatic enhancements in a number of CI-VSLS, including CH₂Cl₂, 1,2dichloroethane (CH₂ClCH₂Cl), trichloromethane (CHCl₃) and tetrachloroethene (C₂Cl₄). 136 Furthermore we will demonstrate how pollution from China and the surrounding region can 137 rapidly, and regularly, be transported across the South China Sea and subsequently uplifted 138 to altitudes of 11-12 km, the region close to the lower TTL. Using the NAME particle dispersion 139 140 model we will also investigate the origin of the observed CI-VSLS and examine the frequency and duration of cold surge events. Finally we present some new estimates of CH₂Cl₂ 141

emissions from East Asia and use these to estimate the likely emissions of CH_2CICH_2CI , for which there is little information in the recent literature.

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145 2. <u>Methods</u>

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Between 2012 and 2014, air samples were collected at various times at (1) two coastal sites
in Taiwan, Hengchun (22.0547°N, 120.6995°E) and Fuguei Cape (25.297°N, 121.538°E); (2)
the Bachok Marine Research Station on the Northeast coast of Peninsular Malaysia (6.009°N,
102.425°E); and (3) during several flights of the IAGOS-CARIBIC aircraft between Germany
and Thailand/Malaysia. IAGOS-CARIBIC is a European project making regular measurements
from an in-service passenger aircraft operated by Lufthansa (Airbus A340-600;
Brenninkmeijer et al., 2007; http://www.caribic-atmospheric.com/).

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155 A total of 21 samples were collected at Hengchun between 7 March and 5 April 2013 with a further 22 samples taken at Cape Fuguei between 11 March and 4 April 2014. 28 samples 156 were collected at Bachok between 20 January and 5 February 2014, during the period of the 157 158 NE winter monsoon. The approximate location of each surface site is shown in Figure 1. The CARIBIC aircraft samples were collected during seven return flights between (i) Frankfurt 159 (Germany) and Bangkok (Thailand), and (ii) Bangkok and Kuala Lumpur (Malaysia) during the 160 periods December 2012 - March 2013 (4 flights) and November 2013 - January 2014 (3 161 flights). All CARIBIC flights in this region between December 2012 and January 2014 have 162 been included in this analysis. With the exception of 3 samples that were taken at altitudes 163 164 between 8.5 and 9.8 km, the CARIBIC samples were all (n=179) collected at altitudes between 10 and 12.3 km. 165

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167 2.1 Sample collection

Air samples from Taiwan and Malaysia were collected in 3.2 litre silco-treated stainless steel 168 canisters (Restek) at a pressure of approximately 2 bar using a battery-powered diaphragm 169 170 pump (Air Dimensions, B series). In Taiwan the samples were collected from the surface via a 1m x ¹/₄" OD Dekabon sampling line, whilst in Bachok the samples were collected from the 171 top of an 18 m tower via a 5 m x 1/4" OD Dekabon sampling line. In both cases the tubing was 172 173 flushed for at least 5 minutes prior to sampling. The sampling integrity was confirmed by 174 sampling high purity air (BTCA-178, BOC) through the inlet tubing and pump. Samples were collected within 50 m of the sea and only when the prevailing winds were from the sea, 175 176 minimising the impact of any local emissions. The CARIBIC aircraft samples were collected in 177 2.7 litre glass flasks at a pressure of 4.5 bar using a two-stage metal bellows pumping system (Brenninkmeijer et al., 2007; Baker et al., 2010). 178

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180 2.2 Sample analysis

The collected air samples were shipped to UEA and analysed for their halocarbon content by 181 gas chromatography - mass spectrometry (GC-MS) following trace gas enrichment using 182 183 previously published methods. All samples (i.e. Taiwan, Bachok and CARIBIC) were analysed 184 for CH₂Cl₂, CHCl₃ and C₂Cl₄ using an Entech-Agilent GC-MS system operating in electron 185 ionisation (EI) mode, as described in Leedham-Elvidge et al., (2015). 1 litre samples were dried and pre-concentrated before injection onto a 30 m x 0.32 mm GS Gas Pro capillary 186 column (Agilent), temperature ramped from -10°C to 200°C. Samples were interspersed with 187 188 repeated analyses of a working standard (SX-706070), a high pressure air sample contained in a 34 litre electropolished stainless steel cylinder (Essex Industries) provided by the Earth 189

190 System Research Laboratory of the National Oceanic and Atmospheric Administration 191 (NOAA-ESRL, Boulder, CO, USA). CH_2Cl_2 , $CHCl_3$ and C_2Cl_4 were quantified on ions with a 192 mass-to-charge ratio of 84 ($CH_2^{35}Cl_2^+$), 83 ($CH^{35}Cl_2^+$ and 166 ($C_2^{35}Cl_3^{37}Cl^+$) respectively. Mean 193 analytical precisions were ± 2% for CH_2Cl_2 and C_2Cl_4 , and ± 3% for CHCl_3. Instrument blanks, 194 determined by analysing 1 litre aliquots of high purity nitrogen (BOC, Research grade), were 195 always below the detection limit of the instrument.

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197 Some of the ground-based samples and a subset of the CARIBIC samples were also analysed 198 for a range of halocarbons, including the newly-identified CH₂CICH₂CI, using a preconcentration/GC system coupled to a Waters AutoSpec magnetic sector MS instrument, also 199 operating in EI mode, but run at a mass resolution of 1000 at 5 % peak height. Samples (using 200 between 200 and 250 ml of air) were analysed on an identical GS GasPro column following a 201 previously described method (Laube et al., 2010; Laube et al., 2012; Leedham-Elvidge et al., 202 203 2015). CH₂CICH₂CI was monitored on the ions with mass-to-charge ratios of 61.99 (C₂H₃³⁵CI⁺, qualifier) and 63.99 (C₂H₃³⁷Cl⁺, quantifier). Mean analytical precision was 1.4 % for 204 CH₂CICH₂CI and the average blank signal was 0.07 ppt (as quantified using regular 205 206 measurements of research-grade helium) and was corrected for on a daily basis.

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208 2.3 Calibration and quality assurance

CH₂Cl₂, CHCl₃ and C₂Cl₄ data are reported on the latest (2003) calibration scales provided by 209 NOAA-ESRL. As was shown in Leedham-Elvidge et al., (2015) our CH₂Cl₂ measurements 210 compare very well with those of NOAA-ESRL at our mutual long-term sampling site at Cape 211 212 Grim, Tasmania over more than 6 years. As a recognised international calibration scale for CH₂CICH₂CI is not yet available this compound was calibrated at UEA using the established 213 214 static dilution technique recently described (Laube et al., 2012). CH₂CICH₂CI was obtained 215 from Sigma Aldrich with a stated purity of 99.8 %. Three dilutions were prepared at 7.1, 11.9 and 15.8 ppt. The mixing ratio assigned to our working standard from these dilutions was 5.67 216 ppt with a 1 σ standard deviation of 1.8 %. CFC-11 was added to the dilutions as an internal 217 218 reference compound and the CFC-11 mixing ratios assigned to the working standard through these dilutions agreed with the value assigned by NOAA-ESRL within 4.3 %. This is well within 219 220 the estimated uncertainty of the calibration system of 7 % (Laube et al., 2012). In addition the 221 mixing ratios of CH₂CICH₂CI in the working standard were compared with those in three other 222 high-pressure canisters (internal surface was either electropolished stainless steel or 223 passivated aluminium) over the whole measurement period. The ratios between standards did 224 not change within the 2 σ standard deviation of the measurements for any of the canisters analysed indicating very good long-term stability for CH₂CICH₂CI. This was also the case for 225 CHCl₃ and C₂Cl₄. As noted in Leedham-Elvidge et al., (2015) mixing ratios of CH₂Cl₂ were 226 227 found to change over longer timescales in some of our standard canisters, but this drift has been successfully quantified and corrected for as indicated by the very good comparability 228 229 with NOAA-ESRL measurements at the Cape Grim site noted above.

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231 **<u>3. Results</u>**

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Figure 1 shows the location of the three surface observation stations as well as the location of the CARIBIC samples. The aircraft sampling points have been coloured by their CH_2Cl_2 concentration (see later discussion). Data from the surface stations and from the CARIBIC aircraft flights are summarised in Table 1, together with a summary of published observations as reported in the most recent Scientific Assessment of Stratospheric Ozone Depletion (Carpenter and Reimann et al., 2015). It should be noted that CH₂CICH₂CI was only analysed
 for in selected samples and no data is available from Hengchun 2013 or from CARIBIC flights
 between Bangkok and Kuala Lumpur. In addition, only 16 Bachok samples were analysed for
 CH₂CICH₂CI.

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243 The highest concentrations of chlorinated VSLS were measured in samples collected in Taiwan, suggesting that Taiwan is located relatively close to major emission regions. Figure 2 244 245 shows the March/April 2014 data from Cape Fuguei. The Numerical Atmospheric-dispersion Modelling Environment model (NAME, see supplementary material) can be used to infer the 246 recent transport history of this pollution. Our NAME analysis (Fig. 2 b-d) indicates that most of 247 the samples that contained high concentrations of CI-VSLS had originated from regions to the 248 north of Taiwan, primarily the East Asian mainland. The median sum of chlorine from the 4 249 250 VSLS listed above (SCI_{VSLS}) in 22 samples collected at Cape Fuguei in March/April 2014 was 251 756 ppt (range 232-2178 ppt). Similarly high concentrations and variation were seen in the 21 252 samples collected at Hengchun in March/April 2013 (Fig. S1 in the supplementary material). 253 To put these concentrations in a global context, the total organic chlorine derived from all 254 known source gases in the background troposphere (including CFCs, HCFCs, and longerlived chlorocarbons) is currently around 3300 ppt, with a typical CI-VSLS contribution in the 255 remote marine boundary layer of approximately 3 % (Carpenter and Reimann et al., 2015). 256 257 Of the four VSLS measured, the two largest contributors to ΣCI_{VSLS} in Taiwan were CH₂Cl₂ 258 (55-76 %) and CH₂CICH₂CI (14-30 %).

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260 Figure 3 shows the CI-VSLS data from 28 samples collected at Bachok, Malaysia during the winter monsoon season in late January/ early February 2014. During this phase of the East 261 Asian monsoon the prevailing winds are from the northeast and, as described earlier, are often 262 263 impacted by emissions further to the north, including from mainland China. As can be seen in Figure 3, there was a 7 day period between 19 and 26 January when significantly enhanced 264 concentrations of CI-VSLS were observed. During this period NAME back trajectories show 265 air travelling from continental East Asia and across the South China Sea before arriving at 266 Bachok. Three examples during this cold surge event are shown in Fig. 3 (b-d). These 267 trajectories often pass over Taiwan and, in some instances, also over parts of Indochina where 268 269 additional emissions could have been picked up. As in the Taiwan samples, CH₂Cl₂ is the 270 largest contributor to ΣCI_{VSLS} (59-66 %), having a mean concentration of 179.9 ± 71.9 ppt 271 (range 94.0 – 354.9 ppt, 9 samples) during the 7-day period of the pollution event. The mean 272 concentration of CH₂ClCH₂Cl was 64.4 ± 23.9 ppt (range 30.2 - 119.5 ppt), accounting for 19-273 23 % of ΣCI_{VSLS} . These abundances are substantially higher than those typically found in the marine boundary layer. For example, the range of ΣCI_{VSLS} from the 4 compounds listed above 274 275 in the tropical marine boundary layer reported in WMO (2014) is 70-134 ppt. The range observed at Bachok over the entire sampling period was 207-1078 ppt, with medians of 546 276 277 ppt and 243 ppt during the polluted (20-26 Jan) and less-polluted (27 Jan - 5 Feb) periods respectively (see Table 1). It is interesting to note that even in the period after the cold surge 278 279 event (Fig.3 e,f), the levels of CI-VSLS are still significantly higher than would be expected, suggesting that this region of the South China Sea is widely impacted by emissions from E 280 Asia. 281

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The pollution or "cold surge" event observed at Bachok lasted for 6-7 days and the back trajectories shown in Figure 3 are typical of those arriving at Bachok during the winter monsoon period (see NAME animations in supplement). To further investigate the frequency 286 and typical duration of these events a NAME trajectory analysis using carbon monoxide (CO) as a tracer of industrial emissions from regions north of 20°N was conducted for the entire 287 winter season (see supplementary information for details). Figure 4(a) shows a time series of 288 this industrial CO tracer for winter 2013/2014 and suggests that the observed event in January, 289 during which there was a strong correlation between the industrial CO tracer and CH₂Cl₂ (Fig. 290 4b), is likely to be repeated regularly throughout the winter. An analysis of a further 5 winters 291 (Fig. 4c) demonstrates that 2013/14 was not unusual and that the events depicted in Figure 292 293 3a occur repeatedly every year (Fig. S2 in the supplementary material).

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The Bachok measurements clearly demonstrate the rapid long-range transport of highly 295 elevated concentrations of CI-VSLS for several thousand kilometres across the South China 296 Sea, as predicted by Ashfold et al., (2015). However, to have an impact on stratospheric ozone 297 298 it is necessary to demonstrate that these high concentrations of CI-VSLS can be rapidly lifted 299 to the upper tropical troposphere (lower TTL) or above. Such evidence can be found in samples from several recent CARIBIC aircraft flights in the Southeast Asia region. Figure 1 300 301 shows significant enhancements of CH₂Cl₂ during flights over northern India and the Bay of 302 Bengal, and also between Bangkok and Kuala Lumpur. The same data is plotted against longitude in Figure 5(a) which shows that elevated concentrations were observed in the seven 303 CARIBIC flights in the region during the periods Dec 2012 - Mar 2013 and Nov 2013 - Jan 304 2014. The samples were collected in the altitude range 10-12.3 km, showing that recent 305 industrial emissions can regularly reach the lower boundary of the TTL. Although CH₂CICH₂CI 306 was only analysed for in a selection of samples during the flights from Germany to Bangkok, 307 308 elevated mixing ratios coinciding with the high levels of CH₂Cl₂ were clearly observed (Fig. 309 5b). CHCl₃ and C₂Cl₄ were also enhanced during these flights (Table 1), with ΣCl_{VSLS} being in 310 the range 48-330 ppt (Fig. 5c). This is up to 3.2 times higher than that previously found in the 311 lower TTL (36-103 ppt; Carpenter and Reimann et al., 2015). The highest abundances of Cl-VSLS were seen in samples collected over the Bay of Bengal, and on flights between Bangkok 312 and Kuala Lumpur (Fig. 5a). NAME back trajectories (Fig. 5d) indicate that in these cases the 313 314 sampled air had almost always been transported from the east, and had often been impacted by emissions from East Asia, with possible contributions from other countries including the 315 Philippines, Malaysia and Indochina. 316

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318 4. Discussion

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320 The high mixing ratios of CH_2CI_2 observed in the Taiwan samples are not entirely unexpected. 321 Previous studies have found very high levels (> 1 ppb) of CH₂Cl₂ in various Chinese cities (Barletta et al., 2006) and in the Pearl River Delta region (Shao et al., 2011). Elevated levels 322 323 (several hundred ppt) were also observed in aircraft measurements in polluted air emanating from China during the TRACE-P campaign in 2001 (Barletta et al., 2006). These studies took 324 325 place in the early 2000s and emissions may be expected to have grown significantly since. CH₂Cl₂ is predominantly (~90%) anthropogenic in origin, and is widely used as a chemical 326 327 solvent, a paint stripper and as a degreasing agent (McCulloch and Midgely, 1996; Montzka et al., 2011). Other uses include foam blowing and agricultural fumigation. A growing use of 328 CH₂CL₂ is in the production of HFC-32 (CH₂F₂), an ozone friendly replacement for HCFC-22 329 (CHF₂Cl) in refrigeration applications. Around 10% of global CH₂Cl₂ emissions come from 330 natural marine and biomass burning sources (Simmonds et al., 2006; Montzka and Reimann 331 332 et al., 2011).

334 Whilst the strong enhancements of CH_2CI_2 are not entirely unexpected, the presence of high concentrations of CH₂CICH₂CI most certainly are. There are very few previously reported 335 measurements of CH₂CICH₂CI, particularly in recent years. Elevated levels have been 336 observed in urban environments close to known emission sources (Singh et al., 1981) and, 337 more recently, Xue et al., (2011) reported elevated levels (91 ± 79 ppt) in air samples collected 338 in the boundary layer over north-eastern China in 2007. The few reported measurements of 339 CH₂CICH₂CI in the remote marine boundary layer are typically in the low ppt range (see Table 340 341 1) but these were mostly made well over a decade ago. No long-term atmospheric 342 measurements of CH₂CICH₂CI have been reported, and CH₂CICH₂CI is not reported by the main surface monitoring networks (AGAGE and NOAA), so current background 343 concentrations and longer term trends are unknown. CH₂CICH₂CI is predominantly 344 anthropogenic in origin, its primary use being in the manufacture of vinyl chloride, the 345 precursor to polyvinyl chloride (PVC), and a number of chlorinated solvents. CH₂CICH₂Cl also 346 347 finds use as a solvent, dispersant and has historically been added to leaded petrol as a lead scavenger (EPA, 1984). In common with CH₂Cl₂ it has also used as a cleaning/degreasing 348 349 agent and as a fumigant. China is the world's largest producer of PVC accounting for 27% of 350 global production in 2009 (DCE, 2017). Production has increased rapidly in recent years (14% per year over the period 2000-2009; DCE, 2017), which could potentially have led to increased 351 atmospheric emissions of CH₂CICH₂CI. Simpson et al. (2011) observed a small enhancement 352 in CH₂CICH₂CI in Canadian boreal forest fire plumes (background average, June-July 2008, 353 9.9 ± 0.3 ppt, plume average 10.6 ± 0.3 ppt) and estimated a global boreal fire source of 0.23 354 ± 0.19 kilotonnes (kt) yr⁻¹. 355

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357 The other CI-VSLS presented here are C₂Cl₄ and CHCl₃. In contrast to CH₂CICH₂Cl, long-term 358 atmospheric data records are available for these compounds, although there are few data 359 from the SE Asia region. Current trends show that C₂Cl₄ is declining in the background troposphere (~6 % yr⁻¹), whilst CHCl₃ is approximately constant (Carpenter and Reimann et 360 al., 2015). However, both compounds were elevated in the samples containing high 361 concentrations of CH₂Cl₂ and CH₂ClCH₂Cl, suggesting that significant, co-located sources 362 remain. Like CH₂CICH₂CI, C₂CI₄ is almost exclusively anthropogenic in origin, used primarily 363 as a solvent in the dry cleaning industry, as a metal degreasing agent and as a chemical 364 365 intermediate, for example in the manufacture of the hydrofluorocarbons HFC-134a and HFC-125. CHCl₃ is believed to be largely natural in origin (seawater, soils, macroalgae), but 366 367 potential anthropogenic sources include the pulp and paper industry, water treatment facilities 368 and HFC production (McCulloch, 2003; Worton et al., 2006; Montzka et al., 2011).

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4.1 Regional emissions of CH₂Cl₂ and CH₂ClCH₂Cl

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China does not report production or emission figures for CH₂Cl₂. However emissions of CH₂Cl₂ 372 can be estimated from known Chinese production of HCFC-22 (CHCIF₂). This is possible 373 because the production of HCFC-22 requires CHCl₃ as feedstock (1 kg HCFC-22 requires 1.5 374 375 kg CHCl₃) and because CHCl₃ is produced almost entirely (>99%) for HCFC-22 production. Production of chloromethanes by any manufacturing process leads to the inevitable co-376 production of CH₂Cl₂ and CHCl₃, with smaller (3-5%) co-production of carbon tetrachloride 377 (CCl₄). The production ratios vary by individual plant but are within the range 30:70-70:30 (% 378 CH₂Cl₂:CHCl₃). Chinese chloromethanes plants, which together represent some 60% of global 379 380 capacity and production, are generally built to a 40:60 - 60:40 flexibility ratio. With falling CFM demand due to diminished feedstock demand for HCFC-22 production, and based on regular 381

- discussions with the individual large producers, ratios in China have been switching in recent years from the traditional 40:60 towards 50:50 (CH₂Cl₂:CHCl₃; Nolan Sherry 2016).
- 384

It can be calculated that in 2015 China produced approximately 600 kt of HCFC-22 for all uses 385 (Nolan Sherry 2016), which would require 900 kt of CHCl₃ as feedstock. By subtracting 386 Chinese imports of CHCl₃ (40 kt; Comtrade 2016) and allowing for some limited emissive 387 solvent use (15 kt) suggests that China produced around 875 kt of CHCl₃ in 2015. As noted 388 389 above, in the chlorocarbon industry CH₂Cl₂ and CHCl₃ are produced in the same 390 manufacturing process and in China this is currently moving from a historic production ratio of around 40:60 towards 50:50. Using a production ratio of 45:55 it can therefore be estimated 391 that China produced around 715 kt of CH₂Cl₂ in 2015. Approximately 90 kt of this was exported 392 (Comtrade 2016) and another 170 kt was used for the production of HFC-32 (CH_2F_2), which 393 394 is a non-emissive application (Nolan Sherry 2016). This leaves an estimated 455 kt (± 10%) 395 of CH₂Cl₂ which is used almost exclusively in emissive applications such as paint stripping, foam blowing, pharmaceuticals, solvent use, etc. Although there is no specific industry-based 396 aggregation of these numbers, they have been verified in discussion with Chinese and other 397 398 industry sources. A similar method has recently been used to assess emissions of CCl₄ 399 (SPARC 2016).

400

There is a strong linear correlation between the observed CH₂Cl₂ and CH₂ClCH₂Cl data at 401 both Bachok ($R^2 = 0.9799$) and Cape Fuguei ($R^2 = 0.9189$). Combining the datasets yields a 402 slope of 0.4456 \pm 0.0194 (R² =0.9228). Using the emissions for CH₂Cl₂ derived above (455 kt) 403 404 and making the assumptions that (1) all emissions originate in China and (2) there are no 405 significant relative losses of the two compounds since emission (lifetimes are 144 days for 406 CH₂Cl₂ and 65 days for CH₂ClCH₂Cl) we can estimate Chinese emissions of CH₂ClCH₂Cl to 407 be of the order of 203 \pm 9 kt yr⁻¹. If accurate, the scale of these emissions is a major surprise as CH₂CICH₂CI is highly toxic (suggesting that local emissions would be minimised) and 408 409 believed to be used almost exclusively in non-emissive applications.

410

411 5. Concluding remarks

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413 When calculating the VSLS contribution to stratospheric chlorine, it is usual to assume an 414 average concentration in the region of the TTL known as the level of zero radiative heating 415 (LZRH). The LZRH is located at the transition between clear-sky radiative cooling and clear-416 sky radiative heating. This occurs at an approximate altitude of 15 km and it is believed that 417 air masses above this level will go on to enter the stratosphere (Carpenter and Reimann et al., 2015). As noted above there are very few measurements in this region and, furthermore, 418 419 many of the available measurements were made over a decade ago and assumptions based on surface temporal trends have to be made in order to estimate present day values 420 421 (Carpenter and Reimann et al., 2015: Hossaini et al., 2015). Another key deficiency in this 422 estimation of VSLS concentrations entering the stratosphere is that most of the reported 423 measurements have not been made in the two key regions where the strongest troposphere 424 to stratosphere transport occurs. Although we have no data from the region of the LZRH, the 425 CARIBIC data over northern India and SE Asia suggests that the contribution of VSLS to 426 stratospheric chlorine loading may be significantly higher than is currently estimated (50-95 ppt, Carpenter and Reimann et al., 2015). It is also interesting to note that the much-discussed 427 428 contribution of VSLS-Br compounds to stratospheric bromine is approximately 5 ppt, which is equivalent to 300 ppt of chlorine (1 ppt of bromine is roughly equivalent to 60 ppt chlorine, 429

430 Sinnhuber et al., 2009). The CARIBIC measurements suggest that CI-VSLS could currently,431 on occasion, contribute a similar amount.

432

These new measurements of CI-VSLS in Taiwan, Malaysia and from an aircraft flying above South-East Asia show that there are substantial regional emissions of these compounds; that these emissions can be rapidly transported long distances into the deep tropics; and that an equally rapid vertical transport to the upper tropical troposphere is a regular occurrence. Although the focus of this paper is short-lived chlorinated gases, there are many other chemical pollutants contained in these air masses which will have a large impact on such things as regional air quality.

440

450

441 Unlike the bromine-containing VSLS which are largely natural in origin, the CI-VSLS reported 442 here are mainly anthropogenic and consequently it would be possible to control their 443 production and/or release to the atmosphere. Of particular concern are the rapidly growing emissions of CH₂Cl₂, and potentially CH₂ClCH₂Cl, especially when considering the 444 445 geographical location of these emissions, close to the major uplift regions of the western 446 Pacific (winter) and the Indian sub-continent (summer). Without a change in industrial practices the contribution of CI-VSLS to stratospheric chlorine loading is likely to increase 447 substantially in the coming years, thereby endangering some of the hard-won gains achieved, 448 449 and anticipated, under the Montreal Protocol.

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- 615
- **Table 1:** Summary of CI-VSLS data from the 3 surface stations and the 7 CARIBIC flights. For
- 617 comparison, the ranges reported in the most recent WMO Ozone Assessment (Carpenter and
- Reimann et al., 2015) for the marine boundary layer (MBL) and lower Tropical Tropopause
- Layer (TTL, 12-14 km altitude) are also shown. All data are reported as mole fractions (ppt).
- 620
- **Figure 1:** Map of the region showing the location of each CARIBIC sample. The markers have been coloured according to their CH_2CI_2 concentration to highlight the regions where enhanced levels of VSLS were observed. Also shown are the approximate locations of the 3 surface stations (orange crosses).
- 625 626 **Figure 2**:

627 *Upper panel (a):* Mole fractions (ppt) of the 4 chlorinated VSLS in air samples collected at 628 Cape Fuguei, Taiwan in March/April 2014. The error bars are ± 1 standard deviation. The 629 black arrows show the dates of the footprint maps shown below.

- *Lower panel (b-d)*: NAME footprint maps indicating the likely origin of the air sampled at Cape
 Fuguei. Figures (b, 13 March) and (c, 30 March) show examples where the observed VSLS
 levels are very high and suggest a strong influence from continental East Asia. Figure (d) is
- 633 from 29 March where the influence of the mainland is much lower and the VSLS mole fractions
- are much closer to the expected background level.
- The location of Cape Fuguei is indicated with a blue circle (see also Figure 1).
- 636637 Figure 3:

638 *Upper panel (a)*: Mole fractions (ppt) of the 4 chlorinated VSLS in air samples collected at 639 Bachok in Jan/Feb 2014. Strongly enhanced levels of all 4 compounds were seen for a 7-640 day period at the beginning of the campaign (20-26 Jan). Also shown (dashed line) are the 641 approximate median background concentrations in the remote marine boundary layer in

- 642 2012 (from Carpenter and Reimann et al., 2015).
- 643 *Lower panels (b-f)*: NAME footprint maps indicating the likely origin of the air sampled at 644 Bachok. During the pollution episode (b = 21 Jan; c = 23 Jan; d = 24 Jan) the samples would 645 have been heavily impacted by emissions from the East Asian mainland, whilst this influence 646 is much reduced during the cleaner, non-polluted periods (e = 3 Feb; f = 5 Feb). Note that 647 even after the main pollution event, the abundance of the VSLS remain significantly above
- 647 even after the main pollution event, the abundance of the VSLS remain significantly abc 648 true background levels for much of the time, suggesting a widespread influence from
- 649 industrial emissions on a regional scale.
- The location of Bachok is indicated with a blue circle (see also Figure 1).
- 651 **Figure 4:**
- (a) Time-series of the modelled carbon monoxide (CO) anomaly at Bachok (i.e. that due only

to industrial emissions from north of 20°N in the previous 12 days) for winter 2013/14. The

- CH_2CI_2 data (grey squares) from the Bachok sampling period are overlaid. The dashed lines show the 25 ppb and 50 ppb thresholds referred to in 3c (see supplement for further details).
- (b) Correlation of the modelled CO anomaly with the observed CH_2CI_2 .
- (c) Average number of days each month, averaged over six consecutive winters (2009/10 –
 2014/15) where the modelled carbon monoxide anomaly at Bachok is above a particular

threshold (25 ppb and 50 ppb which, from the regression in 3b, correspond to 176 ppt and 315 ppt of CH_2Cl_2). The 2013/14 winter is shown separately for comparison with the 6-year average.

662 **Figure 5**

(a) Mole fractions (ppt) of CH₂Cl₂ in CARIBIC air samples collected at 10-12km altitude over
 Northern India, the Bay of Bengal and SE Asia. The samples are plotted against longitude
 and have been coloured by date.

- (b) Mole fraction (ppt) of CH₂CICH₂Cl in selected CARIBIC samples (note: CH₂CICH₂Cl was
 not monitored in the samples collected between Bangkok to Kuala Lumpur, and only in a
 selection of samples on the Frankfurt-Bangkok route).
- (c) Total CI-VSLS derived from the 4 compounds of interest in the CARIBIC samples (note:
 total CI-VSLS could only be calculated for the samples shown in Fig 5b above).
- (d) NAME footprint maps indicating the likely origin of the air sampled by the CARIBIC
- aircraft. NAME footprints at this altitude, and particularly in regions of strong sub-grid-scale
- 673 convection not captured fully in the gridded meteorological input data, may be less reliable
- than those at the surface sites. This makes pinpointing particular emission regions more
- difficult. The central panel therefore shows a composite footprint derived from the samples
- that contained the highest levels of CH_2Cl_2 (90th percentile, $[CH_2Cl_2] > 75.6$ ppt), with the
- composite footprint from the remaining samples ($[CH_2CI_2] < 75.6$ ppt) shown in the left hand
- 678 panel. To emphasise the likely source regions the right hand panel shows the difference
- 679 between the middle and left hand panels. The geographical location of each sample
- 680 included in the composite analysis are shown in blue circles.

681 Table 1

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	Taiwan 2013		Taiwan 2014		Bachok 2014			MBL (WMO 2014) ^(b)	
	Median	Range	Median	Range	Median (CS) ^(a)	Median (non-CS)	Range	Median	Range
CH ₂ Cl ₂	226.6	68 - 624	227.4	70 - 639	170.4	81.9	64.8 – 355	28.4	21.8 - 34.4
CH ₂ ClCH ₂ Cl	-	-	85.4	16.7 – 309	62.2	21.7	16.4 – 120 ^(c)	3.7	0.7 - 14.5 ^(d)
CHCl₃	33.0	11.6 – 232	35.1	13.8 - 103	22.8	14.7	12.8 - 30.5	7.5	7.3 – 7.8
C ₂ Cl ₄	4.4	1.7 – 16.6	5.5	1.7 – 18.6	4.5	1.9	1.5 – 9.5	1.3	0.8 – 1.7
Σ Cl _{VSLS}	-	-	755.8	232 -2178	546.0	243.1	207 – 1078 ^(c)	93.4	70 - 134

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	CARI	BIC (FRA-BKK	, 65-97°E)	CARIBIC (BKK-KUL, 100-105°E)			Lower TTL (WMO 2014) ^(b)	
		10-12 km	ı		10-12 km	12-14 km		
	Mean	Median	Range	Mean	Median	Range	Mean	Range
CH ₂ Cl ₂	43.2	31.6	14.6 - 121	50.4	46.5	22.5 - 100	17.1	7.8 – 38.1
CH ₂ CICH ₂ CI ^(e)	9.9	6.1	0.4 – 29.1	-	-	-	3.6	0.8 - 7.0
CHCl₃	7.0	6.0	2.0 - 15.6	9.3	8.7	3.7 - 46.6	6.8	5.3 - 8.2
C ₂ Cl ₄	0.87	0.65	0.1 - 4.4	1.6	1.5	0.2 - 5.9	1.1	0.7 – 1.3
Σ Cl _{VSLS} ^(e)	153.7	119.3	48.4 - 330				67	36 - 103
Σ Cl _{VSLS} * (f)	<mark>110.9</mark>	<mark>81.4</mark>	<mark>35.2 - 301</mark>	134.8	127.8	56.6 – 251	-	-

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^(a) CS and non-CS refer to the cold surge (polluted) and non-cold surge periods at Bachok

(b) The WMO data is a compilation of all reported global measurements up to, and including, the year 2012. The
 range represents the smallest mean minus one standard deviation and the largest mean plus one standard
 deviation of all considered datasets. Data from the TTL was derived from various aircraft and balloon
 campaigns.

692 ^(c) CH₂CICH₂CI was only analysed for in 16 of the 28 samples collected at Bachok

^(d) Note that the CH₂CICH₂CI MBL data actually date back to the early 2000s. No recent data was reported.

- ^(e) CH₂CICH₂CI was only analysed for in selected samples from the Frankfurt-Bangkok flights and in no samples
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 ^(e) CH₂CICH₂CI was only analysed for in selected samples from the Frankfurt-Bangkok flights and in no samples
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 ^(e) CH₂CICH₂CI was only analysed for in selected samples
 ^(e) CH₂CICH₂
- 697 ^(f) Σ Cl_{VSLS}* is defined as the sum of CI-VSLS excluding the contribution from CH₂CICH₂CI. Statistics derived from
 698 all samples (98 FRA-BKK; 81 BKK-KUL).

699

700









 (a)















- 733 Figure 5





A growing threat to the ozone layer from short-lived anthropogenic chlorocarbons (Oram et al.)

- 742 Supplementary material
- 743

744 **1. Additional measurements**

As mentioned in the main document, 21 air samples were also collected at the Hengchun station in southern Taiwan (22.0547N, 120.6995E) during March and early April 2013 (Figure S1). Unfortunately CH_2CICH_2CI was not analysed for in the 2013 samples, but the absolute concentrations and variability of CH_2CI_2 , $CHCI_3$ and C_2CI_4 were very similar to those observed at Cape Fuguei in 2014 (see also Table 1).

750

751 2. Modelling

752 The Numerical Atmospheric-dispersion Modelling Environment (NAME, Jones et al. 2007) is a Lagrangian particle dispersion model, used here to understand the origin of the sampled air 753 masses. For each air sample, NAME was used to calculate batches of 60000 inert backward 754 trajectories. For the ground samples the trajectories started at the measurement site within an 755 756 altitude range of 0-100 m and were started throughout a 3 hour period encompassing the 757 sample time. For the aircraft samples the trajectories were started at the exact time, horizontal coordinates and altitude at which the sample was collected. The surface sample trajectories 758 ran for 12 days and the aircraft sample trajectories for 20 days. Every 15 minutes the location 759 of all trajectories within the lowest 100 m of the model atmosphere was recorded on a grid 760 with a horizontal resolution of 0.5625° longitude by 0.375° latitude. From this information, and 761 assuming a uniform surface air density consistent with a pressure of 1000 hPa and a 762 763 temperature of 25°C, the sensitivity of the sampled air mass to surface emissions occurring in 764 the previous 12 or 20 days within a particular grid cell can be derived (units sm²g⁻¹).

765

The trajectories were calculated using three-dimensional meteorological fields produced by 766 the UK Meteorological Office's Numerical Weather Prediction tool, the Unified Model (UM). 767 768 These fields have a horizontal grid resolution of 0.35° longitude by 0.23° latitude and 59 769 vertical levels below ~30 km, and are available at 3 h intervals. Vertical velocities are obtained from the UM and available at grid nodes. The sub-grid-scale process of turbulence is 770 parameterised in NAME (Morrison and Webster 2005). Another sub-grid scale process, 771 772 convection, is not parameterised in our NAME calculations. However, past work (Heyes et al. 2009, Ashfold et al. 2012, Navarro et al. 2015) has shown atmospheric composition in the 773 tropics can be interpreted using trajectories calculated with wind fields that, while not resolving 774 775 individual up- and down-draughts, are consistent with large-scale convective activity.

776

777 2.1 Multi-year NAME calculations

By combining the emission sensitivities derived from NAME with a distribution of emissions it 778 779 is possible to calculate a modelled mixing ratio of the emitted species, due only to emissions 780 occurring within the timescale of the backward trajectories, at the measurement site (dimensionally, $sm^2g^{-1} x gm^{-2}s^{-1}$ = dimensionless mixing ratio). We have used an inventory of 781 industrial and combustion carbon monoxide (CO) emissions (RCP8.5 for 2005; Granier et al. 782 783 2011, Riahi et al. 2011), which are likely to be similarly distributed to VSLS-Cl (e.g. regression in Figure 4b and Shao et al. 2011), to model anomalous CO volume mixing ratios (i.e. those 784 due only to these industrial emissions north of 20°N) at Bachok at 3-hourly resolution for six 785 recent NH winters (Oct-Apr 2009/10-2014/15). Figure 4a in the main paper shows this 786 787 modelled quantity over winter 2013/14, during which the Bachok observations were made, as an example. The observed peak in VSLS-Cl is 1) captured well by the model, and 2) likely to 788

be a regularly repeated event. In total during this winter there are ~57 days (i.e. 453 3-hour periods) with a modelled mixing ratio above a threshold of 25 ppb, and 19 days above 50 ppb.
To demonstrate that winter 2013/14 was not unusual, the modelled CO anomalies for the other 5 winters examined are shown in Figure S2.

794 **2.2 NAME animations** 795

The two animations (Jan2014.mp4 and Feb2014.mp4) show 3-hourly NAME footprints of air arriving at Bachok in January and February 2014 and indicate where surface emissions have an influence on the composition of air arriving at the site. The animations give an indication of the frequency that air arriving at Bachok has been influenced by emissions from East Asia.

800 801

793

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- region, China. *Atmos. Chem. Phys.*, **11**, 5011-5025, 2011.
- 830





Figure S1: Mole fractions (ppt) of 3 chlorinated VSLS in air samples collected at Hengchun,

Taiwan in March/April 2013. Note that CH₂CICH₂CI was not monitored in the 2013 samples. The error bars are ± 1 standard deviation.



839

Figure S2: Time-series of the modelled carbon monoxide (CO) anomaly at Bachok, due only to industrial emissions from north of 20°N in the previous 12 days, for six winter seasons. The period of observations at Bachok during Jan and Feb 2014 is shaded in grey. Also shown are the number of days in each winter which exceed the 25 ppb and 50 ppb thresholds which, using the regression equation in Figure 4b, correspond to 176 ppt and 315 ppt of CH₂Cl₂.

846	Response to Reviewers
847	We would like to thank the reviewers for their very helpful comments and we have

addressed these as follows (reviewer's comment in bold):

849 **<u>Reviewer 1</u>** (Sinnhuber)

As one general comment I feel that the presentation of CH2Cl2 emission estimates and the correlation with CH2ClCH2Cl to infer new CH2ClCH2Cl emission estimates (lines 337 - 343) deserves (and requires) more detail, given its importance. Part of the information on estimating the CH2Cl2 emissions given in the supplementary material should be included in the main text and a bit more explanation on the "simple correlation" should be included.

We have moved the discussion about the emission estimates from the supplement into the manuscript as suggested. We have also made a slight modification to our analysis and rather than giving the extreme range of potential CH_2Cl_2 emissions as before (based on the 40:60 and 50:50 production ratios) we have opted to give an estimate based on the more likely ratio of 45:55. This leads us to production and emission figures of 715 kt and 455 kt respectively, with an approximate uncertainty of \pm 10%.

We have also expanded the section on the correlation between the two compounds to include the following text

"There is a strong linear correlation between the observed CH₂Cl₂ and CH₂ClCH₂Cl data at 864 both Bachok ($R^2 = 0.9799$) and Cape Fuguei ($R^2 = 0.9189$). Combining the datasets yields a 865 slope of 0.4456 \pm 0.0194 (R^2 =0.9228). Using the emissions for CH₂Cl₂ derived above (455) 866 kt) and making the assumptions that (1) all emissions originate in China and (2) there are no 867 significant relative losses of the two compounds since emission (lifetimes are 144 days for 868 CH_2CI_2 and 65 days for CH_2CICH_2CI) we can estimate Chinese emissions of CH_2CICH_2CI to 869 be of the order of 203 \pm 9 kt yr⁻¹. If accurate, the scale of these emissions is a major surprise 870 as CH_2CICH_2CI is highly toxic (suggesting that local emissions would be minimised) and 871 believed to be used almost exclusively in non-emissive applications." 872

Abstract, I.31: "higher than expected": what is this expectation based upon? Based on previously reported measurements? On line 360 and following it is discussed that many of the previous measurements have been made over a decade ago and in different regions ("... not the 2 key regions ..."). I believe it would be good to make a bit clearer from the start if the enhancements seen in this study are likely because of recent increases in emissions, regional differences, or both.

By "higher than expected" we do indeed mean higher than previously reported data. This 879 was stated earlier in the sentence and refers to both our surface and aircraft measurements. 880 881 We compare our data with the most recent review (WMO, 2015) as we describe in the text (Section 3, line 236-238) and in Table 1. The enhancements we observe are likely to result 882 from a combination of increasing emissions and the location of the measurements, although 883 884 based on our measurements alone, which are over a relatively short period of time, we have 885 no evidence that emissions are increasing. We do know from previous work that atmospheric levels of CH₂Cl₂ have increased, which implies increasing emissions of this 886 compound. The long term trend of CH₂ClCH₂Cl is unknown. 887

888 I.36: define "CI-VSLS" when first used. Moreover better use this consistently 889 throughout (e.g. Table 1 uses VSLS-CL, which I suppose means the same)

890 Done

- 891 I.45: you may want to cite also the recent study by Hossaini et al., The increasing
- threat to stratospheric ozone from dichloromethane, Nature communications, 2017,
 that was published after submission of the present manuscript.
- As our paper was published in ACPD before the Hossaini et al paper it does not seem appropriate to include it in our reference list at this stage.
- 896 I.104: I suggest to break the sentence in two: "... in the TTL. Surface measurements
 897 ..."
- 898 Done
- 899 I. 126: I don't understand the meaning of "globally" here.
- 900 We have removed the word "globally".

901 I.129: "shorter lifetimes" could be misleading here, as it may imply lifetimes shorter

- than the 10 days for air masses to travel from East Asia to the TTL, which is probably
 not what is meant?
- 904 Text changed to "despite their relatively short atmospheric lifetimes"

905 I.150: "the CARIBIC aircraft": better include a sentence or two on the CARIBIC project,
 906 describing that these are measurements from in-service aircrafts, ideally including a
 907 reference paper (in addition to the http link).

908 Done

909 I.328: "CH2CICH2CI is exclusively anthropogenic in origin ...": WMO (2014) lists also
 910 biomass burning as a source of CH2CICH2CI. Can you include references on
 911 additional sources?

912 We have added the following text: "Simpson et al. (2011) observed a small enhancement in

913 CH₂CICH₂CI in Canadian boreal forest fire plumes (background average, June-July 2008,

914 9.9 \pm 0.3 ppt, plume average 10.6 \pm 0.3 ppt) and estimated a global boreal fire source of

915 0.23 ± 0.19 kilotonnes (kt) yr¹." Of the references given in WMO 2014 (Chapter 1, page

- 916 1.38), this is the only one that reported CH_2CICH_2CI .
- 917 We have also changed the word "*exclusively*" to "*predominantly*" when referring to 918 anthropogenic sources of CH₂CICH₂CI.
- 919 I.334: "Production has increased rapidly ...": Can you give a reference for this
 920 increase in production?
- 921 Reference added (same as in the next sentence)

922 I. 362: Does the superscript "1" have any meaning? Footnote?

- The superscript was actually a missing reference, which has now been added (Carpenter and Reimann et al. 2015).
- 925 I. 367: "2"-> "two"
- 926 Done
- Table 1: Why not use the IATA code "FRA" for Frankfurt (rather than "FFT", which is
 the IATA code for Frankfort, Kentucky)?
- 929 Done

Table 1: Why is the sum of VSLS-CL excluding CH2CICH2CI not given for the other data for comparison?

932 The sum of CI-VSLS excluding CH₂CICH₂CI is not a widely used number so it was not

included in the Table apart from where necessary. There is no equivalent number reported in
 WMO. However, we have added the information for the CARIBIC flights between Frankfurt

and Bangkok for comparison as suggested.

936 **Reviewer 2**

In general, all the information on the samples: time, locations for all the data, number
 of flights for CARIBIC data, number of sampling at the ground-based stations should
 be added in this section. For instance, the 7 IAGOS-CARIBIC flights time should be
 mentioned. Days and months of samples should be specified in this section as well. It

941 will help the reader to get the general feature of the sampling.

The altitude of CARIBIC needs to be shown. Have you filtered IAGOS-CARIBIC data to analyze data between 10 and 12 km only? The statistics of the sampling in this layer is needed.

945 We have added more detail at the beginning of the methods section:

- 946 "A total of 21 samples were collected at Hengchun between 7 March and 5 April 2013 with a
- 947 further 22 samples taken at Cape Fuguei between 11 March and 4 April 2014. 28 samples
- were collected at Bachok between 20 January and 5 February 2014, during the period of the
- NE winter monsoon. The approximate location of each surface site is shown in Figure 1. The
- 950 CARIBIC aircraft samples were collected during seven return flights between (i) Frankfurt
- 951 (Germany) and Bangkok (Thailand), and (ii) Bangkok and Kuala Lumpur (Malaysia) during
- the periods December 2012 March 2013 (4 flights) and November 2013 January 2014 (3 flights) All CAPIPIC flights in this region between December 2012 and January 2014 hours
- 953 flights). All CARIBIC flights in this region between December 2012 and January 2014 have 954 been included in this analysis. With the exception of 3 samples that were taken at altitudes
- been included in this analysis. With the exception of 3 samples that were taken at altitudes
 between 8.5 and 9.8 km, the CARIBIC samples were all (n=179) collected at altitudes

956 between 10 and 12.3 km."

957 The full CARIBIC flight dates have also been added to Figure 5.

958 In the text it is mentioned that 10-12km over East Asia is the lower boundary of the

959 TTL. It would be very helpful to show a map of TTL or a figure of TTL and aircraft

960 altitudes together with respect of the flight tracks (latitude). It would be also useful to

961 directly refer to Box 1-3, Figure 1 of Carpenter and Reimann et al. (2015) that shows

- 962 the altitude range of the TTL.
- We have added a reference to the Figure from Carpenter and Reimann et al. (2015). We do not think it is necessary to reproduce a similar Figure here.
- About the results shown in Figure 3, a sentence explaining that three days have been
 chosen out of the seven days of the cold surge event would be helpful. The term "cold
 surge" should be mentioned.
- 968 New text added: "Three examples during this cold surge event are shown in Fig. 3 (b-d)."

In general, "see in supplement" is largely used in the manuscript but I would suggest
 to refer to figure number and section names of the supplement materials to help the
 reader.

972 Done

- 973 Results from Carpenter and Reimann et al., 2015 are cited as reference for CI-VSLS
- 974 last information. For results at the lower TTL, it would be useful to recall the type of
- 975 observations used in the assessment report: aircraft campaigns and balloons.
- We have added the following sentence to Table 1: "*Data from the TTL was derived from various aircraft and balloon campaigns.*"
- **Table: Units need to be added in Table 1 and its caption**
- 979 Done
- In general, the way to write CI-VSLS should be consistent along the captions and the
 text (sometimes chlorinated VSLS or VSLS-CI).
- 982 Done
- Figure 1: "(red crosses)" for surface sites on the map need to be added in the caption to guide the reader.
- 985 Done
- 986 **Figure 2: We don't see the blue circle on the map.**
- We apologise that the incorrect Figure was included in the original submission. The blue circles have now been added.
- 989 Figures 2,3,5: Helpful to have a recall of ground-based stations location.
- Blue circles have been added to Figures 2 and 3 showing the location of the surfacesampling sites.

Figure 2 and 3: I would rather use letter to name the panels in the caption and I would rather use numbers to link plots in upper panel with maps of the bottom panels.

- We use letters to label the arrows which refer to the individual NAME plots underneath.Mixing letters and numbers would, in our opinion, be more confusing.
- Figure 3: Use arrows as for Figure 2 to help the reader to find the days that are
 chosen for air masses origin (map below). In the caption it is mentioned "true
 background levels", how these levels are estimated? Figure 3 a) is not specified, "a)"
 should be added on the figure.
- Arrows have been added to Figure 3 as requested. The baseline levels are taken from WMO 2015, based on tropical mean background levels. The actual expected background is difficult to define for VSLS as they would be expected to vary substantially across the globe (i.e. with latitude and with distance from source). We have added a reference to WMO in the Figure caption.
- 1005 The missing (a) has been added to Figure 3.

1006 Figure 4 and S2: What does CO anomaly mean? What is the reference value?

By the term "CO anomaly" we mean the fraction of CO observed at Bachok from industrial emissions from regions north of 20N. This is explained in Section 3 (lines 285-288) and in the supplement. We have added a sentence in the caption for Figure 4 to remind readers.

1010 Line 126: "Both in the western Pacific region and globally". It is not clear what 1011 globally means.

- 1012 We have removed the word "globally".
- 1013 Line 147: "Various time" needs to be specified (see general comments).
- 1014 More sampling information has been added (see earlier response to reviewer 2))
- Line 150: Change "CARIBIC aircraft" to "IAGOS-CARIBIC aircraft" as CARIBIC is part
 of the IAGOS program.
- 1017 Done

1018 Line 230: Change "shows the 2014 data …" to "shows data from CAPE Fuguei in the 1019 end of March, beginning of April 2014".

- 1020 Text has been changed to "*shows the March/April 2014 data…*". In addition the sampling 1021 dates have been defined more clearly in the text (see earlier comment of reviewer 2). The 1022 dates are also evident in the Figures.
- 1023 Line 237: Change "March/April 2013" to "mid of March/beginning of April 2013".

1024 To be consistent with the text in the previous comment we have not made this change but 1025 note that the sampling dates have been defined more clearly in the text (see above). The 1026 dates are also evident on the Figures.

- Line 246: "January/February": the entire months are not shown so "end of
 January/beginning of February" would be more appropriate. "During this phase of the
 monsoon": A recall about the Asian Monsoon phases and references would be useful,
 maybe recall that it is the East Asian winter monsoon circulation as mentioned in the
 introduction.
- 1032 Text has been changed to *"late January/ early February 2014"*. Sampling dates have been clarified in methods section.
- 1034 We have added the words "*East Asian*" when referring to the monsoon and reminded 1035 readers that this was described earlier.
- Line 252: change "often" to "most of the case studies we are analyzing here" or "for N days out of Ntotal days of observations" or give the information in Line 277:
- We would prefer to keep the word "often". This was deliberately vague as we have not done any specific analysis to determine how many times the air may have picked up emissions from Taiwan. The observation was also based on the NAME animations which are referred to in line 285.

1042 Change "(see supplement for further details)" to "(Fig. S2 in the supplement 1043 material)".

- 1044 Done
- 1045 Line 287: Change "in all CARIBIC flights" to "in the seven CARIBIC flights" and 1046 remove "(7)".
- 1047 Done
- 1048 Line 334: Need a reference.
- 1049 Reference added
- 1050 Line 362: Need a reference.

1051 Reference added

1052 Line 383: Examples of other chemical pollutants would be useful.

1053 The pollution we were referring to are chemicals that are routinely found in industrialised

1054 countries. These include CO, O₃, CH₄, volatile organic compounds (VOCs) including

1055 hydrocarbons, oxygenated hydrocarbons and certain halocarbons. These measurements are

- 1056 likely to feature in a future publications.
- 1057 Line 384: Remove "etc".
- 1058 Done