

Reply to reviewer #1: Suggestions for revision or reasons for rejection of “The Horizontal Ice Nucleation Chamber HINC: INP measurements at Conditions Relevant for Mixed-Phase Clouds at the High Altitude Research Station Jungfraujoch” by Larissa Lacher et al.”

5 Reviewer comments are reproduced in **bold** and our responses in normal typeface; extracts from the revised manuscript in *blue italic*.

I have read through the reviewer's reply to my suggestions / comments.

10 **It was useful to have the explanation that we need to consider the particles experiencing "constant" conditions for approximately 8 seconds, that each particle does not experience the "mean RH", and also the explanation about deliquesce occurring but not appearing in the measurement (because of the same problem). These difficulties are acknowledged by the authors and explanations are given.**

15 **However, I still feel the authors need to be careful that some of the figures are not over interpreted by readers, and caveat them accordingly, given that the errors in RH are such that it is difficult to quantitatively assess homogeneous nucleation and droplet activation when comparing with theory. Instrument scientists working in this area may be well aware of these difficulties, but modellers may not.**

Given the discussion about particles not experiencing the "mean humidity" the implications therefore are that the instrument would generally underestimate both the amount of droplet activation and homogeneous nucleation at a measured RH; however, I don't think this is actually said in the manuscript - not clearly anyway - for instance perhaps it should be brought more to the fore in the conclusions. This is the main point I have.

20 We dedicate an entire section of the manuscript to the uncertainties and variation of RH in the chamber (see section 2.2.2.) where we discuss the range of RH experienced by the particles. In addition, each figure where AF curves as a function of RH are shown, we explicitly show a representation of this uncertainty by grey or hashed shading. We believe this is a very clear indication to the reader. Also, it is not true that the instrument always underestimates the amount of nucleation/activation because the droplets could be exposed to e.g. $RH_w \pm 2\%$
25 because of the range of RH_w experienced. See figure below for example where theoretically calculated AF using nucleation rates from Koop et al. (2000) are lower than those measured in HINC. The AF that theoretical

calculations would predict at $RH_w = 99.9$, we already measure those at $RH_w = 99\%$. To clearly emphasize this we added the following in section 2.2.4, page 15, lines 13 – 16:

5 *“As can be seen from Fig. 6, some deviation (all within instrument uncertainty) of the data points does occur. This is due to the fact that in the region of ice nucleation or droplet activation the AF (nucleation rate) is a very steep function of RH and a small change in water activity by for example 0.01 (~RH of 1% at equilibrium) can change the nucleation rate by a factor of 10 or 15.”*

We do agree with the reviewer, that this can be mentioned more clearly in the conclusions and we now do this on page 34, lines 13 – 18:

10 *“The uncertainty in INP measurements in HINC arises from the variation in temperature and RH, to which the aerosols in the chamber are exposed to, and are $T \pm 0.4$ K, and $RH_w \pm 1.5\%$ ($RH_i \pm 3\%$) for temperatures $T > 235$ K. These variations can lead to uncertainties in activated fractions in the region of activation and nucleation where nucleation rates are a steep function of RH. Conducting field measurements of INPs with ambient aerosols at 242 K and RH 104%, where we ensure all particles are fully activated to droplets, reduces the associated uncertainties*
15 *in particle concentrations arising from RH variation in the chamber.”*

My specific replies to previous points raised are that:

1. In response to my comment about figure 2: "Our experiments reveal ... excellent agreement". I would not call this "excellent" agreement. By those arguments, given the range in uncertainty in RH you could have
20 presented any value between zero and 1 for the activated fraction and called it excellent agreement. I would say that the general behaviour is consistent with that expected, but it is difficult to use it quantitatively due to errors.

We agree with the reviewer and deleted “excellent” on page 8, line 19.

25 **2. Figure 1.2 in the reply. I also do not agree with the red and blue stars:**

Koop et al (2000) suggests the nucleation rate is around $1.5 \times 10^{16} \text{ m}^{-3} \text{ s}^{-1}$ at these conditions. The volume of water on the particles is around $1.8 \times 10^{-20} \text{ m}^3$ so this suggests a frozen fraction of 2×10^{-3} for a residence time of 8s (an order of magnitude less than in their Figure).

Please note that my calculations also had the same range of growth factors predicted EAIM (using 2 for the van Hoff factor gives a growth factor of 3, using 3 gives a growth factor of 4, so there is no discrepancy in that calculation)

We agree with the reviewer and found an error in our code. The nucleation AF and water activities were interchanged in our previous calculations arriving at the wrong AF using theoretical calculations. We thank the reviewer for their thorough comments and have now presented below the theoretical AFs of 1 and 8 seconds.

We agree with the reviewer that the calculated AF at 233 K and $\text{RH}_w = 98.6\%$ is one order of magnitude lower compared to what we observe in HINC for 8 seconds (and 2 orders of magnitude lower for 1 second). As noted by the given uncertainties of $\text{RH}_w \pm 2\%$, this would represent a shift in RH of 1 % (water activity of 0.1). Because nucleation rate is such a steep function of water activity in this region, a small uncertainty or range of RH experienced results in this change in AF. We now acknowledge this on page 15, lines 13 – 16 (section 2.2.4) and page 34, line 13 – 18 (in the conclusions).

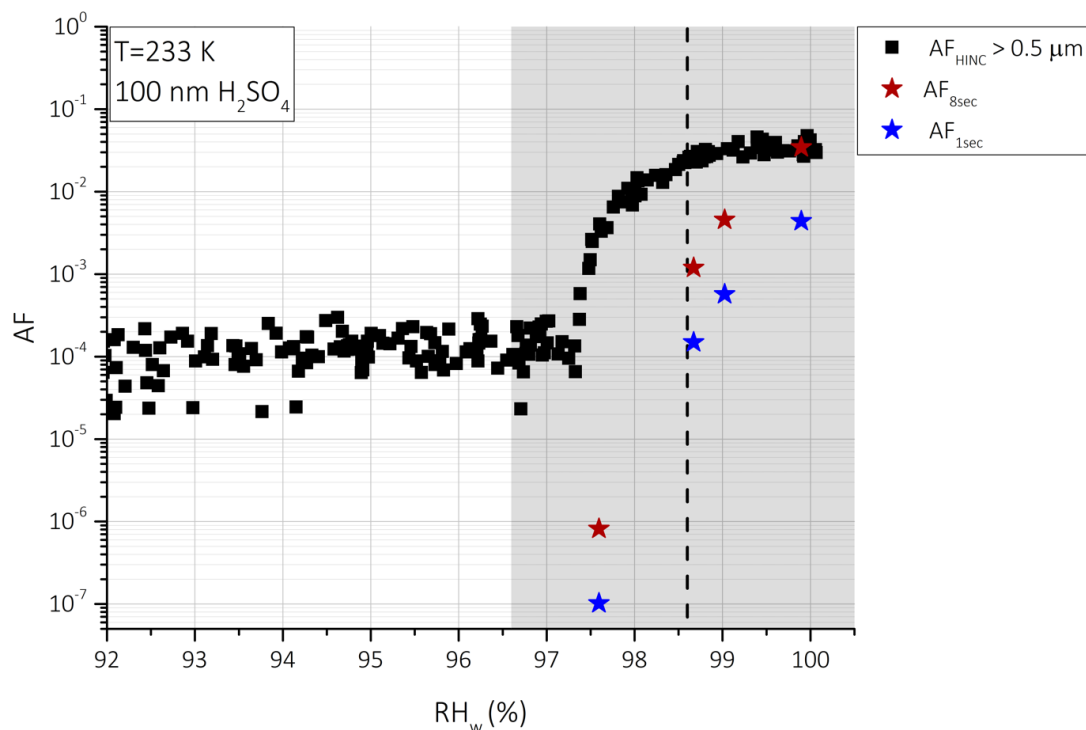


Figure 1: AF as a sum as function of RH_w at 233 K for initial 100 nm H_2SO_4 ; black markers refer to observed AF in OPC size channel $> 0.5 \mu m$ in HINC; blue stars are based on a nucleation time of 1 second, and red stars on 8 seconds; grey area refers to the calculated RH variation and uncertainty; droplet volumes are calculated with E-
 5 AIM Koehler model (<http://www.aim.env.uea.ac.uk/aim/aim.php>), and nucleation rates are inferred from Koop et al. (2000a).

3. The liquid activation curves (figure 3 in the manuscript), have the potential to be confusing. They show particles activating up to an RH of 102%. However, from another point of view they show how the instrument behaves and hence are useful if presented in the right way, with a discussion of time-scales.

10 In order to clarify the reviewer's concern we have now added the following on page 8, line 30 – page 9, line 3:

“As discussed above, the exposure of the aerosol particles to this variable RH leads to droplet activation not only at $RH = 100\%$ but in a range of RH pertaining to the variation and uncertainty in HINC. As such aerosol particles grow within the residence time of 8 seconds to different sizes, which is depicted by the difference in the OPC size channels. In addition, this is also the reason why at $RH_w = 101\%$ for example (Fig. 3), the AF is still observed to be

increasing because some particles in the aerosol layer are still exposed to $RH_w = 99 - 100\%$, explaining why droplet activation is observed up to an RH of 102%.”

References:

- 5 Koop, T., Luo, B., Tsias, A., and Peter, T.: Water activity as the determinant for homogeneous ice nucleation in aqueous solutions, *Nature*, 406, 611-614, 10.1038/35020537, 2000a.

The Horizontal Ice Nucleation Chamber HINC: INP measurements at Conditions Relevant for Mixed-Phase Clouds at the High Altitude Research Station Jungfraujoch

Larissa Lacher¹, Ulrike Lohmann¹, Yvonne Boose^{1~}, Assaf Zipori^{2^}, Erik Herrmann³, Nicolas Bukowiecki³, Martin Steinbacher⁴ and Zamin A. Kanji¹

¹Institute for Atmospheric and Climate Science, ETHZ, Zurich, 8092, Switzerland

²Institute for Earth Science, Hebrew University, Jerusalem, 76100, Israel

³Laboratory of Atmospheric Chemistry, Paul Scherrer Institute, Villigen, 5232, Switzerland

10 ⁴Empa, Swiss Federal Laboratories for Materials Science and Technology, Duebendorf, 8600, Switzerland

[~]now at Karlsruhe Institute of Technology (KIT), Institute of Meteorology and Climate Research, Garmisch-Partenkirchen, 82467, Germany

[^]now at Weizmann Institute of Science, Department of Earth and Planetary Sciences, Rehovot, 7610001, Israel

15 Correspondence to: Larissa Lacher (larissa.lacher@env.ethz.ch), Zamin. A. Kanji (zamin.kanji@env.ethz.ch)

Abstract. In this work we describe the Horizontal Ice Nucleation Chamber, HINC, as a new instrument to measure ambient ice nucleating particle (INP) concentrations for conditions relevant to mixed-phase clouds. Laboratory verification and validation experiments confirm the accuracy of the thermodynamic conditions of temperature (T) and relative humidity (RH) in HINC with uncertainties in temperature of ± 0.4 K and in RH with respect to water (RH_w) of $\pm 1.5\%$, which translates into an uncertainty in RH with respect to ice (RH_i) of $\pm 3.0\%$ at $T > 235$ K. For further validation of HINC as a field instrument, two measurement campaigns were conducted in winters 2015 and 2016 at the High Altitude Research Station Jungfraujoch (JFJ; Switzerland, 3580 m a.s.l.) to sample ambient INPs. During winters 2015 and 2016 the site encountered free tropospheric conditions 92% and 79% of the time, respectively. We measured INP concentrations at 242 K at water subsaturated conditions ($RH_w = 94\%$), relevant for the formation of ice clouds, and in the water supersaturated regime ($RH_w = 103\text{--}104\%$) to represent ice formation occurring under mixed-phase cloud conditions. In winters 2015 and 2016 the median INP concentrations at $RH_w = 94\%$ was below the minimum detectable concentration. At $RH_w = 104\%$, INP concentrations were an order of magnitude higher, with median concentrations in winter 2015 of 2.8 per standard liter (stdL^{-1} ; normalized to standard temperature $T = 273$ K and pressure $p = 1013$ hPa) and 4.7 stdL^{-1} in winter 2016. The measurements are in agreement with previous winter measurements obtained with the Portable Ice Nucleation Chamber, PINC, of 2.2 stdL^{-1} at the same location. During winter 2015, two events caused the INP concentrations at $RH_w = 103\text{--}104\%$ to significantly increase above the campaign average. First, an increase to 72.1 stdL^{-1} was measured during an event influenced by marine air, ~~coming~~ arriving at the JFJ from the Northern Sea and the Norwegian Sea. [The contribution from anthropogenic or other sources can thereby not be ruled out.](#) Second, INP concentrations up to 146.2 stdL^{-1} were observed during a Saharan dust event. To our knowledge this is the first time that a clear enrichment in ambient INP concentration [in remote regions of the atmosphere](#) is observed during a time of marine air mass influence, ~~indicating~~ [suggesting](#) the importance of marine particles on ice nucleation in the free troposphere.

1 Introduction

Clouds and aerosols continue to cause the largest uncertainty in the current assessment of global climate change (e.g. Boucher et al., 2013). Despite their importance in the Earth's system, fundamental knowledge on cloud formation and evolution is still missing. Clouds containing ice can have a positive or negative effect on the Earth's radiative budget, depending on their micro- and macrophysical properties (Lohmann et al., 2016). Cloud microphysical processes are highly variable depending on the available amount of water vapor and the presence of supercooled cloud droplets and ice crystals. In addition, cloud microphysical processes can change during the development of a cloud, and the first formation of ice in clouds is still not completely understood.

Different processes leading to ice formation from the vapor or liquid phase are possible. In the absence of ice nucleating particles (INPs) (Pruppacher and Klett, 1997; Lohmann et al., 2016) freezing of supercooled droplets occurs homogeneously, which is relevant in pristine atmospheric environments. It requires temperatures $T < 235$ K and a relative humidity (RH) with respect to ice $(RH_i) > 140\%$ where the nucleation rate is high enough to outcompete heterogeneous nucleation. In the presence of INPs, heterogeneous ice nucleation is favored, since the particles can lower the energy barrier of the phase change. This freezing pathway is dominant in the mixed-phase cloud regime at temperatures $T > 235$ K, where ice and supercooled water can co-exist and homogeneous freezing rates are negligible. Currently, four different heterogeneous freezing mechanisms are distinguished: Deposition nucleation, contact freezing, immersion freezing and condensation freezing (for a detailed description see Vali et al., 2015). Deposition nucleation is relevant for the formation of cirrus clouds as inferred by Cziczo et al. (2013), but is typically not relevant for the formation of mixed-phase clouds, as lidar observations show that the liquid phase is present before ice crystals form (Ansmann et al., 2008). The two most likely freezing modes in mixed-phase clouds are immersion/condensation freezing, where the INP initiates the freezing from within a supercooled droplet. At present it is questioned if there is a physical difference between immersion and condensation freezing (Welti et al., 2014; Wex et al., 2014; Vali et al., 2015; Burkert-Kohn et al., 2017) but as the ice germ should form from the liquid phase in both cases, it is not expected so.

In addition to different possible ice formation pathways, the identification of ambient INPs remains challenging, since only a small fraction of aerosol particles (~ 1 out of 10^5) nucleates ice (Rogers et al., 1998; DeMott et al., 2010), and the exact properties rendering them ice-active are not known. INPs can be solid and water-insoluble, or also soluble and crystalline (Kanji et al., 2017), and their ice nucleation ability has been linked to a crystal lattice match to ice, surface defects which increase the density of adsorbed water molecules locally, or by functional groups which increase the chemical affinity to ice via hydrogen bonds (Pruppacher and Klett, 1997). From the large variety of ambient aerosol classes (Kanji et al., 2017), mineral dust particles were observed to nucleate ice **below 258 K efficiently** (Hoose and Möhler, 2012 and references therein), and it has been found that K-feldspars are the most efficient INPs out of many tested minerals (Atkinson et al., 2013b; Yakobi-Hancock et al., 2013; Zolles et al., 2015; Harrison et al., 2016; Kaufmann et al., 2016). Furthermore, due to its abundance in the lower free troposphere (FT), it is thought that mineral dust plays a key role in atmospheric ice nucleation (e.g. DeMott et

al., 2003a; Kamphus et al., 2010). Particles of biological origin, like certain bacteria, fungal spores and pollen, were found to be efficient INPs at temperatures $T > 263$ K (Hoose and Möhler, 2012), however, the atmospheric concentration from whole and intact biological particles which are ice-active is temporally and spatially variable, and their influence on ice formation is therefore rather seasonal and local in nature (Després et al., 2012). Nanometer scaled fragments from biological particles are present in much higher concentrations and might have atmospheric implications (Pummer et al., 2012; Augustin et al., 2013; O'Sullivan et al., 2015; Fröhlich-Nowoisky et al., 2015; Wilson et al., 2015). Additionally, bacteria have been found in Saharan (Meola et al., 2015) and soil dust aerosols (Conen et al., 2011), possibly influencing their ice nucleation activity.

The role of marine aerosol as a source of INPs has been reported for the first time more than five decades ago (Brier and Kline, 1959; Cziczo and Froyd, 2014) and has been re-emphasized recently and observed in various field and laboratory studies (Cziczo et al., 2013; Knopf et al., 2014; Wilson et al., 2015, DeMott et al., 2016). Recent field studies or studies of field samples in the laboratory (Cziczo et al., 2013; Knopf et al., 2014; Wilson et al., 2015; Ladino et al., 2016; DeMott et al., 2016) have shown that particles and organic matter sampled or emitted from the sea surface can be a source of INPs. Marine aerosols are produced via a bubble-bursting mechanism (e.g. de Leeuw et al., 2011; Gantt and Meskhidze, 2013; Aller et al., 2005; Cunliffe et al., 2013) when entrained air bubbles rise through the sea surface microlayer and burst upon contact with the atmosphere. The sea surface microlayer is usually enriched in biogenic material leading to the emission of these in the atmosphere as aerosol particles. A source of these marine particles can be microorganisms like phytoplankton and bacteria, exopolymer secretion, colloidal aggregates, glassy organic aerosols, crystalline hydrated sodium chloride particles and frost flowers (summarized in Burrows et al., 2013). Cells or cell fragments and exudates of phytoplankton species were found to be ice-active (Knopf et al., 2011; Alpert et al., 2011; Wilson et al., 2015), and biological material during phytoplankton blooms might also play an important role for ice nucleation (Prather et al., 2013; DeMott et al., 2016). These marine aerosols can be sub-micrometer in size (e.g. $0.02 - 0.2$ μm , Wilson et al., 2015; $0.25 - 1$ μm , DeLeon-Rodriguez et al., 2013), a size range which is transported to higher altitudes. Burrows et al. (2013) state that marine biogenic particles are an important source of INPs in remote marine areas in the absence of other efficient INPs such as mineral dust. During airborne measurements, Cziczo et al. (2013) found sea salt in ice residuals from tropical tropopause cirrus clouds, especially over the open ocean, but also in reduced concentrations over land.

In addition to laboratory studies, which aim to understand the physical processes of ice nucleation and determine key aspects of aerosols acting as INPs, it is crucial to quantify the total number concentration of ambient INPs in an environment relevant for clouds containing ice and to address the question of their variability in space and time. Several studies exist from airborne platforms (e.g. Bigg, 1967; Rogers et al., 1998; Prenni et al., 2009; DeMott et al., 2010; Avramov et al., 2011, Schrod et al., 2017), and ground-based observations (e.g. DeMott et al., 2003b; Chou et al., 2011; Ardon-Dryer and Levin, 2014; Mason et al., 2016; Boose et al., 2016a; Boose et al., 2016b) quantifying the number concentration of INPs and identifying their potential sources. Typically, filter sampling with subsequent offline freezing methods, and online measurements with continuous-flow-diffusion chambers (CFDCs) are used as INP measurement techniques. For filter sampling, aerosols are collected for a certain

time and known air volume, after which the collected particulate is cooled and exposed to controlled temperature and RH conditions (e.g. Bigg, 1967; Santachiara et al., 2010; Conen et al., 2011; Bingemer et al., 2012; Ardon-Dryer and Levin, 2014; Knopf et al., 2014; Mason et al., 2015). Filter techniques observe the onset freezing temperature of a sample with a very large number of particles resulting in a very sensitive detection limit. However, this comes at the cost of a low temporal resolution since the sampling times of the filters often are on the order of a few hours or longer. CFDCs measure INP concentrations in real-time with a higher temporal resolution, on the order of a few to tens of minutes (e.g. Rogers, 1988; Rogers et al., 2001; Chou et al., 2011), but their total sampling volume is lower, and their sensitivity to detect INPs is limited at low concentrations (Boose et al., 2016a). This in particular is challenging at low supercooling or in areas where INP concentrations are lower than 0.1 – 1 per standard liter (stdL^{-1} ; normalized to standard temperature $T = 273 \text{ K}$ and pressure $p = 1013 \text{ hPa}$).

Measurements of INP concentrations during different seasons in an environment which is relevant for the formation of mixed-phase clouds are rare. Conen et al. (2015) collected filters at different elevations in the (partly) FT, namely at Mt. Chaumont and at the Jungfrauoch (JFJ) in Switzerland (1171 m and 3580 m, respectively) and at the Izaña observatory on Tenerife, Canary Islands (2373 m). They sampled for one year with one sample representing 24 hours, and INP measurements were reported for the temperature range 265 K to 269 K. At the JFJ, and for the same temperature range, they found a seasonal cycle with INP concentrations ranging from 0.001 to 0.01 stdL^{-1} , with a maximum in summer and a minimum in winter. This variation is attributed to previous INP activation and subsequent fall-out from advected air masses prior to reaching the JFJ, leaving an air mass which is depleted in INPs upon arrival to the JFJ during wintertime (Conen et al., 2015; Stopelli et al., 2015). Also at the JFJ, INP measurements were performed at temperatures $T = 241 \text{ K}$ during winters 2012, 2013 and 2014 with the CFDC, Portable Ice Nucleation Chamber (PINC; Boose et al., 2016a). INP concentrations were sampled in the deposition nucleation mode in winters 2012 – 2014, and in winter 2014 also in the condensation freezing mode. Median INP concentrations below (above) water saturation were in the range of $\leq 0.05 - 0.1 (4.2) \text{ stdL}^{-1}$. To extend these measurements and to establish a longer time series of measurements at the JFJ, INP concentrations are measured since summer 2014 at the same temperature and RH conditions, with the newly built Horizontal Ice Nucleation Chamber (HINC) based on the design of Kanji and Abbatt (2009). In this study, the new chamber is characterized to be used ~~not only for laboratory studies, but also~~ as a field instrument. To complement the validation and verification experiments performed in the laboratory, two field campaigns in winters 2015 and 2016 were performed, and results are compared to previous winter measurements from the same location discussed in Boose et al. (2016a). In addition, two events of anomalously high INP concentrations from the winter 2015 campaign are discussed to investigate the origin of these INPs.

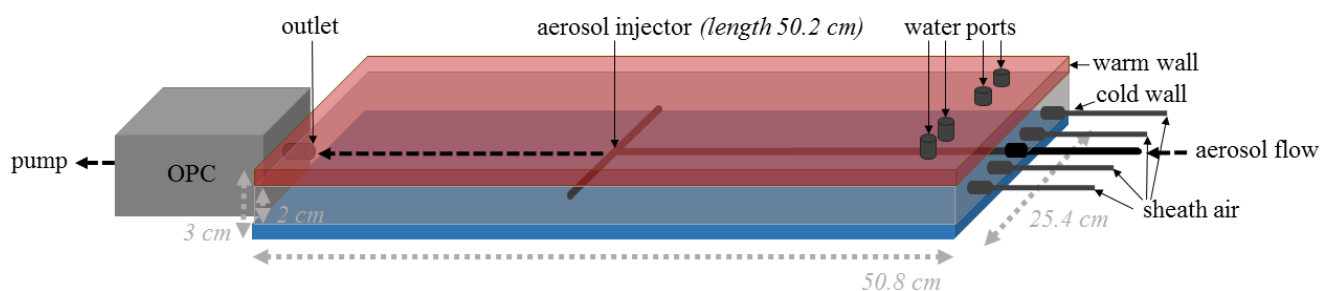
30 **2 Ice nucleation measurements**

2.1 Technical description

HINC is a continuous flow thermal gradient diffusion chamber, based on the design of the UT-CFDC (Kanji and Abbatt, 2009). A schematic of HINC is shown in Fig. 1, including the outer dimensions of the chamber. Inner dimensions and more detailed design aspects can be found in Kanji and Abbatt (2009). HINC consists of two horizontally oriented copper plates which are cooled by an external re-circulating ethanol chiller (LAUDA, RP 890 C). Self-adhering glass fiber filter papers (PALL 66217) mounted on the inner walls of the chamber are wetted prior to an experiment to create an ice layer upon cooling the walls. For the wetting procedure, the walls ~~temperatures~~ are kept at room temperature and the chamber is tilted to an angle of 45°, and approximately 100 ml of double-deionized water is used to wet the filter papers via four water ports, which are in contact with the filter papers of ~~the cold and warm~~ the upper and lower walls. The chamber is kept in this position for 30 minutes to drain excess water via the outlet port downstream of the chamber. After draining, the chamber is brought back into a horizontal position and the outlet port is dried to ensure no residual water drops are retained. Following the wetting procedure, an optical particle counter (OPC, MetOne, GT-526S) is attached to the outlet port, and the walls are cooled down to the desired set point temperature $T < 273$ K. To establish a RH_i and $RH_w > 100\%$, a temperature gradient ΔT is applied between the two ice coated walls (both walls below 273 K), with the upper wall set to the warmer temperature. The horizontal orientation of the chamber ensures no internal convection. For an experiment where the RH should be increased ~~at a constant center~~ temperature, which is typical to determine the onset RH of INPs, the ΔT is achieved by a temperature increase and decrease of the respective walls at equal rates. This ensures that the temperature in the center, where the air containing the aerosols is injected, remains constant. Aerosol enters the chamber via a movable injector, and are released into the chamber via a slit in the front of the injector. The cross section area of the slit is smaller than the cross section of the inner diameter on the injector, which creates a small overpressure at the particles exiting through the slit promoting an equal distribution of the aerosols over the width of the slit. The outer diameter of the injector is 6.35 mm (inner diameter 3.175 mm) which ensures that at a flow rate of 2.8 stdL min⁻¹, and at 242 K the turbulence regime is not encountered by air flowing over the aerosol injector (Reynolds number 880, well below the threshold of 2000 for turbulent flow). Furthermore given the chamber dimensions, before and after the injector, the Reynolds number is 66, well below the critical number. It is expected that the minor disturbances in the flow by the injector will not result in transitioning from the laminar to turbulent regime. The centre flow containing the aerosols (aerosol flow) is layered in between a dry particle-free sheath nitrogen (purity 5.0, 99.999%, H₂O ≤ 3 ppm) flow, with a sheath-to-aerosol flow ratio of typically between ~~10:1 and~~ 12:1, which ensures the aerosol flow remains laminar and is exposed to the constant center temperature and RH conditions in the chamber. The sheath air is controlled by a mass flow controller (MFC; MKS, MF1, full scale flow of 5 stdL min⁻¹). The position of the injector thereby determines the residence time of the aerosols in the chamber. Within this residence time ~~the aerosols can nucleate into~~ ice crystals may nucleate on aerosol particles and grow to larger sizes, allowing for discrimination by size with the OPC. For the field measurements reported here, particles detected by the OPC in size bin > 5 μm in diameter are classified as ice (see sect. 2.2.3). Counts in smaller size channels could be contaminated by unactivated aerosol particles and, at water saturated conditions, the size bins up to 3-4 μm could be contaminated by droplets. For measurements reported here, HINC was kept at sufficiently low RH to ensure water droplets

did not contaminate the signal in the 5 μm channel. The OPC is calibrated for a total flow of 2.8 stdL min^{-1} which is set by an external pump. The MFC is used to set the 92% sheath air flow, so that the remaining 8% is made up by the aerosol flow sampled (pulled) into the chamber.

5 LabVIEW® is used to control the re-circulating cooler temperature and resulting RH in the chamber by regulating the cold and warm wall temperatures. Integrated into the LabVIEW® control panel are the flow rate of the sheath flow through the MFC and a motorised valve to direct the aerosol flow through a HEPA filter to quantify the noise for the signal to noise ratio (see sect. 2.3). Additionally, the counts in all size bins of the OPC are read out, and all set and output parameters are logged into a single file which is later used for data analysis.



10

Fig. 1: Schematic and outer dimensions of HINC (main chamber).

2.2 HINC validation and verification

Here we present laboratory measurements, using a variety of aerosol particles, to verify the accuracy in temperature T and RH for the newly built INP counter. We define the operational range, where it reliably measures INPs in the water saturated regime. 15 To confirm these operation settings in the field, additional tests with ambient particles at the JFJ were performed.

2.2.1 Sample preparation

To validate the temperature and RH conditions in HINC, hygroscopic growth upon deliquescence, cloud droplet activation and 20 homogeneous freezing experiments with size selected ammonium sulfate ($(\text{NH}_4)_2\text{SO}_4$), sodium chloride (NaCl) and sulfuric acid (H_2SO_4) particles were conducted, as well as experiments with polydisperse ambient particles in the field. Except for the latter, particles were generated as aqueous solutions (0.05625% w/w for $(\text{NH}_4)_2\text{SO}_4$, 5% w/w for NaCl , 60% w/w for H_2SO_4), atomized and dried by diffusion to $\text{RH}_w < 2\%$, before they were size selected by a differential mobility analyzer (DMA; TSI,

3081). For the deliquescence and cloud droplet activation experiments 200 nm particles, and for homogeneous freezing experiments 100 nm H₂SO₄ were used. Results from these experiments are presented for different size channels which show the growth of the particles at different RHs to various sizes. E.g. for the homogeneous freezing experiments at 233K, ice particles are observed in size channels > 0.5 μm at RH_w > 97.5%, while in the experiments at 242/243 K and at RH_w = 100 – 107% cloud droplets are measured in size channels < 5 μm, but not in the size channel > 5 μm which is thus used to detect ice crystals. The activated fraction (AF) which is the ratio of aerosol particles which activated into cloud droplets or nucleated ice crystals to the number of total particles, counted by a condensation particle counter (CPC; TSI 3772) in parallel, are reported.

10 2.2.2 Accuracy of temperature and RH in HINC

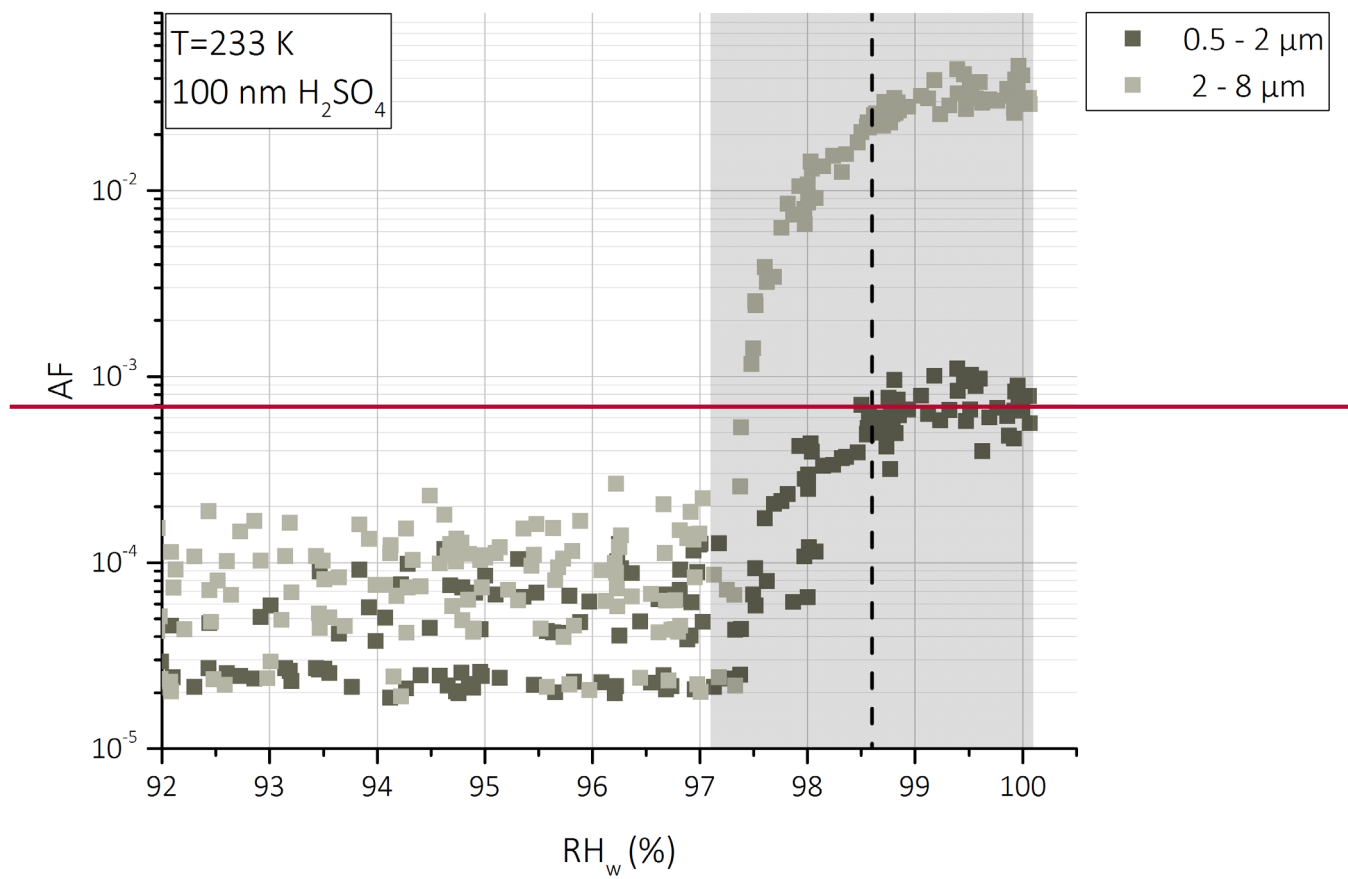
At temperatures T < 235 K, homogeneous freezing experiments were conducted to compare the onset of freezing observed in HINC to those reported and modeled in the literature (Koop et al., 2000a). In the respective experiments the RH_w conditions in HINC were increased at an approximate rate of 0.5% min⁻¹, so that an increase in RH_w of ~10% was achieved over a total time of 20 minutes, which implies that during the 8-second aerosol residence time in HINC, the particles experienced constant RH conditions. Experiments with 100 nm H₂SO₄ particles at 233 K (Fig. 2) revealed that the onset of freezing in HINC occurs within RH_w ± 1.5% compared to of the expected RH_w required to freeze of the freezing of solution droplets of the same initial dry size of the solute particles (Koop et al., 2000a), based on a fixed nucleation rate coefficient of 10¹⁰ cm⁻³s⁻¹ (Koop et al., 2000a), which is RH_w = 98.6% as shown by the dashed line in Fig. 2. The reported RH_w on the x-axis in the figure represents the nominal conditions at the centre line of the chamber, which is the centre of the aerosol layer. Due to the width of the aerosol layer, the particles are exposed to a variation in RH_w ± 1%, and to an uncertainty in RH_w ± 1% due to the temperature uncertainty. The grey shading in Fig. 2 represents this total calculated uncertainty of RH_w ± 2%, for a prescribed RH_w = 98.6%. When the chamber is set to RH_w = 98.6%, the aerosols can be exposed to the RH_w range of 96.6 – 100.6%. Our experiments reveal an increase in the AF of particles between 2 – 8 μm starting at 97.5%, and reaching a plateau value at 99.5%, which is in agreement to the expected range of freezing within the aerosol layer. According to theoretical calculations (Koop et al., 2000a) at 233 K and RH_w = 98.6% (99%) and using anucleation rate of 10¹⁰ cm⁻³s⁻¹, we expect for initial 100 nm H₂SO₄ particles at a residence time of 8 seconds an AF of 0.001 (0.005), which an order of magnitude below the AFs observed in Fig. 2. However, this AF is within the uncertainty range given that in this region where the AF is a steep function of RH, an uncertainty of RH_w of 1% can result in a change in AF by a factor of 10-100.

Above homogeneous freezing temperatures (T > 235 K), we expect cloud droplet activation for 200 nm (NH₄)₂SO₄ and H₂SO₄ particles at RH_w = 100%. We thereby report the onset of cloud droplet formation, which is the first observed increase in the AF at a given size in the OPC channels. This is observed in the smaller OPC channels (0.5 – 2 μm) as an increase in the AF when the conditions in the chamber approach RH_w = 99 - 100% as observed for H₂SO₄ (Fig. 3) and for ambient particles (Fig.

4). The grey shaded area in Fig. 3, 4 and 5 also includes the calculated variation and uncertainty in RH_w for the given temperatures of 242 and 243 K, respectively. As discussed above, the exposure of the aerosol particles to this variable RH leads to droplet activation not only at $RH = 100\%$ but in a range of RH pertaining to the variation and uncertainty in HINC. As such aerosol particles grow within the residence time of 8 seconds to different sizes, which is depicted by the difference in the OPC size channels. In addition, this is also the reason why at $RH_w = 101\%$ for example (Fig. 3), the AF is still observed to be increasing because some particles in the aerosol layer are still exposed to $RH_w = 99 - 100\%$, explaining why droplet activation is observed up to an RH of 102%. We note that an increase in the AF of initial 200 nm H_2SO_4 is observed prior to $RH_w = 100\%$ in the 0.5 and 1 μm channels (Fig. 3) which is to be expected due to hygroscopic growth of the H_2SO_4 particles. Therefore, the increase in size for ~~at~~ $RH_w < 100\%$ is only observed in the smaller size channels occurring prior to droplet activation at $RH_w < 100\%$, while an increase at $RH_w = 100\%$ in the $> 2 \mu m$ channel is observed due to cloud droplet activation. On the other hand, the ambient particles show droplet activation in the $> 0.5 \mu m$, $> 1 \mu m$ and $> 2 \mu m$ channels at $RH_w = 101.5\%$ (Fig. 4). This is likely due to the lower hygroscopicity of the ambient particles compared to H_2SO_4 and due to a larger fraction of the sampled ambient particles being $\ll 100 \text{ nm}$, requiring higher RH for the droplets to activate and grow to detectable cloud droplet sizes at this temperature. ~~but~~ In addition, the experiments could also be ~~compounded~~ influenced by RH uncertainties (see sect. 2.3).

At lower RH_w , hygroscopic growth due to deliquescence was also observed as an increase in particle concentrations in the smallest OPC channels of 0.3 and 1 μm , which occurred for 200 nm NaCl in the range of $RH_w = 79.5 - 81\%$ (Fig. 5). The observed increase in the particle fraction ~~AF~~ due to deliquescence and hygroscopic growth compares well to literature results reported to be $RH_w = 77 \pm 2.5\%$ (Koop et al., 2000b). We observe a first strong increase in the particle fraction > 0.3 and $> 1 \mu m$ at 80 - 81%, followed by a gradual increase in the particle fraction to unity (within uncertainties) at $RH_w \approx 94\%$ for the $> 0.3 \mu m$ trace. Deliquescence is a phase change and not a growth process, and a delay as compared to the literature value (dashed line Fig. 5) is expected, since the deliquesced particles need to grow to a size $> 0.3 \mu m$ to be detected in the OPC. In theory we would expect all particles to grow to sizes larger than 0.3 μm at $RH_w \geq 80\%$ since the deliquescence and growth threshold has been reached. However, we note that due to an uncertainty in sizing of up to 3.5% in the DMA, particles between 193 and 207 nm for a nominal size of 200 nm will be sampled. In reality, we expect an even broader size distribution because dried NaCl particles are aspherical and result in larger sizing errors (Ardon-Dryer et al., 2015). Due to the size selection method with the DMA, a non-negligible fraction of larger particles (10%) between 320 - 440 nm (from double and triple charged particles) will also be sampled by HINC. This breadth in size distribution may explain the initial increase in particle fraction at $RH_w = 80\%$ arising from the multiply charged particles followed by a progressive increase in the particle fraction up to $RH_w = 94\%$ where all the particles grow to sizes $> 0.3 \mu m$. The same can be said for the $> 1 \mu m$ trace. Note that complete activation in this trace occurs at $RH_w > 100\%$, which is expected from the variation in RH in the aerosol layer. Finally, we note that the goal of this experiment is to demonstrate that HINC can achieve prescribed RH conditions with reasonable accuracy by controlling the wall temperature as is seen by the onset in growth at $RH_w = 80\%$ in Fig. 5. Additionally, we acknowledge that

the fraction of particles $> 0.3 \mu\text{m}$ reaches a maximum at higher RH_w than theoretically expected, which can also be attributed to the sizing and counting uncertainty of the OPC, which is most pronounced at these small particle sizes, when the wavelength of the laser (780 nm) is similar to the diameter of detectable particles. Due to the generation method of NaCl at $T > 273 \text{ K}$ and immediate exposure of the NaCl to the respective temperature and humidity, when the aerosol particles are injected in HINC, we believe that we deliquesce NaCl anhydrate, and not NaCl dihydrate (Bode et al., 2015). For 200 nm $(\text{NH}_4)_2\text{SO}_4$ particles, deliquescence and hygroscopic growth was also observed at $\text{RH}_w = 82 - 85\%$, consistent with the $(\text{NH}_4)_2\text{SO}_4$ deliquescence $\text{RH}_w = 82 - 84\%$ (Cziczo and Abbatt, 1999). All validation experiments to verify the RH and temperature accuracy in HINC are summarized in Fig. 6.



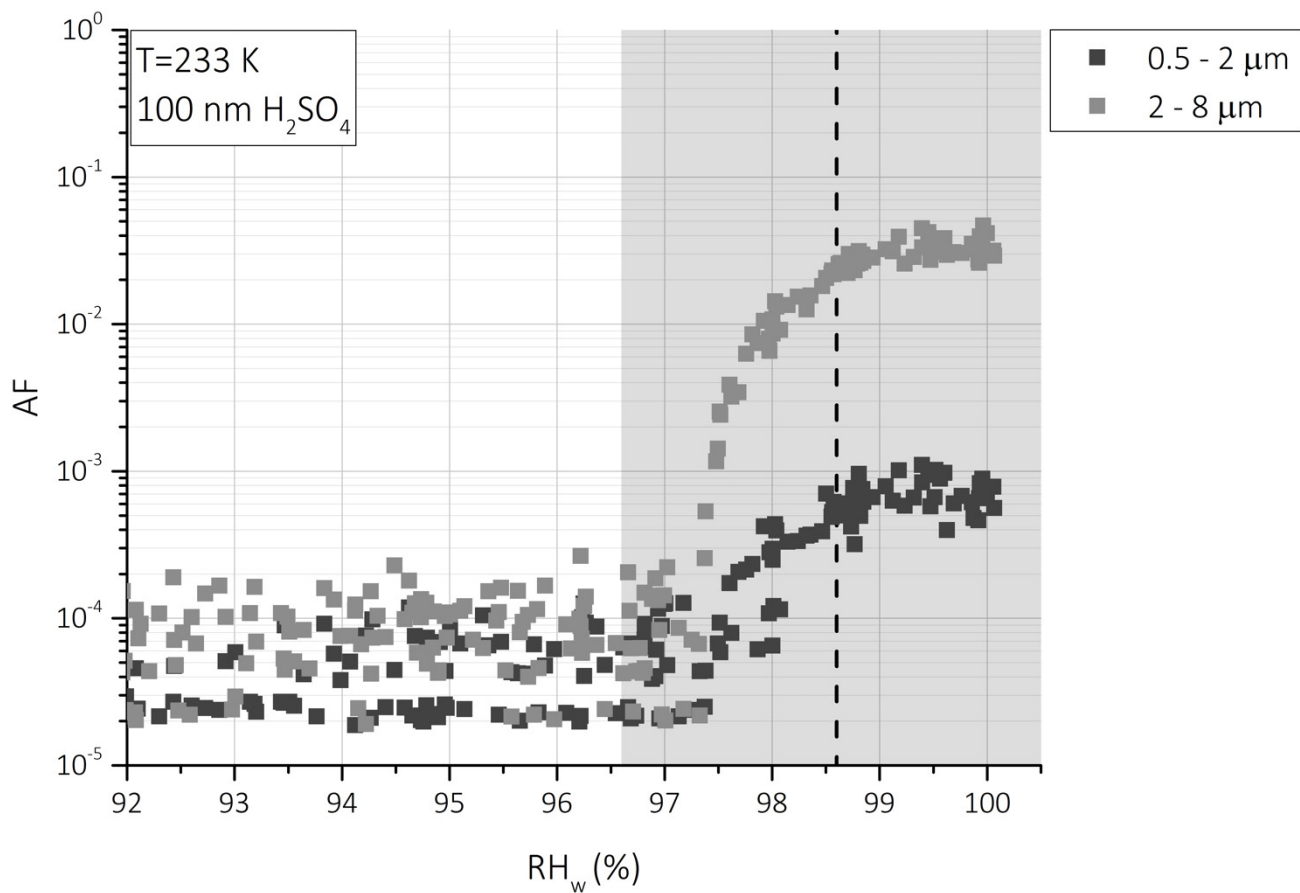
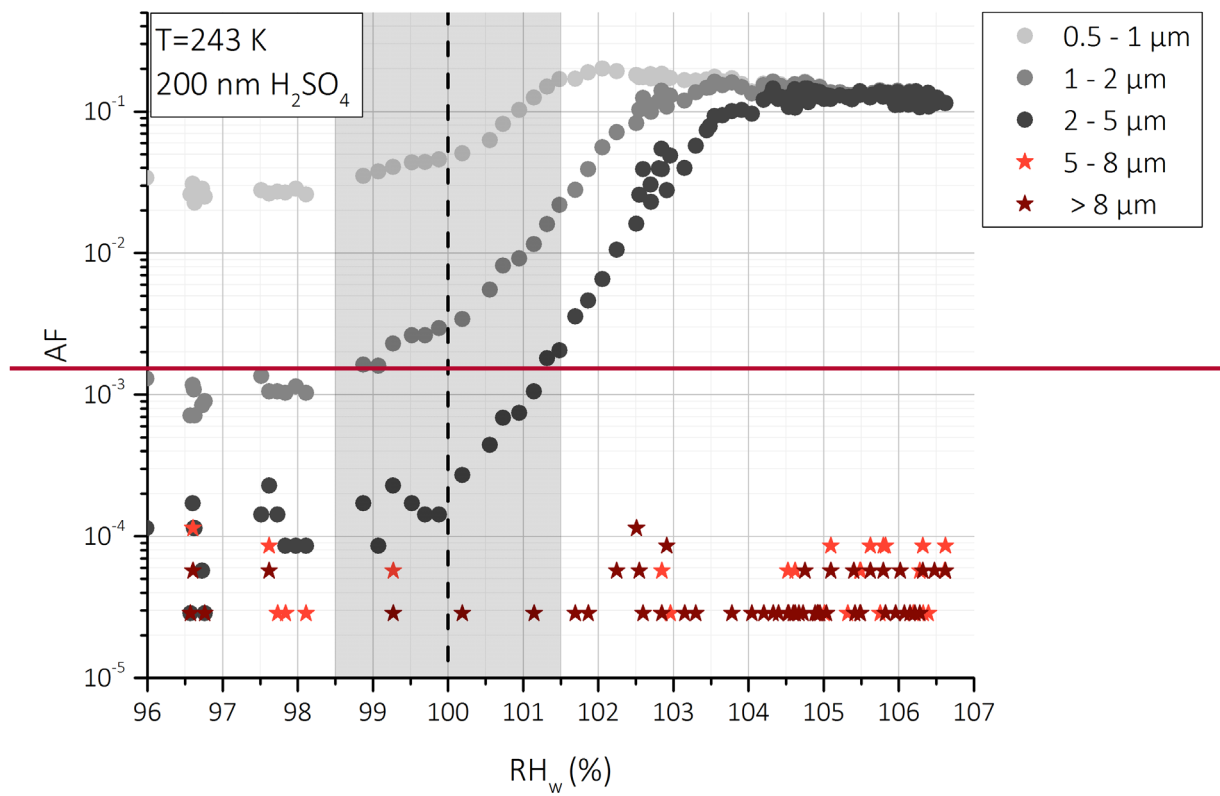


Fig. 2: Homogeneous freezing curve of 100 nm dry diameter H_2SO_4 particles, shown as the AF as a function of RH_w at 233 K in the size channel 0.5 – 2 μm (dark grey) and 2 – 8 μm (light grey). The dashed line ($RH_w = 98.6\%$) represents the expected RH_w for homogeneous freezing of dilute solution drops of an initial dry diameter of 100 nm (Koop et al., 2000a), and the shaded region indicates the calculated range of RH_w and uncertainty to which the particles in the aerosol layer in HINC are exposed to. The shaded region indicates the range of uncertainty in RH_w in HINC (see sect. 2.3)



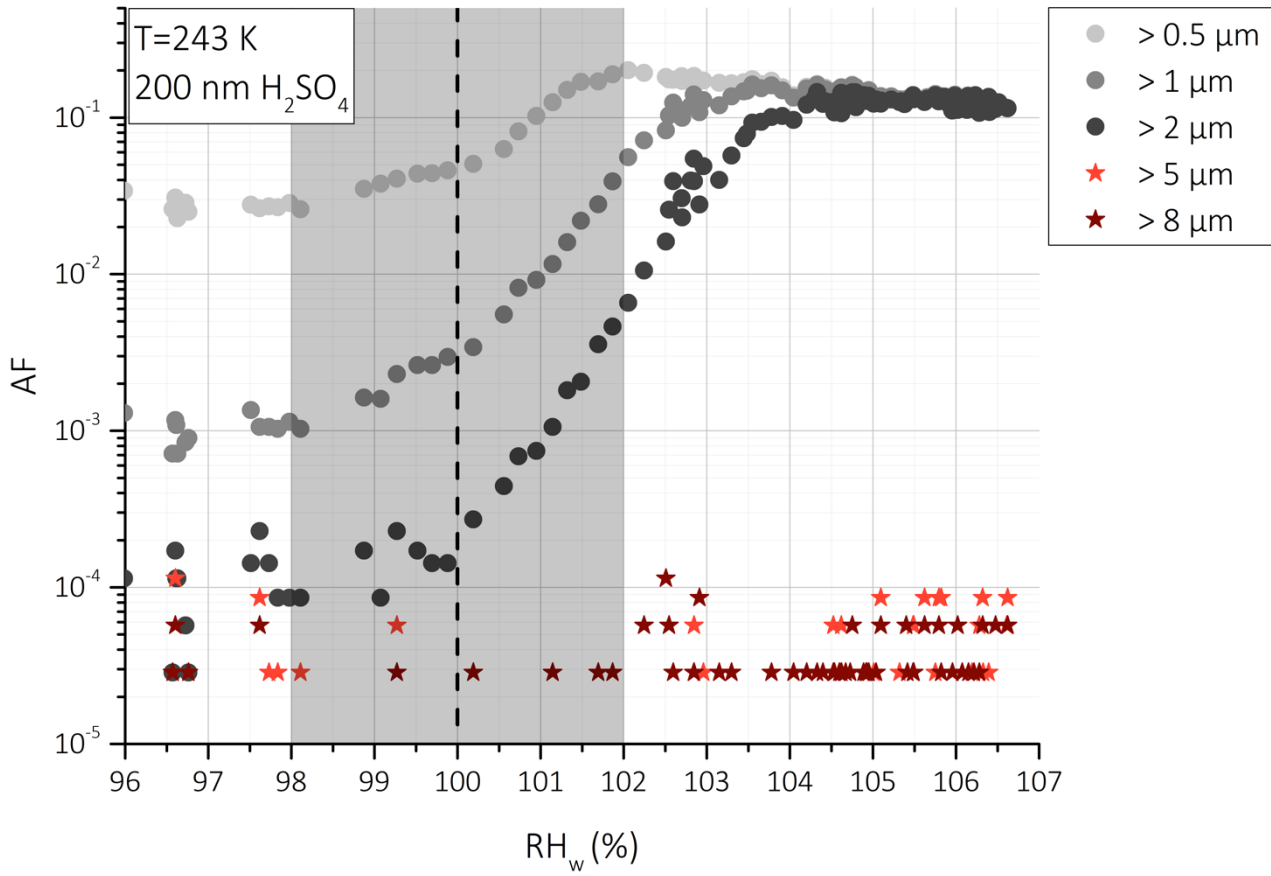
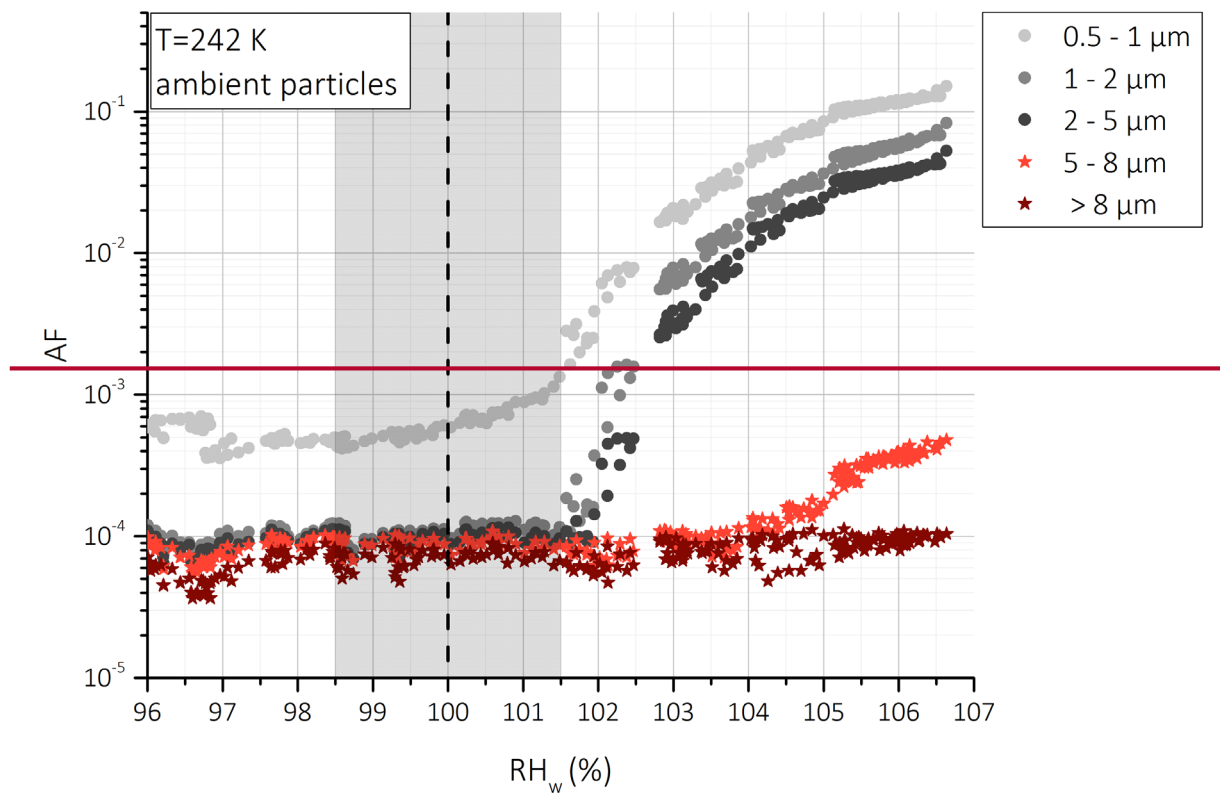


Fig. 3: Water droplet (or ice crystal for $> 5 \mu m$) activation fraction and subsequent growth as function of RH_w at 243 K for 200 nm dry diameter H_2SO_4 particles, shown for all size channels $> 0.5 \mu m$ in the OPC. Vertical dashed line represents expected onset for cloud droplet formation; grey area refers to the calculated variation and uncertainty of RH in the aerosol layer.



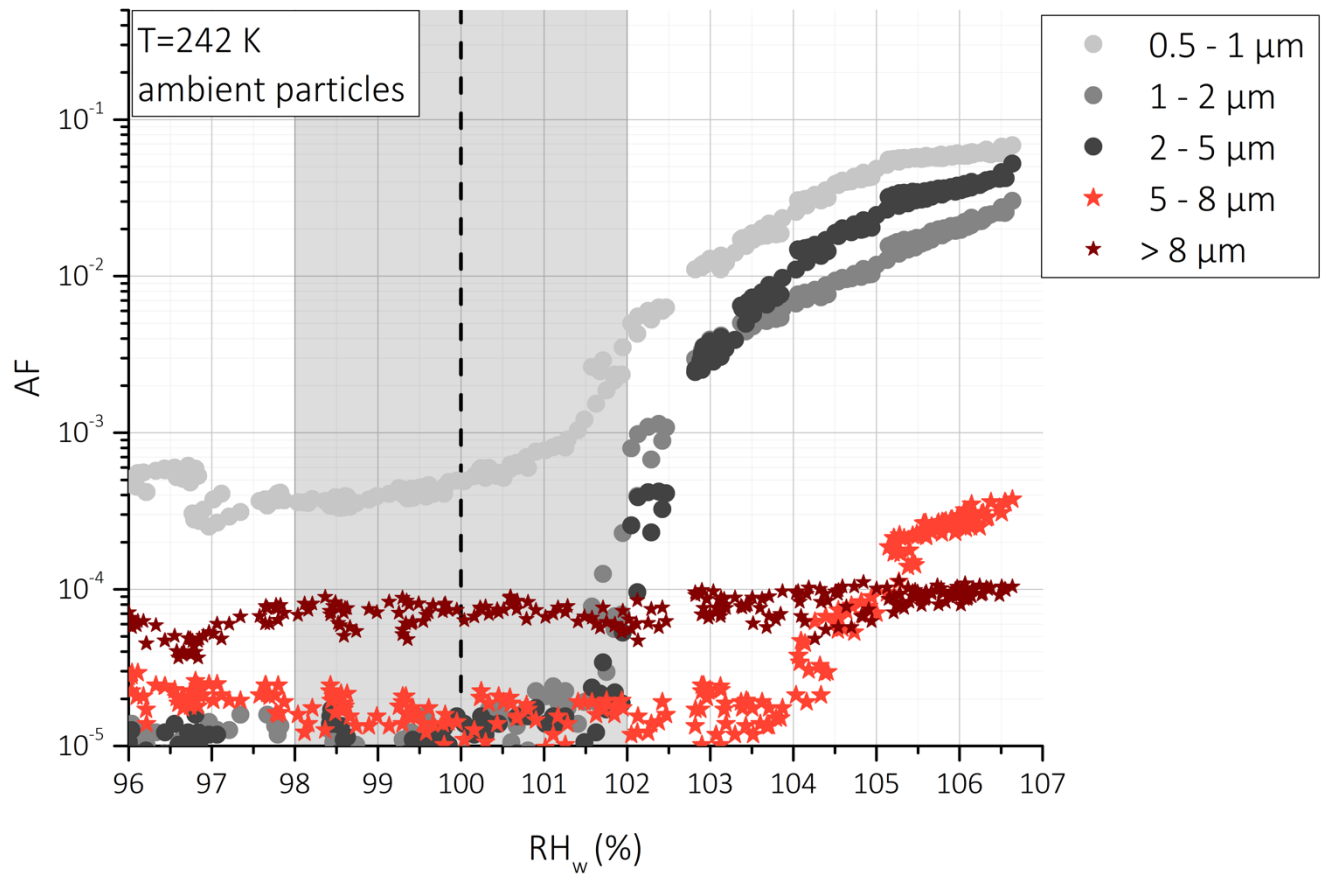
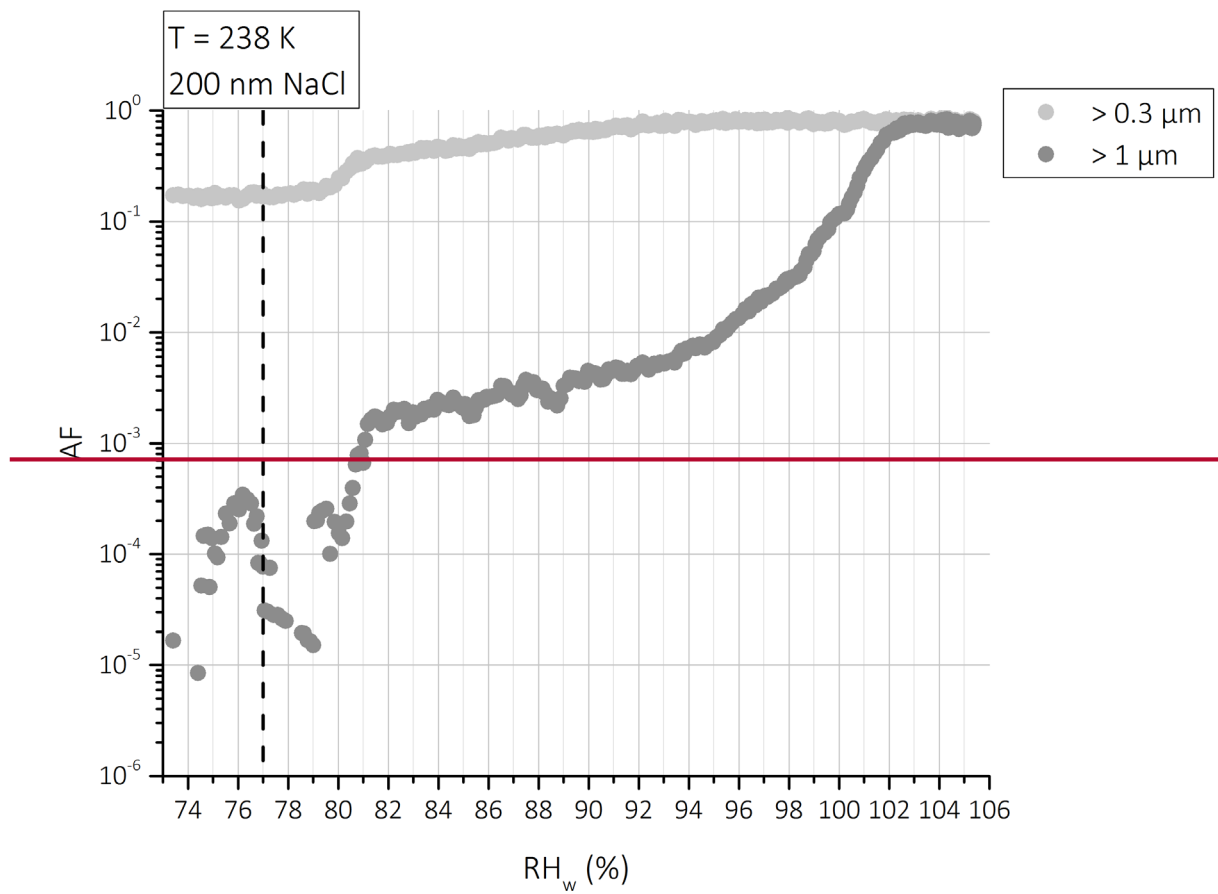


Fig. 4: Water droplet activation fraction and subsequent growth as function of RH_w at 242 K for ambient polydisperse particles sampled at the JFJ, shown for all size channels in the OPC. Vertical dashed line represents expected onset for cloud droplet formation; grey area refers to the calculated variation and uncertainty of RH in the aerosol layer.



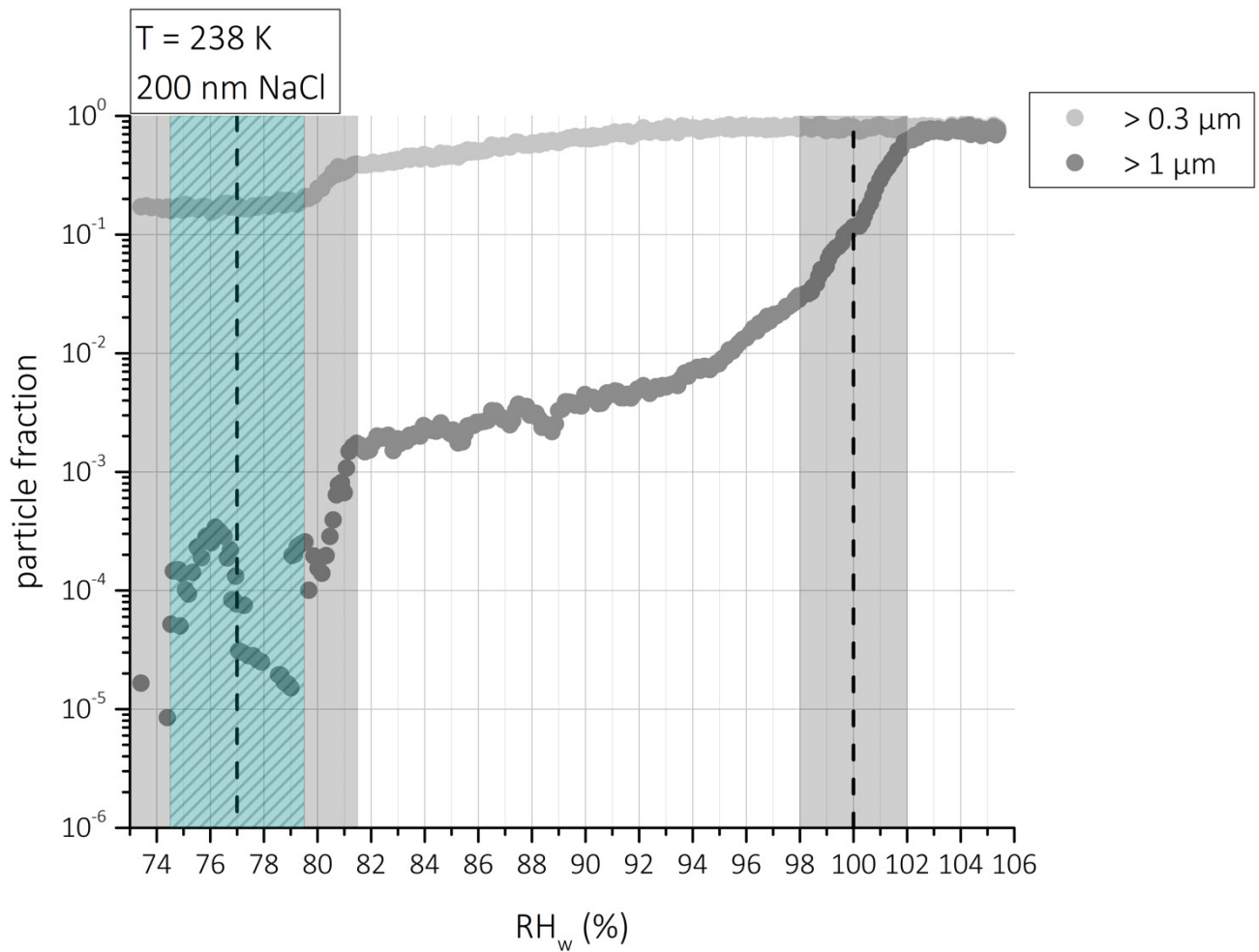


Fig. 5: ~~AF-Particle fraction showing due to deliquescence and subsequent~~ hygroscopic growth after deliquescence as function of RH_w at 238 K for 200 nm NaCl particles. Vertical dashed lines represent deliquescence RH_w of NaCl at $RH_w = 77\%$ and onset of cloud droplet formation at $RH_w = 100\%$; green shaded area refers to observed uncertainty in deliquescence by Koop et al. (2000b); grey area refers to the calculated variation and uncertainty of RH in HINC in the aerosol layer.

2.2.3 Upper RH limit for ice crystal detection: water drop survival (WDS)

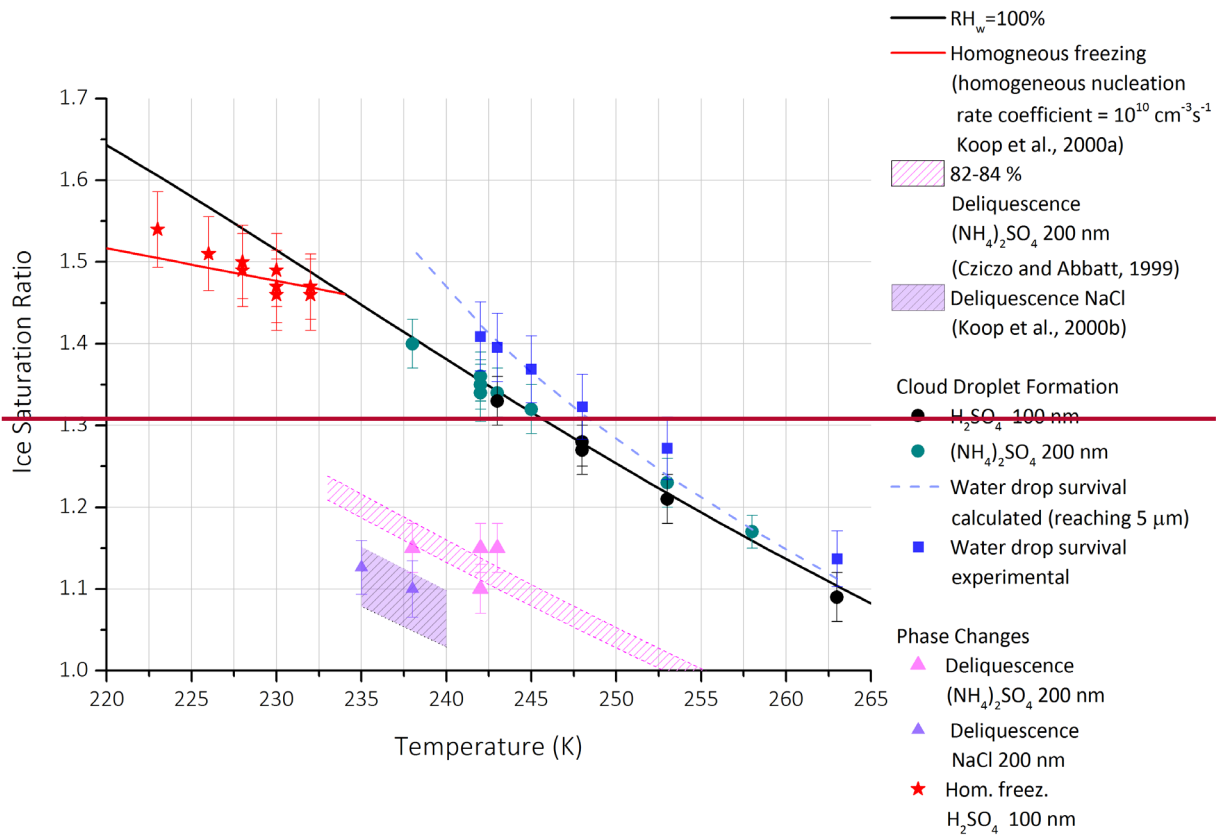
The upper RH achievable to reliably detect ice crystals is limited by the possible activation and subsequent diffusional growth of water drops, since only the optical size is used to discriminate between ice crystals (larger) and water droplets (smaller) at the same temperature and RH conditions. To identify the maximum operation RH, experiments are conducted with 200 nm $(NH_4)_2SO_4$ and H_2SO_4 particles at $T > 235$ K where homogeneous freezing is ruled out insignificant. For these experiments the

RH is increased until activated water droplets grow to a diameter which is detected in the OPC size channel used to detect ice crystals. This is referred to as the WDS point. Based on diffusional growth calculations (Rogers and Yau, 1989) activated cloud droplets of an initial diameter of 200 nm can grow to a size of 4 μm in HINC at 242 K, $\text{RH}_w = 104\%$ for a residence time of 8 seconds (conditions used for field experiments reported here), giving us confidence that droplets are not detected in the 5 μm channel. Only at an RH_w of 107% cloud droplets grow to $> 5 \mu\text{m}$, and therefore by conducting our experiments at $\text{RH}_w = 104\%$, we only detect ice crystals in the 5 μm OPC channel. As a confirmation, no counts in the size channel $> 5 \mu\text{m}$ were observed for H_2SO_4 particles (Fig. 3) even up to an RH_w of 107%. With sampling ambient particles an increase in AF for particles $> 5 \mu\text{m}$ at $\text{RH}_w = 104 - 105\%$ is observed (see Fig. 4), which can be caused by ice crystals forming heterogeneously, since water droplets cannot grow to this size at the respective conditions in HINC. These calculations also reveal that the diffusional growth of the activated cloud droplets, and hence the final size of the cloud droplets of interest, which is 5 μm for the discussed field experiments, are insensitive to the initial dry diameter of the aerosol particles, since the final droplet size at 242K, $\text{RH}_w = 104\%$ and a residence time of 8 seconds of an initial 50 nm, 200 nm and 800 nm is 4.035, 4.038 and 4.08 μm , respectively. Also effects of the particle chemistry are assumed to be negligible for the droplet activation, since we conduct our experiments at $\text{RH}_w \geq 104\%$ where a variety of aerosol chemical compositions should activate into droplets. Thus water droplets contaminating the 5 μm channel should not occur even with varying hygroscopicities and sizes of aerosol populations. ~~An example of an increase in RH_w to $> 106\%$ at 243 K is shown in Fig. 3, where WDS is not observed in the OPC channel $> 5 \mu\text{m}$. This is likely due to settling of the larger liquid droplets out of the aerosol flow, which grow to sizes too large to be sampled by the OPC due to the hygroscopic nature of H_2SO_4 .~~ Note that the AF even at low RH is non-zero, which is caused by either unactivated sample particles (particularly in the smallest size channels) or by internal background counts (see section 2.3). ~~The same experiment was performed in the field with polydisperse ambient particles at 242 K (Fig. 4). This test reveals an increase in AF of a factor of 4 in the OPC size channel 5 – 8 μm at $\text{RH}_w = 105\%$, presumably caused by growing water droplets. However, in this test ice crystals are likely contaminating the signal and cannot be completely ruled out since the exact composition of the ambient particles is not known. This tests together with the H_2SO_4 tests, shown in Fig. 3, These experiments~~ give us confidence that by operating at a temperature $T = 242 \text{ K}$ and a $\text{RH}_w \leq 104\%$, the OPC size channel $> 5 \mu\text{m}$ is suited to reliably detect ice crystals for field measurements. Diffusional growth of ice crystals for the respective conditions in HINC reveal that ice crystals can be lost to settling within 7 seconds. Still, residence time experiments with 400 nm microcline particles at 242 K and $\text{RH}_w = 104\%$ show that the AF is at a maximum value at 8 seconds (Fig. A1, appendix), which informed the 8 second residence time in the field experiments. This discrepancy to the theoretical calculation of 7 seconds is expected due to assumptions in the diffusional growth calculations, such as immediate activation upon entering the chamber and assuming spherical ice crystals. The residence time of 8 seconds therefore should include consideration for the equilibration time of the particles to the center supersaturation (~ 0.2 seconds), the growth time of ice crystals to $> 5 \mu\text{m}$ (2 seconds) and time dependence for nucleation (up to ~ 6 seconds). Thus the residence time of 8 seconds should also minimize the number of ice crystals $< 5 \mu\text{m}$, since ice crystals only need 2 seconds to grow by diffusion to sizes $> 5 \mu\text{m}$ at this high $\text{RH}_i = 140\%$ ($\text{RH}_w = 104\%$). We believe that undercounting INPs due to ice crystals $< 5 \mu\text{m}$ should not significantly influence the

INP concentrations reported especially given the day to day variability in INP concentrations found at the field site studied in this work. ~~The field measurements above water saturation presented in this work are therefore conducted at RH_w 103–104 %.~~

5 2.2.4 Summary of validation and verification experiments

In Fig. 6, we show the results from all the validation experiments performed. The observed phase changes are in the expected range of homogeneous freezing of solution droplets (Koop et al., 2000a), cloud droplet formation (Lohmann et al., 2016), and deliquescence of NaCl (Koop et al., 2000b) and $(NH_4)_2SO_4$ (Cziczo and Abbatt, 1999). The ice onset for homogeneous freezing of H_2SO_4 was observed within a range of $RH_w \pm 1.5\%$ of the value reported in Koop et al. (2000a). At colder temperatures, the onset of homogeneous freezing is observed to shift to higher RH_w but still remains well within the range of uncertainty in RH (see sect. 2.3) of HINC. Above 235 K, where freezing of dilute water droplets is not expected, droplet formation for both H_2SO_4 and $(NH_4)_2SO_4$ as well as ambient particles was observed at $RH_w = 100 \pm 1.5\%$ where particles initially not detectable in the OPC size channel $> 1 \mu m$ activate to droplets and grow large enough to be detected. At lower RH_w we observe hygroscopic growth upon deliquescence of NaCl and $(NH_4)_2SO_4$ particles as an increase in the OPC size channel $> 0.3 \mu m$. For NaCl such an increase was observed at $RH_w = 79 - 81\%$, as compared to a $RH_w = 77 \pm 2.5\%$ (Koop et al., 2000b). For $(NH_4)_2SO_4$ particles a growth was observed at $RH_w = 81 - 85\%$ as compared to $RH_w = 82 - 84\%$ (Cziczo and Abbatt, 1999). Both the phase change and cloud droplet formation experiments with H_2SO_4 , $(NH_4)_2SO_4$, NaCl and ambient particles verify that HINC operates reliably at the discussed settings of temperature and RH. As can be seen from Fig. 6, some deviation (all within instrument uncertainty) of the data points does occur. This is due to the fact that in the region of ice nucleation or droplet activation the AF (nucleation rate) is a very steep function of RH and a small change in water activity by for example 0.01 ($\sim RH$ of 1% at equilibrium) can change the nucleation rate by a factor of 10 or 15.



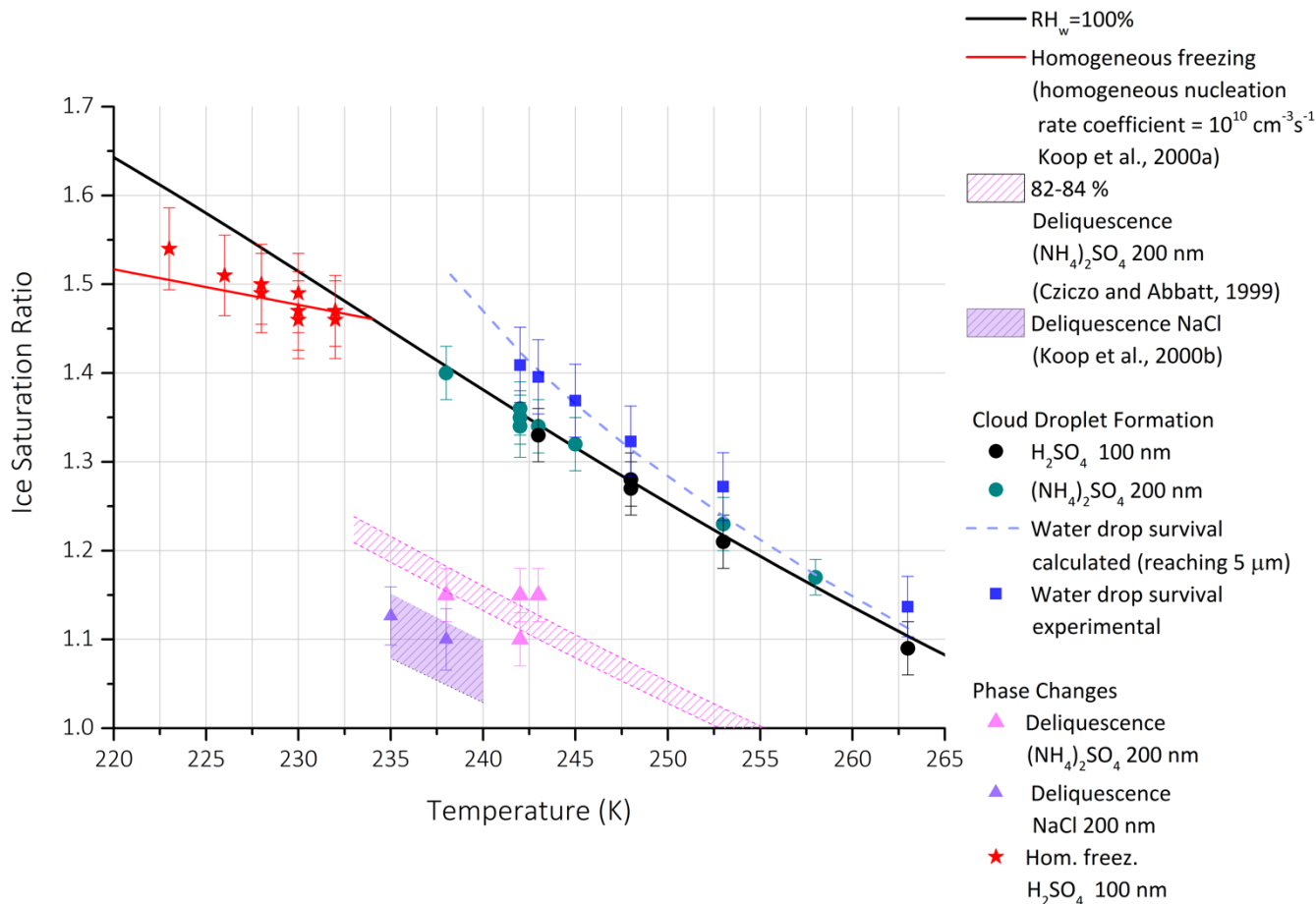


Fig. 6: Summary of characterization experiments and comparison to literature and theoretical values of phase changes and growth processes. Data from experiments reflect the first appearance of ice (ice onset) when the AF increases above the chamber background levels.

5

2.3 Uncertainties and limit of detection (LOD)

Uncertainties in temperature and RH to which the aerosol particles are exposed to can arise from the set wall temperatures of HINC that are measured by two thermocouples on each wall in the activation/growth section of the chamber. The thermocouples have an uncertainty of ± 0.1 K. This translates into an uncertainty in RH at the center location of $RH_w \pm 1\%$ ($RH_i = \pm 1.2\%$) at a center temperature $T = 242$ K and at $RH_w = 104\%$. In addition to this, the aerosol lamina layer is only 1/13th of the total flow (for a typical sheath-to-aerosol ratio of 12:1), and since a linear temperature gradient establishes between the warm and the cold wall, there is a temperature variation of ± 0.4 K across the aerosol lamina layer for the temperature conditions (242 K) used in the field measurements presented here. The variation in temperature ~~relates to~~ causes a variation in

10

RH_w of $\pm 1.21\%$ ($RH_i \pm 2.7\%$). This translates into a calculated total uncertainty of $RH_w \pm 2\%$ ($RH_i \pm 3\%$) at 242 K and $RH_w = 104\%$. The validation experiments of homogeneous freezing and cloud droplet activation reveal an uncertainty of $RH_w \pm 1.5\%$, which is slightly lower than the calculated uncertainty and shows that HINC is able to precisely establish prescribed supersaturations. Thus we take the uncertainties to be ± 0.4 K and $RH_w \pm 1.5\%$ ($RH_i \pm 3\%$), respectively.

5 ~~Considering the validation experiments of homogeneous freezing and cloud droplet activation also reveal an absolute upper limit uncertainty, we take the temperature and RH_w uncertainty to be ± 0.4 K and $\pm 1.5\%$ ($RH_i \pm 3\%$) respectively which is slightly higher than the calculated uncertainties.~~

The OPC, which is used to classify and count detected hydrometeors downstream of HINC, has a relative counting accuracy of $\pm 10\%$, and a relative uncertainty in the sizing channels of $\pm 10\%$. The CPC, which was used to measure the total particle concentration in parallel to the INP measurements, has a relative counting accuracy of $\pm 10\%$, resulting in a relative uncertainty in the AF of 14%. The DMA, which was used to size select the $(NH_4)_2SO_4$ and H_2SO_4 particles, has a relative sizing uncertainty of 3 – 3.5%.

15 During an ice nucleation experiment, erroneous counts in the OPC ice channel can arise from electrical noise in the OPC, or from internal ice sources such as frost falling off the warmer chamber wall giving rise to particle counts that are falsely classified as ice. In order to assess and to correct for the contribution of such false counts, filter measurements are conducted regularly before and after each sampling period to determine a background count in the following way: The instrument is set to its target temperature and RH_i , and the aerosol flow is sampled through a particle filter placed upstream of the aerosol injector for 10 minutes before and after each aerosol-sampling period of 20 minutes. It is observed that the background counts do not change significantly over this time and follow a Poisson distribution. The mean background (μ) and the standard deviation (σ) are derived from the two-10 minute filter periods. The mean ambient INP counts (\overline{INP}) are calculated from the ~~mean~~-counts (ω) during a 20-minute sampling by taking the background into consideration, resulting in $\overline{INP} = \omega - \mu$. The values of μ and σ are used to assess if the signal, i.e. \overline{INP} , is significantly different from the noise. Therefore, the instrument's LOD for ambient INP following Poisson statistics is calculated as $LOD = \sigma$.

Comparing \overline{INP} to the LOD can result in one of three scenarios:

- 25
- 1) $\overline{INP} > \sigma$
 - 2) $\overline{INP} < 0$
 - 3) $0 < \overline{INP} < \sigma$

Scenario 1 is considered as a quantifiable INP measurement with significance because $\omega > \mu + \sigma$ and therefore above the LOD. In scenario 2, INP counts are considered to be non-quantifiable for the volume of air sampled during the 20 minute period. In 30 scenario 3, INP counts are below the LOD but above μ therefore quantifiable but not with significant confidence. INP concentrations below the instrument LOD (scenario 2 and 3) are included in calculations of field campaign averages. We

believe that this is crucial since ambient INP concentrations are typically quite low in the FT, and scenarios 2 and 3 occurring frequently actually speaks to this low observed INP concentration. Complete ignorance of values from scenarios 2 and 3 would lead to an artificially positive bias in reporting INP concentrations (for detailed discussion see Boose et al., 2016a). The concentrations below the LOD are taken into account as their measured value (scenario 3). In the case of scenario 2, instead of using a value of zero for calculating campaign averages, a minimum quantifiable concentration for a 20 minute period is used. This is determined by taking the minimum count possible in the OPC ice channel (1 count) and normalizing to the volume of ambient air during a 20-minute sampling period. By doing so we acknowledge that the true concentration could be below this minimum value (shown in Table 2). In addition, by accounting for the minimum quantifiable concentration in the manner described above, we take into consideration the increase in sampled volume of ambient aerosol flow due to use of an aerosol concentrator, applied in winters 2013 and 2014 (Boose et al., 2016a), which lowers the LOD by a certain concentration factor. For transparency we show average INP concentrations including and excluding the values below the LOD. Finally, ambient INP counts and LODs are converted to concentrations in stdL^{-1} .

3 Field measurements

To further validate the chamber performance for field measurements, INP measurements were conducted at the JFJ during the winters of 2015 and 2016 with HINC, and are compared to earlier measurements conducted at the same location and sampling conditions with PINC (Boose et al., 2016a) during winters 2012, 2013 and 2014. The ice nucleation measurements were performed in the deposition nucleation mode, and since winter 2014 also in the condensation freezing mode. Detailed measurement dates and sampling periods for the campaigns are given in Table 1.

Table 1: Field measurement period and respective total sampling time of INP measurements and cloud water samples. Measurements were performed with PINC (winter 2012 – 2014) and HINC (winter 2015 and 2016).

measurements	start	end	breaks	total sampling time (h)			cloud water
				PINC/HINC			
				T (K)	RH _w 93/94%	RH _w 103% 104%	
winter 2012*	12.01.	27.01.	-	241	62.3	-	
winter 2013*	24.01.	27.02.	-	241	138.9	-	
winter 2014*	24.01.	16.02.	-	241	54.6	28.5	67.8

winter 2015	24.01.	09.02.	-	242	16	26	146
winter 2016	13.01.	06.03.	01.02.- 26.02.	242	17.1	99	42.5

* Boose et al. (2016a) measured with PINC

HINC was setup in the field as shown in the schematic in Fig. 7. ~~HINC~~ Particles were sampled from a total aerosol inlet, which is described in detail by Weingartner et al. (1999). Ambient interstitial and cloud-phase particles with diameters $< 40 \mu\text{m}$ at wind velocities $< 20 \text{ m s}^{-1}$ were sampled through an inlet heated to 293 K to evaporate cloud droplets and ice crystals. To exclude an additional humidity source from ambient air, the aerosol flow was passed through a diffusion dryer ($\text{RH}_w < 2\%$) and was then split into HINC ($0.22 \text{ stdL min}^{-1}$ aerosol flow, $2.83 \text{ stdL min}^{-1}$ total flow) and a CPC (TSI 3772, 1 stdL min^{-1}), counting the total particle concentration in parallel.

The measurement conditions were set such that the aerosol flow experienced a constant temperature $T = 242 \text{ K}$ and $\text{RH}_w = 94 \pm 1\%$ ($\text{RH}_i = 127 \pm 2\%$), relevant for heterogeneous nucleation of ice clouds, and at $T = 242 \text{ K}$ and $\text{RH}_w = 104 \pm 1.5\%$ ($\text{RH}_i = 140 \pm 3\%$) in winters 2015 and 2016, relevant for the mixed-phase cloud regime. The injector position was set to an optimal residence time (8 sec) for the aerosol particles, which takes into account prevention of ice crystal losses due to gravitational settling in the chamber but yet allowing for enough growth time to reach an optical diameter of $\geq 5 \mu\text{m}$. Experiments with two OPCs, one in parallel and one downstream of the ice chamber, were performed in order to obtain the difference in concentrations due to aerosol particle losses. For this test, HINC was set to its field campaign temperature $T = 242 \text{ K}$ and well below water saturation to prevent any activation of particles as droplets or ice crystals. The experiments revealed a particle loss of 26% for $1 \mu\text{m}$ particles, and 44% for $2 \mu\text{m}$ particles, and 100% for particles $> 5 \mu\text{m}$, therefore the OPC channel used to detect ice should not be contaminated with large ($> 5 \mu\text{m}$) unactivated ambient particles. As such the INP measurements reported here are representative for ambient particles below $2 \mu\text{m}$. This size range is characteristic for ambient aerosols at the JFJ, since particle concentrations $> 1 \mu\text{m}$ are naturally very low (e.g. Nyeki et al., 1998, Baltensperger et al., 1998). In addition, calculations with the Particle Loss Calculator (von der Weiden et al., 2009) revealed that 0.8% of $1 \mu\text{m}$ particles, and 2.6% of $2 \mu\text{m}$ particles, and 14 – 50% of $5 - 10 \mu\text{m}$ particles should be lost in the inlet and tubing upstream of HINC which we consider to be negligible in light of the low abundance of particles $> 5 \mu\text{m}$ (on the order of 0.05 stdL^{-1}) ~~uncertainty in particle counting from the OPC and CPC, and losses through the chamber injector.~~

For the field measurements, after the icing procedure (see sect. 2.1) the chamber walls were set to their target temperature, such that a center aerosol temperature $T = 242 \text{ K}$ and $\text{RH}_w = 94\%$ was achieved. While the chamber cooled down to these conditions, it was flooded with filtered dry air to prevent moist room air from contaminating the iced chamber walls. When the target temperature and RH was reached, a 10-minute filter measurement, to quantify the background counts, was conducted,

followed by a 20-minute aerosol measurement, and another 10-minute filter measurement. Following this procedure, the RH was further increased to 104%, and the background-sample-background measurement cycle was repeated.

~~Due to the non-automated operation of~~To avoid depletion of the ice layers in HINC, sampling time was limited to a maximum of 14 hours, of which approximately 50% were performed during nighttime, namely from 7 pm to 7 am.

5

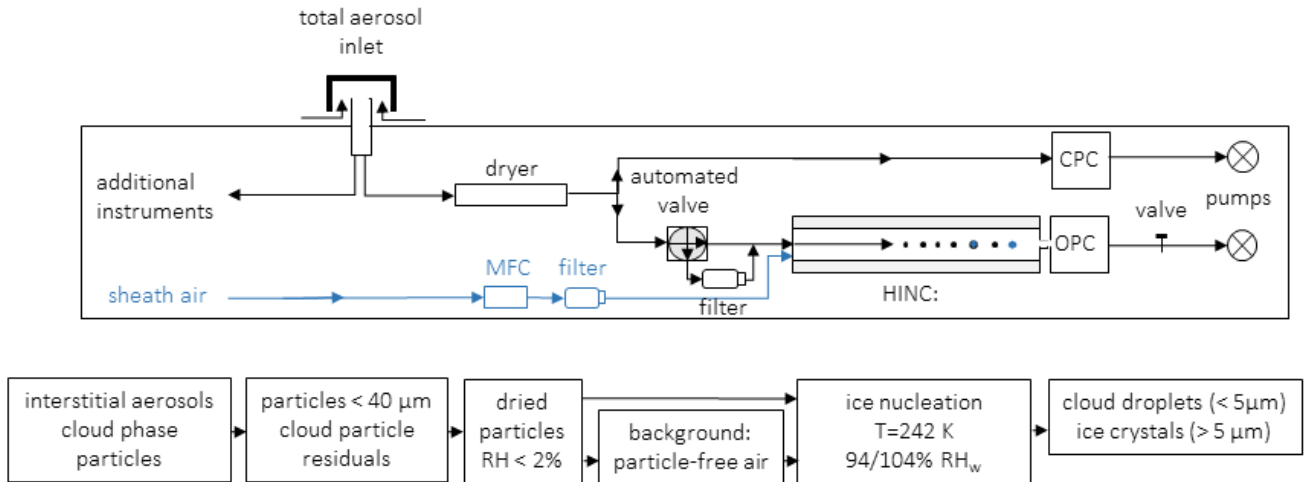


Fig. 7: Schematic of aerosol flow and instrument setup during field measurements at the JFJ research station.

10 3.1 Location

Measurements were performed at the JFJ, located in the Bernese Alps (3580 m a.s.l.; 46°33'N, 7°59'E). The research facility is a Global Atmospheric Watch (GAW) monitoring station and part of the ACTRIS2 Infrastructure (European Research Infrastructure for the observation of Aerosol, Clouds, and Trace gases), the Swiss National Air Pollution Monitoring Network (NABEL) and the SwissMetNet meteorological network. The station is located on an exposed mountain col, only surrounded by firm ice and rocks, without noticeable influence from local vegetation. Due to its elevation, the site is mostly located in the FT and represents background aerosol concentrations (Baltensperger et al., 1997). It can be influenced by local emissions due to daytime tourist activities and boundary layer injections in the warmer season (Lugauer et al., 1998; Zellweger et al., 2003; Collaud Coen et al., 2011; Griffiths et al., 2013; Herrmann et al., 2015). In addition, the station is regularly affected by Saharan dust events (SDEs) where Saharan dust is transported within the FT to the JFJ (Collaud Coen et al., 2004). Continuous measurements of aerosol physical properties (e.g. Baltensperger et al., 1997; Bukowiecki et al., 2016), trace gases (Steinbacher

et al., 2017) and meteorological conditions (Appenzeller et al., 2008) are conducted and give additional information on aerosol properties and air mass origin to complement the INP measurements conducted.

3.2 Aerosol particle measurements

5 A custom built scanning mobility particle sizer (SMPS), consisting of a DMA (TSI 3071) and a CPC (TSI, 3775), measured the aerosol size distributions between 20 to 600 nm in diameter with a time resolution of 6 min (Herrmann et al., 2015). Larger sized particles were measured by an OPC (GRIMM Dust Monitor 1.108; size range 0.23 – 16.4 μm). To merge the respective size distributions, the mobility and optical diameters were converted to volume equivalent diameters, assuming a particle density of 1565 kg m^{-3} (Sjogren et al., 2008) and a unity shape factor. An integrating nephelometer (TSI, 3563) and an
10 aethalometer (MAGEE scientific, AE31) measured the total aerosol scattering coefficients (three wavelengths) and the absorption coefficients (seven wavelengths), respectively. From this the single scattering albedo (SSA) at 450 nm, 550 nm and 700 nm is derived, as well as the SSA Ångström exponent, as described by Collaud Coen et al. (2004). For the normal background aerosol, the SSA increases with wavelength, resulting in a positive SSA Ångström exponent, while for Saharan dust particles, due to their larger size and different optical properties, the SSA decreases with wavelength and its exponent
15 becomes negative. A SDE is declared if the SSA exponent is negative for more than 4 consecutive hours. The aethalometer also measures the equivalent black carbon (eBC) mass concentration, derived from the attenuation measurement by applying the factory standard mass attenuation cross sect. of 16.6 $\text{m}^2 \text{g}^{-1}$ at 880 nm.

3.3 Cloud water samples

20 Cloud water samples at the JFJ were collected to determine the air mass origin and aerosol source regions by analyzing the samples for trace chemical elements (e.g. Zipori et al., 2015), and as such were taken in parallel to the ice nucleation measurements during cloudy periods (see Table 1 for sampling times). Samples were collected on the terrace next to the laboratories at the JFJ, with a home-built plexiglass plate (20x20x0.5 cm) which was attached vertically to the railing, facing the windward side. Prior to the sampling, the plate was cleaned with a super pure (65%) nitric acid solution (0.1-% w/w) and
25 double-deionized water, and a blank sample with double-deionized water was taken by pouring it over the sampler. The sampling method works only for supercooled cloud droplets which freeze upon contact with the plate, but not for ice crystals and precipitation particles, since ice crystals are deflected, and snowflakes are too heavy to stick to the sampler. The supercooled cloud droplets remain as an ice layer on the sampler, which is melted into pre-rinsed plastic bags and stored in Falcon® tubes. The samples were sent to the clean room laboratory at the Hebrew University of Jerusalem, where they were
30 analyzed for 23 trace metals ~~concentration~~ using inductively coupled plasma mass spectrometer (ICP-MS, Agilent, 7500cx). Detailed information regarding sample handling, analysis protocol and quality control can be found in Zipori et al., 2012 and Zipori et al. (2015).

The interpretation of the analysis of elemental concentrations was focused on sodium (Na), aluminium (Al), lead (Pb) and strontium (Sr). Sodium chloride (NaCl) is a natural component of the ocean, and the positive ion Na⁺ is found in an appreciable quantity in sea spray aerosol. In addition, Na⁺ found in cloud samples is always accompanied by Sr. The fraction of Sr coming from sea salt ($f(\text{Sr})_{ss}$) is used as an indication for air masses influence, and is calculated using the following equation:

$$5 \quad f(\text{Sr})_{ss} = \left[\frac{\text{Sr}}{\text{Na}} \right]_{ss} \times \left[\frac{\text{Na}}{\text{Sr}} \right]_{\text{sample}} \quad (1)$$

where $f(\text{Sr})_{ss}$ is the fraction of Sr contributed from sea salt, $[\text{Sr}/\text{Na}]_{ss}$ is the Sr to Na ratio found in sea salt, and $[\text{Sr}/\text{Na}]_{\text{sample}}$ are the concentrations of Na and Sr found in the samples (Herut et al., 1993). Furthermore, elemental ratios such Na/Al, Pb/Al and Pb/Na were used as indications for marine/dust, anthropogenic/dust and anthropogenic/marine influence in the samples, respectively.

10 In addition to the chemical analysis, Sr isotopic ratios were also measured with a multi-collector inductively coupled plasma mass spectrometer (MC-ICP-MS, NEPTUNE Plus). Sr separation was done with Sr-Spec resin flowing the method described by Stein et al., 1997. Since marine ⁸⁷Sr/⁸⁶Sr is constant with a value of 0.70917 (Hodell et al., 1990), while basalt and volcanic rocks has lower ratio and Saharan dust have higher ratio (Capo et al., 1998 and references therein), this parameter can be used to determine the prevailing aerosol type in the sample due to scavenging.

15

3.4 Back trajectories and source ~~emission~~ sensitivities

To obtain information on the trajectories of the air masses arriving at the JFJ, an ensemble of 10-day back trajectories were calculated every 6 hours with the LAGRANTO model (Wernli and Davies, 1997), based on ECMWF Integrated Forecast System wind fields. Back trajectories at five different locations, one ending at the JFJ and four displaced by 0.5° to the North and South, are started at four different altitude levels of 654 hPa, 704 hPa, 604 hPa and 754 hPa. In addition, source sensitivities, [which determine the potential contribution of ground-based regions to have emitted the particles arriving at the JFJ](#), are derived from the Lagrangian particle dispersion model, FLEXPART, products browser at EMPA ([http://lagrange.empa.ch/FLEXPART\ browser/](http://lagrange.empa.ch/FLEXPART_browser/); Stohl et al., 2005; Sturm et al., 2013; Pandey Deolal et al., 2014). [The source sensitivities, which thus indicate](#) give the possible origin of the particles as a probability of the geographical regions from which the aerosol particles were emitted. It simulates the release of 50,000 particles every 3 hours at the JFJ, and traces the particles backwards driven by ECMWF Integrated Forecast System wind fields.

25

3.5 Assessment of free tropospheric conditions

Different proxies are used in this study to qualitatively assess the exposure of the site to the FT. The ratio of total reactive nitrogen (NO_y , as the sum of nitrogen oxide, nitrogen dioxide and its atmospheric oxidation products) to carbon dioxide (CO) is commonly used as an indicator for boundary layer injections into the FT at elevated stations (Zellweger et al., 2003; Zanis et al., 2007; Pandey Deolal et al., 2013; Griffiths et al., 2014; Herrmann et al., 2015; Boose et al., 2016a). Both tracers are
5 subject to emissions from anthropogenic sources, however, the NO_y/CO ratio decreases with increasing transport (aging) of the air mass as CO is inert within the timescale of interest (days) while the concentration decay rate of NO_y is higher. Thus, a NO_y/CO ratio of 0.0057 ppb/ppb was chosen to distinguish between FT conditions and boundary layer influence, in accordance with the value reported for winter time measurements at the JFJ by Zellweger et al. (2003). NO_y/CO ratios below 0.0057 indicate FT conditions while an influence of boundary layer is likely for ratios above this value.

10 The concentration of particles > 90 nm was also used to identify FT conditions, since particles of this size are not formed in the FT, but are transported from the boundary layer, and therefore gives information on boundary layer influence (Herrmann et al., 2015). A threshold of 100 cm^{-3} was chosen, below which the air mass is assumed to be free tropospheric. It should be mentioned that the concentration of particles > 90 nm can be influenced by the occurrence of larger sized dust particles, and should therefore be considered with care during SDEs which are transported in the free troposphere.

15

4 Results

This is the first study where a chamber of HINC's design has been characterized and used for field measurements at conditions relevant to the mixed-phase cloud regime ($T > 235 \text{ K}$ and $\text{RH}_w > 100\%$). An identical chamber (Kanji and Abbatt, 2009) has been used for online field studies (Ladino et al., 2016) and processing of re-suspended field samples (Wilson et al., 2015), at
20 233 K and water subsaturated conditions. Here we compare results from two field campaigns in winter 2015 and 2016 to previously conducted INP measurements in the same season with PINC (Boose et al., 2016a). These results extend the time series of INP measurements below water saturation since winter 2012, and above water saturation since winter 2014 at the JFJ, which also contributes to the monitoring of INPs during winter months.

In winters 2015 and 2016, air masses containing high INP concentrations were sampled, which were excluded from the
25 comparison of the [background](#) campaign average INP concentrations. Furthermore, two such air masses from winter 2015 are discussed to relate the observed increase in INP concentrations to aerosol properties and air mass origin, and therefore examining the possible sources of ambient INPs. In winter 2015, the JFJ experienced FT conditions during 79% of the sampling time, with no specific increase in INP concentrations during boundary layer influence, while in winter 2016 the site was in the FT 92% of the time, and an event of increased INP concentration was observed during boundary layer influence, which is also
30 excluded from the comparison of campaign averages in this study.

4.1 Field measurements of INPs: Winters 2015 and 2016

4.1.1 INP concentrations at water subsaturated conditions (deposition nucleation)

Measurements below water saturation performed with HINC during winters 2015 and 2016 are shown in Fig. 8, and are compared to the measurements performed with PINC in winters 2012 – 2014 taken from Boose et al. (2016a). Campaign median and mean INP concentrations are given in Table 2, panel 1 and 3. The solid boxes in Fig. 8 include the entire distribution of INP concentrations measured including those below the LOD, whereas the dashed ones include only the INP concentrations above the LOD (see sect. 2.3). Excluding INP below the LOD artificially positively biases the data to higher INP values, therefore in discussing the results below the averages that include INP concentrations below the LOD are considered. During winters 2013 and 2014 an aerosol concentrator was used, which increased the signal-to-noise ratio by a factor of 3, and therefore the LOD for ambient INP was lowered.

The median (mean) INP concentration below water saturation, for all five field campaigns, is 0.1 (0.6) stdL^{-1} . The PINC measurements in winter 2012 – 2014 give a median (mean) INP concentration of 0.1 (0.2) stdL^{-1} , as compared to a HINC median (mean) concentration of ≤ 0.2 (1.2) stdL^{-1} for winters 2015 and 2016 (Table 2, third row). The natural variation in reported INP concentrations at a given temperature and RH condition can be as much as an order of magnitude after accounting for contributions from known INP sources. ~~The factor of 3-5 observed here between HINC and PINC suggests INP concentrations were comparable~~ good agreement between the chambers given that the measurements are done in different years, given that the minimum and maximum INP concentrations below water saturation overlap.

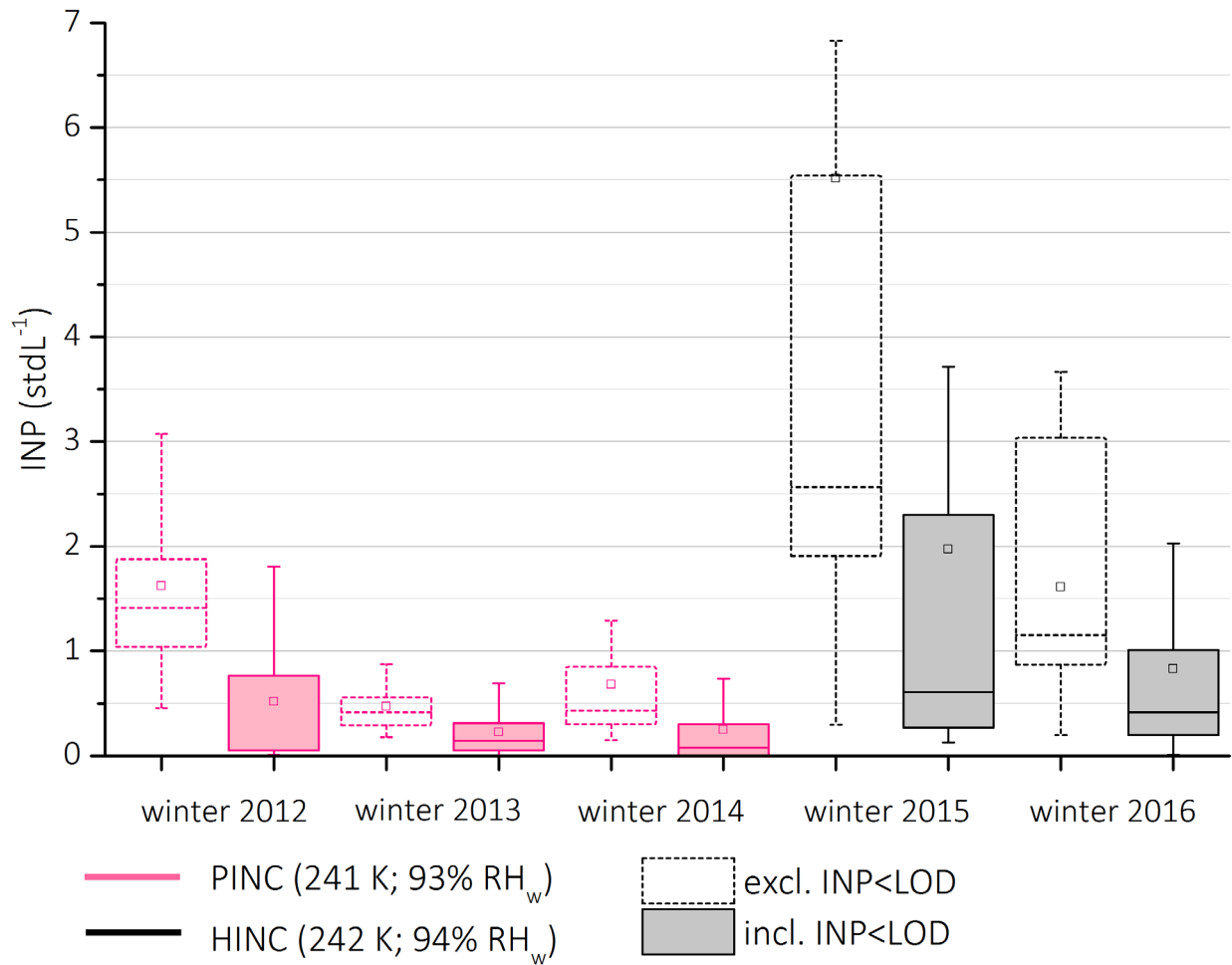


Fig. 8: Averaged INP concentrations observed below water saturation (see legend for exact temperature and RH conditions). Dashed boxplots represent only INPs > LOD, solid boxplots include INPs < LOD (see sect. 2.3 for details). Median: middle bar, mean: open square data point, box: interquartile range (25th to 75th percentile), whiskers: 5th and 95th percentile. Data used to produce the distributions exclude contributions from measurements during periods of [Saharan dust events and air masses of marine influence](#) ~~known air masses arrived~~ at the JFJ (see sect. 4.2) with anomalously high INP concentrations. See Table 1 for field campaign sampling times.

Table 2: Campaign INP concentrations (stdL⁻¹) excluding known cases of high INP concentrations, as measured with PINC (winter 2012 – 2014) and HINC (winter 2015 and 2016). Values are given for each field campaign (rows 1 and 2) and as averages over the PINC and HINC campaigns (rows 3 and 4). Measurements above water saturation were not conducted prior to 2014. INP concentrations here consider data below the LOD, and therefore can differ from values reported in Boose et al. (2016a).

RH _w	INP (stdL ⁻¹)	winter 2012	winter 2013	winter 2014	winter 2015	winter 2016
93/94%	median	< 0.05*	0.1	0.1	≤ 0.2*	≤ 0.2*
	mean	0.2	0.2	0.3	1.7	0.7
103 /104%	median	-		2.2	2.8	4.7
	mean	-		4.2	5.0	8.2
93/94%	median	0.1			≤ 0.2*	
	mean	0.2			1.2	
103 /104%	median	-		2.2	3.8	
	mean	-		4.2	6.6	

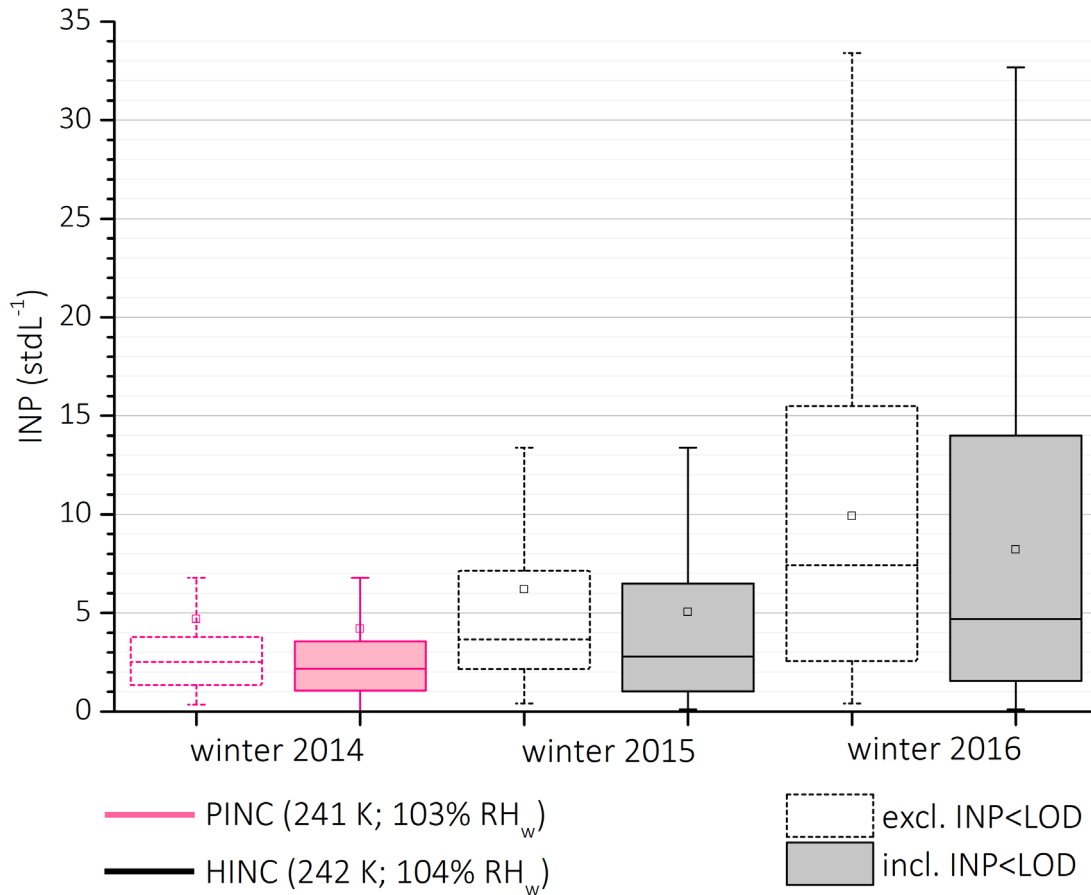
*Averaged INP concentrations reported as the minimum quantifiable INP concentration, for data points that fall under scenario 2 (see sect. 2.3) which are included in the averaging as this minimum.

4.1.2 INP concentrations at water saturated conditions (condensation freezing)

Figure 9 presents INP measurements above water saturation for winter 2014 (Boose et al., 2016a) and for winters 2015 and 2016 from this study. INP concentrations **above water saturation** are typically higher by approximately a factor of 10 as compared to water subsaturated conditions (Fig. 8) yielding a much higher signal-to-noise ratio with only few data points falling below the LOD. As such the differences between the solid and dashed box plots in Fig. 9 are small.

The median (mean) INP concentration in winter 2014 as measured with PINC was 2.2 (4.2) stdL⁻¹, and in winters 2015 and 2016, as measured with HINC, was 2.8 (5.0) and 4.7 (8.2) stdL⁻¹, respectively (Table 2, panels 2 and 4). In winter 2016, the INP concentrations are higher compared to the previous two winters. We explain this by the higher frequency of dust aerosol during that particular season. During the time of the measurements, two Saharan dust events were detected based on the SSA Ångström exponent criteria, but it is possible that the station was under the influence of dust particles without identification of a dust event which requires the condition of 4 continuous hours of a negative exponent of the SSA. This would imply that the majority of ambient particles were non-dust particles but dust could still contribute to the total aerosol loading. This is supported by source emission sensitivities derived by FLEXPART, indicating the Sahara as a source region for several days in addition to the declared SDEs, and by the concurrent increase in the particle concentrations > 0.4 μm, which is typical for dust. Therefore, this difference in INP concentrations between 2015 and 2016 can be explained by a natural inter-annual

variation. However, the difference in median and mean INP concentration is only a factor of 2. As compared to the PINC measurements in winter 2014, HINC measures higher median and mean INP concentrations by up to a factor of 2, which is considered low given that the possible range of observed INP concentrations for a given temperature can be much higher, as observed by e.g. Schrod et al. (2017) and DeMott et al. (2010).



5

Fig. 9: Averaged INP concentrations observed above water saturation (see legend for exact temperature and RH conditions). Dashed boxplots represent only INPs > LOD, solid boxplots include INPs < LOD (see sect. 2.3 for details). Median: middle bar, mean: open square data point, box: interquartile range (25th to 75th percentile), whiskers: 5th and 95th percentile. Data used to produce the distributions exclude contributions from measurements during periods of known air masses arrived at the JFJ (see sect. 4.2) with anomalously high INP concentrations. See Table 1 for field campaign sampling times.

10

4.2 Case studies

INP measurements were performed with HINC from 24th January to 9th February 2015 at the JFJ (Fig. 10). In this section, results from two events during winter 2015 are presented, for which INP concentrations above water saturation increased significantly above the campaign average. The events, which lasted several hours with higher INP concentrations, are shown in Fig. 10 (panel a). We discuss the air mass characteristics and aerosol properties that identify the most likely sources of the observed increase in ice-active particles.

4.2.1 Case 2nd February 2015: marine air

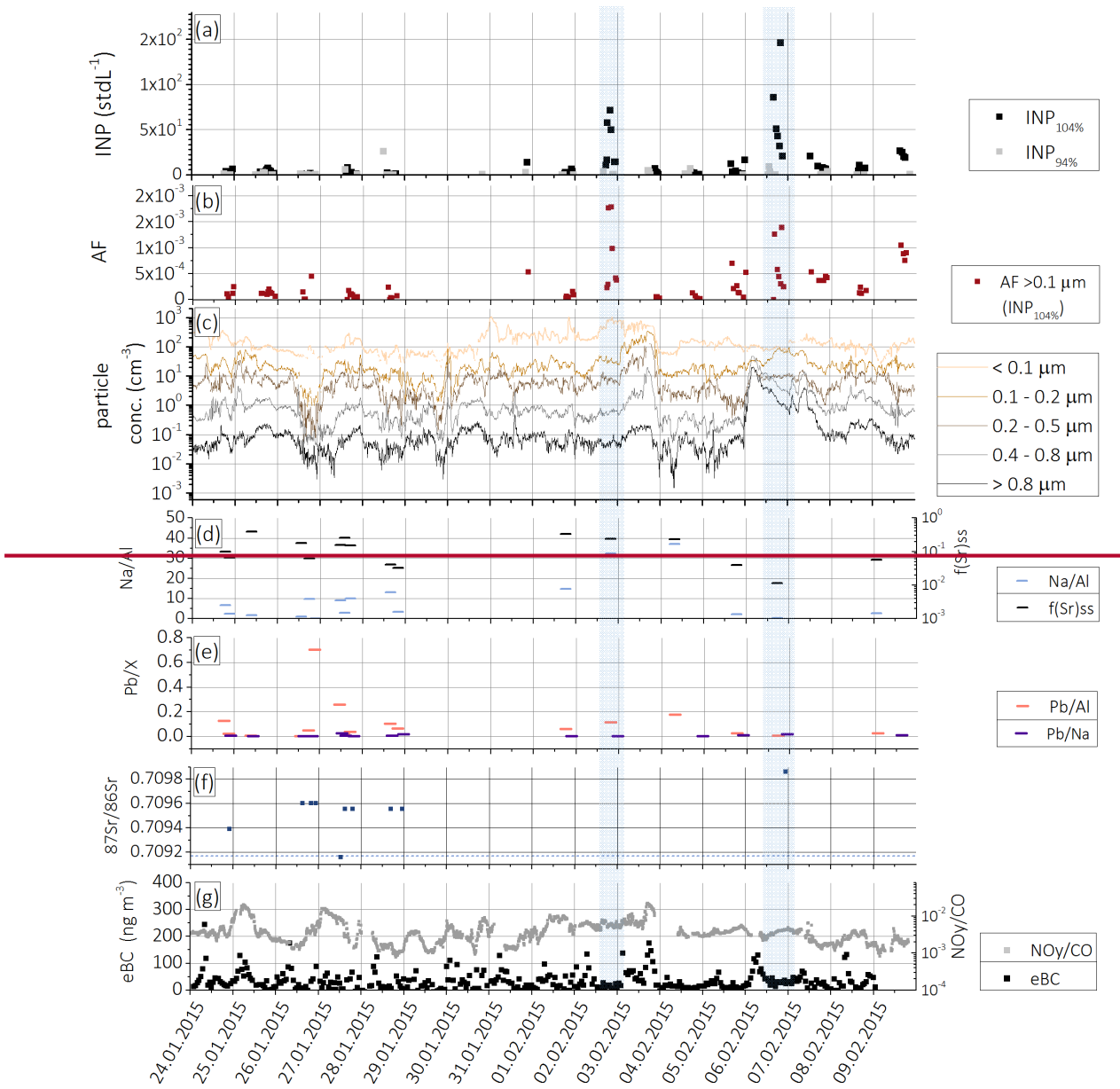
On 2nd February INP concentrations increased up to 72.1 stdL⁻¹ with median (mean) concentrations of 16.34 (33.3) stdL⁻¹ over a time of 5 hours (Fig. 10, panel a). Also an increase in the AF (considering particles > 0.1 μm) by an order of magnitude was observed from the campaign median (mean) value of 1.3×10^{-4} (2.8×10^{-4}) to a maximum of 1.8×10^{-3} indicating that the increase was not only due to a new increase in particle number, but a higher fraction of the aerosol being ice active (Fig. 10, panel b).

The SSA showed a wavelength dependent increase, which is typical for background aerosol conditions at the JFJ, and is an indication for the absence of Saharan dust. The particle concentration in the size bin 0.4 – 0.8 μm and > 0.8 μm did not increase (Fig. 10, panel c) confirming no influence from larger (dust) particles. The ratio Na/Al derived from the cloud water samples during this sampling period increased (Fig. 10, panel d), and at the same time, $f(\text{Sr})_{\text{ss}}$ increased (Fig. 10, panel d), which suggests that the air mass arriving at the JFJ was of marine origin. Sr isotopic ratios, which could strengthen the identification of a marine source, were not available for that day, due to the small volume of cloud water collected which was not sufficient for the isotopic analyses. However, source sensitivities (Fig. 11, panel a) indicate most sources over the Northern Sea followed by the Northern Atlantic and Norwegian Sea. In addition to the source sensitivities, 10-day back trajectories also support marine sources and also reveal that the air parcel travelled over Northern Europe, England and France to the JFJ (Fig. 11, panel c), and could have been subject to aging and anthropogenic emissions. Indeed on that day an increase in particle concentrations < 0.1 μm was observed (Fig. 10, panel c) which could be an indication for an anthropogenic influence ~~on the aerosol~~. This is supported by the finding of an increased NO_y/CO ratio (Fig. 10, panel g), which is a tracer for anthropogenic influence of the air mass arriving at the JFJ. Because eBC mass concentrations were low during that time (Fig. 10, panel g), a higher contribution of marine particles than of the anthropogenic emissions to the observed INPs is likely. However, the influence of aging processes resulting in internal mixing of the marine particles with anthropogenic emissions arriving at the JFJ cannot be ruled out.

The cloud sample analysis of the Na/Al ratio and the $f(\text{Sr})_{\text{ss}}$, as well as the isotopic ratio of Sr (Fig. 10, panel d, e, f, respectively) revealed a marine source of particles on the morning of Jan 27th. Unfortunately, no INP measurements are available for that time, only later from the same day, when the marine influence decreased, and INP measurements were within the campaign average. Another marine air mass event was detected in winter 2016, on 6th March (data not shown here), when the indicators

discussed above for marine influence (i.e. from cloud water samples) were similar. Using the same methods discussed above, we identified the INP concentration during the winter 2016 marine event to be on the same order of magnitude as during the winter 2015 marine event, with a median (mean) INP concentration of 14.9 (25.5) stdL^{-1} , and a maximum concentration of 176.8 stdL^{-1} .

- 5 In Fig. 12 we compare the INP concentrations from the periods of marine influence at the JFJ to those reported in DeMott et al. (2016) for marine aerosols from online and offline INP measurements from different laboratory and field samples of sea waters. We find good agreement at 242 K for the measured concentrations at the JFJ during the two marine events, compared to the laboratory and field measurements of marine INPs. ~~This is somewhat surprising given the measurements from DeMott et al. (2016) are taken directly over ocean sources, or from ocean waters in an ocean tank simulator. However, w~~We also note
- 10 that during the occurrence of a marine event at the JFJ an increase in the AF (Fig. 10, panel b) is observed simultaneously, indicating an enrichment of ice active particles compared to the background levels of INP. Furthermore these observations might also suggest that long range transport of small marine particles to the JFJ does not result in a suppression of their ice nucleation abilities, but rather the INPs retain their ice nucleation abilities during transport to the JFJ despite possible mixing with particles from anthropogenic emissions. Modelling studies have reported that marine aerosols as INPs are relevant on a
- 15 global scale (Yun and Penner, 2013), especially in remote marine areas where dust abundance is low (Burrows et al., 2013; Wilson et al., 2015; Vergara-Temprado et al., 2017).



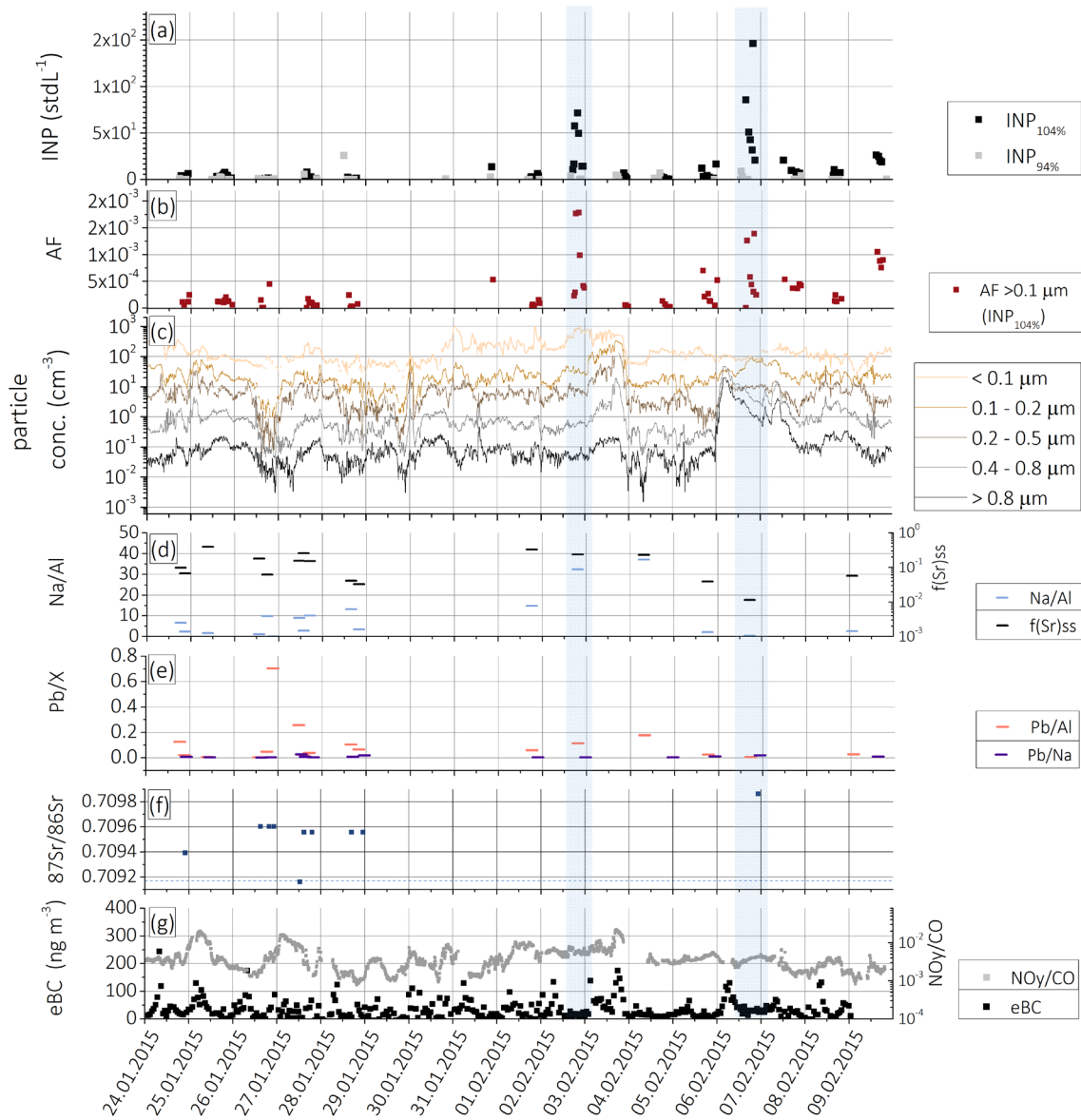


Fig. 10: Time series of (a) INP concentrations at 242 K and $RH_w = 104\%$ and 94% measured with HINC; (b) AF of $INP_{104\%}$ considering particles $> 0.1 \mu\text{m}$; (c) particle concentrations $< 0.1 \mu\text{m}$, $0.1 - 0.2 \mu\text{m}$, $0.2 - 0.5 \mu\text{m}$ (SMPS) and $0.4 - 0.8 \mu\text{m}$ and $> 0.8 \mu\text{m}$ (OPC), given in volume equivalent diameter; cloud water sample analysis, given in ratios of mass of (d) Na/Al and $f(\text{Sr})_{\text{ss}}$ and (e) Pb/Al and Pb/Na; (f) isotopic ratio $^{87}\text{Sr}/^{86}\text{Sr}$, the dashed line represents the constant value for marine sea salt ($=0.70917$); (g) eBC mass concentration and NO_y/CO ratio.

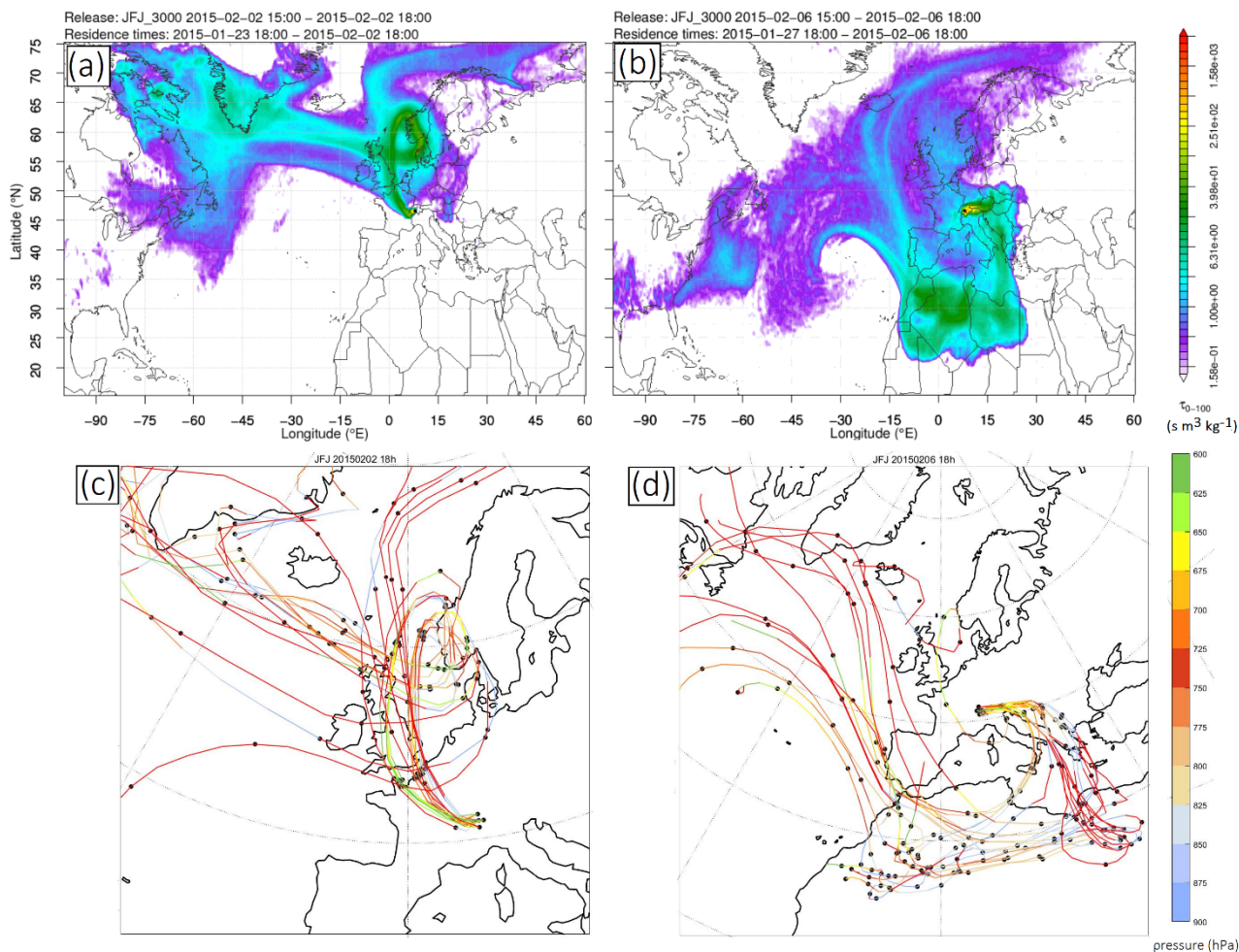
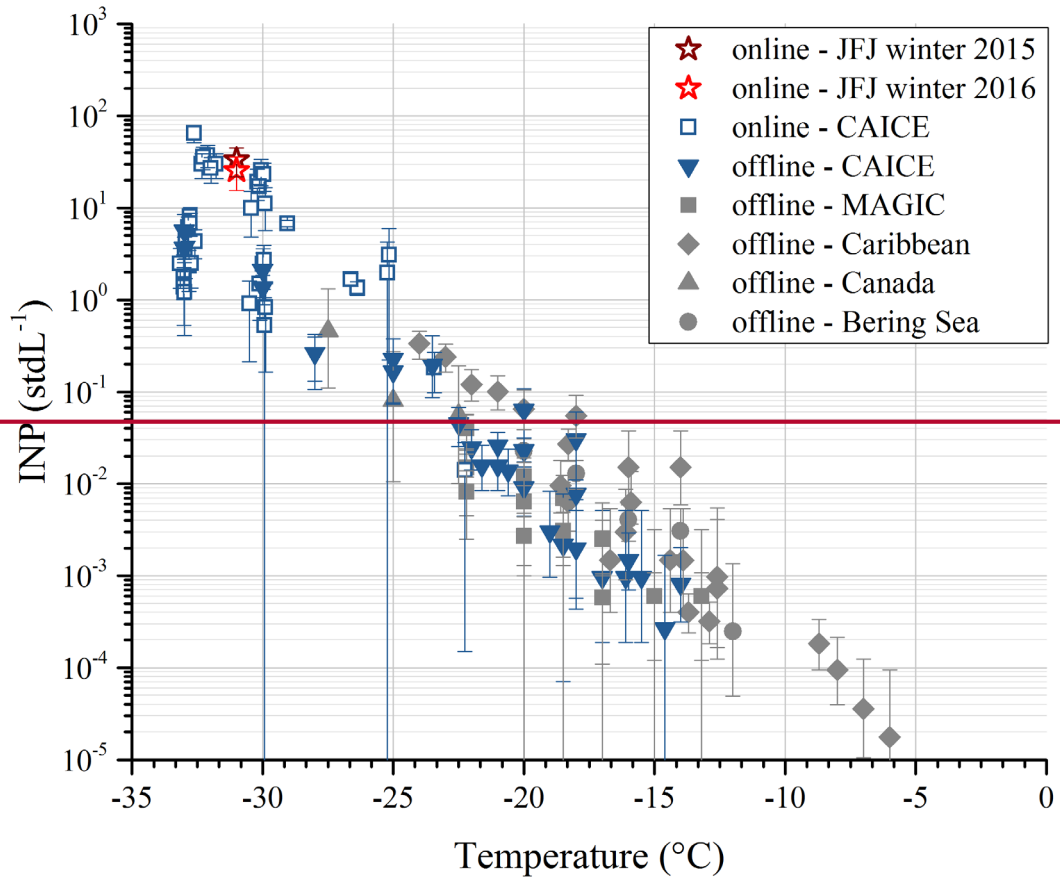


Fig. 11: (a and b) FLEXPART emission sensitivity fields for 2nd (a) and 6th (b) February 2015, calculated 100 m above model ground level ([http://lagrange.empa.ch/FLEXPART\ browser/](http://lagrange.empa.ch/FLEXPART_browser/); Stohl et al., 2005; Sturm et al., 2013; Pandey Deolal et al., 2014); the colour code represents the strength of source region contributions to the aerosol burden given in a unit flux per area. (c and d) 10-day back trajectories for 2nd (c) and 6th (d) February 2015 calculated with LAGRANTO (Wernli and Davies, 1997); the colour code represents the trajectory pressure above model ground, black points indicate each 24-hour back calculation.



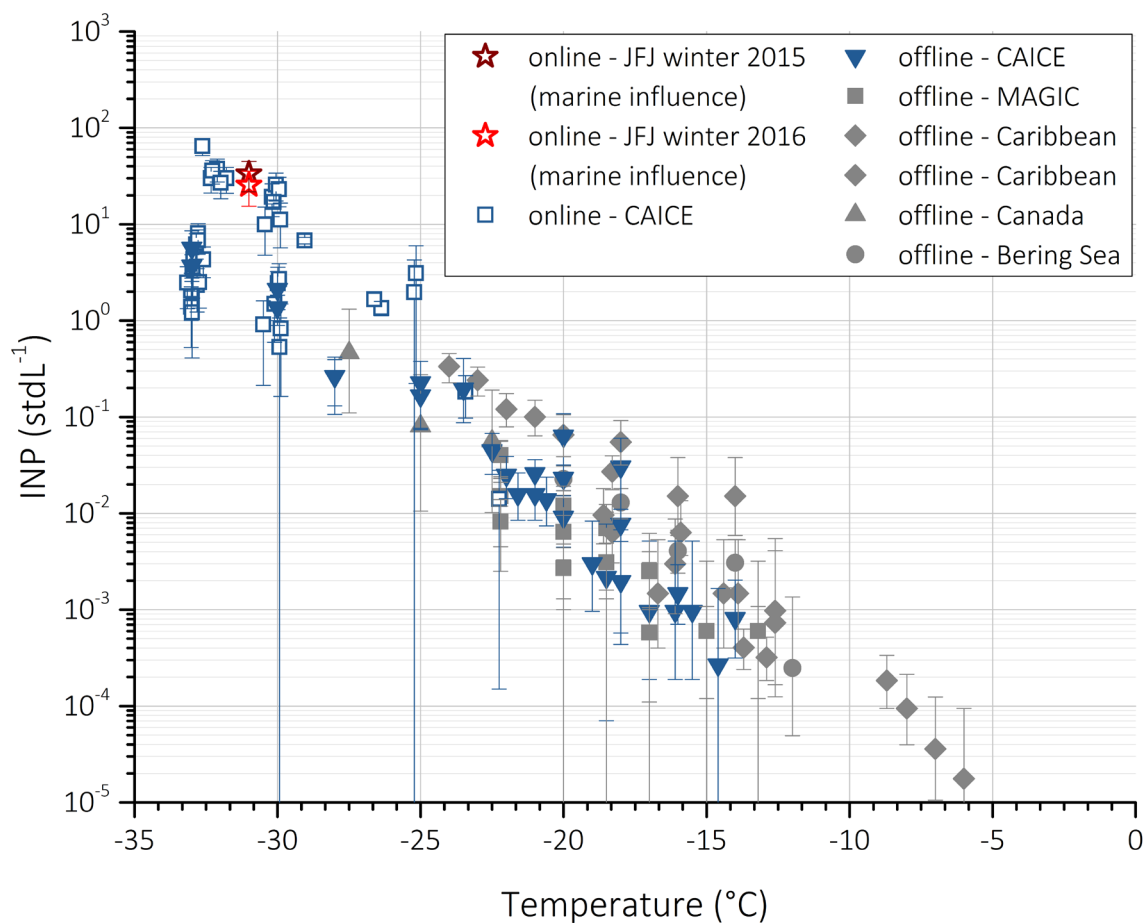


Fig. 12: INP concentrations as a function of sampling temperature. Data represent measurements of sea spray particles (DeMott et al., 2016) in the laboratory (blue) at the Center for Aerosol Impacts on Climate and the Environment (CAICE), and for ambient marine boundary layer particles (grey), during different campaigns (see label for respective field campaign name), and two marine events obtained at the JFJ (stars, this study). Measurements are differentiated between online (open symbols) and offline (filled symbols) freezing methods. Laboratory data are normalized to total particle concentrations of 150 cm⁻³. Error bars are given for twice the Poisson sampling error, and give up-to-date values, which can differ from published ones in DeMott et al. (2016) (personal communication with the author).

4.2.2 Case 6th February 2015: SDE

An increase in INP concentrations was measured on 6th February with values up to 146.2 stdL⁻¹ and with a median (mean) concentration of 42.6 (55.4) stdL⁻¹ (Fig. 10, panel a). The AF also increased by a factor of 10, up to 1.4x10⁻³ (Fig. 10, panel b). During that time a SDE was detected based on the SSA Ångström exponent criterion (see sect. 3.2). An increase in the particle concentration 0.4 – 0.8 µm and > 0.8 µm (Fig. 10, panel c) supports the presence of larger mineral dust particles, as well as a decrease in the Na/Al ratio (Fig. 10, panel d) indicating a dusty air mass rich in alumina silicate minerals. Emission sensitivities (Fig. 11, panel b) identify a large area over the central Sahara as a particle source region, and back trajectories calculated for this day (Fig. 11, panel d) show that the air parcel was traveling from the Saharan Desert over the Mediterranean to the Alps. An influence from the ocean on the air mass composition cannot be excluded according to the back trajectories, as the height of the calculated back trajectories over the Mediterranean Sea was > 950 hPa, which indicates some contact with boundary layer air. In addition, source sensitivities show a possible influence from this region. Chemical analysis of cloud water sampled at the JFJ during the arrival of the respective air mass reveals that the f(Sr)_{ss} is low (Fig. 10, panel d), which confirms a low marine influence. In addition, the Sr isotopic ratio ⁸⁷Sr/⁸⁶Sr is increased to a value of 0.70986, (Fig. 10, panel f), which is an indication for Saharan dust (Capo et al., 1998 and references therein). Furthermore, ratios of Na/Al and Pb/Al were low, due to an enrichment of Al in the water sample. The eBC mass concentration and the NO_y/CO ratio are low (Fig. 10, panel g), which also supports ~~for no anthropogenic influence~~ a pristine air mass.

Below water saturation, four measurements of INP concentrations were taken during the dust event, two measurements being below the LOD, and two measurements with increased concentrations of 3 and 8.8 stdL⁻¹, as compared to a campaign median (mean) value of ≤ 0.2 stdL⁻¹ (1.7 stdL⁻¹). For conditions below water saturation, dust events yielding higher INP concentrations at the JFJ have been reported before (Chou et al., 2011) and in the Saharan air layer by aircraft sampling (DeMott et al., 2003b). At water saturated conditions, this is the first study to clearly show that during a SDE at the JFJ an increase in INPs is observed at 242 K. This agrees with the findings of; ~~however not a surprising result as~~ an increase in INP concentrations ~~was~~ observed during minor influence of Saharan dust (Boose et al., 2016b). We note that at 265 K an increase in immersion INPs ~~concentrations~~ at the JFJ was not observed during SDEs (Conen et al., 2015) indicating that dust contributes to ice nucleation at colder temperatures as has been previously suggested in numerous studies (see references in Hoose and Möhler, 2012, Murray et al., 2012, Kanji et al., 2017).

A calculation based on the size distribution of ambient particles from the field campaigns at the JFJ reveals that the maximum contribution of 1% aerosol particles > 5 µm remain unactivated in HINC would be 0.026 stdL⁻¹ (0.285 stdL⁻¹) in winter 2015 (winter 2016), during a time when INP concentrations reached 85.5 stdL⁻¹ (154.5 stdL⁻¹). Thus a positive bias of larger unactivated particles to INP concentrations should be insignificant.

Also an increase in larger particles was observed on 3rd February (Fig. 10, panel c), particularly in the size range 0.1-0.8 µm. During this period, ~~however~~, the SSA increased with wavelength, which is atypical for Saharan dust, and also no decrease in

the Na/Al ratio was observed. In addition, neither source sensitivities nor back trajectories (see Fig. A32 in the appendix) showed influence from the Saharan desert. However, during this time construction work on the tunnel systems in the Alps under the JFJ was conducted, possibly leading to the abrasion of rocks and the emission of larger particles. These particles were not ice-active at the sampling conditions in HINC, since neither an increase in INP concentration above nor below water saturation was observed.

5 Conclusions

This is the first study in which an INP counter of HINC's design has been used to quantify ambient INP concentrations at temperature and RH conditions relevant for mixed-phase cloud formation, where both liquid and ice particles can co-exist. We demonstrated that HINC, based on the design of the UT-CFDC (Kanji and Abbatt, 2009), was successfully deployed to sample ambient INP concentrations. The RH and temperature accuracy was determined for the temperature range 223 - 263 K and at conditions of sub- and supersaturation with respect to water by observing droplet activation of sulfuric acid particles, and deliquescence of sodium chloride and ammonium sulfate particles. In addition, homogenous freezing of sulfuric acid aerosols at temperatures < 235 K also validated accurate conditions in HINC for ice formation. The uncertainty in INP measurements in HINC arises from the variation in temperature and RH, to which the aerosols in the chamber are exposed to, and are $T \pm 0.4$ K, and $RH_w \pm 1.5\%$ ($RH_i \pm 3\%$) for temperatures $T > 235$ K. These variations can lead to uncertainties in activated fractions in the region of activation and nucleation where nucleation rates are a steep function of RH. Conducting field measurements of INPs with ambient aerosols at 242 K and RH 104%, where we ensure all particles are fully activated to droplets, reduces the associated uncertainties in INP concentrations arising from the RH variation in the chamber. HINC was characterized for an optimum residence time to maximize the growth time of the ice crystals but avoid particle losses due to gravitational settling in the horizontally oriented chamber. INP concentration measurements with HINC from winters 2015 and 2016 at the JFJ were presented at 242 K for $RH_w = 94\%$ and 104%. Median INP concentrations, excluding specific events of high INP concentrations, were on average ≤ 0.2 stdL⁻¹ below water saturation, and these low concentrations are within the range of the median INP concentration of 0.1 stdL⁻¹ measured at the same site before, with the Portable Ice Nucleation Chamber, PINC (Boose et al., 2016a), during winters 2012-2014. Above water saturation, INP concentrations are in general an order of magnitude higher compared to below water saturation, with a median concentration of 3.87 stdL⁻¹ for winters 2015 and 2016 (HINC), and 2.2 stdL⁻¹ during winter 2014 (PINC; Boose et al., 2016a). The small differences in INP concentrations from year-to-year are expected to occur due to natural variability.

In winter 2015, an increase in INP concentrations above water saturation was observed during the influence of an air mass of marine origin, with up to 72.1 stdL⁻¹, and a median concentration of 16.3 stdL⁻¹. The support of marine influence was based on chemical analysis of cloud water samples with a high $f(\text{Sr})_{\text{ss}}$, and an increased Na/Al ratio. Model calculations of back trajectories and air mass origin further support our conclusion of a marine source, but cannot exclude contributions from

anthropogenic or other sources due to entrained particles, which can result in chemical ageing processes during transport to the JFJ. Another marine event was identified in winter 2016, when INP concentrations increased to values up to 176.8 stdL⁻¹. During winter 2014, Boose et al. (2016a) identified a marine influenced air mass arriving at the JFJ, but the INP concentrations were within the campaign average. ~~however, with increased ice-active surface site densities, compared to periods when Saharan dust could have been present.~~ Our findings ~~together with those from Boose et al. (2016a)~~ suggest that the JFJ could regularly be affected by marine aerosols, which can therefore contribute to bursts of increased INP populations in the free troposphere.

An air mass with an increase of INPs was sampled during a SDE in winter 2015, with median INP concentrations reaching 42.6 stdL⁻¹, and a peak INP concentration of 146.2 stdL⁻¹. The identification of the dust laden air mass was supported by several independent measures of aerosol physical and optical properties such as an increase in larger sized particles (0.4 – 0.8 µm and > 0.8 µm), and a negative SSA Ångström exponent, and chemical analysis of cloud water samples, as well as a decrease in the Na/Al and Pb/Al ratios due to an enrichment in Al. The dust source was further supported by air mass back trajectory and source sensitivity calculations revealing the Saharan desert as a source region.

To extend measurements to warmer temperatures, a significantly improved LOD must be achieved as INP are rarer ~~at warmer temperatures~~. To quantify INP concentrations in the range between 253 and 273 K, a technique that is sensitive at warmer temperatures must be used, such as offline techniques of drop freezing (e.g. Mason et al., 2015; Conen et al., 2015). However, improving the LOD of online counters can also be achieved by means of using an aerosol concentrator upstream of an INP counter. This has been done in winters 2013 and 2014 (Boose et al., 2016a), achieving a concentration factor of 3. Recently a more efficient aerosol concentrator was implemented during a field campaign at the JFJ (winter 2017) and will be the subject of a separate study. The ability to conduct field measurements with HINC will aid future measurements with increasing frequency at the JFJ, to determine diurnal, and inter-annual variabilities in INP concentrations at this location.

Acknowledgements

This research was funded by the Global Atmospheric Watch, Switzerland (MeteoSwiss GAW-CH+ 2014-2017). We thank the International Foundation High Altitude Research Station Jungfrauoch and Gornergrat (HFJG) for the opportunity to perform the measurements, and the custodians Maria and Urs Otz, Joan and Martin Fischer, Susanne and Felix Seiler for their support and help. For providing meteorological data we thank MeteoSwiss. Y. Boose and U. Lohmann acknowledge funding from the European Union's Seventh Framework Programme (FP7/2007-2013) under grant agreement no. 603445 (BACCHUS). Trace gases measured at the JFJ are part of the Swiss National Air Pollution Monitoring Network which is jointly run by EMPA and the Swiss Federal Office for the Environment. This project has also received funding from the European Union's Horizon 2020 research and innovation program under grant agreement No 654109, and was as well supported by the Swiss State Secretariat for Education, Research and Innovation (SERI) under contract number 15.0159-1. The opinions expressed and arguments employed herein do not necessarily reflect the official views of the Swiss Government. We acknowledge James Atkinson, Robert David, Fabian Mahrt, [Franz Friebe](#), Nadine Borduas and Claudia Marcolli for useful discussions. For technical support we would like to thank Hannes Wydler, whose expertise greatly helped to improve the instrument.

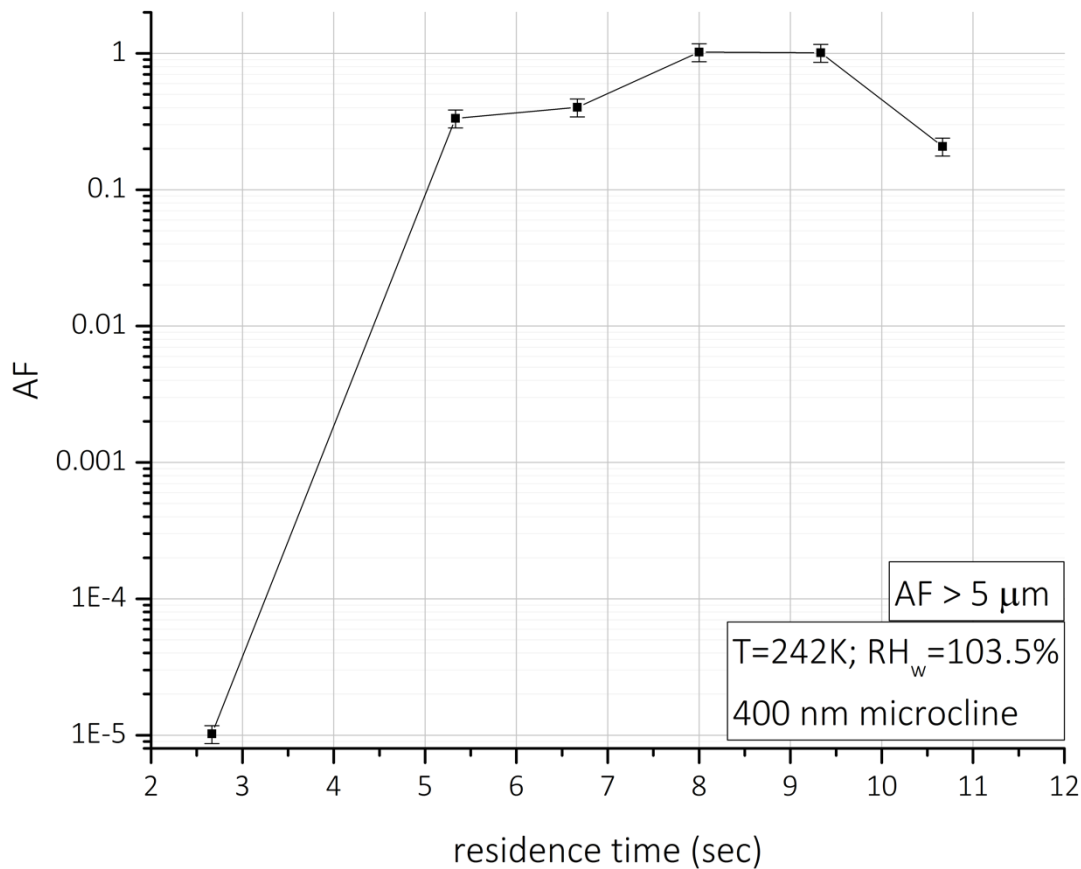
Author Contributions

LL wrote the manuscript, with contributions from ZAK. ZAK and UL conceived the field study. ZAK and LL designed the laboratory experiments. Field measurements were designed by LL, YB and ZAK. LL conducted all INP measurements and analyzed all INP data. LL, ZAK and UL interpreted the INP data. AZ conducted part of the cloud water sampling, and analyzed and interpreted all the data. EH contributed data on size distributions. NB contributed data on absorption characteristics. MS contributed data on trace gases. ZAK oversaw the overall project.

Appendix A

A1: Residence time experiments for 242 K and 104% RH_w

The optimum residence time for the ice crystal detection at the conditions used in the field experiments was determined by using 400 nm microcline particles which are INPs at this temperature (e.g. Atkinson et al., 2013a). The position of the aerosol injector and thus the residence time was chosen accordingly, and the AF of the ice crystal concentration in the OPC size channel $> 5 \mu\text{m}$ was determined (Fig. A1, appendix). The tests revealed the maximum AF at 8 seconds residence time, which gives the aerosols enough time to activate into ice crystals and grow to sizes $> 5 \mu\text{m}$, at the same time preventing gravitational settling losses.

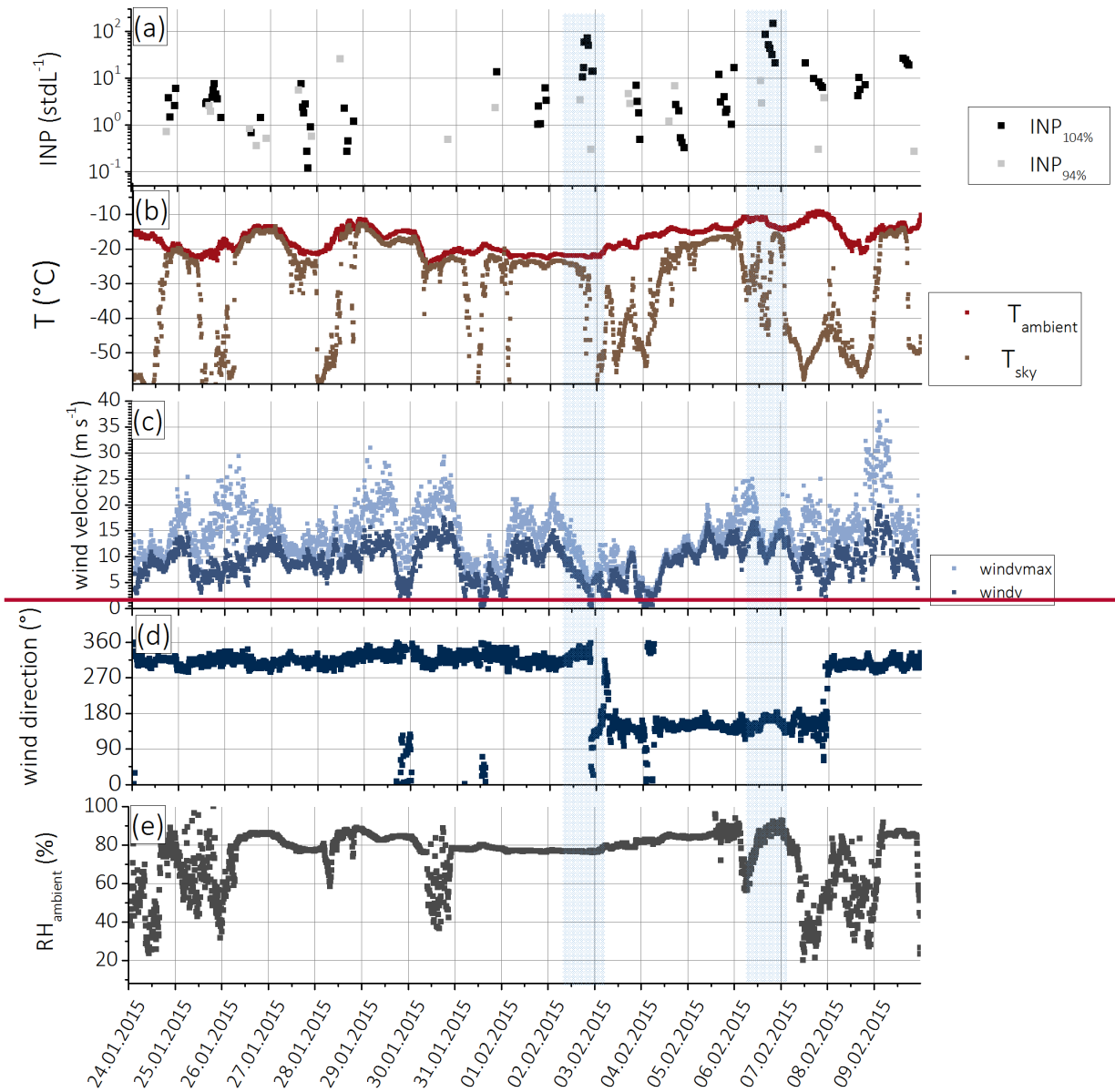


10 Fig. A1: AF as function of residence time for 400 nm microcline particles at 242 K and RH_w = 104%; data is the average over a total of three experiments at each residence time.

A1-A2 Meteorological conditions

For a complete description of the INP measurements in winter 2015 the meteorological data during the same sampling period are presented in Fig. A1A2.

- 5 Ambient temperatures (T_{ambient}) stayed for the whole campaign duration below 0°C , ranging between -9°C to -24°C (Fig. A1A2, panel b). The sky temperature (T_{sky}) is calculated from the longwave radiation measured at the site, and is used to discriminate between in-cloud and out-of-cloud conditions (see Herrmann et al., 2015). During times when the site is in clouds, one would expect the difference between T_{ambient} and T_{sky} to be small, since the longwave radiation received presents the temperature of the cloud surrounding the site and having a similar temperature as the ambient air. INP measurements in- and out-of-cloud conditions do not show significant differences.
- 10 The wind velocity (windv) as well as the hourly maximum wind velocity (windvmax) also do not show a correlation with INP concentrations, and a relationship between them is excluded (Fig. A1A2, panel c), also excluding the role of blowing snow (Lloyd et al., 2015) on our measurements. The wind direction ~~on~~ at the JFJ (Fig. A1A2, panel d) represents the two typical wind directions from Northwest and Southeast, which is a result of the orientation of the surrounding terrain. The SDE on 6th February was transported in a south-easterly flow, as expected, despite no relation to INP concentration can be concluded.
- 15 Also the ambient RH does not show an influence on INP concentrations (Fig. A1A2, panel e).



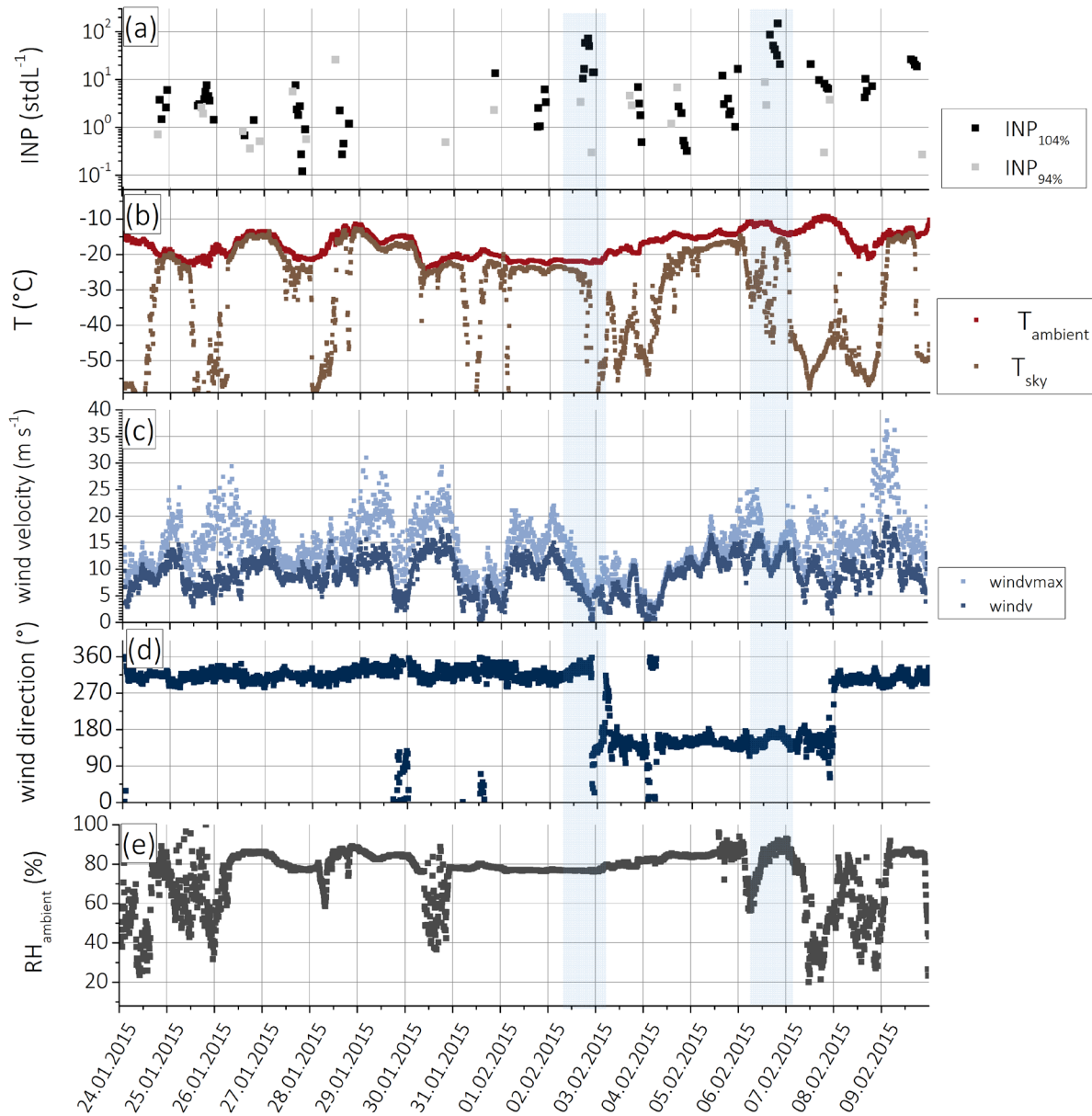


Fig. A1A2: Time series of meteorological data, taken by MeteoSwiss: (a) INP concentrations at 242 K and $RH_w = 104\%$ and 94% as measured with HINC; (b) ambient temperature (T_{ambient}) and sky temperature (T_{sky}) (Herrmann et al., 2015); (c) hourly maximum wind velocity (windmax) and wind velocity (windv); (d) wind direction; (e) ambient relative humidity (RH_{ambient}).

A2–A3 Back trajectory and source emission sensitivities for 3rd February 2015

During February 3rd 2015 an increase in aerosol particles $> 0.5 \mu\text{m}$ was observed, which was not leading to an increase in INP concentrations below and above water saturation. To exclude influence of Saharan dust particles, which is also indicated in an increase in larger particles, we show here source emission sensitivities (Fig. A23, panel a) and back trajectories (Fig. A32, panel b) for the respective day, which indicate the air mass arriving at the JFJ originated in Northern Europe, and therefore excluding the Sahara as a source region.

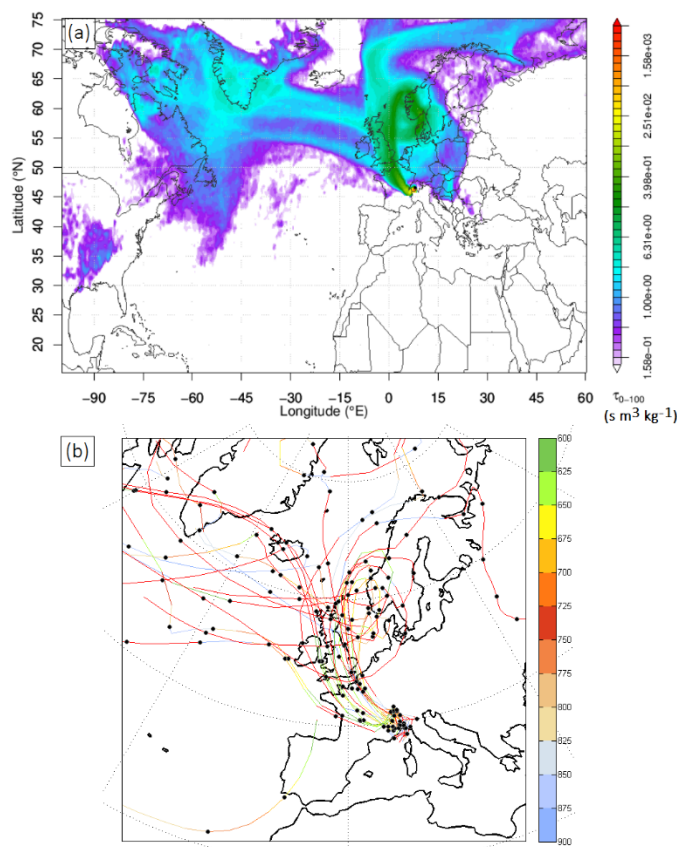


Fig. A2A3: (a) FLEXPART emission sensitivity fields for 3rd February 2015 calculated 100 m above model ground level (http://lagrange.empa.ch/FLEXPART_browser/; Stohl et al., 2005; Sturm et al., 2013; Pandey Deolal et al., 2014), colour code represents the strength of source region contributions to the aerosol burden given in a unit flux per area. (b) 10-day back trajectories calculated with LAGRANTO (Wernli and Davies, 1997); the colour code represents the trajectory pressure above model ground, black points indicate each 24 hour back calculation.

References

- Aller, J. Y., Kuznetsova, M. R., Jahns, C. J., and Kemp, P. F.: The sea surface microlayer as a source of viral and bacterial enrichment in marine aerosols, *Journal of Aerosol Science*, 36, 801-812, 10.1016/j.jaerosci.2004.10.012, 2005.
- Alpert, P. A., Aller, J. Y., and Knopf, D. A.: Initiation of the ice phase by marine biogenic surfaces in supersaturated gas and supercooled aqueous phases, *Physical Chemistry Chemical Physics*, 13, 19882-19894, 10.1039/c1cp21844a, 2011.
- Ansmann, A., Tesche, M., Althausen, D., Muller, D., Seifert, P., Freudenthaler, V., Heese, B., Wiegner, M., Pisani, G., Knippertz, P., and Dubovik, O.: Influence of Saharan dust on cloud glaciation in southern Morocco during the Saharan Mineral Dust Experiment, *Journal of Geophysical Research-Atmospheres*, 113, 10.1029/2007jd008785, 2008.
- Appenzeller, C., Begert, M., Zenklusen, E., and Scherrer, S. C.: Monitoring climate at Jungfraujoch in the high Swiss Alpine region, *Sci. Total Environ.*, 391, 262 - 268, 10.1016/j.scitotenv.2007.10.005, 2008.
- Ardon-Dryer, K., and Levin, Z.: Ground-based measurements of immersion freezing in the eastern Mediterranean, *Atmos. Chem. Phys.*, 14, 5217-5231, 10.5194/acp-14-5217-2014, 2014.
- Ardon-Dryer, K., Garimella, S., Huang, Y. W., Christopoulos, C., and Cziczo, D. J.: Evaluation of DMA Size Selection of Dry Dispersed Mineral Dust Particles, *Aerosol Science and Technology*, 49, 828-841, 10.1080/02786826.2015.1077927, 2015.
- Atkinson, J. D., Murray, B. J., Woodhouse, M. T., Whale, T. F., Baustian, K. J., Carslaw, K. S., Dobbie, S., O'Sullivan, D., and Malkin, T. L.: The importance of feldspar for ice nucleation by mineral dust in mixed-phase clouds, *Nature*, 498, 355-358, 10.1038/nature12278
2013a.
- Atkinson, J. D., Murray, B. J., Woodhouse, M. T., Whale, T. F., Baustian, K. J., Carslaw, K. S., Dobbie, S., O'Sullivan, D., and Malkin, T. L.: The importance of feldspar for ice nucleation by mineral dust in mixed-phase clouds, *Nature*, 498, 355 - 358, 10.1038/nature12384, 2013b.
- Augustin, S., Wex, H., Niedermeier, D., Pummer, B., Grothe, H., Hartmann, S., Tomsche, L., Clauss, T., Voigtländer, J., Ignatius, K., and Stratmann, F.: Immersion freezing of birch pollen washing water, *Atmos. Chem. Phys.*, 13, 10989 -11003, 10.5194/acp-13-10989-2013, 2013.

- Avramov, A., Ackerman, A. S., Fridlind, A. M., van Dierenhoven, B., Botta, G., Aydin, K., Verlinde, J., Korolev, A. V., Strapp, J. W., McFarquhar, G. M., Jackson, R., Brooks, S. D., Glen, A., and Wolde, M.: Toward ice formation closure in Arctic mixed-phase boundary layer clouds during ISDAC, *J. Geophys. Res. Atmos.*, 116, 10.1029/2011JD015910, 2011.
- 5 Baltensperger, U., Gäggeler, H. W., Jost, D. T. a. L., M., Schwikowski, M., and Weingartner, E.: Aerosol climatology at the high-alpine site Jungfraujoch, Switzerland, *J. Geophys. Res.*, 102, 19 707 - 719 715, 10.1029/97JD00928, 1997.
- Baltensperger, U., Schwikowski, M., Jost, D. T., Nyeki, S., Gäggeler, H. W., and Poulida, O.: Scavenging of atmospheric constituents in mixed phase clouds at the high-alpine site jungfraujoch part I: Basic concept and aerosol scavenging by clouds, *Atmos. Environ.*, 32, 3975 - 3983, 10.1016/S1352-2310(98)00051-X, 1998.
- Bigg, E. K.: Cross Sections of Ice Nucleus Concentrations at Altitude over Long Paths, *J. Atmos. Sci.*, 24, 226-229,
10 10.1175/1520-0469(1967)024<0226:CSOINC>2.0.CO;2, 1967.
- Bingemer, H., Klein, H., Ebert, M. a. H., W., Bundke, U., Herrmann, T., Kandler, K., Müller-Ebert, D., Weinbruch, S., Judt, A., Nillius, B., Ardon-Dryer, K., Levin, Z., and Curtius, J.: Atmospheric ice nuclei in the Eyjafjallajökull volcanic ash plume, *Atmos. Chem. Phys.*, 12, 857 - 867, 10.5194/acp-12-857-2012, 2012.
- Bode, A. A. C., Pulles, P. G. M., Lutz, M., Poulisse, W. J. M., Jiang, S., Meijer, J. A. M., van Enckevort, W. J. P., and Vlieg,
15 E.: Sodium Chloride Dihydrate Crystals: Morphology, Nucleation, Growth, and Inhibition, *Cryst. Growth Des.*, 15, 3166-3174, 10.1021/acs.cgd.5b00061, 2015.
- Boose, Y., Kanji, Z. A., Kohn, M., Sierau, B., Zipori, A., Crawford, I., Lloyd, G., Bukowiecki, N., Herrmann, E., Kupiszewski, P., Steinbacher, M., and Lohmann, U.: Ice Nucleating Particle Measurements at 241 K during Winter Months at 3580 m MSL in the Swiss Alps, *J. Atmos. Sci.*, 73, 2203-2228, 10.1175/JAS-D-15-0236.1, 2016a.
- 20 Boose, Y., Sierau, B., Garcia, M. I., Rodriguez, S., Alastuey, A., Linke, C., Schnaiter, M., Kupiszewski, P., Kanji, Z. A., and Lohmann, U.: Ice nucleating particles in the Saharan Air Layer, *Atmos. Chem. Phys.*, 16, 9067-9087, 10.5194/acp-16-9067-2016, 2016b.
- Boucher, O., Randall, D., Artaxo, P., Bretherton, C., Feingold, G., Forster, P., Kerminen, V.-M., Kondo, Y., Liao, H., Lohmann, U., Rasch, P., Satheesh, S. K., Sherwood, S., Stevens, B., and Zhang, X. Y.: Clouds and Aerosols, in: *Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the*
- 25

- Intergovernmental Panel on Climate Change, edited by: Stocker, T. F., Qin, D., Plattner, G.-K., Tignor, M., Allen, S. K., Boschung, J., Nauels, A., Xia, Y., Bex, V., and Midgley, P. M., Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 571–658, 2013.
- 5 Brier, G. W., and Kline, D. B.: Ocean water as a source for ice nuclei, *Science*, 130, 717 - 718, 10.1126/science.130.3377.717, 1959.
- Bukowiecki, N., Weingartner, E., Gysel, M., Collaud Coen, M., Zieger, P., Herrmann, E., Steinbacher, M., Gägeler, H. W., and Baltensperger, U.: A Review of More than 20 Years of Aerosol Observation at the High Altitude Research Station Jungfraujoch, Switzerland (3580 m asl), *Aerosol Air Qual. Res.*, 16, 764-788, 10.4209/aaqr.2015.05.0305, 2016.
- 10 Burkert-Kohn, M., Wex, H., Welti, A., Hartmann, S., Grawe, S., Hellner, L., Herenz, P., Atkinson, J. D., Stratmann, F., and Kanji, Z. A.: Leipzig Ice Nucleation chamber Comparison (LINC): Inter-comparison of four online ice nucleation counters, *Atmos. Chem. Phys. Discuss.*, 2017, 1-30, 10.5194/acp-2017-358, 2017.
- Burrows, S. M., Hoose, C., Pöschl, U., and Lawrence, M. G.: Ice nuclei in marine air: biogenic particles or dust?, *Atmos. Chem. Phys.*, 13, 245-267, 10.5194/acp-13-245-2013, 2013.
- 15 Capo, R. C., Stewart, B. W., and Chadwick, O. A.: Strontium isotopes as tracers of ecosystem processes: theory and methods, *Geoderma*, 82, 197 - 225, 10.1016/S0016-7061(97)00102-X, 1998.
- Chou, C., Stetzer, O., Weingartner, E., Juranyi, Z., Kanji, Z. A., and Lohmann, U.: Ice nuclei properties within a Sahara dust event at the Jungfraujoch in the Swiss Alps, *Atmos. Chem. Phys.*, 11, 4725 - 4738, 10.5194/acp-11-4725-2011, 2011.
- 20 Collaud Coen, M., Weingartner, E., Schaub, D., Hueglin, C., Corrigan, C., Henning, S., Schwikowski, M., and Baltensperger, U.: Saharan dust events at the Jungfraujoch: detection by wavelength dependence of the single scattering albedo and first climatology analysis, *Atmos. Chem. Phys.*, 4, 2465-2480, 10.5194/acp-4-2465-2004, 2004.
- Collaud Coen, M., Weingartner, E., Furger, M., Nyeki, S., Prévôt, A. S. H., Steinbacher, M., and Baltensperger, U.: Aerosol climatology and planetary boundary influence at the Jungfraujoch analyzed by synoptic weather types, *Atmos. Chem. Phys.*, 11, 5931 - 5944, 10.5194/acp-11-5931-2011, 2011.

- Conen, F., Morris, C. E., Leifeld, J., Yakutin, M. V., and Alewell, C.: Biological residues define the ice nucleation properties of soil dust, *Atmos. Chem. Phys.*, 11, 9643-9648, 10.5194/acp-11-9643-2011, 2011.
- Conen, F., Rodríguez, S., Hüglin, C., Henne, S., Herrmann, E., Bukowiecki, N., and Alewell, C.: Atmospheric ice nuclei at the high-altitude observatory Jungfraujoch, Switzerland, *Tellus B*, 67, 10.3402/tellusb.v67.25014, 2015.
- 5 Cunliffe, M., Engel, A., Frka, S., Gašparovič, B., Guitart, C., Murrell, J. C., Salter, M., Stolle, C., Upstill-Goddard, R., and Wurl, O.: Sea surface microlayers: A unified physicochemical and biological perspective of the air–ocean interface, *Prog. Oceanogr.*, 109, 104 - 116, 10.1016/j.pocean.2012.08.004, 2013.
- Cziczo, D. J., and Abbatt, J. P. D.: Deliquescence, efflorescence, and supercooling of ammonium sulfate aerosols at low temperature: Implications for cirrus cloud formation and aerosol phase in the atmosphere, *J. Geophys. Res. Atmos.*, 104, 13781-10 13790, 10.1029/1999JD900112, 1999.
- Cziczo, D. J., Froyd, K. D., Hoose, C., Jensen, E. J., Diao, M. H., Zondlo, M. A., Smith, J. B., Twohy, C. H., and Murphy, D. M.: Clarifying the Dominant Sources and Mechanisms of Cirrus Cloud Formation, *Science*, 340, 1320 - 1324, 10.1126/science.1234145, 2013.
- Cziczo, D. J., and Froyd, K. D.: Sampling the composition of cirrus ice residuals, *Atmos. Res.*, 142, 15 - 31, 15 10.1016/j.atmosres.2013.06.012, 2014.
- de Leeuw, G., Andreas, E. L., Anguelova, M. D., Fairall, C. W., Lewis, E. R., O'Dowd, C., Schulz, M., and Schwartz, S. E.: Production flux of sea spray aerosol, *Rev. Geophys.*, 49, 10.1029/2010RG000349, 2011.
- DeLeon-Rodriguez, N., Latham, T. L., Rodriguez-R, L. M., Barazesh, J. M., Anderson, B. E., Beyersdorf, A. J., Ziemba, L. D., Bergin, M., Nenes, A., and Konstantinidis, K. T.: Microbiome of the upper troposphere: Species composition and 20 prevalence, effects of tropical storms, and atmospheric implications, *Proc. Natl. Acad. Sci.*, 110, 2575-2580, 10.1073/pnas.1212089110, 2013.
- DeMott, P. J., Cziczo, D. J., Prenni, A. J., Murphy, D. M., Kreidenweis, S. M., Thomson, D. S., Borys, R., and Rogers, D. C.: Measurements of the concentration and composition of nuclei for cirrus formation, *Proc. Natl. Acad. Sci.*, 100, 14655-14660, 10.1073/pnas.2532677100, 2003a.

- DeMott, P. J., Sassen, K., Poellot, M. R., Baumgardner, D., Rogers, D. C., Brooks, S. D., Prenni, A. J., and Kreidenweis, S. M.: African dust aerosols as atmospheric ice nuclei, *Geophys. Res. Lett.*, 30, 10.1029/2009GL037639, 2003b.
- DeMott, P. J., Prenni, A. J., Liu, X., Kreidenweis, S. M., Petters, M. D. a. T., C. H., Richardson, M. S., Eidhammer, T., and Rogers, D. C.: Predicting global atmospheric ice nuclei distributions and their impacts on climate, *Proc. Natl. Acad. Sci.*, 5 10.1073/pnas.0910818107, 2010.
- DeMott, P. J., Hill, T. C. J., McCluskey, C. S., Prather, K. A., Collins, D. B., Sullivan, R. C., Ruppel, M. J., Mason, R. H., Irish, V. E., Lee, T., Hwang, C. Y., Rhee, T. S., Snider, J. R., McMeeking, G. R., Dhaniyala, S., Lewis, E. R., Wentzell, J. J. B., Abbatt, J., Lee, C., Sultana, C. M., Ault, A. P., Axson, J. L., Diaz Martinez, M., Venero, I., Santos-Figueroa, G., Stokes, M. D., Deane, G. B., Mayol-Bracero, O. L., Grassian, V. H., Bertram, T. H., Bertram, A. K., Moffett, B. F., and Franc, G. D.:
10 Sea spray aerosol as a unique source of ice nucleating particles, *Proc. Natl. Acad. Sci.*, 113, 5797-5803, 10.1073/pnas.1514034112, 2016.
- Després, V., Huffman, J., Burrows, S., Hoose, C., Safatov, A., Buryak, G., Fröhlich-Nowoisky, J., Elbert, W., Andreae, M., Pöschl, U., and Jaenicke, R.: Primary biological aerosol particles in the atmosphere: a review, *Tellus B*, 64, 10.3402/tellusb.v64i0.15598, 2012.
- 15 Fröhlich-Nowoisky, J., Hill, T. C. J., Pummer, B. G., Yordanova, P., Franc, G. D., and Pöschl, U.: Ice nucleation activity in the widespread soil fungus *Mortierella alpina*, *Biogeosciences*, 12, 1057 -1071, 10.5194/bg-12-1057-2015, 2015.
- Gantt, B., and Meskhidze, N.: The physical and chemical characteristics of marine primary organic aerosol: a review, *Atmos. Chem. Phys.*, 13, 3979 -3996, 10.5194/acp-13-3979-2013, 2013.
- Griffiths, A. D., Parkes, S. D., Chambers, S. D., McCabe, M. F., and Williams, A. G.: Improved mixing height monitoring
20 through a combination of lidar and radon measurements, *Atmos. Meas. Tech.*, 6, 207 - 218, 10.5194/amt-6-207-2013, 2013.
- Griffiths, A. D., Conen, F., Weingartner, E., Zimmermann, L., Chambers, S. D., Williams, A. G., and Steinbacher, M.: Surface-to-mountaintop transport characterised by radon observations at the Jungfrauoch, *Atmos. Chem. Phys.*, 14, 12763-12779, 10.5194/acp-14-12763-2014, 2014.

- Harrison, A. D., Whale, T. F., Carpenter, M. A., Holden, M. A., Neve, L., O'Sullivan, D., Vergara Temprado, J., and Murray, B. J.: Not all feldspars are equal: a survey of ice nucleating properties across the feldspar group of minerals, *Atmos. Chem. Phys.*, 16, 10927-10940, 10.5194/acp-16-10927-2016, 2016.
- Herrmann, E., Weingartner, E., Henne, S., Vuilleumier, L., Bukowiecki, N., Steinbacher, M., Conen, F., Collaud Coen, M.,
5 Hammer, E., Jurányi, Z., Baltensperger, U., and Gysel, M.: Analysis of long-term aerosol size distribution data from Jungfraujoch with emphasis on free tropospheric conditions, cloud influence, and air mass transport, *J. Geophys. Res. Atmos.*, 120, 9459-9480, 10.1002/2015JD023660, 2015.
- Herut, B., Starinsky, A., and Katz, A.: Strontium in rainwater from Israel: Sources, isotopes and chemistry, *Earth Planet. Sc. Lett.*, 120, 77 - 84, 10.1016/0012-821X(93)90024-4, 1993.
- 10 Hodell, D. A., Mead, G. A., and Mueller, P. A.: Variation in the strontium isotopic composition of seawater (8 Ma to present) : Implications for chemical weathering rates and dissolved fluxes to the oceans, *Chemical Geology: Isotope Geoscience section*, 80, 291 - 307, 10.1016/0168-9622(90)90011-Z, 1990.
- Hoose, C., and Möhler, O.: Heterogeneous ice nucleation on atmospheric aerosols: a review of results from laboratory experiments, *Atmos. Chem. Phys.*, 12, 9817 - 9854, 10.5194/acp-12-9817-2012, 2012.
- 15 Kamphus, M., Ettner-Mahl, M., Klimach, T., Drewnick, F., Keller, L., Cziczo, D. J., Mertes, S., Borrmann, S., and Curtius, J.: Chemical composition of ambient aerosol, ice residues and cloud droplet residues in mixed-phase clouds: single particle analysis during the Cloud and Aerosol Characterization Experiment (CLACE 6), *Atmos. Chem. Phys.*, 10, 8077-8095, 10.5194/acp-10-8077-2010, 2010.
- 20 Kanji, Z. A., and Abbatt, J. P. D.: The University of Toronto Continuous Flow Diffusion Chamber (UT-CFDC): A Simple Design for Ice Nucleation Studies, *Aerosol Sci. Tech.*, 43, 730 - 738, 10.1080/02786820902889861, 2009.
- Kanji, Z. A., Ladino, L. A., Wex, H., Boose, Y., Burkert-Kohn, M., Cziczo, D. J., and Krämer, M.: Chapter 1: Overview of Ice Nucleating Particles, *Meteor. Mon.*, Early online release, 10.1175/amsmonographs-d-16-0006.1, 2017.
- Kaufmann, L., Marcolli, C., Hofer, J., Pinti, V., Hoyle, C. R., and Peter, T.: Ice nucleation efficiency of natural dust samples in the immersion mode, *Atmos. Chem. Phys.*, 16, 11177-11206, 10.5194/acp-16-11177-2016, 2016.

- Knopf, D. A., Alpert, P. A., Wang, B., and Aller, J. Y.: Stimulation of ice nucleation by marine diatoms, *Nature Geosci*, 4, 88-90, 10.1038/NGEO1037, 2011.
- Knopf, D. A., Alpert, P. A., Wang, B., O'Brien, R. E., Kelly, S. T., Laskin, A., Gilles, M. K., and Moffet, R. C.: Microspectroscopic imaging and characterization of individually identified ice nucleating particles from a case field study, *J. Geophys. Res. Atmos*, 119, 365-310,381, 10.1002/2014JD021866, 2014.
- 5 Koop, T., Luo, B., Tsias, A., and Peter, T.: Water activity as the determinant for homogeneous ice nucleation in aqueous solutions, *Nature*, 406, 611-614, 10.1038/35020537, 2000a.
- Koop, T., Kapilashrami, A., Molina, L. T., and Molina, M. J.: Phase transitions of sea-salt/water mixtures at low temperatures: Implications for ozone chemistry in the polar marine boundary layer, *J. Geophys. Res. Atmos.*, 105, 26393-26402, 10.1029/2000JD900413, 2000b.
- 10 Ladino, L. A., Yakobi-Hancock, J. D., Kilthau, W. P., Mason, R. H., Si, M., Li, J., Miller, L. A., Schiller, C. L., Huffman, J. A., Aller, J. Y., Knopf, D. A., Bertram, A. K., and Abbatt, J. P. D.: Addressing the ice nucleating abilities of marine aerosol: A combination of deposition mode laboratory and field measurements, *Atmos. Environ.*, 132, 1 - 10, 10.1016/j.atmosenv.2016.02.028, 2016.
- 15 Lloyd, G., Chouarton, T. W., Bower, K. N., Gallagher, M. W., Connolly, P. J., Flynn, M., Farrington, R., Crosier, J., Schlenczek, O., Fugal, J., and Henneberger, J.: The origins of ice crystals measured in mixed-phase clouds at the high-alpine site Jungfraujoch, *Atmos. Chem. Phys.*, 15, 12953-12969, 10.5194/acp-15-12953-2015, 2015.
- Lohmann, U., Lüönd, F., and Mahrt, F.: *An Introduction to Clouds: From the Microscale to Climate*, Cambridge University Press, Cambridge, 2016.
- 20 Lugauer, M., Baltensperger, U., Furger, M., Gäggeler, H. W., Jost, D. T., Schwikowski, M., and Wanner, H.: Aerosol transport to the high Alpine sites Jungfraujoch (3454ma.s.l.) and Colle Gnifetti (4452ma.s.l.), *Tellus B*, 50, 76 - 92, 10.1034/j.1600-0889.1998.00006.x, 1998.
- Mason, R. H., Chou, C., McCluskey, C. S., Levin, E. J. T., Schiller, C. L., Hill, T. C. J., Huffman, J. A., DeMott, P. J., and Bertram, A. K.: The micro-orifice uniform deposit impactor–droplet freezing technique (MOUDI-DFT) for measuring

- concentrations of ice nucleating particles as a function of size: improvements and initial validation, *Atmos. Meas. Tech.*, 8, 2449-2462, 10.5194/amt-8-2449-2015, 2015.
- Mason, R. H., Si, M., Chou, C., Irish, V. E., Dickie, R., Elizondo, P., Wong, R., Brintnell, M., Elsasser, M., Lassar, W. M., Pierce, K. M., Leaitch, W. R., MacDonald, A. M., Platt, A., Toom-Sauntry, D., Sarda-Estève, R., Schiller, C. L., Suski, K. J., Hill, T. C. J., Abbatt, J. P. D., Huffman, J. A., DeMott, P. J., and Bertram, A. K.: Size-resolved measurements of ice-nucleating particles at six locations in North America and one in Europe, *Atmos. Chem. Phys.*, 16, 1637-1651, 10.5194/acp-16-1637-2016, 2016.
- Meola, M., Lazzaro, A., and Zeyer, J.: Bacterial Composition and Survival on Sahara Dust Particles Transported to the European Alps, *Front. Microbiol.*, 6, 10.3389/fmicb.2015.01454, 2015.
- 10 Murray, B. J., O'Sullivan, D., Atkinson, J. D., and Webb, M. E.: Ice nucleation by particles immersed in supercooled cloud droplets, *Chemical Society Reviews*, 41, 6519-6554, 10.1039/C2CS35200A, 2012.
- Nyeki, S., Li, F., Weingartner, E., Streit, N., Colbeck, I., Gäggeler, H. W., and Baltensperger, U.: The background aerosol size distribution in the free troposphere: An analysis of the annual cycle at a high-alpine site, *J. Geophys. Res.*, 103, 31749-31761, 10.1029/1998JD200029, 1998.
- 15 O'Sullivan, D., Murray, B. J., Ross, J. F., Whale, T. F., Price, H. C., Atkinson, J. D., Umo, N. S., and Webb, M. E.: The relevance of nanoscale biological fragments for ice nucleation in clouds, *Sci. Rep.*, 5, 8082, 10.1038/srep08082, 2015.
- Pandey Deolal, S., Staehelin, J., Brunner, D., Cui, J., Steinbacher, M., Zellweger, C., Henne, S., and Vollmer, M. K.: Transport of PAN and NO_y from different source regions to the Swiss high alpine site Jungfrauoch, *Atmospheric Environment*, 64, 103-115, 10.1016/j.atmosenv.2012.08.021, 2013.
- 20 Pandey Deolal, S., Henne, S., Ries, L., Gilge, S., Weers, U., Steinbacher, M., Staehelin, J., and Peter, T.: Analysis of elevated springtime levels of Peroxyacetyl nitrate (PAN) at the high Alpine research sites Jungfrauoch and Zugspitze, *Atmos. Chem. Phys.*, 14, 12553-12571, 10.5194/acp-14-12553-2014, 2014.
- Prather, K. A., Bertram, T. H., Grassian, V. H., Deane, G. B., Stokes, M. D., DeMott, P. J., Aluwihare, L. I., Palenik, B. P., Azam, F., Seinfeld, J. H., Moffet, R. C., Molina, M. J., Cappa, C. D., Geiger, F. M., Roberts, G. C., Russell, L. M., Ault, A. P., Baltrusaitis, J., Collins, D. B., Corrigan, C. E., Cuadra-Rodriguez, L. A., Ebben, C. J., Forestieri, S. D., Guasco, T. L.,
- 25

- Hersey, S. P., Kim, M. J., Lambert, W. F., Modini, R. L., Mui, W., Pedler, B. E., Ruppel, M. J., Ryder, O. S., Schoepp, N. G., Sullivan, R. C., and Zhao, D.: Bringing the ocean into the laboratory to probe the chemical complexity of sea spray aerosol, *Proc. Natl. Acad. Sci.*, 110, 7550-7555, 2013.
- 5 Prenni, A., DeMott, P., Rogers, D., Kreidenweis, S. M., McFarquhar, G. M., Zhang, G., and Poellot, M. R.: Ice nuclei characteristics from M-PACE and their relation to ice formation in clouds, *Tellus B*, 61, 436-448, 10.1111/j.1600-0889.2009.00415.x, 2009.
- Pruppacher, H. R., and Klett, J. D.: *Microphysics of Clouds and Precipitation*, Kluwer Acad. Norwell, Mass, 1997.
- Pummer, B. G., Bauer, H., Bernardi, J., Bleicher, S., and Grothe, H.: Suspendable macromolecules are responsible for ice nucleation activity of birch and conifer pollen, *Atmos. Chem. Phys.*, 12, 2541-2550, 10.5194/acp-12-2541-2012, 2012.
- 10 Rogers, D. C.: Development of a continuous flow thermal gradient diffusion chamber for ice nucleation studies, *Atmos. Res.*, 22, 149 - 181, 10.1016/0169-8095(88)90005-1, 1988.
- Rogers, D. C., DeMott, P. J., Kreidenweis, S. M., and Chen, Y.: Measurements of ice nucleating aerosols during SUCCESS, *Geophys. Res. Lett.*, 25, 1383-1386, 10.1029/97GL03478, 1998.
- Rogers, D. C., DeMott, P. J., Kreidenweis, S. M., and Chen, Y.: A Continuous-Flow Diffusion Chamber for Airborne
15 Measurements of Ice Nuclei, *J. Atmos. Oceanic Technol.*, 18, 725-741, 10.1175/1520-0426(2001)018<0725:ACFDCF>2.0.CO;2, 2001.
- Rogers, R. R., and Yau, M. K.: *A Short Course in Cloud Physics*, Pergamon, 1989.
- Santachiara, G., Matteo, L. D., Prodi, F., and Belosi, F.: Atmospheric particles acting as Ice Forming Nuclei in different size ranges, *Atmos. Res.*, 96, 266 - 272, 10.1016/j.atmosres.2009.08.004, 2010.
- 20 Schrod, J., Weber, D., Drücke, J., Keleshis, C., Pikridas, M., Ebert, M., Cvetkovic, B., Nickovic, S., Marinou, E., Baars, H., Ansmann, A., Vrekoussis, M., Mihalopoulos, N., Sciare, J., Curtius, J., and Bingemer, H. G.: Ice nucleating particles over the Eastern Mediterranean measured by unmanned aircraft systems, *Atmos. Chem. Phys.*, 17, 4817-4835, 10.5194/acp-17-4817-2017, 2017.

- Sjogren, S., Gysel, M., Weingartner, E., Alfarra, M. R., Duplissy, J., Cozic, J., Crosier, J., Coe, H., and Baltensperger, U.: Hygroscopicity of the submicrometer aerosol at the high-alpine site Jungfraujoch, 3580 m a.s.l., Switzerland, *Atmos. Chem. Phys.*, 8, 5715-5729, 10.5194/acp-8-5715-2008, 2008.
- Stein, M., Starinsky, A., Katz, A., Goldstein, S. L., Machlus, M., and Schramm, A.: Strontium isotopic, chemical, and sedimentological evidence for the evolution of Lake Lisan and the Dead Sea, *Geochim. Cosmochim. Ac.*, 61, 3975 - 3992, 10.1016/S0016-7037(97)00191-9, 1997.
- Steinbacher, M., Wyss, S., Emmenegger, L., and Hüglin, C.: National Air Pollution Monitoring Network (NABEL), International Foundation HFJG, annual report, 2017.
- Stohl, A., Forster, C., Frank, A., Seibert, P., and Wotawa, G.: Technical note: The Lagrangian particle dispersion model FLEXPART version 6.2, *Atmos. Chem. Phys.*, 5, 2461-2474, 10.5194/acp-5-2461-2005, 2005.
- Stopelli, E., Conen, F., Morris, C. E., Herrmann, E., Bukowiecki, N., and Alewell, C.: Ice nucleation active particles are efficiently removed by precipitating clouds, 5, 16433, 10.1038/srep16433, 2015.
- Sturm, P., Tuzson, B., Henne, S., and Emmenegger, L.: Tracking isotopic signatures of CO₂ at the high altitude site Jungfraujoch with laser spectroscopy: analytical improvements and representative results, *Atmos. Meas. Tech.*, 6, 1659-1671, 10.5194/amt-6-1659-2013, 2013.
- Vali, G., DeMott, P. J., Möhler, O., and Whale, T. F.: Technical Note: A proposal for ice nucleation terminology, *Atmos. Chem. Phys.*, 15, 10263-10270, 10.5194/acp-15-10263-2015, 2015.
- Vergara-Temprado, J., Murray, B. J., Wilson, T. W., O'Sullivan, D., Browse, J., Pringle, K. J., Ardon-Dryer, K., Bertram, A. K., Burrows, S. M., Ceburnis, D., DeMott, P. J., Mason, R. H., O'Dowd, C. D., Rinaldi, M., and Carslaw, K. S.: Contribution of feldspar and marine organic aerosols to global ice nucleating particle concentrations, *Atmos. Chem. Phys.*, 17, 3637-3658, 10.5194/acp-17-3637-2017, 2017.
- von der Weiden, S.-L., Drewnick, F., and Borrmann, S.: Particle Loss Calculator – a new software tool for the assessment of the performance of aerosol inlet systems, *Atmos. Meas. Tech.*, 2, 479-494, 10.5194/amt-2-479-2009, 2009.

- Weingartner, E., Nyeki, S., and Baltensperger, U.: Seasonal and diurnal variation of aerosol size distributions ($10 < D < 750$ nm) at a high-alpine site (Jungfraujoch 3580 m asl), *J. Geophys. Res.*, 104, 26809-26820, 10.1029/1999JD900170, 1999.
- Welti, A., Kanji, Z. A., Lüönd, F., Stetzer, O., and Lohmann, U.: Exploring the mechanisms of ice nucleation on kaolinite: from deposition nucleation to condensation freezing, *J. Atmos. Sci.*, 71, 16-36, 10.1175/JAS-D-12-0252.1, 2014.
- 5 Wernli, B. H., and Davies, H. C.: A lagrangian-based analysis of extratropical cyclones. I: The method and some applications, *Quart. J. Roy. Meteor. Soc.*, 123, 467-489, 10.1256/smsqj.53810, 1997.
- Wex, H., DeMott, P. J., Tobo, Y., Hartmann, S., Rösch, M., Clauss, T., Tomsche, L., Niedermeier, D., and Stratmann, F.: Kaolinite particles as ice nuclei: learning from the use of different kaolinite samples and different coatings, *Atmos. Chem. Phys.*, 14, 5529-5546, 10.5194/acp-14-5529-2014, 2014.
- 10 Wilson, T. W., Ladino, L. A., Alpert, P. A., Breckels, M. N., Brooks, I. M., Browse, J., Burrows, S. M., Carslaw, K. S., Huffman, J. A., Judd, C., Kilthau, W. P., Mason, R. H., McFiggans, G., Miller, L. A., Najera, J. J., Polishchuk, E., Rae, S., Schiller, C. L., Si, M., Temprado, J. V., Whale, T. F., Wong, J. P. S., Wurl, O., Yakobi-Hancock, J. D., Abbatt, J. P. D., Aller, J. Y., Bertram, A. K., Knopf, D. A., and Murray, B. J.: A marine biogenic source of atmospheric ice-nucleating particles, *Nature*, 525, 234-238, 10.1038/nature14986, 2015.
- 15 Yakobi-Hancock, J. D., Ladino, L. A., and Abbatt, J. P. D.: Feldspar minerals as efficient deposition ice nuclei, *Atmos. Chem. Phys.*, 13, 11175-11185, 10.5194/acp-13-11175-2013, 2013.
- Yun, Y., and Penner, J. E.: An evaluation of the potential radiative forcing and climatic impact of marine organic aerosols as heterogeneous ice nuclei, *Geophys. Res. Lett.*, 40, 4121-4126, 10.1002/grl.50794, 2013.
- Zanis, P., Ganser, A., Zellweger, C., Henne, S., Steinbacher, M., and Staehelin, J.: Seasonal variability of measured ozone
20 production efficiencies in the lower free troposphere of Central Europe, *Atmos. Chem. Phys.*, 7, 223-236, 10.5194/acp-7-223-2007, 2007.
- Zellweger, C., Forrer, J., Hofer, P., Nyeki, S., Schwarzenbach, B., Weingartner, E., Ammann, M., and Baltensperger, U.: Partitioning of reactive nitrogen (NO_y) and dependence on meteorological conditions in the lower free troposphere, *Atmos. Chem. Phys.*, 3, 779 - 796, 10.5194/acp-3-779-2003, 2003.

Zipori, A., Rosenfeld, D., Shpund, J., Steinberg, D. M., and Erel, Y.: Targeting and impacts of AgI cloud seeding based on rain chemical composition and cloud top phase characterization, *Atmos. Res.*, 114–115, 119 - 130, 10.1016/j.atmosres.2012.05.023, 2012.

Zipori, A., Rosenfeld, D., Tirosh, O., Teutsch, N., and Erel, Y.: Effects of aerosol sources and chemical compositions on cloud drop sizes and glaciation temperatures, *J. Geophys. Res. Atmos.*, 120, 9653-9669, 10.1002/2015JD023270, 2015.

Zolles, T., Burkart, J., Häusler, T., Pummer, B., Hitzemberger, R., and Grother, H.: Identification of Ice Nucleation Active Sites on Feldspar Dust Particles, *J. Phys. Chem. A*, 119, 2692-2700, 10.1021/jp509839x, 2015.