



32 investigation shows that these simulated episodic NO_x spikes are always associated with
33 transport events from Europe, but the exact cause remains unclear. The model systematically
34 overestimates C_2H_6 mixing ratios by approximately 20% relative to observations. This
35 discrepancy can be resolved by decreasing anthropogenic C_2H_6 emissions over Asia and the US
36 by $\sim 20\%$, from 5.4 to 4.4 Tg/yr. GEOS-Chem is able to reproduce the seasonal variability of O_3
37 and its spring maximum. However, compared with observations, it underestimates surface O_3 by
38 approximately 13% (6.5 ppbv) from April to July. This low bias appears to be driven by several
39 factors including missing snowpack emissions for NO_x and nitrous acid, the coarse model
40 resolution, model overestimated O_3 dry deposition velocity during springtime, as well as the
41 uncertainties in the stratosphere-to-troposphere exchange scheme for O_3 .

42 1. Introduction

43 Ozone (O_3) and its precursors (e.g., $\text{NO}_x = \text{NO} + \text{NO}_2$ and volatile organic compounds) are
44 important atmospheric species affecting both air quality and climate (e.g., Jacob et al., 1992;
45 Fiore et al., 2002; Unger et al., 2006). Tropospheric O_3 is a potent greenhouse gas and it also has
46 detrimental effects on human health and vegetation (Knowlton et al., 2004; Hollaway et al.,
47 2012; Yue and Unger, 2014). NO_x is an important precursor for O_3 production and peroxyacetyl
48 nitrate (PAN), which serves as a reservoir for NO_x . PAN, O_3 , as well as some of their precursors,
49 have relatively long lifetimes in the atmosphere, enabling them to be transported long distance to
50 remote regions such as the Arctic.

51 Recent studies have shown some significant challenges for atmospheric chemical transport
52 models to simulate O_3 and its precursors in the Arctic (e.g., Shindell et al., 2008; Alvarado et al.,
53 2010; Walker et al., 2012; Wespes et al., 2012; Fischer et al., 2014; Monks et al., 2015). In the
54 multi-model assessment by Shindell et al. (2008), more than a dozen models all showed
55 systematic and persistent underestimation of O_3 at the GEOSummit station, Greenland (hereafter
56 referred to as Summit). Alvarado et al. (2010) used NO_x and PAN measurements from ARCTAS
57 (Arctic Research of the Composition of the Troposphere from Aircraft and Satellites) in the
58 summer to compare with model simulations. They found that model simulated NO_x mixing ratios
59 were higher than observations, while PAN mixing ratios were lower in fresh boreal fire plumes.
60 In terms of global PAN simulations, Fischer et al. (2014) directly partitioned 40% of NO_x
61 emissions from wildfires to PAN formation, which improved the agreement between model and



62 observations. However, the model still underestimated PAN surface mixing ratios during
63 springtime in the Arctic. Walker et al. (2012) reported that model simulated O₃ mixing ratios
64 were biased low when compared with balloon data during summertime from two high-latitude
65 sites at Eureka (80°N, 86°W) and Ny-Ålesund (79°, 12°E). Wespes et al. (2012) also revealed
66 that model simulated O₃ mixing ratios below the boundary layer in the Arctic are significantly
67 underestimated during spring-summer, compared with ARCTAS measurements. More recently,
68 Monks et al. (2015) further demonstrated that model simulated O₃ mixing ratios in the Arctic at
69 the surface and in the upper troposphere were generally lower than the observations.

70 Global anthropogenic ethane (C₂H₆) emission estimates range from 5.7 Tg/yr to 16.2 Tg/yr
71 (Blake and Rowland, 1986; Kanakidou et al., 1991; Rudolph, 1995; Gutpa et al., 1998; Xiao et
72 al., 2008; Etiope and Ciccioli, 2009; Pozzer et al., 2010; Aydin et al., 2011; Simpson et al., 2012;
73 Emmons et al., 2015; Franco et al., 2016; Tzompa-Sosa et al., 2017), with a decreasing trend
74 from 1980 to 2009 (Simpson et al., 2012; Helmig et al., 2014a). However, since 2009, global
75 anthropogenic C₂H₆ emissions began to increase (Franco et al., 2015; Hausmann et al., 2016;
76 Helmig et al., 2016). The RETRO (REanalysis of the TROpospheric chemical composition)
77 global emission inventory used to be the global default anthropogenic C₂H₆ emission inventory,
78 the annual budget of which has been shown too low compared with observations (Xiao et al.,
79 2008; Fischer et al., 2014; Franco et al., 2015, 2016), whereas the emission inventory from Xiao
80 et al. (2008) has been demonstrated to match observations during 1988-2004. Model simulated
81 C₂H₆ mixing ratios are particularly biased low in the remote high latitude regions, when
82 compared with observations (Emmons et al., 2015).

83 Field measurements at Summit show that snowpack emits gas-phase NO_x, PAN, nitrous acid
84 (HONO), as well as hydrogen peroxide (H₂O₂) during spring-summer, when polar sun rises
85 (Ford et al., 2002; Honrath et al., 2002). Although several 1-D models (Thomas et al., 2011,
86 2012; Frey et al., 2013; Murray et al., 2015) have validated its significant importance for surface
87 NO_x as well as O₃ formation, current global chemical transport models (CTMs) usually do not
88 include this emission source (Zatko et al., 2016).

89 In this study, we employ a global chemical transport model, GEOS-Chem CTM, to evaluate the
90 model performance for surface O₃ and its precursors over Summit, in conjunction with two years
91 in-situ measurements during 2008-2010. This paper is organized as follows: section 2 describes



92 model methods and observations, followed by detailed comparisons of model simulations against
93 observations for O₃ and O₃ precursors in section 3; conclusions are summarized in section 4.

94 **2. Observational data and model simulations**

95 In situ measurements of NO_x, PAN, and non-methane hydrocarbons (NMHCs) were performed
96 at Summit from July 2008 to June 2010 (Helmig et al., 2014b; Kramer et al., 2015). An
97 automated O₃ chemiluminescence detection system was used to measure NO_x (Ridley and
98 Grahek, 1990); a commercial PAN gas chromatography analyzer (PAN-GC, Metcon, In.,
99 Boulder, CO) was employed for the measurement of PAN. Measurements of NMHC relied on an
100 automated GC-Flame Ionization Detection (FID) system. Readers are referred to Kramer et al.
101 (2015) and Helmig et al. (2014b) for the details of the measurement techniques and equipment
102 setup. Surface measurements of O₃ using ultraviolet light absorption at 254 nm
103 (Petrovavlovskikh and Oltmans, 2012), and CO by GC (Novellie and Masarie, 2015) are from
104 the National Oceanic and Atmospheric Administration (NOAA). Hourly averaged O₃ and flask
105 sampled CO between July 2008 and June 2010 were downloaded from the NOAA Earth System
106 Research Laboratory (ESRL) Global Monitoring Division (GMD) website
107 (<http://www.esrl.noaa.gov/gmd/dv/data/>). Vertical ozonesonde data profiles were also
108 downloaded from NOAA ESRL GMD (McClure-Begley et al., 2014).

109 The GEOS-Chem CTM (Bey et al., 2001) was used to simulate the seasonal cycles of O₃ and
110 related species (NO_x, PAN, NMHCs) at Summit. The GEOS-Chem model has fully coupled O₃-
111 NO_x-VOC-Aerosol chemistry mechanism and is driven by assimilated meteorological data from
112 the Goddard Earth Observing System version 5 (GEOS-5) of the NASA Global Modeling
113 Assimilation Office. The GEOS-Chem model has been extensively evaluated and applied in a
114 wide range of applications (Martin et al., 2002; Park et al., 2004; Wu et al., 2007; Hudman et al.,
115 2009; Johnson et al., 2010; Huang et al., 2013; Kumar et al., 2013; Zhang et al., 2014; Hickman
116 et al., 2017). GEOS-Chem v10-1 with grid resolution of 4° latitude by 5° longitude, and 47
117 vertical layers was used for the model control simulation. Following McLinden et al. (2000), the
118 Linoz stratospheric O₃ chemistry scheme was used. The simulation was run from June 2007 to
119 June 2010 and the results from the last two years were used in the final analysis. Time series data
120 were archived with 3-hr temporal resolution at the Summit grid box.



121 Global anthropogenic emissions of NO_x , SO_2 , NH_3 , and CO in the model are based on the
122 Emission Database for Global Atmospheric Research (EDGAR) v4.2 inventory, which is
123 overwritten by regional emission inventories where applicable, such as the BRAVO inventory
124 for Mexico (Kuhns et al., 2005), the CAC over Canada, the EMEP emissions over Europe, the
125 Model Inter-comparison Study for Asia Phase III (MIX) emissions over Asia (Li et al., 2017),
126 and the US EPA NEI 2011 (NEI11) emission inventory (Simon et al., 2010). Soil NO_x emission
127 scheme follows Hudman et al. (2012). Lightning NO_x emissions are calculated per flash rate
128 based on GEOS-5 computed cloud-top heights (Price and Rind, 1992), which are determined by
129 deep convection and constrained by satellite observations for monthly average flash rates from
130 the Lightning Imaging Sensor and Optical Transient Detector (OTD/LIS) (Sauvage et al., 2007;
131 Murray et al., 2012). Biomass burning emissions are from the Global Fire Emission Database
132 version 4 (GFED4) inventory with monthly resolution (Giglio et al., 2013). The RETRO global
133 anthropogenic NMHC emission inventory (van het Bolscher et al., 2008) was used except for
134 C_2H_6 and propane (C_3H_8), which follows Xiao et al. (2008, hereafter referred to as X08) for the
135 year 2001. Global biofuel emission inventory follows Yevich and Logan (2003), which includes
136 emissions for C_2H_6 and C_3H_8 . For biogenic VOC emissions, the Model of Emissions of Gases
137 and Aerosols from Nature (MEGAN) scheme (Guenther et al., 2006) was used. Dry deposition
138 of species in GEOS-Chem uses a standard resistance-in-series scheme (Wesely, 1989), as
139 implemented in Wang et al. (1998). Wet scavenging follows Liu et al. (2001), including
140 scavenging in convective updraft, rainout (in-cloud) and washout (below-cloud) from convective
141 anvils and large-scale precipitation.

142 We first ran the standard GEOS-Chem model with a-priori emissions and compare the
143 simulation results against observations for various species (including NO_x , PAN, C_2H_6 , C_3H_8 ,
144 CO , and O_3 , as shown in Fig. 1). Then we focus on the model-observation discrepancies, and
145 where applicable, make revisions to the model simulations and further evaluate the improvement
146 in model performance, as discussed in details below.

147 **3. Results and Discussions**

148 3.1 NO_x



149 We first combine the two years of data for July 2008 – June 2010 and analyze their seasonal
150 variations. As shown in Figure 1a, the GEOS-Chem model in general captures the abundance
151 and seasonal variation of NO_x for July-October. However, compared to observations the model
152 results significantly (by a factor of 2) overestimate NO_x mixing ratios for November-January,
153 while underestimating the data in spring and early summer by over 50%. Another challenge for
154 the model simulation is that it does not capture the decrease of NO_x for May-December. We find
155 that during the 2009-2010 winter season, model simulations show several high NO_x spikes with
156 peak NO_x mixing ratios reaching ~ 0.15 ppbv or higher, which is ~ 15 times greater than typical
157 background levels (Fig. 2). These large peaks in NO_x were not observed in the data. Similar peaks
158 were also seen in the model simulations during the 2008-2009 winter season; however, there are
159 no measurement data available for this period to compare with.

160 Further analyses show that the model-simulated high NO_x spikes during wintertime are all
161 associated with transport events from Europe. We carried out a sensitivity study to examine the
162 impacts of European emissions on Arctic NO_x by manually reducing anthropogenic NO_x
163 emissions from the EMEP emission inventory over Europe by 50% (EMEP50). Results show
164 that surface peak NO_x mixing ratios over Summit during the spike events (e.g., dates around
165 12/09/2009, 12/15/2009, 1/15/2010 and 1/22/2010) from EMEP50 almost decline proportionally
166 by approximately 50% during 2009/12/01-2010/01/31 (Fig. 2), which confirms that the modeled
167 NO_x spikes at Summit during wintertime are associated with transport from Europe. However,
168 the model simulated NO_x is still significantly higher than observations. Comparisons for surface
169 NO₂ mixing ratios between model simulations and 11 in-situ observational sites over Europe
170 during this period were conducted with data downloaded from <http://ebas.nilu.no>. For detailed
171 site information, NO₂ measurement technique and resolution, refer to Table 1. Measurement data
172 over these two months for each site were averaged to compare with the corresponding grid cell in
173 the model. As shown in Figure 3a, GEOS-Chem overestimates surface NO₂ mixing ratios at
174 these sites by over 66%, compared with observations.

175 Instead of using EMEP, we carried out another sensitivity study to force anthropogenic NO_x
176 emissions over Europe following EDGAR v4.2 (EURO_EDGAR), with other model
177 configurations identical to control simulations. As shown in Figure 2, the NO_x mixing ratios over
178 Summit during 12/2009-01/2010 agree much better with observations, especially for January



179 2010 where the model captures the magnitudes of observational peaks (Fig. 2). This is because
180 NO_x emissions from EDGAR over Europe (1.97 Tg NO) are 12% lower than that from EMEP
181 (2.24 Tg NO) for the months of 12/2009 and 01/2010. Furthermore, the discrepancy for the
182 differences of surface NO₂ mixing ratios over Europe between EURO_EDGAR and observations
183 is further reduced (by 50%), relative to the control runs (Fig. 3b). Similarly, we also tested the
184 sensitivity of surface NO_x mixing ratios over Summit in response to the changes in the
185 anthropogenic NO_x emissions from NEI11 over US and MIX over Asia (including Siberia)
186 during these two months, and found that surface NO_x mixing ratio over Summit during these two
187 months were quite close to the control simulations (not shown), reflecting insensitivity to
188 emission perturbations from US and Asia. Therefore, we conclude that uncertainties in fossil fuel
189 NO_x emissions of EMEP associated with transport events from Europe in the model are the most
190 likely cause for the wintertime NO_x spikes over Summit.

191 For April-July, model simulated monthly mean NO_x mixing ratios over Summit are a factor of
192 two lower than the observations (Fig. 4a). In-situ measurements at Summit by Honrath et al.
193 (1999, 2000a, 2000b, 2002) showed upward fluxes of NO_x (2.52×10^8 molecules cm⁻² s⁻¹) from
194 photolysis of nitrate in snowpack during the summertime, leading to enhancement in NO_x levels
195 in the surface layer by approximately 20 pptv, which is comparable to surface NO_x mixing ratios
196 in the Arctic from other sources. Similar results were found over the East Antarctic Plateau, a
197 remote Antarctic location (75.1° S, 123.3° E) covered by snow/ice sheet (Frey et al., 2013;
198 Legrand et al., 2014). The standard GEOS-Chem model does not include the photolysis of nitrate
199 from snowpack, implying a missing source for NO_x in the Arctic/Antarctic boundary layer.

200 In order to test the sensitivity of model simulated surface NO_x mixing ratios to the snowpack
201 emissions, we implement in the model a constant 24-hr NO_x flux $\sim 2.52 \times 10^8$ molecules cm⁻² s⁻¹
202 during April-July over Greenland (60-85° N, 20-60° W), following the measurements conducted
203 at Summit during summertime by Honrath et al. (2002). As a result, we find that on average, the
204 model simulated surface NO_x mixing ratios for April to July over Summit are more than double
205 that from the control simulation, which improves the agreement between model and observations
206 (Fig. 4a). However, the model is still not able to reproduce the decreasing trends of NO_x for
207 May-October. This may be caused by the seasonally decreasing NO_x production rate in the



208 snowpack from spring to fall from a gradual depletion of the snowpack NO_x reservoir (Van Dam
209 et al., 2015), whereas the model considers a simple constant NO_x emission flux.

210 3.2 PAN

211 Figure 1b shows the comparison of model simulated monthly mean PAN mixing ratios with the
212 measurement data. The model captures the seasonal variation of PAN well, although
213 significantly (by ~30%) underestimating the PAN mixing ratios for April-June. By running the
214 model simulation with higher horizontal resolution at 2° latitude by 2.5° longitude (hereafter
215 referred to as GEOS-Chem 2x2.5), we find that the monthly mean PAN mixing ratios over
216 Summit during April-July are increased by up to 23.3 pptv compared to the 4x5 simulation (Fig.
217 4b). This can be explained by two reasons. First, coarse model resolution (e.g., 4x5 horizontal
218 resolution) could artificially smear the intense emission sources throughout the entire grid cell
219 (e.g., over urban regions), leading to underestimates of downwind concentrations for species,
220 e.g., O₃ and O₃ precursors (Jang et al., 1995; Yu et al., 2016). Second, ventilation of lower
221 atmosphere could be better resolved by fine model resolution, leading to more efficient vertical
222 advection (Wang et al., 2004; Chen et al., 2009; Yu et al., 2016). However, on average, monthly
223 mean model simulated PAN mixing ratios are still underestimated by 20% during this period,
224 compared with observations.

225 Snowpack can emit not only NO_x, but also PAN, based on field studies at Summit during
226 summertime by Ford et al. (2002). GEOS-Chem does not contain snowpack PAN emissions and
227 chemistry. For a sensitivity study, similar to snowpack NO_x emissions as discussed in section
228 3.1, we considered a 24-hr constant flux of 2.52×10^8 molecules cm⁻² s⁻¹ of PAN from April to
229 July, following Ford et al. (2002). As a result, model simulated PAN mixing ratios agree much
230 better with observations (Fig. 4b). Note that there are also other possible reasons that lead to
231 model bias. For instance, a study by Fischer et al. (2014) showed that an improved agreement
232 between modeled and measured PAN in the high latitudes can be found when the model was
233 forced to emit a portion of the fire emissions above the boundary layer as well as by directly
234 partitioning 40% of NO_x emissions from fires into PAN. However, in our case, we did not find
235 much difference between a sensitivity study following this method and our control runs.

236 3.3 NMHC



237 Comparisons of observed surface C_2H_6 and C_3H_8 mixing ratios with GEOS-Chem simulations at
238 Summit are shown in Figures 1c and d. The model simulations agree well with surface
239 measurements of C_3H_8 , but systematically overestimate C_2H_6 (by approximately 25% annually),
240 with the largest bias (0.48 ppbv) occurring during summer. This is consistent with the study from
241 Tzompa-Sosa et al., (2017), which used the same model as our study and pointed out that using
242 X08 as global anthropogenic C_2H_6 emission inventory systematically overestimated surface C_2H_6
243 mixing ratios over the Northern Hemisphere, compared with ground-based observations.
244 Anthropogenic C_2H_6 emissions over US from NEI11 are shown to geographically match the
245 distribution of active oil and natural wells (Tzompa-Sosa et al., 2017), and the most recent MIX
246 has been updated to synergize anthropogenic C_2H_6 emissions from various countries over Asia
247 (Li et al., 2017). Therefore, instead of using global anthropogenic fossil fuel emissions of C_2H_6
248 following X08, we first conducted sensitivity simulations by overwriting global emission
249 inventories by NEI11 over US, and MIX over Asia (hereafter referred to as NEI11_MIX). Both
250 NEI11 and MIX contain emissions for the years from 2008 to 2010, which could realistically
251 represent the annual and seasonal variations of C_2H_6 emissions over the US and Asia, thus
252 spatially and temporally better representative of anthropogenic C_2H_6 emissions from mid-
253 latitudes transported to the Arctic regions. In general, model control simulations overestimate
254 annual mean surface C_2H_6 mixing ratios primarily in the Northern Hemisphere, with large
255 differences occurring over Asia and US by up to 5 ppbv, compared with NEI11_MIX during the
256 period of 07/2008-06/2010 (Fig. S1). All the above changes are driven by the substantial
257 reductions of anthropogenic C_2H_6 emissions between emission inventories, from 3.5 (X08) to 2.5
258 Tg/yr (MIX) over Asia, and from 1.9 Tg/yr (X08) to 1.4 Tg/yr (NEI11) over US, reflecting
259 decreasing trend of anthropogenic C_2H_6 emissions during 2001-2009 because X08 emission
260 inventory is based on the year 2001, which is consistent with Helmig et al. (2014a). Substantial
261 changes in surface C_2H_6 mixing ratios over the US between control simulations and NEI11_MIX
262 reflects that there exist temporal changes of C_2H_6 emissions from oil and gas productions
263 during the period of 2001-2009. A similar pattern was also found by Tzompa-Sosa et al. (2017).
264 In contrast to the control simulations, NEI11_MIX model simulations show that monthly mean
265 C_2H_6 mixing ratios over Summit are systematically underestimated by 24%, compared with
266 observations (Fig. 5). Tzompa-Sosa et al. (2017) reported that NEI11 for C_2H_6 emissions were
267 likely underestimated by 40%, compared with in-situ and aircraft observations over the US. With



268 NEI11 C₂H₆ emissions increases by 40%, however, model simulated annual mean C₂H₆ mixing
269 ratios over Summit only increase by 6% during the period of 07/2008-06/2010, relative to
270 NEI11_MIX.

271 Similar to NEI11_MIX, we further conducted sensitivity studies by only replacing the regional
272 emission inventory for C₂H₆ over the US, with other regions still following X08 (hereafter
273 referred to as NEI11_ONLY). Consequently, model simulated surface C₂H₆ mixing ratios over
274 Summit agree better with observations during winter-spring (Fig. 5), decreasing the bias from
275 +15% (control simulations) to +6%. However, model simulated C₂H₆ mixing ratios during
276 summer-fall are higher than the observations by over 30%.

277 We then scaled up the MIX emissions for C₂H₆ by 20% over Asia, with other model
278 configurations identical to NEI11_MIX (hereafter referred to as NEI11_MIX20). By doing this,
279 we increase fossil fuel C₂H₆ emissions from 2.5 to 3 Tg/yr. We find that annual mean C₂H₆
280 mixing ratios from NEI11_MIX20 agree quite well with observations over Summit, with bias
281 less than 1% (Fig. 5). This implies that further assessments of anthropogenic C₂H₆ emissions
282 from MIX over Asia are needed and a more accurate global anthropogenic C₂H₆ emission
283 inventory should be developed and validated to replace X08 in the future. Note that this standard
284 version of GEOS-Chem does not account for the sink of C₂H₆ from the reaction with chlorine,
285 which could reduce the global annual mean surface C₂H₆ mixing ratios by 0-30%, and the global
286 burden of C₂H₆ by about 20%, compared with the simulation without considering the halogen
287 chemistry (Sherwen et al., 2016), which introduces additional uncertainty for our measurement-
288 model comparison, together with the highly uncertain seasonality of C₂H₆ chemistry.

289 3.4 CO

290 Figure 1e shows the comparison of model simulated CO mixing ratios with observations over
291 Summit. Overall, the model generally captures the seasonal trend and annual mean of CO, with
292 annual mean model simulated CO mixing ratios slightly overestimated by up to 3 ppbv,
293 compared with observations.

294 3.5 O₃



295 Surface O₃ mixing ratios from model simulations and surface observations are compared in
296 Figure 1f. The GEOS-Chem model captures the seasonal variation of O₃ including the spring
297 peak. However, the model shows a systematic low bias for most time of the year, in particular for
298 April–July when the surface O₃ mixing ratios are underestimated by approximately 13% (~ 6.5
299 ppbv). Here we focus our analysis for the possible causes that lead to the model low bias during
300 April–July.

301 As discussed earlier, snowpack emissions due to the photolysis of nitrate in the snow during late
302 spring and summer could contribute to NO_x and HONO levels in the ambient air which could
303 significantly enhance O₃ production (Crawford et al., 2001; Zhou et al., 2001; Dibb et al., 2002;
304 Honrath et al., 2002; Yang et al., 2002; Grannas et al., 2007; Helmig et al., 2008; Legrand et al.,
305 2014). We ran a sensitivity study to test the response of surface O₃ mixing ratios to the
306 perturbations of NO_x and HONO from snowpack emissions. In addition to snowpack NO_x
307 emissions that are described in Section 3.1, we implemented in the model a constant flux of
308 HONO (4.64×10^7 molecules cm⁻² s⁻¹) from April to July over Summit as well (Honrath et al.,
309 2002). As a result, monthly mean model simulated surface O₃ mixing ratios increase by up to 3
310 ppbv during this period (Fig. 6). The largest effect occurs in July due to relatively strong solar
311 radiation. O₃ formation due to snowpack emissions in our study is slightly higher than that in
312 Zatkan et al. (2016) because HONO from snowpack emissions is not considered in their study.
313 However, for the months of April and May, surface O₃ mixing ratios only increase by ~ 1 ppbv,
314 compared with the control runs. That is, even after accounting for the snowpack emissions, the
315 model simulated O₃ mixing ratios are still significantly lower than the observations.

316 Comparison of the model simulations with different resolutions (4x5 vs. 2x2.5) shows that the
317 finer resolution simulations substantially increase monthly mean O₃ mixing ratios over Summit
318 by up to 6 ppbv for the months of June and July (Fig. 6). As discussed in section 3.2, fine model
319 resolution can better resolve the emission strengths, which could significantly affect downwind
320 chemical reactions, e.g., O₃ production efficiency (Liang and Jacobson, 2000). Moreover, terrain
321 elevations from fine model resolution are better represented (thus better representative of
322 Summit's elevation) and more efficient vertical ventilation of O₃ and O₃ precursors can be
323 achieved (Wang et al., 2004). Together with the impact of snowpack chemistry, this brings
324 model simulated surface O₃ mixing ratios over Summit in much better agreement with



325 observations for these two months. Unfortunately, there is still a low bias in the model for the
326 months of April and May.

327 Another possible cause for the O₃ biases between model simulations and observations is the
328 stratosphere-to-troposphere exchange scheme (STE) in the model for O₃. Liang et al. (2011)
329 have pointed out that STE could be a significant direct sources of O₃ in the Arctic during spring-
330 summer. We retrieved vertical profiles of O₃ mixing ratios and specific humidity from
331 ozonesondes (0-5 km elevation above the Summit surface) launched at Summit for the months of
332 June and July in 2008 and compared those data with model control runs. Ozonesondes were
333 launched intensively during these two months (a total of 19 times). As shown in Figure 7,
334 compared with observations, model simulated O₃ mixing ratios averaged over 0-5 km above the
335 ground level are underestimated by 3% and 9% in June and July 2008 (Fig. 7a). However,
336 specific humidity in GEOS-5 is overestimated by 50% and 81% (Fig. 7b) respectively.
337 Ozonesonde data show that Summit frequently encounters high O₃/low water vapors events (e.g.,
338 July 9-11, 2008), which are likely of upper tropospheric/stratospheric origins (Helmig et al.,
339 2007b), but these are not captured by the model, which implies that GEOS-Chem possibly
340 underestimates STE for O₃ over Summit.

341 Boundary layer height is another factor that could potentially affect the discrepancy of O₃ mixing
342 ratios between model and observations. The mean springtime afternoon (12:00-14:00, local time)
343 boundary layer height in the model at Summit for the year 2009 is 160 m, which agrees fairly
344 well with observations (156 m) at Summit conducted in spring 2005 (Cohen et al., 2007).
345 Therefore, we exclude that model uncertainties in boundary layer height representation in
346 springtime cause the low bias of O₃ mixing ratios between model and observations.

347 Lastly, we also compared O₃ dry deposition velocity (V_{dry}) in the model in springtime with
348 observations at Summit. For spring 2009, mean O₃ V_{dry} in the model at Summit is 0.009 cm s⁻¹,
349 which is within the range of -0.01-0.01 cm s⁻¹ observed at Summit (Helmig et al., 2009). Helmig
350 et al. (2009) revealed that afternoon (12:00-18:00, local time) O₃ V_{dry} during springtime was
351 close to 0.01 cm s⁻¹. For other times during the day, O₃ V_{dry} was either close to zero or negative
352 (i.e., O₃ production over the snow outweighs its dry deposition). As a result, the net mean O₃ V_{dry}
353 from observations during springtime is about 4-6 times lower than model simulations. Therefore,



354 model overestimation of O₃ loss via surface uptake is another factor that contributes to the low
355 bias of surface O₃ mixing ratios at Summit in the model during springtime.

356 **4. Conclusions**

357 We combine model simulations with two-year (July 2008-June 2010) ground based
358 measurements at Summit to better understand the abundance and seasonal variations of
359 tropospheric O₃ and related species in the Arctic. In general, the GEOS-Chem model is capable
360 of reproducing the seasonal cycles of NO_x, PAN, C₂H₆, C₃H₈, CO, and O₃. However, some major
361 discrepancies between model and observations, especially for NO_x, PAN, C₂H₆, and O₃ are also
362 identified.

363 There are significant differences between model simulated NO_x mixing ratios and observations
364 for the spring and winter seasons. The model underestimates NO_x mixing ratios by
365 approximately 50% during late spring to early summer, which is likely due to the missing NO_x
366 emissions from nitrate photolysis in the snowpack. At the same time, the model overestimates
367 NO_x mixing ratios by more than a factor of two in wintertime. Model simulations indicate
368 episodic but frequent transport events from Europe in wintertime leading to NO_x spikes reaching
369 15 times typical NO_x mixing ratios at Summit; these large NO_x spikes are not seen in the
370 observational data. We have carried out multiple sensitivity model studies but are still unable to
371 reconcile this discrepancy.

372 The model successfully captures the seasonal cycles and the spring maximum PAN mixing
373 ratios, although it underestimates PAN by over 30% during late spring and early summer. Model
374 sensitivity studies reveal that this discrepancy could be largely resolved by accounting for PAN
375 emissions from snowpack.

376 For C₃H₈ and CO, model simulations overall agree well with the surface measurements,
377 however, the model tends to systematically overestimate surface C₂H₆ mixing ratios by
378 approximately 20% on an annual average, compared with observations. This may be explained
379 by that annual emission budgets of C₂H₆ over US and Asia from X08 emission inventory are
380 higher than those from NEI11 and MIX by over 40%. By replacing X08 over US with NEI11 for
381 C₂H₆, and scaling up MIX by 20%, the model-observation bias can be resolved, with annual
382 mean bias less than 1%. However, care must be taken to interpret this result because we do not



383 take into account other factors that may influence the discrepancy of surface C₂H₆ mixing ratios
384 at Summit between model and observations, such as the C₂H₆ chemistry with chlorine.

385 GEOS-Chem is able to reproduce the seasonal variation of surface O₃ at Summit but persistently
386 underestimates O₃ mixing ratios by approximately 13% (~ 6.5 ppbv) from April to July. This low
387 bias is likely caused by a combination of misrepresentations, including the missing snowpack
388 emissions of NO_x and HONO, inaccurate representation of Summit's elevation from coarser
389 model resolutions, model overestimated O₃ dry deposition velocity during springtime, as well as
390 the underestimated STE.

391 All the results presented above reveal the importance of local snowpack emissions in regulating
392 the air quality over Arctic. Improvements in global CTM could likely be achieved by coupling
393 snowpack emissions of reactive gases and photochemistry modules in order to better simulate O₃
394 and O₃ precursors over snow and ice in the Arctic (Zatko et al., 2016). Moreover, this study also
395 demonstrates that anthropogenic emissions from midlatitudes play an important role in affecting
396 the Arctic air quality. However, further investigations in anthropogenic NO_x emissions over
397 Europe and C₂H₆ emissions over Asia and North America are needed. The uncertainties in O₃
398 dry deposition and STE scheme in GEOS-Chem are warranted to be better quantified in our
399 future study.

400 **Acknowledgements** This research was funded by U.S. EPA grant 83518901). Its contents are
401 solely the responsibility of the grantee and do not necessarily represent the official views of the
402 U.S. EPA. Further, U.S. EPA does not endorse the purchase of any commercial products or
403 services mentioned in the publication. Superior, a high performance computing cluster at
404 Michigan Technological University, was used in obtaining results presented in this publication.
405 L. J. Kramer, D. Helmig and R. E. Honrath thank NASA (grant NNX07AR26G) for supporting
406 the measurements at Summit. We acknowledge the observational dataset of O₃ and CO provided
407 by NOAA ESRL. Technical supports from M. Sulprizio and C. Keller are also acknowledged.

408

409

410



411 **References**

- 412 Alvarado, M. J., Logan, J. A., Mao, J., Apel, E., Riemer, D., Blake, D., Cohen, R. C., Min, K.-E.,
413 Perring, A. E., Browne, E. C., Wooldridge, P. J., Diskin, G. S., Sachse, G. W., Fuelberg, H.,
414 Sessions, W. R., Harrigan, D. L., Huey, G., Liao, J., Case-Hanks, A., Jimmenez, J. L.,
415 Cubison, M. J., Vay, S. A., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Flocke, F. M.,
416 Pollack, I. B., Wennberg, P. O., Kurten, A., Crouse, J., St. Clair, J. M., Wisthaler, A.,
417 Mikoviny, T., Yantosca, R. M., Carouge, C. C., and Le Sager, P.: Nitrogen oxides and PAN in
418 plumes from boreal fires during ARCTAS-B and their impact on ozone: an integrated analysis
419 of aircraft and satellite observations, *Atmos. Chem. Phys.*, 10, 9739-9760,
420 doi:10.5194/acp-10-9739-2010, 2010.
- 421 Aydin, M., Verhulst, K. R., Saltzman, E. S., Battle, M. O., Montzka, S. A., Blake, D. R., Tang,
422 Q., and Prather, M. J.: Recent decreases in fossil-fuel emissions of ethane and methane
423 derived from firn air, *Nature*, 476, 198–201, doi:10.1038/nature10352, 2011.
- 424 Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q., Liu, H.,
425 Mickley L. J., and Schultz, M. G.: Global modeling of tropospheric chemistry with assimilated
426 meteorology: Model description and evaluation, *J. Geophys. Res.-Atmos.*, 106, 23073-23095,
427 2001.
- 428 Blake, D. R., and Rowland, F. S.: Global atmospheric concentrations and source strength of
429 ethane, *Nature*, 321, 231–233, 1986.
- 430 Chen, D., Wang, Y., McElroy, M. B., He, K., Yantosca, R. M., and Le Sager, P.: Regional CO
431 pollution and export in China simulated by the high-resolution nested-grid GEOS-Chem



- 432 model, Atmos. Chem. Phys., 9, 3825–3839, doi:10.5194/acp-9-3825-2009, 2009.
- 433 Cohen, L., Helmig, D., Neff, W., Grachev, A., and Fairall, C.: Boundary-layer dynamics and its
434 influence on atmospheric chemistry at Summit, Greenland, Atmos. Environ., 41, 5044–5060,
435 2007.
- 436 Crawford, J. H., Davis, D. D., Chen, G., Buhr, M., Oltmans, S., Weller, R., Mauldin, L., Eisele,
437 F., Shetter, R., Lefer, B., Ari- moto, R., and Hogan, A.: Evidence for photochemical produc-
438 tion of ozone at the South Pole surface, Geophys. Res. Lett., 28, 3641–3644, 2001.
- 439 Dibb, J. E., Arsenault, M., Peterson, M. C., and Honrath, R. E.: Fast nitrogen oxide
440 photochemistry in Summit, Greenland snow, Atmospheric Environment, 36, 2501–2511,
441 2002.
- 442 Emmons, L. K., Arnold, S. R., Monks, S. A., Huijnen, V., Tilmes, S., Law, K. S., Thomas, J. L.,
443 Raut, J.-C., Bouarar, I., Turquety, S., Long, Y., Duncan, B., Steenrod, S., Strode, S.,
444 Flemming, J., Mao, J., Langner, J., Thompson, A. M., Tarasick, D., Apel, E. C., Blake, D. R.,
445 Cohen, R. C., Dibb, J., Diskin, G. S., Fried, A., Hall, S. R., Huey, L. G., Weinheimer, A. J.,
446 Wisthaler, A., Mikoviny, T., Nowak, J., Peischl, J., Roberts, J. M., Ryerson, T., Warneke, C.,
447 and Helmig, D.: The POLARCAT Model Intercomparison Project (POLMIP): overview and
448 evaluation with observations, Atmos. Chem. Phys., 15, 6721–6744, doi:10.5194/acp-15-6721-
449 2015, 2015.
- 450 Etiope, G., and Ciccioli, P.: Earth’s degassing: A missing ethane and propane source, Science,
451 323, 1, 2009.
- 452 Fiore, A. M., Jacob, D. J., Field, B. D., Streets, D. G., Fernandes, S. D., and Jang, C.: Linking



- 453 ozone pollution and climate change: The case for controlling methane, *Geophys. Res. Lett.*,
454 29, 1919, doi:10.1029/2002GL015601, 2002.
- 455 Fischer, E. V., Jacob, D. J., Yantosca, R. M., Sulprizio, M. P., Millet, D. B., Mao, J., Paulot, F.,
456 Singh, H. B., Roiger, A., Ries, L., Talbot, R. W., Dzepina, K., and Pandey Deolal, S.:
457 Atmospheric peroxyacetyl nitrate (PAN): a global budget and source attribution, *Atmos.*
458 *Chem. Phys.*, 14, 2679-2698, doi:10.5194/acp-14-2679-2014, 2014.
- 459 Ford, K. M., Shepson, P. B., Bertman, S. B., Honrath, R. E., Peterson, M., Dibb, J. E., and
460 Bottenheim, J. W.: Studies of peroxyacetyl nitrate (PAN) and its interaction with the
461 snowpack at Summit, Greenland, *J. Geophys. Res.*, 107, ACH6, doi:10.1029/2001JD000547,
462 2002.
- 463 Franco, B., Bader, W., Toon, G., Bray, C., Perrin, A., Fischer, E., Sudo, K., Boone, C., Bovy, B.,
464 Lejeune, B., Servais, C., and Mahieu, E.: Retrieval of ethane from ground-based FTIR solar
465 spectra using improved spectroscopy: Recent burden increase above Jungfraujoeh, *J. Quant.*
466 *Spectrosc. Ra.*, 160, 36–49, doi:10.1016/j.jqsrt.2015.03.017, 2015.
- 467 Franco, B., Mahieu, E., Emmons, L. K., Tzompa-Sosa, Z. A., Fischer, E. V., Sudo, K., Bovy, B.,
468 Conway, S., Griffin, D., Hannigan, J. W., Strong, K., and Walker, K. A.: Evaluating ethane
469 and methane emissions associated with the development of oil and natural gas extraction in
470 North America, *Environ. Res. Lett.*, 11, doi:10.1088/1748-9326/11/4/044010, 2016.
- 471 Frey, M. M., Brough, N., France, J. L., Anderson, P. S., Traulle, O., King, M. D., Jones, A. E.,
472 Wolff, E. W., and Savarino, J.: The diurnal variability of atmospheric nitrogen oxides (NO and
473 NO₂) above the Antarctic Plateau driven by atmospheric stability and snow emissions, *Atmos.*



- 474 Chem. Phys., 13, 3045-3042, doi:10.5194/acp-13-3045-2013, 2013.
- 475 Giglio, L., Randerson, J. T., and van der Werf, G. R.: Analysis of daily, monthly, and annual
476 burned area using the fourth-generation global fire emissions database (GFED4), J. Geophys.
477 Res. Biogeosci., 118, 1, 317-328, doi:10.1002/jgrg.20042, 2013.
- 478 Grannas, A. M., Jones, A. E., Dibb, J., Ammann, M., Anastasio, C., Beine, H. J., Bergin, M.,
479 Bottenheim, J., Boxe, C. S., Carver, G., Chen, G., Crawford, J. H., Domine, F., Frey, M. M.,
480 Guzman, M. I., Heard, D. E., Helmig, D., Hoffmann, M.R., Honrath, R. E., Huey, L. G.,
481 Hutterli, M., Jacobi, H. W., Klán, P., Lefer, B., McConnell, J., Plane, J., Sander, R., Savarino,
482 J., Shepson, P. B., Simpson, W. R., Sodeau, J. R., von Glasow, R., Weller, R., Wolff, E. W.,
483 and Zhu, T.: An overview of snow photochemistry: evidence, mechanisms and impacts,
484 Atmos. Chem. Phys., 7, 4329–4373, doi:10.5194/acp-7-4329-2007, 2007.
- 485 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron C.: Estimate of
486 global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
487 Aerosols from Nature), Atmos. Chem. Phys., 6, 3181-3210, doi:10.5194/acp-6-3181-2006,
488 2006.
- 489 Gupta, M. L., Cicerone, R. J., Blake, D. R., Rowland, F. S., and Isaksen, I. S. A.: Global
490 atmospheric distributions and source strengths of light hydrocarbons and tetrachloroethene, J.
491 Geophys. Res., 103, 28219–28235, 1998.
- 492 Hausmann, P., Sussmann, R., and Smale, D.: Contribution of oil and natural gas production to
493 renewed increase in atmospheric methane (2007–2014): top–down estimate from ethane and
494 methane column observations, Atmos. Chem. Phys., 16, 3227–3244, doi:10.5194/acp-16-



- 495 3227-2016, 2016.
- 496 Helmig, D., Oltmans, S. J., Carlson, D., Lamarque, J. F., Jones, A., Labuschagne, C., Anlauf, K.,
497 Hayden, K.: A review of surface ozone in the polar regions, *Atmospheric Environment*, 41,
498 5138-5161, 2007a.
- 499 Helmig, D., Oltmans, S. J., Morse, T. O., and Dibb, J. E.: What is causing high ozone at Summit,
500 Greenland?, *Atmos. Environ.*, 41, 5031-5043, doi:10.1016/j.atmosenv.2006.05.084, 2007b.
- 501 Helmig, D., Ganzeveld, L., Butler, T., and Oltmans, S. J.: The role of ozone atmosphere-snow
502 gas exchange on polar, boundary-layer tropospheric ozone- a review and sensitivity analysis,
503 *Atmos. Chem. Phys.*, 7, 15-30, doi:10.5194/acp-7-15-2007, 2007c.
- 504 Helmig, D., Johnson, B., Oltmans, S. J., Neff, W., Eisele, F., and Davis, D. D.: Elevated ozone in
505 the boundary-layer at South Pole, *Atmos. Environ.*, 42, 2788–2803, 2008.
- 506 Helmig, D., Cohen, L. D., Bocquet, F., Oltmans, S., Grachev, A., and Neff, W.: Spring and
507 summertime diurnal surface ozone fluxes over the polar snow at Summit, Greenland,
508 *Geophys. Res. Lett.*, 36, L08809, doi:10.1029/2008GL036549, 2009.
- 509 Helmig, D., Petrenko, V., Martinerie, P., Witrant, E., Röckmann, T., Zuiderweg, A., Holzinger,
510 R., Hueber, J., Thompson, C., White, J. W. C., Sturges, W., Baker, A., Blunier, T., Etheridge,
511 D., Rubino, M., and Tans, P.: Reconstruction of Northern Hemisphere 1950–2010
512 atmospheric non-methane hydrocarbons, *Atmos. Chem. Phys.*, 14, 1463–1483,
513 doi:10.5194/acp-14-1463-2014, 2014a.
- 514 Helmig, D., Stephens, C., Caramore, J., and Hueber, J.: Seasonal behavior of non-methane
515 hydrocarbons in the firn air at Summit, Greenland, *Atmos. Environ.*, 85, 234-246,



- 516 doi:10.1016/j.atmosenv.2013.11.021, 2014b.
- 517 Helmig, D., Rossabi, S., Hueber, J., Tans, P., Montzka, S. A., Masarie, K., Thoning, K., Plass-
- 518 Duelmer, C., Claude, A., Car-
519 penter, L. J., Lewis, A. C., Punjabi, S., Reimann, S., Vollmer,
520 M. K., Steinbrecher, R., Hannigan, J. W., Emmons, L. K., Mahieu, E., Franco, B., Smale, D.,
521 and Pozzer, A.: Reversal of global atmospheric ethane and propane trends largely due to US
522 oil and natural gas production, *Nat. Geosci.*, 9, 490–495, doi:10.1038/ngeo2721, 2016.
- 523 Hickman, J. E., Huang, Y., Wu, S., Diru, W., Groffman, P. M., Tully, K. L., and Palm, C. A.:
- 524 Nonlinear response of nitric oxide fluxes to fertilizer inputs and the impacts of agricultural
525 intensification on tropospheric ozone pollution in Kenya, *Glob. Change Biol.*, 23, 3193-3204,
526 doi:10.1111/gcb.13644, 2017.
- 527 Hollaway, M. J., Arnold, S. R., Challinor, A. J., and Emberson, L. D.: Intercontinental trans-
528 boundary contributions to ozone-induced crop yield losses in the North Hemisphere,
529 *Biogeosciences*, 9, 271-292, doi: 10.5194/bg-9-271-2012, 2012.
- 530 Honrath, R. E., Peterson, M. C., Guo, S., Dibb, J. E., Shepson, P. B., and Campbell, B.: Evidence
531 of NO_x production within or upon ice particles in the Greenland snowpack, *Geophys. Res.*
532 *Lett.*, 26, 695-698, 1999.
- 533 Honrath, R. E., Guo, S., Peterson, M. C., Dziobak, M. P., Dibb, J. E., and Arsenault, M. A.:
- 534 Photochemical production of gas phase NO_x from ice crystal NO₃⁻, *J. Geophys. Res.*, 105,
535 24183–24190, 2000a.
- 536 Honrath, R. E., Peterson, M. C., Dziobak, M. P., Dibb, J. E., Arsenault, M. A., and Green, S. A.:
- 537 Release of NO_x from Sunlight-irradiated Midlatitude Snow, *Geophys. Res. Lett.*, 27, 2237–
538 2240, 2000b.



- 538 Honrath, R. E., Lu, Y., Peterson, M. C., Dibb, J. E., Arsenault, M. A., Cullen, N. J., and Steffen,
539 K.: Vertical fluxes of NO_x, HONO, and HNO₃ above the snowpack at Summit, Greenland,
540 Atmos. Environ., 36, 2629-2640, doi:10.1016/S1352-2310(02)00132-2, 2002.
- 541 Huang, Y., Wu, S., Dubey, M. K., and French, N. H. F.: Impact of aging mechanism on model
542 simulated carbonaceous aerosols, Atmos. Chem. Phys., 13, 6329–6343, doi:10.5194/acp-13-
543 6329-2013, 2013.
- 544 Hudman, R. C., Moore, N. E., Mebust, A. K., Martin, R. V., Russell, A. R., Valin, L. C., and
545 Cohen, R. C.: Steps towards a mechanistic model of global soil nitric oxide emissions:
546 implementation and space based-constraints, Atmos. Chem. Phys., 12, 7779-7795, doi:
547 10.5194/acp-12-7779- 2012, 2012.
- 548 Hudman, R. C., Murray, L. T., Jacob, D. J., Turquety, S., Wu, S., Millet, D. B., Avery, M.,
549 Goldstein, A. H., and Holloway, J.: North American influence on tropospheric ozone and the
550 effects of recent emission reductions: Constraints from ICARTT observations, J. Geophys.
551 Res., 114, D07302, doi:10.1029/2008JD010126, 2009.
- 552 Jacob, D. J., Wofsy, S. C., Bakwin, P. S., Fan, S.-M., Harriss, R. C., Talbot, R. W., Bradshaw, J.
553 D., Sandholm, S. T., Singh, H. B., Browell, E. V., Gregory, G. L., Sachse, G. W., Shipham,
554 M. C., Blake, D. R., and Fitzjarrald, D. R.: Summertime photochemistry of the troposphere at
555 high northern latitudes, J. Geophys. Res., 97, D15, 16421-16431, doi:10.1029/91JD01968,
556 1992.
- 557 Jang, J.-C., Jeffries, H., Byun, D., and Pleim, J.: Sensitivity of ozone to model grid resolution – I.
558 Application of high resolution regional acid deposition model, Atmos. Environ., 29, 3085-
559 3100, doi:10.1016/1352-2310(95)00118-I, 1995.



- 560 Johnson, M.S., Meskhidze, N., Solmon, F., Gasso, S., Chuang, P. Y., Gaiero, D. M., Yantosca,
561 R. M., Wu, S., Wang, X., Carouge, C.: Modeling Dust and Soluble Iron Deposition to the
562 South Atlantic Ocean, *J. Geophys. Res.*, 115, D15202, doi:10.1029/2009JD013311, 2010.
- 563 Kanakidou, M., Singh, H. B., Valentin, K. M., and Crutzen, P. J.: A two-dimensional study of
564 ethane and propane oxidation in the troposphere, *J. Geophys. Res.*, 96, 15, 395–315, 413,
565 doi:10.1029/91JD01345, 1991.
- 566 Knowlton, K, Rosenthal, J. E., Hogrefe, C., Lynn, B., Gaffin, S., Goldberg, R., Rosenzweig, C.,
567 Civerolo, K., Ku, J. Y., Kinney, P. L.: Assessing ozone-related health impacts under a
568 changing climate, *Environ. Health Persp.*, 112, 1557–1563, 2004.
- 569 Kramer, L. J., Helmig, D., Burkhardt, J. F., Stohl, A., Oltmans, S., and Honrath, R. E.: Seasonal
570 variability of atmospheric nitrogen oxides and non-methane hydrocarbons at the GEOSummit
571 station, Greenland, *Atmos. Chem. Phys.*, 15, 6827–6849, doi:10.5194/acp-15-6827-2015,
572 2015.
- 573 Kuhns, H., Knipping, E. M., and Vukovich, J. M.: Development of a United States-Mexico emis-
574 sions inventory for the Big Bend Regional Aerosol and Visibility Observational (BRAVO)
575 Study, *JAPCA J. Air Waste M.*, 55, 677–692, 2005.
- 576 Kumar, A., Wu, S., Weise, M. F., Honrath, R., Owen, R. C., Helmig, D., Kramer, L., Val Martin,
577 M., and Li, Q.: Free-troposphere ozone and carbon monoxide over the North Atlantic for
578 2001–2011, *Atmos. Chem. Phys.*, 13, 12537–12547, doi:10.5194/acp-13-12537-2013, 2013.
- 579 Legrand, M., Preunkert, S., Frey, M., Bartels-Rausch, Th., Kukui, A., King, M. D., Savarino, J.,
580 Kerbrat, M., and Jourdain, B.: Large mixing ratios of atmospheric nitrous acid (HONO) at



- 581 Concordia (East Antarctic Plateau) in summer: a strong source from surface snow?, Atmos.
582 Chem. Phys., 14, 9963-9976, doi:10.5194/acp-14-9963-2014, 2014.
- 583 Li, M., Zhang, Q., Kurokawa, J., Woo, J. H., He, K. B., Lu, Z., Ohara, T., Song, Y., Streets, D.
584 G., Carmichael, G. R., Cheng, Y. F., Hong, C. P., Huo, H., Jiang, X. J., Kang, S. C., Liu, F.,
585 Su, H., and Zheng, B.: MIX: a mosaic Asian anthropogenic emission inventory under the
586 international collaboration framework of the MICS-Asia and the HTAP, Atmos. Chem. Phys.,
587 17, 935-963, doi:10.5194/acpd-17-935-2017.
- 588 Liang, J., and Jacobson, M. Z.: Effects of subgrid segregation on ozone production efficiency in
589 a chemical model, Atmos. Environ., 34, 2975–2982, doi:10.1016/S1352-2310(99)00520-8,
590 2000.
- 591 Liang, Q., Rodriguez, J. M., Douglass, A. R., Crawford, J. H., Olson, J. R., Apel, E., Bian, H.,
592 Blake, D. R., Brune, W., Chin, M., Colarco, P. R., da Silva, A., Diskin, G. S., Duncan, B. N.,
593 Huey, L. G., Knapp, D. J., Montzka, D. D., Nielsen, J. E., Pawson, S., Riemer, D. D.,
594 Weinheimer, A. J., and Wisthaler, A.: Reactive nitrogen, ozone and ozone production in the
595 Arctic troposphere and the impact of stratosphere-troposphere exchange, Atmos. Chem. Phys.,
596 11, 13181-13199, doi:10.5194/acp-11-13181-2011, 2011.
- 597 Liu, H. Y., Jacob, D. J., Bey, I., and Yantosca, R. M.: Constraints from pb-210 and Be-7 on wet
598 deposition and transport in a global three-dimensional chemical tracer model driven by
599 assimilated meteorological fields, J. Geophys. Res.-Atmos., 106, 12109-12128, doi:
600 10.1029/2000JD900839, 2001.
- 601 Martin, R. V., Jacob, D. J., Logan, J. A., Bey, I., Yantosca, R. M., Staudt, A. C., Li, Q., Fiore, A.



- 602 M., Duncan, B. N., and Liu, H.: Interpretation of TOMS observations of tropical tropospheric
603 ozone with a global model and in situ observations, *J. Geophys. Res.*, 107(D18), ACH 4-1-
604 ACH 4-27, doi: 10.1029/2001JD001480, 2002.
- 605 McClure-Begley, A., Petropavlovskikh, I., Oltmans, S.: NOAA Global Monitoring Surface
606 Ozone Network. 1973-2014. National Oceanic and Atmospheric Administration, Earth
607 Systems Research Laboratory Global Monitoring Division. Boulder, CO. DATE ACCESSED:
608 4/23/2017, <http://dx.doi.org/10.7289/V57P8WBF>, 2014.
- 609 McLinden, C. A., Olsen, S. C., Hannegan, B., Wild, O., Prather, M. J., and Sundet, J.:
610 Stratospheric ozone in 3-D models: A simple chemistry and the cross-tropopause flux, *J.*
611 *Geophys. Res.*, 105, D11, 14653-14665, doi:10.1029/2000JD900124, 2000.
- 612 Monks, S. A., Arnold, S. R., Emmons, L. K., Law, K. S., Turquety, S., Duncan, B. N.,
613 Flemming, J., Huijnen, V., Tilmes, S., Langner, J., Mao, J., Long, Y., Thomas, J. L., Steenrod,
614 S. D., Raut, J. C., Wilson, C., Chipperfield, M. P., Diskin, G. S., Weinheimer, A., Schlager,
615 H., and Ancellet, G.: Multi-model study of chemical and physical controls on transport of
616 anthropogenic and biomass burning pollution to the Arctic, *Atmos. Chem. Phys.*, 15, 3575-
617 3603, doi:10.5194/acp-15-3575-2015, 2015.
- 618 Murray, K. A., Kramer, L. J., Doskey, P. V., Ganzeveld, L., Seok, B., Van Dam, B., and Helmig,
619 D.: Dynamics of ozone and nitrogen oxides at Summit, Greenland. II. Simulating snowpack
620 chemistry during a spring high ozone event with a 1-D process-scale model, *Atmos. Environ.*,
621 117, 110–123, doi:10.1016/j.atmosenv.2015.07.004, 2015.
- 622 Murray, L. T., Jacob, D. J., Logan, J. A., Hudman, R. C., and Koshak, W. J.: Optimized regional



- 623 and interannual variability of lightning in a global chemical transport constrained by LIS/OTD
624 satellite data, *J. Geophys. Res.*, 117, D20307, doi:10.1029/2012JD017934, 2012.
- 625 Novelli, P. C. and Masarie, K. A.: Atmospheric Carbon Monoxide Dry Air Mole Fraction from
626 the NOAA ESRL Carbon Cycle Cooperative Global Air Sampling Network, 1988-2014,
627 Version: 2015-08-04, NOAA, available at
628 ftp://aftp.cmdl.noaa.gov/data/trace_gases/co/flask/surface/, 2015.
- 629 Park, R. J., Jacob, D. J., Field, B. D., Yantosca, R. M., and Chin, M.: Natural and transboundary
630 pollution influences on sulfate-nitrate-ammonium aerosols in the United States: Implications
631 for policy, *J. Geophys. Res.-Atmos.*, 109, D15204, doi:10.1029/2003JD004473, 2004.
- 632 Petropavlovskikh, I. and Oltmans, S. J.: Tropospheric Ozone Measurements, 1973-2011,
633 Version: 2012-07-10, NOAA, available at: <ftp://aftp.cmdl.noaa.gov/data/ozwv/SurfaceOzone/>,
634 2012.
- 635 Pozzer, A., Pollmann, J., Taraborrelli, D., Jöckel, P., Helmig, D., Tans, P., Hueber, J., and
636 Lelieveld, J.: Observed and simulated global distribution and budget of atmospheric C₂–C₅
637 alkanes, *Atmos. Chem. Phys.*, 10, 4403–4422, doi:10.5194/acp-10-4403-2010, 2010.
- 638 Price, C. and Rind, D.: A simple lightning parameterization for calculating global lightning
639 distributions, *J. Geophys. Res.*, 97, 9919-9933, doi:10.1029/92JD00719, 1992.
- 640 Ridley, B. A., and Grahek, F.: A small, low flow, high sensitivity reaction vessel for NO
641 chemiluminescence detectors, *Am. Meteorol. Soc.*, 7, 307-311, 1990.
- 642 Rudolph, J.: The tropospheric distribution and budget of ethane, *J. Geophys. Res.*, 100, D6,
643 11369-11381, 1995.



- 644 Sauvage, B., Martin, R. V., van Donkelaar, A., Liu, X., Chance, K., Jaeglé, L., Palmer, P. I., Wu,
645 S., and Fu, T.-M.: Remote sensed and in situ constraints on processes affecting tropical tropo-
646 spheric ozone, *Atmos. Chem. Phys.*, 7, 815–838, doi:10.5194/acp-7-815-2007, 2007.
- 647 Sherwen, T., Schmidt, J. A., Evans, M. J., Carpenter, L. J., Großmann, K., Eastham, S. D.,
648 Jacob, D. J., Dix, B., Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A.,
649 Prados-Roman, C., Mahajan, A. S., and Ordóñez, C.: Global impacts of tropospheric halogens
650 (Cl, Br, I) on oxidants and composition in GEOS-Chem, *Atmos. Chem. Phys.*, 16, 12239–
651 12271, doi:10.5194/acp-16-12239-2016, 2016.
- 652 Shindell, D. T., Chin, M., Dentener, F., Doherty, R. M., Faluvegi, G., Fiore, A. M., Hess, P.,
653 Koch, D. M., MacKenzie, I. A., Sanderson, M. G., Schultz, M., Stevenson, D. S., Teich, H.,
654 Textor, C., Wild, O., Bergmann, D. J., Bey, I., Bian, H., Cuvelier, C., Duncan, B. N., Folberth,
655 G., Horowitz, L. W., Jonson, J., Kaminski, J. W., Marmer, E., Park, R., Pringle, K. J.,
656 Schroeder, S., Szopa, S., Takemura, T., Zeng, G., Keating, T. J., and Zuber, A.: A multi-model
657 assessment of pollution transport to the Arctic, *Atmos. Chem. Phys.*, 8, 5353–5372,
658 doi:10.5194/acp-8-5353-2008, 2008.
- 659 Simon, H., Beck, L., Bhave, P. V., Divita, F., Hsu, Y., Luecken, D., Mobley, J. D., Pouliot, G.
660 A., Reff, A., Sarwar, G., and Strum, M.: The development and uses of EPA’s SPECIATE
661 database, *Atmospheric Pollution Research*, 196–206, 10.5094/apr.2010.026, 2010.
- 662 Simpson, I. J., Sulbaek Andersen, M. P., Meinardi, S., Bruhwiler, L., Blake, N. J., Helmig, D.,
663 Rowland, F. S., and Blake, D. R.: Long-term decline of global atmospheric ethane
664 concentrations and implications for methane, *Nature*, 488, 490–494, doi:10.1038/nature11342,



- 665 2012.
- 666 Thomas, J. L., Stutz, J., Lefer, B., Huey, L. G., Toyota, K., Dibb, J. E., and von Glasow, R.:
- 667 Modeling chemistry in and above snow at Summit, Greenland – Part 1: Model description and
- 668 results, *Atmos. Chem. Phys.*, 11, 4899–4914, doi:10.5194/acp-11-4899-2011, 2011.
- 669 Thomas, J. L., Dibb, J. E., Huey, L. G., Liao, J., Tanner, D., Lefer, B., von Glasow, R., and
- 670 Stutz, J.: Modeling chemistry in and above snow at Summit, Greenland – Part 2: Impact of
- 671 snowpack chemistry on the oxidation capacity of the boundary layer, *Atmos. Chem. Phys.*, 12,
- 672 6537–6554, doi:10.5194/acp-12-6537-2012, 2012.
- 673 Tzompa-Sosa, Z. A., Mahieu, E., Franco, B., Keller, C. A., Turner, A. J., Helmig, D., Fried, A.,
- 674 Richter, D., Weibring, P., Walega, J., Yacovitch, T. I., Herndon, S. C., Blake, D. R., Hase, F.,
- 675 Hannigan, J. W., Conway, S., Strong, K., Schneider, M., and Fischer, E. V.: Revisiting global
- 676 fossil fuel and biofuel emissions of ethane, *J. Geophys. Res. Atmos.*, 122,
- 677 doi:10.1002/2016JD025767, 2017.
- 678 Unger, N., Shindell, D. T., Koch, D. M., and Streets, D. G.: Cross influences of ozone and
- 679 sulfate precursor emissions changes on air quality and climate, *Proc. Natl. Acad. Sci.*, 103,
- 680 4377–4380, doi:10.1073/pnas.0508769103, 2006.
- 681 van het Bolscher, M., Pereira, J., Spesso, A., Dalsoren, S., van Noije, T., and Szopa, S.:
- 682 REanalysis of the TROpospheric chemical composition over the past 40 years: A long-term
- 683 global modeling study of tropospheric chemistry, Max Plank Inst. For Meteorology, Hamburg,
- 684 Germany, 77, 2008.
- 685 Van Dam, B., Helmig, D., Toro, C., Doskey, P., Kramer, L., Murray, K., Ganzeveld, L., and



- 686 Seok, B.: Dynamics of ozone and nitrogen oxides at Summit, Greenland: I. Multi-year
687 observations in the snowpack, *Atmos. Environ.*, 123, 268-284,
688 doi:10.1016/j.atmosenv.2015.09.060, 2015.
- 689 Walker, T. W., Jones, D. B. A., Parrington, M., Henze, D. K., Murray, L. T., Bottenheim, J. W.,
690 Anlauf, K., Worden, J. R., Bowman, K. W., Shim, C., Singh, K., Kopacz, M., Tarasick, D. W.,
691 Davies, J., von der Gathen, P., Thompson, A. M., and Carouge, C. C.: Impacts of midlatitude
692 precursor emissions and local photochemistry on ozone abundances in the Arctic, *J. Geophys.*
693 *Res.*, 117, D01305, doi:10.1029/2011JD016370, 2012.
- 694 Wang, Y. H., Jacob, D. J., and Logan, J. A.: Global simulation of tropospheric O₃-NO_x-hydro-
695 carbon chemistry 1. Model formulation, *J. Geophys. Res.-Atmos.*, 103, 10713-10725, doi:
696 10.1029/98JD00158, 1998.
- 697 Wang, Y. X., McElroy, M. B., Jacob, D. J., and Yantosca, R. M.: A nested grid formulation for
698 chemical transport over Asia: Applications to CO, *J. Geophys. Res.*, 109, D22307,
699 doi:10.1029/2004JD005237, 2004.
- 700 Wesely, M. L.: Parameterization of surface resistances to gaseous dry deposition in regional-
701 scale numerical-models, *Atmos. Environ.*, 23, 1293-1304, doi:10.1016/0004-6981(89)90153-
702 4, 1989.
- 703 Wespes, C., Emmons, L., Edwards, D. P., Hannigan, J., Hurtmans, D., Saunio, M., Coheur, P.-
704 F., Clerbaux, C., Coffey, M. T., Batchelor, R. L., Lindenmaier, R., Strong, K., Weinheimer, A.
705 J., Nowak, J. B., Ryerson, T. B., Crounse, J. D., and Wennberg, P. O.: Analysis of ozone and
706 nitric acid in spring and summer Arctic pollution using aircraft, ground-based, satellite



- 707 observations and MOZART-4 model: source attribution and partitioning, *Atmos. Chem. Phys.*,
708 12, 237–259, doi:10.5194/acp-12-237-2012, 2012.
- 709 Wu, S., Mickley, L. J., Jacob, D. J., Logan, J. A., Yantosca, R. M., and Rind, D.: Why are there
710 large differences between models in global budgets of tropospheric ozone?, *J. Geophys. Res.*,
711 112(D5), D05302, doi:10.1029/02006JD007801, 2007.
- 712 Xiao, Y., Logan, J. A., Jacob, D. J., Hudman, R. C., Yantosca, R., and Blake, D. R.: The global
713 budget of ethane and regional constraints on U.S. sources, *J. Geophys. Res.*, 113, D21306,
714 doi:10.1029/2007JD009415, 2008.
- 715 Yang, J., Honrath, R. E., Peterson, M. C., Dibb, J. E., Sumner, A. L., Shepson, P. B., Frey, M.,
716 Jacobi, H.-W., Swanson, A., and Blake, N.: Impacts of snowpack emissions on deduced levels
717 of OH and peroxy radicals at Summit, Greenland, *Atmos. Environ.*, 36, 2523-2534,
718 doi:10.1016/S1352-2310(02)00128-0, 2002.
- 719 Yevich, R., and Logan, J. A.: An assessment of biofuel use and burning of agricultural waste in
720 the developing world, *Global Biogeochem. Cy.*, 17, 1095, doi:10.1029/2002GB001952, 2003.
- 721 Yu, K., Jacob, D. J., Fisher, J. A., Kim, P. S., Marais, E. A., Miller, C. C., Travis, K. R., Zhu, L.,
722 Yantosca, R. M., Sulprizio, M. P., Cohen, R. C., Dibb, J. E., Fried, A., Mikoviny, T., Ryerson,
723 T. B., Wennberg, P. O., and Wisthaler, A.: Sensitivity to grid resolution in the ability of a
724 chemical transport model to simulate observed oxidant chemistry under high-isoprene
725 conditions, *Atmos. Chem. Phys.*, 16, 4369–4378, doi:10.5194/acp-16-4369-2016, 2016.
- 726 Yue, X., and Unger, N.: Ozone vegetation damage effects on gross primary productivity in the
727 United States, *Atmos. Chem. Phys.*, 14, 9137-9153, doi:10.5194/acp-14-9137-2014, 2014.



728 Zatkan, M. C., Geng, L., Alexander, B., Sofen, E. D., and Klein, K.: The impact of snow nitrate
729 photolysis on boundary layer chemistry and the recycling and redistribution of reactive
730 nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem.
731 Phys., 16, 2819-2842, doi:10.5194/acpd-16-2819-2016, 2016.

732 Zhang, H., Wu, S., Huang, Y., and Wang, Y.: Effects of stratospheric ozone recovery on
733 photochemistry and ozone air quality in the troposphere, Atmos. Chem. Phys., 14, 4079-4086,
734 doi:10.5194/acp-14-4079-2014, 2014.

735 Zhou, X., Beine, H. J., Honrath, R. E., Fuentes, J., Simpson, W., Shepson, P. B., and Bottenheim,
736 J. W.: snowpack photochemical production of HONO: a major source of OH in the Arctic
737 boundary layer in springtime, Geophys. Res. Lett., 28, 21, 4087-4090, 2001.

738
739
740
741
742
743
744
745
746
747
748

749 **Table 1.** Surface NO₂ measurements over Europe during 2009/12/01-2010/01/31.

Site ID	Site name	Lat. (°N)	Lon.(°E)	Altitude a.s.l (m)	Technique	Resolution
BE0001R	Offagne	49.88	5.20	430	chemiluminescence	hourly
BE0032R	Eupen	50.63	6	295	chemiluminescence	hourly
DE0001R	Westerland	54.93	8.31	12	NaJ_solution	daily
DK0008R	Anholt	56.72	11.52	40	UV_fluorescence	hourly
FI0096G	Pallas	67.97	24.12	340	chemiluminescence	hourly
GB0014R	High Muffles	54.33	-0.8	267	chemiluminescence	daily
NL0009R	Kollumerwaard	53.33	6.28	1	chemiluminescence	hourly
NO0001R	Birkenes	58.38	8.25	190	glass sinter	daily
NO0039R	Kårvatn	62.78	8.88	210	glass sinter	daily
NO0056R	Hurdal	60.37	11.08	300	glass sinter	daily
SE0005R	Bredkålen	63.85	15.3	404	abs_tube	daily

750

751

752

753

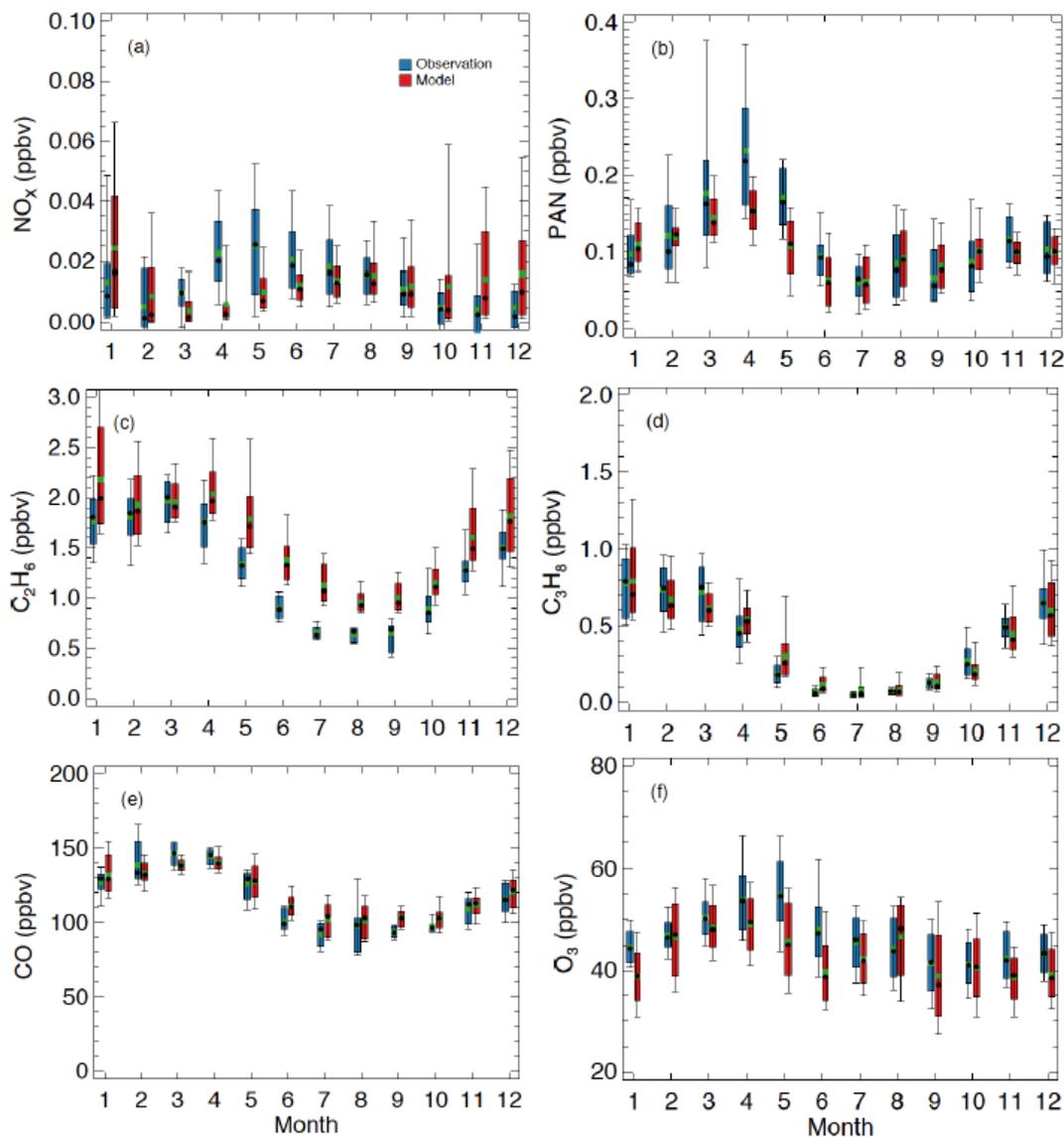
754

755

756

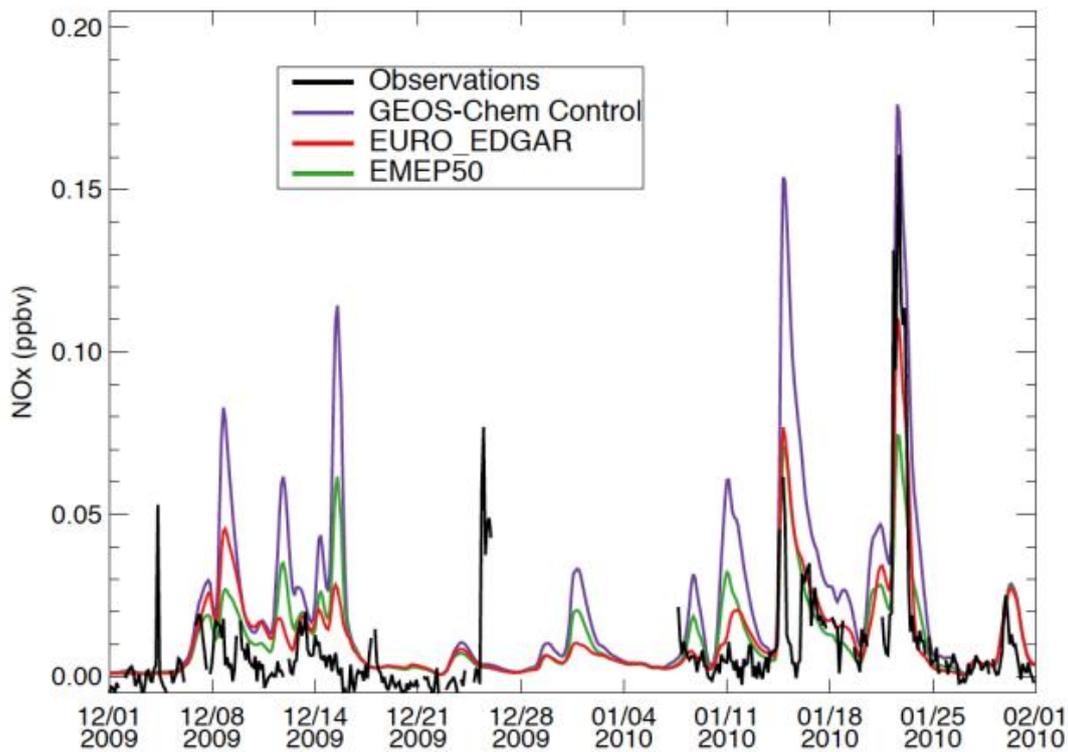
757

758



759

760 **Fig. 1.** Box plot comparison for seasonal variations of (a) NO_x , (b) PAN, (c) C_2H_6 , (d) C_3H_8 , (e)
761 CO, and (f) O_3 between GEOS-Chem model simulations (red) and in-situ measurements (blue)
762 over Summit for the period of 2008/07-2010/06. Data shown are monthly averages during this
763 period. The thick (thin) bars represent the 67% (95%) confidence intervals. Black and green dots
764 represent median and mean values, respectively. The statistics are based on daily averages.



765

766 **Figure 2.** Timeseries of surface NO_x mixing ratios over Summit from observations, GEOS-
767 Chem model control simulations, EURO_EDGAR, and EMEP50 during 2009/12/01-2010/01/31.

768

769

770

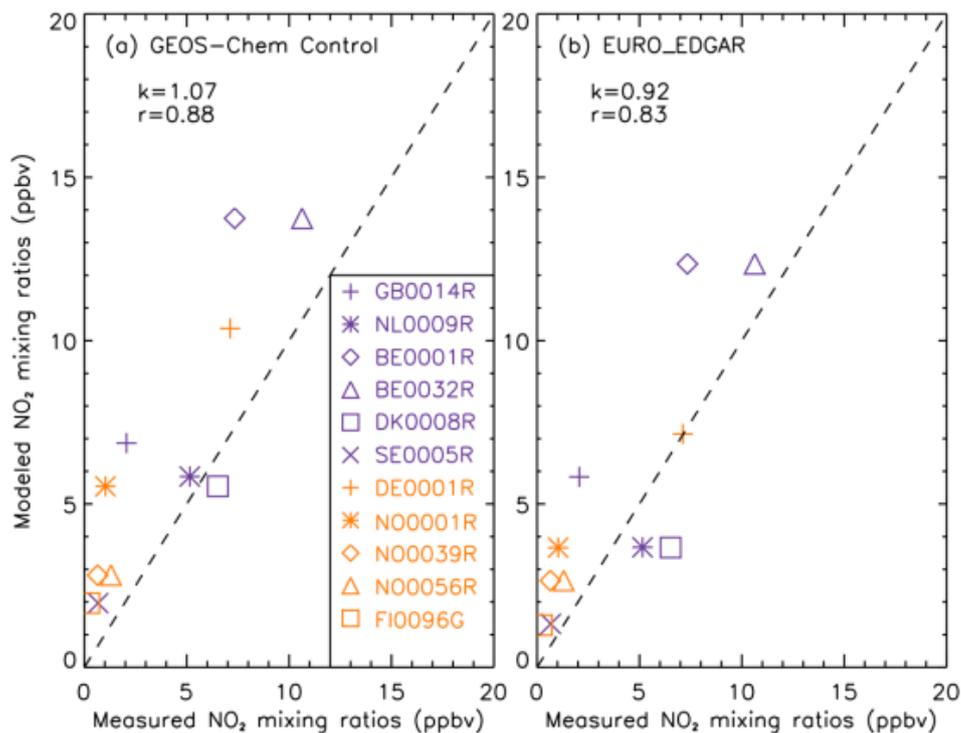
771

772

773

774

775



776

777 **Figure 3.** Scatter plots between measured monthly mean NO₂ mixing ratios at 11 observational
778 sites over Europe and model simulations from (a) GEOS-Chem control simulations and (b)
779 EURO_EDGAR during 2009/12/01-2010/01/31; also shown is the corresponding model-to-
780 observation slopes (k) and correlation coefficients (r) for each panel. The dash line is the 1:1
781 ratio. Explanations of site abbreviations are listed in Table 1.

782

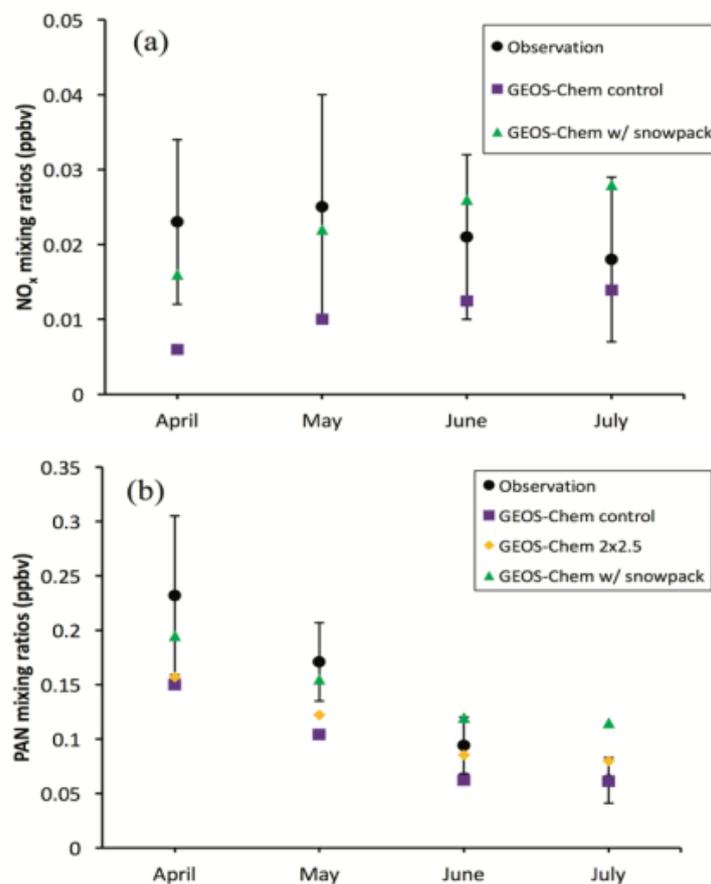
783

784

785

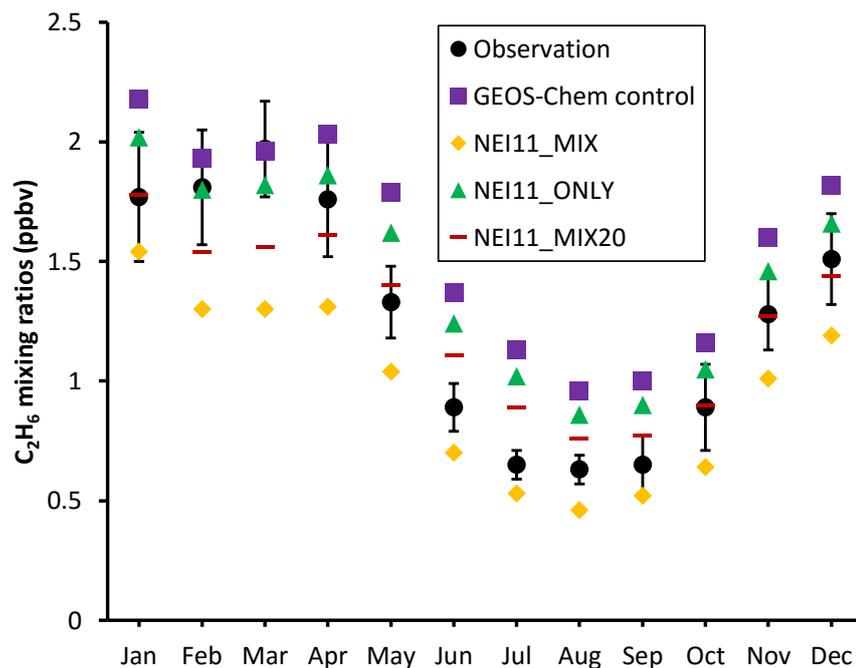
786

787



788

789 **Figure 4.** Monthly mean surface (a) NO_x and (b) PAN mixing ratios from observations (black
790 circles), simulations with (green triangles) /without (purple squares) snowpack emissions, and
791 GEOS-Chem simulations with horizontal grid resolution 2° x 2.5° (orange diamonds) over the
792 period of April- July during 07/2008-06/2010. Vertical bars denote standard deviations over the
793 course of observations for each month.



794

795 **Figure 5.** Monthly mean surface C_2H_6 mixing ratios at Summit from observations (black
796 circles), GEOS-Chem model control simulations (purple squares), NEI11_MIX (orange
797 diamond), and NEI11_ONLY (green triangles) simulations during 2008-2010; vertical bars
798 denote the standard deviation over the course of observations for each month.

799

800

801

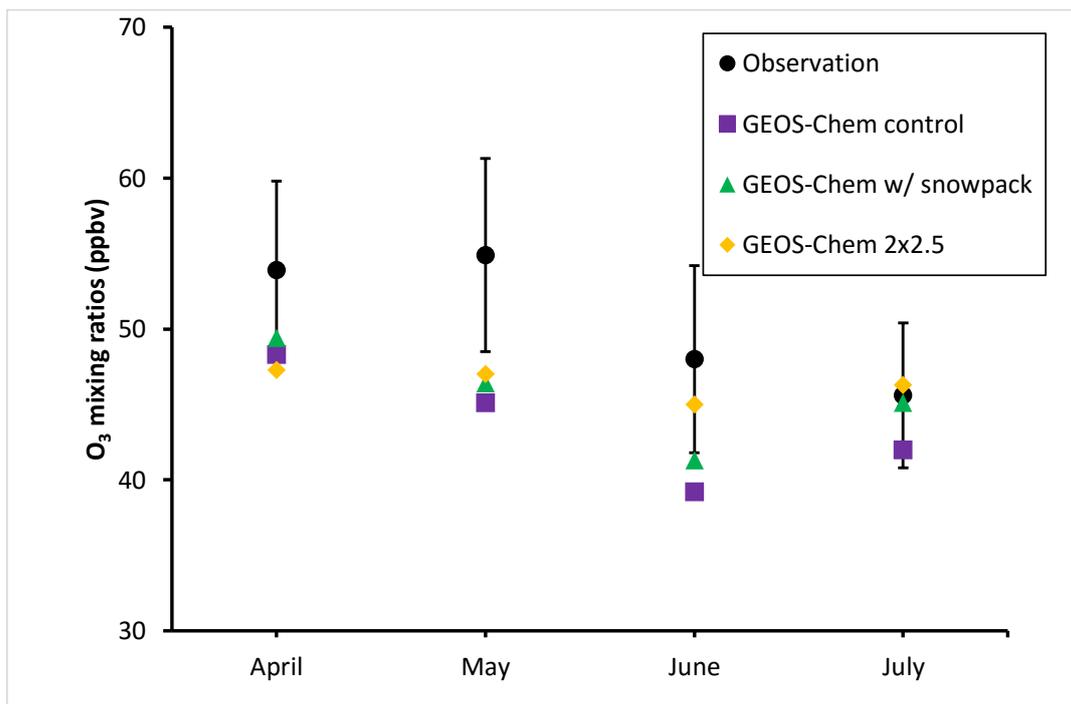
802

803

804

805

806



807

808 **Figure 6.** Monthly mean surface O₃ mixing ratios from observations (black circles), GEOS-
809 Chem control runs (purple squares), with snowpack chemistry (green triangles), and horizontal
810 grid resolution 2° x 2.5° (orange diamonds) for April-July. Vertical bars denote the variability
811 over the course of observations for each month.

812

813

814

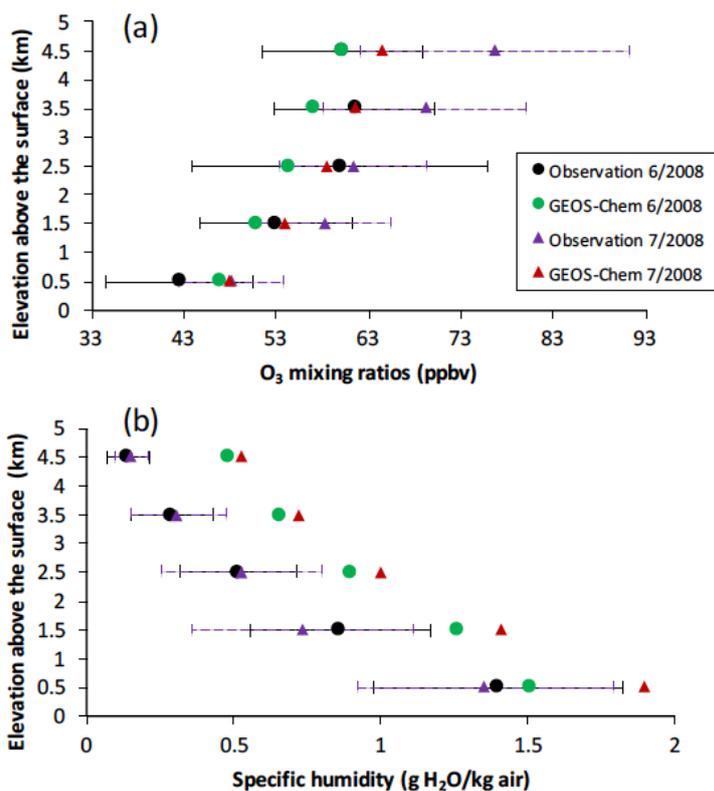
815

816

817

818

819



820

821 **Figure 7.** Comparisons of vertical profiles of (a) O₃ and (b) specific humidity between GEOS-
822 Chem simulations and ozonesondes in June and July 2008 respectively, averaged over 1-km
823 altitude bins. Black and green solid circles represent observations and simulations in June 2008
824 while purple and red triangles denote observations and simulations for July 2008 respectively.
825 Solid and dash horizontal error bars represent observational standard deviations for June and July
826 respectively.

827