

1 **Surface ozone and its precursors at Summit, Greenland: comparison between observations**
2 **and model simulations**

3 Yaoxian Huang^{1,a}, Shiliang Wu^{1,2,3}, Louisa J. Kramer^{1,2,b}, Detlev Helmig⁴, and Richard E.
4 Honrath^{1,2,†}

5 ¹Department of Geological and Mining Engineering and Sciences, Michigan Technological
6 University, Houghton, Michigan, USA

7 ²Atmospheric Sciences Program, Michigan Technological University, Houghton, Michigan,
8 USA

9 ³College of Environmental Science and Engineering, Ocean University of China, Qingdao, China

10 ⁴Institute of Arctic and Alpine Research, University of Colorado, Boulder, Colorado, USA

11 ^anow at: Department of Climate and Space Sciences and Engineering, University of Michigan,
12 Ann Arbor, Michigan, USA

13 ^bnow at: University of Birmingham, Birmingham, UK

14 [†]deceased

15
16 *Correspondence to:* S. Wu (slwu@mtu.edu) and Y. Huang (yaoxianh@mtu.edu)
17

18 **Abstract.** Recent studies have shown significant challenges for atmospheric models to simulate
19 tropospheric ozone (O₃) and some of its precursors in the Arctic. In this study, ground based data
20 are combined with a global 3-D chemical transport model (GEOS-Chem) to examine the
21 abundance and seasonal variations of O₃ and its precursors at Summit, Greenland (72.34° N,
22 38.29° W, 3212 m.a.s.l). Model simulations for atmospheric nitrogen oxides (NO_x), peroxyacetyl
23 nitrate (PAN), ethane (C₂H₆), propane (C₃H₈), carbon monoxide (CO), and O₃ for the period of
24 07/2008-06/2010 are compared with observations. The model performs well in simulating certain
25 species (such as CO and C₃H₈), but some significant discrepancies are identified for other
26 species and further investigated. The model generally underestimates NO_x and PAN (by ~ 50%
27 and 30%, respectively) for March-June. Likely contributing factors to the low bias include
28 missing NO_x and PAN emissions from snowpack chemistry in the model. At the same time, the
29 model overestimates NO_x mixing ratios by more than a factor of two in wintertime, with episodic
30 NO_x mixing ratios up to 15 times higher than the typical NO_x levels at Summit. Further
31 investigation shows that these simulated episodic NO_x spikes are always associated with

32 transport events from Europe, but the exact cause remains unclear. The model systematically
33 overestimates C_2H_6 mixing ratios by approximately 20% relative to observations. This
34 discrepancy can be resolved by decreasing anthropogenic C_2H_6 emissions over Asia and the US
35 by $\sim 20\%$, from 5.4 to 4.4 Tg/yr. GEOS-Chem is able to reproduce the seasonal variability of O_3
36 and its spring maximum. However, compared with observations, it underestimates surface O_3 by
37 approximately 13% (6.5 ppbv) from April to July. This low bias appears to be driven by several
38 factors including missing snowpack emissions of NO_x and nitrous acid in the model, the weak
39 simulated stratosphere-to-troposphere exchange flux of O_3 over the summit, as well as the coarse
40 model resolution.

41 **1. Introduction**

42 Tropospheric ozone (O_3) and its precursors, including nitrogen oxides ($NO_x = NO + NO_2$),
43 carbon monoxide (CO), and volatile organic compounds (VOCs, such as ethane, propane, etc.)
44 are important atmospheric species affecting both air quality and climate (e.g., Jacob et al., 1992;
45 Fiore et al., 2002; Unger et al., 2006; Hollaway et al., 2012). Tropospheric O_3 is mainly
46 produced by photochemical oxidation of CO and VOCs in the presence of NO_x , with additional
47 contribution by transport from the stratosphere. Its major sinks include chemical reactions and
48 dry deposition. As a reservoir species for NO_x , peroxyacetyl nitrate (PAN) also plays an
49 important role in atmospheric chemistry. PAN and O_3 , as well as some of their precursors, have
50 relatively long lifetimes in the atmosphere, enabling them to be transported long distance to
51 remote regions such as the Arctic.

52 Recent studies have shown some significant challenges for atmospheric chemical transport
53 models to simulate O_3 and its precursors in the Arctic (e.g., Shindell et al., 2008; Alvarado et al.,
54 2010; Walker et al., 2012; Wespes et al., 2012; Fischer et al., 2014; Monks et al., 2015), but the
55 causes remain unclear. In the multi-model assessment by Shindell et al. (2008), more than a
56 dozen models all showed systematic and persistent underestimation of O_3 at the GEOSummit
57 station, Greenland (hereafter referred to as Summit). Alvarado et al. (2010) used NO_x and PAN
58 measurements from the ARCTAS (Arctic Research of the Composition of the Troposphere from
59 Aircraft and Satellites) mission in the summer to compare with model simulations. They found
60 that model simulated NO_x mixing ratios were higher than observations, while PAN mixing ratios
61 were lower than the observations in fresh boreal fire plumes. In terms of global PAN

62 simulations, Fischer et al. (2014) directly partitioned 40% of NO_x emissions from wildfires to
63 PAN formation, which improved the agreement between model and observations. However, the
64 model still underestimated PAN surface mixing ratios during springtime in the Arctic. Walker et
65 al. (2012) reported that model simulated O₃ mixing ratios were biased low when compared with
66 balloon data during summertime from two high-latitude sites at Eureka (80°N, 86°W) and Ny-
67 Ålesund (79°, 12°E). Wespes et al. (2012) also revealed that model simulated O₃ mixing ratios
68 within the boundary layer were significantly underestimated during spring-summer, compared
69 with ARCTAS measurements. More recently, Monks et al. (2015) further demonstrated that
70 model simulated O₃ mixing ratios in the Arctic at the surface and in the upper troposphere were
71 generally lower than the observations. In addition, a recent study by Christian et al. (2017)
72 compared O₃ observations from the ARCTAS campaign to GEOS-Chem model simulations and
73 found consistent low biases with the model simulated O₃ at all altitudes except the surface.

74 Field measurements at Summit show that the snowpack emits gas-phase NO_x, PAN, nitrous acid
75 (HONO), as well as hydrogen peroxide (H₂O₂) during spring-summer, when polar sun rises
76 (Ford et al., 2002; Honrath et al., 2002). Although several 1-D models (Thomas et al., 2011,
77 2012; Frey et al., 2013; Murray et al., 2015) have validated the importance of snowpack
78 emissions for surface NO_x as well as O₃ formation, current global chemical transport models
79 (CTMs) usually do not include these emission sources (Zatko et al., 2016).

80 In this study, we examine the abundance and seasonal variations of O₃ and its precursors at
81 Summit with a global chemical transport model, GEOS-Chem CTM, in conjunction with two
82 years of in-situ measurement data for 2008-2010. We first evaluate the model performance in
83 simulating surface O₃ and its precursors, and then implement a series of model updates to resolve
84 the identified model biases. This paper is organized as follows: section 2 describes model
85 methods and observations, followed by detailed comparisons of model simulations against
86 observations for O₃ and O₃ precursors in section 3; conclusions are summarized in section 4.

87 **2. Observational data and model simulations**

88 In situ measurements of NO_x, PAN, and non-methane hydrocarbons (NMHCs) were performed
89 at Summit from July 2008 to June 2010 (Helmig et al., 2014b; Kramer et al., 2015). An
90 automated chemiluminescence instrument was used to measure NO_x (Ridley and Grahek, 1990);

91 a commercial PAN gas chromatography analyzer (PAN-GC, Metcon, In., Boulder, CO) was
92 employed for the measurement of PAN. Measurements of NMHC relied on an automated Gas
93 Chromatography-Flame Ionization Detection (GC-FID) system. Readers are referred to Kramer
94 et al. (2015) and Helmig et al. (2014b) for the details of the measurement techniques and
95 equipment setup. Surface measurements of O₃ using ultraviolet light absorption
96 (Petropavlovskikh and Oltmans, 2012), and CO by GC (Novellie and Masarie, 2015) are from
97 the National Oceanic and Atmospheric Administration (NOAA). Hourly averaged O₃, and flask
98 data for CO between July 2008 and June 2010 were downloaded from the NOAA Earth System
99 Research Laboratory (ESRL) Global Monitoring Division (GMD) website
100 (<http://www.esrl.noaa.gov/gmd/dv/data/>). Vertical ozonesonde data profiles were also
101 downloaded from NOAA ESRL GMD (McClure-Begley et al., 2014).

102 Simulations of O₃ and related species (NO_x, PAN, NMHCs) are conducted using the GEOS-
103 Chem model (Bey et al., 2001) with coupled O₃-NO_x-VOC-Aerosol chemistry mechanism (i.e.
104 these species interact with each other in the model). The GEOS-Chem CTM is driven by
105 assimilated meteorological data from the Goddard Earth Observing System version 5.2.0
106 (GEOS-5.2.0) of the NASA Global Modeling Assimilation Office. The GEOS-Chem model has
107 been extensively evaluated and applied in a wide range of applications (Martin et al., 2002; Park
108 et al., 2004; Wu et al., 2007; Hudman et al., 2009; Johnson et al., 2010; Huang et al., 2013;
109 Kumar et al., 2013; Zhang et al., 2014; Hickman et al., 2017), including the studies in the Arctic
110 (e.g., Alvarado et al., 2010; Monks et al., 2015; Christian et al., 2017). GEOS-Chem v10-1 with
111 grid resolution of 4° latitude by 5° longitude, and 47 vertical layers was used for the model
112 control simulation. Following McLinden et al. (2000), the Linoz stratospheric O₃ chemistry
113 scheme was used. The simulation was run from June 2007 to June 2010 and the results from the
114 last two years were used in the final analysis. Time series data were archived with 3-hr temporal
115 resolution at the Summit grid box for each model vertical level, including the model bottom
116 layer. For comparison with surface observations at Summit, Greenland, we sampled the data for
117 the model bottom layer. We acknowledge that the topography is not well resolved at coarse
118 model resolution, and we believe that the diagnostics for the model bottom layer would work
119 better than the model vertical layer that is about 3212 m above the sea level for comparison to
120 the surface measurements.

121 Global anthropogenic emissions of NO_x, SO₂, NH₃, and CO in the model are based on the
122 Emission Database for Global Atmospheric Research (EDGAR) v4.2 inventory, which is
123 overwritten by regional emission inventories where applicable, such as the BRAVO inventory
124 for Mexico (Kuhns et al., 2005), the CAC over Canada, the EMEP emissions over Europe, the
125 Model Inter-comparison Study for Asia Phase III (MIX) emissions over Asia (Li et al., 2017),
126 and the US EPA NEI 2011 (NEI11) emission inventory (Simon et al., 2010). The soil NO_x
127 emission scheme follows Hudman et al. (2012). Lightning NO_x emissions are calculated per
128 flash rate based on GEOS-5 computed cloud-top heights (Price and Rind, 1992), which are
129 determined by deep convection and constrained by satellite observations for monthly average
130 flash rates from the Lightning Imaging Sensor and Optical Transient Detector (OTD/LIS)
131 (Sauvage et al., 2007; Murray et al., 2012). Biomass burning emissions are from the Global Fire
132 Emission Database version 4 (GFED4) inventory with monthly resolution (Giglio et al., 2013).
133 The RETRO (Reanalysis of the TROpospheric chemical composition) global anthropogenic
134 NMHC emission inventory (van het Bolscher et al., 2008) was used except for ethane (C₂H₆) and
135 propane (C₃H₈), which follows Xiao et al. (2008, hereafter referred to as X08) for the year 2001.
136 In GEOS-Chem, RETRO used to serve as the default global anthropogenic emission inventory
137 for C₂H₆, however, the annual budget of which has been shown too low compared with
138 observations. Global biofuel emission inventory follows Yevich and Logan (2003), which
139 includes emissions for C₂H₆ and C₃H₈. For biogenic VOC emissions, the Model of Emissions of
140 Gases and Aerosols from Nature (MEGAN) scheme (Guenther et al., 2006) was used. Dry
141 deposition of species in GEOS-Chem uses a standard resistance-in-series scheme (Wesely,
142 1989), as implemented in Wang et al. (1998). Wet scavenging follows Liu et al. (2001),
143 including scavenging in convective updraft, rainout (in-cloud) and washout (below-cloud) from
144 convective anvils and large-scale precipitation.

145 We first run the standard GEOS-Chem model with a-priori emissions and compare the
146 simulation results against observations for various species (including NO_x, PAN, C₂H₆, C₃H₈,
147 CO, and O₃, as shown in Fig. 1). Then we focus on the model-observation discrepancies, and
148 where applicable, make revisions to the model simulations and further evaluate the improvement
149 in model performance, as discussed in details below.

150 **3. Results and Discussions**

151 3.1 NO_x

152 We first combine the two years of data for July 2008 – June 2010 and analyze their seasonal
153 variations. As shown in Figure 1a, the GEOS-Chem model simulated NO_x agree well with the
154 observations for July-October. However, compared to observations, the model results
155 significantly overestimate NO_x mixing ratios for November-January by about 150%, while
156 underestimating the data in spring and early summer by approximately 60%. Another challenge
157 for the model simulation is that it does not capture the decrease of NO_x for May - November. We
158 find that during the 2009-2010 winter season, model simulations show several high NO_x spikes
159 with peak NO_x mixing ratios reaching ~ 0.15 ppbv or higher, which is ~ 15 times greater than
160 typical background levels (Fig. 2). These large peaks in NO_x were not observed in the data.
161 Similar peaks were also seen in the model simulations during the 2008-2009 winter season;
162 however, there are no measurements available for this period to compare with.

163 Further analyses show that the model-simulated high NO_x spikes during wintertime are all
164 associated with transport events from Europe. We carry out a sensitivity study to examine the
165 impacts of European emissions on Arctic NO_x by manually reducing anthropogenic NO_x
166 emissions from the EMEP emission inventory over Europe by 50% (EMEP50). Results show
167 that surface peak NO_x mixing ratios over Summit during the spike events (e.g., dates around
168 12/09/2009, 12/15/2009, 1/15/2010 and 1/22/2010) from EMEP50 almost decline proportionally
169 by ~ 50% during 2009/12/01-2010/01/31 (Fig. 2), which confirms that the modeled NO_x spikes
170 at Summit during wintertime are associated with transport from Europe. However, the model
171 simulated NO_x is still significantly higher than observations. Comparisons for surface NO₂
172 mixing ratios between model simulations and 11 in-situ observational sites over Europe during
173 this period were conducted with data downloaded from <http://ebas.nilu.no>. For detailed site
174 information, NO₂ measurement technique and resolution, refer to Table 1. Measurement data over
175 these two months for each site were averaged to compare with the corresponding grid cell in the
176 model. As shown in Figure 3a, GEOS-Chem overestimates surface NO₂ mixing ratios at these
177 sites by over 66%, compared with observations (slope=1.07; correlation coefficient=0.88).

178 Instead of using EMEP, we carry out another sensitivity study to force anthropogenic NO_x
179 emissions over Europe following EDGAR v4.2 (EURO_EDGAR), with other model
180 configurations identical to control simulations. As shown in Figure 2, the NO_x mixing ratios over

181 Summit during 12/2009-01/2010 agree much better with observations, especially for January
182 2010, where the model captures the magnitudes of observational peaks. This is because NO_x
183 emissions from EDGAR over Europe (1.97 Tg NO) are 12% lower than those from EMEP (2.24
184 Tg NO) for the months of 12/2009 and 01/2010. Furthermore, the discrepancy for the differences
185 of surface NO₂ mixing ratios over Europe between EURO_EDGAR and observations is further
186 reduced (by 50%), relative to the control runs, with a model-to-observation slope of 0.92 and a
187 correlation coefficient of 0.83 (Fig. 3b). Similarly, we also tested the sensitivity of surface NO_x
188 mixing ratios over Summit in response to the changes in the anthropogenic NO_x emissions from
189 NEI11 over US and MIX over Asia (including Siberia) during these two months, and found that
190 surface NO_x mixing ratios over Summit during these two months were quite close to the control
191 simulations (not shown), reflecting insensitivity to emission perturbations from the US and Asia.
192 Therefore, we conclude that uncertainties in fossil fuel NO_x emissions of EMEP associated with
193 transport events from Europe in the model are the most likely cause for the wintertime NO_x
194 spikes over Summit.

195 For April-July, model simulated monthly mean NO_x mixing ratios over Summit are a factor of
196 two lower than the observations (Fig. 4a). In-situ measurements at Summit by Honrath et al.
197 (1999, 2000a, 2000b, 2002) showed upward fluxes of NO_x (2.52×10^8 molecules cm⁻² s⁻¹) from
198 photolysis of nitrate in snowpack during the summertime, leading to enhancement in NO_x levels
199 in the surface layer by approximately 20 pptv, which is comparable to surface NO_x mixing ratios
200 in the Arctic from other sources. Similar results were found over the East Antarctic Plateau
201 snow/ice sheet (Frey et al., 2013; Legrand et al., 2014). The standard GEOS-Chem model does
202 not include the photolysis of nitrate from snowpack, implying a missing source for NO_x in the
203 Arctic/Antarctic boundary layer.

204 In order to test the sensitivity of model simulated surface NO_x mixing ratios to the snowpack
205 emissions, we implement in the model a constant NO_x flux of $\sim 2.52 \times 10^8$ molecules cm⁻² s⁻¹
206 during April-July over Greenland (60-85° N, 20-60° W), following the measurements conducted
207 at Summit during summertime by Honrath et al. (2002). As a result, we find that on average, the
208 model simulated surface NO_x mixing ratios for April to July over Summit more than double that
209 from the control simulation, which improves the agreement between model and observations for
210 April-June (Fig. 4a). However, the assumed NO_x flux from snowpack in the model leads to

211 overestimate of NO_x mixing ratios in July and the model is still not able to reproduce the
212 decreasing trend of NO_x for May-October. This decreasing trend of NO_x may be driven by the
213 decreasing NO_x production rate in snowpack resulting from a gradual depletion of snowpack
214 NO_x reservoir (Van Dam et al., 2015), which is not reflected in the model since we implement a
215 simple constant NO_x emission flux. Dibb et al. (2007) reported that nitrate concentrations in the
216 Summit snowpack peaked in June and declined toward fall by ~ 1/3. Van Dam et al. (2015)
217 further showed decreasing trend for NO_x mixing ratios within the snowpack at Summit from
218 June to October. This may partially explain why we would see the declining trend of surface
219 NO_x mixing ratios over Summit from June toward fall. The NO_x emissions from snowpack are
220 affected by a number of factors including nitrate concentrations and solar radiation available and
221 the responses can be very non-linear. Further investigations are needed to account for the
222 seasonal variations of snowpack NO_x emissions from nitrate photolysis in the model, i.e.,
223 constrained by seasonal snowpack NO_x emission flux measurements in the future.

224 **3.2 PAN**

225 We then examine the model performance for PAN, which serves as a reservoir for NO_x. Figure
226 1b shows the comparison of model simulated monthly mean PAN mixing ratios with the
227 measurement data. The model captures the seasonal variation of PAN well, although
228 significantly (by ~ 30%) underestimating the PAN mixing ratios for April-June. By running the
229 model simulation with higher horizontal resolution at 2° latitude by 2.5° longitude (hereafter
230 referred to as GEOS-Chem 2x2.5), we find that the monthly mean PAN mixing ratios over
231 Summit during April-July increased by up to 23.3 pptv compared to the 4x5 simulation (Fig. 4b).
232 This can be explained by two reasons. First, coarse model resolution (e.g., 4x5 horizontal
233 resolution) could artificially smear the intense emission sources throughout the entire grid cell
234 (e.g., over urban regions), leading to underestimates of downwind concentrations for species,
235 e.g., O₃ and O₃ precursors (Jang et al., 1995; Yu et al., 2016). Second, ventilation of the lower
236 atmosphere could be better resolved by a finer model resolution, leading to more efficient
237 vertical advection (Wang et al., 2004; Chen et al., 2009; Yu et al., 2016). However, on average,
238 monthly mean model simulated PAN mixing ratios are still underestimated by 20% during this
239 period, compared with observations. This is consistent with the study by Arnold et al. (2015),
240 which reported that model simulated PAN mixing ratios in GEOS-Chem were lower than

241 ARCTAS observations over high-latitude atmosphere in the Arctic. Meanwhile, this study also
242 revealed that GEOS-Chem produced less PAN relative to CO in Arctic air parcels that were
243 influenced by fires, compared with other models.

244 Snowpack can emit not only NO_x, but also PAN, based on field studies at Summit during
245 summertime by Ford et al. (2002). GEOS-Chem does not contain snowpack PAN emissions and
246 chemistry. For a sensitivity study, similar to snowpack NO_x emissions as discussed in section
247 3.1, we consider a 24-hr constant flux of 2.52×10^8 molecules cm⁻² s⁻¹ of PAN over Greenland
248 from April to July, following Ford et al. (2002). As a result, model simulated PAN mixing ratios
249 agree much better with observations (Fig. 4b). Note that there are also other possible reasons that
250 lead to model bias. For instance, a study by Fischer et al. (2014) showed improved agreement
251 between modeled and measured PAN in the high latitudes when assigning a portion of the fire
252 emissions in the model above the boundary layer and also directly partitioning 40% of NO_x
253 emissions from fires into PAN. We carried out a sensitivity test with similar treatments, but no
254 significant improvements in the model simulated surface PAN were observed at the Summit site.
255 Therefore, we did not include the PAN updates from Fischer et al. (2014) in other model
256 simulations in this study.

257 **3.3 NMHC**

258 Comparisons of observed surface C₂H₆ and C₃H₈ mixing ratios with GEOS-Chem simulations at
259 Summit are shown in Figures 1c and 1d. The model simulations agree well with surface
260 measurements of C₃H₈, but systematically overestimate C₂H₆ (by approximately 25% annually),
261 with the largest bias (0.48 ppbv) occurring during summer. This is consistent with the study from
262 Tzompa-Sosa et al., (2017), which used the same model as our study and pointed out that using
263 X08 as global anthropogenic C₂H₆ emission inventory systematically overestimated surface C₂H₆
264 mixing ratios over the Northern Hemisphere, compared with ground-based observations.
265 Anthropogenic C₂H₆ emissions over US from NEI11 are shown to geographically match the
266 distribution of active oil and natural wells (Tzompa-Sosa et al., 2017), and the most recent MIX
267 has been updated to synergize anthropogenic C₂H₆ emissions from various countries over Asia
268 (Li et al., 2017). Therefore, instead of using global anthropogenic fossil fuel emissions of C₂H₆
269 following X08, we first conduct sensitivity simulations by overwriting global emission

270 inventories by NEI11 over the US, and MIX over Asia (hereafter referred to as NEI11_MIX).
271 Both NEI11 and MIX contain emissions for the years from 2008 to 2010, which could
272 realistically represent the annual and seasonal variations of C₂H₆ emissions over the US and
273 Asia, thus spatially and temporally better representative of anthropogenic C₂H₆ emissions from
274 mid-latitudes transported to the Arctic regions. In general, model control simulations
275 overestimate annual mean surface C₂H₆ mixing ratios primarily in the Northern Hemisphere,
276 with large differences occurring over Asia and the US by up to 5 ppbv, compared with
277 NEI11_MIX during the period of 07/2008-06/2010 (Fig. S1). All the above changes are driven
278 by the substantial reductions of anthropogenic C₂H₆ emissions between emission inventories,
279 from 3.5 (X08) to 2.5 Tg/yr (MIX) over Asia, and from 1.9 Tg/yr (X08) to 1.4 Tg/yr (NEI11)
280 over US, reflecting the decreasing trend of anthropogenic C₂H₆ emissions during 2001-2009
281 (Helmig et al., 2014a), because the X08 emission inventory is based on the year 2001.
282 Substantial changes in surface C₂H₆ mixing ratios over the US between control simulations and
283 NEI11_MIX reflect that there exist temporal changes of C₂H₆ emissions from oil and gas
284 productions during the period of 2001-2009. A similar pattern was also found by Tsompa-Sosa et
285 al. (2017). In contrast to the control simulations, NEI11_MIX model simulations show that
286 monthly mean C₂H₆ mixing ratios over Summit are systematically underestimated by 24%,
287 compared with observations (Fig. 5). Tsompa-Sosa et al. (2017) reported that NEI11 for C₂H₆
288 emissions were likely underestimated by 40%, compared with in-situ and aircraft observations
289 over the US. We therefore run a sensitivity simulation by increasing the NEI11 C₂H₆ emissions
290 by 40% and keeping other model configuration identical to NEI11_MIX (hereafter referred to as
291 NEI11_40_MIX). We find this update leads to an increase in the model simulated annual mean
292 surface C₂H₆ mixing ratios over Summit by only 6% during the period of 07/2008-06/2010
293 (figure not shown), still not able to explain the high model bias.

294 Similar to NEI11_MIX, we further conduct sensitivity studies by only replacing the regional
295 emission inventory for C₂H₆ over the US, with other regions still following X08 (hereafter
296 referred to as NEI11_ONLY). Consequently, model simulated surface C₂H₆ mixing ratios over
297 Summit agree better with observations during winter-spring (Fig. 5), decreasing the bias from
298 +15% (control simulations) to +6%. However, model simulated C₂H₆ mixing ratios during
299 summer-fall are higher than the observations by over 30%.

300 We then scale up the MIX emissions for C₂H₆ by 20% over Asia, with other model
301 configurations identical to NEI11_MIX (hereafter referred to as NEI11_MIX20). By doing this,
302 we increase fossil fuel C₂H₆ emissions from 2.5 to 3 Tg/yr. We find that the simulated annual
303 mean surface C₂H₆ mixing ratios at Summit from NEI11_MIX20 agree quite well with
304 observations (within 1%). Similarly, better agreement between model and observations are found
305 for monthly average values for October - January. However, the new simulation is not able to
306 reproduce the seasonal cycle of C₂H₆ - the model significantly underestimates in February – April
307 but overestimates in June – September (Fig. 5). This implies that further assessments of
308 anthropogenic C₂H₆ emissions from MIX over Asia are needed and a more accurate global
309 anthropogenic C₂H₆ emission inventory should be developed and validated to replace X08 in the
310 future. Note that this standard version of GEOS-Chem does not account for the sink of C₂H₆
311 from the reaction with chlorine, which could reduce the global annual mean surface C₂H₆ mixing
312 ratios by 0-30%, and the global burden of C₂H₆ by about 20% (Sherwen et al., 2016). This may
313 introduce additional uncertainty for our measurement-model comparison, together with
314 uncertainty in the seasonality of C₂H₆ chemistry.

315 **3.4 CO**

316 Figure 1e shows the comparison of model simulated CO mixing ratios with observations over
317 Summit. Overall, the model generally captures the abundance and seasonal variation of CO.
318 Compared with observations, the annual mean CO mixing ratio is slightly overestimated by
319 about 3 ppbv in the model.

320 **3.5 O₃**

321 Surface O₃ mixing ratios from model simulations and surface observations are compared in
322 Figure 1f. The GEOS-Chem model captures the seasonal variation of O₃ including the spring
323 peak. However, the model shows a systematic low bias for most of the year, in particular for
324 April–July when the surface O₃ mixing ratios are underestimated by ~ 13% (~ 6.5 ppbv). Here
325 we focus our analysis for the possible causes that lead to the model low bias during April-July.

326 As discussed earlier, snowpack emissions due to the photolysis of nitrate in the snow during late
327 spring and summer could contribute to NO_x and HONO levels in the ambient air, which could
328 enhance O₃ production (Crawford et al., 2001; Zhou et al., 2001; Dibb et al., 2002; Honrath et

329 al., 2002; Yang et al., 2002; Grannas et al., 2007; Helmig et al., 2008; Legrand et al., 2014). We
330 run a sensitivity study to test the response of surface O₃ mixing ratios to the perturbations of NO_x
331 and HONO from snowpack emissions. In addition to snowpack NO_x emissions that are described
332 in Section 3.1, we implement in the model a constant flux of HONO (4.64×10^7 molecules cm⁻²
333 s⁻¹) from April to July (Honrath et al., 2002). As a result, monthly mean model simulated surface
334 O₃ mixing ratios increase by up to 3 ppbv during this period (Fig. 6). The largest effect occurs in
335 July due to relatively strong solar radiation. O₃ formation due to snowpack emissions in our
336 study is slightly higher than that in Zatzko et al. (2016) because HONO from snowpack emissions
337 is not considered in their study. However, for the months of April and May, surface O₃ mixing
338 ratios only increase by ~ 1 ppbv, compared with the control runs. That is, even after accounting
339 for the snowpack emissions, the model simulated O₃ mixing ratios are still significantly lower
340 than the observations.

341 Comparison of the model simulations at different resolutions (4x5 vs. 2x2.5) shows that the finer
342 resolution simulations substantially increase monthly mean O₃ mixing ratios over Summit by up
343 to 6 ppbv for the months of June and July (Fig. 6). As discussed in section 3.2, fine model
344 resolution can better resolve the emission strengths, which could significantly affect downwind
345 chemical reactions, e.g., O₃ production efficiency (Liang and Jacobson, 2000). Moreover, terrain
346 elevations from fine model resolution are better represented (thus better representative of
347 Summit's elevation) and more efficient vertical ventilation of O₃ and O₃ precursors can be
348 achieved (Wang et al., 2004). Together with the impact of snowpack chemistry, this brings
349 model simulated surface O₃ mixing ratios over Summit to much better agreement with
350 observations for June - July. However, there is still a low bias in the model for the months of
351 April and May.

352 Another possible cause for the low O₃ biases in model simulations is the calculated stratosphere-
353 to-troposphere exchange (STE) O₃ flux in the model. Liang et al. (2011) have pointed out that
354 STE could be a significant direct sources of O₃ in the Arctic during spring - summer. We retrieve
355 vertical profiles of O₃ mixing ratios and specific humidity from ozonesondes (0-5 km elevation
356 above the Summit surface) launched at Summit for the months of June and July in 2008 and
357 compare those data with model control runs. Ozonesondes were launched intensively during
358 these two months (a total of 19 times). As shown in Figure 7, compared with observations,

359 model simulated O₃ mixing ratios averaged over 0-5 km above ground level are underestimated
360 by 3% and 9% in June and July 2008 (Fig. 7a). However, specific humidity in GEOS-5 is
361 overestimated by 50% and 81% (Fig. 7b) respectively. Ozonesonde data show that Summit
362 frequently encounters high O₃/low water vapors events (e.g., July 9-11, 2008), which are likely
363 of upper tropospheric/stratospheric origin (Helmig et al., 2007b), but these are not captured by
364 the model, which implies that GEOS-Chem possibly underestimates STE for O₃ over Summit.
365 This is consistent with the study by Choi et al. (2017), which found low bias with model
366 simulated O₃ mixing ratios over high-latitude upper troposphere of the Northern Hemisphere,
367 compared with ozonesonde data, and attributed the low bias to weak STE in the model.

368 Boundary layer height is another factor that could lead to model-data discrepancy in O₃ mixing
369 ratios (Grannas et al., 2007; Helmig et al., 2007a, c). The mean springtime afternoon (12:00-
370 14:00, local time) boundary layer height in the model at Summit for the year 2009 is 160 m,
371 which agrees fairly well with observations (156 m) at Summit conducted in spring 2005 (Cohen
372 et al., 2007). Therefore, we exclude that model uncertainties in boundary layer height
373 representation in springtime cause the low bias of O₃ mixing ratios between model and
374 observations.

375 **4. Conclusions**

376 We combine model simulations with two-year (July 2008 - June 2010) ground based
377 measurements at Summit, Greenland, to investigate the abundance and seasonal variations of
378 surface O₃ and related species in the Arctic. In general, the GEOS-Chem model is capable of
379 reproducing the seasonal cycles of NO_x, PAN, C₂H₆, C₃H₈, CO, and O₃. However, some major
380 discrepancies between model and observations, especially for NO_x, PAN, C₂H₆, and O₃ are
381 identified.

382 There are significant differences between model simulated NO_x mixing ratios and observations
383 for the spring and winter seasons. The model underestimates NO_x mixing ratios by
384 approximately 50% during late spring to early summer, which is likely due to the missing NO_x
385 emissions from nitrate photolysis in the snowpack. At the same time, the model overestimates
386 NO_x mixing ratios by more than a factor of two in wintertime. Model simulations indicate
387 episodic but frequent transport events from Europe in wintertime leading to NO_x spikes reaching

388 15 times typical NO_x mixing ratios at Summit; these large NO_x spikes are not seen in the
389 observations. We have carried out multiple sensitivity model studies but are still unable to fully
390 reconcile this discrepancy.

391 The model successfully captures the seasonal cycles and the spring maximum PAN mixing
392 ratios, although it underestimates PAN by over 30% during late spring and early summer. Model
393 sensitivity studies reveal that this discrepancy could be largely resolved by accounting for PAN
394 emissions from snowpack.

395 For C₃H₈ and CO, model simulations overall agree well with the surface measurements.
396 However, the model tends to systematically overestimate surface C₂H₆ mixing ratios by ~ 20%
397 on an annual average, compared with observations. This may be explained by that annual
398 emission budgets of C₂H₆ over US and Asia from X08 emission inventory are higher than those
399 from NEI11 and MIX by over 40%. By replacing X08 over the US with NEI11 for C₂H₆, and
400 scaling up MIX by 20%, the model-observation bias can be resolved, resulting in an annual mean
401 bias of less than 1%. However, care must be taken to interpret this result because we do not take
402 into account other factors that may influence the discrepancy of surface C₂H₆ mixing ratios at
403 Summit between model and observations, such as the C₂H₆ chemistry with chlorine.

404 GEOS-Chem is able to reproduce the seasonal variation of surface O₃ at Summit but persistently
405 underestimates O₃ mixing ratios by ~ 13% (~ 6.5 ppbv) from April to July. This low bias is
406 likely caused by a combination of misrepresentations, including the missing snowpack emissions
407 of NO_x and HONO, inaccurate representation of Summit's elevation with a too coarse model
408 resolution, as well as the underestimated STE.

409 All the results presented above reveal the importance of local snowpack emissions in regulating
410 the air quality over the Arctic. Improvements in global CTM could likely be achieved by
411 coupling snowpack emissions of reactive gases and photochemistry modules in order to better
412 simulate O₃ and O₃ precursors over snow and ice (Zatko et al., 2016). Moreover, this study also
413 demonstrates that anthropogenic emissions from midlatitudes play an important role in affecting
414 the Arctic air quality.

415 **Acknowledgements** This research was funded by U.S. EPA grant 83518901. Findings are solely
416 the responsibility of the grantee and do not necessarily represent the official views of the U.S.

417 EPA. Further, U.S. EPA does not endorse the purchase of any commercial products or services
418 mentioned in the publication. Superior, a high performance computing cluster at Michigan
419 Technological University, was used in obtaining results presented in this publication. L. J.
420 Kramer, D. Helmig and R. E. Honrath thank NASA (grant NNX07AR26G) for supporting the
421 measurements at Summit. S. Wu acknowledges the sabbatical fellowship from the Ocean
422 University of China. We acknowledge the observational dataset of O₃ and CO provided by
423 NOAA ESRL. Technical supports from M. Sulprizio and C. Keller are also acknowledged.

424

425 **References**

- 426 Alvarado, M. J., Logan, J. A., Mao, J., Apel, E., Riemer, D., Blake, D., Cohen, R. C., Min, K.-E.,
427 Perring, A. E., Browne, E. C., Wooldridge, P. J., Diskin, G. S., Sachse, G. W., Fuelberg, H.,
428 Sessions, W. R., Harrigan, D. L., Huey, G., Liao, J., Case-Hanks, A., Jimenez, J. L.,
429 Cubison, M. J., Vay, S. A., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Flocke, F. M.,
430 Pollack, I. B., Wennberg, P. O., Kurten, A., Crouse, J., St. Clair, J. M., Wisthaler, A.,
431 Mikoviny, T., Yantosca, R. M., Carouge, C. C., and Le Sager, P.: Nitrogen oxides and PAN in
432 plumes from boreal fires during ARCTAS-B and their impact on ozone: an integrated analysis
433 of aircraft and satellite observations, *Atmos. Chem. Phys.*, 10, 9739-9760,
434 doi:10.5194/acp-10-9739-2010, 2010.
- 435 Arnold, S. R., Emmons, L. K., Monks, S. A., Law, K. S., Ridley, D. A., Turquety, S., Tilmes, S.,
436 Thomas, J. L., Bouarar, I., Flemming, J., Huijnen, V., Mao, J., Duncan, B. N., Steenrod, S.,
437 Yoshida, Y., Langner, J., and Long, Y.: Biomass burning influence on high-latitude
438 tropospheric ozone and reactive nitrogen in summer 2008: a multimodel analysis based on
439 POLMIP simulations, *Atmos. Chem. Phys.*, 15, 6047-6068, doi:10.5194/acp-15-6047-2015,

440 2015.

441 Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q., Liu, H.,
442 Mickley L. J., and Schultz, M. G.: Global modeling of tropospheric chemistry with assimilated
443 meteorology: Model description and evaluation, *J. Geophys. Res.-Atmos.*, 106, 23073-23095,
444 2001.

445 Chen, D., Wang, Y., McElroy, M. B., He, K., Yantosca, R. M., and Le Sager, P.: Regional CO
446 pollution and export in China simulated by the high-resolution nested-grid GEOS-Chem
447 model, *Atmos. Chem. Phys.*, 9, 3825–3839, doi:10.5194/acp-9-3825-2009, 2009.

448 Choi, H., Liu, H., Crawford, J. H., Considine, D. B., Allen, D. J., Duncan, B. N., Horowitz, L.
449 W., Rodriguez, J. M., Strahan, S. E., Zhang, L., Liu, X., Damon, M. R., and Steenrod, S. D.:
450 Global O₃-CO correlations in a chemistry and transport model during July-August: evaluation
451 with TES satellite observations and sensitivity to input meteorological data and emissions,
452 *Atmos. Chem. Phys.*, 17, 8429–8452, doi:10.5194/acp-17-8429-2017, 2017.

453 Christian, K. E., Brune, W. H., and Mao, J.: Global sensitivity analysis of the GEOS-Chem
454 chemical transport model: ozone and hydrogen oxides during ARCTAS (2008), *Atmos. Chem.*
455 *Phys.*, 17, 3769-3784, doi:10.5194/acp-17-3769-2017, 2017.

456 Cohen, L., Helmig, D., Neff, W., Grachev, A., and Fairall, C.: Boundary-layer dynamics and its
457 influence on atmospheric chemistry at Summit, Greenland, *Atmos. Environ.*, 41, 5044–5060,
458 2007.

459 Crawford, J. H., Davis, D. D., Chen, G., Buhr, M., Oltmans, S., Weller, R., Mauldin, L., Eisele,
460 F., Shetter, R., Lefer, B., Ari-moto, R., and Hogan, A.: Evidence for photochemical produc-

461 tion of ozone at the South Pole surface, *Geophys. Res. Lett.*, 28, 3641–3644, 2001.

462 Dibb, J. E., Arsenault, M., Peterson, M. C., and Honrath, R. E.: Fast nitrogen oxide
463 photochemistry in Summit, Greenland snow, *Atmospheric Environment*, 36, 2501-2511, 2002.

464 Dibb, J. E., Whitlow, S. I., Arsenault, M.: Seasonal variations in the soluble ion content of snow
465 at Summit, Greenland: Constraints from three years of daily surface snow samples, *Atmos.*
466 *Environ.*, 41, 5007-5019, doi:10.1016/j.atmosenv.2006.12.010, 2007.

467 Fiore, A. M., Jacob, D. J., Field, B. D., Streets, D. G., Fernandes, S. D., and Jang, C.: Linking
468 ozone pollution and climate change: The case for controlling methane, *Geophys. Res. Lett.*,
469 29, 1919, doi:10.1029/2002GL015601, 2002.

470 Fischer, E. V., Jacob, D. J., Yantosca, R. M., Sulprizio, M. P., Millet, D. B., Mao, J., Paulot, F.,
471 Singh, H. B., Roiger, A., Ries, L., Talbot, R. W., Dzepina, K., and Pandey Deolal, S.:
472 Atmospheric peroxyacetyl nitrate (PAN): a global budget and source attribution, *Atmos.*
473 *Chem. Phys.*, 14, 2679-2698, doi:10.5194/acp-14-2679-2014, 2014.

474 Ford, K. M., Shepson, P. B., Bertman, S. B., Honrath, R. E., Peterson, M., Dibb, J. E., and
475 Bottenheim, J. W.: Studies of peroxyacetyl nitrate (PAN) and its interaction with the
476 snowpack at Summit, Greenland, *J. Geophys. Res.*, 107, ACH6, doi:10.1029/2001JD000547,
477 2002.

478 Frey, M. M., Brough, N., France, J. L., Anderson, P. S., Traulle, O., King, M. D., Jones, A. E.,
479 Wolff, E. W., and Savarino, J.: The diurnal variability of atmospheric nitrogen oxides (NO and
480 NO₂) above the Antarctic Plateau driven by atmospheric stability and snow emissions, *Atmos.*
481 *Chem. Phys.*, 13, 3045-3042, doi:10.5194/acp-13-3045-2013, 2013.

482 Giglio, L., Randerson, J. T., and van der Werf, G. R.: Analysis of daily, monthly, and annual
483 burned area using the fourth-generation global fire emissions database (GFED4), *J. Geophys.*
484 *Res. Biogeosci.*, 118, 1, 317-328, doi:10.1002/jgrg.20042, 2013.

485 Grannas, A. M., Jones, A. E., Dibb, J., Ammann, M., Anastasio, C., Beine, H. J., Bergin, M.,
486 Bottenheim, J., Boxe, C. S., Carver, G., Chen, G., Crawford, J. H., Domine, F., Frey, M. M.,
487 Guzman, M. I., Heard, D. E., Helmig, D., Hoffmann, M.R., Honrath, R. E., Huey, L. G.,
488 Hutterli, M., Jacobi, H. W., Klán, P., Lefer, B., McConnell, J., Plane, J., Sander, R., Savarino,
489 J., Shepson, P. B., Simpson, W. R., Sodeau, J. R., von Glasow, R., Weller, R., Wolff, E. W.,
490 and Zhu, T.: An overview of snow photochemistry: evidence, mechanisms and impacts,
491 *Atmos. Chem. Phys.*, 7, 4329–4373, doi:10.5194/acp-7-4329-2007, 2007.

492 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron C.: Estimate of
493 global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
494 Aerosols from Nature), *Atmos. Chem. Phys.*, 6, 3181-3210, doi:10.5194/acp-6-3181-2006,
495 2006.

496 Helmig, D., Oltmans, S. J., Carlson, D., Lamarque, J. F., Jones, A., Labuschagne, C., Anlauf, K.,
497 Hayden, K.: A review of surface ozone in the polar regions, *Atmospheric Environment*, 41,
498 5138-5161, 2007a.

499 Helmig, D., Oltmans, S. J., Morse, T. O., and Dibb, J. E.: What is causing high ozone at Summit,
500 Greenland?, *Atmos. Environ.*, 41, 5031-5043, doi:10.1016/j.atmosenv.2006.05.084, 2007b.

501 Helmig, D., Ganzeveld, L., Butler, T., and Oltmans, S. J.: The role of ozone atmosphere-snow
502 gas exchange on polar, boundary-layer tropospheric ozone- a review and sensitivity analysis,

503 Atmos. Chem. Phys., 7, 15-30, doi:10.5194/acp-7-15-2007, 2007c.

504 Helmig, D., Johnson, B., Oltmans, S. J., Neff, W., Eisele, F., and Davis, D. D.: Elevated ozone in
505 the boundary-layer at South Pole, Atmos. Environ., 42, 2788–2803, 2008.

506 Helmig, D., Petrenko, V., Martinerie, P., Witrant, E., Röckmann, T., Zuiderweg, A., Holzinger,
507 R., Hueber, J., Thompson, C., White, J. W. C., Sturges, W., Baker, A., Blunier, T., Etheridge,
508 D., Rubino, M., and Tans, P.: Reconstruction of Northern Hemisphere 1950–2010
509 atmospheric non-methane hydrocarbons, Atmos. Chem. Phys., 14, 1463–1483,
510 doi:10.5194/acp-14-1463-2014, 2014a.

511 Helmig, D., Stephens, C., Caramore, J., and Hueber, J.: Seasonal behavior of non-methane
512 hydrocarbons in the firn air at Summit, Greenland, Atmos. Environ., 85, 234-246,
513 doi:10.1016/j.atmosenv.2013.11.021, 2014b.

514 Hickman, J. E., Huang, Y., Wu, S., Diru, W., Groffman, P. M., Tully, K. L., and Palm, C. A.:
515 Nonlinear response of nitric oxide fluxes to fertilizer inputs and the impacts of agricultural
516 intensification on tropospheric ozone pollution in Kenya, Glob. Change Biol., 23, 3193-3204,
517 doi:10.1111/gcb.13644, 2017.

518 Hollaway, M. J., Arnold, S. R., Challinor, A. J., and Emberson, L. D.: Intercontinental trans-
519 boundary contributions to ozone-induced crop yield losses in the North Hemisphere,
520 Biogeosciences, 9, 271-292, doi: 10.5194/bg-9-271-2012, 2012.

521 Honrath, R. E., Peterson, M. C., Guo, S., Dibb, J. E., Shepson, P. B., and Campbell, B.: Evidence
522 of NO_x production within or upon ice particles in the Greenland snowpack, Geophys. Res.
523 Lett., 26, 695-698, 1999.

524 Honrath, R. E., Guo, S., Peterson, M. C., Dziobak, M. P., Dibb, J. E., and Arsenault, M. A.:
525 Photochemical production of gas phase NO_x from ice crystal NO₃⁻, *J. Geophys. Res.*, 105,
526 24183–24190, 2000a.

527 Honrath, R. E., Peterson, M. C., Dziobak, M. P., Dibb, J. E., Arsenault, M. A., and Green, S. A.:
528 Release of NO_x from Sunlight-irradiated Midlatitude Snow, *Geophys. Res. Lett.*, 27, 2237–
529 2240, 2000b.

530 Honrath, R. E., Lu, Y., Peterson, M. C., Dibb, J. E., Arsenault, M. A., Cullen, N. J., and Steffen,
531 K.: Vertical fluxes of NO_x, HONO, and HNO₃ above the snowpack at Summit, Greenland,
532 *Atmos. Environ.*, 36, 2629-2640, doi:10.1016/S1352-2310(02)00132-2, 2002.

533 Huang, Y., Wu, S., Dubey, M. K., and French, N. H. F.: Impact of aging mechanism on model
534 simulated carbonaceous aerosols, *Atmos. Chem. Phys.*, 13, 6329–6343, doi:10.5194/acp-13-
535 6329-2013, 2013.

536 Hudman, R. C., Moore, N. E., Mebust, A. K., Martin, R. V., Russell, A. R., Valin, L. C., and
537 Cohen, R. C.: Steps towards a mechanistic model of global soil nitric oxide emissions:
538 implementation and space based-constraints, *Atmos. Chem. Phys.*, 12, 7779-7795, doi:
539 10.5194/acp-12-7779- 2012, 2012.

540 Hudman, R. C., Murray, L. T., Jacob, D. J., Turquety, S., Wu, S., Millet, D. B., Avery, M.,
541 Goldstein, A. H., and Holloway, J.: North American influence on tropospheric ozone and the
542 effects of recent emission reductions: Constraints from ICARTT observations, *J. Geophys.*
543 *Res.*, 114, D07302, doi:10.1029/2008JD010126, 2009.

544 Jacob, D. J., Wofsy, S. C., Bakwin, P. S., Fan, S.-M., Harriss, R. C., Talbot, R. W., Bradshaw, J.
545 D., Sandholm, S. T., Singh, H. B., Browell, E. V., Gregory, G. L., Sachse, G. W., Shipham,

546 M. C., Blake, D. R., and Fitzjarrald, D. R.: Summertime photochemistry of the troposphere at
547 high northern latitudes, *J. Geophys. Res.*, 97, D15, 16421-16431, doi:10.1029/91JD01968,
548 1992.

549 Jang, J.-C., Jeffries, H., Byun, D., and Pleim, J.: Sensitivity of ozone to model grid resolution – I.
550 Application of high resolution regional acid deposition model, *Atmos. Environ.*, 29, 3085-
551 3100, doi:10.1016/1352-2310(95)00118-I, 1995.

552 Johnson, M.S., Meskhidze, N., Solmon, F., Gasso, S., Chuang, P. Y., Gaiero, D. M., Yantosca,
553 R. M., Wu, S., Wang, X., Carouge, C.: Modeling Dust and Soluble Iron Deposition to the
554 South Atlantic Ocean, *J. Geophys. Res.*, 115, D15202, doi:10.1029/2009JD013311, 2010.

555 Kramer, L. J., Helmig, D., Burkhart, J. F., Stohl, A., Oltmans, S., and Honrath, R. E.: Seasonal
556 variability of atmospheric nitrogen oxides and non-methane hydrocarbons at the GEOSummit
557 station, Greenland, *Atmos. Chem. Phys.*, 15, 6827-6849, doi:10.5194/acp-15-6827-2015,
558 2015.

559 Kuhns, H., Knipping, E. M., and Vukovich, J. M.: Development of a United States-Mexico emis-
560 sions inventory for the Big Bend Regional Aerosol and Visibility Observational (BRAVO)
561 Study, *JAPCA J. Air Waste M.*, 55, 677–692, 2005.

562 Kumar, A., Wu, S., Weise, M. F., Honrath, R., Owen, R. C., Helmig, D., Kramer, L., Val Martin,
563 M., and Li, Q.: Free-troposphere ozone and carbon monoxide over the North Atlantic for
564 2001–2011, *Atmos. Chem. Phys.*, 13, 12537-12547, doi:10.5194/acp-13-12537-2013, 2013.

565 Legrand, M., Preunkert, S., Frey, M., Bartels-Rausch, Th., Kukui, A., King, M. D., Savarino, J.,
566 Kerbrat, M., and Jourdain, B.: Large mixing ratios of atmospheric nitrous acid (HONO) at

567 Concordia (East Antarctic Plateau) in summer: a strong source from surface snow?, *Atmos.*
568 *Chem. Phys.*, 14, 9963-9976, doi:10.5194/acp-14-9963-2014, 2014.

569 Li, M., Zhang, Q., Kurokawa, J., Woo, J. H., He, K. B., Lu, Z., Ohara, T., Song, Y., Streets, D.
570 G., Carmichael, G. R., Cheng, Y. F., Hong, C. P., Huo, H., Jiang, X. J., Kang, S. C., Liu, F.,
571 Su, H., and Zheng, B.: MIX: a mosaic Asian anthropogenic emission inventory under the
572 international collaboration framework of the MICS-Asia and the HTAP, *Atmos. Chem. Phys.*,
573 17, 935-963, doi:10.5194/acpd-17-935-2017.

574 Liang, J., and Jacobson, M. Z.: Effects of subgrid segregation on ozone production efficiency in
575 a chemical model, *Atmos. Environ.*, 34, 2975–2982, doi:10.1016/S1352-2310(99)00520-8,
576 2000.

577 Liang, Q., Rodriguez, J. M., Douglass, A. R., Crawford, J. H., Olson, J. R., Apel, E., Bian, H.,
578 Blake, D. R., Brune, W., Chin, M., Colarco, P. R., da Silva, A., Diskin, G. S., Duncan, B. N.,
579 Huey, L. G., Knapp, D. J., Montzka, D. D., Nielsen, J. E., Pawson, S., Riemer, D. D.,
580 Weinheimer, A. J., and Wisthaler, A.: Reactive nitrogen, ozone and ozone production in the
581 Arctic troposphere and the impact of stratosphere-troposphere exchange, *Atmos. Chem. Phys.*,
582 11, 13181-13199, doi:10.5194/acp-11-13181-2011, 2011.

583 Liu, H. Y., Jacob, D. J., Bey, I., and Yantosca, R. M.: Constraints from pb-210 and Be-7 on wet
584 deposition and transport in a global three-dimensional chemical tracer model driven by
585 assimilated meteorological fields, *J. Geophys. Res.-Atmos.*, 106, 12109-12128, doi:
586 10.1029/2000JD900839, 2001.

587 Martin, R. V., Jacob, D. J., Logan, J. A., Bey, I., Yantosca, R. M., Staudt, A. C., Li, Q., Fiore, A.

588 M., Duncan, B. N., and Liu, H.: Interpretation of TOMS observations of tropical tropospheric
589 ozone with a global model and in situ observations, *J. Geophys. Res.*, 107(D18), ACH 4-1-
590 ACH 4-27, doi: 10.1029/2001JD001480, 2002.

591 McClure-Begley, A., Petropavlovskikh, I., Oltmans, S.: NOAA Global Monitoring Surface
592 Ozone Network. 1973-2014. National Oceanic and Atmospheric Administration, Earth
593 Systems Research Laboratory Global Monitoring Division. Boulder, CO. DATE ACCESSED:
594 4/23/2017, <http://dx.doi.org/10.7289/V57P8WBF>, 2014.

595 McLinden, C. A., Olsen, S. C., Hannegan, B., Wild, O., Prather, M. J., and Sundet, J.:
596 Stratospheric ozone in 3-D models: A simple chemistry and the cross-tropopause flux, *J.*
597 *Geophys. Res.*, 105, D11, 14653-14665, doi:10.1029/2000JD900124, 2000.

598 Monks, S. A., Arnold, S. R., Emmons, L. K., Law, K. S., Turquety, S., Duncan, B. N.,
599 Flemming, J., Huijnen, V., Tilmes, S., Langner, J., Mao, J., Long, Y., Thomas, J. L., Steenrod,
600 S. D., Raut, J. C., Wilson, C., Chipperfield, M. P., Diskin, G. S., Weinheimer, A., Schlager,
601 H., and Ancellet, G.: Multi-model study of chemical and physical controls on transport of
602 anthropogenic and biomass burning pollution to the Arctic, *Atmos. Chem. Phys.*, 15, 3575-
603 3603, doi:10.5194/acp-15-3575-2015, 2015.

604 Murray, K. A., Kramer, L. J., Doskey, P. V., Ganzeveld, L., Seok, B., Van Dam, B., and Helmig,
605 D.: Dynamics of ozone and nitrogen oxides at Summit, Greenland. II. Simulating snowpack
606 chemistry during a spring high ozone event with a 1-D process-scale model, *Atmos. Environ.*,
607 117, 110–123, doi:10.1016/j.atmosenv.2015.07.004, 2015.

608 Murray, L. T., Jacob, D. J., Logan, J. A., Hudman, R. C., and Koshak, W. J.: Optimized regional

609 and interannual variability of lightning in a global chemical transport constrained by LIS/OTD
610 satellite data, *J. Geophys. Res.*, 117, D20307, doi:10.1029/2012JD017934, 2012.

611 Novelli, P. C. and Masarie, K. A.: Atmospheric Carbon Monoxide Dry Air Mole Fraction from
612 the NOAA ESRL Carbon Cycle Cooperative Global Air Sampling Network, 1988-2014,
613 Version: 2015-08-04, NOAA, available at
614 ftp://aftp.cmdl.noaa.gov/data/trace_gases/co/flask/surface/, 2015.

615 Park, R. J., Jacob, D. J., Field, B. D., Yantosca, R. M., and Chin, M.: Natural and transboundary
616 pollution influences on sulfate-nitrate-ammonium aerosols in the United States: Implications
617 for policy, *J. Geophys. Res.-Atmos.*, 109, D15204, doi:10.1029/2003JD004473, 2004.

618 Petropavlovskikh, I. and Oltmans, S. J.: Tropospheric Ozone Measurements, 1973-2011,
619 Version: 2012-07-10, NOAA, available at: <ftp://aftp.cmdl.noaa.gov/data/ozwv/SurfaceOzone/>,
620 2012.

621 Price, C. and Rind, D.: A simple lightning parameterization for calculating global lightning
622 distributions, *J. Geophys. Res.*, 97, 9919-9933, doi:10.1029/92JD00719, 1992.

623 Ridley, B. A., and Grahek, F.: A small, low flow, high sensitivity reaction vessel for NO
624 chemiluminescence detectors, *Am. Meteorol. Soc.*, 7, 307-311, 1990.

625 Sauvage, B., Martin, R. V., van Donkelaar, A., Liu, X., Chance, K., Jaeglé, L., Palmer, P. I., Wu,
626 S., and Fu, T.-M.: Remote sensed and in situ constraints on processes affecting tropical tropo-
627 spheric ozone, *Atmos. Chem. Phys.*, 7, 815-838, doi:10.5194/acp-7-815-2007, 2007.

628 Sherwen, T., Schmidt, J. A., Evans, M. J., Carpenter, L. J., Großmann, K., Eastham, S. D.,
629 Jacob, D. J., Dix, B., Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A.,

630 Prados-Roman, C., Mahajan, A. S., and Ordóñez, C.: Global impacts of tropospheric halogens
631 (Cl, Br, I) on oxidants and composition in GEOS-Chem, *Atmos. Chem. Phys.*, 16, 12239–
632 12271, doi:10.5194/acp-16-12239-2016, 2016.

633 Shindell, D. T., Chin, M., Dentener, F., Doherty, R. M., Faluvegi, G., Fiore, A. M., Hess, P.,
634 Koch, D. M., MacKenzie, I. A., Sanderson, M. G., Schultz, M., Stevenson, D. S., Teich, H.,
635 Textor, C., Wild, O., Bergmann, D. J., Bey, I., Bian, H., Cuvelier, C., Duncan, B. N., Folberth,
636 G., Horowitz, L. W., Jonson, J., Kaminski, J. W., Marmer, E., Park, R., Pringle, K. J.,
637 Schroeder, S., Szopa, S., Takemura, T., Zeng, G., Keating, T. J., and Zuber, A.: A multi-model
638 assessment of pollution transport to the Arctic, *Atmos. Chem. Phys.*, 8, 5353-5372,
639 doi:10.5194/acp-8-5353-2008, 2008.

640 Simon, H., Beck, L., Bhave, P. V., Divita, F., Hsu, Y., Luecken, D., Mobley, J. D., Pouliot, G.
641 A., Reff, A., Sarwar, G., and Strum, M.: The development and uses of EPA's SPECIATE
642 database, *Atmospheric Pollution Research*, 196-206, 10.5094/apr.2010.026, 2010.

643 Thomas, J. L., Stutz, J., Lefer, B., Huey, L. G., Toyota, K., Dibb, J. E., and von Glasow, R.:
644 Modeling chemistry in and above snow at Summit, Greenland – Part 1: Model description and
645 results, *Atmos. Chem. Phys.*, 11, 4899–4914, doi:10.5194/acp-11-4899-2011, 2011.

646 Thomas, J. L., Dibb, J. E., Huey, L. G., Liao, J., Tanner, D., Lefer, B., von Glasow, R., and
647 Stutz, J.: Modeling chemistry in and above snow at Summit, Greenland – Part 2: Impact of
648 snowpack chemistry on the oxidation capacity of the boundary layer, *Atmos. Chem. Phys.*, 12,
649 6537–6554, doi:10.5194/acp-12-6537-2012, 2012.

650 Tzompa-Sosa, Z. A., Mahieu, E., Franco, B., Keller, C. A., Turner, A. J., Helmig, D., Fried, A.,

651 Richter, D., Weibring, P., Walega, J., Yacovitch, T. I., Herndon, S. C., Blake, D. R., Hase, F.,
652 Hannigan, J. W., Conway, S., Strong, K., Schneider, M., and Fischer, E. V.: Revisiting global
653 fossil fuel and biofuel emissions of ethane, *J. Geophys. Res. Atmos.*, 122,
654 doi:10.1002/2016JD025767, 2017.

655 Unger, N., Shindell, D. T., Koch, D. M., and Streets, D. G.: Cross influences of ozone and
656 sulfate precursor emissions changes on air quality and climate, *Proc. Natl. Acad. Sci.*, 103,
657 4377-4380, doi:10.1073/pnas.0508769103, 2006.

658 van het Bolscher, M., Pereira, J., Spesso, A., Dalsoren, S., van Noije, T., and Szopa, S.:
659 REanalysis of the TROpospheric chemical composition over the past 40 years: A long-term
660 global modeling study of tropospheric chemistry, Max Plank Inst. For Meteorology, Hamburg,
661 Germany, 77, 2008.

662 Van Dam, B., Helmig, D., Toro, C., Doskey, P., Kramer, L., Murray, K., Ganzeveld, L., and
663 Seok, B.: Dynamics of ozone and nitrogen oxides at Summit, Greenland: I. Multi-year
664 observations in the snowpack, *Atmos. Environ.*, 123, 268-284,
665 doi:10.1016/j.atmosenv.2015.09.060, 2015.

666 Walker, T. W., Jones, D. B. A., Parrington, M., Henze, D. K., Murray, L. T., Bottenheim, J. W.,
667 Anlauf, K., Worden, J. R., Bowman, K. W., Shim, C., Singh, K., Kopacz, M., Tarasick, D. W.,
668 Davies, J., von der Gathen, P., Thompson, A. M., and Carouge, C. C.: Impacts of midlatitude
669 precursor emissions and local photochemistry on ozone abundances in the Arctic, *J. Geophys.*
670 *Res.*, 117, D01305, doi:10.1029/2011JD016370, 2012.

671 Wang, Y. H., Jacob, D. J., and Logan, J. A.: Global simulation of tropospheric O₃-NO_x-hydro-

672 carbon chemistry 1. Model formulation, *J. Geophys. Res.-Atmos.*, 103, 10713-10725, doi:
673 10.1029/98JD00158, 1998.

674 Wang, Y. X., McElroy, M. B., Jacob, D. J., and Yantosca, R. M.: A nested grid formulation for
675 chemical transport over Asia: Applications to CO, *J. Geophys. Res.*, 109, D22307,
676 doi:10.1029/2004JD005237, 2004.

677 Wesely, M. L.: Parameterization of surface resistances to gaseous dry deposition in regional-
678 scale numerical-models, *Atmos. Environ.*, 23, 1293-1304, doi:10.1016/0004-6981(89)90153-
679 4, 1989.

680 Wespes, C., Emmons, L., Edwards, D. P., Hannigan, J., Hurtmans, D., Saunio, M., Coheur, P.-
681 F., Clerbaux, C., Coffey, M. T., Batchelor, R. L., Lindenmaier, R., Strong, K., Weinheimer, A.
682 J., Nowak, J. B., Ryerson, T. B., Crouse, J. D., and Wennberg, P. O.: Analysis of ozone and
683 nitric acid in spring and summer Arctic pollution using aircraft, ground-based, satellite
684 observations and MOZART-4 model: source attribution and partitioning, *Atmos. Chem. Phys.*,
685 12, 237–259, doi:10.5194/acp- 12-237-2012, 2012.

686 Wu, S., Mickley, L. J., Jacob, D. J., Logan, J. A., Yantosca, R. M., and Rind, D.: Why are there
687 large differences between models in global budgets of tropospheric ozone?, *J. Geophys. Res.*,
688 112(D5), D05302, doi:10.1029/02006JD007801, 2007.

689 Xiao, Y., Logan, J. A., Jacob, D. J., Hudman, R. C., Yantosca, R., and Blake, D. R.: The global
690 budget of ethane and regional constraints on U.S. sources, *J. Geophys. Res.*, 113, D21306,
691 doi:10.1029/2007JD009415, 2008.

692 Yang, J., Honrath, R. E., Peterson, M. C., Dibb, J. E., Sumner, A. L., Shepson, P. B., Frey, M.,

693 Jacobi, H.-W., Swanson, A., and Blake, N.: Impacts of snowpack emissions on deduced levels
694 of OH and peroxy radicals at Summit, Greenland, *Atmos. Environ.*, 36, 2523-2534,
695 doi:10.1016/S1352-2310(02)00128-0, 2002.

696 Yevich, R., and Logan, J. A.: An assesment of biofuel use and burning of agricultural waste in
697 the developing world, *Global Biogeochem. Cy.*, 17, 1095, doi:10.1029/2002GB001952, 2003.

698 Yu, K., Jacob, D. J., Fisher, J. A., Kim, P. S., Marais, E. A., Miller, C. C., Travis, K. R., Zhu, L.,
699 Yantosca, R. M., Sulprizio, M. P., Cohen, R. C., Dibb, J. E., Fried, A., Mikoviny, T., Ryerson,
700 T. B., Wennberg, P. O., and Wisthaler, A.: Sensitivity to grid resolution in the ability of a
701 chemical transport model to simu- late observed oxidant chemistry under high-isoprene
702 conditions, *Atmos. Chem. Phys.*, 16, 4369–4378, doi:10.5194/acp-16-4369- 2016, 2016.

703 Zatzko, M. C., Geng, L., Alexander, B., Sofen, E. D., and Klein, K.: The impact of snow nitrate
704 photolysis on boundary layer chemistry and the recycling and redistribution of reactive
705 nitrogen across Antarctica and Greenland in a global chemical transport model, *Atmos. Chem.*
706 *Phys.*, 16, 2819-2842, doi:10.5194/acpd-16-2819-2016, 2016.

707 Zhang, H., Wu, S., Huang, Y., and Wang, Y.: Effects of stratospheric ozone recovery on
708 photochemistry and ozone air quality in the troposphere, *Atmos. Chem. Phys.*, 14, 4079-4086,
709 doi:10.5194/acp-14-4079-2014, 2014.

710 Zhou, X., Beine, H. J., Honrath, R. E., Fuentes, J., Simpson, W., Shepson, P. B., and Bottenheim,
711 J. W.: snowpack photochemical production of HONO: a major source of OH in the Arctic
712 boundary layer in springtime, *Geophys. Res. Lett.*, 28, 21, 4087-4090, 2001.

713

714 **Table 1.** Surface NO₂ measurements over Europe during 2009/12/01-2010/01/31.

Site ID	Site name	Lat. (°N)	Lon. (°E)	Altitude a.s.l (m)	Technique	Resolution
BE0001R	Offagne	49.88	5.20	430	chemiluminescence	hourly
BE0032R	Eupen	50.63	6	295	chemiluminescence	hourly
DE0001R	Westerland	54.93	8.31	12	NaJ_solution	daily
DK0008R	Anholt	56.72	11.52	40	UV_fluorescence	hourly
FI0096G	Pallas	67.97	24.12	340	chemiluminescence	hourly
GB0014R	High Muffles	54.33	-0.8	267	chemiluminescence	daily
NL0009R	Kollumerwaard	53.33	6.28	1	chemiluminescence	hourly
NO0001R	Birkenes	58.38	8.25	190	glass sinter	daily
NO0039R	Kårvatn	62.78	8.88	210	glass sinter	daily
NO0056R	Hurdal	60.37	11.08	300	glass sinter	daily
SE0005R	Bredkälén	63.85	15.3	404	abs_tube	daily

715

716

717

718

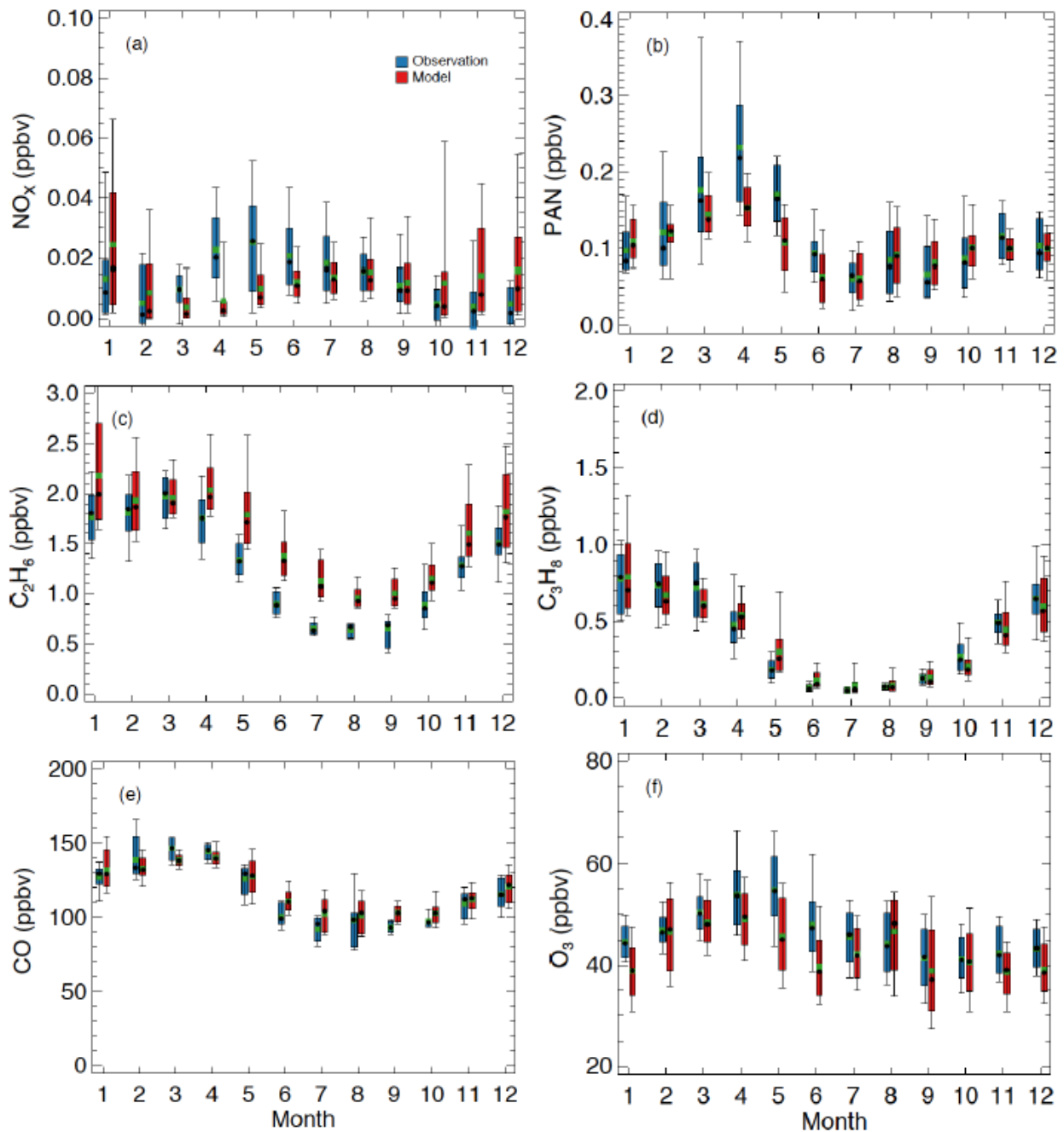
719

720

721

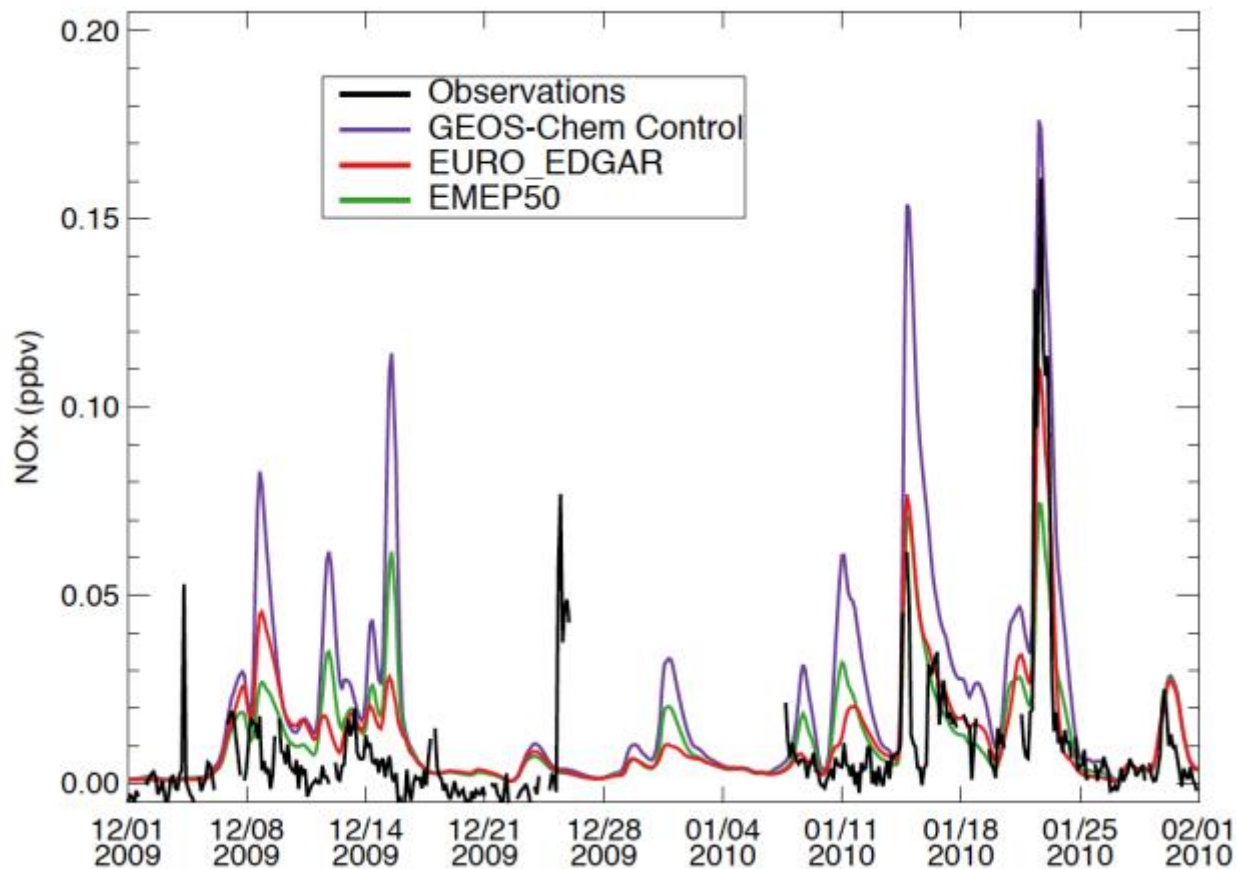
722

723



724

725 **Fig. 1.** Box plot comparison for seasonal variations of (a) NO_x , (b) PAN, (c) C_2H_6 , (d) C_3H_8 , (e)
 726 CO, and (f) O_3 between GEOS-Chem model simulations (red) and measurements (blue) at
 727 Summit for the period of 2008/07-2010/06. Data shown are monthly averages during this period.
 728 The thick (thin) bars represent the 67% (95%) confidence intervals. Black and green dots
 729 represent median and mean values, respectively. The statistics are based on daily averages.



730

731 **Figure 2.** Timeseries of surface NO_x mixing ratios over Summit from observations, GEOS-
 732 Chem model control simulations, EURO_EDGAR, and EMEP50 during 2009/12/01-2010/01/31.
 733 EURO_EDGAR represents simulations with anthropogenic NO_x emissions over Europe
 734 following EDGAR v4.2, while EMEP50 denotes simulations with anthropogenic NO_x emissions
 735 from the EMEP emission inventory over Europe reduced by 50%, with other model
 736 configurations identical to the control simulations. Readers are referred to the text for details.

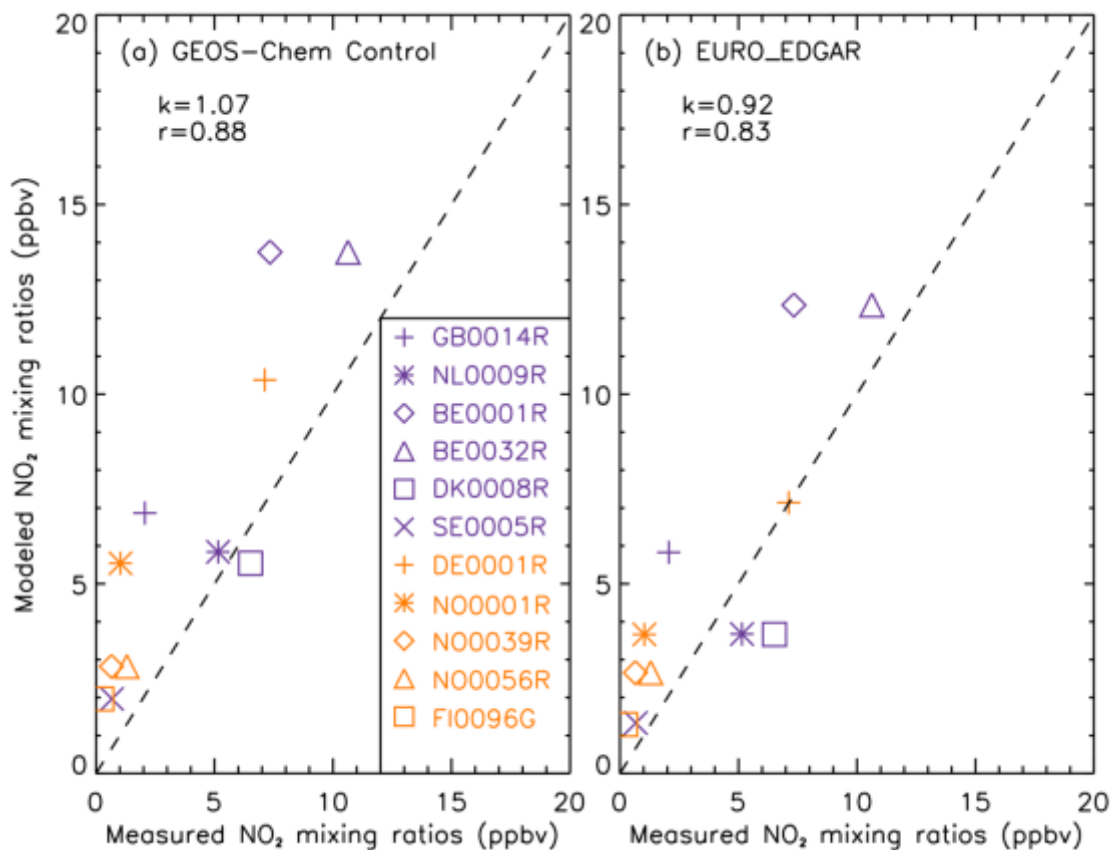
737

738

739

740

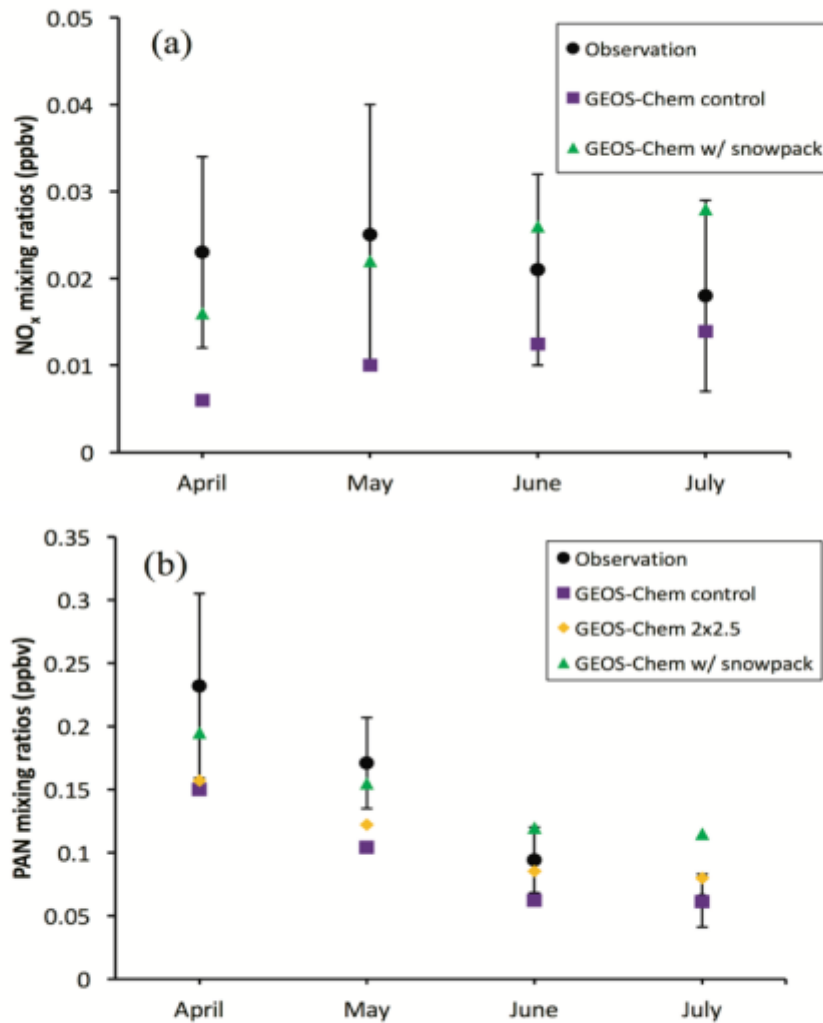
741



742

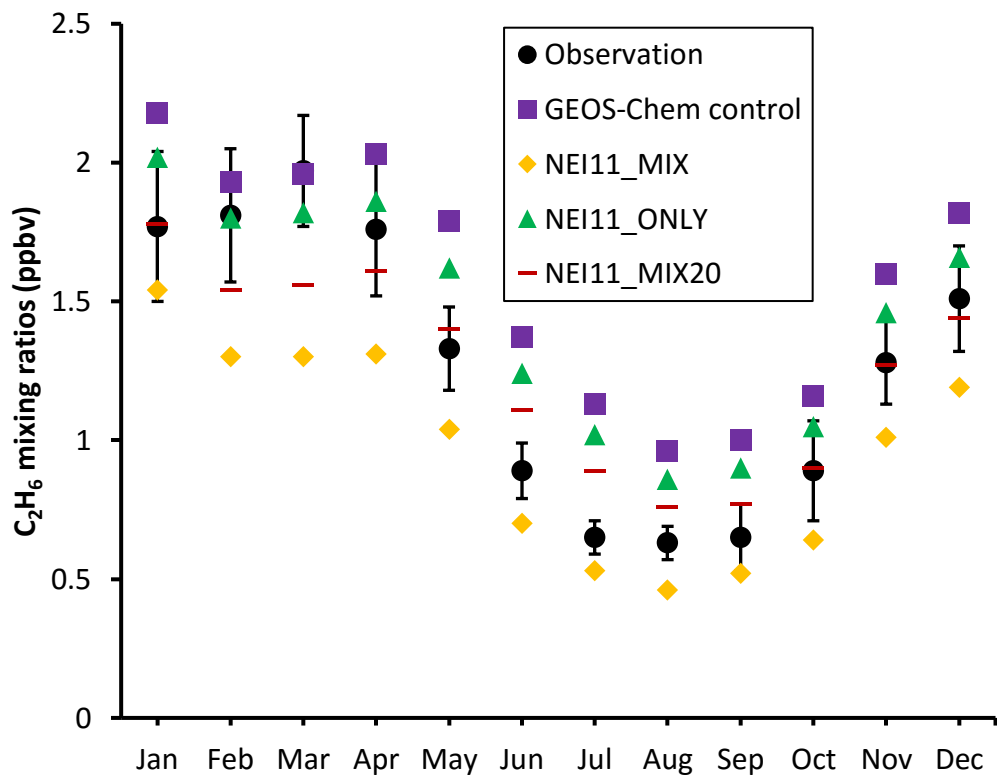
743 **Figure 3.** Scatter plots of model simulations from (a) GEOS-Chem control simulations and (b)
 744 EURO_EDGAR during 2009/12/01-2010/01/31 and measured monthly mean NO₂ mixing ratios
 745 at 11 observational sites over Europe; also shown is the corresponding model-to-observation
 746 slopes (k) and correlation coefficients (r) for each panel. The dashed line is the 1:1 ratio.
 747 Explanations of site abbreviations are listed in Table 1. EURO_EDGAR represents simulations
 748 with anthropogenic NO_x emissions over Europe following EDGAR v4.2, with other model
 749 configurations identical to the control simulations.

750



751

752 **Figure 4.** Monthly mean surface (a) NO_x and (b) PAN mixing ratios from observations (black
 753 circles), simulations with (green triangles) /without (purple squares) snowpack emissions, and
 754 GEOS-Chem simulations with horizontal grid resolution 2° x 2.5° (orange diamonds) for April-
 755 July during 07/2008-06/2010. Vertical bars denote standard deviations over the course of
 756 observations for each month.



757

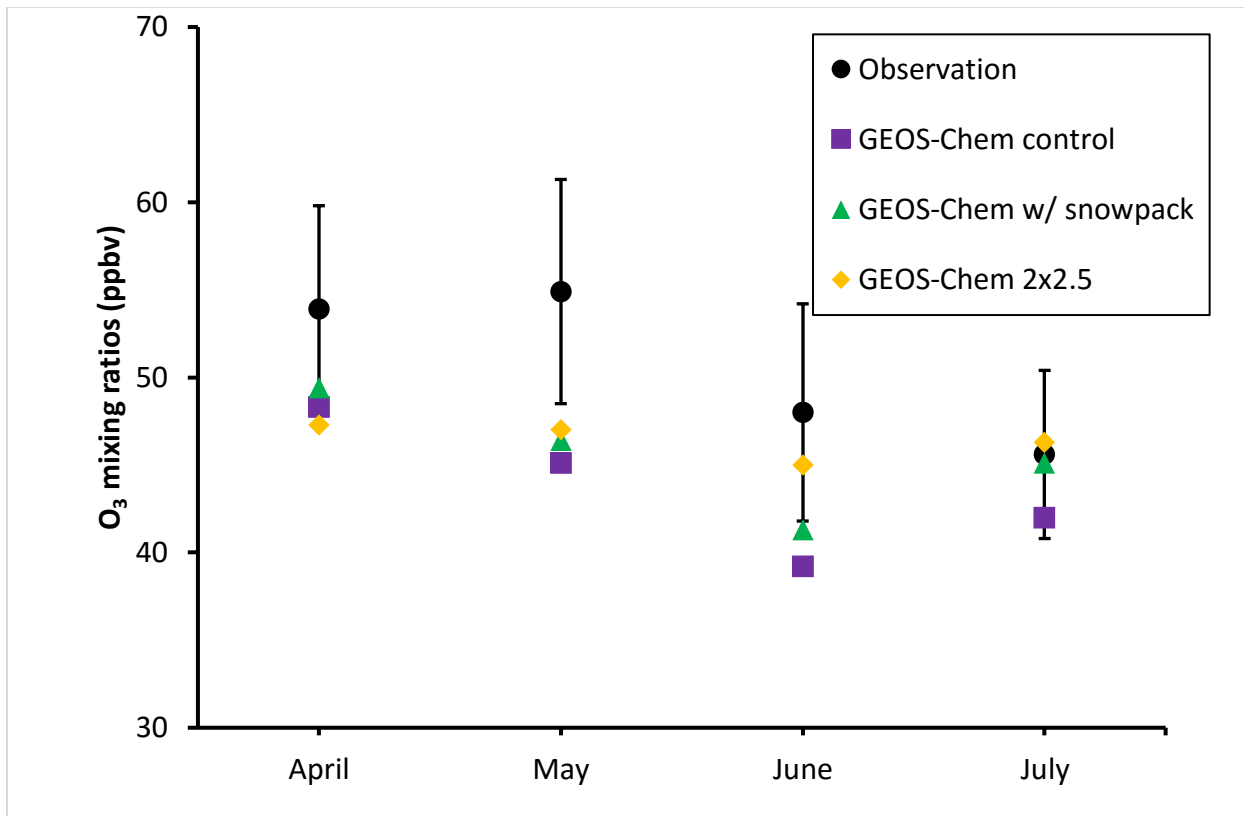
758 **Figure 5.** Monthly mean surface C₂H₆ mixing ratios at Summit from observations (black
 759 circles), GEOS-Chem model control simulations (purple squares), NEI11_MIX (orange
 760 diamond), and NEI11_ONLY (green triangles) simulations during 2008-2010; vertical bars
 761 denote the standard deviation over the course of observations for each month. NEI11_MIX
 762 represents model perturbations with global C₂H₆ emission inventories overwritten by NEI11 over
 763 US and by MIX over Asia, with other model configurations identical to the control simulations.
 764 NEI11_ONLY denotes the simulation that is the same with the control simulation except that the
 765 C₂H₆ emission inventory over US is overwritten by NEI11. NEI11_MIX20 is the simulation that
 766 is identical to NEI11_MIX except the 20% increased MIX C₂H₆ emission inventory over Asia.
 767 Readers are referred to the text for details.

768

769

770

771



772

773 **Figure 6.** Monthly mean surface O₃ mixing ratios from observations (black circles), GEOS-
 774 Chem control runs (purple squares), with snowpack chemistry (green triangles), and horizontal
 775 grid resolution 2° x 2.5° (orange diamonds) for April-July. Vertical bars denote the variability
 776 over the course of observations for each month.

777

778

779

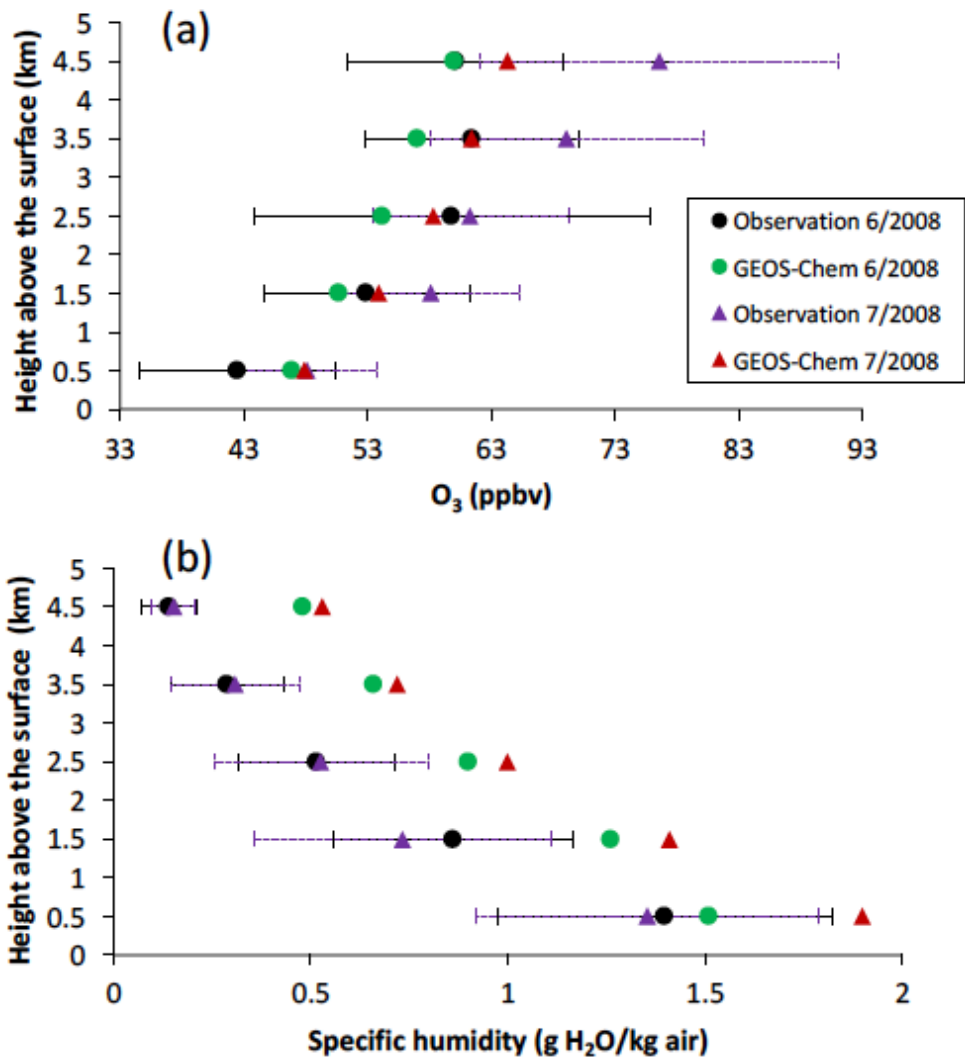
780

781

782

783

784



785

786 **Figure 7.** Comparisons of vertical profiles of (a) O₃ and (b) specific humidity between GEOS-
 787 Chem simulations and ozonesondes in June and July 2008 respectively, averaged over 1-km
 788 altitude bins. Black and green solid circles represent observations and simulations in June 2008
 789 while purple and red triangles denote observations and simulations for July 2008 respectively.
 790 Solid and dashed horizontal error bars represent observational standard deviations for June and
 791 July respectively.

792