Response to the editor:

We thank the editor for his helpful comments. Our responses to the comments are provided below, with editor's comments italicized and our responses in plain and bold fonts.

The response to the reviewer comments was adequate, except for two comments from Dr. Liu for which you responded well, but did not change the text of the paper. To avoid confusion, I'll just ask that you modify the text of the paper for these two comments before it can be accepted for publication:

3). Section 2, 2nd paragraph: "Time series data were archived with 3-hr temporal resolution at the Summit grid box" — I think you meant "grid column". Moreover, it is not clear how the model output was sampled in the vertical. The elevation of Summit is 3212m asl. Did you sample the model bottom layer, or the model vertical layer that is about 3212m above the sea level? The latter may very well be different than the former because the topography is not well resolved at coarse resolution. Would the results about model overestimates or underestimates found in this paper be different if the alternative way of model sampling is used (e.g., lines 206-207)?

Response: Thanks for pointing this out. We have modified the text according to our responses to Dr. Liu's comments as "Time series data were archived with 3-hr temporal resolution at the Summit grid box for each model vertical level, including the model bottom layer. For comparison with surface observations at Summit, Greenland, we sampled the data for the model bottom layer. We acknowledge that the topography is not well resolved at coarse model resolution, and we believe that the diagnostics for the model bottom layer would work better than the model vertical layer that is about 3212 m above the sea level for comparison to the surface measurements." (lines 161-166)

Line 77: "....used to be the global default anthropogenic C2H6 emission inventory" - Do you mean "default" in GEOS-Chem or any other models?

Response: We have revised the text as "In GEOS-Chem, RETRO used to serve as the default global anthropogenic emission inventory for C₂H₆, however, the annual budget of which has been shown too low compared with observations." (lines 189-191)

Response to reviewer #1:

We thank Reviewer #1 for his/her valuable and thoughtful comments. Our responses to the comments are provided below, with the reviewer's comments italicized and our responses in plain and bold fonts.

This paper describes an evaluation of tropospheric ozone and its precursor species simulated by the GEOS-Chem global chemical transport model (CTM) at the Summit observatory station in Greenland. Based on evaluation of the standard GEOS-Chem model, and deficiencies identified through comparison with observations, a number of model changes are implemented (mostly to emissions) which are shown to improve the model performance. The paper serves as a useful documentation of Greenland surface ozone, NO_y and VOC sensitivity to a number of key processes, and highlights processes that warrant further investigation to improve understanding of the surface Arctic ozone budget. These issues are important in light of recent studies demonstrating poor model performance for Arctic tropospheric ozone, as cited by the authors. The paper is generally well written, logically structured and is suitable for the journal. I would recommend publication of this manuscript in ACP, once the following minor issues have been addressed.

1) Paragraph beginning Line 69. The discussion of ethane appears a bit out of the blue. The authors should explain more clearly in the manuscript the importance and relevance of ethane to the previous discussion. i.e. give some context for how ethane is relevant to the study - which is motivated by understanding Arctic tropospheric ozone. i.e. as has been done for NOx, PAN.

Response: We thank the reviewer for pointing this out. We have reorganized the flow of the text related to ethane in the introduction section - a) we have deleted some discussions on ethane that are not closely related to our study here; b) We have added discussions on the importance of volatile organic compounds (e.g., ethane and propane) for the productions of ozone (lines 47-52) –

"Tropospheric ozone (O₃) and its precursors, including nitrogen oxides (NO_x = NO + NO₂), carbon monoxide (CO), and volatile organic compounds (VOCs, such as ethane, propane, etc.) are important atmospheric species affecting both air quality and climate (e.g., Jacob et al., 1992; Fiore et al., 2002; Unger et al., 2006; Hollaway et al., 2012). Tropospheric O₃ is

mainly produced by photochemical oxidation of CO and VOCs in the presence of NO_x , with additional contribution by transport from the stratosphere."

2) Lines 109-112: It is unclear here what is meant by fully coupled aerosol? Does this include size-resolved modal aerosol for example? Heterogeneous chemistry, semivolatile nitrate..?

Response: We have clarified this part to "Simulations of O_3 and related species (NO_x , PAN, NMHCs) are conducted using the GEOS-Chem model (Bey et al., 2001) with coupled O_3 -NO_x-VOC-Aerosol chemistry mechanism (i.e. these species interact with each other in the model)." (lines 148-150)

3) Lines 112-115: Discussion of previous GEOS-Chem evaluation. It would be helpful here to provide a few sentences for a brief but more critical review of what has been shown in terms of model performance with previous studies specifically using GEOS-Chem in the Arctic. e.g. sensitivity analysis by Christian et al., (2107), the recent POLMIP evaluation (see Monks et al., 2015). These have shown some important limitations and strengths that it would be useful to point out for context.

Response: This is an excellent point. We have added the descriptions of previous GEOS-Chem evaluations in the text. In the Introduction part, we have included the discussions of Monks et al. (2015) and Christian et al. (2017) as "More recently, Monks et al. (2015) further demonstrated that model simulated O₃ mixing ratios in the Arctic at the surface and in the upper troposphere were generally lower than the observations. In addition, a recent study by Christian et al. (2017) compared O₃ observations from the ARCTAS campaign to GEOS-Chem model simulations and found consistent low biases with the model simulated O₃ at all altitudes except the surface." (lines 87-91)

In Section 2, we have modified text in lines 112-115 in ACPD as "The GEOS-Chem model has been extensively evaluated and applied in a wide range of applications (Martin et al., 2002; Park et al., 2004; Wu et al., 2007; Hudman et al., 2009; Johnson et al., 2010; Huang et al., 2013; Kumar et al., 2013; Zhang et al., 2014; Hickman et al., 2017), including the studies

in the Arctic (e.g., Alvarado et al., 2010; Monks et al., 2015; Christian et al., 2017)." (lines 152-156)

4) Discussion of model NOx bias (first paragraph of page 5). Perhaps here quote the obs/model slope or model bias. You give figures for the slopes / correlations in the panels of Fig 3 but don't mention the numbers in the text.

Response: We agree with the reviewer. Now we have included model NO_x bias in the text as "As shown in Figure 1a, the GEOS-Chem model simulated NO_x agree well with the observations for July-October. However, compared to observations, the model results significantly overestimate NO_x mixing ratios for November-January by about 150%, while underestimating the data in spring and early summer by approximately 60%." (lines 208-211)

For Fig. 3, we have included the NO_x model-to-observations slopes and correlation coefficients in the text as "As shown in Figure 3a, GEOS-Chem overestimates surface NO₂ mixing ratios at these sites by over 66%, compared with observations (slope=1.07; correlation coefficient=0.88)." (lines 231-232) and "Furthermore, the discrepancy for the differences of surface NO₂ mixing ratios over Europe between EURO_EDGAR and observations is further reduced (by 50%), relative to the control runs, with a model-to-observation slope of 0.92 and a correlation coefficient of 0.83 (Fig. 3b)." (lines 248-249)

5) Lines 206-208: Is the magnitude of the snowpack NOx reservoir depletion of right order to explain this? Is the source linearly dependent on the reservoir? Would it be hard to test this in the model to see if it improves the model bias? i.e. can you scale the monthly emissions according to this finding? Perhaps not necessary, but a brief discussion of the order of magnitude of depletion and how that relates to the model bias would be helpful.

Response: Thanks for the excellent questions. Snowpack nitrate photolysis plays an important role in affecting the surface NO_x mixing ratios during late spring and summer over Summit, Greenland. Dibb et al. (2007) demonstrated that nitrate concentrations in the snowpack peaked in June and declined toward fall by ~ 1/3. Moreover, Van Dam et al. (2015)

offered the direct evidence that NO_x mixing ratios within the snowpack showed declining trend from June to October, which may partially explain why we would see the declining trend of surface NO_x mixing ratios over Summit from May-October. We have therefore added this discussions in the text "Dibb et al. (2007) reported that nitrate concentrations in the Summit snowpack peaked in June and declined toward fall by $\sim 1/3$. Van Dam et al. (2015) further showed decreasing trend for NO_x mixing ratios within the snowpack at Summit from June to October. This may partially explain why we would see the declining trend of surface NO_x mixing ratios over Summit from June toward fall. The NO_x emissions from snowpack are affected by a number of factors including nitrate concentrations and solar radiation available and the responses can be very non-linear. Further investigations are needed to account for the seasonal variations of snowpack NO_x emissions from nitrate photolysis in the model, i.e., constrained by seasonal snowpack NO_x emission flux measurements in the future." (lines 281-289)

6) Does this model include the PAN budget updates from the Fischer et al. study that is mentioned? This should be stated clearly. Arnold et al., (2105) showed that GEOS-Chem produces less PAN relative to CO than other models in Arctic air masses influenced by fires. It would be useful to refer back to this here to give context to the model performance relative to that found for other models.

Response: Points are well taken. We have added clarification and discussion in the text - "For instance, a study by Fischer et al. (2014) showed improved agreement between modeled and measured PAN in the high latitudes when assignining a portion of the fire emissions in the model above the boundary layer and also directly partitioning 40% of NO_x emissions from fires into PAN. We carried out a sensitivity test with similar treatments, but no significant improvements in the model simulated surface PAN were observed at the Summit site. Therefore, we did not include the PAN updates from Fischer et al. (2014) in other model simulations in this study." (lines 321-327)

We have also added discussion on the reference of Arnold et al. (2015) - "This is consistent with the study by Arnold et al. (2015), which reported that model simulated PAN mixing ratios in GEOS-Chem were lower than ARCTAS observations over high-latitude

atmosphere in the Arctic. Meanwhile, this study also revealed that GEOS-Chem produced

less PAN relative to CO in Arctic air parcels that were influenced by fires, compared with

other models." (lines 305-314)

Typographical / editorial corrections:

Line 58: ":::while PAN mixing ratios were lower in fresh boreal fire plumes." This sentence

in unclear. Lower than observed? Lower than in other air mass types simulated in the model?

Response: Thanks for pointing this out. Model simulated PAN mixing ratios were lower than

the observations. Therefore, we have modified the whole sentence as "They found that model

simulated NO_x mixing ratios were higher than observations, while PAN mixing ratios were

lower than the observations in fresh boreal fire plumes." (lines 64-66)

Line 82: ".. that the snowpack emits.."

Response: Done.

Paragraph beginning Line 141 contains mixed (past / present) tenses. Please adjust the text to

make it consistent.

Response: Points are well taken. We have corrected the paragraph as "We first run the

standard GEOS-Chem model with a-priori emissions and compare the simulation results

against observations for various species (including NO_x, PAN, C₂H₆, C₃H₈, CO, and O₃, as

shown in Fig. 1). Then we focus on the model-observation discrepancies, and where

applicable, made revisions to the model simulations and further evaluate the improvement

in model performance, as discussed in details below." (lines 198-202)

Line 156: "not observed in the data." Better to simply say ".. not observed".

Response: Agree. We have deleted "in the data" in the revised text.

Line 161: Omit word "mannually" (which should be spelled "manually" in any case).

Response: Typo has been corrected.

Response to Dr. Hongyu Liu's comments:

We thank Dr. Hongyu Liu for his valuable and thoughtful comments. Our responses to the comments are provided below, with Dr. Hongyu Liu's comments italicized and our responses in plain and bold fonts.

This paper presents a GEOS-Chem model analysis of surface ozone and its precursors (NOx, PAN, C2H6, C3H8, CO) observed at Summit, Greenland during the period of July 2008 - June 2010, with a focus on their concentrations and seasonal variations. Modeling tropospheric ozone in the Arctic has been challenging, and it is very interesting to use a state-of-the art chemical transport model to test and improve our understanding of its sources and variability. The authors identified the discrepancies between the GEOS-Chem simulations and observations, which were then examined using various model perturbation experiments. The results are original, and the paper is concise and very well written. I recommend its publication on ACP with some minor modifications, as itemized below.

1). Title - Using "tropospheric ozone" in the title is a bit misleading. Although this study also compared the model vertical profiles of ozone and specific humidity with ozonesonde observations, the main scope of this paper is "surface ozone".

Response: We agree with the reviewer's comment. We therefore change the title as "Surface ozone and its precursors at Summit, Greenland: comparison between observations and model simulations".

2). Section 2: It is not clear which version of the GEOS-5 meteorological data archive was used. Is it GEOS-5.1.0 or GEOS-5.2.0? See e.g., http://wiki.seas.harvard.edu/geos-chem/index.php/GEOS-5_met_field_reprocessing and "http://wiki.seas.harvard.edu/geos-chem/index.php/List_of_GEOS-5_met_fields".

Response: Thanks for pointing this out. It is GEOS-5.2.0. We have added this in Section 2 as "The GEOS-Chem model has fully coupled O_3 -NO_x-VOC-Aerosol chemistry mechanism and is driven by assimilated meteorological data from the Goddard Earth Observing System

version 5.2.0 (GEOS-5.2.0) of the NASA Global Modeling Assimilation Office." (lines 150-

152)

3). Section 2, 2nd paragraph: "Time series data were archived with 3-hr temporal resolution at

the Summit grid box" — I think you meant "grid column". Moreover, it is not clear how the model

output was sampled in the vertical. The elevation of Summit is 3212m asl. Did you sample the

model bottom layer, or the model vertical layer that is about 3212m above the sea level? The latter

may very well be different than the former because the topography is not well resolved at coarse

resolution. Would the results about model overestimates or underestimates found in this paper be

different if the alternative way of model sampling is used (e.g., lines 206-207)?

Response: Good points. We archived the time series data with 3-hr temporal resolution at

Summit grid box for each model vertical level, including the model bottom layer. For

comparison with surface observations at Summit, Greenland, we sampled the data for the

model bottom layer. Indeed, the topography is not very well resolved at coarse model

resolution, and we believe diagnostics for the model bottom layer would work better than

those for the 3212 m level for comparison to the surface measurements.

Minor editorial comments:

Line 43: change the "and" before "volatile organic compounds" to comma.

Response: Changed.

Line 56: the ARCTAS mission

Response: Corrected.

Line 66: What do you mean "O3 mixing ratios below the boundary layer"? Within the boundary

layer?

Response: Yes, it is within the boundary layer. We have therefore corrected the sentence as

"Wespes et al. (2012) also revealed that model simulated O₃ mixing ratios within the

boundary layer were significantly underestimated during spring-summer, compared with

ARCTAS measurements." (lines 85-87)

Line 77: "....used to be the global default anthropogenic C2H6 emission inventory" - Do you mean

"default" in GEOS-Chem or any other models?

Response: Yes, it is the GEOS-Chem default anthropogenic C_2H_6 emission inventory.

Line 228: add "over Greenland" after "PAN".

Response: Added.

Line2 268-270: "relative to NEI11 MIX" – isn't this relative to NEI11? Remove it?

Response: Well, it is relative to NEI11_MIX. In terms of emissions, you are right because we

don't change the MIX emissions in this sensitivity simulation. Therefore, in order to avoid

confusion, we give a simulation name for this sensitivity run and change this sentence as "We

therefore run a sensitivity simulation by increasing the NEI11 C2H6 emissions by 40% and

keeping other model configuration identical to NEI11_MIX (hereafter referred to as

NEI11 40 MIX). We find this update leads to an increase in the model simulated annual

mean surface C₂H₆ mixing ratios over Summit by only 6% during the period of 07/2008-

06/2010 (figure not shown), still not able to explain the high model bias." (lines 366-370)

Line 2279-281: The annual mean agrees quite well with observations, but the simulation is worse

in summer.

Response: Points are well taken. We have changed this sentence to "We find that the

simulated annual mean surface C₂H₆ mixing ratios at Summit from NEI11 MIX20 agree

quite well with observations (within 1%). Similarly, better agreement between model and

observations are found for monthly average values for October - January. However, the new

simulation is not able to reproduce the seasonal cycle of C₂H₆ - the model signficantly

underestimates in February – April but overestimates in June – September (Fig. 5)." (lines

384-389)

Lines 283-288: Good point, but this long sentence needs a break.

Response: Thanks for pointing this out. We have divided the long sentence to "Note that this standard version of GEOS-Chem does not account for the sink of C_2H_6 from the reaction with chlorine, which could reduce the global annual mean surface C_2H_6 mixing ratio by 0-30%, and the global burden of C_2H_6 by about 20% (Sherwen et al., 2016). However, this may introduce additional uncertainty for our measurement-model comparison, together with the uncertainty in the seasonality of C_2H_6 chemistry." (lines 392-396)

Line 325: "Unfortunately, ..." – "However, ..."

Response: Corrected.

Lines 339-340: "..., which implies that GEOS-Chem possibly underestimates STE for O3 over Summit" — This is interesting and appears consistent with Choi et al., ACP 2017 (https://www.atmos-chem-phys.net/17/8429/2017/, see their Fig. 6), where the GMI CTM driven by MERRA (GEOS-5.2.0) underestimates ozonesonde-observed ozone in the Northern Hemisphere high-latitude upper troposphere.

Response: Thank you for providing us a reference source, which also attributed the model low bias to STE. We have included a discussion about this study in the text as "This is consistent with the study by Choi et al. (2017), which found low bias with model simulated O₃ mixing ratios over high-latitude upper troposphere of the Northern Hemisphere, compared with ozonesonde data, and attributed the low bias to weak STE in the model." (lines 471-473)

Lines 358-359: Summit, Greenland; surface ozone

Response: Good suggestion. We have changed the whole sentence to "We combine model simulations with two-year (July 2008 - June 2010) ground based measurements at Summit, Greenland, to investigate the abundance and seasonal variations of surface O_3 and related species in the Arctic." (lines 482-484)

Figures 2, 3,5, S1: In the caption, state briefly what the perturbation simulations are

and refer the reader to the text for details.

Response: We have added additional descriptions in the captions for Figures 2, 3, 5, and S1. Please refer to our revised manuscripts for details.

T	Surface ozone and its precursors at Summit, Greenland: comparison between observations		
2	and model simulations		
3	Yaoxian Huang 1,a , Shiliang $Wu^{1,2,3}$, Louisa J. Kramer 1,2,b , Detlev $Helmig^4$, and Richard E. Honrath $^{1,2,\uparrow}$		
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Abstract. Recent studies have shown some significant challenges for atmospheric models to 18 simulate tropospheric ozone (O₃) and some of its precursors in the Arctic. In this study, ground 19 based data are combined with a global 3-D chemical transport model (GEOS-Chem) to examine 20 21 the abundance and seasonal variations of O₃ and its precursors at Summit, Greenland (72.34° N, 38.29° W, 3212 m.a.s.l). Model simulations for atmospheric nitrogen oxides (NO_x), peroxyacetyl 22 nitrate (PAN), ethane (C₂H₆), propane (C₃H₈), carbon monoxide (CO), and O₃ for the period of 23 07/2008-06/2010 are compared with observations. The model performs well in simulating certain 24 species (such as CO and C₃H₈), but some significant discrepancies are identified for other 25 species and further investigated. The model generally underestimates NO_x and PAN (by around 26 50% and 30%, respectively) for March-June. Likely contributing factors to the low bias include 27 missing NO_x and PAN emissions from snowpack chemistry in the model. At the same time, the 28 model overestimates NO_x mixing ratios by more than a factor of 2 in wintertime, with episodic 29 NO_x mixing ratios up to 15 times higher than the typical NO_x levels at Summit. Further 30 investigation shows that these simulated episodic NO_x spikes are always associated with 31

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transport events from Europe, but the exact cause remains unclear. The model systematically 37 overestimates C₂H₆ mixing ratios by approximately 20% relative to observations. This 38 discrepancy can be resolved by decreasing anthropogenic C₂H₆ emissions over Asia and the US 39 by $\sim 20\%$, from 5.4 to 4.4 Tg/yr. GEOS-Chem is able to reproduce the seasonal variability of O_3 40 and its spring maximum. However, compared with observations, it underestimates surface O₃ by 41 approximately 13% (6.5 ppbv) from April to July. This low bias appears to be driven by several 42 factors including missing snowpack emissions for NO_x and nitrous acid in the model, the weak 43 44 simulated stratosphere-to-troposphere exchange flux of O₃ over the Summit, as well as the coarse 45 model resolution.

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1. Introduction

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Tropospheric ozone (O_3) and its precursors, including nitrogen oxides $(NO_x = NO + NO_2)_x$ carbon monoxide $(CO)_x$, and volatile organic compounds (VOCs, such as ethane, propane, etc.) are important atmospheric species affecting both air quality and climate (e.g., Jacob et al., 1992; Fiore et al., 2002; Unger et al., 2006; Hollaway et al., 2012). Tropospheric O_3 is mainly produced by photochemical oxidation of CO and VOCs in the presence of NO_x , with additional contribution by transport from the stratosphere. Its major sinks include chemical reactions and dry deposition. As a reservoir species for NO_x , peroxyacetyl nitrate (PAN) also plays an important role in atmospheric chemistry, PAN and O_3 , as well as some of their precursors, have relatively long lifetimes in the atmosphere, enabling them to be transported long distance to remote regions such as the Arctic.

Recent studies have shown some significant challenges for atmospheric chemical transport 57 models to simulate O₃ and its precursors in the Arctic (e.g., Shindell et al., 2008; Alvarado et al., 58 2010; Walker et al., 2012; Wespes et al., 2012; Fischer et al., 2014; Monks et al., 2015), but the 59 causes remain unclear. In the multi-model assessment by Shindell et al. (2008), more than a 60 dozen models all showed systematic and persistent underestimation of O₃ at the GEOSummit 61 station, Greenland (hereafter referred to as Summit). Alvarado et al. (2010) used NO_x and PAN 62 63 measurements from ARCTAS (Arctic Research of the Composition of the Troposphere from Aircraft and Satellites) in the summer to compare with model simulations. They found that 64 model simulated NOx mixing ratios were higher than observations, while PAN mixing ratios 65 66 were lower than the observations in fresh boreal fire plumes. In terms of global PAN Deleted: O

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simulations, Fischer et al. (2014) directly partitioned 40% of NO_x emissions from wildfires to 80 PAN formation, which improved the agreement between model and observations. However, the 81 model still underestimated PAN surface mixing ratios during springtime in the Arctic. Walker et 82 83 al. (2012) reported that model simulated O₃ mixing ratios were biased low when compared with balloon data during summertime from two high-latitude sites at Eureka (80°N, 86°W) and Ny-84 Ålesund (79°, 12°E). Wespes et al. (2012) also revealed that model simulated O₃ mixing ratios 85 86 within the boundary layer were significantly underestimated during spring-summer, compared 87 with ARCTAS measurements. More recently, Monks et al. (2015) further demonstrated that model simulated O₃ mixing ratios in the Arctic at the surface and in the upper troposphere were 88

generally lower than the observations. <u>In addition, a recent study by Christian et al. (2017)</u> compared O₃ observations from the ARCTAS campaign to GEOS-Chem model simulations and

91 found consistent low biases with the model simulated O₃ at all altitudes except the surface.

92 Field measurements at Summit show that the snowpack emits gas-phase NO_x, PAN, nitrous acid

93 (HONO), as well as hydrogen peroxide (H₂O₂) during spring-summer, when polar sun rises

94 (Ford et al., 2002; Honrath et al., 2002). Although several 1-D models (Thomas et al., 2011,

95 2012; Frey et al., 2013; Murray et al., 2015) have validated the importance of snowpack

 $\underline{\text{emissions}}$ for surface NO_x as well as O_3 formation, current global chemical transport models

97 (CTMs) usually do not include this emission source (Zatko et al., 2016).

In this study, we examine the abudance and seasonal variations of O₃ and its precursors at

Summit with a global chemical transport model, GEOS-Chem CTM, in conjunction with two

years in-situ measurement data for 2008-2010. We first evaluate the model performance in

simulating surface O₃ and its precursors, and then implement a series of model updates to resolve

the identified model biases. This paper is organized as follows: section 2 describes model

methods and observations, followed by detailed comparisons of model simulations against

observations for O_3 and O_3 precursors in section 3; conclusions are summarized in section 4.

2. Observational data and model simulations

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In situ measurements of NO_x, PAN, and non-methane hydrocarbons (NMHCs) were performed at Summit from July 2008 to June 2010 (Helmig et al., 2014b; Kramer et al., 2015). An automated chemiluminescence instrument was used to measure NO_x (Ridley and Grahek, 1990);

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a commercial PAN gas chromatography analyzer (PAN-GC, Metcon, In., Boulder, CO) was 137 138 employed for the measurement of PAN. Measurements of NMHC relied on an automated Gas 139 Chromatography-Flame Ionization Detection (FID) system. Readers are referred to Kramer et al. 140 (2015) and Helmig et al. (2014b) for the details of the measurement techniques and equipment setup. Surface measurements of O₃ using ultraviolet light absorption at 254 nm 141 142 (Petropavlovskikh and Oltmans, 2012), and CO by GC (Novellie and Masarie, 2015) are from 143 the National Oceanic and Atmospheric Administration (NOAA). Hourly averaged O₃ and flask 144 sampled CO between July 2008 and June 2010 were downloaded from the NOAA Earth System (ESRL) Global Division Research Laboratory Monitoring (GMD) 145 (http://www.esrl.noaa.gov/gmd/dv/data/). Vertical ozonesonde data profiles were also 146 147 downloaded from NOAA ESRL GMD (McClure-Begley et al., 2014).

148 Simulations of O3 and related species (NOx, PAN, NMHCs) are conducted using the GEOS-Chem model (Bey et al., 2001) with coupled O₃-NO_x-VOC-Aerosol chemistry mechanism (i.e. 149 150 these species interact with each other in the model). The GEOS-Chem CTM, is driven by 151 assimilated meteorological data from the Goddard Earth Observing System version 5.2.0 152 (GEOS-5.2.0) of the NASA Global Modeling Assimilation Office. The GEOS-Chem model has 153 been extensively evaluated and applied in a wide range of applications (Martin et al., 2002; Park et al., 2004; Wu et al., 2007; Hudman et al., 2009; Johnson et al., 2010; Huang et al., 2013; 154 155 Kumar et al., 2013; Zhang et al., 2014; Hickman et al., 2017), including the studies in the Arctic 156 (e.g., Alvarado et al., 2010; Monks et al., 2015; Christian et al., 2017). GEOS-Chem v10-1 with grid resolution of 4° latitude by 5° longitude, and 47 vertical layers was used for the model 157 control simulation. Following McLinden et al. (2000), the Linoz stratospheric O₃ chemistry 158 scheme was used. The simulation was run from June 2007 to June 2010 and the results from the 159 last two years were used in the final analysis. Time series data were archived with 3-hr temporal 160 resolution at the Summit grid box for each model vertical level, including the model bottom 161 layer. For comparison with surface observations at Summit, Greenland, we sampled the data for 162 the model bottom layer. We acknowledge that the topography is not well resolved at coarse 163 164 model resolution, and we believe that the diagnostics for the model bottom layer would work 165 better than the model vertical layer that is about 3212 m above the sea level for comparison to 166 the surface measurements.

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Deleted: was used to simulate the seasonal cycles of O_3 and related species (NO_s, PAN, NMHCs) at Summit. The GEOS-Chem model has fully

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174 Global anthropogenic emissions of NO_x, SO₂, NH₃, and CO in the model are based on the Emission Database for Global Atmospheric Research (EDGAR) v4.2 inventory, which is 175 176 overwritten by regional emission inventories where applicable, such as the BRAVO inventory 177 for Mexico (Kuhns et al., 2005), the CAC over Canada, the EMEP emissions over Europe, the Model Inter-comparison Study for Asia Phase III (MIX) emissions over Asia (Li et al., 2017), 178 179 and the US EPA NEI 2011 (NEI11) emission inventory (Simon et al., 2010). The soil NO_x 180 emission scheme follows Hudman et al. (2012). Lightning NO_x emissions are calculated per flash rate based on GEOS-5 computed cloud-top heights (Price and Rind, 1992), which are 181 determined by deep convection and constrained by satellite observations for monthly average 182 183 flash rates from the Lightning Imaging Sensor and Optical Transient Detector (OTD/LIS) (Sauvage et al., 2007; Murray et al., 2012). Biomass burning emissions are from the Global Fire 184 Emission Database version 4 (GFED4) inventory with monthly resolution (Giglio et al., 2013). 185 The RETRO (Reanalysis of the TROpospheric chemical composition) global anthropogenic 186 187 NMHC emission inventory (van het Bolscher et al., 2008) was used except for ethane (C₂H₆) and 188 propane (C₃H₈), which follows Xiao et al. (2008, hereafter referred to as X08) for the year 2001. 189 In GEOS-Chem, RETRO used to serve as the default global anthropogenic emission inventory 190 for C₂H₆, however, the annual budget of which has been shown too low compared with 191 observations. Global biofuel emission inventory follows Yevich and Logan (2003), which includes emissions for C₂H₆ and C₃H₈. For biogenic VOC emissions, the Model of Emissions of 192 Gases and Aerosols from Nature (MEGAN) scheme (Guenther et al., 2006) was used. Dry 193 194 deposition of species in GEOS-Chem uses a standard resistance-in-series scheme (Wesely, 1989), as implemented in Wang et al. (1998). Wet scavenging follows Liu et al. (2001), 195 including scavenging in convective updraft, rainout (in-cloud) and washout (below-cloud) from 196 197 convective anvils and large-scale precipitation.

We first run the standard GEOS-Chem model with a-priori emissions and compare the

simulation results against observations for various species (including NO_x, PAN, C₂H₆, C₃H₈,

CO, and O₃, as shown in Fig. 1). Then we focus on the model-observation discrepancies, and where applicable, make revisions to the model simulations and further evaluate the improvement

in model performance, as discussed in details below.

3. Results and Discussions

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 $3.1 \quad NO_x$

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We first combine the two years of data for July 2008 - June 2010 and anylaze their seasonal 207 variations. As shown in Figure 1a, the GEOS-Chem model simulated NO_x agree well with the 208 209 observations, for July-October. However, compared to observations, the model results 210 significantly, overestimate NO_x mixing ratios for November-January by about 150%, while 211 underestimating the data in spring and early summer by approximately 60%. Another challenge 212 for the model simulation is that it does not capture the decrease of NO_x for May - November, We find that during the 2009-2010 winter season, model simulations show several high NO_x spikes 213 with peak NO_x mixing ratios reaching ~ 0.15 ppbv or higher, which is ~ 15 times greater than 214 215 typical backgound levels (Fig. 2). These large peaks in NO_x were not observed in the data. Similar peaks were also seen in the model simulations during the 2008-2009 winter season; 216 217 however, there are no measurement data available for this period to compare with.

218 Further analyses show that the model-simulated high NO_x spikes during wintertime are all associated with transport events from Europe. We carried out a sensitivity study to examine the 219 impacts of European emissions on Arctic NO_x by manually reducing anthropogenic NO_x 220 221 emissions from the EMEP emission inventory over Europe by 50% (EMEP50). Results show 222 that surface peak NO_x mixing ratios over Summit during the spike events (e.g., dates around 12/09/2009, 12/15/2009, 1/15/2010 and 1/22/2010) from EMEP50 almost decline proportionally 223 by approximately 50% during 2009/12/01-2010/01/31 (Fig. 2), which confirms that the modeled 224 225 NO_x spikes at Summit during wintertime are associated with transport from Europe. However, the model simulated NO_x is still significantly higher than observations. Comparisons for surface 226 NO₂ mixing ratios between model simulations and 11 in-situ observational sites over Europe 227 during this period were conducted with data downloaded from http://ebas.nilu.no. For detailed 228 229 site information, NO₂ measurement technique and resolution, refer to Table 1. Measurement data 230 over these two months for each site were averaged to compare with the corresponding grid cell in 231 the model. As shown in Figure 3a, GEOS-Chem overestimates surface NO2 mixing ratios at 232 these sites by over 66%, compared with observations (slope=1.07; correlation coefficient=0.88).

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Instead of using EMEP, we carried out another sensitivity study to force anthropogenic NO_x

emissions over Europe following EDGAR v4.2 (EURO EDGAR), with other model

configurations identical to control simulations. As shown in Figure 2, the NO_x mixing ratios over 242 Summit during 12/2009-01/2010 agree much better with observations, especially for January 243 244 2010, where the model captures the magnitudes of observational peaks. This is because NOx 245 emissions from EDGAR over Europe (1.97 Tg NO) are 12% lower than that from EMEP (2.24 Tg NO) for the months of 12/2009 and 01/2010. Furthermore, the discrepancy for the differences 246 247 of surface NO2 mixing ratios over Europe between EURO EDGAR and observations is further 248 reduced (by 50%), relative to the control runs, with a model-to-observation slope of 0.92 and a correlation coefficient of 0.83 (Fig. 3b). Similarly, we also tested the sensitivty of surface NO_x 249 mixing ratios over Summit in response to the changes in the anthropogenic NO_x emissions from 250 251 NEI11 over US and MIX over Asia (including Siberia) during these two months, and found that surface NO_x mixing ratio over Summit during these two months were quite close to the control 252 253 simulations (not shown), reflecting insenstivity to emission perturbations from US and Asia. Therefore, we conclude that uncertainties in fossil fuel NO_x emissions of EMEP associated with 254 transport events from Europe in the model are the most likely cause for the wintertime NO_x 255 spikes over Summit. 256 For April-July, model simulated monthly mean NO_x mixing ratios over Summit are a factor of 257

two lower than the observations (Fig. 4a). In-situ measurements at Summit by Honrath et al. 258 (1999, 2000a, 2000b, 2002) showed upward fluxes of NO_x $(2.52 \times 10^8 \text{ molecules cm}^{-2} \text{ s}^{-1})$ from 259 photolysis of nitrate in snowpack during the summertime, leading to enhancement in NO_x levels 260 in the surface layer by approximately 20 pptv, which is comparable to surface NO_x mixing ratios 261 262 in the Arctic from other sources. Similar results were found over the East Antarctic Plateau. snow/ice sheet (Frey et al., 2013; Legrand et al., 2014). The standard GEOS-Chem model does 263 not include the photolysis of nitrate from snowpack, implying a missing source for NO_x in the 264 Arctic/Antarctic boundary layer.

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In order to test the sensitivity of model simulated surface NO_x mixing ratios to the snowpack emissions, we implement in the model a constant 24-hr NO_x flux $\sim 2.52 \times 10^8$ molecules cm⁻² s⁻¹ during April-July over Greenland (60-85° N, 20-60° W), following the measurements conducted at Summit during summertime by Honrath et al. (2002). As a result, we find that on average, the model simulated surface NO_x mixing ratios for April to July over Summit more than double that from the control simulation, which improves the agreement between model and observations for Deleted: (Fig. 2)

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276 April-June (Fig. 4a). However, the assumed NO_x flux from snowpack in the model leads to 277 overestimate of NOx mixing ratios in July and the model is still not able to reproduce the 278 decreasing trends of NO_x for May-October. This decreasing trend of NO_x may be driven by the 279 decreasing NO_x production rate in snowpack resulting from a gradual depletion of the snowpack NO_x reservoir (Van Dam et al., 2015), which is not reflected in the model since we implement a 280 281 simple constant NO_x emission flux. Dibb et al. (2007) reported that nitrate concentrations in the 282 Summit snowpack peaked in June and declined toward fall by ~ 1/3. Van Dam et al. (2015) further showed decreasing trend for NO_x mixing ratios within the snowpack at Summit from 283 June to October. This may partially explain why we would see the declining trend of surface 284 285 NO_x mixing ratios over Summit from June toward fall. The NO_x emissions from snowpack are affected by a number of factors including nitrate concentrations and solar radiation available and 286 the responses can be very non-linear. Further investigations are needed to account for the 287 288 seasonal variations of snowpack NO_x emissions from nitrate photolysis in the model, i.e., constrained by seasonal snowpack NO_x emission flux measurements in the future. 289

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290 3.2 PAN

291 We then examine the model performance for PAN, which serves as a reservoir for NO_{sc}. Figure 1b shows the comparison of model simulated monthly mean PAN mixing ratios with the 292 293 measurement data. The model captures the seasonal variation of PAN well, although significantly (by ~30%) underestimting the PAN mixing ratios for April-June. By running the 294 model simulation with higher horizontal resolution at 2° latitude by 2.5° longitude (hereafter 295 referred to as GEOS-Chem 2x2.5), we find that the monthly mean PAN mixing ratios over 296 297 Summit during April-July increased by up to 23.3 pptv compared to the 4x5 simulation (Fig. 4b). This can be explained by two reasons. First, coarse model resolution (e.g., 4x5 horizontal 298 resolution) could artificially smear the intense emission sources throughout the entire grid cell 299 (e.g., over urban regions), leading to underestimates of downwind concentrations for species, 300 e.g., O₃ and O₃ precursors (Jang et al., 1995; Yu et al., 2016). Second, ventilation of lower 301 302 atmosphere could be better resolved by a finer model resolution, leading to more efficient vertical advection (Wang et al., 2004; Chen et al., 2009; Yu et al., 2016). However, on average, 303 304 monthly mean model simulated PAN mixing ratios are still underestimated by 20% during this 305 period, compared with observations. This is consistent with the study by Arnold et al. (2015), Formatted: Subscript

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which reported that model simulated PAN mixing ratios in GEOS-Chem were lower than
ARCTAS observations over high-latitude atmosphere in the Arctic. Meanwhile, this study also
revealed that GEOS-Chem produced less PAN relative to CO in Arctic air parcels that were
influenced by fires, compared with other models.

Snowpack can emit not only NO_x, but also PAN, based on field studies at Summit during summertime by Ford et al. (2002). GEOS-Chem does not contain snowpack PAN emissions and chemistry. For a sensitivity study, similar to snowpack NO_x emissions as discussed in section 3.1, we considered a 24-hr constant flux of 2.52 x 10⁸ molecules cm⁻² s⁻¹ of PAN over Greenland from April to July, following Ford et al. (2002). As a result, model simulated PAN mixing ratios agree much better with observations (Fig. 4b). Note that there are also other possible reasons that lead to model bias. For instance, a study by Fischer et al. (2014) showed improved agreement between modeled and measured PAN in the high latitudes can be found when assigning a portion of the fire emissions in the model above the boundary layer and directly partitioning 40% of NO_x emissions from fires into PAN. We carried out a sensitivity test with similar treatments, but no

325 significant improvements in the model simulated surface PAN were observed at the Summit site.

Therefore, we did not include the PAN updates from Fischer et al. (2014) in other model

327 <u>simulations in this study</u>

3.3 NMHC

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Comparisons of observed surface C₂H₆ and C₃H₈ mixing ratios with GEOS-Chem simulations at 329 Summit are shown in Figures 1c and d. The model simulations agree well with surface 330 331 measurements of C₃H₈, but systematically overestimate C₂H₆ (by approximately 25% annually), 332 with the largest bias (0.48 ppbv) occurring during summer. This is consistent with the study from Tzompa-Sosa et al., (2017), which used the same model as our study and pointed out that using 333 334 X08 as global anthropogenic C₂H₆ emission inventory systematically overestimated surface C₂H₆ mixing ratios over the Northern Hemisphere, compared with ground-based observations. 335 Anthropogenic C₂H₆ emissions over US from NEI11 are shown to geographically match the 336 337 distribution of active oil and natural wells (Tzompa-Sosa et al., 2017), and the most recent MIX has been updated to synergize anthropogenic C₂H₆ emissions from various countries over Asia 338 (Li et al., 2017). Therefore, instead of using global anthropogenic fossil fuel emissions of C₂H₆ 339

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following X08, we first conducted sensitivity simulations by overwritting global emission 346 347 inventories by NEI11 over the US, and MIX over Asia (hereafter referred to as NEI11 MIX). 348 Both NEI11 and MIX contain emissions for the years from 2008 to 2010, which could 349 realistically represent the annual and seasonal variations of C₂H₆ emissions over the US and Asia, thus spatially and temporally better representative of anthropogenic C₂H₆ emissions from 350 351 mid-latitudes transported to the Arctic regions. In general, model control simulations 352 overestimate annual mean surface C₂H₆ mixing ratios primarily in the Northern Hemisphere, 353 with large differences occurring over Asia and the US by up to 5 ppbv, compared with NEI11 MIX during the period of 07/2008-06/2010 (Fig. S1). All the above changes are driven 354 by the substantial reductions of anthropogenic C₂H₆ emissions between emission inventories, 355 from 3.5 (X08) to 2.5 Tg/yr (MIX) over Asia, and from 1.9 Tg/yr (X08) to 1.4 Tg/yr (NEI11) 356 over US, reflecting the decreasing trend of anthropogenic C₂H₆ emissions during 2001-2009 357 (Helmig et al., 2014a), because the X08 emission inventory is based on the year 2001, 358 Substantial changes in surface C₂H₆ mixing ratios over the US bewteen control simulations and 359 NEI11 MIX reflects that there exist tempospatial changes of C₂H₆ emissions from oil and gas 360 productions during the period of 2001-2009. A similar pattern was also found by Tsompa-Sosa et 361 al. (2017). In contrast to the control simulations, NEI11 MIX model simulations show that 362 monthly mean C₂H₆ mixing ratios over Summit are systematically underestimated by 24%, 363 compared with observations (Fig. 5). Tsompa-Sosa et al. (2017) reported that NEI11 for C₂H₆ 364 emissions were likely underestimated by 40%, compared with in-situ and aircraft observations 365 366 over the US. We therefore run a sensitivity simulation by increasing the NEI11 C₂H₆ emissions by 40% and keeping other model configuration identical to NEI11 MIX (hereafter referred to as 367 368 NEI11 40 MIX). We find this update leads to an increase in the model simulated annual mean 369 surface C₂H₆ mixing ratios over Summit by only 6% during the period of 07/2008-06/2010 370 (figure not shown), still not able to explain the high model bias.

Similar to NEI11_MIX, we further conducted sensitivity studies by only replacing the regional emission inventory for C₂H₆ over the US, with other regions still following X08 (hereafter referred to as NEI11_ONLY). Consequently, model simulated surface C₂H₆ mixing ratios over Summit agree better with observations during winter-spring (Fig. 5), decreasing the bias from +15% (control simulations) to +6%. However, model simulated C₂H₆ mixing ratios during summer-fall are higher than the observations by over 30%.

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Deleted: With NEI11 C_2H_6 emissions increases by 40%, however, model simulated annual mean C_2H_8 mixing ratios over Summit only increase by 6% during the period of 07/2008-06/2010, relative to NEI11_MIX.

382 We then scale, up the MIX emissions for C₂H₆ by 20% over Asia, with other model configurations identical to NEI11 MIX (hereafter referred to as NEI11 MIX20). By doing this, 383 384 we increase fossil fuel C₂H₆ emissions from 2.5 to 3 Tg/yr. We find that the simulated annual 385 mean surface C₂H₆ mixing ratios at Summit from NEI11 MIX20 agree quite well with observations (within 1%). Similarly, better agreement between model and observations are found 386 387 for monthly average values for October - January. However, the new simulation is not able to 388 reproduce the seasonal cycle of C₂H₆ - the model significantly underestimates in February – April but overestimates in June - September (Fig. 5). This implies that further assessments of 389 anthropogenic C₂H₆ emissions from MIX over Asia are needed and a more accurate global 390 391 anthropogenic C₂H₆ emission inventory should be developed and validated to replace X08 in the future. Note that this standard version of GEOS-Chem does not account for the sink of C₂H₆ 392 393 from the reaction with chlorine, which could reduce the global annual mean surface C₂H₆ mixing ratios by 0-30%, and the global burden of C₂H₆ by about 20% (Sherwen et al., 2016). This may 394 395 introduce additional uncertainty for our measurement-model comparison, together with the highly uncertain seasonality of C₂H₆ chemistry. 396

397 **3.4 CO**

Figure 1e shows the comparison of model simulated CO mixing ratios with observations over

Summit. Overall, the model generally captures the <u>abundance and</u> seasoanl <u>variation</u> of CO.

400 Compared with observations, the annual mean CO mixing ratios is slightly overestimated by

401 <u>about 3 ppbv in the model.</u>

402 **3.5 O**₃

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403 Surface O₃ mixing ratios from model simulations and surface observations are compared in

404 Figure 1f. The GEOS-Chem model captures the seasonal variation of O₃ including the spring

405 peak. However, the model shows a systematic low bias for most time of the year, in particualr for

406 April – July when the surface O_3 mixing ratios are underestimated by ~13% (~ 6.5 ppbv). Here

we focus our analysis for the possible causes that lead to the model low bias during April-July.

408 As discussed earlier, snowpack emissions due to the photolysis of nitrate in the snow during late

409 spring and summer could contribute to NO_x and HONO levels in the ambient air which could

410 enhance O₃ production (Crawford et al., 2001; Zhou et al., 2001; Dibb et al., 2002; Honrath et

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al., 2002; Yang et al., 2002; Grannas et al., 2007; Helmig et al., 2008; Legrand et al., 2014). We 426 ran a sensitivity study to test the response of surface O₃ mixing ratios to the perturbations of NO_x 427 428 and HONO from snowpack emissions. In addition to snowpack NO_x emissions that are described 429 in Section 3.1, we implement in the model a constant flux of HONO (4.64 x 10⁷ molecules cm⁻² s⁻¹) from April to July (Honrath et al., 2002). As a result, monthly mean model simulated surface 430 O₃ mixing ratios increase by up to 3 ppbv during this period (Fig. 6). The largest effect occurs in 431 432 July due to relatively strong solar radiation. O₃ formation due to snowpack emissions in our 433 study is slightly higher than that in Zatko et al. (2016) because HONO from snowpack emissions is not considered in their study. However, for the months of April and May, surface O₃ mixing 434 ratios only increase by ~ 1 ppbv, compared with the control runs. That is, even after accounting 435 for the snowpack emissions, the model simulated O₃ mixing ratios are still significantly lower 436 437 than the observations.

438 Comparison of the model simulations at different resolutions (4x5 vs. 2x2.5) shows that the finer resolution simulations substantially increase monthly mean O₃ mixing ratios over Summit by up 439 to 6 ppbv for the months of June and July (Fig. 6). As discussed in section 3.2, fine model 440 resolution can better resolve the emission strengths, which could significantly affect downwind 441 chemical reactions, e.g., O₃ production efficiency (Liang and Jacobson, 2000). Moreover, terrain 442 elevations from fine model resolution are better represented (thus better representative of 443 Summit's elevation) and more efficient vertical ventilation of O₃ and O₃ precursors can be 444 achieved (Wang et al., 2004). Together with the impact of snowpack chemistry, this brings 445 model simulated surface O₃ mixing ratios over Summit in much better agreement with 446 447 observations for June - July, However, there is still a low bias in the model for the months of 448 April and May.

Another possible cause for the <u>low</u> O₃ biases <u>in model simulations</u> is the <u>calculated</u> stratosphereto-troposphere exchange (STE) O₃ flux in the model. Liang et al. (2011) have pointed out that STE could be a significant direct sources of O₃ in the Arctic during spring-summer. We retrieved vertical profiles of O₃ mixing ratios and specific humidity from ozonesondes (0-5 km elevation above the Summit surface) launched at Summit for the months of June and July in 2008 and compared those data with model control runs. Ozonesondes were launched intensively during these two months (a total of 19 times). As shown in Figure 7, compared with observations, Deleted: ed

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465 model simulated O₃ mixing ratios averaged over 0-5 km above ground level are underestimated by 3% and 9% in June and July 2008 (Fig. 7a). However, specific humidity in GEOS-5 is 466 467 overestimated by 50% and 81% (Fig. 7b) respectively. Ozonesonde data show that Summit 468 frequently encounters high O₃/low water vapors events (e.g., July 9-11, 2008), which are likely 469 of upper tropospheric/stratospheric origin, (Helmig et al., 2007b), but these are not captured by 470 the model, which implies that GEOS-Chem possibly underestimates STE for O₃ over Summit. 471 This is consistent with the study by Choi et al. (2017), which found low bias with model 472 simulated O₃ mixing ratios over high-latitude upper troposphere of the Northern Hemisphere, compared with ozonesonde data, and attributed the low bias to weak STE in the model. 473

Boundary layer height is another factor that could <u>lead to model-data</u> discrepancy in O₃ mixing ratios (Grannas et al., 2007; Helmig et al., 2207a, c). The mean springtime afternoon (12:00-14:00, local time) boundary layer height in the model at Summit for the year 2009 is 160 m, which agrees fairly well with observations (156 m) at Summit conducted in spring 2005 (Cohen et al., 2007). Therefore, we exclude that model uncertainties in boundary layer height representation in springtime cause the low bias of O₃ mixing ratios between model and observations.

4. Conclusions

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492 493 We combine model simulations with two-year (July 2008_-_June 2010) ground based measurements at Summit to better understand the abundance and seasonal variations of tropospheric O₃ and related species in the Arctic. In general, the GEOS-Chem model is capable of reproducing the seasonal cycles of NO_x, PAN, C₂H₆, C₃H₈, CO, and O₃. However, some major discrepancies between model and observations, especially for NO_x, PAN, C₂H₆, and O₃ are also identified.

There are significant differences between model simulated NO_x mixing ratios and observations for the spring and winter seasons. The model underestimates NO_x mixing ratios by approximately 50% during late spring to early summer, which is likely due to the missing NO_x emissions from nitrate photolysis in the snowpack. At the same time, the model overestimates NO_x mixing ratios by more than a factor of two in wintertime. Model simulations indicate episodic but frequent transport events from Europe in wintertime leading to NO_x spikes reaching

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Deleted: Lastly, we also compared O_3 dry deposition velocity (V_{dry}) in the model in springtime with observations at Summit. For spring 2009, mean O_3 V_{dry} in the model at Summit is 0.009 cm s², which is within the range of -0.01-0.01 cm s² observed at Summit (Helmig et al., 2009). Helmig et al. (2009) revealed that afternoon (12-00-18:00, local time) O_3 V_{dry} during springtime was close to 0.01 cm s². For other times during the day, O_3 V_{dry} was either close to zero or negative (i.e., O_3 production over the snow outweighs its dry deposition). As a result, the net mean O_3 V_{dry} from observations during springtime is about 4-6 times lower than model simulations. Therefore, model overestimation of O_3 loss via surface uptake is another factor that contributes to the low bias of surface O_3 mixing ratios at Summit in the model during springtime.

514 observations. We have carried out multiple sensitivity model studies but are still unable to fully Deleted: al data 515 reconcile this discrepancy. 516 The model successfully captures the seasonal cycles and the spring maximum PAN mixing 517 ratios, although it underestimates PAN by over 30% during late spring and early summer. Model 518 sensitivity studies reveal that this discrepancy could be largely resolved by accounting for PAN 519 emissions from snowpack. 520 For C₃H₈ and CO, model simulations overall agree well with the surface measurements. Deleted: 521 However, the model tends to systematically overestimate surface C_2H_6 mixing ratios by $\sim 20\%$ Deleted: h **Deleted:** approximately 522 on an annual average, compared with observations. This may be explained by that annual 523 emission budgets of C₂H₆ over US and Asia from X08 emission inventory are higher than those 524 from NEI11 and MIX by over 40%. By replacing X08 over the US with NEI11 for C2H6, and 525 scaling up MIX by 20%, the model-observation bias can be resolved, resulting in annual mean Deleted: with 526 bias less than 1%. However, care must be taken to interpret this result because we do not take 527 into account other factors that may influence the discrepancy of surface C₂H₆ mixing ratios at 528 Summit between model and observations, such as the C₂H₆ chemistry with chlorine. Deleted: o 529 GEOS-Chem is able to reproduce the seasonal variation of surface O₃ at Summit but persistently 530 underestimates O₃ mixing ratios by \sim 13% (~ 6.5 ppbv) from April to July. This low bias is **Deleted:** approximately 531 likely caused by a combination of misrepresentations, including the missing snowpack emissions 532 of NO_x and HONO, inaccurate representation of Summit's elevation with a too coarse model Deleted: from Deleted: r 533 resolutions, as well as the underestimated STE. Deleted: model overestimated O3 dry deposition velocity during All the results presented above reveal the importance of local snowpack emissions in regulating 534 535 the air quality over the Arctic. Improvements in global CTM could likely be achieved by 536 coupling snowpack emissions of reactive gases and photochemistry modules in order to better simulate O₃ and O₃ precursors over snow and ice in the Arctic (Zatko et al., 2016). Moreover, 537 538 this study also demonstrates that anthropogenic emissions from midlatitudes play an important 539 role in affecting the Arctic air quality. Deleted: However, further investigations in anthropogenic NO_x emissions over Europe and C_2H_6 emissions over Asia and North America are needed. The uncertainties in O_3 dry deposition and STE Acknowledgements This research was funded by U.S. EPA grant 83518901). Its contents are scheme in GEOS-Chem are warranted to be better quantified in our 540 future study

15 times typical NO_x mixing ratios at Summit; these large NO_x spikes are not seen in the

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References

- 567 Alvarado, M. J., Logan, J. A., Mao, J., Apel, E., Riemer, D., Blake, D., Cohen, R. C., Min, K.-E.,
- Perring, A. E., Browne, E. C., Wooldridge, P. J., Diskin, G. S., Sachse, G. W., Fuelberg, H.,
- Sessions, W. R., Harrigan, D. L., Huey, G., Liao, J., Case-Hanks, A., Jimmenez, J. L.,
- Cubison, M. J., Vay, S. A., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Flocke, F. M.,
- Pollack, I. B., Wennberg, P. O., Kurten, A., Crounse, J., St. Clair, J. M., Wisthaler, A.,
- 572 Mikoviny, T., Yantosca, R. M., Carouge, C. C., and Le Sager, P.: Nitrogen oxides and PAN in
- 573 plumes from boreal fires during ARCTAS-B and their impact on ozone: an integrated analysis
- of aircraft and satellite observations, Atmos. Chem. Phys., 10, 9739-9760,
- 575 doi:10.5194/acp-10-9739-2010, 2010.
- 576 Arnold, S. R., Emmons, L. K., Monks, S. A., Law, K. S., Ridley, D. A., Turquety, S., Tilmes, S.,
- 577 Thomas, J. L., Bouarar, I., Flemming, J., Huijnen, V., Mao, J., Duncan, B. N., Steenrod, S.,
- Yoshida, Y., Langner, J., and Long, Y.: Biomass burning influence on high-latitude
- tropospheric ozone and reactive nitrogen in summer 2008: a multimodel analysis based on
- 580 POLMIP simulations, Atmos. Chem. Phys., 15, 6047-6068, doi:10.5194/acp-15-6047-2015,

Deleted: Aydin, M., Verhulst, K. R., Saltzman, E. S., Battle, M. O., Montzka, S. A., Blake, D. R., Tang,

584	_ 2015.	
585	Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q., Liu, H.,	
586	Mickley L. J., and Schultz, M. G.: Global modeling of tropospheric chemistry with assimilated	
587	meteorology: Model description and evaluation, J. Geophys. ResAtmos., 106, 23073-23095,	
588	2001.	
589	Chen, D., Wang, Y., McElroy, M. B., He, K., Yantosca, R. M., and Le Sager, P.: Regional CO	Deleted: Blake, D. R., and Rowland, F. S.: Global atmospheric concentrations and source strength of
590	pollution and export in China simu- lated by the high-resolution nested-grid GEOS-Chem	
591	model, Atmos. Chem. Phys., 9, 3825–3839, doi:10.5194/acp-9-3825-2009, 2009.	
592	Choi, H., Liu, H., Crawford, J. H., Considine, D. B., Allen, D. J., Duncan, B. N., Horowitz, L.	
593	W., Rodriguez, J. M., Strahan, S. E., Zhang, L., Liu, X., Damon, M. R., and Steenrod, S. D.:	
594	Global O ₃ -CO correlations in a chemistry and transport model during July-August: evaluation	
595	with TES satellite observations and sensitivity to input meteorological data and emissions,	
596	Atmos. Chem. Phys., 17, 8429–8452, doi:10.5194/acp-17-8429-2017, 2017.	
597	Christian, K. E., Brune, W. H., and Mao, J.: Global sensitivity analysis of the GEOS-Chem	
598	chemical transport model: ozone and hydrogen oxides during ARCTAS (2008), Atmos. Chem.	
599	Phys., 17, 3769-3784, doi:10.5194/acp-17-3769-2017, 2017.	
600	Cohen, L., Helmig, D., Neff, W., Grachev, A., and Fairall, C.: Boundary-layer dynamics and its	
601	influence on atmospheric chemistry at Summit, Greenland, Atmos. Environ., 41, 5044-5060,	
602	2007.	
603	Crawford, J. H., Davis, D. D., Chen, G., Buhr, M., Oltmans, S., Weller, R., Mauldin, L., Eisele,	Deleted: .

F., Shetter, R., Lefer, B., Ari- moto, R., and Hogan, A.: Evidence for photochemical produc-

609 tion of ozone at the South Pole surface, Geophys. Res. Lett., 28, 3641-3644, 2001. Dibb, J. E., Arsenault, M., Peterson, M. C., and Honrath, R. E.: Fast nitrogen oxide 610 611 photochemistry, in Summit, Greenland snow, Atmospheric Environment, 36, 2501-2511, 2002. Deleted: Deleted: [3] 612 Dibb, J. E., Whitlow, S. I., Arsenault, M.: Seasonal variations in the soluble ion content of snow 613 at Summit. Greenland: Constraints from three years of daily surface snow samples, Atmos. 614 Environ., 41, 5007-5019, doi:10.1016/j.atmosenv.2006.12.010, 2007. Formatted: Widow/Orphan control, Adjust space between Latin and Asian text, Adjust space between Asian text and Fiore, A. M., Jacob, D. J., Field, B. D., Streets, D. G., Fernandes, S. D., and Jang, C.: Linking 615 **Deleted:** Emmons, L. K., Arnold, S. R., Monks, S. A., Huijnen, V., Tilmes, S., Law, K. S., Thomas, J. L., ... [4] ozone pollution and climate change: The case for controlling methane, Geophys. Res. Lett., 616 29, 1919, doi:10.1029/2002GL015601, 2002. 617 618 Fischer, E. V., Jacob, D. J., Yantosca, R. M., Sulprizio, M. P., Millet, D. B., Mao, J., Paulot, F., Singh, H. B., Roiger, A., Ries, L., Talbot, R. W., Dzepina, K., and Pandey Deolal, S.: 619 620 Atmospheric peroxyacetyl nitrate (PAN): a global budget and source attribution, Atmos. Chem. Phys., 14, 2679-2698, doi:10.5194/acp-14-2679-2014, 2014. 621 Ford, K. M., Shepson, P. B., Bertman, S. B., Honrath, R. E., Peterson, M., Dibb, J. E., and 622 Bottenheim, J. W.: Studies of peroxyacetyl nitrate (PAN) and its interaction with the 623 snowpack at Summit, Greenland, J. Geophys. Res., 107, ACH6, doi:10.1029/2001JD000547, 624 2002. 625 626 Frey, M. M., Brough, N., France, J. L., Anderson, P. S., Traulle, O., King, M. D., Jones, A. E., **Deleted:** Franco, B., Bader, W., Toon, G., Bray, C., Perrin, A Fischer, E., Sudo, K., Boone, C., Bovy, B., 627 Wolff, E. W., and Savarino, J.: The diurnal variability of atmospheric nitrogen oxides (NO and

Chem. Phys., 13, 3045-3042, doi:10.5194/acp-13-3045-2013, 2013.

NO₂) above the Antarctic Plateau driven by atmospheric stability and snow emissions, Atmos.

628

- 639 Giglio, L., Randerson, J. T., and van der Werf, G. R.: Analysis of daily, monthly, and annual
- burned area using the fourth-generation global fire emissions database (GFED4), J. Geophys.
- Res. Biogeosci., 118, 1, 317-328, doi:10.1002/jgrg.20042, 2013.
- 642 Grannas, A. M., Jones, A. E., Dibb, J., Ammann, M., Anastasio, C., Beine, H. J., Bergin, M.,
- Bottenheim, J., Boxe, C. S., Carver, G., Chen, G., Crawford, J. H., Domine, F., Frey, M. M.,
- Guzman, M. I., Heard, D. E., Helmig, D., Hoffmann, M.R., Honrath, R. E., Huey, L. G.,
- Hutterli, M., Jacobi, H. W., Klán, P., Lefer, B., McConnell, J., Plane, J., Sander, R., Savarino,
- J., Shepson, P. B., Simpson, W. R., Sodeau, J. R., von Glasow, R., Weller, R., Wolff, E. W.,
- and Zhu, T.: An overview of snow photochemistry: evidence, mechanisms and impacts,
- Atmos. Chem. Phys., 7, 4329–4373, doi:10.5194/acp-7-4329-2007, 2007.
- 649 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron C.: Estimate of
- 650 global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and
- 651 Aerosols from Nature), Atmos. Chem. Phys., 6, 3181-3210, doi:10.5194/acp-6-3181-2006,
- 652 2006.
- Helmig, D., Oltmans, S. J., Carlson, D., Lamarque, J. F., Jones, A., Labuschagne, C., Anlauf, K.,
- Hayden, K.: A review of surface ozone in the polar regions, Atmospheric Environment, 41,
- 655 5138-5161, 2007a.
- 656 Helmig, D., Oltmans, S. J., Morse, T. O., and Dibb, J. E.: What is causing high ozone at Summit,
- 657 Greenland?, Atmos. Environ., 41, 5031-5043, doi:10.1016/j.atmosenv.2006.05.084, 2007b.
- 658 Helmig, D., Ganzeveld, L., Butler, T., and Oltmans, S. J.: The role of ozone atmosphere-snow
- gas exchange on polar, boundary–layer tropospheric ozone- a review and sensitivity analysis,

Deleted: Gupta, M. L., Cicerone, R. J., Blake, D. R., Rowland, F. S., and Isaksen, I. S. A.: Global[6]

- Atmos. Chem. Phys., 7, 15-30, doi:10.5194/acp-7-15-2007, 2007c.
- Helmig, D., Johnson, B., Oltmans, S. J., Neff, W., Eisele, F., and Davis, D. D.: Elevated ozone in
- the boundary-layer at South Pole, Atmos. Environ., 42, 2788–2803, 2008.
- Helmig, D., Petrenko, V., Martinerie, P., Witrant, E., Röckmann, T., Zuiderweg, A., Holzinger,
- R., Hueber, J., Thompson, C., White, J. W. C., Sturges, W., Baker, A., Blunier, T., Etheridge,
- D., Rubino, M., and Tans, P.: Reconstruction of Northern Hemisphere 1950–2010
- atmospheric non-methane hydrocarbons, Atmos. Chem. Phys., 14, 1463–1483,
- doi:10.5194/acp-14-1463-2014, 2014a.
- 671 Helmig, D., Stephens, C., Caramore, J., and Hueber, J.: Seasonal behavior of non-methane
- 672 hydrocarbons in the firn air at Summit, Greenland, Atmos. Environ., 85, 234-246,
- doi:10.1016/j.atmosenv.2013.11.021, 2014b.
- Hickman, J. E., Huang, Y., Wu, S., Diru, W., Groffman, P. M., Tully, K. L., and Palm, C. A.:
- Nonlinear response of nitric oxide fluxes to fertilizer inputs and the impacts of agricultural
- intensification on tropospheric ozone pollution in Kenya, Glob. Change Biol., 23, 3193-3204,
- doi:10.1111/gcb.13644, 2017.
- 678 Hollaway, M. J., Arnold, S. R., Challinor, A. J., and Emberson, L. D.: Intercontinental trans-
- boundary contributions to ozone-induced crop yild losses in the North Hemisphere,
- Biogeosciences, 9, 271-292, doi: 10.5194/bg-9-271-2012, 2012.
- 681 Honrath, R. E., Peterson, M. C., Guo, S., Dibb, J. E., Shepson, P. B., and Campbell, B.: Evidence
- of NO_x production within or upon ice particles in the Greenland snowpack, Geophys. Res.
- 683 Lett., 26, 695-698, 1999.

Deleted: Helmig, D., Rossabi, S., Hueber, J., Tans, P., Montzka, S. A., Masarie, K., Thoning, K., Plass-

- 690 Honrath, R. E., Guo, S., Peterson, M. C., Dziobak, M. P., Dibb, J. E., and Arsenault, M. A.:
- Photochemical production of gas phase NO_x from ice crystal NO₃, J. Geophys. Res., 105,
- 692 24183–24190, 2000a.
- 693 Honrath, R. E., Peterson, M. C., Dziobak, M. P., Dibb, J. E., Arsenault, M. A., and Green, S. A.:
- Release of NO_x from Sunlight-irradiated Midlatitude Snow, Geophys. Res. Lett., 27, 2237–
- 695 2240, 2000b.
- 696 Honrath, R. E., Lu, Y., Peterson, M. C., Dibb, J. E., Arsenault, M. A., Cullen, N. J., and Steffen,
- 697 K.: Vertical fluxes of NO_x, HONO, and HNO₃ above the snowpack at Summit, Greenland,
- 698 Atmos. Environ., 36, 2629-2640, doi:10.1016/S1352-2310(02)00132-2, 2002.
- 699 Huang, Y., Wu, S., Dubey, M. K., and French, N. H. F.: Impact of aging mechanism on model
- simulated carbonaceous aerosols, Atmos. Chem. Phys., 13, 6329–6343, doi:10.5194/acp-13-
- 701 6329-2013, 2013.
- 702 Hudman, R. C., Moore, N. E., Mebust, A. K., Martin, R. V., Russell, A. R., Valin, L. C., and
- 703 Cohen, R. C.: Steps towards a mechanistic model of global soil nitric oxide emissions:
- implementation and space based-constraints, Atmos. Chem. Phys., 12, 7779-7795, doi:
- 705 10.5194/acp-12-7779- 2012, 2012.
- Hudman, R. C., Murray, L. T., Jacob, D. J., Turquety, S., Wu, S., Millet, D. B., Avery, M.,
- 707 Goldstein, A. H., and Holloway, J.: North American influence on tropospheric ozone and the
- 708 effects of recent emission reductions: Constraints from ICARTT observations, J. Geophys.
- 709 Res., 114, D07302, doi:10.1029/2008JD010126, 2009.
- 710 Jacob, D. J., Wofsy, S. C., Bakwin, P. S., Fan, S.-M., Harriss, R. C., Talbot, R. W., Bradshaw, J.
- 711 D., Sandholm, S. T., Singh, H. B., Browell, E. V., Gregory, G. L., Sachse, G. W., Shipham,

- 712 M. C., Blake, D. R., and Fitzjarrald, D. R.: Summertime photochemistry of the troposphere at
- 713 high northern latitudes, J. Geophys. Res., 97, D15, 16421-16431, doi:10.1029/91JD01968,
- 714 1992.
- 715 Jang, J.-C., Jeffries, H., Byun, D., and Pleim, J.: Sensitivity of ozone to model grid resolution I.
- Application of high resolution regional acid deposition model, Atmos. Environ., 29, 3085-
- 717 3100, doi:10.1016/1352-2310(95)00118-I, 1995.
- 718 Johnson, M.S., Meskhidze, N., Solmon, F., Gasso, S., Chuang, P. Y., Gaiero, D. M., Yantosca,
- 719 R. M., Wu, S., Wang, X., Carouge, C.: Modeling Dust and Soluble Iron Deposition to the
- 720 South Atlantic Ocean, J. Geophys. Res., 115, D15202, doi:10.1029/2009JD013311, 2010.
- Kramer, L. J., Helmig, D., Burkhart, J. F., Stohl, A., Oltmans, S., and Honrath, R. E.: Seasonal
- variability of atmospheric nitrogen oxides and non-methane hydrocarbons at the GEOSummit
- station, Greenland, Atmos. Chem. Phys., 15, 6827-6849, doi:10.5194/acp-15-6827-2015,
- 724 2015.
- 725 Kuhns, H., Knipping, E. M., and Vukovich, J. M.: Development of a United States-Mexico emis-
- sions inventory for the Big Bend Regional Aerosol and Visibility Observational (BRAVO)
- 727 Study, JAPCA J. Air Waste M., 55, 677–692, 2005.
- 728 Kumar, A., Wu, S., Weise, M. F., Honrath, R., Owen, R. C., Helmig, D., Kramer, L., Val Martin,
- 729 M., and Li, Q.: Free-troposphere ozone and carbon monoxide over the North Atlantic for
- 730 2001–2011, Atmos. Chem. Phys., 13, 12537-12547, doi:10.5194/acp-13-12537-2013, 2013.
- 731 Legrand, M., Preunkert, S., Frey, M., Bartels-Rausch, Th., Kukui, A., King, M. D., Savarino, J.,
- Kerbrat, M., and Jourdain, B.: Large mixing ratios of atmospheric nitrous acid (HONO) at

Deleted: Kanakidou, M., Singh, H. B., Valentin, K. M., and Crutzen, P. J.: A two-dimensional study of

- 736 Concordia (East Antarctic Plateau) in summer: a strong source from surface snow?, Atmos.
- 737 Chem. Phys., 14, 9963-9976, doi:10.5194/acp-14-9963-2014, 2014.
- 738 Li, M., Zhang, Q., Kurokawa, J., Woo, J. H., He, K. B., Lu, Z., Ohara, T., Song, Y., Streets, D.
- 739 G., Carmichael, G. R., Cheng, Y. F., Hong, C. P., Huo, H., Jiang, X. J., Kang, S. C., Liu, F.,
- Su, H., and Zheng, B.: MIX: a mosaic Asian anthropogenic emission inventory under the
- international collaboration framework of the MICS-Asia and the HTAP, Atmos. Chem. Phys.,
- 742 17, 935-963, doi:10.5194/acpd-17-935-2017.
- 743 Liang, J., and Jacobson, M. Z.: Effects of subgrid segregation on ozone production efficiency in
- a chemical model, Atmos. Environ., 34, 2975–2982, doi:10.1016/S1352-2310(99)00520-8,
- 745 2000.
- 746 Liang, Q., Rodriguez, J. M., Douglass, A. R., Crawford, J. H., Olson, J. R., Apel, E., Bian, H.,
- Plake, D. R., Brune, W., Chin, M., Colarco, P. R., da Silva, A., Diskin, G. S., Duncan, B. N.,
- Huey, L. G., Knapp, D. J., Montzka, D. D., Nielsen, J. E., Pawson, S., Riemer, D. D.,
- Weinheimer, A. J., and Wisthaler, A.: Reactive nitrogen, ozone and ozone production in the
- 750 Arctic troposphere and the impact of stratosphere-troposphere exchange, Atmos. Chem. Phys.,
- 751 11, 13181-13199, doi:10.5194/acp-11-13181-2011, 2011.
- 752 Liu, H. Y., Jacob, D. J., Bey, I., and Yantosca, R. M.: Constraints from pb-210 and Be-7 on wet
- deposition and transport in a global three-dimensional chemical tracer model driven by
- assimilated meteorological fields, J. Geophys. Res.-Atmos., 106, 12109-12128, doi:
- 755 10.1029/2000JD900839, 2001.
- 756 Martin, R. V., Jacob, D. J., Logan, J. A., Bey, I., Yantosca, R. M., Staudt, A. C., Li, Q., Fiore, A.

- 757 M., Duncan, B. N., and Liu, H.: Interpretation of TOMS observations of tropical tropospheric
- ozone with a global model and in situ observations, J. Geophys. Res., 107(D18), ACH 4-1-
- 759 ACH 4-27, doi: 10.1029/2001JD001480, 2002.
- 760 McClure-Begley, A., Petropavlovskikh, I., Oltmans, S.: NOAA Global Monitoring Surface
- Ozone Network. 1973-2014. National Oceanic and Atmospheric Administration, Earth
- 762 Systems Research Laboratory Global Monitoring Division. Boulder, CO. DATE ACCESSED:
- 763 4/23/2017, http://dx.doi.org/10.7289/V57P8WBF, 2014.
- McLinden, C. A., Olsen, S. C., Hannegan, B., Wild, O., Prather, M. J., and Sundet, J.:
- Stratospheric ozone in 3-D models: A simple chemistry and the cross-tropopause flux, J.
- Geophys. Res., 105, D11, 14653-14665, doi:10.1029/2000JD900124, 2000.
- Monks, S. A., Arnold, S. R., Emmons, L. K., Law, K. S., Turquety, S., Duncan, B. N.,
- Flemming, J., Huijnen, V., Tilmes, S., Langner, J., Mao, J., Long, Y., Thomas, J. L., Steenrod,
- S. D., Raut, J. C., Wilson, C., Chipperfield, M. P., Diskin, G. S., Weinheimer, A., Schlager,
- 770 H., and Ancellet, G.: Multi-model study of chemical and physical controls on transport of
- anthropogenic and biomass burning pollution to the Arctic, Atmos. Chem. Phys., 15, 3575-
- 772 3603, doi:10.5194/acp-15-3575-2015, 2015.
- 773 Murray, K. A., Kramer, L. J., Doskey, P. V., Ganzeveld, L., Seok, B., Van Dam, B., and Helmig,
- D.: Dynamics of ozone and nitrogen oxides at Summit, Greenland. II. Simulating snowpack
- chemistry during a spring high ozone event with a 1-D process-scale model, Atmos. Environ.,
- 776 117, 110–123, doi:10.1016/j.atmosenv.2015.07.004, 2015.
- 777 Murray, L. T., Jacob, D. J., Logan, J. A., Hudman, R. C., and Koshak, W. J.: Optimized regional

- and interannual variability of lightning in a global chemical transport constrained by LIS/OTD
- satellite data, J. Geophys. Res., 117, D20307, doi:10.1029/2012JD017934, 2012.
- 780 Novelli, P. C. and Masarie, K. A.: Atmospheric Carbon Monoxide Dry Air Mole Fraction from
- the NOAA ESRL Carbon Cycle Cooperative Global Air Sampling Network, 1988-2014,
- 782 Version: 2015-08-04, NOAA, available at
- 783 ftp://aftp.cmdl.noaa.gov/data/trace_gases/co/flask/surface/, 2015.
- 784 Park, R. J., Jacob, D. J., Field, B. D., Yantosca, R. M., and Chin, M.: Natural and transboundary
- 785 pollution influences on sulfate-nitrate-ammonium aerosols in the United States: Implications
- 786 for policy, J. Geophys. Res.-Atmos., 109, D15204, doi:10.1029/2003JD004473, 2004.
- 787 Petropavlovskikh, I. and Oltmans, S. J.: Tropospheric Ozone Measurements, 1973-2011,
- 788 Version: 2012-07-10, NOAA, available at: ftp://aftp.cmdl.noaa.gov/data/ozwv/SurfaceOzone/,
- 789 2012.
- 790 Price, C. and Rind, D.: A simple lightning parameterization for calculating global lightning
- distributions, J. Geophys. Res., 97, 9919-9933, doi:10.1029/92JD00719,1992.
- 792 Ridley, B. A., and Grahek, F.: A small, low flow, high sensitivity reaction vessel for NO
- chemiluminescence detectors, Am. Meteorol. Soc., 7, 307-311, 1990.
- 794 Sauvage, B., Martin, R. V., van Donkelaar, A., Liu, X., Chance, K., Jaeglé, L., Palmer, P. I., Wu,
- 795 S., and Fu, T.-M.: Remote sensed and in situ constraints on processes affecting tropical tropo-
- 796 spheric ozone, Atmos. Chem. Phys., 7, 815–838, doi:10.5194/acp-7-815-2007, 2007.
- 797 Sherwen, T., Schmidt, J. A., Evans, M. J., Carpenter, L. J., Groß- mann, K., Eastham, S. D.,
- Jacob, D. J., Dix, B., Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A.,

Deleted: Pozzer, A., Pollmann, J., Taraborrelli, D., Jöckel, P., Helmig, D., Tans, P., Hueber, J., and

Deleted: Rudolph, J.: The tropospheric distribution and budget of ethane, J. Geophys. Res., 100, D6, ... [11]

- Prados-Roman, C., Mahajan, A. S., and Ordóñez, C.: Global impacts of tropospheric halogens
- 806 (Cl, Br, I) on oxidants and composition in GEOS-Chem, Atmos. Chem. Phys., 16, 12239–
- 807 12271, doi:10.5194/acp-16-12239-2016, 2016.
- 808 Shindell, D. T., Chin, M., Dentener, F., Doherty, R. M., Faluvegi, G., Fiore, A. M., Hess, P.,
- Koch, D. M., MacKenzie, I. A., Sanderson, M. G., Schultz, M., Stevenson, D. S., Teich, H.,
- 810 Textor, C., Wild, O., Bergmann, D. J., Bey, I., Bian, H., Cuvelier, C., Duncan, B. N., Folberth,
- G., Horowitz, L. W., Jonson, J., Kaminski, J. W., Marmer, E., Park, R., Pringle, K. J.,
- 812 Schroeder, S., Szopa, S., Takemura, T., Zeng, G., Keating, T. J., and Zuber, A.: A multi-model
- assessment of pollution transport to the Arctic, Atmos. Chem. Phys., 8, 5353-5372,
- doi:10.5194/acp-8-5353-2008, 2008.
- Simon, H., Beck, L., Bhave, P. V., Divita, F., Hsu, Y., Luecken, D., Mobley, J. D., Pouliot, G.
- A., Reff, A., Sarwar, G., and Strum, M.: The development and uses of EPA's SPECIATE
- database, Atmospheric Pollution Research, 196-206, 10.5094/apr.2010.026, 2010.
- Thomas, J. L., Stutz, J., Lefer, B., Huey, L. G., Toyota, K., Dibb, J. E., and von Glasow, R.:
- Modeling chemistry in and above snow at Summit, Greenland Part 1: Model description and
- results, Atmos. Chem. Phys., 11, 4899–4914, doi:10.5194/acp-11-4899-2011, 2011.
- Thomas, J. L., Dibb, J. E., Huey, L. G., Liao, J., Tanner, D., Lefer, B., von Glasow, R., and
- Stutz, J.: Modeling chemistry in and above snow at Summit, Greenland Part 2: Impact of
- snowpack chemistry on the oxidation capacity of the boundary layer, Atmos. Chem. Phys., 12,
- 824 6537–6554, doi:10.5194/acp-12-6537-2012, 2012.
- Tzompa-Sosa, Z. A., Mahieu, E., Franco, B., Keller, C. A., Turner, A. J., Helmig, D., Fried, A.,

Deleted: Simpson, I. J., Sulbaek Andersen, M. P., Meinardi, S., Bruhwiler, L., Blake, N. J., Helmig, D.,

- Richter, D., Weibring, P., Walega, J., Yacovitch, T. I., Herndon, S. C., Blake, D. R., Hase, F.,
- Hannigan, J. W., Conway, S., Strong, K., Schneider, M., and Fischer, E. V.: Revisiting global
- fossil fuel and biofuel emissions of ethane, J. Geophys. Res. Atmos., 122,
- doi:10.1002/2016JD025767, 2017.
- 833 Unger, N., Shindell, D. T., Koch, D. M., and Streets, D. G.: Cross influences of ozone and
- sulfate precursor emissions changes on air quality and climate, Proc. Natl. Acad. Sci., 103,
- 4377-4380, doi:10.1073/pnas.0508769103, 2006.
- van het Bolscher, M., Pereira, J., Spesso, A., Dalsoren, S., van Noije, T., and Szopa, S.:
- REanalysis of the TROpospheric chemical composition over the past 40 years: A long-term
- global modeling study of tropospheric chemistry, Max Plank Inst. For Meteorology, Hamburg,
- 839 Germany, 77, 2008.
- 840 Van Dam, B., Helmig, D., Toro, C., Doskey, P., Kramer, L., Murray, K., Ganzeveld, L., and
- Seok, B.: Dynamics of ozone and nitrogen oxides at Summit, Greenland: I. Multi-year
- observations in the snowpack, Atmos. Environ., 123, 268-284,
- doi:10.1016/j.atmosenv.2015.09.060, 2015.
- Walker, T. W., Jones, D. B. A., Parrington, M., Henze, D. K., Murray, L. T., Bottenheim, J. W.,
- Anlauf, K., Worden, J. R., Bowman, K. W., Shim, C., Singh, K., Kopacz, M., Tarasick, D. W.,
- Davies, J., von der Gathen, P., Thompson, A. M., and Carouge, C. C.: Impacts of midlatitude
- precursor emissions and local photochemistry on ozone abundances in the Arctic, J. Geophys.
- Res., 117, D01305, doi:10.1029/2011JD016370, 2012.
- 849 Wang, Y. H., Jacob, D. J., and Logan, J. A.: Global simulation of tropospheric O₃-NOx-hydro-

- carbon chemistry 1. Model formulation, J. Geophys. Res.-Atmos., 103, 10713-10725, doi:
- 851 10.1029/98JD00158, 1998.
- 852 Wang, Y. X., McElroy, M. B., Jacob, D. J., and Yantosca, R. M.: A nested grid formulation for
- chemical transport over Asia: Applications to CO, J. Geophys. Res., 109, D22307,
- doi:10.1029/2004JD005237, 2004.
- 855 Wesely, M. L.: Parameterization of surface resistances to gaseous dry deposition in regional-
- scale numerical-models, Atmos. Environ., 23, 1293-1304, doi:10.1016/0004-6981(89)90153-
- 857 4, 1989.
- Wespes, C., Emmons, L., Edwards, D. P., Hannigan, J., Hurtmans, D., Saunois, M., Coheur, P.-
- F., Clerbaux, C., Coffey, M. T., Batchelor, R. L., Lindenmaier, R., Strong, K., Weinheimer, A.
- J., Nowak, J. B., Ryerson, T. B., Crounse, J. D., and Wennberg, P. O.: Analysis of ozone and
- 861 nitric acid in spring and summer Arctic pollution using aircraft, ground-based, satellite
- observations and MOZART-4 model: source attribution and partitioning, Atmos. Chem. Phys.,
- 863 12, 237–259, doi:10.5194/acp- 12-237-2012, 2012.
- 864 Wu, S., Mickley, L. J., Jacob, D. J., Logan, J. A., Yantosca, R. M., and Rind, D.: Why are there
- large differences between models in global budgets of tropospheric ozone?, J. Geophys. Res.,
- 866 112(D5), D05302, doi:10.1029/02006JD007801, 2007.
- Xiao, Y., Logan, J. A., Jacob, D. J., Hudman, R. C., Yantosca, R., and Blake, D. R.: The global
- budget of ethane and regional constraints on U.S. sources, J. Geophys. Res., 113, D21306,
- doi:10.1029/2007JD009415, 2008.
- Yang, J., Honrath, R. E., Peterson, M. C., Dibb, J. E., Sumner, A. L., Shepson, P. B., Frey, M.,

871	Jacobi, HW., Swanson, A., and Blake, N.: Impacts of snowpack emissions on deduced levels	
872	of OH and peroxy radicals at Summit, Greenland, Atmos. Environ., 36, 2523-2534,	
873	doi:10.1016/S1352-2310(02)00128-0, 2002.	
874	Yevich, R., and Logan, J. A.: An assessment of biofuel use and burning of agricultural waste in	
875	the developing world, Global Biogeochem. Cy., 17, 1095, doi:10.1029/2002GB001952, 2003.	
876	Yu, K., Jacob, D. J., Fisher, J. A., Kim, P. S., Marais, E. A., Miller, C. C., Travis, K. R., Zhu, L.,	
877	Yantosca, R. M., Sulprizio, M. P., Cohen, R. C., Dibb, J. E., Fried, A., Mikoviny, T., Ryerson,	
878	T. B., Wennberg, P. O., and Wisthaler, A.: Sensitivity to grid resolution in the ability of a	
879	chemical transport model to simu- late observed oxidant chemistry under high-isoprene	
880	conditions, Atmos. Chem. Phys., 16, 4369-4378, doi:10.5194/acp-16-4369-2016, 2016.	
881	Zatko, M. C., Geng, L., Alexander, B., Sofen, E. D., and Klein, K.: The impact of snow nitrate	Deleted: Yue, X., and Unger, N.: Ozone vegetation damage effects on gross primary productivity in the
881 882	Zatko, M. C., Geng, L., Alexander, B., Sofen, E. D., and Klein, K.: The impact of snow nitrate photolysis on boundary layer chemistry and the recycling and redistribution of reactive	
	<u> </u>	
882	photolysis on boundary layer chemistry and the recycling and redistribution of reactive	
882 883	photolysis on boundary layer chemistry and the recycling and redistribution of reactive nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem.	
882 883 884	photolysis on boundary layer chemistry and the recycling and redistribution of reactive nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem. Phys., 16, 2819-2842, doi:10.5194/acpd-16-2819-2016, 2016.	
882 883 884 885	photolysis on boundary layer chemistry and the recycling and redistribution of reactive nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem. Phys., 16, 2819-2842, doi:10.5194/acpd-16-2819-2016, 2016. Zhang, H., Wu, S., Huang, Y., and Wang, Y.: Effects of stratospheric ozone recovery on	
882 883 884 885 886	photolysis on boundary layer chemistry and the recycling and redistribution of reactive nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem. Phys., 16, 2819-2842, doi:10.5194/acpd-16-2819-2016, 2016. Zhang, H., Wu, S., Huang, Y., and Wang, Y.: Effects of stratospheric ozone recovery on photochemistry and ozone air quality in the troposphere, Atmos. Chem. Phys., 14, 4079-4086,	
882 883 884 885 886	photolysis on boundary layer chemistry and the recycling and redistribution of reactive nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem. Phys., 16, 2819-2842, doi:10.5194/acpd-16-2819-2016, 2016. Zhang, H., Wu, S., Huang, Y., and Wang, Y.: Effects of stratospheric ozone recovery on photochemistry and ozone air quality in the troposphere, Atmos. Chem. Phys., 14, 4079-4086, doi:10.5194/acp-14-4079-2014, 2014.	

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Table 1. Surface NO₂ measurements over Europe during 2009/12/01-2010/01/31.

Site ID	Site name	Lat. (°N)	Lon.(°E)	Altitude a.s.l (m)	Technique	Resolution
BE0001R	Offagne	49.88	5.20	430	chemiluminescence	hourly
BE0032R	Eupen	50.63	6	295	chemiluminescence	hourly
DE0001R	Westerland	54.93	8.31	12	NaJ_solution	daily
DK0008R	Anholt	56.72	11.52	40	UV_fluorescence	hourly
FI0096G	Pallas	67.97	24.12	340	chemiluminescence	hourly
GB0014R	High Muffles	54.33	-0.8	267	chemiluminescence	daily
NL0009R	Kollumerwaard	53.33	6.28	1	chemiluminescence	hourly
NO0001R	Birkenes	58.38	8.25	190	glass sinter	daily
NO0039R	Kårvatn	62.78	8.88	210	glass sinter	daily
NO0056R	Hurdal	60.37	11.08	300	glass sinter	daily
SE0005R	Bredkälen	63.85	15.3	404	abs_tube	daily

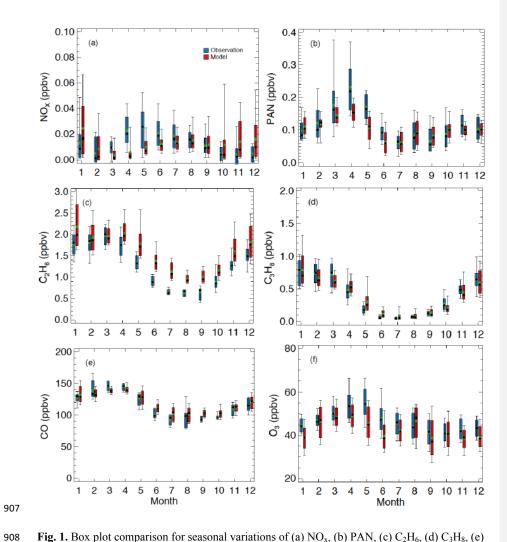


Fig. 1. Box plot comparison for seasonal variations of (a) NO_x , (b) PAN, (c) C_2H_6 , (d) C_3H_8 , (e) CO, and (f) O_3 between GEOS-Chem model simulations (red) and in-situ measurements (blue) over Summit for the period of 2008/07-2010/06. Data shown are monthly averages during this period. The thick (thin) bars represent the 67% (95%) confidence intervals. Black and green dots represent median and mean values, respectively. The statistics are based on daily averages.

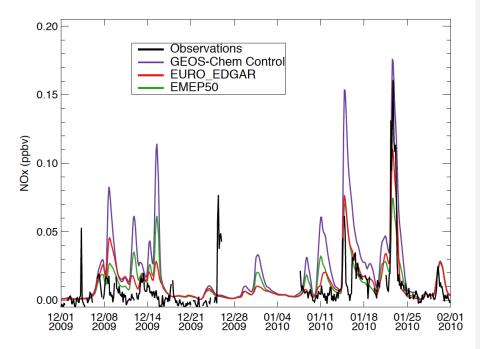


Figure 2. Timeseries of surface NO_x mixing ratios over Summit from observations, GEOS-Chem model control simulations, EURO_EDGAR, and EMEP50 during 2009/12/01-2010/01/31. EURO EDGAR represents simulations with anthropogenic NO_x emissions over Europe following EDGAR v4.2, while EMEP50 denotes simulations with anthropogenic NO_x emissions from the EMEP emission inventory over Europe reduced by 50%, with other model configurations identical to the control simulations. Readers are referred to the text for details.

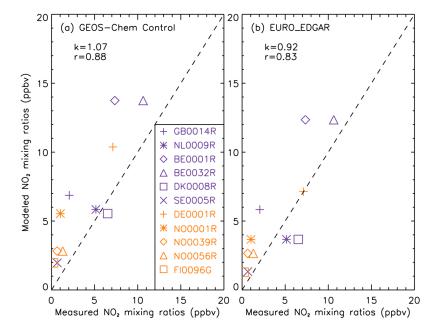


Figure 3. Scatter plots between measured monthly mean NO₂ mixing ratios at 11 observational sites over Europe and model simulations from (a) GEOS-Chem control simulations and (b) EURO_EDGAR during 2009/12/01-2010/01/31; also shown is the corresponding model-to-observation slopes (k) and correlation coefficients (r) for each panel. The dashed line is the 1:1 ratio. Explanations of site abbreviations are listed in Table 1. <u>EURO EDGAR represents simulations with anthropogenic NO_x emissions over Europe following EDGAR v4.2, with other model configurations identical to the control simulations.</u>

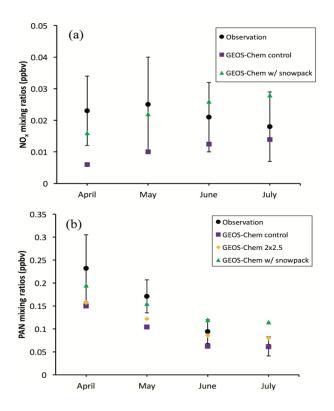


Figure 4. Monthly mean surface (a) NO_x and (b) PAN mixing ratios from observations (black circles), simulations with (green triangles) /without (purple squares) snowpack emissions, and GEOS-Chem simulations with horizontal grid resolution 2° x 2.5° (orange diamonds) over the period of April- July during 07/2008-06/2010. Vertical bars denote standard deviations over the course of observations for each month.

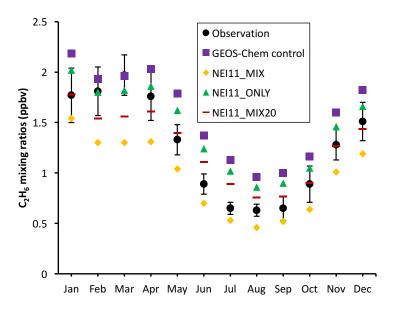


Figure 5. Monthly mean surface C_2H_6 mixing ratios at Summit from observations (black circles), GEOS-Chem model control simulations (purple squares), NEI11_MIX (orange diamond), and NEI11_ONLY (green triangles) simulations during 2008-2010; vertical bars denote the standard deviation over the course of observations for each month. NEI11_MIX represents model perturbations with global C_2H_6 emission inventories overwritten by NEI11 over US and by MIX over Asia, with other model configurations identical to the control simulations. NEI11_ONLY denotes the simulation that is the same with the control simulation except that the C_2H_6 emission inventory over US is overwritten by NEI11. NEI11_MIX20 is the simulation that is identical to NEI11_MIX except the 20% increased MIX C_2H_6 emission inventory over Asia. Readers are referred to the text for details.

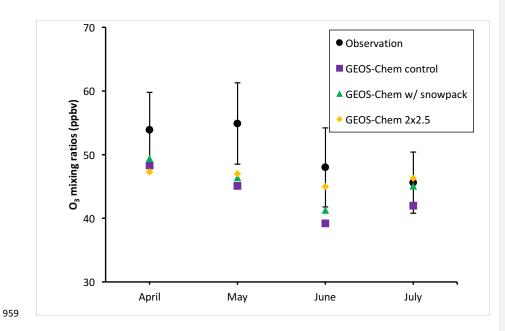


Figure 6. Monthly mean surface O_3 mixing ratios from observations (black circles), GEOS-Chem control runs (purple squares), with snowpack chemistry (green triangles), and horizonal grid resolution 2° x 2.5° (orange diamonds) for April-July. Vertical bars denote the variability over the course of observations for each month.

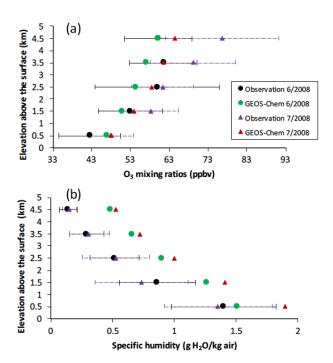


Figure 7. Comparisons of vertical profiles of (a) O_3 and (b) specific humidity between GEOS-Chem simulations and ozonesondes in June and July 2008 respectively, averaged over 1-km altitude bins. Black and green solid circles represent observations and simulations in June 2008 while purple and red triangles denote observations and simulations for July 2008 respectively. Solid and dash horizontal error bars represent observational standard deviations for June and July respectively.