

1 Environmentally dependent dust chemistry of a super Asian dust storm in
2 March 2010: observation and simulation

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17
18 **Abstract**

19 Near surface and vertical *in situ* measurements of atmospheric particles were
20 conducted in Shanghai during March 19-23, 2010 to explore the transport and
21 chemical evolution of dust particles in a super dust storm. An air quality model with
22 optimized physical dust emission scheme and newly implemented dust chemistry was
23 utilized to study the impact of dust chemistry on regional air quality. Two
24 discontinuous dust periods were observed with one travelling over Northern China
25 (DS1) and the other passing over the coastal regions of Eastern China (DS2). Stronger
26 mixing extents between dust and anthropogenic emissions were found in DS2,
27 reflecting by the higher SO₂/PM₁₀ and NO₂/PM₁₀ ratios as well as typical pollution
28 elemental species such as As, Cd, Pb, and Zn. As a result, the concentrations of SO₄²⁻

29 and NO_3^- and the ratio of Ca^{2+}/Ca were more elevated in DS2 than in DS1 but
30 opposite for the $[\text{NH}_4^+]/[\text{SO}_4^{2-}+\text{NO}_3^-]$ ratio, suggesting the heterogeneous reactions
31 between calcites and acid gases were significantly promoted in DS2 due to the higher
32 level of relative humidity and gaseous pollution precursors. Lidar observation showed
33 a columnar effect on the vertical structure of particle optical properties in DS1 that
34 dust dominantly accounted for ~80-90% of the total particle extinction from near the
35 ground to ~700m. In contrast, the dust plumes in DS2 were refrained within lower
36 altitudes while the extinction from spherical particles exhibited maximum at a high
37 altitude of ~800m. The model simulation reproduced relatively consistent results with
38 observations that strong impacts of dust heterogeneous reactions on secondary aerosol
39 formation occurred in areas where the anthropogenic emissions were intensive.
40 Compared to the sulfate simulation, the nitrate formation on dust is suggested to be
41 improved in the future modeling efforts.

42

43 **1. Introduction**

44 Asian dust originating from the arid and semiarid areas in Mongolia and China can
45 be transported for long distances, reaching Beijing (Sun et al., 2010), Shanghai (Fu et
46 al., 2010), Xiamen (Zhao et al., 2011), Taiwan (Tsai et al., 2012; Tsai et al., 2014), and
47 even as far as North America (Uno et al., 2009; Wu et al., 2015), exerting significant
48 impacts on the air quality of both densely populated habitations and remote regions.
49 Huang et al. (2014) showed that Asian dust could transport from the Qilian Mountain
50 or from the Qaidam Basin through Qinghai and Gansu provinces to reach the Pacific
51 Ocean, and that dust originating from the Taklimakan Desert could travel across the
52 Hexi Corridor and Loess Plateau to reach southeastern China. Zhao et al. (2009)
53 demonstrated that the deserts in Mongolia and in western and northern China were the
54 major sources of Asian dust particles in East Asia and estimated that 26% of the dust
55 particles emitted from Asian dust sources was transported to the Pacific Ocean.
56 Eguchi et al. (2009) reported that the dust plume from the Gobi Desert in East Asia

57 was transported at low altitudes of 4-6km to North America and mixed with Asian
58 anthropogenic air pollutants during its transport. Fu et al. (2014) simulated that during
59 a dust event from May 1 to 6, 2011, the transported dust particles accounted for
60 78.9% of the surface layer PM₁₀ over the Yangtze River Delta.

61 Dust particles can significantly influence the regional/global climate directly by
62 absorbing and scattering solar radiation (Bi et al., 2016) and also indirectly by
63 influencing the formation of ice nuclei, cloud, and precipitation (Creamean et al.,
64 2013; Li and Min, 2010; Wang et al., 2010). In addition, deposition of transported
65 dust particles into the ocean can enhance phytoplankton blooms due to the existence
66 of bioavailable iron (Wang et al., 2012; Zhuang et al., 1992), which indirectly impacts
67 on global climate change. The effects of dust particles on climate change depend
68 critically on their physical and chemical properties. Natural dust particles with limited
69 contamination have low light-absorption, with single-scattering albedo of 0.91-0.97 at
70 500nm and 550nm (Bi et al., 2014; Uchiyama et al., 2005). During the long-range
71 transport, dust particles are often modified by their mixing with anthropogenic
72 emissions over the downwind areas (Fischer et al., 2011; Formenti et al., 2011; Huang
73 et al., 2010b; Tobo et al., 2010), resulting in high uncertainties in evaluating the
74 climatic effects of dust particles. It was estimated that mineral dust had a radiative
75 forcing of $-0.1 \pm 0.2 \text{ Wm}^{-2}$ (IPCC, 2013), of which the uncertainty was as high as
76 200%. Obviously, the characteristics of dust particles and their evolution during the
77 transport are not well understood.

78 In March 2010, a super dust storm swept China, invading extensive areas from
79 Northern China to Southern China including Fujian and Guangdong provinces, and
80 lasting for ~4 days from March 19 to 23 (Li et al., 2011). The dust plumes further
81 extended to the South China Sea (Wang et al., 2011), Taiwan (Tsai et al., 2013), Korea
82 (Tatarov et al., 2012), Japan (Zaizen et al., 2014), and even to North America (Wu et
83 al., 2015). This dust storm was as strong as the one in March 20-21, 2002 and
84 attracted considerable attentions. Chen et al. (2017) used WRF-Chem to simulate the

85 emission and transport of dust particles over the Taklimakan Desert and Gobi Desert.
86 The results indicated that the Gobi Desert dust particles were easily lifted to 4km and
87 subject to the long-range transport, which contributed much more to the dust plumes
88 over East Asia than the Taklimakan Desert dust. Lidar observations revealed that this
89 super dust storm was transported within a low altitude (Tatarov et al., 2012; Wang et
90 al., 2011), which could benefit the mixing and interaction between dust particles and
91 anthropogenic pollutants. Indeed, modifications of dust particles during the transport
92 of this dust storm were suggested based on in situ measurements. Zhao et al. (2011)
93 displayed substantial increases of particulate sulfate and nitrate when the dust plumes
94 arrived at Xiamen city of Fujian province, implying the mixing and interaction
95 between dust particles and anthropogenic pollutants. Wang et al. (2011) indicated that
96 the dust particles detected at the Dongsha Island over the South China Sea were
97 mixed with anthropogenic and marine particles. Observations of this dust storm at
98 Tsukuba and Mt. Haruna, Japan showed that most of the transported dust particles in
99 lower altitudes were internally mixed with sulfate or seasalt (Zaizen et al., 2014).

100 Most of the studies on this super dust storm focused on investigating the dust
101 particles reaching Southeastern China and the South China Sea, and relied on single
102 method, e.g. aerosol chemistry measurement, optical property inversion, or model
103 simulation. In this study, we investigated this super dust storm in Shanghai, a coastal
104 city in Eastern China. A synergy of measurement techniques was applied, including in
105 situ measurements of pollutant gaseous precursors, particle with its major chemical
106 components, and Lidar observation of particle optical properties. To corroborate the
107 observational evidence, a regional numerical model was used to simulate the impact
108 of dust chemistry on the perturbation of regional air quality. What interested us is that
109 there were two discontinuous dust periods observed in Shanghai with distinctly
110 different transport pathways, providing a great opportunity to study the chemical
111 evolution of transported dust particles under different environmental conditions.

112

113 **2. Methodology**

114 **2.1. Field measurement**

115 **2.1.1. Lidar observation**

116 A dual-wavelength depolarization Lidar (Model:L2S-SMII) developed by the
117 National Institute for Environmental Studies (NIES) of Japan was installed on the roof
118 (~20m above ground level) of a teaching building on the campus of Fudan University
119 in the Yangpu District of Shanghai (Fig. 1b). The Lidar measurement was performed
120 every 15 min (at 00, 15, 30, and 45 minutes every hour) with a height resolution of 6
121 m. Attenuated backscattering coefficient (β), volume depolarization ratio (δv), particle
122 depolarization ratio (δp), and particle extinction coefficient (σ) at the wavelength of
123 532 nm were obtained by the measurement. More details about the Lidar system have
124 been described in Huang et al. (2012). δv is calculated using the parallel (I_p) and
125 perpendicular (I_s) components of backscatter intensity, and I_p and I_s were calibrated
126 before the calculation. Briefly, a sheet polarizer with the polarizing direction set at 45
127 degree (then -45 degree) to the polarizing plane of the emitted light was installed in
128 front of the beam splitter cube, and two sets of backscatter signal profiles from the sky
129 were obtained for the calibration. Detailed calibration procedure has been described in
130 Shimizu et al. (2004) and Shimizu et al. (2017). σ was derived by the Fernald
131 inversion method (Fernald, 1984) with the lidar ratio (extinction-to-backscatter ratio)
132 set as 50 sr (Liu et al., 2002) in the inversion process. The total particle extinction
133 coefficient can be split to non-spherical particle (dust particle, σ_d) and spherical
134 particle (mostly pollution particle, σ_s) fractions based on the value of δp . The splitting
135 method was described in detail by Sugimoto et al. (2002) and Shimizu et al. (2004).

136 To solve the problem that overlap of the laser beam and the view field of telescope
137 is insufficient for near surface observation, a compensation function $Y(z)$ was applied.
138 Function $Y(z)$ was derived from the signal profiles that observed on a day when the
139 planetary boundary layer was well developed. With the compensation, the optical
140 properties of the particles above 120 m altitude were provided. Detailed correction

141 procedure has been described in Shimizu et al. (2017).

142

143 **2.1.2. Online particle and gases monitoring**

144 Continuous PM₁₀ concentrations were measured by a TEOM (Tapered Element
145 Oscillating Microbalance) 1405D monitor (Thermo Scientific, USA). Trace gases SO₂
146 and NO₂ were measured by a 43i SO₂ analyzer (Thermo Scientific, USA) and a 42i
147 NO-NO₂-NO_x analyzer (Thermo Scientific, USA), respectively. All the measured
148 PM₁₀ and gases concentrations were averaged and used at intervals of 1hr in this
149 study.

150

151 **2.1.3. Aerosol sampling**

152 TSP (Total Suspended Particles) samples were collected during March 19-27, 2010
153 at the Fudan observational site co-located with all the other instruments. The aerosol
154 samples were collected for 12 hours (normally from 8:00 to 20:00LST (Local
155 Standard Time) in daytime and from 20:00 to 8:00 LST of the next day in nighttime)
156 or 24 hours (normally from 8:00 to 8:00 LST of the next day) on Whatman 41 filters
157 (Whatman Inc., Maidstone, UK) by a medium-volume sampler (Beijing Geological
158 Instrument-Dickel Co., Ltd.; model: TSP/PM₁₀/PM_{2.5}-2; flow rate: 77.59 L min⁻¹). All
159 the samples were put in polyethylene plastic bags immediately after sampling and
160 then reserved in a refrigerator. The filters were weighed before and after sampling
161 using an analytical balance (Model: Sartorius 2004MP; reading precision: 10μg) after
162 stabilizing in constant temperature (20±1°C) and humidity (40±2%) for 48 hours. All
163 the procedures were strictly quality controlled to avoid the possible contamination of
164 the samples.

165

166 **2.2. Chemical analysis**

167 **2.2.1. Ion analysis**

168 One fourth of each aerosol sample and blank filters were extracted ultrasonically by

169 10 ml deionized water ($18 \text{ M}\Omega \text{ cm}^{-1}$). Inorganic ions of SO_4^{2-} , NO_3^- , Cl^- , Na^+ , NH_4^+ ,
170 K^+ , Mg^{2+} , and Ca^{2+} were analyzed by an Ion Chromatography (Dionex ICS 3000,
171 USA) with a separation column of DionexIonpac AS 11, a guard column of Dionex
172 Ionpac AG 11, a self-regenerating suppressed conductivity detector of Dionex Ionpac
173 ED50, and a gradient pump of Dionex Ionpac GP50. The detailed analytical
174 procedures can be found in Yuan et al. (2003).

175

176 **2.2.2. Element analysis**

177 Half of each aerosol sample and blank filters were digested at $170 \text{ }^\circ\text{C}$ for 4 hours in
178 a high pressure Teflon digestion vessel with 3ml concentrated HNO_3 , 1ml
179 concentrated HClO_4 , and 1 ml concentrated HF. The solutions were dried, and then
180 diluted to 10 ml with deionized water ($18 \text{ M}\Omega \text{ cm}^{-1}$). Fifteen elements (Al, As, Ca, Cd,
181 Cu, Fe, Mg, Mn, Na, Ni, Pb, Sr, Ti, V, and Zn) were measured by an inductively
182 coupled plasma optical emission spectroscopy (ICP-OES; SPECTRO, Germany). The
183 detailed analytical procedures were described in Sun et al. (2004a) and Zhuang et al.
184 (2001).

185

186 **2.3. Nation-wide daily PM_{10} data**

187 Air pollution index (API) data in 86 major cities (locations shown in Fig. 1a) over
188 China were obtained from the data center of Ministry of Environmental Protection of
189 China (<http://datacenter.mep.gov.cn/>). In China, six grades of air pollution of
190 excellent, good, slightly polluted, lightly polluted, moderately polluted, and heavily
191 polluted were set corresponding to the API scales of 0-50, 51-100, 101-150, 151-200,
192 201-300, and >300 , respectively. Both API values of 50, 100, 200, 300, 400, and 500
193 and their corresponding concentrations of air pollutants were defined in the API
194 grading limited value table as shown in Table S1. According to the definition of API
195 in China, the API value of air pollutants was calculated as

$$196 \quad I_x = (I_{x, \text{high}} - I_{x, \text{low}})(C_x - C_{x, \text{low}})/(C_{x, \text{high}} - C_{x, \text{low}}) + I_{x, \text{low}},$$

197 where C_x and I_x are the concentration and the API value of air pollutant X in Table S1,
198 respectively. $I_{x, high}$ and $I_{x, low}$ stand for the two values in the API grading limited value
199 table (Table S1) that mostly approach to value I_x , respectively. $C_{x, high}$ and $C_{x, low}$
200 represent the concentration of X corresponding to $I_{x, high}$ and $I_{x, low}$, respectively. And
201 the daily API value is defined as

$$202 \quad \text{API} = \text{Max}(I_{\text{PM}_{10}}, I_{\text{SO}_2}, I_{\text{NO}_2}, I_{\text{CO}}, I_{\text{O}_3}).$$

203 According to API data records from March 20 to 23, 2010, PM_{10} was the premier air
204 pollutant in most of the 86 cities over China, i.e. $\text{API} = I_{\text{PM}_{10}}$. Thus, the API value can
205 be converted to PM_{10} concentration as

$$206 \quad C = (I - I_{\text{low}})(C_{\text{high}} - C_{\text{low}})/(I_{\text{high}} - I_{\text{low}}) + C_{\text{low}}.$$

207 It should be noted that API was recorded with a maximum value of 500, which
208 corresponded to the PM_{10} concentration of $600 \mu\text{g m}^{-3}$.

209

210 **2.4. Backward trajectory analysis**

211 48 hours back trajectories of the air masses at both 250 m and 1000 m AGL (Above
212 Ground Level) during dust days starting at Shanghai were computed by the HYSPLIT
213 model (<http://ready.arl.noaa.gov/hypub-bin/trajtype.pl?runtype=archive>), using the
214 meteorological data of GDAS (1 degree, global, 2006-present). Four trajectories
215 ending at 0, 6, 12, and 18 LST were calculated for each day.

216

217 **2.5. Model simulation**

218 The WRF/CMAQ modeling system was applied to simulate the dust chemistry in
219 this study. The Weather Research and Forecasting model (WRFv3.4) was used to
220 produce the meteorology fields by digesting the reanalysis data from National Centers
221 for Environmental Prediction (NCEP). The Community Multiscale Air Quality
222 Modeling System (CMAQv5.0.1) was configured with the 2005 carbon bond
223 gas-phase mechanism (CB05) and aerosol module AE6. The default CMAQ model
224 doesn't include dust chemistry reactions but only the dust emission module. We have

225 implemented dust chemistry in the CMAQ model and corrected the dust emission
226 module in our earlier work (Dong et al., 2015), which was used to simulate the
227 heterogeneous reactions on dust during the long-range transport in this study. Briefly,
228 the major developments included: (1) The default dust emission module in CMAQ
229 was found to strongly underestimate the dust emissions. By removing the double
230 counting of soil moisture in the default dust emission module and conducting a
231 reanalysis of field data, the threshold friction velocities over various land covers were
232 re-adjusted. (2) The source-dependent speciation profiles of dust particles from the
233 Taklimakan and Gobi Desert were implemented based on field measurement data. (3)
234 Thirteen dust heterogeneous reactions were implemented, including dust reactions
235 with O₃, OH, H₂O₂, CH₃COOH, CH₃OH, CH₂O, HNO₃, N₂O₅, NO₂, NO₃, HO₂, and
236 SO₂. The uptake coefficients of gases onto the surface of dust particles were taken
237 from previous published studies in deserts of China. More details of the technical
238 development of this dust emission and chemistry module in CMAQ and the
239 evaluation of model performance can be found in Dong et al. (2015).

240 The modeling domain includes whole China with a horizontal grid resolution of 36
241 km×36 km and 34 vertical layers with a model top at 50 hPa. The chemical initial and
242 boundary conditions for CMAQ were downscaled from the GEOS-Chem global
243 model. Emissions inputs included anthropogenic emissions from Zhao et al. (2013)
244 over China, biogenic emissions from MEGAN2.1 (Guenther et al., 2006), and
245 biomass burning emissions from FLAMBE (Reid et al., 2009).

246

247 **3. Results and Discussion**

248 **3.1. Spatiotemporal evolution of the 2010 spring dust storm over China**

249 On March 19, 2010, a super dust storm outbreak over the Gobi Desert of Inner
250 Mongolia, China and southern Mongolia. It was driven by a strong cold front and
251 revealed evidently by the OMI (Ozone Monitoring Instrument) aerosol index (AI)
252 from space detection (Tatarov et al., 2012). Along with the cold front, the dust plumes

253 moved southeastward, invading extensive areas from Northern China to Southern
254 China. Fig. 2 shows the daily PM₁₀ concentrations over 86 cities of China from March
255 19 to 23, 2010, displaying the spatiotemporal evolution of this dust storm event. On
256 March 20, daily PM₁₀ concentrations of higher than 200 μgm⁻³ were observed over
257 widespread areas of Northern China, close to the dust source regions. The PM₁₀
258 concentrations of those heavily polluted cities, including Yinchuan (YC), Datong
259 (DT), Hohhot (HHT), and Beijing (BJ) (red circles in Fig. 1a,) reached the threshold of
260 600 μgm⁻³, the maximum value recorded by API (Fig. 2a). As the dust plumes
261 transported, high levels of PM₁₀ started to emerge over Central and Eastern China on
262 March 21, when heavy pollution with daily PM₁₀ concentrations higher than
263 420μgm⁻³ were observed over the Yangtze River Delta region as well as in Jiangxi,
264 Hunan, and Hubei provinces. Specifically, the PM₁₀ concentrations in Shanghai,
265 Nanjing (NJ), Hefei (HF), Wuhan (WH), Hangzhou (HZ), and so on (blue circles in
266 Fig. 1a) all reached the threshold of 600 μgm⁻³ (Fig. 2c). On March 22 and 23, the
267 dust plumes drifted towards Fujian and Guangdong provinces in Southern China,
268 where PM₁₀ concentrations of over 600μgm⁻³ were even observed in coastal cities
269 such as Fuzhou (FZ), Xiamen (XM), and Shantou (ST) (pink circles in Fig.1a). The
270 severe air pollution over China caused by this super dust storm lasted for ~4 days
271 until March 23, resulting in significant impacts on the regional air quality and possible
272 perturbation on regional meteorology.

273

274 **3.2. Identification of two dust plumes with distinct transport pathways**

275 Fig. 3a shows the time-height cross-section of δv (volume depolarization ratio)
276 measured at the wavelength of 532 nm from March 19 to 23 in Shanghai. δv is
277 frequently used to identify dust events and a threshold value of 10% is used to
278 distinguish dust from other types of particles (Shimizu et al., 2004). As shown in Fig.
279 3a, there were evidently two discontinuous periods with δv higher than 10%,
280 consistent with the two peaks of PM₁₀ concentrations measured near the ground (Fig.

281 4c). The first dust episode (DS1) started from ~16:00 LST, March 20 to ~10:00LST,
282 March 21 and the second dust episode (DS2) started from ~6:00 LST, March 22 to
283 ~0:00 LST, March 23.

284 The 48-hours backward trajectories of the air masses during March 20-23 in
285 Shanghai are shown in Fig. 2. It is interesting to note that the transport pathways of
286 the dust plumes in DS1 and DS2 were distinctly different. In DS1, the dust plumes at
287 both low altitudes (i.e. 250m denoted by the black lines in Fig. 2c-d) and high
288 altitudes (i.e. 1000m denoted by the red lines in Fig. 2c-d) were mostly transported
289 from the dust source regions in the Gobi Desert. This is one of the typical inland
290 transport pathways of Asian dust, which passed over Northern China that is
291 characterized of intense anthropogenic emission rates, e.g. from Shanxi, Hebei, and
292 Shandong provinces.

293 In DS2, the air masses reaching Shanghai are shown in Fig. 2e – h. From the
294 starting of DS2 to the midday of March 22 (Fig. 2e-f), the backward trajectories were
295 still mainly from the north but we noticed that the low altitude trajectories which
296 originated from the Gobi Desert travelled over the Yellow Sea and East China Sea
297 and then circled back to Shanghai. After the midday of March 22 (Fig. 2g-h), the
298 trajectories became much shorter and restricted within the costal and offshore areas.
299 The low-altitude trajectories were mainly from the ocean and the high-altitude
300 trajectories shifted from the south. As indicated by the surface observation (Fig. 4a),
301 the southeasterly winds prevailed until ~18:00 LST, March 22 when PM_{10} climbed to
302 reach its peak value in DS2, quite different from DS1 when the northerly winds
303 dominated. The CALIPSO (Cloud-Aerosol Lidar and Infrared Pathfinder Satellite
304 Observations) transect at around 13:00 – 13:30 LST, March 21 specified that the
305 major aerosol type was dust over the East China Sea and South China Sea (Fig. S1).
306 Therefore, it was evident that DS2 passed over the coastal regions of Eastern China
307 and the ocean and it should be a humid dust plume.

308 As visualized in Fig. 3, the two dust plumes were mostly transported at low

309 altitudes below 1km. It is expected that this type of transport would benefit for the
310 mixing and interactions between dust particles and anthropogenic pollutants, as the
311 anthropogenic emissions were mainly trapped within the boundary layer. Transport
312 pathway is one of the most important factors accounting for the evolution of dust
313 particles and the resulting environmental effects (Zaizen et al., 2014; Zhang et al.,
314 2010). In this super dust storm, the two discontinuous dust plumes reaching Shanghai
315 were distinctly different in their transport pathways, providing a great opportunity to
316 study the chemical evolution of dust particles under different transport conditions.

317

318 **3.3. Distinct behaviors of gaseous pollutants between two dust episodes**

319 Fig. 4 illustrates the different behaviors of the two identified dust episodes by
320 plotting the temporal variations of emission precursors (i.e. SO₂ and NO₂) as well as
321 PM₁₀ and crucial meteorological parameters. As shown in Fig. 4a, the winds prevailed
322 from the south before the onset of DS1 at ~10:00LST, March 20. Starting from early
323 morning of March 19, SO₂ and NO₂ continuously climbed up due to the enhanced
324 human activities. From 18:00 LST of March 19 to 6:00 LST of March 20, the
325 concentrations of NO₂ and SO₂ synchronously decreased. This was partly attributed to
326 the prevailing southeast winds, i.e. sea breezes which had clean effects on the local
327 pollution. Furthermore, the decrease of NO₂ and SO₂ occurred during night, when the
328 anthropogenic emissions were relatively low during a day. During this period, SO₂
329 and NO₂ concentrations were relatively high with mean concentrations of 39 ± 19 and
330 $70 \pm 25 \mu\text{gm}^{-3}$, respectively. In the meantime, PM₁₀ was at a moderate level of $112 \pm$
331 $54 \mu\text{gm}^{-3}$, suggesting the dominance by anthropogenic emissions. It was observed that
332 the peak hourly concentrations of SO₂ and NO₂ up to $97 \mu\text{gm}^{-3}$ and $116 \mu\text{gm}^{-3}$ occurred
333 just a few hours before the sharp increase of PM₁₀. This was due to that air pollutants
334 north of Shanghai were pushed by the cold front and accumulated before the dust
335 plumes invaded (Guo et al., 2004), which was also evident from the time difference
336 between the values of β and δv as shown in Fig. 3.

337 From 10:00 LST, March 20 to 11:00 LST, March 21, the winds shifted from the
338 north and northwest accompanied with a quick drop of relative humidity (RH) with a
339 minimum of 25% at 21:00 LST, March 20, indicating the invasion of a strong cold
340 front from Northern China. Correspondingly, PM₁₀ abruptly climbed since 16:00 LST
341 and reached 1000 μgm⁻³ (a maximum value of 1000 μgm⁻³ was set in the TEOM
342 1405D monitor) within four hours. This high level of PM₁₀ lasted for ~11 hours till
343 04:00 LST, March 21. During this dust episode, the concentrations of SO₂ and NO₂
344 decreased substantially and the lowest values of 17 μgm⁻³ and 27 μgm⁻³ (Fig. 4c) were
345 recorded at 04:00 LST, March 21, due to the strong dilution effect of dust plumes. The
346 ratios of SO₂/PM₁₀ and NO₂/PM₁₀ were as low as 0.04 ± 0.02 and 0.06 ± 0.03 in DS1
347 compared to that of 0.30 ± 0.05 and 0.57 ± 0.28 before DS1. After 12:00 LST, March
348 21, the wind direction started to shift again from the southeast with an increase of RH.
349 The PM₁₀ concentrations quickly decreased below 200 μgm⁻³ within five hours,
350 indicating the pass of the first dust episode over Shanghai.

351 The second wave of dust (DS2) commenced after about half a day as indicated by
352 the temporal variation of PM₁₀ climbing quickly since 6:00 LST, March 22. PM₁₀
353 reached its peak of 530 μgm⁻³ observed at ~18:00 LST, March 22, much lower than the
354 peak values during DS1. It should be noted that during the growth of DS2 (i.e. from
355 06:00 to 18:00 LST), southerly winds sustained until 18:00 LST, March 22, opposite
356 to DS1. As discussed above, DS2 had gone through the maritime environment during
357 most of its transport trajectory. This can be corroborated by the moderately high RH
358 (Fig. 4b). After 18:00 LST, March 22, PM₁₀ underwent a quick decline followed by
359 the prevailing northeast winds and continuously elevated RH. This suggested the sea
360 breezes after 18:00 LST were almost free of dust particles and acted as cleaner for the
361 dust pollution that was previously accumulated. Compared to the mean RH of 46% ±
362 18% in DS1, RH in DS2 was much higher of 69% ± 8%. This was mainly attributed
363 to the different transport pathways of the two dust episodes.

364 One interesting phenomenon that has been rarely observed was that the temporal

365 variations of SO₂ and NO₂ varied fairly consistent with that of PM₁₀ in DS2 (Fig. 4c),
366 quite different from DS1 and previous studies (Fu et al., 2010; Guo et al., 2004) that
367 dust usually had a clean effect on the local gaseous pollutants, causing inverse
368 relationship between SO₂/NO₂ and PM₁₀. In this case, NO₂ reached its maximum
369 hourly concentration of 131 μg m⁻³ along with the maximum PM₁₀ in DS2, the highest
370 during the whole study period. This probably indicated that the dust particles in DS2
371 were externally mixed or “coated” with abundant gaseous pollutants. As a result, the
372 SO₂/PM₁₀ and NO₂/PM₁₀ ratios reached 0.11 ± 0.03 and 0.20 ± 0.04 in DS2,
373 respectively, much higher than those in DS1 and in a super dust day of April 2, 2007
374 in Shanghai (Fu et al., 2010), revealing that the dust plumes in DS2 were much more
375 polluted.

376 In the following sections, we will investigate deeply into the chemical
377 characteristics and evolution of the two dust episodes.

378

379 **3.4. Chemical evolution of dust particles**

380 **3.4.1. Pollution elements significantly enhanced in dust periods**

381 Table 1 lists the concentrations of measured elements in TSP before, during, and
382 after the two dust episodes in Shanghai. D and N represent the samples that collected
383 in daytime (~8:00 to ~20:00 LST) and nighttime (~20:00 to 8:00 LST in the next day),
384 respectively. NDS represents the non-dust days from March 25 to 27 and the mean
385 concentrations are shown in Table 1. To identify whether these elements were mainly
386 derived from the crustal source or anthropogenic sources, enrichment factors of
387 elements were calculated. Enrichment factor (EF) is defined as
388 $EF = (X/X_{Ref})_{particle} / (X/X_{Ref})_{crust}$, where $(X/X_{Ref})_{particle}$ is the mass ratio of a given
389 element X to the reference element in particle and $(X/X_{Ref})_{crust}$ is the ratio in the crust
390 (Lida, 2006), and Al is the reference element used in this study. EFs of Al, Cu, Ca, Fe,
391 Mg, Na, Mn, Ti, Sr, Ni, and V were calculated to be lower than 5, indicating these
392 eleven elements were mostly from the crustal source, while EFs of As, Cd, Pb, and Zn

393 were higher than 10 even during the dust periods (Fig.5c), indicating these four
394 elements were significantly influenced by anthropogenic sources. Hence, Al, Cu, Ca,
395 Fe, Mg, Na, Mn, Ti, Sr, Ni, and V were classified as “crustal elements” while As, Cd,
396 Pb, and Zn were classified as “pollution elements”. Due to the dilution effect of dust
397 plumes, EFs of the four pollution elements decreased from ~100-150 in NDS to
398 ~10-50 in DS1 and ~20-115 in DS2.

399 As shown in Table 1, the concentrations of the crustal elements before the dust
400 event (19N) were as low as those in NDS, but substantially increased in DS1 and DS2,
401 about 3-13 and 1-7 folds of those in NDS, respectively. The highest concentration of
402 Al, widely used as a tracer for dust, reached $67.5 \mu\text{g m}^{-3}$ in DS1 (20N), comparable to
403 the Al concentration in the super dust day of April 2, 2007 in Shanghai (Fu et al.,
404 2010). The highest concentrations of crustal elements in DS1 (20N) were ~2 folds of
405 those in DS2 (22D), corroborating that the intensity of DS1 was stronger than in DS2.

406 As for the pollution elements, the non-crustal part (nc-) of As, Cd, Pb, and Zn were
407 estimated as $\text{nc-X} = X - \text{Al} \times (\text{X}/\text{Al})_{\text{crust}}$. The concentrations of nc-As, nc-Cd, nc-Pb,
408 and nc-Zn increased in both DS1 and DS2, about 1-4folds of those in NDS (Fig. 5a).
409 As these elements were mainly derived from anthropogenic sources such as coal
410 combustion, industrial processing, vehicle emissions, etc., their enhancement
411 indicated that abundant pollutants had been transported to the downwind regions
412 along with the dust plumes. The mean concentrations of nc-As, nc-Cd, nc-Pb, and
413 nc-Zn were 23.5, 3.6, 154.2, and 580.1 ng m^{-3} in DS2, higher than those of 18.3, 3.1,
414 119.7, and 447.4 ng m^{-3} in DS1, indicating DS2 was more polluted. Consistently, the
415 mass ratios of nc-As, nc-Cd, nc-Pb, and nc-Zn in TSP were $5.1 \times 10^{-3}\%$, $7.2 \times 10^{-4}\%$,
416 $3.1 \times 10^{-2}\%$, and $1.2 \times 10^{-1}\%$ in DS2, higher than those of $1.9 \times 10^{-3}\%$, $3.7 \times 10^{-4}\%$,
417 $1.5 \times 10^{-2}\%$, and $6.2 \times 10^{-2}\%$ in DS1 (Fig. 5b).

418

419 **3.4.2. Crustal vs. secondary water-soluble ions**

420 Fig. 6 shows the evolution of major water-soluble ions in TSP during the whole

421 dust event. The concentrations of Na^+ , K^+ , Mg^{2+} , and Ca^{2+} increased the most among
422 all the ions in both DS1 and DS2 due to their crustal origin, ~ 2 -5 and ~ 2 -3 folds of
423 those in NDS. However, the mass ratios of these ions in TSP generally decreased in
424 the dust periods (Fig. 6d), as the masses of dust particles were dominated by
425 water-insoluble matters. Different from NDS, the concentrations of Ca^{2+} exceeded
426 NH_4^+ and ranked as the most abundant cation in the dust periods. This was because
427 Asian dust particles are rich of calcium carbonate (Wang et al., 2005), which could
428 react with acidic nitrogen and sulfur compounds to form $\text{Ca}(\text{NO}_3)_2$ and CaSO_4 . It has
429 been found Ca-rich particles in Asian dust transported to Japan were mostly in the
430 spherical shape (Zaizen et al., 2014), and many of the spherical Ca-rich particles
431 contained sulfur and nitrogen compounds (Matsuki et al., 2006; Zaizen et al., 2014)
432 due to the heterogeneous reactions. Recently, Pan et al. (2017) reported that the
433 concentrations of both NO_3^- and Ca^{2+} increased in coarse mode mineral dust in
434 Beijing, particularly at high RH condition due to the interaction between nitric acid
435 and Ca-rich particles. It is further suggested that the impact of nitrate on modifying
436 the morphology of dust particles have become increasingly important, as the NOx
437 emissions in East Asia have been rapidly increasing. It is noted that the concentrations
438 of Ca^{2+} in DS2 were comparable to that in DS1 (Fig. 6b), although the intensity of
439 DS1 was much stronger than DS2. In addition, the mass ratios of Ca^{2+} in TSP were
440 even higher in DS2 than in DS1 (Fig. 6d). Collectively, these results indicated that
441 more calcium in its soluble form was produced via the reactions between calcium
442 carbonate and acids. This could be also revealed by the ratio of Ca^{2+}/Ca that was
443 higher in DS2 (0.2-0.5) than in DS1 (0.1-0.2) (Fig.7a). In DS2, the dust plumes
444 travelled over the ocean and carried higher amount of water vapor as well as SO_2 and
445 NO_2 than DS1 as discussed earlier. In this regard, the heterogeneous reaction between
446 calcium carbonate and acidic gases was enhanced more in DS2, resulting in a higher
447 fraction of calcium carbonate from dust particles that could be transformed to soluble
448 calcium.

449 Cl⁻ during the dust periods was moderately enhanced by ~1-2 folds compared to
450 NDS, indicating that Cl⁻ was also impacted by the invasion of dust plumes. As shown
451 in Fig. 6a & 6c, both the concentrations of Cl⁻ and its mass ratios in TSP were higher
452 in DS2 than in DS1. As the air masses of DS2 passed over the ocean, part of the high
453 Cl⁻ concentrations should be attributed to the contribution from seasalts. The average
454 equivalent ratio of Cl⁻ to Na⁺ ([Cl⁻]/[Na⁺]) was 1.65 in NDS and 2.17 in 19N, higher
455 than the ratio in seawater (1.17), indicating that anthropogenic sources such as coal
456 combustion (Sun et al., 2014; Yao et al., 2002) contributed significantly to Cl⁻ in the
457 non-dust days. The [Cl⁻]/[Na⁺] ratio dropped to ~1.0 in DS1 as massive continental
458 particles invaded Shanghai, while in DS2 the ratio was elevated to 1.5. Thus, both
459 seasalts and anthropogenic sources should contribute to the particulate Cl⁻ in DS2.

460 As for the secondary inorganic ions (i.e. SO₄²⁻, NO₃⁻, and NH₄⁺), their
461 concentrations in TSP were elevated to be 16.2, 15.4, and 7.2 μg m⁻³ in the daytime of
462 March 20 (20D) in DS1. The sampling time of 20D was from ~8:00 to 20:00 LST,
463 including the short period of pollutants accumulation before the arrival of DS1. As a
464 result, the high concentrations of SO₄²⁻ and NO₃⁻ measured in 20D were contributed
465 from both local and transported sulfate and nitrate, as well as from the heterogeneous
466 reactions on dust particles. In the nighttime of March 20 (20N), the concentrations of
467 NO₃⁻ and NH₄⁺ sharply decreased by 57.8% and 43.2%, respectively. Although SO₄²⁻
468 also decreased, its reduction extent was much weaker of 27.8% and was still ~2 folds
469 of that in NDS (Fig. 6a). Nitrate formation on dust is strongly dependent on the
470 ambient conditions. Low temperature, low relative humidity, strong wind and low
471 concentrations of pollution gases did not favor the heterogeneous reaction (Huang et
472 al., 2010a; Yuan et al., 2008). Hence, strong dust events usually diluted the
473 concentrations of nitrate (Duvall et al., 2008; Huang et al., 2010a; Wang et al., 2006).
474 As for the sulfate formation on dust, SO₂ could interact with various mineral
475 components of dust particles to produce sulfate, and sulfate became mixed with the
476 dust first competing with nitrate (Dupart et al., 2012; Sullivan et al., 2007). Hence,

477 although the secondary particle components were diluted by the invasion of dust, the
478 formation of sulfate could still be compensated by the heterogeneous reaction on dust
479 to some extent.

480 In the daytime of March 22 (22D) during DS2, the concentrations of SO_4^{2-} and
481 NO_3^- reached the highest during the study period with values of 16.6 and 19.3 μgm^{-3} ,
482 respectively. Although SO_4^{2-} and NO_3^- decreased in the nighttime of March 22 (22N),
483 their levels were still higher than the non-dust days (Fig. 6a). Also, the mass ratios of
484 SO_4^{2-} and NO_3^- in TSP were higher in DS2 than DS1 (Fig. 6c). These results indicated
485 that SO_4^{2-} and NO_3^- were more favorably formed in DS2. It was observed that the
486 concentration of NO_3^- was even higher than that of SO_4^{2-} in both daytime and
487 nighttime samples during DS2, showing a contrary behavior to DS1. As discussed
488 above, the formation of nitrate on dust strongly depends on the ambient conditions.
489 During DS2, RH was at a moderately high level, enhancing the efficiency of aqueous
490 processing on the particles. In addition, the abnormally high concentrations of NO_2
491 (Fig. 4c) suggested the emission precursors were sufficient for the production of
492 nitrate. Finally, high concentrations of nitrate during DS2 may be partially attributed
493 to the reaction between sea salts and nitric acid (Hsu et al., 2014; Huang et al., 2010a)
494 as the dust plumes travelled over the ocean. The formation of secondary aerosol
495 species during DS2 should have involved the complex interactions between pollutant
496 precursors, dust, and sea salts.

497 Fig. 7b shows the equivalent ratio of the total anions to the total cations (A/C) in
498 TSP. The A/C ratio dropped to ~ 0.5 in DS1 from ~ 0.9 in 19N, which was attributed to
499 the existence of abundant CO_3^{2-} in dust particles that can't be detected by Ion
500 Chromatography (Huang et al., 2010a). The A/C ratio reached 0.85 and 0.80 in 22D
501 and 22N in DS2, respectively, much higher than those in DS1, suggesting that more
502 fraction of CaCO_3 in the dust particles in DS2 was transformed to CaSO_4 and
503 $\text{Ca}(\text{NO}_3)_2$ than in DS1, consistent with the higher Ca^{2+}/Ca ratio in DS2 than in DS1
504 (Fig. 7a). The equivalent ratio of NH_4^+ to the sum of SO_4^{2-} and NO_3^- ($[\text{NH}_4^+]/$

505 $[\text{SO}_4^{2-} + \text{NO}_3^-]$) in TSP also indicated that the heterogeneous reactions between
506 carbonate and acid gases were more promoted in DS2 compared to DS1. As shown in
507 Fig. 7c, the $[\text{NH}_4^+]/[\text{SO}_4^{2-} + \text{NO}_3^-]$ ratio was slightly higher than one in the non-dust
508 days, indicating that SO_4^{2-} and NO_3^- could be completely neutralized by NH_4^+ . In DS1,
509 the $[\text{NH}_4^+]/[\text{SO}_4^{2-} + \text{NO}_3^-]$ ratio was 0.68 and 0.65 in 20D and 20N, while in DS2 it was
510 even lower of 0.45 and 0.55 in 22D and 22N, respectively. Apparently, NH_4^+ was
511 insufficient for completely neutralizing SO_4^{2-} and NO_3^- in both dust episodes,
512 particularly in DS2. The ammonium deficiency was also observed in Kinmen and
513 Zhuhai (Hsu et al., 2014), an island site and a coastal site, respectively, during this
514 super dust storm. We further investigated the $[\text{NH}_4^+ + \text{Ca}^{2+} + \text{Mg}^{2+}]/[\text{SO}_4^{2-} + \text{NO}_3^-]$ ratio.
515 As shown in Fig. 7d, with the addition of Ca^{2+} and Mg^{2+} , sulfate and nitrate had been
516 completely neutralized, implying the important role of alkaline calcium and
517 magnesium as the medium of dust heterogeneous reactions. By estimating the
518 neutralization efficiency of Ca^{2+} and Mg^{2+} ($\text{NE}_{\text{Ca\&Mg}}$) as $\text{NE}_{\text{Ca\&Mg}} = 1 -$
519 $[\text{NH}_4^+]/[\text{SO}_4^{2-} + \text{NO}_3^-]$, the average value of $\text{NE}_{\text{Ca\&Mg}}$ in DS1 and DS2 was 0.34 and
520 0.50, respectively. The higher $\text{NE}_{\text{Ca\&Mg}}$ in DS2 also suggested the chemical
521 processing via dust was efficient under the environmental conditions such as DS2 in
522 this study.

523

524 3.5. Vertical evolution of dust particles

525 In both DS1 and DS2, dust particles were mostly refrained below the altitude of
526 1km. Strong vertical gradients of δv and β were observed (Fig. 3). To quantitatively
527 investigate the evolution of vertical dust profiles, we selected four typical episodes as
528 follows: 1. 9:00-15:00 LST of March 20: an episode right before the arrival of DS1; 2.
529 16:00 - 17:45 LST of March 20: an episode before the onset of maximum hourly
530 PM_{10} concentrations in DS1; 3. 18:00 LST of March 20 - 4:45 LST of March 21, an
531 episode that covers the highest PM_{10} concentrations hours ($>1000\mu\text{g}\text{m}^{-3}$) in DS1; 4.
532 6:00 - 18:00 LST of March 22: an episode in DS2.

533 As shown in Fig. 8a, the mean σ between 9:00 - 15:00LST, March 20 ranged from
534 around 0.12 to 0.72 km^{-1} from near ground level to 1km. During this period, the mean
535 δp was relatively low of $\sim 0.02 - 0.08$, suggesting spherical particles, i.e. pollution
536 particles dominated. Accordingly, the contribution of dust to the total particle
537 extinction (dust ratio) was less than 20% (Fig. 8e). As discussed earlier, the high
538 values of light extinction caused by pollution particles before the onset of DS1 were
539 due to the pre-accumulation of local and transported pollutants brought by the invaded
540 cold front. On the pathway of DS1 (Fig. 2c), the air masses travelled over large areas
541 with intense anthropogenic emissions and thus a large quantity of pollutants could be
542 pushed to the downstream areas and accumulated before the dust plume arrived.

543 As DS1 invaded, the mean σ further increased and reached 0.73 km^{-1} near the
544 surface level (Fig. 8b-c). In the meantime, the values of δp were elevated to be higher
545 than 10% from near the ground to $\sim 1\text{km}$ (Fig.8f-g). When PM_{10} reached its highest
546 concentrations ($>1000\mu\text{gm}^{-3}$), the mean δp was $\sim 23\text{-}27\%$ extending from near the
547 ground to $\sim 700\text{m}$ (Fig.8g). However, compared to the δp value of $\sim 30\text{-}35\%$ for the
548 relatively pure Asian dust (Murayama et al., 2003; Sakai et al., 2003), the δp in DS1
549 was still lower, suggesting the mixing of dust with pollution at a certain extent. As
550 shown in Fig. 8b-c, the mean σ_d obviously overwhelmed the mean σ_s and could be up
551 to 0.82 km^{-1} near the surface level, while the mean σ_s dropped to less than 0.1 km^{-1} .
552 From near the ground to $\sim 700\text{m}$, no significant vertical gradients of δp and dust ratio
553 were observed. This indicated DS1 had a “top-down” effect on modifying the bulk
554 particle optical properties in the lower troposphere in this case. Within this altitude
555 range, dust could account for $\sim 80 - 90\%$ of the total particle extinction coefficients
556 while only a minor fraction of $\sim 10 - 20\%$ was attributed to the pollution particles.
557 Above 700m, δp and the dust ratio quickly decreased, suggesting the impact of dust on
558 the light extinction was much weakened.

559 In DS2, the maximum σ and δp averaged between 6:00 and 18:00 LST, March 22
560 was $\sim 0.61\text{km}^{-1}$ and 26%, respectively, observed at an altitude of $\sim 300\text{m}$. Different

561 from DS1, both δp and the dust ratio had a significant vertical gradient in DS2. As
562 showed in Fig. 8h, δp decreased quickly from its maximum value at $\sim 250\text{m}$ to 5% at
563 $\sim 750\text{m}$. Correspondingly, the dust ratios decreased from 88% to 25%. Moreover, the
564 vertical profiles of σ_d and σ_s showed distinctly different behaviors during this period.
565 As shown in Fig. 8d, the maximum σ_d ($\sim 0.55 \text{ km}^{-1}$) showed at the altitude of $\sim 250\text{m}$,
566 while that of σ_s ($\sim 0.25 \text{ km}^{-1}$) appeared at a higher altitude of $\sim 800\text{m}$. As compared to
567 DS1 (Fig. 8b-c), the σ_s in DS2 showed similar magnitudes below $\sim 270\text{m}$. Above this
568 altitude, the σ_s in DS2 gradually increased while that in DS1 varied relatively stable
569 with the altitudes. The enhancement of pollution particles in the middle layer during
570 DS2 should be mainly attributed to its unique transport pathway. As shown in Fig. 2e,
571 substantial air masses in the upper layer transported from the south and may bring
572 more moisture. The sounding data at a meteorology station (31.40°N , 121.46°E) in
573 Shanghai supported this statement as shown in Fig. S2.

574 Opposite to the low relative humidity (RH) and its decreasing trend with altitude in
575 DS1, RH in DS2 (measured at 8:00 and 20:00 LST of March 22, respectively) showed
576 much higher values and an increasing trend with altitude (Fig. S2). This phenomenon
577 corroborated our discussions above that the meteorological conditions were more
578 favorable for promoting the dust chemistry in DS2. As a result, the secondary aerosol
579 formation via heterogeneous reaction yielded stronger particle extinction in DS2,
580 especially in the middle and upper layers through 270m till the top (2km) where the
581 particle extinctions caused by spherical particles were still significant. On the other
582 hand, due to the higher humidity in DS2, the soluble particle components should
583 undergo stronger hygroscopic growth and thus partly explain the structure of vertical
584 profile of spherical particles as shown in Fig. 8d. It is commonly regarded that in a
585 dry and less oxidative environment that dust storms are usually associated with, the
586 formation and growth of secondary particles are often depressed. However, under
587 certain favorable conditions, new particle formation during dust events could be still
588 discernible (Nie et al., 2014). In this study, vertical profiles of crucial meteorological

589 parameters, pollutant precursors and particle numbers were not available for
590 diagnosing the new particle formation. Tethered balloon-based measurement (Li et al.,
591 2015) could be a good platform for investigating the particle formation during dust
592 events at different altitudes in the future research.

593

594 **3.6 Impact of dust chemistry on regional air quality**

595 Chemical transport modeling (Methods in Section 2.5) was utilized to assess the
596 impact of dust chemistry on the perturbation of air quality at the regional scale. The
597 model performance of CMAQ with improved dust module has been evaluated against
598 various observational datasets and it was demonstrated that the model has relatively
599 good capability in capturing both magnitudes and temporal variability of bulk aerosol
600 (e.g. PM₁₀, AOD) during the spring season over China (Dong et al., 2016). It has to be
601 noted that the model only simulated the particle size up to 10 μm while the
602 observation of particle chemistry included all the sizes (i.e. TSP), hence the mismatch
603 of size distribution between the model and observation precluded the evaluation of the
604 simulated particle chemical species in this study. In the following discussions, we
605 focused on the qualitative assessment of the impact of dust chemistry on the regional
606 air quality.

607 Fig. 9 shows the spatial distribution of simulated mineral particles from March 20 –
608 22, 2010, respectively. Accordingly, we show the spatial distribution of Ultraviolet
609 Aerosol Index (UVAI) retrieved from OMI during the same period. UVAI is sensitive
610 to absorbing particles, i.e. black carbon and mineral dust (Torres et al., 2007). Hence,
611 the comparison between simulated mineral particles and observed UVAI could
612 illustrate how the transport of dust was reproduced. However, based on the definition
613 of Aerosol Index, it is a parameter based on the difference between radiance at two
614 near ultraviolet wavelengths. Thus, the comparison between satellite Aerosol Index
615 and simulated mineral dust is only qualitative. As seen from Fig. 9, high values of
616 UVAI were always observed over the Indo-China Peninsula and this was ascribed to

617 the black carbon particles emitted from strong biomass burning over this region
618 during the spring season (Fu et al., 2012; Huang et al., 2013; Tsay et al., 2013). What
619 we concern are the dust and its downwind regions over East Asia. On March 20, high
620 UVAI values stretched from the Gobi Desert to the North China Plain and the Yangtze
621 River Delta. This transport pathway was relatively well reproduced by the model as
622 we can see high concentrations of mineral particles over similar areas as well. On the
623 following day of March 21, the satellite observation illustrated the movement of the
624 high UVAI zones further southward and the drifting of dust plumes off the coastline
625 of East China (Fig. 9c). Accordingly, the simulation showed similar behavior with
626 strengthened concentrations of mineral particles over the Gobi Desert and downwind
627 areas of the Yangtze River Delta (Fig. 9d). On March 22, although UVAI signals were
628 absent over most areas of East China due to the satellite swath, high UVAI values
629 could still be observed over the tip of the Yangtze River Delta and the East China Sea.
630 This is also relatively consistent with the model simulation that high concentrations of
631 mineral particles hovered over the coastlines from YRD to the Taiwan Strait. Overall,
632 we demonstrated that the model is capable of capturing the spatial distribution of dust
633 during the long-range transport.

634 Since we specifically focus on the interaction between dust and anthropogenic
635 pollutants in this study, we have performed two simulations, i.e. one with dust
636 emissions but without dust chemistry and the other one with dust chemistry. The
637 impact of dust chemistry on particle chemical components could thus be quantified
638 via the difference between these two simulations. Fig. 10a&c shows the spatial
639 distribution of sulfate via the formation pathway of dust heterogeneous reactions, as
640 well as for nitrate in Fig. 10b&d. Different from the spatial distribution of mineral
641 particles (Fig. 9a-c), the formation of sulfate and nitrate via dust chemistry mainly
642 occurred over Northeastern China. This is expected that although the major source
643 region of mineral dust is from the Gobi Desert in Northern China, less anthropogenic
644 emission sources existed there, hence relatively weak atmospheric chemical

645 processing was simulated over the dust source region. On the other hand, drier climate
646 in Northern China also suppressed the extent of heterogeneous reactions on the
647 surface of dust. Hence, the strongest impact from dust heterogeneous reactions on the
648 formation of secondary particles occurred in those populous areas where the
649 interaction of anthropogenic precursors and dust was the strongest.

650 As for sulfate, the simulated concentration over Shanghai averaged 8.1, 3.1, and 8.5
651 μgm^{-3} from March 20 – 22, respectively. This temporal variation corresponded to that
652 from observation as discussed in Section 3.4.2. While it has to be noted again that the
653 simulated particle species contained particulate sizes less than 10 μm , close match
654 between the simulation and observation was not expected. As we compare DS1 (Fig.
655 10a) and DS2 (Fig. 10c), the simulated sulfate during DS2 was evidently more intense
656 than that during DS1 at a larger geographic region, which was fairly consistent with
657 the observation. Since the daily emission rates digested by the model were almost
658 constant during this period, meteorological conditions should be the determining
659 factor, of which elevated humidity during DS2 has been diagnosed as the most
660 important factor responsible for stronger dust chemistry.

661 As for nitrate, its spatial distribution pattern was as similar as that of sulfate at a
662 certain extent. The simulated concentration of nitrate averaged 3.1, 2.3, and 5.2 μgm^{-3}
663 over Shanghai from March 20 – 22, respectively. Although the simulation results and
664 observational data couldn't be statistically inter-compared due to the size difference as
665 stated above, we think the simulated nitrate should be largely underestimated. The
666 ratio of simulated nitrate between DS2 and DS1 was 1.7, similar to that of ~ 1.6 from
667 observation. This suggested the favorable meteorological conditions (e.g. higher
668 humidity) on facilitating the dust heterogeneous reaction during DS2 have been
669 accounted by the model. In this regard, we ascribe the low- biased nitrate mainly to
670 several aspects. First, the NO_x emissions could have been underestimated. As
671 indicated in Dong et al. (2015) which used the same anthropogenic emission
672 inventory as this study, the model showed some underestimation of the total NO_2

673 columns as compared to the OMI observation, especially over Northeastern China
674 (See Fig. 8 in Dong et al.(2015)). Secondly, HNO₃, N₂O₅, NO₂ and NO₃ were the
675 major precursors of nitrate as implemented in the dust module. Of which, reactions
676 via HNO₃ and N₂O₅ were the dominant pathways of nitrate formation due to their
677 relatively high uptake coefficients on the dust. It was possible that the formation of
678 HNO₃ was underestimated, thus lowering the production of nitrate. However, this is
679 just a guess as observation of gaseous HNO₃ was not available in this study. Lastly,
680 we think the underestimation of nitrate may be due to the omission of nitrate
681 processing on the surface of sea salt, which was especially important for costal cities
682 such as Shanghai (Buseck and Posfai, 1999).

683

684

685 **4. Conclusions**

686 During March 19-23, 2010, a super dust storm swept extensive areas over China
687 from Northern China to Southern China. Two separate dust periods were observed in
688 Shanghai, one from ~16:00 LST, March 20 to ~10:00 LST March 21 (DS1) and the
689 other from ~6:00 LST, March 22 to ~0:00 LST, March 23 (DS2). In DS1, the dust
690 plumes mostly transported over those areas characterized of high pollution emissions
691 in Northern China, while the dust plumes in DS2 transported over the coastal regions
692 of Eastern China with higher RH. The ratios of SO₂/PM₁₀ and NO₂/PM₁₀ were up to
693 0.11 ± 0.03 and 0.20 ± 0.04 in DS2, much higher than the values in DS1. In addition,
694 the concentrations of typical elemental species such as As, Cd, Pb, and Zn were also
695 more enriched in DS2 than in DS1, indicating the stronger mixing extent between dust
696 and primary anthropogenic emissions in DS2. Due to the higher level of gaseous
697 pollutant precursors associated with moderate relative humidity, SO₄²⁻ and NO₃⁻
698 exhibited higher concentrations in DS2 than in DS1. The higher Ca²⁺/Ca ratio and
699 lower [NH₄⁺]/[SO₄²⁻+NO₃⁻] ratio in DS2 suggested the heterogeneous reactions
700 between calcites and acid gases were significantly promoted in DS2. Particularly,
701 more NO₃⁻ than SO₄²⁻ was formed in DS2, probably due to the high concentrations of

702 NO₂ and the complex reaction among dust, seasalts, and nitric acid. Vertical profiles
703 of particle optical properties from Lidar measurement retrieved high depolarization
704 ratios of 20-25% extending from near the ground to ~700m in DS1 and it was
705 estimated that dust dominantly accounted for ~80-90% of the total particle extinction.
706 In DS2, the vertical structure of particles changed dramatically from that of DS1. The
707 maximum extinction coefficient of dust particles emerged at the altitude of ~250m in
708 DS2, while that of pollution particles showed at a higher altitude of ~800m. The
709 abnormal increase of humidity as a function of altitude from the southerly winds
710 explained this phenomenon due to promoted heterogeneous reactions on dust particles
711 and the subsequent particle hygroscopic growth.

712 Simulated mineral particles were compared to the remote sensing UVAI from
713 satellite, showing consistent spatial patterns between model and observation. By
714 applying the dust scheme with explicit chemistry mechanisms, evident impacts of dust
715 heterogeneous reactions on secondary aerosol formation were reproduced over
716 widespread areas of Northeastern China where the anthropogenic emissions were
717 intensive. The sulfate formation was relatively well simulated while the nitrate
718 formation on dust was believed to be largely underestimated. More research on
719 laboratory kinetic studies of nitrate dust chemistry is suggested. Also, the feedback
720 between dust chemistry and regional climate change needs to be investigated in the
721 future.

722

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731

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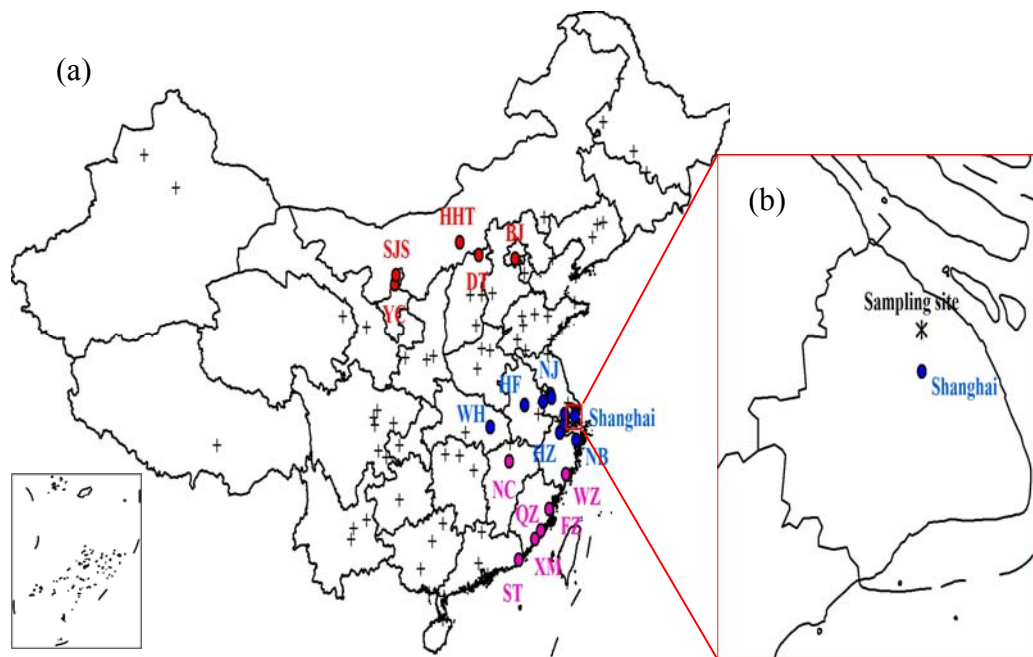
957 Table 1 Concentrations of the measured elements ($\mu\text{g}/\text{m}^3$) in TSP before, during, and
 958 after the dust periods from March 19 to 22 and in non-dust days (NDS) in Shanghai.
 959 N, D, and NDS represent the nighttime samples collected from 20:00 to 8:00 in the
 960 next day, the daytime samples collected from 8:00 to 20:00, and the average of the
 961 non-dust samples collected from March 25 to 27, respectively.

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	DS1					DS2		NDS
	19 N	20 D	20 N	21 D	21 N	22 D	22 N	
TSP	168.5	605.0	1306.7	155.1	194.3	683.3	348.2	160.5
Al	6.2	36.4	67.5	9.6	9.3	35.8	12.8	5.2
Ca	8.5	37.3	56.9	9.4	8.6	33.2	12.0	8.0
Fe	3.2	23.3	42.1	6.8	6.0	23.7	11.6	4.2
Mg	1.4	9.5	14.4	2.2	2.3	8.1	3.4	1.4
Na	1.3	7.9	19.1	2.8	3.9	7.9	3.5	2.4
Mn	0.2	1.0	2.2	0.4	0.3	1.1	0.6	0.2
Ti	0.2	1.9	4.3	0.6	0.5	1.9	0.7	0.3
Sr	4.3×10^{-2}	1.6×10^{-1}	3.3×10^{-1}	4.8×10^{-2}	4.3×10^{-2}	1.9×10^{-1}	6.4×10^{-2}	3.8×10^{-2}
V	1.5×10^{-2}	5.2×10^{-2}	1.0×10^{-1}	1.7×10^{-2}	1.5×10^{-2}	4.7×10^{-2}	2.7×10^{-2}	1.3×10^{-2}
Ni	6.7×10^{-3}	2.5×10^{-2}	4.6×10^{-2}	1.1×10^{-2}	3.0×10^{-3}	2.8×10^{-2}	1.6×10^{-2}	6.5×10^{-3}
Zn	7.5×10^{-1}	6.5×10^{-1}	3.4×10^{-1}	1.9×10^{-1}	2.3×10^{-1}	6.3×10^{-1}	5.7×10^{-1}	4.0×10^{-1}
Pb	1.3×10^{-1}	1.5×10^{-1}	1.1×10^{-1}	4.0×10^{-2}	3.7×10^{-2}	2.0×10^{-1}	1.2×10^{-1}	8.6×10^{-2}
Cu	1.3×10^{-2}	6.3×10^{-2}	7.2×10^{-2}	4.0×10^{-2}	2.4×10^{-2}	6.1×10^{-2}	4.7×10^{-2}	3.7×10^{-2}
As	1.2×10^{-2}	1.3×10^{-2}	2.6×10^{-2}	9.2×10^{-3}	1.3×10^{-2}	2.5×10^{-2}	2.4×10^{-2}	1.1×10^{-2}
Cd	2.8×10^{-3}	3.1×10^{-3}	3.4×10^{-3}	7.4×10^{-4}	1.2×10^{-3}	4.7×10^{-3}	2.7×10^{-3}	1.3×10^{-3}

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967 Fig.1 (a) Locations of the 86 major cities over China. The cities with PM₁₀

968 concentrations up to 600 $\mu\text{g}\text{m}^{-3}$ on March 20, 21, and 22 were marked by red, blue,

969 and pink circles, respectively. (b) The site location of the ground-based measurement

970 in Shanghai.

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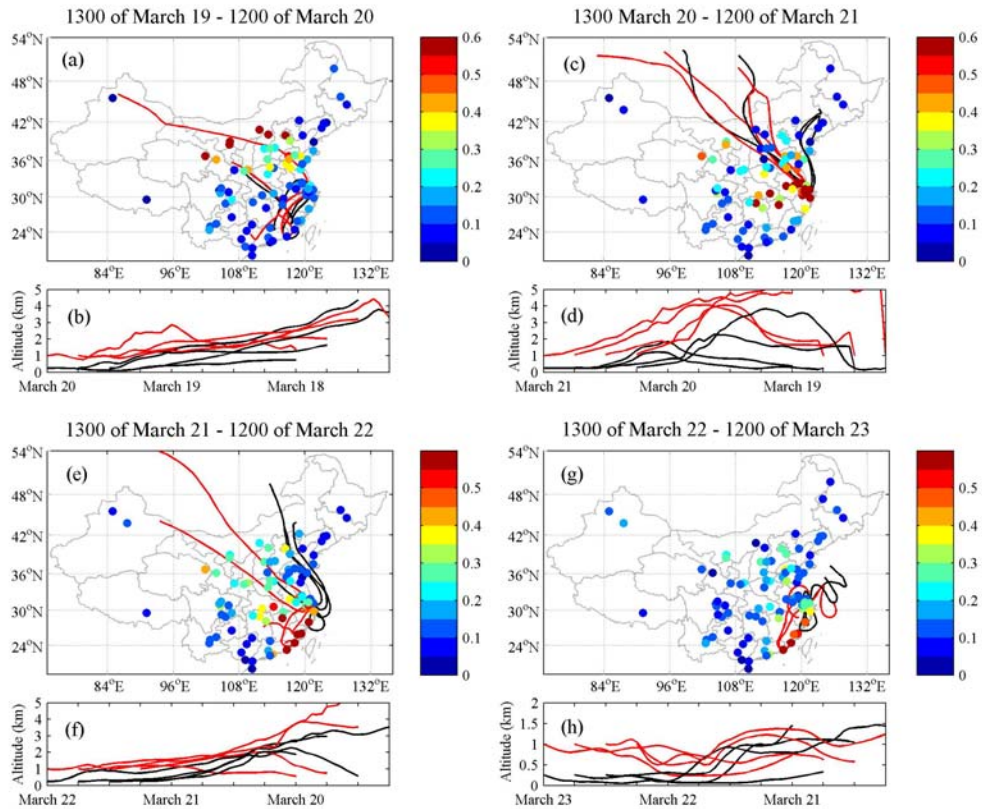
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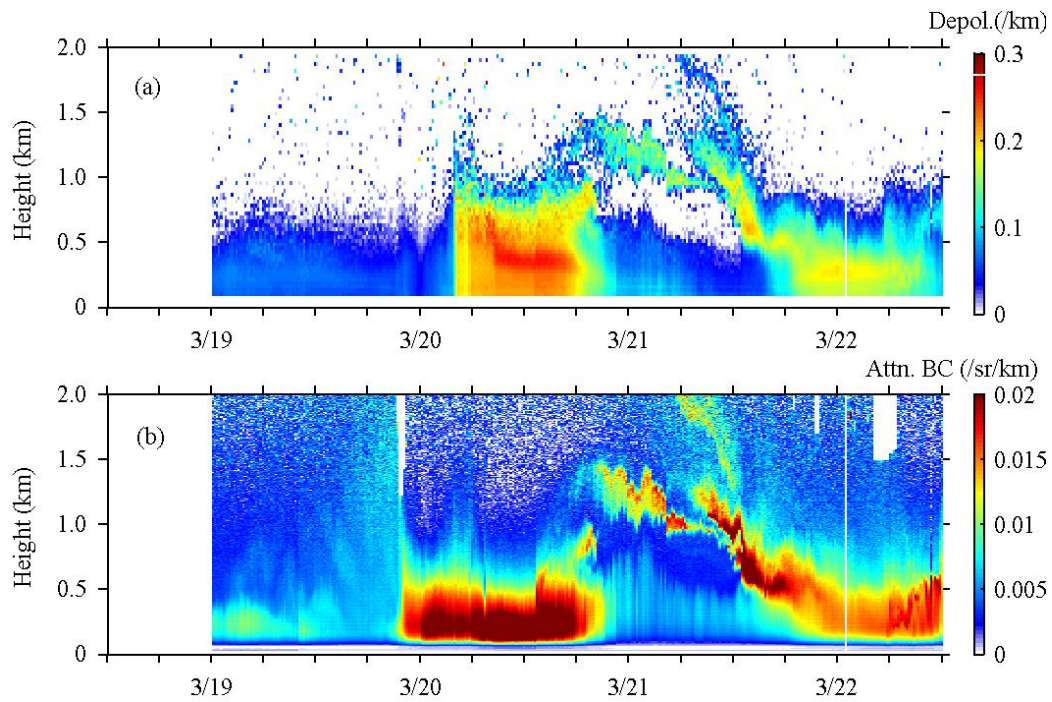
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992 Fig.2 Daily PM_{10} concentrations ($mg\ m^{-3}$) in 86 major cities over China and 48 hours
 993 backward trajectories of the air masses at both 250m (black lines) and 1000m (red
 994 lines) above ground level over Shanghai during March 19-23, 2010. Four trajectories
 995 ending at 00, 06, 12, and 18 LST are computed for each day.

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1016 Fig.3 Time-height cross-section of (a) volume depolarization ratios and (b) attenuated
 1017 aerosol backscattering coefficients measured by a Lidar at the wavelength of 532 nm
 1018 during March 19-22, 2010 (Local Standard Time).

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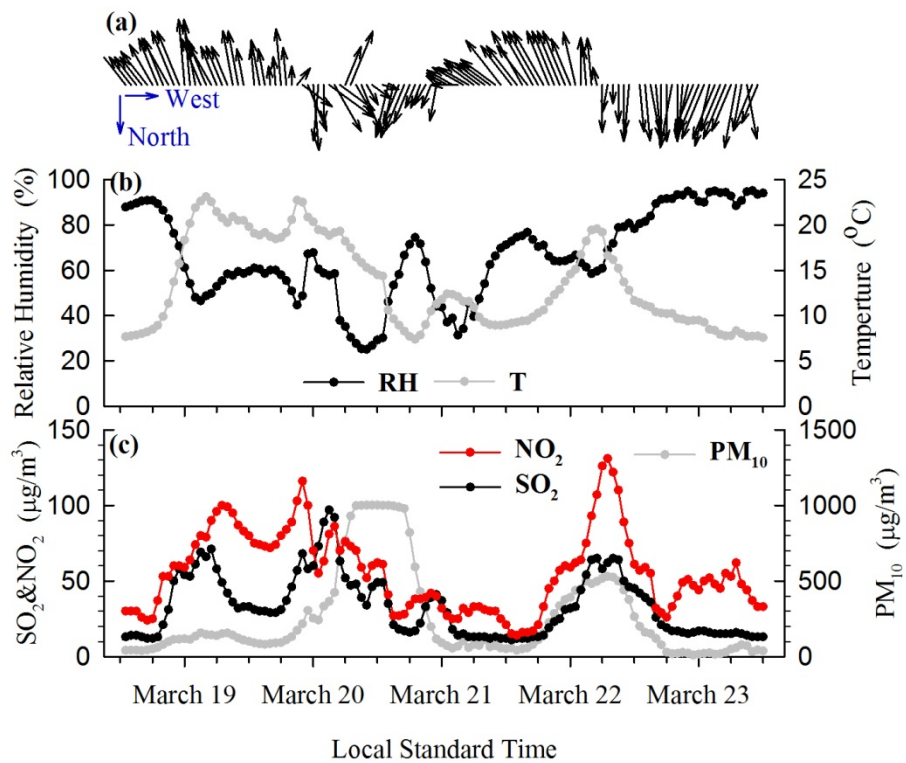
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1037 Fig.4 Time-series of PM₁₀, SO₂, NO₂, and meteorological factors including wind
 1038 conditions, relative humidity (RH), and ambient temperature (T) in Shanghai during
 1039 March 19-23, 2010.

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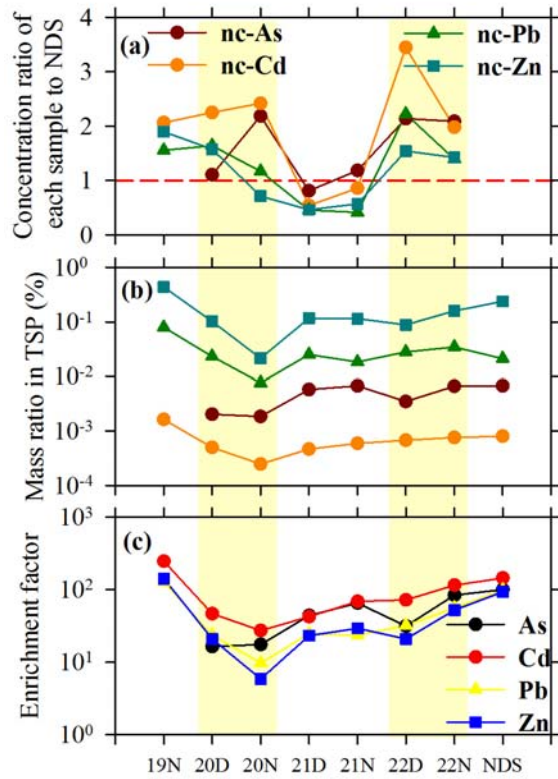
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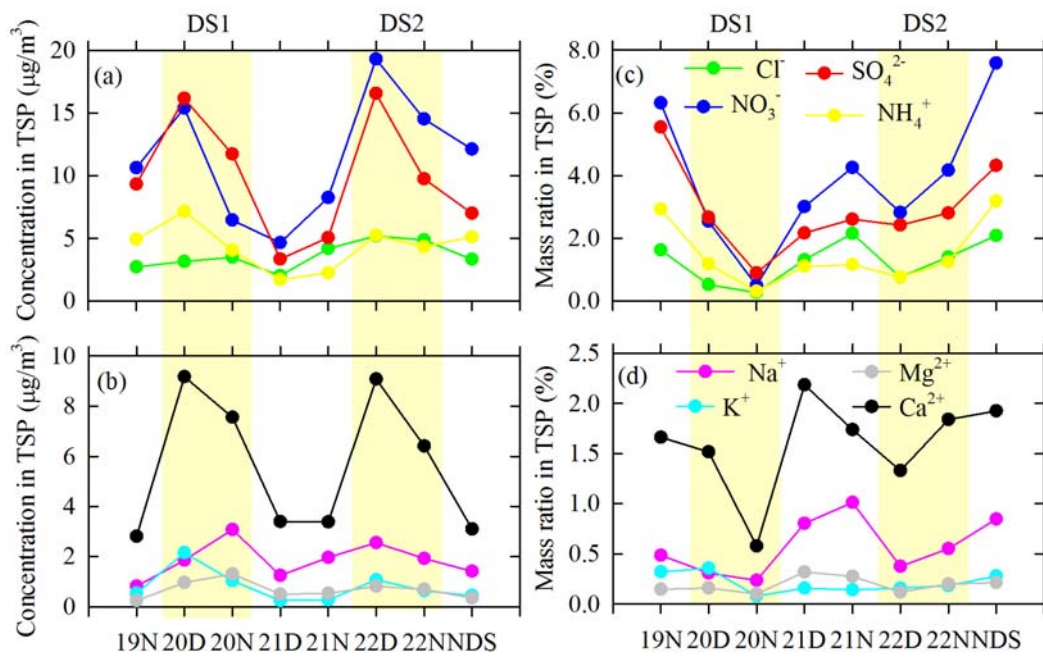
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Fig.5 Variations of (a) the mass concentration ratio of each sample to the NDS sample for nc-As, nc-Cd, nc-Pb, and nc-Zn, (b) the mass ratios, and (c) enrichment factors of pollution elements As, Cd, Pb, and Zn in TSP during March 19-22 and non-dust days (NDS). D, N, and NDS represent the daytime samples collected from 8:00 to 20:00 LST, the nighttime samples collected from 20:00 to 8:00 LST in the next day, and the average of the non-dust samples collected from March 25 to 27, respectively.

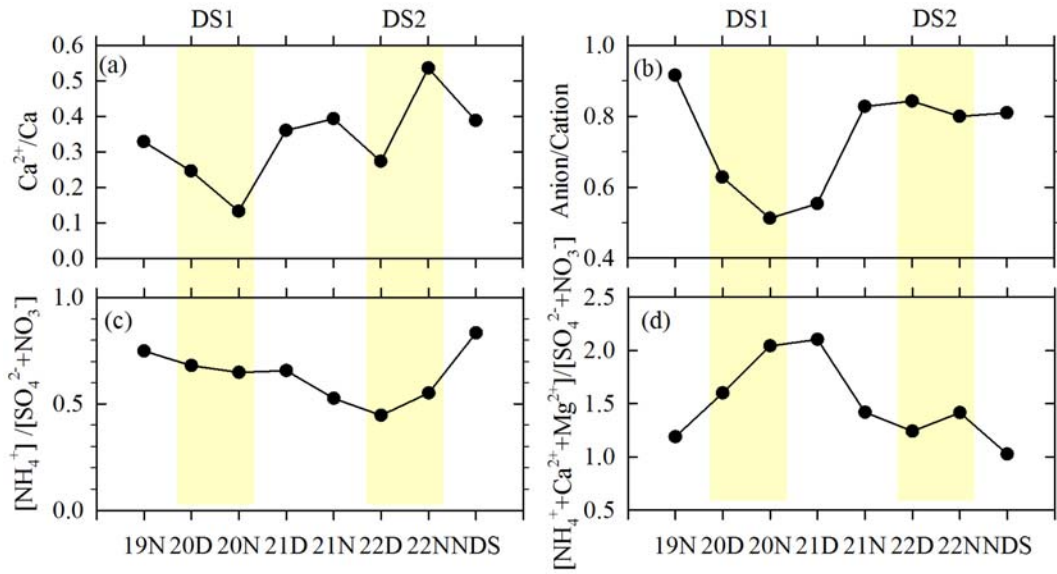
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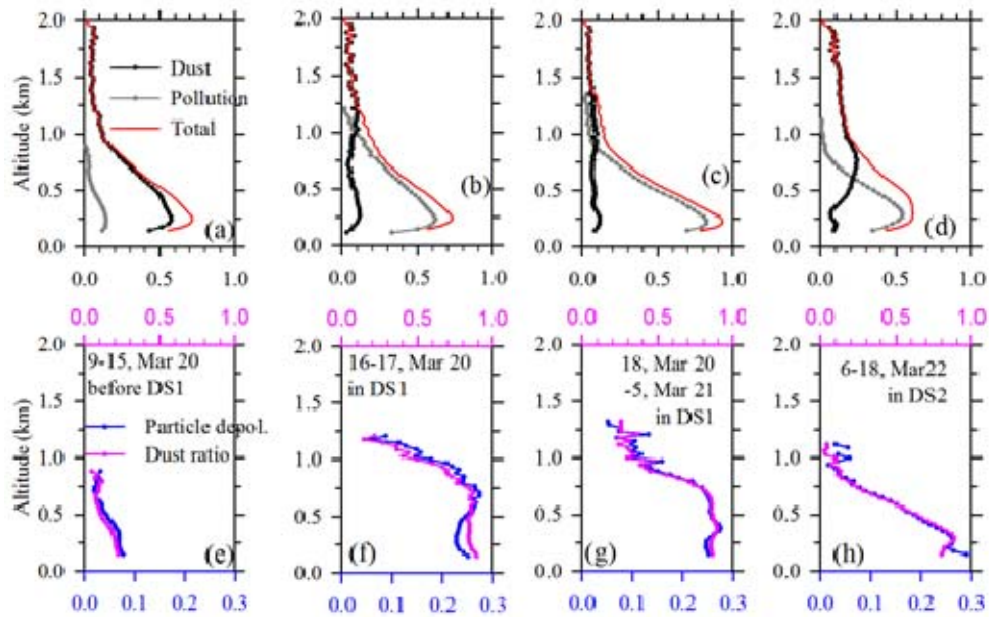
Fig.6 Variations of the concentrations of water-soluble ions and their mass ratios in TSP during March 19-22 and NDS, 2010.

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Fig.7 Variations of (a) the ratios of Ca^{2+}/Ca , (b) the equivalent concentrations of the total anions to the total cations (Anion/Cation), (c) the equivalent concentrations of NH_4^+ to the sum of SO_4^{2-} and NO_3^- ($[\text{NH}_4^+]/[\text{SO}_4^{2-}+\text{NO}_3^-]$), and (d) the equivalent concentrations of the sum of NH_4^+ , Ca^{2+} , and Mg^{2+} to the sum of SO_4^{2-} and NO_3^- ($[\text{NH}_4^++\text{Ca}^{2+}+\text{Mg}^{2+}]/[\text{SO}_4^{2-}+\text{NO}_3^-]$) in TSP during March 19-22 and NDS, 2010.



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1136 Fig.8 Vertical profiles of the average extinction coefficients of dust, pollution, and
 1137 total particles (km⁻¹), particle depolarization ratios (Particle depol., unitless), and the
 1138 ratio of the dust extinction in the total extinction (Dust ratio, unitless) in four periods
 1139 of 9:00 - 15:00 of March 20 (before DS1), 16:00 - 17:45 of March 20 (before the
 1140 highest PM₁₀ concentration in DS1), 18:00 of March 20 - 4:45 of March 21 (during
 1141 the highest PM₁₀ concentration in DS1), and 6:00 - 18:00 of March 22 in DS2.

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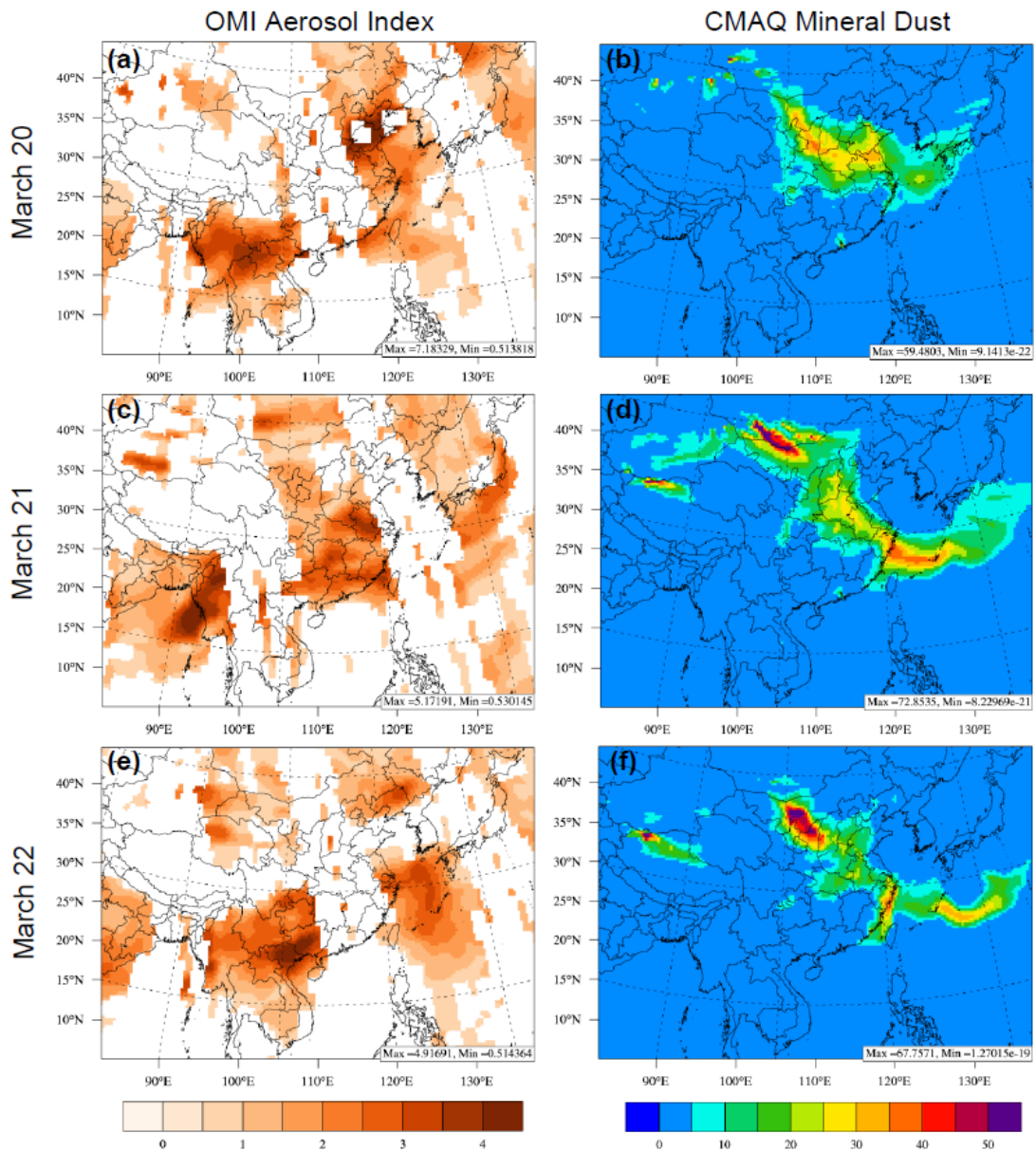


Fig. 9. The spatial distribution of OMI Ultraviolet Aerosol Index (unitless) and simulated mineral aerosols (μgm^{-3}) from March 20 – 22, 2010, respectively.

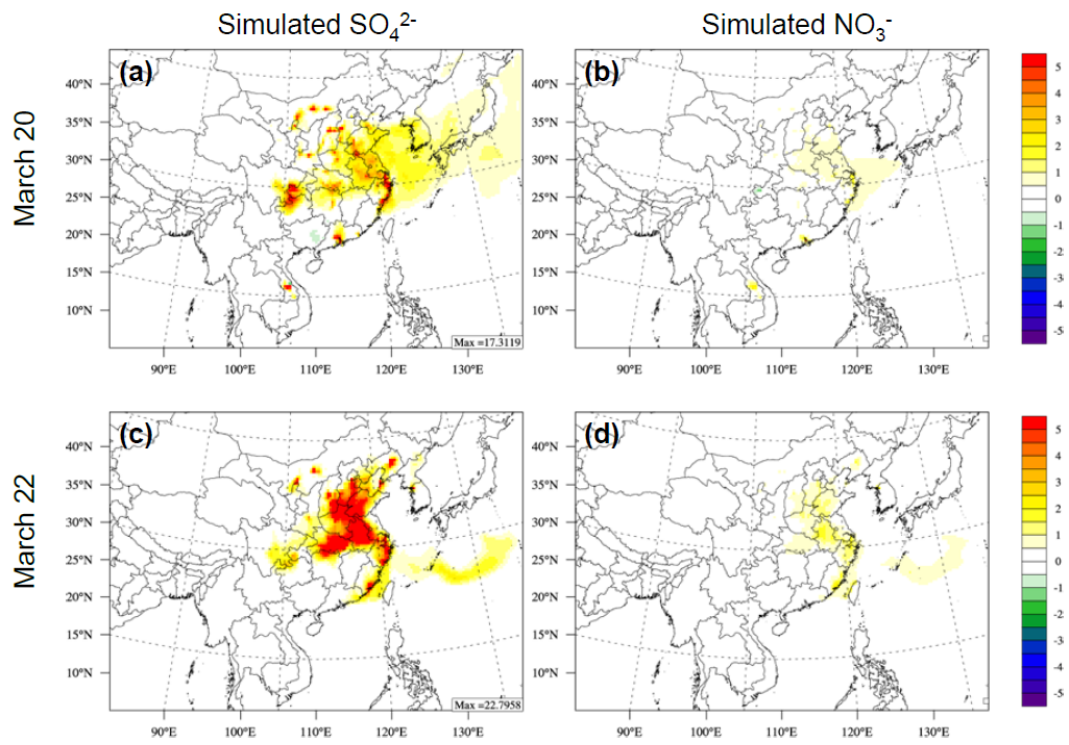


Fig. 10. Simulated SO_4^{2-} and NO_3^- ($\mu\text{g m}^{-3}$) from dust heterogeneous reactions during DS1 and DS2.