

Jonathan Williams
Editor of Atmospheric Chemistry and Physics
Max Planck Institute for Chemistry, Atmospheric Chemistry

Dear Dr. Williams,

Please find below our detailed responses to the three reviews of manuscript acp-2017-328. The reviewer's comments are in bold type and our responses are in normal text. The marked-up manuscript is at the end. We thank the reviewers for their careful reading of our manuscript and for their comments. It appears that reviewer 2 had significant reservations about our work and the fitting analysis that we employed; therefore, we have added a section to the supplementary material to describe the analysis process in more detail, and to show example spectra and fits. We trust that we have satisfactorily addressed all concerns at this point.

Sincerely,

Ralf Staebler
Research Scientist
Environment and Climate Change Canada

Comments from Referee #1

1. Line 113-line 116: this needs slight elaboration – the stray light problem should be very minor and only from internal reflections (since the light is modulated before being sent across the open-path. This means ambient scattered radiation will not be modulated and thus not detected by the instrument).

Yes, the stray light influence on concentration retrieval is generally rather minor, but since it is possible to correct for this, we did. These sentences in the main text have been revised as follows: "This stray light spectrum accounts for radiation back to the detector from reflections by internal parts inside the spectrometer, i.e. not from the retroreflector array, and was subtracted from all the measurement spectra before performing further analysis. Stray light affected final mixing ratios by < 3 % in this study."

2. The MALT program (which you say that your analysis is based on) can calculate the reference spectra directly from the HITRAN database using the ambient temperature measured at the same time as the spectra were recorded. Does the Bruker software really not allow this? This may only introduce uncertainties of 10% but it is an unnecessary added uncertainty, since you have accurate temperature measurements available.

This was a mistake in the manuscript; our analysis is based strictly on Bruker software, not MALT. The Bruker software, OPUS_RS, uses a non-linear fitting method. Reference spectra were fitted to the

measured spectra using a model to calculate the instrumental line shape (ILS) (Harig et al., 2005) followed by a non-linear curve fitting method to retrieve concentrations of pollutants.

In the OPUS_RS software, there is an option of setting up “temperature dependent reference files”. These files include either PNNL (5 °C, 25 °C, 50 °C) or HITRAN files at specific temperatures. Then the program takes the current temperature from a sensor (or temperature data file), and interpolates the high-resolution reference spectrum to the current temperature from those temperature-specific reference spectra which were included in “temperature dependent reference files”.

Harig, R., Rusch, P., Schäfer, K., Flores-Jardines, E.: "Method for on-site determination of the instrument line shape of mobile remote sensing Fourier transform spectrometers", SPIE 5979, 432-441, 2005.

3. There are other uncertainties inherent to open-path FTIR measurements (like those that come from the HITRAN database and fitting errors). These are not mentioned in the text but should at least be referred to as existing even if a full uncertainty analysis is not given.

Yes, some uncertainties are associated with spectrum fitting errors; the thresholds of the correlation coefficients used in fitting analysis in each pollutant in Table 1 give an indication of this. There are also uncertainties dependent on environmental conditions. In this study, we have considered ambient temperature and pressure, as well as H₂O vapor as an interfering gas, in our retrieval analysis. We have stated that for pollutants such as NO and NO₂, water vapor interfered so much that we were not able to get good mixing ratio retrievals. The signal intensity differences related to the distance between spectrometer and retroreflector play a minor role on the detection limits of pollutants studied, since the distance in this study was near the optimal range for this instrument. These detection limits were updated in Table 1 and the text has been revised as follows: “Besides fitting errors and the effect of ambient temperature on the reference spectrum, other environmental conditions may also contribute to uncertainties, such as interference from ambient water vapor.”

4. Section 2.2 – at what time resolution are these calculations made?

As mentioned in the main text, the LEDs were operated in the continuous mode. H, u* and were calculated at a 1-minute resolution in this study.

To clarify, the main text has been revised: “Sensible heat flux (H), friction velocity (u*) and Obukhov length (L) were calculated from scintillometer measurements at a 1-minute resolution in this study.” In addition, to make the manuscript more concise, we decided to move the theory of scintillometer to the Supplementary Material Section 2.

5. Line 212 – Isn’t an estimate using WindTrax and CO mole fractions a “top-down” estimate? You then compare it to one based on traffic volumes – isn’t this one “bottom-up”?

Yes, that was a typo and has been corrected. WindTrax and CO mixing ratio is a “top-down” estimate. A traffic-volume-based estimate is ‘bottom-up’.

6. Line 247: do they generally agree? The level of agreement is not quantified. From looking at the time-series the variations certainly seem to be well captured, but it would be good to give a correlation coefficient for this.

We have done this analysis, and now we include Fig. S2a, CO_NAPS vs. CO_FTIR, in the supplemental material. The r^2 is 0.67 when the wind came across the highway to the NAPS trailer, and 0.57 when the wind came from other directions.

7. In fact some basic statistics for the model's skill level would improve the manuscript. The authors have used the "open-air" package and this has some great tools for quick evaluation of a model's performance against observations.

This is a good suggestion. We have added mean bias and root mean square error of comparison between CO measurement and CO model in the supplemental material, Fig. S2b and c.

8. Reading this discussion about model to measurement spatial differences begs the question as to why a comparison of model to open-path FTIR is not shown. This will suffer similar problems but should be much less than the in-situ observations.

In the supplementary material, we have included Fig. S2c on CO_model vs. CO_FTIR with statistical results, and Fig. S4 polar plot of (CO_model – CO_FTIR) vs. wind direction. In the comparison of CO_model and CO_NAPS, the slope when the wind came from the highway and the slope when the wind came from other direction was very different (0.72 vs. 1.20), indicating the strong spatial differences. In the comparison of CO_model and CO_FTIR, the slopes are much closer (1.20 vs. 1.09), indicating the sampling spatial difference is smaller when comparing path-integrated mixing ratio with the volume-averaged chemistry transport model. From the polar plot of CO_model – CO_FTIR vs. wind direction and speed, it also can be seen that positive differences (yellow) occur mainly when the wind is from other directions, and the dependence of (CO_model – CO_FTIR) on the wind direction is not that strong. This discussion is also included as a new paragraph in the main text, Section 3.2.1.

9. Line 297. Are the traffic volumes similar on the weekends? This is surprising and you have not actually stated that clearly before. Can you clarify?

Yes, traffic volumes on the weekend were similar to weekdays except for the morning period (from around 6:00 to 11:00), as shown as the black dashed line in Fig. 5 (bottom). We also wrote "The median CO mixing ratio on weekends was close to that on weekdays, except for the early morning period." And to make it clearer to readers, we have included the actual traffic volume number in this sentence: "On weekends, traffic volume increased more gradually during the morning until plateauing around 11:30 and on average remained high with about 21800 vehicles h⁻¹ until after 22:00."

10. Line 323: is it worth showing the correlation plot at least in the supplementary data?

Agreed. A supplemental figure of linear regression of NH₃ and CO mixing ratio from FTIR has been added to the supplemental material (Fig. S5).

11. Line 332: Are you also assuming that traffic is the only source of CO emissions above background?

Yes, we assumed that highway traffic emission at this spatial scale is the only source of CO above background, to estimate traffic-related NH₃ emission. We observed a good linear relationship between mixing ratio of NH₃ and CO, shown in Fig. S5.

12. Line 438: Why only 3 days? I assume that these are the best steady wind conditions? Whatever the reason for the choice of these days, it should be stated briefly in the text.

We only picked three days because the dispersion model is not easily automated and requires lots of manual labour. We have revised that paragraph and added the following sentences to the text: “July 22 was chosen because the wind direction was steadily from northwest, and a traffic jam occurred for added interest. July 28 and 29 were chosen because they are two of the highest days for temperature and O₃ during this project.”

13. Line 529: “reasonable” correlations observed. You need to provide some actual statistics to back this up somewhere in the manuscript.

We meant to point out that wind direction affects the comparison, so we revised this sentence as follows: “Model results and measurement results are not expected to be directly comparable for all wind regimes, and comparisons can be better explained after separating wind directions.”

14. Consider changing “mixing ratio” to “mole fraction” throughout, as I believe this is now the preferred terminology.

Thank you. We have seen both terms. This is noted in line 53 in the revised manuscript.

15. Line110: the “fraction” of the path is not actually a fraction but a distance, consider rephrasing.

This was rephrased into “the length of the path that was directly over...”

Comments from referee # 2:

1. The greatest shortcoming of the paper, however, is that there appear to be serious flaws in the data or data analysis, particularly the FTIR spectral analysis. Indeed, it appears that the FTIR data were not analysed at all, but simply taken “as is” from the system, and the IR data appear to have large systematic errors.

When writing this manuscript, we made the decision not to dwell on the minutiae of the FTIR analysis but to focus on the results. However, if more detail is required, we can certainly provide this. Quite a significant amount of work was done on optimizing the fittings and quantifying concentrations of pollutants. We have now included a section in the supplemental material to describe the FTIR spectral analysis by the OPUS-RS software in more detail. Examples of a measured spectrum and model fit spectra along with residuals for CO, O₃, and NH₃ are also shown in Fig. S1. To summarize briefly, the software’s analysis algorithm is based on an iterative nonlinear fit of the measured IR spectra in compound-specified spectral windows. High-resolution spectra of the target and interfering gases, as well as baseline functions are fitted to those spectral windows in the measured spectra, making use of the known instrument response, the instrument line shape to meet the spectral resolution of the measurement, i.e., 0.5 cm⁻¹ in this study. More detailed descriptions and spectrum examples are now given in the supplemental material.

2. With regards to novelty of the study, the crux of the paper is to collect FTIR measurements over an urban highway. Such works have been previously reported – see for example work by Bishop et al., Stremme et al., Colman et al., Grutter et al. and Chaney et al, as well as by other analytical methods (for example, the seminal paper by Stedman et al., the tunnel study by Popa et al and the newly released paper by Haugen et al.). That is to say, there exists significant literature regarding traffic emissions, and the present authors need to cite more of these studies, and also need to cite other studies that use open-path FTIR to measure similar compounds from other sources

We believe our study contains several novel aspects: first, the measurement site is the busiest segment of the busiest highway in North America, as mentioned in the Introduction. Secondly, we conducted measurements for 16 days continuously. Thirdly, what we measured was pollution in the open ambient air over a highway from a real-world, un-manipulated mix of vehicle types. Our study is different from previous tunnel studies, because the air mass is more confined in the tunnel. Mixing processes and solar radiation conditions in the tunnel are different than in the open environment. We also observed and quantified additional pollutants, such as ammonia, formaldehyde, methanol and hydrogen cyanide, for which not much literature exists. These species are not well understood, and their levels can be different at different regions and countries due to different vehicle emission control regulations, fuels, and technologies. Another unique point of this study is that we combined detailed measurement of turbulent mixing (not only wind speed and direction) at the site simultaneous with pollution measurements from the FTIR, to show the role of turbulent mixing on the local air quality. We thank the reviewer for suggesting additional relevant background papers, and they have been cited in the revised manuscript.

3. To add a more unique aspect to their study, we suggest that the authors i) emphasize more the aspect of monitoring near such a large motorway (e.g. perhaps there are atmospheric reactions / products only seen due to higher order rate constants requiring very high CO or O3 levels?), and ii) present a more detailed analysis of the instrumental comparison, i.e. the “shoot-out” between the measurements collected from the FTIR and from the NAPS.

We believe there are many unique aspects to this study. As we have stated in the manuscript and explained in point 2 above, one unique aspect of this study is that we combined direct measurements of mixing ratios of gas-phase pollutants from such a busy segment of highway with detailed micrometeorological and turbulence measurement to show that not only traffic emission rates, but also turbulent mixing conditions in the surface layer play an important role on the accumulation and build-up of air pollutants over this highway. We also observed several interesting pollutants, such as ammonia, formaldehyde, and hydrogen cyanide, which have important implications to atmospheric chemistry and population health, and for which in-situ measurements over a busy highway have not been commonly reported. Another unique aspect is our evaluation of a “top-down” approach to calculate the emission rates of a few primary pollutants with a dispersion model.

The “shoot-out” aspect will be discussed below points 4, 5, 6, 7, and 8.

4. We believe (at least some of) the data as reported are not correct. In particular, looking at the CO plot in Figure 3, the trends for the two instruments follow one another in a nearly identical manner with overall very similar diurnal profiles. If the NAPS were as sensitive to wind direction as the authors purport, then its corresponding diurnal profile would not manifest the same diurnal trend as the FTIR: The NAPS profile would better reflect the wind direction dependence, yet the NAPS values never trend down as the FTIR values goes up (or vice versa). That is to say, at no time is there an obvious anti-correlation seen in the data. Furthermore, CO is a very well mixed gas and the NAPS instrument (Figure 1) is physically located in the middle of the FTIR optical path. Since the path for the FTIR includes both the highway and a stretch of land greater in length than the highway, the averaged CO mixing ratio obtain by FTIR over the entire path should be similar to the NAPS value, but possible lower due to mixing in cleaner adjacent air. Inspection of Figure 3 directly points to this – while the magnitudes and offsets differ (significantly!!), both the NAPS and FTIR data maxima and minima track each other very well, having the same diurnal patterns. However, the fact that the CO mixing ratio values (in terms of offset and amplitude) do not agree at a quantitative value suggest that either the data or data analysis is likely incorrect and it is in this regard we criticise the paper.

The long-path FTIR provides a path-integrated concentration, whereas the NAPS trailer measured concentration at a point located at the south edge of the highway. Therefore, concentration measured by the NAPS trailer was more dependent on the wind direction than concentration obtained from the FTIR. But we must keep in mind that even a point measurement has a footprint. The footprint covers more highway when the wind comes from the north, and covers much less highway (but more parking lots, buildings, a park and golf club), when the wind was from the south. The polar plot of the (CO_FTIR – CO_NAPS) in Fig. 4a already clearly shows the dependence of the difference between the two measurements on the wind direction. We have stated that when the wind direction was from highway to NAPS trailer, the difference was smaller, and when the wind was from other directions, the difference

was greater. When the wind was from the south, the NAPS trailer measured less footprint covering the highway, but still measured some traffic pollutants. Therefore, the NAPS measurements of CO (primary pollutant) still demonstrated a similar diurnal cycle on weekdays as the FTIR measurements. There is no reason to think that CO_NAPS and CO_FTIR should be exactly anti-correlated, since both instruments sample overlapping plumes.

The FTIR measures a path-integrated concentration, and its footprint was not completely independent of wind direction. The dependence is weaker compared to the NAPS trailer. As the reviewer has noticed, the path of our set up included “a stretch of land”, and this stretch of land was downwind when the wind came from north. Therefore, the downwind length of the FTIR path was longer when the wind came from north. To illustrate the different footprints the FTIR picked up with different wind directions, a plot based on the geometry of the path is shown below:

Portion of the measurement path of the FT-IR catching CO emitted from the highway, depending on the wind direction

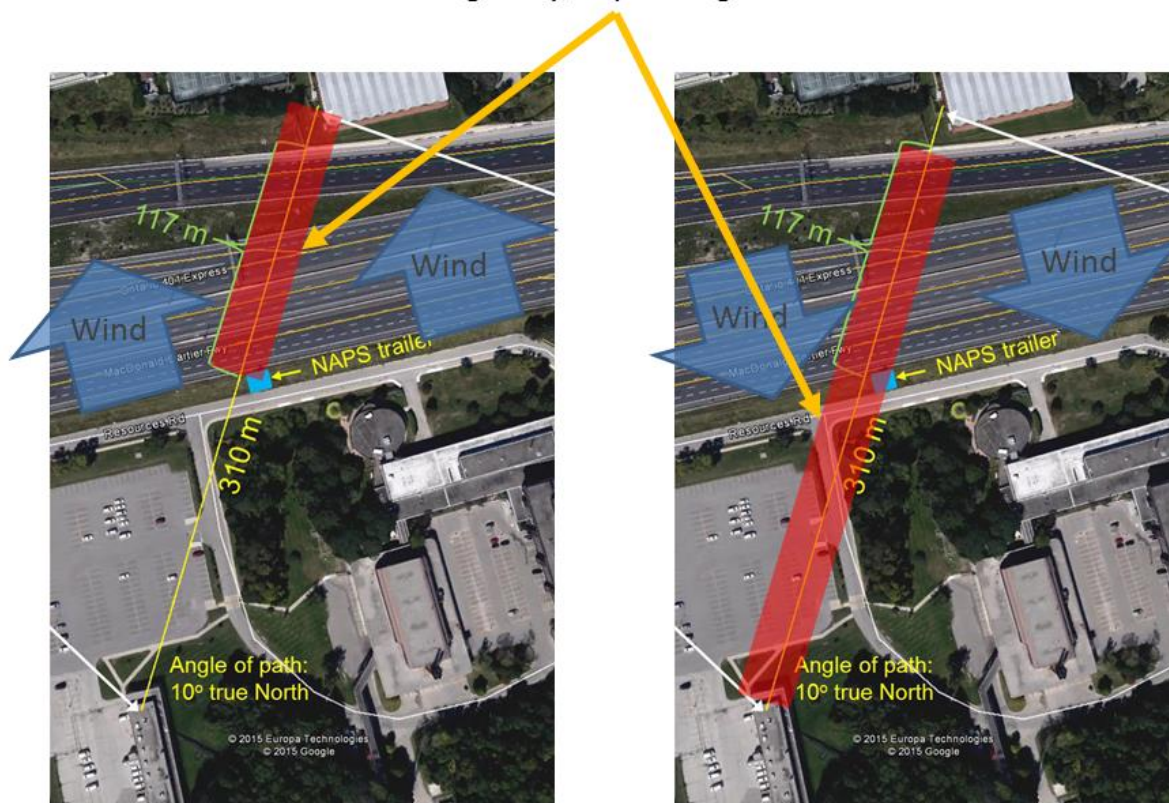


Figure: Dependence of the footprint of the FTIR path measurements on wind direction.

Regardless of wind direction, the FTIR path always picked up some pollution emitted from the highway, and this segment of highway always has traffic (daily minimum is around 3500 vehicle per hour at around 2 to 4 a.m.). So the FTIR measurement never actually sees mixing ratios at the level of urban background air. The NAPS trailer observed a lower pollution level than the FTIR when the wind came from the south because the footprint does not include the highway. This explains the offset between minimum levels of

CO_FTIR and CO_NAPS. Again, CO_NAPS and CO_FTIR were not expected to be anti-correlated at any time or wind direction.

In the supplementary material Fig. S1, we have included examples of a spectrum, and spectral fitting of CO, O₃, and NH₃, at 16:01:45 on July 29, when the wind came from south and a significant offset was observed. These plots of fittings do not show any significant residuals or problems.

We also have checked that NAPS measurement of CO had zero calibrations four times each day. The average of the zero readings was 12.6 ppb with a standard deviation of 7.4 ppb. So the NAPS measurement of CO did not have significant offset and we estimate the uncertainty of CO_NAPS to be 22 ppb (three times of the standard deviation during zero calibration).

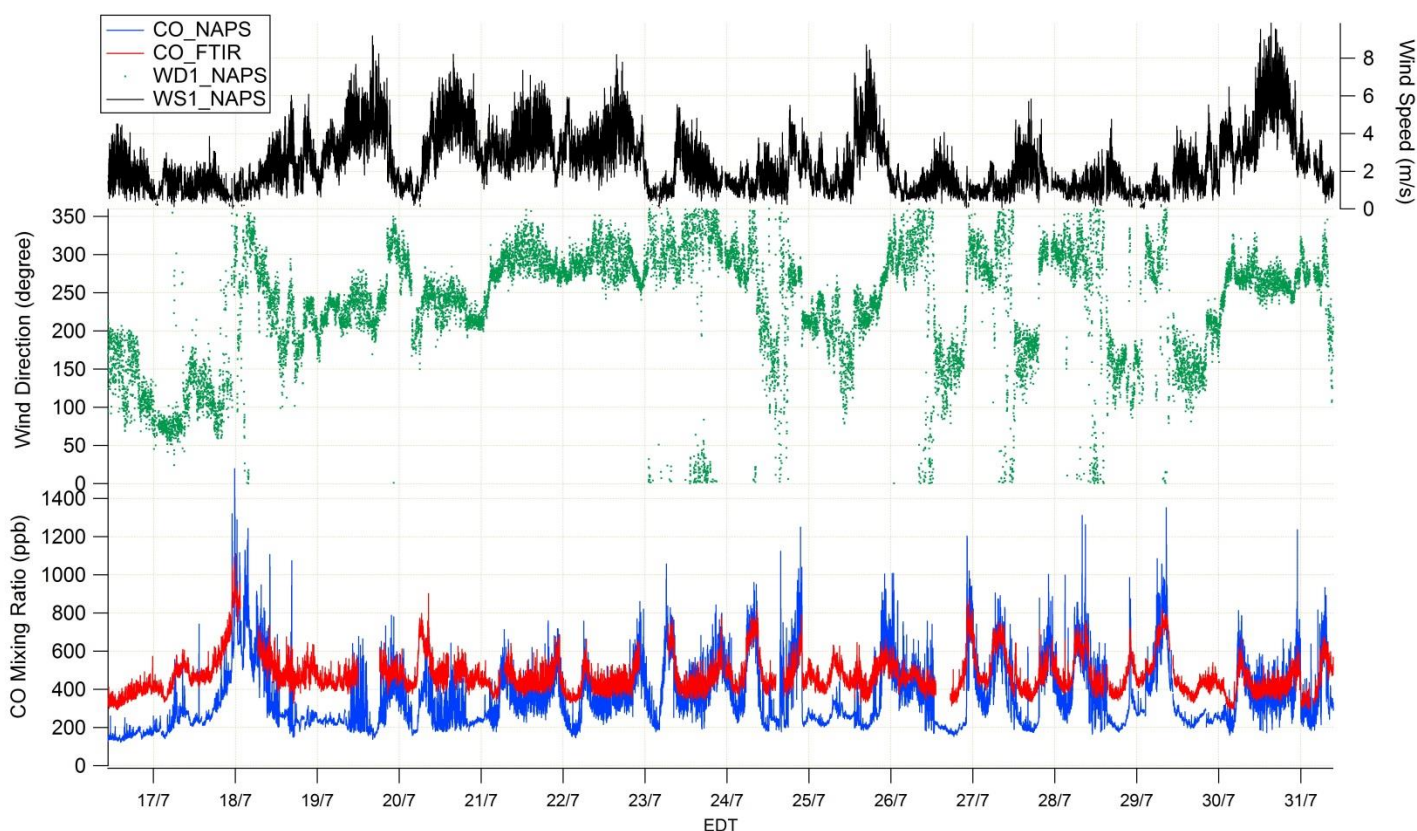


Figure: Time series of CO_FTIR, CO_NAPS, wind speed and wind direction.

5. Specifically, if one observes the CO plot in Figure 3 focusing only on off hours (weekends and during the deep night-time hours), one can draw horizontal lines through the mixing ratio minima for the two methods. While this reviewer is limited to a pencil and paper for the analysis, for such lines “urban minimum baseline” mixing ratio we find the minimum value for the FTIR is ~380 ppb and for the NAPS data the off-hours minimum ~190 ppb, almost exactly a factor of 2.0 difference for this “clean air” baseline.

As stated above, there really was no urban minimum baseline for the FTIR CO mixing ratio, since it always picked up pollution from the highway, even at night, with a minimum of 3500 vehicles passing by per hour. Therefore, over this extremely busy highway, the concept of “off hours” does not apply. For NAPS, if the wind came from the south, it could get some significantly lower levels than the FTIR.

6. While one could argue that this is due to different dispersion / mixing, this is clearly not the case because: 1) the NAPS point source measurements are 1/2 the FTIR values (always lower), and the FTIR values should be greater, representing increased dilution of CO, i.e. mixed with cleaner air further from the motorway, and

This is not quite correct; the NAPS measurements were not always lower than FTIR. For example, 7/24 around 01:00, 7/26 around 03:00, and 7/27 around 0:30, NAPS and FTIR CO were very close. These periods may correspond to the “off hours” mentioned by the reviewer. However, we think there are no “off hours”, since this segment of highway always has traffic (daily minimum is around 3500 vehicle per hour at around 2 to 4 am). NAPS and FTIR CO were very close during these periods mainly because the wind came from north during these periods. It has been shown in the polar plot Fig. 4a.

7. 2) the overall diurnal variations track each other over the entire time interval. We suggest that the difference is likely due to either a systematic negative offset for the CO measurements via NAPS, or (more likely) a systematic large positive offset for the CO measurements with FTIR. It is unclear which of the two instruments is out of calibration, but we suggest the FTIR values are systematically offset and have the incorrect scaling factor, as nowhere in the paper is a calibration procedure reported. There are differences in the offsets for the other gases as well, but CO appears to have the highest. Furthermore, the mixing ratio range for CO from the FTIR is ca. 60% of the NAPS. Again, just by “eyeballing it” the data would appear to show a relative instrument response relation is of the order $Y_{FTIR} = (0.60) * Y_{NAPS} + 180$ ppb. Picking the EDT of 24/7-25/7 (CO from Figure 3), the minima and maxima values for the NAPS are ~ 220 and ~900 ppb, respectively, which is a range of ~ 680 ppb. While the minima and maxima values for the FTIR are ~ 400 and ~780 ppb, respectively, which is a range of ~ 380 ppb or 56% the dynamic range of the NAPS.

We agree with the reviewer that the FTIR CO “offset” seems suspicious; this is something that bothered us from the beginning. Because of this, we have expended significant effort to ensure that there are no biases introduced by our spectral analysis. An example fit is shown in Fig. S1. Interference by H₂O was investigated and found to be insignificant; and different spectral windows suitable for CO were investigated (2068 – 2198 cm⁻¹, with correlation coefficient threshold = 0.7, the difference is 14 to 26 ppb (25 to 75 percentile)), but all led to the same result. After eliminating the possibility of instrumental biases, the most logical explanation remaining is that the higher offset in the FTIR CO is due to the persistent traffic throughout the night, which always affects at least part of the FTIR optical path.

The issue of the slope of 0.6 (rather than 1.0) is again explained by path-integration vs. point measurement. Due to the proximity of NAPS to the highway, larger maxima in CO_NAPS are to be expected, while the CO_FTIR is always an average that includes diluted air.

We have revised the text in the manuscript that describes how we came to the conclusion that this CO offset between FTIR and NAPS appears to be real:

“A major contributing reason for the differences of CO mixing ratios between the FTIR and the NAPS is that the FTIR and the NAPS were not sampling the exact same air, i.e., the measurements represented different footprints. The FTIR measured the air along the path across and above Highway 401, which always included some pollutants emitted from traffic. In contrast, NAPS numbers represented point measurements beside the south edge of the highway. Therefore, CO mixing ratios measured by the NAPS trailer were more dependent on the wind direction than mixing ratios obtained from the FTIR. When the wind was from the south and towards the highway, the NAPS trailer was mostly blind to the highway; when the wind was from the north, it was immediately downwind it. Therefore, CO mixing ratios from the NAPS are expected to be lower than mixing ratios obtained from the FTIR when the wind is from the south and towards the highway.

The path-integrating approach of the FTIR also has a dilution effect since a significant fraction of the path is not above the source (i.e. the highway). Therefore, the CO mixing ratios obtained from the FTIR should be less than CO mixing ratios from NAPS during winds from the highway towards the NAPS trailer. The polar plot in Fig. 4a clearly shows the dependence of the CO mixing-ratio difference between the FTIR and the NAPS on wind direction. When the wind came from the north over the highway towards the NAPS trailer (above the dashed line), CO mixing ratios from the FTIR were close to or lower than mixing ratios from the NAPS. When the wind was from the south and towards the trailer (below the dashed line), the CO mixing ratios from FTIR were higher than CO mixing ratios from NAPS.”

8. Ozone has less of an offset, but the scaling factor is greater. Using the same EDT as above, the dynamic ranges for NAPS and FTIR are 48 and 20 ppb, respectively, which correspond to the FTIR being 42% the range of NAPS. Clearly, there is (are) a systematic flaw(s) present that biases the results by factors of 1.7 (CO) and 2.4 (ozone) Again, since the measurements do not agree, it is unclear what the mixing ratios are for CO and ozone at this certain location at this specific point in time.

If one calls the difference between point measurements and path-integrated measurements a systematic flaw, then we agree. They cannot be directly compared in the sense of a cross-calibration; differences are due to the fundamental difference in the measurement itself, and as previously discussed, the observed differences can be explained by the differing footprints, point measurement right next to the highway vs. path-integrated/diluted measurement incorporating 1/3 highway and 2/3 parking lot, which, depending on wind direction, will contain highway emissions or not.

The offset of O₃ is mainly due to the different footprint. As already shown in polar plot Figure 7 (a), the difference is small when the wind came from the north, and the main offset occurred when wind came from the south. A time series plot of O₃ with wind direction and speed is also shown below. Both polar plot and time series show that the big “offset” occurred when the wind came from around 120 to 220 degrees. The two measurements were close during July 21 to 24 when the wind was steadily from north. We also admit, as indicated in the updated Table 1, that we have less confidence in the accuracy of the FTIR measurements of O₃.

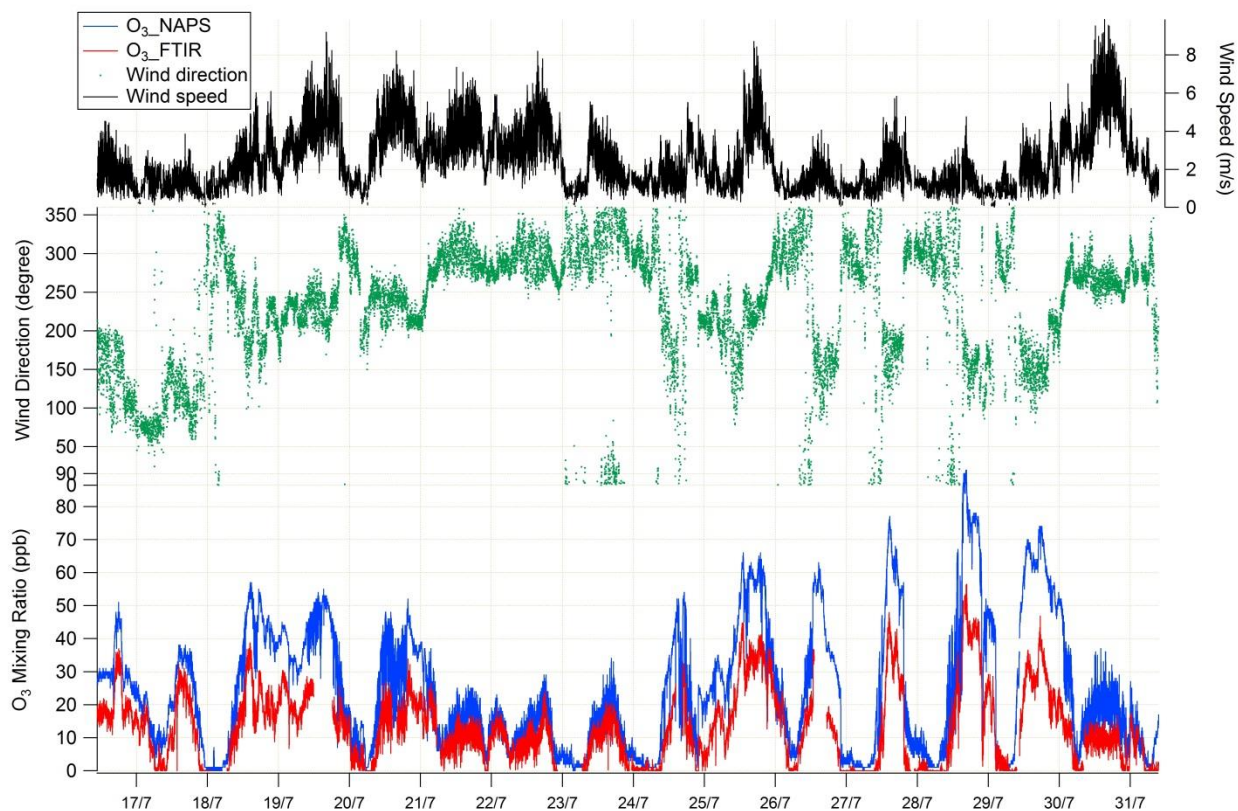


Figure: Time series of O₃_FTIR, O₃_NAPS, wind speed and wind direction.

9. In the paper, there are no explicit data evaluation plots that show what the classic least squares fit looks like relative to the measured spectrum as well as the associated residual plot. It would appear that there was minimal to none of the “hands on” analysis of the FTIR spectra, and as if the results of the Bruker FTIR software (OPS) were used without inspection or vetting. There may thus be miscalculations present that initially went unnoticed. FTIR spectral analysis of gaseous mixtures is not yet a fully automated procedure, but is an interactive process that is subtle and prone to mistakes, interferences, etc. There are several more sophisticated gas analysis programs that may be used to independently confirm or refute the results calculated by OPS, and actual evaluation of the spectra is required.

As already mentioned in our response to comment point #1, when we wrote the manuscript we made the decision not to get into all the details of the FTIR analysis but to focus on the results. To address the reviewer’s concerns, we have included a section in the supplemental material to describe the FTIR spectral analysis by the OPS software. Supplemental text and Fig. S1 have been included to show details of fitting analysis and example spectrum, fitting results, and residual plots. The selected example plot is a spectrum taken at 16:01:45 on July 29, when the reviewer observed a big “systematic offset” of CO and O₃. The wind was from the south. Fig. S1 shows reasonable fits and small residuals of CO, O₃, and NH₃.

10. Furthermore, while authors did correct the raw mixing ratios for temperature and pressure, it appears that the temperature/pressure corrections need to be processed on the reference spectra as well, prior to the fitting. In other words, each reference spectrum (from HITRAN or PNNL) need to be scaled to the correct temperature/pressure then used for quantification. For example, the high resolution reference spectra need to be deresolved to the same resolution of the measured spectra, which in this case is 0.5 cm⁻¹. In the paper, it was not evident if the HITRAN or PNNL reference data were correctly fit to the instrument parameters and instrument lineshape (ILS). It is critical to create the same (instrumental) lineshape and resolution for the fit. Reference spectra may be deresolved using a Gaussian, Lorentzian or Voigt profile, and after deresolving them it is a good idea to check the FWHM to confirm the resolution.

Actually the ILS fit was indeed performed and is included in the quantitative evaluation. We have now included ILS fit information in the supplemental material. Each reference spectrum (from HITRAN or PNNL) was converted to a spectrum with the same resolution (0.5 cm⁻¹ in this study) by the ILS model in OPS software. We did not calculate new reference spectra based on the actual temperature before fitting, but we provided conservative estimations (8.9 % for NH₃, 4.2 % for CH₄, 8.3 % for CO, and 4.1 % for CO₂) of temperature related uncertainty of the retrieved mixing ratios in the range of 5 to 50 °C, which is much wider than the range of difference between ambient temperature and 296 K during our study. We also corrected the raw mixing ratio from FTIR due to the change of air density with change of ambient temperature and pressure.

11. Finally, it appears that one or both instruments were used without any calibration. In order for the scientific community to have confidence in the values obtained, it is important that the instrument(s) undergo some sort of on-site calibration. For the FTIR a wavenumber calibration is necessary and e.g. uses a small gas cells containing NH₃ are used for this purpose; other compounds such as H₂O, CO or CO₂ can also be used.

A wavenumber/line position calibration was performed by using the spectrum fitting analysis of H₂O from the ambient air. In the Bruker software, OPUS_RS, this process is included in the ILS function, as described in the supplementary material. A gas-cell calibration is of course desirable but is not nearly as straightforward as it sounds, since specialized cell windows are required, making this an expensive proposition.

12. Abstract Pg. 1, sent.21: In previous studies, emission factors have units of g kg⁻¹, here the emission factors have units of g km⁻¹. Please explicitly define the emission factor that you are estimating somewhere in the manuscript.

One sentence has been added to the text in Section 3.7 indicating the units of emission factor reported in the references were gram mile⁻¹ and gram km⁻¹, and they were converted into gram km⁻¹. The references in Table 2 also reported emission factors in grams per distance, and they were all converted into grams km⁻¹ to compare to our results. In the last third paragraph in Section 3.7, one sentence was added to explain again that the emission factors results shown in Table 2 were calculated from the emission rate estimates obtained from WindTrax. These emission factors were calculated to compare with previous reported values.

13. Pg. 1 sent 29: There are more pollutants associated with motor vehicles that are not listed here. (Please see Review Article)

Yes, thank you. We have written it out more clearly (such as semi- and low-volatile organic compounds, aromatics and PAHs) and cited more references.

14. Pg. 2 sent 40: change “of” to “in”

It has been corrected.

15. Pg. 2 sent 60: change “spectrometry” to “spectroscopy”

It has been corrected.

16. Pg. 3 sent 65: remove “however” from the sentence.

We keep “however” here, since we are contrasting our study to those in the previous paragraph.

17. Pg. 3 sent 85-86: remove “which were” from the sentence

It has been removed.

18. Pg. 3 sent 87-88: please state what NAPS measures

This sentence has been revised in more specific: “to conduct an in-depth comparison of a range of pollutants (CO, O₃ and NO_x) measured by both the path-integrating FTIR instrument and the in -situ station.”

19. Pg. 3 sent 89: change “in-situ” to “in situ”

It has been corrected.

20. Pg. 3 sent 95: the objectives have already been done (paper by Griffith et al. and Yokelson et al.).

The description of objective point 1 has been revised: “1) to evaluate the capabilities of the long-path FTIR spectroscopy for quantifying the mixing ratios of gaseous pollutants in a heavily polluted open urban traffic environment for a length of time sufficient to cover a range of environmental conditions (16 days)”

Experimental

21. Pg. 4 sent 104: The Global temperature between 1200 and 1500C is its varying state, however, it will not be in its varying state during the measurements.

The exact temperature is not well known, so we decided to leave it out.

22. Pg. 4 sent 105: This is called a bistatic configuration.

Thank you. It has been noted in the revised manuscript.

23. Pg. 4 sent 116: Please add a reference to the end of this sentence. For example: Akagi, S.; Yokelson, R. J.; Burling, I.; Meinardi, S.; Simpson, I.; Blake, D. R.; McMeeking, G.; Sullivan, A.; Lee, T.; Kreidenweis, S., Measurements of reactive trace gases and variable O₃ formation rates in some South Carolina biomass burning plumes. Atmos. Chem. and Phys. 2013, 13 (3), 1141-1165.

We don't think this reference is supportive to that sentence, so we added this reference to the biomass burning sentence in line 65 in the revised manuscript.

24. Pg. 4 sent 119: Please add references to the previous studies that used absorption features in spectral window for analysis.

Yes, I have added a column for references to Table 1.

25. Pg. 4 sent 122: Please add reference for the HITRAN database

I had the reference at line 56, and I added the reference here again this time.

26. Pg. 4 sent 123: Please add reference for the PNNL database

I had a reference at line 57; I added this reference here again this time. I also included Johnson et al. (2010).

27. Pg. 5 sent 126: How were these values adjusted for temperature/pressure? The reference spectra need to be adjusted to the correct temperature/pressure and used in the fitting process. From this sentence, it appears that the reference spectra were not corrected, but instead the reference spectra were used as is to calculate the mixing ratios, which were then adjusted for temperature/pressure.

The reference spectra were not corrected by ambient temperature in the analysis shown in this manuscript. Mixing ratios were calculated by reference spectrum from two databases at 296 K and 298 K, and then were corrected by the density of air depending on ambient temperature and pressure. The effect of temperature on the reference spectrum is small for the range of ambient temperature in this study. We estimated the uncertainty due to the difference between the ambient temperature and reference spectrum in a conservative way, as described in section 2.1 and supplementary material.

28. Pg. 5 sent 131: cite PNNL and HITRAN databases.

They are cited here now. They were already cited in line 56 and 57 in page 2.

29. Pg. 5 sent 135: This is a huge uncertainty for greenhouse gases. For example, is the CO₂ 400? Or 440 ppm?

As we have stated in the text, the maximum uncertainty of 10 % is a conservative estimation because we used a much wider temperature range than the actual ambient temperature range. In addition, we did not report any CO₂ mixing ratio in the manuscript.

30. Pg. 5 sent 146: sensible heat flux. . . what is this?

Sensible heat flux is the turbulent flux of “heat” across a warm-to-cold air gradient. It is a component of the atmosphere/surface energy budget (net radiation = sensible + latent heat + soil heat flux). It is typically measured using eddy covariance, but in this case, we are calculating it using similarity theory, based on the scintillometer data. (Stull, 2003)

Stull, R.B.: An Introduction To Boundary Layer Meteorology, Kluwer Academic Publishers, Netherland, 670 pp., 2003.

31. Pg. 7 sent 174: does this make a difference?

The location of the weather station was written out here to describe the source of solar radiation data. 9 km is close enough at the relevant time scales.

32. Pg. 7 NAPS measurements: please provide type of analysers used at the NAPS station.

Information on the NAPS analyzers has been added to Section 2.3 as follows: “The CO analyzer [Model 48iTrace level-Enhanced, Thermo Fisher Scientific, USA], operates based on infrared absorption and gas filter correlation; the NO_x analyzer [Model 42i Trace Level, Thermo Fisher Scientific, USA] on chemiluminescence; the O₃ analyzer [Model 49i, Thermo Fisher Scientific, USA] on UV absorption; the SO₂ analyzer, [Model 43i, Thermo Fisher Scientific, USA] on UV fluorescence.; and the PM_{2.5} analyzer [Model SHARP 5030, Thermo Fisher Scientific, USA] on light scattering and beta attenuation. Meteorological parameters including air temperature, pressure, relative humidity, and wind speed and direction, were monitored using a WXT520 weather station [Vaisala, Finland].”

Results and Discussion

33. Pg. 9 sent 225: the measurement are off and so what is the point of the study?

As we have explained in points 3, 4, and 5, the offset between two measurements can be explained. They agreed well when wind came from north (highway). The main points of this study have been explained in our responses to comments 2 and 3. Text has been added in lines 89-95 on page 3 in the revised manuscript.

34. Pg. 9 sentence 244 add references to this sentence.

Yes, these references have been included now.

35. Pg. 9 sentence 247 “generally agree with each other, but with a significant offset. . .” what does this mean? The language used here is vague and does not tell us anything.

I have revised this sentence in the text to be more specific: “As shown in Fig. 3, many mixing ratios peaks of CO from the FTIR and the NAPS matched well, and mixing ratios generally correlated with each other

(Fig. S2 a), but with a significant offset and amplitude difference when the wind came from the south (more detailed comparison in the next paragraph).”

36. Pg. 10 paragraph1/looking at figure 3: CO from FTIR has an offset of 390 ppb and the NAPS has an offset of 190 ppb?

Please see our response above in points 4, 5, 6 and 7.

37. Pg. 12 sent. 320-321: Please cite the studies that you are referring to here. Throughout the paper whenever referring to studies, please cite them.

Yes, now I have cited there again. I meant the two references/studies discussed in line 321-330.

38. Pg. 12 sent. 331: Fig 6 should be Fig 5?

Yes, thanks. We have corrected it into Fig. 5 in the revised manuscript.

39. Pg. 13 sent 364: Here you are not comparing the FTIR results (due to water interference) to the model, yet FTIR is mentioned here.

Yes, the reference to the FTIR path has been removed.

40. Pg. 13 sent 369: remove “and a” from the sentence.

Sorry, a typeset mistake. Should have been “significant decrease in the middle of the day and a secondary peak...” It has been revised: “... reaching a peak over 100 ppb from 6:00 to 8:00 followed by significant decrease in the middle of the day and a secondary peak between 20:00 and 23:00.”

41. Pg. 17 sent. 471: Here it states the differences, and the range is large, yet you state that it “within the range of estimates”. Cite some of those ranges to support your claim.

The ranges have been cited there.

42. Pg. 18 sent. 498: change “are in range” to “in the range”

It has been corrected.

43. Pg. 18 sent. 499: change “of” to “the”

It has been corrected.

44. Pg. 19 sent. 529: change “comparable” to “compare”

This sentence has been revised to: “...model results and measurement results are not expected to directly compare for all wind regimes, and comparisons can be better explained after separating wind directions.”

Reviewer 3:

General comments:

1. I think the analysis of the data in some cases has been stretched to the limit of credibility; in particular the comparison of weekday / weekend effects from such a limited data set. The authors should put their 15 days worth of data into context using the longer record of data from the near road site.

The authors recognize that 4 days on weekends are not statistically robust enough to show the difference between weekdays and weekends, especially for O₃. Therefore, we extracted the summer O₃ data in 2013, 2014, and 2015 from a regular NAPS station near the highway. Please see the details in point 12 below, the revised manuscript, and the supplemental material.

Minor comments:

2. P3. In the interest of brevity, the section on page 3 describing how PBL dynamics can impact surface concentration of pollutants is probably unnecessary for the readership of this journal. I found this introductory material unnecessary and I think it adds to manuscript bloat.

We believe this section introduces an essential aspect of our study, and would like to keep it in. We have shortened it by removing 3 sentences. We have also moved most of section 2.2 to the supplementary material.

3. P3. "... first direct comparison of this kind..." It would be good to check the publication of M. Grutter at Centro de Ciencias de la Atmósfera, UNAM, Ciudad Universitaria, Mexico City. He also uses OP FTIR and there had been some big field international air quality field experiments in Mexico City over the last 15 years that would have likely produced opportunities for OP FTIR / fixed point measurement comparisons. I know he has done this for formaldehyde.

We thank the reviewer for the references. Yes, Grutter et al. (2005) did a comparison of op-FTIR with a point measurement at the same site for formaldehyde mixing ratio in downtown Mexico City in 2003. They also did a comparison of op-FTIR measurement of CO with a point measurement of CO at a different site.

We have included Grutter et al. (2005) in the main text. At the end of 5th paragraph in the Introduction, "Grutter et al. (2005) measured the formaldehyde mixing ratio by op-FTIR in downtown Mexico City in 2003 and compared it with a point measurement at the same site." And in the second last paragraph of Introduction, "To our knowledge, (Grutter et al., 2005) presents the first of very few direct comparisons of this kind to be published."

4. P4. Experimental section should list dates of the study period.

The dates of the study have been added to the first paragraph in the Experimental section.

5. P8. It is not clear why the GEM-MACH model results for CO was averaged over 3 hours (1 hour period on each side of the h1-hr period of interest) to get a running average to compare with the 1-hr averages of the data?

This is a misunderstanding. GEM-MACH model outputs were not averaged over 3 hours. This version of GEM-MACH produced concentration of pollutants every hour hh:00 (a snapshot), and we wanted an averaged concentration over 60 minutes, so we took two consecutive outputs from GEM-MACH, averaged these two outputs, and used this as the average concentration for the 60 minutes between these snapshots. This has been described in the second paragraph of Section 2.4.

6. P8. WindTrax. The discussion didn't make clear how the concentration at the measurement site was apportioned to the source area (highway lanes) of interest. Wouldn't the back trajectory model need a high resolution emission model to determine what mass of CO measured at the site was from the emission area of interest? This needs to be clarified for the reader who hasn't used WindTrax. Why is this model needed for equation (9) if the denominator is being determined by another model (the bLS model)? I found this section confusing.

WindTrax is a software tool which incorporates a backward Lagrangian Stochastic (bLS) model (Flesch et al., 1995) with a graphical interface. In this interface, we defined the paved surface of Highway 401 as the only source of CO and other primary pollutants listed in Table 2. The bLS model then releases a large number of virtual particles from the point of measurement and calculates individual trajectories backward in time, based on the input meteorological conditions (wind direction, u_* , L, temperature). The fraction of trajectories that originate from the defined source area is then determined, and used to calculate the simulated relationship between source strength and concentration, as given in the denominator of equation (1). (Please note that equation (9) in the previous version of the manuscript is now equation (1) in the revised manuscript, since we decided to put the theory and equations for the scintillometer in the supplementary material.) This equation can then be used to calculate the actual source strength, given the observed and background concentrations.

To make it clear to readers, Section 2.5 has been revised. "We used a backward Lagrangian Stochastic (bLS) model (WindTrax, <http://www.thunderbeachscientific.com> ; Flesch et al., 1995) which calculates the emission rate Q through formula (1) where C is the concentration of a pollutant at the measurement site, C_b is the background concentration, and $(C/Q)_{sim}$ is the simulated ratio of concentration at the site to the emission rate upwind."

7. Figure 2. I can't tell the difference between the line for z/L and the line for u_* .

The color of the line for u_* has been changed to red, to distinguish it from z/L .

8. Fig 4. This is a nice figure but it isn't clear from the text what is actually plotted – the image looks smoothed to color code difference ranges rather than being a collection of individual data points.

Figure 4 shows the difference of CO concentrations (as the color code) between FTIR measurements and NAPS measurements, as well as GEM-MACH model results and NAPS measurements. Yes, it is correct that the color has been smoothed. The angle is the wind direction, and the radius is the wind speed. We used this figure to show that the difference of concentration from these two sources depends on the wind direction (i. e. across the highway or not). Since the color smoothing may be confusing to other readers as well, we have changed these figures back to show individual data points.

9. P11. Ambient temperature. It is well known that traffic emission of CO can be influenced by temperature but this is primarily due to start emissions when catalytic converters are still cold (< 200 C). Vehicle running emissions of CO are not strongly influenced by ambient temperature. This section has an odd reference “Choi pdf” accessed from the internet. It would be better to cite an actual EPA report on MOVES temperature parameterization of vehicle emissions. One suggested reference is “MOVES2010 Highway Vehicle Temperature, Humidity, Air Conditioning, and Inspection and Maintenance Adjustments”, EPA-420-R-10-027.

Thank you! This EPA report has been added to the references of the revised manuscript.

10. P12. Why was a background value of 256 ppbv used for CO – what is the reasoning for this as a “background” values for the airshed or for upwind of the FTIR beam? Do you get the same value for a CO vs NOx regressions? Air entering the urban airshed or crossing the highway will contain NH3 and CO – shouldn’t these background values be subtracted from both to reveal increase due to local traffic emissions? This background value is an important number as it is later used in the WindTrax calculations so it deserves better definition.

In the revised supplemental material, we have added the plot of [NH₃] vs. [CO] from FTIR results (Fig. S5). From the linear relationship analysis, 265 ppb is the intercept of [CO] ([CO] = 256 ppb when [NH₃] = 0 ppb). Our reasoning was that we assumed that traffic emissions were the only source of CO above background at this spatial scale, and all the NH₃ from traffic emissions were associated with CO. The slope of the linear regression between [NH₃] and [CO] has also been reported by Livingston et al. (2009) and Perrino et al. (2002). We have compared our slope with those slopes, as discussed in Section 3.3 of our manuscript.

From our FTIR observations, [NH₃] reached 0 ppb sometimes (see time series of [NH₃] in Fig. 6), so we did not consider NH₃ background from other sources in this analysis. We used a linear relationship between ([CO]-[CO_background]) and [NH₃], to estimate NH₃ from traffic emissions, which is shown in Fig. 5. We stated “ Overall, there is no indication of a background offset of NH₃, and most measured NH₃ at this site can be accounted for by traffic emissions ” in the manuscript.

As the reviewer suggested, we did another regression analysis: [CO] vs. [NO_x] from NAPS measurement. The intercept of CO on weekdays and weekends are 196 and 186 ppb. This is discussed in point 11 below.

11. P14. It is more common in the literature to report CO vs NOx regressions and to discuss CO-to-NOx molar ratios (cf. the papers by D.D. Parrish or Wallace et al Atmos Environ. 2012). A

ratio of ~5 would be expected for running emissions at your site. I think it would be better to show Figure 10 in the traditional way (NO_x vs CO) so that your slopes could be compared with the literature and vehicle emission inventory.

As mentioned in Section 3.4, we did not retrieve reliable mixing ratios of NO and NO₂ from the FTIR, so we could not do the same regression analysis of [CO] vs. [NO_x] for FTIR measurements. However, we did a regression analysis of [NO_x] vs. [CO] from NAPS measurements as shown in Fig. 10. These slopes have been compared to previous values in Kim et al. (2016) in Section 3.4 in the discussion manuscript. There is another recent publication, Hassler et al. (2016) (D. D. Parrish is one of the authors) showing a long-term observation of [NO_x] and [CO]. In that paper, they showed [NO_x] / [CO] from LA basin observations as well as from near road monitoring stations from Paris and London. Our slopes of 0.1- 0.2 agree well with observation data from LA Basin after 2010.

Figure 10 is in the orientation of [NO_x] vs. [CO] already. Perhaps the reviewer meant a plot of [CO] vs. [NO_x]? We did another regression analysis with NAPS measurement [CO] vs. [NO_x]. The slopes are 3.14 and 7.75 for weekdays and weekends, respectively, i.e., bracketing the expected slope of 5. In Parrish et al. (2002), the slopes are in the range of 6.3 to 18.9, given data from 1987 to 1999. In Wallace et al. (2012), the slope is 4.2 from morning rush hours in Meridian, Idaho, USA in 2008- 2009.

The ratio of [NO_x] / [CO] depends on the mix of vehicle types, fuels and engines, driving cycles, mileage, and emission control techniques. Figure 10 also shows the difference of [NO_x] / [CO] between weekdays and weekend, due to decreased numbers of heavy-duty diesel vehicles on the weekends (as discussed in Section 3.4). It is also consistent with Hassler et al. (2016) showing ratios of [NO_x] / [CO] in Paris and London are higher than ratios in the LA Basin after 2010, due to larger fractions of diesel vehicles in Paris and London than in the LA Basin. Since there are recent publications of [NO_x] vs. [CO], and our slopes can be compared to them, we think either way is acceptable.

We have added this discussion on [CO] vs. [NO_x] in the main text at the end of the third paragraph in Section 3.4: “ Hassler et al. (2016) showed the trend of [NO_x] / [CO] in the Los Angeles Basin, and the ratio is between 0.1 and 0.2 after 2010, which agrees well with our results. There are also some previous studies showing ratio of [CO] / [NO_x] from regions near heavy traffic emission. Parrish et al. (2012) reported the slope of [CO] vs. [NO_x] was in the range of 6.3 to 18.9 for the measurement from 1987 to 1999. Wallace et al. (2012) reported the slope of [CO] vs. [NO_x] was 4.2 in the morning rush hours in 2009. The slopes of [CO] vs. [NO_x] in this study are 3.14 and 7.75 for weekdays and weekends, respectively. Therefore, our results on NO_x and CO are comparable with these previous studies.”

Hassler, B., McDonald, B.C., Frost, G.J., Borbon, A., Carslaw, D.C., Civerolo, K., Granier, C., Monks, P.S., Monks, S., Parrish, D.D., Pollack, I.B., Rosenlof, K.H., Ryerson, T.B., von Schneidemesser, E. and Trainer, M.: Analysis of long-term observations of NO_x and CO in megacities and application to constraining emissions inventories, *Geophys. Res. Lett.*, 43, 9920-9930, doi: 10.1002/2016GL069894, 2016.

12. P14. The analysis of the weekend / weekday comparison of ozone is perhaps more than what the data can support. There were only 2 weekend periods. Is this really enough data to statistically demonstrate that weekends have different ozone production rates that weekdays? Isn't the production and accumulation of ozone in the airshed also affected by meteorology (irradiance,

dispersion)? How were these factors accounted for? You state poor statistics in explaining HCHO patterns. The weekday /weekend difference of vehicle emission on ozone production is interesting but you do not have a statistically relevant difference with 15 days of data. This should be recognized in this section. I would recommend you can place the campaign data into context with ozone data from the NAPS site for a multi-year summer period.

Yes, we agree that only 16 days of results, including 4 weekend days, may not be statistically robust enough to show the difference on ozone production. We also think that the different levels of NO_x between weekdays and weekends affect the ozone production. O_3 production actually goes down at high NO_x (on weekdays), because the OH is reacting with NO_2 to form HNO_3 and not with VOCs to form O_3 and HCHO. Weekend NO_x levels more efficiently form O_3 and HCHO.

Yes, the production and accumulation of O_3 in the surface air probably was also affected by meteorology. As has been discussed in the CO and NH_3 sections, turbulent mixing was much stronger during the day, and sensible heat flux H reached maximum during 12 to 14:00 EDT (Fig. 2). Turbulent mixing would “dilute” the pollutants accumulated in the surface air, so they would not contribute to the peak of O_3 in the early afternoon.

The NAPS station which provided measurement results for this manuscript was a recently installed station, so it was there only for that summer. However, there is another nearby NAPS station which has been in operation for several years. This station is situated about 175 m south of Highway 401, and about 200 m southwest of the new NAPS research trailer. We have extracted the O_3 data from that NAPS station for July and August from 2013 to 2015, to show some multi-year statistics.

Diurnal plots of O_3 on weekdays and weekends in the summer of 2013, 2014 and 2015 are shown in Fig. S7. For 2014 and 2015, the median of the diurnal average on weekends from 7 am to midnight are higher than median on weekdays, which is consistent with the results from the new research NAPS trailer shown in the main text and Fig.9. There is no resolvable difference between weekdays and weekends in 2013. Therefore, the difference between weekdays and weekends we observed in our 16-day measurement agrees with longer observations in 2014 and 2015, which suggests it is representative. We are not expecting the results from these three summers to be same as results from the research NAPS trailer, since this NAPS station is considerably farther from the highway. It is also interesting to see that for 2015, the maximum median [O_3] on weekends (50 ppb) from the regular NAPS station was lower than weekends median [O_3] (54 ppb) from the highway NAPS research trailer. Both stations show differences of 20 – 24 ppb of maximum median [O_3] between weekdays and weekends.

We recognize we do not have enough weekend data to be statistically robust, so as suggested, we have included this discussion above in the supplemental material, and in Fig. S7. We also have added two sentences to the main text in Section 3.4 to summarize these results and thoughts. “To evaluate how representative the contrasts between weekends and weekdays based on this 16-day data set are compared to longer timeframes, 3 summers of O_3 measurements from a nearby NAPS station in Toronto West were extracted and analyzed (Supplementary material Section 3 and Fig. S7). Similar diurnal patterns and differences were observed in 2 of the 3 years, suggesting that the analysis presented above is representative of longer terms as well.”

13. P15. If the gas phase mechanism in the GEM-MACH model does not explicitly represent HCHO then it shouldn't be portrayed as HCHO in Figure 6, that is somewhat misleading if one doesn't read the fine print. What other compounds are included with HCHO, methacrolein and methyl vinyl ketone? If this is the case then I suggest leaving out the model data in Fig 6 for "HCHO".

Methacrolein is partially included in the HCHO lumped model species. As this reviewer suggested, HCHO GEM-MACH results have been removed from Fig. 6.

14. P15. The HCN section is very brief. Any idea why it is so variable; most data appear below DL of instrument except for 3 days at the end of the campaign. If HCN is from vehicle exhaust why isn't it elevated when CO was elevated? It is hard to tell from the figure, but it doesn't seem to follow CO.

As the reviewer noted, most HCN mixing ratios were below detection limit. HCN also did not follow CO, since it was below 1 ppb for most of the period we measured. The variable HCN mixing ratio from FTIR spectrum analysis is probably due to operating so close to its detection limit, which makes it hard to fit HCN's absorption features among much more abundant interfering gases. However, indications are that we did detect a real HCN signal for at least a few short periods. We are very interested in HCN because it has adverse health effects and our colleagues have observed and modeled HCN in emissions from vehicles in the past (Moussa et al., 2016). One sentence has been added to the text in this section: "Only on July 28, 29 and 30th, the HCN was observed above its detection limit".

15. P15. I don't understand the reasoning behind the statement " ...flat on weekends, indicating that a large component of CH₃OH may have come from traffic emissions". Methanol doesn't co-vary with CO from examination of the figures. I find it hard to believe all CH₃OH in an urban area is due to vehicles. What are other sources of methanol? As far as I know methanol is not included as a compound in vehicle emission inventories by the US EPA but perhaps this is different in Canada? Trees emit methanol. You would probably measure similar levels of methanol outside of the Toronto urban area as a result. Are there other urban sources of methanol that are relevant, solvent use for example?

The reasoning here is comparing the diurnal cycles on weekdays and weekends; weekends did not have the early morning peak which corresponds to rush hour traffic on weekdays, so the early morning peak of CH₃OH on weekdays was probably from traffic emissions. In addition, in the early morning on a few weekdays, we found linear relationship between CO and CH₃OH, and the slopes have been shown in Table 2 to estimate emission rate of CH₃OH from traffic. The linear relationship in the early morning on weekdays also suggests at least some CH₃OH was from traffic emissions.

Besides the two references in the manuscript, there are more studies on urban VOC fluxes. Rantala et al. (2016) studied urban VOC fluxes in urban Helsinki and found methanol fluxes were correlated with traffic and with CO fluxes. Traffic could partially explain the observed methanol. Sahu and Saxena (2015) also reported CH₃OH mixing ratios at Ahmedabad (an urban site in India), and both traffic emission and the transport from biomass burning and biogenic sources outside the city contributed to

CH₃OH. Reyes et al. (2006) reported vehicle emission of non-regulated pollutants, including methanol, by using local gasoline and driving conditions in Mexico City.

Other sources of methanol include biogenic sources, such as trees and forest plants, residential wood combustion, and biomass burning. Our measurement could not directly distinguish different sources. We found no relationship of [CH₃OH] with ambient temperature (Fig. S6c), and we observed a linear relationship between [CH₃OH] and [CO] in the early morning on some weekdays. Therefore we concluded that at least some of the observed CH₃OH was from the traffic.

We have revised Section 3.6: “As shown in Fig. 6, mixing ratios of CH₃OH from the FTIR were between 2 and 20 ppb most of the time, with some high spikes. Figure 12 presents the corresponding average weekday and weekend diurnal cycles of CH₃OH for the study period. This plot shows the mixing ratio reached a peak (maximum of 20 ppb at 7:30) from 7:00 to 9:00 on weekdays whereas there was no peak in the mornings on weekends. In addition, a linear relationship between [CH₃OH] and [CO] was observed during the early morning rush hours on some weekdays (Table 2). These results suggest that at least a fraction of observed CH₃OH was from traffic emissions. Observations of methanol associated with traffic have been reported in other studies. Rogers et al. (2006) reported CH₃OH in the diluted pipeline exhaust of a mobile laboratory. CH₃OH may also come from non-engine sources, such as windshield wiper fluid. Durant et al. (2010) measured gas and particle pollutants near Interstate 93 in Massachusetts. They reported CH₃OH was above 20 ppb at 7:20 50 m downwind of the highway, possibly with contributions from some other local sources. Reyes et al. (2006) reported vehicle emission of non-regulated pollutants, including methanol, by using local gasoline and driving conditions in Mexico City. Rantala et al. (2016) studied urban VOC fluxes in urban Helsinki and found methanol fluxes were correlated with traffic and with CO fluxes, traffic could partially explain the observed methanol. Sahu and Saxena (2015) also reported CH₃OH mixing ratios at Ahmedabad (an urban site in India), and both traffic emission and the transport from biomass burning and biogenic sources outside the city contributed to CH₃OH. The mixing ratio of CH₃OH we observed did not correlate with ambient temperature (Fig. S6c), so there was no strong indication of biogenic sources.”

Rantala, P., Järvi, L., Taipale, R., Laurila, T.K., Patokoski, J., Kajos, M.K., Kurppa, M., Haapanala, S., Siivola, E., Petäjä, T., Ruuskanen, T.M. and Rinne, J.: Anthropogenic and biogenic influence on VOC fluxes at an urban background site in Helsinki, Finland, *Atmos. Chem. Phys.*, 16, 7981-8007, doi: 10.5194/acp-16-7981-2016, 2016.

Sahu, L.K. and Saxena, P.: High time and mass resolved PTR-TOF-MS measurements of VOCs at an urban site of India during winter: Role of anthropogenic, biomass burning, biogenic and photochemical sources, *Atmos. Res.*, 164-165, 84-94, doi: 10.1016/j.atmosres.2015.04.021, 2015.

Reyes, F., Grutter, M., Jazcilevich, A. and González-Oropeza, R.: Technical Note: Analysis of non-regulated vehicular emissions by extractive FTIR spectrometry: Tests on a hybrid car in Mexico City, *Atmos. Chem. Phys.*, 6, 5339-5346, 2006.

Long-path measurements of pollutants and micrometeorology over Highway 401 in Toronto

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Abstract

Traffic emissions contribute significantly to urban air pollution. Measurements were conducted over Highway 401 in Toronto, Canada, with a long-path Fourier Transform Infra-Red Spectrometer (FTIR) combined with a suite of micrometeorological instruments, to identify and quantify a range of air pollutants. Results were compared with simultaneous in-situ observations at a roadside monitoring station, and with output from a special version of the operational Canadian air quality forecast model (GEM-MACH). Elevated mixing ratios of ammonia (0-23 ppb) were observed, of which 76 % were associated with traffic emissions. Hydrogen cyanide was identified at mixing ratios between 0 and 4 ppb. Using a simple dispersion model, an integrated emission factor of on average 2.6 g km⁻¹ carbon monoxide was calculated for this defined section of Highway 401, which agreed well with estimates based on vehicular emission factors and observed traffic volumes. Based on the same dispersion calculations, vehicular average emission factors of 0.04, 0.36 and 0.15 g km⁻¹ were calculated for ammonia, nitrogen oxide, and methanol respectively.

1. Introduction

In 1996, 45.2 % of the population of Toronto, Canada's largest city, lived within 500 m of a highway or within 100 m of a major road (HEI, 2010). This percentage was updated to 40 % in 2002 and 2005 (Su et al., 2015). Therefore, a significant portion of the population is exposed to traffic-related air pollution. Pollutants that have been previously reported from motor vehicles include nitrogen oxides (NO_x), carbon monoxide (CO), ultrafine particles ~~and~~ PM_{2.5}, black carbon, ~~and~~ volatile organic compounds (VOCs), semi- and low-volatile organic compounds, aromatics, polycyclic aromatic hydrocarbons (PAHs) and greenhouse gases (Brugge et al., 2007; Zhou and Levy, 2007; Karner et al., 2010, [Gentner et al., 2012 and 2017](#); [Popa et al., 2014](#)). Motor-vehicle-related emissions contributed about 40 % of the PM_{2.5} in Toronto during 2000 to 2001 according to Lee et al. (2003). A study on a global scale indicated that traffic emissions are important contributors to outdoor air pollution

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(ozone (O₃) and PM_{2.5}) associated with premature mortality in 2010 for the U.S.A., Germany, and the U.K. (Lelieveld et al., 2015).

Exposure to these air pollutants is associated with negative health effects. Laboratory studies have indicated that inhalation of fine particles and O₃ even for a short time causes acute conduit artery vasoconstriction (Brook, 2002). Studies in Toronto have shown that exposure to traffic-related air pollution is associated with respiratory conditions (Buckeridge et al., 2002), increased risk of circulatory mortality (Jerrett et al., 2009), cardiovascular mortality (Chen et al., 2013), ischemic heart disease (Beckerman et al., 2012), and childhood atopic asthma (Shankardass et al., 2015). Research results in other locations have also shown associated negative health effects, such as asthma (Lin et al., 2002; McConnell et al., 2006), cancer and leukemia (Pearson et al., 2000) ~~in~~ children, and development of obesity in children (Jerrett et al., 2014). Exposure to traffic-related air pollution may also be associated with increased risk of dementia. Chen et al. (2017) studied a large adult population of Ontario between 2001 and 2012, and they found that incident dementia was 7 % higher for people living within 50 m away from major roads than for the general population.

The segment of Highway 401 crossing Toronto is the busiest highway in North America, with annual average daily traffic (AADT) counts of 410,000 (Ontario Ministry of Transportation, 2013). A few studies on air pollution have been conducted near Highway 401 in the Greater Toronto Area in the past. Beckerman et al. (2008) measured air pollutants at the same location as the current study presented here. They showed elevated nitrogen dioxide (NO₂) and VOCs levels both upwind and downwind of Highway 401, and pollutants did not decay to background levels until 300-500 m downwind.

~~The main~~One focus of this study was to measure gaseous pollutants from a [highway segment with very high traffic](#) through the use of a long-path Fourier Transform Infra-Red (FTIR) spectrometer [for 16 days](#). Compared to off-line post analytical methods, FTIR can measure mixing ratios (~~is also referred to as~~ “mole fractions” ~~in some studies~~) of a variety of gaseous pollutants in near-real time simultaneously, without a container or tubing and without experimental contamination after sampling ([Griffith and Jamie, 2000](#)). Another advantage of FTIR is that it retrieves path-averaged mixing ratios instead of point measurements, so it is less dependent on wind direction. A common approach to retrieve mixing ratios of species from FTIR measurements is to compare the measured spectra with reference spectra obtained in the laboratory at a given temperature and pressure with a known mixing ratio. The Pacific Northwest National Laboratory (PNNL) established a database of gas-phase infrared spectra for pure compounds (Sharpe et al., 2004; [Johnson et al., 2010](#)). Another source of reference spectra is the molecular absorption database HITRAN (HIgh resolution TRANsmission molecular absorption database) (Rothman et al., 1998, 2013). A major weakness of FTIR is the interference from water vapour, which can be too strong for some species and some absorption features, for example when quantifying mixing ratios of nitrogen oxide (NO) and NO₂ in a humid environment.

65 FTIR ~~spectrometry-spectroscopy~~ has been used to quantify the mixing ratios of various trace species emitted by forest biomass burning ([Griffith et al., 1991](#); Yokelson et al., 1996, 1997, [2007, 2008, 2013](#); [Goode et al., 1999](#); Yokelson, 1999; [Burling et al., 2010](#); [Johnson et al., 2010](#); Akagi et al., [2013 and 2014](#); Paton-Walsh et al., 2014; Smith et al., 2014), volcanoes (Horrocks et al., 1999; [Oppenheimer and Kyle, 2008](#)), industrial parks (Wu et al., 1995), and in urban areas ([Grutter et al., 2003](#); Hong et al., 2004; [Coleman et al., 2015](#)). FTIR was also used in flux measurements by the gradient technique at agriculture sites ([Griffith and Galle, 2000](#)). High resolution FTIR has also been used to obtain ozone profiles in the Canadian Arctic ([Lindenmaier et al., 2010](#)).- Vehicle emissions have also been investigated in a tunnel by ~~open~~long-path FTIR ([Bishop et al., 1996](#); [Popa et al., 2014](#)). Bradley et al. (2000) performed a three-hour measurement in the morning beside a road in Denver using long-path FTIR and quantified mixing ratios of CO, carbon dioxide (CO₂), and nitrous oxide (N₂O). [Grutter et al. \(2005\) measured the formaldehyde mixing ratios by open-path FTIR in downtown Mexico City in 2003 and compared it with a point measurement at the same site.](#)

75 There are very few studies, however, that combine direct measurements of mixing ratios of gas-phase pollutants from highway emissions with detailed information on the micrometeorology at the same time and same location. Micrometeorological conditions will be shown here to have a significant effect on modulating the observed mixing ratios. Baldauf et al. (2008) studied the effect of traffic emission and meteorological conditions on the local air quality near a road in Raleigh, North Carolina, U.S.A. in 2006 using long-path FTIR. ~~The CO and NH₃ mixing ratios reached their peaks around 7:00, corresponding to the morning commuter rush hour. Another peak of CO around 17:00 corresponded to a traffic peak as well. Their results showed that horizontal turbulence intensities were large between noon and 17:00 when measured pollutant mixing ratios were low.~~ Brachtel et al. (2009) measured ~~polycyclic aromatic hydrocarbon~~PAH mixing ratios along with CO, sulfur dioxide (SO₂), NO_x, and PM_{2.5} at ~~the side of a road~~roadside for four days in Quito, Ecuador. An early morning peak followed by a sharp drop of mixing ratios was observed, corresponding to the sharp increase of solar irradiation after 7:00. Their results also showed
85 another weaker peak of CO, NO_x and PM_{2.5} between 20:00 and 21:00, after solar irradiation decreased to zero and temperature dropped. Other studies monitored ambient temperature and wind speed to understand meteorological and mixing conditions, and changes in pollutant mixing ratios were found to correlate with these conditions. Gentner et al. (2009) measured CO and VOC mixing ratios 1 km from a highway for two month-long periods in Riverside, California in 2005. They attributed the minimum CO mixing ratios observed in the afternoon to increased mixing and dilution. Durant et al. (2010) presented one-day
90 measurements of pollutant mixing ratios, wind speed, and ambient temperature, along with traffic density. They observed an increase of pollutant levels before sunrise and a sharp decrease after sunrise. Hu et al. (2009) monitored pollutant mixing ratios, wind, and ambient temperature in the early morning period for three days. They found mixing ratios were much higher before sunrise even though traffic volume was lower than later during the daytime.

95 In this study we conducted measurements of gaseous pollutants, along with ~~ierometeorological-turbulent mixing~~ conditions in the surface layer, continuously over two weeks from July 16 to July 31, 2015 across Highway 401 in Toronto. Quantified

pollutants ~~which were~~ discussed in the text included: CO, NH₃, O₃, formaldehyde (HCHO), hydrogen cyanide (HCN), and methanol (CH₃OH). [NH₃, HCHO, and HCN have important implications to atmospheric chemistry and population health, and for which in-situ measurements over busy highways have not been commonly reported for these \(see details and references in the following sections\).](#) In addition, we used the proximity of a NAPS (National Air Pollution Surveillance) surface measurement station, which was located near the middle of the FTIR path next to the highway, to conduct an in-depth comparison of ~~a range of~~ pollutants (~~CO, O₃, and NO_x~~) measured by both the path-integrating FTIR instrument and the in-situ station (~~CO, O₃, and NO_x~~). To our knowledge, ~~Grutter et al. (2005) this presents~~ the first [of very few](#) direct comparisons of this kind to be published. These data are then used to evaluate a research version of the GEM-MACH (Global Environmental Multiscale model-Modelling Air quality and CHEmistry) air quality forecast model (Moran et al., 2010; Gong et al., 2015; Makar et al., 2015a). Finally, highway-integrated emission rates [of a few primary pollutants](#) are estimated [by a “top-down” approach](#) using a backward Lagrangian stochastic dispersion model, and compared with previously published engine emission results scaled by traffic volume.

The objectives to be addressed with this analysis are: 1) to evaluate the capabilities of the long-path FTIR ~~spectrometer spectroscopy~~ for quantifying the mixing ratios of gaseous pollutants [in a heavily polluted open urban traffic environment for a length of time sufficient to cover a range of environmental conditions \(16 days\)](#); 2) to quantify gaseous-pollutant mixing ratios as a function of traffic volume and micrometeorological conditions; 3) to compare mixing ratios from these direct measurements to GEM-MACH model results; and 4) to evaluate the feasibility of deriving emission rate estimates from these measurements using an inverse dispersion model.

2. Experimental

2.1 Long-path FTIR setup and analysis

As shown in Fig. 1, the FTIR and scintillometer instruments were set up on the south side of Highway 401 at 125 Resources Road (43.711° N, 79.543° W) in Toronto, Ontario, Canada. Our study [was from July 16 to 31, in 2015. The FTIR measurements were taken by](#) ~~with used~~ a commercial Open Path FTIR ~~spectrometer Spectrometer~~ (Open Path Air Monitoring System (OPS), Bruker, Germany). The infrared source is an air-cooled Global ~~at a fixed temperature between 1200 and 1500 °C~~. The emitted radiation is directed ~~with a lens~~ through the interferometer where it is modulated, travels along the measurement path across the highway, reaches a retroreflector array that reflects the radiation back, travels back across the highway, and enters a Stirling-cooled mercury cadmium telluride (MCT) detector ([bistatic configuration](#)). The FTIR spectrometer was set up on the roof of a building, about 8 m above the ground, while the retroreflector array was mounted on a mast at 4 m above ground level north of Highway 401. The distance between the spectrometer and retroreflector array was 310 m, resulting in a path length of 623.7 m, which includes 3.7 m of internal reflections. The [fraction-length](#) of the path that was directly over the highway was 117 m (Fig. 1).

In this study, spectra were measured at a resolution of 0.5 cm^{-1} with 250 scans co-added to increase signal-to-noise ratio, resulting in roughly a one-minute temporal resolution. Before July 23rd, 100 scan co-adds were used. At the beginning of the measurement period, a stray light spectrum was recorded by pointing the spectrometer away from the retroreflector. This stray light spectrum accounts for radiation [back to the detector from entering the spectrometer from reflections by internal parts inside the spectrometer, i. e. not from the retroreflector array, and; this](#) was subtracted from all the measurement spectra before performing further analysis. [Stray light had a retrieved affected retrieved final mixing ratios by < 3% in this study. The Bruker OPUS software analyzes spectra to derive the mixing ratios of the gases of interest using a quantitative method described in detail in Griffith \(1996\).](#) Spectral ranges for retrieval analysis [in the Bruker software, OPUS_RS,](#) for each target gas were chosen based on prominent absorption features of the target gas and spectral windows as found in previous studies [as shown in Table 1.](#) Reference spectra were fitted to the measured spectra using [classie least squares \(CLS\) non-linear curve fitting](#) methods within the chosen window.

For each gas of interest, a reference file was made including spectra of target and interference gases. High-resolution reference spectra at 296 K and 1013.25 hPa were taken from the HITRAN database when available. For species not available in the HITRAN database, the reference spectra were taken from the PNNL database. Spectral ranges for fitting, interference gases, and detection limit based on Bruker's results for each pollutant retrieved in this work are listed in Table 1. [Examples of measured spectra, model fit spectra in the optimum spectral ranges, and residuals are shown in Fig. S1.](#) Raw estimates of mixing ratios of gases of interest were retrieved assuming an ambient temperature of 296 K and air pressure of 1013.25 hPa. These values were then corrected for the actual temperature and pressure measured at the NAPS station using the ideal gas law.

The air temperature also has a secondary effect on the signature of the IR spectrum of individual gases. The population of the higher vibrational and rotational states increases as temperature increases. However, the sensitivity of temperature on those signatures depends on the individual gas and the range of temperature change. [HITRAN and PNNL r](#)Reference spectra at different temperatures are available for a limited number of species at 278, 298, and 323 K [\(Rothman et al., 1998, 2013; Sharpe et al., 2004; Johnson et al., 2010\).](#) Temperature-dependent reference files can be made in the [OPUS_RS](#) software to combine reference spectra at these three temperatures. To test the effect of temperature on the retrieved mixing ratio, spectra during the last eight days of July were analyzed for NH_3 , CH_4 , CO , and CO_2 by using these temperature-dependent reference files. The maximum difference in retrieved mixing ratio for the 45°C range is 8.9 % for NH_3 , 4.2 % for CH_4 , 8.3 % for CO , and 4.1 % for CO_2 . Based on this test, we estimate that using reference spectra at standard temperature and pressure contributed to an uncertainty of less than 10 % in the final mixing ratio results. [Besides fitting errors and the effect of ambient temperature on reference spectrum, other environmental conditions may also contribute to uncertainties, such as interference from ambient water vapor.](#)

2.2 Scintillometer theory and setup

Simultaneous long-path turbulence measurements were made using a Boundary Layer Scintillometer (BLS 900, Scintec, Germany). The scintillometer receiver was set up next to the FTIR spectrometer, on the south side of Highway 401 (Fig. 1). The transmitter, with two disks of 924 LEDs emitting at 880 nm, was set up on the north side of Highway 401 just above the FTIR retroreflector. The mean height of the scintillometer path was 8 m above ground level. In this study, the LEDs were operated in continuous mode. Radiation is directed onto the photodiodes in the receiver, which quantify the turbulence-induced fluctuations in the optical refractive index between the transmitter and receiver. [The theory of scintillometer measurements and calculations are included in the Supplementary Material, Section 2. Sensible heat flux \(H\), friction velocity \(\$u_*\$ \), and Obukhov length \(L\) were calculated from scintillometer measurements at a 1-minute resolution in this study.](#)

The structure function of the optical refractive index C_n^2 is determined from the fluctuations of the light beam intensities received by the scintillometer receiver. The structure parameter of temperature C_T^2 can be derived from C_n^2 given the ambient temperature, humidity, pressure, and wavelength (Thiermann and Grassl, 1992). The determination of the sensible heat flux H based on C_T^2 needs additional assumptions based on Monin-Obukhov Similarity Theory (MOST).

H is defined as

$$H = -\rho C_p \theta_* u_* \quad (1)$$

where ρ is the air density, C_p is the specific heat capacity of air at constant pressure, θ_* is the temperature scale, and u_* is the friction velocity. θ_* and u_* are determined by the MOST functions (Wood et al., 2013):

$$\frac{C_T^2 z^{2/3}}{\theta_*^2} = \Psi_H \left(\frac{z}{L} \right) \quad (2)$$

$$\frac{\kappa z U(z)}{\ln \left(\frac{z}{z_0} \right) u_*} = \Psi_M \left(\frac{z}{L} \right) \quad (3)$$

where the Obukov length L is defined as

$$L = \frac{T u_*^3}{\kappa g \theta_*} \quad (4)$$

and z is the height above the surface, z_0 is the roughness length, the von Kármán constant $\kappa=0.4$, g is the gravitational constant, and $U(z)$ is the mean horizontal wind speed. The scaling functions Ψ_H and Ψ_M can be calculated by Eqs. (5) and (6) (Thiermann and Grassl, 1992) and Eqs. (7) (Paulson, 1970) and (8) (Businger et al., 1971):

$$\Psi_H = 4\beta_1 \left[1 - 7\frac{z}{L} + 75\left(\frac{z}{L}\right)^2 \right]^{-1/3} \quad \text{for } \frac{z}{L} < 0 \quad (5)$$

$$\Psi_H = 4\beta_1 \left[1 - 7\frac{z}{L} + 20\left(\frac{z}{L}\right)^2 \right]^{1/3} \quad \text{for } \frac{z}{L} > 0 \quad (6)$$

with $\beta_1=0.86$

$$\Psi_M = -2 \ln\left(\frac{1+\chi}{2}\right) - \ln\left(\frac{1+\chi^2}{2}\right) + 2 \arctan(\chi) - \frac{\pi}{2} \quad \text{for } \frac{z}{L} < 0 \quad (7)$$

$$\text{where } \chi = \left(1 - 15\frac{z}{L}\right)^{1/4}$$

$$\text{and } \Psi_M = 4.7\frac{z}{L} \quad \text{for } \frac{z}{L} > 0 \quad (8)$$

These calculations were performed with the software provided by Scintec (SRUN, version 1.31; see <http://www.scintec.com/english/web/scintec/Details/A012000.aspx>). The procedure to calculate the sensible heat flux H from measurements is as follows:

An initial $|L|=|L_{\text{min}}|=1000$ m is assumed, where the signs of L and H are determined by the simultaneous measurement of the vertical temperature gradient $\Delta T/\Delta z$. Then Ψ_H is calculated from Eqs. (5) and (6) using L and path height z . θ_x can then be calculated with C_x^2 using Eq. (2). Next Ψ_M is calculated from Eqs. (7) and (8). Friction velocity u_x is then calculated by Eq. (3) given the measured wind speed (from the NAPS station) and z_0 . A new L can then be calculated from Eq. (4) using the calculated θ_x and u_x . This process is then repeated iteratively until the change in L is smaller than 0.1. Finally, H is calculated from Eq. (1) using the last calculated values for θ_x and u_x .

z/L (where z is the height above the surface) is a surface-layer scaling parameter describing the dynamic stability of the surface layer (Stull, 2003). Negative z/L values indicate an unstable surface layer, while positive z/L values indicate a stable surface layer. The closer the value of z/L is to zero, the closer conditions are to neutral stability. In this work, H and z/L were used to determine the strength of turbulence and mixing in the surface layer (Fig. 2). Solar radiation data were taken from a York University weather station (<http://www.yorku.ca/pat/weatherStation/index.asp>) situated about 9 km north-west-east of our site.

We used the downwelling short wavelength radiation data to quantify cloudiness during the study. During the 16-day measurement period, only July 17th had some rain, and all other days were mostly sunny.

2.3 NAPS measurements

The NAPS program aims to provide accurate and long-term air quality data with uniform standards across Canada by coordinating the data collection from existing air quality monitoring networks (Galarneau et al., 2016). The first NAPS measurements were conducted in 1972, focusing on SO₂ and particulate matter. Currently, SO₂, CO, NO₂, O₃, and PM_{2.5} are continuously measured at more than 200 sites across Canada (Environment and Climate Change Canada, <http://www.ec.gc.ca/rnsps-naps/default.asp?lang=En&n=8BA86647-1>, last accessed March 25, 2017). The data shown in this study come from the NAPS trailer located right beside the FTIR path on the south edge of the Highway 401 (see Fig. 1). Pollutants monitored by this NAPS trailer include CO, NO, NO₂, O₃, SO₂, and PM_{2.5} at one-minute temporal resolution. The CO analyzer [Model 48i Trace level-Enhanced, Thermo Fisher Scientific, USA], operates based on infrared absorption and gas filter correlation; -the NO_x analyzer [Model 42i Trace Level, Thermo Fisher Scientific, USA] on chemiluminescence; the O₃ analyzer [Model 49i, Thermo Fisher Scientific, USA] on UV absorption-; the SO₂ analyzer, [Model 43i, Thermo Fisher Scientific, USA] on UV fluorescence-; and the PM_{2.5} analyzer [Model SHARP 5030, Thermo Fisher Scientific, USA] on light scattering and beta attenuation. Meteorological parameters including air temperature, pressure, relative humidity, and wind speed and direction, were monitored using a WXT520 weather station [Vaisala, Finland].

2.4 GEM-MACH model

GEM-MACH is a chemical transport model embedded within the GEM (Global Environmental Multiscale) numerical weather forecast model of Environment and Climate Change Canada (ECCC) (Côté et al., 1998a,b). Meteorological conditions (Makar et al., 2015b) and air quality processes, including gas-phase, aqueous-phase, and heterogeneous chemistry and size-resolved aerosol processes, are included in GEM-MACH (Moran et al., 2010; Gong et al., 2015; Makar et al., 2015a). GEM-MACH is used operationally by ECCC for short-term air quality forecasting on a North American grid with 10-km horizontal grid spacing (Moran et al., 2014; Pavlovic et al., 2016). In this study, a research version of GEM-MACH simulated concentrations of pollutants with a horizontal grid-cell size of 2.5 km within a 40 m layer above ground level. Our measurement site was located within one model grid cell. Hourly outputs were obtained from GEM-MACH in this study.

GEM-MACH outputs instantaneous pollutant mixing ratio fields once an hour, including CO, O₃, NH₃, HCHO, NO, and NO₂. The FTIR and the NAPS measured mixing ratios once a minute. In order to compare model results and measurements for similar periods, GEM-MACH results were averaged over the two bracketing timestamps to get an estimate of the average mixing ratio of pollutants over each hour, while the measurements results were averaged every hour to match the temporal resolution of GEM-MACH results.

2.5 WindTrax estimation of source emission rate from mixing ratio measurements

Various approaches have been developed to deduce source emission rates from pollutant concentrations, including inverse dispersion models (cf. Flesch et al., 2004). We used a backward Lagrangian Stochastic (bLS) model (WindTrax, <http://www.thunderbeachscientific.com>; Flesch et al., 1995) which calculates the emission rate Q by the formula through

$$Q = \frac{(C - C_b)}{(C/Q)_{sim}} \quad (19)$$

where C is the concentration of a pollutant at the measurement site, C_b is the background concentration, and $(C/Q)_{sim}$ is the simulated ratio of concentration at the site to the emission rate upwind. In this study, $(C/Q)_{sim}$ is calculated by using a the backward Lagrangian Stochastic (bLS) model (Flesch et al., 1995). In the bLS model, a large numbers of virtual particles are is released at the site, and individual upwind trajectories are calculated backward in time from the site. Then the fraction of trajectories that originate from the user-predetermined source area is determined/calculated, which in turn factors into the calculation of $(C/Q)_{sim}$. WindTrax can handle complex source-area shapes but not variations in topography. The micrometeorological condition inputs for the bLS model are u_s (friction velocity) and L (Obukhov length) obtained from the scintillometer measurements (supplementary material Section 2) as well as wind ~~direction~~ and ambient temperature data from the NAPS trailer.

In this study, we used the mixing ratio of CO measured by the FTIR to estimate the CO emission rate from the highway (a “bottom-up/top-down” approach). The estimated emission rates are then compared to the emission rates derived from traffic volumes combined with published emission factors of vehicle engines, i.e. a “bottom-up” approach (Section 3.67). -These results will help evaluate the capability of deducing emission rates from our measurements.

2.6 Traffic volume data

Traffic volume data were provided by the Ontario Ministry of Transport, in units of vehicles per hour passing a point on Highway 401. Before July 20th, counts were available at the Islington Avenue intersection, about 700 metres to the west of our site. However, after July 20th, data at Islington Avenue were not available, and we instead used traffic volume data at a nearby intersection to the east of our site at Avenue Road, which showed a linear relationship with traffic volumes at Islington Avenue. Therefore, the traffic volume data before July 20th were measured at the Islington Avenue intersection, while data after July 20th for the same location ~~are~~ were estimated.

3. Results and discussion

3.1 Micrometeorology

255 During the study, the mean wind speed measured at the NAPS trailer was 2.5 m s^{-1} , with a range from 0 to 9.9 m s^{-1} and quartiles of 1.3 and 3.3 m s^{-1} . The mean friction velocity u_* during the study was 0.40 m s^{-1} , with a range from 0.02 to 1.31 m s^{-1} and quartiles of 0.25 and 0.52 m s^{-1} . The mean ambient temperature was 24°C , with a range from 14 to 33°C .

260 In Toronto in late July, sunrise occurs at about 6:00 EDT (Eastern Daylight saving Time, same time labels were used for the entire study), solar noon occurs at about 13:30, and sunset occurs at about 21:00. As shown in Fig. 2, sensible heat flux H started to increase beginning at 6:00, reached its maximum in the early afternoon around 13:30, and then decreased to its minimum after 23:00. The downwelling shortwave solar radiation started to increase at 6:30 and reached a peak around 13:00. It is notable that H remained positive throughout the night and started to increase before sunrise. We surmise that this is due to traffic providing a source of sensible heat and mechanical and convective turbulence, as well as slow release of heat from the pavement at night (Sailor and Lu, 2004; Khalifa et al., 2016). A rough estimation of 33 W m^{-2} (56 % of H) at 5:30 (before the sunrise) was contributed by vehicles on the highway, based on the traffic volume, the ratio of energy loss from gasoline engine, and the typical fuel consumption of gasoline vehicles. Result of z/L remained negative throughout the night, indicating that the surface layer was always unstable or neutral. u_* also varied diurnally, with higher values from 08:00 to 21:00 and lower values during the night, which also suggests stronger turbulence in the daytime and is correlated with H. All of these measurements show that mixing and turbulence started to increase quickly after sunrise, reached a maximum in the early afternoon, and decreased to a minimum after 23:00.

270 3.2 CO

3.2.1 Comparison between FTIR, NAPS and GEM-MACH

275 CO is directly emitted by vehicles, and CO emission ~~factors~~ from vehicles [and urban activities](#) have been reported in [various previous studies \(for example Chaney, 1983; Stedman, 1989; Stremme et al., 2013; Haugen et al., 2017\)](#). Among those studies, Bradley et al. (2000) and Baldauf et al. (2008) measured CO from traffic by open-path FTIR. CO has also been used as a reference pollutant to determine emission factors of other primary pollutants by calculating concentration ratios of pollutants to CO (Warneke et al., 2007; Baker et al., 2008; Gentner et al., 2013). As shown in Fig. 3, [many mixing ratios peaks](#) of CO from the FTIR and the NAPS [matched well, and mixing ratios](#) generally ~~agree~~ [correlated](#) with each other ([Fig. S2 a](#)), but [these mixing ratios were also](#) with a significant offset and amplitude difference [when the wind came from the south \(more detailed discussion in the next paragraph\)](#). The GEM-MACH simulation predictions and the measurements of CO mixing ratio agree well in general (Fig. 3). The GEM-MACH simulated most of the peak mixing ratios consistent with measurements.

280 ~~One~~ [A major](#) contributing reason for the differences of CO mixing ratios between the FTIR and the NAPS ~~may be~~ is that the FTIR and the NAPS were not sampling the exact same air, i.e., the measurements represented different footprints. The FTIR measured the air along the path across and above the Highway 401, which always included some pollutants emitted from

285 traffic. In contrast, NAPS numbers represented point measurements beside the south edge of the highway. Therefore, CO mixing ratios measured by the NAPS trailer were more dependent on the wind direction than mixing ratios obtained from the FTIR. When the wind was from the south and towards the highway, the NAPS trailer was mostly blind to the highway; when the wind was from the north, it was immediately downwind it. Since CO is directly emitted from the highway Therefore, CO mixing ratios from the NAPS might be expected to be lower than those mixing ratios obtained from the FTIR, particularly when the wind is from the south and towards the highway.

290 The path-integrating approach of FTIR also has a dilution effect since a significant fraction of the path is not above the source (i. e. the highway). Therefore, the CO mixing ratios obtained from the FTIR should be less than CO mixing ratios from NAPS, during the wind from highway towards the NAPS trailer. The polar plot in Fig. 4a clearly shows the dependence of the CO mixing-ratio difference between the FTIR and the NAPS on wind direction. When the wind came from the north over the highway towards the NAPS trailer (above the dashed line), CO mixing ratios from the FTIR were still greater than CO
295 from close to or lower than mixing ratios from the NAPS. When the wind was from the south and towards the trailer (below the dashed line), the CO mixing ratios from FTIR were higher than CO mixing ratios from NAPS.

Spatial incommensurability remains an issue when comparing gridded air quality model predictions with measurements. A GEM-MACH surface-level mixing ratio represents a mean value over a grid-cell volume that is 2.5 km by 2.5 km by 40 m in size whereas the FTIR measurements are averages over a line that is an order of magnitude shorter than the length of the side of
300 GEM-MACH grid cell and the NAPS measurements correspond to values at a single point right at the south edge of the highway. In addition, the emissions considered by GEM-MACH for a particular grid cell include the contributions of all point, line, area, and volume sources contained within that grid cell, and the sum of these multiple sources is assumed to be distributed uniformly across the grid cell (see maps of grid cells Fig. S3+). Thus, the artificial mixing and dilution of emissions within a model grid cell, subgrid-scale variations in wind direction, and the locations of emissions sources relative to measurement
305 locations may impact the comparison between model results and measurements, particularly for primary pollutants.

To investigate the effect of wind direction on the difference of CO mixing ratios between from NAPS and GEM-MACH, the difference was plotted as a function of wind direction (Fig. 4b). GEM-MACH predictions were lower than the NAPS measurements when the wind direction was from the highway towards the NAPS trailer (cold colors), while GEM-MACH predictions were greater than the NAPS measurements when the wind was from other directions (warm colors). A linear
310 regression analysis of CO mixing ratios from GEM-MACH and NAPS stratified by wind direction is shown in Fig. S2b. The slope of the best-fit line when the wind was from the highway to the trailer was less than 1.0 and the mean bias was negative; for winds from other directions, the slope was greater than 1.0 and the mean bias was positive. These results are consistent with the above discussion about point-measurement representativeness vs. model grid-cell averages. When the wind was from the

highway, CO measurements by the NAPS were directly influenced by the trailer's close proximity to heavy traffic emissions, a subgrid-scale emissions feature that could not be well represented by the air quality model.

The dependence of the difference of CO mixing ratios between from FTIR and GEM_MACH on the wind direction was also studied. Figure S2c shows the difference of slopes in GEM-MACH vs. FTIR between conditions when wind from the highway and wind from other directions are smaller than the difference of slopes in GEM-MACH vs. NAPS (Fig S2b), indicating the sampling spatial difference is smaller when comparing 2.5 km grid cell in the model with path-integrated mixing ratio than when comparing 2.5 km grid cell in the model with a fixed point mixing ratio. From the polar plot of (CO_model - CO_FTIR) vs. wind direction in Fig. S4, it also can be seen that positive differences (warm colors) ~~is~~ mainly occur when the wind was from other (non-highway) directions, and the dependence of (CO_model - CO_FTIR) on the wind direction is not as strong as (CO_model - CO_NAPS) (Fig. 4b).

3.2.2 Average diurnal cycles

During weekdays (Fig. 5), the minimum traffic volume was about 5000 vehicles h⁻¹ between 2:00 and 5:00; traffic started to increase after 5:00 and reached a maximum 23800 vehicles h⁻¹ from 7:00 to 8:00. After reaching the morning peak, traffic volume remained high through most of the day, starting to drop after 21:00. The CO mixing ratio on weekdays rapidly reached a peak between 6:00 and 8:00. H and u_* during this period were still low compared to the middle of the day (see Fig. 2), indicating that turbulence was weak compared to the afternoon. This suggests that the peaks of CO mixing ratio observe in the early morning were due to rapid increase and accumulation of emissions of CO while there was still little convection, before stronger mixing started later in the morning. Similar observations have been previously reported (Janhäll et al., 2006; Hu et al., 2009; Durant et al., 2010). In the afternoon on weekdays, when the traffic volume was still high, the CO mixing ratio dropped significantly, compared to early morning rush hour. Turbulence was strong at noon and in the afternoon, so emitted pollutants were diluted efficiently. Therefore CO mixing ratio in the afternoon was lower than in the morning despite similar traffic volumes. In the late evening (21:00 to 0:00), there was a secondary peak in mixing ratios of primary pollutants from traffic even as traffic volume started to drop, again due to diminished vertical mixing leading to accumulation in the surface layer after sunset (Gentner et al., 2009).

The average weekday diurnal cycles of CO, traffic volume and turbulence/mixing clearly show that turbulence and mixing played an important role on the mixing ratios of primary pollutants above the highway. On weekends, traffic volume increased more gradually during the morning until plateauing around 11:30 and on average remained high with about 21800 vehicles h⁻¹ until after 22:00. The diurnal patterns of CO mixing ratio were flatter but with greater variability, compared to weekdays. The median CO mixing ratio on weekends was close to that on weekdays, except for the early morning period. These comparable

CO levels for weekdays and weekends for comparable-similar traffic volumes suggest that traffic was the main emission source of CO.

Ambient temperature may also affect emissions from vehicles and hence pollutant mixing ratios near traffic ([U.S. EPA, 2010](#); [Choi et al., 2010](#); Rubin et al., 2006). However, since the range of ambient temperatures was small during the study period (from 15 to 32 °C), the effect of temperature on the average diurnal cycle of CO mixing ratio was likely weak.

3.3 NH₃

NH₃ can form secondary aerosols that are associated with negative health effects (Seinfeld and Pandis, 2006; Behera and Sharma, 2012; Liu et al., 2015) as well as radiative forcing impacts. According to the U.S. Environmental Protection Agency (U.S. EPA)'s trends data for 2016, 2.4 % of U.S. national NH₃ emissions are from vehicles which are more important sources in urban regions. After the three-way catalytic converter (TWC) was introduced to gasoline vehicles in 1981 and became used widely, NH₃ (as a product formed in TWC from the reaction of NO with CO and H₂O) emissions from vehicles increased (Moeckli et al., 1996; Fraser and Cass, 1998; Kean et al., 2000). NH₃ is also involved as a reagent in the reduction processes for NO in selective catalytic reduction converters (SCR) in diesel vehicles. Therefore, diesel vehicles could also contribute to NH₃ emissions, due to the aging of catalysts and over-doping of urea. However, they play only a minor role in NH₃ traffic emissions compared to gasoline vehicles based on the emission inventory used by GEM-MACH over Greater Toronto and Hamilton Area (ECCC, 2014). NH₃ is gaining importance as a pollutant from traffic due to the gaining use of emission control systems, but previous studies which directly measured NH₃ mixing ratio from traffic are rare. Elevated mixing ratios of NH₃ between 0 and 23 ppb were observed with the FTIR in this study (Fig. 6). Baldauf et al. (2008) showed diurnal plots for traffic volume and mixing ratios of NH₃ measured by open-path FTIR 20 m and 300 m from a main road. The NH₃ mixing ratio they reported was to be between 10 and 35 ppb, comparable to our results.

Traffic emissions appear to be very important to NH₃ in urban environments, although residential garbage collection (Reche et al., 2012), soil and fertilizers, biomass burning, natural ecosystems, sewage and landfill, and direct emissions by humans and animals could also contribute (Sutton et al., 2000). Yao et al. (2013) found a good linear correlation between mixing ratios of NH₃ and NO during periods in the morning at the same site beside Highway 401. CO has been used as a common reference pollutant from vehicle emissions as discussed in Section 3.2.1, and studies have also used [NH₃] / [CO] ratio to correlate NH₃ to traffic emissions ([Perrino et al. 2002](#); [Livingston et al. 2009](#)). A linear correlation between emission factors of CO and NH₃ was found in light and medium-duty vehicles in the California South Coast air basin by Livingston et al. (2009). Perrino et al. (2002) found a linear relationship between mixing ratios of NH₃ and CO at a traffic site in Rome. A linear relationship between NH₃ and CO mixing ratios from the FTIR over the whole period was also observed in this study (slope=0.023, r²=0.60, [Fig. S5](#)). This linear relationship suggests that NH₃ and CO shared a common source, which in this case, a significant fraction (76% during morning rush hour: see discussion in the next paragraph) of NH₃ came from traffic. The slope of 0.023 ([NH₃] / [CO]) is

close to values previously reported (Livingston et al., 2009). NH₃ emission factors from vehicles in the literature are in the range of 0 to 0.144 g km⁻¹ depending on various factors such as fuel type, driving cycle, vehicle engine power, engine temperature, and catalyst aging (Durbin et al., 2002; Huai et al., 2003). Therefore, differences in slopes among studies are to be expected.

Average diurnal cycles of NH₃ mixing ratios on weekdays and weekends are shown in Fig. 5. To estimate the NH₃ due to traffic emissions, it was assumed that [traffic emission was the only source of CO above background at this spatial scale and all NH₃ emissions from traffic emissions are were](#) correlated with CO. Thus, a background CO mixing ratio of 265 ppb was subtracted from the retrieved CO mixing ratio and the result was regressed against NH₃ mixing ratios, resulting in a traffic-related NH₃ being estimated as 0.023×([CO]_{FTIR} - 265) ppb. The 265 ppb CO background was the intercept of CO from the linear regression of NH₃ with CO. The resulting weekday and weekend diurnal cycles of the estimated NH₃ mixing ratio due to traffic emissions are plotted in Fig. 5. During the morning rush hour and late at night on weekdays, traffic emissions contributed more to NH₃ levels than during other times of day. On weekends, the diurnal variation of total NH₃ was weaker, and estimated NH₃ from traffic accounted for essentially all NH₃ observed. Overall, there is no indication of a background offset of NH₃, and most measured NH₃ at this site can be accounted for by traffic emissions.

NH₃ measurements by the FTIR also agreed well with GEM-MACH model simulations (Fig. 6). The analysis results of the traffic contribution to NH₃ around the site based on the FTIR measurements are consistent with the GEM-MACH model NH₃ input emissions, which show that the main source of NH₃ at this location is vehicular (Fig. S34).

3.4 O₃, NO, NO₂, and HCHO

O₃ is a secondary pollutant and is not emitted directly by vehicles. NO reacts with O₃ forming NO₂ on a time scale of a few minutes during the day. Photochemistry between VOCs and ambient oxidants produces O₃, and HCHO is one of the products from these photochemical reactions. The chemistry of titration and photochemical production of O₃ has been discussed previously in detail (Marr and Harley, 2002b; Fujita et al., 2003; Seinfeld and Pandis, 2006; Murphy et al., 2007). The time series of O₃ mixing ratio from the FTIR also agrees broadly with the NAPS O₃ measurements (Fig. 3). However, the polar plot in Fig. 7a shows that O₃ mixing ratios measured by the FTIR and the NAPS were close when the wind was from the highway, whereas O₃ from the FTIR was much lower than O₃ from the NAPS, when the wind was from other directions. These results can be explained by the titration of O₃ over the highway by NO emissions from vehicles: when the wind is from the north, the O₃ reaching the NAPS trailer has been titrated, but when the wind is from the south, O₃ measured at the NAPS site is titrated over the highway downwind of the measurement point.

Mixing ratios of NO and NO₂ can be retrieved from the FTIR spectra, but the correlation coefficients of fitting are less than 0.1 and estimated mixing ratios contain large offsets and biases, probably due to the strong interference from water vapor.

Therefore, mixing ratios of NO and NO₂ from the FTIR are not shown here. The GEM-MACH simulations and NAPS measurements for NO and NO₂ often do not agree well (Fig. 8). The disagreements can again be partially explained by the influence of wind direction. Like CO and NH₃, NO is directly emitted from vehicles, but it reacts in the atmosphere much more quickly than CO or NH₃. Polar plots for NO and NO₂ (Figs. 7b and 7c) show the effect of wind direction on the mixing-ratio differences between GEM-MACH results and NAPS measurements. -When the wind blew from the NAPS trailer towards the highway, the difference was close to zero, but when the wind blew across the highway towards the trailer, GEM-MACH predictions were significantly lower than NAPS measurements. Similar to the CO comparison, the NAPS measurements were strongly influenced by traffic emissions when the wind came from highway compared to GEM-MACH. Note that GEM-MACH simulates mean pollutant mixing ratios within a 40-m layer ~~whereas the FTIR path was located about 8 m above the highway pavement~~ and the inlet of the NAPS trailer was about 3 m above the ground. These different heights also contribute to the disagreement between measurements and model results.

The NO_x (nitrogen oxides, NO_x = NO+NO₂) mixing ratio measured at the NAPS station on weekdays showed a similar average diurnal cycle (Fig. 9) to CO by the FTIR, reaching a peak over 100 ppb from 6:00 to 8:00 followed by significant decrease in the middle of the [day](#) and a secondary peak between 20:00 and 23:00. The diurnal cycle of NO_x on weekends with mixing ratios of 0-35 ppb over the whole day, significantly lower than weekday NO_x levels, was also less variable. Reduced NO_x levels on weekends may have been due to fewer diesel vehicles operating on weekends; this pattern has been reported in studies in California (Marr and Harley, 2002a; Harley et al., 2005; Kim et al., 2016). Zhang et al. (2012) found that fewer diesel vehicles were observed on weekends on another major highway in the Toronto area. The annual sales of fuel used for on-road motor vehicles in Canada in 2015 were 42.6 billion litres of gasoline and 18.0 billion litres of diesel (Statistics Canada 2016), i.e., a significant fraction of fuel burned is diesel. Therefore, lower diesel vehicle volumes on weekends may have contributed to different emissions of NO_x on Highway 401 near our site. The [NO_x] / [CO] ratio also has been used to check the chemical conditions related to O₃ production. Figure 10 shows that the ratios in this study are 0.20 and 0.10 for weekdays and weekends, respectively. The lower ratio on weekends is likely due to reduced numbers of diesel vehicles, which is consistent with a previous study by Kim et al. (2016). Our [NO_x] / [CO] ratios during both weekdays and weekends are greater than their results (0.11 and 0.033), but our study only focused on near-surface observations over a short defined section of Highway 401 while their observations were at 1 km above the ground level with a bigger footprint which included off-road emissions and other local sources. [Hassler et al. \(2016\) showed the trend of \[NO_x\] / \[CO\] in the Los Angeles Basin, and the ratio is between 0.1 and 0.2 after 2010, which agrees well with our results. There are also some previous studies showing ratio of \[CO\] / \[NO_x\] from regions near heavy traffic emission. Parrish et al. \(2012\) reported the slope of \[CO\] vs. \[NO_x\] was in the range of 6.3 to 18.9 for the measurement from 1987 to 1999. Wallace et al. \(2012\) reported the slope of \[CO\] vs. \[NO_x\] was 4.2 in the morning rush hours in 2009. The slopes of \[CO\] vs. \[NO_x\] in this study are 3.14 and 7.75 for weekdays and weekends, respectively. Therefore our results on NO_x and CO are comparable with these previous studies.](#)

Figure 9 also shows average weekday and weekend diurnal cycles for O_3 measured at the NAPS station. One interesting feature is that the median diurnal O_3 mixing ratios on weekends were consistently greater than on weekdays. Also, the diurnal cycles of O_3 were inversely correlated with those for NO_x . The low O_3 mixing ratios in the mornings of weekdays can be explained by titration with high fresh emissions of NO from traffic, whereas the afternoon maximum is mainly due to production of O_3 through increased levels of photochemistry with VOCs. The diurnal cycle of odd oxygen ($O_x = O_3 + NO_2$) shown in Fig. 9 can be used to separate the contributions of titration and photochemistry with VOCs to O_3 mixing ratios. Titration does not increase the sum of O_3 and NO_2 , whereas photochemistry with VOCs does. Therefore, variations in O_x levels indicate that photochemistry with VOCs is important (Pollack et al., 2012). The average diurnal mixing ratios of O_3 and O_x from NAPS measurements showing a maximum in the afternoon and being slightly higher on the weekends is also consistent with the average diurnal mixing ratios of HCHO showing a peak in the afternoon on weekends (Fig. 11). These diurnal results also suggest that the photochemistry with VOCs producing NO_2 and O_3 was important especially in the afternoon and on weekends. In addition, O_x levels peaked in the afternoon, also consistent with diurnal cycles of sunlight intensity (Fig. 2) which is a critical condition of photochemistry to produce O_3 . Other VOCs besides HCHO that were emitted by traffic and other local sources may also have contributed to the photochemical production of O_3 , but they were not quantified. Similar differences of O_3 and O_x mixing ratios between weekdays and weekends were reported in the South Coast air basin (Pollack et al., 2012; Warneke et al., 2013). Temperature also affects O_3 production, but given the great variation of O_3 mixing ratio through the day, this was a secondary effect here based on box model calculations in the temperature range of our study (Coates et al., 2016).

To evaluate how representative the contrasts between weekends and weekdays based on this 16-day data set are compared to longer timeframes, 3 summers of O_3 measurements from a nearby NAPS station in Toronto West were extracted and analyzed (Supplementary material Section 3 and Fig. S7). Similar diurnal patterns and differences were observed in 2 of the 3 years, suggesting that the analysis presented above is representative of longer terms as well.

Time series of HCHO mixing ratios retrieved from the FTIR shown in Fig. 6 were between 0 and 5 ppb. The average diurnal cycle of HCHO during weekends reached a peak in the early afternoon (shown in Fig. 11), which did not correspond to the average diurnal cycles of either traffic volume or primary pollutant CO, but rather to the sunlight intensity (i.e., actinic radiation: see Fig. 2). This indicates that photochemistry of VOCs with oxidants was a dominant source (Stroud et al., 2016). The lifetime of HCHO in the atmosphere is on the scale of hours to days depending on the levels of ambient oxidants (Seinfeld and Pandis, 2006). Stroud et al. (2016) reported on levels of HCHO in Toronto and Egbert, Ontario and source apportionment. Primary mobile emissions were found to contribute ~ 12% of HCHO in Toronto. Previous research also showed that both light-duty and heavy-duty vehicles emit HCHO (Grosjean et al., 2001).

The correlation between HCHO mixing ratio and ambient temperature was moderate (Fig. S6a, $r^2 = 0.42$). HCHO levels were low for a few weekdays (July 22-24) with lower temperature as compared to the four warmer days sampled on the weekends.

Therefore, the difference between the average weekday and weekend diurnal cycles shown in Fig. 11 may be due in part to sample size, and it is possible that other local HCHO emission sources which depend on the temperature may also have contributed to the HCHO observed, especially in the afternoons on weekends.

GEM-MACH simulations of HCHO mixing ratio are always greater than the FTIR measurements (~~Fig. 6~~), but in the GEM-MACH HCHO model species from the ADOM-II gas-phase chemistry mechanism is actually a lumped species that also includes isoprene oxidation products. [Therefore, the GEM-MACH results of HCHO are not shown here.](#)

3.5 HCN

The mixing ratios of HCN retrieved from FTIR measurements were between 0 and 4 ppb (Fig. 6). [Only on July 28, 29 and 30th the HCN was observed above its detection limit.](#) HCN has severe adverse effects on human health, and chronic exposure to low cyanide can cause abnormal thyroid function and neurological problems (El Ghawabi et al., 1975; Blanc et al., 1985; Banerjee et al., 1997; U.S. EPA, 2010). HCN has been reported previously in vehicle exhaust (Bradow and Stump, 1977; Keirns and Holt, 1978; Cadle et al., 1979; Urban and Garbe, 1979, 1980; Karlsson, 2004; Baum et al., 2007; Moussa et al. 2016). It may form over the catalytic converters in the vehicle emission control systems (Voorhoeve et al., 1975; Suárez and Löffler, 1986; Baum et al., 2007). A recent study for Toronto reported comparable HCN mixing ratio values (Moussa et al., 2016). These HCN measurements contribute to the few studies reporting measurements of HCN mixing ratio beside a highway in urban ambient air.

3.6 CH₃OH

As shown in Fig. 6, mixing ratios of CH₃OH from the FTIR were between 2 and 20 ppb most of the time, with some high spikes. Figure 12 presents the corresponding average weekday and weekend diurnal cycles of CH₃OH for the study period.

This plot shows the mixing ratio reached a peak (maximum of 20 ppb at 7:30) from 7:00 to 9:00 on weekdays whereas ~~it there~~ was [generally flat no peak in the mornings](#) on weekends. [In addition, a linear relationship between \[CH₃OH\] and \[CO\] was observed during the early morning rush hours on some weekdays \(Table 2\), indicating](#) These results suggest that [at least a large component fraction of observed CH₃OH may have come was](#) from traffic emissions. Observations of methanol associated with traffic have been reported in other studies. Rogers et al. (2006) reported CH₃OH in the diluted pipeline exhaust of a mobile laboratory. CH₃OH may also come from non-engine sources, such as windshield wiper fluid. Durant et al. (2010) measured gas and particle pollutants near Interstate 93 in Massachusetts. They reported CH₃OH was above 20 ppb at 7:20 50 m downwind of the highway, possibly with contributions from some other local sources. [Reyes et al. \(2006\) reported vehicle emission of non-regulated pollutants, including methanol, by using local gasoline and driving conditions in Mexico City. Rantala et al. \(2016\) studied urban VOC fluxes in urban Helsinki and found methanol fluxes were correlated with traffic and with CO fluxes, traffic could partially explain the observed methanol. Sahu and Saxena \(2015\) also reported CH₃OH mixing ratios at Ahmedabad \(an](#)

[urban site in India](#)), and both traffic emission and the transport from biomass burning and biogenic sources outside the city contributed to CH₃OH. The mixing ratio of CH₃OH we observed did not correlate with ambient temperature (Fig. S6c), so there was no strong indication of biogenic sources.

3.7 Estimation of emission factors

To evaluate the feasibility of using measurement data from this study to estimate emission rates, we picked measurements for three days (July 22, 28 and 29) from this study to use as inputs to a backward Lagrangian stochastic dispersion model (WindTrax, <http://www.thunderbeachscientific.com/>). July 22 was chosen, because the wind direction was steadily from northwest and a traffic jam occurred for added interest. July 28 and 29 were chosen, because they are two of the highest days for temperature and O₃ during this project. -The following were included as the inputs: CO mixing ratio from the FTIR; background mixing ratio of CO; winds and temperature from the NAPS trailer; and atmospheric stability (u_z and L) from the scintillometer. -The surface roughness (z_0) was set to 0.15 m, the maximum allowed by WindTrax. The defined section of Highway 401, which was assumed to be the only CO source in the footprint, is about 1870 m long and 110 m wide. The FTIR path is roughly in the centre of the defined section. WindTrax was then used to estimate the emission rate of CO from this defined section. In the model, 50000 virtual particle trajectories were calculated upwind of the FTIR path with the given meteorological conditions (wind direction and temperature) and surface-layer turbulence (u_z and L), to determine what fraction of trajectories originated from the designated source area.

Over three days, July 22, 28, and 29, CO emission rate estimates ($\text{grams hour}^{-1} \text{m}^{-2}$) were calculated by ~~the~~ WindTrax with a one-minute resolution for ten-minute periods, and the average estimates over those ten-minute periods were calculated and shown as the markers in Figure 13. The constant background used in WindTrax was 265 ppb, which was determined from the CO intercept of the linear regression analysis of NH₃ with CO (see Section 3.3). The mixing ratio of changing background used in WindTrax was determined using a more dynamic definition of background based on wind direction. When the wind was from the south, the background was chosen as the NAPS measurement for July 28 and 29. In the morning on July 28 when the wind was from the northwest, the background was chosen as 415 ppb, the minimum mixing ratio of that morning measured by the FTIR. When the wind direction varied greatly, the background value of the previous hour was chosen. On July 22, the wind was consistently from the northwest, and the minimum of 329 ppb over the whole day from the FTIR was chosen as the changing background.

Liu and Frey (2015) reported vehicular [empirical cycle average emission factor for specific pollutant, vehicle, and driving cycle in grams per mile as well as average, minimum and maximum values, based on empirical data measured between 2008 and 2013 in the Raleigh and Research Triangle Park area \(North Carolina, U.S.\) for 100 vehicles with a range of model years and accumulated mileage. We converted these results into \$\text{g km}^{-1}\$ emission factors ranging from 0.003 to 5.1 \$\text{g km}^{-1}\$ with an average value of 0.62 \$\text{g km}^{-1}\$ based on empirical data measured between 2008 and 2013 in the Raleigh and Research Triangle Park area](#)

530 ~~(North Carolina, U.S.) for 100 vehicles with a range of model years and accumulated mileage.~~ They also reported simulated CO emission factors ranging from 0.004 to 6.87 g km⁻¹ with an average value of 1.99 g km⁻¹ by using the U.S. EPA's Motor Vehicle Emission Simulator (MOVES) emission factor model. Moussa et al. (2016) reported that emission factors of CO measured from several gasoline light duty vehicles with different driving cycles ranged from 0.1 to 3.0 g km⁻¹ with an average value of 0.9 g km⁻¹.

Using emission factors from the MOVES model, traffic volume estimates, and the width of Highway 401 at the site, a "bottom-up" estimate of the emission rate (grams hour⁻¹ m⁻²) was calculated and compared to the dispersion model results (Fig. 13).

535 There is a good agreement amongst the different approaches ~~(-140% to 460%, and -110% to 70%, see Fig. 13 and discussion below)~~. On July 22, the emission rate estimates from the WindTrax are close to the MOVES results. The MOVES estimates show a sharp drop between 14:00 and 15:30 due to decrease of traffic volume to 14 % of at the value at 13:30. During this time, the emission rate estimates from the WindTrax do not fluctuate much. This result suggests that vehicle numbers passing by a fixed point may not be the best indicator of emissions since they do not account for the traffic speed: an extreme example would be a traffic jam with zero traffic flow but nonzero emissions. On July 28 and 29, the estimates from the WindTrax are greater than the MOVES estimates. The difference between the emission rates estimated by using the WindTrax with changing background and by using 0.9 g km⁻¹ and 3.0 g km⁻¹ from Moussa et al. (2016) is in the range of -140 % to 460 %, and -110 % to 70 %, respectively. These results suggest that WindTrax dispersion calculation results based on CO mixing ratio measurements from the FTIR and in-situ micrometeorology are well within the range of estimates based on traffic volume and emission factors of various vehicles.

545 The input background mixing ratio assumed at each hour influences the emission rate estimates. Especially during the night around 0:00 to 6:00 when the wind was from north, it is difficult to determine the background mixing ratios of CO, since no measurements were available upwind. Both emission rate estimates with constant background and changing background were calculated to check the sensitivity of background mixing ratio on the emission rate estimates by the WindTrax. Our changing-background approach should be closer to reality, since conditions around the highway do change with time. Figure 13 shows that both estimates from the WindTrax using changing background and constant background agree in general with the bottom-up estimates, except for the period of the morning on July 29 when WindTrax estimates with the constant background are greater than the other estimates. The wind was from the south during this period and the assumed constant background of 265 ppb was significantly lower than the NAPS measurements, resulting in an overestimate by the WindTrax. Beside the uncertainties in background mixing ratio, the variations in emission rate estimates are most likely due to changes in wind direction over short periods. When the wind direction changed quickly, the input wind direction used by the WindTrax may not be representative and hence may bias the calculated emission rate.

Emission rates of other primary pollutants from traffic can be determined by using the concentration ratios of these pollutants to CO and emission rate estimates of CO, as mentioned in Section 3.2.1. We found that [NH₃] (as discussed in Section 3.3), [CH₃OH], and early morning periods of [NO] had linear relationships with [CO]. With these ratios and emission rate estimates of CO obtained from the WindTrax (Figure 13), the ten-minute average emission rate estimates of NH₃, NO, and CH₃OH were calculated. The minimum-to-maximum range and the average of these ten-minute average estimates are shown in Table 2.

[Emission factors in g km⁻¹ were calculated from WindTrax emission rates estimates by using the 110 m width of that section of highway and traffic volume, to compare our estimates to previously reported emission factors.](#)

The inputs of u_s and L from the scintillometer measurements also contain uncertainties. A realistic estimate of the uncertainty of u_s is $\pm 20\%$ to 30% (Andreas, 1992). We conducted a sensitivity study by varying u_s from 0.7 to $1.3 u_{s(\text{obs})}$ and L from 2.20 to $0.34 L_{(\text{obs})}$ corresponding to change of u_s while keeping heat flux fixed, which resulted in emission rates from 0.69 to 1.37 times the original emission rate estimates. We also investigated the sensitivity of heat flux on emission rates estimates by varying L from 0.5 to $2L_{(\text{obs})}$ with fixed $u_{s(\text{obs})}$, which resulted in emission rates estimates are in [the](#) range of 0.70 to 1.45 times [of the](#) original emission rate estimates. Therefore, even with conservative uncertainty estimates about surface-layer stability, the calculated emission rates from the WindTrax are still within 45 % of the bottom-up estimates.

WindTrax limits z_0 to a maximum value of 0.15 m. However, the z_0 of the actual measurement site over the highway is around 0.6 m based on urban-scale meteorological model results for Toronto (Leroyer et al., 2016). This difference between the actual z_0 and the input z_0 used by WindTrax likely also contributes to the uncertainty of the emission rate estimates.

Better emission-rate estimates might also be obtained if traffic information included vehicle types and vehicle speed. Speed and speed variation of different vehicle types are known to affect emission rates of CO, hydrocarbon and NO_x (Zhang et al., 2011), but consideration of these factors would have required more sophisticated traffic quantification.

4. Summary and Conclusions

This study demonstrated the utility of combining long-path FTIR spectroscopy with micrometeorological measurements to identify and quantify pollutants emitted by moving traffic, and to calculate emission rates in a representative real-world setting. We retrieved mixing ratios of eight air pollutants over Highway 401 in Toronto, Canada. Traffic emissions were shown to contribute quantifiable levels of NH₃, HCN, HCHO, and CH₃OH to the urban mix of pollutants. Of particular interest was the quantification of species such as HCN, a toxic pollutant with severe health implications, and NH₃, which may be gaining in importance due to the increasing use of catalytic converters which reduce vehicular NO_x emissions. Very few ambient data sets on these species from traffic-dominated environments are available in the published literature, and the methods described here can fill a significant gap.

Differences between weekdays and weekends in the average diurnal cycles of some of the pollutants mixing ratios (CO, NO_x, O₃, NH₃, and HCHO) were observed. The biggest differences are that on weekdays, the mixing ratios of primary pollutants from traffic, such as CO and NH₃, showed an obvious peak in the early morning around 6:00 to 9:00, corresponding to the sharp increase of traffic volume during morning rush hour, while on weekends, mixing ratios varied less throughout the day and no obvious peaks in the early morning were observed. Combined FTIR analysis and turbulence results clearly elucidated the role of turbulence in the build-up and dispersion of traffic emissions.

A comparison of the path-averaged FTIR with single-point NAPS measurements showed general agreement of the variations in mixing ratio, but also showed differences due to the difference in measurement footprint. This comparison also uncovered some issues with offsets and amplitude differences between the FTIR and in-situ analyzers that are likely due to pervasive H₂O interference across the FTIR spectrum, especially for NO and NO₂.

The modelled pollutant concentrations at the study site from a high-resolution version of the GEM-MACH air quality model agreed well in general with the measurements, especially for CO, O₃, and NH₃. Given that the version of GEM-MACH considered here employed 2.5 km by 2.5 km grid cells, model results and measurement results are not expected to be directly comparable for all wind regimes, and comparisons can be better explained after separating wind directions, but reasonable correlations were observed.

Lastly, by combining mixing ratio with micrometeorological measurements and a simple dispersion model, we demonstrated the calculation of real-world, spatially representative vehicular emission rates using CO as an example, and derived emission rates of NH₃, NO and CH₃OH.

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Tables

Table 1. Regions of long-path FTIR spectra used to retrieve mixing ratios of target gases in this study.

| Gases of Interest | Spectral Region Fitted (cm ⁻¹) | Interference Gases Fitted | Correlation Threshold (r) ^a | Detection Limit (ppb) ^b | Reference for Spectral Region Fitted (cm ⁻¹) |
|---|--|---|--|------------------------------------|--|
| CO (Carbon monoxide) | 2142-2241 | H ₂ O, CO ₂ , N ₂ O | 0.7 | 0.7 0.8 | Smith et al. (2011) |
| CO ₂ (Carbon dioxide) | 2224-2255 | H ₂ O, N ₂ O, CO | 0.97 | | Griffith (1996) |
| CH ₄ (Methane) | 2904-3024 | H ₂ O | 0.95 | 0.70 0.8 | |
| O ₃ (Ozone) | 1040-1065 | H ₂ O, NH ₃ , CH ₃ OH, benzene, HCHO | 0.47 | 4.44.9 | |
| NO (Nitrogen oxide) | 1893-1913 | H ₂ O | 0.1 | 5.86 5 | |
| NO ₂ (Nitrogen dioxide) | 1595-1607 | H ₂ O, NH ₃ , CH ₃ OH | 0 | 7.38 2 | |
| SO ₂ (sulfur dioxide) | 2465-2550 | H ₂ O, N ₂ O | 0.5 | 1.51 7 | |
| CH ₃ OH (Methanol) | 980-1080 | H ₂ O, NH ₃ , O ₃ | 0.7 | 0.70 0.8 | |
| NH ₃ (Ammonia) | 910-990 | H ₂ O | 0.7 | 0.70 0.8 | Smith et al. (2014) |
| HCN (Hydrogen cyanide) | 710-717 | H ₂ O, N ₂ O, CO ₂ , C ₂ H ₂ , NH ₃ , NO ₂ | 0.3 | 2.93 2 | Akagi et al. (2014) |
| HCHO (Formaldehyde) | 2740-2840 | H ₂ O, CO ₂ , CH ₄ | 0.3 | 1.51 7 | Akagi et al. (2014) |
| N ₂ O (Nitrous oxide) | 2198-2223 | H ₂ O, CO ₂ , CO | 0.97 | 0.50 6 | Griffith (1996) |
| CH ₃ (CO)CH ₃ Acetone | 870-940 | H ₂ O, NH ₃ , C ₂ H ₆ , C ₂ H ₄ | 0.3 | 2.22 5 | |
| C ₂ H ₂ (Acetylene) | 680-780 | H ₂ O, CO ₂ | 0.3 | 0.60 7 | |
| C ₂ H ₆ (Ethane) | 800-850 | H ₂ O, CO ₂ | 0 | 1.41 6 | |
| C ₃ H ₈ (Propane) | 2860-2975 | H ₂ O, CH ₄ , C ₂ H ₆ , HCHO, NO ₂ | 0 | 0.70 0.8 | |

^a Correlation thresholds are inputs for OPUS_RS used when retrieving the mixing ratios from FTIR spectra. When the correlation between the measured spectrum and reference spectrum in that spectral range is below this threshold, that pollutant is not “identified” and the mixing ratio will be reported as zero.

^b Detection limit is [calculated by converting](#) 3σ of the noise for measurements with a retroreflector distance of [310-225 m](#) ~~according to Bruker~~ [to \$3\sigma\$ of the noise with 310 m in our setup, and then includes the dependence of the signal intensity on the distance between the spectrometer and retro-reflector, which is 12% in this case.](#)

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Table 2. Pollutant emission rates

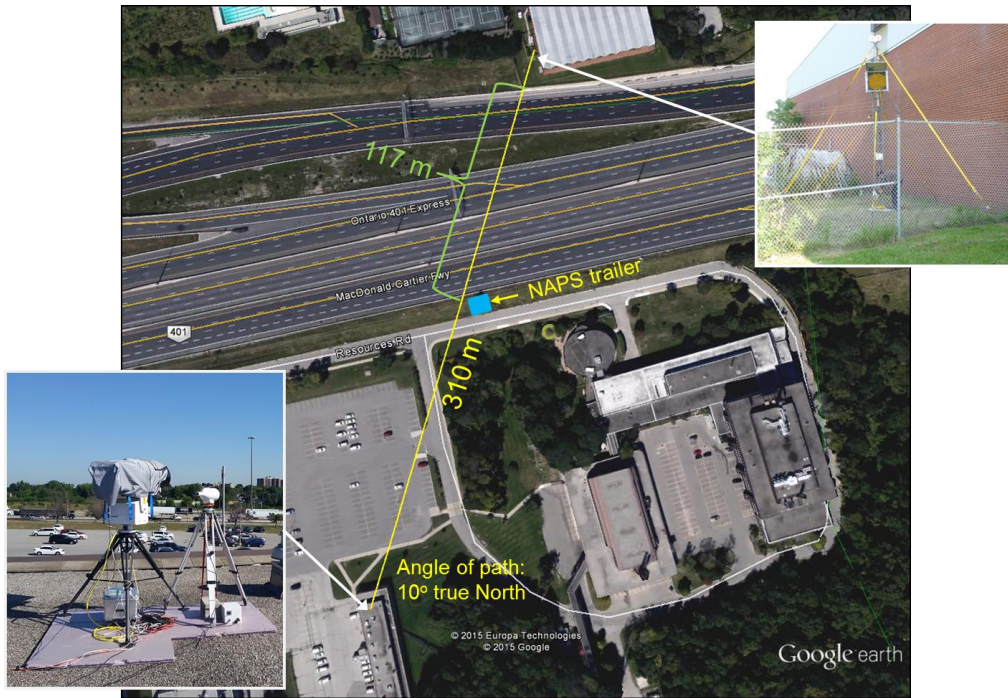
| Pollutant | [pollutant]/ [CO] (ppbv/ppbv) | Emission rates (g m ⁻² h ⁻¹) | Emission factors (average) (g km ⁻¹) | Emission factors (g km ⁻¹) previously reported |
|---------------------------------|-------------------------------------|--|---|---|
| CO | 1 | 0-0.90 | 0-6.97 (2.6) | 0.004-6.84 (Liu and Frey 2015) 0.1-3.0 (Moussa et al. 2016) |
| NH ₃ | ^a 0.026 | 0-0.01 | 0-0.11 (0.04) | 0-0.11 (Durbin et al. 2002) 0-0.144 (Huai et al. 2003) 0-0.26 (Livingston et al. 2009) 0.004-0.062 (Suarez-Bertoa et al. 2014) |
| NO | ^b 0.128 | 0-0.12 | 0-0.96 (0.36) | 0.008-1.26 Frey et al. (2003) |
| ^c CH ₃ OH | ^c 0.051 | 0-0.05 | 0-0.41 (0.15) | N/A 0.0015-0.0067 Reyes et al. (2006) |

^a The ratio was found as the slope of the linear fit over the entire study period.

^b The ratio was found as the average slope of the linear fits for the three early morning periods on July 22, 28 and 29.

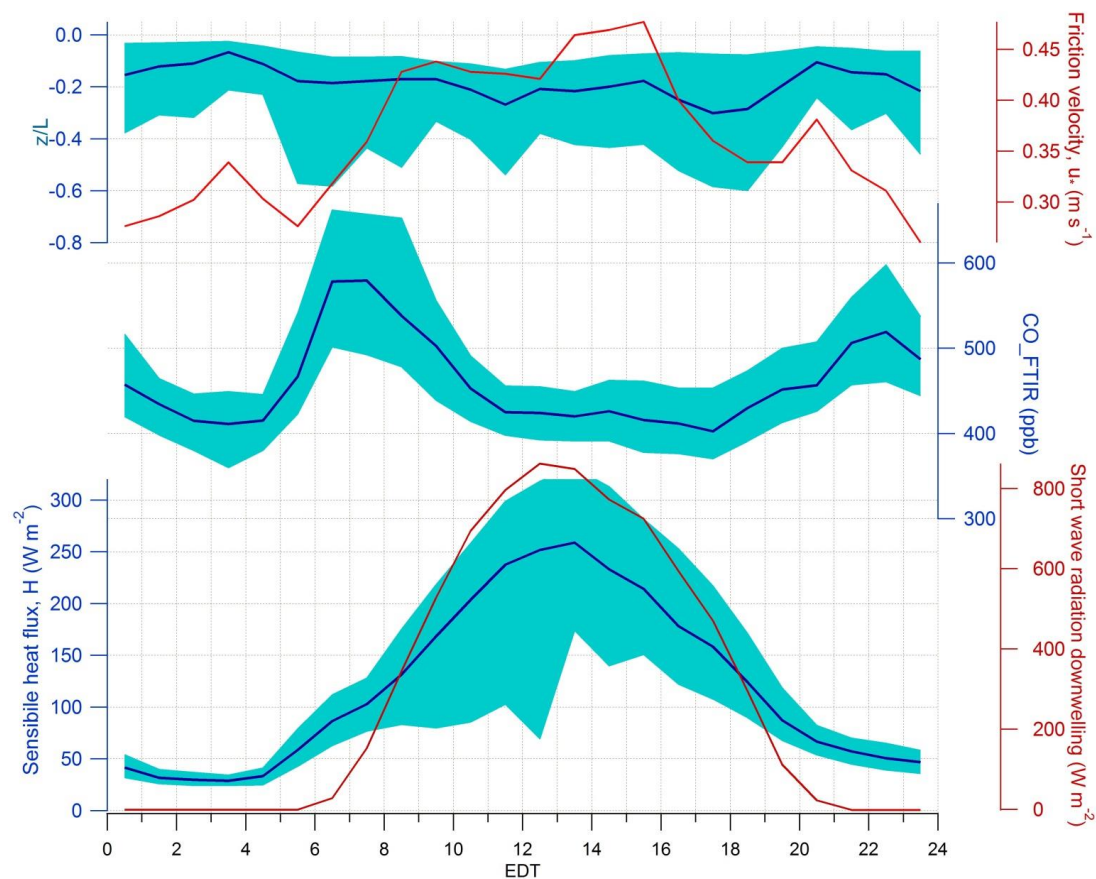
^c The ratio was found as the slope of the linear fit for July 28 and 29.

Figures



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Figure 1: Setup of the FTIR, scintillometer (see Section 2.2), and the NAPS trailer near Highway 401.



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Figure 2: Average diurnal cycles (one-hour averages) of z/L and u_* (top), CO mixing ratio from the FTIR (middle), as well as sensible heat flux H and downwelling shortwave radiation (bottom) for 16-31 July 2015. Lines represent the medians, and the shaded regions are the interquartile ranges for z/L , CO, and H .

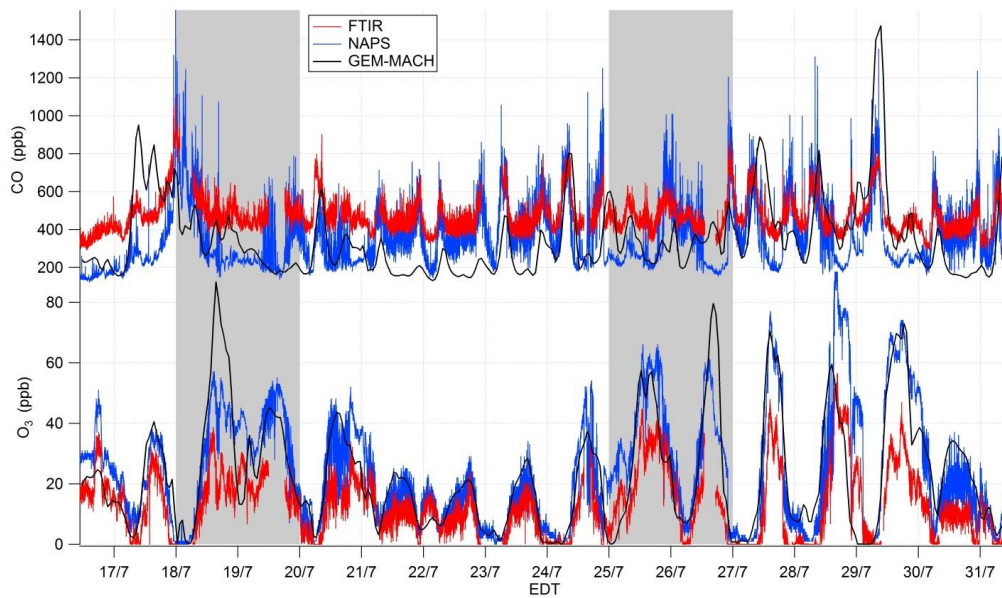
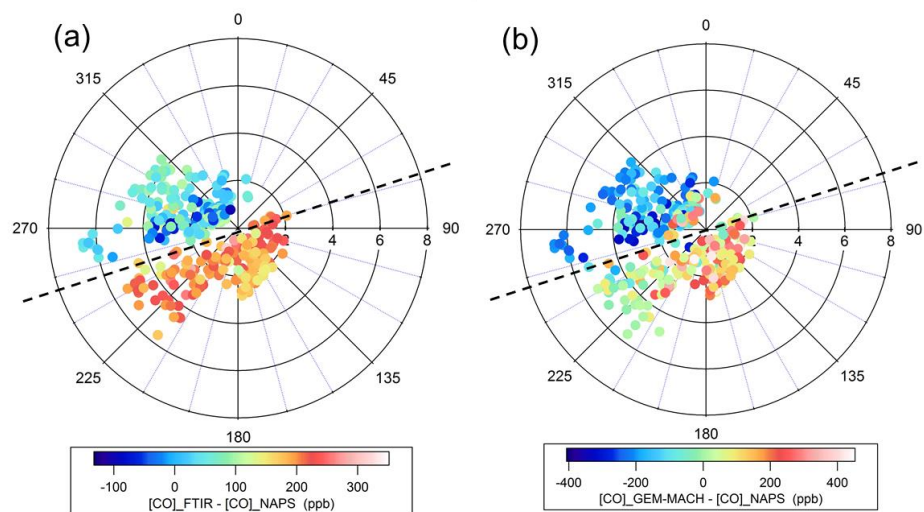


Figure 3: Time series of mixing ratios of CO (top) and O₃ (bottom) for the full study period. The red traces are mixing ratios retrieved from the FTIR spectra. The blue traces are measurements from the NAPS station. The black traces are output from GEM-MACH. Grey shaded areas highlight the weekend periods.

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Figure 4: Polar plots of CO mixing-ratio difference between measurements from the FTIR and NAPS (a), between GEM-MACH output and NAPS measurements (b). Azimuth angle represents wind direction (meteorological convention: 0° = wind from north, 90° = wind from east, etc.), and radius indicates wind speed (m s^{-1}). The color shows the CO mixing ratio difference. The center corresponds to the location of the NAPS trailer. The black dashed line shows the orientation of the highway: above this line, the wind came from the highway towards the trailer.

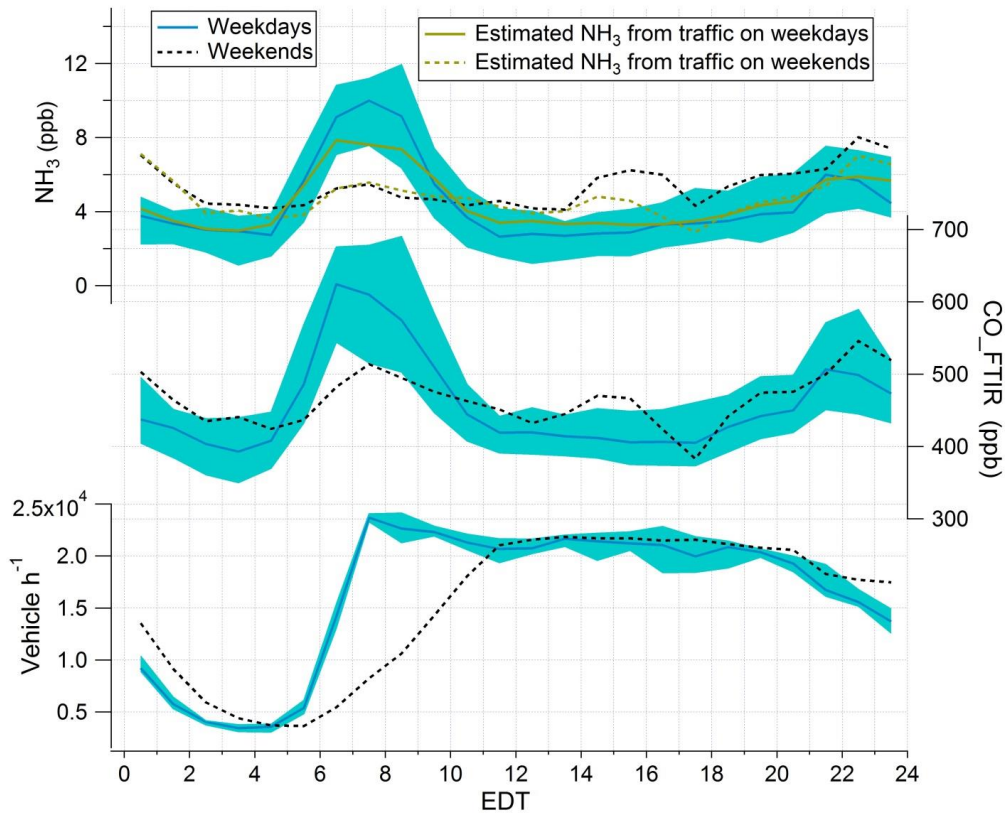
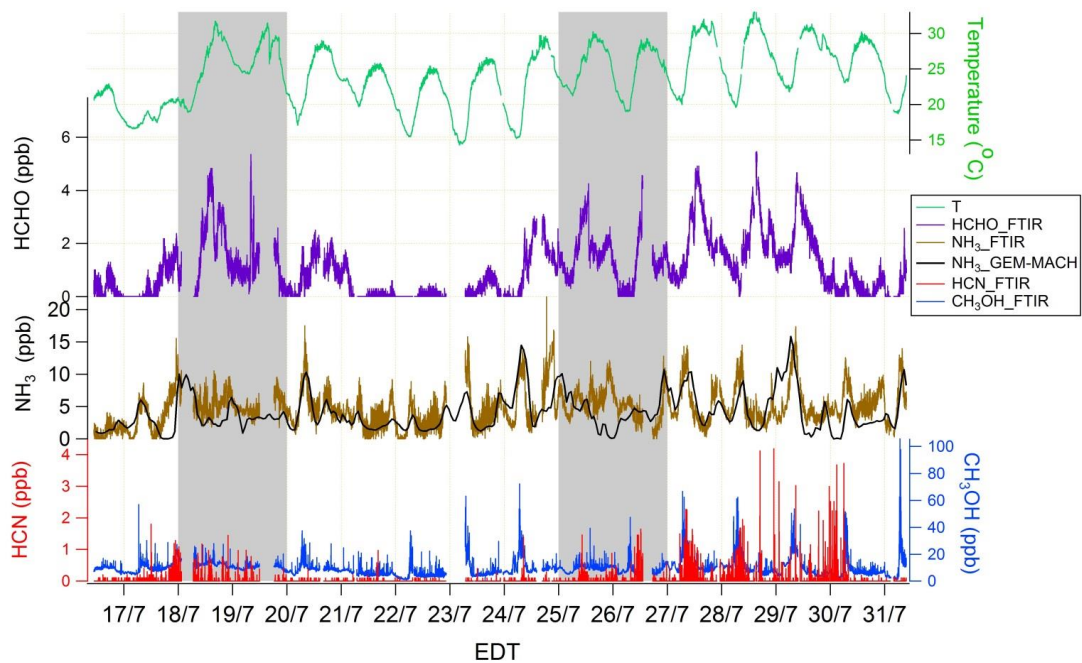


Figure 5: Average weekday and weekend diurnal cycles of mixing ratio of NH_3 (top) and CO (middle) from the FTIR and traffic volume (bottom) for the 16-day study period. Blue solid lines are medians, and the shaded areas show the interquartile ranges for weekdays; black dashed lines are the medians for weekends. The brown solid line is an estimation of NH_3 levels associated with traffic emissions on weekdays; the brown dashed line corresponds to weekends.

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Figure 6: Time series of ambient temperature, mixing ratio of HCHO, NH₃, HCN, and CH₃OH for the 16-day study period. Grey shaded areas indicate the weekend periods.

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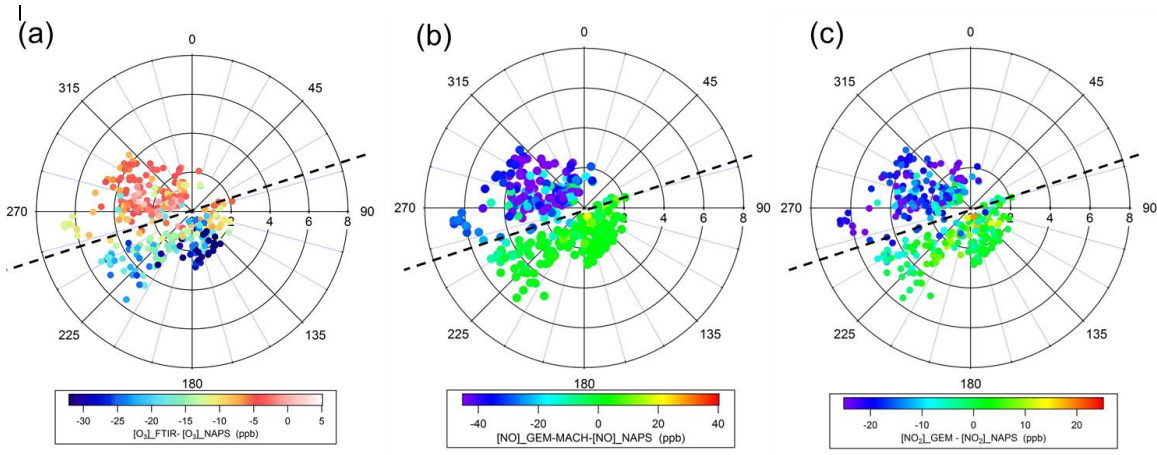


Figure 7: Polar plots of O_3 mixing ratio difference between measurements from the FTIR and the NAPS (a); mixing-ratio difference between results from GEM-MACH predictions and hourly averaged measurements from NAPS for NO (b) and NO_2 (c). Azimuth angle represents wind direction (meteorological convention), and radius indicates wind speed (m s^{-1}). The center of each plot corresponds to the location of the NAPS trailer. The black dashed line shows the orientation of the highway: above this line, the wind came from across the highway towards the trailer.

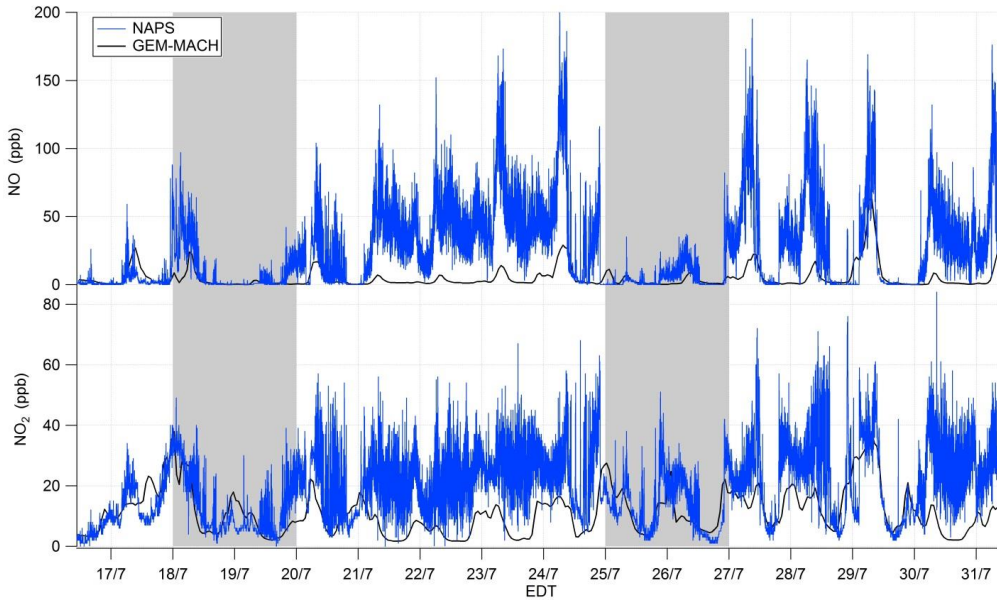
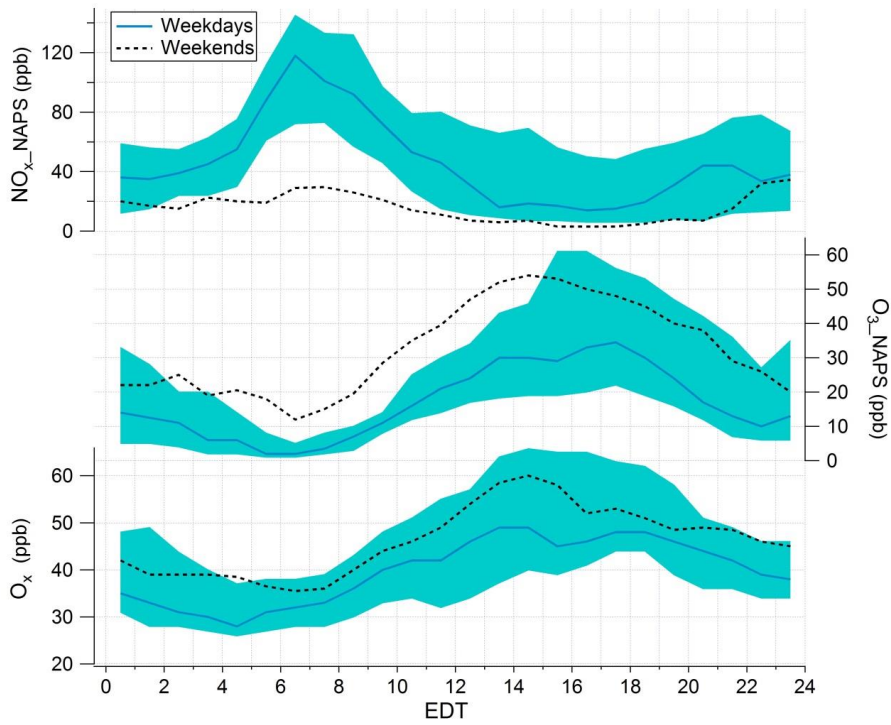
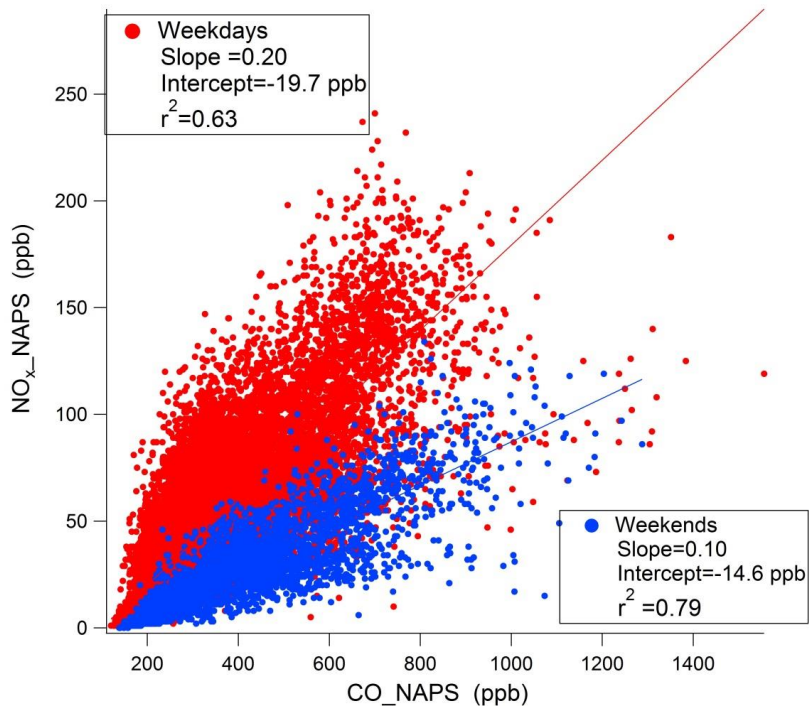


Figure 8: Time series of mixing ratios of NO (top) and NO₂ (bottom). Grey shaded areas indicate the weekend periods.



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Figure 9: Average weekday and weekend diurnal cycles of mixing ratios of NO_x (top), O_3 (middle), and O_x (bottom) from the NAPS for the 16-day study period. Solid green lines are medians and the shaded areas are the interquartile ranges on weekdays; dashed black lines are medians on weekends.



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Figure 10: Scatterplot of NO_x vs. CO mixing ratios from the NAPS on weekdays (red) and weekends (blue). Lines are the linear regression results for weekdays (red) and weekends (blue).

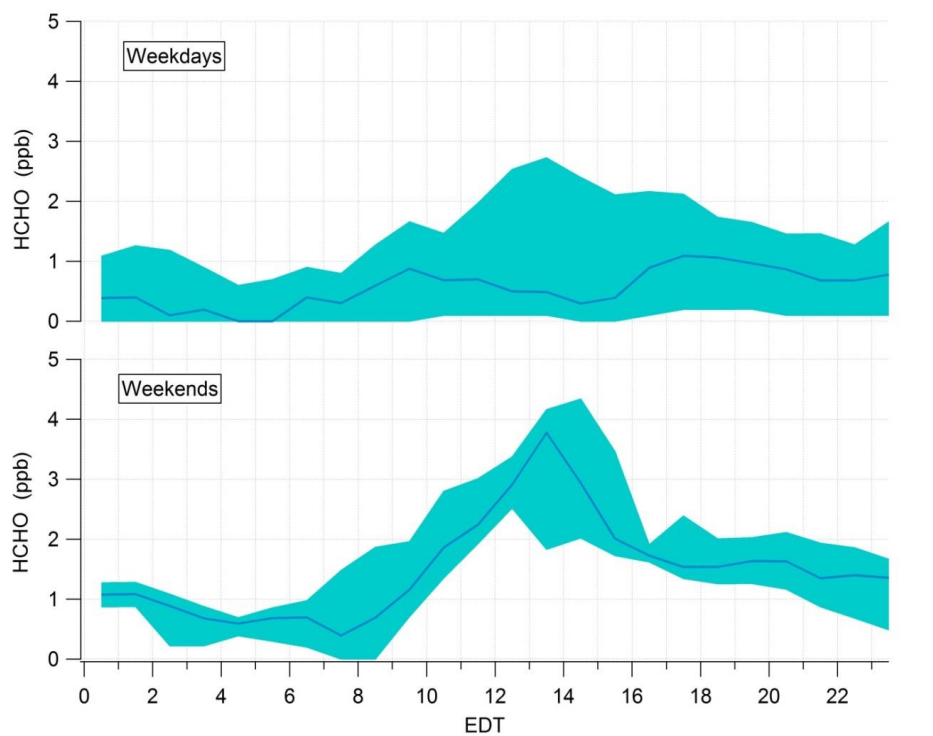


Figure 11: Average diurnal cycles of HCHO on weekdays (top), and on weekends (bottom) for the 16-day study period. Solid green lines are medians, and the shaded areas are the interquartile ranges.

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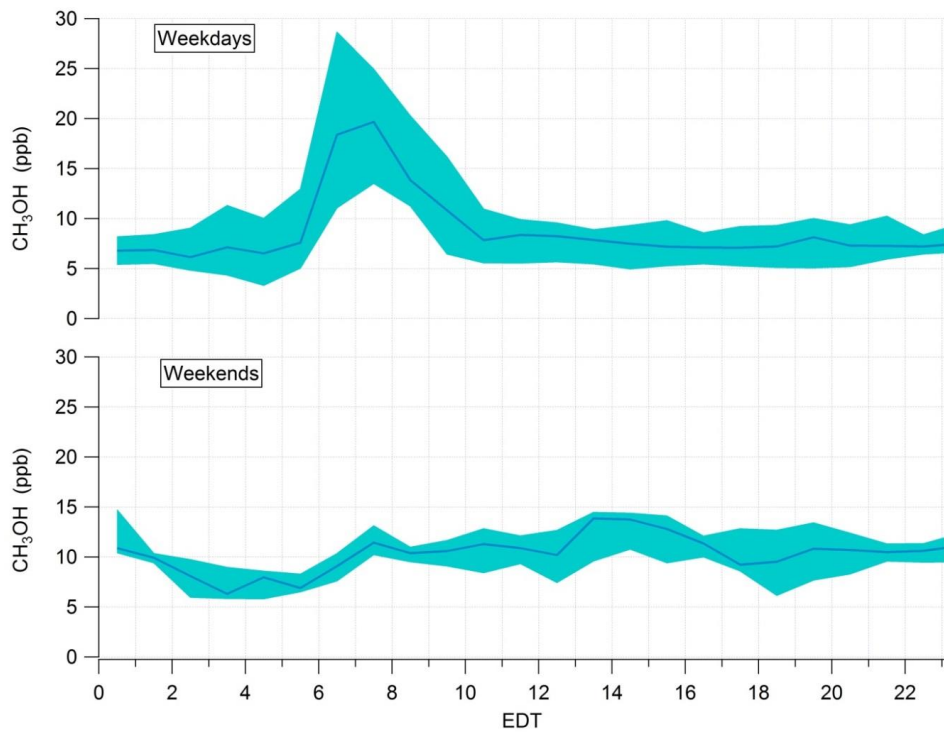


Figure 12: Average diurnal cycles of CH_3OH mixing ratio on weekdays (top) and weekends (bottom) for the 16-day study period. Solid green lines are medians, and the shaded areas are the interquartile ranges.

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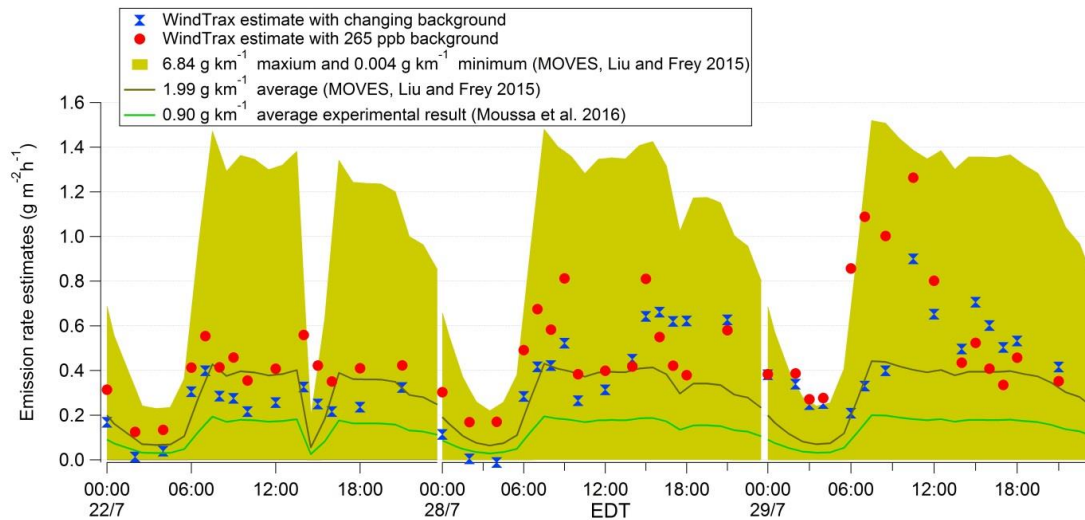


Figure 13: CO emission rate estimates over three days. Red dots are CO emission rates simulated by WindTrax using CO mixing ratios from the FTIR and a constant CO background of 265 ppb (see the text). Blue markers are CO emission rates simulated by the WindTrax using changing CO background values. The brown line is the CO emission rate estimated by using traffic volume estimates and emission factors from the average MOVES results in Liu and Frey (2015); the brown shade is the range of CO emission rates estimates obtained by using the maximum and minimum CO emission factor results from MOVES in Liu and Frey (2015). The green line is the CO emission rate simulated by using traffic volume estimates and the average CO emission factor from Moussa et al. (2016).

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