

# **(1) comments from Referees**

## **Anonymous Referee #1**

**Received and published: 19 May 2017**

### **General comment**

This paper reports an analysis of PM<sub>2.5</sub> and PM<sub>10</sub> chemical compositions in Sao Paulo (Brazil). The information is used for source apportionment using PMF receptor model and to investigate some atmospheric processes. The approach is not particularly new, however, the chemical investigation is very detailed and used to interpret some processes involving fine and coarse fractions of aerosol. Therefore, I believe that it would be interesting for readers and scientist. There are some parts of the paper that are not completely clear and interpretations not fully discussed (see my specific comments). In conclusion, I suggest to consider the paper for publication only after a revision that addresses all my specific comments.

### **Specific comments**

- 1- Number of chapters and subchapters are inconsistent.
- 2- It would be better to mention that also metals are analysed in the abstract otherwise the reader should wait several pages before to understand that also some metals are investigated.
- 3- Section 1.2. The height of the back-trajectories used start at 500 m that are not exactly near the ground. Why a lower starting point has not been chosen?
- 4- Section 1.4 (line 180). It is reported that PMF was applied to PM<sub>10</sub> but I understand that all data including PM<sub>2.5</sub> were used together. Please correct this incongruence.
- 5- Equation (3). The symbol EC was already used for elemental carbon, another symbol should be used in this equation otherwise it is confusing.
- 6- Section 3.1 line 210. Why correlation with minimum relative humidity and not with the average relative humidity measured during the collection of samples?
- 7- Section 3.1 lines 226-227. It is not clear why a comparison with London and Madrid? If a comparison with other urban areas is needed it would be better to expand it or explain why choosing specific towns.
- 8- Section 3.3 lines 275-279. This aspect should be discussed in more detail, are authors suggesting that the missing negative charges could be associated to carbonates, like calcium and/or magnesium carbonates? This could be put in evidence looking at the correlation between anion deficit and nss-Ca<sup>2+</sup> concentrations as done, for example, in Contini et al. (Science of the Total Environment 472, 2014, pp. 248–261).
- 9- Section 3.3. line 312. What are IC<sub>2.5</sub> and EC<sub>2.5</sub>?
- 10- Section 3.3. Line 298-299. This sentence is strange. It is not clear if authors speaks about PM<sub>2.5</sub> or PM<sub>10</sub>.
- 11- Section 3.3. Line 323. I do not believe that there are anomalous enrichments, there are only enriched or not enriched elements if a single threshold approach is used.
- 12- Section 3.4. Lines 342-343. The ratio OC/EC is also strongly depending on the measurement protocol used. This aspect should be discussed.
- 13- Section 3.3. Lines 345-349. The mass closure analysis has been done considering metal oxides and possible contribution of carbonates like done, for example, in Cesari et al. (Environmental Science and Pollution Research 23, 15, pp 15133–15148, 2016)? The Si concentrations could be evaluated starting from Al concentrations. I believe that some more details are necessary to actually interpret the unaccounted mass.
- 14- Section 3.4.3 regarding the PAHs risks. It could be useful if the calculated values are compared with typical values found in other areas, see for example, Gregoris et al (Science of the Total Environment 476–477 (2014) 393–405) and references therein.
- 15- Section 3.6 Lines 471-475. It is not clear why only a few species are included in this analysis and not all measured components. This is a little confusing, I can understand that some species were eliminated in PMF for one reason or the other but it is not clear why the other Figures and Tables reports different groups of species.
- 16- Section 3.7 line 503. Why it is necessary to add such a large uncertainty? Could this be due to an underestimation of the uncertainties of chemical determination of the different species?
- 17- Section 3.7 lines 515-516. Mention what is the relative change in Q due to the application of constraints.
- 18- Section 4 line 567. Better high than higher.

- 19- Table 2 is not very useful because the detection limits vary for several order of magnitudes and it is not clear which species have low detection limits and which species have high detection limits. I believe that a more descriptive table could be constructed.
- 20- Table 4. It is not clear the difference between under detection limit and not detected. Essentially not detected for what reason? Because it is under detection limit?
- 21- The correlation table in supplementary materials is not readable too many small numbers.
- 22- In table S7. I do not understand the last line with the percentages

## **Anonymous Referee #2**

**Received and published: 15 June 2017**

### **General comments**

The manuscript entitled "Airborne particles in the Brazilian city of São Paulo: One-year investigation for the chemical composition and source apportionment" by Pereira et al. investigates the chemical composition of PM<sub>2.5</sub> and PM<sub>10</sub> filter samples collected in an extensive and an intensive campaign on a rooftop inside the USP campus in Sao Paulo City over one year. This year was characterized by low precipitation, high temperatures during the summer, resulting in high concentrations of air pollutants over the whole year. Source apportionment of a high number of investigated pollutants was performed using PMF, where 5 factors could be identified. Findings were also used for health risk assessments. Although the number of samples is limited over the year, the results give a good and comprehensive general overview of the amount and sources of a high number of investigated pollutants in Sao Paulo City over the seasons. Overall, the work is within the scope of work published by ACP. Therefore, I recommend publication once the comments and questions below are addressed.

The manuscript will need copy-editing, as there are numerous grammatical errors, a few are mentioned below. In addition, some of the figures and tables are not readable without zooming in and should be revised.

### **Specific comments:**

- 1- Revise the numeration of chapters and subchapters.
- 2- Abstract: Please explain in short the location of the sampling site in the abstract, and that there is only one site where the samples were collected. As Sao Paulo is a megacity with plenty of possible sites with potentially different sources of pollutants, it should be clear for the reader, that this study is different from several other papers publishing e.g. measurements in tunnels in Sao Paulo. The word airborne in the title and abstract is a bit misleading, as it is mostly associated with measurements on airplanes. It is very important to give the sampling height in the abstract and chapter 1.1.
- 3- Page 2, line 67: please write which primary sources PAHs have.
- 4- Table 2: As some species measured from the same instrument have apparently huge DL-differences, it is not helpful to just give the range of DLs. Additionally, add horizontal lines between the different instruments for a better separation.
- 5- Table 2: For the IC, do all ions really have exactly the same DL (6.4)? If so, the authors should write (in short) the reason for equal DLs and/or refer to a publication for this measurement technique.
- 6- Page 3, line 135: I guess, all carbon should be 'oxidized' to CO<sub>2</sub>, not 'volatilized'?
- 7- Page 5, line 188: At that point, it is not clear for me how "missing data" occur. Do you mean missing filter samples for certain days, or missing data from specific species of a filter sample? If it is the last, how can that happen? Were there some species from a filter measured, but others not?
- 8- Table S1: Please explain "Therm. Ampl." and the possible reason for its correlation to PM<sub>2.5</sub> and PM<sub>10</sub>. I am asking as these values are highlighted and not mentioned anywhere in the manuscript.
- 9- Page 6, line 229: Here the authors should make clear, that their data set does not have the data coverage over the year to be fully representative and comparable with the mentioned guidelines. To my knowledge, for both the EU and WHO guidelines, a coverage of 75% for one day and 90% for one calendar year, respectively, is mandatory for a proper risk assessment. Regarding the annual mean limits, the presented data is quite far away from that, considering there is only one day of sampling per week during the extensive campaign.
- 10- Page 7, line 279: Which other cationic species are you suggesting?
- 11- Page 8, line 291: Where and at which height were of gaseous species from CETESB measured? Collocated to the filter sampling?

- 12- Page 8, line 298: A  $R=0.35$  is not relatively higher than  $R=0.78$ . Which correlation(s) are you comparing with  $R=0.35$ ?
- 13- Table 4: What is the difference between "below detection limit" and "not detected?" Beside that "DL" is not used in table 4, but solely in Table 5, and there, "DL" is not explained.
- 14- Page 9, line 347: If the unaccounted part may be attributed to absorbed water, can you see a correlation between this part and ambient humidity?
- 15- Page 10, line 447: I do not think it is necessary to mention the location of a chamber (Florida), but rather the institute.
- 16- Page 13, line 500ff: The authors mention the strong and weak variables. Do all species, which are not mentioned, belong to the bad variables or are they simply not taken into account? Why and how did you define the 25% additional uncertainty? You combine species from different instruments, which have different uncertainties, as you pointed out in Table 2. Inserting these uncertainties into the error matrix usually cause that PMF strongly under- and overestimated the importance of variables within the solutions. Did you scale in some way the uncertainties to each other? Please explain also, why the organic character of Lev and Man justify to set them as weak variables.
- 17- Table S7: What do the percentages mean in the last row?
- 18- General aspects to PMF: I understood that the source of both the RD and VE factor is the traffic. It is not clearly seen in the polar plots (maybe because of the very high RD concentration point coming from NW) how well both factors are correlated to each other. The authors should add a comment on a possible correlation. I also do not understand why all factors were not compared to the gaseous CO and NO<sub>x</sub> time series, as these gases are commonly used to be correlated with primarily emitted factors, especially traffic sources. This would also justify additionally the author's choice of the 5-factor-solution.

### Technical corrections:

Page 2, line 63. Revise this sentence: "Particulate organic carbon includes key species including polycyclic aromatic hydrocarbons (PAHs) and monosaccharides. The last are considered as biomass burning tracers (such as levoglucosan, mannosan, and galactosan)".

Page 3, line 118: use the present tense: intensifies

Page 3, line 118: Revise this sentence: "is presented" should be in the end of the sentence.

Table 1: Its hard to distinguish and attribute some information of the columns to the sources. Add horizontal lines between the rows belonging to each source.

Page 6, line 208. Likewise above: Revise this sentence: "is presented" should be in the end of the sentence.

Figure 1: The 1-2 highest precipitation values increase the scale of the plot too large to see any more variations. Revise it to make the plot more readable, e.g. by changing the y-axis scale above 10 mm.

Page 6, line 234: between the words "study" and "was", the word "it" is missing.

Page 7, line 272: Revise the sentence beginning with "It has..."

Page 7, line 276: Begin the sentence with "The ratio..."

Page 7, line 283: write "potassium ions"

Page 8, line 288: write "the intensive campaign"

Table S2: This table is not readable as the letters and values are too small.

Page 8, line 305: write "increase of..."

Page 9, line 338: delete the word "be"

Figure 9a: Write the factors in the legend on the top of the graph in opposite order, in the figure caption as well, because RD is factor 1, IN is factor 2 etc.

Page 13, line 534: Revise the first part of the sentence.

Page 13, line 542: I guess there is the word "but" missing between the comma and "more"

Page 13, line 549: Rephrase the part in brackets into something like "(as seen by high mass loadings of...)"

Figure S3: Neither the axes nor the legends are readable.

## **(2) author's response**

### **Author comments on referee #1**

- 1- Number of chapters and subchapters are inconsistent.

**AC:** This was corrected for the reviewed version.

- 2- It would be better to mention that also metals are analyzed in the abstract otherwise the reader should wait several pages before to understand that also some metals are investigated.

**AC:** Metals are mentioned on the reviewed abstract:

"...thermal-optical analysis. Trace elements were determined by inductively coupled plasma mass spectrometry. The associated risks..."

- 3- Section 1.2. The height of the back-trajectories used start at 500 m that are not exactly near the ground. Why a lower starting point has not been chosen?

**AC:** Using a lower starting point can be challenging, the height of trajectories was selected to avoid the characteristics of the urban surface (with high rugosity height inducing mechanical turbulence).

- 4- Section 1.4 (line 180). It is reported that PMF was applied to PM<sub>10</sub> but I understand that all data including PM<sub>2.5</sub> were used together. Please correct this incongruence.

**AC:** The PMF was applied considering all data (PM<sub>2.5</sub> and PM<sub>10</sub>), this incongruence was corrected in the reviewed text:

"The widely used source apportionment model, positive matrix factorization (PMF), was applied to all samples dataset (Paatero and Tapper, 1994)."

- 5- Equation (3). The symbol EC was already used for elemental carbon, another symbol should be used in this equation otherwise it is confusing.

**AC:** This symbol for Element Concentration was be changed from EC to C in the reviewed equation:

" $Unc = ([EF \times C]^2 + [0.5 \times DL]^2)^{1/2}$ . Where EF is the error fractions and C is the element concentration."

- 6- Section 3.1 line 210. Why correlation with minimum relative humidity and not with the average relative humidity measured during the collection of samples?

**AC:** Correlations with average and minimum relative humidity will be included on the reviewed text:

"There were moderate negative correlations between PM<sub>10</sub> and average wind speed, minimum and average relative humidity; and between PM<sub>2.5</sub> and average wind speed and minimum relative humidity (Table S1).

- 7- Section 3.1 lines 226-227. It is not clear why a comparison with London and Madrid? If a comparison with other urban areas is needed it would be better to expand it or explain why choosing specific towns.

**AC:** The authors tried to compare particulate matter concentrations with those found in metropolis in different continents, such as Europe and Asia. It will be rewritten in order to explain the choice over those cities:

“The average values for PM<sub>2.5</sub> were higher than those obtained in a year study done in traffic sites in two European metropolis: London and Madrid in 2005 (warm period: 19.40 and 20.63 µg m<sup>-3</sup> for PM<sub>2.5</sub>, respectively) (Kassomenos et al., 2014). The European Union has a more restrictive control of pollutant emissions compared to Latin American countries, since an annual mean of 40 µg m<sup>-3</sup> is established for PM<sub>10</sub> and a limit value of 25 µg m<sup>-3</sup> is imposed for PM<sub>2.5</sub> (Kassomenos et al., 2014). However, these averages in São Paulo are lower than the observed in year-round studies performed in Chinese megacities...”

- 8- Section 3.3 lines 275-279. This aspect should be discussed in more detail, are authors suggesting that the missing negative charges could be associated to carbonates, like calcium and/or magnesium carbonates? This could be put in evidence looking at the correlation between anion deficit and nss-Ca<sup>2+</sup> concentrations as done, for example, in Contini et al. (Science of the Total Environment 472, 2014, pp. 248–261).

**AC:** Authors suggested that the missing charges are associated to calcium and magnesium carbonates. Unfortunately we were unable to determine ionic Ca<sup>2+</sup> and Mg<sup>2+</sup>. The authors decided to withdraw this balance.

- 9- Section 3.3. line 312. What are IC<sub>2.5</sub> and EC<sub>2.5</sub>?

**AC:** They were the previous terms for extensive and intensive campaigns, and were changed to Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub> for the final text. This was corrected for the reviewed version:

“As observed for nss-K<sup>+</sup>, elemental K average concentration was more than twice higher in Int<sub>2.5</sub> than Ext<sub>2.5</sub> (p < 0.05).”

- 10- Section 3.3. Line 298-299. This sentence is strange. It is not clear if authors speaks about PM<sub>2.5</sub> or PM<sub>10</sub>.

**AC:** This paragraph refers to Ext<sub>2.5</sub> (extensive fine, PM<sub>2.5</sub>) and Ext<sub>2.5-10</sub> (extensive coarse, PM<sub>2.5-10</sub>), and will be rewritten in order to make it clearer:

“Na<sup>+</sup> was strongly correlated with Cl<sup>-</sup> in Ext<sub>2.5</sub> (R = 0.78) and had relatively higher correlations with this species (R = 0.35) in Ext<sub>2.5-10</sub>.”

- 11- Section 3.3. Line 323. I do not believe that there are anomalous enrichments, there are only enriched or not enriched elements if a single threshold approach is used.

**AC:** This terminology was done similarly as in Pereira et al. (Atmospheric Environment 41, 2007, pp. 7837–7850) and is also found in Odabasi et al. (Atmospheric Environment 36, 2002, pp. 5841–5851).

- 12- Section 3.4. Lines 342-343. The ratio OC/EC is also strongly depending on the measurement protocol used. This aspect should be discussed.

**AC:** The OC/EC ratios were determined with equivalent measurement protocols in Pio et al. (Atmospheric Environment 45, 2011, pp. 6121–6132). We emphasize this in the reviewed version. Some improvements were done; the references Querol et al., 2013 and Viana et al. 2007 were replaced by Amato et al., 2016 (Atmospheric Chemistry and Physics 16, 2016, pp. 3289–3309), which is a work where equivalent techniques were used:

“Ratios lower than 1 are constantly observed in roadway tunnels and are assumed to describe the composition of fresh traffic emissions (Pio et al., 2011). Amato et al. 2016 found values ranging from 1.8 and 3.7 at the urban background sites using equivalent measurement protocols. It was attributed to the distance from main roads, which can increase the influence of secondary OC (Pio, 2011). In this way, the values for OC/EC found in the present study may be due to vehicle emissions with contribution of secondary organic aerosols.”

- 13- Section 3.3. Lines 345-349. The mass closure analysis has been done considering metal oxides and possible contribution of carbonates like done, for example, in Cesari et al. (Environmental Science and Pollution Research

23, 15, pp 15133–15148, 2016)? The Si concentrations could be evaluated starting from Al concentrations. I believe that some more details are necessary to actually interpret the unaccounted mass.

**AC:** Only the oxides were considered for the mass closure analysis, as in Alves et al. (*Atmospheric Research* 153, 2015, pp. 134–144). The concentrations of Si were evaluated from Al considering a ratio of 3.4 as in Clements et al. (*Atmospheric Environment* 89, 2014, pp. 373–381), but it may have been overestimated this way; the total accounted mass exceeds in 20% the PM<sub>2.5</sub> concentration for the intensive campaign.

- 14-** Section 3.4.3 regarding the PAHs risks. It could be useful if the calculated values are compared with typical values found in other areas, see for example, Gregoris et al (*Science of the Total Environment* 476–477 (2014) 393–405) and references therein.

**AC:** The authors improved the paragraph considering more references:

“DBA had the largest contribution to carcinogenic potential and BaP the highest to mutagenic potential. In studies performed in Italian urban areas, BaP was the compound that most contributed to total carcinogenicity in PM, although the TEF used for DBA was lower in those cases (Cincinelli et al., 2007; Gregoris et al., 2014). LCR from exposure to atmospheric PAH was estimated by multiplying BaP-TEQ and BaP-MEQ by the unit risk ( $87 \times 10^{-6}$  (ng m<sup>-3</sup>)<sup>-1</sup>) for exposure to BaP established by WHO (de Oliveira Alves et al., 2015; WHO, 2000) (Fig. 5) and was possible to observe an increase during the intensive campaign. In all campaigns, the values observed were higher than the observed in studies done in the Amazon during dry season with events of biomass burning (de Oliveira Alves et al., 2015); studies done at different seasons in other urban areas as New York and Madrid pointed carcinogenic risks within the recommended by environmental and health agencies (Jung et al., 2010; Mirante et al., 2013)”.

- 15-** Section 3.6 Lines 471-475. It is not clear why only a few species are included in this analysis and not all measured components. This is a little confusing, I can understand that some species were eliminated in PMF for one reason or the other but it is not clear why the other Figures and Tables reports different groups of species.

**AC:** More than 80 species were determined in this study, so they had to be selected in order to be included in this graph because of the lack of space for it. The ratios were calculated for all species, but only the species more important were discussed on the text: for being source tracers, having higher concentrations and/or health effects.

- 16-** Section 3.7 line 503. Why it is necessary to add such a large uncertainty? Could this be due to an underestimation of the uncertainties of chemical determination of the different species?

**AC:** The uncertainties were increased in order to avoid discarding measurements that have poor data quality, due to some measurements below detection limits, this procedure was done according to reference of Paatero and Hopke (*Analytica Chimica Acta* 490, 2003, pp. 277-289).

- 17-** Section 3.7 lines 515-516. Mention what is the relative change in Q due to the application of constraints.

**AC:** The relative change in Q ( $dQ = 0.4\%$ ) due to application of constraints is mentioned in the reviewed version:

“Lev and Man were pulled up maximally in the biomass burning factor in order to have a better separation between both factors, relative change in Q was of 0.4 %”.

- 18-** Section 4 line 567. Better high than higher.

**AC:** Corrected in the reviewed text.

- 19-** Table 2 is not very useful because the detection limits vary for several order of magnitudes and it is not clear which species have low detection limits and which species have high detection limits. I believe that a more descriptive table could be constructed.

**AC:** The authors suggest keeping this table, showing the range of those detection limits. More than 80 species were determined.

- 20-** Table 4. It is not clear the difference between under detection limit and not detected. Essentially not detected for what reason? Because it is under detection limit?

**AC:** The term “under detection limit” was adopted for all tables in the reviewed version.

- 21-** The correlation table in supplementary materials is not readable too many small numbers.

**AC:** The table file in excel will be uploaded as supplementary material, so it can be better understood.

- 22-** In table S7. I do not understand the last line with the percentages.

**AC:** These are the percentages over the reduction on the ratio  $Q_{robust}/Q_{expected}$  as the number of factor increases, in the Q value analysis. These percentages could be removed for the reviewed final version of supplementary information file.

## **Author comments on referee #2**

- 1-** Revise the numeration of chapters and subchapters.

**AC.** Corrected in the reviewed version.

- 2-** Abstract: Please explain in short the location of the sampling site in the abstract, and that there is only one site where the samples were collected. As São Paulo is a megacity with plenty of possible sites with potentially different sources of pollutants, it should be clear for the reader, that this study is different from several other papers publishing e.g. measurements in tunnels in São Paulo. The word airborne in the title and abstract is a bit misleading, as it is mostly associated with measurements on airplanes. It is very important to give the sampling height in the abstract and chapter 1.1.

**AC.** Modifications in the text:

The title is changed to: “Particulate pollutants in the Brazilian city of São Paulo: One-year investigation for the chemical composition and source apportionment”

Part of the Abstract was modified to: “In order to evaluate the sources of particulate air pollution and related health risks, a year-round sampling was done at the University of São Paulo campus (20 m above ground level), a green area near an important expressway. The sampling was performed for  $PM_{2.5}$  ( $\leq 2.5 \mu m$ ) and  $PM_{10}$  ( $\leq 10 \mu m$ ) in 2014 through intensive (every day sampling in wintertime) and extensive campaigns (once a week for the whole year) with 24 h of sampling”.

In the item 2 “Methodology”, section 2.1. Sampling campaigns: “Aerosol samples were collected at a São Paulo site (SPA, 23°33'34”S and 46°44'01”W) located on the ~ rooftop of the Atmospheric Sciences Department (about 20 m above ground level), at Institute of Astronomy and Atmospheric Sciences (IAG-USP) building, within the campus of University of São Paulo.”

- 3-** Page 2, line 67: please write which primary sources PAHs have.

**AC.** Modifications in the text:

“Particulate organic carbon includes key species including polycyclic aromatic hydrocarbons (PAHs) and monosaccharides. The last are considered as biomass burning tracers (such as levoglucosan, mannosan, and galactosan) (Simoneit et al., 1999). PAHs have natural sources (synthesis by plants and bacteria, degradation of

plants, forest fires and volcanic emissions), but are mostly emitted by anthropogenic sources in urban sites (such as domestic, mobile, industrial and agricultural sources) (Abdel-shafy and Mansour, 2016; Ravindra et al., 2008).”

- 4- Table 2: As some species measured from the same instrument have apparently huge DL-differences, it is not helpful to just give the range of DLs. Additionally, add horizontal lines between the different instruments for a better separation.

**AC.** More than 80 species were determined and it would be difficult to present all detection limits. That is why the authors presented in such way. Lines were added between the different instruments.

- 5- Table 2: For the IC, do all ions really have exactly the same DL (6.4)? If so, the authors should write (in short) the reason for equal DLs and/or refer to a publication for this measurement technique.

**AC.** I noticed there was a typo in this table, in fact the DL value is  $1.27 \text{ ng m}^{-3}$  for all ions (corrected in the reviewed version), and this was calculated with basis on detection limit of  $1 \text{ ng ml}^{-1}$  (per volume of analyte solution on the IC). These values were provided by the FMI research group and are based on the signal to noise ratio of the IC baseline. Similar DLs were observed for the ions  $\text{Na}^+$ ,  $\text{NH}_4^+$ ,  $\text{K}^+$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$  and MSA.

- 6- Page 3, line 135: I guess, all carbon should be 'oxidized' to  $\text{CO}_2$ , not 'volatilized'?

**AC.** Corrected in the reviewed version.

- 7- Page 5, line 188: At that point, it is not clear for me how "missing data" occur. Do you mean missing filter samples for certain days, or missing data from specific species of a filter sample? If it is the last, how can that happen? Were there some species from a filter measured, but others not?

**AC.** The group decided to determine elements after few samples ( $n = 8$ ) were collected in fiber glass filters. In these filters it is not possible to determine trace elements (the blank filters presented high levels of elements). To avoid discarding these samples (other species) we decided to do the statistical treatment as suggested by Brown et al. (Science of Total Environment 518–519, 2015, pp. 626-635). This paper suggests that the missing values are replaced by the median values of the valid samples.

- 8- Table S1: Please explain "Therm. Ampl." and the possible reason for its correlation to  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$ . I am asking as these values are highlighted and not mentioned anywhere in the manuscript.

**AC.** This is the thermal amplitude on the sampling days (maximum temperature minus minimum temperature) and will be removed from the supplementary information.

- 9- Page 6, line 229: Here the authors should make clear, that their data set does not have the data coverage over the year to be fully representative and comparable with the mentioned guidelines. To my knowledge, for both the EU and WHO guidelines, a coverage of 75 % for one day and 90 % for one calendar year, respectively, is mandatory for a proper risk assessment. Regarding the annual mean limits, the presented data is quite far away from that, considering there is only one day of sampling per week during the extensive campaign.

**AC.** The comparisons were removed from the paragraph, but the WHO guidelines were kept cited as a reference. The text:

“In the extensive campaign, the PM mass concentrations exhibited a wide range of concentrations. The concentrations in the  $\text{Ext}_{2.5}$  ranged from 8 to  $78 \text{ } \mu\text{g m}^{-3}$  (average  $30 \text{ } \mu\text{g m}^{-3}$ ), whereas  $\text{Ext}_{10}$  values varied between 12 and  $113 \text{ } \mu\text{g m}^{-3}$  (average  $44 \text{ } \mu\text{g m}^{-3}$ ) (Fig. 3). The World Health Organization (WHO) recommends a daily limit for  $\text{PM}_{10}$  of  $50 \text{ } \mu\text{g m}^{-3}$  and of  $25 \text{ } \mu\text{g m}^{-3}$  for  $\text{PM}_{2.5}$ , (WHO, 2006) while the Brazilian Environmental Agency (CONAMA) recommends a threshold of  $150 \text{ } \mu\text{g m}^{-3}$  for  $\text{PM}_{10}$  (CONAMA, 1990; Pacheco et al., 2017). When considering the CONAMA standards, only one day in the extensive campaign was near the target limit. The  $\text{Ext}_{10}$  campaign was divided into two periods: dry (April to September) and rainy (October to March). It was observed that the average  $\text{PM}_{10}$  was  $52 \text{ } \mu\text{g m}^{-3}$  in the dry period and of  $35 \text{ } \mu\text{g m}^{-3}$  in the rainy period.”



10- Page 7, line 279: Which other cationic species are you suggesting?

**AC.** The authors decided to remove the cation/anion ratio discussion from the manuscript. Important species are missing for a complete analysis.

11- Page 8, line 291: Where and at which height were of gaseous species from CETESB measured? Collocated to the filter sampling?

**AC.** The gaseous species data were taken from two monitoring stations (CETESB). The first one ( $\text{NO}_x$ ) is located inside the university campus (800 m far from the sampling site, at ground level) and another station is 3.2 km far (for CO, at 2m height). The sources that affect these stations are similar to those that affect the sampling site (Marginal Pinheiros expressway and local traffic). Other studies identified high correlations of pollutants concentrations between IPEN station and IAG sampling site; Oyama et al, (Atmospheric Chemistry and Physics, 16, 2016, 14397-14408) and Vara-Vela et al. (Atmospheric Chemistry and Physics, 16, 2016, 777-797).

12- Page 8, line 298: A  $R=0.35$  is not relatively higher than  $R=0.78$ . Which correlation(s) are you comparing with  $R=0.35$ ?

**AC.** It was not properly written, leading to the misunderstanding that these correlations were compared, but they were not. This part was rewritten:

“ $\text{Na}^+$  was strongly correlated with  $\text{Cl}^-$  in  $\text{Ext}_{2.5}$  ( $R = 0.78$ ) and in  $\text{Ext}_{2.5-10}$  presented a relatively moderate correlation ( $R = 0.35$ )”.

13- Table 4: What is the difference between "below detection limit" and "not detected?" Beside that "DL" is not used in table 4, but solely in Table 5, and there, "DL" is not explained.

**AC.** Corrected in the reviewed version.

14- Page 9, line 347: If the unaccounted part may be attributed to absorbed water, can you see a correlation between this part and ambient humidity?

**AC.** This correlation was determined and it was very low ( $R=0.2$ ).

15- Page 10, line 447: I do not think it is necessary to mention the location of a chamber (Florida), but rather the institute.

**AC.** Modifications in the text:

“The Lev/Man ratios are characteristic of each type of biomass. The ratios were similar to that obtained in a chamber study with sugarcane burning (Lev/Man = 10, Hall et al., 2012), and also to that reported for the 2013 intensive campaign (Lev/Man = 12, Pereira et al., 2017).”

16- Page 13, line 500: The authors mention the strong and weak variables. Do all species, which are not mentioned, belong to the bad variables or are they simple not taken into account? Why and how did you define the 25% additional uncertainty? You combine species from different instruments, which have different uncertainties, as you pointed out in Table 2. Inserting these uncertainties into the error matrix usually cause that PMF strongly under- and overestimated the importance of variables within the solutions. Did you scale in some way the uncertainties to each other? Please explain also, why the organic character of Lev and Man justify to set them as weak variables.

**AC.** Some species were rejected due to low S/N ratio and other could not be included due to the relative high number of species compared to the number of samples. Some variables were not considered in the analysis and considered as bad variables, because they are highly correlated with other variables (high co-linearity and redundancies) and don't give us more information related to the sources. PAHs could also not be included because of the great effect of gas-particulate partitioning that is influenced by meteorological conditions. Maybe it was not well understood, but the organic character of Lev and Man mean they are a little partitioned between gas and

particle and also can be decomposed in the atmosphere, as discussed in Pio et al. (Atmospheric Environment 42, 2008, pp. 7530-7543) that's why they were considered as weak species. The uncertainties were increased in order to avoid discarding measurements that have poor data quality, due to some measurements below detection limits; this procedure was done according to reference of Paatero and Hopke (Analytica Chimica Acta 490, 2003, pp. 277-289). The additional uncertainty was applied to all the variables increasing proportionally the uncertainties. Besides the increase of the uncertainty, the physical solution does not change, presenting the same factors and results. We concluded that the solution is stable, because the same sources could be identified in most of the solutions generated, with different additional uncertainties.

17- Table S7: What do the percentages mean in the last row?

**AC.** Percentage of reduction over Qrobust/Qexpected with the increase in the number of factors. It was withdraw in the new version (Supplementary information).

18- General aspects to PMF: I understood that the source of both the RD and VE factor is the traffic. It is not clearly seen in the polar plots (maybe because of the very high RD concentration point coming from NW) how well both factors are correlated to each other. The authors should add a comment on a possible correlation. I also do not understand why all factors were not compared to the gaseous CO and NO<sub>x</sub> time series, as these gases are commonly used to be correlated with primarily emitted factors, especially traffic sources. This would also justify additionally the author's choice of the 5-factor-solution.

**AC.** There was a weak correlation between RD and VE and maybe it should not be expected to have such correlation, since the aerosol emitted by RD has a larger aerodynamic diameter than the aerosol related to VE. RD sources are primary and VE can be primary but they also can be secondarily formed from vehicular primarily emitted precursors. RD can be influenced by traffic but also by high wind speed. High correlations were observed between CO and NO<sub>x</sub> with primary sources factors as VE and BB. It may be due to the difficulty in separating these factors from each other, since their contributions increase with meteorological conditions (wind direction). Correlations were moderate between the gases and IN and RD and no correlation was observed between these gases and the SP (secondary processes). These findings will be included in the PMF discussions (section 3.7):

“...Vehicular source seemed to increase with winds coming from the North and Northwest, passing by the expressway, but decreased with SE winds, as observed previously (Sánchez-Ccoyllo and Andrade, 2002). The polar plots profiles of VE and RD factors presented a different pattern, since the aerosol from road dust suspension has a larger aerodynamic diameter (Karanasiou et al., 2009) and tends to increase with wind speed”.

“High correlations ( $R > 0.8$ ) were observed between the gases CO and NO<sub>x</sub> and the primary sources factors VE and BB. These gases are related to vehicular emissions (Alonso et al., 2010) and the correlations with the biomass burning factors may be due to the fact that it increases with the same wind direction as the vehicular factor. No correlations were found between these gases and SP factor.”

### **(3) author's changes in manuscript**

The most relevant changes made in the text are listed below and can be better observed on the attached marked-up version of the revised manuscript:

- 1- References to “airborne particles” were removed (title and in line 22) as suggested by referee #2.
- 2- More information was added to the abstract (sampling location and analytical method for element determination) (lines 24-25 and 32-33).
- 3- The numeration of chapters and subchapters was revised.
- 4- More information was added to PAHs sources description in the introduction (section 1 / lines 65-67).
- 5- Sampling height is described in the reviewed methodology (section 2.1 / line 99).
- 6- The acronym for element concentration was changed from *EC* to *C* (section 2.4 / lines 198-199).
- 7- Correlations of PM with average humidity were included in the discussion (section 3.1 / lines 213-214).
- 8- Comparisons with WHO guidelines were removed, since referee #2 stated that the coverage of the samplings was not sufficient for a proper risk assessment (section 3.1 / lines 221-222 and 224-225).
- 9- This paragraph was modified in order to explain the choices over those sites (section 3.1 / lines 230-237).  $\Sigma$ cations/ $\Sigma$ anions discussion was removed (section 3.3 / line 280-282). If available,  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  concentrations could explain the missing charges associated with carbonates.
- 10- OC/EC ratio discussion was revised (section 3.4 / lines 344-348). References from Querol et al., 2013 and Viana et al., 2007 were removed, Amato et al. 2016 was considered for the OC/EC ratios comparison.
- 11- PAHs risk assessment discussions were improved (section 3.5.3 / lines 441-443 and lines 446-449).
- 12- Relative change in Q value was added (section 3.8 / line 529).
- 13- A comment on polar plot profiles of VE and RD factors was included (section 3.8 / lines 552-554).
- 14- Discussions on the correlations between gases and PMF factors were inserted (section 3.8 / lines 563-568) A table was also inserted (Table S8).
- 15- The Y-axis of the precipitation plot (Figure 2c) was broken above 15 mm to make it more readable.
- 16- The legend was altered in order to correct the factors order (Figure 9a).
- 17- Horizontal lines were added between rows on the tables (Tables 1 and 2).
- 18- Detection limits for Ion Chromatography were corrected (Table 2).
- 19- The acronyms ND were changed to <DL (Table 4).
- 20- Table S2 will be submitted as an Excel file, since it was not readable in the Supplementary Information.
- 21- Figure S3 charts were enlarged in order to make them more legible in the Supplementary Information.
- 22- Other minor technical corrections were made along the text (as suggested by referees).

# ~~Airborne particles~~ **Particulate pollutants** in the Brazilian city of São Paulo: One-year investigation for the chemical composition and source apportionment

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**Abstract.** São Paulo in Brazil has relatively relaxed regulations for ambient air pollution standards and often experiences high air pollution levels due to emissions of ~~airborne~~ **particulate pollutants** from local sources and long-range transport of biomass burning-impacted air masses. In order to evaluate the sources of particulate air pollution (~~PM~~) and related health risks, a year-round sampling was done at the University of São Paulo campus (20 m above ground level), a green area near an important expressway. The sampling- was performed for PM<sub>2.5</sub> ( $\leq 2.5 \mu\text{m}$ ) and PM<sub>10</sub> ( $\leq 10 \mu\text{m}$ ) in 2014 through intensive (every day sampling in wintertime) and extensive campaigns (once a week for the whole year) with 24 h of sampling. This year was characterized to have lower average precipitation comparing to meteorological data, and high pollution episodes were observed all year round, with a significant increase of pollution level in the intensive campaign, which was performed during wintertime. Different chemical constituents, such as carbonaceous species, polycyclic aromatic hydrocarbons (PAHs) and derivatives, water-soluble ions and biomass burning tracers were identified in order to evaluate health risks and to apportion sources. The species such as PAHs, inorganic and organic ions and monosaccharides were determined by chromatographic techniques and carbonaceous species by thermal-optical analysis. Trace elements were determined by inductively coupled plasma mass spectrometry. The associated risks to particulate matter exposure based on PAH concentrations were also assessed, along with indexes such as the benzo[a]pyrene equivalent (BaPE) and lung cancer risk (LCR). High BaPE and LCR were observed in most of the samples, rising to critical values in the wintertime. Also, biomass burning tracers and PAHs were higher in this season, while secondarily formed ions presented low variation throughout the year. Meanwhile, vehicular tracer species were also higher in the intensive campaign suggesting the influence of lower dispersion conditions in that period. Source apportionment was done by Positive Matrix Factorization (PMF), which indicated five different factors: road dust, industrial emissions, vehicular exhaust, biomass burning and secondary processes. The results highlighted the contribution of vehicular emissions and the significant input from biomass combustion in wintertime, suggesting that most of the particulate matter is due to local sources, besides the influence of pre-harvest sugarcane burning.

## 1 Introduction

45 Air pollution caused by atmospheric particulate matter (PM) is one of the major environmental problems encountered in Latin American cities such as São Paulo (Brazil), Mexico City (Mexico), Bogota (Colombia) and Santiago (Chile) (Romero-Lankao et al., 2013; Vasconcellos et al., 2010, 2011a; Villalobos et al., 2015). The air pollution thresholds in most of the Latin American cities are not very stringent compared to international standards or guidelines (Alvarez et al., 2013; Kumar et al., 2016). Several studies have highlighted a statistical relation between PM and health problems, including respiratory and cardiovascular diseases and genotoxic risks (Newby et al., 2015; de Oliveira Alves et al., 2014; Pope, 2000). In this context, PM<sub>2.5</sub> (PM with aerodynamic diameter smaller than 2.5 µm) and PM<sub>10</sub> (PM with aerodynamic diameter smaller than 10 µm) are particles that are able to penetrate in the respiratory system, with PM<sub>2.5</sub> reaching alveoli in the lungs, and induce adverse impacts on human health (Cai et al., 2015; Kumar et al., 2014). The elderly and the children are more prone to be susceptible individuals to the health effects resulting from PM<sub>2.5</sub> (Cançado et al., 2006; Segalin et al., 2017). Considering that elderly population has grown in São Paulo over the last decades (SEADE, 2016; Segalin et al., 2017), the PM health-related issues can become more relevant. PM also plays an important role in ecosystem biogeochemistry, hydrological cycle, cloud formation and atmospheric circulation (Pöschl, 2005).

Carbonaceous species as organic and elemental carbons (OC and EC) represent a large fraction of PM and play an important role in the formation of haze, interaction with climate and adverse human health effects (Bisht et al., 2015; Liu et al., 2016; Seinfeld and Pandis, 2006). Water-soluble ions (WSI) account for another major fraction of aerosols in urban areas and are able to affect visibility, particle hygroscopicity, cloud formation; they also influence acidity in rainwater and impact climate (Cheng et al., 2011; Jung et al., 2009; Khoder and Hassan, 2008; Tan et al., 2009; Tang et al., 2016; Yang et al., 2015).

Particulate organic carbon includes key species including polycyclic aromatic hydrocarbons (PAHs) and monosaccharides. The last are considered as biomass burning tracers (such as levoglucosan, mannosan, and galactosan) (Simoneit et al., 1999). PAHs have natural sources (synthesis by plants and bacteria, degradation of plants, forest fires and volcanic emissions), but are mostly formed by emitted by anthropogenic emissions sources in urban sites (such as domestic, mobile, industrial and agricultural sources) (Abdel-shafy and Mansour, 2016; Ravindra et al., 2008). They have been studied because of their carcinogenic properties (de Oliveira Alves et al., 2014; Seinfeld and Pandis, 2006). The nitrated and oxygenated PAHs (nitro- and oxy-PAHs) are emitted as primary species or are formed in situ as secondary compounds (Kojima et al., 2010; Souza et al., 2014b; Zhou and Wenger, 2013; Zimmermann et al., 2013). They are potentially more mutagenic and/or carcinogenic than their PAH precursors (Franco et al., 2010).

Chemical speciation and PAH risk assessment have been performed in several Latin American sites, specifically in urban São Paulo, Bogota, Buenos Aires (Vasconcellos et al., 2011a; Vasconcellos et al., 2011b) and in forested areas such as the Amazon region (de Oliveira Alves et al., 2015). Biomass burning tracers have been detected in high concentrations in São Paulo during the dry season and are attributed to the long-range transport of aerosols from areas affected by sugarcane burning. Source apportionment studies have been carried out in São Paulo (Table 1) in the last three decades, but not in as much detail as in other megacities. Detailed characterization of the organic fraction of aerosols is still scarce.

A previous study performed in São Paulo in 1989 highlights the relative importance of the emissions from residual oil and diesel in PM<sub>2.5</sub> and soil dust in the coarse grain size (Andrade et al., 1994). Da Rocha et al. (2012) studied the emission sources of fuel and biomass burning, the gas-to-particle conversion, and sea spray emissions in PM in São Paulo, in one year period (between 2003 and 2004). Another study conducted in the winter of 2003 pointed out a strong impact of local sources in three sites in the state of São Paulo, besides the influence of remote sources (Vasconcellos et al., 2007). A source apportionment for PAHs in the winter of 2002 reported a predominance of diesel emissions for the polyaromatics in PM<sub>2.5</sub> (Bourotte et al., 2005). In turn, Castanho and Artaxo (2001), in their study of 1997 and 1998 in São Paulo city, reported no

85 significant differences in the main air pollution sources (i.e. automobile traffic and soil dust) between wintertime and summertime. The main sources for PM<sub>2.5</sub> were automobile traffic and soil dust. However, biomass burning was not considered as a potential source by the authors.

The current study presents a more comprehensive study that should lead to a better understanding of the main PM sources and atmospheric processes occurring in the São Paulo megacity than previous studies reported in the literature. A year of  
90 extensive sampling of aerosol (PM<sub>2.5</sub> and PM<sub>10</sub>) and a wintertime intensive campaign were performed. Different classes of chemical components in PM were determined such as carbonaceous species, WSI, monosaccharides, PAHs, and their derivatives. Meteorological data were also collected during the sampling days. Moreover, the benzo[a]pyrene equivalent (BaPE) and lung cancer risk (LCR) indexes were calculated in order to assess the potential toxicity of PAHs. Positive Matrix Factorization (PMF) analysis was also used for the source apportionment of PM<sub>10</sub> during the extensive campaign.

## 95 | **2** Methodology

### ~~1.12.1~~ **1.2.1** Sampling campaigns

Aerosol samples were collected at a São Paulo site (SPA, 23°33'34"S and 46°44'01"W) located on the rooftop of the Atmospheric Sciences Department (about 20 m above ground level), at Institute of Astronomy and Atmospheric Sciences  
100 (IAG-USP) building, within the campus of University of São Paulo. The location is inside a green area and approximately 2 km away from an important expressway (Marginal Pinheiros) (Fig. 1). Aerosols were collected in intensive (every day) and extensive campaigns (once a week) throughout 2014. Firstly, the extensive campaign was performed weekly. Accordingly, samples were collected every Tuesday for PM<sub>2.5</sub> (termed Ext<sub>2.5</sub> in this study) and PM<sub>10</sub> (termed Ext<sub>10</sub>). However, due to equipment breaking down, the PM<sub>2.5</sub> sampling was stopped in September while the PM<sub>10</sub> sampling continued until December  
105 ( $n = 32$  and  $38$ , respectively). Secondly, the intensive campaign (termed Int<sub>2.5</sub>) took place between 01 and 18 July, 2014 ( $n = 12$ ), only for PM<sub>2.5</sub> due to problems with PM<sub>10</sub> equipment. However, there were four days (between 08 and 11 July) for which data were not collected due to heavy rain.

PM samples were collected for a period of 24 h, with high-volume air samplers (Hi-Vol), with a flow rate of 1.13 m<sup>3</sup> min<sup>-1</sup>, with 2.5 and 10 μm size selective inlets (Thermo Andersen, USA). Prior to sampling, quartz fiber filters (20 cm × 25 cm, Millipore, USA) were baked for 8 h at 800 °C to remove the organics. In addition, filters were equilibrated at room  
110 temperature and weighed in a microbalance before and after the sampling, in order to estimate the PM concentration. After sampling and weighing, the filters were wrapped in aluminum foil and stored in a refrigerator at 5 °C until chemical analyzes were performed.

### ~~1.22.2~~ **1.2.2** Meteorological data ~~and gaseous species~~

The meteorological data (ambient temperature, relative humidity, precipitation and wind speed) were collected from the climatological bulletin of IAG/USP meteorological station (IAG, 2014). The climate of São Paulo is often classified as humid subtropical (Andrade et al., 2012a). The wintertime in the city is characterized by a slight decrease in temperatures, together with considerably lower relative humidity and precipitations, with more thermodynamic stability, often resulting in accumulation of air pollutants in the lower troposphere, being also subjected to thermal inversion episodes (Miranda et al.,  
120 2012). The local air circulation is mainly associated with the Atlantic Ocean breeze and cold fronts in wintertime often intensified that, with winds generally coming from Southeast (Vasconcellos et al., 2003). In Fig. S1 ~~is presented the a~~ comparison between the average climatological temperature and the data for 2014 is presented (IAG, 2014). During the 2014 campaign, the summer was atypically warmer and drier.

In order to analyze the long-range transport of air pollutants, backward air mass trajectories (96 h) were run using the HYSPLIT model (Draxler and Rolph, 2003), through READY (Real-time Environmental Applications and Display System) platform from NOAA (National Oceanic and Atmospheric Administration). The considered heights were 500, 1500 and 3000 m, corresponding to trajectories near the ground, upper boundary layer and low free troposphere, respectively (Cabello et al., 2016; Toledano et al., 2009).

### **1.3.2.3 Analytical procedures, reagents and standards**

After sampling, the filters were punched for the chemical analysis, as shown in Table 2, which lists all substances determined and their respective analytical techniques as well as their detection limits (DL).

Carbonaceous species were determined at the University of Aveiro, with two punches of 9 mm diameter. Firstly, the carbonates were removed with hydrochloric acid fumes and then OC and EC were determined by a thermal-optical transmission equipment developed at the university. The system comprises a quartz tube with two heating zones, a pulsed laser and a non-dispersive infrared CO<sub>2</sub> analyzer (NDIR). The filters were placed into the first heating zone of the quartz tube then heated to 600 °C in a nitrogen atmosphere for the organic fraction to vaporize, which was quantified as OC. EC was determined with a sequential heating at 850 °C in an atmosphere containing 4 % O<sub>2</sub>. The other heating zone was filled with cupric oxide and was maintained at 650 °C in a 4 % O<sub>2</sub> atmosphere, to assure that all carbon is ~~volatilized~~ oxidized to CO<sub>2</sub>, which is quantified by a NDIR analyzer (Alves et al., 2015).

The determination of polycyclic aromatic hydrocarbons and their derivatives was done at Federal University of Bahia, Brazil, and is summarized in Table 2. Briefly, samples were extracted for 23 min in an ultrasonic bath (4.2 cm<sup>2</sup> punches) with a 500 µL solution of 18 % of acetonitrile in dichloromethane, employing miniaturized extraction devices (Whatmann Mini<sup>TM</sup> UniPrep Filters, Whatman, USA). Their quantification was carried out by gas chromatography with high-resolution mass spectrometer detection (GC-MS). The procedure is described in more details in Santos et al. (2016). BeP was quantified with the same calibration curve as BaP since they have similar fragmentation pattern in the MS detector (Robbat and Wilton, 2014).

US Environmental Protection Agency (EPA) 610 PAH mix in methanol:dichloromethane (1:1), containing 2000 µg mL<sup>-1</sup> each, was purchased from Supelco (St. Louis, USA). Individual standards of 50 µg mL<sup>-1</sup> coronene (Cor) and 1000 µg mL<sup>-1</sup> perylene (Per) and two deuterated compounds, pyrene D10 (Pyr d10) and fluorene D10 (Flu d10) were purchased from Sigma-Aldrich (St. Louis, USA). Quinones investigated in this study were purchased from Fluka (St. Louis, USA). Nitro-PAH certified standard solutions SRM 2264 (aromatic hydrocarbons nitrated in methylene chloride I) and SRM 2265 (polycyclic aromatic hydrocarbons nitrated in methylene chloride II) were purchased from the National Institute of Standards and Technology (NIST, USA).

Monosaccharides and WSI were determined at Finnish Meteorological Institute. From quartz fiber filters samples, 1 cm<sup>2</sup> filter pieces were punched for both analyzes. Concentrations of monosaccharides were determined using a Dionex ICS-3000 system coupled to a quadrupole mass spectrometer (Dionex MSQ<sup>TM</sup>) by high-performance anion exchange chromatography (HPAEC-MS). Levoglucosan (1,6-anhydro-β-D-glucopyranose, purity 99 %; Acros Organics, NJ, USA), mannosan (1,6-anhydro-β-D-mannopyranose; purity 99 %; Sigma-Aldrich Co., MO, USA), and galactosan (1,6-anhydro-β-D-galactopyranose; Sigma-Aldrich Co.) were used for the calibration. The 1 cm<sup>2</sup> punches were extracted with 5 mL of deionized water (Milli-Q water; resistivity 18.2 MΩ.cm at 25 °C, Merck Millipore, MA, USA) with methyl-β-D-arabinopyranoside as internal standard (purity 99 %; Aldrich Chemical Co., WI, USA), and 10 min of gentle rotation. The extract was filtered through an IC Acrodisc® syringe filter (13 mm, 0.45 µm Supor® (PES) membrane, Pall Sciences) (Saarnio et al., 2010).

165 In order to determine the water-soluble ions ( $\text{Cl}^-$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{C}_2\text{O}_4^{2-}$ , methylsulfonate,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{NH}_4^+$ ) 10 mL of deionized water was used to extract the sample aliquots, with 10 min of gentle rotation. The ions were determined using two ion chromatography systems (ICS 2000 system, Dionex) simultaneously; cations were analyzed using a CG12A/CS12A column with an electrochemical suppressor (CSRS ULTRA II, 4 mm) and anions using an AG11/AS11 column with an electrochemical suppressor (ASRS ULTRA II, 4 mm).

170 Finally, trace elements in the samples were extracted using a microwave digestion system (MLS-1200 mega, Milestone Inc., Italy) at National University of Singapore. Punctures of the filters were cut into small pieces and added into PTFE vessels with 4 mL  $\text{HNO}_3$  (Merck), 2 mL  $\text{H}_2\text{O}_2$  (Merck) and 0.2 mL HF (Merck). The vessels were then subjected to a three-stage digestion inside the microwave digester (250 W for 5 min, 400 W for 5 min, and 600 W for 2 min). Following the digestion procedure, extracts were filtered with 0.45  $\mu\text{m}$  PTFE syringe filters, diluted 8 times and stored in the 4 °C cold room. The concentrations of trace elements were quantified by ICP-MS (Agilent 7700, USA) in triplicates. The instrumental parameters maintained during sample runs using the ICP-MS analysis were: plasma gas (15.0 L  $\text{min}^{-1}$ ), auxiliary gas (1.0 L  $\text{min}^{-1}$ ), and nebulizer gas (1.0 L  $\text{min}^{-1}$ ). Clean ceramic scissors and forceps were used to handle all PM samples. ICP-MS standards (purchased from High-Purity Standards, USA) were used for calibration.

#### 1.42.4 Statistical analysis and receptor model

180 Pearson coefficients were calculated to verify the correlation between all the species (software STATISTICA). It determines the extent to which values of the variables are linearly correlated. The coefficients ( $r$ ) were considered significant when  $p < 0.05$ . Two-tailed t-tests were also employed in order to evaluate equal and unequal variances ( $p < 0.05$ ). Polar plots considered the mass concentrations as a function of wind speed and direction (software R x64 3.3.2).

185 The widely used source apportionment model, positive matrix factorization (PMF), was applied to ~~all the~~ **PM<sub>10</sub>-samples** dataset (Paatero and Tapper, 1994). In this study, specifically, the EPA PMF5.0 software was used. Variables were classified as *strong*, *weak* and *bad* according to the signal-to-noise ratio ( $S/N$ ), number of samples below the detection limit (Amato et al., 2016; Contini et al., 2016; Paatero and Hopke, 2003) and thermal stability of the species. The species were categorized as *bad* when the  $S/N$  ratios were less than 0.2 and *weak* when the  $S/N$  ratios were greater than 0.2 but less than 2 (Lang et al., 2015). Accordingly, species with  $S/N$  ratios higher than 2 were considered *strong*. *Bad* variables were excluded from the model and the *weak* ones had their uncertainty increased by a factor of 3, as described in the EPA PMF Fundamentals and User Guide (Norris et al., 2014).

190 When concentrations were below the detection limits, they were substituted by half the detection limit ( $DL$ ). Missing data were replaced by the median ( $M$ ) of the whole dataset for that species. Uncertainties were calculated by Eq. (1) according to Norris et al. (2014), when the concentrations were below the detection limits:

$$Unc = 5/6 \times DL \quad (1)$$

195 Uncertainty for missing data (Brown et al., 2015) is given by Eq. (2):

$$Unc = 4 \times M \quad (2)$$

When the concentrations were above the detection limit, uncertainty is determined from Eq. (3):

$$Unc = ([EF \times EC]^2 + [0.5 \times DL]^2)^{1/2} \quad (3)$$

Where  $EF$  is the error fractions and  $EC$  is the element concentration.

200 Q robust value ( $Q_R$ ) is the goodness-of-fit parameter computed with the exclusion of points not fitted by the model. To evaluate the number of factors,  $Q_R$  was compared to  $Q_T$  ( $Q$  theoretical value). At the point when changes in the ratio  $Q_R/Q_T$



become smaller with the increase of the number of factors, it can be demonstrative that there might be an excessive number of factors being fitted (Brown et al., 2015).  $Q_T$  was estimated as in Lang et al. (2015), given by Eq. (4):

$$Q_T = (n_s \times n_e) - ([n_s \times n_f] + [n_e \times n_f]) \quad (4)$$

205 Where  $n_s$  is the number of samples,  $n_e$  is the number of *strong* elements, and  $n_f$  is the number of factors.

## 23 Results and discussions

### 3.1 Concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> during extensive campaigns

210 The extensive campaigns (Ext<sub>2.5</sub> and Ext<sub>10</sub>) were carried out over a whole year, during which the meteorological conditions varied largely. The average temperature during the sampling days in all campaigns ranged from 14 to 26 °C and the wind speed varied between 0.6 and 2.6 m s<sup>-1</sup>; most of the sampling was carried out on days without rainfall. In Fig. 2 ~~is presented~~ the meteorological variables, PM<sub>10</sub> and PM<sub>2.5</sub> concentrations ~~s~~ for all analyzed days ~~are presented~~.

215 There ~~were~~ ~~a~~ moderate negative correlations ~~between PM<sub>10</sub>, PM<sub>2.5</sub> and minimum relative humidity and average wind speed, average and minimum relative humidity; and between PM<sub>2.5</sub> and average wind speed and minimum relative humidity~~ (Table S1). This observation is in agreement with the fact that days with lower relative humidity and lower wind speed present higher PM<sub>2.5</sub> and PM<sub>10</sub> levels than in more humid and with more ventilation conditions.

220 In the extensive campaign, the PM mass concentrations exhibited a wide range of concentrations. For example, Ext<sub>2.5</sub> ranged from 8 to 78 µg m<sup>-3</sup> (average 30 µg m<sup>-3</sup>), whereas Ext<sub>10</sub> values varied between 12 and 113 µg m<sup>-3</sup> (average 44 µg m<sup>-3</sup>) (Fig. 3). The World Health Organization (WHO) recommends a daily limit for PM<sub>10</sub> of 50 µg m<sup>-3</sup> and of 25 µg m<sup>-3</sup> for PM<sub>2.5</sub>, (WHO, 2006) while the Brazilian Environmental Agency (CONAMA) recommends a threshold of 150 µg m<sup>-3</sup> for PM<sub>10</sub> (CONAMA, 1990; Pacheco et al., 2017). ~~In our study, 50 % of the Ext<sub>2.5</sub> and 30 % of the Ext<sub>10</sub> samples were above the guidelines recommended by WHO.~~ When considering the CONAMA standards, only one day in the extensive campaign was near the target limit. The Ext<sub>10</sub> campaign was divided into two periods: dry (April to September) and rainy (October to March). It was observed that the average PM<sub>10</sub> was 52 µg m<sup>-3</sup> in the dry period ~~(above the guideline recommended by WHO)~~ and of 35 µg m<sup>-3</sup> in the rainy period ~~(below the same guideline)~~.

225 A study done by Vasconcellos et al. (2011b) about a decade ago (2003/2004) in the city, showed a similar average of PM<sub>10</sub> (46 µg m<sup>-3</sup>). According to CETESB (São Paulo State Environmental Agency), the annual average PM<sub>10</sub> concentrations (considering all monitoring stations in the São Paulo Metropolitan Area) ranged from 33 to 41 µg m<sup>-3</sup>, between the years of 2005 and 2014, showing no significant differences (CETESB, 2015).

230 The average values for PM<sub>2.5</sub> were higher than those obtained in a year study done in traffic sites in two European ~~cities~~ ~~metropolis~~: London and Madrid in 2005 (warm period: 19.40 and 20.63 µg m<sup>-3</sup> for PM<sub>2.5</sub>, respectively) (Kassomenos et al., 2014). The European Union has a more restrictive control of pollutant emissions ~~compared to Latin American countries~~, since an annual mean of 40 µg m<sup>-3</sup> is established for PM<sub>10</sub> and a limit value of 25 µg m<sup>-3</sup> is imposed for PM<sub>2.5</sub> (Kassomenos et al., 2014). However, these averages in São Paulo are lower than ~~in some~~ ~~the~~ ~~observed~~ ~~in~~ year-round studies performed ~~at~~ ~~sites~~ ~~from~~ ~~in~~ Chinese megacities, such as Shanghai (83 µg m<sup>-3</sup> for PM<sub>2.5</sub> and 123 µg m<sup>-3</sup> for PM<sub>10</sub>) and Nanjing (222 µg m<sup>-3</sup> for PM<sub>2.5</sub> and 316 µg m<sup>-3</sup> for PM<sub>10</sub>) (Shi et al., 2015; Wang et al., 2003, 2013a). Indeed, in 2014, Zheng et al. (2016) assessed the PM<sub>2.5</sub> concentrations in 161 Chinese cities reporting an annual average concentration of 62 µg m<sup>-3</sup>.

235 In this study ~~it~~ was found that, on average, more than 60 % of the total mass PM is within the PM<sub>2.5</sub>; it is consistent with a previous study done at this site (dry season, 2008) when this value was 69 % (Souza et al., 2014a). On the other hand, in a two-year study conducted in 10 urban sites in Rio de Janeiro, the coarse fraction represented from 60 to 70 % of the PM<sub>10</sub> mass concentration (Godoy et al., 2009). The PM<sub>2.5</sub>/PM<sub>10</sub> ratio found in other urban Brazilian sites with different

characteristics (biomass burning, coastal environment) were close to 40 %, considerably lower than in São Paulo Metropolitan Area, according to the local environmental agency (CETESB, 2015), highlighting the importance of fine particulate matter over São Paulo city aerosol.

### 245 3.2 Concentrations of PM<sub>2.5</sub> during intensive campaign

The winter campaign began with high PM<sub>2.5</sub> concentrations (a maximum of 88 µg m<sup>-3</sup> on 02 July) and low relative humidity (minimum of 21 %). The average temperatures ranged from 15 to 21 °C and the wind speed ranged between 0.6 and 2.2 km h<sup>-1</sup>. The concentrations of PM<sub>2.5</sub> in the intensive campaign ranged from 15 to 88 µg m<sup>-3</sup> (average 45 µg m<sup>-3</sup>), with a similar average to that obtained in another intensive study in 2008, 47 µg m<sup>-3</sup> (Souza et al., 2014a). The levels of PM<sub>2.5</sub> in this  
250 campaign were above the recommended by WHO in 90 % of the sampling days.

The average concentration of PM<sub>2.5</sub> was higher in Int<sub>2.5</sub> than in Ext<sub>2.5</sub>, which can be explained by the fact that the campaign took place in the dry season (winter). In winter, the meteorological conditions are more unfavorable to the dispersion of pollutants and also due to the predominance of sugarcane burning (da Rocha et al., 2005, 2012; Sánchez-Ccoyllo and Andrade, 2002; Vasconcellos et al., 2010).

### 255 3.3 WSI and trace elements

The WSI represent a large fraction in the aerosol mass and have already been suggested to present ability to form CCN (cloud condensation nuclei) and fog (Rastogi et al., 2014). The secondary inorganic components, sulfate, nitrate and ammonium (SNA) were the most abundant ions in all campaigns (Table 3), which has already been observed in previous studies for this site (Vasconcellos et al., 2011a). SNA accounted for 74, 82 and 79 % of the total mass of inorganic species in  
260 the Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub> campaigns, respectively. The SNA were also found to be the major portion of the WSI in other studies around the world. For instance, Zheng et al. (2016) assessed PM<sub>2.5</sub> concentrations at 17 diversified sites in China. An average contribution of SNA of more than 90 % of total ions was obtained, which represented 50 % of PM<sub>2.5</sub>. The levels of SNA in aerosols from urban sites are highly influenced by the anthropogenic emissions of precursors (SO<sub>2</sub>, NO<sub>x</sub>, and NH<sub>3</sub>) (Wang et al., 2013b), although they may also be directly emitted for different sources, such as automobile or industrial  
265 sources.

Sulfate average concentrations appear to vary less than nitrate comparing Int<sub>2.5</sub> to Ext<sub>2.5</sub> and were not statistically different ( $p \sim 0.8$ ). The same trend in sulfate in this study, was also observed by Villalobos et al. (2015) for Santiago, Chile, 2013. In that study, the annual average concentration of sulfate (2000 ng m<sup>-3</sup>) is considerably lower than that observed in São Paulo extensive campaigns in 2014. The sulfate concentrations in Santiago aerosols have reduced since air quality regulations limited to 15 ppm the sulfur content in diesel and gasoline (MMA, 2014). In Brazil, since 2013 the S-10 diesel (10 ppm of  
270 ~~Sulphur~~Sulfur) substituted the S-50 diesel (50 ppm), whereas in 2014 the S-50 gasoline replaced the S-800 gasoline (800 ppm), although older vehicles are still allowed to use S-500 (500 ppm) diesel (CETESB, 2015). During the studies done in several urban sites in China, sulfate concentrations varied between 4200 and 23000 ng m<sup>-3</sup>. These values are higher than those of this study and also 5 to 10 times higher than the measured concentrations in Europe and United States (Hidy, 2009; Putaud et al., 2004; Zheng et al., 2016).

The SO<sub>4</sub><sup>2-</sup>/NO<sub>3</sub><sup>-</sup> ratio was nearly twice higher in the Ext<sub>2.5</sub> campaign than in Int<sub>2.5</sub>. ~~If-This pattern~~ has already been observed, in warmer ambient conditions the fine NO<sub>3</sub><sup>-</sup> aerosols can be volatilized, increasing the ratio between these species (Rastogi and Sarin, 2009; Souza et al., 2014a). NH<sub>4</sub>NO<sub>3</sub> exists in a reversible equilibrium between HNO<sub>3</sub> and NH<sub>3</sub> (Tang et al., 2016). Ammonium concentrations were not significantly different ( $p \sim 0.3$ ) and were slightly higher in Int<sub>2.5</sub> (1712 ng m<sup>-3</sup>) than in  
280 Ext<sub>2.5</sub> (1370 ng m<sup>-3</sup>). ~~Ratio  $\sum cations / \sum anions$  (ion balance) was calculated similarly to what was done in the previous study~~

~~by da Rocha et al. (2012). In the present study, the ratios were lower than 1.0 in all the campaigns, and considerably higher in Int<sub>2.5</sub>. It is suggested that it occurred due to the lack of other cationic species data.~~

Non-sea-salt potassium was calculated (nss-K<sup>+</sup>) based on seawater ion ratios [ss-K<sup>+</sup>] = 0.036 [Na<sup>+</sup>] (Nayebare et al., 2016; Seinfeld and Pandis, 2006). Concentrations of nss-K<sup>+</sup> were significantly higher in Int<sub>2.5</sub> than in Ext<sub>2.5</sub>, with average concentrations of 809 and 366 ng m<sup>-3</sup>, respectively ( $p < 0.01$ ). The higher concentrations found in the intensive campaigns have already been attributed to biomass burning in previous studies (Pereira et al., 2017; Vasconcellos et al., 2011a). However, potassium ions can also come from soil resuspension (Ram et al., 2010; Tiwari et al., 2016), which becomes important in PM<sub>2.5-10</sub>. Higher concentrations of chloride in fine particles (964 and 330 ng m<sup>-3</sup> for Int<sub>2.5</sub> and Ext<sub>2.5</sub>, respectively) were observed in campaign Int<sub>2.5</sub> (although the value of  $p$  was slightly above 0.05), probably due to a higher influence of biomass burning (Allen et al., 2004). On the other hand, chloride in coarse particles is mostly attributed to marine aerosols. Cl<sup>-</sup>/Na<sup>+</sup> ratios were below 1.8 in Ext<sub>2.5</sub> and Ext<sub>10</sub> and higher in Int<sub>2.5</sub>; although Cl<sup>-</sup>/Na<sup>+</sup> ratios are attributed to increased sea salt contribution (Souza et al., 2014a), the higher contribution in the intensive campaign may be explained by a higher contribution of other sources of chloride in that period, such as biomass burning.

Pearson correlations were obtained for all determined species in Ext<sub>2.5</sub> and Ext<sub>2.5-10</sub> (coarse mode), including meteorological data such as temperature, relative humidity and wind speed, some gaseous species such NO<sub>x</sub> and CO, were obtained from CETESB database and were also included. NH<sub>4</sub><sup>+</sup> was from moderately to strongly correlated with C<sub>2</sub>O<sub>4</sub><sup>2-</sup> (oxalate), Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> in Ext<sub>2.5</sub> (R = 0.66, 0.62, 0.85 and 0.79, respectively) suggesting the neutralization of oxalic, hydrochloric, nitric and sulfuric acids by NH<sub>3</sub> (Table S2). The formation of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, a non-volatile species, could represent a gas-to-particle process and can account for the formation of new particles through nucleation (Mkoma et al., 2014; da Rocha et al., 2005) and can lead to CCN (cloud condensation nuclei) formation. NH<sub>4</sub>NO<sub>3</sub> and NH<sub>4</sub>Cl also have an important influence on Earth's acid deposition (Tang et al., 2016).

Na<sup>+</sup> was strongly correlated with Cl<sup>-</sup> in Ext<sub>2.5</sub> (R = 0.78) and ~~presented~~ had relatively ~~moderate~~ higher correlations ~~with this species~~ (R = 0.35) in ~~the coarse fraction~~ Ext<sub>2.5-10</sub>. These species are often associated with marine aerosol, which is mainly in the coarse mode (Godoy et al., 2009; da Rocha et al., 2012). Although it was observed that ocean influence is not the only source of Na<sup>+</sup> in the site, this species may have vehicular sources (Vieira-Filho et al., 2016). C<sub>2</sub>O<sub>4</sub><sup>2-</sup> was also moderately correlated with NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup> and K<sup>+</sup> (R = 0.67, 0.61 and 0.68, respectively), this species reported sources can be biomass burning and secondary conversion of natural and anthropogenic gases (Custódio et al., 2016). The secondarily formed species were negatively correlated with wind speed (from R = -0.40 to R = -0.70); lower wind speed can increase of the formation of secondary ionic species due to an increase the precursor species concentrations (Yu et al., 2016).

Average, maximum and minimum trace element concentrations are presented in Table 4. Mg, Al, K, Ca, Fe, Cu and Zn were the most abundant elements in all campaigns, similar to those observed by Vasconcellos et al. (2011a) for the intensive campaign in 2008. All of them had higher concentrations in Int<sub>2.5</sub> than in Ext<sub>2.5</sub>. However, crustal elements were significantly higher; Al and Ca had concentrations nearly 3 times higher in intensive campaign ( $p < 0.05$ ). A similar trend was observed between wintertime and summertime campaigns by Castanho and Artaxo (2001). They reported higher concentrations of soil resuspension elements during wintertime. An increase in soil resuspension is expected in drier conditions.

As observed for nss-K<sup>+</sup>, elemental K average concentration was more than twice higher in Int<sub>2.5</sub> than Ext<sub>2.5</sub> ( $p < 0.05$ ). This may be explained by a higher biomass burning contribution during the intensive campaign since sugar cane burning is significantly increased in this time of the year. Cu has been attributed to vehicular emissions in São Paulo (Castanho and Artaxo, 2001), because it may be present in the ethanol, which is mixed with gasoline and used in light-duty vehicles in Brazil. Cu has also been related to wear emissions of road traffic (Pio et al., 2013). This element was approximately 70 % higher in Int<sub>2.5</sub> than Ext<sub>2.5</sub>. Although there is no significant difference in vehicular emissions all year round, the meteorological conditions are more unfavorable to pollutant dispersion in winter season.

Enrichment factor (EF) is an approximation often used in order to identify the degree to which an element in an aerosol is enriched or depleted regarding a specific source. EFs are calculated based on a reference metal (Al as a soil tracer in this study), considering crustal element composition (Lee, 1999). A convention often adopted is considering that when elements have EFs below 10 they have significant crustal source and are often called non-enriched elements (NEEs) and when the elements have EFs above 10 they have a higher non-crustal character and are referred as anomalously enriched elements (AEEs) (Pereira et al., 2007). Values were higher than 10 for Cr (except for Int<sub>2.5</sub>), Cu, Zn, As, Se, Cd, Sn, Tl, Pb and Bi, meaning that they can be attributed to anthropogenic sources as vehicular and industries emissions (Table S3). Elements like K, Mn, Ni, Rb, Sr, Cs, Li, Mg, Ca, Fe, Co and Sr had EFs lower than 10 and could be attributed to soil resuspension (da Rocha et al., 2012).

Strong correlations were observed between Al and Li, Mg, K, Ca, Mn, Fe, Rb and Sr ( $R > 0.85$ ) in Ext<sub>2.5</sub>. Al also had strong correlations with Li, Mg, K, Ca and Fe in Ext<sub>2.5-10</sub> ( $R > 0.70$ ). Strong correlations were observed between species like Cl<sup>-</sup> and NO<sub>3</sub><sup>-</sup> with Mg, Al, Ca and Fe ( $R > 0.7$ ); atmospheric reactions can occur between acids (HCl and HNO<sub>3</sub>) and soil particles that have alkaline character (Rao et al., 2016). Mg, Al, K, Sr and Fe were negatively correlated with relative humidity ( $R \leq -0.60$ ), suggesting strong influence of drier conditions over these species.

### 3.4 Carbonaceous species and mass balance

Higher concentrations of OC and EC were observed in Int<sub>2.5</sub> than in Ext<sub>2.5</sub> with average values of 10.2  $\mu\text{g m}^{-3}$  for OC and 7.0  $\mu\text{g m}^{-3}$  for EC (Fig. 4 and Table S4). However, the difference of carbonaceous species concentrations was not considered statistically significant between the campaigns ( $p \sim 0.1$ ). The OC/EC ratios were 1.5, 1.7, and 1.8 for Int<sub>2.5</sub>, Ext<sub>2.5</sub>, and Ext<sub>10</sub>, respectively. Since the ratio values were similar, as well as the absolute OC and EC concentrations were higher in intensive than extensive campaigns, this may indicate similar sources of OC and EC are contributing all year long but with higher concentrations during Int<sub>2.5</sub>. Ratios lower than 1 are constantly observed in roadway tunnels and are assumed to describe the composition of fresh traffic emissions (Pio et al., 2011). ~~(Amato et al., (2016) found values ranging from 1.8 and 3.7 at the urban background sites using equivalent measurement protocols. It was attributed to the distance from main roads, which can increase the influence of secondary OC (Pio et al., 2011). Ratios ranging from 2 to 5 are commonly observed in urban background atmospheres and are assumed to indicate a significant contribution of secondary aerosol sources (Pio et al., 2011; Querol et al., 2013; Viana et al., 2007).~~ In this way, the values for OC/EC found in the present study may be due to vehicle emissions with contribution of secondary organic aerosols.

TOM (total organic matter) was calculated by multiplying the organic carbon content by 1.6 (Timonen et al., 2013) and represented 36, 36 and 28 % of the total PM, respectively (Fig. S2). Mass balance was determined for the aerosol considering trace elements as if they all existed as oxides (Alves et al., 2015). The unaccounted part was of 6, 15 and 26 % for Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub>, respectively. This unaccounted part can be attributed to adsorbed water or the fact that abundant species as carbonates and Si were not determined, similarly as observed in Pio et al. (2013).

OC and EC were well correlated in Ext<sub>2.5</sub>, with values above 0.8. This suggests that a large amount of OC is emitted by a dominant primary source at this site (Aurela et al., 2011; Kumar and Attri, 2016). The studied site is strongly affected by vehicle emissions and, during the winter months biomass burning also contributes to these species (Pereira et al., 2017). Correlations were strong between the carbonaceous species with vehicular emitted gases such as NO<sub>x</sub> and CO ( $R > 0.85$ ). OC also had good correlations with soil elements (Mg and Al) and also nss-K<sup>+</sup> ( $R > 0.8$ ), suggesting association with the resuspension of road dust and also a significant biomass burning contribution.

#### ~~3.4.13.5.1~~ Polycyclic aromatic hydrocarbons and derivatives

The PAH and derivatives concentrations are presented in Table 5. The total PAHs were higher in Int<sub>2.5</sub> than in Ext<sub>2.5</sub>, 23.3 ng m<sup>-3</sup> and 18.4 ng m<sup>-3</sup>, respectively (although not significantly different, with  $p > 0.05$ ). The total PAH concentration for the Ext<sub>10</sub> was 24.3 ng m<sup>-3</sup>. The lowest total PAH concentration of 2.6 ng m<sup>-3</sup> was observed in Ext<sub>2.5</sub>, while the maximum of 115.3 ng m<sup>-3</sup> was observed in Ext<sub>10</sub>. These levels were similar to those obtained in past studies at the same site, as 25.9 ng m<sup>-3</sup> in PM<sub>10</sub> samples during the intensive campaign in the winter of 2008 (Vasconcellos et al., 2011a) and 27.4 ng m<sup>-3</sup> for PM<sub>10</sub> in the winter of 2003 (Vasconcellos et al., 2011b). In addition, the total PAH levels from the present study is higher than in 2013 and 2012 intensive campaigns (8.7 ng m<sup>-3</sup> and 8.2 ng m<sup>-3</sup> in PM<sub>10</sub>) (Pereira et al., 2017). Total PAHs represented 0.23, 0.27 and 0.31 % of OC for Int<sub>2.5</sub>, Ext<sub>2.5</sub>, and Ext<sub>10</sub>, respectively. In spite of accounting for a small fraction of organic carbon, it is important to observe that PAHs are among the pollutants of major concern due to their carcinogenic and mutagenic effects.

BbF was the most abundant PAH (the BbF percentages in relation to total PAHs were 13, 12, 12 % for Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub>, respectively) in all the campaigns. This compound has carcinogenic properties already reported in other studies (Ravindra et al., 2008). Its concentrations reached the values of 6.1, 6.4 and 13.3 ng m<sup>-3</sup> in Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub>. BbF was also the most abundant PAH in 2013 intensive campaign (Pereira et al., 2017). This species was also among the most abundant PAHs in the study performed at Jânio Quadros tunnel, with a predominance of light-duty vehicles (Brito et al., 2013). Correlations were strong between all PAHs heavier than Flt ( $R > 0.8$ ); suggesting different sources from the PAHs with lower molecular weight at this site. Most of the heavier PAHs appeared to have negative correlations with temperature; the condensation of organic compounds in the aerosol is influenced by lower temperatures (Bandowe et al., 2014). Coronene, a PAH often used as a vehicular fuel marker (Ravindra et al., 2006) was correlated to vehicular related species as Cu and Pb ( $R > 0.7$ ).

BaP, the PAH most studied due to its proven carcinogenic potential, was considerably higher in Int<sub>2.5</sub> than in Ext<sub>2.5</sub>. It reached the mean values of 5.5, 7.6 and 12.5 ng m<sup>-3</sup> in Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub>, respectively. In the tunnels, its presence was associated with the higher contribution of LDV emissions (Brito et al., 2013). 2-NFlu and 2-NBP were among the nitro-PAHs with highest concentrations. 2-NFlu is a major component of diesel exhaust particles, such as the nitropyrenes, and is known as a carcinogenic nitro-PAH (Draper, 1986; Fujimoto et al., 2003). 2-NFlt was moderately correlated with Flt ( $R = 0.4$ ); this species is produced from reactions between Flt and NO<sub>2</sub> (Albinet et al., 2008). The ratios 2-NFlt/1-NPyr were close to 1; ratios lower than 5 indicate a predominance of primary emissions of nitro-PAHs (Ringuet et al., 2012). The compound 9,10-AQ was the most abundant oxy-PAH found in this study. It can be either primarily emitted or secondarily formed. A recent study showed that it can be formed from the heterogeneous reaction between NO<sub>2</sub> and Ant adsorbed on NaCl particles (sea salt) (Chen and Zhu, 2014). A moderate correlation was found between 9,10-AQ and Ant ( $R = 0.54$ ).

### **3.4.23.5.2 PAH diagnostic ratios**

The PAH diagnostic ratios (Table 5) were obtained for all the campaigns, since they can point to some emission sources, such as oil products, fossil fuels, coal or biomass combustion. However, these ratios should be used with caution, due to the peculiarity of fuel compositions in Brazilian's car fleet. The values of PAH ratios can also be affected by changes of phase, transport and degradation (Tobiszewski and Namieśnik, 2012). The ratio BaP/(BaP+BeP) is related to the aerosol photolysis. Most of the local PAH emissions contain equal concentrations of BeP and BaP. However, BaP is more likely to undergo photolysis or oxidation (Oliveira et al., 2011). The average BaP/(BaP+BeP) was close to 0.4 for the three campaigns. This ratio was slightly lower than the ratio obtained in the 2013 intensive campaign, although still very close to 0.5 (Pereira et al., 2017); it is suggested that the PAHs found in the site are mostly emitted locally.

The Flt/(Flt+Pyr) and InP/(InP+BPe) ratios were reported to be the most conservative by Tobiszewski and Namieśnik (2012). The Flt/(Flt+Pyr) ratio for all the campaigns were close to 0.5, falling within the range for fossil fuel combustion

(0.4–0.5) (de la Torre-Roche et al., 2009). The ratio InP/(InP+BPe) represented values close to 0.5, similar to the ratio obtained for JQ tunnel (0.55) impacted by LDV (the ratio found for MM tunnel was 0.36). The average BaA/Chr ratio ranged between 0.5–0.6 in the 2014 campaigns, also approaching that of JQ tunnel (0.48) (Brito et al., 2013), whilst a value of 0.79 was obtained for MM tunnel. The BaA/(BaA+Chr) ratio was reported to be sensitive to photodegradation (Tobiszewski and Namieśnik, 2012). However, it is possible to consider that this degradation was not significant due to proximity to the emission sources (the expressway). All ratios suggested a greater contribution of LDV to PAHs at the sampling site.

The  $\Sigma$ LMW/ $\Sigma$ HMW ratios (LMW - Low Molecular Weight PAHs with three and four aromatic rings and HMW - High Molecular Weight PAHs with more than four rings) were considerably low in all campaigns (predominance of HMW PAHs). It is known that LMW PAHs have higher concentrations in the gas phase while HMW PAHs are preferentially present in PM (Agudelo-Castañeda and Teixeira, 2014; Duan et al., 2007). The HMW PAHs contribution was higher in winter, just as the ratios were lower, corroborating the results of some previous studies (Chen et al., 2016; Teixeira et al., 2013). In turn, HMW PAHs are more likely to be retained in particles due to its lower vapor pressure than LMW PAHs. LMW PAHs are also mostly associated with diesel engines, while HMW PAHs are predominantly emitted by gasoline exhaust (Chen et al., 2013; Cui et al., 2016; Miguel et al., 1998). The ratio between BPe and BaP for all campaigns (all close to 1) was very similar to that found in a study with Brazilian light-duty diesel vehicles exhaust (1.13) (de Abrantes et al., 2004) and also found in a campaign performed in São Paulo (1.11) (de Martinis et al., 2002). This may be a characteristic fingerprint for local vehicular emissions.

### 3.4.33.5.3 PAHs risk assessment

BaPE is a parameter introduced to quantify the aerosol carcinogenicity related to all carcinogenic PAHs instead of BaP solely. BaPE values above 1.0 ng m<sup>-3</sup> represent an increased cancer risk. The carcinogenic nitro-PAHs (1-NPyr, 4-NPyr, 6-NChr) were below the detection limit in most part of the extensive campaign samples, so they were not considered in the risk assessment. BaPE is calculated according to Eq. (5), given by Yassaa et al. (2001) and Vasconcellos et al. (2011a) :

$$BaPE = ([BaA] \times 0.06) + ([BbF] \times 0.07) + ([BkF] \times 0.07) + ([BaP] \times 1) + ([DBA] \times 0.6) + ([InP] \times 0.08) \quad (5)$$

The BaPE values for the Int<sub>2,5</sub> ranged between 0.6 and 8.0 ng m<sup>-3</sup> and for Ext<sub>2,5</sub>, between 0.3 and 10.5 ng m<sup>-3</sup>, while the average of BaPE for the Int<sub>2,5</sub> was considerably higher than in Ext<sub>2,5</sub> (3.4 and 2.4 ng m<sup>-3</sup>, respectively). In the Ext<sub>10</sub> this index ranged between 0.5 and 18.3 ng m<sup>-3</sup>. The maximum value was even higher than the value of 12.1 ng m<sup>-3</sup> in PM<sub>10</sub>, obtained in São Paulo in an intensive campaign conducted in 2008 (Vasconcellos et al., 2011a). More than 70 % of the samples in the Ext<sub>10</sub> had BaPE indexes higher than 1 ng m<sup>-3</sup>. The year 2014 was a relatively dry year, with an annual rainfall 13 % below the average (IAG, 2014). The average values for BaPE in PM<sub>10</sub> at the site were 1.9 and 3.7 ng m<sup>-3</sup> in the intensive campaigns of 2007 and 2008, respectively. On the other hand, at forested areas in São Paulo state the value can be as low as 0.1 ng m<sup>-3</sup> (Vasconcellos et al., 2010).

The lifetime lung cancer risks (LCR) were assessed from the carcinogenic potential (BaP-TEQ) and mutagenic potential (BaP-MEQ), through equations (6) and (7) (Jung et al., 2010):

$$(BaP - TEQ) = ([BaA] \times 0.1) + ([Chr] \times 0.01) + ([BbF] \times 0.1) + ([BkF] \times 0.1) + ([BaP] \times 1) + ([InP] \times 0.1) + ([DBA] \times 5) + ([BPe] \times 0.01) \quad (6)$$

$$(BaP - MEQ) = ([BaA] \times 0.082) + ([Chr] \times 0.017) + ([BbF] \times 0.25) + ([BkF] \times 0.11) + ([BaP] \times 1) + ([InP] \times 0.31) + ([DBA] \times 0.29) + ([BPe] \times 0.19) \quad (7)$$

DBA had the largest contribution to carcinogenic potential and BaP for mutagenic potential, in studies performed in Italian urban areas BaP was the compound that most contributed to total carcinogenicity in PM, although the TEF used for DBA was lower in that cases. (Cincinelli et al., 2007; Gregoris et al., 2014). LCR from exposure to atmospheric PAH was estimated by multiplying BaP-TEQ and BaP-MEQ by the unit risk ( $87 \times 10^{-6} \text{ (ng m}^{-3}\text{)}^{-1}$ ) for exposure to BaP established by WHO (de Oliveira Alves et al., 2015; WHO, 2000) (Fig. 5) and was possible to observe an increase during the intensive campaign. In all campaigns, the values observed were higher than the observed in studies done in the Amazon during dry season with events of biomass burning (de Oliveira Alves et al., 2015); studies done at different seasons in other urban areas as New York and Madrid pointed carcinogenic risks within the recommended by environmental and health agencies (Jung et al., 2010; Mirante et al., 2013).

### **3.53.6 Biomass burning tracers**

The highest concentrations of biomass burning tracers (levoglucosan, means  $509 \text{ ng m}^{-3}$ ; mannosan  $45 \text{ ng m}^{-3}$ ; galactosan  $33 \text{ ng m}^{-3}$ ) were observed in the  $\text{Int}_{2.5}$  ( $p \sim 0.05$ ), during the biomass burning period (Fig. 6 - Table S5). In the intensive campaign period, 1364 fire spots were registered in São Paulo state, with an average of 72 fires per day (INPE, 2014). In the same way, 65 % of the sampling days the backward air masses passed through regions with biomass burning. The average concentration of levoglucosan obtained in the  $\text{Int}_{2.5}$  ( $509 \text{ ng m}^{-3}$ ) were higher than those of the intensive  $\text{PM}_{10}$  sampling campaigns in 2013 and 2012 ( $474 \text{ ng m}^{-3}$  and  $331 \text{ ng m}^{-3}$ ) (Caumo et al., 2016; Pereira et al., 2017), as well as more than twice than the values obtained in the 2008 intensive campaign (Vasconcellos et al., 2010).

The Lev/Man ratios are characteristic of each type of biomass. The ratios were similar to that obtained in a chamber study with sugarcane burning -in Florida (Lev/Man = 10) (Hall et al., 2012), and also to that reported for the 2013 intensive campaign (Lev/Man = 12) (Pereira et al., 2017). Nss- $\text{K}^+$ /Lev ratios were 1.6, 1.4 and 1.3 for  $\text{Int}_{2.5}$ ,  $\text{Ext}_{2.5}$  and  $\text{Ext}_{10}$ , respectively. These ratios are similar to those obtained in the previous  $\text{PM}_{10}$  intensive campaign (1.4) in 2013, which was attributed to a combination of smouldering (flameless combustion) and flaming processes during the combustion of biomass (Kundu et al., 2010; Pereira et al., 2017). The flaming combustion is predominant for sugarcane leaves (Hall et al., 2012; Urban et al., 2016).

Correlations between potassium and monosaccharides in  $\text{Ext}_{2.5}$  were high ( $R > 0.8$ ), indicating that, most of the year, potassium in  $\text{PM}_{2.5}$  can be linked to biomass burning. Coarse fraction potassium, more related to soil sources, did not present strong correlations with levoglucosan. Local burning also can affect the site since some restaurants consume wood for cooking (pizzerias and steakhouses) (Kumar et al., 2016). There is a stronger correlation between chloride and other biomass burning tracers in  $\text{Ext}_{2.5}$  than in  $\text{Ext}_{2.5-10}$ . Chloride is also a major emission from biomass burning, in the form of KCl (Allen et al., 2004) and is also emitted as HCl in garbage burning (Calvo et al., 2013). Carbonaceous species presented high correlations ( $R > 0.75$ ) with levoglucosan in  $\text{Ext}_{2.5}$ . This suggests that some of these species may be also linked to biomass burning emissions.

During the intensive campaign, the backward trajectories pointed to air masses passing by regions affected by biomass burning on 65 % of the sampling days. The highest concentrations of biomass burning tracers were found on 1<sup>st</sup> of July, when the levoglucosan level reached  $1263 \text{ ng m}^{-3}$ . On that day, about 100 fire spots (INPE, 2014) were observed in the state of São Paulo and the back trajectories revealed air masses crossing the West and Northwest of the state (Fig. 7a), where the fire spots were observed. In this same sampling day, local fire spots were observed, possibly landfill burning.

On 12<sup>th</sup> of July, the air masses travelled through the Atlantic Ocean before reaching the site. In the same period, the  $\text{PM}_{2.5}$  and biomass burning tracers concentrations dropped. Figure 7b shows the trajectories for 13<sup>th</sup> of July. Some of the lowest

concentrations of levoglucosan (80 and 74 ng m<sup>-3</sup>) and PM<sub>2.5</sub> (28 and 26 µg m<sup>-3</sup>) were observed on 12<sup>th</sup> and 13<sup>th</sup> of July, respectively.

### **3.6.3.7 Distribution of species in fine and coarse particles during extensive campaigns**

Figure 8 shows the mass percentage of tracers in fine (PM<sub>2.5</sub>) and coarse particles (PM<sub>2.5-10</sub>) in the extensive campaign; their values are presented in SI (Table S6). The biomass burning tracers, levoglucosan and mannosan were present mostly in PM<sub>2.5</sub> mass fractions (over 75 %). In this study, 73 % of nss-K<sup>+</sup> mass was in PM<sub>2.5</sub>. This species may also be attributed to biomass burning, although coarse potassium may be from soil dust resuspension (Souza et al., 2014a; Vasconcellos et al., 2011a).

Species related to vehicular emissions as coronene and Cu (Brito et al., 2013; Ravindra et al., 2006) were also predominantly found in PM<sub>2.5</sub> (73 and 61 %, respectively). More than 50 % of Fe and Ca, crustal elements, was found in PM<sub>2.5-10</sub>. A previous source apportionment study in Southern European cities (AIRUSE-LIFE+ project) also pointed out soil dust as a significant source, accounting for 2–7 % of PM<sub>2.5</sub> at suburban and urban background sites and 15 % at a traffic impacted station. In the case of PM<sub>10</sub>, these percentages increased to 7–12 % and 19 %, respectively (Amato et al., 2016).

In a previous winter campaign in 2008, in São Paulo (Souza et al., 2014a) levoglucosan and mannosan were also mostly present in PM<sub>2.5</sub>. Urban et al. (2014) found for an agro-industrial region in São Paulo state that between 58 and 83 % of levoglucosan was present in particles smaller than 1.5 µm. It is similar to the values observed in other studies done in the state of São Paulo and in the Amazon region (Decesari et al., 2006; Schkolnik et al., 2005; Urban et al., 2012).

Sulfate and ammonium were predominant in PM<sub>2.5</sub> (over 65% and 80%). Sulfate was also predominant in PM<sub>2.5</sub> in a previous study done in São Paulo between 1997 and 1998; it was attributed to the gas-to-particle conversion of vehicular SO<sub>2</sub> (Castanho and Artaxo, 2001). Both ions may be present as (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> in the fine mode. Nitrate is well distributed in both phases, likely resulting from reactions of HNO<sub>3</sub> with soil species (Tang et al., 2016). In turn, fine mode nitrate is often present in the form of ammonium nitrate, which is a thermally unstable species (Maenhaut et al., 2008).

Other ions, such as sodium and chloride were halved in each mode. These species are related to sea salt aerosols and are more often present in the coarse mode, as observed in the study done at urban sites in Rio de Janeiro (Godoy et al., 2009).

Chloride in PM<sub>2.5</sub> can also be originated by biomass burning emissions (Allen et al., 2004). OC and EC, which are mainly related to vehicular emissions in São Paulo (Castanho and Artaxo, 2001), were mostly in the fine particles. OC and EC were also associated with biomass burning in a recent study (Pereira et al., 2017).

BaP was found mainly in the PM<sub>2.5</sub> (over 80 %), which are able to be deposited in the tracheobronchial region of the human respiratory tract, representing an increased health risk (Sarigiannis et al., 2015). In turn, As, Cd and Pb, identified as elements that can cause carcinogenic health effects (Behera et al., 2015), were also found predominantly in the PM<sub>2.5</sub> (over 75 %). In this way, they may be indicative of higher carcinogenicity of fine over coarse particles.

### **3.7.3.8 Source apportionment by PMF and polar plots**

Source apportionment was performed with PMF including all data. Then, the factor contributions were separated for each campaign ( $n = 78$ ). Eleven *strong* species were considered (SO<sub>4</sub><sup>2-</sup>, nss-K<sup>+</sup>, Mg, Cr, Mn, Fe, Ni, Cd, Pb, OC and EC), six were considered *weak* (Lev, Man, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, Ca and Cu) and the PM concentrations were set as a total variable. An extra modelling uncertainty of 25 % was added to the model.

Considering the limited number of samples, a restricted number of species had to be chosen. Elements already studied and attributed to sources in São Paulo were preferred. In some of the base model runs it was possible to observe a sea salt profile with Na<sup>+</sup> and Cl<sup>-</sup>, but after they were removed, other profiles were clearly improved. PAHs were firstly included in the



520 model but it created a factor associated with temperature conditions, increasing in the dry season since the lower dispersion conditions in the period favor the accumulation of HMW-PAHs in suspended particles (Agudelo-Castañeda and Teixeira, 2014; Ravindra et al., 2006) Lev and man were set as *weak* due to their organic character, and  $\text{NO}_3^-$  and  $\text{NH}_4^+$  due to their thermal instability. Ca and Cu also had to be set as *weak* in order to have a convergent base model run.

Solutions with three to eight factors were tested. The ratio of robust to theoretical parameters ( $Q_R/Q_T$ ) reduced between simulations when increasing the number of factors. A solution with five factors was found to have more meaningful results;  $Q_R$  and  $Q_T$  values were 367 (Table S7). The source profiles obtained in the PMF analysis and the contribution of each factor to  $\text{PM}_{10}$  concentrations are found in Fig. 9. Constraints were applied, Cu was pulled up maximally in the vehicular factor and Lev and Man were pulled up maximally in the biomass burning factor in order to have a better separation between both factors. relative change in Q was of 0.4 %. The PMF result charts are presented in Fig. S3.

530 Factor 1 presented higher loadings for Mg, Ca and Fe, elements associated with soil resuspension in previous studies (da Rocha et al., 2012). The factor was also mixed with vehicular related species, such as Cu and OC, which can be attributed to the resuspension of road dust by traffic. Accounting for 24.3, 12.5 and 25.7 % of  $\text{Int}_{2.5}$ ,  $\text{Ext}_{2.5}$  and  $\text{Ext}_{10}$ , respectively, it was the most important source more relevant for the  $\text{PM}_{10}$  campaign. In some runs, it was possible to observe Li and Tl in this factor, but these species were not considered in the final model. This soil contribution was similar to that obtained for  $\text{PM}_{10}$  in a year round inventory in the city (CETESB, 2015). High loadings for ions, such as  $\text{nss-K}^+$  and  $\text{NO}_3^-$ , were also present in the factor. Gaseous  $\text{HNO}_3$  can interact with soil particles and form coarse nitrates (Tang et al., 2016). The factor contribution appeared to increase with wind speed from NW and decrease with SE winds (Fig. 10). Soil dust and vegetation sources also tended to reduce with SE winds, as observed previously by Sánchez-Ccoyllo and Andrade (2002).

535 Factor 2 shows high loads for Ni, Pb and Cr, which are often attributed to industrial emissions (Bourotte et al., 2011; Castanho and Artaxo, 2001). This factor had some of the lowest contributions, 10.5, 9.7 and 9.5 % for  $\text{Int}_{2.5}$ ,  $\text{Ext}_{2.5}$  and  $\text{Ext}_{10}$  and appeared to increase with SE winds, passing through nearby industrial regions (southeast of the city). The growth of industries have been limited in the last years and vehicle fleet is expected to be a main source of atmospheric pollutants in the area (Kumar et al., 2016).

540 Factor 3 showed high loadings for vehicular related tracers, such as Cu, Fe, OC and EC (with a higher load on EC and Cu). Cu and Fe were found in the LDV impacted tunnel study in São Paulo and Cu is emitted from brake pads, in stop-and-go driving in the expressways (Andrade et al., 2012b; Brito et al., 2013) and also are present in ethanol after the processing of copper tanks. Loadings still relatively high in this were observed for levoglucosan and mannosan were observed in this factor, which precluded the total separation of from the biomass burning tracers factor. On days with NW winds, both source contributions tended to increase as observed in the polar plots. This factor represented 30.9, 39.1 and 39.2 % contribution for  $\text{Int}_{2.5}$ ,  $\text{Ext}_{2.5}$  and  $\text{Ext}_{10}$ , and had a constant contribution comparing dry and wet period in the  $\text{Ext}_{10}$  campaign. Vehicular source seemed to increase with winds coming from the North and Northwest, passing by the expressway, but decreased with SE winds, as observed previously (Sánchez-Ccoyllo and Andrade, 2002). The polar plots profiles of VE and RD factors presented a different pattern, since the aerosol from road dust suspension has a larger aerodynamic diameter (Karanasiou et al., 2009) and tends to increase with wind speed.

550 Factor 4 was associated with biomass burning due to the loadings for levoglucosan, mannosan and non-sea-salt potassium, OC and EC. It is also noteworthy the loading of Cd in this factor; wood burning was pointed out as a possible source of this metal in a previous study in Belgium (Maenhaut et al., 2016), but more studies are necessary in order to explain the biomass burning contribution to this species in São Paulo. This factor represented 18.3, 11.6 and 7.6 % for  $\text{Int}_{2.5}$ ,  $\text{Ext}_{2.5}$  and  $\text{Ext}_{10}$ , respectively. The contributions of this factor were higher in the intensive campaign (sugarcane burning period), but were also present in the other periods, suggesting other biomass burning sources in the city, such as waste burning and wood stoves (Kumar et al., 2016). Several fire spots were registered in São Paulo state in the intensive campaign, some of them in

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the neighboring towns (INPE, 2014). The polar plot showed that this factor tended to increase with NW winds, passing through the inland of São Paulo state, and decrease with SE (more humid) winds from the sea. High correlations ( $R > 0.8$ ) were observed between the gases CO and NO<sub>x</sub> and the primary sources factors VE and BB (Table S8). These gases are related to vehicular emissions (Alonso et al., 2010) and the correlations with the biomass burning factors may be due to the fact that it increases with the same wind direction as the vehicular factor. No correlations were found between these gases and SP factor.

Factor 5 was attributed to the secondary inorganic aerosol formation processes (as seen by high mass loadings for NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup> and NH<sub>4</sub><sup>+</sup>) and also OC (secondary organic carbon). The contributions were 15.9, 27.1 and 17.9 % for Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub>, respectively. The contributions of this profile did not follow any seasonal trend (Fig. S3a). In 2014, 78 % of NO<sub>x</sub> and 43 % of SO<sub>x</sub> emissions in Greater São Paulo were attributed to the vehicle fleet (CETESB, 2015). Taking into account that in São Paulo SO<sub>x</sub> and NO<sub>x</sub> concentrations are similar all year round, this could explain the lack of seasonality of this factor. The polar plot showed a centralized profile, increasing with lower wind speed, which suggests a local secondary process.

Other polar plots were obtained for individual species and are presented in Fig. 11. It is possible to see that Na<sup>+</sup> tended to increase with stronger winds coming from the sea, while Cl<sup>-</sup> had a different pattern. Chloride in the marine aerosol can be depleted after atmospheric reactions with acids (Calvo et al., 2013; White, 2008). It is noteworthy that MSA was associated with NW winds, This species is often associated with the decomposition of DMS, emitted by the sea (Bardouki et al., 2003). More studies are needed in order to identify MSA sources on this site. Similarly, as for biomass burning factor, levoglucosan tended to increase with NW winds. However, it is also possible to observe local sources for this species due to its high levels, even with lower speed wind. Secondarily formed species such as NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> had a much-centered profile and tended to increase with lower wind speed. EC, Chr and Cor seemed to be emitted by local sources, likely vehicular emissions (Alves et al., 2016; Ravindra et al., 2008). On the other hand, Flt (a light molecular mass PAH), seemed to be influenced by different air masses, suggesting different sources.

#### 4 Summary and conclusions

Particulate matter (PM<sub>2.5</sub> and PM<sub>10</sub>) was collected throughout the year 2014 to determine different chemical constituents, including carbonaceous species, WSI, monosaccharides, PAHs, and their derivatives. The risks of PAHs for human health were assessed with levels exceeding the suggested guidelines. High concentrations of biomass burning species were found in the fine particles during the campaigns. Good correlations were found between the monosaccharides and OC and EC, highlighting their contributions to carbonaceous species. Non-sea-salt potassium was also well correlated with the biomass burning species, corroborating the input from this source.

PMF analysis was performed and source profiles were obtained for Int<sub>2.5</sub>, Ext<sub>2.5</sub> and Ext<sub>10</sub>. Five factors were identified: road dust, industrial, vehicular, biomass burning and secondary processes. Almost 20 % of biomass burning contribution was observed for the PM<sub>2.5</sub> intensive sampling campaign. The source apportionment led to the identification of traffic-related sources, as expected for the site, since the samples were collected during weekdays. The considerable biomass burning contribution suggests not only the importance of long-range transport of emissions from sugarcane burning, but also the input from local biomass burning sources, such as waste burning and wood stoves in restaurants. More studies are needed on the impact of local sources of biomass burning, in order to identify the different inputs.

**Data availability:** All data are available by the authors Guilherme Martins Pereira (martinspereira2@hotmail.com) and Pérola de Castro Vasconcellos (perola@iq.usp.br). Meteorological data can be found at CETESB, 2014, bulletin (<http://www.estacao.iag.usp.br/Boletins/2014.pdf>). Gaseous species concentration can be found at CETESB ([http://sistemasinter.cetesb.sp.gov.br/Ar/php/ar\\_dados\\_horarios.php](http://sistemasinter.cetesb.sp.gov.br/Ar/php/ar_dados_horarios.php)).

**Author contributions:** Mr. Guilherme M. Pereira performed the PAHs, WSI and monosaccharides determination and the PMF analysis and Prof. Dr. Pérola Vasconcellos is his advisor and laboratory supervisor in Brazil. Dr. Kimmo Teinillä was responsible for WSI and monosaccharides determination with Mr. Guilherme Pereira, and Dr. Risto Hillamo is the laboratory supervisor. Dr. Danilo Custódio was responsible for carbonaceous species determination and polar plots, Dr. Célia Alves is the department supervisor in Portugal. Dr. Aldenor Gomes Santos developed the method for PAHs determination; Prof. Gisele Olímpio da Rocha and Prof. Jailson Bittencourt de Andrade are the supervisors at UFBA. Prof. Prashant Kumar and Prof. Maria de Fátima Andrade contributed with source apportionment. Ms. Huang Xian and Prof. Rajasekhar Balasubramanian were responsible for elements determination.

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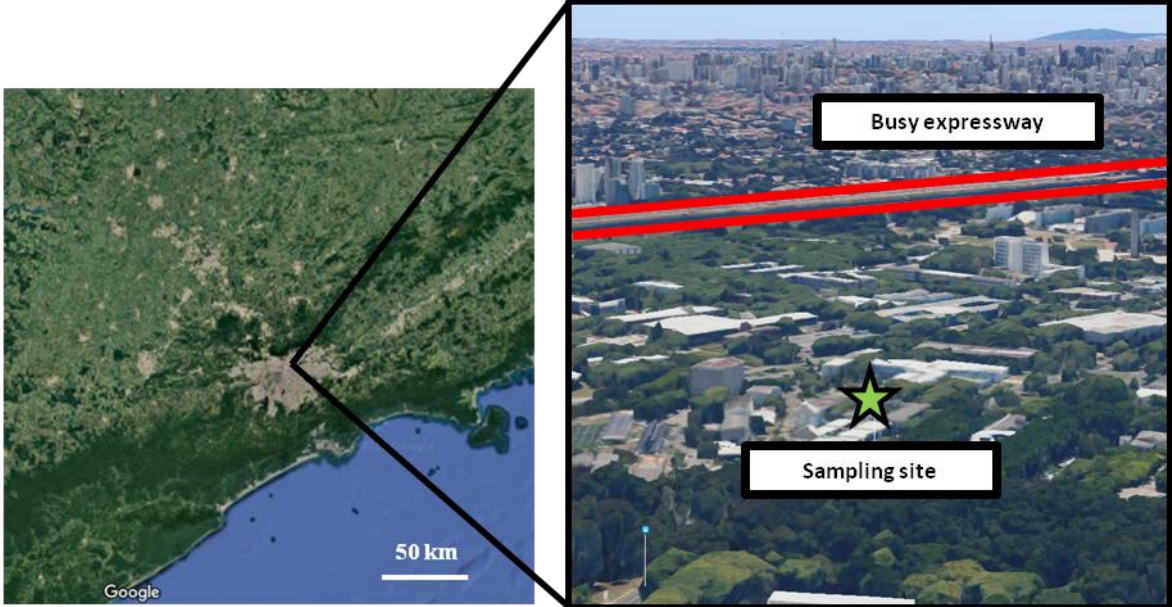


Figure 1. Location of the sampling site. Maps are a courtesy of Google maps.

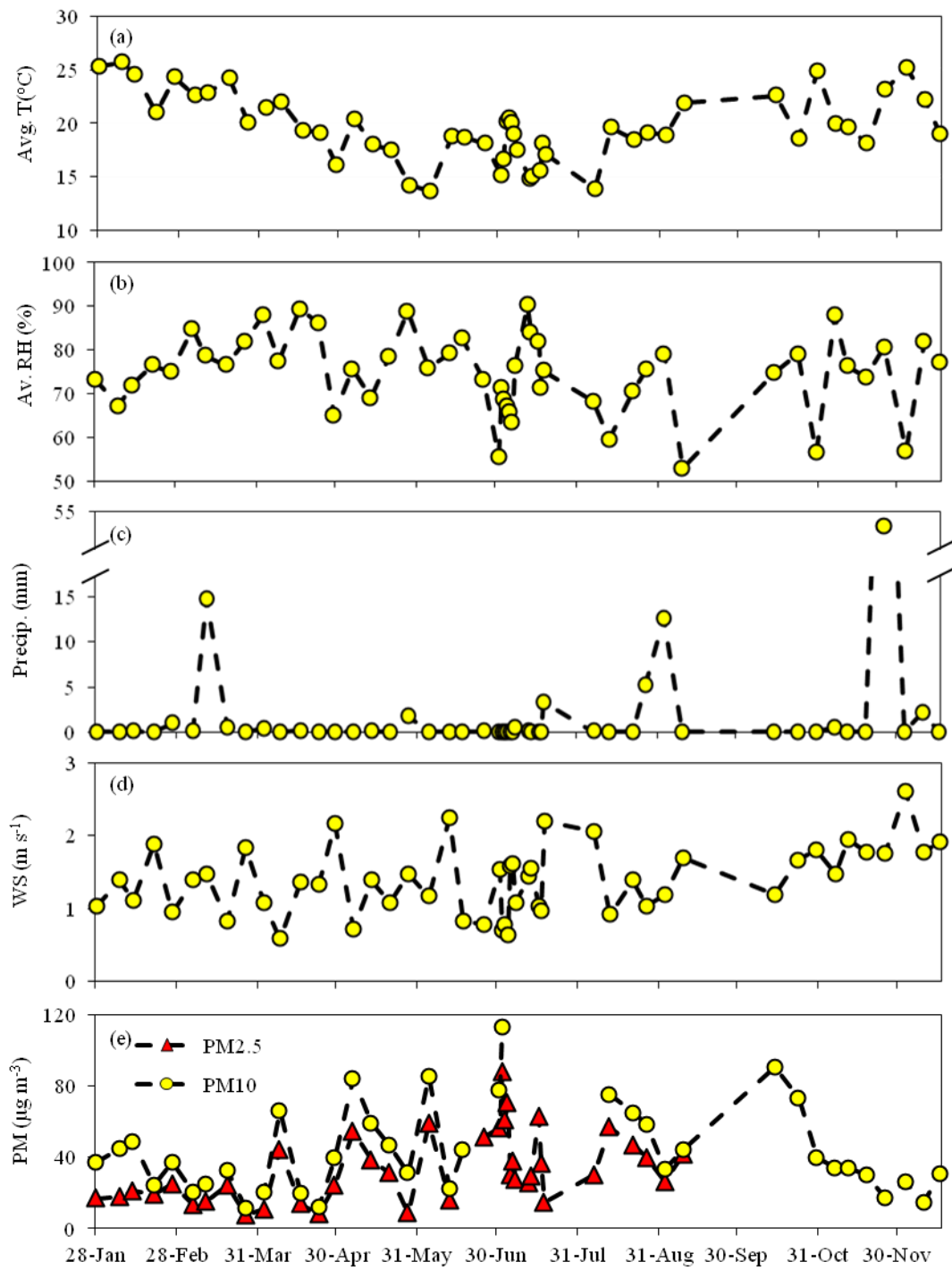


Figure 2. (a) Daily average temperatures, (b) relative humidity (RH), (c) precipitation, (d) wind speed and (e) PM<sub>10</sub> and PM<sub>2.5</sub> concentrations for the three campaigns.

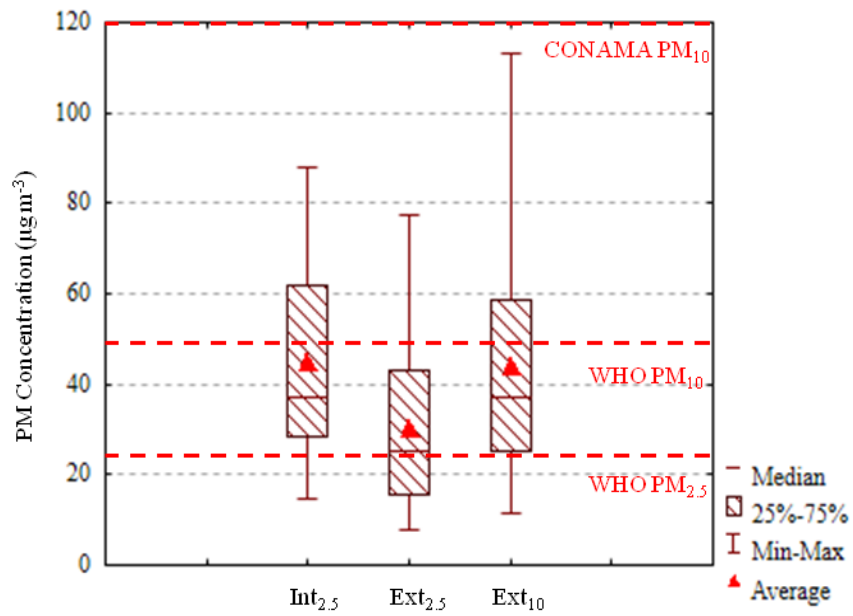


Figure 3. Box plot for particulate matter concentrations in the intensive and extensive campaigns.

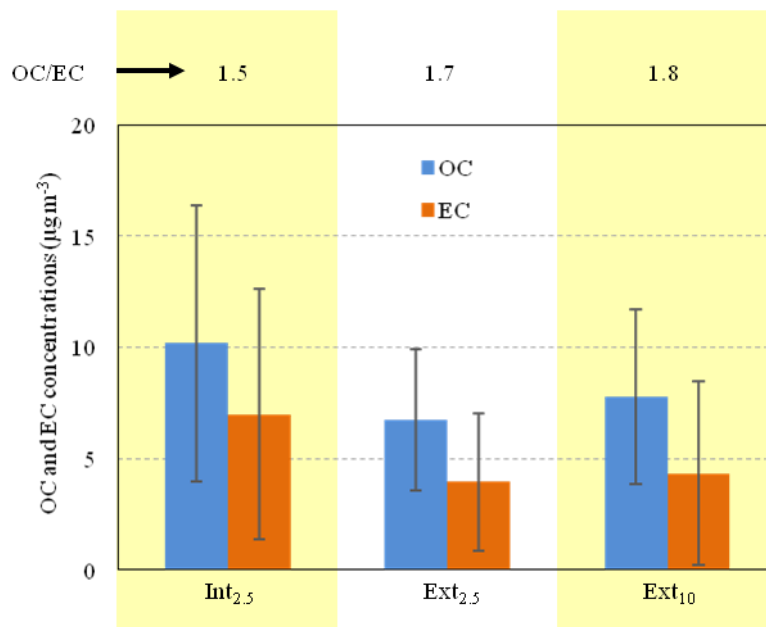


Figure 4. Carbonaceous species concentrations for all campaigns.

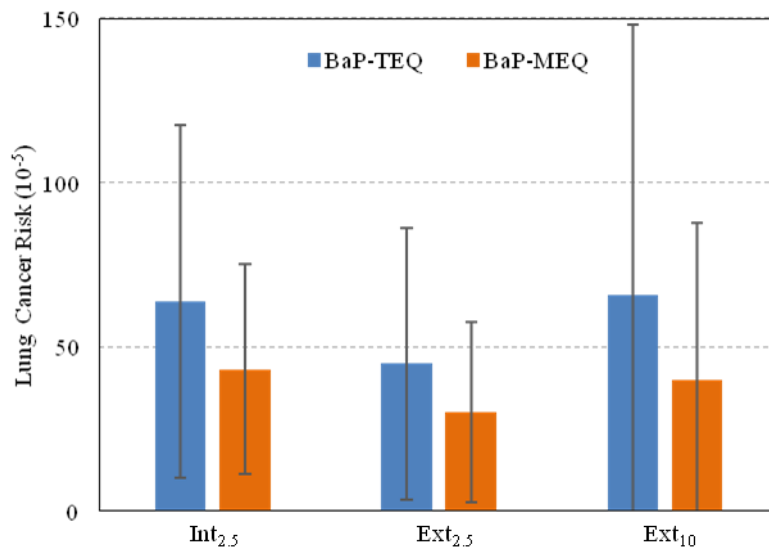


Figure 5. Lung cancer risk from the BaP-TEQ and BaP-MEQ for all the three campaigns.

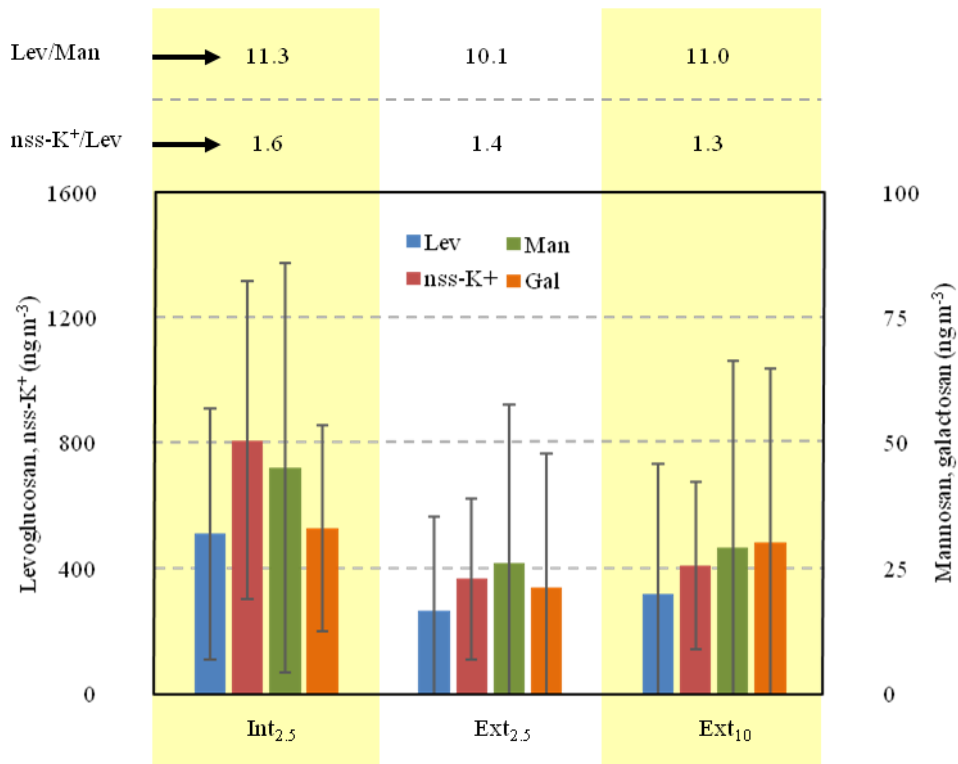
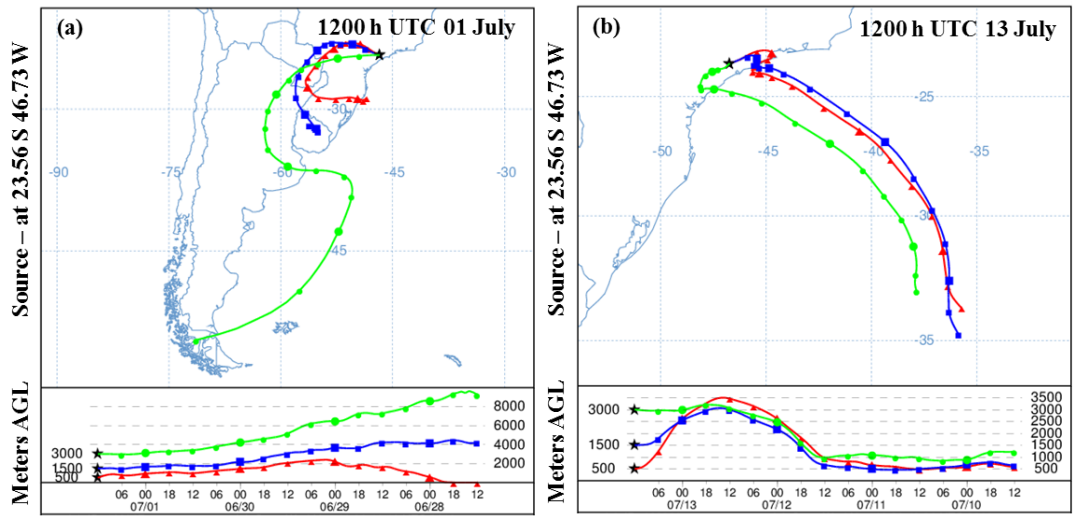
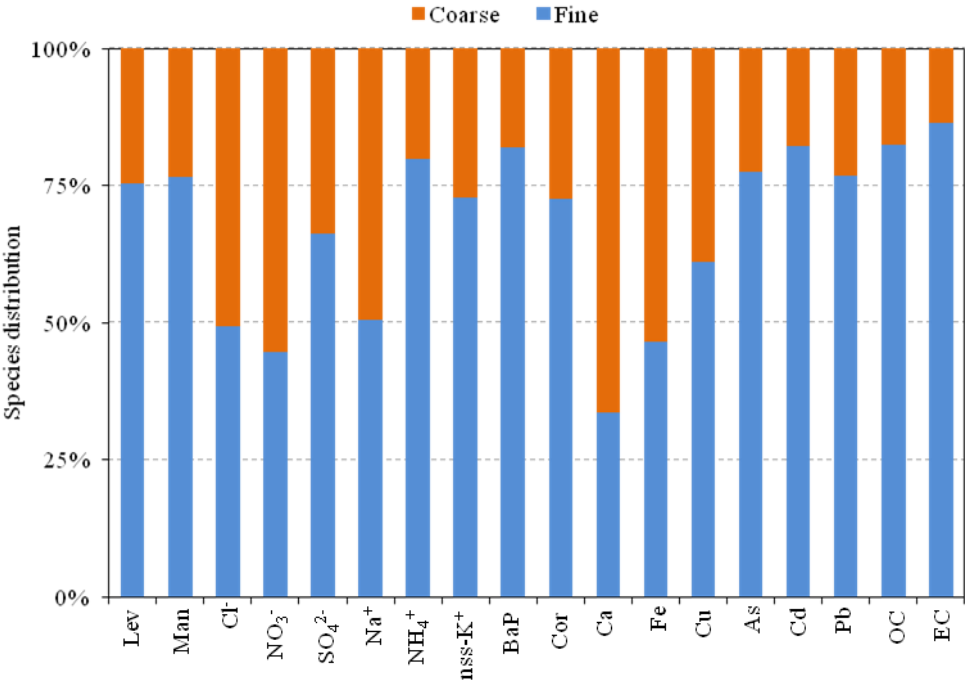


Figure 6. Concentrations of biomass burning tracers for all campaigns.



1045 Figure 7. Backward air mass trajectories for the days (a) 01 July and (b) 13 July.



1050 Figure 8. Mass percentage distribution of species in the fine and coarse particles.

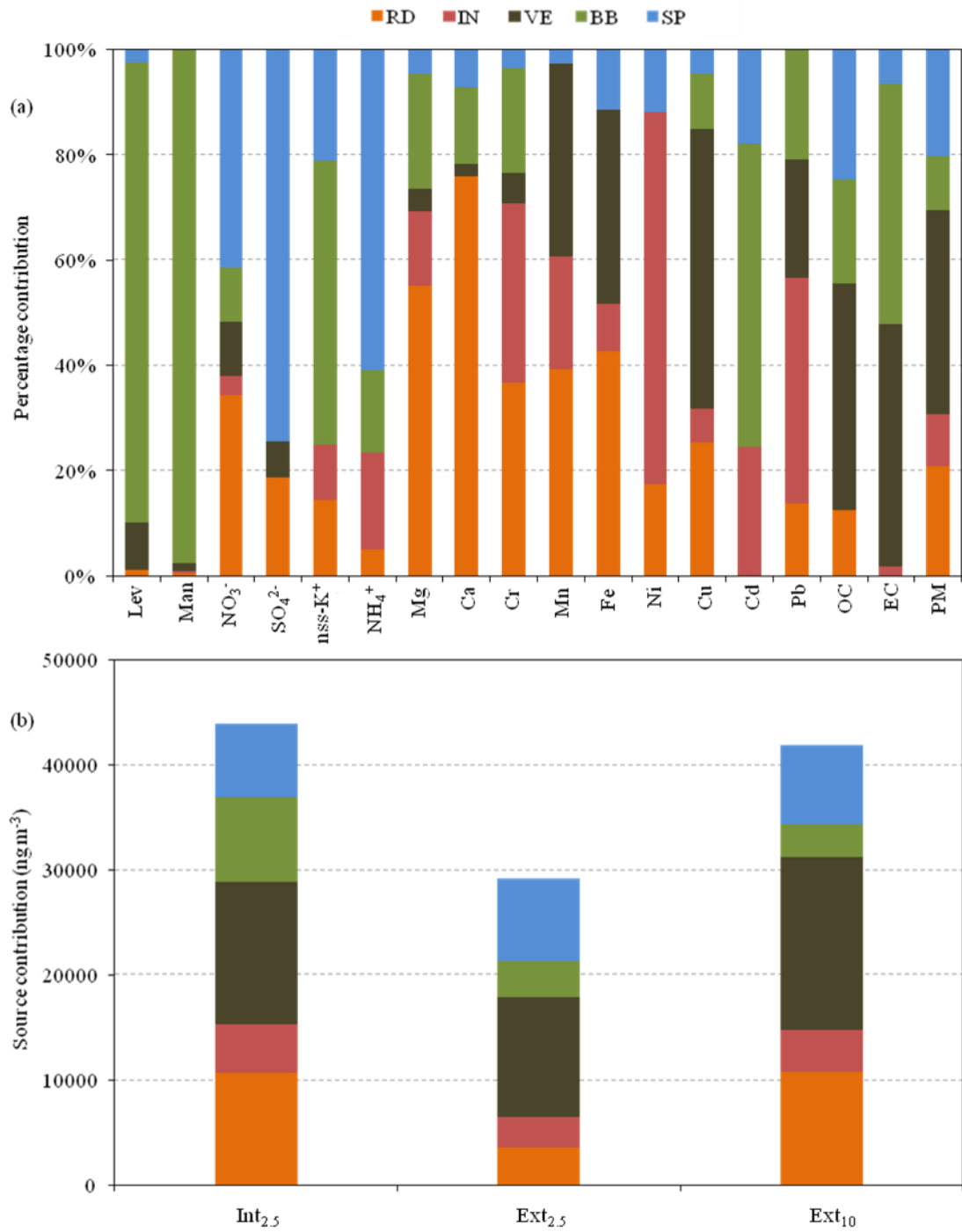


Figure 9. (a) Profile of species for each source (SP - Secondary processes, BB - Biomass burning, VE - Vehicular, IN - Industrial and RD - Road dust) (b) Contribution of sources for each campaign.

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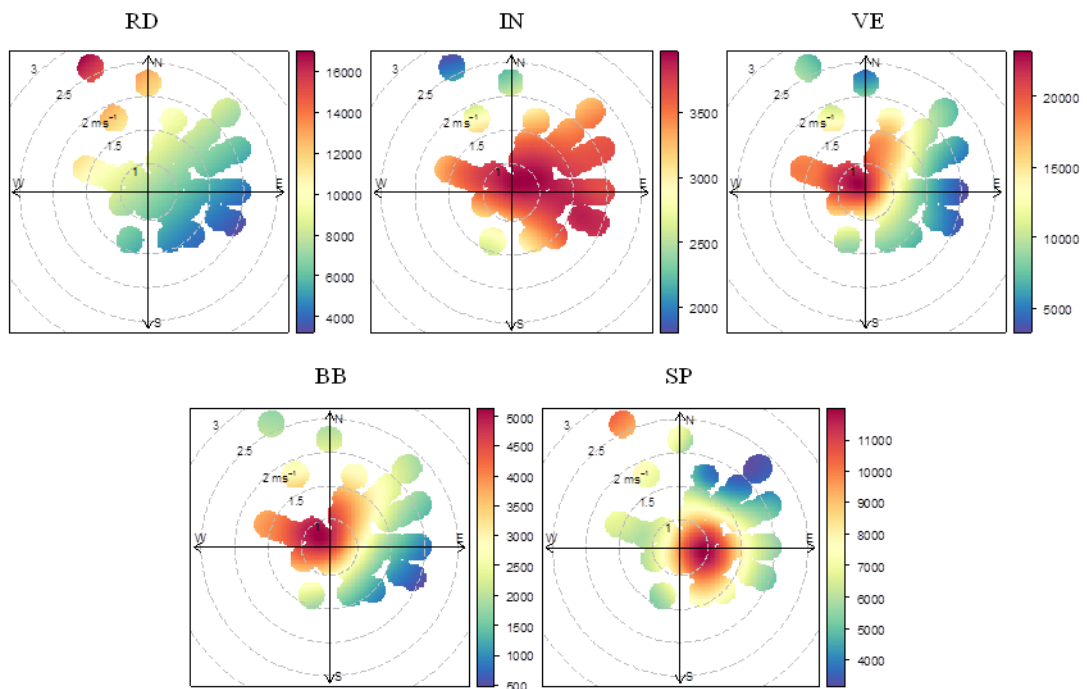
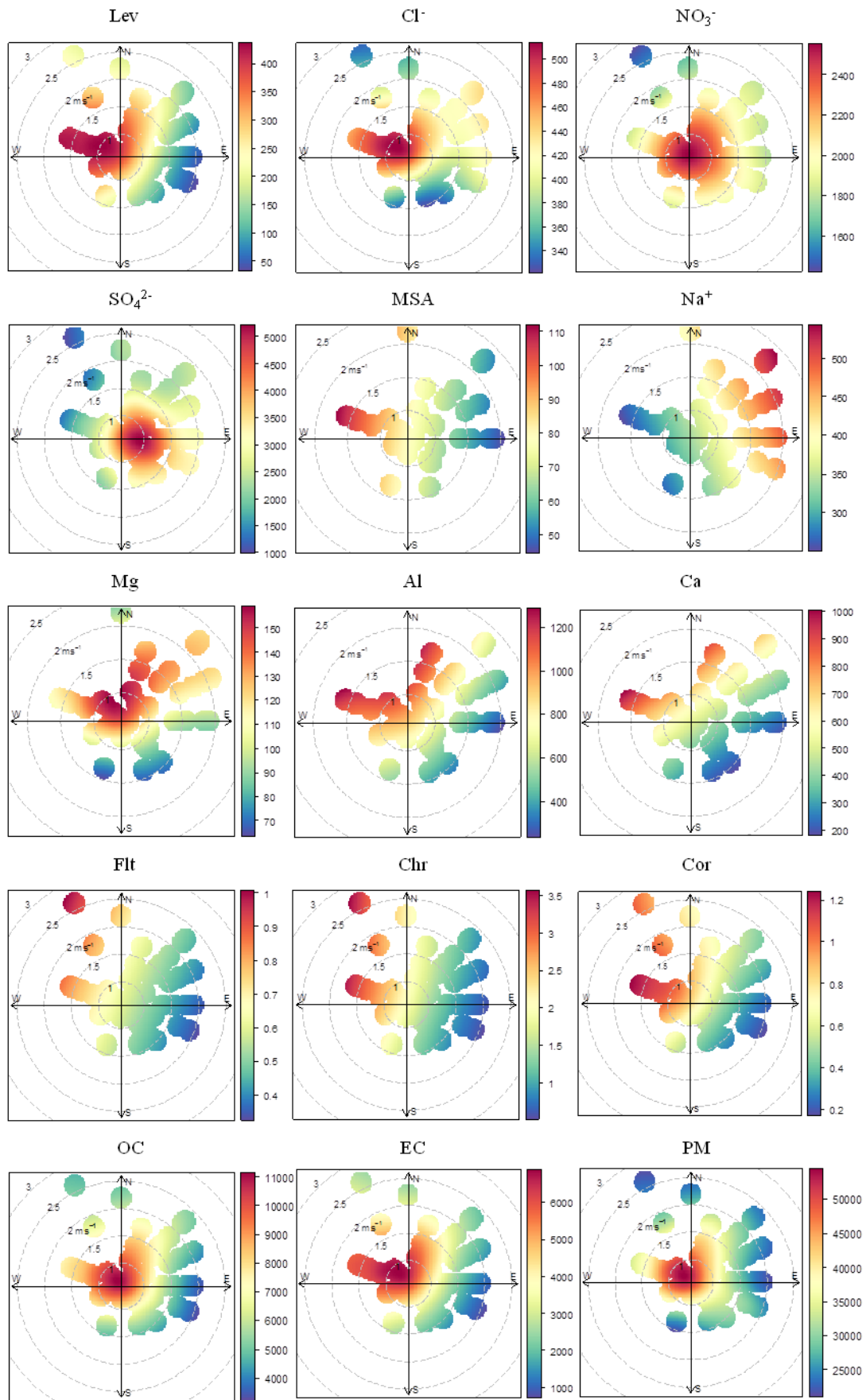


Figure 10. Polar plots of sources contributions in São Paulo ( $\text{ng m}^{-3}$  and  $\text{m s}^{-1}$ ).





1060 Figure 11. Polar plots for different species ( $\text{ng m}^{-3}$  and  $\text{m s}^{-1}$ ).

**Table 1. Results of previous source apportionment studies in São Paulo city.**

Site	Year	Instruments	Species	Range	Identified Sources	Source
University of São Paulo - Atmospheric Sciences Department (campus)	1989 (winter)	PIXE	Elements	PM <sub>2.5</sub>	Factor analysis - five sources: industrial emissions (13 %), emissions from residual oil and diesel (41 %), resuspended soil dust (28 %) and emissions of Cu and Mg (18 %).	Andrade et al. (1994)
				PM <sub>2.5-15</sub>	Four sources: soil dust (59%), industrial emissions (19 %), oil burning (8 %) and sea salt aerosol (14 %).	
University - Medicine School (downtown)	1994 (winter)	PIXE, reflectance	Elements, black carbon	PM <sub>2</sub>	Factor analysis - five sources: vehicles, garbage incineration, vegetation, suspended soil dust and burning of fuel oil.	Sánchez-Ccoyllo and Andrade (2002)
University - Medicine School building and Atmospheric Sciences Department	1997 (winter) and 1998 (summer)	TEOM; PIXE; ACPM; Aethalometer	OC, EC, elements and gaseous species	PM <sub>2.5</sub>	Factor analysis - five sources: motor vehicle (28 and 24 %, for winter and summer), resuspended soil dust (25 and 30 %), oil combustion source (18 and 21 %), sulfates (23% and 17%) and industrial emissions (5% and 6%).	Castanho and Artaxo (2001)
				PM <sub>2.5-10</sub>	Resuspended soil dust represented a large fraction (75–78 %).	
University of São Paulo - Atmospheric Sciences Department	2002 (winter)	GC-MS	PAHs	PM <sub>2.5</sub>	Factor analysis - four factors: diesel emissions, stationary combustion source, vehicular emissions, natural gas combustion and biomass burning.	Bourotte et al. (2005)
				PM <sub>2.5-10</sub>	Two factors: vehicular emissions and mixture of combustion sources (natural gas combustion, incineration emissions and oil combustion).	
University of São Paulo - Atmospheric Sciences Department	2003 (winter)	IC; ICPMS	WSI and Elements	PM <sub>10</sub>	Principal component analysis – Two factors (48.5 % of variance): local and remote sources.	Vasconcelos et al. (2007)

University of São Paulo - Atmospheric Sciences Department	2003-2004 (year round)	IC; CCD ICP	WSI and Elements	PM <sub>10</sub>	Principal component analysis - three principal components: biomass burning and/or automobile fuel burning (40.3 %), gas-to-particle conversion (12.7 %) and sea spray contribution (11.7 %).	da Rocha et al., (2012)
University - Medicine School building	2007-2008 (year round)	X-ray spectrometry, reflectance	Elements and black carbon	PM <sub>2.5</sub>	APCA - four factors: crustal emission (soil and construction) (13 %); oil-burning boilers, industrial emissions and secondary aerosol formation (13 %); light-duty vehicle emissions (12 %) and heavy-duty diesel fleet (28 %).	Andrade et al. (2012)

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**Table 2. Details of the analyzed species, analytical methods and detection limits.**

Analytical method	Detection limits (ng m <sup>-3</sup> )	Determined species
Thermal-optical analysis	14 (EC) /	<b>Carbonaceous species:</b> OC and EC.
	262 (OC)	
GC-MS	0.01–0.06	<b>PAHs:</b> naphthalene (Nap), acenaphthene (Ace), acenaphthylene (Acy), Fluorene (Flu), phenanthrene (Phe), anthracene (Ant), fluoranthene (Flt), pyrene (Pyr), benzo[a]anthracene (BaA), chrysene (Chr), benzo[b]fluoranthene (BbF), benzo[k]fluoranthene (BkF), benzo[e]pyrene (BeP), benzo[a]pyrene (BaP), perylene (Per), indeno[1,2,3-cd]pyrene (InP), dibenzo[ah]anthracene (DBA), benzo[ghi]perylene (BPe) and coronene (Cor). <b>Nitro-PAHs:</b> (1-nitronaphthalene (1-NNap), 2-nitronaphthalene (2-NNap), 1-methyl-4-nitronaphthalene (1-Methyl-4-NNap), 1-methyl-5-nitronaphthalene (1-Methyl-5-NNap), 1-methyl-6-nitronaphthalene (1-Methyl-6-NNap), 2-methyl-4-nitronaphthalene (2-methyl-4-NNap), 2-nitrobiphenyl (2-NBP), 3-nitrobiphenyl (3-NBP), 4-nitrobiphenyl (4-NBP), 5-nitroacenaphthene (5-NAce), 2-nitrofluorene (2-Nflu), 2-nitrophenanthrene (2-NPhe), 3-nitrophenanthrene (3-NPhe), 9-nitrophenanthrene (9-NPhe), 2-nitroanthracene (2-NAnt), 9-nitroanthracene (9-NAnt), 2-nitrofluoranthene (2-NFlt), 3-nitrofluoranthene (3-NFlt), 1-nitropyrene (1-NPyr), 2-nitropyrene (2-NPyr), 4-nitropyrene (4-NPyr), 6-nitrochrysene (6-NChr), 7-nitrobenz[a]anthracene (7-NBaA), 3-nitrobenzanthrone (3-NBA), 6-nitrobenzo[a]pyrene (6-NBaPyr), 1-nitrobenzo[e]pyrene (1-NBePyr), and 3-nitrobenzo[e]pyrene (3-NBePyr).
	0.01–0.50	
	0.3–10.3	<b>Oxy-PAHs:</b> 1,4-benzoquinone (1,4-BQ), 9,10-phenanthraquinone (9,10-PQ), 9,10-anthraquinone (9,10-AQ), 1,2-naphthoquinone (1,2-NQ) and 1,4-naphthoquinone (1,4-NQ).
IC	<del>1.3–1.36.4–6.4</del>	<b>WSI:</b> Cl <sup>-</sup> , NO <sub>3</sub> <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup> , C <sub>2</sub> O <sub>4</sub> <sup>2-</sup> , methylsulfonate (MSA <sup>-</sup> ), Na <sup>+</sup> , K <sup>+</sup> , NH <sub>4</sub> <sup>+</sup> .
HPAEC-MS	1.3–2.5	<b>Monosaccharides:</b> levoglucosan, mannosan and galactosan.
ICP-MS	0.0002–2.3	<b>Elements:</b> Li, Mg, Al, K, Ca, Cr, Mn, Fe, Co, Ni, Cu, Zn, As, Se, Rb, Sr, Cd, Sn, Cs, Tl, Pb, Bi.

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**Table 3. Concentrations of WSI in all campaigns.**

(ng m <sup>-3</sup> )	Int <sub>2.5</sub>	Ext <sub>2.5</sub>	Ext <sub>10</sub>
	Average (Min–Max)	Average (Min–Max)	Average (Min–Max)
Cl <sup>-</sup>	964 (107–4549)	330 (16–1427)	641 (76–5904)
NO <sub>3</sub> <sup>-</sup>	2678 (667–6873)	1430 (183–3419)	2872(437–8880)
SO <sub>4</sub> <sup>2-</sup>	3266 (1252–5959)	3197 (922–6300)	3680 (569–9361)
MSA <sup>-</sup>	84 (15–214)	63 (13–226)	107 (28–444)*
C <sub>2</sub> O <sub>4</sub> <sup>2-</sup>	478 (176–753)	282 (57–726)	367 (50–1180)
Na <sup>+</sup>	350 (46–869)	238 (64–512)	571 (76–1908)
NH <sub>4</sub> <sup>+</sup>	1712 (613–4075)	1370 (281–2845)	1336 (57–4436)
nss-K <sup>+</sup>	809 (237–2007)	366 (49–1137)	413 (63–1181)
SNA	7655	5997	7888
Total	10334	7276	9986
SO <sub>4</sub> <sup>2-</sup> /NO <sub>3</sub> <sup>-</sup>	1.2	2.2	1.3
Cl/Na <sup>+</sup>	2.7	1.4	1.1
$\Sigma$ cations/ $\Sigma$ anions	0.88	0.58	0.64
SNA/Total (%)	74	82	79

\*Data was not determined after 21 August.

**Table 4. Average, minimum and maximum concentrations of tracer elements for all campaigns.**

	<b>Int<sub>2,5</sub></b>	<b>Ext<sub>2,5</sub></b>	<b>Ext<sub>10</sub></b>
(ng m <sup>-3</sup> )	<b>Average</b> (Min–Max)	<b>Average</b> (Min–Max)	<b>Average</b> (Min–Max)
<b>Li</b>	0.48 (<DLND –1.12)	0.27 (<DLND –0.70)	0.40 (<NDL –1.25)
<b>Mg</b>	210 (5–469)	93 (5–356)	154 (ND<DL–377)
<b>Al</b>	1851 (<DLND –2782)	691 (<DLND –2712)	981 (ND<DL–3014)
<b>K</b>	1431 (191–3833)	500 (<DLND –1967)	600 (ND<DL–1682)
<b>Ca</b>	1164 (<DLND –3204)	397 (<DLND –1671)	666 (ND<DL–2160)
<b>Cr</b>	23 (1–60)	13 (1–60)	20 (ND<DL–54)
<b>Mn</b>	30 (<DLND –64)	17 (<DLND –49)	33 (4–175)
<b>Fe</b>	962 (173–2056)	581 (140–1408)	1269 (240–3578)
<b>Co</b>	0.45 (0.03–1.06)	0.23 (0.01–0.78)	0.59 (0.07–1.74)
<b>Ni</b>	7.3 (2.3–14.8)	4.6 (<DLND –16.1)	6.6 (ND<DL–25.9)
<b>Cu</b>	181 (7–390)	109 (7–308)	188 (32 –976)
<b>Zn</b>	284 (<DLND –673)	110 (<DLND –279)	193 (ND<DL–716)
<b>As</b>	2.8 (0.06–5.7)	1.9 (<DLND –7.1)	2.2 (ND<DL–7.9)
<b>Se</b>	5.6 (<DLND –13.2)	2.6 (<DLND –7.5)	2.6 (ND<DL–7.9)
<b>Rb</b>	5.7 (0.4–12.3)	2.2 (0.1–8.9)	2.6 (0.2–8.9)
<b>Sr</b>	6.6 (0.4–13.4)	3.0 (0.2–12.2)	4.8 (0.4–14.3)
<b>Cd</b>	2.5 (0.2–15.1)	0.8 (0.1–3.0)	1.2 (0.2–10.6)
<b>Sn</b>	19.5 (3.2–40.2)	8.8 (0.3–35.9)	12.3 (1.6–41.8)
<b>Cs</b>	0.28 (0.07–1.01)	0.14 (<DLND –0.51)	0.19 (0.02–0.77)
<b>Tl</b>	0.21 (<DLND –0.75)	0.13 (<DLND –0.38)	0.15 (0.03 –0.65)
<b>Pb</b>	54 (3–172)	31 (3–71)	42 (4–176)
<b>Bi</b>	0.76 (0.06–3.03)	0.47 (<DLND –3.03)	0.83 (0.12–3.24)

<DL - below detection limit, ND – not detected.

1105

1110

1115

1120

**Table 5. Concentrations of PAHs and derivatives for all campaigns.**

(ng m <sup>-3</sup> )	<b>Int<sub>2,5</sub></b>	<b>Ext<sub>2,5</sub></b>	<b>Ext<sub>10</sub></b>
	<b>Average</b>	<b>Average</b>	<b>Average</b>
	<b>(Min - Max)</b>	<b>(Min - Max)</b>	<b>(Min - Max)</b>
<b>Nap</b>	0.30 (0.17–0.77)	0.36 (0.02–0.77)	0.41 (0.09–0.81)
<b>Acy</b>	0.09 (0.06–0.12)	0.10 (0.03–0.19)	0.12 (0.05–0.34)
<b>Ace</b>	0.03 (0.02–0.08)	0.05 (0.02–0.16)	0.07 (0.02–0.23)
<b>Flu</b>	0.27 (0.15–1.03)	0.31 (0.06–1.44)	0.51 (0.10–1.75)
<b>Phe</b>	0.65 (0.30–2.48)	0.74 (0.12–3.55)	1.28 (0.28–4.08)
<b>Ant</b>	0.17 (0.10–0.44)	0.16 (0.06–0.60)	0.25 (0.08–0.67)
<b>Flt</b>	0.48 (0.21–0.86)	0.53 (0.06–1.40)	0.73 (0.19–2.21)
<b>Pyr</b>	0.52 (0.20–0.99)	0.54 (0.07–1.54)	0.71 (0.19–2.45)
<b>BaA</b>	1.0 (0.3–2.4)	0.9 (0.1–4.8)	1.2 (0.3–5.9)
<b>Chr</b>	1.8 (0.5–4.4)	1.6 (0.3–5.7)	2.1 (0.5–10.5)
<b>BbF</b>	3.0 (0.9–6.1)	2.3 (0.5–6.4)	3.0 (0.7–13.3)
<b>BkF</b>	2.5 (0.6–5.2)	1.9 (0.2–7.4)	2.5 (0.4–11.8)
<b>BeP</b>	2.8 (0.6–6.1)	2.2 (0.3–7.3)	2.8 (0.5–14.4)
<b>BaP</b>	2.3 (0.4–5.5)	1.6 (0.2–7.6)	2.0 (0.3–12.5)
<b>Per</b>	0.35 (0.04–0.79)	0.27 (<DL–1.27)	0.38 (0.05–1.90)
<b>InP</b>	2.9 (0.6–6.0)	1.8 (0.3–6.3)	2.4 (0.4–13.2)
<b>DBA</b>	0.8 (0.1–2.3)	0.6 (0.0–2.0)	0.9 (0.0–5.1)
<b>BPe</b>	2.4 (0.5–4.8)	1.6 (0.2–5.5)	2.1 (0.4–10.5)
<b>Cor</b>	1.0 (0.1–2.4)	0.7 (0.0–2.4)	0.9 (0.1–5.2)
<b>Total</b>	23.3 (6.0–48.8)	18.4 (2.6–61.6)	24.3 (5.4–115.3)
<b>BaPE</b>	3.4 (0.6–8.0)	2.4 (0.3–10.5)	3.2 (0.5–18.3)
<b>1-NNap</b>	<DL	<DL	<DL
<b>1-Methyl-4-NNap</b>	<DL	<DL	<DL
<b>2-NNap</b>	<DL	<DL	<DL
<b>2-NBP</b>	0.56 (<DL–1.36)	0.56 (<DL–1.36)	1.23 (0.47–2.47)
<b>1-Methyl-5-NNap</b>	0.18 (<DL–0.28)	<DL	<DL
<b>1-Methyl-6-NNap</b>	0.36 (<DL–0.40)	0.27 (<DL–0.41)	0.29 (<DL–0.86)
<b>2-Methyl-4-NNap</b>	0.45 (<DL–0.45)	0.36 (<DL–0.44)	0.42 (<DL–1.26)
<b>3-NBP</b>	0.60 (0.48–0.88)	0.52 (<DL–0.87)	0.55 (<DL–1.58)
<b>4-NBP</b>	<DL	<DL	0.18 (<DL–0.41)
<b>5-NAce</b>	<DL	<DL	0.20 (<DL–0.52)
<b>2-NFlu</b>	0.98 (0.78–1.39)	0.99 (0.38–1.56)	1.09 (0.54–1.79)
<b>2-NPhe</b>	0.43 (0.30–0.67)	0.51 (0.19–1.40)	0.61 (<DL–1.80)
<b>3-NPhe</b>	0.43 (<DL–0.46)	0.44 (<DL–0.68)	0.47 (<DL–1.11)
<b>9-NPhe</b>	0.62 (<DL–0.64)	0.52 (<DL–0.64)	0.55 (<DL–0.82)
<b>2-Nant</b>	0.66 (<DL–0.80)	0.56 (<DL–0.80)	0.61 (<DL–0.88)
<b>9-Nant</b>	0.44 (<DL–0.57)	0.42 (<DL–0.69)	0.46 (<DL–1.15)
<b>2-NFlt</b>	1.19 (<DL–1.35)	0.98 (<DL–1.25)	1.02 (<DL–1.43)

1125	<b>3-NFlt</b>	1.45 (<DL-1.48)	1.05 (<DL-1.48)	1.02 (<DL-1.11)
	<b>1-NPyr</b>	0.98 (<DL-1.12)	0.73 (<DL-0.88)	0.79 (<DL-1.28)
	<b>2-NPyr</b>	0.94 (<DL-0.99)	0.76 (<DL-0.99)	0.78 (<DL-1.27)
	<b>4-NPyr</b>	1.61 (<DL-1.67)	1.27 (<DL-1.34)	1.28 (<DL-1.72)
	<b>7-NBaA</b>	1.19 (<DL-1.34)	0.91 (<DL-1.06)	1.01 (<DL-1.67)
	<b>6-NChr</b>	<DL	0.60 (<DL-0.67)	0.69 (0.58-1.10)
	<b>3-NBA</b>	<DL	<DL	<DL
	<b>6-NBaPyr</b>	<DL	<DL	1.01 (<DL-1.19)
	<b>1-NBaPyr</b>	<DL	<DL	<DL
	<b>3NBePyr</b>	<DL	<DL	<DL
1130	<b>1,4-BQ</b>	<DL	<DL	<DL
	<b>1,4-NQ</b>	0.54 (0.43-0.72)	0.44 (0.28-0.67)	0.46 (0.31-1.08)
	<b>1,2-NQ</b>	<DL	<DL	<DL
	<b>9,10-AQ</b>	1.6 (0.8-3.7)	2.5 (0.3-8.0)	2.6 (0.4-10.9)
	<b>9,10-PQ</b>	<DL	<DL	<DL
	<b>Total PAHs/OC (%)</b>	0.23	0.27	0.31
1135	<b>ΣLMW/ΣHMW</b>	0.32	0.41	0.43
	<b>Flt/(Flt+Pyr)</b>	0.5	0.5	0.5
	<b>BaA/Chr</b>	0.5	0.6	0.5
	<b>InP/(InP+BPe)</b>	0.5	0.5	0.5
	<b>BaP/(BaP+BeP)</b>	0.4	0.4	0.4
	<b>BPe/BaP</b>	1	1	1
	<b>2-NFlt /1-NPyr</b>	1.3	1.3	1.3

1140