

## RESPONSE TO REVIEWERS

### REVIEWER 1:

#### L. Horváth (Referee)

5 The authors thank Prof. Horváth for his constructive comments and for taking the time to look at all the details described in the manuscript. We have carefully considered all comments. Please refer to the specific responses.

#### 1) General observations

*“Emission sources According to the MS (Fig. 10b) the share of fertilizers is 16.2% together with ‘other animals’.*

10 *This is a simplification, these two sources have to split, since it takes 4-5 times higher contribution than that of sheep. So I miss displaying fertilizers from some figures (7, 10, 16).”*

**Author Response:** There is one NAMN site only classed as dominated by emissions from the “fertiliser” emission source sector (see section 3.1 describing the classification of each NAMN sites according to one of seven specific dominant emission source sectors). At this site, NH<sub>3</sub> is measured but not NH<sub>4</sub><sup>+</sup> (see Supp. Fig. 2a). Temporal and trend analysis have therefore not been carried out for the “fertilisers” source sector. The focus of Fig. 10b is on three specific dominant emission source sectors (Cattle, Pigs & Poultry, Sheep), compared with other emissions.

15 Since the “fertiliser” category is not considered in the paper (due to too few sites), “other animals + fertilisers” in the pi-chart have been grouped together for simplicity. “Fertiliser” category has not been considered separately in Figs 7 and 16 for the same reason.

20 UK annual ammonia emissions data are downloaded from <http://naei.defra.gov.uk/data/>:

Other animals (horses) = 1.39 % and fertilisers = 14.86 %.

25 Sum of “other animals + fertiliser” = 16.2 % (Fig. 10b).

30 *“Long-term trend analysis Ammonia emission in UK decreased substantially during the examined period while concentration remained at same level as it have been observed in other countries in Europe. Authors mentioned, it is the effect of sulphur dioxide emission and concentration decrease. It is true, but I miss a more detailed explanation of this mechanism. Fowler et al., 2001 (Water Air Soil Poll, Focus 1, 39-48) pointed out firstly the importance of co-deposition of ammonia and sulphur dioxide. I.e. there is a direct proportion between the SO<sub>2</sub> concentration and the dry deposition velocity of NH<sub>3</sub> onto natural surfaces that strongly influences the ammonia level in the atmosphere.*

35 **Author Response:** See response to Reviewer Comment 2 (p9).

40

5 “Seasonal trend analysis Source strength of ammonia – of course – strongly depends on temperature, so seasonal  
trend of  $\text{NH}_x$  is mainly determined by this factor. But, as to the ammonia/ammonium transformation it is partly an  
equilibrium process due to the  $\text{NH}_3 + \text{HNO}_3 : \text{NH}_4\text{NO}_3$  reaction as it mentioned in the first paragraph of 3.5.6.  
The dissociation constant of ammonium nitrate depends on temperature, relative humidity and particle size  
10 (Mozurkewich, Atmos Envir 27A:261-270, 1993). At low relative air humidity (r.h. <60-70) ammonium nitrate  
does not exist in air at all. This phenomenon may strongly effect on seasonal variation of  $\text{NH}_3$  and  $\text{NH}_4^+$   
concentrations as a consequence of difference of summer/winter humidity even if part of ammonium is associated  
with hydrogen sulphate or sulphate ions. Authors should also describe this mechanism in the interpretation of  $\text{NH}_x$   
seasonal trend. A sulphate/nitrate ratio in aerosol phase in different seasons would give a good qualitative picture.  
15 In Fig. 18a we can observe nitrate dominance against the sulphate (2:1 in case of ammonium nitrate-ammonium  
hydrogen sulphate regime) that underlines the importance of ammonium nitrate in controlling the  
ammonium/ammonia ratio. Spring maxima for particulate ammonium has observed and explained by the effect of non-  
domestic (continental) sources (after Vieno et al., 2014). But, the reason of that did not mentioned. How is the  
possible mechanism responsible for high continental ammonium (or ammonia) concentrations and transport from  
the continent in spring?”

**Author Response:** Section 3.3, paragraph 5 has been expanded to include explanation of the equilibrium between  
gas and aqueous aerosol phase as drivers in the seasonal variations of particulate  $\text{NH}_4^+$ .

20 “For particulate  $\text{NH}_4^+$ , as expected for a secondary pollutant, concentrations are more decoupled from  
the dominant  $\text{NH}_3$  source sectors in the vicinity of a site. Although the formation of particulate  $\text{NH}_4^+$   
primarily depends on the occurrence of  $\text{NH}_3$  in the atmosphere, synoptic meteorology and long range  
transboundary transport from continental Europe are important drivers influencing the seasonal  
25 variations of  $\text{NH}_4^+$  across the UK, due to its’ longer lifetime. The seasonal trends in particulate  $\text{NH}_4^+$  are  
seen to be broadly similar for the four different emission source sectors (Figure 8b), with the magnitude  
of the  $\text{NH}_4^+$  concentrations reflecting  $\text{NH}_3$  concentrations at a regional level. In the atmosphere,  
particulate  $\text{NH}_4^+$  are primarily in the form of  $(\text{NH}_4)_2\text{SO}_4$  and  $\text{NH}_4\text{NO}_3$ , formed when the acid gases  $\text{HNO}_3$   
and  $\text{H}_2\text{SO}_4$  in the atmosphere are neutralised by  $\text{NH}_3$  (Putaud et al., 2010).  $\text{NH}_3$  preferentially neutralizes  
30  $\text{H}_2\text{SO}_4$  due to its low saturation vapour pressure (forming  $\text{NH}_4\text{HSO}_4$  followed by  $(\text{NH}_4)_2\text{SO}_4$ ), while  
 $\text{NH}_4\text{NO}_3$  is formed when abundant  $\text{NH}_3$  is available. In contrast to  $(\text{NH}_4)_2\text{SO}_4$ ,  $\text{NH}_4\text{NO}_3$  is a semi-volatile  
component (Stelson & Seinfeld, 1982). Long-term data from the UK Acid Gas and Aerosol Network  
(AGANet, Conolly et al., 2016) shows a change in the particulate phase of  $\text{NH}_4^+$  from  $(\text{NH}_4)_2\text{SO}_4$  to  
 $\text{NH}_4\text{NO}_3$ , with particulate nitrate concentrations exceeding that of particulate sulphate approximately  
35 three-fold (on a molar basis) (Fig. 18a). This suggests that the thermodynamic equilibrium between the  
gas phase  $\text{NH}_3$  and  $\text{HNO}_3$  and the aerosol phase  $\text{NH}_4\text{NO}_3$  will have a much greater effect on the seasonal  
concentrations of  $\text{NH}_4^+$  than  $(\text{NH}_4)_2\text{SO}_4$ . The formation and dissociation of  $\text{NH}_4\text{NO}_3$  depend strongly on  
ambient temperature and humidity (Stelson and Seinfeld, 1982). Warm, dry weather in summer  
promotes dissociation, decreasing particulate phase  $\text{NH}_4\text{NO}_3$  relative to gas phase  $\text{NH}_3$  and  $\text{HNO}_3$ .  
40 During the winter months, low temperature and high humidity favour the formation of  $\text{NH}_4\text{NO}_3$  from  
the gas phase  $\text{NH}_3$  and  $\text{HNO}_3$ . By contrast, the spring peak in  $\text{NH}_4^+$  concentrations may be attributed to  
photochemical processes (elevated ozone) leading to enhanced formation of  $\text{HNO}_3$  during this period  
(Pope et al., 2016) and also to import of particulate  $\text{NO}_3^-$  through long-range transboundary transport,  
e.g. from continental Europe, as discussed in Vieno et al. (2014). Nevertheless, it is notable that the  
45 winter minima for  $\text{NH}_4^+$  aerosol concentrations at sheep and background sites are more pronounced than  
that for pig, poultry and cattle dominated sites. This may be a result of a combination of smaller  $\text{NH}_3$   
emissions in winter in these areas (as indicated by Figure 8a) and differences in long-range transport to  
the more remote areas in winter conditions.”

References added:

50 Pope, R.J., Butt, E.W., Chipperfield, M.P., Doherty, R.M., Fenech, S., Schmidt, A., Arnold, S.R and  
Savage, N.H. The impact of synoptic weather on UK surface ozone and implications for premature  
mortality. Environmental Research Letters. 11, doi:10.1088/1748-9326/11/12/124004, 2016.

Putaud, J.P., Van Dingenen, R., Alastuey, A., Bauer, H., Birmili, W., Cyrys, J., Flentje, H., Fuzzi, S., Gehrig, R., Hansson, H.C. and Harrison, R.M. A European aerosol phenomenology-3: Physical and chemical characteristics of particulate matter from 60 rural, urban, and kerbside sites across Europe. *Atmospheric Environment*, 44(10), 1308-1320, doi:10.1016/j.atmosenv.2009.12.011, 2010.

Stelson, A. W., and Seinfeld, J. H.: Relative humidity and temperature dependence of the ammonium nitrate dissociation constant, *Atmospheric Environment*, 16, 983-992, doi: 10.1016/0004-6981(82)90184-6, 1982.

“Sampling networking. Because the short lifetime of ammonia it is difficult to find a “representative” measurement site for comparison with modelled concentrations on a 5\_5 km grid. Authors mention a reason of the discrepancy between modelled concentration for the whole UK and concentration for the grids involving one or more measurement sites. This happens in the low and high concentrations regimes (<0.5 and >3.0  $\mu\text{g}/\text{m}^3$ ) in different directions (over- or underestimation), as it also appears clearly in Figure 5. Authors describe some reasons of that (page 9 lines 39-41 though page 10 lines 1-2), mentioning that samplings were influenced by nearby emission sources. In this case some sites are not representative for the given 5\_5 km grid. It is illustrated by the relationship between modelled and measured concentrations in the lower range (selected for the range of between the range of 0 and 4.5  $\mu\text{g}/\text{m}^3$ ) where the relationship is stronger. Further analysis is needed how FRAME model correlated with measured NH<sub>3</sub> concentrations in the work of Dore et al. 2015. Is there any discrepancy between modelled and measured concentrations in low and high ranges? How the model was validated? At sites with low concentrations samplings were performed in a clearing of forests. Question is: do model predict concentrations for layer above the canopy or for the ground level, where effect of deposition of the nearby forest is substantial? It would be the source of another bias between the modelled and measured ammonia concentrations. Other possible source of bias could be derived by the difference between the monthly sampling applied in the NAMN network and the sampling/measurement method for the validation of model. Are there any inter-comparison among the methods described in this manuscript and other methods based on daily or shorter time basis? In any case, taking into account that the modelled and measured concentrations agree well in middle range and in the average for the whole UK, the network seems to be suitable to establish trends for ammonia/ammonium concentrations.”

Re: “Further analysis is needed how FRAME model correlated with measured NH<sub>3</sub> concentrations in the work of Dore et al. 2015. Is there any discrepancy between modelled and measured concentrations in low and high ranges? How the model was validated?”

**Author Response:** Validation of the atmospheric transport model FRAME (Fine Resolution Ammonia Exchange) in estimating atmospheric concentrations and deposition rates of gaseous NH<sub>3</sub> and particulate NH<sub>4</sub><sup>+</sup> have previously been made by comparison with measurements from the UK NAMN (Dore et al., 2007) and by comparison with other models (Dore et al., 2015). When compared with other atmospheric chemistry transport models, FRAME was found to correlate well with measured NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> concentrations from NAMN (Dore et al., 2015). The comparison of NAMN NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> measurements with modelled NH<sub>3</sub> concentrations from the FRAME model in the paper is made for an example year of 2012 in the paper, This updates an earlier inter-comparison assessment carried out by Dore et al. (2007) for the year 2002, and demonstrates that the FRAME model is performing well in describing the spatial distribution of both NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup>.

The FRAME model uses a database of NH<sub>3</sub> emissions with a 5×5 km grid-square resolution as input. In the present comparison of the FRAME model estimates (based on 2012 UK AENEID NH<sub>3</sub> emission data) with the NAMN measurement results for 2012 (Figure 5), the network annual mean concentrations for each site is compared against the model estimate for the 5-km grid square in which it occurs, and the point is classified according to the estimated dominant source sector of the grid square. Both the model outputs and the measurement agree that background and sheep sites are characterised by small NH<sub>3</sub> concentrations (< 1  $\mu\text{g NH}_3 \text{ m}^{-3}$  annual mean), while agricultural areas, particularly areas with intensive pig and poultry areas, are associated with large NH<sub>3</sub> concentrations (up to 8  $\mu\text{g NH}_3 \text{ m}^{-3}$  annual mean). Overall, the comparison suggests a fairly good fit with regard to both the magnitude and spatial variability of NH<sub>3</sub> concentrations at a national scale, with an R<sup>2</sup> value of 0.6 (**Error! Reference source not found.**). The results of the network thus broadly support the predictions of the FRAME model, lending support to the AENEID model outputs. There are however, systematic differences in the comparison of FRAME and the measurements, depending on the air concentration and dominant source. Figure 5 shows that concentrations are overestimated by FRAME in areas dominated by cattle, pig and poultry, compared with the measurement data,

while the results agree well in sheep and non-agricultural areas. Possible reasons for the overestimation of cattle, pig and poultry farming by FRAME compared to the measurements may be due to the following:

5 a) spatial location of the sampling site relative to the distribution of sources. Ammonia exhibits large sub-grid spatial variability (Dragosits et al. 2002), with the result that single site measurements may not reflect the concentrations across the 5 km grid squares. For example, at many of the sites where the model overestimates concentrations, the measurements are in fact made in nature reserves, which would on average be more distant from sources than assumed in the FRAME 5 km average estimates, thereby underestimating concentrations. This effect is particularly important in areas with high local variability in ammonia emissions, such as intensive agricultural areas. This illustrates the importance of having a large number of sites for comparison,

b) accuracy of the emissions data that are critical to the performance of the model. For example accuracy of emission factors for different livestock classes affecting the model estimates, or

15 c) that dispersion in the model is slightly underestimated. Clearly further work is required to address these questions.

Dore, A. J., Vieno, M., Tang, Y. S., Dragosits, U., Dosio, A., Weston, K. J., & Sutton, M. A. (2007). Modelling the atmospheric transport and deposition of sulphur and nitrogen over the United Kingdom and assessment of the influence of SO<sub>2</sub> emissions from international shipping. *Atmospheric Environment*, 41(11), 2355-2367.

20 Dore, A. J., Carslaw, D. C., Braban, C., Cain, M., Chemel, C., Conolly, C. & Lawrence, S. (2015). Evaluation of the performance of different atmospheric chemical transport models and inter-comparison of nitrogen and sulphur deposition estimates for the UK. *Atmospheric Environment*, 119, 131-143.

25  
30 *Re: "At sites with low concentrations samplings were performed in a clearing of forests. Question is: do model predict concentrations for layer above the canopy or for the ground level, where effect of deposition of the nearby forest is substantial? It would be the source of another bias between the modelled and measured ammonia concentrations."*

35 **Author Response:** FRAME is a Lagrangian model that incorporates horizontal and vertical gradients of NH<sub>3</sub> and calculates vertical concentration profiles with diffusion through 33 layers of varying depth. The modelled concentrations output is from the 1-2 m layer, used to compare with NAMN measurements that are made at approx. 1.5 m above ground.

40 Additional text has been added to Section 2.1, paragraph 1, after the first sentence to provide further information on siting of sites to provide representative measurements:

45 *"The network covers a wide distribution of monitoring sites with measurements in both agricultural and semi-natural areas. Monitoring locations are sited away from point sources (> 150 m) such as farm buildings, which avoids overestimating NH<sub>3</sub> concentrations compared with the grid square, since the aim is to provide meso-scale and regional patterns. In addition, where sampling is carried out in woodland areas, it is made in clearings."*

50 *Re: "Other possible source of bias could be derived by the difference between the monthly sampling applied in the NAMN network and the sampling/measurement method for the validation of model. Are there any inter-comparison among the methods described in this manuscript and other methods based on daily or shorter time basis?"*

55 **Author Response:** Comparison between model and measurement discussed in this paper is based on annual concentrations. Annual mean concentrations from NAMN are derived from the mean of monthly measured concentrations.

Inter-comparison among the methods described in this manuscript:

ALPHA vs Daily Annular Denuder Method – see section 2.2.2, last paragraph:

In the USA (Puchalski et al., 2011), the ALPHA samplers performed well against a reference annular denuder method with a median relative percent difference of -2.4%.

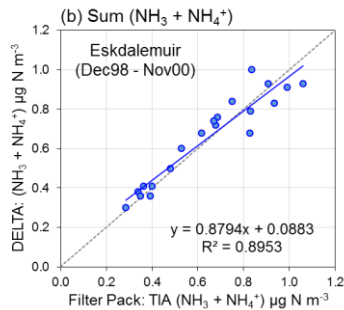
5 The ALPHA and DELTA methods described in this manuscript have also been compared with other methods with shorter time resolution and performed well. Some examples and references are given below.

Comparison with different methods with daily timescales:

DELTA vs Daily Annular Denuder method – see Tang et al., 2009

10 Tang, Y. S., Simmons, I., van Dijk, N., Di Marco, C., Nemitz, E., Dammgén, U., Gilke, K., Djuricic, V., Vidic, S., Gliha, Z., Borovecki, D., Mitosinkova, M., Hanssen, J. E., Uggerud, T. H., Sanz, M. J., Sanz, P., Chorda, J. V., Flechard, C. R., Fauvel, Y., Ferm, M., Perrino, C., and Sutton, M. A.: European scale application of atmospheric reactive nitrogen measurements in a low-cost approach to infer dry deposition fluxes, *Agriculture Ecosystems & Environment*, 133, 183-195, doi.org/10.1016/j.agee.2009.04.027, 2009.

15 DELTA vs Daily Filter Pack (EMEP method) – see Tang et al., 2017 (unpublished data, paper in prep)



20 Comparison of total inorganic ammonium, TIA (sum of  $\text{NH}_3 + \text{NH}_4^+$ ) concentrations at the Eskdalemuir monitoring station (EMEP station code = GB0002R; UK-AIR ID = UKA00130) measured under the EMEP program with concentrations of the corresponding gas and aerosol from the UK National Ammonia Monitoring Network (NAMN,  $\text{NH}_3$  and  $\text{NH}_4^+$ ). EMEP values (data downloaded from <http://ebas.nilu.no/>) are means of daily measurements for TIA by the EMEP filter pack method, matched to the NAMN sampling periods (monthly). Filter pack measurements at Eskdalemuir terminated in December 2000.

Comparison with different methods with shorter timescales:

DELTA and ALPHA vs AMOR at Zegfeld (ID 633; Dutch National Air Quality Monitoring Network, LML):

25 An intercomparison of  $\text{NH}_3$  measurements by the RIVM AMOR system (hourly) and the CEH DELTA sampling system (monthly) have been carried out at the Zegfeld site in the Netherlands since July 2003. Since September 2012, CEH ALPHA measurements have also been included. To compare results, monthly mean concentrations were derived from the average of hourly AMOR data for the corresponding DELTA and ALPHA monthly sampling periods. The long-term comparison with the AMOR at Zegfeld, NL, has been added to the Supplementary Material.

30 The following text is added to the Manuscript in Section 2.2.4 and the Graph is Supp. Figure S6)

“An intercomparison of  $\text{NH}_3$  measurements by the RIVM AMOR system (hourly, Wyers et al., 1993;) and the DELTA sampling system (monthly) have been carried out at the Zegfeld site (ID 633) in the Dutch National Air Quality Monitoring Network (van Zanten et al., 2017) since July 2003. Since September 2012, ALPHA

35 measurements have also been included. To compare results, monthly mean concentrations were derived from the average of hourly AMOR data for the corresponding DELTA and ALPHA monthly sampling periods with good agreement (supp. Figure S6).”

40

*Re: “Methods The sampling and analytical methods need more detailed descriptions. Detection limit precision, sensitivity if any should be mentioned for all sampling and analytical procedures (where appropriate).”*

5 **Author Response:** Detailed descriptions of the DELTA and ALPHA methodologies are available from the references provided in the paper (Sutton et al., 2001a, 2001c and Tang et al., 2001a). We feel that sufficient analytical details have already been provided in section 2.2.1. DELTA methods and section 2.2.2. Passive Methods. Some additional text describing sample analysis and LODs has been added at end of Section 2.2.3,

10 Chemical analysis:

15 “The extracted samples were analysed for  $\text{NH}_4^+$  against a series of  $\text{NH}_4^+$  standards and quality controls. Parallel analysis of laboratory and field blank (unexposed) samples were used to determine the amounts of  $\text{NH}_4^+$  derived from  $\text{NH}_3$  and  $\text{NH}_4^+$  in the atmosphere during transport and storage. The limit of detection (LOD) calculation of the ALPHA and DELTA methodologies are determined as three times the standard deviations of the laboratory blanks. For the DELTA method, the LODs were  $0.01 \mu\text{g m}^{-3}$  for gaseous  $\text{NH}_3$  and  $0.02 \mu\text{g m}^{-3}$  for particulate  $\text{NH}_4^+$ . For the ALPHA method, the LOD was determined as  $0.03 \mu\text{g m}^{-3}$ ”

5 *Re: "Interpretations Manuscript has too many figures and tables. I suggest to reduce them. For example Fig 11 relationships among rainfall amount, temperature and ammonia emission can hardly be seen. Moreover this kind of relations have still demonstrated by Fig. 9. Also, for figures 12 and 13. One of them is unnecessary. It should be decided what is the more representative interpretation statistically, the trend of yearly or monthly data. I believe the latter. Do not repeat information both by figures and in tables. Statistical parameters are displayed in figures and also in tables (e.g. figures 13, 14, 15 and corresponding tables). Also there are redundancies with figures 17a and 18a."*

10 **Author Response:** The authors feel that there is justification for the number of graphs presented as each has a particular purpose. We have explained this for each graph below, though we have agreed that Figure 17a was redundant as similar data was also shown in Figure 18a.

*Re: "Figure 11"*

15 **Author Response:** Figure 1 demonstrates the relationships between monitored monthly mean NH<sub>3</sub> concentrations with locally available monthly temperature and rainfall data at an example site. Figure 11 on the other hand provides an important comparison on a national level the annual mean NH<sub>3</sub> concentrations of all NAMN sites with UK annual mean temperature and rainfall. We strongly suggest to retain this graph, since it shows the strong inter- and intra-annual variability in the parameters considered. The annually averaged data of all sites masks considerable spatial and seasonal variability in NH<sub>3</sub> concentrations. Drivers contributing to this variability include the influence of climate on emissions, variations in management practice for a particular emission source, and influence of local emission sources and chemical interactions with other chemical species on NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> concentrations that are discussed in the paper.

*Re: "Figures 12 and 13 - one of them is unnecessary. It should be decided what is the more representative interpretation statistically, the trend of yearly or monthly data. I believe the latter"*

25 **Author Response:** The authors feel that both annual and seasonal variability are of equal interest. Since ammonia is strongly seasonally variable, it is important to demonstrate this graphically. In addition, the monthly data allows the seasonality to be accounted for in the seasonal Mann-Kendall test, which takes into account the 12 month seasonality and tests whether there is a trend not due to seasonality.

30 *Re: "Statistical parameters are displayed in figures and also in tables (e.g. figures 13, 14, 15 and corresponding tables)."*

**Author Response:** The data has been shown graphically and in table form in order for data transparency and so that readers may use the parameterisations if they wish.

*Re: "Also there are redundancies with figures 17a and 18a."*

**Author Response:** Agree, Figure 17a deleted.

35 *Re: "Conclusion Too long and overlaps with discussions. It has to cut insisting only on the most important findings."*

40 **Author Response:** We feel that the conclusions are not too long and are not particularly discursive. Therefore we would prefer to leave them as they are.

2) Specific comments:

*"Page 2: First paragraph: Authors should describe the mechanism, how SO<sub>2</sub> reduction influences the concentration and deposition of ammonia; here or/and in line 31, in 3.5.6.,line 15 on page 17. Namely the decreased efficiency of co-deposition of SO<sub>2</sub> and NH<sub>3</sub> onto surfaces."*

**Author Response:** the following text has been added section 3.5.6, after paragraph 2:

"Dry deposition of SO<sub>2</sub> and NH<sub>3</sub> are enhanced in the presence of both gases, an interaction referred to as "co-deposition" (Fowler et al., 2001). The acid-base neutralization by each of the gases provides an efficient sink for dry deposition on leaf surfaces and deposition enhancement for each gas depends on the relative air concentrations of NH<sub>3</sub> and SO<sub>2</sub>. For SO<sub>2</sub>, the dry deposition process has been shown to be strongly influenced by ambient concentrations of NH<sub>3</sub> because the surface resistance is regulated mainly by uptake in moisture on foliar surfaces, which, in turn, is strongly influenced by the presence of NH<sub>3</sub>. The large reduction in SO<sub>2</sub> emissions and ambient concentrations, compared with the relative stagnation in NH<sub>3</sub> emissions and concentrations over the same period has meant that the SO<sub>2</sub>/NH<sub>3</sub> ratio has decreased dramatically. This has led to a systematic decrease in canopy resistance to uptake of SO<sub>2</sub> on surfaces, increasing dry deposition of SO<sub>2</sub> in the UK (ROTAP 2012). The underlying cause of the decrease in surface resistance is that the ambient NH<sub>3</sub> is sufficient to neutralize acidity from the solution and oxidation of deposited SO<sub>2</sub>, maintaining large rates of deposition."

*"Line 13: SO<sub>2</sub> and NO<sub>x</sub> are not acids, but acid anhydrides (as to SO<sub>2</sub> and NO<sub>2</sub>)."*

**Author Response:** text changed from "acids" to "acid gases"

*"Line 13: emitted "mainly" form combustion processes. (Do not forget natural sources esp. on global scale)."*

**Author Response:** text added "and from natural sources"

*"Line 14: Primary product of neutralization is the NH<sub>4</sub>HSO<sub>4</sub> followed by forming (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> only in case when ammonia is available in quantity enough."*

**Author Response:** text added. See Author response 1c. on page 2.

*"Line 16: do not forget the role of PM in cloud/for formation as condensation nuclei."*

**Author Response:** We have modified the sentence ....

*"The effects of PM on atmospheric visibility, radiative scattering, cloud formation (and resultant climate effects)..."*



Page 4:

“Lines 17-18: The sentence “the network has a good representation in the middle air concentration classes of 3-4  $\mu\text{g m}^{-3}$ ” does not agree with Fig. 1c where measured concentration in the range of 3-4.5  $\mu\text{g m}^{-3}$  is doubled. I would state instead “the network has a good representation in the middle air concentration classes of 0.5-3  $\mu\text{g m}^{-3}$ ”, so it is true. Otherwise it would make questionable the statement in lines 24-25, but this correlation should be justified also by figures.”

**Author Response:** We changed the sentence to:

“..the network has a good representation in the middle air concentration classes of 0.5 – 1.5  $\mu\text{g m}^{-3}$  (33 % of NAMN sites, compared with 29 % of all FRAME 5 km x 5 km grid squares) and 1.5 - 3  $\mu\text{g m}^{-3}$  (32 % of NAMN sites, compared with 39 % of all FRAME 5 km x 5 km grid squares), but with .... (Figure 2c).

“Page 5: line 18: clarify the filter pack. I suppose the first filter is a Teflon one to capture particles.”

**Author Response:** The filter is cellulose impregnated with citric acid. The word “cellulose” has been added to the sentence. There is no second filter.

“Page 7: lines 12-13: was the two instruments inter-calibrated?”

**Author Response:** Yes the passive diffusion tube and ALPHA method are calibrated against the active sampling DELTA method on a monthly basis as discussed in Section 2.2.2 paragraph 3.

“Page 18, lines 4-5: the formation of ammonia takes place by the same procedure with the same kinetic parameters, so cannot be “slower” rather less effective.”

**Author Response:** The authors agree and think this was a typo. We have revised the text changing “slower” to “lower”

“Fig. 4 is not demonstrative to me, e.g. the relation between discrete measurement points and emission map is hardly seen. An iso-line picture for ammonia instead of discrete figures would show better the situation, but the comparison of emission with concentration in this way has not much sense, because the effect of transport and transformation processes.”

Re: “An iso-line picture for ammonia instead of discrete figures would show better the situation”

**Author Response:** Interpolated concentration maps have not been used since the interpolation of the discrete measurement points (e.g. using bilinear interpolation) will give the reader a false sense of the spatial variability of air concentrations from the limited number of measurement locations. The number of sites required to replicate the spatial resolution of the emissions map (5 km x 5 km grid resolution) will be impossibly high. The measured annual concentrations have therefore been shown as coloured dots on the map to show the observed spatial variability across the UK. Interpolated concentration maps can be produced from the discrete measurement points and added in Supplementary Materials, if required.

Re: *“but the comparison of emission with concentration in this way has not much sense, because the effect of transport and transformation processes”*

5 **Author Response:** The authors acknowledge that concentrations are affected by effect of transport and transformation processes, but at the same time, concentrations are also largely driven by emissions. The large variability in NH<sub>3</sub> emissions across the UK is reflected by both modelled (FRAME) and observed spatial variability in NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> concentrations, with largest concentrations in the largest emission source areas and lowest in background areas. The comparison of the measured concentrations (shown as discrete point data) with the emissions maps support this in Figure 4.

10 The FRAME model uses as input annual ammonia emissions data from the UK National Atmospheric Emissions Inventory (<http://naei.defra.gov.uk/>) and incorporates the main atmospheric processes (emission, diffusion, chemistry and deposition) to calculate annual NH<sub>3</sub> concentration fields in the UK at a 5 km x 5 km grid resolution. The spatial variability estimated by the FRAME model mirrors the variability in emissions across the country, with largest concentrations in the largest emission source areas.

20 *“Figures involving temperature relating ammonia concentration: not mentioned but I believe they are air temperatures. But, emission of ammonia rather depends on soil surface temperature since decomposition of manure happens in the upper layer of soil. I know, soil temperature strongly correlates with air temperature but it has to be mentioned.”*

25 **Author Response:** We confirm that the temperatures used are air temperatures, which are more available across the domain than soil temperatures. As Prof. Horvath points out, the soil and air temperatures do correlate.

*“Figure 16b: mean NH3 of what? Square bracket suggests it is molar concentration, but mass concentration was used all over the MS. Better to name “NH3 concentration” on the axis and avoid bracket.”*

30 **Author Response:** The authors agree and have adjusted the graph axis label and the figure caption.

35 *“How concentrations in Fig. 16b were calculated? Did authors split the ammonia concentration among the number of animals, taking into account the variation of the latter? How other sources were taken into account? Others than cows, pigs, poultry takes 1/3 of total emission. What does it mean “Total” in Fig.16. I suppose this is the total of cows, pigs, and poultry only rather than total emission from all of sources. Otherwise the blue line on Fig 16b should be uniform with pink one on 17a. Explain please in the legend. On the other hand concentrations does not directly relate to emission to compare.”*

40 **Author Response:** The authors apologise, we omitted a description of Figure 16 in the text. This has now been added to Section 3.5.5 paragraph 3 as follows:

45 *“In Figure 16, the relative changes in UK emissions between 1998 and 2014 are compared with relative changes in mean measured NH<sub>3</sub> concentrations for all NAMN sites, and for grouped sites classified as dominated by cattle, pigs & poultry, and sheep.”*

*Fig. 18a: nanomoles per what? Cubic meter?*

50 **Author Response:** Nanomoles per cubic metre. Axis on graph corrected.

*3) Technical comments*

*“Figures: use unambiguous and uniform in legends of vertical axes. E.g. concentration or emission of something (dimension in bracket).”*

55 **Author Response:** Most graphs are systematically labelled but we have adjusted Figure 18 to put the percentage in brackets.

*“Fig. 7, 8: Split the two figures (a and b), vertical axis of “b” is too close to “a””*

**Author response:** Thank you for spotting this. We have adjusted the a and b so they are separated more widely.

*“Use greek mü instead u for micro in all figures”*

5

**Author Response:** Yes, we have checked and updated all figures.  
Figures 8a, 8b corrected  
Supp. Figure 1a, 1b corrected

10

REVIEWER 2:

**Anonymous Referee #2**

Received and published: 9 June 2017

5 **RESPONSE TO REVIEWER**

We have carefully considered Referee #2's comments.

*"The trend analysis in this study is superficial and does not meet a criteria for publishing in a high-impacted journals such as ACP."*

10

The objective of the statistical trend analysis presented in our research paper was to identify trends in the long-term datasets (univariate monotonic, see e.g. Hirsch et al., 1991), estimate the rate of change and to address the question of whether trends in NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> concentrations (if any) are consistent with the changes in estimated UK annual NH<sub>3</sub> emissions (data downloaded from: <http://naei.beis.gov.uk/data/data-selector-results?q=101505>)?" The dataset is sufficiently long-term (i.e. gaseous NH<sub>3</sub>: 17 years and particulate NH<sub>4</sub><sup>+</sup>: 16 years) and collected by consistent methods, to allow for effective statistical trend analyses to be carried out.

15

20

To identify and quantify monotonic trends in the paper, trend assessment was carried out using (i) linear regression (LR), (ii) Mann-Kendall (MK) test (Hirsch et al., 1981; Gilbert, 1987) on annually averaged and monthly mean data, and (iii) Seasonal Mann-Kendall (SMK) test (Hirsch et al., 1982) on monthly data only. We think that this is not a superficial trend analysis - rather we applied the relevant methodologies. We referred to overviews of some of the more widely used techniques in time series modelling and analysis are widely available (see e.g., Chatfield, 2016; Hamilton, 1994; Meals et al., 2011). Online resources (e.g. <https://cran.r-project.org/web/views/TimeSeries.html>) also provide information on the range of statistical tests to identify and quantify trends in environmental data.

25

30

It is noted that the non-parametric Mann-Kendall (MK) statistical approach is also commonly employed to detect monotonic trends in series of environmental data in many papers and scientific reports (e.g. Colette et al., 2012., Gurreiro et al., 2014, Li et al., 2016, Meals et al., 2011; Serrano et al., 1999; Torseth et al., 2012., Yao et al., 2016) and hydrological data (e.g. Hirsch et al., 1981, 1982).

35

Trend analysis using the Mann-Kendall approach are also described in publications by ACP (e.g., Gurreiro et al., 2014, Li et al., 2016, Torseth et al., 2012., Yao et al., 2016). The advantages of the MK approach over linear regression for trend assessments are in that (i) it does not require normally distributed data, (ii) it is not affected by outliers, and (iii) it removes the effect of temporal auto-correlation in the data. The Seasonal Kendall test deployed also is highly robust and relatively powerful, recommended for water quality trend monitoring (Meals et al., 2011) and most recently applied in air pollution trend assessments in Europe (Colette et al., 2016; Torseth et al., 2012).

40

The cause of the trends were subsequently interpreted in terms of three main drivers:

- 1) Meteorological: influence of temperature/rainfall
- 2) Changes in emissions from 3 dominant source sectors (cattle, pigs & poultry, sheep)
- 3) Changes in chemical climate, e.g. effects of large decrease in SO<sub>2</sub> emissions and concentrations on co-deposition relationship of NH<sub>3</sub> with SO<sub>2</sub>, and shift in form of particulate NH<sub>4</sub><sup>+</sup> from (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> to NH<sub>4</sub>NO<sub>3</sub>.

45

*"The authors are strongly encouraged to conduct a literature review for which trend analysis tools are the most suitable for this work."*

A literature review for trend analysis tools as suggested by the reviewer is considered outwith the scope of this research paper.

We have added a sentences discussing the previous use of trend analysis methods by EMEP and in UK Air quality monitoring network reports - primary users of these datasets. As noted in the text of the manuscript both analysis methods lead to similar results.

5 Incorrect adopting trend analysis tools also leads that several discussion such as "Trends in NH<sub>3</sub> concentrations vs trends in NH<sub>3</sub> emissions", "Influence of climate" and "Influence of local emission sources" is full of augments and lack of solid scientific values. The reviewer believes a substantial revision to be required to make the current version publishable

10 Given our opinion that we have used appropriate methods, and a lack of detailed critique by Reviewer #2, we are unable to directly respond to this comment, however do not think a substantial revision is required.

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45

## Manuscript changes:

- Section 1 (paragraph 2): text changed from “acids” to “acid gases”
- Section 1 (paragraph 2): text added “and from natural sources”
- Section 1 paragraph 2: Text modified:

5 “The effects of PM on atmospheric visibility, radiative scattering, cloud formation (and resultant climate effects)…”

- Section 2.1, paragraph 1: additional text added:

10 “The network covers a wide distribution of monitoring sites with measurements in both agricultural and semi-natural areas. Monitoring locations are sited away from point sources (> 150 m) such as farm buildings, which avoids overestimating NH<sub>3</sub> concentrations compared with the grid square, since the aim is to provide meso-scale and regional patterns. In addition, where sampling is carried out in woodland areas, it is made in clearings.”

- Section 2.1 paragraph 2: Text modified to:

15 “..the network has a good representation in the middle air concentration classes of 0.5 – 1.5 µg m<sup>-3</sup> (33 % of NAMN sites, compared with 29 % of all FRAME 5 km x 5 km grid squares) and 1.5 - 3 µg m<sup>-3</sup> (32 % of NAMN sites, compared with 39 % of all FRAME 5 km x 5 km grid squares), but with .... (Figure 2c).

- Section 2.2.1 paragraph 3: The word “cellulose” added in.

- Section 2.2.3:

20 “The extracted samples were analysed for NH<sub>4</sub><sup>+</sup> against a series of NH<sub>4</sub><sup>+</sup> standards and quality controls. Parallel analysis of laboratory and field blank (unexposed) samples were used to determine the amounts of NH<sub>4</sub><sup>+</sup> derived from NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> in the atmosphere during transport and storage. The limit of detection (LOD) calculation of the ALPHA and DELTA methodologies are determined as three times the standard deviations of the laboratory blanks. For the DELTA method, the LODs were 0.01 µg m<sup>-3</sup> for gaseous NH<sub>3</sub> and 0.02 µg m<sup>-3</sup> for particulate NH<sub>4</sub><sup>+</sup>. For the ALPHA method, the LOD was determined as 0.03 µg m<sup>-3</sup>”

- Section 2.2.4 and Graph is Supp. Figure S6

30 “An intercomparison of NH<sub>3</sub> measurements by the RIVM AMOR system (hourly, Wyers et al., 1993;) and the DELTA sampling system (monthly) have been carried out at the Zegveld site (ID 633) in the Dutch National Air Quality Monitoring Network (van Zanten et al., 2017) since July 2003. Since September 2012, ALPHA measurements have also been included. To compare results, monthly mean concentrations were derived from the average of hourly AMOR data for the corresponding DELTA and ALPHA monthly sampling periods with good agreement (supp. Figure S6).”

- Section 2.2.5 paragraph 1 first sentence: word “out” added in.

- Section 2.2.5, paragraph 2: text modified and added to:

40 *The main advantages, as discussed in the literature of the MK approach over linear regression for trend assessments are that (i) it does not require normally distributed data, (ii) it is not affected by outliers, and (iii) it removes the effect of temporal auto-correlation in the data. The MK approach are widely used in environmental time series assessments, e.g. long-term trends in precipitation (Serrano et al. 1999) and long-term trends in European air quality (EMEP Colette et al., 2016; Torseth et al., 2012). However linear trend assessment have been used in UK air quality monitoring network reports (e.g. Conolly et al., 2016), therefore both approaches were used in this paper primarily as a quality assurance check.*

- Section 3.3, paragraph 5: has been expanded to include explanation of the equilibrium between gas and aqueous aerosol phase as drivers in the seasonal variations of particulate NH<sub>4</sub><sup>+</sup>.

45 Text deleted:

50 *For particulate NH<sub>4</sub><sup>+</sup>, as expected for a secondary pollutant, concentrations are more decoupled from the dominant NH<sub>3</sub> source sectors in the vicinity of a site. The seasonal trends in NH<sub>4</sub><sup>+</sup> are broadly similar for the four source sectors shown in Figure 7b, with the magnitude of the NH<sub>4</sub><sup>+</sup> concentrations reflecting NH<sub>3</sub> concentrations at a regional level. Here the spring peak is more driven by long-range transboundary transport, e.g. influence from continental Europe (Vieno et al., 2014). Nevertheless, it is notable that the winter minima for*

*NH<sub>4</sub><sup>+</sup> aerosol concentrations at sheep and background sites are more pronounced than that for pig, poultry and cattle dominated sites. This may be a result of a combination of smaller NH<sub>3</sub> emissions in winter in these areas (as indicated by Figure 7a) and differences in long-range transport to the more remote areas in winter conditions.*

Text added:

5 *For particulate NH<sub>4</sub><sup>+</sup>, as expected for a secondary pollutant, concentrations are more decoupled from the*  
*dominant NH<sub>3</sub> source sectors in the vicinity of a site. Although the formation of particulate NH<sub>4</sub><sup>+</sup> primarily*  
*depends on the occurrence of NH<sub>3</sub> in the atmosphere, synoptic meteorology and long range transboundary*  
*transport from continental Europe are important drivers influencing the seasonal variations of NH<sub>4</sub><sup>+</sup> across the*  
*UK, due to its' longer lifetime. The seasonal trends in particulate NH<sub>4</sub><sup>+</sup> are seen to be broadly similar for the*  
10 *four different emission source sectors (Figure 8b), with the magnitude of the NH<sub>4</sub><sup>+</sup> concentrations reflecting NH<sub>3</sub>*  
*concentrations at a regional level. In the atmosphere, particulate NH<sub>4</sub><sup>+</sup> are primarily in the form of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>*  
*and NH<sub>4</sub>NO<sub>3</sub>, formed when the acid gases HNO<sub>3</sub> and H<sub>2</sub>SO<sub>4</sub>.in the atmosphere are neutralised by NH<sub>3</sub> (Putaud*  
*et al., 2010). NH<sub>3</sub> preferentially neutralizes H<sub>2</sub>SO<sub>4</sub> due to its low saturation vapour pressure (forming NH<sub>4</sub>HSO<sub>4</sub>*  
*then (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>), while NH<sub>4</sub>NO<sub>3</sub> is formed when abundant NH<sub>3</sub> is available. In contrast to (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>,*  
15 *NH<sub>4</sub>NO<sub>3</sub> is a semi-volatile component (Stelson & Seinfeld, 1982). Long-term data from the UK Acid Gas and*  
*Aerosol Network (AGANet, Conolly et al., 2016) shows a change in the particulate phase of NH<sub>4</sub><sup>+</sup> from*  
*(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> to NH<sub>4</sub>NO<sub>3</sub>, with particulate nitrate concentrations exceeding that of particulate sulphate*  
*approximately three-fold (on a molar basis) (Fig. 18a). This suggests that the thermodynamic equilibrium between*  
*the gas phase NH<sub>3</sub> and HNO<sub>3</sub> and the aerosol phase NH<sub>4</sub>NO<sub>3</sub> will have a much greater effect on the seasonal*  
20 *concentrations of NH<sub>4</sub><sup>+</sup> than (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>. The formation and dissociation of NH<sub>4</sub>NO<sub>3</sub> depend strongly on*  
*ambient temperature and humidity (Stelson and Seinfeld, 1982). Warm, dry weather in summer promotes*  
*dissociation, decreasing particulate phase NH<sub>4</sub>NO<sub>3</sub> relative to gas phase NH<sub>3</sub> and HNO<sub>3</sub>. During the winter*  
*months, low temperature and high humidity favour the formation of NH<sub>4</sub>NO<sub>3</sub> from the gas phase NH<sub>3</sub> and HNO<sub>3</sub>.*  
*By contrast, the spring peak in NH<sub>4</sub><sup>+</sup> concentrations may be attributed to photochemical processes (elevated*  
25 *ozone) leading to enhanced formation of HNO<sub>3</sub> during this period (Pope et al., 2016) and also to import of*  
*particulate NO<sub>3</sub>- through long-range transboundary transport, e.g. from continental Europe, as discussed in*  
*Vieno et al. (2014). Nevertheless, it is notable that the winter minima for NH<sub>4</sub><sup>+</sup> aerosol concentrations at sheep*  
*and background sites are more pronounced than that for pig, poultry and cattle dominated sites. This may be a*  
*result of a combination of smaller NH<sub>3</sub> emissions in winter in these areas (as indicated by Figure 8a) and*  
30 *differences in long-range transport to the more remote areas in winter conditions.*

- Section 3.5.5 text added:

*"In Figure 16, the relative changes in UK emissions between 1998 and 2014 are compared with relative changes in mean measured NH<sub>3</sub> concentrations for all NAMN sites, and for grouped sites classified as dominated by cattle, pigs & poultry, and sheep."*

- Section 3.5.6, after paragraph 2:

*"Dry deposition of SO<sub>2</sub> and NH<sub>3</sub> are enhanced in the presence of both gases, an interaction referred to as "co-deposition" (Fowler et al., 2001). The acid-base neutralization by each of the gases provides an efficient sink for dry deposition on leaf surfaces and deposition enhancement for each gas depends on the relative air concentrations of NH<sub>3</sub> and SO<sub>2</sub>. For SO<sub>2</sub>, the dry deposition process has been shown to be strongly influenced by ambient concentrations of NH<sub>3</sub> because the surface resistance is regulated mainly by uptake in moisture on foliar surfaces, which, in turn, is strongly influenced by the presence of NH<sub>3</sub>. The large reduction in SO<sub>2</sub> emissions and ambient concentrations, compared with the relative stagnation in NH<sub>3</sub> emissions and concentrations over the same period has meant that the SO<sub>2</sub>/NH<sub>3</sub> ratio has decreased dramatically. This has led to a systematic decrease in canopy resistance to uptake of SO<sub>2</sub> on surfaces, increasing dry deposition of SO<sub>2</sub> in the UK (ROTAP 2012). The underlying cause of the decrease in surface resistance is that the ambient NH<sub>3</sub> is sufficient to neutralize acidity from the solution and oxidation of deposited SO<sub>2</sub>, maintaining large rates of deposition."*

- Section 3.5.6 paragraph 5: reference to Figure 17a and SuppFigS4 removed
- Section 3.5.6, paragraph 6: "slower" corrected to "lower"
- Section 4 paragraph 1: References Vieno et al 2014 and Dore et al 2015 removed
- Section 4 paragraph 5: space removed before "Overall"

- Acknowledgements: “colleagues at RIVM” added
- References added:

5 Colette, A., Aas, W., Banin, L., Braban, C.F., Ferm, M., Ortiz, A., Ilyin, I., Mar, K., Pandolfi, M., Putaud, J.P., and Shatalov, V.: Air pollution trends in the EMEP region between 1990 and 2012. Joint Report of the EMEP Task Force on Measurements and Modelling (TFMM), Chemical Co-ordinating Centre (CCC), Meteorological Synthesizing Centre-East (MSC-E), Meteorological Synthesizing Centre-West (MSC-W), EMEP/CCC-Report 1/2016, 2016.

10 Fowler, D., Sutton, M.A., Flechard, C., Cape, J.N., Storeton-West, R., Coyle, M., and Smith, R.I.: The control of SO<sub>2</sub> dry deposition on to natural surfaces by NH<sub>3</sub> and its effects on regional deposition. *Water, Air, & Soil Pollution: Focus*, 1(5), 39-48, 2001.

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20 Putaud, J.P., Van Dingenen, R., Alastuey, A., Bauer, H., Birmili, W., Cyrys, J., Flentje, H., Fuzzi, S., Gehrig, R., Hansson, H.C. and Harrison, R.M. A European aerosol phenomenology–3: Physical and chemical characteristics of particulate matter from 60 rural, urban, and kerbside sites across Europe. *Atmospheric Environment*, 44(10), 1308-1320, doi:10.1016/j.atmosenv.2009.12.011, 2010.

25 Stelson, A. W., and Seinfeld, J. H.: Relative humidity and temperature dependence of the ammonium nitrate dissociation constant, *Atmospheric Environment*, 16, 983-992, doi: 10.1016/0004-6981(82)90184-6, 1982.

- References removed:

30 EMEP: Air pollution trends in the EMEP region between 1990 and 2012. Joint Report of the EMEP Task Force on Measurements and Modelling (TFMM), Chemical Co-ordinating Centre (CCC), Meteorological Synthesizing Centre-East (MSC-E), Meteorological Synthesizing Centre-West (MSC-W). EMEP/CCC-Report 1/2016, 2016.

- Figures added:

S6

- Figures deleted:

Figure 17a

- Figures Modified:

Figure 7: adjusted the a and b so they are separated more widely

Figure 8: adjusted the a and b so they are separated more widely; greek mu inserted for u

35 Figure 16b: graph axis label and the figure caption modified.

Figure 18 a: Axis on graph modified

Figure S1 greek mu inserted for u

40



## Drivers for spatial, temporal and long-term trends in atmospheric ammonia and ammonium in the UK

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**Abstract.** A unique long-term dataset from the UK National Ammonia Monitoring Network (NAMN) is used here to assess spatial, seasonal and long-term variability in atmospheric ammonia (NH<sub>3</sub>: 1998-2014) and particulate ammonium (NH<sub>4</sub><sup>+</sup>: 1999-  
15 2014) across the UK. Extensive spatial heterogeneity in NH<sub>3</sub> concentrations is observed, with lowest annual mean concentrations at remote sites (< 0.2 µg m<sup>-3</sup>) and highest in the areas with intensive agriculture (up to 22 µg m<sup>-3</sup>), while NH<sub>4</sub><sup>+</sup> concentrations show less spatial variability (e.g. range of 0.14 to 1.8 µg m<sup>-3</sup> annual mean in 2005). Temporally, NH<sub>3</sub> concentrations are influenced by environmental conditions and local emission sources. In particular, peak NH<sub>3</sub> concentrations are observed in summer at background sites (defined by 5 km grid average NH<sub>3</sub> emissions <1 kg N ha<sup>-1</sup> y<sup>-1</sup>) and in areas  
20 dominated by sheep farming, driven by increased volatilization of NH<sub>3</sub> in warmer summer temperatures. In areas where cattle, pig and poultry farming is dominant, the largest NH<sub>3</sub> concentrations are in spring and autumn, matching periods of manure application to fields. By contrast, peak concentrations of NH<sub>4</sub><sup>+</sup> aerosol occur in spring, associated with long-range transboundary sources. An estimated decrease in NH<sub>3</sub> emissions by 16 % between 1998 and 2014 was reported by the UK National Atmospheric Emissions Inventory. Annually averaged NH<sub>3</sub> data from NAMN sites operational over the same period  
25 ( $n = 59$ ) show an indicative downward trend, although the reduction in NH<sub>3</sub> concentrations is smaller and non-significant (-6.3 % (Mann-Kendall, MK); -3.1 % (linear regression, LR)). In areas dominated by pig and poultry farming, a significant reduction in NH<sub>3</sub> concentrations between 1998 and 2014 (-22 % (MK); -21 % (LR): annually averaged NH<sub>3</sub>) is consistent with, but not as large as the decrease in estimated NH<sub>3</sub> emissions from this sector over the same period (-39 %). By contrast, in cattle-dominated areas there is a slight upward trend (non-significant) in NH<sub>3</sub> concentrations (+12%, (MK); +3.6% (LR):  
30 annually averaged NH<sub>3</sub>), despite the estimated decline in NH<sub>3</sub> emissions from this sector since 1998 (-11%). At background and sheep dominated sites, NH<sub>3</sub> concentrations increased over the monitoring period. These increases (non-significant) at background (+17 % (MK); +13 % (LR): annually averaged data) and sheep dominated sites (+15 % (MK); +19 % (LR): annually averaged data) would be consistent with the concomitant reduction in SO<sub>2</sub> emissions over the same period, leading to a longer atmospheric lifetime of NH<sub>3</sub>, thereby increasing NH<sub>3</sub> concentrations in remote areas. The observations for NH<sub>3</sub>  
35 concentrations not decreasing as fast as estimated emission trends are consistent with a larger downward trend in annual particulate NH<sub>4</sub><sup>+</sup> concentrations (1999-2014: -47 % (MK); -49 % (LR),  $p < 0.01$ ,  $n = 23$ ), associated with a slower formation of particulate NH<sub>4</sub><sup>+</sup> in the atmosphere from gas-phase NH<sub>3</sub>.

## 1 Introduction

Atmospheric ammonia ( $\text{NH}_3$ ) gas is assuming increasing importance in the global pollution climate, with effects on local to international (transboundary) scales (Fowler et al., 2016). While substantial reductions in  $\text{SO}_2$  emissions and limited reductions in  $\text{NO}_x$  emissions have been achieved in Europe and North America following legislation designed to improve air quality,  $\text{NH}_3$  emissions, primarily from the agricultural sectors (94 % of total  $\text{NH}_3$  emissions in Europe in 2014) have seen much smaller reductions (EEA, 2016). In the period 2000-2014,  $\text{NH}_3$  emissions are estimated to have decreased in the EU-28 (28 member states of the European Union) by only 8 % from 4.3 to 3.9 million tonnes, with the UK contributing 7.2 % in 2014 (EEA, 2016).  $\text{SO}_2$  emissions are estimated to have declined by 69 % and  $\text{NO}_x$  by 39 % across the EU-28 over the same period.

$\text{NH}_3$  is known to contribute significantly to total nitrogen (N) deposition to the environment, and causes harmful effects through eutrophication and acidification of land and freshwaters. This can lead to a reduction in both soil and water quality, loss of biodiversity and ecosystem change (e.g. Pitcairn et al., 1998; Sheppard et al., 2011). In the atmosphere,  $\text{NH}_3$  is the major base for neutralization of atmospheric acid gases, such as  $\text{SO}_2$  and  $\text{NO}_x$  emitted from combustion processes (vehicular and industrial) and from natural sources, to form ammonium-containing particulate matter (PM); primarily ammonium sulphate ( $(\text{NH}_4)_2\text{SO}_4$ ) and ammonium nitrate ( $\text{NH}_4\text{NO}_3$ ). This secondary PM is mainly in the 'fine' mode with diameters of less than  $2.5 \mu\text{m}$  (i.e.  $\text{PM}_{2.5}$  fraction) (Vieno et al., 2014). The effects of PM on atmospheric visibility, radiative scattering, cloud formation (and resultant the greenhouse climate effects) and on human health (bronchitis, asthma, coughing) are well documented (e.g. Kim et al., 2015; Brunekreef et al., 2015). Inputs of  $\text{NH}_3$  and  $\text{NH}_4^+$  (collectively termed  $\text{NH}_x$ ) are the dominant drivers of ecological effects of deposited N, compared with wet deposited  $\text{NH}_4^+$  in rain (UNECE, 2016) and the importance of  $\text{NH}_x$  can be expected to increase further, relative to oxidised N, as  $\text{NO}_x$  emissions have been decreasing faster than  $\text{NH}_3$  emissions (Reis et al., 2012; EEA, 2016; EU, 2016).

In gaseous form,  $\text{NH}_3$  has a short atmospheric lifetime of about 24 hours (Wichink Kruit et al., 2012). It is primarily emitted at ground level in the rural environment, and is associated with large dry deposition velocities to vegetation (Sutton and Fowler, 2002). High  $\text{NH}_3$  concentrations can lead to acute problems at a local scale to, for example, nature reserves located in intensive agricultural landscapes (Sutton et al., 1998; Cape et al., 2009a; Hallsworth et al., 2010; Vogt et al., 2013). The  $\text{NH}_3$  remaining in the atmosphere generally partitions to PM where the  $\text{NH}_4^+$  can have a lifetime of several days (Vieno et al., 2014). Although  $\text{NH}_4^+$  dry deposits at the surface, the primary removal mechanism for  $\text{NH}_4^+$  is thought to be through scavenging of PM by cloud and rain, leading to wet deposition of  $\text{NH}_4^+$  (Smith et al., 2000). Characterising the relationship between  $\text{NH}_3$  emissions and the formation of PM is, however, not straight forward; an increase in  $\text{NH}_3$  emissions does not automatically translate to a proportionate increase in  $\text{NH}_4^+$  (Bleeker et al., 2009). The relationship depends on climate and meteorology as well as the concentration of other precursors to PM formation such as  $\text{SO}_2$  and  $\text{NO}_x$  (Fowler et al., 2009). While it is clear that reductions in  $\text{NH}_3$  emissions will lead to reductions in overall  $\text{NH}_4^+$  concentrations, the relative changes in gaseous  $\text{NH}_3$  and  $\text{NH}_4^+$  particles remains poorly quantified.

International targets have been agreed to reduce  $\text{NH}_3$  emissions to move towards protection against its harmful effects. These include the UNECE Convention on Long-Range Transboundary Air Pollution (CLRTAP) Gothenburg Protocol and the recently revised EU National Emission Ceilings Directive (NECD 2016/2284) (EU, 2016). The 1999 UNECE Gothenburg Protocol is a multi-pollutant protocol to reduce acidification, eutrophication and ground-level ozone by setting emissions ceilings for sulphur dioxide, nitrogen oxides, volatile organic compounds and ammonia, which are to be met by 2020. Revised in 2012, the protocol requires national parties to jointly reduce emissions of  $\text{NH}_3$ , in the case of the EU-28 by 6 % between 2005 and 2020 (Reis et al., 2012). Under the revised NECD (EU, 2016), the EU is also committed to reductions of 6% for

NH<sub>3</sub>, but by a later date of 2029, as well as an additional 13% reduction in NH<sub>3</sub> emission beyond 2030 compared with a 2005 baseline.

Although this demonstrates that there is currently no strong commitment to reduce NH<sub>3</sub> emissions compared with SO<sub>2</sub> and NO<sub>x</sub>, other supporting measures should also be noted including the Industrial Emissions Directive 2010/75/EU (IED), which requires pig and poultry farms (above stated size thresholds) to reduce emissions using Best Available Techniques. The IED applies to around 70 % of the European poultry industry and around 25 % of the pigs industry (UNECE, 2010). In tandem, revised UNECE 'Critical Levels' (CLE) of NH<sub>3</sub> concentrations to protect sensitive vegetation and ecosystems were adopted in 2007 (UNECE, 2007). These set limits of NH<sub>3</sub> concentrations to 1 µg NH<sub>3</sub> m<sup>-3</sup> and 3 µg NH<sub>3</sub> m<sup>-3</sup> annual mean for the protection of lichens-bryophytes and other vegetation, respectively (Cape et al., 2009b). The new CLEs replaced the previous single value of 8 µg NH<sub>3</sub> m<sup>-3</sup> (annual mean) and have since been adopted as part of the revised Gothenburg Protocol. Such CLEs for NH<sub>3</sub> are widely exceeded, including over the areas designated as Special Areas of Conservation (SAC) under the Habitats Directive, and indicates a significant threat to the Natura 2000 network established by that directive (Bleeker et al., 2009; Hallsworth et al., 2010; van Zanten et al., 2017).

Few countries have established systematic networks to measure NH<sub>3</sub> across their domains. In the Netherlands, a continuous wet annular denuder method (AMOR, replaced by the DOAS (Differential Optical Absorption Spectroscopy) device in 2015) has been used at 8 stations in the Dutch National Air Quality Monitoring Network (Van Pul et al., 2004; van Zanten et al., 2017). The Ammonia in Nature (MAN) network established in 2005 in the Netherlands monitors NH<sub>3</sub> with passive diffusion tubes in Natura 2000 areas (Lolkema et al., 2015). In the USA, the Ambient Ammonia Monitoring Network (AMoN) uses passive (Radiello) samplers at 50 sites since Oct 2010 (Puchalski et al., 2011). Hungary (Horvath et al., 2009), Belgium (den Bril et al., 2011), Switzerland (Thöni et al., 2004), West Africa (Senegal and Mali under the Pollution of African Capitals program; Adon et al., 2016) and China (Xu et al., 2016) also have long-term NH<sub>3</sub> measurements (see review by Bleeker et al., 2009).

In the UK, the National Ammonia Monitoring Network (NAMN) was established in September 1996 with the aim of establishing long-term continuous monthly measurements of atmospheric NH<sub>3</sub> gas (Sutton et al., 2001a). Particulate NH<sub>4</sub><sup>+</sup> measurements were added in 1999, since this was expected to exhibit different spatial patterns and temporal trends to gaseous NH<sub>3</sub> (Sutton et al., 2001b). The NAMN thus provides a unique and important long-term record for examining responses to changing agricultural practice and allows assessment of the compliance of NH<sub>3</sub> emissions with targets established by international policies on emissions abatement. Measurements of NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> in the NAMN also address spatial patterns, covering both source and sink areas to test performance of atmospheric transport models, to support estimation of dry deposition of NH<sub>x</sub>, to improve estimation of the UK NH<sub>x</sub> budget (Fowler et al., 1998; Smith et al., 2000; Sutton et al., 2001b) and to assist with the assessment of exceedance of critical loads and critical levels (UNECE, 2007).

This paper provides an analysis on the state of atmospheric concentrations of NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> in the UK from 1998 to 2014 and their spatial and temporal trends. Overall, 17 years of continuous long-term NH<sub>3</sub> measurement data and 16 years of continuous long-term NH<sub>4</sub><sup>+</sup> measurement data from the NAMN are analysed to assess trends in concentrations in relation to estimated changes in emissions. The long-term measurement dataset is also used to explore spatial and temporal patterns in NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> across the UK in relation to regional variability in emission source sectors.

## 2 Material and Methods

### 2.1 Network structure and site requirements

The design strategy for NAMN was to sample at a large number of sites (>70) using low-frequency (monthly) sampling for cost-efficient assessment of temporal patterns and long-term trends. The network covers a wide distribution of monitoring sites with measurements in both agricultural and semi-natural areas. Monitoring locations are sited away from point sources (> 150 m) such as farm buildings, which avoids overestimating NH<sub>3</sub> concentrations compared with the grid square, since the aim is to provide meso-scale and regional patterns. In addition, where sampling is carried out in woodland areas, it is made in clearings. It was also recognised that the location of the network sites needed to consider the extent of sub-grid variability and the representativeness of sampling points. Spatially detailed local-scale NH<sub>3</sub> monitoring was therefore also carried out at a sub-1 km level to assess the extent to which a monitoring location is representative (Tang et al., 2001b). The NAMN started with 70 sites. Over time, new sites were added to fill gaps in the map, some sites were closed following reviews and some sites had to be relocated due to local reasons, for example land ownership changes or site re-development. The number of sites peaked at 93 in 2000, but since 2009 has been stable at 85 sites. The locations of the NAMN sites for NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> in 2012 are shown in Figure 2a & b.

<INSERT FIGURE 1>

The selection of NAMN sites to provide a representative concentration field across the UK was aided by the availability of an estimated UK NH<sub>3</sub> concentration field at a 5 km by 5 km grid resolution provided by the Fine Resolution Atmospheric Multi-pollutant Exchange (FRAME) model (Singles et al., 1998; Fournier et al., 2002). A comparison of FRAME modelled NH<sub>3</sub> concentrations for NAMN sites with FRAME modelled concentrations for the whole of the UK shows that the network has a good representation in the middle air concentration classes of 0.5 – 1.5 µg m<sup>-3</sup> (33 % of NAMN sites, compared with 29 % of all FRAME 5 km x 5 km grid squares) and 1.5 - 3 µg m<sup>-3</sup> (32 % of NAMN sites, compared with 39 % of all FRAME 5km x 5 km grid squares) good representation in the middle air concentration classes of 3-4 µg m<sup>-3</sup>, but with an over-representation at high concentrations and under-representation at low concentrations (Figure 2c). Since air concentrations are more variable in high concentration areas, a larger number of monitoring sites were located in these areas than in remote low concentration areas where air concentrations are more homogeneous. Similarly, the monitoring sites were strategically selected to cover source areas of expected high concentrations and variability on the basis of the FRAME model NH<sub>3</sub> concentration estimates (Figure 2a & b), and this approach was expected to provide additional evidence to test the performance of atmospheric dispersion models (Fournier et al., 2005; Dore et al., 2015). When compared with other atmospheric chemistry transport models, FRAME was found to correlate well with measured NH<sub>3</sub> concentrations (Dore et al. 2015). The NAMN sites were also similarly checked for representativeness of particulate NH<sub>4</sub><sup>+</sup> by comparing FRAME modelled NH<sub>4</sub><sup>+</sup> concentrations at NAMN sites with modelled concentrations for the whole of the UK, which demonstrates a good representation across the range of expected concentrations (Figure 2d).

### 2.2 Atmospheric NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> measurements

Monthly time-integrated measurements of atmospheric NH<sub>3</sub> are made in the NAMN using a combination of passive samplers (Sutton et al., 2001a; Tang et al., 2001a) and an active diffusion denuder method referred to as the DENuder for Long Term Atmospheric (DELTA) sampler (Sutton et al., 2001a & c). In terms of passive samplers, membrane diffusion tubes (3.5 cm long) with a limit of detection (LOD) around 1 µg NH<sub>3</sub> m<sup>-3</sup> (Sutton et al., 2001a) were used in the first 4 years (September 1996 – April 2000). These were replaced in May 2000 with the more sensitive Adapted Low-cost, Passive High Absorption (ALPHA, LOD = 0.03 µg NH<sub>3</sub> m<sup>-3</sup>) diffusive samplers (Tang et al., 2001a; Tang and Sutton, 2003), following a period of parallel testing (Sutton et al., 2001c).

Particulate  $\text{NH}_4^+$  measurement was added to the NAMN in 1999 at all DELTA sites (50) in the first two years (1999 and 2000). Following this initial period, the sampling density was reduced during early 2001 to 37 sites and has been stable at 30 sites since 2006. Although not presented in this paper, the DELTA samplers additionally provide concentrations of acid gases (HNO<sub>3</sub>, SO<sub>2</sub>, HCl) and aerosols (NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, Cl<sup>-</sup>, Na<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>) for the UK Acid Gas and Aerosol monitoring network (AGANet) at a subset of NAMN DELTA sites (Tang et al., 2015; Conolly et al., 2016). Measurement data from the AGANet are used to aid interpretation of NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> results in Sect. 3.5.6.

### 2.2.1 DELTA method

The DELTA method uses a small pump to sample air (0.2 to 0.4 L min<sup>-1</sup>) in combination with a high-sensitivity gas meter to record sampled volume (Sutton et al., 2001c). Two citric acid coated denuders (10 cm long borosilicate glass tubes) in series are used to collect NH<sub>3</sub> gas and to check the collection efficiency. A collection efficiency correction is applied to the measurement (Sutton et al., 2001d). The corrected air concentration ( $\chi_a$  (corrected)) is determined as in Equation 1:

$$\chi_a \text{ (corrected)} = \chi_a \text{ (Denuder 1)} * \frac{1}{1 - \chi_a \left[ \frac{\chi_a \text{ (Denuder 2)}}{\chi_a \text{ (Denuder 1)}} \right]} \quad (1)$$

Typically, denuder collection efficiency is better than 90% (Conolly et al., 2016). At 90 % collection efficiency, the correction represents < 1 % of the corrected air concentration. Individual measurements with collection efficiency < 75 % (correction amounts to 11 % of the total at 75%) are flagged as valid, but less certain (Tang and Sutton, 2003). Where less than 60 % of the total capture is recorded in the first denuder, the correction factor amounts to greater than 50 % and is not applied. The air concentration of ( $\chi_a$ ) of NH<sub>3</sub> is then determined as the sum of NH<sub>3</sub> in denuders 1 and 2 (Equation 2):

$$\chi_a = \chi_a \text{ (Denuder 1)} + \chi_a \text{ (Denuder 2)} \quad (2)$$

At sites where particulate NH<sub>4</sub><sup>+</sup> is also sampled, a 25 mm filter pack with a citric acid impregnated cellulose filter is added after the denuders to capture the NH<sub>4</sub><sup>+</sup>. The calculated air concentrations ( $Y_a$ ) of NH<sub>4</sub><sup>+</sup> is corrected for incomplete capture of NH<sub>3</sub> by the double denuder. The corrected air concentrations ( $Y_a$  (corrected)) of NH<sub>4</sub><sup>+</sup> is determined as in Equation 3:

$$Y_a \text{ (corrected NH}_4^+) = Y_a \text{ (NH}_4^+) - [(\chi_a \text{ (corrected NH}_3) - (\chi_a \text{ (Denuder 1 NH}_3) + \chi_a \text{ (Denuder 2 NH}_3)))] * (18/17) \quad (3)$$

For NH<sub>4</sub><sup>+</sup> sampling, loss of NH<sub>3</sub> due to volatilisation of NH<sub>4</sub><sup>+</sup> from the acid impregnated filter has been investigated, by adding a third citric acid coated denuder after the filter pack which was found to be negligible. At DELTA sites where additional simultaneous sampling of acid gases and particulate phase components are made for AGANet, ion balance checks between anions and cations in the particulate phase are performed to provide an indication of the quality of the particulate measurements. For the acid and base particulate components, close coupling is expected between NH<sub>4</sub><sup>+</sup> and the sum of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>, as NH<sub>3</sub> is neutralised by HNO<sub>3</sub> and H<sub>2</sub>SO<sub>4</sub> to form NH<sub>4</sub>NO<sub>3</sub> and (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, respectively (Conolly et al., 2016).

At the Bush OTC site in Scotland (UK-AIR ID = UKA00128), duplicate DELTA measurements are made to assess the reproducibility of the method. For continuous monthly measurements between 1999 and 2014, the R<sup>2</sup> between the duplicate systems was 0.96 for both NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> (supp. Figure S1).

### 2.2.2 Passive methods

The NH<sub>3</sub> membrane diffusion tubes deployed in the NAMN from 1996 to 2000 are hollow cylindrical tubes (FEP, 3.5 cm long). A cap at the top end holds in place two stainless steel grids coated with sulphuric acid. The lower air-inlet end of the tube is capped with a gas-permeable membrane (Sutton et al., 2001a; Tang et al., 2001a; Thijssse, 1996). In comparison, the ALPHA passive sampler is a badge-type high sensitivity sampler with an uptake rate that is ~20 times faster than the diffusion tube. It consists of a cylindrical low-density polyethylene body. An internal ridge supports a cellulose filter coated with citric acid, which is held in place with a polyethylene ring. The open end is capped with a PTFE membrane, providing a diffusion path length of 6 mm between the membrane and absorbent surface (Tang et al., 2001a).

10 Triplicate passive samplers are deployed for every measurement in the NAMN. Where the % coefficient of variation (CV) of the triplicate samplers is greater than 30% for the diffusion tubes or greater than 15% for the ALPHA samplers, the sample run is classed as failing the quality control test. Large discrepancies are most likely due to contamination of samples and data from contaminated samples are excluded from the assessment in this paper.

15 The passive methods are calibrated against the DELTA method in the NAMN by ongoing comparison at several sites representing a wide range of ambient NH<sub>3</sub> concentrations (see Sect. 2.2.4). Since 2009, the number of inter-comparison sites has been nine. These are Auchencorth (UKA00451), Bush OTC (UKA00128), Glensaugh (UKA00348), Lagganlia (UKA00290), Llyncllys Common (UKA00270), Moorhouse (UKA00357), Rothamsted (UKA00275), Sourhope (UKA00347) and Stoke Ferry (UKA00317). The inter-comparison is used to establish a regression between the active and passive methods, with the DELTA samplers as the reference system, since the air volume sampled is accurately measured with high sensitivity gas meters. The calibration is necessary to account for the fact that the sampling path length in the passive samplers is longer than the distance between the membrane and adsorbent, due to the additional resistance to molecular diffusion imposed by the turbulence damping membrane at the inlet and the presence of a laminar boundary layer of air on the outside of the sampler (Tang et al., 2001a). In addition, parallel measurements were made at a high NH<sub>3</sub> concentration farm site (1998-2007) to extend the calibration range, and to ascertain linearity of response to high concentrations. To ensure that no bias is introduced in the sampling and to maintain the validity of long-term trends, the calibration is evaluated on an annual basis (Tang and Sutton, 2003; Conolly et al., 2016).

For the period up to 2000 when the diffusion tubes were implemented in the NAMN, their calibration (at 10 µg m<sup>-3</sup>) amounts to an average of 1.5 % compared with the DELTA system. The mean ALPHA sampler calibration (at 10 µg m<sup>-3</sup>), compared with the DELTA system, amounts to a correction of 10 % (ALP1: prototype 1, 1998-2000), 15 % (ALP2: injection mould 1, 2001-2005), 17 % (ALP3: injection mould 2, 2006), 34 % (ALP4: injection mould 2 + new membrane, 2007-2008) and 40 % (ALP5: injection mould 2 + new membrane + new lab/instrument FloRRia, 2010-2014), respectively. The new PTFE membrane (5 µm pore size) is supported on a regular polypropylene grid and is thicker (305 µm) than the earlier PTFE membrane (also 5 µm pore size, but 265 µm thickness) used which was supported instead on a randomly arranged polypropylene support material. The difference in calibration was therefore due to the extra resistance to gas diffusion imposed by the new thicker membrane. The annual calibration of the methods shows both high precision and constancy between years (Figure 3), which is important to support the detection of temporal trends in NH<sub>3</sub> concentrations. There is no systematic trend over time in either of the passive method calibrations.

<INSERT FIGURE 2>

The comparison of monthly measurement data between the DELTA and calibrated passive measurements demonstrated a close agreement (Figure 4). The correlation ( $R^2$ ) between DELTA and calibrated diffusion tubes was 0.91 (Figure 4a), while the correlation between DELTA and calibrated ALPHA samplers was 0.92 (Figure 4b). From the calibrated results, the intercept for the diffusion tubes was  $0.10 \mu\text{g NH}_3 \text{ m}^{-3}$ , while that for the ALPHA samplers was  $0.03 \mu\text{g NH}_3 \text{ m}^{-3}$ , demonstrating the improvement in sensitivity with the ALPHA samplers compared with the diffusion tubes (Figure 4). In the present case the value of the intercepts, even for diffusion tubes, is much less than typical  $\text{NH}_3$  air concentrations (see Sect. 3). However, this cannot be assumed to be the case in other implementations of the same methods. Experience from other studies using the lower sensitivity diffusion tubes indicates a tendency to overestimate  $\text{NH}_3$  concentrations under clean conditions (RGAR, 1990; Thijsse et al., 1996; Tang et al., 2001a; Lolkema et al., 2015). This observation points to the need for any application of  $\text{NH}_3$  passive sampling for ambient monitoring to be accompanied by testing and calibration against a verified active sampling method. In independent assessments, for example in the USA (Puchalski et al., 2011), the ALPHA samplers performed well against a reference annular denuder method with a median relative percent difference of  $-2.4\%$ .

<INSERT FIGURE 3>

### 2.2.3 Chemical analysis

$\text{NH}_3$  gas captured on the acid coating of the denuder (DELTA), grid (diffusion tubes) or filter paper (ALPHA), and particulate  $\text{NH}_4^+$  captured on the DELTA aerosol filter, are extracted into deionised water and analysed for  $\text{NH}_4^+$  on an ammonia flow injection analysis system. The analytical instrument has changed over the network's operational period from the AMFIA (ECN, NL) to the FloRRIA (Mechatronics, NL), an updated model based on AMFIA (Conolly et al., 2016). The principles of operation of both instruments are the same and are based on selective diffusion of  $\text{NH}_4^+$  across a PTFE membrane at c. pH 13 into a counter-flow of deionized water, allowing selective detection of  $\text{NH}_4^+$  by conductivity (Wyers et al., 1993). The extracted samples were analysed for  $\text{NH}_4^+$  against a series of  $\text{NH}_4^+$  standards and quality controls. Parallel analysis of laboratory and field blank (unexposed) samples were used to determine the amounts of  $\text{NH}_4^+$  derived from  $\text{NH}_3$  and  $\text{NH}_4^+$  in the atmosphere during transport and storage. The limit of detection (LOD) calculation of the ALPHA and DELTA methodologies are determined as three times the standard deviations of the laboratory blanks. For the DELTA method, the LODs were  $0.01 \mu\text{g m}^{-3}$  for gaseous  $\text{NH}_3$  and  $0.02 \mu\text{g m}^{-3}$  for particulate  $\text{NH}_4^+$ . For the ALPHA method, the LOD was determined as  $0.03 \mu\text{g m}^{-3}$ .

### 2.2.4 Data Quality Control

Measurement data are checked and screened, based on the quality management system applied in the UK air monitoring networks (Tang and Sutton, 2003). Data quality is assessed against the following set quality control criteria: a) DELTA system: monitoring of the air flow rate and the use of two denuders in every sample to assess capture efficiency for  $\text{NH}_3$ , and b) passive samplers: use of triplicate samplers for monitoring  $\text{NH}_3$  concentrations at every site, to allow an assessment of sampling precision, and c) ongoing calibration of passive samplers against the DELTA. Data flags are applied to the dataset; a full list of these is available from the EMEP website (<http://www.nilu.no/projects/ccc/flags/index.html>). Following the quality control checks and data flagging on the collected dataset, the annually ratified data from the NAMN are made publically available on the Department for Environment, Food & Rural Affairs (Defra) UK-AIR website (<https://uk-air.defra.gov.uk/>) and are also in the process of being made available on the EMEP website (<http://ebas.nilu.no/>).

An intercomparison of  $\text{NH}_3$  measurements by the RIVM AMOR system (hourly, Wyers et al., 1993) and the DELTA sampling system (monthly) have been carried out at the Zegveld site (ID 633) in the Dutch National Air Quality Monitoring Network (van Zanten et al., 2017) since July 2003. Since September 2012, ALPHA measurements have also been included. To compare

results, monthly mean concentrations were derived from the average of hourly AMOR data for the corresponding DELTA and ALPHA monthly sampling periods with good agreement (supp. Figure S6).

### 2.2.5 Trend Analyses

5 Trend analyses were carried out using (i) linear regression (LR), (ii) Mann-Kendall (MK) test (Gilbert, 1987) on annually averaged and monthly mean data, and (iii) Seasonal Mann-Kendall (SMK) test (Hirsch et al., 1982) on monthly data only. Mann-Kendall tests were performed using the 'Kendall' package (McLeod, 2015) in the R software. Computation of the Sen's slope and confidence interval (for non-seasonal Sen's slope only) of the linear trend were performed using the R 'Trend' package (Pohlert, 2016). The SMK test (Hirsch et al., 1982) takes into account a 12 month seasonality in the time series data  
10 by computing the MK test on each of monthly 'seasons' separately, and then combining the results. So for monthly 'seasons', January data are compared only with January, February only with February, etc. No comparisons are made across season boundaries.

The Sen's slope is the fitted median slope of a linear regression joining all pairs of observations. For the SMK, an estimate of  
15 the seasonal Sen's trend slope over time is computed as the median of all slopes between data pairs within the same season (i.e. January compared only with January etc.). Therefore no cross-season slopes contribute to the overall estimate of the SMK trend slope. The main advantages, as discussed in the literature of the MK approach over linear regression for trend assessments are that (i) it does not require normally distributed data, (ii) it is not affected by outliers, and (iii) it removes the effect of temporal auto-correlation in the data. The MK approach are widely used in environmental time series assessments, e.g. long-term trends in precipitation (Serrano et al. 1999) and long-term trends in European air quality (EMEP Colette et al., 2016; Torseth et al., 2012). However linear trend assessment have been used in UK air quality monitoring network reports (e.g. Conolly et al., 2016), therefore both approaches were used in this paper primarily as a quality assurance check.

## 3 Results and discussion

25 In order to summarise and discuss the NAMN dataset, the spatial patterns in the measurements of  $\text{NH}_3$  and  $\text{NH}_4^+$  are considered in Sect. 3.1 (comparison with emission estimates) and Sect. 3.2 (comparison with modelled concentration estimates), seasonal patterns are discussed in Sect. 3.3, and long-term trends across the UK in Sect. 3.4.

### 3.1 Spatial variability in $\text{NH}_3$ and $\text{NH}_4^+$ concentrations in relation to estimated emissions

30 As a primary pollutant emitted from ground-level sources,  $\text{NH}_3$  exhibits high spatial variability in concentrations (Sutton et al., 2001b; Hellsten et al., 2008; Vogt et al., 2013), confirmed by  $\text{NH}_3$  data from the NAMN (e.g. range of 0.06 – 8.8  $\mu\text{g m}^{-3}$  annual mean in 2005) (Figure 5a). The observed variability is consistent with the large regional variability in  $\text{NH}_3$  emissions and sources (Figure 5c & d). With agriculture being the main source of  $\text{NH}_3$  emissions, Figure 5a shows the largest concentrations of measured  $\text{NH}_3$  in parts of the UK with the highest livestock emissions, such as eastern England (East Anglia), north-west England (Eden Valley, Cumbria) and the border area between England and Wales (Shropshire) (Figure 5d). By  
35 contrast, the lowest  $\text{NH}_3$  measured concentrations are found in the north-west Scottish Highlands ( $< 0.2 \mu\text{g m}^{-3}$ ), which is consistent with the emissions map (Figure 5c). The 2005 data show exceedance of the Critical Levels for annual mean  $\text{NH}_3$  concentrations of 1 and 3  $\mu\text{g NH}_3 \text{ m}^{-3}$  for the protection of lichens-bryophytes and vegetation, respectively (UNECE, 2007) at many of the sites (53 %  $> 1 \mu\text{g NH}_3 \text{ m}^{-3}$  and 13 %  $> 3 \mu\text{g NH}_3 \text{ m}^{-3}$ ). In 2014, exceedance of the 1 and 3  $\mu\text{g NH}_3 \text{ m}^{-3}$  CLe increased to 60 % and 16 %, respectively. The widespread exceedance of the CLe for  $\text{NH}_3$  concentrations across the UK thus



represents an ongoing threat to the integrity of sites designated under the Habitats Directive, as well as nationally designated Sites of Special Scientific Interest (SSSI) and other sensitive habitats.

Concentrations of  $\text{NH}_4^+$  are less spatially heterogeneous than those of  $\text{NH}_3$ , based on data from 30 sites (e.g. range of 0.14 to 1.8  $\mu\text{g m}^{-3}$  annual mean in 2005) with a more coherent pattern of variation across the country, reflecting regional differences in  $\text{NH}_3$  concentrations (Figure 5b). Thus there is a general decreasing gradient from the south-east to the north-west of the UK, due to both  $\text{NH}_3$  sources in England and import of particulate matter from Europe (Vieno et al., 2014; Dore et al., 2015). The limited variation across the UK for the annual average  $\text{NH}_4^+$  concentrations can be attributed to the atmospheric formation process (providing a diffuse source) and its longer atmospheric lifetime.

**<INSERT FIGURE 4>**

A similar picture is reported by the Dutch National Air Quality Monitoring Network (van Zanten et al., 2017), with large spatial variability of  $\text{NH}_3$  concentrations (2 – 20  $\mu\text{g NH}_3 \text{ m}^{-3}$ ) across the country and a more homogeneous distribution of particulate  $\text{NH}_4^+$  (1-2  $\mu\text{g NH}_4^+ \text{ m}^{-3}$  in 2014), although the number of Dutch monitoring sites reported there is much smaller with only 8 stations providing continuous measurements. Both  $\text{NH}_3$  and  $\text{NH}_4^+$  concentrations were correlated with emission density, but the correlation was smaller for  $\text{NH}_4^+$  than for  $\text{NH}_3$  because of the larger contribution to  $\text{NH}_4^+$  concentrations from long-range transport in the Netherlands.

The UK  $\text{NH}_3$  emissions inventory is calculated and spatially distributed annually. Agricultural sources at a 5 km by 5 km grid resolution are combined with a large number of non-agricultural sources (Sutton et al., 2000; Tsagatakis et al., 2016) at a 1 or 5 km resolution to produce the annual  $\text{NH}_3$  emissions data, and maps at a 1 km by 1 km grid resolution are reported by the official UK National Atmospheric Emissions Inventory (NAEI; <http://naei.defra.gov.uk/data/mapping>). In the UK, agriculture accounts for > 80% of total  $\text{NH}_3$  emissions and is estimated by the National Ammonia Reduction Strategy Evaluation System (NARSES) model (Webb & Misselbrook 2004; Misselbrook et al., 2015). For the agricultural  $\text{NH}_3$  emission maps, parish statistics on livestock numbers and crop areas are combined with satellite-based land cover data to model emissions at a 1 km resolution, using the AENEID model (Dragosits et al., 1998; Hellsten et al., 2007). For reasons of data confidentiality, the 1 km data need to be aggregated to produce annual agricultural  $\text{NH}_3$  emissions maps at a 5 km by 5 km grid resolution. National emission estimates for  $\text{NH}_3$  are submitted to both the European Commission under the NECD (2001/81/EC) and the United Nations Economic Commission for Europe (UN/ECE) under the Convention on Long-Range Transboundary Air Pollution (CLRTAP).

The AENEID approach (Dragosits et al. 1998) can further be used to classify each 5 km by 5 km grid square in the UK into dominant  $\text{NH}_3$  emission source categories (Figure 5d), following the method of Hellsten et al. (2008), where grid squares with >45% from a given category are referred to as dominated by that source. The seven categories are: cattle, pigs & poultry (combined for data disclosivity reasons), sheep, fertilizer application to crops and grassland, non-agricultural sources, as well as a mixed category where no single source dominates, and background. Background grid squares are defined by very low  $\text{NH}_3$  emissions of <1  $\text{kg N ha}^{-1} \text{ y}^{-1}$ .

Using the dominant emission sources map, each site in the NAMN is classified to one of the seven categories just described. This provides information of the main emission source type expected in the 5 km by 5 km grid square containing the monitoring site and is useful for assessing whether the network has a good representation of key emission source categories (Supp. Figure

S2a & b). Over the period since the NAMN was established, from 1996 to present, there have been substantial changes in emissions estimated for the different source sectors. For analysis in this paper, the dominant sources map for 2005 emission year was used as representing the mid-point of the data series (1998-2014) and compared with the classification from other years for consistency. This categorization of sites is used further in the interpretation of the monitored  $\text{NH}_3$  and  $\text{NH}_4^+$  concentrations and their long-term trends in the next sections.

<INSERT FIGURE 5>

### 3.2 Spatial variability in $\text{NH}_3$ and $\text{NH}_4^+$ concentrations in relation to modelled concentrations

Comparison of measurements with modelled  $\text{NH}_3$  concentrations from the FRAME model for an example year of 2012 showed significant scatter when considering the full network of sites ( $n = 85$ ,  $R^2 = 0.62$ ) (Figure 6a). In this graph, each point is colour-coded according to the estimated dominant  $\text{NH}_3$  emission source category for the 5 km by 5 km grid square. This updates a similar comparison from Sutton et al. (2001b) for the year 2000. The scatter may be explained by the large local spatial variability of  $\text{NH}_3$ , related primarily to rapid decreases of  $\text{NH}_3$  concentrations with distance from a source (see e.g. Pitcairn et al., 1998; Dragosits et al., 2002), with the result that a single site measurement only gives an approximate indication of concentrations across the model grid square it is located in. At many of the sites where the model overestimates concentrations, the measurements are in fact carried out in nature reserves, or in clearings inside forests. The monitoring sites in these sink areas are typically well away from local sources. Conversely, some of the outliers where measurements are larger than the model predictions show indications of being affected by nearby emission sources, as was established by investigations during site visits.

<INSERT FIGURE 6>

Figure 7 considers measured  $\text{NH}_3$  concentrations at a subset of sites (44 out of the full 85 sites) that are located away from nearby local sources, in forest or semi-natural areas, following the site classification and assessment by Hallsworth et al. (2010). For this restricted set of sites,  $R^2 = 0.76$  for 2012 which is higher than the correlation for the overall UK network. The improvement in correlation between measured and modelled  $\text{NH}_3$  concentrations for this subset of sites can be explained by the monitoring locations typically being further away from sources, so that uncertainties in local emissions estimates are to some extent averaged out. This observation is also consistent with the findings of Vieno et al. (2009).

In contrast to  $\text{NH}_3$ , the correlation between NAMN measurements and FRAME model output is stronger for particulate  $\text{NH}_4^+$  concentrations ( $R^2 = 0.87$ ). However, measured concentrations are generally larger than the modelled ones (slope 1.1, intercept  $-0.16 \mu\text{g m}^{-3}$ ) (Figure 6b). One reason for the better agreement for  $\text{NH}_4^+$  is the more slowly changing spatial patterns in concentrations, which are not expected to vary on a finer scale than the model's 5 km by km grid, improving the representativeness of site-based measurements. The 2012 comparison shown here updates an earlier inter-comparison assessment carried out by Dore et al. (2007) for the year 2002 and demonstrates that the FRAME model is performing well in describing the spatial distribution of  $\text{NH}_4^+$ . However, for the 2012 inter-comparison, the FRAME model appears to underestimate  $\text{NH}_4^+$  at sites with concentrations  $< 0.6 \mu\text{g NH}_4^+ \text{ m}^{-3}$ , with better agreement at concentrations above  $0.6 \mu\text{g NH}_4^+ \text{ m}^{-3}$ . This suggests either too low a formation rate for  $\text{NH}_4^+$  in the model at cleaner sites, or too high a removal rate for  $\text{NH}_4^+$ , or a combination of both. The presence of higher measured  $\text{NH}_4^+$  concentrations in remote areas than shown by the model may also indicate that  $\text{NH}_4^+$  has a longer residence time than treated in the model. Similar regressions between NAMN and FRAME  $\text{NH}_4^+$  aerosol concentrations were observed for other years. For example, for 2008 the FRAME model underestimated  $\text{NH}_4^+$

at concentrations  $< 0.7 \mu\text{g NH}_4^+ \text{ m}^{-3}$  (slope 1.2, intercept  $-0.26 \mu\text{g}^{-3}$ ;  $R^2 = 0.89$ , range =  $0.2 - 1.4 \mu\text{g m}^{-3}$ ). Changes in the chemical climate, such as reduced emissions of  $\text{SO}_2$  in the UK, are postulated to affect conversion rates of  $\text{NH}_3$  into  $\text{NH}_4^+$ , as well as the dry deposition rates, leading to more  $\text{NH}_3$  remaining in the atmosphere (van Zanten et al., 2017). This is discussed further in Sect. 3.5.6.

### 5 3.3 Seasonal variability in measured UK $\text{NH}_3$ and $\text{NH}_4^+$ concentrations

A comprehensive account of the seasonal variability of  $\text{NH}_3$  and  $\text{NH}_4^+$  for different regions across the UK is provided by the NAMN. In Figure 8, the average seasonal cycles of grouped sites from four different emission source categories are compared for  $\text{NH}_3$  and  $\text{NH}_4^+$ .

#### 10 <INSERT FIGURE 7>

In addition to substantial differences in the overall magnitude of  $\text{NH}_3$  concentrations, where the largest concentrations in the network are found at sites dominated by pig and poultry farming, followed by areas where cattle farming predominates, it is clear that the seasonal patterns of  $\text{NH}_3$  also vary depending on the dominant source type (Figure 8a). For background sites  
15 (defined as located in grid squares with  $\text{NH}_3$  emissions  $< 1 \text{ kg N ha}^{-1} \text{ y}^{-1}$ ), a clear summer maximum in  $\text{NH}_3$  concentrations can be observed, with minimum concentrations occurring in winter. The summer peak is probably related to increased land surface  $\text{NH}_3$  emissions in warm, dry summer conditions, both from the presence of low-density grazing livestock and wildlife. It is also related to surface factors such as the compensation point for vegetation, which is defined as the concentration below which growing plants start to emit  $\text{NH}_3$  into the atmosphere (Sutton et al., 1995). The interaction between atmospheric  $\text{NH}_3$   
20 concentrations and vegetation is complex, leading to both emission and deposition fluxes, depending on relative differences in concentrations. However, it is well established that warm, dry conditions promote  $\text{NH}_3$  emission from vegetation (e.g. Massad et al., 2010; Flechard et al., 2013). It is therefore possible that bi-directional exchange with vegetation is at least partly controlling  $\text{NH}_3$  concentrations at remote sites distant from intensive livestock farming.

25 The possibility for such interactions can be considered further using the example of Inverpolly (UKA00457), a remote background site in the NW Scottish Highlands. This site shows a very clear seasonal cycle with peak concentrations in July when warmer, drier conditions prevail, while lowest concentrations occur during the cooler and wetter winter months (Figure 9a & b). A smaller peak in  $\text{NH}_3$  can also be seen annually in April, which indicates potential longer range influences of manure spreading in spring, even at this remote location (Figure 9b). Although there is substantial scatter, Figure 10 shows that there  
30 is significant correlation between monthly  $\text{NH}_3$  concentrations and both temperature ( $R^2 = 0.33$ ,  $n = 231$ ,  $p < 0.05$ ) and precipitation ( $R^2 = 0.19$ ,  $n = 231$ ,  $p < 0.05$ ). The influence of temperature and rainfall on  $\text{NH}_3$  emission and concentrations is well characterised (e.g. see Sutton et al., 2013; van Zanten et al., 2017).

#### <INSERT FIGURE 8>

35

#### <INSERT FIGURE 9>

For sites dominated by emissions from sheep farming, the seasonal profile in  $\text{NH}_3$  concentrations is similar to that for background sites, although the summer maximum in  $\text{NH}_3$  is larger than background sites, because grazing emissions are larger  
40 (Hellsten et al., 2008). It is notable that the peak  $\text{NH}_3$  concentration occurs later in the year for background areas (July-September) than for sheep areas (June-August). This may be related to the seasonal presence of lambs, which are often only present for the first part of the summer. In areas with more intensive livestock farming, where emissions comes from either

cattle or from pig & poultry farming, the largest concentrations are observed in spring and autumn, corresponding to periods of manure application to land. The spring peak in March is larger than the autumn peak in September, which coincides with the main period for manure application being in spring, before the sowing of arable crops or early on in the grass-growing period (Hellsten et al., 2007). Ammonia concentrations in these areas are also larger in summer than winter, due to warmer conditions promoting volatilization. Interestingly, the dip in concentrations in June matches a period when crops will be actively growing with possible uptake and removal of  $\text{NH}_3$  from the atmosphere.

For particulate  $\text{NH}_4^+$ , as expected for a secondary pollutant, concentrations are more decoupled from the dominant  $\text{NH}_3$  source sectors in the vicinity of a site. The seasonal trends in  $\text{NH}_4^+$  are broadly similar for the four source sectors shown in Figure 7b, with the magnitude of the  $\text{NH}_4^+$  concentrations reflecting  $\text{NH}_3$  concentrations at a regional level. Here the spring peak is more driven by long-range transboundary transport, e.g. influence from continental Europe (Vieno et al., 2014). Nevertheless, it is notable that the winter minima for  $\text{NH}_4^+$  aerosol concentrations at sheep and background sites are more pronounced than that for pig, poultry and cattle dominated sites. This may be a result of a combination of smaller  $\text{NH}_3$  emissions in winter in these areas (as indicated by Figure 7a) and differences in long-range transport to the more remote areas in winter conditions.

For particulate  $\text{NH}_4^+$ , as expected for a secondary pollutant, concentrations are more decoupled from the dominant  $\text{NH}_3$  source sectors in the vicinity of a site. Although the formation of particulate  $\text{NH}_4^+$  primarily depends on the occurrence of  $\text{NH}_3$  in the atmosphere, synoptic meteorology and long range transboundary transport from continental Europe are important drivers influencing the seasonal variations of  $\text{NH}_4^+$  across the UK, due to its' longer lifetime. The seasonal trends in particulate  $\text{NH}_4^+$  are seen to be broadly similar for the four different emission source sectors (Figure 8b), with the magnitude of the  $\text{NH}_4^+$  concentrations reflecting  $\text{NH}_3$  concentrations at a regional level. In the atmosphere, particulate  $\text{NH}_4^+$  are primarily in the form of  $(\text{NH}_4)_2\text{SO}_4$  and  $\text{NH}_4\text{NO}_3$ , formed when the acid gases  $\text{HNO}_3$  and  $\text{H}_2\text{SO}_4$  in the atmosphere are neutralised by  $\text{NH}_3$  (Putaud et al., 2010).  $\text{NH}_3$  preferentially neutralizes  $\text{H}_2\text{SO}_4$  due to its low saturation vapour pressure (forming  $\text{NH}_4\text{HSO}_4$  then  $(\text{NH}_4)_2\text{SO}_4$ ), while  $\text{NH}_4\text{NO}_3$  is formed when abundant  $\text{NH}_3$  is available. In contrast to  $(\text{NH}_4)_2\text{SO}_4$ ,  $\text{NH}_4\text{NO}_3$  is a semi-volatile component (Stelson & Seinfeld, 1982). Long-term data from the UK Acid Gas and Aerosol Network (AGANet, Conolly et al., 2016) shows a change in the particulate phase of  $\text{NH}_4^+$  from  $(\text{NH}_4)_2\text{SO}_4$  to  $\text{NH}_4\text{NO}_3$  with particulate nitrate concentrations exceeding that of particulate sulphate approximately three-fold (on a molar basis) (Fig. 18a). This suggests that the thermodynamic equilibrium between the gas phase  $\text{NH}_3$  and  $\text{HNO}_3$  and the aerosol phase  $\text{NH}_4\text{NO}_3$  will have a much greater effect on the seasonal concentrations of  $\text{NH}_4^+$  than  $(\text{NH}_4)_2\text{SO}_4$ . The formation and dissociation of  $\text{NH}_4\text{NO}_3$  depend strongly on ambient temperature and humidity (Stelson and Seinfeld, 1982). Warm, dry weather in summer promotes dissociation, decreasing particulate phase  $\text{NH}_4\text{NO}_3$  relative to gas phase  $\text{NH}_3$  and  $\text{HNO}_3$ . During the winter months, low temperature and high humidity favour the formation of  $\text{NH}_4\text{NO}_3$  from the gas phase  $\text{NH}_3$  and  $\text{HNO}_3$ . By contrast, the spring peak in  $\text{NH}_4^+$  concentrations may be attributed to photochemical processes (elevated ozone) leading to enhanced formation of  $\text{HNO}_3$  during this period (Pope et al., 2016) and also to import of particulate  $\text{NO}_3^-$  through long-range transboundary transport, e.g. from continental Europe, as discussed in Vieno et al. (2014). Nevertheless, it is notable that the winter minima for  $\text{NH}_4^+$  aerosol concentrations at sheep and background sites are more pronounced than that for pig, poultry and cattle dominated sites. This may be a result of a combination of smaller  $\text{NH}_3$  emissions in winter in these areas (as indicated by Figure 8a) and differences in long-range transport to the more remote areas in winter conditions.

Overall, the seasonal distributions show that  $\text{NH}_3$  concentrations are mostly governed by local emission sources and by changes in environmental conditions, with warm, dry weather favouring increased volatilisation. By contrast, particulate  $\text{NH}_4^+$  concentrations are largely determined by more distant sources through long-range transport and synoptic meteorology.

### 3.4 Long-term trends in estimated UK NH<sub>3</sub> emissions

UK NH<sub>3</sub> emissions are estimated to have fallen by 16 % between 1998 and 2014, from 336 to 281 kt (Figure 11a) (<http://naei.defra.gov.uk/>). The most significant cause of the estimated reductions has been decreasing cattle, pig and poultry numbers in the UK over this period. Between 2013 and 2014, the decreasing trend in UK NH<sub>3</sub> emissions was however reversed with an increase of 3.3 % from 272 to 281 kt NH<sub>3</sub> due to an increase in emissions from the agricultural sector from 224 kt in 2013 to 234 kt in 2014. This is attributed to an increase in dairy cow numbers (and dairy cow N excretion) and increase in fertiliser N use (particularly urea, which is associated with a higher emission factor than other fertilisers types used in the UK) (Misselbrook et al. 2015; <http://naei.defra.gov.uk/>).

#### 10 <INSERT FIGURE 10>

Although the UK met the 2010 emission ceilings target of 297 kt NH<sub>3</sub> emission per year set out under the Gothenburg Protocol and NEC Directive, it is committed to a further emission reduction by 2020 of 8 % from the 2005 total under the 2012 revised Gothenburg Protocol, and by 17% after 2030 under the revised 2016 NEC Directive (EU, 2016). The revised 2020 target of 282 kt NH<sub>3</sub> (8% reduction of the baseline figure of 307 kt NH<sub>3</sub> emissions total in 2005) may require emission strategies to be implemented, rather than relying on decreasing livestock populations as during the recent decades.

Agricultural emissions are by far the largest NH<sub>3</sub> sources in the UK's emission inventory, accounting for 86 % and 83 % of the total NH<sub>3</sub> emissions in 1998 and 2014, respectively. The primary source of agricultural emissions is livestock manure management, in particular from cattle which make up approximately 46 % of the total agricultural emissions, followed by pigs and poultry contributing another 18 % in 2014 (Defra, 2015; Misselbrook et al., 2015) (Figure 11b). Over the period 1998 to 2014, NH<sub>3</sub> emissions from cattle are estimated to have decreased by 11 % (from 144 to 128 kt), with emissions estimated to have remained relatively stable since 2008, followed by a modest 2 % increase between 2013 and 2014 from 125 kt to 128 kt (Figure 11a; Figure 17). Emissions from pigs and poultry showed a large downward trend between 1998 and 2014, with a decrease of 39 % (from 82.7 kt to 50.3kt) (Figure 11a; Figure 17), although the decreasing trend was reversed between 2012 and 2014, with an increase of 6 % from 46.7 kt to 50.3 kt. The sheep sector is a minor source, contributing 3.6 % to the total agricultural emissions. NH<sub>3</sub> emissions from this sector are estimated to have decreased by 24 % in 2014 relative to 1998 (from 13.3 to 10.1 kt).

### 30 3.5 Long-term trends in measured NH<sub>3</sub> concentrations

The UK NAMN dataset was analysed to compare levels and trends against the NH<sub>3</sub> emission inventory. To avoid bias due to changes in the number and locations of sites over the duration of the network, sites with incomplete data runs over selected periods for analysis are excluded. Based on these exclusion criteria, the number of sites with complete data runs was 59 for the period 1998 to 2014, 66 sites for 1999 to 2014, and 75 sites for the period 2000 to 2014. To ensure consistency in the trend analysis, several combinations of the available data were used:

- 1a. 1998 – 2014 (59 sites): annually averaged data
- 1b. 1998 – 2014 (59 sites): monthly mean data
- 2a. 1999 - 2014 (66 sites): annually averaged data
- 2b. 1999 - 2014 (66 sites): monthly mean data
- 3a. 2000 - 2014 (75 sites): annually averaged data
- 3b. 2000 – 2014 (75 sites): monthly mean data

A visualization of the time series according to dataset 1a is summarized in Figure 12. This shows the mean UK monitored annual NH<sub>3</sub> concentrations of 59 sites with complete data runs from 1998 (first complete year of monitoring) to 2014, summarised in a boxplot, together with annual mean UK rainfall and temperature data and compared with NH<sub>3</sub> emissions trends over the same period. The interquartile ranges and the spread of the NH<sub>3</sub> concentrations can be seen to be variable from year to year, demonstrating both substantial inter- and intra-annual variability.

<INSERT FIGURE 11>

### 3.5.1 Mann-Kendall non-parametric time series analysis

To detect trends and to indicate the significance level of the trends in the long-term NAMN data, the non-parametric Mann-Kendall (MK) approach was used combined with the Sen's slope method for estimating the trend and confidence interval of the linear trend (see Sect. 2.2.5). The classic MK test was used on the annually averaged data (datasets 1b, 2b, 3b), while both the classic MK and seasonal Mann-Kendall (SMK) tests were applied to the monthly averaged data (datasets 1a, 2a, 3a).

Results of the Mann-Kendall tests are summarised in Table 1. For each time series, the median annual trend (in units of  $\mu\text{g NH}_3^{-1}\text{y}^{-1}$ ) is estimated from the Sen's slope and intercept of the MK linear trend. To assess the relative change over time, the % relative median change was calculated from the estimated NH<sub>3</sub> concentration at the start ( $y_0$ ) and at the end ( $y_i$ ) of the selected time period ( $100 * [(y_i - y_0) / y_0]$ ) computed from the Sen's slope and intercept. This approach was adopted instead of a direct comparison of actual observed NH<sub>3</sub> concentrations at the start ( $y_0$ ) and at the end ( $y_i$ ) of the time series, since there is substantial inter-annual variability in the data (Figure 11, Figure 17). Using the estimated concentrations at the start and end from the fitted Sen's slope allows using a reference that is less sensitive to inter-annual variability than the actual observed concentrations.

<INSERT TABLE 1>

For the annually averaged NH<sub>3</sub> concentrations across the UK, dataset 1a (1998-2014, 59 sites) show a small, but non-significant decreasing trend (relative median change = -6.3 %), while datasets 2a (1999-2014, 66 sites) and 3a (2000-2014, 75 sites) show no discernible trends (median relative change = 0.0 % for both) (Table 1). Results from the analysis of monthly data from all three different data groupings (1b, 2b, 3b) (relative median change = -4.2 to -8.2 %) are similar to results for dataset 1a, based on analysis of annual data (Table 1). In the SMK tests on monthly data, two monthly "seasons" (January and April) in dataset 1b (1998-2014, 59 sites) are significant ( $p < 0.05$ ) with a third monthly "season" (August) near-significant at  $p = 0.06$ . For datasets 2b (1999-2014, 66 sites) and 3b (2000-2014, 75 sites), August is the only monthly "season" in either time series to be close to significance at  $p = 0.06$ . Trends in individual monthly "seasons" are therefore weak and results between the MK and seasonal MK tests on monthly data are similar (Table 1).

### 3.5.2 Linear regression parametric time series analysis

<INSERT TABLE 2>

The parametric linear regression time series trend analysis was also performed on the different data groupings. Results of the linear regression tests are summarised in Table 2, and a comparison of trends from the Mann-Kendall with the linear regression approach is provided in Figure 13 for annual datasets 1a, 2a, 3a, and Figure 14 for monthly datasets 1b, 2b, 3b. A similar approach to the Mann-Kendall was taken to assess the relative change, by calculating the % relative change from the estimated NH<sub>3</sub> concentration at the start ( $y_0$ ) and at the end ( $y_i$ ) of the time series ( $100 * [(y_i - y_0) / y_0]$ ) computed from the linear regression slope and intercept. The different data groupings all show small, but non-significant decreasing trends (relative change = -2.4 % to -5.3 %), similar to the trends and % relative median change from the MK and SMK analysis (Figure 13, Figure 14). This suggests that the errors in the NAMN data are normally distributed and that no or few outliers are present, since the results from the non-parametric Mann-Kendall are very similar to the parametric least squares linear regression.

<INSERT FIGURE 12>

<INSERT FIGURE 13>

### 3.5.3 Trends in NH<sub>3</sub> concentrations vs trends in NH<sub>3</sub> emissions

Overall, the long-term NH<sub>3</sub> concentration data from the UK NAMN suggests evidence of a small, but non-significant decreasing trend (Figure 13, Figure 14). The level of reduction observed in the datasets is however less than the 16.3 %, 15.6 % and 13.1 % reduction in estimated UK NH<sub>3</sub> emissions over the periods 1998-2014, 1999-2014 and 2000-2014, respectively (Tables 1,2). Inventories have inherent uncertainties such as uncertainties in activity data and emission factors, or may be missing emission sources. In terms of measurement data, it has already been shown in Sects. 3.1 and 3.3 that the annually averaged data mask considerable spatial and seasonal variability in NH<sub>3</sub> concentrations. Drivers contributing to this variability include the influence of climate on emissions, variations in management practice for a particular emission source, and influence of local emission sources and interactions on concentrations at a site. In addition, once emissions have taken place, the resulting atmospheric NH<sub>3</sub> concentrations are influenced by local deposition, which is in turn affected by receptor surfaces and by concentrations of interacting chemical species that affect atmospheric lifetime and transport distance of NH<sub>3</sub> and physical dispersion (e.g. Bleeker et al., 2009; Sutton et al., 2013). In the following sections, we consider the possibility of interactions with climate, emission source type and chemical interactions as this may affect long term trends in NH<sub>3</sub> concentrations.

### 3.5.4 Influence of climate

UK temperature and rainfall varied from year to year over the period 1998 to 2014 (Figure 12), with no clear relationship with NH<sub>3</sub> easily visible in the graph. Plotting the annual mean NH<sub>3</sub> concentrations against the average temperature and rainfall however does show indicatively that elevated annual mean NH<sub>3</sub> concentrations are observed in warmer years, and reduced annual mean NH<sub>3</sub> concentrations are observed in wetter years (Supp. Figure S3). This analysis for the full network is therefore consistent with the observation at a remote site (Inverpolly, Figure 10). The thermodynamic equilibrium shifts NH<sub>3</sub> from the aqueous (or particulate) phase to the gas phase with increased temperature, hence emissions from animal manures, soils and vegetation increase with increasing temperature (Asman et al., 1998; Sutton et al., 1993). Conversely, increases in precipitation decrease NH<sub>3</sub> emissions because rain events dilute the available NH<sub>3</sub> pool, while having the potential to wash urea and NH<sub>x</sub> in solution from the surface. As NH<sub>3</sub> is soluble and washed out of the atmosphere by rainfall, this should also contribute to reduced NH<sub>3</sub> concentrations during wet periods.

An exception to this relationship can occur where N is excreted as uric acid from birds (e.g. poultry). In this case, sufficient water is needed to allow hydrolysis to form NH<sub>3</sub> (Riddick et al., 2014). In this situation, the arrival of rain promoted uric acid hydrolysis from seabird guano surfaces, which was limited in the absence of soil moisture. It is possible that this interaction could lead to NH<sub>3</sub> emissions from field spreading of poultry litter to be larger in wetter years. In a recent trend analysis of NH<sub>3</sub> concentrations from the Dutch Air Quality Monitoring Network, an attempt was also made to correct for meteorological (temperature and rainfall) influences for the eight monitoring stations, which broadly produced similar results with slightly enhanced statistical significance for the trends (van Zanten et al., 2017).

### 3.5.5 Influence of local emission sources

<INSERT FIGURE 14>

<INSERT FIGURE 15>

The inter- and intra-annual variability is also expected to be linked to influences from local emission source and activities. It has already been shown in Sect. 3.1 that the concentrations of NH<sub>3</sub> in air are greatest in parts of the country with a large presence of livestock farming, particularly in areas of pig, poultry and cattle farming. Using the classification of NAMN sites according to dominant emission source sectors described in Sect. 3.1, the long-term change in NH<sub>3</sub> concentrations at sites grouped into four different emission source sectors: background, sheep, cattle, and pigs and poultry are compared in Figure 15 (annual mean data) and Figure 15 (monthly mean data). Results of the Mann-Kendall time series trend analysis are summarised in Table 3 and results of linear regression analysis are summarised in Table 4. A comparison of trends in measured NH<sub>3</sub> concentrations with trends in NH<sub>3</sub> emissions for the different source types then provided indicative evidence to support and inform the national emission inventory compilation. [In Figure 16, the relative changes in UK emissions between 1998 and 2014 are compared with relative changes in mean measured NH<sub>3</sub> concentrations for all NAMN sites, and for grouped sites classified as dominated by cattle, pigs & poultry, and sheep.](#)

<INSERT TABLE 3>

<INSERT TABLE 4>

For the 17 sites in cattle dominated areas, there is an increasing, but non-significant trend. Overall, based on MK analysis of annual data, the relative change from 1998 to 2014 is a 12 % increase (Table 3, Figure 15), compared with a smaller increase of 4 % from linear regression (Table 4, Figure 15). With the monthly data, there is no discernible trend (−0.9 % (MK); 1.4 % (LR)). In the seasonal MK test on monthly data (% relative median change = 3.9 %), no monthly “seasons” are significant, with only January approaching significance at  $p = 0.07$ . The near-significant trend for January is likely to be due to unusually high NH<sub>3</sub> concentrations recorded in January at some sites in the first few months of the time series, attributed to manure spreading activities taking place in the winter months when the ground was frozen (confirmed by local observations), in direct contravention of good farming practice.

In terms of UK cattle NH<sub>3</sub> emission, this has a decreasing trend with an estimated 11% decrease since 1998 (Figure 17, Table 5), and is therefore clearly in contrast to the non-discernible or small increasing trend (non-significant) in NH<sub>3</sub> concentrations from cattle sites. In principle, a signal related to substantial livestock changes associated with the 2000 outbreak of Foot and Mouth Disease might have been expected. However, this outbreak was actually rather localized in north-west England and



south-west England, and was followed by substantial restocking from 2001 (Sutton et al., 2006) and there was no detectable signal of FMD in the average for cattle-dominated areas.

<INSERT FIGURE 16>

5 <INSERT TABLE 5>

By contrast, in pig and poultry dominated areas (9 sites) there is a decreasing trend with significant reduction in measured NH<sub>3</sub> concentrations between 1998 and 2014 (-22 % (MK),  $p = 0.02$ , Table 3; -21 % (LR),  $p = 0.06$ , Table 4) from analysis of annual data (Figure 15). For the monthly data, the overall change based on linear regression is also a 22 % decrease ( $p = 0.02$ ) (Table 4, Figure 16), compared with a larger level of decrease based on MK analysis (-32 %,  $p = 0.01$ ) (Table 3, Figure 16). The SMK test also show a significant decreasing trend (-11 %, overall  $p < 0.001$ ), with 6 of the 12 monthly “seasons” showing significant trends (Feb, Jun, Nov, Dec:  $p < 0.05$ , Oct:  $p < 0.01$ , Jan:  $p < 0.001$ ). A decrease in emissions from pig and poultry of 39 % between 1998 and 2014 (Figure 17, Table 5) is therefore broadly supported, although not matched by a similar decrease in measured NH<sub>3</sub> concentrations.

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For sheep dominated sites (4 sites), there is an increasing trend in NH<sub>3</sub> (MK: +16 %,  $p = 0.17$ , Table 3; LR: 20 %,  $p = 0.09$ , Table 4) between 1998 and 2014 in the annual data (Figure 15). The monthly data also show a similar upward trend (Figure 15) with relative change in concentrations of +19% based on MK ( $p = 0.10$ ) (Table 3) and +17% based on LR ( $p = 0.14$ ) (Table 4). The increasing trend at sheep sites is therefore in contrast to the estimated 24 % decrease in NH<sub>3</sub> emissions from this sector since 1998 (Figure 17, Table 5). For the SMK test, no individual monthly “seasons” were significant, although 3 of the monthly “seasons” approached the significance level (Apr, Dec:  $p = 0.08$ , Oct:  $p = 0.09$ ). Overall, the increasing trend from the SMK test is significant at  $p < 0.01$ . While the Sen’s trend slope from both MK and SMK tests were comparable, at 0.0036 and 0.0033  $\mu\text{g NH}_3 \text{ y}^{-1}$ , respectively, the % relative median change results computed from them are very different (MK = 16 % *cf* SMK = 210 %), because the intercepts of the fitted Sen’s trend slopes are different (MK = 0.289  $\mu\text{g NH}_3 \text{ m}^{-3}$  *cf* SMK = -0.0267  $\mu\text{g NH}_3 \text{ m}^{-3}$ ). Caution therefore needs to be exercised when interpreting the % relative change results, especially at sites with low NH<sub>3</sub> concentrations, which must be examined together with the fitted trends.

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At background sites (5 sites where total NH<sub>3</sub> emissions for the respective 5 km grid squares are estimated at  $< 1 \text{ kg N ha}^{-1} \text{ y}^{-1}$ ), NH<sub>3</sub> concentrations also appear to have increased (non-significant). Based on the MK analysis for the period 1998 to 2014, NH<sub>3</sub> concentrations increased overall by 18 % and 13 % from the analysis of annual and monthly data, respectively (Table 3). Results from linear regression were similar, with an overall increase of 13 % and 12 % from analysis of the annual and monthly data, respectively (Table 4). Similar to sheep sites, the % relative median change estimated from the seasonal MK Sen’s slope and intercept (+ 49%) is larger than from the classic MK Sen’s slope (+13%) due to differences in the intercepts of the fitted trend lines (MK = 0.1528  $\mu\text{g NH}_3 \text{ m}^{-3}$  *cf* SMK = 0.0388  $\mu\text{g NH}_3 \text{ m}^{-3}$ ) since the trend slopes are the same (0.0012  $\mu\text{g NH}_3 \text{ y}^{-1}$ ). Overall, the SMK test show a significant increasing trend in the monthly data ( $p = 0.05$ ). No individual monthly “seasons” were significant, with March, April and November monthly “seasons” approaching the significance level ( $p = 0.09$ ).

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As with the annual UK-wide long-term datasets (Sect. 3.5), it is useful to consider the significance of the NH<sub>3</sub> trends for the groupings of sites according to dominant emission source sectors. Table 3 and Table 4 shows that neither the annual nor the monthly time series showed a significant change in NH<sub>3</sub> concentrations for the cattle dominated sites. In the case of pig and poultry dominated sites, the decrease in measured NH<sub>3</sub> concentrations was significant for both the annual and monthly datasets. For sheep dominated and backgrounds sites, the estimated increase in NH<sub>3</sub> concentrations was not significant based on the

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MK and linear regression tests on the annual and monthly data, but was significant based on the SMK test of the monthly data. Overall, these statistics confirm significant differences between NH<sub>3</sub> trends for sites dominated by different source types, with concentrations decreasing at pig and poultry dominated sites, concentrations increasing at sheep dominated and background sites, and no significant trend at cattle dominated sites (Table 5).

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### 3.5.6 Changing chemical climate and effects on long-term trends in NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup>

Other pollutants that affect NH<sub>3</sub> concentrations in the atmosphere include SO<sub>2</sub> and NO<sub>x</sub> emissions, which determine rates of secondary inorganic aerosol formation and therefore the lifetime of NH<sub>3</sub> in the atmosphere. UK emissions of SO<sub>2</sub> are estimated to have declined significantly by 81 % from 1.6 million tonnes in 1998 to 0.3 million tonnes in 2014 (Defra, 2015). Similarly, NO<sub>x</sub> emissions over the same period are estimated to have fallen by 50 % from 2 million tonnes to 1 million tonnes (Defra, 2015). The reaction of NH<sub>3</sub> with H<sub>2</sub>SO<sub>4</sub> to form (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> is effectively irreversible (in the absence of in-cloud reprocessing), whereas an equilibrium exists between gaseous NH<sub>3</sub> and particulate NH<sub>4</sub>NO<sub>3</sub> and NH<sub>4</sub>Cl components which are appreciably volatile at ambient temperatures. A change in the particulate phase from (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> to NH<sub>4</sub>NO<sub>3</sub> suggests that NH<sub>3</sub> will remain longer in the atmosphere, since NH<sub>4</sub>NO<sub>3</sub> is volatile and releases NH<sub>3</sub> in warm weather.

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Elsewhere, a mismatch between reported trends in emissions and measurement data have similarly been investigated. The question of the 'Ammonia Gap' in the Netherlands was debated over a number of years. There, the estimated reduction in emissions due to mitigation measures was not matched by expected decreases in measured NH<sub>3</sub> concentrations in air and/or NH<sub>4</sub><sup>+</sup> in precipitation (Erisman et al., 2001; Bleeker et al., 2009; van Zanten et al., 2017). Similarly in Hungary, monitored NH<sub>3</sub> concentrations from long-term measurements did not match the estimated reduction in NH<sub>3</sub> emissions following the decline in agricultural livestock population and fertiliser usage after political changes in 1989 (Horvath and Sutton, 1998). This was subsequently attributed to a reduction in SO<sub>2</sub> emissions over the same period, increasing the atmospheric lifetime of NH<sub>3</sub> (Horvath et al., 2009).

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Dry deposition of SO<sub>2</sub> and NH<sub>3</sub> are enhanced in the presence of both gases, an interaction referred to as "co-deposition" (Fowler et al., 2001). The acid-base neutralization by each of the gases provides an efficient sink for dry deposition on leaf surfaces and deposition enhancement for each gas depends on the relative air concentrations of NH<sub>3</sub> and SO<sub>2</sub>. For SO<sub>2</sub>, the dry deposition process has been shown to be strongly influenced by ambient concentrations of NH<sub>3</sub> because the surface resistance is regulated mainly by uptake in moisture on foliar surfaces, which, in turn, is strongly influenced by the presence of NH<sub>3</sub>. The large reduction in SO<sub>2</sub> emissions and ambient concentrations, compared with the relative stagnation in NH<sub>3</sub> emissions and concentrations over the same period has meant that the SO<sub>2</sub>/NH<sub>3</sub> ratio has decreased dramatically. This has led to a systematic decrease in canopy resistance to uptake of SO<sub>2</sub> on surfaces, increasing dry deposition of SO<sub>2</sub> in the UK (ROTAP 2012). The underlying cause of the decrease in surface resistance is that the ambient NH<sub>3</sub> is sufficient to neutralize acidity from the solution and oxidation of deposited SO<sub>2</sub>, maintaining large rates of deposition.

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Similar interactions are seen to be occurring in the UK based on the NAMN data, where the concurrent reduction in SO<sub>2</sub> and NO<sub>x</sub> emissions over the same period (Figure 19b) should theoretically lead to a longer atmospheric lifetime of NH<sub>3</sub>, thereby increasing NH<sub>3</sub> concentrations in the UK, especially in remote areas. The interpretation of the NH<sub>3</sub> measurement data can further be aided by comparison with particulate nitrate (NO<sub>3</sub><sup>-</sup>) and sulphate (SO<sub>4</sub><sup>2-</sup>) data from the UK AGANet that are made concurrently with the NAMN NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> measurements at 30 sites (see Sect. 2.2). For particulate NH<sub>4</sub><sup>+</sup>, it has already been shown in Sect. 3.3 that this regional species has less of a relationship to the dominant NH<sub>3</sub> source sectors; trend analysis was therefore undertaken using all NH<sub>4</sub><sup>+</sup> site data combined. As with the NH<sub>3</sub> time series analysis, sites with incomplete data runs

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for particulate  $\text{NH}_4^+$  due to reduced density of  $\text{NH}_4^+$  measurements and site changes occurring from 2001-2006 were excluded (see Sect. 2.2.1).

<INSERT TABLE 6>

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Two data series for NAMN  $\text{NH}_4^+$  data were selected for analysis, i) 23 sites with complete  $\text{NH}_4^+$  time series from 1999 to 2014, and ii) 30 sites with complete  $\text{NH}_4^+$  time series from 2006 to 2014. Both time series show a large significant downward trend in  $\text{NH}_4^+$  ( $p < 0.01$ ) (Table 6, Supp. Figure S4). Overall, MK and LR tests show a significant decrease in  $\text{NH}_4^+$  concentrations by 47 % and 49 %, respectively, between 1999 and 2014 and by 44 % and 43 %, respectively, between 2006 and 2014 (Table 6, Supp. Figure S4). By contrast, concurrent  $\text{NH}_3$  data from the same sites over the same time periods showed a much smaller, non-significant downward trend between 1999 and 2014 (-17 % (MK); -18 % (LR)), and no discernible trend between 2006 and 2014 (+ 3 % (MK and LR)) (Figure 17a, Table 6, Supp. Figure S4). This reduction in particulate  $\text{NH}_4^+$  can be seen to be closely associated with parallel decreases in particulate  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  concentrations from AGANet (Table 7, Figure 19a), which are themselves associated with reductions in  $\text{SO}_2$  and  $\text{NO}_x$  emissions (Table 7, Figure 19b).

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<INSERT FIGURE 17>

<INSERT FIGURE 18>

<INSERT TABLE 7>

20 The comparisons above therefore suggest that reductions in  $\text{SO}_2$  and  $\text{NO}_x$  emissions over the period have led to a slower formation of particulate  $\text{NH}_4^+$  in the atmosphere. Further evidence in support of this is indicated by plotting the ratio of  $\text{NH}_3/\text{NH}_4^+$  (Figure 17b), which has increased from 1.8 in 1999 to 2.8 in 2014. This demonstrates how a larger fraction of the reduced N is staying in the gas phase as  $\text{NH}_3$ , increasing its atmospheric residence time and maintaining  $\text{NH}_3$  concentrations at a higher level than solely based on  $\text{NH}_3$  emission trends. Although the overall changes in  $\text{NH}_3$  concentrations in the UK dataset are small and in many cases not significant for particular data groupings, they are consistent with similar phenomena observed in Hungary, the Netherlands and Denmark (Horvath et al., 2009; Erisman et al., 2001; Sutton et al., 2003; Bleeker et al., 2009).

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#### 30 4. CONCLUSIONS

Spatial and temporal trends in  $\text{NH}_3$  are found to be related to variability in emission source types across the UK and also to be influenced by changes in environmental conditions. Extensive spatial heterogeneity in  $\text{NH}_3$  concentrations was observed, with lowest annual mean concentrations at remote sites ( $< 0.2 \mu\text{g m}^{-3}$ ) and highest in the areas with intensive agriculture (up to  $22 \mu\text{g m}^{-3}$ ).  $\text{NH}_4^+$  concentrations show less spatial variability (e.g. range of 0.14 to  $1.8 \mu\text{g m}^{-3}$  annual mean in 2005) with a general decreasing gradient from the south-east to the north-west of the UK, due to both regional differences in  $\text{NH}_3$  concentrations and import of particulate matter into south-east England from Europe (Vieno et al., 2014; Dore et al., 2015).

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Peak  $\text{NH}_3$  concentrations are observed in summer at background sites (defined by 5 km grid average  $\text{NH}_3$  emissions  $< 1 \text{ kg N ha}^{-1} \text{ y}^{-1}$ ) and in areas dominated by sheep farming, driven by increased volatilization of  $\text{NH}_3$  in warmer summer temperatures.

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In areas where cattle, pig and poultry farming is dominant, the largest  $\text{NH}_3$  concentrations are in spring and autumn, matching periods of manure application to fields. By contrast, peak concentrations of  $\text{NH}_4^+$  aerosol occur in spring from long-range

transboundary sources. The spatial and seasonal patterns established for sites influenced by different emission source sectors are important for providing a foundation to understanding NH<sub>3</sub> exchange processes, impacts and the UK NH<sub>3</sub> budget, and to inform abatement strategies.

5 Official published estimates of UK NH<sub>3</sub> emissions are estimated to have declined by 16.3 % between 1998 and 2014. The long-term NH<sub>3</sub> concentration data from the UK NAMN suggests evidence of a smaller, but non-significant decreasing trend (−6.3 % (MK); −3.1 % (LR)), based on analysis of annually averaged data (*n* = 59) over the same period (Table 2). Analysis of annually averaged data for different groupings of the NAMN dataset for the time periods 1999-2014 (*n* = 66) and 2000-2014 (*n* = 75) also gave similar results. In each case, the level of reduction observed in the datasets (1999-2014: 0.0 % (MK) vs −3.0 % (LR); 2000-2014: 0.0 % (MK) vs −2.8 % (LR)) is less than the 15.6 % and 13.1 % reduction in estimated UK NH<sub>3</sub> emissions over the periods 1999-2014 and 2000-2014, respectively (Table 2).

In areas with intensive pig and poultry farming, there is a significant downward trend in NH<sub>3</sub> concentrations from the analysis of annually averaged data (−22 % (MK), *p* = 0.02; −21 % (LR), *p* = 0.06) that is consistent with, but not as large as the decrease in estimated NH<sub>3</sub> emissions from this sector over the same period (−39 %) (Table 5). By contrast, in cattle-dominated areas, there is evidence of a small increasing, but non-significant trend in NH<sub>3</sub> concentrations (+12 % (MK); +3.6 % (LR): annually averaged data), despite the decline in NH<sub>3</sub> emissions from this sector since 1998 (−11%) (Table 5). At background and sheep dominated sites, NH<sub>3</sub> concentrations increased (non-significant) over the monitoring period (Table 5). These increases in NH<sub>3</sub> concentrations at background (+17 % (MK); +13 % (LR): annually averaged data) and sheep dominated sites (+15 % (MK); +19 % (LR): annually averaged data) are consistent with decreasing SO<sub>2</sub> emissions (and to a lesser extent NO<sub>x</sub> emissions) associated with a change in the PM from (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> to NH<sub>4</sub>NO<sub>3</sub>, the latter being volatile and releasing NH<sub>3</sub> in warm weather.

Particulate NH<sub>4</sub><sup>+</sup> represents a secondary pollutant formed from NH<sub>3</sub> and oxidation products of acidic gases such as SO<sub>2</sub> and NO<sub>x</sub>. As the emissions of these acidic gases have reduced over the past years, the ratio between NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> has increased from 1.8 to 2.8 between 1999 and 2014. These changes are consistent with observed decreases in particulate SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> concentrations that are associated with decline in SO<sub>2</sub> and NO<sub>x</sub> emissions over the same period. This effect appears to be of sufficient magnitude to explain the lack of overall decrease in NH<sub>3</sub> concentrations, where the decrease in NH<sub>4</sub><sup>+</sup> is larger than for NH<sub>3</sub> at corresponding sites. Overall, UK annual particulate NH<sub>4</sub><sup>+</sup> concentrations decreased by −47 % (MK) and −49 % (LR) for period 1999 -2014, associated with a slower formation of particulate NH<sub>4</sub><sup>+</sup> in the atmosphere from gas-phase NH<sub>3</sub>. The findings are consistent with a parallel change in partitioning from particulate NH<sub>4</sub><sup>+</sup> to gaseous NH<sub>3</sub> as also detected in Hungary, the Netherlands and Denmark.

Until now, only a modest commitment has been agreed to reduce European NH<sub>3</sub> emissions. By contrast, SO<sub>2</sub> and NO<sub>x</sub> emissions have decreased over Europe over the past decades, and are projected to decrease further under the revised Gothenburg Protocol and revised NECD. As a result, the importance of NH<sub>3</sub> relative to oxidised N and SO<sub>2</sub> emissions is expected to continue to increase over the next decades, playing a significant role in the formation of fine PM and contributing to ecosystem effects through N deposition. With longer atmospheric lifetimes of gaseous NH<sub>3</sub> and little commitment to reduce emissions, combined with climate warming effects tending to increase NH<sub>3</sub> emissions, there is a substantial risk that exceedance of the NH<sub>3</sub> critical levels may increase in the future, exacerbating the threat to the most sensitive semi-natural habitats. The growing relative importance of reduced nitrogen to total acidic and total nitrogen deposition indicates that future strategies to tackle acidification and eutrophication will need to include measures to abate emissions of NH<sub>3</sub>.

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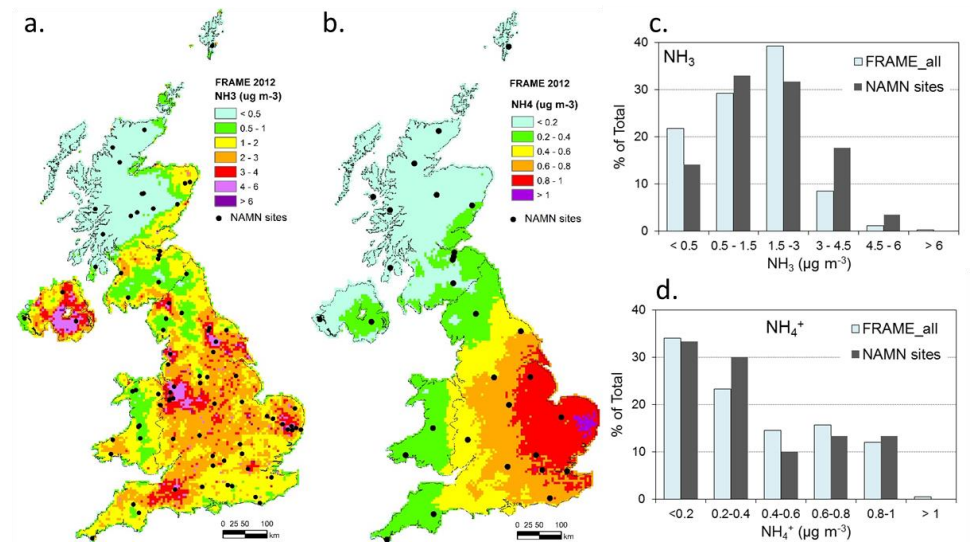


Figure 2: Maps of modelled annual mean concentrations of (a)  $\text{NH}_3$  and (b)  $\text{NH}_4^+$  at  $5 \text{ km} \times 5 \text{ km}$  grid resolution from the FRAME atmospheric transport model using 2012 UK emissions data, based on Dore et al. (2008), overlaid with the National Ammonia Monitoring Network (NAMN) measurement sites, and frequency distributions of the modelled concentrations of (c)  $\text{NH}_3$  and (d)  $\text{NH}_4^+$  for the FRAME  $5 \text{ km}$  grid squares containing a NAMN site (85 and 30 sites, respectively, in 2012) and for all model grid squares over the UK.

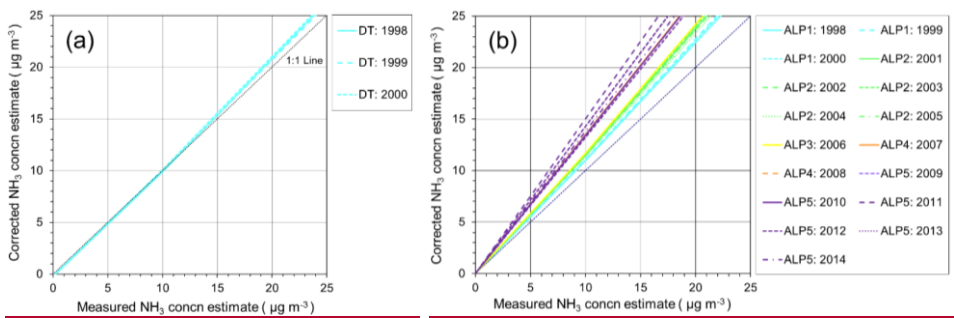


Figure 3: Comparison of annual empirical calibration curves for the passive samplers against the reference estimates from DELTA sampling at  $> 8$  sites in the UK National Ammonia Monitoring Network (NAMN). (a) DT = Diffusion Tubes. (b) ALP = ALPHA samplers, ALP1 is prototype 1 (1998-2000), ALP2 (2001-2005) and ALP3-ALP5 were manufactured from injection moulds 1 and 2, respectively. ALP4 and ALP 5 = new inlet PTFE membrane (Swiftlab 07-OPM-027:  $305 \mu\text{m}$ , regular polypropylene grid support material) that replaced the previous TE38 PTFE membrane ( $265 \mu\text{m}$ , randomly arranged polypropylene support material). ALP5 = new laboratory with analysis on FloRRia (previously on AMFIA).

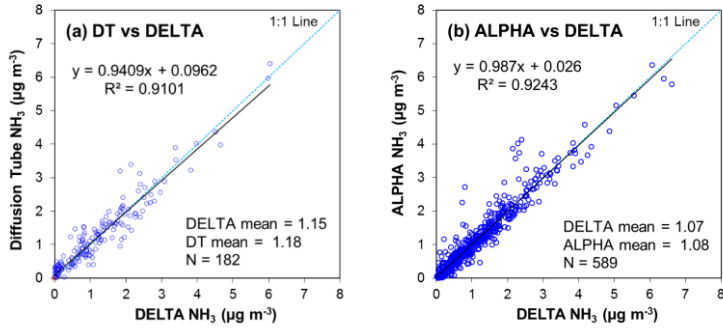


Figure 4: Regression of passive samplers vs DELTA measurements at >8 sites in the UK National Ammonia Monitoring Network (NAMN), showing results for (a) diffusion tubes (DT), used during the early years of the network (1998-2000), and (b) for ALPHA samplers (results shown are for 2009-2014 where all analyses were carried out at a new laboratory). All passive data shown are the monthly measured concentrations for each site using the calibrated data for the respective passive methods.

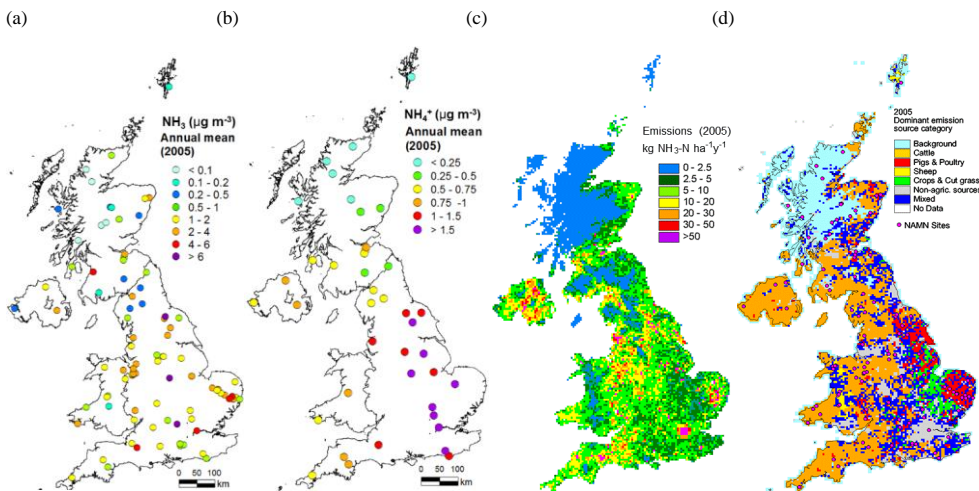
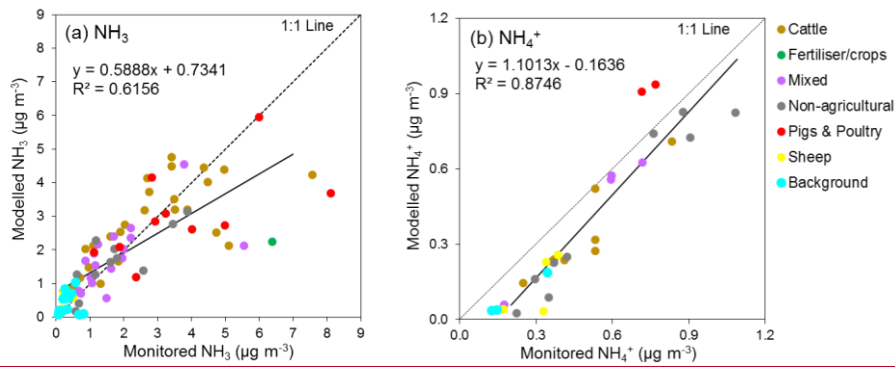
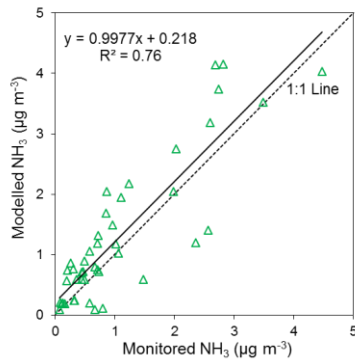


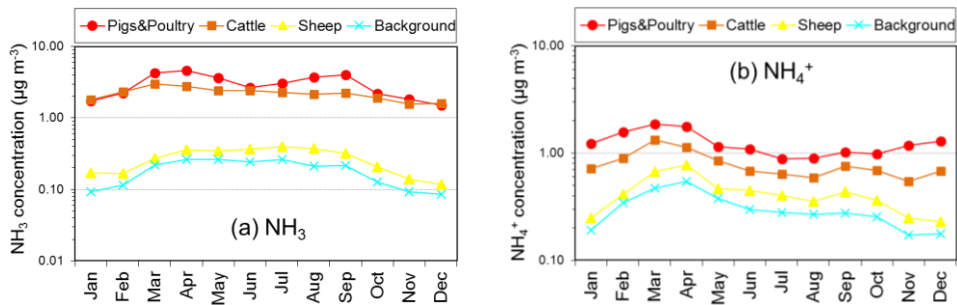
Figure 5: Measured annual mean concentrations from the UK National Ammonia Monitoring Network (NAMN) for 2005 for (a)  $\text{NH}_3$  and (b) particulate  $\text{NH}_4^+$ , and maps at 5 km by 5 km grid resolution for 2005 of (c) the estimated annual  $\text{NH}_3$  emissions (Dragosits et al. 2005) and (d) the dominant  $\text{NH}_3$  emission source category (based on Hellsten et al., 2008), indicating the relationships between measured air concentrations and spatial variability in  $\text{NH}_3$  sources emissions.



5 **Figure 6: Comparison of 2012 annual mean concentrations of (a) NH<sub>3</sub> and (b) NH<sub>4</sub><sup>+</sup> modelled using the FRAME atmospheric model with 2012 measurements from the UK National Ammonia Monitoring Network (NAMN) for all sites according to dominant emission source classification.**



10 **Figure 7: Comparison of 2012 annual mean concentrations of NH<sub>3</sub> from output of the FRAME atmospheric model with measurements from the UK National Ammonia Monitoring Network (NAMN) for a subset of sites classified as located in semi-natural or forest locations.**



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Figure 8: Seasonal trends in (a)  $\text{NH}_3$  (mean monthly data for 1998-2014) and (b)  $\text{NH}_4^+$  (mean monthly data for 1999-2014) concentrations of sites in the UK National Ammonia Monitoring Network (NAMN) classified according to four key emission source categories: cattle, sheep, pigs & poultry and background (based on 2005 dominant emission source classification). The concentrations are plotted on a log scale for better visualisation of the low concentration background and sheep profiles.

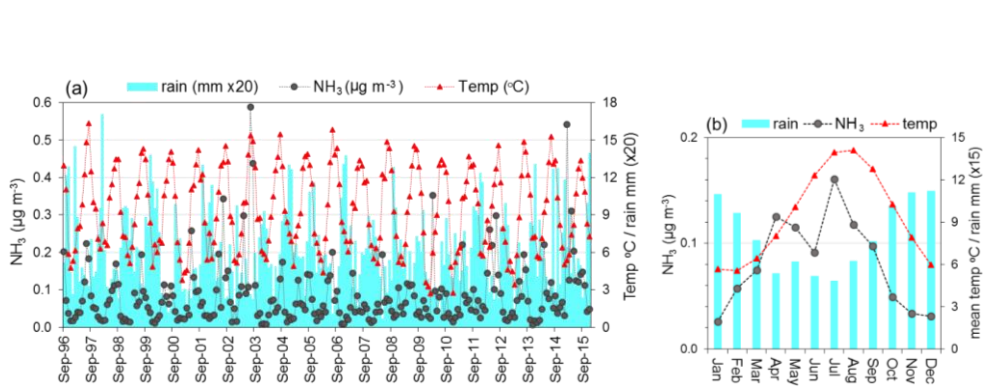


Figure 9: (a) Long-term trends in measured monthly-mean  $\text{NH}_3$  concentrations at the remote background Inverpolly site in NW Scotland (UKA00457), demonstrating strong intra- and inter-annual variability, from the UK National Ammonia Monitoring Network (NAMN). Also plotted for comparison are monthly rainfall and temperature data from the nearby Aultbea meteorological station (ID no. 52; MetOffice, 2016). (b) Comparison of seasonal trends in  $\text{NH}_3$  concentrations with temperature and rainfall at Inverpolly. Data shown are averaged over the period 1996 – 2015. Peak concentrations of  $\text{NH}_3$  can be seen to coincide with summer maxima in the temperature profile, while the lowest concentration occur in winter when the temperature is lowest and also when rainfall is generally highest.



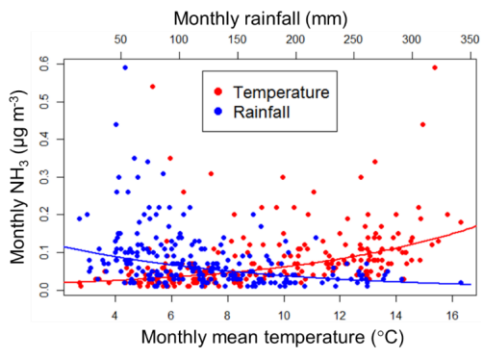
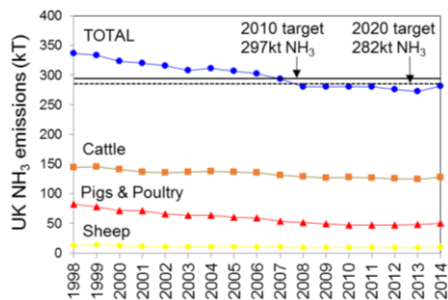


Figure 10. Relationships between measured monthly-mean  $\text{NH}_3$  concentrations from the UK National Ammonia Monitoring Network (NAMN) and mean monthly temperature and rainfall at Inverpolly (UKA00457).  $\text{NH}_3$  was negatively correlated with rainfall (blue line:  $\text{Log}(\text{NH}_3) = -0.0059 \cdot \text{Log}(\text{rain}) - 2.1612$ ,  $R^2 = 0.19$ ,  $n = 231$ ,  $p < 0.05$ ) and positively correlated with temperature (red line:  $\text{Log}(\text{NH}_3) = 0.1482 \cdot \text{Log}(\text{temp}) - 4.2708$ ,  $R^2 = 0.33$ ,  $n = 231$ ,  $p < 0.05$ ). Rain and temperature data are from the nearby Aultbea meteorological station (ID no. 52; MetOffice, 2016).

5

10 (a)



(b)

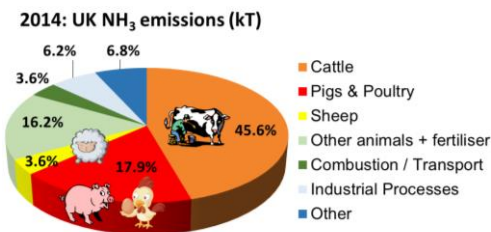


Figure 11: (a) Trends between 1998 and 2014 in the UK National Atmospheric Emission Inventory (NAEI) for total UK  $\text{NH}_3$  emissions and selected sub-sources: cattle, pigs & poultry and sheep. The 2010  $\text{NH}_3$  national emissions ceilings target of 297 kt (Gothenburg protocol and NECD) and the 2020 target of 282 kt (revised Gothenburg protocol) area are also shown for comparison. (b) UK  $\text{NH}_3$  emission sources in 2014. Data from <http://naei.defra.gov.uk/> and Misselbrook et al. 2015.

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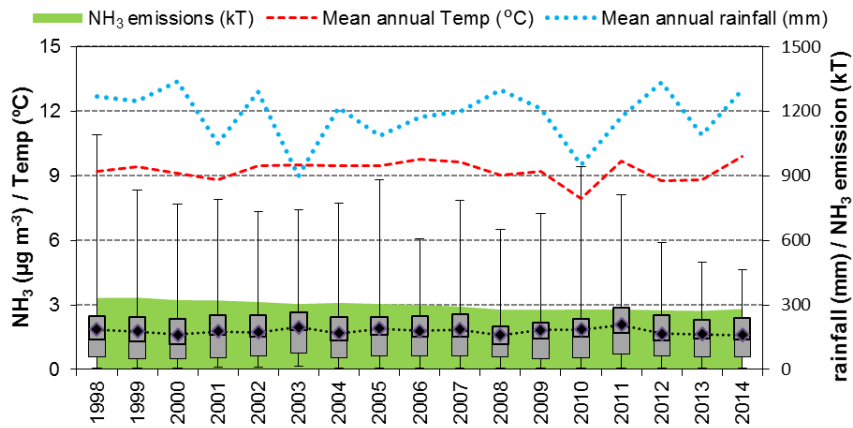


Figure 12: Changes in annual mean atmospheric NH<sub>3</sub> concentrations averaged over all sites in the National Ammonia Monitoring Network (NAMN) operational between 1998 and 2014 (59 sites). The diamonds show the mean NH<sub>3</sub> concentration, with the grey box indicating the median and interquartile range, while the error bars show the range (minimum and maximum) of measured mean concentrations. Annual mean UK meteorological data (source <http://www.metoffice.gov.uk/>) are also plotted for comparison over the same period. 2010 was an unusual year, characterised by a considerably lower than average mean annual temperature of 7.9 °C due to exceptionally cold winters, with Dec 2010 recorded as the coldest for over 100 years (*cf.* mean = 9.2 °C for 1998 to 2014) and lower than average rainfall of 950 mm (*cf.* mean = 1190 mm for 1998 to 2014).

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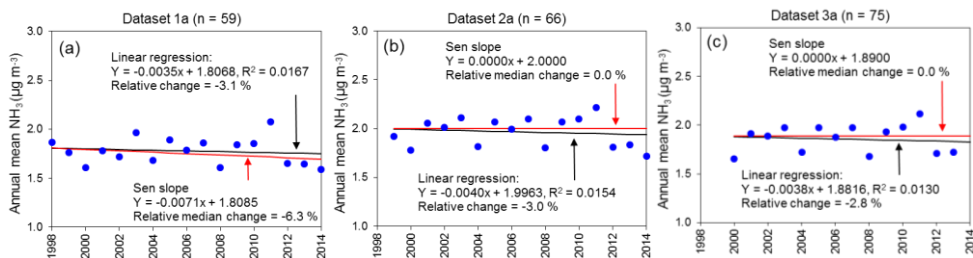


Figure 13: Time series trend analysis by non-parametric Mann-Kendall Sen slope vs parametric linear regression on annually averaged  $\text{NH}_3$  concentrations from the UK National Ammonia Monitoring Network (NAMN) for a) dataset 1a (1998-2014,  $n=59$ ), b) dataset 2a (1999-2014,  $n=66$ ) and c) dataset 3a (2000-2014,  $n=75$ ). Individual data points are annually averaged  $\text{NH}_3$  concentrations.

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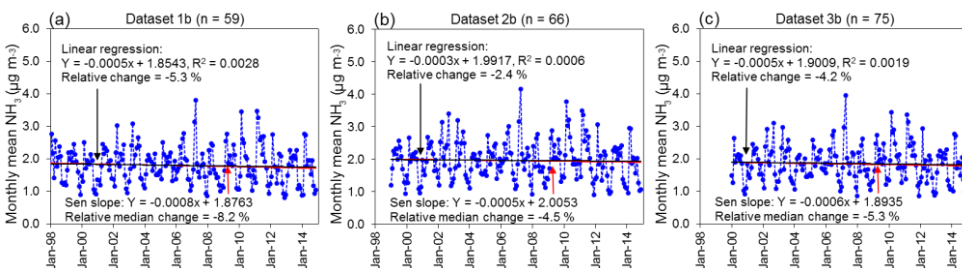
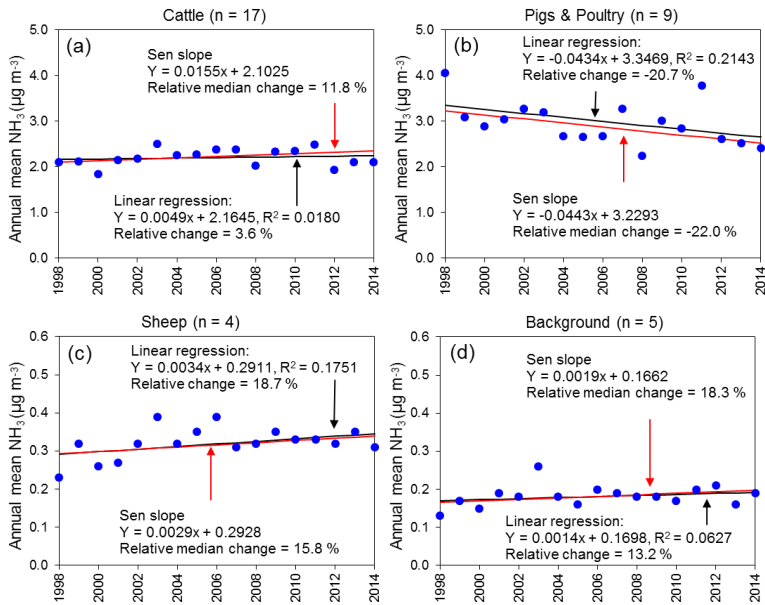


Figure 14: Time series trend analysis by non-parametric Mann-Kendall Sen slope vs parametric linear regression on monthly mean  $\text{NH}_3$  concentrations from the UK National Ammonia Monitoring Network (NAMN) for a) dataset 1b (1998-2014,  $n=59$ ), b) dataset 2b (1999-2014,  $n=66$ ) and c) dataset 3b (2000-2014,  $n=75$ ). Individual data points are monthly mean  $\text{NH}_3$  concentrations.

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5 Figure 15: Time series trend analysis by non-parametric Mann-Kendall Sen slope vs parametric linear regression on annually averaged  $\text{NH}_3$  concentrations from the UK National Ammonia Monitoring Network (NAMN) for sites in 5 km grid squares classed as dominated by (a) cattle (> 45 % of total  $\text{NH}_3$  emissions from this category in a grid square); (b) pigs & poultry (> 45 % of total  $\text{NH}_3$  emissions from this category in a grid square); (c) sheep (> 45 % of total  $\text{NH}_3$  emissions from sheep in a grid square); (d) NAMN sites in grid squares classed as background (defined as grid squares with average  $\text{NH}_3$  emissions <1 kg N ha<sup>-1</sup> y<sup>-1</sup>). Individual data points are annually averaged  $\text{NH}_3$  concentrations.

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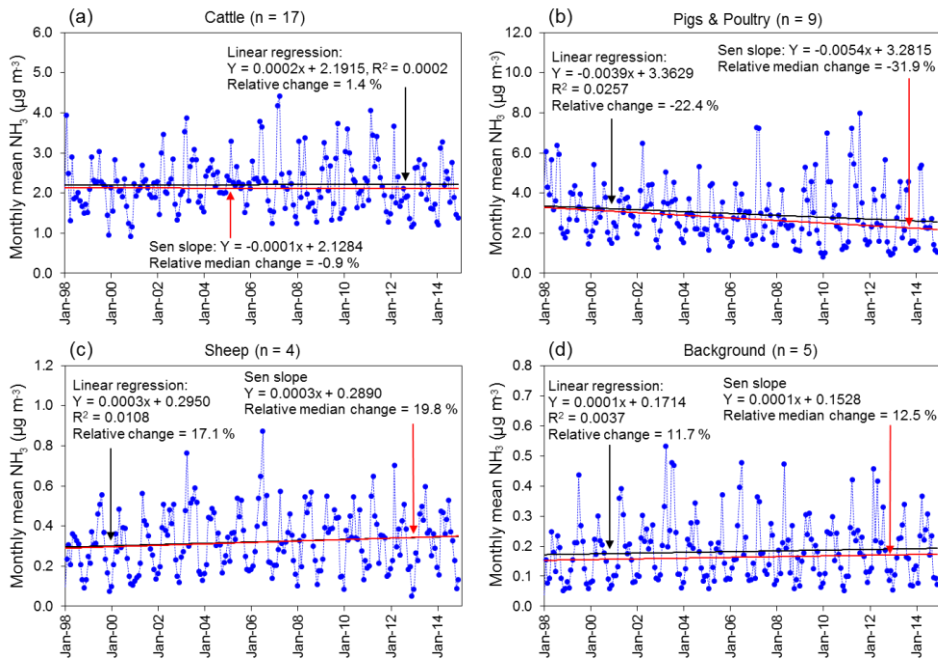


Figure 16: Time series trend analysis by non-parametric Mann-Kendall Sen slope vs parametric least squares linear regression on annually averaged  $\text{NH}_3$  concentrations from the UK National Ammonia Monitoring Network (NAMN) for sites in 5 km grid squares classed as dominated by (a) cattle (> 45 % of total  $\text{NH}_3$  emissions from this category in a grid square); (b) pigs & poultry (> 45 % of total  $\text{NH}_3$  emissions from this category in a grid square); (c) sheep (> 45 % of total  $\text{NH}_3$  emissions from sheep in a grid square); (d) NAMN sites in grid squares classed as background (defined as grid squares with average  $\text{NH}_3$  emissions < 1 kg N ha<sup>-1</sup> y<sup>-1</sup>). Individual data points are monthly mean  $\text{NH}_3$  concentrations.

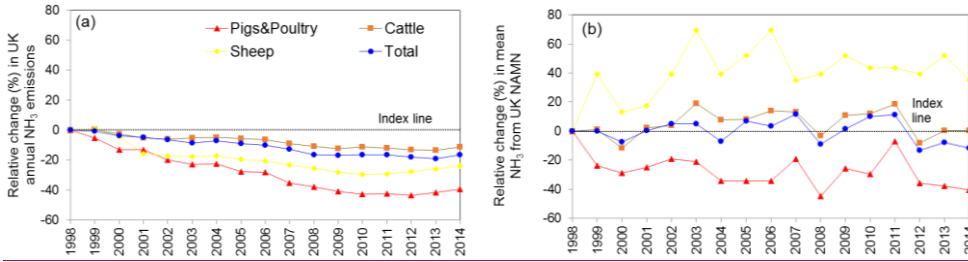


Figure 17: (a) Relative trends between 1998 and 2014 in NH<sub>3</sub> emissions from the UK National Atmospheric Emission Inventory (NAEI) for total emissions (all NH<sub>3</sub> sources) and emissions from cattle, pigs & poultry, and sheep separately (data from: <http://naei.defra.gov.uk/> and Misselbrook et al, 2015). (b) Relative trends between 1998 and 2014 in measured annual mean NH<sub>3</sub> concentrations ( $\mu\text{g NH}_3 \text{ m}^{-3}$ ) for all UK National Ammonia Monitoring Network (NAMN) sites, and for grouped sites classified as dominated by cattle, pigs & poultry, and sheep. Both figures are plotted with the same scale to allow direct comparison of the relative magnitudes in trends.

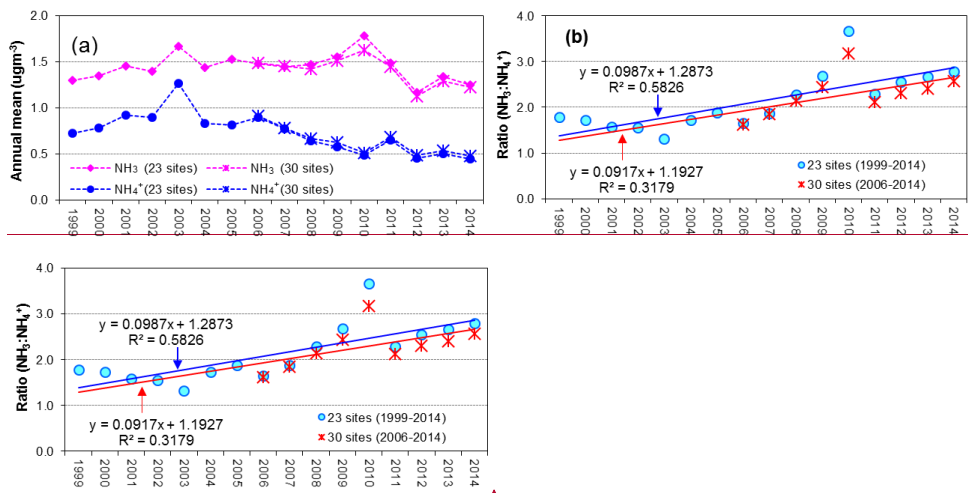


Figure 18: (a) Long-term trends in annual mean concentrations of measured particulate NH<sub>4</sub><sup>+</sup> from the UK National Ammonia Monitoring Network (NAMN) comparing i) 23 sites with complete NH<sub>4</sub><sup>+</sup> time series from 1999 to 2014, and ii) 30 sites with complete NH<sub>4</sub><sup>+</sup> time series from 2006 to 2014. For comparison, long-term trends in annual mean concentrations of measured NH<sub>3</sub> from the same sites over the same time periods are also shown. (b) Long-term trends in Plot of ratio of NH<sub>3</sub>:NH<sub>4</sub><sup>+</sup>, based on data from graph a, indicating an increase in this ratio with time. The comparisons shown is for datasets i) 23 sites with complete NH<sub>4</sub><sup>+</sup> time series from 1999 to 2014, and ii) 30 sites with complete NH<sub>4</sub><sup>+</sup> time series from 2006 to 2014.

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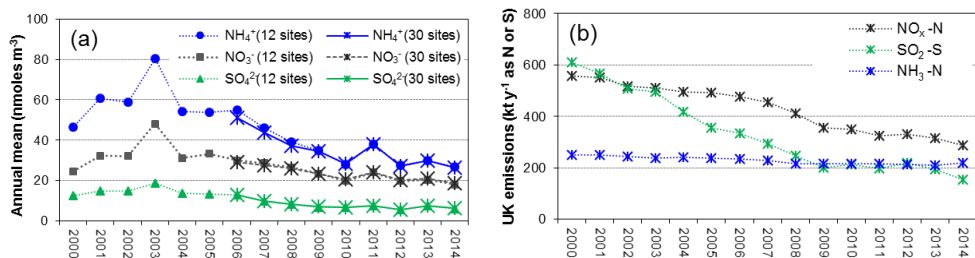


Figure 19: (a) Long-term trends in particulate NH<sub>4</sub><sup>+</sup> from the UK National Ammonia Monitoring Network (NAMN) compared with particulate NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> concentrations from the UK Acid Gas and Aerosol Network (AGANet; Connolly et al. 2016) measured at the same time. Each data point represents the averaged monthly measurements from all AGANet sites (increased from 12 to 30 sites since Jan 2006) and also the original 12 AGANet sites in the network (1999 data were excluded as measurements started in September 1999). (b) Trends in total UK emissions of NH<sub>3</sub>, NO<sub>x</sub> and SO<sub>2</sub> over the same period (2000-2014). Data from the National Atmospheric Emission Inventory (NAEI, <http://naei.defra.gov.uk/>).

**Table 1: Summary of Mann-Kendall (MK) and Seasonal Mann-Kendall (SMK) time series trend analysis on NH<sub>3</sub> data (annually averaged datasets 1a, 2a, 3a and monthly mean datasets 1b, 2b, 3b) from the UK National Ammonia Monitoring Network (NAMN). The following are shown: the *p*-value, median annual trend (Sen's slope, in  $\mu\text{g NH}_3 \text{ y}^{-1}$ ) and the relative median change over the selected time period (in %). For the MK tests, the 95% confidence interval (CI) for the trend and relative change are also estimated. For comparison, the reduction in estimated UK NH<sub>3</sub> emissions over the periods 1998-2014, 1999-2014 and 2000-2014 are 16.3 %, 15.6 % and 13.1 % respectively.**

Dataset	Time series	<sup>a</sup> Number of sites	<i>p</i> -value	Significant trend ( <i>p</i> <0.05)	<sup>b</sup> Median annual trend & [95% CI] ( $\mu\text{g NH}_3 \text{ y}^{-1}$ )	<sup>c</sup> Relative median change over the period & [95% CI] (%)
1a: annual (MK)	1998-2014	59	0.46	no	-0.0071 [-0.0200, 0.0125]	-6.3 [-16, 12]
1b: monthly (MK)	1998-2014	59	0.22	no	-0.0096 [-0.0264, 0.0060]	-8.2 [-21, 5.5]
1b: monthly (SMK)	1998-2014	59	0.10	no	-0.0100	-5.8
2a: annual (MK)	1999-2014	66	1.00	no	0.0000 [-0.0227, 0.0200]	0.0 [-16, 16]
2b: monthly (MK)	1999-2014	66	0.51	no	-0.0060 [-0.0252, 0.0132]	-4.5 [-18, 11]
2b: monthly (SMK)	1999-2014	66	0.25	no	-0.0073	-4.2
3a: annual (MK)	2000-2014	75	1.00	no	0.0000 [-0.0283, 0.0175]	0.0 [-19, 14]
3b: monthly (MK)	2000-2014	75	0.43	no	-0.0072 [-0.0264, 0.0120]	-5.3 [-18, 9.5]
3b: monthly (SMK)	2000-2014	75	0.15	no	-0.0079	-4.5

<sup>a</sup>Number of sites providing complete data runs over the time period.

<sup>b</sup>Median annual trend = fitted Sen's slope of Mann-Kendall linear trend (unit =  $\mu\text{g NH}_3 \text{ y}^{-1}$ )

<sup>c</sup>Relative median change calculated based on the NH<sub>3</sub> concentration at the start ( $y_0$ ) and at the end ( $y_i$ ) of time series computed from the Sen's slope and intercept ( $=100*[(y_i - y_0) / y_0]$ ).

**Table 2: Summary of linear regression time series trend analysis on NH<sub>3</sub> data (annually averaged datasets 1a, 2a, 3a and monthly mean datasets 1b, 2b, 3b) from the UK National Ammonia Monitoring Network (NAMN). The following are shown: the *p*-value, annual trend (fitted slope, in  $\mu\text{g NH}_3 \text{ y}^{-1}$ ), *R*<sup>2</sup>, and the relative change over the selected time period (in %). For comparison, the reduction in estimated UK NH<sub>3</sub> emissions over the periods 1998-2014, 1999-2014 and 2000-2014 are 16.3 %, 15.6 % and 13.1 % respectively.**

Dataset	Time series	<sup>a</sup> Number of sites	<i>p</i> -value	Significant trend ( <i>p</i> <0.05)	<sup>b</sup> Annual Trend ( $\mu\text{g NH}_3 \text{ y}^{-1}$ )	<i>R</i> <sup>2</sup>	<sup>c</sup> Relative change over the period (%)
1a: annual	1998-2014	59	0.62	no	-0.0035	0.0167	-3.1
1b: monthly	1998-2014	59	0.45	no	-0.0062	0.0028	-5.3
2a: annual	1999-2014	66	0.65	no	-0.0040	0.0154	-3.0
2b: monthly	1999-2014	66	0.74	no	-0.0031	0.0006	-2.4
3a: annual	2000-2014	75	0.69	no	-0.0038	0.0130	-2.8
3b: monthly	2000-2014	75	0.56	no	-0.0057	0.0019	-4.2

<sup>a</sup>Number of sites providing complete data runs over the time period.

<sup>b</sup>Annual trend = fitted slope of linear regression (unit =  $\mu\text{g NH}_3 \text{ y}^{-1}$ )

<sup>c</sup>Relative change calculated based on the estimated annual NH<sub>3</sub> concentration at the start ( $y_0$ ) and at the end ( $y_i$ ) of time series ( $=100*[(y_i - y_0) / y_0]$ ) computed from the slope and intercept ( $=100*[(y_i - y_0) / y_0]$ ).



Table 3: Summary of Mann-Kendall (MK) and Seasonal Mann-Kendall (SMK) time series trend analysis on grouped NH<sub>3</sub> concentration data (annually averaged and monthly mean data) from the UK National Ammonia Monitoring Network (NAMN) for four different emission source sectors. The following are shown: the *p*-value, median annual trend (Sen's slope, in  $\mu\text{g NH}_3 \text{ y}^{-1}$ ) and the relative median change over the selected time period (in %). For the MK tests, the 95% confidence interval (CI) for the trend and relative change are also estimated.

Source sector	Time series (1998-2014)	<sup>a</sup> Number of sites	<i>p</i> -value	Significant trend ( $p < 0.05$ )	<sup>b</sup> Median annual trend & [95% CI] ( $\mu\text{g NH}_3 \text{ y}^{-1}$ )	<sup>c</sup> Relative median change over the period & [95% CI] (%)
Cattle	Annual (MK)	17	0.46	no	0.0155 [-0.0150, 0.0300]	12 [-10, 24]
Cattle	Monthly (MK)	17	0.90	no	-0.0012 [-0.0192, 0.0168]	-0.9 [-14, 13]
Cattle	Monthly (SMK)	17	0.51	no	0.0043	3.9
Pigs&Poultry	Annual (MK)	9	0.02	yes	-0.0043 [-0.1008, -0.0071]	-22 [-42, -3.9]
Pigs&Poultry	Monthly (MK)	9	< 0.001	yes	-0.0648 [-0.0984, -0.0300]	-32 [-46, -16]
Pigs&Poultry	Monthly (SMK)	9	< 0.001	yes	-0.0588	-11
Sheep	Annual (MK)	4	0.17	no	0.0029 [0.0000, 0.0069]	16 [0.0, 46]
Sheep	Monthly (MK)	4	0.10	no	0.0036 [0.0000, 0.0072]	20 [0.0, 45]
Sheep	Monthly (SMK)	4	< 0.01	yes	0.0033	210
Background	Annual (MK)	5	0.20	no	0.0019 [-0.0012, 0.0038]	18 [-10, 41]
Background	Monthly (MK)	5	0.23	no	0.0012 [-0.0012, 0.0036]	13 [-11, 42]
Background	Monthly (SMK)	5	0.05	yes	0.0012	49

<sup>a</sup>Number of sites providing complete data runs over the period 1998 to 2014.

<sup>b</sup>Median annual trend = fitted Sen's slope of Mann-Kendall linear trend (unit =  $\mu\text{g NH}_3 \text{ y}^{-1}$ )

<sup>c</sup>Relative median change calculated based on the annual NH<sub>3</sub> concentration at the start ( $y_0$ ) and at the end ( $y_i$ ) of time series computed from the Sen's slope and intercept ( $=100 * [(y_i - y_0) / y_0]$ ).

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*Cattle sites:* Bickerton Hill (UKA00297), Brown Moss (UKA00369), Castle Cary (UKA00328), Cwmystwyth (UKA00325), Fenn's Moss (UKA00291), High Muffles (UKA00169), Hillsborough (UKA00293), Little Budworth (UKA00298), Llyncllys Common (UKA00270), Lough Navar (UKA00166), Myerscough (UKA00356), Northallerton (UKA00316), North Wyke (UKA00269), Penallt (UKA00324), Wardlow Hay Cop (UKA00119), Wem Moss (UKA00299), Yarnar Wood (UKA00168).

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*Pig & Poultry sites:* Beddingfield (UKA00334), Dennington (UKA00331), Dunwich Heath (UKA00308), Fressingfield (UKA00335), Mere Sands Wood (UKA00280), Redgrave + Lopham (UKA00311), Sibton (UKA00012), Stoke Ferry (UKA00317), Stanford (UKA00476).

*Sheep sites:* Glensaugh (UKA00348; 2005 classification = background, but 1km radius is predominantly sheep from local landuse information), Moorhouse (UKA00357) and Sourhope (UKA00347) (2015 classification = cattle, but 1km radius around site is sheep from local landuse information), (Shetland UKA00486).

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*Background sites:* Allt a Mharcaidh (UKA00086), Dumfries (UKA00368), Eskdalemuir (UKA00130), Inverpolly (UKA00457), Strathvaich (UKA00162).

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**Table 4: Summary of linear regression time series trend analysis on grouped NH<sub>3</sub> concentration data (annually averaged data and also monthly mean data) from the UK National Ammonia Monitoring Network (NAMN) for four different emission source sectors. The following are shown: the *p*-value, annual trend (fitted slope, in  $\mu\text{g NH}_3 \text{ y}^{-1}$ ), *R*<sup>2</sup>, and the relative change over the selected time period (in %).**

Source sector	Time series (1998-2014)	<sup>a</sup> Number of sites	<i>p</i> -value	Significant trend ( <i>p</i> <0.05)	<sup>b</sup> Annual Trend ( $\mu\text{g NH}_3 \text{ y}^{-1}$ )	<i>R</i> <sup>2</sup>	<sup>b</sup> Relative change over the period [%]
Cattle	annual	17	0.61	no	0.0049	0.0180	3.6
Cattle	monthly	17	0.84	no	0.0019	0.0002	1.4
Pigs&Poultry	annual	9	0.06	no	-0.0434	0.2143	-21
Pigs&Poultry	monthly	9	0.02	yes	-0.0466	0.0257	-22
Sheep	annual	4	0.09	no	0.0034	0.1751	19
Sheep	monthly	4	0.14	no	0.0032	0.0108	17
Background	annual	5	0.33	no	0.0014	0.0627	13
Background	monthly	5	0.39	no	0.0013	0.0037	12

5 <sup>a</sup>Number of sites providing complete data runs over the specified time period in analysis

<sup>b</sup>Annual trend = fitted slope of linear regression (unit =  $\mu\text{g NH}_3 \text{ y}^{-1}$ )

<sup>c</sup>Relative change calculated based on the estimated annual NH<sub>3</sub> concentration at the start (*y*<sub>0</sub>) and at the end (*y*<sub>1</sub>) of time series (=100\*[(*y*<sub>1</sub>-*y*<sub>0</sub>)/*y*<sub>0</sub>]) computed from the slope and intercept (=100\*[(*y*<sub>1</sub>-*y*<sub>0</sub>)/*y*<sub>0</sub>]).

10 *Cattle sites*: Bickerton Hill (UKA00297), Brown Moss (UKA00369), Castle Cary (UKA00328), Cwmystwyth (UKA00325), Fenn's Moss (UKA00291), High Muffles (UKA00169), Hillsborough (UKA00293), Little Budworth (UKA00298), Llyncllys Common (UKA00270), Lough Navar (UKA00166), Myerscough (UKA00356), Northallerton (UKA00316), North Wyke (UKA00269), Penalt (UKA00324), Wardlow Hay Cop (UKA00119), Wem Moss (UKA00299), Yarnor Wood (UKA00168).

15 *Pig & Poultry sites*: Bedlingfield (UKA00334), Dennington (UKA00331), Dunwich Heath (UKA00308), Fressingfield (UKA00335), Mere Sands Wood (UKA00280), Redgrave + Lopham (UKA00311), Sibton (UKA00012), Stoke Ferry (UKA00317), Stanford (UKA00476).

*Sheep sites*: Glensaugh (UKA00348; 2005 classification = background, but 1km radius is predominantly sheep from local landuse information), Moorhouse (UKA00357) and Sourhope (UKA00347) (2015 classification = cattle, but 1km radius around site is sheep from local landuse information), (Shetland UKA00486).

20 *Background sites*: Allt a Mharcaidh (UKA00086), Dumfries (UKA00368), Eskdalemuir (UKA00130), Inverpolly (UKA00457), Strathvaich (UKA00162).

**Table 5: Comparison of % change in estimated UK NH<sub>3</sub> emissions reported by the National Atmospheric Emission Inventory (NAEI) (data from: <http://naei.defra.gov.uk/>) with % change between 1998 and 2014 in annually averaged NH<sub>3</sub> concentration data from the UK National Ammonia Monitoring Network (NAMN) for all NAMN sites (dataset 1a) and for grouped sites in four different emission source sectors.**

Comparison period: 1998 - 2014	All sites (dataset 1a: n = 59)	Cattle (n=17)	Pigs & Poultry (n=9)	Sheep (n=4)	Background (n=5)
UK NH <sub>3</sub> emissions: % change relative to 1998	-16	-11	-39	-24	no data
UK NAMN NH <sub>3</sub> : % relative median change estimated from MK Sen's slope and intercept	-6.3 (see Table 1)	12 (see Table 3)	-22* (see Table 3)	15 (see Table 3)	17 (see Table 3)
UK NAMN NH <sub>3</sub> : % relative change estimated from linear regression slope and intercept	-3.1 (see Table 2)	3.6 (see Table 4)	-21 <sup>Δ</sup> (see Table 4)	19 (see Table 4)	13 (see table 4)

5 Significance: \* $p < 0.05$ , \*\* $p < 0.01$ , \*\*\* $p < 0.001$ , <sup>Δ</sup> $p = 0.06$ .

**Table 6: Comparison of % change in UK NH<sub>3</sub>, SO<sub>2</sub> and NO<sub>x</sub> emissions reported by the National Atmospheric Emission Inventory (NAEI) (data from: <http://naei.defra.gov.uk/>) with % change in annually averaged NH<sub>4</sub><sup>+</sup> and NH<sub>3</sub> concentration data from the UK National Ammonia Monitoring Network (NAMN) for sites with complete data runs of both NH<sub>4</sub><sup>+</sup> and NH<sub>3</sub> over the specified time periods.**

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	NH <sub>4</sub> <sup>+</sup> (23 sites) (1999-2014)	NH <sub>3</sub> (23 sites) (1999-2014)	NH <sub>4</sub> <sup>+</sup> (30 sites) (2006-2014)	NH <sub>3</sub> (30 sites) (2006-2014)
UK emissions: % change over the time period	-16 (NH <sub>3</sub> ), -75 (SO <sub>2</sub> ), -53 (NO <sub>x</sub> )		-7 (NH <sub>3</sub> ), -54 (SO <sub>2</sub> ), -39 (NO <sub>x</sub> )	
UK NAMN: % relative median change estimated from MK Sen's slope and intercept	-47**	3.0	-44**	-17
UK NAMN: % relative change estimated from linear regression slope and intercept	-49**	3.0	-43**	-18 <sup>Δ</sup>

Significance: \* $p < 0.05$ , \*\* $p < 0.01$ , \*\*\* $p < 0.001$ , <sup>Δ</sup> $p = 0.06$ .

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**Table 7: Comparison of % change in UK NH<sub>3</sub>, SO<sub>2</sub> and NO<sub>x</sub> emissions reported by the National Atmospheric Emission Inventory (NAEI) (data from: <http://naei.defra.gov.uk/>) with % change in annually averaged NH<sub>4</sub><sup>+</sup> concentration data from the UK National Ammonia Monitoring Network (NAMN) and SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> concentration data from the UK Acid Gas and Aerosol Network (AGANet) for sites with complete concurrent data runs over the specified time periods.**

	NH <sub>4</sub> <sup>+</sup> (12 sites) (2000-2014)	SO <sub>4</sub> <sup>2-</sup> (12 sites) (2000-2014)	NO <sub>3</sub> <sup>-</sup> (12 sites) (2000-2014)	NH <sub>4</sub> <sup>+</sup> (30 sites) (2006-2014)	SO <sub>4</sub> <sup>2-</sup> (30 sites) (2006-2014)	NO <sub>3</sub> <sup>-</sup> (30 sites) (2006-2014)
UK emissions: % change over the time period	-16 (NH <sub>3</sub> )	-75 (SO <sub>2</sub> )	-53 (NO <sub>x</sub> )	-7 (NH <sub>3</sub> )	-54 (SO <sub>2</sub> )	-39 (NO <sub>x</sub> )
UK NAMN: % relative median change estimated from MK Sen's slope and intercept	-56**	-63***	-46***	-44**	-45*	-35**
UK NAMN: % relative change estimated from linear regression slope and intercept	-58**	-65***	-45**	-43**	-46**	-33***

20 Significance: \* $p < 0.05$ , \*\* $p < 0.01$ , \*\*\* $p < 0.001$ .