Dear Editor:

We have carefully revised the manuscript and addressed all of the comments provided by the reviewer. The details can be found in our enclosed responses to the reviewer's comments. For your and the reviewer's convenience in reviewing the changes, a copy of the paper with track-changes is also attached below.

Thank you for taking care of the review process for this paper.

Sincerely,

Huanbo Wang and co-authors

### **Response to Referee #4**

We greatly appreciate the helpful comments from the reviewer, which have helped us improve the paper. We have addressed all of the comments carefully, as detailed below. Our responses start with "R:".

Sichuan Basin is one of the regions in China with severe aerosol pollution, while the chemical component analysis is still rare. This study fills the gap in this area. Under this condition, the short-term (four selected months) and low time resolution sampling is still acceptable. Thus, I suggest it published after major revision. In general, the manuscript is not well organized. Authors should put related results together to support one conclusion. The results now are presented with several short paragraphs which are not carefully ordered, e.g., section 3.1.2. From line 197 to line 234, it shows the results of SNA, SNA, carbonaceous aerosols, and SNA again. The readers can hardly find the sufficient information to know and understand the results. Presentation should be modified by native speaker before publication.

**R:** We have better organized sections 3.1.1, 3.1.2, 3.1.3, and 3.2.1 following this comment. Several short paragraphs in sections 3.1.1 and 3.1.2 were combined based on their contents. Six short paragraphs in section 3.1.3 were reorganized into four paragraphs following the order of similarity, difference (SNA and carbonaceous component) and correlation analysis between CD and CQ sampling site. Section 3.2.1 was also integrated according to the pollution episodes during the one-year sampling periods and key chemical compositions caused PM<sub>2.5</sub> pollution in winter.

We have rewritten section 3.1.2 following this order: we first summarized the seasonal variations of PM<sub>2.5</sub> concentrations and the impacts of the meteorological parameters on the seasonal trends. We then explored the seasonal trends of major components of PM<sub>2.5</sub> with discussions for the same chemical components grouped into one place, as recommended by the reviewer.

In addition, our discussion basically kept the same sequence for the major chemical components groups: we first discussed the fine soil due to their different seasonal trends and spatial variations at CD and CQ, then SNA and individual SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>; and finally, carbonaceous components including organic carbon and elemental carbon.

We have also carefully proofread the revised paper to minimize any grammar errors, and have asked a native speaker to have a final check of the paper.

There are many qualitative speculations in the article. Without robust evidence, to do such speculations is unreliable and may even mislead readers. Many conclusions are not based on strict logic.

Line 203: "The majority of PM<sub>2.5</sub> components showed a summer minimum, which was caused by high planetary boundary layer height favoring pollutants dispersion and abundant precipitation favoring wet scavenging." I think it will be more appropriate to show precipitation to support this statement.

**R:** We have added quantitative discussion wherever possible. For example, we have added seasonal precipitation amount in Table 1 and related discussions. The precipitation amount was the highest in summer at both CD and CQ, which supported our conclusion. We admit that there are still places with qualitative discussions due to the nature of the study, the lack of data or the limitation for the current knowledge on this topic.

Line 208 and 217. The statement could not only be supported by other references. Other mechanisms can also lead to similar or opposite results. More observed results are needed.

**R**: We have added the observed results such as gaseous precursors to explain the seasonal trends of major chemical components in  $PM_{2.5}$ . Detailed discussions are included in section 3.1.2 in the revised manuscript.

Line 237: "Although none of the two sites alone can represent the whole region of the Sichuan basin, the similarities in the characteristics of the major pollutants between the two sites should represent the regional-scale characteristics of urban-environment pollution while the differences between the two sites should reflect the sub-regional characteristics of urban pollution." As the authors response to previous suggestions, "The two monitoring sites selected in this study should represent the typical urban environment in their respective cities". Thus, many similarities reflect the characteristics of urban region instead of the whole region.

**R:** Yes, the selected sampling sites at CD and CQ represent the typical urban environment. We have changed the sentence "a moderate similarity was observed in autumn, winter and spring on the basis of low COD values (0.15-0.18), indicating the regional-scale PM<sub>2.5</sub> pollution pattern in Sichuan Basin" to "a moderate similarity was observed in autumn, winter and spring on the basis of low COD values (0.15-0.18), indicating the regional-scale PM<sub>2.5</sub> pollution pattern in Sichuan Basin" to "a moderate similarity was observed in autumn, winter and spring on the basis of low COD values (0.15-0.18), indicating limited differences between the two urban environments in the Sichuan Basin".

Line 403, the interaction between aerosol and ozone are quite complicated. Aerosols can provide an interface for the heterogeneous reaction of ozone products. Previous study shows that this mechanism usually contributes more than the impact through photochemical process. Fig S5 also could not support that "the formation of  $SO_4^{2-}$  during the polluted periods was dominated by heterogeneous aqueous processes rather than photochemical reactions".

**R:** Yes, we agree with the reviewer's comments and added the explanation on why high PM<sub>2.5</sub> concentrations associated with lower O<sub>3</sub> concentrations: On the one hand, aerosols are generally considered as a constraining factor to O<sub>3</sub> production due to their absorption and scattering of UV radiation, which reduce solar radiation and

consequently decrease photochemical activity. On the other hand, aerosols can provide an interface for the heterogeneous reaction, in accordance with O<sub>3</sub> consumption and secondary aerosol formation, which would result in decreasing O<sub>3</sub> concentrations and increasing PM<sub>2.5</sub> concentrations.

Fig. S5 could not give a solid evidence for the formation mechanism of SO4<sup>2-</sup>. However, it can be found that the ambient RH remained high at low O<sub>3</sub> concentrations, which was beneficial to form SO4<sup>2-</sup> through heterogeneous aqueous processes, resulting high SOR value at high RH accompanied with low O<sub>3</sub> concentrations.

Line 365, "Similar diurnal variations were also found between clean and polluted periods for CO (Fig. S4), suggesting no significant extra CO emission during polluted periods." Is that possible to give a quantitative value to support that the variations are similar? Also, I don't consider this can support no extra CO emission directly.

**R:** The diurnal variations of CO between clean and pollution periods were significant correlation with Pearson coefficient of 0.83 and 0.50 at both CD and CQ (p<0.05) based on statistical analysis. As suggested by the reviewer, it also could not support the statement that no extra CO emission occurred only according to the diurnal variations of CO. Thus, we deleted those explanations in the revised manuscript. Nevertheless, in practice scaling the concentrations of other pollutants to that of CO (or BC) was also an effective method to reduce the impact of atmospheric physical processes on other pollutants, which has been used widely in the previous study (Zhang et al., 2015; Zheng et al., 2015; Liggio et al., 2016; Hu et al., 2013), and we added those citations in the revised manuscript.

Zhang, Q., Quan, J. N., Tie, X. X., Li, X., Liu, Q., Gao, Y., and Zhao, D. L.: Effects of meteorology and secondary particle formation on visibility during heavy haze events in Beijing, China, Sci Total Environ, 502, 578-584, 10.1016/j.scitotenv.2014.09.079, 2015.

Zheng, G. J., Duan, F. K., Su, H., Ma, Y. L., Cheng, Y., Zheng, B., Zhang, Q., Huang, T., Kimoto, T., Chang, D., Poschl, U., Cheng, Y. F., and He, K. B.: Exploring the severe winter haze in Beijing: the impact of synoptic weather, regional transport and heterogeneous reactions, Atmos Chem Phys, 15, 2969-2983, 10.5194/acp-15-2969-2015, 2015b.

Liggio, J., Li, S.M., Hayden, K., Taha, Y.M., Stroud, C., Darlington, A., Drollette, B.D., Gordon, M., Lee, P., Liu, P., Leithead, A., Moussa, S.G., Wang, D., O'Brien, J., Mittermeier, R.L., Brook, J.R., Lu, G., Staebler, R.M., Han, Y.M., Tokarek, T.W., Osthoff, H.D., Makar, P.A., Zhang, J.H., Plata, D.L., Gentner, D.R., 2016. Oil sands operations as a large source of secondary organic aerosols. Nature 534, 91-94.

Hu, W.W., Hu, M., Yuan, B., Jimenez, J.L., Tang, Q., Peng, J.F., Hu, W., Shao, M., Wang, M., Zeng, L.M., Wu, Y.S., Gong, Z.H., Huang, X.F., He, L.Y., 2013. Insights on organic aerosol aging and the influence of coal combustion at a regional receptor site of central eastern China. Atmospheric Chemistry and Physics 13, 10095-10112.

Some speculations are even self-contradictory. Line 208: "It is also noted that the seasonal variations of NO<sub>3</sub> were much larger than those of SO<sub>4</sub> and NH<sub>4</sub>. This can be explained by the enhanced formation of NO<sub>3</sub> under high RH in winter, and volatility of NH<sub>4</sub>NO<sub>3</sub> in summer under high temperature condition." And Line 348: "In contrast, RH remained high during clean or polluted periods in the present study, suggesting that high RH might not be the driving force for the pollution episodes in Sichuan Basin." Also, it is not convinced to claim RH might not be the driving force supported by this. Many factors including meteorological conditions and emission are combined together to form polluted episodes.

**R:** Line 208 and line 348 is not self-contradictory. As shown in Table 1, the average ambient RH in winter was 12% higher than that in summer. Therefore, compared with that in summer, the increased ambient RH in winter would enhance the formation of  $NO_3^-$  through the heterogeneous aqueous processes.

However, the average ambient RH during clean period and pollution period in winter was 81.9% and 80.2% at CD, and 70.2% and 68.0% at CQ, respectively (Table S1). No significant variations of ambient RH between clean period and pollution period were observed (less than 3%). Thus, the high ambient RH during the whole sampling period in winter was beneficial to the NO<sub>3</sub><sup>-</sup> formation, but it might not be the primary cause of the dramatic increase of PM<sub>2.5</sub> concentrations during the pollution period.

The conclusions part are all qualitative even without a number.

**R:** We feel that Conclusion should be rewritten differently from the Abstract. In the Abstract quantitative statement summarizing major results should be presented while in the Conclusion section, results not covered in the abstract, uncertainty discussion, future research recommendations should be presented. However, we have added some quantitative statement following this comment.

Besides, I have some specific comments on the manuscript as follows: 1. Line 238, "regional" instead of "reginal".

## R: Corrected.

2. Line 453, It is better for understanding if "Neijiang, Zigong, Yibin and Luzhou" could be marked in Fig 1.

## **R:** We have added the mentioned cities in Fig 1.

3. The period should be stated in caption of Fig 6 and Fig 7.

**R:** We have added the clean period (CP) and pollution period (PP) in the caption of Fig 6 and Fig 7. At CD: CP, 6 January and 27 January-2 February 2015; PP, 7-26

January 2015. At CQ: CP, 6-7 January and 27 January-2 February 2015; PP, 8-26 January 2015.

4. Line 415, the ref should not be in brackets.

**R:** Corrected.

# Seasonal characteristics, formation mechanisms and source origins of PM<sub>2.5</sub> in two megacities in Sichuan Basin, China

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1	Abstract. To investigate the characteristics of PM <sub>2.5</sub> and its major chemical components, formation
2	mechanisms, and geographical origins in the two biggest citiesmegacities, Chengdu (CD) and
3	Chongqing (CQ), in Sichuan Basin <u>of southwest China</u> , daily PM2.5 samples were collected
4	simultaneously at one urban site in each city for four consecutive seasons from autumn 2014 to summer
5	2015. Annual mean concentrations of PM_{2.5} were 67.0 $\pm$ 43.4 and 70.9 $\pm$ 41.4 $\mu g$ m $^{-3}$ at CD and CQ,
6	respectively. Secondary inorganic aerosols (SNA) and organic matter (OM) accounted for 41.1% and
7	26.1% of PM <sub>2.5</sub> mass at CD, and 37.4% and 29.6% at CQ, respectively. Seasonal variations of PM <sub>2.5</sub> and
8	major chemical components were significant, usually with the highest mass concentration in winter and
9	the lowest in summer. Daily PM2.5 concentration exceeded the national air quality standard on 30% of
10	the sampling days at both sites, and most of the pollution events were at the regional scale within the
11	basin formed under stagnant meteorological conditions. The concentrations of carbonaceous
12	components were higher at CQ than CD, likely partially caused by emissions from the large amount of
13	motorcycles and spraying process during automobile production in CQ. Heterogeneous-
14	gas-phase reactions probably played an important role in the formation of SO4 <sup>2-</sup> , while both
15	homogeneous and heterogeneous reactions contributed to the formation of NO3 <sup>-</sup> . Geographical origins
16	of emissions sources contributing to high PM2.5 masses at both sites were identified to be mainly
17	distributed within the basin based on potential source contribution function (PSCF) analysis.
18	

## 19 **1 Introduction**

Fine particles (PM<sub>2.5</sub>, particulate matter with an aerodynamic diameter smaller than 2.5 µm) have 20 adverse effects on human health (Anderson et al., 2012;Lepeule et al., 2012;Taus et al., 2008), 21 deteriorate air quality (Zhang et al., 2008; Paraskevopoulou et al., 2015), reduce atmospheric visibility 22 (Fu et al., 2016;Cao et al., 2012;Baumer et al., 2008), impact climate (Ramanathan and Feng, 23 2009;Hitzenberger et al., 1999;Mahowald, 2011), and affect ecosystem (Larssen et al., 2006). In the past 24 two decades, China has experienced serious PM<sub>2.5</sub> pollution due to the rapidily increasing energy 25 consumption through econmic development, industrialization and urbanization (Tie and Cao, 2009). 26 The National Ambient Air Quality Standards (NAAQS) for PM2.5 was promulgated by the Chinese 27 government in 2012, and strict strategies have been implemented nationwide, e.g. controling SO<sub>2</sub> 28 emissions by installing desulphurization system in coal-fired power plants and conversion of fuel to 29 natural gas (Lu et al., 2011), mitigating NO<sub>x</sub> emissions through traffic restrictions, and reducing 30 biomass burning through straw shredding. Despite these efforts, there are still many cities that have not 31 yet met the current NAAQS (Tao et al., 2017). According to the '2013-2015 Reports on the State of 32 33 Environment of China', annual mean concentration of PM<sub>2.5</sub> in all the 74 major cities overacross China was 72, 64, and 50 µg m<sup>-3</sup> in 2013, 2014 and 2015, respectively, and only 4.1%, 12.2% and 22.5% of 34 the monitored cities met the NAAOS ( $35 \mu g m^{-3}$ ). 35

Previous studies showed that Beijing-Tianjin-Hebei area (BTH), Yangtze River Delta (YRD), Pearl 36 River Delta (PRD), and Sichuan Basin were the four main regions in China with severe aerosol 37 pollution (Tao et al., 2017). While many studies have been conducted in BTH, PRD and YRD regions 38 to understand the general characteristics of PM2.5 and its chemical components, formation mechanism, 39 and sources (Ji et al., 2016;Li et al., 2015;Quan et al., 2015;Tan et al., 2016;Yang et al., 2015;Zhang et 40 al., 2013;Zhao et al., 2015;Zhao et al., 2013a;Cheng et al., 2015;Zheng et al., 2015a;Yang et al., 2011a), 41 only a few studies have focused on the Sichuan Basin (Tao et al., 2014; Tian et al., 2013; Yang et al., 42 2011b). Covering an area of 260,000 km<sup>2</sup> and with a population of around 100 million, the Sichuan 43 Basin is the most populated basin in China. It is a subtropical expanse of low hills and plains and is 44 completely encircled by high mountains and plateaus. It is also characterized by persistently high 45 relative humidity and extremely low wind speeds all the year-round (Guo et al., 2016;Chen and Xie, 46 47 2013). The characteristics of  $PM_{2.5}$  in the Sichuan Basin are supposed to be very different from those in eastern coastal China (i.e. PRD and YRD) and North China Plain (i.e. BTH) due to the special 48

3

topography and meteorological conditions, besides emission sources, in the basin. Furthermore, the
terrain in the two megacities is also distinct from each other significantly, i.e., Chongqing is a
mountainous city lying on the eastern margin of the basin while Chengdu is a flat city on the western
margin of the basin. Therefore, there is a great interest in comparing the chemical components of PM<sub>2.5</sub>
and characterizing pollution episodes <u>betweenin</u> the two cities.

54 The present study aims to fill this gap by measuring chemically-resolved PM<sub>2.5</sub> in Chengdu and Chongqing in four consecutive seasons during 2014-2015. The main objectives are to: (1) characterize 55 PM<sub>2.5</sub> mass and major chemical components in urban environemntss of Chengdu and Chongqing; (2) 56 compare PM<sub>2.5</sub> chemical compositions under different pollution levels and identify major chemical 57 components responsible for long-lasting PM2.5 pollution episodes in winter; (3) explore the possible 58 formation mechanism of the secondary aerosols; and (4) reveal the geographical source regions 59 contibuting to the high PM<sub>2.5</sub> levels through potential source contribution function (PSCF) analysis. 60 Knowledge gained in this study provides scientific basis for making future emision control plocies 61 aiming to allivating heavy PM<sub>2.5</sub> pollution in this unique basin. 62

# 63 **2 Methodology**

# 64 **2.1 Sampling sites**

PM<sub>2.5</sub> samples were collected at two urban sites, one in Chengdu and another in Chongqing, the two 65 largest cities in Sichuan Basin, southwest China. The two sampling sites are located 260 km apart (Fig. 66 1). The sampling site in Chengdu (CD) is located on the roof of a sixth floor building in the Sichuan 67 Academy of Environmental Science (104°4' E, 30°37' N) with no large surrounding industries but heavy 68 traffic. The closest main road (Renmin South road of Chengdu) is about 20 m east of the samling site. 69 The sampling site in Chongqing (CQ) is located on the rooftop of Chongqing Monitoring Center 70 (106°30' E, 29°37' N). The highway G50 is 250 m away from this sampling site. The two selected 71 sampling sites are considered to represent typical urban environments in their respective cities (Tao et 72 al., 2014;Chen et al., 2017). 73

#### 74 **2.2 Sample collection**

75 Daily (23-h) integrated PM<sub>2.5</sub> samples were collected in four months, each in a different season: autumn

- 76 (23 October to 18 November, 2014), winter (6 January to 2 February, 2015), spring (2 to 29 April,
- 2015), and summer (2 to 30 July, 2015). At both sites, PM<sub>2.5</sub> samples were collected in parallel on
- 78 Teflon filters (Whatman Corp., 47 mm) and quartz filters (Whatman Corp., 47 mm). At CD site, PM<sub>2.5</sub>

sampling was carried out using a versatile air pollutant sampler (Wang et al., 2017). One channel was 79 used to load PM<sub>2.5</sub> sample on Teflon filter for mass and trace elements anlysis and the other one was 80 equipped with quatz filter for water-soluble inorganic ions and carbonaceous components analysis. The 81 sampler was running at 15 L min<sup>-1</sup> for each channel. At CQ site, a low-volume aerosol sampler (BGI 82 Corp., frmOmni, USA) operating at a flow rate of 5 L min<sup>-1</sup> was used to collect PM<sub>2.5</sub> samples on 83 Teflon filter, and another sampler (Thermo Scientific Corp. Partisol 2000i, USA) with a flow rate of 84 16.7 L min<sup>-1</sup> was used to collect PM<sub>2.5</sub> samples on quartz filter. A total of 112 samples and 8 field 85 blanks, nearly equally distributed in the four seasons, were collected at each site during the campaign. 86 In addition, three lab blank filters in each campaign were stored in a clean Petri slides in the dark and 87 analysed in the same ways as the collected samples to evaluate the background contamination. 88

Before sampling, all the quartz filters were preheated at 450°C for 4 h to remove the organic
compounds. All sampled filters were stored in clean Petri slides in the dark and at -18°C until analysis
to prevent the evaporation of volatile compounds. Before and after sample collection, all the Teflon
filters were weighted at least three times using an microbalance (Sartorius, ME 5-F, Germany) after
their stabilization for 48 h under controlled conditions (temperature: 20~23°C, relative humidity:
45~50%). Differences among replicate weights were mostly less than 15 µg for each sample.

#### 95 **2.3 Chemical analysis**

For the analysis of water-soluble inorganic ions, a quarter of each quartz filter was first extracted using 96 ultrapure water in an ultrasonic bath for 30 min, and then filtered through a 0.45 µm pore syringe filter. 97 Anions (SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and Cl<sup>-</sup>) and cations (Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup> and Ca<sup>2+</sup>) were determined using ion 98 chromatograph (Dionex Corp., Dionex 600, USA). Anions were separated using AS11-HC column with 99 30 mM KOH as an eluent at a flow rate of 1.0 ml min<sup>-1</sup>. Cations were determined using CS12A column 100 with 20 mM MSA (methanesulfonic acid) at a flow rate of 1.0 ml min<sup>-1</sup>. Individual standard solutions of 101 all investigated anions and cations (1000 mg L<sup>-1</sup>, o2si, USA) were diluted to construct the calibration 102 curves. The correlation coefficients of the linear regression of the standard curves were all above 0.999. 103 Field blanks were prepared and analyzed together with the samples and then subtracted from the 104 105 samples. The concentrations of the water-soluble inorganic ions in the field blanks were in the range of 106 0.008-0.13 µg m<sup>-3</sup>. The relative standard deviation of each ion was better than 8% for the reproducibility 107 test.

108 Organic carbon (OC) and elemental carbon (EC) were measured by thermal-optical reflectance

(TOR) method using a DRI OC/EC analyzer (Atmoslytic Inc., USA). The methodology for OC/EC
analysis was based on TOR method as described in Chow et al. (2007). For calibration and quality
control, measurement with filter blank, standard sucrose solution and replicate analysis were performed.
Blank corrections were performed by subtracting the blank values from the sampled ones. The
concentration of EC in field blanks was zero while OC was below 0.7 µg C cm<sup>-2</sup>. The repeatability was
better than 15%.

115 The elements including Al, Si, Ca, Fe, and Ti were analyzed on Teflon filter using X-ray fluorescence analyzer (Epsilon 5ED-XRF, PAN'alytical Corp., Netherlands), the QA/QC procedures of 116 the XRF analysis have been described in Cao et al. (2012). The gaseous species were continuously 117 measured by a set of online gas analyzers, including EC9850 SO<sub>2</sub> analyzer, 9841 NO/NO<sub>2</sub>/NO<sub>x</sub> 118 analyzer, 9830 CO analyzer, and 9810 O<sub>3</sub> analyzer (Ecotech Corp., Australia) at CD, and Thermo 42i 119 NO/NO<sub>2</sub>/NO<sub>x</sub> analyzer, 43i SO<sub>2</sub> analyzer, 48i CO analyzer, and 49i O<sub>3</sub> analyzer (Thermo Scientific 120 121 Corp., USA) at CQ. The mass concentrations of PM2.5 was were automatically measured by online particulate monitor instruments (BAM1020, Met one Corp., USA, at CD and 5030 SHARP, Thermo 122 123 Scientific Corp, USA, at CQ). Hourly meteorological parameters, including ambient temperature (T), relative humidity (RH), wind speed (WS) and direction, barometric pressure (P), and solar radiation 124 (SR) were obtained from an automatic weather station (Lufft Corp. WS501, Germany) at each site. 125 Hourly precipitation data were recorded at the nearest weather station operated by China Meteorology 126 127 Administration (http://www.weather.com.cn/). Planetary boundary layer height (PBLH) was obtained from the HYSPLIT model (http://ready.arl.noaa.gov/HYSPLIT.php). 128

# 129 **2.4 Data analysis**

- 130 The EC-tracer method has been widely used to estimate SOC (Turpin and Lim, 2001;Castro et al.,
- 131 1999), which can be expressed as
- 132  $POC=(OC/EC)_{prim} \times EC$  (1)
- 133 SOC=OC-POC

Where POC, SOC and OC represent <u>the</u> estimated primary OC, secondary OC and measured total OC, respectively. (OC/EC)<sub>min</sub> was simplified as the (OC/EC)<sub>prim</sub> to estimate SOC in this study. (OC/EC)<sub>min</sub> was 2.4, 2.6, 1.6 and 2.2 in autumn, winter, spring and summer at CD, respectively, and 1.9, 2.8, 1.1 and 1.5 at CQ. The estimated SOC was only an approximation with uncertainties, e.g., from influence of biomass burning (Ding et al., 2012).

(2)

139 The coefficient of divergence (COD) has been used to evaluate the spatial similarity of chemical 140 compositions at different sites (Wongphatarakul et al., 1998;Qu et al., 2015), which is defined as

141 
$$COD_{jk} = \sqrt{\frac{1}{p} \sum_{1}^{p} (\frac{x_{ij} - x_{ik}}{x_{ij} + x_{ik}})^2}$$
 (3)

142 Where  $x_{ij}$  and  $x_{ik}$  represent the average concentration for a chemical component *i* at site *j* and *k*,

respectively, *p* is the number of chemical components. Generally, a COD value lower than 0.2 indicates

144 a relatively similarity of spatial distribution.

# 145 2.5 Geographical origins of PM<sub>2.5</sub>

146 72-h air mass back trajectories were generated based on the Hybrid Single Particle Lagrangian

147 Integrated Trajectory (HYSPLIT) model using 0.5°×0.5° meteorological data for <u>the period of PM<sub>2.5</sub></u>

148 measurements from October 2014 to July 2015 when PM<sub>2.5</sub> measurements were made at both sites. Four

trajectories at 04:00, 10:00, 16:00, and 22:00 UTC every day with the starting height of 300 m aboveground level were calculated (Squizzato and Masiol, 2015).

PSCF is substantially a conditional probability that trajectories with pollutant concentrations larger than a given criterion passed through a grid cell (i,j) (Ashbaugh et al., 1985;Polissar et al., 1999), that means a grill cell (i,j) with high PSCF values are mostly potential source locations of pollutants. PSCF is defined as follows,

155 
$$PSCF_{ij} = \frac{m_{ij}}{n_{ij}}$$
(4)

Where  $n_{ij}$  is the total number of endpoints falling in the grid cell (i,j) and  $m_{ij}$  denotes the number of endpoints that are associated with samples exceeding the threshold criterion in the same cell. To reduce the PSCF uncertainties associated with small  $n_{ij}$  values, weighting function was adopted as follows,

159 
$$W_{ij} = \begin{cases} 1.0 & 3n_{ave} < n_{ij} \\ 0.7 & 1.5n_{ave} < n_{ij} \le 3n_{ave} \\ 0.42 & n_{ave} < n_{ij} \le 1.5n_{ave} \\ 0.2 & n_{ij} \le n_{ave} \end{cases}$$
(5)

160 Where  $n_{ave}$  is the average number of endpoints in each grid cell.

161 The trajectories coupled with daily pollutants concentrations were used for PSCF analysis, with the 162 threshold criterion in PSCF analysis being set at the upper 50% of PM<sub>2.5</sub> and other pollutants. The 163 trajectory covered area was in the range of 20-45° N and 90-120° E and divided into  $0.5^{\circ} \times 0.5^{\circ}$  grid cells. 164 **3 Results and discussion** 

## 165 **3.1 PM<sub>2.5</sub> mass concentration and chemical composition**

## 166 **3.1.1 Overview**

Table 1 presents seasonal and annual mean concentrations of PM2.5 and its major chemical components 167 at CD and CQ during the sampling periods. Daily PM<sub>2.5</sub> ranged from 11.6 to 224.7 µg m<sup>-3</sup> with annual 168 average being 67.0  $\pm$  43.4 µg m<sup>-3</sup> at CD and 70.9  $\pm$  41.4 µg m<sup>-3</sup> at CQ, which were about two times the 169 NAAOS annual limit. Secondary inorganic aerosol (SNA, the sum of SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>) and 170 carbonaceous species together represented more than 70% of PM<sub>2.5</sub> mass at both sites (Fig. 2). The 171 annual mean concentrations of SNA were 27.6 µg m<sup>-3</sup> at CD and 26.5 µg m<sup>-3</sup> at CO, contributing 41.1% 172 and 37.4% to PM<sub>2.5</sub> mass, respectively.  $SO_4^{2-}$ ,  $NO_3^{-}$  and  $NH_4^{+}$  accounted for 16.8%, 13.6% and 10.8%, 173 respectively, of PM<sub>2.5</sub> mass at CD, and 17.2%, 10.9% and 9.2%, respectively, at CO. Organic matters 174 (OM), estimated from OC using a conversion factor of 1.6 to account for other elements presented in 175 organic compounds (Turpin and Lim, 2001), were the most abundant species in PM<sub>2.5</sub>, accounting for 176 177 26.1% and 29.6% of PM<sub>2.5</sub> mass at CD and CQ, respectively. In contrast, EC only comprised of around 178 6% at both sites. The annual mean concentrations of OC and EC were 20% and 25%, respectively, and were higher at CQ than CD. 179

180 The annual mean concentration of fine soil (FS), calculated by summing the oxides of major crustal

elements, i.e., Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, CaO, FeO, Fe<sub>2</sub>O<sub>3</sub>, and TiO<sub>2</sub> (Huang et al., 2014), was 6.7  $\mu$ g m<sup>-3</sup> (9.5% of

182 PM<sub>2.5</sub> mass) at CQ. It is noted that this was about two times that at CD ( $3.8 \mu g m^{-3}$ , 5.7% of PM<sub>2.5</sub>

mass). The minor components such as  $K^+$  and  $Cl^-$  constituted less than 5% of PM<sub>2.5</sub>. The unaccounted

portions of PM<sub>2.5</sub> reached 18.3% at CD and 15.3% at CQ, which were likely related to the uncertainties

in the multiplication factors used for estimating OM and FS, other unidentified species, and

186 measurement uncertainties.

#### 187 **3.1.2 Seasonal variations**

Figure 3 shows the seasonal variations in mass concentrations of  $PM_{2.5}$  and its major chemical components at CD and CQ. Seasonal variations of any pollutants should bewere influenced by the seasonal variations in source emission intensities, atmospheric processes and meteorological conditions. Unlike in northern China, there were no extensive coal combustion or wood burning for domestic heating in winter due to the warm temperature (around 10°C on average) in the Sichuan Basin, hence atmospheric processes and meteorological conditions played a-vital roles in the seasonal variations of PM<sub>2.5</sub>. On a seasonal basis, PM<sub>2.5</sub> mass was <u>the highest in winter at both CD and CQ</u>, which was 1.8-2.5 times of those
in the other seasons. In contrast, its seasonal differences among the other three seasons were generally
small, i.e., less than 40%. Stagnant air conditions with frequent calm winds and low planetary boundary
layer heights were was the major causes of the highest PM<sub>2.5</sub> mass in winter (Table 1) (Liao et al.,
2017;Chen and Xie, 2013;Li et al., 2017b).

All the major PM<sub>2.5</sub> components except FS followed the seasonal pattern of PM<sub>2.5</sub> mass with subtle differences. The highest FS concentrations were observed in spring at both sites. The relatively high wind speed and lower RH in spring were conducive for re-suspension of crustal dust and resulted in higher FS concentrations. In addition, frequent spring dust storms originated in the northwestern China was able to reach Sichuan Basin via long-rang transport, and cause<u>d</u> the elevated FS concentrations (Chen et al., 2015;Tao et al., 2013). <u>The highest contributions from FS to PM<sub>2.5</sub> mass was more than</u> <u>10%, appeared in spring at both sites.</u>

The majority of PM<sub>2.5</sub> components showed a summer minimum, which was caused by high 206 207 planetary boundary layer height favoring pollutants dispersion and abundant precipitation favoring wet scavenging (Table 1). In contrastHowever, the lowest seasonal average concentrations did not appear in-208 summer for One exception was  $SO_4^{2-}$ , which had a minimum in spring at CD and in autumn at CO, 209 likely due to the enhanced photochemical reactions associated with high temperature and strong solar 210 211 radiation in summer. High O<sub>3</sub> concentrations in summer also supported this seasonal trend. It is also noted that the seasonal variations of  $NO_3^-$  were much larger than those of  $SO_4^{2-}$  and  $NH_4^+$ . <u>SO4<sup>2-</sup> and</u> 212 SO<sub>2</sub> showed similar seasonal trends, with their concentrations 1.4-2.0 times higher in winter than in the 213 other seasons (Table 1). In contrast, the seasonal variations of NO<sub>3</sub><sup>-</sup> were much larger than that of NO<sub>2</sub>, 214 e.g., while the concentrations of NO<sub>2</sub> were 1.2-1.6 times higher in winter than in the other seasons, 215 those of NO<sub>3</sub><sup>-</sup> were 9.6 times higher in winter than in summer at CO. Thus, seasonal variations of SO<sub>2</sub> 216 217 and NO<sub>x</sub> emissions were comparable, but the atmospheric chemical processes caused the much larger seasonal variations in NO<sub>3</sub><sup>-</sup>. The concentration of NO<sub>3</sub><sup>-</sup> could be enhanced in winter under high RH 218 through heterogeneous aqueous processes and decreased in summer due to volatility of NH4NO3 under 219 220 high temperature, which increased the seasonal differences in NO<sub>3</sub><sup>-</sup> concentrations between winter and summer (Pathak et al., 2009; Quan et al., 2015; Squizzato et al., 2013). In addition, thermodynamically 221 222 driven behavior of NH<sub>4</sub>NO<sub>3</sub> was another factor for the lower NO<sub>3</sub><sup>-</sup> concentrations in summer (Wang et al., 2016; Kuprov et al., 2014). As shown in Fig. 2, <u>Tthe seasonal average contributions of SNA to PM2.5</u> 223

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only varied within a small range from 39.5% to 43.2% at CD, whereas in a relatively larger range from
 31.0% in summer to 37.1-41.5% in the other seasons at CQ. The smaller contribution in summer at CQ
 was mainly due to the lower NO<sub>3</sub><sup>-</sup> concentrations. At both CD and CQ, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> showed the
 highest contributions in winter and the lowest ones in summer, whereas an opposite trend was found for
 SO<sub>4</sub><sup>2-</sup>.

Both OC and EC exhibited the highest concentrations in winter at CD and CQ, around 1.9-3.1 times 229 230 of those in the other seasons whereas seasonal differences of those carbonaceous components were lessdistinct in other seasons, e.g. the variations of OC and EC among the other three seasons were less than-231 30%. Seasonal average SOC was unexpectedly also the highest in winter at both sites, similarly to what 232 observed for OC, which. This seasonal trends were can be partly explained by the enhanced 233 condensation process forming SOC under low temperature- that different from the anticipated high-234 value in summer in consideration of strong photochemical reaction. Condensation of semi-volatile-235 organic aerosols in winter seemed to play a larger role than photochemical reaction in summer, knowing 236 that low temperature in winter favors favored condensation process and was beneficial to form SOC-237 238 (Sahu et al., 2011;Cesari et al., 2016). Although high O3 and strong solar radiation condition in summerwas conducive to strong photochemical reactions, In contrast, high temperature in summer favored gas-239 particle partitioning in the gas phase and thus limited the formationincrease of SOC (Strader et al., 240 1999). Seasonal average relative contributions of major chemical components to PM2.5 are depicted in-241 Fig. 2. The seasonal average contributions of SNA to PM2.5 only varied within a small range from-242 39.5% to 43.2% at CD, whereas in a relatively larger range from 31.0% in summer to 37.1-41.5% in-243 other seasons at CQ. The smaller contribution in summer at CQ was mainly due to the lower NO2-244 concentrations. At both CD and CO, NO<sub>4</sub><sup>±</sup> and NH<sub>4</sub><sup>±</sup> showed the highest contributions in winter and the 245 <del>lowest ones in summer, whereas an opposite trend was found for  $SO_4^2$ .</del> The contributions of 246 carbonaceous components (the sum of OC and EC) generally followed the seasonal patterns of SNA, 247 accounting for 26.7-38.8% of PM2.5 mass. Among these, OM showed the lowest fractions in PM2.5 in 248 249 spring (21.1%) at CD and the highest value in winter (33.6%) at CQ, while the percentages of OM in other seasons were similar at both sites, around 27%. The seasonal variations of EC fractions were not 250 251 obvious, with a slightly higher value in spring. The highest contributions from FS was more than 10%, appeared in spring at both sites. 252

### 253 **3.1.3 Similarities and differences between the two sites**

254 Although none of the two sites alone can represent the whole region of the Sichuan basinBasin, the 255 similarities in the characteristics of the major pollutants between the two sites should represent the 256 regional-scale characteristics of urban-environment pollution while the differences between the two sites should reflect the sub-regional characteristics of urban pollution. A comparison between the two sites in 257 258 terms of seasonal-average concentrations of major chemical components is shown in Fig. 4 and discussed 259 in detail below. Despite the 260 km distance between the two sampling sites, a moderate similarity was 260 observed in autumn, winter and spring on the basis of low COD values (0.15-0.18), indicating the regional-scale PM<sub>2.5</sub> pollution pattern there was nolimited significant regional differences between the 261 two urban environments in the Sichuan Basin and the similarities in major emission sources for both sites. 262 263 The regional similar pollution patterns observed at both CD and CQ was-were likely to be related to the similar meteorological parameters and special topography of the basin, which is a closed lowland 264 surrounded by high mountains on all sides (Fig. 1). The mean elevation in the basin is about 200-700 m, 265 while the surrounded mountains are around a scope of 1000-3000 m elevation. The Tibetan Plateau lies 266 267 close to the western Sichuan Basin, with an elevation above 4000 m. Such a Plateau-Basin topography forms a barrier for the dispersion of pollutants and causes air stagnation within the basin, thereby 268 facilitating regional scale pollution events in the Basinbasin. 72-h air masses back trajectoryies analysis 269 (18:00 local time) showed that air masses reaching at CD and CQ mainly originated from local areas in 270 271 the basin (Fig. S1), confirming the influence of the high mountainous surrounding the basin. These results were consistent with those found in earlier studies in Chengdu and Chongqing (Tian et al., 2017;Liao et 272 al., 2017), which suggested that air masses had short-range trajectories and primarily originated from 273 inside the Sichuan Basin, highlighting the impacts of the special topography on PM<sub>2.5</sub> pollution. A similar 274 275 case has also been found elsewhere, such as in Po Valley, Italy (Ricciardelli et al., 2017). Po Valley is bordered by the Alps in the north and Apennines in the south, which caused a strong homogeneity in 276 277 PM<sub>2.5</sub> pollution at the regional scale.

It is worth to note that the COD values used to identify the similarities or differences of <u>the</u> two sites were calculated based on seasonal-average concentrations of all <u>the</u> components in PM<sub>2.5</sub>. However, if focusing on individual components, several chemical species in PM<sub>2.5</sub> differed by up to a

factor of 2.5 in their season-average concentrations between CD and CQ, e.g. OC and EC in winter and spring, and Cl<sup>-</sup> and FS in all the four seasons. In summer, the differences for several major chemical

components (FS, OC,  $SO_4^{2-}$ ,  $NO_3^{--}$  and EC) between the two sites were larger than in the other seasons, 283 284 causing a high COD value (0.33). These discrepancies were partly caused by the different atmospheric 285 chemical processes, local sources and meteorological parameters between the two sites. Specifically, FS mostly deviated from the 1:1 straight line in all the seasons, with substantially higher concentrations at 286 CQ than CD (Fig. 4). The main cause of this difference was the higher wind speeds and lower RH at CQ 287 than CD. There was no significant difference in NH<sub>4</sub><sup>+</sup> concentrations between CD and CQ, but 288 considerable differences in SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> in spring and summer. SO<sub>2</sub> concentration was around 25% 289 higher at CQ than CD in spring and summer, which partially explained the site-differences in  $SO_4^2$ . In 290 contrast, NO<sub>2</sub> concentration was comparable at both sites in summer, but NO<sub>3</sub><sup>-</sup> concentration was 58% 291 lower in CQ than CD. The site-differences in NO3<sup>-</sup> concentration was caused by NH4NO3 292 thermodynamic equilibrium controlled by ambient temperature and RH, instead of by its gaseous 293 precursors. The equilibrium would be shifted toward the particulate phase when ambient RH was above 294 the deliquescence relative humidity (DRH) of NH4NO3, and the dissociation constant decreased by 295 296 about one order of magnitude when RH was above 75% (Kuprov et al., 2014). DRH was calculated 297 from ambient RH temperature following Mozurkewich (1993). As shown in Table 1, the average temperature was comparable at CD and CQ during the summer period, hence leading to similar DRH 298 values of NH4NO3, ranging from 59% to 64% with an average value of 60.7%. However, the ambient 299 RH was substantially lower at CQ (61%) than CD (72%), causing lower NO<sub>3</sub><sup>-</sup> concentration at CQ. As 300 shown in Fig. S2, 53% of the hourly data in summer having ambient RH lower than DRH at CQ, while 301 only 19% such data at CD, which explained the different NO<sub>3</sub><sup>-</sup> concentrations between CD and CQ. 302

Fig.4 shows higher concentrations of carbonaceous component (OC and EC) at CQ than CD in all 303 804 the seasons except OC in autumn and EC in winter. OC and EC mainly originate from fossil fuel combustion and biomass burning. K<sup>+</sup> is usually regarded as a tracer of biomass burning (Tao et al., 305 2016). During the sampling campaign, no significant differences in K<sup>+</sup> levels were observed between 306 CD and CQ (Table 1), suggesting that biomass burning was not be the major cause of the higher 307 concentrations of carbonaceous component at CQ. Motorcycle traffic was likely a major source of 308 309 volatile organic compounds (VOCs) in CQ since it is a famous mountain city where most people use B10 motorcycle as daily traffic tools. According to the Chongqing Statistical Yearbook 2015, the The number of motorcycles was 2.0 million (among the total of 2.3 million motor vehicles) in Chongging in B11 2014, which was much higher than those (0.7 million) in Chengdu (Chongqing and Chengdu Statistical B12

Yearbook 2015). Moreover, Chongqing has become China's largest automobile production base, which likely also emit VOCs from the spraying process. Higher concentrations of VOCs in CQ would cause higher concentrations of secondary organic carbon via photochemical reaction under high temperature or vapor condensation under low temperature. This hypothesis is supported by the large differences in OC concentrations in winter between the two sites.

Correlation analysis may also provide an insight into the similarities/differences between the two sites over an intensive sampling period. Good correlations between the two sites were found for daily SNA, OC, EC and K<sup>+</sup> concentrations in autumn, winter and spring (Table <u>2S1</u>). However, for NO<sub>3</sub><sup>-</sup>, a significant correlation was identified only in autumn, likely due to the strong impact of local vehicle emissions and the subsequent atmospheric processes forming NO<sub>3</sub><sup>-</sup>. Similarly, a moderate correlation was observed just in winter for both Cl<sup>-</sup> and FS. In summer, weak or no correlations were identified between the two sites for almost all chemical components.

## 325 **3.2** PM<sub>2.5</sub> formation mechanisms and geographical origins

## 326 **3.2.1** Pollution episodes and key chemical components

327 Pollution episodes during the campaign are highlighted with shaded areas in Fig. 5. These pollution periods (PP) were defined as daily PM<sub>2.5</sub> concentration being above the NAAOS guideline value of 75 328  $\mu$ g m<sup>-3</sup>. Similarly, the days with PM<sub>2.5</sub> concentration below 75  $\mu$ g m<sup>-3</sup> were characterized to be clean 329 periods (CP). A total of seven pollution episodes were identified during the campaign at each site. There 330 331 were three long-lasting pollution episodes occurred simultaneously at the two sites on 23-27 October 832 2014, 7(8)-26 January, and 26-28 (29) April 2015. A total of 34 and 31 polluted-pollution days were counted at CD and CQ site, respectively, accounting for 30.4% and 28.6% of the entire sampling days 333 834 (112 days). The number of polluted pollution days at CD was 8, 21, 4 and 1 in autumn, winter, spring 335 and summer, accounting for 29.6%, 75%, 14.3% and 3.4% of the total sampling days in each season, respectively, and at CQ they were 4, 19, 6 and 2, accounting for 14.8%, 67.9%, 21.4% and 6.9%. 336 837 Stagnant atmosphere and high relative humidityRH were important factors causing PM<sub>2.5</sub> pollution 338 events, as was found in this and earlier studies (Zheng et al., 2015b;Chen et al., 2017;Liao et al., 2017). 339 Compared with the clean periods, the pollution periods were usually characterized by low planetary 840 boundary layer height and weak wind speed (Table <u>S1S2</u>), which suppressed pollutants dispersion vertically and horizontally. Temperature increased during the long-lasting pollution episodes, which 341 342 promoted gas-to-particle transformation, forming secondary aerosols. RH remained high (68-88%)

during pollution episodes (except in spring at CQ), although not much different from clean periods,
which was also conducive for aqueous-phase reactions converting gaseous pollutants into aerosols
(Chen et al., 2017;Tian et al., 2017).

Looking more closely at a regional-scale long-lasting pollution episode in winter, from 8 January to 346 26 January 2015, the concentrations of PM2.5 and major chemical components increased dramatically 347 compared with clean periods (Fig. 6). PM<sub>2.5</sub> concentrations were more than three times higher at both 348 349 sites, with the two dominant groups of components, SNA and OC, being 2.5-2.8 times higher at CD and 1.7-2.7 times higher at CQ. The enhancement of SNA and OC during pollution periods were similar at 350 CD, but OC increased much more than SNA at CQ, indicating some different contributing factors to the 351 high PM2.5 pollution at the two sites. Pollutants accumulation under stagnant meteorological conditions 352 might be a main factor at CD based on the similar magnitudes of the enhancements of PM<sub>2.5</sub> and its 353 dominant components, while additional processes should have increased OC more than other 354 components at CQ. The percentage contributions of SNA to PM<sub>2.5</sub> were similar during clean and 355 B56 polluted pollution periods, 38-41% at CD and CQ (Fig. S3). However, the percentage contributions of B57 OM to PM<sub>2.5</sub> decreased from 30.1% on clean days to 27.5% on polluted pollution days at CD, and increased from 26.9% to 34.9% at CO. Concentrations of the individual SNA species (SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and 358 NH4<sup>+</sup>) increased by a factor of 2.5-3.3 on polluted pollution days compared with clean days in all the 859 cases (Fig. 6). But the percentage contributions differed among the species as NO<sub>3</sub><sup>-</sup> increased and SO<sub>4</sub><sup>2-</sup> 360 361 decreased on polluted pollution days. The percentage contributions of SNA and OM to PM<sub>2.5</sub> discussed 362 above were different from those found in eastern coastal China and North China Plain, where 863 considerable increases were found for SNA and decreases for OM on polluted pollution days than clean days (Tan et al., 2009; Wang et al., 2015a; Quan et al., 2014; Zhang et al., 2015; Zhang et al., 2016; Cheng 364 365 et al., 2015). The polluted pollution periods in eastern coastal China and North China Plain were 366 accompanied with sharp increases of RH, which would promote aqueous-phase formation of secondary inorganic aerosols and resulted in rapid elevation of SO4<sup>2-</sup> and NO3<sup>-</sup> concentrations (Zheng et al., 367 2015b;Zheng et al., 2015a;Zhao et al., 2013b;Li et al., 2017a). In contrast, RH remained high during 868 869 both clean or polluted pollution periods in the present study, suggesting that high RH might not be the 870 primary cause of the dramatic increase inof PM2.5 concentrations during the pollution period the driving force for the pollution episodes in the Sichuan Basin. Another point that need to be mentioned is that, as 871 872 shown in Fig. S1, local sources were the main contributors to the pollution episodes in the Sichuan

basin-Basin while sources outside local regions frequently contributed to pollution episodes in eastern
coastal China and North China Plain through long/medium range transport (Gao et al., 2015;Hua et al.,
2015;Wang et al., 2015b).

## **376 3.2.2 Transformation mechanisms of secondary aerosols**

377 In most cases, meteorological conditions, atmospheric chemical processes and long-range transport are 878 all responsible for PM<sub>2.5</sub> accumulation (Zheng et al., 2015b). CO and EC areis directly emitted from 879 combustion processes and areis not very reactive., thus, their Its concentrations in the air are strongly 880 controlled by meteorological parameters within a relatively short period, which make itthem a good 381 tracer that can be used for separating different dominant factors contributing to pollutants accumulation 882 (Zheng et al., 2015b;Zhang et al., 2015;Hu et al., 2013;Liggio et al., 2016). CO concentration increased by a factor of 1.8 at CD and 1.5 at CO during polluted periods than clean periods. In comparison, other-883 chemical species (except NO<sub>2</sub>) increased by a factor of 1.8-3.3. A similar contrast between CO and 884 885 other pollutants was also seen in . Similar diurnal variations were also found between clean and polluted periods for CO (Fig. S4), suggesting no significant extra CO emission during polluted periods. Thus, the 886 B87 increased concentrations of CO during polluted periods were primarily driven by meteorologicalconditions. Therefore, tThe impact of atmospheric physical processes on other pollutants can be 888 revealed<del>reduced</del> by scaling the concentrations of other pollutants to that of CO, meaning that the impact 889 of chemical reactions can then be captured based on the ratio of other pollutants to CO concentrations. 890 For example, PM<sub>2.5</sub> was enhanced by a factor of 2.7 on pollutioned days at both sites, but the CO-scaled 891

PM<sub>2.5</sub> (the ratio of PM<sub>2.5</sub> to CO concentration) only showed an enhancement of a factor of 1.6-1.8 (Fig.
7), and the latter values were likely from the enhanced secondary aerosol formation.

894 As shown in Fig. 7, the CO-scaled SNA was 60-90% higher on polluted pollution days with 395 individual species 40-120% higher, even though their gaseous precursor (SO<sub>2</sub> and NO<sub>2</sub>, no data for 896 NH<sub>3</sub>) were only less than 30% higher. This suggests suggested stronger chemical transformation from gaseous precursors to particle formation on polluted pollution days. Sulfur oxidation ratio (SOR = n-897  $SO_4^{2-}/(n-SO_4^{2-}+n-SO_2))$  and nitrogen oxidation ratio (NOR =  $n-NO_3^{-}/(n-NO_3^{-}+n-NO_2)$ ) were defined to 398 399 evaluate the degree of secondary transformation (n refers to as the molar concentration) (Hu et al., 400 2014). NOR increased from 0.09 on clean days to 0.16 on pollutioned days at CD and from 0.07 to 0.14 401 at CQ. SOR increased only slightly, from 0.31 to 0.35 at CD and 0.28 to 0.35 at CQ. The CO-scaled 402 SOC increased by a factor of 2.6 on polluted pollution days at CQ, but no significant change was found

at CD (Fig. 7). The different patterns in SOC (or SOC/OC) than SNA (or SOR and NOR) suggested that secondary organic aerosols (SOA) production was of less important than SNA production at CD.

405  $SO_4^{2-}$  is predominantly formed via homogeneous gas-phase oxidation. In this pathway,  $SO_2$  is firstly oxidized by OH radical to SO<sub>3</sub>, and then to H<sub>2</sub>SO<sub>4</sub> (Stockwell and Calvert, 1983;Blitz et al., 406 2003). Apart from homogeneous reaction, particulate  $SO_4^{2-}$  can also be formed through heterogeneous 407 reactions with dissolved O<sub>3</sub> or H<sub>2</sub>O<sub>2</sub> under the catalysis of transition metal and in-cloud process 408 409 (Ianniello et al., 2011). HNO<sub>3</sub> is primarily produced from the reactions between NO<sub>2</sub> and OH radical during the daytime and later combines with NH<sub>3</sub> to produce particulate NO<sub>3</sub><sup>-</sup> (Calvert and Stockwell, 410 1983). Particulate NO<sub>3</sub><sup>-</sup> can also be formed through heterogeneous hydrolysis of N<sub>2</sub>O<sub>5</sub> on moist and 411 acidic aerosols during nighttime (Ravishankara, 1997;Brown and Stutz, 2012). Similarly, SOA is 412 mainly formed through photochemical oxidation of primary VOCs followed by condensation of SVOC 413 onto particles as well as through aqueous-phase reactions (Ervens et al., 2011). While photochemical 414 reactions are mostly influenced by temperature and oxidants amount, heterogeneous reactions always 415 416 depends on ambient RH. To further explore the formation mechanisms of secondary aerosols, SOR, 417 NOR and SOC/OC data were grouped with temperature (at 2°C interval). RH (at 5% interval) and daytime O<sub>3</sub> concentration (at 5 µg m<sup>-3</sup> interval) bins (Fig. 8). An obvious increase of SOR with 418 419 increasing RH was found at both sites, but this was not the case for temperature and O<sub>3</sub> concentration, suggesting heterogeneous processes played important roles in the formation of SO<sub>4</sub><sup>2-</sup>, as was suggested 420 421 in many previous studies (Quan et al., 2015; Zheng et al., 2015a; Zhao et al., 2013b). Interestingly, SOR exhibited a decreasing trend with increasing  $O_3$  concentration at  $O_3$  concentrations lower than 15 µg m<sup>-3</sup> 422 and an opposite trend was found at O<sub>3</sub> concentrations above 20 µg m<sup>-3</sup> (Fig. 8). Additionally, high PM<sub>2.5</sub> 423 424 concentrations were mostly associated with lower O<sub>3</sub> concentrations. This behavior might be explained by the complicated interactions between aerosol and O<sub>3</sub>. On the one hand, Aerosols aerosols are 425 426 generally considered as a constraining factor to O<sub>3</sub> production due to their absorption and scattering of 427 UV radiation, which reduce solar radiation and consequently decrease photochemical activity. On the other hand, aerosols can provide an interface for the heterogeneous reaction, in accordance with O<sub>3</sub> 428 429 consumption and secondary aerosol formation, which would result in decreaseding O<sub>3</sub> concentrations 430 and increaseding secondary aerosols (Zheng et al., 2015b). Apparently, other factors could become dominant in O<sub>3</sub> production and alter such a trend. Low It was further found that the ambient RH 431 remained high at low O<sub>3</sub> concentrations were further found to be associated with high RH (Fig. S5S4), 432

which was beneficial to form-SO<sub>4</sub><sup>2-</sup> formation through indicating that the formation of SO<sub>4</sub><sup>2-</sup> during the polluted periods was dominated by heterogeneous aqueous processes rather than photochemical reactions at both sites, as suggested in many previous studies (Quan et al., 2015;Zheng et al., 2015a;Zheng et al., 2015a;Zheng et al., 2015b). consistent with the observed results that high SOR value occurred at low O<sub>3</sub> concentrations.

438 Unlike SOR, NOR increased with both temperature and RH, suggesting the combined the effects 439 from homogeneous and heterogeneous reactions. However, under the very high temperature and RH 440 conditions, NOR exhibited a decreasing trend with increasing temperature and RH. Volatilization of 441 NH4NO3 at high temperature should be the major cause of such a phenomenon, but it is not clear – 442 about the cause of the decreasing trend of NOR under high RH. Pathak et al. (2009) investigated the 443 formation mechanism of NO<sub>3</sub><sup>-</sup> in ammonium-rich and ammonium-poor samples, suggesting 444 homogeneous gas-phase reaction became evident to form NO3<sup>-</sup> under the former conditions while 445 heterogeneous process dominated the NO<sub>3</sub><sup>-</sup> formation under the latter conditions. As shown in Fig. 9, SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> were almost completely neutralized by NH<sub>4</sub><sup>+</sup>, indicating an ammonium-rich 446 447 environment during the sampling campaign. The ammonium-rich environment was also confirmed by the molar ratios of  $[NO_3^-]/[SO_4^{2-}]$  and  $[NH_4^+]/[SO_4^{2-}]$ . The molar ratio  $[NO_3^-]/[SO_4^{2-}]$  as a function of 448  $[NH_4^+]/[SO_4^{2-}]$  is depicted in Fig.  $\frac{0}{9}$ , which shows significant positive correlations (R<sup>2</sup>=0.82-0.83 at the 449 two sites). Linear regression equations were obtained as  $[NO_3^-]/[SO_4^{2-}] = 0.85[NH_4^+]/[SO_4^{2-}]-1.89$  at 450 CD and  $[NO_3^-]/[SO_4^{2-}] = 0.92[NH_4^+]/[SO_4^{2-}]-1.82$  at CQ. Based on these equations, the molar ratio of 451  $[NH_4^+]/[SO_4^{2-}]$  was defined as the threshold value separating ammonium-rich and ammonium-poor 452 conditions when the value of  $[NO_3^-]/[SO_4^{2-}]$  was zero. In the present study, the threshold value was 2.2 453 and 2.0 at CD and CQ, respectively. The molar ratio  $[NH_4^+]/[SO_4^{2-}]$  was higher than the threshold value 454 at both sites, suggesting the prevalence of ammonium-rich condition. The near-perfect fitting between 455 456 the molar ratios of  $[NO_3^-]/[SO_4^{2-}]$  and  $[NH_4^+]/[SO_4^{2-}]$  further demonstrated the characteristics of  $NO_3^-$ 457 formed through homogenous gas-phase reaction. Moreover, NO<sub>3</sub><sup>-</sup> showed a strong correlation with 458 excess NH4<sup>+</sup> with correlation coefficients of 0.98-0.99 at both sites, providing an insight into the gas-459 phase reactions of ambient NH<sub>3</sub> and HNO<sub>3</sub>. Using high-resolution inorganic ions data, Tian et al. (2017) 460 demonstrated that NO3<sup>-</sup> was primarily formed via homogeneous reaction when RH below 75% and through heterogeneous processes when RH was above 75%. The increases in NOR with RH at both 461 462 sites also revealed the possibility of the heterogeneous processes, although this cannot be verified

463 directly due to the lack of high-resolution data.

The ratio of SOC/OC decreased with <u>increasing</u> temperature at CD but increased <u>with temperature</u> at CQ when <u>T-temperature</u> was lower than 10°C. Although SOC/OC did not correlate well with RH, an opposite trend was also found between CD and CQ at high RH conditions. Heterogeneous reactions seemed to be dominant in the formation of SOA at CD, whereas homogeneous reactions were prevalent at CQ. SOC/OC showed no apparent dependency on O<sub>3</sub> concentrations at either site, indicating more complex formation mechanisms of SOA than  $SO4^{2-}$  and  $NO3^{-}$ .

# 470 **3.2.3 Geographical origins of high PM<sub>2.5</sub> pollution**

PSCF analysis was applied to investigate the potential source regions contributing to high PM<sub>2.5</sub> 471 pollution. As can be seen from the PSCF maps in Fig. 10, all the pollutants including PM<sub>2.5</sub> and its 472 chemical components as well as gaseous precursors had similar spatial patterns of potential source 473 areas. Basically, all the major source areas for high pollutants concentrations were distributed within the 474 basin. Long-range transports as seen in North Plain and eastern coastal regions were not observed at CD 475 and CQ (Zhao et al., 2015; Zhang et al., 2013). At CD, the major source areas in winter included the 476 477 areas of the northeastern, southeastern and southern Chengdu and in some areas of eastern Chongqing. 478 A similar spatial distribution of PM<sub>2.5</sub> potential sources was also found by Liao et al. (2017) through PSCF analysis in winter 2013, in which high PM2.5 concentrations were mostly associated with sources 479 broadly located in the southeast of the basin, covering Neijiang, Zigong, Yibin, Luzhou and east part of 480 Chongqing. At CQ, the northeast area of Chongqing was identified as strong sources, where a number 481 of industries were located, such as Changshou chemical industrial ozone. Overall, PM<sub>2.5</sub> pollution at CQ 482 was characterized by significant local contribution from major sources located in or nearby Chongqing. 483 In contrast, regional transport in Sichuan Basin from southeast, south and southwest of Chengdu had a 484 485 major impact on PM<sub>2.5</sub> pollution at CD.

## 486 **4** Conclusions

Chemically-resolved PM<sub>2.5</sub> data collected during four seasons at two urban sites in Sichuan basin, southwest China were analyzed in the present study. On about 30% of the days, daily PM<sub>2.5</sub> exceeded the national air quality standard, with annual mean concentrations of  $67.0 \pm 43.4$  and  $70.9 \pm 41.4 \ \mu g m^{-3}$  at CD and CQ, respectively.  $SO_4^{2-}$ ,  $NO_3^{-}$ ,  $NH_4^+$ , OM, EC and FS awere the major chemical components of PM<sub>2.5</sub>, accounting for 16.8%, 13.6%, 10.8%, 26.1%, 5.4%, and 5.7% of PM<sub>2.5</sub> at CD, and 17.2%, 10.9%, 9.2%, 29.6%, 6.4%, and 9.5% at CQ, on annual average, respectively. The 493 concurrent occurrences of heavy pollution events at the two sites and similarities in pollutants 494 characteristics between the two sites demonstrated the regional-scale phenomenon of PM<sub>2.5</sub> episodes inside the Basin, which was mainly caused by the surrounding mountainous topography under typical 495 496 stagnant meteorological conditions. Such a finding was also supported by back trajectory analysis which showed that air masses reaching at both sites were originated within the basin and only traveled for short 497 distances on heavy polluted days. Differences between the two sites with regards to several major 498 499 chemical components provided evidences of sub-regional characteristics of emission sources and chemical transformation processes under different meteorological conditions. For example, an additional 500 source factor from motorcycle traffic was identified for VOCs emission in Chongqing, which led to higher 501 502 high OC concentrations, and lower relative humidity in Chongqing caused lower nitrate NO3<sup>-</sup> 503 concentration in this city despite similar levels of its gaseous precursors in the two cities. The present 504 study also identified different driving mechanisms for the polluted PM2.5 pollution episodes in the 505 Sichuan Basin than in the other regions of China. For example, sharply increased relative humidity was 506 thought to be one of the main factors causing high inorganic aerosol concentrations during the polluted 507 pollution periods in eastern coastal China and North China Plain, while in the Sichuan Basin the special topography and meteorological conditions are driving forces for such events considering relative humidity 508 was high throughout the year and did not differ much between polluted pollution and clear clean periods. 509 However, on annual basis heterogeneous reactions might be more important in this basin than in the other 510 regions of China due to the consistent high humidity conditions, as revealed in the case of sulfate  $SO_4^{2-}$ 511 formation in the present study. Future studies should use high-resolution data to verify the findings related 512 to chemical transformation paths proposed here. More importantly, a detailed emission inventory of 513 aerosol particles and related gaseous precursors in the basin should be developed promptly, which is 514 515 needed for further investigating PM2.5 formation mechanisms and for making future emission control policies. Source-receptor analysis using monitored chemical-resolved PM2.5 data should be conducted to 516 verify such emission inventories. 517

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519 *Competing interests*. The authors declare that they have no conflict of interest.

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Table 1 Meteorological parameters, annual and seasonal mean concentrations of PM<sub>2.5</sub>, gaseous pollutants and major chemical components at CD and CQ during 2014-2015.

		CD			CQ					
	Autumn	Winter	Spring	Summer	Annual	Autumn	Winter	Spring	Summer	Annual
Meteorological para	umeters									
T (°C)	15.8±2.9	9.3±2.5	20.4±4.4	28.3±2.9	18.5±7.7	16.0±3.2	10.0±2.3	20.5±4.5	28.4±3.4	18.8±7.6
P (hPa)	960±3.8	963±4.7	954±7.9	946±2.1	955±8.1	981±4.2	984±5.2	974±8.4	963±2.2	975±7.7
RH (%)	81.9±9.0	80.9±6.8	70.5±8.6	72.2±11.3	76.3±10.3	76.1±5.7	68.7±8.8	60.7±13.6	61.0±13.3	66.5±12.5
SR (w m <sup>-2</sup> )	54.9±40.3	37.8±27.2	128.9±65.0	123.6±94.2	67.2±56.7	na.	na.	na.	na.	na.
WS (m s <sup>-1</sup> )	0.5±0.2	0.4±0.3	$0.7 \pm 0.4$	0.5±0.2	0.5±0.3	$0.7 \pm 0.2$	0.7±0.3	1.0±0.4	0.7±0.3	0.8±0.3
Precipitation (mm)	<u>76.3</u>	<u>18.3</u>	<u>56.6</u>	<u>247.8</u>	<u>na.</u>	<u>73.3</u>	<u>22.0</u>	<u>104.6</u>	<u>206.3</u>	<u>na.</u>
PBLH <sub>max</sub> (m)	890±305	852±273	1296±491	1422±529	1119±481	821±252	928±260	1310±491	1329±505	1101±453
Concentrations of gaseous pollutants (µg m <sup>-3</sup> )										
O <sub>3</sub>	19.3±10.5	11.9±7.6	69.3±22.9	90.5±33.3	48.2±39.6	13.3±8.9	12.5±7.7	56.3±23.5	42.8±17.2	31.5±24.5
$SO_2$	15.8±7.0	21.5±9.5	11.2±6.3	11.3±4.7	14.9±3.7	16.4±4.6	23.3±9.2	13.9±5.3	14.4±5.4	17.0±7.3
$NO_2$	60.2±18.7	75.3±24.5	51.8±26.8	54.2±9.4	60.4±22.5	66.5±15.0	81.3±19.8	50.8±16.7	51.7±20.8	62.4±22.0

Concentrations of PM\_{2.5} and chemical compositions ( $\mu g \ m^{-3}$ )

PM2.5	62.1±38.4	113.5±47.8	48.0±25.2	45.1±15.2	67.0±43.4	56.3±23.6	115.1±53.9	58.3±24.6	54.2±16.2	70.9±41.4
SO4 <sup>2-</sup>	$10.5 \pm 6.5$	16.4±7.1	8.3±5.9	9.7±4.7	11.2±6.8	9.9±4.7	17.5±7.4	10.4±6.5	11.1±5.7	12.2±6.8
NO <sub>3</sub> -	9.3±7.4	17.5±8.8	5.9±3.6	3.9±2.2	9.1±8.0	7.8±3.8	15.8±9.5	5.9±5.0	1.6±1.3	7.7±7.6
$\mathrm{NH4}^+$	6.9±4.8	12.7±5.4	5.1±3.2	4.2±1.9	7.2±5.2	5.7±2.8	11.3±5.2	5.2±3.0	4.0±2.1	6.6±4.4
Cl-	$1.9 \pm 1.2$	3.4±1.9	0.6±0.4	0.2±0.2	1.5±1.7	0.8±0.4	$1.6 \pm 1.2$	$0.5 \pm 0.5$	0.04±0.03	$0.7\pm0.9$
$\mathbf{K}^+$	0.6±0.4	1.2±0.6	0.6±0.5	0.5±0.2	$0.7 \pm 0.5$	$0.5\pm0.2$	1.2±0.7	$0.5\pm0.2$	0.3±0.1	0.6±0.5
OC	10.4±6.1	19.7±8.4	6.3±3.7	7.4±1.5	10.9±7.6	9.7±4.7	24.2±13.6	$10.0 \pm 5.1$	8.5±3.4	13.1±10.0
EC	3.0±2.1	6.3±3.0	2.7±2.3	2.5±0.7	3.6±2.7	3.8±1.7	5.9±3.2	4.7 ±3.0	3.7±1.5	4.5±2.6
FS	3.2±1.6	4.5±2.0	4.8±3.0	2.7±1.5	3.8±2.2	5.0±2.8	6.3±3.3	9.1±7.6	6.5±4.0	6.7±5.0

na. means no data.

Table 2 Pearson's correlation analysis for daily concentrations of major components in PM<sub>2.5</sub> between CD and CQ, significant correlation (p<0.01) are bold faced.

	<del>SO</del> 4 <sup>2-</sup>	<del>NO₃⁻</del>	$\mathbf{NH4}^{+}$	<del>C1</del> -	<mark>₭</mark> +	<del>OC</del>	EC	<del>FS</del>
Autumn	<del>0.78</del>	<del>0.72</del>	<del>0.75</del>	<del>0.10</del>	<del>0.76</del>	<del>0.87</del>	<del>0.79</del>	<del>0.08</del>
Winter	<del>0.63</del>	<del>0.49</del>	<del>0.60</del>	<del>0.51</del>	<del>0.77</del>	<del>0.83</del>	<del>0.80</del>	<del>0.6</del> 4
<b>Spring</b>	<del>0.76</del>	<del>0.39</del>	<del>0.59</del>	<del>0.09</del>	<del>0.68</del>	<del>0.78</del>	<del>0.74</del>	<del>0.29</del>
Summer	<del>0. 49</del>	<del>-0.13</del>	<del>0.40</del>	<del>0.37</del>	<del>0.36</del>	<del>0.43</del>	<del>0.23</del>	-0.02
Annual	<del>0.72</del>	<del>0.71</del>	<del>0.76</del>	<del>0.67</del>	<del>0.79</del>	<del>0.80</del>	<del>0.72</del>	<del>0.31</del>

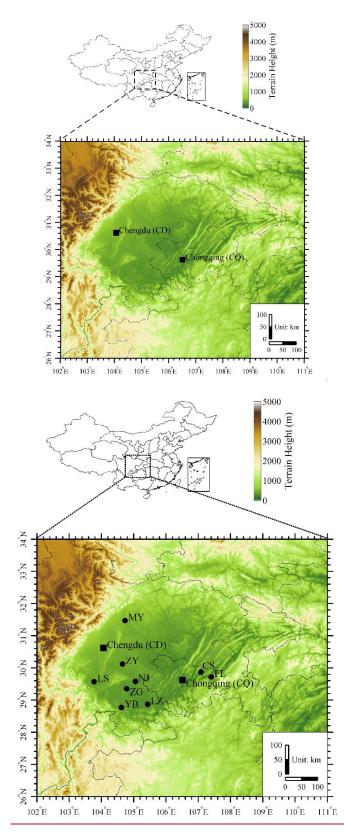


Figure 1. Locations of the sampling sites in Chengdu (CD) and Chongqing (CQ) and major cities in the Sichuan Basin. MY, Mianyang; ZY, Ziyang; LS, Leshan; NJ, Neijiang; ZG, Zigong; YB, Yibin; LZ, Luzhou; CS, Changshou; FL, Fuling.

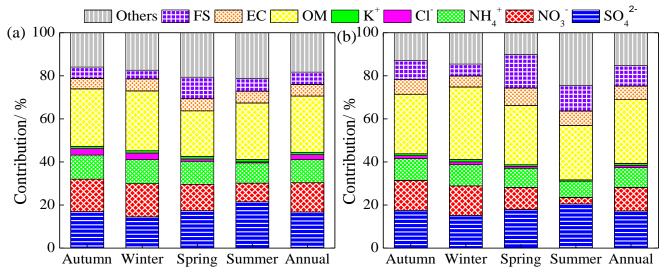


Figure 2. Seasonal and annual contributions of individual chemical components to  $PM_{2.5}$  at CD (a) and CQ (b).

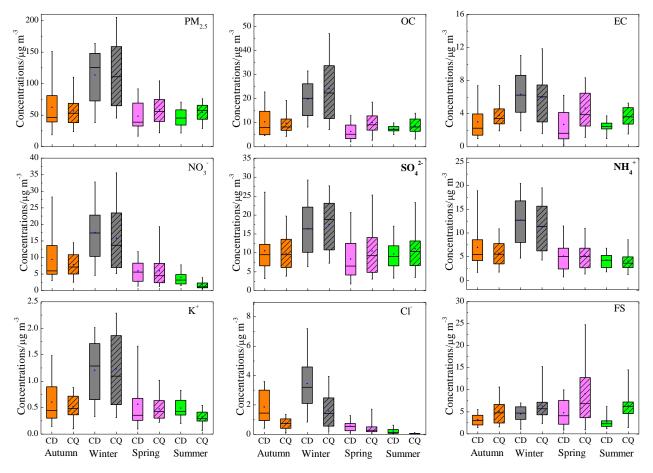


Figure 3. Seasonal distributions of  $PM_{2.5}$  and its major chemical components. Shown in each sub-figure are mean (dot symbol), median (horizontal line), the central 50% data ( $25^{th}$  - $75^{th}$  percentiles, box), and the central 90% data ( $5^{th}$ - $95^{th}$  percentile, whisker)

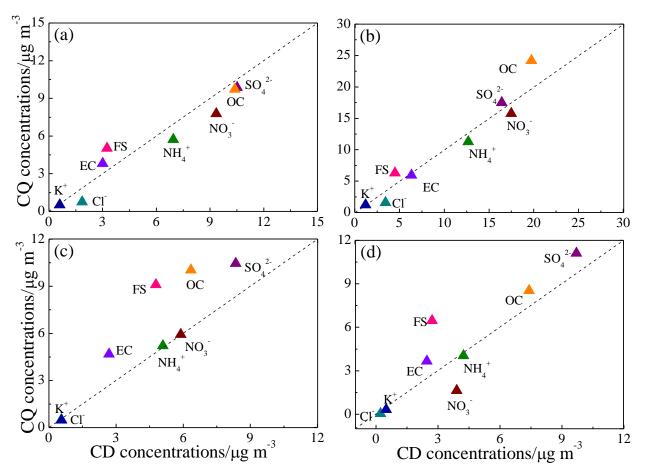


Figure 4. Seasonal mean concentrations of major components in autumn (a), winter (b), spring (c), and summer (d) at CD and CQ sites.

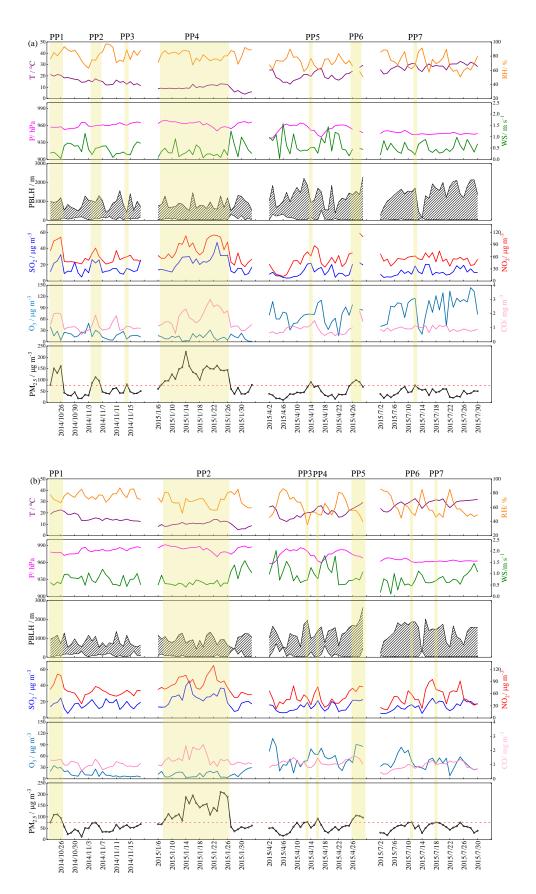


Figure 5 Temporal variations of meteorological parameters, <u>gas-phasegaseous</u> pollutants and  $PM_{2.5}$  during the campaign at CD (a) and CQ (b). Pollution episodes are highlighted by shaded areas.

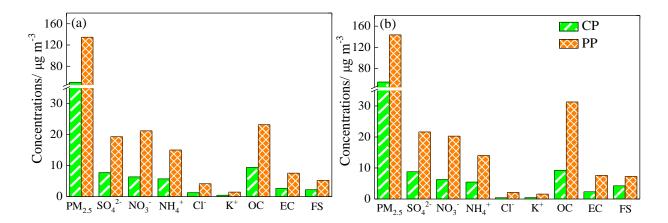


Figure 6. PM<sub>2.5</sub> and major chemical components during clean periods (CP) and <u>polluted-pollution</u> periods (PP) in winter at CD (a) and CQ (b). <u>At CD: CP, 6 January and 27 January-2 February 2015; PP,</u> <u>7-26 January 2015. At CQ: CP, 6-7 January and 27 January-2 February 2015; PP, 8-26 January 2015.</u>

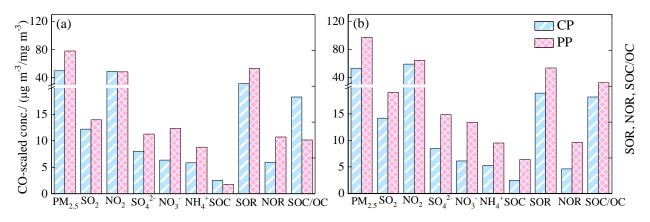


Figure 7. CO-scaled concentrations of various pollutants and the values of SOR, NOR, and SOC/OC in winter at CD (a) and CQ (b). <u>CP and PP is the same period as Figure 6.</u>

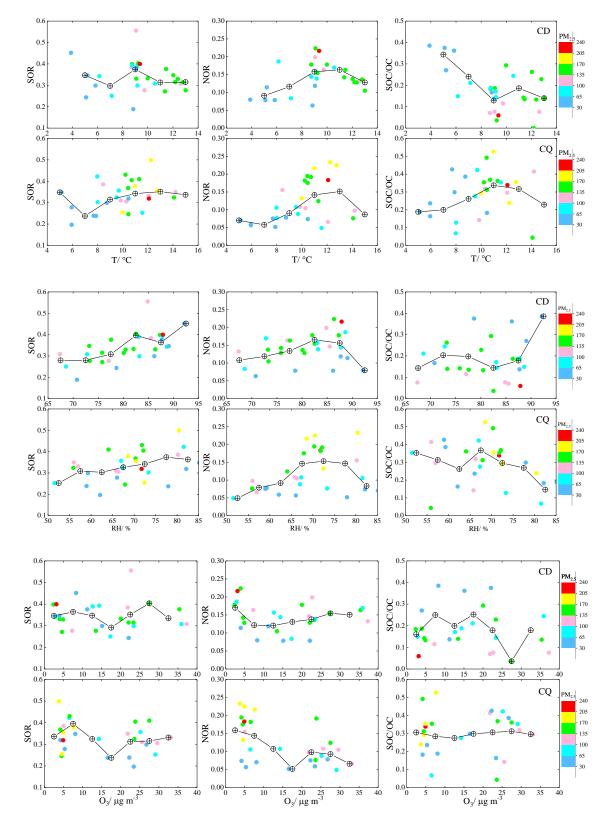


Figure 8. Correlations of SOR, NOR and SOC/OC against temperature <u>(upper)</u>, RH<u>(middle)</u> and O<sub>3</sub> concentration<u>(bottom) in winter</u> at CD and CQ.

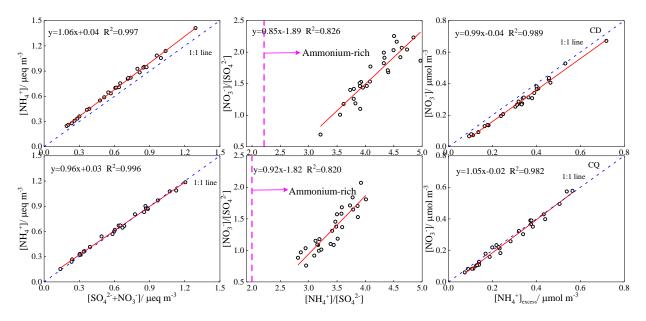


Figure 9.  $NH_4^+$  concentration as a function of the sum of  $SO_4^{2-}$  and  $NO_3^-$  in equivalent concentrations (left column), molar ratio  $NO_3^-/SO_4^{2-}$  as a function of  $NH_4^+/SO_4^{2-}$  (middle column), and  $NO_3^-$  concentration as a function of  $NH_4^+$  (right column) at CD (upper row) and CQ (lower row).

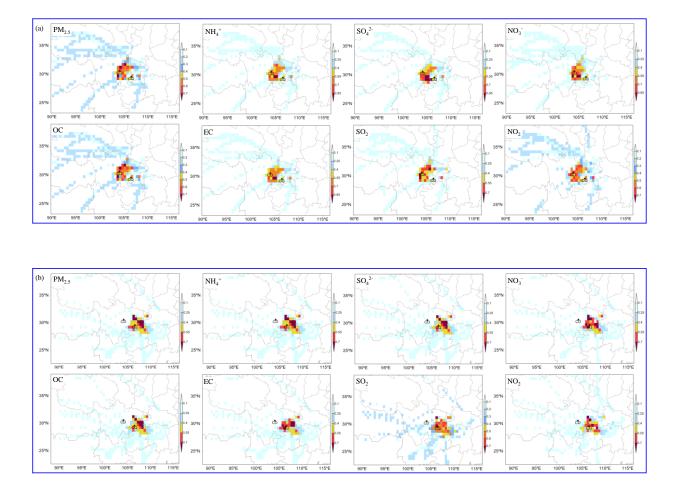


Figure 10. PSCF distribution of PM<sub>2.5</sub>, its chemical components, and gaseous precursors in winter at CD (a) and CQ (b).