We thank the reviewer for the constructive and thoughtful comments. Please find below detailed responses to each comment or question, including notations of improvements to the manuscript. Reviewer comments are in blue fonts. Changes to the text are highlighted in yellow.

General comments:

Study by Thalman et al. presents results of κ_{CCN} and mixing state derived from size resolved CCN observations collected during the GoAmazon2014/5 campaign. From κ_{CCN} and additional aerosol chemical composition measurements, the authors then estimate the hygroscopicity of the organic fraction κ_{org} . The results presented in the study, together with those presented by Pöhlker et al (2016) in another experimental site, represent the first long-term measurement of hygroscopic behaviour of aerosols in the Amazon. Overall, it is a well written manuscript on an important subject. Therefore, I recommend it for publication in ACP after minor corrections.

Specific comments:

Since chemical composition of aerosols in the study was largely dominated by the organic fraction, a discussion on the uncertainties related to limited solubility or surface active species could be relevant, even if no quantitative assessment of these uncertainties is provided.

Yes, limited solubility and reduced surface tension could impact the CCN activity and hygroscopicity of organics. As the κ value was derived from dry particle diameter and critical supersaturation assuming the surface tension of pure water, κ value reported in this manuscript represents the "apparent hygroscopicity", which includes the potential impact due to limited solubility and reduction of surface tension (Sullivan et al., 2009). We have clarified this by including the following sentence in the text:

As hygroscopicities reported in this study were derived from particle dry diameter and critical supersaturation, they represent "apparent hygroscopicity", which includes the potential impact due to the limited solubility of organics and the reduction of surface tension by surface active species (Sullivan et al., 2009).

P 21 L 431-440 Given that the importance of hygroscopicity to the aerosol CCN behavior is enhanced for κ values below 0.2, the choice of an adequate κ value to represent the Amazonian aerosol CCN behavior is important, for instance, to modeling studies. In this sense, it could be interesting to compare the presented results with κ_{HTDMA} results from campaigns previous to 2008 as well, since κ have already been calculated from the original data of some of these campaigns in other works (see, for example, Gunthe et al. (2009) and the Supplement of Sánchez Gácita et al. (2017)). In the authors opinion, differences between presented results and κ_{HTDMA} from earlier campaigns are solely due to the use of H-TDMA or CCN technique, or other factors could be important as well?

We thank the reviewer for this suggestion. We added particle hygroscopicity (κ_{HTDMA}) derived from HTDMA measurements during campaigns before 2008 in Figure 2 for comparison,

including κ_{HTDMA} measured in July 2001, during a "recent biomass burning period" of the CLAIRE-2001 study (Rissler et al., 2004), and that measured from 11 September to 8 October 2002, during the dry period of the LBA-SMOCC (Rissler et al., 2006). During the above two periods, aerosols were strongly influenced by fresh local biomass burning emissions. The measurements during CLAIRE-2001 took place at Balbina, about 125 km northeast of Manaus, and the κ_{HTDMA} during the "recent biomass burning period" agrees well with κ_{CCN} derived at the T3 site for air masses with strong local biomass burning influences. In contrast, κ_{HTDMA} observed during the dry period of LBA-SMOCC is substantially lower than κ_{CCN} at T3 for local biomass burning air mass at all sizes. As LBA-SMOCC took place in the state of Rondônia in southwestern Amazonia with extensive biomass burning activities during the dry season, the difference in κ may be due to the differences in fire condition and the type of vegetation burned. Previous studies show particles sometime exhibit larger κ values for droplet activation (derived from CCN measurements under supersaturated conditions) than for particle growth (derived from particle GF under sub-saturated conditions), this could also contribute to the difference in κ values. We have updated Figure 2, and included following discussion in the text:

For the air masses with strong influence from local biomass burning, the value of κ_{CCN} and its size dependence are consistent with the κ value derived from particle growth factor measurements in July 2001, during a "recent biomass burning period" of the CLAIRE-2001 study (Rissler et al., 2004), which took place at Balbina, about 125 km northeast of Manaus. In contrast, κ values derived from particle growth factor measurements from 11 September to 8 October 2002, during the dry period of the LBA-SMOCC (Rissler et al., 2006) are substantially lower than κ_{CCN} observed at the T3 site for local biomass burning air masses at all sizes. As LBA-SMOCC took place in the state of Rondônia in southwestern Amazonia with extensive biomass burning activities during the dry season, the difference in κ could be due to the differences in fire condition and the type of vegetation burned. Previous studies show particles sometime exhibit larger κ values for droplet activation (derived from particle growth factor under sub-saturated conditions), this could also contribute to the differences in κ values.

Supp. P10 L172 Please provide a reference for equation S1 of supplement

The reference to the supplementary material in Mei et al. (2013) has been added, the formula as given in equation S1 converts the formula in the previous paper to the hygroscopicity dispersion instead of the critical supersaturation dispersion and hence is somewhat different.

Supp. P10 L174-175 The use of equation S2 of supplement was suggested by Petters and Kreidenweis (2007) for $\kappa > 0$:2. Considering that for most conditions in this study κ_{CCN} was below this threshold, perhaps an iterative approach on the original κ -Köhler equation (see, for example, Carrico 2008) would have been more adequate. A sensitivity test on a small fraction of the data would be desirable to show that uncertainties derived from the use of this simplified approach are indeed low or negligible.

We thank the reviewer for this suggestion. We carried out the sensitivity study using an iterative approach based on the original κ -Köhler equation as suggested by the reviewer. The relative and absolute uncertainties in κ_{CCN} derived using equation S2 are shown in Fig 1A and 1B below for each particle diameter. The lower κ_{CCN} limit of each curve represents 1st percentile of κ_{CCN} values derived at the corresponding particle diameter. The results show that the relatively uncertainty is less than 7% for 99% of the data at all particle diameters. Furthermore, the main conclusions of the manuscript are based on κ_{CCN} derived at 112, 142, and 171 nm (e.g., Figures 3-12), and the relative uncertainty is less than 4.5% for 99% of data at the three sizes. Therefore, the uncertainties derived from the use of the simplified approach are low and negligible.



Figure 1: (a) Relative and (b) absolute uncertainties of κ_{CCN} derived using equation S2 at each of the particle diameters. The lower κ_{CCN} limit of the curves represents the 1st percentile of the κ_{CCN} value at corresponding particle diameter.

References

Mei, F., Setyan, A., Zhang, Q., and Wang, J.: CCN activity of organic aerosols observed downwind of urban emissions during CARES, Atmospheric Chemistry and Physics, 13, 12155-12169, 2013.

Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic growth and cloud condensation nucleus activity, Atmos. Chem. Phys., 7, 1961-1971, 2007.

Rissler, J., Swietlicki, E., Zhou, J., Roberts, G., Andreae, M. O., Gatti, L. V., and Artaxo, P.: Physical properties of the sub-micrometer aerosol over the Amazon rain forest during the wet-todry season transition - comparison of modeled and measured CCN concentrations, Atmospheric Chemistry and Physics, 4, 2119-2143, 2004. Rissler, J., Vestin, A., Swietlicki, E., Fisch, G., Zhou, J., Artaxo, P., and Andreae, M. O.: Size distribution and hygroscopic properties of aerosol particles from dry-season biomass burning in Amazonia, Atmos. Chem. Phys., 6, 471-491, 2006.

Sullivan, R. C., Moore, M. J. K., Petters, M. D., Kreidenweis, S. M., Roberts, G. C., and Prather, K. A.: Effect of chemical mixing state on the hygroscopicity and cloud nucleation properties of calcium mineral dust particles, Atmos. Chem. Phys., 9, 3303-3316, 2009.

We thank the reviewer for the constructive and thoughtful comments. Please find below detailed responses to each comment or question, including notations of improvements to the manuscript. Reviewer comments are in blue fonts. Changes to the text are highlighted in yellow.

Overall: The manuscript in overall is very well written, and in agreement with scope of ACP. The scientific issues were well addressed, and are relevant. Several important measurements were performed, which were well treated and discussed, being an important scientific contribution. Therefore, I recommend publish the manuscript in ACP. However, I have few specific questions and technical corrections.

Specific comments and questions:

Line 105: Include references to support the affirmation "The hygroscopicities of typical inorganic in ambient particles are relatively well known"

Reference to Petters and Kreidenweis (2007) is now included.

Line 106: Please comment this sentence and clarify: "atmospheric aerosols consist of a large number of organic compounds, which often dominate the total fine aerosol mass". Which type of region, urban, rural, forest, etc?

The organics often dominate the total fine aerosol mass, especially in forested areas such as in Hyytiälä, Finland and the Amazon basin (Jimenez et al., 2009; de Sá et al. 2016). Organics also dominate total fine aerosol mass in some of the urban areas, such as the Mexico City, Edinburgh, and Zurich during summer time (Jimenez et al., 2009). We have updated the sentence and included additional references. The sentence is changed to:

...atmospheric aerosols consist of a large number of organic compounds, which often dominate the total fine aerosol mass, especially in forested areas (e.g., de Sá et al., 2016; Jimenez et al., 2009; Zhang et al., 2007).

Line 303-305: Why the authors use the same size distribution shape? What is the potential implications?

Given the low volume fraction of nitrate (see Figs 1 and 3-7), vast majority of the ammonium was associated with sulfate. Therefore, the assumption that ammonium had the same size distribution shape as sulfate should be appropriate.

Line 352: The authors said that it was correlated with concentrations of levoglucosan, vanillin and anthropogenic emission of aromatics. However, not mentioned in the section 2.4. this measurements. Please included.

These measurements and correlations are detailed in a manuscript that is in preparation (de Sá et al., 2017). The reference is included.

Line 393: Define CN

The definition (condensation nuclei) has been added to the text.

Line 428: The authors could explain or comment why no clear seasonal trend was observed for sulfate volume fraction?

This is a good question. The lack of a clear seasonal trend of sulfate volume fraction is consistent with observations at the ATTO site (Andreae et al., 2015). At this point, it is unclear what causes the relatively constant sulfate volume fraction despite the large variation in aerosol volume concentration. During dry season, there is a strong increase in aerosol volume concentration, and some of the sulfate is obviously from biomass burning. However the ratio of sulfate to organics is substantially higher than that of biomass burning aerosols, and additional sulfate from fossil fuel burning is likely needed to explain the ratio. This additional sulfate may come from a mix of emissions from the populated regions of northeastern Brazil, emissions along the river, and long range transport from Africa. A detailed analysis of the seasonal cycle of sulfate sources is outside of the scope of this manuscript. We have included the following sentence in the text:

The lack of clear seasonal trends of sulfate and organic fractions are consistent with observations at the T0a site (Andreae et al., 2015).

Lines: 431-432 and 442-443 its look contradictory as well in the conclusion, in lines 699-701 and 704-705. The authors said in lines 699-701: "The KCCN increased with particle size during all seasons, consistent with decreasing organic volume fraction: : :..".However, when Korg increase the KCCN also increase according to text. Please clarify.

The statements are not contradictory. They address the impact on κ_{CCN} by organic volume fraction and κ_{org} value, respectively. Because of the relatively low κ_{org} value compared to those of inorganic species (e.g., sulfate), either a decreasing organic volume fraction or an increasing κ_{org} could lead to an increasing κ_{CCN} .

Technical corrections:

Line 105: ": : :typical inorganic in ambient particles: : :" better ": : :.typical inorganic particles in ambient: : :"

"typical inorganic in ambient particles" refers to inorganic species in ambient particles, which are often mixtures of both inorganic and organic species. Instead, "typical inorganic particles in ambient" implies ambient particles consisting of inorganic species only. Therefore, the first description is more accurate. We have changed "inorganics" to "inorganic species" for clarification.

Line 221: use PM2.5

The text already reads PM2.5 as suggested.

Line 228: missing comma or and

The platform described does have a lot of acronyms, but MAOS is a subset of AMF-1, as such, no and or comma is necessary.

Line 656: Use GoAmazon2014/5 as in others places (lines 687, 292, 249, etc)

Done.

Table 1: Put the meaning of K (Bkgd) in table footnotes

Done.

Figs. 9 and 10: increase the size of the word Korg as in fig. 11

Figures 9 and 10 have been updated.

References

Andreae, M. O., Acevedo, O. C., Araùjo, A., Artaxo, P., Barbosa, C. G. G., Barbosa, H. M. J., Brito, J., Carbone, S., Chi, X., Cintra, B. B. L., Silva, N. F. d., Dias, N. L., Dias-Júnior, C. Q., Ditas, F., Ditz, R., Godoi, A. F. L., Godoi, R. H. M., Heimann, M., Hoffmann, T., Kesselmeier, J., Könemann, T., Krüger, M. L., Lavric, J. V., Manzi, A. O., Moran-Zuloaga, D., Nölscher, A. C., Nogueira, D. S., Piedade, M. T. F., Pöhlker, C., Pöschl, U., Rizzo, L. V., Ro, C.-U., Ruckteschler, N., Sá, L. D. A., Sá, M. D. O., Sales, C. B., Santos, R. M. N. D., Saturno, J., Schöngart, J., Sörgel, M., Souza, C. M. d., Souza, R. A. F. d., H. Su, N., Targhetta, Tóta, J., Trebs, I., Trumbore, S., Eijck, A. v., Walter, D., Wang, Z., Weber, B., Williams, J., Winderlich, J., Wittmann, F., Wolff, S., and Yáñez-Serrano, A. M.: The Amazon Tall Tower Observatory (ATTO): overview of pilot measurements on ecosystem ecology, meteorology, trace gases, and aerosols, Atmospheric Chemistry and Physics, 15, 10723-10776, 2015. de Sá, S. S., Palm, B. B., Campuzano-Jost, P., Day, D. A., Newburn, M. K., Hu, W., Isaacman-VanWertz, G., Yee, L. D., Thalman, R., Brito, J., Carbone, S., Artaxo, P., Goldstein, A. H., Manzi, A. O., Souza, R. A. F., Mei, F., Shilling, J. E., Springston, S. R., Wang, J., Surratt, J. D., Alexander, M. L., Jimenez, J. L., and Martin, S. T.: Influence of urban pollution on the production of organic particulate matter from isoprene epoxydiols in central Amazonia, Atmos. Chem. Phys. Discuss., 2016, 1-58, 2016.

Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prevot, A. S. H., Zhang, Q., Kroll, J. H., DeCarlo, P. F., Allan, J. D., Coe, H., Ng, N. L., Aiken, A. C., Docherty, K. S., Ulbrich, I. M., Grieshop, A. P., Robinson, A. L., Duplissy, J., Smith, J. D., Wilson, K. R., Lanz, V. A., Hueglin, C., Sun, Y. L., Tian, J., Laaksonen, A., Raatikainen, T., Rautiainen, J., Vaattovaara, P., Ehn, M., Kulmala, M., Tomlinson, J. M., Collins, D. R., Cubison, M. J., Dunlea, E. J., Huffman, J. A., Onasch, T. B., Alfarra, M. R., Williams, P. I., Bower, K., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K., Salcedo, D., Cottrell, L., Griffin, R., Takami, A., Miyoshi, T., Hatakeyama, S., Shimono, A., Sun, J. Y., Zhang, Y. M., Dzepina, K., Kimmel, J. R., Sueper, D., Jayne, J. T., Herndon, S. C., Trimborn, A. M., Williams, L. R., Wood, E. C., Middlebrook, A. M., Kolb, C. E., Baltensperger, U., and Worsnop, D. R.: Evolution of Organic Aerosols in the Atmosphere, Science, 326, 1525-1529, 2009.

Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic growth and cloud condensation nucleus activity, Atmos. Chem. Phys., 7, 1961-1971, 2007.

Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Allan, J. D., Coe, H., Ulbrich, I., Alfarra, M. R., Takami, A., Middlebrook, A. M., Sun, Y. L., Dzepina, K., Dunlea, E., Docherty, K., DeCarlo, P. F., Salcedo, D., Onasch, T., Jayne, J. T., Miyoshi, T., Shimono, A., Hatakeyama, S., Takegawa, N., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K., Williams, P., Bower, K., Bahreini, R., Cottrell, L., Griffin, R. J., Rautiainen, J., Sun, J. Y., Zhang, Y. M., and Worsnop, D. R.: Ubiquity and dominance of oxygenated species in organic aerosols in anthropogenically-influenced Northern Hemisphere midlatitudes, Geophys. Res. Lett., 34, L13801, 2007.

2	tral Amazonia: Seasonal and diel variations and impact of anthropogenic emissions
3	
4	Ryan Thalman ^{1,#} , Suzane S. de Sá ² , Brett B. Palm ³ , Henrique M. J. Barbosa ⁴ , Mira L. Pöhlker ⁵ ,
5	M. Lizabeth Alexander ⁷ , Joel Brito ^{4*} , Samara Carbone ⁴ , Paulo Castillo ¹ , Douglas A. Day ³ ,
6	Chongai Kuang ¹ , Antonio Manzi ⁸ , Nga Lee Ng ^{9,10} , Arthur J. Sedlacek III ¹ , Rodrigo Souza ¹¹ ,
7	Stephen Springston ¹ , Thomas Watson ¹ , Christopher Pöhlker ⁵ , Ulrich Pöschl ^{5,6} , Meinrat O. An-
8	dreae ^{5,12} , Paulo Artaxo ⁴ Jose L. Jimenez ³ , Scot T. Martin ^{2,13} , Jian Wang ¹
9	
10	¹ Environmental and Climate Sciences Department, Brookhaven National Laboratory, Upton,
11	NY, USA
12	² School of Engineering and Applied Sciences, Harvard University, Cambridge, MA, USA
13	³ Department of Chemistry and Biochemistry and Cooperative Institute for Research in Environ-
14	mental Sciences (CIRES), University of Colorado Boulder, Boulder, CO, USA
15	⁴ Physics Institute, University of São Paulo, São Paulo, Brazil
16	⁵ Biogeochemistry and Multiphase Chemistry Departments, Max Planck Institute for Chemistry,
17	Mainz, Germany
18	⁶ Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA, USA
19	⁷ Pacific Northwest National Laboratory, Richland, WA, USA
20	⁸ National Institute of Amazonian Research, Manaus, Amazonas, Brazil
21	⁹ School of Chemical and Biomolecular Engineering, Georgia Institute of Technology, Atlanta,
22	GA, USA

CCN activity and organic hygroscopicity of aerosols downwind of an urban region in cen-

²³ ¹⁰School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA,

24 USA

- 25 ¹¹ Amazonas State University, Manaus, Amazonas, Brazil
- ²⁶ ¹²Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA, USA
- ¹³Department of Earth and Planetary Sciences, Harvard University, Cambridge, MA, USA

28

- 29 [#]Now at Department of Chemistry, Snow College, Richfield, UT, USA
- 30 *Now at Laboratory for Meteorological Physics, University Clermont Auvergne, Clermont-
- 31 Ferrand, France
- 32 Correspondence to: J. Wang (jian@bnl.gov)

- 33 Abstract
- 34

35 During the Observations and Modeling of the Green Ocean Amazon (GoAmazon2014/5) 36 campaign, size-resolved cloud condensation nuclei (CCN) spectra were characterized at a research site (T3) 60 km downwind of the city of Manaus, Brazil, in central Amazonia for one year 37 38 (12 March 2014 to 3 March 2015). Particle hygroscopicity (κ_{CCN}) and mixing state were derived 39 from the size-resolved CCN spectra, and the hygroscopicity of the organic component of the aer-40 osol (κ_{org}) was then calculated from κ_{CCN} and concurrent chemical composition measurements. 41 The annual average $\kappa_{\rm CCN}$ increased from 0.13 at 75 nm to 0.17 at 171 nm, and the increase was 42 largely due to an increase in sulfate volume fraction. During both wet and dry seasons, $\kappa_{\rm CCN}$, $\kappa_{\rm org}$, 43 and particle composition under background conditions exhibited essentially no diel variations. 44 The constant κ_{org} of ~ 0.15 is consistent with the largely uniform and high O:C value (~0.8), in-45 dicating that the aerosols under background conditions are dominated by the aged regional aero-46 sol particles consisting of highly oxygenated organic compounds. For air masses strongly influ-47 enced by urban pollution and/or local biomass burning, lower values of κ_{org} and organic O:C 48 atomic ratio were observed during night, due to accumulation of freshly emitted particles, domi-49 nated by primary organic aerosol (POA) with low hygroscopicity, within a shallow nocturnal 50 boundary layer. The O:C, κ_{org} , and κ_{CCN} increased from the early morning hours and peaked 51 around noon, driven by the formation and aging of secondary organic aerosol (SOA) and dilution 52 of POA emissions into a deeper boundary layer, while the development of the boundary layer, 53 which leads to mixing with aged particles from the residual layer aloft, likely also contributed to 54 the increases. The hygroscopicities associated with individual organic factors, derived from PMF 55 analysis of AMS spectra, were estimated through multi-variable linear regression. For the SOA 56 factors, the variation of the κ value with O:C agrees well with the linear relationship reported 57 from earlier laboratory studies of SOA hygroscopicity. On the other hand, the variation in O:C of 58 ambient aerosol organics is largely driven by the variation in the volume fractions of POA and 59 SOA factors, which have very different O:C values. As POA factors have hygroscopicity values 60 well below the linear relationship between SOA hygroscopicity and O:C, mixtures with different 61 POA and SOA fractions exhibit a steeper slope for the increase of κ_{org} with O:C, as observed 62 during this and earlier field studies. This finding helps better understand and reconcile the differ-63 ences in the relationships between κ_{org} and O:C observed in laboratory and field studies, therefore providing a basis for improved parameterization in global models, especially in a tropical 64 65 context.

66

67 **1 Introduction**

Atmospheric aerosols have a major impact on the radiative balance of the Earth's climate 68 69 system by changing the microphysical structure, lifetime, and coverage of clouds. For the same 70 liquid water content, high aerosol concentration leads to more, smaller cloud droplets, and there-71 fore higher cloud albedo (Twomey, 1977). The smaller droplet size also delays or inhibits warm 72 precipitation, leading to increases in both cloud lifetime and coverage (Albrecht, 1989), and ul-73 timately invigoration of convective clouds (Rosenfeld et al., 2008). Currently, the effects of aer-74 osol on clouds remain one of the largest uncertainties in simulated climate change during indus-75 trial era, and a large portion of this uncertainty is due to the natural aerosol properties and pro-76 cesses represented in models (Carslaw et al., 2013; Ghan et al., 2013). The Amazon represents 77 more than half of the planet's rainforest and is a rapidly changing region where deforestation, 78 human activity and natural resource needs are all at play in changing the ecosystem (Andreae et 79 al., 2015; Batistella et al., 2013; Davidson et al., 2012). The Amazon basin also represents at 80 times one of the cleanest continental regions on the planet where it is still possible to find ex-81 tended periods of little or no impact of anthropogenic activity, although the long-distance 82 transport of pollution is occasionally observed (Andreae et al., 2015; Hamilton et al., 2014; 83 Martin et al., 2010b; Wang et al., 2016a; Wang et al., 2016b; Williams et al., 2002). This makes 84 the Amazon Basin an ideal location to characterize aerosol under near natural conditions and as-85 sess the impact due to urban emissions and biomass burning (Kuhn et al., 2010). The biogenic 86 activity of this region makes it a major source of organic carbon released into the atmosphere via 87 isoprene and monoterpenes (Guenther et al., 2006; Guenther et al., 2012; Kesselmeier et al., 88 2002; Kuhn et al., 2007) which are mediated by biotic stress through heat, sunlight and changes 89 in CO_2 (Heald et al., 2009).

5

90 To understand the impact of aerosol on clouds and climate requires knowledge of the concen-91 tration of cloud condensation nuclei, which are particles that are able to form cloud droplets un-92 der relevant atmospheric conditions. The minimum supersaturation required to activate a particle 93 into a cloud droplet can be predicted using κ -Köhler theory based on particle size and the single 94 hygroscopicity parameter κ , which combines a number of thermodynamic properties required for 95 the description of water activity of the growing droplets (Petters and Kreidenweis, 2007). The 96 value of κ is determined by the physicochemical properties of the solutes, including their molar 97 volume, activity coefficient, and the effect on surface tension. For multi-component particles, κ 98 is the volume average of participating species. Hygroscopicity also describes particle growth un-99 der sub-saturated conditions and can be derived from the particle growth factor (GF). However, 100 particles sometime exhibit larger κ values for droplet activation (derived from CCN measure-101 ments under supersaturated conditions) than for particle growth (derived from particle GF under 102 sub-saturated conditions) (e.g., Duplissy et al., 2008; Good et al., 2010; Mikhailov et al., 2013; 103 Pajunoja et al., 2015; Wex et al., 2009). In this paper, "hygroscopicity" represents κ associated 104 with droplet activation derived from CCN measurements unless noted otherwise.

105 The hygroscopicities of typical inorganics species in ambient particles are relatively well 106 known (Petters and Kreidenweis, 2007). However, atmospheric aerosols consist of a large num-107 ber of organic compounds, which often dominate the total fine aerosol mass, especially in forest-108 ed areas (e.g., de Sá et al., 2016; Jimenez et al., 2009; Zhang et al., 2007). The hygroscopicity of 109 aerosol organics (κ_{org}) have been examined in both laboratory (e.g., Asa-Awuku et al., 2009; 110 Duplissy et al., 2011; King et al., 2009; Lambe et al., 2011; Massoli et al., 2010; Prenni et al., 111 2007; Raymond and Pandis, 2003) and field studies (e.g., Cerully et al., 2015; Chang et al., 2010; 112 Dusek et al., 2010; Gunthe et al., 2009; Jimenez et al., 2009; Lathem et al., 2013a; Mei et al.,

113 2013a; Mei et al., 2013b; Moore et al., 2011; Moore et al., 2012; Pöhlker et al., 2016; Rose et al., 114 2010; Shantz et al., 2008; Wang et al., 2008). Overall, these studies show that aerosol organics 115 exhibit a wide range of κ values from 0 to ~0.3, and κ_{org} often increases substantially during aer-116 osol aging in the atmosphere (e.g., Duplissy et al., 2011; Jimenez et al., 2009; Lambe et al., 117 2011; Massoli et al., 2010; Mei et al., 2013a).

118 A number of recent studies examined the sensitivity of predicted CCN concentration and 119 cloud droplet number concentration to aerosol properties (e.g., Ervens et al., 2010; Kammermann 120 et al., 2010; McFiggans et al., 2006; Mei et al., 2013a; Reutter et al., 2009; Rissman et al., 2004; 121 Roberts et al., 2002; Wang, 2007; Wang et al., 2008). These studies show that the predicted CCN 122 concentration is often sensitive to $\kappa_{\rm org}$, especially for aerosol under background conditions where 123 organics tend to dominate submicron aerosol mass (Liu and Wang, 2010; Mei et al., 2013a). Us-124 ing a constant κ_{org} may lead to large biases in predicted CCN concentrations and aerosol indirect 125 forcing (Liu and Wang, 2010). Therefore, it is imperative to understand organic hygroscopicity 126 under background conditions, such as in the Amazon forest, as well as the variation of organic 127 hygroscopicity due to anthropogenic emissions.

128 There have been several studies of aerosol hygroscopicity in the Amazon Basin over the past 129 20 years (Gunthe et al., 2009; Mikhailov et al., 2013; Pöhlker et al., 2016; Rissler et al., 2006b; 130 Roberts et al., 2001; Vestin et al., 2007; Whitehead et al., 2016; Zhou et al., 2002). Gunthe et al. 131 (2009) performed size resolved CCN measurements during the wet season in February and 132 March 2008 as part of the AMAZE-08 campaign (Martin et al., 2010a). That study reported no 133 diel cycle in the CCN concentration during periods with little or no influence of pollution. 134 Pöhlker et al. (2016) measured size-resolved CCN spectra at a remote background site (Amazon Tall Tower Observatory ATTO) over a one-year period from March 2014 to February 2015 and 135

136 observed no diel cycle and only weak seasonal trends in derived particle hygroscopicity, while 137 CCN concentrations had a pronounced seasonal cycle as the background aerosol concentration 138 was strongly influenced by regional biomass burning during the dry season. During the SMOCC-139 2002 campaign (Large scale Biosphere atmosphere experiment in Amazonia – SMOke, aerosols, 140 Clouds, rainfall, and Climate) particle hygroscopicity was derived from HTDMA measurements 141 in the state of Rondônia in the southwest of the Amazon region during the dry season from Sep-142 tember to November of 2002 (Rissler et al., 2006b; Vestin et al., 2007). The study concluded that 143 the diel variation in the aerosol hygroscopicity could be linked to the structure and dynamics of 144 the boundary layer. Local sources dominated night-time aerosol properties with downward mix-145 ing from the residual layer aloft as the day progressed. All of these studies found that particle 146 hygroscopicity increased with particle size (from the Aitken to accumulation modes), consistent 147 with higher sulfate content at larger sizes (Gunthe et al., 2009). The same boundary layer evolu-148 tion has been found to influence particle number and CCN evolution in a number of other related 149 studies (Fisch et al., 2004; Martin et al., 2010a; Rissler et al., 2006b; Vestin et al., 2007; 150 Whitehead et al., 2010; Zhou et al., 2002).

151 In this study we present measurements of size resolved CCN spectra at five particle diame-152 ters ranging from 75 to 171 nm downwind of Manaus, Brazil, in central Amazonia for a period 153 of one year from March 12, 2014 to March 3, 2015. Particle hygroscopicity, mixing state, and 154 organic hygroscopicity are derived from the size-resolved CCN activated fraction and concurrent 155 aerosol composition measurements. The diel variations of these properties are examined for dif-156 ferent seasons (i.e., wet season, dry season, and transition seasons) and for different types of rep-157 resentative air masses, including background conditions, as well as influences of urban pollution 158 plumes and/or local biomass burning. During the wet season, the background air mass represents

near natural conditions, with occasional impact from anthropogenic emissions, while in the dry season, the background is dominated by regional biomass burning aerosol particles. The relationship between organic hygroscopicity and particle oxidation level (i.e., O:C atomic ratio) is examined for both dry and wet seasons. Hygroscopicities associated with organic factors from Aerosol Mass Spectrometry (AMS) Positive Matrix Factorization (PMF) analysis are derived, and their relationship with the O:C ratio is compared with those from previous laboratory studies.

165 2 Experimental Setup

166 2.1 Measurement Sites

167 Observations and Modeling of the Green Ocean Amazon (GoAmazon2014/5), sponsored by 168 the US Department of Energy (DOE) and several Brazilian and German agencies, took place at 169 multiple surface sites surrounding Manaus, Brazil, from January 2014 through December 2015 170 (Martin et al., 2016b). This work focuses on the measurements carried out at a downwind site 171 (T3, 3°12'47.82"S, 60°35'55.32"W, 60 km west of Manaus) from March 2014 to March 2015. 172 Depending on the wind direction, the T3 site experienced conditions ranging from nearly natural 173 to heavily polluted. More detailed characterizations of aerosol and gas phase chemical composi-174 tion were carried out at the T3 site during Intensive Operating Periods (IOPs) from February 1, 175 2014 to March 31, 2014 and from August 15, 2014 to October 15, 2014. In addition, data from 176 two sites normally upwind of Manaus are also used in this study. These background sites include 177 the T0a site, (ATTO, 2° 8'47.88"S, 59° 0'18.00"W) (Andreae et al., 2015) and the T0t site, 178 (2.6091°S, 60.2093°W) (Martin et al., 2010a).

179 **2.2 Activated Fraction of size-selected Particles**

180 The CCN activation fraction of size-selected particles was measured using a Differential 181 Mobility Analyzer (DMA) coupled to a condensation particle counter (CPC, TSI Inc., 3010) and 182 a cloud condensation nuclei counter (CCNC, Droplet Measurement Technologies, Boulder, CO) 183 (Frank et al., 2006; Mei et al., 2013a; Moore et al., 2010; Petters et al., 2007). Aerosol particles were sampled with a total flow rate (Q_a) of 1.53 L min⁻¹ from a height of 5 m above ground level 184 185 and were dried to RH below 20% by a Nafion dryer immediately upon entering the instrument 186 container. The dried aerosol particles then reached steady state charge distribution inside a Kr-85 187 aerosol charger (TSI, model 3077A), prior to being introduced into the DMA operated with a sheath flow rate (Q_{sh}) of 15.3 L min⁻¹ to maintain a 10:1 sheath to aerosol flow ratio (Q_{sh}/Q_a) . 188 189 The aerosol particles were size-selected by the DMA and the size-selected particles were simultaneously characterized by a CPC ($Q_{CPC} = 0.53 \text{ Lmin}^{-1}$) and a CCNC (see Fig. S1 in the Sup-190 191 plementary Information for further details). This system had been operated in previous field 192 campaigns by scanning the particle size while CCNC supersaturation was held constant (Mei et 193 al., 2013a; Mei et al., 2013b). During GoAmazon2014/5 the particle size classified by the DMA 194 was stepped through seven particle diameters (51, 75, 94, 112, 142, 171, and 222 nm), while the CCNC supersaturation was also changed at each diameter by stepping the flow rates (Q_{CCN} rang-195 196 ing from 0.2 to 1.0 L min⁻¹) and temperature gradient ($\Delta T = 4.5, 5.5, 6.5, 8.0$ and 10.0°C). At a 197 given supersaturation, data were acquired for a minimum of 30 s and until 1500 particles were 198 counted by the CPC or up to a maximum time of 120 s. Depending on the aerosol number size 199 distribution, the measurement cycled through the seven particle sizes in 1 - 2 h (see Fig. S1 and 200 S2 for further details of the measurement setup and sampling protocol). The sampling sequence 201 was designed so that the change of CCNC supersaturation was mostly accomplished by stepping

202 flow rates, as the CCNC reaches steady state faster following flow changes than temperature 203 changes. Change of the temperature gradient was kept at a minimum frequency, but was neces-204 sary given the wide range of supersaturation explored. Given the low particle number concentration (e.g., $\sim 200 \text{ cm}^{-3}$ under background conditions during the wet season), these approaches were 205 206 important to achieve adequate counting statistics with good time resolution to capture changes of 207 air mass within 10 - 20 minutes (Liu et al., 2016). The supersaturation of the CCNC was cali-208 brated using ammonium sulfate aerosol, as described previously in the literature (e.g., Mei et al., 209 2013a), at each operational set point (Q_{CCN} and ΔT), ranging from 0.075 – 1.1 %. Fluctuation of 210 the temperature inside the instrument container, ranging from 20 to 30°C over the course of a 211 day, led to substantial variation in the absolute temperature inside the CCNC growth chamber. 212 Calibrations were therefore repeated under a range of container and associated growth chamber 213 temperatures. The dependence of the supersaturation on the temperature at the top of CCNC col-214 umn (instrument temperature T_1) was derived for each Q_{CCN} and ΔT pair and used to retrieve the 215 supersaturation over the range of the instrument operating conditions (see Fig. S3 and further 216 description in the Supplementary Information).

217 **2.3 Aerosol Chemical Composition**

Non-refractory sub-micron aerosol composition (organics, sulfate, nitrate, ammonium, and chloride) was measured by a high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS, Aerodyne Research Inc. (DeCarlo et al., 2006) during the two IOPs, and by an Aerosol Chemical Speciation Monitor (ACSM, Aerodyne Research, Inc., Ng et al., 2011) from July 2014 to March 2015. The AMS sampled from an inlet equipped with a PM2.5 cyclone located at 5 m above ground level. The ambient sample was first dried outside the container by a poly-tube Nafion dryer (Perma Pure, model PD-100T). Once inside the container the sample was further dried by a mono-tube Nafion dryer (Perma Pure, model MD-110) to achieve RH < 40% and was split between the AMS and a Scanning Mobility Particle Sizer (SMPS, TSI, model 3081). Ambient measurements were obtained every 4 of 8 minutes. Further details of the AMS set up and operation are described in de Sá et al. (2016).

229 The ACSM was a part of the Aerosol Radiation Measurement (ARM) Mobile Facility-1 230 (AMF-1) Mobile Aerosol Observation System (MAOS). Aerosol was sampled through an inlet 231 located 10 m above the ground. The aerosol sample was first dried through five large (40 x 1.75 232 cm I.D.) Nafion dryers before being distributed among various instruments including the ACSM. 233 The ACSM sampling alternated between with and without an in-line filter using a 3-way valve, 234 such that aerosol-free background could be subtracted from the ambient measurement. Twenty-235 eight ambient and background scans of the quadrupole mass spectrometer (unit mass resolution) 236 were averaged to give one measurement every 30 min. The mass concentrations of organic spe-237 cies, sulfate, nitrate, ammonium, and chloride were derived from measurements using approach-238 es described in Ng et al. (2011).

239 Refractory black carbon (rBC) was measured using both a Single Particle Soot Photometer 240 (SP2, Droplet Measurement Technologies, Boulder, CO) and an aethalometer (Magee Scientific) 241 co-located with the AMS and ACSM. The SP2 measures rBC using laser-induced incandescence, 242 whereas the aethalometer measures equivalent black carbon (BCe) (Andreae and Gelencsér, 243 2006) using light absorption from particles collected onto a filter. While these are fundamentally 244 different aerosol properties, both species (rBC and BC_e) were treated as equivalent in this study, 245 and BC_e concentration was adjusted to match that of rBC using the approach detailed in Section 246 2.2 of the Supplementary Information.

247 **2.4 Additional Relevant Measurements**

248 Additional measurements of aerosol microphysics, trace gas concentrations, and atmospheric 249 conditions used in this study are briefly described here. These measurements were part of the de-250 ployment of the ARM AMF-1 facility during GoAmazon 2014/5 (Martin et al., 2016b). Relevant 251 aerosol measurements include dry particle number-size distributions from 10 to 480 nm by a 252 SMPS (TSI Inc. Model 3081) and the number concentration of particles with diameters greater 253 than 10 nm by a CPC (TSI Inc. Model 3772). Mixing ratios of CO and O₃ were characterized by 254 an Off-Axis Integrated Cavity Output Spectroscopy CO, N₂O and H₂O analyzer (model number 255 908-0014, Los Gatos) and UV Photometry O₃ analyzer (model 49i, Thermo Scientific Inc.), re-256 spectively. Oxides of nitrogen (NO, NO_x, NO₂, NO_y) were measured using a chemiluminescence 257 technique (details given in Section S2.1 of the Supplementary Information). Meteorological data 258 included relative humidity (RH), ambient temperature, wind speed and direction, and rain accu-259 mulation. The vertical profile of atmospheric backscatter (clouds and aerosol) and boundary lay-260 er heights were estimated from ceilometer (model CL31, Vaisala) measurements.

261 3 Methods

262 **3.1 Derivation of Particle Hygroscopicity and Mixing State**

263 The particle hygroscopicity parameter, κ (Petters and Kreidenweis, 2007), was derived from 264 the activation spectrum (i.e., activated fraction as a function of supersaturation *S*) at the individu-265 al particle sizes using approaches detailed in the literature (Bougiatioti et al., 2011; Cerully et al., 266 2011; Lance et al., 2013; Mei et al., 2013a; Rose et al., 2008b). The activation spectrum of size-267 selected particles was first corrected for the influence of multiply charged particles, which is es-268 timated using the size distribution measured by the SMPS in MAOS and the activation spectrum 269 measured at the sizes of the doubly and triply charged particles (See Section S3.1). The corrected 270 activation spectrum of size selected particles was then fit with a cumulative lognormal (Mei et 271 al., 2013a; Rose et al., 2008a) functional form (See Fig. S6 for examples):

272
$$R_{a}(S) = \frac{E}{2} \left[1 + erf\left(\frac{\ln S - \ln S^{*}}{\sqrt{2\sigma_{s}^{2}}}\right) \right]$$
(1)

273 where R_a is the activated fraction as a function of supersaturation S, E is maximum activated 274 fraction, and (1-E) represents the number fraction of particles consisting of only non-hygroscopic 275 species (e.g., uncoated rBC) that cannot serve as CCN under typical atmospheric supersatura-276 tions. S* is the supersaturation at which R_a reaches 50% of E, and represents the median critical 277 supersaturation of size-selected particles that serve as CCN. The value of σ_s is related to the slope of the increasing R_a with S near S^{*} and reflects the heterogeneity of critical supersaturation, 278 279 which to a large degree arises from the heterogeneity of the hygroscopicity among size-selected 280 particles (Cerully et al., 2011; Mei et al., 2013a). The probability density function of hygroscopicity for size-selected particles is derived from the $R_a(S)$. The average hygroscopicity $\overline{\kappa_{\rm CCN}}$ and 281 dispersion of the hygroscopicity $\sigma(\kappa)/\overline{\kappa_{_{\rm CCN}}}$ for the size-selected CCN were then derived from 282 283 the probability density function of hygroscopicity using the approach detailed in Section S3.4 in 284 Supplementary Information. The dispersion of the hygroscopicity reflects the composition heter-285 ogeneity (i.e., mixing state) among size-selected particles (Mei et al., 2013a). For simplicity, we use $\kappa_{\rm CCN}$ to represent the average hygroscopicity of size-selected CCN in the following sections. 286 287 As hygroscopicities reported in this study were derived from particle dry diameter and critical 288 supersaturation, they represent "apparent hygroscopicity", which includes the potential impact 289 <u>due to the limited solubility of organics and the reduction of surface tension by surface active</u>
 290 <u>species (Sullivan et al., 2009).</u>

291

292 **3.2 Derivation of Organic Aerosol Hygroscopicity**

293 The average particle hygroscopicity was then combined with the chemical composition data to derive the hygroscopicity of the organic component of the size-selected particles, $\kappa_{\rm org}$. Collec-294 295 tively, the AMS, ACSM, SP2, and aethalometer provided mass concentrations of organic spe-296 cies, sulfate, nitrate, ammonium, and rBC. The concentration of chloride was negligible (<<1% 297 of aerosol mass) and was not included in the analysis. Given the low concentrations during 298 GoAmazon 2014/5, size-resolved mass concentrations at the time resolution of the CCN meas-299 urement were not directly derived from AMS particle time of flight (PToF) mode data. For IOP 300 1, measurements were classified into three groups based on bulk organic mass fraction and the 301 characteristic mass size distribution of each species was averaged from measurements in each 302 group. For the dry season, the measurements were classified into three groups each for day and 303 night periods based on the bulk aerosol organic mass fraction, and the mass size distribution of 304 each species was averaged from measurements in each of the six groups. The size-resolved mass 305 concentrations of sulfate, nitrate, and organics at the time resolution of the CCN measurement 306 were then derived by scaling the total mass concentration using the average mass size distribu-307 tions for the corresponding group (based on the bulk organic mass fraction) of either wet or dry season (de Sá, 2017). The shape of the NH₄⁺ mass size distribution was assumed to be the same 308 309 as that of sulfate, as ammonium cations were primarily associated with sulfate. rBC was as-310 sumed to have the same size distribution shape as the total aerosol mass (i.e., mass fraction of 311 rBC was independent of particle size); though this assumption may not always be appropriate, 312 the effect is expected to be very small as the monthly average volume fraction of rBC was al-313 ways less than 4% (Fig. 1). A detailed description of the derivation of the size-resolved mass 314 concentrations is given in Section S4.1.

In most cases, NH_4^+ was insufficient to completely neutralize SO_4^{2-} . The concentrations of 315 both organonitrate and inorganic nitrate during the two IOPs were retrieved from AMS data 316 based on the ratio of ions NO^+ and NO_2^+ (de Sá, 2017; Fry et al., 2009). When the inorganic ni-317 trate mass concentration was negligible (i.e., less than 30 ng m⁻³), as in most of the cases, the 318 319 contributions of ammonium sulfate and ammonium bisulfate were calculated based on the mass concentrations of SO_4^{2-} and NH_4^+ (Nenes et al., 1998). In rare cases when the mass concentra-320 tion of inorganic nitrate was greater than 30 ng m⁻³, sulfate was assumed to be ammonium sul-321 322 fate. During non-IOP periods, only the total nitrate mass concentration is available, all nitrate 323 was assumed to be organonitrate (Cerully et al., 2015; Lathem et al., 2013b; Nenes et al., 1998; 324 Zhang et al., 2005), and sulfate fully neutralized by ammonium (see Section S4.2 in the Supple-325 ment for a sensitivity study of these assumptions for non-IOP periods). The -ONO₂ portion of the organonitrate was added back to the organic mass. We note that the amount of -ONO₂ added 326 327 back was typically small given the low mass fraction of nitrate in the aerosol. Cloud condensa-328 tion nuclei were assumed to be internal mixtures of (NH₄)₂SO₄, NH₄HSO₄, NH₄NO₃, organics, 329 and rBC, and the volume fractions of the species were derived from the mass concentrations and 330 densities. Densities of organics were estimated from the ratios of O:C and H:C measured by the AMS (Kuwata et al., 2012) and were on average 1450 ± 100 and 1470 ± 80 kg m⁻³ for IOP1 and 331 IOP2, respectively. Densities of 1770, 1790, 1730, 1800 kg m⁻³ were used for (NH₄)₂SO₄, 332 NH₄HSO₄, NH₄NO₃, and rBC (Bond and Bergstrom, 2006; Park et al., 2004), respectively. In 333

334 very rare cases, E was less than 100%, suggesting some of the size-selected particles were non-335 hygroscopic. The non-hygroscopic particles were assumed to consist entirely of rBC (Mei et al., 336 2013a). The volume concentration of the non-hygroscopic particles was derived as the product of 337 (1-E) and the total volume concentration (i.e., the sum of volume concentrations of organics, 338 $(NH_4)_2SO_4$, NH_4HSO_4 , NH_4NO_3 and rBC_8) at the size classified by the DMA. The volume con-339 centration of rBC internally mixed within the CCN-active particles was then calculated as the 340 difference between the total rBC volume concentration and the volume concentration of the non-341 hygroscopic particles. (Mei et al., 2013a; Mei et al., 2013b). Assuming a κ value of zero for rBC, 342 we can derive the hygroscopicity of the organic component of the CCN κ_{org} as:

343
$$\kappa_{\rm org} = \frac{1}{x_{\rm org}} \left(\kappa_{\rm CCN} - x_{(\rm NH_4)_2 SO_4} \kappa_{(\rm NH_4)_2 SO_4} - x_{\rm NH_4 HSO_4} \kappa_{\rm NH_4 HSO_4} - x_{\rm NH_4 NO_3} \kappa_{\rm NH_4 NO_3} \right)$$
(2)

where x_i is the volume fraction of the respective species. The κ values are 0.61, 0.7 and 0.67 for (NH₄)₂SO₄, NH₄HSO₄, and NH₄NO₃, respectively (Petters and Kreidenweis, 2007). The uncertainty in κ_{org} using these calculations has been derived using the approach detailed in earlier studies (Mei et al., 2013a; Mei et al., 2013b) and is on the order of 0.01 – 0.02 (which was generally between 10 and 20%) for this data set.

349 **3.2.1 Derivation of κ for AMS PMF Factors**

PMF was applied to the AMS mass spectra (Lanz et al., 2008; Ulbrich et al., 2009), and six organic factors were identified for each of the two IOPs (de Sá, 2017). For IOP2 the PMF analysis included data from 24 August to 15 October 15, 2014, excluding a major regional biomass burning event from 16 to 23 August, which was treated separately in the PMF analysis. For IOP1 (wet season), the six factors were isoprene-epoxydiol-derived secondary organic aerosol (IE- 355 POX-SOA), more-oxidized oxygenated organic aerosol (MO-OOA, i.e., highly oxidized organ-356 ics), less-oxidized oxygenated organic aerosol (LO-OOA), biomass burning organic aerosol 357 (BBOA) with characteristic peaks at m/z = 60 and 73 and correlated with the concentrations of 358 levoglucosan and vanillin, a factor with high contribution from m/z = 91 (Fac91) and correlated 359 with anthropogenic emissions of aromatics, and hydrocarbon-like organic aerosol (HOA). The 360 six factors for IOP2 included IEPOX-SOA, MO-OOA, LO-OOA, an aged biomass burning or-361 ganic aerosol factor (aged-BBOA), a fresh biomass burning organic aerosol factor (fresh-362 BBOA), and HOA. Further details of PMF analysis and the characteristics of the factors can be 363 found in de Sá (2017). The O:C ratio and calculated density for each factor are presented in Ta-364 ble 1. In this study, the O:C ratio was derived using Improved-Ambient Method (Canagaratna et 365 al., 2015).

For each IOP, hygroscopicities associated with the six factors were attributed based on multilinear regression of κ_{org} with respect to the volume fractions of the factors (Levenberg-Marquardt algorithm, IGOR Pro, Wavemetrics):

$$369 \qquad \kappa_{\rm org} = \sum_{i}^{n} \kappa_{i} x_{i} \tag{3}$$

where κ_i and x_i are the hygroscopicity and volume fraction of the individual organic PMF factors. The volume fraction was derived from mass concentrations and the densities of the factors. κ_{org} represents the average organic hygroscopicity at particle diameters (D_p) of 142 and 171 nm. As the PMF analysis is based on the mass spectra of the bulk sub-micrometer aerosol (i.e., MS mode measurements), an implicit assumption of Eq. (3) is that the bulk volume fractions of the factors represented those over the sizes at which κ_{org} was derived (i.e., $D_p = 142$ and 171 nm). The va376 lidity of this assumption is discussed in the results section. The robustness of the factor hygro-377 scopicity derived through linear regression depends on the variation of the factor volume fraction 378 during the measurement period. The HOA hygroscopicity was assumed as zero based on the re-379 sults from previous studies (Cappa et al., 2009; Cappa et al., 2011; Jimenez et al., 2009), and the 380 hygroscopicity of the other five factors were derived by multilinear regression as described 381 above.

382 **3.3 Classification of Seasons and Air Masses**

383 The one-year sampling period was divided into different seasons by grouping months accord-384 ing to the similarity of the aerosol properties and trace gas concentrations measured at the two 385 background sites, T0a and T0t, as well as monthly accumulated rainfall. In this study, the sea-386 sons were defined as follows: the first wet season: March, April and May of 2014; the first tran-387 sition season: June and July 2014; the dry season: August and September of 2014; the second 388 transition season: October, November and the first half of December 2014; and the second wet 389 season: the second half of December 2014 and January, February and the first few days of March 390 2015.

For each season, the air masses arriving at the T3 site were classified into three different types: background, urban pollution, and local biomass burning based on trace gas and aerosol measurements at all sites. During the wet season, the background air mass represented near natural conditions, with occasional impact from anthropogenic emissions, while in the dry season, the background was dominated by regional biomass burning aerosol particles. Polluted air masses represent those with strong influence from urban emissions, which were mostly from Manaus. The local-biomass-burning type describes those air masses strongly influenced by local (i.e.,

398 fresh) biomass burning activities, which dominated over the impact from urban pollution, if any. 399 For each season, background conditions were identified when CO and CN concentrations were 400 below the thresholds derived from measurements at the background T0a and T0t sites, and the 401 NO_v mixing ratio was below 1.5 ppb. Non-background conditions were identified by condensa-402 tion nuclei (CN) and CO concentrations above the respective threshold levels. As biomass burn-403 ing aerosol typically has a higher fraction of accumulation mode particles, and the emissions 404 from Manaus were more dominated by Aitken mode particles, the fraction of particles with di-405 ameter less than 70 nm was used to differentiate air masses strongly influenced by local biomass 406 burning from those with more impact from urban pollution (see Table S2 in Supplementary In-407 formation for details). Contamination by the emissions from an on-site diesel generator, grass 408 cutting activities, tractors, and other vehicles were evidenced by rBC concentrations above 1.0 µg m⁻³ or CN concentration above 10,000 cm⁻³. Over the one-year measurement period, back-409 410 ground, urban pollution, and local biomass burning represented 12.4%, 38.5%, and 28.4% of the 411 CCN measurements, respectively (Table S3). We note that the air masses arriving at the T3 site 412 often included contributions from different sources. The classification of the air masses using the 413 above three types clearly represents a simplification, but is very helpful for understanding the 414 properties of aerosols influenced by the various major sources.

415 **4 Results**

416 **4.1 Seasonal Trend and Size Dependence of Hygroscopicity and Chemical Composition**

417 The monthly average κ_{CCN} at the T3 site varied from 0.1 to 0.2 at five particle diameters 418 ranging from 75 nm to 171 nm (Fig. 1a), and was substantially lower than the value of 0.3 ± 0.1 419 suggested for continental sites (Andreae and Rosenfeld, 2008). This was due to the large organic volume fraction, up to 95% observed at the T3 site. In this study, measurements at 51 and 222 nm were not included, because the range of supersaturation sampled inside the CCN counter only adequately captured the activation spectrum for 51/222 nm particles with relatively high/low $\kappa_{\rm CCN}$ values, leading to a positive/negative bias of the average $\kappa_{\rm CCN}$. The value of $\kappa_{\rm CCN}$ exhibited similar seasonal variations at all five sizes. During the transition from wet to dry season, $\kappa_{\rm CCN}$ decreased by 20-30% with the absolute minimum of 0.116 occurring at 75 nm in September and October (Fig. 1a).

427 The seasonal trend of $\kappa_{\rm CCN}$ was mainly driven by the variation of $\kappa_{\rm org}$, which shows the low-428 est value in September during the dry season (Fig. 1b). Despite a strong increase of aerosol mass 429 concentration from wet to dry season due to biomass burning emissions, the organic volume 430 fraction exhibited little season variation, and was ~90% or higher at the four sizes from 94 to 431 171 nm (Fig. 1c). A minor increase in organic volume fraction in October might have also contributed to the lower κ_{CCN} value observed. The species volume fractions at 75 nm are not shown 432 433 due to the very low signal-to-noise ratio of the AMS PToF data in the small particle diameter 434 range. No clear seasonal trend was observed for sulfate volume fraction, which ranged from 3% 435 to 9% at the four sizes. The lack of clear seasonal trends of sulfate and organic fractions are con-436 sistent with observations at the T0a site (Andreae et al., 2015). Nitrate and rBC represented a small fraction of aerosol volume, and were less than 1% and ~4%, respectively. 437

The average κ_{CCN} increased with increasing particle size for all three air mass types and during all the seasons (Fig. 2), consistent with decreasing organic volume fraction with increasing particle size (Fig. 1c). The κ_{CCN} at 75, 94, 112, 142, and 171 nm averaged for the one-year measurement period were 0.130 ± 0.028, 0.144 ± 0.039, 0.148 ± 0.043, 0.164 ± 0.046, and 0.167 ± 0.042 respectively. The value of κ_{CCN} and its size dependence under background conditions were

443	largely consistent among different seasons, and were in good agreement with those observed un-
444	der near natural conditions during the AMAZE-08 campaign at T0t in the wet season (Gunthe et
445	al., 2009) and during the one-year period from March 2014 to February 2015 at the background
446	T0a site (Pöhlker et al., 2016). For the air masses with strong influence from local biomass burn-
447	ing, the value of $\kappa_{\rm CCN}$ and its size dependence are consistent with the κ value derived from parti-
448	cle growth factor measurements in July 2001, during a "recent biomass burning period" of the
449	CLAIRE-2001 study (Rissler et al., 2004), which took place at Balbina, about 125 km northeast
450	of Manaus. In contrast, κ values derived from particle growth factor measurements from 11 Sep-
451	tember to 8 October 2002, during the dry period of the LBA-SMOCC (Rissler et al., 2006) are
452	substantially lower than κ_{CCN} observed at the T3 site for local biomass burning air masses at all
453	sizes. As LBA-SMOCC took place in the state of Rondônia in southwestern Amazonia with ex-
454	tensive biomass burning activities during the dry season, the difference in κ could be due to the
455	differences in fire condition and the type of vegetation burned. Previous studies show particles
456	sometime exhibit larger κ values for droplet activation (derived from CCN measurements under
457	supersaturated conditions) than for particle growth (derived from particle growth factor under
458	sub-saturated conditions), this could also contribute to the differences in κ values. Compared to
459	κ_{CCN} , κ_{org} was largely independent of particle size for all three air mass types, indicating that the
460	size dependence of $\kappa_{\rm CCN}$ was mainly due to the size dependence of the organic volume fraction
461	and particle composition (Fig. $1c - f$). During the dry season, aerosols classified as urban pollu-
462	tion and local biomass burning exhibited lower $\kappa_{\rm org}$ values compared to background aerosols,
463	contributing to the lower values of overall κ_{CCN} .

464 **4.2 Diel Trends of Particle and Organic Hygroscopicities**

The diel variations of aerosol properties are presented in Figs. 3-7 for different air masses during the two IOPs. Aerosol properties derived from the size-resolved CCN measurements, including κ_{CCN} , $\sigma_{\kappa_{\text{CCN}}}$, $\overline{\kappa}_{\text{CCN}}$, and κ_{org} , and the volume fractions of different species were averaged at the three largest sizes ($D_p = 112$, 142, and 171 nm). The fraction of organic mass at m/z = 44 (f_{44}) and O:C were derived from the AMS bulk measurements. Also shown are diel variations of planetary boundary layer height, CN, and aerosol volume concentrations based on 5-minute average data.

472 **4.2.1 Wet Season Aerosol**

473 Background conditions

474 Figures 3 and 4 show the diel variations of aerosol properties during the wet season of 2014 475 for background and urban pollution air masses, respectively. There were only 0.7% of data clas-476 sified as local biomass burning (see Table S3), which is insufficient to evaluate the diel trends. 477 During the wet season, the background air mass represents near-natural conditions, with mini-478 mum impact from anthropogenic emissions, although long-distance transport of African biomass 479 burning may contribute to the aerosols observed (Chen et al., 2009; Wang et al., 2016b). Back-480 ground aerosol constantly exhibited relatively high hygroscopicity of ~0.19 throughout the day. 481 The lack of a diel trend in $\kappa_{\rm CCN}$ is also in agreement with the results from the T0a site (ATTO), 482 which is upwind of Manaus and served as a background site (Fig. 8). The particle composition 483 averaged for the three particles diameters was dominated by organics, representing ~ 90% of the 484 aerosol volume. The lack of a diel trend in $\kappa_{\rm CCN}$ and $\kappa_{\rm org}$ suggest little variation in particle com-485 position throughout the day. The nearly constant κ_{org} of ~0.16 is also consistent with the lack of a 486 diel trend in f_{44} and O:C. The values of f_{44} and O:C are ~0.2 and ~0.8, respectively, indicating 487 that the aerosol under background condition during the wet season was dominated by the aged 488 regional aerosol particles consisting of highly oxygenated organic compounds.

Aerosol number and volume concentrations exhibited a minimum at \sim 310 cm⁻³ and \sim 0.3 μ m³ 489 cm⁻³, respectively, just before sunrise. The number and volume concentrations started increasing 490 after sun rise, and peaking at 400 cm⁻³ and 0.8 um³ cm⁻³ in the afternoon. These diel variations 491 492 are partially attributed to the wet scavenging of accumulation mode particles, which dominate 493 the submicron particle concentrations under background conditions, and the mixing of the parti-494 cles from the residual layer aloft down to the surface as the boundary layer develops in the morn-495 ing. During the night, the radiative cooling at the surface leads to a shallow nocturnal boundary 496 layer with low and variable winds. RH near surface was near 100%, and fog or mist was identi-497 fied by the weather station (Present Weather Detector, Visalia) 62% of the time during the one-498 year measurement period. The gradual decrease of particle number and volume concentration 499 during these fog events were due to the wet deposition of the accumulation mode particles acti-500 vated into droplets. Similar decreases of particle number concentration were previously reported 501 during night fog events in the tropical rainforest in Borneo (Whitehead et al., 2010). After sun-502 rise, the boundary layer deepened on average from less than 200 up to 800 m as a result of solar 503 heating (Fig. 3g). Consequently, particles in the residual layer aloft (Fisch et al., 2004; Rissler et 504 al., 2006b), which were not impacted by the fog, were mixed down to the surface, leading in part 505 to the observed increases in both number and volume concentrations. Such mixing of particles 506 from the residual layer in the morning had been observed previously in the Amazon basin during 507 the dry season (Rissler et al., 2006a). The formation of secondary organic aerosol (SOA) as a 508 result of photochemical oxidation (in both gas and particle phases) likely contributed to the increase in volume concentration (Chen et al., 2015; Chen et al., 2009; Martin et al., 2010a; Pöschl
et al., 2010).

511 Air masses impacted by urban pollution

512 Air masses arriving at the T3 site frequently had passed over urban and industrial areas up-513 wind. When the air mass was influenced by the urban pollution, $\kappa_{\rm CCN}$ and its dispersion exhibited 514 clear diel variations (Fig. 4). The value of $\kappa_{\rm CCN}$ was lower during the night at 0.15, and it in-515 creased from the early morning hours, peaking at a value of 0.19 around noon (local time, UTC – 516 4 h). The dispersion was anti-correlated with $\kappa_{\rm CCN}$, exhibiting higher values (i.e., increased het-517 erogeneity in particle chemical composition) during night and a minimum value around noon. To a large degree, the diel trend of $\kappa_{\rm CCN}$ was due to the variation of $\kappa_{\rm org}$. The value of $\kappa_{\rm org}$ was lower 518 519 during night at 0.10 and increased to 0.16 at noon. The increase of κ_{org} is consistent with the var-520 iation of O:C, which increased during the early morning and reached the highest value of 0.8 521 around noon time. The pollution strongly affected the particle number and volume concentra-522 tions, both exhibiting similar diel trends. Under polluted conditions, particle number concentration ranged on average from 1500 to 2300 cm⁻³, an increase by a factor of ~5 from that under 523 524 background conditions. In comparison, the increase of the volume concentration was only about a factor of 2 (i.e., from a range of 0.3 - 0.75 to 1.0 to 1.3 μ m³ cm⁻³), as the urban pollution is 525 526 dominated by Aitken mode particles that make a relatively small contribution to aerosol mass 527 and volume concentration.

528 The diel variations of κ_{CCN} , its dispersion, κ_{org} , and O:C are explained as follows. During 529 night, particles in freshly emitted pollution, which are dominated by primary organic aerosol 530 (POA) and have low hygroscopicity, are mixed with more aged particles within a shallow noc-

531 turnal boundary layer (Bateman et al., 2016). In the absence of photochemical oxidation and ag-532 ing, this external mixture leads to higher dispersion of particle hygroscopicity, as well as overall 533 lower O:C and κ_{org} . As the pollution aerosols are mainly from isolated point sources, they are 534 confined in the shallow nocturnal boundary layer during night, and the residual layer above the 535 T3 site is expected to consist of aged background aerosols. Therefore, unlike under background 536 conditions, the mixing of aerosol aloft in the residual layer down to the surface cannot by itself 537 explain the increase in particle number and volume concentration during the day, both of which 538 were substantially above the background values. These increases under polluted conditions might 539 be due to the stronger urban influence at T3 during the day. The strong increase of CN concen-540 tration at 16:00 UTC (local time 12:00) could be caused by the arrival at the T3 site of the Ma-541 naus plume emitted during early morning traffic hours. To a large degree, the increases in O:C 542 and $\kappa_{\rm org}$ are due to the formation and aging of SOA in the pollution, while the development of the 543 daytime boundary layer, which leads to dilution of pollution and mixing with aged particles from 544 the residual layer, can also contribute to the increases. The condensation of secondary species 545 and photochemical aging also lead to more homogenous composition among particles (Mei et al., 546 2013a), and therefore lower dispersion of κ_{CCN} , as was observed. The O:C reached a maximum 547 average value of 0.8, similar to that under the background conditions. This suggests that the for-548 mation and photochemical aging of SOA quickly led to highly oxygenated organic compounds 549 (i.e. within several hours) (de Sá, 2017).

550 **4.2.2 Dry Season Aerosol**

551 Background conditions

552 Figures 5-7 show the diel variations of the aerosol properties observed during the dry season 553 for background, urban pollution, and local biomass burning air masses, respectively. During the 554 dry season, the background aerosol (Fig. 5) is strongly influenced by regional biomass burning, 555 and air masses arriving at the T3 site often pass through urban and industrial areas along the 556 Amazon River and in northeast Brazil (Andreae et al., 2015), indicating that the background aer-557 osol is also impacted by more aged urban and industrial emissions (Martin et al., 2016a). Despite different aerosol sources and processes, the particle hygroscopicity, dispersion, and κ_{org} exhibited 558 559 similar values as those of background aerosol during the wet seasons, and a lack of obvious diel 560 variations. This is also consistent with the absence of a significant diel trend of κ_{CCN} observed at 561 T0a (ATTO) during the dry season (Fig. 8). The O:C value increased slightly from 0.8 during 562 night to 0.9 in the afternoon, possibly due to further oxidation and aging of background aerosols 563 during the day time. The high value of O:C is consistent with the relatively high value of κ_{org} 564 (0.15) and is close to that observed under background condition during the wet season, indicating 565 highly oxygenated organic aerosol. The number and volume concentrations were lower just be-566 fore dawn and increased during the early morning, again a result of wet scavenging of particles 567 by fog followed by the mixing of aerosol aloft in the residual layer down to the surface as the 568 boundary layer develops in the morning.

569 Air masses impacted by urban pollution and local biomass burning

570 For urban pollution (Fig. 6) and local biomass burning (Fig. 7) air masses during the dry sea-571 son, κ_{CCN} (urban pollution: 0.12 – 0.20, local biomass burning: 0.10 – 0.17), its dispersion (urban

pollution: 0.4 – 1.0, local biomass burning: 0.4 – 0.9), κ_{org} (urban pollution: 0.10 – 0.15; local 572 573 biomass burning: 0.08 - 0.14), and O:C (polluted: 0.7 - 0.85; biomass burning: 0.7 - 0.86) 574 showed similar values and diel variations as those under polluted conditions during the wet sea-575 son. This is consistent with the picture that freshly emitted particles (in either the case of urban 576 pollution or local biomass burning) lead to overall lower O:C, κ_{org} , and higher κ_{CCN} dispersion 577 during night, followed by increases in O:C, κ_{org} , and a decrease in the dispersion during day time, 578 which are mainly driven by the formation and photochemical aging of SOA, with contributions 579 from the mixing of background aerosol aloft in the residual layer down to the surface and dilu-580 tion of fresh emission as the boundary layer develops. Compared to urban pollution, local bio-581 mass burning air masses exhibited lower κ_{CCN} and κ_{org} values during night and stronger diel vari-582 ations. In the afternoon, κ_{org} and O:C reached high values of 0.14 and 0.86, respectively, as ob-583 served for background aerosols.

584 For urban pollution air masses, aerosol number and volume concentrations showed similar 585 trends for both seasons. The increases of number and volume concentration from early morning to noon were similar, about 1000 cm⁻³ and 0.5 μ m³/cm³, respectively, for both wet and dry sea-586 587 sons. The percentage increases were less pronounced in the dry season due to the higher back-588 ground values. In contrast to urban pollution, local biomass burning showed higher aerosol num-589 ber and volume concentrations at night, and decreased during the morning. Local biomass burn-590 ing activities typically peaked during evening hours, consistent with frequent classification of the 591 night time aerosol as local biomass burning (Fig. 7i) (Vestin et al., 2007). Despite the wet re-592 moval of particles by fog, the strong emission from local biomass burning, largely confined with-593 in the shallow nocturnal boundary layer, led to higher surface aerosol concentrations than those 594 in the residue layer aloft, which likely represented the regional background. As the boundary
layer deepened in the morning, the mixing with aerosol from the residual layer led to decreasesin both aerosol number and volume concentration observed at the surface (Fig. 7e,f).

4.3 Hygroscopicity of PMF Factors and the Variation of Organic Hygroscopicity with Oxi-

598 **dation Level**

597

599 The hygroscopicities associated with the AMS PMF factors were estimated through multi-600 variable linear regression using different subsets of the data, as well as the entire dataset for each 601 of the two IOPs (IOP1 in Fig. 9 and IOP2 in Fig. 10). The different subsets included measure-602 ments during day, night, certain sampling periods, and ranges in particle hygroscopicity disper-603 sion. Comparison of the hygroscopicities derived from the different subsets of data allowed us to 604 examine the robustness of this approach. Uncertainty in the derived κ for individual factors was 605 determined by the number of points available to fit in the time series, with greater data coverage 606 and therefore lower uncertainty during the dry season. For the wet season (IOP1), the hygrosco-607 picity associated with PMF factors derived using different subsets of the data are largely in 608 agreement with those derived from the entire dataset. There are notable differences between the 609 hygroscopicities of MO-OOA and Fac91 factors derived using data under background conditions 610 only and those derived using the entire dataset. Such difference could be partially due to the lim-611 ited data under the background conditions during IOP1 (Fig. 3). For the dry season (IOP2), the 612 hygroscopicities of PMF factors derived using measurements under background conditions or 613 data with hygroscopicity dispersion less than 0.4 are quite different from those derived using 614 other data subsets and the entire dry season dataset. The agreement among the PMF factor hy-615 groscopicities derived using different sub datasets during the wet season and the disagreements 616 for the dry season are attributed to the applicability of the underlying assumption that the bulk 617 volume fractions of PMF factors (i.e., derived from MS mode data) represented those at the sizes 618 of CCN measurements. For the wet season, the average f_{44} was largely independent of particle 619 size from 130 to 400 nm (Fig. S10), the size range that dominated bulk aerosol mass concentra-620 tion measured by AMS. This is consistent with the assumption that the bulk volume fractions of 621 the PMF factors represent those at the two CCN sizes (142 and 171 nm). For the dry season, the 622 f_{44} averaged over local biomass burning air masses and the entire IOP2 exhibited an increase 623 with particle diameter from 100 to 300 nm (Fig. S10). For periods with hygroscopicity dispersion 624 less than 0.4 or under background conditions, the average size distribution of f_{44} was noisier due to fewer data points. Nevertheless, the size distribution shows f_{44} was largely independent of the 625 626 particle size under these conditions, consistent with the assumption described above. In the fol-627 lowing discussion, we focus on the PMF factor hygroscopicities derived using all data during the 628 wet season, and under background conditions in the dry season.

629 The MO-OOA factors for the two IOPs exhibit very similar O:C and κ values. The O:C val-630 ues were 1.19 and 1.24, and κ values were 0.20 and 0.21 for IOP1 and 2, respectively (Table 1). 631 The O:C and κ values are consistent with those of some typical SOA compounds, such as malo-632 nic acid, which has an O:C value of 1.33 and a κ value of 0.23 (Kumar et al., 2003), and succinic 633 acid, which has an O:C value of 1 and a κ value of 0.23 (Hori et al., 2003). For the LO-OOA and 634 IEPOX-SOA factors, the hygroscopicities vary between the two IOPs. The κ of IEPOX-SOA were 0.18 and 0.08 during IOP1 and IOP2, respectively, and the κ of LO-OOA factor varied 635 636 from 0.12 to 0.20 between IOP 1 and IOP 2. The difference in κ may be partially due to the 637 change of O:C values of the factors derived for the two IOPs. Difference in SOA precursors and 638 therefore composition in LO-OOA (Ng et al., 2010) may also contribute to the difference in its κ 639 values between the two IOPs. The variation of IEPOX-SOA κ between the two IOPs could be a result of the different RH conditions, which may strongly influence the composition of IEPOX-SOA (Riva et al., 2016).

During the wet season, a factor with high contribution from m/z = 91 (Fac91) was identified. The Fac91 factor correlates with several tracers for anthropogenic emissions, including NO_x, benzene, toluene, trimethylbenzene (TMB) and xylenes, but not high NO_x isoprene products (e.g. methylglyceric acid) (de Sá, 2017). This factor likely represents SOA formed from aromatics emitted from urban areas, possibly combined with a mixture of freshly oxidized biogenic compounds within the urban-influenced air (de Sá, 2017). The Fac91 factor has a hygroscopicity value of 0.10, and a much lower O:C ratio of 0.328 compared to those of MO-OOA and LO-OOA.

649 Less oxidized organic factors identified by the PMF analysis were HOA of both IOPs, 650 BBOA of IOP1 and fresh and aged BBOA of IOP2. These factors represent primary OA (POA), 651 except that the aged BBOA of IOP2 likely included contributions from oxidized POA or SOA. 652 The hygroscopicity of the HOA factors was fixed as zero in the multivariate regressions. All 653 BBOA factors have a distinctive m/z = 60 peak and correlate with biomass burning traces includ-654 ing levoglucosan and vanillin (de Sá, 2017). The retrieved hygroscopicity values for the BBOA 655 factors are substantially lower than those of SOA factors, especially for the fresh BBOA factor 656 during IOP2. The extremely low hygroscopicity suggests that the fresh BBOA, likely produced 657 by local fires, behaves very similar to HOA in term of CCN activation despite a substantially 658 higher O:C.

Figures 11 and 12 show that for SOA factors, including IEPOX-SOA, LO-OOA, MO-OOA for both IOPs and Fac91 for IOP1, the κ value increases with increasing O:C, and the variation of κ with O:C agrees with the linear relationship derived from laboratory studies of SOA CCN ac662 tivities (Lambe et al., 2011). The low hygroscopicity of the HOA and the BBOA factors, which 663 are below the linear relationship for SOAs, are also consistent with laboratory results of POA 664 and oxidized POA (Lambe et al., 2011). Cerully et al. (2015) derived κ of LO-OOA, MO-OOA, 665 and IEPOX-SOA factors from data collected in the southeast US during Southern Oxidant and 666 Aerosol Study (SOAS). A different name, Isoprene-OA, was used for the IEPOX-SOA factor in 667 Cerully et al. (2015), as while this factor is mainly attributed to SOA formed from IEPOX up-668 take, it might not be entirely due to IEPOX (Schwantes et al., 2015; Xu et al., 2015a; Xu et al., 669 2015b). The O:C values calculated using the Improved-Ambient method (Canagaratna et al., 670 2015) are 0.59, 0.61, and 0.92 for the IEPOX-SOA, LO-OOA, and MO-OOA factors reported in 671 Cerully et al. (2015), respectively (personal communication, L. Xu and N.L. Ng). For the LO-672 OOA, MO-OOA factors reported in Cerully et al. (2015), κ and O:C values are largely consistent 673 with the linear relationship between κ and O:C derived from Lambe et al. (2011). Cerully et al. 674 (2015) also reported IEPOX-SOA (i.e., called Isoprene-OA in their study) factor κ of 0.2, simi-675 lar to 0.18 derived for IOP1. The O:C value of the IEPOX-SOA factor during the SOAS study 676 was 0.59, somewhat lower than those derived from both IOPs of GoAmazon 2014/5. While the IEPOX-SOA factors identified using different datasets share many similar features, they are not 677 678 identical and can consist of different groups of compounds. Such differences may be due to vary-679 ing degrees of oxidation in different environment between the two field campaigns.

For comparison with earlier field studies, the values of κ_{org} and O:C were averaged according to the hours of the day over particle diameters of 142 and 171 nm for data under polluted conditions during IOP1 and all data during IOP2 (Figs. 11 and 12). For the one-hour diel averages, the slope of κ_{org} vs. O:C, derived through a bivariate least squares fit (i.e., Orthogonal distance regression), is steeper than that derived from laboratory studies of SOA hygroscopicity, especially

685 during IOP2. This steep slope during IOP2 is consistent with the results from earlier field studies 686 (Mei et al., 2013b), although there is a clear offset between the two relationships. The O:C ratios 687 from Mei et al. (2013b) and Lambe et al. (2011) were scaled by a factor of 1.27 to account for 688 changes in the method of calculating the O:C ratio (Improved-Ambient, Canagaratna et al., 2015) 689 while all O:C values from this work were calculated using the Improved-Ambient method. This 690 offset between the field studies may be partially due to the different precursors of the SOA for 691 the campaigns, with higher anthropogenic VOC fraction expected for CalNex and CARES, 692 which took place near Los Angeles and Sacramento, respectively. In addition, biomass burning 693 represented a much smaller fraction of the organics during CalNex and CARES (Mei et al., 694 2013a; Mei et al., 2013b). The factors associated with secondary processes (e.g., MO-OOA, LO-695 OOA, and IEPOX-SOA), which have higher O:C values, exhibited higher volume fraction dur-696 ing the day, whereas the factors associated with primary emissions (e.g. HOA and BBOA), 697 which have lower O:C, had higher volume fractions during the night (de Sá, 2017). As a result, 698 the diel trend of overall O:C was to a large degree driven by the variations in volume fractions of 699 the POA and SOA factors with very different O:C values. This is in contrast to laboratory stud-700 ies, in which the increase of O:C was mainly driven by oxidation. As POA exhibits hygrosco-701 picity values well below the linear fit between SOA hygroscopicity and O:C, mixtures with dif-702 ferent POA and SOA fractions lead to a steeper slope for the increase of κ_{org} with O:C, as shown 703 by the results from this and previous field studies (Mei et al., 2013b).

704 **5 Conclusions**

Size-resolved CCN spectra at five particle diameters ranging from 75 to 171 nm were
characterized down-wind of Manaus, Brazil, in central Amazonia for a period of one year from
March 12, 2014 to March 3, 2015 during GoAmazon2014/5. For each season, the air masses ar-

708 riving at the site were classified into different types, including background, urban pollution, and 709 local biomass burning. During the wet season, the background air mass represented near natural 710 conditions, at times with impact from anthropogenic emissions, while in the dry season, the 711 background was dominated by regional and long-distance biomass burning aerosol particles. Pol-712 luted air masses represented those with strong influence from urban emissions, which were most-713 ly from Manaus. The local-biomass-burning type describes those strongly influenced by local 714 (i.e., fresh) biomass mass burning activities that dominate over the impact from urban pollution, 715 if any.

716 Particle hygroscopicity (κ_{CCN}), mixing state, and organic hygroscopicity (κ_{org}) were derived 717 from size-resolved CCN activation fraction and concurrent aerosol composition measurements. 718 The monthly mean κ_{CCN} exhibits the lowest values during the dry season, largely due to lower $\kappa_{\rm org}$ when aerosol was often strongly influenced by local biomass burning. The $\kappa_{\rm CCN}$ increased 719 720 with particle size during all seasons, consistent with decreasing organic volume fraction with in-721 creasing particle size. Under background conditions, the value of κ_{CCN} and its size dependence 722 were largely consistent among different seasons, despite the very different aerosol sources. Dur-723 ing the dry season, aerosols classified as urban pollution and local biomass burning exhibited 724 lower κ_{org} values compared to background aerosols, contributing to the lower values of overall 725 $\kappa_{\rm CCN}$.

Under background conditions during both wet and dry seasons, the largely constant diel trends of κ_{CCN} and κ_{org} suggest little variation in particle composition throughout the day. The constant κ_{org} of ~ 0.15 is consistent with the lack of a diel trend in f_{44} and O:C. The high values of f_{44} and O:C indicate that the aerosols under background conditions are dominated by the aged regional aerosol particles consisting of highly oxygenated organic compounds. When the air 731 mass is influenced by urban pollution or local biomass burning, κ_{CCN} , κ_{org} , f_{44} , and O:C exhibit 732 clear diel variations. The value of κ_{CCN} (0.1 - 0.2) is lower during the night, and increases from 733 the early morning hours, peaking around noon (local time, UTC - 4 h). This diel trend of κ_{CCN} is 734 largely driven by the variation in κ_{org} (0.08-0.15), consistent with the variation of O:C. The dis-735 persion of $\kappa_{\rm CCN}$ is anti-correlated with $\kappa_{\rm CCN}$, exhibiting higher values during night and a mini-736 mum value around noon, indicating an increased heterogeneity in particle chemical composition 737 during night time. These diel variations for air masses strongly influenced by urban pollution and 738 local biomass burning indicate that during night, freshly emitted particles, dominated by POA 739 and with low hygroscopicity, are mixed with more aged particles within a shallow nocturnal 740 boundary layer. In the absence of photochemical oxidation and aging, this external mixture leads 741 to higher dispersion of particle hygroscopicity, as well as overall lower O:C and κ_{org} . The in-742 creases in O:C and $\kappa_{\rm org}$ during daytime are driven by the formation and aging of SOA and dilu-743 tion of POA emissions into a deeper boundary layer, while the development of the boundary lay-744 er, which leads to mixing with aged particles from the residual layer, likley also contributes to 745 the increases.

746 The hygroscopicities associated with individual PMF organic factors were derived through 747 multi-variable linear regression. For the SOA factors, κ increases within increasing O:C, and the 748 variation of κ with O:C agrees well with the linear relationship derived from laboratory studies of 749 SOA hygroscopicity (Lambe et al., 2011). The low hygroscopicity of HOA and the BBOA fac-750 tors, which are below the linear relationship, are also consistent with laboratory results of POA 751 and oxidized POA (Lambe et al., 2011). In contrast, the slope of κ_{org} (i.e., overall organic hygro-752 scopicity) vs O:C is much steeper when compared to that derived from laboratory studies of 753 SOA hygroscopicity, especially for IOP2. Such difference is because the increase of O:C was

754	driven primarily by oxidation in laboratory SOA studies, while the variation in O:C of ambient
755	organics is to a large degree due to the variations in volume fractions of POA and SOA factors,
756	which have very different O:C values. As POA factors show hygroscopicity values well below
757	the linear fit between SOA hygroscopicity and O:C, mixtures with different POA and SOA frac-
758	tions lead to a steeper slope for the increase of κ_{org} with O:C, as shown by the results from this
759	and earlier field studies (Mei et al., 2013).

762 Acknowledgements

763 We acknowledge the support from the Central Office of the Large Scale Biosphere Atmosphere 764 Experiment in Amazonia (LBA), the National Institute of Amazonian Research (INPA), Amazo-765 nas State University (UEA), and the Max Planck Society (MPG). The Office of Biological and 766 Environmental Research of the Office of Science of the United States Department of Energy is 767 acknowledged for funding, specifically the Atmospheric Radiation Measurement (ARM) Climate 768 Research Facility and the Atmospheric System Research (ASR) Program. The work was con-769 ducted under scientific license 001030/2012-4 of the Brazilian National Council for Scientific 770 and Technological Development (CNPq). NLN acknowledged support from NSF grant 1242258 771 and US EPA STAR grant RD-83540301. This publication's contents are solely the responsibility 772 of the grantee and do not necessarily represent the official views of the US EPA. Further, US 773 EPA does not endorse the purchase of any commercial products or services mentioned in the 774 publication. PA and HB acknowledge the support from FAPESP under research grants 775 13/50510-5 and 13/05014-0, and from Royal Society under research grant NA 140450.

777 **References**

Albrecht, B. A.: Aerosols, Cloud Microphysics, and Fractional Cloudiness, Science, 245, 1227-1230, 1989.

780 Andreae, M. O., Acevedo, O. C., Araùjo, A., Artaxo, P., Barbosa, C. G. G., Barbosa, H. M. J., 781 Brito, J., Carbone, S., Chi, X., Cintra, B. B. L., Silva, N. F. d., Dias, N. L., Dias-Júnior, C. Q., 782 Ditas, F., Ditz, R., Godoi, A. F. L., Godoi, R. H. M., Heimann, M., Hoffmann, T., Kesselmeier, 783 J., Könemann, T., Krüger, M. L., Lavric, J. V., Manzi, A. O., Moran-Zuloaga, D., Nölscher, A. C., Nogueira, D. S., Piedade, M. T. F., Pöhlker, C., Pöschl, U., Rizzo, L. V., Ro, C.-U., 784 785 Ruckteschler, N., Sá, L. D. A., Sá, M. D. O., Sales, C. B., Santos, R. M. N. D., Saturno, J., 786 Schöngart, J., Sörgel, M., Souza, C. M. d., Souza, R. A. F. d., H. Su, N., Targhetta, Tóta, J., 787 Trebs, I., Trumbore, S., Eijck, A. v., Walter, D., Wang, Z., Weber, B., Williams, J., Winderlich, J., Wittmann, F., Wolff, S., and Yáñez-Serrano, A. M.: The Amazon Tall Tower Observatory 788 789 (ATTO): overview of pilot measurements on ecosystem ecology, meteorology, trace gases, and 790 aerosols, Atmospheric Chemistry and Physics, 15, 10723-10776, 2015.

Andreae, M. O. and Gelencsér, A.: Black carbon or brown carbon? The nature of light-absorbing
 carbonaceous aerosols, Atmos. Chem. Phys., 6, 3131-3148, 2006.

Andreae, M. O. and Rosenfeld, D.: Aerosol-cloud-precipitation interactions. Part 1. The nature and sources of cloud-active aerosols, Earth-Science Reviews, 89, 13-41, 2008.

Asa-Awuku, A., Engelhart, G. J., Lee, B. H., Pandis, S. N., and Nenes, A.: Relating CCN
activity, volatility, and droplet growth kinetics of beta-caryophyllene secondary organic aerosol,
Atmos. Chem. Phys., 9, 795-812, 2009.

Bateman, A. P., Gong, Z., Harder, T. H., de Sá, S. S., Wang, B., Castillo, P., China, S., Liu, Y.,
O'Brien, R. E., Palm, B., Shiu, H.-W., Silva, G. d., Thalman, R., Adachi, K., Alexander, M. L.,
Artaxo, P., Bertram, A. K., Buseck, P. R., Gilles, M. K., Jimenez, J. L., Laskin, A., Manzi, A. O.,
Sedlacek, A., Souza, R. A. F., Wang, J., Zaveri, R., and Martin, S. T.: Anthropogenic influences
on the physical state of submicron particulate matter over a tropical forest, 2016. submitted,
2016.

Batistella, M., Artaxo, P., Nobre, C., Bustamante, M., and Luizão, F.: Results from LBA and a
Vision for Future Amazonian Research. In: Amazonia and Global Change, American
Geophysical Union, 2013.

807 Bond, T. C. and Bergstrom, R. W.: Light Absorption by Carbonaceous Particles: An 808 Investigative Review, Aerosol Science and Technology, 40, 27-67, 2006.

809 Bougiatioti, A., Nenes, A., Fountoukis, C., Kalivitis, N., Pandis, S. N., and Mihalopoulos, N.:

- 810 Size-resolved CCN distributions and activation kinetics of aged continental and marine aerosol,
- 811 Atmospheric Chemistry and Physics, 11, 8791-8808, 2011.
- Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P.,
 Hildebrandt Ruiz, L., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue, N. M.,
 Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of organic compounds using

- 815 aerosol mass spectrometry: characterization, improved calibration, and implications, Atmos.
 816 Chem. Phys., 15, 253-272, 2015.
- Cappa, C. D., Bates, T. S., Quinn, P. K., and Lack, D. A.: Source characterization from ambient
 measurements of aerosol optical properties, Geophysical Research Letters, 36, n/a-n/a, 2009.
- Cappa, C. D., Che, D. L., Kessler, S. H., Kroll, J. H., and Wilson, K. R.: Variations in organic
 aerosol optical and hygroscopic properties upon heterogeneous OH oxidation, Journal of
 Geophysical Research: Atmospheres, 116, n/a-n/a, 2011.
- 621 Ocophysical Research. Athospheres, 110, 11/a-11/a, 2011.
- 822 Carslaw, K. S., Lee, L. A., Reddington, C. L., Pringle, K. J., Rap, A., Forster, P. M., Mann, G.
- 823 W., Spracklen, D. V., Woodhouse, M. T., Regayre, L. A., and Pierce, J. R.: Large contribution of 824 natural aerosols to uncertainty in indirect forcing, Nature, 503, 67-71, 2013.
- 825 Cerully, K. M., Bougiatioti, A., Hite, J. R., Guo, H., Xu, L., Ng, N. L., Weber, R., and Nenes, A.:
- 826 On the link between hygroscopicity, volatility, and oxidation state of ambient and water-soluble
- aerosols in the southeastern United States, Atmospheric Chemistry and Physics, 15, 8679-8694,
- 828 2015.
- 829 Cerully, K. M., Raatikainen, T., Lance, S., Tkacik, D., Tiitta, P., Petäjä, T., Ehn, M., Kulmala,
- 830 M., Worsnop, D. R., Laaksonen, A., Smith, J. N., and Nenes, A.: Aerosol hygroscopicity and
- 831 CCN activation kinetics in a boreal forest environment during the 2007 EUCAARI campaign,
 832 Atmos. Chem. Phys., 11, 12369-12386, 2011.
- 833 Chang, R. Y. W., Slowik, J. G., Shantz, N. C., Vlasenko, A., Liggio, J., Sjostedt, S. J., Leaitch,
- W. R., and Abbatt, J. P. D.: The hygroscopicity parameter (κ) of ambient organic aerosol at a field site subject to biogenic and anthropogenic influences: relationship to degree of aerosol
- 836 oxidation, Atmospheric Chemistry and Physics, 10, 5047-5064, 2010.
- Chen, Q., Farmer, D. K., Rizzo, L. V., Pauliquevis, T., Kuwata, M., Karl, T. G., Guenther, A.,
 Allan, J. D., Coe, H., Andreae, M. O., Pöschl, U., Jimenez, J. L., Artaxo, P., and Martin, S. T.:
- Submicron particle mass concentrations and sources in the Amazonian wet season (AMAZE-08),
 Atmos. Chem. Phys., 15, 3687-3701, 2015.
- Chen, Q., Farmer, D. K., Schneider, J., Zorn, S. R., Heald, C. L., Karl, T. G., Guenther, A.,
 Allan, J. D., Robinson, N., Coe, H., Kimmel, J. R., Pauliquevis, T., Borrmann, S., Pöschl, U.,
 Andreae, M. O., Artaxo, P., Jimenez, J. L., and Martin, S. T.: Mass spectral characterization of
 submicron biogenic organic particles in the Amazon Basin, Geophysical Research Letters, 36,
 n/a-n/a, 2009.
- Balch, J. K., Brown, I. F., C. Bustamante, M. M.,
 Coe, M. T., DeFries, R. S., Keller, M., Longo, M., Munger, J. W., Schroeder, W., Soares-Filho,
 B. S., Souza, C. M., and Wofsy, S. C.: The Amazon basin in transition, Nature, 481, 321-328,
 2012.
- de Sá, S. S.: Atmos. Chem. Phys. Discuss., in preparation, 2017.

- de Sá, S. S., Palm, B. B., Campuzano-Jost, P., Day, D. A., Newburn, M. K., Hu, W., Isaacman-
- 852 VanWertz, G., Yee, L. D., Thalman, R., Brito, J., Carbone, S., Artaxo, P., Goldstein, A. H.,
- 853 Manzi, A. O., Souza, R. A. F., Mei, F., Shilling, J. E., Springston, S. R., Wang, J., Surratt, J. D.,
- Alexander, M. L., Jimenez, J. L., and Martin, S. T.: Influence of urban pollution on the production of organic particulate matter from isoprene epoxydiols in central Amazonia, Atmos.
- 256 Cham Dhys Discuss 2016 1 58 2016
- 856 Chem. Phys. Discuss., 2016, 1-58, 2016.
- B57 DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C., Gonin,
 M., Fuhrer, K., Horvath, T., Docherty, K. S., Worsnop, D. R., and Jimenez, J. L.: FieldDeployable, High-Resolution, Time-of-Flight Aerosol Mass Spectrometer, Analytical Chemistry,
 78, 8281-8289, 2006.
- Buplissy, J., DeCarlo, P. F., Dommen, J., Alfarra, M. R., Metzger, A., Barmpadimos, I., Prevot,
 A. S. H., Weingartner, E., Tritscher, T., Gysel, M., Aiken, A. C., Jimenez, J. L., Canagaratna, M.
 R., Worsnop, D. R., Collins, D. R., Tomlinson, J., and Baltensperger, U.: Relating
 hygroscopicity and composition of organic aerosol particulate matter, Atmospheric Chemistry
 and Physics, 11, 1155-1165, 2011.
- Buplissy, J., Gysel, M., Alfarra, M. R., Dommen, J., Metzger, A., Prevot, A. S. H., Weingartner,
 E., Laaksonen, A., Raatikainen, T., Good, N., Turner, S. F., McFiggans, G., and Baltensperger,
 U.: Cloud forming potential of secondary organic aerosol under near atmospheric conditions,
 Geophys. Res. Lett., 35, L03818, 2008.
- Bronn, S., and Pöschl, U.: Enhanced organic mass fraction and decreased
 hygroscopicity of cloud condensation nuclei (CCN) during new particle formation events,
 Geophysical Research Letters, 37, n/a-n/a, 2010.
- Ervens, B., Cubison, M. J., Andrews, E., Feingold, G., Ogren, J. A., Jimenez, J. L., Quinn, P. K.,
 Bates, T. S., Wang, J., Zhang, Q., Coe, H., Flynn, M., and Allan, J. D.: CCN predictions using
 simplified assumptions of organic aerosol composition and mixing state: a synthesis from six
 different locations, Atmospheric Chemistry and Physics, 10, 4795-4807, 2010.
- Fisch, G., Tota, J., Machado, L. A. T., Silva Dias, M. A. F., da F. Lyra, R. F., Nobre, C. A.,
 Dolman, A. J., and Gash, J. H. C.: The convective boundary layer over pasture and forest in
 Amazonia, Theor Appl Climatol, 78, 47-59, 2004.
- Frank, G. P., Dusek, U., and Andreae, M. O.: Technical note: A method for measuring sizeresolved CCN in the atmosphere, Atmos. Chem. Phys. Discuss., 6, 4879-4895, 2006.
- Fry, J. L., Kiendler-Scharr, A., Rollins, A. W., Wooldridge, P. J., Brown, S. S., Fuchs, H., Dube,
 W., Mensah, A., dal Maso, M., Tillmann, R., Dorn, H. P., Brauers, T., and Cohen, R. C.: Organic
 nitrate and secondary organic aerosol yield from NO3 oxidation of beta-pinene evaluated using a
 gas-phase kinetics/aerosol partitioning model, Atmospheric Chemistry and Physics, 9, 14311449, 2009.

- 888 Ghan, S. J., Smith, S. J., Wang, M. H., Zhang, K., Pringle, K. J., Carslaw, K. S., Pierce, J. R.,
- 889 Bauer, S. E., and Adams, P. J.: A simple model of global aerosol indirect effects, Journal of 890
- Geophysical Research-Atmospheres, 118, 6688-6707, 2013.
- 891 Good, N., Topping, D. O., Allan, J. D., Flynn, M., Fuentes, E., Irwin, M., Williams, P. I., Coe,
- 892 H., and McFiggans, G.: Consistency between parameterisations of aerosol hygroscopicity and 893 CCN activity during the RHaMBLe discovery cruise, Atmospheric Chemistry and Physics, 10,
- 894 3189-3203, 2010.
- 895 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates of global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and Aerosols 896 897 from Nature), Atmospheric Chemistry and Physics, 6, 3181-3210, 2006.
- 898 Guenther, A. B., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L. K., and 899 Wang, X.: The Model of Emissions of Gases and Aerosols from Nature version 2.1 900 (MEGAN2.1): an extended and updated framework for modeling biogenic emissions, 901 Geoscientific Model Development, 5, 1471-1492, 2012.
- 902 Gunthe, S. S., King, S. M., Rose, D., Chen, Q., Roldin, P., Farmer, D. K., Jimenez, J. L., Artaxo,
- 903 P., Andreae, M. O., Martin, S. T., and Poschl, U.: Cloud condensation nuclei in pristine tropical 904 rainforest air of Amazonia: size-resolved measurements and modeling of atmospheric aerosol
- 905 composition and CCN activity, Atmos. Chem. Phys., 9, 7551-7575, 2009.
 - 906 Hamilton, D. S., Lee, L. A., Pringle, K. J., Reddington, C. L., Spracklen, D. V., and Carslaw, K. 907 S.: Occurrence of pristine aerosol environments on a polluted planet, Proceedings of the National 908 Academy of Sciences, 111, 18466-18471, 2014.
 - 909 Heald, C. L., Wilkinson, M. J., Monson, R. K., Alo, C. A., Wang, G., and Guenther, A.: 910 Response of isoprene emission to ambient CO2 changes and implications for global budgets,
 - 911 Global Change Biology, 15, 1127-1140, 2009.
 - 912 Hori, M., Ohta, S., Murao, N., and Yamagata, S.: Activation capability of water soluble organic 913 substances as CCN, J. Aerosol Sci., 34, 419-448, 2003.
 - 914 Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prevot, A. S. H., Zhang, Q., Kroll, J. H., 915 DeCarlo, P. F., Allan, J. D., Coe, H., Ng, N. L., Aiken, A. C., Docherty, K. S., Ulbrich, I. M., 916 Grieshop, A. P., Robinson, A. L., Duplissy, J., Smith, J. D., Wilson, K. R., Lanz, V. A., Hueglin, 917 C., Sun, Y. L., Tian, J., Laaksonen, A., Raatikainen, T., Rautiainen, J., Vaattovaara, P., Ehn, M., 918 Kulmala, M., Tomlinson, J. M., Collins, D. R., Cubison, M. J., Dunlea, E. J., Huffman, J. A., 919 Onasch, T. B., Alfarra, M. R., Williams, P. I., Bower, K., Kondo, Y., Schneider, J., Drewnick, F., 920 Borrmann, S., Weimer, S., Demerjian, K., Salcedo, D., Cottrell, L., Griffin, R., Takami, A., 921 Miyoshi, T., Hatakeyama, S., Shimono, A., Sun, J. Y., Zhang, Y. M., Dzepina, K., Kimmel, J. 922 R., Sueper, D., Jayne, J. T., Herndon, S. C., Trimborn, A. M., Williams, L. R., Wood, E. C., 923 Middlebrook, A. M., Kolb, C. E., Baltensperger, U., and Worsnop, D. R.: Evolution of Organic
 - 924 Aerosols in the Atmosphere, Science, 326, 1525-1529, 2009.
 - 925 Kammermann, L., Gysel, M., Weingartner, E., Herich, H., Cziczo, D. J., Holst, T., 926 Svenningsson, B., Arneth, A., and Baltensperger, U.: Subarctic atmospheric aerosol composition:

- 927 3. Measured and modeled properties of cloud condensation nuclei, Journal of Geophysical928 Research, 115, D04202 2010.
- Kesselmeier, J., Kuhn, U., Rottenberger, S., Biesenthal, T., Wolf, A., Schebeske, G., Andreae,
 M. O., Ciccioli, P., Brancaleoni, E., Frattoni, M., Oliva, S. T., Botelho, M. L., Silva, C. M. A.,
 and Tavares, T. M.: Concentrations and species composition of atmospheric volatile organic
 compounds (VOCs) as observed during the wet and dry season in Rondonia (Amazonia), Journal
 of Geophysical Research-Atmospheres, 107, 2002.
- King, S. M., Rosenoern, T., Shilling, J. E., Chen, Q., and Martin, S. T.: Increased cloud
 activation potential of secondary organic aerosol for atmospheric mass loadings, Atmos. Chem.
 Phys., 9, 2959-2971, 2009.
- Kuhn, U., Andreae, M. O., Ammann, C., Araújo, A. C., Brancaleoni, E., Ciccioli, P., Dindorf, T.,
 Frattoni, M., Gatti, L. V., Ganzeveld, L., Kruijt, B., Lelieveld, J., Lloyd, J., Meixner, F. X.,
 Nobre, A. D., Pöschl, U., Spirig, C., Stefani, P., Thielmann, A., Valentini, R., and Kesselmeier,
 J.: Isoprene and monoterpene fluxes from Central Amazonian rainforest inferred from towerbased and airborne measurements, and implications on the atmospheric chemistry and the local
 carbon budget, Atmos. Chem. Phys., 7, 2855-2879, 2007.
- Kuhn, U., Ganzeveld, L., Thielmann, A., Dindorf, T., Schebeske, G., Welling, M., Sciare, J.,
 Roberts, G., Meixner, F. X., Kesselmeier, J., Lelieveld, J., Kolle, O., Ciccioli, P., Lloyd, J.,
 Trentmann, J., Artaxo, P., and Andreae, M. O.: Impact of Manaus City on the Amazon Green
 Ocean atmosphere: ozone production, precursor sensitivity and aerosol load, Atmos. Chem.
 Phys., 10, 9251-9282, 2010.
- Kumar, P. P., Broekhuizen, K., and Abbatt, J. P. D.: Organic acids as cloud condensation nuclei:
 Laboratory studies of highly soluble and insoluble species, Atmos. Chem. Phys., 3, 509-520,
 2003.
- Kuwata, M., Zorn, S. R., and Martin, S. T.: Using Elemental Ratios to Predict the Density of
 Organic Material Composed of Carbon, Hydrogen, and Oxygen, Environmental Science &
 Technology, 46, 787-794, 2012.
- Lambe, A. T., Onasch, T. B., Massoli, P., Croasdale, D. R., Wright, J. P., Ahern, A. T., Williams, L. R., Worsnop, D. R., Brune, W. H., and Davidovits, P.: Laboratory studies of the chemical composition and cloud condensation nuclei (CCN) activity of secondary organic aerosol (SOA) and oxidized primary organic aerosol (OPOA), Atmospheric Chemistry and Physics, 11, 8913-8928, 2011.
- Lance, S., Raatikainen, T., Onasch, T., Worsnop, D., Yu, X. Y., Alexander, M. L., Stolzenburg,
 M. R., McMurry, P. H., Smith, J. N., and Nenes, A.: Aerosol mixing-state, hygroscopic growth
 and cloud activation efficiency during MIRAGE 2006, Atmospheric Chemistry and Physics, 13,
 5049-5062 2013.
- Lanz, V. A., Alfarra, M. R., Baltensperger, U., Buchmann, B., Hueglin, C., Szidat, S., Wehrli,
 M. N., Wacker, L., Weimer, S., Caseiro, A., Puxbaum, H., and Prevot, A. S. H.: Source

Attribution of Submicron Organic Aerosols during Wintertime Inversions by Advanced Factor
 Analysis of Aerosol Mass Spectra, Environmental Science & Technology, 42, 214-220, 2008.

Lathem, T. L., Beyersdorf, A. J., Thornhill, K. L., Winstead, E. L., Cubison, M. J., Hecobian, A.,
Jimenez, J. L., Weber, R. J., Anderson, B. E., and Nenes, A.: Analysis of CCN activity of Arctic
aerosol and Canadian biomass burning during summer 2008, Atmospheric Chemistry and
Physics, 13, 2735-2756, 2013a.

- Lathem, T. L., Beyersdorf, A. J., Thornhill, K. L., Winstead, E. L., Cubison, M. J., Hecobian, A.,
 Jimenez, J. L., Weber, R. J., Anderson, B. E., and Nenes, A.: Analysis of CCN activity of Arctic
 aerosol and Canadian biomass burning during summer 2008, Atmos. Chem. Phys., 13, 27352756, 2013b.
- Liu, X. and Wang, J.: How important is organic aerosol hygroscopicity to aerosol indirectforcing?, Environmental Research Letters, 5, 044010, 2010.
- 977 Liu, Y., Brito, J., Dorris, M. R., Rivera-Rios, J. C., Seco, R., Bates, K. H., Artaxo, P., Duvoisin,
- 978 S., Keutsch, F. N., Kim, S., Goldstein, A. H., Guenther, A. B., Manzi, A. O., Souza, R. A. F.,
- 979 Springston, S. R., Watson, T. B., McKinney, K. A., and Martin, S. T.: Isoprene photochemistry
- over the Amazon rainforest, Proceedings of the National Academy of Sciences, 113, 6125-6130,2016.
- Martin, S. T., Andreae, M. O., Althausen, D., Artaxo, P., Baars, H., Borrmann, S., Chen, Q.,
 Farmer, D. K., Guenther, A., Gunthe, S. S., Jimenez, J. L., Karl, T., Longo, K., Manzi, A.,
 Müller, T., Pauliquevis, T., Petters, M. D., Prenni, A. J., Pöschl, U., Rizzo, L. V., Schneider, J.,
 Smith, J. N., Swietlicki, E., Tota, J., Wang, J., Wiedensohler, A., and Zorn, S. R.: An overview
 of the Amazonian Aerosol Characterization Experiment 2008 (AMAZE-08), Atmos. Chem.
 Phys., 10, 11415-11438, 2010a.
- Martin, S. T., Andreae, M. O., Artaxo, P., Baumgardner, D., Chen, Q., Goldstein, A. H.,
 Guenther, A., Heald, C. L., Mayol-Bracero, O. L., McMurry, P. H., Pauliquevis, T., Pöschl, U.,
 Prather, K. A., Roberts, G. C., Saleska, S. R., Silva Dias, M. A., Spracklen, D. V., Swietlicki, E.,
 and Trebs, I.: Sources and properties of Amazonian aerosol particles, Reviews of Geophysics,
 48, n/a-n/a, 2010b.
- 993 Martin, S. T., Artaxo, P., Machado, L., Manzi, A. O., Souza, R. A. F., Schumacher, C., Wang, J., 994 Biscaro, T., Brito, J., Calheiros, A., Jardine, K., Medeiros, A., Portela, B., Sá, S. S. d., Adachi, 995 K., Aiken, A. C., Albrecht, R., Alexander, L., Andreae, M. O., Barbosa, H. M. J., Buseck, P., 996 Chand, D., Comstock, J. M., Day, D. A., Dubey, M., Fan, J., Fast, J., Fisch, G., Fortner, E., 997 Giangrande, S., Gilles, M., Goldstein, A. H., Guenther, A., Hubbe, J., Jensen, M., Jimenez, J. L., 998 Keutsch, F. N., Kim, S., Kuang, C., Laskin, A., McKinney, K., Mei, F., Miller, M., Nascimento, 999 R., Pauliquevis, T., Pekour, M., Peres, J., Petäjä, T., Pöhlker, C., Pöschl, U., Rizzo, L., Schmid, 1000 B., Shilling, J. E., Dias, M. A. S., Smith, J. N., Tomlinson, J. M., Tóta, J., and Wendisch, M.: 1001 The Green Ocean Amazon Experiment (GoAmazon2014/5) Observes Pollution Affecting Gases, 1002 Aerosols, Clouds, and Rainfall over the Rain Forest, Bulletin of the American Meteorological 1003 Society, 0, null, 2016a.

Martin, S. T., Artaxo, P., Machado, L. A. T., Manzi, A. O., Souza, R. A. F., Schumacher, C.,
Wang, J., Andreae, M. O., Barbosa, H. M. J., Fan, J., Fisch, G., Goldstein, A. H., Guenther, A.,
Jimenez, J. L., Pöschl, U., Silva Dias, M. A., Smith, J. N., and Wendisch, M.: Introduction:
Observations and Modeling of the Green Ocean Amazon (GoAmazon2014/5), Atmospheric
Chemistry and Physics, 16, 4785-4797, 2016b.

1009 Massoli, P., Lambe, A. T., Ahern, A. T., Williams, L. R., Ehn, M., Mikkila, J., Canagaratna, M.

1010 R., Brune, W. H., Onasch, T. B., Jayne, J. T., Petaja, T., Kulmala, M., Laaksonen, A., Kolb, C.

1011 E., Davidovits, P., and Worsnop, D. R.: Relationship between aerosol oxidation level and

1012 hygroscopic properties of laboratory generated secondary organic aerosol (SOA) particles,

1013 Geophysical Research Letters, 37, L24801, 2010.

1014 McFiggans, G., Artaxo, P., Baltensperger, U., Coe, H., Facchini, M. C., Feingold, G., Fuzzi, S.,

- 1015 Gysel, M., Laaksonen, A., Lohmann, U., Mentel, T. F., Murphy, D. M., O'Dowd, C. D., Snider,
- 1016 J. R., and Weingartner, E.: The effect of physical and chemical aerosol properties on warm cloud
- 1017 droplet activation, Atmos. Chem. Phys., 6, 2593-2649, 2006.
- 1018 Mei, F., Hayes, P. L., Ortega, A. M., Taylor, J. W., Allan, J. D., Gilman, J. B., Kuster, W. C., de
- Gouw, J. A., Jimenez, J. L., and Wang, J.: Droplet activation properties of organic aerosols
 observed at an urban site during CalNex-LA, Journal of Geophysical Research, 118, 2903-2917
 2013a.
- Mei, F., Setyan, A., Zhang, Q., and Wang, J.: CCN activity of organic aerosols observed
 downwind of urban emissions during CARES, Atmospheric Chemistry and Physics, 13, 121551024 12169, 2013b.

Mikhailov, E., Vlasenko, S., Rose, D., and Pöschl, U.: Mass-based hygroscopicity parameter
interaction model and measurement of atmospheric aerosol water uptake, Atmos. Chem. Phys.,
13, 717-740, 2013.

Moore, R. H., Bahreini, R., Brock, C. A., Froyd, K. D., Cozic, J., Holloway, J. S., Middlebrook,
A. M., Murphy, D. M., and Nenes, A.: Hygroscopicity and composition of Alaskan Arctic CCN
during April 2008, Atmospheric Chemistry and Physics, 11, 11807-11825, 2011.

Moore, R. H., Cerully, K., Bahreini, R., Brock, C. A., Middlebrook, A. M., and Nenes, A.:
Hygroscopicity and composition of California CCN during summer 2010, Journal of
Geophysical Research, 117, D00V12, 2012.

Moore, R. H., Nenes, A., and Medina, J.: Scanning Mobility CCN Analysis—A Method for Fast
 Measurements of Size-Resolved CCN Distributions and Activation Kinetics, Aerosol Science
 and Technology, 44, 861-871, 2010.

1037 Nenes, A., Pandis, S., and Pilinis, C.: ISORROPIA: A New Thermodynamic Equilibrium Model
1038 for Multiphase Multicomponent Inorganic Aerosols, Aquatic Geochemistry, 4, 123-152, 1998.

Ng, N. L., Canagaratna, M. R., Zhang, Q., Jimenez, J. L., Tian, J., Ulbrich, I. M., Kroll, J. H.,
Docherty, K. S., Chhabra, P. S., Bahreini, R., Murphy, S. M., Seinfeld, J. H., Hildebrandt, L.,
Donahue, N. M., DeCarlo, P. F., Lanz, V. A., Prevot, A. S. H., Dinar, E., Rudich, Y., and

Worsnop, D. R.: Organic aerosol components observed in Northern Hemispheric datasets from
 Aerosol Mass Spectrometry, Atmospheric Chemistry and Physics, 10, 4625-4641, 2010.

Ng, N. L., Herndon, S. C., Trimborn, A., Canagaratna, M. R., Croteau, P. L., Onasch, T. B.,
Sueper, D., Worsnop, D. R., Zhang, Q., Sun, Y. L., and Jayne, J. T.: An Aerosol Chemical
Speciation Monitor (ACSM) for Routine Monitoring of the Composition and Mass
Concentrations of Ambient Aerosol, Aerosol Science and Technology, 45, 780-794, 2011.

- 1048 Pajunoja, A., Lambe, A. T., Hakala, J., Rastak, N., Cummings, M. J., Brogan, J. F., Hao, L. Q.,
- 1049 Paramonov, M., Hong, J., Prisle, N. L., Malila, J., Romakkaniemi, S., Lehtinen, K. E. J.,
- 1050 Laaksonen, A., Kulmala, M., Massoli, P., Onasch, T. B., Donahue, N. M., Riipinen, I.,
- 1051 Davidovits, P., Worsnop, D. R., Petaja, T., and Virtanen, A.: Adsorptive uptake of water by
- semisolid secondary organic aerosols, Geophysical Research Letters, 42, 3063-3068, 2015.
- Park, K., Kittelson, D., Zachariah, M., and McMurry, P.: Measurement of Inherent MaterialDensity of Nanoparticle Agglomerates, J Nanopart Res, 6, 267-272, 2004.
- Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic growth and cloud condensation nucleus activity, Atmos. Chem. Phys., 7, 1961-1971, 2007.
- 1057 Petters, M. D., Prenni, A. J., Kreidenweis, S. M., and DeMott, P. J.: On Measuring the Critical
- 1058 Diameter of Cloud Condensation Nuclei Using Mobility Selected Aerosol, Aerosol Science and1059 Technology, 41, 907-913, 2007.
- Pöhlker, M. L., Pöhlker, C., Klimach, T., Hrabe de Angelis, I., Barbosa, H. M. J., Brito, J.,
 Carbone, S., Cheng, Y., Chi, X., Ditas, F., Ditz, R., Gunthe, S. S., Kesselmeier, J., Könemann,
 T., Lavrič, J. V., Martin, S. T., Moran-Zuloaga, D., Rose, D., Saturno, J., Su, H., Thalman, R.,
 Walter, D., Wang, J., Wolff, S., Artaxo, P., Andreae, M. O., and and Pöschl, U.: Long-term
 observations of cloud condensation nuclei in the Amazon rain forest Part 1: Aerosol size
 distribution, hygroscopicity, and new model parametrizations for CCN prediction Atmospheric
 Chemistry and Physics, 16, 15709-15740, 2016.
- Pöschl, U., Martin, S. T., Sinha, B., Chen, Q., Gunthe, S. S., Huffman, J. A., Borrmann, S.,
 Farmer, D. K., Garland, R. M., Helas, G., Jimenez, J. L., King, S. M., Manzi, A., Mikhailov, E.,
 Pauliquevis, T., Petters, M. D., Prenni, A. J., Roldin, P., Rose, D., Schneider, J., Su, H., Zorn, S.
 R., Artaxo, P., and Andreae, M. O.: Rainforest aerosols as biogenic nuclei of clouds and
 precipitation in the Amazon, Science, 329, 1513-1516, 2010.
- Prenni, A. J., Petters, M. D., Kreidenweis, S. M., DeMott, P. J., and Ziemann, P. J.: Cloud
 droplet activation of secondary organic aerosol, J. Geophys. Res., 112, D10223, 2007.
- 1074 Raymond, T. M. and Pandis, S. N.: Formation of cloud droplets by multicomponent organic1075 particles, Journal of Geophysical Research, 108, 4469, 2003.
- 1076 Reutter, P., Su, H., Trentmann, J., Simmel, M., Rose, D., Gunthe, S. S., Wernli, H., Andreae, M.
- 1077 O., and Poschl, U.: Aerosol- and updraft-limited regimes of cloud droplet formation: influence of
- 1078 particle number, size and hygroscopicity on the activation of cloud condensation nuclei (CCN),
- 1079 Atmospheric Chemistry And Physics, 9, 7067-7080, 2009.

- 1080 Rissler, J., Vestin, A., Swietlicki, E., Fisch, G., Zhou, J., Artaxo, P., and Andreae, M. O.: Size
- distribution and hygroscopic properties of aerosol particles from dry-season biomass burning in
 Amazonia, Atmos. Chem. Phys., 6, 471-491, 2006a.
- Rissler, J., Vestin, A., Swietlicki, E., Fisch, G., Zhou, J., Artaxo, P., and Andreae, M. O.: Size
 distribution and hygroscopic properties of aerosol particles from dry-season biomass burning in
 Amazonia, Atmos. Chem. Phys., 6, 471-491, 2006b.
- 1086 Rissman, T. A., Nenes, A., and Seinfeld, J. H.: Chemical Amplification (or Dampening) of the
 1087 Twomey Effect: Conditions Derived from Droplet Activation Theory, Journal of the
 1088 Atmospheric Sciences, 61, 919-930, 2004.
- Riva, M., Bell, D. M., Hansen, A. M. K., Drozd, G. T., Zhang, Z. F., Gold, A., Imre, D., Surratt,
 J. D., Glasius, M., and Zelenyuk, A.: Effect of Organic Coatings, Humidity and Aerosol Acidity
 on Multiphase Chemistry of Isoprene Epoxydiols, Environmental Science & Technology, 50,
 5580-5588, 2016.
- Roberts, G. C., Andreae, M. O., Zhou, J., and Artaxo, P.: Cloud condensation nuclei in the
 Amazon Basin: "marine" conditions over a continent?, Geophysical Research Letters, 28, 28072810, 2001.
- Roberts, G. C., Artaxo, P., Zhou, J., Swietlicki, E., and Andreae, M. O.: Sensitivity of CCN
 spectra on chemical and physical properties of aerosol: A case study from the Amazon Basin,
 Journal of Geophysical Research: Atmospheres, 107, LBA 37-31-LBA 37-18, 2002.
- Rose, D., Gunthe, S. S., Mikhailov, E., Frank, G. P., Dusek, U., Andreae, M. O., and Poschl, U.:
 Calibration and measurement uncertainties of a continuous-flow cloud condensation nuclei
 counter (DMT-CCNC): CCN activation of ammonium sulfate and sodium chloride aerosol
 particles in theory and experiment, Atmos. Chem. Phys., 8, 1153-1179, 2008a.
- Rose, D., Gunthe, S. S., Mikhailov, E., Frank, G. P., Dusek, U., Andreae, M. O., and Poschl, U.:
 Calibration and measurement uncertainties of a continuous-flow cloud condensation nuclei
 counter (DMT-CCNC): CCN activation of ammonium sulfate and sodium chloride aerosol
 particles in theory and experiment, Atmos. Chem. Phys., 8, 1153-1179, 2008b.
- Rose, D., Nowak, A., Achtert, P., Wiedensohler, A., Hu, M., Shao, M., Zhang, Y., Andreae, M.
 O., and Poschl, U.: Cloud condensation nuclei in polluted air and biomass burning smoke near
 the mega-city Guangzhou, China Part 1: Size-resolved measurements and implications for the
 modeling of aerosol particle hygroscopicity and CCN activity, Atmospheric Chemistry and
 Physics, 10, 3365-3383, 2010.
- Rosenfeld, D., Lohmann, U., Raga, G. B., Dowd, C. D., Kulmala, M., Fuzzi, S., Reissell, A., and
 Andreae, M. O.: Flood or Drought: How Do Aerosols Affect Precipitation?, Science, 321, 1309,
 2008.
- Schwantes, R. H., Teng, A. P., Nguyen, T. B., Coggon, M. M., Crounse, J. D., St Clair, J. M.,
 Zhang, X., Schilling, K. A., Seinfeld, J. H., and Wennberg, P. O.: Isoprene NO3 Oxidation

- Products from the RO2 + HO2 Pathway, Journal of Physical Chemistry A, 119, 10158-10171,2015.
- Shantz, N. C., Leaitch, W. R., Phinney, L., Mozurkewich, M., and Toom-Sauntry, D.: The effect
 of organic compounds on the growth rate of cloud droplets in marine and forest settings, Atmos.
 Chem. Phys., 8, 5869-5887, 2008.
- 1122 Sullivan, R. C., Moore, M. J. K., Petters, M. D., Kreidenweis, S. M., Roberts, G. C., and Prather,
- 1123 K. A.: Effect of chemical mixing state on the hygroscopicity and cloud nucleation properties of 1124 calcium mineral dust particles, Atmospheric Chemistry And Physics, 9, 3303-3316, 2009.
- 1125 Twomey, S.: The Influence of Pollution on the Shortwave Albedo of Clouds, Journal of the 1126 Atmospheric Sciences, 34, 1149-1152, 1977.
- 1127 Ulbrich, I. M., Canagaratna, M. R., Zhang, Q., Worsnop, D. R., and Jimenez, J. L.: Interpretation
- 1128 of organic components from Positive Matrix Factorization of aerosol mass spectrometric data,
- 1129 Atmos. Chem. Phys., 9, 2891-2918, 2009.
- 1130 Vestin, A., Rissler, J., Swietlicki, E., Frank, G. P., and Andreae, M. O.: Cloud-nucleating
- 1131 properties of the Amazonian biomass burning aerosol: Cloud condensation nuclei measurements
- and modeling, Journal of Geophysical Research: Atmospheres, 112, 2007.
- Wang, J.: Effects of spatial and temporal variations in aerosol properties on mean cloud albedo,J. Geophys. Res., 112, D16201, 2007.
- 1135 Wang, J., Krejci, R., Giangrandel, S., Kuang, C., Barbosa, H. M. J., Brito, J., Carbone, S., Chi,
- 1136 X. G., Comstock, J., Ditas, F., Lavric, J., Manninen, H. E., Mei, F., Moran-Zuloaga, D., Pohlker,
- 1137 C., Pohlker, M. L., Saturno, J., Schmid, B., Souza, R. A. F., Springston, S. R., Tomlinson, J. M.,
- 1138 Toto, T., Walter, D., Wimmer, D., Smith, J. N., Kulmala, M., Machado, L. A. T., Artaxo, P.,
- 1139 Andreae, M. O., Petaja, T., and Martin, S. T.: Amazon boundary layer aerosol concentration
- sustained by vertical transport during rainfall, Nature, 539, 416-419, 2016a.
- Wang, J., Lee, Y. N., Daum, P. H., Jayne, J., and Alexander, M. L.: Effects of aerosol organics
 on cloud condensation nucleus (CCN) concentration and first indirect aerosol effect, Atmos.
 Chem. Phys., 8, 6325-6339, 2008.
- Wang, Q. Q., Saturno, J., Chi, X., Walter, D., Lavric, J. V., Moran-Zuloaga, D., Ditas, F.,
 Pohlker, C., Brito, J., Carbone, S., Artaxo, P., and Andreae, M. O.: Modeling investigation of
 light-absorbing aerosols in the Amazon Basin during the wet season, Atmospheric Chemistry
 and Physics, 16, 14775-14794, 2016b.
- Wex, H., Petters, M. D., Carrico, C. M., Hallbauer, E., Massling, A., McMeeking, G. R.,
 Poulain, L., Wu, Z., Kreidenweis, S. M., and Stratmann, F.: Towards closing the gap between
 hygroscopic growth and activation for secondary organic aerosol: Part 1-Evidence from
- 1151 measurements, Atmos. Chem. Phys., 9, 3987-3997, 2009.
- Whitehead, J. D., Darbyshire, E., Brito, J., Barbosa, H. M. J., Crawford, I., Stern, R., Gallagher,
 M. W., Kaye, P. H., Allan, J. D., Coe, H., Artaxo, P., and McFiggans, G.: Biogenic cloud nuclei

in the central Amazon during the transition from wet to dry season, Atmospheric Chemistry andPhysics, 16, 9727-9743, 2016.

Whitehead, J. D., Gallagher, M. W., Dorsey, J. R., Robinson, N., Gabey, A. M., Coe, H.,
McFiggans, G., Flynn, M. J., Ryder, J., Nemitz, E., and Davies, F.: Aerosol fluxes and dynamics
within and above a tropical rainforest in South-East Asia, Atmos. Chem. Phys., 10, 9369-9382,
2010.

Williams, E., Rosenfeld, D., Madden, N., Gerlach, J., Gears, N., Atkinson, L., Dunnemann, N.,
Frostrom, G., Antonio, M., Biazon, B., Camargo, R., Franca, H., Gomes, A., Lima, M.,
Machado, R., Manhaes, S., Nachtigall, L., Piva, H., Quintiliano, W., Machado, L., Artaxo, P.,
Roberts, G., Renno, N., Blakeslee, R., Bailey, J., Boccippio, D., Betts, A., Wolff, D., Roy, B.,
Halverson, J., Rickenbach, T., Fuentes, J., and Avelino, E.: Contrasting convective regimes over
the Amazon: Implications for cloud electrification, Journal of Geophysical Research:
Atmospheres, 107, 8082, 2002.

- Xu, L., Guo, H. Y., Boyd, C. M., Klein, M., Bougiatioti, A., Cerully, K. M., Hite, J. R.,
 Isaacman-VanWertz, G., Kreisberg, N. M., Knote, C., Olson, K., Koss, A., Goldstein, A. H.,
 Hering, S. V., de Gouw, J., Baumann, K., Lee, S. H., Nenes, A., Weber, R. J., and Ng, N. L.:
 Effects of anthropogenic emissions on aerosol formation from isoprene and monoterpenes in the
 southeastern United States, Proceedings of the National Academy of Sciences of the United
 States of America, 112, 37-42, 2015a.
- 1173 Xu, L., Suresh, S., Guo, H., Weber, R. J., and Ng, N. L.: Aerosol characterization over the 1174 southeastern United States using high-resolution aerosol mass spectrometry: spatial and seasonal 1175 variation of aerosol composition and sources with a focus on organic nitrates, Atmospheric 1176 Chemistry and Physics, 15, 7307-7336, 2015b.
- Zhang, Q., Canagaratna, M. R., Jayne, J. T., Worsnop, D. R., and Jimenez, J.-L.: Time- and sizeresolved chemical composition of submicron particles in Pittsburgh: Implications for aerosol
 sources and processes, Journal of Geophysical Research: Atmospheres, 110, 2005.
- 1180 Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Allan, J. D., Coe, H., Ulbrich, I., Alfarra, M. R., 1181 Takami, A., Middlebrook, A. M., Sun, Y. L., Dzepina, K., Dunlea, E., Docherty, K., DeCarlo, P. F., Salcedo, D., Onasch, T., Jayne, J. T., Miyoshi, T., Shimono, A., Hatakeyama, S., Takegawa, 1182 N., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K., Williams, 1183 1184 P., Bower, K., Bahreini, R., Cottrell, L., Griffin, R. J., Rautiainen, J., Sun, J. Y., Zhang, Y. M., 1185 and Worsnop, D. R.: Ubiquity and dominance of oxygenated species in organic aerosols in 1186 anthropogenically-influenced Northern Hemisphere midlatitudes, Geophys. Res. Lett., 34, 1187 L13801, 2007.
- 1188 Zhou, J., Swietlicki, E., Hansson, H. C., and Artaxo, P.: Submicrometer aerosol particle size 1189 distribution and hygroscopic growth measured in the Amazon rain forest during the wet season,
- 1190 Journal of Geophysical Research: Atmospheres, 107, LBA 22-21-LBA 22-10, 2002.
- 1191
- 1192

1194 **Tables**

Table 1: Density, O:C ratio and hygroscopicity associated with organic factors derived from Positive Matrix Factorization (PMF) analysis. For IOP1 and IOP2, hygroscopicities of PMF organic
factors were derived from time series of particle hygroscopicity under all conditions and background conditions, respectively.

IOP1 (wet season)				IOP2 (dry season)			
PMF Factor	ρ (g cm ⁻³)	O:C	К	PMF Factor	ρ (g cm ⁻³)	O:C	κ (Bkgd) <u>*</u>
IEPOX-SOA	1.47	0.798	0.18±0.02	IEPOX-SOA	1.42	0.711	0.08±0.03
MO-OOA	1.80	1.19	0.20±0.02	MO-OOA	1.81	1.24	0.21±0.03
LO-OOA	1.48	0.786	0.12±0.02	LO-OOA	1.52	0.883	0.20±0.03
BBOA	1.42	0.712	0.04±0.03	Aged BBOA	1.37	0.666	0.08±0.03
Fac91	1.14	0.328	0.10±0.03	Fresh BBOA	1.23	0.536	0.00 ± 0.07
НОА	0.95	0.163	0	НОА	1.02	0.223	0

 $\frac{1199}{1200} \frac{\kappa (Bkgd) \text{ refers to } \kappa \text{ values of PMF factors derived from the time series of particle hygroscopicity under background conditions}{(see Sections 4.2.2 and 4.3).}$

1201 Figures

1202



Figure 1: Seasonal variations of aerosol properties observed at the T3 site from March 2014 to March 2015, including (a) κ_{CCN} , (b) κ_{org} , and size resolved volume fraction of (c) organics, (d) sulfate, including (NH₄)₂SO₄ and NH₄HSO₄, (e) nitrate, and (f) refractory Black Carbon. Data points are the monthly mean; error bars represent the 25th and 75th percentiles.

1208

1203



Figure 2: The variation of κ_{CCN} and κ_{org} with particle diameter during different seasons for each of three air mass types. Data points are the mean values; error bars are the 25 and 75 percentiles. The top left panel also includes κ_{CCN} observed under near natural conditions during the AMAZE-

1	
1215	08 campaign at T0t in the wet season (Gunthe et al., 2009) and during the one-year period from
1415	to eampaigh at 10t in the wet season (Gunthe et al., 2007) and during the one year period from

- 1216 March 2014 to February 2015 at the background T0a site (Pöhlker et al., 2016). The top right
- 1217 panel includes κ derived from particle growth factor measurements in July 2001, during a "recent
- 1218 biomass burning period" of the CLAIRE-2001 study (Rissler et al., 2004) and from 11 Septem-
- 1219 ber to 8 October 2002, during the dry period of the LBA-SMOCC (Rissler et al., 2006).

1221



1223 1224 Figure 3: Diel variations of aerosol properties and meteorological parameter under background conditions during the wet season, including (a) $\kappa_{\rm CCN}$, (b) $\sigma_{\kappa_{\rm CCN}}$, (c) fraction of the organic 1225 mass at m/z = 44 (f_{44}) and the elemental ratio O:C, (d) κ_{org} derived using size resolved particle 1226 1227 composition, (e) the total number of condensation nuclei $(N_{\rm CN})$, (f) the total aerosol volume de-1228 rived from size distribution measured by the SMPS in MAOS, (g) planetary boundary layer 1229 height as estimated using the ceilometer data, (h) the volume fractions of aerosol species, and (i) 1230 the number of samples in each hour bin corresponding to the data by the same colors and symbols in their respective panel. The values of κ_{CCN} , $\sigma_{\kappa_{\text{CCN}}}$, κ_{org} , and volume fraction of aero-1231 sol species were averaged over three particle diameters of 112, 142 and 171 nm. The values of f_{44} 1232 1233 and O:C were derived from the AMS bulk measurements. Data include the last two weeks of March 2014 when valid data from both size-resolved CCN system and AMS were available. Da-1234 1235 ta points are hourly averaged mean values; error bars represent the 25 and 75 percentiles of the 1236 data. Yellow shading represents the local daytime (10:00 - 22:00 UTC).



1237Wet Season - Polluted Conditions1238Figure 4: Diel variations of aerosol properties and meteorological parameters for urban pollution

1239 air masses during the wet season (analogous to Fig. 3).



Figure 5: Diel variations of aerosol properties and meteorological parameters under backgroundconditions during the dry season (analogous to Fig. 3).





1248 Figure 6: Diel variations of aerosol properties and meteorological parameters for urban pollution

1249 air masses during the dry season (analogous to Fig. 3).



Figure 7: Diel variations of aerosol properties and meteorological parameters for local biomassburning air masses during the dry season (analogous to Fig. 3).



Figure 8: Comparison κ_{CCN} values derived from measurements at the T0a (ATTO) site (Pöhlker et al., 2016) and at the T3 site under background conditions (this study) during the (a) wet season (April and May 2014) and (b) dry season. The size resolved CCN data at T0a was collected by stepping the particle size at given CCNC supersaturations (Rose et al., 2008b). Data displayed for T0a is averaged over critical particle diameters ranging from 44 to 175 nm, while the T3 data is averaged from measurements at 112, 142, and 171 nm.

1264



Figure 9: Hygroscopicity of AMS PMF factors for IOP1 (i.e., wet season) retrieved by multilinear regressions using all data (red square), data from UTC 12:00-24:00 (cyan circle), data
from UTC 0:00-12:00 (blue triangle), data under background conditions (red triangle), data un-

- 1271 der polluted conditions (red diamond), and data with hygroscopicity dispersion $\sigma_{\kappa_{CCN}} / \bar{\kappa}_{CCN}$ less
- 1272 than 0.6 (black diamond).



Figure 10: Hygroscopicity of PMF factors for IOP2 (i.e., dry season) retrieved by multi-linear regressions using all data (red square), data from UTC 12:00-24:00 (cyan circle), data from UTC 0:00-12:00 (blue triangle), data with strong influence from local biomass burning (red right triangle), data under background conditions (red upside-down triangle), data from Aug 21 to Sep

- 1280 14, 2014 only (red left pointing triangle), data from Sep 15 to Oct 15 only (brown diamond), and
- 1281 data with a dispersion ($\sigma_{\kappa_{CCN}} / \overline{\kappa}_{CCN}$) < 0.4 (black diamond).



Figure 11: The variation of PMF factor hygroscopicity, 1-hour diel average of organic hygrosco-picity and O:C ratio at 142 and 171 nm for urban pollution air masses. Also shown are the rela-

- 1285 tionships between κ_{org} and O:C reported by earlier field and laboratory studies.


Figure 12: The variation of PMF factor hygroscopicity, 1-hour diel average of organic hygroscopicity, and O:C ratio at 142 and 171 nm for local biomass burning air masses. Also shown are the relationships between κ_{org} and O:C reported by earlier field and laboratory studies.

Supplement to:

CCN activity and organic hygroscopicity of aerosols downwind of an urban region in
 central Amazonia: Seasonal and diel variations and impact of anthropogenic emissions

4

5 Ryan Thalman^{1,#}, Suzane S. de Sá², Brett B. Palm³, Henrique M. J. Barbosa⁴, Mira L. Pöhlker⁵,

- 6 M. Lizabeth Alexander⁷, Joel Brito^{4*}, Samara Carbone⁴, Paulo Castillo¹, Douglas A. Day³,
- 7 Chongai Kuang¹, Antonio Manzi⁸, Nga Lee Ng^{9,10}, Arthur J. Sedlacek III¹, Rodrigo Souza¹¹,
- 8 Stephen Springston¹, Thomas Watson¹, Christopher Pöhlker⁵, Ulrich Pöschl^{5,6}, Meinrat O.

9 Andreae^{5,12}, Paulo Artaxo⁴ Jose L. Jimenez³, Scot T. Martin^{2,13}, Jian Wang¹

- 10
- ¹Environmental and Climate Sciences Department, Brookhaven National Laboratory, Upton,
 NY, USA
- ¹³ ²School of Engineering and Applied Sciences, Harvard University, Cambridge, MA, USA
- 14 ³Department of Chemistry and Biochemistry and Cooperative Institute for Research in
- 15 Environmental Sciences (CIRES), University of Colorado Boulder, Boulder, CO, USA
- ⁴Physics Institute, University of São Paulo, São Paulo, Brazil
- ⁵Biogeochemistry and Multiphase Chemistry Departments, Max Planck Institute for Chemistry,
 Mainz, Germany
- ⁶Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA, USA
- 20 ⁷Pacific Northwest National Laboratory, Richland, WA, USA
- 21 ⁸National Institute of Amazonian Research, Manaus, Amazonas, Brazil
- ⁹School of Chemical and Biomolecular Engineering, Georgia Institute of Technology, Atlanta,
 GA, USA
- ¹⁰School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA,
 USA
- 26 ¹¹ Amazonas State University, Manaus, Amazonas, Brazil
- ¹²Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA, USA
- ¹³Department of Earth and Planetary Sciences, Harvard University, Cambridge, MA, USA
- 29
- 30 [#]Now at Department of Chemistry, Snow College, Richfield, UT, USA
- 31 *Now at Laboratory for Meteorological Physics, University Clermont Auvergne, Clermont-
- 32 Ferrand, France
- 33 Correspondence to: J. Wang (jian@bnl.gov)

34

35

37 1 Calibration and Operation of the CCN counter

38 The system used to measure size-resolved CCN spectra during GoAmazon 2014/5 is shown in Fig. S1, and the operation sequences of DMA classified particle diameter, $Q_{\rm CCN}$ and ΔT are 39 40 present in Fig. S2. The CCN counter was calibrated at each flow and temperature setting using 41 ammonium sulfate as an inorganic aerosol standard (Lance et al., 2006; Mei et al., 2013; Roberts 42 and Nenes, 2005; Shilling et al., 2007). As the temperatures at the top of the CCN counter 43 column (T_1) varies with room temperature, the calibration was repeated for each pair of Q_{CCN} and 44 ΔT over a range of T₁ typically encountered during GoAmazon 2014/5. The calibration results 45 are shown as a function of Q_{CCN} and ΔT conditions at $T_1 = 25^{\circ}\text{C}$ in Fig. S3, and the variation of 46 instrument supersaturation (SS) with respect to T_1 is present in Fig. S4. Instrument 47 supersaturation during the measurements was calculated using the Q_{CCN} , ΔT , and T_1 based on the 48 calibrations.



Figure S1: Schematic of the system for measuring size-resolved CCN spectra during GoAmazon2014/5.



Figure S2: Instrument operation sequence showing set points of DMA classified particle diameter, Q_{CCN} and ΔT for one full measurement cycle. Red vertical lines mark the start and finish of one cycle.



52

57 Figure S3: Calibrated CCN counter supersaturation as a function of ΔT at T_1 of 25°C for Q_{CCN}

58 ranging from 0.2 to 1.0 LPM.



59

Figure S4: Calibrated instrument supersaturation as a function of T_1 for different ΔT and Q_{CCN} values. The symbol colors of purple, dark blue, light blue, green and red correspond to ΔT values of 4.5, 5.5, 6.5, 8.0, and 10.0°C, respectively. Calibrations were carried out throughout the entire campaign (February/March 2014 - •; October 2014 - •; March 2015 - •). The

64 lines represent least square fits of the calibration results.

65 **2 ARM Facility Instruments**

66 2.1 Measurements of Oxides of Nitrogen

67 Oxides of nitrogen were measured by catalytic conversion to NO and chemiluminescence at 68 one minute time resolution. The NO/NO₂/NO_y system is based on two Thermo Scientific 43i instruments and customized by Air Quality Design, Inc. (Wheatridge, CO) for detection of NO. 69 70 The inlet/converter box is external and located at 10 meters above ground level and is 71 environmentally controlled. A dedicated channel converts total NOy (NO + NO₂ + HNO₃ + 72 peroxy acyl nitrates + aerosol nitrates, but not NH_4) to NO via a Mo catalyst heated to 325°C. A 73 second channel converts NO₂ to NO by a light emitting diode (LED) photolysis cell with 50% 74 conversion efficiency (C.E.). The LED is toggled on and off on a two minute cycle to measure 75 ambient NO and NO + C.E.× NO₂, and the NO₂ mixing ratio is derived from the difference of 76 the measurements. The inlet box contains mass flow controllers (MFCs) so that the residence 77 time to the chemiluminescence cells is minimal and occurs at reduced pressure. The unit is 78 calibrated daily with standard additions of NO and NO₂ to measure the conversion efficiency of 79 both the Mo converter (1.00 ± 0.02) and the photolysis cell.

80 2.2 Refractory Black Carbon Measurements

The incandescence signal measured by SP2 is proportional to the mass of rBC in the particles irrespective of whether the rBC is coated or not (Moteki et al., 2007; Schwarz et al., 2010; Stephens et al., 2003). The SP2 was calibrated using size-selected fullerene soot with no corrections for non-rBC content. The detectable size range of the SP2 is 75 - 600 nm (volume equivalent diameter), which typically provides a direct measurement of > 90% of rBC mass concentrations. If necessary, the complete mass size distribution is extrapolated to include rBC outside of this size range assuming a log-normal distribution (Mei et al., 2013). During

88 GoAmazon 2014/5, rBC mass was mostly within the detectable range. In contrast to the SP2, the 89 seven-wavelength Aethalometer (Arnott et al., 2005; Hansen et al., 1984) reports an equivalent 90 black carbon mass loading by converting a light absorption measurement from aerosols 91 accumulated on a filter relative to a manufacturer-supplied conversion. However, there is 92 substantial and growing evidence relating to measurement artifacts attributed to aerosol loading 93 effects and enhanced light absorption through multiple-scattering of photons (Collaud Coen et 94 al., 2010; Rizzo et al., 2011; Schmid et al., 2006; Virkkula et al., 2015; Virkkula et al., 2007; 95 Weingartner et al., 2003). Given the more complete dataset of the Aethalometer, our analysis 96 utilized the Aethalometer data only when SP2 data was unavailable. The available concurrent 97 SP2 and Aethalometer datasets were used to derive a linear relationship to correct for 98 Aethalometer measurement artifacts cited above. The mass concentrations of refractory black 99 carbon (rBC) derived from SP2 and Aethalometer data were averaged to a common time base 100 and the least square linear fit between both rBC mass concentrations is shown in Fig. S5 for both 101 the wet and dry seasons. The linear relationship exhibited a seasonal dependence, which might 102 be driven by variations in RH, the ratio of mass concentrations of organics to BC, or a 103 combination of both. The rBC mass concentration derived from the Aethalometer measurement 104 was used to fill the gaps in the SP2 data after scaled using the linear relationship corresponding 105 to the respective season.

106



Figure S5: Correlation between rBC mass concentrations derived from the Aethalometer and SP2 measurements. Data points are colored by the time of the measurements. The red and green lines represent least square fit of data collected during the dry and wet seasons, respectively.

108

113 **3 Reduction of the size resolved CCN activation spectrum**

114 **3.1 Correction for multiple charging**

115 Size-resolved CCN activation fraction was characterized at particle diameters of 51, 75, 94, 116 112, 142, 171 and 222 nm. These sampling sizes were selected to form a progression of particle 117 diameters corresponding to doubly and triply charged particles (e.g. 112 nm corresponds to the 118 diameter of doubly charged particles, when the DMA is set to select particles with a diameter of 119 75 nm. See Table S1 for more details). The contribution of doubly and triply charged particles to 120 the activation fraction was corrected using the size distribution measured by the SMPS (Model 121 3080 TSI, Inc) in MAOS and particle charging probabilities (Wiedensohler, 1988). In cases 122 when the size distribution data were not available, we used the concentrations of classified

123 particles measured by the CPC in the size-resolved CCN system when sampling at corresponding 124 multiple-charged sizes. The impact of multiple-charged particles on activation fraction was then 125 accounted for using the concentration of multiple-charged particles and the measured activation 126 fractions at the corresponding sizes. This is similar to the methods established in the literature 127 (Rose et al., 2008). When direct measurements of the activation fraction of multiple charged 128 particles were not available, the multiple charged particles were assumed to have the same 129 distribution of κ as the singly charge particles, and the activation fraction was corrected using the 130 following iterative approach. First, the activation fractions of multiple charged particles were 131 derived from the measured activation factions, taken into consideration the same κ distribution 132 but larger particle diameters. These derived activation fractions at multiple charged sizes were 133 then used to correct measured activation fraction using the approach described above. This 134 process was continued until the corrected activation fraction of singly charged particles 135 converged.

136

137 **3.2 Data quality control criteria**

138 Several quality-control criteria were applied to the data and results from the fitting of 139 activation spectrum. The measurement of the entire activation spectrum at a single particle 140 diameter took about 5-20 minutes. Individual points (i.e., activation fraction at a single S) of an 141 activation spectrum were excluded from the fitting if the standard deviation of 1-sec 142 measurements of the classified particle concentration is greater than 30% of the mean value for 143 the current size, or single data points are more than 1.5 times the mean. This is to remove large 144 fluctuation or spikes in CPC concentration due to contamination by emission of site generator or 145 traffic. The removal of measurement at single supersaturation typically does not prevent fitting

146	of the activation spectrum using data at remaining supersaturation points. Parameters derived
147	from the fitting of activation spectrum were flagged and excluded from further analysis in this
148	study for the following scenarios:
149	1. The fitted S^* is outside the supersaturation range of the measured activation spectrum.
150	2. The fitted maximum activated fraction (E) is not within 15% of any activated fraction
151	measured at the three highest supersaturations. This indicates that the activation fraction
152	did not reach a plateau even at the highest supersaturations sampled, and therefore E is
153	not well constrained by the activation spectrum.
154	3. The residue of the fit is greater than 0.1. This indicates relatively poor fit, which
155	represented less 1% of the activation spectra.
156	4. The fitted E is less than 0.5, indicating external mixtures with substantial fraction of fresh
157	black carbon and primary organic aerosols emitted by the generator, vehicles, and grass
158	cutting activities at or near the site.
159	For the one-year measurement period, 6.5 % of the total activation curves were removed by
160	the above data quality filters.
161	
162 163	3.4 Deriving average particle hygroscopicity and hygroscopicity dispersion from activation spectrum
164	For particles with the same size and composition (i.e., hygroscopicity), a step increase of R_a
165	from 0% to 100% with increasing supersaturation is expected because all particles would have
166	the identical critical supersaturation (S_c) . The measurements of ambient aerosols show more
167	gradual increase in R_a (i.e., instead of a step change), suggesting heterogeneity in particle S_c . The
168	heterogeneity in particle S_c , as described by σ_s is due to a combination of width of DMA transfer

169 function (particles classified by a DMA do not have exactly the same size), instrument non-170 idealities, and the heterogeneity in particle hygroscopicity. The value of σ_s due to the 171 heterogeneity in particle hygroscopicity alone can be estimated by:

172
$$\sigma_s = \sqrt{\sigma_{s,m}^2 - \sigma_{s,AS}^2}$$
(S1)

173 Where $\sigma_{s,m}$ and $\sigma_{s,AS}$ are the values corresponding to the ambient measurement and the calibration 174 using ammonium sulfate, respectively (Supplementary Information, Mei et al., 2013). Particle 175 critical supersaturation S_c is related to hygroscopicity κ by (Petters and Kreidenweis, 2007):

176

$$S_{c}(\kappa) = \left(\frac{4A^{3}}{27D_{p}^{3}\kappa}\right)^{1/2},$$

$$A = \frac{4\sigma_{w}M_{w}}{RT\rho_{w}},$$
(S2)

177 Where D_p is the particle diameter, M_w , ρ_w and σ_w are the molecular weight, density and surface 178 tension of water, respectively. The probability distribution function of particle hygroscopicity 179 $p(\kappa)$ can be derived as the following:

180
$$p(\kappa) = -\frac{dR_a(S_c(\kappa))}{d\kappa}$$
 (S3)

181 Where R_a is the fitted active spectrum with σ_s adjusted using Eq. (S1). We note that Eq. (S2) 182 represents an approximation (Petters and Kreidenweis, 2007). For this study, the uncertainty of 183 derived κ values due to this approximation is negligible. The dispersion of κ , which describes the 184 heterogeneity of hygroscopicity for activated particles, is given by (Mei et al., 2013):

185
$$\sigma(\kappa)/\overline{\kappa_{\rm CCN}} = \left[\exp(4\sigma_s^2) - 1\right]^{1/2}$$
 (S4)

186 where $\sigma(\kappa)$ and $\overline{\kappa_{CCN}}$ are the standard deviation and average value of the hygroscopicity for 187 activated particles.

It is worth noting that because of the skewness of the lognormal distribution, the average particle hygroscopicity $\overline{\kappa_{\text{CCN}}}$ is higher than κ^* , which corresponds to the fitted S^* , and represents the median hygroscopicity of activated particles (Fig. S6). The difference between $\overline{\kappa_{\text{CCN}}}$ and κ^* becomes significant at high σ_s values (Fig. S7). In this study, $\overline{\kappa_{\text{CCN}}}$ is derived from the probability density function using the following equation:

193
$$\overline{\kappa_{\rm CCN}} = \frac{\int_0^{0.65} \kappa \cdot p(\kappa) d\kappa}{\int_0^{0.65} p(\kappa) d\kappa}$$
(S5)

194 The upper limit of the integration is limited to 0.65, which reflects the maximum particle 195 hygroscopicity expected at the T3 site. The organic hygroscopicity is derived from the average 196 particle hygroscopicity $\overline{\kappa_{CCN}}$ and average chemical composition.



197

Figure S6: Sample activation spectra for particles with diameter of 142 nm. The black dots and green triangles represent spectra with a low dispersion ($\sigma_s = 0.2$; $\sigma(\kappa)/\overline{\kappa_{CCN}} = 0.4$) and a much higher dispersion ($\sigma_s = 0.4$; $\sigma(\kappa)/\overline{\kappa_{CCN}} = 0.9$), respectively.



Figure S7: Probability density distributions of particle hygroscopicity derived from the fitted activation spectra shown in Figure S6. As dispersion increases, the distribution becomes increasingly asymmetric, and the difference between $\overline{\kappa_{CCN}}$ and κ^* increases.

205 4 Aerosol composition derived from AMS and rBC measurements

206 4.1 Bulk and size-resolved mass concentration of aerosol species

Given the very low aerosol mass concentration in the Amazon basin, the signal to noise ratio 207 208 of particle time-of-flight (P-ToF) mode measurement was not sufficient to directly provide size 209 resolved species mass concentrations for individual CCN activation spectrum. Instead, the size 210 resolved species mass concentrations were constructed by combing the bulk mass concentrations 211 measured in MS mode, which have higher signal to noise ratio, and P-ToF mode size resolved 212 species mass distributions averaged over extended time periods. For both IOPs, the bulk organics 213 mass fraction was derived from MS mode portion of individual AMS measurement. Based on the 214 bulk organic mass fraction, measurements during IOP1 were then classified into three groups 215 with equal number of measurements, and the characteristic mass size distribution of each species 216 (i.e., organics, SO₄, NO₃, and NH₄) was averaged from P-ToF measurements within each group. 217 For IOP2, the measurements were classified into three groups each for day and night periods 218 based on the bulk aerosol organic mass fraction, and the mass size distribution of each species 219 was averaged from P-ToF measurements in each of the six groups. The average mass size 220 distributions of the four species for the three groups during the daytime of IOP2 are shown in 221 Fig. S8 as examples. The underlying assumption is that observed aerosols exhibiting similar 222 bulk composition (i.e., organic mass fraction) during the IOPs also had species mass size 223 distributions with the same shapes. The vacuum aerodynamic diameter (D_{va}) measured by the 224 AMS was converted to particle mobility diameter (D_m) and volume equivalent diameter (D_v) 225 using the particle density with the assumption of spherical particles, which is reasonable given 226 most of the aerosol observed at T3 site were quite aged. The average densities were calculated using the approach described in Kuwata et al. (2012), and were 1.45 μ g m³ and 1.47 during IOP 227 228 1 and 2, respectively. In this study, unless otherwise indicated, particle diameter D_p represents

229 the mobility diameter (D_m) , which is equivalent to the volume average diameter given the 230 assumption of spherical particles.



231

Figure S8: Size resolved mass concentrations of organics, SO₄, NO₃, and NH₄ averaged from P-ToF measurements for each of three groups during the day time of IOP2 (dry season). The three groups were classified based on bulk organics mass fraction ($f_{M,org}$).

235

For individual AMS measurements, the size resolve chemical composition was then derived as follows: the organic mass concentration at the particle diameter of CCN measurements, $m_{org} (D_p)$, was calculated as:

239
$$m_{org} \left(D_{p} \right) = M_{org,b} \times \frac{\overline{m}_{org} \left(D_{p} \right)}{\int_{D_{p,\min}}^{D_{p,\max}} \overline{m}_{org} \left(D_{p}^{'} \right) d \log D_{p}^{'}}$$
(S6)

where $M_{org,b}$ is the bulk organic mass concentration from MS mode measurement, $\overline{m}_{org}(D_p)$ is the average organic mass size distribution with respect to $\log D_p$, $D_{p,max}$ and $D_{p,min}$ are the maximum and minimum diameters of the average mass size distribution. Using the same approach, the mass concentration of SO_4^{2-} , NO_3^- , and NH_4^+ at specific D_p are calculated using the corresponding average mass size distributions. Mass concentration for black carbon $m_{rBC}(D_p)$ is derived with the assumption that the mass size distribution of rBC has the same shape of the total mass size distribution:

247
$$f_{M,rBC} = \frac{M_{rBC,b}}{M_{total,b}} = \frac{m_{rBC}(D_p)}{m_{SO_4^{2-}}(D_p) + m_{NO_3^{-}}(D_p) + m_{NH_4^{+}}(D_p) + m_{org}(D_p) + m_{rBC}(D_p)}$$
(S7)

where $f_{M,rBC}$ is the bulk mass fraction of refractory black carbon, $M_{rBC,b}$ and $M_{total,b}$ are the bulk rBC and total aerosol mass concentrations, respectively. From Eq. (S7), $m_{rBC}(D_p)$ can be derived as:

251
$$m_{rBC}(D_{p}) = \frac{f_{M,rBC}\left[m_{SO_{4}^{2-}}(D_{p}) + m_{NO_{3}^{-}}(D_{p}) + m_{NH_{4}^{+}}(D_{p}) + m_{org}(D_{p})\right]}{(1 - f_{M,rBC})}$$
(S8)

The fractional chemical composition at the size of CCN measurement D_p was then derived from the species mass concentrations calculated above. While not perfect, this approach allowed us to take both the temporal variation and size dependence of species mass concentrations into consideration when deriving particle composition at the sizes of CCN measurements. This is important because the organic volume fraction, required to derive the organic hygroscopicity was often much higher at the sizes of CCN measurements than that based on the bulk measurements.

258

260 **4.2 Sulfate and nitrate**

261 Outside of the IOP periods, particle composition was derived from ACSM data, and the 262 retrieval of organonitrate mass concentration is not available. In addition, the detection limit of $\mathrm{NH}^{\scriptscriptstyle +}_4$ for the ACSM is very high, such that the measurement of $\mathrm{NH}^{\scriptscriptstyle +}_4$ mass concentration was 263 too noisy for determining aerosol ionic balance. The sensitivity of derived $\kappa_{\rm org}$ to the assumption 264 265 on nitrate was examined using the AMS data from the two IOPs to inform the appropriate 266 assumptions for the analysis outside the IOP periods. Organic hygroscopicity was calculated 267 using three different approaches: (1) the approach based on the retrievals of both inorganic and 268 organic nitrate mass concentrations from AMS, as described in section 3.2 of the main text, (2) all NO_3^- was assumed from NH_4NO_3 and all sulfate from ammonium sulfate, and (3) all NO_3^- 269 was from organonitrate with all sulfate from ammonium sulfate. $\kappa_{\rm org}$ derived using the three 270 different assumptions are shown in Figure S9 for both IOPs. When all NO_3^- is assumed from 271 272 organonitrate (i.e., approach 3), the derived κ_{org} values are very close to those based on retrieved 273 organonitrate and inorganic nitrate concentrations from AMS data (i.e., approach 1). As a result, 274 for measurements outside of the two IOPs when only ACSM measurements are available, all measured NO_3^- was assumed from organonitrate and sulfate was assumed from ammonium 275 276 sulfate.



Figure S9: Diel variations of organic hygroscopicity derived using the three different assumptions on nitrate described in the text (i.e., mixed inorganic/organic nitrates, all organic nitrate, and all inorganic nitrate).



Figure S10: Size dependence of the fraction of the organic mass at m/z = 44 (f_{44}), which is indicative of the extent of oxidation. The f_{44} averaged during IOP1 was essentially independent of particle diameter D_p from 140 to 400 nm, a size range that dominated bulk organic mass concentration and encompasses the diameters of CCN measurements (i.e., 142 and 171 nm). During the IOP2, f_{44} averaged for all data and local biomass burning air mass type showed a clear dependence on D_p . In comparison, f_{44} averaged for the background condition and the periods with low hygroscopicity dispersion ($\sigma_{\overline{\kappa}}/\overline{\kappa} < 0.4$) during IOP2 had lower signal to noise ratio due

to the low mass loading, but were largely independent of particle size.

293 Tables

Table S1: Particle diameter classified by the DMA and the corresponding diameters for doubly and triply charged particles. Diameters in the parenthesis represent the measurements (if available) used to correct the contribution of doubly and triply charged particles to the measured activation fractions.

DMA classified singly charged particles, D_p (nm)	Doubly charged particles D_p (nm)	Triply charged particles D_p (nm)
51	75 (75)	94 (94)
75	112 (112)	142 (142)
94	142 (142)	182(171)
112	171 (171)	222 (222)
142	221 (222)	293 (N/A)
171	272 (N/A)	365 (N/A)
222	364 (N/A)	498 (N/A)

298

299

301 Table S2: Criteria used to classify air mass type for different seasons. The threshold values of CN

302	and CO concentrations used to identify background conditions are the mean plus one standard deviation
303	of respective measurements at the background sites T0a and T0t.

	Season			
Air mass type	Wet/Wet 2	Transition 1	Dry	Transition 2
Background	CN<500 cm ⁻³ CO<0.14 ppm NO _y <1.5ppbv	CN<900 cm ⁻³ CO<0.14 ppm NO _y <1.5ppbv	CN<1500 cm ⁻³ NO _y <1.5ppbv	CN<1500 cm ⁻³ NO _y <1.5ppbv
Urban Pollution	CN>500 cm ⁻³ * $f_{<70}$ > 0.45	$CN>900 \text{ cm}^{-3}$ $f_{<70} > 0.45$	$CN>1500 \text{ cm}^{-3}$ $f_{<70} > 0.45$	$CN>1500 \text{ cm}^{-3}$ $f_{<70} > 0.45$
Local biomass burning	CN>500 cm ⁻³ $f_{<70} < 0.45$ CO>0.14 ppm	CN>900 cm ⁻³ $f_{<70} < 0.45$ CO>0.14 ppm	$CN>1500 \text{ cm}^{-3}$ $f_{<70} < 0.45$	$CN>1500 \text{ cm}^{-3}$ $f_{<70} < 0.45$
Site Contamination	$BC>1 \ \mu g \ m^{-3}$ $CN>10^4 \ cm^{-3}$	$BC>1 \ \mu g \ m^{-3}$ $CN>10^4 \ cm^{-3}$	$BC>1 \ \mu g \ m^{-3}$ $CN>10^4 \ cm^{-3}$	BC>1 μ g m ⁻³ CN>10 ⁴ cm ⁻³

- 304 * $f_{<70}$: number fraction of particles with diameters less than 70 nm, derived from SMPS 305 measurement.

Table S3: Classification of air masses for size resolved CCN measurements at 112, 142, and 171

nm, and relevant measurements (CN, CO, SMPS, rBC) averaged to 5-min intervals.

		Percentage of air mass types		
Season	Classification	Size resolved CCN	Relevant measurements	
	Background	10.2%	17.5%	
	Urban Pollution	65.1%	63.7%	
Wet Season	Local Biomass Burning	0.7%	0.6%	
2014	Site Contamination	1.1%	1.3%	
	Invalid CCN spectra Fit [#]	12.5%	N/A	
	Unclassified ^{\$}	10.5%	17.0%	
	Background	8.9%	11.8%	
—	Urban Pollution	59.2%	59.1%	
Transition Season 1	Local Biomass Burning	3.4%	3.6%	
Season 1	Site Contamination	1.0%	1.2%	
	Invalid CCN spectra Fit [#]	4.0%	N/A	
	Unclassified ^{\$}	23.6%	24.2%	
	Background	7.9%	9.7%	
	Urban Pollution	14.4%	16.4%	
D	Local Biomass Burning	68.1%	58.6%	
Dry Season	Site Contamination	1.3%	1.1%	
	Invalid CCN spectra Fit [#]	2.9%	N/A	
	Unclassified ^{\$}	5.5%	14.3%	
	Background	26.5%	26.9%	
	Urban Pollution	11.0%	10.1%	
Transition	Local Biomass Burning	50.4%	53.8%	
Season 2	Site Contamination	1.7%	2.0%	
	Invalid CCN spectra Fit [#]	3.3%	N/A	
	Unclassified ^{\$}	7.0%	7.3%	
	Background	8.1%	12.6%	
	Urban Pollution	39.6%	41.2%	
Wet Season	Local Biomass Burning	25.6%	24.5%	
2015	Site Contamination	0.8%	0.9%	
	Invalid CCN spectra Fit [#]	8.3%	N/A	
	Unclassified ^{\$}	17.6%	20.8%	
	Background	12.4%	16.5%	
	Urban Pollution	38.5%	41.2%	
Mar 2014 -	Local Biomass Burning	28.4%	24.7%	
Mar 2015	Site Contamination	1.2%	1.3%	
	Invalid CCN spectra Fit [#]	6.5%	N/A	
	Unclassified ^{\$}	13.0%	16.3%	

- [#] An invalid fit of the CCN actiation spectrum (Sec. S3.2)
 ^{\$} One or more of the measurements (CN, SMPS, CO, rBC) neede to classify air mass is missing
- or air mass is not classifed as one of the four types using the the criteria described in Table S2.

324 **References**

Arnott, W. P., Hamasha, K., Moosmüller, H., Sheridan, P. J., and Ogren, J. A.: Towards Aerosol
Light-Absorption Measurements with a 7-Wavelength Aethalometer: Evaluation with a
Photoacoustic Instrument and 3-Wavelength Nephelometer, Aerosol Science and Technology,
39, 17-29, 2005.

- 329 Collaud Coen, M., Weingartner, E., Apituley, A., Ceburnis, D., Fierz-Schmidhauser, R., Flentje,
- H., Henzing, J. S., Jennings, S. G., Moerman, M., Petzold, A., Schmid, O., and Baltensperger,
 U.: Minimizing light absorption measurement artifacts of the Aethalometer: evaluation of five
- 332 correction algorithms, Atmos. Meas. Tech., 3, 457-474, 2010.
- Hansen, A. D. A., Rosen, H., and Novakov, T.: The aethalometer An instrument for the realtime measurement of optical absorption by aerosol particles, Science of The Total Environment,
 36, 191-196, 1984.
- 336 Kuwata, M., Zorn, S. R., and Martin, S. T.: Using Elemental Ratios to Predict the Density of
- 337 Organic Material Composed of Carbon, Hydrogen, and Oxygen, Environmental Science &
- 338 Technology, 46, 787-794, 2012.
- Lance, S., Nenes, A., Medina, J., and Smith, J. N.: Mapping the Operation of the DMT
 Continuous Flow CCN Counter, Aerosol Science and Technology, 40, 242-254, 2006.
- Mei, F., Hayes, P. L., Ortega, A., Taylor, J. W., Allan, J. D., Gilman, J., Kuster, W., de Gouw, J.,
 Jimenez, J. L., and Wang, J.: Droplet activation properties of organic aerosols observed at an
 urban site during CalNex-LA, Journal of Geophysical Research: Atmospheres, 118, 2903-2917,
 2013.
- Moteki, N., Kondo, Y., Miyazaki, Y., Takegawa, N., Komazaki, Y., Kurata, G., Shirai, T., Blake,
 D. R., Miyakawa, T., and Koike, M.: Evolution of mixing state of black carbon particles:
 Aircraft measurements over the western Pacific in March 2004, Geophys. Res. Lett., 34, L11803
 2007.
- Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic growth
 and cloud condensation nucleus activity, Atmos. Chem. Phys., 7, 1961-1971, 2007.
- Rizzo, L. V., Correia, A. L., Artaxo, P., Procopio, A. S., and Andreae, M. O.: Spectral
 dependence of aerosol light absorption over the Amazon Basin, Atmos. Chem. Phys., 11, 88998912, 2011.
- Roberts, G. C. and Nenes, A.: A Continuous-Flow Streamwise Thermal-Gradient CCN Chamber
 for Atmospheric Measurements, Aerosol Science and Technology, 39, 206-221, 2005.
- Rose, D., Gunthe, S. S., Mikhailov, E., Frank, G. P., Dusek, U., Andreae, M. O., and Poschl, U.:
 Calibration and measurement uncertainties of a continuous-flow cloud condensation nuclei
 counter (DMT-CCNC): CCN activation of ammonium sulfate and sodium chloride aerosol
 particles in theory and experiment, Atmos. Chem. Phys., 8, 1153-1179, 2008.

- Schmid, O., Artaxo, P., Arnott, W. P., Chand, D., Gatti, L. V., Frank, G. P., Hoffer, A., Schnaiter, M., and Andreae, M. O.: Spectral light absorption by ambient aerosols influenced by biomass burning in the Amazon Basin. I: Comparison and field calibration of absorption measurement techniques. Atmos. Chem. Phys. 6, 3443, 3462, 2006
- 363 measurement techniques, Atmos. Chem. Phys., 6, 3443-3462, 2006.
- 364 Schwarz, J. P., Spackman, J. R., Gao, R. S., Perring, A. E., Cross, E., Onasch, T. B., Ahern, A.,
- 365 Wrobel, W., Davidovits, P., Olfert, J., Dubey, M. K., Mazzoleni, C., and Fahey, D. W.: The
- 366 Detection Efficiency of the Single Particle Soot Photometer, Aerosol Sci. Technol., 44, 612-628,
- 367 2010.
 - 368 Shilling, J. E., King, S. M., Mochida, M., Worsnop, D. R., and Martin, S. T.: Mass Spectral
 - 369 Evidence That Small Changes in Composition Caused by Oxidative Aging Processes Alter
 - Aerosol CCN Properties, The Journal of Physical Chemistry A, 111, 3358-3368, 2007.
 - 371 Stephens, M., Turner, N., and Sandberg, J.: Particle identification by laser-induced 372 incandescence in a solid-state laser cavity, Appl. Optics, 42, 3726-3736, 2003.
 - 373 Virkkula, A., Chi, X., Ding, A., Shen, Y., Nie, W., Qi, X., Zheng, L., Huang, X., Xie, Y., Wang,
 - 374 J., Petaja, T., and Kulmala, M.: On the interpretation of the loading correction of the
 - aethalometer, Atmos. Meas. Tech., 8, 4415-4427, 2015.
 - 376 Virkkula, A., Mäkelä, T., Hillamo, R., Yli-Tuomi, T., Hirsikko, A., Hämeri, K., and Koponen, I.
 - 377 K.: A Simple Procedure for Correcting Loading Effects of Aethalometer Data, Journal of the Air
 - 378 & Waste Management Association, 57, 1214-1222, 2007.
 - Weingartner, E., Saathoff, H., Schnaiter, M., Streit, N., Bitnar, B., and Baltensperger, U.:
 Absorption of light by soot particles: determination of the absorption coefficient by means of
 aethalometers, Journal of Aerosol Science, 34, 1445-1463, 2003.
 - 382 Wiedensohler, A.: An approximation of the bipolar charge distribution for particles in the 383 submicron range, J. Aerosol Sci., 19, 387-390, 1988.
 - 384