## Response to anonymous referee #1

The authors greatly thank the reviewer for the interesting and constructive comments on the manuscript. We will try below to answer the questions and propose solutions. The reviewer's question is in italic, while the author's answer is below.

### General comments:

The interpretation of presented fluxes largely depends on the quality of the flux measurements. For the determination of fluxes using the closed-dynamic chamber technique the authors use several assumptions without discussing their validity. For example, it cannot be assumed that there are no interactions of NH3 with the chamber walls without adequate test experiments. While I highly acknowledge the fact that measurements in the present study region are challenging, I believe the validity of the used assumptions has to be tested. An accurate flux error assessment is especially necessary for NH3, which is subject to bi-directional exchange, and might explain some of the strong variability of the presented results.

The referee is totally right in stating that experimental tests are lacking on ammonia adsorption on Teflon chamber walls, in comparison with other surfaces. However, we have trusted the literature on that specific question.

Vaittinen et al. (2013, and references therein), have systematically assessed the adsorption of gas phase NH<sub>3</sub> on various surface materials. They have stated that polymers generally adsorb less NH<sub>3</sub> than stainless steel surfaces (giving that adsorption is the difference between the inlet and outlet mixing ratio).

As an example, in their Table 2, they assess that the adsorption of ammonia is  $13.9.10^{12}$  molecules.cm<sup>-2</sup> on PFA. The surface of our chamber is  $2700 \text{ cm}^2$ , which gives a value of  $37530.10^{12}$  molecules adsorbed (for a given concentration of 8.5 ppb, which is in the upper range of concentrations measured on the field). The volume of the chamber is  $12300 \text{ cm}^3$  (12.3 liters). One mole of gas occupies 22.4 liters, therefore the chamber contains 0.55 moles of gas, i.e.  $3.3.10^{23}$  molecules.

 $3.3.10^{23}$  molecules compared to  $37530.10^{12}$  gives an amount of  $1.14.10^{-5}$  % adsorbed on the surface of the chamber.

The surface of the Teflon tubing is  $2\pi*0.22*400 \text{ cm}^2 = 553 \text{ cm}^2$ , i.e. it may adsorb  $553*13.9.10^{12} = 7685.10^{12} \text{ molecules } (2.3.10^{-7} \%)$ .

We therefore suggest that the flux error, relative to ammonia adsorption on Teflon walls, is negligible.

## The following text has been added line 181:

No deposition occurs onto the Teflon walls chamber, as Vaittinen et al. (2013, and references therein) have demonstrated that the adsorption of ammonia on Teflon walls is negligible.

The authors use the measured NO and NH3 fluxes for a stepwise linear multiple regression analysis, upscaling to country-wide soil fluxes, and comparison with soil emission estimates from the GEOS-Chem model. These analyses give valuable information on the importance of soil NO and NH3 exchange and our current knowledge about them. However, while the authors state that a process understanding of the NO and NH3 fluxes is not within the scope of the presented study, in my opinion it is important to understand the underlying processes of the measured fluxes. For example, the estimated emissions from soil characteristics only poorly agree with the measured fluxes in some parts, which indicates that a more detailed process understanding is necessary.

The authors agree that a close understanding of the underlying processes of NO and NH<sub>3</sub> exchanges is important. There is a misunderstanding of what we wanted to state by writing that the understanding of these processes is not in the scope of this study. Indeed, in our study, we could not find any strong correlation between soil moisture and fluxes, or soil temperature and fluxes, because of the spatial set up of the experiment (with 4 types of soil described). However, this set up is useful to represent the spatial variability of fluxes. This may justify why we did not present any plot of flux magnitude vs environmental variable, but rather temporal evolution of fluxes and comparison with modeling results.

Line 499, we have removed the sentence "which is not in the scope of this study"

The study focuses on soil fluxes, which is why the authors do not discuss the impact of vegetation on the NO and NH3 fluxes. Especially for NH3, a present canopy may significantly alter the net ecosystem flux and I suggest to add a note including this aspect in the discussion of the manuscript.

Yes indeed, the role of vegetation in regulating NH<sub>3</sub> and NO fluxes from soils is of prime importance.

## **Line 409** we added the following paragraph:

The NO flux estimated in this study does not consider the impact of vegetation on the net ecosystem flux, as we focus on soil fluxes only. However, the net emission to the atmosphere should take into account the oxidation of NO to NO2 and the eventual re-deposition of NO2 on the vegetation, i.e. what is called Canopy Reduction Factor and is assumed to be a linear function of the Leaf Area Index (e.g. Yienger and Levy, 1995, and Ganzeveld et al., 2002)).

## **Line 446** we added the following paragraph:

Our measurements were conducted without vegetation inside the chambers, but vegetation was present in the fields. It is important to mention that the role of vegetation on NH3 bidirectional fluxes is essential, especially during the wet season (time of the experiment), when deposition on the vegetation through stomata and cuticles dominate the exchange (during rain events, the cuticular resistance becomes small and cuticular deposition dominates), due to an increase of the deposition velocity of NH3 (consecutive to the humidity response of the surface) and a decrease of the canopy compensation point, sensitive to the surface temperature and the surface wetness (Wichink-Kruit et al., 2007).

## Specific comments

L. 50: The study by Oswald et al. (2013) is on soil HONO emissions. Please cite here the original publication for NO emissions (IPCC or other original source).

The reference has been changed by the IPCC reference.

L. 159: Active charcoal is mainly suitable for medium to high molecular weight compounds and compounds with low volatility. Hence, I am surprised that the active charcoal was enough to remove all ambient air NH3. Was the quality of zero air source also tested against other methods? If so, please state this in the manuscript.

**Line 159,** the following explanations have been added:

"....and a second one to validate the efficiency of the NH<sub>3</sub> converter with a NH<sub>3</sub>/N<sub>2</sub> mixture diluted in pure air (Alphagaz 1, Airliquide). The zero air for NO, NO<sub>2</sub> calibration was obtained by filtering ambient air, previously passed on charcoal and desiccant cartridges."

L. 180: The assumption that the concentration in the chamber is equal to the concentration leaving the chamber to the analyzer is questionable. Due to the low flow rate required for the practical use of the closed-dynamic chamber technique, the residence time within the chamber is substantial (17-18 min). As no active mixing (e.g. with fan) is used, the chamber geometry in relation to the positioning of the ambient air inlet and sample outlet is of importance.

As described in Delon et al. (2017): The external volume of the chamber was  $40 \text{ cm} \times 20 \text{ cm} \times 20 \text{ cm}$ . The useful volume was  $18 \times 38 \times 18 \text{ cm} 3$  (12.31 or 0.0123m3), due to the thickness of the Teflonwalls. The air inlet is on one side of the chamber (a small vent of 4 mm in diameter provided the pressure equilibrium between the inside and outside of the chamber). The air outlet on the other side (40 cm away) is connected to the analyzer with a 4 m Teflon tube.

These specifications are not recalled in the manuscript, we just added:

Line 181: the details of the calculation "and the chamber design" are given in Delon et al. (2017).

L. 181-182: Especially NH3 is known to be a very sticky molecules and it cannot necessarily be assumed that it does not adsorb to Teflon material. E.g. from online NH3 measurements there is strong evidence that NH3 significantly interacts with the walls of used inlet Teflon tubing, already on a short time scale. The adsorption strength is thought to depend mainly on temperature, presence of NH3 and particulate matter on the Teflon surface, or relative humidity. Likewise, there could be a substantial effect depending on whether the manual chamber was cleaned before each measurement or not. Potential wall effects on NH3 fluxes are an important issue and should be addressed, e.g. by performing a field blank test or adequate laboratory test experiments.

This point was mentioned in the general comments, and considering the literature, we propose to neglect the adsorption of  $NH_3$  on Teflon walls and Teflon tubing (for particular temperature, humidity and concentration conditions). Of course, the temperature and humidity conditions in our experiment are different, but even if the number of adsorbed molecules increases, it may not bring discredit on our results A sentence has been added in the manuscript **line 181**. During this experiment, unfortunately no field blank test has been made.

## Line 173 we added the following text:

The Teflon chamber was cleaned at the beginning of each day of measurement, and during the day when the deposition of sand could potentially interfere with the measurements.

## Line 225 we added the following text:

The measurements have not been corrected from a possible interaction with particulate matter.

L. 190-193: Both the dilution effect and the detection limit are directly linked to the considered time interval. To my understanding, with longer time intervals the dilution effect increases and the detection limit decreases. As stated in the manuscript, for NO a shorter time interval (120s) was chosen than for NH3 (180-300s) (note here: in Delon et al. (2017), the time intervals for NO and NH3 were the opposite). According to this, the dilution effect is larger for NH3 than for NO, however, the stated detection limit is smaller for an NO than NH3, which should be the opposite. Please correct these inconsistencies or explain the differences in the revised manuscript.

## Thanks for mentioning this inconsistency.

Actually, the considered time interval varies in a range between 100 and 300s both for NH<sub>3</sub> and NO (this is different from Delon et al., 2017). The dilution effect is calculated separately for each flux

and the correction is 7.7 ( $\pm 1.7$ ) % in average for NH<sub>3</sub> fluxes, and 6.7 ( $\pm 1.6$ ) % in average for NO fluxes.

The minimum flux measurable is calculated from the precision of the instrument (and not from the detection limit), which is  $\pm 0.4$  ppbv f, for a 10s time interval. It does not depend on the time interval used for the calculation of the linear regression. The minimum flux detected by this device would therefore be 0.4 ngN m<sup>-2</sup> s<sup>-1</sup> for NH3 and for NO.

## Line 190 the paragraph has been changed into:

The linear regression is calculated over a 100 to 300 s time interval after the installation of the chamber on soil for both NO and NH3. Based on the methodology developed in Delon et al. (2017), the dilution effect due to mixing of outside air in the chamber is calculated for each flux separately and is in average  $6.7(\pm 1.6)$  % for NO and  $7.7(\pm 1.7)$  % for NH3. Considering the precision of the analyzer ( $\pm 0.4$  ppbv), the detection limit is 0.4 ngN m-2 s-1 for NO and NH3 fluxes

This is different from Delon et al., 2017, but fluxes were anyway superior to this value.

L. 202-203: Please state if a 1-sigma or 3-sigma detection limit is given here.

A 2-sigma detection limit is given, added line 203

L. 227-228: Key for the quality of the closed-dynamic chamber technique is the accurate determination of the initial concentration slope after the chamber installation. For this reason, the authors correctly omit fluxes where the slope is below a threshold correlation coefficient and the measured concentration difference is low. However, especially for NH3, where a R2 threshold value of 0.4 was chosen, the knowledge of the flux error is important for the further interpretation and might explain some of the presented flux variations. Therefore, the authors should include an estimate of the flux error associated with the linear regression and take that into account for the discussion of results.

## Line 229, the following point was added:

A flux error was estimated by calculating the dispersion of points around the linear regression's slope. According to this method, the dispersion for NO flux calculation is comprised between 5 and 12%, and the dispersion for NH<sub>3</sub> flux calculation is comprised between 15 and 20%...

L. 248-250: This assumption seems brave if it was not tested with a set of test experiments. Although the microbial activity is reduced due to the dry conditions, there is a chance that NH3 volatilizes with the drying of the soil sample material.

The authors are aware of this particular problem of NH<sub>3</sub> volatilization. The analyses have been made as soon as possible after the field campaign. A direct analysis was not possible, due to missing infrastructure. Freezing the samples would have been the best solution, but considering the difficulty of organizing the campaign in a place where the minimum material was installed, we could not afford to bring a freezer.

Line 257, new references have been added in the text:

Some authors have also published results of ammonium concentrations measured in soils that were dried in ambient air (Bai et al., 2010, Dick et al., 2006, Cassity-Duffrey et al., 2015).

Very few values of ammonium concentrations in African soils are available in the literature, it is therefore very difficult to compare.

Line 360, the following references have been added:

Dick et al. (2006) have found NH<sub>4</sub><sup>+</sup> concentrations between 2 and 8 mgN.kg<sup>-1</sup> in Senegalese soils, which is very close from our results. Vanlauwe et al. (2002) have found values between 0.8 and 1.4 mgN.kg<sup>-1</sup> in West African moist savanna soils (in Togo and Nigeria)..

L. 368-439: Presentation of NO and NH3 flux results: The flux at the soil-atmosphere interface is governed next to processes in the soil by the ambient trace gas concentration above the soil surface. The authors report relevant soil properties, while the atmospheric NO and NH3 mixing ratios from the chamber measurement are not reported. As they might significantly impact the magnitude and sign of the fluxes, the authors should include this information in the figures and the manuscript. This is especially important for the interpretation of the NH3 fluxes which are subject to bidirectional exchange and might explain some of the large flux variations observed.

Figure 2 to 5 have been modified and concentrations have been added (see new figures at the end of this document)

## **In section 3.4 line 391** we have added the following paragraph:

Daily means of NO concentrations are measured close to the soil (0.1m, half height of the chamber) and reported in fig. 2 to 5. Daily means of NO concentration vary from 1.28 to 5.40 ppb for all sites. The average concentration during the whole campaign on all sites is  $2.70 \pm 1.03$  ppb. Average NO concentration is  $2.97 \pm 1.49$  ppb on bare soil,  $2.57 \pm 0.96$  ppb on grassland,  $2.55 \pm 0.83$  on maize, and  $2.76 \pm 0.65$  ppb on forest soil. The concentrations are quasi equivalent for all sites. As these concentrations are low, they do not lead to NO deposition on soil and the NO flux stays positive. In fact, NO deposition has been measured in other studies only in the case of high NO concentrations (>60ppb, Laville et al., 2011).

## **In section 3.5 line 442** we have added the following paragraph:

As for NO concentrations, NH<sub>3</sub> concentrations are reported in fig. 2 to 5. Daily means of NH<sub>3</sub> concentration vary from 0 to 12.46 ppb for all sites, and the average concentration is  $4.42 \pm 3.23$  ppb during the whole campaign. Average NH<sub>3</sub> concentration is  $6.28 \pm 3.90$  ppb for bare soils,  $3.28 \pm 1.79$  ppb for grassland,  $4.36 \pm 3.99$  for the maize field, and  $3.68 \pm 2.13$  440 ppb for forest.. The largest deposition fluxes are found on bare soils, where the largest concentrations are measured.

L. 385-387: Why are the underground roots especially important for bare soil and the maize field? Are they more dominant than roots at the grassland and forest site?

### Line 403 the paragraph has been modified:

The spatial variability of NO fluxes is high, for bare soil, forest and the maize field where underground roots (not visible at the surface) are heterogeneously distributed. These roots are likely to influence the ammonium content of the soil and the subsequent NO flux measurement. Standard deviation is generally smaller for grassland (except for two days, July 9<sup>th</sup> and 13<sup>th</sup>), where the vegetation (and the root distribution) is more homogeneous.

L. 431-434: I agree with the authors in addressing the issue of the NH4+ adsorption capacity of soil particles when interpreting the results from the soil measurements. However, in this context it is also important what method was chosen to determine the soil NH4+. E.g. some common methods use a potassium chloride solution, to extract the soil NH4+. As a consequence, using a strong extraction solution might result in an overestimation of the emission potential.

Material and Method section mentions that mineral and organic nitrogen are determined following norm NF ISO 13878.

There is an error in the text. NF ISO 13878 is used for Total N. This has been corrected.

Line 263 the following text has been added:

Mineral nitrogen is determined following an internal method MT-AZM adapted from norm NF ISO 14256-2. This method uses a potassium chloride solution and is COFRAC certified.

PS: the concentration is unknown because the lab does not want to communicate on it.

L. 435-436: The authors bring up the potential of NH3 deposition on water film on vegetation surfaces, although the study focuses on soil emissions. Hence, it is important to mention (e.g. in method section) in case the chamber measurements also incorporated lower growing plant species (e.g. for grassland site) and include that in the interpretation of the results (i.e, stomatal emission potential).

All the measurements were made on direct soil without vegetation, even in the grassland field (the grass was too high to be included in the chamber). However, some short stems or leaves sometimes subsist within the chamber and were not removed to avoid any disturbance of the soil condition, but does not justify to take into account stomatal emission potential (because in too low quantity). The hypothesis of a water film on the vegetation is removed from the sentence; the hypothesis of a water film on the soil surface is left. (line 466).

Technical comments: corrections have been done.

## The following references were added:

Bai Junhong, Haifeng Gao, Wei Deng, Zhifeng Yang, Baoshan Cui, Rong Xiao, Nitrification potential of marsh soils from two natural saline—alkaline wetlands, Biol Fertil Soils (2010) 46:525–529.

Cassity-Duffey Kate, Miguel Cabrera, John Rema, Ammonia Volatilization from Broiler Litter: Effect of Soil Water Content and Humidity, Soil Sci. Soc. Am. J. 79:543–550, 2014.

Dick Jan, Ute Skiba, Robert Munro and Douglas Deans, Effect of N-fixing and non N-fixing trees and crops on NO and N2O emissions from Senegalese soils, Journal of Biogeography (J. Biogeogr.) (2006) 33,416–423.

Ganzeveld L.N., J. Lelieveld, F. J. Dentener, M. C. Krol, A. J. Bouwman, and G.-J. Roelofs, Global soil-biogenic NOx emissions and the role of canopy processes, JOURNAL OF GEOPHYSICAL RESEARCH, VOL. 107, NO. D16, 10.1029/2001JD001289, 2002.

Vaittinen O., M. Metsa la, S. Persijn, M. Vainio, L. Halonen, Adsorption of ammonia on treated stainless steel and polymer surfaces, Appl. Phys. B, DOI 10.1007/s00340-013-5590-3, 2013.

Vanlauwe B., J. Diels, O. Lyasse, K. Aihou, E.N.O. Iwuafor, N. Sanginga, R. Merckx & J. Deckers, Fertility status of soils of the derived savanna and northern guinea savanna and response to major plant nutrients, as influenced by soil type and land use management, Nutrient Cycling in Agroecosystems 62: 139–150, 2002.

Wichink Kruit, R. J., van Pul, W. A. J., Otjes, R. P., Hofschreuder, P., Jacobs, A. F. G., and Holtslag, A. A. M (2007), Ammonia fluxes and derived canopy compensation points over non-fertilized agricultural grassland in the Netherlands using the new gradient ammonia – high accuracy – monitor (GRAHAM), Atmos. Environ., 41, 1275–1287.

Yienger, J. J., and H. Levy II, Global inventory of soil-biogenic NOx emissions, J. Geophys. Res., 100, 11,447–11,464, 1995.

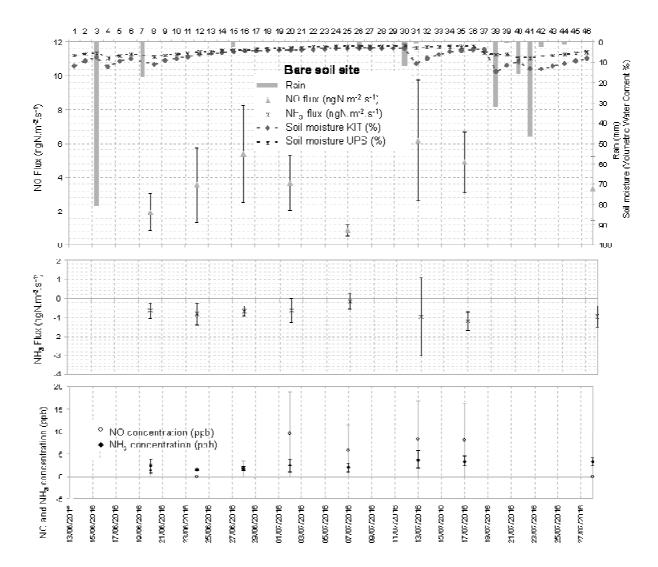


Fig. 2 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH $_3$  fluxes in ngN m $^{-2}$  s $^{-1}$  measured at the bare soil site; Lower panel: daily mean NO and NH $_3$  concentrations in ppb measured at the bare soil site. Vertical bars show the standard deviation from individual fluxes and concentrations.

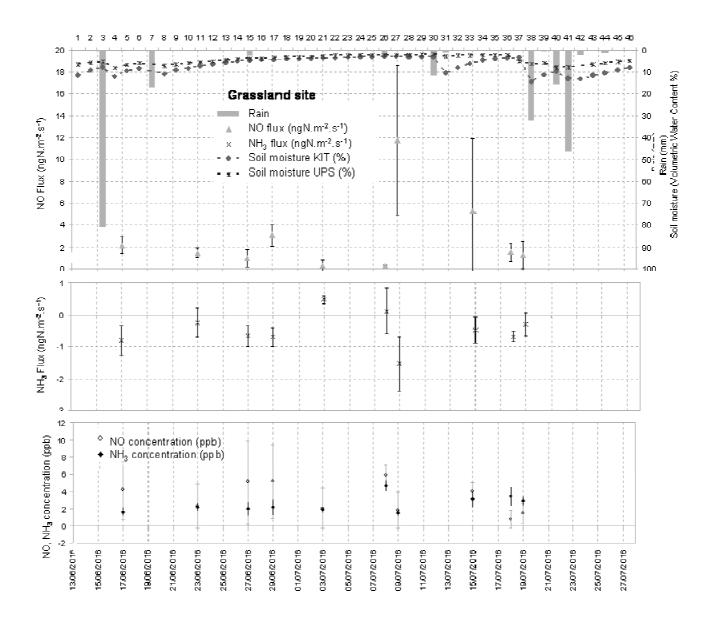


Fig. 3 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH $_3$  fluxes in ngN m $^{-2}$  s $^{-1}$  measured at the grassland site; Lower panel: daily mean NO and NH $_3$  concentrations in ppb measured at the grassland site. Vertical bars show the standard deviation from individual fluxes and concentrations.

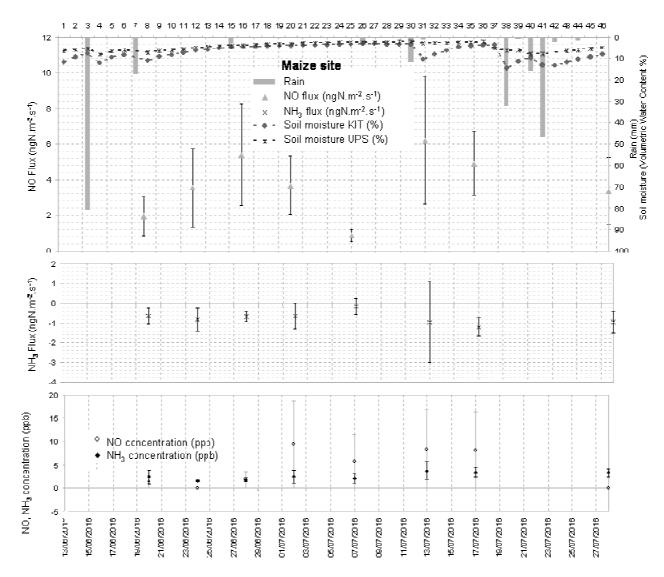


Fig. 4 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH $_3$  fluxes in ngN m $^{-2}$  s $^{-1}$  measured at the maize field site; Lower panel: daily mean NO and NH $_3$  concentrations in ppb measured at the maize field site. Vertical bars show the standard deviation from individual fluxes and concentrations.

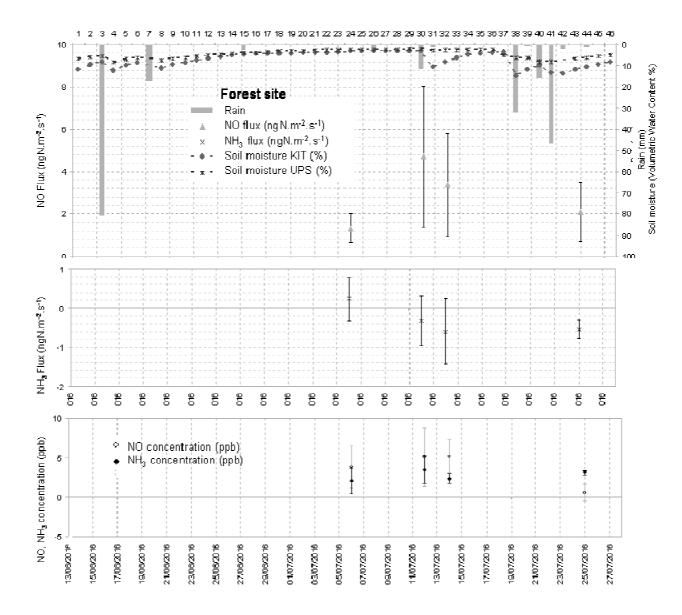


Fig. 5 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH $_3$  fluxes in ngN m $^2$  s $^{-1}$  measured at the forest site; Lower panel: daily mean NO and NH $_3$  concentrations in ppb measured at the forest site. Vertical bars show the standard deviation from individual fluxes and concentrations.

## Response to anonymous referee #2

The authors greatly thank the reviewer for the interesting and constructive comments on the manuscript. We will try below to answer the questions and propose solutions. The reviewer's question is in italic, while the author's answer is below. Line numbers where modifications are made are relative to the new version of the manuscript.

### General comments:

corrected.

While I believe the methodologies associated with atmospheric and gas flux measurements are sound, I am concerned with the soil sampling protocol and subsequent characterization, especially as it relates to mineral-N. It is not clear if the authors took 3-4 replicates for each landscape element each day, or a total of 3-4 replicates over the course of the campaign for each landscape element (lines 243-244).

The authors took a total of 3-4 replicates over the course of the campaign. On one particular day, samples were collected at the 4 different ecosystems. For example, on July 19<sup>th</sup> and 28<sup>th</sup>, samples were collected in bare soil, grassland, forest and maize field, whereas the forest site was not sampled on July 9<sup>th</sup>, and maize field was not sampled on July 6<sup>th</sup>. Moreover, the first collection date for grassland should be July 6<sup>th</sup> (instead of 7<sup>th</sup>). This will be

Line 246, the sentence has been modified in

Samples were collected at the four different land cover types, three to four times during the campaign

Soil C should not change (to the extent reported) over the course of 3 days (Bare Soil: 06/07-09/07) and can only be attributed to environmental heterogeneity

Actually samples were collected in the same area, but not exactly at the same place (another hole was dug, close to the first one). Therefore, the spatial heterogeneity of samples may explain de different results of organic carbon.

While I am sure location and resources had much to do with this, air-drying may result in large changes to ammonium concentrations. Additionally, significant changes in the amounts of ammonium can take place over prolonged storage at room temperature, even if soils are dried. It seems that the authors are aware of this issue and attempted to justify their method by citing a meta-analysis of warming experiments on N-cycle activity. (Bai et al. 2013). However, this meta-analysis found that warming and moisture reduction had no significant effect on mineralization (Bai et al, 2013: Table 1), indicating even in dried samples, pools of inorganic-N may change over time. To remedy this, the authors could have compared their ammonium concentrations to similar studies from this region; however, this was not included in the results/discussion.

The authors are aware of this particular problem of NH<sub>3</sub> volatilization. The analyses have been made as soon as possible after the field campaign. A direct analysis was not possible, due to missing infrastructure. Freezing the samples would have been the best solution, but considering the difficulty of organizing the campaign in a place where the minimum material was installed, we could not afford to bring a freezer.

Some authors have also published results of ammonium concentrations measured in soils that were dried in ambient air (Bai et al., 2010, Dick et al., 2006, Cassity-Duffrey et al., 2015).

Moreover, very few values are available in the literature, it is therefore very difficult to compare. The results presented in section 3.3 are consistent with Massad et al. (2010) which may provide a certain confidence in analysis. Dick et al. (2006) have also found NH<sub>4</sub><sup>+</sup> concentrations between 2 and 8 mgN.kg<sup>-1</sup> in Senegalese soils, which is very close from our results. Vanlauwe et al. (2002) have found values between 0.8 and 1.4 mgN.kg<sup>-1</sup> in West African moist savanna soils (in Togo and Nigeria).

## Line 255, new references have been added in the text:

Some authors have also published results of ammonium concentrations measured in soils that were dried in ambient air (Bai et al., 2010, Dick et al., 2006, Cassity-Duffrey et al., 2015).

## **Line 360**, the following references have been added:

Dick et al. (2006) have found NH<sub>4</sub><sup>+</sup> concentrations between 2 and 8 mgN.kg<sup>-1</sup> in Senegalese soils, which is very close from our results. Vanlauwe et al. (2002) have found values between 0.8 and 1.4 mgN.kg<sup>-1</sup> in West African moist savanna soils (in Togo and Nigeria)..

In regards to the tables and figures, the authors should strongly consider merging Figures 2-5. As they currently sit, there is a large amount of redundancy.

As reviewer #1 asked to add NO and NH<sub>3</sub> concentrations in the figures, we chose not to merge figures 2 to 5, it would have been impossible to read.

### References

Bai Junhong, Haifeng Gao, Wei Deng, Zhifeng Yang, Baoshan Cui, Rong Xiao, Nitrification potential of marsh soils from two natural saline—alkaline wetlands, Biol Fertil Soils (2010) 46:525–529.

Cassity-Duffey Kate, Miguel Cabrera, John Rema, Ammonia Volatilization from Broiler Litter: Effect of Soil Water Content and Humidity, Soil Sci. Soc. Am. J. 79:543–550, 2014.

Dick Jan, Ute Skiba, Robert Munro and Douglas Deans, Effect of N-fixing and non N-fixing trees and crops on NO and N2O emissions from Senegalese soils, Journal of Biogeography (J. Biogeogr.) (2006) 33, 416–423.

Vanlauwe B., J. Diels, O. Lyasse, K. Aihou, E.N.O. Iwuafor, N. Sanginga, R. Merckx & J. Deckers, Fertility status of soils of the derived savanna and northern guinea savanna and response to major plant nutrients, as influenced by soil type and land use management, Nutrient Cycling in Agroecosystems 62: 139–150, 2002.

# Measurements of nitric oxide and ammonia soil fluxes from a wet savanna ecosystem site in West Africa during the DACCIWA field campaign.

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### 15 Correspondence to: Claire Delon (claire.delon@aero.obs-mip.fr)

**Abstract.** It is important to correctly simulate biogenic fluxes from soil in atmospheric chemistry models at a local and regional scale to study air pollution and climate in an area of the world, West Africa, that has been subject to a strong increase in anthropogenic emissions due to a massive growth in population and urbanization. Anthropogenic pollutants are transported inland and northward from the mega cities located on the coast, where the reaction with biogenic emissions may lead to enhanced ozone production outside urban areas, as well as secondary organic aerosols formation, with detrimental effects on humans, animals, natural vegetation and crops.

Here we present field measurements of soil fluxes of nitric oxide (NO) and ammonia (NH<sub>3</sub>) observed over four different land cover types, i.e. bare soil, grassland, maize field and forest, at an inland rural site in Benin, West Africa, during the DACCIWA field campaign in June and July 2016.

We observe NO fluxes up to 48.05 ngN m<sup>-2</sup> s<sup>-1</sup>. NO fluxes averaged over all land cover types are  $4.79 \pm 5.59$  ngN m<sup>-2</sup> s<sup>-1</sup>, maximum soil emissions of NO are recorded over bare soil. NH<sub>3</sub> is dominated by deposition for all land cover types. NH<sub>3</sub> fluxes range between -6.59 and 4.96 ngN m<sup>-2</sup> s<sup>-1</sup>. NH<sub>3</sub> fluxes averaged over all land cover types are -0.91  $\pm$  1.27 ngN m<sup>-2</sup> s<sup>-1</sup> and maximum NH<sub>3</sub> deposition is measured over bare soil. The observations show high spatial variability even for the same soil type, same day and same meteorological conditions.

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We compare point daily average measurements of NO emissions recorded during the field campaign with those simulated by GEOS-Chem (Goddard Earth Observing System Chemistry Model) for the same site and find good agreement. In an attempt to quantify NO emissions at the regional and national scale, we also provide a tentative estimate of total NO emissions for the entire country of Benin for the month of July using two distinct methods: upscaling point measurements and using the GEOS-Chem model. The two methods give similar results:  $1.17 \pm 0.6$  GgN/month and 1.44 GgN/month, respectively. Total NH<sub>3</sub> deposition estimated by upscaling point measurements for the month of July is 0.21 GgN/month.

### 1 Introduction

- Biogenic soil fluxes of nitric oxide (NO) and ammonia (NH<sub>3</sub>) play an important role on tropospheric chemistry. Nitric Oxide emitted by soil influences the concentration of nitrogen oxides (NO<sub>x</sub>) in the atmosphere, consequently modifying the rates of ozone (O<sub>3</sub>) production, where O<sub>3</sub> is a pollutant, harmful to humans and plants, and also a greenhouse gas (Steinkamp et al., 2009). The production and consumption of NO in soil is regulated by microbial activity, mainly nitrification/denitrification processes, and chemical reactions (Pilegaard et al., 2013). Measurements using soil chambers in the field and laboratory experiments show that nitrification/denitrification, and consequently NO emissions, vary greatly with climate and soil conditions, in particular they are strongly correlated with nitrogen (N) availability, temperature and soil moisture, making soil NO emissions dependent on regional temperature and precipitation patterns, and fertilizer management practices (e.g., Bouwman et al., 2002; Meixner and Yang, 2006; Hudman et al., 2010).
- Soil NO emissions are about 20% of total NO sources to the atmosphere (PCC, 2007) and almost of the same order of magnitude of fossil fuel NO emissions. Soil emission of biogenic NO plays a prominent role in the regional atmospheric chemistry of non-urbanized areas, where anthropogenic emissions are negligible (Pilegaard, 2013). The main inputs of N compounds onto semi arid uncultivated soils, like the savanna ecosystem, are biological nitrogen fixation, atmospheric wet and dry deposition and lightning. NO fluxes are considered as one way only, even if NO deposition exists in very specific conditions (Grote et al., 2009).

Soil N losses towards the atmosphere also involve NH<sub>3</sub>. The largest source of NH<sub>3</sub> emissions is agriculture, via the application of synthetic fertilizer. When released into the atmosphere, NH<sub>3</sub> increases the level of air pollution. In the atmosphere NH<sub>3</sub> has a relatively short life time of less than five days and high deposition rates, it is converted into ammonium (NH<sub>4</sub><sup>+</sup>) aerosols, which has a life time of the order of fifteen days, can travel long distances and it is relevant for air quality and climate (Fuzzi et al., 2015). The exchange of soil NH<sub>3</sub> is bi-directional as it also includes deposition. NH<sub>3</sub> returned to the surface by deposition can potentially cause eutrophication, reducing biodiversity and water quality (Sutton et al., 2009a).

65 The net flux of NH<sub>3</sub> is the combination of different exchange pathways between plant (cuticle and stomata), soil, leaf litter and atmosphere. The overall NH<sub>3</sub> flux for a given surface may switch from net emission to net deposition at sub-hourly, diurnal and seasonal scales. Moreover, NH<sub>3</sub> can be rapidly deposited onto cuticles due to its high solubility (e.g. Sutton et al., 2009b; Massad et al., 2010; Loubet et al., 2012).

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The direction and magnitude of NH<sub>3</sub> exchanges depend on the difference in NH<sub>3</sub> concentration between the canopy and the atmosphere, and on a large range of environmental factors, in particular air humidity, which influence surface wetness, and soil moisture conditions, but also vegetation cover and soil characteristics. The relationships between NO and NH<sub>3</sub> soil fluxes have been identified through the ammonium content in the soil (McCalley and Sparks, 2008). Ammonia is mainly emitted by agricultural activities, and also by the decomposition of litter and volatilization of animal excreta (Sutton et al., 2009b; Massad et al., 2010).

Soil fluxes in West Africa have only been measured in a limited number of studies due to the challenging experimental conditions (remote sites, no power supply, very hot temperatures), and mainly with manual chamber techniques rather than more complex micrometeorological techniques (Serça et al., 1998, Le Roux et al. 1995 for NO, Delon et al., 2017 for NO and NH<sub>3</sub>). However, tropical savanna has been recognized as one of the ecosystems characterized by the largest NO emissions (Davidson and Kingerlee, 1997, Hudman et al., 2012).

Anthropogenic emissions of pollutants from mega cities located on the Guinean coast in South West Africa have been increasing, and are likely to keep increasing in the next decades, due to a strong anthropogenic pressure, land use change and urbanization. When transported northward on the African continent, polluted air masses meet biogenic emissions from rural areas which contributes to increased O<sub>3</sub> and secondary organic aerosol production, in high temperature and solar radiation conditions, highly favorable to enhance photochemistry (Knippertz et al., 2015a, 2015b).

The objectives of this study are to quantify soil fluxes of NO and NH<sub>3</sub> for the different land cover types typical of rural West 90 Africa, suggest a tentative strategy to scale point measurements in the field to ecosystem and larger regional scale, and provide data for inventories and model evaluation to improve air quality and climate modelling.

In this paper we present the soil fluxes of NO and NH<sub>3</sub> measured in a rural site near the city of Savè, Benin, West Africa, during the DACCIWA (Dynamics-Aerosol-Chemistry-Cloud Interactions in West Africa) field campaign which lasted from 14<sup>th</sup> June to 30<sup>th</sup> July 2016 (wet season). The DACCIWA campaign was lead to investigate the possible role of local air pollution on climate change in West Africa, focusing on atmospheric composition, air pollution and cloud–aerosol interactions over several sites in the region (Knippertz et al., 2015a, 2015b, 2017). The Savè site is part of the savanna ecosystem, where grassland is intercut with crops and degraded forest. Biogenic soil fluxes measurements were taken using the manual chamber technique, which is robust and of reduced costs (Delon et al., 2017). Along with these observations we also present measurements of soil characteristics and meteorological variables from the same site. We include the comparison of measured NO soil emissions with those simulated by the Hudman et al. (2012) process-based model for NO soil emission implemented into GEOS-Chem.

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### 2 Material and method

### 2.1 Site description

The Savè site for ground-based observations is located in a hinterland area of Benin, 6 km south west from the city of Savè (8°02'03" N, 2°29'11" E, 166 m a.s.l.). The Savè ground-based observation site is located within the Gobè site managed by the Institut National des Recherches Agricoles du Bénin (INRAB).

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The site is characterized by a wet savanna ecosystem. The climate of the region is Sudano-Guinean, with a rainy season from March to October and a dry season from November to February (Michels et al., 2000). The average annual rainfall is about 1100 mm (Savè weather station, data averaged from 1969 to 2004, Michels et al., 2000 and Säidou et al., 2004) and the average yearly temperature is about 27.5 °C with little variation from year to year (data averaged from 1984 to 2004, Säidou et al., 2004). Average minimum temperature, based on 1969-1990 data, is 21.5 °C and mean maximum temperature is 35.5 °C.

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The tree coverage in the Savè region is low with most of the land occupied by subsistence agriculture and grassland (CILSS, 2016). Four land cover types are identified at the observation site: bare soil, grassland, maize field and degraded forest. Bare soil is defined as a patch of land of minimum five by five meters wide, without vegetation growing or hanging over the plot. Ground photographs of the four land cover types are shown in Fig.1.

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The most abundant tree species next to the grassland site and in the forest are: Anacardium occidentale, Daniellia oliveri, and Pterocarpus erinaceus; while the most abundant tree species next to the maize field are: Mangifera indica, Cocos nucifera, Carica papaya L., Tectona grandis, and Azadirachta indica. The herbaceous vegetation is dominated by Cleome sp., Crotalaria sp., Mucuna sp., Imperata cylindrica and Rhynchelytrum repens next to the grassland site and in the forest, and Commelina benghalensis, Euphorbia sp., Boerhavia diffusa, Phyllanthus amarus, Digitaria horizontalis by maize field. In the maize field, the main species, Zea Mays, is intercropped with Sesamum indicum and, to a lesser extent, with other species: Dioscorea sp., Manihot esculenta, Arachis hypogaea, Vigna unguiculata, Gossypium sp., Sorghum sp. and Solanum lycopersicum. The maize field was not treated with mineral fertilizer. The only livestock consist of a few dozens of domestic fowls belonging to small subsistence-oriented family farms, mainly grazing in the maize field.

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At the Savè site, the soil is sandy, with 87% of sand and 4.1% of clay (the rest being silt) for the 0-5 cm horizon. Surface pH ranges from 6.32 to 8.46, depending on the place where the measurement is done. Mean meteorological and average soil characteristics for the observation site are reported in Table 1, dominant vegetation species and soil composition for each land cover type are given in Table 2 and 3, respectively. Sunrise and sunset UTC time at the beginning and at the end of the campaign are: 05:33 and 18:08 on 14<sup>th</sup> June, and 05:42 and 18:11 on 30<sup>th</sup> July.

### 2.2 Sampling sites

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The samples were taken at the four land cover types (bare soil, grassland, maize field and forest), one location per day. Two to three samplings spots were chosen each day for each location, collecting eight to twenty-five flux measurements for both NO and NH<sub>3</sub> soil fluxes each day. Each location was sampled during daytime, approximately from 7 a.m. to 6 p.m., alternating measurements at the four different land cover types from one day to the other, over the entire campaign. Bare soil and the maize field were sampled for both NO and NH<sub>3</sub> soil fluxes on eight different, generally non-consecutive, days, grassland on ten days, and the forest site on four different days.

#### 2.3 Chamber flux measurements

The technique used to measure NO and NH<sub>3</sub> soil fluxes makes use of a Thermoscientific 17i (ThermoFischer Scientific, MA, USA) to measure the concentration of NO and NH<sub>3</sub> through chemiluminescence, and the closed dynamic chamber technique to calculate fluxes. The details of this technique are fully described in Delon et al. (2017).

The remoteness of the study site limited installation of permanent structures and we were unable to automate our chamber measurements, thus all measurements were made manually. The instrument was powered by a generator (>100 m away) and carried around on a wheeled-table to reach the locations of the four soil types where the NO and NH<sub>3</sub> soil fluxes were measured. The analyzer was connected via a Teflon tube to the Teflon chamber which was put on the ground to detect the fluxes. The external sides of the chamber were covered with sand or soil to isolate it during the measurement. The soil under the chamber was left unperturbed. Adjustments were taken in order to make sure the analyzer did not reach temperatures that would invalidate the measurements.

The calibration of the NO sensor of the 17i analyzer was made before and after the campaign, with a reference  $NO_2$  air mixture, i.e. NO in  $N_2$  diluted with zero air. Two post-campaign calibrations were made: a first one to validate the efficiency of the  $NO_2$  converter using a reference dilution of  $NO_2$  in zero air, and a second one to validate the efficiency of the  $NH_3$  converter with a  $NH_3/N_2$  mixture diluted in pure air (Alphagaz 1, Airliquide). The zero air for  $NO_2$  calibration was obtained by filtering ambient air, previously passed on charcoal and desiccant cartridges. The dilution for all the calibration experiments was made with the 146i module (ThermoFischer Scientific, MA, USA) and the dilution module, equipped with certified mass flow meters, on board of the ATR-42 research aircraft during an inter-calibration with other  $NO_x$  instrumentation of the DACCIWA campaign (i.e. the instrumentation on the Savè measurement site tower and the instrumentation on the ATR-42 aircraft; Brito et al., 2017, Derrien et al., 2016). Reference  $NO_2$  and  $NO_2$  were ISO 6141:2015 certified at 8.73 and 8.58 ppm for  $NO_2$  before and after the campaign, respectively, and 9.28 ppm for  $NO_2$ , both with 5% precision. Reference  $NH_3$  mixture was certified at 14.78 ppm with 2% precision for  $NH_3$ . Multipoint (at least 4

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points) calibrations between 50 to 250 ppb were done to ensure the linearity of the response, obtaining regression coefficients over 0.9993 for both NO and NO<sub>2</sub>. The global precision of the analyzer is  $\pm 0.4$  ppb.

The air inlet is located on one side of the chamber, where a small vent of 4 mm in diameter provided the pressure equilibrium between the inside and outside of the chamber. The air outlet on the other side is connected to the analyzer with a 4 m Teflon tube. The chamber is continuously swept with an air flow of 0.7 L min<sup>-1</sup> insured by the instrument pump, and the air flow is controlled inside the analyzer by a flow meter. The Teflon chamber was cleaned at the beginning of each day of measurement, and during the day when the deposition of sand could potentially interfere with the measurements.

The opaque walls minimize photochemical reactions inside the chamber, which are therefore considered as negligible. The chamber is placed on the soil for 10 min. After 10 min, the chamber is turned over to let the analyzer be swept by ambient air for 5 min, then the chamber is placed again on the soil to begin a new cycle.

The calculation of the fluxes are based on the closed dynamic chamber technique, with the following assumptions: the concentration in the chamber is equal to the concentration leaving the chamber to the analyzer, no deposition occurs onto the Teflon walls chamber as Vaittinen et al. (2013, and references therein) have demonstrated that the adsorption of ammonia on Teflon is negligible; chemical reactions in the gas phase inside the chamber are limited, thanks to the opaque walls. All the details of the calculation and the chamber design are given in Delon et al. (2017). In brief:

$$F_{x} = \frac{V}{A_{0}} \frac{\delta C_{x}}{\delta t} \tag{1}$$

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Where  $F_X$  is the flux (NO or NH<sub>3</sub>) in nmol m<sup>-2</sup> s<sup>-1</sup>,  $\delta C_X$  is the concentration increase in the chamber in nmol m<sup>-3</sup> during the temporal interval  $\delta t$ .  $A_0$ =0.0684 m<sup>2</sup> is the surface of the ground covered by the chamber, V=0.0123 m<sup>3</sup> is the volume of the chamber. This equation is similar to the one in Davidson et al. (1990). The flux is then converted to ngN m<sup>-2</sup> s<sup>-1</sup>.

The linear regression is calculated over a 100 to 300 s time interval after the installation of the chamber on soil for both NO  $_{11}^{11}$  and NH<sub>2</sub>. Based on the methodology developed in Delon et al. (2017), the dilution effect due to mixing of outside air in the  $_{11}^{11}$  chamber is calculated for each flux separately and is in average 6.7(±1.6) % for NO and 7.7(±1.7) % for NH<sub>3</sub>. Considering  $_{11}^{11}$  the precision of the analyzer (±0.4 ppbv), the detection limit is 0.4 ngN  $_{12}^{-2}$  s<sup>-1</sup> for NO and NH<sub>3</sub> fluxes.

The chemical reactions inside the chamber can determine NO consumption, and consequently an underestimation of the NO fluxes calculated with our method. This underestimation is taken into account and calculated following the method by Pape et al. (2009) with the relation  $k \cdot [NO] \cdot [O_3]$ . In this relation k is the temperature-dependent reaction rate constant (Pape et al., 2009, Atkinson et al., 2004), [NO] is measured by the Thermoscientific 17i at soil level just before positioning the chamber

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for the measurement of soil fluxes, and [O<sub>3</sub>] at soil level is derived by measurements of NO and NO<sub>2</sub> at soil level made with the Thermoscientific 17i and measurements of NO, NO<sub>2</sub> and O<sub>3</sub> taken on an 8 m high tower. On the 8 m high tower, NO and NO<sub>2</sub> were measured with a Model 42C TraceLevel NO-NO<sub>2</sub>-NOx by Thermo-environmental Instruments Inc., calibrated with the same method as the Thermoscientific 17i and with 0.05 ppb (2-sigma) detection limit. Ozone was measured on the tower with a Model 49i Ozone Analyzer by Thermo-environmental Instruments Inc. with 1 ppb detection limit. The Model 49i Ozone Analyzer was calibrated by comparison with a Thermo Scientific Model 49PS reference instrument. The reference instrument is sent twice a year to the French Laboratoire national d'Essais (LNE) for comparison with a National Institute of Standards and technology (NIST). All data on the tower were sampled at 10 seconds. [O<sub>3</sub>] at soil level was then calculated considering the diurnal steady state of the reactions described in equation (2) and (3), using equation (4):

210 NO + 
$$O_3 \rightarrow NO_2 + O_2$$
 (2)

$$NO_2 \rightarrow NO + O$$
 (3)

$$[O_3]_{il} = \frac{[NO]_{i}[O_3]_{i}}{[NO_2]_{il}} \frac{[NO_2]_{il}}{[NO]_{il}}$$
(4)

Where []<sub>sl</sub> is the concentration at the soil level and []<sub>tl</sub> is the concentration measured on the tower. In conclusion, we correct NO fluxes for the underestimation of NO fluxes due to chemical reactions inside the chamber with values ranging between 0 and 63% (8% on average for the whole campaign).

As studied by Kristensen et al. (2010a) and Kristensen et al. (2010b), O<sub>3</sub> deposition can decrease O<sub>3</sub> concentration close to soil surface further. However, considering that O<sub>3</sub> concentrations calculated near the soil are already very low (1 ppb at soil level compared to 24 ppb at 8 m, averaged for the entire measurement campaign), O<sub>3</sub> deposition has been considered of secondary importance in this calculation and has not been included. If O<sub>3</sub> deposition were to be included it would possibly decrease the correction of NO fluxes and consequently slightly decrease NO emissions in a negligible proportion compared to the correction already applied for the chemical reactions inside the chamber.

The measurements have not been corrected from a possible interaction with particulate matter.

## 225 **2.4 Data quality check**

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A quality check method based on the following criteria is used to select observed fluxes (Delon et al., 2017):

- The coefficient of determination for linear regression  $R^2$  has to be higher than 0.4 (considered as a significant correlation) for NH<sub>3</sub> fluxes, and higher than 0.8 for NO fluxes.

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- A flux error was estimated by calculating the dispersion of points around the linear regression's slope. According to this method, the dispersion for NO flux calculation is comprised between 5 and 12%, and the dispersion for NH<sub>8</sub> flux calculation is comprised between 15 and 20%

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- The concentration difference between the last and the first  $NH_3$  measurement point has to be more than 0.4 ppb (sensitivity of the analyzer).  $R^2$  was generally lower than 0.4 for concentration differences below 0.4 ppb.

Finally, 351/488 (72%) NH<sub>3</sub> flux measurements and 459/488 (94%) NO flux measurements are considered valid.

### 235 2.5 Meteorological station

Continuous in situ observations of meteorological variables, including air and soil temperature and moisture, rainfall, wind speed, wind direction, radiation and energy balance components were taken at the Savè site as part of the DACCIWA campaign. Data are provided as 1 min averages, apart from energy fluxes which are given as 30 min averages (Derrien et al., 2016, Kohler et al., 2016, Handwerker et al., 2016, Wieser et al., 2016). An overview of the complete set of instrumentation and measurements is given by Brooks et al. (2017), while a summary of the available ground-based meteorological observations is given by Kalthoff et al. (2017). In this study we present soil moisture measured in two distinct locations of the Savè site by the Karlsruhe Institute of Technology (KIT) instrumentation at 5cm depth on grassland, and average soil moisture, between 0 and 30 cm, measured by the Université Paul Sabatier (UPS) instrumentation in the maize field. Details of the instrumentation is given by Brooks et al. (2017). We include soil moisture measured with both systems as the intercomparison of the two methods is out of the scope of this study.

#### 2.6 Soil characteristics (texture, pH, N content)

Soil samples were collected during the measurement campaign to analyze the biogeochemical characteristics of the site. Soil samples (0-5 cm) were taken for each land cover type where NO and NH<sub>3</sub> fluxes were measured. Samples were collected at the four different land cover types, three to four times during the campaign.

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Samples were dried in ambient conditions (mean day-time temperature is approximately 26 °C, Kalthoff et al., 2017), and stored in the dark for the determination of texture, ammonium concentrations [NH<sub>4</sub><sup>+</sup>], C/N ratio, total C, total N, and pH at the GALYS Laboratoire (http://www.galys-laboratoire.fr, NF EN ISO/CEI 17025: 2005). The analyses were performed two months after sampling. We assume that the ammonium content in litter or soils is not modified by volatilization or chemical transformation during transport and storage, because of the very low soil moisture level in samples. Past studies have demonstrated that microbial activity and mineralization processes are inhibited in low soil moisture conditions, even when soil temperature is high (Bai et al., 2013, and references therein). Some authors have also published results of ammonium concentrations measured in soils that were dried in ambient air (Bai et al., 2010, Dick et al., 2006, Cassity-Duffrey et al., 2015). Soil texture is determined following norm NF X 31e107. Clay (<2 μm), fine silt (2 to 20 μm), coarse silt (20 to 50 μm) and total sand (50 to 2000 μm) are determined without decarbonation. Organic carbon and total carbon are determined

**Supprimé:** Three to four samples were collected for each land cover type on different days during the campaign.

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following norm NF ISO 10694. The whole carbon of the sample is transformed into CO<sub>2</sub>. Then CO<sub>2</sub> is measured by thermal conductibility. NF ISO 13878 is used for Total N. Mineral nitrogen is determined following an internal method MT-AZM adapted from norm NF ISO 14256-2. This method uses a potassium chloride solution and is COFRAC certified. The sample is heated at 1000°C with O<sub>2</sub>. Products of combustion or decomposition are reduced in N<sub>2</sub>. N<sub>2</sub> is then measured by thermal conductibility (catharometer). pH is determined according to norm NF ISO 10390, with soil samples stirred with water (ratio 1/5).

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### 2.7 Soil ammonia emission potential $\Gamma_g$ and compensation point $\chi_g$

Measurements of soil pH and ammonium concentrations  $[NH_4^+]$  are used to quantify the soil emission potentials for the different land cover types at the measurement site. The soil emission potential  $\Gamma_g$  is the ratio of  $[NH_4^+]$  to  $[H^+]$  concentrations in the water solution of the soil (mol  $L^{-1}$ ). A large  $\Gamma_g$  indicates that the soil has a high propensity to emit  $NH_3$ , considering that the potential emission of  $NH_3$  depends on the availability of ammonium in the soil and on pH.

The soil compensation point  $(\chi_g)$  has been calculated from the emission potential  $\Gamma_g$ , as a function of soil surface temperature  $(T_g \text{ in } K)$  according to Wentworth et al. (2014):

$$\chi_g \text{ (ppb)} = 13587 \cdot \Gamma_g \cdot e^{-(10396K/Tg)} \times 10^9,$$
 (5)

The soil compensation point indicates the equilibrium between gaseous  $NH_3$  in the soil pore space and  $[NH_4^+]$  in the soil solution, i.e. the concentration of  $NH_3$  for which the  $NH_3$  flux switches from emission to deposition (or viceversa).

### 280 2.8 Stepwise multiple regression analysis

A stepwise linear multiple regression analysis was performed between daily averaged gas fluxes of NO and  $NH_3$  and relevant available daily averaged variables such as wind speed, soil temperature at 5 cm, soil moisture at 5 cm, soil heat flux, outgoing longwave radiation and incoming shortwave radiation. Soil parameters such as mineral nitrogen, total N and organic C, soil texture and pH could not be used for this regression analysis since their relative measurements did not have the same temporal resolution as the other parameters. The R software (http://www.R-project.org) was used to provide the results of this linear regression analysis.

## 2.9 GEOS-Chem

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GEOS-Chem is a global three-dimensional model of tropospheric chemistry driven by meteorological input from the NASA Goddard Earth Observing System (www.geos-chem.org, Bey et al., 2001). In this study we use GEOS-Chem Version 10-01 which includes the process-based parameterization of soil NO emission by Hudman et al. (2012). This parameterization represents available nitrogen (N) in soils using biome specific emission factors, online wet and dry deposition of N, and

fertilizer and manure N derived from a spatially explicit dataset, distributed using seasonality derived from data obtained by the Moderate Resolution Imaging Spectrometer (MODIS). Emissions are a smooth function of soil moisture and temperature consistent with point measurements and ecosystem scale experiments. This parameterization also included pulsing following soil wetting by rain or irrigation, represented as a function dependent on dry spell length. The parameterization by Hudman et al. (2012) was successfully evaluated for pulsing events in central Sahel (0–30° W, 12–18° N).

Boundary conditions for our experiment are generated from a global GEOS-Chem simulation at 4° x 5° horizontal resolution. The regional GEOS-Chem model for West Africa runs at a horizontal resolution of 0.25° x 0.3125° (latitudes 6°S–16°N, longitudes 18.125°W–26.875°E) and a vertical resolution of 47 levels (up to 0.01hPa). Meteorology is driven by the NASA GMAO (Global Modeling and Assimilation Office) GEOS-FP (Forward Processing) assimilated meteorological data. The global model is spun-up from 1<sup>st</sup> May 2015 to 1<sup>st</sup> May 2016. The global simulation is then run from 1<sup>st</sup> May 2016 to 1<sup>st</sup> August 2016, outputting boundary condition files for West Africa. The regional West Africa simulation is then run from 1<sup>st</sup> May 2016 to 1<sup>st</sup> August 2016 using the 4° x 5° boundary conditions from the global simulation. All simulations use the GEOS-FP meteorology which has a three-hour time resolution. We used the same MODIS/Koppen land cover map as in Hudman et al. (2012; http://glcf.umd.edu/data/lc) which includes 24 land cover types. In this simulation we use EDGAR v4.2 (EC-JRC/PBL, 2011) for anthropogenic emissions, GFED4 (Giglio et al., 2013) for biomass burning emissions and MEGAN v2.1 (Guenther et al., 2012) for biogenic emissions of volatile organic compounds. The same emission inventories are used for both the boundary conditions and the West Africa simulation.

#### 3 Results and discussion

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#### 3.1 Meteorological data

Mean air temperature averaged over the whole campaign was  $25.4 \pm 2.6$  °C, mean wind speed was  $1.3 \pm 0.6$  m s<sup>-1</sup>, mean relative air humidity is  $86.3 \pm 10.5$  %, mean soil temperature was  $25.2 \pm 3.4$  °C, mean KIT soil moisture at 5 cm was  $7.1 \pm 3.6$  %, while mean UPS soil moisture averaged between 0 and 30 cm was  $4.5 \pm 2.8$  %. Total KIT precipitation was 198 mm for the whole campaign, and total UPS precipitation was 215 mm.

Median diurnal cycles of air temperature, specific humidity and precipitation are reported in Kalthoff et al., (2017). Knippertz et al. (2017) distinguish four different phases of the monsoon season during the DACCIWA campaign (14<sup>th</sup> June to 30<sup>th</sup> July 2016) over the DACCIWA focus region (5-10° N, 8° W - 8° E), which covers a wide area of West Africa (see Fig. 1, Knippertz et al. (2017)). The division into phases is mainly based on the north–south precipitation difference between the coastal zone (0–7.5° N) and the Sudanian–Sahelian zone (7.5–15° N), both averaged across the longitude range 8° W–8° E. Savè (8.03° N) is located very close to the border between the two zones, with a rainfall pattern that seems to follow more closely that of the coastal zone rather than that of the northern inland Sudanian–Sahelian zone. These four phases are: the

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pre-onset phase characterized by a rainfall maximum near the coast (before 21<sup>st</sup> June, phase 1); the post-onset phase during which the rainfall maximum occurred inland (22<sup>nd</sup> June - 20<sup>th</sup> July, phase 2); the wet westerly regime when the rainfall maximum shifted back to the coast (21<sup>st</sup> - 26<sup>th</sup> July, phase 3); and the recovery of the monsoon with a shift of the rainfall maximum inland (27<sup>th</sup> July until the end of the campaign, phase 4). A specific period within phase 2 is indicated "vortex", during which an unusual development occurred (09<sup>th</sup> - 16<sup>th</sup> July): in the north, a cyclonic feature slowly propagated from eastern Mali to Cape Verde and in the south, an anticyclonic vortex tracked in the west-northwesterly direction along the 330 Guinean coast (see Knippertz et al. 2017 for a more detailed description). At the Savè site the most intense rainfall events happened the day before the first soil fluxes observation, on 15<sup>th</sup> June 2016, and towards the end of the measurement campaign between 20<sup>th</sup> and 23<sup>rd</sup> July 2016. Other minor rainfall events are recorded on 19<sup>th</sup> and 27<sup>th</sup> June, 8<sup>th</sup>, 12<sup>th</sup>, 13<sup>th</sup>, 24<sup>th</sup> and 26<sup>th</sup> July. Daily rainfall measurements are reported in Figs. 2 to 5.

### 3.2 Soil texture, soil organic carbon, total nitrogen, pH and ammonium content

- Bare soil recorded the lower amount of total sand  $(83.75 \pm 1.82 \%)$  and the higher amount of clay  $(5.13 \pm 0.63 \%)$ , fine  $(5.13 \pm 0.96\%)$  and coarse silt  $(5.98 \pm 0.51\%)$ . Grassland recorded the higher amount of total sand  $(89.20 \pm 0.71 \%)$  and the lower amount of clay  $(3.15 \pm 0.50\%)$  and fine silt  $(2.93 \pm 0.32\%)$ , while intermediate values were found for the maize field and forest (Table 3). These values determine the classification of sandy soil for all measurements sites.
- Soil organic carbon (C) and total nitrogen (N) are respectively  $12.2 \pm 5.7$  g kg<sup>-1</sup> and  $0.95 \pm 0.51$  g kg<sup>-1</sup>, averaged for all land 340 cover types over the entire campaign. Table 4 gives soil characteristics for each land cover types, including individual values of C/N ratio, soil organic C and total N for the entire field campaign. The highest average soil organic C was measured for bare soil  $(17.3 \pm 5.9 \text{ g kg}^{-1})$  and the lowest soil organic C was measured for grassland  $(6.2 \pm 1.3 \text{ g kg}^{-1})$ , while the maize field and forest site accounted for  $14.1 \pm 2.9 \text{ g kg}^{-1}$  and  $11.4 \pm 4.5 \text{ g kg}^{-1}$  soil organic C, respectively. The highest average total N was measured for bare soil  $(1.44 \pm 0.51~g~kg^{-1})$  and the lowest total N was measured for grassland  $(0.44 \pm 0.04~g~kg^{-1})$ , while the maize field and forest site accounted similar amounts of total N,  $0.99 \pm 0.19$  g kg<sup>-1</sup> and  $0.94 \pm 0.48$  g kg<sup>-1</sup>, respectively. Values of C/N, soil organic C and total N recorded for grassland at the Savè site compare closely to those reported by Delon et al. (2017, table 2) for the semi-arid site of Dahra (15°24' N 15°25' W), Senegal. Our values of C/N and total N for grassland are also close to those reported by Le Roux et al. (1995, table 1) and Lata et al. (2004) for the wet savanna 350 ecosystem of Lamto (6°13' N, 5°20' W), Ivory Coast, although we observe lower values of soil organic C compared to Le Roux et al. (1995) and Lata et al. (2004). Values of C/N and soil organic C recorded for the maize field at the Savè site are slightly higher than those recorded by Barthes et al. (2004) in a maize field at Agonkanmey (6°24' N, 2°20' E), near Cotonou in southern Benin.
- All the sites listed in the comparison in the previous paragraph are sandy, as the Savè site. The Dahra site (Delon et al., 2017) also shows similar pH than our site (Table 5), while lower pH (acidic or near-neutral) was recorded at the sites of Lamto (Le

Roux et al., 1995, Lata et al., 2004) and Agonkanmey (Barthes et al., 2004). Table 5 provides individual values of pH, [NH<sub>4</sub>+],  $\Gamma$ g and  $\chi_g$  for the entire field campaign. The highest average pH was observed for bare soil (8.23) and the lowest for the forest site (7.07), while measured average pH was 7.27 for grassland and 7.70 for the maize field. The [NH<sub>4</sub>+] content averaged for all land cover types over the entire campaign is 5.33 ± 4 mg kg<sup>-1</sup>. The highest average [NH<sub>4</sub>+] was recorded for the maize field (7.9 ± 6 mg kg<sup>-1</sup>) and the lowest for grassland (2.0 ± 0.3 mg kg<sup>-1</sup>). Average [NH<sub>4</sub>+] is 6.2 ± 5 mg kg<sup>-1</sup> and 7.0 ± 2.2 mg kg<sup>-1</sup> for forest and bare soil, respectively. Dick et al. (2006) have found NH<sub>4</sub>+ concentrations between 2 and 8 mgN.kg<sup>-1</sup> in Senegalese soils, which is very close from our results. Vanlauwe et al. (2002) have found values between 0.8 and 1.4 mgN.kg<sup>-1</sup> in West African moist savanna soils (in Togo and Nigeria).

Higher soil organic C and N over bare soil could be due to the fact that these bare soil patches experienced recent burning (Santín and Doerr, 2016). The higher  $[NH_4^+]$  over the maize field can be caused by chicken excreta, as chickens were roaming over the maize field (Paillat et al., 2005, Tiquia and Tam, 2000).

### 3.3 Soil emission potential $\Gamma_g$ and compensation point $\chi_g$

The mean soil emission potentials for the Savè site is 43 714  $\pm$  58 077, with values ranging from 380 to 159 343. The highest values of soil emission potentials are observed for bare soil (113 672  $\pm$  67 788), followed by maize field (33 880  $\pm$  20 680), forest (11 982  $\pm$  11 061) and grassland (4 929  $\pm$  4 409). The ammonia compensation point ranges between 5 to 2 215 ppb, with soil temperatures between 25 and 29 °C. The highest values of  $\chi_g$  are observed for bare soil (1 607  $\pm$  993), followed by maize field (473  $\pm$  317), forest (175  $\pm$  167) and grassland (58  $\pm$  47). Our values of soil emission potential for bare soil and maize (no fertilization) are comparable with those presented in Massad et al. (2010, table 4), although those data come from measurements taken on different ecosystems. Both  $\Gamma_g$  and  $\chi_g$  values recorded at the Savè site exceed those recorded by Delon et al. (2017) over a grazed semi-arid Sahelian ecosystem in Senegal.

#### 3.4 NO fluxes

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NO fluxes from soil measured during the field campaign range between 0 and  $48.05 \text{ ngN m}^2 \text{ s}^{-1}$ . NO fluxes averaged over all land cover types are  $4.79 \pm 5.59 \text{ ngN m}^{-2} \text{ s}^{-1}$ , while average NO fluxes for each land cover type are:  $8.05 \pm 3.49 \text{ ngN m}^{-2} \text{ s}^{-1}$  for bare soil,  $3.73 \pm 1.76 \text{ ngN m}^{-2} \text{ s}^{-1}$  for the maize field,  $2.87 \pm 1.49 \text{ ngN m}^{-2} \text{ s}^{-1}$  for forest and  $2.82 \pm 3.46 \text{ ngN m}^{-2} \text{ s}^{-1}$  for grassland. Soil emissions of NO from the different land cover types provide similar values, NO emissions from bare soil are higher on average, but have a larger standard deviation.

Other measurements of biogenic NO soil emissions from the West African wet savanna can be found in Delon et al. (2012, table 7). We find that our measured NO soil emissions averaged over all land cover types are higher than those measured from other wet savanna sites. Our measurements are in better agreement with emissions from dry savanna grasslands (Delon et al., 2012), and with measurements from a semi-arid savanna, with over 80% sandy soil, in South Africa (Parsons et al.,

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1996, Scholes et al., 1997). However, these studies measured NO emissions during different seasons and soil moisture conditions compared to our study. For example, Parsons et al. (1996) recorded NO emissions up to 20 ngN m<sup>-2</sup> s<sup>-1</sup> over an open savanna during the period going from the end of the dry season to the beginning of the wet season. Nitric oxide emissions of the same magnitude as in our study were also recorded over a grazed semi-arid Sahelian ecosystem in Senegal during the month of July by Delon et al. (2017): 5.7 ± 3.1 ngN m<sup>-2</sup> s<sup>-1</sup> in July 2012 and 5.1 ± 2.1 ngN m<sup>-2</sup> s<sup>-1</sup> in July 2013.

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Daily means of NO concentrations are measured close to the soil (0.1m, half height of the chamber) and reported in fig. 2 to 5. Daily means of NO concentration vary from 1.28 to 5.40 ppb for all sites. The average concentration during the whole campaign on all sites is  $2.70 \pm 1.03$  ppb. Average NO concentration is  $2.97 \pm 1.49$  ppb on bare soil,  $2.57 \pm 0.96$  ppb on grassland,  $2.55 \pm 0.83$  on maize, and  $2.76 \pm 0.65$  ppb on forest soil. The concentrations are quasi equivalent for all sites. As these concentrations are low, they do not lead to NO deposition on soil and the NO flux stays positive. In fact, NO deposition has been measured in other studies only in the case of high NO concentrations (>60ppb, Laville et al., 2011).

Figures 2 to 5 show daily averaged NO and NH<sub>3</sub> fluxes (±1 standard deviation) for each land cover type, along with precipitation and soil moisture. The spatial variability of NO fluxes is high, especially for bare soil, forest and the maize field where underground roots, not visible at the surface, are heterogeneously distributed. These roots are likely to influence the ammonium content of the soil, and the subsequent NO flux measurement. Standard deviation is generally smaller for grassland (except for two days, July 9th and 13th), where the vegetation (and the root distribution) is more homogeneous. The variation of soil moisture is consistent with the presence of rain events, showing a sharper increase of soil moisture at 5 cm, especially after rainfall following dry periods.

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NO emissions from bare soil and grassland show an increase, sharper for grassland, one to two days after the rain event on 8<sup>th</sup> July. The longer rain event between 20<sup>th</sup> and 24<sup>th</sup> July does not seem to produce an increase in NO emissions (data available only for maize field and forest). This might be linked with the non-linear relationship between NO biogenic soil emissions and soil water content (Oswald et al., 2013). In fact, a light precipitation event (5-15 mm) occurring on dry soils can result in a large flux of NO (Meixner & Yang, 2006, Hartley & Sclesinger, 2000). However, when soil moisture stays at an equivalent level, after several rain events, pulse emissions do not occur (Millet et al., 2004).

A multiple linear regression analysis was performed between daily mean NO fluxes and the following variables: wind speed, soil temperature at 5 cm, soil moisture at 5 cm, soil heat flux, upward longwave radiation and downward shortwave radiation. This regression gives R<sup>2</sup>=0.49 (p-value=0.004), indicating a weak but existing relationship between those variables and NO soil emissions. However, the aim of our experiment was not to reproduce the known relationships between soil fluxes and meteorological variables, like soil temperature and soil moisture, but to give an estimate of soil fluxes at the larger ecosystem scale. For this reason the location of the soil flux measurements was not kept constant even for the same land

cover type on the same measurement day. Moreover, soil temperature and soil moisture were not measured on the same soil parcel where the soil fluxes where measured. Therefore, our experiment does not show the details of microbial and physical processes driving soil fluxes at a single point, but aims to estimate the spatial variability of fluxes at the ecosystem scale.

The NO flux estimated in this study does not consider the impact of vegetation on the net ecosystem flux, as we focus on soil fluxes only. However, the net emission to the atmosphere should take into account the oxidation of NO to NO2 and the eventual re-deposition of NO2 on the vegetation, i.e. what is called Canopy Reduction Factor and is assumed to be a linear function of the Leaf Area Index (e.g. Yienger and Levy, 1995, and Ganzeveld et al., 2002)).

### 3.5 NH<sub>3</sub> fluxes

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NH<sub>3</sub> fluxes measured during the field campaign range between -6.59 and 4.96 ngN m<sup>-2</sup> s<sup>-1</sup>. Ammonia fluxes averaged over all land cover types are -0.91  $\pm$  1.27 ngN m<sup>-2</sup> s<sup>-1</sup>, showing a predominance of NH<sub>3</sub> deposition over emission, which is verified for every land cover type, with an average value of: -1.33  $\pm$  0.86 ngN m<sup>-2</sup> s<sup>-1</sup> for bare soil, -0.75  $\pm$  0.31 ngN m<sup>-2</sup> s<sup>-1</sup> for the maize field, -0.48  $\pm$  0.55 ngN m<sup>-2</sup> s<sup>-1</sup> for grassland, and -0.30  $\pm$  0.38 ngN m<sup>-2</sup> s<sup>-1</sup> for forest. Low positive ammonia fluxes, indicating average NH<sub>3</sub> emission, are only recorded during three days, between 6<sup>th</sup> and 8<sup>th</sup> July, after the longest dry period of the measurement campaign (Figures 2 to 5).

To our knowledge, NH<sub>3</sub> soil fluxes from west African wet savanna are not available in the scientific literature. In Delon et al. (2017) NH<sub>3</sub> soil fluxes measured in Dahra (15°24' N 15°25' W), Senegal, on a dry savanna ecosystem, show low fluxes with a predominance of NH<sub>3</sub> emission: 1.3 ± 1.1 ngN m<sup>-2</sup> s<sup>-1</sup>, -0.1 ± 1.1 ngN m<sup>-2</sup> s<sup>-1</sup> and 0.7 ± 0.5 ngN m<sup>-2</sup> s<sup>-1</sup> over three different measurements campaigns. However, Sutton et al. (2007) shows how pre-cut grassland is characterized by NH<sub>3</sub> deposition, as in our study, in contrast to post-cut grassland, which is marked by NH<sub>3</sub> emission. It is interesting to notice that the literature provides up to about 700 ngN m<sup>-2</sup> s<sup>-1</sup> NH<sub>3</sub> emission for fertilized Zea Mays fields (Walker et al., 2013) while in our study site NH<sub>3</sub> deposition was recorded for the maize field, which is not treated with mineral fertilizer.

As for NO concentrations, NH<sub>3</sub> concentrations are reported in fig. 2 to 5. Daily means of NH<sub>3</sub> concentration vary from 0 to 12.46 ppb for all sites, and the average concentration is  $4.42 \pm 3.23$  ppb during the whole campaign. Average NH<sub>3</sub> concentration is  $6.28 \pm 3.90$  ppb for bare soils,  $3.28 \pm 1.79$  ppb for grassland,  $4.36 \pm 3.99$  for the maize field, and  $3.68 \pm 2.13$  440 ppb for forest. The largest deposition fluxes are found on bare soils, where the largest concentrations are measured.

A multiple linear regression analysis was performed between daily mean NH<sub>3</sub> fluxes and the following variables: wind speed, soil temperature at 5 cm, soil moisture at 5 cm, soil heat flux, outgoing longwave radiation and incoming shortwave radiation). This regression gives a weak but existing relationship, with  $R^2$ =0.37 (p-value=0.03). This correlation highlights

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the link between NH<sub>3</sub> fluxes and relevant environmental parameters. However, the same considerations explained in Sect. 3.4 for NO emissions are also valid for the correlation between NH<sub>3</sub> fluxes and meteorological variables.

According to the current parameterization of soil ammonia emission potential (Sect. 2.7), high values of pH and  $[NH_4^+]$  in the liquid phase will determine high values of  $\Gamma_g$  indicating that the soil has a high propensity to emit NH<sub>3</sub>. However, despite the high values of  $\Gamma_g$  recorded, our measurement site remains a net sink for NH<sub>3</sub>. The reasons for this can be manifold. One explanation could be that soil particles on our site may have a high adsorption capacity limiting the amount of soil gaseous NH<sub>3</sub> concentrations (Neftel et al., 1998) and the largest part of the estimated ammonium content in the soil may not be in the liquid phase, but adsorbed by solid soil particles. In these conditions ammonium will not be available for gas exchange to open porosity and the atmosphere (Flechard et al., 2013). Another explanation could be given by the presence of a water film at the soil surface (linked to high air humidity at the site), which will increase the net deposition process. David et al. (2009) conclude from their measurements that the bare soil can be a significant source of NH<sub>3</sub> only for a limited period and only when the cut vegetation is removed, but not if the soil surface remains covered by grass. Measurements in Ferrara et al. (2014) show other occurrences of high soil ammonia emission potential and NH<sub>3</sub> deposition.

Our measurements were conducted without vegetation inside the chambers, but vegetation was present in the fields. It is important to mention that the role of vegetation on NH<sub>3</sub> bidirectional fluxes is essential, especially during the wet season (time of the experiment), when deposition on the vegetation through stomata and cuticles dominate the exchange (during rain events, the cuticular resistance becomes small and cuticular deposition dominates), due to an increase of the deposition velocity of NH<sub>3</sub> (consecutive to the humidity response of the surface) and a decrease of the canopy compensation point, sensitive to the surface temperature and the surface wetness (Wichink-Kruit et al., 2007).

### 3.6 Comparison of observed and modelled NO soil emissions

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We have compared observed daily averaged (8 a.m. to 6 p.m.) soil NO emissions with those modelled by GEOS-Chem for the entire period of the campaign over the model grid box including the measurement site. The model grid box is positioned at latitude  $8.0^{\circ}$  N  $- 8.25^{\circ}$  N and longitude  $2.19^{\circ}$  E  $- 2.5^{\circ}$  E. The area of this grid box is 958 km<sup>2</sup>. The land cover type within this grid box is classified as "Savannah (Warm)" but the surrounding area also consists of "Woody Savannah", while the observations were taken over the four land cover types representative of the region: bare soil, grassland, maize field and forest.

The model is able to reproduce mean air temperature (25.3 ± 0.7 °C) and the main rain events. Soil emissions of NO are well simulated in magnitude. Simulated NO emissions are often higher than those recorded over the grassland areas, however, simulated NO emissions are often within the error bars of measurements (Fig. 6). It appears that when the model is able to

reproduce the length and the intensity of the rain events, NO emissions are especially well simulated, e.g. the model is able to reproduce the longest rain period (from 20<sup>th</sup> to 26<sup>th</sup> July 2016) and the decrease of emissions at the end of the measurement campaign.

### 90 3.7 Estimate of total NO soil emissions and NH<sub>3</sub> deposition for Benin

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In order to give a tentative estimate of NO and NH3 soil fluxes for Benin we have used the land use/land cover map of Benin provided by the US Geographical Survey Atlas: Landscapes of West Africa - A Window on a Changing World (CILSS, 2016, Fig. 7a). The method of mapping land use/land cover used in this atlas was based on Landsat imagery and expert visual interpretation. In particular, these maps provide an accurate indication of cropland distribution using visual interpretation. According to CILSS (2016) Benin's present-day (2013) land cover is mainly savanna, almost 60%, followed by agricultural land, 31%, while forest is only a small fraction under 1% (the rest of the surface is mainly gallery forest and, on a smaller extent, settlements). In the Atlas (CILSS, 2016) bare soils are defined as those surfaces that are bare even in the green/rainy season. For Benin, the amount of bare soil estimated by CILSS (2016) is very small, not big enough to appear on the Atlas' maps. We have multiplied average NO emissions measured at the Savè site for each land cover type by an estimate of the land cover area of each class given by the Atlas. We have made some approximations: as the land use/land cover maps do not distinguish between shrub savanna, tree savanna, and wooded savanna, we have considered NO soil emissions from Savè's grassland savanna to be representative of the general savanna category in CILSS (2016). Moreover, the Atlas has a crop category that does not distinguish the type of crop and we only have observations of NO soil emissions from an intercropped maize field. We have taken NO soil emissions from the maize field as representative of NO emissions of Benin's agricultural land, but other cultures are present in other parts of the country, e.g. oil palm plantations, where possibly stronger fertilization could determine higher NO soil emissions. This tentative calculation gives that Benin's NO soil emissions for the month of July (wet season) is 1.17 ± 0.6 GgN/month, i.e. 0.09% of the average global monthly NO soil emissions as given by Davidson et al., (1997).

We have also calculated Benin's total monthly NO soil emissions with GEOS-Chem adding together the NO soil emissions from the grid boxes where 50% or greater of the box lies within Benin. Benin's total monthly NO soil emissions calculated with GEOS-Chem for the month of July are 1.44 GgN/month and agree with the tentative calculation given above (1.54 ± 0.8 GgN/month). However, the land cover types covering Benin in GEOS-Chem differ from those in the US Geographical Survey Atlas (CILSS, 2016). In GEOS-Chem Benin is cover by 60.9% savanna, 31.4% woody savannah, 4.5% grassland, 1.3% mixed forest and 0.6% urban and built-up lands. Benin's total monthly NO soil emissions calculated with GEOS-Chem for the months of May and June are higher, 3.51 GgN/month and 2.59 GgN/month, respectively, given those months are at the beginning of the wet season and are characterized by more predominant pulse emissions. Using the same method described above we have upscaled point measurements of NH<sub>3</sub> fluxes with relevant land cover surfaces from CILSS (2016) and obtained that total NH<sub>3</sub> dry deposition for the month of July is 0.21 ± 0.11 GgN/month (0.22 kgN ha<sup>-1</sup> yr<sup>-1</sup>). This value is

about ten times smaller than the estimation of NH<sub>3</sub> dry deposition given in Adon et al. (2013) for the wet savanna site of Djougou (Benin, 9.7°N, 1.7°E) for the month of July, which is around 2.5 kgN ha<sup>-1</sup> yr<sup>-1</sup>.

### 4. Conclusion

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We provide soil flux measurements along with soil characteristics for a land cover type, savanna, that is considered to have large NO emissions (Davidson and Kingerlee, 1997), and for an area of the world, West Africa, with little observations. The aim of this study is to contribute to our knowledge in biogenic soil nitrogen exchanges, provide data for inventories and model evaluation to improve air quality and climate modelling.

In situ measurements were made in a wet savanna site in central Benin from mid-June to the end of July 2016. Complementary to these exchange fluxes, soil N and C content, as well as soil pH, soil moisture, soil temperature and meteorological data were measured. Soil fluxes of NO and NH<sub>3</sub> were measured over four different land cover types in order to give a tentative estimate of regional soil fluxes.

Given the set up of the experiment, the known relationships between soil fluxes, soil temperature and soil moisture were not reproduced. Rather than looking at the microbial and physical processes behind soil fluxes, we are able to provide observations that are representative of a bigger surface area and that represent the spatial variability of fluxes. However, we observe that while shorter rain events determine an increase in NO soil emissions, the longer rain event at the end of the campaign (20<sup>th</sup> to 24<sup>th</sup> July 2016) is accompanied by a decrease in NO soil emissions, in agreement with the fact that the relationship between NO soil emissions and soil moisture is not univocal. Soil emissions of NO increase until an optimum value of soil moisture is reached and then decrease (Oswald et al., 2013).

NH<sub>3</sub> emissions measured in this study probably underestimate total NH<sub>3</sub> emissions for the entire country, as possibly higher localized NH<sub>3</sub> emissions are present in the south of the country where industrial scale agriculture would probably deploy mineral N fertilization.

Soil NO emissions simulated by GEOS-Chem are in good agreement with the local observations taken at the site of Savè, providing a good baseline for simulating local atmospheric chemistry. Moreover, GEOS-Chem is also in good agreement with the tentative total monthly NO soil emission estimate for Benin for the month of July made with local observation in Savè and US Geographical Survey Atlas (CILSS, 2016). All these elements contribute to improve our confidence in the results of modelling studies of local and regional air quality and climate over this region.

Agriculture is the first form of economic activity in Benin, occupying a majority of the active population. The most obvious recent change in land cover is the major expansion of agricultural land across most regions of Benin. Agricultural areas

(including plantations and irrigated agriculture) progressed from 9.2 to 27.1 % of the total country area between 1975 and 2013, improving food security. Oil palm trees are the main crop, and oil palms farmland already covered most of the southern Terre de Barre plateau of Benin by 1975, and increased by about 28 percent over the following 38-year period. A century or more ago, Benin was covered by dense, biologically diverse forest. Since then, Benin has lost nearly all of that forest cover, by 2013, 58 % of the 1975 forest cover had been lost, leaving only 0.2 percent of the country covered with dense forest. Savanna area has also decreased by 23 percent since 1975, but it still remains the dominant land cover type in Benin and covers more than half of the country (CILSS, 2016).

More measurements of NO and NH<sub>3</sub> exchanges between soil-vegetation-atmosphere in areas of Benin (or West Africa) interested by land-use change could improve our estimate of the impact of biogenic soil emissions on air quality and climate, as biogenic soil fluxes influence for example the amount of aerosol and tropospheric O<sub>3</sub>, a greenhouse gas and pollutant, in the atmosphere. Management practices of agriculture affect biogenic soil emissions. Moreover, loosing savanna to oil palm plantations or other crop would have different impacts on air quality, carbon budget and climate than the conversion of forest into crop or oil palm plantation. Furthermore, oil palm plantations are generally closer to the coast and likely to be more influenced by anthropogenic emissions from industry and coastal cities (Knippertz et al., 2015a, 2015b). Oil palm trees are also a strong isoprene emitters. Isoprene emissions influence ozone concentration and the oxidizing capacity of the atmosphere, and it is a source of secondary organic aerosol, thus affecting local air quality and global climate. Large-scale land use change in the tropics – specifically the conversion of tropical rain forest to oil palm plantations in Malaysia – were shown to cause changes in atmospheric composition and chemistry (Hewitt et al., 2009), indicating that the management of the emissions of reactive nitrogen species is essential to prevent damaging levels of ground-level ozone in those regions.

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#### References

- Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G., Jenkin, M. E., Rossi, M. J., and Troe, J.: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume I gas phase reactions of Ox, HOx, NOx and SOx species, Atmos. Chem. Phys., 4, 1461–1738, 2004, http://www.atmos-chem-phys.net/4/1461/2004/.
  - Bai, E., Li, S., Xu, W., Li, W., Dai, W., Jiang, P.: A meta-analysis of experimental warming effects on terrestrial nitrogen pools and dynamics, New Phytol. 199, 441-451. http://dx.doi.org/10.1111/nph.12252, 2013.
- 585 Bai Junhong, Haifeng Gao, Wei Deng, Zhifeng Yang, Baoshan Cui, Rong Xiao, Nitrification potential of marsh soils from two natural saline–alkaline wetlands, Biol Fertil Soils, 46:525–529, 2010.
  - Barthès, B., Azontonde, A., Blanchart, E., Girardin, C., Villenave, C., Lesaint, S., Oliver, R. and Feller, C.: Effect of a legume cover crop (*Mucuna pruriens* var. *utilis*) on soil carbon in an Ultisol under maize cultivation in southern Benin. Soil Use and Manage., 20, 231-239, 2004.
- 590 Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q., Liu, H. Y., Mickley, L. J. and Schultz, M. G.: Global modeling of tropospheric chemistry with assimilated meteorology: Model description and evaluation, J. Geophys. Res., 106 (D19), 23,073 23,095, doi:10.1029/2001JD000807, 2001.
  - Bouwman, A.F., Boumans, L.J.M., Batjes, N.H.: Emissions of N<sub>2</sub>O and NO from fertilized fields: summary of available measurement data. Glob. Biogeochem. Cycles 16 (4), 1058. http://dx.doi.org/10.1029/2001GB001811, 2002.
- 595 Brito, J., Freney, E., Dominutti, P., Borbon, A., Haslett, S. L., Colomb, A., Dupuy, R., Denjean, C., Burnet, F., Bourriane, T., Deroubaix, A., Sellegri, K., Coe, H., Flamant, C., Knippertz, P. and Schwarzenboeck, A., Assessing the role of anthropogenic and biogenic sources on PM<sub>1</sub> over Southern West Africa using aircraft measurements, Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2017-717, 2017.
- Brooks, B., Bessardon, G., Smith, V., Groves, J., Sharpe, S., Kalthoff, N., Adler, B., Handwerker, J., Kohler, M., Kunka, N.,
  Tan, N., Wieser, A., Lohou, F., Bezombes, Y., Bret, G., Delon, C., Derrien, S., Dione, C., Durand, P., Etienne, P., Gabella, O.,
  Jambert, C., Leclercq, J., Lothon, M., Medina, P., Pedruzo, X., Reinares, I., Jegede, G., Ayoola, M., Sunmonu, L., Ajao, A.,
  Abiye, O., Ajileye, O., Aryee, J., Fosu-Amankwah, K., Cayle-Aethelhard, F., Atiah, W. A., Amekudzi, L., Danuor, S.: A new high-quality dataset of the southern West African atmospheric boundary layer diurnal cycle during the Monsoon season, Sci. Data, 2017, in preparation.

- 605 Cassity-Duffey Kate, Miguel Cabrera, John Rema, Ammonia Volatilization from Broiler Litter: Effect of Soil Water Content and Humidity, Soil Sci. Soc. Am. J. 79:543–550, 2014.
  - CILSS (2016). Landscapes of West Africa A Window on a Changing World. U.S. Geological Survey, EROS, 47914 252nd St, Garretson, SD 57030, UNITED STATES, 2016.
- David, M., Loubet, B., Cellier, P., Mattsson, M., Schjoerring, J.K., Nemitz, E., Roche, R., Riedo, M., Sutton, M.A.:

  610 Ammonia sources and sinks in an intensively managed grassland canopy. Biogeosciences 6, 1903-1915, 2009.
  - Davidson, E. A. and Kingerlee, W.: A global inventory of nitric oxide emissions from soils, Nutr Cycl Agroecosys, 48: 37–50, 1997.
- Delon, C., Galy-Lacaux, C., Boone, A., Liousse, C., Serça, D., Adon, M., Diop, B., Akpo, A., Lavenu, F., Mougin, E., and Timouk, F.: Atmospheric nitrogen budget in Sahelian dry savannas, Atmos. Chem. Phys., 10, 2691–2708, doi:10.5194/acp-10-2691-2010, 2010.
  - Delon, C., Galy-Lacaux, C., Adon, M., Liousse, C., Serça, D., Diop, B., and Akpo, A.: Nitrogen compounds emission and deposition in West African ecosystems: comparison between wet and dry savanna, Biogeosciences, 9, 385–402, doi:10.5194/bg-9-385-2012, 2012.
- Delon, C., Galy-Lacaux, C., Serça, D., Loubet, B., Camara, N., Gardrat, E., Saneh, I., Fensholt, R., Tagesson, T., Le Dantec,
   V., Sambou, B., Diop, C., Mougin, E.: Soil and vegetation-atmosphere exchange of NO, NH<sub>3</sub>, and N<sub>2</sub>O from field measurements in a semi arid grazed ecosystem in Senegal, Atmos. Environ., 156, 36–51, 2017.
  - Derrien, S., Bezombes, Y., Bret, G., Gabella, O., Jarnot, C., Medina, P., Pique, E., Delon, C., Dione, C., Campistron, B., Durand, P., Jambert, C., Lohou, F., Lothon, M., Pacifico, F., Meyerfeld, Y.: DACCIWA field campaign, Savè super-site, UPS instrumentation; SEDOO OMP. https://doi.org/10.6096/DACCIWA.1618, 2016.
- 625 Dick Jan, Ute Skiba, Robert Munro and Douglas Deans, Effect of N-fixing and non N-fixing trees and crops on NO and N2O emissions from Senegalese soils, Journal of Biogeography (J. Biogeogr.), 33, 416–423, 2006.
  - EC-JRC / PBL: Emission Database for Global Atmospheric Research (EDGAR), release version 4.2. European Commission, Joint Research Centre (JRC)/Netherlands Environmental Assessment Agency (PBL), http://edgar.jrc.ec.europa.eu., 2011.
- Ferrara, R. M., Loubet, B., Decuq, C., Palumbo, A. D., Di Tommasi, P., Magliulo, V., Masson, S., Personne, E., Cellier, P., 630 Rana, G.: Ammonia volatilisation following urea fertilisation in an irrigated sorghum crop in Italy, Agr. Forest Meteorol. 195–196, 179–191, 2014.
  - Flechard, C.R., Massad, R.-S., Loubet, B., Personne, E., Simpson, D., Bash, J.O., Cooter, E.J., Nemitz, E., Sutton, M.A.: Advances in understanding, models and parameterizations of biosphere-atmosphere ammonia exchange, Biogeosciences, 10, 5183-5225, 2013.

- 635 Fuzzi, S., Baltensperger, U., Carslaw, K., Decesari, S., Denier van der Gon, H., Facchini, M. C., Fowler, D., Koren, I., Langford, B., Lohmann, U., Nemitz, E., Pandis, S., Riipinen, I., Rudich, Y., Schaap, M., Slowik, J. G., Spracklen, D. V., Vignati, E., Wild, M., Williams, M., Gilardoni, S.: Particulate matter, air quality and climate: lessons learned and future needs. Atmos. Chem. Phys. 15, 8217-8299. http://dx.doi.org/10.5194/acp-15-8217-2015, 2015.
- Ganzeveld L.N., J. Lelieveld, F. J. Dentener, M. C. Krol, A. J. Bouwman, and G.-J. Roelofs, Global soil-biogenic NOx emissions and the role of canopy processes, JOURNAL OF GEOPHYSICAL RESEARCH, VOL. 107, NO. D16, 10.1029/2001JD001289, 2002.
  - Giglio, L., Randerson, J. T. and van der Werf, G. R.: Analysis of daily, monthly, and annual burned area using the fourth-generation global fire emissions database (GFED4), J. Geophys. Res. Biogeosci., 118, 317–328, doi:10.1002/jgrg.20042, 2013.
- 645 Grote, R., Lehmann, E., Brümmer, C., Brüggemann, N., Szarzynski, J., Kunstmann, H.: Modelling and observation of biosphereeatmosphere interactions in natural savannah in Burkina Faso, West Africa. Phys. Chem. Earth 34, 251-260, 2009.
  - Guenther, A. B., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L. K., and Wang, X.: The Model of Emissions of Gases and Aerosols from Nature version 2.1 (MEGAN2.1): an extended and updated framework for modeling biogenic emissions, Geosci. Model Dev., 5, 1471-1492, doi:10.5194/gmd-5-1471-2012, 2012.
- 650 Jegede, O. O., Ayoola, M. A., Sunmonu, L. A., Ajao A. I., and Babić K.: DACCIWA field campaign, Ile-Ife super-site, Surface measurements; SEDOO OMP.
  - Kalthoff, N., Lohou, F., Brooks, B., Jegede, G., Adler, B., Babić, K., Dione, C., Ajao, A., Amekudzi, L. K., Aryee, J. N. A., Ayoola, M., Bessardon, G., Danuor, S. K., Handwerker, J., Kohler, M., Lothon, M., Pedruzo-Bagazgoitia, X., Smith, V., Sunmonu, L., Wieser, A., Fink, A. H.1, and Knippertz, P.: An overview of the diurnal cycle of the atmospheric boundary
- layer during the West African monsoon season: Results from the 2016 observational campaign, Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2017-631, 2017.
  - Kohler, M., Kalthoff, N., Seringer, J., and Kraut, S.: DACCIWA field campaign, Savè super-site, Surface measurements; SEDOO OMP. https://doi.org/10.6096/DACCIWA.1690, 2016.
- Knippertz, P., Coe, H., Chiu, J. C., Evans, M. J., Fink, A. H., Kalthoff, N., Liousse, C., Mari, C., Allan, R. P., Brooks, B.,
  Danour, S., Flamant, C., Jegede, O. O., Lohou, F., and Marsham, J. H.: The DACCIWA project: Dynamics-aerosol-chemistry-cloud interactions in West Africa, B. Am. Meteorol. Soc., 96, 1451–1460, https://doi.org/10.1175/BAMS-D-14-00108.1, 2015a.
  - Knippertz, P., Evans, M. J., Field, P. R., Fink, A. H., Liousse, C., and Marsham, J. H.: The possible role of local air pollution in climate change in West Africa, Nature Climate Change, 5, 815–822, https://doi.org/10.1038/nclimate2727, 2015b.

- Knippertz, P., Fink, A. H., Affo-Dogo, A., Bahaga, T., Brosse, F., Deroubaix, A., Evans, M., Gaetani, M., Guebsi, R., Latifou, I., Lavaysse, C., Maranan, M., Mari, M., Marsham, J. H., Meynadier, R., Morris, E., Rosenberg, P. D., Schlueter, A., and Tocquer, F.: A meteorological and chemical overview of the DACCIWA field campaign in West Africa in June–July 2016, Atmos. Chem. Phys. Discuss., doi.org/10.5194/acp-2017-345, 2017.
- Handwerker, J., Scheer, S., and Gamer, T.: DACCIWA field campaign, Savè super-site, Cloud and precipitation; SEDOO OMP. https://doi.org/10.6096/DACCIWA.1686, 2016.
  - Hewitt, C. N., MacKenzie, A. R., Di Carlo, P., Di Marco, C. F., Dorsey, J. R., Evans, M., Fowler, D., Gallagher, M. W., Hopkins, J. R., Jones, C. E., Langford, B., Lee, J. D., Lewis, A. C., Lim, S. F., McQuaid, J., Misztal, P., Moller, S. J., Monks, P. S., Nemitz, E., Oram, D. E., Owen, S. M., Phillips, G. J., Pugh, T. A. M., Pyle, J. A., Reeves, C. E., Ryder, J., Siong, J., Skiba, U., and Stewart, D. J.: Nitrogen Management Is Essential to Prevent Tropical Oil Palm Plantations from Causing
- 675 Ground-Level Ozone Pollution, P. Natl. Acad. Sci., 106, 18447–18451, 2009.
  - Hudman, R.C., Russell, A.R., Valin, L.C., Cohen, R.C.: Interannual variability in soil nitric oxide emissions over the United States as viewed from space. Atmos. Chem. Phys. 10, 9943e9952. http://dx.doi.org/10.5194/acp-10-9943-2010, 2010.
  - Hudman, R. C., Moore, N. E., Mebust, A. K., Martin, R. V., Russell, A. R., Valin, L. C., and Cohen, R. C.: Steps towards a mechanistic model of global soil nitric oxide emissions: implementation and space based-constraints, Atmos. Chem. Phys., 12, 7779-7795, doi:10.5194/acp-12-7779-2012, 2012.
  - Kristensen, L., Lenschow, D. H., Gurarie, D., Jensen, N. O.,: A simple model for the reactive species in the convective atmospheric boundary layer, Bound-Lay Meteorol., 134, 195-221, doi:10.1007/s10546-009-9443-x, 2010a.
  - Kristensen, L., Lenschow, D. H., Gurarie, D., Jensen, N. O.: Erratum to: A Simple Model for the Vertical Transport of Reactive Species in the Convective Atmospheric Boundary Layer, Bound-Lay Meteorol., 135, 181–183, DOI 10.1007/s10546-010-9473-4, 2010b.
  - Lata, J.-C., Degrange, V. Raynaud, X., Maron, P.-A., Lensis, R. and Abbadie, L.: Grass populations control nitrification in savanna soils, Funct. Ecol., 18, 605-611, 2004.
  - Le Roux, X., Abbadie, L., Lensi, R., Serça, D.: Emission of nitrogen monoxide from African tropical ecosystems: control of emission by soil characteristics in humid and dry savannas of West Africa. J. Geophys. Res. 100, 23133-23142, 1995.
- 690 Loubet, B., Decuq, C., Personne, E., Massad, R.S., Flechard, C., Fanucci, O., Mascher, N., Gueudet, J.-C., Masson, S., Durand, B., Genermont, S., Fauvel, Y., Cellier, P.:. Investigating the stomatal, cuticular and soil ammonia fluxes over a growing tritical crop under high acidic loads. Biogeosciences 9, 1537-1552, 2012.
  - Massad, R.-S., Nemitz, E., Sutton, M. A.: Review and parameterisation of bi-directional ammonia exchange between vegetation and the atmosphere. Atmos. Chem. Phys. 10, 10359-10386. http://dx.doi.org/10.5194/acp-10-10359-2010, 2010.

- 695 McCalley, C. K. and Sparks, J. P.: Controls over nitric oxide and ammonia emissions from Mojave Desert soils. Oecologia 156, 871-881, 2008.
  - Meixner, F.X., Yang, W.X., In: DO doroco, P., Porporato, A. (Eds.), Biogenic Emissions of Nitric Oxide and Ditrous Oxide from Arid and Semi-arid Land, Dryland Ecohydrology. Kluwer Academic Publishers B.V, Dordrecht, The Netherlands, pp. 23-46, 2006.
- Michiels, B., Babatounde, S., Dahouda, M., Chabi, S. L. W. and Buldgen, A.: Botanical composition and nutritive value of forage consumed by sheep during the rainy season in a Sudano-guinean savanna (central Benin), Trop Grasslands, 34, 43-47, 2000.
  - Neftel, A., Blatter, A., Gut, A., Högger, D., Meixnera, F., Ammann, C., Nathaus, F. J.: NH<sub>3</sub> soil and soil surface gas measurements in a triticale wheat field, Atmos. Environ., 32, 3, 499-505, 1998.
- Oswald, R., Behrendt, T., Ermel, M., Wu, D., Su, H., Cheng, Y., et al.: HONO Emissions from Soil Bacteria as a Major Source of Atmospheric Reactive Nitrogen. Science, 341(6151), 1233-1235, 2013.
  - Paillat, J, -M, , Robin, P., Hassouna, M., Leterme, P.: Predicting ammonia and carbon dioxide emissions from carbon and nitrogen biodegradability during animal waste composting, Atmos. Environ. 39, 6833–6842, 2005.
- Pape, L., Ammann, C., Nyfeler-Brunner, A., Spirig, C., Hens, K., Meixner, F.X.,: An automated dynamic chamber system for surface exchange measurement of non-reactive and reactive trace gases of grassland ecosystems. Biogeosciences, 6, 405e429. http://dx.doi.org/10.5194/bg-6-405-2009, 2009.
  - Parsons, D. A. B., Scholes, M. C. and Levine, J. S.: Biogenic NO emissions from savanna soils as a function of fire regime, soil type, soil nitrogen and water status, J Geophys Res 101: 23683–23688, 1996.
  - Pilegaard, K.: Processes regulating nitric oxide emissions from soils, Phil. Trans. R. Soc. B 2013 (368), 20130126, 2013.
- 715 Säidou, A., Kuyper, T. W., Kossou, D. K., Tossou, R., and Richards, P.: Sustainable soil fertility management in Benin: learning from farmers, NJAS-WAGEN J LIFE SC, 52-3/4, 349-369, 2004.
  - Santín C., and Doerr, S. H.: Fire effects on soils: the human dimension. Phil. Trans. R. Soc. B 371: 20150171. http://dx.doi.org/10.1098/rstb.2015.0171, 2016.
- Scholes, M. C., Scholes, R. J., Parsons, D., Martin, R., and Winstead, E.: NO and N<sub>2</sub>O emissions from savanna soils following the first rains. Nut. Cycling Agroecosys. 48: 115–122, 1997.
  - Serça, D., Delmas, R., Le Roux, X., Parsons, D. A. B., Scholes, M. C., Abbadie, L., Lensi, R., Ronce, O., Labroue, L.: Comparison of nitrogen monoxide emissions from several African tropical ecosystems and influence of season and fire. Glob. Biogeochem. Cy. 12 (4), 637-651, 1998.

- Steinkamp, J., Ganzeveld, L.N., Wilcke, W., Lawrence, M.G.: Influence of modelled soil biogenic NO emissions on related trace gases and the atmospheric oxidizing efficiency. Atmos. Chem. Phys. 9, 2663e2677. http://dx.doi.org/10.5194/acp-9-2663-2009, 2009.
  - Sutton, M. A., Nemitz, E., Erisman, J. W., Beier, C., Butterbach Bahl, K., Cellier, C., de Vries, W., Cotrufo, F., Skiba, U., Di Marco, C., Jones, S., Laville, P., Soussana, J. F., Loubet, B., Twigg, M., Famulari, D., Whitehead, J., Gallagher, M. W., Neftel, A., Flechard, C., Herrmann, B., Calanca, P.L., Schjoerring, J. K., Daemmgen, U., Horvath, L., Tang, Y. S., Emmett, B.
- A., Tietema, A., Pe' nuelas, J., Kesik, M., Brueggemann, N., Pilegaard, K., Vesala, T., Campbell, C. L., Olesen, J. E., Dragosits, U., Theobald, M. R., Levy, P., Mobbs, D. C., Milne, R., Viovy, N., Vuichard, N., Smith, J. U., Smith, P. E., Bergamaschi, P., Fowler, D., and Reis, S.: Challenges in quantifying biosphere-atmosphere exchange of nitrogen species, Environ. Pollut., 150, 125–139, 2007.
- Sutton, M., Reis, S., and Baker, S. M. H.: Atmospheric Ammonia, detecting emission changes and environmental impacts.

  Results of an expert workshop under the convention of long range transboundary air pollution, Springer Edition, New York, ISBN 978-1-4020-9120-9, 2009a.
  - Sutton, M. A., Nemitz, E., Milford, C., Campbell, C., Erisman, J. W., Hensen, A., Cellier, P., David, M., Loubet, B., Personne, E., Schjoerring, J. K., Mattsson, M., Dorsey, J. R., Gallagher, M. W., Horvath, L., Weidinger, T., Meszaros, R., Dammgen, U., Neftel, A., Herrmann, B., Lehman, B. E., Flechard, C., Burkhardt: Dynamics of ammonia exchange with cut grassland: synthesis of results and conclusions of the GRAMINAE integrated experiment, Biogeosciences 6, 2907-2934, 2009b.

- Tiquia, S. M. and Tam, N. F. Y.: Fate of nitrogen during composting of chicken litter, Environ. Pollut., 110, 535-541, 2000.

  Vaittinen O., M. Metsa la, S. Persijn, M. Vainio, L. Halonen, Adsorption of ammonia on treated stainless steel and polymer surfaces, Appl. Phys. B, DOI 10.1007/s00340-013-5590-3, 2013.
- Vanlauwe B., J. Diels, O. Lyasse, K. Aihou, E.N.O. Iwuafor, N. Sanginga, R. Merckx & J. Deckers, Fertility status of soils of the derived savanna and northern guinea savanna and response to major plant nutrients, as influenced by soil type and land use management, Nutrient Cycling in Agroecosystems 62: 139–150, 2002.
  - Walker, J. T., Jones, M. R., Bash, J. O., Myles, L., Meyers, T., Schwede, D., Herrick, J., Nemitz, E., and W. Robarge: Processes of ammonia air-surface exchange in a fertilized Zea mays canopy, Biogeosciences, 10, 981–998, doi:10.5194/bg-10-981-2013, 2013.
  - Wentworth, G. R., Murphy, J. G., Gregoire, P. K., Cheyne, C. A. L., Tevlin, A. G. and Hems, R.: Soil-atmosphere exchange of ammonia in a non-fertilized grassland: measured emission potentials and inferred fluxes, Biogeosciences, 11, 5675-5686, 2014.

- Wichink Kruit, R. J., van Pul, W. A. J., Otjes, R. P., Hofschreuder, P., Jacobs, A. F. G., and Holtslag, A. A. M, Ammonia 755 fluxes and derived canopy compensation points over non-fertilized agricultural grassland in the Netherlands using the new gradient ammonia high accuracy monitor (GRAHAM), Atmos. Environ., 41, 1275–1287, 2007.
  - Wieser, A., Adler, B., and Deny, B.: DACCIWA field campaign, Savè super-site, Thermodynamic data sets; SEDOO OMP. https://doi.org/10.6096/DACCIWA.1659, 2016.
- Yienger, J. J., and H. Levy II, Global inventory of soil-biogenic NOx emissions, J. Geophys. Res., 100, 11,447–11,464, 760 1995.

	Savè ground-based observation site
Location	8°02'03" N, 2°29'11" E
Elevation	166 m a.s.l.
Mean annual precipitation	1100 mm
Mean annual temperature	27.5 °C
Soil type	sandy
Sand percentage	87%
Clay percentage	4.1%

Table 1

765 Main characteristics of the Savè site.

Soil type		Plant family	Plant species	Common name/s
Next to grassland and forest	Dominant tree species	Anacardiaceae	Anacardium occidentale	cashew tree
		Fabaceae	Daniellia oliveri	African copaiba balsam tree
		•	Pterocarpus erinaceus	barwood, muninga, vène, mukwa
	Dominant ground species	Cleomaceae	Cleome sp.	spider flowers, spider plants
		Fabaceae	Crotalaria sp.	rattlepod or rattlebox
	•	•	Mucuna sp.	velvet bean
	•	Poaceae	Imperata cylindrica	cogon grass, cotton wool grass, kura-kura
		•	Rhynchelytrum repens	rose natal grass
Next to maize field	Dominant tree species	Anacardiaceae	Mangifera indica	mango
		Arecaceae	Cocos nucifera	coconut tree
		Caricaceae	Carica papaya L.	papaya
	•	Lamiaceae	Tectona grandis	teak
	•	Meliaceae	Azadirachta indica	neem, nimtree, Indian lilac
	Dominant ground species	Commelinaceae	Commelina benghalensis	benghal dayflower, tropical spiderwort
		Euphorbiaceae	Euphorbia sp.	spurge
		Nyctaginaceae	Boerhavia diffusa	punarnava, red spiderling
	•	Phyllanthaceae	Phyllanthus amarus	gale of the wind, stonebreaker
	•	Poaceae	Digitaria horizontalis	Jamaican crabgrass
	Crops	Dioscoreaceae	Dioscorea sp.	yam
		Euphobiaceae	Manihot esculenta	cassava
	· 	Fabaceae	Arachis hypogaea	peanut
			Vigna unguiculata	cowpea
		Malvaceae	Gossypium sp.	cooton
		Pedaliaceae	Sesamum indicum	sesame
		Poaceae	Zea mays	maize
			Sorghum sp.	sorghum
		Solanaceae	Solanum lycopersicum	tomato

Table 2

List of plant species at the Savè site. The list of common names is not considered to be exhaustive.

	Bare Soil	Grassland	Maize Field	Forest
Clay (<2µm) (%)	5.13±0.63	3.15±0.50	4.40±0.35	3.70±1.25
Fine Silt (2 to 20μm) (%)	5.13±0.96	2.93±0.32	4.13±1.00	3.40±1.21
Coarse Silt (20 to 50μm) (%)	5.98±0.51	4.78±0.66	4.37±0.38	4.67±1.05
Total Sand (50 to 2000 μm) (%)	83.75±1.82	89.20±0.71	87.13±0.99	88.20±3.50

Table 3
List of soil characteristics for each land cover type at the Savè site, including standard deviation.

Soil Type	Date	C/N ratio	Organic C (g kg <sup>-1</sup> )	Total N (g kg <sup>-1</sup> )
Bare soil	06/07/2016	14.90	24.28	1.63
	09/07/2016	12.20	10.47	0.86
	19/07/2016	12.30	15.05	1.22
	28/07/2016	9.50	19.30	2.04
Grassland	07/07/2016	15.20	5.78	0.38
	09/07/2016	16.00	7.36	0.46
	19/07/2016	14.90	7.16	0.48
	28/07/2016	10.90	4.56	0.42
Maize	09/07/2016	14.80	17.33	1.17
	19/07/2016	16.40	13.08	0.8
	28/07/2016	11.80	11.83	1.00
Forest	06/07/2016	14.80	7.98	0.54
	19/07/2016	11.90	9.62	0.81
	28/07/2016	11.30	16.56	1.47

## 785 Table 4

List of soil characteristics for each land cover type at the Savè site for each soil sampling day: Carbon-to-Nitrogen ratio (C/N), organic carbon  $(g \ kg^{-1})$  and total nitrogen  $(g \ kg^{-1})$ . The accuracy for the C/N ratio is 14%. The measurement accuracy for organic carbon and total nitrogen is 14 and 13%, respectively.

Soil Type	Date	pН	$[NH_4^+] (mg kg^{-1})$	$\Gamma_g[N{H_4}^+]/[H^+]$	$X_g$ (ppb)
Bare soil	06/07/2016	8.38	6.82	136 334	1891
	09/07/2016	7.73	2.90	12 978	134
	19/07/2016	8.46	6.63	159 343	2188
	28/07/2016	8.34	8.01	146 033	2215
	07/07/2016	7.15	1.96	2 307	29
	09/07/2016	7.91	1.55	10 499	108
	19/07/2016	7.52	2.28	6 291	86
	28/07/2016	6.51	2.30	620	9
Maize	09/07/2016	7.46	4.40	10 575	109
	19/07/2016	7.61	14.74	50 040	687
	28/07/2016	8.04	4.49	41 027	622
Forest	06/07/2016	6.32	2.18	380	5
	19/07/2016	7.51	4.88	13 159	181
	28/07/2016	7.37	11.47	22 407	340

Table 5

List of soil pH, ammonium concentrations [NH<sub>4</sub><sup>+</sup>] (mg kg<sup>-1</sup>), soil emission potential  $\Gamma_g$  and soil compensation point  $\chi_g$  (ppb) for each land cover type at the Savè site for each soil sampling day. The measurement accuracy for pH is 0.15 when pH  $\leq$  7 and 0.20 when pH > 7. The accuracy for ammonium concentrations [NH<sub>4</sub><sup>+</sup>], soil emission potential  $\Gamma_g$  and soil compensation point  $\chi_g$  is 25%.

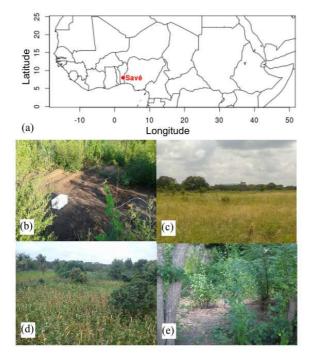


Fig. 1(a) Location of the Savè site in West Africa, (b) one of the bare soil sampling sites, (c) the grassland sampling site, (d) the maize field sampling site and (e) the forest sampling site at the Savè site.

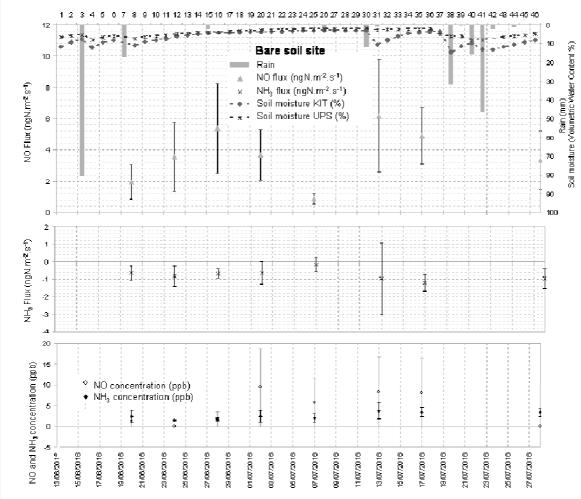


Fig. 2 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH<sub>3</sub> fluxes in ngN m-2 s-1 measured at the bare soil site; Lower panel: daily mean NO and NH<sub>3</sub> concentrations in ppb measured at the bare soil site. Vertical bars show the standard deviation from individual fluxes and concentrations

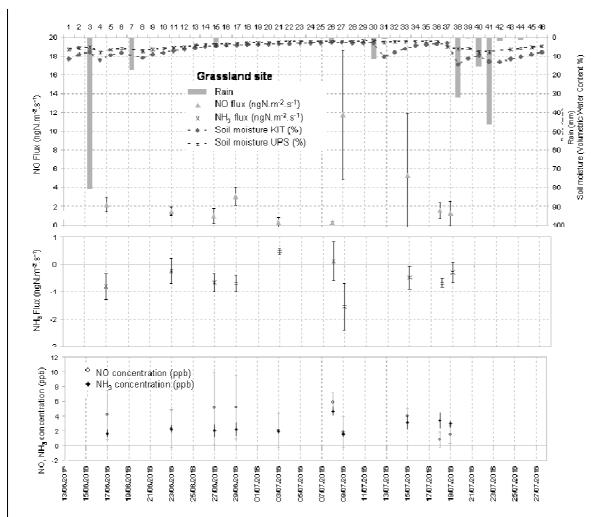


Fig. 3 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH<sub>3</sub> fluxes in ngN m-2 s-1 measured at the grassland site; Lower panel: daily mean NO and NH<sub>3</sub> concentrations in ppb measured at the grassland site. Vertical bars show the standard deviation from individual fluxes and concentrations.

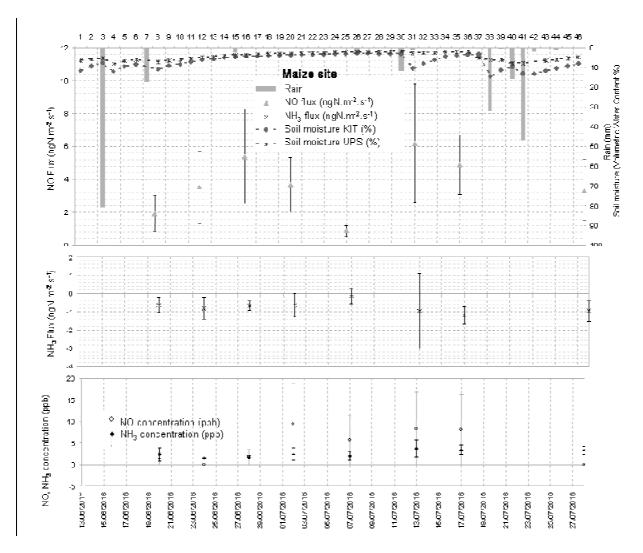


Fig. 4 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH<sub>3</sub> fluxes in ngN m-2 s-1 measured at the maize field site; Lower panel: daily mean NO and NH<sub>3</sub> concentrations in ppb measured at the maize field site. Vertical bars show the standard deviation from individual fluxes and concentrations.

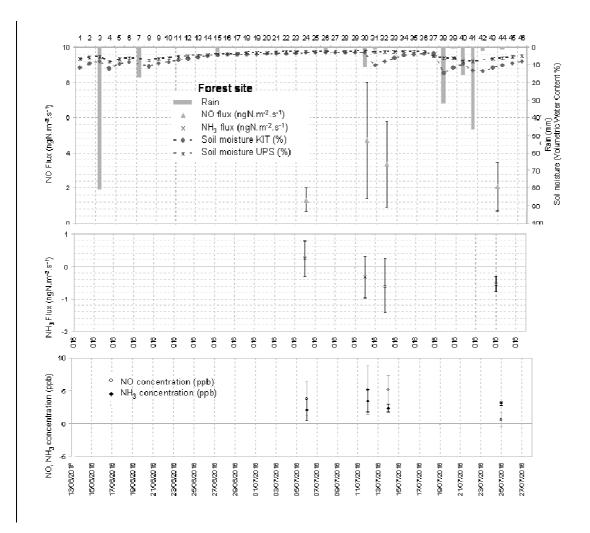


Fig. 5 Upper panel: Daily total precipitation (mm), daily mean soil moisture at 5 cm (%) measured by the Karlsruhe Institute of Technology (KIT), daily mean soil moisture averaged between 0 and 30 cm measured by the Université Paul Sabatier (UPS) instrumentation; Middle panel: daily mean NO and NH<sub>3</sub> fluxes in ngN m-2 s-1 measured at the forest site; Lower panel: daily mean NO and NH<sub>3</sub> concentrations in ppb measured at the forest site. Vertical bars show the standard deviation from individual fluxes and concentrations.

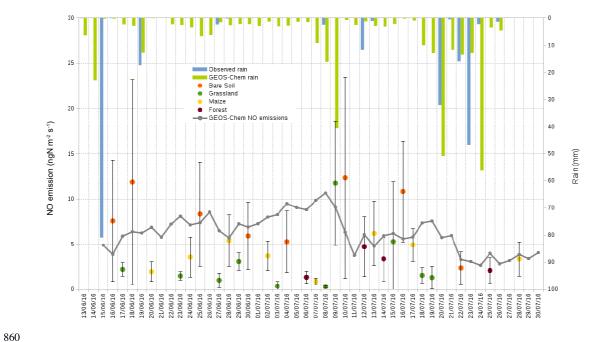


Fig. 6 Nitric oxide emissions measured over each land cover type (orange dot for bare soil, green for grassland, yellow for the maize field and brown for forest) and simulated with GEOS-Chem, along with rainfall measured and modelled with GEOS-Chem. Soil NO emissions are daily average between 8 a.m. and 6 p.m..

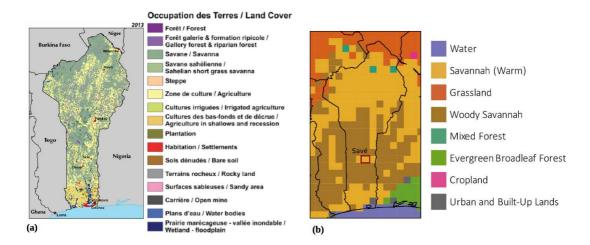


Fig. 7 (a) Land cover map of Benin for 2013 from the US Geographical Survey Atlas: Landscapes of West Africa – A Window on a Changing World (CILSS, 2016) and (b) land cover map of Benin used in the GEOS-Chem simulation.