#### **Response to editor:**

Dear Dr. Yugo Kanaya,

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We thank you for the constructive comment. We have revised the manuscript following the comment as described below. The comment is shown in blue. Our response is shown in black. The revised text is shown in italics.

Point regarding isoprene chemistry mechanism needs further justification: Besides sensitivity on nitrate yield, production of radicals from Isoprene RO2 radicals should be discussed. For example, see CheT2 mechanism (ISO2 + N2/O2), in Table 2 of Squire et al. (2015).

Thanks for the comment. The production of radicals from isoprene peroxy radicals (ISOPO2) might be important for secondary pollutants (e.g. O<sub>3</sub> and secondary organic aerosol), particularly in isoprene-rich regions. Squire et al. (2015) compared the effects of four different reduced isoprene chemical mechanisms on tropospheric O<sub>3</sub>. One of the four mechanisms included the formation of hydroperoxy-aldehydes (HPALDs) from ISOPO2 and subsequent rapid release of OH. In addition, Kanaya et al. (2012) evaluated the potential effect of isomerization of ISOPO2 on HO and HO2, in comparison model results with HOx measurements using multiple instruments. They revised the model by adding detailed reactions of isomerization of ISOPO2 and photolysis of HPALDs following Peeters and Müller (2010), and found that the revision could increase OH and HO2 concentrations by 28-38% for daytime. However, Kanaya et al. (2012) also pointed out that the effects of isomerization of ISOPO2 proposed by Peeters and Müller (2010) might be overestimated.

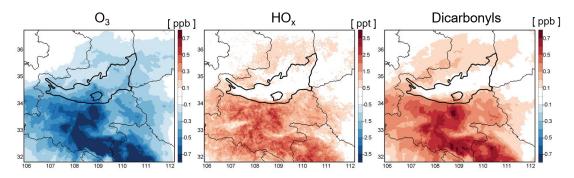
To evaluate the potential impacts of radical from ISOPO2 on tropospheric O<sub>3</sub> in the GZ basin, we modified the standard RADM2 mechanism by adding a reduced ISOPO2 isomerization reaction following a very recent study (Li et al., 2018), as follows:

ISOPO2 > 2\*HO2+HCHO+0.33\*MGLY+0.5\*GLYALD+0.25\*GLYX+1.5\*HACET

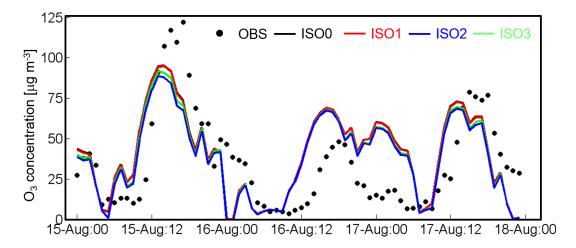
with reaction rate coefficient  $k = 4.07 \times 10^8 \times \exp(-7694/T)$  (see Table R1).

We conducted four sensitivity simulations (15<sup>th</sup>-17<sup>th</sup> Aug.), namely ISO0 (standard RADM2 mechanism), ISO1 (including the above ISOPO2 isomerization reaction), ISO2 (including reactions of isoprene nitrate with the yield of 4%), and ISO3 (including both revisions in ISO1 and ISO2), to explore the effects of the advances in isoprene chemistry. Figure R1 shows spatial distributions of the changes of O<sub>3</sub>, HO<sub>x</sub> and dicarbonyls (MGLY+GLYX+GLYALD) due to ISOPO2 isomerization reaction (ISO1-ISO0). We found that ISOPO2 isomerization slightly decreased O<sub>3</sub> concentration, less than 0.5 ppb for urban Xi'an (see Table R2), as a result of the HOx increasement. Figure R2 and Table R2 compare the simulated O<sub>3</sub> concentration at urban Xi'an from different sensitivity runs. Daily peak O<sub>3</sub> concentration decreased by 2.9 ppb (7%) in ISO3 (mix ISOPN production and ISOPO2 isomerization) compared with the ISO0 run. It can be interesting to analyze the budgets of O<sub>3</sub> and HOx in future with updated isoprene chemistry in a modified chemistry transport model.

We added the discussion above in Section 2.2 and added Tables R1 and R2 and



**Figure R1.** Concentration changes of O<sub>3</sub>, HO<sub>x</sub> (HO+HO<sub>2</sub>) and dicarbonyls due to ISOPO2 isomerization reaction (ISO1-ISO0) averaged for O<sub>3</sub> peak time (14:00-18:00 local time) in the Guanzhong basin.



**Figure R2.** O<sub>3</sub> concentrations changes due to updated isoprene chemistry at Xi'an Jiaotong University for the period of 15<sup>th</sup> to 17<sup>th</sup> August 2011.

**Table R1.** Updated isoprene oxidation chemistry (unit for reaction rate is molecule<sup>-1</sup> cm<sup>3</sup> s<sup>-1</sup>)

Reactions	Reaction rates		
ISOPO2+NO >	2.7×10 <sup>-12</sup> ×exp(350/T)		
ALD+HCHO+HO2+0.96*N			
ISOPO2 >	4.07×10 <sup>8</sup> ×exp(-7694/T)		
2*HO2+HCHO+0.33*MGL			
+1.5*HACET			
Species name			
T	Temperature (K)		

NO	Nitric oxide
NO2	Nitrogen dioxide
HO2	Hydroperoxy radical
ISOPO2	Hydroperoxy radicals from isoprene oxidation by OH
ISOPN	Organic nitrate
НСНО	Formaldehyde
ALD	Acetaldehyde and higher aldehydes
HACET	Hydroxyacetone
MGLY	Methylglyoxal
GLYALD	Glycolaldehyde
GLYX	Glyoxal

**Table R2.** O<sub>3</sub> concentration changes due to updated isoprene chemistry averaged for urban Xi'an

O <sub>3</sub> concentration (ppb)	ISO1-ISO0	ISO2-ISO0	ISO3-ISO0
Daily peak	-0.47	-4.6	-2.9
24h average	-0.27	-2.9	-2.0

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We noted some advances in isoprene chemistry recently. Some studies pointed out that isoprene reacts with OH radical to form hydroperoxy radicals (ISOPO2). Subsequently, in the presence of NOx, ISOPO2 reacts with NO leading to the production of hydroxynitrates (ISOPN) by a minor branch, which sequesters  $NO_x$  and therefore regulates  $O_3$  formation. A number of laboratory, filed observation and simulation studies (e.g. Paulot et al., 2009a, b, 2012; Horowitz et al., 2007; Hudman et al., 2009; Fisher et al., 2016; Travis et al., 2016) highlighted the importance of isoprene nitrate chemistry and all agreed there were still large uncertainties (for example, the estimates of ISOPN yield (4%-15%)). Horowitz et al. (2007) found a 4% ISOPN yield best captured the alkyl and multifunctional nitrates measured by aircraft, and Hudman et al. (2009) pointed out that unreasonably high ISOPN yield (18%) would let ISOPN be a terminal sink for  $NO_x$ . In addition, the production of radicals from ISOPO2 is regarded to be important for HOx and further for  $O_3$  (e.g. Squire et al., 2015). Kanaya et al. (2012) quantified the formation of hydroperoxy-aldehydes (HPALDs) from ISOPO2 and subsequent rapid release of HOx. They revised the model by adding detailed reactions of isomerization of ISOPO2 and photolysis of HPALDs following Peeters and Müller (2010), and found the revision could increase OH and HO2 concentrations by 28-38% for daytime. However, Kanaya et al. (2012) also pointed out that isomerization of ISOPO2 at the rates proposed by Peeters and Müller (2010) might be overestimated.

Back to our study, these advances in isoprene chemistry were not contained in the standard RADM2 mechanism. To assess the potential uncertainties, we modified the RADM2 mechanism by adding formation pathway of ISOPN from ISOPO2+NO (the yield of 4%) following Horowitz et al. (2007) and ISOPO2 isomerization reaction following Li et al. (2018). Details of these updates can be found in Table S1. We conducted four sensitivity simulations (15<sup>th</sup>-17<sup>th</sup> Aug.), namely ISOO (standard RAMD2 mechanism), ISO1 (including the ISOPO2 isomerization reaction), ISO2 (including reactions of isoprene nitrate), and ISO3 (including both revisions in ISO1 and ISO2), to explore the effects on O3 in the GZ basin. The results show that near surface O3 concentration was decreased by 2.9 ppb (7%) averaged for urban Xi'an in August after implementing these updates (ISO3-ISOO), and we thought this would not crucially impact conclusions of this study. Details of the sensitivity analysis results can be found in Table S2 and Fig. S1.

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### Manuscript with changes marked

# Impacts of Biogenic and Anthropogenic Emissions on Summertime Ozone Formation in the Guanzhong Basin, China

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#### Abstract

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This study is the first attempt to understand the synergistic impact of anthropogenic and biogenic emissions on summertime ozone (O<sub>3</sub>) formation in the Guanzhong (GZ) basin where Xi'an, the oldest and the most populous city (with a population of 9 million) in the northwest China, is located. Month-long (August 2011) WRF-Chem simulations with different sensitivity experiments were conducted and compared with near-surface measurements. Biogenic volatile organic compounds (VOCs) concentrations were characterized from 6 surface sites among the Qinling Mountains, and urban air composition was measured in the Xi'an city at a tower 100 m above the surface. The WRF-Chem control experiment reasonably reproduced the magnitudes and variations of observed O<sub>3</sub>, VOCs, NO<sub>x</sub>, PM<sub>2.5</sub> and meteorological parameters, with normalized mean biases for each parameter within ±21%. Subsequent analysis employed the factor separation approach (FSA) to quantitatively disentangle the pure and synergistic impacts of anthropogenic and/or biogenic sources on summertime O<sub>3</sub> formation. The impact of anthropogenic sources alone was found to be dominant for O<sub>3</sub> formation. Although anthropogenic particles reduced NO<sub>2</sub> photolysis by up to 60%, the anthropogenic sources contributed 19.1 ppb O<sub>3</sub> formation on average for urban Xi'an. The abundant biogenic VOCs from the nearby forests promoted O<sub>3</sub> formation in urban areas by interacting with the anthropogenic NO<sub>x</sub>. The calculated synergistic contribution (from both biogenic and anthropogenic sources) was up to 14.4 ppb in urban Xi'an, peaking in the afternoon. Our study reveals that the synergistic impact of individual source contributions to O<sub>3</sub> formation should be considered in the formation of air pollution control strategies, especially for big cities in the vicinity of forests.

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Keywords: ozone, emission, biogenic volatile organic compounds, WRF-Chem

#### 1. Introduction

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Elevated ozone (O<sub>3</sub>) levels in China has drawn increasing attention in recent years (e.g. Xue et al., 2014; Hu et al., 2016; Wang T. et al., 2016; Wang M. et al., 2016). O<sub>3</sub>, a secondary pollutant, is mainly formed by complex photochemical reactions of nitrogen oxides  $(NO_x = NO + NO_2)$  and volatile organic compounds (VOCs). High O<sub>3</sub> concentration at ground level is harmful to human health and ecosystems (WHO, 2005; Feng et al., 2015; Brauer et al., 2016). O<sub>3</sub> in the troposphere is an important greenhouse gas that has the third-highest radiative forcing after CO<sub>2</sub> and CH<sub>4</sub> (Stevenson et al., 2013; IPCC, 2013). In addition, O<sub>3</sub> is the primary source of the hydroxyl radical (OH), which has a major influence on the oxidizing capacity of the atmosphere and thus impacts the oxidation chemistry of secondary pollutants (e.g. sulfate and secondary organic aerosol) (Ehhalt et al., 2000; Rohrer et al., 2006). In recent years, the surface O<sub>3</sub> level has been increasing in most Chinese cities. For instance, in the highly urbanized areas in China, maximum 8h O<sub>3</sub> concentration increased by 19% (16.9 ppb) from 2015 to 2017 in the Beijing-Tianji-Hebei (BTH) region, the growth trend was 14% (11.3 ppb) for the Pearl River Delta (PRD) region (10.5)ppb) for the Yangtze River Delta (YRD) region (http://datacenter.mep.gov.cn). In the future, the pollution trend is likely to worsen due to potential changes of climate and emissions (Wang et al., 2013; Liu et al., 2013; Zhu et al., 2016, 2017).

On the global scale, VOC emissions from natural vegetation is estimated to be one order greater than that from anthropogenic activities, in spite of large uncertainties in different studies (Guenther et al., 2006; Wu et al., 2007, 2008; Jiang et al., 2013; Zhu et al., 2017). In addition, biogenic VOCs (e.g. isoprene) are highly reactive, reacting more efficiently with OH than most anthropogenic VOC species (Carlo et al., 2004). Previous studies have demonstrated the significant impacts of biogenic VOCs on surface O<sub>3</sub> formation under strong solar radiation, high temperature and NO<sub>x</sub> level (e.g. Fiore et al., 2005, 2011; Wang et al., 2008; Curci et al., 2009; Geng et al., 2011; Strong et al., 2013; Squire et al., 2014; Lee et al., 2014; Zhang et al., 2017). In China, biogenic VOC emissions are estimated to be 17-44 TgC yr<sup>-1</sup> (Guenther et al., 1995; Lin et al., 2008; Fu et al., 2012, 2104; Li et al., 2014), and are concentrated in the warm summer season. Current studies report that biogenic VOCs contribute to surface O<sub>3</sub> concentrations in China (Geng et al., 2011; Qu et al.,

2013, 2014) and play an important role in intercontinental transport of  $O_3$  (Zhu et al., 2017).

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The impacts of biogenic VOCs on O<sub>3</sub> formation may vary in different regions and different seasons (Im et al., 2011; Strong et al., 2013; Wagner et al., 2014; Lee et al., 2014). Qu et al. (2013) employed the RAQM model to examine the influence of biogenic emissions on daily maximum surface O<sub>3</sub> concentration in China. Their calculations showed that in general the impact from biogenic sources on O<sub>3</sub> were more obvious in South China than in North China, but the O<sub>3</sub> increments in different regions didn't follow the same seasonality. Geng et al. (2011) used the WRF-Chem model to evaluate the effect of biogenic emissions on O<sub>3</sub> production in Shanghai in summer, and suggested that the carbonyls produced by the continuous oxidation of isoprene have important impacts on O<sub>3</sub> level in the city. In addition, some studies suggest that biogenic emissions may increase due to global warming and land-use change, and the impact on O<sub>3</sub> formation could be more significant in the future (Lin et al., 2008; Fiore et al., 2011; Liu et al., 2013; Fu et al; 2015). Fiore et al. (2011) pointed out that the potential increases in biogenic isoprene in North American (NA) could offset the regional and intercontinental surface O<sub>3</sub> decreases produced by controls on NA anthropogenic emissions during warm seasons.

The Guanzhong (GZ) basin is the most developed region in northwestern China. In the past few years, air pollution has grown up to be a severe issue in the GZ basin (Wang et al., 2012; Xue et al., 2017), due to its specific basin topography and abundant anthropogenic emissions (Li et al., 2017). According to the data from national environmental monitoring stations in the GZ basin, 43% of days in 2013-2017 have AQI > 100 (e.g., unhealthy air quality category), and in summer O<sub>3</sub> was regarded as the primary pollutant in 70% of polluted cases. In this study, we employed a regional chemical model WRF-Chem to simulate O<sub>3</sub> concentration in the GZ basin for summer 2011. Our aim is the source apportionment of urban O<sub>3</sub> formation in this city surrounded by forests, specifically quantifying the individual and synergistic contributions of anthropogenic and biogenic sources. The paper is organized as follows. We first describe the sampling campaign, the chemical model and the emission data used for driving the model (Section 2). We then evaluate the model performance by comparing the observed urban air quality and biogenic VOCs with the simulated results (Section 3). Finally, we analyze the sensitivity of

#### 2. Methodology

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#### 2.1 Sampling sites and descriptions

The study was conducted in Xi'an, one of the oldest cities in the world. Xi'an is the most populous city in the northwest of China with a population of greater than 9 million. It is located in the heartland of the GZ basin between the Qinling Mountains and the Loess Plateau (Fig. 1). The topographical features result in air pollutants often being trapped in the valley with limited dispersion. The city borders the northern foot of the Qinling Mountains, only around 50 kilometers from downtown to the foothills. The Qinling Mountains are an east-west mountain range of 1600 kilometers in length and 300 kilometers in width and are regarded as a natural boundary between Northern and Southern China. Climate, and even culture are significantly different from the north to the south. The Xi'an city has high temperatures and strong solar intensity in summer, making it an ideal location to assess the importance of biogenic contributions to urban air quality.

#### 2.1.1 Biogenic VOC measurements in the Qinling Mountains

We selected six field sites in the Qinling Mountains (Fig. 1b, the triangles) and collected one ambient air sample at each site on 6<sup>th</sup> – 7<sup>th</sup> August 2011 under sunny weather conditions (details are presented in Table 1). Sampling was conducted between 9:30 am to 16:30 pm local time (each sampling lasted for 30 minutes) to target expected daily maximum isoprene concentrations. At each site, ambient air samples were pulled in parallel onto three cartridges filled with Tenax GR and Carbograph 5TD solid adsorbents using a mass flow controlled pump for 30 min. Samples were shipped to the lab at NCAR (Boulder CO, USA) for chemical analysis. Cartridges were desorbed using an Ultra<sup>TM</sup> TD auto sampler with a Unity thermal desorption system (MARKES International Series 2, Llantrisant, UK) interfaced with a temperature programmed Agilent 7890A series Gas Chromatograph with a 5975C Electron Impact Mass Spectrometer and a Flame Ionization Detector (GC-MS/FID, Agilent Technologies, Santa Clara, CA, USA). We used nitrogen as a carrier gas at the flow rate of 3 mL min<sup>-1</sup>. Isoprene and monoterpene identifications were based on the

comparison of retention time of authentic standards and mass spectra in the National Institute of Standards and Technologies (NIST) databases. Quantifications were calculated using FID calibrated with a NIST traceable standard.

#### 2.1.2 Air quality monitoring in urban Xi'an

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We set up an urban air quality monitoring site (Fig. 1b) at the roof (107 m above ground) of the main building (108.984°E, 34.245°N) on the campus of Xi'an Jiaotong University to minimize the ground level influences of local emissions. The campus is in the southeastern part of the downtown surrounded by residential areas. We obtained reliable observations of the concentrations of O<sub>3</sub>, NO<sub>x</sub> and PM<sub>2.5</sub> during 15<sup>th</sup> - 30<sup>th</sup> August 2011. Gases were measured by Ecotech analyzers (Ecotech Pty Ltd, Australia). O<sub>3</sub> were measured by an UV photometric analyzer EC9801. NO<sub>x</sub> was measured by a gas-phase chemiluminescence detection analyzer EC9841, coupled with a hot molybdenum converter. We collected 24h PM<sub>2.5</sub> filter samples by a mini-volume sampler (Airmetrics, USA) at a flow rate of 5 L min<sup>-1</sup> using both 47 mm quartz-fiber (Whatman, Mid-dlesex, UK) and teflon-membrane (Gelman, Ann Arbor, MI) filters. We calculated the PM<sub>2.5</sub> mass concentrations gravimetrically by weighing the teflon-membrane filters pre and post-collection for at least 4 times using an electronic microbalance (MC5, Sartorius, Göttingen, Germany) with ±1 μg sensitivity under controlled conditions. EC and OC concentrations were analyzed based on a 0.5 cm<sup>2</sup> punch from the quartz-fiber filter following the IMPROVE\_A (Interagency Monitoring of Protected Visual Environments) thermal/optical reflectance (TOR) protocol (Chow et al., 2007) using a DRI model 2001 Carbon Analyzer. The concentrations of ions were quantified from a 10.8 cm<sup>2</sup> of the teflon-membrane filter by a Dionex DX-600 ion chromatography (Dionex Inc., Sunnyvale, CA, USA) (Zhang et al., 2014).

#### 2.2 The WRF-Chem model

We employed the WRF-Chem model to study biogenic VOC emissions from the Qinling Mountains and their contributions to regional O<sub>3</sub> formation in urban Xi'an. WRF-Chem is a 3-D online-coupled meteorology and chemistry model consisting of the components of emission, transport, chemical transformation, photolysis and radiation (Tie et al., 2003, Li et al., 2011), dry and wet deposition (Wesely, 1989), and

aerosol interactions (replaced with CMAQ aerosol module, Binkowski and Roselle, 2003; Li et al., 2010). WRF is a non-hydrostatic mesoscale dynamical system with various options for physical parameterizations (Skamarock et al., 2008). The chemical modules were implemented into the WRF framework obeying the same schemes for the simultaneous simulations (Grell et al., 2005).

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We adopted RADM2 (Regional Acid Deposition Model) as the gas phase chemical mechanism to predict O<sub>3</sub> formation. RADM2 is an aggregated species type using the reactivity based weighting scheme to adjust for lumping (Stockwell et al., 1990). The mechanism implemented in our WRF-Chem model covers 158 reactions among 36 species, containing the complete reaction paths for isoprene, monoterpenes and the relevant inorganic reactions. As an explicit species, isoprene chemistry is based on an updated CB4 gas-phase mechanism (Carter and Atkinson., 1996). We noted some advances in isoprene chemistry recently. Some studies pointed out that isoprene reacts with OH radical to form hydroperoxy radicals (ISOPO2). Subsequently, in the presence of NO<sub>x</sub>, ISOPO2 reacts with NO leading to the production of hydroxynitrates (ISOPN) by a minor branch, which sequesters NO<sub>x</sub> and therefore regulates O<sub>3</sub> formation. A number of laboratory, filed observation and simulation studies (e.g. Paulot et al., 2009a, b, 2012; Horowitz et al., 2007; Hudman et al., 2009; Fisher et al., 2016; Travis et al., 2016) highlighted the importance of isoprene nitrate chemistry and all agreed there were still large uncertainties (for example, the estimates of ISOPN yield (4%-15%)). Horowitz et al. (2007) found a 4% ISOPN yield best captured the alkyl and multifunctional nitrates measured by aircraft, and Hudman et al. (2009) pointed out that unreasonably high ISOPN yield (18%) would let ISOPN be a terminal sink for NO<sub>x</sub>. In addition, the production of radicals from ISOPO2 is regarded to be important for HOx and further for secondary pollutants (e.g. O<sub>3</sub> and secondary organic aerosol (SOA)) (e.g. Squire et al., 2015). Kanaya et al. (2012) quantified the formation of hydroperoxy-aldehydes (HPALDs) from ISOPO2 and subsequent rapid release of HOx. They revised the model by adding detailed reactions of isomerization of ISOPO2 and photolysis of HPALDs following Peeters and Müller (2010), and found the revision could increase OH and HO2 concentrations by 28-38% for daytime. However, Kanaya et al. (2012) also pointed out that isomerization of ISOPO2 at the rates proposed by Peeters and Müller (2010) might be overestimated.

Back to our study, these advances in isoprene chemistry were not contained in the standard RADM2 mechanism. To assess the potential uncertainties, we modified the RADM2 mechanism by adding formation pathway of ISOPN from ISOPO2+NO (the yield of 4%) following Horowitz et al. (2007) and ISOPO2 isomerization reaction following Li et al. (2018). Details of these updates can be found in Table S1. We conducted four sensitivity simulations (15<sup>th</sup>-17<sup>th</sup> Aug.), namely ISOO (standard RAMD2 mechanism), ISO1 (including the ISOPO2 isomerization reaction), ISO2 (including reactions of isoprene nitrate), and ISO3 (including both revisions in ISO1 and ISO2), to explore the effects on O<sub>3</sub> in the GZ basin. The results show that near surface O<sub>3</sub> concentration was decreased by 2.9 ppb (7%) averaged for urban Xi'an in August after implementing these updates (ISO3-ISOO), and we thought this would not crucially impact conclusions of this study. Details of the sensitivity analysis results can be found in Table S2 and Fig. S1.

The simulated domain (Fig. 1) is  $600\times600~\text{km}^2$  centered on urban Xi'an with 3 km horizontal grid spacing. We set up 28 vertical layers from the surface up to 50 hPa with 7 layers below 1 km to assure a high near-ground vertical resolution. The National Centers for Environmental Prediction (NCEP) FNL Operational Global Analysis data provided the initial and boundary fields of meteorology. Initial and boundary conditions of chemistry were derived by a global chemical transport model (Model for Ozone and Related chemical Tracers, MOZART) (Emmons et al., 2010). We considered the first 7 days as spin-up period, and the study focused on  $6^{th} - 7^{th}$  and  $15^{th} - 30^{th}$  August 2011 because of the available field observation datasets (as described in Section 2.1).

#### 2.3 Biogenic and anthropogenic emissions

Biogenic emissions were quantified by the widely used model MEGAN (Model of Emissions of Gas and Aerosols from Nature) (Guenther et al., 2006). MEGAN coupled into the WRF-Chem model, referred to here as WRF-MEGAN2, provides on-line estimates of the net landscape-averaged biogenic emissions from terrestrial ecosystems into the above-canopy atmosphere. The on-line estimated emissions of isoprene, individual monoterpenes and other biogenic VOCs serve as the inputs for the further chemistry simulation. To drive MEGAN, we need the following inputs:

emission factors (EFs), leaf area index (LAI), plant functional types (PFTs), as well as meteorology conditions. The meteorology was obtained from WRF simulations and the LAI and PFT data were extracted from MODIS (Tian et al., 2004). We adopted the canopy-scale emission factors of dominant species from Guenther et al. (2006).

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Estimated the whole year 2011 by WRF-MEGAN2, the isoprene and monoterpene emissions in the Qinling Mountains were mostly concentrated in summer (71% and 58%, respectively). During our simulation period, the isoprene emission from the domain is 157 Gg mon<sup>-1</sup>, accounting for ~80% of total biogenic VOC emissions (Table S3). The rest are monoterpenes and other biogenic VOCs (e.g. acetone and MBO (2-methyl-3-buten-2-ol)). Figures 2a and 2b show the spatial distributions of biogenic isoprene and monoterpenes emission fluxes during the simulation period, indicating the high emission zone of isoprene in the Qinling Mountains lying to the south of the Xi'an city.

The anthropogenic emissions were obtained from the Multi-resolution Emission Inventory for China (MEIC, Li et al., 2017) for the year of 2010, which was downscaled to a resolution of 3 km using locations of point sources and various spatial proxies (Geng et al., 2017). The upgraded highly resolved emission data were based on a collection of statistics and newly developed emission factors. The emission inventory used in our model includes all major anthropogenic sources, but excluded open biomass burning which occupies a low proportion in the GZ basin during our period (estimated Fire simulating by Inventory from NCAR, https://www2.acom.ucar.edu/modeling/finn-fire-inventory-ncar). The anthropogenic emission sources are composed of power, industry, residential, transportation, and agriculture. The emission estimates and uncertainties of VOCs, SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, and PM<sub>2.5</sub> in the domain during the simulation period are summarized in Table S3, and the potential impacts of emission uncertainty on simulation will be discussed in Section 3.3. The estimated anthropogenic VOCs emissions are 72.2 Gg, contributing up to ~30% of total VOC emissions. Figures 2c and 2d present the spatial distributions of anthropogenic VOC and NO<sub>x</sub> emissions in the simulation period. The highest emission intensity of anthropogenic VOCs and NOx are in Xi'an city and the GZ basin due to the frequent vehicle and industrial activities of this area.

#### 2.4 Factor Separation Technique

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 $O_3$  is formed by complicated nonlinear reactions of anthropogenic and biogenic precursors (NO<sub>x</sub> and VOCs) in the presence of sunlight. The approach referred to as the "brute-force" method (sensitivity analysis used to measure the model output response to emission changes) is traditionally used in air quality model to identify source contributions from specific non-reactive species in a linear process, but it cannot straightforward apply to secondary species due to the non-linearity in responses. In practice, the actual impact of one factor in a nonlinear process in the presence of others can be separated into 1) pure impact from the factor and 2) interactional impacts from all those factors. In this study, we adopted the factor separation approach (FSA) (Stein and Alpert, 1993) to decompose the pure contribution of a factor from its interaction with other factors.

We considered anthropogenic and biogenic sources as two interactional factors to influence the  $O_3$  formation.  $f_{anth-bio}$ ,  $f_{anth}$ ,  $f_{bio}$  and  $f_0$  are the simulation results including both anthropogenic and biogenic sources, anthropogenic source only, biogenic source only, and neither, respectively. Pure contributions of anthropogenic and biogenic sources are expressed as Eq. (1) and (2), respectively:

$$f'_{anth} = f_{anth} - f_0 \tag{1}$$

$$f'_{bio} = f_{bio} - f_0 \tag{2}$$

The calculated result including both anthropogenic and biogenic sources should include both pure contributions of the two factors, the synergistic impact, and the impact of background transport (Eq. 3):

$$f_{anth-bio} = f'_{anth} + f'_{bio} + f'_{anth-bio} + f_0$$
 (3)

Thus, the synergistic effect between anthropogenic and biogenic sources is represented as:

$$f'_{anth-bio} = f_{anth-bio} - f'_{anth} - f'_{bio} - f_0$$

$$= f_{anth-bio} - (f_{anth} - f_0) - (f_{bio} - f_0) - f_0$$

$$= f_{anth-bio} - f_{anth} - f_{bio} + f_0$$
(4)

Based on the FSA, we conducted four simulations, namely BASE, ANTH, BIO and NEITHER, to explore the pure and synergistic impacts of anthropogenic and/or biogenic sources on O<sub>3</sub> production in the GZ basin. Detailed simulation settings and the various contribution definitions are summarized in Table 2.

#### 3. Observation data and model validation

#### 3.1 Meteorology

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The specific topographical features of Xi'an make the meteorological conditions crucial for the accumulation and dispersion of urban pollutants. To validate our model performance in wind, temperature and relative humidity, we compared the hourly meteorological data (http://www.meteomanz.com) observed at the Jinghe site (108.58°E, 34.26°N, in the west of Xi'an) with model simulations. Figure 3c illustrates the observed and simulated near-surface wind speed and directions during 15<sup>th</sup> – 30<sup>th</sup> August 2011. The WRF-Chem model successfully captured the prevailing wind direction from north and northeast, consistent with the in-situ observations. It should be noted that in our simulation period the prevailing wind blew from south, which enhanced the transport of biogenic emissions from the Qinling Mountains to urban Xi'an. In addition, a continuous rainfall event during 18<sup>th</sup>-22<sup>nd</sup> August (green shadow in Fig. 3) was characterized by lower temperature and near-saturated humidity.

We conducted the statistical verification of meteorological variables in Table 3, including the r (correlation coefficient), NMB (normalized mean bias) and RMSE (root mean square errors). Modeled meteorological variables were in good agreement with observations (Fig. 3a-c) with the NMB less than  $\pm 6\%$ .

#### 3.2 Biogenic VOC concentrations in the Qinling Mountains

Samples from the Qinling Mountains show that the dominant VOC species was isoprene, and α-pinene was the main constituent of monoterpenes (Table 1). The ratio of isoprene to monoterpenes varies considerably. In general, different terpene emitters are not homogeneously distributed in a kilometer-scale grid and the point measurements are influenced by the microenvironment and meteorology (Zare et al., 2012; Kota et al., 2015). However, in this study, our goal is to estimate the biogenic effects on urban O<sub>3</sub> 50 km away from the foothills, which requests more concern on the regional scale VOC level, rather than the microenvironment-scale variability in either the observation or the simulation. Thus, we compared the average of VOC

measurements with model simulations to validate whether the calculated results were reasonable. The isoprene mean concentration simulated in the six grids (corresponding to the time of observations) was 1.4 ppb, which is close to the observed average value of 1.3 ppb at the six sampling sites. Monoterpenes performed quite similarly, simulated 0.22 ppb comparing with observed 0.21 ppb. We also analyzed the temporal variation of simulated biogenic VOC during the whole simulation period and found the sub-month variability was relatively small (the standard deviation < 25%). The evaluation indicates that biogenic VOCs simulations reasonably agreed with the observations in the Qinling Mountains, on average, which provides a basis for us to further evaluate biogenic effects on O<sub>3</sub>.

#### 3.3 Gaseous and particulate pollutants in urban Xi'an

The sampling campaign was organized in summer. Based on the gaseous and particulate pollutant observations, the daily mean PM<sub>2.5</sub> concentration was 90.0±53.5  $\mu$ g m<sup>-3</sup>, with 57% of days exceeding the WHO Interim target-1 (IT-1) 75  $\mu$ g m<sup>-3</sup>. The daily mean NO<sub>x</sub> concentrations were 25.8-63.2 ppb, with 40% of days exceeding the guideline 48.7 ppb ( $\approx$ 100  $\mu$ g m<sup>-3</sup>, GB 3095-2012). The maximum 8h O<sub>3</sub> concentration was 3.5-95.6 ppb, with most of the values around the national first grade standard of 46.6 ppb ( $\approx$ 100  $\mu$ g m<sup>-3</sup>, GB 3095-2012). Summer in Xi'an is monitored as the least polluted season of the year, and the case we picked is regarded as a typical situation in summer Xi'an.

Figure 3 compares the simulated hourly  $O_3$  and  $NO_x$  concentrations with in-situ observations. During the rainy episode (the green shadow in Fig. 3), our model overestimated  $NO_x$  concentration and underestimated  $O_3$  concentration. The deviation can be explained by the failure to simulate the precipitation in the WRF model resulting in underestimates in wet deposition. So, we focused our analysis on the period excluding the rainy period. During the no-raining days, our model well reproduced the diurnal variations and magnitudes of  $O_3$  and  $NO_x$  concentrations. The calculated  $O_3$  averaged for no-raining period was 38.7 ppb, ~20% higher than the observed value of 31.5 ppb. Our simulated  $O_3$  also reproduced the temporal variation of the observed  $O_3$  (r = 0.72). For  $NO_x$ , the simulated hourly  $NO_x$  averaged for the no-raining period was 46.6 ppb, close to the observed 47.0 ppb (NMB=-1%), which

suggests no systematic bias in NO<sub>x</sub> emissions. It is worth noting that the observed NO<sub>x</sub> were detected by a chemiluminescence analyzer coupled with a hot molybdenum converter (MoO), and this method was recognized to cause higher NO<sub>2</sub> concentration due to the positive interference of other nitrogen-containing components (NO<sub>z</sub>, e.g. PAN, HNO<sub>3</sub> and HONO). Xu et al. (2013) found that the uncertainty caused by the MoO converter was much smaller at urban sites (less than 10%) than that at suburban and background sites (more than 30%). In the GZ basin, to evaluate the uncertainty, we estimated the ratio of NO<sub>z</sub>/(NO<sub>x</sub>+NO<sub>z</sub>) from the model. The calculated results indicated that the NO<sub>z</sub> accounted for 11% of the total NO<sub>x</sub>+NO<sub>z</sub> at urban Xi'an during the no-raining period. We noted the uncertainty in our NO<sub>x</sub> measurements, but considered this would not crucially impact the model-measurement comparison.

Unfortunately, the anthropogenic VOC was not included in our samples as the observations are primarily targeted to biogenic VOC in the Qinling forest. Alternatively, three sensitivity simulations are conducted (for 15<sup>th</sup>-17<sup>th</sup> August), namely VOC0 (using standard MEIC emission estimates), VOC1 (with an increase of anthropogenic VOC emission by 50%) and VOC2 (with a decrease of anthropogenic VOC emission by 33%), to explore the sensitivity of simulated VOC and O<sub>3</sub> concentrations to anthropogenic VOC emissions. We found that 50% increases of anthropogenic VOC emission could lead to a 22% increase of urban VOC concentration, while the 33% emission decrease resulted in a 24% decrease of concentration (Table S4). It is worth noting that the concentration of O<sub>3</sub> stayed almost the same (because the O<sub>3</sub> production regime is NO<sub>x</sub>-limit). We addressed that the uncertainties of VOC emission obviously affected the VOC concentrations; however, MEIC inventory is the most updated available emission for China so far, and quantifying its uncertainties can be done in future studies (possibly with satellite-based measurement of HCHO (Miller et al., 2008)).

We analyzed PM<sub>2.5</sub> concentration and composition (sulfate, nitrate, ammonium, EC, organic matter) with the filter-base measurements. The model predicted PM<sub>2.5</sub> concentration to be 94.6±28.2 μg m<sup>-3</sup>, slightly lower (NMB=-12%) than measured 107 μg m<sup>-3</sup> averaged for the no-raining period, but didn't perform well in capturing temporal correlation (r=0.17). The simulated PM<sub>2.5</sub> showed the similar compositions to the observation (Fig. S2b and S2c). Sulfate is the dominant constituent of both simulated (32%) and observed (37%) PM<sub>2.5</sub>. High sulfate concentration was mainly

attributed to the high SO<sub>2</sub> emission in the GZ basin as well as the humid weather conditions (Wang et al., 2014). The secondary constituent of observed PM<sub>2.5</sub> is organic matter, which accounted for 16% of the total observed PM<sub>2.5</sub>, close to the simulated result (14%). Secondary organic matter contributed half to total simulated organic matter, mainly due to the abundant precursor (i.e. VOCs) emissions and the high atmospheric oxidation capacity in summer.

#### 4. Impacts of biogenic and anthropogenic sources on O<sub>3</sub> formation

In this section, we analyze the results from the four simulations (BASE, ANTH, BIO and NEITHER, Table 2) to characterize the fate of O<sub>3</sub> and its precursors in the GZ basin and to quantify the pure and synergistic impacts of anthropogenic and/or biogenic sources on summertime O<sub>3</sub> formation.

#### 4.1 Base simulation of O<sub>3</sub>

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Firstly, we discuss the spatial and temporal characteristics of the simulated  $O_3$  and the precursors (VOCs and  $NO_x$ ) in the GZ basin in the BASE simulation.

Figure 4a shows spatial distribution of the simulated VOCs during the no-raining period, overlaid with the simulated wind vectors. The highest concentration (more than 50 ppb, with ethane being the dominant species) was in urban Xi'an and its downwind region (the southwest of urban Xi'an), due to anthropogenic activities. In addition, another high-value area (~30 ppb, with isoprene being the dominant species) was found in the Qinling Mountains, which was probably due to biogenic sources. To better understand the composition of VOCs, we analyzed some typical individual VOC species. Figure S3a shows the spatial distribution of xylenes, representing anthropogenic VOCs, and Fig. S4a and S5a show isoprene and monoterpenes, representing biogenic VOCs. The anthropogenic xylenes were mainly distributed in the GZ basin, while the high biogenic isoprene and monoterpene concentrations were found over the Qinling Mountains. These results explain the spatial feature of total VOCs and the dominant sources. Detailed discussion of source apportionment is given in Section 4.2

The spatial distribution of  $NO_x$  was slightly different (Fig. 5a). The highest concentrations of  $NO_x$  were in the GZ basin (average of 11.1 ppb), especially in urban

Xi'an (averaged of 30.1 ppb), while among the Qinling Mountains, NO<sub>x</sub> was low and dominated by biogenic sources.

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PM, even though not directly involved in the formation pathways of O<sub>3</sub>, influences the chemical equilibrium indirectly. In the daytime, NO<sub>2</sub> photolysis frequency (*J*(NO<sub>2</sub>)) is determined by the solar radiation influenced by PM via scattering and absorption. Figure 6 shows the changes of *J*(NO<sub>2</sub>) (calculated by model track output photor\_no<sub>2</sub>) with the participation of PM (concentration, 102 μg m<sup>-3</sup>; aerosol optical depth (AOD) at 550 nm, 1.92; single scattering albedo (SSA) at 550 nm, 0.92) averaged for urban Xi'an. *J*(NO<sub>2</sub>) was reduced by 40-60%, most significantly in morning and evening rush hours. In the night time, PM<sub>2.5</sub> can remove N<sub>2</sub>O<sub>5</sub> from the NO<sub>x</sub> cycle via heterogeneous reactions, as one of the major NO<sub>x</sub> sinks in the atmosphere (Xue et al., 2014). Figure S6a shows the spatial feature of PM<sub>2.5</sub>. The densest area was urban Xi'an (averaged for 102 μg m<sup>-3</sup>) followed by the western part of the GZ basin. The spatial distribution of high-values of PM<sub>2.5</sub> was similar to that of NO<sub>x</sub>, but covered a wider area mostly in the downwind region of urban Xi'an, which is expected due to longer lifetime of aerosols compared with NO<sub>x</sub> and the time required for secondary aerosol formation, thus further dispersion.

The typical diurnal variation of O<sub>3</sub> (Fig. 7b) demonstrates there are higher concentrations in the afternoon and lower at night. For better understanding of O<sub>3</sub> concentration characteristics and source/sink mechanisms, we discussed two different time scales: 1) O<sub>3</sub> peak time (14:00-18:00) (Fig. 8) and 2) O<sub>3</sub> 24-hour average (Fig. 9). During the peak time, simulated near-surface O<sub>3</sub> was high in the GZ basin, with averaged concentration of 75 ppb. In the downwind region of high NO<sub>x</sub> and VOCs in the west of urban Xi'an, the concentration reached up to 110 ppb. We employed the ratio of H<sub>2</sub>O<sub>2</sub>/HNO<sub>3</sub> to investigate the chemistry regime of O<sub>3</sub> formation (Sillman 1995; Wang et al., 2017). If the ratio is greater than 0.5, the O<sub>3</sub> production regime is considered NO<sub>x</sub>-controlled, otherwise VOC-controlled if the ratio less than 0.3. The range between 0.3 and 0.5 is defined as the transition regime from NO<sub>x</sub>- to VOC-controlled, indicating the competition of both NO<sub>x</sub> and VOCs in O<sub>3</sub> production. Figure 10a shows the spatial distribution of the simulated H<sub>2</sub>O<sub>2</sub>/HNO<sub>3</sub> ratio during the O<sub>3</sub> peak time. The west and southeast of the GZ basin were right in the transition regime with a complicated O<sub>3</sub> production mechanism sensitive to both NO<sub>x</sub> and VOCs. Most of the rest of the simulation region was VOC-controlled, excluding the

Yuncheng and Hejing cities in the neighboring Shanxi provinces.

On the 24h average scale, the spatial distribution of O<sub>3</sub> presented a different picture (Fig. 9). The original high-value area during the peak time in the GZ city cluster shifted to low-value region due to the consumption of O<sub>3</sub> by abundant NO<sub>x</sub> emissions. At night time, the titration effect of freshly emitted NO dominates, and the O<sub>3</sub> concentration tends to drop to a lower level. The high value of 24h averaged O<sub>3</sub> converged in the south and northwest outside of the GZ Basin. Those areas have elevated O<sub>3</sub> due to high daytime production, similar to the nearby zone of peak O<sub>3</sub>, but also have lower emissions of NO resulting in lower loss of O<sub>3</sub>.

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#### 4.2 Pure impact of biogenic or anthropogenic sources

Using the FSA method, we evaluated the pure contribution of anthropogenic or biogenic sources to the summertime O<sub>3</sub> formation in the GZ basin. In the scenario of pure contribution of anthropogenic emissions, the VOC concentrations were mostly distributed over the GZ city cluster (8.0 ppb), especially in urban Xi'an (26.4 ppb) (Fig. 4c). In the scenario of pure contribution of biogenic emissions, the VOCs were widely dispersed over the Qinling Mountains (Fig. 4d), with a calculated 9.9 ppb for the GZ basin and 12.4 ppb for urban Xi'an. NO<sub>x</sub> concentration has the similar pattern as VOCs in the scenario of pure contribution of anthropogenic emissions, with averaged concentrations of 11.0 ppb for the GZ basin and 30.3 ppb for urban Xi'an (Fig. 5c). However, in the GZ basin and urban Xi'an, biogenic sources contributed less than 0.2 ppb to NO<sub>x</sub> concentration (Fig. 5d). In the scenario of pure contribution of anthropogenic emissions, PM<sub>2.5</sub> spread over a wider area (Fig. S6c), due to the time required for secondary aerosol formation. In the scenario of pure contribution of biogenic emissions, PM<sub>2.5</sub> was mostly distributed among the Qinling Mountains (Fig. S6d), but the concentration was lower by one order of magnitude (Table 4, Fig. S7).

In the scenario of pure contribution of anthropogenic emissions, daily peak O<sub>3</sub> accumulated in the downwind region in the center and western GZ due to high VOCs and NO<sub>x</sub> concentrations. Daily peak O<sub>3</sub> concentrations reached 22.6 ppb for GZ and 19.1 ppb for urban Xi'an (Fig. 8c). Lower daily peak O<sub>3</sub> concentration was found outside of GZ basin where less anthropogenic VOCs and NO<sub>x</sub> was emitted. In contrast, daily peak O<sub>3</sub> was negligible (less than 3 ppb) in the scenario of pure contribution of

biogenic emissions (Fig. 8d) due to the low NO<sub>x</sub> emissions. However, the distribution of 24h averaged O<sub>3</sub> was different from daily peak O<sub>3</sub>. 24h averaged O<sub>3</sub> concentration in the scenario of pure contribution of anthropogenic emissions was more diluted in the GZ city cluster than for surrounding areas (Fig. 9c). Due to the abundant NO emission and its titration effect on O<sub>3</sub>, the pure effect of anthropogenic sources was negative, calculated to be -2.2 ppb in urban Xi'an.

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## 4.3 Synergistic impact of the interaction between biogenic and anthropogenic sources

The synergistic impact on O<sub>3</sub> formation includes the interactions between anthropogenic and biogenic sources. In other word, it reflects the potential production trend of either "O<sub>3</sub>-promoted" or "O<sub>3</sub>-suppressed" under the natural coexistence of all emission sources. In the cases of NO<sub>x</sub>, VOCs and PM<sub>2.5</sub>, the synergistic impacts contributed less than  $\pm 3\%$  of total concentrations (Fig. 4b, 5b, S5b, S6, and Table 4). However, the synergistic impact on O<sub>3</sub> played a remarkable role showing positive impacts for both daily peak (Fig. 8b) and 24h averaged O<sub>3</sub> (Fig. 9b). It means that the mixed state of anthropogenic and biogenic sources potentially enhanced the O<sub>3</sub> production more than each single source. To make it more specific, we started the discussion from the result of ANTH simulation without the biogenic sources (Table 2). Figure 10b shows the O<sub>3</sub> production regime in the ANTH simulation. VOC-controlled O<sub>3</sub> production regime covered the west of urban Xi'an and the southeast of the GZ basin. In the rest of the GZ basin and the neighboring Shanxi provinces, the O<sub>3</sub> production was in the transition regime, controlled by both NO<sub>x</sub> and VOC. NO<sub>x</sub>-controlled O<sub>3</sub> production regime dominated the rest of the region. After we included biogenic VOC emissions in the simulation, the O<sub>3</sub> concentration was significantly enhanced in the VOC-controlled regions, and partly enhanced in the mix-controlled region. However, in the VOC-controlled region, the synergistic impact contributed little.

The synergistic impact is of great importance, approximately the same magnitude as the impact from pure contributions of anthropogenic sources. The synergistic impact contributed daily peak O<sub>3</sub> concentrations of 10.5 ppb for the GZ basin and 14.3 ppb for urban Xi'an, while the pure anthropogenic impact contributed 22.6 ppb

for GZ basin and 19.1 ppb for urban Xi'an. However, the extent was ~50% smaller on the 24h averaged scale, but still increased O<sub>3</sub> concentration by 5.8 ppb for the GZ basin and 6.8 ppb for urban Xi'an. Figure 7 shows the diurnal variation of the observed and simulated O<sub>3</sub> concentration at Xi'an Jiaotong University, as well as the tested contributing components. Transport dominated O<sub>3</sub>, constantly contributing 30-40 ppb as background. The impact of pure anthropogenic sources was positive on O<sub>3</sub> production during 13:00-19:00 but negative during the rest of the time, and the impact of pure biogenic sources was negligible. Synergistic impact of both anthropogenic and biogenic sources resulted in a positive contribution during 10:00-21:00, comparable to the impact of pure anthropogenic sources.

It is worth noting that the biogenic contribution to PM<sub>2.5</sub> is not obvious (less than 3%) in GZ basin, which might be different from some other regions (e.g. Fu et al., 2012; Li et al., 2013). The main reasons are that 1) organic matter, the most important biogenic PM<sub>2.5</sub> component, only accounted for 14-16% of PM<sub>2.5</sub> in GZ basin in August; 2) Undeniably, uncertainties still exist in organic matter simulations in the model.

#### 5. Conclusions

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The GZ basin is a representative region in the northwest of China, suffering serious air pollution in recent years. Geographically, the GZ basin borders the northern foot of the Qinling Mountains. For this reason, in addition to the anthropogenic emissions from metropolitan areas, biogenic emissions are of great importance in the region, especially in warm season with active photochemistry. In this study, we used the WRF-Chem model to simulate O<sub>3</sub> in the GZ basin and compared the results to near-surface measurements, with the aim of quantifying the pure and synergistic impacts of anthropogenic and/or biogenic sources on summertime O<sub>3</sub> formation. The simulation was driven by the best currently available inventory of anthropogenic emissions and online calculated biogenic emissions. Near-surface measurements were captured from 6 surface sites among the Qinling Mountains for biogenic VOCs and one 100-m-high site in the Xi'an city for air quality (NOx, VOCs, O<sub>3</sub> and PM<sub>2.5</sub>).

Our model successfully reproduced the observed air quality and meteorological

parameters. The biogenic VOCs simulation showed a reasonable agreement. Our model also well-reproduced the magnitudes and variations of  $O_3$ ,  $NO_x$  and  $PM_{2.5}$  concentrations excluding rainy days, with normalized mean bias less than  $\pm 21\%$ .

We further conducted three scenario simulations to explore the pure and synergistic impacts of anthropogenic and/or biogenic sources on O<sub>3</sub> and the precursors, by using the factor separation approach (FSA). The results concluded that, for the precursors, pure impact of anthropogenic source contributed 99% of NO<sub>x</sub>, 80% of PM<sub>2.5</sub>, and 33% of VOCs in the GZ basin, and pure impact of biogenic source contributed 40% of VOCs but only 1-5% of PM<sub>2.5</sub> and NO<sub>x</sub>. Meanwhile, synergistic impacts from the combination of anthropogenic and biogenic sources did not bring significant changes on NO<sub>x</sub>, VOCs and PM<sub>2.5</sub> (less than ±4%). In the case of daily peak O<sub>3</sub>, the pure impact of anthropogenic source remained the dominant contributor (19.1 ppb for urban Xi'an), even after anthropogenic particles reduced the NO<sub>2</sub> photolysis by up to 60%. The abundant biogenic VOCs from the nearby forests promoted the O<sub>3</sub> formation by interaction with anthropogenic NO<sub>x</sub>, contributing 14.4 ppb to O<sub>3</sub> in urban Xi'an. This synergistic impact presented a positive contribution to O<sub>3</sub> production throughout the day and the positive effect was much more prominent during 12:00-19:00.

O<sub>3</sub> pollution in China has been raising increasing concerns in recent years. Some scientists hold the view that excessive concentration of PM<sub>2.5</sub> suppressed the formation of O<sub>3</sub> in the past, hiding the problem temporally. However, with the effective control of PM<sub>2.5</sub>, O<sub>3</sub> pollution is manifested. The phenomenon can also be demonstrated by the government control action during G20 summit (The Group of Twenty Finance Ministers and Central Bank Governors) in Hangzhou in 2016. The concentration of PM<sub>2.5</sub> was depressed sharply under the strict emission control, but O<sub>3</sub> concentration was even higher than usual. Better understanding of O<sub>3</sub> pollution sources/sinks and formation mechanisms in high PM<sub>2.5</sub> exposed area in China will benefit and guide the implementation of PM<sub>2.5</sub>/O<sub>3</sub> cooperative control. Our results suggest that, in big cities geographically close to forest, O<sub>3</sub> pollution can be enhanced by the synergistic impact from the combination of biogenic and anthropogenic sources. The synergistic contribution of each single source to O<sub>3</sub> formation cannot be neglected when making pollution control strategies.

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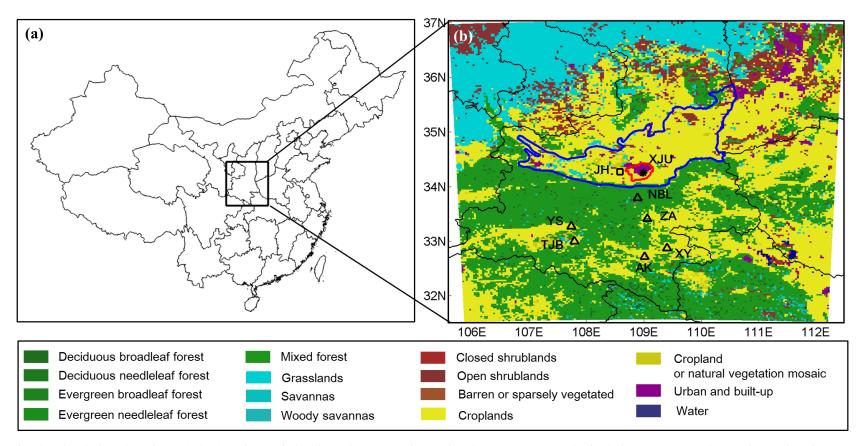
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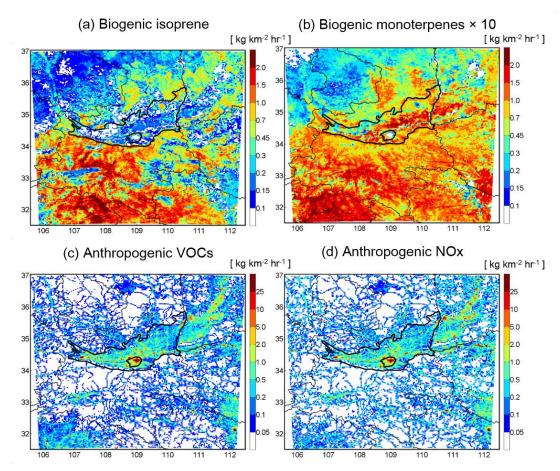
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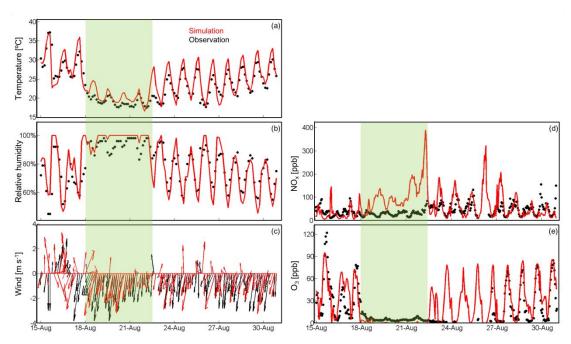
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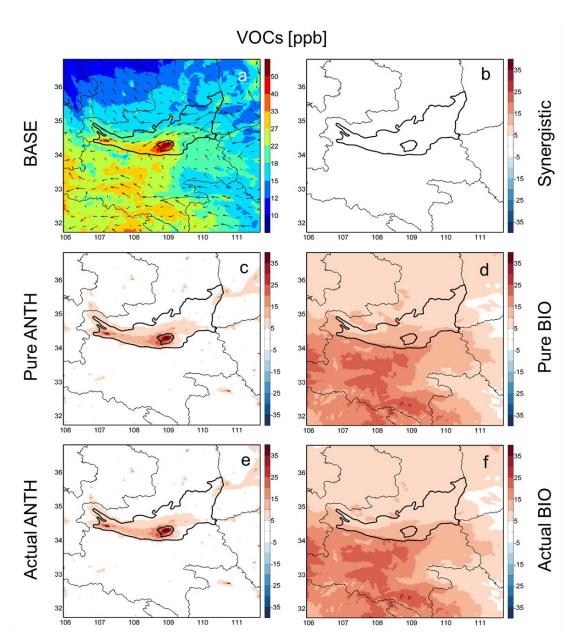
**Figure 1.** The simulation domain and the locations of six biogenic VOCs sites (triangles), one meteorological site (square) and one air quality site (snowflake). Underlain are land types from MODIS. The area of red line indicates the urban area of the Xi'an city. The area of blue line indicates the GZ basin.



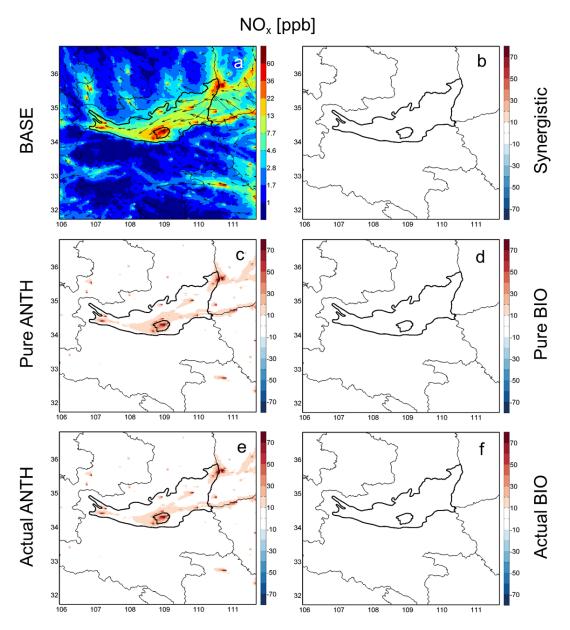
**Figure 2.** Monthly mean emissions of (a) biogenic isoprene, (b) biogenic monoterpenes, (c) anthropogenic VOCs and (d) anthropogenic  $NO_x$  in the GZ basin and surrounding areas in August 2011.



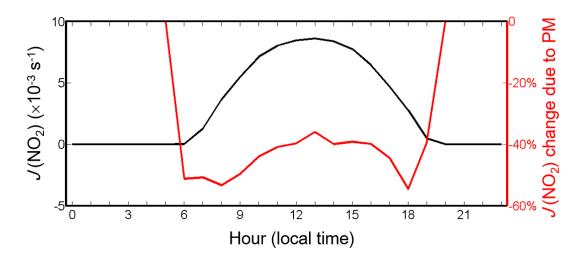
**Figure 3.** Observed (black) and simulated (red) temporal patterns of temperature (a), relative humidity (b) and wind (c) at the Jinghe site and  $NO_x$  (d) and  $O_3$  (e) concentrations at Xi'an Jiaotong University during the period from  $15^{th}$  to  $30^{th}$  August 2011. The green shadow ( $18^{th}$  - $22^{nd}$  August) indicates rainy days.



**Figure 4.** Spatial distributions of monthly mean concentrations of VOCs in August 2011. (a) is the result from the BASE simulation, overlaid with simulated wind vectors. (b)-(f) are simulated VOCs concentrations contributed from synergistic anthropogenic and biogenic, pure anthropogenic, pure biogenic, actual anthropogenic and actual biogenic sources, respectively.



**Figure 5.** Spatial distributions of monthly mean concentrations of  $NO_x$  in August 2011. (a) is the result from the BASE simulation, overlaid with simulated wind vectors. (b)-(f) are simulated  $NO_x$  concentrations contributed from synergistic anthropogenic and biogenic, pure anthropogenic, pure biogenic, actual anthropogenic and actual biogenic sources, respectively.



**Figure 6.** Diurnal variations of  $J(NO_2)$  (black) and the changes in  $J(NO_2)$  (red) averaged in urban Xi'an due to PM effects in August 2011.

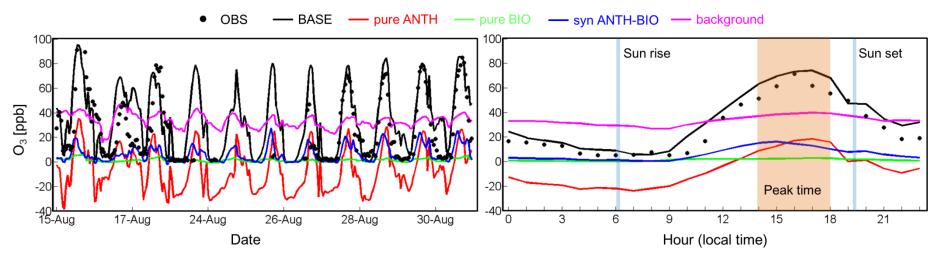
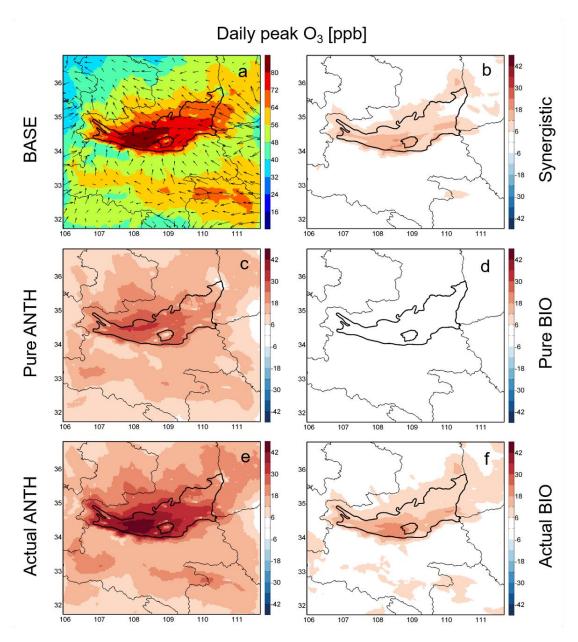
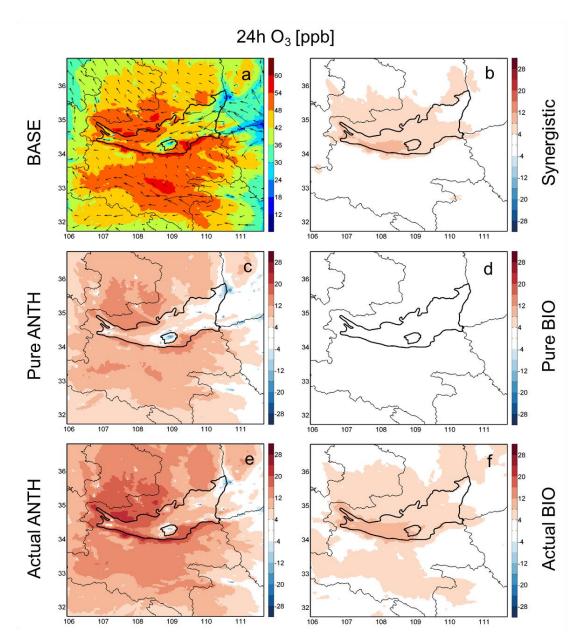


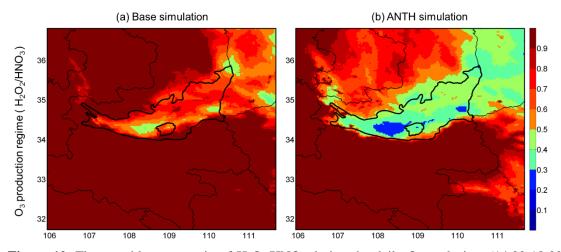
Figure 7. Temporal patterns of the simulated  $O_3$  concentrations and the tested contributing components during the period from 15<sup>th</sup> to 30<sup>th</sup> August 2011, excluding the rainy days (18<sup>th</sup> -22<sup>nd</sup> August). The orange shadow (14:00-18:00) indicates daily  $O_3$  peak time.



**Figure 8.** Spatial distributions of monthly mean concentrations of daily peak  $O_3$  in August 2011. (a) is the result from the BASE simulation, overlaid with simulated wind vectors. (b)-(f) are simulated daily peak  $O_3$  concentrations contributed from synergistic anthropogenic and biogenic, pure anthropogenic, pure biogenic, actual anthropogenic and actual biogenic sources, respectively.



**Figure 9.** Spatial distributions of monthly mean concentrations of 24h averaged  $O_3$  in August 2011. (a) is the result from the BASE simulation, overlaid with simulated wind vectors. (b)-(f) are simulated 24h averaged  $O_3$  concentrations contributed from synergistic anthropogenic and biogenic, pure anthropogenic, pure biogenic, actual anthropogenic and actual biogenic sources, respectively.



**Figure 10.** The monthly mean ratio of  $H_2O_2/HNO_3$  during the daily  $O_3$  peak time (14:00-18:00 local time) in August 2011 in the (a) Base simulation and (b) the simulation without biogenic sources.

**Table 1.** Ambient biogenic VOCs observations in the Qinling Mountains during 6<sup>th</sup> – 7<sup>th</sup> August 2011

Site <sup>a</sup>	Date	Start Time <sup>b</sup>	Location	Isoprene (ppb)		Monoterpenes (ppb)		Dominant monoterpenes <sup>c</sup>
				Observation	Simulation	Observation	Simulation	
NBL	2011/8/6	10:20	33.78°E, 108.88°N	3.8	4.5	0.42	0.29	α-pinene
ZA	2011/8/6	12:44	33.40°E, 109.05°N	0.1	0.9	0.16	0.22	$\alpha$ -pinene, limonene, menthone
XY	2011/8/6	16:14	32.87°E, 109.40°N	1.0	0.3	0.24	0.13	α-pinene
AK	2011/8/7	09:45	32.71°E, 109.01°N	0.8	0.4	0.10	0.08	α-pinene
TJB	2011/8/7	13:00	32.99°E, 107.78°N	0.5	1.1	0.27	0.26	α-pinene
YS	2011/8/7	14:50	33.27°E, 107.73°N	1.6	0.9	0.04	0.32	α-pinene
Average				1.3	1.4	0.21	0.22	

<sup>&</sup>lt;sup>a</sup> Site names: NBL (Niubeiliang), ZA (Zhenan), XY (Xunyang), AK (Ankang), TJB (Tangjiaba), YS (Youshui)

<sup>&</sup>lt;sup>b</sup> The sampling duration is 30 minutes.

<sup>&</sup>lt;sup>c</sup> Other monoterpenes detected (0.1% to 9%) include tricyclene, α-thujene, camphene, sabinene, myrcene, α-phellandrene,  $\Delta$ -carene, o-cymene, β-ocimene, cineole, isopulegol, isomenthone.

**Table 2.** Summary of different simulation settings and definitions of the various contributions from anthropogenic and/or biogenic sources.

Simulation	Simulation results	Anthropogenic emission	Biogenic emission
BASE	$f_{ m anth-bio}$	✓	✓
ANTH	$f_{ m anth}$	✓	×
BIO	$f_{ m bio}$	×	$\checkmark$
NEITHER	$f_0$	*	×
Contribution			
$f_{ m anth-bio}-f_{ m bio}$		Actual contribution of anthropoge	enic emissions
$f_{ m anth-bio}$ $-f_{ m anth}$		Actual contribution of biogenic en	missions
$f'_{0} = f_{0}$		The contribution of background to	ransport
$f'_{\text{anth}} = f_{\text{anth}} - f_0$		Pure contribution of anthropogen	ic emissions
$f'_{\text{bio}} = f_{\text{bio}} - f_0$		Pure contribution of biogenic emi	issions
$f'_{\text{anth-bio}} = f_{\text{anth-bio}} -$	$(f_{\text{anth}} + f_{\text{bio}}) + f_0$	Synergistic contribution of an	thropogenic and biogenic
		emissions	

Table 3. Statistics of meteorological and air quality variables over the GZ basin in August 2011<sup>a</sup>

	Me	an	r <sup>d</sup>	NMB <sup>d</sup>	RMSE d	
	Observation	Simulation	1 -	INIVID	KIVISE "	
Meteorology <sup>b</sup>						
Wind speed (m s <sup>-1</sup> )	2.6	2.5	-	-6%	1.8	
Temperature (°C)	25.1	24.2	0.86	4%	2.5	
Relative humidity	73.6%	74.2%	0.72	1%	12%	
Air quality <sup>c</sup>						
$NO_{x}$ (ppb)	47.0	46.6	0.36	-1%	18.1	
O <sub>3</sub> (ppb)	31.5	38.7	0.72	21%	8.1	
$PM_{2.5} (\mu g m^{-3})$	107	94.6	-	-12%	49.3	

<sup>&</sup>lt;sup>a</sup> Averaged for the period from 15<sup>th</sup> to 30<sup>th</sup> August 2011, excluding the rainy days.

 $<sup>^{\</sup>rm b}$  Meteorological data were obtained from the hourly surface measurements at Jinghe station (108.58°E, 34.26°N)

<sup>&</sup>lt;sup>c</sup> Air quality data were measured at the roof (107 m above ground) of the main building (108.98°E, 34.25°N) on the campus of Xi'an Jiaotong University

<sup>&</sup>lt;sup>d</sup> r: correlation coefficient; NMB: normalized mean bias; RMSE: root mean square errors.

**Table 4.** The various contribution components of the simulated  $O_3$  (and the precursors) and  $PM_{2.5}$  in August 2011

	$NO_x$	VOCs	O <sub>3</sub> [ppb]		$PM_{2.5}$
	[ppb]	[ppb]	daily peak <sup>a</sup>	24h	_ [μg m <sup>-3</sup> ]
The GZ basin					
Base	11.1	24.5	74.1	44.4	65.1
Pure ANTH	11.0	8.0	22.6	5.0	52.0
Pure BIO	0.1	9.9	2.0	1.1	3.3
Actual ANTH	10.6	7.7	33.0	10.8	53.5
Actual BIO	-0.3	9.6	12.5	7.0	4.9
Syn ANTH-BIO	-0.4	-0.3	10.5	5.8	1.5
Urban Xi'an					
Base	30.1	44.8	74.7	38.7	102
Pure ANTH	30.3	26.4	19.1	-2.2	88.7
Pure BIO	0.15	12.4	2.6	1.4	3.4
Actual ANTH	29.6	26.2	33.4	4.6	91.1
Actual BIO	-0.6	12.2	16.8	8.2	5.8
Syn ANTH-BIO	-0.7	-0.2	14.3	6.8	2.4

<sup>&</sup>lt;sup>a</sup> Daily O<sub>3</sub> peak time is from 14:00 to 18:00 local time