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- 22 Abstract
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24 This study investigates aerosol hygroscopicity, mixing state, and cloud condensation nucleation as part of the Atmosphere-Aerosol-Boundary Layer-Cloud Interaction Joint 25 Experiment done in the summer of 2016 at Xingtai (XT), a suburban site located in 26 the center of the North China Plain (NCP). In general, the probability density function 27 (PDF) of the hygroscopicity parameter ( $\kappa$ ) for 40–200-nm particles had a unimodal 28 distribution, and mean  $\kappa$ -PDF patterns for different sizes were similar, suggesting that 29 the particles were highly aged and internally mixed because of strong photochemical 30 reactions. The  $\kappa$  calculated from the hygroscopic growth factor in the daytime and at 31 night suggests that photochemical reactions largely enhanced the aerosol 32 hygroscopicity. This effect became weaker as the particle size increased. In addition, 33 the aerosol hygroscopicity was much larger at XT than at other sites in the NCP. This 34 is because new particle formation takes place much more frequently in the central 35 NCP, which is heavily polluted from industrial activities, than elsewhere in the region. 36 37 The evolution of the planetary boundary layer played a dominant role in dictating aerosol mass concentration. Particle size was the most important factor influencing 38 the ability of aerosols to activate, whereas the effect of chemical composition was 39 secondary, especially when supersaturation was high. Using a fixed value of  $\kappa$  = 40 0.31 to calculate the cloud condensation nuclei number concentration in this region 41 suffices. 42

## 43 **1. Introduction**

Aerosols, defined as the mixture of solid and liquid particles suspended in air, are 44 ubiquitous in the atmosphere because of direct emissions from biogenic and 45 anthropogenic sources, and the secondary transformation from gas precursors. Aerosol 46 particles play an important role in climate change through direct and indirect effects 47 (e.g., Ramanathan et al., 2001; Rosenfeld et al., 2008; Li et al., 2016). However, the 48 impact of aerosols on climate change is difficult to simulate because of the highly 49 variable physical and chemical properties of aerosols, and complex aerosol-cloud 50 interactions (IPCC, 2013; Lebo et al., 2017). 51

The hygroscopic growth and mixing state of aerosol particles are important for 52 estimating the direct radiative effect of aerosols on Earth's climate. This is because 53 54 the growth and mixing can change the particle size and optical properties of aerosol 55 particles which influences the solar radiation budget and atmospheric visibility. In addition, aerosol particles can be activated as cloud condensation nuclei (CCN) under 56 supersaturation (SS) conditions. The variability in CCN number concentration ( $N_{\text{CCN}}$ ) 57 can modify both cloud microphysical properties (Twomey, 1974; Albrecht, 1989) and 58 morphology (Rosenfeld et al., 2008) and can lead to a broad impact on a wide range 59 60 of meteorological variables including severe weather events (Li et al., 2017a).

61 Previous studies have addressed three main aerosol properties influencing the 62 CCN activation, namely, particle size, chemical composition, and mixing state. 63 However, their relative importance is different under different environmental

conditions (e.g., Dusek et al., 2006; Ervens et al., 2007; Cubison et al., 2008; Deng et
al., 2011; Zhang et al., 2014; Schmale et al., 2018).

66 Ambient aerosols are composed of different species, including inorganic ions, organic components, black carbon (BC), and mineral dust. Inorganics mainly contain 67 sulfate, nitrate, and ammonium, while organic aerosols (OA) consist of thousands of 68 chemicals (Jacobson et al., 2000). The hygroscopicity and CCN activity of a single 69 component can be characterized according to laboratory studies (e.g., Petters and 70 Kreidenweis, 2007), but the properties of their mixtures are hard to estimate because 71 72 of the different chemical species and mixing states of particles in the atmosphere. Therefore, aerosol hygroscopicity and CCN activity are very different in different 73 regions due to different chemical compositions. Comprehensive field measurements 74 75 of aerosol properties in different regions are thus necessary to improve models.

76 China, especially the North China Plain (NCP), has been suffering from severe air pollution over the last couple of decades due to rapid industrialization and 77 urbanization. Diverse sources and aging processes make aerosol properties 78 particularly diverse and complex in this part of the world. As such, the region has 79 drawn much attention regarding the aerosol mixing state, hygroscopicity, and CCN 80 activity (Deng et al., 2011; Liu et al., 2011; Zhang et al., 2014; F. Zhang et al., 2016; 81 S.L. Zhang et al., 2016; Wu et al., 2016; Y. Wang et al., 2017). Liu et al. (2011) and 82 Y. Wang et al. (2017) have suggested that ambient particles are mostly an external 83 mixture with different hygroscopicities. Deng et al. (2011) have shown that the 84 aerosol number size distribution is critical in the prediction of  $N_{\rm CCN}$  while Zhang et al. 85

(2014, 2017) have highlighted the importance of chemical composition in determining 86 particle activation properties. However, these studies were done using data from the 87 88 northern part of the NCP. Few studies have focused on the central region of the NCP. Compared to the northern part of the NCP, the central part of the NCP is more affected 89 by industrial emissions because a dense cluster of China's heavy industries exists 90 there (Fu et al., 2014). Measurements of aerosol properties in the central part of the 91 NCP are thus critically needed to investigate the impact of air pollution on the 92 environment and climate change. 93

Xingtai (XT), a city located in the center of the NCP, is considered one of the
most polluted cities in China. Local industrial and domestic sources of pollution are
the greatest contributors to severe haze events in that region (Wang et al., 2014). A
field experiment called the Atmosphere-Aerosol-Boundary Layer-Cloud (A<sup>2</sup>BC)
Interaction Joint Experiment was done at a suburban site in XT in the summer of 2016.
Differences in aerosol properties at this site and at sites in the northern part of the
NCP were found.

101 The paper is organized as follows. Sections 2 and 3 describe the measurement 102 method and data analysis theory. Section 4 presents and discusses the measurement 103 results, which includes data time series, aerosol mixing state, hygroscopicity, CCN 104 prediction and its sensitivity to chemical composition. A summary and conclusions are 105 given in section 5.

#### 106 2. Measurements

## 107 2.1. Sampling site and meteorology

The A<sup>2</sup>BC experiment was done at the National Meteorological Basic Station 108 located in XT (37.18°N, 114.37°E, 180 m above sea level) from 1 May to 15 June of 109 2016. This suburban site is situated ~17 km northwest of the XT urban area in 110 southern Heibei Province, which is located in the central part of the NCP and to the 111 east of the Taihang Mountains (Fig. 1a). This region is heavily populated, urbanized, 112 and industrialized. Major industrial manufacturers include coal-based power plants, 113 steel and iron works, glassworks, and cement mills. Weak diffusion conditions and 114 heavy industrial emissions lead to exceptionally high concentrations of particulate 115 116 matter (PM) with diameters less than 10 µm and 2.5 µm (PM<sub>2.5</sub>), as well as gas pollutants such as sulfur dioxide (SO<sub>2</sub>), volatile organic compounds (VOCs), and 117 nitrogen oxides (NO<sub>x</sub>) during the frequently occurring haze episodes in this region 118 (Wang et al., 2014; Fu et al., 2014). Figure 1b shows the mean distribution of SO<sub>2</sub> 119 concentrations from May of 2012 to 2016 which confirms that the measurement site is 120 located in one of the pollution centers in this region. A detailed analysis of gas 121 122 precursors and aerosol chemical species shows that this station is a representative site in this region (Zhang et al., 2018). 123

Time series of meteorological variables measured at this meteorological station are shown in Fig. S1. This site is strongly affected by mountain-valley winds. Southeasterly winds prevail during the day and at night northwesterly winds prevail (Fig. S1 and Fig. S2). There was almost no precipitation during the study period. The
ambient temperature (*T*) and relative humidity (RH) time series show opposing trends.
Campaign-mean values of *T* and RH are 21.9°C and 51.6%, respectively.

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## 2.2. Instrumentation and operation

## 131 **2.2.1.** Aerosol hygroscopicity measurements

The custom-built hygroscopicity tandem differential mobility analyzer (HTDMA) 132 used in this study has been described in detail by others (Tan et al., 2013; Y. Wang et 133 al., 2017). Briefly, ambient aerosols are first dried and neutralized by a Nafion dryer 134 and a soft X-ray charger. A differential mobility analyzer (DMA<sub>1</sub>, model 3081L, TSI 135 Inc.) is used to select monodispersed particles of a certain diameter. The 136 monodispersed particles are then passed through a Nafion humidifier with a controlled 137 higher RH and are humidified. A second DMA (DMA<sub>2</sub>, same model as the DMA<sub>1</sub>) 138 and a water-based condensation particle counter (WCPC, model 3787, TSI Inc.) are 139 used to measure the number size distribution of the humidified particles. The DMA<sub>1</sub> 140 and WCPC can also be connected directly to measure the 10-400-nm particle number 141 size distribution (PNSD). In this study, the dry diameters selected by the DMA<sub>1</sub> were 142 40, 80, 110, 150, and 200 nm, and the humidified RH was set to 85%. The RH 143 calibration with ammonium sulfate for the H-TDMA is shown in Fig. S3. 144

145 The hygroscopic growth factor (GF) is defined as the ratio of the humidified146 diameter at a given RH to the dry diameter:

147 
$$GF = \frac{D_p(RH)}{D_{po}},$$
 (1)

where  $D_{\rm p}(\rm RH)$  is the particle diameter at the given RH and  $D_{\rm p0}$  is the dry diameter selected by the DMA<sub>1</sub>. The measured distribution function versus GF can be calculated with WCPC data downstream from the DMA<sub>1</sub> and DMA<sub>2</sub>. The GF probability density function is then retrieved using the TDMAFIT algorithm (Stolzenburg and McMurry, 1988, 2008).

#### 153 **2.2.2.** Aerosol chemical composition measurements

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An Aerosol Chemical Speciation Monitor (ACSM) was used to measure 154 non-refractory submicron aerosol species (sulfate, nitrate, ammonium, chloride, and 155 organics) in real-time. A PM2.5 URG cyclone (model URG-2000-30ED) was installed 156 in the front of the sampling inlet to remove coarse particles (> 2.5  $\mu$ m in diameter). 157 Before sampling into the ACSM, aerosol particles were dried (below 40% RH) by a 158 silica gel diffusion dryer. The ACSM was calibrated routinely with pure ammonium 159 nitrate to determine its ionization efficiency. More detailed descriptions about the 160 ACSM are given by Ng et al. (2011) and Sun et al. (2012). A positive matrix factor 161 analysis was used to analyze the organic spectral matrices according to Ulbrich et al. 162 (2009). Three factors, i.e., hydrocarbon-like OA (HOA), cooking OA (COA), and 163 oxygenated OA (OOA), are chosen as the ACSM dataset. HOA and COA are both 164 primary organic aerosols (POA) while OOA is the secondary organic aerosol (SOA). 165 166 The ACSM does not detect refractory material such as BC, so a seven-wavelength

168  $\mu$ m (PM<sub>1</sub>) cyclone was used to measure the BC mass concentration of BC particles

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aethalometer (AE-33, Magee Scientific Corp.) with a PM with diameters less than 1

with diameters < 1.0 μm. Mineral dust and sea salt are the other refractory species, but</li>
they typically exist in the coarse mode and contribute negligibly to PM<sub>1</sub> (Juranyi et al.,
2010; Meng et al., 2014).

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## 2.2.3. Aerosol size distribution and CCN measurements

173 The aerosol particle number size distribution (15-685 nm) was measured by a scanning mobility particle sizer (SMPS) that was equipped with a long DMA (model 174 3081L, TSI Inc.) and a condensation particle counter (model 3775, TSI Inc.). A 175 single-column continuous-flow thermal-gradient cloud condensation nuclei counter 176 (model CCNC-100, DMT Inc.) was used to measure the bulk CCN number 177 concentration. Five SS levels, i.e., 0.07, 0.1, 0.2, 0.4, and 0.8%, were set in the CCNC 178 and the running time was 10 min for each SS level. The SS levels in the CCNC were 179 180 calibrated with pure ammonium sulfate (Rose et al., 2008) before and after the measurement campaign. The corrected SS levels were 0.11, 0.13, 0.22, 0.40, and 181 0.75%. 182

The aerosol activation ratio (AR) at a certain SS is calculated as  $N_{\text{CCN}}$  divided by the total particle number concentration in the 15–685-nm range ( $N_{15-685 \text{ nm}}$ ), i.e., AR =  $N_{\text{CCN}}/N_{15-685 \text{ nm}}$ . The SMPS does not measure particle number concentrations below 15 nm. Since the activation critical diameter is always larger than 15 nm at these SS levels (Zhang et al., 2014), this does not affect the calculated  $N_{\text{CCN}}$ . Aerosol particles with diameters greater than 685 nm are also not detected by the SMPS. These larger particles will always act as CCN due to their larger dry sizes. Note that the number concentration above 685 nm in the atmosphere is always negligible (Juranyi et al.,2010).

192 2.2.4. Other measurements

193 In this study, a micro-pulse lidar (MPL-4B, Sigmaspace Corp.) was used to study the evolution of the planetary boundary layer (PBL) which plays a crucial role in 194 modulating surface air quality (Z. Li et al., 2017b). The pulse repetition rate of the 195 MPL was 2.5 kHz at a visible wavelength of 532 nm. The peak value of the optical 196 energy of the laser beam was 8 µJ. The pulse duration ranged from 10 to 100 ns, and 197 the pulse interval was set to 200 ns, corresponding to a spatial resolution of 30 m. The 198 MPL-retrieved PBL height is the altitude where a sudden decrease in the scattering 199 coefficient occurs (Brooks, 2003; Quan et al., 2013). Trace gas analyzers 200 201 (manufactured by ECOTECH) were used to measure the gaseous species of ozone, SO<sub>2</sub>, NO<sub>x</sub>, NO, and carbon monoxide. More detailed descriptions about the analyzers 202 are given by Zhu et al. (2016). 203

Two containers at ground level housed all sampling instruments. Two air conditioners maintained the temperature at  $20-25^{\circ}$ C inside the containers. All stainless tube inlets were ~1.5 m above the top of the containers.

207 **3.** Theory

#### 208 3.1. Hygroscopicity parameter

209 To link hygroscopicity measurements below and above the water vapor saturation,

210 the Köhler theory (Köhler, 1936) is parameterized using the hygroscopicity parameter 211  $\kappa$  (Petters and Kreidenweis, 2007). This is known as the  $\kappa$ -Köhler theory. According 212 to the theory, the equilibrium equation for a solution droplet at a saturation ratio S(D)213 is

214 
$$S(D) = \frac{D^3 - D_d^3}{D^3 - D_d^3 (1 - \kappa)} \exp\left(\frac{4\sigma_{s/a} M_w}{RT \rho_w D}\right) , \qquad (2)$$

where *D* and  $D_d$  are the wet and dry droplet diameters, respectively,  $\sigma_{s/a}$  is the surface tension coefficient,  $M_w$  is the mole mass of water, *R* is the universal gas constant, *T* is the temperature, and  $\rho_w$  is the density of water.

Below the water vapor saturation, S(D) is RH, D is  $D_p(RH)$ , and  $D_d$  is  $D_{p0}$ from Eq. (1). The  $\kappa$  parameter is then calculated using H-TDMA data according to Eq. (1) and Eq. (2):

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$$\kappa_{\rm gf} = (\rm GF^3 - 1) \cdot \left[\frac{1}{\rm RH} \exp\left(\frac{4\sigma_{\rm s/a}M_{\rm w}}{RT\rho_{\rm w}D_{\rm d}\rm GF}\right) - 1\right] \quad . \tag{3}$$

For a multicomponent particle, the Zdanovskii–Stokes–Robinson (ZSR) mixing rule (Stokes and Robinson, 1966) can also estimate  $\kappa$  using chemical composition data:

225  $\kappa_{\rm chem} = \sum_i \varepsilon_i \kappa_i, \tag{4}$ 

where  $\varepsilon_i$  and  $\kappa_i$  are the volume fraction and  $\kappa$  for the *i*th chemical component, respectively. The ACSM provides the mass concentrations of inorganic ions and organics. A simplified ion-pairing scheme such as that described by Gysel et al. (2007) is applied to convert ion mass concentrations to mass concentrations of their corresponding inorganic salts (see Table S1 in the supplement). Table S1 also lists  $\kappa$ and the gravimetric density of each individual component under supersaturated conditions. In the following discussions,  $\kappa_{gf}$  and  $\kappa_{chem}$  denote the hygroscopicity parameters derived from H-TDMA measurements and estimated using the ZSR mixing rule, respectively.

## 235 **3.2.** CCN estimation

The critical supersaturation ( $s_c$ ,  $s_c = S_c-1$ ) for the  $D_d$  of a particle with 236 237 hygroscopicity  $\kappa$  is calculated from the maximum of the  $\kappa$ -Köhler curve (Eq. 2; Petters and Kreidenweis, 2007). The  $D_d$  is also the critical diameter corresponding to 238 the  $s_c$  when  $\kappa$  is known. The  $s_c$ - $D_d$  relationship can thus be established. According to 239 this relationship, the critical diameter  $(D_{0,crit})$  can be calculated using the estimated 240  $\kappa_{\rm chem}$  (Eq. 4) at a given SS. All particles larger than  $D_{0,{\rm crit}}$  will activate as CCN, 241 assuming that aerosols are internally mixed. Then the CCN number concentration can 242 be estimated from the integral of the aerosol size distribution provided by the SMPS 243 244 from  $D_{0,crit}$  to the maximum measured size  $(D_{max})$ :

$$N_{\rm CCN}(\rm SS) = \int_{D_{0,\rm crit}(\rm SS)}^{D_{\rm max}} \frac{dN(D)}{d\log(D)} d\log(D) \ . \tag{5}$$

 $N_{\rm CCN}(\rm SS)$  can then be compared to the number of CCN at the same SS measured by the CCNC.

#### 248 **4. Results and discussion**

#### 249 **4.1. Overview**

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Figures 2 and 3 show time series of the main aerosol properties measured during the field campaign. The PNSD changes dramatically (Fig. 2a) and the aerosol number

concentration in the 15–50 nm range ( $N_{15-50 \text{ nm}}$ ) increases sharply in the morning 252 almost every day (Fig. 2b). The time series of the mean diameter  $(D_m)$  of particles also 253 254 shows that a growth process occurs after the sharp increase in  $N_{15-50 \text{ nm}}$ . All these 255 phenomena suggest that new particle formation (NPF) events frequently occurred at 256 XT during the field experiment (Kulmala et al., 2012; Y. Li et al., 2017). This is likely related to the high concentration of gas precursors from mainly local emissions. High 257 emissions of SO<sub>2</sub> and VOCs associated with the high oxidation capacity in a polluted 258 atmosphere make NPF events occur more frequently in northern China (Z. Wang et al., 259 260 2017).

Figure 2c-d shows time series of the probability density functions (PDFs) of  $\kappa_{gf}$ ( $\kappa$ -PDF) for 40-nm and 150-nm particles, respectively. In general, mono-modal  $\kappa$ -PDFs were observed. This is different from  $\kappa$ -PDFs at other sites in China where bi- and tri-modal distributions dominate (Liu et al., 2011; Ye et al., 2013; Jiang et al., 2016; S. L. Zhang et al., 2016; Y. Wang et al., 2017). Differences in the aerosol mixing state explain this (see section 4.2).

Figure 3a shows the bulk mass concentrations of organics, sulfate, nitrate, ammonium, and chloride measured by the ACSM and the BC mass concentration measured by the AE-33. Organics and sulfate were the dominant chemical species with mass fractions in PM<sub>1</sub> of 39.1% and 24.7%, respectively. Figure 3b-c shows the volume fractions of paired chemical compositions and  $\kappa_{chem}$ , respectively. The average volume fraction of inorganics ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>+NH<sub>4</sub>HSO<sub>4</sub>+H<sub>2</sub>SO<sub>4</sub>+NH<sub>4</sub>NO<sub>4</sub>) was similar to that of organics (POA+SOA), but their volume fractions changed diurnally. In 274 general, the volume fraction of inorganics increased during daytime while the volume 275 fraction of organics decreased. SOA was the dominant contributor to OA, accounting 276 for ~69% of the organics volume. This shows that photochemical reactions were 277 strong at XT during the field campaign (Huang et al., 2014). The mean  $\kappa_{chem}$  in Fig. 278 3c was 0.31 with values ranging from 0.20 to 0.40. The trend in  $\kappa_{chem}$  was similar to 279 that of the volume fraction of inorganics. This suggests that inorganics played a key 280 role in  $\kappa_{chem}$ . This is consistent with the study by Wu et al. (2016).

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## 4.2. Aerosol mixing state and hygroscopicity

Figure 4 shows mean  $\kappa$ -PDFs for different particle sizes derived from H-TDMA 282 data. For all particle sizes considered,  $\kappa_{gf}$  ranged from 0 to 0.8, and the  $\kappa$ -PDF 283 patterns were similar. This suggests that hygroscopic compounds in different particle 284 285 size modes were similar at XT. In general,  $\kappa$ -PDF patterns show only one hydrophilic mode with a weak hydrophobic mode occasionally appearing at night when 286 photochemical reactions are weak (Fig. S4). The *k*-PDF patterns always show bi- or 287 tri-modal distributions at other sites in China (Liu et al., 2011; Ye et al., 2013; Jiang et 288 al., 2016; Zhang et al., 2016; Y. Wang et al., 2017). Based on previous studies (Liu et 289 al., 2011; Y. Wang et al., 2017), ambient aerosols can be classified into three groups 290 291 according to their  $\kappa_{gf}$  values:

292 — nearly hydrophobic (NH): 
$$\kappa_{gf} < 0.1$$

293 — less hygroscopic (LH): 
$$0.1 \leq \kappa_{gf} < 0.2$$

294 — more hygroscopic (MH):  $0.2 \leq \kappa_{gf}$ 

Table 1 gives the number fractions of each group for different particle sizes. The MH 295 group dominated all particle sizes. The number fractions of the NH and LH groups 296 297 were both less than 6.0%. However, the volume fractions of hydrophobic BC and low-hygroscopic organics (where  $\kappa_{BC}$  is approximately zero and  $\kappa_{organic}$  is 298 typically less than 0.1) were  $\sim 10.1\%$  and 47.4%, respectively, according to chemical 299 composition measurements (Fig. 3b). This suggests that the particles were highly aged 300 and internally mixed at XT during the field campaign. The coating of sulfates and 301 302 secondary organics during the aging process changes the structure of BC and makes 303 these particles grow which can significantly enhance the hygroscopicities of particles (e.g., Zhang et al., 2008; Jimenez et al., 2009; Tritscher et al., 2011; Guo et al., 2016). 304 The observed unimodal distribution of  $\kappa$ -PDF also suggests the internal mixing state 305 306 of the particles (Swietlicki et al., 2008).

Figure 5 shows the average size-resolved  $\kappa_{gf}$  derived from H-TDMA data at XT and other sites in China. At XT,  $\kappa_{gf}$  for different particle sizes were larger in the daytime than at night, and the difference between daytime and nighttime decreased with increasing particle size. This suggests that the impact of photochemical reactions on aerosol hygroscopicity is strong. The effect is weaker with increasing particle size because most of the larger particles are always well aged.

The magnitude of  $\kappa_{gf}$  was larger at XT than at other sites in China. In particular, the magnitude of  $\kappa_{gf}$  was much larger at XT than at sites in the northern part of the NCP, i.e., Beijing, Wuqing, and Xianghe. The lower  $\kappa_{gf}$  in the Beijing urban area is likely related to the more severe traffic emissions there (Ye et al., 2013; Wu et al.,

2016). Wuqing and Xianghe are located in the suburban area between the two 317 megacities of Beijing and Tianjin and are simultaneously affected by traffic and 318 319 industrial emissions. The magnitudes of  $\kappa_{gf}$  at these two sites are higher than at Beijing but lower than at XT. Although distant from these megacities, XT is situated 320 321 in the industrial center of the NCP, so particles there are more internally mixed and 322 highly aged due to the higher concentrations of precursors and strong photochemical reactions. This is why  $\kappa_{gf}$  at XT is larger than at other sites. This suggests that the 323 324 hygroscopicities of particles from different emissions and chemical processes differ in 325 the NCP. Forty nm particles were always more hygroscopic than 80-nm particles at XT, especially in the daytime. This differed from other sites likely because the coating 326 effect of sulfates and secondary organics is more significant for smaller particles 327 328 (Tritscher et al., 2011; Guo et al., 2016). Furthermore, since the field measurements took place in a locality with heavy industrial activities, it is possible that amine 329 contributed significantly to the hygroscopicity of 40-nm particles. Several studies 330 331 have shown that amine compounds in aerosol phase can be hygroscopic, sometimes at 332 even low RH (e.g., Qiu and Zhang, 2012; Chu et al., 2015; Gomez-Hernandez et al., 2016). 333

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## 4.3. Diurnal variations in aerosol properties

4.3.1. Diurnal variations in aerosol number and mass concentrations

Figure 6a shows the diurnal variation in MPL-derived PBL height. The PBL height is the altitude where a sudden decrease in the MPL-measured scattering

coefficient occurs (Cohn and Angevine, 2000; Brooks, 2003). Note that the retrieved 338 PBL height is only valid from 0700 local time (LT) to 1900 LT (Quan et al., 2013). 339 340 The retrieved PBL height at night is not accurate because of the likely influence of residual aerosols within the nocturnal PBL. The evolution of PBL height from 0700 341 342 LT to 1900 LT is sufficient to analyze its link with the change in aerosol number and mass concentrations during the daytime. Figure 6b shows diurnal variations in aerosol 343 number and mass concentrations in the 15–685 nm range ( $N_{15-685 \text{ nm}}$  and  $PM_{15-685 \text{ nm}}$ , 344 respectively). Variations in  $N_{15-685 \text{ nm}}$  and  $PM_{15-685 \text{ nm}}$  trended opposite from each other. 345 346 From 0800 LT to 1400 LT, the PBL height lifted from ~0.5 km to ~0.6 km, while  $PM_{15-685\ nm}$  generally decreased from ~24  $\mu g\ m^{-3}$  to ~19  $\mu g\ m^{-3}.$  This suggests the 347 important effect of PBL evolution on PM<sub>15-685 nm</sub>. However, N<sub>15-685 nm</sub> sharply 348 increased from ~7600 cm<sup>-3</sup> at 0700 LT to ~13,000 cm<sup>-3</sup> at 1300 LT. This is related to 349 the sudden burst of small Aitken-mode particles (< 50 nm) during NPF events. Newly 350 formed fine particles contribute little to PM15-685 nm. In the evening, PM15-685 nm 351 352 increased gradually while  $N_{15-685 \text{ nm}}$  decreased. The decline of the nocturnal PBL and particle coagulation and growth explains this. In other words, the evolution of the 353 PBL influenced the aerosol mass concentration, while particle formation and growth 354 had a greater influence on the variation in aerosol number concentration. 355

## 356 **4.3.2.** Diurnal variation in aerosol hygroscopicity

Figure 6c shows diurnal variations in  $\kappa_{gf}$  and  $\kappa_{chem}$ . Values of  $\kappa_{gf}$  for different particle sizes increased in the morning when the NPF event started. The increase was 359 sharpest for 40-nm particles. The increase in  $\kappa_{gf}$  in the morning synchronized with the particle number concentration ( $N_{15-685 \text{ nm}}$ ) but not with the PBL height, further 360 361 suggesting the impact of photochemical reactions on aerosol hygroscopicity. The  $\kappa_{gf}$ for 40-nm particles increased from ~0.32 at 0700 LT to ~0.44 at 1500 LT and 362 approached the  $\kappa$  value of pure ammonium sulfate. This also suggests that a large 363 amount of hygroscopic compounds were produced during NPF events. Fig. S5 shows 364 sharply increased concentrations of SO2 and VOCs in the morning and the enhanced 365 atmospheric oxidation capacity under high RH and low T conditions. The production 366 367 of sulfate and SOAs resulted. This is why aerosol hygroscopicity and the occurrence of NPF events increased. Zhang et al. (2018) characterized the aerosol chemistry 368 during NPF events in this field campaign. The diurnal pattern in  $\kappa_{\rm gf}$  for 80–200 nm 369 370 particles differs from that of 40-nm particles. The differences in  $\kappa_{gf}$  for 80–200 nm particles in the early morning were large but gradually decreased as the sun rose. The 371  $\kappa_{gf}$  for 80–200-nm particles were similar but lower than that for 40-nm particles after 372 373 1100 LT. The condensation of sulfates and secondary organics likely caused the enhanced hygroscopicity of the 40-200-nm particles, especially of 40-nm particles 374 (Fig. 6d). 375

Figure 6c also shows that the  $\kappa_{chem}$  for PM<sub>1</sub> was lower than the  $\kappa_{gf}$  for 40– 200-nm particles and had a weaker diurnal variation. This feature was stronger at noon when atmospheric oxidation and the aging process were more rapid. The simple ZSR mixing rule is responsible for this. During the daytime, the condensation of sulfuric acid on organics or BC greatly enhances their hygroscopicities (Zhang et al., 2008; Zhang et al., 2017). The ZSR model cannot accurately represent this phenomenon. Cruz and Pandis (2000) have shown that the measured  $\kappa_{gf}$  of internally mixed (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>-organic aerosols is larger than the predicted  $\kappa_{chem}$  based on the ZSR model.

In summary, the ample supply of SO<sub>2</sub> and VOCs provided sufficient precursors for the strong photochemical reactions at XT during this field campaign, and the production and condensation of sulfate and SOAs greatly enhanced aerosol hygroscopicity, especially during the daytime. The oxidation of precursors likely induced the observed frequent NPF events.

## **4.3.3.** Diurnal variation in CCN number concentration and activation ratio

Figure 7a shows the diurnal variations in N<sub>CCN</sub> and AR at different SS. In the 391 morning, N<sub>CCN</sub> first decreased then increased while AR showed the opposite trend. 392 This is related to the evolution of the PBL and NPF events. At the initial stage of an 393 NPF event, the newly formed particles were less than 15 nm in size which was below 394 the detection limit of the SMPS. As a result,  $N_{15-685 \text{ nm}}$  decreased (Fig. 6b) as the PBL 395 lifted, and  $N_{\rm CCN}$  also decreased. However, the mixing of aged particles within the PBL 396 made the particle size (Fig. 7b) and AR increase slightly. Condensation and the 397 growth of new particles caused the number of fine particles detected by the SMPS to 398 399 increase rapidly. However, because of their smaller sizes, some of these particles were not activated. Therefore, N<sub>CCN</sub> increased, but AR decreased from 0800 LT to 1400 LT. 400 In the afternoon and evening,  $N_{\rm CCN}$  and AR increased slightly as particle sizes 401

402 increased (Fig. 7b). These trends weakened as SS decreased because the critical 403 diameter is larger at low SS and the influence of aerosol size distribution on  $N_{\rm CCN}$  and 404 AR is relatively weaker. Particle size was the most important factor influencing 405 aerosol activation and CCN number concentrations, especially at larger SS. Figure 6S 406 shows the results from a sensitivity test of particle size in a CCN closure study similar 407 to that done by Dusek et al. (2006).

#### 408 **4.4.** CCN estimation from chemical composition data

This section presents a CCN closure study and a discussion of the impact of 409 chemical composition on  $N_{\rm CCN}$ . It is reasonable to assume that aerosols are internally 410 411 mixed when estimating  $N_{\rm CCN}$  because H-TDMA data showed that this was the case at XT. Figure 8a shows estimated  $N_{\rm CCN}$  as a function of measured  $N_{\rm CCN}$  using real-time 412  $\kappa_{chem}$ . The estimated N<sub>CCN</sub> correlates well with measurements ( $R^2 \ge 0.85$ ), but is 413 generally overestimated. The slope of each linearly fitted line is greater than 1.10 and 414 increases with increasing SS. The relative deviation (RD) increases from 16.2% to 415 25.2% as SS increases from 0.13% to 0.75%, suggesting that estimates become worse 416 at larger SS. The large measurement uncertainties of CCNC mainly cause the 417 overestimation of  $N_{\text{CCN}}$ : (1) The temperature or high flow rates in the CCNC may not 418 allow enough time for particles to reach sizes large enough to be counted by the 419 420 optical particle counter at the exit of the CCN chamber (Lance et al., 2006; Cubison et al., 2008) and (2) in high particle number concentration environments, water depletion 421 in the CCNC may reduce the counting rate of the CCNC (Deng et al., 2011). These 422

423 uncertainties make measured  $N_{\rm CCN}$  lower than the actual  $N_{\rm CCN}$ . At larger SS, activated 424 aerosols in the cloud chamber of the CCNC are greater in number and smaller in size, 425 so the impact of these uncertainties is greater. Figure S7 shows results from the  $N_{\rm CCN}$ 426 closure study for separated  $N_{\rm CCN}$ . The CCN closure is reasonable when  $N_{\rm CCN} < 5500$ 427 cm<sup>-3</sup>.

Figure 8b shows estimated  $N_{\text{CCN}}$  using the mean value for  $\kappa_{\text{chem}}$  ( $\kappa_{\text{chem}} = 0.31$ ). 428 429 Compared with results using real-time values for  $\kappa_{chem}$ , the fit parameters and RD change slightly, suggesting that the effect of chemical composition on  $N_{\rm CCN}$  is weaker 430 431 relative to the particle size. Figure 9 shows the sensitivity of estimated  $N_{\rm CCN}$  to the variability in chemical composition. The variability in the equipotential lines of RD 432 suggests that the sensitivity of  $N_{\rm CCN}$  is strongly time dependent. This is attributed to 433 434 the variability of the shape of the aerosol size distribution (Juranyi et al., 2010) which further demonstrates the importance of particle size to  $N_{\rm CCN}$ . The sensitivity of  $N_{\rm CCN}$ 435 to chemical composition ( $\kappa_{chem}$ ) becomes weaker with increasing SS, suggesting that 436 437 chemical composition becomes less important in  $N_{\rm CCN}$  estimates at larger SS. RD is always less than 10% when estimating  $N_{\rm CCN}$  using the mean value of  $\kappa_{\rm chem}$ . The 438 value  $\kappa = 0.31$  is thus a good reference value to model N<sub>CCN</sub> in this region. 439

In summary, the particle size is the most important factor influencing aerosol activation at XT, especially at larger SS. The chemical composition was not as important when estimating  $N_{\rm CCN}$  because particles were highly aged and internally mixed at XT. Aerosol hygroscopicity was not sensitive to estimates of  $N_{\rm CCN}$ .

#### 444 **5.** Summary and conclusions

The Atmosphere-Aerosol-Boundary Layer-Cloud (A<sup>2</sup>BC) Interaction Joint Experiment was done at a suburban site (Xingtai, or XT) located in the central North China Plain (NCP) from 1 May to 15 June of 2016. The study investigated aerosol hygroscopicity, the mixing state, and CCN activity at XT.

In general, the probability density function (PDF) of the hygroscopicity parameter 449  $\kappa$  ( $\kappa$ -PDF) for 40–200-nm particles was a unimodal distribution, which is different 450 from distributions at other sites in China. Particles of all sizes covered a large range of 451  $\kappa_{\rm gf}$  (the hygroscopicity parameter derived from H-TDMA measurements; mostly 452 from 0 to 0.8) and showed similar  $\kappa$ -PDF patterns, suggesting that the hygroscopic 453 compounds in these particles from 40 nm to 200 nm were similar at XT. The  $\kappa$ -PDF 454 455 patterns also suggest that particles were highly aged and internally mixed at XT 456 during the field campaign. This is likely related to strong photochemical reactions.

The mean  $\kappa_{gf}$  for different particle sizes were larger in the daytime than at night. 457 458 Daytime and nighttime  $\kappa_{gf}$  differences decreased with increasing particle size. The impact of photochemical reactions on aerosol hygroscopicity was strong, and the 459 effect became weaker as particle size increased. The coating of sulfates or secondary 460 organics likely enhanced the hygroscopicities of 40-200-nm particles. This effect was 461 more significant for 40-nm particles. Compared with other sites in China, the aerosol 462 hygroscopicity was much larger at XT because of the sufficient amount of precursors 463 strong atmospheric oxidation. The comparison also shows that the 464 and

465 hygroscopicities of particles from different emissions and chemical processes differed466 greatly.

New particle formation events occurred frequently at XT during this field 467 campaign. The evolution of the planetary boundary layer influenced the aerosol mass 468 concentration, while particle formation and growth had a greater influence on the 469 variation in aerosol number concentration. Particle size was the most important factor 470 influencing aerosol activation and the CCN number concentration (N<sub>CCN</sub>) at XT, 471 especially at larger supersaturations (SS). Although estimated  $N_{\rm CCN}$  correlated well 472 with measurements ( $\mathbb{R}^2 \ge 0.85$ ), N<sub>CCN</sub> was overestimated because of measurement 473 uncertainties. The effect of chemical composition on  $N_{\text{CCN}}$  was weaker relative to the 474 particle size. Sensitivity tests show that the impact of chemical composition on  $N_{\rm CCN}$ 475 476 became weaker as SS increased, suggesting that the effect of chemical composition on the estimation of  $N_{\text{CCN}}$  is less important at larger SS. The value  $\kappa = 0.31$  is a good 477 proxy for  $N_{\rm CCN}$  in this region. 478

479 XT is located in the most polluted region in China. The multitude of factories in the 480 region generates strong emissions. The plenitude of gas precursors and strong 481 photochemical reactions at XT make aerosol properties there unique. More field 482 measurements on gas-particle transformation and aerosol properties in this region are 483 needed for studying haze formation mechanisms and climate effects.

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485 *Data availability*. Data used in the study are available from the first author upon
486 request (wang.yuying@mail.bnu.edu.cn).

488 *Competing interests*. The authors declare that they have no conflict of interest.489

Author contribution. Z. L. and Y. W. designed the experiment; Y. W., Y. Z., and W. D.
carried it out and analyzed the data; other co-authors participated in science discussions
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749 **Table 1.** Number fractions of different hygroscopic groups for different particle sizes.

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		40 nm	80 nm	110 nm	150 nm	200 nm
	NH	5.1 %	5.0 %	5.1 %	5.0 %	5.7 %
	LH	4.8 %	4.2 %	4.3 %	4.7 %	5.1 %
	MH	90.1 %	90.8 %	90.6 %	90.3 %	89.2 %

750	NH: nearly hydrophobic; LH: less hygroscopic; MH: more hygroscopic
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Figure 1. (a) Map showing the location of the sampling site and (b) the distribution of

mean  $SO_2$  concentrations from May of 2012 to 2016.





Figure 2. The time series of (a) the particle number size distribution, (b) the aerosol number concentration in the 15–50 nm range ( $N_{15-50 \text{ nm}}$ ) and the geometric mean diameter ( $D_{\text{m}}$ ), (c) the probability density function of  $\kappa_{\text{gf}}$  for 40-nm and (d) 150-nm particles from 6 May to 15 June of 2016.



Figure 3. Time series of (a) the bulk mass concentration of aerosol species in PM<sub>1</sub>, (b) the volume fractions of POA, SOA, BC, and inorganics with the simplified ion-pairing scheme, and (c) the hygroscopicity parameter derived from the chemical composition ( $\kappa_{chem}$ ).





Figure 4. Mean probability density functions of  $\kappa_{gf}$  ( $\kappa$ -PDF) for different particle sizes and their standard deviations (shaded areas) derived from H-TDMA data and measured at RH = 85 %.



Figure 5. Size-resolved aerosol hygroscopicity parameter derived from H-TDMA
data at XT and at other sites in China.



Figure 6. Diurnal variations in (a) planetary boundary layer (PBL) height retrieved from micropulse lidar data, (b) aerosol number and mass concentrations in the 15–685 nm range ( $N_{15-685 \text{ nm}}$  and PM<sub>15-685 nm</sub>, respectively) derived from the SMPS (an aerosol density of 1.6 g cm<sup>-3</sup> is assumed), (c) the hygroscopicity parameter derived from the hygroscopic growth factor ( $\kappa_{gf}$ ) and predicted from the bulk chemical composition ( $\kappa_{chem}$ ), and (d) the mass fractions of different species.





Figure 7. Diurnal variations in (a) CCN number concentration  $(N_{\text{CCN}})$  and activation 

ratio (AR), and (b) the normalized aerosol size distribution in the 15-685-nm particle 

- size range.



Figure 8. Estimated versus measured cloud condensation nuclei (CCN) number concentrations (N<sub>CCN</sub>) for ambient aerosols at four different supersaturation (SS) levels. The  $N_{\rm CCN}$  is estimated based on  $\kappa$ -Köhler theory using the real-time  $\kappa_{\rm chem}$ (a1-a4) and the mean  $\kappa_{chem}$  (b1-b4). The slope and coefficient of determination (R<sup>2</sup>) 

of the linear regression and the relative deviation (RD) of estimated  $N_{\rm CCN}$  (RD =  $|N_{\rm CCN\_estimated} - N_{\rm CCN\_measured}| / N_{\rm CCN\_measured}$ ) are shown in each panel. The regression line is overlaid on the measurements (solid line) and the dashed line is the 1:1 line.

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Figure 9. Sensitivity of N<sub>CCN</sub> estimates to  $\kappa_{chem}$  as a function of time at (a) SS = 0.22% and (b) SS = 0.75%. The color scale indicates the relative deviation (RD) of CCN estimates using the  $\kappa_{chem}$  value shown on the ordinate. In each panel, open circles show the real-time  $\kappa_{chem}$ . Note that RD is by definition zero at these points. The black line is  $\kappa$  at RD = 10% and the red line is the mean value for  $\kappa_{chem}$  (0.31). Figure S8 in the supplement shows the same plots but for SS = 0.13% and 0.40%.