

## Reply to RC1

This study reported field measurements results of the chemical speciation, hygroscopicity and CCN properties of ambient particles at a suburban site in the central North China Plain (NCP). The probability density function of the hygroscopicity parameter kappa-PDF was then derived from measurement data and showed only a singular hygrophilic mode which was very different from profiles observed in other regions of the NCP (that were normally bi- or tri-modal). Among the possible factors affecting aerosol hygroscopicity (the mixing state, chemical composition and particle size), particle size was identified as the key factor influencing the particle CCN activation. This study explored the aerosol microphysical properties in a region that was not previously studied and the results can be useful when compared with existing data to understand aerosol aging and its impact on particle microphysics and the climate. The topic is relevant to the scope of the journal of Atmospheric Chemistry and Physics and should be considered for publication.

(1) The title may be changed to “Characterization of aerosol hygroscopicity, mixing state, and CCN activity at a suburban site in the central North China Plain” to reflect the unique location of this study.

Re: It's a good suggestion, the title has been revised. Very thanks.

(2) Section 4.1, Lines 317-321: While it's been shown that aging of BC will enhance its hygroscopicity and CCN activation, the actual determination of the GF of aged BC could be challenging (see, for example, Torsten et al., Environ. Res. Lett., 2011) as the DMA mobility size change may be marginal. A few chambers studies on soot SOA from anthropogenic VOCs may provide some insights here (Guo et al., ES&T, 2016 and Qiu et al., ES&T, 2012). In general, the knowledge on particle morphology is useful, and in principle, ACSM and DMA data can be combine to retrieve morphology information.

Re: Thanks for your suggestion and the recommended references. We read the recommended papers carefully and indeed found a more comprehensive interpretation for the enhanced hygroscopicity. The lines have been revised as “This suggests that the particles were highly aged and internally mixed at XT during this campaign. Coating of sulfates and secondary organics during the aging process changes the structure of BC and makes it grow, which can significantly enhance the hygroscopicity of particles (e.g., Zhang et al., 2008; Jimenez et al., 2009; Tritscher et al., 2011; Guo et al., 2016).

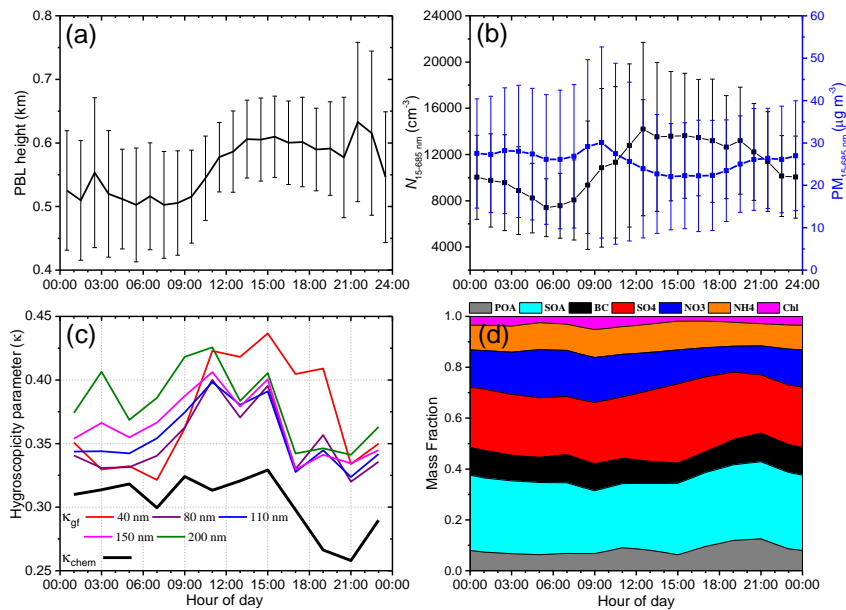
(3) Section 4.1, Lines 347-351: As pointed out by recent studies, amines may contribute significantly to the NPF events (e.g., Zhang et al., Chem Rev., 2015). Several studies have shown that amine compounds in aerosol phase can be hygroscopic, sometimes even at event low RH (e.g., Gomez-Hernandez et al., ES&T, 2016; Chu et al., PCCP, 2015; Qiu and Zhang, ES&T, 2013). Since the

44 reported field measurements took place in a local with heavy industrial activities,  
45 it is possible that amine may contribute significantly to the hygroscopicity of the  
46 40-nm particles

47 Re: This is a good point interpreting the hygroscopicity difference of 40 nm  
48 particles with other size particles. According to the recommended references, the  
49 lines have been revised as “40 nm particles were always more hygroscopic than 80  
50 nm particles at XT, especially in the daytime, which was also different from other  
51 sites. This is likely because the coating effect of sulfates and secondary organics is  
52 more significant on smaller particles (Tritscher et al., 2011; Guo et al., 2016).  
53 Furthermore, since the field measurements took place in a local with heavy  
54 industrial activities, it is possible that amine contributes significantly to the  
55 hygroscopicity of 40 nm particles. Several studies have shown that amine  
56 compounds in aerosol phase can be hygroscopic, sometimes at even low RH (Qiu  
57 et al., 2012; Chu et al., 2015; Gomez-Hernandez et al., 2016).”.

58  
59 (4) Section 4.3: It would make more sense to merge Figures 6&7 as the discussions  
60 on the two figures are closely related.

61 Re: It’s a good suggestion, we have merge them, as:



62  
63 As the reviewer’s suggestion, the merging make more sense. For example, we find the  
64 increase of hygroscopicity parameter ( $\kappa_{\text{gr}}$ ) in the morning was synchronous with the  
65 particle number concentration ( $N_{15-685 \text{ nm}}$ ), but was not with the PBL height, further  
66 suggesting the impact of photochemical reactions on aerosol hygroscopicity.  
67

68 (5) Section 4.4: It seems odd that kappa was not derived from CCN data as described  
69 by Petters and Kreidenweis (2007). A side-by-side comparison of kappa values

70 derived from HTDMA, chemical speciation and CCN may be more  
71 straightforward. Also, CCN derived kappa values can also provide basis for  
72 comparison with other studies that may only had CCN results.

73 **Re: Yes, we will obtain more information about aerosol hygroscopicity if kappa**  
74 **from CCN data can be derived. However, it needs to connect DMA and CCNc to**  
75 **measure the size-CCN number concentration. Unfortunately, we only measure the**  
76 **bulk CCN number concentration in the campaign. We will do the work in the**  
77 **future campaigns.**  
78

## 79 **References:**

80 Chu Y., Sauerwein M. and Chan C.K.: Hygroscopic and phase transition properties of alkyl aminium  
81 sulfates at low relative humidities, *Phys Chem Chem Phys*, 17, 19789-19796,  
82 <https://doi.org/10.1039/c5cp02404h>, 2015.

83 Gomez-Hernandez M., McKeown M., Secest J., Marrero-Ortiz W., Lavi A., Rudich Y., Collins D.R.  
84 and Zhang R.: Hygroscopic Characteristics of Alkylaminium Carboxylate Aerosols, *Environ Sci*  
85 *Technol*, 50, 2292-2300, <https://dx.doi.org/10.1021/acs.est.5b04691>, 2016.

86 Guo S., Hu M., Lin Y., Gomez-Hernandez M., Zamora M.L., Peng J., Collins D.R. and Zhang R.:  
87 OH-Initiated Oxidation of m-Xylene on Black Carbon Aging, *Environ Sci Technol*, 50, 8605-8612,  
88 <https://dx.doi.org/10.1021/acs.est.6b01272>, 2016.

89 Qiu C. and Zhang R.: Physicochemical Properties of Alkylaminium Sulfates: Hygroscopicity,  
90 Thermostability, and Density, *Environ Sci Technol*, 46, 4474-4480,  
91 <https://dx.doi.org/10.1021/es3004377>, 2012.

92 Tritscher T., Juranyi Z., Martin M., Chirico R., Gysel M., Heringa M.F., DeCarlo P.F., Sierau B.,  
93 Prevot A.S.H., Weingartner E. and Baltensperger U.: Changes of hygroscopicity and morphology  
94 during ageing of diesel soot, *Environ. Res. Lett.*, 6, <https://dx.doi.org/10.1088/1748-9326/6/3/034026>,  
95 2011.

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## Reply to RC2

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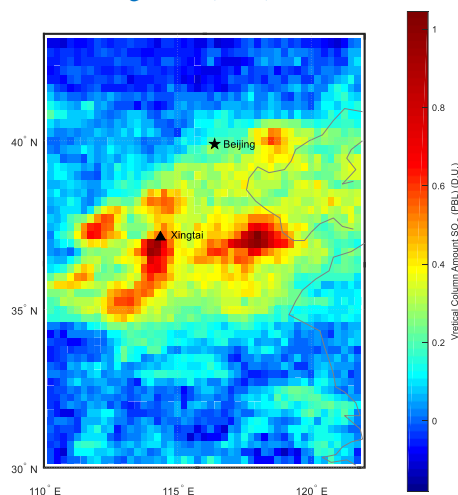
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Based on a field campaign conducted in Xingtai in the central North China Plain, the authors discussed about the chemical composition, hygroscopicity and CCN activity of aerosol particles. There have been a number of studies talked about the north part of NCP but very few about the central part. And it was found that aerosol mixing state and hygroscopicity in Xingtai largely differs from that in the north part of NCP. My major concerns are:

- 1) Xingtai locates at the western boundary of NCP and is heavily affected by the mountain-valley wind (L126). Can the measurement well represent the background aerosol in the central NCP?

Re: The influence of mountain-valley wind is regional. Figure 1 in this reply shows pollution in the NCP is rather non-uniform, the central region is more serious than that in northern and southern regions. Our sampling site (Xingtai) is located in one of the pollution centers and thus represents the condition near an emission source region. The average  $PM_{2.5}$  mass concentration at this station was  $45.2 \mu\text{g m}^{-3}$  in this campaign, which was only 15% lower than that ( $53.3 \mu\text{g m}^{-3}$ ) measured at the urban site in Xingtai. In addition, OOA and sulfate were the most important organic and inorganic species respectively. Figure 2 in this reply shows that the diurnal cycle of OOA was flat and the diurnal cycle of sulfate was also smoother compared with nitrate, also reflecting the regional characteristics of the main pollutants. All these suggest that the influence of mountain-valley wind is limited and this site is a good representative in this region. Detailed discussion about gas precursors and aerosol chemical species in this region can be found in Zhang et al., (2018).

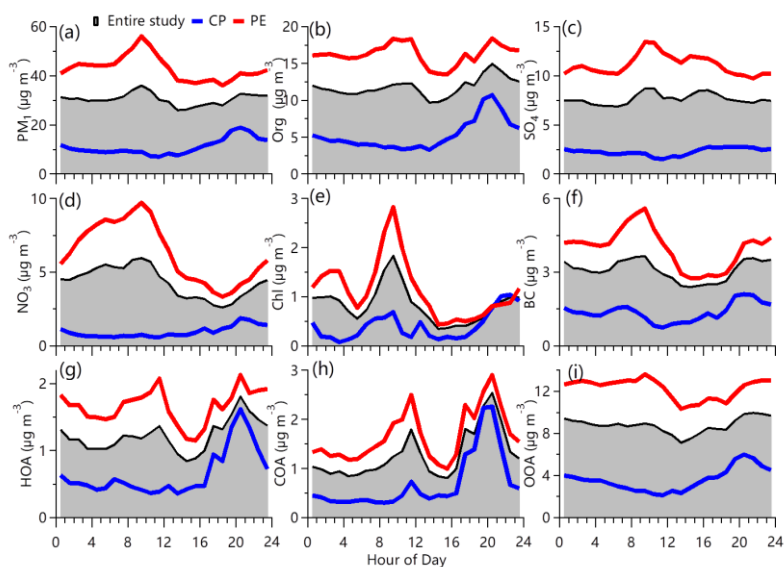


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Figure 1. The distribution of mean  $SO_2$  concentrations of May from 2012 to 2016. The map of NCP can refer to Fig. 4 in this reply.



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134 Figure 2. Diurnal cycles of chemical species of  $PM_{10}$  and OA factors during entire  
 135 study, PE (polluted events) and CP (clean periods). The figure is from Zhang et al.,  
 136 (2018).  
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138 2) I did not find a strong connection between the sections of CCN (section 4.3.3 and  
 139 4.4) and HTDMA (section 4.1 to 4.3.2). They look like two independent works  
 140 but each of them a too weak to be an individual study.

141 Re: Aerosol mixing state is one of three factors influencing aerosol activation ability.  
 142 In this campaign, we only measured bulk CCN data rather than size-resolved CCN  
 143 data. The bulk CCN data cannot provide the information about aerosol mixing state  
 144 but HTDMA data can. HTDMA data in this campaign suggests the aerosol is highly  
 145 internal-mixed and aged, which is the base to do CCN closure studies in this paper. In  
 146 addition, the HTDMA data are very useful for the analysis of the activation rate  
 147 diurnal variations. These are important links between the sections of HTDMA and  
 148 CCN.

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150 Specific comments:

151 L51: defined as the mixture of solid and liquid particles suspended in air,

152 Re: Good suggestion, we have corrected.

153 L79: I did not see any causal relationship between L72-78 and L79-81

154 Re: L79-81 had been corrected as “aerosol hygroscopicity and CCN activity are very  
 155 different in different regions due to different chemical compositions”.

156 L113: the author need to give more detailed information of the station. Does it locate  
 157 in urban, sub-urban or rural environment? What about the surroundings, any roads,  
 158 industrial or residential activities nearby?

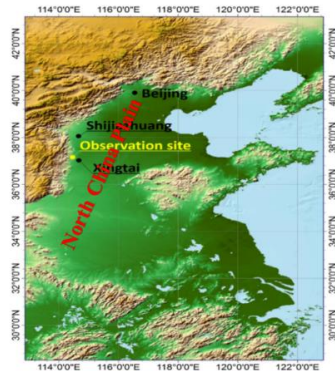
159 Re: This suburban site is situated ~17 km northwest of Xingtai urban area in southern  
160 Hebei Province. A provincial road is ~400 m southeast of the sample site, a school  
161 adjoins the east, and a town is ~600 m in the northwest (Fig. 3 in this reply). There are  
162 many industrial manufacturers in this region including coal-based power plants, steel  
163 and iron works, glassworks, and cement mills. We have added more environment  
164 information about the sample site in the manuscript.



165  
166 **Figure 3.** The surroundings at the sample site.  
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168 L114: how do the authors define the NCP? From the map in Fig. 1 it seems the station  
169 locates at the southern boundary of the plain. In L126 the authors also state the station  
170 is “heavily affected by the mountain-valley wind”.

171 Re: NCP is one of the largest plains in China. It covers an area of about 0.3 million  
172 square kilometers. The Fig.1 in the manuscript only shows a part of NCP, a full image  
173 can refer to Fig. 4 in this reply. This figure shows the observation site is in the central  
174 NCP although it is not far away from the Taihang Mountains. This station is still a  
175 good representative site in this region although it is affected by the mountain-valley  
176 wind (c.f. response to the Question 1).



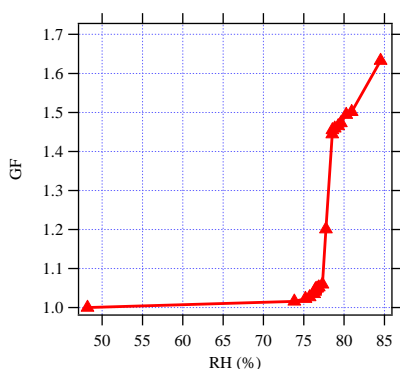
177  
178 **Figure 4.** Map showing the observation site and the scale of North China Plain (NCP).  
179 The original figure is from Wang et al., (2018).  
180

181 L138: are then passed – pass

182 Re: Yes, we have corrected it.

183 L144: Normally the RH of the HTDMA should be calibrated with ammonium sulfate,  
184 especially for high RH measurement. Did the authors calibrate the system during the  
185 campaign?

186 Re: Yes, the RH calibration is important for HTDMA running. Figure 5 in this reply  
187 shows the calibration result with ammonium sulfate during our campaign. It shows the  
188 deliquescence point of ammonium sulfate is  $78 \pm 1\%$ , this is consistent with previous  
189 studies (Badger et al., 2006; Tan et al., 2013).



190

191 Figure 5. Humidogram of ammonium sulfate for 150-nm particles measured with the  
192 HTDMA.

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194 L147: I suggest to use single letter for variables, for example,  $f_{\{g\}}$  for growth factor.

195 Re: It's a good suggestion, but we tried to follow the conventional acronyms such as  
196 "GF" or "g<sub>f</sub>" that has been used in most previous papers. "f<sub>g</sub>" is a good suggestion but  
197 it's easily confused with the expression of the aerosol scattering enhancement factor  
198 ( $f_{RH}$ ). Therefore, we'd prefer to continue the use of "GF" in this manuscript.

199 L154: An Aerosol Chemical. . .

200 Re: Corrected per the comment.

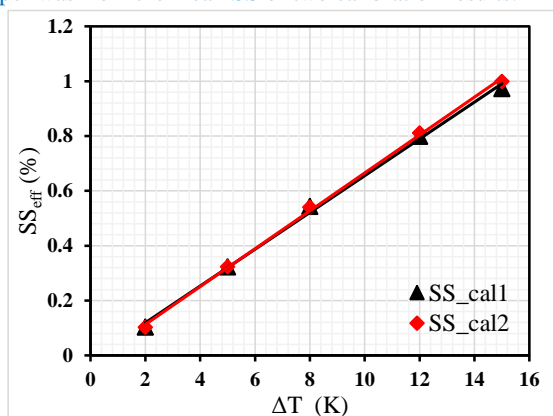
201 L156: Is the cyclone for all the aerosol instruments or only for the ACSM?

202 Re: ACSM and AE33 had their separate cyclones. SMPS had an impactor in the inlet,  
203 which can also remove the most particles larger than its measurement range. HTDMA  
204 had no cyclone which has a minor effect on its measurement. CCNc had not also a  
205 cyclone because it needed to measure the bulk CCN number concentration in this  
206 campaign.

207 L164: Is there a separate inlet line with PM<sub>1</sub> size cut for the aethalometer? I think the  
208 authors need to give a clear description of the inlets and sampling line for all the  
209 instruments. Now I am a bit confused.

210 Re: Yes, aethalometer had a separate inlet headed by a PM<sub>1</sub> for filtering particle size  
211 great than 1 μm. CCNc and HTDMA shared the same inlet. During this campaign, all  
212 sampling instruments were placed in two containers at ground level and two air  
213 conditioners were used to maintain the temperature at 20–25 °C inside containers. All  
214 stainless tube inlets were 1.5 m above the top of containers.

215 L181: I do not know why the SS needs to be corrected. Normally after applying the  
 216 calibration parameters to the system, the set SS is the true effective SS in the chamber.  
 217 No more correction is needed. Is the SS corrected with the first or second calibration  
 218 result? Is there any large difference between the two calibrations?  
 219 Re: Actually, the SS is influenced by flow rates and the temperature gradient in the  
 220 cloud column. Therefore, the flow and temperature calibrations are also needed,  
 221 which had been conducted before this campaign and the corresponding parameters  
 222 were applied in the system. However, the SS maybe changed if the CCNc runs  
 223 unsteadily, so we did two SS campaigns before and after the campaign instead of  
 224 using the SS parameters from a single calibration. Figure 6 in this reply shows the  
 225 results of the SS calibrations, the calibration method is the same as Rose et al., (2008).  
 226 The results show CCNc run well and steadily during this campaign. The calibrated SS  
 227 used in this paper was from the mean SS of two calibration results.



228  
 229 Figure 6. The results of SS calibration experiments with ammonium sulfate: CCN  
 230 efficiency spectra measured at 5 different temperature gradient ( $\Delta T$ ). SS\_cal1 and  
 231 SS\_cal2 are the calibration results before and after the campaign respectively.

232  
 233 L183: AR defined in the manuscript is determined both on chemical composition and  
 234 PNSD. I suggest using critical diameter.

235 Re: The critical diameter ( $D_{0,crit}$ ) used in this study was derived from the  
 236 hygroscopicity parameter ( $\kappa_{chem}$ ), not measured directly. AR is used for a preliminary  
 237 analysis of its relationships with nucleation events and PBL height. In the further  
 238 campaigns, we will conduct the size-resolved CCN measurements through connecting  
 239 DMA and CCNc. The critical diameter will be retrieved from size-CCN data, which  
 240 can study the relationships in detail.

241 L235: I guess here the authors mean the kappa-Sc relationship.

242 Re: The relationships of  $\kappa$ ,  $S_c$  ( $S_c = s_c - 1$ ) and  $D_d$  is shown in Fig.1 in Petters et al.  
 243 (2007). Actually, the  $D_d$  is the critical diameter corresponding to the critical  
 244 supersaturation when  $\kappa$  is known. Therefore, here we first establish the  $s_c$ - $D_d$   
 245 relationship, not  $\kappa$ - $s_c$  relationship.

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247 L246: How do the authors define the mean diameter.

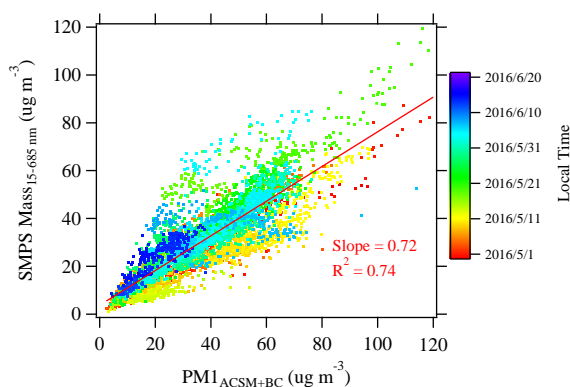
248 Re: Here the mean diameter is the geometric mean diameter ( $D_m$ ). It is defined as:

249 
$$D_m = \frac{\int_{15\text{nm}}^{685\text{nm}} n(\log D_p) D_p d\log D_p}{\int_{15\text{nm}}^{685\text{nm}} n(\log D_p) d\log D_p}$$

250 We found a mistake in its calculation previously, now it has been corrected in the  
 251 revised manuscript for which we are very grateful to the reviewer.

252 L261: Is there undefined species? Maybe the authors can do a simple mass closure  
 253 between ACSM and SMPS data by assuming a typical aerosol density in that region,  
 254 to check if there is anything missing in the aerosol mass defined by ACSM. This is  
 255 important since the authors use this data later to estimate kappa.

256 Re: The mass closure between ACSM and SMPS is hard to realize because aerosol  
 257 density and morphology are unknown during this campaign. Even so, we try to do a  
 258 simple mass closure according to the suggestion of the reviewer. Here we assume  
 259 aerosol particles are spherical and the density is 1.6 g/cm<sup>3</sup>. The closure result (Fig. 8  
 260 in this reply) shows the two masses are related dependent on time. For most of the  
 261 time, their correlation is good although the SMPS mass was always lower than that of  
 262 ACSM+BC. The biases are likely caused by different measurement size range, and  
 263 the variations of aerosol density and morphology.



264  
 265 Figure 7. Comparison of the mass concentrations of PM<sub>1</sub> (=NR-PM<sub>1</sub>+BC) with those  
 266 derived from SMPS measurements ( $D_p = 15-685$  nm).  
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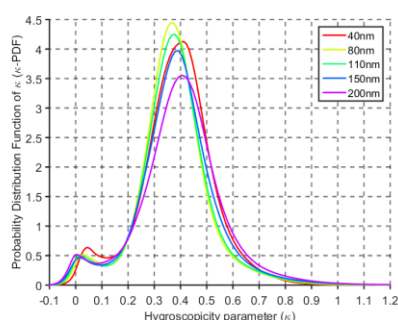
268 L267: How did the authors assume the kappa for HOA, SOA and OOA?

269 Re: The gravimetric densities ( $\rho$ ) and hygroscopicity parameters ( $\kappa$ ) of all species  
 270 used in this study for CCN closure (note: POA = HOA+COA, SOA = OOA).

Species	NH <sub>4</sub> NO <sub>3</sub>	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	NH <sub>4</sub> HSO <sub>4</sub>	H <sub>2</sub> SO <sub>4</sub>	POA	SOA	BC
$\rho$ (kg m <sup>-3</sup> )	1720	1769	1780	1830	1000	1400	1700
K	0.67	0.61	0.61	0.9	0	0.1	0

271

272 L277: the hydrophobic mode locates at kappa of 0.05 for Dd of 40 nm, and shift  
273 towards 0 for large particles. Do the authors have any explanation on this?  
274 Re: The direct comparison of the  $\kappa$ -PDF is depicted in Fig. 9 in this reply. There is an  
275 obvious difference of the hydrophobic mode as the reviewer found. This is related to  
276 the different chemical compositions. The hydrophobic mode of 40-nm particles is  
277 mainly caused by organics, while that of the larger particles is mainly caused by BC.  
278 BC is fully hydrophobic but some organics has a limited water uptake ability. This  
279 can be verified from previous studies, Wu et al., (2017) reported the fresh emitted BC  
280 are mainly in the accumulation mode.



281  
282 Figure 8. Comparison of  $\kappa$ -PDF of different size particles.

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284 L305: the difference decreases with increasing size because most of the larger  
285 particles are well aged.

286 Re: Good point. We have added the corresponding expression.

287 L316: of precursors

288 Re: We have corrected.

289 L316: is there trace gas (SO<sub>2</sub>, NO<sub>x</sub> etc.) measurement during the campaign to support  
290 the hypothesis here?

291 Re: Yes. Figure 10 in this reply shows the diurnal variations of trace gases and  
292 meteorological variables ( $T$  and RH) in this campaign. Affected by the  
293 mountain-valley wind, prevailing winds shifted from the northwest to the southeast in  
294 the early morning. There are more industrial emissions to the southeast of the  
295 measurement site than to the northwest. Therefore, the CO, SO<sub>2</sub> concentrations  
296 increased sharply in the morning after the wind shift. The increased CO suggests the  
297 increase of VOCs because of their similar sources. The O<sub>3</sub> concentration increased  
298 gradually after sunrise during the day, reflecting the enhancement of atmospheric  
299 oxidation capacity. The ample supply of effluent SO<sub>2</sub> and VOCs precursors and the  
300 strong atmospheric oxidation capacity under high RH and low  $T$  conditions made  
301 plenty of sulfate and SOA produce (Wang et al., 2016; Wang et al., 2017). This is the  
302 reason in the frequent occurrence of NPF events and the enhancement of aerosol  
303 hygroscopicity during the daytime at XT.

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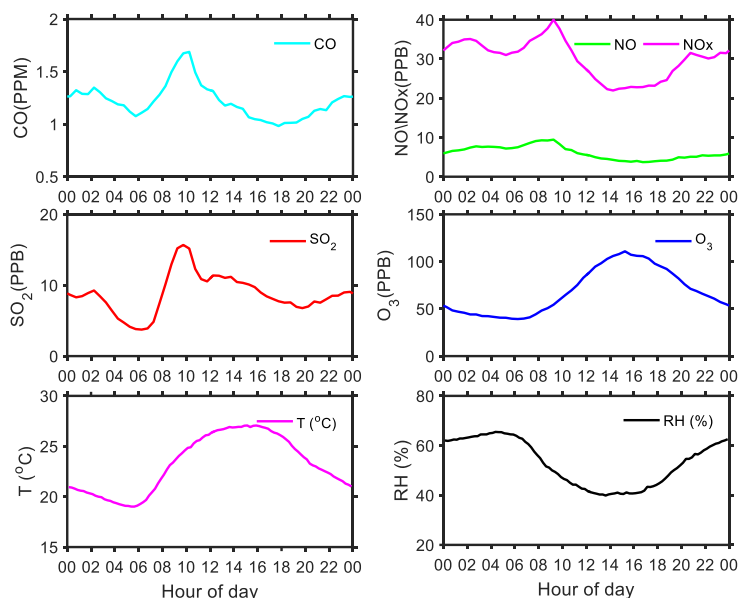


Figure 9. Diurnal variations in trace gases (CO, NO<sub>x</sub>/NO, SO<sub>2</sub> and O<sub>3</sub>) and meteorological variables (*T* and RH).

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309 L346: 100 nm is too much for nucleation mode.

310 Re: It has been corrected as “small Aitken mode particles (< 50 nm)”.

311 L359: From Fig. 6 it seems kappa already starts decrease at 12:00. The secondary  
 312 aerosol production is also active in the afternoon (Fig. 7b). Why does kappa for larger  
 313 particles decrease?

314 Re: This is related with the local emissions, especially the emitted organics. As is  
 315 shown in the diurnal variations (Fig.2 in this reply), the primary organics (COA and  
 316 HOA) increase sharply near noon. This is the reason why  $\kappa$  has a short-time decrease  
 317 at near 12:00.

318 L360: in Fig. 6c, kappa of larger particles also reaches 0.4 around noon, which is  
 319 also very close to that of pure AS. But from Fig. 6d we can see there is still a large  
 320 fraction of hydrophobic and less-hygroscopic species.

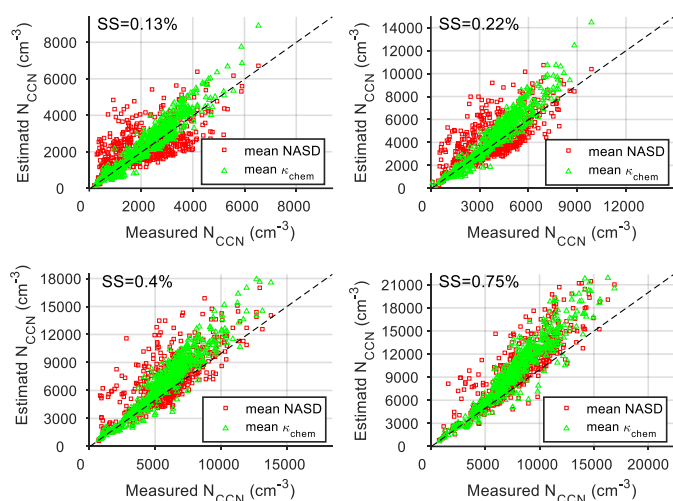
321 Re: As discussed in section 4.2, aerosol particles in this site were highly  
 322 internal-mixed and aged.  $\kappa$ -PDF for all size particles showed only one hydrophilic  
 323 mode during daytime although there was still a large fraction of hydrophobic and  
 324 less-hygroscopic species. This is because the condensation of sulfuric acid on  
 325 organics or BC can greatly enhance their hygroscopicity (Peng et al., 2016; Zhang et  
 326 al., 2017).

327 L370: the growth of the newly formed particles is also driven by condensation of  
 328 sulfate and organics.

329 Re: Yes, we have corrected.

330 L374: why does the PM1 composition differs largely from 40-200 nm particles? 200

331 nm is in the accumulation mode which is the main contributor of PM1 mass. So I  
 332 would not expect a large difference between  $\kappa_{\text{chem}}$  and  $\kappa_{\text{gf,200nm}}$ .  
 333 Re: The reviewer is right. We have deleted the expression “the bulk chemical  
 334 compositions of PM<sub>1</sub> and of 40–200 nm particles differ greatly” in the manuscript.  
 335 The difference was mainly induced by the simple ZSR mixing rule. This feature was  
 336 stronger at noon when atmospheric oxidation and the aging process were more rapid.  
 337 We have studied this phenomenon deeply in another paper (Zhang et al., 2017).  
 338 L387: Use critical diameter in stead of AR in this section.  
 339 Re: As mentioned above, we did not take size-resolved CCN measurement in this  
 340 campaign, so critical diameter cannot be got directly. We will do the work in the  
 341 future campaigns.  
 342 L394: the number concentration of fine particles. . .  
 343 Re: We have corrected.  
 344 L400: I do not think one can get this conclusion based on the discussion in this section.  
 345 The result is in consistent with it but can not directly prove it. Also I think the paper  
 346 of Dusek et al., (2006) should be cited here. Dusek U, Frank G P, Hildebrandt L, et al.  
 347 Size matters more than chemistry for cloud-nucleating ability of aerosol particles[J].  
 348 Science, 2006, 312(5778):1375.  
 349 Re: Figure 11 in this reply depicts the sensibility tests of aerosol size distribution and  
 350 chemical composition to the CCN number concentration estimation, which are similar  
 351 with that of Dusek et al., (2006). The figure shows the closure correlation using mean  
 352 NASD (normalized aerosol size distribution) is weaker than that using mean  $\kappa_{\text{chem}}$ ,  
 353 reflecting the more important effect of particle size for aerosol CCN activity than  
 354 chemical composition. We have added the figure in the supplement and cited this  
 355 paper in the revised manuscript.



356  
 357 Figure 10. Correlation between the measured and estimated CCN number  
 358 concentration. The later was from the mean NASD (normalized aerosol size  
 359 distribution) but variable composition ( $\kappa_{\text{chem}}$ ) or using the mean  $\kappa_{\text{chem}}$  but variable

360 NASD.

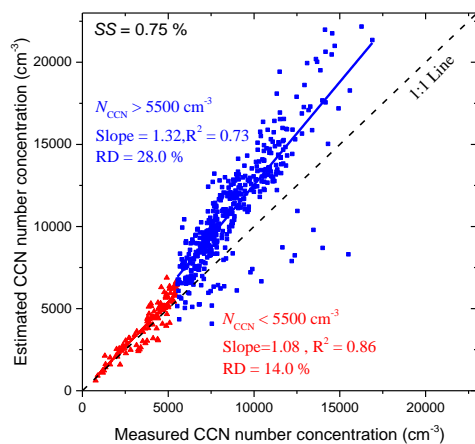
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362 L407: the critical diameters of aerosol for SS from 0.07% to 0.80% range from about  
363 30 nm to 200 nm. As shown in Fig.6, there is a large difference between  
364  $\kappa_{\text{chem}}$  and  $\kappa_{\text{gf}}$  and the authors claimed that PM1 composition “differ  
365 greatly from 40-200 nm particles” (L374). It means one can not use the PM1  
366 composition in the CCN closure. I think a better way might be to compare the kappa  
367 obtained from HTDMA and CCN measurements. Although some species may exhibit  
368 different kappa in sub- and super-saturation, at least the two measurements are in the  
369 same size range.

370 Re: The related expression “PM<sub>1</sub> composition differ greatly from 40-200 nm particles”  
371 is not appropriate, which has been deleted in this manuscript. The  $\kappa$  closure from  
372 HTDMA and CCN data is useful to analyze the aerosol hygroscopicity in detail.  
373 However, it cannot be realized in this campaign because no size-resolved CCN data. It  
374 will be done in the future campaigns.

375 L434: From this paragraph what I understand is, 0.31 is the average  $\kappa_{\text{chem}}$   
376 and calculated  $N_{\text{ccn}}$  is not sensitive on the variation of  $\kappa_{\text{chem}}$ . But how can  
377 you make sure that 0.31 is a good proxy for the calculation of  $N_{\text{ccn}}$  in XT? You found  
378 some discrepancies between the calculated and measured  $N_{\text{ccn}}$  and claimed that there  
379 are some biases in measured  $N_{\text{ccn}}$ . So, there is no reference to check which kappa  
380 value is appropriate. Normally, water depletion effect can be neglected if  $N_{\text{ccn}} < 1e4$ .  
381 Maybe you can try to do the fit for data points  $< 1e4$  and  $> 1e4$  separately.

382 Re: Figure 8 in the manuscript shows there is a limited difference for the CCN closure  
383 using variable or mean  $\kappa_{\text{chem}}$ , so  $N_{\text{CCN}}$  is not sensitive on the variation of  $\kappa_{\text{chem}}$ . 0.31 is  
384 only a reference value for people who need to calculate the CCN concentration in this  
385 region in their models. As the reviewer says, the CCN closure can be categorized  
386 according to  $N_{\text{CCN}}$ . Actually, the measured  $N_{\text{CCN}}$  is biased when  $N_{\text{CCN}} > 5500 \text{ cm}^{-3}$   
387 (Fig. 12 in this reply), this is coincident with the report from DMT-CCNC manual.  
388 Figure 12 in this reply also shows the CCN closure is very good when  $N_{\text{CCN}} < 5500$   
389  $\text{cm}^{-3}$ , reflecting the validation of the closure method used in this study.



390

391 Figure 11. Estimated versus measured CCN number concentrations at SS = 0.75 %.  
 392 The  $N_{CCN}$  is estimated based on  $\kappa$ -Köhler theory, using the real-time  $\kappa_{chem}$ . Here, the  
 393 critical value of  $N_{CCN} = 5500 \text{ cm}^{-3}$  is used to separate the points into two groups. A  
 394 separate linear regression analysis is done on each group. The slopes, correlation  
 395 coefficients ( $R^2$ ), and relative deviations (RD) are shown in the figure.

396

397 L437: the sensitivity of  $N_{ccn}$  on aerosol mixing state is not examined in this section.

398 Re: According to the HTDMA measurement results, aerosols in this region are highly  
 399 internal-mixed and aged. Therefore, we directly assume aerosols are internally mixed  
 400 when calculating the CCN number concentration as description in section 3.2.

401 Another work of our group have suggested the mixing state had a minor effect for the  
 402 CCN estimation in Beijing (Ren et al., 2017).

403

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452 **Characterization of aerosol hygroscopicity, mixing state, and**  
453 **CCN activity at a suburban site in the central North China Plain**

454

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473

474 **Abstract.** Aerosol hygroscopicity, mixing state and CCN activity were investigated as  
475 a part of the Atmosphere-Aerosol-Boundary Layer-Cloud (A<sup>2</sup>BC) Interaction Joint  
476 Experiment carried out at Xingtai (XT), a suburban site in the center of the North  
477 China Plain (NCP). In general, the probability density function of the hygroscopicity  
478 parameter ( $\kappa$ -PDF) for 40–200 nm particles had a unimodal distribution and mean  
479  $\kappa$ -PDF patterns for different sizes were similar, suggesting that the particles were  
480 highly aged and internally mixed because of strong photochemical reactions. The  $\kappa$   
481 calculated from ~~the~~ hygroscopic growth factor in the daytime and at nighttime showed  
482 that photochemical reactions largely enhanced the aerosol hygroscopicity, and the  
483 effect became weaker as the particle size increased. In addition, the aerosol  
484 hygroscopicity was much larger at XT than at sites in the northern part of the NCP,  
485 illustrating that the hygroscopicity of particles varies largely due to different  
486 emissions and chemical processes in the NCP.

487 Measurement results also showed that new particle formation events occurred  
488 frequently at XT, one of the most polluted city in China. The evolution of the  
489 planetary boundary layer played a dominant role in aerosol mass concentration  
490 changes while particle formation and growth had a greater influence on the variation  
491 in aerosol number concentrations. Particle size was the most important factor  
492 influencing the ability of aerosols to activate, especially at higher levels of  
493 supersaturation (SS). The cloud condensation nuclei (CCN) number concentration  
494 ( $N_{CCN}$ ) derived from chemical composition was highly correlated with the measured  
495  $N_{CCN}$  ( $R^2 \geq 0.85$ ), but was generally overestimated due to measurement uncertainties.

496 The effect of chemical composition on  $N_{CCN}$  was weaker relative to the particle size.  
497  $N_{CCN}$  sensitivity tests showed that the impact of chemical composition on  $N_{CCN}$   
498 became weaker with increasing SS, suggesting that chemical composition played a  
499 less role in  $N_{CCN}$  estimation~~ss~~ at higher SS level~~ss~~. A good proxy for the chemical  
500 comical composition ( $\kappa = 0.31$ ) was found, which can simplify the calculation of  
501  $N_{CCN}$  on models.

## 502 1. Introduction

503 Aerosols, defined as the mixture of solid and liquid particles suspended in air, are  
504 ubiquitously present in the atmosphere because of direct emissions from biogenic and  
505 anthropogenic sources and the secondary transformation from gas precursors. Aerosol  
506 particles play an important role in climate changes through direct and indirect effects  
507 (e.g. Ramanathan et al., 2001; Daniel et al., 2008; ~~Z~~-Li et al., 2016). However, the  
508 impact of aerosols on climate change is difficult to simulate because of the highly  
509 variable physical and chemical properties of aerosols, and complex aerosol-cloud  
510 interactions (IPCC, 2013; Lebo et al., 2017).

511 The hygroscopic growth and mixing state of aerosol particles are important for  
512 estimating the direct climate effect of aerosols. This is because the growth and mixing  
513 can change the particle size and optical properties of aerosol particles, directly  
514 influencing the terrestrial radiation budget and degrading the atmospheric visibility  
515 (e.g. Covert et al., 1972; Stock et al., 2011; Peng et al., 2016; Z. Li et al., 2017). In  
516 addition, aerosol particles can be activated as cloud condensation nuclei (CCN) under

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517 supersaturation (SS) conditions. The variability in CCN number concentration ( $N_{CCN}$ )  
518 can modify cloud microphysical properties, thereby causing an indirect radiative  
519 forcing (Twomey, 1974; Albrecht, 1989). Previous studies have addressed three main  
520 aerosol properties influencing the CCN activation, namely, particle size, chemical  
521 composition, and mixing state. However, their relative importance is different in  
522 different environments (e.g. [Dusek et al., 2006](#); Ervens et al., 2007; Cubison et al.,  
523 2008; Deng et al., 2011; Zhang et al., 2014; [Schmale et al., 2018](#)).

524 Ambient aerosols are composed of different species, including inorganic ions,  
525 organic components, black carbon (BC), and mineral dust. Inorganics mainly contain  
526 sulfate, nitrate, and ammonium, while organic aerosols (OA) consist of thousands of  
527 chemicals (Jacobson et al., 2000). The hygroscopicity and CCN activity of a single  
528 component can be characterized according to laboratory studies (e.g. Petters and  
529 Kreidenweis, 2007), but the properties of their mixtures are hard to estimate because  
530 of the different chemical species and mixing states of particles in the atmosphere.  
531 Therefore, aerosol hygroscopicity and CCN activity are very different in different  
532 regions [due to different chemical compositions](#). Comprehensive field measurements  
533 of aerosol properties in different areas are necessary to improve models.

534 China, especially the North China Plain (NCP), has been suffered from severe air  
535 pollution since its rapid industrialization and urbanization in the last couple of  
536 decades, where diverse sources and aging processes make aerosol properties  
537 particularly diverse and complex. As such, the region has drawn much attention in  
538 studying the aerosol mixing state, hygroscopicity, and CCN activity (Deng et al., 2011;

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539 Liu et al., 2011; Zhang et al., 2014; F. Zhang et al., 2016; S.L. Zhang et al., 2016; Wu  
540 et al., 2016; Y. Wang et al., 2017). Liu et al. (2011) and Y. Wang et al. (2017) have  
541 suggested that ambient particles are mostly an external mixture with different  
542 hygroscopicities. Deng et al. (2011) has shown that the aerosol number size  
543 distribution is critical in the prediction of  $N_{CCN}$  while Zhang et al. (2014, 2017) have  
544 highlighted the importance of chemical composition in determining particle activation  
545 properties. However, all these studies were done using data from the northern part of  
546 the NCP. Few studies have focused on the central region of the NCP. Compared to the  
547 northern part of the NCP, the central part of the NCP is more affected by industrial  
548 emissions where a dense cluster of China's heavy industries exist (Fu et al., 2014).  
549 Measurement of aerosol properties in the central part of the NCP are critically needed  
550 to investigate the impact of air pollution on the environment and climate changes.

551 Xingtai (XT), a city located in the ~~middle-central area~~ of the NCP, often ranks in  
552 the top of polluted cities in China. Local industrial and domestic sources are the  
553 greatest contributors to severe haze events (Wang et al., 2014). A field experiment  
554 called the Atmosphere-Aerosol-Boundary Layer-Cloud ( $A^2BC$ ) Interaction Joint  
555 Experiment was carried out at a suburban site in Xingtai in the summer of 2016.  
556 Differences in aerosol properties at this site and at sites in the northern part of the  
557 NCP were found in this study.

558 The paper is organized as follows. Sections 2 and 3 describe the measurement  
559 method and data analysis theory. Section 4 presents and discusses the measurement  
560 results, which includes the data time series, aerosol mixing state, hygroscopicity, CCN

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561 prediction and its sensitivity to chemical composition. A summary and conclusions are  
562 given in section 5.

## 563 2. Measurements

### 564 2.1. Sampling site and meteorology

565 The A<sup>2</sup>BC was carried out at the National Meteorological Basic Station~~national~~  
566 ~~weather station~~ located in XT (37.18° N, 114.~~36°~~37° E, 180 m ASL) from 1 May to 15  
567 June of 2016. This suburban site is situated ~ 17 km northwest of Xingtai urban area  
568 in southern Hebei Province, located in the central part of the NCP and to the east of  
569 Taihang Mountains (Fig. 1a). This region is heavily populated, urbanized, and  
570 industrialized. The major industrial manufacturers include coal-based power plants,  
571 steel and iron works, glassworks, and cement mills. The weak diffusion conditions  
572 and heavy industrial emissions lead to exceptionally high concentrations of particulate  
573 matter (PM) with diameter less than 10 μm (PM<sub>10</sub>) and 2.5 μm (PM<sub>2.5</sub>), as well as gas  
574 pollutants such as sulfur dioxide (SO<sub>2</sub>), volatile organic compounds (VOCs) and  
575 nitrogen oxides (NO<sub>x</sub>) during the frequent occurring haze episodes in this region  
576 (Wang et al., 2014; Fu et al., 2014). Figure 1b shows the mean distribution of SO<sub>2</sub>  
577 concentrations from May of 2012 to 2016, confirming that the measurement site is  
578 located in one of the pollution centers ~~in~~of this region. The detailed analysis of gas  
579 precursors and aerosol chemical species shows this station is a good representative  
580 site in this region (Zhang et al., 2018).

581 Time series of meteorological variables measured at ~~the~~this ~~weather~~

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582 meteorological station are shown in Fig. S1. This site is heavily affected by the  
583 mountain-valley wind, showing a prevailing southeasterly wind during the day and a  
584 northwesterly wind at night (Fig. S1 and Fig. S2). There was almost no precipitation  
585 during the study period. The ambient temperature ( $T$ ) and relative humidity (RH) time  
586 series show opposing trends. Campaign-mean values of  $T$  and RH are 21.9 °C and  
587 51.6 %, respectively.

## 588 2.2. Instrumentation and operation

### 589 2.2.1. Aerosol hygroscopicity measurements

590 ~~The~~ The custom-built hygroscopicity tandem differential mobility analyzer  
591 (H-TDMA) used in this study has been described in detail by others (Tan et al., 2013;  
592 Y. Wang et al., 2017). Briefly, ambient aerosols are first dried and neutralized by a  
593 Nafion dryer and a soft X-ray charger. A differential mobility analyzer (DMA<sub>1</sub>, model  
594 3081L, TSI Inc.) is used to select monodispersed particles of a certain diameter ( $D_{p0}$ ).  
595 The monodispersed particles are then passed through ~~aa n~~ Nafion humidifier with a  
596 controlled higher RH and are humidified. A second DMA (DMA<sub>2</sub>, same model as the  
597 DMA<sub>1</sub>) and a water-based condensation particle counter (WCPC, model 3787, TSI  
598 Inc.) are used to measure the number size distribution of the humidified particles. The  
599 DMA<sub>1</sub> and WCPC can also be connected directly to measure the 10–400 nm particle  
600 number size distribution (PNSD). In this study, the dry diameters selected by the  
601 DMA<sub>1</sub> are 40, 80, 110, 150, and 200 nm. The humidified RH is set to 85 %, the RH  
602 calibration with ammonium sulfate for the HTDMA is shown in Fig. S3 in the

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603 [supplement](#).

604 The hygroscopic growth factor (GF) is defined as the ratio of the humidified  
605 diameter at a given RH to the dry diameter:

$$606 \quad GF = \frac{D_p(RH)}{D_{p0}}, \quad (1)$$

607 where  $D_p(RH)$  is the particle diameter at the given RH and  $D_{p0}$  is the dry diameter  
608 selected by the DMA<sub>1</sub>. The measured distribution function versus GF (GF-MDF) can  
609 be calculated with WCPC data downstream from the DMA<sub>1</sub> and DMA<sub>2</sub>. The GF  
610 probability density function [\(GF-PDF\)](#) is then retrieved using the TDMAFIT  
611 algorithm [\(Stolzenburg and McMurry, 1988, 2008\)](#).

### 612 2.2.2. Aerosol chemical composition measurements

613 [TheAn](#) Aerosol Chemical Speciation Monitor (ACSM) was deployed to measure  
614 the non-refractory submicron aerosol (NR-PM<sub>1</sub>) species (sulfate, nitrate, ammonium,  
615 chloride, and organics) in real-time. A PM<sub>2.5</sub> URG cyclone (model URG-2000-30ED)  
616 was installed in the front of the sampling inlet to remove coarse particles (> 2.5 μm in  
617 diameter). Before sampling into the ACSM, aerosol particles were dried (below 40 %  
618 RH) by a silica gel diffusion dryer. In addition, the ACSM was calibrated routinely  
619 with pure ammonium nitrate to determine its ionization efficiency. More detailed  
620 descriptions about the ACSM are given by [Ng et al., \(2011\)](#) and [Sun et al., \(2012\)](#). A  
621 positive matrix factor analysis is used to analyze the organic spectral matrices  
622 according to [Ulbrich et al., \(2009\)](#). Three factors, i.e., hydrocarbon-like OA (HOA),  
623 cooking OA (COA), and oxygenated OA (OOA), are chosen as the ACSM dataset.

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624 HOA and COA are both ~~anthropogenic~~ primary organic aerosols (POA) while OOA is  
625 the secondary organic aerosol (SOA).

626 The ACSM does not detect refractory material such as BC, so a seven-wavelength  
627 aethalometer (AE-33, Magee Scientific Corp.) was used to measure the BC mass  
628 concentration of BC particles with diameters  $< 1.0 \mu\text{m}$  (BC  $\text{PM}_{10}$ ). Mineral dust and  
629 sea salt are the other refractory species, but they typically exist in the coarse mode and  
630 make negligible contributions to  $\text{PM}_{10}$  (Juranyi et al., 2010; Meng et al., 2014).

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### 631 2.2.3. Aerosol size distribution and CCN measurements

632 The aerosol particle number size distribution (15–685 nm) was measured by a  
633 scanning mobility particle sizer (SMPS) that was equipped with a long DMA (model  
634 3081L, TSI Inc.) and a condensation particle counter (CPC, model 3775, TSI Inc.). A  
635 single-column continuous-flow ~~stream-wise~~ thermal-gradient cloud condensation  
636 nuclei counter (CCNC-100, DMT Inc.) was applied to measure the bulk CCN number  
637 concentration. Five SS levels, i.e., 0.07, 0.1, 0.2, 0.4, and 0.8 %, were set in the  
638 CCNC and the running time was 10 min for each SS level. The SS in the CCNC ~~are~~  
639 were calibrated with pure ammonium sulfate (Rose et al., 2008) before and after the  
640 measurement campaign. The corrected SS levels ~~are-were~~ 0.11, 0.13, 0.22, 0.40, and  
641 0.75 %, respectively.

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642 The aerosol activation ratio (AR) at a certain SS is calculated as  $N_{\text{CCN}}$  divided by  
643 the total particle number concentration in the 15–685 nm range ( $N_{15-685 \text{ nm}}$ ), i.e.,  $\text{AR} =$   
644  $N_{\text{CCN}} / N_{15-685 \text{ nm}}$ . The particle number concentration below 15 nm is not measured by



645 the SMPS, but this does not affect the calculated  $N_{CCN}$  because the activation critical  
646 diameter is always larger than 15 nm at these SS levels (Zhang et al., 2014). Aerosol  
647 particles with diameters larger than 685 nm are also not detected by the SMPS. These  
648 larger particles will always act as CCN due to their larger dry sizes. However, the  
649 number concentration above 685 nm in the atmosphere is always negligible (Juranyi  
650 et al., 2010).

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#### 651 2.2.4. Other measurements

652 In this study, a micro-pulse lidar (MPL-4B, Sigmaspace Corp.) was used to study  
653 the evolution of the planetary boundary layer (PBL). The pulse repetition rate of the  
654 MPL ~~w~~was 2.5 kHz at a visible wavelength of 532 nm. The peak value of the optical  
655 energy of the laser beam was 8  $\mu$ J. The pulse duration ranged from 10 to 100 ns, and  
656 the pulse interval was set to 200 ns, corresponding to a spatial resolution of 30 m. The  
657 MPL-retrieved PBL height is the altitude where a sudden decrease in the scattering  
658 coefficient occurs (Brooks, 2003; Quan et al., 2013).

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659 Trace gas analyzers ~~(manufactured by ECOTECH)~~ were used to measure the  
660 gaseous species of ~~ozone ( $O_3$ ) and  $SO_2$ ,  $NO_x$ ,  $NO$  and  $CO$ .~~  ~~$SO_2$  was measured~~  
661 ~~using an  $SO_2$  analyzer with a fluorescence cell (Ecotech model 9850A) and  $O_3$  was~~  
662 ~~measured using an  $O_3$  analyzer (Ecotech model 9810B) with ultraviolet absorption~~  
663 ~~technology.~~ More detailed descriptions about the ~~trace gas~~ analyzers are given by Zhu  
664 et al., (2016).

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665 During this campaign, all sampling instruments were placed in two containers at

666 ground level and two air conditioners were used to maintain the temperature at 20–25  
667 °C inside containers. All stainless tube inlets were ~ 1.5 m above the top of containers.

### 668 3. Theory

#### 669 3.1. Hygroscopicity parameter

670 To link hygroscopicity measurements below and above water vapor saturation,  
671 the Köhler theory (Köhler, 1936) is parameterized using the hygroscopicity parameter  
672  $\kappa$  (Petters and Kreidenweis, 2007). This is known as the  $\kappa$ -Köhler theory. According  
673 to the theory, the equilibrium equation over a solution droplet at a saturation ratio  
674  $S(D)$  is

$$675 S(D) = \frac{D^3 - D_d^3}{D^3 - D_d^3(1 - \kappa)} \exp\left(\frac{4\sigma_{s/a}M_w}{RT\rho_w D}\right), \quad (2)$$

676 where  $D$  and  $D_d$  are the wet and dry droplet diameters, respectively,  $\sigma_{s/a}$  is the  
677 surface tension coefficient,  $M_w$  is the mole mass of water,  $R$  is the universal gas  
678 constant,  $T$  is the temperature, and  $\rho_w$  is the density of water.

679 Below the water vapor saturation,  $S(D)$  is RH,  $D$  is  $D_p(\text{RH})$ , and  $D_d$  is  $D_{p0}$   
680 in Eq. (1). The  $\kappa$  parameter is then calculated using H-TDMA data according to Eq. (1)  
681 and Eq. (2):

$$682 \kappa_{\text{gt}} = (\text{GF}^3 - 1) \cdot \left[ \frac{1}{\text{RH}} \exp\left(\frac{4\sigma_{s/a}M_w}{RT\rho_w D_d \text{GF}}\right) - 1 \right]. \quad (3)$$

683 For a multicomponent particle, the Zdanovskii–Stokes–Robinson (ZSR) mixing  
684 rule (Stokes and Robinson, 1966) can also estimate  $\kappa$  using chemical composition  
685 data:

$$686 \kappa_{\text{chem}} = \sum_i \varepsilon_i \kappa_{i\text{c}} \quad (4)$$

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687 where  $\varepsilon_i$  and  $\kappa_i$  are the volume fraction and hygroscopicity parameter for the  $i$ th  
 688 chemical component. The ACSM provides the mass concentrations of inorganic ions  
 689 and organics. A simplified ion-pairing scheme such as that described by Gysel et al.  
 690 (2007) ~~was~~ is applied to convert ion mass concentrations to mass concentrations of  
 691 their corresponding inorganic salts (see Table S1 in the supplement). Table S1 also  
 692 lists  $\kappa$  and the gravimetric density of each individual component under supersaturated  
 693 conditions. In the following discussions,  $\kappa_{gf}$  and  $\kappa_{chem}$  denote the hygroscopicity  
 694 parameters derived from H-TDMA measurements and estimated using the ZSR  
 695 mixing rule, respectively.

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### 696 3.2. CCN estimation

697 The critical supersaturation ( $s_c$ ,  ~~$s_c$~~   ~~$s_c$~~   ~~$s_c$~~   $s_c = S_c - 1$ ) for a dry diameter ( $D_d$ ) of a particle  
 698 with hygroscopicity  $\kappa$  is calculated from the maximum of the  $\kappa$ -Köhler curve (Eq.  
 699 ~~2~~) (Petters and Kreidenweis, 2007). ~~The  $D_d$  is also the critical diameter~~  
 700 ~~corresponding to the  $s_c$  when  $\kappa$  is known, so the  $s_c$ - $D_d$  relationship is then can~~  
 701 ~~be~~ established. According to this relationship, the critical diameter ( $D_{0,crit}$ ) can be  
 702 calculated using the estimated  $\kappa_{chem}$  (Eq. 4) at ~~a given~~ ~~the SS~~ ~~set in the CCNC~~. All  
 703 particles larger than  $D_{0,crit}$  will activate as CCN, assuming that aerosols are  
 704 internally mixed. Then the CCN number concentration can be estimated from the  
 705 integral of the aerosol size distribution provided by the SMPS from  $D_{0,crit}$  to the  
 706 ~~largest maximum~~ measured size ( $D_{max}$ ) following Eq. (5):

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$$N_{CCN}(SS) = \int_{D_{0,crit}(SS)}^{D_{max}} \frac{dN(D)}{d\log(D)} d\log(D) \quad (5)$$

708  $N_{CCN}(SS)$  can then be compared to the number of CCN at the same SS measured by

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709 the CCNC (i.e. a closure study).-

710

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## 711 4. Results and discussion

### 712 4.1. Overview

713 Figures 2 and 3 show the time series of the main aerosol properties during the  
714 field experimentthis campaign. The PNSD changes dramatically (Fig. 2a) and the  
715 aerosol number concentration in the 15–50 nm range ( $N_{15-50 \text{ nm}}$ ) increases sharply in  
716 the morning almost every day (Fig. 2b). The time series of the mean diameter ( $D_p$  $D_m$ )  
717 of particles also shows that a growth process occurs after the sharp increase in  $N_{15-50}$   
718 nm. All these phenomena suggest that new particle formation (NPF) events occurred  
719 frequently at XT during the field experiment (Kulmala et al., 2012; Y. Li et al., 2017).

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720 This is likely related to the high concentration of gas precursors mainly from local  
721 emissions. High emissions of SO<sub>2</sub> and volatile organic compounds (VOCs) associated  
722 with the high oxidation capacity in a polluted atmosphere make NPF events occur  
723 more frequently in north China (Z. Wang et al., 2017).

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724 Figure 2c-d shows the time series of the probability density function of  $\kappa_{\text{gf}}$   
725 ( $\kappa$ -PDF) for 40 nm and 150 nm particles, respectively. In general, mono-modal  
726  $\kappa$ -PDFs are-were observed. This is different from  $\kappa$ -PDFs at other sites in China  
727 where bi- and tri-modal distributions are dominant (Liu et al., 2011; Ye et al., 2013;  
728 Jiang et al., 2016; S.L. Zhang et al., 2016; Y. Wang et al., 2017). This is due to  
729 differences in the aerosol mixing state, which will be discussed in section 4.2.

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730 The bulk mass concentrations of organics, sulfate, nitrate, ammonium, and  
731 chloride measured by the ACSM are shown in Fig. 3a, along with the BC mass  
732 concentration measured with the AE-33. Organics and sulfate ~~are-were~~ the dominant  
733 chemical species with mass fractions in PM<sub>1</sub> of 39.1 % and 24.7 %, respectively.  
734 Figure 3b-c shows the volume fractions of paired chemical compositions and the  
735 hygroscopicity parameter— ( $\kappa_{\text{chem}}$ ) derived from ~~chemical compositions~~ $\kappa_{\text{chem}}$ ,  
736 respectively. The average volume fraction of inorganics  
737 ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>+NH<sub>4</sub>HSO<sub>4</sub>+H<sub>2</sub>SO<sub>4</sub>+NH<sub>4</sub>NO<sub>4</sub>) ~~is-was~~ similar to that of organics  
738 (POA+SOA), but their volume fractions changed diurnally. ~~In general, t~~The volume  
739 fraction of inorganics ~~increases-increased~~ during daytime while the volume fraction of  
740 organics ~~decreases-decreased~~. In addition, SOA ~~is-was~~ the dominant contributor to OA,  
741 accounting for ~69 % of the organics volume. This shows that photochemical  
742 reactions were strong at XT during ~~this campaign~~~~the field experiment~~ (Huang et al.,  
743 2014). The mean  $\kappa_{\text{chem}}$  in Fig. 3c ~~is-was~~ 0.31 with values ranging from 0.20 to 0.40.  
744 The trend in  $\kappa_{\text{chem}}$  ~~is-was~~ similar to that of the volume fraction of inorganics,  
745 suggesting that inorganics ~~played-plays~~ a key role when it ~~comes-comes to~~  $\kappa_{\text{chem}}$ .  
746 ~~this is consistent with the study of~~ (Wu et al., (2016).

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#### 747 4.2. Aerosol mixing state and hygroscopicity

748 The average probability density functions of  $\kappa_{\text{gf}}$  ( $\kappa$ -PDF) for different particle  
749 sizes derived from H-TDMA data are shown in Fig. 4. For all particle sizes considered,  
750  $\kappa_{\text{gf}}$  ~~ranges-ranged~~ from 0 to 0.8 and the  $\kappa$ -PDF patterns ~~are-were~~ similar, suggesting

751 that the hygroscopic compounds in different particle size mode ~~were~~ similar at  
752 XT ~~during the field experiment~~. In general,  $\kappa$ -PDF patterns show only one  
753 hydrophilic mode with ~~the~~ weak hydrophobic modes occasionally appearing at night  
754 when photochemical reactions are weak (Fig. ~~S3S4~~). This is different from what has  
755 been reported at other sites in China (Liu et al., 2011; Ye et al., 2013; Jiang et al.,  
756 2016; Zhang et al., 2016; Y. Wang et al., 2017) where the  $\kappa$ -PDF patterns always  
757 show bi- or tri-modal distributions. Based on previous studies (Liu et al., 2011; Y.  
758 Wang et al., 2017), ambient aerosols can be classified into three groups according to  
759 their  $\kappa_{gf}$  values:

- 760 — nearly hydrophobic (NH):  $\kappa_{gf} < 0.1$
- 761 — less hygroscopic (LH):  $0.1 \leq \kappa_{gf} < 0.2$
- 762 — more hygroscopic (MH):  $0.2 \leq \kappa_{gf}$

763 Table 1 gives the number fractions of each group for different particle sizes. The MH  
764 group ~~dominates~~ ~~dominated~~ all particle sizes. The number fractions of the NH and LH  
765 groups ~~are~~ ~~were~~ less than 6.0 % each. However, the volume fractions of hydrophobic  
766 BC and low-hygroscopic organics (where  $\kappa_{BC}$  is approximately zero and  $\kappa_{organic}$  is  
767 typically less than 0.1) ~~are~~ ~~were~~ ~10.1 % and 47.4 % according to chemical  
768 composition measurements (Fig. 3b). This suggests that the particles were highly aged  
769 and internally mixed at XT during ~~the field experimentis~~ ~~campaign~~. Coating of  
770 sulfates and secondary organics during the aging process changes the structure of BC  
771 and makes it grow, which can significantly enhance the hygroscopicity of particles  
772 (e.g., Zhang et al., 2008; Jimenez et al., 2009; Tritscher et al., 2011; Guo et al., 2016).

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773 In addition, the observed unimodal distribution of  $\kappa$ -PDF also suggests the ~~highly~~  
774 ~~internal~~internally mixed state of the particles (Swietlicki et al., 2008).

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775 Figure 5 shows the average size-resolved  $\kappa_{gf}$  derived from H-TDMA data at XT  
776 and at other sites in China. At XT,  $\kappa_{gf}$  for different particle sizes ~~are~~were larger in  
777 the daytime than at night and the difference between daytime and nighttime ~~decreases~~  
778 ~~decreased~~ with increasing particle size. This suggests that the impact of  
779 photochemical reactions on aerosol hygroscopicity ~~i~~was strong and that the effect  
780 ~~i~~was weaker with increasing particle size because most of the larger particles are  
781 always well aged.

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782 The magnitude of  $\kappa_{gf}$  ~~is~~was larger at XT than at other sites of China. In  
783 particular, the magnitude of  $\kappa_{gf}$  ~~wa~~is much larger at XT than at sites in the northern  
784 part of the NCP, i.e., Beijing, Wuqing, and Xianghe. The lower  $\kappa_{gf}$  in the urban area  
785 of Beijing is likely related to the more severe traffic emissions (Ye et al., 2013; Wu et  
786 al., 2016). Wuqing and Xianghe are located in the suburban area between the two  
787 megacities of Beijing and Tianjin and are simultaneously affected by traffic and  
788 industrial emissions. The magnitude of  $\kappa_{gf}$  at these two sites are higher than at  
789 Beijing but lower than at XT. Although XT is located far away from these megacities,  
790 it is situated in the industrial center of the NCP, so the higher concentrations of  
791 precursors and strong photo chemical reactions make the particles more internally  
792 mixed and highly aged. This is why  $\kappa_{gf}$  in XT is larger than at other sites. This  
793 suggests that the hygroscopicity of particles from different emissions and chemical  
794 processes differ in NCP. In addition, 40 nm particles ~~are~~were always more

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795 hygroscopic than 80 nm particles at XT, especially in the daytime, which ~~is~~was also  
796 different from other sites. This is likely because the coating effect of sulfates and  
797 secondary organics is more significant on smaller particles (Tritscher et al., 2011; Guo  
798 et al., 2016). Furthermore, since the field measurements took place in a local with  
799 heavy industrial activities, it is possible that amine contributes significantly to the  
800 hygroscopicity of 40 nm particles. Several studies have shown that amine compounds  
801 in aerosol phase can be hygroscopic, sometimes at even low RH (e.g. Qiu and Zhang,  
802 2012; Chu et al., 2015; Gomez-Hernandez et al., 2016). —

### 803 4.3. Diurnal variations in aerosol properties

#### 804 4.3.1. Diurnal variations in aerosol number and mass concentrations

805 Figure 6a shows the diurnal variation in MPL-derived PBL height. PBL height  
806 can be determined at the altitude where a sudden decrease in the scattering coefficient  
807 occurs from the MPL data (Cohn and Angevine, 2000; Brooks, 2003). Note that the  
808 retrieved PBL height is only valid from 07:00 local time (LT) to 19:00 LT (Quan et al.,  
809 2013). The retrieved PBL height at night is not accurate because of the likely  
810 influence of residual aerosols within the nocturnal PBL. The evolution of PBL height  
811 from 07:00 LT to 19:00 LT is sufficient to analyze its link with the change in aerosol  
812 number and mass concentrations during the daytime. Figure 6b shows diurnal  
813 variations in aerosol number and mass concentrations in the 15–685 nm range ( $N_{15-685}$   
814 nm and  $PM_{15-685}$  nm, respectively). Variations in the  $N_{15-685}$  nm and  $PM_{15-685}$  nm trendsed  
815 oppose each other. From 08:00 LT to 14:00 LT, the PBL height ~~lifts~~lifted from ~0.5

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816 km to ~0.6 km, while  $PM_{15-685\text{ nm}}$  ~~decreaseds~~ from  $\sim 24\ \mu\text{g m}^{-3}$  to  $\sim 19\ \mu\text{g m}^{-3}$  although  
817 there ~~wais~~ a slight increase at the beginning of the period. This suggests the important  
818 effect of PBL evolution on  $PM_{15-685\text{ nm}}$ . However,  $N_{15-685\text{ nm}}$  sharply ~~increase~~s from  
819  $\sim 7600\ \text{cm}^{-3}$  at 07:00 LT to  $\sim 13000\ \text{cm}^{-3}$  at 13:00 LT. This is related to the sudden burst  
820 of ~~nucleations~~ small Aitken mode particles ( $< 100-50\ \text{nm}$ ) when NPF events occurred.  
821 Newly formed fine particles contribute little to  $PM_{15-685\text{ nm}}$ . In the evening,  $PM_{15-685\text{ nm}}$   
822 ~~increaseds~~ gradually while  $N_{15-685\text{ nm}}$  ~~decreases~~ ~~decreased~~. This is attributed to the  
823 declining trend in the nocturnal PBL and particle coagulation and growth. In other  
824 words, the evolution of the PBL ~~plays~~ ~~played~~ a dominant role on the aerosol mass  
825 concentration, while particle formation and growth ~~has~~ ~~had~~ a greater influence on the  
826 variation in aerosol number concentration.

#### 827 4.3.2. Diurnal variation in aerosol hygroscopicity

828 Figure 6c shows diurnal variations in  $\kappa_{\text{gf}}$  and  $\kappa_{\text{chem}}$ . All sized  $\kappa_{\text{gf}}$  ~~increases~~  
829 ~~increased~~ beginning from the NPF event, especially for the 40 nm particles. The  
830 increase of  $\kappa_{\text{gf}}$  in the morning ~~wawas~~ synchronous with the particle number  
831 concentration ( $N_{15-685\text{ nm}}$ ) but not with the PBL height, further suggesting the impact  
832 of photochemical reactions on aerosol hygroscopicity. The  $\kappa_{\text{gf}}$  for 40 nm particles  
833 ~~increases~~ ~~increased~~ from  $\sim 0.32$  at 07:00 LT to  $\sim 0.44$  at 15:00 LT, and ~~approaches~~  
834 ~~approached~~ the  $\kappa$  of pure ammonium sulfate, ~~also~~ ~~[\kappa\_{\text{gf}}(\text{NH}\_4)\_2\text{SO}\_4 = 0.48~~ (Wu et al.,  
835 ~~2016]). This suggest~~ ~~ings~~ that a large amount of ~~hygroscopic compounds~~ ~~sulfates~~ were  
836 produced ~~through the photochemical reactions of precursors~~ ~~during NPF events~~. Fig.

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837 ~~S5 in the supplement shows. This can be verified from the diurnal variation in the~~  
838 ~~sharply increased concentrations of SO<sub>2</sub> and VOCs in the morning and the enhanced~~  
839 ~~atmospheric oxidation capacity under high RH and low T made plenty of sulfate and~~  
840 ~~SOA produced. This is the reason in the increase of aerosol hygroscopicity and the~~  
841 ~~frequent occurrence of NPF events~~ chemical composition (Fig. 6d), which shows that  
842 ~~the mass fraction of sulfate increased.~~ Detailed characterization of aerosol chemistry  
843 ~~during NPF events in this campaign has been studied in Zhang et al. (2018). T-during~~  
844 ~~the daytime while the mass fraction of organics (POA or SOA) varied more weakly.~~  
845 ~~The diurnal variation in SO<sub>2</sub> precursor also shows this (Fig. S4).~~ The diurnal variation  
846 ~~pattern~~ in  $\kappa_{gf}$  for 80–200 nm particles ~~differs~~ from that of 40 nm particles. The  
847 differences in  $\kappa_{gf}$  between 80–200 nm particles in the early morning ~~are were~~ large but  
848 gradually decrease as the sun rises. After 11:00 LT, the  $\kappa_{gf}$  for 80–200 nm particles ~~are~~  
849 ~~were~~ similar but lower than that of 40 nm particles. ~~All these,~~ suggest~~ing that~~ the  
850 enhanced hygroscopicity in the ~~80~~40–200 nm particles ~~was~~ likely~~ly~~ caused by  
851 the condensation of sulfates and secondary organics (Fig. 6d) ~~while and the effect was~~  
852 ~~more significant for 40 nm particles~~ that of the 40 nm particles ~~was caused by the~~  
853 ~~growth of the new particles.~~

854 Figure 6c also shows that the  $\kappa_{chem}$  for PM<sub>1</sub> ~~is was~~ lower than the  $\kappa_{gf}$  for 40–  
855 200 nm particles and ~~has had~~ a weaker diurnal variation. ~~This feature was stronger at~~  
856 ~~noon when atmospheric oxidation and the aging process were more rapid. Two~~  
857 ~~reasons may explain this: (1) the bulk chemical compositions of PM<sub>1</sub> and of 40–200~~  
858 ~~nm particles differ greatly and (2) The difference was mainly induced by the simple~~

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859 ~~ZSR mixing rule. the ZSR model cannot describe the impact of condensation on~~  
860 ~~aerosol hygroscopicity very well.~~ During the daytime, the condensation of sulfuric  
861 acid on organics or BC greatly enhances their hygroscopicity (Zhang et al., 2008;  
862 Zhang et al., ~~2009~~2017). ~~This phenomenon can't be described accurately by the ZSR~~  
863 ~~model.~~ Cruz and Pandis (2000) have shown that the measured  $\kappa_{gf}$  of internally  
864 mixed  $(\text{NH}_4)_2\text{SO}_4$ -organic aerosols is larger than the predicted  $\kappa_{chem}$  based on the  
865 ZSR model.

866 In summary, the ample supply of effluent  $\text{SO}_2$  and VOCs provided ~~de~~ sufficient  
867 precursors for the strong photochemical reactions at XT during ~~this campagne-field~~  
868 ~~experiment,~~ and the produce ~~and d sulfates and the~~ condensation of sulfate ~~and~~  
869 ~~SOAuric acid~~ enhanced ~~de~~ aerosol hygroscopicity largely, especially during the daytime.  
870 This also suggests that the observed frequent NPF events ~~were-were~~ mainly induced  
871 by the oxidation of precursors.

#### 872 4.3.3. Diurnal variation in CCN number concentration and activation ratio

873 Figure 7a shows the diurnal variations in  $N_{CCN}$  and AR at different SS. In the  
874 morning,  $N_{CCN}$  first ~~decreases-decreased~~ then ~~increases-increased~~ while AR ~~shows~~  
875 ~~showed~~ the opposite trend. This is related to the evolution of the PBL and NPF events.  
876 At the initial stage of an NPF event, the newly formed particles ~~are-were~~ less than 15  
877 nm in size, which ~~is-was~~ below the detection limit of the SMPS. As a result,  $N_{15-685\text{ nm}}$   
878 ~~decreases-decreased~~ (Fig. 6b) as the PBL lifts and  $N_{CCN}$  also ~~decreasesdecreased~~.  
879 However, the mixing of aged particles within the PBL ~~madekes~~ the particle size (Fig.

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880 7b) and AR increase slightly. With condensation and the growth of new particles, the  
881 number of fine particles detected by the SMPS increased rapidly but a portion of them  
882 cannot be activated because their smaller size. Therefore, So  $N_{CCN}$  increases-increased  
883 but AR decreases-decreased from 08:00 LT to 14:00 LT. In the afternoon and evening,  
884  $N_{CCN}$  and AR increased slightly with the increase in particle size (Fig. 7b). However,  
885 these trends become-became weaker as SS decreases-decreased, this is likely because  
886 the critical diameter is larger at low SS and the influence of aerosol size distribution  
887 on  $N_{CCN}$  and AR is relatively weaker. This demonstrates that the particle size is-was  
888 the most important factor influencing the aerosol activation ability and the CCN  
889 number concentration, especially at larger SS levels. The sensitivity test of particle  
890 size in CCN closure study similar with that in Dusek et al., (2006) was shown in Fig.  
891 S6.

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#### 892 4.4. CCN estimation from chemical composition data

893 The three main factors influencing CCN activation are particle size, mixing state,  
894 and chemical composition. As discussed in the above sections, particles are-were  
895 highly internally mixed at XT and particle size has-had a great influence on  $N_{CCN}$ . In  
896 this section, a CCN closure study is-is conducted and the impact of chemical  
897 composition on  $N_{CCN}$  is discussed. Figure 8a shows estimated  $N_{CCN}$  as a function of  
898 measured  $N_{CCN}$  using real-time  $\kappa_{chem}$ . The estimated  $N_{CCN}$  correlates well with  
899 measurements ( $R^2 \geq 0.85$ ) but is generally overestimated. The slope of each linearly  
900 fitted line is greater than 1.10 and increases with increasing SS. In addition, the

901 relative deviation (RD) increases from 16.2 % to 25.2 % as SS increases from 0.13 %  
902 to 0.75 %, suggesting that estimates become worse at larger SS. The overestimation of  
903  $N_{CCN}$  is mainly caused by large measurement uncertainties of CCNC: (1) the  
904 temperature or high flow rates in the CCNC may not allow enough time for particles  
905 to reach sizes large enough to be counted by the OPC at the exit of the CCN chamber  
906 (Lance et al., 2006; Cubison et al., 2008) and (2) in high particle number  
907 concentration environments, water depletion in the CCNC may reduce the counting  
908 rate of the CCNC (Deng et al., 2011). These uncertainties make measured  $N_{CCN}$  lower  
909 than the actual  $N_{CCN}$ . At larger SS, those activated aerosols in the cloud chamber of  
910 CCNC are greater in number and smaller in size, so the impact of these uncertainties  
911 is greater. ~~The separated  $N_{CCN}$  closure study is shown in Another discussion about this  
912 problem can be found in the supplement (Fig. S5S7). Figure S7 suggests the CCN  
913 closure is very good when  $N_{CCN} < 5500 \text{ cm}^{-3}$ , reflecting the validation of the CCN  
914 closure method in this study.~~

915 Figure 8b shows estimated  $N_{CCN}$  using the mean value for  $\kappa_{chem}$  ( $\kappa_{chem} = 0.31$ ).  
916 Compared with results using real-time values for  $\kappa_{chem}$ , the fit parameters and RD  
917 change slightly, suggesting that the effect of chemical composition on  $N_{CCN}$  is weaker  
918 relative to the particle size. The sensitivity of estimated  $N_{CCN}$  to the variability in  
919 chemical composition ( $\kappa_{chem}$ ) is further investigated (Fig. 9). In this figure, the  
920 variability of the equipotential lines in RD suggests that the sensitivity of  $N_{CCN}$  is  
921 strongly time dependent. This is attributed to the variability of the shape of the aerosol  
922 size distribution (Juranyi et al., 2010), ~~further verifying the importance of particle size~~

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923 ~~to  $N_{CCN}$~~ . The sensitivity of  $N_{CCN}$  to chemical composition ( $\kappa_{chem}$ ) becomes weaker  
924 with increasing SS, suggesting that chemical composition becomes less important in  
925  $N_{CCN}$  estimates at larger SS. In addition, the RD is always less than 10 % when  
926 estimating  $N_{CCN}$  using the mean value of  $\kappa_{chem}$ , suggesting that  $\kappa = 0.31$  is a good  
927 proxy for chemical composition when estimating  $N_{CCN}$  at XT.

928 In summary, particle size is the most important factor influencing the aerosol  
929 activation ability at XT, especially at larger SS levels. The mixing state and chemical  
930 composition were not as important when estimating  $N_{CCN}$  because the particles were  
931 highly aged and internally mixed at XT ~~during the field experiment~~, and aerosol  
932 hygroscopicity was not sensitive to estimates of  $N_{CCN}$ .

## 933 5. Summary and conclusions

934 The Atmosphere-Aerosol-Boundary Layer-Cloud ( $A^2BC$ ) Interaction Joint  
935 Experiment was carried out at a ~~polluted-suburban~~ site located in the central North  
936 China Plain (NCP) from 1 May to 15 June of 2016. The aerosol hygroscopicity,  
937 mixing state and CCN activity at the site Xingtai (XT) ~~were-were~~ investigated in this  
938 study.

939 In general, the probability density function of the hygroscopicity parameter  
940 ( $\kappa$ -PDF) for 40–200 nm particles ~~is-was~~ a unimodal distribution, which is different  
941 from distributions at other sites in China. Particles of all sizes covered a large range of  
942  $\kappa_{gf}$  (mostly from 0 to 0.8) and showed similar  $\kappa$ -PDF patterns, suggesting that the  
943 hygroscopic compounds in these particles from 40 nm to 200 nm were similar at XT.

944 The  $\kappa$ -PDF patterns also suggests that the particles were highly aged and internally  
945 mixed at XT during the ~~field experiment~~this campaign. This is likely related to strong  
946 photochemical reactions.

947 The mean  $\kappa_{gf}$  for different particle sizes ~~are~~were larger in the daytime than at  
948 night. Daytime and nighttime  $\kappa_{gf}$  differences ~~decreased~~ with increasing particle size.  
949 This illustrates that the impact of photochemical reactions on aerosol hygroscopicity  
950 ~~was~~was strong and that the effect ~~became~~became weaker as particle sizes  
951 ~~increased~~increases. The enhanced hygroscopicity of 40–200 nm particles ~~was~~was  
952 likely caused by the coating of sulfates or secondary organics ~~while~~and the effect ~~was~~  
953 was more significant for ~~–~~ 40 nm particles. Compared with other sites in China, the  
954 aerosol hygroscopicity ~~was~~was much larger at XT because of ~~the strong~~  
955 ~~photochemical reactions and~~ the sufficient precursors and strong atmospheric  
956 oxidation capacity. The comparison also shows that the hygroscopicity of particles  
957 from different emissions and chemical processes differed largely.

958 New particle formation events occurred frequently at XT during ~~the field~~  
959 ~~experiment~~this campaign. The evolution of the planetary boundary layer (PBL) ~~played~~  
960 played a dominant role on the aerosol mass concentration, while particle formation  
961 and growth ~~had~~had a greater influence on the variation in the aerosol number  
962 concentration. Particle size ~~was~~was the most important factor influencing the aerosol  
963 activation ability and the CCN number concentration at XT during the field  
964 experiment, especially at larger supersaturations (SS). Although the estimated  $N_{CCN}$   
965 ~~correlated~~correlates well with measurements ( $R^2 \geq 0.85$ ),  $N_{CCN}$  ~~was~~is

966 overestimated because of measurement uncertainties. The effect of chemical  
967 composition on  $N_{CCN}$  ~~was~~is weaker relative to the particle size. Sensitivity tests show  
968 that the impact of chemical composition on  $N_{CCN}$  ~~became~~becomes weaker as SS  
969 ~~increased~~increases, suggesting that the effect of chemical composition on  $N_{CCN}$   
970 estimates is less important at larger SS. The value  $\kappa = 0.31$  is a good proxy for  
971 chemical composition when estimating  $N_{CCN}$  for the model at XT.

972 Our results show that aerosol properties in the middle of the NCP differ from  
973 those in the northern part of the NCP and other regions in China. This is because there  
974 are more industrial emissions in the central NCP. The plenitude of gas precursors and  
975 strong photochemical reactions at XT make aerosol properties there different from  
976 those at sites under other polluted conditions. More field measurements on  
977 gas-particle transformation and aerosol properties in this region are needed, which  
978 would be meaningful for studying the haze formation mechanism and climate change  
979 in the NCP.

980  
981 *Data availability.* The data used in the study are available from the first author upon  
982 request ([wang.yuying@mail.bnu.edu.cn](mailto:wang.yuying@mail.bnu.edu.cn)).

983  
984 *Competing interests.* The authors declare that they have no conflict of interest.

985  
986 *Author contribution.* Z.L. and Y.W. designed the experiment, Y.W., Y.Z., and W.D.  
987 carried it out and analyzed the data, other co-authors participated in science discussions  
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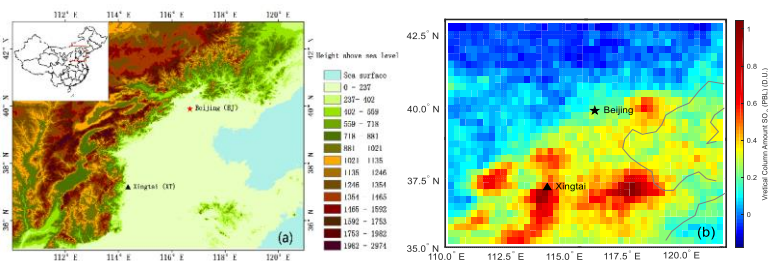
1205  
 1206 **Table 1.** The number fractions of different hygroscopic groups for different particle  
 1207 sizes.

	40 nm	80 nm	110 nm	150 nm	200 nm
NH	5.1 %	5.0 %	5.1 %	5.0 %	5.7 %
LH	4.8 %	4.2 %	4.3 %	4.7 %	5.1 %
MH	90.1 %	90.8 %	90.6 %	90.3 %	89.2 %

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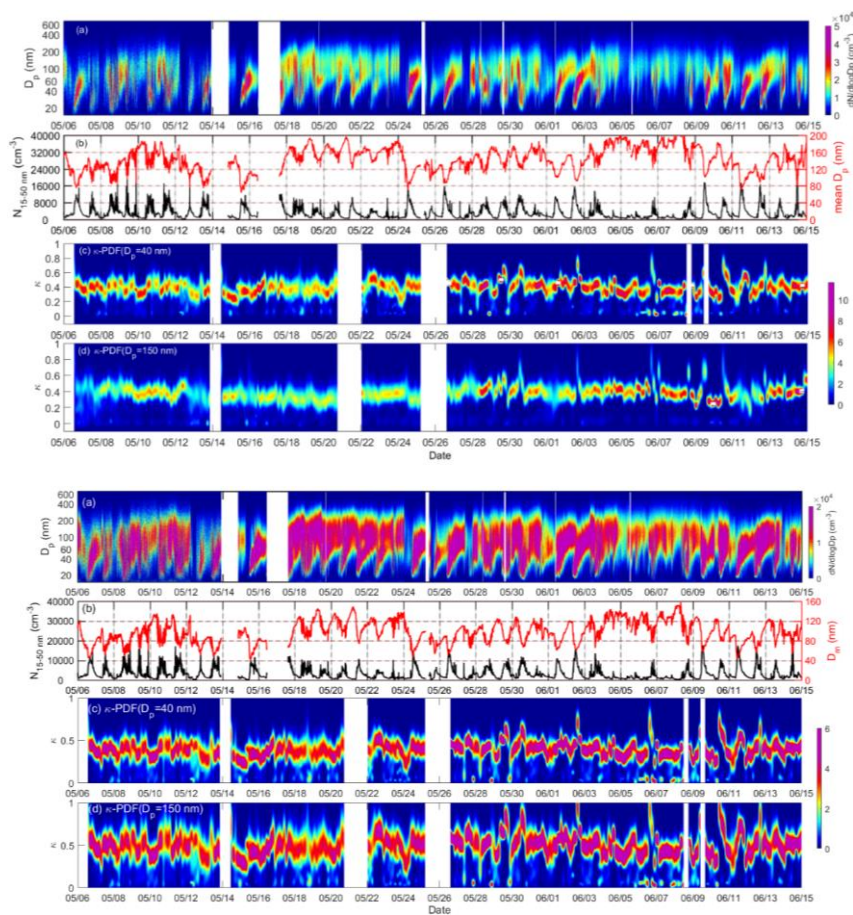
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**Figure 12.** (a) Map showing the locations of the sampling sites and (b) the distribution of mean SO<sub>2</sub> concentrations ~~offrom~~ ~~from-of~~ May ~~from-of~~ 2012 to 2016.

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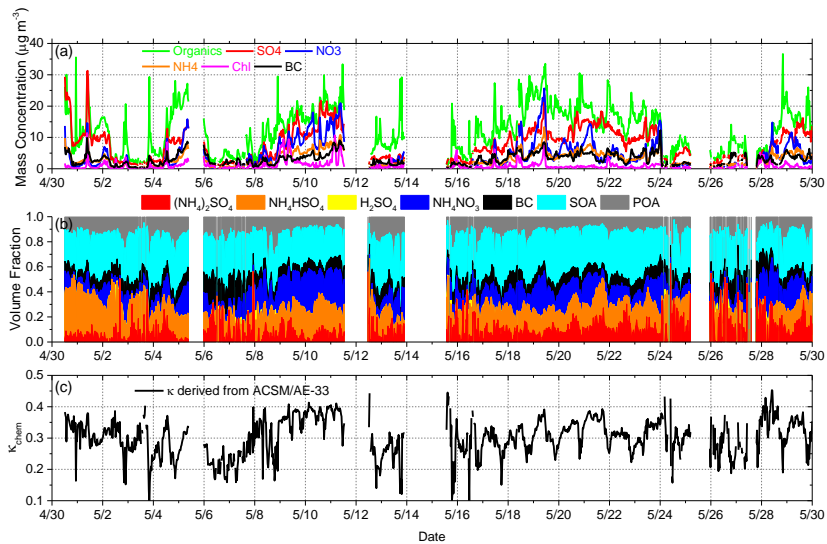
1217 **Figure 13.** The time series of (a) particle number size distribution (PNSD), (b)  
 1218 aerosol number concentration in the 15–50 nm range ( $N_{15-50 \text{ nm}}$ ) and the geometric  
 1219 mean diameter ( $D_{mp}$ ), (c) the probability density function of  $\kappa_{gf}$  ( $\kappa$ -PDF) for 40 nm  
 1220 and (d) 150 nm particles from 6 May to 15 June of 2016.

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1224 **Figure 14.** Time series of (a) the bulk mass concentration of aerosol species in  $\text{PM}_{10}$ ,

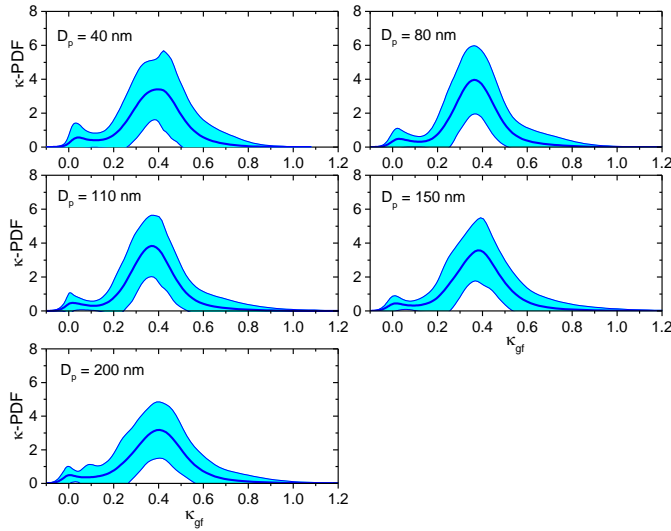
1225 (b) the volume fractions of POA, SOA, BC, and inorganics with the simplified ion

1226 pairing scheme, and (c) the hygroscopicity parameter derived from the chemical

1227 compositions ( $\kappa_{\text{chem}}$ ).

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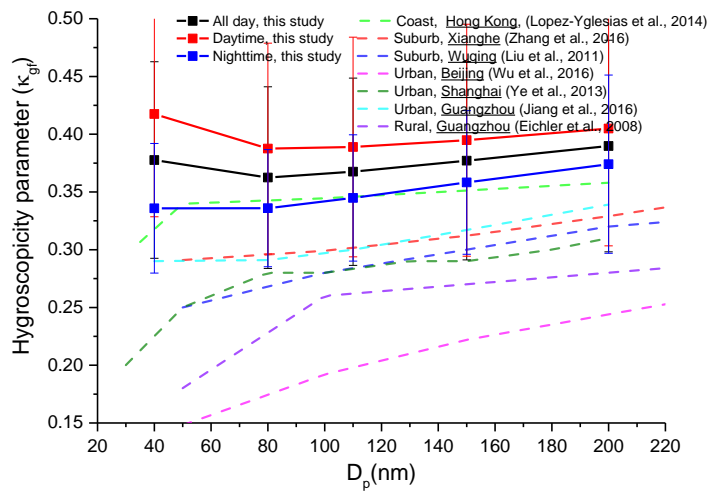
1230 **Figure 15.** Mean probability density functions of  $\kappa_{gf}$  ( $\kappa$ -PDF) for different particle

1231 sizes and their standard deviations (shaded areas) derived from H-TDMA data and

1232 measured at RH = 85 %.

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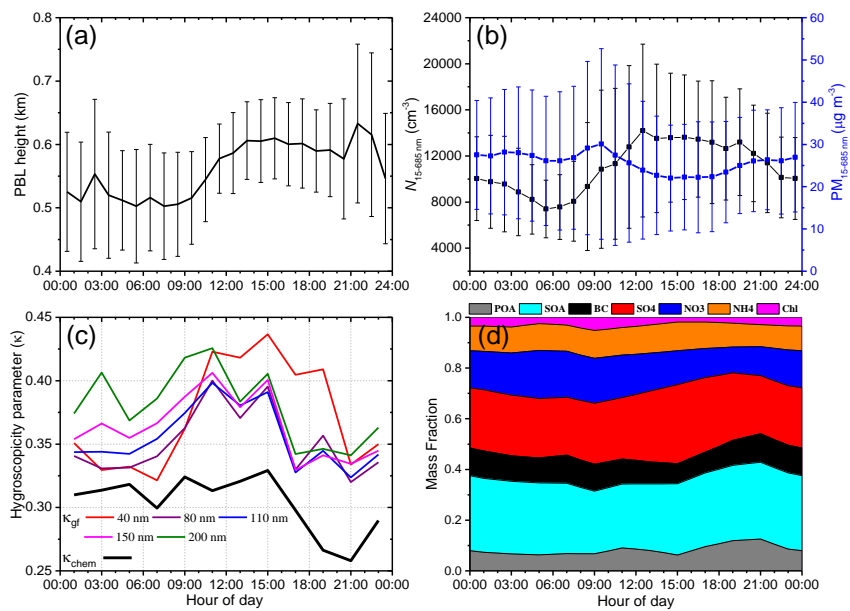
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1235 **Figure 16.** Size-resolved aerosol hygroscopicity parameter ( $\kappa_{gf}$ ) derived from

1236 H-TDMA data at XT and at other sites in China.

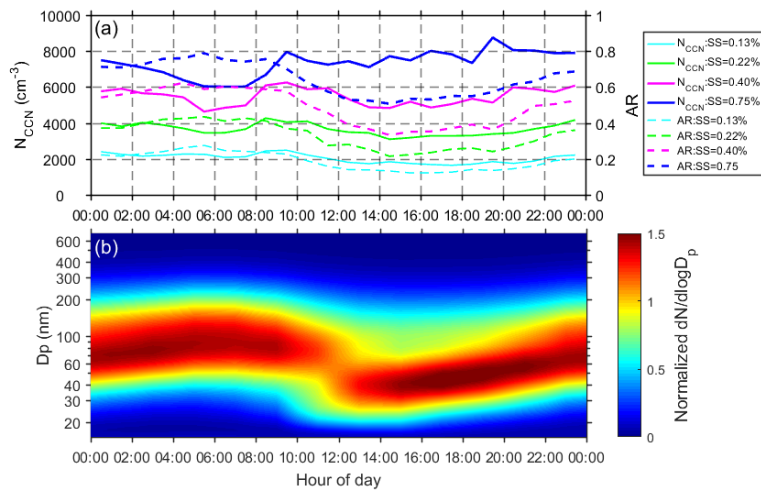
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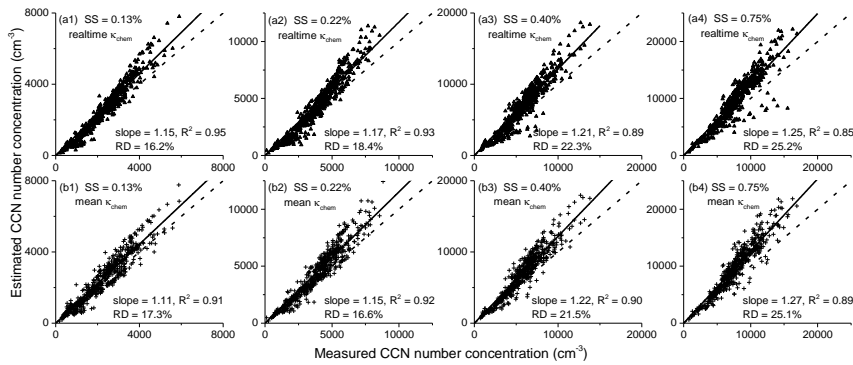
1238  
 1239 **Figure 17.** Diurnal variations in (a) planetary boundary layer (PBL) height retrieved  
 1240 from the MPL, (b) aerosol number and mass concentrations in the 15–685 nm  
 1241 ( $N_{15-685\text{ nm}}$  and  $\text{PM}_{15-685\text{ nm}}$ , respectively) derived from the SMPS (an aerosol density  
 1242 of  $1.6\text{ g cm}^{-3}$  is assumed), (c) the hygroscopicity parameter derived from the  
 1243 hygroscopic growth factor ( $\kappa_{\text{gf}}$ ) and predicted from the bulk chemical composition  
 1244 ( $\kappa_{\text{chem}}$ ), and (d) the mass fractions of different species.  
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 1247 **Figure 18.** Diurnal variations in (a) CCN number concentration ( $N_{CCN}$ ) and activation  
 1248 ratio (AR), and (b) the normalized aerosol size distribution in the 15–685 nm particle  
 1249 size range.

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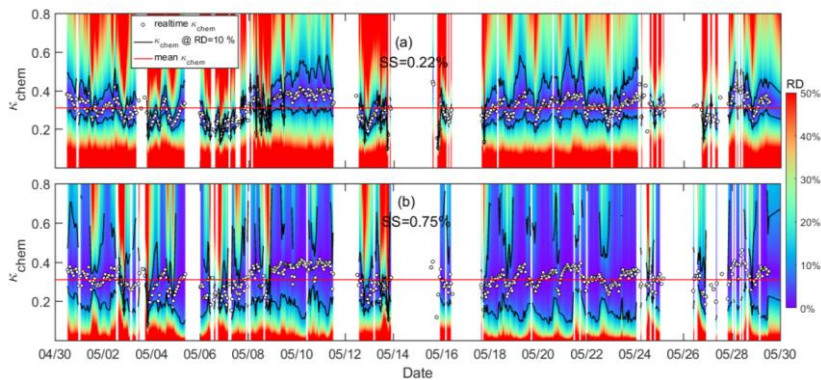


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 1253 **Figure 19.** Estimated versus measured CCN number concentration for ambient  
 1254 aerosols at four different supersaturation levels. The  $N_{CCN}$  is estimated based on  
 1255  $\kappa$ -Köhler theory, using the real-time  $\kappa_{chem}$  (a1-a4) and the mean  $\kappa_{chem}$  (b1-b4).  
 1256 The slope and correlation coefficient ( $R^2$ ) of the linear regression, and the relative

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1257 deviation of estimated  $N_{CCN}$  ( $RD = |N_{CCN\_estimated} - N_{CCN\_measured}| / N_{CCN\_measured}$ ) are  
 1258 shown in each panel. The regression line is overlaid on the measurements (solid line)  
 1259 and the dashed line is the 1:1 line.

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1262  
 1263 **Figure 20.** Sensitivity of  $N_{CCN}$  estimates to  $\kappa_{chem}$  as a function of time at (a)  $SS =$   
 1264  $0.22\%$  and (b)  $SS = 0.75\%$ . The color scale indicates the relative deviation (RD) of  
 1265 the CCN estimates using the  $\kappa_{chem}$  value shown on the ordinate. In each panel, open  
 1266 circles show the real-time  $\kappa_{chem}$ . Note that RD is by definition zero at these points.  
 1267 The black line is  $\kappa$  at  $RD = 10\%$  and the red line is the mean value for  $\kappa_{chem}$  ( $0.31$ ).  
 1268 Figure S6-S8 in the supplement shows the same plots but for  $SS = 0.13\%$  and  $0.40\%$ .  
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