



1 Vertical characterization of Highly Oxygenated Molecules

(HOMs) below and above a boreal forest canopy

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23 and secondary organic aerosol (SOA) formation is not in dispute, the interplay between 24 HOM chemistry and atmospheric conditions continues to draw significant research 25 attention. During the Influence of Biosphere-Atmosphere Interactions on the Reactive 26 Nitrogen budget (IBAIRN) campaign, profile measurements of neutral HOM molecules 27 below and above the forest canopy were performed for the first time in the boreal forest 28 SMEAR II station during September 2016. The HOM concentrations and composition 29 distributions below and above the canopy were similar, supporting a well-mixed 30 boundary layer approximation during daytime. However, much lower HOM





31 concentration were frequently observed at ground level due to the formation of a 32 shallow decoupled layer below the canopy attached to the forest floor. Near ground 33 HOMs were influenced by the changes in the precursors and oxidants, and enhancement 34 of the loss on surfaces in this layer, while the HOMs above the canopy top were not 35 significantly affected. Our findings also illustrate that near-ground HOM measurements 36 conducted in strong stably stratified conditions might only be representative of a small 37 fraction of the entire nocturnal boundary layer. This might, in turn, influence the growth 38 of newly formed particles and SOA formation below the canopy where a large majority 39 of measurements are typically conducted.

40 2 Introduction

41 Highly oxygenated molecules (HOMs), a sub-group of the oxidation products of 42 volatile organic compounds (VOCs) identified by their high oxidation states, have been 43 recognized as important precursors for organic aerosol in the atmosphere (Ehn et al., 44 2014). They have also been found to enhance new particle formation (NPF) and growth 45 (Kulmala et al., 2013; Zhao et al., 2013; Ehn et al., 2014; Bianchi et al., 2016; Kirkby 46 et al., 2016; Tröstl et al., 2016). The importance of HOMs has been confirmed in 47 ambient environments, especially in monoterpene-dominated regions such as the boreal 48 forest (Kulmala et al., 2013; Ehn et al., 2014), but also in high altitude mountain regions 49 (Bianchi et al., 2016) and in rural areas (Jokinen et al., 2014a; Kürten et al., 2016). In 50 laboratory studies, HOM formation has been observed from various precursor 51 molecules (Ehn et al., 2017), including both biogenic and anthropogenic emissions.

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The direct observation of HOMs has only recently become possible, following the developments of the Atmospheric-Pressure-interface Time-Of-Flight (APi-TOF, measures the charged HOM clusters) (Junninen et al., 2010) and Chemical Ionization Atmospheric-Pressure-interface Time-Of-Flight (CI-APi-TOF, measures the neutral HOM molecules) (Jokinen et al., 2012) mass spectrometers. Ehn et al. (2010) and





- 58 Bianchi et al. (2017) found that the naturally charged HOM clusters could be observed 59 every night in boreal forest during spring. Out of the observed ambient mass spectra, a 60 significant part could be reproduced in a chamber by introducing the monoterpene α -61 pinene (C₁₀H₁₆, the major biogenic VOC in the boreal forest) and ozone (O₃) (Ehn et 62 al., 2012).
- 63

64 Further investigations of HOM formation chemistry have been done in both laboratory 65 and field studies. Based on current understanding from laboratory experiments, the 66 formation of HOM molecules involves three main steps: 1) initial formation of peroxy radicals (RO₂)from VOC oxidation; 2) RO₂ auto-oxidation, that is, the isomerization of 67 68 the RO_2 via intramolecular H-shifts and the subsequent oxygen (O_2) additions; and 3) 69 radical termination, forming closed-shell molecules (Crounse et al., 2013; Ehn et al., 70 2014; Jokinen et al., 2014b, 2016; Rissanen et al., 2014; Mentel et al., 2015). In the 71 atmosphere, HOM formation studies are complicated by the plethora of different 72 compounds and processes taking place. However, recent ambient measurements 73 together with factor analysis were able to shed light on the HOM formation pathways 74 in the boreal forest (Yan et al., 2016). They showed that the majority of the daytime 75 production of HOMs was from reactions initiated by the oxidation of monoterpenes 76 (MT) with hydroxyl radical (OH) or O₃. The RO₂ after auto-oxidation was either 77 terminated by hydroperoxyl radical (HO2) or self-termination (Orlando and Tyndall, 78 2012), to form a non-nitrate HOM monomer (CHOmonomer, mainly C9 and C10 79 compounds, masses between 290-450 Th after clustering with the charging ion (NO₃⁻) 80 of the instrument); or reacting with nitrogen oxides (NO_x) to form an organonitrate 81 HOM monomer (CHON_{monomer}). During nighttime, MT were mainly oxidized by O₃ 82 and NO3 radical. Furthermore, due to the lower nocturnal HO2 and NOx concentrations, 83 besides the production of CHONmonomer, the RO2 products could also react with another 84 RO2 to form a non-nitrate HOM dimer (CHOdimer, mainly C16-20 compounds, masses 85 between 450-600 Th after clustering with NO3) or an organonitrate HOM dimer





86 (CHON_{dimer}), depending on the oxidants of the RO₂ radical. (Ehn et al., 2014; Yan et

- 87 al., 2016).
- 88

89 Beyond those chemical pathways, varied meteorological conditions are also factors 90 influencing the MT and oxidants at different heights above the forest floor. 91 Unsurprisingly, the oxidants of HOMs (e.g. O₃) were found almost uniformly 92 distributed within the well-mixed daytime boundary layer (Chen et al., 2017). In 93 contrast, the nocturnal boundary layer was shallow with stability regimes that depended 94 on radiative cooling within the canopy and turbulent shear stresses at the canopy top. 95 In Hyytiälä, the depletions of O₃ in the sub-canopy level were frequently observed 96 during nighttime, while the O₃ above the canopy was less affected. The MT 97 concentration at ground level increased when O₃ was depleted (Eerdekens et al., 2009). 98 The inhomogeneous distribution of the precursors and oxidants below and above the 99 canopy might further impact nocturnal HOM distributions, which frames the scope of 100 this study. Until now, all CI-APi-TOF deployments have been at ground level, and the 101 main subject of inquiry here is the vertical information on HOMs and the role of 102 meteorological condition in shaping them. A characterization of the HOMs at different 103 heights provides a decisive advantage in disentangling the role of non-uniform mixing 104 within the atmospheric layers impacted by strong thermal stratification, especially 105 inside the canopy volume.

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107 The first measurements of the HOM concentrations at two different heights (36 m and 108 1.5 m a.g.l.) during September 2016 are presented and discussed. The influence of 109 boundary layer dynamics on the HOMs at these different heights are explicitly analyzed 110 and characterized in conjunction with auxiliary turbulence and micrometeorological 111 measurements.





112 **3** Experimental

113 **3.1 Measurement site description**

114 The measurements were performed at the SMEAR II station (Station for Measuring 115 Ecosystem-Atmosphere Relations) in the boreal forest in Hyytiälä, southern Finland 116 (61°51' N, 24°17' E, 181 m a.s.l., Hari and Kulmala, 2005; Hari et al., 2013) during 117 September 2016. There is no large anthropogenic emission source at or near the site. 118 The closest sources are the two sawmills ~5 km southeast from the site and emission 119 from the city area of Tampere (~60 km away). The forest surrounding the station is 120 primarily Scots pine with a mean canopy height of ~17.5 m (Bäck et al., 2012). The planetary boundary layer height at the SMEAR II station has been determined from 121 122 previous studies using radiosondes (Lauros et al., 2007; Ouwersloot et al., 2012) and 123 balloon soundings (Eerdekens et al., 2009). Roughly, these heights span some 400 m (March) to 1700 m (August) at noontime, and 100 m (March) to <160 m (April) at 124 125 midnight.

126 **3.2 Instrumentation**

127 Concentration of HOM molecules were measured with two nitrate-ion based CI-APi-128 TOF mass spectrometers. The CI-APi-TOF measuring at higher altitude was deployed 129 to the top of a 35 m tower located ~20 m horizontally from the ground measurement location. The inlets of the two instruments were pointed to the southeast direction and 130 131 fixed at \sim 36 m and \sim 1 m above ground. The tower measurement is about twice the 132 canopy height, which is still within the roughness sublayer of the forest (Raupach and Thom, 1981). The instrument setup of the two CI-APi-TOF mass spectrometers were 133 134 similar. In brief, the CI-APi-TOF was the combination of a chemical ionization (CI) 135 inlet, and an atmospheric pressure interface time-of-flight (APi-TOF) mass 136 spectrometer (Aerodyne Research Inc., USA, and Tofwerk AG, Switzerland). The 137 ambient air was first drawn into the inlet with a sample flow of 7 lpm (liter per minute),





138 and then converged to the center of an ion reaction tube surrounded by sheath flow 139 (filtered air, 35 lpm). Meanwhile, the nitrate ions carried by the sheath gas, which were 140 generated by exposing the nitric acid (HNO₃) to soft x-ray radiation, were guided into 141 the sample gas by an electrical field at ambient pressure (~ 100 ms reaction time). 142 Neutral molecules (M) in the sample air were ionized by either clustering with charged 143 nitrate/nitric acid ((HNO₃)_{n=0-2}·NO₃⁻) to form (M)·NO₃⁻ cluster ions, or losing a proton 144 to the charging ions to form deprotonated ions (e.g., $H_2SO_4+NO_3^-\rightarrow HSO_4^++HNO_3$). 145 The ions then entered the APi part, which was a three-stage vacuum chamber, through a pinhole. In the APi, two quadrupoles and a stack of ion lenses guide the ions into the 146 TOF mass spectrometer, where ions were separated based on their mass-to-charge (m/z)147 148 ratios. A more detailed description of this instrument has been given by Junninen et al. 149 (2010) and Jokinen et al. (2012). Mass spectra obtained from the instrument were analyzed using the 'tofTools' program described in Junninen et al. (2010). 150 151 Determination of the concentration of a measured molecule M was based on the 152 following equation:

153
$$[M] = \frac{M}{\sum reagent ion count rates} \times C$$
(1)

154 where the sum of ion count rates was an inclusion of all detected ions relating to 155 compound M, whether deprotonated or in clusters with reagent ions, and the sum of 156 reagent ion count rates is the total signal of the charged nitric acid ions. C was the 157 calibration coefficient, which was assigned the same value for all detected compounds. 158 This assignment is only valid for compounds that cluster with the reagent ions at the 159 collision limit, such as H₂SO₄ (Viggiano et al., 1997) and have equal collision rates. 160 The collision rates of nitrate ions with H₂SO₄ and with HOMs are expected to be very close (Ehn et al., 2014). Here, a calibration coefficient of 1×10^{10} molec cm⁻³, estimated 161 162 from previous calibrations with similar settings using sulfuric acid and theoretical constraints (Ehn et al., 2014), with an uncertainty of -50%/+100%, was used in 163 164 calculating the HOM concentrations for both instruments. Ultimately, the absolute HOM concentrations in this work are of secondary importance, as we focus on the 165





166 relative comparison of HOM concentrations measured at different heights. However, 167 the comparability of the two CI-APi-TOF instruments is of great importance, and 168 results cannot be allowed to vary e.g. as a result of inevitable differences in the mass-169 dependent transmission efficiency (TE). To this end, instead of directly evaluating the 170 TE of each instrument, a "relative" TE of the two CI-APi-TOFs was used for data 171 correction: we selected a time period at noon-time on September 9 with well-mixed 172 boundary layer condition, identified with the clear and sunny weather and 173 homogeneous vertical distribution of monoterpene and other trace gases, and assumed 174 the HOM concentrations at the two heights to be the same. Thus, the relative TE was obtained from the concentration ratio between the two CI-APi-TOFs at each m/z (Figure 175 176 1). Weaker correlation was obtained in the sub-150 Th mass range, but in the mass range where most of the HOMs were located (290-600 Th) there is very little scatter around 177 the fitted curve, clearly suggesting that observed differences in the two instruments 178 179 responses were mainly due to differences in TE. Additionally, an inter-comparison between the two instruments with a permeation tube was conducted after the campaign, 180 181 and the results showed good agreement with the relative TE, lending confidence to the 182 method used here. Finally, it should be noted that the difference in TE between the two 183 instruments was larger than one would normally expect, since the tower CI-APi-TOF 184 had been tuned for maximum sensitivity at the largest masses (at the expense of 185 transmission at the lower masses).







188 Figure 1 The relative TE curve between the two CI-APi-TOFs for data correction.

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In comparison to the direct determination of TE (Heinritzi et al., 2016), this method
increases the uncertainty in the quantification of HOM concentrations. However, as
mentioned, a more accurate knowledge of the exact HOM concentrations would not
influence the main findings of this study.

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195 The MT, trace gases, and meteorological parameters were continuously monitored at 196 the different heights (4.2 m, 8.4 m, 16.8 m, 33.6 m, 50.4 m, 67.2 m, 125 m) on a 126 m 197 mast ~100 m away from the location of the CI-APi-TOFs. The data at 4.2 m and 33.6 198 m were used in this study to represent the concentrations at near ground and tower level, 199 respectively. Ambient MT concentration was measured every third hour using a proton 200 transfer reaction mass spectrometer with the lowest detection limit of 1 pptv (PTR-MS, 201 Ionicon Analytik GmbH; Taipale et al., 2008). The O₃ concentration was measured with 202 an UV light absorption analyzer that had a lowest detection limit of 1 ppbv (TEI model 203 49C, Thermo Fisher Scientific, USA). The NOx measurement was conducted using a 204 chemiluminescence analyzer (TEI model 42C TL, Thermo Fisher Scientific, USA). The 205 lowest detection limit of the NO_x analyzer is 50 pptv. The CO₂ measurement was





206 performed using an infrared light absorption analyzer (URAS4 CO2, Hartmann & 207 Braun, Germany). The aerosol number concentration size distributions were obtained 208 with a twin differential mobility particle sizer (twin-DMPS) for the size range from 3-209 1000 nm (Aalto et al., 2001) at 8 m height above ground, and was used to calculate 210 condensation sink (CS) based on the method from Kulmala et al. (2001). Air 211 temperature was measured with PT-100 resistance thermometers. Air relative humidity 212 (RH) was measured with RH sensors (Rotronic Hygromet model MP102H with 213 Hygroclip HC2-S3, Rotronic AG, Switzerland). Global radiation (solar radiation in 214 wavelength range of 0.3-4.8 µm) was obtained with a Pyranometer (Reemann TP3, Astrodata, Estonia) above the canopy top at 18 m. All the data presented are at 10 min 215 216 averaging interval except for the MT (in 1-hour averaging interval).

217

218 4 Results and discussion

219 4.1 Data overview

220 The Influence of Biosphere-Atmosphere Interactions on the Reactive Nitrogen budget 221 (IBAIRN) campaign was conducted from September 1 to 25, 2016. After data quality 222 checks, only the measurements collected after September 5 were used. Figure 2 shows 223 the overall time series of the meteorological parameters measured at ground and tower 224 levels, including the temperature, RH, global radiation, concentrations of trace gases, 225 MT, and total HOMs. The weather was generally sunny and clear during the campaign 226 except for a few cloudy (September 10, 15, and 22-23) and drizzling (September 24 227 and 25) days. The mean air temperature and RH of 10.8 ± 3.3 °C and $87.3 \pm 13.4\%$ (1 σ 228 standard deviation) were observed at ground level, and 10.5 ± 3.0 °C and 88.2 ± 14.4 % 229 at tower level. The O₃ concentrations measured at ground and tower levels were $21.2 \pm$ 230 7.6 ppbv and 24.9 ± 6.0 ppbv, respectively. The air temperature, RH and O₃ measured 231 at the two heights were close to each other during daytime. The NO_x concentrations 232 were quite low throughout the campaign, the mean NO_x concentrations were mostly





- around the reported detection limit at 0.38 ± 0.35 ppbv (ground) and 0.44 ± 0.45 ppbv (tower), yet showed an overall good agreement between the measurements at the different heights. The MT concentrations at ground level (0.38 ± 0.34 ppbv on average)
- 236 was generally higher than that above the canopy level $(0.20 \pm 0.16 \text{ ppbv})$.
- 237





240 concentrations at the ground (green) and tower (brown) levels.





241

242	The total HOM concentration is representative for the overall concentration level of
243	HOMs, and is defined as the sum of the detected signals between ions from m/z 200 to
244	600 after removing the identified background peaks. The gaps in the ground total HOM
245	data were due to automatic zero-check. During the campaign, a significant difference
246	was found in the total HOM concentrations below and above the canopy (mean and
247	median concentrations of 1.1 \pm 1.7 $\times 10^8$ cm^{-3} and 7.6 \times 10 7 cm^{-3} at ground level, 1.7 \pm
248	$1.3\times10^8\text{cm}^{\text{-3}}$ and $1.3\times10^8\text{cm}^{\text{-3}}$ at tower level). The causes of these differences frame
249	the upcoming discussion.

250

4.2 Inter-comparison of total HOM concentrations

252 The total HOM concentrations at the two heights were not different during the day, 253 which validates the use of only one day of data for scaling the TE of the ground CI-254 APi-TOF to match the HOM signals. The good daytime agreement throughout the 255 campaign period also verifies that the response of each instrument stayed stable. Contrary to the daytime results, the total HOM concentration at ground level usually 256 257 diverged from the tower measurement in the nocturnal boundary layer. The 258 concentration below the canopy became even lower when temperature inversions were 259 observed, accompanied by a decreasing ground-level O3 and increasing MT 260 concentrations. Figure 3 shows the correlation between the total HOM concentrations 261 observed at two heights. Herein, good agreement could be found for the group of points representing the concentrations around noontime. The points indicating the nighttime 262 263 total HOM concentrations were scattered, and the ground concentrations were found to 264 be much lower than the tower ones.









267 Figure 3 Correlation between ground (x-axis) and tower (y-axis) measurements of the total HOM

268 concentrations. The black line denotes the 1:1 ratio. Color code indicates the sampling time of HOMs.

269

270 Figure 4a shows the mean spectra (in unit mass resolution, UMR, between m/z 200 – 271 600) obtained from the ground and tower HOM measurements. It is worth mentioning 272 that there might be some signals not attributable to HOMs in the plotted spectra, but 273 only in little proportion. Only selected periods (09:00-15:00 for daytime and 21:00-274 03:00 for nighttime, local winter time (UTC +2)) are included in the averaging period 275 to eliminate the effect of sunrise and sunset periods. During daytime, a good agreement 276 $(R^2 = 0.87)$ was obtained from the mass-by-mass comparison (shown in Figure 4b) using the UMR concentrations extracted from daytime mean spectra, suggesting a 277 278 uniform composition distribution in the daytime boundary layer condition. During 279 nighttime, the mean concentrations of all HOM molecules in the ground mean spectra 280 were much lower than the tower spectra. The HOM concentrations shown in the ground 281 and tower mean spectra were also less correlated in Figure 4c. Therefore, a logical 282 outcome is that the conditions below and above the canopy are experiencing different 283 turbulent mixing strength or source-sink regimes during night.







Figure 4 (a) Mean mass spectra with the averaging periods of daytime (09:00-15:00) and nighttime
(21:00-03:00) at ground and tower levels; (b) mass-by-mass comparison between ground and tower
UMR concentrations extracted from daytime mean spectra in Figure 4a; (c) mass-by-mass comparison
between ground and tower UMR concentrations extracted from nighttime mean spectra in Figure 4a.





290 4.3 Influence of nocturnal boundary layer dynamics and micrometeorological

291 processes

292 The nighttime HOMs at ground level are influenced by transport processes below the 293 canopy, since the total HOM concentrations were found much lower in the nights when 294 temperature inversions were observed. To further investigate the potential impact from 295 such micrometeorological phenomena on the ground HOMs, all the nights during the 296 campaign, after ruling out those with precipitation and instrument failures, were 297 selected (12 nights in total) and categorized into 2 types based on the appearance of the 298 temperature inversion: 1) the "non-inversion night" type included 6 nights when no 299 temperature inversion was recorded; 2) the "inversion night" type contained 6 nights 300 that had encountered temperature inversions, and the ground temperatures were 301 generally ~1 °C lower compared to tower temperatures during these nights.

302

303 4.3.1 Statistics of the "non-inversion night" and "inversion night" types

304 Table 1 shows the overall statistics including the mean, median, 25% percentile and 75% 305 percentile values of the temperatures, O₃, NO_x, MT and total HOM concentrations for 306 the "non-inversion night" and "inversion night" types. In the non-inversion nights, the 307 air below and above the canopy were relatively well-mixed. The mean and median 308 concentrations of the ground O_3 (21.3 \pm 7.7 ppbv and 21.8 ppbv) were close to the tower 309 values (24.9 ± 6.2 ppbv and 23.9 ppbv). The slight difference might be attributed to the 310 higher VOC emissions and larger sink near ground level. In contrast, during the 311 inversion nights, the mean total HOM concentration and O₃ at ground level were 312 generally much lower, only \sim 33% and \sim 69% of the tower concentrations, respectively. 313 Instead, the mean and median ground MT concentration (0.70 ± 0.28 ppbv and 0.70314 ppbv) were ~ 3 times higher than the tower ones (0.24 \pm 0.04 ppbv and 0.23 ppbv), 315 respectively. The measured NOx levels were similar in both types and different heights,





- though the ambient concentrations were close to the detection limit and therefore small
- 317 differences might not be observable.
- 318

319 4.3.2 Case study

320 Two individual nights representing the "non-inversion night" and "inversion night" 321 types were selected and further compared. Figure 5a shows the time series of the 322 meteorological parameters, trace gases and HOMs measured at ground and tower levels of one selected night of "non-inversion night" type (September 11-12, from 21:00 323 324 to 03:00). A number of measures can be used to assess the local atmospheric stability 325 conditions at a given layer. These measures are commonly based on either the Obukhov 326 length and its associated atmospheric stability parameter or a Richardson number (flux-327 based, gradient-based, or bulk). Because of its simplicity and the availability of high 328 resolution mean air temperature profiles, the bulk Richardson number (Ri) was used 329 here (Mahrt et al., 2001; Mammarella et al., 2007; Vickers et al., 2012; Alekseychik et 330 al., 2013). It is calculated using:

331
$$Ri = \frac{g\Delta\overline{\theta}\Delta z}{\overline{\theta}(\overline{u})^2}$$
(2)

where g is the gravitational acceleration, $\Delta \overline{\theta}$ and Δz are the mean potential 332 333 temperature (10 min averaging interval, same as measurement data) and height difference between the ground and tower levels, respectively, $\overline{\theta}$ and \overline{u} are the mean 334 335 potential temperature and mean wind velocity at tower level, respectively. During the 336 selected "non-inversion" night, Ri was generally positive but close to 0 (shown in Figure 5a), indicating a weakly stable and relatively well-mixed (i.e. $\Delta \overline{\theta} \rightarrow 0$) 337 338 condition (Mahrt, 1998; Mammarella et al., 2007). This was also confirmed using the 339 well correlated ground and tower MT and trace gases concentrations.





	Type		Non	ı-inversion nigl	ht			Ι	nversion night		
	Date		Septembe	r 6, 7, 9, 11, 15,	16, 21*			September	5, 8, 10, 12, 13	3, 14, 19**	
	Dammeteree	Temperature	03	NOx	MT	Total HOM	Temperature	03	NOx	MT	Total HOM
	Farameters	[°C]	[ppbv]	[vdqq]	[vdqq]	$[10^8 \text{ cm}^{-3}]$	[.c]	[ppbv]	[ppbv]	[ppbv]	$[10^{8} { m cm}^{3}]$
	Mean $\pm 1\sigma$ standard deviation	10.2 ± 2.6	24.9 ± 6.2	0.50 ± 0.46	0.31 ± 0.31	2.9 ± 1.9	9.5 ± 1.7	23.5 ± 2.4	0.42 ± 0.26	0.24 ± 0.04	2.4 ± 0.8
Tower	Median	10.9	23.9	0.38	0.17	2.8	9.2	23.3	0.34	0.23	2.3
	25%/75% percentile	7.9 / 12.4	21.3 / 27.4	0.16 / 0.67	0.15 / 0.24	2.2/3.2	8.1 / 10.9	22.0 / 25.3	0.22 / 0.54	0.21 / 0.26	1.8 / 2.8
	Mean $\pm 1\sigma$ standard deviation	10.6 ± 2.7	21.3 ± 7.7	0.43 ± 0.40	0.52 ± 0.74	1.6 ± 0.6	8.3 ± 2.2	16.2 ± 6.2	0.33 ± 0.24	0.70 ± 0.28	0.8 ± 0.4
Ground	Median	11.5	21.8	0.31	0.22	1.7	8.5	16.8	0.27	0.70	0.7
	25%/ 75% percentile	8.1 / 12.8	158.1 / 25.9	0.15 / 0.53	0.21 / 0.31	0.9/3.2	6.6/9.9	10.8/21.8	0.17 / 0.44	0.46 / 0.82	0.5 / 1.2
delieve	le on Sentember 5 o	nd 10		-							

Table 1 Summary of the "Non-inversion night" and "Inversion night" types. 341

*MT data not available on September 5 and 19.

**MT data not available on September 15 and 16. 342 343

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- 344 Selected HOM molecules representing the major HOM types (and formation pathways)
- 345 were summed up and categorized into 4 groups, as shown in Table 2. Each pathway
- 346 might be influenced differently by boundary layer dynamics and micrometeorological
- 347 processes. In this study, OH-initiated HOMs were assumed negligible due to the very
- 348 low OH level in the nocturnal boundary layer.
- 349

350	Table 2 Compositions of selected HOM molecules and their main oxidants (Yan et al., 2016).

	Molecule compositions	Main Main terminators		
	Molecule compositions	oxidants	Wall ter limators	
CHO _{monomer}	$C_{10}H_{14}O_7, C_{10}H_{14}O_9$	O ₃	Self-terminate or RO ₂	
CHON _{monomer}	$C_{10}H_{15}O_9N_1, C_{10}H_{15}O_{11}N_1$	O ₃ or NO ₃	NO or Self-terminate/RO ₂	
CHO _{dimer}	$C_{19}H_{28}O_{11}, C_{20}H_{30}O_{14}$	O ₃	RO ₂	
CHON _{dimer}	$C_{20}H_{32}O_{12}N_2,C_{20}H_{31}O_{13}N_1$	NO ₃	RO ₂	

351

All the HOM groups in Figure 5a show stable patterns, and good agreement is observed between the ground and tower measurements in the first half of the night. Variations were observed when air mass change occurred at around 01:00, as indicated by the drop of NO_x concentration and CS, and wind shift (not shown here). However, the HOM groups were still well correlated with each other, suggesting the unchanged well-mixed condition in the non-inversion night.

358

Figure 5b shows the time series of the trace gases, MT, and HOM groups during an "inversion night" case (September 8-9, from 21:00 to 03:00). *Ri* was generally higher during this night, and increased from ~0.03 (indicating weakly stable condition, Mammarella et al., 2007), at around midnight, to a maximum of ~1.13 (indicating very stable condition) in the remaining night period. Roughly, *Ri* in excess of unity indicate that stably stratified condition appreciably diminish the inverse turbulent Prandtl number (*Pr*) and the efficiency of turbulence to mix heat when compared to momentum





- 366 (Katul et al., 2014). The parameters measured above the canopy did not show strong *Ri*
- 367 fluctuations throughout the night, in contrast, significant variations were observed at
- 368 ground level.
- 369





371 Figure 5 (a) Time series of the selected "non-inversion night" case (September 11), and (b) Time series

372 of the selected "inversion night" case (September 8).





373

374 The ground O₃ concentration experienced a rapid decrease at midnight. In about an hour 375 (from 23:30-00:30), ground O_3 concentration dropped by more than half (from 20.1 376 ppbv to 8.6 ppbv), and CO₂ concentration increased as well (from 404 ppbv to 423 377 ppbv). To the contrary, the MT concentration at ground level was almost doubled (from 378 0.49 ppbv to 0.80 ppbv) during the same period. Theoretically, the enhancement of 379 HOM precursor and decrease of oxidant would compensate each other if the sink 380 remained the same, and the ground HOM concentration should also keep constant. 381 However, all the HOM groups showed significant decrease from midnight. In particular, the concentration of the CHO_{monomer} group dropped ~80%, from 8.6 \times 10⁶ cm⁻³ to 1.7 \times 382 383 10^6 cm⁻³, and the concentration of the CHO_{dimer} group decreased from 1.5×10^6 cm⁻³ to $\sim 1.0 \times 10^5$ cm⁻³. The concentrations of the CHON_{monomer} and CHON_{dimer} groups also 384 experienced large declines (~34% and ~50%, respectively), in the latter half of the night. 385 386 At 03:00, the CHON_{dimer} concentration was already below the detection limit (1×10^4) cm⁻³). Therefore, the much lower ground HOM concentration might not be totally 387 388 explained by the change of HOM production, but also some other processes.

389

390 A previous study by Alekseychik et al. (2013) showed that nocturnal decoupled air 391 layers were frequently (with a fraction of 18.6% based on a long-term dataset) observed 392 under high *Ri* condition in the boreal forest. The decoupled layer could strongly 393 influence the ground O₃, MT, and CO₂ concentrations (Rannik et al., 2009, 2012; 394 Alekseychik et al., 2013; Chen et al., 2017), and could also explain the occurrence of 395 the strong temperature inversion during the inversion nights. To explore the possible 396 mechanism resulting in the significant different O3, MT and HOM concentrations in the 397 sub-canopy level, the mean continuity equation for high Reynolds number flows within 398 the canopy is formulated as (e.g. Katul et al. 2006):

$$399 \quad \frac{\partial \bar{c}}{\partial t} + \bar{U} \frac{\partial \bar{c}}{\partial x} + \bar{W} \frac{\partial \bar{c}}{\partial z} = -S - \frac{\partial \overline{w'c'}}{\partial z} - \frac{\partial \overline{u'c'}}{\partial x}$$
(3)

$$400 T_1 + T_2 + T_3 = T_4 + T_5 + T_6 (4)$$





401 where t is time, x and z are the longitudinal and vertical directions, respectively, C402 is the scalar concentration, U and W are the longitudinal and vertical velocity components, $\overline{w'c'}$ and $\overline{u'c'}$ are the turbulent scalar fluxes in the vertical and 403 404 horizontal, respectively, and S represents the net sources or sinks (physical, chemical, 405 and biological) of C, and overline represents time averaging over turbulent scales. The 406 6 terms in this equation represent the following (left to right): local rate of change (= T_1), 407 horizontal advection by the mean velocity $(= T_2)$, vertical advection by the mean 408 velocity $(=T_3)$, net sources or sinks $(=T_4)$, net vertical transport by the vertical 409 turbulent flux gradient $(=T_5)$, net horizontal transport by the horizontal turbulent flux 410 gradient (= T_6). Generally, $|T_6| \ll |T_5|$, and is hereafter ignored in the discussion.

411

412 During the non-inversion night, the ground O₃ could be replenished either by vertical turbulent transport (T_5) , mean vertical advection from upper boundary layer (T_3) , or 413 414 horizontal advection below canopy (T_2) (as shown in Figure 6). However, for highly 415 stratified flows, T_5 becomes small, as the efficiency of turbulence to transport O₃ to 416 layers near the ground becomes weak (Katul et al., 2014). Vertical and horizontal 417 advection were also small within such a stable layer, and the reduced mean velocity 418 would result in smaller contributions from T_2 and T_3 . Noted that these advective terms 419 tend to be opposite in sign by the virtue of the mean fluid continuity equation (Katul et 420 al., 2006). Instead, the sink of $O_3(T_4)$ was stronger because of the increasing loss due 421 to a higher surface area-to-volume density (S/V) in this shallow decoupled layer. Under 422 this circumstance, the ground O₃ concentration dramatically decreased when the air 423 layer was forming, and eventually reached a much lower concentration. The decoupled 424 layer also affected the sub-canopy level MT and CO₂ in the inversion night, but resulted 425 in concentration increases as opposed to O_3 . The weakened vertical turbulence (T_5) 426 tended to retain the emissions from ground and understory vegetation within the layer, 427 though T_4 also increased. In general, the increased CO₂ (primary source from the 428 ground) and MT (primary source from the canopy) at ground level are good indicators





- 429 for the extent of the mixing in the shallow decoupled layer. At the same time, the strong
- 430 decrease of O₃ shows how the sinks in this layer are no longer balanced by a large flux
- 431 of O₃ from upper layers. However, the stabilization of ground-level O₃ concentrations
- 432 at non-zero values after the initial fast decrease suggests that a small amount of inflow,
- 433 either via T_3 or T_5 , is still taking place.





435

Figure 6 Schematic figure showing how vertical mixing, vertical advection, and horizontal advection
influence ground O₃ concentrations differently in non-inversion night and inversion night in boreal forest.

Therefore, the differences between the ground and tower measurements were due to the joint effects of: (i) decoupling between the stably stratified near-ground layer and the canopy top, and the consequent formation of a shallow layer, (ii) weakening of advective and turbulent flux transport terms thereby inhibiting mass exchange between the ground decoupled layer and the remaining nocturnal boundary layer, and (iii) increased surface area to volume within the decoupled layer thereby enhancing T_4 .







446

Figure 7 Mass defect (MD) plots of the selected "non-inversion night" case (September 11), at (a) tower and (b) ground levels; and "inversion night" case (September 8), at (c) tower and (d) ground levels. The grey shade area denotes the dimer range (m/z 450-600).

450

451 Examination of the selected HOM molecules was useful and efficient to assess the 452 changes in HOMs, however, it might only stand for the major formation pathways. 453 Hence, it was also worthwhile to have a holistic view on the whole mass spectra and all 454 the detected HOMs. The mass defect (MD) plot, with the exact masses of the 455 compounds on the x-axis, the deviation from the integer mass on the y-axis, the compounds plotted in circles and the areas scaled by concentrations, shows the 456 457 abundance and chemical speciation of all the detected HOMs in the spectra. Figure 7a 458 and 7b are MD plots showing the mean spectra of the selected non-inversion night 459 (September 11) at tower and ground levels. Without the formation of a decoupled layer,





460 nearly identical composition distribution of HOMs were observed. In contrast, during
461 the inversion night (September 8, Figure 7c and 7d), large difference could be found
462 between the two measurement heights. Moreover, a significant fraction of the ground
463 HOMs disappeared in the inversion night, and the concentrations of the remaining
464 HOMs were also lower, confirming the aforementioned results obtained with the
465 selected HOM groups.

466

467 4.4 Study limitations

Several limitations still existed in this study. The contribution from the potential 468 micrometeorological processes in the layer between 1.5 m and 4.2 m (between the 469 470 sampling heights of the ground HOMs and other parameters) could not be estimated 471 with the current experiment design (i.e., only two measurement heights). Similarly, the influence from horizontal advection could not be entirely ruled out for the reduced 472 473 ground-level HOM concentration (and other largely changed species), because of the 474 horizontal inhomogeneity of HOM precursors and oxidants in the sub-canopy level. 475 However, our conclusion was confirmed by the incompatibility between the increasing 476 ground MT and CO₂ concentrations and the advection hypothesis (i.e., all species would 477 show similar tendencies if advection played a major role), indicating the influence of 478 horizontal and vertical advection might be minor when compared to the increasing sink. 479 However, more direct evidence was still needed for further validation, which also 480 highlighted the needs for joint vertical-planar HOM studies.

481

482 **5** Conclusion

Highly oxygenated molecules (HOMs) were measured above the canopy and at ground
level (below canopy) in a boreal forest environment during the IBAIRN campaign,
September 2016. Boundary layer dynamics and micrometeorology were found to be





486 important factors that influence the HOMs at ground level, by perturbing both their 487 sources and sinks. In the well-mixed boundary layer (e.g. during daytime or nights 488 without strong inversion), HOM concentrations and other measured species were 489 overall similar between the ground and tower measurements. In contrast, much lower 490 ground level HOM concentrations were observed when nighttime temperature 491 inversion and formation of a decoupled layer occurred in the sub-canopy level. On one 492 hand, the production of the ground-level HOMs could be affected by the decreasing O3 493 concentrations and the increasing MT concentration in the shallow layer. On the other 494 hand, the surface area to volume ratio dramatically increased in the shallow layer 495 compared to the nocturnal boundary layer. The possibility of losses on surfaces for 496 ground-level HOMs became much larger than usual during inversion nights. The 497 enhanced interaction of air in the decoupled layer with the forest floor was supported 498 by increased concentrations of CO₂, emitted mainly from the ground, in this layer.

499

500 We have presented the first detailed measurements of HOMs below and above the 501 canopy across a wide range of atmospheric stability conditions. The results highlight 502 the significance of near-ground boundary layer dynamics and micrometeorological 503 processes to the ambient HOMs, showing that ground-based HOM measurement might 504 not be representative for the entire nocturnal boundary layer. Conventionally, field 505 measurements of HOMs and other parameters are mostly performed close to ground, 506 and the effect from boundary layer dynamics and micrometeorological processes to the 507 HOM measurements have rarely been considered. Aerosol particle growth and SOA 508 formation rates at ground level are likely to be influenced by the reduced HOM 509 concentrations. Clearly, more vertical and planar measurements of HOMs are needed 510 to confirm the emerging picture presented here. Influence from boundary layer 511 dynamics should be better characterized and evaluated in future field campaigns.





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