

1 **Chemical characterization of long-range transport biomass burning** 2 **emissions to the Himalayas: insights from high-resolution aerosol** 3 **mass spectrometry**

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13 **Abstract**

14 An intensive field measurement was conducted at a remote, background, and high-altitude site
15 (Qomolangma station, QOMS, 4276 m a.s.l.) in the northern Himalayas, using an Aerodyne high-
16 resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS) along with other collocated
17 instruments. The field measurement was performed from April 12 to May 12, 2016 to chemically
18 characterize the high time-resolved submicron particulate matter (PM₁) and obtain the dynamic
19 processes (emissions, transport, and chemical evolution) of biomass burning (BB), frequently
20 transported from South Asia to the Himalayas during pre-monsoon season. Overall, the average (\pm
21 1σ) PM₁ mass concentration was 4.44 (\pm 4.54) $\mu\text{g m}^{-3}$ for the entire study, comparable with those
22 observed at other remote sites worldwide. Organic aerosol (OA) was the dominant PM₁ species
23 (accounting for 54.3% of total PM₁ on average) followed by black carbon (BC) (25.0%), sulfate
24 (9.3%), ammonium (5.8%), nitrate (5.1%), and chloride (0.4%). The average size distributions of
25 PM₁ species all peaked at an overlapping accumulation mode (\sim 500 nm), suggesting that aerosol
26 particles were internally well-mixed and aged during long-range transport. Positive matrix
27 factorization (PMF) analysis on the high-resolution organic mass spectra identified three distinct
28 OA factors, including a BB-related OA (BBOA, 43.7%), a nitrogen-containing OA (NOA, 13.9%)
29 and a more-oxidized oxygenated OA (MO-OOA, 42.4%). Two polluted episodes with enhanced
30 PM₁ mass loadings and elevated BBOA contributions from the west and southwest of QOMS
31 during the study were observed. A typical BB plume was investigated in detail to illustrate the
32 chemical evolution of aerosol characteristics under distinct air mass origins, meteorological
33 conditions and atmospheric oxidation processes.

34 **1 Introduction**

35 The Tibetan Plateau and Himalayas (TPH), generally called the “third pole”, is the highest
36 (average altitude of more than 4000 m a.s.l.) and largest (\sim 2 500 000 km²) plateau in the world.
37 This region has been recognized as one of the most pristine region in the world due to its high
38 altitude, sparse population and minor influence of anthropogenic activities (Yao et al., 2012a).

39 Consideration on the intense dynamical and thermal forcing effects, the TPH not only plays a key
40 role in the formation of Asian monsoon systems, but also impacts the large-scale atmospheric
41 circulation, hydrological cycle, as well as global climate (Duan and Wu, 2005; Wu et al., 2007).
42 Over the past decades, more attentions have been paid to the environment and climate change in
43 the TPH since this region is very susceptibility to the global climate change such as fast air
44 temperature rise and dramatic glacier shrinkage (Xu et al., 2009; Kang et al., 2010; Yao et al.,
45 2012b; Yang et al., 2014). Atmospheric environment in the TPH, albeit which is one of the most
46 pristine region in the world, has been thought to be influenced variably due to the worse air
47 pollution in its surrounding countries (Hou et al., 2003; Lau et al., 2008). For example, polluted air
48 mass, particularly from South and Southeast Asia regions, had been observed frequently to
49 transport to the Himalayas (Bonasoni et al., 2010; Cong et al., 2015), heat the aloft air masses over
50 the TPH (Lau et al., 2006; Ramanathan and Carmichael, 2008) and decline the surface albedo after
51 its deposition onto snow and glacier (Xu et al., 2009). As a consequence, characterizing the
52 aerosol physicochemical properties in the TPH, including mass loading, chemical composition,
53 size distribution and source, are of great importance to better understand the aerosol chemistry,
54 estimate the aerosol radiative forcing, and finally evaluate the effect of polluted air mass on the
55 ecology and environment in the TPH region.

56 Numerous aerosol measurements have been conducted in the TPH region in past decades to
57 characterize the physicochemical properties, sources and transport pathways of ambient aerosol
58 (Liu et al., 2008; Decesari et al., 2010; Marcq et al., 2010; Marinoni et al., 2013; Putero et al.,
59 2014; Xu et al., 2017; Zhang et al., 2017a). South and Southeast Asia are two major polluted
60 regions due to their intense biomass burning (BB) activities from natural forest fires and
61 traditional human burning activities for residential heating and cooking (Engling et al., 2011;
62 Yadav et al., 2017). The polluted feature of South and Southeast Asia during April 12 to May 12,
63 2016 can be further revealed by the distribution of average aerosol optical depth in Fig. 1. During
64 the pre-monsoon period, atmospheric pollutants associated with BB emissions in South Asia are
65 generally advected by regional and long-range transport (e.g., westerlies and South Asian
66 monsoon system) to Himalayas and built up in the southern foothills, then pollutants are lifted up
67 to high altitude by the Himalayan topography and the typical valley wind circulation (Zhao et al.,
68 2013; Cong et al., 2015; Liu et al., 2017). However, the chemical properties of aerosol particles
69 are still not well understood and limited in the Himalayas region due to its remote and harsh
70 environments, challenging weather conditions and logistic difficulties. In addition, most of the
71 available studies are mainly based on the off-line filter sampling of ambient aerosol or snow/ice
72 samples following by laboratory analyses (Decesari et al., 2010; Ram et al., 2010; Li et al., 2016;
73 Wan et al., 2017). These studies usually had a relatively low-time resolution (days to weeks).
74 Therefore, real-time consecutive field measurement, especially focusing on the high-resolution
75 size-resolved chemical characteristics of aerosol particles, is of great importance and necessary to
76 give insight into the sources and the dynamic chemical evolution of ambient aerosol.

77 Online real-time instrument such as Aerodyne aerosol mass spectrometer (AMS), which can
78 be used to characterize the chemical properties and sources of submicron aerosol particles with
79 high time resolution and sensitivity, has been greatly developed and widely implemented
80 worldwide (Canagaratna et al., 2007; Zhang et al., 2007a; Jimenez et al., 2009; Li et al., 2017).
81 Although the deployments of the AMS in China have started since 2006, most of these studies in

82 China are conducted in urban areas, including Beijing–Tianjin–Hebei (Sun et al., 2013; Sun et al.,
83 2016), Yangtze River Delta (Wang et al., 2016a; Wang et al., 2016b), Pearl River Delta regions
84 (Huang et al., 2011), and Lanzhou (Xu et al., 2014; Xu et al., 2016; Zhang et al., 2017b) as shown
85 in Fig. S1, whereas just few studies deployed in remote sites so far, such as Menyuan (Du et al.,
86 2015), Mt. Yulong (Zheng et al., 2017), and Nam Co (Xu et al., 2017; Wang et al., 2017). In this
87 paper, an Aerodyne high-resolution time-of-flight mass spectrometer (HR-ToF-AMS) was
88 deployed at the Qomolangma Station for Atmospheric and Environmental Observation and
89 Research (QOMS) in the north slope of the Himalayas to fill the vacancy of real-time mass
90 spectrometer measurement at high elevation site and evaluate the significant impacts of BBs from
91 polluted areas in the South Asia on the TPH aerosol properties during the pre-monsoon season.
92 Here, we report an overview of the 5-min real-time chemical and physical characteristics of
93 submicron aerosols (PM_1), including mass loading, composition, size distribution, acidity as well
94 as temporal and diurnal variations. The sources of organic aerosols (OA) are also investigated
95 using positive matrix factorization analysis on the high-resolution OA mass spectrum. BB
96 influence and chemical evolution of aerosols in polluted plume are examined via combining back
97 trajectory analysis of air masses and fire hotspots information, respectively.

98 **2 Experimental methods**

99 **2.1 Sampling site**

100 The QOMS (28.36° N, 86.95° E, 4276 m a.s.l.; Fig. 1), which is located in the northern slope of
101 Mt. Everest (~ 30 km away), was established for atmospheric and environmental observation since
102 2005 (Ma et al., 2008). The geomorphic and climate features around the QOMS are typical alpine
103 cold and arid areas covered by sandy soil with sparse vegetation. The QOMS is located in a long
104 river valley and isolated from residential areas due to its harsh environment with a small village
105 (with a population of ~ 300) to the south (~ 10 km). The closest town, Dingri County, is ~ 100 km
106 south from the QOMS. A freeway is located at the front of the QOMS for tourism with increased
107 tourist during summer. The measurements were conducted from April 12 to May 12, 2016. Since
108 this period was within the typical pre-monsoon season of the TPH, the large-scale atmospheric
109 circulation pattern was dominated by westerly or southwesterly winds with limited precipitation.
110 Owing to a distinct thermal forcing from the southern mountains and glaciers, the QOMS was
111 locally dominated by strongly mountain-valley circulation with down-slope wind prevailing
112 during the daytime, especially in the afternoon (Fig. 1c and S2) (Zou et al., 2008), which would
113 make the valley as an efficient channel for the down transport of air mass from high-altitude
114 troposphere.

115 **2.2 Instrumentation**

116 A suite of real-time instruments were co-located to measure the physiochemical properties of fine
117 particles at the QOMS, including an Aerodyne HR-ToF-AMS (Aerodyne Research Inc., Billerica,
118 MA, USA) for 5-min size-resolved chemical compositions (organics, sulfate, nitrate, ammonium,
119 and chloride) of non-refractory submicron particulate matter (NR- PM_1), a scanning mobility
120 particle sizer (SMPS, model 3936, TSI Inc., Shoreview, MN, USA) for 5-min particle number
121 concentration and size distribution between 14.6 and 661.2 nm in mobility diameter (D_m), and a
122 photoacoustic extinctionsimeter (PAX, DMT Inc., Boulder, CO, USA) for particle light absorption

123 and scattering coefficient (b_{abs} and b_{scat}) at 405 nm and further deriving black carbon (BC) mass
124 concentration. All instruments were placed in an air-conditioned room with temperature
125 maintaining at ~ 20 °C. Ambient aerosol particles were introduced through a 0.5 inch copper tube
126 which stemmed out of the rooftop by about 1.5 m. A $\text{PM}_{2.5}$ cyclone (model URG-2000-30EH,
127 URG Corp., Chapel Hill, NC, USA) was used in front of the sampling inlet for removing coarse
128 particles with size cutoffs of 2.5 μm in aerodynamic diameter (D_{va}). A diffusion dryer was placed
129 following the cyclone to dry the ambient air and eliminate potential humidity effect on particles.
130 The total length of the sampling line was about 5 m and the retention time of particles was less
131 than 2.5 s in the whole inlet. The total air flow rate from the sampling inlet was about 10 L min^{-1} ,
132 with part of flow shared by the HR-ToF-AMS and the SMPS while the remaining flow exhausted
133 by an external pump. The meteorology data including wind speed (WS), wind direction (WD),
134 relative humidity (RH), temperature (T), and solar radiation (SR) during this study were obtained
135 from a Vantage Pro2 weather station (Davis Instruments Corp., Hayward, CA, USA). Note that all
136 the date and time used in this study are reported in Beijing Time (BJT: UTC + 8 h).

137 **2.3 HR-ToF-AMS operation and data analysis**

138 **2.3.1 HR-ToF-AMS operation**

139 A detailed instrumental description of the Aerodyne HR-ToF-AMS can be found elsewhere
140 (DeCarlo et al., 2006) and only a brief summary is provided here. Briefly, the HR-ToF-AMS
141 consists of three main parts: an aerosol sampling inlet, a particle sizing vacuum chamber, and a
142 particle composition detection section (Jimenez et al., 2003). Ambient particles are sampled into
143 the instrument through a critical orifice (130 μm in this study for enhancing the transmission
144 efficiency at the high-altitude area) and focus into a concentrated and narrow beam through an
145 aerodynamic lens. Then particles are accelerated into the sizing vacuum chamber and obtain
146 different velocities for particles with different sizes due to the supersonic expansion induced by
147 different pressure between the two chambers. Meanwhile, a mechanical chopper with two radial
148 slits located 180° apart is used to intercept the focused particle, and then the time of flight (P-ToF)
149 from the chopper to the vaporizer is measured to obtain the aerodynamic size of particles. After
150 passing through the sizing chamber, particles are directed onto a resistively heated surface (\sim
151 600 °C) under a high vacuum and ionized by a 70 eV electron impact, and finally detect by the
152 high-resolution time-of-flight mass spectrometer. In this study, the HR-ToF-AMS was only
153 toggled under the high sensitive V-mode (detection limits $\sim 10 \text{ ng m}^{-3}$). Under the V-mode
154 operation, the instrument also switched between the mass spectrum (MS) mode and the particle P-
155 ToF mode every 15 s, spending 6 and 9 s on each, to obtain the mass concentrations and size
156 distributions of the non-refractory species, respectively.

157 The HR-ToF-AMS was calibrated for ionization efficiency (IE) and particle sizing at the
158 beginning, in the middle, and at the end of this study according to the standard protocols (Jayne et
159 al., 2000). Both the calibrations of IE and particle sizing were performed using mono-dispersed
160 ammonium nitrate particles with nominal diameters of 70–300 nm. Default relative ionization
161 efficiency (RIE) values were assumed in this study as 1.1 for nitrate, 1.3 for chloride, and 1.4 for
162 organics. The RIE values of 3.9 and 4.2 were used for ammonium based on the results of two IE
163 calibrations at the beginning and in the middle of this study, while RIE values of 1.6 and 1.4 were
164 determined similarly for sulfate by using mono-dispersed ammonium sulfate particles,

165 respectively.

166 **2.3.2 HR-ToF-AMS data analysis**

167 The mass concentrations and size distributions of NR-PM₁ species and the ion-speciated mass
168 spectra, composition and elemental composition of organics were determined from the HR-ToF-
169 AMS data by using the standard ToF-AMS analysis toolkit SQUIRREL (v1.56) and PIKA (v1.15c)
170 modules written in Igor Pro (Wavemetrics Inc., Lake Oswego, OR, USA). An empirical particle
171 collection efficiency (CE) of 0.5 was used to compensate for the incomplete transmission and
172 detection of particles due to particle bouncing at the vaporizer and partial transmission through the
173 aerodynamic lens, which has been widely used in field studies employing AMS with a dryer
174 installed in front of the inlet (Xu et al., 2014; Xu et al., 2016). The elemental ratios of oxygen-to-
175 carbon (O/C), hydrogen-to-carbon (H/C), nitrogen-to-carbon (N/C), and organic mass-to-organic
176 carbon (OM/OC) for this study were determined using the “improved-ambient” method (referred
177 as I-A method) (Canagaratna et al., 2015), which increased O/C on average by 34%, H/C on
178 average by 15%, and OM/OC on average by 17% (Fig. S3) compared with those determined from
179 the “Aiken ambient” method (referred as A-A method) (Aiken et al., 2008).

180 Positive matrix factorization (PMF) analysis using the PMF2.exe algorithm (v4.2) (Paatero
181 and Tapper, 1994) in robust mode was conducted on the high resolution mass spectra (HRMS) to
182 determine distinct OA components in this study. The analysis was performed using an Igor Pro-
183 based PMF Evaluation Tool (PET, v2.03) (Ulbrich et al., 2009), downloaded from the webpage
184 (http://cires.colorado.edu/jimenez-group/wiki/index.php/PMF-AMS_Analysis_Guide). The data
185 and error matrices input into the PMF analysis were generated from analyzing the V-mode data via
186 PIKA fitting. Detailed PMF analysis was thoroughly evaluated following the procedures
187 summarized in Table 1 of Zhang et al. (2011). Isotopic ions were generally excluded and the four
188 ions of O⁺, HO⁺, H₂O⁺, and CO⁺ were downweighted in PMF analysis, because they were
189 determined according to the relationship with CO₂⁺ signal (Ulbrich et al., 2009). The “bad” ions
190 with *S/N* less than 0.2 were removed from the HRMS data and error matrices before PMF analysis,
191 and “weak” ions with *S/N* between 0.2 and 2 were downweighted by increasing their errors. In
192 addition, some runs with huge mass loading spikes were also removed from the data and error
193 matrices. The detailed matrix preparation and data pretreatment can also refer to Xu et al. (2014).

194 A summary of key diagnostic plots of the PMF results for this study is presented in Fig. S4.
195 Overall, the PMF solutions were investigated for 1 to 8 factors and for the rotational parameter
196 (fPeak) varying from -1 to 1 with a step of 0.1. Besides examining the model residuals, scaled
197 residuals, and the Q/Q_{exp} contributions for each *m/z* and time following procedures detailed in
198 Table 1 of Zhang et al. (2011), the optimum solution can also be evaluated via comparing the mass
199 spectra of individual factors with reference spectra from specific sources or other ambient AMS
200 measurements, comparing the time series of individual factors with the known external tracers,
201 and analyzing the diurnal variations of individual factors. Finally, the 3-factor solution with fPeak
202 = 0 was chosen in this work. The direct comparisons of the mass spectra, time series, and diurnal
203 variations for 2-factor and 4-factor solution were also shown in Fig. S5 and S6, respectively. The
204 2-factor solution does not resolve the small, yet distinct nitrogen-containing OA, while the 4-
205 factor solution shows a splitting factor from the BB OA resolved in the 3-factor solution and
206 seems just like a simple separation of the two BB polluted episodes.

207 **2.4 Other relevant data**

208 The Hybrid Single Particle Lagrangian Integrated Trajectory (HYSPLIT4) model developed by
209 the National Oceanic and Atmospheric Administration (NOAA) (Draxler and Rolph, 2003) was
210 used to investigate the origins of air masses in this study, using the meteorological data from the
211 NOAA Global Data Assimilation System (GDAS). The back trajectories were calculated every 6 h
212 at an ending height of 500 m above ground level at the QOMS during the entire campaign, and
213 then clustered them according to their similarity in spatial distribution. Finally, a four-cluster
214 solution was adopted according to its small total spatial variance.

215 Aerosol optical depth (AOD) at 550 nm was derived from the observations made by National
216 Aeronautics and Space Administration (NASA) Moderate Resolution Imaging Spectroradiometer
217 (MODIS) onboard the Terra satellite. The distribution of average aerosol optical depth (AOD) in a
218 large range areas (20°–45° N, 60°–110° E) around the TPH during the entire period of this study is
219 given in Fig. 1d.

220 Various active fire hotspots were detected over South and Southeast Asia by the Fire
221 Information for Resource Management System (FIRMS) provided by MODIS satellite
222 (<https://firms.modaps.eosdis.nasa.gov>), demonstrating the possibility that active wildfires or BBs
223 from South and Southeast Asia may have significant impacts on the air conditions in the TPH
224 region.

225 The aerosol liquid water content (ALWC) was estimated with the Extended AIM (E-AIM)
226 Aerosol Thermodynamics Model (<http://www.aim.env.uea.ac.uk/aim/aim.php>). The input data
227 included the concentrations of sulfate, nitrate, ammonium, and chloride measured by the HR-ToF-
228 AMS as well as the relative humidity (RH) and temperature of ambient air.

229 **3 Results and discussion**

230 **3.1 Overview of the study**

231 **3.1.1 Meteorological conditions**

232 The measurement period in our study was within the typical pre-monsoon season of the TPH. The
233 meteorological conditions were therefore characterized by a relatively cold, dry and windy
234 weather, and the westerlies dominated the large-scale atmospheric circulation patterns with little
235 precipitation, as displayed in Fig. 2. During the study, the averaged diurnal air temperature ranged
236 from –2.0 to 12.5 °C with an average ($\pm 1\sigma$) of 5.7 (± 5.0) °C, and the RH ranged from 15.3 to
237 67.5% with an average of $39.8 \pm 18.8\%$. Only two light precipitation events (1 and 0.5 mm d⁻¹)
238 occurred on 1 and 8 May, respectively. The WDs at QOMS were predominantly by southwesterly,
239 which were mainly associated with the thermally driven mountain-valley winds and glacier winds
240 (Zou et al., 2008). For the diurnal variation of wind conditions, a nearly calm wind period (hourly
241 average WS less than 2 m s⁻¹) was observed in the early morning time; after sunrise to noon time,
242 there was a weak up-slope wind period (from the north); the diurnal wind cycles in the rest time
243 were dominated by the down-slope wind (from the southwest) with the maximum value of hourly
244 average WS up to 7 m s⁻¹ (Fig. 2b and S2).

245 **3.1.2 Inter-comparisons between different instruments**

246 An inter-comparison of the total PM₁ (NR-PM₁ + BC) mass concentrations measured by the HR-
247 ToF-AMS (CE = 0.5) and the PAX with particle volumes (assuming spherical particles)
248 determined from the SMPS is shown in Fig. S7. Overall, the PM₁ mass is closely correlated ($R^2 =$
249 0.97) with that of SMPS particle volume during the entire campaign, with a linear regression slope
250 of 2.86. This slope is significantly higher than the estimated average PM₁ density of 1.44 g cm⁻³,
251 which is calculated based on the measured particle compositions in this study and the assumed
252 particle densities of 1.2 for organics, 1.78 for (NH₄)₂SO₄, 1.72 for NH₄NO₃, 1.52 for NH₄Cl and
253 1.8 g cm⁻³ for BC (Zhang et al., 2005b; Xu et al., 2016). This discrepancy is likely introduced by
254 various factors, including different transmission sizes between HR-ToF-AMS and SMPS (up to ~
255 1.0 μm in D_{va} for AMS vs. limited size range of 14.6–661.2 nm in D_m for SMPS), rough
256 calculation of PM₁ density using assumed composition densities and spherical shape without
257 consideration the particle porosity, as well as the using of empirical and constant CE value of 0.5
258 in this study. This phenomenon was also observed at other sites in previous studies (Ge et al., 2012;
259 Huang et al., 2012; Xu et al., 2014; Du et al., 2015).

260 3.1.3 Mass concentration and chemical composition of PM₁

261 As shown in Fig. 2, the mass concentrations of PM₁ and all PM₁ species, as well as their mass
262 fractions in PM₁ varied dynamically throughout this study. Two polluted periods (PP1 and PP2)
263 were identified according to their high PM₁ mass concentrations (daily average PM₁ mass is larger
264 than 5 μg m⁻³), high contributions from BBOA and unique back trajectories. The rest periods
265 characterized by low PM₁ mass concentrations were considered as clear periods (CP1 and CP2).
266 The 5-min total PM₁ mass concentration ranged from 0.18 to 27.97 μg m⁻³ for the study, with an
267 average ($\pm 1\sigma$) value of 4.44 \pm 4.54 μg m⁻³. This average value was more than two times lower
268 than most of the PM₁ mass concentrations measured with Aerodyne AMS or aerosol chemical
269 speciation monitor (ACSM) instruments at various urban, suburban, rural or background sites in
270 China (10.9–138.8 μg m⁻³) (Fig. S1), except slightly lower than that at Mt. Yulong (5.7 μg m⁻³)
271 located at the southeastern edge of the TPH, whereas higher than that at Nam Co Station (2.0 μg
272 m⁻³) located in the central of the TPH. Moreover, as shown in table S1, the PM₁ mass
273 concentration in this study was also lower than those measured at the three remote island sites in
274 Asia which were frequently influenced by outflow from China, Korea and Japan (i.e., 7.9 μg m⁻³
275 for Okinawa island, 12.0 μg m⁻³ for Fukue island in Japan, and 10.7 μg m⁻³ for Jeju island in
276 Korea) (Takami et al., 2005; Jimenez et al., 2009), as well as the PM₁ mass concentration (15.1 μg
277 m⁻³) obtained at the Bachelor mountain in United States which was heavily impacted by wildfire
278 smoke plumes (Zhou et al., 2017). However, it was higher than those reported at other coastal,
279 high elevation, forest or remote background sites in North America and Europe (0.55–2.91 μg m⁻³)
280 (Zhang et al., 2007a; Sun et al., 2009; Fröhlich et al., 2015). Although these measurements
281 mentioned above were conducted at various sites worldwide during different seasons, these
282 comparisons further demonstrate that QOMS is a typical high elevation and remote background
283 site in Asia.

284 Overall, organics and BC were the two dominant PM₁ species (averagely contributed 54.3%
285 and 25.0% to the total PM₁ mass, respectively) followed by sulfate (9.3%), ammonium (5.8%),
286 nitrate (5.1%), and chloride (0.4%) (Fig. 3a). The high contributions of organics and BC at QOMS
287 were significantly associated with the active BB emissions by long-range transport from polluted

288 areas in South Asia. Organic compounds and BC have been revealed as two dominant components
289 of BB aerosols and generally used to identify BB events in previous studies (Bond et al., 2004;
290 Bougiatioti et al., 2014). In addition, biomass burning at high elevation regions of Himalayas and
291 south Asia was more incomplete burning and could emit amount of BC. This conclusion can be
292 further revealed by their enhanced mass concentrations and contributions, especially for organics,
293 during the two distinct polluted episodes influenced by active BB plumes. Figure 3b showed the
294 mass contributions of PM₁ species as a function of total PM₁ mass concentrations. The PM₁ mass
295 loadings in this study were mostly below 6 μg m⁻³ (accounted for ~ 77%); The mass contribution
296 of organics increased significantly with the increase of total PM₁ mass loading whereas the rest
297 species showed relatively stable or decrease trends, suggesting the dominant contributions of
298 organics in the polluted episodes at QOMS.

299 **3.1.4 Acidity and size distributions of submicron aerosols**

300 To evaluate the bulk acidity of NR-PM₁ in this study, we calculated the NH₄⁺ concentration
301 (NH₄⁺_{calc}) based on the mass concentrations of sulfate, nitrate and chloride measured by the HR-
302 ToF-AMS and assumed full neutralization of these anions by ammonium (Zhang et al., 2007b).
303 The scatter plot of the measured NH₄⁺ (NH₄⁺_{meas}) concentration versus the NH₄⁺_{calc} concentration
304 for the entire campaign was shown in Fig. S7. A tight correlation ($R^2 = 0.97$) existed between
305 NH₄⁺_{meas} and NH₄⁺_{calc} with a linear regression slope of 1.2, indicating that there were excess of
306 ammonium in the submicron particle. This slightly high NH₄⁺_{meas}/NH₄⁺_{calc} ratio was quite different
307 with those results from various urban and rural sites in China, where bulk aerosols were overall
308 neutralized or acidic due to the enrich gaseous precursors of SO₂ and NO_x that could be further
309 oxidized to sulfate and nitrate (Sun et al., 2013; Xu et al., 2014; Du et al., 2015; Zhang et al.,
310 2017b). The excess ammonium at QOMS might relate to the important contributions of organic
311 acids in this area (Cong et al., 2015), which could underestimate the NH₄⁺_{calc} due to the neglect of
312 organic acids in the ion-balance calculation, and the non-negligible contributions of nitrogen-
313 containing organic compounds to NH_x⁺ which finally overestimated the NH₄⁺_{meas} (Sun et al., 2009;
314 Ge et al., 2012). As mentioned above, atmospheric aerosols in the TPH region were significantly
315 influenced by BB emissions from South Asia during the sampling periods. BBs would emit large
316 amounts of nitrogen-containing organic compounds (Fleming et al., 2017; Zhou et al., 2017) and
317 as discussed in section 3.2.

318 Figure 4 shows the average size distributions of NR-PM₁ species and their mass contributions
319 as the function of size distribution. Overall, all chemical species showed a nearly consistent but
320 narrow accumulation mode peaking at ~ 500 nm in D_{va} , indicating the well internal-mixed and
321 aged aerosol particles at QOMS. Ultrafine particles (particles with diameter less than 100 nm)
322 were dominated by organics (more than 70%), while the mass contributions of chemical species at
323 the major peak (~ 500 nm) were organics (~ 65%), sulfate (~ 13%), nitrate (~ 11%), ammonium (~
324 10%), and chloride (~ 1%). The contribution of organics decreased with the increase of size mode,
325 while the contributions of three major inorganic species (sulfate, nitrate and ammonium) slightly
326 increased with the increasing sizes (Fig. 4b).

327 **3.1.5 Diurnal variations of chemical species**

328 The average diurnal cycles of meteorological parameters as well as the PM₁ species and their mass
329 fractions for the entire campaign were shown in Fig. 5. All PM₁ species presented a similar diurnal

330 pattern with lower concentrations in the daytime whereas higher concentrations in the nighttime.
331 The mass concentrations reached the minimum values at around 15:00. This pattern was
332 accompanied with the enhanced wind speed and the increased air temperature in the afternoon
333 which could related with the dynamics of planetary boundary layer (PBL). After that, the mass
334 concentrations began to build up and reached to high levels in the nighttime. Note that the mass
335 concentrations of chloride and BC also existed a slight peak during the early morning, which
336 corresponded with the calm wind conditions and the lowest air temperature of the day and could
337 associated with the enhanced local emissions at QOMS in the morning. The diurnal cycles of mass
338 contributions of each PM₁ species were relatively stable for the entire campaign, besides the slight
339 increase of BC from 24% at ~ 08:00 to 30% at ~ 10:00. Overall, organics dominated PM₁
340 throughout the day (49–57%), followed by BC (23–30%), sulfate (9–10%), ammonium (5–6%),
341 nitrate (4–6%), and chloride (0.3–0.8%).

342 3.2 Bulk characteristics of OA

343 Figure 6a and b showed the average mass contributions of the four elements and the six ion
344 categories to total organics, respectively. The organic mass was on average composed of 36.8%
345 carbon, 57.9% oxygen, 4.0% hydrogen, and 1.3% nitrogen. For ionic categories, C_xH_yO₁⁺ ions
346 dominated the total OA accounting for 41.3%, followed by C_xH_yO₂⁺ (24.9%), C_xH_y⁺ (23.9%),
347 H_yO₁⁺ (6.1%), C_xH_yN_p⁺ (2.9%) and C_xH_yO_zN_p⁺ (0.9%). The contributions of oxygen and the two
348 major oxygenated ion fragments (C_xH_yO_z⁺ = C_xH_yO₁⁺+C_xH_yO₂⁺) at QOMS were quite higher than
349 those obtained at other urban or rural sites in China, whereas carbon and C_xH_y⁺ ions had relative
350 lower contributions, e.g., 38% of C_xH_yO_z⁺ and 21% of oxygen versus 56% of C_xH_y⁺ and 70% of
351 carbon in urban Lanzhou (Xu et al., 2014), and 37.4% of C_xH_yO_z⁺ versus 51.2% of C_xH_y⁺ in urban
352 Nanjing (Wang et al., 2016a), suggesting that OA at QOMS were highly aged. Correspondingly,
353 the average high-resolution OA mass spectrum (Fig. 6c) also showed significantly high
354 contribution (~ 25%) at *m/z* 44 signal (one of the most reliable marker of oxygenated OA)
355 compared with other ion fragments, e.g., 5% at *m/z* 43 (indicator for less oxidized compounds),
356 1.7% at *m/z* 55 (important COA fragment), and 0.4% at *m/z* 57 (tracer for traffic-related emission)
357 (Alfarra et al., 2004; Zhang et al., 2005a). The average O/C ratio was 1.07 during this study, which
358 was much higher than those observed at various urban and rural sites in China using the I-A
359 method, e.g., 0.37 in Beijing (Sun et al., 2016), 0.36 in Lanzhou (Xu et al., 2016), 0.35 in Nanjing
360 (Wang et al., 2016a), and 0.65 in Ziyang (Hu et al., 2016)). Moreover, the average O/C ratio was
361 even higher than that of 0.98 at the background site of Mt. Wuzhi in southern China (Zhu et al.,
362 2016), indicating that OA at QOMS was more oxidized and aged during long-range transport. The
363 average H/C, N/C and OM/OC ratios were on average 1.29, 0.026 and 2.55 in this study,
364 determined a nominal chemical formula of OA as C₁H_{1.29}O_{1.07}N_{0.026}.

365 For the diurnal cycles, O/C ratio had two peaks in the early morning and late afternoon, likely
366 related to the production of secondary organic aerosol (SOA) via aqueous-phase reactions or
367 photochemical oxidation processes during these two periods. H/C and N/C ratios yet showed
368 inverse diurnal cycles with that of O/C, namely peaked at around 08:00–10:00 in the morning.
369 The Van Krevelen diagram (H/C versus O/C), which had been used widely to probe the oxidation
370 reaction mechanisms for bulk OA (Heald et al., 2010), showed an apparent anticorrelation ($R^2 =$
371 0.57) with a slope of –0.48 at QOMS (Fig. S8). Ng et al. (2011b) have suggested that a slope of –

372 0.5 indicate a net change in chemical composition from the addition of both acid and
373 alcohol/peroxide functional groups without fragmentation, and/or carboxylic acid groups with
374 fragmentation.

375 **3.3 Organic aerosol source apportionment**

376 Source apportionment via PMF analysis on the high-resolution OA mass spectrum identified three
377 distinct factors in this campaign according to their unique temporary variations, mass spectrum
378 (MS) profiles, element ratios, correlations with tracers, and diurnal patterns, i.e., a BB-related OA
379 (BBOA), a nitrogen-containing OA (NOA) and a more-oxidized oxygenated OA (MO-OOA).
380 Detailed discussion on each factor is given in the following subsections.

381 **3.3.1 BBOA**

382 Although significant high contribution at m/z 44 (mostly CO_2^+) was found in all of the three OA
383 components, the BBOA MS was also characterized by contributions at m/z 60 (mainly $\text{C}_2\text{H}_4\text{O}_2^+$)
384 and tiny m/z 73 (mainly $\text{C}_3\text{H}_5\text{O}_2^+$) (Fig. 7g), which were generally regarded as well-known tracers
385 for BB emissions (Alfarra et al., 2007). The average fraction of the signal at m/z 60 (referred as f_{60})
386 in the BBOA mass spectrum was 0.61%, which was higher than the typical value of $\sim 0.3\%$ in the
387 absence of BB impacts (Cubison et al., 2011). The time series of BBOA correlated tightly with
388 those of $\text{C}_2\text{H}_4\text{O}_2^+$ ($R^2 = 0.91$) and $\text{C}_3\text{H}_5\text{O}_2^+$ ($R^2 = 0.87$) as well as BC ($R^2 = 0.72$) and nitrate ($R^2 =$
389 0.75) (Fig. 7a and Table S2). If ignoring the influence of high contribution at m/z 44, the BBOA
390 mass spectrum in this study correlated well ($R^2 = 0.5\text{--}0.9$) with those BBOA mass spectrum
391 identified at other sites worldwide (Ng et al., 2011a; Mohr et al., 2012; Saarikoski et al., 2012;
392 Crippa et al., 2013; Crippa et al., 2014; Xu et al., 2016), as shown in Fig. S9. The average mass
393 concentration of BBOA was $1.05 \mu\text{g m}^{-3}$ for the entire study and contributed a large fraction
394 (43.7%) of the total OA mass on average (Fig. 8a), indicating that BBOA was an important
395 components of OA during the pre-monsoon season at the QOMS. The diurnal cycle of BBOA
396 showed high concentrations during nighttime whereas relatively low concentrations during
397 daytime (Fig. 7d). Correspondingly, the mass contributions of BBOA to total OA mass decreased
398 distinctly from $\sim 55\%$ at 00:00 to 28% at 15:00 (Fig. 8b). In addition, higher mass concentrations
399 and contributions of BBOA were found during the two polluted episodes (PP1 and PP2) than those
400 during the clear periods, further indicating the important contribution of BBOA to OA in this
401 region. Figure 8c showed the mass fractions of the three OA components as a function of total OA
402 mass during the entire campaign. A continuously increased trend was found for the BBOA
403 contributions with the increasing OA mass, which contributed $\sim 15\%$ when the total OA mass was
404 less than $0.3 \mu\text{g m}^{-3}$, whereas it reached up to more than 75% with the OA mass increased to $9 \mu\text{g}$
405 m^{-3} . This dominant contribution of BBOA during the polluted periods was consistent with those
406 results in previous studies that BB emission were an important source of aerosol to the southern
407 TPH (Engling et al., 2011; Xia et al., 2011; Putero et al., 2014; Cong et al., 2015). The O/C ratio
408 (0.85) of BBOA in this study was quite higher than those BBOA factors identified at other
409 urban/rural sites in previous studies (Aiken et al., 2009; Huang et al., 2011; Mohr et al., 2012; Sun
410 et al., 2016; Xu et al., 2016), suggesting its long-range transport feature. This aged BBOA feature
411 was similar with those obtained at other remote sites worldwide, such as a remote forest site in
412 Finland (Raatikainen et al., 2010), a remote background site in Greece (Bougiatioti et al., 2014),

413 and a national air quality background sites in southern China (Zhu et al., 2016), where OA were
414 generally highly oxidized.

415 **3.3.2 NOA**

416 Besides the two highest signals at m/z 43.99 (CO_2^+) and 27.995 (CO^+) which together contributed
417 half of the total NOA signal due to the highly aged OA nature at QOMS, the NOA MS was also
418 characterized by some nitrogen-containing fragments, such as m/z 27.011 (CHN^+), 41.027
419 ($\text{C}_2\text{H}_3\text{N}^+$), and 43.006 (CHON^+). In total, these three fragments could comprise nearly half of the
420 nitrogen-containing signals in the NOA factor and finally contributed 5% of the total NOA signal.
421 The average O/C ratio of NOA for the entire campaign was 1.10 with the highest N/C ratio (0.068)
422 among the three OA components. This high N/C ratio at QOMS was comparable with those
423 nitrogen-containing OA factor identified in previous studies, such as 0.06 in Mexico City (Aiken
424 et al., 2009), 0.078 in Po Valley, Italy (Saarikoski et al., 2012), and 0.053 in New York (Sun et al.,
425 2012). The time series of NOA showed tightly correlation ($R^2 = 0.62$) with that of estimated
426 organic nitrates, whereas relatively weak correlations with PM_{10} species and OA ions (Table S2). In
427 addition, the f_{60} value ($\sim 0.37\%$) was also slightly higher than the background f_{60} (0.3%) of BB
428 aerosols (Fig. 7h). These results together suggested that this oxygenated OA factor was likely a
429 nitrogen-containing OA and might be related to the aged BB emissions, consistent with the results
430 in previous studies that large amounts of nitrogen-containing organic compounds were found from
431 BB aerosols (Laskin et al., 2009; Gautam et al., 2016; Wang et al., 2017). In recently, Fleming et
432 al. (2017) found dung burning, a very popular activities in Himalayas and India for residential
433 cooking and heating, could emit much more nitrogen-containing OA than wood burning. Our filter
434 samples during high BBOA period analyzed by Fourier Transform Ion Cyclotron Resonance Mass
435 Spectrometry (FTICR-MS) also found amount of ON molecular (in preparation). As shown in Fig.
436 7e and 8b, both the diurnal cycles of mass concentrations and fractions of NOA had distinct
437 increase in the morning, similar with the diurnal patterns of chloride, element ratios of H/C and
438 N/C, and the estimated organic nitrates. This diurnal feature of NOA at QOMS was quite
439 consistent with those NOA factors identified in Po Valley, Italy (Saarikoski et al., 2012) and in
440 Mexico City (Aiken et al., 2009), or less-oxidized oxygenated OA (LO-OOA) in southeastern
441 USA (Xu et al., 2015) where have active BB emissions. NOA contributed $\sim 14\%$ of the total OA
442 mass on average, with an average mass concentration of $0.34 \mu\text{g m}^{-3}$ for the entire study (Fig. 8a).

443 **3.3.3 MO-OOA**

444 An obvious more oxygenated OA factor was also identified in this study according to its
445 significant high signal at m/z 44 ($\sim 25\%$) and the high average O/C ratio of 1.34 (Fig. 7i). The time
446 series of MO-OOA correlated closely ($R^2 = 0.7$) with sulfate and nitrate (Fig. 7c and Table S2).
447 Moreover, the mass spectrum of MO-OOA in this study resembled tightly to those more aged and
448 low-volatility oxygenated OA (LV-OOA) observed using AMS instruments at various sites
449 worldwide (Fig. S9), e.g., with R^2 of 0.89 and 0.97 to those in Lanzhou, China (Xu et al., 2014;
450 Zhang et al., 2017b), 0.96 to that in Paris, France (Crippa et al., 2013), 0.95 to that in Barcelona,
451 Spain (Mohr et al., 2012), as well as 0.70 and 0.71 to the standard LV-OOA mass spectrums
452 obtained from abundant AMS data sets by Ng et al. (2011a) and Crippa et al. (2014). The diurnal
453 variation of MO-OOA was mainly driven by the dynamic of PBL height, with high concentrations
454 during the nighttime yet relatively low concentrations during the daytime (Fig. 7f). This pattern

455 was quite different with those observed in previous studies that LV-OOA generally showed
456 elevated concentrations during the afternoon in accordance with strong photochemical activities,
457 suggesting that SOA at QOMS were mainly oxidized and aged during the long-range transport. On
458 average, MO-OOA contributed by 42.4% of the total OA mass, with an average mass
459 concentration of $1.02 \mu\text{g m}^{-3}$ for the entire study (Fig. 8a). As shown in Fig. 2f and 8c, MO-OOA
460 also displayed enhanced mass contributions during the clear periods, especially for period after
461 May 2 when the average mass fraction of MO-OOA increased up to $\sim 68\%$ of the total OA mass.

462 **3.4 Impact of BB emissions on aerosol characteristics**

463 **3.4.1 Sources of BB aerosols**

464 In order to understand the transport pathways and the potential source areas of aerosol, 3-day back
465 trajectories of air mass were calculated at an ending height of 500 m above ground level every 6 h
466 at the QOMS from April 12 to May 12, 2016. A four-cluster solution and the wildfire hotspots
467 around the QOMS during the entire measurement period were presented in Fig. 9. Cluster 1 and 2
468 (C1 and C2), which originated from the west of the QOMS and passed over many hotspot areas
469 (e.g., Indo-Gangetic Plain and Nepal), represented two polluted clusters. On the contrary, C3 and
470 C4, which accounted for half of the total back trajectories, were identified as clear clusters. C3
471 traveled a short distance from the southwest of the QOMS, whereas C4 was from the north of the
472 QOMS and passed over the inland of the TPH. The average PM_{10} mass concentrations for C1 and
473 C2 were 5.17 and $6.61 \mu\text{g m}^{-3}$, respectively, which were 2–3 times higher than those for the two
474 clear clusters (2.74 and $2.21 \mu\text{g m}^{-3}$). The mass contributions of OA and BBOA during C1 and C2
475 were up to more than 55% and 25% of the total PM_{10} mass on average, whereas weak contributions
476 were found for the clear clusters (C3 and C4), indicating the significant impacts of BB emissions
477 from South Asia on aerosol loadings at QOMS.

478 **3.4.2 Comparison of aerosol characteristics and air mass origins during different episodes**

479 As shown in Fig. 2, the mass concentrations and compositions of PM_{10} varied dynamically during
480 the entire sampling period. Two polluted periods (PP1 and PP2) and two clear periods (CP1 and
481 CP2) were identified. The comparisons of average mass concentrations and other indicators for the
482 four different episodes were presented in box plots in Fig. 10, whereas the corresponding back
483 trajectories of air masses and MODIS fire hotspots belong to each episode period were given in
484 Fig. 11, respectively.

485 During the two polluted periods, PM_{10} mass concentrations were much higher than those in
486 clear periods (8.06 and $7.87 \mu\text{g m}^{-3}$ for PP1 and PP2 vs. 2.76 and $1.82 \mu\text{g m}^{-3}$ for CP1 and CP2;
487 similarly hereinafter), with higher contributions from OA (60.1% and 57.5% vs. 48.1% and 43.9%)
488 and BBOA (38.3% and 36.6% vs. 14.3% and 7.5%) (Fig. 10 and 11). In addition, f_{60} were also
489 higher during polluted periods than those for clear periods (0.34% and 0.34% vs. 0.26% and 0.22%
490 on average) (Fig. 10). Air masses during PP1 and PP2 generally originated from long-range
491 transport to the west of the QOMS, which would pass through intense wildfires areas in South
492 Asia (e.g., Indo-Gangetic Plain and Nepal where showed high AOD values in Fig. 1d and active
493 fire hotspots in Fig. 9 and 11). The fire hotspot number around the air mass trajectories during PP2
494 was more than three times higher than those during other periods. Although the hotspot number
495 around the air mass trajectories during PP1 was not as abundant as that during PP2 and even

496 slightly lower than that during CP1, it was just collected within 3 days for PP1 whereas 8-10 days
497 for another periods. Hence, the BB activities were also more frequent and intense during the short
498 PP1 and finally resulted in the highest average PM₁ mass concentration among these periods.
499 Back trajectories in CP1 also originated from the west of QOMS and passed over the northern
500 India and Nepal, however, both the intensity of fire hotspot number (1089 hotspots in ~ 8 days)
501 and average FRP (19.6) were obvious lower than that in PP2. CP2 was the most clear period, of
502 which average PM₁ mass concentration was more than four times lower than those in polluted
503 periods. Back trajectories during CP2 period were from either the north of QOMS which passed
504 over inland areas of the TPH or the south of QOMS with quite short distance and low WS. These
505 results together suggested the significant roles of air mass sources and BB emissions to aerosol
506 characteristics at QOMS.

507 **3.4.3 Case study on the chemical evolution of BB emission aerosols**

508 In order to examine how atmospheric aging affects the aerosol chemistry characteristics at QOMS,
509 a typical evolution process of BB aerosol plume (referred as BB evolution case) was analyzed
510 from April 30 at 15:00 when a fresh BB plume occurred to May 1 at 18:00 when the BB plume
511 was highly aged after undergoing various atmospheric oxidation processes. The temporal
512 variations of meteorological parameters, mass concentrations and mass contributions of each PM₁
513 species and OA components as well as other chemistry parameters before and during this BB
514 evolution case were all shown in Fig. 12.

515 Before the BB evolution case, all the mass concentrations decreased slowly and synchronously
516 from 00:00 to 10:00 on April 30, which were consistent with the nearly stable trends of mass
517 contributions and other chemistry parameters, indicating the relatively unified air mass sources
518 and stable atmospheric conditions. After that, the wind circulations changed from the thermally-
519 driven down-slope winds (mostly southwest) to the weak up-slope winds (northeast). In this
520 period, BBOA and f_{60} values kept relatively stable in contrast to other species likely due to the
521 weak of air dilution and local sources. All the species reached the minimum at around 15:00 due to
522 the lift of PBL.

523 The BB evolution case in this study was further divided into three different situations (as
524 marked with arrows in Fig. 12 and 13), including the arriving of the fresh BBOA plume (from
525 15:00 to 24:00 on April 30), followed by the aqueous-phase oxidation in the nighttime (from 2:30
526 to 7:10 on May 1) and photochemical oxidation in the daytime (from 10:00 to 18:00 on May 1).
527 All the mass concentrations began to increase from 15:00 and finally reached the maximum PM₁
528 mass loading of 18.4 $\mu\text{g m}^{-3}$ at 24:00, which was about four times higher than the average PM₁
529 mass during the entire campaign. Thus continuous increase was mainly dominated by the dramatic
530 increase of BBOA, which reached up to 10.8 $\mu\text{g m}^{-3}$ and contributed 88% of the total OA mass
531 and 50% of the total PM₁ at 24:00 (Fig. 13a), suggesting a distinct presence of BB emissions
532 during this period. In contrast, the total OA mass was comprised by 78% of MO-OOA and 12% of
533 BBOA at 15:00. Similar continuous increase trend could also be found for the mass concentration
534 of calculated organic nitrate in this stage. In addition, nine aerosol chemistry parameters were
535 presented as a function of BBOA mass concentrations during this period (Fig. 13b). The mass
536 contributions of OA to PM₁ (f_{Org}) and BBOA to total OA (f_{BBOA}), f_{60} , and H/C ratio were all
537 increased with the increasing BBOA mass, whereas the mass contribution of MO-OOA to total

538 OA ($f_{\text{MO-OOA}}$), O/C ratio, carbon oxidation state ($\text{OS}_c = 2 \times \text{O/C} - \text{H/C}$) of OA, and aerosol single
539 scattering albedo (SSA) were decreased obviously, indicating the fresh nature of this BB plume.
540 The significant impacts of fresh BB plume during this period was mainly associated with the
541 unique wind circulation and the long-range transport of air masses. As displayed in Fig. 12b, the
542 wind circulation changed from the weak up-slope winds to the strong down-slope glacier winds on
543 April 30 at 15:00, with the WS increased from ~ 2 to 8 m s^{-1} . Meanwhile, the back trajectories in
544 this period also presented that the long-range transport of air masses passed over the northern
545 India and Nepal where active wildfires occurred, then air masses would accumulate and uplift to
546 cross the Himalayas and finally downward to QOMS with the strong glacier winds.

547 A distinct aqueous-phase oxidation process was found in the nighttime from 02:30 to 07:10 on
548 May 1. Although the total PM_1 and its species showed nearly stable mass concentrations during
549 this period, the BBOA mass decreased gradually (from 82% to 70%) whereas MO-OOA increased
550 constantly (from 14% to 20%) with the significant increase of RH (up to 91%) and aerosol liquid
551 water content (ALWC) (Fig. 12). The scattering plots of the aerosol chemistry parameters versus
552 the logarithmic values of cumulative ALWC, which could be used for the aqueous-phase oxidation
553 during transport, also showed apparent increase trends for $f_{\text{MO-OOA}}$, O/C ratio, OS_c , and SSA that
554 generally indicated the aerosol aging extent. All of these together suggested a distinct aqueous-
555 phase oxidation of BBOA in the nighttime.

556 Since sunrise, all the mass concentrations decreased gradually, mainly related to the increasing
557 PBL height and the clear air mass dilution. The back trajectories indicated that air masses during
558 this period firstly went into the inland of the north of QOMS where had rare wildfires. Moreover,
559 the BB plume would further undergo strong photochemical oxidation in the daytime due to the
560 strong solar radiation. MO-OOA just contributed 26% of the total OA mass at 10:00, but it could
561 increase to 74% at 18:00 after long-time photochemical oxidation. In contrast, BBOA mass
562 contribution decreased from 49% to 20%. The cumulative solar radiation, which denoted the total
563 amount of solar radiation that the plumes were exposed to during transport, could be used as an
564 indicator for the extent of photochemical aging in the daytime (Zhou et al., 2017). Clear increased
565 trend were found for $f_{\text{MO-OOA}}$, O/C ratio, OS_c , and SSA values with the increasing of cumulative
566 solar radiation, whereas decreased trend in f_{BBOA} , f_{NOA} , H/C ratio, and f_{60} values, suggesting a
567 possible oxidation mechanism that the relatively fresh BBOA and NOA oxidized to aged MO-
568 OOA in the daytime. Another interesting phenomenon was the continuous increase of SSA during
569 both the aqueous-phase and photochemical oxidation periods on May 1 (Fig. 12e and 13b),
570 indicating the potential influence of atmospheric aging to aerosol optical property at QOMS.

571 **4 Conclusions**

572 A comprehensive characterization of submicron aerosol chemical compositions and sources was
573 investigated at the QOMS during the pre-monsoon season in 2016. The average mass
574 concentration of PM_1 (NR- PM_1 + BC) was $4.44 (\pm 4.54) \mu\text{g m}^{-3}$ for the entire study, which was
575 much lower than those observed in various sites in China. OA was the dominant PM_1 species
576 (accounted for 54.3% of the total mass on average) and its contributions increased with the
577 increase PM_1 mass loading. The average size distributions of all PM_1 species displayed an
578 overlapping and narrow accumulation mode at $\sim 500 \text{ nm}$, indicating the internally well-mixed and
579 aged aerosol particles at QOMS. All species presented similar diurnal cycles, with lower

580 concentrations in the daytime whereas higher concentrations at the nighttime, mainly attributed to
581 the dynamic variations of PBL height. Three OA factors were identified by PMF analysis on the
582 high-resolution OA mass spectrum, including a relatively fresh BB-related OA (BBOA), a
583 nitrogen-containing OA (NOA) and a more-oxidized oxygenated OA (MO-OOA). BBOA and
584 MO-OOA could respectively account for 43.7% and 42.4% of OA mass on average, however,
585 their contributions to OA showed completely opposite variation trends with the increase of OA
586 mass. A continuously increased trend could be found for BBOA with the increasing OA,
587 suggesting the key role of BBOA during polluted periods when frequent and intense wildfires
588 were observed in South Asia. The significant impact of BB emissions on aerosol characteristics at
589 QOMS have been also illustrated for different air mass origins and periods, respectively. Elevated
590 PM₁ mass concentrations and high contributions of BBOA were found for both polluted clusters
591 and polluted periods. A case study of typical evolution process of BB aerosol plume was
592 investigated in detail to illustrate the chemical evolution of aerosol characteristics at QOMS. The
593 fresh BB plume occurred in the afternoon on April 30 and finally resulted in highly PM₁ mass
594 loading of 18.4 μg m⁻³, which was about four times higher than the average PM₁ mass during the
595 entire campaign. Obvious aqueous-phase oxidation and photochemical oxidation processes were
596 analyzed in the nighttime and daytime on May 1, respectively, both suggesting the oxidation
597 mechanism that fresh BBOA to aged MO-OOA. The continuous increase of SSA during the two
598 oxidation periods suggested the potential influence of atmospheric aging to aerosol optical
599 property at QOMS.

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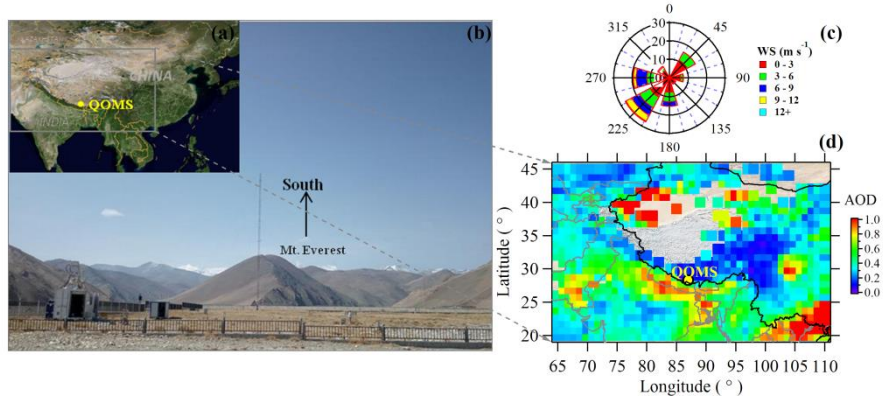
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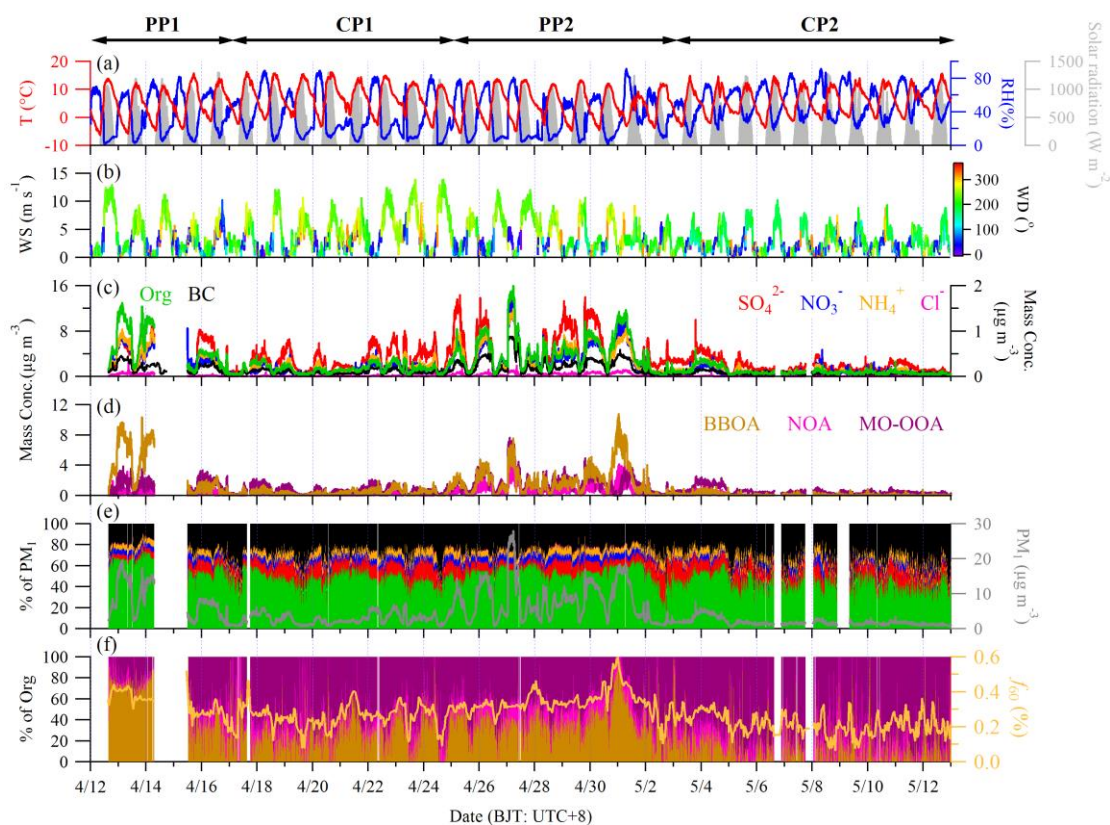
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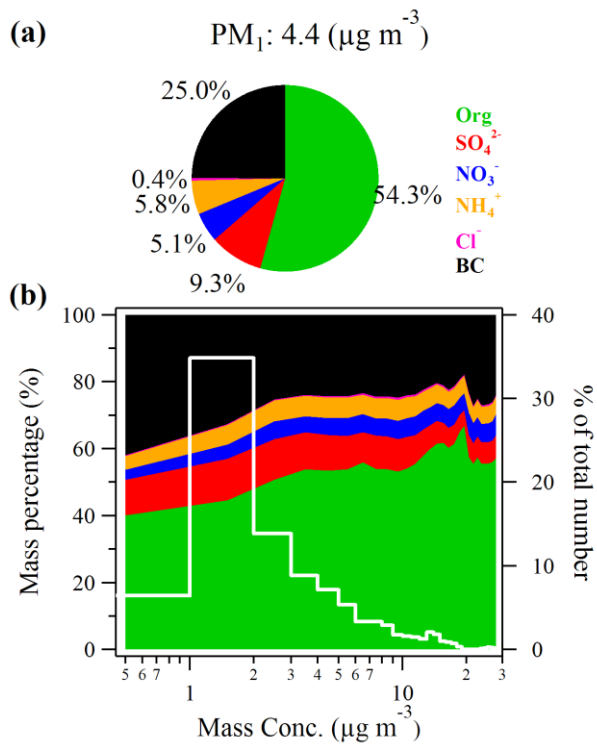
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879 **Figure 1.** (a) Location map for the QOMS, (b) picture for the QOMS and its surrounding, (c) wind rose plot colored by wind
 880 speed in this study, and (d) distribution of the average aerosol optical depth (AOD) around the QOMS retrieved from Terra
 881 MODIS at 550 nm during this study.



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883 **Figure 2.** Summary of meteorological and HR-ToF-AMS data. The 5-min time series of (a) ambient temperature (T), relative
 884 humidity (RH), and solar radiation, (b) wind speed (WS) colored by wind direction (WD), (c) mass concentrations of PM_{10}
 885 species, (d) mass concentrations of organic components, (e) mass contributions of PM_{10} species to total PM_{10}
 886 mass concentrations, and (f) mass contributions of organic components to organics. The time series of hourly average f_{60} (= $C_2H_4O_2^+ / OA$)
 887 values for the entire period is also showed. The markers of PP1 and PP2 represent the two polluted periods while
 888 CPI and CP2 are clear periods, respectively.



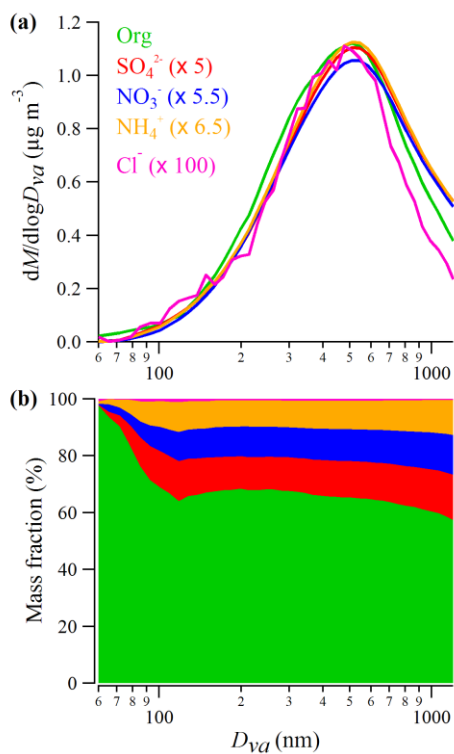
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Figure 3. The average mass contributions of PM_{10} (= NR- PM_{10} + BC) species (a) during the entire sampling period and (b) as a function of the total PM_{10} mass concentrations. The white solid line in (b) shows the percentage of the data number in each mass bins to the total data number.

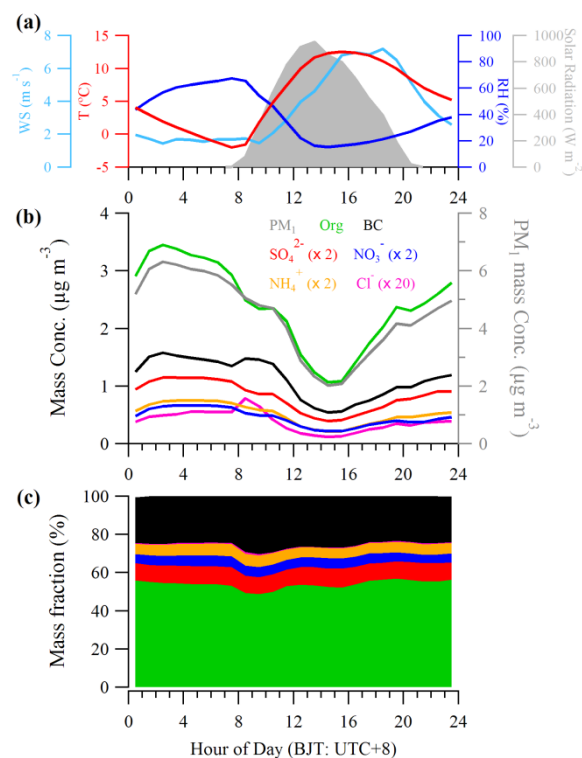


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Figure 4. The average size distributions of (a) mass concentrations and (b) mass contributions of NR- PM_{10} species for the entire study.

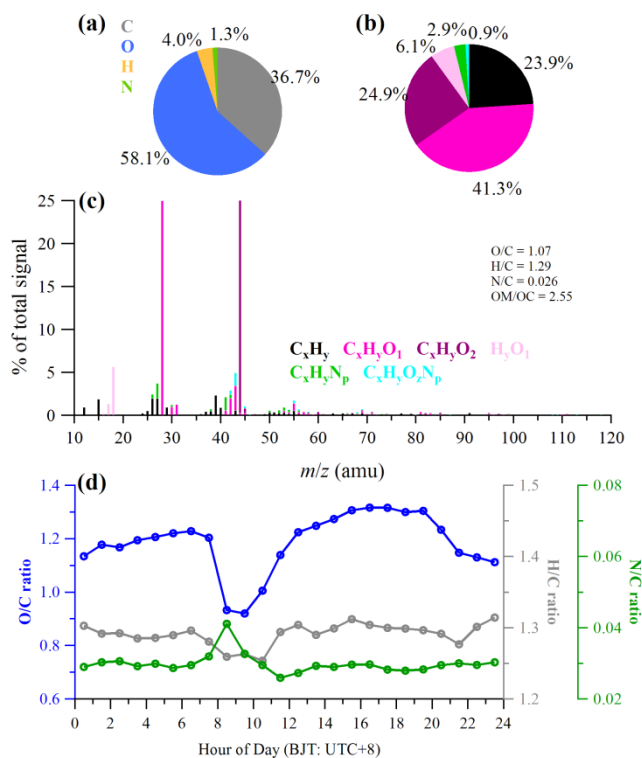


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Figure 5. The diurnal cycles of (a) meteorological parameters (temperature, RH, wind speed, and solar radiation), (b) mass concentrations and (c) mass contributions of PM₁ chemical species for the entire study.



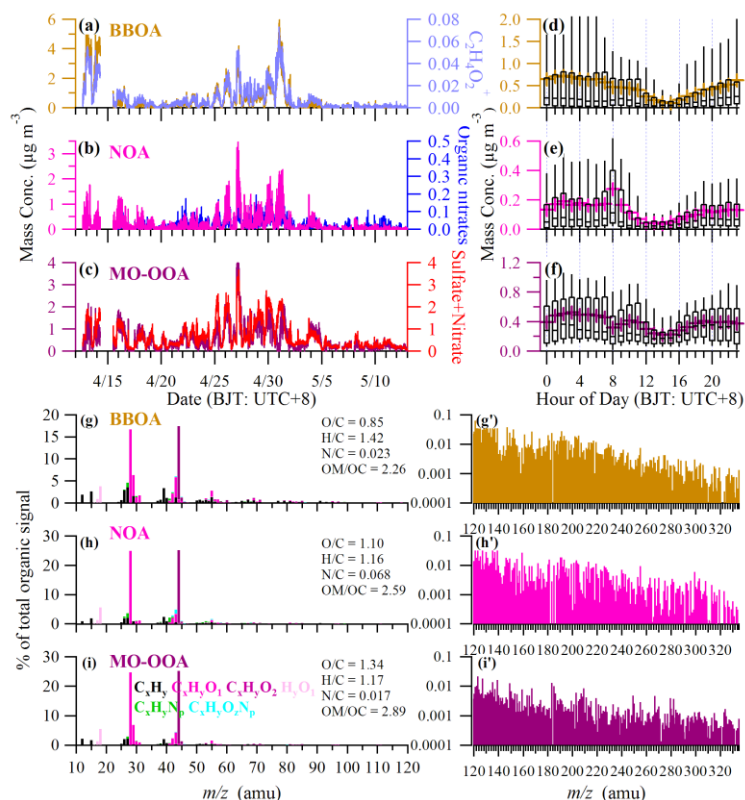
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Figure 6. The average contributions of (a) four elements (C, O, H, and N) and (b) six ion categories (colors as in (c)) to OA for the entire study; (c) the average high-resolution mass spectrum of OA (colors show six ion categories); (d) the diurnal variations of O/C, H/C, and N/C ratios.



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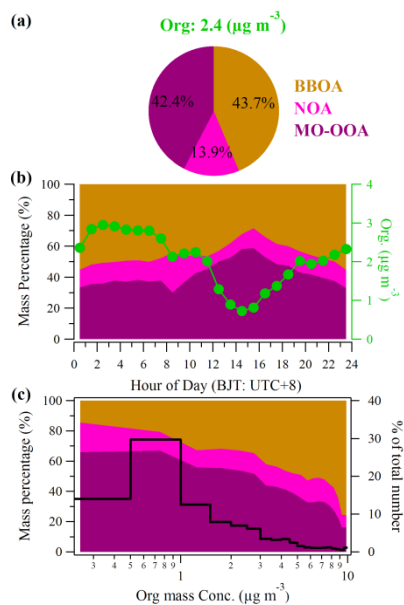
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Figure 7. The PMF results of (a–c) the time series of three OA factors and corresponding tracer species, (d–f) the diurnal variations of the mass concentrations of the three OA factors (the whiskers above and below the boxes indicate the 90th and 10th percentiles, the upper and lower boundaries respectively indicate the 75th and 25th percentiles, the lines in the boxes indicate the median values, and the cross symbols indicate the mean values), (g–i) high-resolution mass spectra of the three OA factors colored by six ion families at $m/z < 120$, and (g'–i') the unit resolution mass spectra at $m/z > 120$ for each OA factor.



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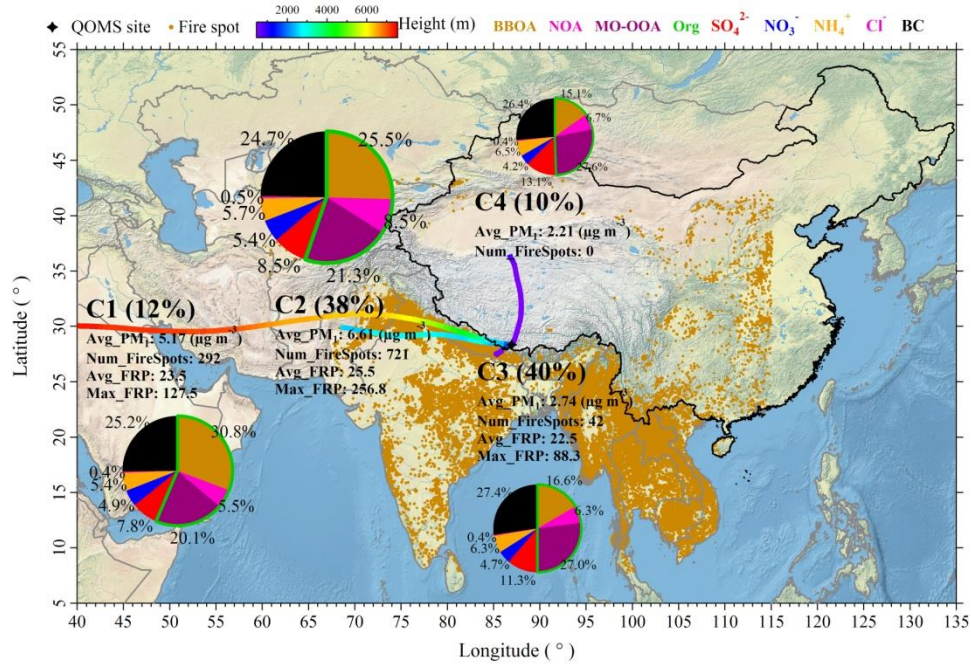
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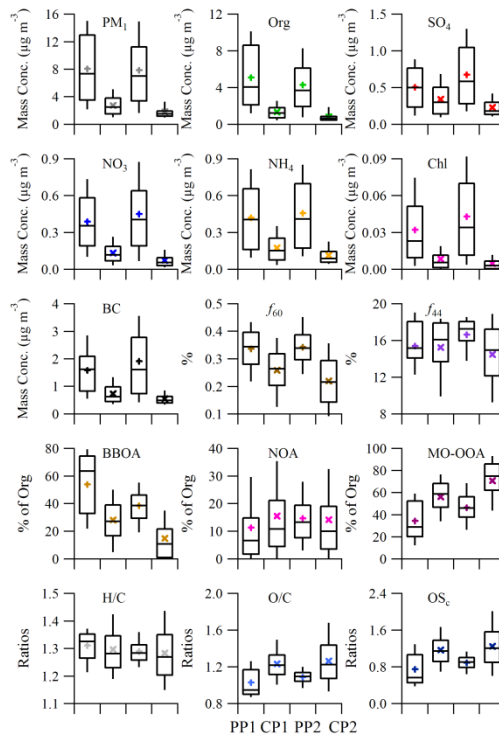
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Figure 8. (a) The average mass concentration of OA factors to total OA; (b) the diurnal variations of mass contributions of three OA factors to total OA and the total OA mass concentration; (c) The average mass contributions of three OA factors as a function of total OA mass concentrations. The black solid line in (c) shows the percentage of the data number in each OA mass bins to the total data number.



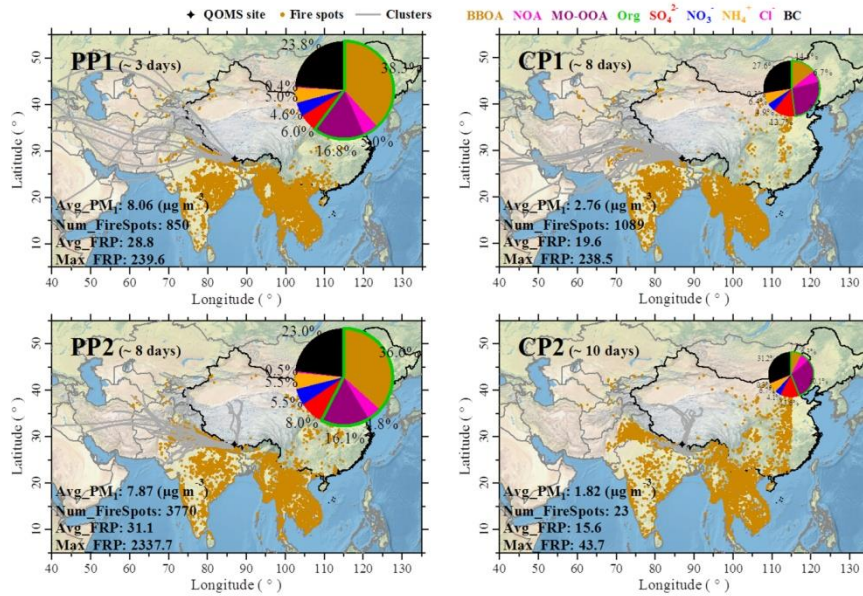
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915 **Figure 9.** The average back trajectory clusters during the entire study and the corresponding mass contributions of PM₁ species
 916 and OA factors to the total PM₁ mass. The areas of each pie charts are scaled by the corresponding average PM₁ mass
 917 concentrations. The average PM₁ mass concentrations, number of fire hotspots as well as the average and maximum fire radiative
 918 powers (FRP) belong to each clusters during the entire measurement period are also given.



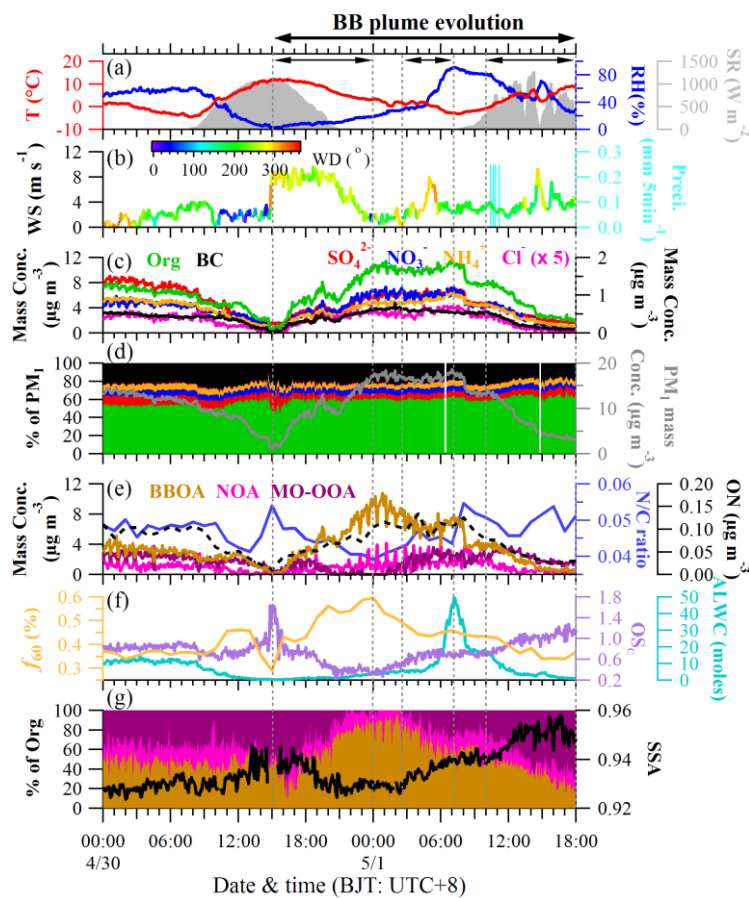
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920 **Figure 10.** Box plots of mass concentrations of total PM₁ and its species, *f*₆₀ and *f*₄₄ values, mass contribution of three OA
 921 components to organics, element ratios (H/C and O/C), and carbon oxidation states (OS_c) among the four polluted and clear
 922 periods. The whiskers indicate the 90th and 10th percentiles, the upper and lower boundaries of boxes indicate the 75th and 25th
 923 percentiles, the lines in the boxes indicate the median values, and the markers indicate the mean values.



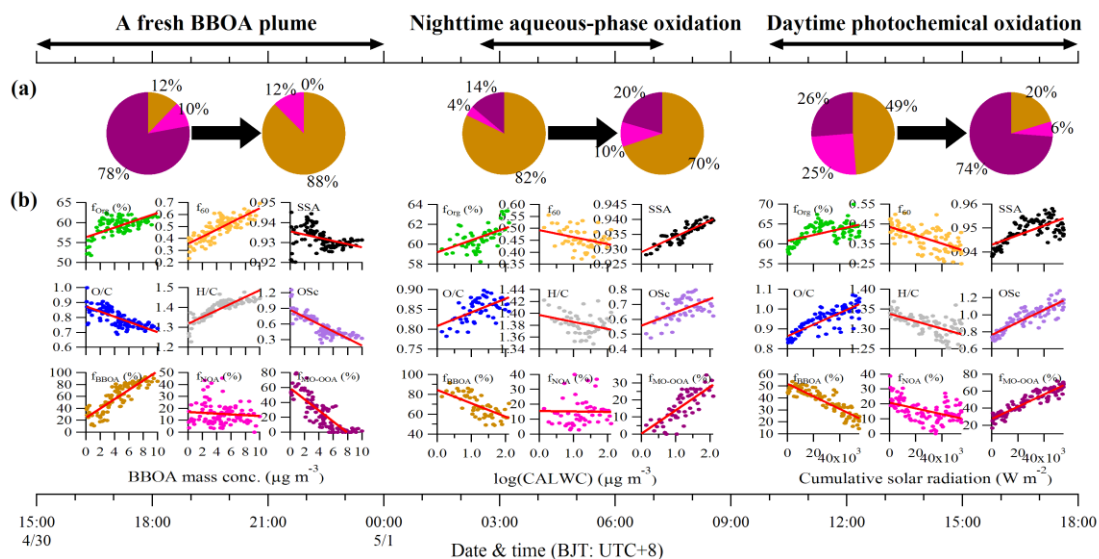
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925 **Figure 11.** The 72-h back trajectories (grey solid lines) calculated every 6 h for the different episodes. Pie charts show the
 926 average mass contributions of PM₁ species and OA factors to the total PM₁ mass for each episodes (scaled by the corresponding
 927 average PM₁ mass concentrations). The average PM₁ mass concentrations, number of fire hotspots as well as the average and
 928 maximum fire radiative powers (FRP) belong to all trajectories for the different episodes are also given.



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930 **Figure 12.** The temporal variations of meteorological parameters, mass concentrations and mass contributions of each PM₁
 931 species and OA components as well as the N/C ratio, f_{60} values, carbon oxidation states (OS_c), aerosol liquid water content
 932 (ALWC) and single scattering albedo (SSA) for the case study period from April 30 at 00:00 to May 1 at 18:00.



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Figure 13. Case study of chemical evolution of BB plume from April 30 at 15:00 to May 1 at 18:00. The periods marked with arrows are three distinct evolution processes. Pie charts in (a) are the mass contributions of three OA factors to total OA during the beginning and end time for each process, respectively. The scattering plots in (b) are the aerosol chemistry parameters as a function of BBOA mass concentration, logarithmic values of cumulative aerosol liquid water content (CALWC), and cumulative solar radiation for the corresponding process.