

1    **Supplementary information**

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3    **Identification of Criegee intermediates as potential  
4    oxidants in the troposphere**

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2 Table SI-1. Average concentrations with  $1\sigma$  standard deviation of measured unsaturated VOC  
 3 during the HUMPPA-COPEC 2010 and HOPE 2012 campaigns, together with the rate  
 4 coefficients of the reaction with ozone (IUPAC recommended values) (Atkinson et al., 2006).

Compound	[molecules cm <sup>-3</sup> ]		Rate coefficient with O <sub>3</sub> [cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> ]
	HUMPPA- COPEC 2010	HOPE 2012	
<b>isoprene</b>	(1.8 ± 1.8) x 10 <sup>9</sup>	(2.2 ± 2.2) x 10 <sup>9</sup>	1 x 10 <sup>-14</sup> exp(-1995/T)
<b>α-pinene</b>	(2.7 ± 3) x 10 <sup>9</sup>	(1.5 ± 1.5) x 10 <sup>9</sup>	8.1 x 10 <sup>-16</sup> exp(-640/T)
<b>β-pinene</b>	(1.9 ± 6.6) x 10 <sup>8</sup>	(9 ± 9) x 10 <sup>8</sup>	1.4 x 10 <sup>-15</sup> exp(-1270/T)
<b>3-carene</b>	(1.7 ± 2) x 10 <sup>9</sup>	(5.6 ± 4.7) x 10 <sup>8</sup>	4.8 x 10 <sup>-17, b</sup>
<b>myrcene</b>	(2.6 ± 2.7) x 10 <sup>8</sup>	(2.2 ± 1.6) x 10 <sup>8</sup>	2.7 x 10 <sup>-15</sup> exp(-520/T)
<b>limonene</b>	n.a.	(2.9 ± 2.1) x 10 <sup>8</sup>	2.8 x 10 <sup>-15</sup> exp(-770/T)
<b>sabinene</b>	n.a.	(9.2 ± 9.6) x 10 <sup>8</sup>	8.2 x 10 <sup>-17, b</sup>
<b>γ-terpinene</b>	n.a.	(1 ± 1) x 10 <sup>8</sup>	1.5 x 10 <sup>-16, b</sup>
<b>2-methylpropene</b>	n.a.	(4.2 ± 2.5) x 10 <sup>8</sup>	2.7 x 10 <sup>-15</sup> exp(-1630/T)
<b>but-1-ene</b>	n.a.	(1.4 ± 4.2) x 10 <sup>8</sup>	1.2 x 10 <sup>-17, a,b</sup>
<b>propene</b>	n.a.	(4.7 ± 3.7) x 10 <sup>8</sup>	5.5 x 10 <sup>-15</sup> exp(-1880/T)
<b>cis-2-butene</b>	n.a.	(6.1 ± 3.0) x 10 <sup>7</sup>	3.2 x 10 <sup>-15</sup> exp(-965/T)
<b>ethene</b>	n.a.	(7.3 ± 9.0) x 10 <sup>9</sup>	9.1 x 10 <sup>-15</sup> exp(-2580/T)

5 a, rate coefficient from Adeniji et al. (1981).

6 b, at 298 K

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3 Table SI-2. Average concentrations with  $1\sigma$  standard deviation of measured trace gas during  
 4 the HUMPPA-COPEC 2010 and HOPE 2012 campaigns, with the rate coefficients of the  
 5 reaction with OH (IUPAC recommended values) (Atkinson et al., 2006; Atkinson et al., 2004)

<b>Compound</b>	<b>[molecules cm<sup>-3</sup>]</b>		<b>Rate coefficient with OH [cm<sup>3</sup> molecule<sup>-1</sup> s<sup>-1</sup>]</b>
	<b>HUMPPA- COPEC 2010</b>	<b>HOPE 2012</b>	
<b>isoprene</b>	(1.8 ± 1.8) x 10 <sup>9</sup>	(2.2 ± 2.0) x 10 <sup>9</sup>	2.7 x 10 <sup>-11</sup> exp(390/T)
<b>α-pinene</b>	(2.7 ± 3) x 10 <sup>9</sup>	(1.5 ± 1.5) x 10 <sup>9</sup>	1.2 x 10 <sup>-11</sup> exp(440/T)
<b>β-pinene</b>	(1.9 ± 6.6) x 10 <sup>8</sup>	(9 ± 9) x 10 <sup>8</sup>	7.4 x 10 <sup>-11, a,b</sup>
<b>3-carene</b>	(1.7 ± 2) x 10 <sup>9</sup>	(5.6 ± 4.7) x 10 <sup>8</sup>	8.8 x 10 <sup>-11, a,b</sup>
<b>myrcene</b>	(2.6 ± 2.7) x 10 <sup>8</sup>	(2.2 ± 1.6) x 10 <sup>8</sup>	3.3 x 10 <sup>-10, b,c</sup>
<b>limonene</b>	n.a.	(2.9 ± 2.1) x 10 <sup>8</sup>	3 x 10 <sup>-11</sup> exp(515/T), <sup>d</sup>
<b>sabinene</b>	n.a.	(9.2 ± 9.6) x 10 <sup>8</sup>	1.2 x 10 <sup>-10, a,b</sup>
<b>γ-terpinene</b>	n.a.	(1 ± 1) x 10 <sup>8</sup>	1.7 x 10 <sup>-10, b</sup>
<b>MACR</b>	(1.0 ± 0.9) x 10 <sup>10</sup>	(1.4 ± 0.9) x 10 <sup>9</sup>	8 x 10 <sup>-12</sup> exp(380/T)
<b>ethanol</b>	(3.6 ± 2.2) x 10 <sup>10</sup>	(1.8 ± 1.1) x 10 <sup>10</sup>	3.2 x 10 <sup>-12</sup> exp(20/T)
<b>methanol</b>	(1.0 ± 1.4) x 10 <sup>11</sup>	(9.0 ± 3.4) x 10 <sup>10</sup>	9 x 10 <sup>-13, b</sup>
<b>ozone</b>	(1.1 ± 0.2) x 10 <sup>12</sup>	(1.1 ± 0.3) x 10 <sup>12</sup>	1.7 x 10 <sup>-12</sup> exp(-940/T)
<b>SO<sub>2</sub></b>	(1.4 ± 1.7) x 10 <sup>10</sup>	(2.3 ± 2.2) x 10 <sup>9</sup>	2 x 10 <sup>-12, b</sup>
<b>H<sub>2</sub>O<sub>2</sub></b>	(1.1 ± 1.0) x 10 <sup>10</sup>	n.a.	1.7 x 10 <sup>-12, b</sup>
<b>HO<sub>2</sub></b>	(9.0 ± 9.5) x 10 <sup>8</sup>	(1.4 ± 8.6) x 10 <sup>8</sup>	4.8 x 10 <sup>-11</sup> exp(250/T)
<b>NO</b>	(6.5 ± 7.0) x 10 <sup>8</sup>	(3.8 ± 5.0) x 10 <sup>9</sup>	1.3 x 10 <sup>-11, b</sup>
<b>NO<sub>2</sub></b>	(9.5 ± 5.0) x 10 <sup>9</sup>	(3.8 ± 2.4) x 10 <sup>10</sup>	1.1 x 10 <sup>-11, b</sup>
<b>CO</b>	(3.0 ± 1.2) x 10 <sup>12</sup>	(2.8 ± 0.4) x 10 <sup>12</sup>	2.1 x 10 <sup>-13, b</sup>

<b>HONO</b>	$(3.4 \pm 3.1) \times 10^9$	n.a.	$6.0 \times 10^{-12, b}$
<b>propanal</b>	n.a.	$(5.8 \pm 3.0) \times 10^9$	$4.9 \times 10^{-12} \exp(405/T)$
<b>acetaldehyde</b>	$(1.8 \pm 1.0) \times 10^{10}$	$(2.9 \pm 1.4) \times 10^{10}$	$1.5 \times 10^{-11, b}$
<b>formaldehyde</b>	$(1.4 \pm 1.6) \times 10^{10}$	$(2.1 \pm 0.4) \times 10^{10}$	$8.5 \times 10^{-12, b}$
<b>acetone</b>	$(8.2 \pm 3.8) \times 10^{10}$	$(6.0 \pm 2.2) \times 10^{10}$	$1.8 \times 10^{-13, b}$
<b>CH<sub>4</sub></b>	$(4.4 \pm 0.07) \times 10^{13}$	$(4.3 \pm 0.1) \times 10^{13}$	$6.4 \times 10^{-15, b}$
<b>2-methylpropene</b>	n.a.	$(4.2 \pm 2.5) \times 10^8$	$6.1 \times 10^{-11, a,b}$
<b>but-1-ene</b>	n.a.	$(1.4 \pm 4.2) \times 10^8$	$3.1 \times 10^{-11, a,b}$
<b>propene</b>	n.a.	$(4.7 \pm 3.7) \times 10^8$	$2.9 \times 10^{-11, b}$
<b>cis-2-butene</b>	n.a.	$(6.1 \pm 3.0) \times 10^7$	$6.4 \times 10^{-11, b}$
<b>ethene</b>	n.a.	$(7.3 \pm 9.0) \times 10^9$	$7.8 \times 10^{-12, b}$
<b>p-xylene</b>	n.a.	$(7.2 \pm 5.2) \times 10^8$	$2.0 \times 10^{-11, a,b}$
<b>benzene</b>	$(2.1 \pm 1.9) \times 10^9$	$(8.0 \pm 4.0) \times 10^8$	$1.2 \times 10^{-12, a,b}$
<b>ethylbenzene</b>	n.a.	$(2.3 \pm 2.1) \times 10^8$	$7.0 \times 10^{-12, a,b}$
<b>Toluene</b>	$(6.1 \pm 3.0) \times 10^9$	$(1.2 \pm 0.7) \times 10^9$	$5.6 \times 10^{-12, a,b}$
<b>ethane</b>	n.a.	$(1.8 \pm 0.3) \times 10^{10}$	$4.8 \times 10^{-11} \exp(250/T), ^a$
<b>propane</b>	n.a.	$(5.6 \pm 3.6) \times 10^9$	$1.1 \times 10^{-12, a,b}$
<b>methylpropane</b>	$(1.8 \pm 2.3) \times 10^9$	$(1.4 \pm 0.9) \times 10^9$	$2.1 \times 10^{-12, a,b}$
<b>butane</b>	$(1.8 \pm 1.6) \times 10^9$	$(2.0 \pm 1.2) \times 10^9$	$2.3 \times 10^{-12, a,b}$
<b>2-methylbutane</b>	$(1.6 \pm 1.2) \times 10^9$	n.a.	$3.6 \times 10^{-12, a,b}$
<b>n-pentane</b>	$(1.0 \pm 0.9) \times 10^9$	$(5.6 \pm 5.0) \times 10^9$	$3.8 \times 10^{-12, a,b}$

1 a, rate coefficient from (Atkinson and Arey, 2003).

2 b, at 298 K.

3 c, rate coefficient from (Hites and Turner, 2009)

4 d, rate coefficient from (Braure et al., 2014)

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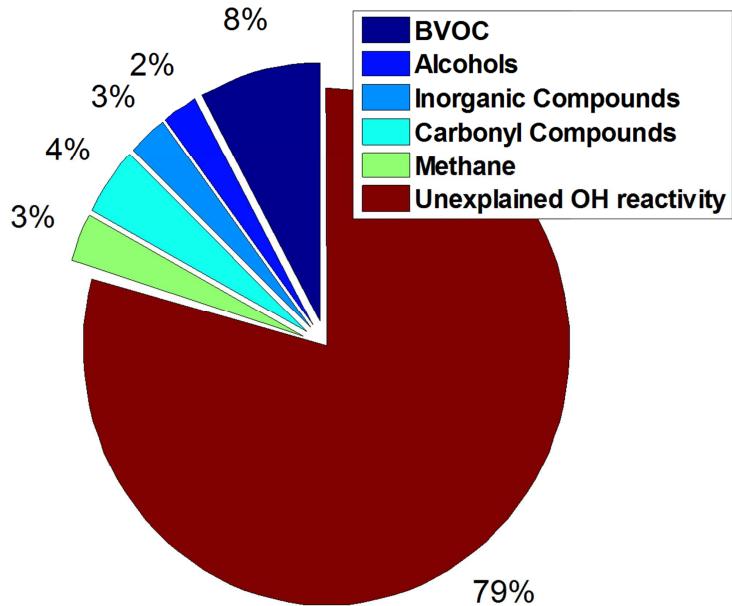
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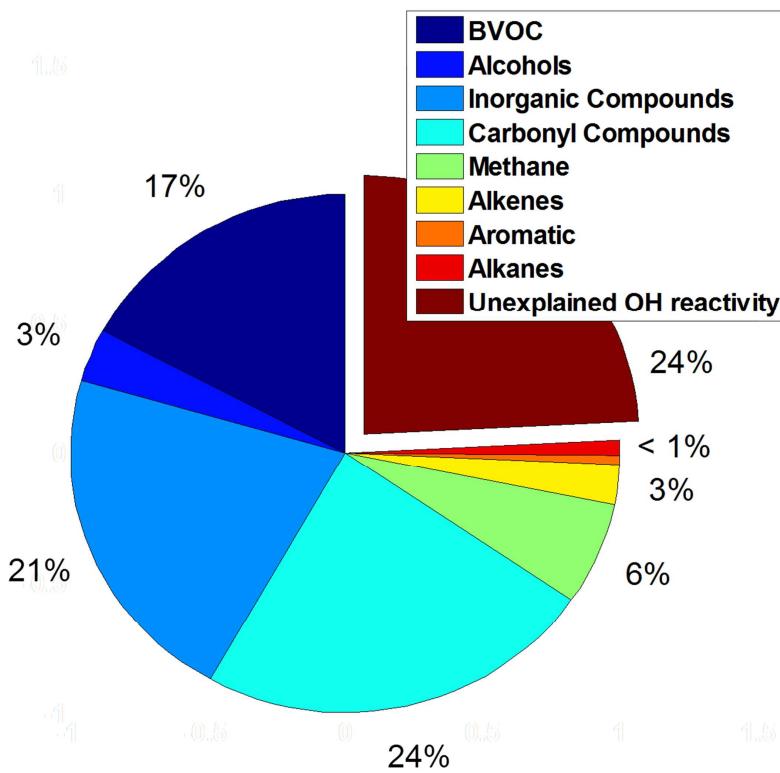
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3 Table SI-3. Average sum of concentrations with  $1\sigma$  standard deviation of BVOC (isoprene,  $\alpha$ -  
4 pinene,  $\beta$ -pinene, 3-carene, myrcene, limonene, sabinene,  $\gamma$ -terpinene) and temperature for the  
5 entire HOPE 2012 field campaign excluding the period between 26<sup>th</sup> to 28<sup>th</sup> of July 2012.

	$\Sigma[\text{VOC}] [\text{molecules cm}^{-3}]$	Temperature [ $^{\circ}\text{C}$ ]
HOPE 2012 campaign	$(5 \pm 4) \times 10^9$	$16 \pm 3$
26 <sup>th</sup> to 28 <sup>th</sup> of July 2012	$(1.3 \pm 0.9) \times 10^{10}$	$22 \pm 3$

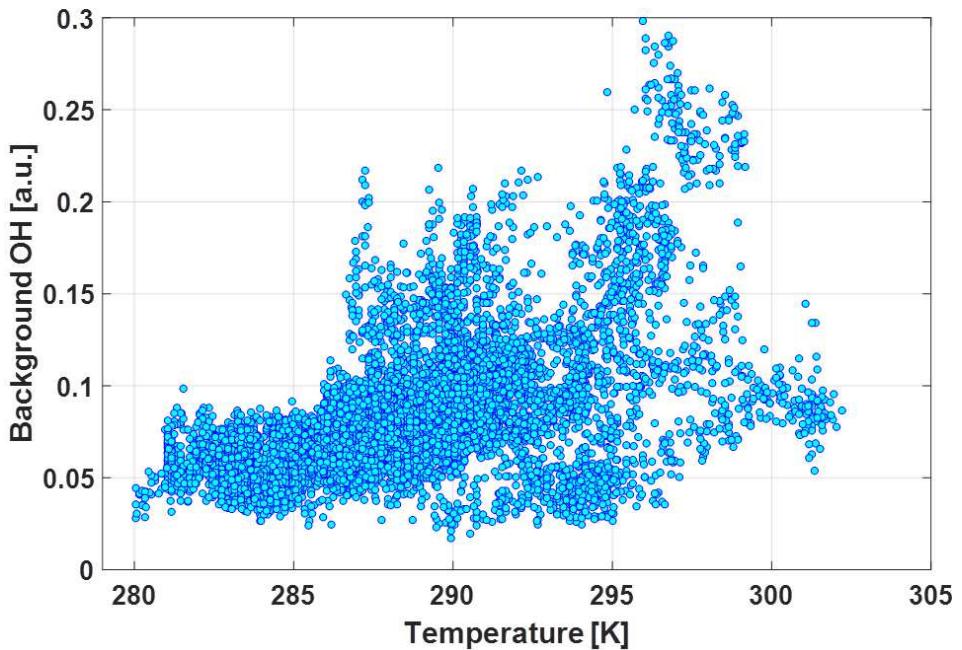
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2 Figure SI-1. Contributions of measured trace gases to the measured OH reactivity during the  
3 HUMPPA-COPEC 2010.

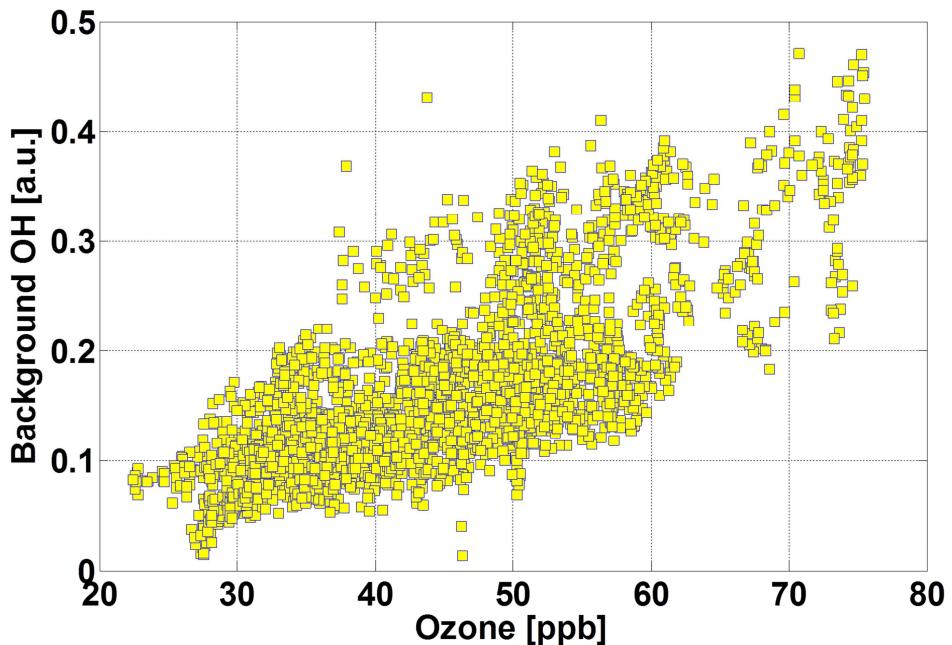


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5 Figure SI-2. Contributions of measured trace gases to the measured OH reactivity during the  
6 HOPE 2012.



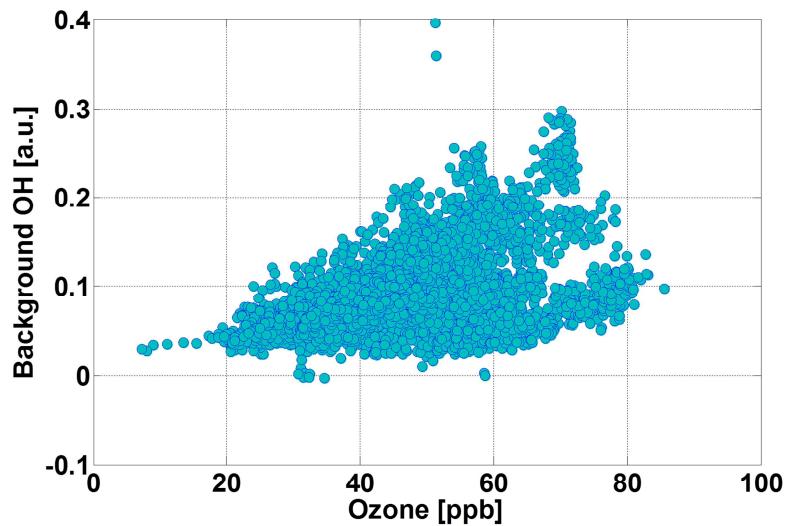
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2 Figure SI-3. Background OH as a function of temperature during the HOPE 2012 campaign.



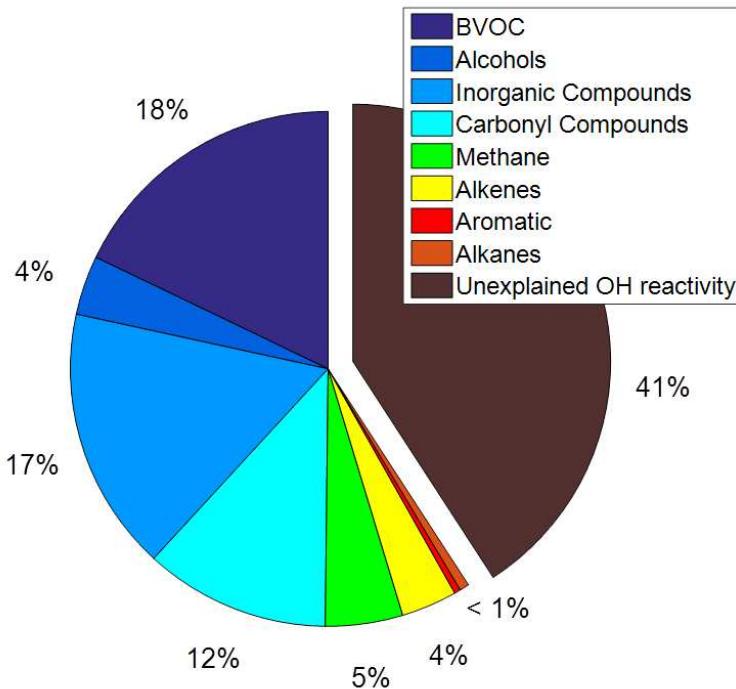
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4 Figure SI-4. Background OH as a function of the ozone concentration during the HUMPPA-  
5 COPEC 2010 campaign.



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2 Figure SI-5. Background OH signal as a function of ozone concentration during the HOPE  
3 2012 campaign.



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5 Figure SI-6. Contribution of measured trace gases to the measured OH reactivity during  
6 HOPE 2012 between the 1<sup>st</sup> and 3<sup>rd</sup> of August 2012.

1    **References**

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