



1 **Title: Biomass burning emissions in north Australia during the early dry season:**
2 **an overview of the 2014 SAFIRED campaign**

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38 **Abstract**

39

40 The SAFIRED (Savannah Fires in the Early Dry Season) campaign took place from
41 29th of May, 2014 until the 30th June, 2014 at the Australian Tropical Atmospheric
42 Research Station (ATARS) in the Northern Territory, Australia. The purpose of this
43 campaign was to investigate emissions from fires in the early dry season in northern
44 Australia. Measurements were made of biomass burning aerosols, volatile organic
45 compounds, polycyclic aromatic carbons, greenhouse gases, radon, mercury cycle, and
46 trace metals. Aspects of the biomass burning aerosol emissions investigated included;
47 emission factors of various emitted species, physical and chemical aerosol properties,
48 aerosol aging, micronutrient supply to the ocean, nucleation, and aerosol water uptake.
49 Over the course of the month-long campaign, biomass burning signals were prevalent
50 and emissions from several large single burning events were observed at ATARS.

51 Biomass burning emissions dominated the gas and aerosol concentrations in this
52 region. Nine major biomass burning events were identified and associated with intense
53 or close individual smoke plumes. Dry season fires are extremely frequent and
54 widespread across the northern region of Australia, which suggests that the measured
55 aerosol and gaseous emissions at ATARS are likely representative of signals across the
56 entire region of north Australia. Air mass forward trajectories show that these biomass
57 burning emissions are carried north west over the Timor Sea and could influence the
58 atmosphere over Indonesia and the tropical atmosphere over the Indian Ocean.

59 The outcomes of this campaign will be numerous. This region is an environment
60 with little human impact and provides a unique look into the characteristics of biomass
61 burning aerosol without the influence of other significant emission sources.
62 Relationships between the aerosol physical and chemical properties, gas concentrations



63 and meteorological data for the entire month will provide fundamental knowledge
64 required to understand the influence of early dry season burning in this tropical region
65 on the atmosphere. In this paper we present characteristics of the biomass burning
66 observed at the sampling site and provide an overview of the more specific outcomes
67 of the SAFIRED campaign.



68 **1. Introduction**

69 Tropical north Australia is dominated by savannah ecosystems. This region consists of
70 dense native and exotic grasslands and scattered trees and shrubs. Conditions are hot,
71 humid and wet in the summer months of December through March with hot, dry
72 conditions for the rest of the year giving rise to frequent fires between June and
73 November each year. Human settlements are relatively scarce in northern Australia,
74 outside of the territory capital, Darwin (population of 146 000). To the north of the
75 continent are the tropical waters of the Timor Sea, as well as the highly populated
76 Indonesian archipelago. South of the savannah grasslands are the Tanami, Simpson and
77 Great Sandy Deserts, spanning hundreds of thousands of square kilometers. Emissions
78 from fires in the savannah regions of northern Australia are therefore the most
79 significant regional source of greenhouse and other trace gases, as well as atmospheric
80 aerosol. Globally, savannah and grassland fires are the largest source of carbon
81 emissions (van der Werf et al., 2010; Shi et al., 2015) and play a significant role in the
82 earth's radiative budget. It is therefore important to quantify, characterise and fully
83 understand the emissions from savannah fires in northern Australia, taking into account
84 the complexity, variability and diversity of the species emitted.

85

86 In Australia approximately 550 000 km² of tropical and arid savannahs burn each year
87 (Meyer et al., 2012; Russell-Smith et al., 2007), representing 7% of the continent's land
88 area. In the tropical north of Australia, the fires during the early dry season in May/June
89 consist of naturally occurring and accidental fires, as well as prescribed burns under
90 strategic fire management practice to reduce the frequency and intensity of more
91 extensive fires in the late dry season in October and November (Andersen et al., 2005).
92 These fires in the early dry season burn with a low to moderate intensity and are



93 normally confined to the grass-layer. Events where fires reach the canopy level are rare.
94 These prescribed burns are an important process for the region and are undertaken by
95 local landholders with permits, as well as government supported bodies and volunteers.
96 There has been a recent push to reinstate traditional Aboriginal fire management
97 regimes in this region (Russell-Smith et al., 2013). Other fire management regimes are
98 implemented in similar environments around the world, such as the savannah
99 ecosystems of Africa (Govender et al., 2006) or the chaparral grasses in the United
100 States (Akagi et al., 2012). In general, fire management regimes are considered to
101 benefit regional biodiversity and can lead to the long-term increase in living biomass,
102 resulting in a reduction of greenhouse gas emissions (Russell-Smith et al., 2013).
103 Quantifying the emissions from dry season fires on regional scales is essential for
104 understanding the impact of these fires on the local and global atmosphere.

105

106 The components and concentrations of emissions from savannah fires are dependent
107 upon the vegetation and burning conditions. While CO₂ is the primary product of
108 biomass burning (BB), incomplete combustion also results in the emission of many
109 other trace gases such as CO, CH₄, NO_x, N₂O as well as non methane organic
110 compounds (NMOCs) and aerosol particles composed of elemental carbon, organic
111 carbon and some inorganic material (Crutzen and Andreae, 1990). The state of organics
112 in biomass burning aerosols can vary significantly due to the type of plant material
113 burned, the characteristics of the fires themselves as well as through aging processes in
114 the atmosphere.

115

116 The effects of these emissions on radiative forcing are complex. The global average
117 radiative forcing due to biomass burning aerosol-radiation interaction is estimated in



118 the 5th International Panel on Climate Change report as 0.0 W m^{-2} with an uncertainty
119 range of -0.20 to $+0.20 \text{ W m}^{-2}$ (Bindoff et al., 2013). It is well known that greenhouse
120 gases have a positive radiative forcing, heating up the atmosphere. Light absorbing
121 carbon in the aerosol phase will also result in a positive radiative forcing (Jacobson,
122 2001) by absorbing shortwave radiation. Conversely, the presence of aerosol organic
123 and inorganic matter can result in a negative radiative forcing by scattering solar
124 radiation (Penner et al., 1998). In addition, biomass burning has been shown to be a
125 significant source of cloud condensation nuclei (CCN), despite typically being
126 composed of weakly hygroscopic substances (Lawson et al., 2015), due to the high
127 number of particles emitted. This can result in a change in cloud droplet concentrations
128 and volume, thereby influencing cloud formation, albedo and lifetime. The contribution
129 of each species to the overall radiative forcing is also likely to change as smoke plumes
130 age (Liousse et al., 1995). Furthermore, not all biomass burning aerosol will interact
131 with radiation in the same way. For example, fresh BB emissions in the tropics has been
132 observed to be more absorbing than those from boreal forest fires (Wong and Li, 2002).
133 The role of biomass burning emissions is not limited to the Earth's radiative budget.
134 Certain species of emissions (e.g., mercury) can be deposited and sequestered in soil
135 (Gustin et al., 2008), vegetation (Rea et al., 2002) or bodies of water (LaRoche and
136 Breitbarth, 2005).

137

138 There is a need for a better scientific understanding of the influence biomass burning
139 has on atmospheric composition and air quality (Kaiser and Keywood, 2015).
140 Furthermore, the tropics are disproportionately under-sampled and the atmospheric and
141 ocean processes in these regions are of both regional and global consequence. The



142 SAFIRED campaign will contribute towards better understanding biomass burning
143 emissions and the atmospheric composition in tropical Australia.

144

145 **2. Description of experiment**

146 **2.1 Site**

147 The Australian Tropical Atmospheric Research Station (ATARS; 12°14'56.6"S,
148 131°02'40.8"E) is located on the Gunn Point peninsula in northern Australia (see Figure
149 1). ATARS is operated by the Australian Bureau of Meteorology and the CSIRO
150 (Commonwealth Scientific and Industrial Research Organisation). Standard
151 meteorological measurements (wind velocity, atmospheric pressure, precipitation) run
152 permanently at ATARS and two laboratories are in place for the installation of other
153 instruments. The SAFIRED campaign took place from 29th May 2014 until the 30th
154 June 2014, with personnel and instruments from nine institutes utilising these
155 laboratories to make comprehensive gaseous and aerosol measurements during this
156 period of the early dry season.

157 **2.2 Instruments and measurements**

158 **2.2.1 Trace Gases**

159 **Greenhouse gases**

160 Continuous measurement of CO₂, CO, CH₄ and N₂O were made using a high precision
161 FTIR trace gas and isotope Spectronus analyser, developed by the Centre for
162 Atmospheric Chemistry at the University of Wollongong. The analyser combines a
163 Fourier Transform Infrared (FTIR) Spectrometer (Bruker IRCube), a pressure and



164 temperature controlled multi-pass cell and an electronically cooled mercury cadmium
165 telluride detector. A detailed description of the instrument and concentration retrieval
166 technique are available in Griffith et al. (2012) and Griffith (1996).

167 **Ozone and other trace gases**

168 A Multi Axis Differential Optical Absorption Spectrometer (MAX-DOAS) was
169 installed on the top of one of the laboratories during the campaign. The technique has
170 been shown to provide vertical profile of nitrogen dioxide, ozone, sulfur dioxide,
171 formaldehyde, glyoxal and aerosol extinction (Sinreich et al., 2005; Honninger et al.,
172 2004). The MAX-DOAS instrument used in this campaign was designed and built at
173 the University of Wollongong. It consists of a vertically rotating prism capturing
174 scattered solar radiation at different angles (1°, 2°, 4°, 8°, 16°, 30° and a reference at
175 90°) into a fibre optic that carries the radiation to a UV-Visible spectrometer (AvaSpec
176 – ULS3648). Furthermore, a Thermo Scientific model 49i UV Photometric Ozone
177 analyser was used to measure ozone concentrations.

178 **Non methane organic compounds**

179 Online NMOC measurements were made using a high sensitivity Proton Transfer
180 Reaction-Mass Spectrometer (PTR-MS; Ionicon Analytik) using H_3O^+ as the primary
181 ion. The inlet was 10 m in length and drew air at 5 L min^{-1} from 2 m above the roof
182 (approx 5.5 m above ground level). The PTR-MS ran with inlet and drift tube
183 temperature of 60 °C, 600 V drift tube, and 2.2 mbar drift tube pressure, which equates
184 to an energy field of 135 Td. The PTR-MS sequentially scanned masses 15-190, with
185 1 second dwell time. The PTR-MS operated with the aid of auxiliary equipment which
186 regulates the flow of air in the sample inlet and controls whether the PTR-MS is
187 sampling ambient or zero air or calibration gas (Galbally et al., 2007). During the



188 campaign the PTR-MS was calibrated once per day for the following compounds using
189 certified gas standards from Apel Riemer Environmental Inc, USA and Air Liquide
190 Specialty Gases, USA: acetaldehyde, acetone, acetonitrile, benzene, methacrolein,
191 methanol, methyl ethyl ketone, toluene, 1,3,5-trimethyl benzene, m-xylene,
192 chlorobenzene, alpha pinene, 1,2-dichlorobenzene, 1,3,4 trichlorobenzene, dimethyl
193 sulphide and isoprene. Calibration data were used to construct sensitivity plots, which
194 were used to calculate approximate response factors for other masses not specifically
195 calibrated.

196

197 During sampling, carbonyls and dicarbonyls were trapped on S10 Supelco cartridges,
198 containing high-purity silica adsorbent coated with 2,4-dinitrophenylhydrazine
199 (DNPH, where they were converted to the hydrazone derivatives. Samples were
200 refrigerated immediately after sampling until analysis. The derivatives were extracted
201 from the cartridge in 2.5 mL of acetonitrile and analysed by high performance liquid
202 chromatography with diode array detection. The diode array detection enables the
203 absorption spectra of each peak to be determined. The difference in the spectra
204 highlights which peaks in the chromatograms are mono- or dicarbonyl DNPH
205 derivatives and, along with retention times, allows the identification of the dicarbonyls
206 glyoxal and methylglyoxal. Further details can be found in Lawson et al. (2015).

207 PAHs

208 PAHs were sampled through a high-volume air sampler (Kimoto Electric Co., LTD.)
209 using a sampling rate typically at $\sim 60 \text{ m}^3 \text{ h}^{-1}$. The sampling rate was calibrated using
210 an orifice plate prior to the sampling campaign and the sampling volume was calculated
211 based on the calibrated sampling rate and sampling duration. A bypass gas meter
212 installed on the sampler was used to monitor any anomalous fluctuation of the sampling



213 rate during the sampling period. Particle-associated and gaseous PAHs were collected
214 on glass fibre filters (Whatman™, 203×254 mm, grade GF/A in sheets) and subsequent
215 polyurethane foam plugs respectively. The glass fibre filters and polyurethane foam,
216 along with the field blank samples, were extracted separately using an Accelerated
217 Solvent Extractor (Thermo Scientific™ Dionex™ ASE™ 350) after being spiked with
218 a solution containing 7 deuterated PAHs (i.e. ²D₁₀-phenanthrene, ²D₁₀-fluoranthene,
219 ²D₁₂-chrysene, ²D₁₂-benzo[b]fluoranthene, ²D₁₂-BaP, ²D₁₂-indeno[1,2,3-cd]pyrene,
220 ²D₁₂-benzo[g,h,i]perylene) at different levels as internal standards for quantification
221 purposes. Concentrated extracts were cleaned up by neutral alumina and neutral silica.
222 Eluents were carefully evaporated to near dryness and refilled with 250 pg of ¹³C₁₂-
223 PCB (polychlorinated biphenyl) 141 (in 25 μL isooctane) employed as the
224 recovery/instrument standard for estimating the recoveries of the spiked internal
225 standards and monitoring the performance of the analytical instrument. Samples were
226 analysed using a Thermo Scientific™ TRACE™ 1310 gas chromatograph coupled to
227 a Thermo Scientific™ double-focusing system™ Magnetic Sector high resolution mass
228 spectrometer. The HRMS was operated in electron impact-multiple ion detection mode
229 and resolution was set to ≥ 10,000 (10% valley definition). An isotopic dilution method
230 was used to quantify 13 PAH analytes including phenanthrene, anthracene,
231 fluoranthene, pyrene, benzo[a]anthracene, chrysene, benzo[b]fluoranthene,
232 benzo[k]fluoranthene, benzo[e]pyrene, BaP, indeno[1,2,3-cd]pyrene,
233 dibenzo[a,h]anthracene, benzo[g,h,i]perylene.

234 **Mercury**

235 Total gaseous mercury, gaseous elemental mercury + gaseous oxidised mercury (TGM;
236 GEM + GOM), was sampled from a 10 m mast and measured via gold pre-concentration
237 and cold vapour atomic fluorescence spectroscopy using a Tekran 2537X instrument.



238 Simultaneously, GEM, GOM and PBM were individually measured using a Tekran
239 2537B connected to a combined Tekran 1130/1135 speciation unit sampling at a 5.4 m
240 height. The sampling train of the 1130/1135 collects first GOM (KCl-coated denuder)
241 then PBM (quartz wool pyrolyser) in series from a 10 L min^{-1} sampling flow, allowing
242 GEM only to flow onwards for detection by subsampling by the 2537B. Due to the
243 small atmospheric concentrations of GOM and PBM, pre-concentration occurred over
244 a 1-hour period with subsequent analysis taking an additional hour. Continuous
245 measurements of GEM at 5-minute resolution were made possible for the 2537B unit
246 by rotating pre-concentration/analysis roles of the two internal gold traps. Both 2537
247 units sampled at 1 L min^{-1} and were calibrated every 23 hours using an internal mercury
248 permeation source. For more information on the 2537 and 1130/1135 systems see
249 Landis et al. (2002) and Steffen et al. (2008).

250

251 GEM fluxes were measured using the methods outlined in Edwards et al. (2005). Air
252 samples were drawn at heights of 5.2 and 8.0 m through 46.4 m of nylon tubing using
253 a PTFE diaphragm pump operating at 10 L min^{-1} . Subsampling from this flow through
254 a $0.2 \mu\text{m}$ PTFE filter at 1 L min^{-1} by a Tekran 2537A, and switching between sample
255 intakes, allowed resolution of a GEM gradient every 30 minutes. The transfer velocity
256 was measured using a Campbell Scientific CSAT3 sonic anemometer and LI-COR
257 7200 closed path infrared gas analyser for CO_2 , both located on the same tower as the
258 gradient intakes at 6.6 m and sampling at 20 Hz.

259 **Radon**

260 Radon-222 (radon) is a naturally occurring radioactive noble gas that arises from the
261 alpha-particle decay of radium-226, which is ubiquitous in most soil and rock types.
262 With a half-life of 3.82 days, radon has thus proven to be an excellent indicator of recent



263 (within 2-3 weeks) terrestrial influences on air masses for observations at coastal or
264 island sites (Chambers et al., 2014). Radon is unreactive and poorly soluble, and its
265 only atmospheric sink is radioactive decay. Furthermore, it has a half-life comparable
266 to the lifetimes of short-lived atmospheric pollutants (e.g., NO_x, SO₂) and atmospheric
267 residence time of water and aerosols. Radon's unique combination of physical
268 characteristics make it an ideal tracer for (i) regional air mass transport studies, in which
269 it is often used in conjunction with air mass back trajectories for fetch analyses
270 (Williams et al., 2009; Chambers et al., 2014); (ii) investigations of vertical mixing
271 processes within the daytime convective boundary layer (Williams et al., 2011) or
272 nocturnal boundary layer (Chambers et al., 2015); (iii) identifying periods of minimal
273 terrestrial influence on a measured air mass ("baseline" studies; (Chambers et al.,
274 2016)); (iv) performing regional flux or inventory analyses for trace atmospheric
275 species with similar source distributions (Biraud et al., 2000) and (v) evaluating the
276 performance of transport and mixing schemes in climate and chemical-transport models
277 (Locatelli et al., 2015).

278

279 In order to measure Radon concentrations, a 700 L dual-flow-loop two-filter radon
280 detector, designed and built by the Australian Nuclear Science and Technology
281 Organisation (Whittlestone and Zahorowski, 1998; Chambers et al., 2014), was installed
282 at the ATARS in 2011 and has been fully operational since July 2012. The detector
283 provided continuous hourly radon concentrations for the duration of the SAFIRED
284 campaign, sampling air at 40 L min⁻¹ from 12 m above ground level through 25 mm
285 high-density polyethylene agricultural pipe. A coarse aerosol filter and dehumidifier
286 were installed "upstream" of the detector, as well as a 400 L delay volume to ensure
287 that thoron (²²⁰Rn, half-life 55 s) concentrations in the inlet air stream were reduced to



288 less than 0.5 % of their ambient values. The detector's response time is around 45
289 minutes, and the lower limit of detection is 40 - 50 mBq m⁻³. Calibrations are performed
290 on a monthly basis by injecting radon from a PYLON 101.15±4% kBq Ra-226 source
291 (12.745 Bq min⁻¹ ²²²Rn), traceable to NIST standards, and instrumental background is
292 checked every 3 months. In post processing, half-hourly raw counts were integrated to
293 hourly values before calibration to activity concentrations (Bq m⁻³).

294 **Aerosols**

295 **Aerosol Drying System**

296 An Automated Regenerating Aerosol Diffusion Dryer (ARADD) is permanently
297 installed on the roof of the laboratory containing the aerosol instrumentation for this
298 campaign. This was used in front of the aerosol manifold to continuously dry the aerosol
299 sample. The ARADD design, similar to that described by Tuch et al. (2009),
300 continuously conditions the aerosol sample to a relative humidity of below 40% with
301 maximum aerosol transmission efficiency. The ARADD utilizes two diffusion drying
302 columns in parallel, each containing 7 stainless steel mesh tubes of 10 mm internal
303 diameter and approximately 800 mm length, surrounded by a cavity packed with silica
304 gel. The aerosol sampled is directed into one column at a time, while the other column
305 is regenerated by an ultra-dry compressed air system. All flows are controlled by
306 software that directs sample flow and compressed air flow to the appropriate column
307 with a series of valves. The ARADD has a Total Suspended Particulate style intake at
308 the inlet of the aerosol sample path, and a sample manifold at the exit of the system to
309 provide sampling take-offs for the various aerosol instruments connected to the
310 ARADD. Flow through the ARADD is provided by the instruments and pumps
311 connected downstream. The ambient and inlet relative humidity for the entire sampling
312 period were logged and are displayed in Supplementary Figure 1.



313 **Aerosol Size**

314 Aerosol size distributions were measured with a Scanning Mobility Particle Sizer
315 (SMPS). A TSI 3071 long-column electrostatic classifier with a TSI 3772 Condensation
316 Particle Counter (CPC) measured the size distribution over a range of 14 nm to 670 nm
317 at a scan interval of 5 minutes.

318

319 In addition to the aerosol size distributions measured by the SMPS, neutral and charged
320 aerosol particle distributions from 0.8 nm to 42 nm were measured using a Neutral
321 cluster and Air Ion Spectrometer (NAIS)(Manninen et al., 2009;Mirme et al., 2007). In
322 this study, the NAIS was set to operate in a cycle of 4 min including ion and neutral
323 particle sampling periods of 2 and 1 minute, respectively, with the remaining minute
324 being an offset period which is required to neutralize and relax the electrodes. The total
325 sampling air flow was 60 L min^{-1} , the high flow rate being used to minimize ion
326 diffusion losses and maximize the measured ion concentration sensitivity. Ion losses
327 are accounted for during post-processing of the data by the software (Mirme et al.,
328 2007).

329 **Aerosol Composition and Water Uptake**

330 PM_{10} and PM_{10} 12-hour filter samples (night and day) were collected on a TAPI 602
331 Beta plus particle measurement system (BAM). Portions of the PM_{10} filters have been
332 analysed for elemental and organic carbon mass loadings using a DRI Model 2001A
333 Thermal-Optical Carbon Analyzer following the IMPROVE-A temperature protocol
334 (Chow et al., 2007b). Additional portions of the PM_{10} filters were extracted in 5 ml of
335 $18.2 \text{ m}\Omega$ de-ionized water and preserved using 1% chloroform. These extracts will be
336 analysed for major water-soluble ions by suppressed ion chromatography and for



337 anhydrous sugars including levoglucosan by high-performance anion-exchange
338 chromatography with pulsed amperometric detection (Iinuma et al., 2009).

339

340 Daily aerosol filters were collected using two Ecotech 3000 high-volume volumetric
341 flow controlled aerosol samplers with PM₁₀ size selective inlets. One high-volume
342 sampler was used to collect aerosols on acid cleaned Whatman 41 filters to determine
343 the soluble and total fraction of trace metals. Soluble trace metals were extracted from
344 a filter aliquot using ultra-pure water (>18.2 mΩ) leaching experiments. Total trace
345 metal concentrations were determined by digesting a second filter aliquot with
346 concentrated nitric and hydrofluoric acids. Leachates and digested solutions were
347 analysed by high resolution inductively couple plasma mass spectrometry. The second
348 sampler was used to collect a set of aerosol samples on quartz filters for elemental and
349 organic carbon analysis following (Chow et al., 2007a), and major anion and cation
350 analysis.

351

352 The volatility and hygroscopicity of 50 nm and 150 nm particles were measured with a
353 custom built Volatility and Hygroscopicity Tandem Differential Mobility Analyser
354 (VH-TDMA). Inlet dried particles were size selected (alternating between 50 and 150
355 nm) using a TSI 3080 electrostatic classifier. Scans alternated between two different
356 sample pathways. In the first, after size selection, particles were passed through a
357 thermodenuder set to 120°C. The sample line was then split so that half went to an
358 SMPS comprised of a TSI 3080 classifier and a TSI 3010 CPC (V-TDMA). The rest of
359 the sample was passed through a humidifying system that exposed the particles to a
360 relative humidity of 90% before being brought into another SMPS with a 3080 classifier
361 and 3010 CPC (H-TDMA). Alternatively, the thermodenuder was bypassed in every



362 second scan so that the V-TDMA was used to verify the size selection and the H-TDMA
363 was able to observe the hygroscopic growth of ambient particles. Each scan ran for 3
364 minutes, giving a full set of data every 12 minutes.

365

366 The chemical composition and properties of non-refractory sub-micron particles were
367 investigated with a compact Time-of-Flight Aerosol Mass Spectrometer (cToF-AMS,
368 Aerodyne Research, Inc.) and a Time of Flight Aerosol Chemical Speciation Monitor
369 (ToF-ACSM, Aerodyne Research, Inc.). Both of these instruments operate with the
370 same principle and have many identical components. An aerodynamic lens in the inlet
371 of each instrument focuses the particles into a beam and differential pumping removes
372 most of the gas phase. Particles are flash vaporized at 600°C and ionized by electron
373 impact before passing through a time-of-flight mass spectrometer to a multi-channel
374 plate detector in the cToF-AMS and a dynode detector in the ToF-ACSM. The cToF-
375 AMS has the added benefit of having a particle Time-of-Flight (pToF) mode, which
376 allows the size resolved chemical composition to be measured. Both instruments
377 sampled through a PM_{2.5} inlet and nafion dryer. In addition, the inlet of the cToF-AMS
378 was incorporated into the VH-TDMA system, so that when the VH-TDMA was
379 measuring ambient particles, the cToF-AMS would draw particles through the
380 thermodenuder set at 120°C and vice-versa. This gives additional information about the
381 chemical composition of the volatile component of submicron particles.

382

383 The number of particles activated to cloud droplets were measured using a Continuous-
384 Flow Steam Wise Thermal Gradient Cloud Condensation Nuclei Counter (CCNC) from
385 Droplet Measurement Technologies Inc. (DMT, model No. 100). Particles were



386 exposed to a 0.5% supersaturation and activated particles greater than 1 μ m were
387 counted with an Optical Particle Counter using a 50 mW, 658 nm laser diode.

388 **Back trajectories**

389 Hourly 10-day air mass back trajectories terminating at ATARS were produced using
390 the NOAA HYSPLIT model (Draxler and Rolph, 2003), and catalogued in a data base
391 for use with the SAFIRED campaign data set. Global Data Assimilation System input
392 files with 0.5° resolution were obtained from NOAA ARL FTP site
393 (<http://ready.arl.noaa.gov/gdas1.php>) to drive the HYSPLIT model.

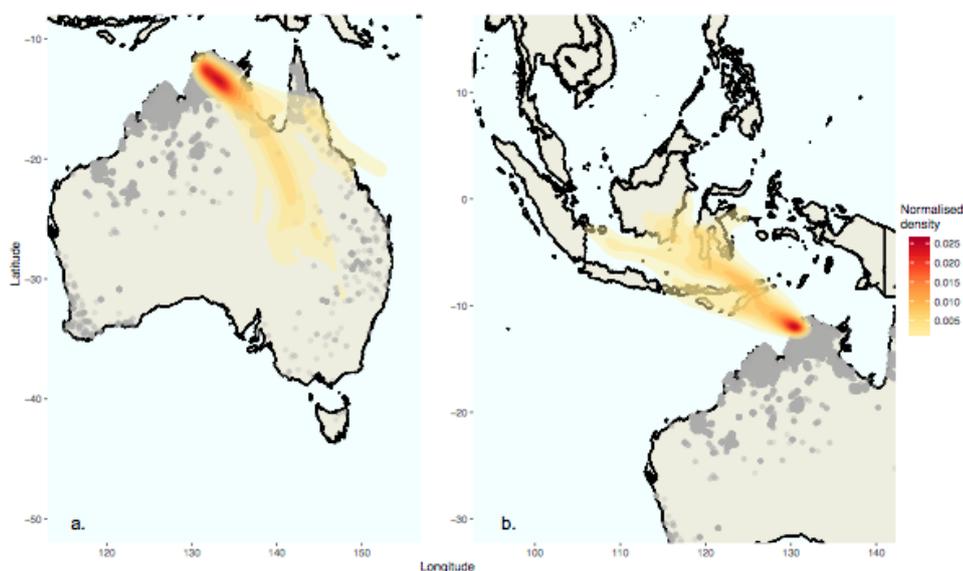
394 **Satellite detection of fires**

395 Data on the location of fires was collected from the Australian national bushfire
396 monitoring system, Sentinel Hotspots. Hotspot locations are derived from the Moderate
397 Resolution Imaging Spectroradiometer (MODIS) sensors on the Terra and Aqua
398 satellites and the Visible Infrared Imaging Radiometer Suite (VIIRS) sensor on the
399 Suomi NPP satellite. The Terra, Aqua and Suomi NPP satellites fly over the region
400 around ATARS at approximately 10:30 am, 3 pm and 2:30 pm, respectively. Detection
401 of fires is therefore limited to those that are flaming during these times.



402 3. Overview of Campaign

403 Fires and air masses



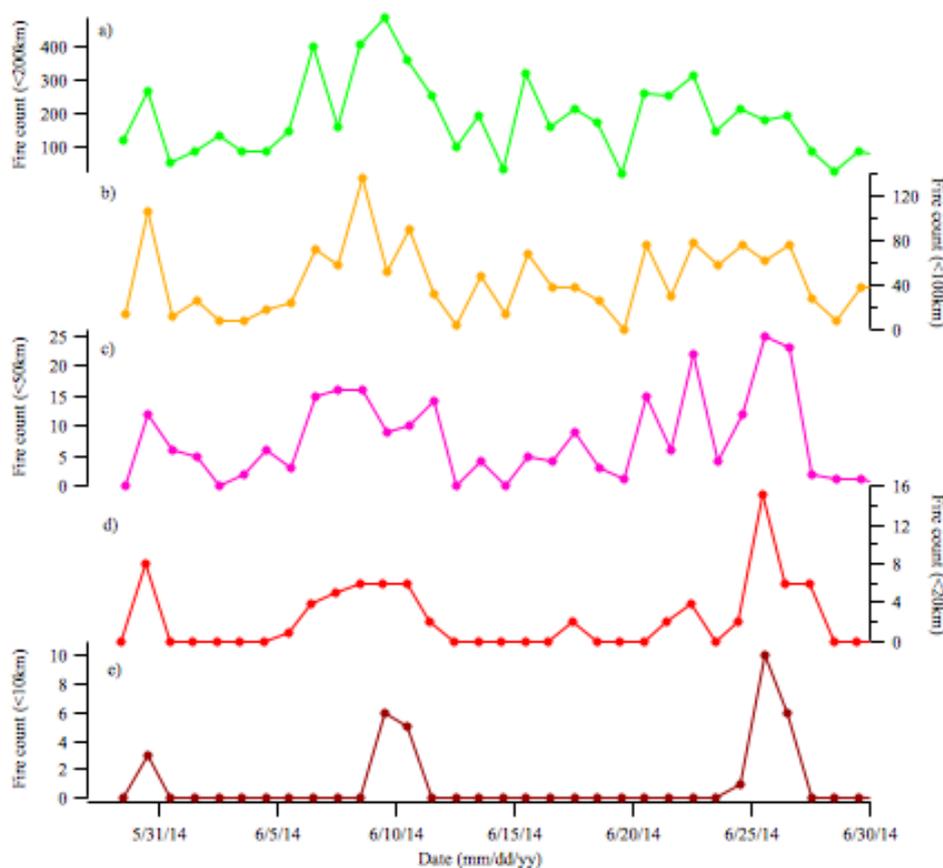
404

405 **Figure 1** The HYSPLIT (a) back and (b) forward trajectories to and from the ATARS site, truncated to five
406 days. The colour scale represents the air mass trajectory density, normalised to 1. Grey dots are MODIS and
407 VIIRS detected hotspots from the 30th of May 2016 until the 29th of June 2016.

408 Thousands of fires were observed in during the period of the SAFIRED campaign in
409 Australia by the MODIS and VIIRS sensors on the Terra and Aqua NASA satellites.
410 The vast majority of these occurred in the savannah regions of northern Australia
411 (Figure 1). The number of detected fires on each day within 10 km, 20 km, 50 km, 100
412 km and 200 km of the sampling location was determined (Figure 2). Several fires within
413 10km were detected on the 30th of May and the 9th, 10th, 25th and the 26th of June. These
414 correspond with the highest gaseous and aerosol signals. The periods between the 11th
415 and 24th of June had very few detected fires within 50km of the station. Satellite flyby
416 times were in the early afternoon local time each day and therefore fires not occurring
417 during these times would not be recorded. Airmass back trajectories from the ATARS



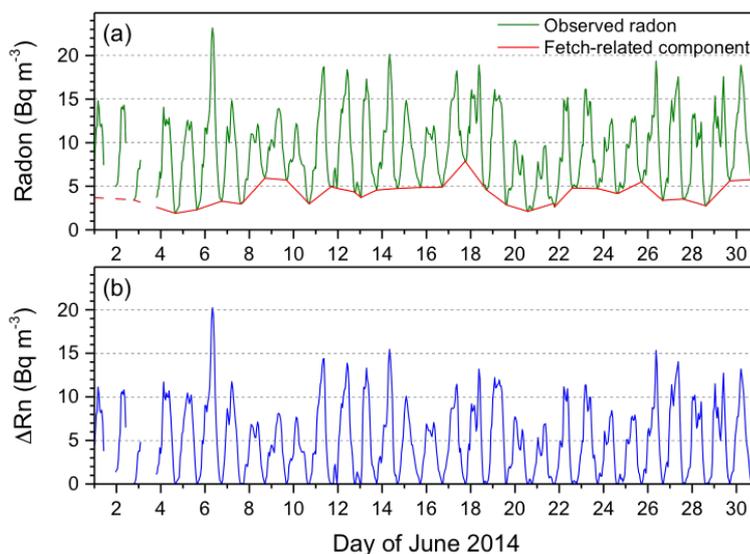
418 site show that air masses over the study period were predominately southeasterly
419 (Figures 1a and 4c), generally over the regions where fires were frequently detected.



420

421 **Figure 2** The number of hotspots observed each day within (a) 200 km, (b) 100 km, (c) 50 km, (d) 20 km and
422 (e) 10 km of the ATARS, as detected by the MODIS and VIIRS sensors on the Terra and Aqua satellites.

423 Afternoon radon concentrations provide further information regarding the regional air
424 mass fetch and the degree of contact with the land surface (red line, Figure 3a). Over
425 the campaign period, air masses with the least terrestrial fetch (low radon indicates
426 strongest oceanic signature) were observed on June 4-6 and 20-22, whereas June 8-9,
427 17-18 and 29-30 represented periods of particularly extensive continental fetch.



428

429 **Figure 3** Hourly ATARS radon observations for June 2014: (a) observed hourly data, and afternoon-to-
430 afternoon interpolated values (indicative of changes in the regional air mass fetch); and (b) difference in radon
431 concentration between the hourly observations and interpolated afternoon values (indicative of diurnal
432 variability).

433 A pronounced diurnal variability can clearly be seen in the ΔRn signal (Figure 3b).

434 Mean hourly diurnal composites of radon concentrations, wind speed, wind direction

435 and dew point temperature at the ATARS site during the period of the SAFIRE

436 campaign are shown in Figure 4. Following the technique described in Chambers et al.

437 (2016), these composites have been computed separately for three diurnal mixing

438 categories based on the mean ΔRn over the 12 hour period 2000-0800 h:

439 Strong mixing: $\Delta Rn_{12} < 5400 \text{ mBq m}^{-3}$

440 Moderate mixing: $5400 \leq \Delta Rn_{12} < 6700 \text{ mBq m}^{-3}$

441 Weak mixing: $\Delta Rn_{12} \geq 6700 \text{ mBq m}^{-3}$

442 The air masses predominantly originated from the southeast as indicated in Figure 1

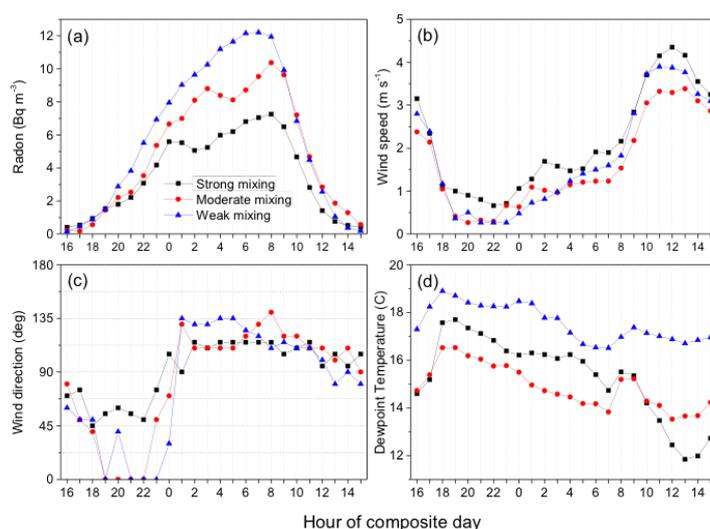
443 and Figure 4c. Starting from approximately 10:00 am each morning, however, sea

444 breeze circulations slowly turn the measured wind direction around from southeast to

445 northeast, before reverting back to the dominant wind direction again at around



446 midnight. Wind speeds reached a maximum just before midday and were at their lowest
447 just before midnight (Figure 4b). The “strong mixing” category was associated with
448 generally higher wind speeds, which cause increased mechanical turbulence leading to
449 deeper nocturnal mixing layers (i.e., hinder the development of a shallow nocturnal
450 inversion layer).



451

452 **Figure 4** Mean hourly diurnal composite (a) radon, (b) wind speed, (c) wind direction, and (d) dew point
453 temperature at ATARS, as a function of radon-based nocturnal mixing categories.

454 **Gas and Aerosol measurements**

455 The campaign average, standard deviation, median and Q25/Q75 values for the major
456 gaseous and aerosol species are shown in Table 1. The median values for each species
457 are likely to be representative of background concentrations in this region. The average
458 concentrations for most species were higher than the median concentrations, due to the
459 periods of close or intense fires. The extent of the influence of these close fires are
460 demonstrated by the maximum concentrations.



461 **Table 1** The campaign average, standard deviation, maximum, median, Q25 and Q75 values for key measured
 462 gas and aerosol species. All parts-per notation refer to mole fractions unless otherwise indicated.

Species (unit)	Average	Standard deviation	Maximum	Median	Q25	Q75
CO (ppb)	229	494	18900	130	87	214
CO ₂ (ppm)	404.68	11.539	513.578	402.454	394.728	411.299
O ₃ (ppbv)	24.616	9.903	99.784	22.771	17.896	29.778
CH ₄ (ppb)	1839.88	68.06	3766.81	1820.11	1802.26	1852.97
N ₂ O (ppb)	326.329	0.449	334.871	326.276	326.121	326.444
GEM (ng m ⁻³)	0.992	0.081	1.734	0.986	0.952	1.020
Acetonitrile (ppb)	0.351	0.629	9.775	0.197	0.129	0.337
Organics (ug m ⁻³)	11.081	22.385	347.657	4.160	2.335	13.279
SO ₄ ²⁻ (ug m ⁻³)	0.514	0.318	2.254	0.411	0.294	0.679
NH ₄ ⁺ (ug m ⁻³)	0.351	0.676	18.17	0.180	0.096	0.415
NO ₃ ⁻ (ug m ⁻³)	0.187	0.456	10.925	0.042	0.004	0.189
Cl ⁻ (ug m ⁻³)	0.166	1.271	53.270	0.029	0.016	0.076
PNC (cm ⁻³)	8182	19031	40300	2032	2032	8335
Mode diameter (nm)	104	31	-	102	85	122
Geom. SD	1.71	0.13	-	1.70	1.65	1.75

463
 464 In order to demonstrate the influence of close fires and the changing inversion layer,
 465 the time series of major greenhouse gases (CO, CO₂, CH₄ and N₂O), gaseous elemental
 466 mercury, acetonitrile and ozone throughout the campaign are shown in Figure 5. Sub-
 467 micron non-refractory aerosol organic, sulfate, ammonium and nitrate mass

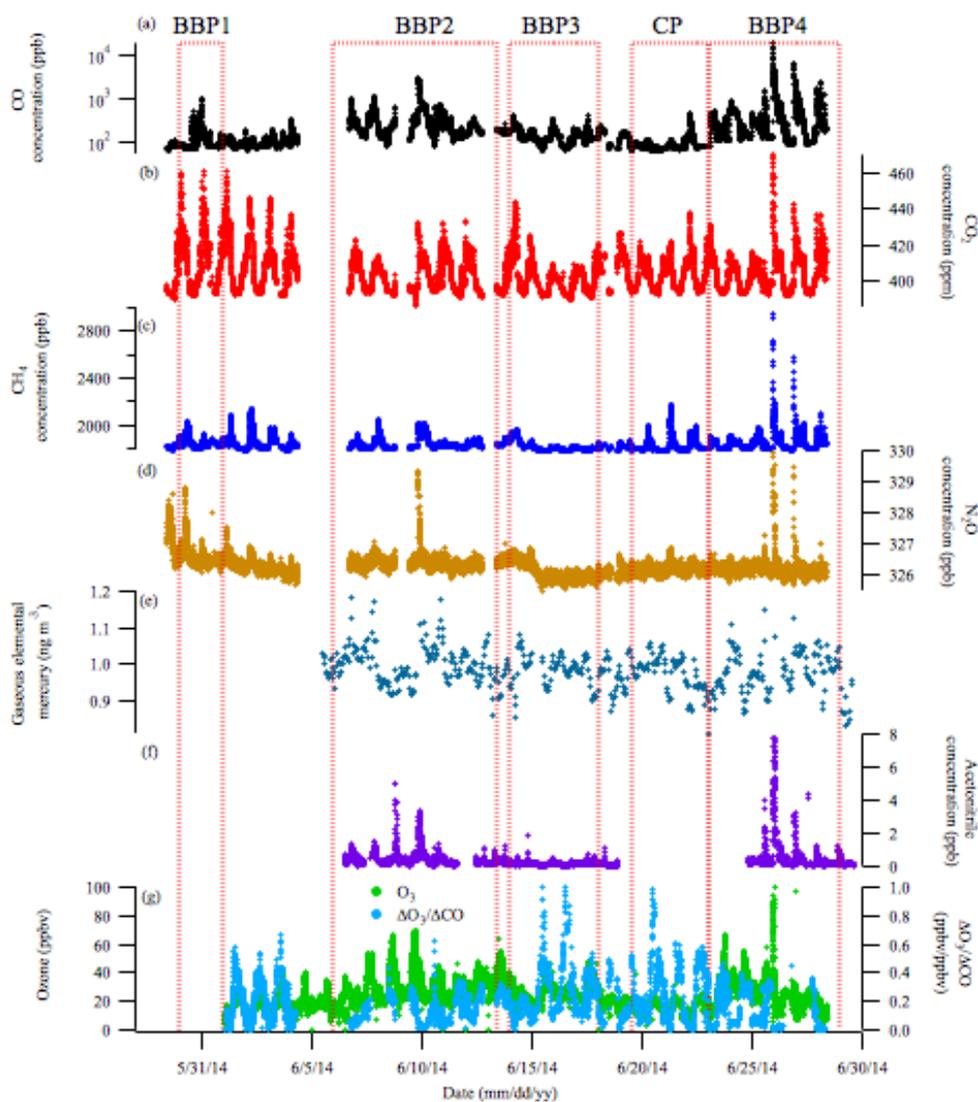


468 concentrations, organic mass fraction, PM₁ OC and EC mass concentrations and
469 particle size distributions for the sampling period are shown in Figure 6. Periods of
470 missing data correspond to times when instruments were not operating. Most of these
471 time series display a clear diurnal trend as a result of the varying inversion layer height.
472 Other enhancements in concentrations can be clearly seen and correspond to periods of
473 frequent close fires (Figure 2). Over the entire sampling period from the 29th of May
474 2014 until the 28th of June 2014, four biomass burning related periods (BBP) and a
475 "coastal" period (CP) have been distinguished. The dates for these periods are shown
476 in Table 2. These periods are also displayed in Figure 5 and Figure 6.

477 **Table 2** The start and end dates for the four identified Biomass Burning Periods (BBP1, BBP2, BBP3 and
478 BBP4) and the Coastal Period (CP).

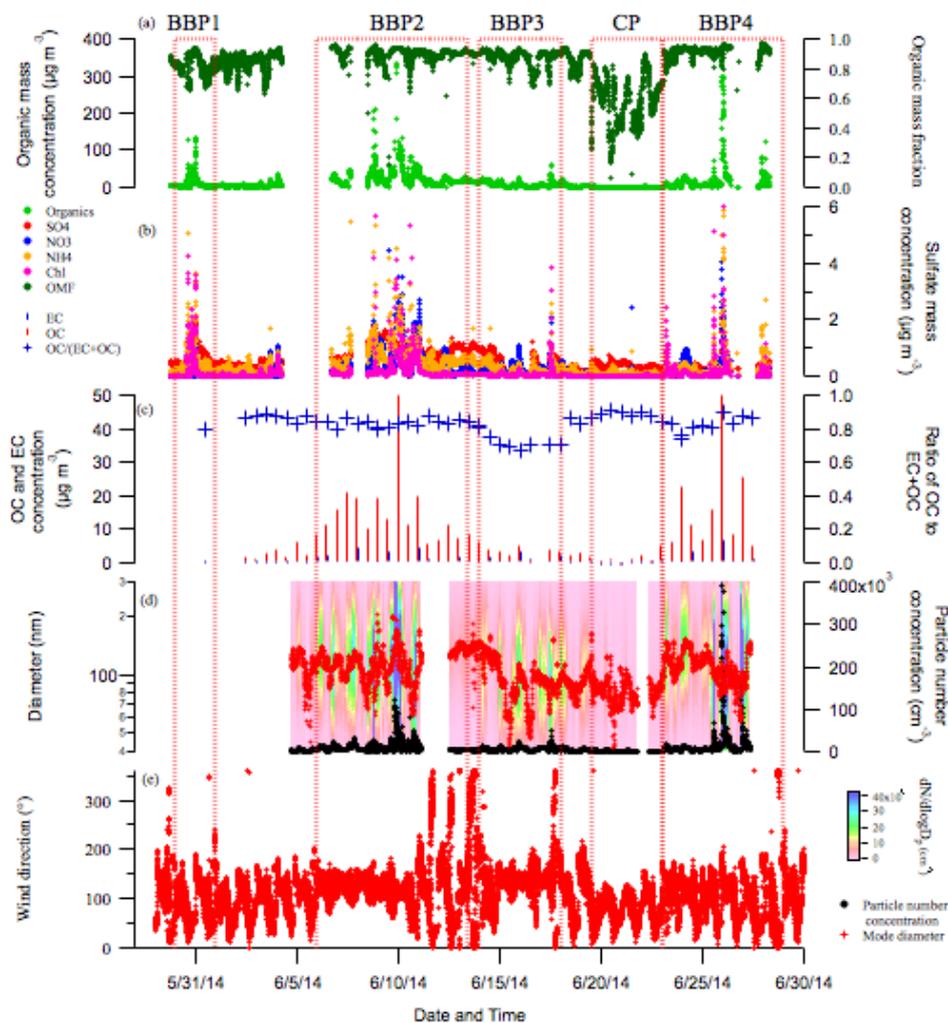
Period	Start date (mm/dd/yy hh:mm)	End date (mm/dd/yy hh:mm)
BBP1	05/30/14 00:00	05/31/14 23:59
BBP2	06/06/14 00:00	06/12/14 23:59
BBP3	06/14/14 00:00	06/17/14 23:59
CP	06/19/14 12:00	06/22/14 23:59
BBP3	06/23/14 00:00	06/28/14 23:59

479



480

481 Figure 5 The time series of the major measured gaseous species during the SAFIRED campaign: (a) carbon
482 monoxide, (b) carbon dioxide, (c) methane, (d) nitrogen dioxide, (e) gaseous elemental mercury, (f) acetonitrile
483 and (g) ozone and $\Delta O_3/\Delta CO$. The biomass burning and coastal periods are indicated by the red dotted lines.
484 All parts-per notation refer to mole fractions unless otherwise indicated.



485

486 Figure 6 The times series of the major aerosol properties during the SAFIRED campaign: (a) the non-
 487 refractory PM₁ organic mass concentration (left) and organic mass fraction (right), b) the inorganic non-
 488 refractory PM₁ mass concentrations, (c) the 12-hour filter OC and EC PM₁ mass concentrations (left) and the
 489 ratio of OC to OC+EC (right), (d) the particle size distributions and particle size mode (left) and the total
 490 particle number concentration (right) and (e) the wind direction at ATARS.

491 Over the campaign organics dominated the non-refractory sub-micron aerosol mass
 492 contributing, on average 90% (median; 86%) of the total mass. Sulphate, nitrates,
 493 ammonium and chloride species contributed the rest of this mass, with the largest
 494 contributions from sulphate and ammonium. Sulphate contributions were very
 495 significant during the coastal period, contributing up to 32% of the total mass. Although



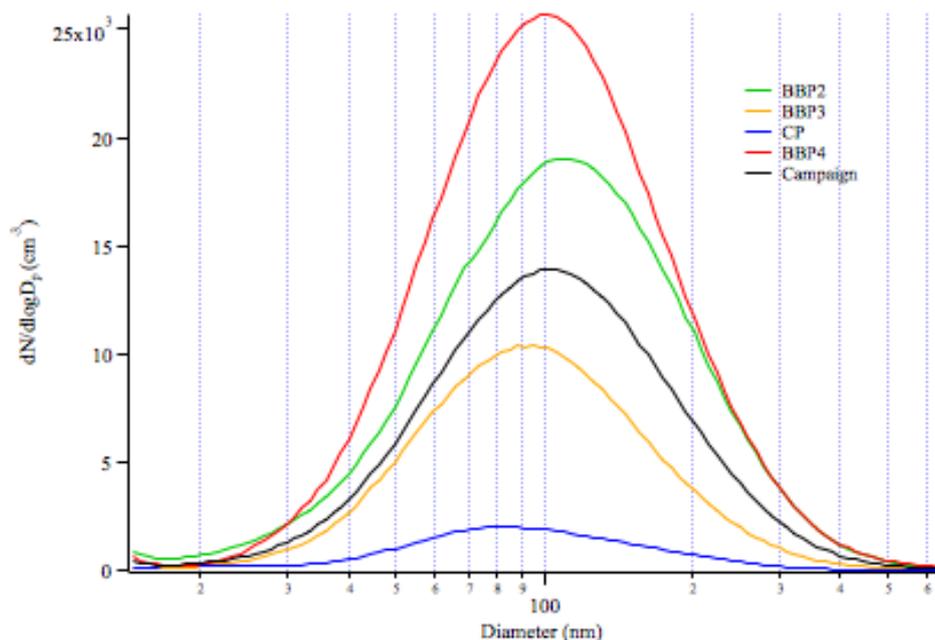
496 chlorides contributed the least to the total mass, on average, during clear biomass
497 burning events where sharp increases in CO and organics were observed, chlorides
498 made up the largest component of inorganic aerosol. The maximum chloride
499 concentration during the campaign reached $53 \mu\text{g m}^{-3}$. High soil and vegetation chloride
500 contents have been observed in savannah and coastal environments (Lobert et al.,
501 1999;Andreae et al., 1996). The strong elevations of chloride signals observed,
502 particularly during burning events BBP1, BBP2 and BBP4 are likely a result of
503 emission of these chloride ions. Outside of the "burning events" where very sharp
504 increases in concentrations were observed, chloride concentrations were very low. This
505 either suggests that these chloride species are short-lived, or only present in fires very
506 close to the coast and therefore the ATARS site.

507

508 Particle size distributions were unimodal for the majority of the sampling period with
509 a mode of approximately 100 nm on average (see Figures 6d and 7). The SMPS was
510 not operational during BBP1. Although the shape of the BBP4 size distribution was
511 similar to the campaign average, concentrations were much higher and a result of close
512 fires. BBP2 had a slightly larger size distribution centered on 110 nm. The size
513 distribution during BBP3 was slightly smaller than the campaign average and BBP2
514 and BBP4, with a mode centered on ~ 95 nm. The contrast between these size
515 distributions could be a result of atmospheric aging and dilution in which organic mass
516 condenses onto or evaporates from the particle. Variations in fuel load or burning
517 conditions could also contribute to this difference. The size and concentration of
518 particles during the Coastal Period (CP) were much smaller than the rest of the
519 campaign. There were two periods during CP where a bimodal size distribution was
520 observed; one from approximately 3 pm until midnight on the 19th of June and the other



521 between 2 pm and 6 pm on the 20th of June. As mentioned earlier, air masses during
522 this time had a stronger oceanic influence, with back trajectories in this period
523 originating along the east coast of Australia and passing over very little land before
524 arriving at the station. The size distributions for both of these periods had a mode at
525 approximately 20 nm and another at approximately 85 nm. Submicron sulfates made
526 up to 32% of the total submicron non-refractory mass concentrations, as reported by
527 the cToF-AMS from the period of midday on the 19th of June until midnight on the 22nd
528 of June, whereas the average sulfate contribution for the rest of the campaign was
529 approximately 8%. The low radon fetch, small particle concentrations, bimodal size
530 distributions and significant contributions of sulfate during this period suggest very
531 little biomass burning signal and a more marine-like aerosol. No particle nucleation
532 events were observed over the entire sampling period (See Supplementary Figure S2).
533 This is likely due to the elevated particle concentrations acting as a condensation sink.



534

535 **Figure 7** The average size distribution during BBP2, BBP3, BBP4, CP and the campaign average. The SMPS
536 was not operational during BBP1.



537 BBP1, BBP2 and BBP4 correspond to the periods when fires were burning within 10
538 km of ATARS. Large enhancements of biomass burning related emissions were
539 observed during these three periods. There were distinct enhancements of all measured
540 gaseous and aerosol species during these periods. Differences between the maximum
541 and background concentrations were very prominent for CO (note the logarithmic scale
542 in Figure 5a), CH₄, N₂O, acetonitrile (an established marker for biomass burning) and
543 organic, nitrate and chloride non-refractory sub-micron aerosol species. Similar
544 enhancements of CH₄ were also observed outside of these BB periods, which suggests
545 another source of methane in this region. Only slight enhancements of GEM
546 concentrations above background were observed during BBP2 and BBP4. Similar to
547 much of the rest of the campaign sampling period, the non-refractory submicron aerosol
548 was dominated by organics, with contributions typically varying between 70% and 95%
549 of the mass. Relative to background concentrations, there were also large enhancements
550 of nitrate and chloride species during these periods. While there were also
551 enhancements of sulfate and ammonium species during these periods, similar
552 enhancements were observed outside of these periods, again indicating a non-fire
553 source of these species. The ratio of O₃ to CO concentrations above background (taken
554 as 10 ppbv and 66 ppbv, respectively) gives an indication of the photochemical age of
555 a smoke plume. $\Delta\text{O}_3/\Delta\text{CO}$ were lowest during BBP2 and BBP4 (and not measured
556 during BBP1) relative to the rest of the campaign, indicating that the biomass burning
557 signals during these periods had not undergone extensive photochemical aging and are
558 therefore characteristic of fresh smoke.

559

560 Elevated signals during BBP1 were likely a result of a series of close fires within 5 km
561 ENE of ATARS. The VIIRS and MODIS sensors on the SUMO NPP, Terra and Aqua



562 satellites observes smaller fires at approximately 2 pm on the 30th of May. Winds were
563 northeasterly during these two events. It is therefore likely that these signals were
564 continuation or evolution of those fires. Burned vegetation was also visually observed
565 the next morning at these locations. The large burst event later on the evening of the
566 31st of May is unlikely to be associated with these fires as the wind direction during
567 this event was from the SSW and SSE. Large clusters of fires were observed at
568 approximately 100 km and 150 km SE of the station by the Terra and Aqua satellites.
569 The signals observed during this event could be a result of the plumes from this fire,
570 although the possibility of a fire ignited after the satellite flyovers, or a combination of
571 these cannot be eliminated.

572

573 Large signal enhancements on the 8th of June during BBP2 is likely a result of a cluster
574 of fires approximately 100 km south east of the station. The MODIS sensors on the
575 Terra and Aqua satellites observed the small cluster of fires along the back-trajectory
576 at 11:14 am and 1:56 pm. The source of BB emissions for the large event on the 9th of
577 June during this period is unclear. Several fires approximately 5 km from the station
578 along the back-trajectory were detected by the MODIS sensor on Aqua and the VIIRS
579 sensor on SUOMI NPP at approximately 2:30 pm on the 9th of June. There were also
580 numerous fires detected between 100 km and 200 km southeast along this trajectory.
581 The signals associated with this event could therefore be a result of the closer fires that
582 started to blaze later in the evening, the distant fires or a combination of both.

583

584 Only one fire within 20km of ATARS was observed during BBP3 on the 17th of June.
585 Numerous fires were observed further than 20km from the station and is possible that
586 the signals during this period were more aged. While photochemical aging and



587 coagulation typically lead to larger particles, particle size distributions were smaller
588 during this period and the ratio of OC to OC+EC was 70%, 10% lower than the ratio
589 during the rest of the campaign. Whether these observations were a result of burn
590 conditions or aging processes (i.e. evaporation of organic compounds from the aerosol
591 phase) is unclear, although the highest $\Delta O_3/\Delta CO$ values during the campaign were
592 observed during BBP3, which indicates photochemical aging was more extensive
593 during this period.

594

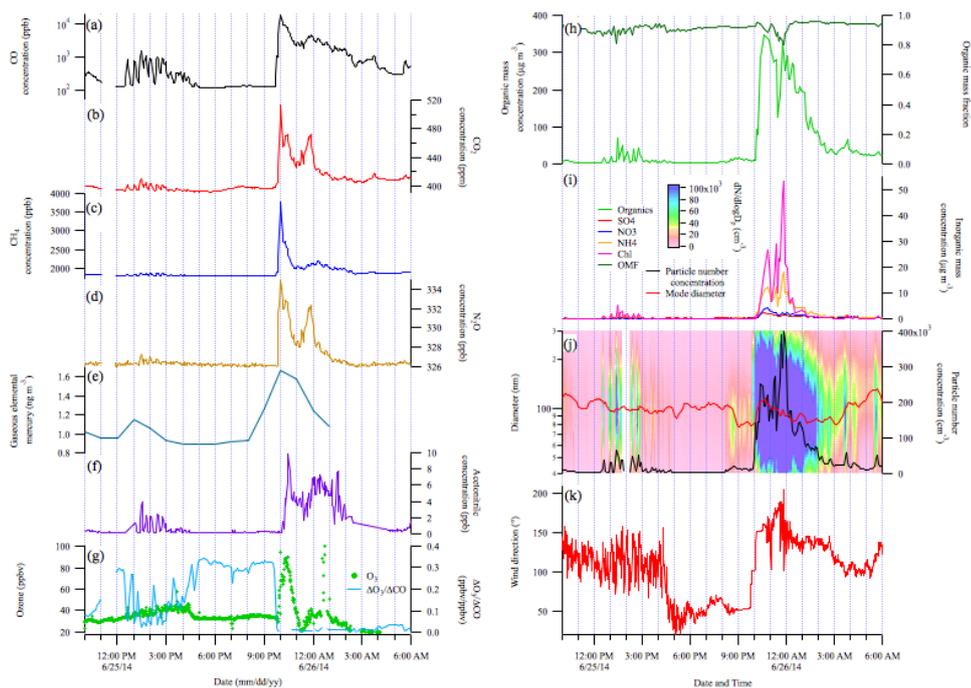
595 One close fire was also observed during CP, however wind directions during this period
596 were typically north-easterly and concentrations were therefore much lower. 5-day
597 HYSPLIT trajectories also show that air mass during the CP originated along the east
598 coast of Australia before travelling towards the sampling station with very little
599 terrestrial influence.

600 **Close proximity fires**

601 With numerous fires occurring across the region and the limitations of once-per-day
602 satellite fly-overs and stationary measurements, it can be difficult to identify the exact
603 source of these elevated signals. For a portion of BBP4, however, fires were burning
604 within several kilometers of ATARS and several plumes were easily observed from the
605 station. The signals from these plumes are shown in Figure 8. The observed
606 enhancements between 12:30 pm and 3:00 pm on the 25th June during BBP4 were a
607 result of grass fires burning approximately 1 km south-east from the station. During
608 this event, the wind direction (Figure 8k) was highly variable, changing between 140°
609 and 80° True Bearing (TB) multiple times. As a result, the sampling changed from
610 measuring the air mass with and without the plume from this fire, which led to sharp
611 increases and decreases in biomass burning-related signals (Figure 8a through 8j).



612 Visually, the fire area and extent of the plume was larger at 4:00pm than earlier,
613 however the wind direction changed to north-easterly which directed the plume away
614 from the station. From 4:00 pm until 10:00 pm, the wind direction was stable at
615 approximately 50° TB. At 10:00 pm, the wind direction rapidly changed to directly
616 south and the largest enhancements for the whole campaign were observed until
617 approximately 2:00 am on the 26th of June. It is very likely that these signals were a
618 result of a continuation and evolution of these fires as the night progressed. Portions of
619 a ~0.25 km² grassland field within 500 m directly south of ATARS were observed to
620 be burned upon arrival at the station on the morning of the 26th of June and we speculate
621 that the burning of this field contributed to the large enhancements in measured biomass
622 burning emissions. The emissions during this portion of BBP4 are likely to be the most
623 representative of fresh biomass burning smoke during the SAFIRED campaign.



624

625 **Figure 8** The major gas and aerosol concentrations measured during two biomass burning events within 1
626 km of ATARS during BBP4. (a) through (g) and (h) through (k) are as per Figures 5 and 6, respectively. All
627 parts-per notation or mole fractions unless otherwise indicated.

628 **4. Outcomes of SAFIRED**

629 The overall aim of this study was to investigate the characteristics of BB emissions in
630 the tropical savannah region of northern Australia during the early dry season. For many
631 gaseous and aerosol species, elevated signals were observed for much of the month-
632 long sampling period due to the high frequency of fires. The more specific outcomes of
633 SAFIRED are outlined below.

634 **Emission factors**

635 Australian fires are responsible for 6% of global CO₂ biomass burning emissions, most
636 of which is due to savannah fires (Shi et al., 2015). Carbon sequestering during
637 regrowth periods is considered to balance carbon emissions in tropical Australia



638 (Haverd et al., 2013). Greenhouse gases emitted from savannah fires that are not
639 sequestered, such as methane (CH_4) and nitrous oxide (N_2O), have been shown to
640 contribute 2-4% of the annual accountable greenhouse gas emissions from Australia
641 (Meyer et al., 2012). Seasonal emission factors for the major greenhouse gases are
642 important for national greenhouse gas inventories and in understanding the impact of
643 savannah fires. Furthermore, emission factors of CO_2 and CO can be used to infer
644 mechanisms behind the emissions of other species, such as the connection between
645 particulate matter and burning conditions.

646

647 The gaseous and aerosol data for the sample period were investigated to identify BB
648 events and determine the emission factors of CO_2 , CO, CH_4 , N_2O , as well as Aitken
649 and Accumulation mode aerosols and submicron particle species (organics, sulfates,
650 nitrates, ammonium and chlorides) for several individual BB events. These emission
651 factors were mostly found to be dependent on the combustion conditions (using the
652 modified combustion efficiency as a proxy) of the fires. These results will be the first
653 set of emission factors for aerosol particles from savannah fires in Australia.
654 Furthermore, the variability in emission factors for different fires calls for a separation
655 of single-value emission factors that are usually reported for savannah fires into grass
656 and shrub components. A full discussion of these results are presented in Desservettaz
657 et al. (2016, submitted).

658 **Non-methane organic compounds (NMOCs)**

659 Biomass burning is the second largest source of NMOCs globally with a recent global
660 estimate of at least 400 Tg year^{-1} , second only to biogenic sources (Akagi et al., 2011).
661 Biomass burning produces a complex mix of NMOCs, which may be saturated or
662 unsaturated, aliphatic or aromatic, and contain substitutions of oxygen, sulfur, nitrogen,



663 halogens and other atoms. NMOC emission rates are strongly tied to the efficiency of
664 combustion, with smouldering fires emitting NMOC at higher rates than flaming fires
665 (Andreae and Merlet, 2001). Biomass burning derived NMOCs fuel the production of
666 tropospheric ozone in diluted, aged biomass burning plumes, with higher ozone
667 enhancements observed when biomass burning plumes interact with NO_x-rich urban
668 plumes (Jaffe and Wigder, 2012; Wigder et al., 2013; Akagi et al., 2013). Oxidation of
669 NMOCs results in lower volatility products that partition to the aerosol phase and
670 contribute significantly to secondary organic aerosol (Hallquist et al., 2009). Biomass
671 burning produces significant amounts of semi-volatile NMOC which can be difficult to
672 quantify and identify with current measurement techniques. However recent studies
673 have shown that including semi volatile NMOC chemistry in models improves the
674 agreement between the modeled and observed organic aerosol (Alvarado et al., 2015;
675 Konovalov et al., 2015) and ozone (Alvarado et al., 2015). High quality NMOC
676 emission factors are crucial for models to assess the impact of biomass burning plumes
677 on air quality and climate.

678 **PAHs**

679 Polycyclic aromatic hydrocarbons (PAHs) are a group of chemicals that are formed and
680 emitted during combustion processes. Globally, major sources include
681 residential/commercial biomass burning, open-field biomass burning and vehicular
682 emissions (Shen et al., 2013). In Oceania in 2007, 31% of PAH emissions were
683 estimated to be attributed to deforestation and wildfires (Shen et al., 2013). With control
684 strategies targeting and reducing vehicular emission of PAHs over the last few decades,
685 the relative contribution of other emission sources, such as savannah fires, has increased
686 (Friedman et al., 2013; Kallenborn et al., 2012; Wang et al., 2016). Although most of
687 these emissions are in the gas-phase (Jenkins et al., 1996; Atkins et al., 2010), the



688 particle-phase PAHs, such as benzo[a]pyrene (BaP), may have high genotoxicity
689 (IARC., 2015). However, field-based studies on emissions of PAHs from open-field
690 biomass burning, including savannah fires remain limited in Australia (Freeman and
691 Cattell, 1990).

692

693 Emission factors of PAHs from biomass burning related to savannah fires in northern
694 Australia will be estimated from the data collected during this campaign. This
695 estimation will be based on the (background subtracted) concentrations of PAHs and
696 CO₂ (and CO) during the events where biomass burning contributes most to these
697 concentrations measured at the sampling site. The concentrations of 13 major PAHs
698 (gaseous plus particle-associated phase) varied from ~ 1 to over 15 ng m⁻³ within
699 different BB events. In the gas phase, 3- and 4-ring compounds typically contributed ~
700 90% to the sum concentrations whereas the particle-associated PAHs were dominated
701 by 5- and 6-ring compounds (> 80%). Measured PAH concentrations were significantly
702 higher (paired *t*-test, *P* < <0.05) during BB events E, F and G. For these events,
703 concentrations of BaP exceeded the monitoring investigation level for atmospheric BaP
704 in Australia (National-Environment-Protection-Council-Service-Corporation, 2011),
705 i.e. 0.30 ng m⁻³, by 66% (BB event E) and 200% (BB events F and G). A full discussion
706 of these results can be found in (Wang et al., 2016, under review).

707 **Mercury**

708 The atmosphere is the dominant transport pathway for mercury globally, with emissions
709 to the atmosphere from both natural and anthropogenic origins (Driscoll et al., 2013).
710 Whilst our understanding of the natural cycling of mercury has improved markedly over
711 the past decades (Pirrone et al., 2010), large uncertainties still exist; specifically, global
712 emission estimates to the atmosphere from biomass burning currently range between



713 300 and 600 Mg year⁻¹ (Driscoll et al., 2013). In the atmosphere, mercury exists as one
714 of three operationally-defined species: gaseous elemental mercury (GEM), gaseous
715 oxidised mercury (GOM) and particulate-bound mercury (PBM), each with differing
716 abundances, solubility and depositional characteristics and with in-air conversion
717 between all three species possible (Lin and Pehkonen, 1999). Mercury can be
718 scavenged from the atmosphere through both wet and dry depositional processes, and
719 the monsoonal climate of northern Australia results in varying significance of each of
720 these processes through the year (Packham et al., 2009). Upon deposition, mercury may
721 be stored in plant tissue via stomatal or cuticular uptake (Rea et al., 2002) or sequestered
722 within soils (Gustin et al., 2008). Release from both of these pools is achieved from
723 burning events that may volatilise or thermally desorb mercury from biomass and soil,
724 respectively (Melendez-Perez et al., 2014). Subsequently this mercury pool is
725 redistributed through the atmospheric pathway to ecosystems that may methylate
726 mercury, thereby enhancing its bioavailability to the local food chain.

727

728 SAFIRED represents the first measurements of atmospheric mercury undertaken in the
729 tropical region of the Australian continent. The mean observed GEM concentration
730 over the study period was 0.99 ± 0.09 ng m⁻³, similar to the average over that month
731 (0.96 ng m⁻³) for 5 other Southern Hemisphere sites and slightly lower than the average
732 (1.15 ng m⁻³) for 5 tropical sites (Sprovieri et al., 2016). Mean GOM and PBM
733 concentrations were 11 ± 5 pg m⁻³ and 6 ± 3 pg m⁻³ respectively, representing 0.6 –
734 3.4% of total observed atmospheric mercury.

735

736 Atmospheric mercury measurements were available only during the final four identified
737 burn events. During these events, spikes in GEM concentrations were observed, though



738 there were no significant increases in GOM or PBM. Emission ratios calculated during
739 the campaign were two orders of magnitude higher than those reported by Andreae and
740 Merlet (2001), though those were from scrub, rather than grass, BB events
741 (Desservettaz et al., 2016). Future outcomes from the SAFIRED campaign will focus
742 on the use of micrometeorological techniques and the passive tracer radon to quantify
743 delivery of atmospheric mercury to tropical savannah ecosystems. ATARS also now
744 serves as an additional site measuring continuous GEM as part of the Global Mercury
745 Observation System (GMOS), one of only two tropical observing sites in the Eastern
746 Hemisphere and the third such site located in Australia.

747 **Aging of aerosols**

748 Atmospheric chemistry and radiative forcing will depend on how gaseous and aerosol
749 emissions from fires age as they move and interact with each other and existing species
750 in the atmosphere. Biomass burning aerosols can be involved in condensation and
751 coagulation (Radhi et al., 2012), undergo water uptake (Mochida and Kawamura, 2004)
752 form cloud droplets (Novakov and Corrigan, 1996), and be exposed to photochemical
753 aging processes, including those involving the gaseous components of fire emissions
754 (Keywood et al., 2011; Keywood et al., 2015). With a reported lifetime of 3.8 ± 0.8
755 days (Edwards et al., 2006), biomass burning aerosols are able to travel intercontinental
756 distances (Rosen et al., 2000) and are therefore present in the atmosphere long enough
757 for substantial changes due to aging. Furthermore, tropical convection is likely to affect
758 the aging of BB emissions in the region around ATARS, due to the immediate
759 proximity to the warm waters in the Timor Sea (Allen et al., 2008). This introduces
760 further uncertainty to the effect of BB emissions on radiation flux.

761



762 Primary organic aerosol directly emitted from biomass burning can interact with
763 NMOCs to change composition and mass, resulting in secondary organic aerosol
764 (Hallquist et al., 2009). Photochemical oxidation of NMOCs occurs during the daytime
765 by either hydroxyl radicals or ozone. Ozone is also typically produced in the aging
766 processes of tropical biomass burning plumes when NMOCs can oxidise to produce
767 peroxy radicals that react with NO. Photochemical reactions also may lead to an overall
768 increase in total aerosol mass through the condensation of NMOCs onto existing
769 particles (Reid et al., 1998; Yokelson et al., 2009; Akagi et al., 2012; DeCarlo et al.,
770 2008). Some studies have shown the opposite, i.e., photo-oxidation can also lead to the
771 evaporation of some primary organic constituents, resulting in an overall mass
772 reduction (Hennigan et al., 2011; Akagi et al., 2012). With thousands of organic
773 compounds in the atmosphere, each with different volatilities and potential reaction
774 mechanisms, our understanding of secondary organic aerosol production is limited
775 (Goldstein and Galbally, 2007; Keywood et al., 2011). Furthermore, secondary organic
776 aerosol can also form through aqueous phase reactions where water-soluble organics
777 dissolve into water on existing particles (Lim et al., 2010).

778

779 Further analysis into the aerosol chemical composition will elucidate the aging of early
780 dry season biomass burning emissions. Fractional analysis (e.g., f₄₄ and f₆₀, the
781 fraction of m/z 44 and m/z 60 to all organic masses, indicated oxygenation and BB
782 sources, respectively) and factor analysis using positive matrix factorisation (PMF) of
783 cToF-AMS data has been investigated over the entire sampling period. Outside of the
784 periods of significant influence from BB events, three PMF-resolved organic aerosol
785 factors were identified. A BB organic aerosol factor was found to comprise 24% of the
786 submicron non-refractory organic mass, with an oxygenated organic aerosol factor and



787 a biogenic isoprene-related secondary organic aerosol factor comprising 47% and 29%,
788 respectively. These results indicate the significant influence of fresh and aged BB on
789 aerosol composition in the early dry season. The emission of precursors from fires is
790 likely responsible for some of the SOA formation. A full discussion of these results can
791 be found in Milic et al. (2016). Future analysis will investigate the gas and particle-
792 phase composition for individual BB events.

793 **Water uptake of aerosols**

794 The water uptake by aerosols is determined by their size and composition, as well as
795 the atmospheric humidity (McFiggans et al., 2006). The hygroscopic properties of all
796 of the different components of an aerosol particle contribute to its total hygroscopicity
797 (Chen et al., 1973; Stokes and Robinson, 1966). The presence of different water-soluble
798 and water-insoluble organics and inorganics will therefore strongly influence water
799 uptake. Furthermore, chamber studies that have investigated emissions from biomass
800 fuels, both separately and in combination, have shown that the hygroscopic response
801 can vary significantly depending on fuel type (Carrico et al., 2010). Understanding the
802 water uptake of atmospheric aerosols is further complicated when considering other
803 aging processes as described previously. Nonetheless, it is important to characterise the
804 water uptake, as this will, in turn, influence other atmospheric chemistry processes,
805 radiation scattering and absorption as well as cloud processing.

806

807 Biomass burning aerosols can act as cloud condensation nuclei if they are large enough
808 for water to easily condense onto their surface, or if the particles have a large affinity
809 for water due to their composition (Novakov and Corrigan, 1996). Ultimately, this
810 means that BB emissions can lead to a higher number of cloud droplets. This is
811 important in reflecting solar radiation and cooling the earth's surface. Cloud albedo is



812 more susceptible to changes when cloud condensation nuclei concentrations are
813 relatively low (Twomey, 1991), such as in marine environments like the Timor Sea off
814 the coast of northern Australia.

815

816 The water uptake of aerosols has been further investigated to identify the possible
817 influence of early dry season BB in this region on cloud formation. The concentrations
818 of cloud condensation nuclei at a constant supersaturation of 0.5% were typically of the
819 order of 2000 cm^{-3} and reached well over 10000 cm^{-3} during intense BB events.
820 Variations in the ratio of aerosol particles activating cloud droplets showed a distinct
821 diurnal trend, with an activation ratio of $40\% \pm 20\%$ during the night and $60\% \pm 20\%$
822 during the day. The particle size distribution and the hygroscopicity of the particles
823 were found to significantly influence this activation ratio. A full discussion of these
824 results can be found in Mallet et al. (2016, submitted). Future analysis will elucidate
825 the contribution of different biomass burning aerosol components on the
826 hygroscopicity.

827 **Trace metal deposition**

828 The deposition and dissolution of aerosols containing trace metals into the ocean may
829 provide important micronutrients required for marine primary production. Conversely,
830 the deposition of soluble iron can trigger toxic algal blooms, such as *Trichodesmium*,
831 in nutrient-poor tropical and subtropical waters (LaRoche and Breitbarth, 2005).
832 *Trichodesmium* blooms require large quantities of soluble iron, of which aerosols are a
833 source (Boyd and Ellwood, 2010; Rubin et al., 2011). To date, most studies have
834 assumed that mineral dust aerosols represent the primary source of soluble iron in the
835 atmosphere (Baker and Croot, 2010); however fire emissions and oil combustion are
836 other likely sources (Ito, 2011; Schroth et al., 2009; Sedwick et al., 2007). A few studies



837 have shown that iron contained in biomass burning emissions is significantly more
838 soluble than mineral dust (Guieu et al., 2005;Luo et al., 2008;Schroth et al., 2009) but,
839 to date, no data exists for Australian fires.

840

841 The aim of the trace metal aerosol component of SAFIRED is to quantify, for the first
842 time, the fractional solubility of aerosol iron, and other trace metals, derived from
843 Australian dry season BB. The fractional iron solubility is an important variable
844 determining iron availability for biological uptake. On a global scale, the large
845 variability in the observed fractional iron solubility results, in part, from a mixture of
846 different aerosol sources. Estimates of fractional iron solubility from fire combustion
847 (1 - 60 %) are thought to be greater than those originating from mineral dust (1 - 2%)
848 (Chuang et al., 2005;Guieu et al., 2005;Sedwick et al., 2007), and may vary in
849 relationship to biomass and fire characteristics as well as that of the underlying terrain
850 (Paris et al., 2010;Ito, 2011). Iron associated with BB may provide information with
851 respect to BB inputs of iron to the ocean (Giglio et al., 2013;e.g. Meyer et al., 2008).
852 The ATARS provides an ideal location to further investigate BB derived fractional iron
853 solubility at the source. The results from this study can be found in Winton et al. (2016)
854 and show that soluble iron concentrations from BB sources are significantly higher than
855 those observed in Southern Ocean baseline air masses from the Cape Grim Baseline
856 Air Pollution Station, Tasmania, Australia (Winton et al., 2015). Aerosol iron at
857 SAFIRED was a mixture of fresh BB, mineral dust, sea spray and industrial pollution
858 sources. The fractional iron solubility (2 - 12%) was relatively high throughout the
859 campaign and the variability was related to the mixing and enhancement of mineral
860 dust iron solubility with BB species.



861 **5. Looking forward**

862 While the specific outcomes of the SAFIRED campaign are reviewed above, the
863 general importance of this study can be discussed in a greater context. This is the first
864 large-scale collaborative project undertaken in this region and draws on the resources
865 and expertise of most of Australia's research institutes focused on atmosphere
866 chemistry and composition. Large scale, multidisciplinary measurement campaigns in
867 the tropics, such as SAFIRED, are needed to make distinctions between different types
868 of fires in different regions to reduce uncertainties in global climate models (Keywood
869 et al., 2013). This need has been recognized with the formation of global collaborative
870 initiatives promoting interdisciplinary collaboration in biomass burning research
871 (Kaiser and Keywood, 2015). As the world moves towards a warmer climate, biomass
872 burning is likely to increase in frequency and intensity, and these emissions will become
873 an increasingly important source of trace gases and aerosols to the atmosphere.
874 SAFIRED lays the foundation for future measurements at ATARS that could make
875 measurements throughout the whole dry season and on a more long-term scale. Future
876 work in this region should focus on 1) the detailed characterisation of individual fires
877 and their emissions, 2) biomass burning emissions throughout the late dry season and
878 3) the vertical and horizontal transport of biomass burning emissions in this region.

879 **Data availability**

880 All data are available upon request from the corresponding authors (Branka Miljevic,
881 b.miljevic@qut.edu.au; Melita D. Keywood; melita.keywood@csiro.au).



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892 b: Contributed to the organisation of the campaign

893 c: Installed and/or operated instrumentation during the sampling period

894 d: Analysed data

895 e: Contributed to the manuscript and/or data interpretation

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898 Competing interests

899 The authors declare that they have no conflict of interest.

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906 6. References

- 907 Akagi, S., Yokelson, R. J., Wiedinmyer, C., Alvarado, M., Reid, J., Karl, T., Crouse,
908 J., and Wennberg, P.: Emission factors for open and domestic biomass burning for use
909 in atmospheric models, *Atmospheric Chemistry and Physics*, 11, 4039-4072, 2011.
- 910 Akagi, S., Yokelson, R. J., Burling, I. R., Meinardi, S., Simpson, I., Blake, D. R.,
911 McMeeking, G., Sullivan, A., Lee, T., and Kreidenweis, S.: Measurements of reactive
912 trace gases and variable O₃ formation rates in some South Carolina biomass burning
913 plumes, *Atmospheric Chemistry and Physics*, 13, 1141-1165, 2013.
- 914 Akagi, S. K., Craven, J. S., Taylor, J. W., McMeeking, G. R., Yokelson, R. J., Burling,
915 I. R., Urbanski, S. P., Wold, C. E., Seinfeld, J. H., Coe, H., Alvarado, M. J., and Weise,
916 D. R.: Evolution of trace gases and particles emitted by a chaparral fire in California,
917 *Atmospheric Chemistry and Physics*, 12, 1397-1421, 2012.
- 918 Allen, G., Vaughan, G., Bower, K., Williams, P., Crosier, J., Flynn, M., Connolly, P.,
919 Hamilton, J., Lee, J., and Saxton, J.: Aerosol and trace-gas measurements in the Darwin
920 area during the wet season, *Journal of Geophysical Research: Atmospheres* (1984–
921 2012), 113, 2008.
- 922 Andersen, A. N., Cook, G. D., Corbett, L. K., Douglas, M. M., Eager, R. W., Russell-
923 Smith, J., Setterfield, S. A., Williams, R. J., and Woinarski, J. C.: Fire frequency and
924 biodiversity conservation in Australian tropical savannas: implications from the
925 Kapalga fire experiment, *Austral Ecology*, 30, 155-167, 2005.
- 926 Andreae, M., Atlas, E., Harris, G., Helas, G., DeKock, A., Koppmann, R., Manó, S.,
927 Pollock, W., Rudolph, J., and Scharffe, D.: Methyl halide emissions from savanna fires
928 in southern Africa, 1996.
- 929 Andreae, M. O., and Merlet, P.: Emission of trace gases and aerosols from biomass
930 burning, *Global Biogeochemical Cycles*, 15, 955-966, 2001.
- 931 Atkins, A., Bignal, K. L., Zhou, J. L., and Cazier, F.: Profiles of polycyclic aromatic
932 hydrocarbons and polychlorinated biphenyls from the combustion of biomass pellets,
933 *Chemosphere*, 78, 1385-1392, 2010.
- 934 Baker, A., and Croot, P.: Atmospheric and marine controls on aerosol iron solubility in
935 seawater, *Marine Chemistry*, 120, 4-13, 2010.
- 936 Bindoff, N. L., Stott, P. A., AchutaRao, K. M., Allen, M. R., Gillett, N., Gutzler, D.,
937 Hansingo, K., Hegerl, G., Hu, Y., Jain, S., Mokhov, I. I., Overland, J., Perlwitz, J.,
938 Sebbari, R., and Zhang, X.: Detection and Attribution of Climate Change: from Global
939 to Regional, in: *Climate Change 2013: The Physical Science Basis. Contribution of*
940 *Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on*
941 *Climate Change*, edited by: Stocker, T. F., Qin, D., Plattner, G.-K., Tignor, M., Allen,
942 S. K., Boschung, J., Nauels, A., Xia, Y., Bex, V., and Midgley, P. M., Cambridge
943 University Press, Cambridge, United Kingdom and New York, NY, USA, 867–952,
944 2013.
- 945 Biraud, S., Ciais, P., Ramonet, M., Simmonds, P., Kazan, V., Monfray, P., O'Doherty,
946 S., Spain, T. G., and Jennings, S. G.: European greenhouse gas emissions estimated



- 947 from continuous atmospheric measurements and radon 222 at Mace Head, Ireland,
948 *Journal of Geophysical Research: Atmospheres* (1984–2012), 105, 1351-1366, 2000.
- 949 Boyd, P., and Ellwood, M.: The biogeochemical cycle of iron in the ocean, *Nature*
950 *Geoscience*, 3, 675-682, 2010.
- 951 Carrico, C., Petters, M., Kreidenweis, S., Sullivan, A., McMeeking, G., Levin, E.,
952 Engling, G., Malm, W., and Collett Jr, J.: Water uptake and chemical composition of
953 fresh aerosols generated in open burning of biomass, *Atmospheric Chemistry and*
954 *Physics*, 10, 5165-5178, 2010.
- 955 Chambers, S., Williams, A., Crawford, J., and Griffiths, A.: On the use of radon for
956 quantifying the effects of atmospheric stability on urban emissions, *Atmospheric*
957 *Chemistry and Physics*, 15, 1175-1190, 2015.
- 958 Chambers, S. D., Hong, S.-B., Williams, A. G., Crawford, J., Griffiths, A. D., and Park,
959 S.-J.: Characterising terrestrial influences on Antarctic air masses using Radon-222
960 measurements at King George Island, *Atmospheric Chemistry and Physics*, 14, 9903-
961 9916, 2014.
- 962 Chambers, S. D., Williams, A. G., Conen, F., Griffiths, A. D., Reimann, S., Steinbacher,
963 M., Krummel, P. B., Steele, L. P., van der Schoot, M. V., and Galbally, I. E.: Towards
964 a Universal “Baseline” Characterisation of Air Masses for High-and Low-Altitude
965 Observing Stations Using Radon-222, *Aerosol Air Quality Research*, 16, 885-899,
966 2016.
- 967 Chen, H., Sangster, J., Teng, T., and Lenzi, F.: A general method of predicting the water
968 activity of ternary aqueous solutions from binary data, *The Canadian Journal of*
969 *Chemical Engineering*, 51, 234-241, 1973.
- 970 Chow, J. C., Watson, J. G., Chen, L. W. A., Chang, M. C. O., Robinson, N. F., Trimble,
971 D., and Kohl, S.: The IMPROVE-A temperature protocol for thermal/optical carbon
972 analysis: maintaining consistency with a long-term database, *Journal of the Air &*
973 *Waste Management Association*, 57, 1014-1023, 2007a.
- 974 Chow, J. C., Watson, J. G., Chen, L. W. A., Chang, M. O., Robinson, N. F., Trimble,
975 D., and Kohl, S.: The IMPROVE-A temperature protocol for thermal/optical carbon
976 analysis: maintaining consistency with a long-term database, *Journal of the Air &*
977 *Waste Management Association*, 57, 1014-1023, 2007b.
- 978 Chuang, P. Y., Duvall, R. M., Shafer, M. M., and Schauer, J. J.: The origin of water
979 soluble particulate iron in the Asian atmospheric outflow, *Geophysical Research*
980 *Letters*, 32, L07813, 2005.
- 981 Crutzen, P. J., and Andreae, M. O.: Biomass burning in the tropics: Impact on
982 atmospheric chemistry and biogeochemical cycles, *Science*, 250, 1669-1678, 1990.
- 983 DeCarlo, P. F., Dunlea, E. J., Kimmel, J. R., Aiken, A. C., Sueper, D., Crouse, J.,
984 Wennberg, P. O., Emmons, L., Shinozuka, Y., Clarke, A., Zhou, J., Tomlinson, J.,
985 Collins, D. R., Knapp, D., Weinheimer, A. J., Montzka, D. D., Campos, T., and
986 Jimenez, J. L.: Fast airborne aerosol size and chemistry measurements above Mexico
987 City and Central Mexico during the MILAGRO campaign, *Atmospheric Chemistry and*
988 *Physics*, 8, 4027-4048, 2008.
- 989 Desservettaz, M. J., Paton-Walsh, C., Griffith, D. W. T., Kettlewell, G., Keywood, M.
990 D., van der Schoot, M. V., Ward, J., Mallet, M. D., Milic, A., Miljevic, B., Ristovski,
991 Z. D., Howard, D., Edwards, G. C., and Atkinson, B.: Emission factors of trace gases
992 and particulates during the dry season in tropical northern Australia, *Journal of*
993 *Geophysical Research: Atmospheres*, 2016, submitted.
- 994 Draxler, R. R., and Rolph, G.: HYSPLIT (HYbrid Single-Particle Lagrangian
995 Integrated Trajectory) model access via NOAA ARL READY website (<http://www/>.



- 996 arl.noaa.gov/ready/hysplit4.html). NOAA Air Resources Laboratory, Silver Spring,
997 in, Md, 2003.
- 998 Driscoll, C. T., Mason, R. P., Chan, H. M., Jacob, D. J., and Pirrone, N.: Mercury as a
999 global pollutant: sources, pathways, and effects, *Environmental science & technology*,
1000 47, 4967-4983, 2013.
- 1001 Edwards, D., Emmons, L., Gille, J., Chu, A., Attié, J. L., Giglio, L., Wood, S.,
1002 Haywood, J., Deeter, M., and Massie, S.: Satellite-observed pollution from Southern
1003 Hemisphere biomass burning, *Journal of Geophysical Research: Atmospheres* (1984–
1004 2012), 111, 2006.
- 1005 Edwards, G. C., Rasmussen, P. E., Schroeder, W. H., Wallace, D. M., Halfpenny-
1006 Mitchell, L., Dias, G. M., Kemp, R. J., and Ausma, S.: Development and evaluation of
1007 a sampling system to determine gaseous Mercury fluxes using an aerodynamic
1008 micrometeorological gradient method, *Journal of Geophysical Research: Atmospheres*,
1009 110, 2005.
- 1010 Freeman, D. J., and Cattell, F. C. R.: Woodburning as a source of atmospheric
1011 polycyclic aromatic hydrocarbons, *Environmental Science and Technology*, 24, 1581-
1012 1585, 1990.
- 1013 Friedman, C. L., Zhang, Y., and Selin, N. E.: Climate change and emissions impacts on
1014 atmospheric PAH transport to the Arctic, *Environmental Science & Technology*, 48,
1015 429-437, 2013.
- 1016 Giglio, L., Randerson, J. T., and Werf, G. R.: Analysis of daily, monthly, and annual
1017 burned area using the fourth-generation global fire emissions database (GFED4),
1018 *Journal of Geophysical Research: Biogeosciences*, 118, 317-328, 2013.
- 1019 Goldstein, A. H., and Galbally, I. E.: Known and unexplored organic constituents in the
1020 earth's atmosphere, *Environmental Science & Technology*, 41, 1514-1521, 2007.
- 1021 Govender, N., Trollope, W. S., and Van Wilgen, B. W.: The effect of fire season, fire
1022 frequency, rainfall and management on fire intensity in savanna vegetation in South
1023 Africa, *Journal of Applied Ecology*, 43, 748-758, 2006.
- 1024 Griffith, D. W. T.: Synthetic calibration and quantitative analysis of gas-phase FT-IR
1025 spectra, *Applied Spectroscopy*, 50, 59-70, 1996.
- 1026 Griffith, D. W. T., Deutscher, N. M., Caldow, C., Kettlewell, G., Riggenbach, M., and
1027 Hammer, S.: A Fourier transform infrared trace gas and isotope analyser for
1028 atmospheric applications, *Atmos. Meas. Tech.*, 5, 2481-2498, 2012.
- 1029 Guieu, C., Bonnet, S., Wagener, T., and Loye-Pilot, M. D.: Biomass burning as a source
1030 of dissolved iron to the open ocean?, *Geophysical Research Letters*, 32, 2005.
- 1031 Gustin, M. S., Lindberg, S. E., and Weisberg, P. J.: An update on the natural sources
1032 and sinks of atmospheric mercury, *Applied Geochemistry*, 23, 482-493, 2008.
- 1033 Hallquist, M., Wenger, J. C., Baltensperger, U., Rudich, Y., Simpson, D., Claeys, M.,
1034 Dommen, J., Donahue, N. M., George, C., Goldstein, A. H., Hamilton, J. F., Herrmann,
1035 H., Hoffmann, T., Iinuma, Y., Jang, M., Jenkin, M. E., Jimenez, J. L., Kiendler-Scharr,
1036 A., Maenhaut, W., McFiggans, G., Mentel, T. F., Monod, A., Prevot, A. S. H., Seinfeld,
1037 J. H., Surratt, J. D., Szmigielski, R., and Wildt, J.: The formation, properties and impact
1038 of secondary organic aerosol: current and emerging issues, *Atmospheric Chemistry and
1039 Physics*, 9, 5155-5236, 2009.
- 1040 Haverd, V., Raupach, M., Briggs, P., Canadell, J., Isaac, P., Pickett-Heaps, C.,
1041 Roxburgh, S., van Gorsel, E., Viscarra Rossel, R., and Wang, Z.: Multiple observation
1042 types reduce uncertainty in Australia's terrestrial carbon and water cycles,
1043 *Biogeosciences*, 10, 2011-2040, 2013.
- 1044 Hennigan, C. J., Miracolo, M. A., Engelhart, G. J., May, A. A., Presto, A. A., Lee, T.,
1045 Sullivan, A. P., McMeeking, G. R., Coe, H., Wold, C. E., Hao, W. M., Gilman, J. B.,



- 1046 Kuster, W. C., de Gouw, J., Schichtel, B. A., Collett, J. L., Kreidenweis, S. M., and
1047 Robinson, A. L.: Chemical and physical transformations of organic aerosol from the
1048 photo-oxidation of open biomass burning emissions in an environmental chamber,
1049 *Atmospheric Chemistry and Physics*, 11, 7669-7686, 2011.
- 1050 Honninger, G., von Friedeburg, C., and Platt, U.: Multi axis differential optical
1051 absorption spectroscopy (MAX-DOAS), *Atmospheric Chemistry and Physics*, 4, 231-
1052 254, 2004.
- 1053 Agents Classified by the IARC Monographs. World Health Organization.
1054 <http://monographs.iarc.fr/ENG/Classification/>, 2015.
- 1055 Iinuma, Y., Engling, G., Puxbaum, H., and Herrmann, H.: A highly resolved anion-
1056 exchange chromatographic method for determination of saccharidic tracers for biomass
1057 combustion and primary bio-particles in atmospheric aerosol, *Atmospheric*
1058 *Environment*, 43, 1367-1371, 2009.
- 1059 Ito, A.: Mega fire emissions in Siberia: potential supply of bioavailable iron from
1060 forests to the ocean, *Biogeosciences*, 8, 1679-1697, 2011.
- 1061 Jacobson, M. Z.: Strong radiative heating due to the mixing state of black carbon in
1062 atmospheric aerosols, *Nature*, 409, 695-697, 2001.
- 1063 Jaffe, D. A., and Wigder, N. L.: Ozone production from wildfires: A critical review,
1064 *Atmospheric Environment*, 51, 1-10, 2012.
- 1065 Jenkins, B. M., Jones, A. D., Turn, S. Q., and Williams, R. B.: Emission Factors for
1066 Polycyclic Aromatic Hydrocarbons from Biomass Burning, *Environmental Science &*
1067 *Technology*, 30, 2462-2469, 1996.
- 1068 Kaiser, J. W., and Keywood, M.: Preface for *Atmos. Env. Special issue on IBBI*,
1069 *Atmospheric Environment*, 121, 1-3, 2015.
- 1070 Kallenborn, R., Halsall, C., Dellong, M., and Carlsson, P.: The influence of climate
1071 change on the global distribution and fate processes of anthropogenic persistent organic
1072 pollutants, *Journal of Environmental Monitoring*, 14, 2854-2869, 2012.
- 1073 Keywood, M., Guyes, H., Selleck, P., and Gillett, R.: Quantification of secondary
1074 organic aerosol in an Australian urban location, *Environmental Chemistry*, 8, 115-126,
1075 2011.
- 1076 Keywood, M., Kanakidou, M., Stohl, A., Dentener, F., Grassi, G., Meyer, C. P.,
1077 Torseth, K., Edwards, D., Thompson, A. M., Lohmann, U., and Burrows, J.: Fire in the
1078 air: Biomass burning impacts in a changing climate, *Critical Reviews in Environmental*
1079 *Science and Technology*, 43, 40-83, 2013.
- 1080 Keywood, M., Cope, M., Meyer, C. M., Iinuma, Y., and Emmerson, K.: When smoke
1081 comes to town: The impact of biomass burning smoke on air quality, *Atmospheric*
1082 *Environment*, 121, 13-21, 2015.
- 1083 Landis, M. S., Stevens, R. K., Schaedlich, F., and Prestbo, E. M.: Development and
1084 characterization of an annular denuder methodology for the measurement of divalent
1085 inorganic reactive gaseous mercury in ambient air, *Environmental science &*
1086 *technology*, 36, 3000-3009, 2002.
- 1087 LaRoche, J., and Breitbarth, E.: Importance of the diazotrophs as a source of new
1088 nitrogen in the ocean, *Journal of Sea Research*, 53, 67-91, 2005.
- 1089 Lawson, S. J., Keywood, M. D., Galbally, I. E., Gras, J. L., Caine, J. M., Cope, M. E.,
1090 Krummel, P. B., Fraser, P. J., Steele, L. P., Bentley, S. T., Meyer, C. P., Ristovski, Z.,
1091 and Goldstein, A. H.: Biomass burning emissions of trace gases and particles in marine
1092 air at Cape Grim, Tasmania, *Atmospheric Chemistry and Physics*, 15, 13393-13411,
1093 2015.



- 1094 Lim, Y., Tan, Y., Perri, M., Seitzinger, S., and Turpin, B.: Aqueous chemistry and its
1095 role in secondary organic aerosol (SOA) formation, *Atmospheric Chemistry and*
1096 *Physics*, 10, 10521-10539, 2010.
- 1097 Lin, C.-J., and Pehkonen, S. O.: The chemistry of atmospheric mercury: a review,
1098 *Atmospheric Environment*, 33, 2067-2079, 1999.
- 1099 Lioussé, C., Devaux, C., Dulac, F., and Cachier, H.: Aging of savanna biomass burning
1100 aerosols: Consequences on their optical properties, *Journal of Atmospheric Chemistry*,
1101 22, 1-17, 1995.
- 1102 Lobert, J. M., Keene, W. C., Logan, J. A., and Yevich, R.: Global chlorine emissions
1103 from biomass burning: Reactive chlorine emissions inventory, *Journal of Geophysical*
1104 *Research: Atmospheres*, 104, 8373-8389, 1999.
- 1105 Locatelli, R., Bousquet, P., Hourdin, F., Saunio, M., Cozic, A., Couvreur, F.,
1106 Grandpeix, J. Y., Lefebvre, M. P., Rio, C., Bergamaschi, P., Chambers, S. D., Karstens,
1107 U., Kazan, V., van der Laan, S., Meijer, H. A. J., Moncrieff, J., Ramonet, M., Scheeren,
1108 H. A., Schlosser, C., Schmidt, M., Vermeulen, A., and Williams, A. G.: Atmospheric
1109 transport and chemistry of trace gases in LMDz5B: evaluation and implications for
1110 inverse modelling, *Geoscientific Model Development*, 8, 129-150, 2015.
- 1111 Luo, C., Mahowald, N., Bond, T., Chuang, P., Artaxo, P., Siefert, R., Chen, Y., and
1112 Schauer, J.: Combustion iron distribution and deposition, *Global Biogeochemical*
1113 *Cycles*, 22, 2008.
- 1114 Mallet, M. D., Cravigan, L. T., Milic, A., Alroe, J., Ward, J., Keywood, M. D.,
1115 Ristovski, Z. D., Williams, L. R., Selleck, P., and Miljevic, B.: Composition, size and
1116 cloud condensation nuclei activity of smoke from north Australian savannah fires,
1117 *Atmospheric Chemistry and Physics*, 2016, submitted.
- 1118 Manninen, H. E., Petaja, T., Asmi, E., Riipinen, I., Nieminen, T., Mikkilä, J., Horrak,
1119 U., Mirme, A., Mirme, S., Laakso, L., Kerminen, V. M., and Kulmala, M.: Long-term
1120 field measurements of charged and neutral clusters using Neutral cluster and Air Ion
1121 Spectrometer (NAIS), *Boreal Environment Research*, 14, 591-605, 2009.
- 1122 McFiggans, G., Artaxo, P., Baltensperger, U., Coe, H., Facchini, M. C., Feingold, G.,
1123 Fuzzi, S., Gysel, M., Laaksonen, A., Lohmann, U., Mentel, T. F., Murphy, D. M.,
1124 O'Dowd, C. D., Snider, J. R., and Weingartner, E.: The effect of physical and chemical
1125 aerosol properties on warm cloud droplet activation, *Atmospheric Chemistry and*
1126 *Physics*, 6, 2593-2649, 2006.
- 1127 Melendez-Perez, J. J., Fostier, A. H., Carvalho, J. A., Windmüller, C. C., Santos, J. C.,
1128 and Carpi, A.: Soil and biomass mercury emissions during a prescribed fire in the
1129 Amazonian rain forest, *Atmospheric Environment*, 96, 415-422, 2014.
- 1130 Meyer, C., Cook, G., Reisen, F., Smith, T., Tattaris, M., Russell-Smith, J., Maier, S.,
1131 Yates, C., and Wooster, M.: Direct measurements of the seasonality of emission factors
1132 from savanna fires in northern Australia, *Journal of Geophysical Research:*
1133 *Atmospheres* (1984–2012), 117, 2012.
- 1134 Meyer, C. P., Luhar, A. K., and Mitchell, R. M.: Biomass burning emissions over
1135 northern Australia constrained by aerosol measurements: I—Modelling the distribution
1136 of hourly emissions, *Atmospheric Environment*, 42, 1629-1646, 2008.
- 1137 Milic, A., Mallet, M. D., Cravigan, L. T., Alroe, J., Ristovski, Z. D., Selleck, P.,
1138 Lawson, S. J., Ward, J., Desservettaz, M. J., Paton-Walsh, C., Williams, L. R.,
1139 Keywood, M. D., and Miljevic, B.: Aging of aerosols emitted from biomass burning in
1140 northern Australia, *Atmospheric Chemistry and Physics Discussions*, 2016, 1-24,
1141 10.5194/acp-2016-730, 2016.



- 1142 Mirme, A., Tamm, E., Mordas, G., Vana, M., Uin, J., Mirme, S., Bernotas, T., Laakso,
1143 L., Hirsikko, A., and Kulmala, M.: A wide-range multi-channel Air Ion Spectrometer,
1144 *Boreal Environmental Research*, 12, 247-264, 2007.
- 1145 Mochida, M., and Kawamura, K.: Hygroscopic properties of levoglucosan and related
1146 organic compounds characteristic to biomass burning aerosol particles, *Journal of*
1147 *Geophysical Research: Atmospheres* (1984–2012), 109, 2004.
- 1148 National-Environment-Protection-Council-Service-Corporation: National
1149 Environment Protection (Air Toxics) Measure, 2011.
- 1150 Novakov, T., and Corrigan, C.: Cloud condensation nucleus activity of the organic
1151 component of biomass smoke particles, *Geophysical Research Letters*, 23, 2141-2144,
1152 1996.
- 1153 Packham, D., Tapper, N., Griepsma, D., Friedli, H., Hellings, J., and Harris, S.: Release
1154 of Mercury from Biomatter after Burning: A Preliminary Investigation of Biomatter
1155 and Soils, *Air Quality and Climate Change*, 24-27, 2009.
- 1156 Paris, R., Desboeufs, K., Formenti, P., Nava, S., and Chou, C.: Chemical
1157 characterisation of iron in dust and biomass burning aerosols during AMMA-
1158 SOP0/DABEX: implication for iron solubility, *Atmospheric Chemistry and Physics*,
1159 10, 4273-4282, 2010.
- 1160 Penner, J., Chuang, C., and Grant, K.: Climate forcing by carbonaceous and sulfate
1161 aerosols, *Climate Dynamics*, 14, 839-851, 1998.
- 1162 Pirrone, N., Cinnirella, S., Feng, X., Finkelman, R. B., Friedli, H. R., Leaner, J., Mason,
1163 R., Mukherjee, A. B., Stracher, G. B., Streets, D. G., and Telmer, K.: Global mercury
1164 emissions to the atmosphere from anthropogenic and natural sources, *Atmospheric*
1165 *Chemistry and Physics*, 10, 5951-5964, 2010.
- 1166 Radhi, M., Box, M. A., Box, G. P., and Mitchell, R. M.: Biomass-burning aerosol over
1167 northern Australia, *Australian Meteorological and Oceanographic Journal*, 62, 25,
1168 2012.
- 1169 Rea, A. W., Lindberg, S. E., Scherbatskoy, T., and Keeler, G. J.: Mercury accumulation
1170 in foliage over time in two northern mixed-hardwood forests, *Water, Air, and Soil*
1171 *Pollution*, 133(1-4), 49-67, 2002.
- 1172 Reid, J. S., Hobbs, P. V., Ferek, R. J., Blake, D. R., Martins, J. V., Dunlap, M. R., and
1173 Liousse, C.: Physical, chemical, and optical properties of regional hazes dominated by
1174 smoke in Brazil, *Journal of Geophysical Research: Atmospheres* (1984–2012), 103,
1175 32059-32080, 1998.
- 1176 Rosen, J., Young, S., Laby, J., Kjome, N., and Gras, J.: Springtime aerosol layers in the
1177 free troposphere over Australia: Mildura Aerosol Tropospheric Experiment (MATE
1178 98), *Journal of Geophysical Research: Atmospheres* (1984–2012), 105, 17833-17842,
1179 2000.
- 1180 Rubin, M., Berman-Frank, I., and Shaked, Y.: Dust-and mineral-iron utilization by the
1181 marine dinitrogen-fixer *Trichodesmium*, *Nature Geoscience*, 4, 529-534, 2011.
- 1182 Russell-Smith, J., Yates, C. P., Whitehead, P. J., Smith, R., Craig, R., Allan, G. E.,
1183 Thackway, R., Frakes, I., Cridland, S., Meyer, M. C. P., and Gill, M.: Bushfires' down
1184 under': patterns and implications of contemporary Australian landscape burning,
1185 *International Journal of Wildland Fire*, 16, 361-377, 2007.
- 1186 Russell-Smith, J., Cook, G. D., Cooke, P. M., Edwards, A. C., Lendrum, M., Meyer,
1187 C., and Whitehead, P. J.: Managing fire regimes in north Australian savannas: applying
1188 Aboriginal approaches to contemporary global problems, *Frontiers in Ecology and the*
1189 *Environment*, 11, e55-e63, 2013.
- 1190 Schroth, A. W., Crusius, J., Sholkovitz, E. R., and Bostick, B. C.: Iron solubility driven
1191 by speciation in dust sources to the ocean, *Nature Geoscience*, 2, 337-340, 2009.



- 1192 Sedwick, P. N., Sholkovitz, E. R., and Church, T. M.: Impact of anthropogenic
1193 combustion emissions on the fractional solubility of aerosol iron: Evidence from the
1194 Sargasso Sea, *Geochemistry, Geophysics, Geosystems*, 8, 2007.
- 1195 Shen, H., Huang, Y., Wang, R., Zhu, D., Li, W., Shen, G., Wang, B., Zhang, Y., Chen,
1196 Y., Lu, Y., Chen, H., Li, T., Sun, K., Li, B., Liu, W., Liu, J., and Tao, S.: Global
1197 Atmospheric Emissions of Polycyclic Aromatic Hydrocarbons from 1960 to 2008 and
1198 Future Predictions, *Environmental Science & Technology*, 47, 6415-6424,
1199 10.1021/es400857z, 2013.
- 1200 Shi, Y., Matsunaga, T., Saito, M., Yamaguchi, Y., and Chen, X.: Comparison of global
1201 inventories of CO₂ emissions from biomass burning during 2002–2011 derived from
1202 multiple satellite products, *Environmental Pollution*, 206, 479-487, 2015.
- 1203 Sinreich, R., Friess, U., Wagner, T., and Platt, U.: Multi axis differential optical
1204 absorption spectroscopy (MAX-DOAS) of gas and aerosol distributions, *Faraday*
1205 *Discuss*, 130, 153-164, 2005.
- 1206 Sprovieri, F., Pirrone, N., Bencardino, M., D'Amore, F., Carbone, F., Cinnirella, S.,
1207 Mannarino, V., Landis, M., Ebinghaus, R., Weigelt, A., Brunke, E. G., Labuschagne,
1208 C., Martin, L., Munthe, J., Wängberg, I., Artaxo, P., Morais, F., Cairns, W., Barbante,
1209 C., Diéguez, M. D. C., Garcia, P. E., Dommergue, A., Angot, H., Magand, O., Skov,
1210 H., Horvat, M., Kotnik, J., Read, K. A., Neves, L. M., Gawlik, B. M., Sena, F.,
1211 Mashyanov, N., Obolkin, V. A., Wip, D., Feng, X. B., Zhang, H., Fu, X.,
1212 Ramachandran, R., Cossa, D., Knoery, J., Maruszczak, N., Nerentorp, M., and Norstrom,
1213 C.: Atmospheric Mercury Concentrations observed at ground-based monitoring sites
1214 globally distributed in the framework of the GMOS network, *Atmospheric Chemistry*
1215 *and Physics Discussions*, 2016, 1-32, 2016.
- 1216 Steffen, A., Douglas, T., Amyot, M., Ariya, P., Aspino, K., Berg, T., Bottenheim, J.,
1217 Brooks, S., Cobbett, F., Dastoor, A., Dommergue, A., Ebinghaus, R., Ferrari, C.,
1218 Gardfeldt, K., Goodsite, M. E., Lean, D., Poulain, A. J., Scherz, C., Skov, H., Sommar,
1219 J., and Temme, C.: A synthesis of atmospheric mercury depletion event chemistry in
1220 the atmosphere and snow, *Atmospheric Chemistry and Physics*, 8, 1445-1482, 2008.
- 1221 Stokes, R., and Robinson, R.: Interactions in aqueous nonelectrolyte solutions. I.
1222 Solute-solvent equilibria, *The Journal of Physical Chemistry*, 70, 2126-2131, 1966.
- 1223 Tuch, T. M., Haudek, A., Müller, T., Nowak, A., Wex, H., and Wiedensohler, A.:
1224 Design and performance of an automatic regenerating adsorption aerosol dryer for
1225 continuous operation at monitoring sites, *Atmos. Meas. Tech.*, 2, 417-422, 2009.
- 1226 Twomey, S.: Aerosols, clouds and radiation, *Atmospheric Environment. Part A.*
1227 *General Topics*, 25, 2435-2442, 1991.
- 1228 van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G., Mu, M., Kasibhatla, P.
1229 S., Morton, D. C., DeFries, R., Jin, Y. v., and van Leeuwen, T. T.: Global fire emissions
1230 and the contribution of deforestation, savanna, forest, agricultural, and peat fires (1997–
1231 2009), *Atmospheric Chemistry and Physics*, 10, 11707-11735, 2010.
- 1232 Wang, X., Thai, P. K., Li, Y., Li, Q., Wainwright, D., Hawker, D. W., and Mueller, J.
1233 F.: Changes in atmospheric concentrations of polycyclic aromatic hydrocarbons and
1234 polychlorinated biphenyls between the 1990s and 2010s in an Australian city and the
1235 role of bushfires as a source, *Environmental Pollution*, 213, 223-231, 2016.
- 1236 Wang, X., P.K., T., Mallet, M. D., Desservettaz, M. J., Hawker, D. W., Keywood, M.
1237 K., Miljevic, B., Paton-Walsh, C., Gallen, M., and Mueller, J. F.: Emissions of selected
1238 semivolatile organic chemicals from forest and savannah fires, *Environmental Science*
1239 *& Technology*, 2016, under review.



- 1240 Whittlestone, S., and Zahorowski, W.: Baseline radon detectors for shipboard use:
1241 Development and deployment in the First Aerosol Characterization Experiment (ACE
1242 1), *Journal of Geophysical Research: Atmospheres*, 103, 16743-16751, 1998.
1243 Wigder, N., Jaffe, D., and Saketa, F.: Ozone and particulate matter enhancements from
1244 regional wildfires observed at Mount Bachelor during 2004–2011, *Atmospheric
1245 Environment*, 75, 24-31, 2013.
1246 Williams, A. G., Chambers, S., Zahorowski, W., Crawford, J., Matsumoto, K., and
1247 Uematsu, M.: Estimating the Asian radon flux density and its latitudinal gradient in
1248 winter using ground-based radon observations at Sado Island, *Tellus B*, 61, 732-746,
1249 2009.
1250 Williams, A. G., Zahorowski, W., Chambers, S., Griffiths, A., Hacker, J. M., Element,
1251 A., and Werczynski, S.: The vertical distribution of radon in clear and cloudy daytime
1252 terrestrial boundary layers, *Journal of the Atmospheric Sciences*, 68, 155-174, 2011.
1253 Winton, V., Bowie, A., Edwards, R., Keywood, M., Townsend, A., van der Merwe, P.,
1254 and Bollhöfer, A.: Fractional iron solubility of atmospheric iron inputs to the Southern
1255 Ocean, *Marine Chemistry*, 177, 20-32, 2015.
1256 Winton, V., Edwards, R., Bowie, A., Keywood, M., Williams, A., Chambers, S.,
1257 Selleck, P., Desservettaz, M., Mallet, M., and Paton-Walsh, C.: Dry season aerosol iron
1258 solubility in tropical northern Australia, *Atmospheric Chemistry and Physics
1259 Discussions*, doi:10.5194/acp-2016-419, 2016.
1260 Wong, J., and Li, Z.: Retrieval of optical depth for heavy smoke aerosol plumes:
1261 uncertainties and sensitivities to the optical properties, *Journal of the Atmospheric
1262 Sciences*, 59, 250-261, 2002.
1263 Yokelson, R. J., Crouse, J. D., DeCarlo, P. F., Karl, T., Urbanski, S., Atlas, E.,
1264 Campos, T., Shinozuka, Y., Kapustin, V., Clarke, A. D., Weinheimer, A., Knapp, D. J.,
1265 Montzka, D. D., Holloway, J., Weibring, P., Flocke, F., Zheng, W., Toohey, D.,
1266 Wennberg, P. O., Wiedinmyer, C., Mauldin, L., Fried, A., Richter, D., Walega, J.,
1267 Jimenez, J. L., Adachi, K., Buseck, P. R., Hall, S. R., and Shetter, R.: Emissions from
1268 biomass burning in the Yucatan, *Atmospheric Chemistry and Physics*, 9, 5785-5812,
1269 2009.
1270

1271