





36 particulate matter, usually called  $PM_{2.5}$  (Huang et al., 2014).  $PM_{2.5}$  can directly reduce atmospheric  
37 visibility by scattering or absorbing solar light (Seinfeld and Pandis, 1998; Buseck and Posfai,  
38 1999; Cheng et al., 2006) and is harmful to human health (Finlayson-Pitts and Pitts, 2000; Nel,  
39 2005; Poschl, 2005; Peplow, 2014).

40 To mitigate the serious pollution status, identification of the sources of  $PM_{2.5}$  is urgently needed for  
41 the effective control measures. Based on field measurements, positive matrix factorization (PMF)  
42 (Yu et al., 2013; Wu et al., 2014; Huang et al., 2014), principal component analysis (PCA) (Wang et  
43 al., 2015) and chemical mass balance (CMB) (Huang et al., 2014; Guo et al., 2012) have been widely  
44 used for identifying the sources of  $PM_{2.5}$ . However, the results of the source apportionment are still  
45 not convincing because there are large uncertainties about the indicators, dominant factors and  
46 emission inventories used for the identification. For example, some studies suggested traffic  
47 emissions in Beijing contributed about 15~20% to the  $PM_{2.5}$  (Yu et al., 2013; Wu et al., 2014), while  
48 only 4% of the contribution was also reported (Huang et al., 2014). Additionally, the current source  
49 apportionment can only present gross contribution of each source classification, but there are  
50 markedly different emissions from individual sources in the same classification. For example, due  
51 to the strict control measures and highly efficient combustion, the emissions of pollutants from  
52 power plants and big boilers fueled by coal must be totally different from the domestic coal stoves  
53 on both the emission strengths and composition of pollutants. Finally, most studies about source  
54 apportionment mainly focused on emissions from traffic, industry, construction and secondary  
55 formation, whereas the emissions from farmers' activities in the NCP were almost neglected.

56 There are about 300,000 km<sup>2</sup> agricultural fields and 0.16 billion farmers in the NCP (Zhang et al.,  
57 2011). The farmers' activities in the NCP are very seasonal, e.g., the fertilization events and harvests



58 mainly focus on June-July and October-November and domestic coal stoves are prevailingly used  
59 for heating in winter. The seasonal activities of farmers in the NCP were suspected to make  
60 significant contribution to deteriorate the regional air quality, e.g., the most serious pollution events  
61 (or haze days) in the NCP were usually coincident with the three seasonal activities of farmers in  
62 recent years (Yang et al., 2015b;Huang et al., 2012;Li et al., 2014;Li et al., 2011;Liu et al., 2013;Sun  
63 et al., 2013). The serious pollution events during harvest seasons were widely ascribed to crop straw  
64 burning (Huang et al., 2012;Li et al., 2014), but the influence of fertilization events and crop straw  
65 returning to fields on the regional air quality during the harvest seasons periods was totally neglected.  
66 Strong ammonia ( $\text{NH}_3$ ) emission from the vast agricultural fields in the NCP has been found during  
67 fertilization events just after harvest of winter wheat in June-July (Zhang et al., 2011), which must  
68 accelerate atmospheric ammonium formation. Although crop straws burning by stealth is still  
69 prevailing, most residual crops are being returned into the agricultural fields under the advocacy of  
70 government for protecting the air quality. Because crop leaves absorbed large quantities of  
71 atmospheric particles during crop growing season, the abrupt release of the particles by smashing  
72 crop straw for returning in the vast area of the NCP must also make striking contribution to  
73 atmospheric particles in the region during the seasonal harvest seasons. In winter, the serious  
74 pollutant emissions from the chimney of the farmers' coal stoves can be easily imagined by the  
75 strong smog. Although domestic coal consumption only accounts for small fraction of the total, e.g.,  
76 ~11% in Beijing-Tianjin-Hebei area (<http://hbdczx.mep.gov.cn/pub/>), the emission strengths of  
77 pollutants from farmers' coal stove is usually about 1-2 magnitude greater than those from power  
78 plants (Xu et al., 2006), and the farmers coal consumption mainly concentrates on the four months  
79 in winter.



80 In this study, to understand the possible influence of farmers' activities on the regional air quality in  
81 the NCP, filter samples of PM<sub>2.5</sub> were daily collected in Beijing city for a whole year of 2014, and  
82 the seasonal variation characteristics of the water-soluble ions (WSIs) in the PM<sub>2.5</sub> samples were  
83 comprehensively investigated in relation to the farmers' activities. The scientific evidences found  
84 in this study will be helpful for future control measures in reducing pollutant emissions from rural  
85 areas in the NCP.

## 86 **2. Materials and methods**

### 87 **2.1. Sampling**

88 The sampling site was chosen on a rooftop (about 25m above ground) in the Research Center for  
89 Eco-Environmental Sciences (RCEES), which is located between the north fourth-ring road and the  
90 north fifth-ring road of Beijing and surrounded by some institutes, campuses, and residential areas  
91 (Pang and Mu, 2006). PM<sub>2.5</sub> samples were collected on Millipore PTFE filters (90mm) by an  
92 artificial intelligence's PM<sub>2.5</sub> sampler (LaoYing-2034) and the sampling flow rate was set to 100L  
93 min<sup>-1</sup>. The duration of each sampling was 24 hours, started at 3:00 p.m. every day and ended at 3:00  
94 p.m. on the next day. All the samples were put in dedicated filter storage containers (90mm,  
95 Millipore) after sampling and preserved in a refrigerator till analysis. A total of 235 PM<sub>2.5</sub> samples  
96 were collected from January to November of 2014, in winter (Jan 9- Mar 15 2014), spring (Mar 16-  
97 May 31 2014), summer (Jun 1- Jun 30, Aug 9- Aug 21 2014) and autumn (Sep 19- Nov 14 2014).

### 98 **2.2. Ion analysis**

99 Sample and blank filters were extracted ultrasonically with 10mL ultrapure water for half an hour.  
100 The solutions were filtered through water micro-porous membrane (pore size, 0.45µm; diameter,  
101 13mm) before analysis and the water-soluble ions (WSIs) in the treated filtrates were analyzed by



102 Ion Chromatography (IC, WAYEE IC6200). Five anions ( $F^-$ ,  $HCOO^-$ ,  $Cl^-$ ,  $NO_3^-$  and  $SO_4^{2-}$ ) were  
103 separated by using an anion column (IC SI-52 4E, 4mmID\*250mm) with the eluent ( $3.6\text{mmol L}^{-1}$   
104  $Na_2CO_3$ ) flow rate of  $0.8\text{mL min}^{-1}$  and column temperature of  $45\text{ }^\circ\text{C}$ . Five cations ( $Na^+$ ,  $NH_4^+$ ,  $Mg^{2+}$ ,  
105  $Ca^{2+}$  and  $K^+$ ) were separated by using a cation column (TSKgelSuperIC-CR, 4.6mmID\*15cm) with  
106 the eluent ( $2.2\text{mmol L}^{-1}$  MSA and  $1\text{mmol L}^{-1}$  18-crown-6) flow rate of  $0.7\text{mL min}^{-1}$  and column  
107 temperature of  $40\text{ }^\circ\text{C}$ . The relative standard deviation (RSD) of each ion was less than 0.5% for the  
108 reproducibility test. The detection limits ( $S/N=3$ ) were less than  $0.001\text{ mg L}^{-1}$  for the anions and  
109 cations. The concentrations of all the ions (less than  $0.03\text{ mg L}^{-1}$  for each ion) in daily field blank  
110 filter were subtracted from sample determination.

### 111 **2.3. Meteorology, trace gases and back trajectory**

112 The meteorological data, including temperature, wind speed, wind direction, relative humidity (RH),  
113 visibility and Air Pollution Index of  $PM_{2.5}$ ,  $SO_2$ ,  $NO_2$ ,  $O_3$  in RCEES were both collected from  
114 Beijing urban ecosystem research station (<http://www.bjurban.rcees.cas.cn/>).

115 To identify the potential influence of air parcel transport, the air mass backward trajectories were  
116 calculated for 72h through the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPPLIT  
117 4) Model of the Air Resources Laboratory of NOAA with NCEP Final analyses data. The backward  
118 trajectories arriving at 500m above sampling position were computed at 0:00h, 6:00h, 12:00h and  
119 18:00h (UTC) each sampling day respectively. A total of 940 backward trajectories with 72 hourly  
120 trajectory endpoints in four seasons were used as input for further analysis.

### 121 **2.4 The TEOM 1405 Monitor**

122 The mass concentration of  $PM_{2.5}$  was monitored by a tapered element oscillating microbalance with  
123 the filter dynamic measurement system (TEOM-FDMS, Thermo; Model 1405). A filter in the



124 TEOM 1405 Monitor is used for collecting and measuring  $PM_{2.5}$  through variation of the oscillation  
125 frequency. To avoid water condensation on the TEOM filter, the temperature of the TEOM filter as  
126 well as the inlet is kept at 50 °C during sampling. In this study, we replaced the TEOM filters every  
127 12 days, and the concentrations of the WSIs on the TEOM filters were analyzed for comparing with  
128 those on the filter collected by the filter sampling method.

### 129 **3. Results and discussion**

#### 130 **3.1. Comparison between WSIs and $PM_{2.5}$**

131 The mass concentrations of WSIs and  $PM_{2.5}$  at the sampling site were simultaneously measured by  
132 the filter sampling method and the TEOM 1405 Monitor for 24 days (Jan 1- Jan 24, 2015). As  
133 shown in Fig. 1a and Fig. 1b, the variation trends of the WSIs and  $PM_{2.5}$  were almost the same  
134 with a correlation coefficient ( $R^2$ ) of 0.908, implying that the concentration of WSIs measured  
135 could well reveal the pollution status of  $PM_{2.5}$  in Beijing. The average mass concentration of WSIs  
136 contributed about 80% to the mass of  $PM_{2.5}$  measured by the TEOM 1405 Monitor, whereas the  
137 WSIs accounted for about 50-60% of the total mass concentration measured by the filter sampling  
138 method in the NCP (Shen et al., 2009; Li et al., 2013). The mass concentration of  $PM_{2.5}$  measured  
139 by the TEOM 1405 Monitor was suspected to be largely underestimated because the volatile even  
140 semi-volatile component in  $PM_{2.5}$  can be easily lost at 50 °C which is designed in the TEOM 1405  
141 Monitor for avoiding water condensation on the filter (Grover et al., 2005; Liu et al., 2014), e.g.,  
142 under clean days after serious pollution episodes, the mass concentration of WSIs was even higher  
143 than the mass concentration of  $PM_{2.5}$  measured by the TEOM 1405 Monitor (Fig. 1a). To verify  
144 above assumption, the concentrations of WSIs on the filters collected by the filter sampling  
145 method and the TEOM 1405 Monitor were comparatively measured, and the results are illustrated



146 in Fig. 1c and Fig. 1d. It is evident that the proportions of  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{Cl}^-$  on the filter  
147 collected by the TEOM 1405 Monitor were dramatically lower than those on the filter collected by  
148 the filter sampling method, whereas  $\text{SO}_4^{2-}$  was on the contrary. It is well documented that  
149 temperature is a key factor affecting the distribution of both  $\text{NH}_4\text{NO}_3$  and  $\text{NH}_4\text{Cl}$  on particle phase  
150 due to their thermo decomposition, e.g., at temperature greater than 35 °C, little  $\text{NH}_4\text{NO}_3$  is  
151 expected under typical ambient conditions (Finlayson-Pitts et al., 1986). The negative  $\text{PM}_{2.5}$   
152 values of the TEOM 1405 Monitor after a serious pollution episode also indicated the serious loss  
153 of the volatile component. Although the TEOM 1405 Monitor is widely used for measuring  
154 atmospheric  $\text{PM}_{2.5}$  in the net stations of China, the pollution levels measured could only represent  
155 the lower limits, especially under the clean days after serious pollution episodes in winter.

### 156 3.2. Daily variations of WSIs in each season

157 The daily variations of WSIs in each season are illustrated in Fig. 2 and the statistic mass  
158 concentrations of the WSIs are summarized in Table 1. It is evident that the daily variations of the  
159 WSIs exhibited significantly periodic fluctuation, indicating meteorological conditions played a  
160 pivotal role in accumulation and dissipation of atmospheric pollutants. For example, the most  
161 frequently high pollution levels of the WSIs in winter were mainly ascribed to the relatively stable  
162 meteorological conditions with the low height of boundary layer which favors pollutants  
163 accumulation (Wang et al., 2013;Quan et al., 2014;Tian et al., 2014;Wang et al., 2014;Zhang et al.,  
164 2015a). Besides meteorological conditions, the extremely high levels of the WSIs during the  
165 pollution episodes revealed strong sources of the pollutants around Beijing.

166 The mean concentrations ( $\mu\text{g m}^{-3}$ ) of WSIs in spring, summer, autumn and winter were  $50.5 \pm 37.3$ ,  
167  $44.2 \pm 28.9$ ,  $78.3 \pm 92.6$ , and  $78.7 \pm 61.2$ , respectively.  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$  and  $\text{NH}_4^+$  were found to be the



168 principal ions, accounted for about 80% to the total WSIs in each season, which were in line with  
169 previous studies (Hu et al., 2014; Yang et al., 2015a; Huang et al., 2016; Yang et al., 2015b). The three  
170 principal ions were mainly ascribed to secondary formation as discussed in the following section.  
171 Although the most intensive photochemical reactivity in summer favors sulfate and nitrate  
172 formation, the relatively low SO<sub>2</sub> concentration, the fast thermal decomposition of ammonium  
173 nitrate and the frequent scavenging by rain events must greatly counteract the contribution of the  
174 secondary formation, resulting in the lowest pollution levels of the WSIs in summer. In comparison  
175 with other seasons, the remarkable elevation of atmospheric SO<sub>2</sub> and NO<sub>x</sub> (see Sect. 3.3) in winter  
176 would override the relatively low atmospheric photo-oxidants for their oxidation rates and resulted  
177 in the highest mean concentration of WSIs. Although the atmospheric concentrations of SO<sub>2</sub> and  
178 NO<sub>x</sub> in autumn were much smaller than in winter and in spring (see Sect. 3.3), the mean  
179 concentration of WSIs in autumn was almost the same as that in winter and nearly twice as those in  
180 spring and summer, indicating that special mechanisms dominated the secondary formation of the  
181 atmospheric principal ions (see Sect. 3.3).

### 182 3.3. The possible sources for the WSIs

183 To disclose the contribution of possible sources to the WSIs, the molar composition of the WSIs,  
184 the seasonal variation characteristics of typical WSIs, the variation characteristics of the three  
185 principal ions during serious pollution episodes, the contribution of secondary formation to  
186 atmospheric WSIs, and backward trajectories of air parcels were comprehensively analyzed.

#### 187 3.3.1. The molar composition of the WSIs

188 The molar composition of water-soluble ions in each season under three pollution levels is illustrated  
189 in Fig. 3. With increasing pollution levels, the noticeable reduction of the proportions of metallic





190 ions (such as  $\text{Ca}^{2+}$ ,  $\text{Na}^+$  and  $\text{Mg}^{2+}$ ) and the evident increase of  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  proportions  
191 revealed that the three principle ions ( $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ ) were mainly from atmospheric  
192 secondary formation. Compared with  $\text{SO}_4^{2-}$ , the fast increase of  $\text{NO}_3^-$  proportion with increasing  
193 pollution levels indicated that the formation rate of nitrate was faster than that of sulfate under higher  
194 pollution levels. It should be mentioned that the increase rate of  $\text{NO}_3^-$  proportion with increasing  
195 pollution levels was much slower in summer than in other seasons, validating that nitrate was easily  
196 thermal decomposed under high temperature. The conspicuous reduction of  $\text{Cl}^-$  proportion with  
197 increasing pollution levels meant  $\text{Cl}^-$  might be mainly from primary sources.

### 198 3.3.2. The seasonal variation characteristics of typical WSIs

199 The seasonal variation characteristics of typical WSIs are illustrated in Fig. 4. For  $\text{Cl}^-$  and  $\text{K}^+$ , their  
200 high concentrations mainly occurred in winter and autumn. It should be mentioned that the  
201 extremely high concentration of  $\text{K}^+$  in winter on 1 February (Fig. 2) was due to firework for  
202 celebrating Chinese lunar year (Jiang et al., 2015; Kong et al., 2015). Sea-salt has long been  
203 considered as the source for atmospheric  $\text{Cl}^-$  (Souza et al., 2014), however, the molar ratio of  $\text{Cl}^-$  to  
204  $\text{Na}^+$  measured by this study (Fig. 5) in each season was above 1.30 which was much greater than  
205 the value of 1.18 in fresh sea-salt particles (Brewer, 1975), indicating sources other than sea-salt  
206 dominated atmospheric  $\text{Cl}^-$  in Beijing. Because  $\text{K}^+$  has been widely used as an indicator for biomass  
207 burning (Gao et al., 2011) and crop straw burning by stealth was prevailing in the countryside around  
208 Beijing during autumn and winter seasons, crop straw burning was suspected to be a common source  
209 for  $\text{K}^+$  and  $\text{Cl}^-$  (Li et al., 2014). The pronounced correlation coefficients ( $r > 0.6$ ,  $p < 0.01$ ) between  
210  $\text{K}^+$  and  $\text{Cl}^-$  in the two seasons might be the circumstantial evidence for above suspicion. Several  
211 studies have reported extremely high emission factors of  $\text{Cl}^-$  (80-300mg  $\text{Cl}^-$ /kg coal) from the coal



212 combustion in China (Huang et al., 2014). Because large fraction of coal consumed by farmers for  
213 heating in winter was the extra source for atmospheric pollutants in the vast area of North China,  
214 the obviously higher  $\text{Cl}^-$  concentrations measured in winter than in other seasons (Fig. 2) indicated  
215 that coal combustion by farmers in winter might make great contribution to atmospheric  $\text{Cl}^-$  in  
216 Beijing. The source of atmospheric  $\text{NO}_x$  in Beijing is dominated by vehicles and relatively stable in  
217 the four seasons, and hence the ratios of  $\text{Cl}^-$  to  $\text{NO}_x$  can largely counteract the influence of  
218 accumulation and dispersion due to variation of meteorological factors for identifying the possible  
219 extra source of  $\text{Cl}^-$ . The ratio of  $\text{Cl}^-$  to  $\text{NO}_x$  in winter was about a factor of 2 greater than those in  
220 other seasons (Fig. 5), confirming that coal combustion by farmers in winter indeed made evident  
221 contribution to atmospheric  $\text{Cl}^-$  in Beijing. Previous field investigations in different areas of Chinese  
222 mainland also found relatively high  $\text{Cl}^-$  concentration in winter, which was also ascribed to coal  
223 combustion (Yu et al., 2013; Wu et al., 2014). In addition, fertilization events in the agricultural  
224 fields around Beijing might also make contribution to atmospheric  $\text{Cl}^-$ , because the volatile  
225 ammonium chloride is a kind of prevailingly used fertilizer in the NCP, e.g., the extremely high  
226 ratios of  $\text{Cl}^-$  to  $\text{Na}^+$  (Fig. 5) were coincident with the cultivation seasons of spring and summer.  
227 For  $\text{Ca}^{2+}$ , remarkably high concentrations occurred in both spring and autumn. The evident elevation  
228 of  $\text{Ca}^{2+}$  concentrations in spring has been usually ascribed to the frequent dust storm (Zhao et al.,  
229 2013b), but there was still no explanation about the extremely high  $\text{Ca}^{2+}$  concentrations in autumn  
230 (Zhao et al., 2013b; Zhang et al., 2013). The three serious pollution events with remarkable elevation  
231 of  $\text{Ca}^{2+}$  (Fig. 2) were coincident with the intensive harvest of maize and tillage of the agricultural  
232 fields for planting winter wheat in the countryside around Beijing, and hence the extremely high  
233  $\text{Ca}^{2+}$  concentrations in autumn were suspected to be from the farmers' activities. Because abundant



234 atmospheric mineral particles were absorbed by crop leaves during crop growing season, especially  
235 in the North China where atmospheric mineral dust is always at high level (Zhang et al., 2013;Zhao  
236 et al., 2013b), a large fraction of the mineral dust absorbed on the leaves of crop would be released  
237 into the atmosphere during harvest with crop straw being crushed into pieces for returning to fields  
238 which is a prevailing cultivation manner under the advocacy of governments for reducing the  
239 influence of crop straw burning on the air quality.

240 For  $\text{NH}_4^+$ ,  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$ , remarkably high concentrations also appeared in both winter and autumn.  
241  $\text{NH}_4^+$  was mainly from the reactions of  $\text{NH}_3$  with acid gases (such as  $\text{HNO}_3$ ) and acid particles, and  
242 hence its variation trend was the same as those of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$ . Although atmospheric  $\text{NH}_3$  has  
243 long been considered to be mainly from agricultural activities, their emissions mainly focus on  
244 warmer seasons (Krupa, 2003). However, the frequently high concentrations of  $\text{NH}_4^+$  appeared in  
245 winter. Beside the slow thermal decomposition of ammonium nitrate, strong  $\text{NH}_3$  emission sources  
246 other than agricultural activities were suspected to be responsible for the frequently high  
247 concentrations of  $\text{NH}_4^+$  in the cold winter. Emissions of  $\text{NH}_3$  from vehicles was regarded as an  
248 important source (Liu et al., 2014). In addition, strong emission of  $\text{NH}_3$  from domestic coal stoves  
249 was indeed found by our preliminary measurements (data were not shown). During the serious  
250 pollution episodes, the concentrations of  $\text{SO}_2$  in autumn were almost the same as those in summer  
251 and about one magnitude lower than in winter (Fig. 6), but the peak concentrations of  $\text{SO}_4^{2-}$  in  
252 autumn were about two times greater than those in summer and at almost the same level as those in  
253 winter. The gaseous phase reaction with OH (Zhao et al., 2013c;Quan et al., 2014), the  
254 heterogeneous reaction on mineral dust (He et al., 2014;Nie et al., 2014), and multiphase reactions  
255 in the water of particulate matters (Zheng et al., 2015a) of  $\text{SO}_2$  have been recognized to be



256 responsible for atmospheric  $\text{SO}_4^{2-}$  formation. The significant elevation of both  $\text{Ca}^{2+}$  and  $\text{SO}_4^{2-}$  in  
257 autumn implied that the heterogeneous reaction of  $\text{SO}_2$  on the mineral dust might greatly accelerate  
258 the conversion of  $\text{SO}_2$  to  $\text{SO}_4^{2-}$ . Although evidently high concentrations of  $\text{Ca}^{2+}$  occurred (Fig. 2 and  
259 Fig. 4) in spring and  $\text{SO}_2$  concentrations were much greater in spring than in autumn (Fig. 6), the  
260  $\text{SO}_4^{2-}$  concentrations were about a factor of 2 less in spring than in autumn. Atmospheric humidity  
261 was suspected to play an important role in the heterogeneous reaction, e.g., the relative humidity  
262 was much higher in autumn than in spring during the serious pollution events (Fig. 6). Similar to  
263  $\text{SO}_4^{2-}$ , the relatively high concentrations of  $\text{NO}_3^-$  during the serious pollution events in autumn were  
264 also ascribed to the heterogeneous reaction of  $\text{NO}_2$  on the mineral dust.

### 265 3.3.3. The variation characteristics of the three principal ions during serious pollution episodes

266 As shown in Fig. 6, the serious pollution episodes with noticeable elevation of various pollutants  
267 usually occurred under slow wind speed (less than  $2 \text{ m s}^{-1}$ ) and high relative humidity. In comparison  
268 with their precursors of  $\text{SO}_2$  and  $\text{NO}_x$ , however the detailed variation trends of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  were  
269 different, indicating that the elevation of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  was not simply ascribed to the physical  
270 process of accumulation. It is interesting to be noted that the increasing rates of  $\text{SO}_4^{2-}$  during some  
271 serious pollution events especially with elevation of  $\text{Ca}^{2+}$  (such as in spring and autumn) were much  
272 slower than those of  $\text{NO}_3^-$ , implying that the atmospheric heterogeneous reaction of  $\text{NO}_2$  on the  
273 mineral dust might be faster than that of  $\text{SO}_2$ . In comparison with summer and winter, the relatively  
274 high ratios of  $\text{NO}_3^-/\text{SO}_4^{2-}$  in spring and autumn (Fig. 5) also supported above assumption.

### 275 3.3.4. Secondary formation for atmospheric sulfate and nitrate

276 The nitrogen oxidation ratio  $\text{NOR} = n\text{NO}_3^- / (n\text{NO}_3^- + n\text{NO}_x)$  (n refers to molar concentration) and  
277 the sulfur oxidation ratio  $\text{SOR} = n\text{SO}_4^{2-} / (n\text{SO}_4^{2-} + n\text{SO}_2)$  have been used to estimate the degree of



278 secondary formation of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ , which can counteract the interference of meteorological  
279 factors (Chan and Yao, 2008; Yu et al., 2013; Guo et al., 2014; Huang et al., 2014; Yang et al.,  
280 2015b; Zheng et al., 2015b). The values of NOR and SOR during haze days and non-haze days in  
281 four seasons are listed in Table 2. Both the values of NOR and SOR on non-haze days were found  
282 to be the highest in summer and the lowest in winter, well reflecting the seasonal variation of  
283 photochemical intensity. Although sunlight intensity greatly reduced at ground level during haze  
284 days, the values of NOR and SOR were about a factor of 2 greater during haze days than during  
285 non-haze days in the four seasons, implying again that the heterogeneous or multiphase reactions of  
286  $\text{SO}_2$  and  $\text{NO}_2$  on atmospheric particles made significant contribution to atmospheric sulfate and  
287 nitrate.

### 288 3.3.5. The influence of air mass transport on the WSIs in Beijing

289 To reveal the air mass transport influence on the WSIs in Beijing, three-day backward trajectories  
290 for clusters and the corresponding mass concentrations of WSIs in each season were analyzed, and  
291 the results are illustrated in Fig. 7. It could be seen that the lowest concentrations of WSIs usually  
292 occurred in the northwest/northeast airflow with long distance transport. Because Beijing is  
293 surrounded by mountains in the north/northwest/northeast directions where the population is sparse,  
294 these clusters brought the relatively clean air mass to accelerate the dissipation of aerosols. The  
295 highest concentrations of WSIs (especially for  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$  and  $\text{NH}_4^+$ ) were usually observed in the  
296 air parcel from southwest/south regions with high density of population. Considering the large  
297 fraction (~30%) of air parcel from the southwest/south regions in each season, the human activities  
298 in the southwest/south regions made evident contribution to the atmospheric WSIs in Beijing.  
299 Besides the industries, the emissions from the high density of farmers in the southwest/south regions



300 of Beijing was also suspected to make evident contribution to the atmospheric WSIs in Beijing, e.g.,  
301 the remarkable elevations of  $\text{Cl}^-$  in winter and  $\text{Ca}^{2+}$  in autumn were probably from farmers' coal  
302 combustion for heating and harvest of maize, respectively.

### 303 **3.4. Comparison with previous studies**

304 The mean concentrations of the three principal ions and some related indicators in Beijing over the  
305 past decade are summarized in Table 3. The seasonal variations of the three principal ions reported  
306 were quite different, e.g., Huang et al. (2016) found the maximal mean concentrations of  $\text{SO}_4^{2-}$  and  
307  $\text{NH}_4^+$  in the summer and of  $\text{NO}_3^-$  in the autumn of 2014, whereas in this study all the maximal mean  
308 concentrations of the three principal ions appeared in autumn. The mean concentrations of the three  
309 ions in autumn in this study were in good agreement with the values reported by Yang et al. (2015).  
310 For the mass concentration ratios of  $\text{NO}_3^-/\text{SO}_4^{2-}$  (denoted as N/S), all the investigations exhibited  
311 relatively high values in autumn and spring, further confirming that the heterogeneous reaction of  
312  $\text{NO}_2$  on mineral dust favored nitrate formation (as discussed above). For NOR and SOR, all  
313 investigations were in good agreement, with the highest values in summer, the lowest in winter and  
314 higher values during haze days than during clean days. Compared with the investigations of 2003,  
315 the evident increase of both the concentration of  $\text{NO}_3^-$  and the ratio of N/S in recent years revealed  
316 the fast increase of vehicle numbers in the decade made significant contribution to atmospheric  
317 nitrate.

### 318 **4. Conclusions**

319 The comparison between the mass concentrations of WSIs measured by the filter method and the  
320 mass concentrations of  $\text{PM}_{2.5}$  measured by the TEOM 1405 Monitor revealed that the mass  
321 concentrations of WSIs could well reflect the pollution status of  $\text{PM}_{2.5}$  and the mass concentrations



322 of PM<sub>2.5</sub> measured by the TEOM 1405 Monitor were evidently underestimated due to the serious  
323 loss of volatile components in the atmospheric particulate matters.

324 The conspicuous daily fluctuation of the WSIs in each season confirmed that meteorological factors  
325 played an important role in governing the accumulation and dispersion of the pollutants. The  
326 extremely high concentrations of the WSIs during the serious pollution episodes indicated there  
327 were strong sources of the pollutants in Beijing. Based on the comprehensive analysis of the data of  
328 the WSIs, the strongly periodic activities of farmers, such as crop harvest, crop straw burning, and  
329 coal combustion for heating, were found to make evident contribution to the atmospheric WSIs in  
330 Beijing. To mitigate the currently serious pollution status in the NCP including Beijing, the strong  
331 emissions of pollutants from the periodic activities of farmers should be aroused great attention.

#### 332 **Author contribution**

333 **Y. J. Mu** designed the experiments and prepared the manuscript. **P. F. Liu** carried out the  
334 experiments and prepared the manuscript. **C. L. Zhang** carried out the experiments. **C. T. Liu, C.**  
335 **Y. Xue, C. Ye, J. F. Liu** and **Y. Y. Zhang** were involved in part of the work. **H. X. Zhang** provided  
336 the meteorological data and trace gases.

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#### 342 **References**

343 Brewer, P. G. (Eds.): Minor elements in sea water, Chemical Oceanography, Academic, San Diego,



- 344 California, 1975.
- 345 Buseck, P. R., and Posfai, M.: Airborne minerals and related aerosol particles: Effects on climate  
346 and the environment, Proceedings of the National Academy of Sciences of the United States of  
347 America, 96, 3372-3379, 10.1073/pnas.96.7.3372, 1999.
- 348 Chan, C. K., and Yao, X.: Air pollution in mega cities in China, Atmospheric Environment, 42, 1-  
349 42, 10.1016/j.atmosenv.2007.09.003, 2008.
- 350 Cheng, Y. F., Eichler, H., Wiedensohler, A., Heintzenberg, J., Zhang, Y. H., Hu, M., Herrmann, H.,  
351 Zeng, L. M., Liu, S., Gnauk, T., Brüggemann, E., and He, L. Y.: Mixing state of elemental carbon  
352 and non-light-absorbing aerosol components derived from in situ particle optical properties at  
353 Xinken in Pearl River Delta of China, Journal of Geophysical Research, 111,  
354 10.1029/2005jd006929, 2006.
- 355 Finlayson-Pitts, B. J., Barbara J.: Atmospheric Chemistry, Library of Congress, Canada, 1986.
- 356 Finlayson-Pitts, B. J., Pitts, J. N. (Eds.): Chemistry of the upper and lower atmosphere, Academic  
357 Press, San Diego, 2000.
- 358 Gao, X., Yang, L., Cheng, S., Gao, R., Zhou, Y., Xue, L., Shou, Y., Wang, J., Wang, X., Nie, W., Xu,  
359 P., and Wang, W.: Semi-continuous measurement of water-soluble ions in PM<sub>2.5</sub> in Jinan, China:  
360 Temporal variations and source apportionments, Atmospheric Environment, 45, 6048-6056,  
361 10.1016/j.atmosenv.2011.07.041, 2011.
- 362 Grover, B. D., Kleinman, M., Eatough, N. L., Eatough, D. J., Hopke, P. K., Long, R. W., Wilson, W.  
363 E., Meyer, M. B., and Ambs, J. L.: Measurement of total PM<sub>2.5</sub> mass (nonvolatile plus semivolatile)  
364 with the Filter Dynamic Measurement System tapered element oscillating microbalance monitor,  
365 Journal of Geophysical Research, 110, 10.1029/2004jd004995, 2005.
- 366 Guo, S., Hu, M., Guo, Q., Zhang, X., Zheng, M., Zheng, J., Chang, C. C., Schauer, J. J., and Zhang,  
367 R.: Primary sources and secondary formation of organic aerosols in Beijing, China, Environ Sci  
368 Technol, 46, 9846-9853, 10.1021/es2042564, 2012.
- 369 Guo, S., Hu, M., Zamora, M. L., Peng, J., Shang, D., Zheng, J., Du, Z., Wu, Z., Shao, M., Zeng, L.,  
370 Molina, M. J., and Zhang, R.: Elucidating severe urban haze formation in China, Proceedings of the  
371 National Academy of Sciences of the United States of America, 111, 17373-17378,  
372 10.1073/pnas.1419604111, 2014.
- 373 He, H., Wang, Y., Ma, Q., Ma, J., Chu, B., Ji, D., Tang, G., Liu, C., Zhang, H., and Hao, J.: Mineral  
374 dust and NO<sub>x</sub> promote the conversion of SO<sub>2</sub> to sulfate in heavy pollution days, Scientific reports,  
375 4, 4172, 10.1038/srep04172, 2014.
- 376 Hu, G., Zhang, Y., Sun, J., Zhang, L., Shen, X., Lin, W., and Yang, Y.: Variability, formation and  
377 acidity of water-soluble ions in PM<sub>2.5</sub> in Beijing based on the semi-continuous observations,  
378 Atmospheric Research, 145-146, 1-11, 10.1016/j.atmosres.2014.03.014, 2014.
- 379 Huang, R. J., Zhang, Y., Bozzetti, C., Ho, K. F., Cao, J. J., Han, Y., Daellenbach, K. R., Slowik, J.  
380 G., Platt, S. M., Canonaco, F., Zotter, P., Wolf, R., Pieber, S. M., Bruns, E. A., Crippa, M., Ciarelli,  
381 G., Piazzalunga, A., Schwikowski, M., Abbaszade, G., Schnelle-Kreis, J., Zimmermann, R., An, Z.,  
382 Szidat, S., Baltensperger, U., El Haddad, I., and Prevot, A. S.: High secondary aerosol contribution  
383 to particulate pollution during haze events in China, Nature, 514, 218-222, 10.1038/nature13774,  
384 2014.
- 385 Huang, W., Bi, X., Zhang, G., Huang, B., Lin, Q., Wang, X., Sheng, G., and Fu, J.: The chemical  
386 composition and stable carbon isotope characteristics of particulate matter from the residential  
387 honeycomb coal briquettes combustion, Geochimica, 43, 640-640, 2014.





- 388 Huang, X., Song, Y., Li, M., Li, J., and Zhu, T.: Harvest season, high polluted season in East China,  
389 Environmental Research Letters, 7, 10.1088/1748-9326/7/4/044033, 2012.
- 390 Huang, X., Liu, Z., Zhang, J., Wen, T., Ji, D., and Wang, Y.: Seasonal variation and secondary  
391 formation of size-segregated aerosol water-soluble inorganic ions during pollution episodes in  
392 Beijing, Atmospheric Research, 168, 70-79, 10.1016/j.atmosres.2015.08.021, 2016.
- 393 Jiang, Q., Sun, Y. L., Wang, Z., and Yin, Y.: Aerosol composition and sources during the Chinese  
394 Spring Festival: fireworks, secondary aerosol, and holiday effects, Atmospheric Chemistry and  
395 Physics, 15, 6023-6034, 10.5194/acp-15-6023-2015, 2015.
- 396 Kong, S. F., Li, L., Li, X. X., Yin, Y., Chen, K., Liu, D. T., Yuan, L., Zhang, Y. J., Shan, Y. P., and  
397 Ji, Y. Q.: The impacts of firework burning at the Chinese Spring Festival on air quality: insights of  
398 tracers, source evolution and aging processes, Atmospheric Chemistry and Physics, 15, 2167-2184,  
399 10.5194/acp-15-2167-2015, 2015.
- 400 Krupa, S. V.: Effects of atmospheric ammonia (NH<sub>3</sub>) on terrestrial vegetation: a review,  
401 Environmental Pollution, 124, 179-221, 10.1016/s0269-7491(02)00434-7, 2003.
- 402 Li, J., Song, Y., Mao, Y., Mao, Z., Wu, Y., Li, M., Huang, X., He, Q., and Hu, M.: Chemical  
403 characteristics and source apportionment of PM<sub>2.5</sub> during the harvest season in eastern China's  
404 agricultural regions, Atmospheric Environment, 92, 442-448, 10.1016/j.atmosenv.2014.04.058,  
405 2014.
- 406 Li, W., Zhou, S., Wang, X., Xu, Z., Yuan, C., Yu, Y., Zhang, Q., and Wang, W.: Integrated evaluation  
407 of aerosols from regional brown hazes over northern China in winter: Concentrations, sources,  
408 transformation, and mixing states, Journal of Geophysical Research, 116, 10.1029/2010jd015099,  
409 2011.
- 410 Li, X., Wang, L., Ji, D., Wen, T., Pan, Y., Sun, Y., and Wang, Y.: Characterization of the size-  
411 segregated water-soluble inorganic ions in the Jing-Jin-Ji urban agglomeration: Spatial/temporal  
412 variability, size distribution and sources, Atmospheric Environment, 77, 250-259,  
413 10.1016/j.atmosenv.2013.03.042, 2013.
- 414 Liang, C. S., Duan, F. K., He, K. B., and Ma, Y. L.: Review on recent progress in observations,  
415 source identifications and countermeasures of PM<sub>2.5</sub>, Environ Int, 86, 150-170,  
416 10.1016/j.envint.2015.10.016, 2016.
- 417 Liu, C.N., Lin, S. F., Awasthi, A., Tsai, C.-J., Wu, Y.-C., and Chen, C.-F.: Sampling and conditioning  
418 artifacts of PM<sub>2.5</sub> in filter-based samplers, Atmospheric Environment, 85, 48-53,  
419 10.1016/j.atmosenv.2013.11.075, 2014.
- 420 Liu, T. Y., Wang, X. M., Wang, B. G., Ding, X., Deng, W., Lu, S. J., and Zhang, Y. L.: Emission  
421 factor of ammonia (NH<sub>3</sub>) from on-road vehicles in China: tunnel tests in urban Guangzhou,  
422 Environmental Research Letters, 9, 8, 10.1088/1748-9326/9/6/064027, 2014.
- 423 Liu, X. G., Li, J., Qu, Y., Han, T., Hou, L., Gu, J., Chen, C., Yang, Y., Liu, X., Yang, T., Zhang, Y.,  
424 Tian, H., and Hu, M.: Formation and evolution mechanism of regional haze: a case study in the  
425 megacity Beijing, China, Atmospheric Chemistry and Physics, 13, 4501-4514, 10.5194/acp-13-  
426 4501-2013, 2013.
- 427 Nel, A.: Air pollution-related illness: Effects of particles, Science, 308, 804-806,  
428 10.1126/science.1108752, 2005.
- 429 Nie, W., Ding, A., Wang, T., Kerminen, V. M., George, C., Xue, L., Wang, W., Zhang, Q., Petaja, T.,  
430 Qi, X., Gao, X., Wang, X., Yang, X., Fu, C., and Kulmala, M.: Polluted dust promotes new particle  
431 formation and growth, Scientific reports, 4, 6634, 10.1038/srep06634, 2014.



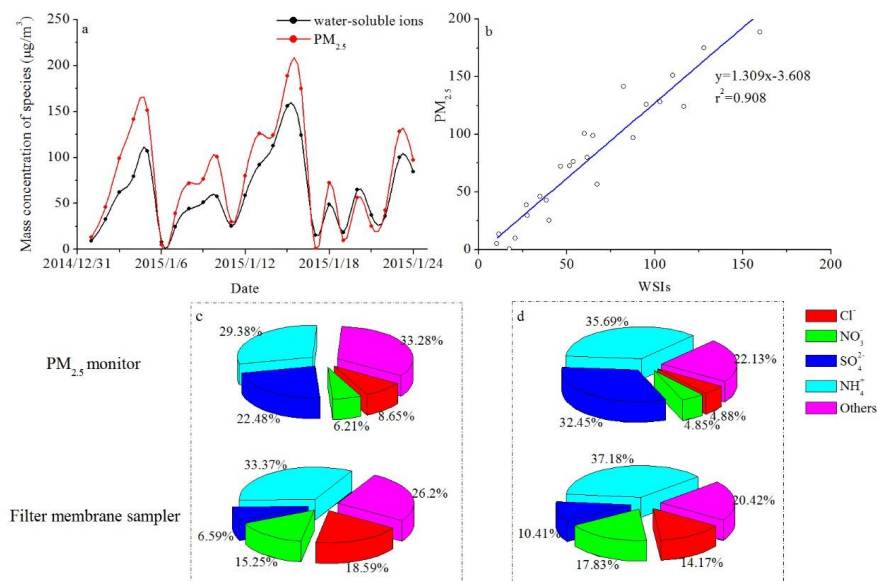
- 432 Pang, X. B., and Mu, Y. J.: Seasonal and diurnal variations of carbonyl compounds in Beijing  
433 ambient air, *Atmospheric Environment*, 40, 6313-6320, 10.1016/j.atmosenv.2006.05.044, 2006.
- 434 Pathak, R. K., Wu, W. S., and Wang, T.: Summertime PM<sub>2.5</sub> ionic species in four major cities of  
435 China: nitrate formation in an ammonia-deficient atmosphere, *Atmospheric Chemistry and Physics*,  
436 9, 1711-1722, 2009.
- 437 Peplow, M.: Beijing smog contains witches' brew of microbes, *Nature*, doi, 10, 2014.
- 438 Poschl, U.: Atmospheric aerosols: composition, transformation, climate and health effects, *Angew*  
439 *Chem Int Ed Engl*, 44, 7520-7540, 10.1002/anie.200501122, 2005.
- 440 Quan, J., Tie, X., Zhang, Q., Liu, Q., Li, X., Gao, Y., and Zhao, D.: Characteristics of heavy aerosol  
441 pollution during the 2012–2013 winter in Beijing, China, *Atmospheric Environment*, 88, 83-89,  
442 10.1016/j.atmosenv.2014.01.058, 2014.
- 443 Seinfeld, J. H., Pandis, S. N. (Eds.): *Atmospheric chemistry and physics*, Wiley, New York, 1998.
- 444 Shen, Z., Cao, J., Arimoto, R., Han, Z., Zhang, R., Han, Y., Liu, S., Okuda, T., Nakao, S., and Tanaka,  
445 S.: Ionic composition of TSP and PM<sub>2.5</sub> during dust storms and air pollution episodes at Xi'an, China,  
446 *Atmospheric Environment*, 43, 2911-2918, 10.1016/j.atmosenv.2009.03.005, 2009.
- 447 Souza, D. Z., Vasconcellos, P. C., Lee, H., Aurela, M., Saarnio, K., Teinila, K., and Hillamo, R.:  
448 Composition of PM<sub>2.5</sub> and PM<sub>10</sub> Collected at Urban Sites in Brazil, *Aerosol and Air Quality*  
449 *Research*, 14, 168-176, 10.4209/aaqr.2013.03.0071, 2014.
- 450 Sun, Y., Zhuang, G., Tang, A., Wang, Y., and An, Z.: Chemical Characteristics of PM<sub>2.5</sub> and PM<sub>10</sub>  
451 in Haze-Fog Episodes in Beijing, *Environ. Sci. Technol.*, 40, 3148-3155, 2006.
- 452 Sun, Y. L., Zhuang, G. S., Ying, W., Han, L. H., Guo, J. H., Mo, D., Zhang, W. J., Wang, Z. F., and  
453 Hao, Z. P.: The air-borne particulate pollution in Beijing - concentration, composition, distribution  
454 and sources, *Atmospheric Environment*, 38, 5991-6004, 10.1016/j.atmosenv.2004.07.009, 2004.
- 455 Sun, Y. L., Wang, Z. F., Fu, P. Q., Yang, T., Jiang, Q., Dong, H. B., Li, J., and Jia, J. J.: Aerosol  
456 composition, sources and processes during wintertime in Beijing, China, *Atmospheric Chemistry*  
457 *and Physics*, 13, 4577-4592, 10.5194/acp-13-4577-2013, 2013.
- 458 Tian, S., Pan, Y., Liu, Z., Wen, T., and Wang, Y.: Size-resolved aerosol chemical analysis of extreme  
459 haze pollution events during early 2013 in urban Beijing, China, *J Hazard Mater*, 279, 452-460,  
460 10.1016/j.jhazmat.2014.07.023, 2014.
- 461 Wang, G., Cheng, S., Li, J., Lang, J., Wen, W., Yang, X., and Tian, L.: Source apportionment and  
462 seasonal variation of PM<sub>2.5</sub> carbonaceous aerosol in the Beijing-Tianjin-Hebei region of China,  
463 *Environ Monit Assess*, 187, 143, 10.1007/s10661-015-4288-x, 2015.
- 464 Wang, H., Tan, S.-C., Wang, Y., Jiang, C., Shi, G.-y., Zhang, M.-X., and Che, H.-Z.: A multisource  
465 observation study of the severe prolonged regional haze episode over eastern China in January 2013,  
466 *Atmospheric Environment*, 89, 807-815, 10.1016/j.atmosenv.2014.03.004, 2014.
- 467 Wang, Y., Zhuang, G. S., Tang, A. H., Yuan, H., Sun, Y. L., Chen, S. A., and Zheng, A. H.: The ion  
468 chemistry and the source of PM<sub>2.5</sub> aerosol in Beijing, *Atmospheric Environment*, 39, 3771-3784,  
469 10.1016/j.atmosenv.2005.03.013, 2005.
- 470 Wang, Y., Yao, L., Wang, L., Liu, Z., Ji, D., Tang, G., Zhang, J., Sun, Y., Hu, B., and Xin, J.:  
471 Mechanism for the formation of the January 2013 heavy haze pollution episode over central and  
472 eastern China, *Science China Earth Sciences*, 57, 14-25, 10.1007/s11430-013-4773-4, 2013.
- 473 Wu, S., Deng, F., Wei, H., Huang, J., Wang, X., Hao, Y., Zheng, C., Qin, Y., Lv, H., Shima, M., and  
474 Guo, X.: Association of cardiopulmonary health effects with source-appointed ambient fine  
475 particulate in Beijing, China: a combined analysis from the Healthy Volunteer Natural Relocation



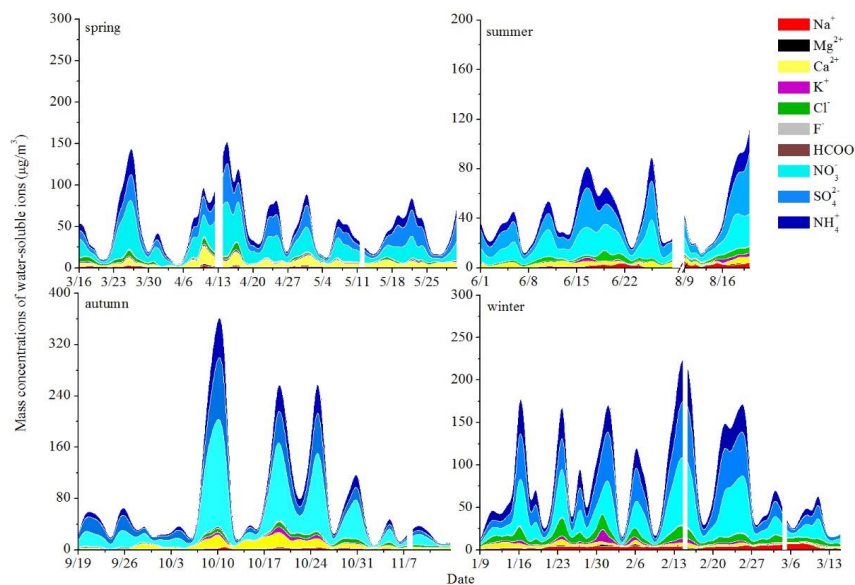
- 476 (HVNR) study, *Environ Sci Technol*, 48, 3438-3448, 10.1021/es404778w, 2014.
- 477 Xu, S. S., Liu, W. X., and Tao, S.: Emission of polycyclic aromatic hydrocarbons in China, *Environ.*  
478 *Sci. Technol.*, 40, 702-708, 10.1021/es0517062, 2006.
- 479 Yang, Y., Zhou, R., Wu, J., Yu, Y., Ma, Z., Zhang, L., and Di, Y.: Seasonal variations and size  
480 distributions of water-soluble ions in atmospheric aerosols in Beijing, 2012, *J Environ Sci (China)*,  
481 34, 197-205, 10.1016/j.jes.2015.01.025, 2015a.
- 482 Yang, Y. R., Liu, X. G., Qu, Y., An, J. L., Jiang, R., Zhang, Y. H., Sun, Y. L., Wu, Z. J., Zhang, F.,  
483 Xu, W. Q., and Ma, Q. X.: Characteristics and formation mechanism of continuous hazes in China:  
484 a case study during the autumn of 2014 in the North China Plain, *Atmospheric Chemistry and*  
485 *Physics*, 15, 8165-8178, 10.5194/acp-15-8165-2015, 2015b.
- 486 Yu, L. D., Wang, G. F., Zhang, R. J., Zhang, L. M., Song, Y., Wu, B. B., Li, X. F., An, K., and Chu,  
487 J. H.: Characterization and Source Apportionment of PM<sub>2.5</sub> in an Urban Environment in Beijing,  
488 *Aerosol and Air Quality Research*, 13, 574-583, 10.4209/aaqr.2012.07.0192, 2013.
- 489 Zhang, J. K., Sun, Y., Liu, Z. R., Ji, D. S., Hu, B., Liu, Q., and Wang, Y. S.: Characterization of  
490 submicron aerosols during a month of serious pollution in Beijing, 2013, *Atmospheric Chemistry*  
491 *and Physics*, 14, 2887-2903, 10.5194/acp-14-2887-2014, 2014.
- 492 Zhang, L., Wang, T., Lv, M., and Zhang, Q.: On the severe haze in Beijing during January 2013:  
493 Unraveling the effects of meteorological anomalies with WRF-Chem, *Atmospheric Environment*,  
494 104, 11-21, 10.1016/j.atmosenv.2015.01.001, 2015a.
- 495 Zhang, R., Jing, J., Tao, J., Hsu, S. C., Wang, G., Cao, J., Lee, C. S. L., Zhu, L., Chen, Z., Zhao, Y.,  
496 and Shen, Z.: Chemical characterization and source apportionment of PM<sub>2.5</sub> in Beijing: seasonal  
497 perspective, *Atmospheric Chemistry and Physics*, 13, 7053-7074, 10.5194/acp-13-7053-2013, 2013.
- 498 Zhang, Y., Liu, J., Mu, Y., Pei, S., Lun, X., and Chai, F.: Emissions of nitrous oxide, nitrogen oxides  
499 and ammonia from a maize field in the North China Plain, *Atmospheric Environment*, 45, 2956-  
500 2961, 10.1016/j.atmosenv.2010.10.052, 2011.
- 501 Zhang, Y. L., Huang, R. J., El Haddad, I., Ho, K. F., Cao, J. J., Han, Y., Zotter, P., Bozzetti, C.,  
502 Daellenbach, K. R., Canonaco, F., Slowik, J. G., Salazar, G., Schwikowski, M., Schnelle-Kreis, J.,  
503 Abbaszade, G., Zimmermann, R., Baltensperger, U., Prévôt, A. S. H., and Szidat, S.: Fossil vs. non-  
504 fossil sources of fine carbonaceous aerosols in four Chinese cities during the extreme winter haze  
505 episode of 2013, *Atmospheric Chemistry and Physics*, 15, 1299-1312, 10.5194/acp-15-1299-2015,  
506 2015b.
- 507 Zhao, P. S., Dong, F., He, D., Zhao, X. J., Zhang, X. L., Zhang, W. Z., Yao, Q., and Liu, H. Y.:  
508 Characteristics of concentrations and chemical compositions for PM<sub>2.5</sub> in the region of Beijing,  
509 Tianjin, and Hebei, China, *Atmospheric Chemistry and Physics*, 13, 4631-4644, 10.5194/acp-13-  
510 4631-2013, 2013a.
- 511 Zhao, P. S., Dong, F., Yang, Y. D., He, D., Zhao, X. J., Zhang, W. Z., Yao, Q., and Liu, H. Y.:  
512 Characteristics of carbonaceous aerosol in the region of Beijing, Tianjin, and Hebei, China,  
513 *Atmospheric Environment*, 71, 389-398, 10.1016/j.atmosenv.2013.02.010, 2013b.
- 514 Zhao, X. J., Zhao, P. S., Xu, J., Meng, W., Pu, W. W., Dong, F., He, D., and Shi, Q. F.: Analysis of  
515 a winter regional haze event and its formation mechanism in the North China Plain, *Atmospheric*  
516 *Chemistry and Physics*, 13, 5685-5696, 10.5194/acp-13-5685-2013, 2013c.
- 517 Zheng, B., Zhang, Q., Zhang, Y., He, K. B., Wang, K., Zheng, G. J., Duan, F. K., Ma, Y. L., and  
518 Kimoto, T.: Heterogeneous chemistry: a mechanism missing in current models to explain secondary  
519 inorganic aerosol formation during the January 2013 haze episode in North China, *Atmospheric*



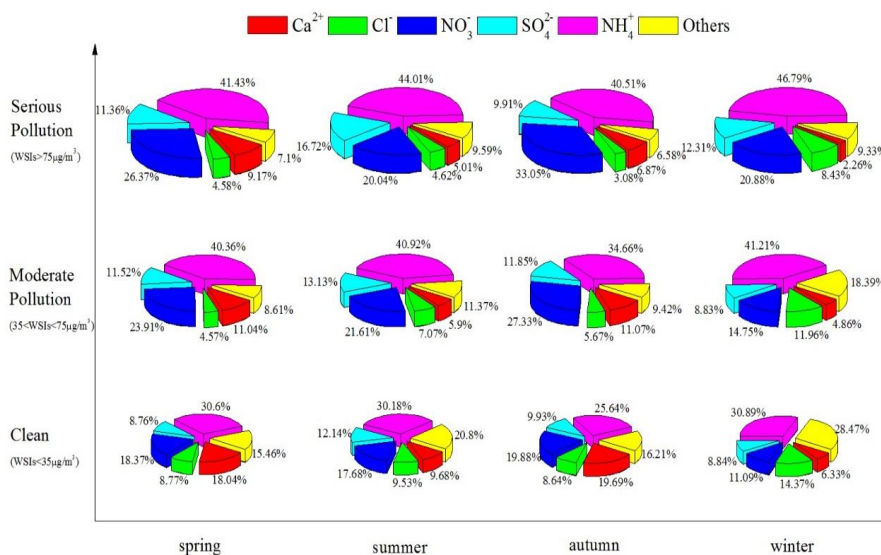
520 Chemistry and Physics, 15, 2031-2049, 10.5194/acp-15-2031-2015, 2015a.  
521 Zheng, G. J., Duan, F. K., Su, H., Ma, Y. L., Cheng, Y., Zheng, B., Zhang, Q., Huang, T., Kimoto,  
522 T., Chang, D., Pöschl, U., Cheng, Y. F., and He, K. B.: Exploring the severe winter haze in Beijing:  
523 the impact of synoptic weather, regional transport and heterogeneous reactions, Atmospheric  
524 Chemistry and Physics, 15, 2969-2983, 10.5194/acp-15-2969-2015, 2015b.  
525  
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 547 **Fig. 1** Comparison between the filter sampling method and the PM<sub>2.5</sub> monitor for the daily average mass  
 548 concentrations of the WSIs and PM<sub>2.5</sub> (Fig. 1a and 1b), and for the 12-day-average molar composition of the WSIs  
 549 on the filters collected by the two methods during the two 12-day sampling periods (Fig. 1c represents the data  
 550 collected during the first 12-day; Fig. 1d represents the data collected during the second 12-day.).

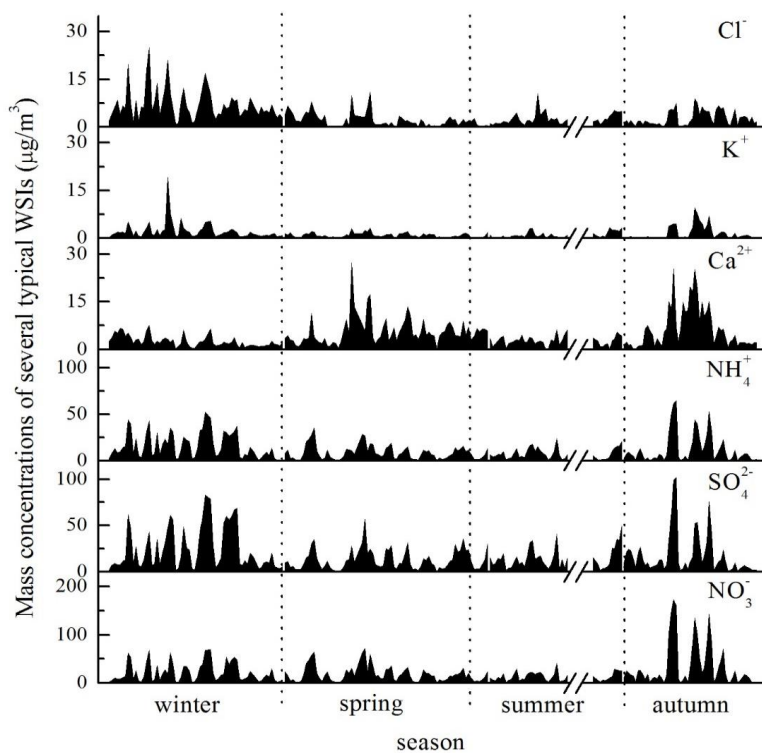


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 552 **Fig. 2** Daily variations of WSIs in each season (the smooth lines for the WSIs were drawn between the points of  
 553 the daily data).



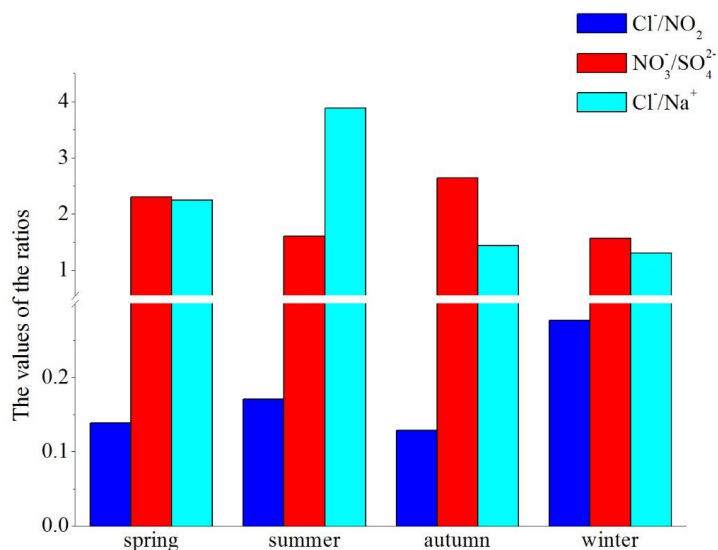
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**Fig. 3** Molar composition of the WSIs under different pollution levels in four seasons (Clean: WSIs < 35 µg m<sup>-3</sup>; Moderate pollution: 35 µg m<sup>-3</sup> < WSIs < 75 µg m<sup>-3</sup>; Serious pollution: WSIs > 75 µg m<sup>-3</sup>)



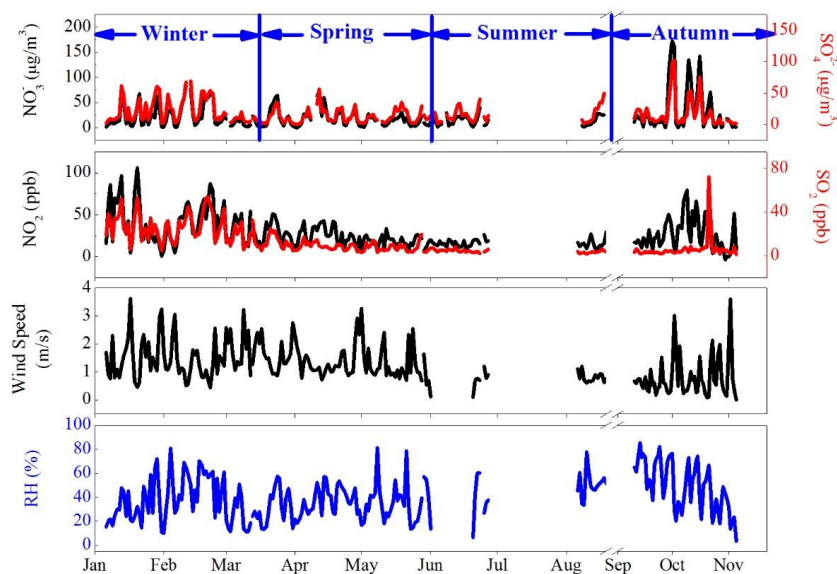
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**Fig. 4** The seasonal variations of the several typical WSIs in the year of 2014



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560 **Fig. 5** the average ratio of Cl<sup>-</sup>/NO<sub>2</sub> (the unit is μg/m<sup>3</sup> and ppb, respectively) and the average molar ratios of Cl<sup>-</sup>/  
 561 /Na<sup>+</sup> and NO<sub>3</sub><sup>-</sup>/SO<sub>4</sub><sup>2-</sup> in each season.



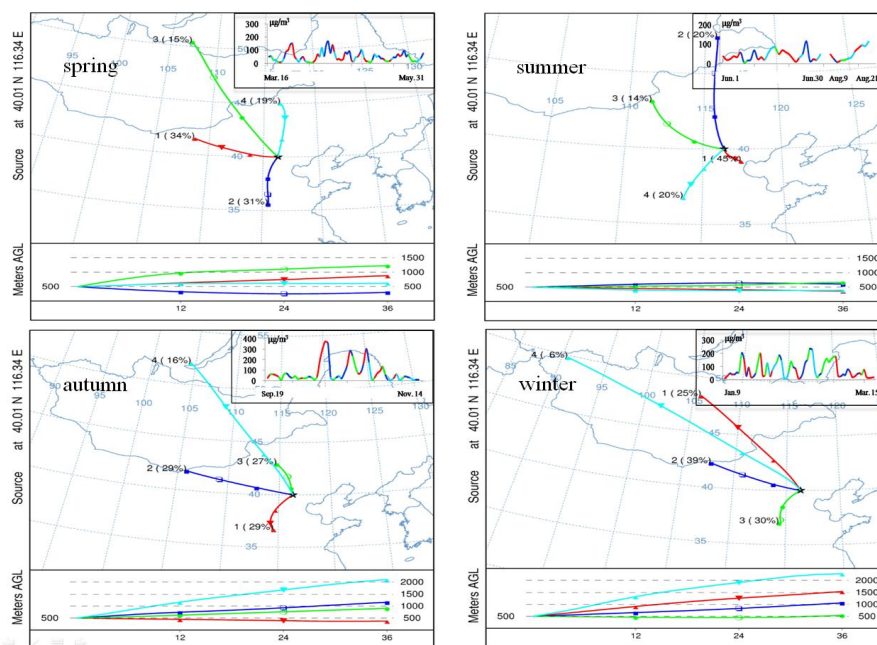
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563 **Fig. 6** Time series of NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, NO<sub>2</sub> and SO<sub>2</sub> and meteorological data (wind speed and relative humidity) in four  
 564 seasons for 2014

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568 **Fig. 7** The back trajectory cluster analysis and the corresponding overall ion mass concentration in four seasons

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573 **Table 1** Concentrations ( $\mu\text{g m}^{-3}$ ) of the WSIs (mean concentrations and standard deviation (SD)) in four seasons in  
 574 Beijing.

Species	Spring (N=74)		Summer (N=41)		Autumn (N=56)		Winter (N=64)		Annual (N=235)	
	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
F <sup>-</sup>	0.3	0.3	0.2	0.1	0.4	0.2	0.2	0.2	0.3	0.2
HCOO <sup>-</sup>	0.2	0.1	0.2	0.1	0.4	0.5	0.3	0.2	0.3	0.3
Cl <sup>-</sup>	2.4	2.2	2.6	1.9	2.8	2.3	7.0	4.9	3.9	3.7
NO <sub>3</sub> <sup>-</sup>	18.4	16.0	13.4	9.3	34.3	45.2	23.8	22.8	22.8	27.7
SO <sub>4</sub> <sup>2-</sup>	13.0	10.9	14.6	11.6	18.1	22.8	22.2	19.6	17.0	17.3
Na <sup>+</sup>	1.2	0.8	2.1	1.4	1.6	1.1	3.8	1.7	2.3	1.8
NH <sub>4</sub> <sup>+</sup>	8.8	7.4	7.6	6.0	12.3	16.3	16.5	13.6	11.5	12.2
Mg <sup>2+</sup>	0.5	0.4	0.3	0.2	0.4	0.3	0.5	0.5	0.4	0.4
Ca <sup>2+</sup>	5.6	4.2	2.9	1.5	6.8	6.4	2.6	1.8	4.6	4.4
K <sup>+</sup>	1.0	0.7	1.1	1.0	1.6	2.2	2.2	2.7	1.5	1.9
Mass	50.5	37.3	44.2	28.9	78.3	92.6	78.7	61.2	63.7	62.0

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577 **Table 2** SOR and NOR during haze days and non-haze days in four seasons.

	Spring		Summer		Autumn		Winter	
	SOR	NOR	SOR	NOR	SOR	NOR	SOR	NOR
Haze days	0.3	0.3	0.7	0.4	0.6	0.4	0.2	0.3
Non-haze days	0.2	0.2	0.3	0.2	0.3	0.2	0.1	0.1
Ratio	1.8	1.8	2.0	2.3	2.0	2.6	2.3	2.5

578 The ratio of values in Haze days to that in Non-haze days.



**Table 3** Summary of these principal ions ( $\mu\text{g m}^{-3}$ ), the mass concentration ratio of  $\text{NO}_3^-/\text{SO}_4^{2-}$  (denoted as N/S), NOR and SOR for four seasons in Beijing.

Year	Spring										Summer										Autumn										Winter										Reference
	$\text{NO}_3^-$	$\text{SO}_4^{2-}$	$\text{NH}_4^+$	N/S <sup>+</sup>	NOR	SOR	$\text{NO}_3^-$	$\text{SO}_4^{2-}$	$\text{NH}_4^+$	N/S	NOR	SOR	$\text{NO}_3^-$	$\text{SO}_4^{2-}$	$\text{NH}_4^+$	N/S	NOR	SOR	$\text{NO}_3^-$	$\text{SO}_4^{2-}$	$\text{NH}_4^+$	N/S	NOR	SOR	$\text{NO}_3^-$	$\text{SO}_4^{2-}$	$\text{NH}_4^+$	N/S	NOR	SOR											
2014	18.4	13.0	8.8	1.4	0.2	0.2	13.4	14.6	7.6	0.9	0.2	0.4	34.3	18.1	12.3	1.9	0.2	0.4	23.8	22.2	16.5	1.1	0.2	0.2	23.8	22.2	16.5	1.1	0.2	0.2	This work										
2014(haze)	30.2	21.6	14.5	1.4	0.3	0.3	25.0	28.8	15.3	0.9	0.4	0.7	73.6	36.0	26.5	2.0	0.4	0.6	37.7	34.5	25.4	1.1	0.3	0.2	37.7	34.5	25.4	1.1	0.3	0.2	This work										
2014(clean)	7.8	5.2	3.5	1.5	0.2	0.2	8.6	8.7	4.4	1.0	0.2	0.3	8.9	6.5	3.2	1.4	0.2	0.3	5.9	6.4	4.5	0.9	0.1	0.1	5.9	6.4	4.5	0.9	0.1	0.1	This work										
2014													35.5	20.0	16.7	1.8	0.2	0.4													Yang et al., 2015b										
2013-2014(haze)	14.7	9.0	10.3	1.6	0.2	0.4	33.9	32.7	24.0	1.0	0.3	0.7	40.0	17.4	22.2	2.3	0.2	0.6	22.0	20.4	18.8	1.1	0.2	0.2	22.0	20.4	18.8	1.1	0.2	0.2	Huang et al., 2016										
2013-2014(clean)	3.6	2.4	4.4	1.5	0.1	0.2	8.8	8.1	11.7	1.1	0.1	0.4	5.5	4.5	5.6	1.2	0.1	0.4	6.6	5.2	6.0	1.3	0.1	0.1	6.6	5.2	6.0	1.3	0.1	0.1	Huang et al., 2016										
2013(haze)	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	26.1	33.3	24.1	0.8	-	-	26.1	33.3	24.1	0.8	-	-	Tian et al., 2014										
2013(clean)	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	4.9	5.0	4.9	1.0	-	-	4.9	5.0	4.9	1.0	-	-	Tian et al., 2014										
2010(clean)	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Zhao et al., 2013a										
2010(haze)	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Zhao et al., 2013a										
2009-2010	15.5	14.7	7.5	1.1	-	-	11.8	23.5	11.0	0.5	-	-	10.7	7.9	4.7	1.4	-	-	7.3	8.5	4.5	0.9	-	-	7.3	8.5	4.5	0.9	-	-	Zhang et al., 2013										
2009	-	-	-	-	-	-	12.7	26.1	9.1	0.5	0.2	0.7	6.1	20.1	4.3	0.3	0.1	0.6	-	-	-	-	-	-	-	-	-	-	-	-	Hu et al., 2014										
2005	-	-	-	-	-	-	9.9	22.6	4.7	0.4	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Pathak et al., 2009										
2001-2003	11.9	13.5	6.5	0.9	0.1	0.1	11.2	18.4	10.1	0.6	0.1	0.4	9.1	12.7	6.3	0.7	0.1	0.2	12.3	21.0	10.6	0.6	0.1	0.1	12.3	21.0	10.6	0.6	0.1	0.1	Wang et al., 2005										
2002-2003	-	-	-	-	-	-	12.2	16.0	10.4	0.8	-	-	-	-	-	-	-	-	17.0	30.4	12.9	0.6	-	-	17.0	30.4	12.9	0.6	-	-	Sun et al., 2004										