We thank the reviewer for the effort to review the manuscript and to provide constructive comments which help to improve the manuscript. Our replies to the comments and our actions taken to revise the paper (in blue) are given below (the original comments are copied here in *Italic*).

Anonymous Referee #2

The authors implement a temperature-dependent organic nucleation parameterization in a global model. Simulations are compared with in situ observations. A climate forcing is calculated. The modification of the parameterisation and the in-situ observation comparison are interesting and useful, but the climate forcing calculation is questionable.

The main problem is that (looking at Yu et al. 2015), there seems to ONLY be organics mediated nucleation in this model set-up. This does not invalidate all of the findings, but one of the main outcomes of the CERN CLOUD experiment is that we know there are many different types of nucleation going on at the same time, and so models which only include one type of nucleation will seem more sensitive to changes than the real atmosphere. Figure 5 shows that the temperature dependent parameterisation does give very good agreement with observations, and I believe the temperature dependence is useful and valid in this sense. But the idea that 40% of boundary layer CCN would be lost to a temperature dependence in the organic nucleation rate only makes sense if no other nucleation is happening in the model, which is very unrealistic for the atmosphere. This also calls into question the quoted radiative forcings (radiative effects, since they do not relate to the pre-industrial?) shown in Figure 6.

Simulations which include other nucleation types should be performed to give a more accurate estimate of the climate estimates, which I am sure will be smaller than those quoted in the current version of the manuscript.

The main focus of this study is to develop a modification of organic nucleation parameterization and investigate the impact of temperature on H2SO4-Organics nucleation rates. We agree with the reviewer about the uncertainties in the radiative forcing (RF) estimates. To address the referee's concern and to follow the referee's suggestion (see below) to place the emphasis on "the realistic improvements in nucleation representation shown in Figure 5, rather than on unrealistically high model sensitivity to an overly simplistic nucleation scheme", we have deleted the RF figure (Figure 6b) and associated text in the revised manuscript.

Following the suggestion of the other reviewer, we have added two new figures (Fig. 4 and Fig. 6 in revised manuscript) and associated discussions to explore the sensitivity of results to ΔH values and to expand comparisons with observations.

Was there any physical basis for setting the maximum value of fT to 10? If the method is valid for higher temperatures, why is it not valid for lower temperatures? If the method gives unrealistic values at low temperatures, why do we trust it at high ones?

The physical basis for setting the maximum value of f_T is that the organics mediated nucleation rates should not exceed the 3-body kinetic collision rate for forming a cluster containing two

 H_2SO_4 molecules and one BioOxOrg molecule (k_{max}). k_{max} depends on T as well as the mass and sizes of colliding molecules. At T=270 K, k_{max} is about a factor of 38 higher than k_m . We have revised the manuscript to set the maximum value of f_T to k_{max}/k_m instead of 10. The modification has negligible effect on the results and conclusions of this study which focuses on the summer month and the boundary layer. The method is valid for both high and low temperature.

It says in Section 2.2 that this is the "first global modeling attempt in studying the effect of temperature on organics-mediated nucleation in the atmosphere", and this assertion is repeated in the summary and discussion section. Section 20 of supplementary materials and figure S9 in Dunne et al. (2016)'s Nature paper on nucleation based on CERN CLOUD experiment contains a temperature dependence of organic nucleation. It was not the main focus of the paper but it should be mentioned.

We have modified Section 2.2 and Discussion to remove "first". The work of Dunne et al. (2016) is now mentioned in the discussion.

With revisions, the paper will make a good contribution to the field and should be published. But the emphasis should be placed on the realistic improvements in nucleation representation shown in Figure 5, rather than on unrealistically high model sensitivity to an overly simplistic nucleation scheme.

Please see our reply above.

Anonymous Referee #3

Review of Yu et al. 2016, "Impact of temperature dependence on the possible contribution of organics to new particle formation in the atmosphere" The authors explore temperature dependence effects on organic-sulfuric acid new particle formation in a chemical transport model with aerosol microphysics. This is an interesting and concise paper that addresses a significant missing piece (T-dependence) of current new particle formation research. However, I have a few major comments that need to be addressed before I can recommend this paper for publication. Once these are addressed, I believe the paper will be stronger and more impactful, and will be ready for publication.

Major comments:

1. There is a lot of uncertainty in the Nucl-OrgT parameterization. The authors do acknowledge this, discussing the uncertainty in the chemical identity of the stable cluster molecule and the subsequent uncertainty of the ΔH and fT values. The authors mention explicitly on page 8 around line 150, suggesting that a change of 5 kcal/mol would result in a 40% difference in fT at 288 K, whereas a 20 kcal/mol change results in a factor of 3.5 difference. Although the acknowledgement of this uncertainty is helpful, why not actually implement this into the simulations? Since the uncertainty in the ΔH values is that large, I would encourage them to explore the sensitivity of their results (nucleation rates, CCN formation, etc) to using different values of ΔH . It would be very interesting and useful to the community to have a range (low ΔH , high ΔH) of results for Tdependent organic-sulfuric acid nucleation. This is a very good suggestion. We have carried out further simulations to explore the sensitivity of nucleation rates, CN10, and CCN concentrations to Δ H. A figure showing the results along with the associated discussions has been added to the revised manuscript.

2. The authors conclude that the temperature-dependent Nucl-OrgT parameterization is "likely" more realistic than Nucl-Org, which does not account for temperature. While it makes sense that adding something so fundamental to nucleation as temperature would improve a parameterization, this is not automatically the case. What I find more concerning is that I do not think the authors have appropriately justified this conclusion that Nucl-OrgT is better with the results presented in this paper. Besides the qualitative argument I mention above, the only evidence in this paper we have comes from Fig 5, where _10 nucleation events at one location (Duke Forest) in spring and summer are used to validate the model. Figure 4 and its associated text mention that the Nucl-OrgT does better against observations but there are no observations actually plotted in Fig. 4! Another paper is referenced, but to make this conclusion for this paper, the comparison needs to be explicitly presented. Thus, I suggest the authors do show this observational data and present statistics on how it compares to the Nucl-OrgT simulation, for more than only the Duke Forest site.

In the paragraph discussing Figure 4, we pointed out that "the previous comparisons of simulated and observed particle size distributions measured in nine forest areas in North America (NA) (Yu et al., 2005) showed that $J_{Nucl-Org}$ parameterization (Eq. 1) over-predicts particle number concentrations at these sites in summer by a factor of around two on average (Yu et al., 2005)" and "the simulated monthly mean CN10 values in the NA boundary layer based on $J_{Nucl-OrgT}$ (Eq. 4) are about a factor of two lower than those based on $J_{Nucl-Org}$ (Eq. 1)". We feel that Figure 4, combined with Figure 5, is adequate to make the point that T-dependence correction improves the agreement of predicted CN10 at the nine forest sites with observations in the summer. To further demonstrate this point and to address the referee's concern, we added in the revised manuscript a figure comparing the observed and simulated CN10 values averaged over the nine forest sites for different Δ H values.

3. The authors tend to take their model and its configuration somewhat uncritically. For example, the introduction paints a clear picture of new particle formation being a "strong" and "significant" contributor to the aerosol indirect effect and CCN concentrations, citing mostly their own work. The authors should perform due diligence to other work which may not agree with their single model findings (some suggested citations below). Indeed, despite changing the mean nucleation rate by nearly an order of magnitude with Nucl-OrgT, CCN0.2 are reported to be changed by only around 10-20% globally, suggesting probably a weaker sensitivity than is introduced by the authors.

Spracklen et al. 2008, JGR, 35, L06808 Westervelt, et al, 2014. Atmos. Chem. Phys., 14, 5577-5597

In the Introduction, we cited "(Spracklen et al., 2008; Pierce and Adams, 2009; Yu and Luo, 2009)" to point out the dominant contribution of secondary particles to particle number concentrations and cited "(Wang and Penner, 2009; Kazil et al., 2010; Yu et al., 2012)" to support the strong effect of nucleation schemes/parameterizations on the aerosol IRF estimations. We feel that we were not citing mostly our own work in these cases.

In the sentence followed, i.e., "Different nucleation schemes, with nucleation rates depending on different variables, predict significantly different spatial patterns and seasonal variations of nucleation rates and CCN concentrations (Yu et al., 2010, 2015)", we emphasized "significantly different <u>spatial patterns and seasonal variations</u>". So this citation is not about the sensitivity of CCN to nucleation schemes.

The exact magnitude of CCN change associated with nucleation rate changes depend on seasons, locations, water supersaturation ratio at which CCN is calculated, assumption of primary particle emissions, and others. The focus of this study is on the impact of temperature dependence on the possible contribution of organics to new particle formation in the atmosphere, rather than the sensitivity of CCN to nucleation schemes. Therefore, we would like not to delve into the discussion of the exact magnitudes of CCN sensitivity to nucleation rates in the present study.

Other comments:

1. Line 35-36, Page 3: While technically true, the impact of aerosols on the surface solar radiation balance and thus atmospheric circulation is a stronger control on the hydrological cycle than the indirect effect.

Point taken but we think it is fine to say here that particles can "affect" the hydrological cycle.

2. Line 67 Page 4: "In the recent study. . . " change to "In a recent study". Also looks like there are too many spaces between "the" and "recent". There are a few other instances of superfluous "the", please proofread carefully.

Modified. Thanks.

3. Line 136, Page 7: Why is fT capped at 10? Seems arbitrary.

The physical basis for setting the maximum value of f_T is that the organics mediated nucleation rates should not exceed the 3-body kinetic collision rate for forming a cluster containing two H_2SO_4 molecules and one BioOxOrg molecule (k_{max}). k_{max} depends on T as well as the mass and sizes of colliding molecules. At T=270 K, k_{max} is about a factor of 38 higher than k_m . We have revised the manuscript to set the maximum value of f_T to k_{max}/k_m instead of 10. The modification has negligible effect on the results and conclusions of this study which focuses on the summer month and the boundary layer.

4. Line 132, Page 7: Along with major comment 1, explain why this deltaH value is used.

Please see our reply to major comment 1.

5. Line 202, Page 10: "agree much better with the observed values". As per major comment 2, need to actually show this.

Please see our reply to major comment 2.

6. Line 260, Page 12: "To our acknowledge". This should say "To our knowledge" Corrected. Thanks.

1	Impact of temperature dependence on the possible contribution of organics to new
2	particle formation in the atmosphere
3	Fangqun Yu ¹ , Gan Luo ¹ , Alexey B. Nadykto ^{1, 2} , and Jason Herb ¹
4	
5	
6	¹ Atmospheric Sciences Research Center, State University of New York, 251 Fuller Road,
7	Albany, New York 12203, USA
8	² Department of Applied Mathematics, Moscow State University of Technology "Stankin",
9	Vadkovsky 1, Moscow, Russia
10	
11	

Abstract. Secondary particles formed via new particle formation (NPF) dominate cloud 13 condensation nuclei (CCN) abundance in most parts of the troposphere and are important for 14 aerosol indirect radiative forcing (IRF). Laboratory measurements have shown that certain 15 organic compounds can significantly enhance binary nucleation of sulfuric acid and H₂O. 16 According to our recent study comparing particle size distributions measured in nine forest areas 17 in North America with those predicted by a global size-resolved aerosol model, current H₂SO₄-18 19 Organics nucleation parameterizations appear to significantly over-predict NPF and particle 20 number concentrations in summer. The lack of the temperature dependence in the current H₂SO₄-21 Organics nucleation parameterization has been suggested to be a possible reason for the observed over-prediction. In this work, H₂SO₄-Organics clustering thermodynamics from quantum-22 chemical studies has been employed to develop a scheme to incorporate temperature dependence 23 into H₂SO₄-Organics nucleation parameterization. We show that temperature has a strong impact 24 on H_2SO_4 -Organics nucleation rates, and may reduce nucleation rate by ~ one order of 25 magnitude per 10 K of the temperature increase. The particle number concentrations in summer 26 over North America based on the revised scheme is a factor of more than two lower, in much 27 better agreement with the observations. With the temperature-dependent H₂SO₄-Organics 28 nucleation parameterization, the summer month CCN concentrations in the lower troposphere in 29 the northern hemisphere are about 10-30% lower compared to the temperature independent one. 30 This study highlights the importance of the temperature effect and its impacts on NPF in global 31 modeling of aerosol number abundance. 32

Deleted: and the aerosol first IRF about 0.5 - 1.0 W/m² less negative

33

Deleted: IRF

37 1. Introduction

38 Atmospheric particles, through acting as cloud condensation nuclei (CCN), modify cloud properties and precipitation and thus, indirectly, affect the hydrological cycle and climate. 39 Aerosol indirect radiative forcing (IRF) remains a major uncertainty in assessing climate change 40 (IPCC, 2013). Secondary particles formed via nucleation dominate particle number 41 concentrations in many parts of troposphere (Spracklen et al., 2008; Pierce and Adams, 2009; Yu 42 and Luo, 2009), and global simulations indicate that nucleation schemes/parameterizations have 43 44 a strong effect on the aerosol IRF estimations (Wang and Penner, 2009; Kazil et al., 2010; Yu et 45 al., 2012). Different nucleation schemes, with nucleation rates depending on different variables, 46 predict significantly different spatial patterns and seasonal variations of nucleation rates and CCN concentrations (Yu et al., 2010, 2015). Therefore, it is important to understand mechanisms 47 of new particle formation (NPF) and the key parameters controlling the contribution of the NPF 48 to CCN formation under wide range of varying atmospheric conditions and to validate their 49 representation in regional and global climate models. 50

A number of laboratory chamber studies indicate that certain organic species can 51 significantly enhance NPF (e.g., Zhang et al., 2004; Riccobono et al., 2014). This finding may 52 have important implications for the interactions of anthropogenic and biogenic emissions and the 53 associated climate forcing. To this regard, it is necessary to assess the ability of organics-54 enhanced nucleation to explain nucleation phenomena observed in the atmosphere and to 55 determine the contribution of organics to atmospheric NPF and climate implications. In several 56 57 laboratory studies, empirical parameterization of formation rate as a function of the concentrations of sulfuric acid and low-volatility highly oxidized organics has been derived 58 (Metzger et al., 2010; Riccobono et al., 2014). One of the most important limitations of these 59

empirical parameterizations is that they were derived from the chamber measurements carried 60 out under limited range of well-controlled conditions and, thus, reliably extrapolating these data 61 to a wide range of atmospheric conditions remains a major issue. It should also be noted that 62 empirical activation and kinetic nucleation formulas $(J = k_1[H_2SO_4] \text{ or } J = k_2[H_2SO_4]^2)$ derived 63 from limited field measurements (e.g., Riipinen et al., 2007; Kuang et al., 2008) also do not 64 account for the impact of temperature variations on computed nucleation rates. Although these 65 66 simple empirical temperature independent nucleation parameterizations have been widely used in 67 global aerosol modeling and aerosol IRF studies (e.g., Spracklen et al., 2008; Wang and Penner, 68 2009; Kazil et al., 2010; Scott et al., 2014; Westervelt et al., 2014; Lupascu et al., 2015), possible 69 impacts of temperature variations were not considered in these studies.

In a recent study comparing particle size distributions measured in nine forest areas in North 70 America with those predicted by a global size-resolved (sectional) aerosol model, Yu et al. 71 (2015) showed that H₂SO₄-Organics nucleation parameterization of Riccobono et al. (2014) 72 significantly over-predict NPF and particle number concentrations in summer (Yu et al., 2005). 73 The lack of temperature dependence in the H₂SO₄-Organics nucleation parameterization has 74 been suggested as a possible reason for the observed over-prediction. The main objectives of the 75 present study are (1) to develop a scheme to incorporate temperature dependence into H_2SO_4 -76 Organics nucleation parameterization, (2) to assess the ability of the modified parameterization 77 in explaining the seasonal variations of NPF in NA, and (3) to study the global implications. 78

79

80 **2.** Methods

81 2.1. Organics-mediated nucleation parameterization with temperature dependence (Nucl-OrgT)

Deleted: the

Based on the CLOUD chamber study of nucleation process involving sulfuric acid and
organic compounds of relatively low volatility from the oxidation of pinanediol, Riccobono et al.
(2014) derived the following organics-mediated nucleation parameterization (Nucl-Org),

$$J_{\text{Nucl-Org}} = k_{\text{m}} \times [\text{H}_2 \text{SO}_4]^2 \times [\text{BioOxOrg}]$$
(1)

where $J_{Nucl-Org}$ is the formation rate (# cm⁻³s⁻¹) of particles of ~ 1.7 nm, k_m is the fitting pre-factor with a value of 3.27×10^{-21} cm⁶ s⁻¹ (90% confidence interval: 1.73×10^{-21} to 6.15×10^{-21} cm⁶ s⁻¹), [H₂SO₄] and [BioOxOrg] are the gas-phase concentrations (# cm⁻³) of H₂SO₄ and biogenic oxidized organic (BioOxOrg) vapors, respectively. In the chamber study reported in Riccobono et al. (2014), BioOxOrg molecules are organic compounds of relatively low volatility from the oxidation of pinanediol (a first-generation oxidation product of α -pinene) and represent latergeneration oxidation products of biogenic monoterpenes.

The Nucl-Org parameterization given in Eq. 1, derived from laboratory chamber studies at 94 T=278 K and RH=39% (Riccobono et al., 2014), does not consider the possible effect of 95 temperature on nucleation rate. According to the nucleation theory, nucleation rates are 96 temperature-dependent unless nucleation is barrierless and limited by collision rates only. 97 However, the value of the pre-factor k_m of 3.27×10^{-21} cm⁶ s⁻¹ is well below the three-body 98 collision rate corresponding to the formation of a cluster containing two H₂SO₄ and one 99 100 BioOxOrg molecules. This indicates that nucleation in the CLOUD chamber under conditions reported in Riccobono et al. (2014) was not barrierless, and, thus, nucleation rates should be 101 102 temperature-dependent.

Based on the classical homogeneous nucleation theory, the rate of nucleation (J) can be generally written in the form

105

$$J = C_1 \exp(-\Delta G/kT)$$
(2)

Deleted: much smaller than

where ΔG is the Gibbs free energy needed to form the critical cluster and C₁ is the pre-factor. With $\Delta G = \Delta H - T \Delta S$, where ΔH and ΔS are associated enthalpy and entropy change, we get

109
$$J = C_1 \exp(-\Delta H/kT + \Delta S/k) = C_1 \exp(\Delta S/k) \exp(-\Delta H/kT) = C_2 \exp(-\Delta H/kT)$$
(3)

The temperature dependence of nucleation rate is dominated by the exponential term in Eq. (3), although C_2 may also weakly depend on temperature. Assuming that C_2 is independent of temperature and using $J_{Nucl-Org}$ given in Eq. (1) as the nucleation rate at the reference temperature $T_0=278$ K, we obtain

$$\mathbf{J}_{\mathrm{Nucl-OrgT}} = \mathbf{J}_{\mathrm{Nucl-Org}} f_{\mathrm{T}} \tag{4}$$

115
$$f_{\rm T} = \exp\left[\frac{\Delta H}{k}\left(\frac{1}{T} - \frac{1}{T_0}\right)\right]$$
 (5)

116 where $f_{\rm T}$ is the correction factor accounting for the temperature dependence.

114

117 One challenge here is to obtain enthalpy change (ΔH) associated with the critical cluster formation because it is quite difficult to determine the chemical identities of BioOxOrg 118 119 molecules involved in atmospheric nucleation (Elm et al, 2014; Riccobono et al., 2014). As a 120 first order approximation, we use 2-Methyl-5-[(1S,2S,3R)-1,2,3,4-tetrahydroxybutyl]-3-furoic 121 acid, a select highly oxidized organic $C_{10}H_{14}O_7$ compound, as a proxy for BioOxOrg molecules. 122 The stability of the cluster composed of two H_2SO_4 and one $C_{10}H_{14}O_7$ molecules has been 123 investigated using the Density Functional theory (DFT) at PW91PW91/6-311++G(3df,3pd) level. The PW91PW91 is the most common density functional used in atmospheric studies that 124 125 predicts structure, vibrational spectrums, dipolar properties and thermodynamics of atmospheric molecules and molecular clusters with high degree of confidence and its predictions, which have 126 127 been systematically validated against experimental and higher level ab initio Gibbs free energies, are in a very good agreement with them for a number of atmospherically relevant molecules and 128

129	clusters (e.g. Herb et al., 2013, Elm et al., 2013, Nadykto et al., 2015; DePalma et al., 2015).
130	Computations have been carried out using the Gaussian 09 suite of programs (Frish et al., 2009).
131	Figure 1 presents the equilibrium geometry of the most stable isomers of heteromolecular
132	trimer composed of $(C_{10}H_{14}O_7)(H_2SO_4)_2$ and Table 1 reports the corresponding the
133	thermodynamic data associated the formation of this cluster. The computational methodology,
134	benchmarks of Gibbs free energy changes and Cartesian geometries of global minima and local
135	minima located within 1 kcal mol ⁻¹ of the global minima, along with interactions of $C_{10}H_{14}O_7$
136	and H ₂ SO ₄ with some base molecules, will be detailed in a separate manuscript. Here, as a first
137	order of approximation, we use ΔH value of -38.30 kcal mol ⁻¹ to calculate the temperature-
138	dependent factor $f_{\rm T}$ in Eq. (5). Figure 2 shows the calculated value of $f_{\rm T}$ as a function of T. It is
139	clear from Fig. 2 that $f_{\rm T}$ decreases significantly as T increases, roughly one order of magnitude
140	per 10 K. When T < 269 K, $f_{\rm T}$ becomes larger than 10 and increases with decreasing T. $f_{\rm T}$ is set
141	to have a maximum value so that J _{Nucl-OrgT} does not exceed the 3-body kinetic collision rate for
142	forming a cluster containing two H ₂ SO ₄ molecules and one BioOxOrg molecule, which depends
143	on T as well as the mass and sizes of colliding molecules. At T=270 K, the maximum value of $f_{\rm T}$
144	is ~ 38, Compared to the original $J_{Nucl-Org}$ parameterization of Riccobono et al. (2014) (Eq. 1)
145	derived from laboratory chamber studies at T=278 K and not taking into account the temperature
146	dependence of nucleation rates (i.e., $f_{\rm T}$ =1, dashed line in Fig. 2), the revised parameterization
147	$J_{\text{Nucl-OrgT}} = f_{\text{T}} J_{\text{Nucl-Org}}$ predicts quite different nucleation rate in the atmosphere, especially in the
148	summer season, when both T and VOC emissions are at peak values.
149	It should be noted that $f_{\rm T}$ shown in Fig. 2 is subject to large uncertainty because of the

potential difference between the molecules involved in the nucleation and the proxy molecule shown in Figure 1. The thermodynamic data for the formation of $(H_2SO_4)_2(BioOxOrg)$ clusters is Deleted: In the present study,

Deleted: of 10

154	quite limited. Elm et al. (2014) investigated the molecular interactions between the α -pinene
155	oxidation product pinic acid and sulfuric acid using computational methods and reported a ΔH
156	value of -42.5 kcal/mol for the formation of $(H_2SO_4)_2$ (Pinic Acid). More negative ΔH implies
157	stronger temperature dependence. The sensitivity of $f_{\rm T}$ values to $\Delta {\rm H}$ can be readily calculated
158	from Eq. (4). For example, a fairly large uncertainty of 5 kcal mol ⁻¹ in ΔH leads to the
159	uncertainty in $f_{\rm T}$ of a factor of ~ 1.4 at T=288 K, while the extremely large 20 kcal mol ⁻¹
160	variation in Δ H alters $f_{\rm T}$ at T=288 K by a factor of ~ 3.5. The sensitivity of predicted nucleation
161	rates and particle number concentrations to ΔH values is presented in Section 3. Despite possible
162	uncertainties in $f_{\rm T}$, the temperature dependent $J_{\rm Nucl-OrgT}$ is likely to be more realistic than $J_{\rm Nucl-Org}$,
163	in which the temperature dependence is neglected.

165 2.2. GEOS-Chem model and global simulations

This work represents <u>a major</u> global modeling attempt in studying the effect of temperature 166 on organics-mediated nucleation in the atmosphere. This study is built upon the work reported in 167 Yu et al. (2015) and, thus, we use the same global model (GEOS-Chem) and configurations as 168 that described in Yu et al. (2015). GEOS-Chem is a global 3-D model of atmospheric 169 composition driven by assimilated meteorological observations from the Goddard Earth 170 Observing System (GEOS) of the NASA Global Modeling and Assimilation Office (GMAO) 171 (e.g., Bey et al., 2001). More detailed information about GEOS-Chem and updates can be found 172 at the model website (http://geos-chem.org/). The aerosol simulation is based on a size-resolved 173 (sectional) advanced particle microphysics (APM) model incorporated into GESO-Chem by Yu 174 and Luo (2009) and considers the successive oxidation aging of the oxidation products of various 175 VOCs (Yu, 2011). In GEOS-Chem v8-03-02, on which this study and previous work (Yu et al., 176

Deleted: the first

178 2015) are based, the concentration of highly oxidized low volatile secondary organic gas from 179 the oxidation products of α -pinenes (LV-SOG_{α -pinene}) is explicitly simulated and used in Eqs. (1) 180 and (4) to calculate organics-mediated nucleation rates. The horizontal resolution of GEOS-181 Chem employed in this study is 2°×2.5° and there are 47 vertical model layers (with 14 layers 182 from surface to ~ 2 km above the surface). Other relevant model configurations (including 183 emission inventories and various schemes) can be found in Yu et al. (2015).

The main difference between the present study and the previous one reported by Yu et al. (2015) is that the present study employs the T-dependent Nucl-Org parameterization given in Eq. (4) instead of T-independent parameterization of Riccobono et al. (2014). In addition, study of Yu et al. (2015) focuses only on the NA region, while in this work, the discussion on organicsmediated nucleation is expanded to the whole globe.

189

190 **3. Results**

191 Figure 3 shows the effect of T-dependent correction factor on simulated global distributions 192 of monthly mean (July, 2006) nucleation rates, particle number and CCN concentrations in the boundary layer (0-1 km above the surface). The high biogenic VOC emissions in the summer 193 coupled with strong photochemistry lead to higher concentrations of LV-SOG_{a-pinene} or 194 BioOxOrg (Yu et al., 2015) and hence, according to the parameterization of Riccobono et al. 195 196 (2014) (i.e., Eq. 1), significant organics-mediated nucleation (Fig. 3a) and higher particle number concentrations (Figs. 3c and 3e). However, the high temperature in the summer substantially 197 lower nucleation rates (Fig. 3b), and reduce the global monthly mean nucleation rate in the 198 boundary layer from 0.17 cm⁻³s⁻¹ (Fig. 3a) to 0.02 cm⁻³s⁻¹, with stronger effect in the northern 199

201	layer decrease by 40% and 30%, respectively.			
202	<u>A ΔH value of -38.3 kcal mol⁻¹ was used in calculating J_{Nucl-OrgT} in Fig. 3b. The impact of ΔH</u>			
203	values (from 0 to 58.3 kcal mol ⁻¹) on J _{Nucl-OrgT} , CN10 and CCN0.4 averaged in the boundary			
204	layer over the whole globe for the same summer month is presented in Figure 4. A zero value of			
205	ΔH corresponds to the case of no T-dependent correction (i.e., $J_{Nucl-Org}$, Fig. 3a). $J_{Nucl-OrgT}$, CN10			
206	and CCN0.4 are more sensitive to ΔH when ΔH is small (~ 30 kcal mol ⁻¹), with J _{Nucl-OrgT}			
207	decreasing by a factor of 5, CN10 by 33% and CCN0.4 by 28% as Δ H increases from 0 to 28.3			
208	kcal mol ⁻¹ . Further increase of ΔH from 28.3 to 58.3 kcal mol ⁻¹ reduces J _{Nucl-OrgT} , CN10 and			
209	CCN0.4 by 100%, 25%, and 10%, respectively. It can be seen from Figure 4 that the T-			
210	dependent correction, even with a smaller value of ΔH , is important. On the other hand, the			
211	effect of potential uncertainty in ΔH around the values derived from quantum calculation (~ 40			
212	kcal mol ⁻¹ , see Section 2.1) is relatively weaker.			
213	As we have pointed out earlier, the previous comparisons of simulated and observed particle			
214	size distributions measured in nine forest areas in North America (NA) (Yu et al., 2005) showed			
215	that J _{Nucl-Org} parameterization (Eq. 1) over-predicts <u>condensation nuclei</u> number concentrations <u>in</u>			
216	the size range of 10 and 100 nm (CN ₁₀₋₁₀₀) at these sites in summer by a factor of around two on			
217	average (Yu et al., 2005). To examine the extent, at which the revised parameterization			
218	considering T-dependence (Eq. 4) can improve the agreement of simulations with measurements,			
219	we present the monthly mean horizontal distributions of CN10 zoomed into the NA region in Fig.			
220	5 and observed and simulated CN ₁₀₋₁₀₀ averaged over the nine forest sites in Fig. 6, It can be			
221	clearly seen from Fig. 5 that the simulated monthly mean CN10 values in the NA boundary layer			
222	based on $J_{Nucl-OrgT}$ (Eq. 4) are about a factor of two lower than those based on $J_{Nucl-Org}$ (Eq. 1),			

hemisphere (Fig. 3b). As a result, the global monthly mean CN10 and CCN0.4 in the boundary

200

Formatted: Superscript

Deleted: particle

 Deleted: 4

 Formatted: Subscript

225	with larger difference in the lower latitude part of the domain, where T is higher. In the case,		
226	6 when the effect of T on Nucl-Org is taken into account, the domain-wide average CN10 value		
227	decreases from 4600 to 2200 #/cm ³ and Figure 6 shows that the simulated CN_{10-100} averaged over		
228	the nine forest sites (with locations marked on Fig. 5, see Yu et al. (2015) for details) agrees		
229	much better with that of observed. It can also be seem from Fig. 6 that CN10-100 over the NA		
230	forest sites is more sensitive to ΔH values than the global mean CN10 shown in Fig. 4, and ΔH		
231	of ~ 35 kcal mol ⁻¹ agrees best with the observations.		

232 To further illustrate the difference and improvement for the cases with and without Tdependent correction, we present in Fig. 7, a set of detailed comparisons of simulated and 233 234 observed evolution of particle size distributions during two ten-day periods in March and July of 235 2006 in Duke Forest (Pillai et al., 2013), along with time series of CN_{10-100} (integrated from 236 PSDs), which give a good overall representation of particle nucleation and growth. The observed 237 PSDs and simulated PSDs based on $J_{Nucl-Org}$ has been discussed in Yu et al. (2015) and are 238 repeated here for comparison with J_{Nucl-OreT} scheme in order to demonstrate the impact of 239 temperature on nucleation and particle number concentrations. Although the present work 240 focuses on the summer month, when the largest difference between $J_{Nucl-Org}$ prediction and 241 observation is observed, we also show in Fig. 7 simulations for a 10-day period in March as well for the comparison purpose. NPF events observed in Duke Forest are much more frequent and 242 concentrations of nucleation mode particles are much higher in the spring than in the summer 243 (Figs. 7a & 7b). The temperature correction (Eq. 4) has small effect in the spring (Figs. 7c, 7e, 244 245 and 7g) but significantly reduces nucleation rate and particle number concentration in summer 246 (Figs. 7d, 7e, and 7f). J_{Nucl-Org} scheme (Eq. 1) predicts strong nucleation events (Fig. 7d) and 247 significant diurnal variations in CN_{10-100} (Fig. <u>7</u>h) almost every day in the summer period that

-{	Deleted: particle number concentrations at				
-{	Deleted: 4				
-(Deleted: the				
-{	Deleted: values				

Deleted: 5

Deleted: the concentration of condensation nuclei (CN) between 10 and 100 nm (**Deleted:**,

Deleted: 5

Deleted: 5
Deleted: 5

obviously contradicts to observations (Fig. 7b). The high nucleation rates in the summer based 267 on J_{Nucl-Org} scheme can be easily explained by the much higher BioOxOrg concentrations as a 268 result of high VOC emissions and stronger photochemistry. Nevertheless, the high T in the 269 summer inhibits nucleation (Eq. 4) and the temperature correction factor substantially improves 270 the agreement of the simulated evolution of PSDs (Figs. 7_{b} , 7_{d} , 7_{f}) and CN_{10-100} (Fig. 7_{h}) with 271 272 observations.

273 Figure & shows the ratios of the CCN concentration in the lower troposphere (0-3 km) based 274 on Nucl-Org to the CCN concentration based on Nucl-OrgT, The CCN concentrations are 275 calculated at a water supersaturation ratio of 0.2% (CCN0.2) from simulated PSDs. As a result of 276 higher nucleation rates, CCN0.2 based on Nucl-Org are about 10-20% higher than those based on Nucl-OrgT in July over most parts of northern hemisphere (Fig. 6a), with the largest 277 difference up to 30-70% reached over part of NA, Europe, and Asia. 278

279 4. Summary and discussion

Simple empirical nucleation parameterizations, which were derived from laboratory or field 280 measurements under limited conditions and do not consider any temperature dependence of 281 nucleation rates, have been widely used in global aerosol modeling and aerosol indirect radiative 282 forcing studies. Based on the classical nucleation theory, temperature should be one of key 283 parameters controlling nucleation rates, unless nucleation is barrierless. A recent study indicates 284 (Yu et al., 2015) that the empirical parameterization of H₂SO₄-Organics nucleation of Riccobono 285 et al. (2014) significantly over-predicts NPF and particle number concentrations in North 286 287 America in summer. The lack of temperature dependence in the parameterization has been suggested as a likely reason for the observed over-prediction. In the present study, H_2SO_4 -288 289 Organics clustering thermodynamics from quantum-chemical studies has been employed to Deleted: 5

Deleted: 5
Deleted: 5
Deleted: 5
Deleted: 5
Deleted: in
Deleted: ¶
Deleted: 6
$\begin{array}{l} \textbf{Deleted:} \ \mathrm{as \ well \ as \ the \ difference \ of \ aerosol \ first \ indirect \ radiative \ forcing \ (FIRF) \ (FIRF_{Nucl-Org} - FIRF_{Nucl-Org} - FIRF_{Nucl-Org} - FIRF_{Nucl-Org} \) \end{array}$
Deleted: and the calculation of aerosol FIRF in GEOS-Chem is based on the approach discussed in Yu et al. (2013)

Deleted: Higher CCN leads to higher cloud droplet number concentrations and enhanced cloud albedo, resulting in an additional negative FIRF (cooling) of $0.5 - 1.0 \text{ W/m}^2$ in a large fraction of NH with a global average of -0.2 W/m².

Deleted: ¶

310	develop a scheme for incorporating the temperature dependence into H ₂ SO ₄ -Organics nucleation		
311	parameterization, which reduces global mean nucleation rate in the boundary layer in a summer		
312	month is by a factor of ~ 8 and improves the agreement of predicted particle number		
313	concentrations over North America with observations. With temperature-dependent $\mathrm{H}_2\mathrm{SO}_4\text{-}$		
314	Organics nucleation parameterization, the summer month CCN concentrations in the lower		
315	troposphere in the northern hemisphere are about 10-30% lower. In view of the potential effects		
316	of changes in CCN concentrations on precipitation (second indirect impact) and cloud cover, it is		
317	important to reduce uncertainties in NPF calculation in regional and global climate models.		

The study highlights the importance of including the temperature dependence of nucleation 318 319 rates in global modeling of NPF and aerosol indirect radiative forcing. In a recent study, Dunne et al. (2016) also showed a substantial impact of the temperature dependence on the contribution 320 of organic nucleation to overall nucleation. The temperature dependence factor derived under 321 322 this study <u>can be applied to</u> study the temperature effect on organics-mediated nucleation in the global atmosphere and improve the agreement of simulated particle number concentrations with 323 observations. Although it may subject to uncertainties due to the possible difference between the 324 molecules involved in the nucleation and the proxy molecule, temperature dependent $J_{\text{Nucl-OroT}}$, 325 likely more realistic than $J_{Nucl-Org}$, in which the temperature dependence is neglected. Further 326 laboratory measurements and theoretical studies are needed to better understand the effect of 327 328 temperature on organics-mediated nucleation in the atmosphere.

Acknowledgments. This study was supported by NASA under grant NNX13AK20G and <u>the</u>
 National Science Foundation (NSF) under grant 1550816. We would like to acknowledge high performance computing support from Yellowstone (ark:/85065/d7wd3xhc) provided by NCAR's

329

333 <u>Computational and Information Systems Laboratory, sponsored by the NSF.</u> The GEOS-Chem

Deleted: , and the aerosol first IRF about 0.5 - 1.0 W/m² less negative compared to temperature independent one

1	Deleted: represents the first (to our acknowledge) attempt of
1	Deleted: ing
-	Deleted: s
1	Deleted: between
Y	Deleted: and
Y	Deleted: in NA

Formatted: Font: Times

Deleted: US

345 model is managed by the Atmospheric Chemistry Modeling Group at Harvard University with

- 346 support from NASA's Atmospheric Chemistry Modeling and Analysis Program.
- 347

348 **References**

- Bey, I., Jacob, D. J. Yantosca, R. M., Logan, J. A. Field, B. Fiore, A. M. Li, Q., Liu, H.,
 Mickley, L. J., and Schultz, M.: Global modeling of tropospheric chemistry with assimilated
 meteorology: Model description and evaluation, Journal of Geophysical Research, 106,
 23,073–23,096, 2001.
- DePalma, J. W., J. Wang, A. S. Wexler, and M.V Johnston, Growth of ammonium bisulfate
 clusters by adsorption of oxygenated organic molecules, The Journal of Physical Chemistry
 A, 119 (45), 11191–11198, DOI: 10.1021/acs.jpca.5b07744, 2015.
- 356 Dune et al., Global atmospheric particle formation from CERN CLOUD measurements, Science,

357 <u>354, 6316, DOI:10.1126/science.aaf2649, 2016.</u>

- Elm, J., M. Bilde, and K. V. Mikkelsen: Assessment of binding energies of atmospherically
 relevant clusters. Physical Chemistry Chemical Physics 15.39, 16442-16445., 2013.
- 360 Elm, J., T. Kurten, M. Bilde and K.V. Mikkelsen: Molecular interaction of pinic acid with
- sulfuric acid: Exploring the thermodynamic landscape of cluster growth, The Journal of
 Physical Chemistry A, 2014, 118, 7892–7900, 2014.
- 363 Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.;
- Scalmani, G.; Barone, V.; Mennucci, B.; et al.: Gaussian 09, Gaussian, Inc., Wallingford CT,
 2009.

- Herb, J., Y. Xu, F. Yu, and A. B. Nadykto, Large hydrogen-bonded pre-nucleation (HSO₄⁻)
- 367 $(H_2SO_4)_m(H_2O)_k$ and $(HSO_4)(NH_3)(H_2SO_4)_m(H_2O)_k$ Clusters in the Earth's Atmosphere, The
- 368 Journal of Physical Chemistry A, *117* (1), 133–152, 2013.
- 369 IPCC, Climate Change 2013: The Physical Scientific Basis, edited by T. F. Stocker, D. Qin, et al.,
 370 Cambridge Univ. Press, New York, USA, 2013.
- 371 Kazil, J., Stier, P., Zhang, K., Quaas, J., Kinne, S., O'Donnell, D., Rast, S., Esch, M., Ferrachat,
- 372 S., Lohmann, U., and Feichter, J.: Aerosol nucleation and its role for clouds and Earth's
- 373 radiative forcing in the aerosol-climate model ECHAM5-HAM, Atmospheric Chemistry and
- 374 Physics, 10, 10733-10752, doi:10.5194/acp-10-10733-2010, 2010.
- Kuang, C., McMurry, P. H., McCormick, A. V., and Eisele, F. L.: Dependence of nucleation
 rates on sulfuric acid vapor concentration in diverse atmospheric locations, J. Geophys. Res.,
 113, D10209, doi:10.1029/2007JD009253, 2008.
- Lupascu, A., Easter, R., Zaveri, R., Shrivastava, M., Pekour, M., Tomlinson, J., Yang, Q., Matsui,
 H., Hodzic, A., Zhang, Q., and Fast, J. D.: Modeling particle nucleation and growth over
 northern California during the 2010 CARES campaign, Atmos. Chem. Phys., 15, 1228312313, doi:10.5194/acp-15-12283-2015, 2015.
- 382 Metzger, A., Verheggen, B., Dommen, J., Duplissy, J., Prevot, A.S., Weingartner, E., Riipinen, I.,
- Kulmala, M., Spracklen, D.V., Carslaw, K.S. and Baltensperger, U.: Evidence for the role of
 organics in aerosol particle formation under atmospheric conditions, Proceedings of the
 National Academy of Sciences, 107(15), pp.6646-6651, 2010.
- 386 Nadykto, A. B., J. Herb, F.Yu, and Y. Xu: Enhancement in the production of nucleating clusters
- 387 due to dimethylamine and large uncertainties in the thermochemistry of amine-enhanced
- nucleation, Chemical Physics Letters, 609, 42-49, 2015.

- Pierce, J.R., and Adams, P.J.: Uncertainty in global CCN concentrations from uncertain aerosol
 nucleation and primary emission rates, Atmospheric Chemistry and Physics, 9, 1339-1356,
 2009.
- Pillai, P., Khlystov, A., Walker, J. and Aneja, V. Observation and analysis of particle nucleation
 at a forest site in southeastern US. Atmosphere. 4(2):72-93, 2013.
- 394 Riccobono, FSchoberberger, S., Scott, C.E., Dommen, J., Ortega, I.K., Rondo, L., Almeida, J.,
- 395 Amorim, A., Bianchi, F., Breitenlechner, M., David, A., Downard, A., Dunne, E.M.,
- 396 Duplissy, J., Ehrhardt, S., Flagan, R.C., Franchin, A., Hansel, A., Juuninen, H., Kajos, M.,
- 397 Keskinen, H., Kupc, A., Kürten, A., Kvashin, A.N., Laaksonen, A., Lehtipalo, K.,
- 398 Makkmutov, V., Mathot, S., Nieminen, T., Onnela, A., Petäjä, T., Praplan, A.P., Santos, F.D.,
- 399 Schallhart, S., Seinfeld, J.H., Sipilä, M., Spracklen, D.V., Stozhkov, Y., Stratmann, F., Tomé,
- 400 A., Tsagkogeorgas, G., Vaattlovaara, P., Viisanen, Y., Vrtala, A., Wagner, P.E., Weingartner,
- 401 E., Wex, H., Wimmer, D., Carslaw, K.S., Curtius, J., Donahue, N.M., Kirkby, J., Kulmala, M.
- Worsnop, D.R., and Baltensperger, U.: Oxidation products of biogenic emissions contribute
 to nucleation of atmospheric particles, Science, 344, 717, 2014.
- 404 Riipinen, I., S.-L. Sihto, M. Kulmala, F. Arnold, M. Dal Maso, W. Birmili, K. Saarnio, K.
- 405Teinilä, V.-M. Kerminen, A. Laaksonen, and K. E. J. Lehtinen: Connections between406atmospheric sulphuric acid and new particle formation during QUEST III–IV campaigns in
- 407 Heidelberg and Hyytiälä, Atmos. Chem. Phys., 7, 1899-1914, 2007.
- 408 Scott, C. E., Rap, A., Spracklen, D. V., Forster, P. M., Carslaw, K. S., Mann, G. W., Pringle, K.
- 409 J., Kivekäs, N., Kulmala, M., Lihavainen, H., and Tunved, P.: The direct and indirect
- 410 radiative effects of biogenic secondary organic aerosol, Atmos. Chem. Phys., 14, 447-470,
- 411 doi:10.5194/acp-14-447-2014, 2014.

- 412 Spracklen, D. V., Carslaw, K. S., Kulmala, M., Kerminen, V.-M., Sihto, Riipinen, I., Merikanto,
- 413 J., Mann, G.W., Chipperfield, M.P., Wiedensohler, A., Birmili, W., and Lihavainen, H.:
- 414 Contribution of particle formation to global cloud condensation nuclei concentrations,
- 415 Geophysical Research Letters, 35, L06808, doi:10.1029/2007GL033038, 2008.
- Wang, M. and Penner, J. E.: Aerosol indirect forcing in a global model with particle nucleation,
 Atmospheric Chemistry and Physics, 9, 239–260, 2009.
- Westervelt, D. M., Pierce, J. R., and Adams, P. J.: Analysis of feedbacks between nucleation rate,
 survival probability and cloud condensation nuclei formation, Atmos. Chem. Phys., 14,
- 420 5577-5597, doi:10.5194/acp-14-5577-2014, 2014.
- Yu, F.: A secondary organic aerosol formation model considering successive oxidation aging and
 kinetic condensation of organic compounds: global scale implications, Atmospheric
 Chemistry and Physics, 11, 1083-1099, doi:10.5194/acp-11-1083-2011, 2011.
- Yu, F., and G. Luo: Simulation of particle size distribution with a global aerosol model:
 contribution of nucleation to aerosol and CCN number concentrations. Atmospheric
 Chemistry and Physics, 9, 7691-7710, 2009.
- 427 Yu, F., G. Luo, T. Bates, B. Anderson, A. Clarke, V. Kapustin, R. Yantosca, Y. Wang, S.
- Wu: Spatial distributions of particle number concentrations in the global troposphere:
 Simulations, observations, and implications for nucleation mechanisms, J. Geophys. Res.,
 115, D17205, doi:10.1029/2009JD013473, 2010.
- 431 Yu, F., Luo, G., Liu, X., Easter, R. C., Ma, X., and Ghan, S. J.: Indirect radiative forcing by ion-
- 432 mediated nucleation of aerosol, Atmospheric Chemistry and Physics, 12, 11451-11463, 2012.

433	3 Yu, F., X. Ma, and G. Luo: Anthropogenic contribution to cloud condensation nuclei and the					
434	first aerosol indirect climate effect, Environmental Research Letters 8 024029					
435	doi:10.1088/1748-9326/8/2/024029, 2013.					
436	436 Yu, F., Luo, G., Pryor, S. C., Pillai, P. R., Lee, S. H., Ortega, J., Schwab, J. J., Hallar, A. G.,					
437	437 Leaitch, W. R., Aneja, V. P., Smith, J. N., Walker, J. T., Hogrefe, O., and Demerjian, K. L.:					
438	38 Spring and summer contrast in new particle formation over nine forest areas in North					
439	America, Atmos. Chem. Phys., 15, 13993-14003, doi:10.5194/acp-15-13993-2015, 2015.					
440	440 Zhang, R., I. Suh, J. Zhao, D. Zhang, E.C. Fortner, X. Tie, L.T. Molina, and M.J. Molina:					
441	Atmospheric new particle formation enhanced by organic acids, Science, 304, 1487-1490,					
442	2004.					
443						

446 Table 1. Changes of enthalpy (Δ H), entropy (Δ S), and Gibbs free energy (Δ G) for the formation

447 of $(C_{10}H_{14}O_7)$ $(H_2SO_4)_2$ cluster under the standard condition (P= 1 atm, T=298 K).

	ΔH (kcal mol ⁻¹)	$\frac{\Delta S}{(cal mol^{-1}K^{-1})}$	ΔG (kcal mol ⁻¹)
$C_{10}H_{14}O_7 + H_2SO_4 + H_2SO_4 \leftrightarrows (C_{10}H_{14}O_7)(H_2SO_4)_2$	-38.30	-75.45	-15.81

Figure Captions 449

Figure 1. Equilibrium geometry of the most stable isomers of heteromolecular trimer composed 450 of (C₁₀H₁₄O₇) (H₂SO₄)₂ obtained at PW91PW91/6-311++G(3df,3pd) level of theory. Bonding 451 452 lengths are in angstroms.

- 453 Figure 2. Calculated temperature dependence correction factor for Nucl-Org parameterization $(f_{\rm T})$ as a function of T. 454
- 455 Figure 3. Horizontal distributions of monthly mean nucleation rates (J) (a, b), concentrations of 456 condensation nuclei larger than 10 nm (CN10) (c, d), and concentrations of cloud condensation 457 nuclei at water supsaturation ratio of 0.4% (CCN0.4) (e, f) in the boundary layer (0-1 km above 458 the surface) in July of 2006 based on two organics-mediated nucleation schemes: $J_{Nucl-Org}$ (left 459
- panels) and J_{Nucl-OrgT} (right panels).
- Figure 4. Dependence of organics-mediated nucleation rates (left axis), CN10 and CCN0.4 460
- (right-axis) averaged in the boundary layer (0-1 km) over the whole globe for July 2006 on ΔH 461
- values assumed in calculating temperature dependence correction factor for Nucl-Org 462
- 463 parameterization ($f_{\rm T}$).
- Figure 5, Horizontal distributions of monthly mean CN10 in the boundary layer (0-1 km above 464
- surface) in July of 2006 based on two organics-mediated nucleation schemes: (a) $J_{Nucl-Org}$ and (b) 465
- J_{Nucl-OrgT}. The locations of 9 forest sites where observed particle size distributions measurements 466
- have been used for comparisons in Yu et al. (2015) are marked. 467
- 468 **Figure 6.** Effect of ΔH values on simulated number concentrations of condensation nuclei in the
- size range of 10 and 100 nm (CN₁₀₋₁₀₀) in the boundary layer (0-1 km) for July 2006 averaged 469
- 470 over nine forest sites in North America (NA) (locations marked on Fig. 5). The horizontal dashed
- 471 line shows the average of CN₁₀₋₁₀₀ observed in a summer month at the 9 sites.

Deleted: 4

Formatted: Don't adjust right indent when grid is defined, Don't adjust space between Latin and Asian text, Don't adjust space between Asian text and numbers, Tab stops: 3.31", Left

473	Figure 7, Particle size distributions (PSDs) observed (a, b) and simulated based on Nucl-Org (c,	 Deleted: 5
474	d) and Nucl-OrgT (e, f) schemes during two ten-day periods in March (a, c, e) and July (b, d, f)	
475	of 2006 in Duke Forest (DUK), along with time series of the concentration of condensation	
476	nuclei between 10 and 100 nm (CN ₁₀₋₁₀₀) (g, h).	
477	Figure 8, Ratios of the concentration of CCN (at water supersaturation ratio of 0.2%) in the	 Deleted: 6
		 Deleted: (a
478	lower troposphere (0-3 km) based on Nucl-Org scheme to those based on Nucl-OrgT scheme,	Deleted:
479		Deleted: , a indirect radia

-	Deleted: 6
\neg	Deleted: (a)
Υ	Deleted:
	$\begin{array}{l} \textbf{Deleted:} \ , \ and \ (b) \ Difference \ of \ aerosol \ first \\ indirect \ radiative \ forcing \ (FIRF) \ (FIRF_{Nucl-Org} - \\ FIRF_{Nucl-Org}) \end{array}$











495 Figure 2.









