

Determination of the atmospheric lifetime and global warming potential of sulphur hexafluoride using a three-dimensional model

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Abstract. We have used the Whole Atmosphere Community Climate Model (WACCM), with an updated treatment of loss processes, to determine the atmospheric lifetime of SF₆. The model includes the following SF₆ removal processes: photolysis, electron attachment and reaction with mesospheric metal atoms. The Sodankylä Ion Chemistry (SIC) model is incorporated into the standard version of WACCM to produce a new version with a detailed *D* region ion chemistry with cluster ions and negative ions. This is used to determine a latitude- and altitude-dependent scaling factor for the electron density in the standard WACCM in order to carry out multi-year SF₆ simulations. The model gives a mean SF₆ lifetime over a 11-year solar cycle (τ) of 1278 years (with a range from 1120 to 1475 years), which is much shorter than the currently widely used value of 3200 years, due to the larger contribution (97.4%) of the modelled electron density to the total atmospheric loss. The loss of SF₆ by reaction with mesospheric metal atoms (Na and K) is far too slow to affect the lifetime. We investigate how this shorter atmospheric lifetime impacts the use of SF₆ to derive stratospheric age-of-air. The age-of-air derived from this shorter lifetime SF₆ tracer is longer by 9% in polar latitudes at 20 km compared to a passive SF₆ tracer. We also present laboratory measurements of the infrared spectrum of SF₆ and find good agreement with previous studies. We calculate the resulting radiative forcings and efficiencies to be, on average, very similar to those reported previously. Our values for the 20, 100 and 500-year global warming potentials are 18,000, 23,800 and 31,300, respectively.

33 **1 Introduction**

34 Sulphur hexafluoride (SF_6) is an anthropogenic greenhouse gas which is mainly used as an
35 electrical insulator, with other applications as a quasi-inert gas. Although its main sources are
36 in the Northern Hemisphere, its atmospheric abundance is increasing globally in response to
37 these emissions and its long atmospheric lifetime (defined as the ratio of the atmospheric
38 burden of a trace gas to its rate of loss from the atmosphere). SF_6 is characterised by large
39 absorption cross-sections for terrestrial infrared radiation such that the presently increasing SF_6
40 abundance will contribute a positive radiative forcing over many centuries. The important
41 known removal sources are electron attachment and photolysis. Recently, Totterdill *et al.*,
42 (2015) have also measured bimolecular rate constants for the reaction of SF_6 with mesospheric
43 metals.

44 Harnisch and Eisenhauer (1998) reported that SF_6 is naturally present in fluorites, and out-
45 gassing from these materials leads to a natural background atmospheric abundance of 0.01
46 pptv. However, at present the anthropogenic emissions of SF_6 exceed the natural ones by a
47 factor of 1000 or more and are responsible for the rapid increase in its atmospheric abundance.
48 Surface measurements show that atmospheric SF_6 has increased steadily since the 1980s
49 (Geller *et al.*, 1997; Maiss and Brenninkmeijer, 1998) and is still increasing fairly linearly. The
50 current global mean surface mixing ratio is 8.60 pptv, which increasing by about 0.33 pptv/year
51 (~4%/yr) (Dlugokencky *et al.*, 2016).

52 SF_6 provides a useful tracer of atmospheric transport in both the troposphere and stratosphere.
53 Rates for transport of pollutants into, within, and out of the stratosphere are important
54 parameters that regulate stratospheric composition. The basic characteristics of the
55 stratospheric Brewer-Dobson (B-D) circulation are known from observations of trace gases
56 such as SF_6 : air enters the stratosphere at the tropical tropopause, rises at tropical latitudes, and
57 descends at middle and high latitudes to return to the troposphere. Understanding the rate of
58 this transport on a global scale is crucial in order to predict the response of stratospheric ozone
59 to climatic or chemical change. SF_6 is essentially inert in the troposphere to middle stratosphere
60 and is removed by electron attachment and photolysis in the upper stratosphere and mesosphere
61 (Ravishankara *et al.*, 1993). This tracer therefore provides an ideal probe of transport on
62 timescales of importance in the stratospheric circulation and quantitative information on mean
63 air mass age for the lower and middle stratosphere.

64 The mean age-of-air (AoA) is the interval between the time when the volume mixing ratio of

65 a linearly increasing atmospheric tracer reaches a certain value at a given location in the
66 stratosphere and an earlier time when this mixing ratio was reached at a reference location.
67 Mean AoA is expressed as (Hall and Plumb, 1994; Waugh and Hall, 2002)

$$68 \quad \text{AoA} = t(\chi, l, z) - t(\chi, l_0, z_0) \quad (\text{E1})$$

69 where t is time, χ is the volume mixing ratio, l and z are latitude and altitude, and the 0 subscripts
70 denote the reference latitude and altitude which are chosen to be the upper tropical troposphere
71 (for the purposes of our simulations we chose the gridpoint latitude = 1°N, altitude = 13.9 km).

72 In principle the trend of AoA can be used to diagnose changes in the strength of Brewer-Dobson
73 circulation (BDC); in practice, however, it is very difficult to obtain unambiguous results on
74 trends from this or any other trace gas (Garcia *et al.*, 2011). Ideally, AoA should be determined
75 experimentally using a tracer with very small (or zero) chemical sink in the stratosphere or
76 mesosphere. Otherwise, a correction must be applied to account for this loss. A correction
77 would also be necessary for any non-linear tropospheric growth. However, for the period
78 considered for diagnosing age-of-air in this paper (2002-2007) the growth of SF₆ is
79 approximately linear, so we can reasonably neglect such a correction for SF₆-derived AoA
80 (Hall and Plumb, 1994).

81 Ravishankara *et al.* (1993) reported the atmospheric lifetime of SF₆ to be 3200 years by
82 considering electron attachment and vacuum ultraviolet (VUV) photolysis. They also studied
83 the loss of SF₆ by reaction with O(¹D) but found the rate too slow to be important. They
84 deduced that electron attachment was the dominant loss process and quantified this process
85 using a 2-D model, wherein they assumed that all SF₆ molecules are destroyed after attachment
86 of an electron (with a rate constant of 10⁻⁹ cm³ molecule⁻¹ s⁻¹). They therefore argued that their
87 lifetime of 3200 years could be a lower limit, but clearly this result depends on the accuracy of
88 the 2-D electron density, which was calculated using only photochemistry. Morris *et al.* (1995)
89 subsequently extended the work of Ravishankara *et al.* (1993) by including an ion chemistry
90 module in the same 2-D model. They also made other assumptions to maximise the impact of
91 electron attachment on SF₆ loss and derived a lifetime as low as 800 years (which could be
92 further sporadically decreased by large solar proton events). Using a 3-D middle atmosphere
93 model, Reddman *et al.* (2001) estimated the lifetime to be 472 years when SF₆ is irreversibly
94 destroyed purely by direct electron attachment and to be 9379 years when SF₆ loss is assumed
95 to occur only via indirect loss (via the formation of SF₆⁻) and ionization via the reactions with
96 O₂⁺ and N₂⁺. They concluded that the estimated lifetime depends strongly on the electron

97 attachment mechanism, because the efficiency of this process as a permanent removal process
98 of SF₆ depends on the competition between reaction of SF₆⁻ with H and HCl, and
99 photodetachment and reaction with O and O₃. Here we extend on the above studies and
100 investigate the SF₆ lifetime using a state-of-the-art 3-D chemistry climate model with a domain
101 from the surface to 140 km. Our modelled electron density is based on results of a detailed ion
102 chemistry model and we use a detailed methodology for treating the atmospheric background
103 electrons, which is based on Troe's formalism (Troe *et al.*, 2007a,b; Viggiano *et al.*, 2007).

104 In addition to determining the SF₆ lifetime, in this study we report new measurements of the
105 infrared absorption cross-sections for SF₆ and input these into a line-by-line radiative transfer
106 model in order to obtain radiative forcings and efficiencies. These values are then used to
107 determine more accurate values of global warming potentials (GWPs) based on their cloudy
108 sky adjusted radiative efficiencies. GWP is the metric used by the World Meteorological
109 Organisation (WMO) and Intergovernmental Panel on Climate Change (IPCC) to compare the
110 potency of a greenhouse gas relative to an equivalent emission of CO₂ over a set time period.
111 The definitions of these radiative terms are given in Appendix A.

112 **2 Methodology**

113 **2.1 WACCM 3D model**

114 To simulate atmospheric SF₆ we have used the Whole Atmosphere Community Climate Model
115 (WACCM). Here we use WACCM 4 (Marsh *et al.*, 2013), which is part of the NCAR
116 Community Earth System Model (CESM; Lamarque *et al.*, 2012), configured to have 88
117 pressure levels from the surface to the lower thermosphere (5.96×10^{-6} Pa, 140 km) and a
118 horizontal resolution of $1.9^\circ \times 2.5^\circ$ (latitude \times longitude). The model contains a detailed
119 treatment of middle atmosphere chemistry including interactive treatments of Na and K (Plane
120 *et al.*, 2015). We use the specified dynamics (SD) version of the model to allow comparison
121 with observations (see Garcia *et al.*, (2014) for details). The SF₆ surface emission flux and
122 initial global vertical profiles were taken from a CCMI (Chemistry Climate Model Initiative)
123 simulation using the same version of SD-WACCM with the same nudging analyses (D.
124 Kinnison, personal communication, 2013).

125 Lyman- α photolysis is the only SF₆ loss reaction in the standard version of WACCM and in
126 this work we have added the additional processes given in **Table 1**. The rate constants for the
127 SF₆ + metal reactions have been measured in our laboratory for mesospheric conditions
128 (Totterdill *et al.*, 2015); here we use the experimental values for the reactions with Na and K.

129 For the photolysis of SF₆ we used the standard WACCM methodology but with the updated
 130 Lyman- α cross section from our laboratory of 1.37×10^{-18} cm² molecule⁻¹ (Totterdill *et al.*,
 131 2015). The WACCM Lyman- α flux is taken from Chabrillat and Kockarts (1997).

132 Electron attachment to SF₆ plays a major role in its atmospheric removal and so both
 133 dissociative and non-dissociative attachment are considered in this study. The detailed method
 134 is described in a recent paper (Totterdill *et al.*, 2015) and here only a brief summary is given.
 135 The removal process by the attachment of low energy electrons to SF₆ can be described using
 136 Troe's theory (Troe *et al.*, 2007a,b; Viggiano *et al.*, 2007). In the middle and lower mesosphere,
 137 electrons are mostly attached to neutral species in the form of anions. However, above 80 km
 138 the concentration of free electrons increases and the direct electron attachment to SF₆ becomes
 139 more likely. This can happen either by associative attachment forming the SF₆⁻ anion which
 140 can then undergo chemical reactions with H, O, O₃ and HCl, or by dissociative attachment
 141 forming the SF₅⁻ anion fragment. The probability β of dissociative attachment when an electron
 142 is captured by SF₆ is given by

$$143 \quad \beta(p, T) = \frac{k_{dis}}{k_{at} + k_{dis}} \quad (E2)$$

144 where k_{dis} is the rate constant for dissociative attachment and k_{at} is the rate constant for
 145 associative attachment. β can be expressed as

$$146 \quad \beta(p, T) = \exp(-4587T + 7.74) \times 10^{\left[4.362 - 0.582 \log_{10}(p/Torr) - 0.0203 \left(\log_{10}\left(\frac{p}{Torr}\right)\right)^2 / 5.26 \times 10^{-4}\right]}$$

147 (E3)

148 where T is the temperature in K and p is the pressure in Torr (Totterdill *et al.*, 2015).

149 We include both associative and dissociative electron attachment using WACCM-predicted
 150 electron concentrations (see **Table 1**). Note that the SF₆⁻ anion is not modelled directly. **As it**
 151 **is short-lived we can treat it as being in steady-state and consider the relative rates of its**
 152 **destruction pathways. Therefore,** the **net** SF₆ attachment loss rate is calculated by multiplying
 153 k_{at} by the **ratio of** permanent destruction of the resulting SF₆⁻ (reactions of SF₆⁻ with H and
 154 HCl) to the **total** sum of these reactions **plus** processes which recycle SF₆⁻ to SF₆ (reactions
 155 with O and O₃, and photodetachment).

156 In order to use a realistic electron concentration, the role of negative ions in the *D* region must
157 be considered. The *D* region is the lowest part of the ionosphere, extending from about 60 to
158 85 km. It is characterized by the appearance of cluster ions (e.g. proton hydrates $H^+(H_2O)_n$,
159 where $n \leq 6$) and negative ions (e.g. O_2^- , CO_3^- and NO_3^-) rather than free electrons. These
160 species predominate because the atmospheric pressure is high enough to facilitate the three-
161 body attachment of ligand species like H_2O to positive ions, and electrons to neutral molecules.
162 Therefore, a scaling factor was introduced that converts the standard WACCM electron
163 concentrations, which are calculated from charge balance with the five major positive *E* region
164 ions (N^+ , N_2^+ , O^+ , O_2^+ and NO^+), to more realistic electron concentrations. We have recently
165 incorporated the Sodankylä Ion Chemistry (SIC) model into the standard version of WACCM
166 to produce a new version (WACCM-SIC) containing a detailed *D* region ion chemistry with
167 cluster ions and negative ions (Kovács *et al.*, 2016). The mesospheric positive and negative
168 ions in WACCM-SIC are listed in **Table 2**. The electron scaling factor in each grid box of
169 WACCM was then defined as the annually averaged ratio of $[e]_{WACCM-SIC}/[e]_{WACCM}$ for the year
170 2013, where $[e]_{WACCM-SIC}$ is the electron density calculated from WACCM-SIC and $[e]_{WACCM}$
171 from the standard WACCM.

172 The scaling factor, which varies with altitude and latitude, is shown in **Figure 1** (bottom panel)
173 together with the electron densities from the standard WACCM (top panel) and WACCM-SIC
174 (middle panel) models. The annually averaged electron concentration in the WACCM-SIC
175 model is significantly smaller in the lower and middle mesosphere than in the standard
176 WACCM, which is expected because of negative ion formation. In the polar regions the scaling
177 factor is around 0.01 at 50 km and 0.5 at 70km. Note that in the upper mesosphere (70 - 80 km)
178 the electron density in WACCM-SIC is larger than WACCM, by over a factor 5 near the poles.
179 This results from the inclusion of medium energy electrons (MEE) (electrons with energy
180 between 30 keV and 2 MeV) in WACCM-SIC. **Figure 2** shows the effect of MEE by
181 comparing WACCM-SIC runs with and without this source of ionization in the upper
182 mesosphere included. To describe the effect of ionization, WACCM-SIC uses ionization rates
183 (I) as a function of time and pressure which were calculated from the spectra based on the
184 proton energy-range measurements in standard air as described by Verronen *et al.* (2005).
185 According to Figure 3 of Meredith *et al.* (2015), the annually averaged medium energy electron
186 flux for 2013 approximately corresponds to the long-term, 20-year average. This allows us to
187 assume that the annually averaged electron density of 2013 from WACCM-SIC can be used to

188 scale the long-term simulations using the standard WACCM aimed at determining the
189 atmospheric lifetime of SF₆.

190 The WACCM simulation included five different SF₆ tracers in order to quantify the importance
191 of different loss processes. All of these tracers used the same emissions but differed in their
192 treatment of SF₆ loss reactions. One SF₆ tracer included no atmospheric loss (i.e. a passive
193 tracer). Three tracers included one of the following loss processes for SF₆: (i) reaction with
194 mesospheric metals (Na, K), (ii) electron attachment, and (iii) UV photolysis. Finally, one ‘total
195 reactive’ SF₆ tracer included all three loss processes. This total reactive tracer should be the
196 most realistic and was used in the radiative forcing calculations. WACCM was run for the
197 period 1990-2007, and the first five years were treated as spin-up. For the analysis the monthly
198 mean model outputs were saved and later globally averaged for the lifetime calculations.

199

200 **2.2 Infrared absorption spectrum and radiative forcing**

201 Previous quantitative infrared absorption spectra of SF₆ have been compared in Hodnebrog *et*
202 *al.* (2013) (their Table 12). There are differences of ~10% between existing integrated cross-
203 section estimates, and the measurements cover different spectral ranges. We therefore
204 performed a more complete set of measurements over a wider spectral range, in order to reduce
205 uncertainty in the absorption spectrum and hence the radiative efficiency of SF₆. Measurements
206 were taken using an experimental configuration consisting of a Bruker Fourier transform
207 spectrometer (Model IFS/ 66), which was fitted with a mid-infrared (MIR) source used to
208 generate radiation which passed through an evacuable gas cell with optical path length 15.9
209 cm. The cell was fitted with KBr windows, which allow excellent transmission between 400
210 and 40,000 cm⁻¹. The choice of source and window were selected so as to admit radiation across
211 the mid IR range where bands of interest are known to occur. Room temperature (296 ± 2 K)
212 measurements were carried out between 400 and 2000 cm⁻¹ at a spectral resolution of 0.1 cm⁻¹
213 and compiled from the averaged total of 128 scans to 32 background scans at a scanner velocity
214 of 1.6 kHz. Gas mixtures were made using between 8 and 675 Torr of SF₆ diluted up to an
215 atmosphere using N₂, according to the method described in Totterdill *et al.* (2016).

216 Radiative forcing calculations were made using the Reference Forward Model (RFM) (Dudhia,
217 2013) which is a line-by-line radiative transfer model based on the previous GENLN2 model
218 (Edwards, 1987). Results obtained from this model were validated against the DISORT

219 radiative transfer solver (Stamnes *et al.*, 2000) included within the libRadtran (Library for
220 Radiative Transfer) package (Mayer and Kylling, 2005). A full description of these models and
221 parameters used alongside discussion of treatment of clouds and model comparison is also
222 given in Totterdill *et al.* (2016).

223 **3 Results**

224 **3.1 Global distributions of SF₆ from WACCM simulations**

225 **Figure 3** shows typical zonal mean profiles of the WACCM SF₆ tracers in the north and south
226 polar regions for different seasons, compared to MIPAS observations for the year 2007 (Haenel
227 *et al.*, 2015). Although the MIPAS SF₆ data provides much more coverage horizontally and
228 vertically compared to in situ aircraft and balloon data, it has only been validated up to 35 km
229 (Stiller *et al.*, 2008). Validation at higher altitudes is not possible due to the lack of suitable
230 reference data. Details of the validation of the MIPAS data version used here (V5h_SF6_20
231 for the full resolution product from 2004 and earlier; V5r_SF6_222 and V5r_SF6_223 for the
232 reduced resolution period of 2005 and later) can be found in Haenel *et al.* (2015), including
233 Figure S-2 of their supplementary material. The WACCM passive SF₆ tracer has a mixing ratio
234 profile that is almost constant with altitude in the stratosphere and lower mesosphere below 60
235 km, then decreases by about 10% by 70 km, after which it decreases more rapidly. Comparison
236 of the tracers that include loss processes show that removal of SF₆ is dominated by electron
237 attachment, with a small contribution (<3%) direct from photolysis. The mesospheric metals
238 make a negligible contribution because the Na and K layers occur in the upper mesosphere
239 above 80 km (with peaks around 90 km), and the concentrations of these metal atoms are too
240 low. **Figure 3** shows that the model mean profile agrees well with the MIPAS mean in the
241 polar lower stratosphere (around 20 km) but exhibits a positive bias of around 1 pptv at higher
242 altitudes in the middle stratosphere. However, the figure also shows that the variability in the
243 observed SF₆ at high latitudes is large. The time variation of modelled SF₆ shown in **Figure 4**
244 corresponds to an emission rate (slope) of 6.5×10^{-3} Tg/year, i.e. a 0.29 pptv/year increase in
245 global mean volume mixing ratio, and a volume mixing ratio of 6.4 pptv by the end of 2007.

246 **Figure 5** shows the zonal mean annual mean SF₆ distribution from the five WACCM tracers
247 and MIPAS observations for 2007. **Figure 5a** (and **Figure 3**) shows that there is a rapid
248 decrease in SF₆ above 75 km even for the inert tracer. This can be explained by diffusive
249 separation, which becomes pronounced in the upper mesosphere because SF₆ is a relatively
250 heavy molecule (molar mass 0.146 kg) compared to the mean mass of air molecules (mean

251 molar mass 0.0288 kg) (cf. Garcia *et al.* (2014), where similar behaviour is seen for CO₂,
252 another relatively heavy molecule). Panels (a)-(c) of the figure all show SF₆ decreasing above
253 ~80 km, and panels (a) and (c) are almost identical, while in panel (b) the decrease begins about
254 4 km lower. This is all consistent with the notion that metals do not affect SF₆ and photolysis
255 contributes only slightly. The fact that diffusive separation prevents SF₆ from reaching altitudes
256 where photolysis is faster must be contributing to the very long lifetime (48,000 years, Table
257 3) found when photolysis is the only loss considered. By contrast, in Figure 5d SF₆ decreases
258 rapidly above 70 km, which is related to the fact that loss via electron attachment is important
259 at these lower altitudes. Thus, in this case, SF₆ loss occurs below the altitudes (~90 km) where
260 diffusive separation is important (and where air density is higher), which makes it a much more
261 effective loss mechanism. The WACCM SF₆ tracer that includes all loss processes (Figure 5e)
262 has a very similar distribution to that which only treats loss due to electron attachment (Figure
263 5d), which emphasises how this process dominates SF₆ loss in the model. This model tracer
264 can be compared to the MIPAS observations in Figure 5f, which shows that WACCM appears
265 to reproduce the general features of the MIPAS distribution (note the smaller altitude range in
266 panels (e) and (f) of Figure 5). However, it is also clear that WACCM SF₆, even with all losses
267 considered, decreases with altitude much more slowly at all latitudes than MIPAS SF₆. This
268 could indicate a problem with the model's meridional transport. However, a too-fast BDC
269 would tend to produce low levels of SF₆ at middle and high latitudes in the descending branch,
270 which does not seem to be the case. Therefore, at least two other possible scenarios could be
271 responsible for the discrepancy: SF₆ loss in WACCM is still somewhat underestimated despite
272 the inclusion of the electron attachment, or MIPAS SF₆ is biased low above ~20 km.

273 3.2 Atmospheric lifetime

274 The atmospheric lifetime is defined as the ratio of the atmospheric burden to the atmospheric
275 loss rate. This definition was used to calculate annual mean lifetime values from the WACCM
276 output containing the individual rates for the different loss processes. During the simulation
277 the total atmospheric burden of SF₆ increased linearly as expected (see Figure 4) from 3.4×10^{32}
278 molecules with an annual increment of 2.3×10^{31} molecules/year. Figure 6 shows the variation
279 in SF₆ lifetime from 1995 to 2007, corresponding to a full solar cycle (the solar minima
280 occurred in May 1996 and January 2008). The figure demonstrates that the lifetime has a strong
281 dependence on solar activity, being anti-correlated with solar activity, for example as measured
282 by the radio flux at 10.7 cm (2800 MHz) (Tapping, 2013) which ranges over $(72 - 183) \times 10^{-22}$
283 W m⁻² Hz⁻¹, with an average value of 90.3×10^{-22} W m⁻² Hz⁻¹. The mean SF₆ lifetime over the

284 same solar cycle period is 1278 years, with a range from 1120 to 1475 years. The annual
285 averaged electron number density in the polar regions is also plotted in **Figure 6**; as expected,
286 it is correlated with the 10.7 cm radio emission (Tapping, 2013).

287 As noted in Section 1, the SF₆ lifetime has been reported to be 3200 years by Ravishankara *et al.*
288 *al.*, (1993). For this they used a total electron attachment rate constant of $k_{EA} = 10^{-9} \text{ cm}^3 \text{ s}^{-1}$. In
289 Morris *et al.* (1995) the calculated lifetime decreased to 800 years by considering ion chemistry
290 and assuming that the associative attachment forming SF₆⁻ does not regenerate the parent
291 molecule, thereby obtaining a lower limit for the lifetime. Reddmann *et al.* (2001) estimated
292 the lifetime to be 472 years when SF₆ is irreversibly destroyed purely by direct electron
293 attachment and to be 9379 years when SF₆ loss is assumed to occur only via indirect loss (via
294 the formation of SF₆⁻) and ionization via the reactions with O₂⁺ and N₂⁺. In the present study
295 we have directly applied Troe's theory (Troe *et al.*, 2007a,b; Viggiano *et al.*, 2007) to determine
296 the efficiency of electron attachment as a function of temperature and pressure, and the
297 branching ratio for dissociative attachment (equation E2), which we extrapolated to
298 mesospheric conditions (Totterdill *et al.*, 2015).

299 Our estimated *partial* lifetime of SF₆ due to photolysis (**i.e. the lifetime calculated considering**
300 **photolysis as the only atmospheric loss process**) for the SF₆ tracer which includes all loss
301 processes is 48,000 yr, which is considerably longer than that the 13,000 yr determined by
302 Ravishankara *et al.* (1993) despite our Lyman- α cross section ($1.37 \times 10^{-18} \text{ cm}^2$, **Table 1**) being
303 only ~22% smaller than the value the value measured by Ravishankara *et al.* ($1.76 \times 10^{-18} \text{ cm}^2$).
304 One reason why our photolysis-related partial lifetime is longer is that WACCM includes
305 diffusive separation, which was not described in the earlier 2-D model study. The inclusion of
306 diffusive separation reduces sharply the abundance of SF₆ at high altitudes, where photolysis
307 is most effective. Another contributing factor could be that the VUV photolysis is important
308 only above 80 km, while in our model runs SF₆ is mostly destroyed by electron attachment,
309 which results in less being transported into this upper mesospheric region. When we analyse
310 our WACCM SF₆ tracer which is subject to photolysis loss only, the resulting steady-state
311 *overall* lifetime (**i.e. lifetime calculated using the rates of all loss processes**) for the last model
312 year (2007) is 17,200 yr which is only 32% larger than the value of Ravishankara *et al.* (1993)
313 and thus more consistent with the difference in the Lyman- α cross sections. Finally, if we do
314 not include the electron scaling factor to reduce the electron density below 80 km due to

315 negative ion formation, then the SF₆ lifetime decreases to 776 years (not shown), which is
316 similar to the value of 800 years obtained by Morris *et al.* (1995).

317 **3.3 Impact of SF₆ loss on mean age of stratospheric air**

318 As SF₆ is a chemically stable molecule in the stratosphere and troposphere, and has an almost
319 linearly increasing tropospheric abundance, its atmospheric mixing ratio is often used to
320 determine the mean age of stratospheric air. This is an important metric in atmospheric science
321 as the distribution of ozone and other greenhouse gases depends significantly on the transport
322 of air into, within, and out of the stratosphere. WACCM contains an idealized, linearly-
323 increasing age-of-air tracer (AOA1) that provides model age values for model experiments
324 (Garcia *et al.*, 2011).

325 Age-of-air has generally been derived from observations by treating SF₆ as a passive (non-
326 reactive) tracer. The assumption is that the global loss rate is too slow to significantly affect
327 the lifetime. This was confirmed by Garcia *et al.* (2011) when only photolysis was included.
328 However, when loss via electron attachment is also considered, the lifetime may become short
329 enough that this assumption is no longer valid, in which case the stratospheric mixing ratio
330 would appear to correspond to an earlier tropospheric mixing ratio than in reality. We have
331 compared the passive WACCM SF₆ tracer with that subject to all loss processes, which yields
332 the new lifetime of 1278 yr. The difference between these two tracers indicates the error in the
333 derived age-of-air that would arise in the real atmosphere if SF₆ is assumed to be a passive
334 tracer. The error caused by chemical removal can be expressed as:

$$335 \quad \Delta(\text{AoA}) = \text{AoA}(\text{reactive tracer}) - \text{AoA}(\text{passive tracer}) \quad (\text{E4})$$

336 where $\Delta(\text{AoA})$ is the difference in the age-of-air value caused by chemical loss, $\text{AoA}(\text{reactive}$
337 $\text{tracer})$ is the calculated age-of-air considering the chemical removal, and $\text{AoA}(\text{passive tracer})$
338 is the value obtained from a non-reactive tracer. The expression for the age-of-air at any point
339 in the stratosphere can be obtained from a simplified version of (E1) that is derived from a
340 Taylor series expansion, retaining only the linear term. It is then expressed as

$$341 \quad \text{AoA} = [(\chi_0(\text{SF}_6) - \chi(\text{SF}_6)) / r(\text{SF}_6)] \quad (\text{E5})$$

342 where $\chi(\text{SF}_6)$ and $\chi_0(\text{SF}_6)$ are the SF₆ volume mixing ratios at the actual and the reference
343 (tropical tropopause) points, respectively, while $r(\text{SF}_6)$ is the rate of increase of tropospheric
344 SF₆. In our simulations $r(\text{SF}_6)$ is 0.29 pptv/year (**Figure 4**), which is an approximation as the

345 growth rate is not constant in reality. Stiller *et al.* (2012) report a value of 0.24 pptv/year based
346 on observations over the MIPAS period. These two simplifications will lead to deviations
347 between WACCM and MIPAS age data. If (E5) is substituted into (E4) then the error in age-
348 of-air will be:

$$349 \quad \Delta(\text{AoA}) = (\chi(\text{SF}_6, \text{passive}) - \chi(\text{SF}_6, \text{reactive})) / r(\text{SF}_6) \quad (\text{E6})$$

350 This error, along with the mean age itself, was calculated from WACCM output for 2007.
351 **Figure 7** shows the annual mean ages determined from the WACCM simulation from 2002-
352 2007 using the total reactive and the inert SF₆ tracers and the idealized AOA1 age tracer. There
353 is a clear difference between the age values derived from the passive SF₆ and the idealized AoA
354 tracer. If equation (E5) is used to determine the age values there is no guarantee that the age
355 values derived from the two tracers will be identical; the rate was determined from the increase
356 of the SF₆ burden (0.29 pptv/year) and this was provided by the linear fit (**Figure 4**), which
357 can misrepresent the growth rate at a specific time. **Figure 7** also shows the difference between
358 the age values obtained from the reactive and inert SF₆ tracers. It can be seen that consideration
359 of the reactive SF₆ tracer does indeed affect the determined mean age values, mostly where
360 electron attachment dominates. The age estimates at high latitudes are most sensitive to
361 chemical loss because the air that reaches these locations has descended from the high altitudes
362 where SF₆ loss predominantly occurs. According to the MIPAS satellite observations (Stiller
363 *et al.*, 2012; Haenel *et al.*, 2015.), the derived age value over the tropical lower stratosphere at
364 25 km is slightly more than 3 years, while the WACCM simulations with the reactive SF₆ tracer
365 predicts 3 years. Comparing **Figures 7a** and **7b**, the effect of chemical removal in this region
366 is minor (0.01 year or 0.5% change) and therefore it does not have much impact on the inferred
367 atmospheric transport. At the poles the effect is much more significant; the difference at 25 km
368 between the reactive and inert SF₆ tracers is up to 0.55 years (9%). In summary, in the
369 troposphere-stratosphere at low latitudes the effect of chemical removal is not very significant
370 and the error on the estimated mean age caused by the assumption of SF₆ being a passive tracer
371 is not important. However, the effect of chemical removal becomes more significant at high
372 latitudes.

373 We can also compare modelled and observed mean age values in the lower stratosphere (20
374 km). **Figure 8** shows the mean age profiles from WACCM tracers, ER-2 observations (Hall *et*
375 *al.*, 2009) and our analysis of MIPAS SF₆ satellite data at 20 km. From this it can be seen that
376 in the tropical region the mean age values are similar between the idealized age tracer and the

377 inert and reactive SF₆ tracers. This is consistent with no loss of SF₆ having occurred in air
378 parcels in the deep tropics. At high latitudes there is up to 0.5 year difference in the modelled
379 mean ages, with the reactive SF₆ tracer producing the oldest apparent age. The differences in
380 mean age between the tracers is larger in the SH polar region than in the NH because the polar
381 region is less well mixed. The tendency is very similar when we compare the WACCM mean
382 ages to the MIPAS observations. Note that the satellite observations show more seasonal
383 variability in the middle and high latitudes than in the tropics.

384 **3.4 Radiative Efficiency and Forcing**

385 To determine the radiative efficiency and global warming potential of SF₆, integrated cross-
386 sections were taken from two public Molecular Spectroscopic Databases: (i) the GEISA-
387 2009/2011 Spectroscopic Database (Jacquinet-Husson *et al.*, 2011), which uses the data of
388 Varanasi (2001) and Hurley (2003), and (ii) the HITRAN-2012 Molecular Spectroscopic
389 Database (Rothman *et al.*, 2012) which uses the data of the Pacific Northwest National
390 Laboratory Infra-Red Database (Sharpe *et al.*, 2004). Values were also measured in this study.
391 The literature values are presented in **Table 4** for comparison with our experimentally
392 determined values and the full SF₆ spectrum obtained in this study is given in **Figure 9**. In our
393 study the spectrometer error is $\pm 1.0\%$ for all experiments, and the uncertainty in the sample
394 concentrations of SF₆ was calculated to be 0.7%. Spectral noise was averaged at $\pm 5 \times 10^{-21}$ cm²
395 molecule⁻¹ per 1 cm⁻¹ band. However, at wavenumbers < 550 cm⁻¹, towards the edge of the mid
396 infrared where opacity of the KBr optics increases, this value was 1×10^{-20} cm² molecule⁻¹ per
397 1 cm⁻¹ band. The error from determining the scaling cross-section was 5%. This results in an
398 average overall error of $\pm 5\%$ in the cross-sections.

399 The intensities of the main SF₆ absorption bands (925-955 cm⁻¹) measured in this study are 7%
400 greater than those reported by Hurley (2003), 1% greater than Varanasi (2001) and 1% lower
401 than those given in HITRAN (Rothman *et al.*, 2012) (**Table 4**). Comparison of our results
402 against Varanasi (2001) between 650 and 2000 cm⁻¹ gives an agreement within 9%. Note that
403 these differences are within the combined error of both experiments.

404 The instantaneous and stratospheric adjusted SF₆ radiative efficiencies in clear and cloudy sky
405 conditions are given in **Table 5**. These are also presented as present-day radiative forcings
406 employing a current surface concentration of 9.3 pptv (NOAA, 2016) (see **Figure 4**). The
407 radiative efficiency was calculated in the RFM for each month between 90°S and 90°N at
408 latitudinal resolutions (on which the data was averaged to obtain the zonal mean vertical

409 profile) of 1.5° and 9.0°. The tropopause used the standard WMO lapse rate definition (see
410 Totterdill *et al.*, 2016). **Figure 10** shows the seasonal-latitudinal variation of the instantaneous
411 clear sky radiative forcing for SF₆ on the high (1.5°) and low (9°) resolution grids. Employing
412 profiles averaged over the lower resolution grid gives an average forcing within 1% of the
413 higher resolution grid. Using only a single annually averaged global mean profile led to a 10%
414 error in radiative forcing when compared to our monthly resolved high resolution profile,
415 supporting the findings of Freckleton *et al.* (1998) and Totterdill *et al.* (2016).

416 A selection of experiments were carried out over a range of months and latitudes to investigate
417 the sensitivity of the forcing calculations to the bands used. The average contributions from the
418 main bands were compared against the calculation with the full measured spectrum. The results
419 showed that the 580 – 640 and 925 – 955 cm⁻¹ bands contribute almost 99% to the instantaneous
420 radiative forcing. Our forcing calculations suggest that the SF₆ minor bands contribute only a
421 small amount to the final value. This means that deviations between our experimentally
422 determined spectra and those in the literature only result in a significant change to previously
423 published radiative forcings and efficiencies when that deviation occurs over a major band.

424 The SF₆ adjusted cloudy sky radiative efficiency published by the IPCC AR5 report and used
425 to determine its GWP values is 0.57 Wm⁻² ppbv⁻¹ (Myhre *et al.*, 2013). This compares to **our**
426 adjusted cloudy sky radiative efficiency of 0.59 Wm⁻² ppbv⁻¹ (**Table 5**). A review on radiative
427 efficiencies and global warming potentials by Hodnebrog *et al.* (2013) provides a
428 comprehensive list of all published values for these parameters for many species including SF₆.
429 They established the range of published radiative efficiencies for SF₆ to be **0.49** – 0.68 Wm⁻²
430 ppbv⁻¹, with a mean value of 0.56 Wm⁻² ppbv⁻¹. They also made their own revised estimate
431 using an average of the HITRAN (Rothman *et al.*, 2012) and GEISA (**Jacquinet-Husson et al.**,
432 **2011**) spectral databases and found a best estimate of (0.565 ± 0.025) Wm⁻² ppbv⁻¹. Their mean
433 value for radiative efficiency is very close to that determined in this study using similar
434 conditions (0.59 Wm⁻² ppbv⁻¹).

435 **3.5 Global Warming Potential**

436 **Table 6** gives our estimates of the 20, 100 and 500-year GWPs based on cloudy sky adjusted
437 radiative efficiencies of SF₆ compared with IPCC AR5 values (IPCC, 2013). Our 20, 100 and
438 500-year global warming potentials for SF₆ are 18,000, 23,800 and 31,300 respectively. The
439 20-year and 100-year values are 3% greater and 1% greater, respectively, than their IPCC
440 counterparts and the 500-year GWP is 4% smaller than its AR4 counterpart (Forster *et al.*,

441 2007). The forcing efficiencies determined in this study are somewhat higher than the
442 previously published values (see Section 3.4), which would imply a higher GWP value.
443 However, our shorter atmospheric lifetimes would lead to a smaller GWP estimate, with a
444 larger effect at longer time horizons when atmospheric loss becomes relevant. The trade-off
445 between these competing effects is apparent in Table 6, where SF₆ exhibits a 20-year GWP
446 that is slightly larger than the IPCC value, while the 500-year GWP is slightly smaller. The
447 radiative efficiency effect is most obvious for the case of the 20-year GWP where, because the
448 atmospheric lifetime of SF₆ is 1278 years, the species does not have time for any significant
449 loss to occur.

450 4 Conclusions

451 The 3D Whole Atmosphere Community Climate Model was used to simulate the SF₆
452 atmospheric distribution over the period of 1995-2007. From the concentrations and the
453 knowledge of the electron attachment, photolysis and metal reaction rates we determined the
454 atmospheric lifetime which shows a significant dependence (-12% to +15%) on the solar cycle
455 due to varying electron density. The mean SF₆ atmospheric lifetime and 1σ variation over a
456 solar cycle were determined to be 1278 years (ranging from 1120 to 1475 years), which is
457 different to previously reported literature values and much shorter than the widely quoted value
458 of 3200 years. The reason is our more detailed treatment of electron attachment using a new
459 formalism to describe both associative and dissociative attachment, and the use of a detailed
460 model of *D* region ion chemistry to evaluate the partitioning of electrons and negative ions
461 below 80 km. Further refinement to this lifetime estimate from modelling studies depends on
462 decadal length simulations of the model with detailed ion chemistry for the realistic long-term
463 estimation of electron density in the upper atmosphere without the need for scaling factors as
464 employed here.

465 Based on this new estimate of the SF₆ lifetime, we find that the derived mean age of
466 stratospheric air from observations can be slightly affected by the atmospheric removal of SF₆.
467 In the polar region the age-of-air values differ by up to 9% when the values from inert and
468 reactive model tracers are compared, suggesting that SF₆ loss does not have a large influence
469 on the age values but that it should be included in detailed analyses.

470 We also re-investigated the radiative efficiency and global warming potential of SF₆. Our
471 radiative efficiency value reported here, $0.59 \pm 0.045 \text{ Wm}^{-2} \text{ ppbv}^{-1}$, is slightly higher than the
472 IPCC AR5 estimate of $0.57 \text{ Wm}^{-2} \text{ ppbv}^{-1}$. The global warming potentials of SF₆ for 20, 100

473 and 500 years have been determined to be 18,000, 23,800 and 31,300, respectively. We find
474 that our revised lifetime and efficiency values somewhat cancel each other out so overall do
475 not play a significant role in modifying the GWP estimates on these time horizons.

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481 were performed on the Archer and University of Leeds HPC systems.

482 Appendix A

483 The climate metrics used in this paper are defined in the following ways.

- 484 • Radiative forcing (RF) is the change in the net irradiance at the tropopause after allowing
485 stratospheric temperatures to adjust to radiative equilibrium (expressed in watts per square
486 metre; Wm⁻²) due to an internal or external change in the forcing of the climate system,
487 such as a change in the concentration of CO₂ or the output of the Sun.
- 488 • Radiative efficiency is the radiative forcing per kg of compound.
- 489 • Global warming potential (GWP) is defined by the expression:

$$490 \quad \text{GWP} = \frac{\int_0^{\text{TH}} a_{\chi}[\chi(t)]dt}{\int_0^{\text{TH}} a_{\text{r}}[r(t)]dt} \quad (\text{E6})$$

491 where TH is time horizon; a_{χ} is radiative forcing due to a unit increase in atmospheric
492 abundance of the compound (Wm⁻² kg⁻¹) and $[\chi(t)]$ is its time-dependent decay in
493 concentration following its instantaneous release at time t=0. The denominator contains the
494 corresponding quantities for CO₂ as a reference gas (Myhre *et al.*, 2013). GWP is the most
495 common metric used by the WMO and IPCC to compare the potency of a greenhouse gas
496 relative to an equivalent emission of CO₂ over a set time period. GWP takes into account
497 species lifetime. This means a species with a very high radiative forcing may still have a low
498 GWP if it also possesses a short atmospheric lifetime. However, GWP does not account for
499 factors such as changes in emission or the introduction of replacement species. Criticisms of
500 GWP are discussed in greater detail by Myhre *et al.* (2013).

501 **References**

- 502 Chabrillat, S. and Kockarts, G.: Simple parameterization of the absorption of the solar Lyman-
503 alpha line, *Geophys. Res. Lett.*, 24, 2659-2662, 1997.
- 504 Dlugokencky, E.J., B.D. Hall, M.J. Crotwell, S.A. Montzka, G. Dutton, J.Mühle and J.W.
505 Elkins: Long-lived greenhouse gases [in “State of the Climate in 2015”], *Bull. Amer.*
506 *Meteor. Soc.*, 97 (8), S44-S46, 2016.
- 507 Dudhia, A.: Reference Forward Model V4.30: <http://www.atm.ox.ac.uk/RFM>, 2013.
- 508 Edwards, D. P.: GENLN2: The new Oxford line-by-line atmospheric transmission/radiance
509 model, Clarendon Laboratory, Oxford, 1987.
- 510 Freckleton, R. S., Highwood, E. J., Shine, K. P., Wild, O., Law, K. S., and Sanderson, M. G.:
511 Greenhouse gas radiative forcing: Effects of averaging and inhomogeneities in trace gas
512 distribution, *Q. J. Roy. Meteor. Soc.*, 124, 2099-2127, 10.1002/qj.49712455014, 1998.
- 513 Forster, P., V. Ramaswamy, P. Artaxo, T. Berntsen, R. Betts, D.W. Fahey, J. Haywood, J.
514 Lean, D.C. Lowe, G. Myhre, J. Nganga, R. Prinn, G. Raga, M. Schulz and R. Van
515 Dorland: Changes in Atmospheric Constituents and in Radiative Forcing. *In: Climate*
516 *Change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth*
517 *Assessment Report of the Intergovernmental Panel on Climate Change*, Cambridge
518 University Press, Cambridge, United Kingdom and New York, NY, USA, 2007.
- 519 Garcia, R. R., Randel, W. J., and Kinnison, D. E.: On the determination of age of air trends
520 from atmospheric trace species, *J. Atmos. Sci.*, 68, 139-154, 2011.
- 521 Garcia, R. R., López-Puertas, M., Funke, B., Marsh, D. R., Kinnison, D. E., Smith, A. K. and
522 Gómez-Galindo, F.: On the distribution of CO₂ and CO in the mesosphere and lower
523 thermosphere, *J. Geophys. Res.*, 119, 5700-5718, 2014.
- 524 Geller, L. S., Elkins, J. W., Lobert, J. M., Clarke, A. D., Hurst, D. F., Butler, J. H., and Myers,
525 R. C.: Tropospheric SF₆: Observed latitudinal distribution and trends, derived emissions and
526 interhemispheric exchange time, *Geophys. Res. Lett.*, 24, 675-678, 1997.
- 527 Haenel, F.J., Stiller, G.P., von Clarmann, T., Funke, B., Eckert, E., Glatthor, N., Grabowski,
528 U., Kellmann, S., Kiefer, M., Linden, A. and Reddman, T.: Reassessment of MIPAS age
529 of air trends and variability, *Atmos. Chem. Phys.*, 15, 13161-13176, 2015.
- 530 Hall, T.M., and Plumb, R.A.: Age as a diagnostic of stratospheric transport. *J. Geophys. Res.*,
531 99, 1059-1070, 1994.

532 Hall, T. M., Waugh, D. W., Boering, K. A., and Plumb, R. A.: Evaluation of transport in
533 stratosphere models, *J. Geophys. Res.*, 104, 18815-18839, 1999.

534 Harnisch, J., and Eisenhauer, A.: Natural CF₄ and SF₆ on Earth, *Geophys. Res. Lett.*, 25, 2401-
535 2404, 1998.

536 Hodnebrog, Ø., Etminan, M., Fuglestvedt, J. S., Marston, G., Myhre, G., Nielsen, C. J., Shine,
537 K. P., and Wallington, T. J.: Global warming potentials and radiative efficiencies of
538 halocarbons and related compounds: A comprehensive review, *Rev. Geophys.*, 51, 300-378,
539 10.1002/rog.20013, 2013.

540 Hurley, M. D.: GEISA : 2011 Spectroscopic Database; SF₆ Infrared Absorption Cross-sections,
541 2003.

542 IPCC: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to
543 the Fifth Assessment Report of the Intergovernmental Panel on Climate Change, Cambridge
544 University Press, Cambridge, United Kingdom and New York, NY, USA, 1535 pp., 2013.

545 Jacquinet-Husson, N., Crepeau, L., Armante, R., Boutammine, C., Chédin, A., Scott, N.A.,
546 Crevoisier, C., Capelle, V., Boone, C., Poulet-Crovisier, N., Barbe, A., Campargue, A., Chris
547 Benner, D., Benilan, Y., Bézard, B., Boudon, V., Brown, L.R., Coudert, L.H., Coustenis,
548 A., Dana, V., Devi, V.M., Fally, S., Fayt, A., Flaud, J.-M., Goldman, A., Herman, M., Harris,
549 G.J., Jacquemart, D., Jolly, A., Kleiner, I., Kleinböhl, A., Kwabia-Tchana, F.,
550 Lavrentieva, N., Lacome, N., Li-Hong, Xu, Lyulin, O.M., Mandin, J.-Y., Maki, A.,
551 Mikhailenko, S., Miller, C.E., Mishina, T., Moazzen-Ahmadi, N., H.S.P. Müller, A.
552 Nikitin, J. Orphal, V. Perevalov, Perrin, D.T. Petkie, A. Predoi-Cross, Rinsland, C.P.,
553 Remedios, J.J., Rotger, M., Smith, M.A.H., Sung, K., Tashkun, S., Tennyson, J., Toth, R.A.,
554 Vandaele, A.-C., J. Vander Auwera, J.: The 2009 edition of the GEISA spectroscopic
555 database. *J. Quant. Spectrosc. Radiat. Transfer*, 112, 2395–2445, 2011.

556 Kovács, T., Plane, J. M.C., Feng, W., Nagy, Chipperfield, M. P., Verronen, P. T., Andersson,
557 M. E., Newnham, D. A., Clilverd, M. A., and Marsh, D. R.: *D*-region ion–neutral coupled
558 chemistry (Sodankylä Ion Chemistry, SIC) within the Whole Atmosphere Community
559 Climate Model (WACCM 4) – WACCM-SIC and WACCM-rSIC, *Geosci. Model Dev.*, 9,
560 3123-3136, doi:10.5194/gmd-9-3123-2016, 2016.

561 Lamarque, J.-F., Emmons, L. K., Hess, P. G., Kinnison, D. E., Tilmes, S., Vitt, F., Heald, C.
562 L., Holland, E. A., Lauritzen, P. H., Neu, J., Orlando, J. J., Rasch, P., and Tyndall, G.: CAM-
563 chem: description and evaluation of interactive atmospheric chemistry in CESM., *Geosci.*
564 *Model Dev.*, 5, 369-411, 2012.

565 Maiss, M., and Brenninkmeijer, C. A. M.: Atmospheric SF₆, trends, sources and prospects.,
566 Environ. Sci. Technol., 32, 3077-3086, 1998.

567 Marsh, D. R., Mills, M. J., Kinnison, D. E., and Lamarque, J.-F.: Climate change from 1850 to
568 2005 simulated in CESM1 (WACCM), J. Climate, 26, 7372-7391, 2013.

569 Mayer, and Kylling: Technical note: The libRadtran software package for radiative transfer
570 calculations - description and examples of use, Atmos. Chem. Phys., 5, 1855-1877, 2005.

571 Meredith, N. P., Horne, R. B., Isles, J. D., and Rodriguez, J. V.: Extreme relativistic electron
572 fluxes at geosynchronous orbit: Analysis of GOES E > 2 MeV electrons, Space Weather,
573 13, 170-184, 2015.

574 Morris, R. A., Miller, T. M., Viggiano, A. A., Paulson, J. F., Solomon, S., and Reid, G.: Effects
575 of electron and ion reactions on atmospheric lifetimes of fully fluorinated compounds, J.
576 Geophys. Res., 100, 1287-1294, 1995.

577 Myhre, G., Shindell, D., Bréon, F.-M., Collins, W., Fuglestedt, J., Huang, J., Koch, D.,
578 Lamarque, J.-F., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G., Takemura,
579 T., and Zhang, H.: Anthropogenic and natural radiative forcing, Cambridge (UK), New
580 York (USA), 2013.

581 NOAA Earth System Research Laboratory, Halocarbons and other atmospheric trace gases,
582 SF₆ – Combined Dataset, <http://www.esrl.noaa.gov/gmd/hats/combined/SF6.html>,
583 [Accessed 1/1/2016], 2016.

584 Plane, J. M. C., Feng, W., and Dawkins, E. C. M.: The mesosphere and metals, Chem. Rev.,
585 115, 4497-4541, 2015.

586 Ravishankara, A. R., Solomon, S., Turnipseed, A. A., and Warren, R. F.: Atmospheric lifetimes
587 of long-lived halogenated species, Science, 259, 194-199, 1993.

588 Reddman, T., Ruhnke, R., and W. Kouker, W.: Three-dimensional model simulations of SF₆
589 with mesospheric chemistry, J. Geophys. Res., 106, 14,525–14,537,
590 doi:10.1029/2000JD900700, 2001.

591 Rothman, L. S., Gordon, I. E., Babikov, Y., Barbe, A., Benner, D. C., Bernath, P. F., Birk, M.,
592 Bizzocchi, L., Boudon, V., Brown, L. R., Campargue, A., Chance, K., Cohen, E. A.,
593 Coudert, L. H., Devi, V. M., Drouin, B. J., Faytl, A., Flaud, J. M., Gamache, R. R., Harrison,
594 J. J., Hartmann, J. M., Hill, C., Hodges, J. T., Jacquemart, D., Jolly, A., Lamouroux, J., Roy,
595 R. J. L., Li, G., Long, D. A., Lyulin, O. M., Mackie, C. J., Massie, S. T., Mikhailenko, S.,
596 Müller, H. S. P., Naumenko, O. V., Nikitin, A. V., Orphal, J., Perevalov, V., Perrin, A.,

597 Polovtseva, E. R., Richard, C., Smith, M. A. H., Starikova, E., Sung, K., Tashkun, S.,
598 Tennyson, J., Toon, G. C., Tyuterev, W. G., and Wagner, G.: The HITRAN 2012 Molecular
599 Spectroscopic Database, *J. Quant. Spectrosc. Radiat. Transfer*, 4 - 50, 2012.

600 Sharpe, S.W., Johnson, T.J., Sams, R.L., Chu, P.M., Rhoderick, J.C.: Gas-Phase Databases for
601 Quantitative Infrared Spectroscopy, *Appl Spectrosc.*, 58(12), 1452-1461, 2004.

602 Stamnes, K., Tsay, S. C., Wiscombe, W., and Laszlo, I.: DISORT, a General-Purpose Fortran
603 Program for Discrete-Ordinate-Method Radiative Transfer in Scattering and Emitting
604 Layered Media: Documentation of Methodology, 2000.

605 Stiller, G. P., von Clarmann, T., Höpfner, M., Glatthor, N., Grabowski, U., Kellmann, S.,
606 Kleinert, A., Linden, A., Milz, M., Reddmann, T., Steck, T., Fischer, H., Funke, B., Lopez-
607 Puertas, M., and Engel, A.: Global distribution of mean age of stratospheric air from MIPAS
608 SF₆ measurements, *Atmos Chem Phys*, 8, 677–695, 2008.

609 Stiller, G. P., von Clarmann, T., Haenel, F., Funke, B., Glatthor, N., Grabowski, U., Kellmann,
610 S., Kiefer, M. Linden, A., Lossow, S. and López-Puertas, M.: Observed temporal evolution
611 of global mean age of stratospheric air for the 2002 to 2010 period, *Atmos. Chem. Phys.*,
612 12, 3311-3331, 2012.

613 Tapping, K. E.: The 10.7 cm solar radio flux (F10.7), *Space Weather*, 11, 394-406, 2013.

614 Totterdill, A., Kovács, T., Gomez-Martin, J. C., Feng, W., and Plane, J. M. C.: Mesospheric
615 removal of very long-lived greenhouse gases SF₆ and CFC-115 by metal reactions, Lyman-
616 α photolysis, and electron attachment, *J. Phys. Chem. A*, 115, 2016-2025, 2015.

617 Totterdill, A., Kovács, T., Feng, W., Dhormse, S., Smith, C., Martin, J. C. G., Chipperfield,
618 M., Forster, P., and Plane, J.: Atmospheric Lifetimes, Infrared Absorption Spectra,
619 Radiative Forcings and Global Warming Potentials of NF₃ and CFC-115, *Atmos. Chem.*
620 *Phys.*, 16, 11,451-11,463, doi:10.5194/acp-2016-231, 2016.

621 Troe, J., Miller, T. M., and Viggiano, A. A.: Low-energy Electron Attachment to SF₆. I. Kinetic
622 modeling of nondissociative attachment, *J. Chem. Phys.*, 127, 244303, 2007a.

623 Troe, J., Miller, T. M., and Viggiano, A. A.: Low-energy electron attachment to SF₆. II.
624 Temperature and pressure dependences of dissociative attachment, *J. Chem. Phys.*, 127,
625 244304, 2007b.

626 Varanasi, P. SF₆ Infrared Absorption Cross-Sections. 2001 [Accessed 03/03/2015] Available
627 from: http://ether.ipsl.jussieu.fr/ether/pubipsl/GEISA/geisa_crossIR_frame_2011_uk.jsp.

628 Verronen, P. T., Seppälä, A., Clilverd, M. A., Rodger, C. J., Kyrölä, E., Enell, C.-F., Ulich, T.,
629 and Turunen, E.: Diurnal variation of ozone depletion during the October-November 2003
630 solar proton events, *J. Geophys. Res.*, 110, A09S32, 2005.

631 Viggiano, A. A., Miller, T. M., Friedman, J. F., and Troe, J.: Low-energy electron attachment
632 to SF₆. III. From thermal detachment to the electron affinity of SF₆, *J. Chem. Phys.*, 127,
633 244305, 2007.

634 Waugh, D., and Hall, T: Age of stratospheric air: Theory, observations and models, *Rev*
635 *Geophys.*, 40, doi:10.1029/2000RG000101, 2002.

636

637 **Tables**638 **Table 1.** SF₆ loss reactions included in WACCM.

Loss process	Rate constant	Reference and comments
Na + SF ₆	$k = 1.80 \times 10^{-11} \exp(-590.5/T)$	From Totterdill <i>et al.</i> , (2015) Refitted for mesospheric temperatures 215-300K.
K + SF ₆	$k = 13.4 \times 10^{-11} \exp(-860.6/T)$	From Totterdill <i>et al.</i> , (2015) Refitted for mesospheric temperatures 215-300K.
Electron attachment	<p>Associative attachment:</p> $k_{EA,ass} = k_{at} \times (k_{(SF_6^- + H)}[H] + k_{(SF_6^- + HCl)}[HCl]) / (j_{PD} + k_{(SF_6^- + H)}[H] + k_{(SF_6^- + HCl)}[HCl] + k_{(SF_6^- + O_3)}[O_3] + k_{(SF_6^- + O)}[O])$ <p>Dissociative attachment:</p> $k_{EA,diss} = k_{at} \times \beta,$ <p>where β is the fraction of SF₆⁻ that dissociates into SF₅⁻.</p>	Totterdill <i>et al.</i> , (2015).
Photolysis	<p>Lyman-α: $\sigma(121.6 \text{ nm}) = 1.37 \times 10^{-18} \text{ cm}^2$</p> <p>Parameterised expression over the range of 115-180 nm, based on previous measurements.</p>	Totterdill <i>et al.</i> , (2015).

639

640 **Table 2.** Positive and negative ions included in WACCM-SIC.

Positive ions	O_2^+ , O_4^+ , NO^+ , $NO^+(H_2O)$, $O_2^+(H_2O)$, $H^+(H_2O)$, $H^+(H_2O)_2$, $H^+(H_2O)_3$, $H^+(H_2O)_4$, $H^+(H_2O)_5$, $H^+(H_2O)_6$, $H_3O^+(H_2O)_2(CO_2)$, $H_3O^+(OH)$, $O_2^+(CO_2)$, $H_3O^+(OH)(CO_2)$, $H_3O^+(OH)(H_2O)$, $O_2^+(H_2O)(CO_2)$, $O_2^+(H_2O)_2$, $O_2^+(N_2)$, $NO^+(H_2O)_2$, $H^+(H_2O)(CO_2)$, O^+ , N^+ , N_2^+ , $NO^+(H_2O)_3$, O_4^+ , $H^+(H_2O)_2(CO_2)$, $H^+(H_2O)_2(N_2)$
Negative ions	O_3^- , O^- , O_2^- , OH^- , $O_2^-(H_2O)$, $O_2^-(H_2O)_2$, O_4^- , CO_3^- , $CO_3^-(H_2O)$, CO_4^- , HCO_3^- , NO_2^- , NO_3^- , $NO_3^-(H_2O)$, $NO_3^-(H_2O)_2$, $NO_3^-(HNO_3)$, NO_3^- $(HNO_3)_2$, Cl^- , ClO^- , $NO_2^-(H_2O)$, $Cl^-(H_2O)$, $Cl^-(CO_2)$, $Cl^-(HCl)$

641

642 **Table 3.** Partial (reactions with electrons, photolysis, and metals (K, Na)) and total atmospheric
 643 lifetimes (years) of SF₆ from different studies. Numbers in parentheses show relative
 644 percentage contribution of loss due to the different processes.

Study	Lifetime / years			
	Photolysis	Electron attachment	Total	Model dimensions
Ravishankara <i>et al.</i> (1993)	13,000 (24%)	4210 (76%)	3200	2D
Morris <i>et al.</i> (1995)	N/A	N/A	800	2D
This work	48,000 (2.6%)	1339 (97.4%)	1278	3D

645

646 **Table 4.** Integrated absorption cross-sections for SF₆ measured in this work and ratios with
 647 values obtained by **GEISA** (Jacquinet-Husson et al., 2011; Hurley 2003; Varanasi 2001) and
 648 HITRAN (Rothman et al., 2012).

Band limits (cm ⁻¹)	Integrated cross-section (10 ⁻¹⁶ cm ² molec ⁻¹ cm ⁻¹)	Ratio of integrated cross-sections in this work to previous studies		
		Hurley (2003)	Varanasi (2001)	HITRAN
925 - 955	2.02	1.07	1.01	0.99
650 - 2000	2.40	-	1.09	-

649

650 **Table 5.** Calculated instantaneous and stratospheric adjusted radiative forcings and radiative
 651 efficiencies of SF₆ in clear and all-sky conditions^a.

	Instantaneous		Stratospheric adjusted	
	Clear	All-sky	Clear	All-sky
Radiative forcing (mWm ⁻²)	76.43	48.91	81.81	56.01
Radiative efficiency (Wm ⁻² ppbv ⁻¹)	0.77	0.50	0.85	0.59

652 a. Based on present day atmospheric SF₆ surface concentration of 9.3 pptv.

653 **Table 6.** Comparison of 20, 100 and 500-year global warming potentials for SF₆ from this work
 654 with values from IPCC (2013).

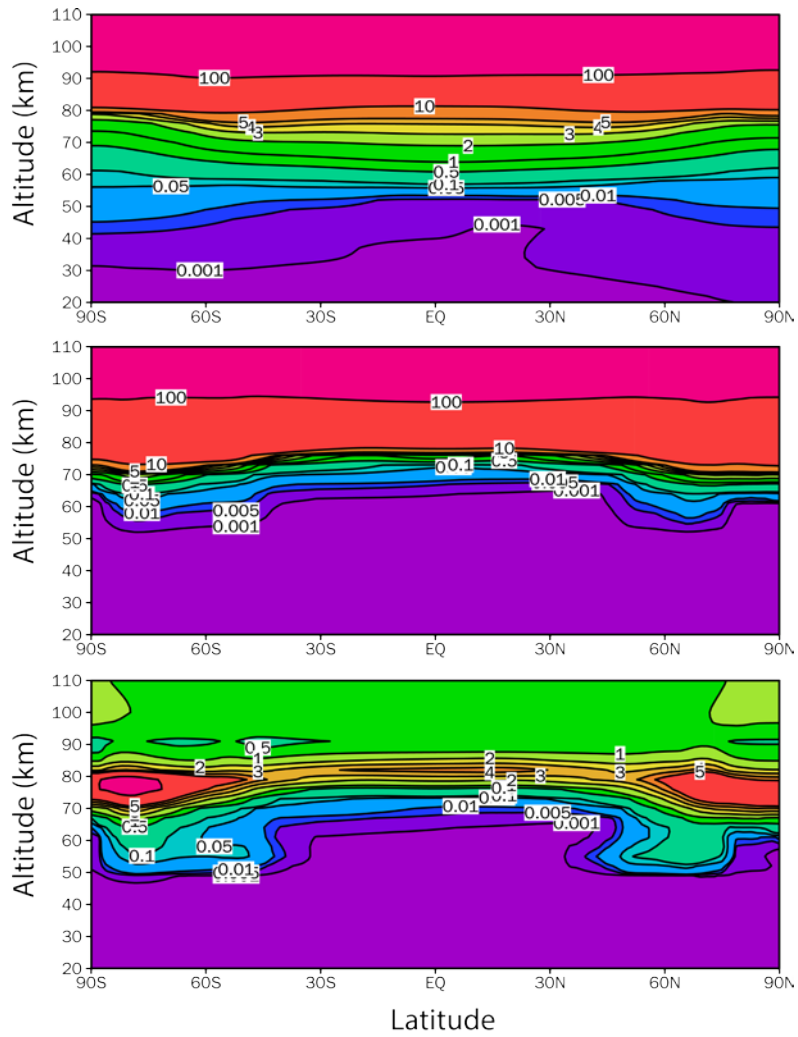
	Global Warming Potential		
	GWP ₂₀	GWP ₁₀₀	GWP ₅₀₀
This work ^a	18000	23700	31300
IPCC (2013) ^b	17500	23500	32600 ^c
Difference (%) (This work – IPCC)	+3%	+1%	-4%

655 ^a Based on our atmospheric lifetime of 1278 yrs and RE of 0.59 Wm⁻² ppbv⁻¹.

656 ^b Based on an atmospheric lifetime of 3200 yrs and RE of 0.57 Wm⁻² ppbv⁻¹.

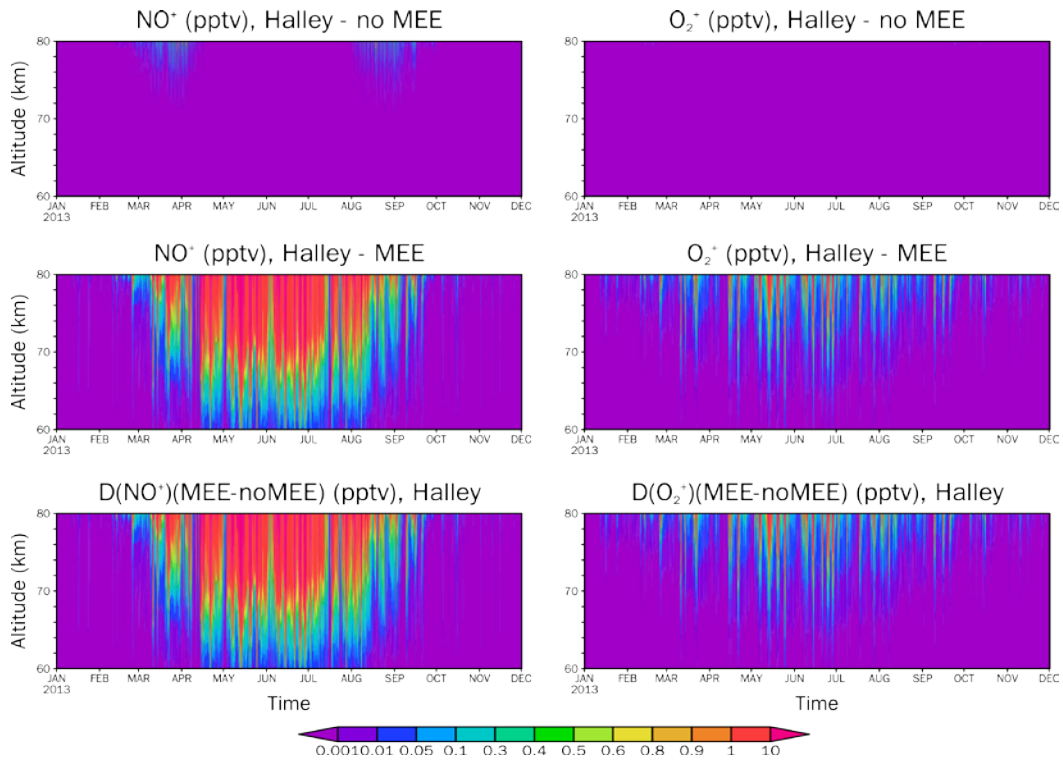
657 ^c Based on an atmospheric lifetime of 3200 yrs and RE of $0.52 \text{ Wm}^{-2} \text{ ppbv}^{-1}$ from IPCC AR4
658 (Forster *et al.*, 2007).

659 **Figures**



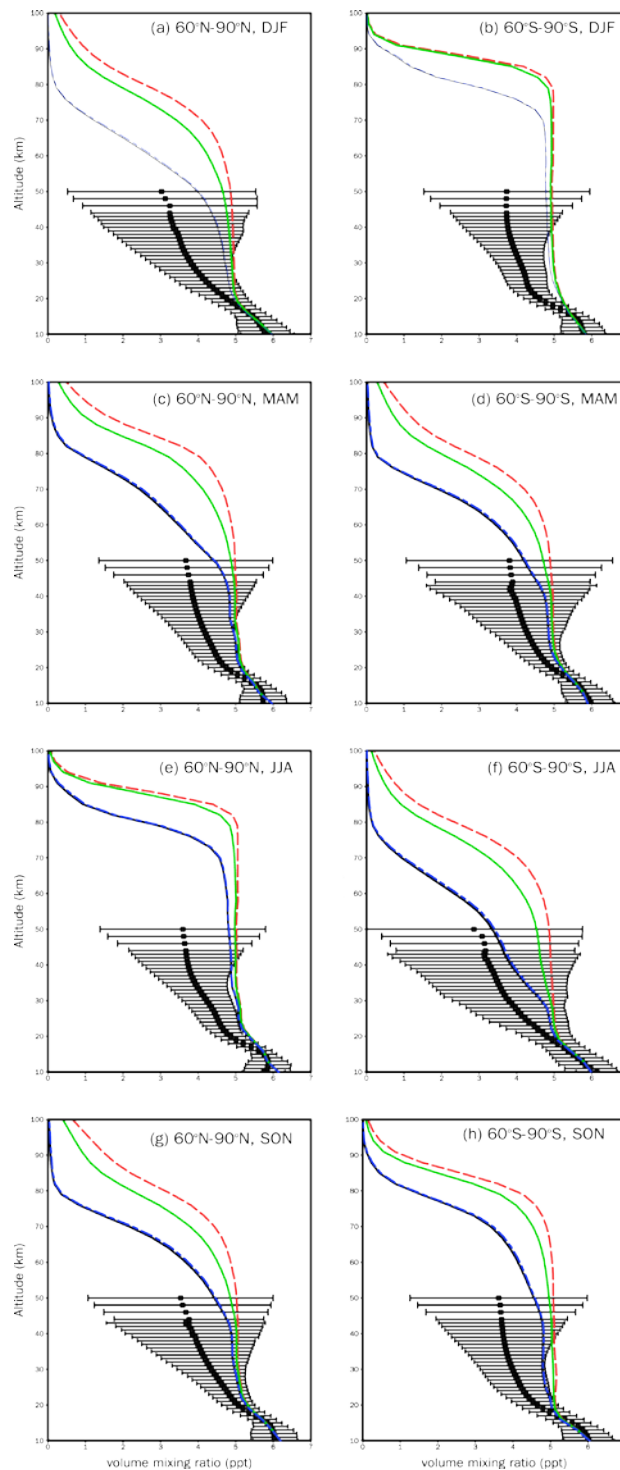
660

661 **Figure 1.** Top: annual average electron concentration for 2013 from the standard WACCM
 662 model (in 10^2 electrons cm^{-3}). Middle: annual average electron concentration for 2013 from
 663 WACCM-SIC model (in 10^2 electrons cm^{-3}). Bottom: annually averaged electron scaling factor
 664 for 2013.



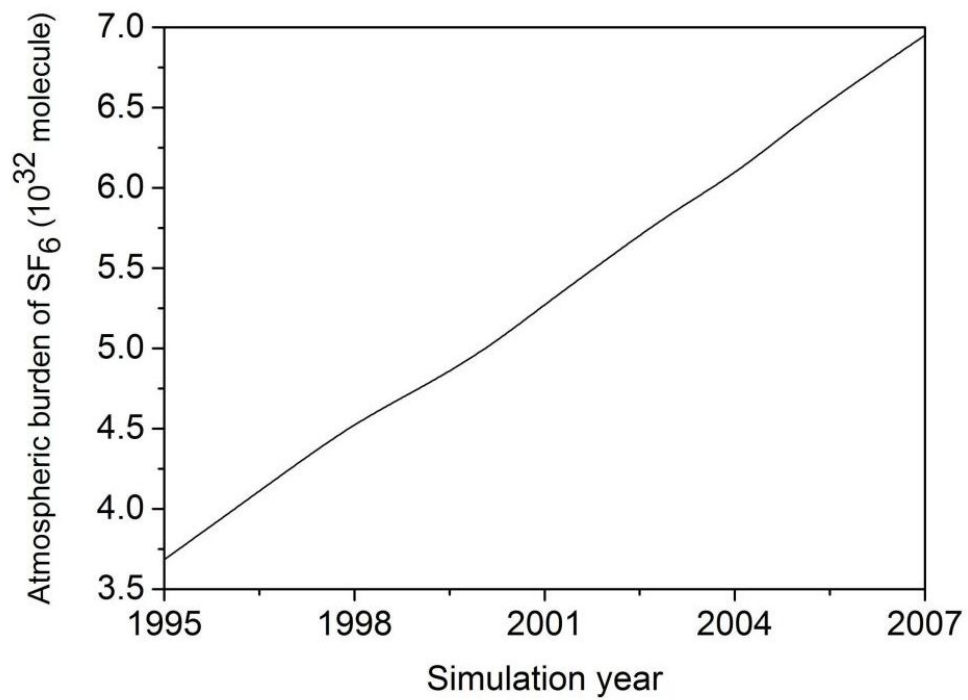
666

667 **Figure 2.** Time series of volume mixing ratio profiles (pptv) of NO^+ (left panels) and O_2^+ (right
 668 panels) above Halley (76°S) from two WACCM-SIC simulations. Top panels show the values
 669 obtained from the model run without medium energy electrons (MEE); the middle panels show
 670 the run with medium energy electrons; and the bottom panels show the absolute differences
 671 between the two model runs.



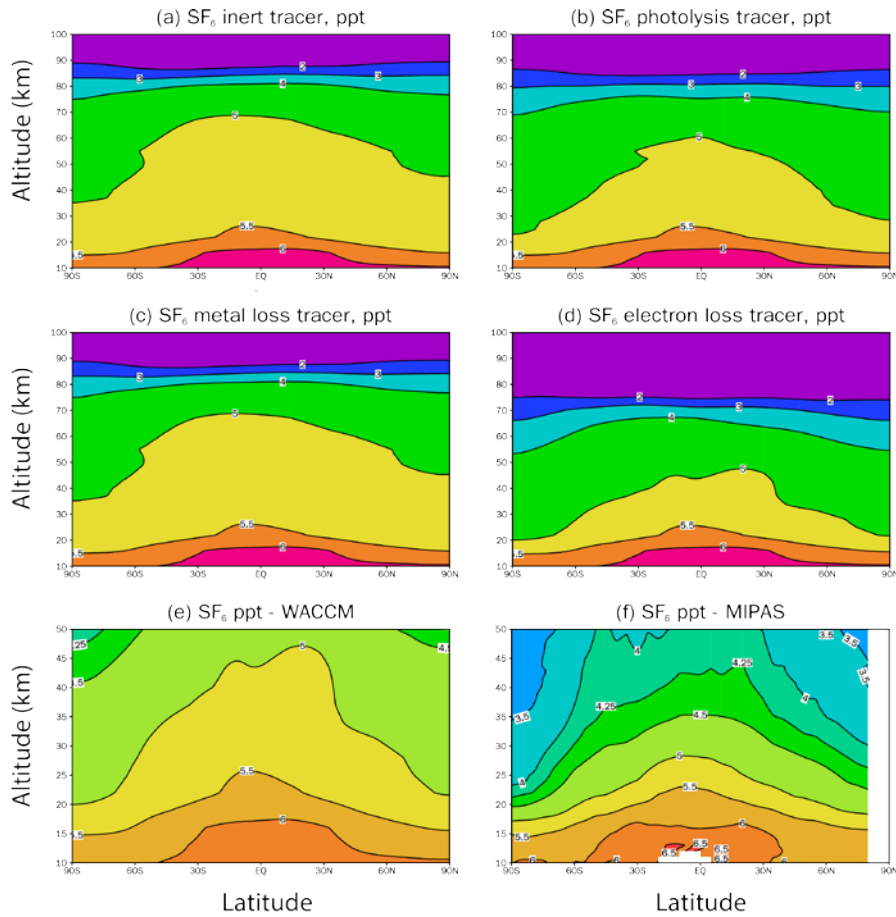
673

674 **Figure 3.** Seasonal mean volume mixing ratios (pptv) of the different SF₆ tracers for the polar
 675 regions (60°N – 90°N and 60°S – 90°S latitudes) in 2007 as a function of altitude for MIPAS
 676 satellite observed SF₆ (black symbols with standard deviations for $\pm 1\sigma$) (Stiller *et al.*, 2012),
 677 the total WACCM-SF₆ (blue solid line), the photolysis WACCM-SF₆ tracer (green solid line)
 678 and the inert WACCM SF₆ tracer (red dashed line).



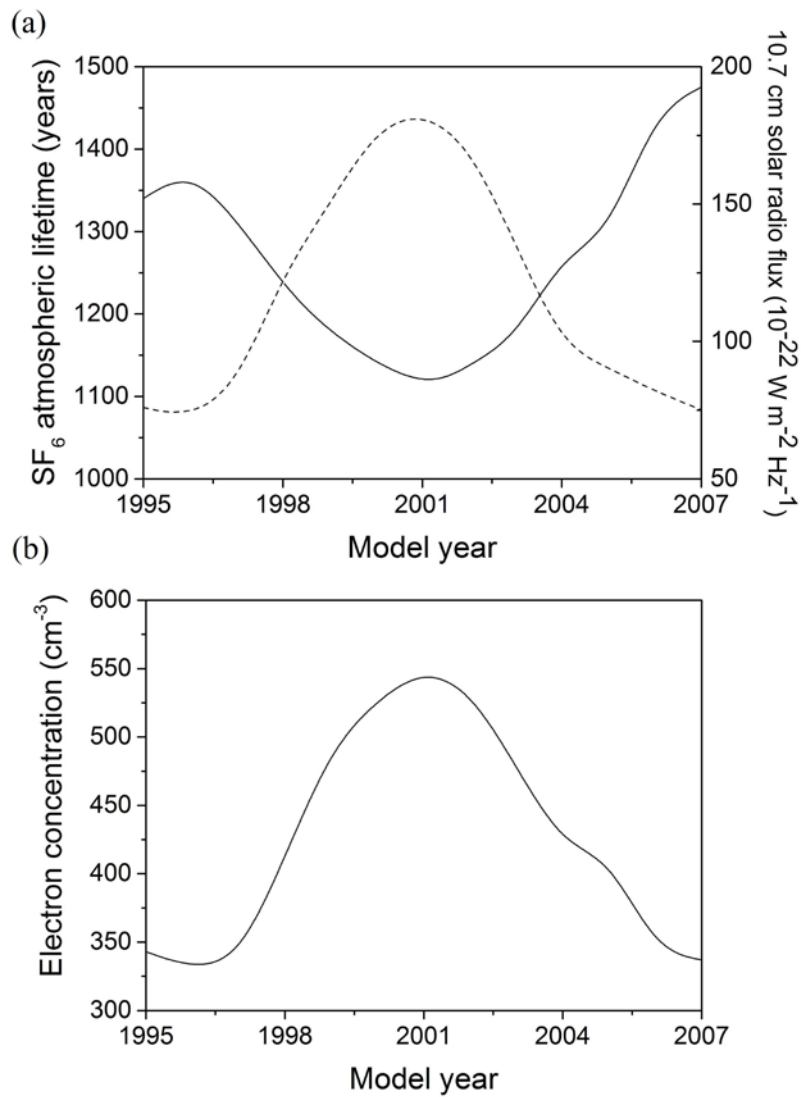
680

681 **Figure 4.** Variation of the total annual atmospheric burden of SF₆ during the simulation from
682 1995 to 2007. According to this the emission rate (slope) was determined to be 6.5×10^{-3}
683 Tg/year, corresponding to 0.29 pptv/year.



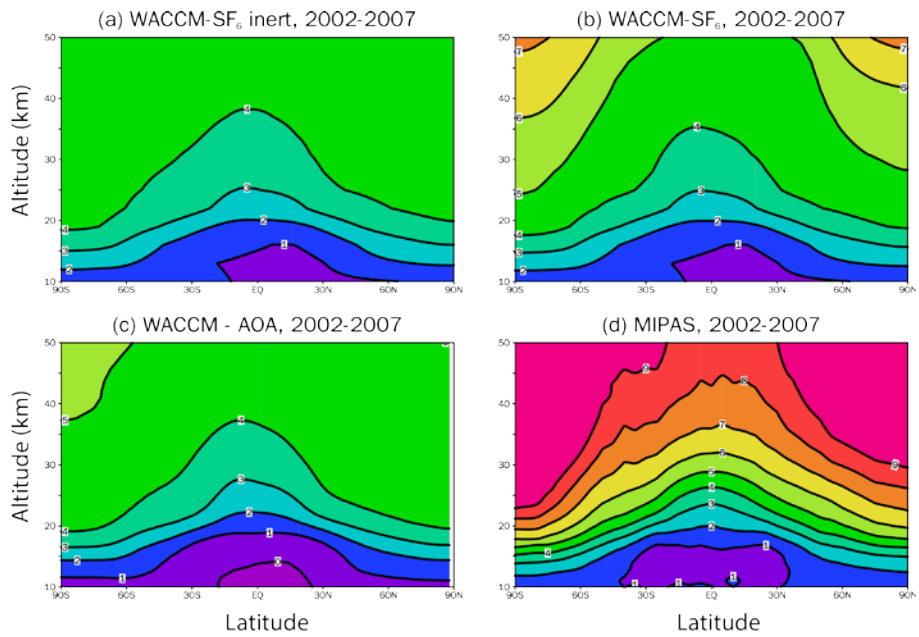
684

685 **Figure 5.** Annual zonal mean latitude-height volume mixing ratios (pptv) of the different
 686 WACCM SF₆ tracers in 2007: (a) inert SF₆ tracer; (b) SF₆ tracer removed by photolysis only;
 687 (c) SF₆ tracer removed by mesospheric metals only; (d) SF₆ tracer removed by electron
 688 attachment only; and (e) total reactive SF₆. Panel (f) shows the SF₆ volume mixing ratio for
 689 2007 from MIPAS observations. Note the different altitude ranges and contour intervals for
 690 panels (a)-(d) versus panels (e)-(f).



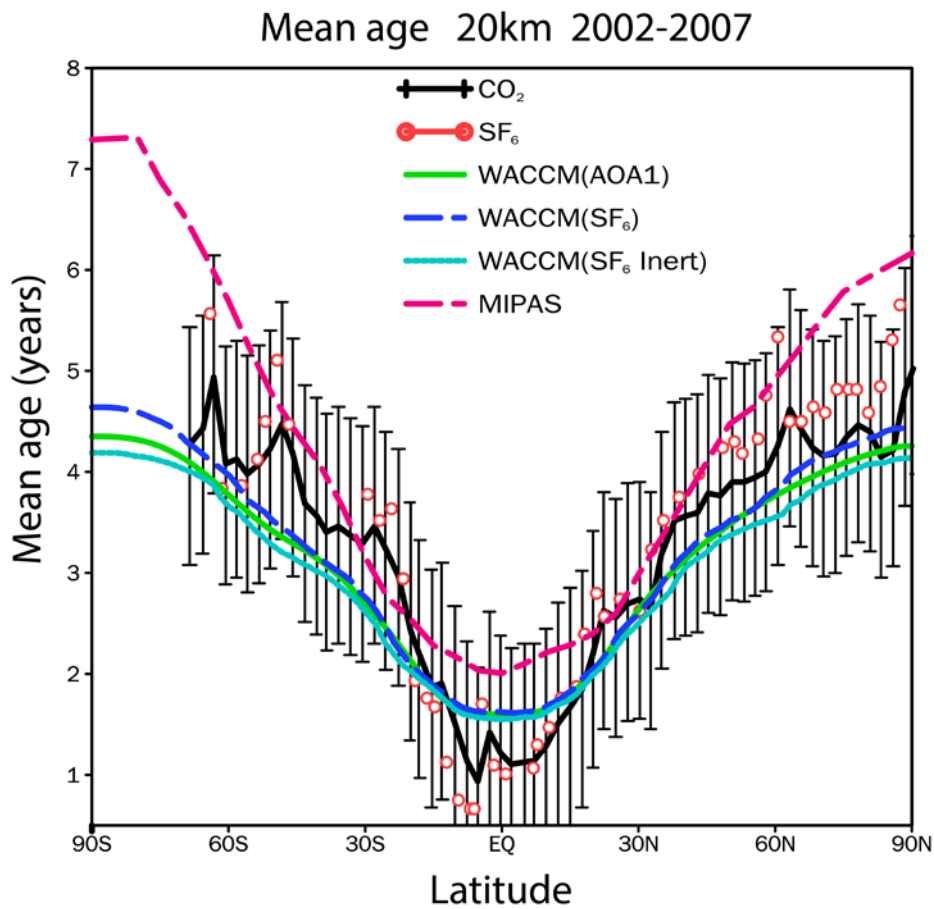
692

693 **Figure 6.** (a) Variation in atmospheric lifetime of SF₆ (solid line) and 10.7 cm solar radio flux
 694 (dashed line) during the WACCM simulation. (b) Variation of the WACCM electron
 695 concentration (cm⁻³) at 80 km, averaged over polar latitudes (60°N – 90°N and 60°S – 90°S).



696

697 **Figure 7.** Annual mean age of stratospheric air (years) for the period of 2002–2007 determined
 698 from a WACCM simulation using: (a) the inert SF₆ tracer; (b) the total reactive SF₆ tracer; (c)
 699 the idealized AOA1 tracer. Panel (d) shows the age values derived for the same period from
 700 our analysis of MIPAS SF₆ observations.

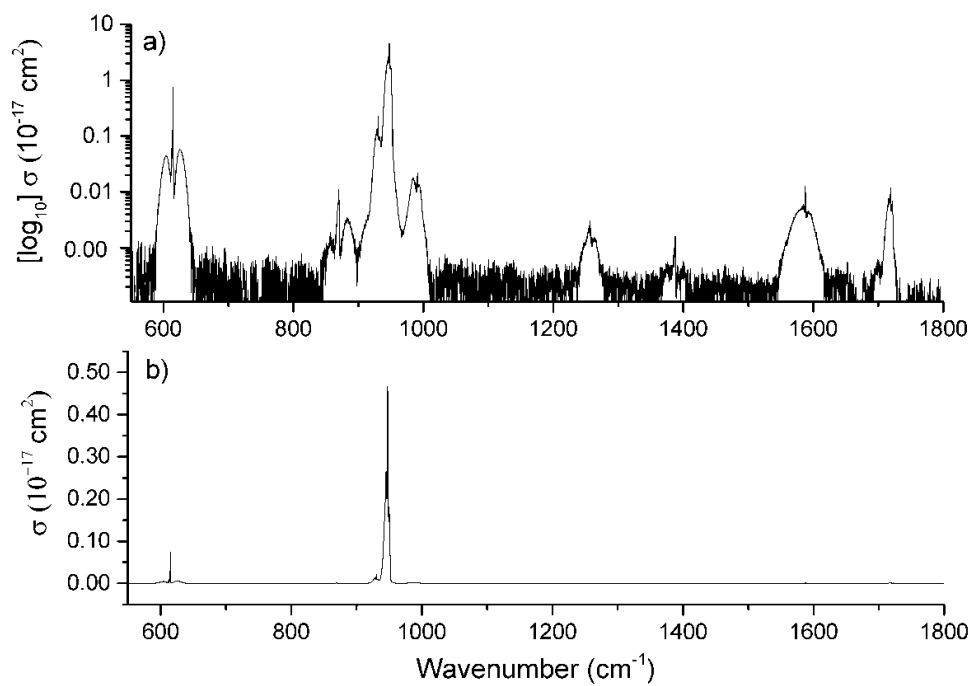


702

703 **Figure 8.** Mean age values at 20 km altitude derived from MIPAS satellite (dashed magenta
 704 line) and ER-2 aircraft observations (SF₆ red open circles, CO₂ black crosses) (Hall *et al.*,
 705 1999). The error bars apply to the age derived from the ER-2 observations. Also shown is the
 706 mean age derived from WACCM tracers: reactive SF₆ (dashed blue line), passive SF₆ (light
 707 blue line) and AOA tracer (solid green line).

708

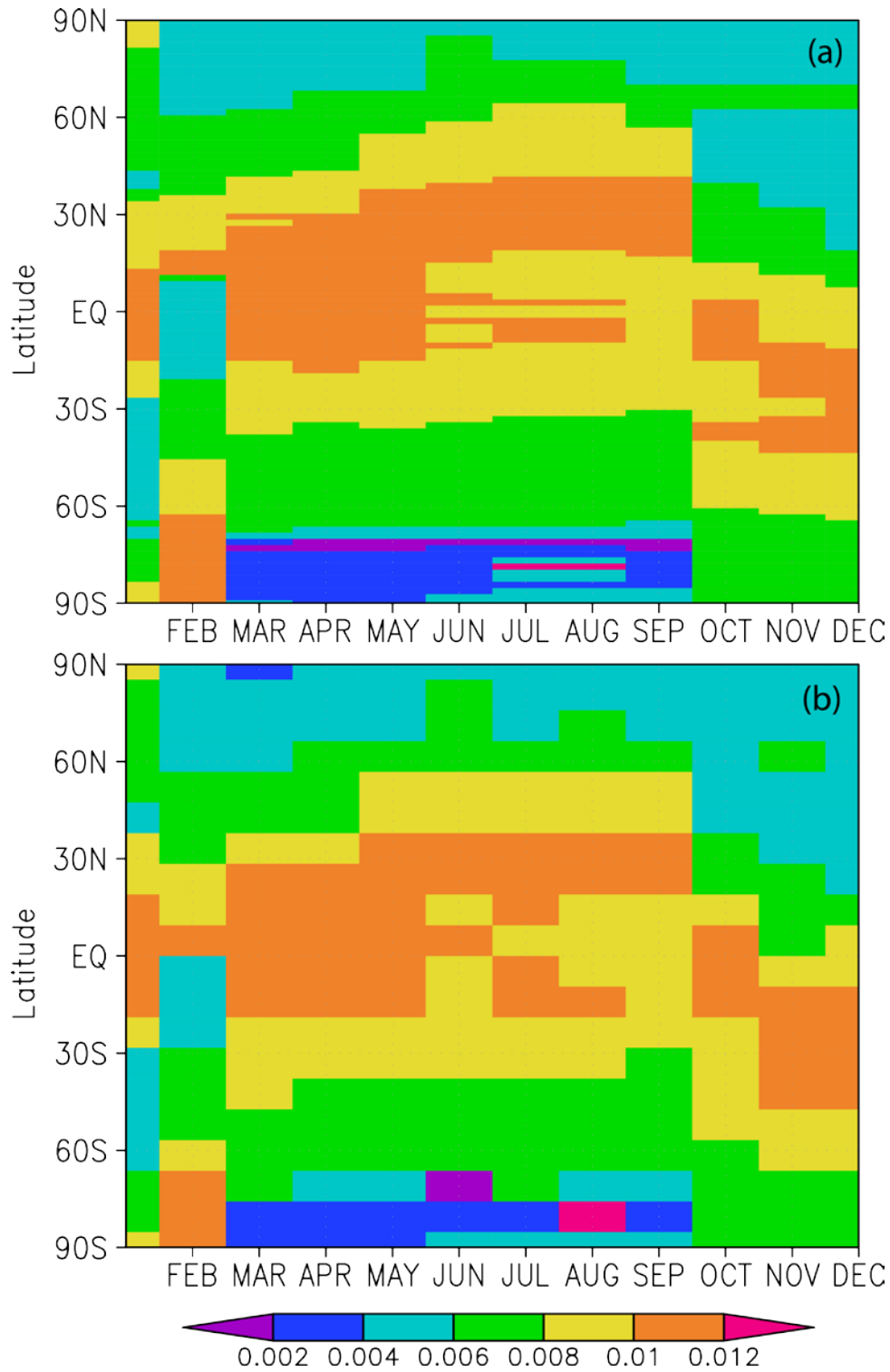
709



710

711

712 **Figure 9.** Infrared absorption spectrum of SF₆ at ~295 K on (a) a logarithmic y axis and (b) a
713 linear y axis. The logarithmic scale in panel (a) highlights the relative positions of the minor
714 bands.



715

716 **Figure 10.** Latitude-time plots for instantaneous radiative forcing (Wm^{-2}) by SF₆ as a function
 717 of latitude and month at (a) high latitude resolution (1.5° spacing) and (b) low latitude
 718 resolution (9° spacing).