

1 **Current Understanding of the Driving Mechanisms for Spatiotemporal**
2 **Variations of Atmospheric Speciated Mercury: A Review**

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14 **Abstract**

15
16 Atmospheric mercury is a global pollutant and thought to be the main source for mercury
17 in oceanic and remote terrestrial systems, where it becomes methylated and bioavailable, and
18 hence atmospheric mercury pollution has global consequences for both human and ecosystem
19 health. Understanding of spatial and temporal variations of atmospheric speciated mercury can
20 advance our knowledge of mercury cycling in various environments. This review summarized
21 spatiotemporal variations of total gaseous mercury or gaseous elemental mercury (TGM/GEM),
22 gaseous oxidized mercury (GOM), and particulate-bound mercury (PBM) in various
23 environments including oceans, continents, high elevation, the free troposphere, and low to high
24 latitudes. In the marine boundary layer (MBL), the oxidation of GEM was generally thought to
25 drive the diurnal and seasonal variations of TGM/GEM and GOM in most oceanic regions,
26 leading to lower GEM and higher GOM from noon to afternoon and higher GEM during winter
27 and higher GOM during spring-summer. At continental sites, the driving mechanisms of
28 TGM/GEM diurnal patterns included surface and local emissions, boundary layer dynamics,
29 GEM oxidation, and for high elevation sites mountain-valley winds, while oxidation of GEM
30 and entrainment of free tropospheric air appeared to control the diurnal patterns of GOM. No
31 pronounced diurnal variation was found for Tekran measured PBM at MBL and continental sites.
32 Seasonal variations in TGM/GEM at continental sites were attributed to increased winter
33 combustion and summertime surface emissions, and monsoons in Asia, while those in GOM
34 controlled by GEM oxidation, free tropospheric transport, anthropogenic emissions, and wet
35 deposition. Increased PBM at continental sites during winter was primarily due to local/regional
36 coal and wood combustion emissions. Long-term TGM measurements from the MBL and
37 continental sites indicated an overall declining trend. Limited measurements suggested

38 TGM/GEM increasing from the southern to northern hemisphere due largely to the vast majority
39 of Hg emissions in the NH, and the latitudinal gradient was insignificant in summer probably as
40 a result of stronger meridional mixing. Aircraft measurements showed no significant vertical
41 variation in GEM over the field campaign regions; however depletion of GEM was observed in
42 stratospherically influenced air masses. In examining the remaining questions and issues,
43 recommendations for future research needs were provided, and among them is the most
44 imminent need for GOM speciation measurements and fundamental understanding of multiphase
45 redox kinetics.

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48 **1. Introduction**

49 Atmospheric mercury (Hg) is a pervasive toxic with comparable natural and
50 anthropogenic sources (UNEP, 2013). It is operationally defined in three forms, gaseous
51 elemental mercury (GEM), gaseous oxidized mercury (GOM), and particulate-bound mercury
52 (PBM). In most environments GEM comprises >95% of total gaseous mercury (TGM =
53 GEM+GOM) with lifetime of 0.5 – 1 year (Driscoll et al., 2013). Besides emissions, GOM and
54 PBM are largely formed from oxidation of GEM, with lifetimes of hours to weeks (Cole et al.,
55 2014). They are highly soluble, and their wet and dry deposition is a major input of Hg to
56 ecosystems and oceans followed by bioaccumulation, where Hg can enter human bodies through
57 the food chain. To ultimately regulate anthropogenic emissions of Hg in order to control the
58 ambient atmospheric concentration of Hg, it is imperative to understand Hg cycling between the
59 atmosphere, ecosystems, and oceans.

60 The pathways of Hg cycling include chemical transformation and transport via air and

61 water in various systems as illustrated in Subir et al. (2011). Mercury can be chemically
62 transformed from one species to another through oxidation/reduction reactions, complex
63 formation, phase transitions, biodegradation, and surface and heterogeneous interactions with
64 aerosols, clouds, snow, and ice. Mercury can also be redistributed between geographic locations
65 and spheres through physical processes such as wind, water runoff, dry and wet deposition, and
66 volatilization. In addition, natural and anthropogenic sources of Hg are distributed vastly uneven
67 as a result of anthropogenic activities and land surface types. The eventual effect of all these
68 processes, some of which are in fact sinks, and sources is manifested in the great heterogeneity
69 of temporal and spatial variations of atmospheric Hg concentrations observed in numerous
70 studies (Sprovieri et al, 2010b, references therein; references in Tables S1 – S7 in the
71 supplementary information (SI)). Characterization and intercomparison of such variations for
72 different geographic and chemical environments can provide a gateway to our understanding of
73 Hg cycling.

74 Numerous measurement studies in the literature have shown distinctly different
75 spatiotemporal variations of GEM, GOM, and PBM in the following environments:

- 76 • Marine boundary layer (MBL)
- 77 • Land: urban, rural, and remote
- 78 • High elevation, high altitude
- 79 • Low, mid-, and high latitudes

80 owing to their respective atmospheric chemical composition, sources, and meteorological
81 conditions. Such differences were attributed to natural and anthropogenic sources of not only Hg
82 but also other reactive chemical compounds that are involved in Hg cycling, meteorological
83 conditions, and chemistry, all of which were highly dependent on geographic locations and

84 surrounding land surface types. Therefore, it is highly complex to delineate the effects of
85 controlling factors determining observed spatiotemporal variations of Hg concentrations.

86 Sprovieri et al. (2010b) reviewed the state of global mercury measurements focusing on
87 instrumentation and techniques, and ranges of concentration levels in studies from different
88 continents and oceanic regions up to 2009. Atmospheric Hg research has since continued to
89 flourish, and in particular longer datasets accumulated in several regions have become available
90 for temporal variability characterization so as to understand the driving mechanisms for such
91 variabilities. Also of importance is the efficacy of emission reductions that have been
92 implemented in North America and Europe for nearly two decades and over shorter periods in
93 East Asia. This paper is, different from Sprovieri et al. (2010b), aimed to provide a global
94 picture of spatiotemporal variations of speciated Hg using measurement-based studies in the
95 literature over ocean, over land, by altitude, and by latitude, and further glean insight on
96 important factors that could potentially contribute to the observed variations.

97 It should be noted that *units were converted for a standard atmosphere* for comparison.
98 One more cautionary note is that Hg data in earlier studies had coarser temporal resolution than
99 in more recent studies, and hence the comparisons should be viewed with this caveat in mind.
100 Though the earlier studies tended to have orders of magnitude larger concentrations, suggesting
101 at higher temporal resolution those concentrations would have been even larger.

102 **2. Marine Boundary Layer**

103 Measured TGM/GEM, GOM, and PBM concentrations in the MBL globally were
104 summarized in Tables S1 – S3 of the supplementary information (SI). Spatiotemporal variations
105 in speciated Hg and the potential causes for these variations were summarized with respect to
106 their ambient concentration levels, continental (including anthropogenic) influence, hemispheric

107 gradient, diurnal to annual cycles, and long term trends, accompanied by discussions on potential
108 causal mechanisms.

109 **2.1 TGM/GEM**

110 TGM and GEM in the MBL atmosphere have been measured since the late 1960s. Near
111 the surface in most environments, except polar springtime and Dead Sea mercury depletion
112 events (MDEs) when strong GEM oxidation occurs, the difference between TGM and GEM was
113 small to negligible (e.g., Temme et al., 2003a; Mao and Talbot, 2012). Concentrations were
114 generally higher in near-coastal regions due largely to anthropogenic influence, which under
115 certain meteorological conditions could extend to even open oceans. Natural emissions
116 including biomass burning, volcanic, and oceanic emissions were suggested to be of influence in
117 some studies. It was also found that meteorological conditions could play important roles in
118 determining ambient concentrations of TGM/GEM via transport, PBL dynamics, and solar
119 radiation, especially in regions nearing emission sources such as the Mediterranean and in
120 springtime Polar Regions. Long term trends have varied over different time periods, speculated
121 to be associated with changing anthropogenic emissions, legacy emissions, and photooxidation.

122 **2.1.1 Concentration Metrics**

123 The mean concentrations of TGM/GEM observed over varying time periods reported
124 from the studies in the literature ranged from 1.05 ng m⁻³ over the *Antarctic* Ocean to 2.34 ng m⁻³
125 over the *West Pacific seas*, as shown in Table S1 (references therein). The concentration
126 averaged for each oceanic region calculated using the values reported from all the studies was
127 the lowest at 1.53 ng m⁻³ over the *Antarctic* Ocean and the largest at 2.36 ng m⁻³ over the *West*
128 *Pacific seas* (**Fig. 1a**). The range of 0.05 – 29 ng m⁻³ over the Atlantic (**Fig. 1a**), obtained from
129 individual studies, appeared to be the largest, although the maximum concentration was from a

130 single event influenced by forest fires in Quebec, Canada at a long term site in the MBL 20 km
131 from the coast of southern New Hampshire, USA (Mao and Talbot, 2012). With that single
132 event removed, the TGM/GEM concentrations were much more variable in the MBL of the
133 *Mediterranean Sea* and its nearby seas (Table S1; references therein).

134 Atmospheric Hg over the *Atlantic* Ocean has been studied most extensively compared to
135 other oceans, largely via shipboard measurements. Concentrations of TGM/GEM ranged from
136 0.05 ng m^{-3} (15-minute average) in Cape Point, South Africa (Brunke et al., 2010) to 29 ng m^{-3}
137 (5-minute average) near the shore of southern New Hampshire, USA (Mao and Talbot, 2012). In
138 the earliest shipboard global study of atmospheric Hg, Seiler et al. (1980) found highly variable
139 TGM concentrations ($1 - 10 \text{ ng m}^{-3}$, 2-4 h average) averaged at 2.8 ng m^{-3} between Hamburg
140 (54°N , 10°E) and Santo Domingo (20°N , 67°W) across the Atlantic Ocean over 11 October - 1
141 November 1973. It should be noted that early studies used very different measurement
142 techniques and hence the magnitude needs to be considered with discretion. During the following
143 40 years, most studies reported TGM/GEM ranging from below LOD to a few ng m^{-3} and higher
144 concentrations in near-coastal regions (Table S1; references therein). The *first* measurements of
145 Hg species was a one month shipboard study over the *South Atlantic* Ocean during polar summer
146 (February) 2001 by Temme et al. (2003b). Their measurements (5-min – 15-min average data)
147 during the cruise from Neumayer to Punta Arenas exhibited very small variation with TGM
148 averaged at $1.1 \pm 0.2 \text{ ng m}^{-3}$ and no significant difference between TGM and GEM. Relatively
149 homogeneous distributions of TGM/GEM were observed over open waters in the *South Atlantic*
150 with mean values hovering around 1 ng m^{-3} and standard deviation $< 0.3 \text{ ng m}^{-3}$ compared to
151 larger mean values ($1.3 - \sim 3 \text{ ng m}^{-3}$) over the *North Atlantic*.

152 Over the *Pacific* Ocean, 1-min to 15-min TGM/GEM concentrations measured over the

153 North and South *Pacific* Ocean ranged from 0.3 ng m⁻³ over 40° – 45°N in July – September
154 2008 (Kang and Xie, 2011) to 7.21 ng m⁻³ in the Los Angeles Port on 27 May 2010 (Weiss-
155 Penzias et al., 2013), with generally higher concentrations near coasts and lower ones over open
156 oceans (Table S1; reference therein). The distribution of TGM/GEM over the South *Pacific*
157 appeared to be quite heterogeneous, where Xia et al. (2010) measured TGM averaged at
158 2.20±0.67 ng m⁻³, a factor of 2 higher than those in Soerensen et al. (2010) that measured a mean
159 of 1.03 ±0.16 ng m⁻³.

160 Over the *South China Sea, the Yellow Sea*, and other neighboring seas, located on the
161 Eastern Asian continental margin in the tropical-subtropical western North Pacific, elevated
162 concentrations of TGM/GEM were observed with mean values varying over 2.08 – 2.62 ng m⁻³
163 (Fu et al., 2010; Nguyen et al., 2011; Ci et al., 2011) (Table S1). TGM concentrations over the
164 *Mediterranean Sea, Adriatic Sea, Dead Sea, Augusta Basin*, and *Baltic Sea* ranged from 0.4 to
165 11 ng m⁻³ (Table S1; references therein).

166 A few studies on Hg over the *Indian* Ocean (Soerensen et al., 2010; Xia et al., 2010; Witt
167 et al.; 2010; Angot et al., 2014) reported a concentration gradient of TGM with increasing
168 concentrations at more northern locations closer to the inter-tropical convergence zone (ITCZ),
169 with a mean concentration of 1.24±0.06 ng m⁻³ over 9°S - 21°S latitudes (Witt et al., 2010).

170 Studies on TGM/GEM over the *Arctic* Ocean showed fairly constant concentrations in
171 January and August – December and reported MDEs in spring and summertime annual
172 maximums (Lindberg et al., 2002; Aspino et al., 2006; Sommar et al., 2010; Steffen et al., 2013;
173 Yu et al., 2014). During the 1998 – 2001 Barrow Atmospheric Mercury Study (BAMS), daily
174 average GEM concentrations ranged from <0.2 to ~3.7 ng m⁻³, averaged between 1.5 – 2 ng m⁻³
175 in January and mid-August – December (Lindberg et al., 2002). The means and ranges measured

176 in summer 2004, 2005, and 2012 (Aspmo et al., 2006; Sommar et al., 2010; Yu et al., 2014) were
177 well within the 1999 summertime range of Lindberg et al. (2002) (Table S1). Different
178 concentrations of GEM over sea ice-covered ($1.81 \pm 0.43 \text{ ng m}^{-3}$) vs. sea ice-free ($1.55 \pm 0.21 \text{ ng}$
179 m^{-3}) Arctic Oceanic waters were measured by Sommar et al. (2010) in summer 2005. In spring
180 2009 (14 – 26 March) a mean 5-min GEM concentration of 0.59 ng m^{-3} was measured with a
181 range of $0.01\text{--}1.51 \text{ ng m}^{-3}$ over sea ice on the Beaufort Sea near Barrow, Alaska, which appeared
182 to be depleted compared to annual Arctic ambient boundary layer concentrations (Steffen et al.,
183 2013).

184 In the *Antarctica*, the first study, conducted by de More et al. (1993), reported a mean
185 TGM concentration of $0.55 \pm 0.28 \text{ ng m}^{-3}$ and a range of $0.02 - 1.85 \text{ ng m}^{-3}$ (24-48 h) at Ross
186 Island during 1987 – 1989. Over November 2000 – January 2001, Sprovieri et al. (2002)
187 reported a similar range but a mean of $0.9 \pm 0.3 \text{ ng m}^{-3}$, twice larger than that of de More (1993) a
188 decade earlier. Similar means and ranges of TGM/GEM concentrations were measured by
189 Ebinghaus et al. (2002b), Temme et al. (2003b), Soerensen et al. (2010), and Xia et al. (2010).
190 Similar mean values but a much wider range ($0.02 - 3.07 \text{ ng m}^{-3}$) were found in the multi-year
191 dataset in Pfaffhuber et al. (2012) (Table S1).

192 2.1.2 Hemispheric Difference

193 Hemispheric gradient over the *Atlantic* and *Pacific* Ocean has been reported since the
194 1980s, with higher concentrations in the North Atlantic attributed to anthropogenic and biomass
195 burning emissions (Seiler et al, 1980; Slemr et al., 1981, 1985, 1995; Slemr and Langer, 1992;
196 Fitzgerald et al., 1996; Lamborg et al., 1999; Temme et al., 2003a; Chand et al., 2008; Xia et al.,
197 2010; Soerensen et al., 2010; Müller et al., 2012). An average gradient of 0.37 ng m^{-3} in TGM
198 was measured in October – November 1973 (Seiler et al., 1980). Measurements from the same

199 cruise paths from Hamburg (54°N) to Buenos Aires (35°S) in 1977, 1978 – 1980, 1992, and
200 1994 consistently showed TGM hemispheric difference, 1.56 ± 0.32 and 1.05 ± 0.22 ng m⁻³ in the
201 NH and SH, respectively, in 1977, increased to 2.25 ± 0.41 and 1.50 ± 0.30 ng m⁻³ in 1992
202 followed by significant decreases to 1.79 ± 0.41 and 1.18 ± 0.17 ng m⁻³ in 1994 (Slemr et al., 1981,
203 1985, 1995; Slemr and Langer, 1992). The hemispheric difference from a NH average of
204 1.32 ± 0.16 ng m⁻³ in summer 2006 and 2.61 ± 0.36 ng m⁻³ in spring 2007, and a SH average of
205 1.27 ± 0.2 ng m⁻³ measured by Soerensen et al. (2010) was close to the 1978 – 1980 hemispheric
206 gradient in Slemr et al. (1985) but lower than the 1990 value in Slemr and Langer (1992).

207 Over the *Pacific* a hemispheric gradient of ~ 0.4 ng m⁻³ was found in early studies by
208 Seiler et al. (1980) and Fitzgerald et al. (1984). Higher concentrations but similar magnitude of
209 hemispheric difference of TGM was measured in December 2007 by Xia et al. (2010) with a
210 mean of 1.746 ± 0.513 ng m⁻³ over the *North Pacific* and 1.471 ± 0.842 ng m⁻³ over the *South*
211 *Indian* Ocean (Note: their cruise passed through the South Indian instead the South Pacific).
212 Around the same time, Soerensen et al. (2010) measured nearly twice lower concentrations over
213 the *South Pacific* (1.11 ± 0.11 ng m⁻³ along the Chilean Coast and up to 1.33 ± 0.24 ng m⁻³ near
214 East Australia) than the *North Atlantic* concentrations (mean values of 2.26 and 2.86 ng m⁻³ over
215 23°N – 59°N; no measurements over the *North Pacific* in the study) from the same study.

216 Studies found higher TGM concentrations up to ~ 2.3 ng m⁻³ over the *equatorial Pacific*
217 in October 1980, markedly higher (>0.5 ng m⁻³) than those outside this region (Fitzgerald et al.,
218 1984; Kim and Fitzgerald, 1988). However, Wang et al. (2014) found no sustained high GEM
219 concentrations indicative of persistently enhanced biotic mercury evasion from the upwelling
220 region over the Galápagos Islands in the equatorial Pacific during February – October 2011.
221 They found GEM concentrations averaged at 1.08 ± 0.17 ng m⁻³, twice lower than the earlier ones.

222 2.1.3 Temporal Variations from Diurnal Cycle to Long-term Trend

223 2.1.3.1 Diurnal variation

224 Early studies on TGM over the *Atlantic* Ocean showed one order of magnitude larger
225 diurnal amplitude than that in more recent studies, with daily peaks of 5 ng m^{-3} at noon and
226 amplitude of $2\text{-}3 \text{ ng m}^{-3}$ across the North and South Atlantic in Seiler et al. (1980). Yet none was
227 observed by Slemr et al. (1981, 1985) and Slemr and Langer (1992). Measurements of TGM at
228 Cape Point, South Africa (Brunke et al., 2010) and GEM at Appledore Island, Maine, USA (Mao
229 and Talbot, 2012) exhibited pronounced diurnal variation in summer with daily peaks
230 (minimums) before sunrise (in the late afternoon) and amplitudes of 0.8 ng m^{-3} and $\sim 10 \text{ ppqv}$
231 ($\sim 0.09 \text{ ng m}^{-3}$) for the two sites, respectively.

232 The opposite diurnal pattern with significant amplitude was observed over the *Pacific*
233 (Fitzgerald et al., 1984; Weiss-Penzias et al., 2003, 2013; Kang and Xie, 2011; Tseng et al., 2012;
234 Wang et al., 2014) with daily peaks ranging from 0.7 ng m^{-3} (5-min) over the Japan Sea (Kang
235 and Xie, 2011) to 2.25 ng m^{-3} (unknown time resolution) in the equatorial region (Fitzgerald et
236 al., 1984). The most pronounced diurnal variation in TGM was reported in Fitzgerald et al. (1984)
237 with daily amplitude of 0.7 ng m^{-3} in the equatorial region ($4^\circ\text{N} - 10^\circ\text{S}$). Similar pattern and
238 magnitude of GEM diurnal variation was observed by Tseng et al. (2012) over the *South China*
239 *Sea* during May 2003 – December 2005, especially in warm seasons. Opposite patterns were
240 observed in Weiss-Penzias et al. (2003, 2013). Laurier et al. (2003) found no diurnal variation
241 during a cruise from Osaka, Japan to Honolulu, Hawaii over 1 May 2002 – 4 June 2002.

242 Over the *Arctic* diurnal variation of GEM was observed by Lindberg et al. (2002) with
243 noontime minimums in spring and summer, diurnal amplitude $\sim 2 \text{ ng m}^{-3}$ on a typical day in
244 January – June. On the other hand, the shipboard measurements from Sommar et al. (2010)

245 suggested very small near none diurnal variation. Similarly, no diurnal variation was found over
246 the *Antarctica* (Pfaffhuber et al., 2012), except one case with influence of in situ human activity.

247 2.1.3.2 Seasonal to Annual Variation

248 Annual cycles of TGM/GEM were reported over *the Atlantic* in both hemispheres.

249 Annual cycles with an annual maximum in austral summer and a minimum in austral winter and
250 average amplitude of 0.134 ng m^{-3} were observed at Cape Point, South Africa (Slemr et al., 2008;
251 Brunke et al., 2010). Opposite annual variation with higher (lower) concentrations in winter
252 (summer) was measured over the *North Atlantic*, such as Mace Head (amplitude 0.097 ng m^{-3}), a
253 remote site on the west coast of Ireland adjacent to the North Atlantic (Ebinghaus et al., 2002a)
254 and the Appledore Island (25 ppqv, i.e. $\sim 0.2 \text{ ng m}^{-3}$) site in Mao and Talbot (2012). Similarly,
255 significant seasonal variation in NH with an annual minimum in July and maximum in January –
256 March and amplitude of $0.3 - 0.4 \text{ ng m}^{-3}$ was measured in a global cruise (Soerensen et al., 2010).

257 Average seasonal difference of 0.19 ng m^{-3} GEM concentrations over the *Pacific* were
258 observed by Wang et al. (2014) with the highest and most variable concentrations over February
259 – May 2011 and the lowest and least variable in October over the Galápagos Islands during 12
260 November 2011 – 11 December 2011. In contrast, a *lack* of seasonal variation in GEM was
261 reported by Weiss-Penzias et al. (2003) using a subset of data of marine origin extracted from
262 one year speciated Hg data (May 2001 – May 2002) at the Cheeka Peak Observatory on the east
263 coast of the Pacific. This was uncharacteristic of midlatitudinal NH sites, but significant
264 interannual variation was noted in this study.

265 Distinct annual variation in GEM over the *South China Sea* was observed in the cruise
266 study by Tseng et al. (2012) over May 2003 – December 2005. The winter maximum was
267 $5.7 \pm 0.2 \text{ ng m}^{-3}$ and summer minimum $2.8 \pm 0.2 \text{ ng m}^{-3}$, 2-3 times higher than global background

268 levels. Difference of 0.4 ng m^{-3} in seasonal average GEM was quantified with higher
269 concentrations in the summer than in the autumn over the *Adriatic Sea* (Sprovieri et al., 2010)
270 and a factor of two less over the *Augusta Basin* (Bagnato et al., 2013). The study by Obrist et al.
271 (2011) was the first to show the occurrence of mercury depletion events (MDEs) in midlatitudes
272 with GEM down to 22 ppqv (0.2 ng m^{-3}) most frequently in summer in the boundary layer of the
273 *Dead Sea*, as opposed to MDEs, as commonly known, occurring in the springtime *Arctic* and
274 *Antarctic* only.

275 Annual variation of GEM over the *Indian Ocean* was reported in Angot et al. (2014) with
276 higher concentrations in winter ($1.06 \pm 0.09 \text{ ng m}^{-3}$) and lower in summer ($1.04 \pm 0.07 \text{ ng m}^{-3}$),
277 opposite of those at Cape Point (Slemr et al., 2008) and Galapagos Islands (Wang et al., 2014)
278 with annual amplitude an order of magnitude smaller.

279 Annual maximum concentrations of GEM occurred in summer over the *Arctic Ocean* and
280 frequent MDEs with GEM depleted to near zero in spring (Lindberg et al., 2002; Aspmo et al.,
281 2006; Cole et al., 2013; Moore et al., 2013). Lindberg et al. (2002) observed GEM
282 concentrations up to 4 ng m^{-3} in June 2000 compared to $1.82 \pm 0.24 \text{ ng m}^{-3}$ in summer 2004
283 (Aspmo et al., 2006) and $1.23 \pm 0.61 \text{ ng m}^{-3}$ in summer 2012 (Yu et al., 2014).

284 Seasonal variation in *Antarctic Hg* suggested large variation in TGM/GEM in spring due
285 to the occurrence of MDEs. The longest continuous data record in the Antarctic started in
286 February 2007 at the Norwegian Antarctic Troll Research Station (TRS) in Queen Maud Land
287 near the Antarctic coast (Pfaffhuber et al., 2012). Concentrations were fairly constant hovering
288 at $\sim 1 \pm 0.07 \text{ ng m}^{-3}$ in late fall through winter and highly variable ranging from 0.02 to 3.04 ng m^{-3}
289 with a mean of $0.86 \pm 0.24 \text{ ng m}^{-3}$ in spring and summer (Pfaffhuber et al., 2012), close to the
290 values from 6 years earlier in Sprovieri et al. (2002) and Temme et al. (2003b).

291 2.1.3.3 Long-term Trends

292 Long-term trends in TGM over the *Atlantic* varied during different time periods of the
293 past decades. TGM concentrations averaged over latitudes from Hamburg, Germany to Punta
294 Arenas, Chile were increasing at a rate of $1.46 \pm 0.17\% \text{ yr}^{-1}$ from 1970 to 1990 (Slemr and Langer,
295 1992) followed by a 22% decrease from 1990 to 1994 (Slemr et al., 1995). In similar latitudinal
296 coverage but over a wider longitudinal span during three cruises in September – November 1996,
297 December 1999 – March 2000, and February 2001, TGM concentrations were averaged at
298 $1.26 \pm 0.1 \text{ ng m}^{-3}$ (Temme et al., 2003a), comparable to the 1977 – 1980 (Slemr et al., 1985) and
299 1994 concentrations (Slemr et al., 1995) but lower than the 1990 ones (Slemr et al., 1992). Over
300 September 1995 – December 2001, a slight increase (4%) in TGM was observed at Mace Head
301 (Ebinghaus et al., 2002a). In the *South Atlantic* at Cape Point a small but significant decrease
302 was reported in TGM annual median from 1.29 ng m^{-3} in 1996 to 1.19 ng m^{-3} in 2004 (Slemr et
303 al., 2008), and at an about three times faster decreasing rate ($-0.034 \pm 0.005 \text{ ng m}^{-3} \text{ yr}^{-1}$) over 1996
304 – 2008 (Slemr et al., 2011). A statistically significant decreasing trend of $-0.028 \pm 0.01 \text{ ng m}^{-3} \text{ yr}^{-1}$
305 ($\sim 1.6\text{-}2.0\% \text{ yr}^{-1}$) in TGM over the *North Atlantic* was reported for the same time period at Mace
306 Head, Ireland (Ebinghaus et al., 2011). In an updated study, Weigelt et al. (2015) presented a
307 relatively smaller decreasing trend of $-0.016 \pm 0.002 \text{ ng m}^{-3} \text{ yr}^{-1}$ in monthly median marine GEM
308 concentrations over 1996 – 2013. In Soerensen et al. (2012) a steep 1990–2009 decline of -
309 $0.046 \pm 0.010 \text{ ng m}^{-3} \text{ yr}^{-1}$ ($-2.5\% \text{ yr}^{-1}$) was found in TGM over the *North Atlantic* (steeper than at
310 NH land sites) but no significant decline over the *South Atlantic*. A recent comparison by Slemr
311 et al. (2015) found smaller trends during shorter time periods and a possible increasing trend at
312 Cape Point for the period 2007–2013, qualitatively consistent with the trend changes observed at
313 Mace Head (Weigelt et al., 2015).

314 Over the *Arctic* Ocean, weak or insignificant declines in TGM at rates of -0.007 ± 0.019
315 and -0.003 ± 0.012 ng m⁻³ yr⁻¹ were found at Alert and Zeppelin, respectively, during 2000 – 2009,
316 significantly smaller than the trends at midlatitude sites (Ebinghaus et al., 2011; Slemr et al.,
317 2011; Soerensen et al., 2012; Cole et al., 2013; Berg et al., 2013; Weigelt et al., 2015).
318 TGM/GEM concentrations over the *Antarctic* Ocean appeared to have increased from the 1980s
319 to the 2000s (Ebinghaus et al., 2002b; Temme et al., 2003b; Soerensen et al., 2010; Xia et al.,
320 2010; Pfaffhuber et al., 2012), and no significant trend was detected over 2007 – 2013 (Slemr et
321 al., 2015).

322 2.1.4 Mechanisms Driving the Observed Temporal Variabilities

323 2.1.4.1 Causes for Episodic Higher Concentrations

324 It has been hypothesized that anthropogenic, biomass burning, and volcanic emissions
325 caused higher concentrations over open waters and near-coastal regions in many cases. Such
326 influences on the atmospheric concentration of Hg was demonstrated using backward trajectories
327 and correlations of TGM/GEM with carbon monoxide (CO), ²²²Rn, black carbon, sulfur dioxide
328 (SO₂), and dimethylsulfide (DMS) (Williston, 1968; Seiler et al., 1980; Fitzgerald et al., 1981;
329 Fitzgerald et al., 1984; Kim and Fitzgerald, 1988; Slemr et al., 1981; Slemr et al., 1985; Slemr
330 and Langer, 1992; Slemr, 1996; Lamborg et al., 1998; Sheu and Mason, 2001; Laurier and
331 Mason, 2007; Soerensen et al., 2010; Mao and Talbot, 2012; Müller et al., 2012; Xia et al., 2010;
332 Chand et al., 2008; Kang and Xie, 2011; Weiss-Penzias et al., 2013; Fu et al., 2010; Nguyen et
333 al., 2011; Ci et al., 2011; Bagnato et al., 2013; Kotnik et al., 2014). Some studies also suggested
334 that oceanic evasion was an important source contributing to higher concentrations (Seiler et al.,
335 1980; Pirrone et al., 2003; Sigler et al., 2009b), while others thought otherwise (Slemr et al.,
336 1981, 1985; Slemr and Langer, 1992). Strong photoreduction could have caused higher

337 TGM/GEM concentrations under sunny, warm and dry conditions with lower amounts of
338 precipitation in the Mediterranean Sea region (Pirrone et al., 2003; Sprovieri et al., 2003;
339 Sprovieri and Pirrone, 2008). These influences often occurred in multitude simultaneously
340 leading to elevated ambient Hg concentrations.

341 2.1.4.2 Diurnal Variation

342 Nearly in all studies diurnal variation over the *Atlantic*, *Pacific*, and *Arctic* was found to
343 be most pronounced in warm seasons, i.e. spring and/or summer. Different combinations of
344 oceanic emissions, photooxidation, biological production, and meteorology were suggested to
345 work together shaping the observed patterns in different oceanic regions. The pattern with
346 daytime peaks was attributed to oceanic emissions and biological production in sea water (Seiler
347 et al., 1980; Fitzgerald et al., 1984; Tseng et al.; 2012; Wang et al., 2014), which was supported
348 by the concurrent measurements of dissolved elemental Hg (Tseng et al., 2012). The opposite
349 pattern with daytime minimums was associated with photooxidation of GEM by abundant
350 halogen radicals and meteorological conditions (Lindberg et al., 2002; Brunke et al., 2010; Mao
351 and Talbot, 2012; Weiss-Penzias et al., 2003, 2013). The most pronounced diurnal variation in
352 TGM in the equatorial area (4°N – 10°S) was demonstrated to be caused by biological
353 production (Fitzgerald et al., 1984).

354 However, Mao et al. (2012) suggested that the predominant effect of oceanic evasion on
355 ambient GEM concentrations was episodic, not necessarily diurnal, because they found, among
356 all physical parameters, the only significant correlation GEM had was with wind speed
357 exceeding 15 m s^{-1} at a marine location, which occurred rather sparsely. This was corroborated
358 by Sigler et al. (2009b) suggesting enhanced oceanic evasion at a rate of $\sim 7 \text{ ppqv hr}^{-1}$ (0.063 ng
359 m^{-3}) leading to 30 – 50 ppqv ($0.27\text{-}0.45 \text{ ng m}^{-3}$) increases in coastal and inland GEM

360 concentrations in southern New Hampshire, USA during the April 2007 Nor'easter.

361 In the study by Laurier et al. (2003) the lack of diurnal variation over the *Pacific* was
362 speculated to be caused by continuous evasion from surface water. Over the *Arctic*, unlike the
363 distinctive diurnal pattern with noontime peaks in the study by Lindberg et al. (2002), very small
364 near none diurnal variation in GEM was manifested in the shipboard measurements of Sommar
365 et al. (2010) and was speculated to result from low in situ oxidation of GEM. No diurnal
366 variation was found over the *Antarctica* due possibly to lack of diurnally varying sources and
367 sinks (Pfaffhuber et al., 2012), except one case with in situ human activity.

368 2.1.4.3 Seasonal to Annual Variation

369 Annual cycles of TGM/GEM in the MBL differed between various oceanic regions and
370 were suggested to be driven predominantly by oceanic evasion, biomass burning, anthropogenic
371 emissions, interhemispheric flux, and/or meteorological conditions (Slemr et al., 2008;
372 Ebinghaus et al., 2002a,b; Sigler et al., 2009a; Brunke et al., 2010; Soerensen et al., 2010; Mao
373 and Talbot, 2012; Angot et al., 2014; Wang et al., 2014). Annual cycles of TGM/GEM with an
374 annual maximum in summer and a minimum in winter observed at Cape Point, South Africa in
375 the *South Atlantic* MBL was hypothesized to be driven predominantly by oceanic emissions,
376 biomass burning, and anthropogenic activities (Brunke et al., 2010), and interhemispheric flux
377 (Slemr et al., 2008; Brunke et al., 2010). Higher concentrations of GEM in the summer over the
378 *Adriatic Sea* (Sprovieri et al., 2010) and over the *Augusta Basin* (Bagnato et al., 2013) were
379 suggested to be caused by stagnant meteorological conditions in the former study and enhanced
380 evasion from sea water in the latter. *Opposite annual variation* with higher (lower)
381 concentrations in winter (summer) was proposed to be determined largely by meteorology
382 (Ebinghaus et al., 2002a, 2011) and photochemical oxidation of GEM (Mao and Talbot, 2012).

383 The same annual cycle over the *Indian* Ocean was speculated to be a result of long range
384 transport of air masses originated from southern Africa biomass burning during the winter
385 months (July – September), and low GEM associated with southerly polar and marine air masses
386 from the remote southern Indian Ocean (Angot et al., 2014). Frequent MDEs in the summertime
387 Dead Sea MBL were observed to be often concurrent with varying concentrations of bromine
388 oxide (BrO) and high temperatures up to 45°C (Obrist et al., 2011). Such high temperatures
389 seemed to be contradictory to the general understanding that Br-initiated GEM oxidation tends to
390 go forward under very cold conditions at temperature < -40°C. Despite that, the authors
391 suggested that Br species were the major oxidants of GEM during depletion events, even when
392 constantly high temperatures were accompanied by sometimes low BrO concentrations.

393 Springtime large variation in *Arctic* and *Antarctic* TGM/GEM was caused by the
394 occurrence of MDEs. Polar MDEs have been generally linked to reactive Br-initiated GEM
395 oxidation in spring when Br explosion occurs producing abundant reactive Br (Schroeder et al.,
396 1998; Ebinghaus et al., 2002b; Lindberg et al., 2002; Temme et al., 2003b; Mao et al., 2010;
397 Steffen et al., 2013; Moore et al., 2014). For Antarctic MDEs, Ebinghaus et al. (2002b) found a
398 strong positive correlation between TGM and O₃ over August – October, accompanied by
399 enhanced Global Ozone Monitoring Experiment (GOME) column BrO. Compared to Arctic
400 MDEs, the first Antarctic MDE occurred about 1-2 months earlier, probably due to the lower
401 latitude of the monitoring site and sea ice, the former allowing earlier sunrise and the latter
402 conducive to Br/BrO formation. Temme et al. (2003b) found that the air masses reaching the
403 station during MDEs had a maximum contact with sea ice (coverage >40%) over the South
404 Atlantic Ocean, which was speculated to contain abundant reactive Br released from sea salt
405 associated with sea ice or sea salt aerosols.

406 Summertime annual maximums of GEM over the *Arctic* and *Antarctic* Ocean were
407 generally associated with enhanced evasion of GEM and from GOM reduction in snow resulting
408 from maximum exposed sea water after snow/ice melt (Lindberg et al., 2002; Aspmo et al., 2006;
409 Soerensen et al., 2010; Cole et al., 2013; Moore et al., 2014), which was also suggested using
410 model simulations by Dastoor and Durnford (2014). A different mechanism of riverine
411 contribution was hypothesized in Fischer et al. (2012) using an atmosphere-ocean coupled model.
412 Yu et al. (2014) observed high TGM concentrations concurrent with low salinity, CO, and high
413 chromophoric dissolved organic matter (CDOM) over the ice-covered central Arctic Ocean and
414 speculated that the relatively high CDOM concentrations associated with river runoff could
415 enhance Hg²⁺ reduction. Moreover they related the summer monthly variability in TGM
416 concentrations to less chemical loss.

417 2.1.4.4 Long-term Trends

418 Four *hypotheses* were made to explain the observed decreasing trends in TGM/GEM
419 during the past decades. First, the global decreasing trend was caused by decreased reemission
420 of legacy mercury as a result of a substantial shift in the biogeochemical cycle of Hg through the
421 atmospheric, oceans, and soil reservoirs, although exactly what may have caused this shift
422 remained unexamined (Slemr et al., 2011). Second, the decreasing trend was linked to
423 increasing tropospheric O₃ (Ebinghaus et al., 2011). However, this speculation was negated by
424 the plausibility of GEM oxidation by O₃ in the atmosphere. Third, based on atmosphere-ocean
425 coupled model simulations, the decreasing trend in TGM over the North Atlantic was caused by
426 decreasing North Atlantic oceanic evasion driven by declining subsurface water Hg
427 concentrations resulting from reduced Hg inputs from rivers and wastewater and from changes in
428 the oxidant chemistry of the atmospheric MBL (Soerensen et al., 2012). However, Amos et al.

429 (2014) suggested that the decrease in riverine input was too small to affect Hg concentrations in
430 the open ocean let alone the declining trend in North Atlantic sea water Hg concentrations. Last,
431 a 20% decrease in total Hg emissions and 30% in anthropogenic Hg^o emissions were estimated
432 for the period of 1990 – 2010, leading to the observed decreasing trends in TGM/GEM, as
433 suggested by a most recent modeling study (Zhang et al., 2016).

434 **2.2 GOM and PBM**

435 2.2.1 Concentration Metrics

436 The mean concentrations of GOM from individual studies varied from below LOD in
437 several studies to 4018 pg m⁻³ (1-h) in the *Dead Sea* MBL (Obrist et al., 2011; Moore et al., 2013)
438 (Table S2; references therein). The GOM concentration averaged for each oceanic region based
439 on values from the literature varied from 3 pg m⁻³ over the *Atlantic Ocean* to 40 pg m⁻³ over the
440 *Antarctica*, and the largest range 0.1 – 4018 pg m⁻³ was over the *Mediterranean* Sea and its
441 neighboring seas (**Fig. 1b**). Note that the small ranges in other oceanic MBL did not necessarily
442 indicate less variability in GOM but merely a result of limited measurement data available (Table
443 S2; references therein).

444 The mean concentrations of PBM from individual studies varied from below LOD in
445 several regions to 394 pg m⁻³ (1-h) over the *Beaufort* Sea (Steffen et al., 2013) (Table S3;
446 references therein). The PBM concentration averaged for each oceanic region based on values in
447 the literature varied from 0.6 pg m⁻³ over the *Indian* to 394 pg m⁻³ over the *Arctic* Ocean (**Fig.**
448 **1c**). No ranges were provided for the *Arctic*, *Antarctic*, and *Indian Ocean* MBL due to limited
449 numbers of studies there. The few studies available indicated that PBM concentrations were in
450 most cases smaller and less variable than GOM.

451 The earliest shipboard measurements of GOM showed dimethyl mercury (DMM)

452 concentrations orders of magnitude larger (Slemr et al., 1981, 1985) than the total GOM
453 concentration measured in the recent two to three decades. Due to the use of very different
454 techniques in early studies, those concentrations were listed in Table S2 (references therein) but
455 were not used for comparison with more recent studies (Table S2; references therein).

456 Same as GEM, GOM concentrations tended to be higher over the North than the South
457 *Atlantic* and in near-coastal regions than open waters (Temme et al., 2003b; Mason et al., 2001;
458 Sheu and Mason, 2001; Mason and Sheu, 2002; Aspmo et al., 2006; Laurier and Mason, 2007;
459 Sigler et al. 2009b; Mao and Talbot, 2012). Hourly GOM concentrations of 1 – 30 pg m⁻³ over
460 the *South Atlantic* Ocean from Neumayer to Punta Arenas in February 2001 (Temme et al.,
461 2003b) were 1 – 2 orders of magnitude smaller than the concentrations (1.38±1.30 pmol m⁻³, i.e.
462 ~300±280 pg m⁻³) near Bermuda in September and December 1999 and March 2000 (Mason et
463 al., 2001). However, at around the same time average values almost an order of magnitude
464 smaller were reported at Bermuda (50±43 pg m⁻³, a few pg m⁻³ to 128 pg m⁻³) (Mason and Sheu,
465 2002) and at a US mid-Atlantic coastal site (40 pg m⁻³) (Sheu and Mason, 2001). In comparison,
466 GOM concentrations were an order of magnitude smaller over the open water and at higher
467 latitude (Aspmo et al., 2006; Laurier and Mason, 2007), comparable to those over the *South*
468 *Atlantic*. Similar magnitude of GOM concentrations were measured at a *North Atlantic* near
469 coastal MBL site with an average of 0.4 ppqv (~3.6 pg m⁻³) (0 – 22 ppqv, i.e. 0 – 196 pg m⁻³, 2-h)
470 during 2007 –2010 (Sigler et al., 2009b; Mao and Talbot, 2012).

471 PBM concentrations (Table S3; references therein) were measured with an average of
472 1.9±0.2 pg m⁻³ during the May-June 1996 *South* and *equatorial Atlantic* cruise (Lamborg et al.,
473 1999) and 1.3 ± 1.7 pg m⁻³ (<0.5 pg m⁻³ (LOD) to 5.2 pg m⁻³) in Bermuda, 30-40 times smaller
474 than the concurrent weekly averaged GOM concentrations (Mason and Sheu, 2002; Sheu, 2001).

475 At higher *North Atlantic* latitudes, PBM concentrations were averaged at 2.4 pg m^{-3} , very close
476 to the concurrent average GOM concentrations but with a factor of 4 smaller varying from
477 <LOD to 6.3 pg m^{-3} in summer 2004 (Aspmo et al., 2006). Mao and Talbot (2012) reported
478 PBM concentrations varying from 0.09 ppqv (0.8 pg m^{-3}) in winter 2010 to 0.52 ppqv (4.6 pg m^{-3})
479 in summer 2010.

480 During the 2000s decade, concentrations of GOM over the *Pacific* decreased by around a
481 factor of 2 from 9.5 pg m^{-3} over open waters in 2002 (Laurier et al., 2003) to around 4 pg m^{-3} at a
482 remote Japanese site downwind of major Asian source regions in spring 2004 (Chand et al., 2008)
483 and in the equatorial region in 2011 (Wang et al., 2014) (Table S2; references therein). The
484 maximum concentration from a decade of studies was 700 pg m^{-3} (3-h) measured in air masses
485 originated from upper air over the *Pacific* (Timonen et al., 2013), about two orders of magnitude
486 larger than what Chand et al. (2008) and Laurier et al. (2003) reported. PBM concentrations
487 over the *Pacific* reached up to 17 pg m^{-3} , comparable to GOM, and on average were three times
488 larger downwind of East Asia ($3.0 \pm 2.5 \text{ pg m}^{-3}$) than in the equatorial *Pacific* MBL (Chand et al.,
489 2008; Wang et al., 2014) (Table S3).

490 In the southern *Indian* Ocean, very low GOM and PBM concentrations were observed,
491 averaged at 0.34 (<LOD ($0.28 - 0.42 \text{ pg m}^{-3}$) – 4.07 pg m^{-3}) and 0.67 pg m^{-3} (<LOD – 12.67 pg
492 m^{-3}), respectively, over two years from a remote location, Amsterdam Island (Angot et al., 2014).
493 These concentrations were at the lower end of the range of *Atlantic* and the *Pacific* MBL
494 measurements.

495 Measurements over the *Mediterranean Sea* and its neighboring seas generally showed
496 much higher concentration levels than over the *Atlantic*, *Pacific*, and *Indian* Ocean, with GOM
497 ranging from 0.1 pg m^{-3} over the *Adriatic* (Sprovieri and Pirrone, 2008) to 4018 pg m^{-3} over the

498 *Dead Sea* (Obrist et al., 2011) (Tables S2 & S3; references therein). Frequency distributions of
499 24-hour average GOM and PBM concentrations from a site situated in the Mediterranean MBL
500 exhibited log-normal distributions with the maximum frequency at around 59 and 48 pg m^{-3} ,
501 respectively (Pirrone et al., 2003). One of the major findings from Sprovieri et al. (2003) was
502 constant presence of GOM averaged at $7.9 \pm 0.8 \text{ pg m}^{-3}$ in the MBL over a 6000 km long cruise
503 path around the Mediterranean Sea. In a one year dataset of 2008, Beldowska et al. (2012)
504 showed 24-h PBM concentrations varied over 2 – 142 pg m^{-3} averaged at $20 \pm 18 \text{ pg m}^{-3}$ with 93%
505 on average in the coarse fraction ($>2 \mu\text{m}$) over the southern *Baltic Sea*.

506 In springtime *Arctic*, the highest concentrations of GOM at 900 – 950 pg m^{-3} were
507 observed during the 1998 – 2001 Barrow Atmospheric Mercury Study (BAMS). Very high
508 springtime PBM concentrations (mean 394 pg m^{-3} , 47 – 900 pg m^{-3} , 1-h) were reported over
509 Beaufort Sea sea ice by Steffen et al. (2013). This was an order of magnitude higher than
510 concurrent GOM concentrations (mean 30 pg m^{-3} , 3.5 – 104.5 pg m^{-3}) and even larger than those
511 in temperate regions, where particle concentrations tended to be large. In comparison, Sommar et
512 al. (2010) found very low GOM and PBM over the summertime Arctic Ocean.

513 Over the *Antarctica*, 2-h GOM concentrations ranged over 10.5 – 334 pg m^{-3} averaged at
514 $116.2 \pm 77.8 \text{ pg m}^{-3}$ in Terra Nova Bay during spring – summer 2000 (Sprovieri et al., 2002), and
515 a similar range was also observed by Temme et al. (2003b) at the Neumayer Station in summer
516 2001 (Table S2). A range of 30 – 140 pg m^{-3} (80-min) was reported for peaks of GOM in
517 summer 2007 (Soerensen et al., 2010). Concentrations of 1-h PBM from Temme et al. (2003b)
518 varied over 15 – 120 pg m^{-3} , a range a factor of 3 smaller than that of concurrent GOM, tracking
519 GOM well only at a lower level. Different from the *Arctic*, summertime GOM concentrations
520 over the *Antarctic* were orders of magnitude larger.

521 2.2.2 Hemispheric Difference

522 Hemispheric gradient has been measured in both GOM and PBM since the early 1980s
523 (Slemr et al., 1985; Soerensen et al., 2010). In the first shipboard study by Slemr et al. (1985),
524 PBM concentrations of 0.013 ± 0.018 and 0.007 ± 0.004 ng m⁻³ over the North and South
525 *Atlantic* Ocean, respectively, were derived from Hg concentrations in rain water. About three
526 decades later Soerensen et al. (2010) reported hemispheric difference of GOM with a NH
527 average of 0.3 ± 3 pg m⁻³ in summer and 0.8 ± 2 pg m⁻³ in spring, and a seasonally invariable SH
528 average of 4.3 ± 0.14 pg m⁻³.

529 2.2.3 Temporal Variations from Diurnal to Long-term Trend

530 2.2.3.1 Diurnal Variation

531 While some studies found a lack of diurnal variation in GOM (Sheu and Mason, 2001;
532 Aspomo et al., 2006; Temme et al., 2003b), many reported distinct diurnal variation with noon-
533 afternoon peaks and nighttime minimums in various oceanic regions (Mason et al., 2001; Mason
534 and Sheu, 2002; Lindberg et al., 2002; Laurier et al., 2003; Sprovieri et al., 2003, 2010; Laurier
535 and Mason, 2007; Mao et al., 2008; Chand et al., 2008; Sigler et al., 2009b; Soerensen et al.,
536 2010; Mao and Talbot, 2012; Wang et al., 2014). Over the *Atlantic* amplitude values varied from
537 0.27 pg m⁻³ in winter 2010 near the coast of southern New Hampshire, USA (Mao and Talbot,
538 2012) to >80 pg m⁻³ on the cruise from Barbados via Bermuda to Baltimore, Maryland, USA
539 (Mason and Sheu, 2002; Laurier and Mason, 2007). Over the *Pacific* amplitude values exceeded
540 80 pg m⁻³ (Laurier et al., 2003; Chand et al., 2008; Wang et al., 2014). Over the *Mediterranean*
541 Sea and its neighboring seas diurnal amplitude reached up to 35 pg m⁻³ (Sprovieri et al., 2003;
542 Sprovieri et al., 2010). The most pronounced diurnal variation was observed in the springtime
543 *Arctic* with noontime peaks up to $900 - 950$ pg m⁻³ and near zero concentrations at night

544 (Lindberg et al., 2002).

545 The diurnal pattern of PBM concentrations, measured using a Tekran speciation unit, at a
546 midlatitude North *Atlantic* near coastal MBL site was in general not consistent between seasons
547 and years with seasonally averaged daily peaks 0.2 – 0.7 ppqv ($1.7 - 6.2 \text{ pg m}^{-3}$) at varying time
548 of a day (Mao and Talbot, 2012). The Tekran PBM instrument measures PBM on particles <
549 2.5 μm . Using a 10-stage impactor, Feddersen et al. (2012), perhaps the first to study the size
550 distribution of PBM in MBL, reported PBM concentrations (up to 0.25 ppqv, i.e. 2.2 pg m^{-3} , in
551 3.3 – 4.7 μm) in ten size fractions (<0.4 μm to >10 μm) at the same MBL location from Mao and
552 Talbot (2012), and found a diurnal cycle with daily maximums at around 16:00 UTC (noon local
553 time) and minimums around sunrise.

554 2.2.3.2 Seasonal to Annual Variation

555 Studies reported distinct seasonal variation in GOM with higher concentrations in
556 warmer months and lower in colder months (Mason et al., 2001; Mason and Sheu, 2002; Pirrone
557 et al., 2003; Laurier and Mason, 2007; Sigler et al., 2009a; Sprovieri et al., 2010; Soerensen et al.,
558 2010; Mao and Talbot, 2012; Obrist et al., 2011; Moore et al., 2013; Wang et al., 2014; Angot et
559 al., 2014). A fairly flat baseline with negligible annual variation in GOM was observed at a
560 midlatitude North *Atlantic* MBL site near southern New Hampshire, USA, in a three year dataset,
561 with more variability in higher mixing ratios and seasonal median values ranging from 0.03 ppqv
562 ($\sim 0.27 \text{ pg m}^{-3}$) in winter 2010 to 0.55 ppqv ($\sim 4.9 \text{ pg m}^{-3}$) in summer 2007 (Mao and Talbot,
563 2012). Over the *Mediterranean* the fall 2004 campaign experienced no production of GOM
564 whereas the summer 2005 one saw very high concentrations varying over 21 – 40 pg m^{-3}
565 (Sprovieri et al., 2010a). In the *Dead Sea* MBL, AMDEs resulting in 1-h GOM up to 700 pg m^{-3}
566 occurred³ occurred more frequently in the summer than in winter (Obrist et al., 2011; Moore et al.,

567 2013).

568 In the *Arctic* MBL, several hundreds of pg m^{-3} GOM concentrations were observed in
569 spring (Lindberg et al., 2002; Steffen et al., 2013) and very low GOM and PBM concentrations
570 in summer (Sommar et al., 2010). Quite differently, summertime GOM concentrations over the
571 *Antarctic* seemed to be orders of magnitude larger (Sprovieri et al, 2002; Temme et al., 2003b;
572 Soerensen et al., 2010).

573 Some studies observed seasonal variation in PBM. Sprovieri et al. (2010a) found PBM
574 concentrations on average were more than a factor of 2 higher during high Hg episodes in the fall
575 than during the summertime ones over the *Mediterranean* Sea. Beldowska et al. (2012)
576 measured an average 24-h PBM of 15 pg m^{-3} and a $3 - 67 \text{ pg m}^{-3}$ range in the non-heating season
577 compared to an average of 24 pg m^{-3} and a range of $2 - 142 \text{ pg m}^{-3}$ in the heating season. The
578 PBM measurements at a *North Atlantic* coastal site using a 10-stage impactor showed distinct
579 seasonal variation with 50-60% of PBM in coarse fractions, $1.1 - 5.8 \mu\text{m}$, composed largely of
580 sea salt aerosols at both sites in summer and 65% in fine fractions in winter (Feddersen et al.,
581 2012). Over the *Indian* Ocean significantly higher concentrations were observed in winter than
582 in summer ($2.18 \pm 1.56 \text{ ng m}^{-3}$ vs. $1.79 \pm 1.15 \text{ pg m}^{-3}$) (Angot et al., 2014.)

583 2.2.4 Mechanisms Driving the Observed Temporal Variabilities

584 2.2.4.1 Factors Causing Episodic High and Low Concentrations

585 Long range transport of air masses of terrestrial origin with high PBM concentrations was
586 evidenced in elevated crustal enrichment factors in the PBM samples (Lamborg et al., 1999).

587 An episode of high GOM concentrations coincided with a passing hurricane was linked to
588 downward mixing of air aloft with higher GOM (Prestbo, 1997; Mason and Sheu, 2002). Low
589 GOM concentrations were found to be concurrent with high humidity (e.g., fog) and rainfall but

590 highest concentrations on the day after such events if temperatures were elevated (Mason and
591 Sheu, 2002). High nighttime concentrations of GOM in the Mediterranean Basin were observed
592 in anthropogenic plumes identified using backward trajectories (Sprovieri et al., 2010a). The
593 GOM concentrations in air masses of marine origin at a site on the East Pacific coast were
594 unusually high ranging over 200 – 700 pg m^{-3} (Timonen et al., 2013). The high GOM
595 concentrations were thought to be partitioned back from the PBM that was accumulated on
596 aqueous super-micron sea salt aerosols in the MBL when being lofted above the MBL, and an
597 anticorrelation between GOM and GEM was found in air masses of marine origin indicating
598 strong in-situ oxidation of GEM.

599 2.2.4.2 Diurnal Variation

600 The lack of GOM diurnal variation was speculated to result from diverse air masses with
601 different concentrations converging at the location leading to the removal of diurnal variation in
602 GOM (Sheu and Mason, 2001), and from low solar radiation ($<200 \text{ W m}^{-2}$) at higher latitudes
603 (Aspmo et al., 2006). The majority of the studies reporting significant diurnal variation in GOM
604 attributed it to photooxidation, loss via dry deposition, and oceanic evasion, which was backed
605 up by modeling studies (Hedgecock et al., 2003, 2005; Laurier et al., 2003; Selin et al., 2007;
606 Strode et al., 2007).

607 It was generally found that GOM concentrations were positively correlated with solar
608 radiation flux and anticorrelated with relative humidity and at times with O_3 (Mason and Sheu,
609 2002; Laurier and Mason, 2007; Soerensen et al., 2010; Mao et al., 2012). The correlation
610 between GOM and UV radiation flux indicated photochemical processes, and the anticorrelation
611 between GOM and O_3 was caused by processes destroying O_3 and producing GOM (Mason and
612 Sheu, 2002; Laurier and Mason, 2007), especially the oxidation reactions in the presence of

613 deliquescent sea salt aerosols (Sheu and Mason, 2004). The fact that GOM daytime peaks over
614 the *Pacific* increased with lower wind speeds and stronger UV radiation suggested that GOM
615 was produced in situ via photochemically driven oxidation (Laurier et al., 2003; Chand et al.,
616 2008). Chand et al. (2008) estimated the magnitude of GOM close to the amount produced from
617 the reaction of GEM + OH alone. Mao and Talbot (2012) suspected that unknown production
618 mechanism(s) of GOM in the nighttime MBL kept the levels above the LOD. Positive
619 correlation between GOM/PBM and temperature indicated possible temperature dependence of
620 certain oxidation reactions and gas-particle partitioning, whereas the anti-correlation between
621 GOM/PBM and wind speed indicated enhanced loss via deposition caused by faster wind speed
622 over water (Mao et al., 2012).

623 No consistent diurnal variation in PBM measured using a Tekran speciation unit
624 suggested more complicated processes than photochemistry involved in PBM budgets (Mao et
625 al., 2012). However, Feddersen et al. (2012) found diurnal variation in 10-stage impactor PBM
626 measurement data and speculated that GEM oxidation drove the PBM daytime maximum at
627 around 16:00 UTC (noon local time) and depositional loss at night without replenishment led to
628 the minimum around sunrise. In the same study, the large peaks of PBM appeared to be of
629 continental origin.

630 2.2.4.3 Seasonal to Annual Variation

631 Larger concentrations of GOM in spring and/or summer were generally associated with
632 stronger photo oxidation, biological activity, biomass burning, oceanic, and anthropogenic
633 emissions whereas low concentrations with wet deposition (Lindberg et al., 2002; Mason and
634 Sheu, 2002; Temme et al., 2003b; Pirrone et al., 2003; Sprovieri et al., 2003; Hedgecock et al.,
635 2004; Laurier and Mason, 2007; Sprovieri and Pirrone, 2008; Sprovieri et al., 2010; Soerensen et

636 al., 2010; Obrist et al., 2011; Mao et al., 2012; Angot et al., 2014; Wang et al., 2014). The
637 positive correlation between GOM concentration and solar radiation was used to explain warm
638 season maximums of GOM based on the same line of reasoning that was used to explain daytime
639 peaks of GOM (Mason and Sheu, 2002; Pirrone et al., 2003; Mao et al., 2012). Observed
640 seasonal variation in PBM was attributed to anthropogenic influence and gas-particle partitioning
641 as well as condensation and coagulation of fine particles (Sprovieri et al., 2010a; Beldowska et
642 al., 2012).

643 Over the *Mediterranean* Sea and its neighboring seas, it was generally thought that
644 meteorological conditions combined with anthropogenic, oceanic, and biomass emissions caused
645 GOM and PBM seasonal variation (e. g. Pirrone et al., 2003; Sprovieri et al., 2003; Hedgecock et
646 al., 2004; Sprovieri and Pirrone, 2008). A case in point is the seasonal contrast of no production
647 and little variation in GOM due likely to strong removal under the wet conditions in fall 2004
648 and very high concentrations due to strong oxidation under dry, sunny conditions in summer
649 2005 (Sprovieri et al., 2010). Sensitivity box model simulations suggested that the Hg + Br
650 controlled the production rate of GOM without contributions from the oxidation reactions by O₃
651 and OH and that HgBr was quickly converted to GOM. In the same study it was brought to
652 attention that biomass burning and ship emissions in the region were not included in the emission
653 inventory but could be important to ambient concentrations (Sprovieri et al., 2010). The authors
654 suggested that ship emissions could become a more important source of contaminants as
655 emissions from other sources were being more stringently controlled, and also the Mediterranean
656 was a place where busy shipping routes ran close to population centers. However, no studies
657 have demonstrated that ship emissions were an important source of Hg.

658 In the *Dead* Sea MBL, frequent occurrences of MDEs in the summer were linked to
659 higher BrO concentrations indicative of Br-initiated oxidation of GEM despite high temperature
660 and sometimes low BrO concentrations (Obrist et al., 2011). There is apparent discrepancy
661 between our theoretical understanding of the conditions required for Br-initiated GEM oxidation
662 and the real atmospheric conditions in the summertime Dead Sea MBL.

663 Wang et al. (2014) proposed iodine in a two-step mercury oxidation mechanism, where
664 BrHgI was hypothetically formed, helped to reconcile the modelled GOM with the observed
665 annual maximum GOM in October over the *equatorial Pacific*. The authors mentioned that HO₂
666 and/or NO₂ aggregation with HgBr from Dibble et al. (2012) could be another possibility and
667 further suggested that a major process in representing Hg oxidation is missing in current models.

668 Lindberg et al. (2002) found that springtime *Arctic* maximum concentrations of GOM at
669 900 – 950 pg m⁻³ corresponded to open leads over sea ice and an extensive area of elevated BrO
670 concentrations under the calmest conditions and strongest UV radiation. Low GOM and
671 unusually large PBM concentrations over Beaufort Sea sea ice in spring 2009 were speculated to
672 be caused by low temperatures and GOM formation followed by adsorption onto available sea
673 salt and sulfate aerosols, as well as ice crystals around the sea ice (Steffen et al., 2013). In
674 contrast, very low summertime Arctic GOM and PBM were due possibly to low in situ oxidation
675 of GEM and enhanced physical scavenging as a result of low visibility and high relative
676 humidity (Sommar et al., 2010).

677 Higher concentrations of GOM over the *Antarctic* Ocean were first proposed by Sprovieri
678 et al. (2002) to be produced from gas-phase oxidation of GEM by O₃, H₂O₂, and OH together
679 with favorable physical conditions such as PBL height. Temme et al. (2003b) found that the
680 highest concentrations of GOM corresponding to the lowest concentration of GEM falling below

681 the LOD (1.1 pg m^{-3}) during MDEs in summer were associated with the air masses having a
682 maximum contact with sea ice (coverage $>40\%$) over the *South Atlantic* Ocean, which was
683 speculated to contain abundant reactive Br, released from sea salt associated with sea ice.
684 Summertime GOM was found to be correlated with GEM due probably to in situ oxidation and
685 build-up (Soerensen et al., 2010) and was also observed to be anti-correlated with GEM due
686 solely to oxidation (Temme et al., 2003b; Sprovieri et al., 2002).

687 **3. Continental Boundary Layer**

688 In this section, continental sites are defined as inland sites located in non-polar regions
689 and exclude locations impacted by the MBL, e.g. coastal sites and oceans.

690 **3.1 TGM/GEM**

691 **3.1.1 Concentration Metrics**

692 Field measurements of TGM/GEM at continental sites were conducted mainly in Asia,
693 Canada, Europe, and USA. Very few TGM/GEM measurements have been made at inland sites
694 in the SH. Of all the four regions, the median concentrations of TGM or GEM were 1.6 ng m^{-3}
695 at remote and rural surface (low elevation) sites, 2.1 ng m^{-3} at urban surface sites, and 1.7 ng m^{-3}
696 at high elevation sites (**Fig. 2a**). TGM/GEM ranged over $0.1\text{-}11.3 \text{ ng m}^{-3}$ at remote sites, 0.2-
697 18.7 ng m^{-3} at rural sites, $0.2\text{-}702 \text{ ng m}^{-3}$ at urban sites, and $0.6\text{-}106 \text{ ng m}^{-3}$ at high elevation sites.
698 Overall these statistics indicate that TGM/GEM at continental urban sites were higher and had
699 larger variability than rural and remote surface sites and high elevation sites in the NH. By
700 geographical region (**Fig. 2b**), the median TGM/GEM in Asia, comprising of sites
701 predominantly in China and a few sites in Korea and Japan, were higher by 26-55% than those in
702 Europe, Canada, and USA in this respective order. Although a higher median TGM/GEM was
703 found in Asia, the maximum single 5-min concentration was recorded in the USA (324 ng m^{-3} ,

704 Engle et al., 2010). The 5-min maximum TGM/GEM among the four regions was the lowest in
705 Europe (23 ng m⁻³, Witt et al., 2010). It is important to note that most urban sites in the literature
706 are located in North America and Europe, and hence the higher TGM/GEM at continental urban
707 sites as shown in **Fig. 2b** were predominantly driven by measurements at those sites (instead of
708 Asian sites). A summary of the mean and the range of TGM/GEM as well as the distribution of
709 mean TGM/GEM at individual continental sites can be found in **Fig. S1** and Table S4. Statistics
710 from studies prior to 2009 are referred to in Sprovieri et al. (2010b).

711 3.1.2 Temporal Variations from Diurnal Cycle to Long-term Trends

712 3.1.2.1 Diurnal Variation

713 At *remote* surface locations, the diurnal variation of TGM/GEM is characterized by a daytime
714 increase reaching a maximum concentration in the afternoon and nighttime decrease
715 (Manolopoulos et al., 2007; Cheng et al., 2012). At *rural* surface and *high elevation* sites,
716 several different diurnal patterns have been reported. The first pattern, similar to remote surface
717 locations, is an early morning minimum, followed by midday to afternoon maximum and
718 decrease at night (Swartzendruber et al., 2006; Yatavelli et al., 2006; Choi et al., 2008, 2013; Fu
719 et al., 2008, 2009, 2010, 2012b; Lyman and Gustin, 2008; Mao et al., 2008; Obrist et al., 2008;
720 Faïn et al., 2009; Sigler et al., 2009; Mazur et al., 2009; Nair et al. 2012; Mao and Talbot, 2012;
721 Eckley et al., 2013; Parsons et al., 2013; Cole et al., 2014; Brown et al., 2015; Zhang et al., 2015).
722 The second diurnal pattern typically observed is a higher nighttime TGM/GEM than daytime.
723 This tends to occur in Asia and more polluted sites outside of Asia, e.g. abandoned Hg mines and
724 cement plants (Lyman and Gustin, 2008; Wan et al., 2009a; Rothenberg et al., 2010; Li et al.,
725 2011; Nguyen et al., 2011; Fu et al., 2012a; Gratz et al., 2013; Zhang et al., 2013; Cole et al.,
726 2014). The third pattern found at rural surface and elevated sites is a weak or lack of diurnal

727 pattern in TGM/GEM (Choi et al., 2008, 2013; Mao et al., 2008; Sigler et al., 2009; Engle et al.,
728 2010; Rothenberg et al., 2010; Mao and Talbot, 2012; Zhang et al., 2013; Han et al., 2014).

729 At *urban* surface sites, the predominant diurnal pattern is an increase in TGM/GEM
730 throughout the night that leads to a maximum in the early morning and a decrease in TGM/GEM
731 in the afternoon (Stamenkovic et al., 2007; Li et al., 2008; Choi et al., 2009; Lyman and Gustin,
732 2009; Song et al., 2009; Liu et al., 2010; Witt et al., 2010; Nguyen et al., 2011; Nair et al., 2012;
733 Zhu et al., 2012; Gratz et al., 2013; Kim et al., 2013; Civerolo et al. 2014; Cole et al., 2014; Han
734 et al., 2014; Lan et al., 2014; Xu et al., 2014; Xu et al., 2015). The diurnal amplitude tends to be
735 higher during summer compared to other seasons (Stamenkovic et al., 2007; Peterson et al. 2009;
736 Civerolo et al. 2014; Lan et al., 2014; Xu et al., 2014). Diurnal variations with daytime
737 maximum and early morning minimum have also been observed at *urban* surface sites (Fostier
738 and Michelazzo, 2006; Rothenberg et al., 2010; Witt et al., 2010; Jiang et al., 2013; Han et al.,
739 2014).

740 3.1.2.2 Seasonal Variation

741 The seasonal variation in TGM/GEM at some continental *remote* surface sites can be
742 characterized by a winter to early-spring maximum and lower summer/fall concentrations
743 (Manolopoulos et al., 2007; Cheng et al., 2012). At other *remote* sites, a completely opposite
744 seasonal pattern was found with higher summer/fall concentrations than winter/spring (Abbott et
745 al., 2008; Cole et al., 2014). The predominant seasonal TGM/GEM trend at *rural* surface and
746 *elevated* sites is the winter to spring maximum and summer/fall minimum (Zielonka et al., 2005;
747 Yatavelli et al., 2006; Choi et al., 2008; Fu et al., 2008, 2009, 2010; Mao et al., 2008; Sigler et al.,
748 2009a; Mazur et al., 2009; Engle et al., 2010; Mao and Talbot, 2012; Nair et al., 2012; Chen et
749 al., 2013; Parson et al., 2013; Cole et al., 2014; Marumoto et al., 2015). Other studies conducted

750 in *rural* sites and *elevated* sites found higher TGM/GEM during warm seasons (spring/summer)
751 than in the winter (Weiss-Penzias et al., 2007; Obrist et al., 2008; Nguyen et al., 2011; Eckley et
752 al., 2013; Zhang et al., 2013; Zhang et al., 2015).

753 The seasonal patterns at continental *urban* surface sites can be vastly different from each
754 other. Five major seasonal patterns have been identified including (1) a winter to spring
755 maximum (Fostier and Michelazzo, 2006; Stamenkovich et al. 2007; Choi et al., 2009; Peterson
756 et al., 2009; Civerolo et al., 2014; Xu et al., 2015), (2) a summer TGM/GEM maximum (Xu and
757 Akhtar, 2010; Jiang et al., 2013), (3) higher TGM during both winter and summer (Xu et al.,
758 2014), (4) higher TGM/GEM during spring/summer (Liu et al., 2007, 2010; Song et al., 2009;
759 Nair et al., 2012; Zhu et al., 2012; Hall et al., 2014), and (5) an absence of a clear seasonal trend
760 (Kim et al., 2013; Civerolo et al., 2014; Marumoto et al., 2015). Table 1 summarizes the
761 predominant diurnal and seasonal patterns observed at *rural*, *urban* and *high elevation*
762 continental sites.

763 3.1.2.3 Long-term Trends

764 At *rural* sites across Canada, TGM decreased at a rate of 0.9-3.3% per year between 1995
765 and 2011, which was determined using 5-15 years of TGM data depending on the location (Cole
766 et al., 2014). A GEM decrease of 0.056 ng m⁻³ yr⁻¹ from 2005-2010 was found at an *elevated* site
767 in New Hampshire (Mao and Talbot, 2012). Widespread declines in GEM across North America
768 between 1997 and 2007 have also been reported (Weiss-Penzias et al., 2016); however, the
769 trends were not determined separately for rural and urban sites. No significant trends in TGM
770 were found at *urban/industrial* sites in the UK from 2003-2013 (Brown et al., 2015) and at
771 another urban site in Seoul, Korea from 2004-2011 (Kim et al., 2013). However, a short-term
772 annual TGM decrease from 2.0 to 1.7 ng m⁻³ was recorded at an urban site in Windsor, Canada

773 from 2007-2009 (Xu et al., 2014). At a chlor-alkali site in the UK, TGM declined by 1.36 ± 0.43
774 $\text{ng m}^{-3} \text{ yr}^{-1}$ from 2003-2012 (Brown et al., 2015). Weigelt et al. (2015) determined annual TGM
775 trends for different air masses arriving at Mace Head, Ireland between 1996 and 2013.
776 Specifically for continental airflows, TGM decreased by $0.0240 \pm 0.0025 \text{ ng m}^{-3} \text{ yr}^{-1}$ for polluted
777 air masses from Europe, which was a slightly faster decline compared to marine airflows from
778 the North Atlantic Ocean ($-0.0209 \pm 0.0019 \text{ ng m}^{-3} \text{ yr}^{-1}$) and the SH ($-0.0161 \pm 0.0020 \text{ ng m}^{-3} \text{ yr}^{-1}$).
779 In certain months, the TGM decreases associated with local and European airflows (0.047 -
780 $0.051 \text{ ng m}^{-3} \text{ yr}^{-1}$) were greater than other months (Weigelt et al., 2015).

781 3.1.3 Mechanisms Driving the Observed Temporal Variabilities

782 3.1.3.1 Diurnal Variation of TGM/GEM

783 TGM/GEM was higher during daytime than nighttime and often declined to a minimum
784 in the early morning at *remote, rural, high elevation, and some urban* surface sites (Table 1).
785 One of the mechanisms driving this diurnal pattern involved meteorological parameters, such as
786 temperature, the increase of which enhances TGM/GEM volatilization (Manolopoulos et al.,
787 2007; Mao et al., 2012; Jiang et al., 2013; Han et al., 2014). Surface emissions of TGM can
788 occur during daytime from soil and snow as temperature and solar radiation increases (Mao et al.,
789 2012; Cole et al. 2014). Solar radiation minimizes the activation energy required for Hg
790 emissions (Zhu et al., 2012) and increases Hg photoreduction in soil and snow (Steffen et al.,
791 2008; Zhu et al., 2012; Hall et al., 2014; Xu et al., 2014; Xu et al., 2015). This process appeared
792 to be especially relevant at sites with elevated Hg in soil (Lyman and Gustin, 2008; Brown et al.,
793 2015) because of a larger flux gradient. Dry deposition of GEM in the night might also played a
794 role since deposition was typically observed in nighttime in contrast to emission during daytime
795 (Zhang et al., 2009). Fog or dew formation occurring in the late summer was believed to have

796 caused GEM depletion in the early morning hours by capturing GEM in fog or dew water
797 (Manolopoulos et al., 2007; Mao and Talbot, 2012). Another driving mechanism of this
798 TGM/GEM diurnal pattern was the change in the boundary layer mixing height. Lower
799 TGM/GEM during nighttime is due to TGM/GEM deposition as the nocturnal inversion layer
800 forms. In the morning, the nocturnal inversion breaks down and mixes with TGM/GEM-rich air
801 in the residual layer and subsequently leads to increasing TGM/GEM during the day (Yatavelli et
802 al., 2006; Mao et al., 2008; Mazur et al., 2009; Mao and Talbot, 2012; Nair et al. 2012; Choi et
803 al., 2008, 2013; Jiang et al., 2013; Cole et al., 2014). At *elevated* sites, there was a transition
804 from the sampling of boundary layer during daytime to free troposphere air at night which was
805 driven by mountain/valley atmospheric patterns (Obrist et al., 2008). During daytime, mountain
806 breeze causes moist air to ascended from the surface to higher altitudes carrying with it GEM
807 from the boundary layer (Swartzendruber et al., 2006; Obrist et al., 2008; Fu et al., 2010, 2012b;
808 Zhang et al., 2015). At night, drier free troposphere air impacted the elevated site leading to
809 lower GEM and water vapor and higher GOM and ozone (Obrist et al., 2008). A lack of diurnal
810 variability was also reported at some rural surface locations, although the driving mechanism is
811 not quite clear. At an elevated site, the sampling of air above the nocturnal boundary layer and
812 lack of anthropogenic sources or GEM oxidants near the site led to constant GEM during most of
813 the time except in the summer (Mao et al., 2008; Sigler et al., 2009a; Mao and Talbot, 2012).
814 Thus this differed from other mountain sites, which were affected by surface emissions and
815 local/regional transport of GEM from the boundary layer during daytime.

816 At most *urban* sites and some *elevated* and polluted rural sites, the nighttime TGM
817 concentrations were higher than daytime, and the maximum concentration typically occurred in
818 the early morning before sunrise (Table 1). This type of diurnal variation was driven by

819 nighttime accumulation of TGM/GEM near the surface due to a shallow nocturnal boundary
820 layer and dilution during the day initiated by convective mixing with cleaner air aloft as the
821 mixing layer increases (Stamenkovic et al., 2007; Li et al., 2008; Lyman and Gustin, 2008, 2009;
822 Choi et al., 2009; Wan et al., 2009a; Rothenberg et al., 2010; Witt et al., 2010; Li et al., 2011;
823 Nguyen et al., 2011; Fu et al., 2012a; Nair et al., 2012; Zhu et al., 2012; Gratz et al., 2013; Kim
824 et al., 2013; Zhang et al., 2013; Cole et al., 2014; Lan et al., 2014; Xu et al., 2014). The shallow
825 nocturnal boundary layer was often associated with high TGM coinciding with low wind speeds
826 at night (Li et al., 2008; Fu et al., 2012a; Lan et al., 2014). Increases in nighttime concentrations
827 could also be driven by nighttime sources, such as emissions from mercury mining regions
828 (Lyman and Gustin, 2008) and local emissions occurring at night (Song et al., 2009; Wan et al.,
829 2009a; Rothenberg et al., 2010; Gratz et al., 2013; Kim et al., 2013). At *urban* surface sites,
830 studies suggested the driving mechanisms for the morning maximum were surface emissions
831 (Zhu et al., 2012; Hall et al., 2014; Xu et al., 2014; Xu et al., 2015), volatilization of Hg from
832 dew (Zhu et al., 2012), and vehicular traffic emissions evident by correlations between
833 TGM/GEM and CO and NO_x (Zhu et al., 2012; Xu et al., 2015). However, there is little research
834 suggesting significant amounts of Hg from vehicular emissions (Conaway et al., 2005; Landis et
835 al., 2007; Won et al., 2007). The general view is that the global contribution from petroleum
836 fuels combustion represented 0.00013% of the total anthropogenic emissions and thus can be
837 neglected in global assessment of Hg emissions (Pirrone et al., 2010). The lower TGM/GEM
838 observed in the afternoon was driven by GEM oxidation (Stamenkovic et al., 2007; Choi et al.,
839 2009; Lyman and Gustin, 2009; Li et al., 2011; Nguyen et al., 2011; Kim et al., 2013; Zhang et
840 al., 2013; Xu et al., 2014; Xu et al., 2015).

841 Many studies conducted in *urban* areas found a larger diurnal amplitude during summer
842 than other seasons. The major driving mechanism for this larger amplitude originated from
843 higher solar radiation and temperature, which increased the boundary layer mixing height in the
844 summer (Civerolo et al., 2014; Xu et al., 2014). Higher solar radiation during summer also
845 increased photochemical reactions, like GEM oxidation. The larger diurnal variation was also
846 attributed to increases in uptake and re-emissions by vegetation and power plant emissions from
847 air conditioner use during summer nights (Xu et al., 2014). The shift in the timing of the
848 TGM/GEM maximum varied with season at some urban sites. During spring in Windsor,
849 Canada, the decrease in TGM earlier in the afternoon was thought to be due to increase
850 photochemical processes resulting from higher solar radiation and lower GEM emissions due to
851 less vegetation coverage in the spring (Xu et al., 2014). In Nanjing, China, the peak
852 concentration occurring later in the morning during spring was driven by prolonged sunlight
853 hours (Zhu et al., 2012).

854 Site characteristics may have different impacts on the diurnal variation. During nighttime,
855 GEM at an *urban* site was significantly higher than a rural site suggesting higher GEM fluxes
856 from buildings and pavement than vegetation and soil (Liu et al., 2010), but may be simply
857 caused by stronger and more anthropogenic sources in urban areas. The diurnal amplitude at an
858 *urban* site was greater than a suburban site in one study; however, the reason was not known
859 (Civerolo et al., 2014). In the same study, nighttime GEM was 25-30% higher than daytime for
860 the urban site close to the Atlantic Ocean, whereas the GEM difference between night and day
861 was only 10% at an inland suburban site (Civerolo et al., 2014). The study suggested that the
862 higher halogen concentrations in marine environments increased GEM oxidation and
863 subsequently, the loss of GEM in the afternoon leading to larger diurnal variation. At a different

864 coastal-urban location, nighttime GEM was only slightly higher than daytime because of the
865 cleaner air transported from the marine environment (Nguyen et al., 2011). These studies
866 suggested that MBL influence could lead to very different diurnal patterns. Sites continuously
867 impacted by Hg point sources likely contributed to the large short-term fluctuations in the diurnal
868 patterns at some urban sites (Rutter et al., 2008; Engle et al., 2010; Witt et al., 2010).

869 3.1.3.2 Seasonal Variation of TGM/GEM

870 The seasonal variation exhibiting a winter to spring maximum in remote, rural, urban and
871 high elevation environments (Table 1) was suggested to be driven by multiple mechanisms,
872 including anthropogenic emissions for winter heating (coal and wood combustion), reduced
873 atmospheric mixing, decreased GEM oxidation, less scavenging, and emissions from soil,
874 vegetation, and melting snow in the spring (Stamenkovic et al., 2007; Choi et al., 2008; Mao et
875 al., 2008; Sigler et al., 2009a; Peterson et al., 2009; Wan et al., 2009a; Cheng et al., 2012; Mao
876 and Talbot, 2012; Civerolo et al., 2014; Cole et al., 2014; Xu et al., 2015). The lower
877 TGM/GEM during summer has been attributed to increased GEM oxidation, uptake by
878 vegetation, and higher wet deposition of GOM (Yatavelli et al., 2006; Fu et al., 2008, 2009;
879 Engle et al., 2010; Xu et al., 2015). While these were the predominant driving mechanisms of
880 the seasonal variations in the NH, the seasonal patterns could also be influenced by changes in
881 the prevailing wind patterns (Fostier and Michelazzo, 2006; Fu et al., 2010, 2015; Sheu et al.,
882 2010; Chen et al., 2013; Zhang et al., 2013; Hall et al., 2014). The impact of combustion
883 emissions from winter heating was ruled out at a subtropical site in the Pearl River Delta region
884 of China; instead, the elevated TGM in the spring was attributed to monsoons which advected
885 southerly marine air masses during summer and northeasterly winds from Siberia during winter
886 (Chen et al., 2013). The transition from cold dry air to warm moist air often led to strong

887 temperature inversion and haze in the spring, which in turn inhibits pollutant dispersion.
888 Summer and spring maxima in TGM/GEM have also been found at remote, rural, and urban
889 atmospheres. This pattern was predominantly driven by meteorology. Higher solar radiation
890 and temperature during summer increased GEM emissions from Hg contaminated soil (Zhu et al.,
891 2012; Eckley et al., 2013), from vegetation at a forested agricultural site (Nguyen et al., 2011),
892 and from urban surfaces such as soil and pavement in Windsor, Canada (Xu and Akhtar, 2010).

893 3.1.3.3 Long-term Trends of TGM/GEM

894 Long-term trends of TGM/GEM over continental regions indicated a declining trend at
895 some sites and no significant trend at others, particularly at urban sites. Previous studies partly
896 attributed the long-term TGM trends to anthropogenic Hg emissions reductions. There has been
897 a 60-70% decrease in anthropogenic Hg emissions from USA and Canada; however only up to
898 15% of those emissions reductions impacted TGM at Canadian sites (Cole et al., 2014). The
899 more rapid decline in TGM measured at Mace Head, Ireland for local and European air masses
900 compared to marine air masses was thought to be driven by Hg emissions reductions in Europe
901 (Weigelt et al., 2015). The baseline TGM at Mace Head decreased at a larger rate in November
902 than other months suggesting that it is related to lower Hg emissions from residential heating in
903 Europe. The 21% decline in TGM from 2006-2012 in urban/industrial areas of the UK was also
904 consistent with the 0.21 Mg yr^{-1} (24%) reduction in Hg emissions from the UK, even though the
905 TGM trend from the 2003-2013 period was not statistically significant (Brown et al., 2015). In
906 Seoul, Korea, no significant trend in TGM was found from 2004-2011, consistent with the slight
907 decrease (1%) in coal consumption in Seoul over the same time frame (Kim et al., 2013). While
908 TGM/GEM trends appear to be aligned with local/regional Hg emission trends, a discrepancy
909 exists when the trend was compared to the increasing global anthropogenic Hg emissions

910 (Sprovieri et al., 2010b; Ebinghaus et al., 2011; Cole et al., 2014). Alternative reasons for the
911 decline in TGM could be due to faster cycling of Hg as O₃ and other oxidants have been
912 increasing or lower emissions of previously-deposited Hg (Sprovieri et al., 2010b; Ebinghaus et
913 al., 2011). Modeling studies indicated global Hg emissions inventory have not accounted for the
914 changes in Hg speciation emission profiles from coal combustion and reduced emissions from
915 products containing Hg (Zhang et al., 2016).

916 **3.2 GOM and PBM**

917 3.2.1 Concentration Metrics

918 The highest median GOM and PBM were found at *high elevation* sites, while the lowest
919 concentrations were found at *rural* surface sites. The median GOM from all locations were 12.1
920 pg m⁻³ at *elevated* sites, 9.9 pg m⁻³ at *urban* sites, 3.8 pg m⁻³ at *remote* sites, and 2.8 pg m⁻³ at
921 *rural* sites (Fig. 2a), and correspondingly the median PBM concentration was 11.0, 10.0, 6.9, and
922 4.6 pg m⁻³. The variabilities in GOM and PBM were greatest at urban locations. 2-3 hour GOM
923 concentrations ranged from <LOD-880 pg m⁻³ at elevated sites, <LOD-8160 pg m⁻³ at urban sites,
924 <LOD-224 pg m⁻³ at remote sites, and <LOD-462 pg m⁻³ at rural sites (see individual site
925 statistics and the map of mean concentrations at all sites in **Fig. S1** and Table S5). 2-3 hour
926 PBM concentrations ranged from <LOD-1001 pg m⁻³ at elevated sites, <LOD-11600 pg m⁻³ at
927 urban sites, <LOD-404 pg m⁻³ at remote sites, and <LOD-205 pg m⁻³ at rural sites (Table S6).
928 By geographical region, the median GOM in Asia was a factor of 1.4-5.1 higher than those in
929 Canada and USA (Fig. 2b). Similarly, the median PBM in Asia was 1.8-8.1 times higher than
930 those in Canada, Europe and USA. This was potentially because one-third of the elevated sites
931 were in China. The GOM and PBM maxima of 8160 pg m⁻³ and 11600 pg m⁻³, respectively,
932 were both observed at an urban site in Illinois, USA (Engle et al., 2010; Table S5 and S6).

933 3.2.2 Temporal Variations from Diurnal Cycle to Seasonal Trends

934 3.2.1.1 Diurnal Variation

935 The predominant diurnal pattern of GOM at *remote*, *rural*, *urban*, and *elevated* sites was
936 an increase in the morning leading to a maximum sometime between midday to late afternoon
937 and eventually decreasing at night (Yatavelli et al., 2006; Manolopoulos et al. 2007; Abbott et al.,
938 2008; Lyman and Gustin, 2008; Faïn et al., 2009; Rothenberg et al., 2010; Cheng et al., 2012; Fu
939 et al., 2012a; Nair et al., 2012; Eckley et al., 2013; Gratz et al., 2013; Cole et al., 2014; Civerolo
940 et al., 2014; Marumoto et al. 2015; Zhang et al., 2015). Late evening increases in GOM were
941 observed at some *urban* and *elevated* sites (Lynam and Keeler, 2005; Song et al., 2009; Gratz et
942 al., 2013). The average GOM was 18-60 pg m⁻³ between midnight and early morning at two
943 *elevated* sites, whereas the average daytime GOM was 9.2-39 pg m⁻³ (Swartzendruber et al.,
944 2006; Sheu et al., 2010).

945 No predominant diurnal pattern was found for PBM, which was mostly measured using
946 the Tekran speciation unit (2537-1135-1130). At *rural* and *urban* sites, the types of diurnal
947 patterns include, daytime/afternoon peak (Yatavelli et al., 2006; Choi et al., 2008; Rothenberg et
948 al. 2010; Cole et al., 2014), increasing during daytime leading to a nighttime peak (Nair et al.,
949 2012; Zhang et al., 2013), or lack of variation (Cobbett and Van Heyst, 2007; Choi et al., 2008;
950 Rothenberg et al., 2010; Cole et al., 2014).

951 3.2.1.2 Seasonal Variation

952 No predominant seasonal pattern in GOM was found at *remote*, *rural*, *urban*, and
953 *elevated* sites. At *remote* sites, some studies observed a winter to early-spring maximum and
954 lower concentrations during summer/fall (Manolopoulos et al., 2007; Cheng et al., 2012),
955 whereas higher summer/fall than winter/spring concentrations were also reported (Abbott et al.,

956 2008). In *rural* and *elevated* sites, the maximum concentration occurred in different seasons. At
957 *urban* sites, the maximum GOM typically occurred in warmer seasons, e.g. spring or summer
958 (Song et al., 2009; Liu et al., 2010; Choi et al., 2013; Wang et al., 2013; Gratz et al., 2013;
959 Civerolo et al., 2014; Han et al., 2014; Marumoto et al., 2015; Xu et al., 2015). Higher PBM and
960 total particulate Hg (TPM) during colder seasons than summer was a highly ubiquitous trend for
961 *remote, rural, urban, and elevated* sites (Zielonka et al., 2005; Choi et al., 2008; Wan et al.,
962 2009b; Liu et al., 2010; Kim et al., 2012; Gratz et al., 2013; Beldowska et al., 2012; Marumoto et
963 al., 2015; Schleicher et al., 2015; Zhang et al., 2015). However, increases in PBM also occurred
964 during summer in a few studies (Song et al., 2009; Huang et al., 2010; Cheng et al., 2012).

965 3.2.3 Mechanisms Driving the Observed Temporal Variabilities

966 3.2.3.1 Diurnal Variations of GOM and PBM

967 The widespread observation of a midday to late afternoon peak in GOM at continental
968 sites (Table 1) often coincided with meteorological parameters, such as solar radiation and
969 temperature, and/or ozone (Yatavelli et al., 2006; Abbott et al., 2008; Wan et al., 2009a; Weiss-
970 Penzias et al., 2009; Nair et al., 2012; Mao et al., 2012; Gratz et al., 2013; Zhang et al., 2013;
971 Civerolo et al., 2014; Cole et al., 2014; Marumoto et al., 2015). At *high elevation* sites, GOM
972 was also inversely correlated with relative humidity, water vapor, or dew point temperature
973 (Swartzendruber et al., 2006; Lyman and Gustin, 2008, 2009; Weiss-Penzias et al., 2009), and in
974 some cases GOM was not correlated with O₃ (Lyman and Gustin, 2009; Peterson et al., 2009; Xu
975 et al., 2015). These diurnal trends indicated daytime *in-situ* photochemical production of GOM
976 or entrainment of GOM from the free troposphere due to convective mixing. Increases in GOM
977 during daytime at a rural site was attributed to local transport from urban areas as indicated by
978 similarities in diurnal patterns between GOM, SO₂, and O₃ and a delay in the timing of the GOM

979 maximum likely resulting from emissions transport (Rothenberg et al., 2010). Short-term
980 fluctuations in the diurnal pattern of GOM also suggested the influence of point sources (Rutter
981 et al., 2008; Engle et al., 2010). Dry deposition and scavenging of GOM by dew played a role in
982 decreasing GOM during nighttime (Liu et al., 2007; Wan et al., 2009b; Weiss-Penzias et al.,
983 2009; Nair et al., 2012; Choi et al., 2013; Civerolo et al., 2014). The stronger diurnal amplitude
984 during the spring/summer coincided with stronger correlations between GOM, solar radiation,
985 temperature and O₃ (Yatavelli et al., 2006; Mao et al., 2012; Gratz et al., 2013; Zhang et al.,
986 2013), which suggested that increased photochemical processes led to higher GOM. Large
987 diurnal variation during summer was also potentially driven by high pressure, drier and cloud-
988 free conditions that are conducive to the buildup of GOM in the free troposphere (Lyman and
989 Gustin, 2009).

990 Nighttime increases in GOM seen exclusively at *urban* and *elevated* sites (Table 1)
991 appeared to be driven by anthropogenic emissions and the free troposphere. Nocturnal emissions
992 and local/regional transport within the boundary layer (Lynam and Keeler, 2005; Song et al.,
993 2009) and reduced vertical mixing in the stable nocturnal boundary layer led to higher GOM at
994 night in *urban* areas (Gratz et al., 2013). At *high elevation* sites, katabatic winds entrained GOM
995 from the free troposphere. In one study, GOM from the free troposphere was believed to
996 originate from *in-situ* photochemical processes due to a strong inverse GEM-GOM correlation
997 and a GOM/GEM slope near unity during an elevated GOM episode (Swartzendruber et al.,
998 2006). While an anti-correlation between GEM and GOM was also found at another elevated
999 site, Sheu et al. (2010) did not observe a complete photochemical conversion of GEM to GOM.
1000 The difference between these two *elevated* sites suggested different sources of GOM in the free
1001 troposphere. Timonen et al. (2013) found that in one type of free troposphere air mass, GEM

1002 oxidation occurred in anthropogenic plumes transported from Asia to Mt. Bachelor Observatory,
1003 USA and converted 20% of the GEM to GOM. A second type of air mass travelling over the
1004 Pacific Ocean resulted in 100% GEM conversion to GOM likely because of GEM oxidation by
1005 bromine.

1006 The driving mechanisms behind the diurnal pattern of PBM were better explored for
1007 *urban* sites than other site categories. Frequent spikes in hourly concentrations during daytime
1008 were attributed to point sources (Rutter et al., 2008; Civerolo et al., 2014). At a valley *urban* site,
1009 higher PBM and GEM during daytime suggested similar emission sources from Hg enriched
1010 areas (Lyman and Gustin, 2009). Higher PBM during daytime in the summer could also be
1011 initiated by photochemical production of GOM followed by absorption on secondary organic
1012 aerosols (Choi et al., 2013). Diurnal patterns exhibiting nighttime increases in PBM in urban
1013 areas could be due to multiple mechanisms and sources, such as nocturnal emissions and
1014 local/regional transport within the boundary layer (Song et al., 2009), reduced vertical mixing in
1015 the stable nocturnal boundary layer (Gratz et al., 2013; Xu et al., 2015), vehicular emissions in
1016 China (Xu et al., 2015), and nighttime street food vending in Beijing (Schleicher et al. 2015).

1017 3.2.3.2 Seasonal Variations of GOM and PBM

1018 The seasonal variation characterized by higher GOM in the warm seasons (Table 1) was
1019 primarily driven by photochemical production due to increased solar radiation, O₃, and likely
1020 other atmospheric oxidants (Liu et al., 2010; Choi et al., 2013; Civerolo et al., 2014; Xu et al.,
1021 2015). Alternative reasons could be attributed to anthropogenic emissions leading to higher
1022 GOM in the summer at *urban* sites (Song et al., 2009; Gratz et al., 2013). Atmospheric mercury
1023 depletion events occurring at *higher latitude* continental sites led to higher GOM during spring
1024 (Cole et al., 2014). Free troposphere transport was a major driving mechanism for higher

1025 reactive Hg at three *high elevation* western U.S. sites (Weiss-Penzias et al., 2015). At *elevated*
1026 sites in China, the occurrence of higher GOM between fall and spring were attributed to coal and
1027 biofuel burning (Wan et al., 2009b) and changes in the prevailing winds that advected GOM
1028 from polluted regions (Fu et al., 2012a; Zhang et al., 2015). Lower GOM during summer was
1029 due to wet deposition (Wan et al., 2009b; Sheu et al., 2010).

1030 Several mechanisms contributed to the increase in PBM or TPM during colder seasons
1031 (Table 1) including, local/regional coal combustion and wood burning emissions, lower mixing
1032 height, less oxidation, and increased gas-particle partitioning (Song et al., 2009; Xiu et al., 2009;
1033 Liu et al., 2010; Cheng et al., 2012; Fu et al., 2012a; Kim et al., 2012; Choi et al., 2013; Gratz et
1034 al., 2013; Wang et al., 2013; Civerolo et al., 2014; Cole et al., 2014; Schleicher et al., 2015; Xu
1035 et al., 2015). Oxidized Hg tended to partition to particles during colder seasons because of lower
1036 temperatures (Rutter et al., 2007), higher relative humidity (Kim et al., 2012), and reduced
1037 volatilization of gaseous Hg (Choi et al., 2013). Similar to GOM, decreases in PBM during
1038 summer at many sites in China were due to wet deposition (Wan et al., 2009b; Schleicher et al.,
1039 2015; Xu et al., 2015; Zhang et al., 2015) and a shift to cleaner marine airflows during summer
1040 (Kim et al., 2012). Higher PBM during warm seasons may be driven by forest fire emissions
1041 (Eckley et al., 2013) and increased PM_{2.5} available for GOM absorption at urban sites (Song et
1042 al., 2009; Schleicher et al., 2015).

1043 **4. Latitudinal Variation**

1044 There are a few shipboard and airborne studies that surveyed latitudinal variation of
1045 TGM/GEM (Slemr et al., 1981, 1985, 1995; Slemr and Langer, 1992; Fitzgerald et al., 1984;
1046 Lamborg et al., 1999; Temme et al., 2003a; Aspmo et al., 2006; Soerensen et al., 2010). Bagnato
1047 et al. (2013) compiled a latitudinal distribution of TGM/GEM using measurement data from a

1048 number of shipboard measurement studies spanning the time period of 1980 – 2012 (**Fig. 3**) and
1049 showed a small but discernible inter-hemispheric gradient, with the highest concentrations (~ 3.5
1050 ng m^{-3}) in NH midlatitudes and the lowest in SH latitudes ($\sim 0.9 \text{ ng m}^{-3}$), resulting from greater
1051 emissions of Hg in the more industrialized NH.

1052 Tropospheric airborne measurements from INTEX-B (Talbot et al., 2007, 2008) and
1053 ARCTAS (Mao et al., 2010), spanning near the surface to 12 km altitude, suggested distinct
1054 seasonal variation in GEM concentrations and latitudinal gradient. On average there was an
1055 increase of ~ 50 ppqv ($\sim 0.5 \text{ ng m}^{-3}$) from lower latitudes ($\sim 20 - 30^\circ\text{N}$) to higher ($60 - 90^\circ\text{N}$)
1056 latitudes in spring while negligible latitudinal variation in summer (**Fig. 4**). It was speculated
1057 that smaller latitudinal gradient of temperature in summer likely enhanced meridional circulation
1058 resulting in smaller latitudinal variation in GEM concentration in the troposphere.

1059 A small gradient was measured in atmospheric GEM concentrations over the *Pacific*
1060 from 1.32 ng m^{-3} in $14 - 20^\circ\text{N}$ latitudes to 1.15 ng m^{-3} in $1-15^\circ\text{S}$ latitudes in October 2011
1061 (Soerensen et al., 2014). Atmospheric GEM elevated in the northern part of the ITCZ was
1062 temporarily influenced by the northeastern trade wind that enhanced oceanic evasion, consistent
1063 with the largest evasion flux in that region.

1064 **5. Altitude Variation**

1065 Airborne measurements of TGM, GEM, and/or GOM have been conducted since 1977
1066 (Seiler et al., 1980) extending from near the surface to ~ 12 km altitude at several geographic
1067 locations (Table S7; references therein). More recent studies showed GEM concentrations
1068 remaining nearly constant *vertically*, slightly decreasing with altitude (Banic et al., 2003; Radke
1069 et al., 2007; Talbot et al., 2007, 2008; Mao et al., 2010). *Seasonal variation* was observed from
1070 surface to 7 km over Canada with $\sim 1.5 \text{ ng m}^{-3}$ in summer, 1.7 ng m^{-3} in winter, $1.7 \text{ ng m}^{-3} > 1 \text{ km}$

1071 altitude and 1.2 ng m^{-3} below 1 km due to widespread MDEs over the sea ice in the springtime
1072 Arctic (Banic et al., 2003). During ARCTAS, Mao et al. (2010) found that the vertical extent of
1073 springtime Arctic MDEs varied from meters to 1 km depending on the thickness of the surface
1074 inversion layer.

1075 Observation of low GEM in stratospherically influenced air led to the hypothesis that the
1076 upper troposphere/lower stratosphere (UTLS) was a Hg sink region (Radke et al., 2007). With
1077 repeated measurements of depleted GEM in stratospherically influenced air coupled with
1078 Murphy et al. (1998, 2006)'s findings of enrichment of PBM in lower stratospheric aerosols,
1079 Talbot et al. (2007) hypothesized that stratospheric GEM depletion was caused by fast oxidation
1080 of GEM by abundant halogen radicals and O_3 and estimated a lifetime of 2 and 0.5 days for 100
1081 ppqv GEM oxidized by O_3 and Br, respectively. Talbot et al. (2007) suggested that stratospheric
1082 intrusion could be a source of tropospheric Hg if PBM was to be transformed back to gaseous Hg.

1083 A $1 - 2 \text{ ng m}^{-3}$ range of upper tropospheric GEM was reported by Ebinghaus et al. (2007)
1084 and elevated GEM concentrations in biomass burning plumes from the same study suggested
1085 biomass burning representing a major mercury source. In the atmosphere of East Asia, Friedli et
1086 al. (2004) was the first to report GEM concentrations from sea level to ~ 7 km altitude under the
1087 influence of continental export from East China, showing concentrations at all altitudes higher
1088 than the global background, with the largest 6.3 ng m^{-3} in an industrial plume mostly from coal
1089 combustion and at times from other sources including dust storms, biomass burning, and
1090 volcanic eruption. On a relevant note, Swartzendruber et al. (2008) suggested that long range
1091 transport of Asian pollution contributed to the higher GEM concentrations above 2.5 km, which
1092 increased with altitude from $1.30 \pm 0.084 \text{ ng m}^{-3}$ in 0 – 0.5 km altitude to $1.52 \pm 0.182 \text{ ng m}^{-3}$ in the
1093 highest layer 5.5 – 6.5 km altitude over the Pacific Northwest over 13 April – 16 May 2006.

1094 Upper air GOM concentrations were first measured in spring by Lindberg et al. (2002) at
1095 1000 m and 100 m altitude immediately northeast of Point Barrow. Six aircraft surveys
1096 consistently showed that GOM concentrations decreased from an average of 70 to 20 to 2 pg m^{-3}
1097 from 5 to 100 to 1000 m altitude, supporting the hypothesis that the Hg oxidation reactions
1098 occurred in the near-surface boundary layer driven by halogen compounds derived from sea-salt
1099 aerosols. In recent years, more studies attributed higher GOM concentrations in higher altitudes
1100 to lack of depositional loss, lower temperature, and/or more abundant Br radicals (Sillman et al.,
1101 2007; Lyman and Jaffe, 2011; Brooks et al., 2014; Gratz et al., 2015; Shah et al., 2016). Sillman
1102 et al. (2007) reported GOM concentrations measured in Florida increasing with height from 10 to
1103 230 pg m^{-3} , which was reproduced using CMAQ model (Bullock and Brehme, 2002) with gas-
1104 phase oxidation reactions GEM+O₃ and GEM+OH, the latter being dominant. Lyman and Jaffe
1105 (2011) found enhanced GOM concentrations of ~450 pg m^{-3} and depleted GEM in one
1106 stratospheric intrusion case and further speculated that the stratosphere was depleted in total Hg
1107 and enriched in GOM, and suggested that stratospheric intrusion could be a source of GOM to
1108 the troposphere. Near Tullahoma, TN, USA the highest GOM concentrations (200 – 500 pg m^{-3})
1109 from flights over a year were observed always at 2 – 4.5 km altitude with a strong *seasonal*
1110 *variation* with a wintertime minimum and a summertime maximum (Brooks et al., 2014). In the
1111 same study, limited PBM measurements exhibited similar levels to GOM at all altitudes.

1112 In a most recent field campaign NOMADSS, the highest Hg(II) concentrations of 300–
1113 680 pg m^{-3} were observed in dry (RH<35 %) and clean air masses during two flights over Texas
1114 at 5–7 km altitude and off the North Carolina coast at 1–3 km altitude (Gratz et al., 2015; Shah et
1115 al., 2016). Gratz et al. (2015) found, using back trajectories, that a segment of air masses with
1116 elevated GOM averaged at $0.266\pm 0.038 \text{ ng m}^{-3}$ and ranging over 0.182 – 0.347 ng m^{-3} at 7 km

1117 altitude over Texas originated from the upper troposphere of the Pacific High. It was speculated
1118 that the stable, dry conditions of large scale anticyclones resulted in a lack of GOM removal by
1119 wet deposition or in-cloud reduction and were thus ideal for GOM accumulation. They
1120 demonstrated that elevated BrOx could persist and that sufficient GOM could be produced
1121 during long-range transport in the Pacific upper troposphere. Their sensitivity analysis suggested
1122 a range of 8 – 13 days required to produce the observed GOM. Shah et al. (2016), using the
1123 GEOS-Chem model with tripled bromine radical concentrations or a faster oxidation rate
1124 constant for GEM + Br, increased modeled Hg(II) concentrations by a factor of 1.5 – 2
1125 improving agreement with the observations, and suggested that the subtropical anticyclones were
1126 significant global sources of Hg(II).

1127 **6. Summary and Recommendations**

1128 This review summarized the general characteristics in GEM, GOM, and PBM
1129 concentrations in the MBL, over land, from low to high latitudes, and from the surface to the
1130 upper troposphere, and further the factors driving such variabilities based on a great wealth of
1131 research in the literature. The Key points are summarized below.

- 1132 1. For MBL TGM/GEM, diurnal variation in most oceanic regions featured noon to
1133 afternoon minimums due probably to in situ oxidation of GEM, while a few
1134 studies showed the opposite pattern over the *Atlantic* and the *equatorial Pacific*
1135 Ocean, attributed to enhanced oceanic evasion linked to enhanced photoreduction
1136 and biological activity. Seasonal to annual variation was generally characterized
1137 as higher (lower) concentrations in colder (warmer) months, which was largely
1138 thought to be caused by less (more) loss via oxidation in colder (warmer) months.
1139 Long term trends have been identified at locations in Mace Head, Ireland,

1140 midlatitudinal Canada, and Cape Point, South Africa, and varied over different
1141 time periods, which was speculated to be associated with changing anthropogenic
1142 and legacy emissions, and redox chemistry.

1143 2. For MBL GOM, diurnal variation was generally characterized with noon to
1144 afternoon peaks and nighttime low values and seasonal variation with higher
1145 concentrations in spring and summer and lower in fall and winter, largely
1146 attributed to GEM photooxidation as often supported by correlation of GOM with
1147 solar radiation and BrO. In one study springtime maximums were also linked to
1148 biological activity and in a few studies annual minimums were associated with
1149 scavenging by precipitation. No long term trends have been reported for oceanic
1150 regions.

1151 3. For MBL PBM, no consistent diurnal and seasonal variation has been identified in
1152 most studies, and only two studies reported seasonal variation with higher
1153 concentrations in fall/winter associated with anthropogenic emissions. One study
1154 showed no consistent diurnal variation in Tekran measurements but a clear diurnal
1155 cycle with maximums at noon and minimums before sunrise using 10-stage
1156 impactor measurements.

1157 4. For continental TGM/GEM, higher concentrations were found at urban sites than
1158 remote, rural, and elevated sites. This result is unbiased by elevated TGM/GEM
1159 from Asian sites. The predominant diurnal pattern was an early morning
1160 minimum and afternoon maximum, opposite to that at urban sites. Diurnal
1161 patterns at surface sites were thought to be driven by surface and local emissions,
1162 boundary layer dynamics, Hg photochemistry, dry deposition, and sequestering by

1163 dew. At elevated sites, mountain-valley winds appeared to be important drivers
1164 of the diurnal cycle. Seasonal variations were influenced by fossil fuel emissions
1165 for winter heating, surface emissions, and monsoons in Asia. At background sites,
1166 long-term declines in TGM were partially attributed to anthropogenic Hg
1167 emission reductions.

1168 5. For continental GOM, concentrations were higher at elevated sites. However, this
1169 result may be biased by a large proportion of high elevation studies from China
1170 where speciated atmospheric mercury are typically elevated. The predominant
1171 diurnal pattern was a noon to mid-afternoon maximum and nighttime minimum,
1172 except for nighttime increases at urban and elevated sites. The driving
1173 mechanisms of the diurnal variations were suggested to include in situ
1174 photochemical production, dry deposition, and scavenging by dew. Entrainment
1175 of GOM from the free troposphere was believed to contribute to nighttime
1176 increases at some elevated sites. No predominant seasonal pattern in GOM was
1177 found, except for higher concentrations in the spring/summer at urban sites.
1178 Photochemical production driven by strong solar radiation and atmospheric
1179 oxidants, free tropospheric transport, anthropogenic emissions, and increased wet
1180 deposition during summer appeared to affect GOM seasonal variation.

1181 6. For continental PBM or TPM, no predominant diurnal pattern was found.
1182 Increases in PBM or TPM were prevalent during colder seasons and were driven
1183 by local/regional coal combustion and wood burning emissions, lower mixing
1184 height, reduced oxidation, and increased gas-particle partitioning.

1185 7. TGM/GEM over the ocean surface decreased from the NH to the SH with the

1186 highest concentrations ($\sim 3.5 \text{ ng m}^{-3}$) in NH midlatitudes and the lowest in SH
1187 ($\sim 0.9 \text{ ng m}^{-3}$). This interhemispheric gradient was believed to suggest the
1188 majority of Hg emissions in NH, contradicting the hypothesis of large oceanic
1189 sources of Hg by previous work. However, in other studies the largest oceanic
1190 source was found in the equatorial region. Airborne measurements of TGM
1191 suggested distinct seasonal variation in latitudinal distributions, a $\sim 50 \text{ ppqv}$ (~ 0.5
1192 ng m^{-3}) increase in GEM concentrations from $\sim 20^\circ\text{N} - 30^\circ\text{N}$ to $60^\circ\text{N} - 90^\circ\text{N}$
1193 latitudes in spring and negligible latitudinal variation in summer. It was
1194 speculated that smaller latitudinal gradient of temperature in summer likely
1195 enhanced meridional circulation resulting in smaller latitudinal variation in GEM
1196 concentration in the troposphere.

1197 8. GEM concentrations remained nearly constant, slightly decreasing with altitude
1198 over the several airborne field campaign regions, and depleted GEM was found in
1199 stratospherically influenced air masses. Abundant GOM has been suggested, but
1200 only very few studies have conducted measurements of free tropospheric GOM
1201 showing concentrations of hundreds of pg m^{-3} , particularly in the area of Pacific
1202 High.

1203 Over two decades of extensive measurements have advanced our knowledge of the
1204 spatiotemporal variation of TGM/GEM, GOM, and PBM in numerous continental and oceanic
1205 environments. However, measurement data, especially those of PBM, remain scarce in the SH,
1206 MBL, and upper air. In oceanic regions most observations, obtained via shipboard
1207 measurements of TGM/GEM with a few exceptions as ground-based on islands, suggested
1208 composite instead of instantaneous variation. Moreover, there are hardly size-fractionated PBM

1209 measurements. The current Tekran speciation unit could only measure PBM <2.5 μm , and
1210 Tekran PBM measurement data from a limited number of MBL and continental monitoring
1211 locations exhibited no definitive diurnal patterns in PBM concentrations. However, impactor
1212 measurements of total PBM in the MBL showed clearly-defined diurnal variation with daily
1213 maximums at around noon and minimums before sunrise. These existing problems impede our
1214 gaining full knowledge of global distributions and temporal variations of speciated Hg.

1215 GEM oxidation is one of the main driving mechanisms of diurnal and seasonal variations
1216 of TGM/GEM and GOM. However, the oxidants that are involved in the photochemical
1217 reactions driving the diurnal and seasonal variations of GOM remain largely unknown/uncertain,
1218 due to the lack of speciated GOM and upper air measurements. This is largely a result of
1219 inadequate technologies and a nebulous understanding of chemical reactions in atmospheric Hg
1220 transformation. Studies such as Chand et al. (2008) estimated GOM concentrations using the
1221 reaction of GEM + OH alone, and Sillman et al. (2007) reproduced observed GOM
1222 concentrations over Florida using CMAQ with gas-phase oxidation of GEM by O₃ and OH only.
1223 However, the reactions of GEM+ O₃ and GEM + OH have been subject to debate between
1224 theoretical and experimental studies, as no mechanism consistent with thermochemistry has been
1225 proposed (Pal and Ariya, 2004; Calvert and Lindberg, 2005; Subir et al., 2011; Ariya et al.,
1226 2015). It was speculated that GEM oxidation in the MBL and the upper troposphere was
1227 possibly largely Br-initiated (Holmes et al., 2009; Gratz et al., 2015; Shah et al., 2016). This
1228 indicated that even if a model reproduced observed concentrations of GOM, the chemistry in the
1229 model was not necessarily correct. So far, most chemical transport models have rarely focused
1230 on diurnal variation of speciated Hg; instead, they mostly focused on reproducing annual and
1231 monthly variations in TGM/GEM (Lei et al., 2013; Song et al., 2015), with large discrepancies

1232 between model simulations and surface measurements of GOM and PBM (Zhang et al., 2012;
1233 Kos et al., 2012). There are too many misrepresentations of Hg science and confounding issues
1234 in current models to gain a full understanding of the driving mechanisms for the observed diurnal
1235 to decadal variation in speciated Hg.

1236 In examining these unresolved questions and issues, the following recommendations for
1237 future research were hence suggested:

- 1238 • Global tropospheric distributions need to be mapped out for TGM/GEM, GOM, and
1239 PBM. Long-term monitoring of atmospheric Hg will need to be continued in time
1240 and space, particularly over oceans and at high altitudes utilizing innovative platforms,
1241 which undoubtedly demands technological breakthroughs in instrumentation.
- 1242 • Future research is warranted on GOM speciation measurements and multiphase redox
1243 kinetics. Field measurement studies need to include more oxidants besides ozone
1244 (and BrO in limited number of studies) in the analysis of diurnal variation.
- 1245 • Monitoring of long-term trends in TGM/GEM needs to continue, and more work is
1246 needed to unravel the causes responsible for the observed trends. Current hypotheses
1247 need to be validated using more extensive, longer datasets and a modeling system that
1248 includes realistic representation of dynamical, physical, and chemical processes in Hg
1249 cycling not only in the atmosphere but also in the ocean and between the two systems.
- 1250 • Size-fractionated PBM measurements are needed, including Hg concentrations on
1251 particles of all sizes, in space and time concurrent with TGM/GEM and GOM
1252 measurements.

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1258

1259 **Table Caption**

1260 Table 1: Summary of predominant temporal patterns of speciated atmospheric mercury at

1261 continental sites in the northern hemisphere

1262

1263 **Figure Captions**

1264 Figure 1. Means and ranges of TGM/GEM (a), GOM (b), and PBM (c) concentrations, estimated
1265 from the values in the literature as shown in Tables S1 – S3, over the Atlantic, Indian, Pacific,
1266 seas over the West Pacific (denoted as Pacific-Seas, only TGM/GEM in this category), seas in
1267 the Mediterranean region (denoted as Mediterranean), Arctic, and Antarctica Ocean. The solid
1268 black squares represent the mean value and the lowest whisker the minimum and the largest the
1269 maximum concentration in the region.

1270

1271 Figure 2. Median and range in TGM/GEM, GOM and PBM by site category (a) and by
1272 geographical region (b). Bar graph represents the median and error bar represents the maximum,
1273 estimated from the values in the literature as shown in Tables S4 – S6.

1274

1275 Figure 3. Compiled values for several marine/oceanic environmental systems. GEM over the
1276 Augusta basin is in red open circles. (Based on the figure from Bagnato et al., 2013)

1277

1278 Figure 4. GEM (ppqv) from the INTEX-B in spring 2006 and ARCTAS in spring and summer
1279 2008 (Data sources: Talbot et al., 2007, 2008; Mao et al., 2010).

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1282 **References**

- 1283 Abbott, M. L., Lin, C. J., Martian, P., and Einerson, J. J.: Atmospheric mercury near Salmon
1284 Falls creek reservoir in southern Idaho, *Appl. Geochem.*, 23(3), 438-453, 2008.
- 1285 Amos, H.M., Jacob, D. J., Kocman, D., Horowitz, H.M., Zhang, Y., Dutkiewicz, S., Horvat, M.,
1286 Corbitt, E.S., Krabbenhoft, D.P. and Sunderland, E.M.: Global biogeochemical implications of
1287 mercury discharges from rivers and sediment burial, M., et al.: Global biogeochemical
1288 implications of mercury discharges from rivers and sediment burial, *Environ. Sci. Technol.*, 48
1289 (16), 9514-9522, 2014.
- 1290 Angot, H., M. Barret¹, O. Magand, M. Ramonet, and A. Dommergue, A 2-year record of
1291 atmospheric mercury species at a background Southern Hemisphere station on Amsterdam Island,
1292 *Atmos. Chem. Phys.*, 14, 11461–11473, 2014.
- 1293 Ariya, P. A., Amyot, M., Dastoor, A., Deeds, D., Feinberg, A., Kos, G., Poulain, A., Ryjkov, A.,
1294 Semeniuk, K., Subir, M., and Toyota, K.: Mercury physicochemical and biogeochemical
1295 transformation in the atmosphere and at atmospheric interfaces: a review and future directions,
1296 *Chem. Rev.*, 115, 3760–3802, doi:10.1021/cr500667e, 2015.
- 1297 Aspino, K., C. Temme, T. Berg, C. Ferrari, P.-A. Gauchard, X. Fain, and G. Wibetoe, Mercury
1298 in the atmosphere, snow, and melt water ponds in the North Atlantic Ocean during Arctic
1299 Summer, *Environ. Sci. Technol.*, 40(13), 4083 – 4089, 2007.
- 1300 Bagnato, E., M. Sproverie, M. Barra, The sea–air exchange of mercury (Hg) in the marine
1301 boundary layer of the Augusta basin (southern Italy): Concentrations and evasion flux,
1302 *Chemosphere*, 93, 2024–2032, 2013.
1303
- 1304 Banic, C. M., S. T. Beauchamp, R. J. Tordon, W. H. Schroeder, A. Steffen, K. A. Anlauf, and H.
1305 K. T. Wong, Vertical distribution of gaseous elemental mercury in Canada, *J. Geophys. Res.*,
1306 108(D9), 4264, doi:10.1029/2002JD002116, 2003.
1307
- 1308 Beldowska, M., D. Saniewska, L. Falkowska, and A. Lewandowska, Mercury in particulate
1309 matter over Polish zone of the southern Baltic Sea, *Atmos. Environ.*, 46, 397-404, 2012.
1310
- 1311 Berg, T., Pfaffhuber, K. A., Cole, A. S., Engelsen, O., and Steffen, A.: Ten-year trends in
1312 atmospheric mercury concentrations, meteorological effects and climate variables at Zeppelin,
1313 Ny-Ålesund, *Atmos. Chem. Phys.*, 13(13), 6575-6586, 2013.
1314
- 1315 Brooks, S. X. Ren, M. Cohen, W. T. Luke, P. Kelley, R. Artz, A. Hynes, W. Landing, and B.
1316 Martos, Airborne Vertical Profiling of Mercury Speciation near Tullahoma, TN, USA, *Atmos.*, 5,
1317 557-574; doi:10.3390/atmos5030557, 2014.
1318

1319 Brown, R. J., Goddard, S. L., Butterfield, D. M., Brown, A. S., Robins, C., Mustoe, C. L., and
1320 McGhee, E. A.: Ten years of mercury measurement at urban and industrial air quality monitoring
1321 stations in the UK, *Atmos. Environ.*, 109, 1-8, 2015.

1322
1323 Brunke, E. G., Labuschagne, C., Ebinghaus, R., Kock, H.H., Slemr, F. Gaseous elemental
1324 mercury depletion events observed at Cape Point during 2007 and 2008, *Atmos. Chem. Phys.* 10,
1325 1121-1131, 2010.

1326
1327 Bullock, O. R. and Brehme, K. A.: Atmospheric mercury simulation using the CMAQ model:
1328 formulation description and analysis of wet deposition results, *Atmos. Environ.*, 36(13), 2135-
1329 2146, 2002.

1330
1331 Calvert, J. G. and Lindberg, S. E.: Mechanisms of mercury removal by O₃ and OH in the
1332 atmosphere, *Atmos. Environ.*, 39(18), 3355-3367, 2005.

1333
1334 Chand, D., et al. (2008), Reactive and particulate mercury in the Asian marine boundary layer,
1335 *Atmos. Environ.*, 28, 7988–7996, doi:10.1016/j.atmosenv.2008.06.048.

1336
1337 Chen, L., Liu, M., Xu, Z., Fan, R., Tao, J., Chen, D., Zhang, D., Xie, D. and Sun, J.: Variation
1338 trends and influencing factors of total gaseous mercury in the Pearl River Delta—A highly
1339 industrialised region in South China influenced by seasonal monsoons, *Atmos. Environ.*, 77,
1340 757-766, 2013.

1341
1342 Chen, L., Y. Zhang, D. J. Jacob, A. L. Soerensen, J. A. Fisher, H.M. Horowitz, E. S. Corbitt, and
1343 X. Wang (2015), A decline in Arctic Ocean mercury suggested by differences in decadal trends
1344 of atmospheric mercury between the Arctic and northern midlatitudes, *Geophys. Res. Lett.*, 42,
1345 6076–6083, doi:10.1002/2015GL064051.

1346
1347 Cheng, I., Zhang, L., Blanchard, P., Dalziel, J., Tordon, R., Huang, J., and Holsen, T. M.:
1348 Comparisons of mercury sources and atmospheric mercury processes between a coastal and
1349 inland site, *J. Geophys. Res. Atmos.*, 118(5), 2434-2443, 2013.

1350
1351 Cheng, I., Zhang, L., Blanchard, P., Graydon, J. A., and St. Louis, V. L.: Source-receptor
1352 relationships for speciated atmospheric mercury at the remote Experimental Lakes Area,
1353 northwestern Ontario, Canada, *Atmos. Chem. Phys.*, 12(4), 1903-1922, 2012.

1354
1355 Cheng, I., Zhang, L., Mao, H., Blanchard, P., Tordon, R., and Dalziel, J.: Seasonal and diurnal
1356 patterns of speciated atmospheric mercury at a coastal-rural and a coastal-urban site, *Atmos.*
1357 *Environ.*, 82, 193-205, 2014.

1358
1359 Choi, E. M., Kim, S. H., Holsen, T. M., and Yi, S. M.: Total gaseous concentrations in mercury
1360 in Seoul, Korea: local sources compared to long-range transport from China and Japan. *Environ.*
1361 *Pollut.*, 157(3), 816-822, 2009.

1362
1363 Choi, H. D., Holsen, T. M., and Hopke, P. K.: Atmospheric mercury (Hg) in the Adirondacks:
1364 Concentrations and sources, *Environ. Sci. Technol.*, 42(15), 5644-5653, 2008.

1365
1366 Choi, H. D., Huang, J., Mondal, S., and Holsen, T. M.: Variation in concentrations of three
1367 mercury (Hg) forms at a rural and a suburban site in New York State, *Sci. Total Environ.*, 448,
1368 96-106, 2013.

1369 Ci, Z. J., Zhang, X. S., Wang, Z. W., Niu, Z. C., Diao, X. Y., Wang, S. W., Distribution and air-
1370 sea exchange of mercury (Hg) in the Yellow Sea, *Atmos. Chem. Phys.*, 11, 2881–2892, doi:
1371 10.5194/acp-11-2881-2011, 2011.

1372 Civerolo, K. L., Rattigan, O. V., Felton, H. D., Hirsch, M. J., and DeSantis, S.: Mercury wet
1373 deposition and speciated air concentrations from two urban sites in New York State: Temporal
1374 patterns and regional context, *Aerosol Air Qual. Res.*, 14(7), 1822-1837, 2014.
1375

1376 Cobbett, F. D., and Van Heyst, B. J.: Measurements of GEM fluxes and atmospheric mercury
1377 concentrations (GEM, RGM and Hgp) from an agricultural field amended with biosolids in
1378 Southern Ont., Canada (October 2004–November 2004), *Atmos. Environ.*, 41(11), 2270-2282,
1379 2007.
1380

1381 Cole, A. S., A. Steffen, K. A. Pfaffhuber, T. Berg, M. Pilote, L. Poissant, R. Tordon, and H.
1382 Hung (2013), Ten-year trends of atmospheric mercury in the high Arctic compared to Canadian
1383 sub-Arctic and mid-latitude sites, *Atmos. Chem. Phys.*, 13, 1535–1545, doi:10.5194/acp-13-
1384 1535-2013.
1385

1386 Cole, A. S., Steffen, A., Eckley, C. S., Narayan, J., Pilote, M., Tordon, R., Graydon, J. A., St.
1387 Louis, V.L., Xu, X., and Branfireun, B. A.: A survey of mercury in air and precipitation across
1388 Canada: patterns and trends, *Atmosphere*, 5(3), 635-668, 2014.

1389 Conaway, C. H., Mason, R. P., Steding, D. J., and Flegal, A. R.: Estimate of mercury emission
1390 from gasoline and diesel fuel consumption, San Francisco Bay area, California, *Atmos. Environ.*,
1391 39, 101–105, 2005.

1392 Dastoor, A. P., and D. A. Durnford, Arctic Ocean: is it a sink of a source of atmospheric mercury?
1393 *Environ. Sci. Technol.*, 48, 1707–1717, 2014.

1394 De More, S. J., J. E. Patterson, D. M. Bibby, Baseline atmospheric mercury studies at Ross
1395 Island, Antarctica, *Antarctic Sci.*, 5(3), 323-326, 1993.

1396 Dibble, T. S., Zelic, M. J., and Mao, H.: Thermodynamics of reactions of ClHg and BrHg
1397 radicals with atmospherically abundant free radicals, *Atmos. Chem. Phys.*, 12(21), 10271-10279,
1398 2012.
1399

1400 Driscoll, C. T., Mason, R. P., Chan, H. M., Jacob, D. J., and Pirrone, N.: Mercury as a global
1401 pollutant: sources, pathways, and effect, *Environ. Sci. Technol.*, 47(10), 4967-4983, 2013.
1402

1403 Ebinghaus, R., and F. Slemr, Aircraft measurements of atmospheric mercury over southern and
1404 eastern Germany, *Atmos. Environ.*, 34, 895-903, 2000.

1405
1406 Ebinghaus, R., F. Slemr, C.A.M. Brenninkmeijer, P. van Velthoven, A. Zahn, M. Hermann, D. A.
1407 O'Sullivan, and D. E. Oram (2007), Emissions of gaseous mercury from biomass burning in
1408 South America in 2005 observed during CARIBIC flights, *Geophys. Res. Lett.*, 34, L08813,
1409 doi:10.1029/2006GL028866.
1410
1411 Ebinghaus, R., Jennings, S. G., Kock, H. H., Derwent, R. G., Manning, A. J., and Spain, T. G.:
1412 Decreasing trends in total gaseous mercury observations in baseline air at Mace Head, Ireland
1413 from 1996 to 2009, *Atmos. Environ.*, 45(20), 3475-3480, 2011.
1414
1415 Ebinghaus, R., Kock, H. H., Coggin, AM, Spain, TG, Jennings, SG, Temme, C., Long term
1416 measurements of atmospheric mercury at Mace Head, Irish west coast, between 1995 and 2001.
1417 *Atmos. Environ.*, 36, 5267 – 76, 2002a.
1418
1419 Ebinghaus, R., Kock, H. H., Temme, C., Einax, J. W., Löwe, A. G., Richter, A., Burrows, J. P.,
1420 Schroeder, W. H., Antarctic springtime depletion of atmospheric mercury, *Environ. Sci.*
1421 *Technol.*, 36, 1238-1244, 2002b.
1422
1423 Eckley, C.S., Parsons, M.T., Mintz, R., Lapalme, M., Mazur, M., Tordon, R., Elleman, R.,
1424 Graydon, J.A., Blanchard, P. and St. Louis, V.: Impact of closing Canada's largest point-source
1425 of mercury emissions on local atmospheric mercury concentrations, *Environ. Sci. Technol.*,
1426 47(18), 10339-10348, 2013.
1427
1428 Engle, M. A., Tate, M. T., Krabbenhoft, D. P., Schauer, J. J., Kolker, A., Shanley, J. B., and
1429 Bothner, M. H.: Comparison of atmospheric mercury speciation and deposition at nine sites
1430 across central and eastern North America, *J. Geophys. Res. Atmos.*, 115(D18), 2010.
1431
1432 Fain, X., Obrist, D., Hallar, A. G., Mccubbin, I., and Rahn, T.: High levels of reactive gaseous
1433 mercury observed at a high elevation research laboratory in the Rocky Mountains, *Atmos. Chem.*
1434 *Phys.*, 9(20), 8049-8060, 2009.
1435
1436 Feddersen, D. M., Talbot, R., Mao, H., and Sive, B. C.: Size distribution of particulate mercury
1437 in marine and coastal atmospheres, *Atmos. Chem. Phys.*, 12(22), 10899-10909, 2012.
1438
1439 Fisher, J. A., D. J. Jacob, A. L. Soerensen, H. M. Amos, E. S. Corbitt, D. G. Streets, Q. Wang, R.
1440 M. Yantosca, and E. M. Sunderland (2013), Factors driving mercury variability in the Arctic
1441 atmosphere and ocean over the past 30 years, *Global Biogeochem. Cycles*, 27, 1226–1235,
1442 doi:10.1002/2013GB004689.
1443
1444 Fisher, J. A., Jacob, D. J., Soerensen, A. L., Amos, H. M., Steffen, A. & Sunderland, E. M.:
1445 Riverine source of Arctic Ocean mercury inferred from atmospheric observations. *Nature*
1446 *Geoscience*, 5(7), 499-504, 2012
1447
1448 Fitzgerald, W.F.: Is mercury increasing in the atmosphere? The need for an atmospheric mercury
1449 network (AMNET). *Water, Air and Soil Pollut.*, 80, 245-254, 1995.
1450

1451 Fitzgerald, W.F., Gill, G.A., Kim, J.P., 1984. An Equatorial Pacific source of atmospheric mercury.
1452 Science, 224, 597-599.
1453

1454 Fitzgerald, W.F., Mason, R.P., 1996. The global mercury cycle: oceanic and anthropogenic
1455 aspects. In: Baeyens, W., Ebinghaus, R., Vasiliev, O. (Eds.), Global and Regional Mercury
1456 Cycles: Sources, Fluxes and Mass Balances. NATO ASI Series 2. Environment, vol. 21. Kluwer
1457 Ac. Pub., Dordrecht, pp. 85-108.
1458

1459 Fostier, A. H. and Michelazzo, P. A.: Gaseous and particulate atmospheric mercury
1460 concentrations in the Campinas Metropolitan Region (Sao Paulo State, Brazil), J. Brazilian
1461 Chem. Soc., 17(5), 886-894, 2006.
1462

1463 Friedli, H. R., L. F. Radke, R. Prescott, P. Li, J.-H. Woo, and G. R. Carmichael (2004), Mercury
1464 in the atmosphere around Japan, Korea, and China as observed during the 2001 ACE-Asia field
1465 campaign: Measurements, distributions, sources, and implications, J. Geophys. Res., 28, D19S25,
1466 doi:10.1029/2003JD004244.
1467

1468 Fu, X. W., Feng, X., Dong, Z. Q., Yin, R. S., Wang, J. X., Yang, Z. R., and Zhang, H.:
1469 Atmospheric gaseous elemental mercury (GEM) concentrations and mercury depositions at a
1470 high-altitude mountain peak in south China, Atmos. Chem. Phys., 10(5), 2425-2437, 2010.
1471

1472 Fu, X. W., Feng, X., Liang, P., Zhang, H., Ji, J., and Liu, P.: Temporal trend and sources of
1473 speciated atmospheric mercury at Waliguan GAW station, Northwestern China, Atmos. Chem.
1474 Phys., 12(4), 1951-1964, 2012a.
1475

1476 Fu, X. W., Feng, X., Shang, L. H., Wang, S. F., and Zhang, H.: Two years of measurements of
1477 atmospheric total gaseous mercury (TGM) at a remote site in Mt. Changbai area, Northeastern
1478 China, Atmos. Chem. Phys., 12, 4215-4226, doi:10.5194/acp-12-4215-2012, 2012b.
1479

1480 Fu, X. W., Zhang, H., Yu, B., Wang, X., Lin, C.-J., and Feng, X. B.: Observations of
1481 atmospheric mercury in China: a critical review, Atmos. Chem. Phys., 15, 9455-9476,
1482 doi:10.5194/acp-15-9455-2015, 2015.
1483

1484 Fu, X., Feng, X., Wang, S., Rothenberg, S., Shang, L., Li, Z., and Qiu, G.: Temporal and spatial
1485 distributions of total gaseous mercury concentrations in ambient air in a mountainous area in
1486 southwestern China: Implications for industrial and domestic mercury emissions in remote areas
1487 in China, Sci. Total Environ., 407(7), 2306-2314, 2009.
1488

1489 Fu, X., Feng, X., Zhang, G., Xu, W., Li, X., Yao, H., Liang, P., Li, J., Sommar, J., Yin, R., and
1490 Liu, N.: Mercury in the marine boundary layer and seawater of the South China Sea:
1491 Concentrations, sea/air flux, and implication for land outflow, J. Geophys. Res., 115, D06303,
1492 doi:10.1029/2009JD012958, 2010.
1493

1494 Fu, X., Feng, X., Zhu, W., Wang, S., and Lu, J.: Total gaseous mercury concentrations in
1495 ambient air in the eastern slope of Mt. Gongga, South-Eastern fringe of the Tibetan plateau,
1496 China, Atmos. Environ., 42(5), 970-979, 2008.

1497
1498 Gay, D. A., Schmeltz, D., Prestbo, E., Olson, M., Sharac, T., and Tordon, R.: The Atmospheric
1499 Mercury Network: measurement and initial examination of an ongoing atmospheric mercury
1500 record across North America, *Atmos. Chem. Phys.*, 13(22), 11339-11349, 2013.
1501 *Geoscience*, 5(7), 499-504.
1502
1503 Gratz, L. E., Ambrose, J.L., Jaffe, D.A., Shah, V., Jaeglé, L., Stutz, J., Festa, J., Spolaor, M., Tsai,
1504 C., Selin, N.E. and Song, S: Oxidation of mercury by bromine in the subtropical Pacific free
1505 troposphere, *Geophys. Res. Lett.*, 42, 10,494–10,502, doi:10.1002/2015GL066645, 2015.
1506
1507 Gratz, L. E., Keeler, G. J., Marsik, F. J., Barres, J. A., and Dvonch, J. T.: Atmospheric transport
1508 of speciated mercury across southern Lake Michigan: Influence from emission sources in the
1509 Chicago/Gary urban area, *Sci. Total Environ.*, 448, 84-95, 2013.
1510
1511 Hall, C.B., Mao, H., Ye, Z., Talbot, R., Ding, A., Zhang, Y., Zhu, J., Wang, T., Lin, C.J., Fu, C.
1512 and Yang, X.: Sources and Dynamic Processes Controlling Background and Peak Concentrations
1513 of TGM in Nanjing, China, *Atmosphere*, 5(1), 124-155, 2014.
1514
1515 Han, Y. J., Kim, J. E., Kim, P. R., Kim, W. J., Yi, S. M., Seo, Y. S., and Kim, S. H.: General
1516 trends of atmospheric mercury concentrations in urban and rural areas in Korea and
1517 characteristics of high-concentration events, *Atmos. Environ.*, 94, 754-764, 2014.
1518
1519 Hedgecock I. M., and N. Pirrone, Chasing quicksilver: modeling the atmospheric lifetime of Hg0
1520 (g) in the marine boundary layer at various latitudes. *Environ. Sci. Technol.*, 38, 69–76, 2004.
1521
1522 Hedgecock, I. M., G. A. Trunfio, N. Pirrone, and F. Sprovieri (2005), Mercury chemistry in the
1523 MBL: Mediterranean case and sensitivity studies using the AMCOTS (Atmospheric Mercury
1524 Chemistry over the Sea) model, *Atmos. Environ.*, 39, 7217– 7230.
1525
1526 Hedgecock, I. M., N. Pirrone, F. Sprovieri, and E. Pesenti (2003), Reactive gaseous mercury in
1527 the marine boundary layer: Modelling and experimental evidence of its formation in the
1528 Mediterranean region, *Atmos. Environ.*, 37, suppl. 1, S41– S49.
1529
1530 Huang, J., Choi, H. D., Hopke, P. K., and Holsen, T. M.: Ambient mercury sources in Rochester,
1531 NY: results from principle components analysis (PCA) of mercury monitoring network data,
1532 *Environ. Sci. Technol.*, 44(22), 8441-8445, 2010.
1533
1534 Holmes, C. D., Jacob, D. J., and Yang, X.: Global lifetime of elemental mercury against
1535 oxidation by atomic bromine in the free troposphere, *Geophys. Res. Lett.*, 33, L20808,
1536 doi:10.1029/2006GL027176, 2006.
1537
1538 Holmes, C. D., Jacob, D. J., Mason, R. P., and Jaffe, D. A.: Sources and deposition of reactive
1539 gaseous mercury in the marine atmosphere, *Atmos. Environ.*, 43(14), 2278-2285, 2009.
1540
1541 Hynes, A. J., Donohoue, D. L., Goodsite, M. E., and Hedgecock, I. M.: Our current
1542 understanding of major chemical and physical processes affecting mercury dynamics in the

1543 atmosphere and at the air-water/terrestrial interfaces, Pirrone, N. and Mason, R. (Eds.), In
1544 Mercury fate and transport in the global atmosphere (pp. 427-457). Springer U.S., 2009.
1545

1546 Jiang, Y., Cizdziel, J. V., and Lu, D.: Temporal patterns of atmospheric mercury species in
1547 northern Mississippi during 2011–2012: Influence of sudden population swings, *Chemosphere*,
1548 93(9), 1694-1700, 2013.

1549 Kang, H, and Z. Xie, Atmospheric mercury over the marine boundary layer observed during the
1550 third China Arctic Research Expedition, *J. Environ. Sci.*, 23(9), 1424–1430, 2011.

1551 Kim, J. and Fitzgerald, W.: Gaseous mercury profiles in the tropical Pacific Ocean, *Geophys.*
1552 *Res. Lett.*, 15(1), 40-43, 1988.

1553 Kim, K. H., Yoon, H. O., Brown, R. J., Jeon, E. C., Sohn, J. R., Jung, K., Park, C. G., and Kim, I.
1554 S.: Simultaneous monitoring of total gaseous mercury at four urban monitoring stations in Seoul,
1555 Korea, *Atmos. Res.*, 132, 199-208, 2013.
1556

1557 Kim, P. R., Han, Y. J., Holsen, T. M., and Yi, S. M.: Atmospheric particulate mercury:
1558 Concentrations and size distributions, *Atmos. Environ.*, 61, 94-102, 2012.
1559

1560 Kolker, A., Olson, M. L., Krabbenhoft, D. P., Tate, M. T., and Engle, M. A.: Patterns of mercury
1561 dispersion from local and regional emission sources, rural Central Wisconsin, USA, *Atmos.*
1562 *Chem. Phys.*, 10(10), 4467-4476, 2010.
1563

1564 Kos G., Ryzhkov A., Dastoor A., Narayan J., Steffen A., Ariya P.A., and Zhang L., 2013.
1565 Evaluation of discrepancy between measured and modelled oxidized mercury species.
1566 *Atmospheric Chemistry and Physics*, 13, 4839-4863.
1567

1568 Kotnik, J., Sprovieri, F., Ogrinc, N., Horvat, M., and Pirrone, N.: Mercury in the Mediterranean,
1569 part I: spatial and temporal trends, *Environ. Sci. Pollut. Res.*, 21(6), 4063-4080, 2014.
1570

1571 Lamborg, C. H., K. R. Rolfhus, and W. F. Fitzgerald, The atmospheric cycling and air-sea
1572 exchange of mercury species in the south and equatorial Atlantic Ocean, *Deep Sea Res.*, Part II,
1573 46, 957– 977, 1999.
1574

1575 Lan, X., Talbot, R., Castro, M., Perry, K., and Luke, W.: Seasonal and diurnal variations of
1576 atmospheric mercury across the US determined from AMNet monitoring data, *Atmos. Chem.*
1577 *Phys.*, 12(21), 10569-10582, 2012.
1578

1579 Lan, X., Talbot, R., Laine, P., Lefer, B., Flynn, J., and Torres, A.: Seasonal and diurnal
1580 variations of total gaseous mercury in urban Houston, TX, USA, *Atmosphere*, 5(2), 399-419,
1581 2014.
1582

1583 Landis, M. S., Lewis, C. W., Stevens, R. K., Keeler, G. J., Dvonch, J. T., and Tremblay, R. T.: Ft.
1584 McHenry tunnel study: Source profiles and mercury emissions from diesel and gasoline powered
1585 vehicles, *Atmos. Environ.*, 41, 8711–8724, 2007.

1586 Laurier, F. J. G., R. P. Mason, L. Whalin, and S. Kato, Reactive gaseous mercury formation in
1587 the North Pacific Ocean's marine boundary layer: A potential role of halogen chemistry, *J.*
1588 *Geophys. Res.*, 108(D17), 4529, doi:10.1029/2003JD003625, 2003.
1589

1590 Laurier, F., and R. Mason, Mercury concentration and speciation in the coastal and open ocean
1591 boundary layer, *J. Geophys. Res.*, 112, D06302, doi:10.1029/2006JD007320.
1592

1593 Lei, H., Liang, X.-Z., Wuebbles, D. J., and Tao, Z.: Model analyses of atmospheric mercury:
1594 present air quality and effects of transpacific transport on the United States, *Atmos. Chem. Phys.*,
1595 13, 10807-10825, doi:10.5194/acp-13-10807-2013, 2013.
1596

1597 Li, J., Sommar, J., Wängberg, I., Lindqvist, O., and Wei, S. Q.: Short-time variation of mercury
1598 speciation in the urban of Göteborg during GÖTE-2005, *Atmos. Environ.*, 42(36), 8382-8388,
1599 2008.
1600

1601 Li, Z., Xia, C., Wang, X., Xiang, Y., and Xie, Z.: Total gaseous mercury in Pearl River Delta
1602 region, China during 2008 winter period, *Atmos. Environ.*, 45(4), 834-838, 2011.
1603

1604 Lindberg, Lindberg, S. E. et al. Dynamic oxidation of gaseous mercury in the Arctic troposphere
1605 at polar sunrise. *Environ. Sci. Tech.* 36, 1245-1256, 2002.
1606

1607 Liu, B., Keeler, G. J., Dvonch, J. T., Barres, J. A., Lynam, M. M., Marsik, F. J., and Morgan, J.
1608 T.: Temporal variability of mercury speciation in urban air. *Atmos. Environ.*, 41(9), 1911-1923,
1609 2007.
1610

1611 Liu, B., Keeler, G. J., Dvonch, J. T., Barres, J. A., Lynam, M. M., Marsik, F. J., and Morgan, J.
1612 T.: Urban-rural differences in atmospheric mercury speciation, *Atmos. Environ.*, 44(16), 2013-
1613 2023, 2010.
1614

1615 Lyman, S. N. and Gustin, M. S.: Speciation of atmospheric mercury at two sites in northern
1616 Nevada, USA, *Atmos. Environ.*, 42(5), 927-939, 2008.
1617

1618 Lyman, S. N., and D. A. Jaffe, Formation and fate of oxidized mercury in the upper troposphere
1619 and lower stratosphere, *Nature Geosci.*, DOI: 10.1038/NGEO1353, 2011.
1620

1621 Lyman, S. N., and Gustin, M. S.: Determinants of atmospheric mercury concentrations in Reno,
1622 Nevada, USA, *Sci. Total Environ.*, 408(2), 431-438, 2009.
1623

1624 Lynam, M. M. and Keeler: Automated speciated mercury measurements in Michigan, *Environ.*
1625 *Sci. Technol.*, 39(23), 9253-9262, 2005.
1626

1627 Malcolm, E. G.; Keeler, G. J.; Landis, M. S. The effects of the coastal environment on the
1628 atmospheric mercury cycle. *J. Geophys. Res.* 2003, 108, article no. 4357.
1629

1630 Manolopoulos, H., Schauer, J. J., Purcell, M. D., Rudolph, T. M., Olson, M. L., Rodger, B., and
1631 Krabbenhoft, D. P.: Local and regional factors affecting atmospheric mercury speciation at a
1632 remote location. *J. Environ. Eng. Sci.*, 6(5), 491-501, 2007.

1633

1634 Mao, H., Talbot, R. W., Sigler, J. M., Sive, B. C., and Hegarty, J. D.: Seasonal and diurnal
1635 variations of Hg over New England, *Atmos. Chem. Phys.*, 8(5), 1403-1421, 2008.

1636

1637 Mao, H. and Talbot, R.: Speciated mercury at marine, coastal, and inland sites in New England–
1638 Part 1: Temporal variability, *Atmos. Chem. Phys.*, 12(11), 5099-5112, 2012.

1639

1640 Mao, H., Talbot, R., Hegarty, J., and Koermer, J.: Speciated mercury at marine, coastal, and
1641 inland sites in New England–Part 2: Relationships with atmospheric physical parameters, *Atmos.*
1642 *Chem. Phys.*, 12(9), 4181-4206, 2012.

1643

1644 Mao, H., Talbot, R. W., Sive, B. C., Kim, S. Y., Blake, D. R., and Weinheimer, A. J.: Arctic
1645 mercury depletion and its quantitative link with halogens, *J. Atmos. Chem.*, 65(2-3), 145-170,
1646 2010.

1647

1648 Marumoto, K., Hayashi, M., and Takami, A.: Atmospheric mercury concentrations at two sites in
1649 the Kyushu Islands, Japan, and evidence of long-range transport from East Asia, *Atmos.*
1650 *Environ.*, 117, 147-155, 2015.

1651

1652 Mason, R. P., and G.-R. Sheu, Role of the ocean in the global mercury cycle, *Global*
1653 *Biogeochem. Cycles*, 16(4), 1093, 10.1029/2001GB001440, 2002.

1654

1655 Mason, R. P.; Lawson, N. M.; Sheu, G.-R. Mercury in the Atlantic Ocean: factors controlling air-
1656 sea exchange of mercury and its distribution in the upper waters. *Deep-Sea Res. II.*, 48, 2829-
1657 2853, 2001.

1658

1659 Mason, R.P., Fitzgerald, W.F., Morel, F.M., 1994. The biogeochemical cycling of elemental
1660 mercury: anthropogenic influences, *Geochimica et Cosmochimica Acta*, 58, 3191–3198.

1661

1662 Mason, R.P., Fitzgerald, W.F., Vandal, G.M., 1992. The sources of mercury in Equatorial Pacific
1663 rain. *Journal of Atmospheric Chemistry* 14, 489-500.

1664

1665 Mazur, M., Mintz, R., Lapalme, M., and Wiens, B.: Ambient air total gaseous mercury
1666 concentrations in the vicinity of coal-fired power plants in Alberta, Canada, *Sci. Total Environ.*,
1667 408(2), 373-381, 2009.

1668

1669 Moore, C., D. Obrist, M. Luria, Atmospheric mercury depletion events at the Dead Sea: Spatial
1670 and temporal aspects, *Atmos. Environ.*, 69, 231-239, 2013.

1671

1672 Müller, D., Wip, D., Warneke, T., Holmes, C. D., Dastoor, A., and Notholt, J.: Sources of
1673 atmospheric mercury in the tropics: continuous observations at a coastal site in Suriname, *Atmos.*
1674 *Chem. Phys.*, 12, 7391–7397, doi:10.5194/acp-12-7391-2012, 2012.

1675

1676 Murphy, D. M., Thomson, D. S., and Mahoney, M. J.: In situ measurements of organics,
1677 meteoritic material, mercury, and other elements in aerosols at 5 to 19 kilometers, *Science*,
1678 282(5394), 1664-1669, 1998.
1679

1680 Murphy, D. M., Hudson, P. K., Thomson, D. S., Sheridan, P. J., and Wilson, J. C.: Observations
1681 of mercury-containing aerosols, *Environ. Sci. Technol.*, 40(10), 3163-3167, 2006.
1682

1683 Nair, U. S., Wu, Y., Walters, J., Jansen, J., and Edgerton, E. S.: Diurnal and seasonal variation
1684 of mercury species at coastal-suburban, urban, and rural sites in the southeastern United States,
1685 *Atmos. Environ.*, 47, 499-508, 2012.
1686

1687

1688 Nguyen, D. L., Kim, J. Y., Shim, S. G., and Zhang, X. S.: Ground and shipboard measurements
1689 of atmospheric gaseous elemental mercury over the Yellow Sea region during 2007–2008.
1690 *Atmos. Environ.*, 45(1), 253-260, 2011.
1691

1692 Obrist, D., E. Tas, M. Peleg, V. Matveev, X. Faïn, D. Asaf, and M. Lur, Bromine-induced
1693 oxidation of mercury in the mid-latitude atmosphere, *Nature Geosci.*, 4, 22 – 26, 2011.
1694

1695 Obrist, D., Hallar, A. G., McCubbin, I., Stephens, B. B., and Rahn, T.: Atmospheric mercury
1696 concentrations at Storm Peak Laboratory in the Rocky Mountains: Evidence for long-range
1697 transport from Asia, boundary layer contributions, and plant mercury uptake, *Atmos. Environ.*,
1698 42(33), 7579-7589, 2008.
1699

1700 Pal, B. and Ariya, P. A.: Gas-phase HO-initiated reactions of elemental mercury: kinetics,
1701 product studies, and atmospheric implications, *Environ. Sci. Technol.*, 38, 5555–5566, 2004.
1702

1703 Parsons, M. T., McLennan, D., Lapalme, M., Mooney, C., Watt, C., and Mintz, R.: Total gaseous
1704 mercury concentration measurements at Fort McMurray, Alberta, Canada, *Atmosphere*, 4(4),
1705 472-493, 2013.
1706

1707 Peterson, C., Gustin, M., and Lyman, S.: Atmospheric mercury concentrations and speciation
1708 measured from 2004 to 2007 in Reno, Nevada, USA, *Atmos. Environ.*, 43(30), 4646-4654, 2009.
1709

1710 Pfaffhuber, K. A., Berg, T., Hirdman, D., and Stohl, A.: Atmospheric mercury observations from
1711 Antarctica: seasonal variation and source and sink region calculations, *Atmos. Chem. Phys.*, 12,
1712 3241–3251, doi:10.5194/acp-12-3241-2012, 2012.
1713

1714 Pirrone, N., Ferrara, R., Hedgecock, I.M., Kallos, G., Mamane, Y., Munthe, J., Pacyna, J.M.,
1715 Pytharoulis, I., Sprovieri, F., Voudouri, A. and Wangberg, I.: Dynamic processes of atmospheric
1716 mercury over the Mediterranean region, *Atmos. Environ.* , 37(S1), 21–40, 2003.
1717

1718 Pirrone, N., Cinnirella, S., Feng, X., Finkelman, R. B., Friedli, H. R., Leaner, J., Mason, R.,
1719 Mukherjee, A. B., Stracher, G. B., Streets, D. G., and Telmer, K.: Global mercury emissions to
1720 the atmosphere from anthropogenic and natural sources, *Atmos. Chem. Phys.*, 10, 5951-5964,
1721 5964, 2010.

1722 Pongratz, R., and K. Heumann, Production of methylated mercury, lead, and cadmium by marine
1723 bacteria as a significant natural source for atmospheric heavy metals in polar regions,
1724 *Chemosphere*, 39, 89– 102, 1999.
1725
1726 Prestbo, E.M. 1997. Mercury speciation in the boundary layer and free troposphere advected to
1727 South Florida: Phase I - Reconnaissance. Report to the Florida Department of Environmental
1728 Protection, Tallahassee. Frontier Geosciences, Seattle, WA.
1729
1730 Radke, L. F., H. R. Friedli, and B. G. Heikes (2007), Atmospheric mercury over the NE Pacific
1731 during spring 2002: Gradients, residence time, upper troposphere lower stratosphere loss, and
1732 long-range transport, *J. Geophys. Res.*, 112, D19305, doi:10.1029/2005JD005828.
1733
1734 Rothenberg, S. E., McKee, L., Gilbreath, A., Yee, D., Connor, M., and Fu, X.: Evidence for
1735 short-range transport of atmospheric mercury to a rural, inland site, *Atmos. Environ.*, 44(10),
1736 1263-1273, 2010.
1737
1738 Rutter, A. P., and Schauer, J. J.: The effect of temperature on the gas–particle partitioning of
1739 reactive mercury in atmospheric aerosols, *Atmos. Environ.*, 41(38), 8647-8657, 2007.
1740
1741 Rutter, A. P., Schauer, J. J., Lough, G. C., Snyder, D. C., Kolb, C. J., Von Klooster, S., Rudolf,
1742 T., Manolopoulos, H., and Olson, M. L.: A comparison of speciated atmospheric mercury at an
1743 urban center and an upwind rural location. *J. Environ. Monit.*, 10(1), 102-108, 2008.
1744
1745 Rutter, A.P., Snyder, D. C., Stone, E.A., Schauer, J.J., Gonzalez-Abraham, R., Molina, L.T.,
1746 Márquez, C., Cárdenas, B. and Foy, B.D.: In situ measurements of speciated atmospheric
1747 mercury and the identification of source regions in the Mexico City Metropolitan Area, *Atmos.*
1748 *Chem. Phys.*, 9(1), 207-220, 2009.
1749
1750 Sander, R.; Keene, W. C.; Pszenny, A. A. P.; Arimoto, R.; Ayers, G. P.; Baboukas, E.; Caine, J.
1751 M.; Crutzen, P. J.; Duce, R. A.; Honninger, G.; Huebert, B. J.; Maenhaut, W.; Mihalopoulos, N.;
1752 Schleicher, N. J., Schäfer, J., Blanc, G., Chen, Y., Chai, F., Cen, K., and Norra, S.: Atmospheric
1753 particulate mercury in the megacity Beijing: Spatio-temporal variations and source
1754 apportionment, *Atmos. Environ.*, 109, 251-261, 2015.
1755
1756 Schroeder, W. H., Anlauf, K. G., Barrie, L. A., Lu, J. Y., Steffen, A., Schneeberger, D. R., and
1757 Berg, T. : Arctic springtime depletion of mercury, *Nature*, 394, 331-332, 1998.
1758
1759 Schroeder, W. H., and Munthe, J.: Atmospheric mercury—an overview, *Atmos. Environ.*, 32(5),
1760 809-822, 1998.

1761 Seiler, W., C. Eberling, and G. Slemr, Global distribution of gaseous mercury in the troposphere,
1762 *Pageoph*, 118, 964 – 974, 1980.

1763 Selin, N. E., D. J. Jacob, R. J. Park, R. M. Yantosca, S. A. Strode, L. Jaegle, and D. Jaffe (2007),
1764 Chemical cycling and deposition of atmospheric mercury: Global constraints from observations,
1765 *J. Geophys. Res.*, 112, D02308, doi:10.1029/2006JD007450.

1766
1767 Shah, V., et al., Origin of oxidized mercury in the summertime free troposphere over the
1768 southeastern US, *Atmos. Chem. Phys.*, 16, 1511–1530, 2016.
1769
1770 Sheu, G. R., Lin, N. H., Wang, J. L., Lee, C. T., Yang, C. F. O., and Wang, S. H.: Temporal
1771 distribution and potential sources of atmospheric mercury measured at a high-elevation
1772 background station in Taiwan, *Atmos. Environ.*, 44(20), 2393-2400, 2010.
1773
1774 Sheu, G.-R., and R. P. Mason (2001), An examination of methods for the measurements of
1775 reactive gaseous mercury in the atmosphere, *Environ. Sci. Technol.*, 35, 1209–1216.
1776
1777 Sheu, G.-R., and R. P. Mason (2004), An examination of the oxidation of elemental mercury in
1778 the presence of halide surfaces, *J. Atmos. Chem.*, 48, 107–130.
1779
1780 Sheu, G.-R., Speciation and distribution of atmospheric mercury: Significance of reactive
1781 gaseous mercury in the global mercury cycle, Ph.D. thesis, 170 pp., Univ. of Md., College Park,
1782 2001.
1783
1784 Sigler, J. M., Mao, H., and Talbot, R.: Gaseous elemental and reactive mercury in Southern New
1785 Hampshire, *Atmos. Chem. Phys.*, 9(6), 1929-1942, 2009a.
1786
1787 Sigler, J. M., Mao, H., Sive, B., and Talbot, R., Oceanic Influence on Atmospheric Mercury at
1788 Coastal and Inland Sites: A Springtime Nor'easter in New England, *Atmos. Chem. Phys.*,
1789 9, 4023-4030, 2009b.
1790
1791 Sillman, S., Marsik, F. J., Al-Wali, K. I., Keeler, G. J., and Landis, M. S.: Reactive mercury in
1792 the troposphere: Model formation and results for Florida, the northeastern United States, and the
1793 Atlantic Ocean, *J. Geophys. Res. Atmos.*, 112(D23305), doi:10.1029/2006JD008227, 2007.
1794
1795 Siudek, P., Frankowski, M., and Siepak, J.: Atmospheric particulate mercury at the urban and
1796 forest sites in central Poland, *Environ. Sci. Pollut. Res.*, 23(3), 2341-2352, 2016.
1797
1798 Slemr, F., Angot, H., Dommergue, A., Magand, O., Barret, M., Weigelt, A., Ebinghaus, R.,
1799 Brunke, E.-G., Pfaffhuber, K. A., Edwards, G., Howard, D., Powell, J., Keywood, M., and Wang,
1800 F.: Comparison of mercury concentrations measured at several sites in Southern Hemisphere,
1801 submitted to *Atmos. Chem. Phys.*, 15, 3125–3133, 2015.
1802
1803 Slemr, F., Brunke, E.-G., Ebinghaus, R., Kuss, J., Worldwide trend of atmospheric mercury since
1804 1995. *Atmos. Chem. Phys.*, 11, 4779-4787, 2011.
1805
1806 Slemr, F., E.-G. Brunke, C. Labuschagne, and R. Ebinghaus (2008), Total gaseous mercury
1807 concentrations at the Cape Point GAW station and their seasonality, *Geophys. Res. Lett.*, 35,
1808 L11807, doi:10.1029/2008GL033741.

1809 Slemr, F., G. Schuster, W. Seiler, Distribution, speciation, and budget of atmospheric mercury, *J.*
1810 *Atmos. Chem.*, 3(4), 407-434, 1985.

1811 Slemr, F., Trends in atmospheric mercury concentrations over the Atlantic ocean and the Wank
1812 summit, and the resulting constraints on the budget of atmospheric mercury, in “*Global and*
1813 *Regional Mercury Cycles: Sources, Fluxes and Mass Balances*”, eds. W. Bayyens, R. Ebinghaus,
1814 and O. Vasiliev, NATO-ASI-Series, Vol. 21, pp 33-84, Kluwer Academic Publishers, Dordrecht,
1815 The Netherlands, 1996.

1816 Slemr, F., W. JunkermannR, .W. H. Schmidt, and R. Sladkovic, Indication of change in global
1817 and regional trends of atmospheric mercury concentrations, *Geophys. Res. Lett.*, 22(16), 2143-
1818 2146, 1995.

1819
1820 Slemr, F. and Langer, E.: Increase in global atmospheric concentrations of mercury inferred from
1821 measurements over the Atlantic Ocean, *Nature* **355**, 434 – 437, 1992.

1822 Slemr, F., W. Seiler, and G. Schuster, Latitudinal distribution of mercury over the Atlantic
1823 Ocean, *J. Geophys. Res.*, 86, C2, 1159-1166, 1981.

1824 Soerensen, A. L., D. J. Jacob, D. G. Streets, M. L. I. Witt, R. Ebinghaus, R. P. Mason, M.
1825 Andersson, and E.M. Sunderland (2012), Multi-decadal decline of mercury in the North Atlantic
1826 atmosphere explained by changing subsurface seawater concentrations, *Geophys. Res. Lett.*, 39,
1827 L21810, doi:10.1029/2012GL053736.

1828
1829 Soerensen, A. L., H. Skov, D. J. Jacob, B. T. Soerensen, M. S. Johnson, Global Concentrations
1830 of Gaseous Elemental Mercury and Reactive Gaseous Mercury in the Marine Boundary Layer,
1831 *Environ Sci Technol.*, 44(19), 7425-30. doi: 10.1021/es903839n, 2010.

1832
1833 Sommar, J., M. E. Anersson, and H.-W. Jacobi, Circumpolar measurements of speciated mercury,
1834 ozone and carbon monoxide in the boundary layer of the Arctic Ocean, *Atmos. Chem. Phys.*, 10,
1835 5031–5045, 2010.

1836
1837 Song, S., Selin, N. E., Soerensen, A. L., Angot, H., Artz, R., Brooks, S., Brunke, E.-G., Conley,
1838 G., Dommergue, A., Ebinghaus, R., Holsen, T. M., Jaffe, D. A., Kang, S., Kelley, P., Luke, W.
1839 T., Magand, O., Marumoto, K., Pfaffhuber, K. A., Ren, X., Sheu, G.-R., Slemr, F., Warneke, T.,
1840 Weigelt, A., Weiss-Penzias, P., Wip, D. C., and Zhang, Q.: Top-down constraints on
1841 atmospheric mercury emissions and implications for global biogeochemical cycling, *Atmos.*
1842 *Chem. Phys.*, 15, 7103-7125, doi:10.5194/acp-15-7103-2015, 2015.

1843
1844 Song, X., Cheng, I., and Lu, J.: Annual atmospheric mercury species in downtown Toronto,
1845 Canada, *J. Environ. Monit.*, 11(3), 660-669, 2009.

1846
1847 Sprovieri, F. and Pirrone, N., Spatial and temporal distribution of atmospheric mercury species
1848 over the Adriatic Sea, *Environ. Fluid Mech.*, 8, 117–128, doi:10.1007/s10652-007-9045-4, 2008.

1849
1850 Sprovieri, F., I. M. Hedgecock, and N. Pirrone, An investigation of the origins of reactive
1851 gaseous mercury in the Mediterranean marine boundary layer, *Atmos. Chem. Phys.*, 10, 3985–
1852 3997, 2010a.

1853

1854 Sprovieri, F., N. Pirrone, I. M. Hedgecock, M. S. Landis, and R. K. Stevens, Intensive
1855 atmospheric mercury measurements at Terra Nova Bay in Antarctica during November and
1856 December 2000, *J. Geophys. Res.*, 107(D23), 4722, doi:10.1029/2002JD002057, 2002.

1857 Sprovieri, F., N. Pirrone, K. Gärfeldt, and J. Sommar, Mercury measurements in the marine
1858 boundary layer along a 6000 km cruise path around the Mediterranean Sea, *Atmos. Environ.*, 37
1859 suppl., S63-S71, 2003.

1860 Sprovieri, F., Pirrone, N., Ebinghaus, R., Kock, H., and Dommergue, A.: A review of worldwide
1861 atmospheric mercury measurements, *Atmos. Chem. Phys.*, 10(17), 8245-8265, 2010b.
1862

1863 Stamenkovic, J., Lyman, S., and Gustin, M. S.: Seasonal and diel variation of atmospheric
1864 mercury concentrations in the Reno (Nevada, USA) airshed, *Atmos. Environ.*, 41(31), 6662-
1865 6672, 2007.
1866

1867 Steffen, A., Douglas, T., Amyot, M., Ariya, P., Aspö, K., Berg, T., Bottenheim, J., Brooks, S.,
1868 Cobbett, F., Dastoor, A. and Dommergue, A.: A synthesis of atmospheric mercury depletion
1869 event chemistry in the atmosphere and snow, *Atmos. Chem. Phys.*, 8(6), 1445-1482, 2008.
1870

1871 Steffen, A., Bottenheim, J., Cole, A., Douglas, T.A., Ebinghaus, R., Friess, U., Netcheva, S.,
1872 Nghiem, S., Sihler, H. and Staebler, R.: Atmospheric mercury over sea ice during the OASIS-
1873 2009 campaign, *Atmos. Chem. Phys.*, 13, 7007–7021, 2013.
1874

1875 Strode, S. A., L. Jaegle, N. E. Selin, D. J. Jacob, R. J. Park, R. M. Yantosca, R. P. Mason, and F.
1876 Slemr (2007), Air-sea exchange in the global mercury cycle, *Global Biogeochem. Cycles*, 21,
1877 GB1017, doi:10.1029/2006GB002766.
1878

1879 Subir, M., Ariya, P. A., and Dastoor, A. P.: A review of uncertainties in atmospheric modeling of
1880 mercury chemistry I. Uncertainties in existing kinetic parameters – fundamental limitations and
1881 the importance of heterogeneous chemistry, *Atmos. Environ.*, 45, 5664–5676, 2011.

1882 Swartzendruber, P. C., D. Chand, D. A. Jaffe, J. Smith, D. Reidmiller, L. Gratz, J. Keeler, S.
1883 Strode, L. Jaegle, and R. Talbot (2008), Vertical distribution of mercury, CO, ozone, and aerosol
1884 scattering coefficient in the Pacific Northwest during the spring 2006 INTEX-B campaign, *J.*
1885 *Geophys. Res.*, 113, D10305, doi:10.1029/2007JD009579.
1886

1887 Swartzendruber, P.C., Jaffe, D.A., Prestbo, E.M., Weiss-Penzias, P., Selin, N.E., Park, R., Jacob,
1888 D.J., Strode, S. and Jaegle, L.: Observations of reactive gaseous mercury in the free troposphere
1889 at the Mount Bachelor Observatory, *J. Geophys. Res. Atmos.*, 111(D24301),
1890 doi:10.1029/2006JD007415, 2006.
1891

1892 Talbot, R., H. Mao, E. Scheuer, J. Dibb, and M. Avery (2007), Total Depletion of Hg^o in the
1893 Upper Troposphere - Lower Stratosphere, *Geophys. Res. Lett.*, 34, L23804,
1894 doi:10.1029/2007GL031366.
1895

1896 Talbot, R., H. Mao, E. Scheuer, J. Dibb, M. Avery, E. Browell, G. Sachse, S. Vay, D. Blake, G.
1897 Huey, and H. Fuelberg (2008), Factors influencing the large-scale distribution of Hg^o in the
1898 Mexico City area and over the North Pacific, *Atmos. Chem. Phys.* 8, 2103-2114.
1899
1900 Temme, C., F. Slemr, R. Ebinghaus, J.W. Einax, Distribution of mercury over the Atlantic Ocean
1901 in 1996 and 1999–2001, *Atmos. Environ.*, 37, 1889–1897, 2003a.
1902
1903 Temme, C., J.W. Einax, R. Ebinghaus, and W. H. Schroeder, Measurements of mercury species
1904 at a coastal site in the Antarctic and over the South Atlantic Ocean in the polar summer, *Environ.*
1905 *Sci. Technol.*, 37, 1, 22 – 31, 2003b.
1906
1907 Timonen, H., J. L. Ambrose, and D. A. Jaffe, Oxidation of elemental Hg in anthropogenic and
1908 marine airmasses, *Atmos. Chem., Phys.*, 13, 2827–2836, 2013.
1909
1910 Tseng, C.M., C.S. Liu, and C. Lamborg (2012) Seasonal changes in gaseous elemental mercury
1911 in relation to monsoon cycling over the northern South China Sea, *Atmospheric Chemistry and*
1912 *Physics*, 12, 7341-7350.
1913
1914 Turekian, V. C.; Van Dingenen, R. Inorganic bromine in the marine boundary layer: a critical
1915 review, *Atmos. Chem. Phys.*, 3, 1301-1336, 2003.
1916
1917 UNEP, Global Mercury Assessment 2013: Sources, Emissions, Releases and Environmental
1918 Transport. UNEP Chemicals Branch, Geneva, Switzerland, 2013.
1919
1920 Valente, R. J., Shea, C., Humes, K. L., and Tanner, R. L.: Atmospheric mercury in the Great
1921 Smoky Mountains compared to regional and global levels, *Atmos. Environ.*, 41(9), 1861-1873,
1922 2007.
1923
1924 Wan, Q., Feng, X., Lu, J., Zheng, W., Song, X., Han, S., and Xu, H.: Atmospheric mercury in
1925 Changbai Mountain area, northeastern China I. The seasonal distribution pattern of total gaseous
1926 mercury and its potential sources, *Environ. Res.*, 109(3), 201-206, 2009a.
1927
1928 Wan, Q., Feng, X., Lu, J., Zheng, W., Song, X., Li, P., Han, S. and Xu, H., Atmospheric mercury
1929 in Changbai Mountain area, northeastern China II. The distribution of reactive gaseous mercury
1930 and particulate mercury and mercury deposition fluxes, *Environ. Res.*, 109(6), 721-727, 2009b.
1931
1932 Wang, F., A. Saiz-Lopez, A. S. Mahajan, J. C. Gómez Martín, D. Armstrong, M. Lemes, T.
1933 Hay, and C. Prados-Roman, Enhanced production of oxidised mercury over the tropical Pacific
1934 Ocean: a key missing oxidation pathway, *Atmos. Chem. Phys.*, 14, 1323–1335, 2014.
1935
1936 Wang, Y., Huang, J., Hopke, P. K., Rattigan, O. V., Chalupa, D. C., Utell, M. J., and Holsen, T.
1937 M.: Effect of the shutdown of a large coal-fired power plant on ambient mercury species,
1938 *Chemosphere*, 92(4), 360-367, 2013.
1939
1940 Weigelt, A., Ebinghaus, R., Manning, A. J., Derwent, R. G., Simmonds, P. G., Spain, T. G.,
1941 Jennings, S. G., and Slemr, F.: Analysis and interpretation of 18 years of mercury observations

1942 since 1996 at Mace Head at the Atlantic Ocean coast of Ireland, *Atmos. Environ.* 100, 85–93,
1943 2015.

1944

1945 Weiss-Penzias, P. S., E. J. Williams, B. M. Lerner, T. S. Bates, C. Gaston, K. Prather, A.
1946 Vlasenko, and S. M. Li (2013), Shipboard measurements of gaseous elemental mercury along the
1947 coast of Central and Southern California. *J. Geophys. Res. Atmos.*, 118, 208–219,
1948 doi:10.1029/2012JD018463.

1949

1950 Weiss-Penzias, P., Amos, H.M., Selin, N.E., Gustin, M.S., Jaffe, D.A., Obrist, D., Sheu, G.R.
1951 and Giang, A.: Use of a global model to understand speciated atmospheric mercury observations
1952 at five high-elevation sites, *Atmos. Chem. Phys.*, 15(3), 1161-1173, 2015.

1953

1954 Weiss-Penzias, P., D. A. Jaffe, A. McClintick, E. M. Presbo, and M. S. Landis, Gaseous
1955 elemental mercury in the Marine Boundary Layer: Evidence for rapid removal in anthropogenic
1956 pollution, *Environ. Sci. Technol.*, 3755 – 3763, 37, 2003.

1957

1958 Weiss-Penzias, P., Gustin, M. S., and Lyman, S. N.: Observations of speciated atmospheric
1959 mercury at three sites in Nevada: Evidence for a free tropospheric source of reactive gaseous
1960 mercury, *J. Geophys. Res. Atmos.*, 114(D14), 2009.

1961

1962 Weiss-Penzias, P., Jaffe, D., Swartzendruber, P., Hafner, W., Chand, D., and Prestbo, E.:
1963 Quantifying Asian and biomass burning sources of mercury using the Hg/CO ratio in pollution
1964 plumes observed at the Mount Bachelor Observatory, *Atmos. Environ.*, 41(21), 4366-4379,
1965 2007.

1966

1967 Weiss-Penzias, P.S., Gay, D.A., Brigham, M.E., Parsons, M.T., Gustin, M.S. and ter Schure, A.:
1968 Trends in mercury wet deposition and mercury air concentrations across the US and Canada. *Sci.*
1969 *Total Environ.*, in press, 2016.

1970

1971 Williston, S. H. (1968), Mercury in the atmosphere, *J. Geophys. Res.*, 73(22), 7051–7055,
1972 doi:10.1029/JB073i022p07051.

1973

1974 Witt, M. L. I., Mather, T. A., Baker, A. R., De Hoog, J. C. M., and Pyle, D. M.: Atmospheric
1975 trace metals over the south-west Indian Ocean: total gaseous mercury, aerosol trace metal
1976 concentrations and lead isotope ratios, *Mar. Chem.*, 121, 2–16, 2010.

1977

1978 Witt, M. L. I., Meheran, N., Mather, T. A., De Hoog, J. C. M., and Pyle, D. M.: Aerosol trace
1979 metals, particle morphology and total gaseous mercury in the atmosphere of Oxford, UK, *Atmos.*
1980 *Environ.*, 44(12), 1524-1538, 2010.

1981

1982 Won, J. H., Park, J. Y., and Lee, T. G., Mercury emissions from automobiles using gasoline,
1983 diesel, and LPG, *Atmos. Environ.*, 41, 7547-7552, 2007.

1984

1985 Xia, C., Z. Xie, and L. Sun, Atmospheric mercury in the marine boundary layer along a cruise
1986 path from Shanghai, China to Prydz Bay, Antarctica, *Atmos. Environ.*, 44, 1815-1821, 2010.

1987

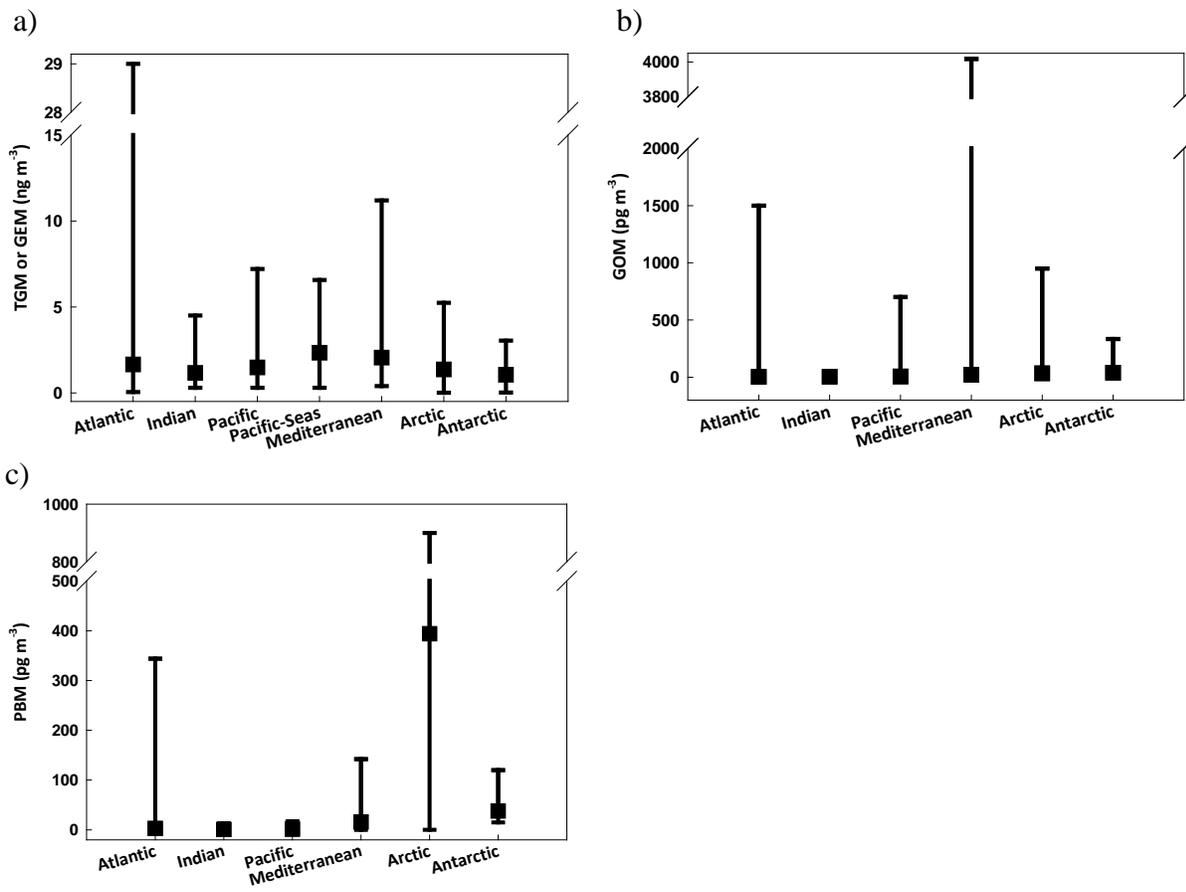
1988 Xiu, G., Cai, J., Zhang, W., Zhang, D., Büeler, A., Lee, S., Shen, Y., Xu, L., Huang, X. and
1989 Zhang, P.: Speciated mercury in size-fractionated particles in Shanghai ambient air, *Atmos.*
1990 *Environ.*, 43(19), 3145-3154, 2009.
1991
1992 Xu, L., Chen, J., Yang, L., Niu, Z., Tong, L., Yin, L., and Chen, Y.: Characteristics and sources
1993 of atmospheric mercury speciation in a coastal city, Xiamen, China, *Chemosphere*, 119, 530-539,
1994 2015.
1995
1996 Xu, X. and Akhtar, U. S.: Identification of potential regional sources of atmospheric total
1997 gaseous mercury in Windsor, Ontario, Canada using hybrid receptor modeling, *Atmos. Chem.*
1998 *Phys.*, 10(15), 7073-7083, 2010.
1999
2000 Xu, X., Akhtar, U., Clark, K., and Wang, X.: Temporal variability of atmospheric total gaseous
2001 mercury in Windsor, ON, Canada, *Atmosphere*, 5(3), 536-556, 2014.
2002 Yang, Y., Chen, H., and Wang, D.: Spatial and temporal distribution of gaseous elemental
2003 mercury in Chongqing, China, *Environ. Monit. Assess.*, 156(1-4), 479-489, 2009.
2004
2005 Yatavelli, R. L., Fahrni, J. K., Kim, M., Crist, K. C., Vickers, C. D., Winter, S. E., and Connell,
2006 D. P.: Mercury, PM 2.5 and gaseous co-pollutants in the Ohio River Valley region: Preliminary
2007 results from the Athens supersite, *Atmos. Environ.*, 40(34), 6650-6665, 2006.
2008
2009 Yu, J. et al., High variability of atmospheric mercury in the summertime boundary layer through
2010 the central Arctic Ocean. *Scientific Reports*, 4, 6091, DOI:10.1038/srep06091, 2014.
2011
2012 Zhang, H., Fu, X. W., Lin, C.-J., Wang, X., and Feng, X. B.: Observation and analysis of
2013 speciated atmospheric mercury in Shangri-La, Tibetan Plateau, China, *Atmos. Chem. Phys.*, 15,
2014 653-665, doi:10.5194/acp-15-653-2015, 2015.
2015
2016 Zhang, L., Wright L.P., and Blanchard P., 2009. A review of current knowledge concerning dry
2017 deposition of atmospheric mercury. *Atmospheric Environment*, 43, 5853-5864.
2018
2019 Zhang, L., Blanchard P., Johnson D., Dastoor A., Ryzhkov A., Lin C.-J., Vijayaraghavan K.,
2020 Gay D., Holsen T.M., Huang J., Graydon J.A., St. Louis V.L., Castro M.S., Miller E.K., Marsik
2021 F., Lu J., Poissant L., Pilote M., and Zhang K.M., 2012. Assessment of modelled mercury
2022 deposition over the Great Lakes region. *Environmental Pollution*, 161, 272-283.
2023
2024 Zhang, L., Wang, S. X., Wang, L., and Hao, J. M.: Atmospheric mercury concentration and
2025 chemical speciation at a rural site in Beijing, China: implications of mercury emission sources,
2026 *Atmos. Chem. Phys.*, 13(20), 10505-10516, 2013.
2027
2028 Zhang, Y., Jacob, D.J., Horowitz, H.M., Chen, L., Amos, H.M., Krabbenhoft, D.P., Slemr, F., St.
2029 Louis, V.L. and Sunderland, E.M.: Observed decrease in atmospheric mercury explained by
2030 global decline in anthropogenic emissions, *Proceed. Natl. Acad. Sci.*, 113(3), 526-531, 2016.
2031

2032 Zhu, J., Wang, T., Talbot, R., Mao, H., Hall, C.B., Yang, X., Fu, C., Zhuang, B., Li, S., Han, Y.
2033 and Huang, X.: Characteristics of atmospheric total gaseous mercury (TGM) observed in urban
2034 Nanjing, China, *Atmos. Chem. Phys.*, 12(24), 12103-12118, 2012.
2035
2036 Zielonka, U., Hlawiczka, S., Fudala, J., Wängberg, I., and Munthe, J.: Seasonal mercury
2037 concentrations measured in rural air in Southern Poland: Contribution from local and regional
2038 coal combustion, *Atmos. Environ.*, 39(39), 7580-7586, 2005.
2039

2040 Table 1: Summary of predominant temporal patterns of speciated atmospheric mercury at
 2041 continental sites in the northern hemisphere

	Diurnal variation	Seasonal variation
<i>TGM/GEM</i>		
Rural	Daytime maximum, nighttime minimum	Winter-spring maximum and summer-fall minimum
Urban	Nighttime maximum, daytime minimum	No predominant pattern
High elevation	Daytime maximum, nighttime minimum	Winter-spring maximum and summer-fall minimum
<i>GOM</i>		
Rural	Midday to late afternoon maximum, nighttime minimum	No predominant pattern
Urban		Spring or summer maximum
High elevation	*Exception: nighttime maximum at urban and elevated sites	No predominant pattern
<i>PBM</i>		
Rural	No predominant pattern	Maximum during heating season
Urban	No predominant pattern	Maximum during heating season
		*Exception: summer maximum
High elevation	No predominant pattern	Maximum during heating season

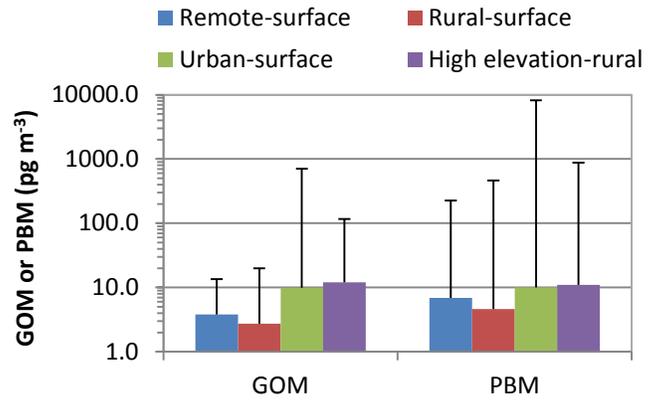
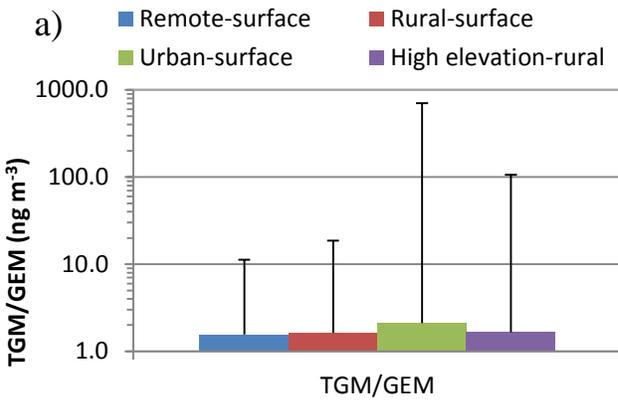
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 2043



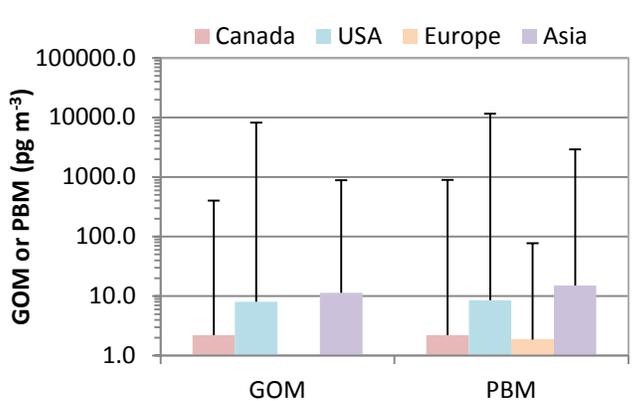
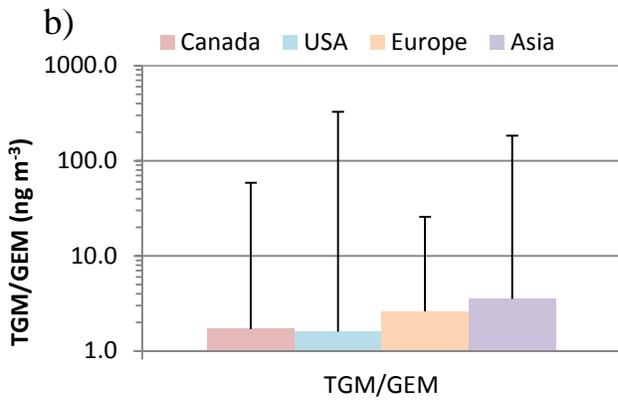
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Figure 1. Means and ranges of TGM/GEM (a), GOM (b), and PBM (c) concentrations, estimated from the values in the literature as shown in Tables S1 – S3, over the Atlantic, Indian, Pacific, seas over the West Pacific (denoted as Pacific-Seas, only TGM/GEM in this category), seas in the Mediterranean region (denoted as Mediterranean), Arctic, and Antarctica Ocean. The solid black squares represent the mean value and the lowest whisker the minimum and the largest the maximum concentration in the region.

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Figure 2. Median and range in TGM/GEM, GOM and PBM by site category (a) and by geographical region (b). Bar graph represents the median and error bar represents the maximum, estimated from the values in the literature as shown in Tables S4 – S6.

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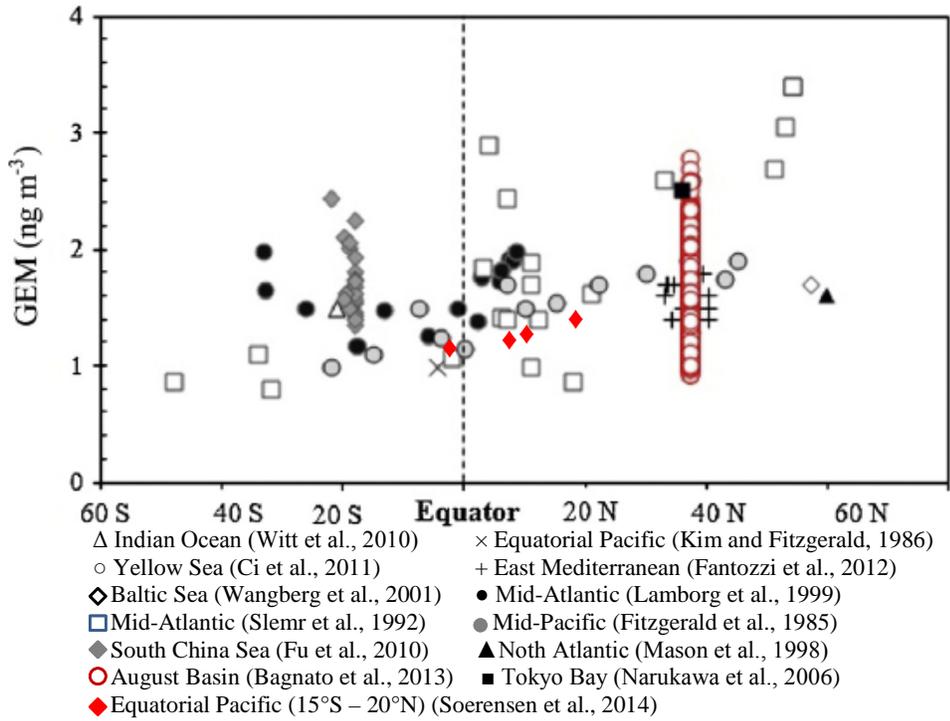


Figure 3. Compiled values for several marine/oceanic environmental systems adapted mostly from Bagnato et al. (2013)

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2100

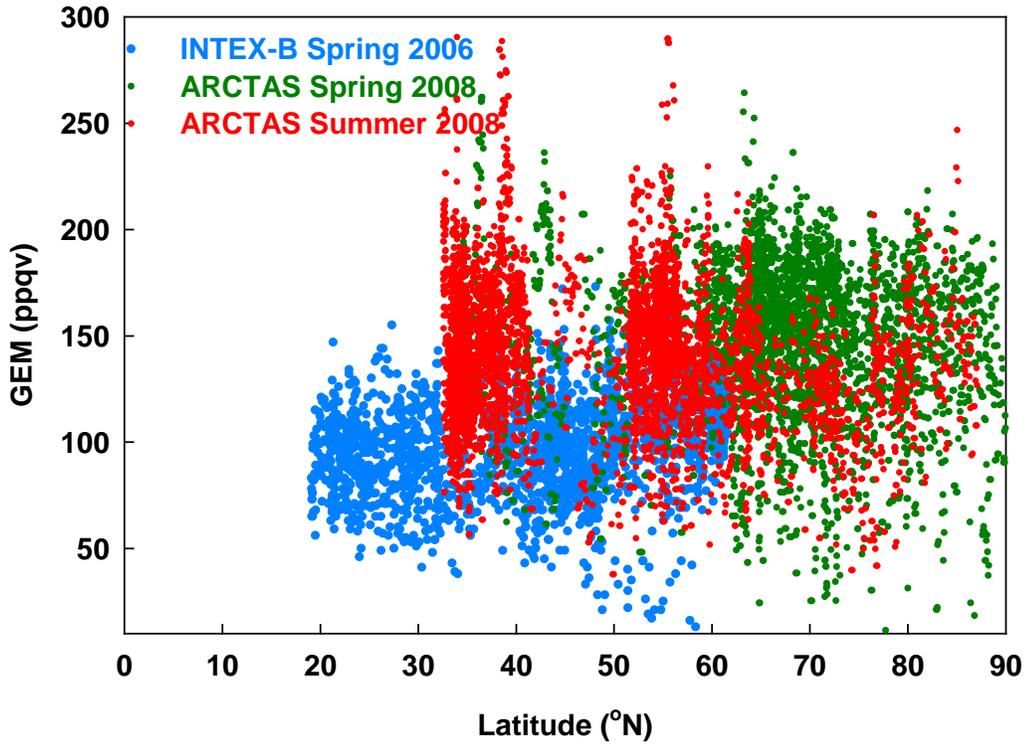


Figure 4. GEM (ppqv) from the INTEX-B in spring 2006 and ARCTAS in spring and summer 2008 (Data sources: Talbot et al., 2007, 2008; Mao et al., 2010).