

1 **Speciated Atmospheric Mercury on Haze and Non-haze Days in an**
2 **Inland City in China**

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25 **Abstract.** Long-term continuous measurements of speciated atmospheric mercury
26 were conducted from July 2013 to June 2014 at Hefei, a mid-latitude inland city in
27 east central China that experiences frequent haze pollution. The mean concentrations
28 (\pm standard deviation) of gaseous elemental mercury (GEM), gaseous oxidized
29 mercury (GOM) and particle-bound mercury (PBM) were $3.95 \pm 1.93 \text{ ng m}^{-3}$, $2.49 \pm$
30 2.41 pg m^{-3} and $23.3 \pm 90.8 \text{ pg m}^{-3}$, respectively, on non-haze days, and $4.74 \pm 1.62 \text{ ng}$
31 m^{-3} , $4.32 \pm 8.36 \text{ pg m}^{-3}$ and $60.2 \pm 131.4 \text{ pg m}^{-3}$, respectively, on haze days. Potential
32 source contribution function (PSCF) analysis suggested that atmospheric mercury
33 pollution on haze days was caused primarily by local emissions, instead of via
34 long-range transport. The poorer mixing conditions on haze days also favored the
35 accumulation of atmospheric mercury. Compared to GEM and GOM, PBM was
36 especially sensitive to haze pollution. The mean PBM concentration on haze days was
37 2.5 times that on non-haze days due to elevated concentrations of particulate matter.
38 PBM also showed a clear seasonal trend; its concentration was the highest in autumn
39 and winter, decreased rapidly in spring, and was the lowest in summer, following the
40 same order in the frequency of haze days in different seasons. On both non-haze and
41 haze days, GOM concentrations remained low at night, but increased rapidly just
42 before sunrise, which could be due to diurnal variation in air exchange between the
43 boundary layer and free troposphere. **However, non-haze and haze days showed**
44 **different trends in daytime GEM and GOM concentrations. On non-haze days, GEM**
45 **and GOM declined synchronously through afternoon, probably due to the retreat of**
46 **the free tropospheric air as the height of the atmospheric boundary layer increases. In**
47 **contrast, on haze days, GOM and GEM showed opposite trends with the highest**
48 **GOM and lowest GEM observed in the afternoon, suggesting the occurrence of**

49 **photochemical oxidation.** This is supported by simple box-model calculations, which
50 showed that oxidation of GEM to GOM does occur and that the transport of free
51 tropospheric GOM alone is not large enough to account for the observed increase in
52 daytime GOM. Our results further postulate that NO₂ aggregation with the HgOH
53 intermediate may be a potential mechanism for the enhanced production of GOM
54 during daytime.

55

56 **1. Introduction**

57 Mercury (Hg) is an environmental pollutant that has received much global
58 attention because of its toxicity and bioaccumulation in the aquatic ecosystems. The
59 most important transport pathway of mercury is via the atmosphere (Schroeder and
60 Munthe, 1998; Lindqvist and Rodhe, 1985). Operationally, atmospheric mercury is
61 commonly differentiated into three forms: gaseous elemental mercury (GEM), gaseous
62 oxidized mercury (GOM) and particle-bound mercury (PBM). The sum of these three
63 atmospheric species of mercury is defined as total atmospheric mercury
64 (TAM=GEM+GOM+PBM), and the sum of GEM and GOM is known as total
65 gaseous mercury (TGM=GEM+GOM) (Gustin and Jaffe, 2010; Gustin et al., 2015).
66 Globally GEM is the dominant form of atmospheric mercury, accounting for over 95%
67 of the total. GEM is stable in the troposphere with a long residence time (0.5–2 yr)
68 and can be transported at the regional to global scale (Schroeder and Munthe,
69 1998; Lindberg et al., 2007). GEM can be photochemically oxidized to GOM, which
70 can be converted to PBM upon adsorption on aerosol surfaces. Different from GEM,
71 GOM and PBM can be readily removed from the air by wet and dry deposition as a
72 result of their high surface affinity and water solubility (Lindqvist and Rodhe, 1985).

73 Thus, chemical transformation between GEM, GOM and PBM will directly influence
74 the atmospheric lifetime of mercury.

75 As a result of rapid industrial development and economic growth in recent
76 decades, China has become one of the major contributors to anthropogenic mercury
77 emissions to the environment (Wu et al., 2006; Pacyna et al., 2006; Pacyna et al.,
78 2010; Zhang et al., 2015b). Atmospheric mercury emissions from anthropogenic
79 sources in China have been estimated to be in the range of 500-700 tons/yr,
80 accounting for 25-30% of the total global anthropogenic mercury emissions (Streets et
81 al., 2005; Wu et al., 2006). Studies of atmospheric mercury in China are therefore
82 critical to the understanding of mercury cycling at both regional and global scales.
83 Long-term observation of atmospheric mercury has been conducted in numerous
84 urban and remote areas in China. TGM concentrations in urban and industrial areas
85 were observed to be in the range of 2.7–35 ng m⁻³, higher than the values reported for
86 North America and Europe, and for the adjacent Asian countries such as Korea and
87 Japan (Weigelt et al., 2013; Fang et al., 2009; Marumoto et al., 2015). TGM and PBM
88 concentrations in remote areas of China were also found to be higher than those
89 observed in North America and Europe (Fu et al., 2008a; Fu et al., 2008b; Fu et al.,
90 2012; Liu et al., 2010).

91 In recent years, haze pollution has become a major concern in China due to its
92 impacts on visibility, air quality, and climate. It is well known that haze formation is
93 mainly dependent on the atmospheric relative humidity (RH) and the concentration of
94 airborne particles (Chen et al., 2003; Sun et al., 2013). Most studies on haze have
95 focused on the measurements of airborne particulate matter; few examined the
96 influence of haze on the chemistry of atmospheric mercury, especially PBM. Here we

97 report a one-year real-time measurement of speciated atmospheric mercury in Hefei,
98 an inland city of China, which experiences frequent haze events. The comparison of
99 atmospheric mercury in haze days and non-haze days allows us to examine the
100 formation and deposition mechanisms of mercury, as well as their temporal variations.

101

102 **2. Methods**

103 **2.1 Study site**

104 Hefei (31°52' N, 117°17' E) is the capital of Anhui Province in east central China,
105 between the Changjiang (Yangtze River) and the Huaihe (Huai River). The region is
106 under a humid subtropical climate with four distinct seasons: summer (June-August),
107 autumn (September-November), winter (December-February), and spring
108 (March-May). The prevailing wind is southeasterly in summer and northwesterly in
109 winter. Like many Chinese cities, Hefei has experienced rapid growth in the past 20
110 years, with a present-day total permanent population of about 7.7 million. The city has
111 also been witnessing an increasing frequency in haze pollution, especially in winter
112 months.

113 The monitoring site was located on the Science Island, a small peninsula on the
114 Dongpu Reservoir in the northwestern outskirts of Hefei (Fig. 1). The sampling and
115 analytical instruments were installed 1.5 m above the rooftop (~ 20 m above the
116 ground) of the main building of Anhui Institute of Optics and Fine Mechanics. Further
117 information about the monitoring site can be found in a previous study (Hu et al.,
118 2014). We chose this area as the monitoring site because it is not adjacent to any
119 direct pollution sources such as power plants, iron and steel works.

120

121 2.2 Measurements of speciated atmospheric mercury

122 From July 2013 to June 2014, simultaneous measurements of GEM, GOM and
123 PBM were carried out by an automated TekranTM mercury speciation system. The
124 system consisted of a Model 2537B mercury analyzer combined with a Model 1130
125 GOM unit and a Model 1135 PBM unit. The system was configured to measure GEM
126 every 5 min., and GOM and PBM every 2 h.

127 The details about the Tekran-based mercury speciation system can be found in
128 Landis et al. (2002). In general, the automated measurement process can be
129 summarized as sample collection, thermal desorption and determination. During the
130 collection period, ambient air is drawn to the system at a typical flow rate of 10 L/min.
131 GOM and PBM in the air are captured by a KCl-coated quartz annular denuder in the
132 1130 unit and a quartz filter in the 1135 unit, respectively, whereas GEM would pass
133 through the denuder and filter and be quantified on the 2537B analyzer by cold-vapor
134 atomic fluorescence spectroscopy (CVAFS). After an hour of sampling, the 1135
135 quartz filter and the 1130 denuder would be switched to the thermal decomposition
136 mode at 800°C and 500°C, respectively, with the resulting Hg⁰ quantified by the 2537B
137 unit in the next hour, while the 1135 and 1130 components are flushed with
138 zero-mercury gas for the next sampling.

139 The instrument maintenance followed typical protocols used in similar
140 studies(Landis et al., 2002;Hu et al., 2014). The quartz annular denuder was recoated
141 every two weeks, the quartz filter was replaced once a month, and the Teflon filter (pore
142 size 0.2 μm) in the sample inlet was changed every two weeks. Automated
143 recalibration of the Tekran 2537B was performed every 25h using an internal mercury
144 permeation source.No calibration standards were available for GOM and PBM, but

145 the 1σ precision for GOM and PBM was about 15 % (Landis et al., 2002). The
146 detection limit in ambient air is about 0.5 ng m^{-3} for GEM (or TGM) at a resolution of
147 5 min, and 1 pg m^{-3} for GOM and PBM at a resolution of 2h (Gustin et al., 2015).
148 Although the Tekran-based mercury speciation technique has been widely used
149 around the world, recent studies have shown that the technique does not efficiently
150 collect all GOM and thus may substantially underestimate the concentration of
151 reactive mercury (Huang et al., 2013;Gustin et al., 2013). Therefore, the GOM values
152 reported in this study should be considered as the lower limit of GOM in the air(Wang
153 et al., 2014).

154

155 **2.3 Ancillary data**

156 Standard meteorological measurements including air temperature, pressure, RH,
157 wind direction and speed were observed with a 5-min resolution. CO was measured by
158 an automated infrared carbon monoxide analyzer (Model EC9830T, Ecotech Inc.,
159 Australia), with a detection limit of 40 ppbv. O_3 was measured every 5 min by an ozone
160 analyzer (Model EC9810B, Ecotech Inc., Australia); its detection limit and accuracy
161 were 0.5 ppbv and 1 ppbv, respectively. NO_2 was measured by a Multi axis differential
162 optical absorption spectroscopy (MAX-DOAS) instrument. The collected spectra
163 were analyzed using the QDOAS spectral fitting software suite developed at
164 BIRA-IASB (<http://uv-vis.aeronomie.be/software/QDOAS/>). $\text{PM}_{2.5}$ (particulate matter
165 of less than $2.5 \mu\text{m}$ in diameter) data were collected from the China air quality online
166 analysis platform (<http://www.aqistudy.cn/historydata/index.php>).

167

168 **2.4 Potential Sources Contribution Function (PSCF) analysis**

169 To identify possible influence of long-range transport on the distribution of
170 atmospheric mercury in Hefei, we calculated backward trajectories of air masses
171 using the HYSPLIT (Hybrid Single-particle Lagrangian Integrated Trajectory) model
172 with the Global Data Assimilation System (GDAS 1°) developed by the National
173 Oceanic and Atmospheric Administration (NOAA)
174 (<http://www.ready.noaa.gov>)(Draxler and Hess, 1998). Considering atmospheric
175 pollutants are mainly concentrated in low altitudes during heavy pollution days, the
176 trajectory arrival heights were set at 500m to represent the boundary layer where
177 atmospheric pollutants were well mixed. In this study, 3-day back-trajectories were
178 generated hourly by the TrajStat software, which employs HYSPLIT for trajectory
179 calculation (Wang et al., 2009).

180 The contributions of other source regions to the atmospheric mercury at Hefei
181 were identified by the Potential Sources Contribution Function (PSCF) analysis with
182 TrajStat. PSCF analysis has been shown to be useful in spatially identifying emission
183 sources for pollutants with a long lifetime such as elemental mercury and CO (Xu and
184 Akhtar, 2010). The study domain is divided into grid cells, and the PSCF value for
185 each cell was calculated by counting the trajectory segment endpoints that terminate
186 within the cell. The PSCF value for the ij_{th} cell is defined as:

$$187 \text{PSCF}_{ij} = \frac{M_{ij}}{N_{ij}} W_{ij} \quad (2)$$

188 where N_{ij} is the number of endpoints that fall in the ij_{th} cell, and M_{ij} is the number of
189 endpoints in the the same cell that has a GEM concentration higher than an arbitrarily
190 set criterion; in this study the criterion was set to be 4 ng m^{-3} which is the mean GEM
191 concentration during the entire study period. W_{ij} is an arbitrary weight function
192 introduced to reduce the effect of small values of N_{ij} to better reflect the uncertainty in

193 the values for these cells (Polissar et al., 2001).The weight function reduces the PSCF
194 values when the total number of endpoints in a particular cell is less than 3 times the
195 average value of the end points per cell:

$$196 \quad W_{ij} = \begin{cases} 1.0 & N_{ij} \geq 3N_{ave} \\ 0.70 & 3N_{ave} > N_{ij} \geq 1.5N_{ave} \\ 0.40 & 1.5N_{ave} > N_{ij} \geq N_{ave} \\ 0.20 & N_{ave} > N_{ij} \end{cases} \quad (3)$$

197

198 3. Results

199 We intended to continuously monitor speciated atmospheric mercury
200 concentrations over the course of a year; however, interruptions were inevitable due to
201 instrument maintenance, which resulted in loss of data for the following four periods:
202 (1) 25 September to 9 October 2013; (2) 5 – 14 November 2013; (3) 9 – 25 February
203 2014; and (4) 1 – 14 April 2014. The rest of the data were grouped into haze days and
204 non-haze days according to the China Meteorological Administration’s haze standard
205 (QX/T 113-2010). Haze days refer to the days when the atmospheric visibility < 10
206 km and RH < 80% (Duan et al., 2016), and non-haze days refer to clear days with the
207 atmospheric visibility > 10 km. The visibility and RH information were collected
208 from the weather history data at the Luogang Airport of Hefei
209 (<http://www.wunderground.com/>).Throughout the study period of almost a year, a
210 total of 56 days were identified to be haze days, and 253 days to be non-haze days. All
211 the times reported herein are local time (UTC + 8 h).

212

213 3.1 Overall characteristics of speciated atmospheric mercury

214 The time series of GEM, GOM and PBM concentrations at the study site
215 throughout the study period are shown in Fig. 2; their frequency distributions are

216 shown in Fig. S1 in the supporting information (SI). The mean (\pm standard deviation)
217 GEM, GOM and PBM concentrations during the entire study period were 4.07 ± 1.91
218 ng m^{-3} , $3.67 \pm 5.11 \text{ pg m}^{-3}$, and $30.0 \pm 100.3 \text{ pg m}^{-3}$, respectively (Table 1). GEM
219 concentrations in different seasons did not differ much, with the highest in autumn
220 ($4.51 \pm 2.10 \text{ ng m}^{-3}$) and the lowest in spring ($3.89 \pm 1.79 \text{ ng m}^{-3}$). GOM concentrations
221 varied greatly, with much higher concentrations in autumn and the lowest in winter. A
222 similar seasonal variation in the GOM concentration was observed at a remote site in
223 Mt. Gongga of southwest China (Fu et al., 2008b). PBM showed the highest degree of
224 seasonal variability; its concentration decreased in the following order: autumn \approx
225 winter > spring > summer. The mean PBM concentrations in autumn and winter were
226 about 20 times that in summer, similar to the findings from many previous studies in
227 China (Zhang et al., 2013; Fu et al., 2011; Fu et al., 2008b; Fang et al., 2001).

228 Comparisons of speciated atmospheric mercury concentrations with other urban
229 and rural areas in China and a few other countries are shown in Table 2. The mean
230 GEM concentration at Hefei is slightly higher than those reported from many remote
231 areas in China (Fu et al., 2008a; Fu et al., 2008b; Fu et al., 2012; Wan et al., 2009a; Wan
232 et al., 2009b; Zhang et al., 2015a), but is much lower than those from urban areas of
233 heavily industrial cities such as Guiyang and Changchun where large point sources of
234 mercury exist (e.g., non-ferrous metal smelting, coal-fired power plants, and
235 residential coal burning) (Feng et al., 2004; Fu et al., 2011; Fang et al., 2004).
236 Although Hefei is geographically close to Shanghai, a mega urban centre in China, it
237 is interesting to note that the TGM concentration of Shanghai is much lower than that
238 of Hefei. This may be due to the fact that Shanghai is a coastal city that is influenced
239 more by cleaner marine air masses (Friedli et al., 2011). Table 2 also shows that the

240 average concentration of GEM in Hefei doubles the typical values reported from the
241 urban and rural areas in Europe and North America (Liu et al., 2010;Li et al.,
242 2008;Brooks et al., 2010).

243

244 **3.2Speciated atmospheric mercury on non-haze days**

245 As shown in Table 1 and Fig.S1(in blue), the mean concentration of GEM on
246 non-haze days was $3.95 \pm 1.93 \text{ ng m}^{-3}$. Its distribution was characterized by large
247 variations ranging from 0.2 to 23.8 ng m^{-3} , although more than half of the GEM
248 values were in the narrow range 2–4 ng m^{-3} . The mean concentration of GOM on
249 non-haze days was $2.49 \pm 2.41 \text{ pg m}^{-3}$ with a range of 0.5–33.5 pg m^{-3} , although
250 most of the values were in the range of 1-4 pg m^{-3} . High concentrations of GOM
251 (exceeding 10 pg m^{-3}) only accounted for 1.4% of the total data points. The mean
252 GOM concentration at Hefei on these non-haze days is much smaller than those
253 reported from other study sites in China (Table 2), but is comparable to the values
254 observed from many European and North American sites (Peterson et al., 2012;Cheng
255 et al., 2014;Ren et al., 2016). The mean PBM concentration on non-haze days was
256 $23.3 \pm 90.8 \text{ pg m}^{-3}$ with an exceptionally large range of 0.5-1827 pg m^{-3} ; high PBM
257 concentrations (i.e., $> 50 \text{ pg m}^{-3}$) accounted for 6.4% of the total data points. The
258 PBM concentration under the non-haze condition in Hefei is generally similar to
259 values reported from remote areas in western China, such as Mt.Gongga,
260 Mt.Waliguan and Shangri-La.

261 Diurnal variations of GEM, PBM and GOM concentrations on non-haze days are
262 shown in Fig. 3. Both GEM and PBM concentrations exhibited similar patterns with
263 elevated concentrations at night. The GOM concentration remained relatively constant

264 at night, but increased rapidly just before sunrise and reached its peak value at ~10:00,
265 followed by a synchronous decline with GEM through afternoon (10:00-18:00).

266

267 3.3 Speciated atmospheric mercury on haze days

268 Haze pollution mainly occurred in December and January at our monitoring site.
269 The four major haze pollution periods were identified in grey in Fig.2. The mean
270 concentrations of GEM, GOM and PBM on these haze days were $4.74 \pm 1.62 \text{ ng m}^{-3}$,
271 $4.32 \pm 8.36 \text{ pg m}^{-3}$ and $60.2 \pm 131.4 \text{ pg m}^{-3}$, respectively (Table 1). The frequency
272 distributions of GEM, GOM and PBM on the haze days are shown in Fig.S1 (in
273 gray).GEM, GOM and PBM concentrations show significant differences between
274 haze and non haze days ($p < 0.001$, t-test) (Fig. S2). On average, the concentration of
275 GEM on haze days was 1.2 times that on non-haze days. Similarly, the concentration
276 of GOM on haze days was about 1-1.7 times those on non-haze days. The largest
277 impact of haze pollution is however on PBM, with the mean PBM concentration on
278 haze days about 2.5 times that of non-haze days. High concentrations of GOM
279 (exceeding 10 pg m^{-3}) and PBM concentrations (exceeding 50 pg m^{-3}) were also more
280 frequently observed on haze days, accounting for 5.9% and 25%, respectively, of the
281 total haze days.

282 As shown in Fig. 3, on haze days the GEM concentration was higher at night and
283 lower during daytime. PBM typically peaked just before sunrise, with the lowest
284 values occurred in the afternoon (14:00-16:00). The opposite pattern was observed for
285 GOM, which showed higher concentrations during daytime than at night. Although on
286 both haze and non-haze days GOM showed rapid increase just before sunrise, they
287 exhibited different trends during daytime from 10:00 to 18:00. On haze days, GOM

288 peaked in the afternoon when GEM was the lowest; the duration of the afternoon
289 GOM peak was also longer on haze days.

290

291 4. Discussion

292 4.1 Influence of atmospheric mercury emission source

293 With the year-long data seasonal mercury emission sources could be inferred from
294 the PSCF analysis. Fig.4a showed the overall spatial contribution of mercury emission
295 sources in China. As Hefei is located in east-central China, its atmospheric mercury
296 concentration could be affected by both north and south emission sources, including
297 those from the North China Plain (especially Shandong Province) and the neighboring
298 provinces of Henan, Jiangsu, Jiangxi and Hubei. The total mercury emissions from
299 Henan and Shandong provinces were estimated to be over 50 and 45 tons in 2010,
300 respectively, making them two largest Hg emitters in China (Zhang et al., 2015b).
301 Long-range transport could also impact seasonal variations of atmospheric mercury in
302 Hefei. As shown in Fig.4, in spring, the major contributors of atmospheric mercury to
303 Hefei were from the southwestern region including the local area and Jiangxi and
304 Hunan Provinces. In summer, the main contributors were from north of Anhui, as well
305 as Henan and Jiangxi Provinces, and even from the Pearl River Delta region in the far
306 south. Since the number of haze days accounts only for 5.6% of the total days in spring
307 and summer, we did not provide haze and non-haze PSCF results for spring and
308 summer seasons. As autumn and winter are the prevalent seasons for haze pollution,
309 one PSCF result for haze days and another for non-haze days are shown for autumn
310 and winter, respectively. The statistically significant difference ($p < 0.001$) in the GEM
311 concentration between non-haze days and haze days suggests that haze pollution

312 could directly affect the concentration of elemental mercury. As shown in Figs. 4d and
313 4f, higher GEM concentration was mainly influenced by local emission sources
314 on haze days. On non-haze days, the most important mercury sources to the monitoring
315 site were not only the local emission sources, but also those from the neighboring
316 region of Shandong, Henan and Jiangxi Provinces (see Figs. 4e and 4g). Therefore, the
317 increase in the GEM concentration on haze days was mainly caused by local
318 emissions.

319 GEM and CO often share similar anthropogenic emission sources, such as
320 industrial coal combustion, domestic coal combustion, iron and steel production and
321 cement production (Wu et al., 2006; Wang et al., 2005). However, they also have their
322 distinct emission sources. For instance, power plants and nonferrous metal smelters
323 emit mercury but hardly any CO; while automobiles contribute greatly to CO
324 emission, they are not a major emitter for mercury. The correlation coefficients and
325 slopes between GEM and CO concentrations during mercury pollution events are
326 shown in Table 3. These mercury pollution episodes were defined as when the hourly
327 average GEM concentration was higher than seasonal average GEM concentration for
328 at least 10 consecutive hours. These episodes could be classified into long-range
329 transport episodes or local episode by using the coefficients of determination (R^2) of
330 linear regression between Hg and CO: a significant positive correlation indicates
331 long-range transport episodes and a poor correlation signals local episodes (Jaffe et al.,
332 2005; Weiss-Penzias et al., 2006; Kim et al., 2009). Using this approach, we identified
333 three local episodes (events: 1-3) characterized by poor correlations between GEM
334 and CO concentrations (R^2 : 0.23-0.29), and four long-range transport episodes (events:
335 4-7) characterized by positive correlations between GEM and CO concentrations

336 ($R^2:0.51-0.79$). These local episodes tend to occur in autumn and winter. The slope of
337 the trend line represents the Hg/CO ratio, which could aid in the identification of
338 specific emission sources. Emissions from power plants typically have a higher
339 Hg/CO ratio (Wu et al., 2006), whereas residential coal and biomass burning
340 combustion have a lower Hg/CO ratio ($0.0013-0.0046 \text{ ng m}^{-3} \text{ ppbv}^{-1}$) due to incomplete
341 combustion (Weiss-Penzias et al., 2007). The Hg/CO ratio for vehicles is close to zero
342 (Zhang et al., 2013). The Hg/CO ratios during the pollution episodes in our study
343 ranged from 0.0001 to $0.005 \text{ ng m}^{-3} \text{ ppbv}^{-1}$ suggesting mercury emission in autumn
344 and winter in Hefei could be related to local incomplete combustion sources, such as
345 residential coal and biomass burning.

346

347 **4.2 Impacts of meteorological factors for atmospheric mercury on haze days**

348 Meteorological conditions, especially wind direction and speed, could also
349 impact atmospheric mercury on haze days. The wind rose for the monitoring site
350 during the study period is shown in Fig.5. Easterly and southeasterly winds
351 represented the prevailing wind directions at the study site. A wind rose diagram of
352 GEM concentrations above the 90th percentile value is shown in Fig.5B. We found
353 that 67% of the high GEM concentrations occurred at low wind speed (below 1.5 m
354 s^{-1}); however, wind speed below 1.5 m s^{-1} accounted for only 1.7% of the study
355 duration. High GOM and PBM concentrations appear not to be related to high wind
356 speed (wind speed: $3-5 \text{ m s}^{-1}$); only 1.4% and 2.6% of the high GOM and PBM
357 concentrations were observed under high wind-speed conditions, respectively (Figs.
358 5C and 5D). The occurrence of high atmospheric mercury levels under low wind
359 speed conditions is to be expected, as this slow wind speed condition is not

360 conductive to the spread and mixing of mercury especially on haze days, and thus
361 favors mercury accumulation in the air. This further supports that atmospheric
362 mercury on haze days is mainly due to by local emissions.

363 Both GEM and PBM concentrations exhibited diurnal variations with elevated
364 concentrations at night or in early morning, regardless of the presence of haze. Such a
365 diurnal variation could be related to changes in the height of the urban boundary layer.
366 The diurnal trend of the boundary layer height (BLH or BLK) in Hefei is shown in
367 Fig. S3, which is typically low in the morning and night, and high during the daytime
368 on both non-haze and haze days. Such diurnal changes in BLH in Hefei and nearby
369 cities have also been observed in previous studies(Yuan et al., 2005;Mao et al., 2006).

370 The maximum PBM concentration (observed at 6:00) was more than 4 times higher
371 than the minimum value (observed at 16:00) on both non-haze and haze days, with a 76%
372 decrease from early morning to the afternoon. However, the reductions of PBM as a
373 result of deposition during haze days was 62.7 pg m^{-3} , which was about 2.4 times that in
374 non-haze days, suggesting that haze pollution could increase the removal of PBM.
375 Although PBM is not the major form of atmospheric mercury, it is crucial in
376 atmospheric mercury transport and removal processes due to its short atmospheric
377 lifetime. As shown in Fig. 6, the highest PBM and $\text{PM}_{2.5}$ concentrations were observed
378 in January, which is most likely due to a shallower boundary layer in January than in
379 other months. The co-variation in February is weaker, possibly due to the loss of PBM
380 data because of instrument maintenance. The PBM concentration co-varied with the
381 $\text{PM}_{2.5}$ concentration, especially in January when all the four PBM peak events were
382 associated with increased $\text{PM}_{2.5}$ concentrations (Fig.6c). These results suggest
383 that $\text{PM}_{2.5}$ may play an important role in the formation of PBM. Thus, elevated PBM

384 concentrations in autumn and winter might be due to the combination of poor mixing
385 conditions and higher PM concentrations.

386

387 **4.3 Enhancement in GOM and the potential GEM oxidation mechanism**

388 Diurnal variations of GEM, GOM, O₃ and CO concentrations on non-haze and
389 haze days are shown in Fig. 7. The weak correlation ($r=0.164$) between GOM and CO
390 suggests that the CO-producing, primary emission is not a major source of GOM in the
391 air. This is clearly shown in Fig. 7 (haze days), where the peak value of GOM
392 coincided with the lowest value of CO. As mentioned earlier, on both non-haze and
393 haze days, GOM concentrations remained relatively constant during night, but
394 increased rapidly prior to sunrise. However, GOM showed different trends during
395 daytime (10:00-18:00) between non-haze and haze days. On non-haze days, a
396 synchronous decline was found between GEM and GOM in the afternoon, but
397 opposite trend was observed between GEM and GOM on haze days. This difference
398 indicates that different GOM formation mechanisms might be at work on non-haze
399 and haze days. Two processes can affect the GOM concentrations in the boundary
400 layer air. The first is the change in the atmospheric boundary layer height, which
401 could change the transport of GOM-enriched free tropospheric air. Secondly, in situ
402 photochemical oxidation of GEM would increase the GOM concentration during
403 daytime. Various atmospheric oxidants are capable of oxidizing GEM to GOM,
404 including halogen radicals, ozone, hydroxyl radicals (OH), among others (Holmes et
405 al., 2010; Wang et al., 2014).

406 It is well established that FT contains higher GOM concentrations than in the
407 boundary layer (e.g., (Murphy et al., 2006; Lyman and Jaffe, 2012; Timonen et al.,

408 2013;Brooks et al., 2014;Shah et al., 2016)).On both non-haze and haze days,it is thus
409 possible that the higher GOM concentrations observed prior to sunrise are due to
410 enhanced admixing of the free tropospheric air as the boundary layer increases in the
411 morning (see Fig. S3). On non-haze days, a synchronous decline in GEM and GOM
412 throughout afternoon (10:00-18:00) might be related to the higher atmospheric
413 boundary layer height during this period.However, on haze days, opposite variation
414 was observed between GEM and GOM from 10:00-18:00, along with the elevated
415 boundary layer height. This suggests that other than FT transport, photochemical
416 oxidation of GEM might also play an important role in the enhancements of GOM. To
417 determine the relative importance of FT transport and in situ photochemical oxidation,
418 we examined the relationship between GOM and the changes in the height of the
419 atmospheric boundary layer and the odd oxygen ($O_x = O_3 + NO_2$) concentrations. We
420 used O_x because it is a more conserved tracer of the extent of photochemical
421 processes in the urban atmosphere (Herndon et al., 2008;Wood et al., 2010), as O_3
422 reacts with NO emitted from automobiles to form NO_2 . Example results are shown in
423 Fig. 8 for 20th November, 2013 (haze day). As can be seen from the figure, GEM and
424 GOM showed opposite trends in the afternoon (12:00-16:00), along with higher
425 O_x concentrations during this period.The height of the atmospheric boundary layer
426 changed very little (less than 0.1 km) over the same period (see Fig. S4). This simple
427 comparison suggests that the transport of FT GOM might be limited and that at least
428 some of the GOM were formed from in situ oxidation of GEM. Note that in our
429 studies we could only calculate daytime O_x concentrations, because NO_2
430 concentrations from MAX-DOAS were only available during daytime.

431 We further investigated the mechanism of GEM oxidation to GOM. Ozone itself
432 is not an efficient oxidant for GEM oxidation due to the low reaction rate (Hall,
433 1995; Holmes et al., 2010). Instead, halogen radicals (especially bromine atoms) and
434 OH radicals, are believed to be the primary oxidants for GEM in the global
435 troposphere (Holmes et al., 2010). Unfortunately, we did not measure halogen radicals
436 in this study. OH radicals are known to be present in the early morning urban
437 boundary layer, primarily from the photolysis of HONO, which accumulates
438 during night (Kleffmann et al., 2005). Therefore, here we consider the oxidation of
439 GEM by OH radicals only. The formation of HgOH as an intermediate product of the
440 $\text{Hg}^0(\text{g}) + \text{OH}$ oxidation reactions has been proposed by Sommar et al., 2001, although
441 HgOH is highly unstable and could decompose back rapidly to Hg^0 and OH (Sommar
442 et al., 2001; Goodsite et al., 2004). It has been proposed that the presence of other
443 gases X (X = NO_2 , HO_2 , RO, RO_2 , or NO) could assist the formation of Hg(II) by
444 forming X-HgOH, which outcompetes the decomposition of HgOH (Calvert and
445 Lindberg, 2005; Dibble et al., 2012; Wang et al., 2014). As an example, we calculated
446 the transformation between GEM and GOM under the influence of NO_2 , using the
447 reactions and rate constants shown in Table S1. As shown in Fig. S5, the production
448 rate of NO_2HgOH , $d[\text{NO}_2\text{HgOH}]/dt$, increased almost linearly with increasing NO_2
449 under low NO_2 concentrations, and eventually reached a steady state when the NO_2
450 concentration is high enough.

451 Based on the production rate of NO_2HgOH , we can estimate the production of
452 NO_2HgOH during the 1hr sampling period when GOM was captured by the
453 KCl-coated denuder in the Tekran 1130 unit. The production of NO_2HgOH and
454 $d[\text{NO}_2\text{HgOH}]/dt$ corresponding to different NO_2 concentrations is shown in Table 4.

455 With the increase of the NO₂ concentration, the contribution of the NO₂HgOH
456 production to GOM will increase. If the NO₂ concentration is within 100 ppbv (from
457 0 to 100 ppbv), the production of NO₂HgOH would be in the range of 0.058-4.81 pg
458 m⁻³ during the 1h sampling period. As illustrated in Table 4, the level of NO₂ observed
459 in our study is high enough to account for the increase in the observed GOM
460 production. Our results thus support a recent study in the tropical equatorial Pacific
461 (Wang et al., 2014) that NO₂ aggregation with HgOH provides a possible mechanism
462 for enhanced production of GOM. If that is true, NO₂ would be expected to play an
463 even more important role in the urban air because of its higher concentration. More
464 laboratory and modeling studies on the mercury oxidation mechanism in the presence
465 of NO₂ and other gases are thus warranted.

466

467 **5. Summary**

468 Continuous measurements of speciated atmospheric mercury were conducted at
469 Hefei, a mid-latitude inland city in central China, from July 2013 to June 2014.
470 Measurements of other trace gases (e.g. CO, O₃, NO₂) and meteorological parameters
471 were employed to better understand the sources and oxidation pathways of
472 atmospheric mercury. The mean GEM, GOM and PBM concentrations during haze
473 days were 4.74 ± 1.62 ng m⁻³, 4.32 ± 8.36 pg m⁻³ and 60.2 ± 131.4 pg m⁻³, respectively.
474 Potential source contribution function (PSCF) analysis suggested that the local
475 mercury emission rather than long-range transport is the most important contributor of
476 atmospheric mercury pollution on haze days at our monitoring site. The low GEM/CO
477 ratio in Hefei could be indicative of local incomplete combustion sources such as
478 residential coal and biomass burning. Haze pollution has a more profound impact on

479 PBM than on GEM and GOM. PBM showed a remarkable seasonal pattern, with
480 higher concentrations in cold seasons and lower in warm seasons. Elevated PBM
481 concentrations might be due to both the high loadings of particle matter and poorer
482 mixing conditions on haze days especially in cold months. Both GEM and PBM
483 concentrations exhibited great variations with elevated concentration during night. The
484 diurnal variations of GEM and PBM might be related to the boundary layer depth; a
485 lower boundary layer depth in the morning and night could elevate the mercury
486 concentration.

487 Different from the diurnal variations of GEM and PBM, GOM concentration
488 remained relatively constant at night, and then increased rapidly prior to the sunrise.
489 The enhancement of GOM during daytime could be due to both the transport of
490 GOM-enriched free tropospheric air to the boundary layer and in situ oxidation of
491 GEM in the boundary layer. Simple photochemical modeling supports the occurrence
492 of daytime oxidation of GEM to GOM. Based on HgOH as an intermediate product,
493 our calculations suggest that NO₂ aggregation with HgOH is a potential mechanism
494 for the enhanced production of GOM in the inland urban air.

495

496

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705

706

707 **Table 1. Summary of GEM, GOM and PBM concentrations measured in Hefei**
708 **from July 2013 to June 2014.**
709

	GEM (ng m ⁻³)			GOM (pg m ⁻³)			PBM (pg m ⁻³)		
	Mean ± σ	Range	N	Mean ± σ	Range	N	Mean ± σ	Range	N
Spring	3.89 ± 1.79	0.2-21.3	7890	4.49 ± 4.22	0.5-69.8	526	8.34 ± 8.97	1.6-130.1	542
Summer	4.08 ± 1.99	0.3-22.9	6050	3.66 ± 4.39	0.5-45.2	511	3.61 ± 4.38	0.5-41.9	570
Autumn	4.51 ± 2.10	0.4-23.8	3632	5.65 ± 8.93	0.5-78.9	274	59.9 ± 153.5	0.5-1615	339
Winter	4.05 ± 1.81	0.9-12.2	6381	2.59 ± 2.58	0.5-9.5	541	56.1 ± 134.9	0.5-1827	639
Total	4.07 ± 1.91	0.2-23.8	23953	3.67 ± 5.11	0.5-78.9	1852	30.02 ± 100.3	0.5-1827	2090
Non-haze	3.95 ± 1.93	0.2-23.8	20345	2.49 ± 2.41	0.5-33.5	1508	23.3 ± 90.76	0.5-1827	1708
Haze	4.74 ± 1.62	2.1-16.5	3608	4.32 ± 8.36	0.5-78.9	344	60.2 ± 131.4	1.6-1615	382

710

711

712 **Table 2. Speciated atmospheric mercury concentrations in Hefei and other urban**
713 **and rural areas.**
714

Location	Classification	Time	TGM (ng m ⁻³)	GEM (ng m ⁻³)	GOM (pg m ⁻³)	PBM (pg m ⁻³)	Reference
Hefei	Suburb	Jul 2013-Jun 2014	4.1	4.07	3.67	30	This study
Hefei	Suburb	Feb-May 2009	2.53	-	-	-	Hu et al. (2014)
Beijing	Rural	Dec2008-Nov2009	3.23	3.22	10.1	98.2	Zhang et al. (2013)
Shanghai	Urban	Aug-Sep 2009	2.7	-	-	-	Friedli et al. (2011)
Nanjing	Urban	Jan-Dec 2011	7.9	-	-	-	Zhu et al. (2012)
Guiyang	Urban	Nov 2001-Nov 2002	8.4	-	-	-	Fenget al. (2004)
Guiyang	Urban	Aug-Dec 2009	-	9.72	35.7	368	Fu et al. (2011)
Changchun	Urban	Jul 1999-Jan 2000	18.4	-	-	276	Fang et al. (2004)
Changchun	Suburb	Jul 1999-Jan 2000	11.7	-	-	109	Fang et al. (2004)
Mt.Changbai	Remote	Aug2005-Jul 2006	3.58	-	65	77	Wan et al. (2009a,b)
Mt.Gongga	Remote	May 2005-July 2006	3.98	-	6.2	30.7	Fu et al. (2008a,b)
Mt.Waliguan	Remote	Sep 2007-Aug 2008	1.98	-	7.4	19.4	Fu et al. (2012a)
Mt.Leigong	Remote	May2008-May 2009	2.8	-	-	-	Fu et al. (2010)
Shangri-La	Remote	Nov 2009-Nov 2010	2.55	-	8.22	38.82	Zhang (2015)
Detroit, USA	Urban	Jan-Dec 2004	-	2.5	15.5	18.1	Liu et al. (2010)
Dexter, USA	Rural	Jan-Dec 2004	-	1.6	3.8	6.1	Liu et al. (2010)
Houston, USA	Urban	Aug-Oct 2006	-	1.66	6.9	2.5	Brooks et al. (2010)
Florida, USA	Urban	Jul 2009-Jul 2010	-	1.3	3	2	Peterson et al. (2012)
Maryland, USA	suburb	2007-2015	-	1.41	4.6	8.6	Ren et al. (2016)
Göteborg,Sweden	Urban	Feb-Mar 2005	-	1.96	2.53	12.5	Li et al. (2008)
Nova Scotia,Canada	Urban	Jan 2010- Dec 2011	-	1.67	2.07	2.32	Cheng et al.(2014)
Northern Hemisphere background value					1.5-1.7		Lindberg et al. (2007)

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716 **Table 3. Coefficients of determination and slopes between GEM and CO**
 717 **concentrations during atmospheric mercury pollution episodes (* $p < 0.01$).**

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Event	Start Time (UTC + 8 hr)	End Time (UTC + 8 hr)	Duration (h)	GEM (ng m ⁻³)	CO (ppbv)	GEM/CO (slope, ng m ⁻³ ppbv ⁻¹)	R ²
1	2013/11/21 03:00	2013/11/22 02:00	23	8.37±2.42	4481.6±717.3	0.0018	0.29*
2	2013/12/07 04:00	2013/12/09 04:00	48	9.21±1.16	5943.8±1394.1	0.0004	0.23*
3	2014/01/17 22:00	2014/01/19 13:00	39	5.80±0.83	5746.3±1626.9	0.0003	0.28*
4	2014/01/25 02:00	2014/01/25 22:00	20	6.03±0.50	8797.9±2244.3	0.0002	0.59*
5	2014/03/16 05:00	2014/03/16 20:00	15	4.46±0.47	2261.7±440.2	0.0010	0.79*
6	2014/03/17 06:00	2014/03/18 12:00	30	8.85±2.46	2697.1±590.3	0.0030	0.51*
7	2014/05/21 00:00	2014/05/21 11:00	11	5.74±0.94	3676.7±1690.0	0.0050	0.79*

719 Notes: these episodes were identified when the hourly average GEM concentration
 720 was higher than the seasonal average GEM concentration for more than 10
 721 consecutive hours

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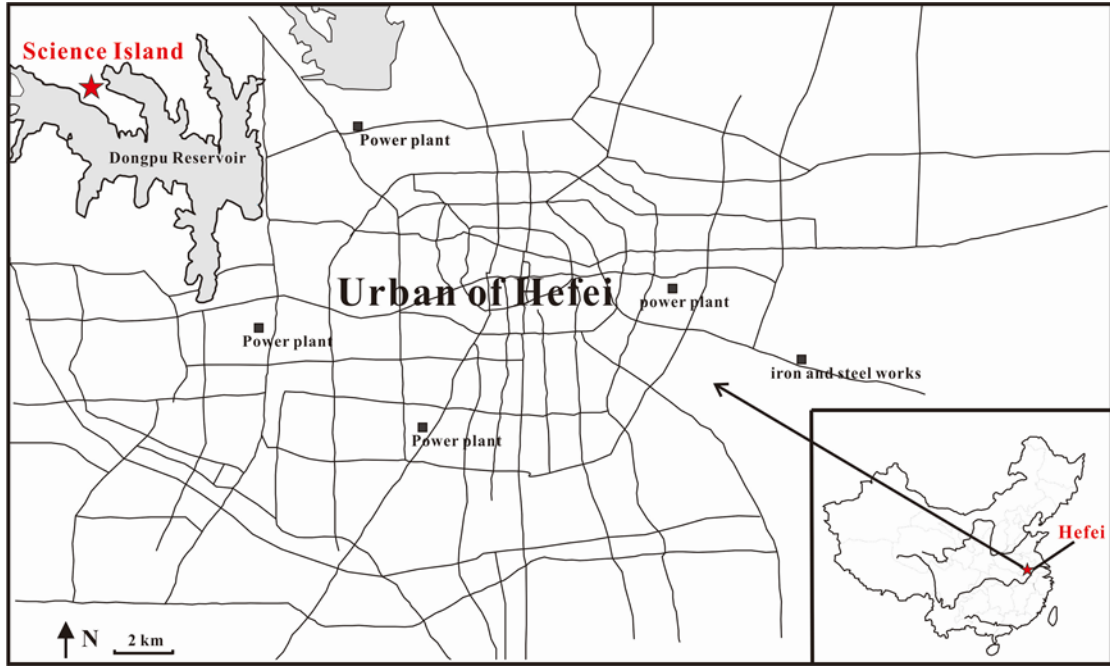
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724 **Table 4.**The production of NO₂HgOH and d[NO₂HgOH]/dtat different NO₂
 725 **concentrations**

NO ₂ (ppbv)	10	20	30	40	50	60	70	80	90	100
d(NO ₂ HgOH)/dt (molecule cm ⁻³ s ⁻¹)	0.36	0.71	1.04	1.37	1.68	1.99	2.28	2.56	2.83	3.10
NO ₂ HgOH (pg m ⁻³ , 1hr)	0.56	1.10	1.63	2.13	2.61	3.08	3.54	3.97	4.40	4.81

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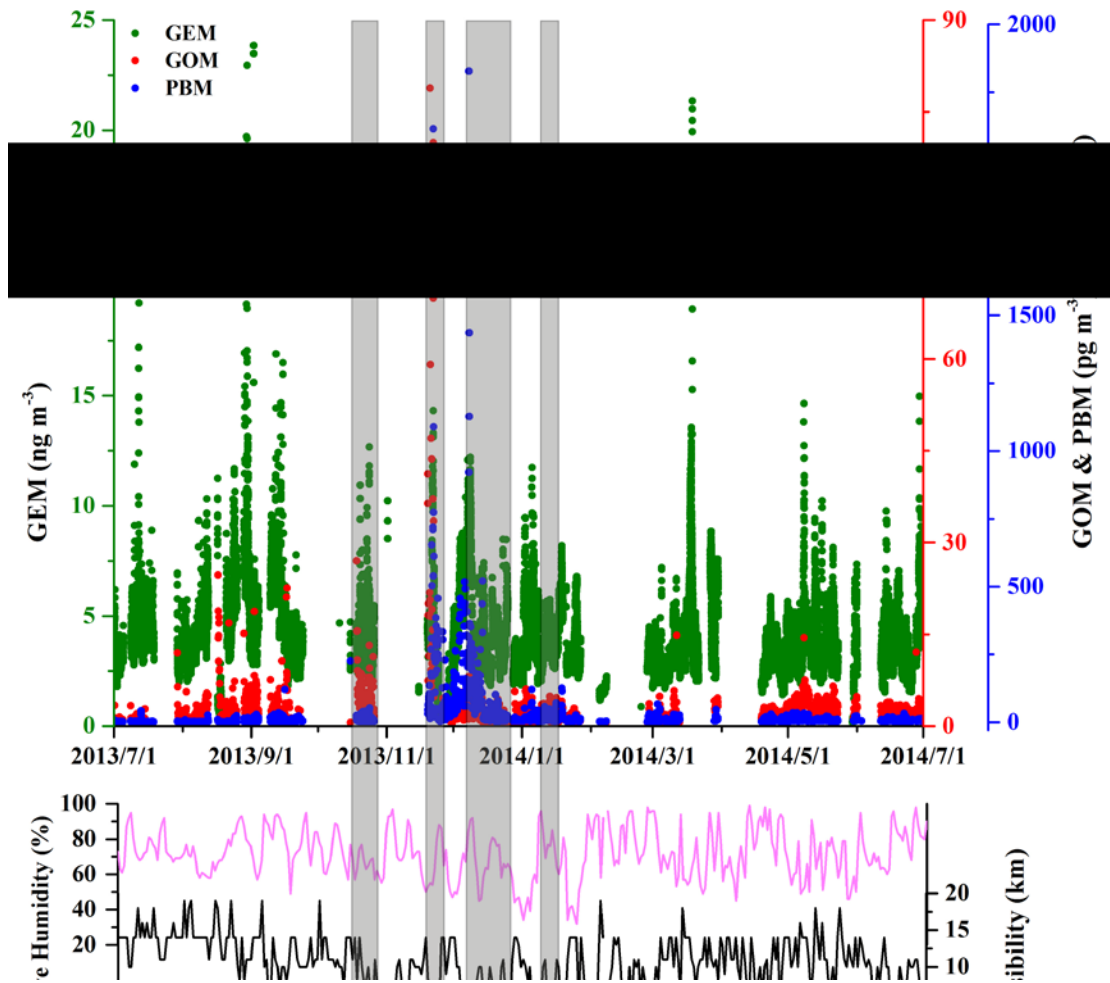
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Fig. 1. Location of the study site in Hefei, China.

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736 **Fig. 2.** Time series of GEM, GOM and PBM concentrations, along with visibility,
737 relative humidity, at the monitoring site in Hefei from July 2013 to June 2014. The
738 GEM data were at a 5-min resolution, and the GOM and PBM data were two-hour
739 averages. The gray columns show the major haze pollution episodes occurred during
740 the study period.

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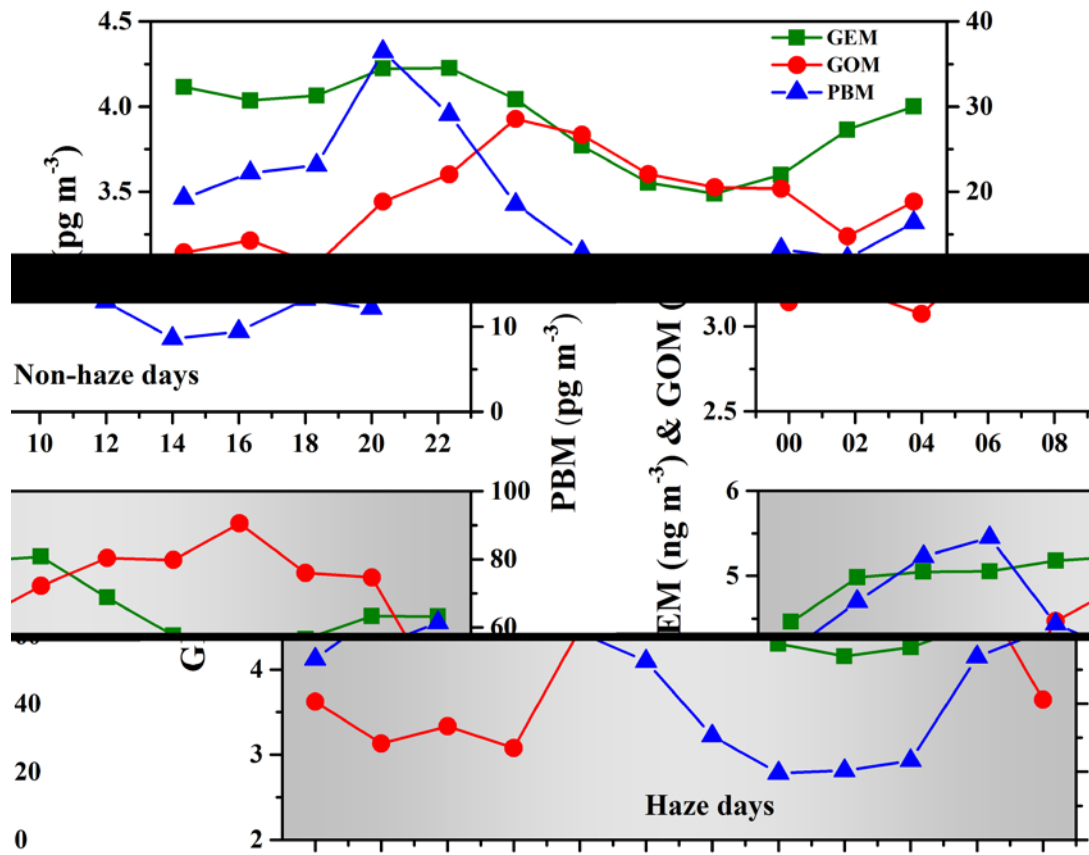
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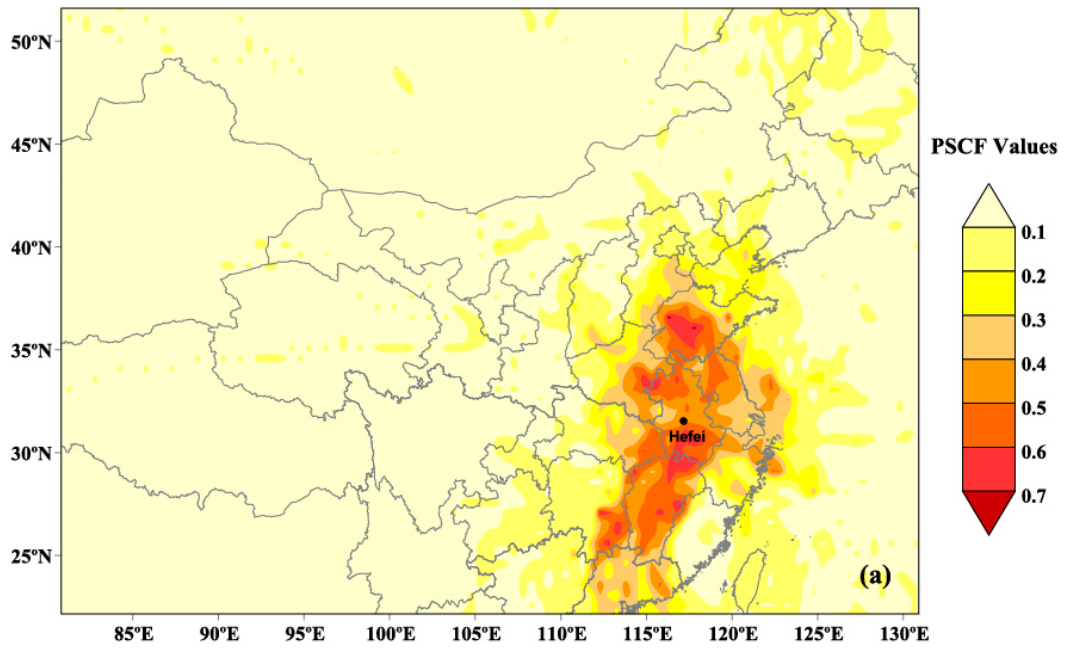
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749 **Fig. 3.** Diurnal trends of GEM, GOM and PBM concentrations in Hefei on non-haze

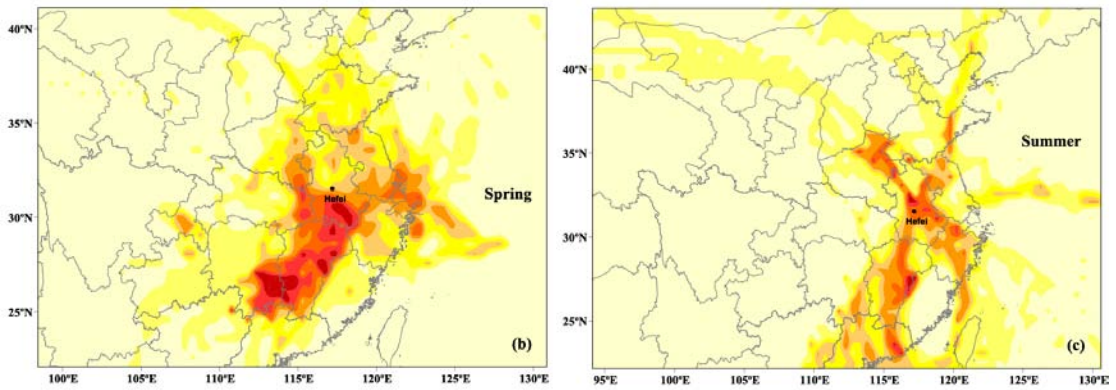
750 and haze days (Local time = UTC + 8 hr). The data were two-hour averages.

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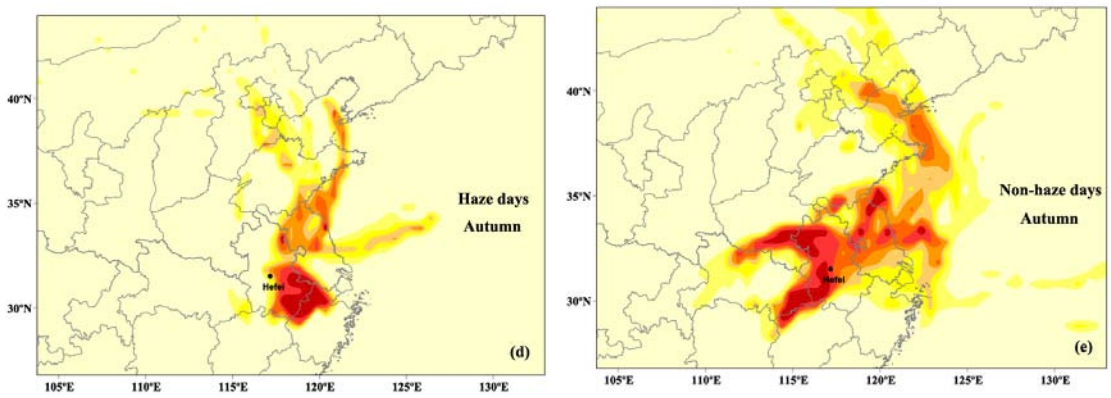


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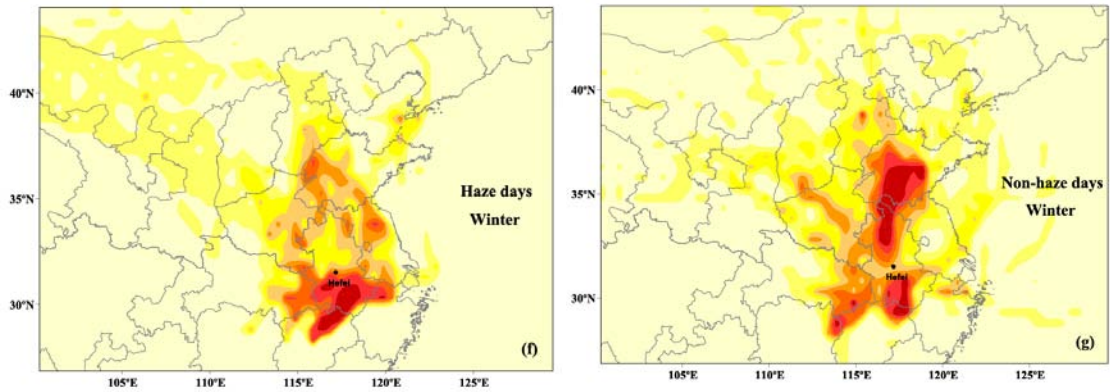


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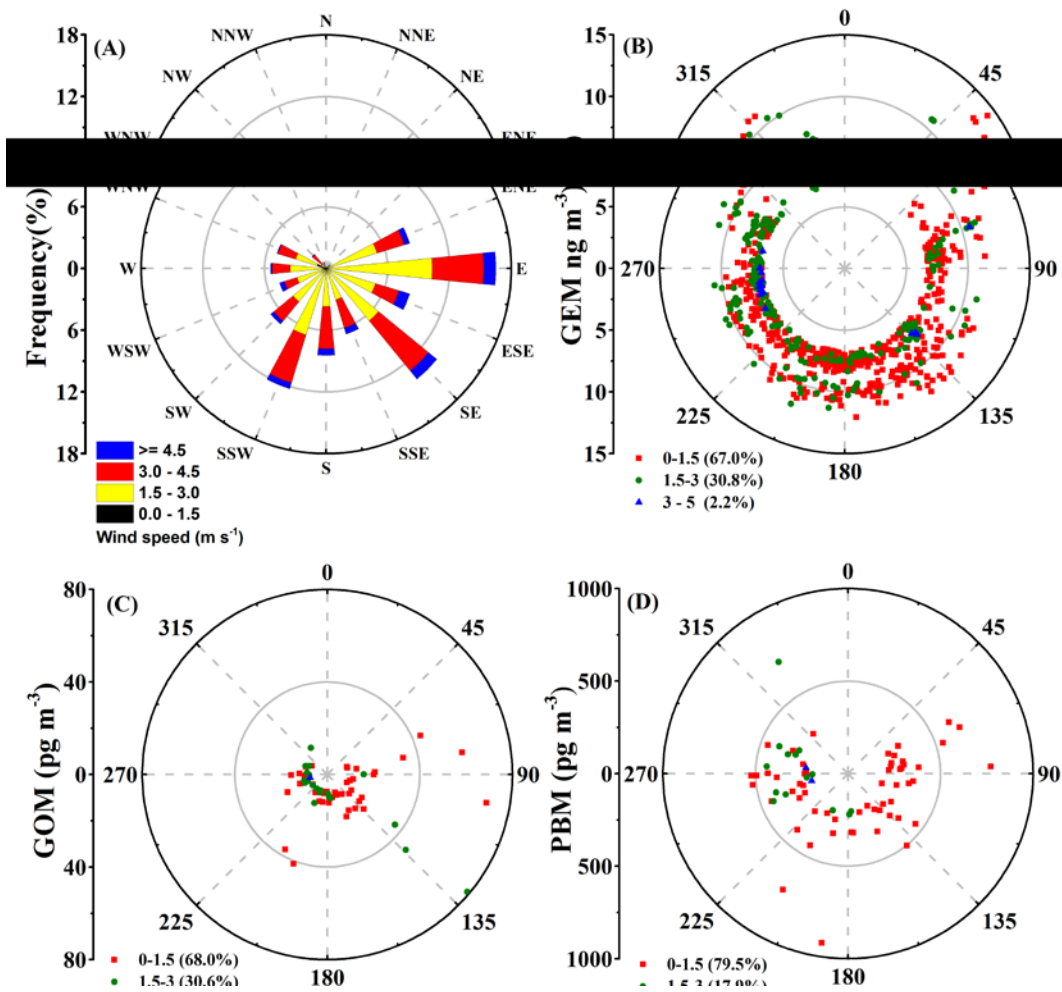
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762 **Fig. 4.** Likely emission source areas of GEM identified by PSCF analysis. (a) overall
763 (from July 2013 to June 2014), (b) spring, (c) summer, (d) haze days in autumn, (e)
764 non-haze days in autumn, (f) haze days in winter, (g) non-haze days in winter.

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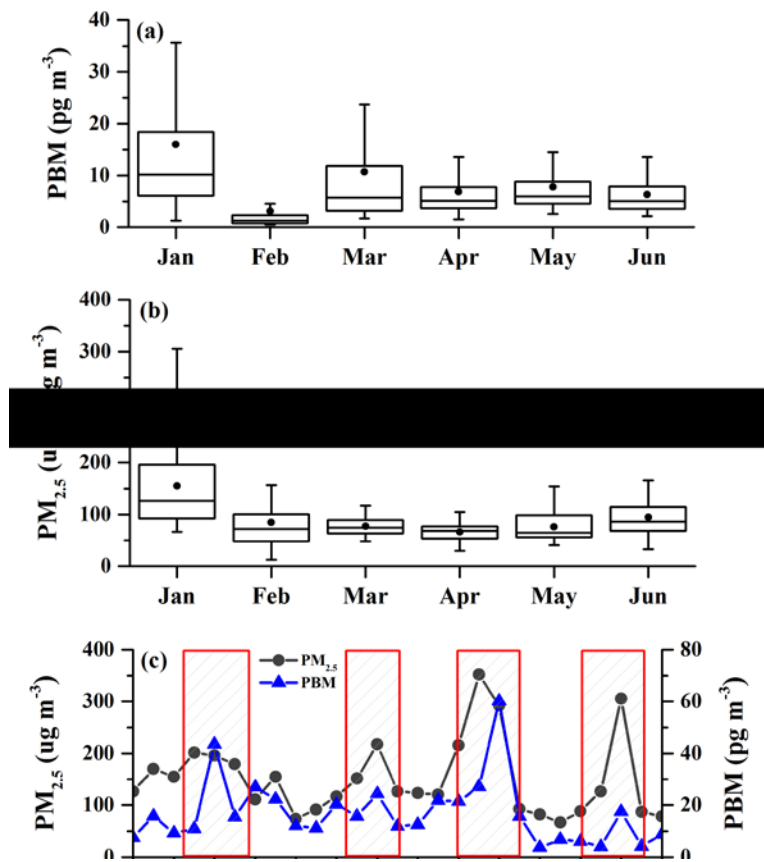


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771 **Fig. 5.** Wind direction and speed at the monitoring station during the study period. (A)
772 the wind rose for the entire study period; (B), (C) and (D) are the wind rose diagrams
773 for GEM, GOM and PBM concentrations above the 90th percentile values,
774 respectively.

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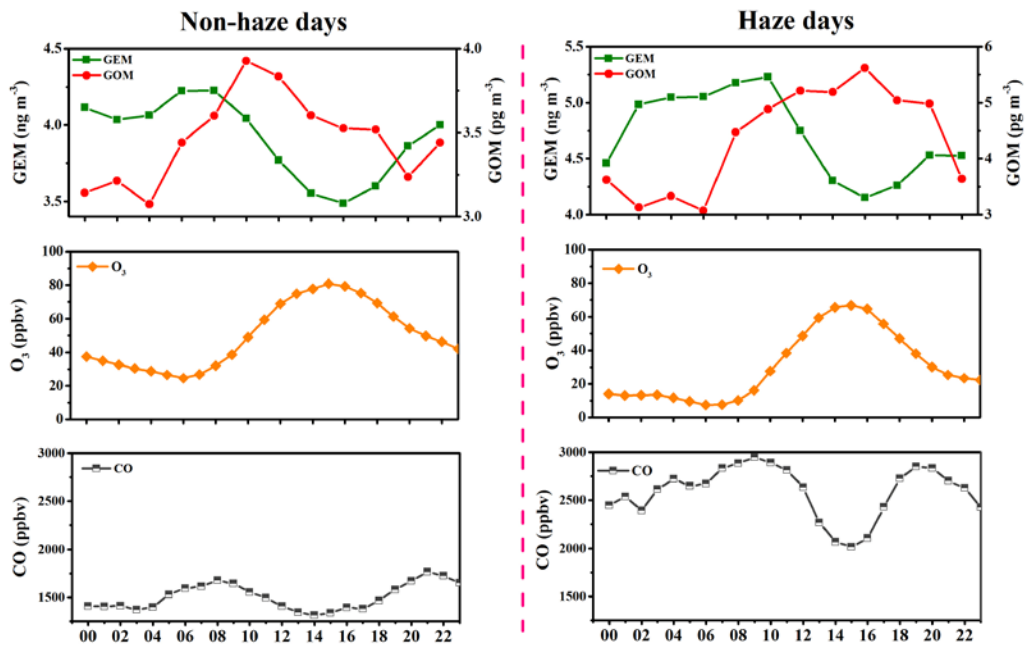
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782 **Fig. 6.** Monthly variation of (a) PBM and (b) $\text{PM}_{2.5}$ concentrations from January to

783 June, 2014, (c) average daily $\text{PM}_{2.5}$ and PBM concentrations in January, 2014.

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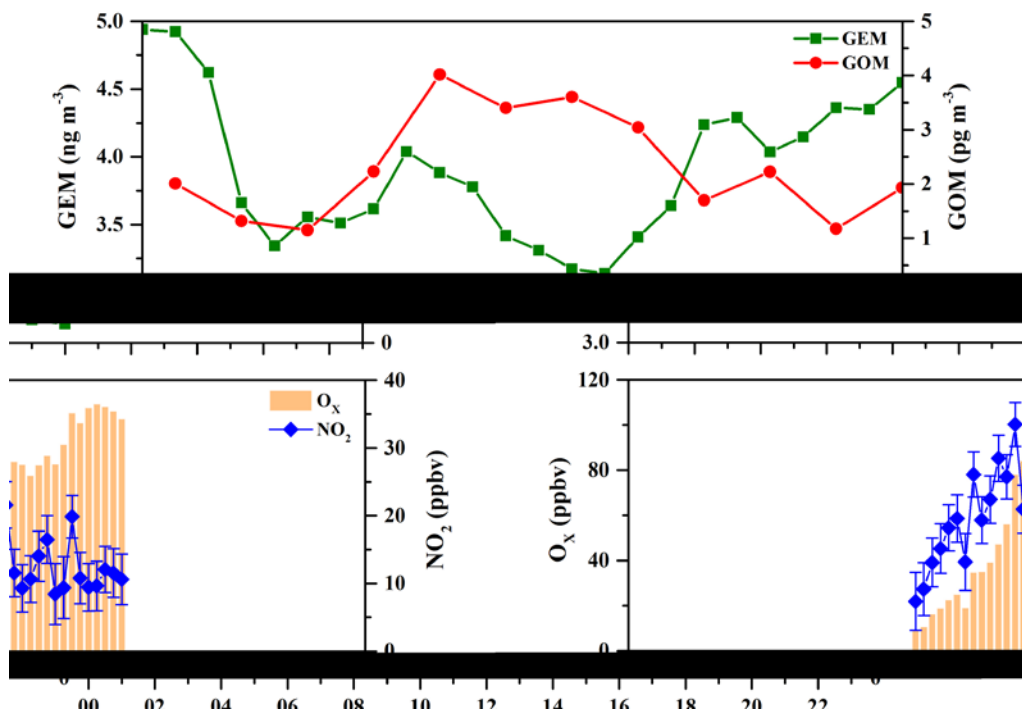


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787 **Fig. 7.** Diurnal variations of GEM, GOM, O₃ and CO concentrations on non-haze and
788 haze days.

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792 **Fig. 8.** A case study of diurnal variations of GEM, GOM, O_x , and NO_2 at Hefei (20th

793 November, 2013). The top panel shows the hourly averaged GEM and GOM

794 concentrations, and the bottom panel shows the O_x ($\text{O}_x = \text{NO}_2 + \text{O}_3$) and the NO_2

795 concentrations. The error bars for NO_2 refer to the NO_2 standard errors.

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