

1 **Response of winter fine particulate matter concentrations to**

2 **emission and meteorology changes in North China**

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1 **Abstract**

2 The winter haze is a growing problem in North China, but the causes have not been well  
3 understood. The chemistry version of the Weather Research and Forecasting model (WRF-  
4 Chem) was applied in North China to examine how the PM<sub>2.5</sub> concentrations change in response  
5 to changes in emissions (sulfur dioxide (SO<sub>2</sub>), black carbon (BC), organic carbon (OC),  
6 ammonia (NH<sub>3</sub>), and nitrogen oxides (NO<sub>x</sub>)), as well as meteorology (temperature, relative  
7 humidity (RH), and wind speeds) changes in winter. From 1960 to 2010, the dramatic changes in  
8 emissions lead to +260% increases in sulfate, +320% increases in nitrate, +300% increases in  
9 ammonium, +160% increases in BC and 50% increases in OC. The responses of PM<sub>2.5</sub> to  
10 individual emission specie indicate that the simultaneous increases in SO<sub>2</sub>, NH<sub>3</sub> and NO<sub>x</sub>  
11 emissions dominated the increases in PM<sub>2.5</sub> concentrations. PM<sub>2.5</sub> shows more notable increases  
12 in response to changes in SO<sub>2</sub> and NH<sub>3</sub> as compared to increases in response to changes in NO<sub>x</sub>  
13 emissions. In addition, OC also accounts for a large fraction in PM<sub>2.5</sub> changes. These results  
14 provide some implications for haze pollution control. The responses of PM<sub>2.5</sub> concentrations to  
15 temperature increases are dominated by changes in wind fields and mixing heights. PM<sub>2.5</sub> shows  
16 relatively smaller changes in response to temperature increases and RH decreases, compared to  
17 changes in response to changes in wind speed and aerosol feedbacks. From 1960 to 2010, aerosol  
18 feedbacks have been significantly enhanced, due to higher aerosol loadings. The discussions in  
19 this study indicate that dramatic changes in emissions are the main cause of increasing haze  
20 events in North China, and long-term trends in atmospheric circulations maybe another  
21 important cause since PM<sub>2.5</sub> is shown to be substantially affected by wind speed and aerosol  
22 feedbacks. More studies are necessary to get a better understanding of the aerosol-circulation  
23 interactions.

# 1 Introduction

2 PM<sub>2.5</sub> (particulate matter with diameter equal to or less than 2.5µm) is a main air pollution  
3 concern due to its adverse effects on public health (Gao et al., 2015; Pope et al., 2009). Pope et  
4 al. (2009) estimated that a decrease of 10µg PM<sub>2.5</sub> is related to about 0.6 year mean life  
5 expectancy increase. PM<sub>2.5</sub> is also associated with visibility reduction and regional climate  
6 (Cheung et al., 2005). Many cities in North China are experiencing severe haze pollution with  
7 exceedingly high PM<sub>2.5</sub> concentrations. In January 2010, a regional haze occurred in North China  
8 and maximum hourly PM<sub>2.5</sub> concentration in Tianjin was over 400µg/m<sup>3</sup> (Zhao et al., 2013). In  
9 January 2013, another unprecedented haze event happened, and the daily PM<sub>2.5</sub> concentrations in  
10 some areas of Beijing and Shijiazhuang reached over 500µg/m<sup>3</sup> (L. T. Wang et al., 2014), and  
11 instantaneous PM<sub>2.5</sub> concentration at some urban measurement sites were over 1000µg/m<sup>3</sup>  
12 (Zheng et al., 2015).

13 It is well known that particulate matter levels are strongly influenced by emissions and  
14 meteorological conditions (Steiner et al., 2006). The PM in the atmosphere can be directly  
15 emitted from sources like wildfires, combustion, wind-blown dust, and sea-salt, or formed from  
16 emitted gases through secondary aerosol formation mechanisms. Meteorology affects PM levels  
17 via changing emissions, chemical reactions, transport and deposition processes (Mu and Liao,  
18 2014). For example, increasing wildfire emission in North America is mainly caused by warmer  
19 temperatures and precipitation changes (Dawson et al., 2014), and increased temperature leads to  
20 higher biogenic emissions, which are important precursors of secondary organic aerosols  
21 (Dawson et al., 2014; Heald et al., 2008; Jacob and Winner, 2009). Increasing temperature also  
22 increases sulfate concentration due to the temperature dependence of SO<sub>2</sub> oxidation and resulting  
23 higher SO<sub>2</sub> oxidation rates (Aw and Kleeman, 2003; Dawson et al., 2007) and semi-volatile

1 aerosols may decrease due to evaporation under higher temperature (Sheehan and Bowman,  
2 2001; Dawson et al., 2007; Tsigaridis and Kanakidou, 2007). Higher relative humidity (RH)  
3 favors the formation of nitrate and increasing precipitation decreases all PM species via wet  
4 scavenging (Dawson et al., 2007; Tai et al., 2010). Furthermore, increasing clouds promote in-  
5 cloud sulfate production (Tai et al., 2010) and changes in wind speed and mixing height  
6 determines the dilution of primary and secondary PM (Jimenez-Guerrero et al., 2012; Megaritis  
7 et al., 2014; Pay et al., 2012).

8 With rapid economic and industrial developments, emissions in China have grown during the  
9 past years. It is estimated that NO<sub>x</sub> emissions in China increased by 70% from 1995 to 2004  
10 (Zhang et al., 2007), Black Carbon (BC) by ~50% from 2000 to 2010 (Lu et al., 2011), Organic  
11 Carbon (OC) by ~30% from 2000 to 2010 (Lu et al., 2011), and SO<sub>2</sub> by ~60% from 2000 to 2006  
12 (Lu et al., 2011). Apart from emission changes, it was observed that the winter is warming up in  
13 China, especially in the northern part (Guo et al., 2013; Hu et al., 2003; Ren et al., 2012). In  
14 addition, wind speed in North China has lowered (Shi et al., 2015; Wang et al., 2004) and RH  
15 has decreased in China (Song et al., 2012; Wang et al., 2004).

16 Many studies have investigated the impacts of emission changes on aerosol formation  
17 (Aksoyoglu et al., 2011; Andreani-Aksoyoglu et al., 2008; Megaritis et al., 2013; Tsimpidi et al.,  
18 2012a; Tsimpidi et al., 2012b) and the effects of climate/meteorology changes on PM<sub>2.5</sub>  
19 concentrations (Dawson et al., 2007; Megaritis et al., 2013; Megaritis et al., 2014; Tagaris et al.,  
20 2007; Tai et al., 2012a; Tai et al., 2012b) in Europe and in the United States. The haze pollution  
21 is growing in China, especially in North China, but the causes of the growth are not well  
22 understood. For haze pollution in China, it has been reported that aerosol feedbacks that change  
23 radiation and temperature can worsen pollution (Gao et al., 2016; Pet äet al., 2016; Xing et al.,

1 2015c; Zhang et al., 2015). In addition, the connections between haze and meteorological  
2 conditions have been established in many former studies (Fu et al., 2014; Jia et al., 2015; Leng et  
3 al., 2015; C. Li et al., 2015; Wang and Chen, 2016; Yang et al., 2016; X. Y. Zhang et al., 2015;  
4 Zhang et al., 2016). However, the roles of the large emission changes during the last 4 to 5  
5 decades and the observed meteorology changes in North China are not known.

6 The main objective of this study is to investigate the responses of  $PM_{2.5}$  and its major species to  
7 changes in emissions, including  $SO_2$ , BC, OC,  $NO_x$  and  $NH_3$ , and to temperature, RH and wind  
8 speed changes in North China region. Winter haze in North China has a large contribution from  
9 secondary inorganic aerosols and secondary inorganic aerosols are influenced by emissions,  
10 temperature and RH. The models used in previous studies of emissions and meteorology  
11 perturbations referenced above are all offline models, which are not capable of considering the  
12 feedbacks of changing meteorology on other meteorological variables, and the impacts of  
13 aerosols on meteorology. However, as pointed by Gao et al. (2016) and J. Wang et al. (2014)  
14 aerosol feedbacks should not be neglected when modeling aerosol in China. In this study, we  
15 consider aerosol feedbacks using the fully online coupled WRF-Chem model.

16 This paper is organized as follows. First, the WRF-Chem model, model settings and domain  
17 settings are briefly described and then in the next section, emission changes from 1960 to 2010  
18 and accordingly  $PM_{2.5}$  changes are discussed. After that, the responses of  $PM_{2.5}$  to changes in  
19 each emission species are analyzed. At last, the impacts of temperature, RH and wind speed  
20 changes on  $PM_{2.5}$  are analyzed and discussed.

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## 1 **2 Methodology**

### 2 **2.1 WRF-Chem model**

3 The WRF-Chem model is the chemistry version of the Weather Research and Forecasting model,  
4 which is fully online coupled that allows gases and aerosols simulations at the same time as  
5 meteorology simulations. In this study, we used a configuration that includes direct and indirect  
6 feedbacks. The gas phase mechanism used in this study is the Carbon Bond Mechanism version  
7 Z (CBM-Z), which includes 67 species and 164 reactions (Zaveri and Peters, 1999; Zaveri et al.,  
8 2008). The gas-particle partitioning module used is the MOSAIC module, which considers all  
9 important aerosol components, such as sulfate, nitrate, ammonium, BC, and OC (Zaveri et al.,  
10 2008). Eight size bins version of MOSAIC was used and the aerosol sizes ranged from 0.039 $\mu$ m  
11 to 10 $\mu$ m. Wind-blown dust was modeled online using the AFWA scheme. Two nested domains  
12 with 81km and 27km horizontal grid resolutions from outer to innermost and 27 vertical grids  
13 were used (Figure S1 in supplementary material). Analysis nudging of meteorology variables  
14 was used for the outer domain. The model physics configurations generally follow the settings in  
15 Gao et al. (2016). Inputs into the model include meteorological boundary and initial conditions  
16 (BCs and ICs) from NCEP FNL 1° $\times$ 1° data and chemical boundary and initial conditions from  
17 MOZART model simulations (Emmons et al., 2010). Chemical BCs and ICs are not changing  
18 along with the sensitivity simulations, but the studied domain (innermost domain) takes  
19 boundary conditions from outer main, where emissions are perturbed. The anthropogenic  
20 emission inventory used is the MACCity (MACC/CityZEN EU projects) emissions dataset,  
21 which provides monthly CO, NO<sub>x</sub>, SO<sub>2</sub>, VOC, BC, OC, and NH<sub>3</sub> emissions from different  
22 sectors for years between 1960 and 2020 (Granier et al., 2011). We compared the MACCity  
23 emission inventory for 2010 (Granier et al., 2011) with MIX emission inventory for 2010 (M. Li

1 et al., 2015) in the China region, and the magnitudes of emissions in China from these two  
2 datasets are very close. For example, the SO<sub>2</sub> emissions in China in 2010 were estimated to be  
3 28663 Gg in the MIX emission inventory, and were 26876.3 Gg in the MACCity emission  
4 inventory. Simulations for evaluating roles of emission changes were conducted using emissions  
5 for year 1960 and year 2010. We assigned emissions to the first 6 layers from surface based on  
6 sectors. For example, emissions from large point sources (such as chimneys) were assigned to  
7 higher layers. Biogenic emissions were estimated online using the MEGAN model (Guenther et  
8 al., 2006). The simulation period was January 2010 and five days in previous month were  
9 modeled as spin-up to overcome the influences of initial conditions.

10

## 11 **2.2 Sensitivity experiments**

12 We explored the sensitivities of the winter time haze event in 2010 to changes in emissions and  
13 meteorology features through a series of simulations using 1960 and 2010 emission baselines.  
14 Specifically, the influences of emission changes of SO<sub>2</sub>, BC, OC, NH<sub>3</sub>, and NO<sub>x</sub>, and  
15 meteorology (temperature, RH and wind speeds) changes on PM<sub>2.5</sub> and its major species were  
16 evaluated using a series of simulations. They are listed and explained in Table 1. All base  
17 simulations use meteorology of January 2010. CTL case uses emissions for year 1960 and  
18 EMI2010 case uses emissions for year 2010. SO<sub>2</sub>, NH<sub>3</sub>, and NO<sub>x</sub> emissions were perturbed  
19 separately from 1960 to 2010 (i.g., SO<sub>2</sub>-2010 NH<sub>3</sub>-2010 NO<sub>x</sub>-2010 cases). In the CTL\_NF and  
20 EMI2010\_NF cases, aerosol-radiation interactions are excluded based on emissions for year  
21 1960 and 2010. It was pointed out surface air temperature in North China increased at the rate of  
22 0.36 °C per decade (Guo et al., 2013), the linear trends coefficient of relative humidity anomaly

1 in North China is about -0.60% per decade (Wang et al., 2004), and national mean wind speed  
2 decreased 16% in the recent 50 years (Wang et al., 2004). To estimate the impacts of changes in  
3 temperature, RH and wind speed that happened in the past several decades, we decreased  
4 temperature by 2 degrees, increased RH by 10%, and increased wind speeds by 20%, to reflect  
5 conditions of early decades (CTL\_T2, CTL\_RH10, CTL\_WS20, EMI2010\_T2, EMI2010\_RH10,  
6 and EMI2010\_WS20 cases). These were conducted by perturbing the initial and boundary  
7 conditions of these individual meteorological variables.

8 At different vertical heights, emission and meteorological variables were uniformly perturbed.  
9 The changes of PM<sub>2.5</sub> and its major components due to perturbations in emissions and  
10 meteorology are analyzed for the North China region. The North China region is defined using  
11 the innermost domain (shown in Figure S1) and the statistics of changes are calculated within  
12 this domain for the January 2010 month.

13

### 14 **2.3 Model Verification**

15 The WRF-Chem model performance has been evaluated using multiple observations, including  
16 surface meteorological, chemical and optical data, and satellite data in Gao et al. (2016). The  
17 model was shown to capture the variations of surface temperature, RH, while wind speed was  
18 slightly overestimated (Gao et al., 2016), which has been reported as a common problem of  
19 current WRF-Chem model under low wind speed conditions. The Root Mean Square Error  
20 (RMSE) of temperature were all less than 3.2K and RMSEs of RH varied from 6.4 to 11.1%.  
21 The RMSE of wind speeds were below the proposed criteria (2m/s) (Emery et al., 2001) at the  
22 Beijing, Tianjin and Baoding stations, but larger than that criteria at the Chengde station. The

1 time series of simulated surface PM<sub>2.5</sub>, NO<sub>2</sub>, and SO<sub>2</sub> showed good agreement with observations  
2 as did simulated aerosol optical depth (AOD) (Gao et al., 2016). Mean Fractional Bias (MFB)  
3 ranged from -21.8% to 0.4% and Mean Fractional Error (MFE) ranged from 26.3% to 50.7%  
4 when comparing against PM<sub>2.5</sub> observations (Gao et al., 2016). In addition, the comparison  
5 between model results and satellite found that the vertical distribution of aerosol and horizontal  
6 distribution were captured well by the model (Gao et al., 2016). Compared with observed PM<sub>2.5</sub>  
7 composition, sulfate and OC were underestimated and nitrate was overestimated by the model  
8 (Gao et al., 2016). The underestimation of sulfate may be due to underestimation of SO<sub>2</sub> gas  
9 phase oxidation, errors in aqueous-phase chemistry, and/or missing heterogeneous sulfate  
10 formation (Gao et al., 2016).

11

## 12 **3 Results and Discussion**

### 13 **3.1 PM<sub>2.5</sub> sensitivity to emission changes from 1960 to 2010**

14 The emission changes of SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, BC and OC and resulting impacts on PM<sub>2.5</sub> from 1960  
15 to 2010 were examined based on the MACCity dataset for years 1960 and 2010. Figure 1(a-e)  
16 displays SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, BC and OC emissions for 1960 and Figure 1(f-j) shows the changes  
17 from 1960 to 2010. Populated regions of North China, such as urban Beijing, urban Tianjin, and  
18 urban Shijiazhuang, exhibit large emissions of SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, BC and OC in 1960. However,  
19 NH<sub>3</sub> emissions exhibit different spatial distribution patterns from SO<sub>2</sub>, NO<sub>x</sub>, BC and OC  
20 emissions, because NH<sub>3</sub> is mainly associated with agriculture while SO<sub>2</sub>, NO<sub>x</sub>, BC and OC are  
21 mainly related with industrial and residential activities. From 1960 to 2010, SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, BC  
22 and OC increased over the entire North China domain and markedly increased in the Jing-Jin-Ji

1 city cluster. In general, the domain averaged  $\text{SO}_2$  emissions and  $\text{NO}_x$  emissions in North China  
2 increased by  $\sim 220\%$  and  $\sim 990\%$  from 1960 to 2010, respectively. The domain averaged  $\text{NH}_3$   
3 emissions in North China increased by  $\sim 390\%$  from 1960 to 2010, but the most significant  
4 increases occurred not in the Jing-Jin-Ji city cluster, but in Inner Mongolia. Unlike  $\text{NH}_3$   
5 emissions, BC emissions increased the most in urban Beijing from 1960 to 2010. This is because  
6 residential sources are the biggest contributor to BC in winter (Li et al., 2016) and the population  
7 in urban Beijing sharply increased with rapid urbanization. From 1960 to 2010, the mean BC  
8 emissions in North China increased by  $\sim 154\%$ . Similar to BC emissions, OC emissions increased  
9 substantially in the center of Beijing, and the domain averaged increasing ratio is about 54%  
10 from 1960 to 2010. The enhancements of  $\text{SO}_2$ ,  $\text{NO}_x$ ,  $\text{NH}_3$ , BC and OC emissions in North China  
11 are expected to result in substantial increase in regional  $\text{PM}_{2.5}$  concentrations.

12 Figure 2 shows the simulated monthly mean concentrations of  $\text{PM}_{2.5}$  and its major components  
13 (sulfate, nitrate, ammonium, BC and OC) based on emissions for year 1960. As listed in Table 2,  
14 the domain averaged concentrations of sulfate, nitrate, ammonium, BC, OC, and  $\text{PM}_{2.5}$  are 1.9,  
15 0.8, 0.8, 1.5, 4.6, and  $19.2 \mu\text{g}/\text{m}^3$ , respectively. For year 1960,  $\text{PM}_{2.5}$  concentrations are mainly  
16 dominated by sulfate, OC and natural dust (the difference between  $\text{PM}_{2.5}$  and the sum of sulfate,  
17 nitrate, ammonium, BC, OC). Figure 3 displays the changes of sulfate, nitrate, ammonium, BC,  
18 OC, and  $\text{PM}_{2.5}$  due to changes in  $\text{SO}_2$ ,  $\text{NO}_x$ , BC and OC emissions from 1960 to 2010. The  
19 predicted monthly mean concentrations of  $\text{PM}_{2.5}$  components and  $\text{PM}_{2.5}$  increase everywhere  
20 over the entire domain due to emission changes resulting from the rapid urbanization and  
21 industrialization from 1960 to 2010 (Figure 3(a-f)). As listed in Table 2, the predicted monthly  
22 domain mean sulfate increases the largest ( $5.0 \mu\text{g}/\text{m}^3$ ), followed by nitrate ( $2.6 \mu\text{g}/\text{m}^3$ ) and OC  
23 ( $2.5 \mu\text{g}/\text{m}^3$ ).

1 From 1960 to 2010, the predicted BC increased by ~157% and OC increased by ~54% due to  
2 153% increase in BC emissions and 52% increase in OC emissions. The nearly linear response of  
3 both BC and OC aerosols to their emissions is due to the omission of a secondary organic aerosol  
4 formation in the chosen CBMZ/MOSAIC mechanism. Thus, both of them were treated as  
5 primary aerosols in these simulations. Our previous analyses indicate that SOA contribution in  
6 this time period was small (Gao et al., 2016). The domain mean PM<sub>2.5</sub> concentrations increased  
7 by 14.7μg/m<sup>3</sup> and the domain maximum increase is about 45μg/m<sup>3</sup> (Figure 3(f) and Table 2).

8 To quantify how much of the changes in Figure 3 are from the impacts of boundary conditions,  
9 we simulated another case with the innermost domain emissions fixed in 1960 and the outer  
10 domain emissions changed from 1960 to 2010. The impacts of boundary conditions mostly occur  
11 around the south boundary and show nearly no impact on PM<sub>2.5</sub> in Beijing (shown in Figure S2),  
12 which are consistent with the continuous weak southerly winds during the study period (Gao et  
13 al., 2016). On domain average, the impacts of boundary conditions result in 5.0ug/m<sup>3</sup> increase in  
14 the study domain, accounting for about 33.9% of the total changes in PM<sub>2.5</sub>.”

15 To explore how emission changes can affect haze days, we calculated the number of haze days in  
16 urban Beijing for the CTL and EMI\_2010 cases, using daily mean thresholds of 35 and 75μg/m<sup>3</sup>  
17 (China National Ambient Air Quality Grade I and Grade II Standard, L. T. Wang et al., 2014). In  
18 urban Beijing, there are 4 days when daily mean PM<sub>2.5</sub> concentrations are above 35μg/m<sup>3</sup>, and 0  
19 days with daily mean PM<sub>2.5</sub> concentrations above 75μg/m<sup>3</sup> for the CTL case. For the EMI\_2010  
20 case, these two numbers increase to 15 and 8, indicating that the large increases in emissions  
21 over the past several decades have significantly affected haze occurrences in Beijing.

22

## 1 **3.2 Sensitivity to changes in individual emission species**

2 The results discussed above show that in the winter period, the concentrations of secondary  
3 inorganic aerosols (sulfate, nitrate, and ammonium) has increased dramatically. Thus it is  
4 important to explore how sensitive secondary inorganic aerosol is to perturbations in precursor  
5 emissions. Three sensitivity simulations (change SO<sub>2</sub>, NH<sub>3</sub> and NH<sub>3</sub> emissions separately) were  
6 conducted to examine how changes in emissions of each species affect aerosol concentrations.  
7 The predicted changes of PM<sub>2.5</sub> and major PM<sub>2.5</sub> components at the ground-level are shown in  
8 Figure 4 and monthly domain mean aerosol changes are summarized in Table 3.

9

### 10 3.2.1 Changes in SO<sub>2</sub> emissions

11 Due to changes in SO<sub>2</sub> emissions from 1960 to 2010, domain averaged sulfate increase by  
12 3.4µg/m<sup>3</sup> (178.3%), nitrate decreases by -0.3µg/m<sup>3</sup> (-32.3%), and ammonium increases by  
13 0.2µg/m<sup>3</sup> (29.4%). NH<sub>3</sub> reacts preferentially with SO<sub>2</sub> (Seinfeld and Pandis, 2006) and free NH<sub>3</sub>  
14 reacts with enhanced H<sub>2</sub>SO<sub>4</sub> due to increasing SO<sub>2</sub>. As a result, ammonium increases and less  
15 HNO<sub>3</sub> gas is transferred to the aerosol phase, which is consistent with the responses to increasing  
16 SO<sub>2</sub> emissions in Kharol et al. (2013).

17

### 18 3.2.2 Changes in NH<sub>3</sub> emissions

19 As shown in Figure 4 and Table 3, changes in NH<sub>3</sub> emissions from 1960 to 2010 result in  
20 significant increases in nitrate (1.5µg/m<sup>3</sup>, +76.0%) and ammonium (0.6µg/m<sup>3</sup>, +84.0%). The  
21 domain mean changes of sulfate due to increase in NH<sub>3</sub> is close to zero (about 0.1µg/m<sup>3</sup>),  
22 because sulfate formation is only indirectly associated with NH<sub>3</sub> availability (Tsimpidi et al.,

1 2007). The significant changes in nitrate and ammonium occurred in south Hebei, Shandong, and  
2 Henan province, where anthropogenic  $\text{NO}_x$  emissions are very high (Figure 1). Although  $\text{NH}_3$   
3 emissions substantially increased in Inner Mongolia (Figure 1), responses of nitrate and  
4 ammonium are not significant there due to trivial  $\text{NO}_x$  emissions. The substantial increases of  
5 nitrate after  $\text{NH}_3$  emission increase indicate that  $\text{NH}_3$  limits the  $\text{NH}_4\text{NO}_3$  formation in North  
6 China region in this period.

7

### 8 3.2.3 Changes in $\text{NO}_x$ emissions

9 After changing  $\text{NO}_x$  emissions from 1960 to 2010 levels, domain mean surface  $\text{PM}_{2.5}$  decreases  
10 by about  $0.2\mu\text{g}/\text{m}^3$ , but the changes of individual  $\text{PM}_{2.5}$  inorganic components vary. The increase  
11 of  $\text{NO}_x$  emissions cause  $0.7\mu\text{g}/\text{m}^3$  (-39.1%) decrease in monthly domain mean sulfate and the  
12 domain peak sulfate reduction is about  $2.9\mu\text{g}/\text{m}^3$ . The OH radical is critical in the sulfate  
13 formation in the regions where  $\text{SO}_2$  concentrations are high and there is a competition between  
14  $\text{NO}_x$  and VOCs to react with OH (Tsimpidi et al. 2012b). When the VOCs/ $\text{NO}_x$  concentration  
15 ratio is close to 5.5:1, the OH reacts with  $\text{NO}_x$  and VOCs at an equal rate (Seinfeld and Pandis,  
16 2006). When the concentration ratio is lower than 5.5:1, the OH primarily reacts with  $\text{NO}_x$ , and  
17 the region with this concentration ratio is called VOC-limited region. In VOC-limited regions, an  
18 increase of  $\text{NO}_x$  will cause a decrease of OH and ozone concentration. When the VOCs/ $\text{NO}_x$   
19 concentration ratio is higher than 5.5:1, the OH will preferentially react with VOCs, and the  
20 region with this high ratio is called  $\text{NO}_x$ -limited region. In the  $\text{NO}_x$ -limited region, an increase of  
21  $\text{NO}_x$  will increase OH and ozone concentrations. In the simulated winter month, biogenic  
22 emissions are low and  $\text{NO}_x$  emissions in North China are very high, leading to lower VOCs to  
23  $\text{NO}_x$  ratios, and it can be considered as VOC-limited region. Fu et al. (2012) pointed out that

1 north East Asia is VOC-limited in January and urban areas of Beijing are VOC-limited in both  
2 January and July. The model averaged VOCs/NO<sub>x</sub> ratio changes from 4.2 to 1.2 due to emission  
3 perturbations from 1960 to 2010 (shown in Figure S3). As a result, the large increases in NO<sub>x</sub>  
4 emissions from 1960 to 2010 result in a 47.9% decrease in daytime surface ozone concentration  
5 and 55.6% decrease in daytime surface OH concentration, which are shown in Figure 5. Over the  
6 entire domain, ozone and OH decrease due to NO<sub>x</sub> emission increases (Figure 5). Consequently,  
7 sulfate aerosol decrease over the entire domain, as shown in Figure 4(i). Decreases in sulfate  
8 might also be related to changes in thermodynamics of the ammonium-sulfate-nitrate system.  
9 Although OH decreases, nitrate still rises ( $0.6\mu\text{g}/\text{m}^3$ , +76.0%) due to the increase in NO<sub>x</sub>  
10 emissions. The domain mean ammonium decreases by about 5.1% ( $-0.04\mu\text{g}/\text{m}^3$ ). The net effects  
11 of NO<sub>x</sub> emission increases bring about  $0.2\mu\text{g}/\text{m}^3$  decrease in monthly domain mean PM<sub>2.5</sub>  
12 concentration and the domain peak decrease is about  $1.1\mu\text{g}/\text{m}^3$  (Table 3).

13

#### 14 3.2.4 Coupled changes in SO<sub>2</sub>, NH<sub>3</sub> and NO<sub>x</sub> emissions

15 As shown above, increasing SO<sub>2</sub> emissions significantly increases PM<sub>2.5</sub> concentrations in North  
16 China region, increasing NH<sub>3</sub> emissions also increases PM<sub>2.5</sub> concentrations but to a lesser  
17 extent, and increasing NO<sub>x</sub> emissions slightly decreases PM<sub>2.5</sub> concentrations. The effects of  
18 coupled changes in SO<sub>2</sub>, NH<sub>3</sub> and NO<sub>x</sub> emissions are not a simple addition of the effect of changing  
19 them separately. As listed in Table 3, the monthly domain mean sulfate, nitrate, ammonium, and  
20 PM<sub>2.5</sub> increases more than the effects of changing emissions separately. Domain mean sulfate  
21 increases by  $5.0\mu\text{g}/\text{m}^3$  (+264.0%), nitrate increases by  $2.6\mu\text{g}/\text{m}^3$  (+322.5%), ammonium  
22 increases by  $2.3\mu\text{g}/\text{m}^3$  (295.2%) and PM<sub>2.5</sub> increases by  $9.9\mu\text{g}/\text{m}^3$  due to coupled changes in SO<sub>2</sub>,  
23 NH<sub>3</sub> and NO<sub>x</sub> emissions from 1960 to 2010. The simultaneous increases in SO<sub>2</sub>, NH<sub>3</sub> and NO<sub>x</sub>

1 emissions promote dramatic increases of secondary inorganic aerosols in North China, which is  
2 partially due to perturbations in VOCs and other species at the same time

3

#### 4 3.2.5 Changes in BC and OC emissions

5 Since BC and OC are treated as primary aerosols in the chosen CBMZ/MOSAIC mechanism,  
6 changes in their emissions do not show any impact on other aerosol components. As listed in  
7 Table 3, monthly domain mean  $PM_{2.5}$  increases by  $2.3\mu g/m^3$  and  $2.5\mu g/m^3$  due to changes in their  
8 emissions from 1960 to 2010, respectively.

9

### 10 **3.3 Effects of temperature increases**

11 The model used in this study is a fully online-coupled model, which simulates meteorological  
12 variables and chemical variables together. Therefore, it is not possible to increase temperature  
13 uniformly, as was done in previous studies using offline models (Dawson et al., 2007; Megartis  
14 et al., 2013; Megartis et al., 2014). To examine the sensitivity of  $PM_{2.5}$  to temperature change  
15 (reflecting the winter warming trends), we decrease temperature by 2 °C in the initial and  
16 boundary conditions to reflect conditions more like 1960. As a result, the monthly domain mean  
17 surface temperature increases 2.0 °C (CTL-CTL\_T2), but in a non-uniform manner. These  
18 responses in domain temperature are partially due to aerosol feedbacks. The spatial distributions  
19 of monthly mean surface temperature and temperature changes are shown in Figure 6(a). The  
20 monthly mean surface temperature increases more along top left domain boundaries and less  
21 over the Bohai sea. The influence of increasing temperature on biogenic emissions is included  
22 using temperature-sensitive biogenic emission model MEGAN (Guenther et al., 2006).

1 Due to the perturbation in temperature as mentioned above, sulfate, nitrate, ammonium and  
2  $PM_{2.5}$  are predicted to increase in most areas of the domain (Figure 7). Predicted monthly mean  
3 sulfate increases by  $0.06\mu\text{g}/\text{m}^3$  (+3.1%), nitrate increases by  $0.03\mu\text{g}/\text{m}^3$  (+4.2%), and ammonium  
4 increases by  $0.02\mu\text{g}/\text{m}^3$  (+2.8%). The increases of sulfate, nitrate and ammonium are mostly  
5 attributed to the increasing OH radicals, as shown in Figure 6(b). After the temperature  
6 perturbation, daytime OH increases by about 3.6% on domain average. It was found that higher  
7 temperature increased volatilization of ammonium nitrate and partitioned it to the gas phase  
8 (Megaritis et al., 2014), but it is not significant here due to the low temperature in winter. In  
9 addition, the increase of sulfate, nitrate, and ammonium could be partially due to accelerated gas-  
10 phase reaction rate at higher temperature (Dawson et al., 2007; Megaritis et al., 2014). It may  
11 also due to enhanced photolysis caused by decreases in cloudiness after temperature  
12 perturbations (as shown in Figure S4: changes in liquid water path).

13 As shown in Figure 7 (d-e), the concentrations of primary aerosols (BC and OC) also increase  
14 after the temperature perturbations. This is due to changes in other physical parameter, such as  
15 wind direction, wind speed, and PBLHs, which are key factors in the diffusion of air pollutants.  
16 Figure 6(c) shows that monthly PBLHs in most North China areas decrease after the temperature  
17 perturbation, and PBLHs over the Bohai sea decrease the most, with monthly mean decrease  
18 over 50 meters. The monthly domain average daytime PBLHs decrease about 2.3% due to  
19 changes in temperature vertical profiles. Surface horizontal winds also change (Figure 6(d)),  
20 which directly affect the distributions and magnitudes of  $PM_{2.5}$  concentrations in North China  
21 along with PBLH changes.

22

1 The responses of PM<sub>2.5</sub> concentrations to temperature perturbation are different from the  
2 responses of sulfate, nitrate, ammonium, BC and OC (Figure 7), with decreases in northwestern  
3 regions and increases in most areas of the North China Plain. This is because natural dust is  
4 dominant in northwestern regions (as shown in Figure 2(f)), and the concentrations of natural  
5 dust decrease under lower horizontal wind speeds (Figure 6(d)). The monthly PM<sub>2.5</sub>  
6 concentration decreases by 0.01 μg/m<sup>3</sup> on domain average due to temperature perturbation.  
7 Because of temperature increase, the numbers of haze days (defined using the daily mean  
8 threshold 35 and 75 μg/m<sup>3</sup>) in urban Beijing do not change.  
9 The discussions shown above are based on emission levels in 1960. The responses to  
10 temperature perturbations were also investigated based on emission levels in 2010, and the  
11 results are shown in Figure S5, S6 and Table 3. The spatial distributions of the changes are  
12 similar to the results shown above, but with larger magnitudes. The domain mean PBL heights  
13 decreases slightly more (-8.6 compared to -8.3 meters). The domain mean PM<sub>2.5</sub> concentrations  
14 and PM<sub>2.5</sub> components exhibit larger increases in North China, although daytime OH  
15 concentrations increases less ( $2.6 \times 10^{-9}$  compared to  $3.3 \times 10^{-9}$  ppmv), suggesting that the  
16 responses of PM<sub>2.5</sub> concentrations are mostly due to changes in PBL heights and wind fields.  
17

### 18 **3.4 Effects of RH decreases**

19 The RH was enhanced by 10% in model initial and boundary conditions to represent RH for the  
20 previous decades. As a result, the simulated monthly mean RH decreases by 9.3% on domain  
21 average. Due to RH perturbation, domain mean PM<sub>2.5</sub> concentration decreases by 0.7 μg/m<sup>3</sup>. As  
22 shown Figure 8(a), PM<sub>2.5</sub> concentrations decrease in the Jing-Jin-Ji region but increase in  
23 southern areas of the domain. The ammonium nitrate formation equilibrium depends on RH (Tai

1 et al., 2010), so  $\text{HNO}_3$  may be shifted to the gas phase under lower RH. In addition, the changes  
2 in RH can also affect the wet deposition rate. The increases in southern areas of the domain are  
3 mainly due to suppressed in-cloud scavenging, as the decreases in RH inhibit the formation of  
4 clouds. As shown in Figure 8(b), liquid water path (LWP) decreases by 75.0%. As a result, the  
5 in-cloud scavenging loss rate decreases. The changes of predicted aerosol optical depth at 600nm  
6 are shown in Figure 8(c). In most regions, visibility decreases due to lower RH. Because of RH  
7 decreases, the numbers of haze days (defined using the daily mean threshold 35 and  $75\mu\text{g}/\text{m}^3$ ) in  
8 urban Beijing do not change. The responses to RH perturbations were also investigated based on  
9 emission levels in 2010, and the results are shown in Figure S7 and Table 3. The responses are  
10 also similar to changes based on emission levels in 1960, but with larger magnitudes.

11

### 12 **3.5 Effects of wind speed decreases**

13 Simulations were also carried out when wind speeds in initial and boundary conditions were  
14 increased to estimate the wind speeds for the previous decades. The predicted domain averaged  
15 monthly mean wind speed decreases by about 0.7 m/s. As shown in Figure 9(a), the monthly  
16 mean near surface horizontal winds are pronounced in mountainous areas (northwest areas of the  
17 domain) and relatively smaller in other areas. Figure 9(b) shows the changes of wind speeds  
18 (CTL-WS20) due to model perturbations. The predicted monthly mean  $\text{PM}_{2.5}$  concentrations  
19 decrease by  $2.3\mu\text{g}/\text{m}^3$  on domain average, but the responses of  $\text{PM}_{2.5}$  vary within the domain. As  
20 shown in Figure 9(c),  $\text{PM}_{2.5}$  concentrations decrease in the northwestern areas because of lower  
21 production of natural dust under lower horizontal wind speeds. However, in most areas of the  
22 North China Plain,  $\text{PM}_{2.5}$  concentrations increase under lower wind speeds (Figure 9(c)). The  
23 domain peak increase is about  $2.4\mu\text{g}/\text{m}^3$ , which is based on low predicted  $\text{PM}_{2.5}$  concentrations

1 using emissions for year 1960. If the concentration in base case is higher, the responses will be  
2 enhanced. As shown in Figure 9(d), the domain maximum increases in  $PM_{2.5}$  increases from 2.4  
3 to  $9.4 \mu\text{g}/\text{m}^3$ . Because of wind speed decreases, number of haze days that daily mean  $PM_{2.5}$   
4 concentrations are above  $35 \mu\text{g}/\text{m}^3$  increases by 1.  
5

### 6 **3.6 Effects of changes in aerosol feedbacks**

7 As mentioned in Gao et al. (2016), high concentrations of aerosol enhance stability of boundary  
8 layer and increase  $PM_{2.5}$  concentrations. Due to dramatic changes in emissions from 1960 to  
9 2010, the strength of aerosol feedbacks may also have changed. To quantify these changes, we  
10 simulated four cases (i.e., CTL, CTL\_NF, EMI2010, and EMI2010\_NF). CTL-CTL\_NF and  
11 EMI2010-EMI2010\_NF are used to represent the contributions of aerosol radiative effects in  
12 1960 and 2010. The changes in monthly mean daytime PBL heights and  $PM_{2.5}$  concentrations are  
13 shown in Figure 10. In 1960, the domain averaged PBL height decreases by 6.7 meters due to  
14 aerosol radiative effects, and the domain maximum decrease is 25.4 meters. Correspondingly, the  
15 domain averaged  $PM_{2.5}$  increases by  $0.1 \mu\text{g}/\text{m}^3$  and the domain maximum increase is  $0.9 \mu\text{g}/\text{m}^3$ . In  
16 2010, the domain averaged PBL height decreases by 13.8 meters and the domain maximum  
17 decrease is 55.2 meters (more than two times compared to 1960). Correspondingly, the domain  
18 averaged  $PM_{2.5}$  increases by  $0.7 \mu\text{g}/\text{m}^3$  and the domain maximum increase is  $5.1 \mu\text{g}/\text{m}^3$ . The  
19 enhanced strength of aerosol feedbacks is another important cause of degraded aerosol pollution.  
20 Thus, controlling emissions will have a co-benefit of reducing strength of aerosol feedbacks.  
21

1 **3.7 Implications for the effects of emission and meteorology changes on PM<sub>2.5</sub>**  
2 **concentrations**

3 The simulated responses of PM<sub>2.5</sub> concentrations to emission changes and meteorology changes  
4 presented here, along with the previous presented effects of aerosol feedbacks (Gao et al. 2016),  
5 provide important implications for the causes of the dramatic increases in winter PM<sub>2.5</sub>  
6 concentrations.

7 We calculated domain maximum changes in PM<sub>2.5</sub> concentration averaged over four stagnant  
8 days (January 16-19) owing to emission changes from 1960-2010 (EMI2010-CTL), temperature  
9 increases (CTL-CTL\_T2), RH decreases (CTL-CTL\_RH10), wind speed decreases (CTL-  
10 CTL\_RH20), and aerosol feedbacks (CTL-CTL\_NF). The values are 137.7, 2.0, 2.6, 7.5 and  
11 4.0 $\mu\text{g}/\text{m}^3$ , respectively. When the perturbations are based on emission levels in 2010, domain  
12 maximum changes in PM<sub>2.5</sub> concentration due to temperature increases (EMI2010-  
13 EMI2010\_T2), RH decreases (EMI2010-EMI2010\_RH10), wind speed decreases (EMI2010-  
14 EMI2010\_WS20), and aerosol feedbacks (EMI2010-EMI2010\_NF) are 4.8, 4.7, 26.4 and  
15 25.5 $\mu\text{g}/\text{m}^3$ . The effects of emission changes on haze formation are dominant and the effects of  
16 aerosol feedbacks are comparable to the effects of wind speed decreases.

17 The comprehensive comparisons of these factors are also summarized in Table 3. Based on the  
18 monthly domain mean responses of PM<sub>2.5</sub> concentrations to these factors, dramatic emission  
19 changes due to urbanization and industrialization are the main causes of degraded air quality and  
20 frequent haze occurrences in in North China. PM<sub>2.5</sub> shows significant responses to changes in  
21 SO<sub>2</sub>, NH<sub>3</sub>, NO<sub>x</sub> emissions than BC and OC (about 106.3% higher). In addition, PM<sub>2.5</sub> shows  
22 significant increases in response to changes in SO<sub>2</sub> and NH<sub>3</sub> emissions, as compared to increases  
23 in response to changes in NO<sub>x</sub> emissions. This region is relatively ammonia-poor in winter, so

1 reducing NH<sub>3</sub> emissions might be effective, which is consistent with previous findings in Europe  
2 (Megaritis et al., 2013). SO<sub>2</sub> is the precursor of sulfate, which accounts for a large fraction of  
3 PM in this region. Thus, they should be preferentially controlled in order to reduce PM<sub>2.5</sub> levels.  
4 To control SO<sub>2</sub> emissions, the usage of natural gas or other clean energy should be promoted to  
5 reduce the usage of coal. NH<sub>3</sub> emissions in China are mainly from agriculture sources (about  
6 90%), including livestock, fertilizer, and agricultural soil (Huang et al., 2012). Lelieveld et al.  
7 (2015) found that agricultural emissions make the largest relative role in PM<sub>2.5</sub> concentration in  
8 eastern USA, Europe, Russia and East Asia. To control NH<sub>3</sub> emissions from agriculture sources,  
9 some animal feeding and animal housing strategies should be taken. In addition, controlling  
10 emissions will also have a co-benefit of reducing strength of aerosol feedbacks.

11 According to the ECLIPSE\_GAINS\_4a emission dataset, SO<sub>2</sub> emissions in China will decrease  
12 by -26%, NO<sub>x</sub> emissions in China will increase by 19%, and NH<sub>3</sub> emissions in China will  
13 increase by 14% from 2010 to 2030. We predicted (by perturbing SO<sub>2</sub>, NO<sub>x</sub> and NH<sub>3</sub> emissions  
14 by -26%, 19% and 14%) that these changes will lead to large decreases in winter sulfate (-  
15 2.3μg/m<sup>3</sup> on domain average). Nitrate will increase by 1.5μg/m<sup>3</sup> and ammonium will slightly  
16 decrease (-0.05μg/m<sup>3</sup>) on domain average. The net change of domain averaged PM<sub>2.5</sub>  
17 concentration is not significant (-0.8μg/m<sup>3</sup>), so more efforts are needed to control these important  
18 gaseous precursors.

19 From the information listed in Table 3, the responses of PM<sub>2.5</sub> concentrations to temperature and  
20 RH perturbations are not as significant as to wind speed perturbations. From Sect. 3.3, we also  
21 found that the effects of temperature perturbation on PM<sub>2.5</sub> concentration are dominant by  
22 changes in PBLH and wind fields. Previous studies have pointed out the occurrences of haze  
23 events are highly associated with atmospheric circulation anomalies (Chen and Wang, 2015;

1 Zhang et al., 2016). Thus, changes in atmospheric circulations may be another important cause of  
2 growing haze pollution, in addition to emission changes. Furthermore, aerosol can also change  
3 atmospheric circulation, especially in severely polluted East Asia. Thus, controlling emission  
4 may have co-benefits of mitigate aerosol effects on atmospheric circulation.  
5 The effects of changing atmospheric circulations on winter haze pollution in China is beyond the  
6 scope of this paper, but should be investigated in future studies.

7

#### 8 **4 Summary**

9 A fully online coupled meteorological and chemical transport model, WRF-Chem was used to  
10 study responses of winter PM<sub>2.5</sub> concentrations to changes in emissions of SO<sub>2</sub>, BC, OC, NH<sub>3</sub>,  
11 and NO<sub>x</sub> and to meteorology (temperature, RH, and wind speeds) changes in North China region,  
12 where people are suffering from severe winter haze pollution.

13 The detailed historical emissions dataset MACCity for year 1960 and 2010 were used to evaluate  
14 the impacts of changes in emissions of SO<sub>2</sub>, BC, and OC. From 1960 to 2010, the dramatic  
15 changes in emissions lead to +264.0% increases in sulfate, +322.5% increases in nitrate,  
16 +295.2% increases in ammonium, +157.0% increases in BC and 54% increases in OC. The  
17 domain mean PM<sub>2.5</sub> concentrations increase by 14.7μg/m<sup>3</sup> and the domain maximum increase is  
18 about 45μg/m<sup>3</sup>. The responses of PM<sub>2.5</sub> to individual emission species indicate that the  
19 simultaneous increases in SO<sub>2</sub>, NH<sub>3</sub> and NO<sub>x</sub> emissions dominated the increases in PM<sub>2.5</sub>  
20 concentrations. PM<sub>2.5</sub> shows significant increases in response to SO<sub>2</sub> and NH<sub>3</sub> emission changes.  
21 The increases in NO<sub>x</sub> emissions may decrease surface ozone concentration and surface OH

1 radical concentrations, because North China region is VOC-limited in the winter. In addition, OC  
2 accounts for a large fraction in  $PM_{2.5}$  changes.

3 The sensitivities of  $PM_{2.5}$  to emission changes of its precursors provide some implications for  
4 haze pollution control.  $SO_2$ ,  $NH_3$  and OC should be preferentially controlled. In China, the  
5 residential sector, particularly biofuel usage is the primary sources of OC (Lu et al., 2011). The  
6 usage of natural gas or other clean energy should be promoted to reduce the usage of coal and  
7 biofuel to reduce  $SO_2$  and OC. To control  $NH_3$  emissions from agriculture sources, some animal  
8 feeding and animal housing strategies should be taken.

9 The effects of changes in winter time meteorology conditions were also studied. Emission  
10 changes from 1960 to 2010 substantially increase numbers of haze days, but meteorology  
11 perturbations do not show any significant impacts. The perturbations in temperature and RH do  
12 change  $PM_{2.5}$  concentrations, but the strength is not as significant as the effects of wind speed  
13 and emission changes. The effects of temperature perturbation are dominated by the changes in  
14 surface wind fields and PBLHs. The effect of aerosol feedbacks is comparable to the effect of  
15 decreasing wind speeds and the strength of aerosol feedbacks significantly increased from 1960  
16 to 2010.

17 The above discussions indicate that aerosol concentrations are mainly controlled by atmospheric  
18 circulations, except emission changes. Thus, long-term trends in atmospheric circulations maybe  
19 another important cause of winter haze events in North China. More studies are necessary to get  
20 a better understanding of the aerosol-circulation interactions.

21 In our previous modeling study of the same period (January 2010), we found that SOA  
22 contribution was small, so we did not include SOA in this study. But this indication might be

1 problematic due to current poorly parameterized SOA scheme. In the future, how changes in  
2 emissions and meteorology variables affect productions of SOA during winter should be further  
3 studied using more advanced SOA schemes. In addition, we did not consider primary PM except  
4 BC and OC in the model because there is no information in the MACCity emission inventory,  
5 which is another direction for improvements in future studies.

6

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13

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Table 1. Simulation cases and descriptions

Cases	Descriptions
CTL	Base case, anthropogenic emissions are from MACCity dataset for year 1960
EMI2010	Anthropogenic emissions are from MACCity dataset for year 2010
SO <sub>2</sub> -2010	Same as CTL case except SO <sub>2</sub> emissions are for year 2010
NH <sub>3</sub> -2010	Same as CTL case except NH <sub>3</sub> emissions are for year 2010
NO <sub>x</sub> -2010	Same as CTL case except NO <sub>x</sub> emissions are for year 2010
CTL_T2	Same as CTL case except temperature BCs and ICs are decreased by 2K
CTL_RH10	Same as CTL case except RH BCs and ICs are increased by 10%
CTL_WS20	Same as CTL case except wind speed BCs and ICs are increased by 20%
CTL_NF	Same as CTL case except aerosol-radiation interactions are excluded
EMI2010_T2	Same as EMI2010 case except temperature BCs and ICs are decreased by 2K
EMI2010_RH10	Same as EMI2010 case except RH BCs and ICs are increased by 10%
EMI2010_WS20	Same as EMI2010 case except wind speed BCs and ICs are increased by 20%
EMI2010_NF	Same as EMI2010 case except aerosol-radiation interactions are excluded

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1 Table 2. Monthly domain mean concentrations of PM<sub>2.5</sub> and its major components for year 1960, and  
 2 domain maximum and mean concentrations for changes from 1960 to 2010 due to emission changes  
 3 (µg/m<sup>3</sup>)

Years		SO <sub>4</sub> <sup>2-</sup>	NO <sub>3</sub> <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>	BC	OC	PM <sub>2.5</sub>
1960	Domain mean	1.9	0.8	0.8	1.5	4.6	19.2
1960-2010	Domain maximum	18.9	7.8	6.8	9.9	11.1	45.0
	Domain mean	5.0 (264.0%)	2.6 (322.5%)	2.3 (295.2%)	2.3 (156.6%)	2.5 (54.0%)	14.7 (76.4%)

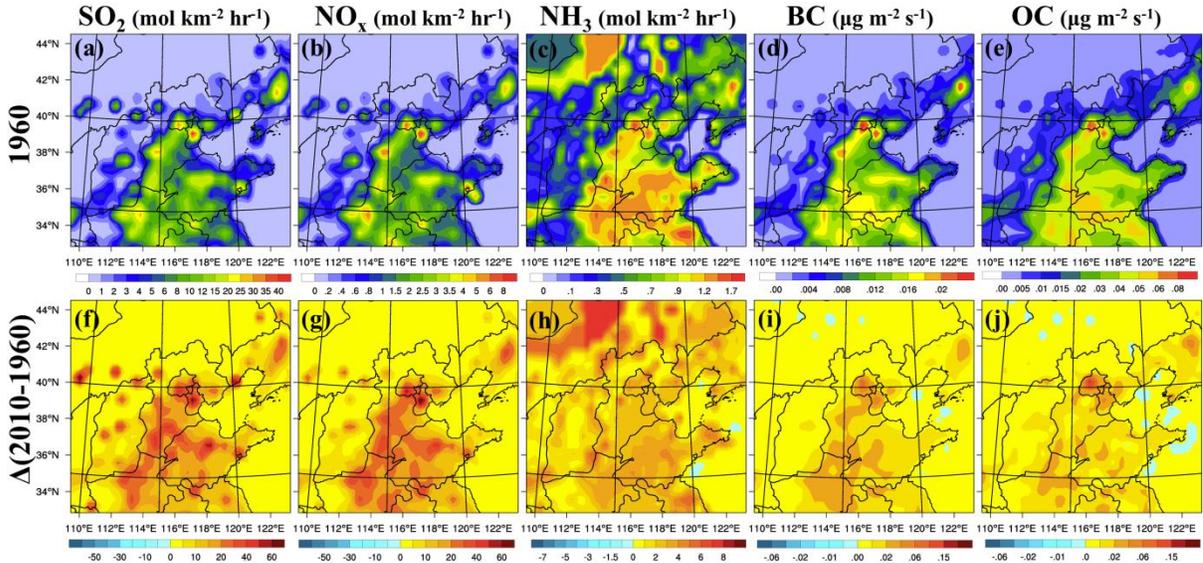
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1 Table 3. Monthly domain mean changes of sulfate, nitrate, ammonium and PM<sub>2.5</sub> concentrations (µg/m<sup>3</sup>)  
 2 due to emission and meteorology perturbations, and aerosol feedbacks (the two values of PM<sub>2.5</sub> changes  
 3 are for meteorology perturbations and aerosol feedbacks based on 1960 and 2010 emission levels,  
 4 respectively)

	SO <sub>4</sub> <sup>2-</sup>	NO <sub>3</sub> <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>	PM <sub>2.5</sub>
Changes in SO <sub>2</sub> emissions	3.4(178.3%)	-0.3 (-32.3%)	0.2 (29.4%)	3.4
Changes in NH <sub>3</sub> emissions	0.1 (5.3%)	1.5 (189.6%)	0.6 (84.0%)	2.3
Changes in NO <sub>x</sub> emissions	-0.7 (-39.1%)	0.6 (76.0%)	-0.04 (-5.1%)	-0.2
Changes in SO <sub>2</sub> , NH <sub>3</sub> , NO <sub>x</sub> emissions	5.0 (264.0%)	2.6 (322.5%)	2.3 (295.2%)	9.9
Changes in BC emissions	-	-	-	2.3
Changes in OC emissions	-	-	-	2.5
Temperature perturbations	-	-	-	-0.01/0.3
RH perturbations	-	-	-	-0.7/-1.1
Wind speed perturbations	-	-	-	-2.3/-0.5
Aerosol feedbacks				0.1/0.7

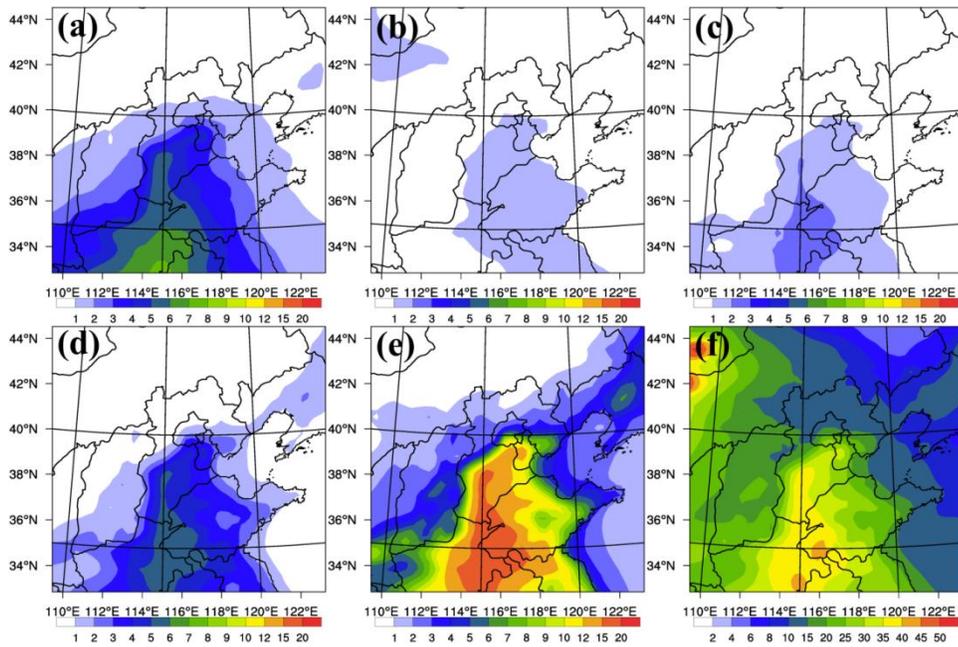
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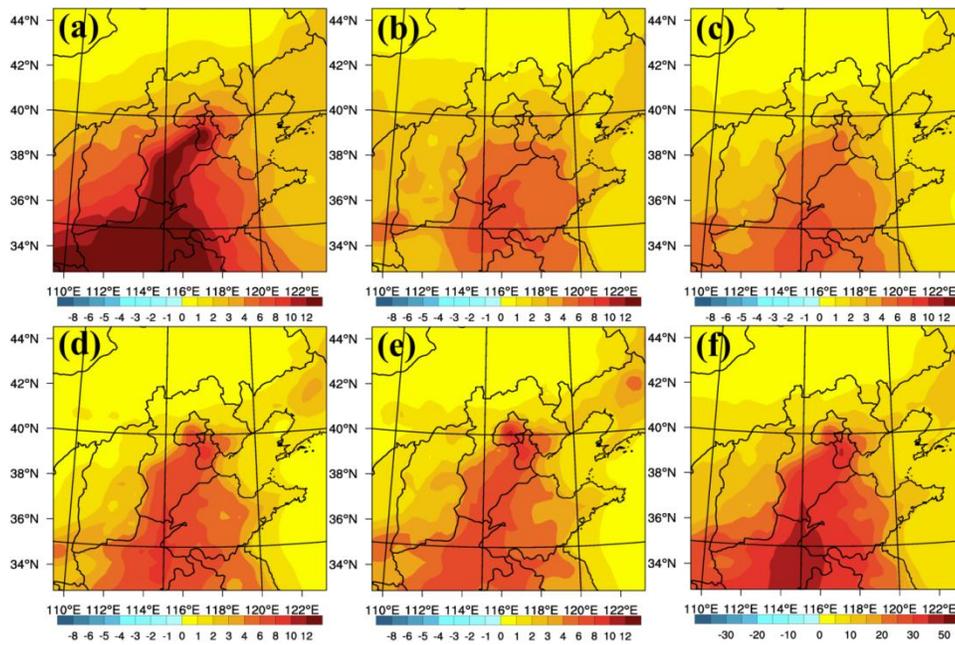
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Figure 1. SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, BC and OC emissions for year 1960 (a-e), and the changes of them from 1960 to 2010 (f-j)



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Figure 2. Predicted monthly mean sulfate (a), nitrate (b), ammonium (c), BC (d), OC (e) and PM<sub>2.5</sub> (f) concentrations based on emissions for year 1960



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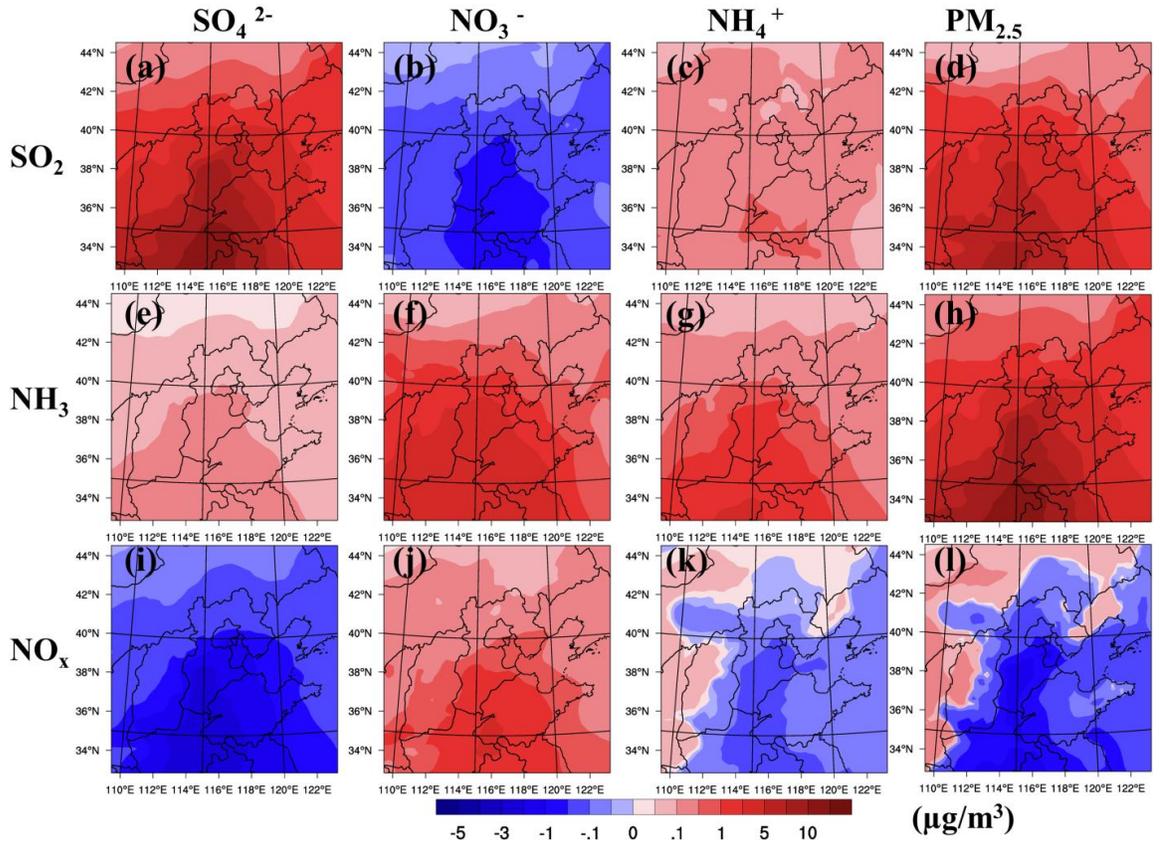
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Figure 3. Predicted monthly mean changes of sulfate (a), nitrate (b), ammonium (c), BC (d), OC (e)

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and  $PM_{2.5}$  (f) due to emission changes from 1960 to 2010

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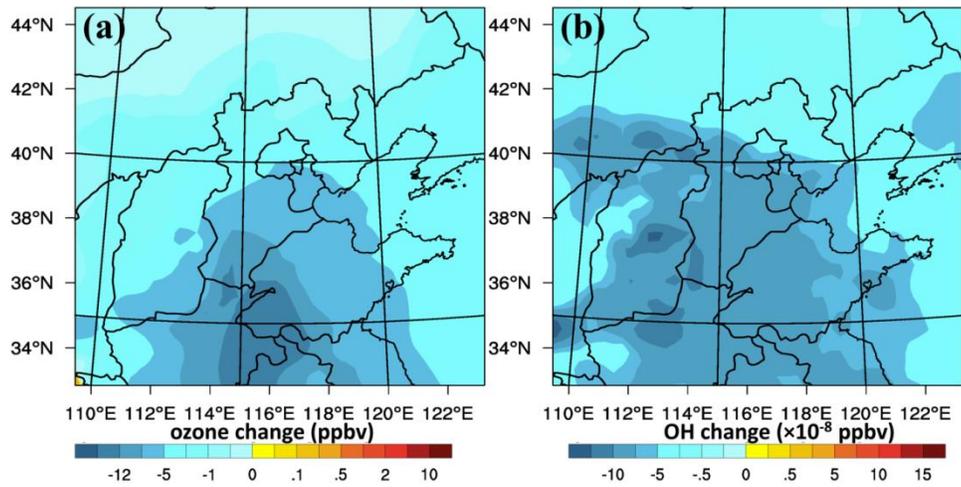
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Figure 4. Responses of PM<sub>2.5</sub> and major PM<sub>2.5</sub> inorganic species (sulfate, nitrate, and ammonium) to individual changes in SO<sub>2</sub>, NH<sub>3</sub> and NO<sub>x</sub> emissions from 1960 to 2010

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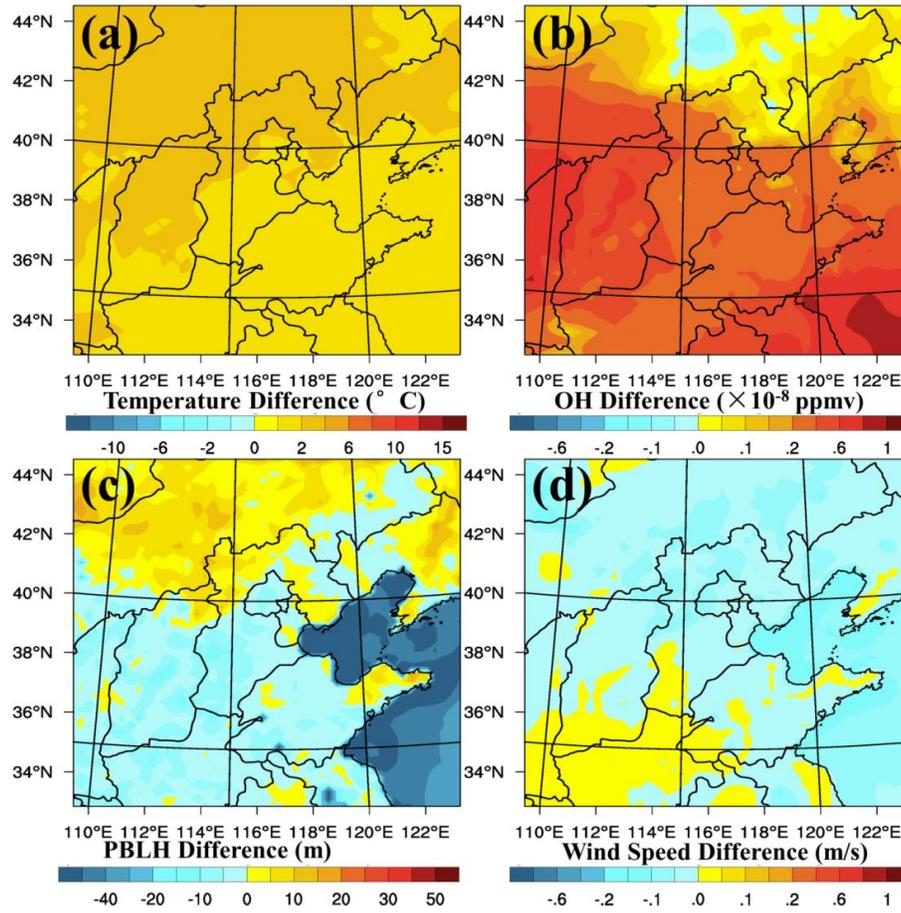


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Figure 5. Daytime ozone (a) and daytime OH (b) changes due to NO<sub>x</sub> emission increases

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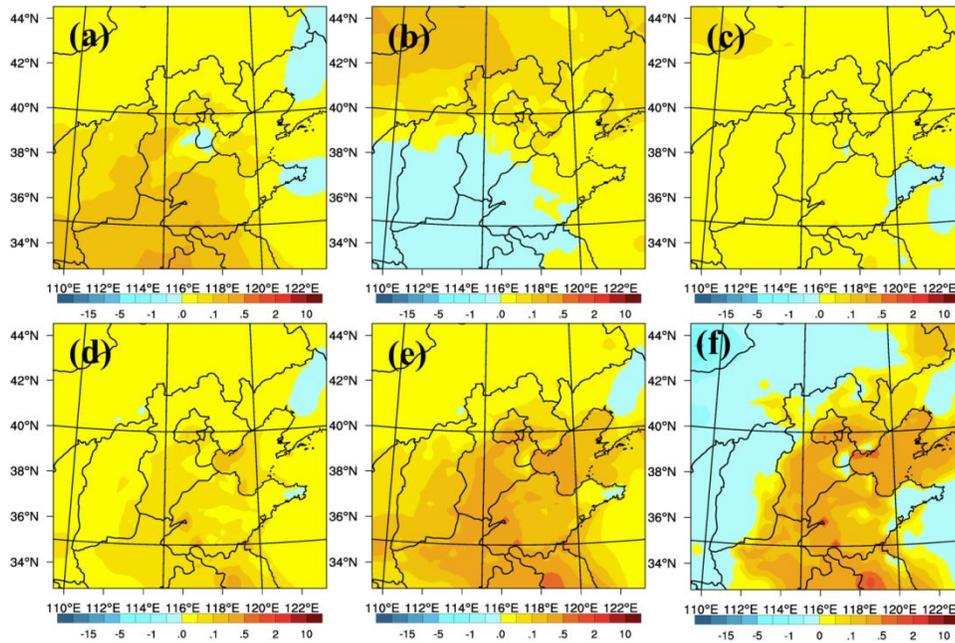
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3 Figure 6. Monthly mean temperature difference due to perturbation in initial and boundary conditions

4 (a), and daily mean OH (b), mean PBLH (c) and mean near surface wind speed changes (d) due to

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temperature increase



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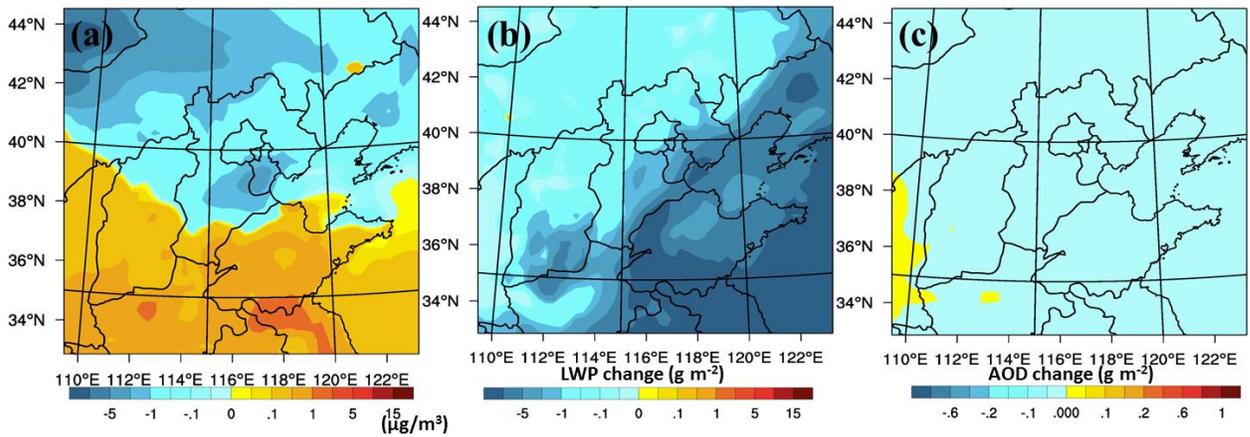
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Figure 7. Monthly mean changes of sulfate (a), nitrate (b), ammonium (c), BC (d), OC (e), and PM<sub>2.5</sub>

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(f) and due to temperature increase

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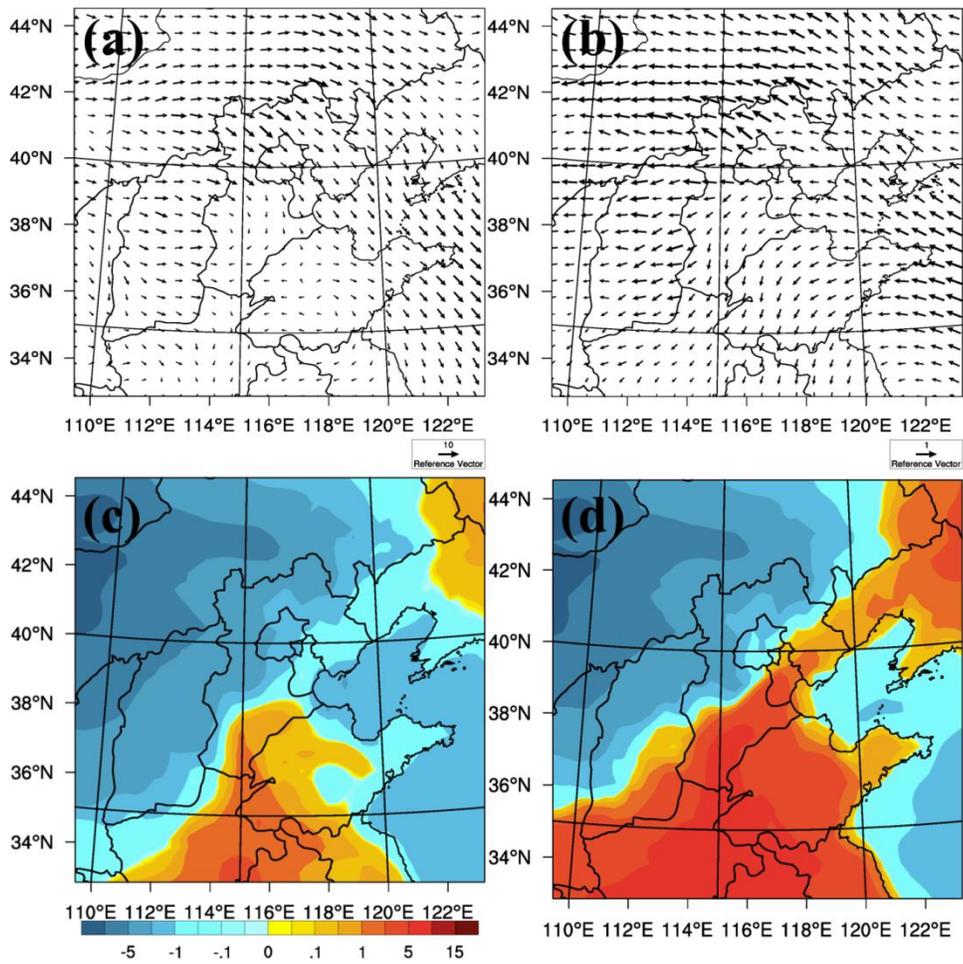


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Figure 8. Monthly mean changes of PM<sub>2.5</sub> (a), LWP (b), and AOD at 600nm (c) due to RH decrease

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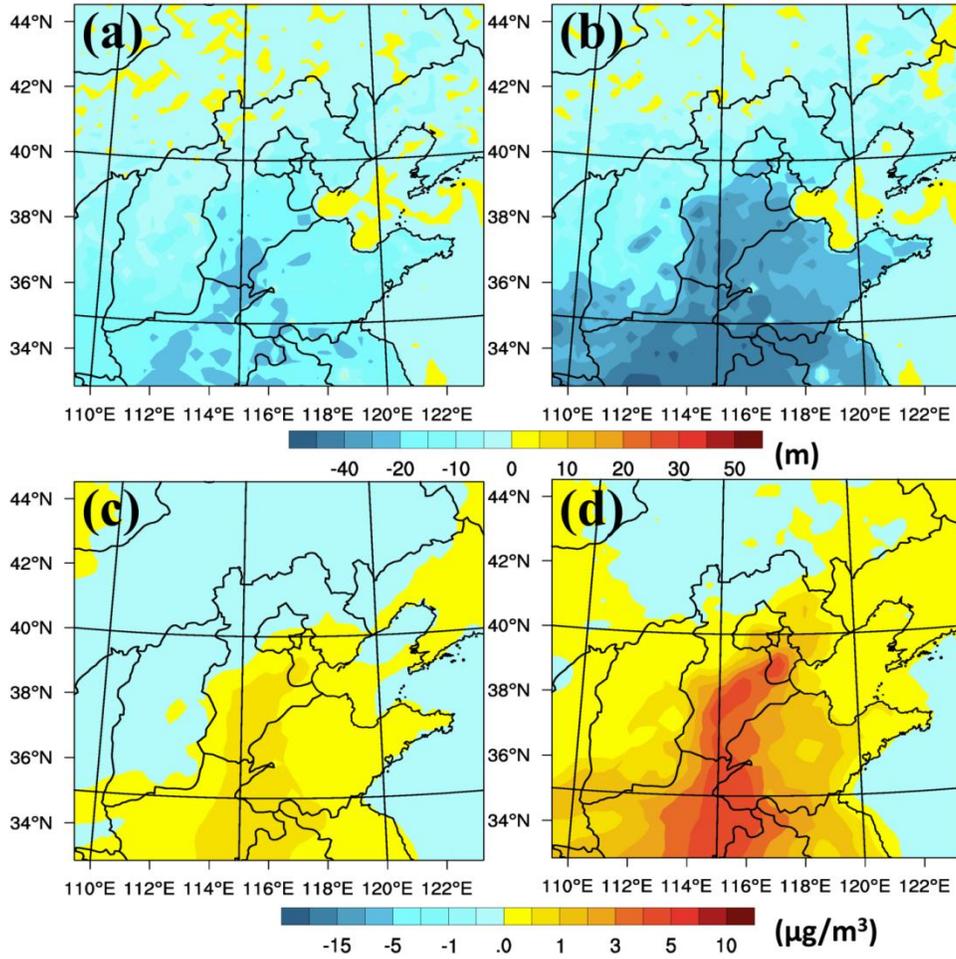
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Figure 9. Monthly mean wind fields for WS20 case (a) and changes of wind speeds (CTL-CTL\_WS20) (b), and mean changes of PM<sub>2.5</sub> concentrations based on 1960 emission levels (c) and 2010 emission levels (d)



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Figure 10. Monthly mean changes of daytime PBL heights for year 1960 (a) and 2010 (b), and of daytime PM<sub>2.5</sub> concentrations for year 1960 (c) and 2010 (d) due to aerosol-radiation interactions