



1 Depletion of atmospheric gaseous elemental mercury by plant uptake at

2 Mt. Changbai, Northeast China

3 Xuewu Fu¹, Wei Zhu¹, Hui Zhang¹, Jonas Sommar¹, Xu Yang¹, Xun Wang^{1,2}, Che-Jen Lin^{1,3,4}, Xinbin Feng^{1,*}

4 ¹State Key Laboratory of Environmental Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences, 99

5 Lincheng West Road, Guiyang, 550081, China

6 ²University of the Chinese Academy of Sciences, Beijing 100049, China

7 ³Department of Civil and Environmental Engineering, Lamar University, Beaumont, Texas 77710, United States

8 ⁴Center for Advances in Water and Air Quality, Lamar University, Beaumont, Texas 77710, United States

9

10 Correspondence to: Xinbin Feng (fengxinbin@vip.skleg.cn)

11

12



13 Abstract: There exists observational evidence that GEM can be readily removed from the
14 atmosphere via chemical oxidation followed by deposition in the polar and sub-polar regions,
15 free troposphere, lower stratosphere, and marine boundary layer under specific environmental
16 conditions. Here we report GEM depletions in a temperate mixed forest at Mt. Changbai,
17 Northeast China. The depletion occurred exclusively at night during leaf-growing season and in
18 the absence of GOM enrichment ($GOM < 3 \text{ pg m}^{-3}$). Vertical gradients of decreasing GEM
19 concentrations from layers above to under forest canopy suggest in situ loss of GEM to forest
20 canopy at Mt. Changbai. Foliar GEM flux measurements showed that the foliage of two
21 predominant tree species is a net sink of GEM at night, with a mean deposition flux of -1.8 ± 0.3
22 $\text{ng m}^2 \text{ h}^{-1}$ over *Fraxinus mandshurica* (deciduous tree species) and $-0.1 \pm 0.2 \text{ ng m}^2 \text{ h}^{-1}$ over
23 *Pinus Koraiensis* (evergreen tree species). Daily integrated GEM $\delta^{202}\text{Hg}$, $\Delta^{199}\text{Hg}$, and $\Delta^{200}\text{Hg}$ at
24 Mt. Changbai ranged from -0.34 to 0.91‰, from -0.11 to -0.04‰ and from -0.06 to 0.01‰,
25 respectively. A large positive shift of GEM $\delta^{202}\text{Hg}$ occurred during the GEM depletion events,
26 whereas $\Delta^{199}\text{Hg}$ and $\Delta^{200}\text{Hg}$ remained essentially unchanged. The observational findings and box
27 model results show that uptake of GEM by forest canopy plays a predominant role in the GEM
28 depletion at Mt. Changbai forest. Such depletion events of GEM are likely to be a widespread
29 phenomenon, suggesting that the forest ecosystem represents one of the largest sinks ($\sim 1930 \text{ Mg}$)
30 of atmospheric Hg at global scale.

31

32



33 1 Introduction

34 Mercury (Hg) is a persistent toxic air pollutant that is ubiquitously distributed in the
35 atmosphere. There are three operationally defined Hg forms: gaseous elemental mercury (GEM),
36 particulate bound mercury (PBM), and gaseous oxidized mercury (GOM). The sum of GEM and
37 GOM is known as total gaseous mercury (TGM). Because of its mild reactivity, high volatility,
38 low dry deposition velocity and water solubility, GEM is the most abundant form of Hg in the
39 atmosphere (Gustin and Jaffe, 2010;Holmes et al., 2010). The cycling of GEM in the atmosphere
40 is largely depending on either direct dry deposition or chemical oxidation followed by wet and
41 dry deposition. The residence time of GEM in the atmosphere is estimated to be in the range of
42 0.5 - 2.0 yr, based on global Hg budget and empirical models (Lindberg et al., 2007). Over the
43 past decades, our understanding regarding the sources and sinks of atmospheric Hg has been
44 improved (Strode et al., 2007;Selin et al., 2008;Holmes et al., 2010;Amos et al., 2013). For
45 instance, the discovery of atmospheric mercury depletion events (AMDEs) in polar and
46 sub-polar regions demonstrated that atmospheric GEM can be readily removed from the
47 atmosphere via reactive halogens-induced oxidation, leading to a deposition of up to 300 Mg yr⁻¹
48 to the arctic (Schroeder et al., 1998;Ebinghaus et al., 2002;Lindberg et al., 2002;Steffen et al.,
49 2008). Similar depletion events occurred in the marine boundary layer at middle latitude to a
50 lesser extent (Brunke et al., 2010;Obrist et al., 2011;Timonen et al., 2013). Fast oxidation of
51 GEM by ozone, reactive halogens and hydroxyl radicals in the free troposphere has also been
52 observed (Swartzendruber et al., 2006;Fain et al., 2009;Swartzendruber et al., 2009b;Lyman and
53 Jaffe, 2012;Shah et al., 2016). These findings indicate that GEM probably has a much shorter
54 atmospheric residence time under specific environmental conditions (Holmes et al., 2010). Dry
55 deposition of GEM (V_d) depends on surface characteristics, meteorological variables, biological
56 and chemical conditions of soil and water. V_d over non-vegetated surfaces (bare soil) and water
57 bodies is typically small (less than 0.03 cm s⁻¹) to counter the emission and re-emission of GEM
58 from these surfaces (Zhang et al., 2009). Therefore soil and water have long been considered a
59 net GEM source (Selin et al., 2007;Holmes et al., 2010). In contrast, strong dry deposition of



60 GEM to vegetated surfaces and wetlands are frequently observed with V_d up to about 2 cm s^{-1}
61 (Zhang et al., 2009), suggesting vegetative surfaces an important sink of GEM.

62 Forest represents a dominant terrestrial ecosystem on the Earth and covers an area of
63 $\sim 4 \times 10^7 \text{ km}^2$. It readily removes trace gases such as CO_2 , ozone, sulfur dioxide, and nitrogen
64 oxides, as well as aerosols from the atmosphere (Munger et al., 1996; Finkelstein et al.,
65 2000; Zhang et al., 2001; Pan et al., 2011). However, there are ongoing debates regarding whether
66 or not forest is a sink or a source of atmospheric Hg. Previous laboratory studies suggested that
67 foliar exchange of GEM is bi-directional with net deposition occurring at elevated Hg
68 concentration and net emission under typical background concentrations (Hanson et al.,
69 1995; Ericksen and Gustin, 2004; Gustin et al., 2004; Graydon et al., 2006). Lindberg et al. (1998)
70 measured GEM fluxes over a mature deciduous forest using the modified Bowen ratio (MBR)
71 method and suggested that global forest is a net source of GEM with an emission ranging from
72 850 to 2000 Mg yr^{-1} . Later, the observation of Hg fluxes in a deciduous forest using a relaxed
73 eddy accumulation (REA) method showed seasonal shift in flux with a net deposition of GEM
74 during leaf-growing season (Bash and Miller, 2009). Although the discrepancy in the measured
75 GEM exchanges between forest and atmosphere is partially attributed to the uncertainties of the
76 flux quantification method (Sommar et al., 2013), there is a need to clarify the role of forest
77 ecosystem in the mass budget of atmospheric GEM. A study in Québec, Canada showed that
78 GEM concentrations at a maple forest site are consistently lower than those measured at an
79 adjacent open site (Poissant et al., 2008). Similarly, the lower GEM concentrations observed in
80 leaf-growing season at many forest sites across the Atmospheric Mercury Network (AMNet) in
81 USA (Lan et al., 2012) also suggest forest a net GEM sink. Currently, it is still unclear whether
82 the loss of GEM over forest is caused by direct dry deposition to canopy or chemical
83 conversions of GEM to GOM (Mao et al., 2008).

84 In this study, we report consistent GEM depletion events during the leaf-growing season in
85 a temperate mixed forest in Northeast China over a time scale of 7 years. Atmospheric Hg
86 speciation, vertical gradient of GEM, foliage/air and soil/air exchange flux of GEM and isotope



87 signatures of GEM samples have also been observed in an intensive campaign to explore the
88 possible mechanisms responsible for the observed GEM depletion.

89 **2 Material and Methods**

90 **2.1 Site description**

91 The study site (42°24'0.1"N, 128°06'25"E, 738 m above sea level) is located in a temperate
92 broadleaf and Korean pine mixed forest on the north slope of Mt. Changbai (Figure S1). The
93 forest is dominated by tree species of *Pinus koraiensis*, *Fraxinus mandshurica*, *Tilia amurensis*,
94 *Acer mono* and *Quercus mongolica*. The height of the forest canopy is 5 - 22 m (mean = 18.3 m)
95 with the heights of mature trees (>50 y) and young trees (<20 y) and shrubs ranging from 15 to
96 22 m and from 5 to 10 m, respectively. Regions to the east and south of the site consist of
97 pristine forest with little anthropogenic influence. Most of the regional industrial sources are
98 located more than 50 km west of the sampling site (Supplementary Figure S1).

99 **2.2 Atmospheric Hg measurements**

100 From Oct 2008 to Jul 2013 and from Jul 2014 to Dec 2015, TGM concentrations were
101 continuously measured using an automated Hg vapor analyzer (Tekran® 2537, Tekran Inc.,
102 Canada). The analyzer has been used extensively for atmospheric TGM measurements
103 worldwide. The analyzer was calibrated automatically every 25 h using the internal Hg⁰
104 permeation source. The permeation rate of the internal source was manually calibrated every 4 -
105 6 months by using an external Hg vapor source (Tekran® 2505). The sampling inlet was
106 mounted at a height of 24 m above ground level (agl, ~3 m above canopy) by using a 25 m
107 Teflon tube and a 15 m heated Teflon tube. Atmospheric TGM consists of GEM and GOM.
108 Gustin et al. (2013;2015) proposed that GOM could be transformed to GEM within the
109 uncovered Teflon tubing, which in turn would be transported efficiently through the tubing and
110 quantified by the Tekran analyzer. However, GOM generally constitutes a small portion of TGM
111 (mean of 0.32% on basis of one year of measurements and will not exceed 1% using a three-fold
112 correction factor to adjust GOM concentrations measured by the Tekran® speciated system)



113 (Gustin et al., 2015). Therefore, we interpret the TGM observations as GEM throughout the
114 paper.

115 GEM, GOM and PBM were measured using the Tekran® 2537/1130/1135 unit (Tekran Inc.,
116 Canada) from Jul 2013 to Jul 2014. The sampling inlet was positioned at 4 m agl in a small
117 clearing plot with tall trees of ~5 m from the system. This system has been widely used and
118 described in detail by many earlier studies (Landis et al., 2002; Lindberg et al., 2002; Lan et al.,
119 2012; Fu et al., 2016). Briefly, GOM, PBM, and GEM in ambient air were collected onto
120 KCl-coated annular denuder, quartz fiber filter and dual gold cartridges in sequence. This system
121 was programmed to collect GOM and PBM at 1-h intervals at a volumetric flow rate of 10 L
122 min⁻¹. GEM was collected from air samples at 5-min intervals at a volumetric flow rate of 1.0 L
123 min⁻¹. Once collected, Hg is thermally decomposed from each unit and detected by cold vapor
124 atomic fluorescence spectroscopy (CVAFS) as Hg⁰. KCl-coated denuder, Teflon coated glass
125 inlet, and impactor plate were replaced bi-weekly and quartz filters were replaced monthly.
126 Denuders and quartz filters were prepared and cleaned before field sampling following the
127 methods in Tekran technical notes. GEM concentrations measured at 4 m agl in the small
128 clearing plot and at 45 m agl (~24 m above canopy) did not bias significantly with each other
129 with a mean difference of 0.03 ng m⁻³ (3% of the mean GEM concentration during the study
130 period) (Supplementary Figure S2). The two Tekran instruments used for this comparison were
131 run side by side for 2 days in the laboratory and showed a mean systematic uncertainty of 1.8 ±
132 1.1% (ranging from 0% to 5.7%). This indicates the measurements at 4 m agl in the small
133 clearing plot did not significantly underestimate the GEM concentrations of ambient air in the
134 study area. In the study area, GEM also has a fast dry deposition velocity within the forest (more
135 details in sections below), although to a lesser extent compared to atmospheric GOM. We
136 therefore assume that the measurements of GOM in the clearing plot didn't result in significantly
137 biased low GOM concentrations and were representative of ambient air in the study area.

138 Vertical profile of GEM concentrations at 1 m, 10 m, 24 m, and 45 m agl within the forest
139 were measured from 10 to 15 Jul 2013 using the Tekran® 2537 analyzer and the Tekran® 1115



140 Synchronized Multi-Port manifold (Tekran Inc., Canada). The sampling duration of GEM during
141 the vertical gradient measurements was programmed to be 2.5 min, and switching of ports of the
142 manifold was made every 5 min.

143 The GEM detection limit for 7.5 L samples measured with Tekran® 2537 analyzer as
144 specified by Tekran Instrument Corporation is 0.1 ng m^{-3} . Due to the lack of understanding of
145 the specific forms and calibration standards of GOM, there are uncertainties regarding the GOM
146 measurements (Gustin et al., 2015). Previous studies suggested that GOM measured by the
147 Tekran system could be biased low and a correction factor of 3 should be applied for adjusting
148 GOM concentrations measured by the Tekran system (Gustin et al., 2013;Huang et al.,
149 2013;Gustin et al., 2015;Huang and Gustin, 2015). Tekran® 2537's default integration at low Hg
150 loading ($\sim 1 \text{ pg}$ per cycle) was reported to have a 25% underestimation of GEM concentration.
151 This could also underestimate GOM concentrations when GOM concentrations were lower than
152 2 pg m^{-3} (Swartzendruber et al., 2009a). These analytical uncertainties are taken into account for
153 the discussions of GEM depletion mechanism in the Results and Discussion section.

154 **2.3 Foliar GEM exchange**

155 Exchange flux of GEM between leaf and the atmosphere was measured using a new
156 dynamic flux bag method described by Graydon et al. (2006) which is thought to maintain
157 normal physiological function of enclosed foliage. Briefly, a Tedlar® gas sampling bag ($\sim 20 \text{ L}$
158 volume, polyvinyl fluoride, DuPont, USA) enclosed living intact leaves, and the foliar GEM flux
159 was obtained via measuring the difference in GEM concentrations at the inlet and outlet of the
160 flux bag. Ambient air was pumped into the flux bag using a Mini Diaphragm vacuum pump
161 (N89 KTDC, KNF, Germany, oil-free, brushless and with diaphragm coated with PTFE). GEM
162 flux was calculated using Equation (1):

$$163 \quad F = (C_o - C_i) \times Q / A \quad (1)$$

164 where F is the foliar GEM flux in $\text{ng m}^{-2} \text{ h}^{-1}$, with positive and negative fluxes representing
165 emission and deposition, respectively, C_o and C_i are the GEM concentrations at the outlet and
166 inlet of the flux bag, respectively, which were measured by the Tekran® 2537 analyzer, Q is the



167 flushing flow rate of air through the flux bag ($0.5 \text{ m}^3 \text{ h}^{-1}$), and A is the single-sided leaf area
168 enclosed by the flux bag in m^2 .

169 Two tree species, *Fraxinus mandshurica* (deciduous tree species) and *Pinus Koraiensis*
170 (evergreen tree species), were selected for the foliar GEM flux measurement. They are the
171 predominant species in the study area with the basal coverage of the *Fraxinus Mandschurica*
172 and *Pinus Koraiensis* accounting for 26.3% and 27.5% of the total basal area (Dai et al., 2011).
173 Both selected species for flux measurement are mature with a height of ~ 20 m. The flux bag
174 was installed at the height of 15 m agl. Foliar GEM fluxes over *Fraxinus mandshurica* and
175 *Pinus Koraiensis* were continuously measured during 16-17 and 17-18 Jul 2013, respectively,
176 and 24-h continuous flux data were obtained for each species. Mean blank of flux chamber
177 measured before and after the field experiment was $-0.02 \pm 0.04 \text{ ng m}^{-2} \text{ h}^{-1}$ ($n=24$), which was
178 indistinguishable from zero and not used to calibrate the measured fluxes.

179 **2.4 Isotopic Composition of Atmospheric GEM**

180 From 8 to 18 July 2013, GEM samples were collected at 4 m agl at the study site for Hg
181 isotopes analysis using a chlorine-impregnated activated carbon (CLC) trap (Fu et al., 2014).
182 Atmospheric GEM was collected daily (24-h sampling duration) at a flow rate of 10 LPM. CLC
183 traps collect GEM at $>95\%$ efficiency at the given sampling flow rate (Fu et al., 2014). To
184 remove air particles, a 47-mm diameter Teflon filter (pore size $0.2 \mu\text{m}$) was installed at the inlet
185 of CLC trap. The CLC trap was kept warm ($50 - 70 \text{ }^\circ\text{C}$) during sampling using a silicone rubber
186 heating pads (RadioSpares) to prevent water condensation. The sampling flow rate of CLC
187 traps was regulated via a gas flow meter installed at the outlet of the vacuum pump, and the
188 total sampling volumes of the CLC traps were recorded using a gas meter, calibrated to
189 standard volumes under a standard pressure of 1013 hPa and a standard temperature of 273.14
190 K using a Bios Defender.

191 After the completion of field sampling, CLC traps were sealed with silicone stoppers and
192 three successive polyethylene bags and stored in a clean environment until pre-concentration
193 into trap solutions for Hg isotope analysis. GEM collected by CLC traps were preconcentrated



194 into reverse aqua regia solution (v/v, 2HNO₃/1HCl) in the laboratory using a double-stage
 195 combustion protocol for Hg isotope analysis (Biswas et al., 2008; Sun et al., 2013; Fu et al.,
 196 2014). Hg isotope ratios were determined by Nu-Plasma MC-ICP-MS following a previously
 197 established method (Yin et al., 2013). Hg isotopic composition is reported in delta notation (δ)
 198 in per mil referenced to the bracketed NIST 3133 Hg standard (Blum and Bergquist, 2007):

$$199 \quad \delta^{xxx}Hg = \left(\frac{\left(\frac{xxxHg}{^{198}Hg} \right)_{sample}}{\left(\frac{xxxHg}{^{198}Hg} \right)_{SRM3133}} - 1 \right) \times 1000\text{‰} \quad (2)$$

200 Mass independent fractionation (MIF) values are expressed by “capital delta (Δ)” notation
 201 (‰), which is the difference between the measured values of δ¹⁹⁹Hg, δ²⁰⁰Hg, δ²⁰¹Hg and those
 202 predicted from δ²⁰²Hg using the kinetic MDF law (Blum and Bergquist, 2007):

$$203 \quad \Delta^{199}Hg (\text{‰}) = \delta^{199}Hg - (0.252 \times \delta^{202}Hg) \quad (3)$$

$$204 \quad \Delta^{200}Hg (\text{‰}) = \delta^{200}Hg - (0.502 \times \delta^{202}Hg) \quad (4)$$

$$205 \quad \Delta^{201}Hg (\text{‰}) = \delta^{201}Hg - (0.752 \times \delta^{202}Hg) \quad (5)$$

206 The analytical uncertainty of isotopic analysis was obtained by repeated analysis of the
 207 UM-Almaden standard. The overall mean values of δ²⁰²Hg and Δ¹⁹⁹Hg for all the UM-Almaden
 208 standards were -0.57 ± 0.09 ‰ and -0.03 ± 0.04 ‰ (2SD, n = 12), respectively, consistent with
 209 previously reported values (Blum and Bergquist, 2007). In the present study, the analytical
 210 uncertainty of CV-MC-ICPMS isotope analysis is the 2SD uncertainty of the UM-Almaden
 211 standard, unless the 2SD uncertainty on repeated analysis of the same sample over different
 212 analytical sessions is larger.

213 3 Results and discussion

214 3.1 Characteristics of depletion events at Mt. Changbai

215 From Oct 2008 to Dec 2015, we observed 52 depletion events with dips of GEM
 216 concentrations <0.5 ng m⁻³. These depletions occurred predominantly from May to September
 217 (Figure 1). GEM concentrations during a typical depletion event decreased rapidly from ~1.50
 218 ng m⁻³ around noon to <0.5 ng m⁻³ at night, corresponding to >65% loss of GEM. Figure 2 shows



219 the representative depletion events in summer of 2010 and 2013. Strong depletion of GEM
220 consistently occurred at night. During the 7 - 13 July, 2010 period, a nearly complete depletion
221 occurred with GEM concentrations decreasing from 1.6 - 2.0 ng m⁻³ at noon to nearly zero at
222 night (removal of GEM averaged 1.83 ± 0.35 ng m⁻³ (n = 7)). The daytime peak GEM
223 concentrations for the depletion events during 9 - 23 Jul, 2013 ranged from 1.50 - 2.31 ng m⁻³,
224 and the lowest GEM concentrations at night were 0.35 - 0.99 ng m⁻³, yielding an averaged
225 removal of GEM of 1.08 ± 0.23 ng m⁻³ (n = 12).

226 GOM concentrations during the nighttime atmospheric GEM depletion events (n = 10,
227 defined as nighttime dips in GEM concentrations <0.5 ng m⁻³) from Jul 2013 to Jul 2014 were
228 typically low (< 3 pg m⁻³ with a mean value of 0.8 pg m⁻³). This is in contrast to previously
229 characterized GEM depletions in the polar and sub-polar regions, marine boundary layer and
230 free troposphere where depletions of GEM were accompanied by strong GOM enhancements
231 (up to 195 - 1200 pg m⁻³) (Lindberg et al., 2002; Swartzendruber et al., 2006; Sheu et al.,
232 2010; Obrist et al., 2011; Lyman and Jaffe, 2012). Wind speed was low (mean of 0.1 m s⁻¹ during
233 7 - 13 Jul, 2010 and 0.4 m s⁻¹ during 9 - 23 Jul, 2013) during the nighttime depletion events at
234 the study site (Figure 2). Shallow nocturnal boundary layer (NBL, see text in the SI) was
235 frequently developed when the depletion occurred with a mean height of 146 m (74 - 200 m)
236 during 7 - 13 Jul, 2010 and 209 m (57 - 300 m) during 9-23, Jul, 2013 (Figure 2). The low winds
237 and shallow NBL limited the transport of air masses at the sampling site and facilitated a
238 continuous depletion of GEM in the presence of vegetative uptake of GEM (more details in
239 sections below). During daytime, the surface wind speed and NBL depth increase due to solar
240 heating (Talbot et al., 2005), enabling the downward transport of GEM from upper air, resulting
241 in the increasing GEM concentrations.

242 Summer nighttime depletion of GEM has also been observed at forest sites in North
243 America (e.g., St. Anicet Maple forest station in Canada, and Piney Reservoir, Huntington
244 Wildlife, Thompson Farm, Kejimikujik National Park, and Stilwell in AMNet, USA) (Mao et al.,
245 2008; Poissant et al., 2008; Lan et al., 2012). Such depletion of GEM in forest ecosystems is



246 likely a widespread phenomenon globally. The depletion at forest sites was different from the
247 atmospheric mercury depletion events (AMDEs) elsewhere. For instance, the AMDEs at Cape
248 Point, coast of South Africa and Dead Sea, Israel are mostly observed during daytime (Brunke et
249 al., 2010; Obrist et al., 2011). The AMDEs in the Polar Regions occur exclusively during Polar
250 sunrise in spring and do not exhibit a well-defined diurnal pattern (Schroeder et al.,
251 1998; Ebinghaus et al., 2002; Lindberg et al., 2002).

252 **3.2 Vertical gradient of GEM observed at Mt. Changbai**

253 A clear vertical gradient of GEM concentrations was observed at the study site, with
254 increasing GEM concentrations with respect to sampling altitude (Figure 3). The average
255 difference in GEM concentrations between 45 m and 1 m (all in agl, $\Delta\text{GEM}_{45-1\text{m}}$) was $0.22 \pm$
256 0.15 ng m^{-3} ($n = 330$), $\sim 20\%$ of the mean GEM concentration at 45 m agl. Average differences in
257 GEM concentrations between 45 m and 24 m ($\Delta\text{GEM}_{45-24\text{m}}$), between 24 m and 10 m
258 ($\Delta\text{GEM}_{24-10\text{m}}$), and between 10 m and 1 m ($\Delta\text{GEM}_{10-1\text{m}}$) were 0.11 ± 0.10 , 0.05 ± 0.09 , and 0.06
259 $\pm 0.11 \text{ ng m}^{-3}$ ($n = 330$), respectively. The observed gradient suggested that the forest at the study
260 site is a net sink for atmospheric GEM, in contrast to the vertical GEM gradients observed in a
261 mature hardwood forest (between 30 and 40 m agl) in Walker Branch Watershed, Tennessee,
262 USA during daytime, which showed decreasing GEM concentrations with sampling altitude
263 above the forest canopy (Lindberg et al., 1998). This difference might be caused by the different
264 forest structure and elevated emission flux of GEM from forest soil ($7.5 \text{ ng m}^{-2} \text{ h}^{-1}$ in Walker
265 Branch Watershed versus $2.8 \text{ ng m}^{-2} \text{ h}^{-1}$ at Mt. Changbai, Supplementary Figure S3) in Walker
266 Branch Watershed (Kim et al., 1995; Lindberg et al., 1998).

267 The vertical gradients of GEM at Mt. Changbai showed clear diurnal trends (Figure 3 and
268 Supplementary Figure S4). $\Delta\text{GEM}_{45-24\text{m}}$ and $\Delta\text{GEM}_{24-10\text{m}}$ values were comparably higher at night
269 (mean = 0.13 and 0.08 ng m^{-3}) than those during daytime (0.09 and 0.02 ng m^{-3}). The smaller
270 daytime $\Delta\text{GEM}_{45-24\text{m}}$ and $\Delta\text{GEM}_{24-10\text{m}}$ were a result of weaker dry deposition of GEM to the
271 forest canopy (more discussion later). A strong negative correlation between the $\Delta\text{GEM}_{24-10\text{m}}$ and
272 wind speed ($r^2 = 0.55$, $p < 0.01$) also suggested stronger vertical mixing during daytime inhibited



273 the buildup of GEM gradient. The diurnal trend of $\Delta\text{GEM}_{10-1\text{m}}$ was opposite to $\Delta\text{GEM}_{45-24\text{m}}$ and
274 $\Delta\text{GEM}_{24-10\text{m}}$, with larger values during daytime (mean = 0.09 ng m^{-3}) and lower values at night
275 (mean = 0.04 ng m^{-3}).

276 3.3 Foliage/air exchange flux of GEM

277 Mean foliar GEM fluxes over *Fraxinus Mandschurica* and *Pinus Koraiensis* were $-1.2 \pm$
278 0.6 (-2.2 to $-0.2 \text{ ng m}^{-2} \text{ h}^{-1}$) and $0.0 \pm 0.4 \text{ ng m}^{-2} \text{ h}^{-1}$ (-0.5 to $2.0 \text{ ng m}^{-2} \text{ h}^{-1}$), respectively (Figure
279 4). Mean ambient GEM concentrations during the flux measurements over *Fraxinus*
280 *Mandschurica* and *Pinus Koraiensis* were 1.42 ± 0.23 and $0.93 \pm 0.28 \text{ ng m}^{-3}$, respectively,
281 below the background concentrations of GEM in the Northern Hemisphere ($1.5 - 1.7 \text{ ng m}^{-3}$)
282 (Lindberg et al., 2007). The low GEM deposition flux over *Pinus Koraiensis* was partially
283 attributed to the low ambient GEM concentration that weakened the deposition flux (Hanson et al.,
284 1995; Ericksen and Gustin, 2004). The mean deposition fluxes over *Fraxinus Mandschurica* (0.7
285 $\pm 0.1 \text{ ng m}^{-2} \text{ h}^{-1}$) was much greater than the that over *Pinus Koraiensis* ($0.0 \pm 0.5 \text{ ng m}^{-2} \text{ h}^{-1}$)
286 given the same GEM ($1.0 - 1.4 \text{ ng m}^{-3}$) range (Figure 4), suggesting that GEM deposition flux
287 varies with tree species with deciduous tree species inducing higher deposition compared to
288 evergreen tree species (Millhollen et al., 2006).

289 The observed foliar GEM fluxes over *Fraxinus Mandschurica* and *Pinus Koraiensis* were
290 within the range of reported values (means = -6 to $3.5 \text{ ng m}^{-2} \text{ h}^{-1}$) (Ericksen et al.,
291 2003; Frescholtz and Gustin, 2004; Gustin et al., 2004; Graydon et al., 2006; Poissant et al.,
292 2008; Stamenkovic and Gustin, 2009). A diurnal pattern with higher deposition fluxes at night
293 was observed for both species. The higher deposition flux at night can be attributed to enhanced
294 foliar GEM uptakes. As seen in Figure 4, GEM concentrations in the outlet stream of the flux
295 bag over *Fraxinus Mandschurica* showed a remarkable decline at night compare to daytime
296 (mean $\text{GEM}_{\text{outlet-night}} = 0.28 \pm 0.11 \text{ ng m}^{-3}$, mean $\text{GEM}_{\text{outlet-daytime}} = 0.75 \pm 0.08 \text{ ng m}^{-3}$). It has
297 been suggested that lower O_3 and higher relative humidity (RH) could facilitate the uptake of
298 GEM by foliage (Lindberg and Stratton, 1998; Stamenkovic and Gustin, 2009). O_3 and RH at the
299 study site showed strong diurnal patterns with decreasing O_3 concentrations and increasing RH at



300 night (Supplementary Figure S5), which may explain the higher deposition fluxes of GEM to
301 foliage at night. Both stomatal and non-stomatal uptakes have been suggested to be responsible
302 for the observed foliage-atmosphere GEM exchange (Zhang et al., 2005; Stamenkovic and
303 Gustin, 2009). Stamenkovic and Gustin (2009) found that GEM deposition flux to foliage
304 remained essentially unchanged whether or not stomata are open. This indicates that
305 non-stomatal route plays an important role in the uptake of GEM by foliage, consistent with the
306 observations in this study. Foliar exchange of GEM is bi-directional with foliage emitting GEM
307 at global background air GEM concentrations (Hanson et al., 1995; Ericksen and Gustin,
308 2004; Graydon et al., 2006). With the GEM concentrations in the range of 0.41 – 1.82 ng m⁻³
309 during this study, however, net deposition was observed except for *Pinus Koraiensis* during
310 daytime when stomata are open. Net emission of GEM from *Pinus Koraiensis* during daytime
311 could be attributed to the enhanced photochemical reduction and re-emission of previously
312 deposited Hg (GEM, GOM and PBM), Hg in dew water and transpiration stream as well as
313 transpiration of Hg⁰ in soil pores (Bishop et al., 1998; Lindberg et al., 1998; Ericksen and Gustin,
314 2004; Stamenkovic and Gustin, 2009).

315 The observed foliar GEM fluxes over *Fraxinus Mandschurica* were negatively correlated
316 with the GEM concentrations in the inlet air (Figure 5A), yielding a compensation point of 0.52
317 ng m⁻³ during daytime and 0.47 ng m⁻³ during nighttime, respectively. No correlation between
318 foliar GEM fluxes and ambient GEM concentrations was observed for *Pinus Koraiensis* during
319 daytime. However, a negative correlation was observed at night when ambient GEM
320 concentrations were higher than 0.98 ng m⁻³ (Figure 5B), which was likely the compensation
321 point for *Pinus Koraiensis* during nighttime. These observed compensation points were
322 comparatively lower than the values (2 - 3 ng m⁻³) measured in laboratory studies (Ericksen and
323 Gustin, 2004; Graydon et al., 2006), but consistent with the field observation at St. Anicet Maple
324 forest, Canada (0.53 ng m⁻³) (Poissant et al., 2008). For *Pinus Koraiensis*, the observed foliar
325 GEM fluxes were not significantly different from zero (mean = -0.1 ± 0.1 ng m⁻² h⁻¹) at GEM
326 concentrations lower than the compensation point (0.98 ng m⁻³). A similar conclusion cannot be



327 reached for *Fraxinus Mandschurica* because the ambient GEM concentrations were higher than
328 the respective compensation points during the entire campaign (Figure 5A). This finding is
329 different from previous results that showed net GEM emissions from foliage at ambient GEM
330 concentrations below the compensation points (Hanson et al., 1995; Graydon et al.,
331 2006; Poissant et al., 2008). Based on the field findings, it is likely that the uptake and emission
332 of GEM over the foliage of *Pinus Koraiensis* reached equilibrium during nighttime when the
333 ambient GEM concentrations were below the compensation point.

334 The total deposition flux of GEM to forest canopy at Mt. Changbai was estimated using
335 Equation (6):

$$336 \quad F = LAI \times \sum_i^n (F_i \times A_i) \quad (6)$$

337 where F is the total deposition flux of GEM in $\text{ng m}^{-2} \text{h}^{-1}$, LAI is the mean leaf area index
338 (dimensionless), F_i is the foliar GEM flux of a tree species (i) in $\text{ng m}^{-2} \text{h}^{-1}$, and A_i is the relative
339 basal area of a tree species (i) in percentile (Dai et al., 2011). In this study, it is assumed that the
340 measured mean foliar GEM fluxes over *Fraxinus Mandschurica* and *Pinus Koraiensis* are
341 representative of deciduous tree species and evergreen tree species, respectively. The measured
342 mean LAI at Mt. Changbai during leaf-growing season was 5.4.

343 The total deposition fluxes of GEM to forest canopy at Mt. Changbai during nighttime and
344 daytime are estimated to be 7.3 (V_d of 0.14 cm s^{-1}) and $2.5 \text{ ng m}^{-2} \text{h}^{-1}$ (V_d of 0.04 cm s^{-1}). We
345 acknowledge that, due to the relatively short field sampling periods for the two selected tree
346 species and the fact that foliar GEM flux may vary with tree species, GEM concentrations and
347 other environmental variables, our estimates may have large uncertainties. Nevertheless, the
348 estimates are generally consistent with the measured deposition flux using Hg accumulated in
349 foliage over time. The mean mass-weighted Hg concentration in litter samples at the study site
350 was $43.0 \pm 29.5 \text{ ng g}^{-1}$ (Supplementary Table S1). With the annual litterfall of 486 g m^{-2} at the
351 site (Zhou et al., 2014), the Hg deposition flux in litterfall was $20.9 \pm 14.3 \text{ } \mu\text{g m}^{-2} \text{yr}^{-1}$. Assuming
352 that the plant foliage had a constant uptake rate of Hg in the leaf-growing season (from May to
353 September), the hourly deposition flux of Hg that end up being contained in litterfall would be



354 $5.7 \text{ ng m}^{-2} \text{ h}^{-1}$, comparable to the GEM deposition flux calculated from flux bag observations
355 (daily mean: $4.9 \text{ ng m}^{-2} \text{ h}^{-1}$).

356 **3.4 Mechanisms for the observed GEM depletion**

357 Oxidation of GEM by reactive halogens and O_3 has been proposed to be an important
358 mechanism for GEM depletions observed elsewhere as evidenced by the elevated GOM
359 concentrations (up to $500 - 1200 \text{ pg m}^{-3}$) associated with the GEM depletion events and an
360 inverse correlation between GOM and GEM concentrations (Lindberg et al., 2002; Obrist et al.,
361 2011; Lyman and Jaffe, 2012). Based on modeling assessments, the nighttime loss of GEM in
362 forest areas has been suggested to be caused by dry deposition and chemical oxidation (by ozone,
363 OH and NO_3) (Mao et al., 2008). However, the GOM concentrations observed during typical
364 nighttime GEM depletion events at Mt. Changbai were extremely low ($< 3 \text{ pg m}^{-3}$ with a mean
365 value of 0.8 pg m^{-3}), similar to those observed at other forest sites (Piney Reservoir, Huntington
366 Wildlife, Thompson Farm, Kejimikujik National Park, and Stilwell) in North America (means =
367 $0.5 - 4 \text{ pg m}^{-3}$ at summertime night) (Lan et al., 2012). In addition, concentrations of many
368 atmospheric oxidants (e.g., O_3 , OH, NO_3 , BrO) at global forest sites were low (Spivakovsky et
369 al., 2000; Yang et al., 2005; Rinne et al., 2012; Hens et al., 2014), which does not support
370 significant conversion of GEM to GOM. Given the environmental condition at Mt. Changbai,
371 the dry deposition flux of GOM was estimated to be $0.034 \text{ ng m}^{-2} \text{ h}^{-1}$, using the mean nighttime
372 GOM concentration (0.8 pg m^{-3}) measured during the GEM depletion events and reported V_d of
373 GOM (0.1 to 5.9 cm s^{-1} with a mean of 1.2 cm s^{-1}) to forest canopy (Lindberg and Stratton,
374 1998; Rea et al., 2000; Zhang et al., 2012). Even with a correction factor of 3 to account for the
375 potential under-estimation of GOM concentration by the Tekran® speciation system (Gustin et
376 al., 2013; Huang et al., 2013; Gustin et al., 2015), the deposition flux contributed by GOM is 0.1
377 $\text{ng m}^{-2} \text{ h}^{-1}$. Assuming that all GOM was formed through in situ oxidation of GEM, the chemical
378 pathway would contribute to merely 1.4% of the measured deposition flux of GEM to forest
379 canopy during the nighttime depletion events.



380 Measurements of GEM isotopic composition also provided insight into the mechanisms
381 responsible for the GEM depletion at Mt. Changbai. $\delta^{202}\text{Hg}$, $\Delta^{199}\text{Hg}$, and $\Delta^{200}\text{Hg}$ of the daily
382 GEM samples from 8 to 18 Jul 2013 were -0.34 to 0.91‰, -0.11 to -0.04‰ and -0.06 to 0.01‰,
383 respectively (n=10, Figure 6, Supplementary Table S2). These are consistent with the
384 observations in the Great Lakes region, Barrow, Alaska, Pensacola, FL and Wisconsin forest in
385 USA ($\delta^{202}\text{Hg}_{\text{GEM}} = -0.12$ to 1.43‰, $\Delta^{199}\text{Hg}_{\text{GEM}} = -0.31$ to -0.01‰, $\Delta^{200}\text{Hg} = -0.11$ to 0.1‰)
386 (Gratz et al., 2010; Sherman et al., 2010; Demers et al., 2013; Demers et al., 2015). A large
387 positive $\delta^{202}\text{Hg}_{\text{GEM}}$ shift was associated with strong GEM depletions; whereas $\Delta^{199}\text{Hg}_{\text{GEM}}$ and
388 $\Delta^{200}\text{Hg}_{\text{GEM}}$ remained unchanged. The $\delta^{202}\text{Hg}_{\text{GEM}}$ was up to 0.91‰ during the most pronounced
389 depletion event (on 13 Jul 2013, daily mean GEM of 0.91 ng m⁻³), 1.05‰ higher than the values
390 at the beginning and end of the sampling period (on 9 and 17 Jul 2013, mean GEM = 1.57 - 1.60
391 ng m⁻³, mean $\delta^{202}\text{Hg}_{\text{GEM}} = -0.14$ ‰). The $\delta^{202}\text{Hg}_{\text{GEM}}$ values were anti-correlated with GEM
392 concentrations ($r^2 = 0.58$, $p < 0.01$), whereas no clear relationship can be established between
393 $\Delta^{199}\text{Hg}$, and $\Delta^{200}\text{Hg}_{\text{GEM}}$ values and atmospheric GEM concentrations (p values for both > 0.05).
394 The lower $\delta^{202}\text{Hg}_{\text{GEM}}$ values at the beginning and end of the sampling period were likely
395 representative of the regional background $\delta^{202}\text{Hg}_{\text{GEM}}$ signatures as the GEM concentrations of
396 the two samples were close to the long-term GEM mean concentration at Mt. Changbai, whereas
397 the positive $\delta^{202}\text{Hg}_{\text{GEM}}$ shifts during 11 - 15 Jul 2013 were most likely due to the uptake of GEM
398 by forest foliage which has been known to induce mass dependent fractionation (MDF, $\delta^{202}\text{Hg}$
399 signature) and negligible MIF ($\Delta^{199}\text{Hg}$, $\Delta^{200}\text{Hg}$ signatures) of Hg isotopes (Demers et al.,
400 2013; Enrico et al., 2016). MDF and MIF of Hg isotopes caused by GEM oxidation have not
401 been well characterized. Studies observed both significant MDF and MIF of Hg isotopes during
402 aqueous- and gas-phase chemical oxidation of elemental Hg (Stathopoulos, 2014; Sun et al.,
403 2016). Our study at Pic du Midi, France (2877 m above sea level) also observed clear shifts of
404 $\delta^{202}\text{Hg}_{\text{GEM}}$ and $\Delta^{199}\text{Hg}_{\text{GEM}}$ during oxidation of GEM to GOM, indicating both MDF and MIF
405 could occur during ‘net oxidation’ of GEM in the ambient air (Sonke *et al.*, manuscript under



406 preparation). Therefore, we conclude foliar uptake of GEM played a predominant role in the
407 GEM depletion at Mt. Changbai.

408 To answer the question whether or not GEM dry deposition to forest canopy alone can
409 explain the GEM depletion at Mt. Changbai, the forced change of GEM concentrations by
410 canopy uptake at the sampling height of 24 m under a typical NBL height of 100 m was
411 simulated using a box model (see text in the SI). The box model results suggest that complete
412 GEM depletions can be achieved by canopy uptake alone in the presence of shallow NBL and
413 low vertical turbulent diffusivity (Figure 7). With a dry deposition GEM flux of $7.3 \text{ ng m}^{-2} \text{ h}^{-1}$
414 and turbulent diffusivity of $0.1 - 1.0 \text{ cm s}^{-1}$ at night (Figure S6), the model predicted that GEM
415 concentrations can be decreased to nearly 0 ng m^{-3} (Figure 7). Depletion cannot occur during
416 daytime mainly due to the low dry deposition flux ($\sim 2.5 \text{ ng m}^{-2} \text{ h}^{-1}$), high vertical turbulent
417 diffusivity ($1 - 100 \text{ cm s}^{-1}$) and absence of shallow NBL (Figure 7). The GEM depletion event at
418 Mt. Changbai showed a seasonal trend with the depletion occurring more frequently and
419 pronouncedly during leaf-growing season from July to August. This can be attributed to: (1)
420 seasonal LAI changes (Figure S7.A), (2) lower wind speed from July to August (Figure S7.B),
421 and (3) the wind direction that inhibited the transport of polluted air from anthropogenic source
422 regions (90° - 202° , natural preserve areas without significant local and regional sources) during
423 leaf-growing season (Figure S7.C). LAI is the highest in July and August (~ 5.4) compared to the
424 non-growing season (< 2 , Figure S8.A) (Shi et al., 2008). Higher LAI values indicate higher dry
425 deposition fluxes of GEM to forest canopy. The low wind speed facilitated the buildup of
426 shallow NBL.

427 **4 Conclusions and implications for the global atmospheric Hg cycling**

428 Depletions of atmospheric GEM were consistently observed during leaf-growing season in
429 Mt. Changbai forest, Northeast China. The depletions occurred exclusively at night in the
430 absence of GOM enrichments. This is in contrast to previously characterized GEM depletions in
431 the polar and sub-polar regions, marine boundary layer and free troposphere where depletions of



432 GEM were mainly caused by fast chemical oxidation of GEM to GOM followed by deposition.
433 The measurements of GEM vertical gradients, foliar GEM fluxes, atmospheric speciated Hg and
434 ambient GEM isotope compositions suggest foliar uptake of GEM played a predominant role in
435 the GEM depletion at Mt. Changbai.

436 Forests cover ~30% (~40 million km²) of the Earth's land surface. There is a need to
437 quantitatively assess the role of global forest in global Hg cycling. Tables S3, S4, and S5
438 summarize the published data of litterfall fluxes at 68 forest sites, throughfall fluxes at 23 forest
439 sites, and emissions from forest floors at 31 forest sites in North America, Europe, Asia, and
440 South America. For the regions (Africa and Oceania) that lack observational data, it is assumed
441 that the median values of the published data are representative. There has not been reliable
442 data on Hg emission from forest canopies via evapotranspiration. We therefore use the observed
443 xylem Hg concentrations and total evapotranspiration from the global forests to estimate Hg
444 emissions from this sector (Bishop et al., 1998; Baldocchi and Ryu, 2011).

445 Using a mass balance approach, we estimated that global inputs of Hg via litterfall and
446 throughfall were 1,232 and 1,338 Mg yr⁻¹, respectively. Hg emissions via the evasion from soil
447 and plant evapotranspiration were 381 and 260 Mgyr⁻¹, respectively. Combining the source and
448 sink terms, the global forest ecosystem represents a net sink of ~1,930 Mg yr⁻¹ of atmospheric
449 Hg. The value is much larger than the estimate of Hg uptake by forest above-ground biomass
450 (Obrist, 2007). The estimate by Obrist (2007) did not include deposition flux by throughfall; and
451 the Hg concentration in biomass used in the study was 2 - 10 times lower than the measured Hg
452 contents in North America, Europe, China and South America (Lindberg et al., 2007; Obrist,
453 2007; Risch et al., 2012; Teixeira et al., 2012; Fu et al., 2015). Our estimate is comparable to the
454 upper limit of atmospheric Hg deposition to terrestrial ecosystem predicted by modeling studies
455 (800-1900 Mg) (Mason and Sheu, 2002; Holmes et al., 2010; Driscoll et al., 2013). This implies
456 that forest ecosystem may be the largest sink of atmospheric Hg in the terrestrial ecosystems,
457 whereas other terrestrial ecosystems may represent net sources.

458 **Supporting Information:**



459 Descriptions of the simulation of NBL, turbulent diffusivity and the box model are shown
460 in supplementary text. The location of the Mt. Changbai forest, GEM concentrations at 4 m agl
461 in a small clearing plot and 24 m and 45 m agl, diurnal trends in vertical GEM gradient, soil/air
462 GEM flux, diurnal variations of meteorological parameters, turbulent diffusivity and seasonal
463 variations in LAI, wind speed and wind direction at Mt. Changbai forest are shown in Figure
464 S1-S7. Litterfall Hg concentrations and litter mass at Mt. Changbai forest, isotopic composition
465 of atmospheric GEM as well as compiled litterfall and throughfall Hg deposition fluxes, and
466 forest soil/air GEM fluxes over the global forests are shown in Table S1-S5.

467 **Acknowledgements.** This work was funded by the National “973” Program of China (2013CB430003); the
468 National Science Foundation of China (41473025, 41273145, and 41428301) and Guangzhou Science and
469 Technology Projects (2014J4100089). We acknowledge Alexandra Steffen from Environment Canada for
470 offering the sampling instruments, Open Research Station of Changbai Mountain Forest Ecosystems for the
471 meteorological parameters and field sampling, CAS, and Hao Xu for assistance with field sampling. The data
472 used are listed in the references, tables, and supplements.

473 **References**

- 474 Amos, H. M., Jacob, D. J., Streets, D. G., and Sunderland, E. M.: Legacy impacts of all-time anthropogenic
475 emissions on the global mercury cycle, *Global Biogeochem Cy*, 27, 410-421, 10.1002/gbc.20040, 2013.
- 476 Baldocchi, D., and Ryu, Y.: A Synthesis of Forest Evaporation Fluxes – from Days to Years – as Measured with
477 Eddy Covariance, in: *Forest Hydrology and Biogeochemistry*, edited by: Levia, D. F., Carlyle-Moses, D., and
478 Tanaka, T., Ecological Studies, Springer Netherlands, 101-116, 2011.
- 479 Bash, J. O., and Miller, D. R.: Growing season total gaseous mercury (TGM) flux measurements over an *Acer*
480 *rubrum* L. stand, *Atmos Environ*, 43, 5953-5961, DOI 10.1016/j.atmosenv.2009.08.008, 2009.
- 481 Bishop, K. H., Lee, Y. H., Munthe, J., and Dambrine, E.: Xylem sap as a pathway for total mercury and
482 methylmercury transport from soils to tree canopy in the boreal forest, *Biogeochemistry*, 40, 101-113, 1998.
- 483 Biswas, A., Blum, J. D., Bergquist, B. A., Keeler, G. J., and Xie, Z. Q.: Natural mercury isotope variation in coal
484 deposits and organic soils, *Environmental Science & Technology*, 42, 8303-8309, Doi 10.1021/Es801444b, 2008.
- 485 Blum, J. D., and Bergquist, B. A.: Reporting of variations in the natural isotopic composition of mercury, *Anal*
486 *Bioanal Chem*, 388, 353-359, DOI 10.1007/s00216-007-1236-9, 2007.
- 487 Brunke, E. G., Labuschagne, C., Ebinghaus, R., Kock, H. H., and Slemr, F.: Gaseous elemental mercury depletion
488 events observed at Cape Point during 2007-2008, *Atmos Chem Phys*, 10, 1121-1131, 2010.
- 489 Dai, L. M., Qi, L., Wang, Q. W., Su, D. K., Yu, D. P., Wang, Y., Ye, Y. J., Jiang, S. W., and Zhao, W.: Changes in
490 forest structure and composition on Changbai Mountain in Northeast China, *Ann Forest Sci*, 68, 889-897, DOI
491 10.1007/s13595-011-0095-x, 2011.



- 492 Demers, J. D., Blum, J. D., and Zak, D. R.: Mercury isotopes in a forested ecosystem: Implications for air-surface
493 exchange dynamics and the global mercury cycle, *Global Biogeochem Cy*, 27, 222-238, Doi 10.1002/Gbc.20021,
494 2013.
- 495 Demers, J. D., Sherman, L. S., Blum, J. D., Marsik, F. J., and Dvonch, J. T.: Coupling atmospheric mercury isotope
496 ratios and meteorology to identify sources of mercury impacting a coastal urban-industrial region near Pensacola,
497 Florida, USA, *Global Biogeochem Cy*, 29, 1689-1705, 2015.
- 498 Driscoll, C. T., Mason, R. P., Chan, H. M., Jacob, D. J., and Pirrone, N.: Mercury as a Global Pollutant: Sources,
499 Pathways, and Effects, *Environmental Science & Technology*, 47, 4967-4983, Doi 10.1021/Es305071v, 2013.
- 500 Ebinghaus, R., Kock, H. H., Temme, C., Einax, J. W., Lowe, A. G., Richter, A., Burrows, J. P., and Schroeder, W. H.:
501 Antarctic springtime depletion of atmospheric mercury, *Environmental Science & Technology*, 36, 1238-1244, Doi
502 10.1021/Es015710z, 2002.
- 503 Enrico, M., Le Roux, G., Maruszczak, N., Heimbürger, L. E., Claustres, A., Fu, X. W., Sun, R. Y., and Sonke, J. E.:
504 Atmospheric Mercury Transfer to Peat Bogs Dominated by Gaseous Elemental Mercury Dry Deposition,
505 *Environmental Science & Technology*, 50, 2405-2412, 10.1021/acs.est.5b06058, 2016.
- 506 Ericksen, J. A., Gustin, M. S., Schorran, D. E., Johnson, D. W., Lindberg, S. E., and Coleman, J. S.: Accumulation
507 of atmospheric mercury in forest foliage, *Atmos Environ*, 37, 1613-1622, 2003.
- 508 Ericksen, J. A., and Gustin, M. S.: Foliar exchange of mercury as a function of soil and air mercury concentrations,
509 *Sci Total Environ*, 324, 271-279, DOI 10.1016/j.scitotenv.200310.034, 2004.
- 510 Fain, X., Obrist, D., Hallar, A. G., Mccubbin, I., and Rahn, T.: High levels of reactive gaseous mercury observed at a
511 high elevation research laboratory in the Rocky Mountains, *Atmos Chem Phys*, 9, 8049-8060, 2009.
- 512 Finkelstein, P. L., Ellestad, T. G., Clarke, J. F., Meyers, T. P., Schwede, D. B., Hebert, E. O., and Neal, J. A.: Ozone
513 and sulfur dioxide dry deposition to forests: Observations and model evaluation, *J Geophys Res-Atmos*, 105,
514 15365-15377, Doi 10.1029/2000jd900185, 2000.
- 515 Frescholtz, T. F., and Gustin, M. S.: Soil and foliar mercury emission as a function of soil concentration, *Water Air
516 Soil Poll*, 155, 223-237, Doi 10.1023/B:Wate.0000026530.85954.3f, 2004.
- 517 Fu, X. W., Sonke, J. E., and Heimbürger, L.-E.: Collection of atmospheric gaseous mercury for stable isotope
518 analysis using iodine- and chlorine-impregnated activated carbon traps, *J Anal Atom Spectrom*, In review, 2014.
- 519 Fu, X. W., Zhang, H., Yu, B., Wang, X., Lin, C. J., and Feng, X. B.: Observations of atmospheric mercury in China:
520 a critical review, *Atmos. Chem. Phys.*, 15, 9455-9476, 10.5194/acp-15-9455-2015, 2015.
- 521 Fu, X. W., Maruszczak, N., Heimbürger, L. E., Sauvage, B., Gheusi, F., Prestbo, E. M., and Sonke, J. E.: Atmospheric
522 mercury speciation dynamics at the high-altitude Pic du Midi Observatory, southern France, *Atmos. Chem. Phys.*,
523 16, 5623-5639, 10.5194/acp-16-5623-2016, 2016.
- 524 Gratz, L. E., Keeler, G. J., Blum, J. D., and Sherman, L. S.: Isotopic composition and fractionation of mercury in
525 Great Lakes precipitation and ambient air, *Environmental Science & Technology*, 44, 7764-7770, Doi
526 10.1021/Es100383w, 2010.
- 527 Graydon, J. A., St Louis, V. L., Lindberg, S. E., Hintelmann, H., and Krabbenhoft, D. P.: Investigation of mercury
528 exchange between forest canopy vegetation and the atmosphere using a new dynamic chamber, *Environmental
529 Science & Technology*, 40, 4680-4688, Doi 10.1021/Es0604616, 2006.
- 530 Gustin, M., and Jaffe, D.: Reducing the Uncertainty in Measurement and Understanding of Mercury in the
531 Atmosphere, *Environmental Science & Technology*, 44, 2222-2227, Doi 10.1021/Es902736k, 2010.
- 532 Gustin, M. S., Ericksen, J. A., Schorran, D. E., Johnson, D. W., Lindberg, S. E., and Coleman, J. S.: Application of



- 533 controlled mesocosms for understanding mercury air-soil-plant exchange, *Environmental Science & Technology*, 38,
534 6044-6050, Doi 10.1021/Es0487933, 2004.
- 535 Gustin, M. S., Huang, J. Y., Miller, M. B., Peterson, C., Jaffe, D. A., Ambrose, J., Finley, B. D., Lyman, S. N., Call,
536 K., Talbot, R., Feddersen, D., Mao, H. T., and Lindberg, S. E.: Do we understand what the mercury speciation
537 instruments are actually measuring? Results of RAMIX, *Environmental Science & Technology*, 47, 7295-7306, Doi
538 10.1021/Es3039104, 2013.
- 539 Gustin, M. S., Amos, H. M., Huang, J., Miller, M. B., and Heidecorn, K.: Measuring and modeling mercury in the
540 atmosphere: a critical review, *Atmos Chem Phys*, 15, 5697-5713, DOI 10.5194/acp-15-5697-2015, 2015.
- 541 Hanson, P. J., Lindberg, S. E., Tabberer, T. A., Owens, J. G., and Kim, K. H.: Foliar Exchange of Mercury-Vapor -
542 Evidence for a Compensation Point, *Water Air Soil Poll*, 80, 373-382, Doi 10.1007/Bf01189687, 1995.
- 543 Hens, K., Novelli, A., Martinez, M., Auld, J., Axinte, R., Bohn, B., Fischer, H., Keronen, P., Kubistin, D., Nolscher,
544 A. C., Oswald, R., Paasonen, P., Petaja, T., Regelin, E., Sander, R., Sinha, V., Sipila, M., Taraborrelli, D., Ernest, C.
545 T., Williams, J., Lelieveld, J., and Harder, H.: Observation and modelling of HOx radicals in a boreal forest, *Atmos
546 Chem Phys*, 14, 8723-8747, 10.5194/acp-14-8723-2014, 2014.
- 547 Holmes, C. D., Jacob, D. J., Corbitt, E. S., Mao, J., Yang, X., Talbot, R., and Slemr, F.: Global atmospheric model
548 for mercury including oxidation by bromine atoms, *Atmos Chem Phys*, 10, 12037-12057, DOI
549 10.5194/acp-10-12037-2010, 2010.
- 550 Huang, J. Y., Miller, M. B., Weiss-Penzias, P., and Gustin, M. S.: Comparison of gaseous oxidized Hg measured by
551 KCl-coated denuders, and Nylon and Cation exchange Membranes, *Environmental Science & Technology*, 47,
552 7307-7316, Doi 10.1021/Es4012349, 2013.
- 553 Huang, J. Y., and Gustin, M. S.: Use of Passive Sampling Methods and Models to Understand Sources of Mercury
554 Deposition to High Elevation Sites in the Western United States, *Environmental Science & Technology*, 49, 432-441,
555 10.1021/es502836w, 2015.
- 556 Kim, K. H., Lindberg, S. E., and Meyers, T. P.: Micrometeorological Measurements of Mercury-Vapor Fluxes over
557 Background Forest Soils in Eastern Tennessee, *Atmos Environ*, 29, 267-282, Doi 10.1016/1352-2310(94)00198-T,
558 1995.
- 559 Lan, X., Talbot, R., Castro, M., Perry, K., and Luke, W.: Seasonal and diurnal variations of atmospheric mercury
560 across the US determined from AMNet monitoring data, *Atmos Chem Phys*, 12, 10569-10582, DOI
561 10.5194/acp-12-10569-2012, 2012.
- 562 Landis, M. S., Stevens, R. K., Schaedlich, F., and Prestbo, E. M.: Development and characterization of an annular
563 denuder methodology for the measurement of divalent inorganic reactive gaseous mercury in ambient air,
564 *Environmental Science & Technology*, 36, 3000-3009, Doi 10.1021/Es015887t, 2002.
- 565 Lindberg, S., Bullock, R., Ebinghaus, R., Engstrom, D., Feng, X. B., Fitzgerald, W., Pirrone, N., Prestbo, E., and
566 Seigneur, C.: A synthesis of progress and uncertainties in attributing the sources of mercury in deposition, *Ambio*,
567 36, 19-32, 2007.
- 568 Lindberg, S. E., Hanson, P. J., Meyers, T. P., and Kim, K. H.: Air/surface exchange of mercury vapor over forests -
569 The need for a reassessment of continental biogenic emissions, *Atmos Environ*, 32, 895-908, Doi
570 10.1016/S1352-2310(97)00173-8, 1998.
- 571 Lindberg, S. E., and Stratton, W. J.: Atmospheric mercury speciation: Concentrations and behavior of reactive
572 gaseous mercury in ambient air, *Environmental Science & Technology*, 32, 49-57, Doi 10.1021/Es970546u, 1998.
- 573 Lindberg, S. E., Brooks, S., Lin, C. J., Scott, K. J., Landis, M. S., Stevens, R. K., Goodsite, M., and Richter, A.:



- 574 Dynamic oxidation of gaseous mercury in the Arctic troposphere at polar sunrise, *Environmental Science &*
575 *Technology*, 36, 1245-1256, Doi 10.1021/Es0111941, 2002.
- 576 Lyman, S. N., and Jaffe, D. A.: Formation and fate of oxidized mercury in the upper troposphere and lower
577 stratosphere, *Nat Geosci*, 5, 114-117, Doi 10.1038/Ngeo1353, 2012.
- 578 Mao, H., Talbot, R. W., Sigler, J. M., Sive, B. C., and Hegarty, J. D.: Seasonal and diurnal variations of Hg degrees
579 over New England, *Atmos Chem Phys*, 8, 1403-1421, 2008.
- 580 Mason, R. P., and Sheu, G. R.: Role of the ocean in the global mercury cycle, *Global Biogeochem Cy*, 16, Artn 1093
581 Doi 10.1029/2001gb001440, 2002.
- 582 Millhollen, A. G., Gustin, M. S., and Obrist, D.: Foliar mercury accumulation and exchange for three tree species,
583 *Environmental Science & Technology*, 40, 6001-6006, Doi 10.1021/Es0609194, 2006.
- 584 Munger, J. W., Wofsy, S. C., Bakwin, P. S., Fan, S. M., Goulden, M. L., Daube, B. C., Goldstein, A. H., Moore, K.
585 E., and Fitzjarrald, D. R.: Atmospheric deposition of reactive nitrogen oxides and ozone in a temperate deciduous
586 forest and a subarctic woodland .1. Measurements and mechanisms, *J Geophys Res-Atmos*, 101, 12639-12657, Doi
587 10.1029/96jd00230, 1996.
- 588 Obrist, D.: Atmospheric mercury pollution due to losses of terrestrial carbon pools?, *Biogeochemistry*, 85, 119-123,
589 DOI 10.1007/s10533-007-9108-0, 2007.
- 590 Obrist, D., Tas, E., Peleg, M., Matveev, V., Fain, X., Asaf, D., and Luria, M.: Bromine-induced oxidation of mercury
591 in the mid-latitude atmosphere, *Nat Geosci*, 4, 22-26, Doi 10.1038/Ngeo1018, 2011.
- 592 Pan, Y. D., Birdsey, R. A., Fang, J. Y., Houghton, R., Kauppi, P. E., Kurz, W. A., Phillips, O. L., Shvidenko, A.,
593 Lewis, S. L., Canadell, J. G., Ciais, P., Jackson, R. B., Pacala, S. W., McGuire, A. D., Piao, S. L., Rautiainen, A.,
594 Sitch, S., and Hayes, D.: A Large and Persistent Carbon Sink in the World's Forests, *Science*, 333, 988-993, DOI
595 10.1126/science.1201609, 2011.
- 596 Poissant, L., Pilote, M., Yumvihoze, E., and Lean, D.: Mercury concentrations and foliage/atmosphere fluxes in a
597 maple forest ecosystem in Quebec, Canada, *J Geophys Res-Atmos*, 113, Artn D10307
598 Doi 10.1029/2007jd009510, 2008.
- 599 Rea, A. W., Lindberg, S. E., and Keeler, G. J.: Assessment of dry deposition and foliar leaching of mercury and
600 selected trace elements based on washed foliar and surrogate surfaces, *Environmental Science & Technology*, 34,
601 2418-2425, 10.1021/es991305k, 2000.
- 602 Rinne, J., Markkanen, T., Ruuskanen, T. M., Petaja, T., Keronen, P., Tang, M. J., Crowley, J. N., Rannik, U., and
603 Vesala, T.: Effect of chemical degradation on fluxes of reactive compounds - a study with a stochastic Lagrangian
604 transport model, *Atmos Chem Phys*, 12, 4843-4854, 10.5194/acp-12-4843-2012, 2012.
- 605 Risch, M. R., DeWild, J. F., Krabbenhoft, D. P., Kolka, R. K., and Zhang, L. M.: Litterfall mercury dry deposition in
606 the eastern USA, *Environ Pollut*, 161, 284-290, DOI 10.1016/j.envpol.2011.06.005, 2012.
- 607 Schroeder, W. H., Anlauf, K. G., Barrie, L. A., Lu, J. Y., Steffen, A., Schneeberger, D. R., and Berg, T.: Arctic
608 springtime depletion of mercury, *Nature*, 394, 331-332, Doi 10.1038/28530, 1998.
- 609 Selin, N. E., Jacob, D. J., Park, R. J., Yantosca, R. M., Strode, S., Jaegle, L., and Jaffe, D.: Chemical cycling and
610 deposition of atmospheric mercury: Global constraints from observations, *J Geophys Res-Atmos*, 112, Artn D02308
611 Doi 10.1029/2006jd007450, 2007.
- 612 Selin, N. E., Jacob, D. J., Yantosca, R. M., Strode, S., Jaegle, L., and Sunderland, E. M.: Global 3-D
613 land-ocean-atmosphere model for mercury: Present-day versus preindustrial cycles and anthropogenic enrichment
614 factors for deposition, *Global Biogeochem Cy*, 22, Artn Gb2011



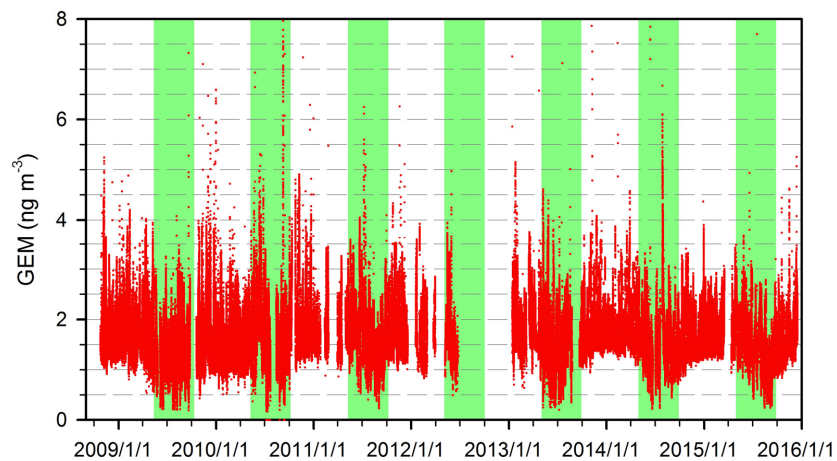
- 615 10.1029/2007gb003040, 2008.
- 616 Shah, V., Jaegle, L., Gratz, L. E., Ambrose, J. L., Jaffè, D. A., Selin, N. E., Song, S., Campos, T. L., Flocke, F. M.,
617 Reeves, M., Stechman, D., Stell, M., Festa, J., Stutz, J., Weinheimer, A. J., Knapp, D. J., Montzka, D. D., Tyndall, G.
618 S., Apel, E. C., Hornbrook, R. S., Hills, A. J., Riemer, D. D., Blake, N. J., Cantrell, C. A., and Mauldin, R. L.:
619 Origin of oxidized mercury in the summertime free troposphere over the southeastern US, *Atmos Chem Phys*, 16,
620 1511-1530, 10.5194/acp-16-1511-2016, 2016.
- 621 Sherman, L. S., Blum, J. D., Johnson, K. P., Keeler, G. J., Barres, J. A., and Douglas, T. A.: Mass-independent
622 fractionation of mercury isotopes in Arctic snow driven by sunlight, *Nat Geosci*, 3, 173-177, Doi 10.1038/Ngeo758,
623 2010.
- 624 Sheu, G. R., Lin, N. H., Wang, J. L., Lee, C. T., Yang, C. F. O., and Wang, S. H.: Temporal distribution and potential
625 sources of atmospheric mercury measured at a high-elevation background station in Taiwan, *Atmos Environ*, 44,
626 2393-2400, DOI 10.1016/j.atmosenv.2010.04.009, 2010.
- 627 Shi, T. T., Guan, D. X., Wang, A. Z., Wu, J. B., Jin, C. J., and Han, S. J.: Comparison of three models to estimate
628 evapotranspiration for a temperate mixed forest, *Hydrol Process*, 22, 3431-3443, 10.1002/hyp.6922, 2008.
- 629 Sommar, J., Zhu, W., Lin, C. J., and Feng, X. B.: Field Approaches to Measure Hg Exchange Between Natural
630 Surfaces and the Atmosphere A Review, *Crit Rev Env Sci Tec*, 43, 1657-1739, Doi 10.1080/10643389.2012.671733,
631 2013.
- 632 Spivakovsky, C. M., Logan, J. A., Montzka, S. A., Balkanski, Y. J., Foreman-Fowler, M., Jones, D. B. A., Horowitz,
633 L. W., Fusco, A. C., Brenninkmeijer, C. A. M., Prather, M. J., Wofsy, S. C., and McElroy, M. B.: Three-dimensional
634 climatological distribution of tropospheric OH: Update and evaluation, *J Geophys Res-Atmos*, 105, 8931-8980, Doi
635 10.1029/1999jd901006, 2000.
- 636 Stamenkovic, J., and Gustin, M. S.: Nonstomatal versus Stomatal Uptake of Atmospheric Mercury, *Environmental*
637 *Science & Technology*, 43, 1367-1372, Doi 10.1021/Es801583a, 2009.
- 638 Stathopoulos, D.: Fractionation of mercury isotopes in an aqueous environment: Chemical Oxidation. , Master
639 Dissertation, Trent University, Peterborough, Ontario, Canada, 2014.
- 640 Steffen, A., Douglas, T., Amyot, M., Ariya, P., Aspö, K., Berg, T., Bottenheim, J., Brooks, S., Cobbett, F., Dastoor,
641 A., Dommergue, A., Ebinghaus, R., Ferrari, C., Gardfeldt, K., Goodsite, M. E., Lean, D., Poulain, A. J., Scherz, C.,
642 Skov, H., Sommar, J., and Temme, C.: A synthesis of atmospheric mercury depletion event chemistry in the
643 atmosphere and snow, *Atmos Chem Phys*, 8, 1445-1482, 2008.
- 644 Strode, S. A., Jaegle, L., Selin, N. E., Jacob, D. J., Park, R. J., Yantosca, R. M., Mason, R. P., and Slemr, F.: Air-sea
645 exchange in the global mercury cycle, *Global Biogeochem Cy*, 21, Artn Gb1017
646 Doi 10.1029/2006gb002766, 2007.
- 647 Sun, G. Y., Sommar, J., Feng, X. B., Lin, C. J., Ge, M. F., Wang, W. G., Fu, X. W., and Shang, L. H.: Mass
648 -dependent and -independent fractionation of mercury isotope during gas-phase oxidation of elemental mercury
649 vapor by atomic Cl and Br, *Environmental Science & Technology*, under revision, 2016.
- 650 Sun, R. Y., Enrico, M., Heimbürger, L. E., Scott, C., and Sonke, J. E.: A double-stage tube furnace-acid-trapping
651 protocol for the pre-concentration of mercury from solid samples for isotopic analysis, *Anal Bioanal Chem*, 405,
652 6771-6781, DOI 10.1007/s00216-013-7152-2, 2013.
- 653 Swartzendruber, P. C., Jaffè, D. A., Prestbo, E. M., Weiss-Penzias, P., Selin, N. E., Park, R., Jacob, D. J., Strode, S.,
654 and Jaegle, L.: Observations of reactive gaseous mercury in the free troposphere at the Mount Bachelor Observatory,
655 *J Geophys Res-Atmos*, 111, D24301, doi 10.1029/2006jd007415, Artn D24302



- 656 Doi 10.1029/2006jd007415, 2006.
- 657 Swartzendruber, P. C., Jaffe, D. A., and Finley, B.: Improved fluorescence peak integration in the Tekran 2537 for
658 applications with sub-optimal sample loadings, *Atmos Environ*, 43, 3648-3651, 10.1016/j.atmosenv.2009.02.063,
659 2009a.
- 660 Swartzendruber, P. C., Jaffe, D. A., and Finley, B.: Development and First Results of an Aircraft-Based, High Time
661 Resolution Technique for Gaseous Elemental and Reactive (Oxidized) Gaseous Mercury, *Environmental Science &*
662 *Technology*, 43, 7484-7489, Doi 10.1021/Es901390t, 2009b.
- 663 Talbot, R., Mao, H. T., and Sive, B.: Diurnal characteristics of surface level O-3 and other important trace gases in
664 New England, *J Geophys Res-Atmos*, 110, 2005.
- 665 Teixeira, D. C., Montezuma, R. C., Oliveira, R. R., and Silva, E. V.: Litterfall mercury deposition in Atlantic forest
666 ecosystem from SE - Brazil, *Environ Pollut*, 164, 11-15, DOI 10.1016/j.envpol.2011.10.032, 2012.
- 667 Timonen, H., Ambrose, J. L., and Jaffe, D. A.: Oxidation of elemental Hg in anthropogenic and marine airmasses,
668 *Atmos. Chem. Phys.*, 13, 2827-2836, 10.5194/acp-13-2827-2013, 2013.
- 669 Yang, X., Cox, R. A., Warwick, N. J., Pyle, J. A., Carver, G. D., O'Connor, F. M., and Savage, N. H.: Tropospheric
670 bromine chemistry and its impacts on ozone: A model study, *J Geophys Res-Atmos*, 110, Artn D23311
671 Doi 10.1029/2005jd006244, 2005.
- 672 Yin, R. S., Feng, X. B., and Meng, B.: Stable Mercury Isotope Variation in Rice Plants (*Oryza sativa* L.) from the
673 Wanshan Mercury Mining District, SW China, *Environmental Science & Technology*, 47, 2238-2245, Doi
674 10.1021/Es304302a, 2013.
- 675 Zhang, H. H., Poissant, L., Xu, X. H., and Pilote, M.: Explorative and innovative dynamic flux bag method
676 development and testing for mercury air-vegetation gas exchange fluxes, *Atmos Environ*, 39, 7481-7493, DOI
677 10.1016/j.atmosenv.2005.07.068, 2005.
- 678 Zhang, L., Blanchard, P., Gay, D. A., Prestbo, E. M., Risch, M. R., Johnson, D., Narayan, J., Zsolway, R., Holsen, T.
679 M., Miller, E. K., Castro, M. S., Graydon, J. A., St Louis, V. L., and Dalziel, J.: Estimation of speciated and total
680 mercury dry deposition at monitoring locations in eastern and central North America, *Atmos Chem Phys*, 12,
681 4327-4340, DOI 10.5194/acp-12-4327-2012, 2012.
- 682 Zhang, L. M., Gong, S. L., Padro, J., and Barrie, L.: A size-segregated particle dry deposition scheme for an
683 atmospheric aerosol module, *Atmos Environ*, 35, 549-560, Doi 10.1016/S1352-2310(00)00326-5, 2001.
- 684 Zhang, L. M., Wright, L. P., and Blanchard, P.: A review of current knowledge concerning dry deposition of
685 atmospheric mercury, *Atmos Environ*, 43, 5853-5864, DOI 10.1016/j.atmosenv.2009.08.019, 2009.
- 686 Zhou, Y., Su, J. Q., Janssens, I. A., Zhou, G. S., and Xiao, C. W.: Fine root and litterfall dynamics of three Korean
687 pine (*Pinus koraiensis*) forests along an altitudinal gradient, *Plant Soil*, 374, 19-32, DOI
688 10.1007/s11104-013-1816-8, 2014.
- 689



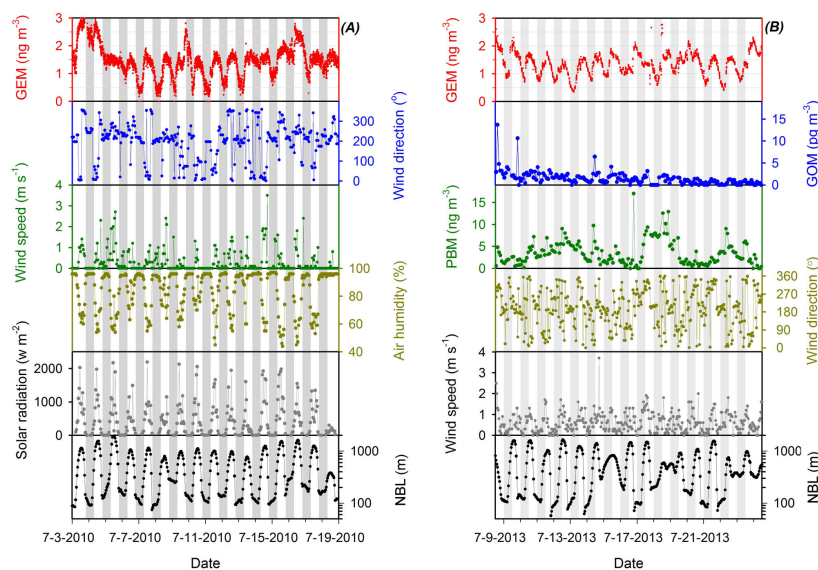
690 Figure 1. Atmospheric 5-min GEM concentrations at Mt. Changbai from Oct 2008 to Dec 2015 (leaf-growing
691 season is marked as the shaded area).



692
693
694



695 Figure 2. Time series of (A) GEM (5-min mean) and meteorological parameters from 3 to 19 July 2010 and (B)
696 speciated atmospheric Hg (GEM, GOM, and PBM) and meteorological parameters 8 to 24 July 2013
697 (nighttime is marked as the shaded area).

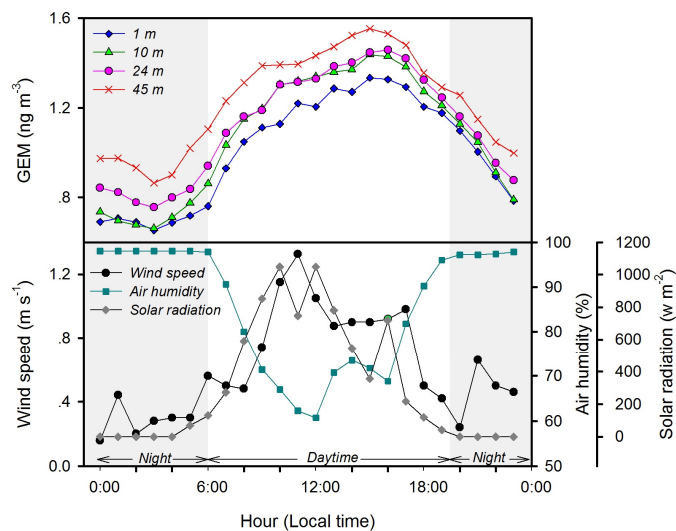


698

699



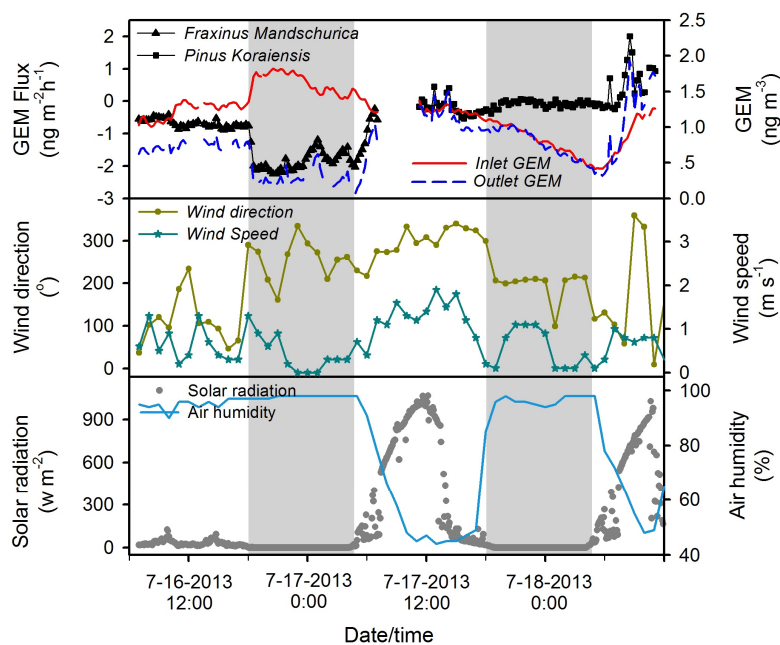
700 Figure 3. Diurnal variations of GEM concentrations at different height and metrological parameters in Mt.
701 Changbai forest from 10 to 15 July 2013 (nighttime is marked as shaded area).



702
703



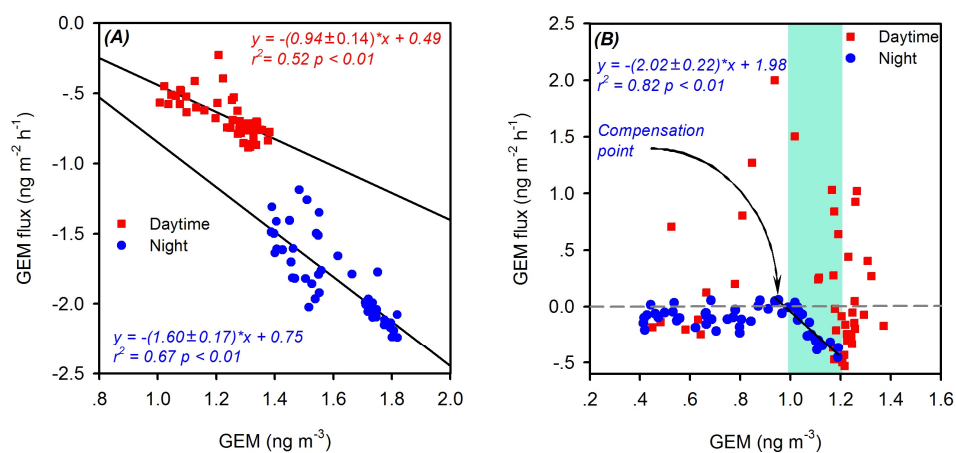
704 Figure 4. Foliar Hg flux over *Fraxinus Mandschurica* and *Pinus Koraiensis*, inlet and outlet GEM
705 concentrations from flux bag and meteorological parameters at Mt. Changbai in July 2013 (nighttime is
706 marked as the shaded area).



707
708
709



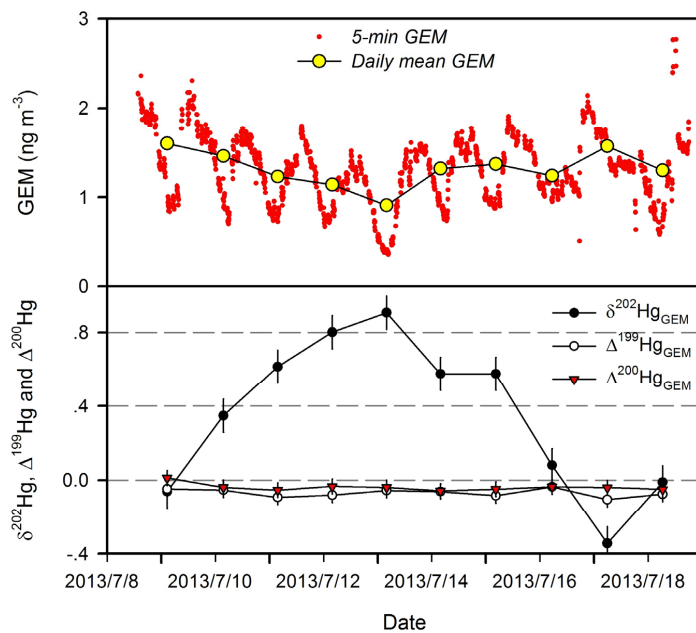
710 Figure 5. Daytime and nighttime correlations between atmospheric GEM concentrations and foliar GEM
711 fluxes over (A) *Fraxinus Mandschurica* and (B) *Pinus Koraiensis*.



712
713
714



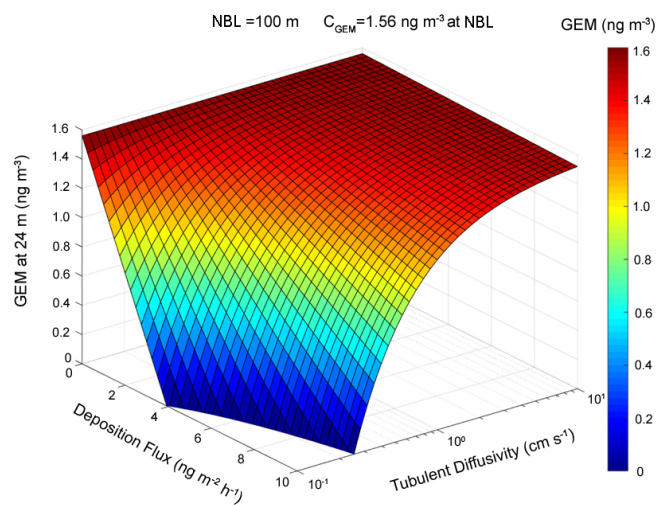
715 Figure 6. Temporal variation in (A) atmospheric GEM concentrations and (B) $\delta^{202}\text{Hg}$, $\Delta^{199}\text{Hg}$ and $\Delta^{200}\text{Hg}$
716 values of daily integrated atmospheric GEM from 9 to 18 July 2013.



717
718
719
720



- 721 Figure 7. Modeling predicted variations of GEM concentration at the height of 24 m agl with dry position
722 fluxes of GEM to forest canopy and vertical turbulent diffusivity under a typical NBL height of 100 m.



723
724