#### Response to Anonymous Referee #3

General comments: This MS conduct an very interesting study of multiple air pollutants, including Hg, SO2, CO2, CO, NOX emissions through the onboard aircraft measurement in the plume downwind a large coal-fired power plant in Germany, and calculated the emission ratios of Hg versus different air pollutants, and the GOM percentage in the plume. Generally, the work provides a lot of information of the multiple air pollutants emissions. Based on the emission ratios, one can calculate one pollutant emission through the other emissions, these make the pollutant estimation much easier.

Specific comments: (1)Since the air pollutant emissions from the coal fired power plant is largely depended on the boiler type, coal property, and the air pollutant control devices (APCDs), so, the result form one plant maybe differs from the others. Hence, please supplement the information about some basic aspects about the studied power plant, especially the coal property such as the proximate and ultimate analysis (if possible), the configuration of APCDs for NOx, PM and SO2 control.

We were not able to get actual data on the composition of the fuel in 2013, i.e. of lignite and sewage sludge, from the operator of the CFPP Lippendorf. Mercury content of the lignite from two seams of "Vereinigtes Schleenhain" open pit was 0.40 and 0.49 ppm (Rösler et al., 1977), within the range of eastern German lignites of 0.16-1.5 ppm (Yudovich and Ketris, 2005). This is now mentioned in the text.

We provide more information about FGD system and refer for details to Schütze et al. (2015). As discussed by Schütze et al. (2015), the chemistry within the FGD system is at least as important as the fuel composition. Schütze et al. (2015) also show a high day-to-day variability of the mercury removal efficiency. Assuming nearly constant FGD operating conditions, this suggests to a large inhomogeneity of the fuel composition.

Mercury emissions of a coal fired power plant in Germany 2 3 Andreas Weigelt<sup>1,\*</sup>, Franz Slemr<sup>2</sup>, Ralf Ebinghaus<sup>1</sup>, Nicola Pirrone<sup>3</sup>, Johannes 4 Bieser<sup>1,4</sup>, Jan Bödewadt<sup>1</sup>, Giulio Esposito<sup>3</sup>, and Peter F.J. van Velthoven<sup>5</sup> 5 <sup>1</sup>Helmholtz-Zentrum Geesthacht (HZG), Institute of Coastal Research, Geesthacht, Germany 6 <sup>2</sup>Max-Planck-Institute for Chemistry (MPI-C), Department of Atmospheric Chemistry, 7 8 Mainz, Germany <sup>3</sup>National Research Council (CNR), Institute of Atmospheric Pollution Research, Rende, Italy 9 <sup>4</sup>Deutsches Zentrum für Luft- und Raumfahrt (DLR), Institute of Atmospheric Physics, 10 11 Oberpfaffenhofen, Germany 12 <sup>5</sup>Royal Netherlands Meteorological Institute (KNMI), Chemistry and Climate Division, De 13 Bilt, Netherlands \*now at: Federal Maritime and Hydrographic Agency (BSH), Hamburg, Germany 14 Correspondence to: A.Weigelt (Andreas.Weigelt@bsh.de), F. Slemr (franz.slemr@mpic.de) 15 16 17 andreas.weigelt@bsh.de 18 franz.slemr@mpic.de 19 ralf.ebinghaus@hzg.de 20 pirrone@iia.cnr.it 21 johannes.Bieser@hzg.de

22

23

24 25 jan.boedewadt@hzg.de esposito@iia.cnr.it

velthove@knmi.nl

## Abstract

Hg/SO<sub>2</sub>, Hg/CO, NOx/SO<sub>2</sub> (NOx being the sum of NO and NO<sub>2</sub>) emission ratios (ERs) in the plume of coal fired power plant (CFPP) Lippendorf near Leipzig in Germany were determined within the European Tropospheric Mercury Experiment (ETMEP) aircraft campaign in August 2013. The gaseous oxidized mercury (GOM) fraction of mercury emissions was also assessed. Measured Hg/SO<sub>2</sub> and Hg/CO ERs were, within the measurement uncertainties, consistent with the ratios calculated from annual emissions in 2013 reported by the CFPP operator, while the NOx/SO<sub>2</sub> ER was somewhat lower. GOM fraction of total mercury emissions, estimated by three independent methods, was ~10% with an upper limit ofbelow ~25%. This result is consistent with findings by others and suggests that GOM fractions of ~40% of CFPP mercury emissions in current emission inventories are overestimated.

#### 1 Introduction

Mercury and especially methyl mercury which bio-accumulates in the aquatic nutritional chain are harmful to humans and animals (e.g. Mergler et al., 2007; Scheuhammer et al., 2007; Selin, 2009; and references therein). Therefore, its Hg emissions are on the priority list of several international agreements and conventions dealing with environmental protection and human health, including the United Nations Environment Program (UNEP) Minamata convention on mercury (www.mercuryconvention.org). Mercury is emitted to the atmosphere from a variety of natural (e.g. volcanic activity, evaporation from ocean and lakes) and anthropogenic sources (e.g. coal and oil combustion) (Mason et al., 2009; Pirrone et al., 2010). Coal-fired power plants (CFPPs) are believed to account for most (≥ 56%) of mercury emitted by stationary combustion sources which constitute 35 – 77% of all anthropogenic Hg emissions (Pirrone et al., 2010; Chen et al., 2014; Ambrose et al., 2015). Mercury from CFPPs is emitted as gaseous elemental mercury (GEM), gaseous oxidized 

mercury (GOM) and particulate bound mercury (PBM). Elemental mercury has a high vapour pressure, is virtually insoluble in water resulting in a long residence time in the atmosphere of about 1 yr (6 - 12 months (Slemr et al., 1985; Lindberg et al., 2007; Selin, 2009; Holmes et al., 2010). GOM with its high solubility and low vapour pressure is readily washed and rained out as are the particles carrying particulate mercury (PM) particle bond mercury. In addition, GOM is also rapidly removed by dry deposition. GOM and PMPBM are believed to be in

- 1 equilibrium (Rutter and Schauer, 2007; Amos et al., 2012). GOM is thus a major driver for
- 2 the global mercury deposition and is estimated to make up more than 50% of the total Hg
- deposition (Zhang et al., 2012a; Bieser et al., 2014).
- 4 There are only two sources of GOM in the atmosphere: primary GOM emissions from
- 5 anthropogenic sources and the oxidation of elemental mercury. The major anthropogenic
- 6 mercury sources on a global scale are small scale artisanal gold mining (SSAG) and coal
- 7 combustion (Pirrone et al. 2010). While SSAG emits solely elemental mercury, the CFPP
- 8 emissions in emission inventories are estimated to have a GOM fraction between 35% and
- 9 40% (Pacyna et al., 2006; Wilson et al., 2010; EPA, 2011). However, global and regional
- model studies have repeatedly indicated that models are overestimating atmospheric GOM
- 11 concentrations (Zhang et al., 2012b; Kos et al., 2013; Bieser et al., 2014). Possible
- 12 explanations for this are an overestimation of the in-plume GEM oxidation rates or the
- overestimation of the amount of GOM emitted by CFPPs. The latter has been hypothesized to
- be due to a fast reduction of GOM inside the plume (Zhang et al., 2012b; Kos et al., 2013).
- 15 While the operators of CFPPs are forced to measure and report the amount of mercury
- 16 released into the atmosphere, there is only little knowledge on the The speciation of these
- 17 emission sources. CFPP emissions is not well known. That is because of varying composition
- 18 of burnt\_coal\_burned, complex chemistry in the stack gases (e.g. Lohman et al., 2006;
- 19 Schofield, 2008; Tatum Ernest Tatum et al<sub>7</sub>., 2014) and the large number of different methods
- 20 used to clean CFPP flue gases with very different percentage of gaseous oxidized mercury
- 21 (GOM) to total mercury ranging from less than 10% up to 90% (Wang et al., 2010,
- 22 SchuetzeSchütze et al., 2012, 2015, and references therein). Analytical problems also
- 23 contribute to the uncertainty: the current emission monitoring systems are not sensitive
- 24 enough to measure and speciate low mercury concentrations in flue gases of modern CFPPs
- 25 (Mayer et al., 2014). Moreover, there has been evidence that the current ambient air
- 26 measurement systems might not capture all oxidized mercury species with similar efficiency
- 27 (Jaffe et al., 2014; Gustin et al., 2015a, 2015; Weiss-Penzias et al., 2015).
- 28 The European Tropospheric Mercury Experiment (ETMEP) was carried out in July/August
- 29 2012 (ETMEP-1) and August 2013 (ETMEP-2) to measure local emissions, vertical
- 30 profile profiles from inside the boundary layer to the lower free troposphere, and horizontal
- 31 distribution of mercury over Europe. In total Altogether 10 measurement flights were
- 32 performed over Italy, Slovenia, and Germany with two propeller aircraft. The ETMEP-1

- 1 campaign focused on volcanic emissions of Etna. The objectives of the ETMEP-2 campaign
- were a) to obtain vertical mercury profiles above several sites in central and southern Europe
- 3 (Weigelt et al., 2016), b) to assess horizontal distribution of mercury concentrations during
- 4 | the flight from flights between Italy to and Germany, and c) to determine mercury emission
- 5 ratios for a coal-fired power plant (CFPP) near Leipzig. Here, we present the measurements of
- 6 CFPP emissions and their speciation.

8

9

10

11

1213

14

15

16

17 18

19

2021

22

23

24

25

26

27

28

29

30

31

32

## 2 Experimental

The power plant under investigation is located in Lippendorf, a small village ca. 15 km south of Leipzig in Germany. The CFPP of Lippendorf consists of two units with 934 MW gross power each. It has been in operation since 2000 and belongs with a net efficiency of 42.6% to one of the most modern and efficient CFPPslignite fuelled power plants in Europe. About 10 million 750 metric tons per hour (t/h) of brown coal with rather high sulphur content-lignite from a nearby open pit mine "Vereinigtes Schleenhain" are burnt annually together with ~ 22 t/h of sewage sludge (Schütze et al., 2015). Mercury content of lignite from two seams of "Vereinigtes Schleenhain" was 0.40 and 0.49 ppm (Rösler et al., 1977), within the range between 0.16 and 1.5 ppm for eastern German lignites (Yudovich and Ketris, 2005). No data about mercury content of the sewage sludge are available. The SO<sub>2</sub> emissions are reduced byflue gas is directed through an electrostatic filter and a flue gas desulfurization (FGD) system to reduce particle and SO<sub>2</sub> emissions. The FGD is using wet washing with CaO suspension, with added sulfidic precipitant and removes ~ 80% of mercury (Schütze et al., 2015). Despite the efficient FGD cleaning, the CFPP of Lippendorf ranks 4<sup>th</sup> most health harmful emitter in Germany (rating based on combined emissions of SO2, NOx, and particulate matter, Preiss et al., 2013) and 14<sup>th</sup> most harmful emitter in Europe according to the European Environment Agency (rating based on combined emissions of SO<sub>2</sub>, NOx, NH<sub>3</sub>, CO<sub>2</sub>, particulate matter, non-methane hydrocarbons, heavy metals, and organic micropollutants, EEA, 2011) with respect to health. Annual emissions reported by the operator of the CFPP Lippendorf for 2013, the year of our measurements, were: 1.18\*10<sup>13</sup> g CO<sub>2</sub>, 1.21\*10<sup>10</sup> g SO<sub>2</sub>, 7.91\*10<sup>9</sup> g NOx, 7.55\*10<sup>8</sup> g CO, and 4.1\*10<sup>5</sup> g Hg, among other pollutants. Mercury limit emission values (LEVs) of large combustion plants in Germany are stipulated by ordinance (Federal Law) from 2004 and its revision in 2013 to 50 µg m<sup>-3</sup> as a half hour average, 30 µg m<sup>-3</sup> as a daily average, and 10 µg m<sup>-3</sup> as an annual average

concentration (Mayer et al., 2014). Continuous monitoring of mercury emissions is mandatory but only annual total (unspeciated) mercury emissions have to be reported. —European Union (EU) wide LEVs of  $< 5 \mu g \text{ m}^{-3}$  for hard coal and  $< 7 \mu g \text{ m}^{-3}$  for lignite fired CFPPs are under discussion (VGB, 2016). The measurement campaign described above was performed with a CASA 212 two engine aircraft (Fig. 1a) operated by Compagnia Generale Ripreseaeree (http://www.terraitaly.it/). The CASA 212 with a maximum payload of 2.7 tons can carry the measurement instruments, different service instruments, the power supply, two pilots, and 5 operators. With a normal cruising speed of ~ 260 km h<sup>-1</sup> its range is ~ 1600 km. Although the maximum flight level of the unpressurized aircraft is 8500 m, the maximum altitude of ETMEP-2 flights without oxygen supply was limited to ~3000 m above sea level (a.s.l.), The aircraft was equipped with a gas inlet system (Fig. 1b) which had been developed and manufactured at the Helmholtz-Zentrum Geesthacht. The gas inlet was designed for the cruising speed of the CASA 212 of  $\sim 72 \text{ m s}^{-1}$ . A diffuser tube reduced the air speed to  $\sim 5 \text{ m}$ s<sup>-1</sup>. About 120 l min<sup>-1</sup> (ambient conditions) enters the inlet at the cruising speed of 260 km h<sup>-1</sup>. The air sample is taken in the centre of the diffuser tube with a flow rate of  $\sim 25 \, \mathrm{l \ min^{-1}}$ . The remaining flow of 951 min<sup>-1</sup> is directed to the back of the inlet where the air speed is increased by a nozzle and the air exits. By replacing the inlet and outlet nozzle with smaller or larger ones, this inlet system can be fitted to other aircraft with a different cruising speed. In the expanded area (behind the main sample line) the air temperature (T), static pressure (p), and relative humidity (rH) are measured. To avoid adsorption losses of sticky trace gases, the internal surface of the inlet system was coated with Teflon and only PFA tubing was used for the sampling lines. The outside of the inlet was coated with copper to avoid electrostatic charging. The inlet was fastened onto a 90 cm long telescope tube (6 cm diameter) which was mounted in a hole on the floor fuselage via a sliding guide. After take-off, the tube was pushed down by ~40 cm from inside the aircraft, to ensure that the inlet nozzle is outside the aircraft boundary layer. Before landing the tube was pulled back into the aircraft to protect it from damage by objects whirled up by the front wheel. The inlet and the telescope tube were equipped with heaters to prevent icing but during the ETMEP measurements the heating was

always switched off because the measurement flights were carried out in summer at altitudes

below 3000 m a.s.l. The tubing from the inlet to instruments (~2.5 m long 3/8" O.D. main

1

2

3 4

5

6

7

8

9 10

11

12

13 14

15

16

17

18 19

20

21

22

23

24

25

2627

28

29

30

- 1 sample tube with PFA manifolds to instruments) was not heated. The temperature inside the
- 2 cabin was 18 to 30°C.
- 3 The aircraft was equipped with three mercury measurement instruments: a Lumex
- 4 RA-915AM, a Tekran 2537B, and a Tekran 2537X -(cf. Tab. 1). The Lumex RA-915 AM is
- 5 based on atomic absorption spectroscopy (AAS) with Zeeman background correction
- 6 (Sholupov et al., 2004) and as such measures specifically only gaseous elemental mercury
- 7 (GEM) with a temporal resolution of 1 s. Its raw signal is noisy (about  $\pm 4$  ng m<sup>-3</sup> with a
- 8 temporal resolution of 1 s) and is dependent on pressure and temperature. Nevertheless, the
- 9 fast response of the instrument is very useful to detect GEM in rather narrow highly
- 10 concentrated plumes at a cruising speed of about 72 m s<sup>-1</sup>. Because of thermal drifts, its zero
- was measured every 4 min for 1 min using an internal active carbon zero air cartridge.
- 12 The Tekran 2537B and 2537X analysers are based on preconcentration of mercury and its
- 13 compounds on gold traps (Slemr et al., 1979), thermodesorption, and detection by cold vapour
- 14 atomic fluorescence spectroscopy (CVAFS). Although CVAFS can detect only GEM,
- 15 mercury compounds are converted to GEM during adsorption or thermodesorption (Slemr et
- 16 al., 1978) and, consequently Tekran instruments <u>can</u> measure total gaseous mercury (TGM).
- 17 The instruments use two gold traps to ensure a continuous measurement: while one is
- 18 adsorbing mercury during sampling, the other one is being analysed and vice versa. The
- 19 highest temporal resolution of the Tekran instruments of  $150 \mathrm{\ s}$  is given by the time necessary
- 20 for the thermodesorption of mercury from the gold traps and their cooling. The Tekran 2527X
- 21 analyser (Tekran 1) was run with quartz wool trap upstream of the instrument, which removes
- 22 gaseous oxidized mercury (GOM) and aerosol particles with particle bound mercury (PBM)
- but no GEM from the air stream (Lyman and Jaffe, 2011; Ambrose et al., 2013). The Tekran
- 24 2537B (Tekran 2) analyser was operated as backup instrument without a quartz wool trap.
- 25 The Teflon made (PFA and PTFE) aircraft gas inlet and tubing system are similar to the
- 26 CARIBIC trace gas inlet for which high GOM transmission was qualitatively demonstrated.
- 27 Based on the short residence time (0.3 sec) in the tubing to the instrument, the conditions as
- 28 during an international field intercomparison (Ebinghaus et al., 1999), and higher GOM
- 29 concentrations in the plume than in ambient air, we presume Tekran measurements without
- 30 quartz wool trap represent total gaseous mercury (TGM = GEM + GOM). Therefore, the
- 31 Tekran 2537B measurement are believed to represent TGM concentrations whereas those by
- 32 Tekran 2537X GEM concentrations, both with an uncertainty of 12.5%. The uncertainty has

1 been calculated by Weigelt et al. (2013) using two different approaches according to ISO

2 20988 type A6 and ISO 20988 Type A2. This uncertainty complies with the quality objective

of the EU air quality directive 2004/107/EC. The instrumental setup in the aircraft was almost

identical and, therefore, we expect the uncertainty to be very similar.

3

4

5

6

7

8 9

10

11

12

13

14

15

16 17

18

19

20

21

2223

24

25

2627

28

29

30

31

3233

Direct estimation of the GOM concentrations was made using three manual KCl denuder samples taken during the vertical profiles (sampling time 1 hour or longer, sampling flow rate 6.4 l/min at standard temperature and pressure (STP; T=273.15 K, p=1013.25 hPa), corresponding to ~ 10 l/min at ambient temperature and pressure in 3000 m a.s.l and controlled using a mass flow controller): one downwind of the Lippendorf CFPP, one upwind over the city of Leipzig (both on August 21, 2013), and one over the GMOS master site "Waldorf" in northern Germany on August 22<del>. Two blank samples were also taken by KCl</del> denuders (Fig. 2). For sampling, the KCl denuders were connected to a bypass of the main sampling line about 1.2 m downstream the above described Teflon coated gas inlet. The sampling flow rate was controlled with a mass flow controller downstream the KCl denuder and was set to 6.4 l/min at standard temperature and pressure (STP; T=273.15 K, p=1013.25 hPa), corresponding to ~ 10 l/min at ambient temperature and pressure in 3000 m a.s.l. The sampling time was 1 hour or longer, corresponding to a total sample volume of 600 litres or more. The KCl denuder was kept at constant temperature of 50°C using a heater band. Two blank samples were also taken using KCl denuders and handled exactly in the same way as the samples (denuder preparation, installation to sampling setup, storage, analysis) but without sucking sample air through them. After all flights had been finished, the KCl denuders were analysed for their total GOM loads in the laboratory. Despite a relatively high uncertainty of about  $\pm 5$  pg m<sup>3</sup>Five days before the ETMEP-2 campaign started all denuders were prepared for sampling by coating with KCl and were purged at 500°C for 60 min in a Tekran 1130 speciation unit with mercury free air from a Tekran active carbon zero air cartridge. During the heating mercury in the flushing air downstream the KCl denuders was measured with a Tekran 2537B mercury analyser to ensure that mercury was quantitatively removed from the KCl denuders. After the campaign the KCl denuders were analysed for their total GOM loads in the laboratory using the same setup as for the denuder preparation. The lower detection limit was estimated to be 1 pg m<sup>-3</sup> and is dominated by the Tekran 2537 lower detection limit (0.1 ng m<sup>-3</sup>). With about  $\pm$  5 pg m<sup>-3</sup> The overall method uncertainty defined as a difference of the two blanks is with about  $\pm 5 \text{ pg m}^{-3}$  relatively high. Nevertheless, the method provides semi-quantitative information about GOM concentration in the plume.

2 interferences. GOM captured by quartz wool can be released by higher air humidity (Ambrose 3 et al., 2015) and KCl traps and denuders can release GOM in presence of high ozone and 4 water concentrations (Lyman et al., 2010; Huang and Gustin, 2015). These interferences may 5 result in overestimation of GEM and underestimation of GOM emissions. GEM measured by 6 Lumex is not subject to any known interference. 7 For the identification and characterization of different air masses carbon monoxide (CO), 8 ozone (O<sub>3</sub>), sulphur dioxide (SO<sub>2</sub>), nitrienitrogen oxide (NO), nitrienitrogen dioxide (NO<sub>2</sub>), 9 and the basic meteorological parameters temperature (T), pressure (p), and relative humidity 10 (rH) were measured simultaneously with high temporal resolution. Instrument details 11 including the estimated measurement uncertainty are summarised in Table 1. Uncertainties 12 were calculated according to the individual instrument uncertainty given by the manufacturer 13 and the calibration gas accuracy (CO, O3, SO2, NO). All instruments were protected from 14 aerosols using PTFE filters (0.2 µm pore size). Model meteorological data like potential 15 vorticity, equivalent potential temperature, relative and specific humidity, cloud cover, cloud 16 water content, three-dimensional wind vector, as well as five day backward trajectories were 17 calculated every 150 s along the aircraft flight tracks for additional information. These 18 calculations are based on meteorological analysis data from the European Centre for Medium-19 Range Weather Forecasts (ECMWF) and the TRAJKS trajectory model (Scheele et al., 1996). 20 Before take-off all instruments were warmed up for at least 45 minutes, using an external 21 ground power supply. During the starting of the engines the power was interrupted for less 22 than 3 minutes. Since 45 minutes were too short to stabilize the Tekran 2537 internal 23 permeation source, thesethe Tekran instruments were calibrated only after each measurement 24 flight before the engine shut down, using the internal permeation source. All data were 25 recalculated, using the post flight ealibration calibrations. Before and after the ETMEP-2 26 campaign the permeation rate of the internal permeation source was checked by manual injection of a known amount of mercury from an external mercury source (Tekran 2505 unit). 27 During the instrument warm up, take-off and landing a Tekran active carbon zero air cartridge 28 29 was inserted upstream of the Tekran instruments to prevent their contamination by the usually 30 dirty air around airports and to enable their zeroing. All mercury instruments reported zero 31 mercury concentration while the cartridge was inserted. The pressure in the fluorescent cells 32 of both Tekran instruments was kept constant using upstream pressure controllers at the exits

We note that both methods used here to estimate GOM concentrations are subject to

of the cells. This eliminated the known pressure dependence of the response signal (Ebinghaus and Slemr, 2000; TalbotRadke et al., 2007). The Lumex analyser has a much shorter warm up time of less than 10 minutes and was, therefore, calibrated before take-off with the internal calibration cell-consisting of a sealed quartz cylinder filled with air saturated with mercury vapor. Unfortunately, the Lumex analyser does not provide the option to verify the internal calibration by injection of mercury saturated air from an external source. However, a comparison of the used Tekran- and Lumex mercury analysers before and after the ETMEP-2 campaign showed a good agreement with a difference of less than 5%. The CO instrument calibration takes 60 seconds and was, therefore, performed during the measurement flights every 20 minutes with external calibration gas. The O<sub>3</sub>, SO<sub>2</sub>, NO/NO<sub>2</sub> instruments have a fairly constant signal response and were thus calibrated before and after the ETMEP-2 measurement campaign. Multipoint SO<sub>2</sub> and NOx calibration was NO calibrations were made using dilution (Environics 300E calibrator) of certified standard gases. NO<sub>2</sub> conversion efficiency was determined using gas phase titration. The factory calibration was used for the pressure, temperature and relative humidity sensors. The measurements were synchronized using their individual delay and response times. Please note that all mercury (TGM, GEM, and GOM) concentrations are reported at standard temperature and pressure (STP; T = 273.15K, p = 1013.25 hPa). At these standard conditions 1 ng m<sup>-3</sup> corresponds to a mixing ratio of 112 ppqv (parts per quadrillion by volume).

20

21

2223

24

25

26

2728

29

30

31

32

1

2

3 4

5

6 7

8

9

10

11

1213

14

1516

17

18 19

# 3 Vertical distribution and Hg/SO<sub>2</sub>, Hg/CO, NOx/SO<sub>2</sub> emission ratios

The measurements were carried out on August 21 and 22, 2013. -On August 21 between 9:30 and 11:20 UTC the aircraft flew many circles at different altitudes downwind of a CFPP Lippendorf -(51°11`N, 12°22`E) followed between 11:25 and 12:20 UTC by a vertical profile upwind of CFPP Lippendorf over the city centre of Leipzig (51.353°N, 12.434 °E). Between 8:30 and 10:00 UTC of August 22 another vertical profile above the GMOS master site "Waldhof" (52°48`N, 10°45`E, about 200 km from Leipzig on the line connecting Leipzig and Hamburg) was flown, followed between 10:00 and 10:35 UTC by additional measurements downwind of the CFPP Lippendorf. Each vertical profile consists of at least seven horizontal flight legs, consisting of circles and altogether lasting 5 - 10 minutes each. The flight legs started inside the boundary layer at about 400 m above ground and ended at 3000 m a.s.l. The tracks of the flight in the CFPP plumeflights on August 21 and August 22 are shown in Figure

2a2 and 2bFigure 3a and 3b, respectively. The CFPP plume was encountered in the distance 1 of  $\sim 7.5$  km from the plant at an altitude of 1900 m a.s.l. on August 21 and in the distance of  $\sim$ 2 5 km at 1500 – 1650 m a.s.l. on August 22. With a horizontal wind speed of 2.4 and 1.5 m s<sup>-1</sup> 3 on August 21 and 22, respectively, the age of the plume was ~0.9 h on both days. 4 5 Figures 34 and 45 show data from the flight sections with CFPP plume encounters on August 21 and 22, 2013, respectively. The plume encounters lasted 1-2 min and are clearly indicated 6 7 by elevated  $SO_2$ , NOx ( $NOx = NO + NO_2$ ), and GEM concentrations measured by Lumex. 8 CO and rH enhancements are hardly visible on August 21 but are clearly recognizable on 9 August 22. Tekran instruments with a temporal resolution of 150 s are too slow to resolve 10 individual plume encounters but they also show a broad peak of enhanced GEM (Tekran 1 11 with quartz wool trap) or TGM (Tekran 2) concentrations. The difference between TGM measured by Tekran 2-without quartz wool trap and GEM measured by Tekran 1-with quartz 12 wool trap is small (on average  $0.087 \pm 0.117$  ng m<sup>-3</sup> (n = 8) on August 21 and  $0.063 \pm 0.079$ 13 ng m<sup>-3</sup> (n = 12) on August 22) and varies between -0.064 and +0.354 ng m<sup>3</sup> on both days. The 14 average differences are not significantly different from zero and neither do the maximum and 15 16 minimum differences exceed the combined uncertainty of the difference of 17.7%. On August 17 21 the plume was encountered several times at an altitude between 1600 and 2500 m a.s.l. The 18 most pronounced encounters numbered 1 - 4 were found at an altitude of 1800 - 2250 m a.s.l. 19 On August 22 the plume was encountered 3 times at a flight level of 1550 m and 3 times at 20 1650 m a.s.l. The numbered plume encounters were selected for quantitative evaluation.

Figure  $\frac{56}{5}$  shows the vertical distribution of the values measured downwind of the Lippendorf CFPP. The vertical profiles above Leipzig and Waldhof are discussed together with further profiles in the Weigelt et al. (2016). In Figure  $\frac{56}{5}$  the squares represent the constant flight level measurement points (2 measurements with 2.5 minutes each). The stars represent the measurements when climbing between two flight levels (2.5 min average). Therefore the The data; indicated as squares are, therefore, more significant and the data illustrated as stars do provide additional information on the vertical structure. Please note that the rH, air temperature (T), and the potential temperature ( $\theta$ ) are plotted with high temporal resolution (1 s) in the rightmost panel. The rH can be used to distinguish between boundary layer- and free tropospheric air. Usually inside Inside the planetary boundary layer (PBL) the relative humidity is usually much higher than in the free troposphere (Spencer and Braswell, 1996).

2122

23

24

25

26

27

28

29

30

1 The lower four horizontal flight legs (570 to 1340 m a.s.l.) show typical northern hemispheric GEM and TGM background concentration of ~1.6 ng m<sup>-3</sup> without any vertical gradient, CO. 2 O<sub>3</sub>, SO<sub>2</sub>, as well as NO and NO<sub>2</sub> also show no vertical gradient, indicating a well-mixed PBL. 3 4 This is in agreement to the other vertical profiles measured during ETMEP-2 campaign (Weigelt et al., 2016). From the fifth flight leg (1630 m a.s.l.) upward the GEM and TGM 5 concentration increases towards the PBL top (GEM (Tekran 1-(GEM): 1.7 ng m<sup>-3</sup> at 1630 m 6 a.s.l.; 2.6 ng m<sup>-3</sup> at 1940 m a.s.l.; TGM (Tekran 2-(TGM): 1.7 ng m<sup>-3</sup> at 1630 m a.s.l.; 2.8 ng 7 m<sup>-3</sup> at 1940 m a.s.l.; <u>GEM (Lumex-(GEM)</u>: 2.1 ng m<sup>-3</sup> at 1630 m a.s.l.; 2.4 ng m<sup>-3</sup> at 1940 m 8 9 a.s.l.). The increasing concentration is also captured by the measurements during the flight level change (GEM (Tekran 1-(GEM)): 1.7 ng m<sup>-3</sup> at 1540 m a.s.l.; 2.1 ng m<sup>-3</sup> at 1800 m a.s.l.; 10 TGM (Tekran 2-(TGM): 1.7 ng m<sup>-3</sup> at 1540 m a.s.l.; 2.3 ng m<sup>-3</sup> at 1800 m a.s.l.; GEM (Lumex 11 (GEM): 1.8 ng m<sup>-3</sup> at 1540 m a.s.l.; 2.2 ng m<sup>-3</sup> at 1800 m a.s.l.; stars in Fig. 5). As indicated 12 by the abrupt decrease of rH, the PBL top was found at 2150 to 2200 m a.s.l.. 13 ThereforeConsequently, the flight leg 7 at 2260 m a.s.l. and leg 8 at 3020 m a.s.l. were 14 performed in free tropospheric air. These two measurements show a typical free tropospheric 15 16 background concentration (~ 1.3 ng/m³, Weigelt et al., 2016 and references therein). The measurements during the flight level change from leg 6 to leg 7 represent a mixture of 17 18 boundary layer- and free tropospheric air (averaged altitude 2150 m a.s.l.). Therefore the 19 GEM (Tekran 1-GEM.), TGM (Tekran 2-TGM.), and GEM (Lumex-GEM) concentration of 2.3 ng m<sup>-3</sup>, 2.4 ng m<sup>-3</sup>, and 1.9 ng m<sup>-3</sup> was strongly influenced by the high concentration 20 21 below the boundary layer top. 22 In the altitude range 1600 m a.s.l. to 2200 m a.s.l. not only mercury, but also SO<sub>2</sub> was 23 significantly increased (from 1.6 ppb to 21.4 ppb), which clearly indicates that the mercury 24 was emitted from the CFPP. Inside the plume (leg 6), the O<sub>3</sub> concentration was slightly 25 decreased to 42.3 ppb. At the same time NO and NO<sub>2</sub> increased to 6.1 ppb and 8.9 ppb, respectively. Outside the plume (e.g. leg 4) O<sub>3</sub> was 48.5 ppb, NO was below the detection 26 27 limit, and NO<sub>2</sub> was ~1.5 ppb. This indicates O<sub>3</sub> depletion due to NO oxidation taking place 28 inside the plume (cf. Fig. 34 and 45). The presence of a temperature inversion at the PBL top 29 is indicated by the changing T and  $\theta$  vertical gradient in Fig.  $\frac{56}{2}$ . This inversion layer prevents a further ascent of the power plant plume. Therefore and, consequently, the highest 30 31 concentration of pollutants was found below the PBL top. As already shown with Fig. 3 and in 32 Figures 4 and 5, during a flight leg in a certain altitude (and during level change) the aircraft

did not remain within the plume all the time. Therefore, the concentrations, given in Fig. 56 1 do represent a mixture of plume and background air. 2 The ratio of concentration enhancements (ERs), ΔHg/ΔSO<sub>2</sub>, ΔHg/ΔCO, and ΔNOx/ΔSO<sub>2</sub> 3 4 represent the emission ratios at the stack if a) chemical reactions during the transport from the 5 stack to the point of interception can be neglected and b) the background concentrations have 6 not changed during the measurement including the transport from the stack to the place of 7 plume encounters. As mentioned above, the transport time from the stack to the location of 8 plume interception was ~ 0.9 h on both days. Based on OH concentrations measured in a 9 CFPP plume, Ambrose et al. (2015) estimated SO<sub>2</sub> and NOx lifetimes of 16 - 43 and 1.8 -10 5.8-h, respectively. The combination of GEM, TGM, and GOM measurements by Lumex, 11 Tekran 2537X (Tekran 1, with quartz wool trap), 2537B (Tekran 2, without quartz wool trap), and KCl denuder, respectively, suggests that there is no substantial conversion of GEM into 12 GOM within the transport time of ~ 0.9 h. The vertical profile over Leipzig, upwind of the 13 14 CFPP, was measured on August 21 -3ca. 1 h after the measurements in the plume. The CO, O<sub>3</sub>, SO<sub>2</sub>, NOx and Hg concentrations in the PBL over Leipzig with ~ 120, 50, 0.5, 3 ppb, 15 1.4-ng-m<sup>-3</sup>, respectively, are similar to respective concentrations found outside of the plume 16 over CFPP Lippendorf. Differences between them for SO2, NOx, and Hg are small when 17 compared with their enhancements in the plumes of ~ 40, 30 ppb, 4 ng m<sup>-3</sup>, respectively. On 18 19 August 22 no vertical profile upwind was measured, but SO2, NOx, and Hg concentrations 20 over Waldhof, ~ 90 km north of Leipzig, measured immediately before the downwind 21 measurements of CFPP Lippendorf, were comparable. We thus conclude that the background 22 concentrations of SO<sub>2</sub>, NOx, and Hg have not changed significantly during the 0.9 h long 23 transport from the stack to the location of aircraft interception and during ~ 20 min of the 24 repeated plume interceptions. In addition, the large SO2, NOx, and Hg enhancements in the plume make the calculated  $\Delta Hg/\Delta SO_2$  and  $\Delta NOx/\Delta SO_2$  ERs insensitive to small changes in 25 background  $SO_2$ , NOx, and Hg concentrations. This is not always the case for small  $\Delta CO$  and 26 negative  $\Delta O_3$  (negative because  $O_3$  is consumed by rapid oxidation of NO to NO<sub>2</sub>) relatively 27 28 to their background mixing ratios. In addition, the CO background mixing ratios changed

31 The ERs are usually calculated as a slope of Hg vs X correlations (e.g. Ambrose et al., 2015).

altitude change.  $\Delta Hg/\Delta CO$  for these plume interceptions was thus not calculated.

substantially from ~123 to 105 ppb during the plume crossing #4 and #5 on August 21 due to

29

30

32 The advantage of this method is that the background concentrations of neither Hg nor X have

1 to be known as long as they remain constant during the measurement. The method, however, 2 is applicable only if the plume crossings are much longer than the response time of the 3 instruments. With the plume transects lasting in our case only 60 - 120 s and effective 4 temporal resolution of 10 s for SO<sub>2</sub> and NOx measurements, however, the signals have to be carefully synchronized. In addition, the correlation slopes for individual plume crossings will 5 6 become quite uncertain because of small number of points. For this reason we apply the 7 correlation method for all (synchronized) points with SO<sub>2</sub> mixing ratios > 10 ppb. This 8 selection provides 35 and 45 points for Hg vs SO<sub>2</sub> correlations on August 21 and 22, 9 respectively. Individual plume crossings are not resolved by this calculation. Correlations 10 made by the bivariate Williamson-York method (Cantrell, 2008) provide a slope and its 11 statistical uncertainty representing ER (Hg/SO<sub>2</sub>) and its uncertainty. 12 An alternative method calculates ERs as a ratio of  $\Delta$ Hg to  $\Delta$ X where  $\Delta$ Hg and  $\Delta$ X are signal 13 enhancements against the background integrated over the plume crossing. This method, called 14 here "integral method", is applicable for measurements with instruments with different 15 response times and we will show that it can use even Tekran measurements with a temporal 16 resolution of 150 s, although not for individual plume crossings. Opposite to the correlation 17 method, no exact synchronization is needed. The disadvantage, however, is that the results are 18 sensitive to the selection of background concentrations. Figures 34 and 45 show that 19 background Hg concentrations are especially difficult to define from the Lumex 20 measurements. We thus use the Hg background concentrations measured by the much-more 21 precise Tekran instrument. As the Lumex instrument measured only GEM, we use the 22 background measured by Tekran instrument with quartz wool (Tekran 1). The other 23 disadvantage of the integral method is that, opposite to the correlation method, the uncertainty 24 of ERs is difficult to quantify. We overcome this difficulty here by averaging the ERs from individual plume crossings and taking thetheir standard deviation as a measure of ER 25 26 uncertainty. The Hg/SO<sub>2</sub> ERs are listed in Table 2. The correlation and integral methods provide similar 27 results with  $5.53 \pm 1.10$  and  $5.56 \pm 1.19$  µmol mol<sup>-1</sup>, respectively, for August 21, and  $7.38 \pm$ 28 0.92 and  $6.32 \pm 1.52$  µmol mol<sup>-1</sup>, respectively for August 22. The integral method with <u>TGM</u> 29 (Tekran 2) and SO<sub>2</sub> integrals over all plume encounters provide somewhat higher Hg/SO<sub>2</sub> ERs 30 31 but still within the uncertainties of the correlation and integral methods. The measured Hg/SO<sub>2</sub> ERs are smaller than the emission ratio of 10.8 µmol mol<sup>-1</sup> calculated from Hg and 32

 $SO_2$  annual emissions reported by the CFPP operator for 2013. They are close to 5.2 - 6.51 umol mol<sup>-1</sup> determined by Ambrose et al. (2015) for Big Brown (BBS) and Dolet Hills 2 Stations (DHS). BBS, a 1187 MW CFPP in Texas, is fired with subbituminous coal and is 3 4 equipped with activated carbon injection flue cleaning. DHS, a 721 MW CFPP in Louisiana, 5 is fired with lignite and is equipped with wet flue gas desulfurizationFGD, similar to the FGD 6 of the CFPP Lippendorf. 7 Hg/CO ERs are frequently used to classify the origin of different plumes (Slemr et al., 2009, 2014; Lai et al., 2011, and references therein) with ERs < 0.25 µmol mol<sup>-1</sup> typical for plumes 8 from biomass burning and ERs  $> 0.6 \mu mol \text{ mol}^{-1}$  characteristic for plumes of urban/industrial 9 origin. The Hg/CO ERs measured in the plume of CFPP Lippendorf are listed in Table 3. The 10 11 correlation method tends to yield somewhat higher Hg/CO ERs than the integral method. 12 Because of changing background on August 21 and changing altitude on August 22, no ERs 13 were calculated by integral method using the Tekran measurements. As mentioned before, the 14 high background CO mixing ratios and relatively small CO enhancement in the plume make 15 the integral method quite sensitive to the chosen background. For this reason we believe 5.2 and 9.4 µmol mol<sup>-1</sup> from correlation method for August 21 and August 22, respectively, to be 16 more reliable. -The Hg/CO emission ratio from the 2013 annual emissions reported by the 17 operator is 7.6 µmol mol<sup>-1</sup>, in reasonable agreement with our measurements. Hg/CO ERs of 18 19 this magnitude have never been observed so far in the plumes detected during the CARIBIC 20 flights (Slemr et al., 2014). This is probably because only large plumes extending over several 21 hundreds to few thousands of km can be detected by these flights. Their Hg/CO ERs are then 22 a mixture of Hg/CO ERs from point sources embedded in plumes from larger industrial and/or 23 urban areas. 24 Simultaneous NOx and SO2 measurements allow us to calculate also the NOx/SO2 ERs which 25 are listed in Table 4. The ERs from the correlations and integral methods are in good agreement with each other on both days. The NOx/SO<sub>2</sub> ER of 0.59 mol mol<sup>-1</sup> on August 21 is 26 almost twice as large as 0.27 mol mol<sup>-1</sup> on August 22, and both ERs are substantially lower 27 than the emission ratio of 0.91 mol mol<sup>-1</sup> calculated from the NOx and SO<sub>2</sub> emissions reported 28 by the CFPP operator for 2013. All these NOx/SO<sub>2</sub> ERs are substantially larger than ~0.08 29 mol mol<sup>-1</sup> reported by Ambrose et al. (2015) for Big Brown CFPP in Texas and corrected for 30 31 the NOx loss during the transport from the stack to the point of the plume interception.

- Ozone is not emitted but the ambient  $O_3$  is consumed by a rapid reaction with NO  $(O_3 + NO =$
- $2 ext{NO}_2 + ext{O}_2$ ) in the plume during the transport from the stack to the point of plume interception.
- 3 The O<sub>3</sub>/NOx ERs thus do not represent emission ratios and they are negative because of O<sub>3</sub>
- 4 consumption. If only NO were emitted the O<sub>3</sub>/NOx ER should be -1 mol mol<sup>-1</sup>. O<sub>3</sub>/NOx ERs
- 5 were not calculated for August 21 because of changing O<sub>3</sub> background mixing ratio. The
- 6 calculated O<sub>3</sub>/NOx ERs for August 22 are listed in Table 5. The correlation method provides a
- slope of  $-0.62 \pm 0.13$  mol mol<sup>-1</sup> while the integral method provides an ER of  $-1.0 \pm 0.6$  mol
- 8  $\text{mol}^{-1}$ . We thus conclude that the emitted NO constitute some 60 100% of NOx emissions.

10

### 4 GOM emissions

- 11 As mentioned earlier, the GOM measurements made here using quartz wool traps and KCl
- 12 coated denuders can be both influenced by high humidity (Huang and Gustin, 2015) and those
- 13 made by KCl additionally by high O<sub>3</sub> concentrations (Lyman et al., 2010). Because of NO
- emissions, the O<sub>3</sub> concentrations in the CFPP plumes will be lower than in ambient air making
- 15 | thisthe O<sub>3</sub> interference unlikely. The humidity interference would lead to an underestimation
- 16 of GOM concentrations measured by KCl denuders and overestimation of GEM
- 17 concentrations measured by Tekran instrument with quartz wool trap. However, specific GEM
- 18 measurements are provided by Lumex, an atomic absorption instrument with Zeeman
- 19 background correction, albeit with a worse precision when compared to Tekran
- 20 measurements.
- 21 Table 6 lists the GOM concentrations measured by the KCl denuders during the vertical
- 22 profiles over Leipzig and in the plume of CFPP Lippendorf on August 21, 2013, and over
- Waldhof on August 22, 2013. Taking into account the uncertainty of  $\pm 5$  pg m<sup>-3</sup> there is hardly
- 24 any difference between GOM concentration of 5.8 pg m<sup>-3</sup> measured during the vertical profile
- over Leipzig and 11.4 pg m<sup>-3</sup> in the plume of CFPP Lippendorf on August 21. The difference
- 26 of 5.6 pg m<sup>-3</sup> is distributed over the vertical profile of 3000 m. Assuming ~ 300 m thick layer
- 27 with Vertical profile in Fig. 6 shows that the CFPP plume and was about 450 m thick.
- 28 Assuming nearly zero GOM concentrations outside of this layer, the GOM concentrations in
- 29 the layer would be  $\sim 6040$  pg m<sup>-3</sup>. This is roughly consistent with the differences between
- Tekran measurements without quartz wool trap and with it. The average difference in the
- 31 | plume was  $87 \pm 117 \text{ pg m}^{-3}$  (n=8) on August 21 and  $63 \pm 79 \text{ pg m}^{-3}$  (n=12), on August 22.
- Related to the average TGM enhancement (Tekran without quartz wool trap) in the plume of

1 0.90 ng m<sup>-3</sup> on August 21 and of 1.03 ng m<sup>-3</sup> on August 22, the GOM concentration would

2 represent ~ 10% and ~ 6% of TGM emissions on August 21 and 22, respectively.

3 An independent assessment of the GOM emissions can be made using Hg/SO<sub>2</sub> ERs listed in

Table 2. On August 21, the Hg/SO<sub>2</sub> ER of  $5.5 \pm 1.1 \,\mu\text{mol mol}^{-1}$  from correlation and  $5.6 \pm 1.2$ 

µmol mol<sup>-1</sup> from integral methods, both based on specific GEM measurements by Lumex, are

within their uncertainties consistent with 6.6 µmol mol<sup>-1</sup> derived from Tekran with quartz

wool trap. On August 22, the Hg/SO<sub>2</sub> ER of  $7.4 \pm 0.9$  µmol mol<sup>-1</sup> from correlation method is

consistent with 8.1  $\mu$ mol mol<sup>-1</sup> determined from Tekran data, while the 6.3  $\pm$  1.5  $\mu$ mol mol<sup>-1</sup>

from the integral method is somewhat lower. Consequently, Hg/SO<sub>2</sub> ERs from less specific

measurements with quartz wool trap tend to be somewhat higher but within their combined

uncertainties comparable with those derived from GEM specific Lumex measurements. A

comparison of Hg/SO<sub>2</sub> ERs measured by Tekran without and with quartz wool trap implies

GOM emissions representing 13 and 9% of TGM emissions on August 21 and 22,

respectively. Taking GEM specific Lumex measurements instead of those made by Tekran

with quartz wool trap would imply GOM emissions representing 27 and 24% on August 21

and 22, respectively, which we consider an upper limit.

4

5

6

7

8

9 10

11

12

13

14

15

17

18

19

20

21

2223

24

25

2627

28

29

30

31

32

In summary, we conclude that GOM represents - 10 represented less than 25 % of the TGM emitted from CFPP Lippendorf on August 21 and 22, 2013. Schütze et al. (2015) provide no numerical value but their Figure 6 shows that GOM represented ~20% of total mercury emissions of the CFPP Lippendorf at operating conditions in 2013, which is consistent with an uncertainty range of 0 - 25%.our measurements. Edgerton et al. (2006) reported GOM fraction of 13, 19, and 21% of total mercury in the plumes from CFPPs Hammond, Crist, and Bowen in the U.S. Stergašek et al. (2008) reported 4% GOM fraction for Hg emissions from CFPP with FGD in Slovenia which was fired by lignite. Wang et al. (2010) found GOM fractions of 6 - 25% of all Hg emissions from five Chinese power plants with FGD. Deeds et al. (2013) found 13% of total mercury being GOM in the plume of CFPP Nanticoke in Canada. They think that discrepancy between this and 43% GOM fraction found in stack gases is due to sampling biases. Tatum Ernest et al. (2014) support their findings using a speciation technique still in development. On the other side Landis et al. (2014) report high GOM fractions of > 86% in stack gases of the CFPP Crist CFPP and 4-40% conversion of GOM into GEM in the plume in 0.6 - 1.3 km distance from the stack. They attribute the difference to a reduction of GOM to GEM- during the plume transport. But the reduction during the plume transport

cannot resolve the difference between 86% and 20% measured by Landis et al. (2014) and Schütze et al. (2015) directly in the stack of the CFPPs Crist and Lippendorf, respectively. We note that Figure 7 of Schütze et al. (2015) shows a large day-to-day variation in mercury removal efficiency of the CFPP Lippendorf which probably also applies to the GOM removal efficiency. Part of the difference GOM in stack gases of CFPP Lippendorf and CFPP Crist can thus result from day-to-day variations in GOM removal efficiency. Putting this unresolved issue aside, low fractions of GOM emissions reported here and by others (Edgerton et al., 2006; Stergašek et al., 2008; Wang et al., 2010; Deeds et al., 2013; Landis et al., 2014) are in contrast to the AMAP/UNEP geospatially distributed mercury emissions dataset "2010v1" (Wilson et al., 2013), splittingwhich splits the speciated mercury emissions from combustion in power plants to 50% GEM, 40% GOM, and 10% PBM. As mentioned before, the flue gas desulphurisation (FGD) in CFPP Lippendorf is made by washing of the flue gas with CaO suspension with added sulfidic precipitant and this type of FGD is known to capture most of GOM (Schütze, 2013). Although no PBM was measured in this study, 10% of mercury being emitted as PBM according to the inventory is probably also an overestimation for CFPPs with FGD (Stergašek et al., 2008; Wang et al., 2010).

## **5 Conclusions**

Plume of the coal fired power plant (CFPP) Lippendorf near Leipzig in Germany was encountered several times on August 21 and 22, 2013. On August 21 the plume was captured at below planetary boundary layer top due to a temperature inversion layer. Hg/SO<sub>2</sub>, Hg/CO, NOx/SO<sub>2</sub> ERs in the plume were determined as a slope of bivariate correlations of the species concentrations and as ratios of integrals over the individual plume crossings. -The measured Hg/SO<sub>2</sub> and Hg/CO ERs were, within the measurement uncertainties, consistent with the ERs calculated from annual emissions reported by the CFPP operator for 2013, the NOx/SO<sub>2</sub> ER was somewhat lower.

GOM fraction of total mercury emissions was estimated a) using GOM measurements by KCl denuders, b) from a difference between Hg measurements by Tekran instruments without and with quartz wool trap, and c) from a difference between Hg measurements by Lumex instrument. Despite large uncertainties in all these estimates we conclude that GOM emissions represent

Formatiert: Standard

-10less than 25% of the total mercury emissions with an uncertainty range of 0 – 25%... This result is consistent with 20% found by Schütze et al. (2015) in stack gases of CFPP Lippendorf in 2013 and findings by others (Edgerton et al., 2006; Stergašek et al., 2008; Wang et al, 2010; Deeds et al., 2013) and). It suggests that GOM fractions of ~40% of CFPP mercury emissions in current emission inventories are overestimated. Although PBM was not measured by us, its inventoried fraction of 10% is too high too for CFPPs with FGD according to the above references too high too for CFPPs with FGD.

8 9

1

2

4

5

6 7

# Acknowledgements

- 10 Measurements were carried out as part of the European Tropospheric Mercury Experiment
- 11 (ETMEP) within the Global Mercury Observation System project (GMOS; www.gmos.eu).
- 12 GMOS is financially supported by the European Union within the seventh framework
- programme (FP-7, Project ENV.2010.4.1.3-2). Special thanks are due to Compagnia Generale
- Ripreseaeree (http://www.terraitaly.it/) in Parma/Italy and the pilots Oscar Gaibazzi and Dario
- 15 Sassi for carrying out the measurement flights.

#### References

- 2 AMAP/UNEP, 2013: AMAP/UNEP geospatially distributed mercury emissions dataset
- 3 2010v1, available online: http://www.amap.no/mercury-emissions/datasets (20.11.2013)
- 4 Ambrose, J.L., Lyman, S.N., Huang, J., Gustin, M.S., and Jaffe, D.A.: Fast time resolution
- 5 oxidized mercury measurements during the Reno Atmospheric Mercury Intercomparison
- 6 Experiment (RAMIX), Environ. Sci. Technol., 47, 7285-7294, 2013.
- 7 Ambrose, J.L., Gratz, L.E., Jaffe, D.A., Campos, T., Flocke, F.M., Knapp, D.J., Stechman,
- 8 D.M., Stell, M., Weinheimer, A., Cantrell, C., and Mauldin, R.L.: Mercury emission ratios
- 9 from coal-fired power plants in the southeastern U.S. during NOMADSS, Environ. Sci.
- 10 Technol., 49, 10389-10397, 2015.
- Amos, H.M., Jacob, D.J., Holmes, C.D., Fisher, J.A., Wang, Q., Yantosca, R.M., Corbitt,
- 12 E.S., Galarneau, E., Rutter, A.P., Gustin, M.S., Steffen A., Schauer, J.J., Graydon, J.A., Louis,
- 13 V.L.St., Talbot, R.W., Edgerton, E.S., Zhang, Y., and Sunderland, E.M.: Gas-particle
- partitioning of atmospheric Hg(II) and its effect on global mercury deposition, Atmos. Chem.
- 15 Phys., 12, 591-603, doi:10.5194/acp-12-591-2012, 2012.
- 16 Bieser, J., DeSimone, F., Gencarelli, C., Geyer, B., Hedgecock, I.M., Matthias, V., Travnikov,
- 17 O., and Weigelt, A.: A diagnostic evaluation of modelled mercury wet depositions in Europe
- 18 using atmospheric speciated high resolution observations, Environ. Sci. Pollut. Res., 21,
- 19 9995-10012,, doi:10.1007/s11356-014-2863-2, 2014.
- 20 Cantrell, C.A.: Technical note: Review of methods for linear least-squares fitting of data and
- 21 application to atmospheric chemistry problems, Atmos. Chem. Phys., 8, 5477-5487, 2008.
- 22 Chen, Y., Wang, R., Shen, H., Li, W., Chen, H., Huang, Y., Zhang, Y., Chen, Y., Su. S., Lin,
- N., Liu, J., Li, B., Wang, X., Coveney Jr., R.M., and Tao, S.: Global mercury emissions from
- combustion in light of international fuel trading, Environ. Sci. Technol., 48, 1727-1735, 2014.
- 25 Deeds, D.A., Banic, C.M., Lu, J., and Daggupaty, S.: Mercury speciation in a coal-fired
- 26 power plant plume: An aircraft-based study of emissions from the 3640 MW Nanticoke
- 27 Generatin Generating Station, Ontario, Canada, J. Geophys. Res., 118, 4919-4935, 2013.
- 28 Ebinghaus, R., Jennings, S.G., Schroeder, W.H., Berg, T., Donaghy, T., Guentzel, J., Kenny,
- 29 C., Kock, H.H., Kvietkus, K., Landing, T., Mühleck, T., Munthe, J., Prestbo, E.M.,
- 30 Schneeberger, D., Slemr, F., Sommar, J., Urba, A., Wallschläger, D., and Xiao, Z.:

- 1 <u>International field intercomparison measurements of atmospheric mercury species at Mace</u>
- 2 Head, Ireland, Atmos. Environ., 33, 3063-3073, 1999...
- 3 Ebinghaus, R. and Slemr, F.: Aircraft measurements of atmospheric mercury over southern
- 4 and eastern Germany, Atmos. Environ., 34, 895–903, doi:10.1016/S1352-2310(99)00347-7,
- 5 2000.
- 6 Ebinghaus, R., Slemr, F., Brenninkmeijer, C. A. M., van Velthoven, P., Zahn, A., Hermann,
- 7 M., O'Sullivan, D. A., and Oram, D. E.: Emissions of gaseous mercury from biomass burning
- 8 in South America in 2005 observed during CARIBIC flights, Geophys. Res. Lett., 34(8), 1–5,
- 9 doi:10.1029/2006GL028866, 2007.
- 10 Edgerton, E.S., Hartsell, B.E., and Jansen, J.J.: Mercury speciation in coal-fired power plant
- 11 plumes observed at three surface sites in the southeastern U.S., Environ. Sci. Technol., 40,
- 12 4563-4570, 2006.
- 13 EEA (European Environmental Agency): Revealing the Costs of Air Pollution from Industrial
- Facilities, Technical Report 15/2011, doi:10.2800/84800, Kopenhagen, 2011.
- 15 Friedli, H. R., Radke, L.F., Prescott, R., Li, P., Woo, J. H., and Carmichael, G.R.: Mercury in
- 16 the atmosphere around Japan, Korea, and China as observed during the 2001 ACE Asia field
- 17 campaign: Measurements, distributions, sources, and implications, J. Geophys. Res.,
- 18 109(D19), 1-13, doi:10.1029/2003JD004244, 2004.
- 19 EPA (Environmental Protection Agency): Electric Generating Utility Mercury Speciation
- 20 Profiles for the Clean Air Mercury Rule, EPA-454/R-11-010, November 2011. Available
- 21 online:
- 22 https://www3.epa.gov/ttn/chief/emch/speciation/EGU\_Hg\_speciation\_summary\_CAMR.pdf
- 23 Gustin, M.S., Huang, J., Miller, M.B., Peterson, C., Jaffe, D.A., Ambrose, J., Finley, B.D.,
- 24 Lyman, S.N., Call, K., Talbot, R., Feddersen, D., Mao, H., and Lindberg, S.E.: Do we
- 25 understand what the mercury speciation instruments are actually measuring? Results of
- 26 RAMIX, Environ. Sci. Technol., 47, 7295-7306, 2013.
- 27 Gustin, M.S., Amos, H.M., Huang, J., Miller, M.B., and Heidekorn, K.: Measuring and
- 28 modelling mercury in the atmosphere: a critical review, Atmos. Chem. Phys. 15, 5697-5713,
- 29 2015.

- 1 Holmes, C.D., Jacob, D.J., Corbitt, E.S., Mao, J., Yang, X., Talbot, R., and Slemr, F.: Global
- 2 atmospheric model for mercury including oxidation by bromine atoms, Atmos. Chem. Phys.,
- 3 10, 12037-12057, 2010.
- 4 Huang, J., and Gustin, M.S.: Uncertainties of gaseous oxidized mercury measurements using
- 5 KCl-coated denuders, cation-exchange membranes, and nylon membranes: Humidity
- 6 influences, Environ. Sci. Technol., 49, 6102-6108, doi:10.1021/acs.est.5b00098, 2015.
- 7 Jaffe, D.A., Lyman, S., Amos, H.M., Gustin, M.S., Huang, J., Selin, N.E., Levin, L., ter
- 8 Schure, A., Mason, R.P., Talbot, R., Rutter, A., Finley, B., Laeglé, L., Shah, V., McClure, C.,
- 9 Ambrose, J., Gratz, L., Lindberg, S., Weiss-Penzias, P., Sheu, G.-R., Feddersen, D., Horvat,
- 10 M., Dastoor, A., Hynes, A.J., Mao, H., Jonke, J.E., Slemr, F., Fisher, J.A., Ebinghaus, R.,
- 21 Zhang, Y., and Edwards, G.: Progress on understanding atmospheric mercury hampered by
- 12 uncertain measurements, Environ. Sci. Technol., 48, 7204-7206, doi:10.1021/es5026432,
- 13 2014.
- 14 Kaiser, R., and Gottschalk, G.: Elementare Tests zur Beurteilung von Meßdaten,
- 15 Hochschultaschenbücher, Band 774, Bibliographisches Institut, Mannheim, 1972.
- 16 Kos, G., Ryzhkov, A., Dastoor, A., Narayan, J., Steffen, A., Ariya, P.A., and Zhang, L.:...
- 17 Evaluation of discrepancy between measured and modelled oxidized mercury species, Atmos.
- 18 Chem. Phys., 13, 4839-4863, doi:10.5194/acp-13-4839-2013, 2013.
- 19 Lai, S.C., Baker, A.K., Schuck, T.J., Slemr, F., Brenninkmeijer, C.A.M., van Velthoven, P.,
- 20 Oram, D.E., Zahn, A., and Ziereis, H.: Characterization and source regions of 51 high-CO
- 21 events observed during the Civil Aircraft for the Regular Investigation of the atmosphere
- 22 Based on the Instrument Container (CARIBIC) flights between South China and the
- 23 Philippines, 2005-2008, J. Geophys. Res., 116, D20308, doi:10.1029/2011JD016375, 2011.
- 24 Landis, M.S., Ryan, J.V., ter Schure, A.F.H., and Laudal, D.: Behavior of mercury emissions
- 25 from a commercial coal-fired power plant: The relationship between stack speciation and
- 26 near-field plume measurements, Environ Sci. Technol., 48, 13540-13548, 2014.
- 27 Lindberg, S., Bullock, R., Ebinghaus, R., Engstrom, D., Feng, X., Fitzgerald, W., Pirrone, N.,
- 28 Prestbo, E. and Seigneur, C.: A synthesis of progress and uncertainties in attributing the
- 29 | sources of mercury in deposition, AMBIO, -36, 19–33, -2007.
- 30 Lohman, K., Seigneur, C., Edgerton, E., and Janssen, J.: Modeling mercury in power plant
- 31 plumes, Environ, Sci. Technol, 40, 3848-3854, 2006.

**Formatiert:** Schriftartfarbe: Automatisch

Formatiert: Leerraum zwischen asiatischem und westlichem Text nicht anpassen, Leerraum zwischen asiatischem Text und Zahlen nicht anpassen

- 1 Lyman, S.N., Jaffe, D.A., and Gustin, D.S.: Release of mercury halides from KCl denuders in
- 2 the presence of ozone, Atmos. Chem. Phys., 10, 8197-8204, doi:10.5194/acp-10-8197-2010,
- 3 2010.
- 4 Lyman, S.N., and Jaffe, D.A.: Formation and fate of oxidized mercury in the upper
- 5 troposphere and lower stratosphere, Nature Geosci., 5, 114-117, doi:10.1038/NGEO1353,
- 6 2012.
- 7 Mason, R.P.: Mercury emissions from natural processes and their importance in the global
- 8 mercury cycle, in Mercury Fate and Transport in the Global Atmosphere, eds. Pirrone, N.,
- 9 and Mason, R., Springer Dordrecht, 2009, pp. 173-191.
- 10 Mayer, J., Hopf, S., van Dijen, F., and Baldini, A.: Measurement of low mercury
- 11 concentrations in flue gases of combustion plants, VGB PowerTech Journal, 3/2014, 64-68,
- 12 2014.
- 13 Mergler, D., Anderson, H.A., Chan, L.H.N., Mahaffey, K.R., Murray, M., Sakamoto, M., and
- 14 Stern, A.H.: Methylmercury exposure and health effects in humans: A worldwide concern,
- 15 Ambio, 36, 3-11, 2007.
- Pacyna, E. G., Pacyna, J. M., Steenhuisen, F., and Wilson, S.: Global anthropogenic mercury
- 17 emission inventory for 2000, Atmos. Environ., 40, 4048 4063, 2006.
- 18 Pirrone, N., Cinnirella, S., Feng, X., Finkelman, R. B., Friedli, H. R., Leaner, J., Mason, R.,
- 19 Mukherjee, A. B., Stracher, G. B., Streets, D. G. and Telmer, K.: Global mercury emissions to
- the atmosphere from anthropogenic and natural sources, Atmos. Chem. Phys., 10, 5951–5964,
- 21 doi:10.5194/acp-10-5951-2010, 2010.
- 22 Preiss, P., Roos, J., and Friedrich, R.: Assessment of Health Impacts of Coal Fired Power
- 23 Stations in Germany, report by the Institute for Energy Economics and Rational Use of
- 24 Energy (IER), University of Stuttgart, March 29<sup>th</sup>, 2013.
- 25 Radke, L. F., Friedli, H. R. and Heikes, B. G.: Atmospheric mercury over the NE Pacific
- during spring 2002: Gradients, residence time, upper troposphere lower stratosphere loss, and
- 27 long-range transport, J. Geophys. Res., 112(D19), 1–17, doi:10.1029/2005JD005828, 2007.
- 28 Rösler, H.J., Beuge, P., Schrön, W., Hahne, K., and Bräutigam, S.: Die anorganischen
- 29 Komponenten der Braunkohlen und ihre Bedeutung für die Braunkohlenerkundung,
- 30 Freiburger Forschungshefte, C331, 53-70, 1977.

- 1 Rutter, A.P., and Schauer, J.J.: The effect of temperature on the gas-particle partitioning of
- 2 reactive mercury in atmospheric aerosols, Atmos. Environ., 41, 8647-8657, 2007.
- 3 Scheele, M. P., Siegmund, P. C. and van Velthoven, P. F. J.: Sensitivity of trajectories to data
- 4 resolution and its dependence on the starting point: In or outside a tropopause fold, Meteorol.
- 5 Appl., 3(3), 267–273, doi:10.1002/met.5060030308, 2007.
- 6 Scheuhammer, A.M., Meyer, M.W., Sandheinrich, M.B., and Murray, M.W.: Effects of
- 7 environmental methylmercury on the health of wild birds, mammals, and fish, Ambio, 36, 12-
- 8 18, 2007.
- 9 Schofield, K.: Fuel-mercury combustion emissions: An important heterogeneous mechanism
- 10 and an overall review of its implications, Environ. Sci. Technol., 42, 9014-9030,
- 11 doi:10.1021/es801440g, 2008.
- 12 SchuetzeSchütze, J., Kunth, D., Weissbach S., and Koeser, H.: Mercury vapor pressure of flue
- 13 gas desulfurization scrubber suspensions: Effects of pH level, gypsum, and iron, Environ. Sci.
- 14 Technol., 46, 3008-3012, doi:10.1021/es203605h, 2012.
- 15 Schütze, J.: Quecksilberabscheidung in der nassen Rauchgasentschwefelung von
- 16 Kohlekraftwerken, Beiträge zum Umweltschutz, Band 6/2013, Shaker Verlag, Aachen, 2013.
- 17 Schütze, J., Schilling, U., Hilbert, L., Strauß, J.H., and Hörtinger, T.: Quecksilberabscheidung
- 18 am Beispiel des Kraftwerks Lippendorf, VGB Power Tech, 81-87, 2015.
- 19 Selin, N. E.: Global biogeochemical cycling of mercury: A review, Ann. Rev. Environ.
- 20 Resour., 34, 43–63, doi:10.1146/annurev.environ.051308.084314, 2009.
- 21 Sholupov, S., Pogarev, S., Ryzhov, V., Mashyanov, N., and Stroganov, A.: Zeeman atomic
- 22 absorption spectrometer RA-915+ for direct determination of mercury in air and complex
- 23 matrix samples, Fuel Process. Technol., 85, 473-485, 2004.
- 24 Slemr, F., Seiler, W., and Schuster, G.: Quecksilber in der troposphere Troposphere, Ber.
- 25 Bunsenges. Phys. Chem., 82, 1142-1146, 1978.
- 26 Slemr, F., Seiler, W., Eberling, C., and Roggendorf, P.: The determination of total gaseous
- 27 mercury in air at background levels, Anal. Chim. Acta., 110, 35-47, 1979.
- 28 Slemr, F., Schuster, G., and Seiler, W: Distribution, speciation, and budget of atmospheric
- 29 mercury, J. Atmos. Chem., 3, 407-434, 1985.

Formatiert: Englisch (USA)

- 1 Slemr, F., Ebinghaus, R., Brenninkmeijer, C.A.M., Hermann, M., Kock, H.H., Martinsson,
- 2 B.G., Schuck, T., Sprung, D., van Velthoven, P., Zahn, A., and Ziereis, H.: Gaseous mercury
- 3 distribution in the upper troposphere and lower stratosphere observed onboard the CARIBIC
- 4 passenger aircraft, Atmos. Chem. Phys, 9, 1957-1969, 2009.
- 5 Slemr, F., Weigelt, A., Ebinghaus, R., Brenninkmeijer, C., Baker, A., Schuck, T., Rauthe-
- 6 Schöch, A., Riede, H., Leedham, E., Hermann, M., van Velthoven, P., Oram, D., O'Sullivan,
- 7 D., Dyroff, C., Zahn, A. and Ziereis, H.: Mercury plumes in the global upper troposphere
- 8 observed during flights with the CARIBIC observatory from May 2005 until June 2013,
- 9 Atmosphere (Basel)., 5(2), 342–369, doi:10.3390/atmos5020342, 2014.
- 10 Slemr, F., Weigelt, A. Ebinghaus, R., Kock, H.H., Bödewadt, J. Brenninkmeijer, C.A.M.,
- 11 Rauthe Schöch, A., Weber, S., Hermann, M., Becker, J., Zahn, A., and Martinsson, B.:
- 12 Atmospheric mercury measurements onboard the CARIBIC passenger aircraft, Atmos-
- 13 Measur. Techn. Discuss., doi:10.594/amt-2015-376, 2016.
- 14 Spencer, R. W. and Braswell, W. D.: How dry is the tropical free troposphere? Implications
- 15 for global warming theory, Bull. Am. Meteorol. Soc., 78, 1097-1106, doi:10.1175/1520-
- 16 0477(1997)078<1097:HDITTF>2.0.CO;2, 1996.
- 17 Stergašek, A., Horvat, M., Kotnik, J., Tratnik, J., Frkal, P., Kocman, D., Jaćimović, R., Fajon,
- 18 V., Ponikvar, M., Hrastel, I., Lenart, J., Debeljak, B., and Čujež, M.: The role of flue gas
- desulphurisation in mercury speciation and distribution in a lignite burning power plant, Fuel,
- 20 87, 3504-3512, 2008.
- 21 Swartzendruber, P. C., Jaffe, D. A., Prestbo, E. M., Weiss Penzias, P., Selin, N. E., Park, R.,
- 22 Jacob, D. J., Strode, S. and Jacglé, L.: Observations of reactive gaseous mercury in the free
- 23 troposphere at the Mount Bachelor Observatory, J. Geophys. Res., 111,
- 24 doi:10.1029/2006JD007415, 2006.
- 25 Swartzendruber, P. C., Chand, D., Jaffe, D. A., Smith, J., Reidmiller, D., Gratz, L., Keeler, J.,
- 26 Strode, S., Jaeglé, L. and Talbot, R.: Vertical distribution of mercury, CO, ozone, and aerosol
- 27 scattering coefficient in the Pacific Northwest during the spring 2006 INTEX B campaign, J.
- 28 Geophys. Res., 113, doi:10.1029/2007JD009579, 2008.
- 29 Swartzendruber, P. C., Jaffe, D. A. and Finley, B.: Development and first results of an
- 30 aircraft based, high time resolution technique for gaseous elemental and reactive (oxidized)
- 31 gaseous mercury, Environ. Sci. Technol., 43(19), 7484-7489, doi:10.1021/es901390t, 2009.

- 1 Tatum Ernest, C., Donohue, D., Bauer, D., Ter Schure, A., and Hynes, A.J.: Programmable
- 2 thermal dissociation of reactive gaseous mercury, a potential approach to chemical speciation:
- 3 Results from a field study, Atmosphere, 5, 575-596, doi:10.3390/atmos5030575, 2014.
- 4 VBG: VGB Initiative "Hg<sup>cap</sup>": Further reduction of mercury emissions from coal-fired power
- 5 plants, Position paper, VBG, Essen, March 2016.
- 6 Wang, S.X., Zhang, L., Li, G.H., Wu, Y., Hao, J.M., Pirrone, M., Sprovieri, F., and Ancora,
- 7 M.P.: Mercury emission and speciation of coal-fired power plants in China, Atmos. Chem.
- 8 Phys., 10, 1183-1192, 2010.
- 9 Weigelt, A., Temme, C., Bieber, E., Schwerin, A., Schuetze, M., Ebinghaus, R. and Kock, H.
- 10 H.: Measurements of atmospheric mercury species at a German rural background site from
- 11 2009 to 2011 methods and results, Environ. Chem., 10(2), 102–110, doi:10.1071/EN12107,
- 12 2013.
- Weigelt, A., Ebinghaus, R., Pirrone, N., Bieser, J., Bödewadt, J., Esposito, G., Slemr, F., van
- 14 Velthoven, P.F.J., Zahn, A., and Ziereis, H.: Tropospheric mercury vertical profiles between
- 15 500 and 10000 m in central Europe, Atmos. Chem. Phys., 16, 4135-4146, 2016.
- Weiss-Penzias, P., Amos, H.M., Selin, N.E., Gustin, M.S., Jaffe, D.A., Obrist, D.,
- 17 Sheu, G.-R., and Giang, A.: Use of a global model to understand speciated
- 18 atmospheric mercury observations at five high-elevation sites, Atmos. Chem. Phys.,
- 19 15, 2225-2225, doi:10.5194/acp-15-2225-2015, 2015.
- 20 Wilson, S., Munthe, J., Sundseth, K., Kindbom, K., Maxson, P., Pacyna, J., and
- 21 Steenhuisen, F.: Updating Historical Global Inventories of Anthropogenic Mercury Emissions
- 22 to Air. Arctic Monitoring and Assessment Programme (AMAP) Technical Report No. 3.,
- 23 AMAP Secretariat, Oslo; Norway, 2010.
- Wilson, S., Kindborn, K., Yaramenka, K., Steenhuisen, F., Telmer, K., and Munthe, J.: Global
- 25 emission of mercury to the atmosphere, in AMAP/UNEP, Technical Background Report for
- 26 the Global Mercury Assessment, Arctic Monitoring and Assessment Programme, Oslo,
- 27 Norway/UNEP Chemicals Branch, Geneva, Switzerland, 2013.
- 28 Yudovich, Y.E., and Ketris, M.P.: Mercury in coal a review: Part 1: Geochemistry, Int. J.
- 29 Coal Geology, 62, 107-134, 2005.

- 1 Zhang, L., Blanchard, P., Gay, D.A., Presbo, E.M., Risch, M.R., Johnson, D., Narayan, J.,
- 2 Zsolway, R., Holsen, T.M., Miller, E.K., Castro, M.S., Graydon, J.A., St. Louis, V.L., and
- 3 Dalziel, J.: Estimation of speciated and total mercury dry deposition at monitoring locations
- 4 in eastern and central North America, Atmos. Chem. Phys., 12, 4327-4340, 2012a.
- 5 Zhang Y., Jaegle L., van Donkelaar A., Martin R.V., Holmes C.D., Amos, H.M., Wang, Q.,
- 6 Talbot, R., Artz, R., Brooks, S., Luke, W., Holsen, T.M., Felton, D., Miller, E.K., Perry, K.D.,
- 7 Schmeltz, D., Steffen, A., Tordon, R., Weiss-Penzias, P., and Zsolway, R.: Nested-grid
- 8 simulation of mercury over North America, Atmos. Chem. Phys., 12, 6095-6111,
- 9 doi:10.5194/acp-12-6095-2012, 2012b.

# 1 Tables

Table 1: List of instruments, installed into in the CASA 212 research aircraft. The acronyms

3 are: GEM = gaseous elemental mercury; GOM = gaseous oxidized mercury.

Parameter	Instrument name	Temporal resolution	Uncertainty	Lower detection limit
GEM	Lumex RA-915AM (modified, T-stabilised by Lumex company)	1 sec (raw signal)	$\pm 4 \text{ ng/m}^3$ (1 s raw signal) $\pm 1 \text{ ng/m}^3$ (10 s average)	0.5 ng/m <sup>3</sup> (120 s average)
GEM	Tekran: 2537X (with upstream quartz wool trap)	150 s	±12.5% of reading	0.1 ng·m <sup>-3</sup>
GEM + unknown amount of GOM*	Tekran 2537B	150 s	±12.5% of reading	0.1 ng·m <sup>-3</sup>
GOM	manually denuder samples	2600 to 3600 s	±5 pg·m <sup>-3</sup> **	1 pg·m <sup>-3</sup>
СО	Aero Laser AL5002	1 s	±3% of reading	1.5 ppb
O <sub>3</sub>	Teledyne API 400E	10 s	±2% of reading	0.6 ppb
SO <sub>2</sub>	Thermo: 43C Trace Level	10 s	±4% of reading	0.2 ppb
NO NO <sub>2</sub>	Teledyne API M200AU	10 s 10 s	±10% of reading	0.05 ppb
Pressure	Sensor Technics CTE7001	1 s	±1% of reading	0 mbar
Temperature	LKM Electronic DTM5080	1 s	±0.13°C	-50°C
Relative Humidity (rH)	Vaisala HMT333	8 s	±1.0% rH (0-90% rH)	0%
			±1.7% rH (90-100% rH)	
GPS data (3d position, speed, heading)	POS AV	1 s	±5 m (horizontal)*** ±15 (vertical)***	

<sup>4 \*</sup> The aircraft inlet system transmission efficiency for GOM was not tested because no GOM sources

<sup>5</sup> were available which would enable for measurements during the flight.

<sup>6 \*\*</sup> Difference of the two blank tests

<sup>7 \*\*\*</sup> The GPS accuracy is dependent on the number of satellites. The given numbers are estimated

<sup>8</sup> values.

Table 2: Hg/SO<sub>2</sub> enhancement ratios (ERs). Correlation method: 10 s average Hg 1 concentrations measured by Lumex correlated with 10 s average SO<sub>2</sub> mixing ratios, only Hg 2 values with SO<sub>2</sub> concentrations > 10 ppb were taken, uncertainties set to 1 ng m<sup>-3</sup> for Lumex 3 and 0.5 ppb for SO<sub>2</sub>. Integral method: 1 s Lumex and SO<sub>2</sub> signals integrated over the duration 4 of Lumex measurement, measurements of Tekran with quartz wool taken as Lumex 5 background concentrations (i.e. 1.27 and 1.25 ng m<sup>-3</sup> for August 21 and 22, respectively). SO<sub>2</sub> 6 7 background mixing ratio was 0.83 and 0.66 ppb on August 21 and 22, respectively. Since 8 Tekran with a temporal resolution of 150 s cannot resolve individual plume crossing, the 9 integral of the Tekran signal encompasses the plumes 1 – 4 on August 21 and the plumes 1 – 10 6 on August 22.

Date	Method	Species	ER 10 <sup>-6</sup> mol mol <sup>-1</sup>	n, R, signif	Comment
A + 21	1	CEM		25.0.6564	
August 21, 2013	correlation	GEM	$5.53 \pm 1.10$	35, 0.6564, >99.9%	
	integral peak 1	GEM	6.67		Lumex
					zeroing
	integral peak 2	GEM	5.72		
	integral peak 3	GEM	5.98		Lumex
					zeroing
	integral peak 4	GEM	3.88		
	integral peak 5	GEM	0.89		
	integral average	GEM	$5.56 \pm 1.19^*$	4*	
	Tekran with	GEM	6.56		
	quartz wool trap				
	Tekran without	TGM	7.55		
	quartz wool trap				
August 22, 2013	correlation	GEM	$7.38 \pm 0.92$	45, 0.7751, >99.9%	
	integral peak 1	GEM	6.44		
	integral peak 2	GEM	4.83		
	integral peak 3	GEM	5.90		Lumex
					zeroing
	integral peak 4	GEM	6.67		
	integral peak 5	GEM	9.03		Lumex
					zeroing
	integral peak 6	GEM	5.02		
	integral average	GEM	$6.32 \pm 1.52$	6	
	Tekran with	GEM	8.13		
	quartz wool trap				
	Tekran without	TGM	8.97		
	quartz wool trap				
2013	reported annual	TGM	10.8		
	emissions				

Formatiert: Englisch (USA)

Formatiert: Englisch (USA)

Formatiert: Englisch (USA)

Formatiert: Englisch (USA)

- $^*$  average without integral of peak 5 which is identified as outlier by Nalimov test (at >95%
- 3 significance level, Kaiser and Gottschalk, 1972)

Table 3: Hg/CO enhancement ratios (ERs). Correlation method: 10 s average Hg concentrations measured by Lumex correlated with 10 s average CO mixing ratios for SO<sub>2</sub> mixing ratios above 10 ppb, uncertainties set to 1 ng m<sup>-3</sup> for Lumex and 1 ppb for CO. Integral method: 1 s Lumex and CO signals integrated over the duration of Lumex measurement, Tekran 1 readings of Tekran with quartz wool taken as Lumex background concentrations (i.e. 1.27 and 1.25 ng m<sup>-3</sup> for August 21 and 22, respectively). CO background mixing ratio was 119.3 ppb on August 21 and 123.8 ppb on August 22.

Date	Method	ER (Hg/CO)		Comment
		10 <sup>-5</sup> mol mol <sup>-1</sup>	n, R, signif	
August 21, 2013	Correlation	$5.19 \pm 0.94$	31, 0.6596,	values only until
			>99.9%	10:40:20
	integral peak 1	3.40		Lumex zeroing
	integral peak 2	4.16		
	integral peak 3	3.33		Lumex zeroing
	integral peak 4			background
				change
	integral peak 5			CO calibration
	integral average	$3.63 \pm 0.46$	3	
August 22, 2013	Correlation	$9.43 \pm 1.07$	37, 0.7880,	
			>99.9%	
	integral peak 1	3.19		
	integral peak 2			CO calibration
	integral peak 3			Lumex zeroing,
				CO calibration
	integral peak 4	7.87		
	integral peak 5	5.61		Lumex zeroing
	integral peak 6	4.75		
	integral average	$5.36 \pm 1.95$	4	
2013	reported annual	7.58		
	emissions			

- 1 Table 4: NOx/SO<sub>2</sub> enhancement ratios (ERs). Correlation method: 10 s average NOx mixing
- 2 ratios correlated with 10 s average SO<sub>2</sub> mixing ratios above 10 ppb, uncertainties set to 1 ppb
- 3 for NOx and 0.5 ppb for SO<sub>2</sub>. Integral method: 1 s NOx and 1 s SO<sub>2</sub> signals integrated over
- 4 the duration of the individual plume intersection, background mixing ratios for SO<sub>2</sub> and NOx
- 5 are 0.83 and 1.78 ppb, respectively, for August 21 and 0.66 and 0.45 ppb, respectively for
- 6 August 22.

Date	Method	ER (NOx/SO <sub>2</sub> )		Comment
		mol mol <sup>-1</sup>	n, R, signif	
August 21, 2013	Correlation	$0.585 \pm 0.038$	34, 0.9379,	
			>99.9%	
	integral peak 1	0.598		
	integral peak 2	0.575		
	integral peak 3	0.725		
	integral peak 4	0.497		
	integral peak 5			
	integral average	$0.598 \pm 0.095$	4	
August 22, 2013	Correlation	$0.262 \pm 0.051$	40, 0.6344,	
			>99.9%	
	integral peak 1	0.297		
	integral peak 2	0.457		
	integral peak 3	0.167		Lumex zeroing
	integral peak 4	0.330		
	integral peak 5	0.133		Lumex zeroing
	integral peak 6	0.317		
	integral average	$0.284 \pm 0.118$	6	
2013	reported annual	0.910		
	emissions			

- 1 Table 5: O<sub>3</sub>/NOx enhancement ratios (ERs). Correlation method: 10 s average O<sub>3</sub> mixing
- 2 ratios correlated with 10 s average SO<sub>2</sub> mixing ratios above 10 ppb, uncertainties set to 1 ppb
- 3 for O<sub>3</sub> and 1 ppb for NOx. Integral method: 1 s O<sub>3</sub> and 1 s NOx signals integrated over the
- 4 duration of the individual plume intersection, background mixing ratios for O<sub>3</sub> and NOx are
- 5 43.09 and 1.78 ppb, respectively, for August 21. Individual O<sub>3</sub> background mixing ratios
- 6 (average of background before and after the peak) varying between 53.9 ppb for peak 1 to
- 7 56.2 ppb for peak 4 were taken for August 22. The NOx background mixing ratio on August
- 8 22 was 0.45 ppb.

Date	Method	ER (O <sub>3</sub> /NOx)		Comment
		mol mol <sup>-1</sup>	n, R, signif	
August 22, 2013	Correlation	$-0.620 \pm 0.134$	40, -0.3776,	
			>95%	
	integral peak 1	-0.979		
	integral peak 2	-0.424		
	integral peak 3	-1.527		
	integral peak 4	-0.686		
	integral peak 5	-2.059		
	integral peak 6	-0.568		
	integral average	$-1.040 \pm 0.633$	6	

1 Table 6: Results of the manual KCl denuder samples during all ETMEP-2 measurement

flights in 2013 over central Europe. GOM data were corrected for denuder blank test,

additionally performed blanks determined over Iskraba/Slovenia and Waldhof/Germany.

4 GOM concentrations are given as a centre of an estimated uncertainty range (in brackets) and

are given at standard temperature and pressure (STP; T=273.15 K, p=1013.25 hPa).

6

7

11

12

5

2

Date	Location	Profile character (relative sampling time in PBL* and FT** air	analysed GOM concentration [pg m <sup>-3</sup> ]
2013-08-21	Lippendorf/Germany	vertical (76% PBL: 24% FT)	11.4 (7.0-15.7)
2013-08-21	Leipzig/Germany	vertical (61% PBL; 39% FT)	5.8 (1.0*** - 10.6)
2013-08-22	Waldhof/Germany	vertical (54% PBL; 46% FT)	31.0 (24.6-37.3)

<sup>\*</sup> planetary boundary layer (PBL)

<sup>8 \*\*</sup> free troposphere (FT)

<sup>9 \*\*\*</sup>If a concentration was found to be below the method lower detection limit of 1.0 pg m<sup>-3</sup>,

<sup>10</sup> the lower detection limit is given.

# **Figures**

Humidity and temperature sensor

Flow direction

Figure 1: For the ETMEP-2 campaign in August 2013 the CASA 212 (a) from the Italian company Compagnia Generale Ripreseaeree (<a href="http://www.terraitaly.it/">http://www.terraitaly.it/</a>) was equipped with specially designed and manufactured trace gas inlet (b).



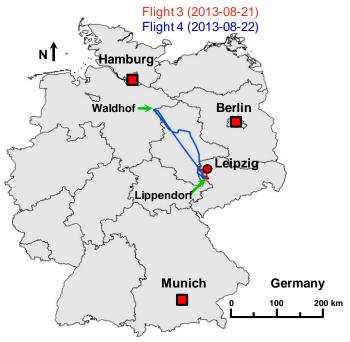
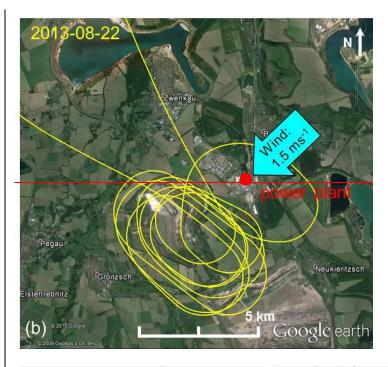


Figure 2: Flight tracks of the ETMEP-2 measurement flights number 3 and 4 over Central and northern Germany. The flights were made from the Leipzig airport.



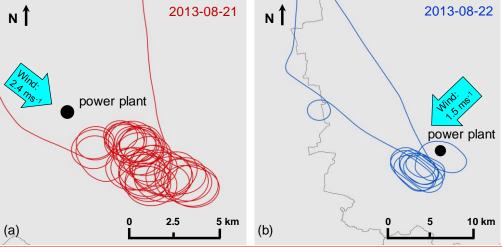
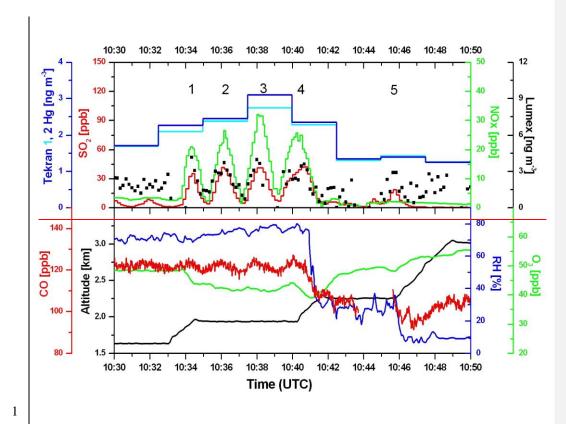


Figure 23: Flight tracktracks of the ETMEP-2 flights on August 21 (a) and 22 (b), 2013, downwind of the coallignite fired power plant "Lippendorf", south of Leipzig, Germany. On both flights the power plant plume was crossed several times.



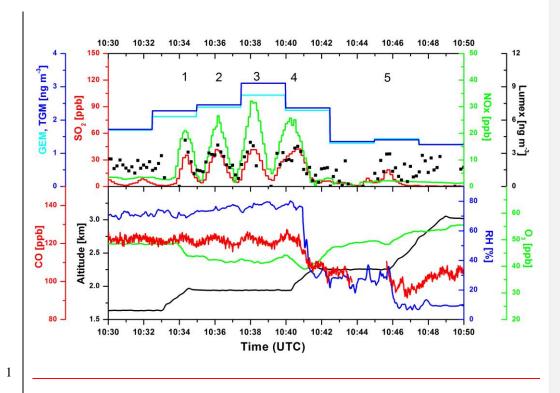
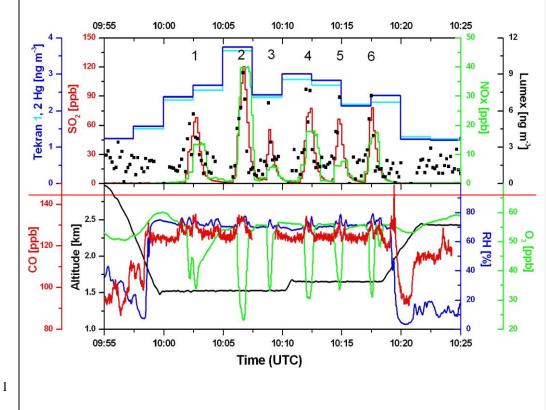


Figure 34: ETMEP-2 eoallignite fired power plant plume measurements on August 21, 2013 south of Leipzig/Germany. -The gaps in the Lumex signal (10 s resolution) are due to internal zero air checks for the correction of the instruments base line drift. GEM was measured using Tekran 1 wasinstrument run with quartz wool trap at the inlet of the instrument which is presumed to remove GOM<sub>7</sub>. TGM was measured by another Tekran 2 without. Tekran 1 and 2 measurements are thus presumed to represent GEM and TGM measurements, respectively instrument with no quartz wool trap at the inlet. All parameters were synchronized using individual instrument delay and response times. All Hg concentrations are given at standard temperature and pressure (STP; T=273.15 K, p=1013.25 hPa).



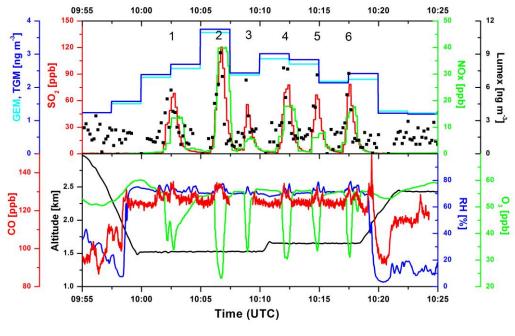


Figure 4: ETMEP 2 coal fired power plant plume measurements on August 22, 2013 south of Leipzig/Germany. The gaps in the Lumex signal (10 s resolution) are due to internal zero air checks for the correction of the instruments base line drift. Tekran 1 was run with quartz wool trap at the inlet of the instrument presumed to remove GOM, Tekran 2 without. Tekran 1 and 2 measurements are thus presumed to represent GEM and TGM measurements, respectively. All parameters were synchronized using individual instrument delay and response times. All Hg concentrations are given at standard temperature and pressure (STP; T=273.15 K, p=1013.25 hPa).

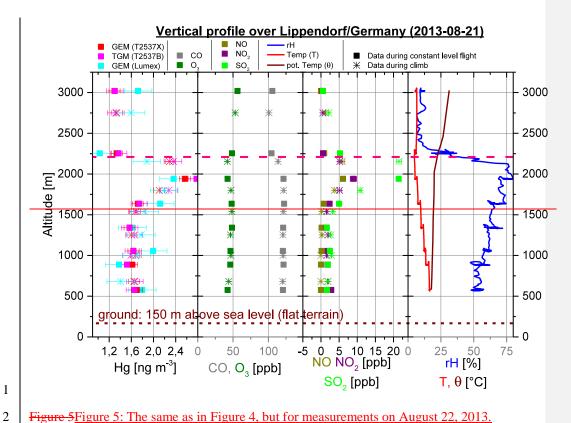
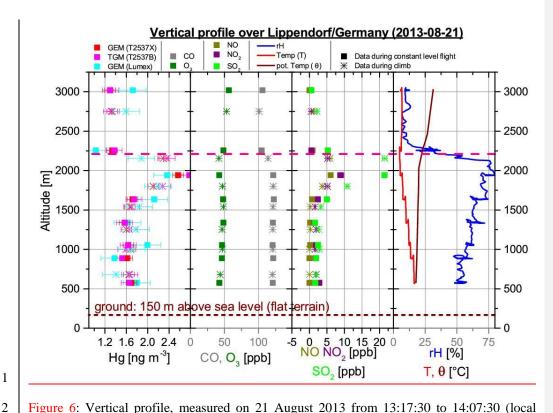


Figure 5: The same as in Figure 4, but for measurements on August 22, 2013.



<u>Figure 6</u>: Vertical profile, measured on 21 August 2013 from 13:17:30 to 14:07:30 (local time) downwind the coal fired power plant Lippendorf (central Germany; 45.561°N, 14.858 °E, elevation: 150 m a.s.l.; flat terrain). Squares represent 300 s averages with horizontal flight leg; stars indicate 150 s averages during climbing between two neighbouring flight legs. The red dashed line indicates the planetary boundary layer (PBL) top, which was determined to be at 2150 to 2250m a.s.l.. All Hg concentrations are given at standard temperature and pressure (STP; T=273.15 K, p=1013.25 hPa).