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2 **Designing global climate and atmospheric chemistry simulations for 1 km and 10**
3 **km diameter asteroid impacts using the properties of ejecta from the K-Pg impact**

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9

10 **Abstract.** About 66 million years ago an asteroid about 10 km in diameter struck the
11 Yucatan Peninsula creating the Chicxulub crater. The crater has been dated and found to
12 be coincident with the Cretaceous-Paleogene (K-Pg) mass extinction event, one of 6 great
13 mass extinctions in the last 600 million years. This event precipitated one of the largest
14 episodes of rapid climate change in Earth history, yet no modern three-dimensional
15 climate calculations have simulated the event. Similarly, while there is an on-going effort
16 to detect asteroids that might hit Earth and to develop methods to stop them, there have
17 been no modern calculations of the sizes of asteroids whose impacts on land would cause
18 devastating effects on Earth. Here we provide the information needed to initialize such
19 calculations for the K-Pg impactor and for a 1 km diameter impactor.

20 There is considerable controversy about the details of the events that followed the
21 Chicxulub impact. We proceed through the data record in the order of confidence that a
22 climatically important material was present in the atmosphere. The climatic importance
23 is roughly proportional to the optical depth of the material. Spherules with diameters of
24 several hundred-microns are found globally in an abundance that would have produced
25 an atmospheric layer with an optical depth around 20, yet their large sizes would only
26 allow them to stay airborne for a few days. They were likely important for triggering
27 global wildfires. Soot, probably from global or near-global wildfires, is found globally in
28 an abundance that would have produced an optical depth near 100, which would
29 effectively prevent sunlight from reaching the surface. Nanometer sized iron particles are
30 also present globally. Theory suggests these particles might be remnants of the vaporized
31 asteroid and target that initially remained as vapor rather than condensing on the
32 hundred-micron spherules when they entered the atmosphere. If present in the greatest
33 abundance allowed by theory, their optical depth would have exceeded 1000. Clastics
34 may be present globally, but only the quartz fraction can be quantified since shock
35 features can identify it. However, it is very difficult to determine the total abundance of
36 clastics. We reconcile previous widely disparate estimates and suggest the clastics may
37 have had an optical depth near 100. Sulfur is predicted to originate about equally from
38 the impactor and from the Yucatan surface materials. By mass, sulfur is less than 10
39 percent of the observed mass of the spheres and estimated mass of nano-particles. Since
40 the sulfur probably reacted on the surfaces of the soot, nano-particles, clastics and
41 spheres, it is likely a minor component of the climate forcing; however, detailed studies
42 of the conversion of sulfur gases to particles are needed to determine if sulfuric acid

43 aerosols dominated in late stages of the evolution of the atmospheric debris. Numerous
44 gases, including CO₂, SO₂ (or SO₃), H₂O, CO₂, Cl, Br, and I, were likely injected into the
45 upper atmosphere by the impact or the immediate effects of the impact such as fires
46 across the planet. Their abundance might have increased relative to current ambient
47 values by a significant fraction for CO₂, and by factors of 100 to 1000 for the other gases.

48 For the 1 km impactor, nano-particles might have had an optical depth of 1.5 if the
49 impact occurred on land. If the impactor struck a densely forested region, soot from the
50 forest fires might have had an optical depth of 0.1. Only S and I would be expected to be
51 perturbed significantly relative to ambient gas phase values. One kilometer asteroids
52 impacting the ocean may inject seawater into the stratosphere as well as halogens that are
53 dissolved in the seawater.

54 For each of the materials mentioned we provide initial abundances and injection altitudes.
55 For particles we suggest initial size distributions and optical constants. We also suggest
56 new observations that could be made to narrow the uncertainties about the particles and
57 gases generated by large impacts.

58

59 **Keywords** Climate modeling; Initial conditions; Asteroid impacts; K-Pg extinction

60

61 **1. Introduction and definitions**

62 About 66 million years ago an asteroid around 10 km in diameter hit the Earth near the
63 present day Yucatan village of Chicxulub and created an immense crater whose age
64 coincides with the Cretaceous-Paleogene (K-Pg) global mass extinction (Alvarez et al.,
65 1980; Schulte et al., 2010; Renne et al., 2013). There is an enormous literature
66 concerning this event and its aftermath. Surprisingly, however, there are very few papers
67 about the changes in climate and atmospheric chemistry caused by the debris from the
68 impact while it was in the atmosphere, and no studies based on modern three-dimensional
69 climate models. Nevertheless, this event was almost certainly one of the largest and most
70 dramatic short-term perturbations to climate and atmospheric chemistry in Earth history.

71 There is substantial evidence for many other impacts in Earth history as large or larger
72 than that at Chicxulub, mostly in the Pre-Cambrian (e.g. Johnson and Melosh, 2012a;
73 Glass and Simonson, 2012). There is also a growing effort to find asteroids smaller than
74 the one that hit Chicxulub, but whose impact might have significant global effects, and to
75 develop techniques to stop any that could hit the Earth. For example, as of November 17,
76 2015 NASA's Near Earth Object Program identifies 13,392 objects whose orbits pass
77 near Earth. Among these objects, 878 have a diameter of about 1 km or larger, and 1640
78 have been identified as Potentially Hazardous Asteroids, which are asteroids that pass the
79 Earth within about 5% of Earth's distance from the sun, and are larger than about 150 m
80 diameter.

81 There is evidence for such smaller impacts in recent geologic history from craters,
82 osmium variations in sea cores (Paquay et al., 2008), and spherule layers (Johnson and
83 Melosh, 2012a; Glass and Simonson, 2012). For instance, a multi-kilometer object
84 formed the Siberian Popigai crater in the Late Eocene and another multi-kilometer object

85 formed the Late Eocene Chesapeake Bay crater in the United States. Size estimates vary
86 between techniques, but within a given technique the Popigai object is generally given a
87 diameter half that of the Chicxulub object. Toon et al. (1997) point out that the
88 environmental effects of impacts scale with the impactor energy, or cube of the diameter,
89 not diameter (or crater size). The Popigai object likely had about 12% of the energy of
90 the Chicxulub object. Surprisingly, except for collisions in the ocean (Pierazzo et al. ,
91 2010), climate models have not been used to determine the destruction that might be
92 caused by objects near 1 km in diameter, a suggested lower limit to the size of an
93 impactor that might do significant worldwide damage (e.g. Toon et al., 1997).

94 Here we describe the parameters that are needed to initialize three-dimensional climate
95 and atmospheric chemistry models for the Chicxulub impact and for a 1-km diameter
96 asteroid impact. Nearly every aspect of the K-Pg impact event is uncertain, and
97 controversial. We will address some of these uncertainties and controversies and make
98 recommendations for the initial conditions that seem most appropriate for a climate
99 model, based upon the geological evidence. We will also suggest the properties of the
100 initial impact debris from a 1 km diameter asteroid.

101 There are numerous observed and predicted components of the Chicxulub impact debris.
102 The distal debris layer, defined to be the debris that is more than 4000 km removed from
103 the impact site, is thought to contain material that remained in the atmosphere long
104 enough to be globally distributed. This distal layer, sometimes called the fireball layer or
105 the magic layer, is typically only a few mm thick (Smit, 1999). As discussed below, the
106 layer includes 200 μm -sized spherules, 50 μm -sized shocked quartz grains, 0.1- μm -sized
107 soot and a 20 nm-sized iron-rich material.

108 We discuss each of the components of the distal layer in detail below. In brief, we find
109 the following: The large spherules are not likely to be of importance to the climate
110 because they would have been removed from the atmosphere in only a few days.
111 However, they may have initiated global wildfires. The shocked quartz grains, one of the
112 definitive pieces of evidence for an impact origin as opposed to volcanic origin of the
113 debris layer, is likely only a small fraction of the clastic debris. It is difficult to identify
114 the rest of the minerals produced by crushing because there is material in the layer that
115 might have been produced long after the impact by erosion and chemical alteration of the
116 large spheres or from the ambient environment. One major controversy surrounding the
117 clastic material is the fraction that is submicron-sized. Particles larger than a micron will
118 not remain in the atmosphere very long and, therefore, are less likely to affect climate.
119 Unfortunately, the sub- μm portion of the clastics in the distal layer, which might linger in
120 the atmosphere for a year or more, has not been directly measured. Our estimate of the
121 mass of submicron-sized clastics suggest that it could have had a very high large optical
122 depth that would be capable of modifying the climate significantly. Nevertheless,
123 submicron clastics are only of modest climatic importance relative to the light absorbing
124 soot and possibly the iron rich nm-scale debris. Submicron soot is observed in the global
125 distal layer in such quantity that it would have had a very great impact on the climate
126 when it was suspended in the atmosphere. The major controversy surrounding the soot is
127 whether it originated from forest fires, or from hydrocarbons at the impact site. The
128 origin of the soot, however, is of secondary importance with regard to its effect on
129 climate. Since the soot layer overlaps the iridium layer in the distal debris it had to have

130 been created within a year or two of the impact, based on the removal time of small
131 particles from the atmosphere (and ocean), and could not have been the result of fires
132 long after the impact. The fireball layer is often colored red and contains abundant iron.
133 Some of the iron has been identified as part of a 20 nm-sized particle phase, possibly
134 representing a portion of the recondensed vaporized impactor and target. However,
135 relatively little work has been done on this material. Its abundance has not been
136 measured, but theoretical work suggests its mass could have been comparable to that of
137 the impactor. Therefore, the nm-sized particles could have been of great importance to
138 the climate. Each of the materials just described is present in the distal layer, and their
139 impacts on the atmosphere were likely additive.

140 There are several other possible components of the distal layer that have not been clearly
141 identified and studied as part of the impact debris, which we discuss below. Water,
142 carbon, sulfur, chlorine, bromine, and iodine were likely present in significant quantities
143 in the atmosphere after the impact. The Chicxulub impact occurred in the sea with depths
144 possibly ranging up to 1 km. The target sediments and the asteroid probably also
145 contained significant amounts of water. Water is an important greenhouse gas, and could
146 condense to form rain, which might have removed materials from the stratosphere.
147 Carbon is present in seawater, in many asteroids and in sediments. Injections as carbon
148 dioxide or methane might have led to an increased greenhouse effect. Sulfur is widely
149 distributed in the ambient environment, and is water-soluble. Therefore, it is difficult to
150 identify extraterrestrial sulfur in the debris layer. However, the impact site contains a lot
151 of sulfur, and asteroids also contain significant amounts of sulfur. Sulfur is noteworthy
152 because it is known to produce atmospheric particulates in today's atmosphere that alter
153 the climate. Chlorine, bromine and iodine can destroy ozone, and their effectiveness as
154 catalysts is enhanced by heterogeneous reactions on sulfuric acid aerosols.

155 In addition to the mm-thick distal layer, there is an intermediate region ranging from
156 2,500-4,000 km from the impact site with a debris layer that is several cm thick (Smit,
157 1999). This layer contains microtektites (molten rock deformed by passage through the
158 air), shocked quartz, as well as clastics such as pulverized and shocked carbonates. Most
159 of this layer originated from the target material in the Yucatan. It is of interest because,
160 like the debris clouds from explosive volcanic eruptions, components of this material
161 may have escaped from the region near the impact site to become part of the global debris
162 layer.

163 Properties of each of these materials need to be known in order to model their effects on
164 the climate and atmospheric chemistry realistically. These properties include the altitude
165 of injection, the size of the injected particles, the mass of injected particles or gases, the
166 density of the particles, and the optical properties of the injected particles and gases. Our
167 best estimates for these properties for the K-Pg impact are summarized in Table 1 for
168 particles and Table 2 for gases, and discussed for each material in Section 2. Tables 3
169 and 4 provide an extrapolation of these properties for an impact of a 1 km sized object.

170 While the mass of the injected material is useful as an input parameter to a model, the
171 optical depth of the particles is needed to quantify their impact on the atmospheric
172 radiation field and, therefore, on the climate. Hence, optical depth is a useful quantity to
173 compare the relative importance of the various materials to the climate. For a

174 monodisperse particle size distribution, the optical depth is given by $\tau = \frac{3Mq_{ext}}{4\rho r}$. Here M
175 is the mass of particles in a column of air (for example, g cm^{-2}), r is the radius of the
176 particles, ρ is the density of the material composing the particles, and q_{ext} is the optical
177 extinction efficiency at the wavelength of interest. The optical extinction efficiency is a
178 function of the size of the particles relative to the wavelength of light of interest, and of
179 the optical constants of the material. The optical extinction efficiency is computed
180 accurately in climate models. However, a rough value of q_{ext} for particles larger than 1
181 μm , is about 2 for visible wavelength light. We use this rough estimate for q_{ext} in Table
182 1 and Table 3 to calculate an optical depth for purposes of qualitatively comparing the
183 importance of the various types of injected particles. We assume in the heuristic
184 calculations of optical depth in Tables 1 and 3 that the particles have a radius of $1\mu\text{m}$
185 because smaller particles will quickly coagulate to a radius near $1\mu\text{m}$ given the large
186 masses of injected material. Particles smaller than $1\mu\text{m}$ would lead to a larger optical
187 depth than given in Tables 1 and 3.

188 Below we define the properties that are needed to perform climate or atmospheric
189 chemistry simulations for each material that might be important.

190

191 **2. Particulate Injections**

192 **2.1 Large spherules**

193 **2.1.1 Large spherules from the Chicxulub impact**

194 The most evident component of the distal and regional debris layers is spherical particles,
195 some of which are large enough to be seen with the naked eye. Due to their spherical
196 shape it is assumed that they are part of the melt debris from the impact or the condensed
197 vapor from the impact (Johnson and Melosh, 2012b; 2014). The particles are not thought
198 to have melted on reentry into the atmosphere since debris launched above the
199 atmosphere by the impact should not reach high enough velocities to melt when it
200 reenters the atmosphere. According to Bohor and Glass (1995) there are two types of
201 spherules, with differing composition and distribution. They identify Type 1 splash-form
202 spherules (tektites or microtektites) that occur in the melt-ejecta (basal or lower) layer of
203 the regional debris layer where it has a two-layered structure. These spherules are found
204 as far from the Chicxulub site as Wyoming, but generally do not extend beyond about
205 4000 km away from Chicxulub. While the type 1 particles are derived from silicic rocks,
206 they are also mixed with sulfur rich carbonates from the upper sediments in the Yucatan.
207 The Type 1 spherules are poor in Ni and Ir, and the lower layer is poor in shocked quartz,
208 consistent with their origin from the lower energy impact ejecta from the crater.
209 Generally, the debris layer within about 4000 km of the crater is almost entirely
210 composed of target material, rather than material from the impactor itself. Type 2
211 spherules, on the other hand, are found in the distal debris layer, and presumably formed
212 primarily from the condensation of rock vapor from the impactor and target (O'Keefe and
213 Ahrens, 1982; Johnson and Melosh, 2012b). There are sub-types of Type 2 spherules
214 that correspond to varying composition of the original source material. Type 2 spherules
215 occur in the upper layer in impact sites near Chicxulub, which merges into the fireball

216 layer at distal sites. The Type 2 spherules are rich in Ni and Ir, while the fireball layer is
217 rich in shocked quartz.

218

219 The formation of the spherical particles may depend on two different processes. Melosh
220 and Vickery (1991) describe one formation mechanism, probably occurring in less
221 heavily shocked portions of the target, when molten material decompresses until it
222 reaches a critical line at which it starts to boil. The gas drag from the rock vapor on the
223 molten rock spheres then tears apart the molten material, just as water droplets break
224 apart when they fall through air. The relative velocities of water drops in air and the melt
225 in vapor are similar, as are the surface tensions. As a result melt droplets are similar in
226 size to drizzle drops in light rain, near $250 \mu\text{m}$. According to Johnson and Melosh
227 (2012b) these spherical particles are most likely to be found within 4000 km of the
228 impact site, and to be chemically related to the target material, and not to the impactor.
229 Such materials are reported across North America as Type 1 spherules (Bohor et al.,
230 1987), sometimes referred to as microtektites. Since these spherules are not global, they
231 likely were not as relevant to climate as the Type 2 spherules.

232

233 Melt droplets can also form in heavily shocked parts of the impact debris as rock vapor
234 condenses to form melt in the fireball, which rises thousands of km above the Earth's
235 surface. These melt droplets form the Type 2 spherules. O'Keefe and Ahrens (1982) first
236 modeled this process, and deduced that particles near a few hundred microns in size
237 would form, as is observed. They also pointed out that the size of the spheres would be
238 proportional to the size of the impactor. Johnson and Melosh (2012b) recently
239 reconsidered this process for forming melt particles. They point out that the large
240 spherules contain iridium (e.g., Smit, 1999), which is consistent with them being
241 composed partially of the vaporized impactor. Their model of the formation and
242 distribution of these particles suggests the particles have a size that varies spatially over
243 the plume. Averaging over the simulated plume yields a mean size of $217 \mu\text{m}$ with a
244 standard deviation of about $47 \mu\text{m}$ for a 10 km diameter impactor hitting at 21 km s^{-1} .
245 From the two examples given by Johnson and Melosh (2012b) it appears that the standard
246 deviation is consistently 22% of the mean radius for asteroids of different sizes. The
247 initial values for the various properties of Type 2 spherules described above are
248 summarized in Table 1 for the K-Pg impactor.

249

250 Smit (1999), who refers to the Type 2 spherules in the distal layer as microkrystites,
251 estimated that these particles typically have a diameter near $250 \mu\text{m}$, and a surface
252 concentration of about 20,000 particles cm^{-2} over the Earth. Unfortunately, we are not
253 aware of studies that measure the dispersion of the size distribution, or the spatial
254 variation of the abundance of these particles. We assume that the particles have the
255 density of CM2 asteroids, since Cr isotope ratios suggest that is the composition of the K-
256 Pg impactor (Trinquier et al., 2006). Assuming this density, $\sim 2.7 \text{ g cm}^{-3}$, the mass of
257 spherules per unit area of the Earth is about 0.4 g cm^{-2} , and the initial optical depth is
258 about 20, as noted in Table 1. These spherules compose about half of the mass of the
259 distal layer. We assume the particles were initially distributed uniformly around the
260 globe, with the initial mixing ratio in the atmosphere varying only in altitude. Some
261 theoretical studies, such as Kring and Durda (2002) and Morgan et al. (2013), suggest

262 that these particles were not uniformly deposited in latitude and longitude, but had
263 focusing points such as the antipodes of the impact site. Unfortunately, we are not aware
264 of quantitative data on the global distribution of the spherules. The study by Morgan et
265 al. (2013) may also be more applicable to the Type 1 spherules since their numerical
266 model does not produce vaporized material from the asteroid impact.

267

268 According to the simulations of Goldin and Melosh (2009), the in-falling spherical
269 particles reached terminal fall velocity near 70km altitude, at which point they begin to
270 behave like individual airborne particles. Kalasnikova et al. (2000) investigated
271 incoming micrometeorites in the present atmosphere, which generally ablate near 85 km.
272 Kalasnikova et al. (2000) find material entering from space stops in the atmosphere after
273 it encounters a mass of air approximately equal to its own mass. Therefore, the altitude
274 distribution is taken to be Gaussian, centered at 70 km and with a half-width of one
275 atmospheric scale height (about 6.6 km based on the U.S. Standard Atmosphere). A scale
276 height is chosen as the half width of the injection profile since it is a natural measure of
277 the density of the atmosphere. Figure 1 illustrates the vertical injection profile of the
278 spherules (green curve). As discussed below we expect several materials with origins
279 similar to those of the spherules to be injected in this same altitude range, but others with
280 origins unrelated to the impact generated plume, such as soot from fires, to be injected at
281 lower altitudes.

282 The 70 km injection altitude refers to the level at which the large spherical
283 particles reached terminal velocity. However, as is evident from the optical depth, many
284 spherules entered through the same air mass. The column mass of the distal layer is $\sim 1\text{g}$
285 cm^{-2} so the air pressure needs to about 1 hPa for the air mass above the altitude in
286 question and the particle mass to be comparable. A pressure of 1hPa occurs at about 48
287 km. Therefore, if the entire distal layer mass is placed into a model above 48 km its mass
288 mixing ratio will be greater than 1, and the atmosphere will be significantly out of
289 hydrostatic balance. We are not aware of any simulations of the first few hours after the
290 impact, but significant turbulence and mixing must have occurred as the atmosphere
291 adjusted to the large mass imbalance. Model initialization should be checked to
292 determine if the planned simulations start out of hydrostatic balance. If so, the injection
293 altitude should be lowered below 70 km.

294

295 The energy release from the reentry of the large spherical particles into the atmosphere
296 was likely responsible for setting most of the above ground terrestrial biosphere on fire.
297 However, due to their size, the spherules could not have remained in the atmosphere for
298 more than a few days. Hence they likely did not have a significant direct impact on the
299 climate, but fell to Earth like a gentle rain.

300

301 **2.1.2 Large spherules from a 1 km diameter asteroid impact**

302 Type 1 spherules, melt droplets, will form from impacts by 1 km diameter asteroids, and
303 produce mm-sized particles in the ejecta curtain layer located near the crater (Johnson
304 and Melosh, 2014). We do not expect an impact by a 1 km diameter asteroid to create a
305 global layer of Type 2 spherules (Toon et al., 1996). Like O’Keefe and Ahrens (1982),
306 Johnson and Melosh (2012b) conclude that the particle size will vary in proportion to the

307 impactor diameter and the impactor velocity. For a 1 km diameter impactor hitting the
308 land at 20 km/s they suggest that the mean diameter of the spherical particles will be
309 about 15 μm , with somewhat larger sizes as the impact velocity increases to 30 km/s.
310 Table 3 provides our assumed properties of the spherules from a hypothetical 1 km
311 diameter impactor hitting the land. It is likely that spherules would not be distributed
312 over all of the globe for the 1 km diameter impact. Johnson and Melosh (2012a) as well
313 as Glass and Simonson (2012) report a spherule layer associated with the Popagai impact
314 in the late Eocene which Johnson and Bowling (2014) suggest was global in extent. This
315 layer contains spherules similar in size or even larger than those associated with the
316 Chicxulub impact. However, this layer is only about 10% as thick as the distal layer from
317 the Chicxulub impact. A 1 km impactor hitting the deep oceans may not produce a layer
318 of spherules.

319

320 **2.2 Soot**

321 **2.2.1 Soot from the Chicxulub impact**

322 Spherical soot (also referred to as black carbon, or elemental carbon) particles were
323 discovered in the boundary layer debris at sites including Denmark, Italy, Spain, Austria,
324 Tunisia, Turkmenistan, the United States and New Zealand, among others by Wolbach et
325 al. (1985; 1988; 1990a, 1990b). Soot was also found in anaerobic deep-sea cores from the
326 mid-Pacific (Wolbach et al., 2003). Soot was apparently lost by oxidation in aerobic
327 deep-water sites in the 66 million years since emplacement. There is debate about
328 whether these particles originated from global wildfires, or from the impact itself
329 (Belcher et al., 2003, 2004, 2005, 2009; Belcher, 2009; Harvey et al., 2008; Robertson et
330 al., 2013a; Pierazzo and Artemieva, 2012; Premovic´ 2012; Morgan et al., 2013; Kaiho et
331 al., 2016). Robertson et al. (2013), Pierazzo and Artemieva (2012), Premovic´ (2012)
332 and Morgan et al. (2013) argue that it is implausible that there was enough carbon at the
333 impact site to produce the amount of soot observed by Wolbach et al. (1988). This
334 debate about the origin of the particles does not greatly affect the impact these particles
335 would have had on the climate when they were suspended in the atmosphere. The
336 particles are small and widely distributed. They are numerous and so must have produced
337 a very large optical depth and, being composed of carbon, they would have been
338 excellent absorbers of sunlight. Whether the soot particles originated from global fires
339 and were deposited in the upper troposphere, or they originated at the impact site and
340 were deposited in the mesosphere, the climate effect of the observed soot would have
341 been very great. Some have suggested that the soot resulted from wildfires in dead and
342 dying trees that occurred well after the impact. However, Wolbach et al. (1988; 1990b)
343 show that soot and iridium are tightly correlated and collocated. Indeed, Wolbach et al.
344 (1990b) suggest the soot and iridium may have coagulated in the atmosphere. The soot
345 and iridium in the distal layer must have been deposited within a few years of the impact,
346 since small particles will not stay in the air much longer. Therefore, any fires must have
347 been very close in time to the impact, and were likely contemporaneous.

348

349 Wolbach et al. (1988) estimated the global mass of elemental carbon (including aciniform
350 soot, charcoal and any unreactive aromatic kerogen) in the debris layer as $7\pm 4 \times 10^4$ Tg of
351 C or equivalently 13 ± 7 mg C cm^{-2} based on data from 5 sites. Wolbach et al. (1990b)

352 updated these mass determinations to $5.6 \pm 1.5 \times 10^4$ Tg or 11 ± 3 mg C cm⁻² based on data
353 from 11 sites. This mass of elemental carbon would require that the bulk of the above
354 ground biomass burned and was partially converted to elemental carbon with an efficiency
355 of about 3%, assuming the biomass is 1.5 g C cm⁻² of above ground, dry organic mass per
356 cm² over the land area of Earth. This biomass density is typical of current tropical forests.
357 This inferred 3% emission factor is about 60 times greater than that suggested by
358 Andreae and Merlet (2001) for current wildfires, but agrees with laboratory and other
359 observations from burning wood under conditions consistent with mass fires (Crutzen et
360 al., 1984; Turco et al., 1990). Mass fires are more intense than forest fires, and consume all
361 the fuel available, possibly including that in the near surface soil. Ivany and Salawitch
362 (1993) argued independently from oceanic carbon isotope ratios that at least 25% of the
363 above ground biomass must have burned at the K-Pg boundary.

364
365 Wolbach et al. (1990b) distinguish several forms of elemental carbon. Aciniform carbon is
366 composed of grape-like clusters of 0.01 to 0.1 μ m spherules. On average, this type of soot
367 is 26.6% of the elemental carbon, yielding a global mass abundance of 1.5×10^4 Tg of
368 aciniform carbon. Charcoal is estimated at 3.3 to 4.1×10^4 Tg, and unreactive kerogen at 0
369 to 0.8×10^4 Tg. Wolbach et al. (2003) discuss a data set from the mid-Pacific that suggests
370 aciniform soot is 9×10^3 Tg, and charcoal is also 9×10^3 Tg. Wolbach et al. directly measure
371 the carbon content of their samples. The aciniform soot to charcoal ratio is determined by
372 using an electron microscope to distinguish small and large particles.

373
374 There are several uncertainties in determining the amount of soot to use in a model. An
375 upper limit of the amount injected into the stratosphere is 7.1×10^4 Tg based on the upper
376 error bar of the Wolbach et al. (1990b) elemental carbon values. An important assumption
377 in this upper limit is that the larger particles found by Wolbach et al. (1990b), are either
378 aggregates of smaller ones, or of the same general size as the aggregates of the smaller ones
379 that occur after coagulation. A lower limit of 1.1×10^4 Tg is obtained using the lower error
380 bar of the elemental carbon from Wolbach et al. (1990b), and assuming 26.6% is aciniform
381 soot. Alternatively, one could argue that this lower limit of aciniform soot should be
382 injected into the stratosphere, along with 3.3×10^4 Tg of charcoal using different size
383 distributions. The most likely value of the aciniform soot in the stratosphere is 1.5×10^4 Tg,
384 and of elemental carbon 5.6×10^4 Tg. We use these most likely values in Table 1.

385
386 Kaiho et al. (2016) argue that the soot came from burning hydrocarbons in the crater and
387 that the total mass emitted was either 5×10^2 , 15×10^2 or 26×10^2 Tg. If we reduce these
388 values by the author's factor of 2.6 to represent the stratospheric emissions, they are 0.4%,
389 1.0% and 1.7% of the globally distributed elemental carbon reported by Wolbach et al.
390 (1990b).

391
392 Kaiho et al. (2016) measured several polycyclic aromatic hydrocarbons (PAHs) that are
393 minor components of soot from one distal site in Caravaca, Spain, and another site at
394 Beloc, Haiti that is about 700 km from the crater. Since the PAHs measured are minor
395 constituents of soot Kaiho et al. (2016) need to use a large correction factor to determine
396 the amount of soot. They first multiply by factors of 2, 5.9, or 10 to account for possible
397 loss of PAH concentrations over time. They presented no data to justify these factors. They

398 then multiply by 3.3×10^3 citing this as the ratio of their measured PAHs to soot in diesel
399 soot. No error bars were presented for this factor, and no values were given for the ratio in
400 biomass soot. The origin of this correction factor is not evident in the cited reference. They
401 then multiplied by another factor of 2.6 to represent the fraction of their soot estimate that
402 they suspect reached the stratosphere. Their overall correction factors were therefore
403 17×10^3 , 50×10^3 , and 86×10^3 . Given these large correction factors, and the lack of
404 information about their uncertainty, it is difficult to compare them with the direct
405 determinations done by Wolbach et al. (1990), which do not require any correction factors.
406

407 As noted in Table 1, the mass of soot found by Wolbach et al. (1988) would produce an
408 optical depth near 100 if the particles coagulated to spheres with a radius of $1 \mu\text{m}$ while
409 they were in the atmosphere. Toon et al. (1997) pointed out that soot clouds with such a
410 large optical depth would reduce light levels at the Earth's surface effectively to zero. The
411 optical and chemical evolution of the particles once in the atmosphere may be influenced
412 by the presence of liquid organics on the soot particles. Bare soot particles coagulate into
413 chains and sheets, while particles that are coated by liquids may form balls. Chains,
414 sheets, and coated balls have very different optical properties than do spheres (Wolf and
415 Toon, 2010; Ackerman and Toon, 1981; Bond and Bergstrom, 2006; Mikhailov et al.,
416 2006). Particulate organic matter can be absorbing, and soot coated with organics can
417 have enhanced absorption relative to soot that is uncoated (Lack et al., 2012; Mikhailov
418 et al., 2006). These fractal shapes, and organic coatings might not be preserved in samples
419 in the distal layer since all the particles have been consolidated in a layer, and even in the
420 current atmosphere the organics have short lifetimes due to rapid oxidation.
421

422 Wolbach et al. (1985) fit the size of the particles they observed, after exposing them to
423 ultrasound to break up agglomerates, to a lognormal size distribution, described by
424

$$425 \quad \frac{dN}{d \ln r} = \frac{N_t}{\ln \sigma \sqrt{2\pi}} \exp\left[-\left(\ln^2\left(\frac{r}{r_m}\right) / 2 \ln^2 \sigma\right)\right]. \quad (1)$$

426
427 Here r is the particle radius, N_t is the total number of particles per unit volume of air, r_m is
428 the mode radius, and σ is the width of the distribution. Wolbach et al. (1985) find $r_m =$
429 $0.11 \mu\text{m}$, and $\sigma = 1.6$ for the soot in the K-Pg boundary layer. We assume this
430 distribution represents the initial sizes of the soot particles. The final size, which would
431 be determined by coagulation while in the atmosphere, might not be preserved in the
432 sediments, and loosely bound clumps of particles would have been destroyed by the
433 ultrasound treatment of the samples.
434

435 The size distribution of soot from the K-Pg boundary is similar to that of smoke nearby
436 present day biomass fires as indicated in Fig. 2 (e.g., Matichuk et al., 2008). This
437 similarity in sizes is somewhat surprising because the present day smoke size distribution
438 includes organic carbon, which is present in addition to the elemental carbon (soot).
439 Generally, in wildfire smoke organic carbon has 5-10 times the mass of soot, so one
440 might anticipate that the K-Pg soot would be about half the size of the present day smoke
441 rather than of similar size since the organic coatings are no longer present, or were never
442 present, on the K-Pg soot. The organics might never have been present, because mass

443 fires are very intense and tend to consume all the available fuel, which might include the
444 organic coatings. Aggregation in the hot fires may have caused this slightly larger than
445 expected size in the K-Pg sediments. Wolbach et al. (1985) suspended their samples in
446 water and subjected them to ultrasound for 15 minutes in a failed attempt to completely
447 break up agglomerates. This failure indicated that the remaining agglomerates might
448 have been flame-welded. Therefore, the K-Pg size distribution from Wolbach et al.
449 (1985) does not represent the monomers in the aggregate soot fractal structures. Rather
450 the K-Pg size distributions represent a combination of monomers and aggregates that may
451 have formed at high temperatures. Possibly the smallest sized particles measured by
452 Wolbach et al. (1985), which have radii of 30-60 nm, represent the soot monomers.
453 These are in the same general range as monomer sizes observed in soot from
454 conventional fires (Bond and Bergstrom, 2006).

455
456 The injection altitude of the soot depends on its source. In a series of papers Belcher et
457 al. (2003; 2004; 2005; 2009) and Belcher (2009) argue from multiple points of view that
458 there were no global forest fires. Harvey et al. (2008) and Kaiho et al. (2016) argue that
459 the soot originated from oil, coal and other organic deposits at the location of the impact.
460 If correct, the soot might have been injected at high altitude along with the large
461 spherules. Recently, Robertson et al. (2013a) reconsidered each of the arguments
462 presented by Belcher et al. and came to the conclusion that global wildfires did indeed
463 occur. Pierazzo and Artemieva (2012), Premovic' (2012), Morgan et al. (2013), as well
464 as Robertson et al. (2013a) have independently argued that oil and other biomass in the
465 crater is quantitatively insufficient to be the source of the soot. Therefore, we assume
466 that the soot indeed originated from burning biomass distributed over the globe. The soot
467 is clearly present in the distal layer material, and therefore was once in the atmosphere
468 where it could cause significant changes to the climate.

469
470 Toon et al. (2007) have outlined the altitudes where one expects large mass fires to inject
471 their smoke. Numerical simulations have shown that mass fires larger than about 5 km in
472 diameter have smoke cloud tops well into the stratosphere. The smoke itself is
473 distributed over a range of heights, however. The details of the injection profiles depend
474 on the rate of fuel burning, the size of the fires, and the meteorological conditions among
475 other factors. In addition, some smoke is quickly removed from the atmosphere by
476 precipitation in pyro-cumulus. However, it is thought that over-seeding of the clouds by
477 smoke prevents precipitation, and that only 20% or so of the smoke injected into the
478 upper troposphere is promptly rained out (Toon et al., 2007). Smoke that is injected near
479 the ground, on the other hand, will be removed by rainfall within days of weeks.

480
481 The K-Pg impact occurred at a time when average biomass density likely was higher than
482 now. Following Small and Heikes (1988; their Figure 3f) and Pittock et al. (1989) one
483 would expect smoke from large area fires burning in high biomass density areas to show
484 a bi-modal smoke injection profile. The smoke at higher levels is injected in the pyro-
485 cumulus and other regions with strong vertical motions. However, once the fires die-
486 down smoke will be emitted in the boundary layer. There are also downdrafts, as well as
487 entrainment and mixing with the environment, that occur in all cumulus and these will
488 carry some smoke into the boundary layer. We simulate this with injections whose

489 vertical distributions are Gaussian functions centered at the tropopause and at the surface,
 490 as illustrated in Fig. 1. The injection at the tropopause (Eq. 2) has a half width of 3 km,
 491 but nothing is injected above about 25 km. We set this upper altitude limit based on the
 492 heights of the stratospheric sulfate clouds from explosive volcanic eruptions, which rise
 493 buoyantly as do smoke plumes. The Gaussian distribution at the ground (Eq. 3) has a
 494 half width of 1 km, assuming that the local boundary layer is relatively shallow. We
 495 assume 50% of the soot is contained in each of these distributions (Eq. 2, and Eq.3) for
 496 the general case, and for the 1 km impact. For the K-Pg, we assume the soot observed in
 497 the distal layer by Wolbach et al (1988, 1990b) was all in the portion of the Gaussian
 498 distribution at the tropopause (Eq. 2).

499 Therefore, the injection profiles are given by:

500
$$I(g s^{-1}km^{-1}) = \frac{I_{T1}}{\eta\sqrt{2\pi}} \left[e^{\left(-0.5\left(\frac{z-z_{trop}}{\eta}\right)^2\right)} \right] \quad (2)$$

501

502
$$I(g s^{-1}km^{-1}) = \frac{I_{T2}}{\mu\sqrt{2\pi}} \left[e^{\left(-0.5\left(\frac{z}{\mu}\right)^2\right)} \right] \quad (3)$$

503

504 Here I is the mass emission rate per km of altitude, I_{T1} and I_{T2} are the total mass emitted
 505 per second into the upper (Eq. 2) or lower (Eq. 3) altitude range after correcting for the
 506 emission altitude range (0-25 km) and grid spacing, μ is 1 km, η is 3 km, and z_{trop} is the
 507 altitude of the tropopause.

508

509 Geographically, we assume for the K-Pg event that all the surface biomass is set on fire.
 510 For the 1 km diameter impact, however, only the region near the impact site would burn
 511 as discussed further below.

512

513 There is also an issue of how long it takes to inject the smoke. Forest fires often burn for
 514 days, advancing along a fire front as winds blow embers far beyond the flames and onto
 515 unburned terrain. Mass fires may not spread because powerful converging winds restrict
 516 the spread. However, little is known observationally about mass fires, and fires can
 517 spread by intense infrared radiation lighting adjacent material. If mass fires are restricted
 518 then they will burn only as long as they have fuel. The present above ground global
 519 biomass in tropical forests is in the range of 0.6-1.2 g C cm⁻² (Houghton, 2005). The
 520 energy content of biomass is on the order of 3x10⁴ J/g C or, given the biomass
 521 concentration just mentioned, about 3x10⁸ J m⁻². Penner et al. (1986) and Small and
 522 Heikes (1988) found that large area mass fires with energy release rates of 0.1 MW m⁻²
 523 would have plumes reaching the lower stratosphere. Hence, it would be necessary to
 524 assume that the fuel burned in an hour or so to achieve these energy releases. Of course,
 525 it might take some time for fires in different places to start fully burning, so considering
 526 the entire region of the mass fire, as opposed to a small individual part of the fires, might
 527 prolong the energy release considerably. For example, it took several hours for the mass
 528 fire in Hiroshima to develop after the explosion of the atom bomb (Toon et al., 2007)

529

530 It should be noted that in simulations of stratospheric injections of soot from nuclear
531 conflicts, soot is self-lofted by sunlight heating the smoke (Robock et al., 2007b).
532 However, in the case of the K-Pg impact, if there are other types of particles injected
533 above the soot, which then block sunlight, the soot may not be self-lofted, which will
534 limit its lifetime. The initial soot distribution that is estimated here does not include the
535 effects of self-lofting, which would continue after the initial injection and should be part
536 of the climate simulation.

537
538 The final property to specify for soot is the optical constants. This issue is complicated
539 by the possible presence of organic material on the soot (Lack et al., 2012). However, it
540 is known that many of these organics are quickly oxidized by ozone, which is plentiful in
541 the ambient stratosphere. The stratosphere after the impact however, may have become
542 depleted in ozone very quickly, so that the organic coatings might have survived. It is
543 also possible that intense fires, such as mass fires, will consume the organic coatings,
544 which may explain why the production of soot in the fires seems to have been so much
545 more efficient than for normal fires. It may therefore be sufficient to treat the soot as
546 fractal agglomerates of elemental carbon (Bond and Bergstrom, 2006). It is known that
547 the optical properties of the agglomerates will not obey Mie theory. However, one may
548 treat their optical properties as well as their microphysical properties using the fractal
549 optics approach described by Wolf and Toon (2010). The optical constants for elemental
550 carbon may then be used for the monomers. Alternatively, one may add the organic mass
551 to the particles, and treat them using core-shell theory (Toon and Ackerman, 1981;
552 Mikhailov et al., 2006).

553
554 Bond and Bergstrom (2006) have thoroughly reviewed the literature on the optical
555 properties of elemental carbon. They conclude that the optical constants are most likely
556 independent of wavelength across the visible, with a value that depends on the bulk
557 density of the particles. Following their range of values for refractive index versus
558 particle density we suggest using a wavelength independent real index of refraction
559 $n=1.80$ and an imaginary index $k=0.67$. We also use these values in the infrared as
560 shown in Figure 3. For the monomers in Tables 1 and 3, we adopt the density suggested
561 by Bond and Bergstrom (2006) for light absorbing material, 1.8 g cm^{-3} .

562

563 **2.2.2 Soot from a 1 km impact**

564 Extrapolations of the soot injection parameters to smaller impactors than the one defining
565 the K-Pg boundary should only involve changes to the mass of soot injected, since the
566 basic properties of the soot at the K-Pg boundary are similar to those of forest fire soot.
567 Therefore, the particle sizes, injection heights, and optical constants recommended in
568 Table 3 for the smaller impact are the same as listed in Table 1 for the Chicxulub impact.
569 The mass of soot injected is estimated from the extrapolations in Toon et al. (1997). For
570 an impactor as small as 1 km diameter, debris from the impact site would not provide
571 sufficient energy to ignite the global biota since the energy of the 1 km impactor is about
572 1000 times less than that of the Chicxulub impactor. Instead, radiation from the ablation
573 of the incoming object and from the rising fireball at the impact site would ignite material
574 that is within visible range of the entering object and the fireball. This ignition

575 mechanism is well understood from nuclear weapons tests (Turco et al., 1990). Hence,
576 for a 1 km diameter impactor the fuel load at the site of the impact becomes critical to
577 evaluate the soot release. No soot would be produced from an impact in the ocean, an ice
578 sheet, or a desert. In Table 3 to compute the smoke emitted (28 Tg), we use equation 12
579 from Toon et al. (1997) to obtain an area of $4.1 \times 10^4 \text{ km}^2$ for the expected area exposed to
580 high thermal radiation density from the fireball for a 1 km diameter impactor with an
581 assumed energy of $6.8 \times 10^4 \text{ Mt}$. We then multiply that area by 3% (the fraction of C in the
582 burned fuel that is converted to smoke) and by 2.25 g C cm^{-2} , (the assumed carbon
583 content per unit area of the dry biomass that burns). The user of Table 3 can choose
584 alternate values of the injected soot by scaling linearly to the biomass concentration they
585 chose.

586

587 Ivany and Salawitch (1993) suggest that the land average, above ground biomass was about
588 $1 \times 10^{18} \text{ g}$ (about 0.7 g C cm^{-2}) at the end of the Cretaceous. The current land average, above
589 ground biomass is about 0.3 to 0.44 g C cm^{-2} (Ciais et al., 2013). An additional 1 to 1.6 g C
590 cm^{-2} is currently present in the soil, while Ivany and Salawitch suggest 1 g C cm^{-2} in the soil
591 in the Cretaceous. Some of the soil biomass may burn in a mass fire. Tropical and boreal
592 forests currently have average biomass concentrations (above ground and in soil) of about
593 2.4 g C cm^{-2} , while temperate forests have about 1.6 g C cm^{-2} including soil carbon (Pan et
594 al., 2011). Soil carbon is 30% of carbon in tropical forests and 60% in boreal forests.
595 Together tropical and boreal forests cover 6% of the Earth's surface, and temperate forests
596 1.5%. These forests cover 26% of Earth's land area. In Table 3 we assume that the
597 biomass that burns is typical of a tropical or boreal forest assuming the soil carbon burns.
598 The reader can make other choices for the biomass by scaling from the fuel load that the
599 reader prefers.

600

601 Another modeling issue of concern is the ability of models to follow the initial evolution
602 of the plume. If we assume that half of the 28 Mt of smoke from the 1 km impact is
603 injected over an area of $4 \times 10^4 \text{ km}^2$, and over a depth of 6 km near the tropopause (Eq. 2)
604 as $0.1 \mu\text{m}$ radius smoke particles, the smoke will have an initial optical depth near 4000,
605 and the number density of particles will be about 10^7 cm^{-3} . (The other half of the smoke
606 mass injected near the ground (Eq. 3) will likely be removed quickly and have little
607 impact on climate). Intense solar heating at the top of the smoke cloud near the
608 tropopause will loft it, while coagulation will reduce the number of particles by a factor
609 of 2 and increase their size proportionately in only one minute. Hence, one needs to
610 model this evolution on sub-minute time scales to accurately follow the initial evolution.
611 Alternatively, but less accurately, one might spread out the injection in time and space, so
612 that the climate model can track the evolving smoke cloud using typical model time
613 steps.

614

615 **2.3 Nano-particles from vaporized impactors**

616 **2.3.1 Nano-particles from the vaporized material following the Chicxulub impact**

617 Johnston and Melosh (2012b) find at the end of their simulations of the rising fireball that
618 about 44% of the rock vapor that was created from the K-Pg asteroid impact remained as
619 vapor rather than condensing to form large spherules. This vapor is about an equal

620 mixture of impactor and asteroid, so the 44% mass fraction is approximately equal to the
621 mass of the impactor. This 44% vapor fraction depends on the pressures reached in the
622 impact, the equation of state of the materials, as well as the detailed evolution of the
623 debris in the fireball. The fate of this vapor phase material is not well understood, and has
624 been little studied. It may simply have condensed on the spherules, or it may have
625 remained as vapor.

626 Presently, 100 μm and larger sized micro-meteoroids ablate to vapor in the upper
627 atmosphere. Hunten et al. (1980), following earlier suggestions, modeled the
628 condensation of these rock vapors as they form nm-sized particles in the mesosphere and
629 stratosphere. Bardeen et al. (2008) produced modern models of their distribution based
630 on injection calculations from Kalashnikova et al. (2000). Hervig et al. (2009) and Neely
631 et al. (2011) showed that these tiny particles are observed as they deposit about 40 tons of
632 very fine-grained material on Earth's surface per day. It is possible that a similar process
633 occurred after the Chicxulub impact. However, in the Chicxulub case the vaporization
634 occurred during the initial asteroid impact at Chicxulub rather than on reentry of the
635 material after the fireball rose thousands of km into space and dispersed over the globe.

636 The presence of 15-25 nm diameter, iron-rich material has been recognized in the fireball
637 layer at a variety of sites by Wdowiak et al. (2001), Verma et al. (2002), Bhandari et al.
638 (2002), Ferrow et al. (2011) and Vajda et al. (2015) among others. The nano-phase iron
639 correlates with iridium, is found worldwide, and therefore is likely a product of the
640 impact process. Unfortunately, these authors have not quantified the amount of this
641 material that is present. Berndt et al. (2011) were able to perform very high-resolution
642 chemical analyses, and also report a component of the platinum group elements that
643 arrived later than the bulk of the ejecta, and was probably the result of submicron sized
644 particles. However, they were not able to size the particles, nor quantify their abundance.

645 In Table 1 we take the upper limit of the injected mass of nano-particles to be 2×10^{18} g.
646 The lower limit is zero. This choice for the upper limit is consistent with the vapor mass
647 left at the end of the simulations by Johnston and Melosh (2012b). We assume an initial
648 diameter of 20 nm, following Wdowiak et al. (2001). We assume the particles are
649 initially injected over the same altitude range as the Type 2 spherules, because we
650 speculate that the small particles would not separate from the bulk of the ejecta in the
651 fireball until the ejecta entered the atmosphere and reached terminal velocity. The mass
652 injected would lead to an optical depth of particles larger than 1000 even if they
653 coagulated into the 1 μm size range. Goldin and Melosh (2009) point out that such an
654 optically thick layer of small particles left behind by the falling large spheres might also
655 be important for determining whether the infrared radiation from the atmosphere heated
656 by the Type 2 spherules is sufficient to start large-scale fires.

657 The optical properties of the nano-particles are not known. We suggest using the optical
658 properties of the small, vaporized particles currently entering the atmosphere from Hervig
659 et al. (2009). These optical constants are plotted in Figure 3. We also assume that the
660 particles have the density of CM2 asteroids, since Cr isotope ratios suggest that is the
661 composition of the K-Pg impactor (Trinquier et al., 2006). This density is 2.7 g cm^{-3} . A
662 significant fraction of the vaporized material may be from the impact site, so using an
663 asteroidal composition to determine the density is an approximation.

664

665 **2.3.2 Nano-particles from the vaporized material from a 1 km impact**

666 Johnson and Melosh (2012b) did not comment on the amount of vapor that would be
667 expected to not condense as spherules from a 1 km diameter impact. From the theory of
668 impacts, it is expected that an amount of impactor plus target that is about twice the mass
669 of the impactor would be converted into vapor from a 1 km diameter impact, just as it is
670 for a 10 km diameter impact. In Table 3 we assume that as an upper limit 35% of the
671 impactor mass plus an equivalent amount of target material, would be left as vapor after
672 spherules form. We chose this mass fraction, which is lower than that for the K-Pg object,
673 because the 1 km impact will have a smaller fireball, and be more confined by the
674 atmosphere. We also assume the injected particles will have a diameter of 20 nm. From
675 simple energy balance along a ballistic trajectory we would expect that the vaporized
676 ejecta in the fireball from a 1 km impact would rise about a thousand km above the
677 Earth's surface. This altitude is consistent with limited numerical calculations for large
678 energy releases, which indicate that the vertical velocity of the fireball is not significantly
679 reduced in passing through the atmosphere (Jones and Kodis, 1982). As the material
680 reenters the atmosphere, the particles will come to rest when they encounter an
681 atmospheric mass comparable to their own mass. Hence it is likely that the altitude
682 distribution of the nano-particles from the 1km impact will be the same as we have
683 assumed for the K-Pg impactor in Table 1, which is also similar to, but slightly lower in
684 altitude than the vertical distribution of micrometeorites on present day Earth as
685 discussed by Bardeen et al. (2008). It is difficult to determine precisely the area that will
686 be covered by this material as it reenters the atmosphere. If we assume that it takes about
687 30 min for the debris to reach peak altitude and return to the Earth, and that the plume is
688 spreading horizontally at about 4 km/s then the debris would enter the atmosphere over
689 an area of about half that of the Earth. These estimates of area covered are consistent
690 with the observations of the SL-9 impact collisions with Jupiter, and the plume from the
691 much less energetic impact at Tunguska, though these are not perfect analogs (Boslough
692 and Crawford, 1997). The optical depth of the nano-particles from the 1 km diameter
693 impact averaged over the Earth is estimated for comparison with the estimates of other
694 types of particles to be relatively large, 1.5, as noted in Table 3.

695

696 **2.4 Submicron clastics**

697 **2.4.1 Submicron clastics from the Chicxulub impact**

698 Another clear component of the K-Pg debris layer is pulverized target material. This
699 clastic material was first recognized from shocked quartz grains (Bohor, 1990), but there
700 are also shocked carbonate particles from the Yucatan Peninsula in the K-Pg boundary
701 layer material (Yancy and Guillemette, 2008; Schulte et al., 2008). Because of chemical
702 alteration of much of this material in the past 65 million years it is difficult to determine
703 the mass and size distribution directly except for the shocked quartz, which is readily
704 identified. The shocked quartz grains generally are large and would not have remained
705 long in the atmosphere. However, the shocked quartz is probably not directly related to
706 the bulk of the clastics. For instance, within 4000 km of Chicxulub the shocked quartz is
707 primarily in the few mm thick fireball layer, which is distinct from the several cm or

708 thicker ejecta layer that is dominated by clastics. The shocked quartz likely came from
709 basement rock, reached higher shock pressures than the bulk of the pulverized ejecta and
710 therefore was distributed globally in the impact fireball along with the melted and
711 vaporized material from the target and impactor. The other pulverized material, in
712 contrast, came mainly from the upper portions of the target along with basement rocks
713 toward the exterior of the crater, and the fragments were distributed locally (within about
714 4000 km of Chicxulub) in the impact ejecta debris.

715
716 The submicron fraction of the clastics is of interest because particles of such size might
717 remain in the atmosphere for months or years and perturb the climate, unlike larger
718 particles that would be removed quickly by sedimentation. For instance, Pueschel et al.
719 (1994) found 3-8 months after the 1991 eruption of Mt. Pinatubo in the Philippines that
720 volcanic dust particles with a mean diameter near $1.5 \mu\text{m}$ were optically important in the
721 lower stratosphere in the Arctic.

722 The optical constants for the injected clastics are suggested from their composition. For
723 the Chicxulub impact the clastic material is largely carbonate evaporates. We suggest
724 using the optical constants of limestone from Orofino et al. (1998). Unfortunately, the
725 values need to be generated from a table of oscillator strengths. They also need to be
726 interpolated into the visible wavelength range. We suggest extending the oscillator
727 predictions into the visible range as done by Querry et al. (1978). The density of
728 limestone is in the range of $2.1\text{-}2.6 \text{ g cm}^{-3}$, while dolomite and anhydrite have densities
729 near 2.9 g cm^{-3} . Granite has a density near $2.6\text{-}2.8$. While each of these materials
730 contribute to the clastic debris, for convenience we assume the pulverized ejecta have a
731 density of 2.7 g cm^{-3} .

732 Pope (2002) and Toon et al. (1997) used two different methods to determine the amount
733 of the submicron-clastic material from the Chicxulub impact. Unfortunately, these
734 estimates disagree by about 4 orders of magnitude, as indicated in Table 5, third row,
735 columns 1 and 2. Toon et al. (1997) used arguments based mainly on impact models, to
736 estimate that more than 10% of the mass of the distal layer ($> 7 \times 10^{17} \text{g}$) is submicron
737 diameter clastics, which would be significant to climate. Pope (2002) estimated that the
738 clastics in the distal layer have a mass that is $< 10^{14} \text{g}$. Pope (2002) used data on shocked
739 quartz to constrain the amount of clastics, which in principle is a better approach than
740 using estimates based on a model as in Toon et al. (1997). The amount of clastics of all
741 sizes in the Pope (2002) model (10^{16}g) is only 12-30 times larger than the clastics of all
742 sizes emitted in the relatively small 1980 Mt. St. Helens eruption. Therefore, based on
743 Pope's (2002) analysis, the submicron fraction would not be of significance to climate.
744 Below we attempt to reconcile these two approaches to better determine the amount of
745 submicron clastics.

746 **2.4.1.1 Potential errors in the Toon et al. (1997) estimate of submicron clastics**

747 Toon et al. (1997) estimated the amount of submicron clastics starting from analytical
748 models of the mass of material injected into the atmosphere by a 45-degree impact. They
749 estimated the mass of melt + vapor per megaton of impact energy ($\sim 0.2 \text{ Tg/Mt}$) and the
750 mass of pulverized material per megaton of impact energy (about 4.5 Tg/Mt). Assuming
751 a $1.5 \times 10^8 \text{ Mt}$ impact, these formulae suggest a melt + vapor amount of $3 \times 10^{19} \text{g}$ ($\sim 1 \times 10^4$

752 km³, assuming a density of 2.7 g cm⁻³) and a pulverized amount of 7x10²⁰g (~2.5x10⁵
753 km³). While sophisticated impact calculations generally agree with the amount of melt +
754 vapor, not all of it is found to reach high enough velocity to be ejected from the crater.
755 For example, Artemieva and Morgan (2009) investigated a number of impact scenarios
756 that created transient craters with diameters of 90-100 km, which they thought to be
757 consistent with the transient diameter of the Chicxulub crater. Considering those cases
758 with oblique impacts from 30-45 degrees with energies of 1.5-2 x10⁸ Mt, they found that
759 the melt was in the range 2.6x10⁴ to 3.8x10⁴ km³. However, the amount that reached high
760 enough speed to be ejected from the crater was in the range 5x10³ to 6x10³ km³ (average
761 5.6x10³ km³, 1.4x10¹⁹g, about 2-10 impactor masses). On average, only about twenty
762 percent of the melt and vapor amount escapes from the crater. Therefore, Toon et al.
763 (1997) may have overestimated the amount of melt escaping from the crater by about a
764 factor of 2. It should be noted that in Artemieva and Morgan (2009) the melt exceeds the
765 mass of the distal layer, which is about 4x10¹⁸g, by about a factor of 5, because much of
766 the melt is deposited as part of the ejecta curtain and never reaches the distal region.

767 Artemieva and Morgan (2009) find that the total mass ejected from the crater is 1.3x10⁴
768 km³ (2.9x10¹⁹ g). Assuming that 90% of this material is pulverized rock their results
769 imply that Toon et al. (2007) overestimated the amount of clastic debris ejected from the
770 crater by a factor of about 25. In column 3 of Table 5 we correct the amount of
771 pulverized material to agree with the Artemieva and Morgan (2009) value of 2.9x10¹⁹ g
772 of clastics escaping the crater. It is interesting to note that the clastic mass from
773 Chicxulub is only a factor of about 10 larger than the minimal estimated mass of clastics
774 ejected in the Toba volcanic eruption about 70,000 years ago (Matthews et al., 2012).

775 Another issue is the fraction of the pulverized debris that is submicron. Toon et al. (1997)
776 computed the amount of pulverized debris whose diameter is smaller than 1 μm from size
777 distributions measured in nuclear debris clouds originating from nuclear tests that were
778 many orders of magnitude lower in energy than the K-Pg impact, and from impact crater
779 studies cited by O'Keefe and Ahrens (1982) based on grain size measurements from
780 craters. Toon et al. (1997) assume that 0.1% of the total clastic material would be
781 submicron. Pope (2002) cited studies of volcanic clouds to conclude that 1% by mass of
782 the pulverized material would be submicron.

783 Rose and Durant (2009) examined the Total Grain Size Distribution (TGSD) from a
784 number of volcanic eruptions and concluded that the amount of fine ash is related to
785 increasing explosivity of the event. The TGSD is supposed to represent the size
786 distribution as the clastics left the crater. Mt. St. Helens is the most likely of the volcanic
787 eruptions they considered to be relevant to the extreme energy release in a large impact.
788 About 2% of the total ejecta from Mt. St. Helens had a diameter smaller than 1 μm. Since
789 the erupted mass was about 3-8x10¹⁴ g, the submicron mass emitted by Mt. St. Helens
790 was about 6-16x10¹²g. Matthews et al. (2012) considered the Toba eruption, whose
791 clastics are within an order of magnitude of those from Chicxulub. Their data shows that
792 1-2% of the mass of the clastics is in particles smaller than 1 μm and 2-6% in clastics
793 smaller than 2.5 μm.

794 In Table 5 we use 2% of the pulverized material as a revised estimate for the fraction of
795 the clastic material that is released as submicron ejecta. This fraction is a factor of 20
796 larger than the one used in Toon et al. (1997). Hence our revised submicron mass

797 estimate for the Chicxulub impact (column 3 row 3) is very similar to the one Toon et al.
798 (2007) estimated (column 2 row 3) because, although we lowered the estimate of the
799 clastic mass exiting the crater to agree with Artemieva and Morgan (2009), we increased
800 the estimate of the fraction that is submicron.

801 A confounding issue is the amount of submicron and other clastics that escapes from the
802 near crater region and is distributed globally. A large fraction of the pulverized debris in
803 the ejecta curtain was removed within 4000 km of the impact crater (Bohor and Glass,
804 1995), and volcanic ejecta is likewise largely removed near the volcanic caldera. For
805 example, there is 4-8 cm of ash 3000 km from the Toba crater, which is not too different
806 from the thickness of the Chicxulub deposits at a similar distance from the crater. If the
807 removal occurred only by individual particle sedimentation, one could simply take the
808 mass in the smaller ranges of the size distribution and assume it spread to the rest of the
809 globe. However, it is clear from volcanic eruption data that a significant fraction of the
810 submicron debris is removed near the volcano by processes other than direct
811 sedimentation (Durant et al, 2009; Rose and Durant, 2009). These processes include
812 rainout of material from water that condenses in the volcanic plume, and also
813 agglomeration possibly enhanced by electrical charges on the particles. It is likewise
814 clear that such localized removal occurred after the K-Pg impact. Yancy and Guillemette
815 (2008) describe accretionary particles that make up a large fraction of the debris layer as
816 far as 2500 km from the Chicxulub crater. These agglomerated particles, which range in
817 size from tens to hundreds of μm , are composed mainly of particles with a radius of 1-4
818 μm . While largely composed of carbonate, the particles are enriched in sulfur.

819 One can use the size distributions from volcanic data, along with the total clastic mass
820 ejected from Chicxulub to compute the particle agglomeration, and thereby follow the
821 particles as they spread across the Earth. Such work is now being done for volcanic
822 events, for example by Folch et al. (2010). They find that they can successfully reproduce
823 mass deposited on the surface from the Mt. St. Helens eruption by including
824 agglomeration. However, such calculations for Chicxulub are difficult for several
825 reasons: the large clastic masses involved exceed the mass of the atmosphere for a
826 considerable distance from the crater, so the debris flows cannot be reproduced in
827 standard climate models; the complexity of the distribution of material in the plume with
828 some material reaching escape velocity and other parts being hurled over a substantial
829 fraction of the planet make it difficult to determine the spatial distribution of the material,
830 and some material is likely lofted well above the tops of most climate models; and the
831 presence of clastics, melt and rock vapor together with sulfur and water produces a
832 chemically complex plume.

833 Eventually it will be necessary to use detailed non-hydrostatic, multiphase plume models
834 including agglomeration to better understand the distribution of Chicxulub ejecta. In the
835 meantime for climate modeling we suggest placing the clastic mass in Table 5 (2.9×10^{19}
836 g) in a circular area with radius of 4000 km, which is 22.4% of the area of Earth. This
837 will result in a column density of 25 g cm^{-2} , or a layer thickness of about 10 cm. The
838 mass density of the atmosphere is about 1000 g cm^{-2} , so this is about a 2.5% perturbation
839 to the mass of the atmosphere. In reality the mass is concentrated near the crater as
840 shown by Hildebrand (1993). However, the observed mass density is relatively constant
841 between 1000 and 4000 km. The initial vertical distribution of this material may be very

842 complex due to density flows within several hundred km of the crater. We suggest
843 initializing models assuming an injection with an altitude independent mass mixing ratio
844 of about 2.5%. Given our suggested vertical distribution 90% of the material will
845 initially lie in the troposphere. Tropospheric material is unlikely to become globally
846 distributed even if it escapes agglomeration, because it will quickly be removed by
847 rainfall.

848 As an alternative to the complexity of modeling the loss of this material in the
849 troposphere and considering the entire size distribution, we suggest simply placing an
850 appropriate mass into the stratosphere. The values for a stratospheric injection are given
851 in the bottom row of Table 5 and the first row of Table 1. For illustration, we have
852 estimated the final optical depth assuming that 10% of the submicron material (the
853 amount placed into the stratosphere) will escape removal. For a size distribution we
854 suggest using the smaller size mode measured in the stratosphere after the Mt. St. Helens
855 eruption as summarized by Turco et al. (1983). This size distribution is log-normal (Eq.
856 1), with a mode radius of $0.5 \mu\text{m}$ and a standard deviation of 1.65. The estimated optical
857 depth of 88 is very large, even though the submicron clastic material in this estimate is
858 only about 1% of the mass of the distal layer.

859

860 **2.4.1.2 Potential errors in the Pope (2002) estimate of submicron clastics**

861 Pope (2002) determined the amount of clastics by modeling the amount of quartz in the
862 distal layer. He found that he needed an initial injection of about 5×10^{15} g of quartz to
863 match the distribution of quartz mass with distance from the impact site. It is not clear
864 how good this estimate is because the removal rate of material in large volcanic clouds, a
865 possible impact analog, does not occur by individual particle sedimentation, but rather by
866 settling of agglomerates (Folch et al. 2010). Hence removal in the region near the impact
867 site may have been larger than Pope estimated, requiring a larger volume of quartz; or the
868 removal of clastics may be different than that of quartz. The value in Artemieva and
869 Morgan (2009) for the pulverized material ejected from the crater is 3 orders of
870 magnitude larger than the estimate of Pope (2002). Most of this material is in the ejecta
871 curtain, not in the impact fireball, and so is deposited close the impact crater. The
872 shocked quartz is primarily associated with the impact fireball, so the bulk of the
873 pulverized material may not be seen in Pope's analysis.

874 Pope assumed that quartz composed 50% of all the clastic debris, so that all of the
875 clastics injected weighed about 10^{16} g. This number is about two orders of magnitude less
876 than the clastics from the Toba eruption (Matthews et al., 2012), and more than 3 orders
877 of magnitude less than the Artemieva and Morgan (2009) estimate for clastics from the
878 Chicxulub impact.

879 The assumption by Pope (2002) that quartz is 50% of all the clastics is likely in error.
880 There is no reason to think there is much quartz in the upper layers of sediment at the
881 Chicxulub site. In the stratigraphic columns shown by Ward et al. (1995) the pre-impact
882 sediments at Chicxulub consist of approximately 3 km of Mesozoic carbonates and
883 evaporites with ~3-4% shale and sandstone. Therefore, it is more likely the quartz
884 originates from the basement rocks. There is also not a strong connection between the
885 physical processes that distributed the quartz (the impact fireball, with high ejection

886 velocity), and those that distributed the pulverized material (the ejecta curtain with low
887 velocity).

888 It is possible that the quartz to clastics ratio is determined by the ratio of quartz to total
889 debris in the samples closest to Chicxulub, since these may have suffered the least removal
890 by sedimentation. Pope suggests these intermediate distance layers contain about 1%
891 quartz, but only considers the fireball layer, which is less than 10% of the total ejecta layer
892 within 1000 km of the crater. The remainder of the intermediate distance layer contains
893 little quartz, so the clastics could be more than 1000 times the mass of the quartz. It is not
894 clear that 1000 is an upper limit to the ratio of clastics to quartz because the quartz and
895 pulverized material move along different paths in the debris cloud. If we accept this ratio
896 of 1000 for the ratio of clastics to quartz, the mass of clastics from Pope's analysis would
897 be 5×10^{18} g, which is within a factor of 6 of the Artemieva and Morgan (2009) value. If 1%
898 of this mass is submicron then 5×10^{16} g of submicron clastics would have been injected into
899 the upper atmosphere.

900 **2.4.1.3 Reconciliation of Pope (2002) and Toon et al. (1997) estimates of submicron** 901 **clastics**

902 Table 5 shows that the new estimate of submicron mass following the procedure of Toon
903 et al. (1997) agrees with the new estimate following the procedure of Pope (2002) within
904 20%. The new estimate is about 12 times less than the Toon et al. (1997) value mainly
905 because Toon et al. (1997) did not consider that most of the pulverized mass would not
906 be ejected from the crater. The new application of the Pope (2002) approach leads to
907 estimated submicron dust emissions that are about 500 times larger than the one derived
908 by Pope (2002). The major difference is that we have assumed the ratio of quartz to
909 clastics is about 1000, rather than 1 as assumed by Pope (2002). Despite the perhaps
910 coincidental agreement of these two estimates, there is substantial uncertainty in the true
911 mass of submicron clastic particles in the K-Pg distal layer. Observations of the
912 submicron material in the distal layer are needed.

913

914 **2.4.2 Submicron pulverized rock from a 1 km diameter impactor**

915 In order to determine the properties of the pulverized ejecta from a 1 km impactor, we
916 use the pulverized mass injection per Tg of impact energy from Toon et al. (1997), but
917 reduce it by the factor of 25 discussed earlier to account for the fraction of the clastic
918 mass with enough velocity to escape the crater. This procedure yields a clastic mass of
919 1.3×10^{16} g. For reference, the volume of clastics from the eruption of Mt. Tambora in
920 1815 is estimated to have been about 150 km^3 , which is a mass of about 3×10^{17} g. Hence
921 the Tambora eruption likely surpasses the clastics from the hypothetical 1 km diameter
922 impactor by more than a factor of 10. The same size distribution for the clastics is
923 recommended for the 1 km impact and the Chicxulub impact, since it seems to hold for a
924 range of volcanic events from Mt. St. Helens to Toba, which span the 1 km diameter
925 impactor in terms of clastics. We also suggest that the mass be initially mixed uniformly
926 in the vertical above the tropopause. According to Stothers (1984) the Tambora clastics
927 were deposited in layers that are centimeters in thickness at distances 500 km from the
928 volcano. Accounting for the drift of the ash downwind, the area of significant ash fall
929 was about $4.5 \times 10^5 \text{ km}^2$. If this same area is used for the initial injection of the clastics for

930 the 1 km impact, then the column mass concentration is about 8.7 g cm^{-2} , which in turn is
931 slightly less than 1% of the atmospheric column mass. The estimated optical depth of the
932 clastics in Table 3 is about 25% of the optical depth from nano-particles originating from
933 vaporized rock. Given that these materials are much less absorbing than soot, and lower
934 in optical depth than nano-particles they can probably be neglected in estimates of the
935 climate changes due to a 1 km diameter impact on land.

936 **3. Gas injections**

937 There are a large number of gases that might be injected into the atmosphere after an
938 impact and might be important to atmospheric chemistry, climate, or both. These can
939 originate from the impactor itself, from ocean or ground water, or from the target
940 sediments. They may also originate in response to environmental perturbations, such as
941 wildfires, or atmospheric heating from the impact fireball and ejecta. Various estimates
942 have been made for each of these sources. However, clear evidence from the distal layer
943 is not available for any gases of potential interest. Some gases, such as carbon dioxide,
944 would have stayed in the gas phase rather than condensing into particulate form. Other
945 gases, such as those containing sulfur, may have reacted on the particles composing the
946 distal layer, or formed independent particles. In either case sulfur is so common in the
947 environment it is difficult to detect an injection. For these reasons all the gas phase
948 injections are uncertain. Below, we first discuss the chemical content of each of the
949 potential sources of gases, and then we discuss the likely amounts of each material
950 injected following an impact. Relevant ambient abundances are given in Tables 2 and 4
951 along with estimated injections for the Chicxulub impact and a 1 km impact. The ambient
952 masses are given to assist the reader in understanding the magnitudes of the injections.
953 Generally ambient concentrations are given in the literature in terms of the mixing ratio.
954 To compute the masses we assume the ambient mixing ratios are constant over the whole
955 atmosphere, or the stratosphere. We then convert the volume mixing ratio to the mass
956 mixing ratio using the molecular weight and then multiply by the mass of the atmosphere
957 above either the surface, or tropopause to obtain the total mass of the gas. The ambient
958 abundances assume the current stratospheric mixing ratio of Cl is 3.7 ppbv (Nassar et al.,
959 2006), Br is 21.5 pptv (Dorf et al., 2006), inorganic I is 0.1pptv (upper limit from Bosch
960 et al. 2003), CO_2 is about 395 ppmv, and methane is about 1.8 ppbv. Stratospheric S,
961 taken from the Pinatubo volcanic eruption, is about 10 Tg (Guo et al., 2004), reactive
962 nitrogen, NO_x , in the stratosphere is difficult to quantify simply. Instead we compare
963 with the ambient abundance of N_2O in the stratosphere, about $2 \times 10^{14} \text{ g N}$. N_2O is a major
964 source of NO_x .

965

966 **3.1 Impactor**

967 **3.1.1 Composition of the impactor**

968 Kring et al. (1996) summarized the S, C, and water contents of a large number of types of
969 asteroids. Trinquier et al. (2006) found from chromium isotopes that the Chicxulub
970 impactor was most likely a carbonaceous chondrite of CM2 type. Such asteroids have
971 3.1wt % S, 1.98 wt% C, 11.9 wt% water, and a density of 2.71 g cm^{-3} . Over the range of
972 chondrites, which constitute 85% of meteorite falls, S varies from 1.57 to 5.67 wt%, C
973 from 0.04 to 3.2 wt %, and water from 0.2 to 16.9 wt %.

974 report that by mass CM carbonaceous chondrites contain about 4ppm Br. Goles et al
975 (1967) report that Cl ranges from 190-840 ppm of carbonaceous chondrites, Br ranges
976 from 0.25 to 5.1 ppm, and Iodine ranges from 170 to 480 ppbm. Table 6 summarizes
977 the composition of asteroids using values for CM2 type carbonaceous chondrites from
978 Kring et al. (1996) for S, C, and water, and for the Mighei (the CM2 type example) from
979 Goles et al. (1967) for Cl, Br and I.

980 **3.1.2 Gases from the impactor**

981 Tables 2 and 4 indicate the direct contributions from 1 and 10 km impactors of a number
982 of chemicals, as discussed further below. We assume that the entire 10 km or 1 km
983 diameter impactor melted or vaporized so that all of the gases are released. For the 10
984 km impactor these gases would have been distributed globally in the hot plume along
985 with the melt spherules within hours. They would reenter with the same vertical
986 distribution as the Type 2 spherules. For the 1 km diameter impactor, the initial injection
987 may have only covered half the Earth, with global distribution over days via wind, after
988 reentry into the upper atmosphere.

989 We further assume that the vapors under consideration do not react with the hot mineral
990 grains either in the plume or in the hot layer at the reentry site. In fact, given the large
991 particle surface areas in the atmosphere over the globe it is possible that there was a
992 significant transfer of material from the gas phase to the surfaces of the mineral grains in
993 a short period of time.

994 As pointed out by Kring et al. (1996) and Toon et al. (1997) the S in a 10 km diameter
995 impactor would exceed that from the Mt. Pinatubo volcanic injection by a factor above
996 1000. Even a 1 km diameter carbonaceous chondrite could deliver several times as much
997 sulfur to the atmosphere as did the Mt. Pinatubo eruption in 1991. Stratospheric water
998 could be enhanced by a factor of more than 100 from the water in a 10 km impactor. Cl
999 could be enhanced by factors above 500, Br by almost 500, and I by more than 50,000.
1000 However, there is not enough C in a 10 km asteroid to affect the global carbon cycle
1001 significantly.

1002 Many investigators have pointed to sulfate as an important aerosol following the
1003 Chicxulub impact. Tables 1 and 3 compare the mass of sulfur from the impactor with the
1004 mass of the spherules and nano-particles. The optical depth, which controls the climate
1005 change following the impact, and the particle surface area, which likely controls
1006 chemistry, are approximately linear with the mass. In our estimates, the sulfate coming
1007 directly from the asteroid could have a large optical depth assuming it was not removed
1008 on the spherules, or large clastics.

1009

1010 **3.2 Seawater**

1011 **3.2.1 Composition and depth of seawater**

1012 The composition of seawater is given in Table 6 (Millero et al., 2008). It is thought that
1013 injections of water into the upper atmosphere will lead to droplet evaporation, with small
1014 crystals of salt left behind. If liquid water is left after a massive injection of water, the
1015 droplets will likely freeze leaving salt behind as particles embedded in ice crystals.

1016 Vaporization of water during the impact may leave behind salt crystals, or the salts may
1017 decompose into their components. As discussed by Birks et al. (2007), complex
1018 simulations are needed to determine how much material is freed from the salt particles to
1019 enter the gas phase where it might destroy ozone. In Tables 2 and 4 we list the total
1020 amounts of several interesting chemicals that might be inserted into the stratosphere.
1021 However, all of them except water vapor are likely to be in the form of a particulate until
1022 photochemical reactions liberate them.

1023 A significant uncertainty related to any oceanic contribution to atmospheric composition
1024 is the depth of the ocean in relation to the size of the impactor, and the water content of
1025 sediments at the crater site. The depth of the ocean at Chicxulub at the time of the impact
1026 is not known. Many investigators have referred to it as a shallow sea. However, Gulick
1027 et al. (2008) estimates that the water depth averaged over the impact site was 650 m,
1028 which is considerably deeper than earlier estimates. We use a water depth of 650 m in
1029 Table 2 to estimate the amounts of material injected by Chicxulub. A 1 km diameter
1030 impactor is smaller than the average depth of the world oceans, which is about 3.7 km.

1031 **3.2.2 Gases from Seawater-Chicxulub**

1032 For the Chicxulub impact, Pope (1997) assumed that the 650 m depth of seawater within
1033 the diameter of the impactor (10 km) will be vaporized, follow the path of the Type 2
1034 spherules, and reenter the atmosphere globally. In Table 2 we compute the water
1035 vaporized following the equations in Toon et al. (1997). These equations, assuming an
1036 impact velocity of 20 km s^{-1} , led to an order of magnitude greater injection of water than
1037 using Pope's estimate. The vaporized water is 0.4 times the impactor mass. During the
1038 vaporization of the seawater we assume the water will be present as water vapor, and that
1039 the materials in the water will be released as vapors. Some of these materials likely
1040 would react quickly with the hot minerals in the fireball or later with the hot minerals in
1041 the reentry layer.

1042 It is also likely that a considerable amount of water was splashed into the upper
1043 atmosphere. Ahrens and O'Keefe (1983) estimated that the water splashed above the
1044 tropopause from a 10 km diameter impact into a 5 km deep ocean would be 30 times the
1045 mass of the impactor. We assume that the amount of water splashed above the tropopause
1046 will scale linearly with the depth of the ocean. Therefore, about 4 times the impactor
1047 mass of water may have been splashed into the upper atmosphere. Much of this water
1048 may immediately condense and rainout, as discussed in Toon et al. (1997). However,
1049 some of the dissolved salts may be released if some of the water evaporates. The
1050 assumed injection of gases, and particulates that might become gases, from the ocean is
1051 summarized in Table 2 for the Chicxulub impact.

1052 **3.2.3 Gases from Seawater-1 km asteroid**

1053 No seawater is injected by the 1 km diameter asteroid impact on land. If a comet hit the
1054 land there would be a water injection.

1055 Pierazzo et al. (2010) estimated that 43 Tg of water would be injected above 15 km by a
1056 1 km asteroid impact into the deep ocean. Of this water, 25% is in the form of vapor and
1057 75% in the form of liquid water. In their modeling the water was assumed to be
1058 distributed with a uniform mixing ratio from the tropopause to the model top. It was also

1059 spread uniformly over an area 6200x6200 km in latitude and longitude. Using the
1060 equations in Toon et al. (1997) for the vaporized water produces a value which is 60% of
1061 the vaporized water from the detailed modeling used in Pierazzo et al. (2010). Given
1062 these water injections we use the composition of sea water to determine the injections of
1063 the various species. Pierazzo et al. (2010) estimate injections of Cl and Br that are more
1064 than an order of magnitude smaller than ours because they consider the amounts that have
1065 been converted into gas phase Cl and Br by photochemical reactions in the atmosphere,
1066 while we estimate the total injections, which initially are likely to be in the particulate
1067 phase.

1068

1069 **3.3 Impact Site**

1070 **3.3.1 Composition of the impact site**

1071 The sea floor at the Chicxulub impact site, like the modern Yucatan, contained abundant
1072 carbonate and sulfate rich deposits. Ward et al. (1995) conclude that 2.5-3 km of
1073 sedimentary rock were present at Chicxulub, composed of 35-40% dolomite, 25-30%
1074 limestone, 25-30% anhydrite, and 3-4% sandstone and shale. The dolomite and
1075 limestone are no doubt porous. Pope et al. (1997) estimate the carbonates in the Yucatan
1076 have a porosity of 20%. The pores would have been filled by seawater since the
1077 sediments were submerged. This ground water produces an equivalent water depth of
1078 about 400 m. The carbon content of limestone is 12% by weight, and of dolomite 15%
1079 by weight. The sulfur content of anhydrite is 23.5% by weight. To our knowledge, trace
1080 species such as Br, Cl, and I have not been reported for these sedimentary rocks, but
1081 would be present in the seawater in the pores.

1082 **3.3.2 Gases from the impact site**

1083 For the 10 km Chicxulub impact we follow Pope et al. (1997) for the abundances of S
1084 and C assuming 30% anhydrite, 30% limestone and 40% dolomite. The composition of
1085 the impact site is given in Table 6. We ignored species other than S and C that might be
1086 in the target material. It is difficult to follow the target debris since some of it is
1087 vaporized, and some melted. We follow Pope (1997) and assume that the upper 3 km of
1088 the target is vaporized within the diameter of the impactor. The gases within this volume
1089 of vaporized material are assumed to be released, and to follow the trajectories of the
1090 Type 2 spherules. Pope et al. (1997) estimated the amount of material that would be
1091 degassed from target material that was melted or crushed in a large impact. We use the
1092 values from Table 3 of Pope et al. (1997) for out of footprint vapors, in our Table 2 for
1093 the degassed impact site emissions. We also assume that the granite underlying the
1094 impact site does not contribute.

1095 The source gases from a 1 km land impact would depend on the composition of the
1096 impact site, so we do not list values in Table 4. We assume nothing would be liberated
1097 from the sea floor in a 1 km impact in the deep ocean.

1098

1099 **3.4 Fires**

1100 **3.4.1 Composition of Smoke**

1101 It is well known that forest fires emit a wide variety of vapors into the atmosphere.
1102 Andreae and Merlet (2001) provide emission ratios (g of material emitted per g of dry
1103 biomass burned) for many vapors expected to be important in the atmosphere as listed in
1104 Table 6. As discussed in section 2.2.1, the soot emission may have been enhanced
1105 relative to wildfire estimates by Andreae and Merlet (2001) after the Chicxulub impact
1106 because the impact-generated fires were mass fires. We do not consider any
1107 enhancements of the gas phase emission ratios, but they may also be impacted by fire
1108 intensity or the types of plants making up the biomass.

1109

1110 **3.4.2 Gases from Fires**

1111 In Tables 2 and 4 we computed the burned mass from Chicxulub assuming that 1.5 g cm^{-2}
1112 of dry biomass burns over the entire land surface area of the Earth, and then used the
1113 emission factors from Andreae and Merlet (2001) to obtain the gas phase emissions. For a
1114 1 km impact we assume the area burned is $4.1 \times 10^4 \text{ km}^2$ (Toon et al., 1997), and the dry
1115 biomass is 2.25 g C cm^{-2} . We then used the emission ratios from Andreae and Merlet
1116 (2001) to compute the gas phase emissions. Comparing the gas phase emissions from
1117 fires in Tables 2 and 4 with ambient values indicates that there would be large
1118 perturbations for all gases for the 10 km diameter impact. Only iodine is significantly
1119 perturbed for the 1 km impact. For the gas phase emissions we suggest using the same
1120 vertical profile as suggested for soot earlier. The emissions would only occur over the
1121 region near the impact site for the 1 km impact.

1122 **3.5 Gases generated by atmospheric heating**

1123 The energy deposited in the upper atmosphere by the initial entry of the bolide, as well as
1124 by the rising fireball, may have converted some N_2 to NO_x . Early studies suggested that
1125 a large fraction of the impact energy would be put into the lower atmosphere, which in
1126 turn led to suggestions that a large amount of nitrogen oxides would be produced from
1127 the heated air. However, it is now understood that most of the energy release from an
1128 impact to the atmosphere will occur at high altitude from reentry of spherules and other
1129 debris. Toon et al. (1997) reviewed the various ways in which NO_x might be generated
1130 following an impact, largely following Zahnle (1990). They concluded that $3 \times 10^{16} \text{ g}$ of
1131 NO might be produced from the atmosphere for a 10 km diameter impact with about half
1132 coming from the plume at the impact site, and half from the reentry of material across the
1133 Earth. We have recorded this value in Table 2. For comparison, Parkos et al. (2015)
1134 conducted detailed evaluations of the NO_x produced by the infalling spherules and
1135 concluded the spherules could produce 1.5×10^{14} moles of NO_x ($3 \times 10^{15} \text{ g}$ if the NO_x is in
1136 the form of NO) which they further concluded was not sufficient to acidify ocean surface
1137 waters. In Table 2 we use the Toon et al. (1997) injection of NO since it includes both
1138 source mechanisms. According to Zahnle et al. (1990) a 1 km impact on land might
1139 produce $0.6 \times 10^{14} \text{ g}$ of NO , largely in the hot plume at the impact site. This value is
1140 entered in Table 4. For comparison, we note that Pierazzo et al. (2010) suggested that the
1141 mass of NO produced by a 1 km ocean impact is about $0.39 \times 10^{14} \text{ g}$.

1142

1143 **3.6 Discussion of gas injections**

1144 Some of the gas phase sources just discussed are easy to apply to an impact. For
1145 example, the emissions from fires simply depend on the area burned, the fuel loading and
1146 the emission factors.

1147 Other sources of gases are more difficult to evaluate. Since we have no measurements for
1148 large impacts, the form of emission can be uncertain. For example, sulfur could be
1149 injected as SO₂ or SO₃. Another difficulty that comes in understanding the contribution
1150 of target material to gases, such as SO₂, is the pressure needed to vaporize the material.
1151 Pope (1997), for example, adopted pressures above 70 GPa to vaporize carbonate, 100
1152 GPa for complete vaporization of anhydrite, and 10 GPa for water vaporization from
1153 pores. These vaporization pressures are higher than suggested by early researchers,
1154 leading to lower amounts of target vaporized. Pierazzo et al. (2003) redid the impact
1155 calculations and also estimated the amounts of materials that might be released, which are
1156 close to those estimated by Pope et al. (1997). The altitude distribution of the ejecta
1157 varies with the source of the material. Finally the chemical form of the emission varies
1158 with thermochemistry in the ejecta plume or fireball, and interactions with hot mineral
1159 surfaces, and for some materials exposure to high temperature on reentry.

1160 Tables 2 and 4 summarize our choices for the injections of the various gases. For each
1161 type of source we also specify the altitude of the expected injection, using a reference to
1162 Tables 1 and 2 for the particle injections. We assume all of the impactor mass entered the
1163 rising fireball, so it would be injected near 60 km altitude along with the spherules. In
1164 some cases, for example for the degassed target material and for splashed seawater, we
1165 consider the material to have been uniformly mixed above the tropopause. For materials
1166 coming from fires we assume the same vertical injection as for soot.

1167 As has been pointed out many times (Kring, 1996; Toon et al., 1997; Pope et al., 1997;
1168 Pierazzo et al., 2003) the sulfur injection from a 10 km impactor might be thousands of
1169 times greater than that from the Pinatubo eruption, and also was likely larger than the
1170 injection from the massive Toba eruption by a factor between 10 and 100. Our sulfur
1171 injection from the target material is about half that of Pope's (1997) estimate of 10¹⁷ g and
1172 slightly less than Pierazzo et al's (2003) estimate for a 15 km diameter impactor of 7.6 x
1173 10¹⁶ g. Our sulfur injection from the asteroid itself is within the range suggested by Pope
1174 et al. (1997) of 2.7-5.9 x 10¹⁶ g. Interestingly, the sulfur injection we estimate for
1175 Chicxulub is about 10 times greater than the yearly emission estimated by Schmidt et al.
1176 (2015) for a large flood basalt from the Deccan traps. Of course, the flood basalt might
1177 continue for a decade or more, bringing the total sulfur emission close to that from the
1178 Chicxulub impact. Table 4 suggests that the sulfur injection from a 1 km impact would
1179 be several times greater than that from the Pinatubo eruption, but that would be only a
1180 modest injection relative to historical volcanic eruptions. In Table 1 and Table 3 we
1181 assume the injected sulfur gas is converted into sulfate. If so it would yield a large
1182 optical depth for the Chicxulub impact. However, for both the 1 km and Chicxulub
1183 impacts, the sulfur injection, if converted to sulfate, would be an order or magnitude less
1184 massive than the nano-particles. Therefore, the sulfate would be an order of magnitude
1185 less important optically than the nano-particles. While it might exceed the soot mass
1186 slightly, soot is much more important optically than sulfate, which is transparent at
1187 visible wavelengths. Therefore, the sulfate in our model is of relatively little importance
1188 optically, unless the sulfur remains in the air after the other particles are removed.

1189 Our estimated C injection (in the form of CO₂) is dominated by emissions from forest
1190 fires. We have the same emission from the impactor as Pope (1997), but we have less
1191 than half the emission from the target material as Pope (1997) or Pierazzo et al. (2003).
1192 All these studies suggest a small impact perturbation relative to the CO₂ 65 million years
1193 ago, which was several times larger than now.

1194 The water vapor injections in Tables 2 and 4 are very large compared with ambient
1195 values in the stratosphere. However, most of the water is from fires, and half will be
1196 injected into the troposphere where it will be quickly removed. The water from the
1197 impactor and target is modest, about 1 cm as a global average depth of rain. The typical
1198 rainfall averaged over the current Earth is about 3 mm day⁻¹. The emissions from the
1199 impactor and from vaporized seawater, both of which would have been injected globally
1200 at the same altitudes as the Type 2 spherules, are capable of saturating the entire ambient
1201 stratosphere. Our water injection is similar to that estimated by Pope (1997), and Pierazzo
1202 et al. (2003). While the water vapor has been largely ignored in previous work on the
1203 Chicxulub impact, it has the ability to alter the thermal balance of the stratosphere by
1204 emitting and absorbing infrared light. Water vapor may have been a factor in the
1205 radiation of thermal energy to the surface during the first few hours after the K-Pg
1206 impact, since Goldin and Melosh (2009) sought an infrared absorber to prevent radiation
1207 from escaping from the top of the atmosphere. Some of the particles in the stratosphere
1208 might be removed by precipitation, but the mass of water injected is comparable to the
1209 mass of the nano-particles and spherules. Therefore, removal by precipitation is probably
1210 not significant since if the water condenses on all the particles it will add only a small
1211 mass, and increase the fall rate only slightly, while if water condenses on only a subset of
1212 the particles it will remove only a subset. The water injection by the 1 km diameter
1213 impact on land is about 15% of the ambient water, but might still lead to some significant
1214 perturbations if it is injected into the upper stratosphere. The 1 km impact in the deep
1215 ocean could inject about 40 times the ambient water into the stratosphere (Pierazzo et al.,
1216 2010), and water should be considered in simulations of such impacts.

1217 For the 10 km diameter impactor, there are injections of Cl, Br, and I that exceed the
1218 ambient values by orders of magnitude. There are significant sources for all three
1219 halogens from fires, the impactor and seawater, so it seems inescapable that large
1220 injections would have occurred. The injections of NO_x from fires, and from heating the
1221 atmosphere are also very large compared with ambient values. For instance, Table 2
1222 shows the NO_x injections are one to two orders of magnitude larger than the stratospheric
1223 burden of N₂O, the principle source of NO_x. For the 1 km diameter land impact only the
1224 injections of I and NO_x appear large enough to perturb the chemistry of the stratosphere.
1225 However, as discussed by Pierazzo et al. (2010) significant Cl and Br injections could
1226 occur for a 1 km impact in the ocean. Seawater injections of Cl, Br, I, and S are
1227 complicated because the salts may be injected in particulate form.

1228

1229 **4. Implications for climate, atmospheric chemistry and numerical modeling, and** 1230 **suggestions for future data analysis**

1231 Since the discovery of the K-Pg impact by Alvarez et al. (1980), many papers have
1232 speculated on which of the many possible effects of the impact on the environment could

1233 have caused the mass extinction. It has become fashionable to claim that one or another
1234 effect is dominant. However, it is quite likely that several effects overlapped, each of
1235 which might have been devastating to a particular species or ecosystem, but which
1236 together made survival very difficult for a broad range of species distributed over the
1237 globe. Here we summarize the environmental perturbations we find likely. However,
1238 there are many uncertainties, and additional data are needed. We outline the data that
1239 would be useful to obtain from the geologic record, and summarize it in Table 7. Also,
1240 models have barely scratched the surface of what is possible in better understanding of
1241 the post impact environment. We summarize the types of modeling work that would be
1242 interesting to pursue. We extend these ideas to smaller impacts since more than 50
1243 impacts of kilometer-sized objects may have occurred since the extinction of the
1244 dinosaurs.

1245 Table 1, shows that spherules, soot, nano-particles, submicron clastics, and sulfates each
1246 may have had very large optical depths. An optical depth greater than unity could have
1247 serious consequences for the environment if maintained for very long. Each of these
1248 materials was likely present in the atmosphere, so they may have interacted.

1249 The spherules are unlikely to have changed climate directly because they would have
1250 been removed quickly from the atmosphere by sedimentation due to their large size.
1251 However, these particles, together with the other impact debris with significant mass,
1252 likely heated the upper atmosphere to temperatures between 1000 and 2000K. The high
1253 temperature upper atmosphere would then have irradiated the surface with near infrared
1254 radiation, causing forest fires. Wolbach et al. (1985) first recognized that the global biota
1255 likely burned after the impact, and Melosh et al. (1990) identified the mechanism for
1256 starting the fires. The recent work by Goldin and Melosh (2009) identified some
1257 complexities in the ignition mechanisms that need further work to be understood. They
1258 pointed out that the light might be blocked by the large spherules falling below the heated
1259 atmospheric layer. However, this is a complex problem since water vapor, and vaporized
1260 impactor would have been present to block radiation escaping to space. Also convection
1261 should occur in such a strongly heated layer, which would act to retard the fall of the
1262 particles as it does for hailstones in tropospheric convection. Moreover, the mass of
1263 debris injected at 70 km, as assumed by Goldin and Melosh (2009), greatly exceeds the
1264 mass of air. This mass distribution is unstable and would lead to rapid stirring of the
1265 atmosphere down to 50 km. These issues all deserve further study with suitable models.
1266 Furthermore, evidence for the nano-particles should be sought as discussed further below.

1267 Robertson et al. (2004) argued that large dinosaurs and other unsheltered animals could
1268 have been killed immediately by the radiation from the sky and the subsequent fires.
1269 However, it is possible there were refugia on the land, either in regions where spherules
1270 did not reenter the atmosphere, as suggested by Kring and Durda (2002) as well as
1271 Morgan et al. (2013), or in regions that happened to have heavy cloud cover which may
1272 have blocked the radiation. To better understand the possibility of refugia, more
1273 complete evidence for the global distribution of spherules would help resolve their
1274 possible non-uniform deposition, as suggested in Table 7. It is known that iridium was
1275 perturbed worldwide following the K-Pg impact. Although iridium concentrations are
1276 spatially variable for a number of reasons, they are basically homogenous over the Earth
1277 and do not fall off with distance from the impact site, or at high latitudes. Similar data on

1278 spherules would be useful to determine if the spherules were injected everywhere, or in
1279 special places. Numerical values of the spherule concentrations and size distributions to
1280 augment the values noted by Smit (1999) would also be of value, as noted in Table 7.
1281 Models of the transmission of the light from the hot debris layer above 60 km through
1282 dense water clouds and the response of the clouds to the heating would be also useful. It
1283 has long been recognized that intense thermal radiation and fires could not have been the
1284 only extinction mechanisms at work, since the mass extinctions in the oceans could not
1285 have occurred in this way, but instead were likely due to the low light levels preventing
1286 photosynthesis (Milne and McKay, 1982; Toon et al., 1982; Pollack et al., 1983; Toon et
1287 al., 1996; Robertson et al., 2013b). The low light levels would have been caused by the
1288 high optical depths of the soot and nano-particles that remained suspended in the air for a
1289 year or more after the impact.

1290 We know from the work of Wolbach et al. (1985; 1988; 1990; 1990b; 2003) that there is
1291 abundant soot in the K-Pg distal layer. It is highly likely that the soot originated from
1292 wildfires (Robertson et al., 2013a), but its origin is of secondary concern for climate. The
1293 widespread distribution of the soot in the layer, and the small size of the particles indicate
1294 this material was almost certainly global in extent. Wolbach et al. (1988; 1990b) show
1295 that soot and iridium are tightly correlated across the K-Pg distal layer. The soot and
1296 iridium in the distal layer must have been deposited within a few years of the impact,
1297 since small particles will not stay in the air much longer. Therefore, any fires must have
1298 been within a year or two of the impact. As noted in Table 7, further examination of the
1299 distributions of soot, iridium and spherules might clarify how long these materials
1300 remained in the atmosphere, which is expected to be days for the spherules, and a few
1301 years for the soot and iridium on small particles. Once in the water column, spherules
1302 would fall to the bottom in days or weeks. However, in the absence of fecal pellets
1303 formed by plankton around the soot, it would take decades for soot to reach the ocean
1304 depths by falling. Currents would likely carry the soot down rather than gravity.

1305 The amount of soot in the K-Pg distal layer would produce a very high optical depth
1306 when it was in the atmosphere. The transmission of light depends not only on the optical
1307 depth, but also on the single scattering albedo of the particles. The single scattering
1308 albedo measures the fraction of the light that is scattered, or absorbed. Scattering light,
1309 which occurs from sulfates that absorb sunlight only weakly, is not nearly as effective in
1310 changing climate as absorbing light.

1311 As discussed by Toon et al. (1997), soot with an optical depth of 100 would prevent any
1312 sunlight from reaching the surface—it would be pitch black. No climate simulations of
1313 such large soot optical depths have ever been conducted. However, there have been
1314 simulations for optical depths in the range of 0.05-1, which show temperatures dropping
1315 to ice age conditions within days, precipitation falling to 50% of normal, and the ozone
1316 layer being destroyed as discussed further below (Robock et al., 2007a,b; Mills et al.,
1317 2008, 2014). There are a number of complexities inherent in climate calculations for soot.
1318 For example, it is important to know how long the soot remained in the atmosphere in
1319 order to determine how long photosynthesis may have been retarded in the oceans. The
1320 lifetime of the soot in turn may depend on the size of the soot particles, their shape, the
1321 amount of rainfall in the lower atmosphere, and the amount of sunlight reaching the soot.
1322 The amount of sunlight reaching the soot matters because heating the soot also heats the

1323 surrounding air, causing it to rise and loft the soot to high altitudes, where it is protected
1324 from rainout (Malone et al. 1985; Robock et al. 2007a,b). These issues can be considered
1325 in modern climate models.

1326 Much of the vaporized impactor and target material is thought to have re-condensed to
1327 250 μm -sized spherules (O’Keefe and Ahrens, 1982; Johnson and Melosh, 2012b), which
1328 are observed, but a significant fraction may have remained as nanometer sized grains
1329 (Johnson and Melosh, 2012b). Iron-rich, nano-phase material with a diameter of 15-25
1330 nm has been identified in the fireball layer at a variety of sites by Wdowiak et al. (2001),
1331 Verma et al., (2002), Bhandari et al. (2002), Ferrow et al. (2011) and Vajda et al. (2015)
1332 among others. However, the abundance of this nano-phase material is not yet constrained
1333 by observations. As noted in Table 7, it is important to quantify the abundance of this
1334 nano-phase material, and to confirm that it is the remnant of the vaporized target and
1335 impactor. If the amount of vapor remaining at the end of the Johnson and Melosh (2012b)
1336 calculation is roughly the amount that remained as rock vapor in the atmosphere, given
1337 the optical depth estimate in Table 1 and its input location in the upper atmosphere above
1338 the soot generated by forest fires, this nano-phase material would be the dominant source
1339 of opacity for changing the climate, and would also greatly affect the amount of radiation
1340 emitted to the surface that could start wildfires in the hours following the impact. The
1341 material contains iron, so it is likely to have been a good absorber of sunlight.
1342 Alternatively, this material might have attached itself to the large spheres and been
1343 quickly removed, though this seems unlikely since the large spheres would separate
1344 gravitationally from the smaller material within hours. No one has yet considered the
1345 effect of this nano-phase material, which is distinct from the clastics envisioned by Toon
1346 et al. (1997) and Pope (2002), on the environment after the K-Pg impact.

1347 The most massive part of the ejecta from the K-Pg crater consisted of clastics: crushed
1348 and pulverized material. Much of this material fell relatively close to the crater, though
1349 significant amounts were emplaced as far a 4000 km from Chicxulub. For comparison the
1350 Toba volcanic eruption about 70,000 years ago is estimated to have released more than
1351 2×10^{18} g of clastics (Matthews et al., 2012), a factor of about 15 less than our estimate for
1352 the Chicxulub impact in Table 1, but more than 200 times greater than the upper limit
1353 previously estimate by Pope (1997) for the clastics generated by Chicxulub.

1354 The Toba eruption may have had a significant impact on the climate, as discussed further
1355 below; however, the magnitude of the effect is controversial. Alvarez et al. (1980), as
1356 well as Toon et al. (1982) and Pollack et al. (1983), thought that the K-Pg layer was
1357 dominated by submicron clastics that caused major loss of sunlight at the surface and
1358 consequently very low temperatures. However, while we don’t know the fraction of the
1359 layer composed of submicron clastics, it is clear that the layer is both thinner than thought
1360 in the years just after its discovery and also dominated by other parts of the impact debris
1361 such as the spherules and the nano-particles. It would be very useful to measure the
1362 amount of submicron clastics in the K-Pg distal layer. Possibly, as suggested in Table 7,
1363 one could start by identifying the amount of submicron quartz in the layer by searching
1364 for small shocked quartz grains. Toon et al. (1997), and Pope (2002) used two differing
1365 indirect approaches to quantify the submicron clastics, and came up with answers that
1366 differ by a factor of about 10^4 . Here we attempted to reconcile these approaches, with the
1367 result shown in Table 1 yielding a significant optical depth. Although the submicron

1368 clastics by themselves would have produced extreme climate changes if they were as
1369 abundant as we estimate, they would have been less important than the soot, and the
1370 nano-particles given our estimates here. The submicron clastics may have been injected
1371 higher than the soot, but lower than the nano-particles on average. Climate calculations
1372 involving all these materials are needed to understand how they may have interacted in
1373 the atmosphere.

1374 The final particulates with large optical depths in Table 1 are sulfates. Pope et al. (1997),
1375 Pierazzo et al. (2003) and others have advocated for the importance of these particles in
1376 recent years. Unfortunately, sulfates in the K-Pg layer have not been traced
1377 unambiguously to the impact, because sulfur is so common in the environment. Possibly
1378 sulfur isotopic studies could distinguish the sulfur in the impactor from sulfur in the
1379 terrestrial environment, but we are not aware of such studies. While there is little doubt
1380 that large amounts of sulfur were present in the target material and in the asteroid, it is
1381 possible that much of it reacted with the hot rock in the impact plume, or the atmospheric
1382 layer heated by re-entering material. Sulfur is present in impact melt spherules and in
1383 carbonaceous clastics, so not all of it was released to the gas phase. Given the large
1384 opacity of the numerous types of particles in the atmosphere, photochemical reactions
1385 would have been inhibited, which would retard the conversion of sulfur dioxide gas into
1386 sulfate particles. It is possible that measurements of the sulfur mass independent
1387 fractionation (MIF) could reveal whether the sulfur quickly reacted with rocks, which
1388 should yield a MIF of zero, or if the sulfur slowly converted to sulfate, which might lead
1389 to MIF not being zero if resolved over the thickness of the distal layer. It is known that a
1390 non-zero MIF can occur following volcanic eruptions due to time dependent movement
1391 of sulfur between changing sulfur reservoirs in the atmosphere (e.g. Pavlov et al., 2005).

1392 It is not clear if SO_3 or SO_2 was the dominant sulfur bearing gas in the ejecta plume.
1393 However, the gas phase reaction of SO_3 and water is not a simple reaction as often
1394 abbreviated in papers about atmospheric sulfur chemistry, but instead involves water
1395 vapor clusters or SO_3 adducts. Sulfur dioxide is observed to convert to particulates with
1396 an e-folding time of less than one month for moderate-sized volcanic eruptions such as
1397 the Mt. Pinatubo eruption. Following the K-Pg impact sulfur dioxide or trioxide gas may
1398 have had an extended lifetime in the atmosphere, due to the lack of sunlight to drive
1399 chemical reactions to convert it to sulfates. Clastics and nano-particles and soot, may
1400 have coagulated to large sizes and fallen out over a year or two. Alternatively, the sulfur
1401 gases may have reacted quickly on all the surfaces present, particularly in hot water
1402 present in the hot radiating layer when the ejecta reentered. Pope et al. (1997) and
1403 Pierazzo et al. (2003) have pointed out the possible importance of the extended lifetime
1404 of the sulfate to causing a prolonged period without photosynthesis in the oceans.
1405 However, clastics or soot need to be present in the sulfate to achieve the loss of sunlight.
1406 Recent work on the Toba eruption (Timmreck et al., 2010) shows that large sulfur
1407 injections do not produce proportionately larger climate perturbations because the climate
1408 effects of sulfur injections are self-limiting, as originally shown by Pinto et al. (1994) and
1409 recognized by Pope et al. (1997) and Pierazzo et al. (2003). Toba probably injected an
1410 amount of sulfur dioxide within an order of magnitude of that from the K-Pg impact.
1411 Larger particles have smaller optical depths, and shorter lifetimes, than smaller particles
1412 that result from smaller SO_2 injections. Further work is needed to understand the

1413 chemistry of the sulfur injected by the Chicxulub impact to determine if it was a
1414 significant factor in the extinction event.

1415 Table 2 shows that significant injections of various ozone destroying chemicals such as
1416 NO_x, Cl, Br, and I, likely occurred. The effects of these gases need to be considered in
1417 calculations but, given the expected darkness, photochemistry may have ceased until the
1418 atmosphere cleared.

1419 Table 3 suggests that the much smaller mass injections from the impact of a 1 km
1420 diameter asteroid on land may produce optical depths that may still be important.
1421 Climate models are needed to fully evaluate these perturbations. At first glance the
1422 injections seem small. For example, the sulfur injection is only about 4 times larger than
1423 that from the Pinatubo eruption. However, the soot injection is very large. Robock et al.
1424 (2007a) and Mills et al. (2014) examined smoke injections at the tropopause of about one
1425 third the 1 km asteroid injection near the tropopause and found that the ozone layer was
1426 severely damaged, and low enough temperatures resulted to damage crops for a decade
1427 after the injection. Table 4 also indicates significant injections of iodine, which may
1428 further damage the ozone layer.

1429 About 50 1-km impacts might have occurred since the demise of the dinosaurs. Based on
1430 the fraction of Earth covered by water, about 35 of these would be expected to have hit
1431 the oceans, perhaps resulting in large ozone losses as discussed by Pierazzo et al. (2010).
1432 Each of the 15 impacts that occurred on land might have led to significant injections of
1433 nano-particles. Paquay et al. (2008) recognized the osmium signature of two large
1434 impacts in the Late Eocene, which produce the 100 km diameter craters at Popigai and
1435 Chesapeake Bay. The osmium indicates a substantial input of vaporized impactor to the
1436 atmosphere from collisions of asteroids larger than 1 km in diameter. Climate model
1437 simulations are needed to evaluate the climate changes that might have occurred. The
1438 effects could have been variable for a variety of reasons, including variability in the light
1439 absorbing properties of rock from differing objects. To have injected significant amounts
1440 of smoke the impactor would need to hit a tropical forest, or at least a heavily forested
1441 region. About 26% of the world is currently forested; about 6% is in tropical rain forest.
1442 Forested area has greatly declined. Tropical rainforests might have covered as much as
1443 20% of the Earth until recently. Hence, about 3 1-km objects might have hit a tropical
1444 rainforest and injected significant amounts of smoke since the K-Pg event.

1445 In this work we have established a set of initial conditions (Tables 1-4) that may be used
1446 for modeling the climate and air chemistry after the K-Pg impact, or the impact of a 1 km
1447 asteroid. Other authors have considered some of these initial conditions, but some, such
1448 as the nano-particles from the vaporized impactor, have not been previously studied in
1449 the detail needed to fully evaluate their importance. Much more work is needed to obtain
1450 field data to further constrain some of parameters, and to resolve remaining differences of
1451 opinion about some of the values. However, simulations using these initial conditions can
1452 now be conducted with modern models of climate and atmospheric chemistry, which
1453 should shed light on the environmental conditions at the K-Pg boundary and the dangers
1454 posed by future impacts. We recently completed such simulations using the Whole
1455 Atmosphere Community Climate Model (WACCM) at the National Center for
1456 Atmospheric Research in a configuration similar to that used by Bardeen et al. (2008) and
1457 Mills et al. (2014).

1458 **Author contributions:** Owen Toon worked to compile the particle and gas emissions.
1459 Charles Bardeen tested them in a climate model to determine if the initial conditions were
1460 specified completely. Rolando Garcia considered the gases that would be important for
1461 atmospheric chemistry.

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1466

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1779

1780 Table 1: K-Pg injection scenario for impactor mass $\sim 1.4 \times 10^{18}$ g, impact energy $\sim 2.8 \times 10^{23}$
 1781 $J = 6.8 \times 10^7$ Mt for 20 km/s impact

Property/ Constituent	Type 2 spherules	Soot	Nano- particles	Clastics, < μm	S
Material amount, g, column density (g cm^{-2})	2.3×10^{18} (0.44)	$1.5\text{-}5.6 \times 10^{16}$ (0.29 to 1.1×10^{-2}) ^{***}	$\sim 2 \times 10^{18}$ ^{**} (0.4)	$< 6 \times 10^{16}$ (0.01)	9×10^{16} (5.4×10^{-2} g SO_4/cm^2)
Global optical depth as $1 \mu\text{m}$ particles *	~ 20 (for $250 \mu\text{m}$ particles)	~ 100	~ 2000	~ 90	~ 450
Vertical distribution	70 km, Gaussian distribution with half width of 6.6 km ^{****}	Eq. 2	Same as Type 2 spherules	Uniformly mixed vertically above tropopause	Same as Type 2 spherules
Optical properties	Not relevant	$n=1.8$ $k=0.67$	Hervig et al., (2009)	Orofino et al. (1998) limestone	Sulfuric acid
Initial Particle size	$250 \mu\text{m}$ diameter	Lognormal, $r_m=0.11 \mu\text{m}$, $\sigma=1.6$; monomers 30-60 nm	20 nm diameter	Lognormal, $r_m=0.5 \mu\text{m}$, $\sigma=1.65$	gas
Material density, g cm^{-3}	2.7	1.8	2.7	2.7	1.8

1782 *Qualitative estimate for comparison purposes only

1783 **This value is an upper limit. The lower limit is zero

1784 ***These values are for aciniform soot, or elemental carbon in the stratosphere, see text.

1785 ****The material may have quickly moved to below 50 km to maintain hydrostatic
 1786 balance. See text.

1787

1788

1789 Table 2: Gas phase emissions (g) from the Chixculub impact

Sources/ Gases ****	S (x10 ¹³)	C (as CO ₂ **) (x10 ¹⁷)	H ₂ O (x10 ¹⁵)	Cl (x10 ¹²)	Br (x10 ¹⁰)	I (x10 ⁷)	N (x10 ¹⁴)	Vertical distribution
Ambient burden (g)	1*	8.4	1.3 strat	2.3 strat	3.1 strat	<2.3 strat	2 as N ₂ O	
Impactor	4x10 ³	0.3	200	7x10 ²	5x10 ²	7x10 ⁴		As Type 2 spherules
Forest fires	40	6	1500	200	1000	9x10 ⁵	10	As soot
Vaporized sea water	60	small	600	1x10 ⁴	5x10 ³	40	-	As Type 2 spherules
Splashed sea water ***	500	small	5x10 ³	1x10 ⁵	4x10 ⁴	3x10 ²	-	Uniformly mixed above tropopause
Impact site (vaporized)	5000	0.6	90	800	400	3		As Type 2 spherules
Impact site (degassed)	500	0.1	120	2x10 ³	1x10 ³	7		Uniformly mixed above tropopause
Air heating							300 as NO _x created from air	Half uniformly mixed, half as Type 2 spherules

1790 * Based on Pinatubo eruption

1791 ** Mass is given in terms of C, but emission is in the form of CO₂

1792 *** S, Cl, Br, I likely injected as particulates

1793 **** The scaling factors given in () apply to all values in column.

1794

1795

1796 Table 3: 1 km land* injection scenario for impactor mass 1.4×10^{15} g; impactor energy
 1797 $\sim 2.8 \times 10^{20}$ J = 6.8×10^4 Mt

Property/ Constituent	Type 2 spherules	Soot**	Nano- particles from vaporized rock***	Clastics, $< \mu\text{m}$ distributed globally	S
Material amount g, column density (g cm^{-2})	1.4×10^{15} (2.6×10^{-4})	2.8×10^{13} (5.6×10^{-6})	1×10^{15} (2×10^{-4})	2.6×10^{13} (5×10^{-6})	4.4×10^{13} (2.6×10^{-5} g $\text{SO}_4 \text{ cm}^{-2}$)
Estimated global optical depth as 1 μm particles	0.2 (as 15 μm particles)	4.7×10^{-2}	1.5	4×10^{-2}	0.22
Vertical & horizontal distributions	Table 1 Over 50% of Earth	50% Eq. 2+50% Eq 3 Over 4×10^4 km^2	Table 1 Over 50% of Earth	Uniformly mixed above tropopause, spread over $4 \times 10^5 \text{ km}^2$	Follow nano- particles
Optical properties	Not relevant	Table 1	Table 1	Depends on impact site	Table 1
Initial particle size (μm)	15 μm	Table 1	20 nm	Table 1	

1798 *We assume a 1 km asteroid impact would not penetrate through the 5km average depth
 1799 of the ocean. Therefore, none of the materials in this Table would be injected into the
 1800 atmosphere for an ocean impact. For the density of all materials follow Table 1.

1801 **The material amount assumes an impact into a region where 2.25 g C cm^{-2} flammable
 1802 biomass is consumed. The material amount can be scaled linearly for other choices of
 1803 available biomass that burns.

1804 ***We assume about 35% of the impactor and an equivalent mass of target would vaporize
 1805 and end up as nano-particles. This value is an upper limit. The lower limit is zero.

1806

1807

1808 Table 4: Gas phase emissions (g) from a 1-km diameter impact

Sources/ Gases****	S (x10 ¹³)	C* (x10 ¹⁷)	H ₂ O (x10 ¹⁵)	Cl (x10 ¹²)	Br (x10 ¹⁰)	I (x10 ⁷)	N (x10 ¹⁴)	Vertical distribution
Ambient burden (g)	1**	8.4	1.3 strat	2.3 strat	3.1 strat	<2.3 strat	2 as N ₂ O**	
Impactor/ land only	4.4	3x10 ⁻²	0.2	0.7	0.5	68	-	As type 2 spherules
Forest fires/land only	2.7 x10 ⁻²	4x10 ⁻³	0.9	0.12	0.62	560	6.9x10 ⁻³	As soot
Vaporized sea water	0.9	small	10	200	80	0.6		Uniformly mixed
Splashed sea water***	3	small	30	600	200	2		
Air heating							0.6	Uniformly mixed

1809 * mass is given in terms of C, but emission is in the form of CO₂

1810 **based on Pinatubo volcanic eruption

1811 ***S, Cl , Br, I may be released as particulates

1812 **** scaling factors given in () apply to all values in column

1813

1814

1815 Table 5: Comparison of Toon et al. (1997) and Pope (2002) estimates of submicron
 1816 clastics.

Method	Quartz based estimate- Pope (2002)	Injected mass-Toon et al. (1997)*	Injected mass - revised	Quartz based estimate-revised	1 km impactor**
Initial clastic debris, g	$<10^{16}$	7×10^{20}	2.9×10^{19}	5×10^{18}	1.3×10^{16}
% clastic $<1 \mu\text{m}$	<1	0.1	2	1	2
Submicron clastics, g	$<10^{14}$	7×10^{17}	5.8×10^{17}	5×10^{16}	2.6×10^{14}
Stratospheric submicron surviving initial removal, g	10^{14}	7×10^{17}	$<5.8 \times 10^{16}$	5×10^{16}	$< 2.6 \times 10^{13}$

1817 * assuming an impact energy of $1.5 \times 10^8 \text{Mt}$, and a velocity of 20 km/s.

1818 ** scaled from Injected Mass Revised using energy scaling assuming an impact energy of
 1819 $6.8 \times 10^4 \text{Mt}$

1820

1821

1822

1823 Table 6: Impactor composition, seawater composition, Yucatan impact site composition
 1824 and forest fire emission ratios

	S	C	H ₂ O	Cl	Br	I	EC	N
Carbonaceous Chondrite (g/g impactor)	3.1 x10 ⁻²	1.98 x10 ⁻²	11.9 x10 ⁻²	4.7 x10 ⁻⁶	3.27 x10 ⁻⁶	4.8 x10 ⁻⁷		
Sea water (g/g sea water)	9.1 x10 ⁻⁴	3 x10 ⁻⁶	0.965	1.9 x10 ⁻²	8.2 x10 ⁻⁵	6.0 x10 ⁻¹⁰	-	-
Impact site (g/g site)	7.1 x10 ⁻²	9.6 x10 ⁻²	0.07					
Emission ratios for forest fires g/g of dry biomass burned	2.9 x10 ^{-4*}	4.3 x10 ⁻¹ as CO ₂ 4.4 x10 ⁻² as CO 5.1 x10 ⁻³ as CH ₄	Highly variable, can equal dry weight	As CH ₃ Cl 1.4 x10 ⁻⁵ to 1.3 x10 ⁻⁴	As CH ₃ Br 6.7 x10 ⁻⁶	As CH ₃ I 6.1 x10 ⁻⁶	6.6 x10 ^{-4**}	7.5 x10 ⁻⁴ as NO 6 x10 ⁻⁵ as N ₂ O

1825 *The mass is given in terms of S, but the emission is in the form of SO₂.

1826 ** We used 0.03 g/g in Table 3, because forest fires will not produce as much soot as mass
 1827 fires.

1828

1829 Table 7 Suggestions for data collection

Property of interest	Rationale
Global distribution of spherules	Some impact models suggest spherules were not distributed globally, limiting area of Earth that might experience fire ignition
Number concentration, size of spherules	Current data are incomplete on number and size of spherules
Soot distribution	Profile soot/iridium/spherule distribution to determine if fires are contemporaneous with iridium fallout
Nano-meter material	Nano-meter material has been detected, but its mass needs to be quantified
Clastics	Submicron component not detected. Possibly search for micron/submicron shocked quartz.
Sulfur	Use sulfur isotopes to search for extraterrestrial sulfur, sulfur MIF to test for prolonged lifetime

1830

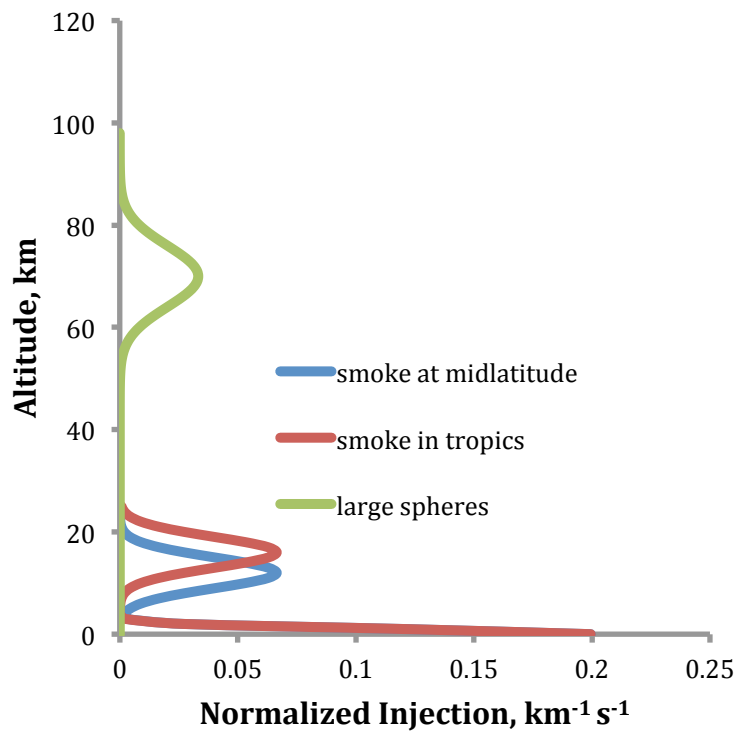


Figure 1. Injection profiles for smoke at midlatitudes and the tropics and for large spherical particles. Many other constituents follow the same vertical profiles as noted in Table 1-4. We suggested clastics be placed above the tropopause using a constant mixing ratio.

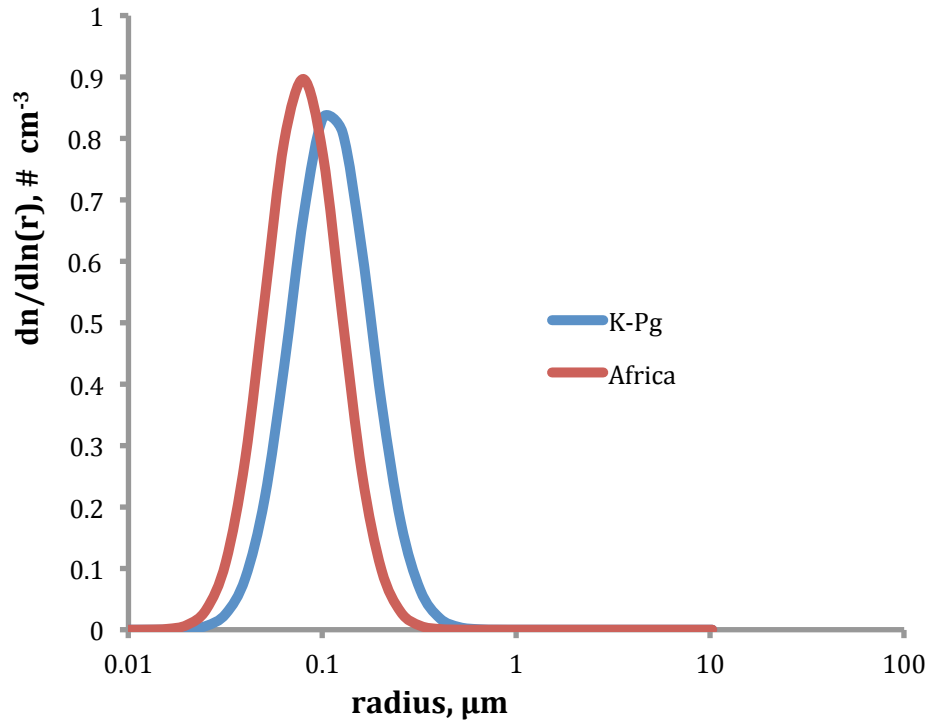


Fig. 2. The size distributions for smoke from modern fires in Africa, and from the K-Pg boundary layer (Wolbach et al., 1985; Matichuk et al., 2008)

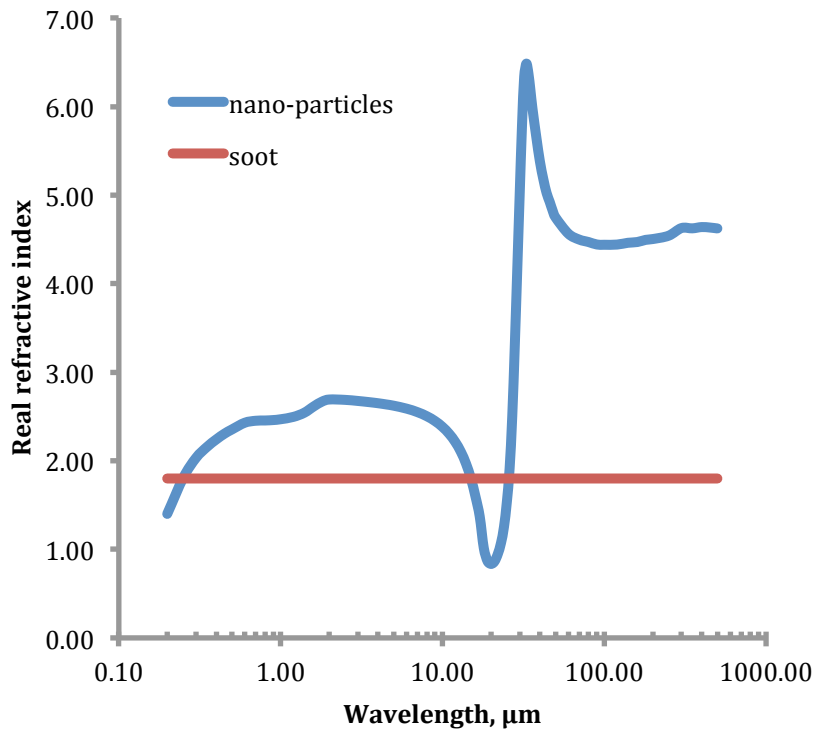
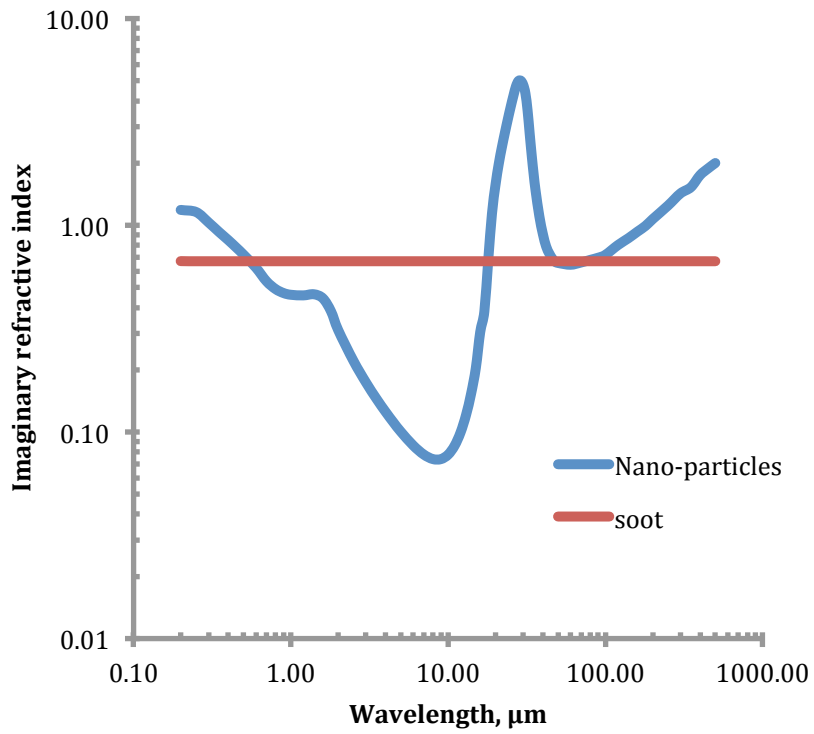


Fig. 3 The real and imaginary parts of the refractive index suggested for nano-particles, and for soot.