

## Responses to Editor

Comments to the Author:

Please be so kind to apply the minor changes suggested by the referee and submit again the revised paper. Also please give more details to the PMF analysis description as requested by the 2 referees and by the one that has revised the last version.

We thank the editor for kind handling of this paper. As suggested, very detailed descriptions about the configuration, run, results and error estimation of the PMF were provided in the last and present versions of the manuscript, including the inputs of concentrations and uncertainties, the S/N ratios, the Q values, the ratio between Q robust and Q true, the agreement between observed and predicted values, the residuals, the G-space plots and the errors estimated with bootstrap method. These criteria and tools fully ensured the reasonability of the source apportionment results. Detailed responses to the Referee #4's comments were provided item by item below.

## Responses to Anonymous Referee #4

Methodology: Please mention that WSI data was not available for summer.

Thanks for the suggestion. Revisions were made as follows.

However, data were not available in May and June, because the instrument was initially deployed in September.

For details, please refer to lines 130-131, page5.

Lines 49, 60: Change chemical compositions to chemical composition

Accepted with thanks.

Line 101: Change deeply to comprehensively

Accepted with thanks.

Lines 103-104: Metal elements?

Thanks for the question. Since some elements were not metals, such as As and Se. "Metal elements" was changed to "elements" throughout the manuscript.

Figure 3: Does this figure refer to autumn data?

Thanks for the question. Actually, this figure shows the data in both summer and autumn. WSIs are represented with green areas, which are not shown in May and June due to the lack of data.

Lines 340-346: Does elemental K correlate with K+?

Good question. During the sampling period,  $K^+$  correlated well with K with  $R^2$  and slope of 0.88 and 0.80, respectively. Revisions were made as follows.

K<sup>+</sup> monitored by the online ion chromatography correlated well ( $R^2 = 0.88$ ; slope = 0.80) with K monitored by the customized metal analyzer. To keep consistency with other elements, K rather than K<sup>+</sup> was used to do the following analyses in this study.

For details, please refer to lines 136-139, page 5.

Line 396: Cu is used as an antioxidant in brake pads. This is not clear. Please elaborate. Refer to the following paper for brake pad composition, and emissions: Grigoratos, T. and Martini, G. (2015) Brake pad particle emissions: a review. Environ Sci Pollut Res Intl, 22:2491-2504.

Thanks for the comment and suggestion of the reference.

Cu has been identified as one of the most abundant metals in both brake linings and the brake wear particles with the concentration of up to 210 mg/g in brake wear dust (Grigoratos and Martini, 2015).

For details, please refer to lines 426-428, page 19 and lines 692-693, page 32.

PMF model: S/N ratio? Were data points not confirming to the residual range (-3 to +3) removed from the analysis?

Thanks for the questions. The signal-to-noise (S/N) ratios were all greater than 1, indicating “good” signal for all the species used for source apportionment, according to the PMF 5.0 User Guide. Data points not confirming to the residual range (-3 to +3) were removed from the analysis. Revisions were made as follows.

The signal-to-noise (S/N) ratios were all greater than 1, indicating “good” signal for all the species involved in source apportionment, according to the PMF 5.0 User Guide.

For details, please refer to lines 190-192, page 7.

Data points not confirming to the residual range (-3 to +3) were removed from the analysis.

For details, please refer to lines 206-207, page 8.

Figure 10: It is a surprising that more EC is apportioned to biomass burning compared to vehicle emissions both in summer and autumn.

We highly appreciate the comment. As a tracer of combustion, EC is generally emitted from vehicle emissions, coal combustion, biomass burning and other combustion activities. Streets et al. (2001) reviewed previous studies and estimated the emission factors of EC in China from different sectors. Results indicated that the average emission factors were 0.08 and 1.1 g/kg fuel for gasoline and diesel vehicular emissions, respectively, while they were 0.90, 0.58 and 0.72 g/kg for the field combustion of wheat, rice and corn residuals, respectively. Since diesel vehicles only accounted for 11.1% of the vehicle fleet in Wuhan (the data were available at <http://www.whepb.gov.cn/u/cms/whepb/201506/051551398i2j.pdf>), it was reasonable that more EC was apportioned to biomass burning compared to vehicle emissions.

On the other hand, May and June in summer and October and November in autumn are the periods with intensive biomass burning in and/or around Wuhan, particularly the combustion of

crop residuals in agricultural provinces in central China. Namely, the intensive biomass burning during the study period was another factor leading to higher EC emitted by biomass burning than vehicular emissions.

Revisions were made as follows.

Both biomass burning and vehicular emissions are important sources of EC. The much lower EC apportioned to vehicular emissions in this study is explained in Section 1 of the Supplement.

For details, please refer to lines 432-434, page 19.

Section 1 Explanation of higher EC apportioned to biomass burning compared to vehicular emissions.

As a tracer of combustion, EC is generally emitted from vehicular emissions, coal combustion, biomass burning and other combustion activities. Streets et al. (2001) reviewed previous studies and estimated the emission factors of EC in China from different sectors. Results indicated that the average emission factors were 0.08 and 1.1 g/kg fuel for gasoline and diesel vehicular emissions, respectively, while they were 0.90, 0.58 and 0.72 g/kg for the field combustion of wheat, rice and corn residuals, respectively. Since diesel vehicles only accounted for 11.1% of the vehicle fleet in Wuhan (the data were available at <http://www.whepb.gov.cn/u/cms/whepb/201506/051551398i2j.pdf>), it was reasonable that more EC was apportioned to biomass burning compared to vehicular emissions.

On the other hand, May and June in summer and October and November in autumn are the periods with intensive biomass burning in and/or around Wuhan, particularly the combustion of crop residuals in agricultural provinces in central China. Namely, the intensive biomass burning during the study period was another factor leading to higher EC emitted by biomass burning than vehicular emissions.

For details, please refer to Section 1 in the Supplement.

Table 4: Contribution of coal is very different between non-episodes 1 and 2. Why?

Many thanks for the question. The source apportionment results indicated that contribution of coal combustion in non-episode 2 was significantly lower than that in non-episode 1 ( $p < 0.05$ ). Bearing in mind the uncertainties caused by the lack of WSIs in summer, the lower contribution of coal combustion in non-episode 2 might be attributable to the National Day holiday from October 1 to 7. During the holiday, the coal-fired boilers in factories and power plants stopped working, which would significantly reduce  $PM_{2.5}$  emissions from coal combustion. The much lower mass contribution of coal combustion in early October (as shown in Figure S9) coincided with this inference.

Revisions were made as follows.

In addition, we noted that the contribution of coal combustion was much lower in non-episode 2 than that in non-episode 1 ( $p < 0.05$ ). The explanation is provided in Section 2 of the Supplement.

For details, please refer to lines 460-462, page 20.

The source apportionment results indicated that contribution of coal combustion in non-episode 2 was significantly lower than that in non-episode 1 ( $p < 0.05$ ). Bearing in mind the uncertainties caused by the lack of WSIs in summer, the lower contribution of coal combustion in non-episode 2 might be attributable to the National Day holiday from October 1 to 7. During the holiday, the coal-fired boilers in factories and power plants stopped working, which would significantly reduce  $PM_{2.5}$  emissions from coal combustion. The much lower mass contribution of coal combustion in early October (as shown in Figure S10) coincided with this inference.

For details, please refer to Section 2 of the Supplement.

#### References:

Streets, D.G., Gupta, S., Waldhoff, S.T., Wang, M.Q., Bond, T.C., and Yiyun, B., 2001. Black carbon emissions in China. *Atmos. Environ.* 35, 4281-4296.

1 Chemical characteristics and causes of airborne particulate pollution in warm  
2 seasons in Wuhan, central China

3

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15 **Abstract:** Continuous measurements of airborne particles and their chemical compositions were  
16 conducted in May, June, October, and November 2014 at an urban site in Wuhan, central China.  
17 The results indicate that particle concentrations remained at a relatively high level in Wuhan,  
18 with averages of  $135.1 \pm 4.4$  (mean  $\pm$  95% confidence interval) and  $118.9 \pm 3.7 \mu\text{g}/\text{m}^3$  for  $\text{PM}_{10}$   
19 and  $81.2 \pm 2.6$  and  $85.3 \pm 2.6 \mu\text{g}/\text{m}^3$  for  $\text{PM}_{2.5}$  in summer and autumn, respectively. Moreover,  
20  $\text{PM}_{2.5}$  levels frequently exceeded the National Standard Level II (i.e., daily average of  $75 \mu\text{g}/\text{m}^3$ ),  
21 and six  $\text{PM}_{2.5}$  episodes (i.e., daily  $\text{PM}_{2.5}$  averages above  $75 \mu\text{g}/\text{m}^3$  for 3 or more consecutive days)  
22 were captured during the sampling campaign. Potassium was the most abundant element in  $\text{PM}_{2.5}$ ,  
23 with an average concentration of  $2060.7 \pm 82.3 \text{ ng}/\text{m}^3$ ; this finding indicates intensive biomass  
24 burning in and around Wuhan during the study period, because almost no correlation was found  
25 between potassium and mineral elements (iron and calcium). The source apportionment results  
26 confirm that biomass burning was the main cause of episodes 1, 3, and 4, with contributions to  
27  $\text{PM}_{2.5}$  of  $46.6\% \pm 3.0\%$ ,  $50.8\% \pm 1.2\%$ , and  $44.8\% \pm 2.6\%$ , respectively, whereas fugitive dust  
28 was the leading factor in episode 2. Episodes 5 and 6 resulted mainly from increases in vehicular  
29 emissions and secondary inorganic aerosols, and the mass and proportion of  $\text{NO}_3^-$  both peaked  
30 during episode 6. The high levels of  $\text{NO}_x$  and  $\text{NH}_3$  and the low temperature during episode 6  
31 were responsible for the increase of  $\text{NO}_3^-$ . Moreover, the formation of secondary organic carbon

32 was found to be dominated by aromatics and isoprene in autumn, and the contribution of  
33 aromatics to secondary organic carbon increased during the episodes.

34 **Keywords:** PM<sub>2.5</sub>; NO<sub>3</sub><sup>-</sup>; SOA; biomass burning; formation mechanism

35

## 36 **1. Introduction**

37 Airborne particulate pollution, also called “haze,” has swept across China in recent years,  
38 particularly over its northern, central, and eastern parts (Cheng et al., 2014; Kang et al., 2013;  
39 Wang et al., 2013). Due to its detrimental effects on human health (Anderson et al., 2012;  
40 Goldberg et al., 2001), the atmosphere (Yang et al., 2012; White and Roberts, 1977), acid  
41 precipitation (Zhang et al., 2007; Kerminen et al., 2001), and climate change (Ramanathan et al.,  
42 2001; Nemesure et al., 1995), particulate pollution has become a major concern of scientific  
43 communities and local governments. China’s national ambient air quality standards issued in  
44 2012 regulate the annual upper limit of PM<sub>10</sub> (i.e., particulate matter with an aerodynamic  
45 diameter of less than 10 μm) and PM<sub>2.5</sub> (i.e., particulate matter with an aerodynamic diameter of  
46 less than 2.5 μm) as 70 μg/m<sup>3</sup> and 35 μg/m<sup>3</sup> and 24-h averages as 150 μg/m<sup>3</sup> and 75 μg/m<sup>3</sup>,  
47 respectively (GB 3095-2012).

48 Numerous studies have been conducted in China to understand the spatiotemporal variations in  
49 particle concentrations, the chemical composition, and the causes of haze events (Cheng et al.,  
50 2014; Cao et al., 2012; Zheng et al., 2005; Yao et al., 2002). In general, particulate pollution is  
51 more severe in winter due to additional emissions (e.g., coal burning) and unfavorable dispersion  
52 conditions (Lyu et al., 2015a; Zheng et al., 2005). Northern China often suffers heavier, longer,  
53 and more frequent haze pollution than southern China (Cao et al., 2012). Chemical analysis  
54 indicates that secondary inorganic aerosol (SIA; i.e., sulfate [SO<sub>4</sub><sup>2-</sup>], nitrate [NO<sub>3</sub><sup>-</sup>], and  
55 ammonium [NH<sub>4</sub><sup>+</sup>]) and secondary organic aerosol (SOA) dominate the total mass of airborne  
56 particles (Zhang et al., 2014; Zhang et al., 2012). However, the composition differs among the  
57 size-segregated particles. In general, secondary species and mineral or sea salt components are  
58 prone to be apportioned in fine and coarse particles (Zhang et al., 2013; Theodosi et al., 2011).  
59 Indeed, the general characteristics of particles (e.g., toxicity, radiative forcing, acidity) are all  
60 tightly associated with their chemical compositions and physical sizes, which therefore have  
61 been extensively studied in the field of aerosols. To better understand and control airborne  
62 particulate pollution, the causes and formation mechanisms have often been investigated (Wang

63 et al., 2014a and b; Kang et al., 2013; Oanh and Leelasakultum, 2011). Apart from the  
64 unfavorable meteorological conditions, emission enhancement was often the major culprit. There  
65 is little doubt that industrial and vehicular emissions contributed greatly to the particle mass via  
66 direct emission and secondary formation of particles from gaseous precursors, such as sulfur  
67 dioxide (SO<sub>2</sub>), nitrogen oxides (NO<sub>x</sub>), and volatile organic compounds (VOCs; Guo et al., 2011a).  
68 In addition, some other sources in specific regions or during specific time periods have also built  
69 up the particle concentrations to a remarkable degree, e.g., coal combustion in north China (Cao  
70 et al., 2005; Zheng et al., 2005) and biomass burning in Southeast Asia (Deng et al., 2008; Koe et  
71 al., 2001). Furthermore, some studies have explored the possible formation mechanisms of the  
72 main particle components (SIA and SOA) and distinguished the contributions of different  
73 formation pathways. For example, Wang et al. (2014) demonstrated that heterogeneous oxidation  
74 of SO<sub>2</sub> on aerosol surfaces was an important supplementary pathway to particle-bound SO<sub>4</sub><sup>2-</sup> in  
75 addition to gas phase oxidation and reactions in clouds. In contrast, it was reported that  
76 homogeneous and heterogeneous reactions dominated the formation of NO<sub>3</sub><sup>-</sup> during the day and  
77 night, respectively (Pathak et al., 2011; Lin et al., 2007; Seinfeld and Pandis, 1998). Furthermore,  
78 biogenic VOCs and aromatics were shown to be the main precursors of SOA (Kanakidou et al.,  
79 2005; Forstner et al., 1997).

80 Despite numerous studies, the full components of airborne particles have seldom been reported  
81 due to the cost of sampling and chemical analysis, resulting in a gap in our understanding of the  
82 chemical characteristics of particles. In addition, although the causes of particle episodes have  
83 been discussed in many case studies (Wang et al., 2014; Deng et al., 2008), the contributions  
84 have rarely been quantified. Furthermore, the formation mechanisms might differ in various  
85 circumstances. Therefore, an overall understanding of the chemical characteristics of airborne  
86 particles, the causes of the particle episodes, and the formation mechanisms of the enhanced  
87 species would be of great value. In addition, the frequent occurrence of haze pollution has  
88 become a regular phenomenon in central China during warm seasons, but the causes have not  
89 been identified and the contributions have not been quantified. Wuhan is the largest megacity in  
90 central China and has suffered from severe particulate pollution in recent years. The data indicate  
91 that the frequency of days in which PM<sub>2.5</sub> exceeded the national standard level II (i.e., a daily  
92 average of 75 µg/m<sup>3</sup>) in Wuhan reached 55.1% in 2014 (Wuhan Environmental Bulletin, 2014).  
93 In the warm seasons of 2014, the hourly maximum PM<sub>2.5</sub> (564 µg/m<sup>3</sup>) was even higher than that

94 in winter ( $383 \mu\text{g}/\text{m}^3$ ), as shown in [Figure S1](#) in the Supplementary Material. Moreover, because  
95 the air quality in Wuhan is strongly influenced by the surrounding cities, the pollution level in  
96 Wuhan also reflects the status of the city clusters in central China. However, previous studies  
97 ([Lyu et al., 2015a](#); [Cheng et al., 2014](#)) did not allow a complete understanding of the properties  
98 of airborne particles in this region, particularly during the warm seasons, nor could they guide  
99 control strategies. It is therefore urgent to understand the chemical characteristics of airborne  
100 particles and to explore the causes and formation mechanisms of the particle episodes in Wuhan.  
101 This study **comprehensively** analyzed the chemical characteristics of  $\text{PM}_{2.5}$  in Wuhan from a full  
102 suite of component measurement data:  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{NH}_4^+$ , organic carbon (OC), including  
103 primary organic carbon (POC) and secondary organic carbon (SOC), elemental carbon (EC), and  
104 **elements**. Furthermore, based on the analysis of meteorological conditions, chemical signatures,  
105 source apportionment, and distribution of wildfires, the causes of the  $\text{PM}_{2.5}$  episodes are  
106 identified and their contributions quantified. Finally, this study used a photochemical box model  
107 incorporating a master chemical mechanism (PBM-MCM) and theoretical calculation to  
108 investigate the formation processes of  $\text{NO}_3^-$  and SOC. Ours is the first study to quantify the  
109 contribution of biomass burning to  $\text{PM}_{2.5}$  and examine the formation mechanisms of both  
110 inorganic and organic components in  $\text{PM}_{2.5}$  in central China.

111

## 112 **2. Methods**

### 113 **2.1 Data collection**

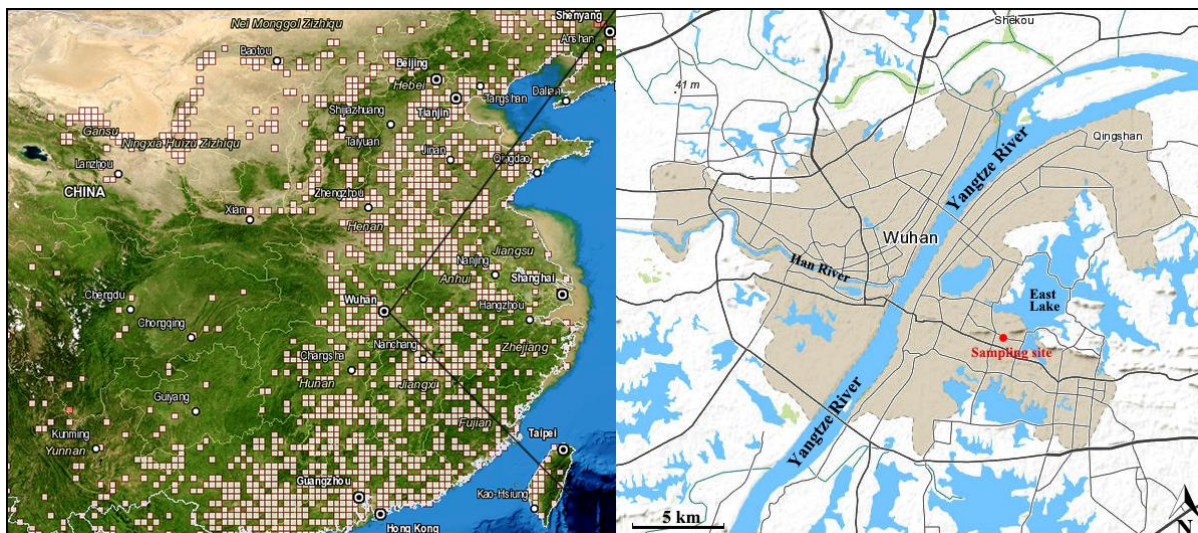
114 The whole set of air pollutants were continuously monitored at an urban site in the largest  
115 megacity of central China, i.e., Wuhan. The measurement covered two periods: May and June in  
116 summer and October and November in autumn of 2014. The measured species included particle-  
117 phase pollutants such as  $\text{PM}_{10}$ ,  $\text{PM}_{2.5}$ , and particle-bound components and gas-phase pollutants,  
118 including VOCs,  $\text{SO}_2$ , CO, NO,  $\text{NO}_2$ ,  $\text{O}_3$ ,  $\text{HNO}_3$  (g),  $\text{NH}_3$  (g), and HCl (g). Hourly data were  
119 obtained for each species. The sampling site ( $30.54^\circ \text{N}$ ,  $114.37^\circ \text{E}$ ) was located in the Hubei  
120 Environmental Monitoring Center Station, as shown in [Figure 1](#), located in a mixed commercial  
121 and residential area in which industries are seldom permitted. The instruments were housed in a  
122 room in a six-story building ( $\sim 18$  m above ground level) adjacent to a main road at a straight-line  
123 distance of  $\sim 15$  m. The traffic volume of the road was around 200 vehicles per hour. However, a  
124 wall ( $\sim 2$  m high) and several rows of trees (7 to 8 m high) were located between the road and the



125 sampling site.

126 PM<sub>10</sub> and PM<sub>2.5</sub> were measured with a continuous ambient particulate monitor (Thermo Fisher-  
127 1405D, USA) integrated with a filter dynamics measurement system to minimize the loss of  
128 semivolatile particulate matter. The water-soluble ions (WSIs) in PM<sub>2.5</sub> and gases including  
129 HNO<sub>3</sub>, HCl, and NH<sub>3</sub> were detected with an online ion chromatography monitor (Metrohm-  
130 MARGA 1S, Switzerland). **However, data were not available in May and June, because the**  
131 **instrument was initially deployed in September. An** aerosol OC/EC online analyzer (Sunset-RT-4,  
132 USA); the NIOSH thermal-optical transmission method was used to resolve the carbonaceous  
133 aerosols (OC and EC). In addition, the elements in PM<sub>2.5</sub> were measured with a customized  
134 metal analyzer. This instrument used a PM<sub>2.5</sub> impactor to collect the airborne particulate samples,  
135 which were analyzed by the  $\beta$ -ray in terms of mass concentrations. The filters loaded with  
136 particles were then sent to an x-ray fluorescence analysis system for quantitative analysis. **K<sup>+</sup>**  
137 **monitored by the online ion chromatography correlated well ( $R^2 = 0.88$ ; slope = 0.80) with K**  
138 **monitored by the customized metal analyzer. To keep consistency with other elements, K rather**  
139 **than K<sup>+</sup> was used to do the following analyses in this study.** For the analysis of trace gases (SO<sub>2</sub>,  
140 CO, NO, NO<sub>2</sub>, and O<sub>3</sub>), we used a suite of commercial analyzers developed by Thermo  
141 Environmental Instruments Inc., which have been described in detail (Lyu et al. 2016; Geng et al.,  
142 2009). Furthermore, a gas chromatography–flame ionization detector–mass spectrometry system  
143 (TH\_PKU-300) was used to resolve the real time data of the ambient VOCs. The details of the  
144 analysis techniques, resolution, detection limits, and the protocol of quality assurance/control  
145 were provided by Lyu et al. (2016) and Wang et al. (2014).

146



147  
 148 Figure 1. Geographic location of the sampling site. White blocks in left panel represent total  
 149 distribution of wildfires in autumn 2014, and urban area in Wuhan is highlighted in gray in the  
 150 right panel.

151  
 152 **2.2 Theoretical calculation and model simulation**

153 Theoretical calculation and model simulation were applied in this study to examine the formation  
 154 mechanisms of  $\text{NO}_3^-$  and SOC. The particle-bound  $\text{NO}_3^-$  was generally combined with  $\text{NH}_3$  or  
 155 presented as  $\text{HNO}_3$  in the ammonia-deficient environment, following the processes described in  
 156 R1 through R3 after  $\text{HNO}_3$  was formed by the oxidation of  $\text{NO}_x$  (Pathak et al., 2011, Lin et al.,  
 157 2010). The production of  $\text{NO}_3^-$  can be calculated with Equations 1-4.

158  $\text{NH}_3(\text{g}) + \text{HNO}_3(\text{g}) \leftrightarrow \text{NH}_4\text{NO}_3(\text{s}) \quad k_1 = \exp [118.87 - 24084/T - 6.025 \ln(T)] \text{ (ppb}^2\text{)} \quad (\text{R1})$

159  $\text{NH}_3(\text{g}) + \text{HNO}_3(\text{g}) \leftrightarrow \text{NH}_4^+ + \text{NO}_3^- \quad k_2 = (P_1 - P_2(1 - a_w) + P_3(1 - a_w)^2) \times (1 - a_w)(1 - a_w)^{1.75} k_1 \text{ (ppb}^2\text{)} \quad (\text{R2})$

161  $\text{N}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow 2\text{HNO}_3 \quad k_3 = \gamma/4(8kT/\pi m_{\text{N}_2\text{O}_5})0.5 A_p \text{ (s}^{-1}\text{)} \quad (\text{R3})$

162  $\ln(P_1) = -135.94 + 8763/T + 19.12 \ln(T) \quad (\text{Eq.1})$

163  $\ln(P_2) = -122.65 + 9969/T + 16.22 \ln(T) \quad (\text{Eq.2})$

164  $\ln(P_3) = -182.61 + 13875/T + 24.46 \ln(T) \quad (\text{Eq.3})$

165  $[\text{NO}_3^-] = 0.775 \left( \frac{[\text{NH}_3] + [\text{HNO}_3] - \sqrt{([\text{NH}_3] + [\text{HNO}_3])^2 - 4([\text{NH}_3][\text{HNO}_3] - k_1(k_2))}}{2} \right) \quad (\text{Eq.4})$

166 where R1 and R2 describe the homogeneous formation of  $\text{NO}_3^-$  in humidity conditions lower  
 167 and higher than the deliquescence relative humidity of  $\text{NH}_4\text{NO}_3$  (i.e., 62%; Tang and Munkelwitz,  
 168 1993), respectively. R3 presents the heterogeneous reaction of  $\text{N}_2\text{O}_5$  on the preexisting aerosol

169 surfaces.  $k_{I-3}$  represents the rate of reactions R1-3.  $T$ ,  $a_w$ , and  $P$  are the temperature, the relative  
170 humidity, and the temperature-related coefficient, respectively. In R3,  $\gamma$  is the reaction  
171 probability of  $N_2O_5$  on aerosol surfaces, assigned as 0.05 and 0.035 on the surface of sulfate  
172 ammonia and element carbon, respectively (Aumont et al., 1999; Hu and Abbatt, 1997).  $k$  is the  
173 Boltzmann constant ( $1.38 \times 10^{-23}$ ),  $m_{N_2O_5}$  is the molecular mass of  $N_2O_5$  ( $1.79 \times 10^{-22}$  g), and  $A_p$   
174 is the aerosol specific surface area ( $cm^2/cm^3$ ).

175 Furthermore, the PBM-MCM model was used to simulate the oxidation products in this study,  
176 i.e.,  $O_3$ ,  $N_2O_5$ , the semi-volatile oxidation products of VOCs (SVOCs), and radicals such as OH,  
177  $HO_2$ , and  $RO_2$ . With full consideration of photochemical mechanisms and real meteorological  
178 conditions, the model has been successfully applied in the study of photochemistry. Details about  
179 the model construction and application were published by Lyu et al. (2015b), Ling et al. (2014),  
180 and Lam et al. (2013).

### 181 **2.3 Source apportionment model**

182 The positive matrix factorization (PMF) model (EPA PMF v5.0) was used to resolve the sources  
183 of  $PM_{2.5}$ . As a receptor model, PMF has been extensively used in the source apportionment of  
184 airborne particles and VOCs (Brown et al., 2007; Lee et al., 1999). Detailed introductions of the  
185 model can be found in Paatero (1997) and Paatero and Tapper (1994). Briefly, it decomposes the  
186 input matrix (X) into matrices of factor contribution (G) and factor profile (F) in  $p$  sources, as  
187 shown in Equation 5. The hourly concentrations of  $PM_{2.5}$  components were included in the input  
188 matrix. Values below the detection limit (DL; see Table S1 in the Supplement) were replaced  
189 with  $DL/2$ . The uncertainties were  $\sqrt{(10\% \times \text{concentration})^2 + DL^2}$  and  $5/6 \times DL$  for the  
190 samples with concentrations higher and lower than DL, respectively. The signal-to-noise (S/N)  
191 ratios were all greater than 1, indicating “good” signal for all the species involved in source  
192 apportionment, according to the PMF 5.0 User Guide. Samples with any missing values were  
193 excluded. In total, 807 and 806 samples were applied for source apportionment in summer and  
194 autumn, respectively.

195 The selection of the factor number and the best solution depended upon the following criteria. (1)  
196 A lower Q value (Equation 6; a function to evaluate the model runs) was preferable. (2) The ratio  
197 between  $Q_{\text{robust}}$  and  $Q_{\text{true}}$  was lower than 1.5. In this study, the ratios were 0.8 and 0.9 for the  
198 summer and autumn data simulation, respectively. (3) Good agreement was shown between the  
199 predicted and observed  $PM_{2.5}$ . The slope and correlation coefficient ( $R^2$ ) for the linear regression

200 were 0.91 and 0.86 in summer and 0.95 and 0.98 in autumn, respectively, as shown in [Figure S2](#)  
 201 in the Supplement. The lower  $R^2$  value seen during the summer might be due to the lack of WSI  
 202 data. (4) The residuals were normally distributed between  $-3$  and  $3$ . [Table S2](#) summarizes the  
 203 percentage of samples with residuals between  $-3$  and  $3$  for each species; the lowest percentages  
 204 were 92.9% and 96.0% for Ni in summer and autumn, respectively. The scaled residuals for  
 205  $PM_{2.5}$  are shown in [Figure S3](#) in the Supplement. The percentage of residuals between  $-3$  and  $3$   
 206 was comparable between summer (97.5%) and autumn (98.1%). **Data points not confirming to**  
 207 **the residual range (-3 to +3) were removed from the analysis.** Finally (5), no correlation was  
 208 found between the factors, which was achieved by examining the G-space plots and controlled  
 209 by the FPEAK model runs. [Figures S4 and S5](#) in the Supplement present the G-space plots in  
 210 summer and autumn, respectively. The low factor contributions and poor correlations indicated  
 211 that rotational ambiguity was effectively controlled.

212 A bootstrap method was used to estimate the model errors, according to which 95% confidence  
 213 intervals (CIs) were calculated. The 95% CI for  $PM_{2.5}$  was  $0.6 \mu g/m^3$  (0.7% of predicted  $PM_{2.5}$ )  
 214 in summer and  $2.6 \mu g/m^3$  (3.2% of predicted  $PM_{2.5}$ ) in autumn.

$$215 \quad x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad (\text{Eq.5})$$

$$216 \quad Q = \sum_{i=1}^n \sum_{j=1}^m \left[ \frac{x_{ij} - \sum_{k=1}^p g_{ik} f_{kj}}{u_{ij}} \right]^2 \quad (\text{Eq.6})$$

217 where  $x_{ij}$  and  $u_{ij}$  are the concentration and uncertainty of  $j$  species (total of  $m$ ) in  $i$  sample (total  
 218 of  $n$ ),  $g_{ik}$  represents the contribution of  $k_{th}$  source to  $i$  sample,  $f_{kj}$  indicates the fraction of  $j$   
 219 species in  $k_{th}$  source, and  $e_{ij}$  is the residual for  $j$  species in the  $i$  sample.

220

### 221 **3. Results and discussion**

#### 222 **3.1 Concentrations of $PM_{10}$ and $PM_{2.5}$**

223 [Table 1](#) shows the mean concentrations of  $PM_{10}$  and  $PM_{2.5}$  in Wuhan and other Chinese cities and  
 224 regions. The mean, maximum and minimum values, and standard deviation or 95% CI were  
 225 provided if available. In general, the concentrations of airborne particles in Wuhan ( $135.1 \pm 4.4$   
 226 and  $118.9 \pm 3.7 \mu g/m^3$  for  $PM_{10}$ ;  $81.2 \pm 2.6$  and  $85.3 \pm 2.6 \mu g/m^3$  for  $PM_{2.5}$  in summer and  
 227 autumn, respectively) were lower than those in northern China (i.e., Beijing and Xi'an),  
 228 comparable to those in eastern China (i.e., Shanghai and Nanjing), and higher than those in  
 229 southern China (i.e., Guangzhou and Hong Kong) and Taiwan. Bearing in mind that the sampling

230 site, period, method, and instrument all interfere with comparisons, the ambient particulate  
 231 pollution in Wuhan was severe.

232 From summer to autumn, PM<sub>10</sub> levels declined considerably from 135.1 ± 4.4 to 118.9 ± 3.7  
 233 µg/m<sup>3</sup>, whereas PM<sub>2.5</sub> remained statistically stable (*p* > 0.05). The higher summer PM<sub>10</sub>  
 234 concentration was probably related to a higher load of fugitive dust. In Wuhan, the temperature  
 235 (25.6 °C ± 0.2 °C) in summer was considerably higher than that (17.5 °C ± 0.3 °C) in autumn (*p* <  
 236 0.05), which led to lower water content in the soil and a higher tendency of dust suspension. In  
 237 addition, the average wind speed in summer (1.2 ± 0.04 vs. 0.8 ± 0.03 m/s in autumn) was also  
 238 higher (*p* < 0.05), which could also have favored the generation of fugitive dust.

239  
 240 Table 1. Comparisons of PM<sub>10</sub> and PM<sub>2.5</sub> (in µg/m<sup>3</sup>) between Wuhan and other Chinese cities and  
 241 regions.

	PM <sub>10</sub>	PM <sub>2.5</sub>	Sampling period
Wuhan	135.1 ± 4.4 <sup>I</sup>	81.2 ± 2.6 <sup>I</sup>	May-Jun. 2014 (this study)
	118.9 ± 3.7 <sup>I</sup>	85.3 ± 2.6 <sup>I</sup>	Oct.-Nov. 2014 (this study)
Beijing	155.9 <sup>II</sup>	73.8 <sup>II</sup>	Jun.-Aug. 2009 <sup>a</sup>
	194.4 <sup>II</sup>	103.9 <sup>II</sup>	Sept.-Nov. 2009 <sup>a</sup>
	133.7 ± 87.8 <sup>III</sup>	71.5 ± 53.6 <sup>III</sup>	2012 whole year <sup>b</sup>
Xi'an	257.8 ± 194.7 <sup>III</sup>	140.9 ± 108.9 <sup>III</sup>	2011 whole year <sup>c</sup>
Shanghai	97.4 to 149.2 <sup>IV</sup>	62.3 to 103.1 <sup>IV</sup>	Jul. 2009-Sept. 2010 <sup>d</sup>
Nanjing	119 to 171 <sup>IV</sup>	87 to 125 <sup>IV</sup>	Jun. 2012 <sup>e</sup>
Guangzhou	23.4 <sup>II</sup>	19.2 <sup>II</sup>	Jun.-Aug. 2010-2013 <sup>f</sup>
	51.0 <sup>II</sup>	41.3 <sup>II</sup>	Sept.-Nov. 2010-2013 <sup>f</sup>
Hong Kong	31.0 ± 16.7 <sup>III</sup>	17.7 ± 12.9 <sup>III</sup>	Jun.-Aug. 2014 <sup>g</sup>
	55.8 ± 23.6 <sup>III</sup>	34.0 ± 17.3 <sup>III</sup>	Sept.-Nov. 2014 <sup>g</sup>
Taiwan	39.5 ± 11.6 <sup>III</sup>	21.8 ± 7.5 <sup>III</sup>	May-Nov. 2011 <sup>h</sup>

242 <sup>I</sup> mean ± 95% confidence interval; <sup>II</sup> arithmetic mean; <sup>III</sup> mean ± standard deviation; <sup>IV</sup> range.

243 <sup>a</sup> Liu et al. (2014); <sup>b</sup> Liu et al. (2015); <sup>c</sup> Wang et al. (2015); <sup>d</sup> Wang et al. (2013); <sup>e</sup> Shen et al.  
 244 (2014); <sup>f</sup> Deng et al. (2015); <sup>g</sup> HKEPD (2014); <sup>h</sup> Gugamsetty et al. (2012).

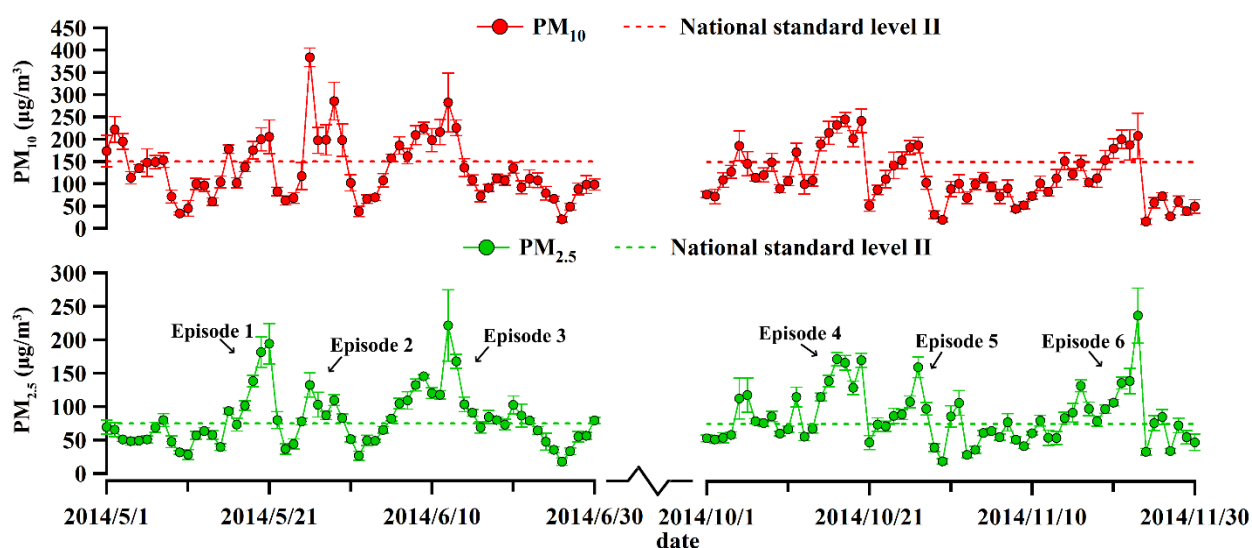
245  
 246 Figure 2 presents the daily concentrations of PM<sub>10</sub> and PM<sub>2.5</sub> during the sampling period in



247 Wuhan, with the National Standard Level II (daily averages of 150 and 75  $\mu\text{g}/\text{m}^3$  for  $\text{PM}_{10}$  and  
 248  $\text{PM}_{2.5}$ , respectively). It was found that the concentrations of  $\text{PM}_{10}$  and  $\text{PM}_{2.5}$  frequently exceeded  
 249 the standard levels, indicating the significance of ambient particulate pollution in Wuhan.  
 250 Because smaller particles tend to pose more harm to human health and to the atmosphere due to  
 251 their larger specific surface areas (Yang et al., 2012; Goldberg et al., 2001), and because the  
 252 chemical compositions in  $\text{PM}_{10}$  were not analyzed, this study focused mainly on  $\text{PM}_{2.5}$ . During  
 253 the sampling campaign, six  $\text{PM}_{2.5}$  episodes, named episodes 1 through 6, with daily averages of  
 254  $\text{PM}_{2.5}$  in excess of 75  $\mu\text{g}/\text{m}^3$ , were captured (Figure 2). It should be noted that to ensure the data  
 255 size of each episode, only the cases in which the daily  $\text{PM}_{2.5}$  average was consecutively higher  
 256 than 75  $\mu\text{g}/\text{m}^3$  for 3 days or longer were treated as  $\text{PM}_{2.5}$  episodes.

257 Table 2 summarizes the concentrations of  $\text{PM}_{10}$  and  $\text{PM}_{2.5}$  and the percentage of  $\text{PM}_{2.5}$  in  $\text{PM}_{10}$ ,  
 258 referred to as  $\text{PM}_{2.5}/\text{PM}_{10}$ , during the summer and autumn episodes and non-episodes.  $\text{PM}_{10}$  and  
 259  $\text{PM}_{2.5}$  concentrations increased significantly ( $p < 0.05$ ) during the episodes in both summer and  
 260 autumn. The  $\text{PM}_{2.5}/\text{PM}_{10}$  value also increased remarkably on episode days compared to that on  
 261 non-episode days, except for episode 2 ( $45.9\% \pm 2.5\%$ ), which suggests that more secondary  
 262 species and/or primary fine particles (e.g., primary OC and EC generated from combustion) were  
 263 generated or released during the episodes. In contrast, the lower  $\text{PM}_{2.5}/\text{PM}_{10}$  value during  
 264 episode 2 might imply a strong source of coarse particles. Indeed, this inference was confirmed  
 265 by the source apportionment analysis in section 3.3.3.

266



267

268 Figure 2. Daily concentrations of  $\text{PM}_{10}$  and  $\text{PM}_{2.5}$  in May, June, October, and November 2014.

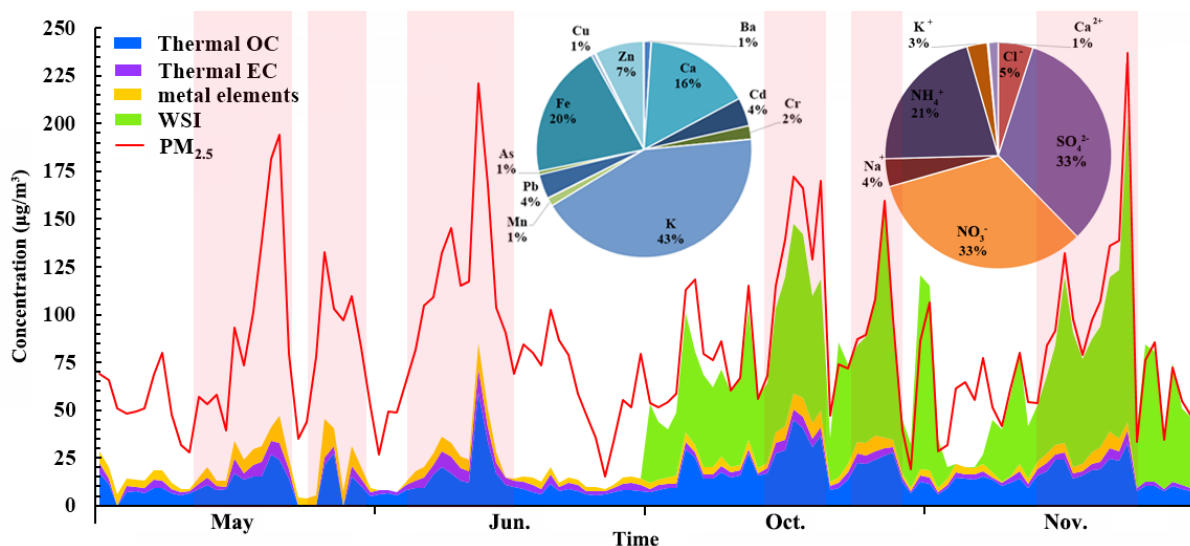
269 Episode 1, May 16 to 22; episode 2, May 25 to 30; episode 3, June 5 to 15; episode 4, October  
 270 15 to 20; episode 5, October 24 to 28; Episode 6, November 14 to 23.

271  
 272 Table 2. Mean PM<sub>10</sub>, PM<sub>2.5</sub>, and PM<sub>2.5</sub>/PM<sub>10</sub> with 95% CI during PM<sub>2.5</sub> episodes and non-  
 273 episodes in Wuhan. Non-episode 1 and Non-episode 2 represent the non-episode periods in  
 274 summer and autumn, respectively.

	PM <sub>10</sub> (μg/m <sup>3</sup> )	PM <sub>2.5</sub> (μg/m <sup>3</sup> )	PM <sub>2.5</sub> /PM <sub>10</sub> (%)
Episode 1	154.3 ± 10.1	123.0 ± 9.1	72.8 ± 2.6
Episode 2	230.1 ± 19.1	98.9 ± 5.7	45.9 ± 2.5
Episode 3	191.4 ± 9.8	126.7 ± 7.0	66.9 ± 1.8
<b>Non-episode 1</b>	<b>98.5 ± 3.9</b>	<b>56.6 ± 1.7</b>	<b>58.9 ± 1.5</b>
Episode 4	221.8 ± 8.9	148.6 ± 5.2	67.9 ± 2.0
Episode 5	154.2 ± 10.4	108.2 ± 6.8	69.3 ± 3.1
Episode 6	157.3 ± 9.0	120.0 ± 7.6	71.2 ± 2.1
<b>Non-episode 2</b>	<b>88.7 ± 3.4</b>	<b>64.2 ± 2.2</b>	<b>65.3 ± 1.3</b>

275  
 276 **3.2 Chemical composition of PM<sub>2.5</sub>**  
 277 [Figure 3](#) shows the daily variations of PM<sub>2.5</sub> and its composition. As the instrument for the  
 278 analysis of WSIs was initially deployed in September 2014, data are not available for May and  
 279 June. The carbonaceous aerosol (18.5 ± 1.2 μg/m<sup>3</sup>) and elements (6.0 ± 0.3 μg/m<sup>3</sup>) accounted for  
 280 19.1% ± 0.6% and 6.2% ± 0.2% of PM<sub>2.5</sub> in summer, respectively. In autumn, WSIs were the  
 281 most abundant component in PM<sub>2.5</sub> (64.4 ± 2.5 μg/m<sup>3</sup>; 68.6% ± 1.9%), followed by carbonaceous  
 282 aerosol (24.3 ± 1.0 μg/m<sup>3</sup>; 25.5% ± 0.8%) and elements (4.5 ± 0.2 μg/m<sup>3</sup>; 4.6% ± 0.1%). The  
 283 secondary inorganic ions SO<sub>4</sub><sup>2-</sup> (18.8 ± 0.6 μg/m<sup>3</sup>), NO<sub>3</sub><sup>-</sup> (18.7 ± 0.8 μg/m<sup>3</sup>), and NH<sub>4</sub><sup>+</sup> (12.0 ±  
 284 0.4 μg/m<sup>3</sup>) dominated the WSIs, with the average contribution of 34.0% ± 0.6%, 30.1% ± 0.5%,  
 285 and 20.4% ± 0.1%, respectively.

286

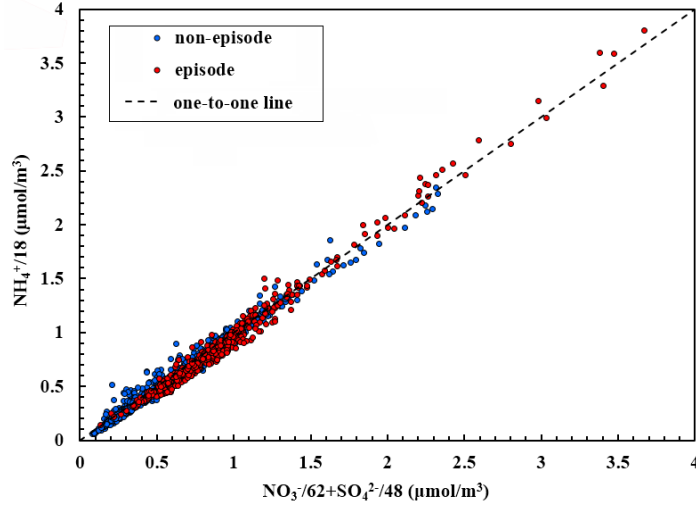


287  
 288 Figure 3. Daily variations of PM<sub>2.5</sub> and its components. Pie charts represent the composition of  
 289 elements and water-soluble ions, respectively. Pink shaded areas represent episodes.

290  
 291 The charge balance between the anions and cations was usually used to predict the existing forms  
 292 of SIAs and the acidity of PM<sub>2.5</sub>. Figure 4 shows the relative abundance of molar charges of SIAs,  
 293 which were located fairly close to the one-to-one line on both episode and non-episode days.  
 294 This finding suggests that NH<sub>4</sub>NO<sub>3</sub> and (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> were coexisting forms of the SIAs in PM<sub>2.5</sub> in  
 295 Wuhan. When extending NH<sub>4</sub><sup>+</sup> to total cations (NH<sub>4</sub><sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Na<sup>+</sup>, and K<sup>+</sup>) and NO<sub>3</sub><sup>-</sup> and  
 296 SO<sub>4</sub><sup>2-</sup> to total anions (NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, and Cl<sup>-</sup>), the molar charges of the cations and anions were  
 297 balanced (slope, 0.98; R<sup>2</sup> = 0.98), as shown in Figure S6 in the Supplement, indicating that PM<sub>2.5</sub>  
 298 was neutralized during autumn in Wuhan.

299





300  
 301 Figure 4. Relative abundance of molar charges of PM<sub>2.5</sub> during autumn in Wuhan.

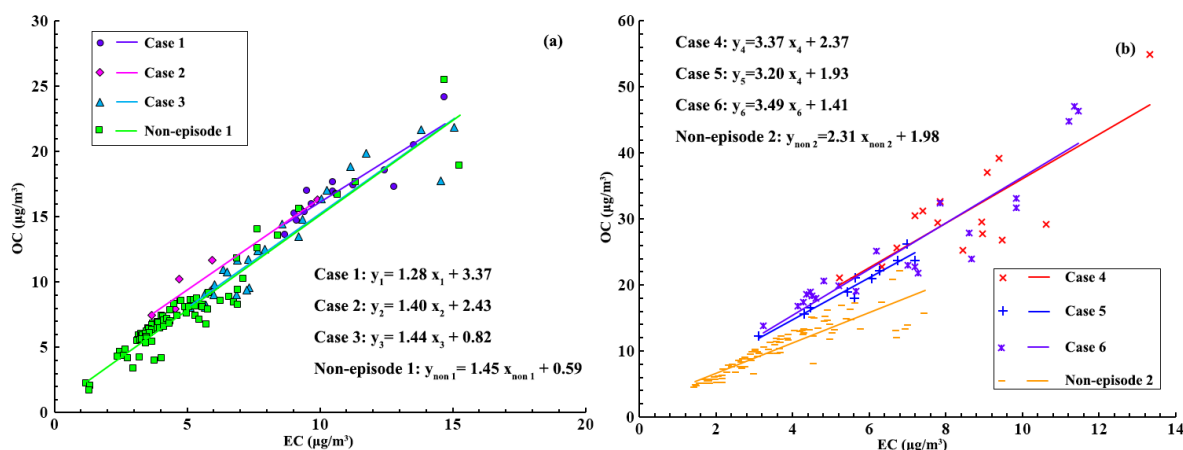
302  
 303 For the carbonaceous aerosol, OC ( $14.8 \pm 0.5 \mu\text{g}/\text{m}^3$ ) and EC ( $3.6 \pm 0.1 \mu\text{g}/\text{m}^3$ ) accounted for  
 304  $79.9\% \pm 0.3\%$  and  $20.2\% \pm 0.3\%$  of the total carbon, respectively. In general, SOC was expected  
 305 to exist when the OC/EC ratio was greater than 2 (Duan et al., 2005; Chow et al., 1996), and the  
 306 proportion of SOC increased with the increase in OC/EC ratio. The average OC/EC ratio was  $4.8$   
 307  $\pm 0.1$  in Wuhan, which suggests that SOC (i.e., carbon fraction of SOA) was an important  
 308 component in PM<sub>2.5</sub>. Indeed, as the constituents of OC, SOC and POC can be distinguished with  
 309 the EC-tracer method, following Equations 7 and 8 (Cabada et al., 2004):

310 
$$\text{POC} = (\text{OC}/\text{EC})_{\text{prim}} \times \text{EC} + \text{OC}_{\text{non-comb}} \quad (\text{Eq.7})$$

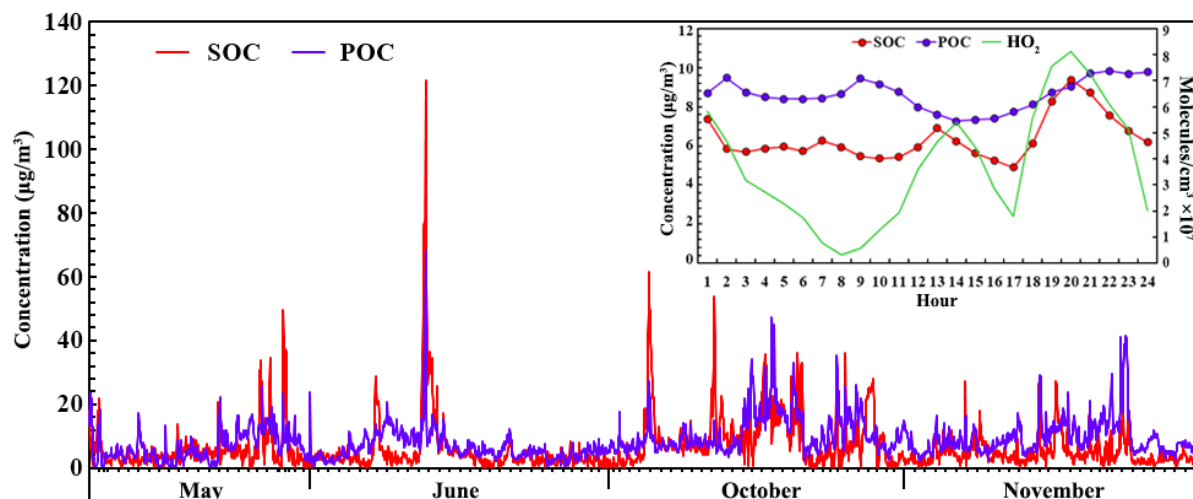
311 
$$\text{SOC} = \text{OC} - \text{POC} \quad (\text{Eq.8})$$

312 where  $(\text{OC}/\text{EC})_{\text{prim}}$  was the ratio of primary OC to EC, obtained from the pairs of OC and EC  
 313 with the OC/EC ratios among the 10% lowest; and  $\text{OC}_{\text{non-comb}}$  was the primary OC that was not  
 314 related to combustion activities. These values were determined by the slope and intercept of the  
 315 linear regression between primary OC and EC, respectively (Figure 5). Because the abundance  
 316 of SOC depended largely upon the oxidative capacity of the atmosphere, the oxidative radical  
 317 ( $\text{HO}_2$ ) was simulated with the PBM-MCM model and compared with the pattern of SOC. More  
 318 details about the simulation are provided in section 3.4. Figure 6 shows the hourly concentrations  
 319 of SOC and POC and the average diurnal patterns of SOC, POC, and  $\text{HO}_2$ . In general, the POC  
 320 levels ( $8.6 \pm 0.2 \mu\text{g}/\text{m}^3$ ) were slightly higher than those of SOC ( $6.4 \pm 0.3 \mu\text{g}/\text{m}^3$ ;  $p < 0.05$ ). The  
 321 difference was greatest in November, when the concentration was  $9.5 \pm 0.4$  and  $4.7 \pm 0.3 \mu\text{g}/\text{m}^3$   
 322 for POC and SOC, respectively. Because the production of SOC was closely related to the

323 atmospheric oxidative capacity, the lowest fraction of SOC in November might be attributable to  
 324 the weakest oxidative capacity; for example, the O<sub>3</sub> level was lowest in November (14.3 ± 1.0  
 325 ppbv). Two peaks were found for the simulated diurnal pattern of HO<sub>2</sub>, which might be caused  
 326 by strong solar radiation at noon and in the early afternoon and by reactions among alkenes and  
 327 O<sub>3</sub> and NO<sub>3</sub> at night (Emmerson et al., 2005; Kanaya et al., 1999). The diurnal patterns of POC  
 328 and SOC revealed that POC levels were relatively stable throughout the day. The increase in the  
 329 POC level in the early morning (06:00 to 08:00) and late afternoon and early evening (16:00 to  
 330 20:00) was likely related to increases in vehicular emissions during rush hours, and the decrease  
 331 from 08:00 to 15:00 might be caused by the extension of the boundary layer. In contrast, the  
 332 SOC level showed two peaks at around 12:00 and 19:00, which was consistent with the diurnal  
 333 variation of the simulated HO<sub>2</sub>, suggesting that the formation of SOC was closely related to the  
 334 oxidative radicals in the atmosphere. (A detailed relationship is discussed in section 3.4.3.)  
 335



336  
 337 Figure 5. Regression between OC and EC with the 10% lowest OC/EC ratios during (a) summer  
 338 and (b) autumn in Wuhan.



339  
 340 Figure 6. Hourly concentrations of SOC and POC. Insert graph presents average diurnal  
 341 variations of SOC, POC, and HO<sub>2</sub>.

342  
 343 Among the elements, potassium (K;  $2060.7 \pm 82.3 \text{ ng/m}^3$ ), iron (Fe;  $996.5 \pm 34.3 \text{ ng/m}^3$ ), and  
 344 calcium (Ca;  $774.1 \pm 39.4 \text{ ng/m}^3$ ) were the most abundant species, accounting for  $47.0\% \pm 2.2\%$ ,  
 345  $21.4\% \pm 0.3\%$ , and  $15.6\% \pm 0.3\%$  of the total analyzed elements, respectively. Correlation  
 346 analysis indicated that Fe had good correlation with Ca ( $R^2 = 0.66$ ; Figure S7 in the Supplement),  
 347 whereas weak correlations of K with Fe ( $R^2 = 0.14$ ) and Ca ( $R^2 = 0.09$ ) were found, suggesting  
 348 that Fe and Ca shared common sources that were different from the sources of K. Because Fe  
 349 and Ca are typical crustal elements, fugitive dust (e.g., dust from traffic, construction and  
 350 demolition works, yards, and bare soil) was their most likely source. In contrast, apart from  
 351 emissions from mineral sources, K is also emitted from biomass burning. As such, K was  
 352 believed to be mainly emitted from biomass burning in this study, which is further supported by  
 353 the moderate correlations of K with OC ( $R^2 = 0.52$ ) and EC ( $R^2 = 0.48$ ) because biomass burning  
 354 also emits OC and EC (Saarikoski et al., 2007; Echalar et al., 1995).

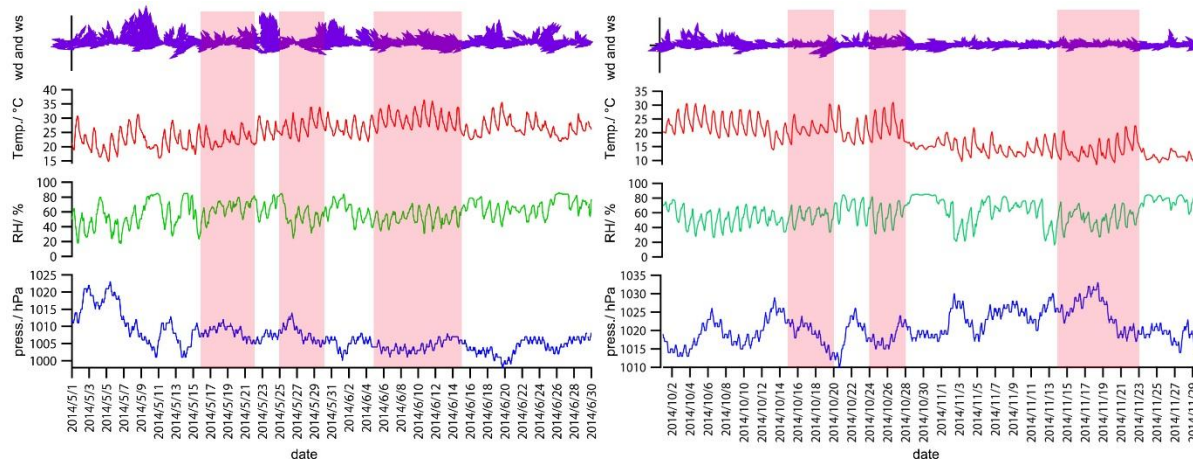
### 355 3.3 Causes of PM<sub>2.5</sub> episodes

#### 356 3.3.1 Meteorological conditions

357 The processes of particle formation, dispersion, and deposition are closely related to  
 358 meteorological conditions. To interpret the possible causes of the PM<sub>2.5</sub> episodes, Figure 8 shows  
 359 the patterns of wind direction and speed, temperature, relative humidity, and atmospheric  
 360 pressure in Wuhan during the monitoring period. In general, southeast winds prevailed at the  
 361 sampling site with a wind speed of approximately 1.0 m/s. The low wind speed indicates the

362 dominance of local air masses. However, due to the high stability and long lifetime of PM<sub>2.5</sub>, the  
 363 regional and superregional impact could not be eliminated. In comparison with those in summer,  
 364 the wind speed (summer, 1.1 ± 0.04 m/s; autumn, 0.8 ± 0.03 m/s) and temperature (summer, 25.6  
 365 ± 0.2 m/s; autumn, 17.5 ± 0.3 m/s) were significantly (*p* < 0.05) lower in autumn, whereas the  
 366 atmospheric pressure (summer, 1006.9 ± 0.2 hPa; autumn, 1020.9 ± 0.2 hPa) was much higher.  
 367 During the episodes, the wind speed was generally lower than during non-episodes, with the  
 368 exception of episode 5. This might be one cause for the episodes, but it does not fully explain the  
 369 great enhancements of PM<sub>2.5</sub>, because the wind speeds were very low and the differences  
 370 between the episodes and non-episodes were minor. The atmospheric pressure was not very high  
 371 during episodes 1 through 5, suggesting that the synoptic system was not responsible for the  
 372 occurrence of these PM<sub>2.5</sub> episodes. However, the atmospheric pressure was remarkably higher  
 373 (*p* < 0.05) in episode 6 (1024 ± 1 hPa) than in non-episode 2 (1021 ± 0.3 hPa), which might have  
 374 suppressed the diffusion of PM<sub>2.5</sub> and the gaseous precursors. In addition, the temperature was  
 375 lower (episode 6, 14.0 °C ± 0.4 °C; non-episode 2, 17.2 °C ± 0.3 °C; *p* < 0.05), which favors the  
 376 gas-to-particle partitioning of semivolatile and non-thermal stabilized species. As a consequence,  
 377 these effects might have elevated the PM<sub>2.5</sub> concentrations in episode 6, which is discussed  
 378 further in section 3.4.2.

379



380

381 Figure 8. Meteorological patterns in Wuhan during the monitoring period. Pink shaded areas  
 382 represent PM<sub>2.5</sub> episodes.

383

### 384 3.3.2 Chemical signatures

385 Table 3 summarizes the mass concentrations and percentages of the main components in PM<sub>2.5</sub>.  
386 The mass concentrations of PM<sub>2.5</sub> components significantly increased from non-episode days to  
387 episode days ( $p < 0.05$ ). In contrast, the percentages of the chemical components in PM<sub>2.5</sub> varied  
388 by species. In summer, the fractions of EC and K in PM<sub>2.5</sub> experienced significant increases from  
389 non-episode 1 (EC, 4.8%  $\pm$  0.2%; K, 2.0%  $\pm$  0.1%) to episode 1 (EC, 5.7%  $\pm$  0.5%; K, 4.4%  $\pm$   
390 0.3%) and episode 3 (EC, 5.3%  $\pm$  0.2%; K, 3.0%  $\pm$  0.2%). Because EC is the tracer of  
391 incomplete combustion (Chow et al., 1996) and K is the indicator of biomass burning (Saarikoski  
392 et al., 2007; Echalar et al., 1995), the higher percentages of EC and K in episodes 1 and 3 imply  
393 the outstanding contribution of biomass burning. In contrast, the fraction of OC in PM<sub>2.5</sub>  
394 remained stable on both episode and non-episode days ( $p > 0.05$ ), possibly because the high  
395 temperatures in summer hindered the gas-to-particle partitioning of semivolatile organics  
396 (Takekawa et al., 2003). Furthermore, the percentages of Ca (2.9%  $\pm$  0.4%) and Fe (2.7%  $\pm$  0.3%)  
397 significantly increased during episode 2 ( $p < 0.05$ ) compared to those in non-episode 1 (Ca, 1.1%  
398  $\pm$  0.1%; Fe, 1.5%  $\pm$  0.1%), which shows that fugitive dust made a considerable contribution to  
399 PM<sub>2.5</sub> in episode 2. In addition, biomass burning might also have contributed to PM<sub>2.5</sub>, in view of  
400 the increase in the percentage of K (non-episode 1, 2.0%  $\pm$  0.1%; episode 2, 3.2%  $\pm$  0.2%).  
401 In autumn, the percentage of K significantly ( $p < 0.05$ ) increased during episode 4 (3.1%  $\pm$  0.1%  
402 vs. 2.1%  $\pm$  0.1% in non-episode 2), as did that of OC (27.3%  $\pm$  0.7% vs. 20.9%  $\pm$  0.8% in non-  
403 episode 2), suggesting the dominant role of biomass burning in episode 4. Furthermore, the  
404 fractions of OC in episode 5 (23.8%  $\pm$  1.5%) and NO<sub>3</sub><sup>-</sup> in episode 6 (26.1%  $\pm$  1.0%) were  
405 obviously higher than those in non-episode 2 (OC, 20.9%  $\pm$  0.8%; NO<sub>3</sub><sup>-</sup>, 19.8%  $\pm$  0.9%). Due to  
406 the complexity of the sources of OC and NO<sub>3</sub><sup>-</sup>, the causes of episodes 5 and 6 are further  
407 explored in the following sections.  
408 In summary, episodes 1, 3, and 4 were greatly affected by biomass burning. This finding was  
409 further confirmed by the significant increases in the gaseous tracers of biomass burning such as  
410 ethyne (C<sub>2</sub>H<sub>2</sub>) and methyl chloride (CH<sub>3</sub>Cl) (Guo et al., 2011b; Simoneit et al., 2002) during  
411 these episodes ( $p < 0.05$ ; see Figure S8 in the Supplement).

412 Table 3. Concentrations ( $\mu\text{g}/\text{m}^3$ ) and percentages (in parentheses) of the main components of  $\text{PM}_{2.5}$  during non-episodes and episodes.  
 413 Bold font demonstrates significant increase in percentage of  $\text{PM}_{2.5}$  components during episodes compared to non-episodes.

	Summer				Autumn			
	Episode 1	Episode 2	Episode 3	Non-episode 1	Episode 4	Episode 5	Episode 6	Non-episode 2
OC	18.5±1.3 (15.1%±0.8%)	16.3±3.3 (14.0%±1.9%)	19.8±2.5 (14.5%±1.0%)	7.9±0.3 (15.4%±0.7%)	<b>35.1±1.7</b> <b>(27.3%±0.7%)</b>	<b>24.9±1.9</b> <b>(23.8%±1.5%)</b>	22.7±1.5 (21.4%±0.8%)	14.6±1.0 (20.9%±0.8%)
EC	<b>6.8±0.5</b> <b>(5.7%±0.5%)</b>	4.3±0.6 (4.1%±0.5%)	<b>6.6±0.5</b> <b>(5.3%±0.2%)</b>	2.8±0.1 (4.8%±0.2%)	5.4±0.4 (4.2%±0.3%)	4.2±0.4 (4.1%±0.4%)	4.2±0.4 (3.9%±0.2%)	2.7±0.2 (4.0%±0.2%)
$\text{SO}_4^{2-}$	NA	NA	NA	NA	28.3±0.9 (22.8%±0.9%)	25.9±2.4 (23.7%±1.1%)	21.3±2.0 (19.2%±0.6%)	18.8±0.9 (26.9%±0.9%)
$\text{NO}_3^-$	NA	NA	NA	NA	23.7±1.9 (17.8%±0.9%)	24.7±2.9 (21.4%±1.1%)	<b>30.8±3.6</b> <b>(26.1%±1.0%)</b>	15.8±1.3 (19.8%±0.9%)
$\text{NH}_4^+$	NA	NA	NA	NA	16.5±0.8 (12.9%±0.2%)	15.6±1.8 (13.7%±0.6%)	16.8±1.8 (14.5%±0.3%)	11.0±0.7 (14.9%±0.3%)
K	<b>5.3±0.6</b> <b>(4.4%±0.3%)</b>	<b>3.4±0.4</b> <b>(3.2%±0.2%)</b>	<b>3.8±0.5</b> <b>(3.0%±0.2%)</b>	1.1±0.1 (2.0%±0.1%)	<b>4.0±0.2</b> <b>(3.1%±0.1%)</b>	2.3±0.2 (2.2%±0.1%)	2.4±0.2 (2.2%±0.1%)	1.4±0.1 (2.1%±0.1%)
Ca	1.2±0.4 (1.1%±0.1%)	<b>3.2±0.4</b> <b>(2.9%±0.4%)</b>	0.9±0.3 (0.8%±0.1%)	0.6±0.04 (1.1%±0.1%)	1.1±0.1 (0.9%±0.1%)	0.8±0.2 (0.8%±0.2%)	0.8±0.1 (0.8%±0.1%)	0.3±0.04 (0.5%±0.1%)
Fe	1.4±0.1 (1.3%±0.1%)	<b>2.8±0.3</b> <b>(2.7%±0.3%)</b>	1.1±0.1 (1.0%±0.1%)	0.8±0.05 (1.5%±0.1%)	1.5±0.1 (1.2%±0.2%)	1.2±0.2 (1.2%±0.2%)	1.2±0.1 (1.1%±0.1%)	0.6±0.04 (0.9%±0.1%)

414

### 415 3.3.3 Source apportionment

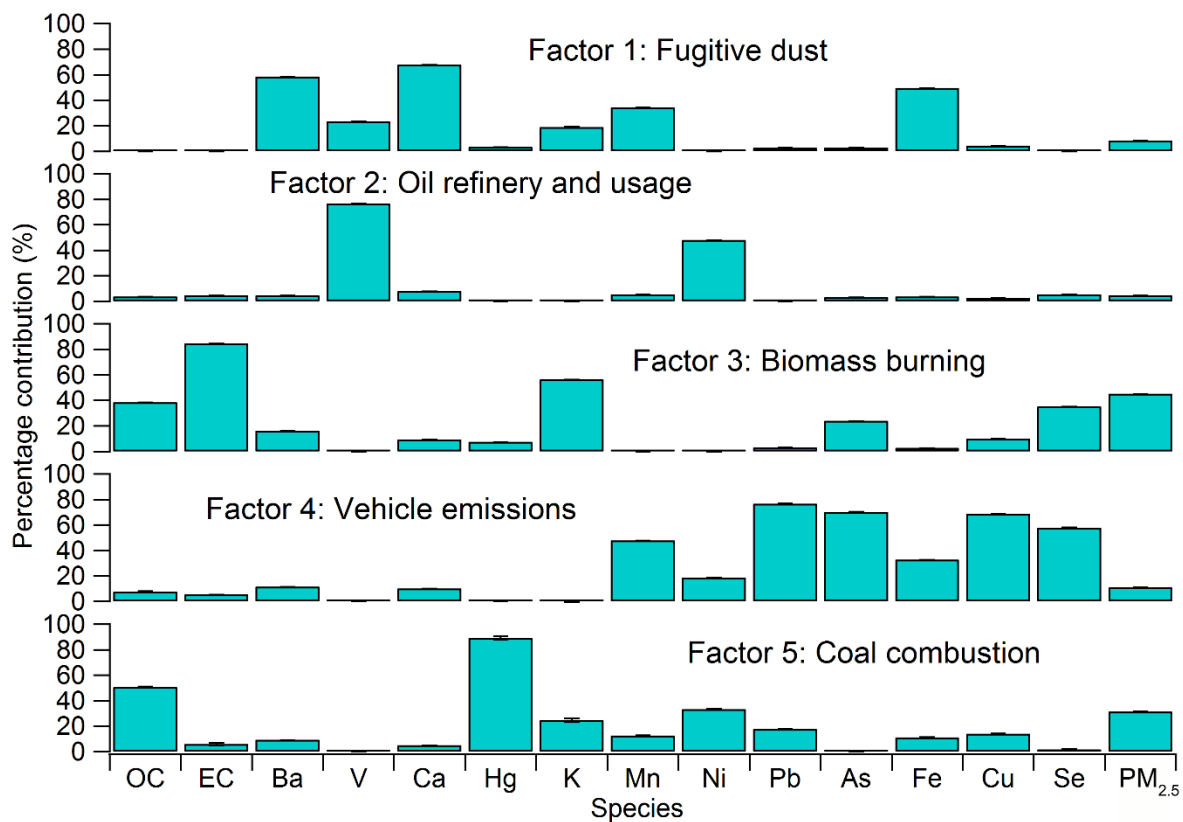
416 To clarify the sources of PM<sub>2.5</sub> and quantify their contributions, the hourly data on PM<sub>2.5</sub>  
417 components were applied to PMF for source apportionment. Five and six sources were resolved  
418 for summer and autumn, respectively, as shown in Figure 10 and Figure 11. The source of SIA  
419 was missing in summer, due to the lack of WSI data. For other sources, the profile of each  
420 corresponding source was similar in summer and autumn. Factor 1 had high loadings of crustal  
421 elements (i.e., Ba, Ca, Mn, and Fe), indicating the source of fugitive dust. Factor 2 was likely  
422 associated with oil refinery and usage, in view of the high percentages of V and Ni, which often  
423 originate from the combustion of heavy oil (Barwise et al., 1990; Nriagu and Pacyna, 1988).  
424 Factor 3 was distinguished by the high loadings of OC, EC, and K, indicating the biomass  
425 burning source (Zhang et al., 2013; Friedli et al., 2003). Factor 4 seemed to represent vehicle  
426 emissions, due to the dominance of Pb, As, Cu, and Se. **Cu has been identified as one of the most**  
427 **abundant metals in both brake linings and the brake wear particles with the concentration of up**  
428 **to 210 mg/g in brake wear dust (Grigoratos and Martini, 2015).** Although lead-containing  
429 gasoline has been forbidden in China since 2000, high levels of Pb are often reported in traffic  
430 sources (Yang et al., 2013; Song et al., 2012), which might be due to the use of Pb-containing  
431 materials in car components, such as lead wheel weights, solder in electronics, and lead-acid  
432 batteries (Song et al., 2012). **Both biomass burning and vehicular emissions are important**  
433 **sources of EC. The much lower EC apportioned to vehicular emissions in this study is explained**  
434 **in Section 1 of the Supplement.** OC and Hg dominated in factor 5, and Cl<sup>-</sup> also showed high  
435 loading in this factor in the autumn profile. It is well documented that Hg and Cl<sup>-</sup> are largely  
436 emitted from coal combustion (Wang et al., 2010; Ye et al., 2003). Hence, this factor was  
437 assigned as coal combustion. Finally, a source of SIA with high loadings of NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, and  
438 NH<sub>4</sub><sup>+</sup> was resolved in autumn.

439 Figures S9 and S10 in the Supplement show the day-to-day variations of the source contributions  
440 in summer and autumn, respectively. Overall, biomass burning was the largest contributor (45.0%  
441 ±0.03%) to PM<sub>2.5</sub> in summer. However, the contribution of biomass burning in autumn (23.7% ±  
442 0.5%) was lower than that of SIA (38.6% ±0.7%). Bearing in mind the uncertainties caused by  
443 the lack of WSIs, the greater contribution of biomass burning in summer might be associated  
444 with the lower WSIs. For example, NO<sub>3</sub><sup>-</sup> was expected to be much lower in summer due to its  
445 thermal decomposition at high temperatures. Table 4 summarizes the source contributions during

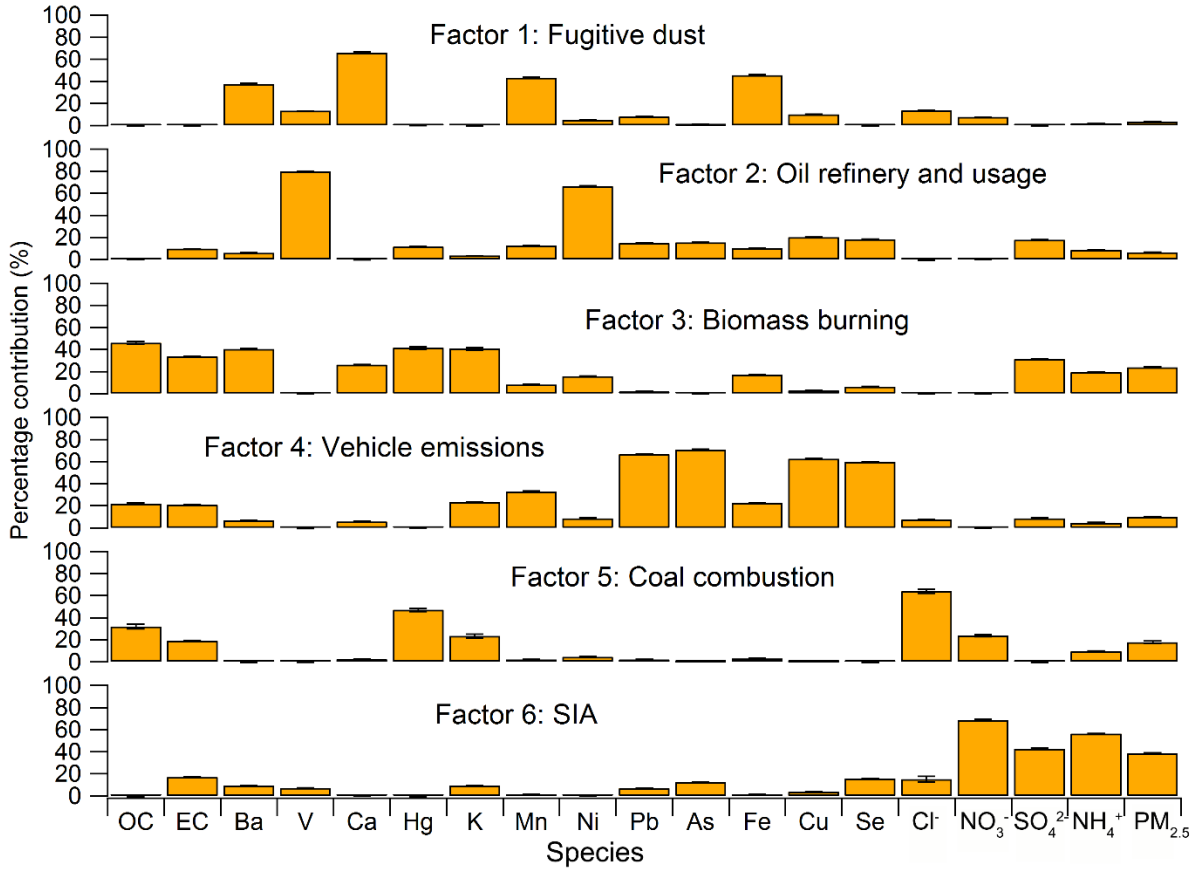


446 episodes and non-episodes. Noticeably, the contributions of biomass burning were significantly  
447 higher ( $p < 0.05$ ) in episode 1 ( $59.2 \pm 6.3 \mu\text{g}/\text{m}^3$ ;  $46.6\% \pm 3.0\%$ ), episode 3 ( $64.9 \pm 3.3 \mu\text{g}/\text{m}^3$ ;  
448  $50.8\% \pm 1.2\%$ ), and episode 4 ( $48.7 \pm 2.9 \mu\text{g}/\text{m}^3$ ;  $44.8\% \pm 2.6\%$ ) than those in the corresponding  
449 non-episodes, confirming that biomass burning was the main cause of these  $\text{PM}_{2.5}$  episodes. In  
450 addition, vehicle emissions made a greater contribution to episode 4 ( $14.9 \pm 1.2 \mu\text{g}/\text{m}^3$ ;  $13.7\% \pm$   
451  $1.1\%$ ) than to non-episode 2. In contrast, the contribution of fugitive dust ( $6.5 \pm 1.3 \mu\text{g}/\text{m}^3$ ;  $5.6\%$   
452  $\pm 1.0\%$ ) in episode 2 was remarkably ( $p < 0.05$ ) higher than in non-episode 1 ( $1.1 \pm 0.1 \mu\text{g}/\text{m}^3$ ;  
453  $1.8\% \pm 0.2\%$ ). This finding was consistent with the inference that episode 2 was attributable to  
454 fugitive dust according to the low  $\text{PM}_{2.5}/\text{PM}_{10}$  ratio ( $45.9\% \pm 2.5\%$ ) and high levels of crustal  
455 elements (Ca,  $2.9\% \pm 0.4\%$ ; Fe,  $2.7\% \pm 0.3\%$ ) in this episode. In addition, vehicle emissions and  
456 SIA both experienced significant increases in episodes 5 and 6 ( $p < 0.05$ ). In fact, the increase of  
457 OC in episode 5 (see Table 3) was mainly caused by vehicle emissions and coal combustion  
458 (Table S3 in the Supplement). For episode 6, in addition to the increase in OC, SIA was also an  
459 important contributor, particularly for  $\text{NO}_3^-$ , which increased from  $9.9 \pm 1.2 \mu\text{g}/\text{m}^3$  in non-  
460 episode 2 to  $21.4 \pm 3.3 \mu\text{g}/\text{m}^3$  in episode 6 (Table S3). **In addition, we noted that the contribution**  
461 **of coal combustion was much lower in non-episode 2 than that in non-episode1 ( $p < 0.05$ ). The**  
462 **explanation is provided in Section 2 of the Supplement.**





463  
 464 Figure 10. Profiles of PM<sub>2.5</sub> sources in summer. Error bars represent 95% CI estimated by  
 465 bootstrap method in PMF.



466

467 Figure 11. Profiles of PM<sub>2.5</sub> sources in autumn. Error bars represent 95% CI estimated by

468 bootstrap method in PMF.

469 Table 4. Mass concentration ( $\mu\text{g}/\text{m}^3$ ) and percentage contribution (in parentheses) of sources to sum of resolved species in  $\text{PM}_{2.5}$ . Bold  
 470 font represents a significant increase in percentage contribution in episodes compared to non-episodes.

	Fugitive dust	Oil refinery and usage	Biomass burning	Vehicle emissions	Coal Combustion	SIA
Episode 1	10.0 $\pm$ 1.3 (8.7% $\pm$ 1.2%)	4.0 $\pm$ 1.2 (2.8% $\pm$ 0.7%)	<b>59.2<math>\pm</math>6.3</b> <b>(46.6% <math>\pm</math>3.0%)</b>	12.2 $\pm$ 1.8 (9.6% $\pm$ 1.2%)	36.0 $\pm$ 2.7 (32.2% $\pm$ 2.5%)	—
Episode 2	<b>29.5<math>\pm</math>6.1</b> <b>(27.5% <math>\pm</math>5.5%)</b>	8.2 $\pm$ 2.6 (6.7% $\pm$ 1.7%)	37.1 $\pm$ 5.8 (35.3% $\pm$ 4.8%)	10.8 $\pm$ 3.1 (11.9% $\pm$ 3.2%)	20.9 $\pm$ 4.7 (18.7% $\pm$ 3.1%)	—
Episode 3	6.4 $\pm$ 0.5 (5.3% $\pm$ 0.4%)	4.9 $\pm$ 1.0 (3.9% $\pm$ 0.7%)	<b>64.9<math>\pm</math>3.3</b> <b>(50.8% <math>\pm</math>1.2%)</b>	8.8 $\pm$ 0.9 (7.9% $\pm$ 0.9%)	41.6 $\pm$ 3.3 (32.1% $\pm$ 1.5%)	—
Non-episode 1	4.8 $\pm$ 0.6 (8.7% $\pm$ 1.0%)	2.9 $\pm$ 0.4 (5.0% $\pm$ 0.7%)	22.2 $\pm$ 1.6 (35.2% $\pm$ 1.9%)	9.3 $\pm$ 0.6 (16.3% $\pm$ 1.0%)	19.5 $\pm$ 1.0 (35.9% $\pm$ 1.7%)	—
Episode 4	3.3 $\pm$ 0.3 (3.0% $\pm$ 0.2%)	7.2 $\pm$ 0.6 (7.0% $\pm$ 0.8%)	<b>48.7<math>\pm</math>2.9</b> <b>(44.8% <math>\pm</math>2.6%)</b>	<b>14.9<math>\pm</math>1.2</b> <b>(13.7% <math>\pm</math>1.1%)</b>	13.8 $\pm$ 1.7 (11.9% $\pm$ 1.3%)	23.3 $\pm$ 3.3 (19.6% $\pm$ 2.6%)
Episode 5	2.3 $\pm$ 0.5 (2.7% $\pm$ 0.5%)	5.3 $\pm$ 0.6 (6.6% $\pm$ 0.8%)	21.1 $\pm$ 2.8 (22.1% $\pm$ 2.3%)	<b>12.2<math>\pm</math>1.9</b> <b>(13.85<math>\pm</math>2.1%)</b>	14.8 $\pm$ 2.0 (17.5% $\pm$ 2.7%)	<b>39.9<math>\pm</math>6.4</b> <b>(37.2% <math>\pm</math>3.0%)</b>
Episode 6	2.6 $\pm$ 0.3 (3.0% $\pm$ 0.3%)	4.7 $\pm$ 0.6 (4.7% $\pm$ 0.4%)	18.4 $\pm$ 2.4 (21.2% $\pm$ 2.5%)	<b>14.2<math>\pm</math>1.3</b> <b>(16.1% <math>\pm</math>1.3%)</b>	17.7 $\pm$ 2.5 (16.3% $\pm$ 1.7%)	<b>44.6<math>\pm</math>6.8</b> <b>(38.7% <math>\pm</math>2.2%)</b>
Non-episode 2	0.9 $\pm$ 0.1 (1.6% $\pm$ 0.2%)	4.7 $\pm$ 0.3 (8.6% $\pm$ 0.6%)	22.2 $\pm$ 1.6 (39.3% $\pm$ 2.5%)	4.2 $\pm$ 0.4 (7.4% $\pm$ 0.7%)	9.8 $\pm$ 0.8 (18.1% $\pm$ 1.8%)	20.5 $\pm$ 2.5 (25.0% $\pm$ 2.3%)

471

472 **3.3.4 Open fires and air mass trajectories**

473 To further confirm the biomass burning activities during the PM<sub>2.5</sub> episodes, the wildfire  
474 distribution (downloaded from NASA Firms Web Fire Mapper, and accessible at  
475 <https://firms.modaps.eosdis.nasa.gov/firemap/>) and 72-h backward air mass trajectories  
476 (simulated by Hysplit v4.9 model) are plotted in Figure 12. Because the concentrations,  
477 compositions and source contributions of PM<sub>2.5</sub> were averaged over the entire period of each  
478 episode, the wildfire distribution and backward trajectories were also averaged for the entire  
479 period of each episode. Consistent with the great contributions to PM<sub>2.5</sub> of biomass burning, the  
480 air masses arriving in Wuhan had passed over the areas where intensive open fires were detected  
481 in episodes 1, 3, and 4. In episode 2, wildfires were widespread in northeast China. However, the  
482 air mass trajectories were mainly from the south and northwest and evaded the burning areas,  
483 which might explain why biomass burning was not a predominant factor in episode 2. The sparse  
484 wildfires in episodes 5 and 6 coincided with the source apportionment result that biomass  
485 burning did not significantly elevate the concentration of PM<sub>2.5</sub>.

486



487  
488 Figure 12. Wildfire distribution and 72-h backward air mass trajectories. Red squares  
489 demonstrate potential areas where biomass burning aggravated particulate pollution in Wuhan.

490

## 491 3.4 Formation mechanisms

### 492 3.4.1 Model validation

493 In this study, the PBM-MCM model was used to help investigate the formation mechanisms of  
494  $\text{NO}_3^-$  and SOC. Before application, the model was validated via  $\text{O}_3$  simulation. Figure S11  
495 compares the daily averages and diurnal variations of  $\text{O}_3$  between the simulation and the  
496 observations. It was found that the model simulated  $\text{O}_3$  variation well in both daily and diurnal  
497 patterns. However, it generally overestimated the  $\text{O}_3$  levels in November. The meteorological  
498 parameters indicated that the frequency of foggy days was extremely high (36.7%) in November,  
499 possibly resulting in a weakening of solar radiation and consequently of photochemical reactivity.  
500 To quantitatively evaluate the performance of the model, the index of agreement (IOA) was  
501 calculated using Equation 9.

$$502 \quad \text{IOA} = 1 - \frac{\sum_{i=1}^n (O_i - S_i)^2}{\sum_{i=1}^n (|O_i - \bar{O}| + |S_i - \bar{O}|)^2} \quad (\text{Eq.9})$$

503 where  $\bar{O}$  is the average of  $n$  samples and  $O_i$  and  $S_i$  represent the observed and simulated values,  
504 respectively. Within the interval of [0, 1], higher IOA values indicated better agreement between  
505 the simulation and observation.

506 By calculation, the IOA reached 0.86, indicating excellent performance of the model in the  $\text{O}_3$   
507 simulation. Because  $\text{O}_3$  production is closely associated with oxidative radicals, intermediates,  
508 and products, the robust  $\text{O}_3$  simulation gave us full confidence to accept the simulated  $\text{N}_2\text{O}_5$ ,  
509  $\text{HO}_2$ , and SVOCs.

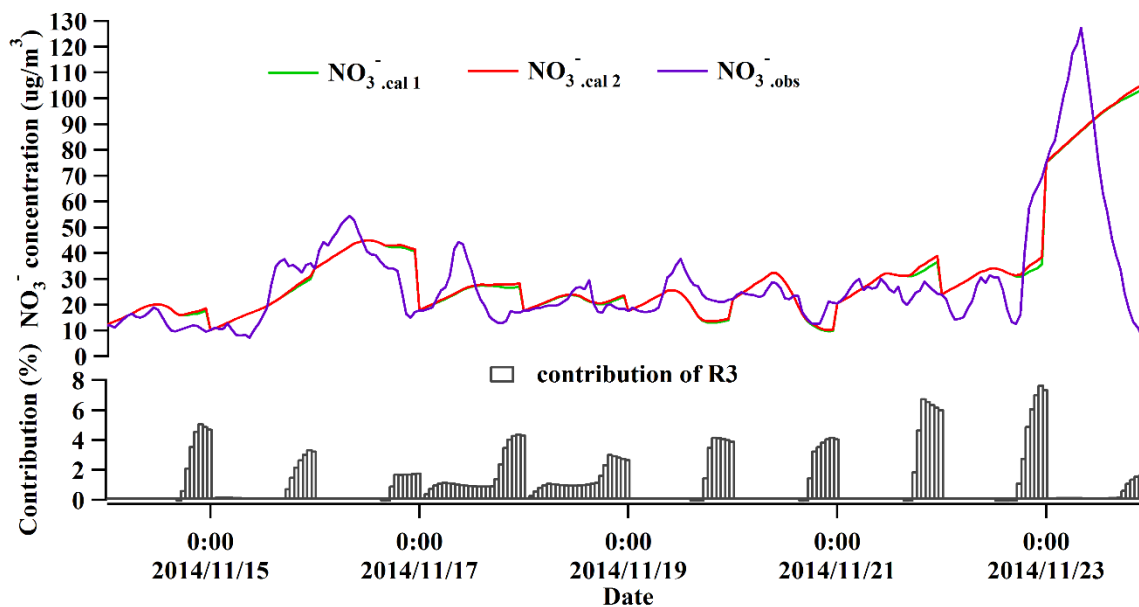
510

### 511 3.4.2 $\text{NO}_3^-$ formation

512 The composition analysis indicated that the proportion of  $\text{NO}_3^-$  increased remarkably during  
513 episode 6. To interpret this phenomenon, the formation mechanisms of  $\text{NO}_3^-$  were investigated.  
514 [Figure 14](#) shows the hourly variations of the calculated and observed  $\text{NO}_3^-$  and the contribution  
515 of R3 (i.e.,  $\text{N}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow 2\text{HNO}_3$ ), among which  $\text{NO}_3^-_{\text{cal 1}}$ ,  $\text{NO}_3^-_{\text{cal 2}}$ , and  $\text{NO}_3^-_{\text{obs}}$  referred to  
516 homogeneous formation (R1 and R2), total formation (R1, R2, and R3), and field measurement  
517 of  $\text{NO}_3^-$ , respectively. Although the particle-bound  $\text{NO}_3^-$  was influenced by many factors (i.e.,  
518 formation, deposition, and dispersion), the calculations generally well reproduced the measured  
519  $\text{NO}_3^-$  in episode 6, with a high correlation coefficient ( $R^2 = 0.63$ ) and an IOA of 0.78. However,  
520 on November 23, 2014, the observed  $\text{NO}_3^-$  decreased rapidly from 09:00, which was not  
521 captured by the calculations. This discrepancy was likely caused by the weather conditions on

522 that day, because (1) the average wind speed increased from 1.7 m/s before 09:00 to 2.7 m/s after  
523 09:00 and even reached 4.0 m/s at 14:00; and (2) moderate rain began at 12:00 and continued  
524 until 23:00, with total precipitation of 24 mm. Indeed, this was the beginning of a 7-day rainy  
525 period, which ended episode 6 with a sharp decrease of the PM<sub>2.5</sub> concentration (approximately  
526 175 µg/m<sup>3</sup>; see [Figure 2](#)).

527 Because the values of NO<sub>3</sub><sup>-</sup><sub>cal 1</sub> were very close to those of NO<sub>3</sub><sup>-</sup><sub>cal 2</sub>, the variation of NO<sub>3</sub><sup>-</sup> in  
528 episode 6 could be well explained by the homogeneous formation (R1 and R2), whereas the  
529 heterogeneous reaction of N<sub>2</sub>O<sub>5</sub> on aerosol surfaces (R3) made only a minor contribution to the  
530 total NO<sub>3</sub><sup>-</sup> (i.e., nearly nil from 0:00 to 17:00 and 3.7% ± 0.6% from 18:00 to 23:00). Because  
531 the homogeneous formation of NO<sub>3</sub><sup>-</sup> was closely related to the concentrations of HNO<sub>3</sub> (g) and  
532 NH<sub>3</sub> (g) and the temperature (see R1 and R2), [Table 4](#) compares the temperature, HNO<sub>3</sub> (g), NH<sub>3</sub>  
533 (g), NO, NO<sub>2</sub>, O<sub>3</sub>, and the simulated OH and HO<sub>2</sub> (a measure of oxidative capacity [[Cheng et al.,](#)  
534 [2010](#)]) between episode 6 and non-episode 2. It was found that the levels of HNO<sub>3</sub> (g) (0.65 ±  
535 0.01 ppbv) and NH<sub>3</sub> (g) (13.48 ± 0.72 ppbv) in episode 6 were significantly higher than those  
536 during the non-episode 2 (0.47 ± 0.03 and 9.54 ± 0.37 ppbv for HNO<sub>3</sub> and NH<sub>3</sub>, respectively),  
537 which might substantially favor the formation of NH<sub>4</sub>NO<sub>3</sub>. Because HNO<sub>3</sub> (g) is generally  
538 formed by oxidation of NO<sub>x</sub>, the production of HNO<sub>3</sub> (g) should be closely related to the  
539 oxidative capacity of the air and the level of NO<sub>x</sub>. In episode 6, the levels of O<sub>3</sub> (17.09 ± 2.04  
540 ppbv), OH ([3.8 ± 1.3] × 10<sup>5</sup> molecules/cm<sup>3</sup>), and HO<sub>2</sub> ([1.1 ± 0.3] × 10<sup>7</sup> molecules/cm<sup>3</sup>) were  
541 noticeably lower than those in non-episode 2 (O<sub>3</sub>, 24.57 ± 1.64 ppbv; OH, [7.2 ± 0.9] × 10<sup>5</sup>  
542 molecules/cm<sup>3</sup>; HO<sub>2</sub>, [2.0 ± 0.2] × 10<sup>7</sup> molecules/cm<sup>3</sup>), indicating a weaker oxidative capacity.  
543 However, the levels of NO (43.55 ± 11.65 ppbv) and NO<sub>2</sub> (44.93 ± 2.29 ppbv) were much higher  
544 than those in non-episode 2 (14.70 ± 2.40 and 29.46 ± 0.95 ppbv for NO and NO<sub>2</sub>, respectively),  
545 possibly leading to the enhancement of HNO<sub>3</sub> (g) in episode 6. Furthermore, the particle-bound  
546 NO<sub>3</sub><sup>-</sup> was of low thermal stability ([Querol et al., 2004](#)), and the temperature lowered ~2.3 C ° in  
547 episode 6, which suppressed the decomposition and volatilization of NH<sub>4</sub>NO<sub>3</sub>. Therefore, the  
548 high levels of NO<sub>x</sub> and NH<sub>3</sub> and the low temperature were both responsible for the increase in  
549 NO<sub>3</sub><sup>-</sup> in episode 6.



550  
 551 Figure 14. Comparison of NO<sub>3</sub><sup>-</sup> between theoretical calculations and observation in episode 6.  
 552  
 553 Table 4. Comparison of temperature, HNO<sub>3</sub> (g), NH<sub>3</sub> (g), NO, NO<sub>2</sub>, O<sub>3</sub>, and simulated OH and  
 554 HO<sub>2</sub> between episode 6 and non-episode 2.

	Episode 6	Non-episode 2
Temperature ( °C)	14.9 ±0.5	17.2 ±0.3
HNO <sub>3</sub> (ppbv)	0.65 ±0.01	0.47 ±0.03
NH <sub>3</sub> (ppbv)	13.48 ±0.72	9.54 ±0.37
NO (ppbv)	43.55 ±11.65	14.70 ±2.40
NO <sub>2</sub> (ppbv)	44.93 ±2.29	29.46 ±0.95
O <sub>3</sub> (ppbv)	17.09 ±2.04	24.57 ±1.64
OH (molecules/cm <sup>3</sup> )	(3.8 ±1.3) ×10 <sup>5</sup>	(7.2 ±0.9) ×10 <sup>5</sup>
HO <sub>2</sub> (molecules/cm <sup>3</sup> )	(1.1 ±0.3) ×10 <sup>7</sup>	(2.0 ±0.2) ×10 <sup>7</sup>

555  
 556 **3.4.3 SOC formation**  
 557 In addition to the high levels of NO<sub>3</sub><sup>-</sup> in episode 6, the proportions of OC also increased during  
 558 the autumn episodes. Because SOC is an important fraction in OC that often grows as an air  
 559 mass ages, it could help to explain the increase of OC in the autumn episodes by exploring the  
 560 possible formation mechanisms of SOC. It is well known that SOC formation is closely related



561 to SVOCs, which are formed from the reactions with oxidative radicals (i.e.,  $\text{RO}_2$ ,  $\text{NO}_3$ , and  $\text{HO}_2$ ;  
562 [Kanakidou et al., 2005](#); [Forstner et al., 1997](#)). Hence, the relationship between SOC and SVOCs  
563 was investigated. The SVOCs were simulated with the PBM-MCM model and SOC was  
564 calculated with the EC-tracer method mentioned in section 3.2.1. The speciation of SVOCs and  
565 their precursors can be found in the [Table S4 in the Supplementary Material](#). Briefly, the  
566 precursors of SVOCs include isoprene, aromatics, and  $\text{C}_7\text{-C}_{12}$  n-alkanes.

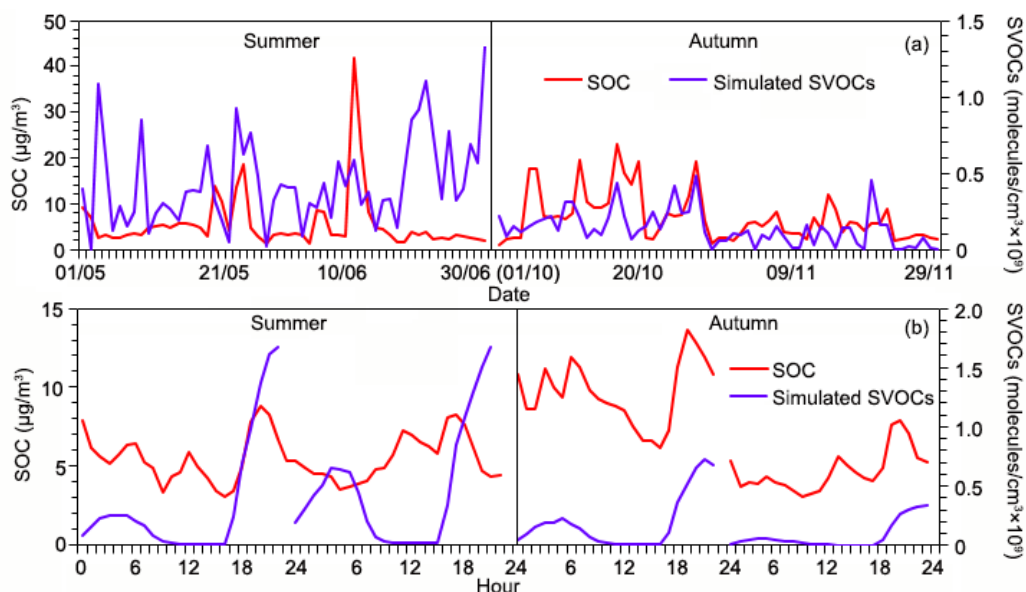
567 [Figure 15](#) presents the daily and diurnal variations in SOC and SVOCs. It was found that SOCs  
568 correlated well with SVOCs in both daily ( $R^2 = 0.52$ ) and diurnal ( $R^2 = 0.63$ ) patterns in autumn,  
569 indicating that the simulated SVOCs were responsible for the production of SOC. The oxidation  
570 products of aromatics and isoprene were the main constituents of the SVOCs, with average  
571 contributions of  $42.5\% \pm 2.8\%$  and  $39.4\% \pm 2.0\%$ , respectively. Among the aromatics, xylenes  
572 made the greatest contribution ( $15.0\% \pm 0.7\%$ ) to the SVOCs, followed by trimethylbenzenes  
573 ( $11.5\% \pm 0.7\%$ ), ethylbenzene ( $8.8\% \pm 0.5\%$ ), toluene ( $5.1\% \pm 0.7\%$ ), and benzene ( $2.2\% \pm$   
574  $0.2\%$ ). Compared to those in non-episode 2 (i.e.,  $40.7\% \pm 3.4\%$  and  $41.1\% \pm 2.4\%$  contributed  
575 by aromatics and isoprene, respectively), the contribution of aromatics to SVOCs increased to  
576  $46.3\% \pm 4.1\%$  during the episodes, whereas the proportion of isoprene oxidation products  
577 decreased to  $36.1\% \pm 3.7\%$ , suggesting that the increase in aromatics was the main cause of the  
578 autumn episodes. To quantify the contribution of biomass burning to SOC, the observed VOCs  
579 were apportioned to different sources, including biomass burning with  $\text{CH}_3\text{CN}$  as the tracer. The  
580 source profiles are provided in [Figure S12 in the Supplementary Material](#). According to the  
581 SVOCs simulated on the basis of VOCs emitted from biomass burning, the SVOCs were  
582 elevated by  $15.4\% \pm 1.3\%$  due to biomass burning during the episodes.

583 In contrast, the correlations were much worse in summer ( $R^2 = 0.01$  and  $0.31$  for daily and  
584 diurnal variations, respectively). The high frequency (50.8%) of rainy days in summer was a  
585 factor for the poor correlation; for example, the level of SOC was low during the late period of  
586 June when the precipitation lasted for about 10 days, and the model overestimated the SVOCs  
587 without considering the influence of precipitation. The correlations between SOC and SVOCs  
588 ( $R^2 = 0.14$  and  $0.19$  for the daily and diurnal variations, respectively) were still poor even after  
589 the rainy days were excluded. Hence, the poor correlation should also be related to other factors,  
590 such as incomplete consideration of the contribution of biogenic VOCs. Although isoprene was  
591 included as a precursor of the SVOCs, other biogenic species (i.e.,  $\alpha$ -pinene,  $\beta$ -pinene, and



592 monoterpenes) that were proven to be important precursors of SOC (Kanakidou et al., 2005)  
 593 were not monitored in this study. Moreover, the level of biogenic VOCs was much higher in  
 594 summer than in autumn. Taking isoprene as an example, the mixing ratio of isoprene was  $66.7 \pm$   
 595  $4.9$  pptv in summer and only  $37.2 \pm 2.6$  pptv in autumn. The higher missing level of biogenic  
 596 VOCs in summer led to a higher deficit of SVOCs, perhaps causing the poorer correlation  
 597 between SOC and SVOCs. Nevertheless, this notion needs requires validation with more  
 598 comprehensive data on biogenic VOCs.

599



600  
 601 Figure 14. Correlations between calculated SOC and simulated SVOCs in (a) day-to-day  
 602 variation and (b) diurnal pattern.

603

#### 604 4. Conclusions

605 In summer and autumn 2014, the concentrations of  $PM_{2.5}$  and its components were continuously  
 606 monitored in Wuhan; six  $PM_{2.5}$  episodes were captured. The analysis of  $PM_{2.5}$  concentrations and  
 607 compositions found that Wuhan suffered from relatively high levels of  $PM_{2.5}$ , even in the warm  
 608 seasons. Secondary inorganic ions were the most predominant species in  $PM_{2.5}$  in the form of  
 609  $NH_4NO_3$  and  $(NH_4)_2SO_4$ . Comparable levels of  $SO_4^{2-}$  and  $NO_3^-$  indicate that stationary and  
 610 mobile sources had equivalent importance in Wuhan. With the EC-tracer method, it was found  
 611 that the POC level was slightly higher than that of SOC, and both increased significantly during  
 612 the episodes. K was the most abundant element in  $PM_{2.5}$ , implying biomass burning in and

613 around Wuhan during the sampling campaign. Indeed, the source apportionment revealed that  
614 biomass burning was the main cause of increases in PM<sub>2.5</sub> in episodes 1, 3, and 4. Fugitive dust  
615 was the leading factor in episode 2. However, episodes 5 and 6 were mainly attributable to  
616 vehicle emissions and SIAs. Study of the formation mechanism of NO<sub>3</sub><sup>-</sup> and SOC found that  
617 NO<sub>3</sub><sup>-</sup> was mainly generated from the homogeneous reactions in episode 6, and the high levels of  
618 NO<sub>x</sub> and NH<sub>3</sub> and the low temperature caused the increase in NO<sub>3</sub><sup>-</sup>. Furthermore, the daily and  
619 diurnal variations of SOC correlated well with those of SVOCs in autumn. Aromatics and  
620 isoprene were the main precursors of SOC, and the contribution of aromatics increased during  
621 the episodes. However, the correlation between SOC and SVOCs was much worse in summer,  
622 possibly as a result of the incompleteness of the biogenic VOC input in the simulation of SVOCs.  
623 This study advances our understanding of the chemical characteristics of PM<sub>2.5</sub> in warm seasons  
624 in Wuhan and for the first time quantifies the contribution of biomass burning to PM<sub>2.5</sub>. The  
625 investigation of SOC formation will also inspire the application of the explicit chemical  
626 mechanisms on the study of SOA.

627  
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