



1

2

3 **Decoupling peroxyacetyl nitrate from ozone in Chinese outflows observed**
4 **at Gosan Climate Observatory**

5

6

7 **Jihyun Han^{1*}, Meehye Lee¹, Gangwoong Lee², Louisa K. Emmons³**

8

9

10 ¹Department of Earth and Environmental Sciences, Korea University, Seoul,
11 Republic of Korea

12 ²Department of Environmental Science, Hankuk University of Foreign Studies, Yongin,
13 Republic of Korea

14 ³Atmospheric Chemistry Observations and Modeling Laboratory, National Center for
15 Atmospheric Research (NCAR), Boulder, CO, USA

16 *now at: Korea Environment Institute, Sejong, Republic of Korea

17

18

19 Correspondence to: M. Lee (meehye@korea.ac.kr)

20

21

22

23 Submitted to Atmospheric Chemistry and Physics

24

December 2016

25



26 **Abstract**

27 We measured peroxyacetyl nitrate (PAN) and other reactive species such as O₃, NO₂, CO,
28 and SO₂ with aerosols including PM₁₀ and PM_{2.5} organic carbon (OC) and elemental carbon
29 (EC) at Gosan Climate Observatory in Korea (33.17°N, E126.10°E) during October 10 to
30 November 6, 2010. PAN was determined through fast gas chromatography with luminol
31 chemiluminescence detection at 425 nm every 2 min. The PAN mixing ratios ranged from 0.1
32 (detection limit) to 2.4 ppbv with a mean of 0.6 ppbv. For all measurements, PAN was
33 unusually better correlated with PM₁₀ (Pearson correlation coefficient, $\gamma = 0.75$) than with O₃
34 ($\gamma = 0.67$). In particular, the O₃ level was highly elevated with SO₂ at midnight, along with a
35 typical midday peak when air was transported rapidly from the Beijing areas. The PAN
36 enhancement was most noticeable during the occurrence of haze under stagnant conditions.
37 In Chinese outflows slowly transported over the Yellow Sea, PAN gradually increased up to
38 2.4 ppbv at night, in excellent correlation with a concentration increase of PM_{2.5} OC and EC,
39 PM_{1.0} K⁺, and PM₁₀ mass. The high K⁺ and OC/EC ratio indicated that the air mass was
40 impacted by biomass combustion. This study highlights PAN decoupling with O₃ in Chinese
41 outflows and suggests PAN as a potential indicator of overall aerosol formation in aged air
42 masses impacted by biomass burning.

43

44 Key words: PAN, O₃, PM₁₀, Chinese outflow, Haze, Biomass combustion



45 1. Introduction

46 At the surface, ozone is primarily photochemically produced, and the contribution from the
47 stratosphere is generally small. Ozone is formed through reactions of various precursors such
48 as CO, CH₄, volatile organic compounds (VOCs), and NO_x (e.g., Brasseur et al., 1999; Jacob,
49 2000; Nielsen et al., 1981). Likewise, peroxyacetyl nitrate (PAN) is a secondary product of
50 urban air pollution and a significant oxidant in the atmosphere (e.g., Hansel and Wisthaler,
51 2000; La Franchi et al., 2009; Lee et al., 2012; Liu et al., 2010; Roberts et al., 2007). PAN is
52 solely produced by the photochemical reaction between the peroxyacetyl radical and nitrogen
53 dioxide, and the peroxyacetyl radical is derived from the OH oxidation or photolysis of
54 VOCs such as acetaldehyde, methylglyoxal, and acetone (e.g., Fischer et al., 2014; La
55 Franchi et al., 2009; Lee et al., 2012). For this reason, PAN is a very useful indicator of
56 photochemical air pollution. As thermal decomposition is a major PAN sink in the
57 troposphere (Beine et al., 1997; Jacob, 2000; Kenley and Hendry, 1982; Talukdar et al., 1995),
58 the lifetime of PAN depends on temperature. For example, the PAN lifetime is ~5 years at
59 -26°C and 1 h at 20°C (Fischer et al., 2010; Zhang et al., 2011). At high altitudes above ~7
60 km, photolysis becomes the most important loss process for PAN (Talukdar, et al., 1995).
61 Thus, PAN can be an indicator of NO_y concentration in the free troposphere in urban areas
62 and a guide for the long-range transport of NO_x in remote regions (Jacob, 1999).

63 In the past decades, PAN was measured not only in urban areas (Aneja et al., 1999;
64 Gaffney et al., 1999; Grosjean et al., 2002; Lee et al., 2008; Zhang et al., 2014) but also in
65 background regions (Fischer et al., 2011; Kanaya et al., 2007; Lee et al., 2012), onboard
66 aircraft (Tereszchuk et al., 2013), and ships (Roberts et al., 2007). PAN concentrations were
67 in the range of a few ppbv in urban areas close to VOCs and NO_x sources (Lee et al., 2008;
68 Zhang et al., 2011). In remote regions, PAN mixing ratios were generally in the range of a



69 few pptv (Gallagher et al., 1990; Mills et al., 2007; Muller and Rudolph, 1992; Staudt et al.,
70 2003).

71 In recent years, NO_x and VOCs have gradually increased in East Asia, particularly China
72 (Akimoto, 2003; Liu et al., 2010; Ohara et al., 2007), leading to an increase in the
73 concentrations of photochemical byproducts such as PAN and O_3 not only in East Asia (Liu et
74 al., 2010; Wang et al., 2010; Zhang et al., 2009; Zhang et al., 2011; Zhang et al., 2014) but
75 also in North America (Fischer et al., 2010; Fischer et al., 2011; Jaffe et al., 2007; Zhang et
76 al., 2008). These results were also demonstrated by the GEOS-Chem model (Zhang et al.,
77 2008). In addition to urban plumes, PAN was reported to be enhanced by biomass
78 combustion (Alvarado et al., 2010; Coheur et al., 2007), such as open burning and use of
79 biofuel, which is used to take place often in China after crop harvesting (Cao et al., 2006;
80 Duan et al., 2004). In this context, PAN is a useful indicator for diagnosing Chinese outflows
81 and assessing their perturbation on regional air quality in the northwestern Pacific region.

82 Gosan Climate Observatory (GCO) is an ideal place to monitor Asian outflows and their
83 transformation and to estimate their impact on air quality over the northern Pacific region
84 (Lee et al., 2007; Lim et al., 2012). In the present study, PAN was first measured
85 continuously at GCO to characterize its variation and source in relation to O_3 and to
86 understand the influence of Chinese outflows on the regional air quality.

87

88 2. Experiments

89 PAN measurements were conducted at GCO (33.17°N, E126.10°E) on Jeju Island from
90 October 10 to November 6, 2010. GCO is located on a cliff at the western edge of Jeju Island.

91 PAN was determined through fast gas chromatography (GC) with luminol
92 chemiluminescence detection, which is described in detail elsewhere (Gaffney et al., 1998;



93 Lee et al., 2008; Marley et al., 2004). Here, we briefly describe the measurement method.
94 Ambient air PAN and NO₂ (and peroxypropyl nitrate (PPN) if present) were separated along
95 a 10-m capillary GC column (DB-1, J&W Scientific, Folsom, CA, USA), whose end was
96 connected to a luminol cell where the column effluent reacted with luminol, giving off
97 luminescent light (Lee et al., 2008; Lee et al., 2012). The concentrations of PAN and other
98 species were determined from the chemiluminescence signals detected by a gated photon
99 counter (HC135-01, Hamamatsu, Bridgewater, NJ, USA) at 425 nm, which was set at 800 V
100 and operated at room temperature (Gaffney et al., 1998; Lee et al., 2012; Lee et al., 2008).
101 PAN was calibrated against standards synthesized by the nitration of peracetic acid in n-
102 tridecane (Gaffney et al., 1984; Gregory, 1990). The nominal detection limit of PAN defined
103 by 3σ of the lowest standard was 100 pptv (Lee et al., 2008).

104 Water-soluble ions of PM_{1.0} were collected by a particle-into-liquid sampler and analyzed
105 by ion chromatography. Gaseous species, including O₃, NO, NO₂, CO, and SO₂, were
106 measured by UV absorption, chemiluminescence with a molybdenum converter, non-
107 dispersive infrared, and pulse UV fluorescence method, respectively (NIER, 2016). Aerosol
108 species, including PM₁₀ mass and PM_{2.5} OC and EC were measured and recorded along with
109 meteorological parameters (relative humidity, temperature, and wind speed). The detailed
110 results of the aerosol measurements can be found in Shang et al. (2017).

111 The three-day backward trajectories of air parcel at 850m a.s.l. for every one hour were
112 calculated using NOAA Air Resources Laboratory (ARL) Hybrid Single-Particle Lagrangian
113 Integrated Trajectory (HYSPLIT) model (version 4) (Draxler and Rolph, 2012; Rolph, 2012,
114 <http://www.arl.noaa.gov/ready/hysplit4.html>).

115

116 3. Results



117 In the present experiments, PAN mixing ratios range from 0.1 to 2.4 ppbv, with an average
118 of 0.6 ppbv. This mean value is lower than those observed in other Asian megacities: Beijing
119 (1.41 ppb in the summer), Pearl River Delta region (1.32 ppb in the summer), and Seoul (0.8
120 ppb in the early summer); similar to those of suburban areas in China, e.g., Lanzhou (0.76
121 ppb in the summer); and higher than those in the western coast of the US, e.g., Sacramento
122 (0.45 ppb in the summer), Mt. Bachelor (0.144 ppb in the spring and early summer), off the
123 western coast of the US (0.65 ppb in the spring), and over the remote North Pacific (total
124 PAN < 0.3 ppb in spring) (Bertram et al., 2013; Fischer et al., 2011; La Franchi et al., 2009;
125 Lee et al., 2008; Roberts et al., 2004; Wang et al., 2010; Zhang et al., 2009; Zhang et al.,
126 2011). Because the PAN lifetime is greatly dependent on temperature, its concentration
127 decreases with increasing distance from the source regions. The PAN mixing ratios calculated
128 in this study thus lie in-between the levels for the East Asian megacities and the northern
129 Pacific. The distributions of all measured species, including PAN and O₃, are presented in Fig.
130 1. In particular, there are several periods characterized by high concentrations of PAN, O₃,
131 and PM₁₀. In terms of PAN, four periods are particularly interesting (Fig. 1). High O₃
132 concentrations were observed during October 31–November 2 [episode 1] but did not
133 coincide with high PAN concentrations. During October 28–29 [episode 2], NO₂ was
134 noticeably increased. On the other hand, PAN and O₃ concentrations were both high during
135 October 20–21 [episode 3] and November 4–5 [episode 4]. Episodes 3 and 4 are
136 characterized by haze, while episodes 1 and 2 are characterized by urban influence in the
137 Korean and Beijing outflows, respectively.

138 In the present study, PAN correlates reasonably well with O₃ ($\gamma = 0.67$) and even better
139 with PM₁₀ ($\gamma = 0.75$). In general, O₃ and PAN exhibit typical diurnal variation with a
140 maximum recorded in the afternoon, which results in a good correlation between the two



141 (Brasseur et al., 1999; Gaffney et al., 1999; Ridley et al., 1990; Schrimpf et al., 1995; Wang
142 et al., 2010). In this study, however, the O₃ peak was often found in the early morning and
143 late afternoon for several days (Fig. 1). Observing the diurnal variations in the entire PAN
144 concentration measurement set (Fig. 2), the maximum was clearly recorded in the morning
145 with the highest outliers, which is rather similar to that of PM₁₀. The diurnal pattern of NO₂
146 shows little variation, even though its concentrations were increased in the morning along
147 with PAN. This first measurement of PAN at GCO reveals that PAN is not always coupled
148 with O₃, which was not typically observed at remote sites in previous studies (e.g., Fischer et
149 al., 2010; Lee et al., 2012).

150

151 4. Discussion

152 4.1. Decoupling of PAN from O₃

153 To examine the detailed mechanism of the decoupling of PAN from O₃, the daily
154 maximum concentrations of PAN and O₃ were further explored. The recorded daily PAN
155 maxima were generally in good correlation with O₃, albeit the relationship did not seem to
156 hold at high concentrations of PAN and O₃ (Fig. 3). The daily maxima were then categorized
157 into four groups according to the time when each O₃ and PAN maximum was recorded: “O₃
158 day-PAN day,” “O₃ day-PAN night,” “O₃ night-PAN day,” and “O₃ night-PAN night.” The
159 day interval started from 08:00 and ended at 18:00 (local time), based on the times of sunrise
160 and sunset during the experiment period. While the high PAN concentrations were associated
161 with the “O₃ day-PAN day” group (cross symbols in Fig. 3), the enhanced O₃ concentration
162 was recorded in the “O₃ night-PAN night” group (star symbols in Fig. 3). The “O₃ night-PAN
163 night” group unexpectedly held more data points than the “O₃ day-PAN day” group, even
164 though the “O₃ night-PAN night” group concentrations were lower (Fig. 3). In addition, there



165 were several days classified in the “O₃ night-PAN day” (marked by diamond) and “O₃ day-
166 PAN night” groups, but with less frequency and lower concentrations. These results indicate
167 that the decoupling of PAN from O₃ was primarily due to the elevated concentrations of O₃
168 and/or PAN at night. The four high PAN and O₃ episodes identified in this study fall under
169 the category of “O₃ night-PAN night” or “O₃ day-PAN day.” This point will be further
170 examined to identify the chemical and physical processes responsible for PAN being
171 decoupled from O₃, instead of being coupled with PM₁₀. The overall characteristics of the
172 four episodes are summarized in Table 1.

173

174 **4.2. Export of O₃ from Asian continents (episodes 1 & 2)**

175 High O₃ concentrations were encountered around midnight on three consecutive days from
176 October 31 to November 2 (episode 1), during which SO₂ reached its maximum
177 concentration (Fig. 1). The backward trajectories of air masses revealed that air passed
178 through the Beijing area during this period (Fig. 4). The wind was strong (13.5 m/s on
179 average) and the recorded O₃ maximum (80.6 ppbv) was concurrent with the PAN maximum
180 (0.9 ppbv) around midnight on November 1st (Fig. 4).

181 All these results indicate that the air was heavily influenced by outflow from the Beijing
182 area, as previously hypothesized (Lim et al., 2012), and that the nighttime enhancement of O₃
183 and PAN resulted from the fast transport of relatively less-aged urban plumes. In this episode,
184 PAN and O₃ could have been formed in urban areas or produced in the outflow while being
185 transported. Because the overall correlation between O₃ and PAN was the best with the
186 highest daily $\Delta O_3/\Delta PAN$ among all cases discussed in this study, episode 1 likely represents
187 an event of rapid transport from the Beijing area (Fig. 5).

188 In previous studies, the nighttime enhancement of O₃ was observed at GCO (e.g., Lee et
189 al., 2007) in association with pollutant-laden air coming from Beijing. Similarly, Banta et al.



190 (1998) pointed out that the evening O₃ maximum was due to long-range transport of O₃ from
191 nearby urban areas. Wang et al. (2011) reported that the O₃ lifetime was about two days in
192 East China during the summer, which is sufficient time for air to travel to GCO. Therefore,
193 the nighttime maximum of O₃ can be attributed to the export of O₃ from megacities in China,
194 causing PAN to be decoupled from O₃. Another night maximum of O₃ was recorded on
195 October 29. Note that NO_x was highly elevated during October 28–29 (episode 2) (Fig. 5b).
196 In contrast, O₃ and PAN levels remained relatively low, leading to the lowest daily
197 $\Delta\text{O}_3/\Delta\text{PAN}$ among all episodes. In this case, air masses passed through the Korean Peninsula,
198 carrying low O₃ being titrated by high NO_x (Brasseur et al., 1999; Jacobson, 2005).

199

200 **4.3. PAN enhancement upon occurrence of haze (episodes 3 & 4)**

201 In this study, two haze events were observed in the very beginning (October 20–21;
202 episode 3) and the end of the study period (November 4–5; episode 4). The first haze event
203 occurred on October 18th and lingered until October 21st, during which O₃ concentrations
204 were gradually elevated. A second peak was recorded around midnight of October 19th and
205 20th, and the maximum was reached in the afternoon of October 20th (Figs. 1 and 3). In this
206 episode, the maximum concentrations of O₃ and PAN were 78.9 ppbv and 2.0 ppbv,
207 respectively, on October 29th, when the highest NO₂ concentration (12.7 ppbv) was observed
208 under low wind speed (6.6 m/s daily average). The air mass trajectories suggest the influence
209 of the Korean Peninsula, particularly the Seoul metropolitan area, in addition to East China
210 (Fig. 4b).

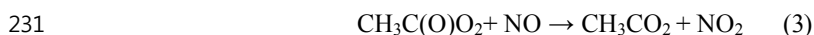
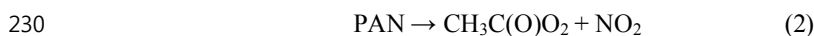
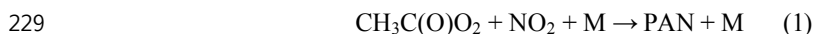
211 In the second haze event (episode 4), an air mass was slowly transported from East China,
212 including the Jiangsu province, under stagnant condition which was developed by an
213 anticyclone system (Fig. 4). We measured the highest concentrations of all aerosol species



214 including the PM₁₀ mass as well as PAN and O₃, which were 170 μg/m³, 2.4 ppbv, and 87.5
215 ppbv, respectively. Other reactive gases such as CO, SO₂, and NO₂ were also highly elevated.
216 Note that PAN and O₃ gradually increased through the night, leading to a nighttime maximum
217 of both species on November 4th. It is likely that the pre-formed PAN and O₃ were
218 continuously transported into Gosan at night.

219 In section 4.2, the nighttime O₃ peak was attributed to the transport from nearby urban
220 areas to Jeju Island. The two haze episodes were also observed in continental outflows.
221 Unlike O₃, however, PAN is linearly correlated with the PM₁₀ mass and major constituents of
222 aerosols including PM_{2.5} OC, PM_{2.5} EC, and PM_{1.0} K⁺, whose concentrations were
223 remarkably high in this episode.

224 PAN is formed through the reaction of the peroxyacetyl radical and nitrogen dioxide (Eq. 1)
225 and decomposed at high temperature (Eq. 2), returning these radicals. Unless the NO
226 concentration is high (Eq. 3), the peroxyacetyl radical recombines with NO₂, producing PAN.
227 Thus, the total lifetime of PAN depends on the NO₂/NO ratio and temperature (Eq. 4)
228 (Brasseur et al., 1999).



232
$$T_{eff} = T_d \left(1 + \frac{k_1[\text{NO}_2]}{k_2[\text{NO}]} \right) [\text{sec}^{-1}] \quad (4)$$

233 where T_d and T_{eff} indicate the lifetime against decomposition and the effective lifetime of
234 PAN (Brasseur et al., 1999). The effective lifetime of PAN was estimated through Eq. 4 using
235 the rate constants proposed by Brasseur et al. (1999), Jacobson (2005), and Maricq and
236 Szente (1996).

237 During the haze event, NO was close to the detection limit, while NO₂ was greatly
238 enhanced. Owing to the high NO₂/NO ratio, the effective lifetime of PAN increased by 57



239 times; this possibly contributed to the gradual increase in PAN through the night on
240 November 4th. In an aged plume, NO_2 is likely to be recycled with O_3 during the day and
241 PAN during the night. Fischer et al. (2014) also reported that, at night, PAN can be produced
242 from the reaction of acetaldehyde with the nitrate radical.

243 Besides PM_{10} , PAN was also well correlated with $\text{PM}_{2.5}$ OC and EC not only during this
244 haze episode but also during the entire measurement period (Fig. 6b and c). Furthermore, the
245 enhancement of PAN was concurrent with that of OC and K^+ , resulting in excellent
246 correlation between them (Fig. 6e and f). In fact, the $\Delta\text{OC}/\Delta\text{EC}$ ratio of episode 4 was much
247 higher (7) than those of the other episodes (~ 2.5) (Fig. 6d). The fraction of $\text{PM}_{2.5}$ against
248 PM_{10} was also the highest in this episode, indicating significant contribution of secondary
249 aerosols. These observations suggest that air masses were affected by biomass combustion
250 (e.g., Ram et al., 2008, 2012; Saarikoski et al., 2008).

251 According to previous studies, PAN can be produced in plumes through biomass
252 combustion (Alvarado et al., 2010; Coheur et al., 2007; Liu et al., 2016; Tereszchuk et al.,
253 2013). In northeast China, open burnings related to agricultural activities frequently occur
254 during the spring and fall (Duan et al., 2004; Yang et al., 2005). Kudo et al. (2014) also
255 reported that, upon burning crop residue in Yangtze region, the levels of oxygenated VOCs
256 were elevated together with NO_x . In addition, biofuel is used for cooking and heating and as
257 an energy source in China's industry (Cao et al., 2006).

258 Therefore, PAN is likely to increase when haze occurs and fine aerosols are transformed as
259 air masses carrying combustion emissions are slowly transported from China over the Yellow
260 Sea. Additionally, the results of this study imply that PAN can be used as a robust tracer for
261 continental outflows in northeast Asia, to identify transport- and chemical transformation-
262 dominant regimes. In a transport-dominant regime, O_3 export was distinguished by the



263 highest levels of primary gaseous species such as SO₂ and relatively low levels of PAN. In
264 contrast, fine aerosol species are enhanced in a chemical transformation regime, leading to
265 haze events with relatively more enhanced PAN compared to O₃.

266 Finally, the measured O₃ and PAN concentrations were compared to results from a global
267 chemistry model, the Community Atmosphere Model with Chemistry (CAM-Chem), a
268 component of the Community Earth System Model (CESM) (Lamarque et al., 2012; Tilmes
269 et al., 2015). The CAM-chem results shown here follow the configuration used for the
270 HTAP2 (Hemispheric Transport of Air Pollution, Phase 2) intercomparison (e.g., Stjern et al.,
271 2016). CAM-chem is nudged to observed meteorology (GEOS-5) to reproduce the actual
272 period of the observations (Oct 2010). The emissions used in the model are the HTAP2
273 inventory (Janssens-Maenhout, et al., 2015), which include the "MIX" Asian emissions
274 inventory. Biomass burning emissions are from the Global Fire Emissions Database (GFED3)
275 (Randerson et al., 2013). In the model simulation, O₃ and PAN were highly underestimated
276 during the episodes observed in Chinese outflows, although the variation around average
277 level of O₃ and PAN was well captured (Fig. 7). The enhancement of PAN during the haze
278 events was not well represented in the model (Oct 20–21 and Nov 4–5). The timing of the O₃
279 diurnal variability was captured by the model, although the magnitude of the variation was
280 underestimated. These results reveal that the current understanding of Chinese outflow is still
281 not sufficient, thereby causing uncertainty in estimating its effect on air quality in the
282 northwestern Pacific Rim.



283 5. Conclusions

284 The first measurements of PAN, reactive gases, and aerosol species were conducted at
285 GCO during October 19 to November 6, 2010. The average concentration of PAN was 0.6
286 ppbv with a maximum of 2.4 ppbv, which was lower than those in major cities in East Asia
287 but much higher than the background concentrations in other regions. In addition, PAN and
288 O₃ concentrations were well correlated ($\gamma = 0.67$). However, the comparison of the daily
289 maxima of PAN and O₃ highlighted that they were not proportionally enhanced. That is,
290 either PAN was relatively more elevated than O₃ or the highest O₃ was associated with low
291 levels of PAN. Unexpectedly, both PAN and O₃ often reached their maxima at night. As a
292 result, PAN was decoupled from O₃ and better correlated with the PM₁₀ mass ($\gamma = 0.75$) than
293 with O₃. In this study, these high-concentration episodes were all encountered in association
294 with continental outflows, and thus, two high-O₃ and two high-PAN events were recorded
295 and investigated in detail.

296 During the O₃ episodes, both O₃ and PAN concentrations reached their maximum values at
297 night. In episode 1 (Oct. 31 to Nov. 2), the O₃ concentration was increased to 80.6 ppbv, with
298 a high SO₂ concentration under strong wind. It was a typical Beijing plume observed in the
299 study region. In comparison, NO₂ was greatly increased in episode 2 (Oct. 28–29) when the
300 air masses were affected by urban emissions from Korean Peninsula. Although the maximum
301 O₃ level was lower during episode 2, these two cases demonstrated well how O₃ was exported
302 from the East Asian continent.

303 The remaining two episodes were highlighted by enhanced PAN concentrations and
304 characterized by haze occurrence. During episode 3 (Oct. 20–21), PAN and O₃ concentrations
305 increased up to 2.0 ppbv and 78.9 ppbv, respectively, with high NO_x levels, probably
306 influenced by emissions from Korea. Episode 4 (Nov. 4–5) was characterized by the highest



307 concentrations of almost all measured species, including PAN, O₃, PM₁₀ mass, and PM_{1.0}
308 species; the maximum recorded concentrations of PAN, O₃, and PM₁₀ mass during this
309 interval were 2.4 ppbv, 87.5 ppbv, and 170 µg/m³, respectively. Note that, along with PM₁₀
310 and O₃, PAN was gradually increased through the night. In this episode, an air mass was
311 slowly transported from eastern China. With depleted NO, the effective lifetime of PAN was
312 greatly extended. In addition, PAN concentration showed good correlation with OC, EC, and
313 K⁺; in fact, the correlation of PAN with K⁺ was comparable to that of OC with K⁺. These
314 results, in conjunction with the high ΔOC/ΔEC (7), imply that the observed haze was mainly
315 caused by the emissions produced by biomass combustion. These results suggest that PAN is
316 a useful tool for distinguishing continental outflows that were typically observed in northeast
317 Asia.

318 The comparison between the measured and calculated concentrations using the CAM-
319 Chem-HTAP2 model showed that the model underestimated the O₃ and PAN levels in
320 Chinese outflows, particularly for haze incidence. These results reveal that Chinese outflows
321 are still poorly understood and not well captured in the model.

322

323 **Acknowledgments**

324 This study was funded by the Korea Meteorological Administration Research and
325 Development Program under Grant KMIPA 2015-6020. The National Center for Atmospheric
326 Research is funded by the National Science Foundation. The authors gratefully acknowledge
327 the NOAA Air Resources Laboratory (ARL) for the provision of the HYSPLIT transport and
328 dispersion model and/or READY website (<http://www.ready.noaa.gov>) used in this
329 publication.

330 **References**

- 331 Akimoto, H.: Global air quality and pollution, *Science*, 302, 1716-1719,
332 doi:10.1126/science.1092666, 2003.
- 333 Alvarado, M. J., Logan, J. A., Mao, J., Apel, E., Riemer, D., Blake, D., Cohen, R. C., Min, K.
334 E., Perring, A. E., Browne, E. C., Wooldridge, P. J., Diskin, G. S., Sachse, G. W.,
335 Fuelberg, H., Sessions, W. R., Harrigan, D. L., Huey, G., Liao, J., Case-Hanks, A.,
336 Jimenez, J. L., Cubison, M. J., Vay, S. A., Weinheimer, A. J., Knapp, D. J., Montzka, D.
337 D., Flocke, F. M., Pollack, I. B., Wennberg, P. O., Kurten, A., Crouse, J., Clair, J. M. S.,
338 Wisthaler, A., Mikoviny, T., Yantosca, R. M., Carouge, C. C., and Le Sager, P.: Nitrogen
339 oxides and PAN in plumes from boreal fires during ARCTAS-B and their impact on
340 ozone: an integrated analysis of aircraft and satellite observations, *Atmos. Chem. Phys.*,
341 10, 9739-9760, doi:10.5194/acp-10-9739-2010, 2010.
- 342 Aneja, V. P., Hartsell, B. E., Kim, D. S., and Grosjean, D.: Peroxyacetyl nitrate in Atlanta,
343 Georgia: Comparison and analysis of ambient data for suburban and downtown
344 locations, *J. Air & Waste Manage. Assoc.*, 49, doi: 177-184,
345 10.1080/10473289.1999.10463786, 1999.
- 346 Banta, R. M., Senff, C. J., White, A. B., Trainer, M., McNider, R. T., Valente, R. J., Mayor, S.
347 D., Alvarez, R. J., Hardesty, R. M., Parrish, D., and Fehsenfeld, F. C.: Daytime buildup
348 and nighttime transport of urban ozone in the boundary layer during a stagnation episode,
349 *J. Geophys. Res. Atmos.*, 103, 22519-22544, doi:10.1029/98jd01020, 1998.
- 350 Beine, H. J., Jaffe, D. A., Herring, J. A., Kelley, J. A., Krognes, T., and Stordal, F.: High-
351 latitude springtime photochemistry .1. NO_x, PAN and ozone relationships, *J. Atmos.*
352 *Chem.*, 27, 127-153, doi:10.1023/a:1005869900567, 1997.
- 353 Bertram, T. H., Perring, A. E., Wooldridge, P. J., Dibb, J., Avery, M. A., and Cohen, R. C.: On
354 the export of reactive nitrogen from Asia: NO_x partitioning and effects on ozone, *Atmos.*
355 *Chem. Phys.*, 13, 4617-4630, doi:10.5194/acp-13-4617-2013, 2013.
- 356 Brasseur, G. P., Orlando, J. J., and Tyndall, G. S.: *Atmospheric chemistry and global change*,
357 Oxford University Press, New York, 235-347 pp., 1999.
- 358 Cao, G., Zhang, X., and Zheng, F.: Inventory of black carbon and organic carbon emissions
359 from China, *Atmos. Environ.*, 40, 6516-6527, doi:10.1016/j.atmosenv.2006.05.070,
360 2006.
- 361 Coheur, P. F., Herbin, H., Clerbaux, C., Hurtmans, D., Wespes, C., Carleer, M., Turquety, S.,
362 Rinsland, C. P., Remedios, J., Hauglustaine, D., Boone, C. D., and Bernath, P. F.: ACE-
363 FTS observation of a young biomass burning plume: first reported measurements of
364 C₂H₄, C₃H₆O, H₂CO and PAN by infrared occultation from space, *Atmos. Chem. Phys.*,
365 7, 5437-5446, doi:10.5194/acp-7-5437-2007, 2007.
- 366 Draxler, R. R., and Rolph, G. D.: HYSPLIT (HYbrid Single-Particle Lagrangian Integrated
367 Trajectory) Model access via NOAA ARL READY Website
368 (<http://ready.arl.noaa.gov/HYSPLIT.php>), NOAA Air Resources Laboratory, Silver
369 Spring, MD., 2012.
- 370 Duan, F., Liu, X., Yu, T., and Cachier, H.: Identification and estimate of biomass burning
371 contribution to the urban aerosol organic carbon concentrations in Beijing, *Atmos.*
372 *Environ.*, 38, 1275-1282, doi:10.1016/j.atmosenv.2003.11.037, 2004.



- 373 Fischer, E. V., Jaffe, D. A., Reidmiller, D. R., and Jaeglé, L.: Meteorological controls on
374 observed peroxyacetyl nitrate at Mount Bachelor during the spring of 2008, *J. Geophys.*
375 *Res.*, 115, D03302, doi:10.1029/2009jd012776, 2010.
- 376 Fischer, E. V., Jaffe, D. A., and Weatherhead, E. C.: Free tropospheric peroxyacetyl nitrate
377 (PAN) and ozone at Mount Bachelor: potential causes of variability and timescale for
378 trend detection, *Atmos. Chem. Phys.*, 11, 5641-5654, doi:10.5194/acp-11-5641-2011,
379 2011.
- 380 Fischer, E. V., Jacob, D. J., Yantosca, R. M., Sulprizio, M. P., Millet, D. B., Mao, J., Paulot, F.,
381 Singh, H. B., Roiger, A., Ries, L., Talbot, R. W., Dzepina, K., and Pandey, D. S.:
382 Atmospheric peroxyacetyl nitrate (PAN): a global budget and source attribution, *Atmos.*
383 *Chem. Phys.*, 14, 2679-2698, doi:10.5194/acp-14-2679-2014, 2014.
- 384 Gaffney, J. S., Fajer, R., and Senum, G. I.: An improved procedure for high purity gaseous
385 peroxyacetyl nitrate production: Use of heavy lipid solvents, *Atmos. Environ.*, 18, 215-
386 218, doi:10.1016/0004-6981(84)90245-2, 1984.
- 387 Gaffney, J. S., Bornick, R. M., Chen, Y. H., and Marley, N. A.: Capillary gas chromatographic
388 analysis of nitrogen dioxide and pans with luminol chemiluminescent detection, *Atmos.*
389 *Environ.*, 32, 1445-1454, doi:10.1016/S1352-2310(97)00098-8, 1998.
- 390 Gaffney, J. S., Marley, N. A., Cunningham, M. M., and Doskey, P. V.: Measurements of
391 peroxyacetyl nitrates (PANs) in Mexico City: implications for megacity air quality
392 impacts on regional scales, *Atmos. Environ.*, 33, 5003-5012, doi:10.1016/S1352-
393 2310(99)00263-0, 1999.
- 394 Gallagher, M. S., Carsey, T. P., and Farmer, M. L.: Peroxyacetyl nitrate in the North Atlantic
395 marine boundary layer, *Global Biogeochem. Cycle.*, 4, 297-308,
396 doi:10.1029/GB004i003p00297, 1990.
- 397 Gregory, G. L.: An intercomparison of airborne PAN measurements, *J. Geophys. Res.*, 95,
398 10077-10087, doi:10.1029/JD095iD07p10077, 1990.
- 399 Grosjean, E., Grosjean, D., Woodhouse, L. F., and Yang, Y.-J.: Peroxyacetyl nitrate and
400 peroxypropionyl nitrate in Porto Alegre, Brazil, *Atmos. Environ.*, 36, 2405-2419,
401 doi:10.1016/S1352-2310(01)00541-6, 2002.
- 402 Hansel, A., and Wisthaler, A.: A method for real-time detection of PAN, PPN and MPAN in
403 ambient air, *Geophys. Res. Lett.*, 27, 895-898, doi:10.1029/1999gl010989, 2000.
- 404 Jacob, D. J.: Introduction to atmospheric chemistry, 199-231 pp., 1999.
- 405 Jacob, D. J.: Heterogeneous chemistry and tropospheric ozone, *Atmos. Environ.*, 34, 2131-
406 2159, doi:10.1016/s1352-2310(99)00462-8, 2000.
- 407 Jacobson, M. Z.: Fundamentals of atmospheric modeling, Second edition, Cambridge, UK,
408 731-738 pp., 2005.
- 409 Jaffe, D. A., Thornton, J., Wolfe, G., Reidmiller, D., Fischer, E. V., Jacob, D. J., Zhang, L.,
410 Cohen, R., Singh, H., Weinheimer, A., and Flocke, F.: Can we detect an Influence over
411 North America from Increasing Asian NOx Emissions?, *Eos Trans, AGU*, 88, 2007.
- 412 Janssens-Maenhout, G., Crippa, M., Guizzardi, D., Dentener, F., Muntean, M., Pouliot, G.,
413 Keating, T., Zhang, Q., Kurokawa, J., Wankmüller, R., Denier van der Gon, H., Kuenen,
414 J. J. P., Klimont, Z., Frost, G., Darras, S., Koffi, B., and Li, M.: HTAP_v2.2: a mosaic of



- 415 regional and global emission grid maps for 2008 and 2010 to study hemispheric
416 transport of air pollution, *Atmos. Chem. Phys.*, 15, 11411-11432, doi:10.5194/acp-15-
417 11411-2015, 2015.
- 418 Kanaya, Y., Tanimoto, H., Matsumoto, J., Furutani, H., Hashimoto, S., Komazaki, Y., Tanaka,
419 S., Yokouchi, Y., Kato, S., Kajii, Y., and Akimoto, H.: Diurnal variations in H₂O₂, O₃,
420 PAN, HNO₃ and aldehyde concentrations and NO/NO₂ ratios at Rishiri Island, Japan:
421 Potential influence from iodine chemistry, *Sci. Total Envir.*, 376, 185-197,
422 doi:10.1016/j.scitotenv.2007.01.073, 2007.
- 423 Kenley, R. A., and Hendry, D. G.: Generation of peroxy radicals from peroxy nitrates
424 (ROONO₂). Decomposition of peroxybenzoyl nitrate (PBzN), *J. Am. Chem. Soc.*, 104,
425 220-224, doi:10.1021/ja00365a040, 1982.
- 426 Kudo, S., Tanimoto, H., Inomata, S., Saito, S., Pan, X., Kanaya, Y., Taketani, F., Wang, Z.,
427 Chen, H., Dong, H., Zhang, M., and Yamaji, K.: Emissions of nonmethane volatile
428 organic compounds from open crop residue burning in the Yangtze River Delta region,
429 China, *J. Geophys. Res.*, 119, 7684–7698, doi:10.1002/2013JD021044, 2014.
- 430 LaFranchi, B. W., Wolfe, G. M., Thornton, J. A., Harrold, S. A., Browne, E. C., Min, K. E.,
431 Wooldridge, P. J., Gilman, J. B., Kuster, W. C., Goldan, P. D., De Gouw, J. A., McKay,
432 M., Goldstein, A. H., Ren, X., Mao, J., and Cohen, R. C.: Closing the peroxy acetyl
433 nitrate budget: Observations of acyl peroxy nitrates (PAN, PPN, and MPAN) during
434 BEARPEX 2007, *Atmos. Chem. Phys.*, 9, 7623-7641, doi:10.5194/acp-9-7623-2009,
435 2009.
- 436 Lamarque, J. F., Emmons, L. K., Hess, P. G., Kinnison, D. E., Tilmes, S., Vitt, F., Heald, C. L.,
437 Holland, E. A., Lauritzen, P. H., Neu, J., Orlando, J. J., Rasch, P. J., and Tyndall, G. K.:
438 CAM-chem: description and evaluation of interactive atmospheric chemistry in the
439 Community Earth System Model, *Geosci. Model Dev.*, 5, 369-411, doi:10.5194/gmd-5-
440 369-2012, 2012.
- 441 Lee, G., Jang, Y., Lee, H., Han, J.-S., Kim, K.-R., and Lee, M.: Characteristic behavior of
442 peroxyacetyl nitrate (PAN) in Seoul megacity, Korea, *Chemosphere*, 73, 619-628,
443 doi:10.1016/j.chemosphere.2008.05.060, 2008.
- 444 Lee, G., Choi, H.-S., Lee, T., Choi, J., Park, J. S., and Ahn, J. Y.: Variations of regional
445 background peroxyacetyl nitrate in marine boundary layer over Baengyeong Island,
446 South Korea, *Atmos. Environ.*, 61, 533-541, doi:10.1016/j.atmosenv.2012.07.075, 2012.
- 447 Lee, M., Song, M., Moon, K. J., Han, J. S., Lee, G., and Kim, K.-R.: Origins and chemical
448 characteristics of fine aerosols during the northeastern Asia regional experiment
449 (Atmospheric Brown Cloud-East Asia Regional Experiment 2005), *J. Geophys. Res.*,
450 112, D22S29, doi:10.1029/2006jd008210, 2007.
- 451 Li, M., Zhang, Q., Kurokawa, J., Woo, J. H., He, K. B., Lu, Z., Ohara, T., Song, Y., Streets, D.
452 G., Carmichael, G. R., Cheng, Y. F., Hong, C. P., Huo, H., Jiang, X. J., Kang, S. C., Liu,
453 F., Su, H., and Zheng, B.: MIX: a mosaic Asian anthropogenic emission inventory for
454 the MICS-Asia and the HTAP projects, *Atmos. Chem. Phys. Discuss.*, 2015, 34813-
455 34869, doi:10.5194/acpd-15-34813-2015, 2015.
- 456 Lim, S., Lee, M., Lee, G., Kim, S., Yoon, S., and Kang, K.: Ionic and carbonaceous
457 compositions of PM₁₀, PM_{2.5} and PM_{1.0} at Gosan ABC Superstation and their ratios as
458 source signature, *Atmos. Chem. Phys.*, 12, 2007-2024, doi:10.5194/acp-12-2007-2012,



- 459 2012.
- 460 Liu, Z., Wang, Y., Gu, D., Zhao, C., Huey, L. G., Stickel, R., Liao, J., Shao, M., Zhu, T., Zeng,
461 L., Liu, S.-C., Chang, C.-C., Amoroso, A., and Costabile, F.: Evidence of reactive
462 aromatics as a major source of peroxy acetyl nitrate over China, *Environ. Sci. Technol.*,
463 44, 7017-7022, doi:10.1021/es1007966, 2010.
- 464 Liu, X., Zhang, Y., Huey, L. G., Yokelson, R. J., Wang, Y., Jimenez, J. L., Campuzano-Jost, P.,
465 Beyersdorf, A. J., Blake, D. R., Choi, Y., St. Clair, J. M., Crouse, J. D., Day, D. A.,
466 Diskin, G. S., Fried, A., Hall, S. R., Hanisco, T. F., King, L. E., Meinardi, S., Mikoviny,
467 T., Palm, B. B., Peischl, J., Perring, A. E., Pollack, I. B., Ryerson, T. B., Sachse, G.,
468 Schwarz, J. P., Simpson, I. J., Tanner, D. J., Thornhill, K. L., Ullmann, K., Weber, R. J.,
469 Wennberg, P. O., Wisthaler, A., Wolfe, G. M., and Ziemba, L. D.: Agricultural fires in
470 the southeastern U.S. during SEAC4RS: Emissions of trace gases and particles and
471 evolution of ozone, reactive nitrogen, and organic aerosol, *Journal of Geophysical
472 Research: Atmospheres*, 121, 7383-7414, doi:10.1002/2016JD025040, 2016.
- 473 Maricq, M. M., and Szente, J. J.: Temperature-dependent study of the $\text{CH}_3\text{C}(\text{O})\text{O}_2 + \text{NO}$
474 reaction, *J. Phys. Chem.*, 100, 12380-12385, doi:10.1021/jp960792c, 1996.
- 475 Marley, N. A., Gaffney, J. S., White, R. V., Rodriguez-Cuadra, L., Herndon, S. E., Dunlea, E.,
476 Volkamer, R. M., Molina, L. T., and Molina, M. J.: Fast gas chromatography with
477 luminol chemiluminescence detection for the simultaneous determination of nitrogen
478 dioxide and peroxyacetyl nitrate in the atmosphere, *Rev. Sci. Instr.*, 75, 4595-4605,
479 doi:10.1063/1.1805271, 2004.
- 480 Mills, G. P., Sturges, W. T., Salmon, R. A., Bauguitte, S. J. B., Read, K. A., and Bandy, B. J.:
481 Seasonal variation of peroxyacetyl nitrate (PAN) in coastal Antarctica measured with a
482 new instrument for the detection of sub-part per trillion mixing ratios of PAN, *Atmos.
483 Chem. Phys.*, 7, 4589-4599, doi:10.5194/acp-7-4589-2007, 2007.
- 484 Muller, K. P., and Rudolph, J.: Measurements of peroxyacetyl nitrate in the marine boundary
485 layer over the Atlantic, *J. Atmos. Chem.*, 15, 361-367, doi:10.1007/BF00115405, 1992.
- 486 Nielsen, T., Samuelsson, U., Grennfelt, P., and Thomsen, E. L.: Peroxyacetyl nitrate in long-
487 range transported polluted air, *Nature*, 293, 553-555, doi:10.1038/293553a0, 1981.
- 488 NIER, 2016, Annual Report of Ambient Air Quality in Korea, 2015, National Institute of
489 Environmental Research, Incheon, Korea, p.10 (in Korean).
- 490 Ohara, T., Akimoto, H., Kurokawa, J., Horii, N., Yamaji, K., Yan, X., and Hayasaka, T.: An
491 Asian emission inventory of anthropogenic emission sources for the period 1980-2020,
492 *Atmos. Chem. Phys.*, 7, 4419-4444, doi:10.5194/acp-7-4419-2007, 2007.
- 493 Ram, K., Sarin, M. M., and Hegde, P.: Atmospheric abundances of primary and secondary
494 carbonaceous species at two high-altitude sites in India: Sources and temporal variability,
495 *Atmos. Environ.*, 42, 6785-6796, doi:10.1016/j.atmosenv.2008.05.031, 2008.
- 496 Ram, K., Sarin, M. M., and Tripathi, S. N.: Temporal trends in atmospheric $\text{PM}_{2.5}$, PM_{10} ,
497 elemental carbon, organic carbon, water-soluble organic carbon, and optical properties:
498 Impact of biomass burning emissions in the Indo-Gangetic Plain, *Environ. Sci. Technol.*,
499 46, 686-695, doi:10.1021/es202857w, 2012.
- 500 Randerson, J. T., van der Werf, G. R., Giglio, L., Collatz, G. J., and Kasibhatla, P. S.: Global
501 Fire Emissions Database, Version 3 (GFEDv3.1). Data set. Available on-line



- 502 [\[http://daac.ornl.gov/\]](http://daac.ornl.gov/) from Oak Ridge National Laboratory Distributed Active Archive
503 Center, Oak Ridge, Tennessee, USA. doi:10.3334/ORNLDAAC/1191, 2013.
- 504 Ridley, B. A., Shetter, J. D., Gandrud, B. W., Salas, L. J., Singh, H. B., Carroll, M. A., Hubler,
505 G., Albritton, D. L., Hastie, D. R., Schiff, H. I., Mackay, G. I., Karechi, D. R., Davis, D.
506 D., Bradshaw, J. D., Rodgers, M. O., Sandholm, S. T., Torres, A. L., Condon, E. P.,
507 Gregory, G. L., and Beck, S. M.: Ratios of peroxyacetyl nitrate to active nitrogen
508 observed during aircraft flights over the Eastern Pacific Oceans and continental United-
509 States, *J. Geophys. Res.*, 95, 10179-10192, doi:10.1029/JD095iD07p10179, 1990.
- 510 Roberts, J. M., Flocke, F., Chen, G., de Gouw, J., Holloway, J. S., Hübler, G., Neuman, J. A.,
511 Nicks, D. K., Nowak, J. B., Parrish, D. D., Ryerson, T. B., Sueper, D. T., Warneke, C.,
512 and Fehsenfeld, F. C.: Measurement of peroxyacetic nitric anhydrides (PANs)
513 during the ITCT 2K2 aircraft intensive experiment, *Journal of Geophysical Research:*
514 *Atmospheres*, 109, D23S21, doi:10.1029/2004JD004960, 2004.
- 515 Roberts, J. M., Marchewka, M., Bertman, S. B., Sommariva, R., Warneke, C., de Gouw, J.,
516 Kuster, W., Goldan, P., Williams, E., Lerner, B. M., Murphy, P., and Fehsenfeld, F. C.:
517 Measurements of PANs during the New England Air Quality Study 2002, *J. Geophys.*
518 *Res.*, 112, D20306, doi:10.1029/2007JD008667, 2007.
- 519 Rolph, G. D.: Real-time Environmental Applications and Display sYstem (READY) Website
520 (<http://ready.arl.noaa.gov>). NOAA Air Resources Laboratory, Silver Spring, MD. , 2012.
- 521 Saarikoski, S., Timonen, H., Saarnio, K., Aurela, M., Järvi, L., Keronen, P., Kerminen, V. M.,
522 and Hillamo, R.: Sources of organic carbon in fine particulate matter in northern
523 European urban air, *Atmos. Chem. Phys.*, 8, 6281-6295, doi:10.5194/acp-8-6281-2008,
524 2008.
- 525 Schrimpf, W., Müller, K. P., Johnen, F. J., Lienaerts, K., and Rudolph, J.: An optimized
526 method for airborne peroxyacetyl nitrate (PAN) measurements, *J. Atmos. Chem.*, 22,
527 303-317, doi:10.1007/bf00696640, 1995.
- 528 Shang, X., Lee, M., Han, J., Kang, E., Gustafsson, Ö., Chang, I.-S., and Han, J.-S.: Water-
529 soluble inorganic and organic compositions of submicron particles observed in Chinese
530 outflows at Gosan Climate Observatory in spring and fall, soon to be submitted.
- 531 Staudt, A. C., Jacob, D. J., Ravetta, F., Logan, J. A., Bachiochi, D., Sandholm, S., Ridley, B.,
532 Singh, H. B., and Talbot, B.: Sources and chemistry of nitrogen oxides over the tropical
533 Pacific, *J. Geophys. Res.*, 108, 8239, doi:10.1029/2002JD002139, 2003.
- 534 Stjern, C. W., Samset, B. H., Myhre, G., Bian, H., Chin, M., Davila, Y., Dentener, F.,
535 Emmons, L., Flemming, J., Haslerud, A. S., Henze, D., Jonson, J. E., Kucsera, T., Lund,
536 M. T., Schulz, M., Sudo, K., Takemura, T., and Tilmes, S.: Global and regional radiative
537 forcing from 20% reductions in BC, OC and SO₄ – an HTAP2 multi-model study,
538 *Atmos. Chem. Phys.*, 16, 13579-13599, doi:10.5194/acp-16-13579-2016, 2016.
- 539 Talukdar, R. K., Burkholder, J. B., Schmoltner, A. M., Roberts, J. M., Wilson, R. R., and
540 Ravishankara, A. R.: Investigation of the loss processes for peroxyacetyl nitrate in the
541 atmosphere: UV photolysis and reaction with OH, *J. Geophys. Res.*, 100, 14163-14173,
542 doi:10.1029/95JD00545, 1995.
- 543 Tereszchuk, K. A., Moore, D. P., Harrison, J. J., Boone, C. D., Park, M., Remedios, J. J.,
544 Randel, W. J., and Bernath, P. F.: Observations of peroxyacetyl nitrate (PAN) in the



- 545 upper troposphere by the Atmospheric Chemistry Experiment-Fourier Transform
546 Spectrometer (ACE-FTS), *Atmos. Chem. Phys.*, 13, 5601-5613, doi:10.5194/acp-13-
547 5601-2013, 2013.
- 548 Tilmes, S., Lamarque, J. F., Emmons, L. K., Kinnison, D. E., Ma, P. L., Liu, X., Ghan, S.,
549 Bardeen, C., Arnold, S., Deeter, M., Vitt, F., Ryerson, T., Elkins, J. W., Moore, F.,
550 Spackman, J. R., and Val Martin, M.: Description and evaluation of tropospheric
551 chemistry and aerosols in the Community Earth System Model (CESM1.2), *Geosci.*
552 *Model Dev.*, 8, 1395-1426, doi:10.5194/gmd-8-1395-2015, 2015.
- 553 Wang, B., Shao, M., Roberts, J. M., Yang, G., Yang, F., Hu, M., Zeng, L., Zhang, Y., and
554 Zhang, J.: Ground-based on-line measurements of peroxyacetyl nitrate (PAN) and
555 peroxypropionyl nitrate (PPN) in the Pearl River Delta, China, *Int. J. Environ. Anal.*
556 *Chem.*, 90, 548-559, doi:10.1080/03067310903194972, 2010.
- 557 Wang, Y., Zhang, Y., Hao, J., and Luo, M.: Seasonal and spatial variability of surface ozone
558 over China: contributions from background and domestic pollution, *Atmos. Chem. Phys.*,
559 11, 3511-3525, doi:10.5194/acp-11-3511-2011, 2011.
- 560 Yang, F., He, K., Ye, B., Chen, X., Cha, L., Cadle, S. H., Chan, T., and Mulawa, P. A.: One-
561 year record of organic and elemental carbon in fine particles in downtown Beijing and
562 Shanghai, *Atmos. Chem. Phys.*, 5, 1449-1457, doi:10.5194/acp-5-1449-2005, 2005.
- 563 Zhang, H., Xu, X., Lin, W., and Wang, Y.: Wintertime peroxyacetyl nitrate (PAN) in the
564 megacity Beijing: Role of photochemical and meteorological processes, *J. Environ. Sci.*,
565 26, 83-96, doi:10.1016/S1001-0742(13)60384-8, 2014.
- 566 Zhang, J. B., Xu, Z., Yang, G., and Wang, B.: Peroxyacetyl nitrate (PAN) and
567 peroxypropionyl nitrate (PPN) in urban and suburban atmospheres of Beijing, China,
568 *Atmos. Chem. Phys. Discuss.*, 11, 8173-8206, doi:10.5194/acpd-11-8173-2011, 2011.
- 569 Zhang, J. M., Wang, T., Ding, A. J., Zhou, X. H., Xue, L. K., Poon, C. N., Wu, W. S., Gao, J.,
570 Zuo, H. C., Chen, J. M., Zhang, X. C., and Fan, S. J.: Continuous measurement of
571 peroxyacetyl nitrate (PAN) in suburban and remote areas of western China, *Atmos.*
572 *Environ.*, 43, 228-237, doi:10.1016/j.atmosenv.2008.09.070, 2009.
- 573 Zhang, L., Jacob, D. J., Boersma, K. F., Jaffe, D. A., Olson, J. R., Bowman, K. W., Worden, J.
574 R., Thompson, A. M., Avery, M. A., Cohen, R. C., Dibb, J. E., Flock, F. M., Fuelberg, H.
575 E., Huey, L. G., McMillan, W. W., Singh, H. B., and Weinheimer, A. J.: Transpacific
576 transport of ozone pollution and the effect of recent Asian emission increases on air
577 quality in North America: An integrated analysis using satellite, aircraft, ozonesonde,
578 and surface observations, *Atmos. Chem. Phys.*, 8, 6117-6136, doi:10.5194/acp-8-6117-
579 2008, 2008.

580 **Tables**

581

582 Table 1. Chemical and meteorological characteristics of the four episodes.

	Episode 1	Episode 2	Episode 3	Episode 4
Type	Transport dominant	Transport dominant	Chemical transformation	Chemical transformation
Event	O ₃ export	O ₃ export	Haze	Haze
O ₃ (ppbv)	60.2 (80.6)	45.6 (62.8)	59.7 (78.9)	61.8 (87.5)
PAN (ppbv)	0.5 (0.9)	0.5 (0.8)	1.2 (2.0)	1.3 (2.4)
PM ₁₀ (µg/m ³)	57 (90)	38 (56)	69 (96)	100 (170)
SO ₂ (ppbv)	4.3 (12.9)	2.0 (4.4)	2.6 (5.4)	4.4 (9.5)
NO ₂ (ppbv)	3.7 (7.3)	6.2 (12.1)	6.2 (12.7)	6.1 (9.9)
Wind Speed (m/s)	13.5 (16.0)	9.5 (16.1)	6.6 (10.2)	5.0 (7.7)

583 *Measurements are given for the average with the maximum in the parenthesis.

584 **Figure Captions**

585

586 Figure 1. Temporal variations of measured species, including PAN, PM₁₀, O₃, NO₂, NO, SO₂,
587 and, CO, and meteorological parameters, including relative humidity, temperature,
588 and wind speed in fall 2010. Episodes 1–4, described in the main text, are shaded
589 in blue and yellow.

590 Figure 2. Diurnal variations in the concentrations of O₃, NO₂, PAN, and PM₁₀, measured at
591 GCO in the fall of 2010 (5 min data of O₃, NO₂, 2 min data of PAN, and 1 h data of
592 PM₁₀).

593 Figure 3. Comparison of O₃ with the PAN daily maxima. The time when the daily maximum
594 appears is classified as daytime (08–18 h) and nighttime (the rest) based on the
595 time of sunrise and sunset. Numerals indicate the days.

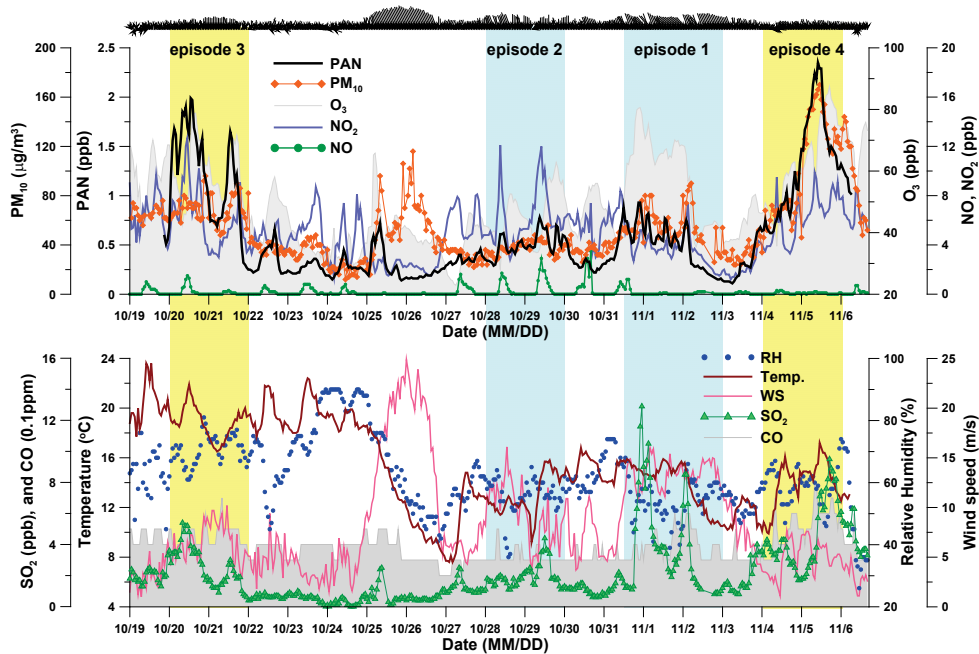
596 Figure 4. The three-day NOAA HYSPLIT backward trajectories of air masses for every one
597 hour observed at GCO during episode 1 (Oct. 31 to Nov. 2), episode 2 (Oct. 28–29),
598 episode 3 (Oct. 20–21), and episode 4 (Nov. 4–5). They are colored according to
599 the level of (a) PAN, (b) O₃, and, (c) NO₂ at GCO at the time of the trajectory
600 initialization. The trajectories north of 50°N are not shown.

601 Figure 5. Correlations among PAN, PM₁₀, O₃, and carbonaceous compounds in PM_{2.5}: (a) O₃
602 and PAN, (b) NO₂ and PAN, and (c) O₃ and PAN. The red lines in (a) and (b)
603 represent linear regression for episode 4.

604 Figure 6. Correlations among PAN, K⁺ ion of PM_{1.0}, and carbon components of PM_{2.5} for
605 three cases: (a) PM₁₀ and PAN, (b) PM_{2.5} OC and PAN, (c) PM_{2.5} EC and PAN, (d)
606 PM_{2.5} OC and EC, (e) PM_{1.0} K⁺ and PAN, and (f) PM_{1.0} K⁺ and PM_{2.5} OC. The red
607 lines represent linear regression for episode 4.

608 Figure 7. Comparison between the observed and calculated (a) PAN and (b) O₃
609 concentrations by CAM-chem model.

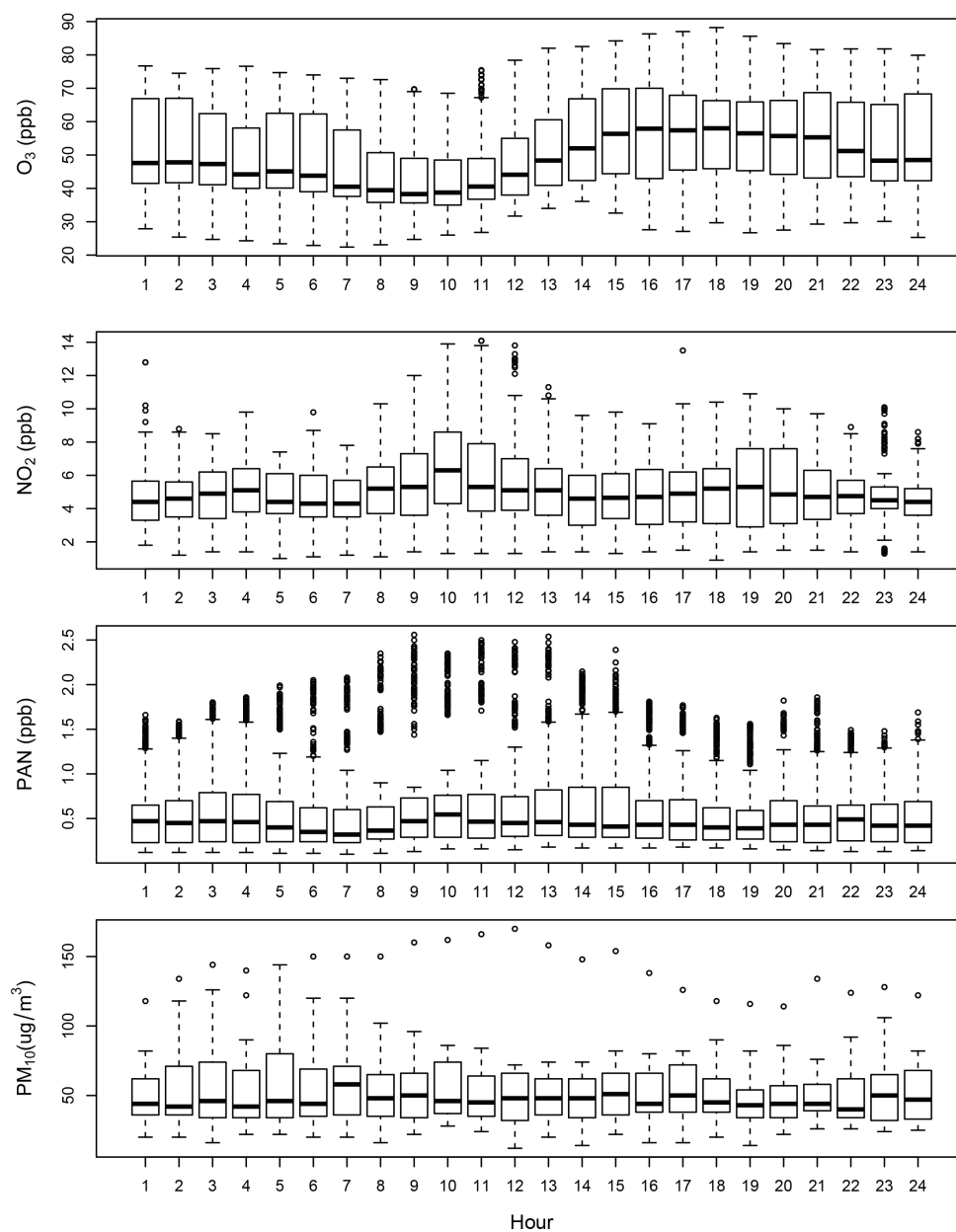
610



611

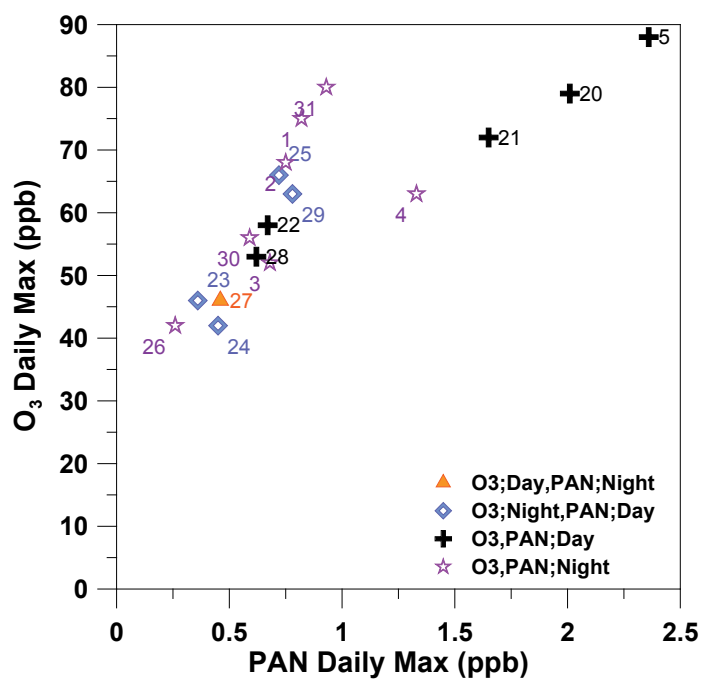
612

613 Figure 1.



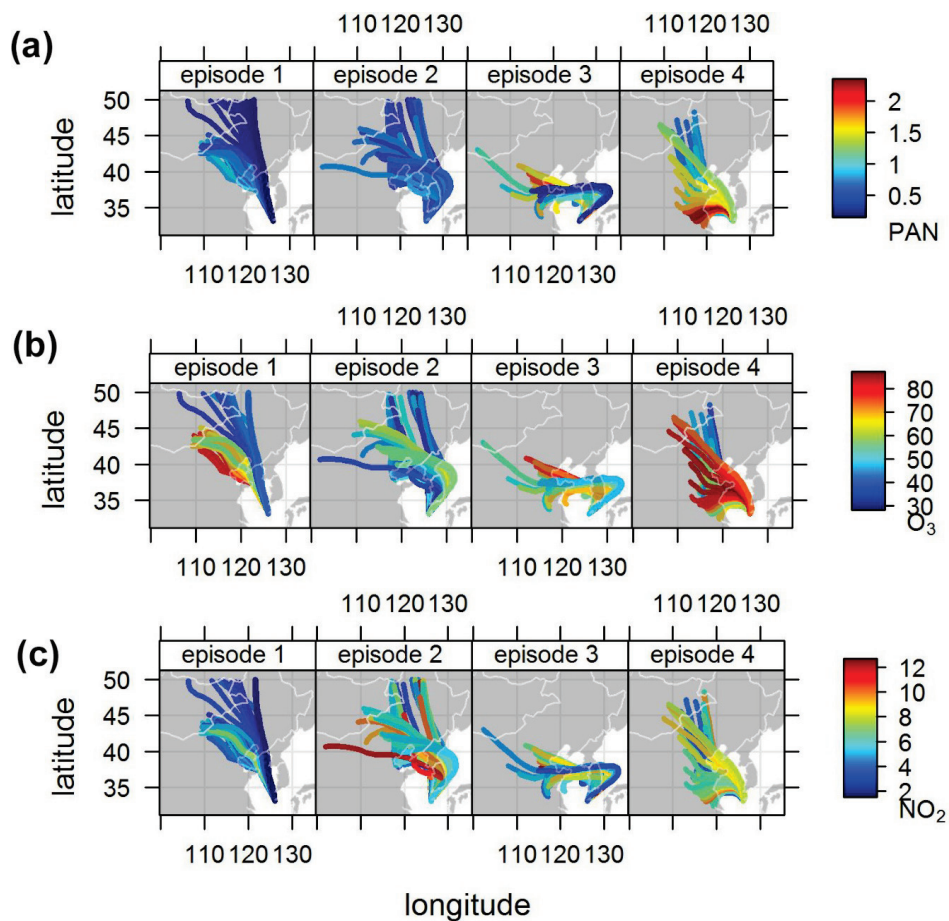
614

615 Figure 2.



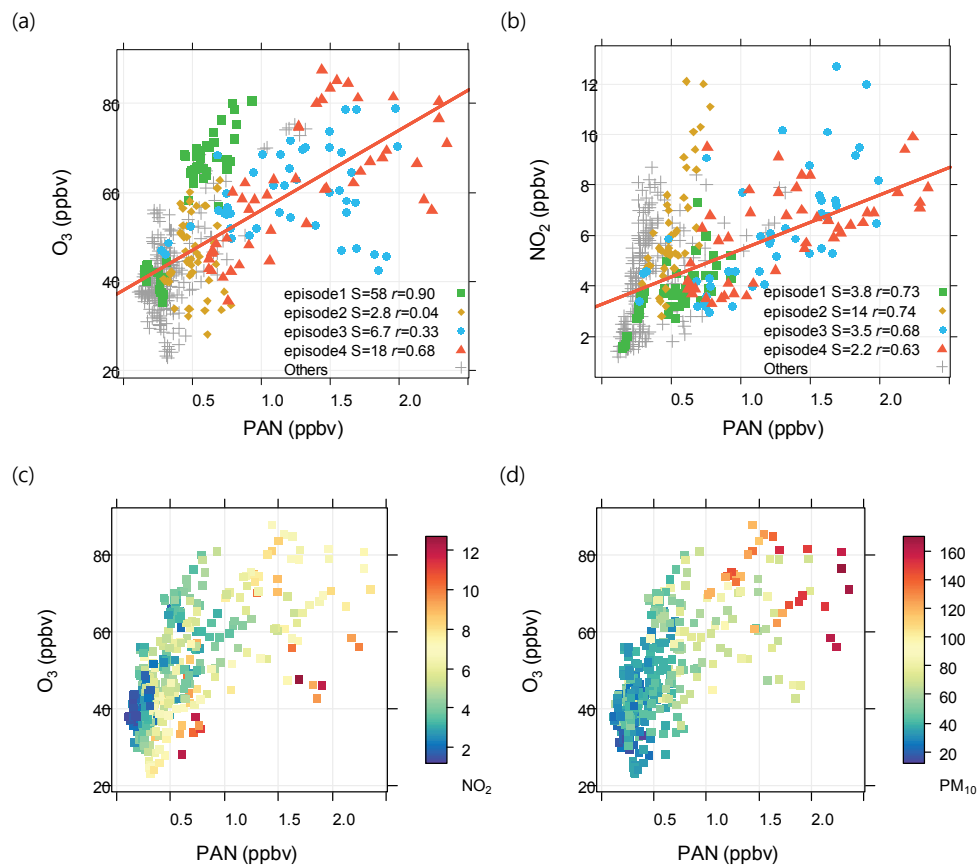
616

617 Figure 3.



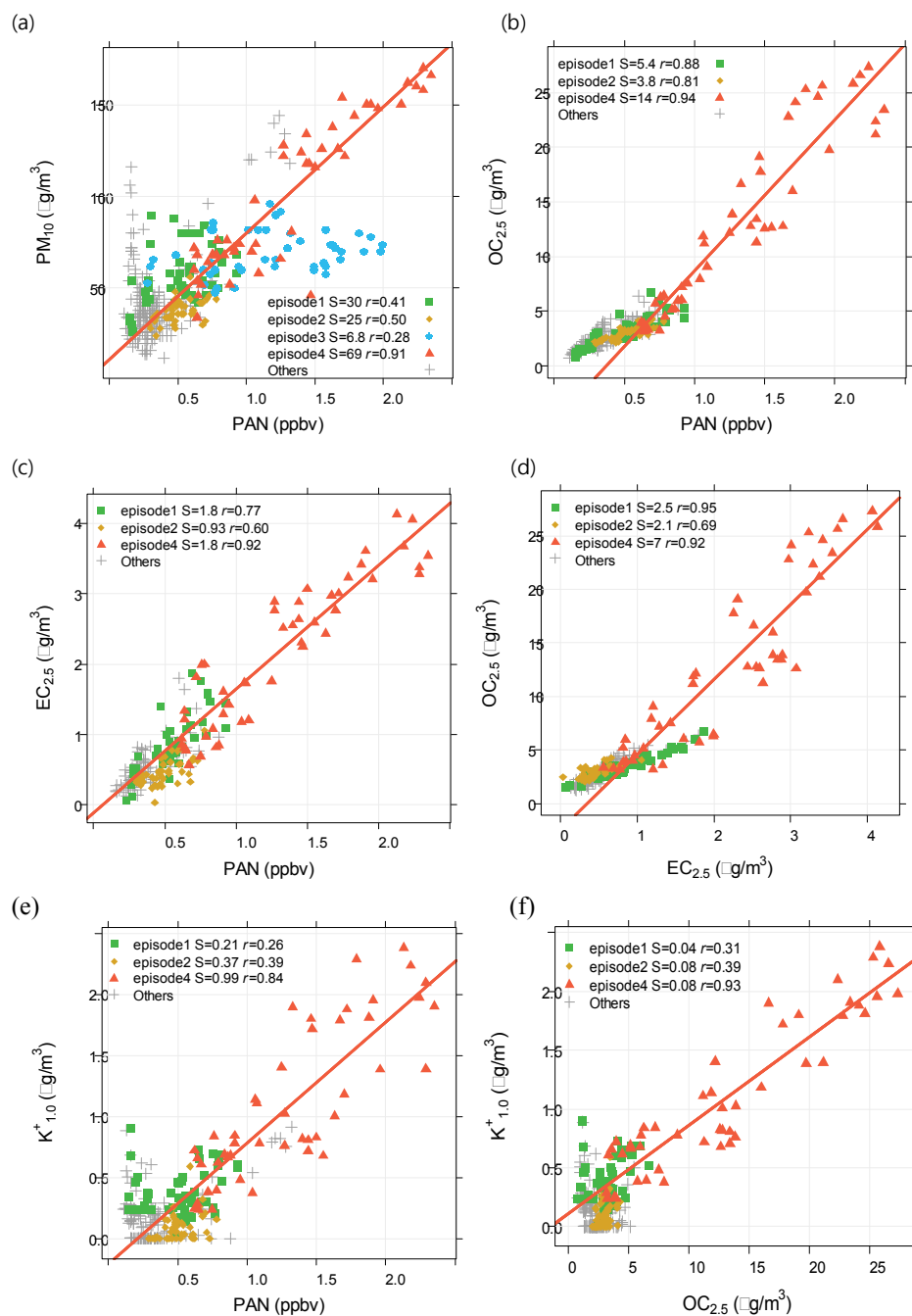
618

619 Figure 4.



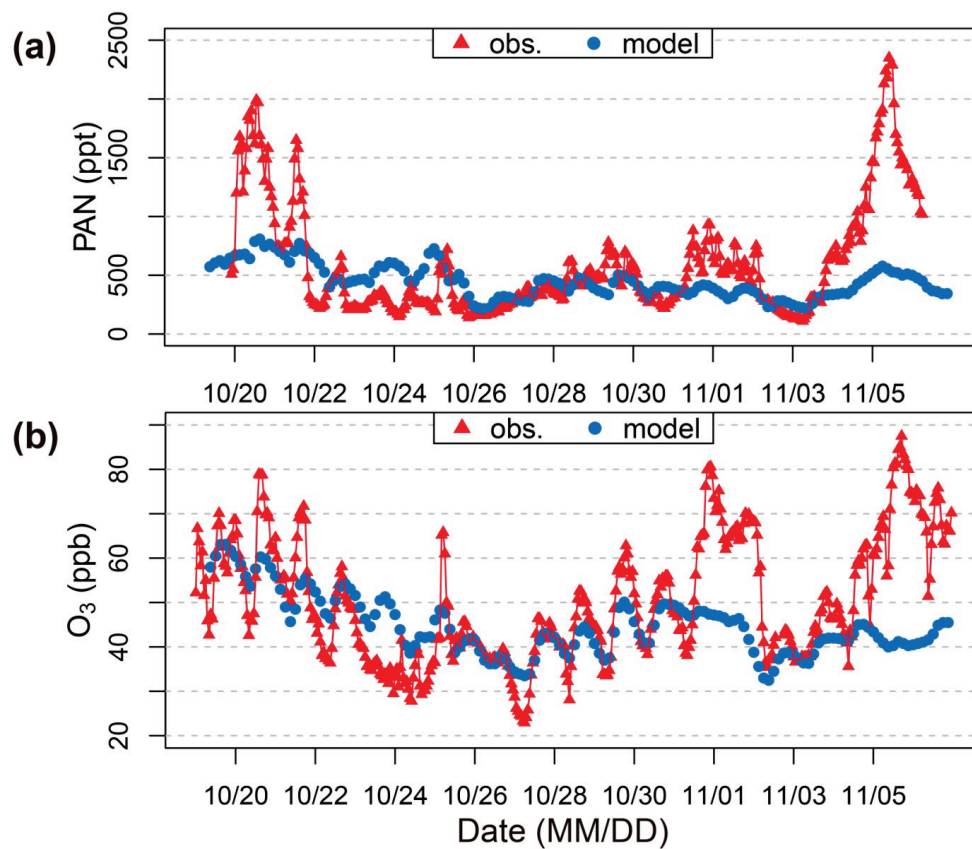
620

621 Figure 5.



622

623 Figure 6.



624

625 Figure 7.