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3 **Decoupling peroxyacetyl nitrate from ozone in Chinese outflows observed**
4 **at Gosan Climate Observatory**
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26 **Abstract**

27 We measured peroxyacetyl nitrate (PAN) and other reactive species such as O₃, NO₂, CO,
28 and SO₂ with aerosols including mass, organic carbon (OC), and elemental carbon (EC) in
29 PM_{2.5} and K⁺ in PM_{1.0} at Gosan Climate Observatory in Korea (33.17°N, 126.10°E) during
30 October 19 to November 6, 2010. PAN was determined through fast gas chromatography
31 with luminol chemiluminescence detection at 425 nm every 2 min. The PAN mixing ratios
32 ranged from 0.1 (detection limit) to 2.4 ppbv with a mean of 0.6 ppbv. For all measurements,
33 PAN was unusually better correlated with PM_{2.5} (Pearson correlation coefficient, $\gamma = 0.79$)
34 than with O₃ ($\gamma = 0.67$). In particular, the O₃ level was highly elevated with SO₂ at midnight,
35 along with a typical midday peak when air was transported rapidly from the Beijing areas.
36 The PAN enhancement was most noticeable during the occurrence of haze under stagnant
37 conditions. In Chinese outflows slowly transported over the Yellow Sea, PAN gradually
38 increased up to 2.4 ppbv at night, in excellent correlation with a concentration increase of
39 PM_{2.5} OC and EC, PM_{2.5} mass, and PM_{1.0} K⁺. The high K⁺ concentration and OC/EC ratio
40 indicated that the air mass was impacted by biomass combustion. This study highlights PAN
41 decoupling with O₃ in Chinese outflows and suggests PAN as a useful indicator for
42 diagnosing continental outflows and assessing their perturbation on regional air quality in
43 northeast Asia.

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45 Key words: PAN, O₃, PM_{2.5}, Chinese outflow, Haze, Biomass combustion

46 1. Introduction

47

48 At the surface, ozone is primarily photochemically produced, and the contribution from the
49 stratosphere is generally small. Ozone is formed through reactions of various precursors such
50 as CO, CH₄, volatile organic compounds (VOCs), and NO_x (e.g., Brasseur et al., 1999; Jacob,
51 2000; Nielsen et al., 1981). Likewise, peroxyacetyl nitrate (PAN) is a secondary product of
52 urban air pollution and a significant oxidant in the atmosphere (e.g., Hansel and Wisthaler,
53 2000; La Franchi et al., 2009; Lee et al., 2012; Liu et al., 2010; Roberts et al., 2007). PAN is
54 solely produced by the photochemical reaction between the peroxyacetyl radical and nitrogen
55 dioxide, and the peroxyacetyl radical is derived from the OH oxidation or photolysis of
56 VOCs such as acetaldehyde, methylglyoxal, and acetone (e.g., Fischer et al., 2014; La
57 Franchi et al., 2009; Lee et al., 2012). For this reason, PAN is a very useful indicator of
58 photochemical air pollution. As thermal decomposition is a major PAN sink in the
59 troposphere (Beine et al., 1997; Jacob, 2000; Kenley and Hendry, 1982; Talukdar et al., 1995),
60 the lifetime of PAN depends on temperature. For example, the PAN lifetime is ~5 years at
61 -26°C and 1 h at 20°C (Fischer et al., 2010; Zhang et al., 2011). At high altitudes above ~7
62 km, photolysis becomes the most important loss process for PAN (Talukdar, et al., 1995).
63 Because of low solubility, PAN is not prone to atmospheric removal, thereby being more
64 efficiently transported to the free troposphere (e.g., Zhu et al., 2017). Thus, PAN can be an
65 indicator of NO_y concentration in the free troposphere and a guide for the long-range
66 transport of NO_x in remote regions (Jacob, 1999).

67 In the past decades, PAN was measured not only in urban areas (Aneja et al., 1999;
68 Gaffney et al., 1999; Grosjean et al., 2002; Lee et al., 2008; Tanimoto et al., 1999; Zhang et
69 al., 2014) but also in background regions (Fischer et al., 2011; Kanaya et al., 2007; Lee et al.,

70 2012; [Tanimoto et al., 2002](#)), onboard aircraft ([Tereszczuk et al., 2013](#)), and ships ([Roberts et](#)
71 [al., 2007](#)). PAN concentrations were in the range of a few ppbv in urban areas close to VOCs
72 and NO_x sources ([Lee et al., 2008](#); [Zhang et al., 2011](#)). In [the most](#) remote regions, PAN
73 [concentrations](#) were generally in the range of a few pptv ([Gallagher et al., 1990](#); [Mills et al.,](#)
74 [2007](#); [Muller and Rudolph, 1992](#); [Staudt et al., 2003](#)).

75 [Although NO_x concentration has recently declined in China](#) ([Gu et al., 2013](#); [Liu et al.,](#)
76 [2016a](#); [Krotkov et al., 2016](#)), NO_x and VOCs have gradually increased in East Asia,
77 particularly China during the last couple of decades ([Akimoto, 2003](#); [Liu et al., 2010](#); [Ohara](#)
78 [et al., 2007](#); [Zhao et al., 2013](#)). It led to an increase in the concentrations of photochemical
79 byproducts such as PAN and O₃ not only in East Asia ([Liu et al., 2010](#); [Wang et al., 2010](#);
80 [Zhang et al., 2009](#); [Zhang et al., 2011](#); [Zhang et al., 2014](#)) but also in North America ([Fischer](#)
81 [et al., 2010](#); [Fischer et al., 2011](#); [Jaffe et al., 2007](#); [Zhang et al., 2008](#)). These results were
82 also demonstrated by the GEOS-Chem model ([Zhang et al., 2008](#)). In addition to urban
83 plumes, PAN was reported to be enhanced by biomass combustion ([Alvarado et al., 2010](#);
84 [Coheur et al., 2007](#); [Zhu et al., 2015](#); [Zhu et al., 2017](#)), such as open burning and use of
85 biofuel, which is used to take place often in China after crop harvesting ([Cao et al., 2006](#);
86 [Duan et al., 2004](#)). [Recent satellite studies have also observed the increased PAN in plumes](#)
87 [associated with anthropogenic emissions in eastern China and boreal fires in Siberia](#) ([Zhu et](#)
88 [al., 2015](#); [Zhu et al., 2017](#)). In this context, PAN is a useful tracer for estimating the impact of
89 Chinese outflows on regional air quality in the northern Pacific region.

90 Gosan Climate Observatory (GCO) is an ideal place to monitor Asian outflows and their
91 transformation and to estimate their impact on air quality over the northern Pacific region
92 ([Lee et al., 2007](#); [Lim et al., 2012](#)). In the present study, PAN was first measured
93 continuously at GCO to characterize its variation and source in relation to O₃ and to

94 understand the influence of Chinese outflows on the regional air quality.

95

96 **2. Experiments**

97

98 PAN measurements were conducted at GCO (33.17°N, E126.10°E) on Jeju Island from
99 October 19 to November 6, 2010. GCO is located on a cliff at the western edge of Jeju Island.
100 PAN was determined through fast gas chromatography (GC) with luminol
101 chemiluminescence detection, which is described in detail elsewhere (Gaffney et al., 1998;
102 Lee et al., 2008; Marley et al., 2004). Here, we briefly describe the measurement method.

103 Ambient air was pumped through a 1.6-m PFA tubing (1/4 inch outer diameter) from the
104 roof of the two-story container into a six-port injection valve (Cheminert C22, Valco
105 Instruments (Houston, TX, USA)) at 100 ml/min controlled by Mass flow controller (Lee et
106 al., 2012; Lee et al., 2008). The residence time of the inlet was less than 2 seconds. PAN and
107 NO₂ (and peroxypropyl nitrate (PPN) if present) were separated along a 10-m capillary GC
108 column (DB-1, J&W Scientific, Folsom, CA, USA), whose end was connected to a luminol
109 cell where the column effluent reacted with luminol, giving off luminescent light (Lee et al.,
110 2008; Lee et al., 2012). The concentrations of PAN and other species were determined from
111 the chemiluminescence signals detected by a gated photon counter (HC135-01, Hamamatsu,
112 Bridgewater, NJ, USA) at 425 nm, which was set at 800 V and operated at room temperature
113 (Gaffney et al., 1998; Lee et al., 2012; Lee et al., 2008).

114 PAN was calibrated against standards synthesized by the nitration of peracetic acid in n-
115 tridecane (Gaffney et al., 1984; Gregory, 1990; Lee et al., 2008). A few microliter aliquots of
116 standard solution were injected through an injection valve and then mixed with zero air
117 (99.999 %) in a 5 L Tedlar bag. After being left for a few minutes for equilibrium, it was

118 injected into GC-luminol instrument and NO_x chemiluminescence instrument with a
119 molybdenum converter (42C, Thermo Electron Corporation, Franklin, MA, USA). The
120 calibration was completed within 5 minutes to prevent thermal decomposition of the PAN
121 (Kourtidis et al., 1993; Lee et al., 2008). These calibration procedures were carried out on the
122 assumption that the PAN was completely converted to NO in the molybdenum converter. The
123 detection limit of PAN defined by 3 σ of the lowest standard was no greater than 100 pptv
124 (Lee et al., 2008). The overall measurement uncertainty and precision was estimated to be 16 %
125 and 5%, respectively (Lee et al., 2012). NO_x instrument was calibrated with NO standard gas.

126 Gaseous species including O₃, NO, NO₂, CO, and SO₂ were measured by UV absorption,
127 chemiluminescence with a molybdenum converter, non-dispersive infrared, and pulse UV
128 fluorescence method, respectively (NIER, 2016a). The measurements were made in
129 compliance with guidelines for installation and operation of air pollution monitoring network
130 (NIER, 2016b). Calibration was conducted before and after the experiment, following the
131 regular checkup procedure. Detection limits of O₃, NO_x, CO, and SO₂ are 2 ppb, 0.1 ppb, 50
132 ppb, 0.1 ppb, respectively (NIER, 2016b). It should be noted that NO₂ concentration reported
133 in the present study is actually the sum of NO₂ and NO_z species due to well-known positive
134 artifact of molybdenum convertor. PAN is one of the major NO_z species and the ratio of PAN
135 to NO₂ was 12 ± 7 % for the whole measurements.

136 Aerosol species, including PM_{2.5} mass and PM_{2.5} OC and EC were measured and recorded
137 along with meteorological parameters (relative humidity, temperature, and wind direction and
138 speed). Water-soluble ions of PM_{1.0} were collected by a particle-into-liquid sampler (PILS)
139 and analyzed by ion chromatography. The detailed results of the aerosol measurements can be
140 found in Shang et al. (2017).

141 For the air parcel at 850 m a.s.l., the three-day backward trajectories were calculated every

142 hour using NOAA Air Resources Laboratory (ARL) Hybrid Single-Particle Lagrangian
143 Integrated Trajectory (HYSPLIT) model (version 4) (Draxler and Rolph, 2012; Rolph, 2012,
144 <http://www.arl.noaa.gov/ready/hysplit4.html>). In addition, O₃ and PAN concentrations were
145 calculated using a global chemistry model, the Community Atmosphere Model with
146 Chemistry (CAM-Chem), a component of the Community Earth System Model (CESM)
147 (Lamarque et al., 2012; Tilmes et al., 2015). The CAM-chem results shown here follow the
148 configuration used for the HTAP2 (Hemispheric Transport of Air Pollution, Phase 2)
149 intercomparison (e.g., Stjern et al., 2016). CAM-chem is nudged to observed meteorology
150 (GEOS-5) to reproduce the actual period of the observations (Oct 2010). The emissions used
151 in the model are the HTAP2 inventory (Janssens-Maenhout, et al., 2015), which include the
152 "MIX" Asian emissions inventory. Biomass burning emissions are from the Global Fire
153 Emissions Database (GFED3) (Randerson et al., 2013).

154

155 **3. Results**

156

157 In the present experiments, PAN concentrations range from 0.1 to 2.4 ppbv, with an
158 average of 0.6 ppbv. This mean value is lower than those observed in other Asian megacities:
159 Beijing (1.41 ppb in the summer), Pearl River Delta region (1.32 ppb in the summer), and
160 Seoul (0.8 ppb in the early summer); similar to those of suburban areas in China, e.g.,
161 Lanzhou (0.76 ppb in the summer); and higher than those of urban and rural sites in Japan,
162 e.g., Tokyo (up to 0.6 ppb in the fall), Rishiri Island (~0.5 ppb in spring) or in the western
163 coast of the US, e.g., Sacramento (0.45 ppb in the summer), Mt. Bachelor (0.144 ppb in the
164 spring and early summer), off the western coast of the US (0.65 ppb in the spring), and over
165 the remote North Pacific (total PAN < 0.3 ppb in spring) (Bertram et al., 2013; Fischer et al.,

166 2011; La Franchi et al., 2009; Lee et al., 2008; Roberts et al., 2004; Tanimoto et al., 1999;
167 Tanimoto et al., 2002; Wang et al., 2010; Zhang et al., 2009; Zhang et al., 2011). Because the
168 PAN lifetime is greatly dependent on temperature, its concentration decreases with increasing
169 distance from the source regions. The PAN concentrations calculated in this study thus lie in-
170 between the levels for the East Asian megacities and the northern Pacific. The distributions of
171 all measured species, including PAN and O₃, are presented in Fig. 1. In particular, there are
172 several periods characterized by high concentrations of PAN, O₃, and PM_{2.5}. In terms of PAN,
173 four periods are particularly interesting (Fig. 1). High O₃ concentrations were observed
174 during October 31–November 2 [episode 1] but did not coincide with high PAN
175 concentrations. During October 28–29 [episode 2], NO₂ was noticeably increased. In
176 comparison, PAN and O₃ concentrations were both high during October 20–21 [episode 3]
177 and November 4–5 [episode 4]. Episodes 3 and 4 are characterized by haze, while episodes 1
178 and 2 are characterized by urban influence in the Korean and Beijing outflows, respectively.
179 Haze is reported by Korea Meteorological Administration (KMA) as a meteorological
180 phenomenon when visibility is 1~10 km and relative humidity is less than 75 %.

181 In the present study, PAN correlates reasonably well with O₃ ($\gamma = 0.67$) and even better
182 with PM_{2.5} ($\gamma = 0.79$). In general, O₃ and PAN exhibit typical diurnal variation with a
183 maximum recorded in the afternoon, which results in a good correlation between the two
184 (Brasseur et al., 1999; Gaffney et al., 1999; Ridley et al., 1990; Schrimpf et al., 1995; Wang
185 et al., 2010). In this study, however, the O₃ peak was often found in the early morning and
186 late afternoon for several days (Fig. 1). Observing the diurnal variations in the entire PAN
187 concentration measurement set (Fig. 2), the maximum was clearly recorded in the morning
188 with the highest outliers, which is rather similar to that of PM_{2.5}. The diurnal pattern of NO₂
189 shows little variation, even though its concentrations were increased in the morning along

190 with PAN. This first measurement of PAN at GCO reveals that PAN is not always coupled
191 with O₃, which was not typically observed at remote sites in previous studies (e.g., Fischer et
192 al., 2010; Lee et al., 2012).

193

194 **4. Discussion**

195 **4.1. Decoupling of PAN from O₃**

196

197 To examine the detailed mechanism of the decoupling of PAN from O₃, the daily
198 maximum concentrations of PAN and O₃ were further explored. The recorded daily PAN
199 maxima were generally in good correlation with O₃, albeit the relationship did not seem to
200 hold at high concentrations of PAN and O₃ (Fig. 3). The daily maxima were then categorized
201 into four groups according to the time when each O₃ and PAN maximum was recorded: “O₃
202 day-PAN day”, “O₃ day-PAN night”, “O₃ night-PAN day”, and “O₃ night-PAN night”. The
203 day interval started from 08:00 and ended at 18:00 (local time), based on the times of sunrise
204 and sunset during the experiment period. While the high PAN concentrations were associated
205 with the “O₃ day-PAN day” group (cross symbols in Fig. 3), the enhanced O₃ concentration
206 was recorded in the “O₃ night-PAN night” group (star symbols in Fig. 3). The “O₃ night-PAN
207 night” group unexpectedly held more data points than the “O₃ day-PAN day” group, even
208 though the “O₃ night-PAN night” group concentrations were lower (Fig. 3). In addition, there
209 were several days classified in the “O₃ night-PAN day” (marked by diamond) and “O₃ day-
210 PAN night” groups, but with less frequency and lower concentrations. These results indicate
211 that the decoupling of PAN from O₃ was primarily **due to the elevated concentrations of O₃**
212 **and PAN at night. While PAN reached the maximum during the day on Oct 20 and Nov 5,**
213 **their concentrations were increased from the previous day through the night.** The four high

214 PAN and O₃ episodes identified in this study fall under the category of “O₃ night-PAN night”
215 or “O₃ day-PAN day”. These two cases will be further examined to identify the chemical and
216 physical processes responsible for PAN being decoupled from O₃, instead of being coupled
217 with PM_{2.5}. The overall characteristics of the four episodes are summarized in Table 1.

218

219 **4.2. Export of O₃ from Asian continents (episodes 1 & 2)**

220

221 High O₃ concentrations were encountered around midnight on three consecutive days from
222 October 31 to November 2 (episode 1), during which SO₂ reached its maximum
223 concentration (Fig. 1). The backward trajectories of air masses revealed that air passed
224 through the Beijing area during this period (Fig. 4). The strong wind (13.5 m/s on average)
225 implies that it would take about a day for air mass leaving Beijing area to arrive at GCO.
226 The recorded O₃ maximum (80.6 ppbv) was concurrent with the PAN maximum (0.9 ppbv)
227 around midnight on November 1st (Fig. 1). All these results indicate that the air was heavily
228 influenced by outflow from the Beijing area, as previously hypothesized (Lim et al., 2012),
229 and that the nighttime enhancement of O₃ and PAN with SO₂ resulted from the fast transport
230 of urban plumes from China.

231 In previous studies, the nighttime enhancement of O₃ was observed at GCO (e.g., Lee et al.,
232 2007) in association with pollutant-laden air coming from Beijing. Similarly, Banta et al.
233 (1998) pointed out that the evening O₃ maximum was due to long-range transport of O₃ from
234 nearby urban areas. Wang et al. (2011) reported that the O₃ lifetime was about two days in
235 East China during the summer, which is sufficient for O₃ to travel to GCO but not for PAN
236 due to its short lifetime. Therefore, the nighttime maximum of O₃ can be attributed to the
237 export of O₃ from megacities in China, causing PAN to be decoupled from O₃. Because the
238 overall correlation between O₃ and PAN was the best with the highest $\Delta O_3 / \Delta PAN$ among all

239 cases discussed in this study (Fig. 5a), episode 1 likely represents an event of rapid transport
240 from the Beijing area.

241 Another night maximum of O₃ was recorded on October 29. Note that NO_x was highly
242 elevated with the lowest SO₂ concentrations during October 28–29 (episode 2) (Fig. 1). In
243 episode 2, O₃ concentrations were much lower and poorly correlated with those of PAN,
244 compared to episode 1. Instead, PAN was best correlated with NO₂ with the highest
245 $\Delta\text{NO}_2/\Delta\text{PAN}$ among all episodes (Fig. 5a, b). In this case, air masses passed through the
246 Korean Peninsula, carrying low O₃ being titrated by high NO_x (Brasseur et al., 1999;
247 Jacobson, 2005). These two episodes illustrate the export of urban plumes in northeast Asia
248 region, which are distinguished by relative enhancement of reactive gases including O₃, PAN
249 and NO_x, depending on the origin and aging of air masses.

250

251 **4.3. PAN enhancement upon occurrence of haze (episodes 3 & 4)**

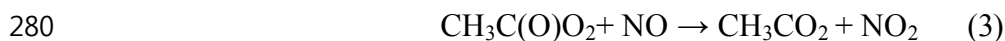
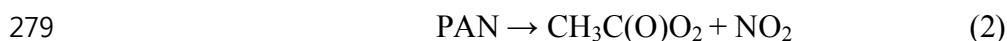
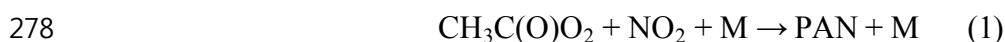
252

253 In this study, two haze events were observed in the very beginning (October 20–21;
254 episode 3) and the end of the study period (November 4–5; episode 4). As the nighttime O₃
255 peak was attributed to the transport from nearby urban areas to Jeju Island, the two haze
256 episodes were also observed in association with continental outflows. The first haze event
257 occurred on October 18th and lingered until October 21st, during which O₃ concentrations
258 were gradually elevated. A second peak was recorded around midnight of October 19th and
259 20th, and the maximum was reached in the afternoon of October 20th (Figs. 1 and 3). In this
260 episode, the maximum concentrations of O₃ and PAN were 78.9 ppbv and 2.0 ppbv,
261 respectively, on October 20th, when the highest NO₂ concentration (12.7 ppbv) was observed
262 under low wind speed (6.6 m/s daily average). The air mass trajectories suggest the influence

263 of the Korean Peninsula, particularly the Seoul metropolitan area, in addition to East China
264 (Fig. 4).

265 In the second haze event (episode 4), an air mass was slowly transported from East China,
266 including the Jiangsu province, under stagnant condition which was developed by an
267 anticyclone system (Fig. 4). We measured the highest concentrations of all aerosol species
268 including the PM_{2.5} mass as well as PAN and O₃, which were 156 µg/m³, 2.4 ppbv, and 87.5
269 ppbv, respectively. Other reactive gases such as CO, SO₂, and NO₂ were also highly elevated.
270 Note that PAN and O₃ gradually increased through the night, leading to a nighttime maximum
271 of both species on November 4th. It is likely that the pre-formed PAN and O₃ were
272 continuously transported into Gosan at night.

273 PAN is formed through the reaction of the peroxyacetyl radical and nitrogen dioxide (Eq. 1)
274 and decomposed at high temperature (Eq. 2), returning these radicals. Unless the NO
275 concentration is high (Eq. 3), the peroxyacetyl radical recombines with NO₂, producing PAN.
276 Thus, the total lifetime of PAN depends on the NO₂/NO ratio and temperature (Eq. 4)
277 (Brasseur et al., 1999).



$$281 \quad T_{eff} = T_d \left(1 + \frac{k_1[\text{NO}_2]}{k_2[\text{NO}]} \right) [\text{sec}^{-1}] \quad (4)$$

282 where T_d and T_{eff} indicate the lifetime against decomposition and the effective lifetime of
283 PAN (Brasseur et al., 1999). The effective lifetime of PAN was estimated through Eq. 4 using
284 the rate constants proposed by Brasseur et al. (1999), Jacobson (2005), and Maricq and
285 Szente (1996).

286 During the haze event, NO was close to the detection limit, while NO₂ was greatly
287 enhanced. Owing to the high NO₂/NO ratio, the effective lifetime of PAN increased by $57 \pm$

288 14 times; this possibly contributed to the gradual increase in PAN through the night on
289 November 4th. For this estimation, PAN concentration was subtracted from the measured
290 NO₂ concentration, considering the positive artifact by molybdenum converter in NO₂
291 measurement. Fischer et al. (2014) also reported that, at night, PAN can be produced from the
292 reaction of acetaldehyde with the nitrate radical.

293 Besides PM_{2.5} mass, PAN was also well correlated with PM_{2.5} OC and EC not only during
294 this haze episode but also during the entire measurement period (Fig. 6a and b). Furthermore,
295 the enhancement of PAN was concurrent with that of OC and K⁺, resulting in excellent
296 correlation between them (Fig. 6b and d). In fact, the ΔOC/ΔEC ratio of episode 4 was much
297 higher (7) than those of the other episodes (~2.5) (Fig. 6c). The fraction of PM_{2.5} against
298 PM₁₀ was also the highest in this episode, indicating significant contribution of secondary
299 aerosols. These observations suggest that air masses were affected by biomass combustion
300 (e.g., Ram et al., 2008, 2012; Saarikoski et al., 2008).

301 According to previous studies, PAN can be produced in plumes through biomass
302 combustion (Alvarado et al., 2010; Coheur et al., 2007; Liu et al., 2016b; Tereszchuk et al.,
303 2013). In northeast China, open burnings related to agricultural activities frequently occur
304 during the spring and fall (Duan et al., 2004; Yang et al., 2005). Kudo et al. (2014) also
305 reported that, upon burning crop residue in Yangtze region, the levels of oxygenated VOCs
306 were elevated together with NO_x. In addition, biofuel is used for cooking and heating and as
307 an energy source in China's industry (Cao et al., 2006).

308 Therefore, PAN is likely to increase when haze occurs and fine aerosols are transformed as
309 air masses carrying combustion emissions are slowly transported from China over the Yellow
310 Sea. Additionally, the results of this study imply that PAN can be used as a robust tracer for
311 continental outflows in northeast Asia, to identify transport- and chemical transformation-

312 dominant regimes. In a transport-dominant regime, O₃ export was distinguished by the
313 highest levels of primary gaseous species such as SO₂ and relatively low levels of PAN. In
314 contrast, fine aerosol species were enhanced in a chemical transformation regime, leading to
315 haze events with relatively more enhanced PAN compared to O₃.

316 Finally, the measured O₃ and PAN concentrations were compared to results from a global
317 chemistry model CAM-Chem. In the model simulation, O₃ and PAN were highly
318 underestimated during the episodes observed in Chinese outflows, although the variation
319 around average level of O₃ and PAN was well captured (Fig. 7). The elevated PAN
320 concentration was underestimated in the model (Oct 20–21 and Nov 4–5), especially when air
321 was impacted by biomass combustion. The timing of the O₃ diurnal variability was captured
322 by the model, although the magnitude of the variation was underestimated. These results
323 reveal that the current understanding of Chinese outflow is still not sufficient, thereby causing
324 uncertainty in estimating its effect on air quality in the northwestern Pacific Rim.

325

326 5. Conclusions

327

328 The first measurements of PAN, reactive gases, and aerosol species were conducted at
329 GCO during October 19 to November 6, 2010. The average concentration of PAN was 0.6
330 ppbv with a maximum of 2.4 ppbv, which was lower than those in major cities in East Asia
331 but much higher than the background concentrations in other regions. Although the hourly
332 concentrations of PAN and O₃ were well correlated ($\gamma = 0.67$), the comparison of the daily
333 maxima of PAN and O₃ highlighted that they were not proportionally enhanced. That is,
334 either PAN was relatively more elevated than O₃ or the highest O₃ was associated with low
335 levels of PAN. Unexpectedly, both PAN and O₃ often reached their maxima at night. In this

336 study, these high concentrations were all encountered in association with continental outflows,
337 where PAN was decoupled from O₃ and better correlated with the PM_{2.5} mass ($\gamma = 0.79$) than
338 with O₃. Thus, two high-O₃ and two high-PAN events were the most clearly distinguished
339 and investigated in detail.

340 During the O₃ episodes, both O₃ and PAN concentrations reached their maximum values at
341 night. In episode 1 (Oct. 31 to Nov. 2), the O₃ concentration was increased to 80.6 ppbv, with
342 a high SO₂ concentration under strong wind. It was a typical Beijing plume observed in the
343 study region. In comparison, NO₂ was greatly increased in episode 2 (Oct. 28–29) when the
344 air masses were affected by urban emissions from Korean Peninsula. Although the maximum
345 O₃ level was lower during episode 2, these two cases demonstrated well how O₃ was exported
346 from the East Asian continent.

347 The remaining two episodes were highlighted by enhanced PAN concentrations and
348 characterized by haze occurrence. During episode 3 (Oct. 20–21), PAN and O₃ concentrations
349 increased up to 2.0 ppbv and 78.9 ppbv, respectively, with high NO_x levels, probably
350 influenced by emissions from Korea. Episode 4 (Nov. 4–5) was characterized by the highest
351 concentrations of almost all measured species, including PAN, O₃, PM_{2.5} mass, and PM_{1.0}
352 species; the maximum recorded concentrations of PAN, O₃, and PM_{2.5} mass during this
353 interval were 2.4 ppbv, 87.5 ppbv, and 156 $\mu\text{g}/\text{m}^3$, respectively. Note that, along with PM_{2.5}
354 and O₃, PAN was gradually increased through the night. In this episode, an air mass was
355 slowly transported from eastern China. With depleted NO, the effective lifetime of PAN was
356 greatly extended. In addition, PAN concentration showed good correlation with OC, EC, and
357 K⁺; in fact, the correlation of PAN with K⁺ was comparable to that of OC with K⁺. These
358 results, in conjunction with the high $\Delta\text{OC}/\text{EC}$ (7), imply that the observed haze was mainly
359 caused by the emissions produced by biomass combustion. These results suggest that PAN is

360 a useful tool for distinguishing continental outflows that were typically observed in northeast
361 Asia.

362 The comparison between the measured and calculated concentrations using the CAM-
363 Chem-HTAP2 model showed that the model underestimated the O₃ and PAN levels in
364 Chinese outflows, particularly for haze incidence. These results reveal that Chinese outflows
365 are still poorly understood and not well captured in the model.

366

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664 **Tables**

665

666 Table 1. Chemical and meteorological characteristics of the four episodes.

	Episode 1	Episode 2	Episode 3	Episode 4
Period	Oct.31 ~ Nov.2	Oct. 28~29	Oct. 20~21	Nov. 4~5
Type	Transport dominant	Transport dominant	Chemical transformation	Chemical transformation
Event	O ₃ export	O ₃ export	Haze	Haze
O ₃ (ppbv)	60.2 (80.6)	45.6 (62.8)	59.7 (78.9)	61.8 (87.5)
PAN (ppbv)	0.5 (0.9)	0.5 (0.8)	1.2 (2.0)	1.3 (2.4)
PM _{2.5} (µg/m ³)	34 (62)	23 (36)	50 (76)	77 (156)
SO ₂ (ppbv)	4.3 (12.9)	2.0 (4.4)	2.6 (5.4)	4.4 (9.5)
NO ₂ (ppbv)	3.7 (7.3)	6.2 (12.1)	6.2 (12.7)	6.1 (9.9)
Wind Speed (m/s)	13.5 (16.0)	9.5 (16.1)	6.6 (10.2)	5.0 (7.7)

667 *Measurements are given for the average with the maximum in the parenthesis.

668 **Figure Captions**

669

670 Figure 1. Temporal variations (against local time) of measured species, including PAN, PM_{2.5},
671 O₃, NO₂, NO, SO₂, and, CO, and meteorological parameters, including relative
672 humidity, temperature, and wind speed in fall 2010. Episodes 1–4, described in the
673 main text, are shaded in blue and yellow.

674 Figure 2. Diurnal variations in the concentrations of O₃, NO₂, PAN, and PM_{2.5}, measured at
675 GCO in the fall of 2010 (5 min data of O₃, NO₂, 2 min data of PAN, and 1 h data of
676 PM_{2.5}).

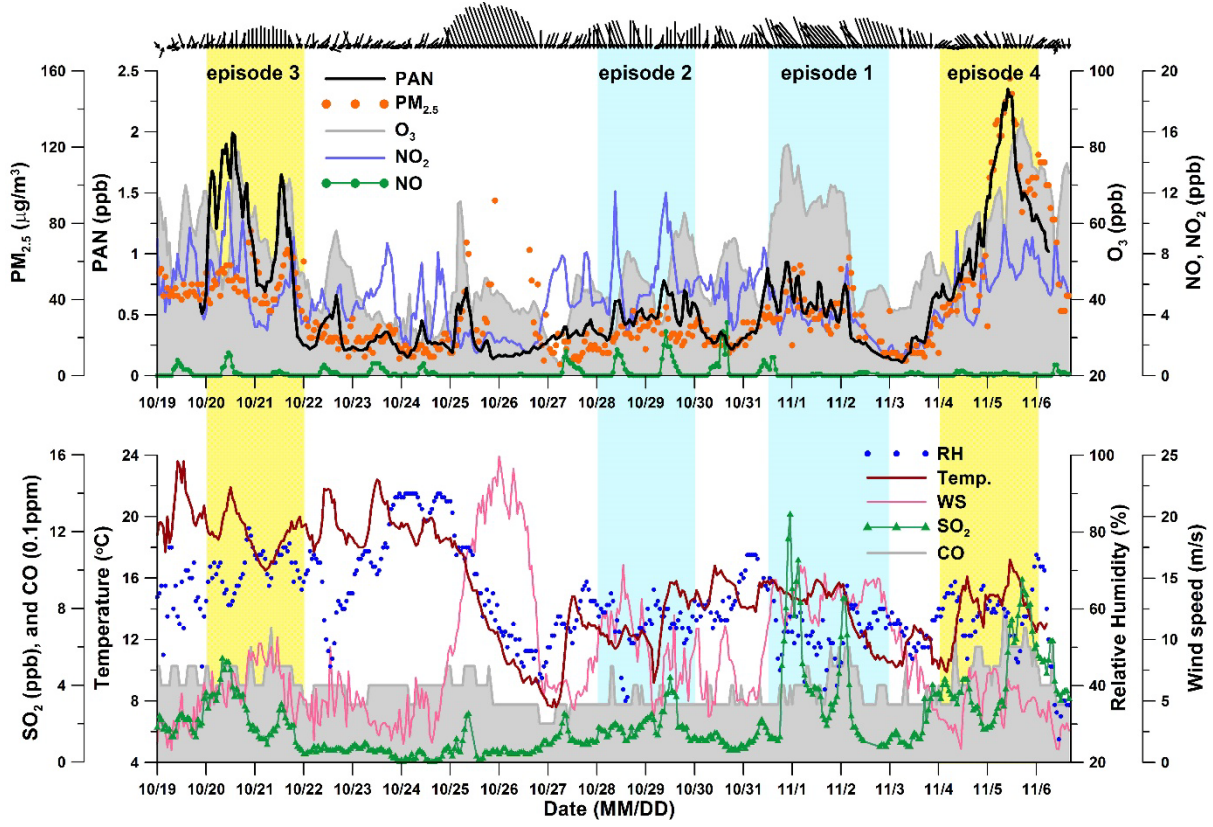
677 Figure 3. Comparison of O₃ with the PAN daily maxima. The time when the daily maximum
678 appears is classified as daytime (08–18 h) and nighttime (the rest) based on the
679 time of sunrise and sunset. Numerals indicate the days.

680 Figure 4. The three-day NOAA HYSPLIT backward trajectories of air masses for every one
681 hour observed at GCO during episode 1 (Oct. 31 to Nov. 2), episode 2 (Oct. 28–29),
682 episode 3 (Oct. 20–21), and episode 4 (Nov. 4–5). They are colored according to
683 the level of (a) PAN, (b) O₃, (c) NO₂, and (d) PM_{2.5} at GCO at the time of the
684 trajectory initialization. The trajectories north of 50°N are not shown. For these
685 horizontal trajectories, (e) vertical heights are given.

686 Figure 5. Correlations between (a) PAN and O₃ and (b) PAN and NO₂ with linear regression
687 line for each episode. Correlations between O₃ and PAN were color coded by the
688 level of (c) NO₂ and (d) PM_{2.5}.

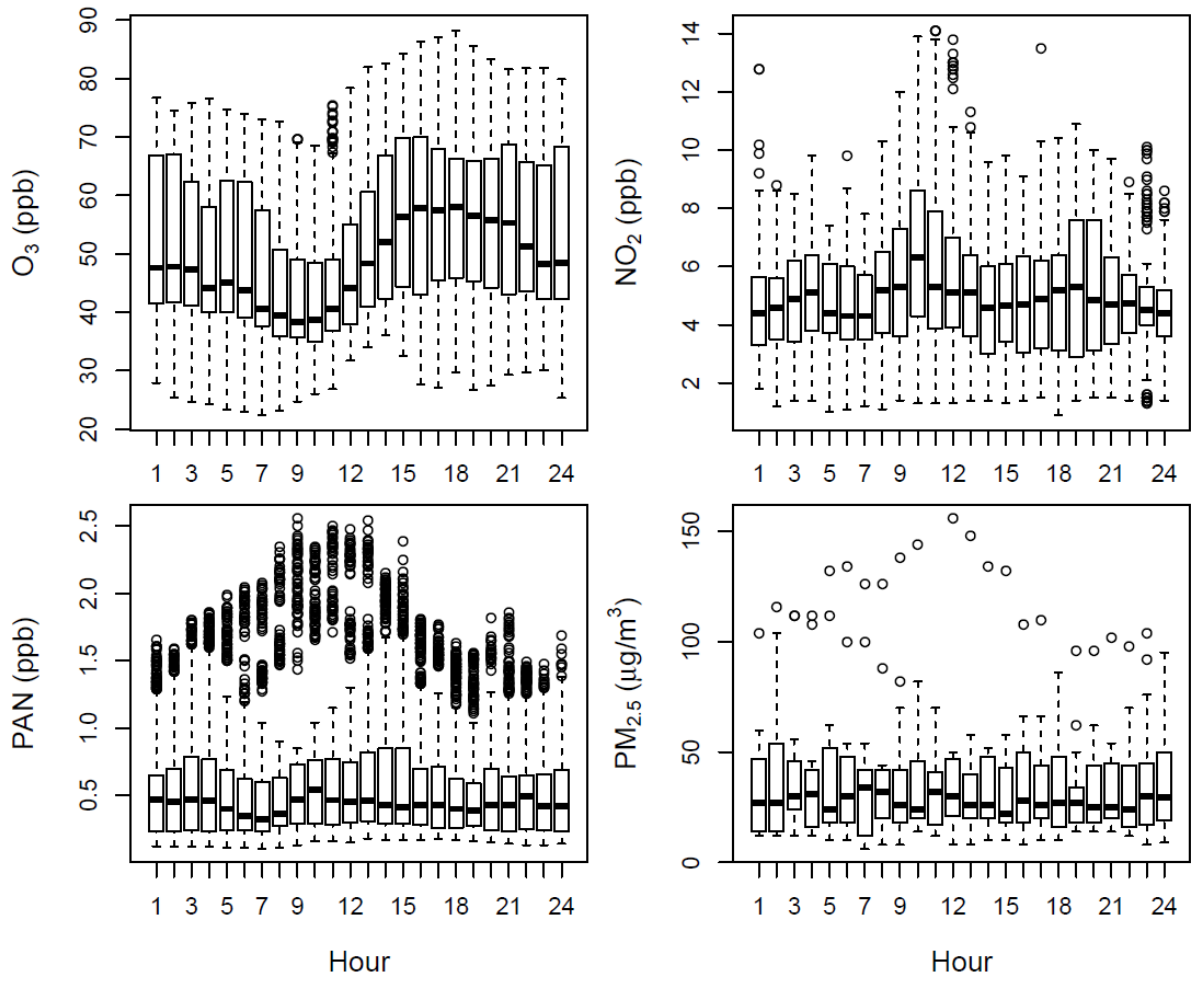
689 Figure 6. Correlations among PAN, K⁺ ion of PM_{1.0}, and carbonaceous components of PM_{2.5}
690 for three cases: (a) PAN and PM_{2.5} mass, (b) PAN and PM_{2.5} OC, (d) PM_{2.5} EC and
691 OC, and (d) PAN and PM_{1.0} K⁺. The lines represent the linear regression for each
692 episode.

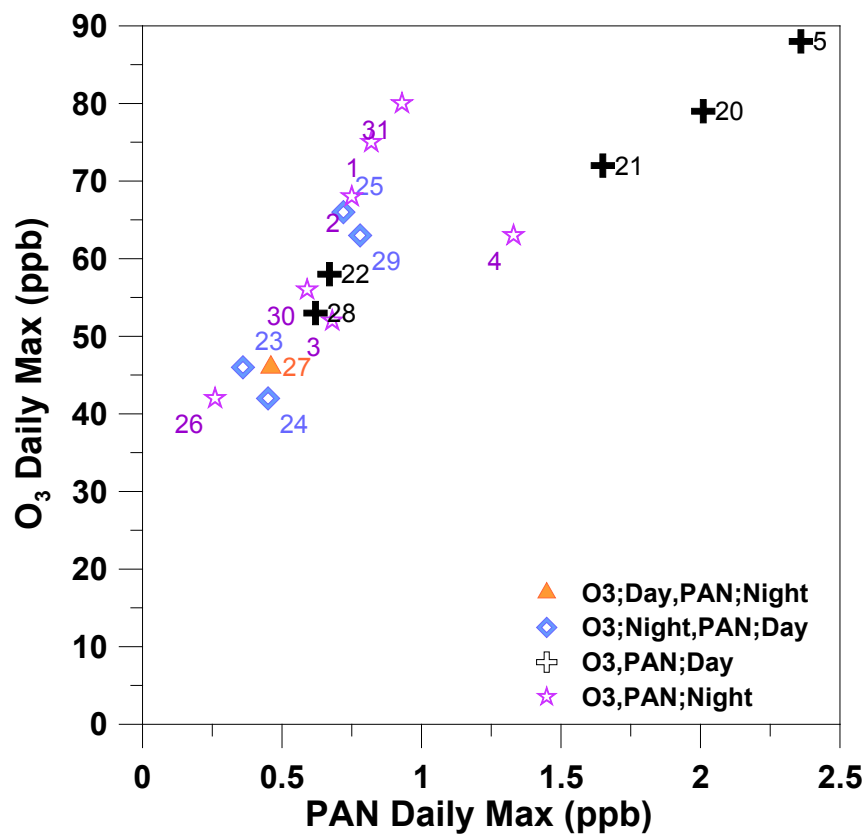
693 Figure 7. Comparison between the observed and calculated (a) PAN and (b) O₃
694 concentrations by CAM-chem model. Time is given in local time and four episodes
695 are shaded.



696

697 Figure 1.

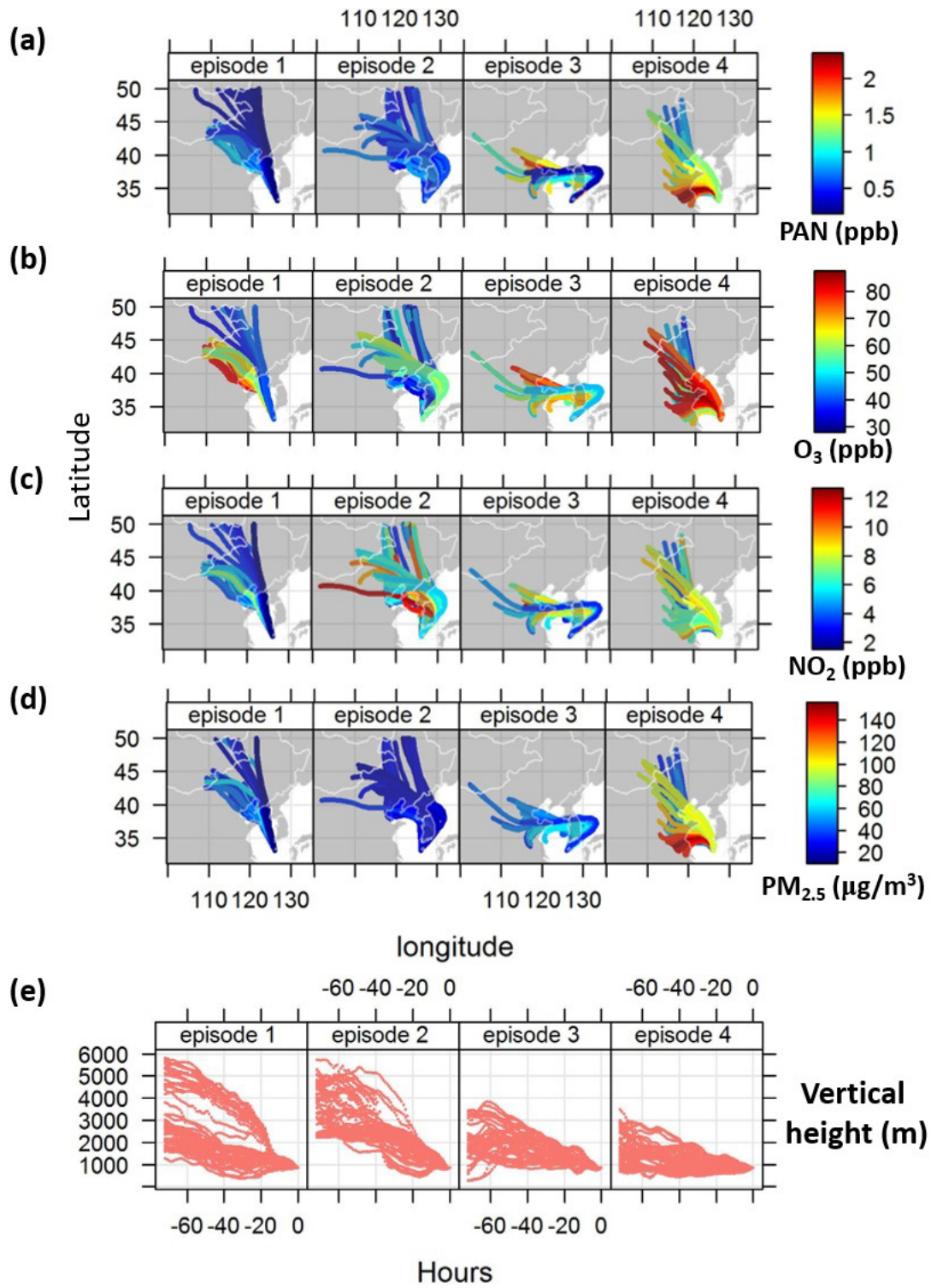




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702 Figure 3.

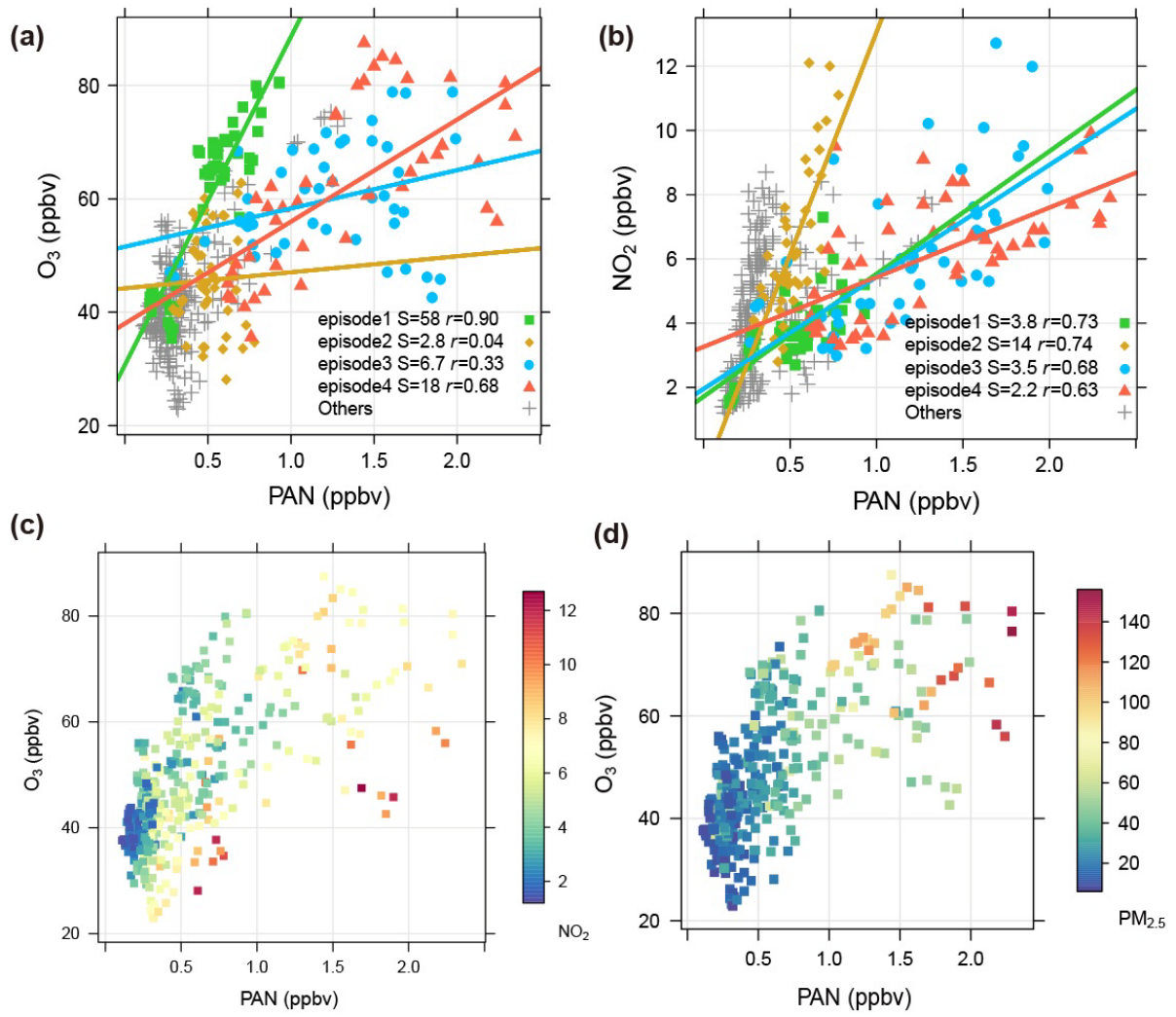
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704 Figure 4.

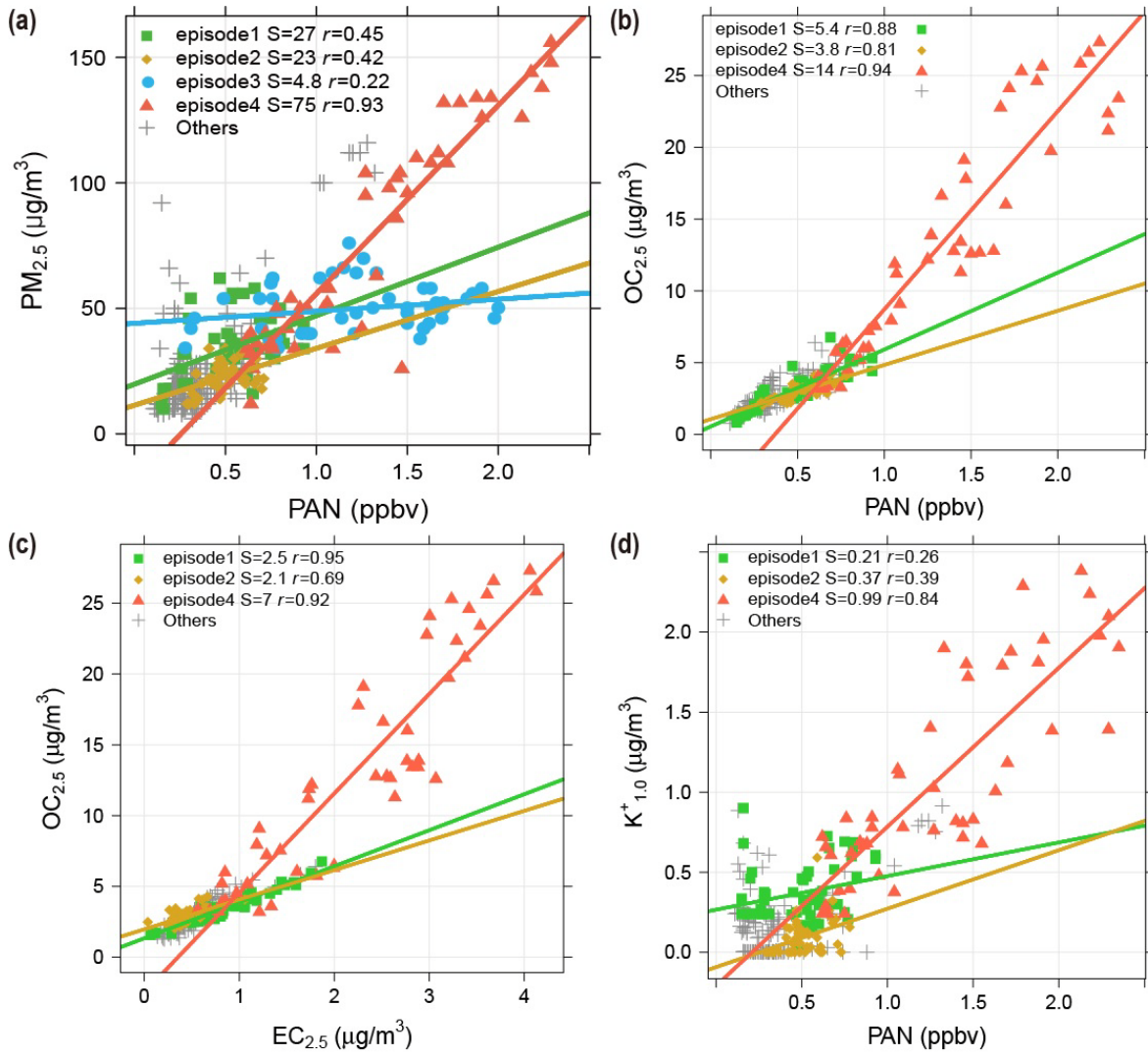
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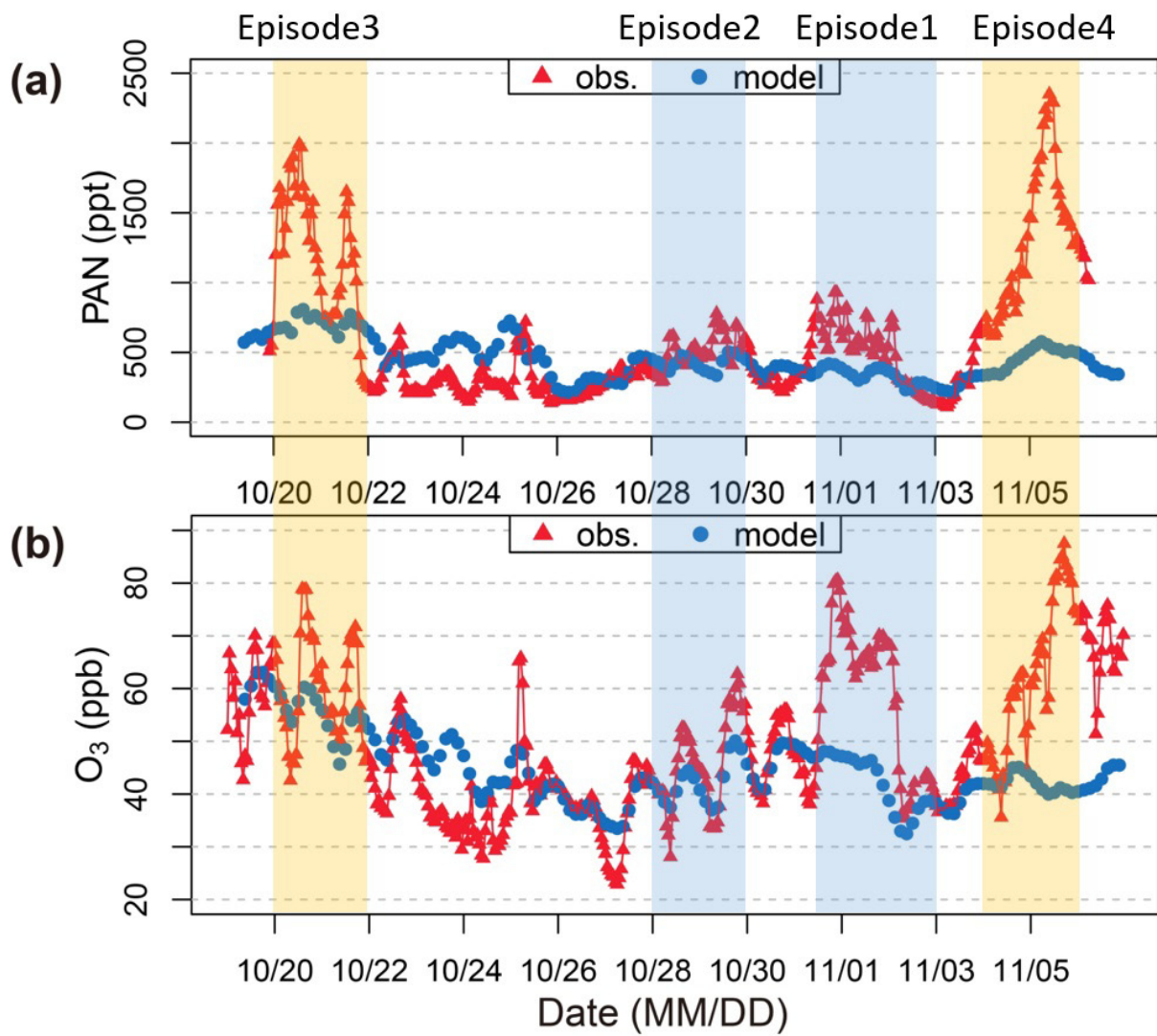
707 Figure 5.

708



709

710 Figure 6.



711

712 Figure 7.